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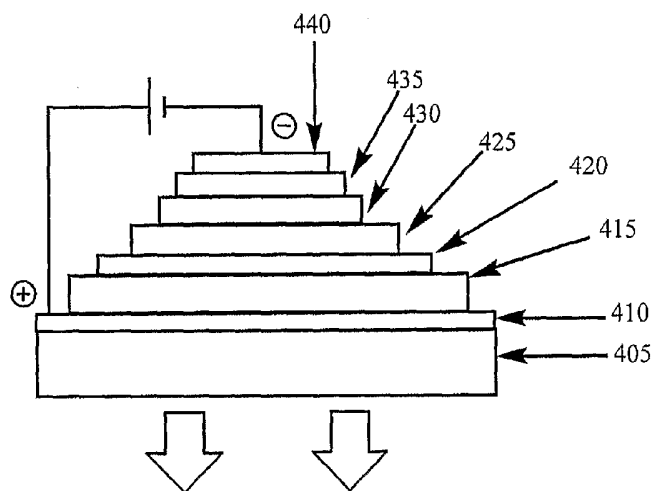
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- (71) Applicant (for all designated States except US): THE UNIVERSITY OF HONG KONG [CN/CN]; G18, Eliot Hall, Pokfulam Road, Hong Kong (CN).
- (72) Inventors: CHE, Chiming; Flat 5, 5/F, Block A, Parkway Court, 4 Park Road, Hong Kong (CN). CHAN, Siuchung; Flat 2421, Wah Sin House, Wah Kwai Estate, Hong Kong (CN).
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[Continued on next page]

(54) Title: ORGANIC LIGHT-EMITTING DEVICES



Electroluminescence

Schematic diagram of OLED in present invention

(57) Abstract: Disclosed are electrophosphorescent organic metal complexes with formula (I), (II), (III) or (IV), of either geometrical isomers, comprising two bidentate NN-type ligands, or two NO-type ligands, or a tetradentate NNNN-type ligand or a tetradentate NOON-type ligand, and a transition metal. These electrophosphorescent materials are valuable to the application in organic light-emitting devices (OLEDs), including red-, orange-, or yellow-light OLEDs.



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— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

Organic Light-Emitting Devices

FIELD OF THE INVENTION

The present invention relates to efficient organic light-emitting devices (OLEDs) which comprise a transition metal complex, wherein the transition metal complex, of either geometrical isomers, comprises two bidentate NN-type ligands, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand, and a transition metal atom as the electrophosphorescent emitter. The invention also relates to methods for preparing thin film OLEDs and their applications such as in liquid crystal displays, plasma panel displays and light-emitting diodes.

BACKGROUND OF THE INVENTION

Organic light-emitting devices (OLEDs) are finding applications as next-generation flat-panel displays (FPDs), liquid crystal displays (LCDs), and plasma display panels (PDPs). This has been driven by their favorable properties including lightweight, fast video response and low power consumption. To this end, organometallic compounds exhibiting electroluminescence are particularly attractive for electrophosphorescent applications, since both the ligand structure and the central metal atom can be varied to modify the properties of the device using these compounds.

An organic light-emitting device (OLED) is an energy conversion device which emits light when current is applied. A multilayer OLED is generally comprised of hole and electron injection layers, hole- and electron-transporting layers, an

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emissive layer, metal oxide layer and metal electrodes. The use of organic small molecules and polymers in the emissive layer has attracted much attention due to their potential applications in full-color large-area flat-panel displays. Tang and VanSlyke first disclosed that organic small molecules can be prepared as thin-films by vacuum deposition to form multilayer organic light-emitting devices (OLEDs) (*see* Tang *et al.*, *Appl. Phys. Lett.* 51:913, (1987)).

Investigations on organic small molecules have been made in order to improve the performance of OLEDs. In general, fluorescent and phosphorescent materials are employed as light emitters in the emissive layer of OLEDs. Light emission from a fluorescent compound occurs as a result of formation of singlet excitons in the emissive layer of the electroluminescent device. U.S. Patent No. 6,310,360 disclosed that theoretically 25 % singlet excitons and 75 % triplet excitons are produced after recombination of holes and electrons in the emissive layer of an electroluminescent device. The singlet excitons transfer their energy to the singlet excited state while the triplet excitons transfer their energy to triplet excited state. Most of the organic small molecules exhibit fluorescence; hence, only 25 % of the generated excitons are utilized resulting in the device with low external efficiency.

In contrast to fluorescent compounds, a series of effective phosphorescent iridium complexes with different color emissions has been reported jointly by Thompson *et al.* at the University of Southern California and Forrest *et al.* at Princeton University (*see* U.S. Patent No. 6,515,298 B2; U.S. Patent Application Publication No.20020182441 A1; Lamansky *et al.*, *J. Am. Chem. Soc.*, 123:4304 (2001); and Xie *et al.*, *Adv. Mat.*, 13:1245 (2001)). Che *et al.* also demonstrated the use of organic metal complexes employing various metal centres such as platinum(II), copper(I), gold(I), and zinc(II) as OLED emitters (*see* U.S. Patent Application

Publication No. 23205707 A1; U.S. Patent Application Publication No. 22179885 A1; Y.-Y. Lin *et al.*, *Chem. Eur. J.*, **9**:1263 (2003); Lu *et al.*, *Chem. Commun.*, 206 (2002); Ma *et al.*, *New J. Chem.*, 263 (1999); Ma *et al.*, *Appl. Phys. Lett.*, **74**:1361 (1999); Ho *et al.*, *Chem. Commun.*, 2101 (1998); and Ma *et al.*, *Chem. Commun.*, 2491 (1998)).

A variety of light-emitting compounds, especially red emitters, have been investigated as active emitters in a number of device structures. U.S. Patent No. 6,048,630 disclosed OLEDs based on phosphorescent Pt(OEP) complex (H₂OEP = octylethylporphyrin) which emits saturated red electroluminescence. Thompson and Forrest *et al.* reported a red phosphorescent material (bis(2-(2'-benzo[4,5-a]thienyl)pyridinato-N, C³) iridium (acetylacetonate) [Btp2Ir(acac)]) with high-efficiency ($\eta_{\text{ext}} = 7.0 \pm 0.5 \%$) (see Adachi *et al.*, *Appl. Phys. Lett.*, **78**:1622 (2001)). In addition, europium complex employed as red emissive dopant in OLED (Eu(TTA)₃phen, TTA = thenoyltrifluoroacetone; phen = 1,10-phenanthroline) was also reported to show sharp red electroluminescence (see Adachi *et al.*, *J. Appl. Phys.*, **87**:8049, (2000)).

Efforts in the development of red phosphorescent emitters with high efficiency for OLEDs are geared towards the full-color flat panel display application. Even though remarkable progress has been made, challenges such as optimization of stability and efficiency of OLEDs need to be met before commercialization. It is, therefore, particularly contemplated to develop phosphorescent materials, which exhibit electroluminescent (EL) emissions in visible light region, with high efficiencies and good stabilities.

Summary of the Invention

The main objective of this invention is to provide organic light-emitting devices (OLEDs) comprising an emissive layer, which employs at least one dopant complex as an electrophosphorescent emitter. The devices should exhibit low turn-on voltages, high luminance, high efficiencies, and desirable colors.

Another objective of the present invention is to provide an OLED structure, which employs an emissive layer comprising at least one electrophosphorescent dopant complex and at least one host material.

Yet another objective is to provide OLEDs that emit desirable colors by varying concentration of the dopant complex in the emissive layer under different applied voltages. It is concerned with the efficiencies of the selected dopant complexes, which can be used at low concentration levels in OLEDs.

In one embodiment, the invention relates to a heterostructured organic light-emitting device comprising:

a substrate upon which a first electrode is placed;

a hole-transporting layer;

at least one emissive layer comprising at least one host material and at least one dopant complex; the dopant complex, of either geometrical isomers, comprising at least one transition metal coordinated to two bidentate NN-type ligands, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand;

a hole-blocking layer;

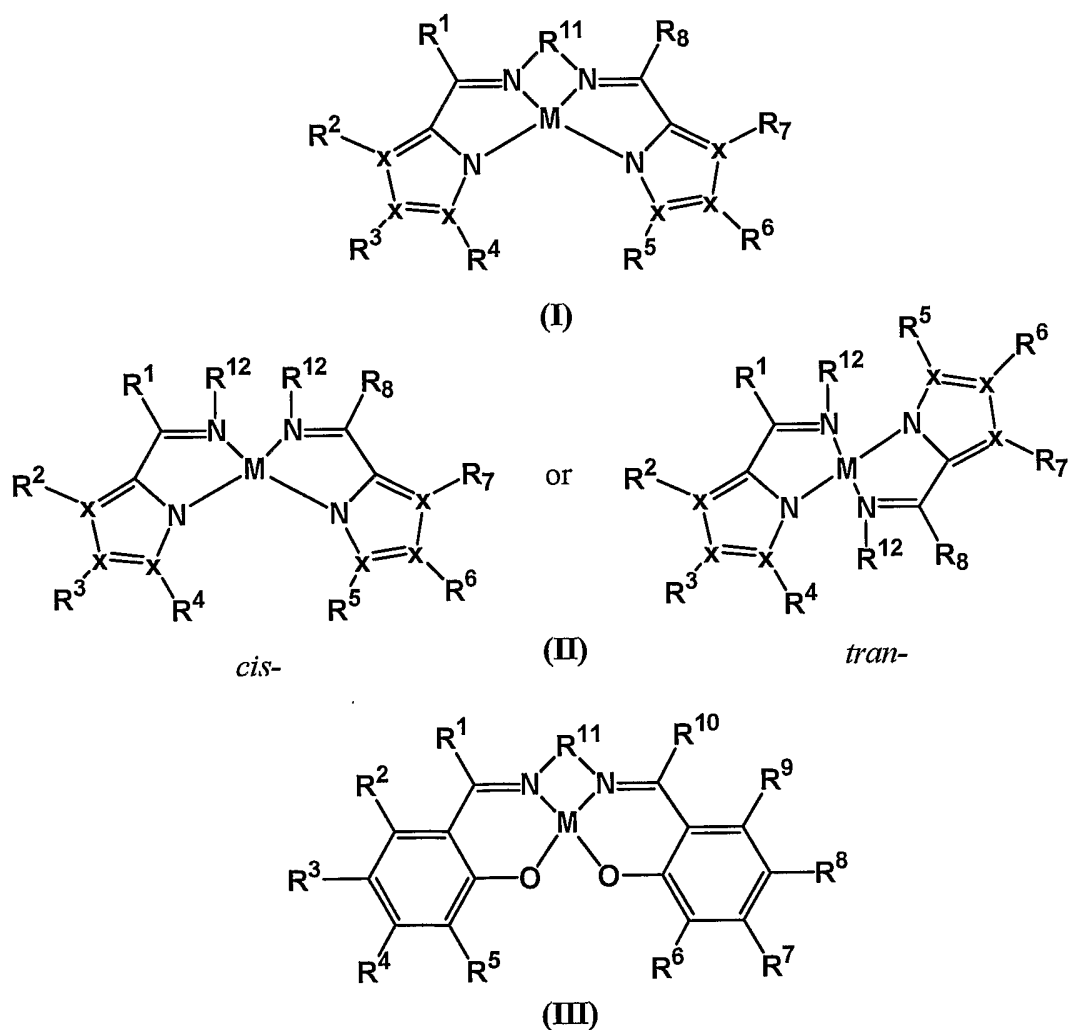
an electron-transporting layer;

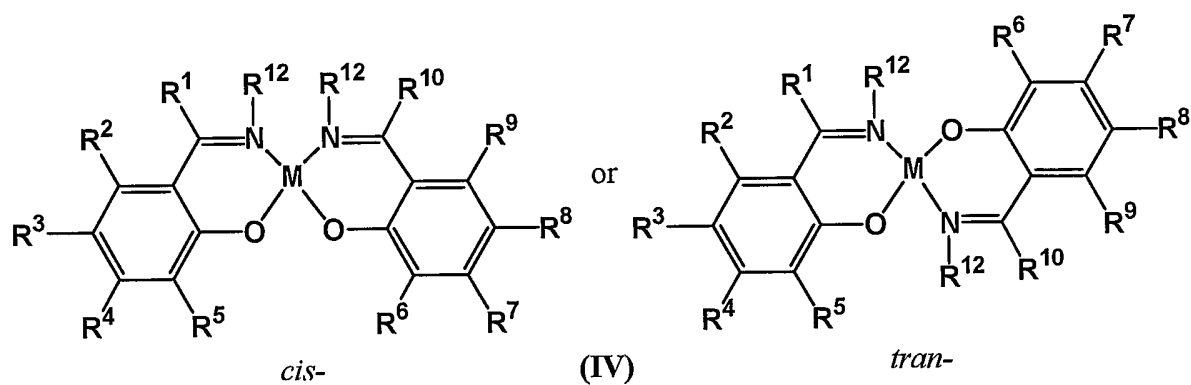
a charge injection layer; and

a second electrode sandwiching the hole-transporting layer, emissive layer, hole-blocking layer, electron-transporting layer and charge injection layer between the first and the second electrode.

In preferred embodiments, the invention relates to OLED comprising an emissive layer which contains at least one transition metal complex. The transition metal complexes, of either geometrical isomers, contain two bidentate NN-type ligand, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand, and a transition metal as the electrophosphorescent dopant complexes.

In a preferred embodiment, the invention relates to a heterostructured OLED comprising one or more dopant complexes of following formulae:





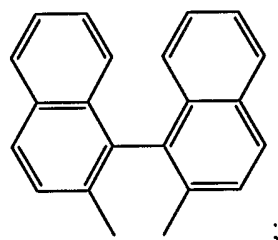
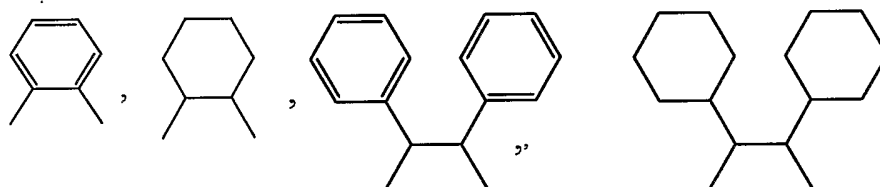
or mixtures thereof, wherein

M is a transition metal selected from the group consisting of Ni, Pd and Pt;

each R^1 - R^{10} is independently -H, -OH, -NH₂, -halogen, -CN, -NO₂, -R¹³, -OR¹⁴, NHR¹⁴,

or -N(R¹⁴)₂;

R¹¹ is -(C(R¹⁵)₂)_n,



or

each R^{12} is independently -H, -(C₁-C₆)alkyl, -phenyl, -naphthyl; -halogen, or -CN;

R¹³ is -(C₁-C₆)alkyl, -phenyl, or -naphthyl, each of which is unsubstituted or

substituted with one or more -(C₁-C₆)alkyl, -phenyl, or -naphthyl;

R¹⁴ is defined as above for R¹³; and

R¹⁵ is defined as above for R¹;

x is independently a carbon or nitrogen atom; and

n is an integer number from 1 to 6.

Another embodiment, the present invention relates to a method of preparing heterostructured organic light emitting devices with yellow, orange or red color emissions. The method includes the steps of:

providing a substrate upon which a first electrode is placed;

providing a hole-transporting layer on top of the first electrode;

forming an emissive layer on top of the hole-transporting layer, the emissive layer comprising at least one host material and at least one dopant complex, the dopant complex, of either geometrical isomers, comprising a transition metal coordinated to two bidentate NN-type ligands, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand.

providing a hole-blocking layer on top of the emissive layer;

providing an electron-transporting layer on top of the hole-blocking layer;

providing a charge injection layer on top of the electron-transporting layer;

and

providing a second electrode on top of the charge injection layer.

In a preferred embodiments of the present invention includes, but is not limited to, OLEDs comprising heterostructures for producing red, orange or yellow electroluminescence; the devices contain an anode (ITO glass substance), a hole-transporting layer (*N,N'*-diphenyl-*N,N'*-bis(2-naphthalene)benzidine (β -NPB)), an emissive layer comprising a host material (4,4'-bis(carbazol-9-yl)biphenyl (CBP)) and an electrophosphorescent dopant complex as illustrated in Formulae **I**, **II**, **III** or **IV** herein), a hole-blocking layer (2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline

(BCP)), an electron-transporting layer (tris(8-hydroxyquinolato) aluminum (Alq_3)), a charge injection layer (lithium fluoride) and a cathode (aluminum metal).

More preferably, the OLEDs employing electrophosphorescent dopant complexes as illustrated in Formulae **I**, **II**, **III** or **IV** herein demonstrate red, orange or yellow emission while a current is applied.

In according with the present invention, the high efficiency OLEDs can be applied to the field of electronic flat panel display, display board for sign lamp or light source.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1. Absorption, excitation and emission spectra of dopant complex **1** in CH_3CN

FIG. 2. Absorption, excitation and emission spectra of dopant complex **2** in CH_3CN

FIG. 3. Absorption, excitation and emission spectra of dopant complex **4** in CH_3CN

FIG 4. Schematic diagram of OLED in present invention

FIG 5. EL spectra of OLED **A** with 4.0 wt.% dopant **1** at different dopant concentrations under 8 V

FIG 6. EL spectra of OLED **A** with 4.0 wt.% dopant **1** at different applied voltages

FIG 7. V-I-B curve of OLED **A** with 4.0 wt.% dopant **1**

FIG 8. External quantum efficiency and luminous efficiency of OLED **A** with 4.0 wt% dopant **1**

FIG 9. EL spectra of OLED **B** with 5 wt.% dopant **2** at different dopant concentrations at 8 V

FIG 10. EL spectra of OLED **B** with 5 wt.% dopant **2** at different applied voltages

FIG 11. V-I-B curve of OLED **B** with 5 wt.% dopant **2**

FIG 12. The external quantum efficiency and luminous efficiency of OLED **B** with 5.0 wt.% dopant **2**

FIG 13. EL spectra of OLED **C** with 3 wt.% dopant **19** at different dopant concentrations at 8 V

FIG 14. EL spectra of OLED **C** with 3 wt.% dopant **19** at different applied voltages

FIG 15. V-I-B curve of OLED **C** with 3 wt.% dopant **19**

FIG 16. The external quantum efficiency and luminous efficiency of OLED **C** with 3.0 wt.% dopant **19**

FIG 17. EL spectra of OLED **D** with 4 wt.% dopant **51** at different dopant concentrations at 8 V

FIG 18. EL spectra of OLED **D** with 4 wt.% dopant **51** at different applied voltages

FIG 19. V-I-B curve of OLED **D** with 4 wt.% dopant **51**

FIG 20. The external quantum efficiency and luminous efficiency of OLED **D** with 4.0 wt.% dopant **51**

FIG. 21 EL spectra of OLED **E** with 1.5 wt.% dopant **99** at different dopant concentrations at 8 V

FIG. 22 EL spectra of OLED **E** with 1.5 wt.% dopant **99** at different applied voltages

FIG. 23 V-I-B curve of OLED **E** with 1.5 wt.% dopant **99**

FIG. 24 External quantum efficiency and luminous efficiency of OLED E with 1.5 wt.% dopant 99

FIG. 25 EL spectra of OLED F with 1.6 wt.% dopant 107 at different dopant concentrations at 8 V

FIG. 26 EL spectra of OLED F with 1.6 wt.% dopant 107 at different applied voltages

FIG. 27 V-I-B curve of OLED F with 1.6 wt.% dopant 104

FIG. 28 External quantum efficiency and luminous efficiency of OLED F with 1.6 wt.% dopant 104

DETAILED DESCRIPTION OF THE INVENTION

Some definitions useful for describing the present invention are provided below:

As used herein, the phrase “bidentate NN-type ligands” refers to a molecule containing an imine group and either a pyrrol group, a pyrazol group, an imidazol group, or a triazol group, which ligand is coordinated to a metal through the nitrogen atoms of these groups.

