



US009249122B1

(12) **United States Patent**
Yang et al.(10) **Patent No.:** **US 9,249,122 B1**
(45) **Date of Patent:** **Feb. 2, 2016**(54) **PROCESS FOR THE PREPARATION OF**
3-(3-CHLORO-1H-PYRAZOL-1-YL)PYRIDINE(71) Applicant: **Dow AgroSciences LLC**, Indianapolis,
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Podhorez, Midland, MI (US)(*) Notice: Subject to any disclaimer, the term of this
patent is extended or adjusted under 35
U.S.C. 154(b) by 0 days.(21) Appl. No.: **14/517,315**(22) Filed: **Oct. 17, 2014****Related U.S. Application Data**(60) Provisional application No. 62/031,547, filed on Jul.
31, 2014.(51) **Int. Cl.****C07D 401/04** (2006.01)
C07D 231/14 (2006.01)
C07D 403/04 (2006.01)
A01N 43/46 (2006.01)
A61K 31/415 (2006.01)(52) **U.S. Cl.**CPC **C07D 401/04** (2013.01)(58) **Field of Classification Search**CPC **C07D 231/14**; **C07D 403/04**; **A01N 43/46**;
A61K 31/415
USPC 514/352, 406; 546/275.4; 548/364.1
See application file for complete search history.(56) **References Cited****U.S. PATENT DOCUMENTS**3,597,341 A 9/1968 Alexis
4,080,457 A 3/1978 Harrison et al.
4,260,765 A 4/1981 Harrison et al.
4,407,803 A 10/1983 Haviv et al.
4,536,506 A 8/1985 Marcoux et al.
4,824,953 A 4/1989 Bronn
5,220,028 A 6/1993 Iwasawa et al.
5,625,074 A 4/1997 Daum et al.
5,631,380 A 5/1997 Haas et al.
5,652,372 A 7/1997 Muller et al.
5,693,657 A 12/1997 Lee et al.
5,750,718 A 5/1998 Muller et al.
5,817,677 A 10/1998 Linz et al.
5,854,264 A 12/1998 Anthony et al.
5,854,265 A 12/1998 Anthony et al.
5,869,681 A 2/1999 Muller et al.
6,040,331 A 3/2000 Yamamoto et al.
6,218,418 B1 4/2001 Pevarello et al.
6,506,747 B1 1/2003 Betageri et al.
6,548,525 B2 4/2003 Galemno, Jr. et al.
6,720,427 B2 4/2004 Sanner et al.
6,878,196 B2 4/2005 Harada et al.
6,916,927 B2 7/2005 Bunnage et al.
6,965,032 B2 11/2005 Freudenberger et al.
7,192,906 B2 3/2007 Hirohara et al.
7,196,104 B2 3/2007 Askew, Jr. et al.7,319,108 B2 1/2008 Schwink et al.
7,774,978 B2 8/2010 Ding et al.
7,803,832 B2 9/2010 Critcher et al.
7,910,606 B2 3/2011 Nazare et al.
7,923,573 B2 4/2011 Tamaki et al.
8,163,756 B2 4/2012 Flynn et al.
8,222,280 B2 7/2012 Liu et al.
8,901,153 B2 12/2014 Buysse et al.
2002/0013326 A1 1/2002 Tiebes et al.
2003/0153464 A1 8/2003 Nakamura et al.
2003/0213405 A1 11/2003 Harada et al.
2004/0043904 A1 3/2004 Yamaguchi et al.
2004/0082629 A1 4/2004 Iwataki et al.
2005/0038059 A1 2/2005 Mueller et al.
2005/0176710 A1 8/2005 Schwink et al.
2006/0135778 A1 6/2006 Schnatterer et al.
2006/0160857 A1 7/2006 Buettelmann et al.
2006/0160875 A1 7/2006 Gaines et al.
2006/0167020 A1 7/2006 Dickerson et al.
2006/0287365 A1 12/2006 Billen et al.
2006/0287541 A1 12/2006 Nishino et al.
2007/0049604 A1 3/2007 Nam et al.
2007/0167426 A1 7/2007 Siddiqui et al.
2008/0004301 A1 1/2008 Tamaki et al.
2008/0027046 A1 1/2008 Annan et al.
2009/0023709 A1 1/2009 Gillespie et al.

