

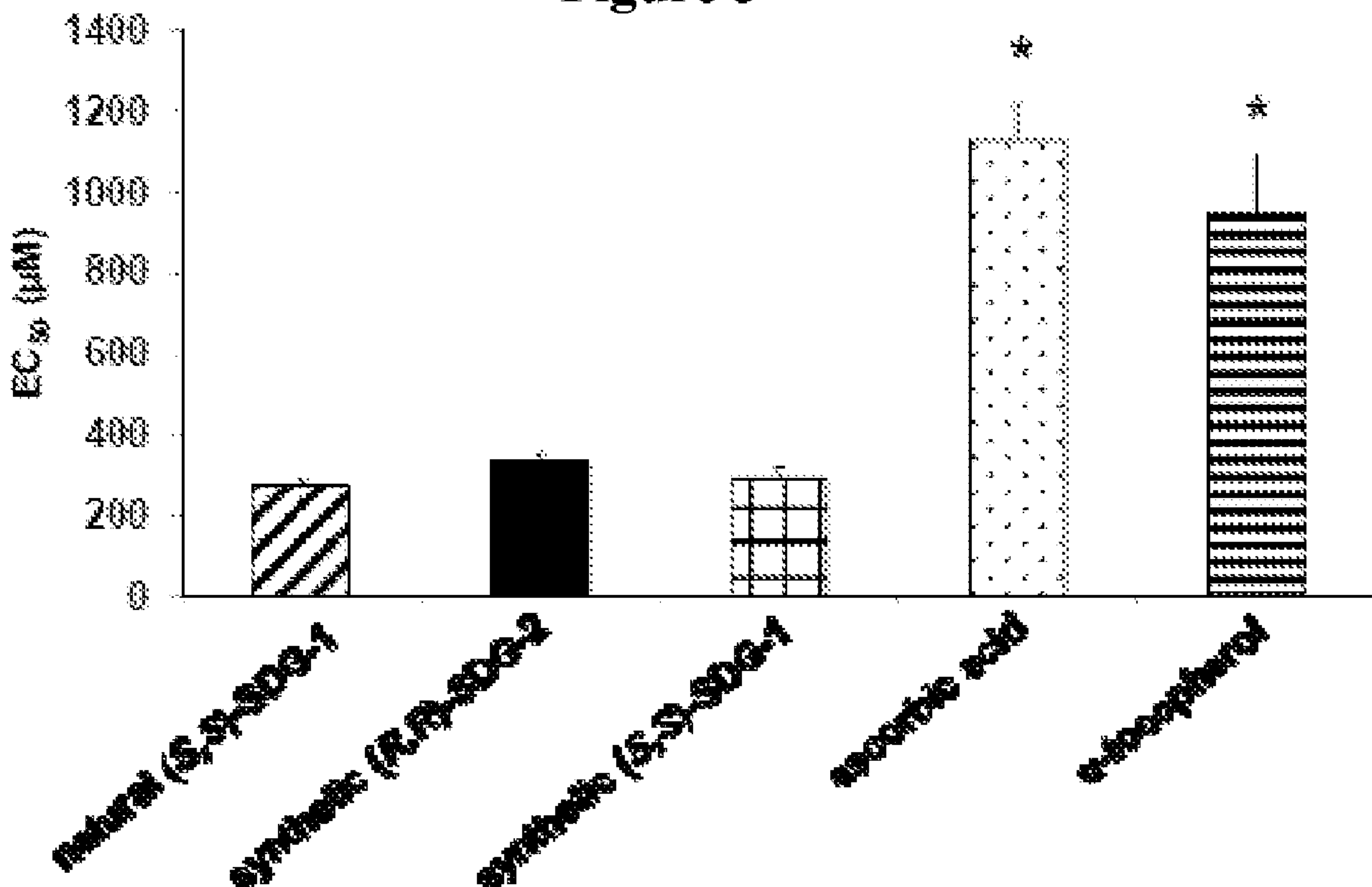


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(54) Titre : PREPARATION DE (S,S)-SECOISOLARICIRESinOL DIGLUCOSIDE ET DE (R,R)-SECOISOLARICIRESinOL DIGLUCOSIDE
(54) Title: PREPARATION OF (S,S)-SECOISOLARICIRESinOL DIGLUCOSIDE AND (R,R)-SECOISOLARICIRESinOL DIGLUCOSIDE

Figure 3



(57) Abrégé/Abstract:

The invention relates to processes for preparing (S,S)-secoisolariciresinol diglucoside and (R,R)-secoisolariciresinol diglucoside and compositions comprising the same.



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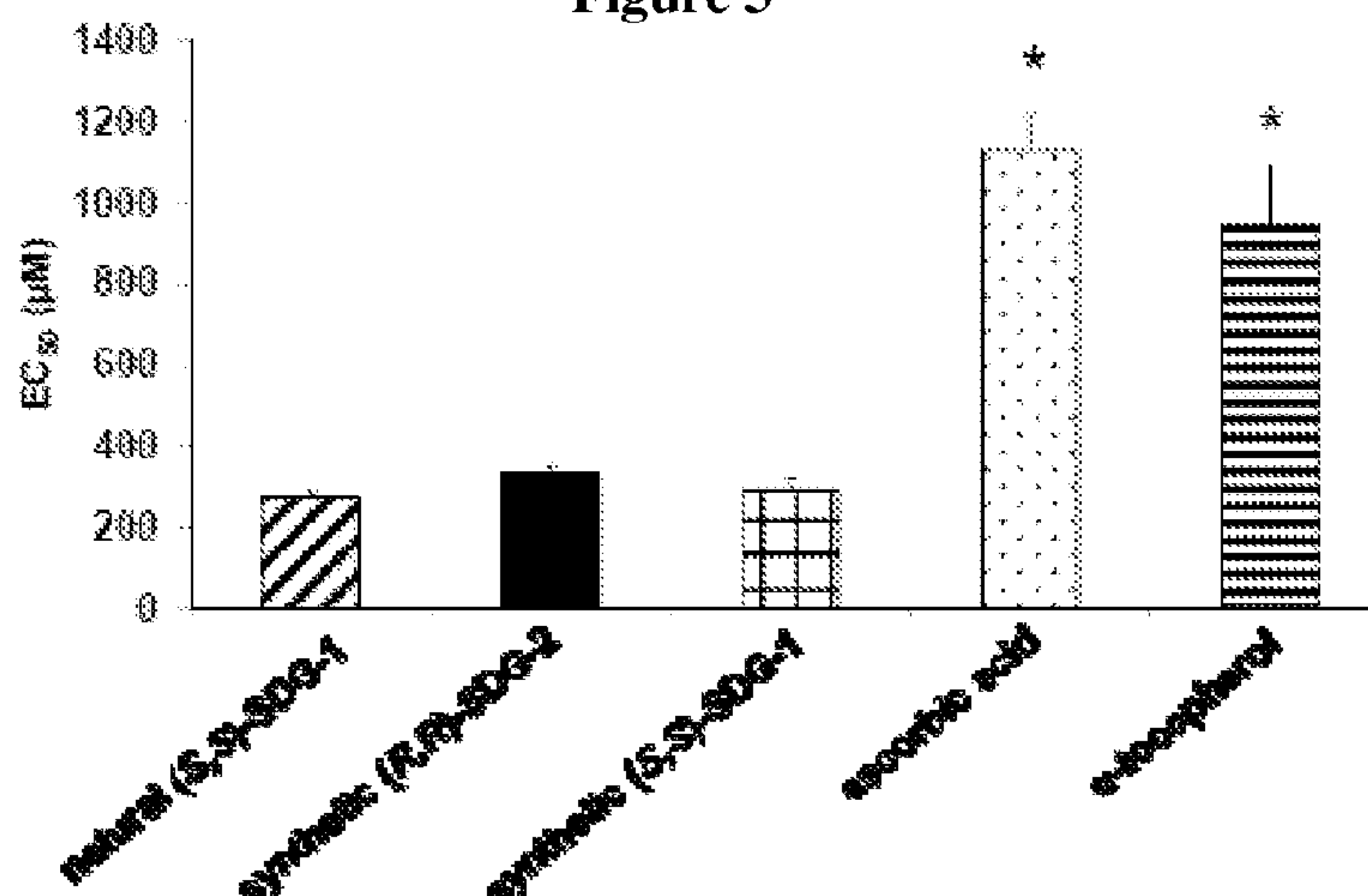
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(54) Title: PREPARATION OF (S,S)-SECOISOLARICIRE SINOL DIGLUCOSIDE AND (R,R)-SECOISOLARICIRE SINOL DIGLUCOSIDE

Figure 3



(57) Abstract: The invention relates to processes for preparing (S,S)-secoisolariciresinol diglucoside and (R,R)-secoisolariciresinol diglucoside and compositions comprising the same.

PREPARATION OF (S,S)-SECOISOLARICIRE SINOL DIGLUCOSIDE
AND (R,R)-SECOISOLARICIRE SINOL DIGLUCOSIDE

GOVERNMENT INTEREST

[0001] This invention was made, in part, with government support under Grant Numbers R01
5 CA133470 and RC1AI081251, both awarded by the National Institutes of Health. The
government has certain rights in the invention.

FIELD OF THE INVENTION

[0002] The invention relates to processes for preparing (S,S)-secoisolariciresinol diglucoside
and (R,R)-secoisolariciresinol diglucoside, as well as compositions comprising the same.

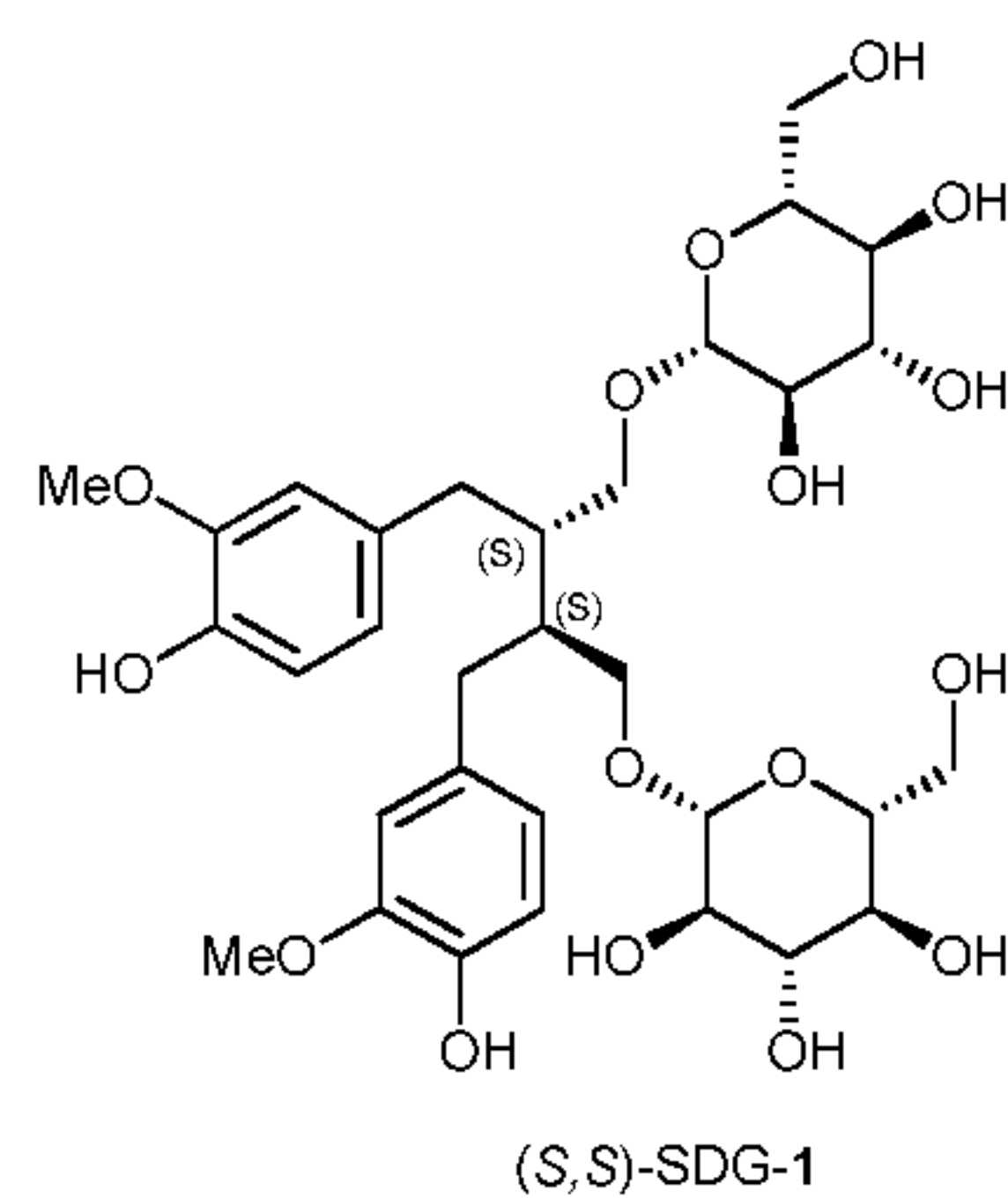
10 **BACKGROUND OF THE INVENTION**

[0003] Ionizing radiation produces a wide range of deleterious effects in living organisms.
Humans are exposed to radiation during diagnostic and therapeutic radiographic procedures,
when using electron devices, from background radiation of nuclear accidents, and during air
and space travel. Current global developments have additionally established terrorism as
15 dangerous means by which potentially large numbers of people can be exposed to lethal
amounts of radiation. It is, therefore, of high importance to identify radioprotective agents
that can be administered both before exposure to radiation, and as treatment after radioactive
exposure.

[0004] Natural products and their analogs are increasingly considered as promising leads for
20 the discovery and development of radioprotectors. Both (2R,3R)- and (2S,3S)- enantiomers
of secoisolariciresinol and meso-secoisolariciresinol have been detected in plants, but only the
(2R,3R)-[(-)secoisolariciresinol]-isomer occurs abundantly. Although this compound is
available by extraction from flaxseed, the necessary subsequent purification is a non-trivial
and tedious process. Accordingly, there exists a need to develop a short and enantioselective
25 synthesis to prepare enantiopure secoisolariciresinol from readily available commercial
materials.

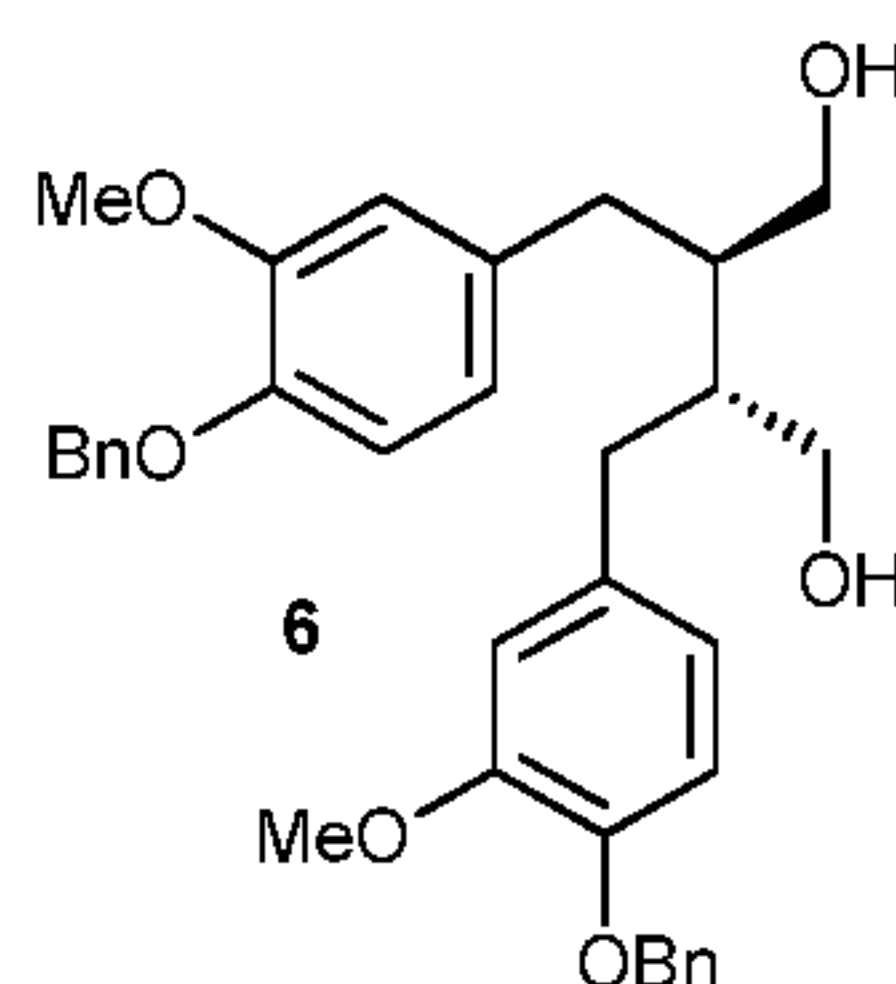
SUMMARY OF THE INVENTION

[0005] The present invention provides a process for preparing a compound of formula ((S,S)-
SDG-1) ((S,S)-secoisolariciresinol diglucosides-1)



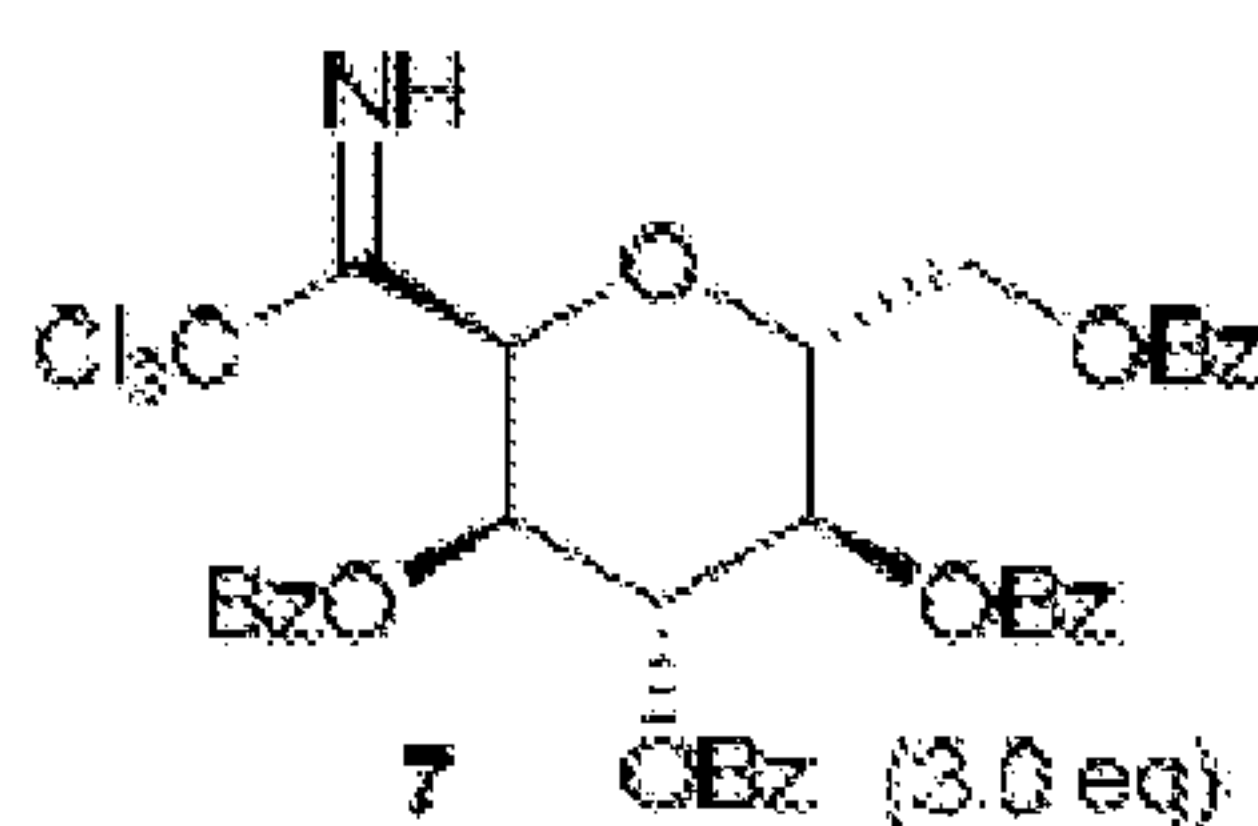
the process comprising:

- a) reacting a compound of formula (6)

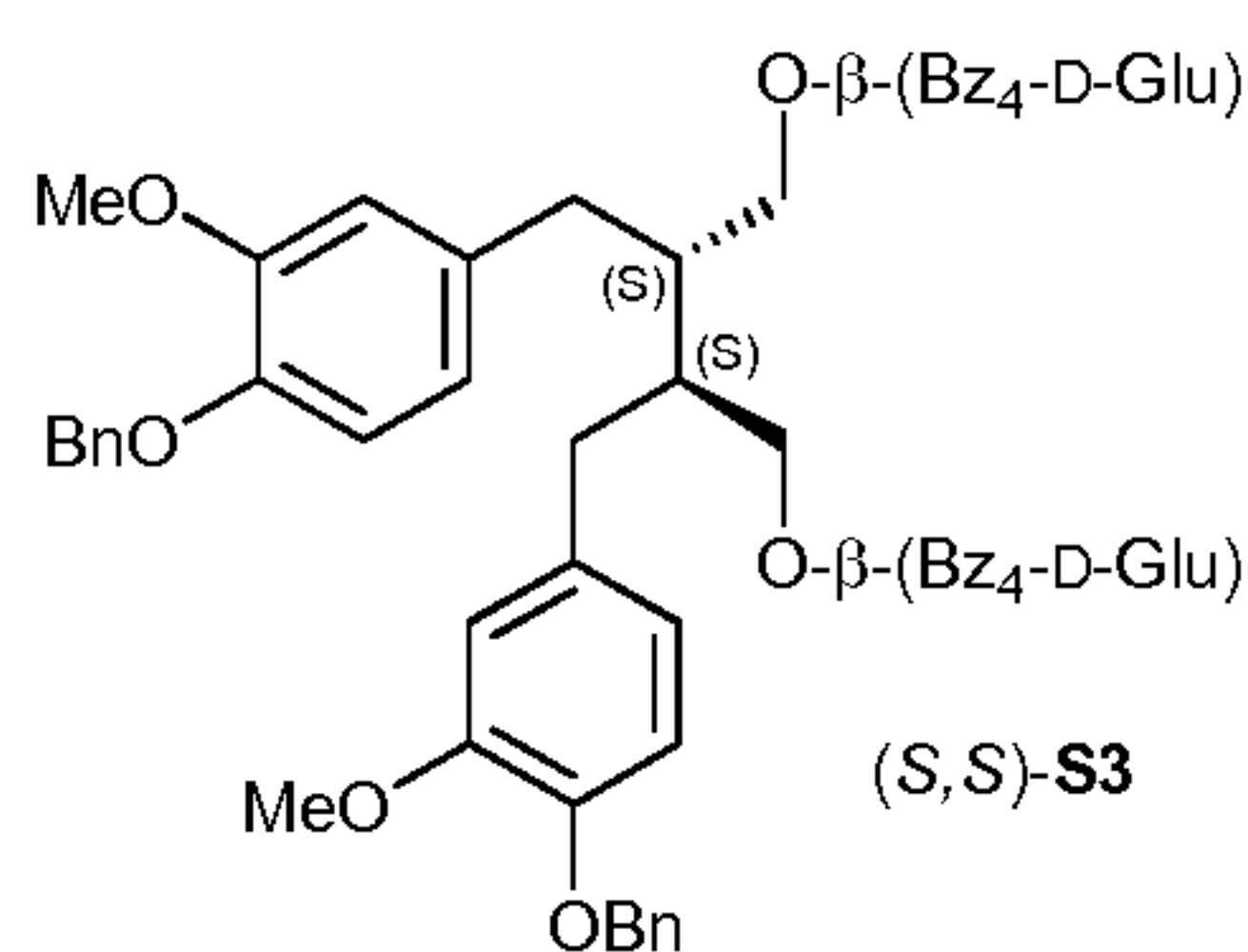


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with a compound of formula (7)

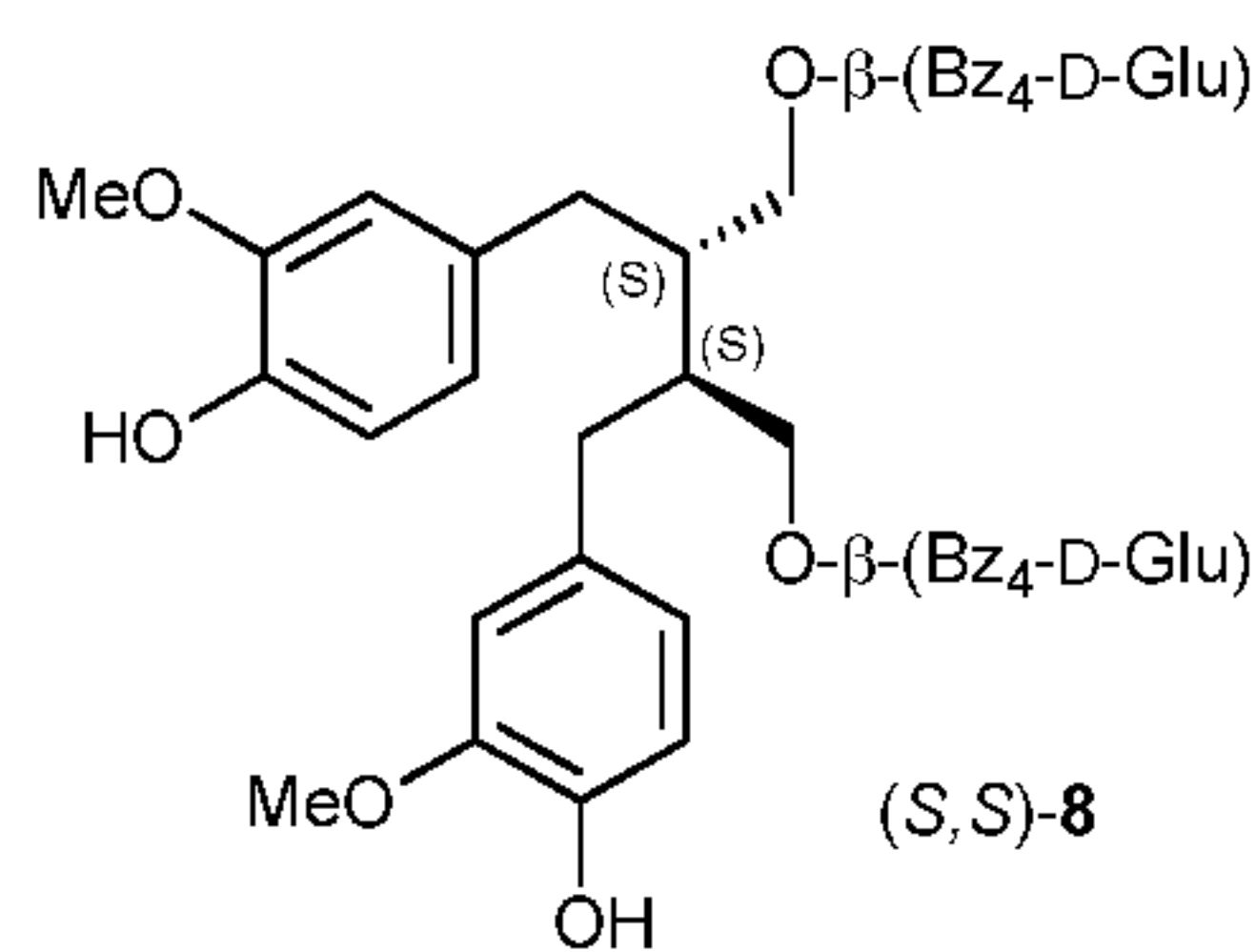


to prepare a compound of formula ((S,S)-S3)



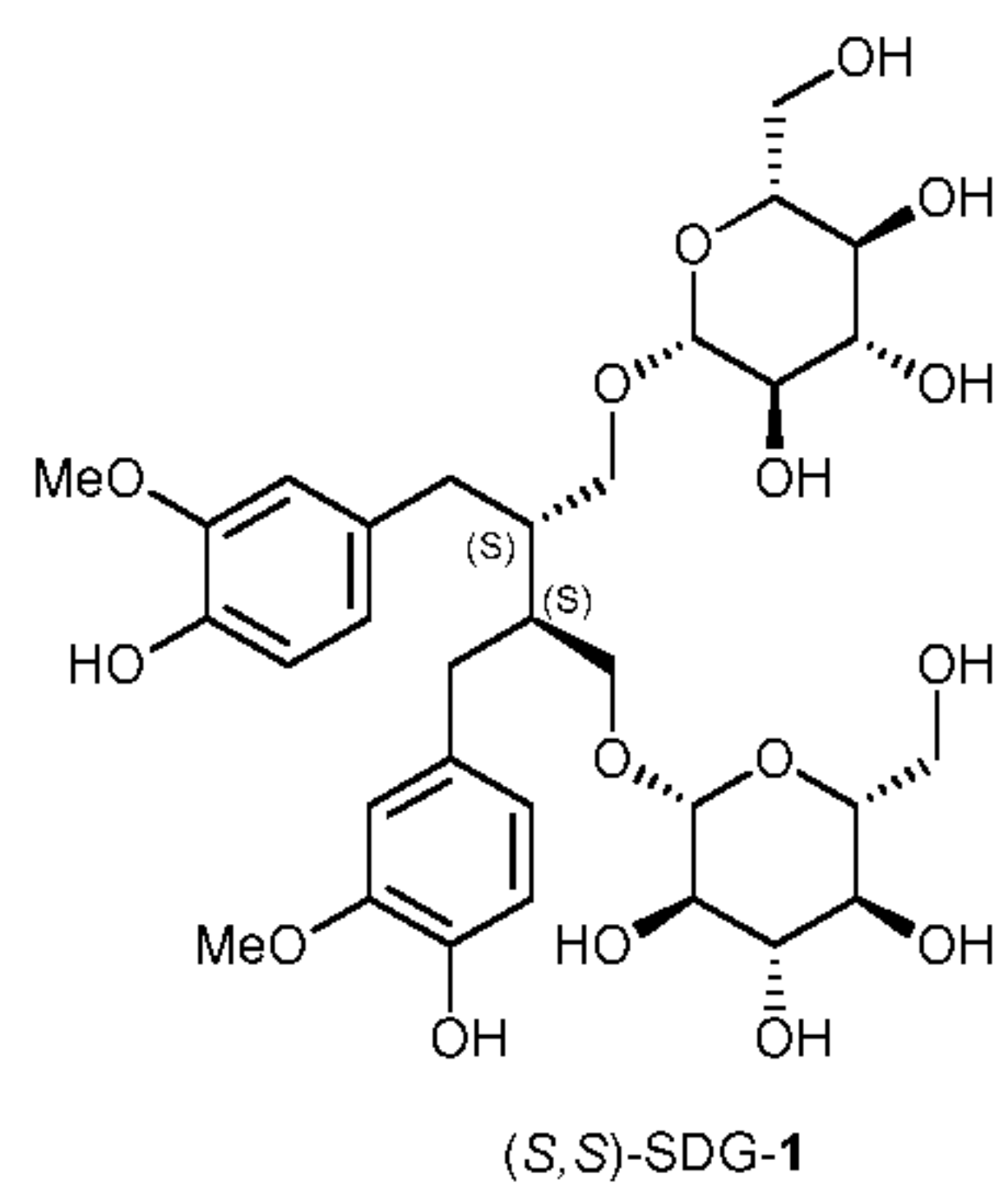
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(b) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare a compound of formula ((S,S)-8)



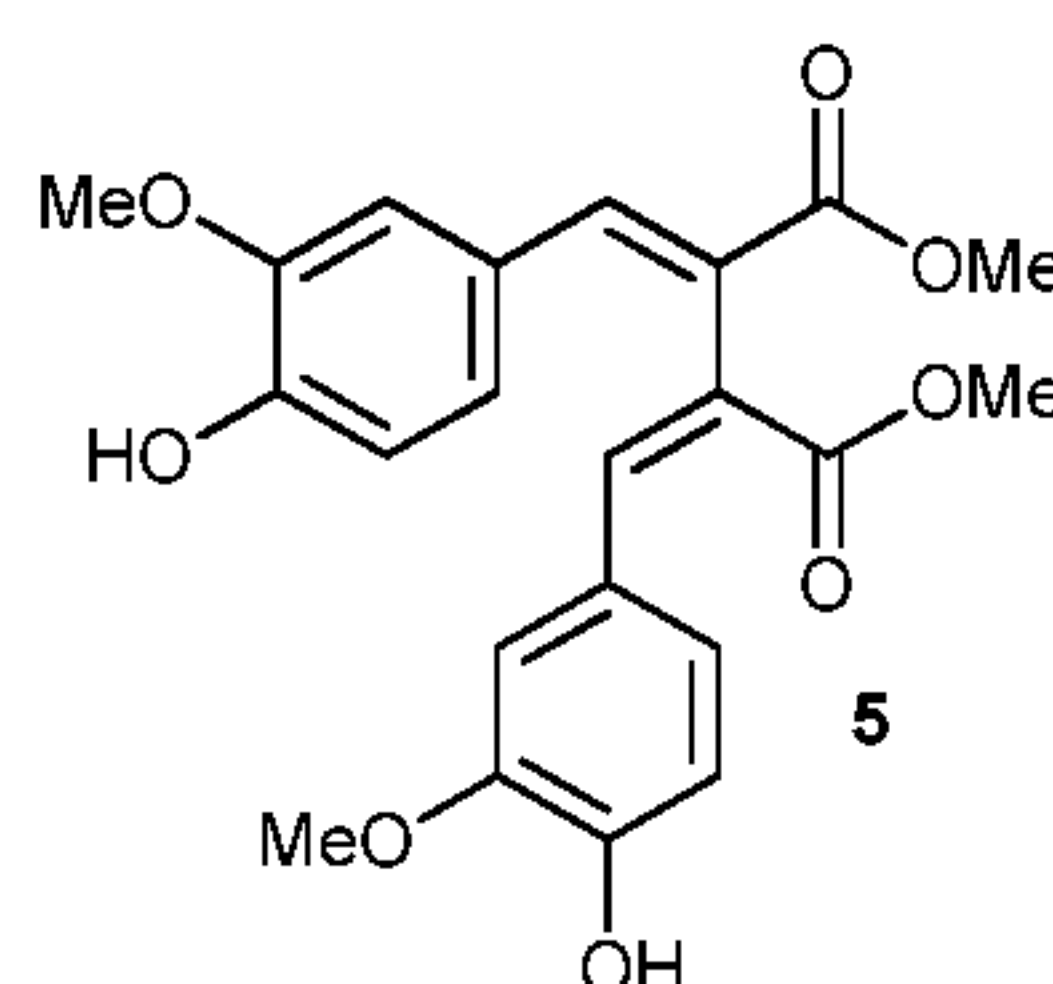
(c) deprotecting the compound of formula ((S,S)-8) to prepare the compound of formula ((S,S)-SDG-1).

[0006] The present invention provides a process for preparing a compound of formula ((*S,S*)-SDG-1)

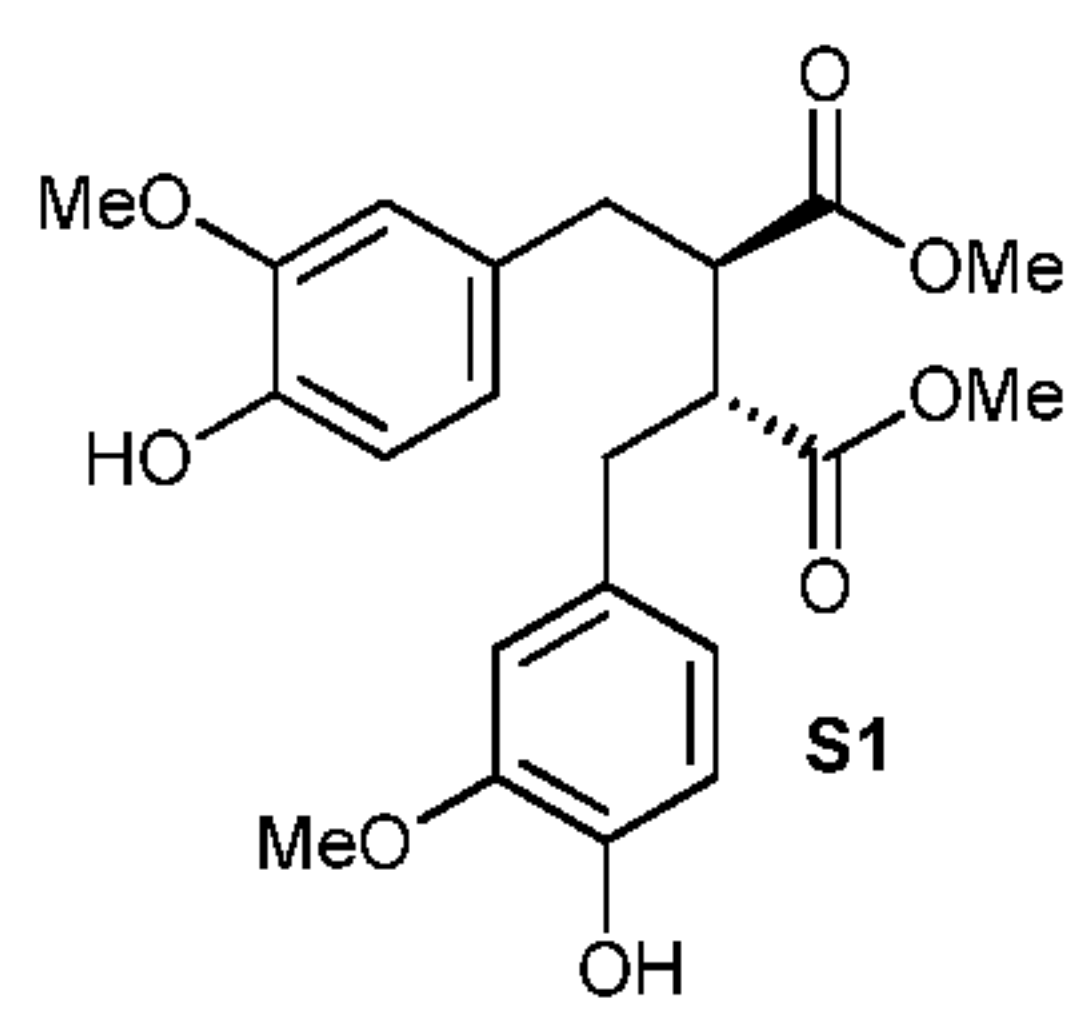


5 the process comprising:

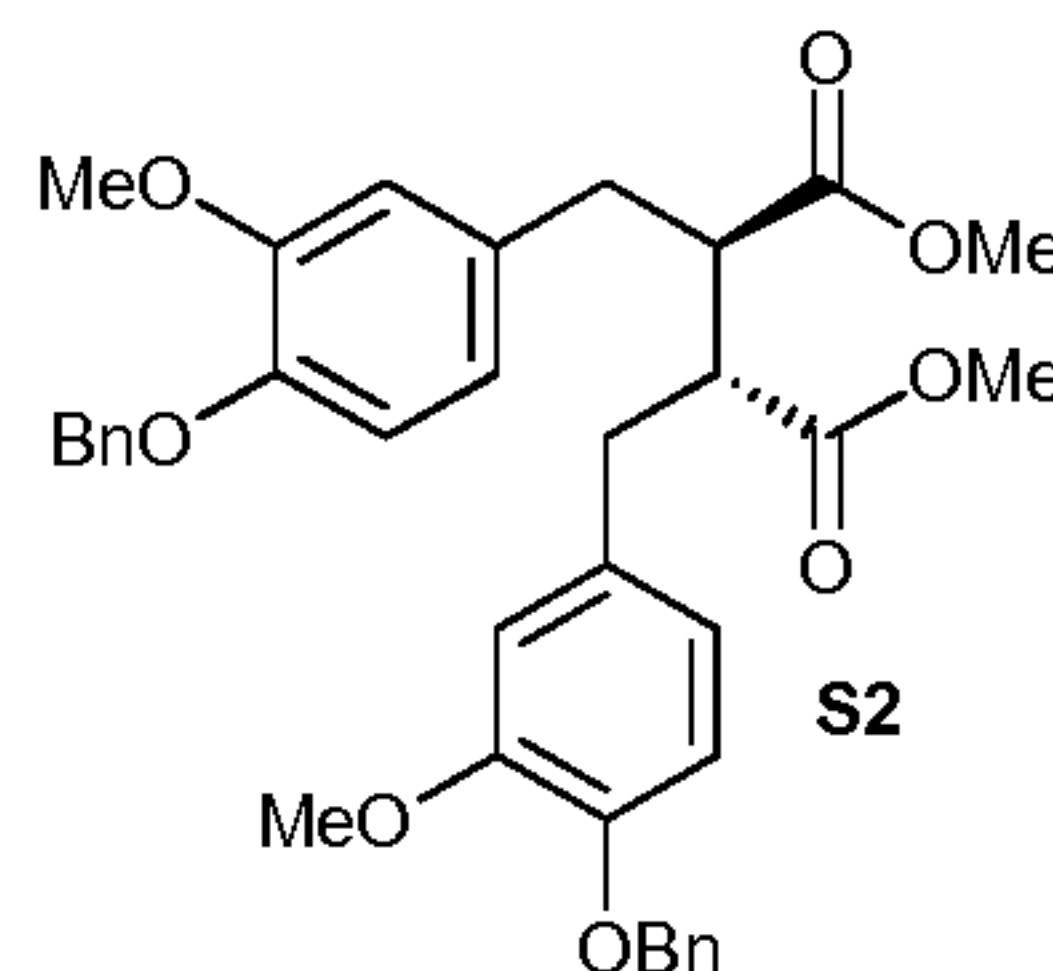
(a) reacting a compound of formula (5)



with a reducing agent to prepare a compound of formula (S1)



10 (b) reacting the compound of formula (S1) with a benzylating agent to prepare a compound of formula (S2)



(c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6);

15 (d) reacting the compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*S,S*)-S3);

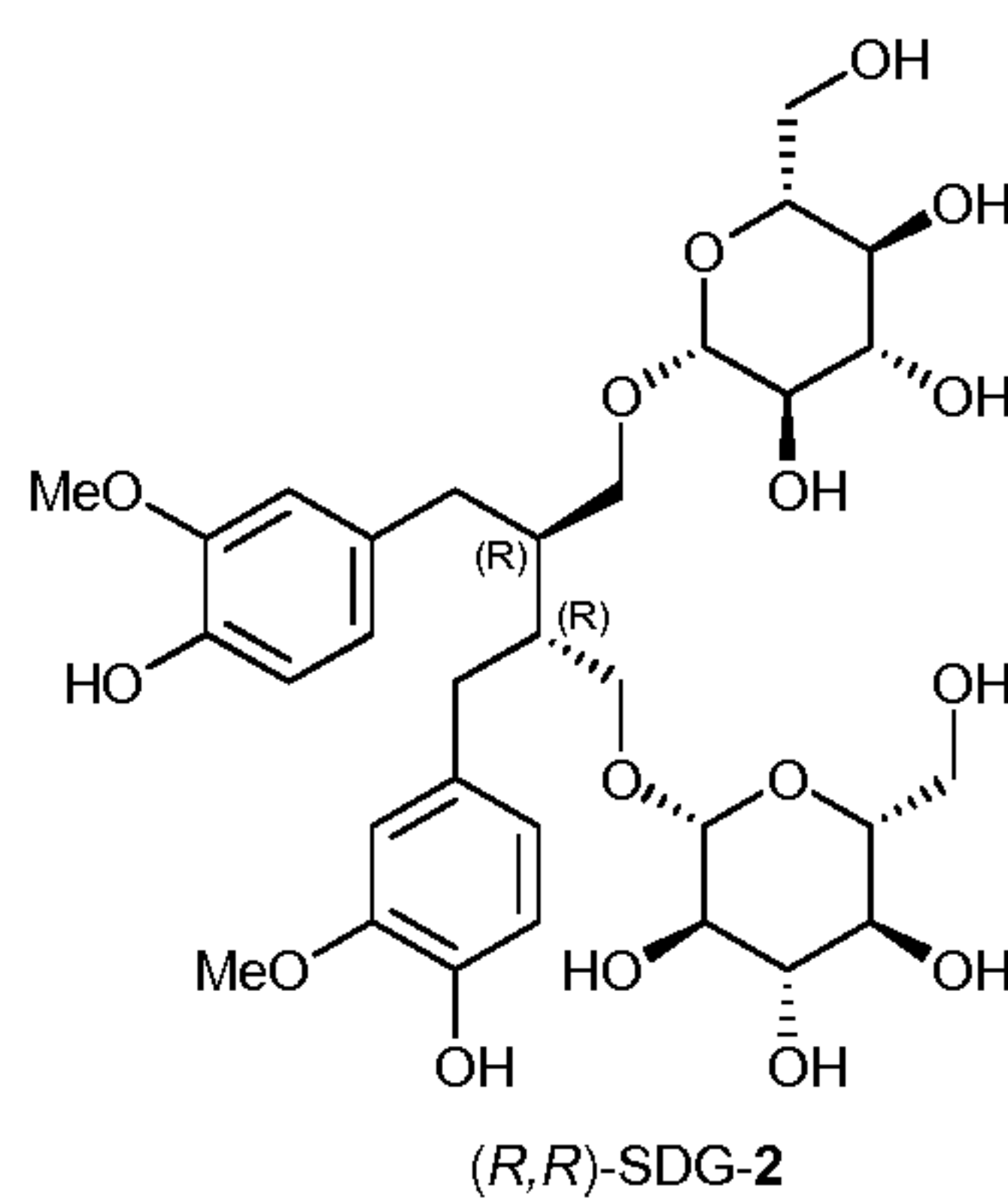
- (e) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare a compound of formula ((*S,S*)-8); and
- (f) deprotecting the compound of formula ((*S,S*)-8) to prepare the compound of formula ((*S,S*)-SDG-1).