As used herein, the phrase “bidentate NO-type ligands” refers to a molecule containing an imine group and a phenoxide group, which ligand is coordinated to a metal through the nitrogen and oxygen atoms of these groups.

As used herein, the phrase “tetradentate NNNN-type ligand” refers to a molecule containing a two imine groups and either two pyrrol groups, two pyrazol groups, two imidazol groups, or two triazol groups, which ligand is coordinated to a metal through the nitrogen atoms of these groups.

As used herein, the phrase “tetradentate NOON-type ligand” refers to a molecule containing two imine groups and two phenoxide groups, which ligand is coordinated to a metal through the nitrogen and oxygen atoms of these groups.

As used herein, the phrase “light-emitting device” refers to structures presenting an asymmetric impedance to current. Typically, such a device allows current to flow more easily in one direction when it is said to be forward biased. However, in some devices of the present invention, significant current may flow in the reverse biased state as well with generation of light.

The present invention is related to a heterostructured OLED comprising an emissive layer, wherein the emissive layer comprises at least one host material and at least one emissive material. Preferably, the emissive material is a dopant complex, of either geometrical isomers, comprising a transition metal coordinated to two bidentate NN-type ligands, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand. The dopant complex can be present as a monomer, a dimer, an oligomer, or mixtures thereof.

In one embodiment, the invention relates to a heterostructured organic light-emitting device comprising:

a substrate upon which a first electrode is placed;

a hole-transporting layer;

at least one an emissive layer comprising at least one host material and at least one dopant complex; the dopant complex, of either geometrical isomers, comprising at least one transition metal coordinated to two bidentate NN-type ligands or two bidentate NO-type ligands or a tetradentate NNNN-type ligand or a tetradentate NOON-type ligand;

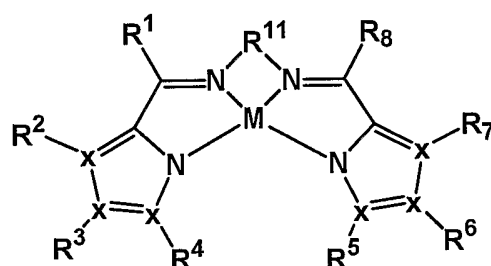
a hole-blocking layer;

an electron-transporting layer;

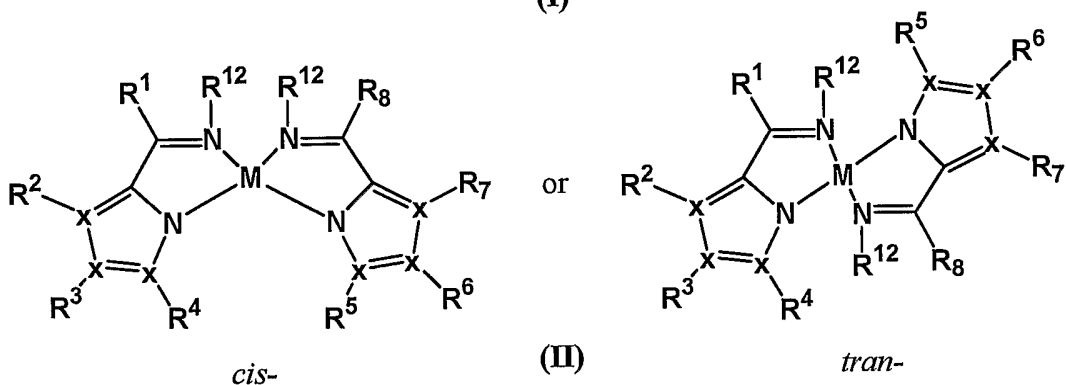
a charge injection layer; and

a second electrode sandwiching the hole-transporting layer, emissive layer, hole-blocking layer, electron-transporting layer and charge injection layer between the first and the second electrode.

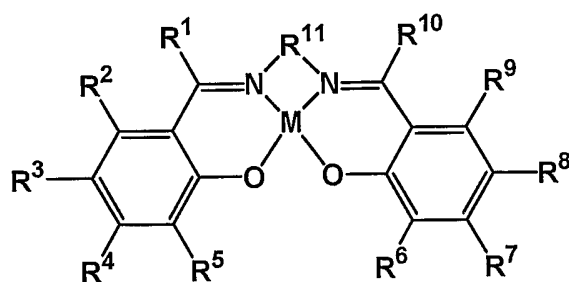
Preferably, the emissive materials are dopant complexes, of either geometrical isomers, comprising two bidentate NN-type ligands or two bidentate NO-type ligands or a tetradentate NNNN-type ligand or a tetradentate NOON-type ligand and a transition metal atom. More preferably, the emissive materials include dopant complexes of the following formulae:



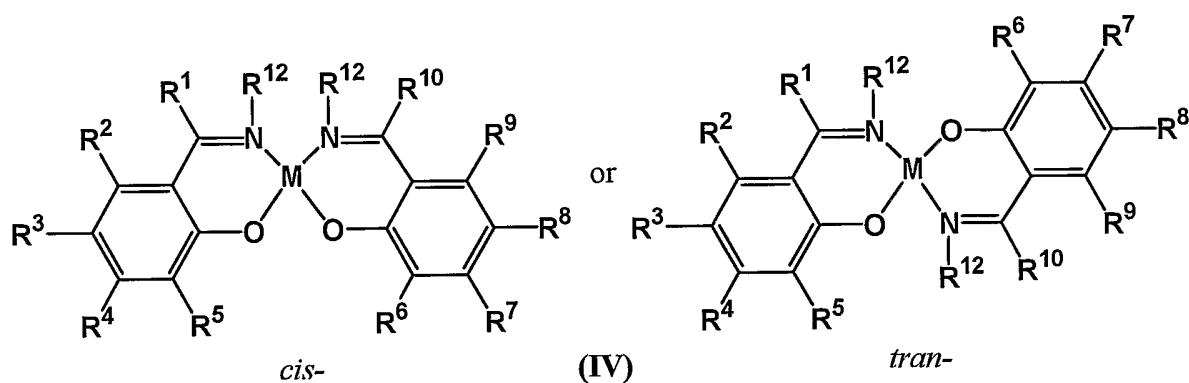
(I)



(II)



(III)



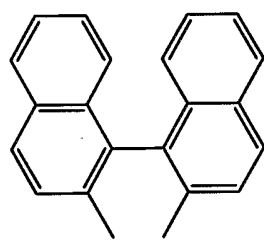
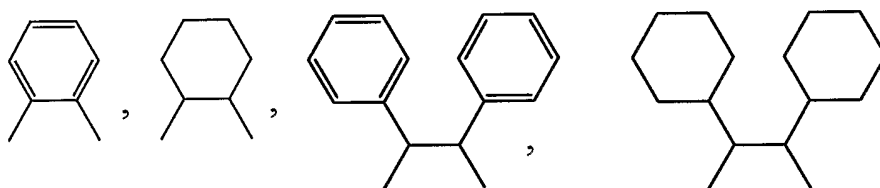
or mixtures thereof, wherein

M is a transition metal selected from the group consisting of Ni, Pd and Pt;

each R^1 - R^{10} is independently -H, -OH, -NH₂, -halogen, -CN, -NO₂, -R¹³, -OR¹⁴, NHR¹⁴,

or -N(R¹⁴)₂;

R¹¹ is -(C(R¹⁵)₂)_n;



or

each R^{12} is independently -H, -(C₁-C₆)alkyl, -phenyl, -naphthyl; -halogen, or -CN;

R^{13} is -(C₁-C₆)alkyl, -phenyl, or -naphthyl, each of which is unsubstituted or substituted with one or more -(C₁-C₆)alkyl, -phenyl, or -naphthyl;

R^{14} is defined as above for R^{13} ; and

R^{15} is defined as above for R^1 ;

x is independently carbon or nitrogen atom; and

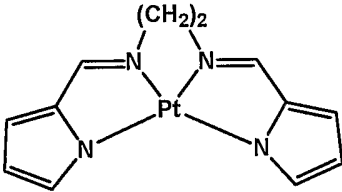
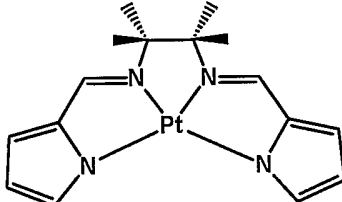
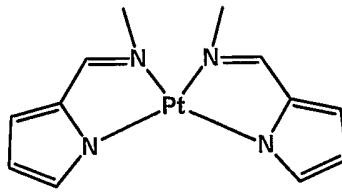
n is an integer from 1 to 6.

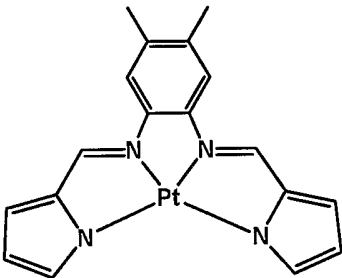
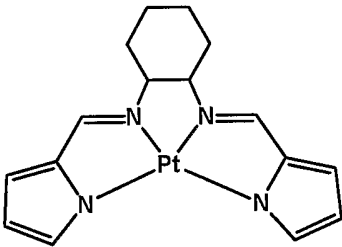
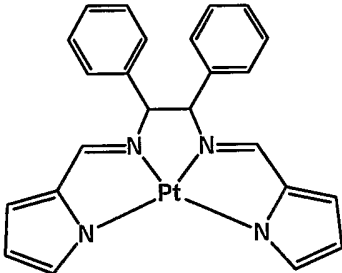
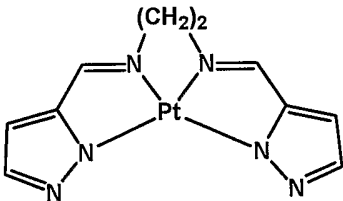
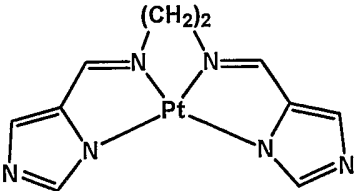
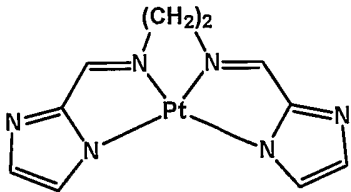
In some embodiments, the compounds of formulae (I), (II), (III) or (IV) may comprise R^1 - R^{10} groups that are electron donors. Non-limiting examples of electron donor groups are amines including $-N(R^{14})_2$ and $-OR^{14}$.

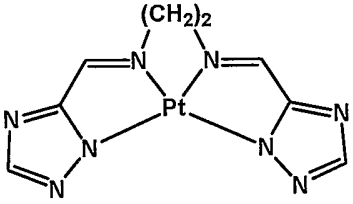
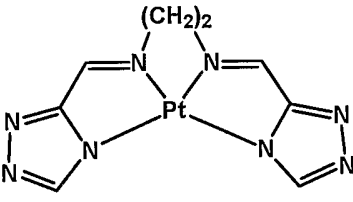
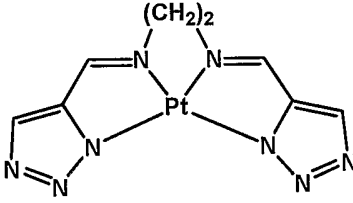
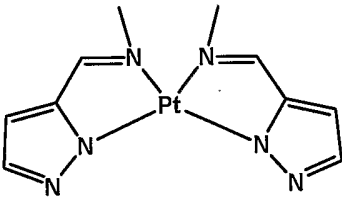
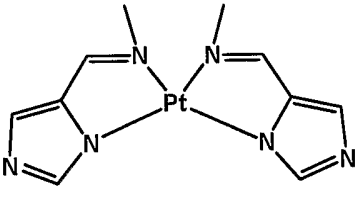
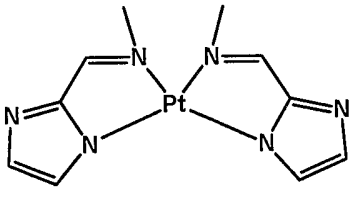
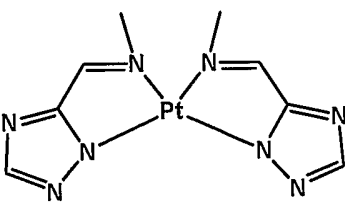
In some embodiments, the compounds of formulae (I), (II), (III) or (IV) may comprise R^1 - R^{10} groups that are electron acceptors. Non-limiting examples of electron acceptor groups include $-F$, $-Cl$, $-Br$, $-I$, $-NO_2$, $-C(O)(C_1-C_6)$, $-C(O)O(C_1-C_6)$, $-SCN$, $-SO_3F$ and $-CN$.

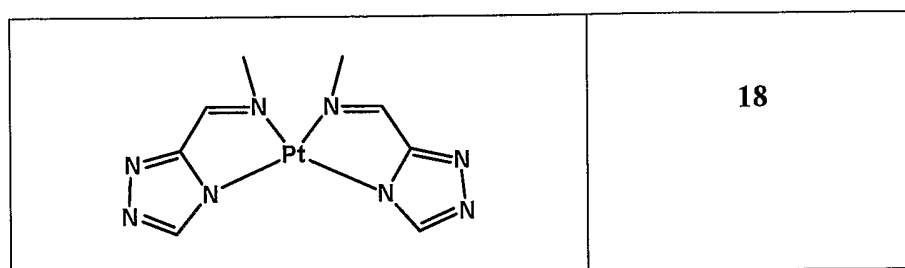
Some illustrative examples and exemplary compounds of formulae (I) and (II) are listed below in Table 1:

Table 1

Structure	Dopant Complex
	1
	2
	3

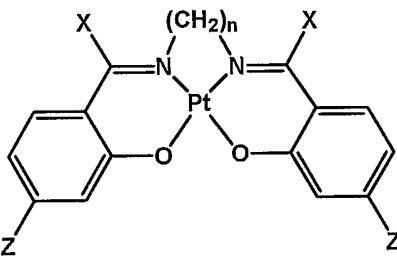
	4
	5
	6
	8
	9
	10

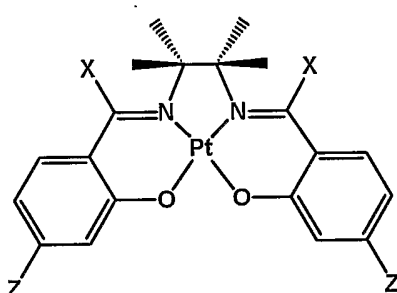
	11
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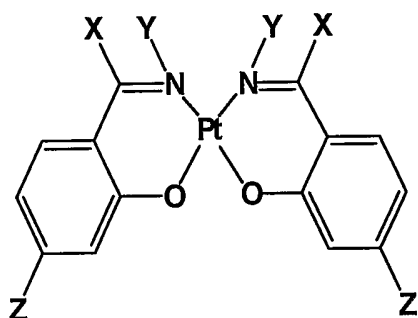


Some illustrative examples and exemplary compounds of formulas (III) and (IV) are listed below in Table 2:

Table 2

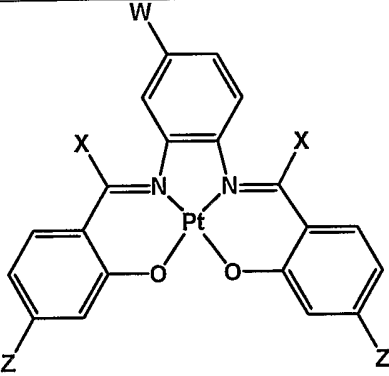
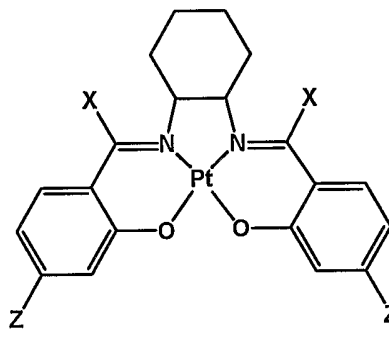
Structure	Dopant Complex
	19: $n = 2$, $X = H$, $Z = H$
	20: $n = 2$, $X = H$, $Z = Cl$
	21: $n = 2$, $X = H$, $Z = Br$
	22: $n = 2$, $X = H$, $Z = F$
	23: $n = 2$, $X = H$, $Z = I$
	24: $n = 2$, $X = H$, $Z = CH_3$
	25: $n = 2$, $X = H$, $Z = t\text{-butyl}$
	26: $n = 2$, $X = H$, $Z = NO_2$
	27: $n = 3$, $X = H$, $Z = H$
	28: $n = 3$, $X = H$, $Z = Cl$
	29: $n = 3$, $X = H$, $Z = Br$
	30: $n = 3$, $X = H$, $Z = F$
	31: $n = 3$, $X = H$, $Z = I$
	32: $n = 3$, $X = H$, $Z = CH_3$
	33: $n = 3$, $X = H$, $Z = t\text{-butyl}$
	34: $n = 3$, $X = H$, $Z = NO_2$
	35: $n = 2$, $X = CH_3$, $Z = H$
	36: $n = 2$, $X = CH_3$, $Z = Cl$
	37: $n = 2$, $X = CH_3$, $Z = Br$
	38: $n = 2$, $X = CH_3$, $Z = F$
	39: $n = 2$, $X = CH_3$, $Z = I$
	40: $n = 2$, $X = CH_3$, $Z = CH_3$
	41: $n = 2$, $X = CH_3$, $Z = t\text{-butyl}$
	42: $n = 2$, $X = CH_3$, $Z = NO_2$
	43: $n = 3$, $X = CH_3$, $Z = H$
	44: $n = 3$, $X = CH_3$, $Z = Cl$
	45: $n = 3$, $X = CH_3$, $Z = Br$
	46: $n = 3$, $X = CH_3$, $Z = F$
	47: $n = 3$, $X = CH_3$, $Z = I$
	48: $n = 3$, $X = CH_3$, $Z = CH_3$
	49: $n = 3$, $X = CH_3$, $Z = t\text{-butyl}$

