

ENZYMATIC METHOD OF MAKING ALDEHYDES FROM FATTY ACIDS

BACKGROUND OF THE INVENTION

Field of the Invention

[0001] The present invention is directed to enzymatic methods of preparing aldehydes from fatty acids.

Description of Related Art

[0002] Microorganism-produced enzymes are widely used as a class of biocatalytic reagents in synthetic organic chemistry in a variety of reactions including, e.g., oxidations, reductions, hydrolyses, and carbon-carbon bond ligations. For example, enzyme reactions catalyzed by esterases may be used either hydrolytically or to synthesize esters, depending on whether the reaction medium is aqueous or organic in composition.

[0003] Biocatalysts are valued for their intrinsic abilities to bind organic substrates and to catalyze highly specific and selective reactions under the mildest of reaction conditions. These selectivities and specificities are realized because of highly rigid interactions occurring between the enzyme active site and the substrate molecule. Biocatalytic reactions are particularly useful when they are used to overcome difficulties encountered in catalysis achieved by the use of traditional chemical approaches.

[0004] The reduction of carboxylic acids by microorganisms is a relatively new biocatalytic reaction that has not yet been widely examined or exploited. Jezo and Zemek reported the reduction of aromatic acids to their corresponding benzaldehyde derivatives by *Actinomyces* in *Chem. Papers* 40(2):279-281 (1986). Kato *et al.* reported the reduction of benzoate to benzyl alcohol by *Nocardia asteroides* JCM 3016 (*Agric. Biol. Chem.* 52(7):1885-1886 (1988)), and Tsuda *et al.* described the reduction of 2-aryloxyacetic acids (*Agric. Biol. Chem.* 48(5):1373-1374 (1984)) and

arylpropionates (*Chem. Pharm. Bull.* 33(11):4657-4661 (1985)) by species of *Glomerella* and *Gloeosporium*. Microbial reductions of aromatic carboxylic acids, typically to their corresponding alcohols, have also been observed with whole cell biotransformations by *Clostridium thermoaceticum* (White *et al.*, *Eur. J. Biochem.* 184:89-96 (1989)) and by *Neurospora* (Bachman *et al.*, *Arch. Biochem. Biophys.* 91:326 (1960)). More recently, carboxylic acid reduction reactions have reportedly been catalyzed by whole cell preparations of *Aspergillus niger*, *Corynespora melonis* and *Coriolus* (Arfmann *et al.*, *Z. Naturforsch* 48c:52-57 (1993); *cf.*, Raman *et al.*, *J. Bacterial* 84:1340-1341 (1962)), and by *Nocardia* (Chen and Rosazza, *Appl. Environ. Microbiol.* 60(4):1292-1296 (1994)).

[0005] Carboxylic acid reductases are complex, multicomponent enzyme systems requiring the initial activation of carboxylic acids via formation of acyl-AMP and often coenzyme A intermediates (see, e.g., Hempel *et al.*, *Protein Sci.* 2:1890-1900 (1993)). The enzymatic reaction offers significant advantages over existing methods used in chemical reductions of carboxylic acids or their derivatives. The carboxylic acid reduction reaction appears to bear the usual desirable features of functional group specificity, and it also functions well under mild reaction conditions and produces a high yield of product. The reduction of the activated carboxylic acid intermediate occurs step-wise to give aldehyde, and then alcohol, products (Gross *et al.*, *Eur. J. Biochem.* 8:413-419; 420-425 (1969); Gross, *Eur. J. Biochem.* 31:585-592 (1972)).

[0006] Fatty alcohols and aldehydes can be commercially produced using one of several methods. However, one of the most widely used commercial methods, catalytic hydrogenation of fatty acids and methyl esters from fats and oils, produces fatty alcohols using high pressures (typically, 25,000-30,000 kP) and high temperatures (typically 250-300 °C). See, e.g., "Fatty Acids and Derivatives from Coconut Oil," in *Bailey's Industrial Oil and Fat Products*, 5th ed., Volume 5, Y.H. Hui, ed., John Wiley & Sons, Inc. (1996). The high pressures and temperatures employed in these methods are harsh in comparison to the milder conditions common with enzymatic reactions and can lead to unwanted side reactions, such as isomerization of one or more double bonds present in unsaturated fatty acid starting materials.

- [0007] There continues to exist a need for improved methods of producing fatty alcohols and aldehydes with improved reaction specificity that lead to higher yields and fewer side reactions.

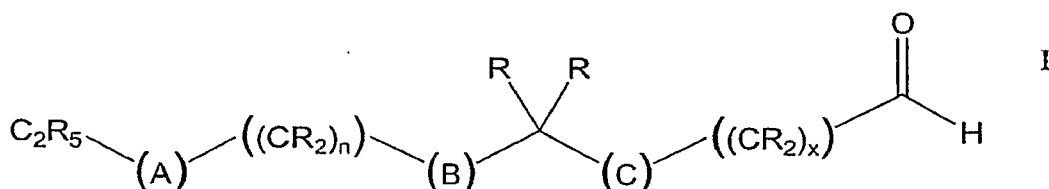
SUMMARY OF THE INVENTION

- [0008] The present invention provides enzymatic methods of preparing aldehydes from fatty acids by utilizing a carboxylic acid reductase enzyme to reduce the fatty acids to their corresponding aldehydes. The present invention also provides aldehydes prepared by the methods of the invention. The carboxylic acid reductase reduces the carboxyl moiety of the fatty acid to an aldehyde moiety which a separate aldehyde reductase can then further reduce to an alcohol moiety (Tao & Rosazza, *J. Indust. Microbiol. Biotechnol.* 25: 328-332 (2000)).
- [0009] Thus, the present invention is directed to a method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material derived from a vegetable oil or animal oil, the method comprising contacting the fatty (C₆-C₃₂)alkyl carboxylic acid material with a carboxylic acid reductase, wherein the (C₆-C₃₂)aldehyde is prepared.
- [0010] The present invention is also directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the vegetable oil is selected from the group consisting of soybean oil, linseed oil, sunflower oil, castor oil, corn oil, canola oil, rapeseed oil, palm kernel oil, cottonseed oil, peanut oil, coconut oil, palm oil, tung oil, safflower oil and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof.
- [0011] The present invention is further directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises a fatty (C₆-C₃₂)alkyl carboxylic acid, and wherein the fatty (C₆-C₃₂)alkyl carboxylic acid is unsaturated. The invention is also directed to the method described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more (C₁₆-C₁₈)alkyl fatty acids.

- [0012] The present invention is also directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more fatty (C₆-C₃₂)alkyl carboxylic acids independently selected from the group consisting of palmitoleic acid, oleic acid, linoleic acid, linolenic acid, eleostearic acid, ricinoleic acid, arachidonic acid, cetoleic acid, eicosapentaenoic acid, docosahexaenoic acid, erucic acid, and derivatives and mixtures thereof.
- [0013] The present invention is also directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has a hydrophilic/lipophilic balance of less than about 3.5.
- [0014] The present invention is also directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has a Hansen polarity value of less than about 4 ($\Delta/\sqrt{\text{MPa}}$) and a Hansen hydrogen bonding value of less than about 8 ($\Delta/\sqrt{\text{MPa}}$).
- [0015] The present invention is also directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has an iodine value greater than about 50.
- [0016] The present invention is directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the reductase is isolated from a *Nocardia* species, a *Neurospora* species, or a *Clostridium* species. The invention is also directed to the method of preparing the (C₆-C₃₂)aldehyde, wherein the reductase is isolated from a *Nocardia* species, including *Nocardia* sp. NRRL 5646. The invention is also directed to the method wherein the reductase is a recombinant reductase.
- [0017] The present invention is directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the method further comprises incubating the fatty (C₆-C₃₂)alkyl carboxylic acid material and the carboxylic acid reductase for a period of time sufficient to reduce at least about 50% of the carboxylic

acid material to the (C₆-C₃₂)aldehyde. The invention is further directed to the method, further comprising separating the (C₆-C₃₂)aldehyde from the fatty (C₆-C₃₂)alkyl carboxylic acid material.

[0018] The present invention is directed to the method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material as described herein, wherein the (C₆-C₃₂)aldehyde has Formula I:



wherein, in each instance, A, B, C, R, n, and x are as defined below.

[0019] The present invention is also directed to a or (C₆-C₃₂)aldehyde made according to the method described herein, wherein the (C₆-C₃₂)aldehyde has Formula I, wherein A, B, C, R, n, and x are as defined below.

DESCRIPTION OF THE FIGURES

[0020] Figures 1A-1D present the results of thin layer chromatography (TLC) analysis of bacterial culture aliquots sampled at various times during the course of the *in vivo* reduction experiments described in Example 1. Key: Lane 1: organism control; Lane 2: oleic acid standard; Lane 3: 4h sample; Lane 4: 8h sample; Lane 5: 24h sample; Lane 6: 24h sample + oleic acid control; Lane 7: octadecanol control.

[0021] Figure 1A presents the results of TLC analysis of bacterial culture aliquots sampled at 4 hours (Lane 3), 8 hours (Lane 4), or 24 hours (Lane 5) after the addition of oleic acid.

[0022] Figure 1B presents the results for a 24 hour sample (Lane 5 and Lane 6) with oleic acid (Lane 2) and octadecanol (Lane 7) standards. TLC plates were visualized by staining with phosphomolybdic acid.

[0023] Figure 1C presents the results for a 24 hour sample (Lane 6) with oleic acid (Lane 2) and octadecanol (Lane 7) controls. TLC plates were visualized by staining with phosphomolybdic acid.

- [0024] Figure 1D presents the TLC results for a 24 hour sample (Lane 5), with oleic acid (Lane 2) and octadecanol (Lane 7) standards. The TLC plate was visualized by staining with p-anisaldehyde.
- [0025] Figures 2A and 2B present ^1H and ^{13}C NMR spectra for Fraction 1 ($R_f = 0.74$), as described in Example 1. Figure 2A presents the ^1H NMR spectrum for Fraction 1, while Figure 2B presents the ^{13}C spectrum for Fraction 1.
- [0026] Figure 3 presents the results of TLC analysis of the reaction mixture taken during the course of the *in vitro* reduction experiment described in Example 1. An aliquot of the reaction mixture was taken 1 hour after incubation of oleic acid with purified recombinant CAR and analyzed by TLC. Key: Lane 2: oleic acid standard; Lane 6: reaction mixture + oleic acid standard; Lane 5: reaction mixture.

DETAILED DESCRIPTION OF THE INVENTION

- [0027] The present invention is directed to enzymatic methods of preparing aldehydes from their corresponding fatty acids by utilizing a carboxylic acid reductase.
- [0028] In the methods of the present invention, a fatty acid is contacted with a carboxylic acid reductase, which enzymatically converts the fatty acid into its corresponding aldehyde. The carboxylic acid reductase reduces the carboxyl moiety of the fatty acid to an aldehyde moiety, which can be further reduced to an alcohol moiety. The fatty aldehydes produced by the methods of the invention and fatty alcohols produced therefrom can be used in a variety of applications. See, e.g., International Publ. No. WO 2006/012917 A1, which describes the use of fatty alcohols as plasticizers. Other applications include incorporation into existing surfactants to produce surfactants with better low temperature performance, resins, dyes, organic acids, perfumes, flavorings, adhesives, papermaking, chemical intermediates, solvents, agricultural chemicals, cosmetic and personal care products, waxes, lubricants, and additives for polymers and coatings with both hydrolytic stability and reactivity.