5

[0007] The present invention further provides a process for preparing a compound of formula ((*S,S*)-SDG-1), the process comprising:

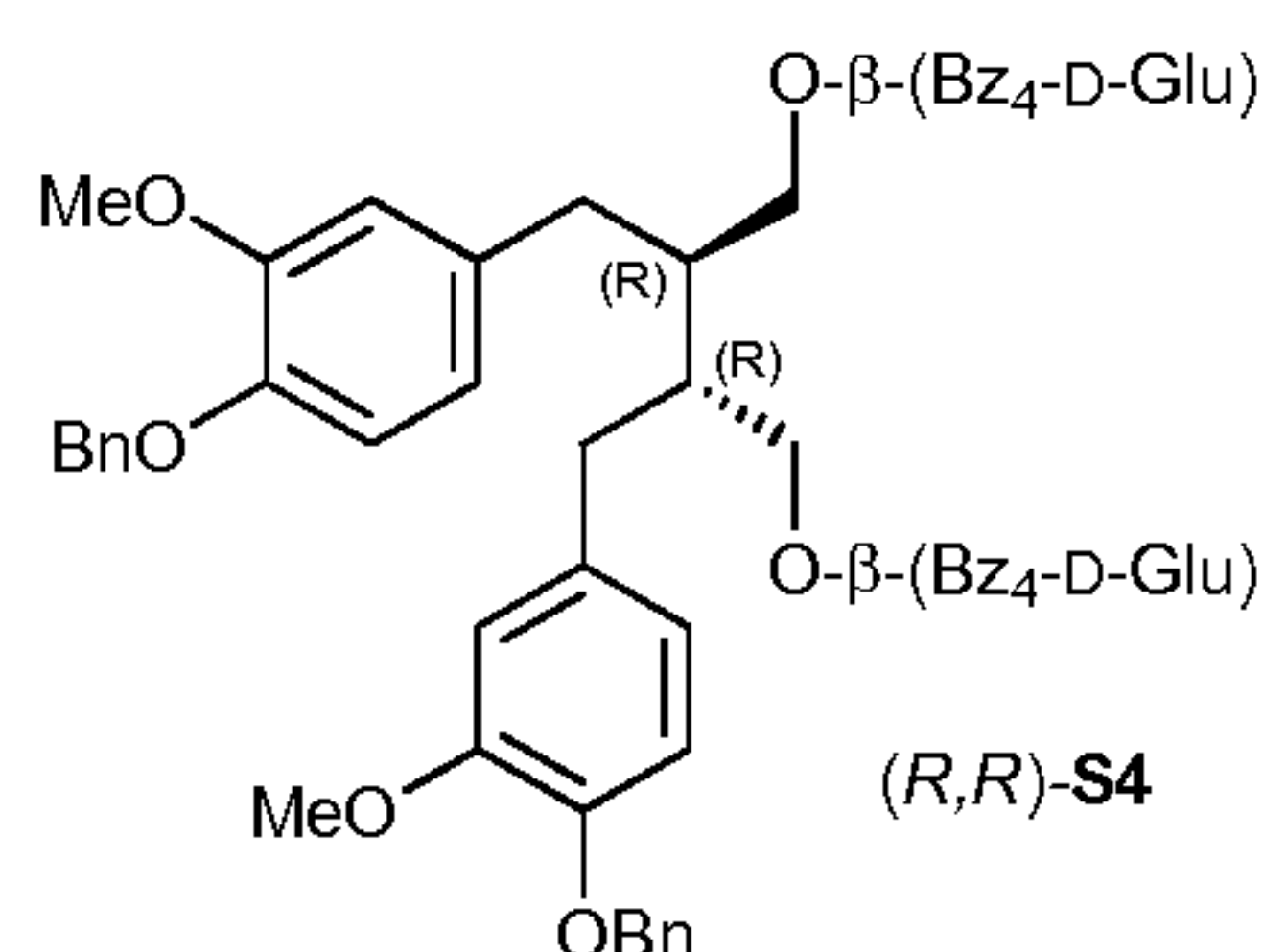
- (a) reacting a compound of formula (S2) with a reducing agent to prepare a compound of formula (6);
- 10 (b) reacting the compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*S,S*)-S3);
- (c) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare a compound of formula ((*S,S*)-8); and
- (d) deprotecting the compound of formula ((*S,S*)-8) to prepare the compound of formula ((*S,S*)-SDG-1).
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[0008] The present invention further provides a process for preparing a compound of formula ((*R,R*)-SDG-2) ((*R,R*)-secoisolariciresinol diglucosides-2)

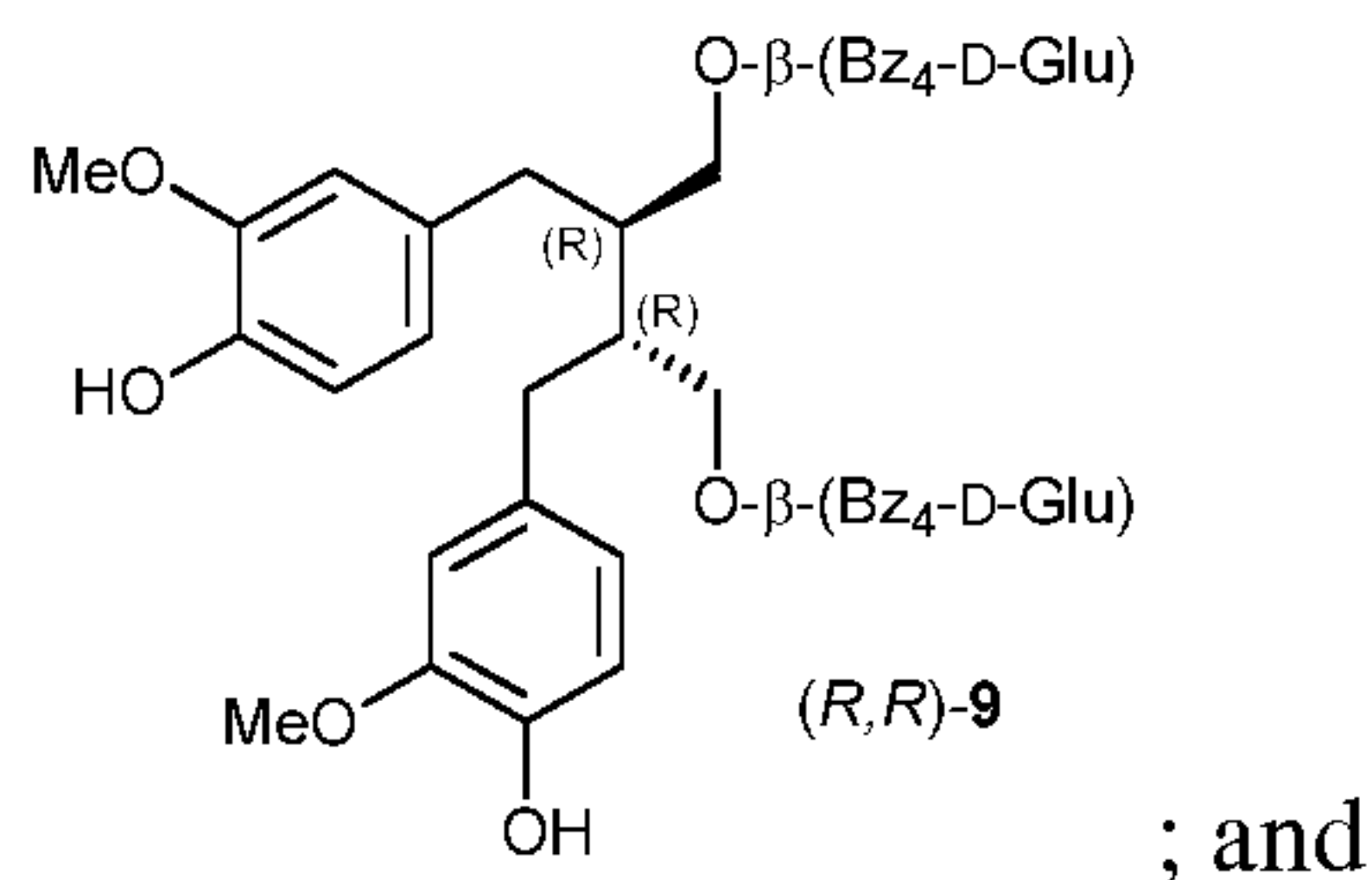


20 the process comprising:

- b) reacting a compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*R,R*)-S4)



- (a) cleaving benzyl ethers of the compound of formula ((*R,R*)-S4), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9)



- (c) deprotecting the compound of formula ((*R,R*)-9) to prepare the compound of formula ((*R,R*)-SDG-2).

[0009] The present invention further provides a process for preparing a compound of formula ((*R,R*)-SDG-2), the process comprising:

- (a) reacting a compound of formula (5) with a reducing agent to prepare a compound of formula (S1);
- (b) reacting the compound of formula (S1), with a benzylating agent to prepare a compound of formula (S2);
- (c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6);
- (d) reacting the compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*R,R*)-S4);
- (e) cleaving benzyl ethers of the compound of formula ((*R,R*)-S3), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9); and
- (f) deprotecting the compound of formula ((*R,R*)-9) to prepare the compound of formula ((*R,R*)-SDG-2).

[0010] The present invention further provides a process for preparing a compound of formula ((*R,R*)-SDG-2), the process comprising:

- (a) reacting a compound of formula (S2), with a reducing agent to prepare a compound of formula (6),
- (b) reacting the compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*R,R*)-S4);
- (c) cleaving benzyl ethers of the compound of formula ((*R,R*)-S4), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9); and

(d) deprotecting the compound of formula ((*R,R*)-9) to prepare the compound of formula ((*R,R*)-SDG-2).

[0011] The present invention further provides a process for preparing a compound of formula (6), the process comprising reacting a compound of formula (S2) with a reducing agent to prepare the compound of formula (6).

[0012] The present invention further provides a process for preparing a compound of formula ((*S,S*)-S3), the process comprising reacting a compound of formula (6) with a compound of formula (7) to prepare the compound of formula ((*S,S*)-S3).

[0013] The present invention further provides a process for preparing a compound of formula ((*S,S*)-8), the process comprising:

(a) reacting a compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*S,S*)-S3);

(b) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare the compound of formula ((*S,S*)-8).

[0014] The present invention provides a process for preparing a compound of formula ((*R,R*)-S4), the process comprising reacting a compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*R,R*)-S4)

[0015] The present invention provides a process for preparing a compound of formula ((*R,R*)-9), the process comprising:

(a) reacting a compound of formula (6) with a compound of formula (7) to prepare a compound of formula ((*R,R*)-S4);

(b) cleaving benzyl ethers of the compound of formula ((*R,R*)-S4), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9).

[0016] The present invention further provides a composition, such as a pharmaceutical composition, comprising a compound of formula ((*S,S*)-SDG-1).

[0017] The present invention further provides a composition, such as a pharmaceutical composition, comprising a compound of formula ((*R,R*)-SDG-2).

[0018] Other features and advantages of the present invention will become apparent from the following detailed description examples and figures. It should be understood, however, that the detailed description and the specific examples while indicating preferred embodiments of the invention are given by way of illustration only, since various changes and modifications within the spirit and scope of the invention will become apparent to those skilled in the art from this detailed description.

BRIEF DESCRIPTION OF THE DRAWINGS

[0019] **Figure 1** depicts an ORTEP representation of dihydroxy compound **6**.

[0020] **Figure 2** illustrates the reducing power of synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2. The increase in absorbance at 700nm indicates increase in reducing power. The results are presented as mean $\pm$ standard deviation ($n = 3$). $*p < 0.05$ significantly lower than natural (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, and synthetic (*S,S*)-SDG-1; $**p < 0.05$ significantly higher than synthetic (*R,R*)-SDG-2, synthetic (*S,S*)-SDG-1, ascorbic acid and α -tocopherol.

[0021] **Figure 3** illustrates the reducing power of synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2. The rate of reaction is linear in the concentration range of 1-100 μ M. Equation of the linear regression was used to determine EC_{50} . The results are presented as mean $\pm$ standard deviation ($n = 3$). Natural (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, and synthetic (*S,S*)-SDG-1 were not significantly different from each other. $*p < 0.05$ significantly higher than natural (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, and synthetic (*S,S*)-SDG-1.

[0022] **Figure 4** depicts DPPH free radical scavenging activity of natural (*S,S*)-SDG-1, synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2, ascorbic acid, and α -tocopherol. The radical scavenging activity was measured as a decrease in the absorbance of DPPH at 517 nm. The results are presented as mean $\pm$ standard deviation ($n = 3$). $*p < 0.05$ significantly lower than all other compounds, $**p < 0.05$ significantly lower than natural (*S,S*)-SDG-1.

[0023] **Figure 5** depicts DPPH free radical scavenging activity of natural (*S,S*)-SDG-1, synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2, ascorbic acid, and α -tocopherol. Equation of the linear regression was used to determine EC_{50} . The rate of reaction is linear in the concentration range of 1-100 μ M. The results are presented as mean $\pm$ standard deviation ($n = 3$). Natural (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, synthetic (*S,S*)-SDG-1, and α -tocopherol

were not significantly different. * $p < 0.05$ significantly higher than natural (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, synthetic (*S,S*)-SDG-1, and α -tocopherol.

[0024] **Figure 6** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound **4**.

5 [0025] **Figure 7** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound **5**.

[0026] **Figure 8** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound **S1**.

10 [0027] **Figure 9** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound **S2**.

[0028] **Figure 10** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound **6**.

[0029] **Figure 11** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of 1:1 (*S,S*)-**S3**/*R,R*)-**S4**.

15 [0030] **Figure 12** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound (*S,S*)-**8**.

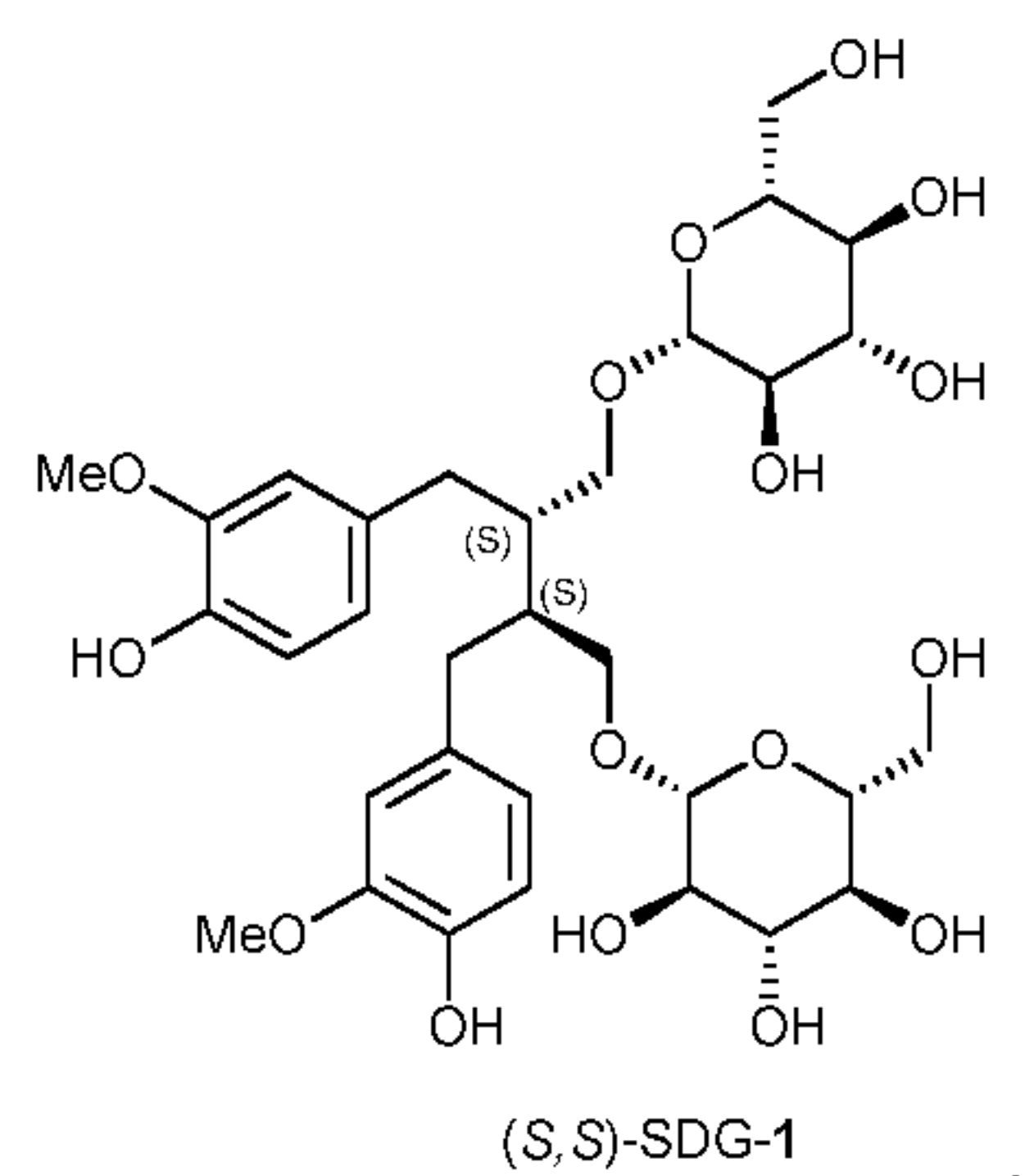
[0031] **Figure 13** shows ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) spectra of compound (*R,R*)-**9**.

20 [0032] **Figure 14** shows ^1H NMR (600 MHz, CD_3OD) and ^{13}C NMR (150 MHz, CD_3OD) spectra of compound (*S,S*)-**SDG-1**.

[0033] **Figure 15** shows ^1H NMR (600 MHz, CD_3OD) and ^{13}C NMR (150 MHz, CD_3OD) spectra of compound (*R,R*)-**SDG-2**.

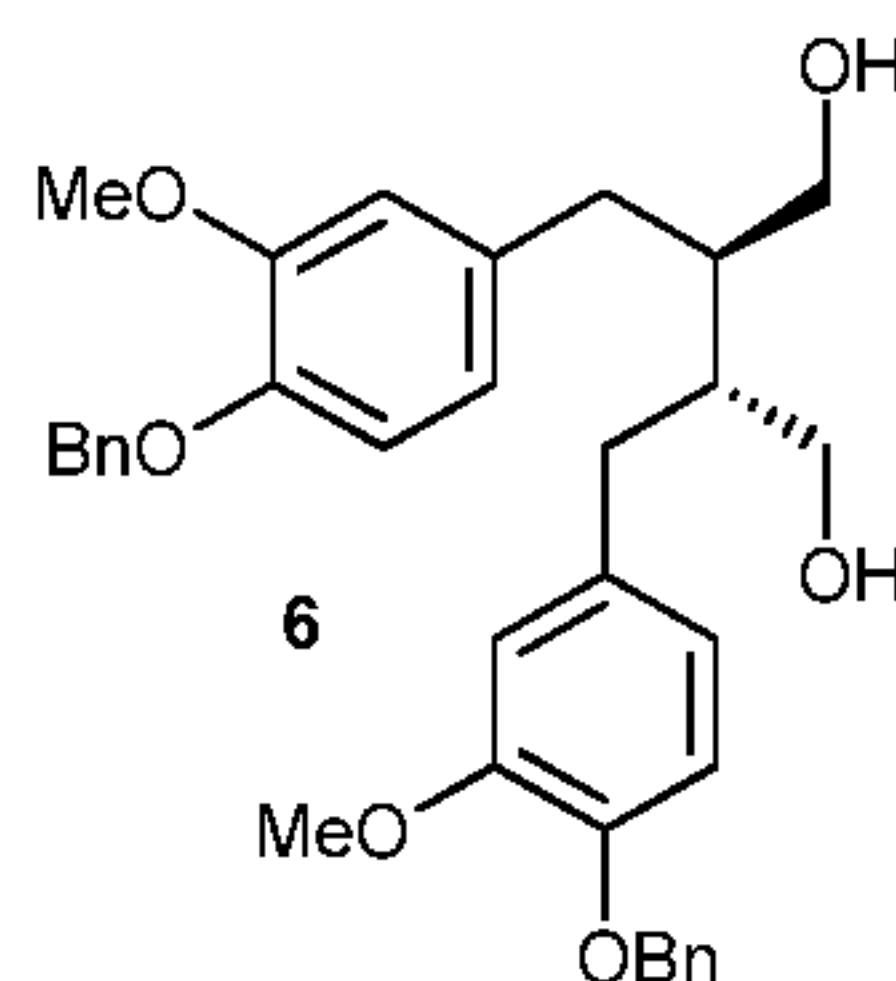
DETAILED DESCRIPTION OF THE INVENTION

25 [0034] The present invention provides a process for preparing a compound of formula ((*S,S*)-SDG-1)



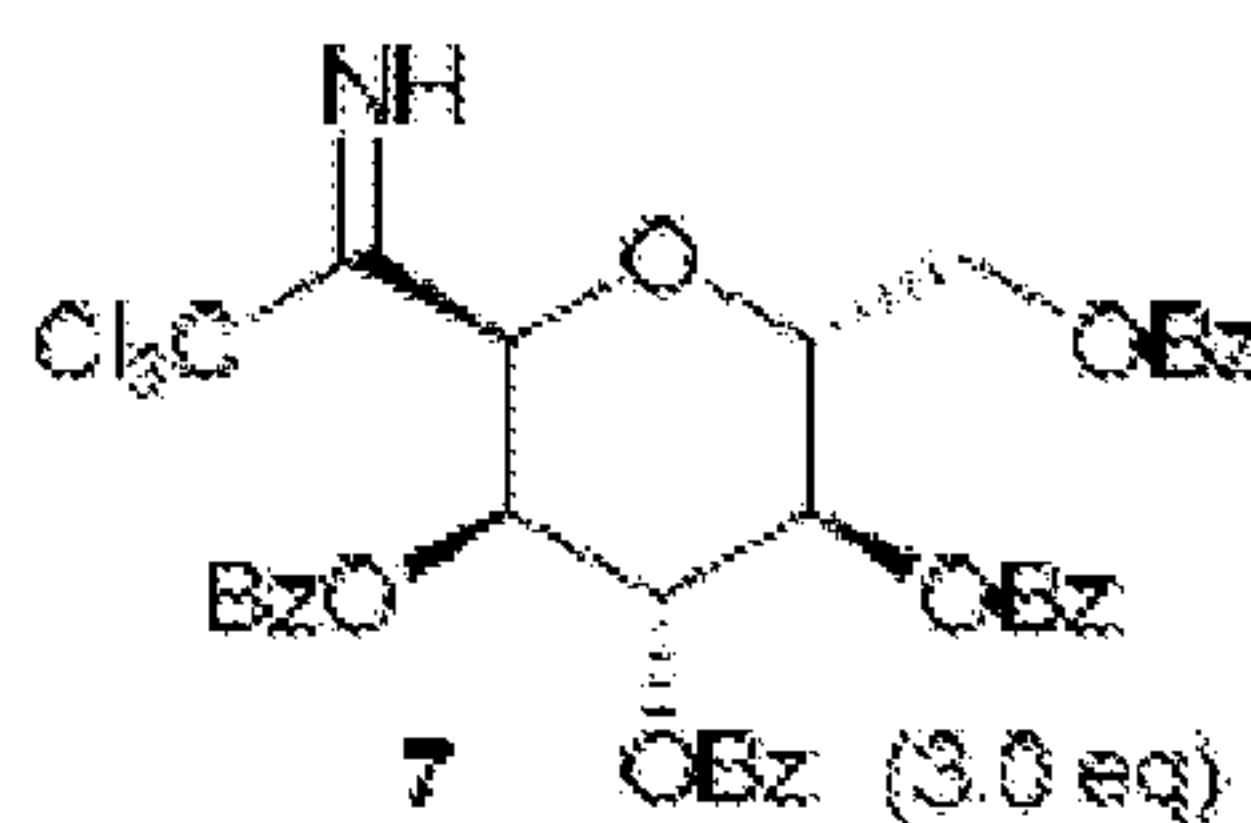
the process comprising:

c) reacting a compound of formula (6)

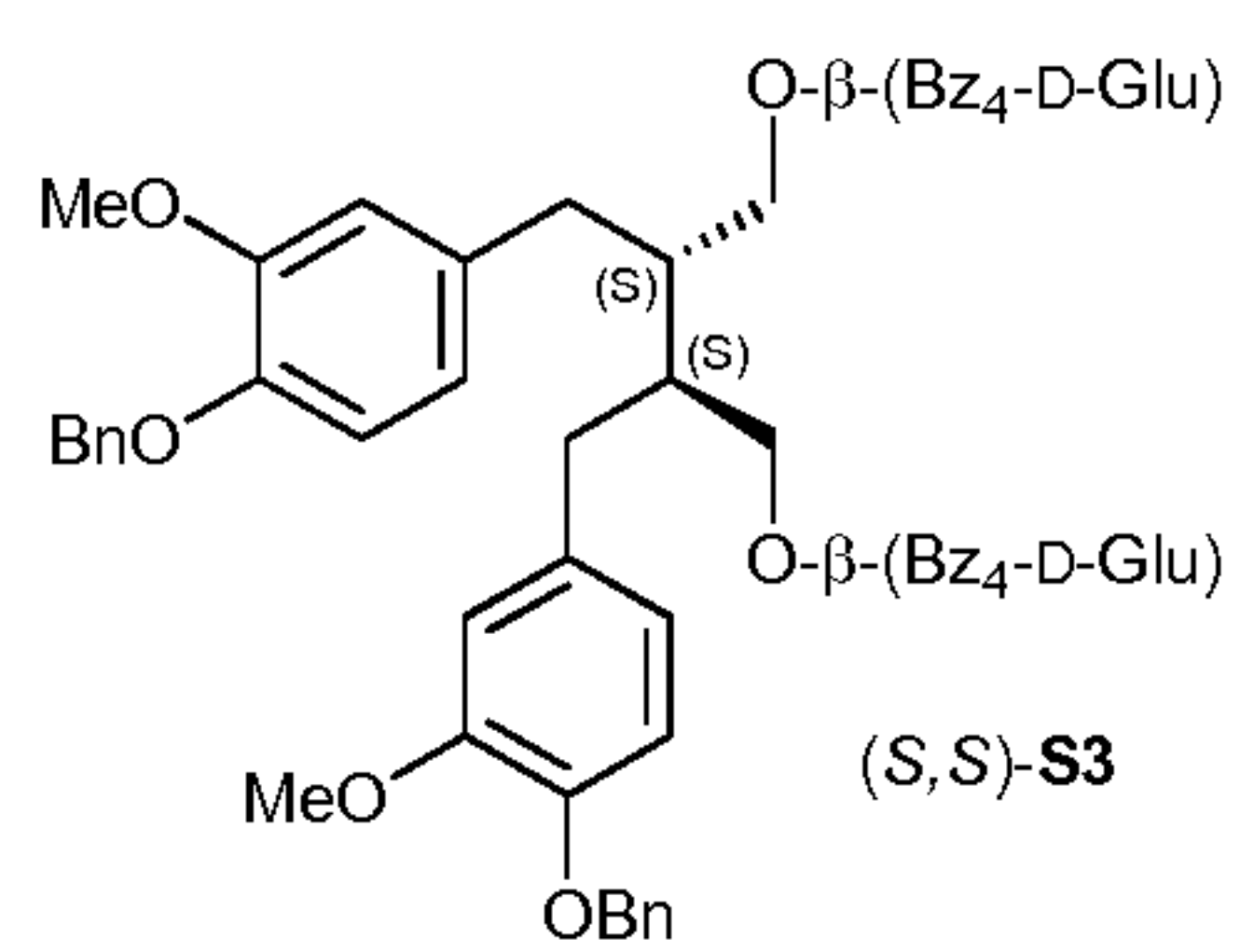


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with a compound of formula (7)

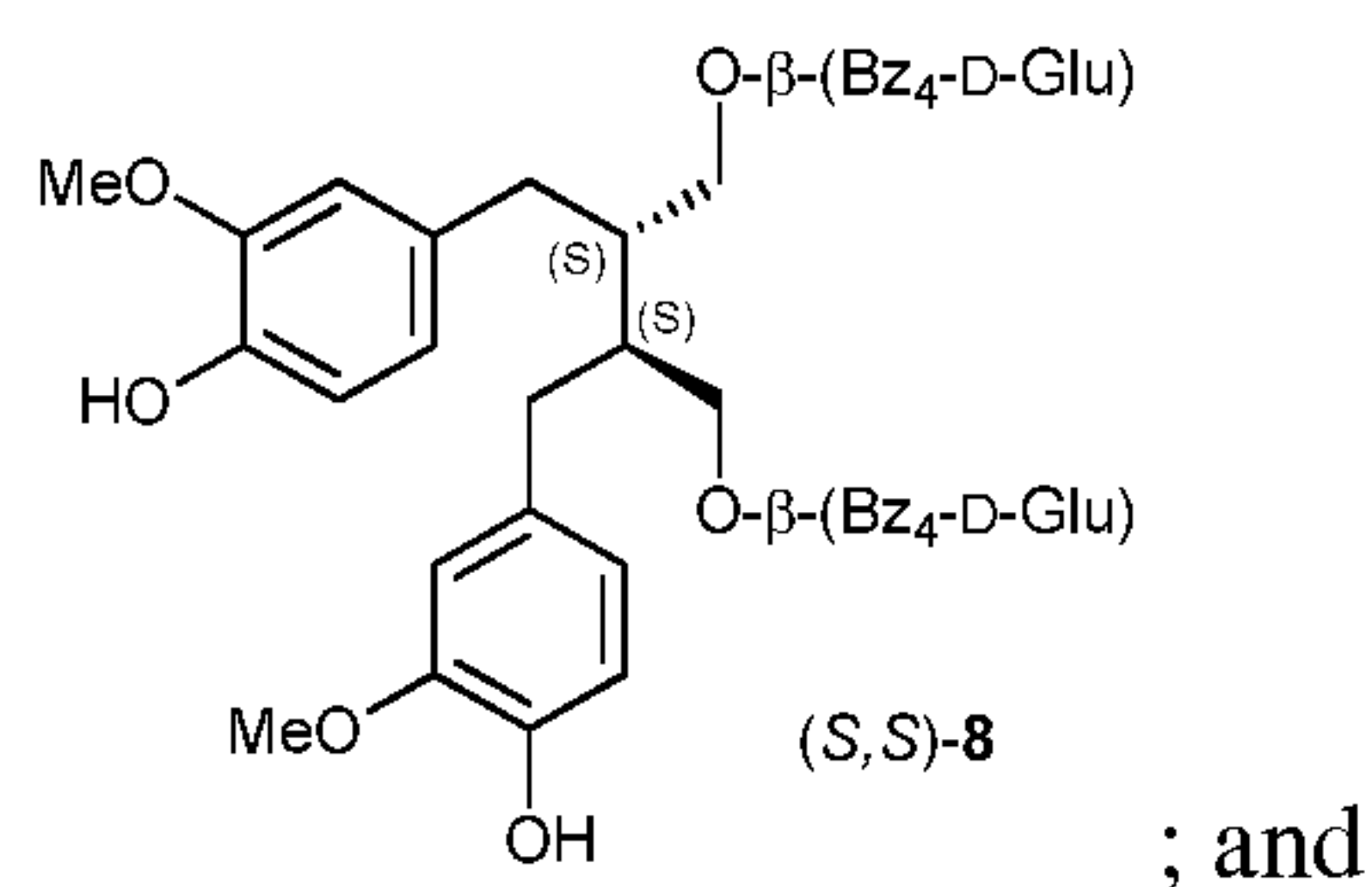


to prepare a compound of formula ((S,S)-S3)



(b) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare a compound of formula ((S,S)-8)

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; and

(c) deprotecting the compound of formula ((*S,S*)-8) to prepare the compound of formula ((*S,S*)-SDG-1).

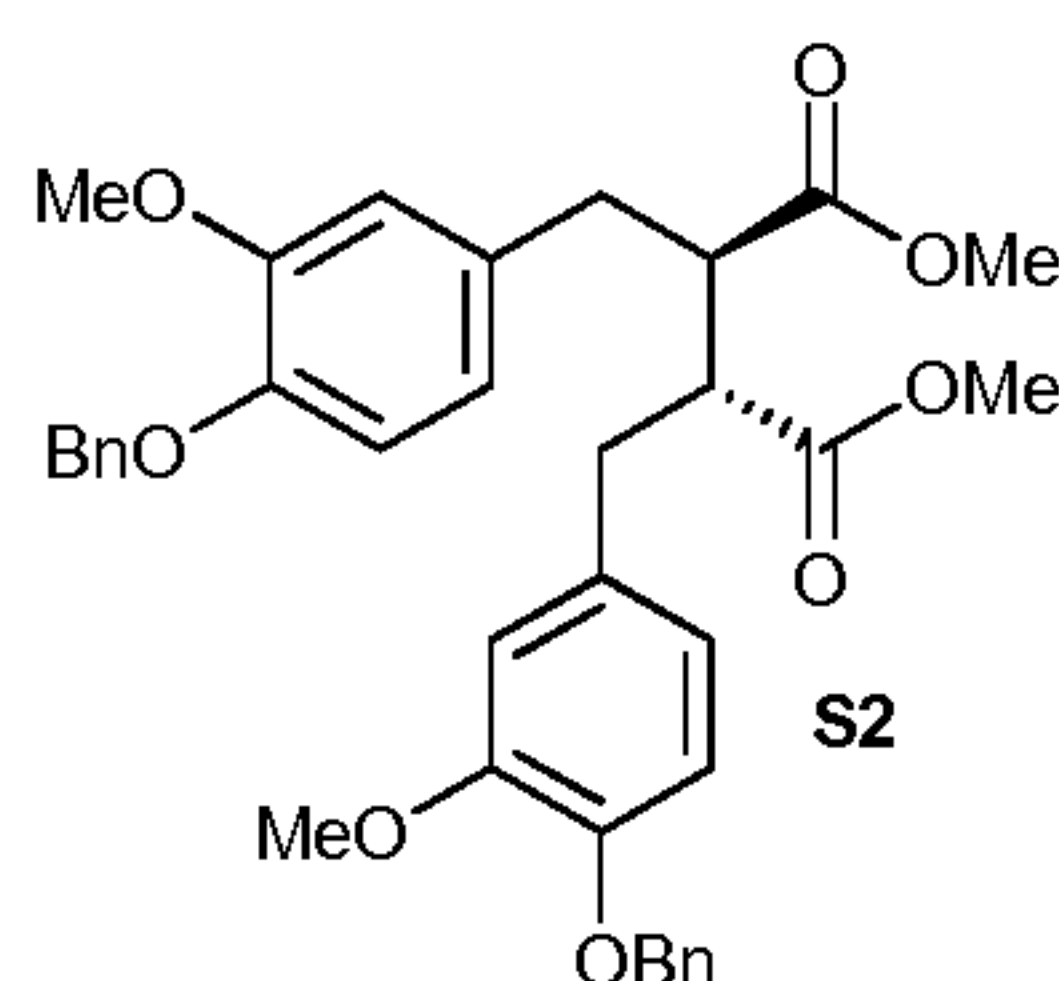
[0035] In some embodiments, said reacting is carried out in the presence of a Lewis acid. In certain embodiments, said Lewis acid is TMSOTf. In some embodiments, said reacting is carried out in the presence of activated molecular sieves.

[0036] In some embodiments, said cleaving is carried out in MeOH in the presence of H₂ and Pd/C.

[0037] In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.

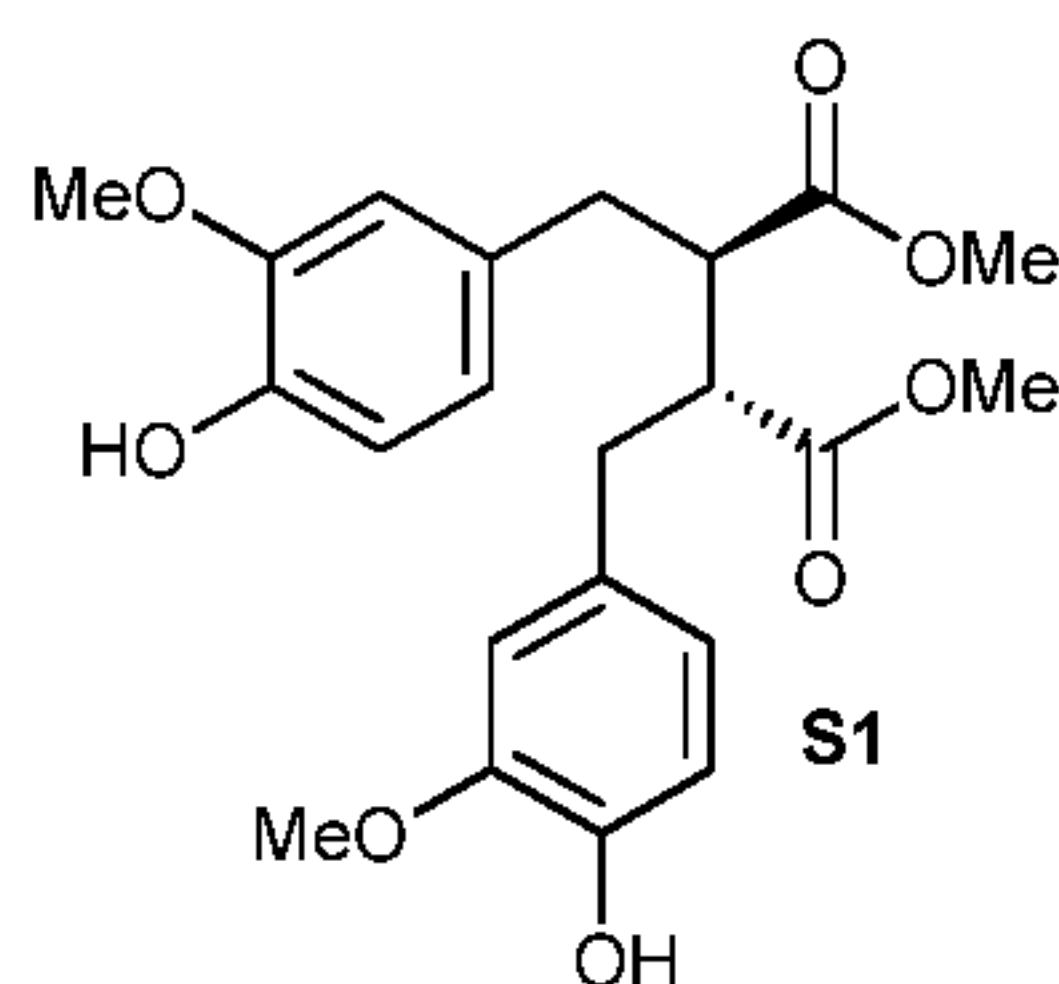
[0038] In some embodiments, said deprotecting is carried out in a solution of NaOMe and MeOH.

[0039] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



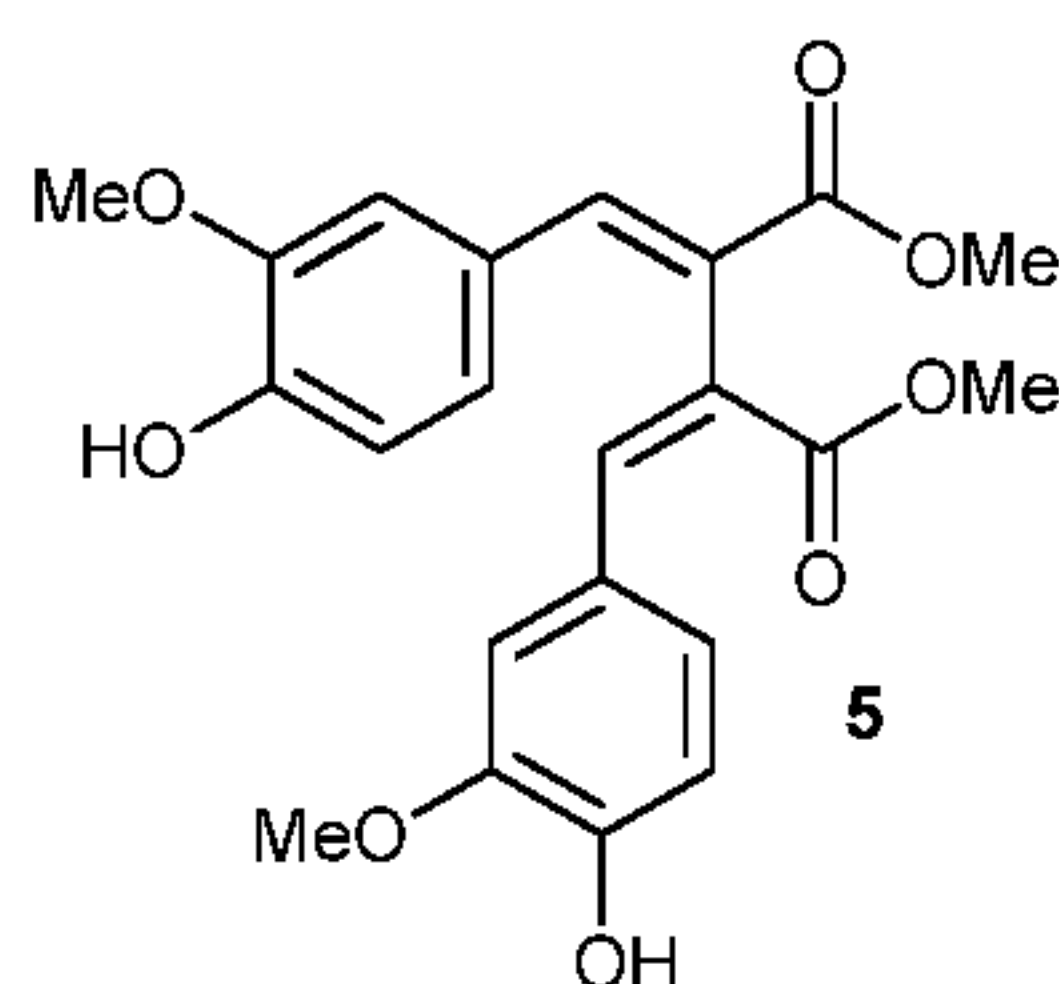
with a reducing agent to prepare the compound of formula (6). In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[0040] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



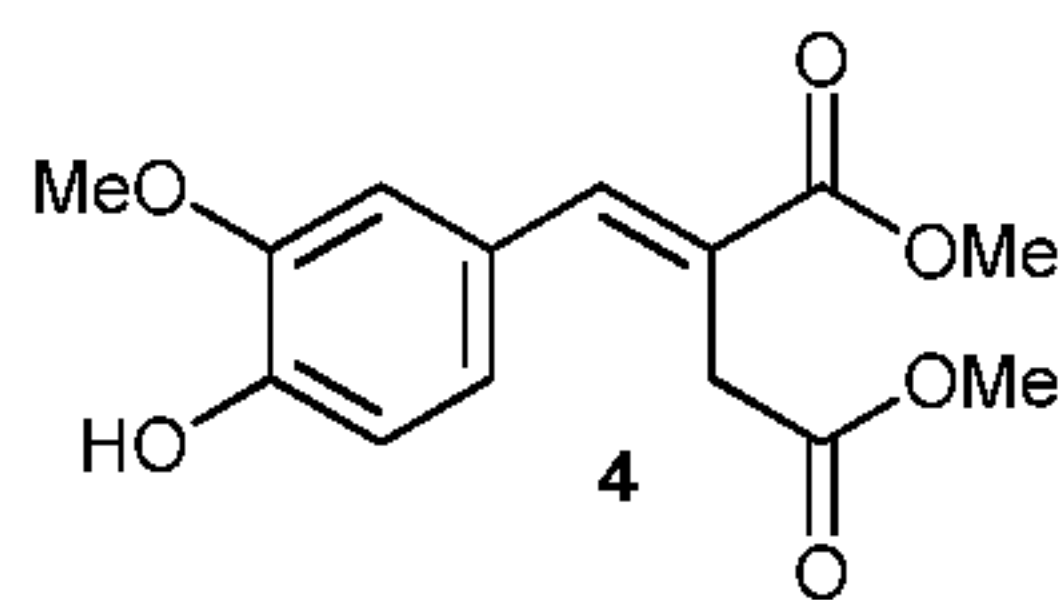
with a benzylating agent to prepare the compound of formula (S2). In some embodiments, said benzylating agent is BnBr and NaH.