(Continued)

FOREIGN PATENT DOCUMENTSEP 0097323 A2 1/1984
EP 0190457 A1 8/1986

(Continued)

OTHER PUBLICATIONSKempe et al. 'Responsive Glyco-poly(2-oxazoline)s: Synthesis,
Cloud Point Tuning, and Lectin Binding,' *Biomacromolecules* 2011,
vol. 12, pp. 2591-2600.
Fields et al. 'Preparation of Trifluoromethyl-Pyrazoles and
-Pyrazolines by the Reaction of 2,2,2-Trifluorodiazoethane With
Carbon—Carbon Multiple Bonds,' *Journal of Fluorine Chemistry*,
1979, vol. 13, pp. 147-158.
Bradbury et al. 'Enzyme-catalysed peptide amidation,' *Eur. J.*
Biochem. 1987, vol. 169, pp. 579-584.
International Search Report and Written Opinion for PCT/US2014/
061005 mailed Dec. 16, 2014.
International Search Report and Written Opinion for PCT/US2014/
061006 mailed Dec. 8, 2014.

(Continued)

Primary Examiner — John Mabry*Assistant Examiner* — Daniel Carcanague(74) *Attorney, Agent, or Firm* — Barnes & Thornburg LLP;
Carl Corvin(57) **ABSTRACT**3-(3-Chloro-1H-pyrazol-1-yl)pyridine is prepared by cycliz-
ing 3-hydrazinopyridine•dihydrochloride with acrylonitrile
to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-
amine, by oxidizing to provide 3-(3-amino-1H-pyrazol-1-yl)
pyridine, and by converting the amino group to a chloro group
by a Sandmeyer reaction.**4 Claims, No Drawings**

(56)

References Cited**U.S. PATENT DOCUMENTS**

2009/0069288	A1	3/2009	Breinlinger et al.
2009/0137524	A1	5/2009	Billen et al.
2009/0325956	A1	12/2009	Taniguchi et al.
2010/0130474	A1	5/2010	Bothmann et al.
2010/0204164	A1	8/2010	Crouse et al.
2010/0286169	A1	11/2010	Guiles et al.
2010/0292253	A1	11/2010	Trullinger et al.
2010/0305200	A1	12/2010	Velicelebi et al.
2011/0021771	A1	1/2011	Mallais et al.
2011/0048261	A1	3/2011	Shimura
2011/0098287	A1	4/2011	Bretschneider et al.
2011/0118290	A1	5/2011	Bretschneider et al.
2011/0166129	A1	7/2011	Machacek et al.
2011/0166143	A1	7/2011	Bretschneider et al.
2011/0184188	A1	7/2011	Wada et al.
2011/0201649	A1	8/2011	Matsuzaki et al.
2011/0212949	A1	9/2011	Bretschneider et al.
2011/0275583	A1	11/2011	Bretschneider et al.
2011/0319428	A1	12/2011	Fublein et al.
2012/0053146	A1	3/2012	Parker et al.
2012/0094837	A1	4/2012	Muhlthau et al.
2012/0095023	A1	4/2012	Bretschneider et al.
2012/0110701	A1	5/2012	Garizi et al.
2012/0110702	A1	5/2012	Yap et al.
2012/0115811	A1	5/2012	Du et al.
2012/0165345	A1	6/2012	Bretschneider et al.
2012/0172218	A1	7/2012	Crouse et al.
2012/0220453	A1	8/2012	Lowe et al.
2012/0252770	A1	10/2012	Berger et al.
2013/0072382	A1	3/2013	Trullinger et al.
2013/0089622	A1	4/2013	Trullinger et al.
2013/0109566	A1	5/2013	Niyaz et al.
2013/0261141	A1	10/2013	Bretschneider et al.
2013/0288893	A1	10/2013	Buyse et al.
2013/0291227	A1	10/2013	Buyse et al.
2013/0324736	A1	12/2013	Ross, Jr. et al.
2013/0324737	A1	12/2013	Ross, Jr. et al.