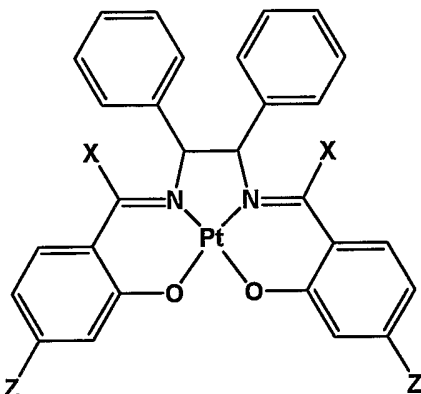
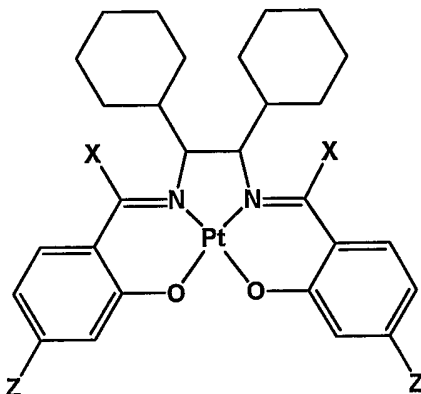
	50: $n = 3$, $X = \text{CH}_3$, $Z = \text{NO}_2$
	51: $X = \text{H}$, $Z = \text{H}$ 52: $X = \text{H}$, $Z = \text{Cl}$ 53: $X = \text{H}$, $Z = \text{Br}$ 54: $X = \text{H}$, $Z = \text{F}$ 55: $X = \text{H}$, $Z = \text{I}$ 56: $X = \text{H}$, $Z = \text{CH}_3$ 57: $X = \text{H}$, $Z = \text{t-butyl}$ 58: $X = \text{H}$, $Z = \text{NO}_2$ 59: $X = \text{CH}_3$, $Z = \text{H}$ 60: $X = \text{CH}_3$, $Z = \text{Cl}$ 61: $X = \text{CH}_3$, $Z = \text{Br}$ 62: $X = \text{CH}_3$, $Z = \text{F}$ 63: $X = \text{CH}_3$, $Z = \text{I}$ 64: $X = \text{CH}_3$, $Z = \text{CH}_3$ 65: $X = \text{CH}_3$, $Z = \text{t-butyl}$ 66: $X = \text{CH}_3$, $Z = \text{NO}_2$
	67: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{H}$ 68: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{Cl}$ 69: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{Br}$ 70: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{F}$ 71: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{I}$ 72: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{CH}_3$ 73: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{t-butyl}$ 74: $X = \text{H}$, $Y = \text{CH}_3$, $Z = \text{NO}_2$ 75: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{H}$ 76: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{Cl}$ 77: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{Br}$ 78: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{F}$ 79: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{I}$ 80: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{CH}_3$ 81: $X = \text{CH}_3$, $Y = \text{CH}_3$, $Z = \text{t-butyl}$



- 82: X = CH₃, Y = CH₃, Z = NO₂
 83: X = CH₃, Y = CN, Z = H
 84: X = CH₃, Y = CN, Z = Cl
 85: X = CH₃, Y = CN, Z = Br
 86: X = CH₃, Y = CN, Z = F
 87: X = CH₃, Y = CN, Z = I
 88: X = CH₃, Y = CN, Z = CH₃
 89: X = CH₃, Y = CN, Z = t-butyl
 90: X = CH₃, Y = CN, Z = NO₂
 91: X = H, Y = CN, Z = H
 92: X = H, Y = CN, Z = Cl
 93: X = H, Y = CN, Z = Br
 94: X = H, Y = CN, Z = F
 95: X = H, Y = CN, Z = I
 96: X = H, Y = CN, Z = CH₃
 97: X = H, Y = CN, Z = t-butyl
 98: X = H, Y = CN, Z = NO₂

- 99: X = H, W = H, Z = H
 100: X = H, W = H, Z = Cl
 101: X = H, W = H, Z = Br
 102: X = H, W = H, Z = F
 103: X = H, W = H, Z = I
 104: X = H, W = H, Z = CH₃
 105: X = H, W = H, Z = t-butyl
 106: X = H, W = H, Z = NO₂
 107: X = CH₃, W = H, Z = H
 108: X = CH₃, W = H, Z = Cl
 109: X = CH₃, W = H, Z = Br
 110: X = CH₃, W = H, Z = F
 111: X = CH₃, W = H, Z = I
 112: X = CH₃, W = H, Z = CH₃
 113: X = CH₃, W = H, Z = t-butyl

	114: X = CH₃, W = H, Z = NO₂ 115: X = H, W = CH₃, Z = H 116: X = H, W = CH₃, Z = Cl 117: X = H, W = CH₃, Z = Br 118: X = H, W = CH₃, Z = F 119: X = H, W = CH₃, Z = I 120: X = H, W = CH₃, Z = CH₃ 121: X = H, W = CH₃, Z = t-butyl 122: X = H, W = CH₃, Z = NO₂ 123: X = CH₃, W = CH₃, Z = H 124: X = CH₃, W = CH₃, Z = Cl 125: X = CH₃, W = CH₃, Z = Br 126: X = CH₃, W = CH₃, Z = F 127: X = CH₃, W = CH₃, Z = I 128: X = CH₃, W = CH₃, Z = CH₃ 129: X = CH₃, W = CH₃, Z = t-butyl 130: X = CH₃, W = CH₃, Z = NO₂
	131: X = H, Z = H 132: X = H, Z = Cl 133: X = H, Z = Br 134: X = H, Z = F 135: X = H, Z = I 136: X = H, Z = CH₃ 137: X = H, Z = t-butyl 138: X = H, Z = NO₂ 139: X = CH₃, Z = H 140: X = CH₃, Z = Cl 141: X = CH₃, Z = Br 142: X = CH₃, Z = F 143: X = CH₃, Z = I 144: X = CH₃, Z = CH₃ 145: X = CH₃, Z = t-butyl 146: X = CH₃, Z = NO₂

	147: X = H, Z = H 148: X = H, Z = Cl 149: X = H, Z = Br 150: X = H, Z = F 151: X = H, Z = I 152: X = H, Z = CH₃ 153: X = H, Z = t-butyl 154: X = H, Z = NO₂ 155: X = CH₃, Z = H 156: X = CH₃, Z = Cl 157: X = CH₃, Z = Br 158: X = CH₃, Z = F 159: X = CH₃, Z = I 160: X = CH₃, Z = CH₃ 161: X = CH₃, Z = t-butyl 162: X = CH₃, Z = NO₂
	163: X = H, Z = H 164: X = H, Z = Cl 165: X = H, Z = Br 166: X = H, Z = F 167: X = H, Z = I 168: X = H, Z = CH₃ 169: X = H, Z = t-butyl 170: X = H, Z = NO₂ 171: X = CH₃, Z = H 172: X = CH₃, Z = Cl 173: X = CH₃, Z = Br 174: X = CH₃, Z = F 175: X = CH₃, Z = I 176: X = CH₃, Z = CH₃ 177: X = CH₃, Z = t-butyl 178: X = CH₃, Z = NO₂

Non-limiting examples of bidentate NN-type ligands include those shown above for dopant complexes **3** and **14** to **18**. For example, NN-type ligands are selected from ligands consisting of at least an unsubstituted 5-membered or 6-membered ring or substituted 5-membered or 6-membered ring; wherein the substituted 5-membered or 6-membered ring includes at least one substituent selected from the groups; a hydrogen, a halogen, a hydroxyl group, an alkyl group, a cycloalkyl group, an aryl group, an acyl group, an alkoxy, an acyloxy group, an amino group, an acyl amino group, an aralkyl group, a cyano group, a carboxyl group, a thio group, a vinyl group, a styryl group, an aminocarbonyl group, a carbonyl group, an aranyl group, an aryloxycarbonyl group, a xylyloxycarbonyl group, a phenoxycarbonyl group or an alkoxycarbonyl group as well as recognized donor or acceptor groups; wherein the substituents, for example, an aryl group, may combine together to form a substituted or unsubstituted, saturated or unsaturated ring with any number of members. In a preferred embodiment, the transition metal is Pt.

Non-limiting examples of bidentate NO-type ligands include those shown above for dopant complexes **67** to **98**. For example, NO-type ligands are selected from ligands consisting of at least an un-substituted 6-membered or 5-membered ring or substituted 6-membered or 5-membered ring; wherein the substituted 6-membered or 5-membered ring includes at least one substituent selected from the groups; a hydrogen, a halogen, a hydroxyl group, an alkyl group, a cycloalkyl group, an aryl group, an acyl group, an alkoxy, an acyloxy group, an amino group, an acyl amino group, an aralkyl group, a cyano group, a carboxyl group, a thio group, a vinyl group, a styryl group, an aminocarbonyl group, a carbonyl group, an aranyl group, an aryloxycarbonyl group, a xylyloxycarbonyl group, a phenoxycarbonyl group or an alkoxycarbonyl group as

well as recognized donor or acceptor groups; wherein the substituents, for example, an aryl group, may combine together to form a substituted or unsubstituted, saturated or unsaturated ring with any number of members. In a preferred embodiment, the transition metal is Pt.

Non-limiting examples NNNN-type ligands include those shown above for dopant complexes **1-2** and **4-13**. For example, NNNN-type ligands are selected from ligands consisting of at least an unsubstituted 5-membered or 6-membered ring or substituted 5-membered or 6-membered ring; wherein those substituted 5-membered or 6-membered ring includes at least a substituent selected from the groups; a hydrogen, a halogen, a hydroxyl group, an alkyl group, a cycloalkyl group, an aryl group, an acyl group, an alkoxy, an acyloxy group, an amino group, an acyl amino group, an aralkyl group, a cyano group, a carboxyl group, a thio group, a vinyl group, a styryl group, an aminocarbonyl group, a carbonyl group, an aranyl group, an aryloxycarbonyl group, a xylyloxycarbonyl group, a phenoxycarbonyl group or an alkoxycarbonyl group as well as recognized donor or acceptor groups; wherein the substituents, for example, an aryl group, may combine together to form a substituted or unsubstituted, saturated or unsaturated ring with any number of members. In a preferred embodiment, the transition metal is Pt.

Non-limiting examples NOON-type ligands include those shown above for dopant complexes **19-66** and **99-178**. For example, NOON-type ligands are selected from ligands consisting of at least an unsubstituted 6-membered or 5-membered ring or substituted 6-membered or 5-membered ring; wherein those substituted 6-membered or 5-membered ring includes at least a substituent selected from the groups; a hydrogen, a halogen, a hydroxyl group, an alkyl group, a cycloalkyl group, an aryl group, an acyl group, an alkoxy, an acyloxy group, an amino group, an acyl amino

group, an aralkyl group, a cyano group, a carboxyl group, a thio group, a vinyl group, a styryl group, an aminocarbonyl group, a carbonyl group, an aranyl group, an aryloxycarbonyl group, a xilyloxycarbonyl group, a phenoxycarbonyl group or an alkoxycarbonyl group as well as recognized donor or acceptor groups; wherein the substituents, for example, an aryl group, may combine together to form a substituted or unsubstituted, saturated or unsaturated ring with any number of members. In a preferred embodiment, the transition metal is Pt.

The present invention is also directed to methods for preparation of OLEDs that can be fabricated by a vapor deposition process.

In one embodiment, OLEDs contain an anode, a hole-transporting layer, an emissive layer comprising at least one host material and at least one dopant complex, a hole-blocking layer, an electron-transporting layer, a charge injection layer and a cathode.

Non-limiting examples of an anode useful for OLEDs are indium-tin-oxide (ITO) and doped polyaniline.

Non-limiting examples of hole-transporting materials useful in the present invention are beryllium bis(2-(2'-hydroxyphenyl)pyridine, 4,4'-bis(carbazol-9-yl)biphenyl (CBP), *N,N*-diphenyl-*N,N*-bis(1-naphthalene)benzidine (α -NPB), *N,N*-diphenyl-*N,N*-bis(2-naphthalene)benzidine (β -NPB), *N,N*-bis(3-methylphenyl)-*N,N*-bis(phenyl)benzidine (TPD), 4,4',4''-tris(*N*-3-methylphenyl-*N*-phenylamino)triphenylamine (m-TDATA) and tetrakis(diaryl-amino)-9,9'-spirobifluorenes.

Non-limiting examples of host materials useful in the present invention include beryllium bis(2-(2'-hydroxyphenyl)pyridine, 4,4'-bis(carbazol-9-yl)biphenyl (CBP), *N,N*-diphenyl-*N,N*-bis(1-naphthalene)benzidine (α -NPB), *N,N*-diphenyl-

N,N-bis(2-naphthalene)benzidine (β -NPB), *N,N'*-bis(3-methylphenyl)-*N,N'*-bis(phenyl)benzidine (TPD), 4,4',4''-tris(*N*-3-methylphenyl-*N*-phenylamino)triphenylamine (m-TDATA) tetrakis(diarylamino)-9,9'-spirobifluorenes, beryllium bis(2-(2'-hydroxyphenyl)pyridine (Bepp₂), 3-phenyl-4-(1'-naphthyl)-5-phenyl-1,2,4-triazole (TAZ); 2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline (BCP), 1,3-bis(*N,N*-*t*-butyl-phenyl)-1,3,4-oxadiazole (OXD7), and 1,3,5-tris(3-methyldiphenylamino)benzene (m-MTDAB).

In this invention, at least one suitable host material was employed in an emissive layer together with at least one dopant complex.

Non-limiting examples of dopant complexes, of either geometrical isomers, comprising a transition metal coordinated to two bidentate NN-type ligands, or two bidentate NO-type ligands, or a tetradentate NNNN-type ligand, or a tetradentate NOON-type ligand include those shown for dopant complexes **1-178** in Table 1 and 2 above. In a preferred embodiment, the dopant complexes are selected from the groups consisting of dopant complexes **1-18, 19, 22, 24-25, 27, 30, 32-33, 35, 38, 40-41, 43, 46, 48-49, 51, 54, 56-57, 59, 62, 64-65, 99, 102, 104-105, 107, 110, 112-113, 115, 118, 120-121, 123, 126, 128-129** and mixtures thereof. More preferably, the dopant complexes are **1, 2, 4, 19, 35, 51, 99** and **104**.

Non-limiting examples of hole-blocking layer suitable for the present invention include 3,4,5-triphenyl-1,2,4-triazole, 3-(biphenyl-4-yl)-4-phenyl-5-(4-tert-butylphenyl)-1,2,4-triazole (TAZ), 2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline (BCP) and 1,3,5-tris[5-(4-(1,1-dimethylethyl)phenyl)-1,3,4-oxadiazol-2-yl]benzene (TBOP).

Non-limiting examples of electron-transporting materials for the present invention include tris(8-hydroxyquinolato) aluminum (Alq_3) and 2-(4-biphenyl)-5-(*p-tert*-butylphenyl)-1,3,4-oxadiazole.

Non-limiting examples of charge injection layer suitable for the present invention include lithium fluoride, cesium fluoride and lithium benzoate.

Non-limiting examples of low work function metals for use as cathode in the present invention include aluminum, potassium, lithium, magnesium, silver, gold, rubidium, beryllium and cesium.

In one preferred embodiment, the, OLED described herein comprises heterostructures for producing electroluminescence which contain an anode (ITO glass substance), a hole-transporting layer (N,N' -diphenyl- N,N' -bis(2-naphthalene)benzidine (β -NPB)), a matrix emissive layer including a host material 4,4'-bis(carbazol-9-yl)biphenyl (CBP)) and an electrophosphorescent dopant complex as illustrated in Formulae **I**, **II**, **III** or **IV** herein, a hole-blocking layer (2,9-dimethyl-4,7-diphenyl-1,10-phenanthroline (BCP)), an electron-transporting layer (tris(8-hydroxyquinolato) aluminum (Alq_3)), a charge injection layer (lithium fluoride) and a cathode (aluminum metal).

Preferably, in present invention, OLEDs comprising dopant complexes as illustrated in Formula **I**, **II**, **III** or **IV** herein exhibit red, orange or yellow electroluminescence. The concentration of the dopant complexes in the emissive layer can range from 0.5 to 8.0 wt.% based on the efficiency of energy conversion between dopant complexes and host materials and molecular structure of dopant complexes. However, other concentrations can be used.

The following examples are set forth to aid in understanding of the inventions but

are not intended to, and should not be interpreted to, limit in any way the claimed invention.

EXAMPLE 1

Example 1 shows the synthesis of dopant complex 1. The tetradentate NNNN-type ligand was prepared according to modification of literature procedures (*see Bacchi et al., Inorganica Chimica Acta. 342:229, (2003); Male et al., J. Chem. Soc., Dalton Trans. 2487, (1997)*).