[0041] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In some embodiments, said reducing agent is H_2 and Pd/C.

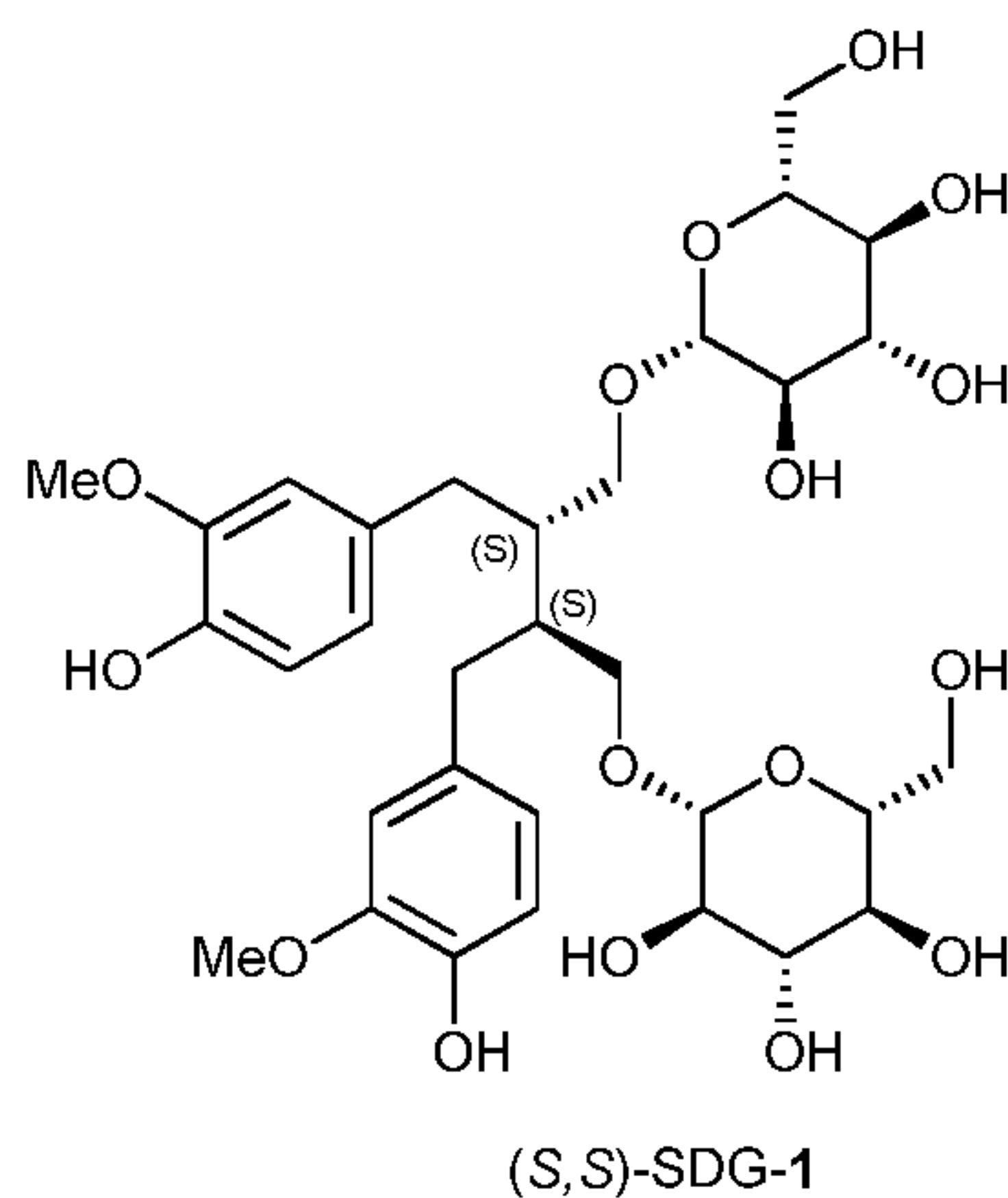
[0042] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium (e.g., lithium wires). In some embodiments, said esterification reaction is carried out with MeOH in the presence of H_2SO_4 .

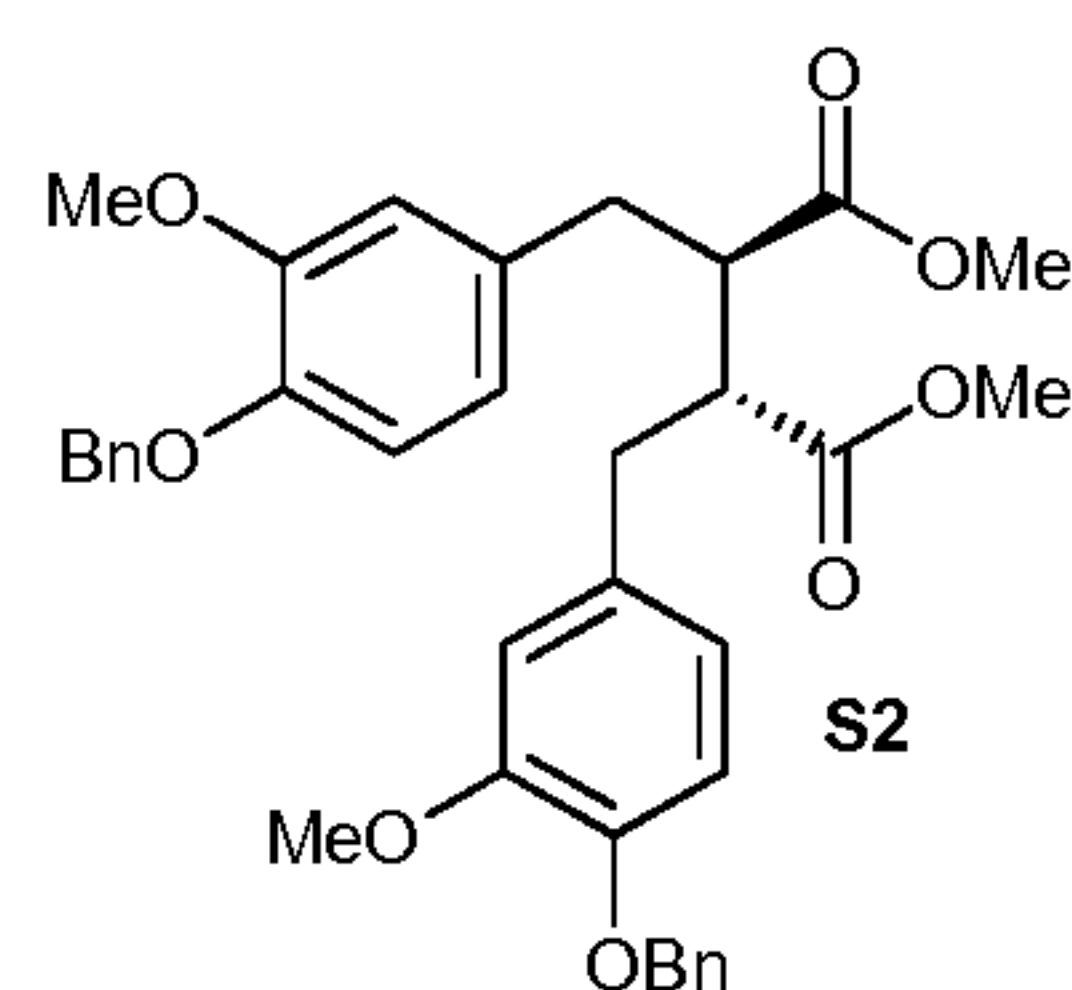
[0043] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H_2SO_4 .

[0044] The present invention provides a process for preparing a compound of formula ((S,S)-SDG-1)

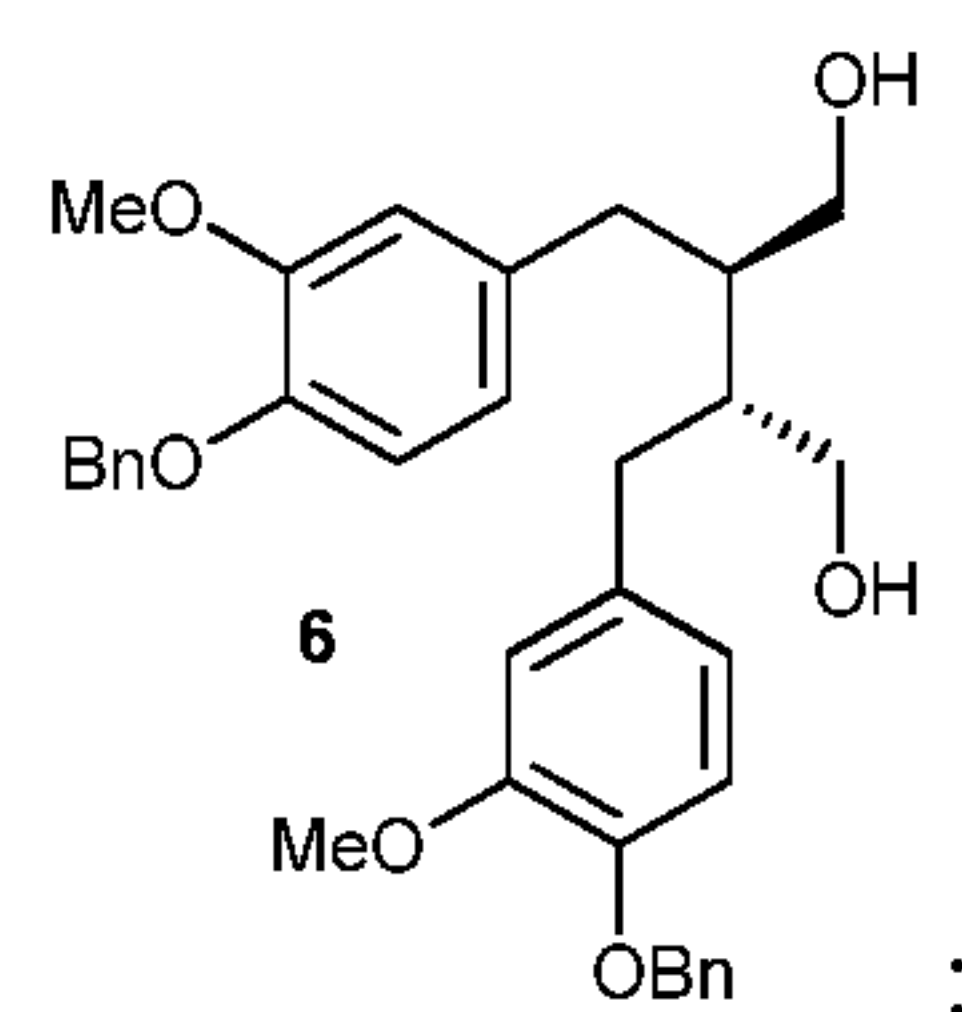


the process comprising:

(a) reacting a compound of formula (S2)

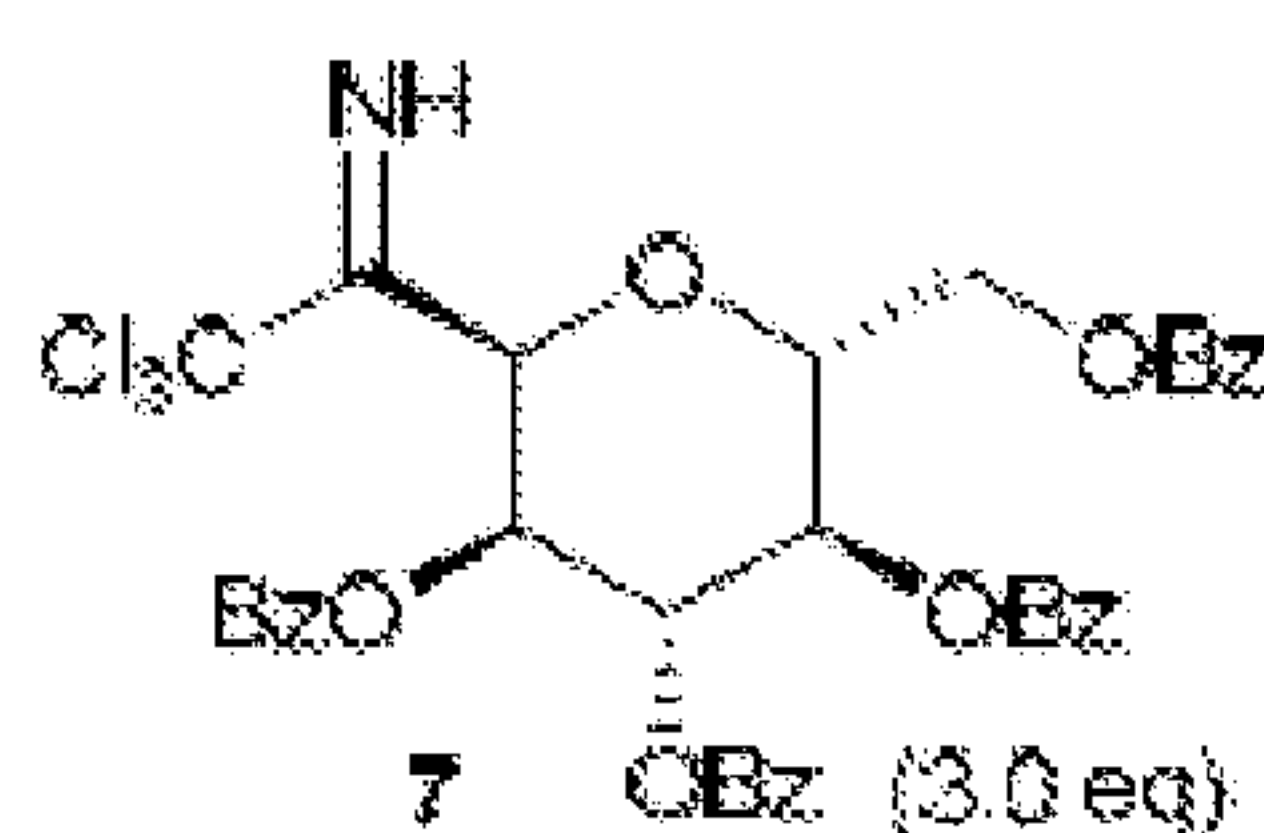


with a reducing agent to prepare a compound of formula (6)

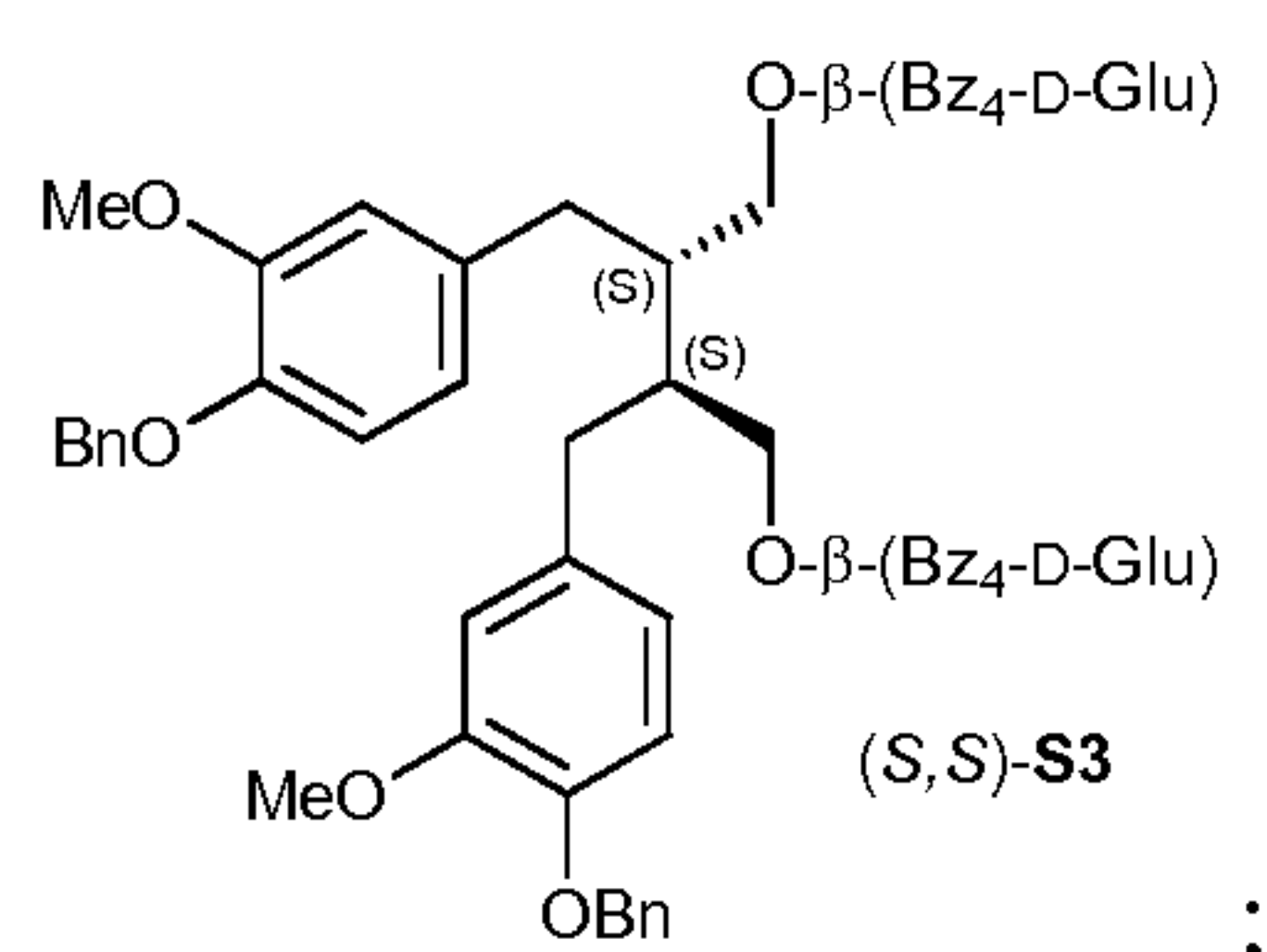


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(b) reacting the compound of formula (6) with a compound of formula (7)

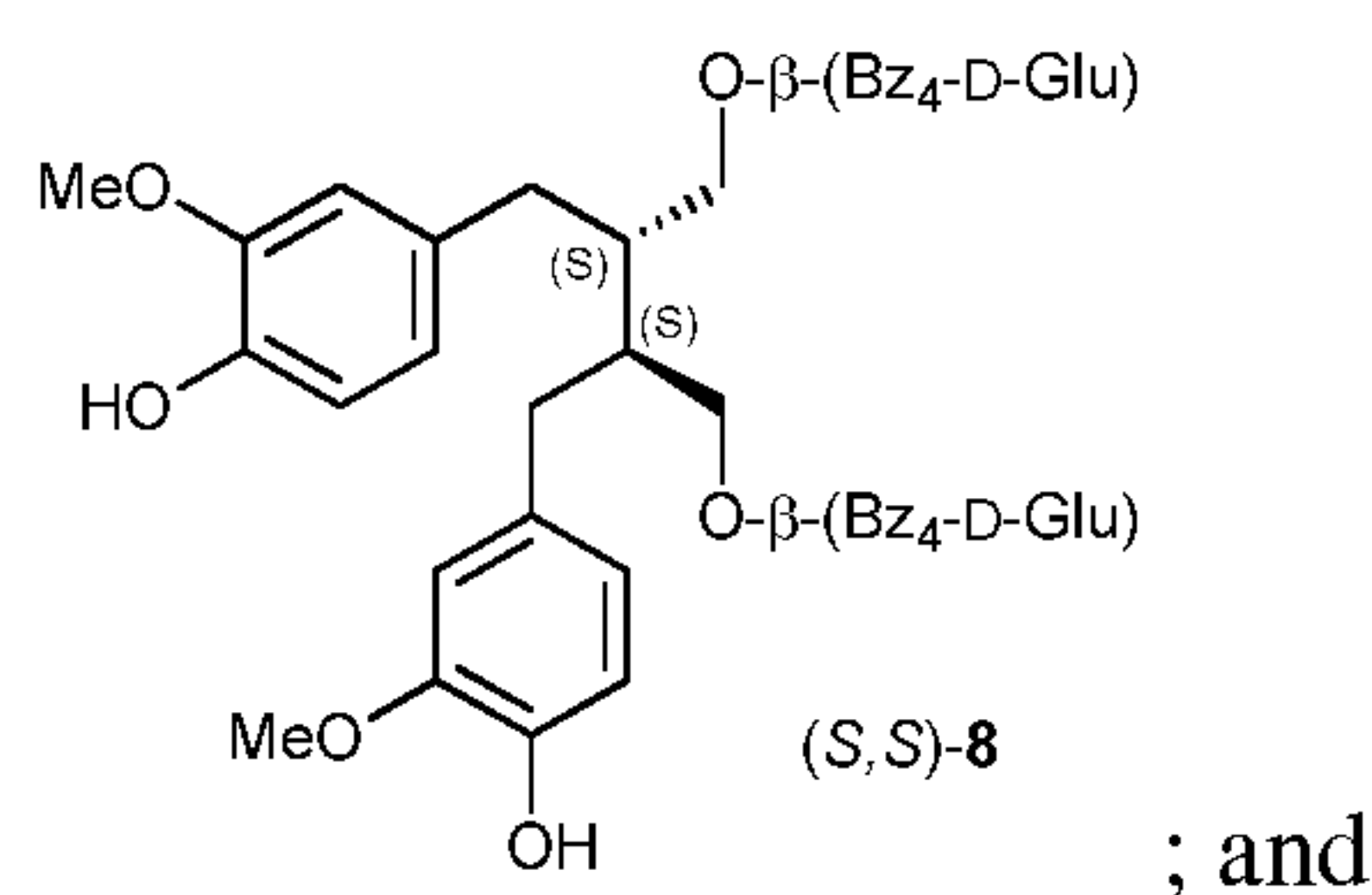


to prepare a compound of formula ((S,S)-S3)



(c) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare a compound of formula ((S,S)-8)

10



(d) deprotecting the compound of formula ((S,S)-8) to prepare the compound of formula ((S,S)-SDG-1).

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[0045] In some embodiments, said reacting of step (a) is carried out in the presence of TMSOTf. In some embodiments, said reacting of step (a) is carried out in the presence of activated molecular sieves.

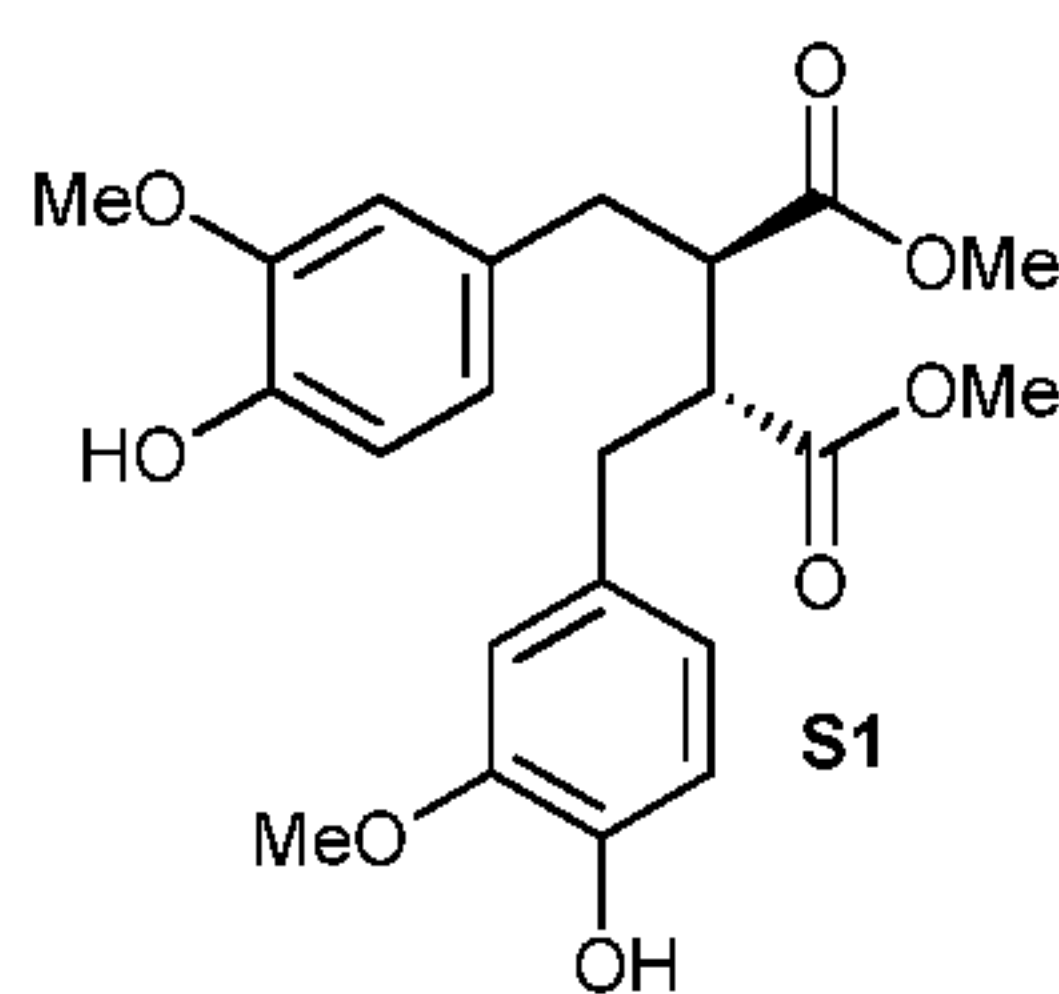
[0046] In some embodiments, said cleaving is carried out in the presence H₂ and Pd/C in MeOH.

[0047] In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.

[0048] In some embodiments, said deprotecting is carried out in a solution of NaOMe and MeOH.

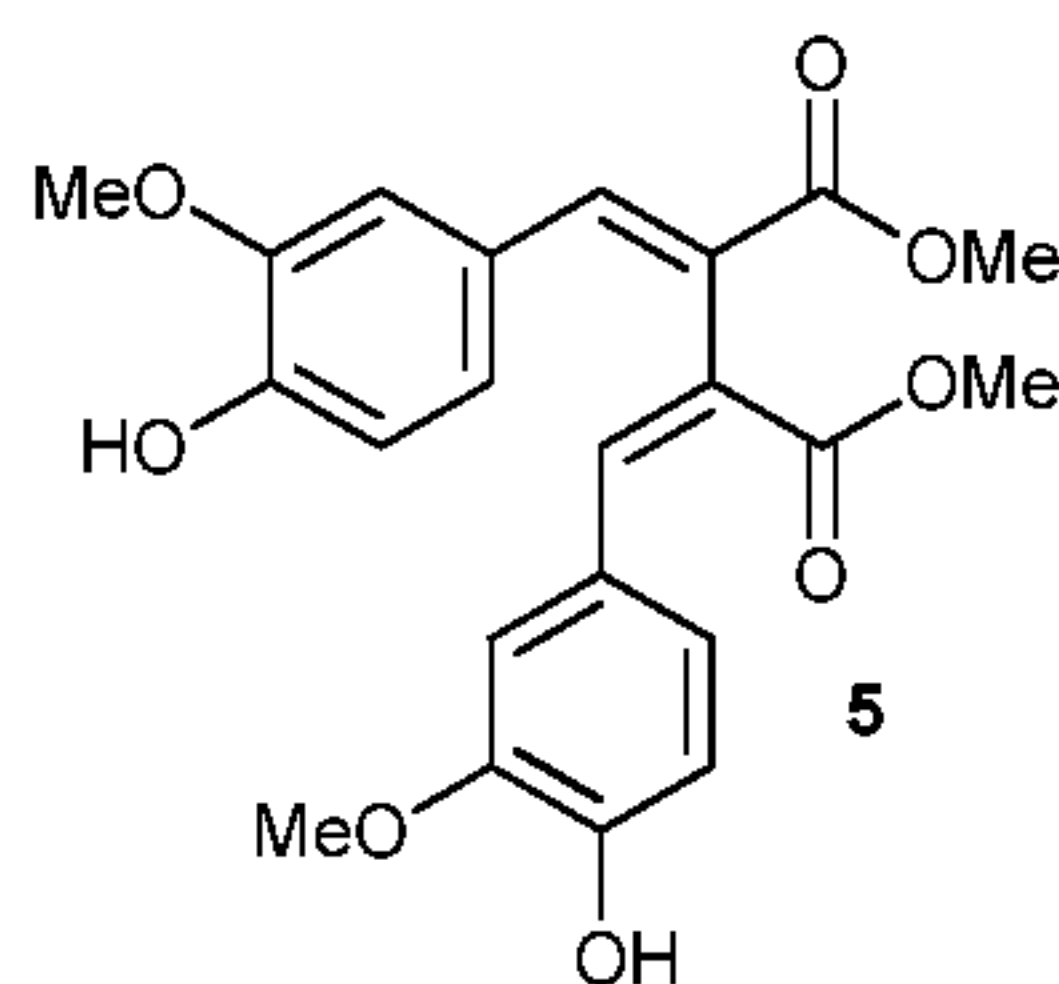
[0049] In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[0050] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



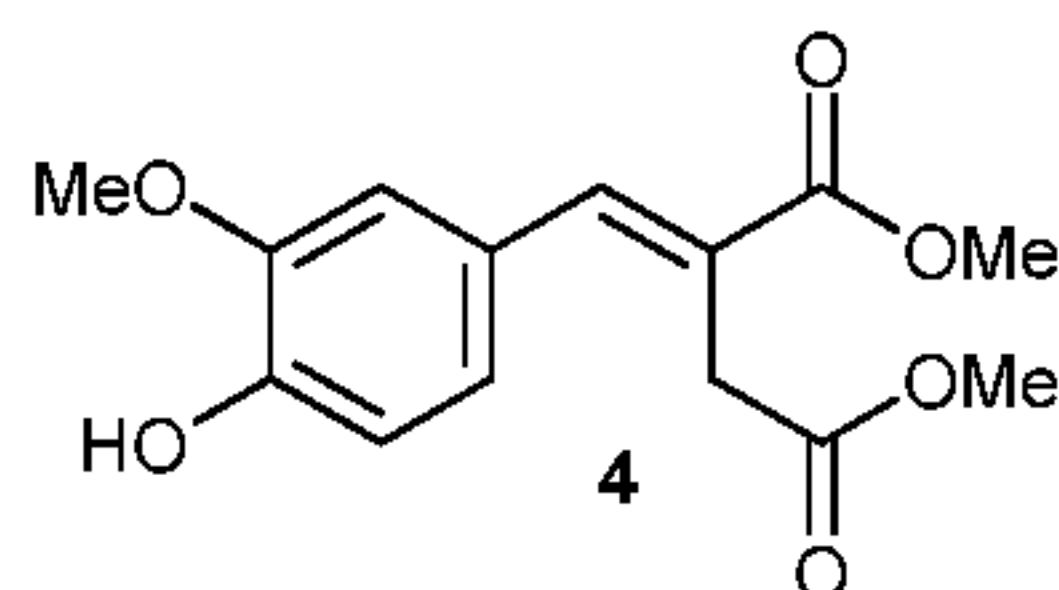
with a benzylating agent to form the compound of formula (S2). In some embodiments, said benzylating agent is BnBr and NaH.

[0051] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In some embodiments, said reducing agent is H₂ and Pd/C.

[0052] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

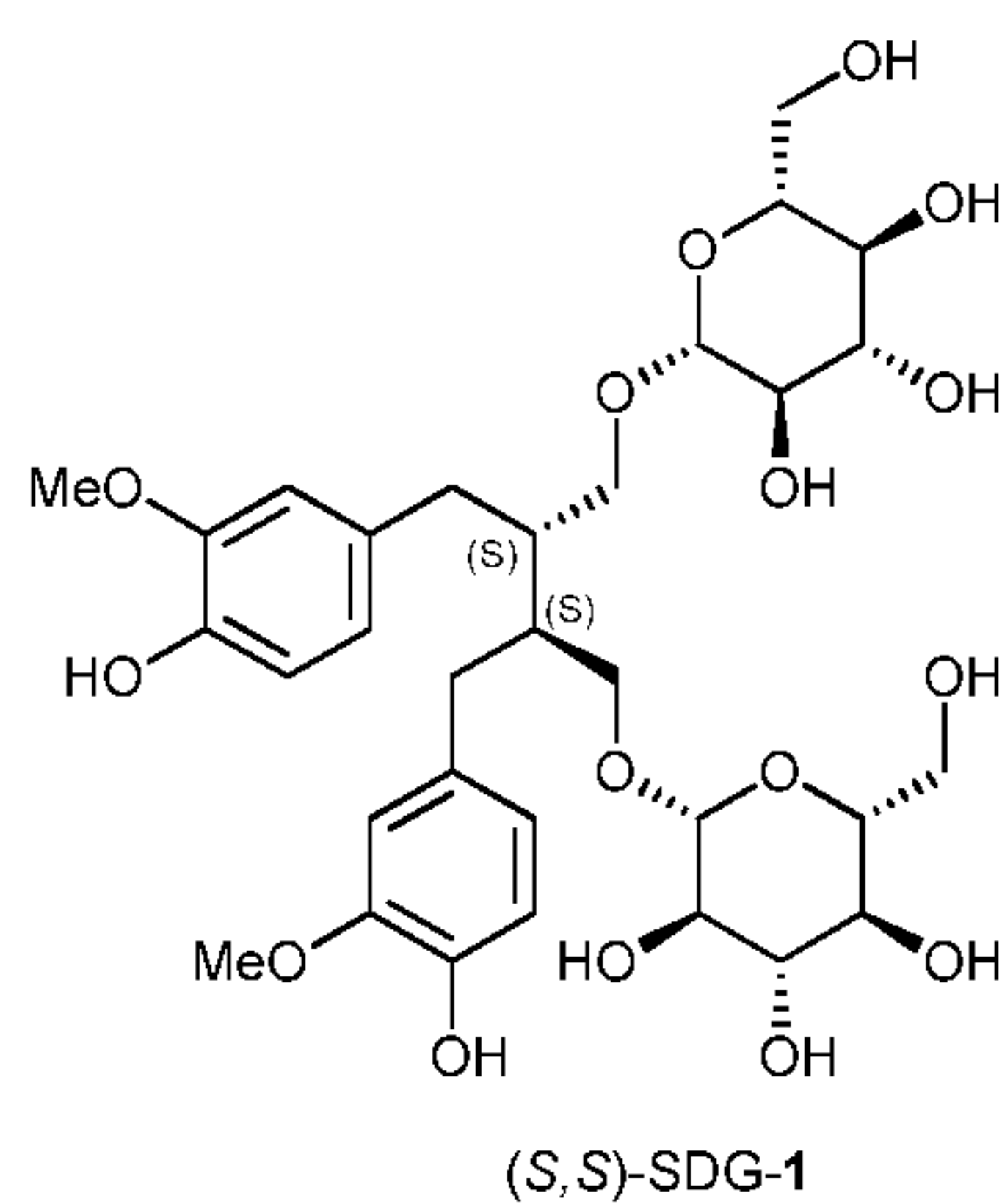
[0053] In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

[0054] In some embodiments, In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0055] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

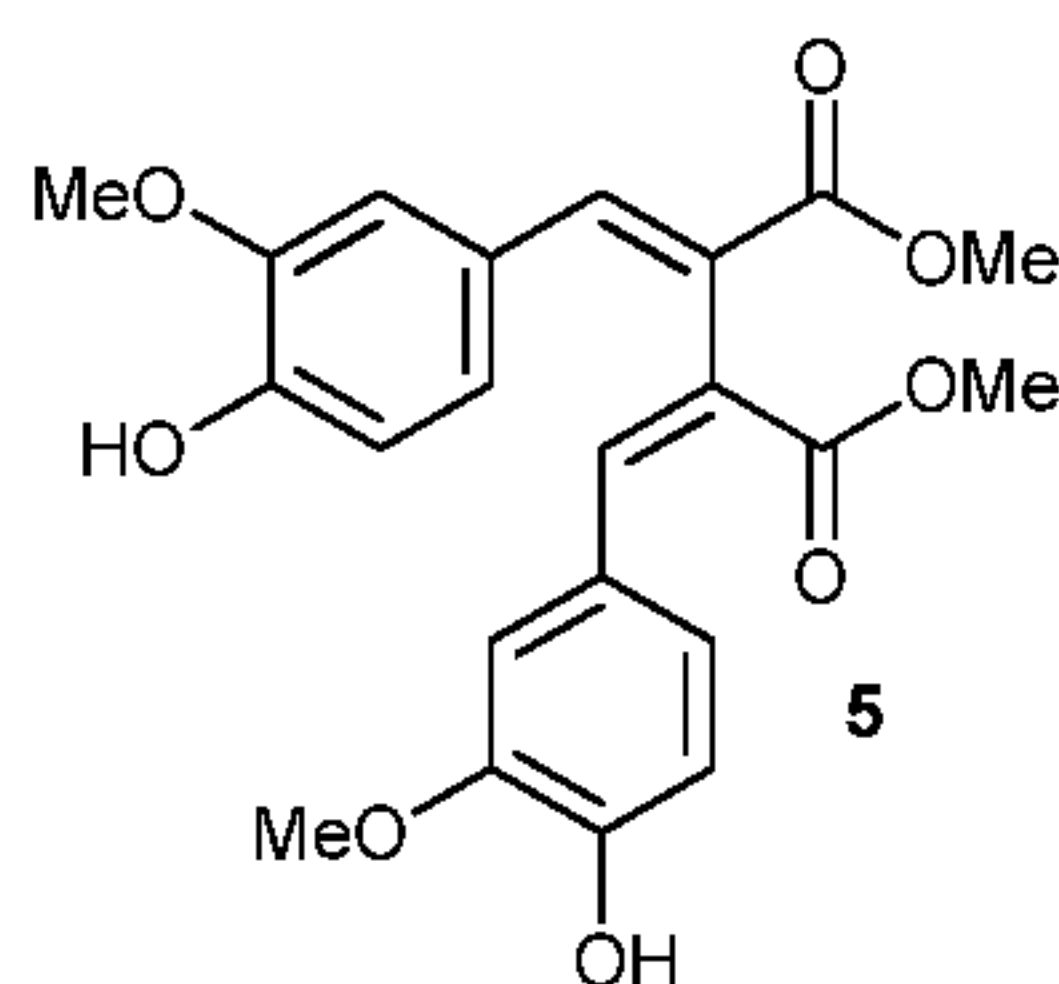
[0056] In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0057] The present invention provides a process for preparing a compound of formula ((S,S)-SDG-1)

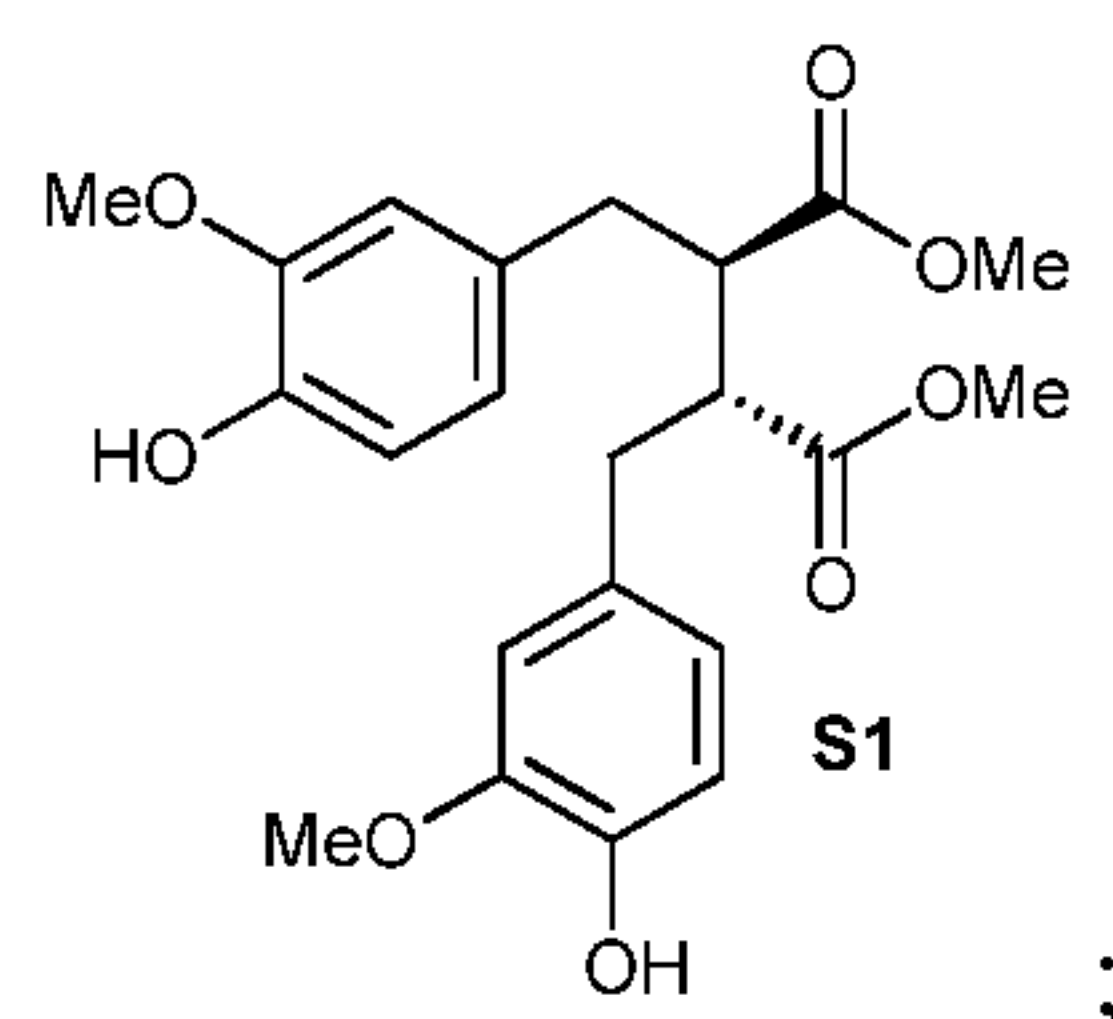


the process comprising:

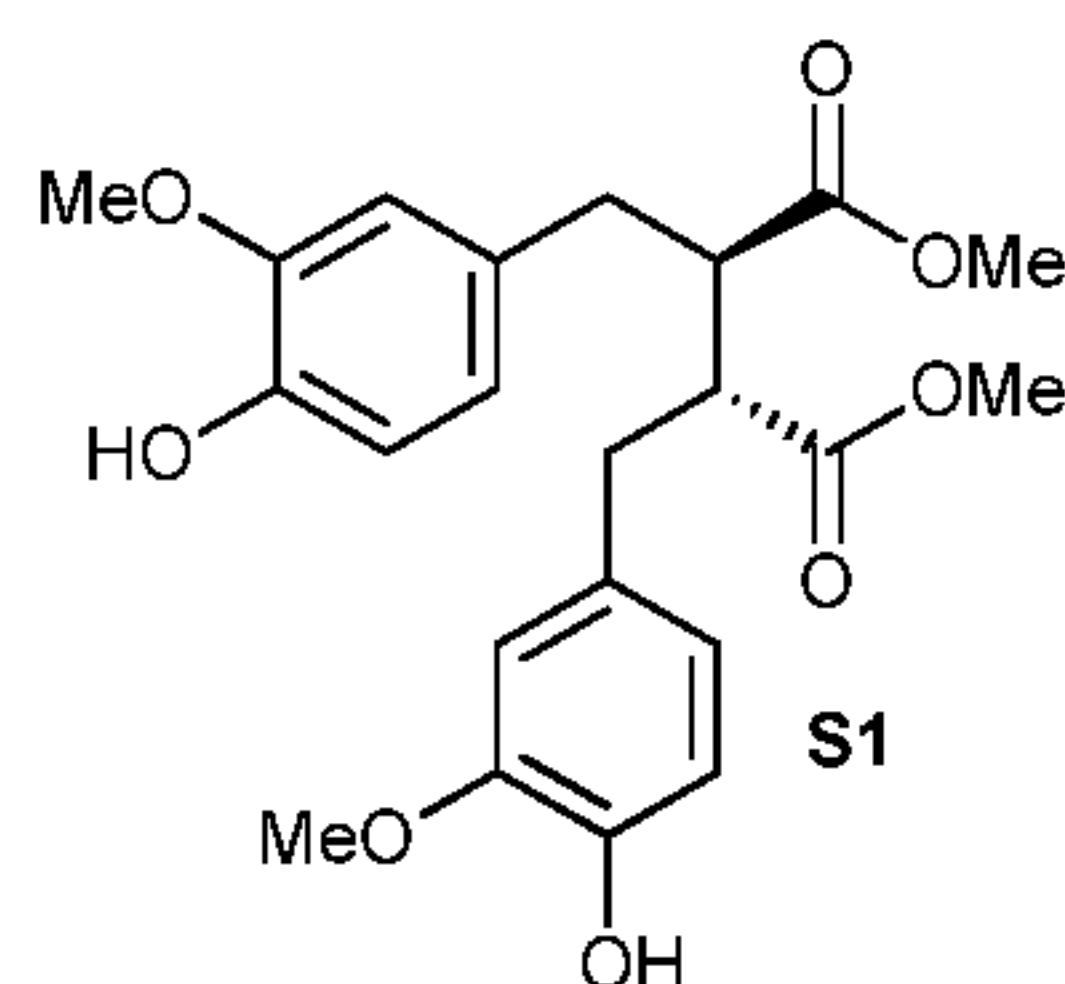
(a) reacting a compound of formula (5)