FOREIGN PATENT DOCUMENTS

EP	0205024	A2	12/1986
EP	0248315	A2	12/1987
EP	0425948	A2	5/1991
EP	1273582	A1	1/2003
EP	1321463	A1	6/2003
EP	1329160	A2	7/2003
JP	1987-153273	A	7/1987
JP	1988-174905	A	7/1988
JP	1989-226815	A	9/1989
JP	2003-212864	A	7/2003
JP	2004-051628	A	2/2004
JP	2004-292703	A	10/2004
JP	2012-188418	A	10/2012
JP	2013-075871	A	4/2013
JP	2013-082699	A	5/2013
JP	2013-082704	A	5/2013
JP	2013-107867	A	6/2013
JP	2013-129651	A	7/2013
JP	2013-129653	A	7/2013
WO	WO 94/13644	A1	6/1994
WO	WO 97/36897	A1	10/1997
WO	WO 98/49166	A1	11/1998
WO	WO 00/35919	A2	6/2000
WO	WO 01/34127	A1	5/2001
WO	WO 01/90078	A1	11/2001
WO	WO 02/083111	A2	10/2002
WO	WO 03/008405	A1	1/2003
WO	WO 03/072102	A1	9/2003
WO	WO 2004/041813	A1	5/2004
WO	WO 2005/070925	A1	8/2005
WO	WO 2005/074875	A2	8/2005
WO	WO 2006/023462	A1	3/2006
WO	WO 2006/033005	A2	3/2006
WO	WO 2006/046593	A1	5/2006

WO	WO 2006/103045	A1	10/2006
WO	WO 2007/005838	A2	1/2007
WO	WO 2007/087427	A2	8/2007
WO	WO 2007/098826	A2	9/2007
WO	WO 2008/005457	A2	1/2008
WO	WO 2008/079277	A1	7/2008
WO	WO 2008/090382	A1	7/2008
WO	WO 2009/149858	A1	12/2009
WO	WO 2010/006713	A2	1/2010
WO	WO 2010/009290	A1	1/2010
WO	WO 2010/012442	A2	2/2010
WO	WO 2010/033360	A1	3/2010
WO	WO 2010/048207	A2	4/2010
WO	WO 2010/060379	A1	6/2010
WO	WO 2010/075376	A2	7/2010
WO	WO 2010/129497	A1	11/2010
WO	WO 2010/133336	A1	11/2010
WO	WO 2010/146236	A1	12/2010
WO	WO 2011/003065	A2	1/2011
WO	WO 2011/043371	A1	4/2011
WO	WO 2011/045224	A1	4/2011
WO	WO 2011/045240	A1	4/2011
WO	WO 2011/091153	A1	7/2011
WO	WO 2011/101229	A1	8/2011
WO	WO 2011/126903	A2	10/2011
WO	WO 2011/128304	A1	10/2011
WO	WO 2011/134964	A1	11/2011
WO	WO 2011/138285	A1	11/2011
WO	WO 2011/163518	A1	12/2011
WO	WO 2012/000896	A2	1/2012
WO	WO 2012/004217	A1	1/2012
WO	WO 2012/007500	A2	1/2012
WO	WO 2012/035011	A1	3/2012
WO	WO 2012/052412	A1	4/2012
WO	WO 2012/061290	A2	5/2012
WO	WO 2012/070114	A1	5/2012
WO	WO 2012/102387	A1	8/2012
WO	WO 2012/108511	A1	8/2012
WO	WO 2012/147107	A2	11/2012
WO	WO 2012/168361	A1	12/2012
WO	WO 2013/000931	A1	1/2013
WO	WO 2013/010946	A2	1/2013
WO	WO 2013/010947	A2	1/2013
WO	WO 2013/062980	A1	5/2013
WO	WO 2013/064324	A1	5/2013
WO	WO 2013/156431	A1	10/2013
WO	WO 2013/156433	A1	10/2013

OTHER PUBLICATIONS

International Search Report and Written Opinion for PCT/US2014/061007 mailed Dec. 31, 2014.

International Search Report and Written Opinion for PCT/US2014/061009 mailed Dec. 8, 2014.

International Search Report and Written Opinion for PCT/US2014/061010 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061012 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061014 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061016 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061022 mailed Dec. 29, 2014.

International Search Report and Written Opinion for PCT/US2014/061023 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061024 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061027 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061029 mailed Dec. 15, 2014.