- [0029] Thus, the present invention is directed to a method of preparing an aldehyde from a corresponding fatty acid or fatty acid material, the method comprising contacting the fatty acid or fatty acid material with a carboxylic acid reductase, wherein the aldehyde is prepared. The fatty acid or fatty acid material is typically derived from a vegetable oil or an animal oil. The fatty acid or fatty acid material also comprises one or more fatty acids that can be saturated or unsaturated.
- [0030] Enzymatic conversion of fatty acids to alcohols and/or aldehydes provides a number of advantages over traditional synthetic methods of producing alcohols and aldehydes from fatty acids, such as high degree of specificity and selectivity and mild reaction conditions.
- [0031] A particular advantage of the enzymatic methods of the present invention is that corresponding alcohols and/or aldehydes can be produced from fatty acids having one or more double bonds along the hydrocarbon backbone (unsaturated fatty acids) or from fatty acid derivatives, *i.e.*, fatty acids bearing functional groups such as, *e.g.*, a hydroxyl group, an amine group, or an epoxide ring attached to the hydrocarbon backbone.
- [0032] For example, the enzymatic methods of the present invention can be used to convert a fatty acid having an epoxide group along the backbone to the corresponding alcohol and/or aldehyde containing the epoxide group at the same position along the backbone. Reactions such as these can be difficult to achieve using conventional commercial methods of converting fatty acids to alcohols.
- [0033] As another example, the methods of the present invention can be used to convert an unsaturated fatty acid having one or more *cis* double bonds along its hydrocarbon chain to the corresponding unsaturated *cis* aldehyde. The aldehyde product retains the same stereochemistry at each double bond as the unsaturated fatty acid starting material, and can be further converted to an alcohol using an enzyme or other suitable catalyst.
- [0034] Current methods of producing alcohols from their corresponding fatty acids can lead to isomerization of one or more double bonds in unsaturated fatty acids, resulting in contamination of the corresponding alcohol and/or aldehyde with unwanted geometric isomers. A commercially produced sample of oleic acid was recently analyzed using gas

chromatography with mass spectroscopy to determine the presence and composition of geometric isomers of oleyl alcohol (C18:1 9c), a C18 alcohol containing one *cis* double bond in its hydrocarbon backbone. In addition to containing small percentages of C12 alcohols (0.26% w/w), C14 alcohols (0.34% w/w), and C16 alcohols (3.74% w/w), the C18 oleyl alcohol (91.75% w/w) was found to contain significant levels of the corresponding *trans* isomer in the following proportions:

Table 1

Isomer	% (w/w)
C18:1 <i>cis</i> isomer	77.75
C18:1 <i>trans</i> isomer	14.02

[0035] Use of the methods of the present invention make possible the production of a specific *cis* or *trans* isomer of an aldehyde from the corresponding fatty acid that is uncontaminated with the corresponding unwanted geometric isomer. The carboxylic acid reductase reduces the carboxyl moiety of the fatty acid to an aldehyde moiety which a separate aldehyde reductase can then further reduce to an alcohol moiety.

[0036] The term "fatty acid" as used herein refers to a saturated or unsaturated aliphatic carboxylic acid having a hydrocarbon backbone from about 4 to about 32 carbon atoms long. The term "medium chain fatty acid" as used herein refers to a saturated or unsaturated aliphatic fatty acid, *i.e.*, aliphatic carboxylic acid, having a hydrocarbon backbone from about 10 to about 22 carbon atoms long. The term "fatty (C₆-C₃₂)alkyl carboxylic acid" as used herein refers to a saturated or unsaturated aliphatic fatty acid, *i.e.*, aliphatic carboxylic acid, having a hydrocarbon backbone from 6 to 32 carbon atoms long. The aliphatic hydrocarbon chain of the fatty acid can be linear or branched, but is typically linear. Examples of fatty acids useful in the present invention include, but are not limited to, palmitoleic acid, oleic acid, linoleic acid, linolenic acid, eleostearic acid, ricinoleic acid, arachidonic acid, cetoleic acid, eicosapentaenoic acid, docosahexaenoic acid, and erucic acid. The term "fatty acid" is also meant to encompass

derivatives of fatty acids, including, e.g., conjugated derivatives and genetically modified derivatives of fatty acids. Examples of fatty acid derivatives useful in the present invention include, but are not limited to, fatty acids having functional groups capable of undergoing reduction, such as fatty acids having internal aldehydes, and epoxides and hydroxides of fatty acids. Such fatty acids having functional bonds capable of undergoing reduction, such as double bonds, may conveniently be sourced from genetically modified plants, such as genetically modified soybean, canola, sunflower, or other plants. Alternatively, fatty acids from milk or tissue from genetically modified animals, such as *Bos taurus*, may be useful.

[0037] The term "about," as used herein, applies to all numeric values, whether or not explicitly indicated. The term "about" generally refers to a range of numbers that one skilled in the art would consider equivalent to the recited value (i.e., having the same function or result) and may include numbers that are rounded to the nearest significant figure.

[0038] The term "fatty acid material" or "fatty (C₆-C₃₂)alkyl carboxylic acid material" as used herein refers to a composition comprising one or more fatty acids or derivatives thereof, or one or more (C₆-C₃₂)alkyl fatty acids or derivatives thereof, respectively, and typically includes mixtures of fatty acids or their derivatives. Examples of fatty acid materials include, but are not limited to, mixtures of fatty acids derived from animal or vegetable oils.

[0039] Fatty acids or fatty acid materials suitable for use in the methods of the present invention are derived from animal oils or vegetable oils, or mixtures or derivatives thereof, including, for example, conjugated derivatives and genetically-modified derivatives.

[0040] For example, in some aspects of the invention, the fatty acid or fatty acid material is derived from a vegetable oil or derivative thereof. The fatty acid or fatty acid material can be derived from triglycerides (including, e.g., triglycerides of medium chain fatty acids), diglycerides, or monoglycerides of vegetable origin, or derivatives or mixtures thereof. In some embodiments, the fatty acid or fatty acid material is enriched in double bonds and is derived from a vegetable oil such as, e.g., cocoa butter, cocoa butter substitutes, illipe fat, kokum butter, mowrah fat, phulwara butter, sal fat, shea fat, borneo tallow, canola oil, castor oil, coconut oil, coriander oil,

corn oil, cottonseed oil, hazelnut oil, hempseed oil, jatropha oil, linseed oil, mango kernel oil, meadowfoam oil, mustard oil, neat's foot oil, olive oil, palm oil, palm kernel oil, peanut oil, rapeseed oil, rice bran oil, safflower oil, sasanqua oil, shea butter, soybean oil, sunflower seed oil, tall oil, tsubaki oil, tung oil, or another vegetable oil, or a derivative, conjugated derivative, genetically-modified derivative or mixture thereof. Thus, in some embodiments, the fatty acid or fatty acid material is derived from a vegetable oil selected from the group consisting of soybean oil, linseed oil, sunflower oil, castor oil, corn oil, canola oil, rapeseed oil, palm kernel oil, cottonseed oil, peanut oil, coconut oil, palm oil, tung oil, safflower oil and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof. In other embodiments, the fatty acid or fatty acid material is derived from triolein palm olein, palm stearin, palm kernel olein, palm kernel stearin, or a derivative, conjugated derivative, genetically-modified derivative or mixture thereof.

[0041] In other aspects of the invention, the fatty acid or fatty acid material is derived from an animal fat or oil or derivative thereof. Examples of animal fats and oils include, but are not limited to, butterfat, chicken fat, milk fat, lard, lanolin, beef tallow, mutton tallow, tallow, and egg yolk, and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof. In some embodiments, for example, the fatty acid or fatty acid material is derived from an animal fat or oil selected from the group consisting of butterfat, chicken fat, tallow, lard, and egg yolk, and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof. In other embodiments, the fatty acid or fatty acid material is derived from a marine or fish oil. Examples of marine or fish oils include, but are not limited to, mackerel oil, salmon oil, striped bass oil, rainbow trout oil, halibut oil, tuna oil, cod oil, cod liver oil, tilapia oil, menhaden oil, candlefish oil, orange roughy oil, pile herd oil, sardine oil, whale oil, herring oil, and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof.

[0042] In some aspects of the present invention, the fatty acid or fatty acid material comprises a fatty (C₆-C₃₂)alkyl carboxylic acid or a mixture of fatty (C₆-C₃₂)alkyl carboxylic acids. In other aspects, the fatty acid or fatty acid

material comprises a fatty (C₆-C₃₀)alkyl carboxylic acid or a mixture of fatty (C₆-C₃₀)alkyl carboxylic acids, or a fatty (C₆-C₂₂)alkyl carboxylic acid or a mixture of fatty (C₆-C₂₂)alkyl carboxylic acids. In yet other aspects of the present invention, the fatty acid or fatty acid material comprises a fatty (C₁₆-C₂₂)alkyl carboxylic acid or a mixture of fatty (C₁₆-C₂₂)alkyl carboxylic acids. In some aspects, the fatty acid or fatty acid material comprises a fatty (C₁₆-C₂₀)alkyl carboxylic acid or a mixture of fatty (C₁₆-C₂₀)alkyl carboxylic acids thereof, or a fatty (C₁₆-C₁₈)alkyl carboxylic acid or a mixture of fatty (C₁₆-C₁₈)alkyl carboxylic acids.

[0043] Thus, in some aspects, the present invention is directed to a method of preparing a (C₆-C₃₂)aldehyde from a corresponding fatty (C₆-C₃₂)alkyl carboxylic acid material derived from a vegetable oil or an animal oil, the method comprising contacting the fatty (C₆-C₃₂)alkyl carboxylic acid material with a carboxylic acid reductase; wherein the (C₆-C₃₂)aldehyde is prepared.

[0044] Fatty acids suitable for use in the present invention can be either saturated fatty acids or unsaturated fatty acids. Examples of suitable saturated fatty acids include, but are not limited to, palmitic acid, stearic acid, arachidic acid, behenic acid, and lignoceric acid.

[0045] Suitable unsaturated fatty acids include olefinically unsaturated fatty acids having one or more double bonds in the fatty acid hydrocarbon backbone. In some embodiments, the unsaturated fatty acids have one to six double bonds. In other embodiments, the unsaturated fatty acids have one to four double bonds, one to three double bonds, or one to two double bonds. Each double bond can have either the *cis* or *trans* configuration. In some embodiments, the *cis* configuration is preferred.