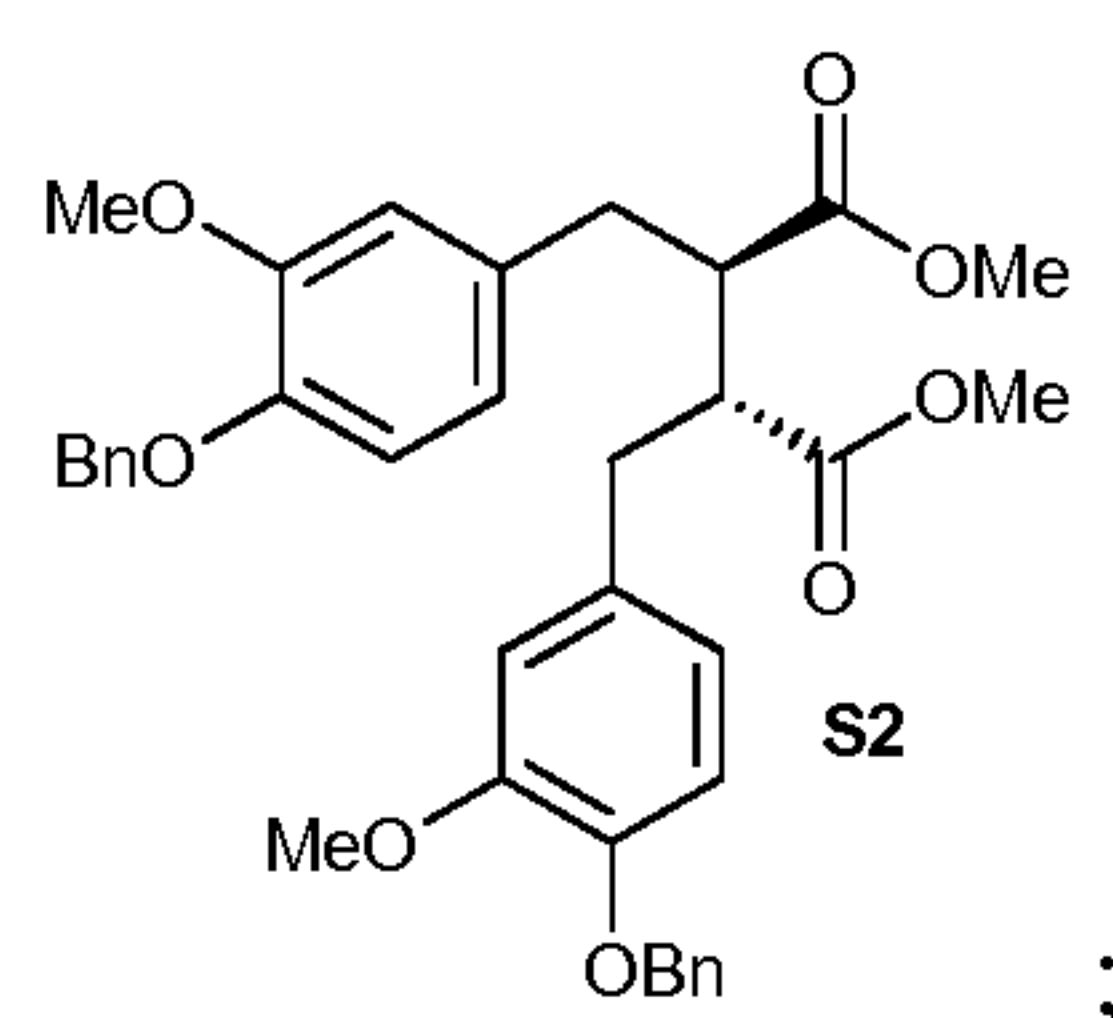
with a reducing agent to prepare a compound of formula (S1)



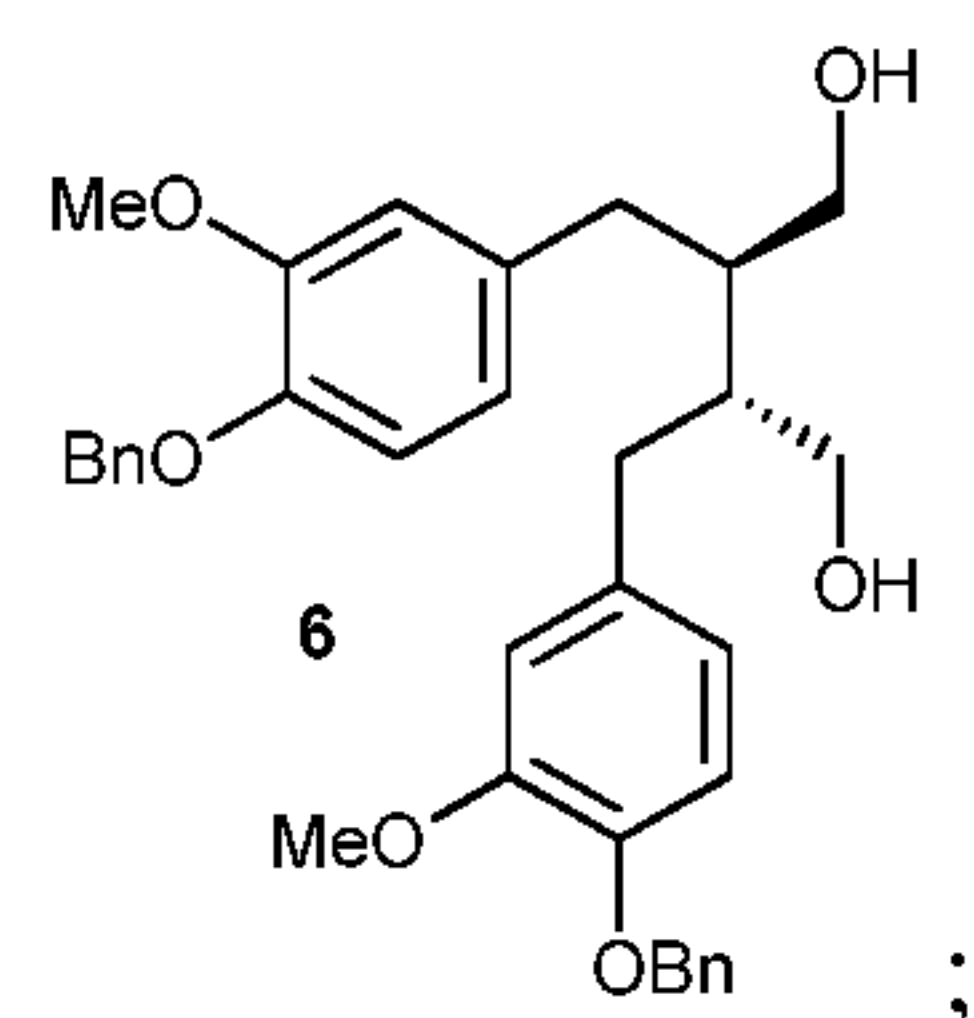
(b) reacting the compound of formula (S1)



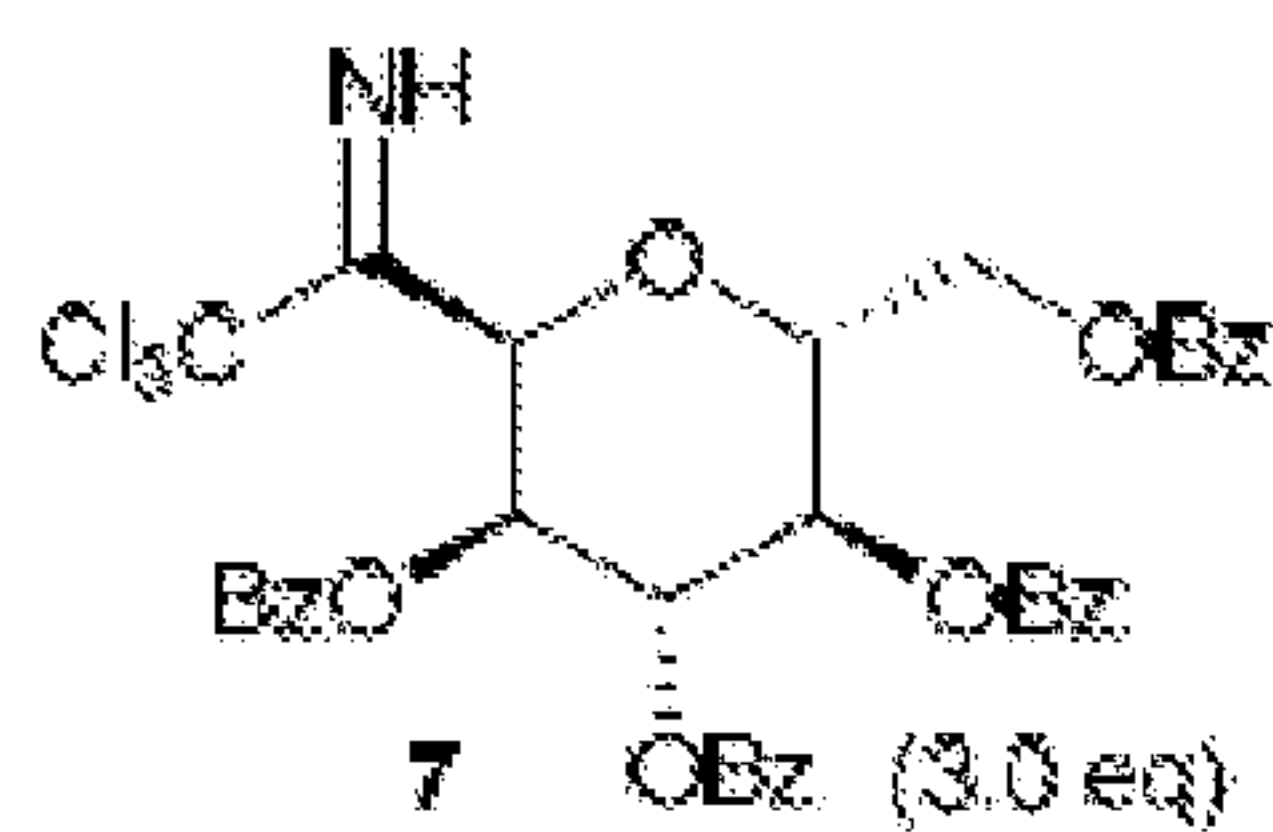
5 with a benzylating agent to prepare a compound of formula (S2)



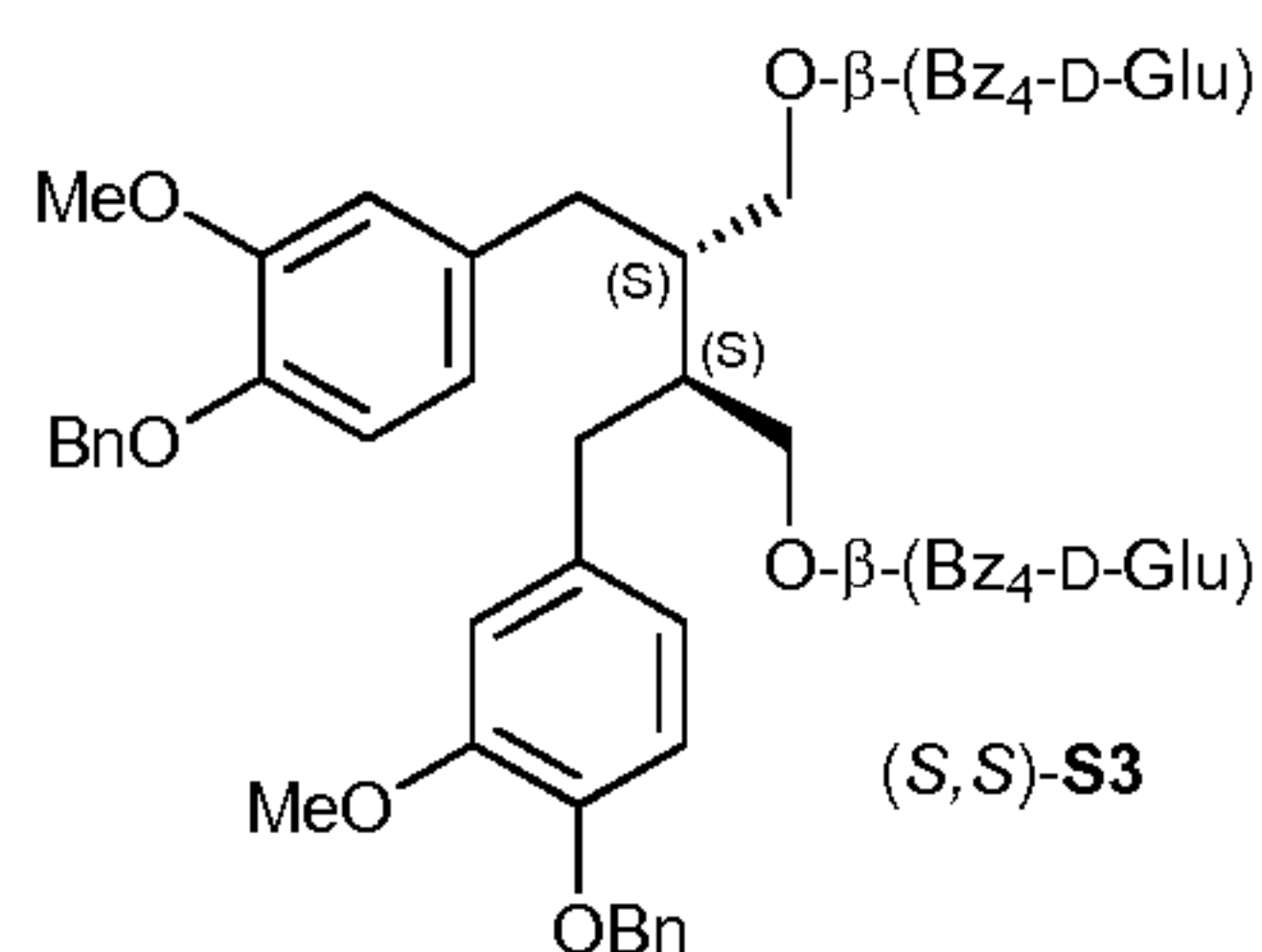
(c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6)



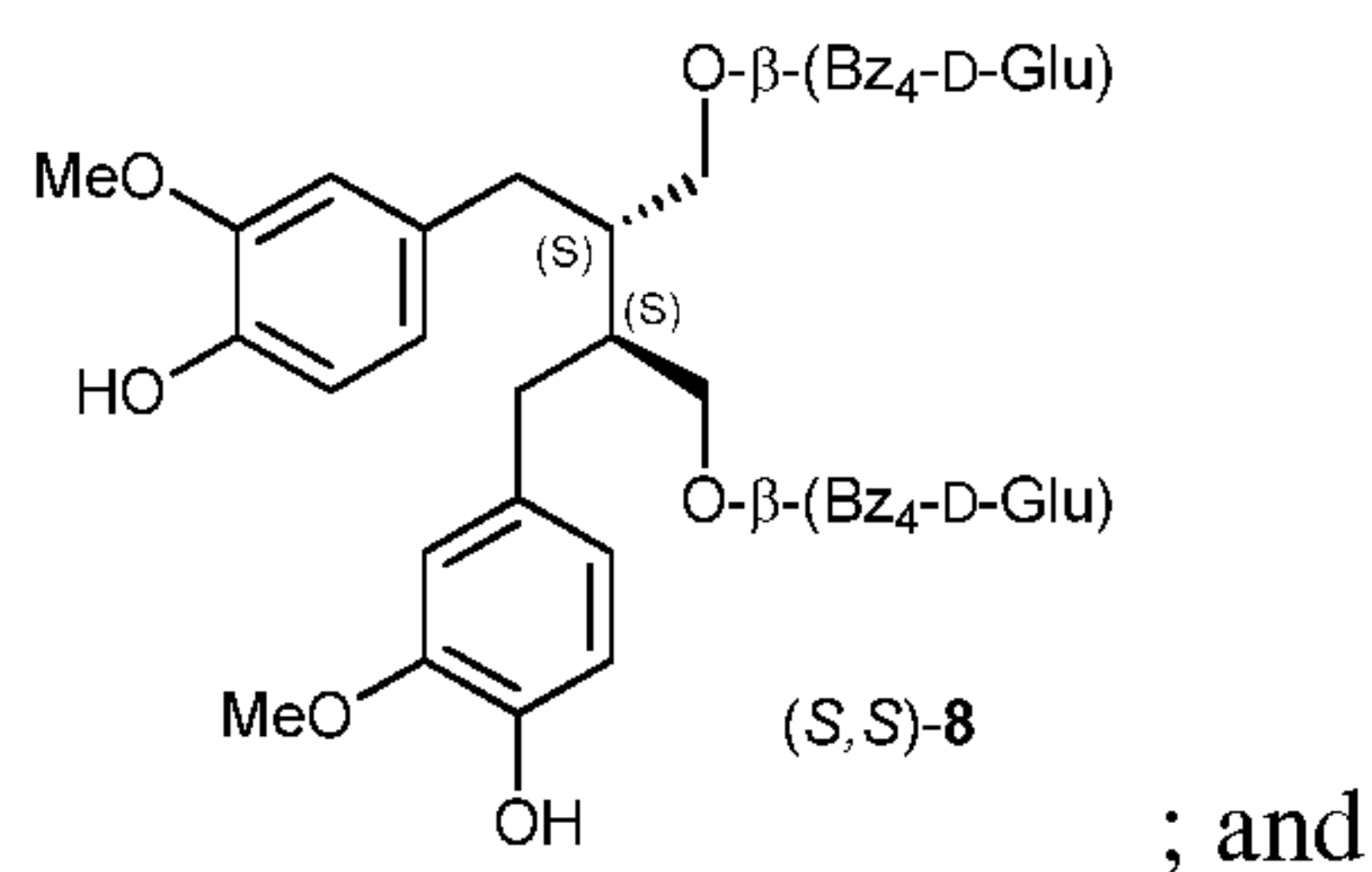
10 (d) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((S,S)-S3)

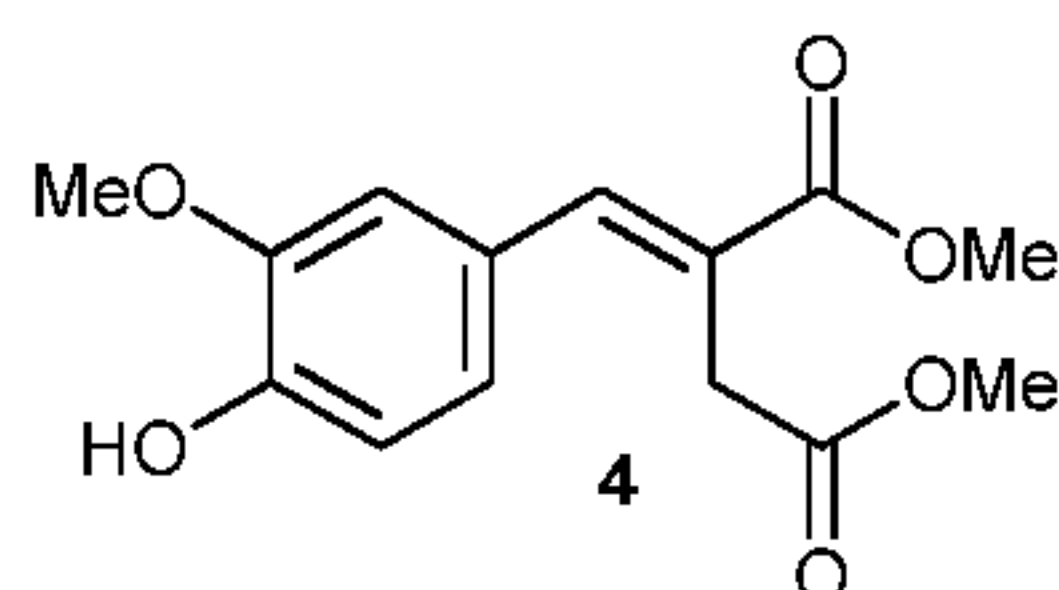


(e) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare a compound of formula ((S,S)-8)



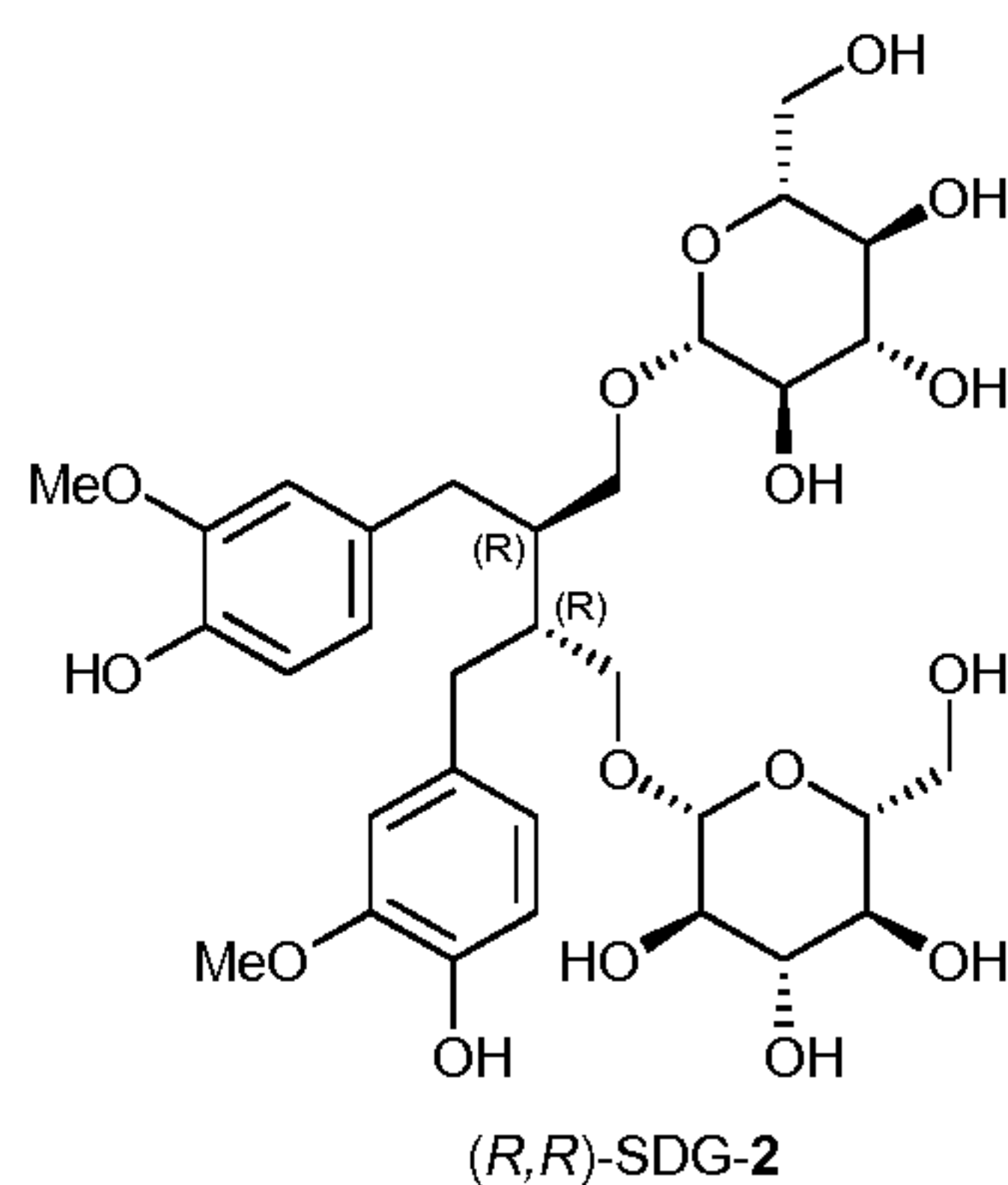
5 (f) deprotecting the compound of formula ((S,S)-8) to prepare the compound of formula ((S,S)-SDG-1).

[0058] In some embodiments, In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



10 with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

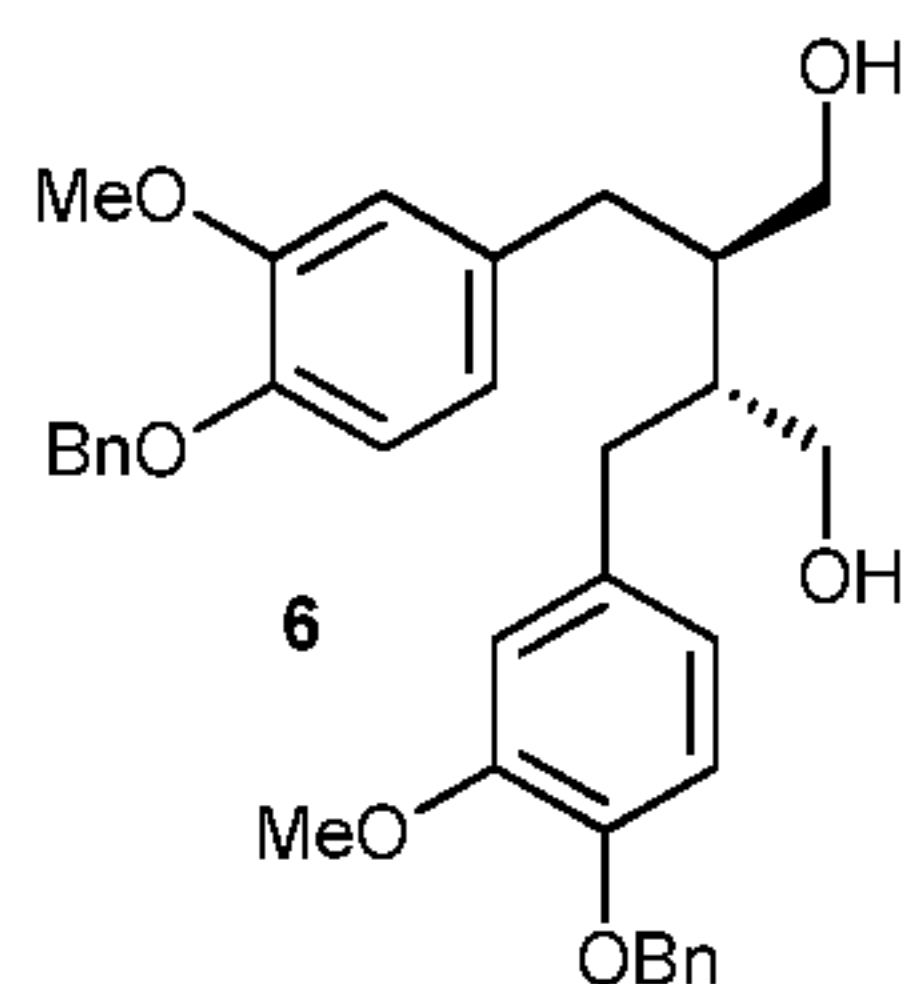
[0059] The present invention provides a process for preparing a compound of formula ((R,R)-SDG-2)



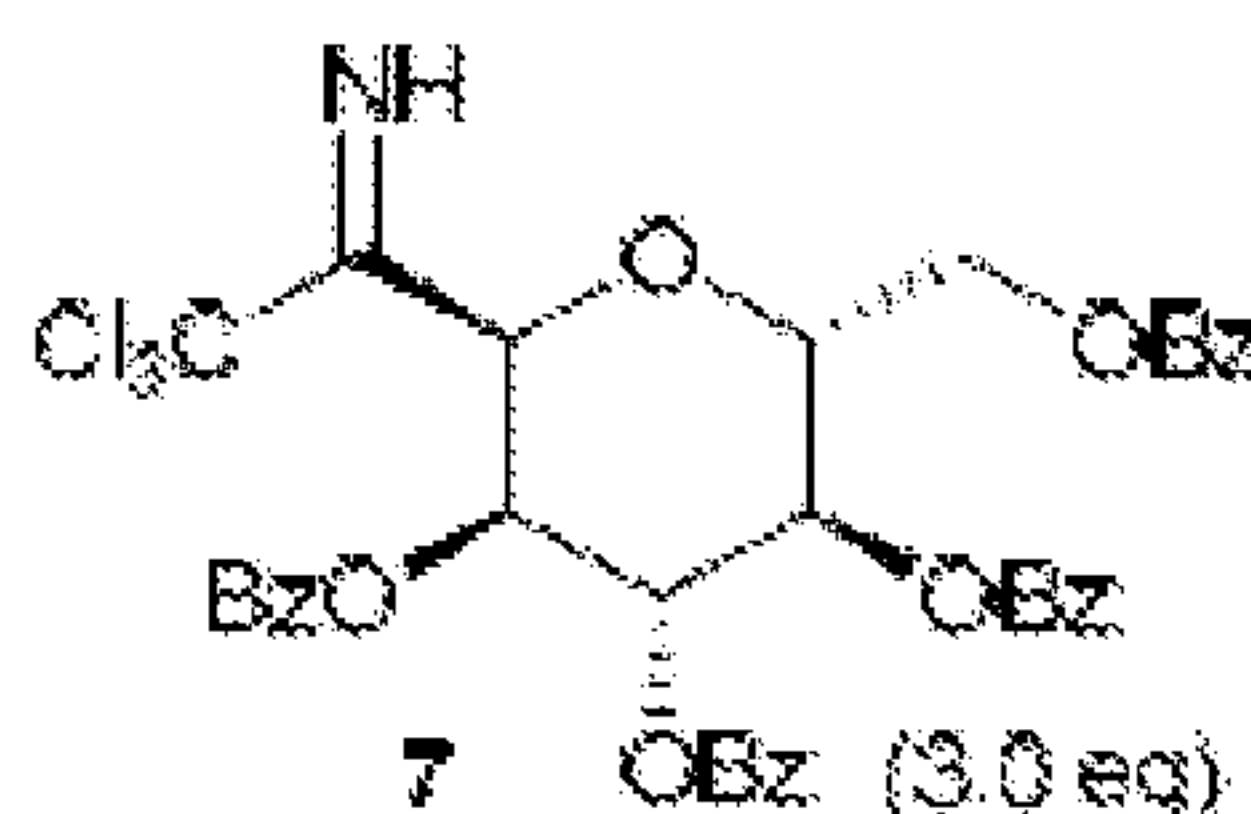
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the process comprising:

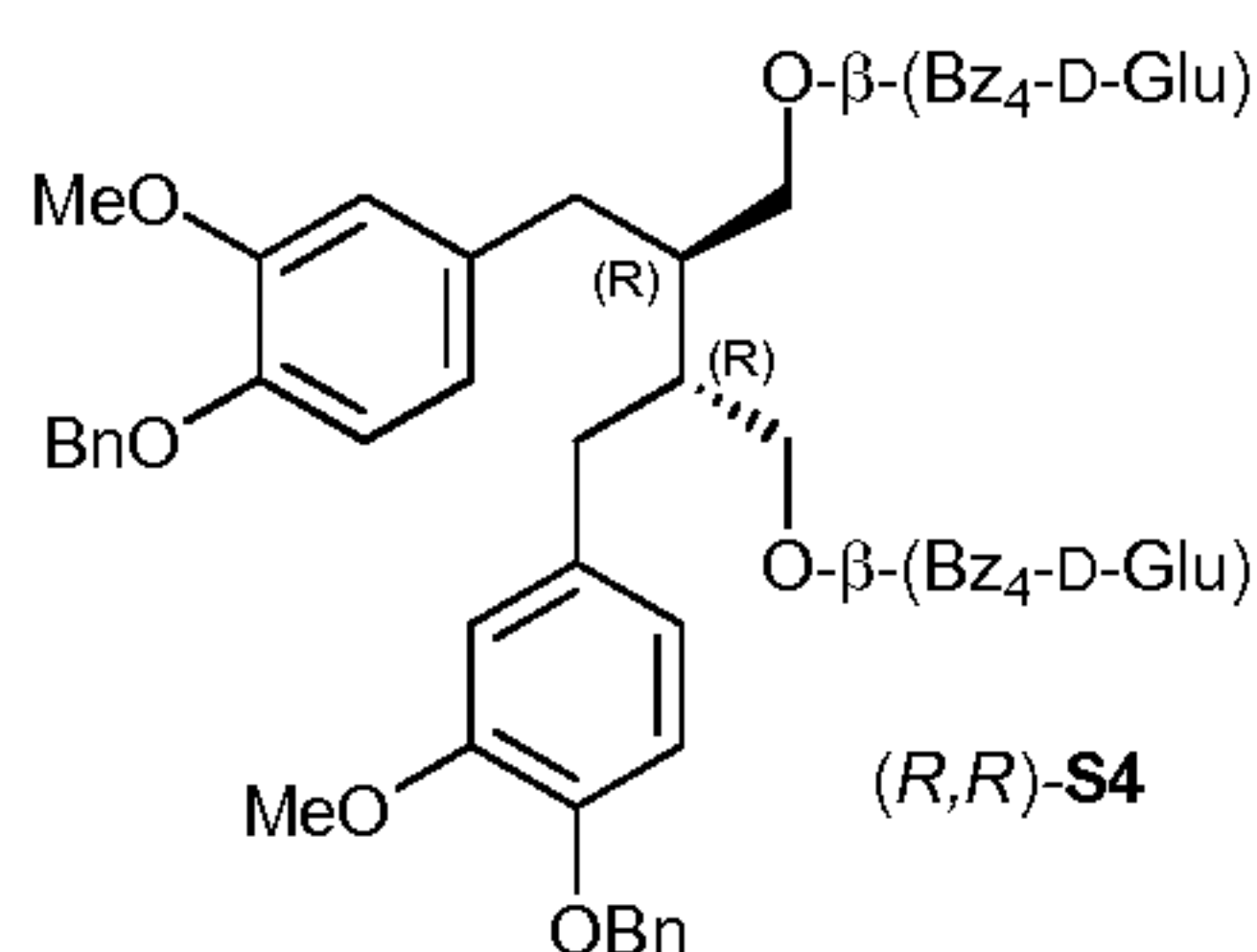
(a) reacting a compound of formula (6)



with a compound of formula (7)

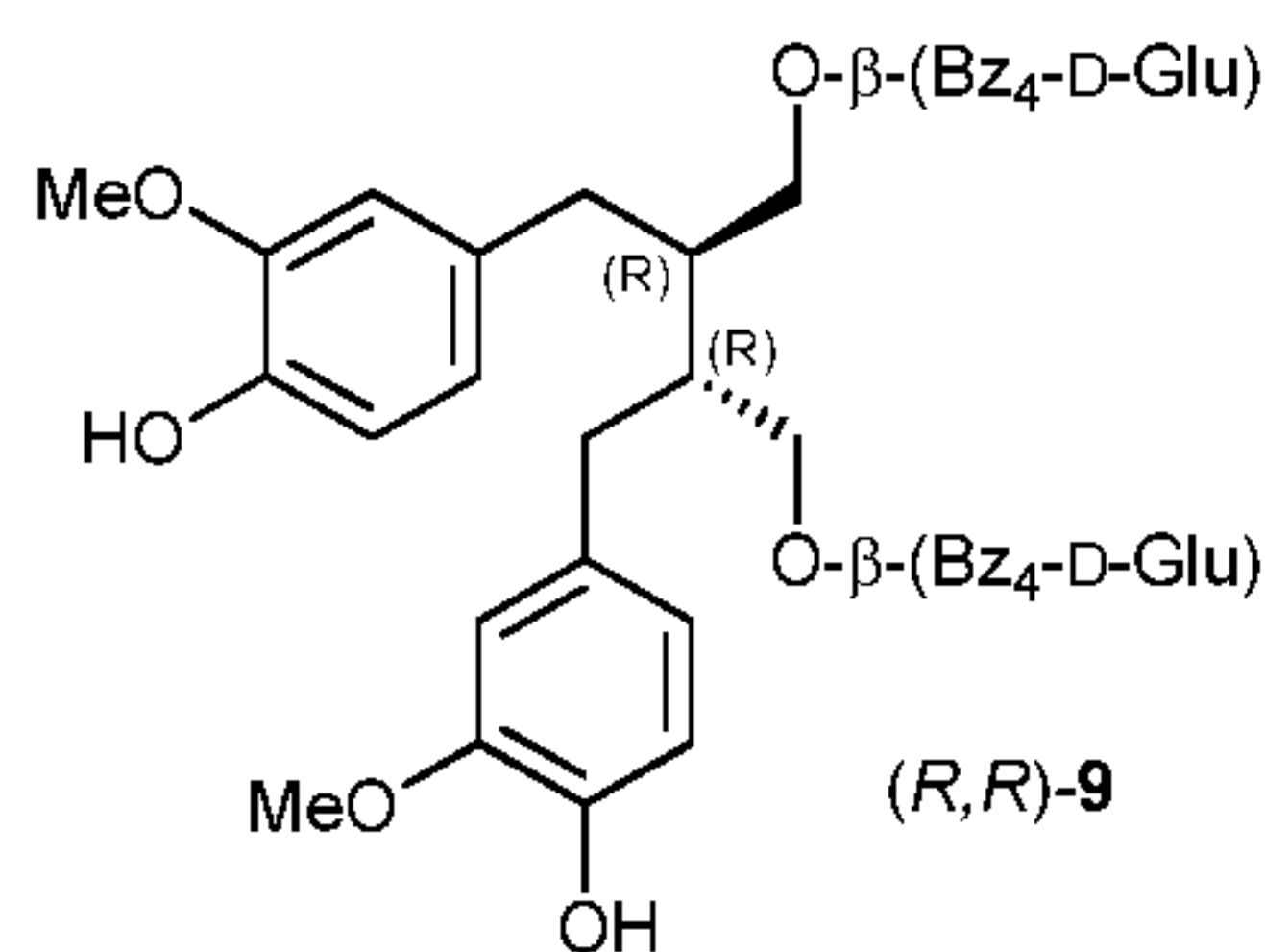


to prepare a compound of formula ((R,R)-S4)



5

(b) cleaving benzyl ethers of the compound of formula ((R,R)-S4), followed by a separation procedure to prepare a compound of formula ((R,R)-9)



; and

(c) deprotecting the compound of formula ((S,S)-9) to prepare the compound of formula ((R,R)-SDG-2).

10

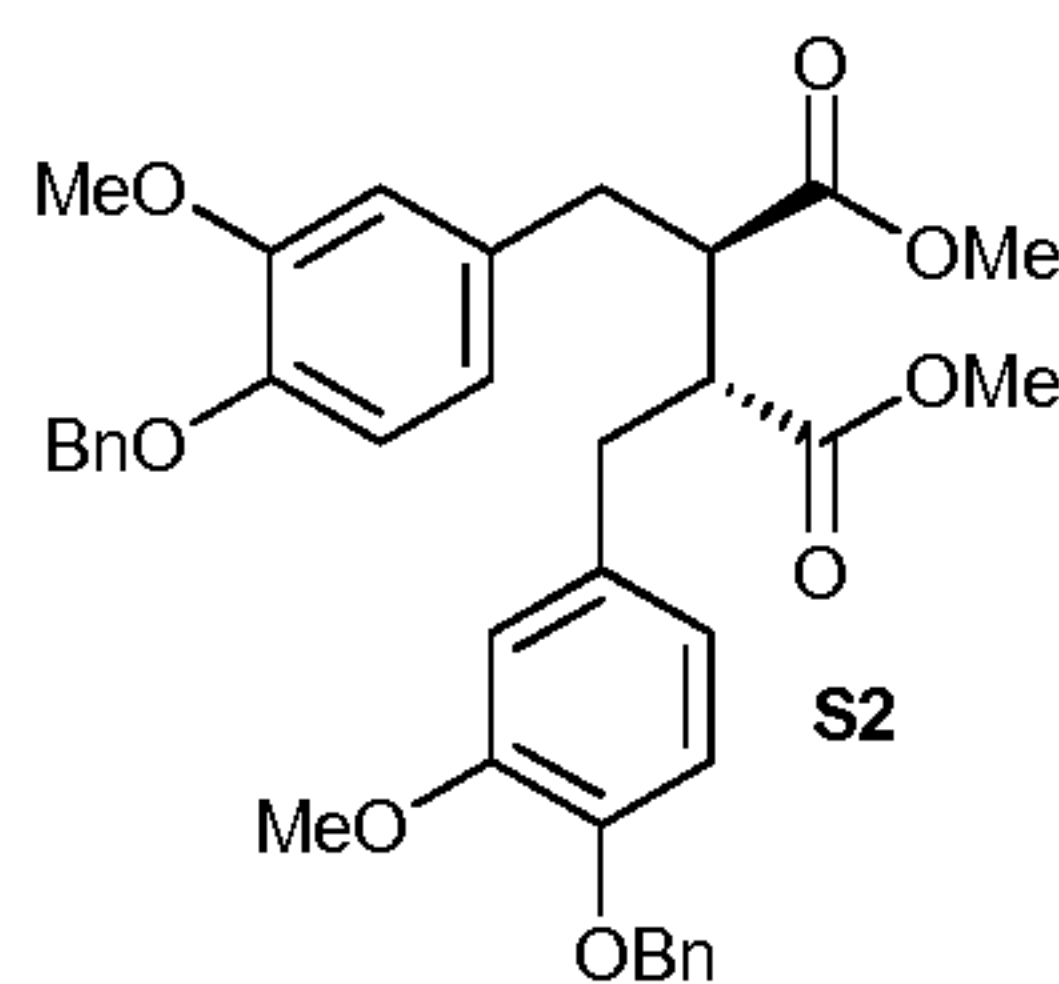
[0060] In some embodiments, said reacting is carried out in the presence of a Lewis acid. In certain embodiments, said Lewis acid is TMSOTf.

[0061] In some embodiments, said reacting is carried out in the presence of activated molecular sieves.

15 [0062] In some embodiment, said cleaving is carried out in the presence of H₂ and Pd/C in MeOH. In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.

[0063] In some embodiments, said deprotecting is carried out in a solution of NaOMe and MeOH.