International Search Report and Written Opinion for PCT/US2014/061030 mailed Dec. 15, 2014.

International Preliminary Report on Patentability for PCT/US2011/058578 mailed Dec. 21, 2012.

International Search Report and Written Opinion for PCT/US2011/058578 mailed Apr. 5, 2012.

International Search Report and Written Opinion for PCT/US2013/029615 mailed May 8, 2013.

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PROCESS FOR THE PREPARATION OF 3-(3-CHLORO-1H-PYRAZOL-1-YL)PYRIDINE

CROSS REFERENCE TO RELATED APPLICATIONS

This Application claims the benefit of the following U.S. Provisional Patent Applications: Ser. No. 62/031,547, filed Jul. 31, 2014, the entire disclosure of which is hereby expressly incorporated by reference into this Application.

BACKGROUND

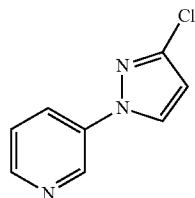
The present invention concerns an improved process for preparing 3-(3-chloro-1H-pyrazol-1-yl)pyridine.

US 20130288893(A1) describes, inter alia, certain (3-halo-1-(pyridin-3-yl)-1H-pyrazol-4-yl)amides and carbamates and their use as pesticides. The route to prepare such compounds involved the preparation of 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b) by the direct coupling of 3-bromopyridine with 3-chloropyrazole. The 3-chloropyrazole was prepared by a) treating 1H-pyrazole with 2-dimethylsulfamoyl chloride and sodium hydride to provide N,N-dimethyl-1H-pyrazole-1-sulfonamide, b) treating the N,N-dimethyl-1H-pyrazole-1-sulfonamide with perchloroethane and n-butyl lithium to provide 3-chloro-N,N-dimethyl-1H-pyrazole-1-sulfonamide, and c) removing the N,N-dimethylsulfonamide from 3-chloro-N,N-dimethyl-1H-pyrazole-1-sulfonamide with trifluoroacetic acid to give the 3-chloropyrazole.

The disclosed process produces low yields, relies on a starting material that is difficult to prepare (3-chloropyrazole) and provides a product that is difficult to isolate in a pure form. It would be desirable to have a process for preparing 3-(3-chloro-1H-pyrazol-1-yl)pyridine that avoids these problems.

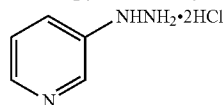
SUMMARY

The present invention provides such an alternative by cyclizing 3-hydrazinopyridine•dihydrochloride with acrylonitrile to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a), by oxidizing to provide 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a), and by converting the amino group to a chloro group by a Sandmeyer reaction. Thus, the present invention concerns a process for preparing 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b),



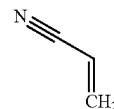
which comprises

a) treating 3-hydrazinopyridine•dihydrochloride



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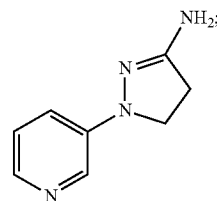
with acrylonitrile



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in a (C₁-C₄) aliphatic alcohol at a temperature of about 25° C. to about 100° C. in the presence of an alkali metal (C₁-C₄) alkoxide to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a)

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(9a)

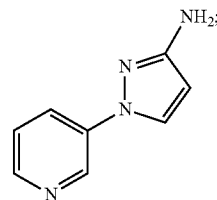
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b) treating the 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a) with an oxidant in an inert organic solvent at a temperature of about 25° C. to about 100° C. to provide 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a)

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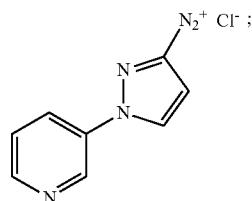


(8a)

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c) treating the 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a) in aqueous hydrochloric acid with sodium nitrite at a temperature of about 0° C. to about 25° C. to provide the diazonium salt (8b)

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(8b)

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and

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d) treating the diazonium salt (8b) with copper chloride at a temperature of about 0° C. to about 25° C.

DETAILED DESCRIPTION

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The present invention provides an improved process for preparing 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b), by

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cyclizing 3-hydrazinopyridine•dihydrochloride with acrylonitrile to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a), by oxidizing to provide 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a), and by converting the amino group to a chloro group by a Sandmeyer reaction.