Synthesis of dopant complex 1

Sodium acetate (0.077 g, 0.94 mmol) was suspended in a DMF (10mL) solution of bidentate ligand, N,N-Bis-(1H-pyrrol-2-ylmethylene)-ethane-1,2-diamine (0.1 g, 0.47 mmol). K₂PtCl₄ (0.19 g, 0.47 mmol) dissolved in DMSO (1 mL) was dropwise added to the suspension at 80 °C dropwise. The resulting yellow solution turned orange-red after being stirred at 80 °C for 4 hours. After cooling, distilled water (50 mL) was then added to the orange-red mixture to afford an orange-brown precipitate. The solid product was filtered and washed with H₂O (2 × 10 mL) to give an orange-brown solid, which was then purified by silica gel column chromatography with CH₂Cl₂ as the eluent. Removal of solvent gave an orange solid. Orange red crystals were obtained by slow evaporation of acetonitrile solution of the orange solid. Yield: 42 mg (22 %). ¹H NMR (CDCl₃): δ = 7.67 (s, 2H, HC=N), 7.10 (m, 2H, pyrrole), 6.71 (d, *J* = 3.3 Hz, 2H, pyrrole), 6.29 (dd, *J* = 3.9, 1.8 Hz, 2H, pyrrole), 4.18 (s, 4H, CH₂). ¹³C NMR (CDCl₃): δ = 155.7, 145.4, 137.7, 119.0, 110.7, 59.4. FAB-MS (*m*-NBA) (*m/z*): 407 {*M*⁺}. IR (KBr): ν = 3107, 3095, 3028, 2998, 2913, 1582, 1570 cm⁻¹. Anal. Calcd (%) for C₁₂H₁₂N₄Pt: C, 35.38; H, 2.97; N, 13.75. Found: C, 34.89; H, 2.98; N, 13.29.

EXAMPLE 2

Example 2 shows the photophysical properties of non-limiting illustrative emissive materials corresponding to dopant complexes 1, 2 and 4 of the present invention. The absorption and photoluminescence properties of dopant complexes are provided in Table 3. UV/vis absorption, excitation and emission spectra of dopant complexes 1, 2 and 4 are shown in FIG. 1 to 3 respectively. The photoluminescence (PL) spectrum is substantially independent of excitation wavelength from 300 to 450 nm. At room temperature, strong PL emissions are obtained with quantum yields (Φ) up to 0.110 in CH₃CN. The emission lifetimes of the dopant complexes range from 0.57 to 4.25 μ s.

Table 3. Physical characterization of dopant complexes 1, 2 and 4.

Dopant Complex	$\lambda_{\text{abs, sol.}} \text{ (nm)}$ $[\epsilon \text{ (} 10^{-4} \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1})]^{\text{a}}$	$\lambda_{\text{em, sol.}}^{\text{a}} \text{ (nm)}$	$\tau \text{ (}\mu\text{s)}^{\text{b}} [\Phi_{\text{em, sol.}}]$
1	278 [1.52], 307 [1.47], 317 [1.56], 372 [sh, 1.34], 388 [1.85], 438 [0.45], 459 [sh, 0.36]	566 (max), 613	4.25 [0.097]
2	279 [1.45], 316 [1.64], 367 [1.36], 383 [1.82], 431 [0.48], 448 [sh, 0.41]	563 (max), 606, 656	3.60 [0.110]
4	246 [1.19], 310 [sh, 1.94], 324 [2.57], 379 [1.99], 390 [sh, 1.80], 478 [sh, 1.20], 498 [1.30], 520 [sh, 1.06]	680 (max), 740, 822	0.57 [0.001]

^a UV/vis and PL emission was measured in acetonitrile (CH₃CN)

^b Lifetime was measured at peak maximum

FIG. 1 shows representative UV/vis absorption, excitation and emission spectra of dopant complex 1 in CH₃CN solution. The solution exhibits strong absorption bands ranged from 278 to 388 nm ($\epsilon = 1.52$ to $1.85 \times 10^{-4} \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$) and moderately intense absorption bands from 438 to 459 nm ($\epsilon = 0.45$ to 0.36×10^{-4}

$\text{dm}^3\text{mol}^{-1}\text{cm}^{-1}$). Upon excitation at 459 nm, an orange photoluminescence (PL) emission is obtained with λ_{max} 566 nm and 613 nm. PL quantum yield (Φ) of complex 1 is 0.097.

FIG. 2 depicts the UV/vis, excitation and emission spectra of dopant complex 2 in CH_3CN solution. The solution exhibits several vibronic absorption transitions ranged from 279 to 383 nm ($\epsilon = 1.45$ to $1.82 \times 10^4 \text{ dm}^3\text{mol}^{-1}\text{cm}^{-1}$) and moderately intense absorption bands from 431 to 448 nm ($\epsilon = 0.48$ to $0.41 \times 10^4 \text{ dm}^3\text{mol}^{-1}\text{cm}^{-1}$). The dopant complex 2 exhibits an orange PL emission bands at λ_{max} 563 nm and 656 nm while the excitation wavelength is at 431 nm. PL quantum yield (Φ) of complex 2 is 0.110.

FIG. 3 shows the UV/vis absorption, excitation and emission spectra of dopant complex 4 in CH_3CN solution. The UV/vis absorption spectrum shows vibronic absorption transitions ranged 246 to 390 nm ($\epsilon = 1.19$ to $1.80 \times 10^4 \text{ dm}^3\text{mol}^{-1}\text{cm}^{-1}$) and moderately intense absorption bands from 478 to 520 nm ($\epsilon = 1.20$ to $1.06 \times 10^4 \text{ dm}^3\text{mol}^{-1}\text{cm}^{-1}$). The PL spectrum of dopant 4 in CH_3CN shows a deep red emission at λ_{max} 680 nm and 740 nm. PL quantum yield (Φ) of complex 4 is 0.001.

EXAMPLE 3

Example 3 illustrates a non-limiting method for preparing an OLED of the present invention. The electroluminescent devices were prepared on patterned indium-tin-oxide (ITO) glass with a sheet resistance of 20 Ω/square . The glass was cleaned sequentially in detergent solution, deionized water, ethanol and acetone. After the wet-cleaning process, the ITO glass was dried at 130 $^{\circ}\text{C}$ for 1 h and treated in UV ozone cleaner for 10 mins. In the practice of the present invention of this example, the device configuration is ITO/NPB (40 nm)/CBP:X wt.% dopant complex as illustrated in formulae (I), (II), (III) or (IV) (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5

nm)/Al (150 nm); all of the layers were grown sequentially by thermal deposition at a deposition rate of about 0.2 Å/sec or about 5 Å /sec under a vacuum of 1×10^{-6} Torr.

The configuration of OLED in the present invention is schematically shown in FIG. 4. The device has multiple layers as shown. In particular, anode layer **410** preferably comprising indium-tin-oxide is deposited upon substrate layer **405**. The substrate can be glass or other material through which the electroluminescence can traverse. Hole-transporting layer **415** comprising NPB is placed on top of layer **410**. Emissive layer **420** employing CBP host and dopant complex is in contact with hole-transporting layer **415**. A hole-blocking layer **425** containing BCP is deposited on the emissive layer **420**. Adjacent to the hole-blocking layer **425**, an electron-transporting layer **430**, preferably Alq_3 , is placed on it. A charge injection layer **435** comprising LiF is then deposited on the layer **430**. On top of the layer **435**, a cathode layer **440** is fabricated. Preferably, the thickness for NPB is 40 nm (hole-transporting layer **415**) and the emissive layer **420** is about 30 nm thick; the hole-blocking layer **425** is 20 nm and electron-transporting layer **430** is 30 nm. The charge transport layer **435** is 0.5 nm thick and cathode layer **440** is preferably about 150 nm thick. The emissive area of device is $3 \times 3 \text{ mm}^2$, which is defined by overlapping area between cathode and anode. Although not shown, glass substrate **405** need not be flat in all embodiments of the invention. In one embodiment, the glass substrate **405** is shaped, for instance, in a concave shape to focus the light generated in emissive layer **420**, which provides even greater light intensity in a small region. In another embodiment, the glass substrate **405** is shaped, for instance, in a convex shape that spreads the generated light more diffusely.

EXAMPLE 4

Example 4 shows a red OLED A employing dopant complex 1 as dopant in a CBP host. The configuration of device A is ITO/NPB (40 nm)/CBP:4 wt.% dopant 1 (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 4 wt.% dopant concentration, there was a red EL emission with a peak maximum at 620 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.62$ and $y = 0.38$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 6.5%, 9.0 cd/A, 4.0 lm/W and 11 000 cd/m², respectively.

FIG. 5 shows EL spectra of the OLED A with 4.0 wt.% dopant 1 at different dopant concentrations under 8 V.

FIG. 6 shows EL spectra of the OLED A with 4.0 wt.% dopant 1 at different applied voltage.

FIG. 7 depicts V-I-B curve of the OLED A with 4.0 wt.% dopant 1.

FIG. 8 shows external quantum efficiency and luminous efficiency of the OLED A with 4.0 wt.% dopant 1.

Table 4 shows the EL performance of OLED A with 4.0 wt.% dopant 1 at different dopant concentrations.

Table 4. EL performance of OLED A with dopant 1 at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
0.5	3.5	8 030	4.1	7.1	3.1
1.0	3.4	9 700	5.5	7.4	3.5
2.0	3.5	10 050	5.9	8.6	3.8
4.0	3.2	11 000	6.5	9.0	4.0
6.0	3.3	9 500	5.6	7.5	3.3

EXAMPLE 5

Example 5 demonstrates an orange OLED **B** employing dopant complex **2** as dopant in a CBP host. The device configuration is ITO/NPB (40 nm)/CBP:5 wt.% dopant **2** (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 5 wt.% dopant concentration, there was a orange EL emission with a peak maximum and a shoulder at 568 and 616 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.52$ and $y = 0.48$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 4.9%, 13.1 cd/A, 5.9 lm/W and 10120 cd/m², respectively.

FIG. 9 shows the EL spectra of OLED **B** with 5 wt.% dopant **2** at different dopant concentrations at 8 V.

FIG. 10 shows the EL spectra of OLEDs **B** with 5 wt.% dopant **2** at different applied voltages.

FIG. 11 shows the V-I-B curve of OLED **B** with 5 wt.% dopant **2**.

FIG. 12 shows the external quantum efficiency and luminous efficiency of OLED **B** with 5.0 wt.% dopant **2**.

Table 5 shows the EL performance of OLED **B** with dopant **2** at different dopant concentrations.

Table 5. The EL performance of OLED **B**
with dopant **2** at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
1.0	3.3	8 200	3.9	10.2	4.8
3.0	3.2	9 150	4.4	11.9	5.3
5.0	2.9	10 120	4.9	13.1	5.9
8.0	3.0	9 250	4.2	11.6	5.1

EXAMPLE 6

Example 6 shows a yellow OLED C employing dopant complex **19** as dopant in a CBP host. The configuration of device C is ITO/NPB (40 nm)/CBP:3 wt.% dopant **19** (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 3 wt.% dopant concentration, there was a yellow EL emission with a peak maximum at 620 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.49$ and $y = 0.50$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 2.3%, 6.1 cd/A, 2.4 lm/W and 9370 cd/m², respectively.

FIG. 13 shows the EL spectra of OLED C with 3 wt. % dopant **19** at different dopant concentrations at 8 V.

FIG. 14 shows the EL spectra of OLED C with 3 wt.% dopant **19** at different applied voltages.

FIG. 15 shows the V-I-B curve of OLED C with 3 wt.% dopant **19**.

FIG. 16 shows the external quantum efficiency and luminous efficiency of OLED C with 3.0 wt.% dopant **19**.

Table 6 shows the EL performance of OLED C with dopant **19** at different dopant concentrations.

Table 6. EL performance of OLED C with dopant **19** at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
1.0	3.4	9 050	2.2	5.9	2.3
3.0	3.4	9 370	2.3	6.1	2.4
5.0	3.5	6 120	1.4	3.8	1.4
8.0	3.6	3 460	0.81	2.1	0.85

EXAMPLE 7

Example 7 shows a yellow OLED **D** employing dopant complex **51** as dopant in a CBP host. The configuration of device **D** is ITO/NPB (40 nm)/CBP:4 wt.% dopant **51** (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 4 wt.% dopant concentration, there was a yellow EL emission with a peak maximum and a shoulder at 550 and 590 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.48$ and $y = 0.52$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 11 %, 31 cd/A, 14 lm/W and 23000 cd/m², respectively.

FIG. 17 shows the EL spectra of OLED **D** with 4 wt.% dopant **51** at different dopant concentrations at 8 V.

FIG. 18 shows the EL spectra of OLED **D** with 4 wt.% dopant **51** at different applied voltages.

FIG. 19 shows the V-I-B curve of OLED **D** with 4 wt.% dopant **51**.

FIG. 20 shows the external quantum efficiency and luminous efficiency of OLED **D** with 4.0 wt.% dopant **51**.

Table 7 shows the EL performance of OLED **D** with dopant **51** at different dopant concentrations.

Table 7. EL performance of OLED **D** with dopant **51** at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
0.5	3.8	4 500	2.0	2.8	1.2
1.0	3.3	11 000	5.4	14	6.2
2.0	2.9	20 500	10	28	13
4.0	2.8	23 000	11	31	14

EXAMPLE 8

Example 8 shows a red OLED **E** employing dopant complex **99** as dopant in a CBP host. The configuration of device **E** is ITO/NPB (40 nm)/CBP:1.5 wt.% dopant **99** (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 1.5 wt.% dopant concentration, there was a red EL emission with a peak maximum at 636 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.65$ and $y = 0.35$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 9.4 %, 11 cd/A, 4.9 lm/W and 17900 cd/m², respectively.

FIG. 21 shows the EL spectra of OLED **E** with 1.5 wt.% dopant **99** at different dopant concentrations at 8 V.

FIG. 22 shows the EL spectra of OLED **E** with 1.5 wt.% dopant **99** at different applied voltages.

FIG. 23 shows the V-I-B curve of OLED **E** with 1.5 wt.% dopant **99**.

FIG. 24 shows the external quantum efficiency and luminous efficiency of OLED **E** with 1.5 wt.% dopant **99**.

Table 8 shows the EL performance of OLED **E** with dopant **99** at different dopant concentrations.

Table 8. EL performance of OLED **E** with dopant **99** at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
0.5	3.2	12 200	6.1	7.2	3.3
1.0	3.2	14 300	7.3	8.5	3.7
1.5	3.1	17 900	9.4	11	4.9
2.5	3.3	15 100	8.1	9.4	3.8
5.0	3.2	10 500	5.5	6.7	2.5

EXAMPLE 9

Example 9 shows a red OLED **F** employing dopant complex **104** as dopant in a CBP host. The configuration of device **F** is ITO/NPB (40 nm)/CBP:1.6 wt.% dopant **104** (30 nm)/BCP (20 nm)/Alq₃ (30 nm)/LiF (0.5 nm)/Al (150 nm). At 1.6 wt.% dopant concentration, there was a red EL emission with a peak maximum at 628 nm corresponding to the 1931 Commission Internationale de L'Eclairage (CIE_1931) coordinates of $x = 0.64$ and $y = 0.35$ is obtained at 8V. The maximum external quantum efficiency (η_{ext}), luminous efficiency (η_L), power efficiency (η_P) and brightness of the device are 6.4 %, 7.5 cd/A, 3.4 lm/W and 13600 cd/m², respectively.

FIG. 25 shows the EL spectra of OLED **F** with 1.6 wt.% dopant **104** at different dopant concentrations at 8 V.

FIG. 26 shows the EL spectra of OLED **F** with 1.6 wt.% dopant **104** at different applied voltages.

FIG. 27 shows the V-I-B curve of OLED **F** with 1.6 wt.% dopant **104**.

FIG. 28 shows the external quantum efficiency and luminous efficiency of OLED **F** with 1.6 wt.% dopant **104**.

Table 9 shows the EL performance of OLED **F** with dopant **104** at different dopant concentrations.

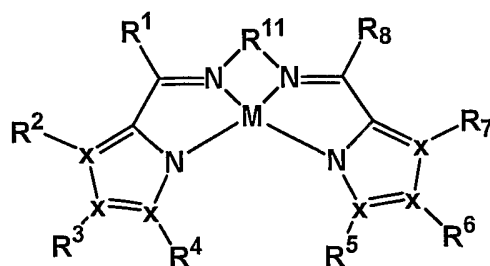
Table 9. EL performance of OLED **F** with dopant **104** at different dopant concentrations

Dopant (wt.%)	V_{on} (V)	B_{max} (cd/m ²)	$\eta_{ext,max}$ (%)	$\eta_{L,max}$ (cd/A)	$\eta_{P,max}$ (lm/W)
1.0	3.1	11 300	5.1	6.1	2.7
1.6	3.0	13 600	6.4	7.5	3.4
2.8	3.1	10 100	4.7	5.5	2.5
5.0	3.0	8 500	4.1	4.7	2.2

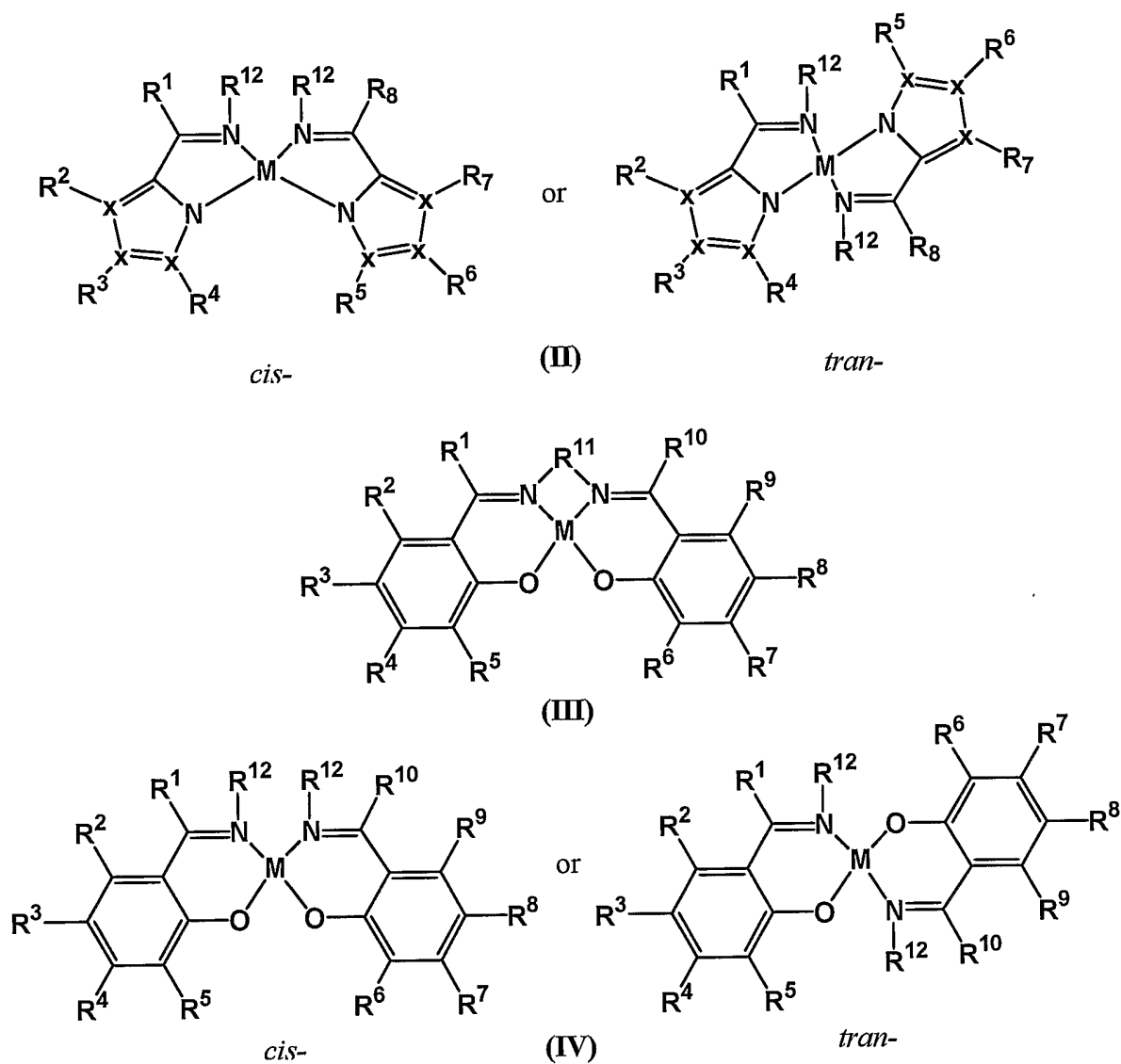
The foregoing description of the preferred embodiments of the present invention has been presented for purposes of illustration and explanation. The various cited references and documents in the preceding description are all incorporated herein by reference in their entirety for all purposes. The description is not intended to be exhaustive nor to limit the invention to the precise form disclosed. As is expected, many modifications and variations will be apparent to those skilled in the art since the embodiments were chosen and described in order to explain the principles of the invention and its practical applications, thereby enabling others skilled in the art to understand the invention. For example, an advantage of the OLEDs of the present invention is that the color of the emitted light may be tuned during fabrication by changing the concentration of the dopant complex. In other embodiments, the color and/or intensity of the emission of the OLEDs of the present invention may be changed by the use of filters, as is known in the art. Various contemplated alternative embodiments and modifications that are suited to a particular use are within the scope of the invention. It is intended that the scope of the invention be defined by the accompanying claims and their equivalents.