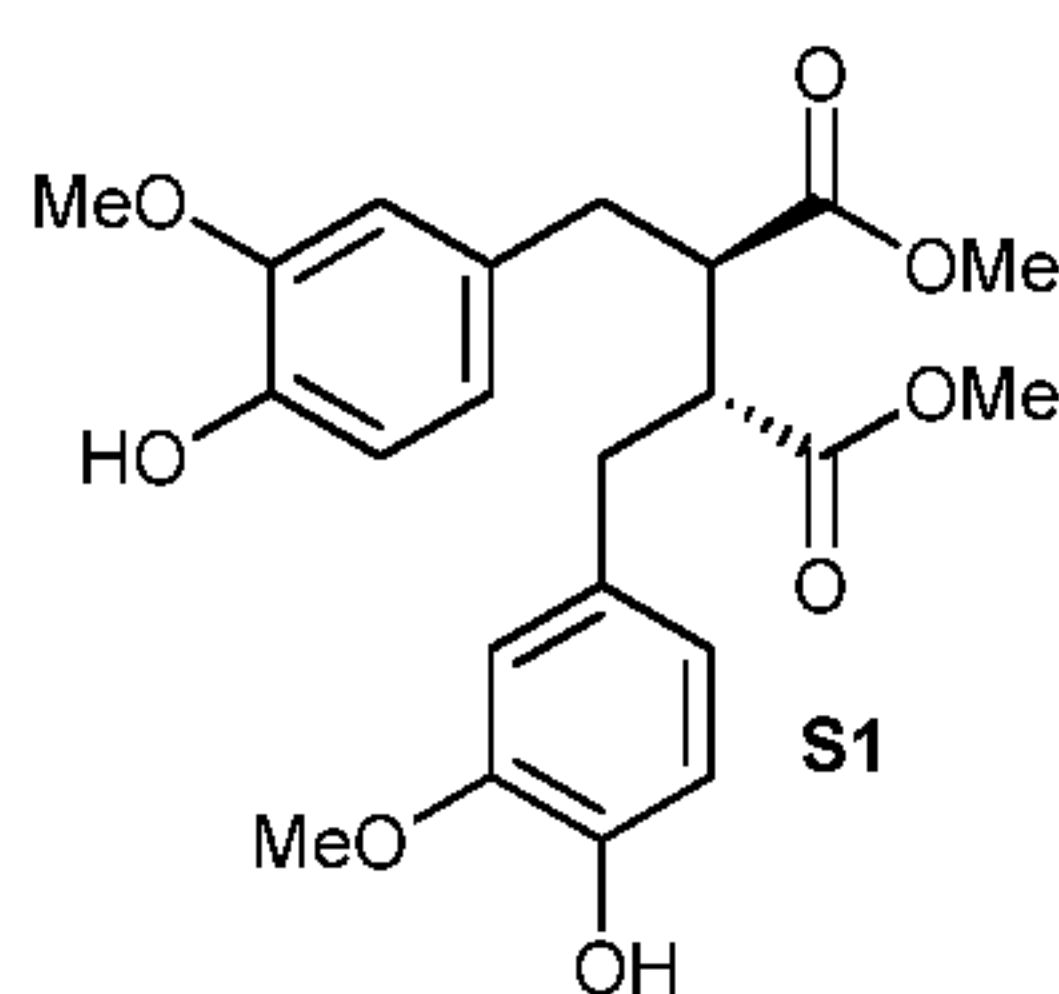
[0064] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

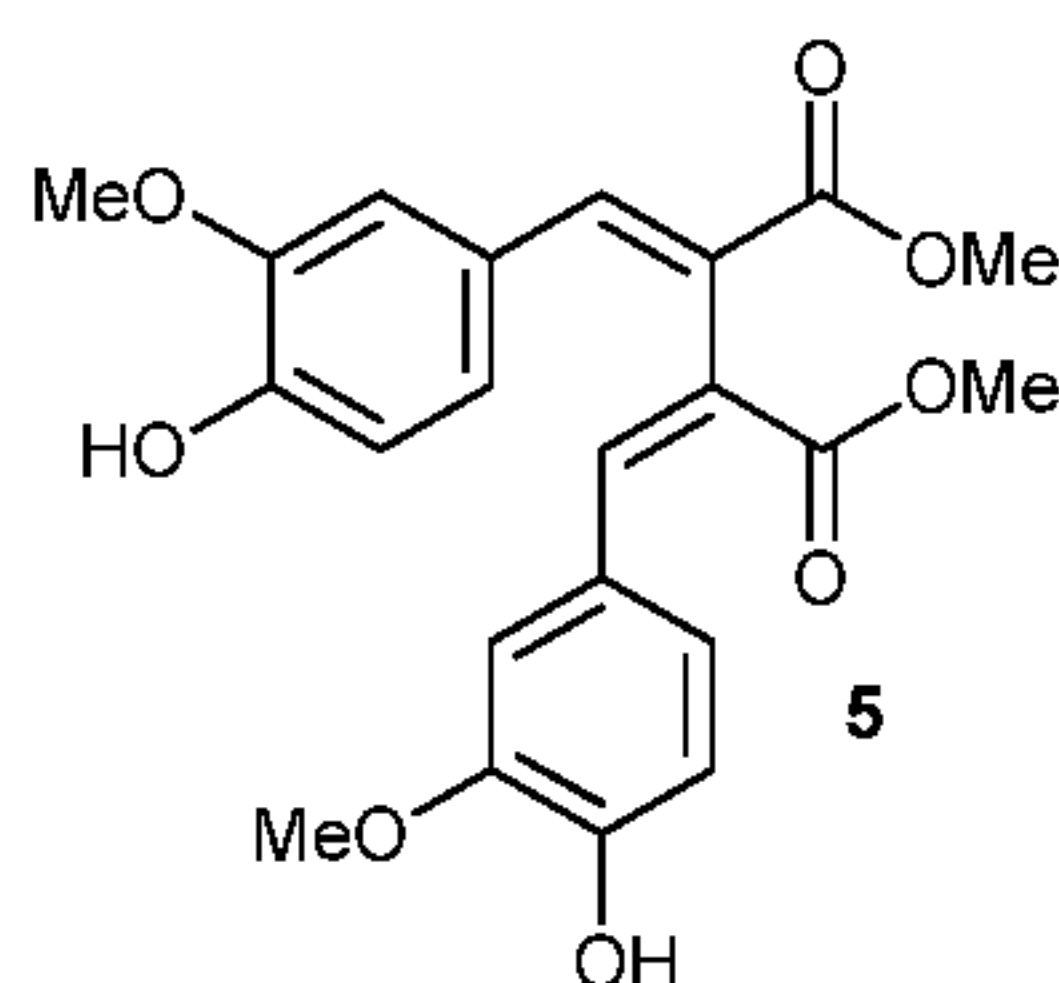
[0065] In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[0066] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



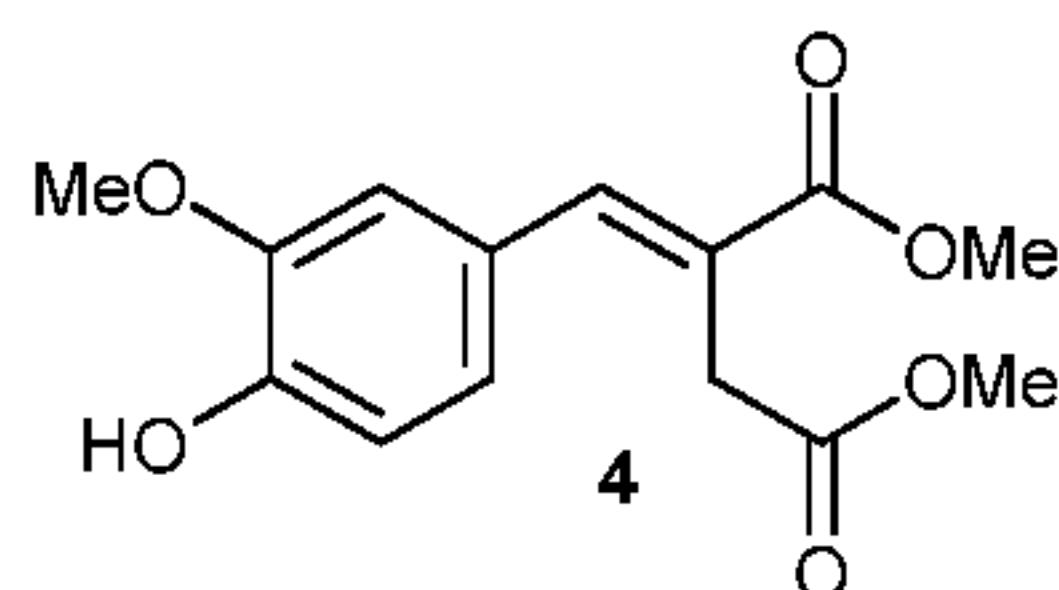
with a benzylating agent to prepare the compound of formula (S2). In some embodiments, said benzylating agent is BnBr and NaH.

[0067] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In some embodiments, said reducing agent is H₂ and Pd/C.

[0068] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

[0069] In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium (e.g., lithium wires).

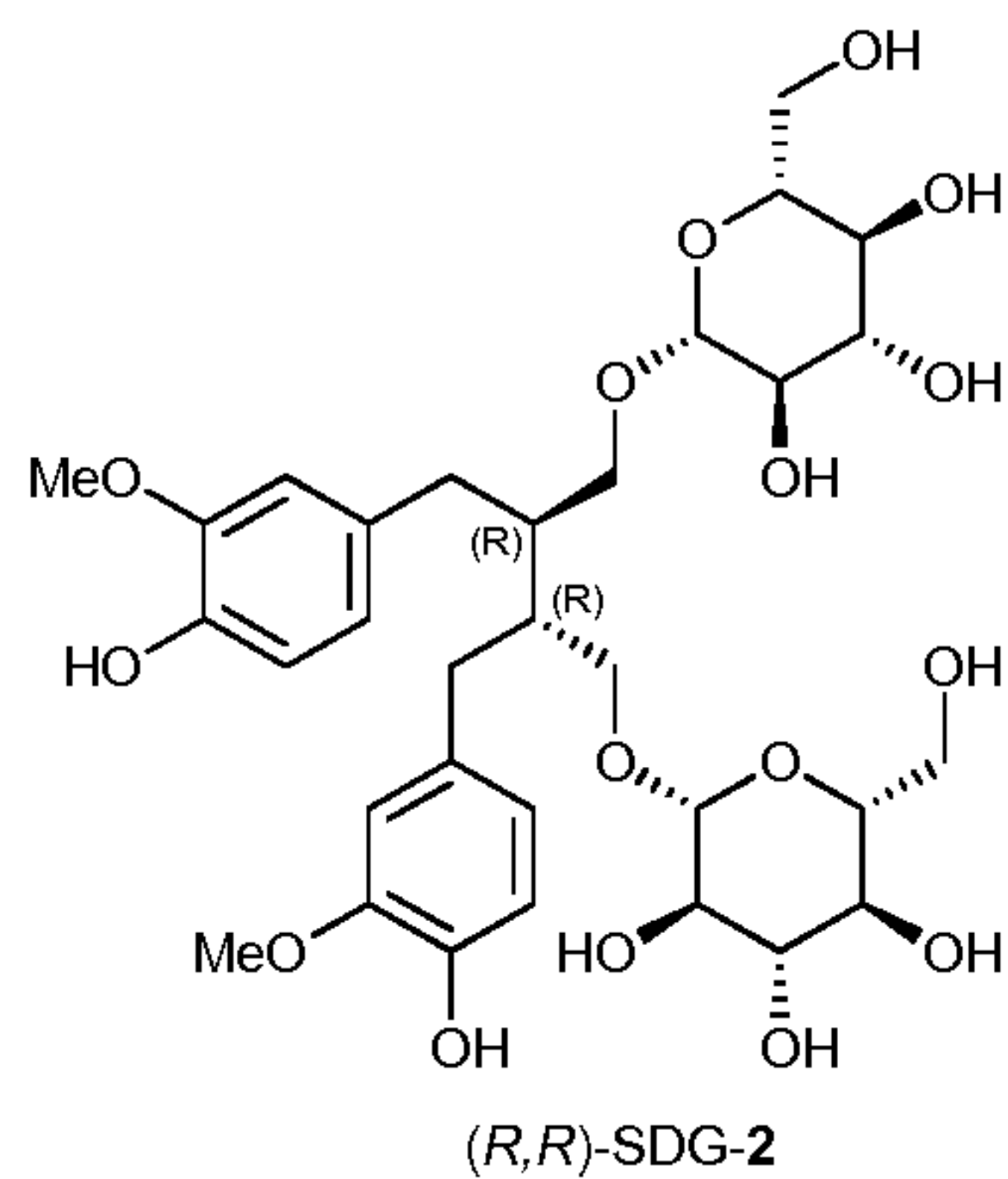
[0070] In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0071] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

[0072] In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

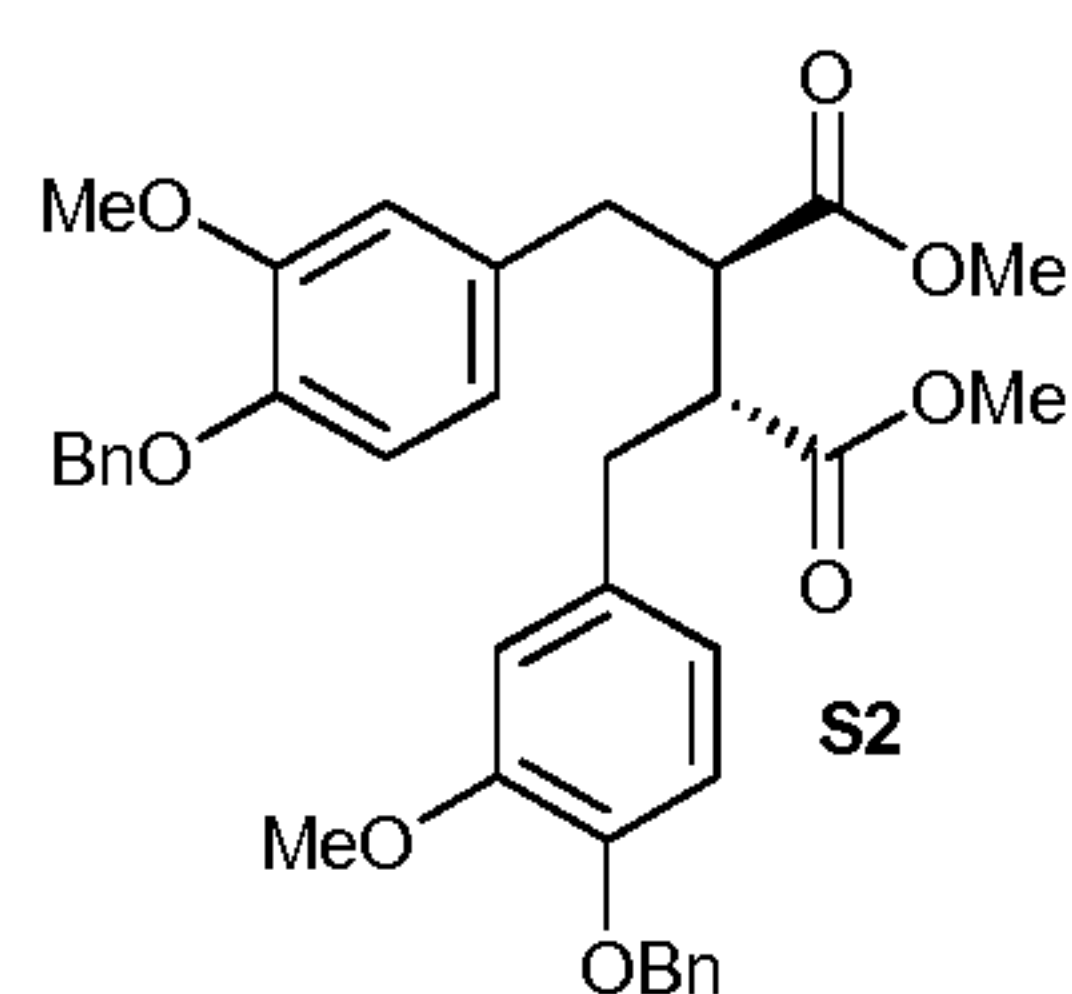
[0073] In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0074] The present invention provides a process for preparing a compound of formula ((*R,R*)-SDG-2)

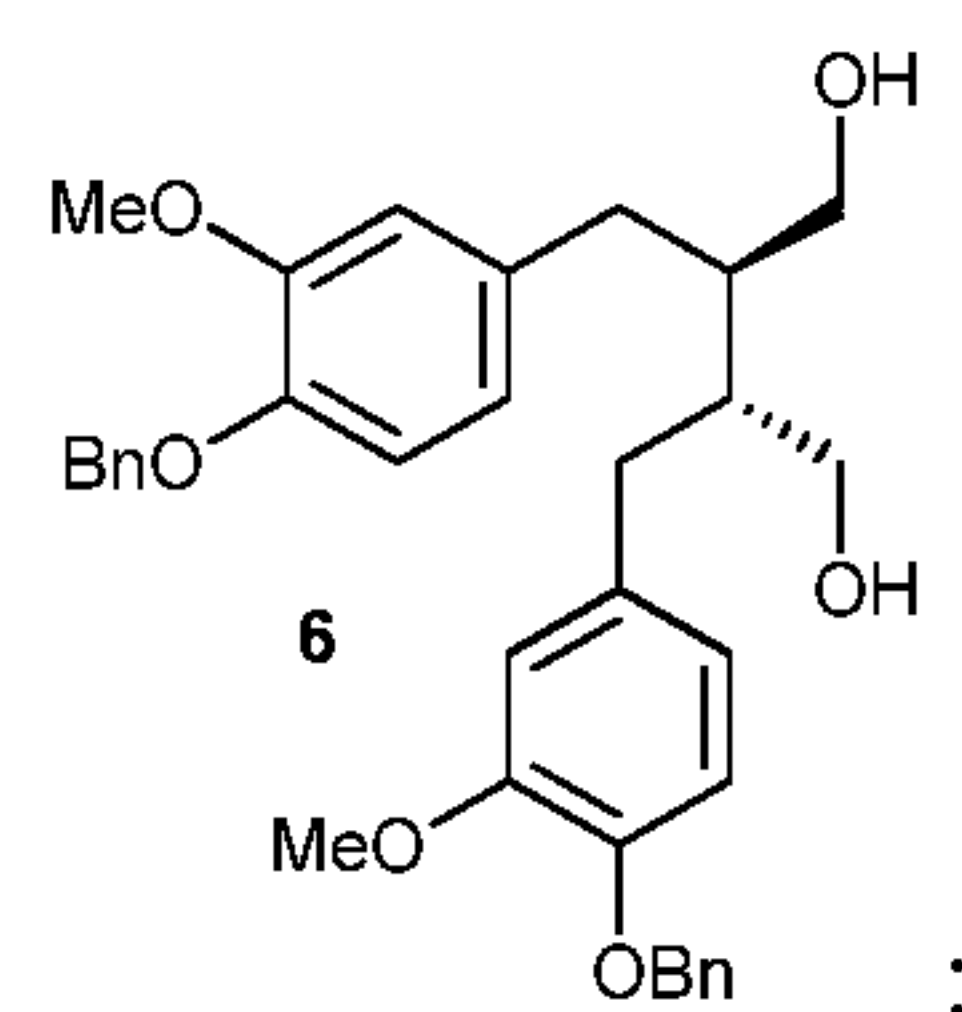


the process comprising:

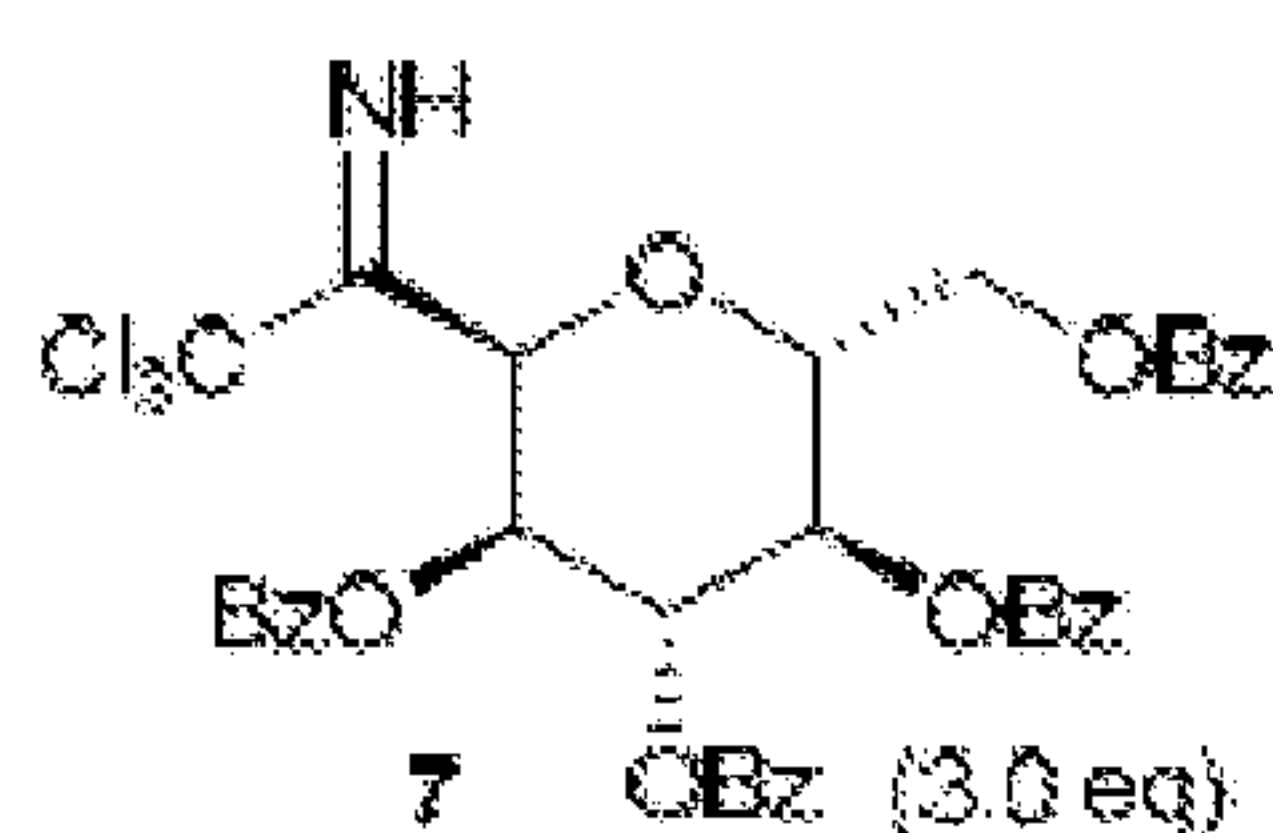
(a) reacting a compound of formula (S2)



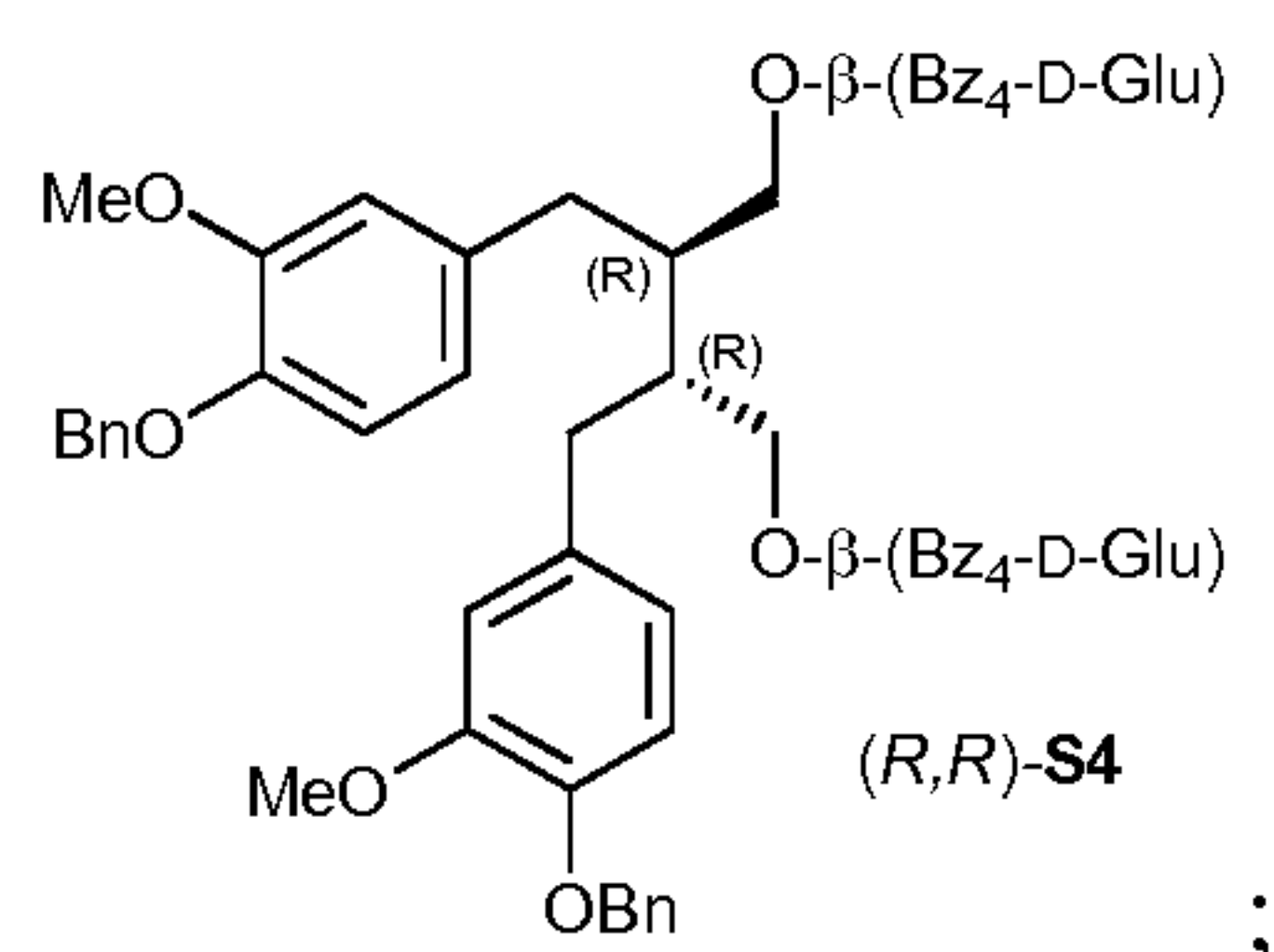
with a reducing agent to prepare a compound of formula (6)



5 (b) reacting the compound of formula (6) with a compound of formula (7)

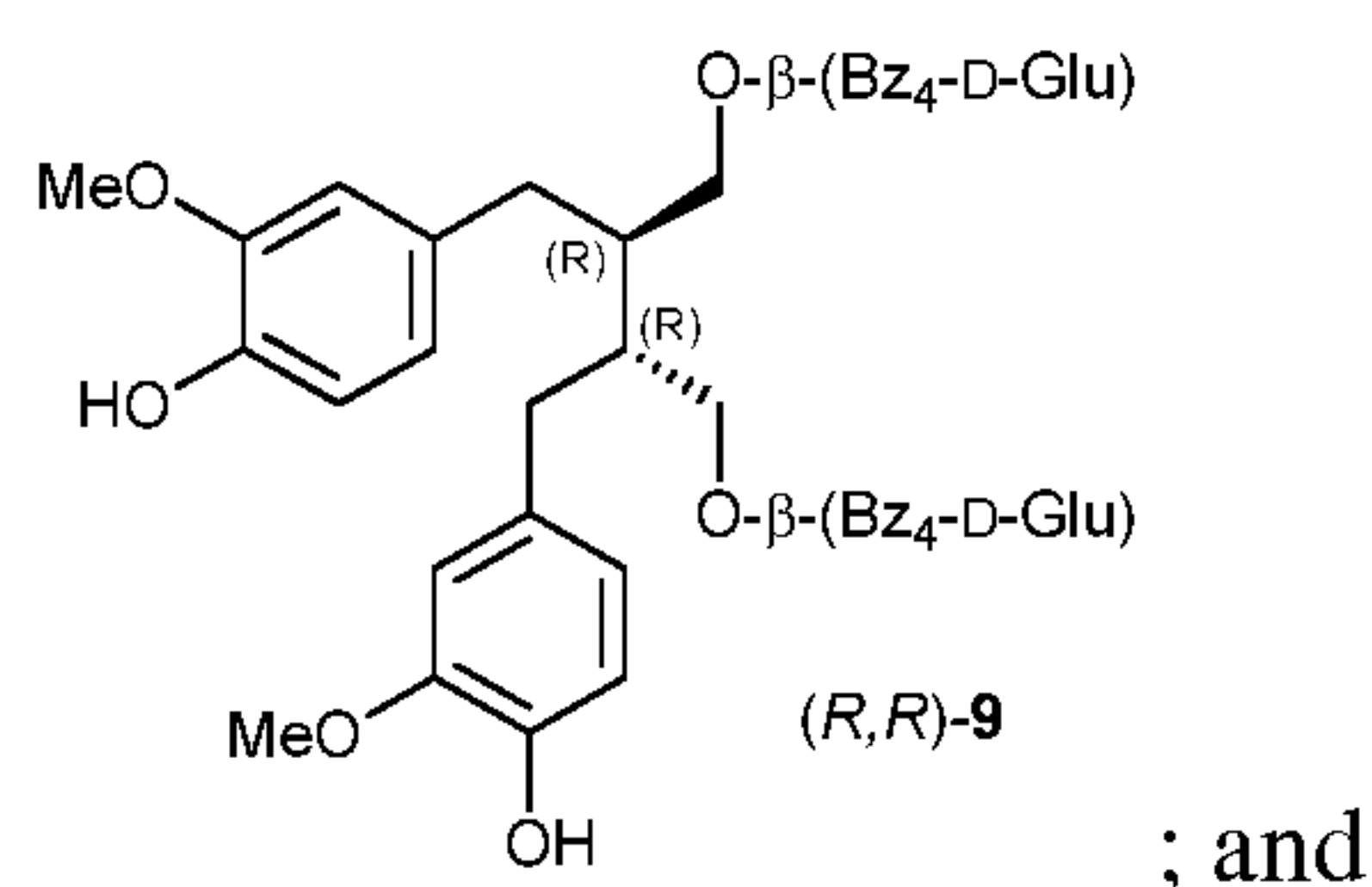


to prepare a compound of formula ((R,R)-S4)



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(c) cleaving benzyl ethers of the compound of formula ((R,R)-S4), followed by a separation procedure to prepare a compound of formula ((R,R)-9)



; and

(d) deprotecting the compound of formula ((R,R)-9) to prepare the compound of formula ((R,R)-SDG-2).

15

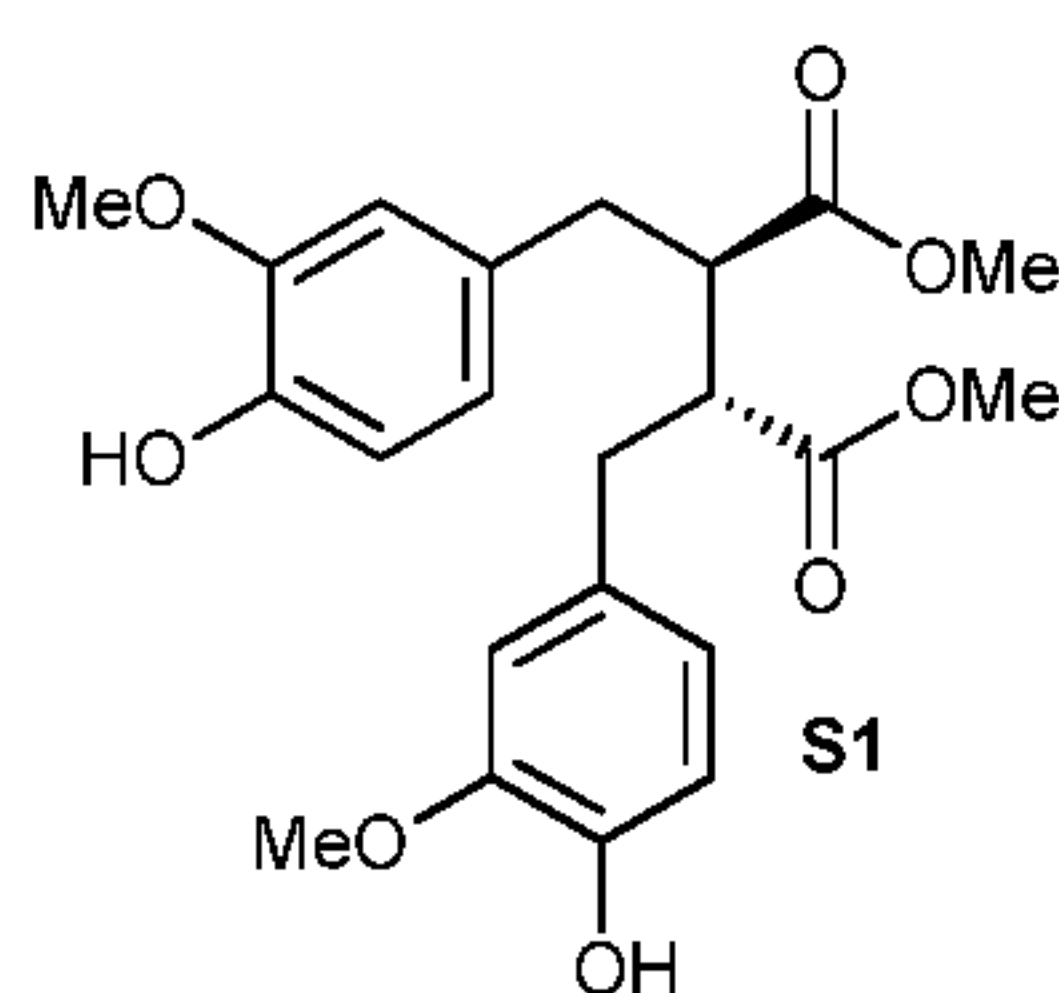
[0075] In some embodiments, said reacting of step (a) is carried out in the presence of TMSOTf. In some embodiments, said reacting of step (a) is carried out in the presence of activated molecular sieves.

[0076] In some embodiments, said cleaving is carried out in the presence H₂ and Pd/C in MeOH. In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.

[0077] In some embodiments, said deprotecting is carried out in a solution of NaOMe and MeOH.

[0078] In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

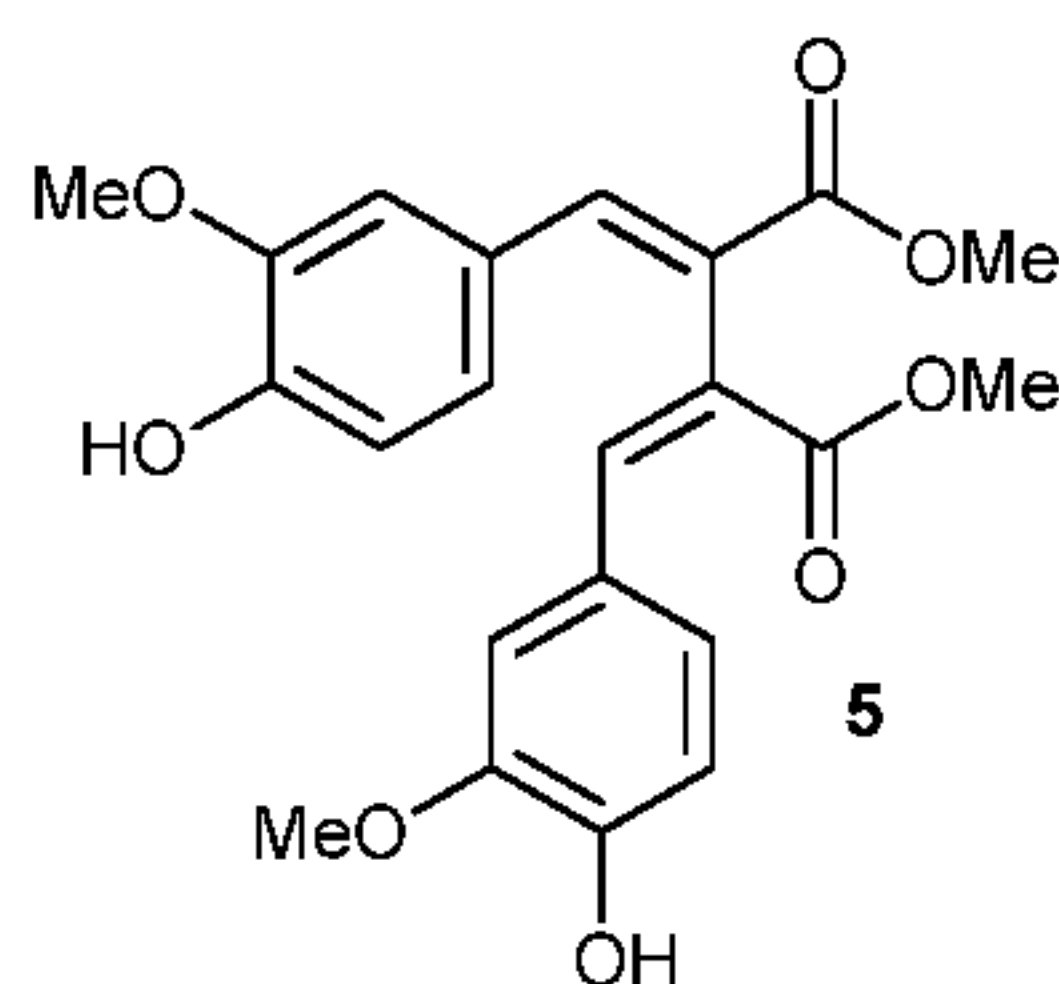
[0079] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to form the compound of formula (S2).

[0080] In some embodiments, said benzylating agent is BnBr and NaH.

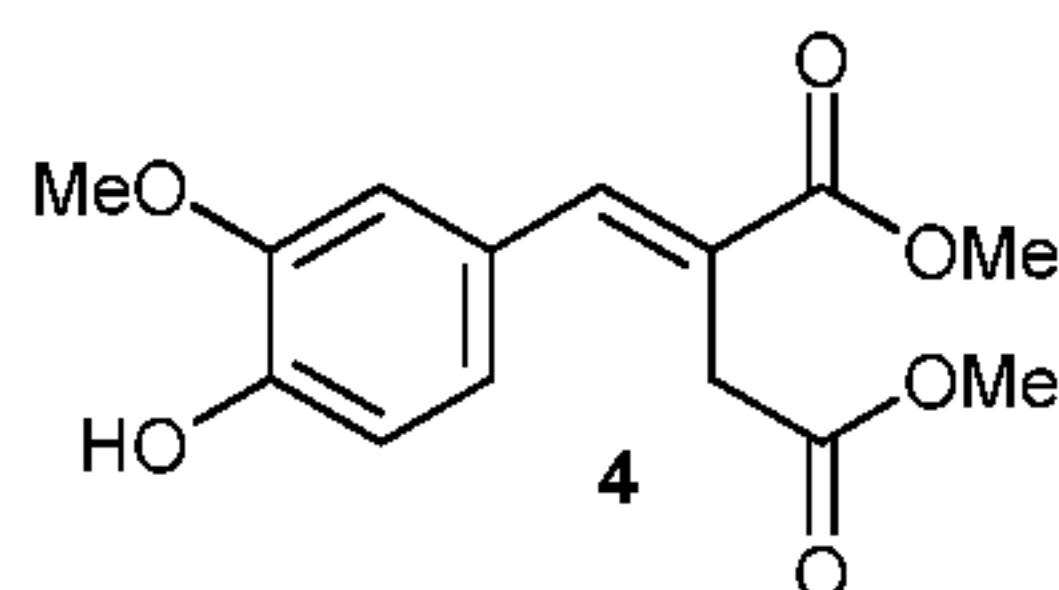
[0081] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

[0082] In some embodiments, said reducing agent is H₂ and Pd/C.

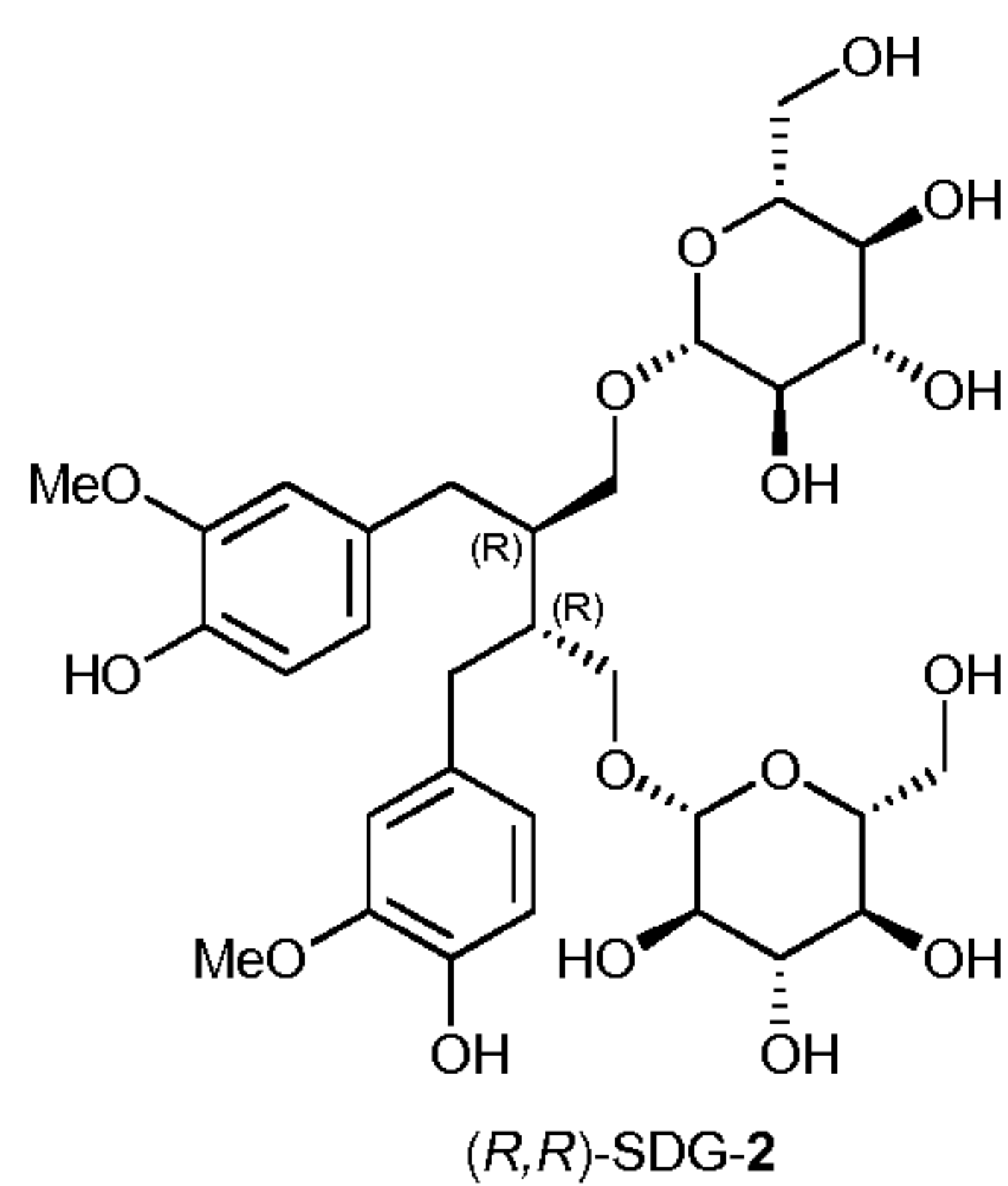
[0083] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

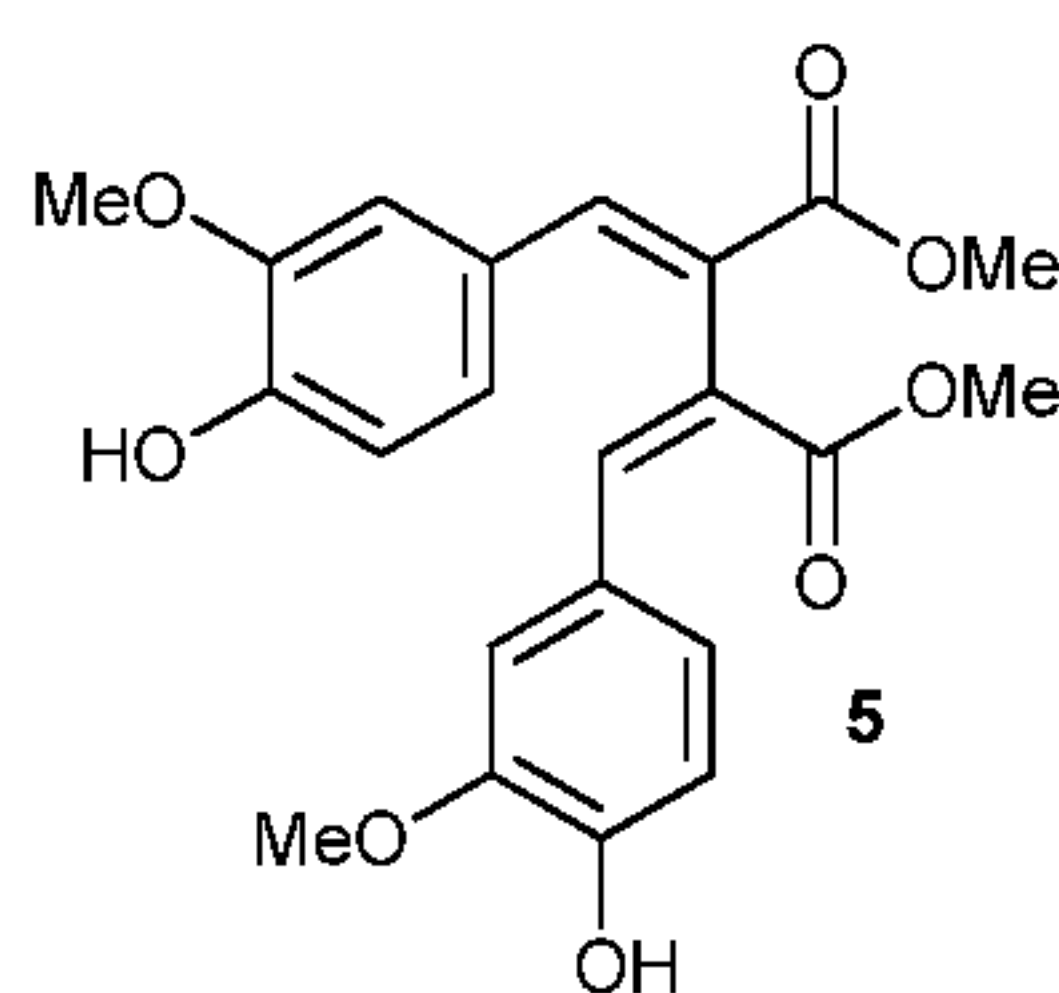
[0084] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0085] The present invention provides a process for preparing a compound of formula ((*R,R*)-SDG-2)