In the first step, 3-hydrazinopyridine•dihydrochloride is treated with acrylonitrile in a (C₁-C₄) aliphatic alcohol at a temperature of about 25° C. to about 100° C. in the presence of an alkali metal (C₁-C₄) alkoxide to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine. While stoichiometric amounts of 3-hydrazinopyridine•dihydrochloride and acrylonitrile are required, it is often convenient to use about a 1.5 fold to about a 2 fold excess of acrylonitrile. The cyclization is run in the presence of an alkali metal (C₁-C₄) alkoxide base. It is often convenient to use about a 2 fold to about a 5 fold excess of base. The cyclization is performed in a (C₁-C₄) aliphatic alcohol. It is most convenient that the alkoxide base and the alcohol solvent be the same, for example, sodium ethoxide in ethanol.

In a typical reaction, 3-hydrazinopyridine•dihydrochloride and an anhydrous alcohol are introduced into a reaction vessel and the alkoxide base is gradually added. The mixture is stirred and the acrylonitrile is added. The mixture is stirred at about 60° C. until most of the 3-hydrazinopyridine has reacted. The mixture is allowed to cool and the excess base is neutralized with acid. The crude 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a) is conveniently isolated and purified by standard techniques.

In the second step, 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a) is treated with an oxidant in an organic solvent at a temperature of about 25° C. to about 100° C. to provide 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a). Suitable oxidants include manganese(IV) oxide, potassium ferricyanide(III), copper(I) chloride in the presence of oxygen, and iron(III) chloride in the presence of oxygen. Manganese(IV) oxide is preferred. It is often convenient to use about a 2 fold to about a 10 fold excess of oxidant. The oxidation is performed in a solvent that is inert to the oxidant. Suitable solvents include nitriles such as acetonitrile or halocarbons such as dichloromethane. With manganese(IV) oxide as the oxidant, acetonitrile is a preferred solvent.

In a typical reaction, 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a) and solvent are mixed with the oxidant and the mixture is heated at about 60° C. until the reaction is completed. The 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a) is conveniently isolated and purified by standard techniques.

The 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a) is then converted to the desired 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b) by treatment in aqueous hydrochloric acid with sodium nitrite at a temperature of about 0° C. to about 25° C. to provide a diazonium salt followed by treatment of the diazonium salt with copper chloride at a temperature of about 0° C. to about 25° C. While stoichiometric amounts of reagents are required, it is often convenient to use an excesses of reagents with respect to the 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a). Thus, aqueous hydrochloric acid is used in large excess as the reaction medium. Sodium nitrite is used in about a 1.3 fold to about a 2 fold excess. Copper chloride is used in about 5 mole percent to about 60 mole percent excess, preferably from about 15 mole percent to about 30 mole percent excess. The copper chloride may be either copper(I) chloride or copper(II) chloride. To suppress foaming during the reaction a water-immiscible organic solvent such as toluene or chloroform can be added during the treatment of the diazonium salt with copper chloride.

In a typical reaction, a mixture of 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a) and aqueous hydrochloric acid are mixed

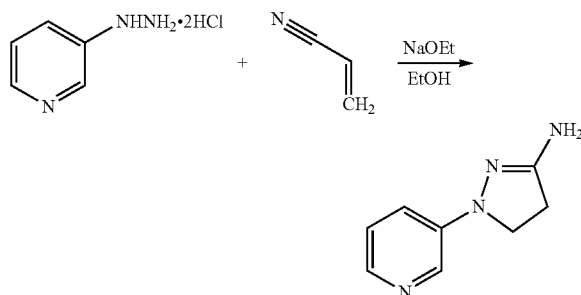
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and cooled to about 0° C. An aqueous solution of sodium nitrite is slowly added maintaining the temperature below about 5° C. The suspension is stirred at about 0° C. for about 2 hours. In a separate vessel, a mixture of copper(II) chloride and toluene is cooled to about 0° C. and the chilled suspension of diazonium salt is added at a rate maintaining the temperature below about 5° C. The mixture is allowed to warm to about ambient temperature. After completion of the reaction, the mixture is treated with aqueous sodium hydroxide to adjust the pH to about 8 to about 10. The resulting solution is extracted with a water-immiscible organic solvent. After removal of the solvent, the 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b) can be used directly in the next reaction or further purified by standard techniques such as flash column chromatography or crystallization.