WHAT IS CLAIMED IS:

1. A heterostructured organic light-emitting device comprising at least one emissive layer comprising at least one host material and at least one dopant complex, wherein the dopant complex comprises a transition metal atom coordinated to two bidentate NN ligands, or two bidentate NO ligands, or a tetradentate NNNN ligand, or a tetradentate NOON ligand.
2. The heterostructured organic light-emitting device of claim 1, wherein the emissive layer comprises one dopant complex, which dopant complex dopes the host material.
3. The heterostructured organic light-emitting device of claim 2, wherein the dopant complex is geometrically in the *cis*-configuration.
4. The heterostructured organic light-emitting device of claim 2, wherein the dopant complex is geometrically in the *trans*-configuration.
5. The heterostructured organic light-emitting device of claim 1, wherein the emissive layer is a sublimation, vacuum deposition, vapor deposition or spin-coating layer.
6. The heterostructured organic light-emitting device of claim 1, wherein the dopant complex is:



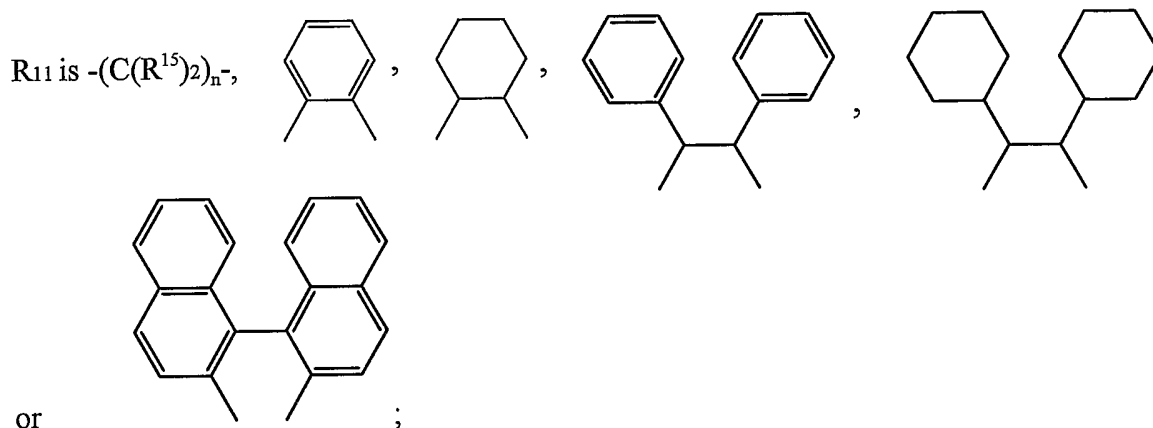
(I)



or a mixture thereof, wherein

M is a transition metal selected from the group consisting of Ni, Pd and Pt;

each R^1-R^{10} is independently -H, -OH, -NH₂, -halogen, -CN, -NO₂, -R¹³, -OR¹⁴, NHR¹⁴,
or -N(R¹⁴)₂;



each R^{12} is independently -H, $-(C_1-C_6)$ alkyl, -phenyl, -naphthyl, -halogen, or -CN;

R^{13} is $-(C_1-C_6)$ alkyl, -phenyl, or -naphthyl, each of which is unsubstituted or substituted with one or more $-(C_1-C_6)$ alkyl, -phenyl, or -naphthyl;

R^{14} is as defined above for R^{13} ;

R^{15} is as defined above for R^1 ;

each x is independently a carbon or nitrogen atom; and

n is an integer from 1 to 6.

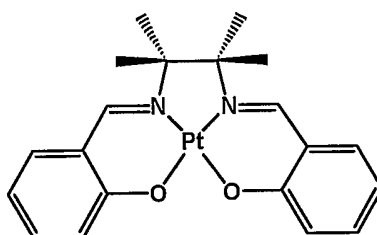
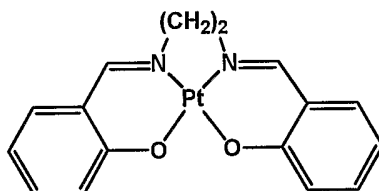
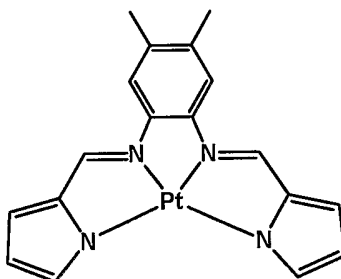
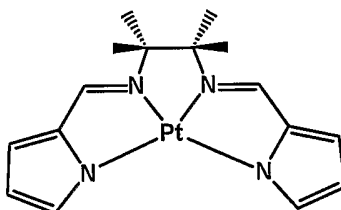
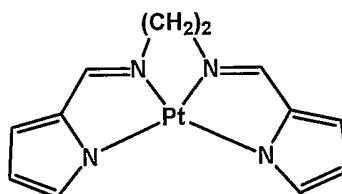
7. The heterostructured organic light-emitting device of claim 6, wherein M is Pt.

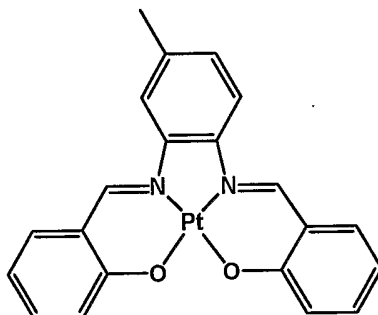
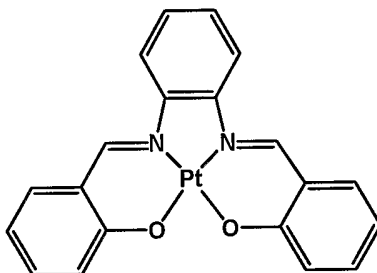
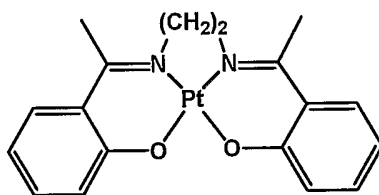
8. The heterostructured organic light-emitting device of claim 7, having structure I or

II.

9. The heterostructured organic light-emitting device of claim 7, having structure III or IV.

10. The heterostructured organic light-emitting device of claim 7, wherein the dopant complex is selected from the group consisting of:





or a mixture thereof.

11. The heterostructured organic light-emitting device of claim 1, wherein the emissive layer comprises 0.5 to 8.0 weight % dopant complex based on the weight of host material.

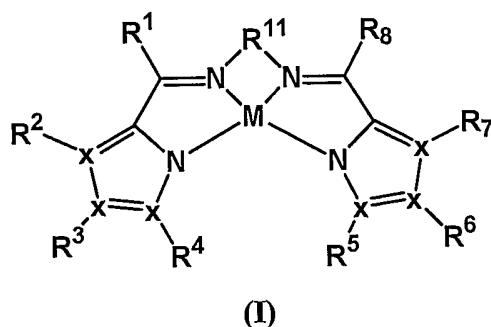
12. The heterostructured organic light-emitting device of claim 1, wherein the emissive layer comprises a dopant complex exhibits an electroluminescence of visible color.

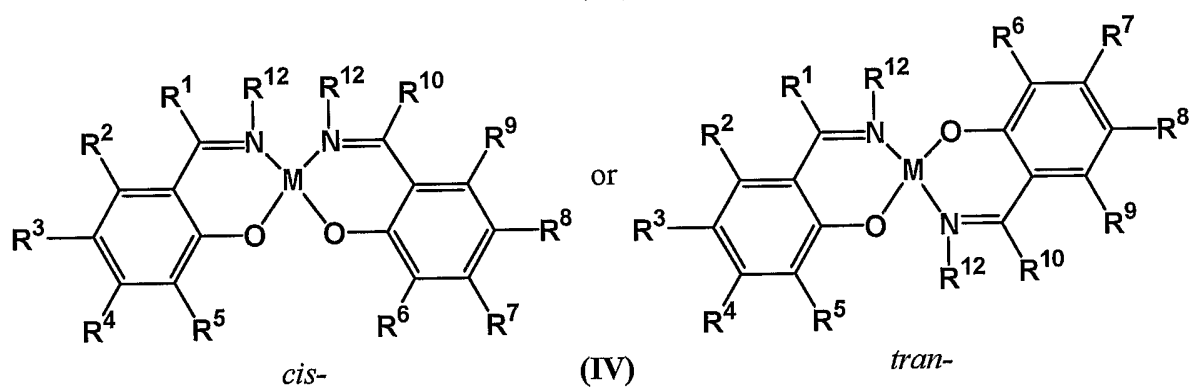
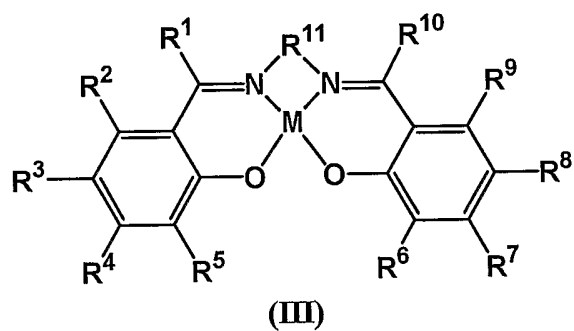
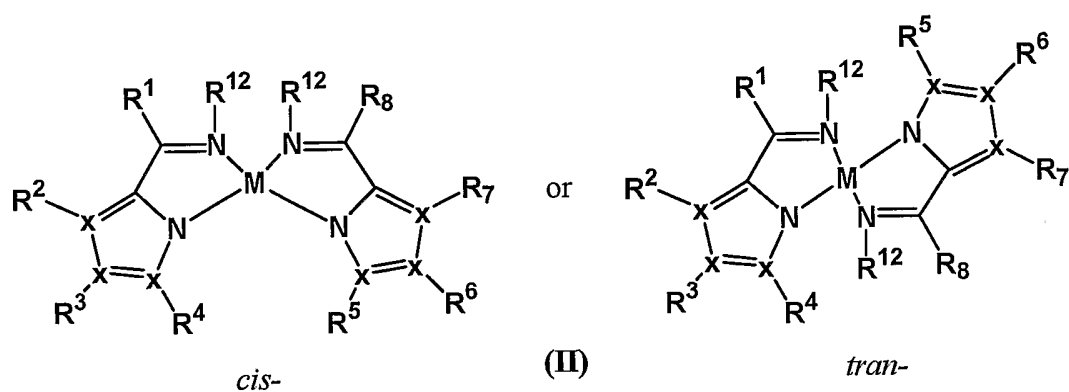
13. The heterostructured organic light-emitting device of claim 1, wherein the emissive layer comprises a dopant complex which exhibits red, orange or yellow electroluminescence.

14. The heterostructured organic light-emitting device of claim 1, wherein the host material is selected from the group consisting of beryllium bis(2-(2'-hydroxyphenyl)pyridine), 4,4'-bis(carbazol-9-yl)biphenyl (CBP), N,N'-diphenyl-N,N'-bis(1-naphthalene)benzidine (α -NPB), N,N'-diphenyl-N,N'-bis(2-naphthalene)benzidine (β -NPB), N,N'-bis(3-methylphenyl)-N,N'-bis(phenyl)benzidine (TPD), 4,4',4''-tris(N-3-methylphenyl-N-phenylamino)triphenylamine (m-TDATA) or tetrakis(diarylamino)-9,9'-spirobifluorene.

15. A method for preparing a heterostructured organic light emitting devices which comprises providing an emissive layer, wherein the emissive layer comprising at least one host material and at least one dopant complex, the dopant complex comprising a transition metal coordinated to two bidentate NN ligands, or two bidentate NO ligands, or a tetradentate NNNN ligand, or a tetradentate NOON ligand.

16. The method of claim 15, wherein the dopant complex in emissive layer is:



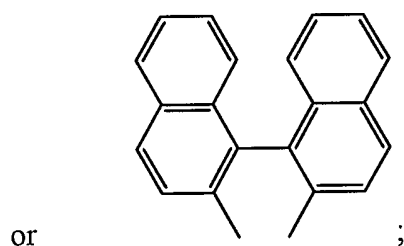
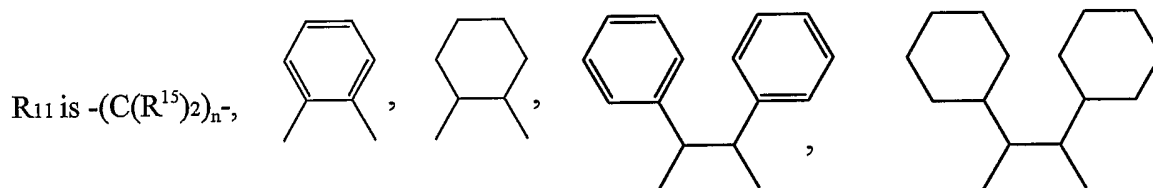


or a mixture thereof, wherein

M is a transition metal selected from the group consisting of Ni, Pd and Pt;

each R^1 - R^{10} is independently -H, -OH, -NH₂, -halogen, -CN, -NO₂, -R¹³, -OR¹⁴, NHR¹⁴,

or -N(R¹⁴)₂;



each R¹² is independently -H, -(C₁-C₆)alkyl, -phenyl, -naphthyl, -halogen, or -CN;

R¹³ is -(C₁-C₆)alkyl, -phenyl, or -naphthyl, each of which is unsubstituted or substituted with one or more -(C₁-C₆)alkyl, -phenyl, or -naphthyl;

R¹⁴ is as defined above for R¹³; and

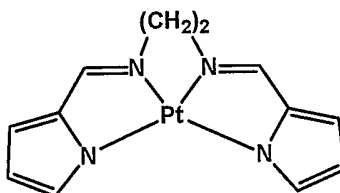
R¹⁵ is as defined above for R¹;

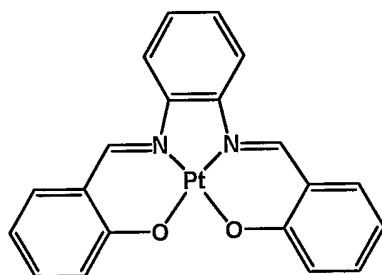
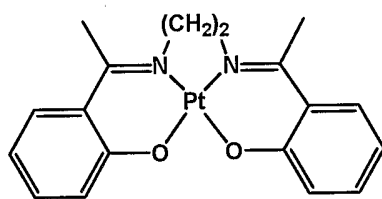
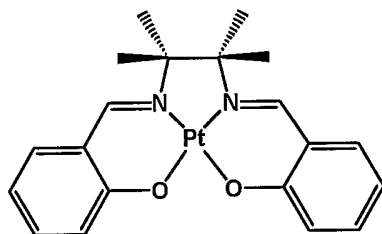
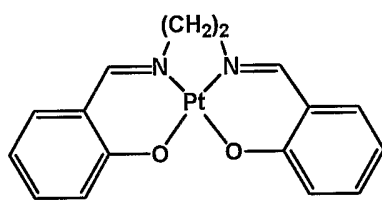
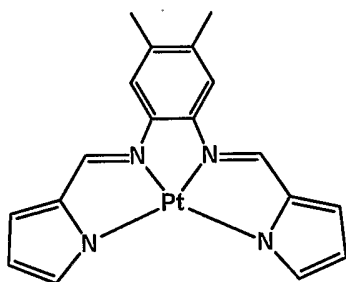
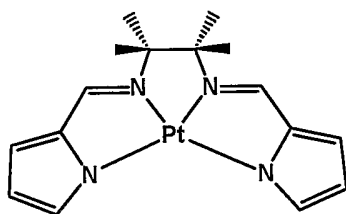
each x is independently a carbon or nitrogen atom; and

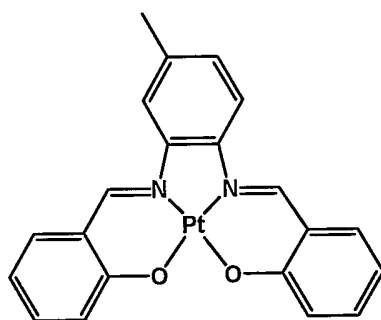
n is an integer number from 1 to 6.