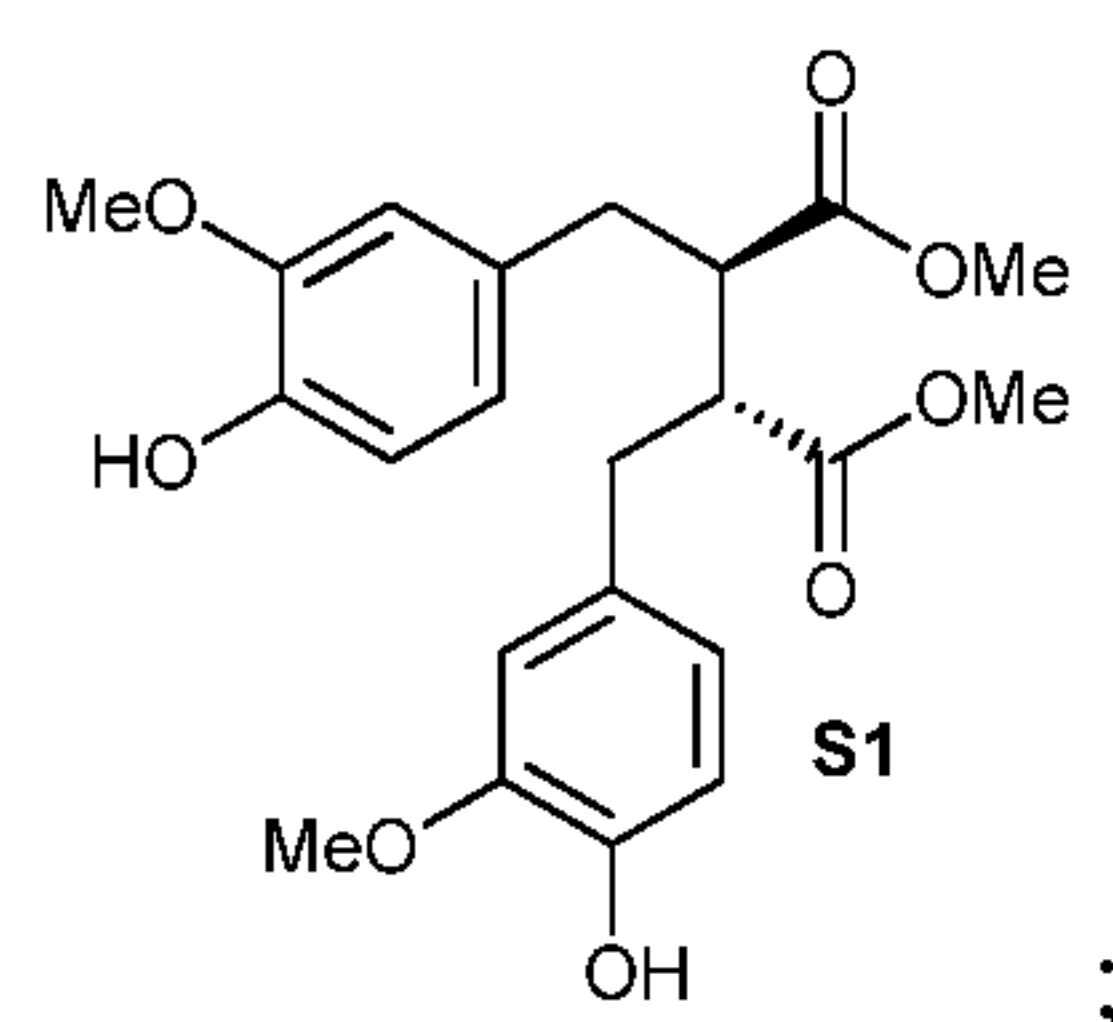


the process comprising:

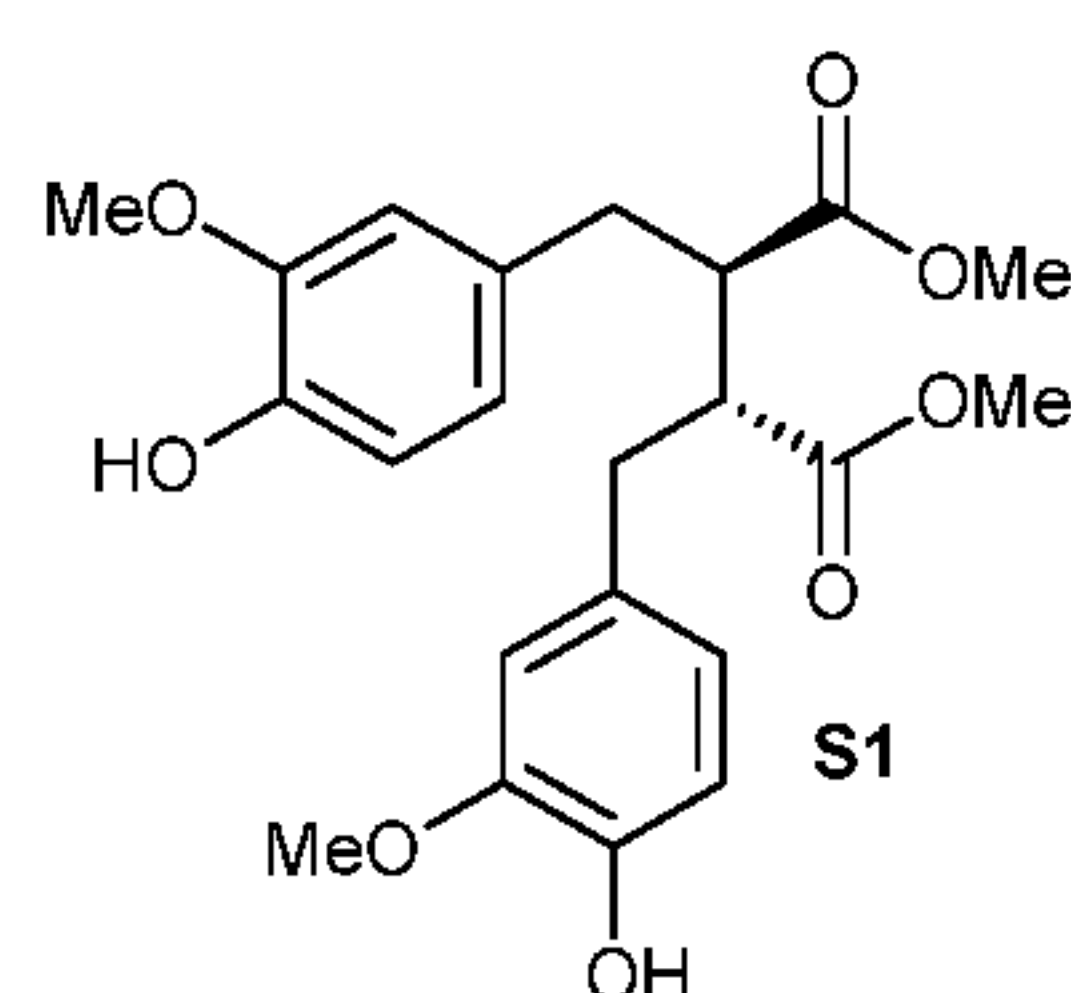
- (a) reacting a compound of formula (5)



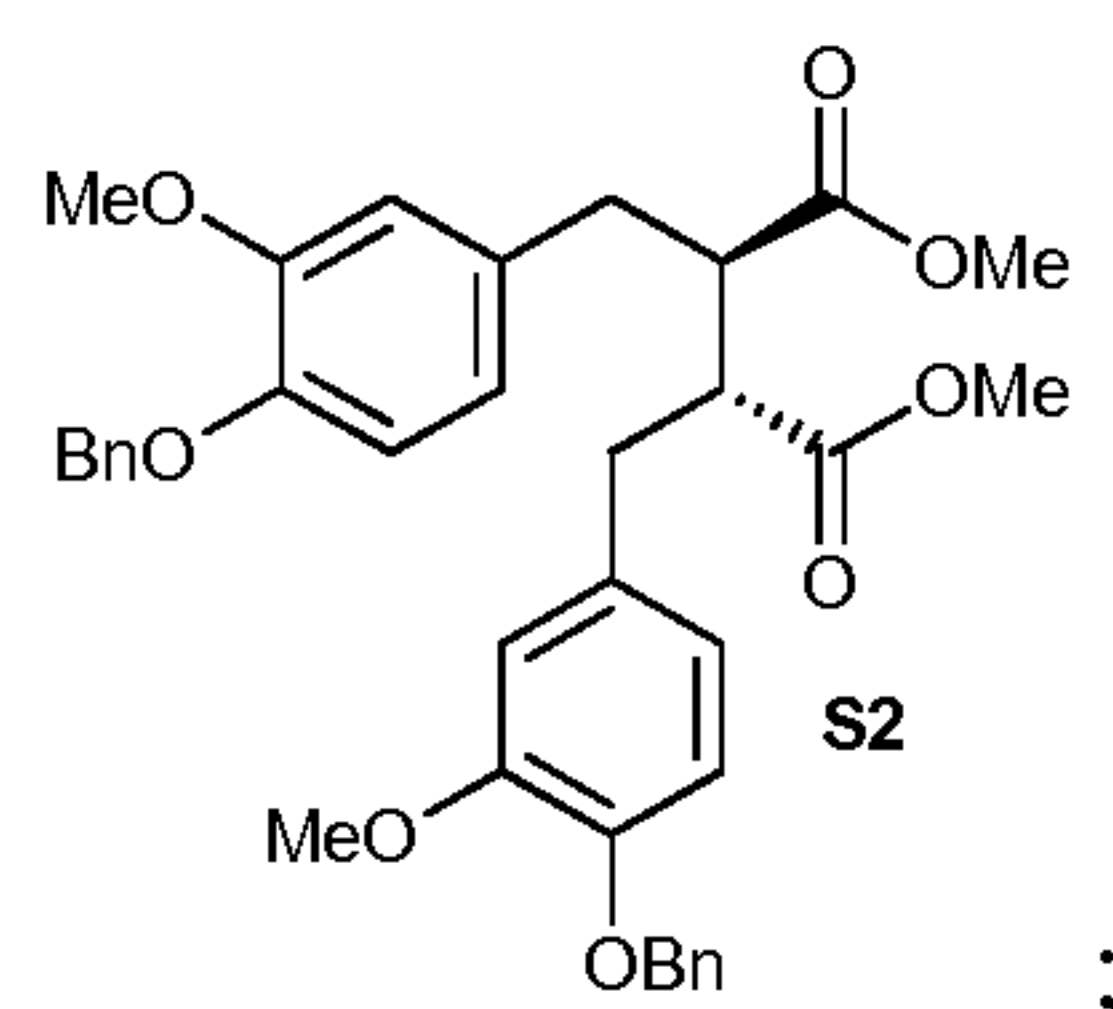
with a reducing agent to prepare a compound of formula (S1)



(b) reacting the compound of formula (S1)

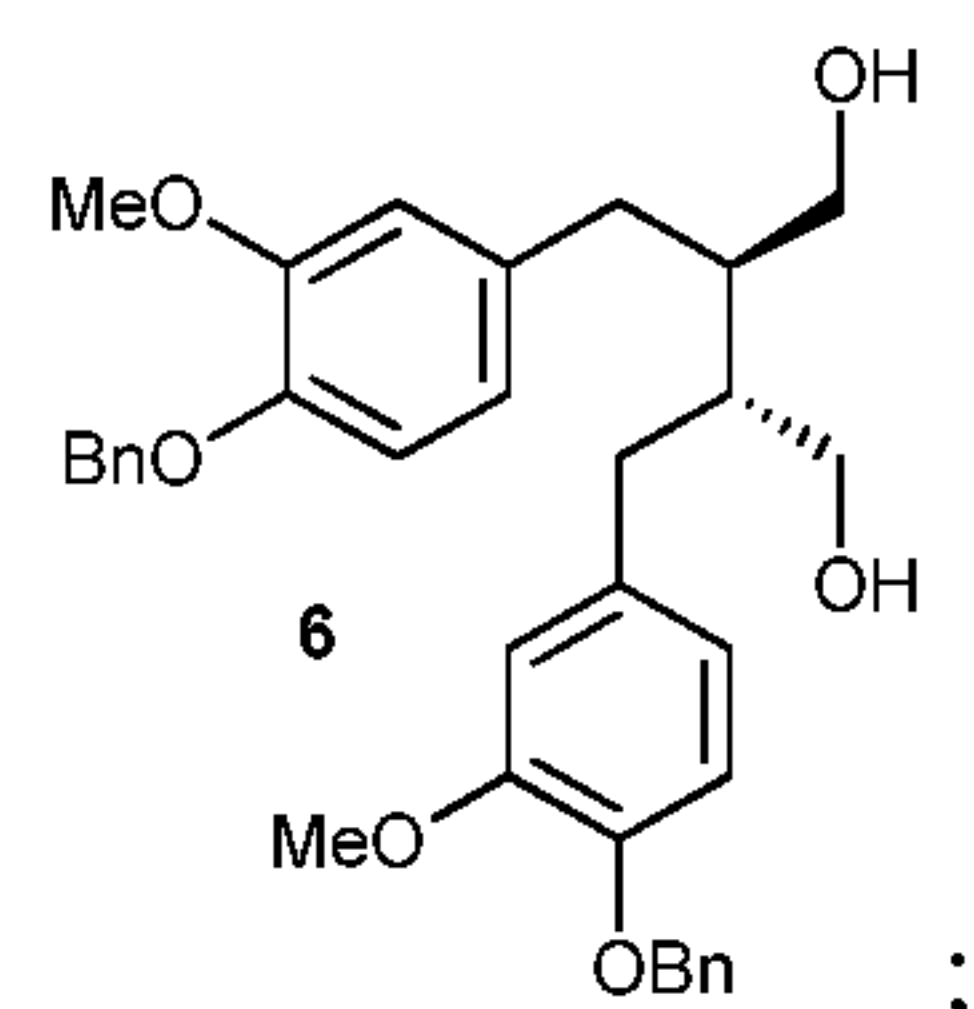


with a benzylating agent to prepare a compound of formula (S2)



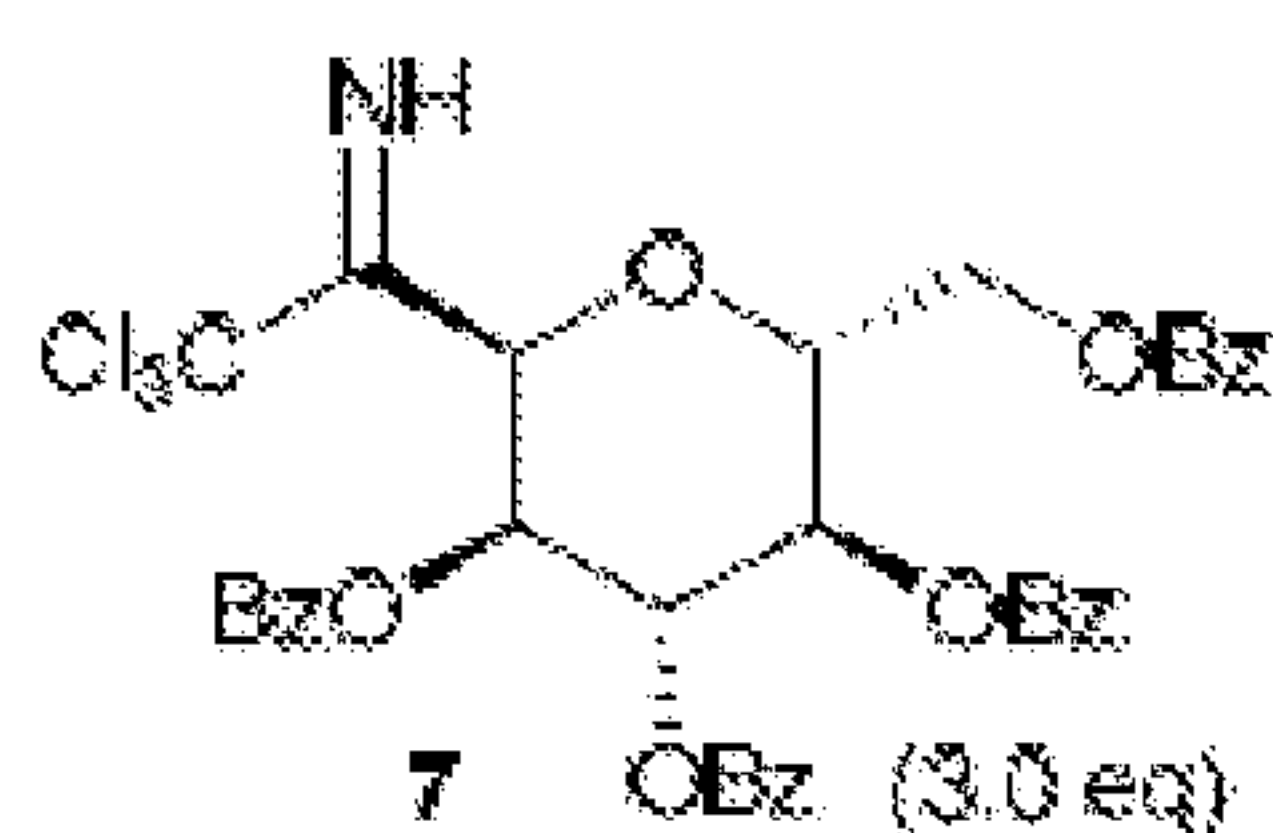
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(c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6)

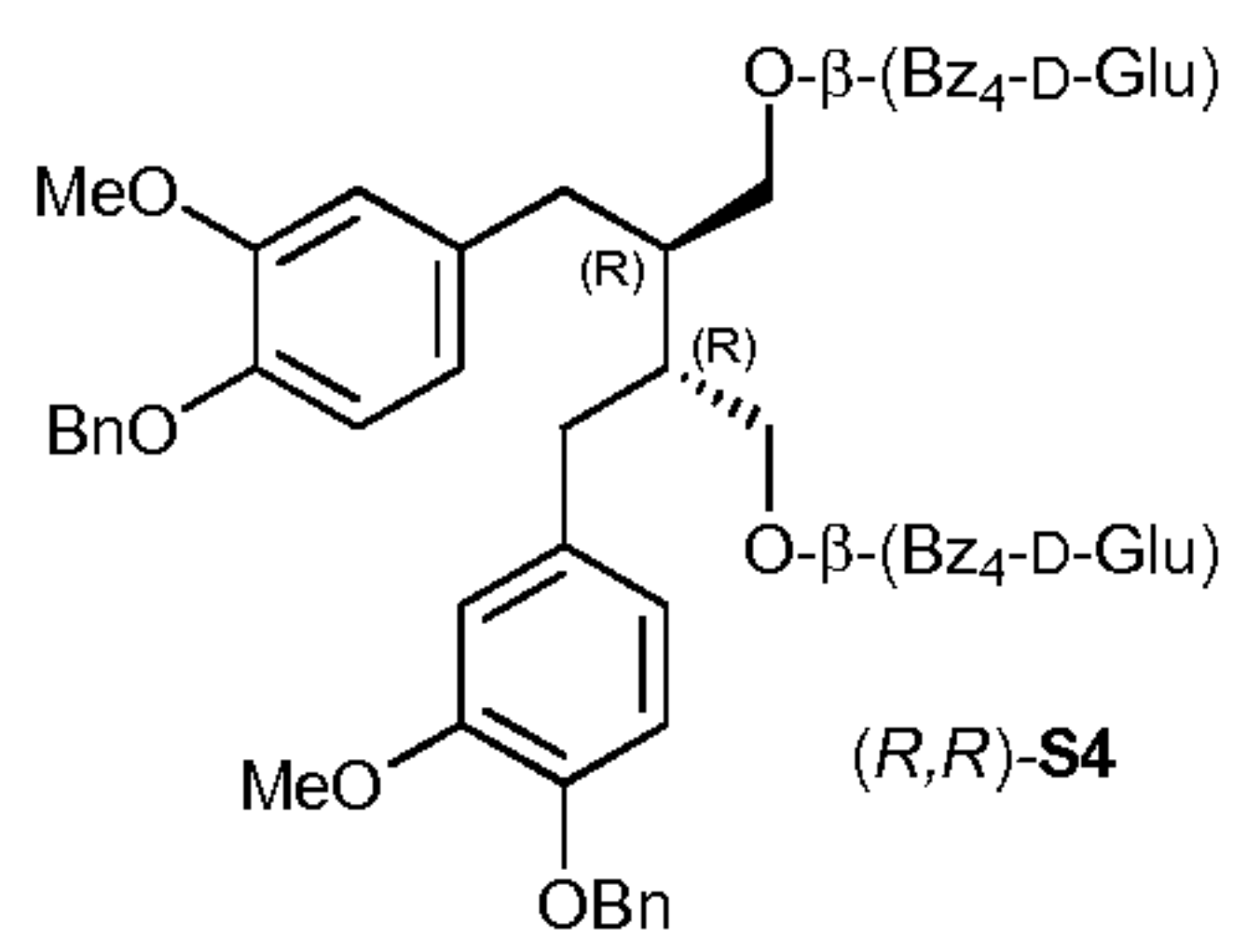


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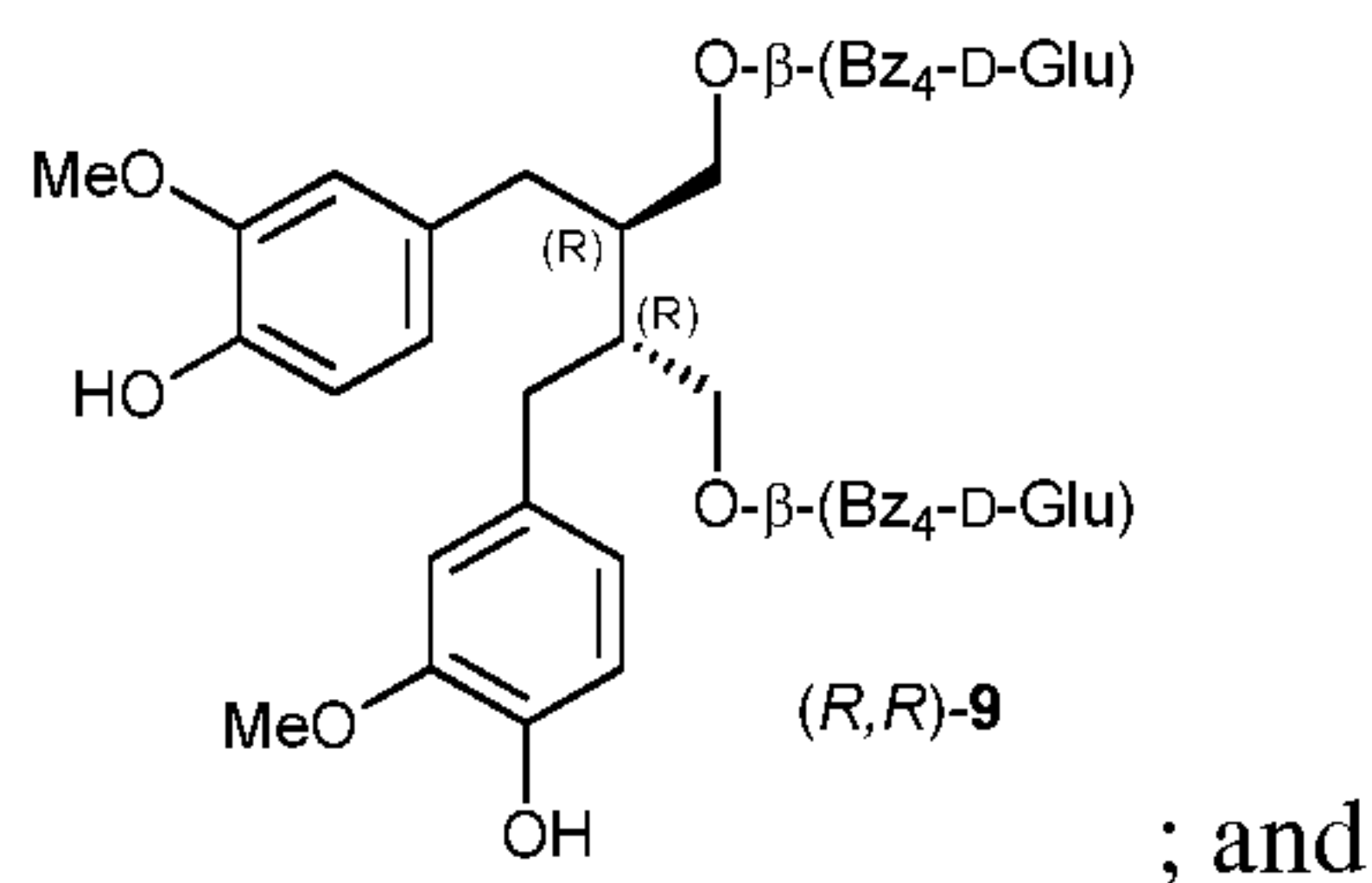
(d) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((R,R)-S4)

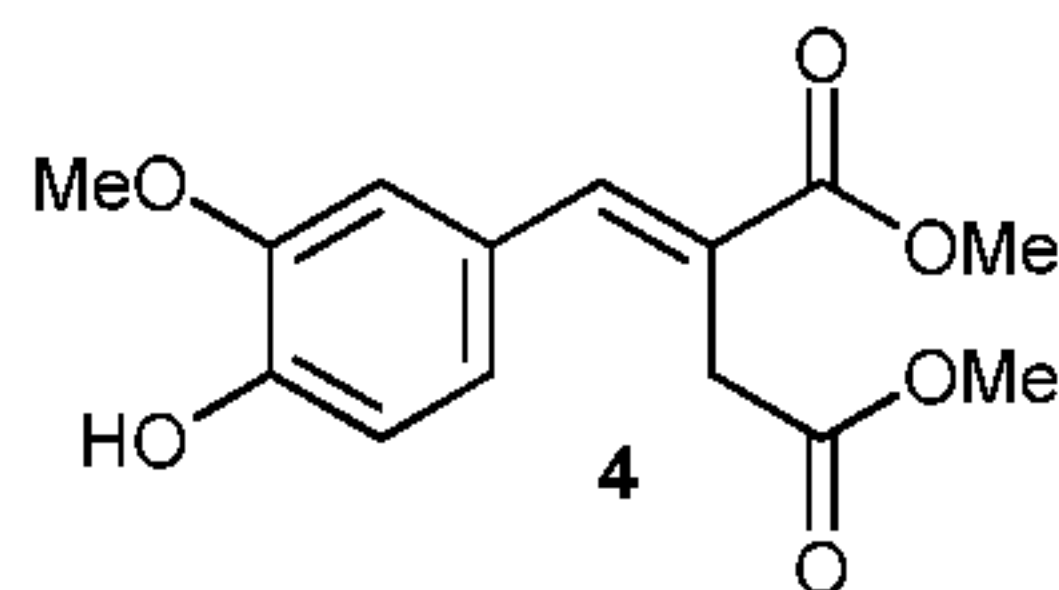


(e) cleaving benzyl ethers of the compound of formula ((R,R)-S3), followed by a separation procedure to prepare a compound of formula ((R,R)-9)



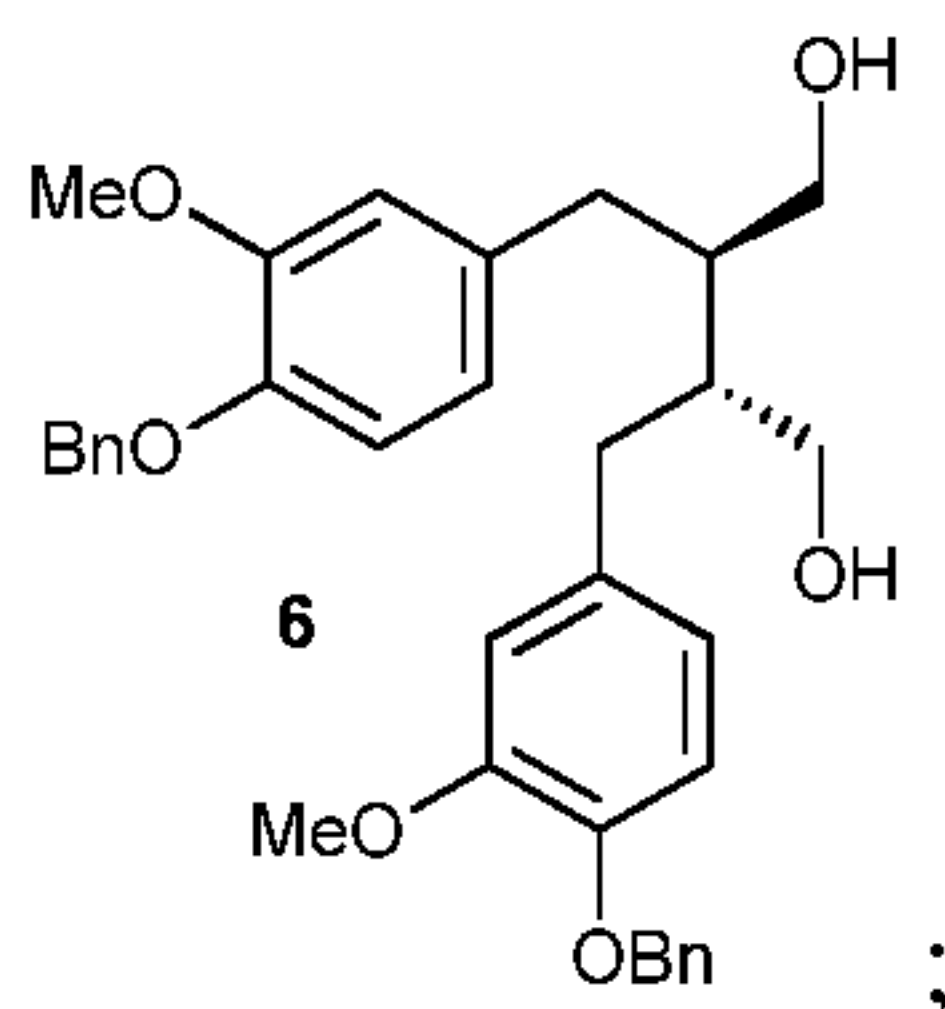
5 (f) deprotecting the compound of formula ((R,R)-9) to prepare the compound of formula ((R,R)-SDG-2).

[0086] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)

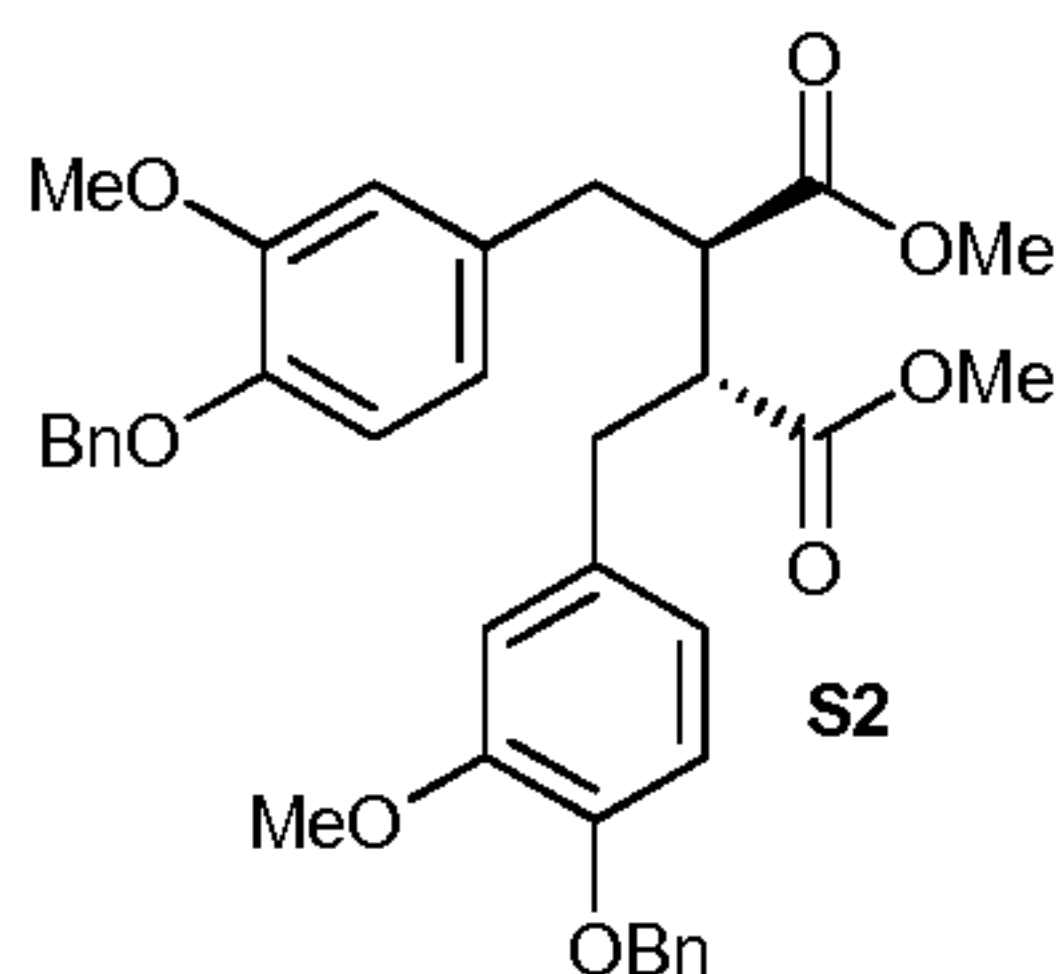


10 with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

[0087] The present invention provides a process for preparing a compound of formula (6)

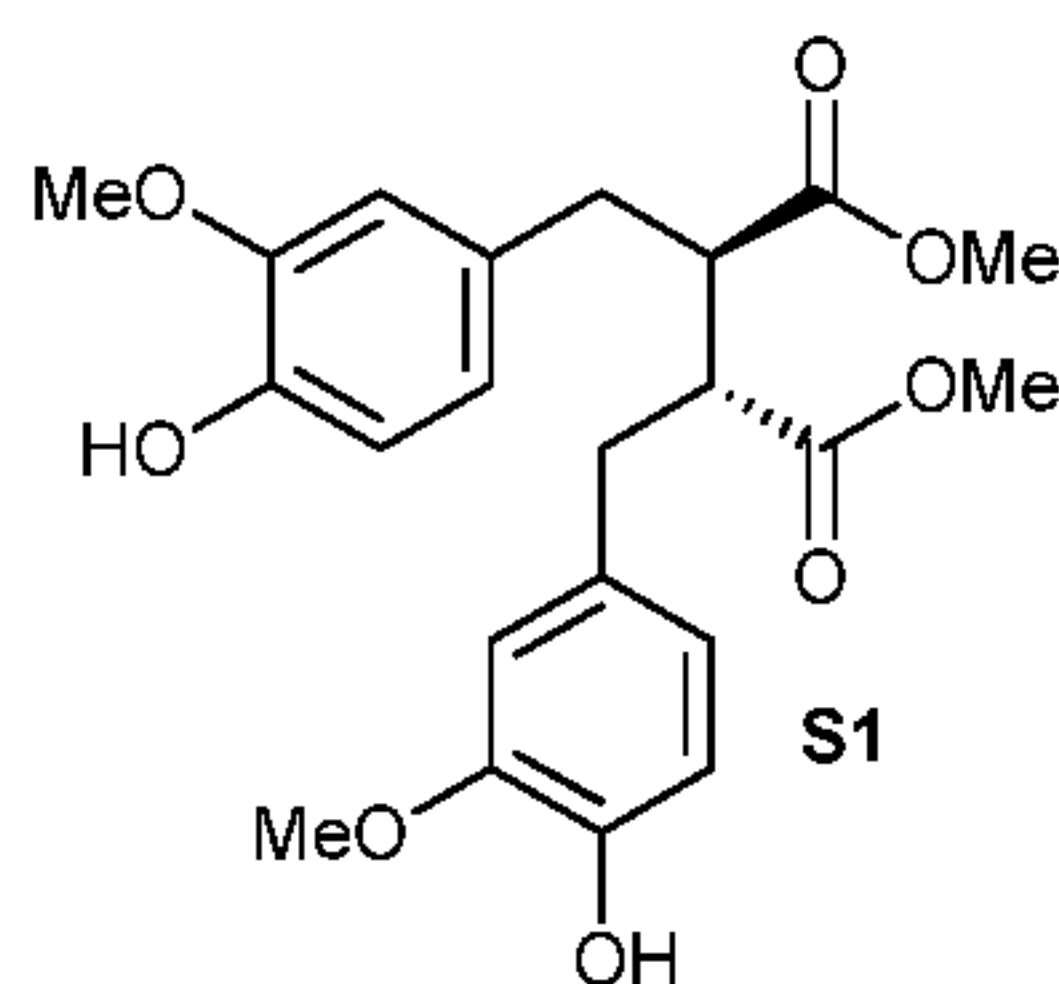


15 the process comprising reacting a compound of formula (S2)



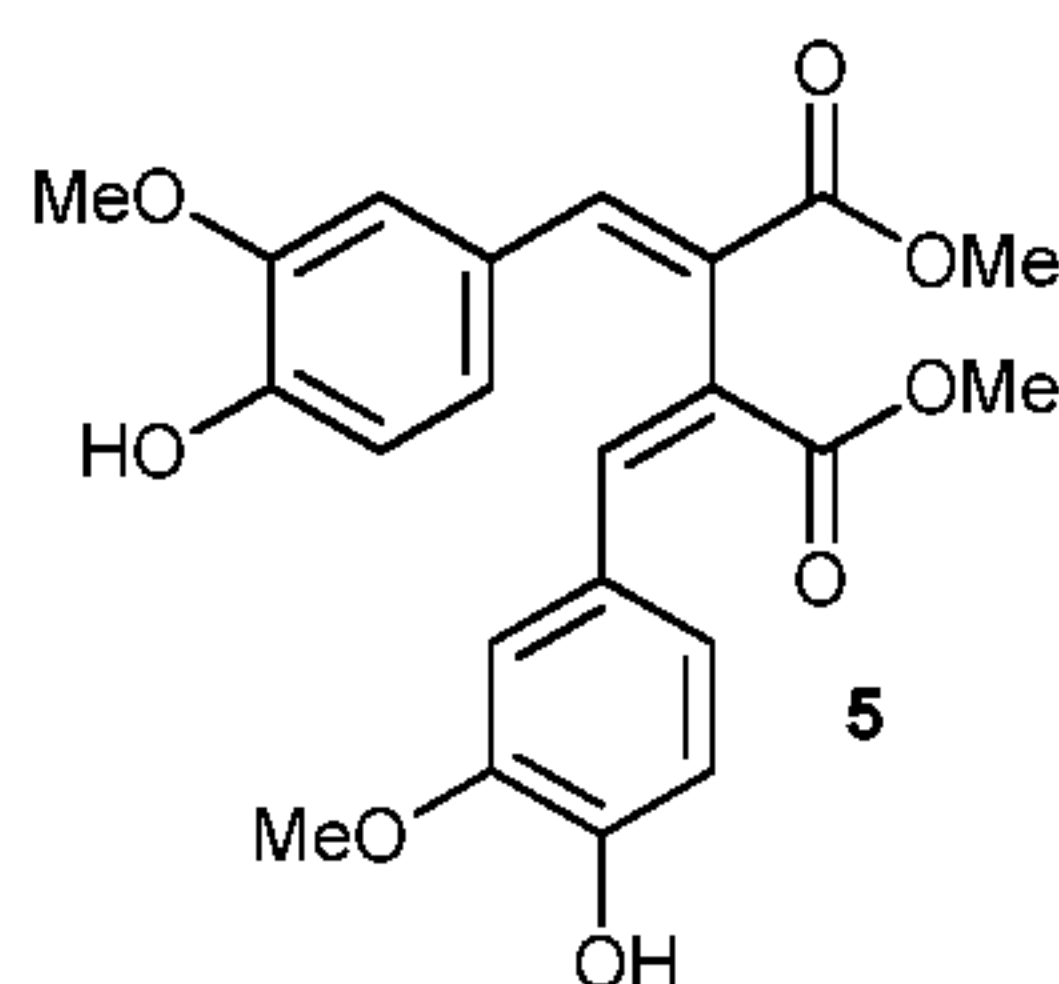
with a reducing agent to prepare the compound of formula (6). In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[0088] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



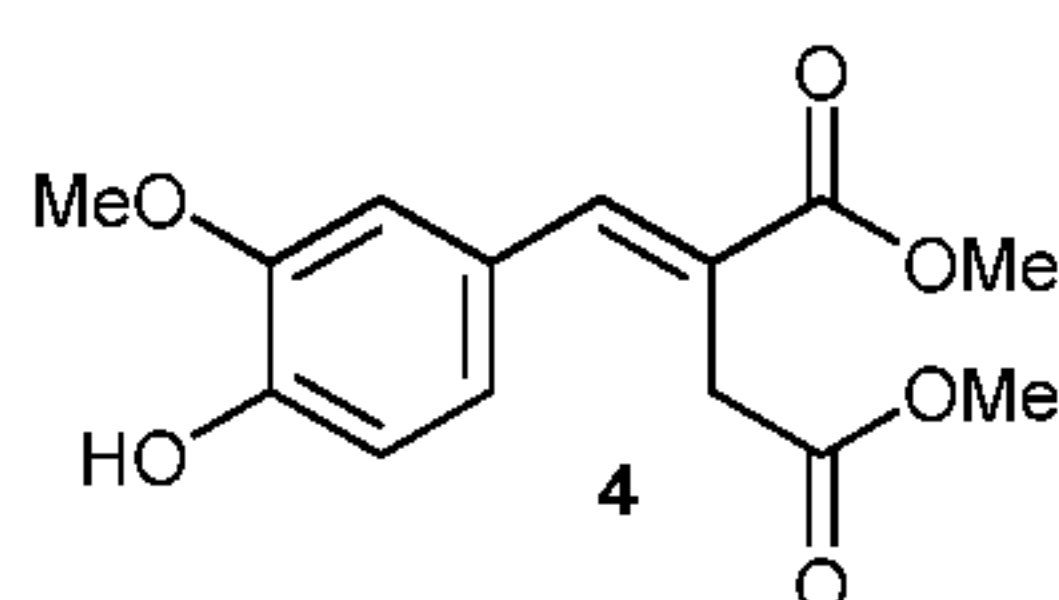
with a benzylating agent to prepare the compound of formula (S2). In certain embodiments, said benzylating agent is BnBr and NaH.

[0089] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In certain embodiments, said reducing agent is H₂ and Pd/C.