[0046] Examples of unsaturated fatty acids suitable for use in the present invention are described in U.S. Pat. No. 5,219,733 and include, but are not limited to, 4-decenoic acid, caproic acid, 4-dodecenoic acid, 5-dodecenoic acid, lauroic acid, 4-tetradecenoic acid, 5-tetradecenoic acid, 9-tetradecenoic acid, palmitoleic acid, 6-octadecenoic acid, oleic acid, 9-octadecenoic acid, 11-octadecenoic acid, 9-eicosenoic acid, *cis*-11-eicosenoic acid, cetoleic acid, 13-docosenoic acid, 15-tetracosenoic acid, 17-hexacosenoic acid, 6,9,12,15-hexadecatetraenoic acid, linoleic acid,

linolenic acid, α -eleostearic acid, β -eleostearic acid, punicic acid, 6,9,12,15-octadecatetraenoic acid, parinaric acid, 5,8,11,14-eicosatetraenoic acid, 5,8,11,14,17-eicosapentaenoic acid (EPA), 7,10,13,16,19-docosapentaenoic acid, 4,7,10,13,16,19-docosahexaenoic acid (DHA) and the like, and mixtures and derivatives thereof. Thus, in some embodiments of the invention, the fatty acid or fatty acid material comprises one or more unsaturated fatty acids selected from the group consisting of palmitoleic acid, oleic acid, linoleic acid, linolenic acid, eleostearic acid, ricinoleic acid, arachidonic acid, eicosapentaenoic acid, docosahexaenoic acid, cetoleic acid and erucic acid and mixtures and derivatives thereof. In a preferred embodiment, the fatty acid or fatty acid material comprises one or more unsaturated fatty acids selected from the group consisting of palmitoleic acid, oleic acid, linoleic acid, and linolenic acid, and mixtures and derivatives thereof.

[0047] Fatty acids suitable for use in the present invention also include derivatives of unsaturated or saturated fatty acids, e.g., including, but not limited to, hydroxy derivatives of fatty acids, or "hydroxy fatty acids." Examples of useful hydroxy fatty acids are described in U.S. Pat. No. 5,219,733 and include, but are not limited to, α -hydroxylauric acid, α -hydroxymyristic acid, α -hydroxypalmitic acid, α -hydroxystearic acid, ω -hydroxylauric acid, α -hydroxyarachidic acid, 9-hydroxy-12-octadecenoic acid, ricinoleic acid, α -hydroxybehenic acid, 9-hydroxy-trans-10,12-octadecadienic acid, kamolenic acid, ipurolic acid, 9,10-dihydroxystearic acid, 12-hydroxystearic acid and the like.

[0048] The fatty acids useful in the present invention also can be optionally substituted with one or more moieties selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl, phenyl(C₁₋₂₆)alkyl, and a dioxanone ring and an oxirane (epoxy, or epoxide) ring formed between two adjacent chain carbons.

[0049] Thus, in some aspects of the present invention, the fatty acid or fatty acid material comprises saturated or unsaturated (C₆-C₃₂)alkyl carboxylic

acids that are optionally substituted with one or more moieties selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl, phenyl(C₁₋₂₆)alkyl, and a dioxanone ring and an oxirane (epoxy, or epoxide) ring formed between two adjacent chain carbons.

[0050] In some embodiments, preferred fatty acids are optionally substituted with one or more moieties selected from the group consisting of a dioxanone or oxirane ring formed between two adjacent chain carbons. In other embodiments of the invention, preferred fatty acids are optionally substituted with one or more moieties selected from the group consisting of C₁₋₂₆ alkoxy, hydroxy, and hydroxy(C₁₋₂₆)alkyl, or selected from the group consisting of amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, and amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl. In yet other embodiments, preferred fatty acids are optionally substituted with one or more moieties selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, and C₃₋₂₆ alkenyl, or selected from the group consisting of C₆₋₁₀ aryl, heteroaryl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl, and phenyl(C₁₋₂₆)alkyl.

[0051] The fatty acids and fatty acid materials useful in the present invention can also be described by their physical characteristics. Useful parameters include solubility, iodine value, and, in particular, hydrophilic/lipophilic balance, since the fatty acids and fatty acid materials of the present invention are, in general, lipophilic and relatively non-water-soluble. For example, fatty acid materials preferred for use in the present invention have a hydrophilic/lipophilic balance of less than about 3.5. Values for hydrophilic/lipophilic balance can be calculated using the "Molecular Modeling Pro Plus" software (version 6.0.6, Norgwyn Montgomery Software Inc.) available from Chem SW, Fairfield, CA.

[0052] Preferred fatty acid materials also have iodine values greater than about 50. Iodine values are well-known measures of the number of double bonds present in fatty (C₆-C₃₂)alkyl carboxylic acids and can be determined according to Method Tg1a-67, or calculated according to Method Cd1c-85, both as recited in the "Official Methods and Recommended Practices of the American Oil Chemists' Society, Fifth Edition, Second Printing, AOCS

Press, Champaign, IL, USA, which is incorporated by reference herein in its entirety. For example, the iodine value of oleic acid is 89; the iodine value of linoleic acid is 173, and the iodine value of linolenic acid is 262. Due to their high degrees of unsaturation, the iodine values of arachidonic acid, eicosapentaenoic acid and docosahexaenoic acid are all greater than the iodine value of linoleic acid.

[0053] The fatty acids and fatty acid materials useful in the present invention can also be described by Hansen solubility parameters. Hansen solubility parameters are described in detail in "Hansen Solubility Parameters: A User's Handbook," by Charles M. Hansen (CRC Press, 1999), which is incorporated by reference herein in its entirety. Preferred alkyl carboxylic acid materials for use in the present invention have Hansen polarity values of less than about 4 ($\delta/\text{sqr}(\text{MPa})$), or less than about 3.5($\delta/\text{sqr}(\text{MPa})$), and Hansen hydrogen bonding values of less than about 8 ($\delta/\text{sqr}(\text{MPa})$), or less than about 7.6 ($\delta/\text{sqr}(\text{MPa})$).

[0054] For example, preferred alkyl carboxylic acid materials can include, but are not limited to, those having Hansen polarity values between about 4.0 and about 1.0 ($\delta/\text{sqr}(\text{MPa})$), or between about 4.0 and about 2.0 ($\delta/\text{sqr}(\text{MPa})$), or between about 3.5 and about 2.5 ($\delta/\text{sqr}(\text{MPa})$), and Hansen hydrogen bonding values between about 8.0 and about 5.0 ($\delta/\text{sqr}(\text{MPa})$), or between about 8.0 and about 6.0 ($\delta/\text{sqr}(\text{MPa})$). Especially preferred are those alkyl carboxylic acids or alkyl carboxylic acid materials having Hansen polarity values of less than about 3.5 ($\delta/\text{sqr}(\text{MPa})$) and Hansen hydrogen bonding values of less than about 7.6 ($\delta/\text{sqr}(\text{MPa})$). Hansen solubility parameters can be calculated using the program "Molecular Modeling Pro Plus" (version 6.0.6, Norgwyn Montgomery Software Inc., available from ChemSW, Inc.) based on values published in the Handbook of Solubility Parameters and Other Parameters by Allan F.M. Barton (CRC Press, 1983) for solvents obtained experimentally by Hansen.

[0055] The Hansen polarity values at 25 °C and Hansen hydrogen bonding values at 25 °C of several alkyl carboxylic acid materials suitable for use in the present invention are listed in Table 1.

Table 1. Hansen polarity values and Hansen hydrogen bonding values of several alkyl carboxylic acid materials.

Hansen Solubility Parameters (25 °C)	oleic acid	linoleic acid	linolenic acid	arachidonic acid	eicosa-pentaenoic acid	docosa-hexaenoic acid
Polarity	3.2384	3.31338	3.37669	3.13706	3.19388	2.97912
Hydrogen bonding	6.02164	6.51409	6.97265	7.07841	7.48207	7.53673

[0056] The Hansen 3-D solubility parameters in Table 1 were calculated using the program "Molecular Modeling Pro Plus" (version 6.0.6, Norgwyn Montgomery Software Inc., available from ChemSW, Inc.) based on values published in the Handbook of Solubility Parameters and Other Parameters by Allan F.M. Barton (CRC Press, 1983) for solvents obtained experimentally by Hansen.

[0057] The methods of the present invention utilize a carboxylic acid reductase enzyme to reduce fatty acids to their corresponding aldehydes. Examples of carboxylic acid reductase enzymes are those from *Nocardia* species, or those from certain *Neurospora* species (e.g., *Neurospora crassa* (Gross, G. & Zenk, M, *Eur. J. Biochem.* 8:420-425 (1969)) or from certain *Clostridium* species (e.g., *Clostridium thermoaceticum* (White, H., *et al.*, *Eur. J. Biochem* 184:89-96 (1989))). In some aspects of the invention, the carboxylic acid reductase is a prokaryotic carboxylic acid reductase from a *Nocardia* species, e.g., *Nocardia asteroides*. The purification and characterization of carboxylic acid reductase from *Nocardia asteroides* is described in Kato, N. *et al.*, *J. Agric. Biol. Chem.* 55:757-762 (1991). A preferred carboxylic acid reductase is the carboxylic acid reductase, or aryl aldehyde oxidoreductase, from *Nocardia* species strain NRRL 5646. This carboxylic acid reductase is described by Rosazza *et al.* in U.S. Pat. No. 6,261,814 B1 and in U.S. Pat. Appl. Publ. No. US 2004/0180400 A1. U.S. Pat. No. 6,261,814 B1 and U.S. Pat. Appl. Publ. No. US 2004/0180400 A1 are incorporated by reference herein in their entireties.

[0058] In some aspects of the present invention, the carboxylic acid reductase is present in its natural form. In some embodiments, the carboxylic acid reductase is isolated and purified prior to its use in the

methods of the present invention. For example, methods for purifying the *Nocardia* sp. NRRL 5646 carboxylic acid reductase are described in Rosazza *et al.*, U.S. Pat. No. 6,261,814 B1.

[0059] In other aspects, the carboxylic acid reductase is a recombinant carboxylic acid reductase. Examples of recombinant *Nocardia* sp. NRRL 5646 carboxylic acid reductase that are suitable for use in the present invention are described in Rosazza *et al.*, U.S. Pat. Appl. Publ. No. US 2004/0180400 A1. U.S. Pat. Appl. Publ. No. US 2004/0180400 A1 also provides methods of making the recombinant enzyme. In some embodiments, the recombinant carboxylic acid reductase is isolated and purified prior to its use in the methods of the present invention.

[0060] In some aspects, the methods disclosed herein for preparing an aldehyde from its corresponding fatty acid further comprise incubating the fatty acid or fatty material and the carboxylic acid reductase. In some embodiments of the invention, the fatty acid or fatty acid material is incubated with the carboxylic acid reductase for a period of time sufficient to reduce at least about 10%, about 20%, about 30%, about 40%, or about 50% of the fatty acid material to its corresponding aldehyde. In other embodiments, the fatty acid or fatty acid material is incubated with the carboxylic acid reductase for a period of time sufficient to reduce at least about 60%, about 70%, about 80%, about 90%, or about 95% of the fatty acid material to its corresponding aldehyde. In preferred embodiments, the fatty acid or fatty acid material is incubated with the carboxylic acid reductase for a period of time sufficient to reduce at least about 50% of the fatty acid material to its corresponding aldehyde.