17. The method of claim 15, wherein M is Pt.

18. The method of claim 17, wherein the dopant complex in emissive layer is selected from the group consisting of:







or a mixture thereof.

19. The method of claim 15, wherein the emissive layer comprises 0.5 to 8.0 weight % dopant complex based on weight of host material.

20. The method of claim 15, wherein the emissive layer comprises a dopant complex exhibits an electroluminescence of visible color.

21. The method of claim 20, wherein the emissive layer comprises a dopant complex which exhibits red, orange or yellow electroluminescence.

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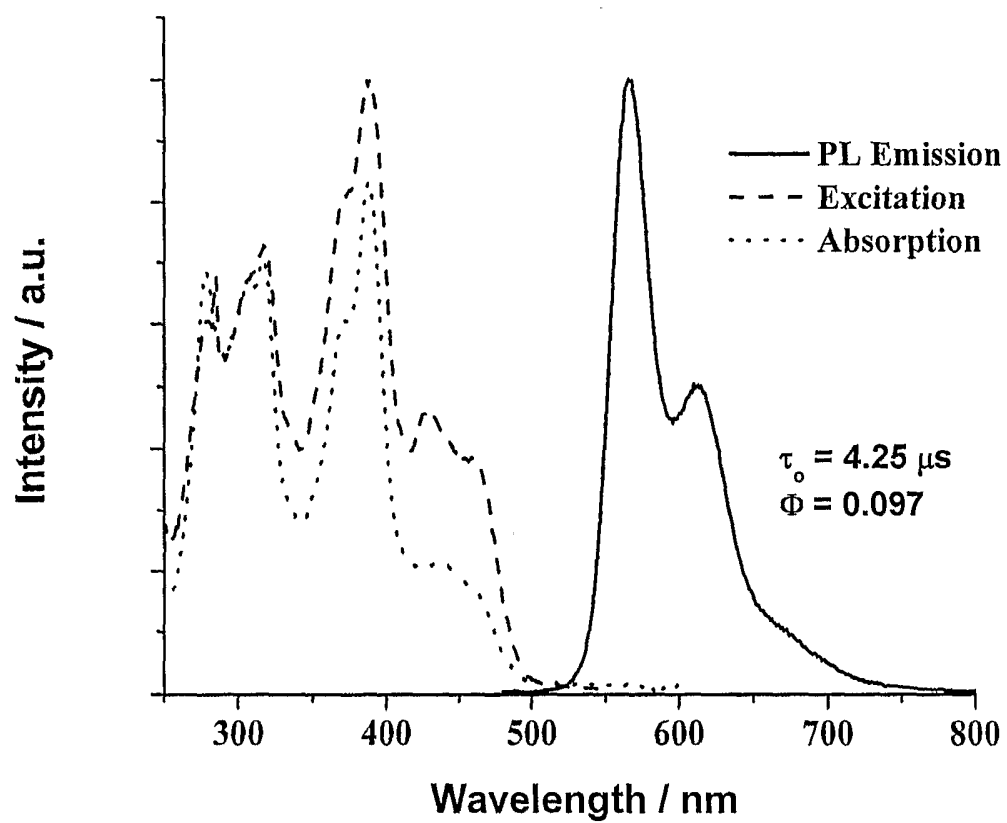


FIG. 1. Absorption, excitation and emission spectra of dopant complex 1 in CH_3CN

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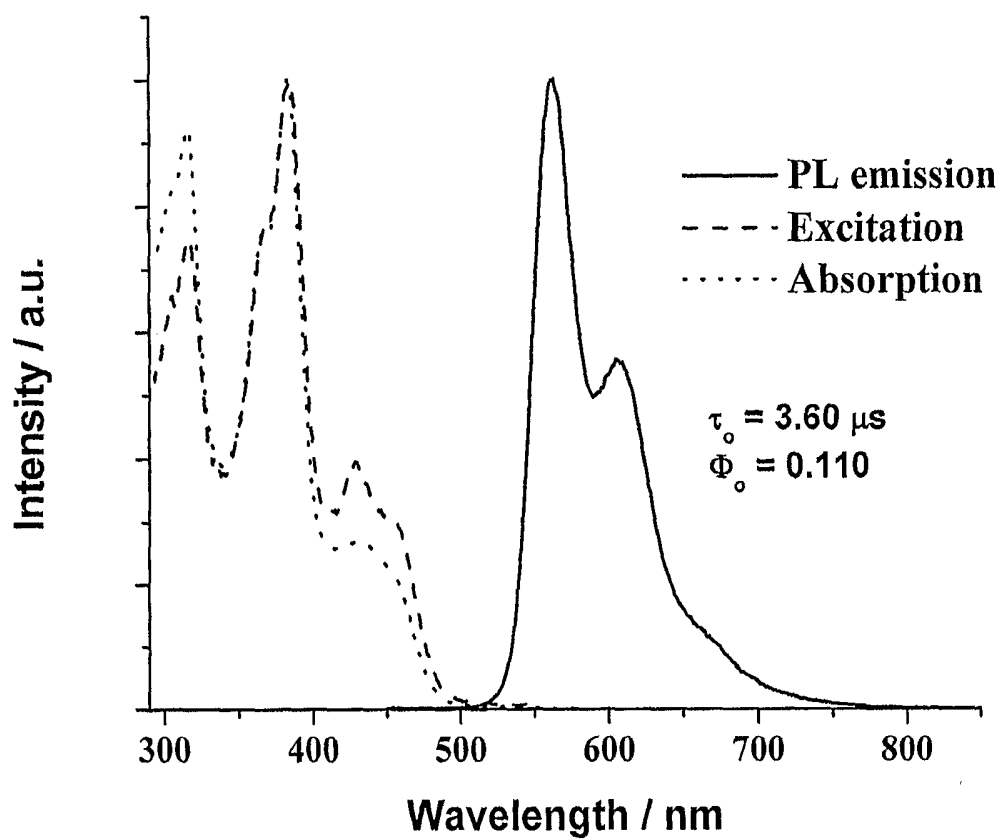


FIG. 2. Absorption, excitation and emission spectra of dopant complex 3 in CH₃CN

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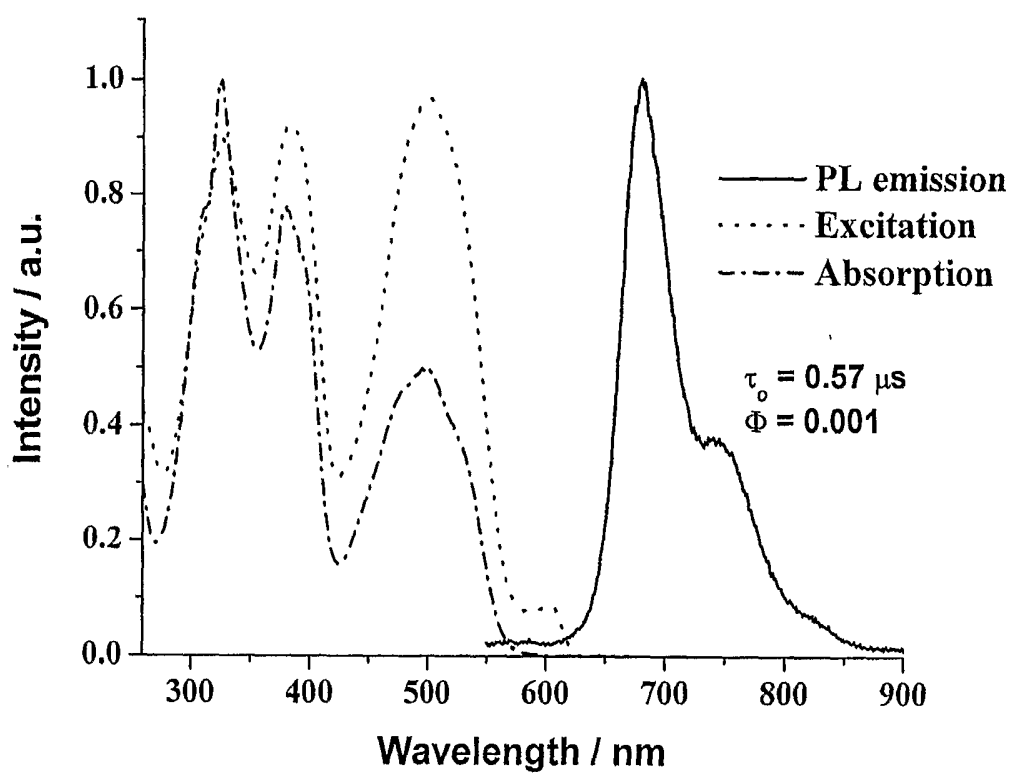


FIG. 3. Absorption, excitation and emission spectra of dopant complex 6 in CH_3CN

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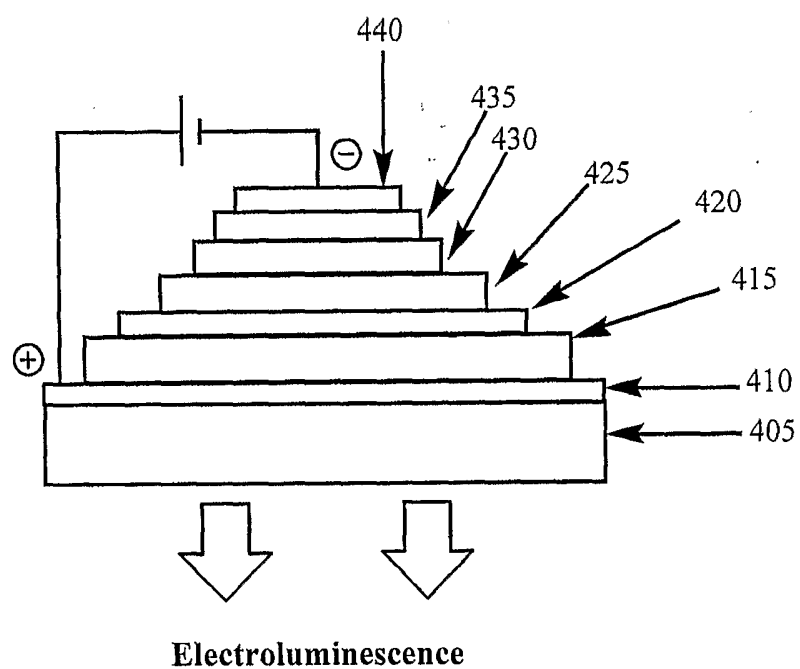


FIG 4. Schematic diagram of OLED in present invention

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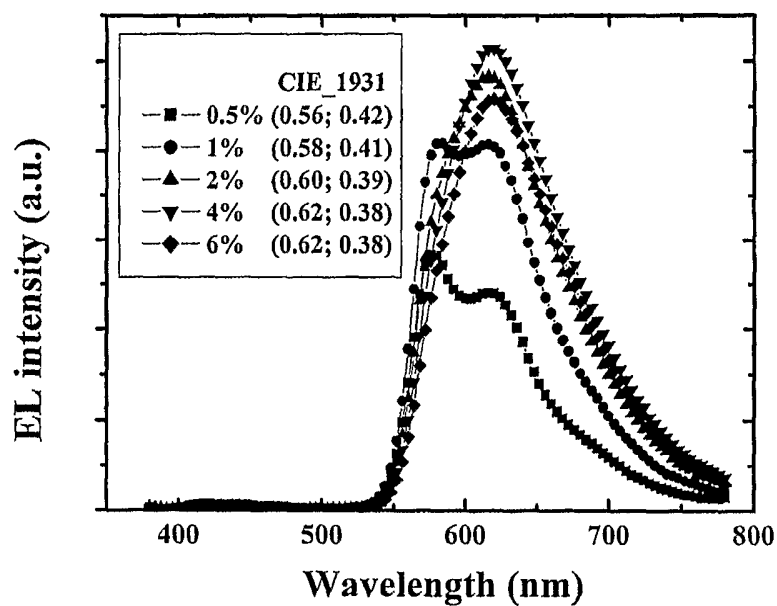


FIG 5. EL spectra of OLED A with 4.0 wt.% dopant 1 at different dopant concentrations under 8 V

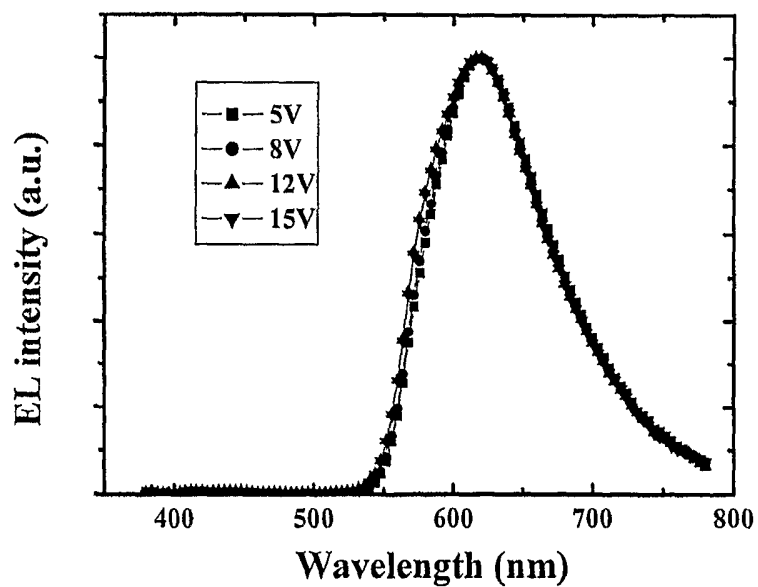


FIG 6. EL spectra of OLED A with 4.0 wt.% dopant 1 at different applied voltages

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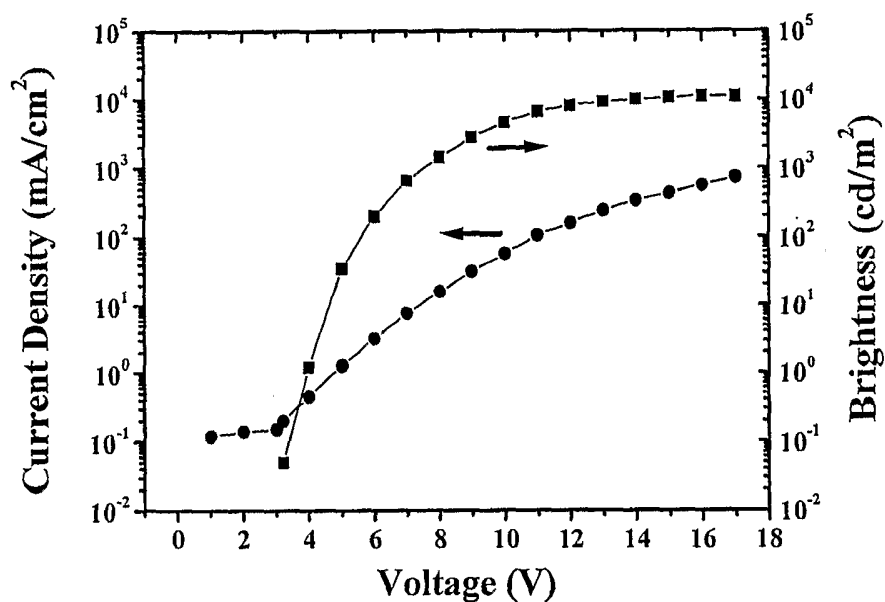


FIG 7. V-I-B curve of OLED A with 4.0 wt.% dopant 1

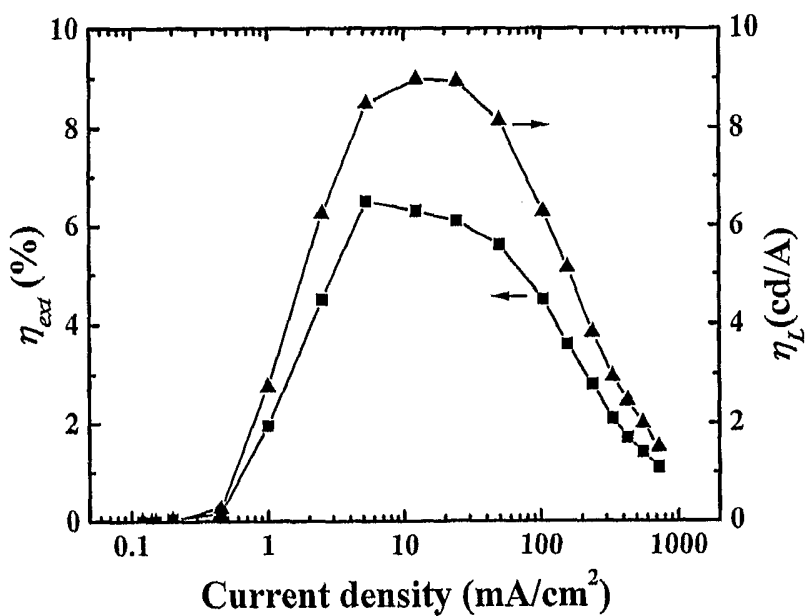


FIG 8. External quantum efficiency and luminous efficiency of OLED A with 4.0 wt.% dopant 1

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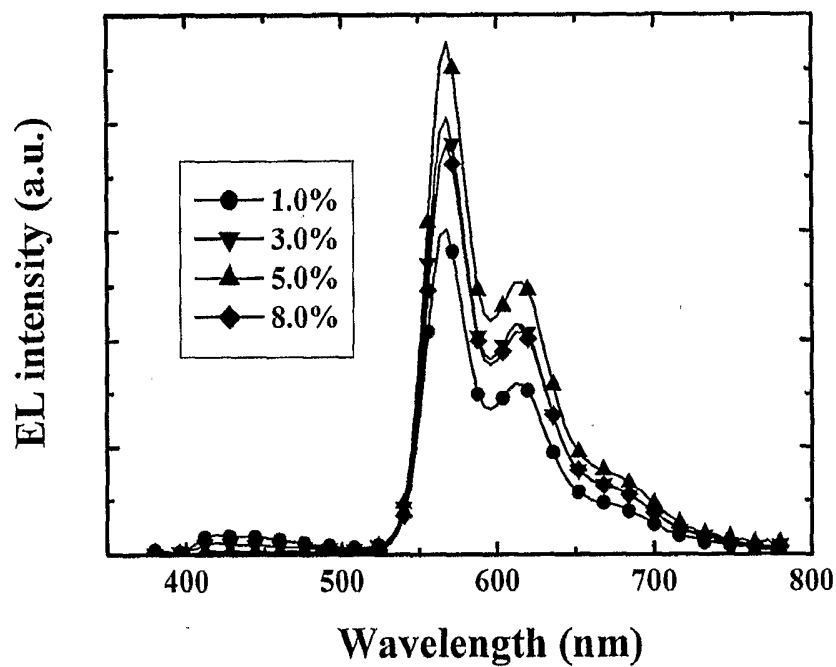