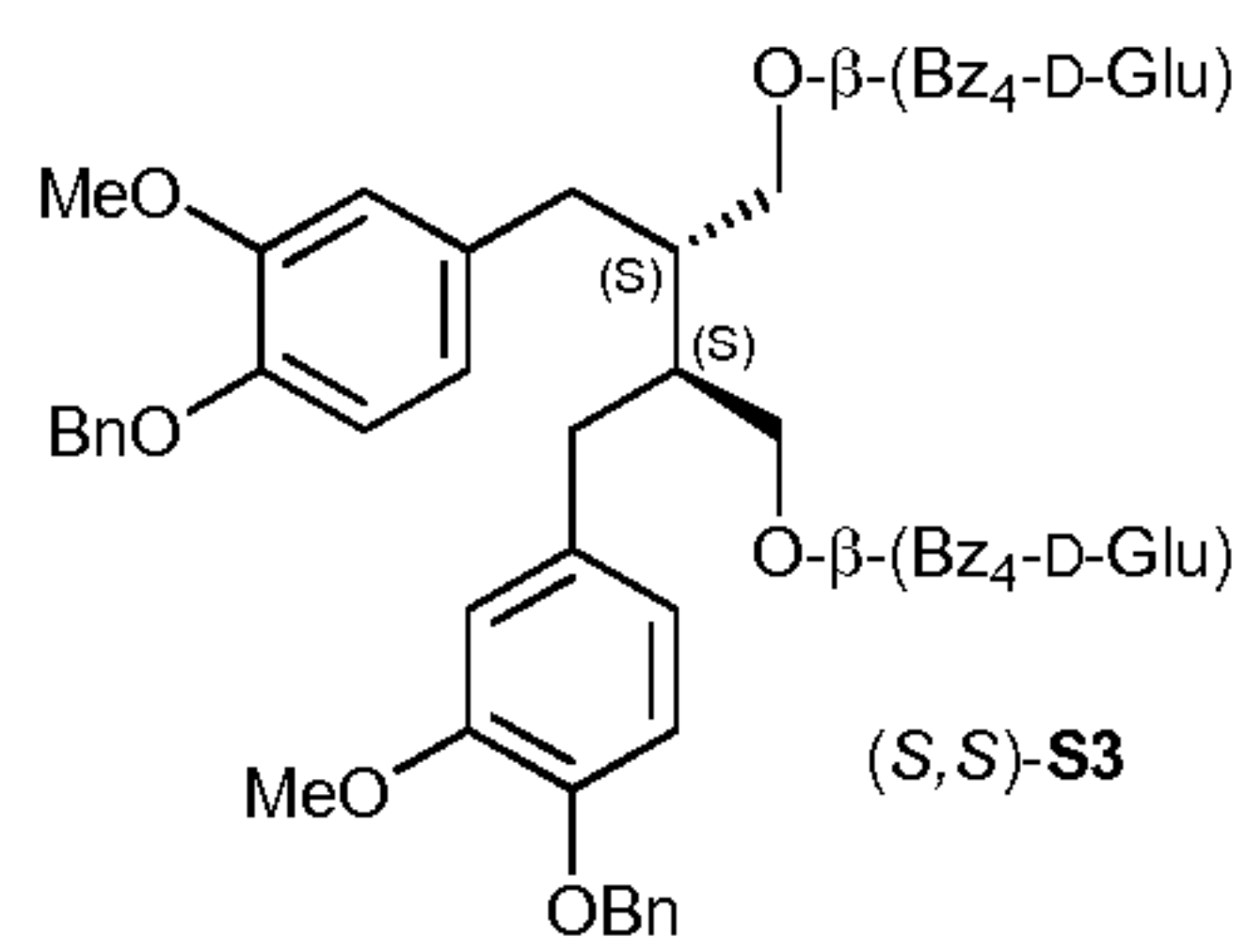
[0090] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



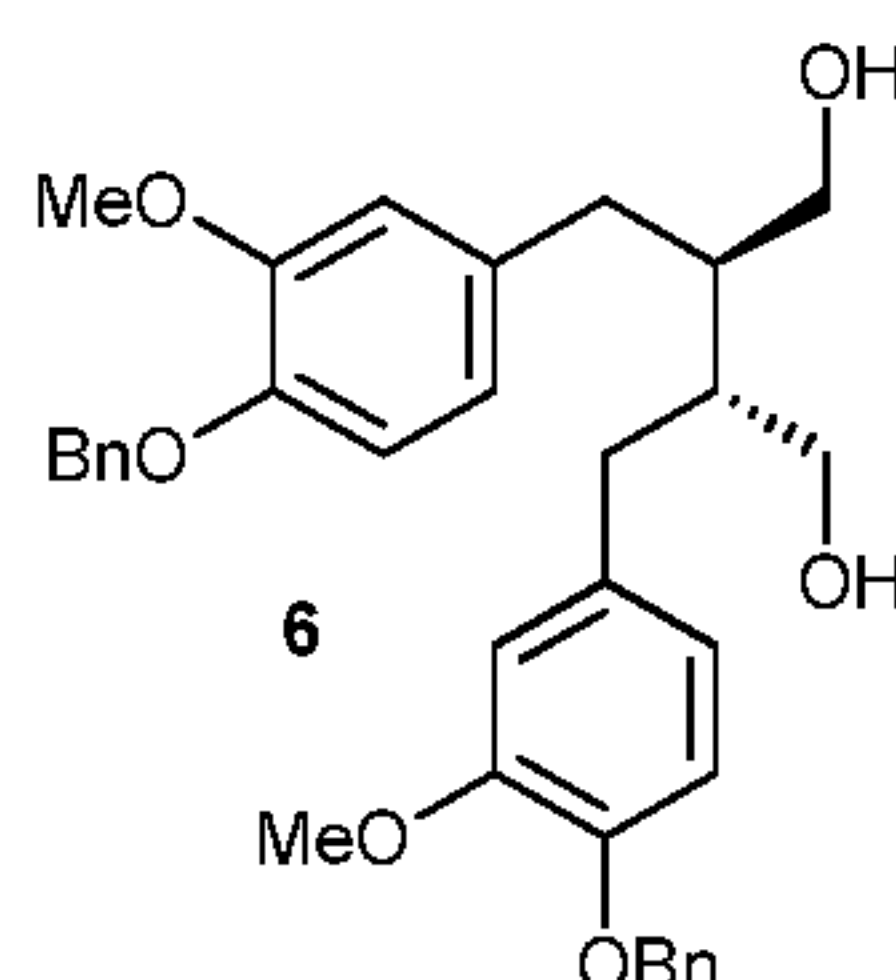
with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0091] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

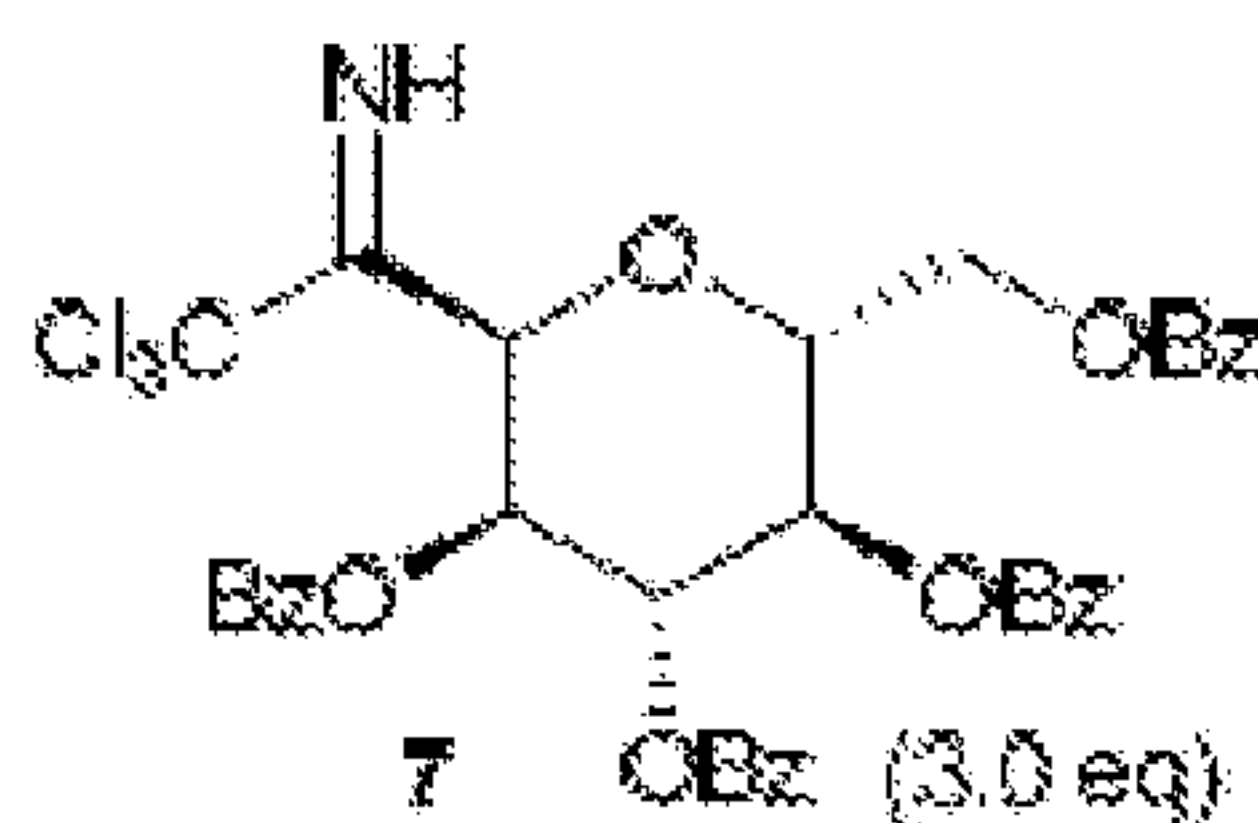
[0092] The present invention provides a process for preparing a compound of formula ((*S,S*)-S3)



5 the process comprising reacting a compound of formula (6)



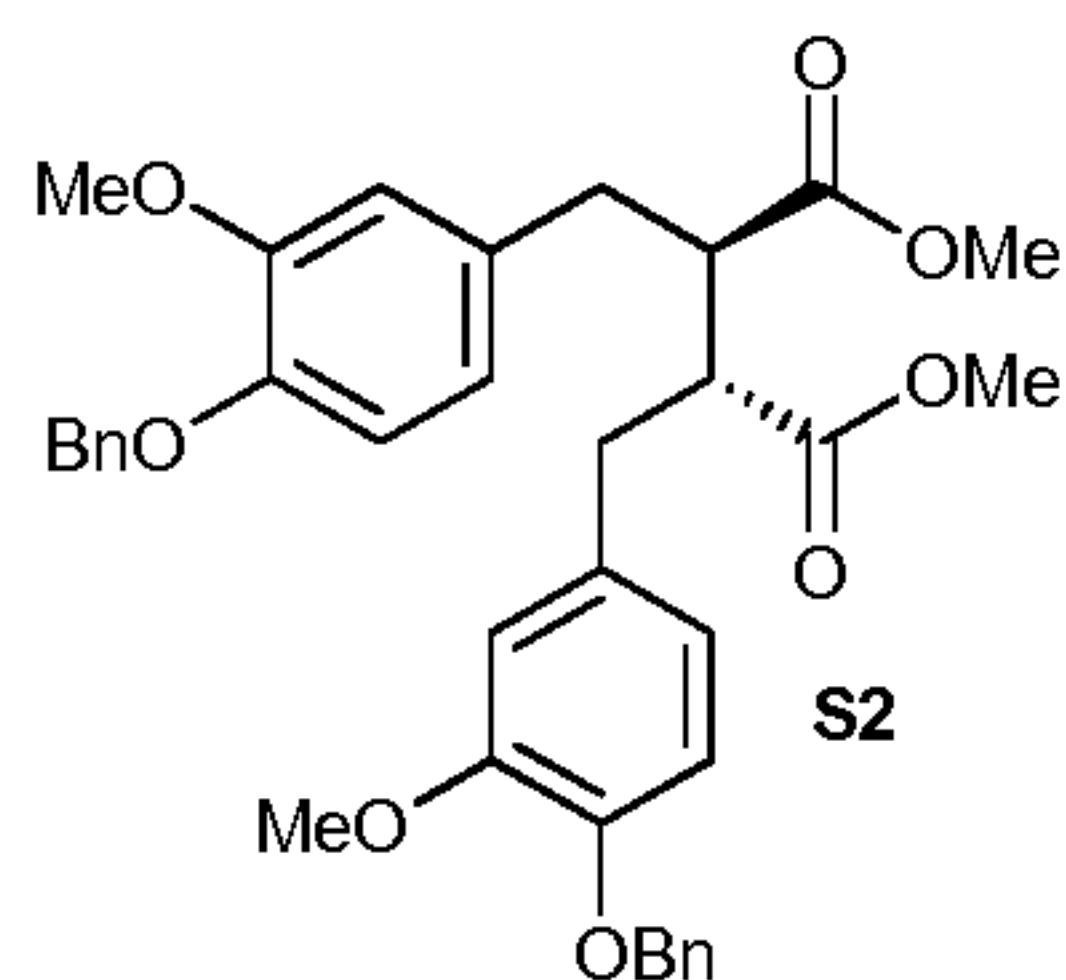
with a compound of formula (7)



to prepare the compound of formula ((*S,S*)-S3).

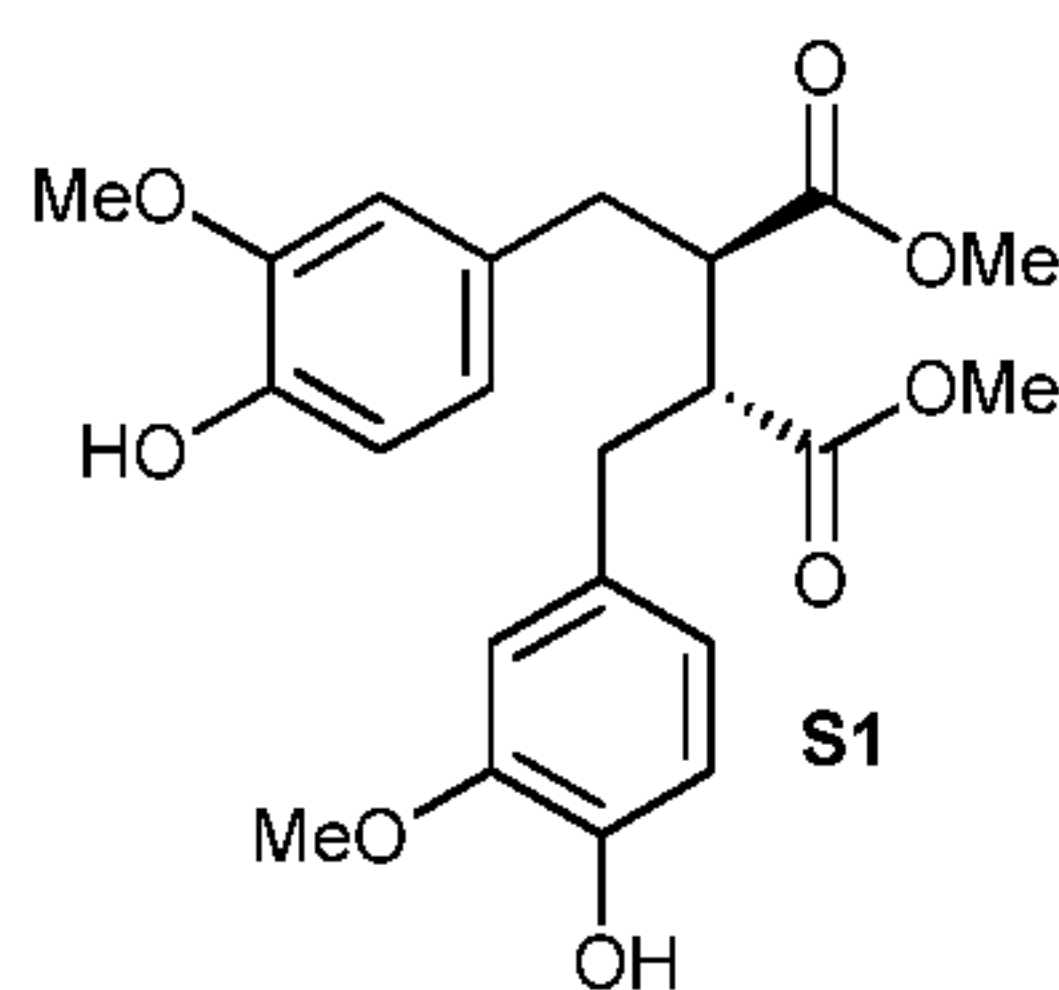
10 [0093] In some embodiments, said reacting is carried out in the presence of a Lewis acid. In some embodiments, said Lewis acid is TMSOTf. In some embodiments, said reacting is carried out in the presence of activated molecular sieves.

[0094] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



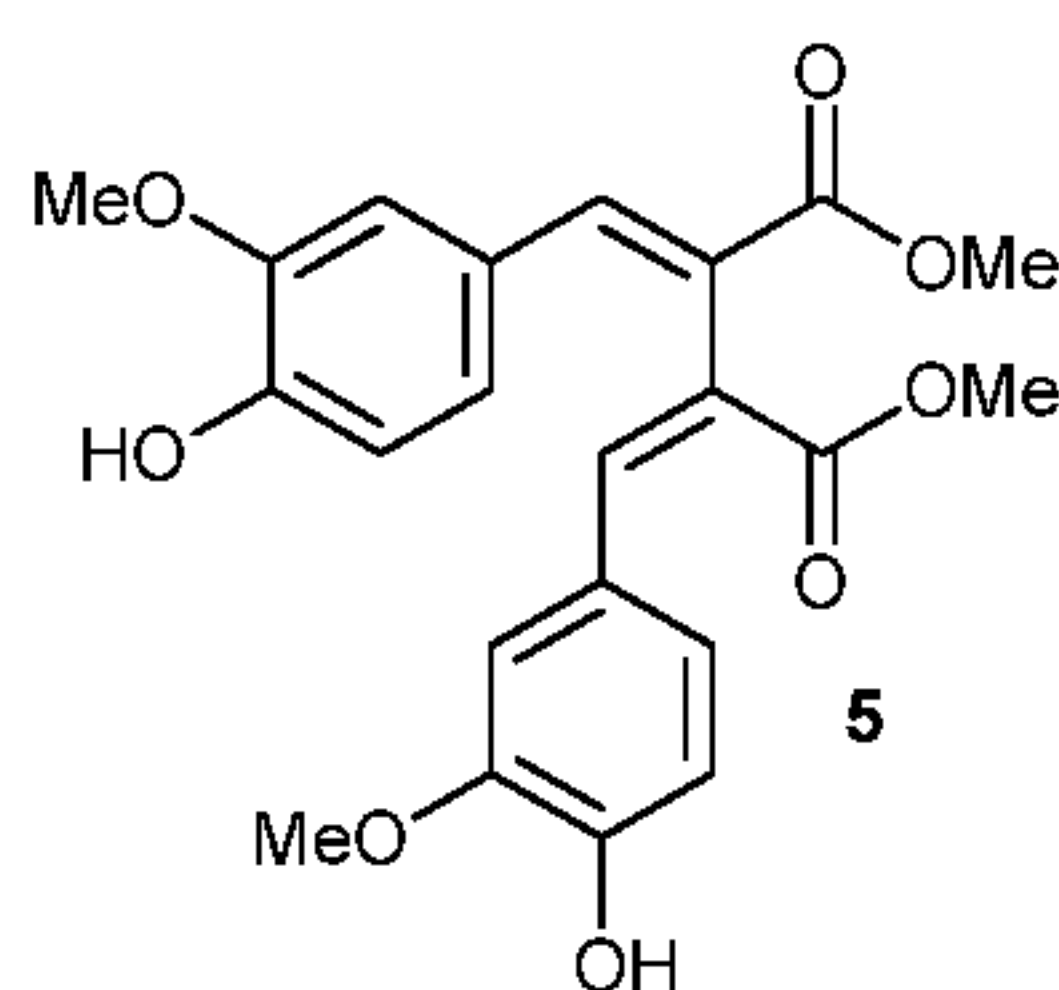
15 with a reducing agent to prepare the compound of formula (6). In some embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[0095] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



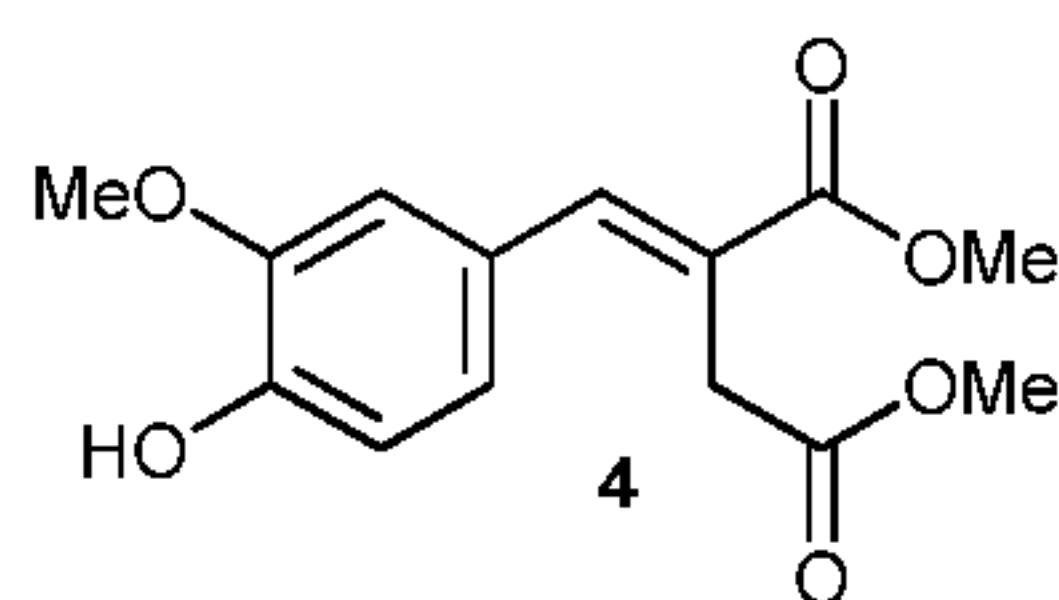
with a benzylating agent to prepare the compound of formula (S2). In certain embodiments, said benzylating agent is BnBr and NaH.

[0096] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In certain embodiments, said reducing agent is H₂ and Pd/C.

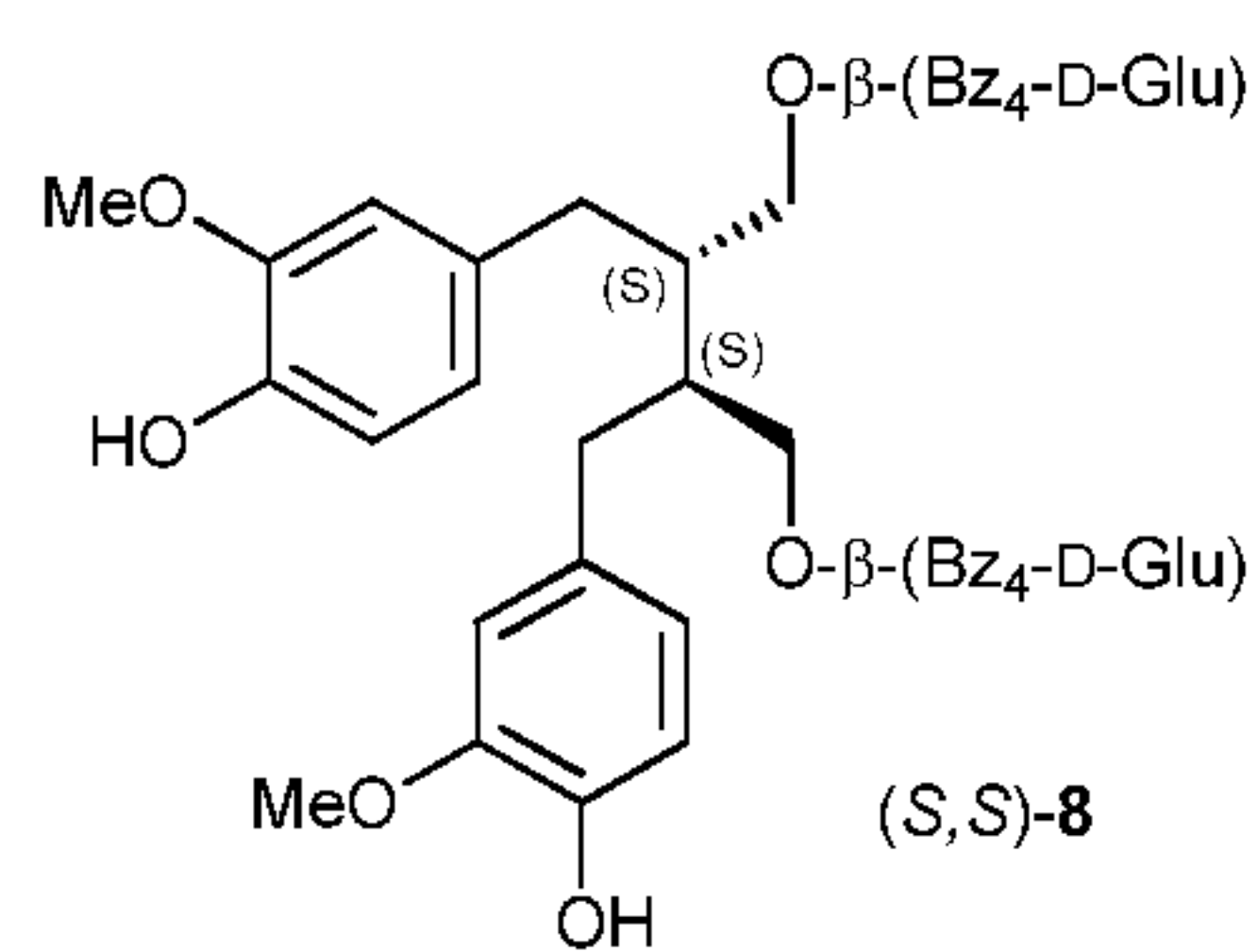
[0097] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

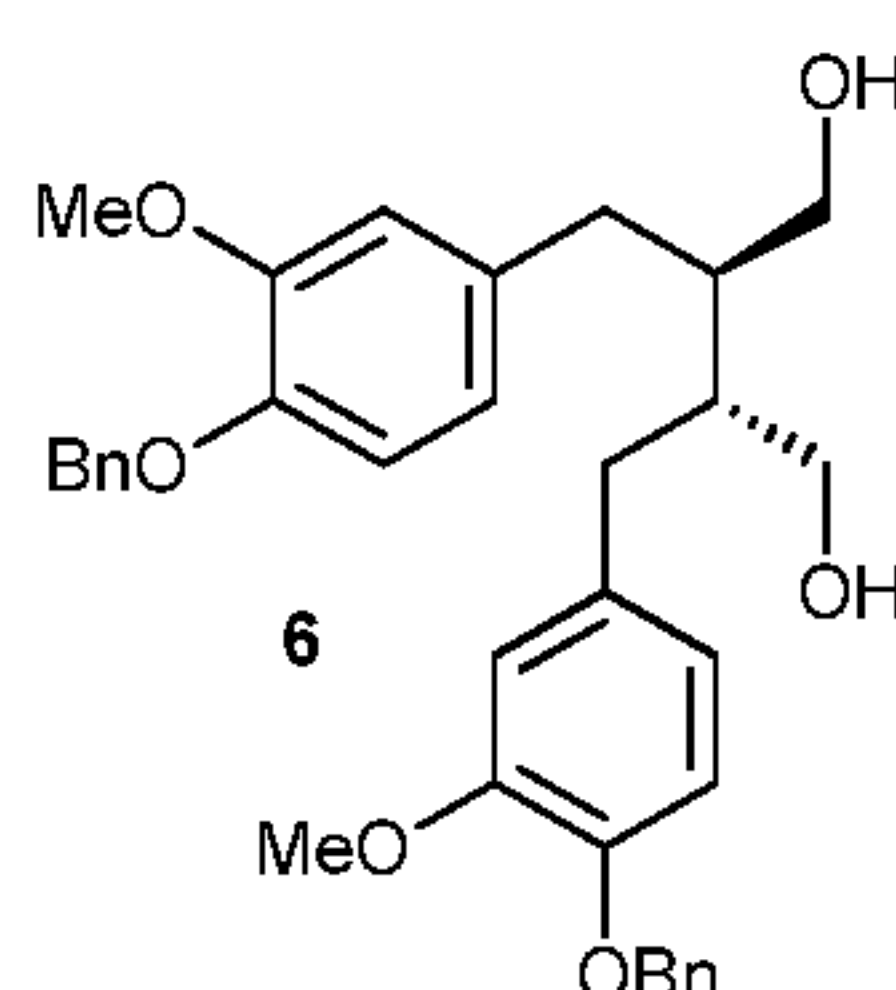
[0098] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[0099] The present invention provides a process for preparing a compound of formula ((*S,S*)-8)

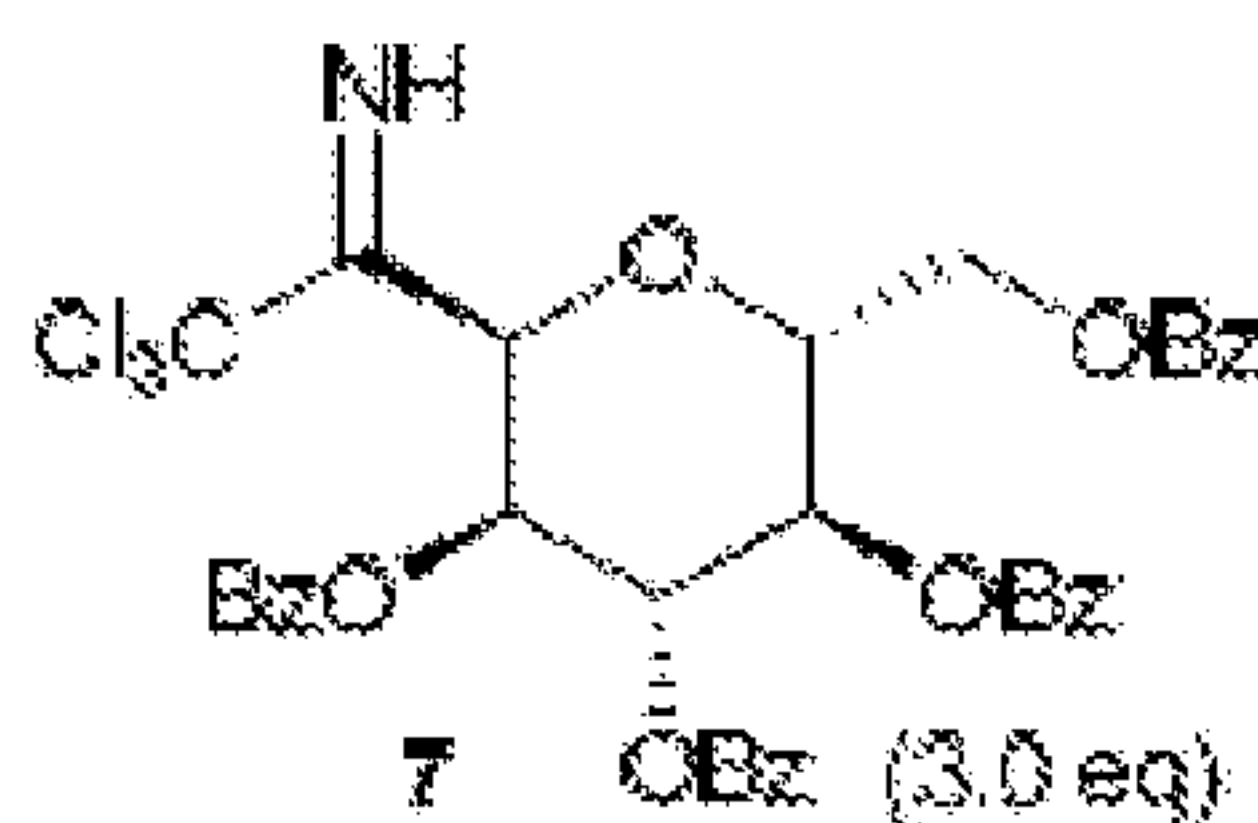


the process comprising:

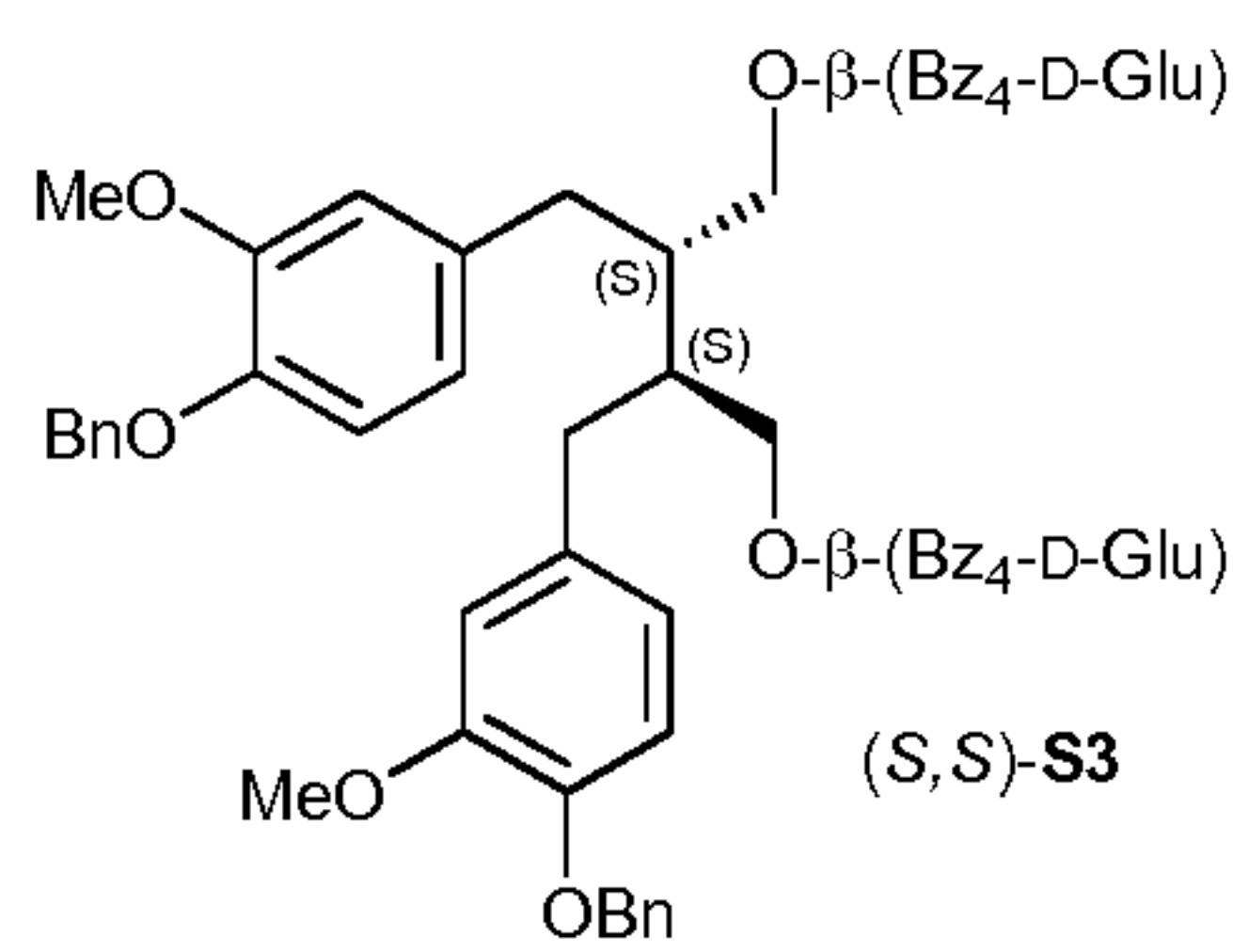
- 5 (a) reacting a compound of formula (6)



with a compound of formula (7)



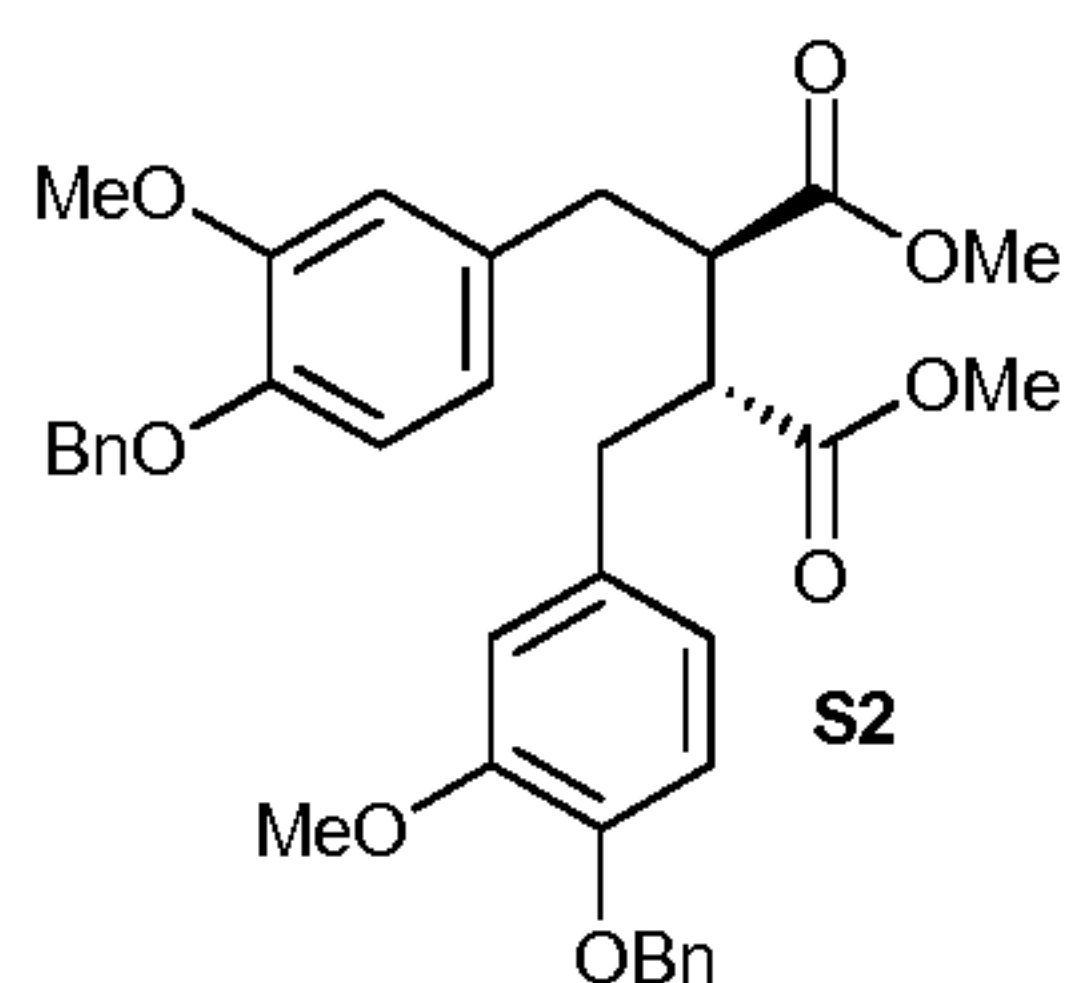
to prepare a compound of formula ((*S,S*)-S3)



(b) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare the compound of formula ((*S,S*)-8).

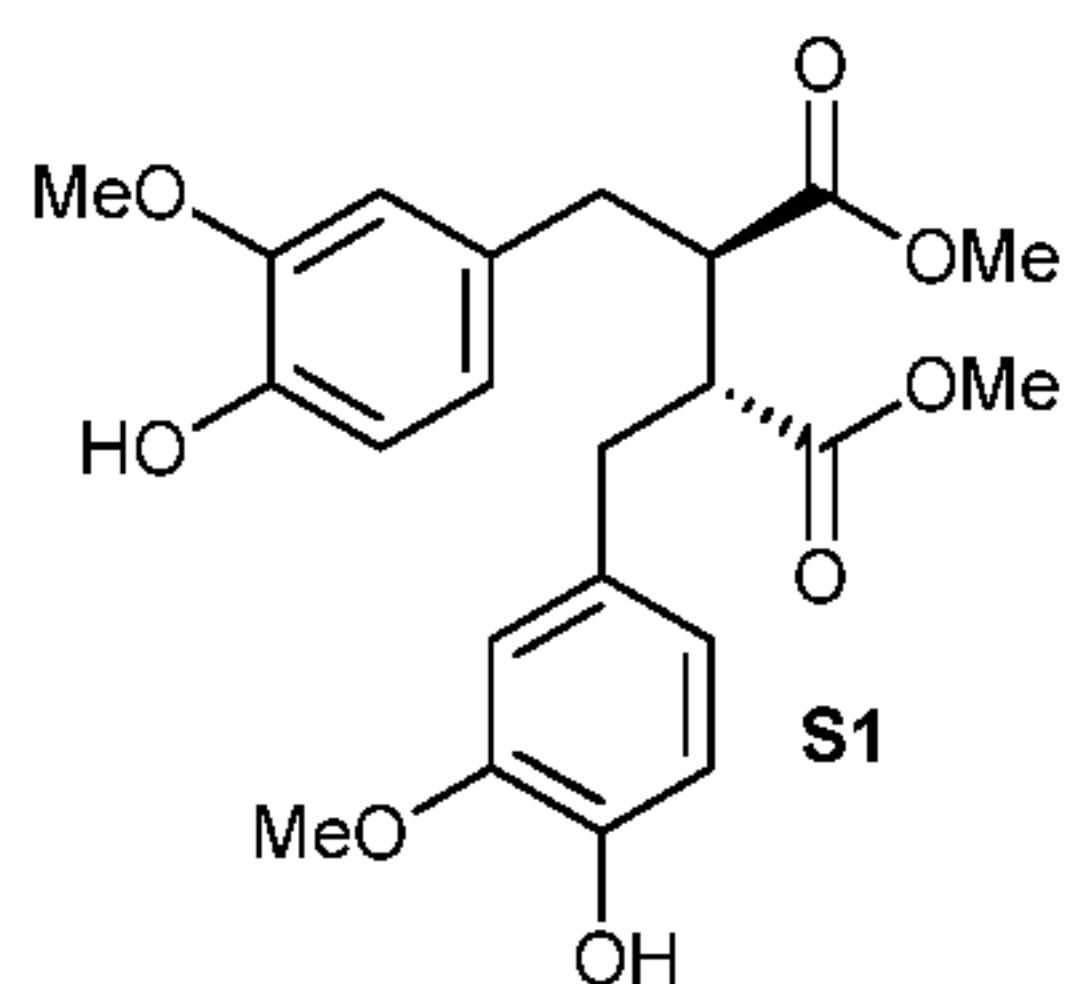
[00100] In some said reacting is carried out in the presence of TMSOTf. In some embodiments, reacting is carried out in the presence of activated molecular sieves. In some
15 embodiments, said cleaving is carried out in the presence of H₂ and Pd/C in MeOH. In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.

[00101] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



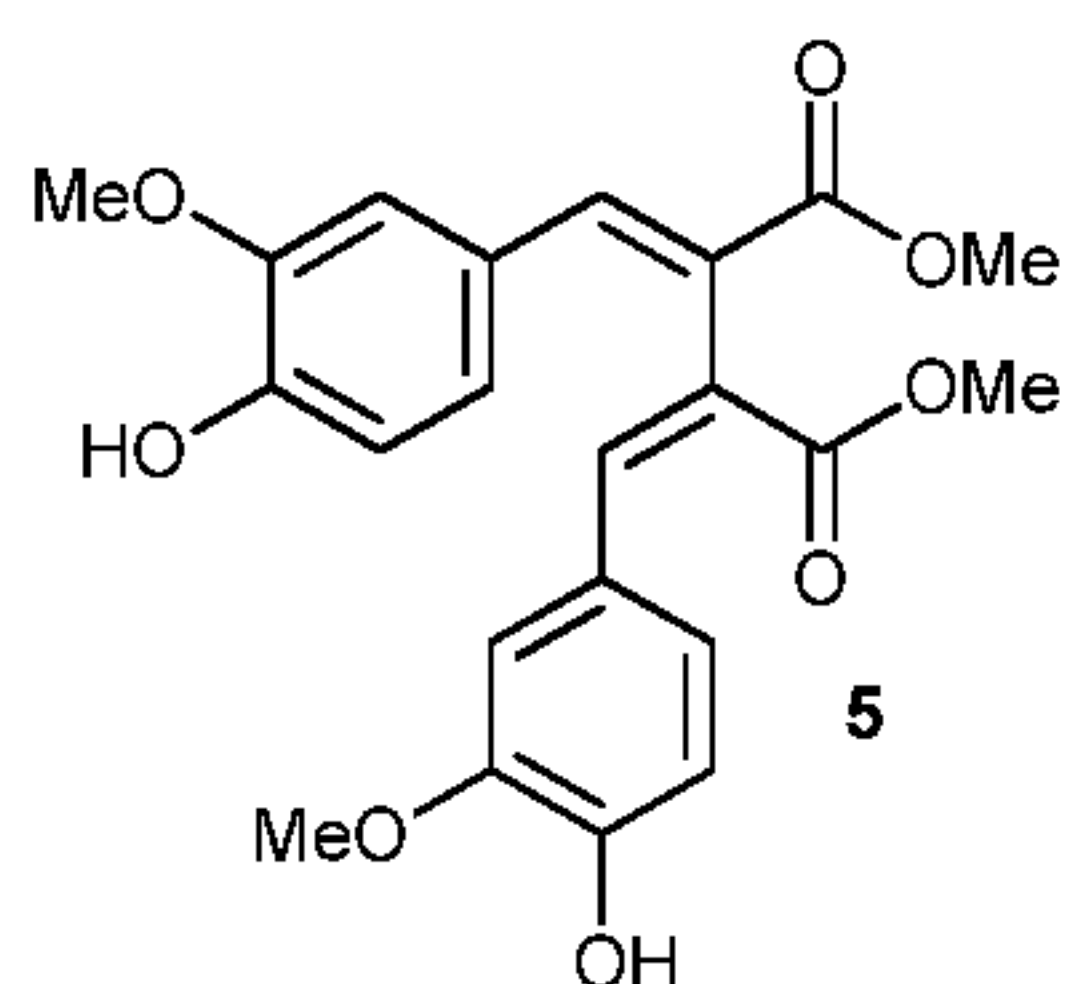
- 5 with a reducing agent to prepare the compound of formula (6). In certain embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[00102] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



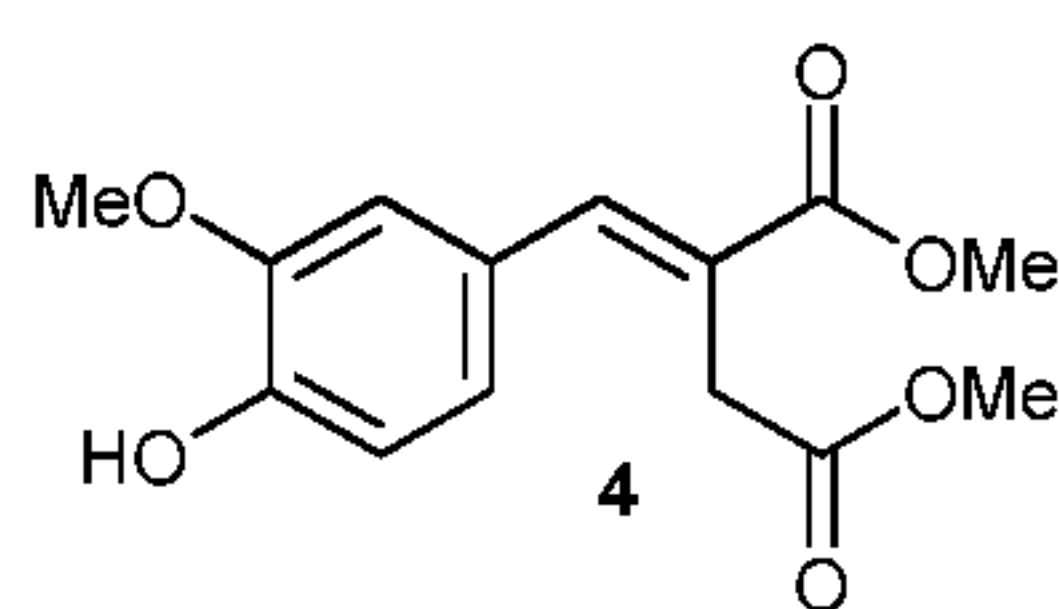
- 10 with a benzylating agent to prepare the compound of formula (S2). In certain embodiments, said benzylating agent is BnBr and NaH.