[0061] Incubation can be performed under *in vitro* or *in vivo* conditions. For example, isolated carboxylic acid reductase can be used to convert fatty acids to their corresponding aldehydes according to the methods of the present invention by incubating the isolated enzyme with fatty acids under *in vitro* conditions. Examples of suitable reaction conditions for *in vitro* reactions are described in U.S. Pat. No. 6,261,814 B1 for reduction reactions of aryl carboxylic acids using isolated *Nocardia* sp. NRRL 5646 carboxylic acid reductase in aqueous systems, as well as in Example 1 below. Incubation can be carried out at temperatures ranging from about -25 °C to

about 70 °C for times ranging from about 30 minutes to several days with or without agitation. For *in vitro* reactions using aqueous buffered solutions, suitable buffers include phosphate buffer, Tris buffer, BIS TRIS buffer, ADA buffer, ACES buffer, PIPES buffer, MOPSO buffer, BES buffer, MOPS buffer, HEPES buffer, TES buffer, DIPSO buffer, and TRISMA buffer. In addition, a non-aqueous solvent may be contacted with the aqueous buffered solution. Examples of such non-aqueous solvents include, but are not limited to, pentane, hexane, heptane, octane, petroleum ether, and isomers and mixtures thereof.

[0062] In other aspects of the present invention, incubation of the carboxylic acid reductase and the fatty acid or fatty acid material is performed under *in vivo* conditions. Example 1 below describes conditions suitable for *in vivo* reactions using *Nocardia* sp. NRRL 5646 in whole cells to convert oleic acid into its corresponding aldehyde forms. Use of whole cells to carry out enzymatic reactions is known to those of skill in the art. For example, the use of bacterial cells to carry out enzymatic reactions is described in European Patent application EP0187525 A2, "Process for producing L-serine" (Ajinomoto Co., Inc, Tokyo, Japan), which describes the enzymatic conversion of 2-oxo-oxazolidine-4-carboxylic acid (OOC) to L-lysine by incubation of whole *Bacillus licheniformis* cells with OOC.

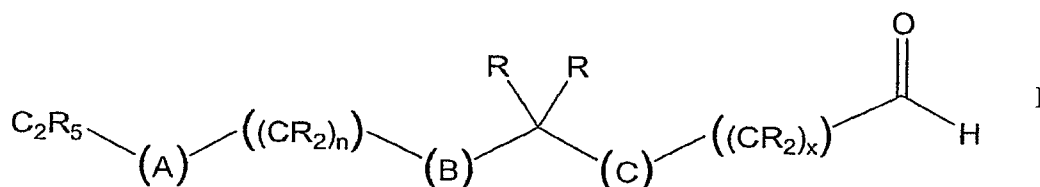
[0063] In some aspects, the methods disclosed herein for preparing an aldehyde from its corresponding fatty acid further comprise separating the aldehyde from the fatty acid or fatty acid material.

[0064] The aldehyde can be separated from the fatty acid or fatty acid material using methods known to those of skill in the art. Examples of suitable purification methods include, but are not limited to, distillation and chromatography, including simulated moving bed chromatography.

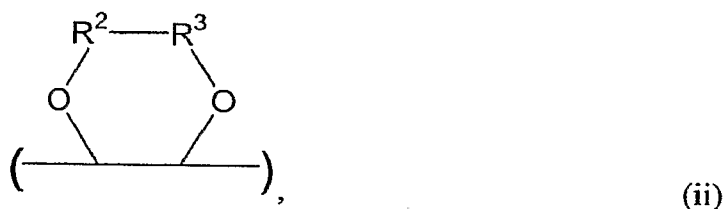
[0065] The aldehydes produced using the methods of the present invention have structures that correspond to the structures of the fatty acid. As described above, the aldehydes produced have hydrocarbon backbones from about 4 to 32 carbon atoms long and be saturated, or unsaturated, or optionally substituted with one or more moieties selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl,

heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl, phenyl(C₁₋₂₆)alkyl, and a dioxanone ring and an oxirane (epoxy or epoxide) ring formed between two adjacent chain carbons.

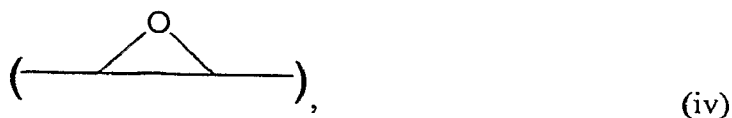
[0066] Thus, in some aspects of the present invention, the aldehydes produced using the methods of the present invention include, but are not limited to (C₆-C₃₂)aldehydes of Formula I:



wherein, in each instance, A, B and C are independently selected from the group consisting of:

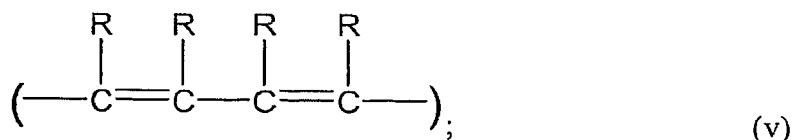


wherein, one of R² and R³ is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl and phenyl(C₁₋₂₆)alkyl;



and

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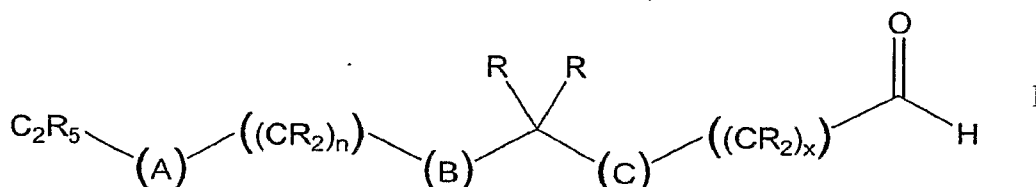
wherein, R is, in each instance, independently selected from the group consisting of: hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl and phenyl(C₁₋₂₆)alkyl;

x is an integer from 5 to 10; and

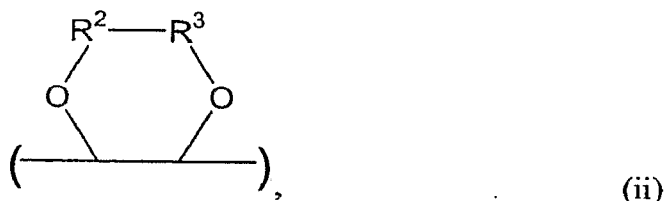
n is an integer from 1 to 6.

[0067] In some embodiments, preferred (C₆-C₃₂)aldehydes of Formula I are those wherein one of R² and R³ of (ii) is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, and C₁₋₂₆ alkoxy.

[0068] Thus, in some aspects of the present invention, the aldehydes produced using the methods of the present invention include, but are not limited to (C₆-C₃₂)aldehydes of Formula I:

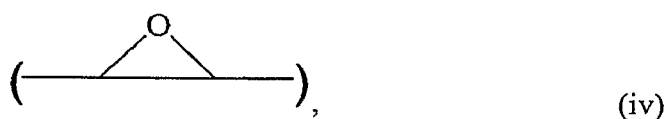
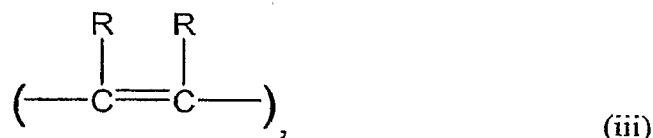


wherein, in each instance, A, B and C are independently selected from the group consisting of:

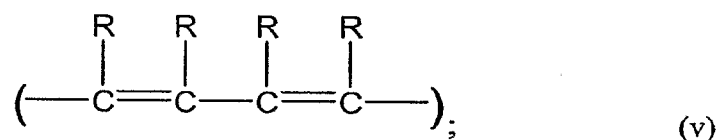


wherein, one of R² and R³ is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, and C₁₋₂₆ alkoxy;

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and

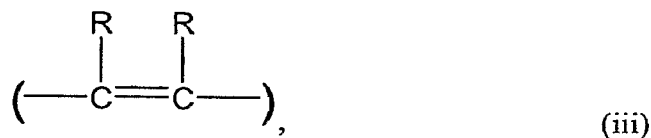


wherein, R is, in each instance, independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl and phenyl(C₁₋₂₆)alkyl;

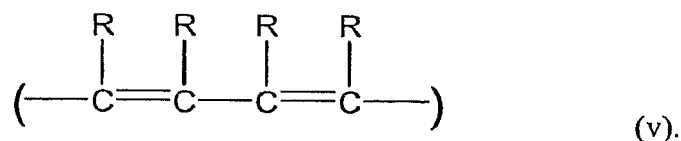
x is an integer from 5 to 10; and

n is an integer from 1 to 6.

[0069] In other embodiments, preferred (C₆-C₃₂)aldehydes of Formula I are those in which A, B, and C are independently

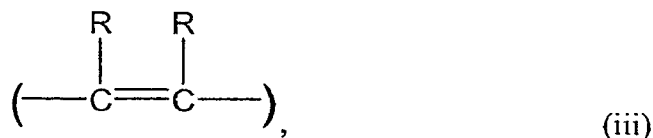


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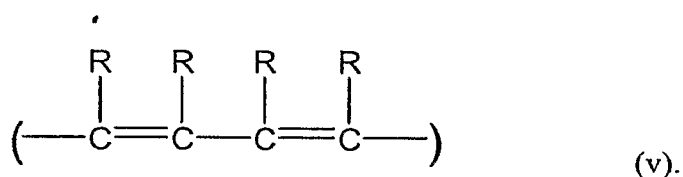


[0070] Especially preferred are those aldehydes in which A, B, and C are independently

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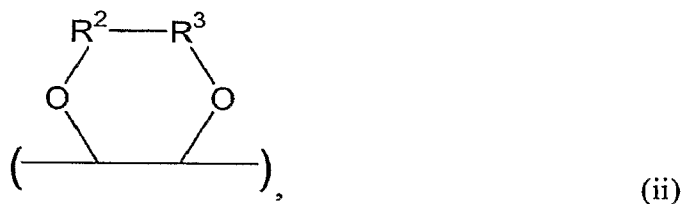


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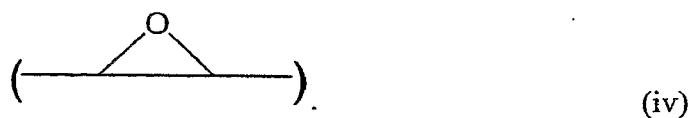


[0071] Aldehydes in which A, B, and C of Formula I are independently (iii) or (v) include all stereoisomers of the aldehydes, e.g., includes those aldehydes having either *cis* or *trans* stereochemistry about each double bond. Particularly preferred are those aldehydes having *cis* stereochemistry.

[0072] In yet other embodiments, preferred (C₆-C₃₂)aldehydes of Formula I are those in which A, B, and C are independently



wherein, one of R² and R³ is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino (C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl (C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl and phenyl (C₁₋₂₆)alkyl; or



[0073] In these embodiments, preferred (C₆-C₃₂)aldehydes of Formula I are those wherein one of R² and R³ of (ii) is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, and C₁₋₂₆ alkoxy.

- [0074] In some embodiments, preferred values of R are selected from the group consisting of C₁₋₂₆ alkoxy, hydroxy, and hydroxy(C₁₋₂₆)alkyl, or selected from the group consisting of amino (C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, and amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl. In yet other embodiments, preferred values of R are selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, and C₃₋₂₆ alkenyl, or selected from the group consisting of C₆₋₁₀ aryl, heteroaryl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl, and phenyl(C₁₋₂₆)alkyl.
- [0075] Preferred values of R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, and C₁₋₂₆ alkoxy.
- [0076] Preferred values of x are 5 to 9, 6 to 9, or 7 to 9. Preferred values of n are 1 to 5, 1 to 4, or 1 to 3.
- [0077] The following example is illustrative, but not limiting, of the methods and compositions of the present invention. Other suitable modifications and adaptations of the variety of conditions and parameters normally encountered and obvious to those skilled in the art are within the spirit and scope of the invention.