FIG 9. EL spectra of OLED B with 5 wt.% dopant 3 at different dopant concentrations at 8 V

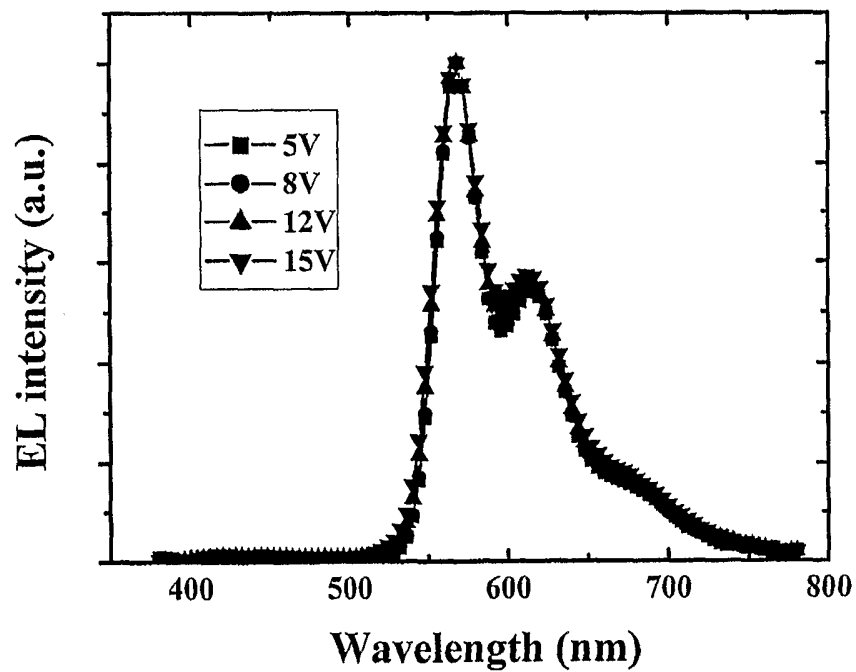


FIG 10. EL spectra of OLED B with 5 wt.% dopant 3 at different applied voltages

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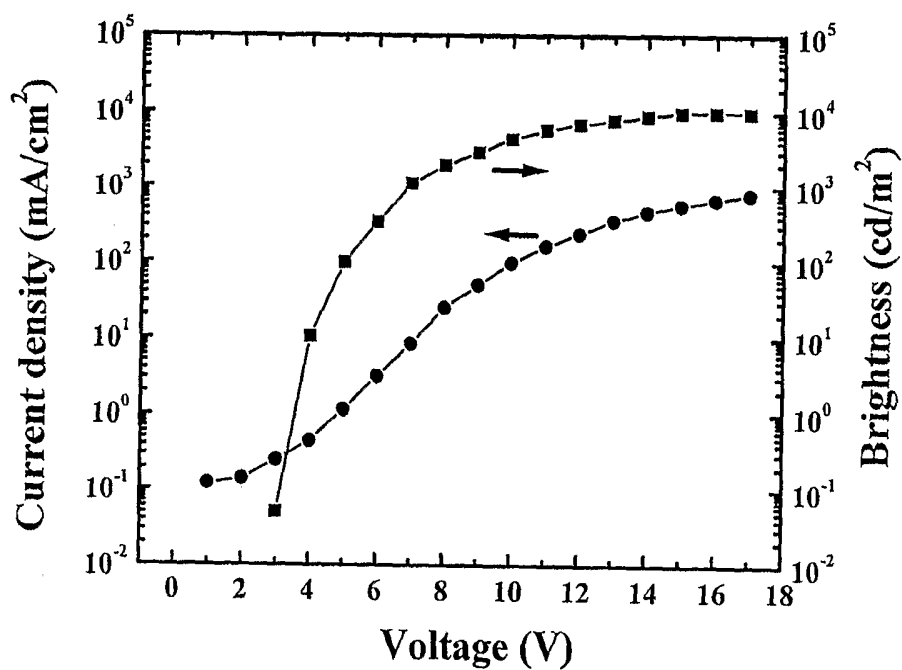


FIG 11. V-I-B curve of OLED B with 5 wt.% dopant 3

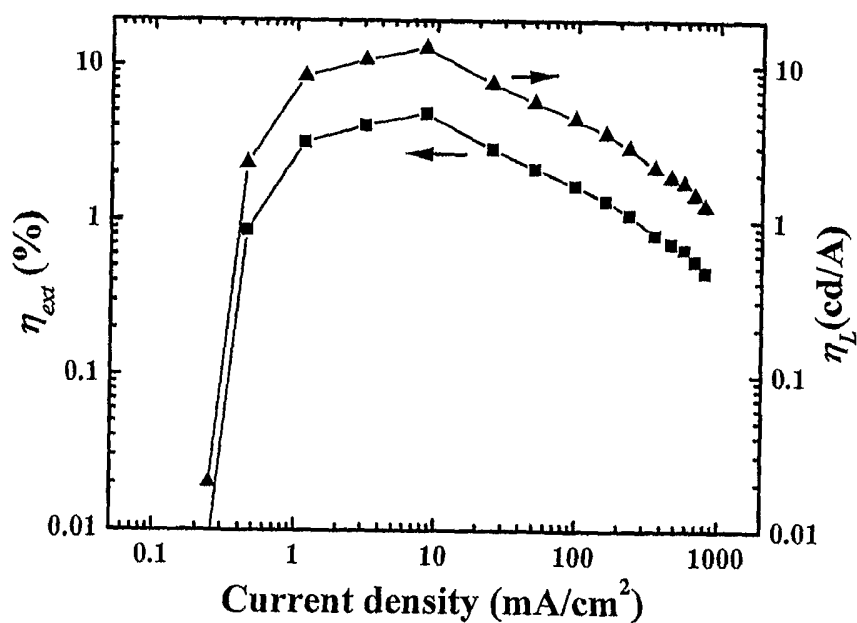


FIG 12. The external quantum efficiency and luminous efficiency of OLED B with 5.0 wt.% dopant 3

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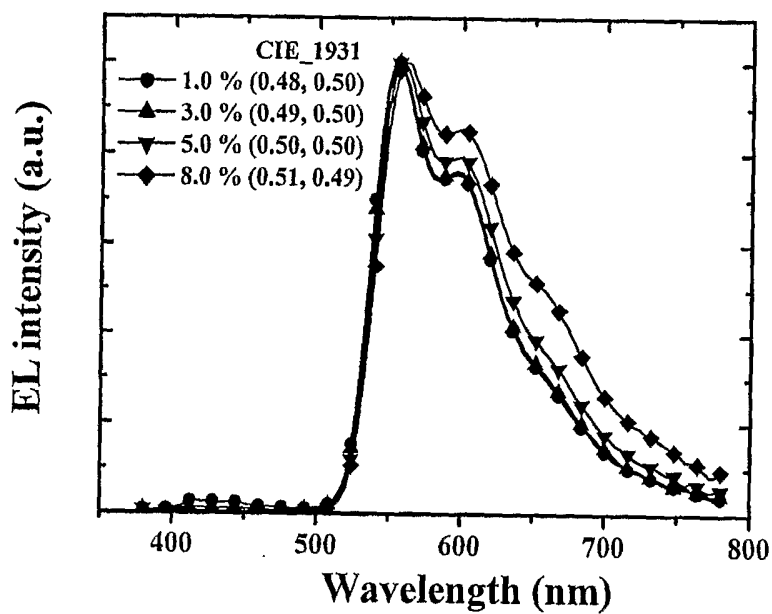


FIG 13. EL spectra of OLED C with 3 wt.% dopant 19 at different dopant concentrations at 8 V

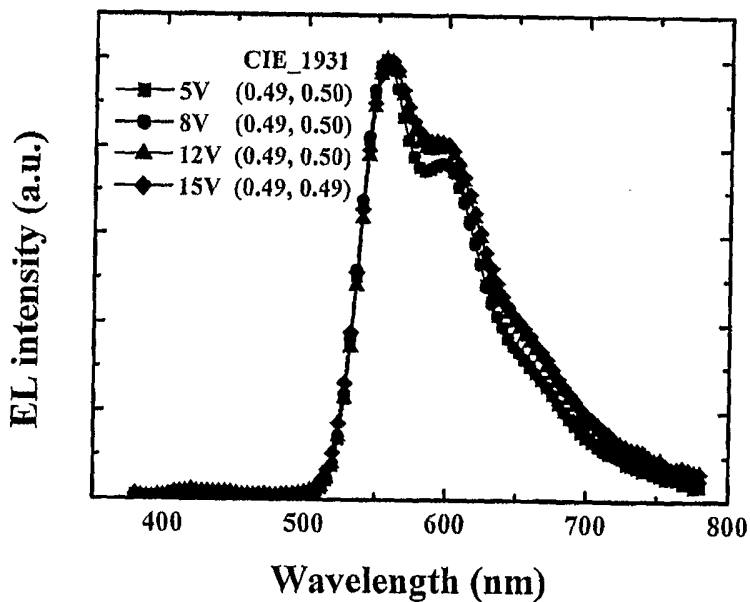


FIG 14. EL spectra of OLED C with 3 wt.% dopant 19 at different applied voltages

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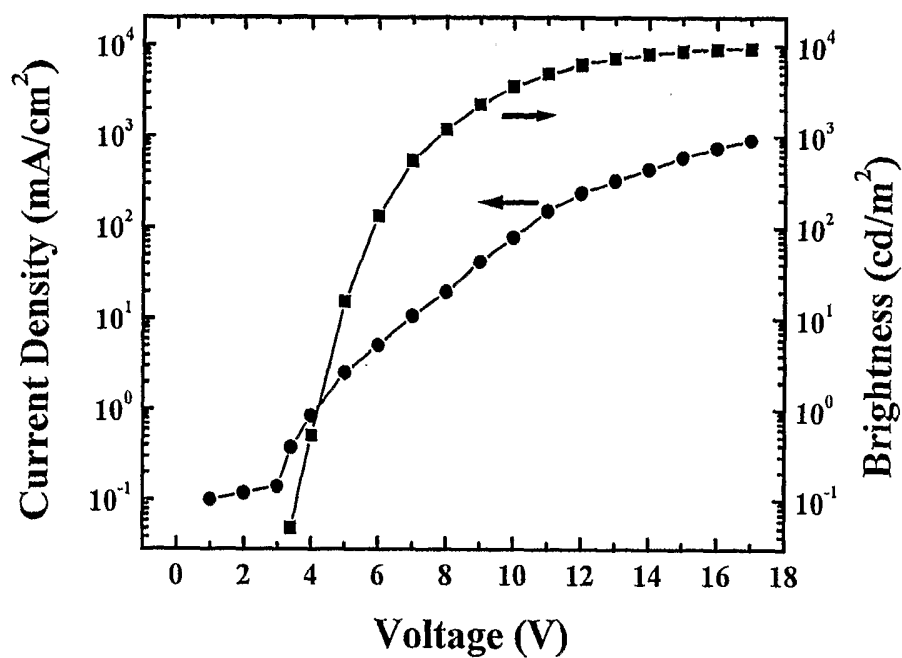


FIG 15. V-I-B curve of OLED C with 3 wt.% dopant 19

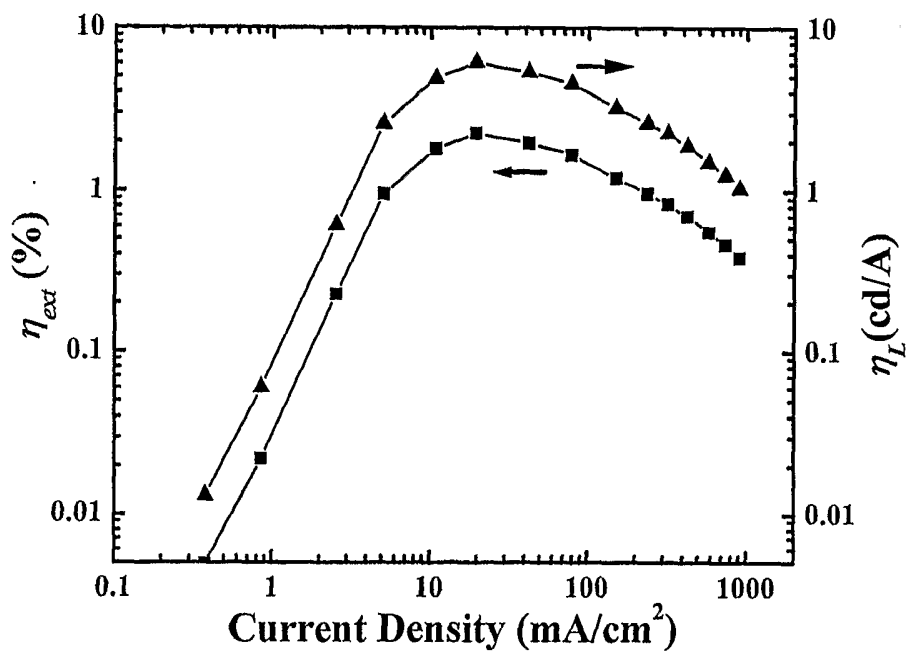


FIG 16. The external quantum efficiency and luminous efficiency of OLED C with 3.0 wt.% dopant 19

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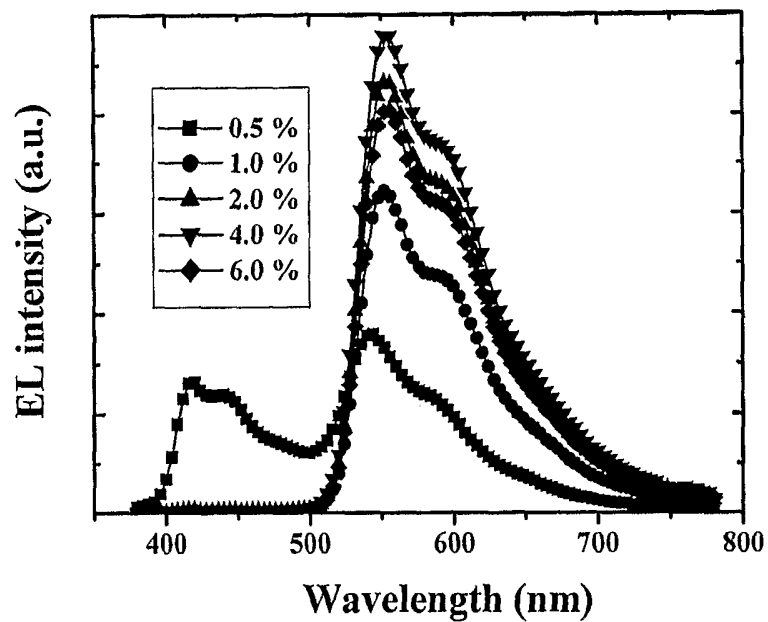


FIG 17. EL spectra of OLED D with 4 wt.% dopant 51 at different dopant concentrations at 8 V

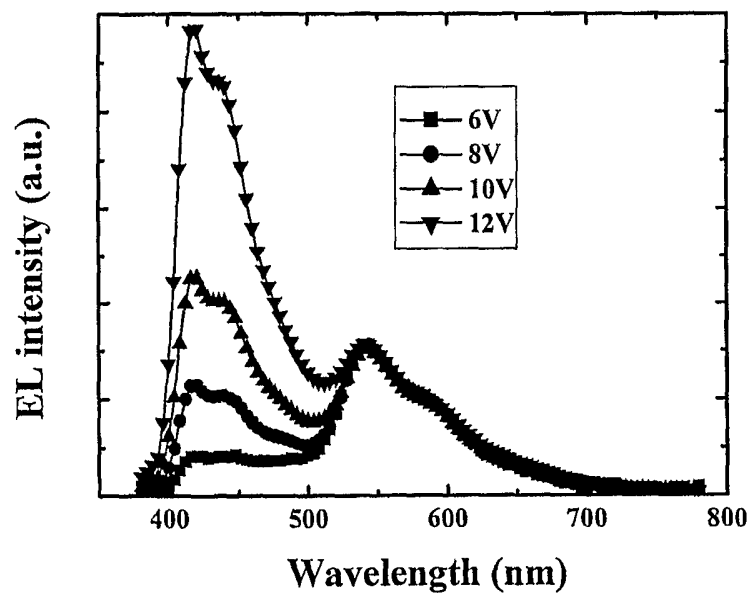


FIG 18. EL spectra of OLED D with 4 wt.% dopant 51 at different applied voltages

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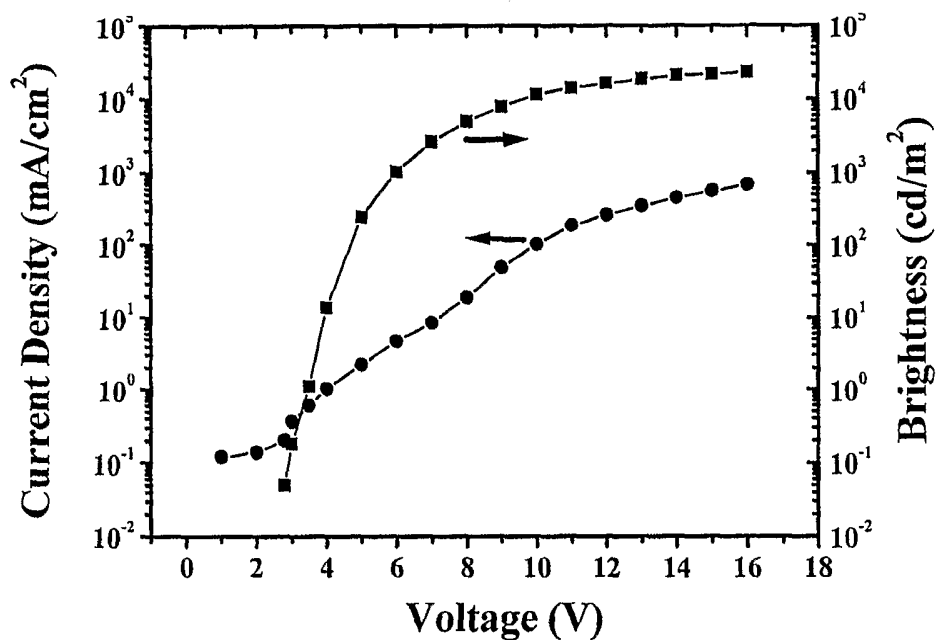