The following examples are presented to illustrate the invention.

EXAMPLES

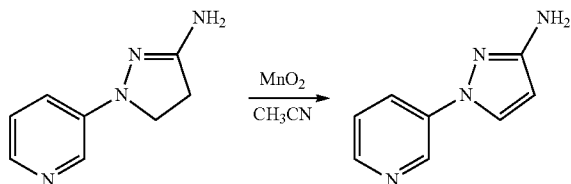
1. Preparation of 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a)



To a 4-neck, round bottomed flask (250 mL) was charged sodium ethanolate (21 wt % in ethanol, 32 mL). 3-Hydrazinopyridine•dihydrochloride (5.00 g, 27.5 mmol) was added, causing an exotherm from 20° C. to 58° C. The mixture was allowed to cool to 20° C. and acrylonitrile (2.91 g, 54.9 mmol) was added. The reaction was heated at 60° C. for 5 hours and cooled to 20° C. The excess sodium ethanolate was quenched with hydrochloric acid (4 M in 1,4-dioxane, 6.88 mL, 27.5 mmol) at <20° C. The mixture was adsorbed on silica gel (10 g) and the crude product was purified by flash column chromatography using 0-10% methanol/dichloromethane as eluent. The fractions containing pure product were concentrated to dryness to afford the title compound as a yellow solid (3.28 g, 74%): mp 156-160° C.; ¹H NMR (400 MHz, CDCl₃) δ 8.24 (dd, J=2.8, 0.8 Hz, 1H), 8.01 (dd, J=4.6, 1.4 Hz, 1H), 7.22 (ddd, J=8.4, 2.8, 1.5 Hz, 1H), 7.12 (ddd, J=8.4, 4.6, 0.8 Hz, 1H), 4.20 (s, 2H), 3.70 (t, J=9.3 Hz, 2H), 2.92 (t, J=9.3 Hz, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 154.23, 144.78, 139.22, 135.08, 123.44, 119.44, 49.23, 32.74; ESIMS m/z 163 ([M+H]⁺).

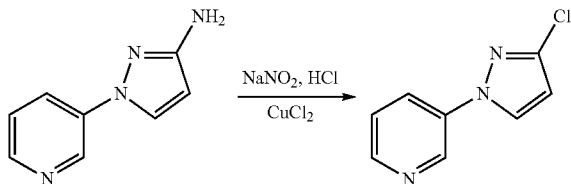
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2. Preparation of 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a)



To a 3-neck, round bottomed flask (100 mL) was charged 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (1.00 g, 6.17 mmol) and acetonitrile (20 mL). Manganese(IV) oxide (2.68 g, 30.8 mmol) was added, causing an exotherm from 20° C. to 25° C. The reaction was stirred at 60° C. for 18 hours, after which it was filtered through a Celite® pad and the pad was rinsed with acetonitrile (20 mL). Water (20 mL) was added to the combined filtrates and the resulting mixture was concentrated to 10 mL. Water (20 mL) was added and the resulting mixture was again concentrated to 10 mL. The resulting suspension was stirred at 20° C. for 18 hours and filtered. The filter cake was rinsed with water (2×5 mL) and dried to afford the title compound as a brown solid (0.68 g, 69%): mp 169-172° C.; ¹H NMR (400 MHz, DMSO-d₆) δ 9.07-8.82 (m, 1H), 8.33 (dd, J=4.6, 1.5 Hz, 1H), 8.24 (d, J=2.6 Hz, 1H), 8.00 (ddd, J=8.4, 2.7, 1.4 Hz, 1H), 7.42 (ddd, J=8.5, 4.6, 0.8 Hz, 1H), 5.80 (d, J=2.6 Hz, 1H), 5.21 (s, 2H); ¹³C NMR (101 MHz, DMSO-d₆) δ 157.67, 144.68, 138.00, 136.22, 128.30, 123.95, 123.17, 97.08; ESIMS m/z 161 ([M+H]⁺).