EXAMPLES

Example 1

- [0078] Recombinant carboxylic acid reductase (CAR) from *Nocardia* sp. NRRL 5646 was tested for its ability to reduce oleic acid under *in vivo* and *in vitro* conditions.

Materials and methods

- [0079] Chromatography: Organic solvents were obtained from Fisher scientific. Thin Layer Chromatography (TLC) was carried on Silica gel on polyester plates (0.25 mm Aldrich) or on Glass Plates (0.25 mm, Whatman). TLC plates were developed in solvent mixture of hexanes: ethylacetate: glacial acetic acid (75:25:0.1 v/v/v) and visualized by spraying with either p-anisaldehyde stain (0.5 mL of p-anisaldehyde and 0.5 mL of H₂SO in 60 mL methanol) or with phosphomolybdic acid (7% (w/v)) in ethanol followed by heating.

- [0080] Fermentation and screening procedures: *E. coli* BL21 Codon Plus[®] (DE3) RP/pPV 1.184 (*car* + *sfp*) cultures were prepared as described in US Pat. Appl. Publ. No. 2004/0180400 A1, which is incorporated by reference herein in its entirety.
- [0081] All solutions were prepared in distilled, deionized water. LB medium (1 L) contained Bacto tryptone (10 g), Bacto yeast extract (5 g), and NaCl (10 g). Ampicillin (Ap) stock solutions (100 mg/mL) were sterilized through 0.22- μ m membranes (Millipore). Chloramphenicol (Cm) (34 mg/mL) was prepared by dissolving in 95% ethanol.
- [0082] For *in vivo* reductions, *E. coli* BL21 Codon Plus[®] (DE3) RP/pPV 1.184 (*car* + *sfp*) cultures were initiated by inoculating a tiny crystal from the glycerol freeze into 10 mL LB medium containing Ap (100 μ g/mL) and Cm (34 μ g/mL) in a stainless steel-capped 25 mL Delong flask. The cultures were incubated with shaking at 250 rpm on New Brunswick Scientific G25 Gyrotory shaker for 16 h at 37°C. 10% inoculum derived from this overnight grown culture was inoculated in 200 mL LB medium containing Ap (100 μ g/mL) and Cm (34 μ g/mL). The cultures were incubated in 37°C, 250 rpm for 4 hours before receiving oleic acid (200 mg) as substrate. A control was prepared consisting of sterile medium incubated with the organism but without substrate.
- [0083] 5 mL samples were withdrawn from the cultures at various time intervals, acidified to pH 2, extracted with ethyl acetate (2 x 5 mL), and the organic layer separated. The organic layer was then removed, evaporated to dryness and reconstituted in 0.5 mL ethyl acetate. Samples were spotted onto TLC plates for analysis.
- [0084] For *in vitro* reductions, reactions were carried out in 50 mM Tris buffer (pH 7.5), 1mM EDTA, 10 mM MgCl₂, 1 mM DTT, 10% (v/v) glycerol, 5 mM ATP, 0.4 mM NADPH, and 5 mM oleic acid in a final volume of 50 mL. The oleic acid substrate was dissolved in 50 μ L of DMF and added to the stirred solution. The pH of the solution was adjusted to 7.5 before addition of substrate. The solution was stirred at 100 rpm at room temperature (rt) for 10 min before the enzyme was added. 1 mg of purified recombinant CAR (specific activity (sp. act.) 0.26 units/mg) was added to

this solution and incubated for 6 h. Purification of CAR was carried out as described in U.S. Pat. Appl. Publ. No. 2004/0180400 A1, which is incorporated by reference herein in its entirety.

[0085] 2 mL samples were withdrawn, acidified to pH 2, extracted with ethyl acetate (2 x 5 mL), and the organic layer separated. The organic layer was then removed, evaporated to dryness and reconstituted in 0.5 mL ethyl acetate. Samples were spotted onto TLC plates for analysis.

Results

[0086] *In vivo* reduction of oleic acid: TLC analysis of samples taken from various aliquots indicated that oleic acid was consumed over a 24 h period. Two new spots were also observed concurrently at $R_f = 0.74$ and 0.38 . (R_f for oleic acid = 0.56 ; R_f for octadecanol = 0.34 .) After 24h of growth, the culture was acidified with 6N HCl to pH 2 and extracted with ethyl acetate (3x 200 mL). The combined organic extract was washed with brine, dried over Na_2SO_4 and concentrated *in vacuo*. The extract (50 mg) was further purified by thin layer chromatography to yield 3 mg of Fraction 1 ($R_f = 0.74$), 10 mg of Fraction 2 ($R_f = 0.56$) and 3 mg of Fraction 3 ($R_f = 0.38$) (Figure 1). ^1H NMR and ^{13}C NMR analysis of the fraction of $R_f = 0.74$ (Fraction 1) showed a peak corresponding to an aldehyde moiety (Figure 2).

[0087] *In vitro* reduction of oleic acid: *In vitro* reduction of oleic acid by purified CAR gave a spot of $R_f = 0.78$ on TLC, but the reaction did not go to completion and the majority of the starting material was recovered (Figure 3). The poor reaction may have been due to the poor solubility of the starting material in the reaction mixture.

[0088] All documents, e.g., scientific publications, patents, patent applications and patent publications, recited herein are hereby incorporated by reference in their entirety to the same extent as if each individual document was specifically and individually indicated to be incorporated by reference in its entirety. Where the document cited only provides the first page of the document, the entire document is intended, including the remaining pages of the document.

WHAT IS CLAIMED IS:

1. A method of preparing a (C₆-C₃₂)aldehyde from a fatty (C₆-C₃₂)alkyl carboxylic acid material derived from a vegetable oil or an animal oil, the method comprising contacting the fatty (C₆-C₃₂)alkyl carboxylic acid material with a carboxylic acid reductase;

wherein the (C₆-C₃₂)aldehyde is prepared.

2. The method of claim 1, wherein the vegetable oil or animal oil is selected from the group consisting of butterfat, cocoa butter, cocoa butter substitutes, illipe fat, kokum butter, milk fat, mowrah fat, phulwara butter, sal fat, shea fat, borneo tallow, lard, lanolin, beef tallow, mutton tallow, tallow, animal fat, canola oil, castor oil, coconut oil, coriander oil, corn oil, cottonseed oil, hazelnut oil, hempseed oil, jatropha oil, linseed oil, mango kernel oil, meadowfoam oil, mustard oil, neat's foot oil, olive oil, palm oil, palm kernel oil, peanut oil, rapeseed oil, rice bran oil, safflower oil, sasanqua oil, shea butter, soybean oil, sunflower seed oil, tall oil, tsubaki oil, tung oil, vegetable oils, marine oils, menhaden oil, candlefish oil, cod-liver oil, orange roughy oil, pile herd oil, sardine oil, whale oils, herring oils, triglyceride, diglyceride, monoglyceride, triolein palm olein, palm stearin, palm kernel olein, palm kernel stearin, triglycerides of medium chain fatty acids, and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof.

3. The method of claim 2, wherein the vegetable oil is selected from the group consisting of soybean oil, linseed oil, sunflower oil, castor oil, corn oil, canola oil, rapeseed oil, palm kernel oil, cottonseed oil, peanut oil, coconut oil, palm oil, tung oil, safflower oil and derivatives, conjugated derivatives, genetically-modified derivatives and mixtures thereof.

4. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises a fatty (C₆-C₃₂)alkyl carboxylic acid.

5. The method of claim 4, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid is unsaturated.

6. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more (C₁₆-C₁₈)alkyl fatty acids.

7. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more fatty (C₆-C₃₂)alkyl carboxylic acids independently selected from the group consisting of palmitoleic acid, oleic acid, linoleic acid, linolenic acid, eleostearic acid, ricinoleic acid, arachidonic acid, cetoleic acid, eicosapentaenoic acid, docosahexaenoic acid, and erucic acid, and derivatives and mixtures thereof

8. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more fatty (C₆-C₃₂)alkyl carboxylic acids independently selected from the group consisting of 4-decenoic acid, caproic acid, 4-dodecenoic acid, 5-dodecenoic acid, lauroleic acid, 4-tetradecenoic acid, 5-tetradecenoic acid, 9-tetradecenoic acid, palmitoleic acid, 6-octadecenoic acid, oleic acid, 9-octadecenoic acid, 11-octadecenoic acid, 9-eicosenoic acid, cis-11-eicosenoic acid, cetoleic acid, 13-docosenoic acid, 15-tetracosenoic acid, 17-hexacosenoic acid, 6,9,12,15-hexadecatetraenoic acid, linoleic acid, linolenic acid, α -eleostearic acid, β -eleostearic acid, punicic acid, 6,9,12,15-octadecatetraenoic acid, parinaric acid, 5,8,11,14-eicosatetraenoic acid, 5,8,11,14,17-eicosapentaenoic acid (EPA), 7,10,13,16,19-docosapentaenoic acid, 4,7,10,13,16,19-docosahexaenoic acid (DHA), and derivatives and mixtures thereof.

9. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material comprises one or more fatty (C₆-C₃₂)alkyl carboxylic acids independently selected from the group consisting of α -hydroxylauric acid, α -hydroxymyristic acid, α -hydroxypalmitic acid, α -hydroxystearic acid, ω -hydroxylauric acid, α -hydroxyarachic acid, 9-hydroxy-12-octadecenoic acid, ricinoleic acid, α -hydroxybehenic acid, 9-hydroxy-trans-10,12-octadecadienic acid, kamolenic acid, ipurolic acid, 9,10-dihydroxystearic acid, and 12-hydroxystearic acid.

10. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has a hydrophilic/lipophilic balance of less than about 3.5.

11. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has a Hansen polarity value of less than about 4 and a Hansen hydrogen bonding value of less than about 8.

12. The method of claim 1, wherein the fatty (C₆-C₃₂)alkyl carboxylic acid material has an iodine value greater than about 50.

13. The method of claim 1, wherein the reductase is isolated from a *Nocardia* species, a *Neurospora* species, or a *Clostridium* species.

14. The method of claim 13, wherein the reductase is isolated from a *Nocardia* species.

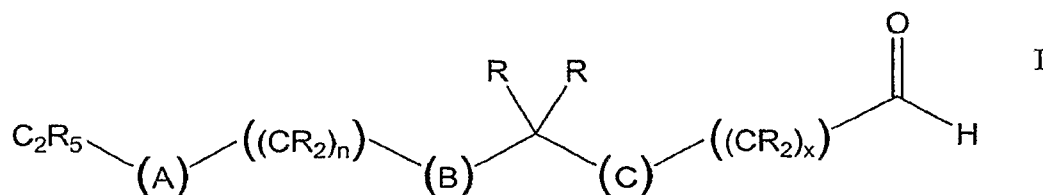
15. The method of claim 14, wherein the reductase is isolated from *Nocardia* sp. NRRL 5646.

16. The method of claim 1, wherein the reductase is a recombinant reductase.

17. The method of claim 1, further comprising incubating the fatty (C₆-C₃₂)alkyl carboxylic acid material and the carboxylic acid reductase for a period of time sufficient to reduce at least about 50% of the carboxylic acid material to the (C₆-C₃₂)aldehyde.