FIG 19. V-I-B curve of OLED D with 4 wt.% dopant 51

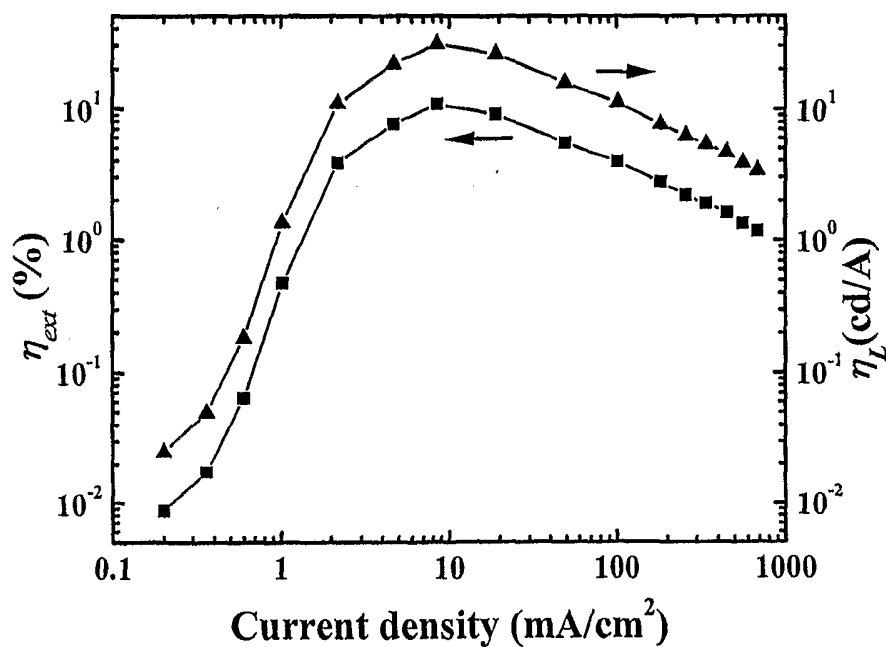


FIG 20. The external quantum efficiency and luminous efficiency of OLED D with 4.0 wt.% dopant 51

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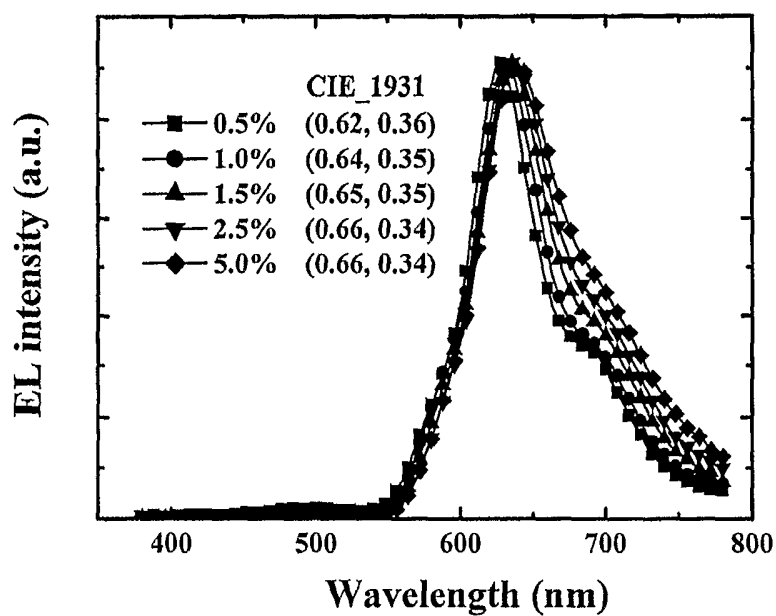


FIG. 21 EL spectra of OLED E with 1.5 wt.% dopant 99 at different dopant concentrations at 8 V

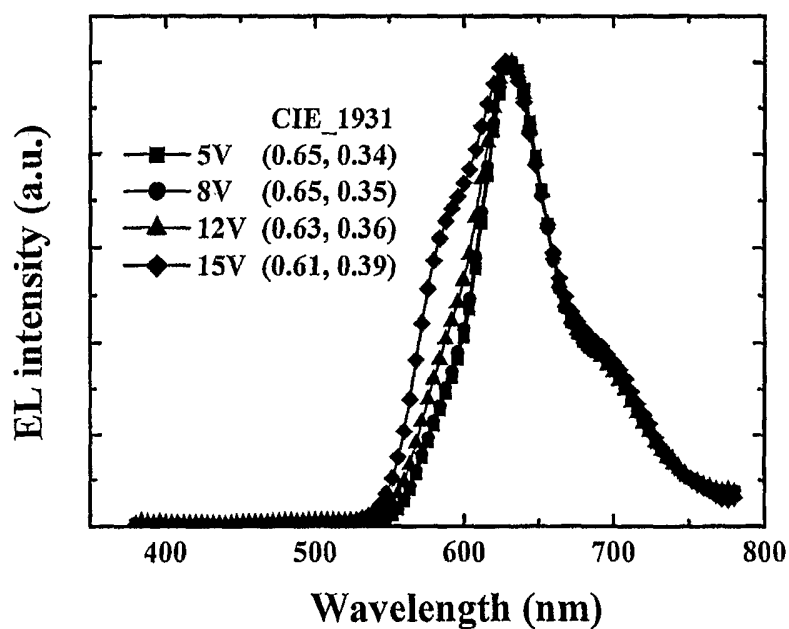


FIG. 22 EL spectra of OLED E with 1.5 wt.% dopant 99 at different applied voltages

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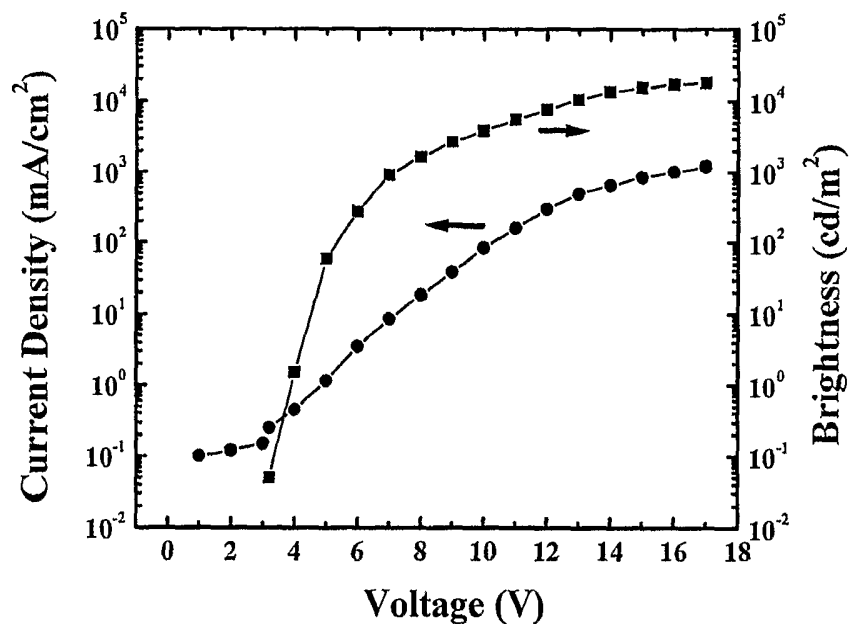


FIG. 23 V-I-B curve of OLED E with 1.5 wt.% dopant 99

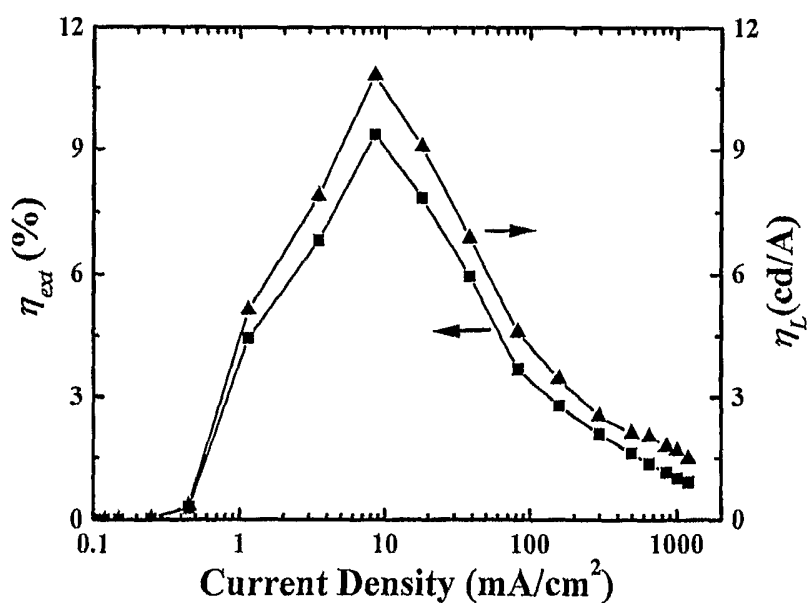


FIG. 24 External quantum efficiency and luminous efficiency of OLED E with 1.5 wt.% dopant 99

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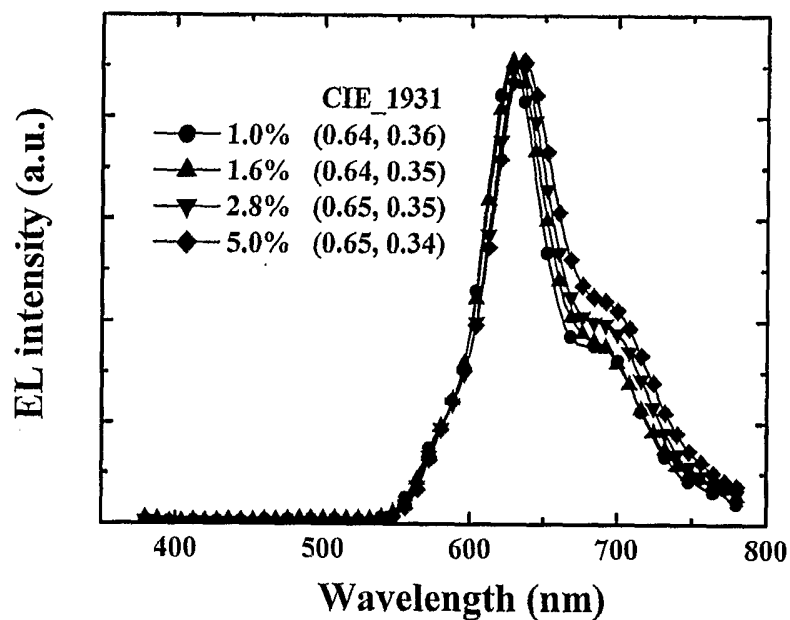


FIG. 25 EL spectra of OLED F with 1.6 wt.% dopant 104 at different dopant concentrations at 8 V

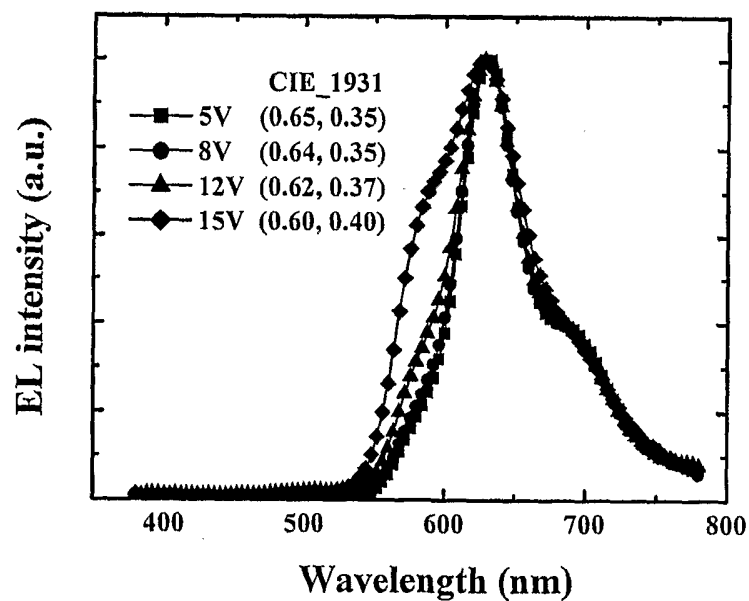


FIG. 26 EL spectra of OLED F with 1.6 wt.% dopant 104 at different applied voltages

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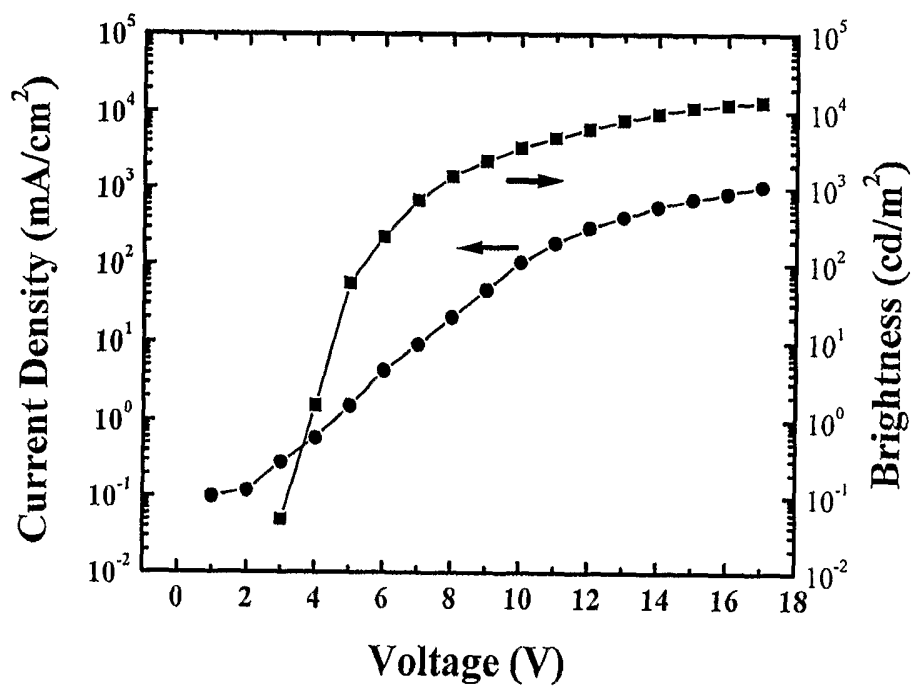


FIG. 27 V-I-B curve of OLED F with 1.6 wt.% dopant 104

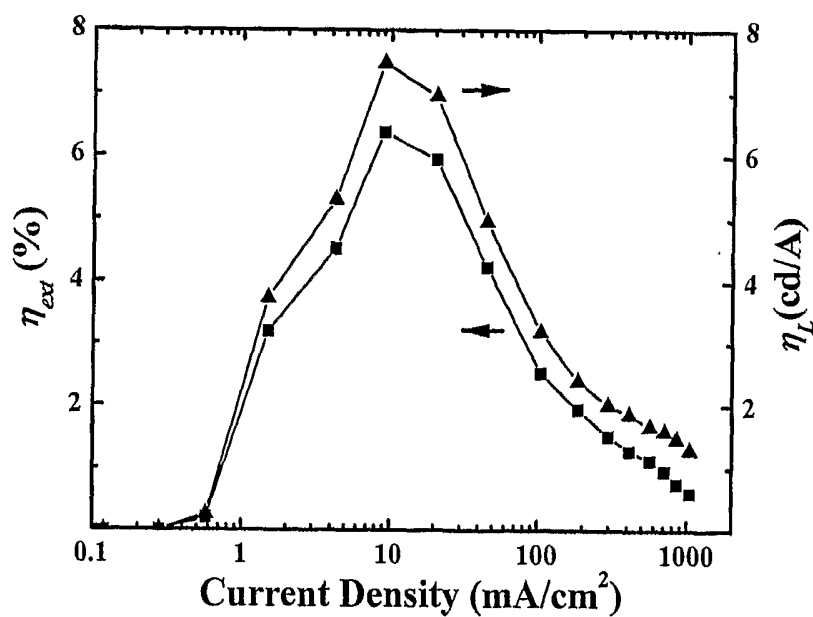


FIG. 28 External quantum efficiency and luminous efficiency of OLED F with 1.6 wt.% dopant 104

INTERNATIONAL SEARCH REPORT

International application No.
PCT/CN2005/000522

A. CLASSIFICATION OF SUBJECT MATTER		
IPC7: H05B33/14, C09K11/06		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED		
Minimum documentation searched (classification system followed by classification symbols)		
IPC7: H05B33, C09K11		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
CPRS		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)		
CPRS, WPI, EPODOC, PAJ: organic light-emitting device(s), OLED, bidentate, tetradentate, dopant, doped, doping, complex		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US6,653,654B1(The university of Hong Kong), 25, Nov. 2003(25.11.2003), See claims 1 and 4, examples 5, 7, 8)	1,2,5,11-15,19-21
X	US6,410,166B1(Futaba Corporation), 25, Jun. 2002(25.06.2002), see column 11 line 19-27, Examples 11-17, claim 2, column 1 line 30-31.	1,2,5,11,12,15,19,20
X	US5,552,547A(Song Q. Shi), 3, Sep. 1996(03.09.1996), see column 2, line 56-60, example 1, Claims 1-2.	1,2,5,15
A	JP2000-229966A(FuJi Photo Film Co., Ltd), 22, Aug. 2000(22.08.2000), see whole document.	1-21
☐ Further documents are listed in the continuation of Box C. ☒ See patent family annex.		
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Date of the actual completion of the international search 9, Aug. 2005(09.08.2005)		Date of mailing of the international search report 25 · AUG 2005 (25 · 08 · 2005)
Name and mailing address of the ISA/CN The State Intellectual Property Office, the P.R.China 6 Xitucheng Rd., Jimen Bridge, Haidian District, Beijing, China 100088 Facsimile No. 86-10-62019451		Authorized officer WU Hongxiu Telephone No. 86-10-62085652



INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.
PCT/CN2005/000522

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		AU2003218866A1	17, Nov. 2003(17.11.2003)
		EP1499624A1	26, Jan. 2005(26.01.2005)
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		JP3651135B2	25, May 2005(25.05.2005)
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