3. Preparation of 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b)



To a 3-neck round bottomed flask (250 mL) was charged 3-(3-amino-1H-pyrazol-1-yl)pyridine (5.00 g, 31.2 mmol) and hydrogen chloride (37 wt %, 15 mL). The mixture was cooled to 0° C. A solution of sodium nitrite (4.31 g, 62.4 mmol) in water (15 mL) was added in portions at <1° C. over 20 minutes and the resulting brown solution was stirred at <0° C. for 2 hours. To a separate 3-neck round bottomed flask (250 mL) was charged copper(II) chloride (5.04 g, 37.5 mmol) and toluene (30 mL). It was cooled to 0° C. and the yellow solution was added in portions at <1° C. over 15 minutes. The resulting mixture was allowed to warm up, off-gassing was observed when the reaction temperature reached 18° C. The reaction was stirred at 20° C. for 18 hours. The reaction was basified with 50 wt % sodium hydroxide to pH ~10. Celite® (10 g) was added and the resulting suspension was stirred for 10 minutes. The suspension was filtered through a Celite® pad (10 g) and the filter cake was rinsed with ethyl acetate (2×50 mL). The layers of the filtrates were separated and the aqueous layer was extracted with ethyl acetate (100 mL). The organic layers were concentrated to dryness and the residue was purified by flash column chromatography using 0-60% ethyl acetate/hexanes as eluent. The fractions containing the desired product were concentrated to give the title compound

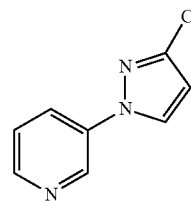
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as a white solid (3.80 g, 68%): mp 104-106° C.; ¹H NMR (400 MHz, CDCl₃) δ 8.93 (d, J=2.7 Hz, 1H), 8.57 (dd, J=4.8, 1.4 Hz, 1H), 8.02 (ddd, J=8.3, 2.7, 1.5 Hz, 1H), 7.91 (d, J=2.6 Hz, 1H), 7.47-7.34 (m, 1H), 6.45 (d, J=2.6 Hz, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 148.01, 142.72, 140.12, 135.99, 128.64, 126.41, 124.01, 108.08; EIMS m/z 179 ([M]⁺).

What is claimed is:

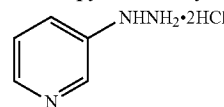
1. A process for preparing 3-(3-chloro-1H-pyrazol-1-yl)pyridine (5b),

(5b)

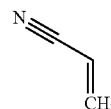


which comprises

a) treating 3-hydrazinopyridine•dihydrochloride

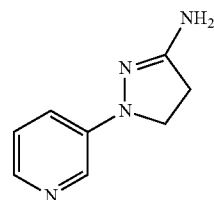


with acrylonitrile



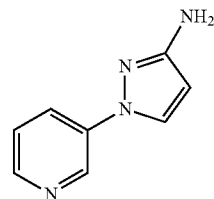
in a (C₁-C₄) aliphatic alcohol at a temperature of about 25° C. to about 100° C. in the presence of an alkali metal (C₁-C₄) alkoxide to provide 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a)

(9a)



b) treating the 1-(pyridin-3-yl)-4,5-dihydro-1H-pyrazol-3-amine (9a) with an oxidant in an organic solvent at a temperature of about 25° C. to about 100° C. to provide 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a)

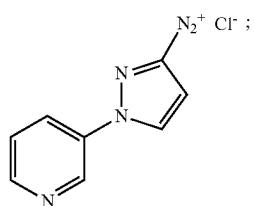
(8a)



c) treating the 3-(3-amino-1H-pyrazol-1-yl)pyridine (8a) in aqueous hydrochloric acid with sodium nitrite at a temperature of about 0° C. to about 25° C. to provide the diazonium salt (8b)

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(8b)

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and

d) treating the diazonium salt (8b) with copper chloride at a temperature of about 0°C . to about 25°C .

2. The process of claim 1 in which the oxidant is manganese(IV) oxide, potassium ferricyanide (III), copper(I) chloride in the presence of oxygen, or iron(III) chloride in the presence of oxygen. 15

3. The process of claim 2 in which the oxidant is manganese(IV) oxide. 20

4. The process of claim 1 in which a water immiscible organic solvent is added in step d) to suppress foaming.

* * * * *