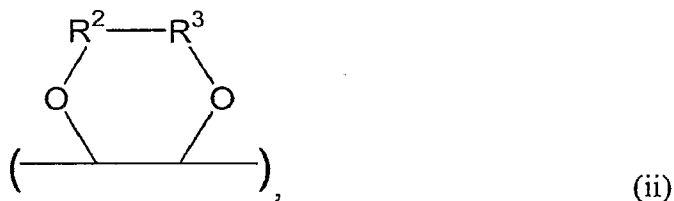
18. The method of claim 17, further comprising separating the (C₆-C₃₂)aldehyde from the fatty (C₆-C₃₂)alkyl carboxylic acid material.

19. The method of claim 1, wherein the (C₆-C₃₂)aldehyde has Formula I:

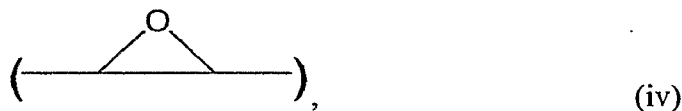
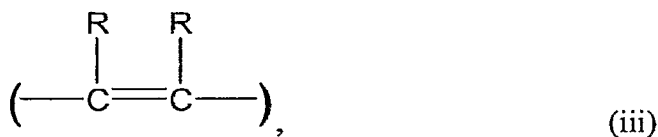


-28-

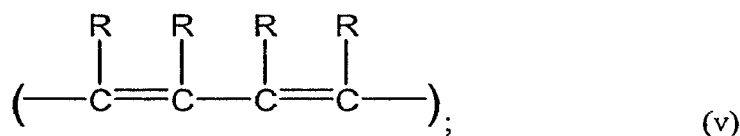
wherein, in each instance, A, B and C are independently selected from the group consisting of:



wherein, one of R^2 and R^3 is a carbonyl and the other of R^2 and R^3 is CR^4R^5 , and wherein R^4 and R^5 are independently selected from the group consisting of hydrogen, C_{1-26} alkyl, C_{3-26} alkenyl, and C_{1-26} alkoxy,



and



wherein, R is, in each instance, independently selected from the group consisting of: hydrogen, C_{1-26} alkyl, C_{3-26} alkenyl, C_{1-26} alkoxy, C_{6-10} aryl, hydroxy, hydroxy(C_{1-26})alkyl, amino(C_{1-26})alkyl, amino(C_{6-10})aryl, heteroaryl, amino(C_{6-10})aryl(C_{1-26})alkyl, heteroaryl(C_{1-26})alkyl, C_{3-6} cycloalkyl and phenyl(C_{1-26})alkyl;

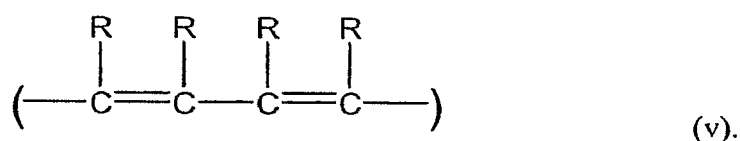
x is an integer from 5 to 10; and

n is an integer from 1 to 6.

20. The method of claim 19, wherein A, B, and C are independently selected from the group consisting of

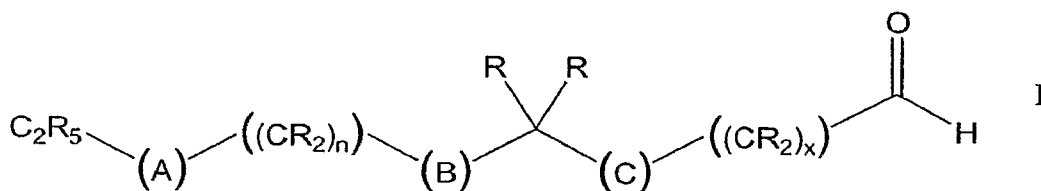


and

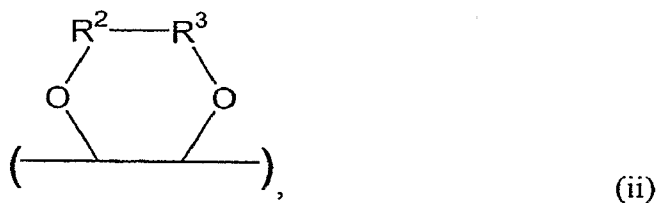


21. A (C₆-C₃₂)aldehyde made according to the method of claim 1.

22. The (C₆-C₃₂)aldehyde of claim 21, wherein the (C₆-C₃₂)aldehyde has Formula I:

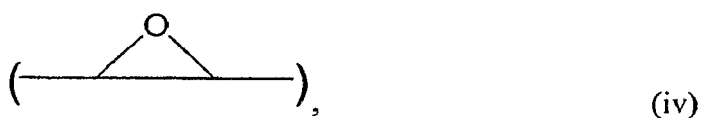
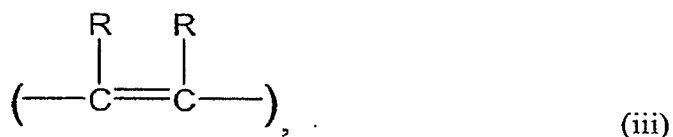


wherein, in each instance, A, B and C are independently selected from the group consisting of:

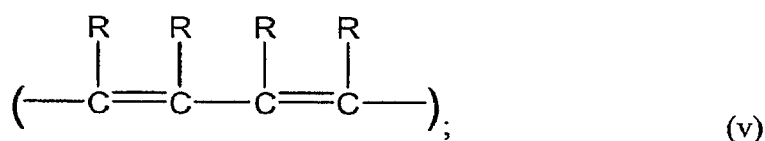


wherein, one of R² and R³ is a carbonyl and the other of R² and R³ is CR⁴R⁵, and wherein R⁴ and R⁵ are independently selected from the group consisting of hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, and C₁₋₂₆ alkoxy,

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and



wherein, R is, in each instance, independently selected from the group consisting of: hydrogen, C₁₋₂₆ alkyl, C₃₋₂₆ alkenyl, C₁₋₂₆ alkoxy, C₆₋₁₀ aryl, hydroxy, hydroxy(C₁₋₂₆)alkyl, amino(C₁₋₂₆)alkyl, amino(C₆₋₁₀)aryl, heteroaryl, amino(C₆₋₁₀)aryl(C₁₋₂₆)alkyl, heteroaryl(C₁₋₂₆)alkyl, C₃₋₆ cycloalkyl and phenyl(C₁₋₂₆)alkyl;

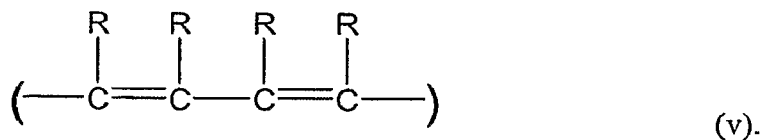
x is an integer from 5 to 10; and

n is an integer from 1 to 6.

23. The (C₆-C₃₂)aldehyde of claim 22, wherein A, B, and C are independently selected from the group consisting of



and



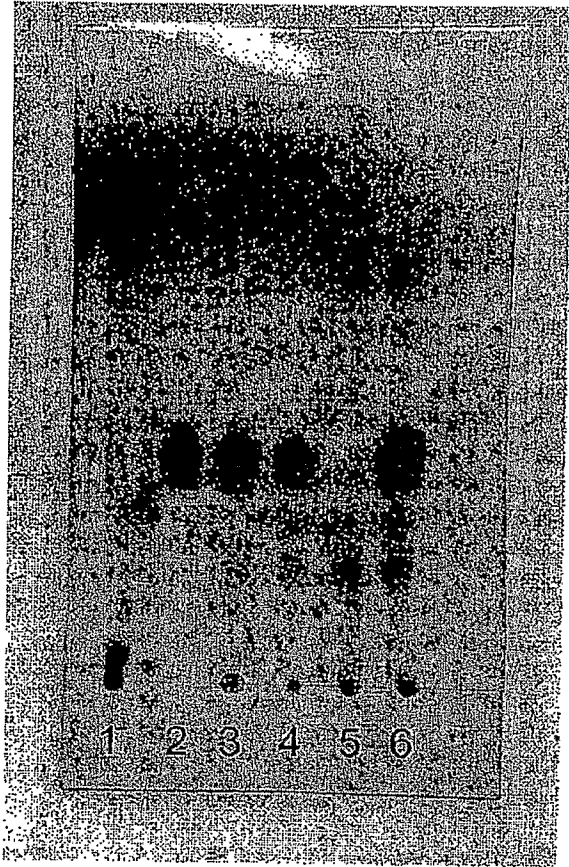


Figure 1A

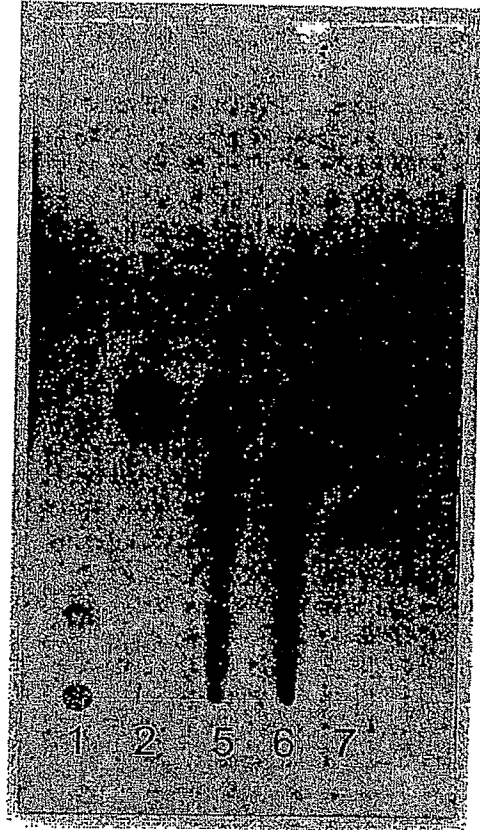


Figure 1B

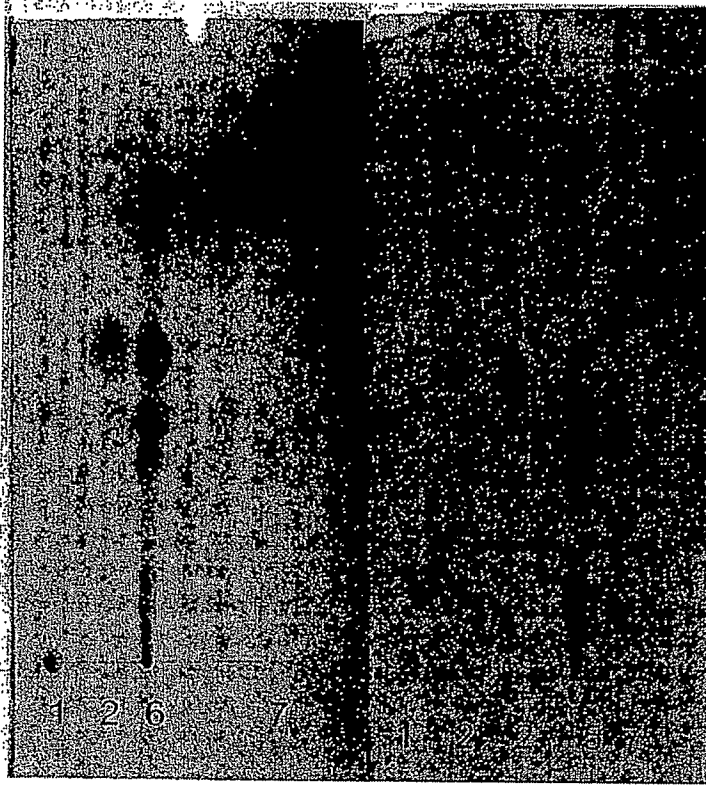


Figure 1C

Figure 1D

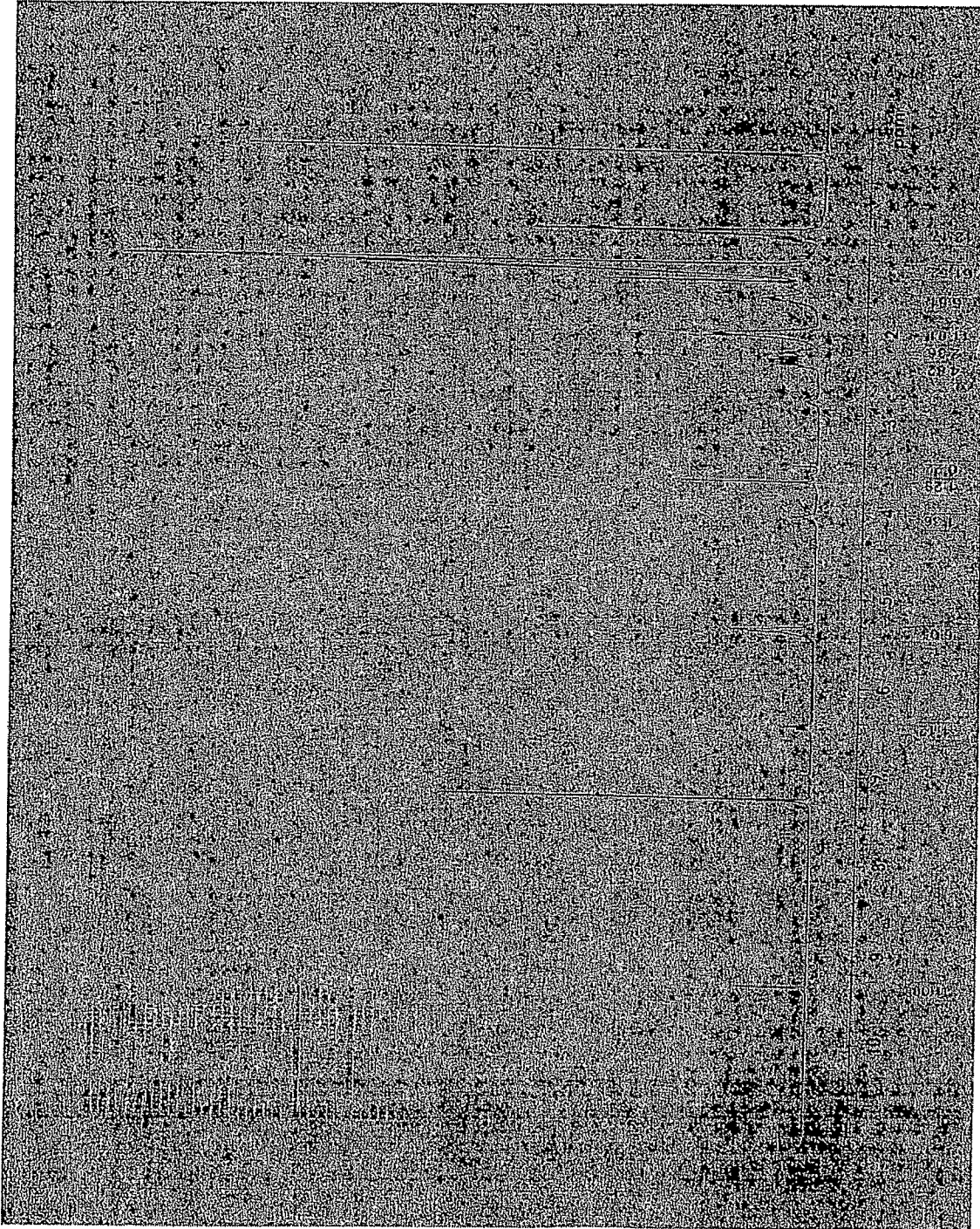


Figure 2A

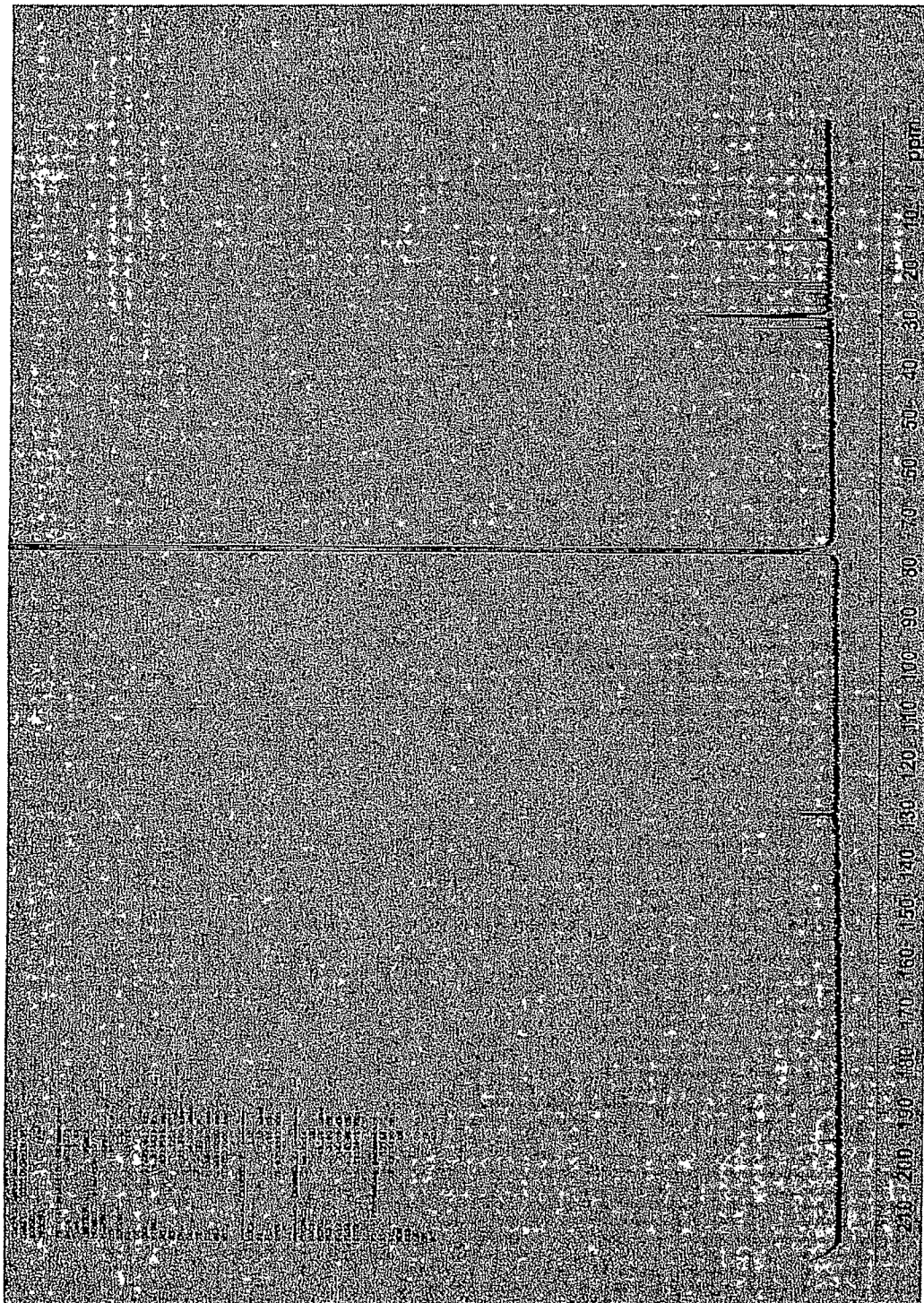


Figure 2B

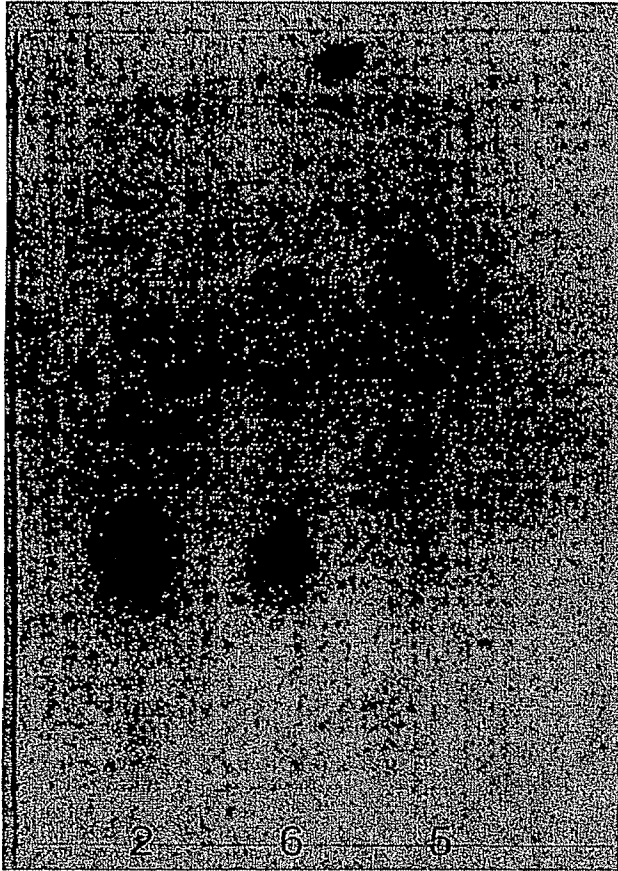


Figure 3

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2007/011556

A. CLASSIFICATION OF SUBJECT MATTER
INV. C12P7/24 C12P7/64

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
C12P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, BIOSIS, EMBASE, WPI Data, CHEM ABS Data, BEILSTEIN Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	RIENDEAU D ET AL: "Enzymatic reduction of fatty acids and acyl-CoAs to long chain aldehydes and alcohols." EXPERIENTIA 15 JUN 1985, vol. 41, no. 6, 15 June 1985 (1985-06-15), pages 707-713, XP002456132 ISSN: 0014-4754 abstract page 707, column 1, paragraph 1 - page 708, column 2, paragraph 1; table 1 page 709, column 2, paragraph 4 - page 711, column 2, paragraph 2	1-8, 10-13, 19-23
Y	----- -/--	16

Further documents are listed in the continuation of Box C.

See patent family annex.

* Special categories of cited documents:

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

26 October 2007

Date of mailing of the international search report

12/11/2007

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Fax: (+31-70) 340-3016

Authorized officer

Schröder, Gunnar

INTERNATIONAL SEARCH REPORT

International application No

PCT/US2007/011556

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DAY J I E ET AL: "PARTIAL PURIFICATION AND PROPERTIES OF ACYL COENZYME A REDUCTASE FROM CLOSTRIDIUM-BUTYRICUM" ARCHIVES OF BIOCHEMISTRY AND BIOPHYSICS, NEW YORK, US, vol. 190, no. 1, 1978, pages 322-331, XP008085140 ISSN: 0003-9861 abstract page 326, column 1, paragraph 4 - page 328, column 1, paragraph 3	1-8, 10-13, 19-23
Y		16
X	----- RIENDEAU D ET AL: "Fatty acid reductase in bioluminescent bacteria. Resolution from aldehyde reductases and characterization of the aldehyde product" CANADIAN JOURNAL OF BIOCHEMISTRY, NATIONAL RESEARCH COUNCIL OF CANADA, OTTAWA,, CA, vol. 59, no. 6, June 1981 (1981-06), pages 440-446, XP008085137 ISSN: 0008-4018 page 442, column 1, paragraph 3 - column 2, paragraph 1	1-5, 10-12, 17,19-23
Y		16
X	----- WHITE HILTRUD ET AL: "On a reversible molybdenum-containing aldehyde oxidoreductase from Clostridium formicoaceticum" ARCHIVES OF MICROBIOLOGY, BERLIN, DE, vol. 159, no. 3, 1993, pages 244-249, XP008084779 ISSN: 0302-8933 abstract page 247, column 1, paragraph 1-3; table 2	1-4
Y		16
X	----- BIERL-LEONHARDT B A ET AL: "Location of double-bond Position in Long-chain aldehydes and acetates by mass spectral analysis of epoxide derivatives" JOURNAL OF CHROMATOGRAPHIC SCIENCE, PRESTON PUBLICATIONS, NILES, IL, US, vol. 18, no. 8, 1980, pages 364-367, XP008084801 ISSN: 0021-9665 page 364, column 2, paragraph 4 - page 365, column 2, paragraph 1; figure 2; table 1	21-23

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INTERNATIONAL SEARCH REPORT

International application No

PCT/US2007/011556

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>EINHORN J ET AL: "From CI-NO+/MS TO CI-NO+/MS-MS to locate an epoxide in long chain epoxide-acetates, alcohols and aldehydes" SPECTROSCOPY, ELSEVIER, AMSTERDAM, NL, vol. 5, no. 1-6, 1988, pages 281-288, XP008084799 ISSN: 0712-4813 page 282, paragraphs 2,3; table 1 page 286, paragraph 1 - page 287, paragraph 1</p>	21-23
X	<p>----- DATABASE WPI Week 198919 Derwent Publications Ltd., London, GB; AN 1989-141726 XP002456142 & JP 01 086884 A (SHIONO KORYO KK) 31 March 1989 (1989-03-31) abstract</p>	21-23
A		1-20
X	<p>----- DATABASE PHEROLIST [Online] PheroNet, Sweden; 2004, WITZGALL P, LINDBLOM T, BENGTTSSON M, TÓTH M: "The Pherolist" XP002456139 retrieved from WWW-PHEROLIST.SLU.SE/ List of identified pheromones of Lepidopteran insects. References for each specific substance are accessible via the website. & "Aldehydes"[Online] 2004, Retrieved from the Internet: URL:http://www-pherolist.slu.se/pherolist. php?command=pherolist&file=compound_index& groupID=4&orderID=1000> [retrieved on 2007-10-22] the whole document</p>	21-23
X	<p>----- NELSON DENNIS R ET AL: "Very long-chain methyl-branched alcohols and their acetate esters in the internal lipids of Lepidopteran pupae: <i>Cochylis hospes</i>, <i>Diatraea grandiosella</i>, <i>Homoeosoma</i> <i>electellum</i>, <i>Heliothis virescens</i> and <i>Helicoverpa zea</i>" COMPARATIVE BIOCHEMISTRY AND PHYSIOLOGY B, vol. 116, no. 2, 1997, pages 243-256, XP002456136 ISSN: 0305-0491 page 245, column 2, paragraph 5 page 247, column 1, paragraph 1; table 4</p>	21-23
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International application No

PCT/US2007/011556

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>NISHIDA TAKANOBU ET AL: "Synthesis and characterization of hexadecadienyl compounds with a conjugated diene system, sex pheromone of the persimmon fruit moth and related compounds." BIOSCIENCE BIOTECHNOLOGY AND BIOCHEMISTRY, vol. 67, no. 4, April 2003 (2003-04), pages 822-829, XP002456137 ISSN: 0916-8451 page 824, column 1, paragraph 1 - page 825, column 2, paragraph 2; figure 3 scheme 1, scheme 2 page 822, column 2, paragraph 2 - page 823, column 1, paragraph 1; table 1</p>	21-23
X	<p>DATABASE BEILSTEIN BEILSTEIN INSTITUTE FOR ORGANIC CHEMISTRY, FRANKFURT-MAIN, DE; 1988, ANDO T ET AL: "Mass Spectra of Lepidopterous Sex Pheromones with a Conjugated Diene System" XP002456141 Database accession no. 6857312 abstract & AGRICULTURAL AND BIOLOGICAL CHEMISTRY, JAPAN SOC. FOR BIOSCIENCE, BIOTECHNOLOGY AND AGROCHEM, TOKYO, JP, vol. 52, no. 6, 1988, pages 1415-1424,</p>	21-23
P,A	<p>VENKITASUBRAMANIAN P ET AL (ED: PATEL R N): "Biocatalytic reduction of carboxylic acids: mechanism and applications" BIOCATALYSIS IN THE PHARMACEUTICAL AND BIOTECHNOLOGY INDUSTRIES, CRC PRESS, BOCA RATON, FL, USA, September 2006 (2006-09), pages 425-440, XP008084608 pages 431-438; tables 15.2,15.3</p>	1-20
A	<p>GROSS G G ET AL: "[Reduction of aromatic acids to aldehydes and alcohols in the cell-free system. 1. Purification and properties of aryl-aldehyde: NADP-oxidoreductase from Neurospora crassa]" EUROPEAN JOURNAL OF BIOCHEMISTRY / FEBS APR 1969, vol. 8, no. 3, April 1969 (1969-04), pages 413-419, XP002456138 ISSN: 0014-2956 cited in the application abstract page 417, column 2, paragraph 2 - page 418, column 2, paragraph 1; table 3</p>	1-13

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International application No
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C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 2004/180400 A1 (ROSAZZA JOHN P [US] ET AL) 16 September 2004 (2004-09-16) cited in the application paragraphs [0005] - [0013], [0148] - [0169]; figure 2; tables 1-3	16
A	-----	1-15
A	WO 98/40472 A (UNIV IOWA RES FOUND [US]) 17 September 1998 (1998-09-17) cited in the application abstract; table 2 page 26, line 11 - page 28, line 14 page 31, lines 6-26 claims 1-7,18,19	13-15
X	----- DATABASE PHEROBASE [Online] Ashraf M. El-Sayed; EL-SAYED A M: "The Pherobase: Database of Insect Pheromones and Semiochemicals" XP002456140 retrieved from HTTP://WWW.PHEROBASE.COM List of behaviour modifying chemicals (including pheromones). References for each specific substance are accessible via the website. & "Semiochemicals - Aldehydes"[Online] 2007, Retrieved from the Internet: URL:http://www.pherobase.com/database/compound/compounds-aldes.php> [retrieved on 2007-10-22] the whole document	21-23
X	& REED D W AND CHISOLM M D: "Attraction of moth species of Tortricidae, Gelechiidae, Geometridae, Drepanidae, Pyralidae, and Gracillariidae families to field traps baited with conjugated dienes" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NE, vol. 11, 1985, pages 1645-1657,	21-23
X	& PRESTWICH G D AND COLLINS M S: "Chemical defense secretions of the termite soldiers of Acorhinotermes and Rhiotermes (Isoptera, Rhiotermitinae): ketones, vinyl ketones, and beta-ketoaldehydes derived from fatty acids" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NL, vol. 8, 1982, pages 147-161,	21
X	& BRAND J M ET AL: "The chemistry of the defensive secretion of the beetle, Drusilla canaliculata" JOURNAL OF INSECT PHYSIOLOGY, PERGAMON PRESS, OXFORD, GB, vol. 19, 1973, pages 369-382, -/--	21

INTERNATIONAL SEARCH REPORT

International application No

PCT/US2007/011556

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	& RITTER ET AL: "Evaluation of social insect pheromones in pest control with special reference to subterranean termites and Pharaoh's ants" TRAIL PHEROMONE OF THE PHARAOH'S ANT, MONOMORIUM PHARAONIS: ISOLATION AND IDENTIFICATION OF FARANAL, A TERPENOID RELATED TO JUVENILE HORMONE II, vol. 30, 1977, pages 2617-2618,	21
X	& SWEDENBORG P D AND JONES R L: "(Z)-4-tridecenal, a pheromonally active air oxidation product from a series of (Z,Z)-9,13-dienes in Macrocentrus grandii Goidanich (Hymenoptera: Braconidae)" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NL, vol. 18, 1992, pages 1913-1931,	21
X	& COSSÉ A A: "Pheromone components of the wheat stem sawfly: identification, electrophysiology, and field bioassay" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NL, vol. 28, 2002, pages 407-423,	21
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X	& MC DANIEL ET AL: "Mandibular gland secretions of the male beeswolves Philanthus crabroniformis, P. barbatus, and P. pulcher (Hymenoptera: Sphecidae)" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NL, vol. 18, 1982, pages 27-37,	21-23
X	& SILVERSTEIN R M ET AL: "Perception by Trogoderma species of chirality and methyl branching at a site far removed from a functional group in a pheromone component" JOURNAL OF CHEMICAL ECOLOGY, KLUWER ACADEMIC PUBLISHERS-PLENUM PUBLISHERS, NL, vol. 6, 1980, pages 911-917,	21-23
X	& SANTANGELO E M ET AL: "Identification, syntheses, and characterization of the geometric isomers of 9,11-hexadecadienal from female pheromone glands of the sugar cane borer Diatraea saccharalis" JOURNAL OF NATURAL PRODUCTS, ACS, WASHINGTON, DC, US, vol. 65, 2002, pages 909-915, <p style="text-align: center;">-/--</p>	-21-23

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International application No

PCT/US2007/011556

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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X	& STEVENSON D E ET AL: "Pecan nut casebearer pheromone monitoring and degree-day model validation across the pecan belt" SOUTHWESTERN ENTOMOLOGIST, WESLACO, TX, US, vol. 57, 2003, page 27,	21-23
X	& LEVINSON H Z AND MORI K: "The pheromone activity of chiral isomers of trogoderma for male khapra beetles" NATURWISSENSCHAFTEN, SPRINGER-VERLAG, BERLIN, DE, vol. 67, 1980, pages 148-149,	21-23
X	& ROSSI R ET AL: "Chirality influences the biological activity of the sex pheromones of the khapra beetle" NATURWISSENSCHAFTEN, SPRINGER-VERLAG, BERLIN, DE, vol. 66, 1979, page 211,	21-23
X	& FRÉROT B AND DEMOLIN G: "Sex pheromone of the processionary moths and biosystematic considerations within the genus Thaumetopoea (Thaumetopoeidae Thaumetopoeinae)" BOLLETTINO DI ZOOLOGIA AGRARIA E DI BACHICOLTURA, SER. II, MILANO, IT, vol. 25, 1993, pages 33-40,	21
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