[00103] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



- 15 with a reducing agent to prepare the compound of formula (S1). In certain embodiments, said reducing agent is H₂ and Pd/C.

[00104] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)

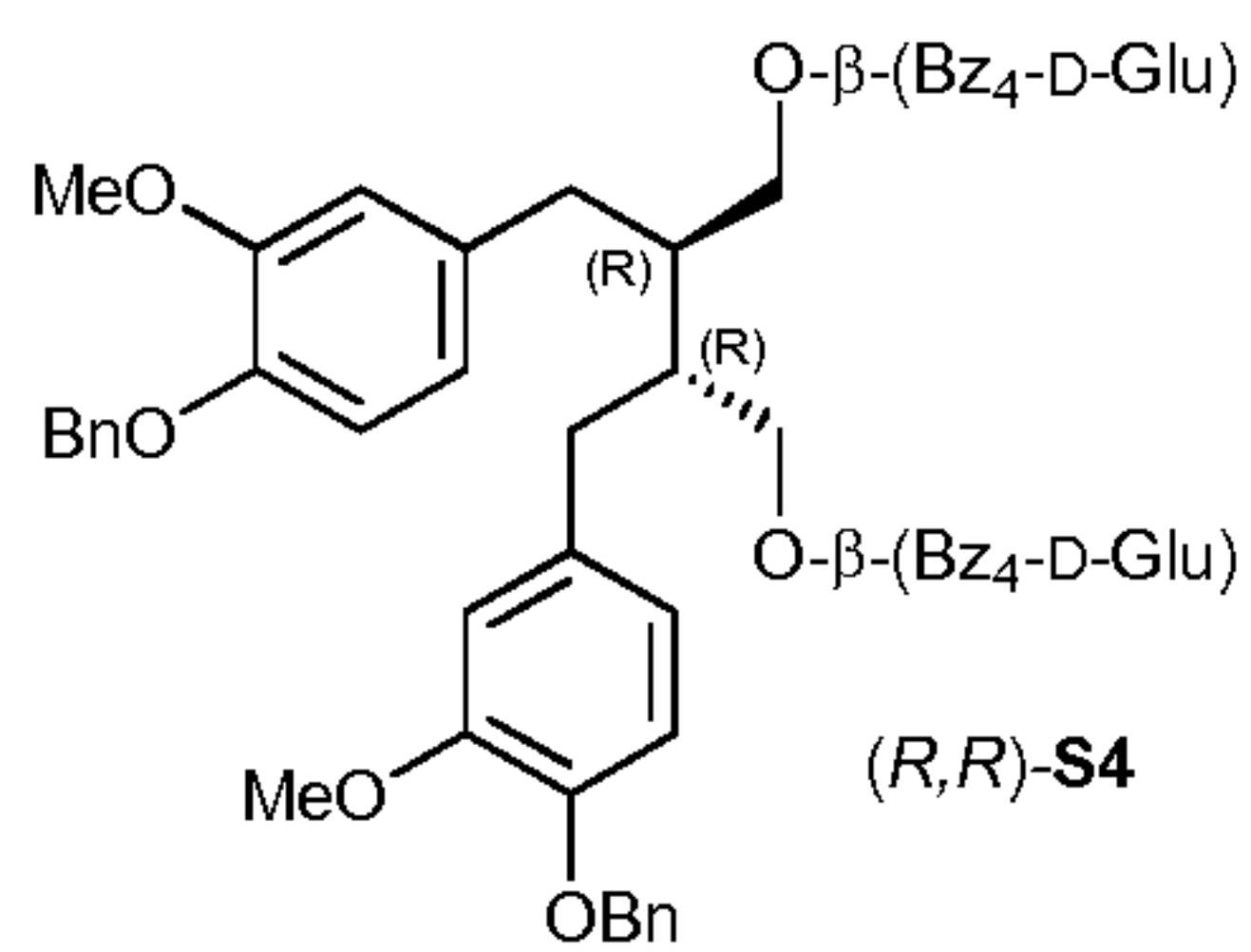


with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

5 [00105] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

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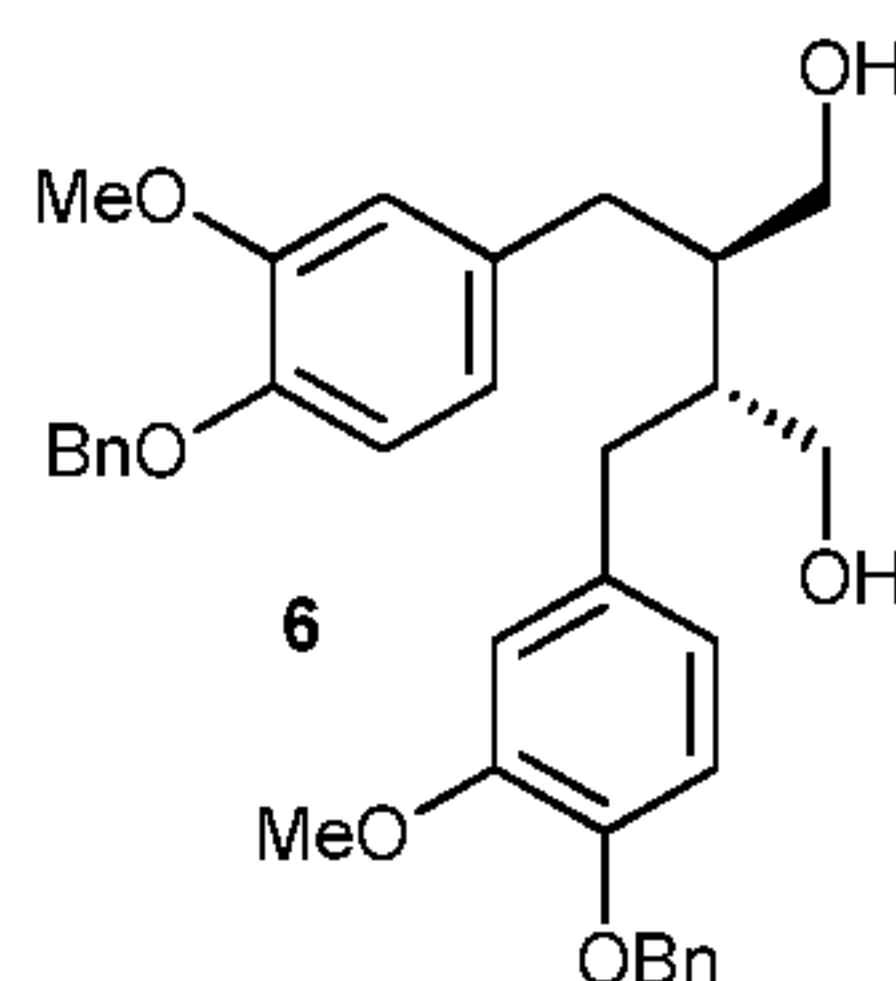
[00106] The present invention provides a process for preparing a compound of formula ((*R,R*)-S4)



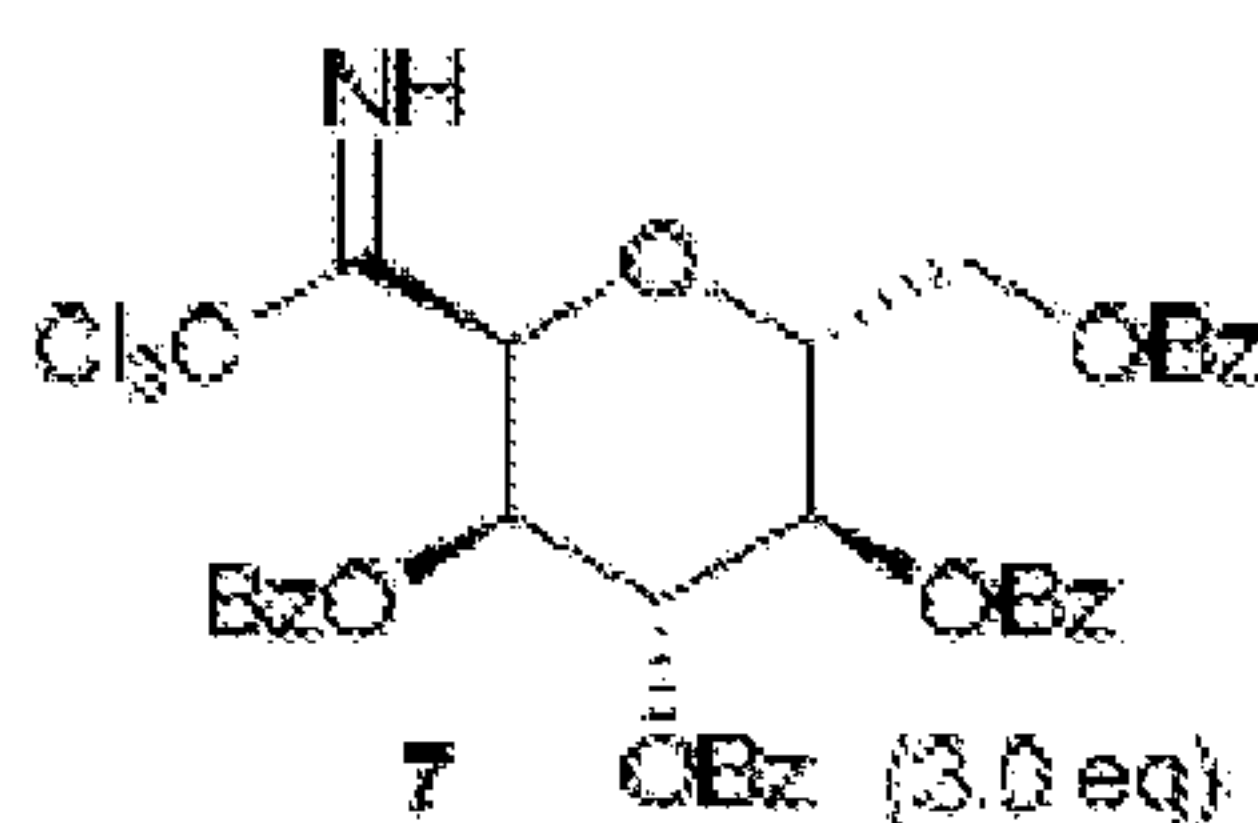
the process comprising:

15

(a) reacting a compound of formula (6)



with a compound of formula (7)



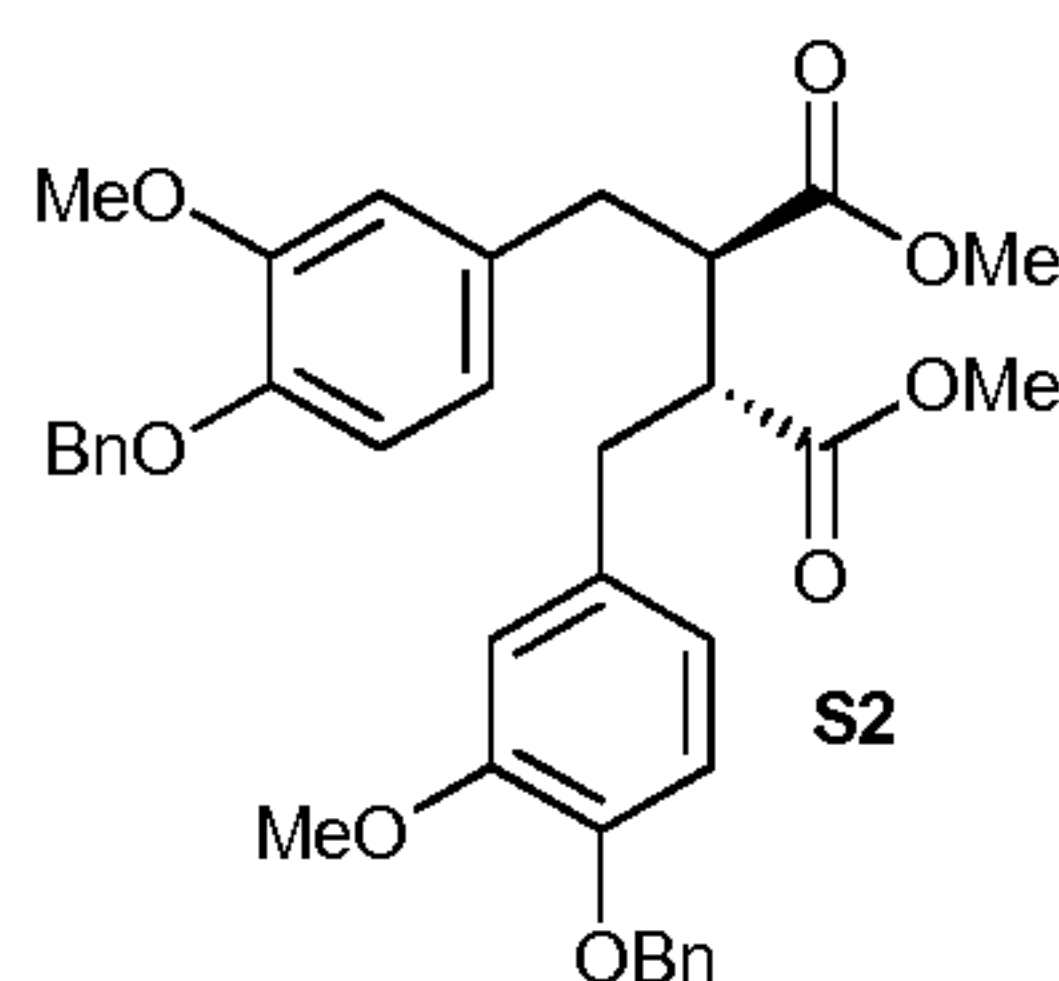
to prepare a compound of formula ((*R,R*)-S4)

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[00107] In some embodiments, said reacting is carried out in the presence of TMSOTf.

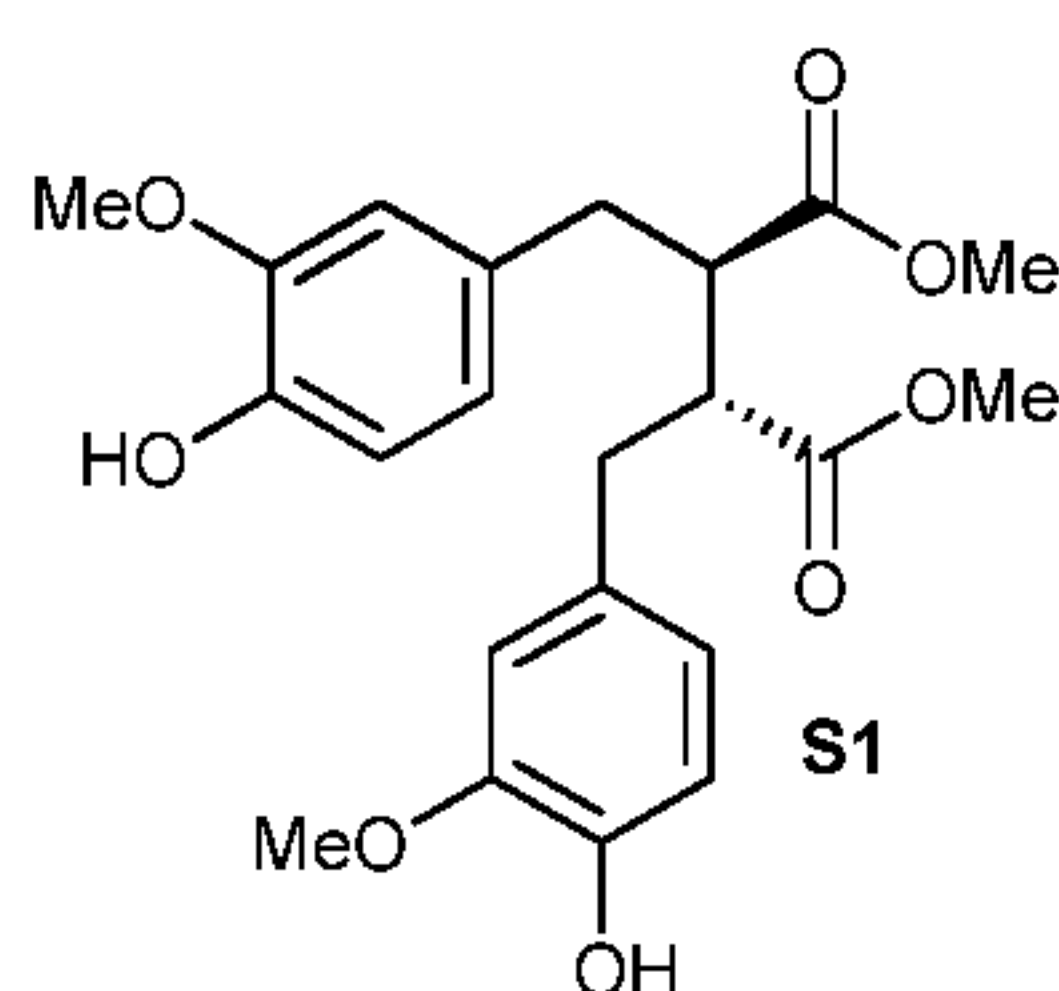
[00108] In some embodiments, said reacting is carried out in the presence of activated molecular sieves.

[00109] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



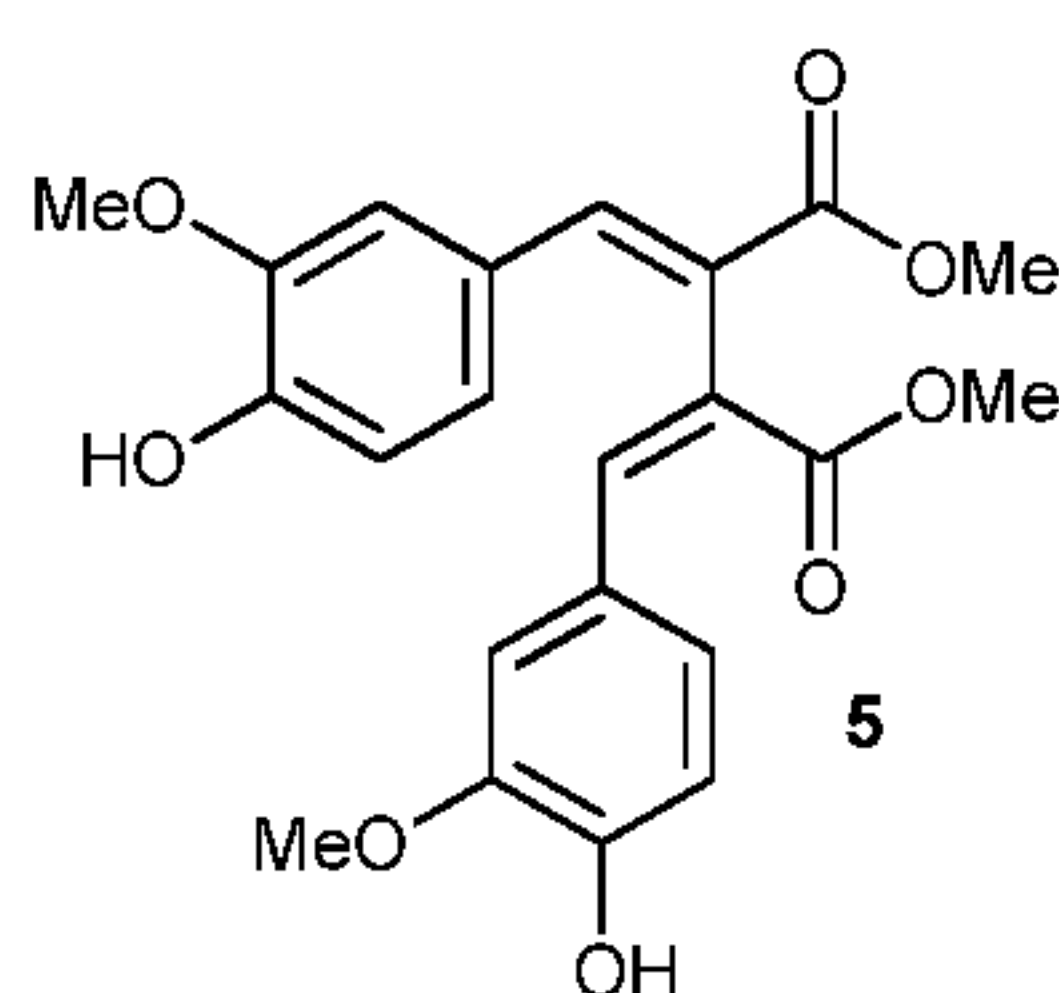
with a reducing agent to prepare the compound of formula (6). In certain embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

[00110] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



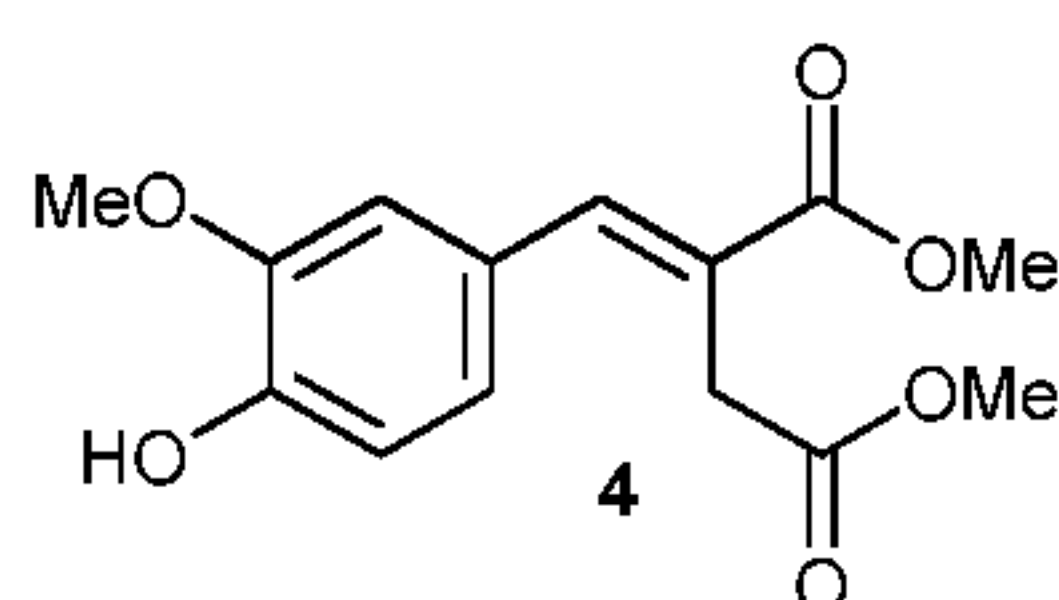
with a benzylating agent to prepare the compound of formula (S2). In certain embodiments, said benzylating agent is BnBr and NaH.

[00111] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In certain embodiments, said reducing agent is H₂ and Pd/C.

[00112] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)

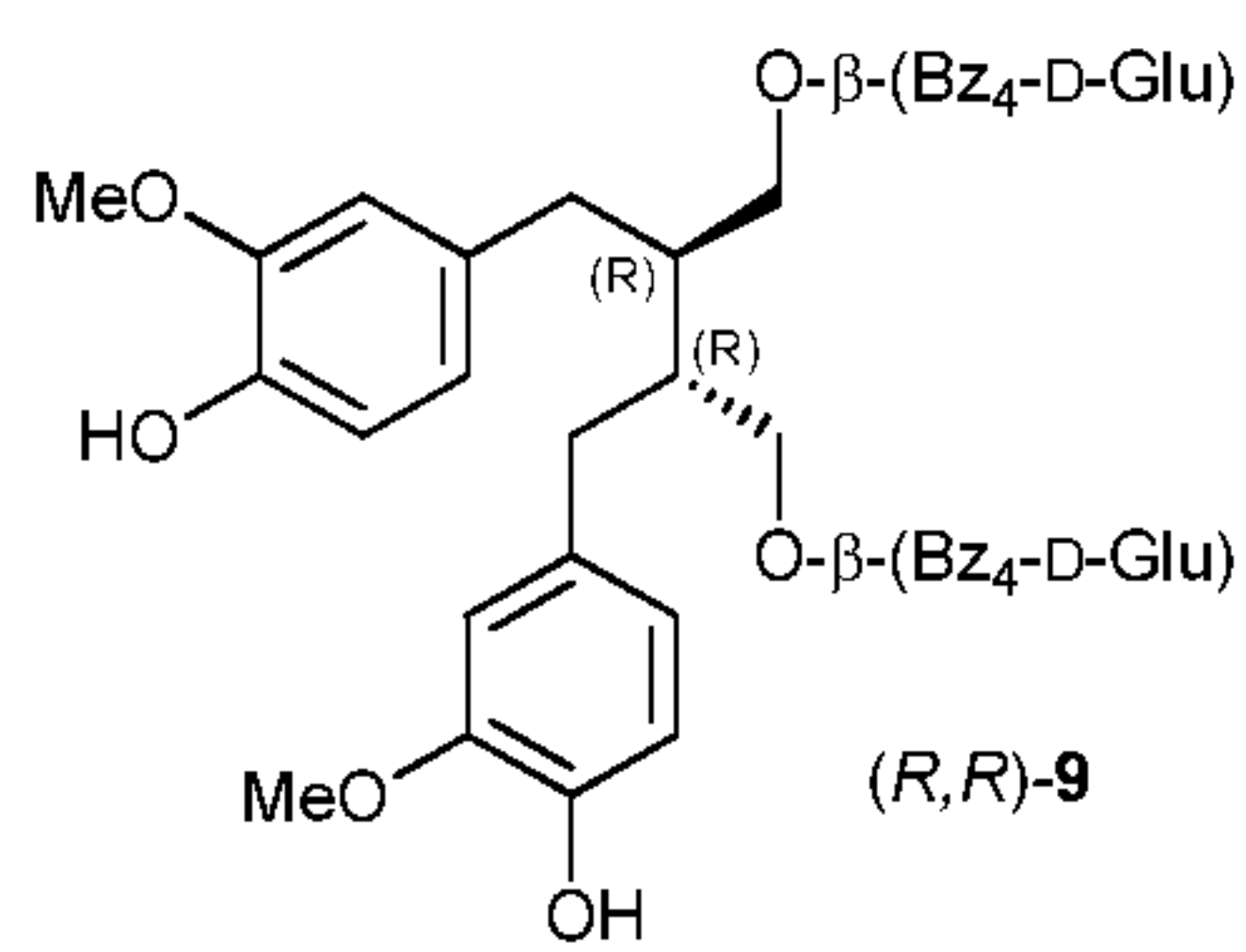


with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

- 5 [00113] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

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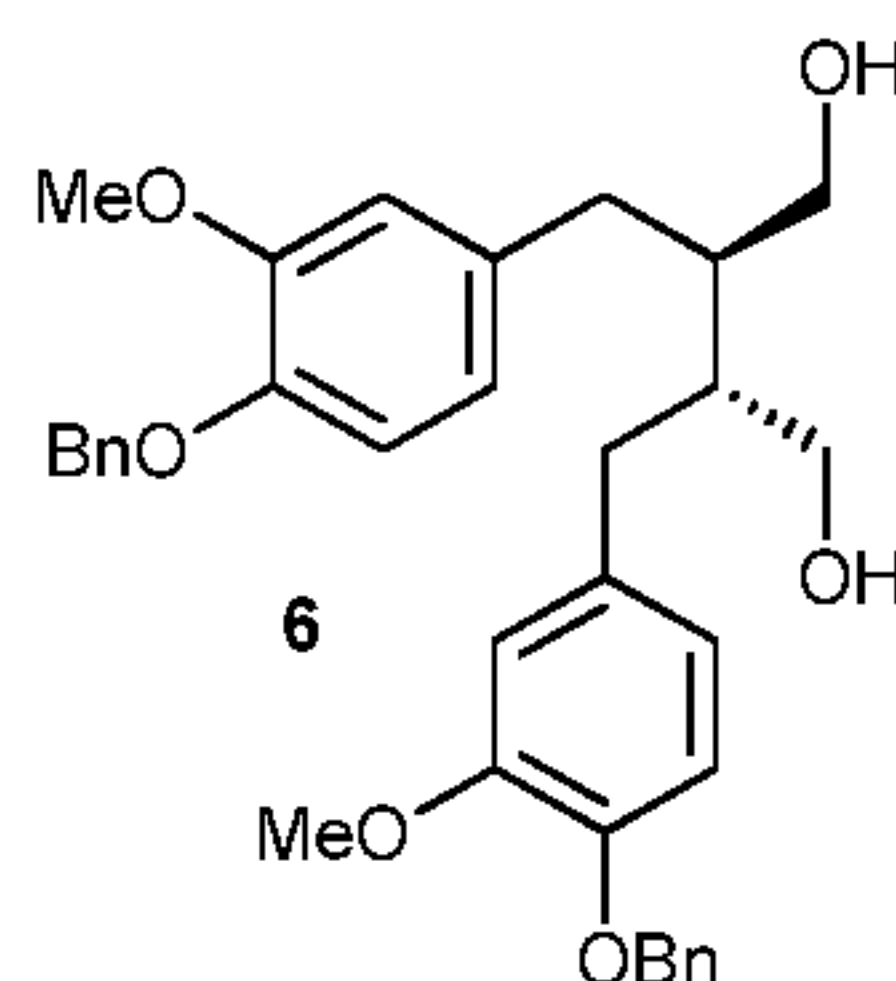
[00114] The present invention provides a process for preparing a compound of formula ((*R,R*)-9)



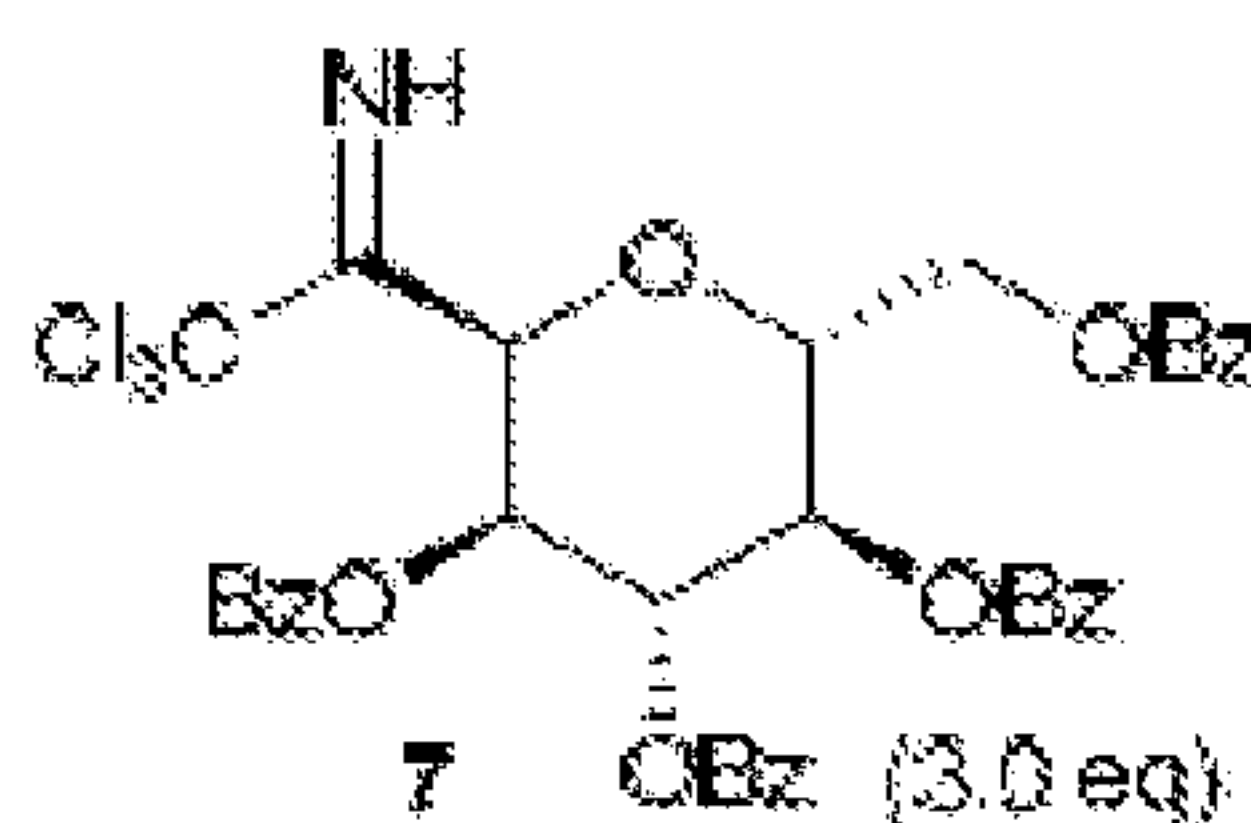
the process comprising:

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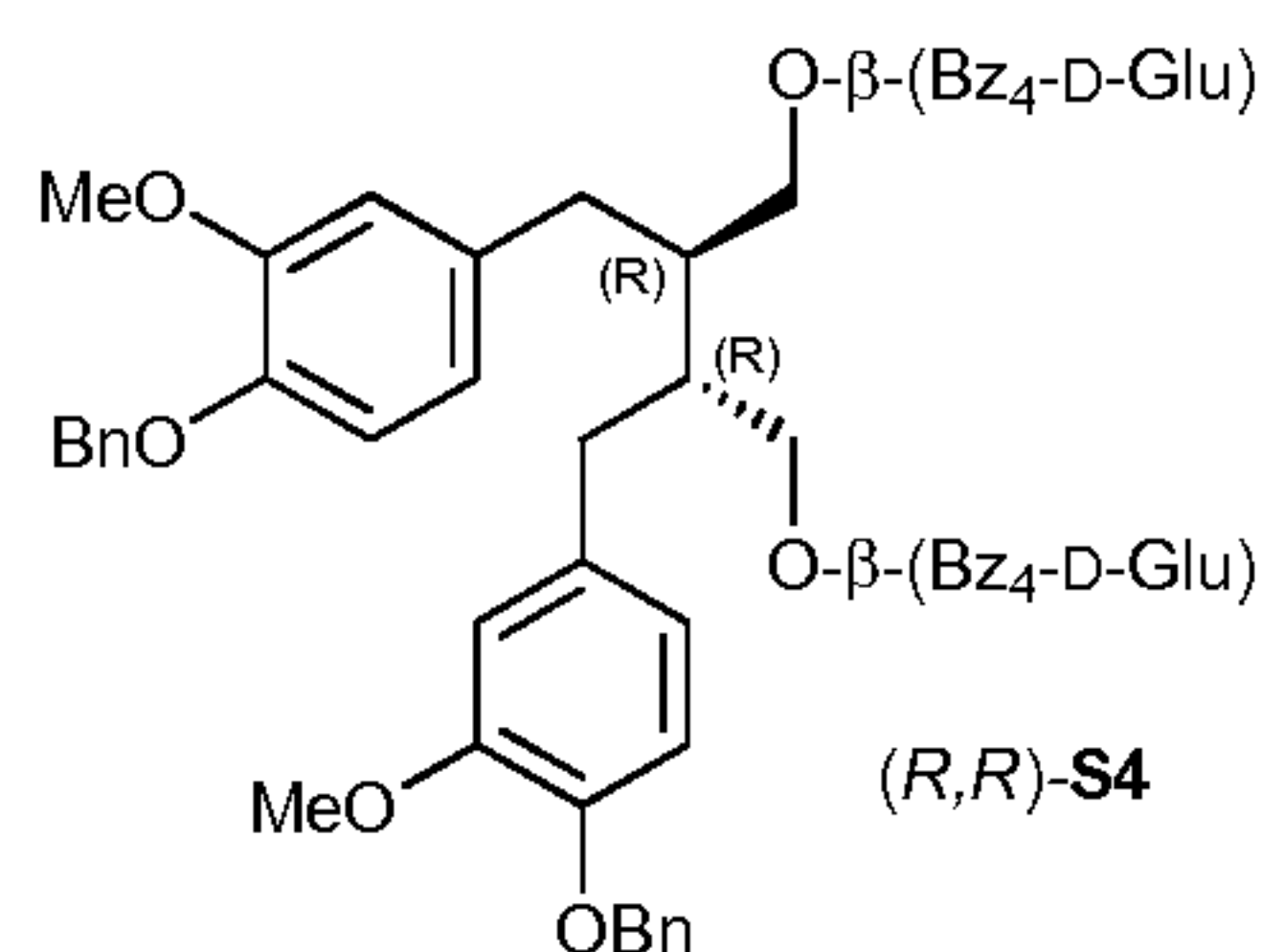
- (a) reacting a compound of formula (6)



with a compound of formula (7)

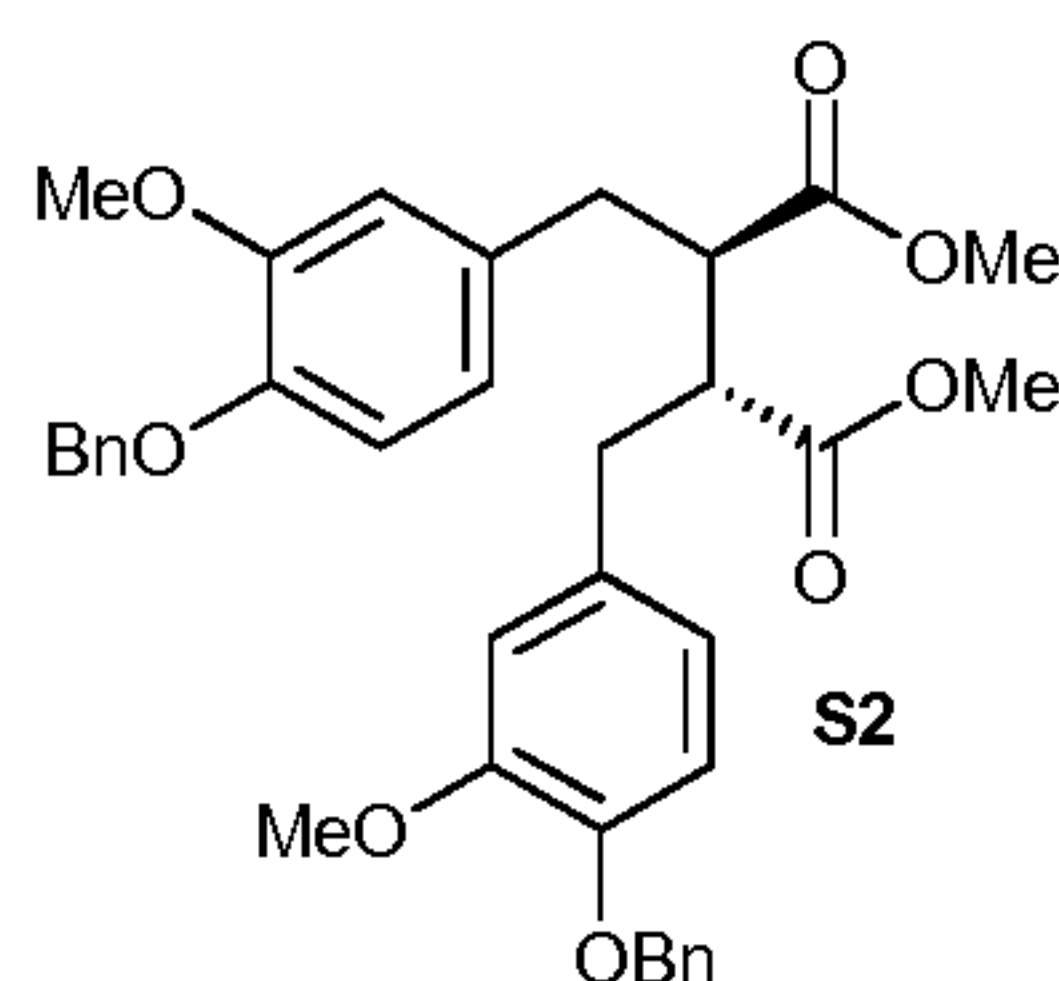


to prepare a compound of formula ((*R,R*)-S4)



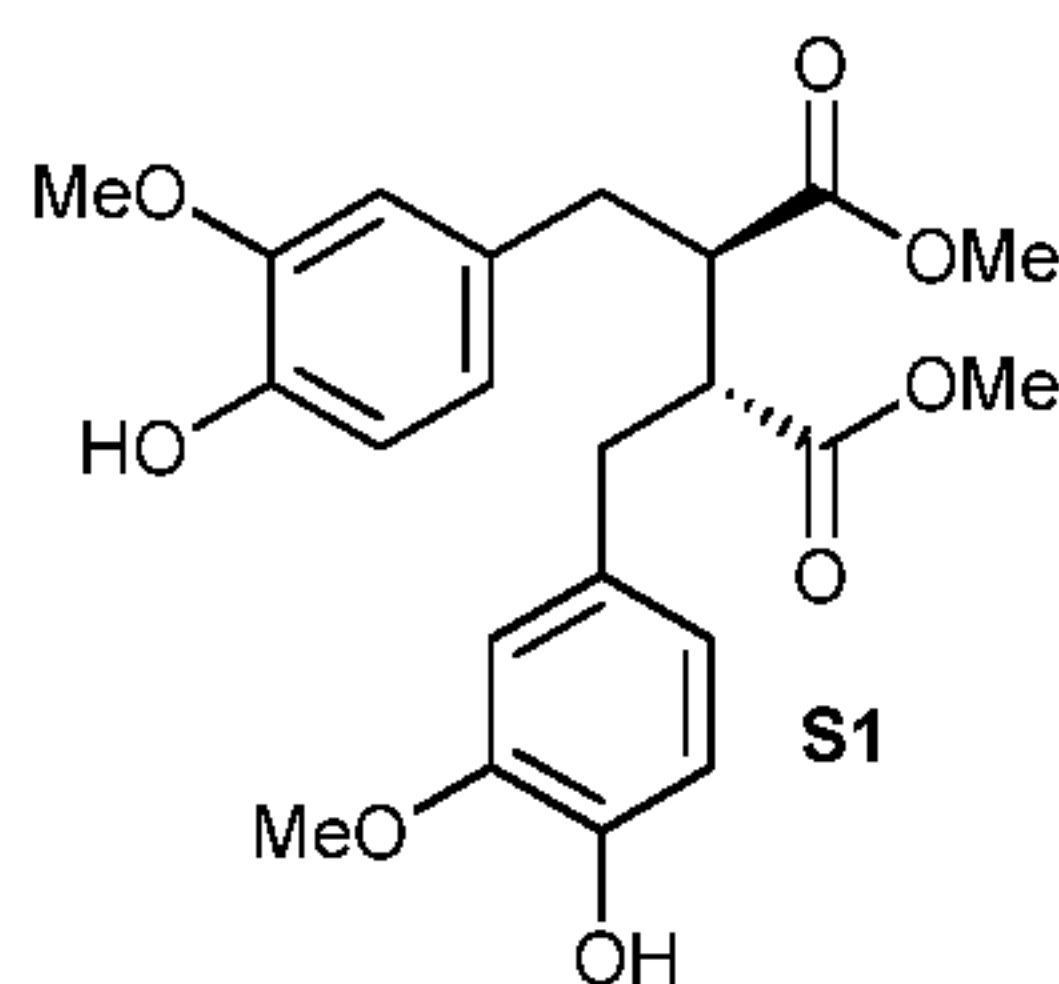
(b) cleaving benzyl ethers of the compound of formula ((R,R)-S4), followed by a separation procedure to prepare a compound of formula ((R,R)-9)

- 5 [00115] In some embodiments, said reacting is carried out in the presence of TMSOTf. In some embodiments, said reacting is carried out in the presence of activated molecular sieves. In some embodiments, said cleaving is carried out in the presence of H₂ and Pd/C in MeOH. In some embodiments, said separation procedure is carried out using preparative thin layer chromatography.
- 10 [00116] In some embodiments, said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



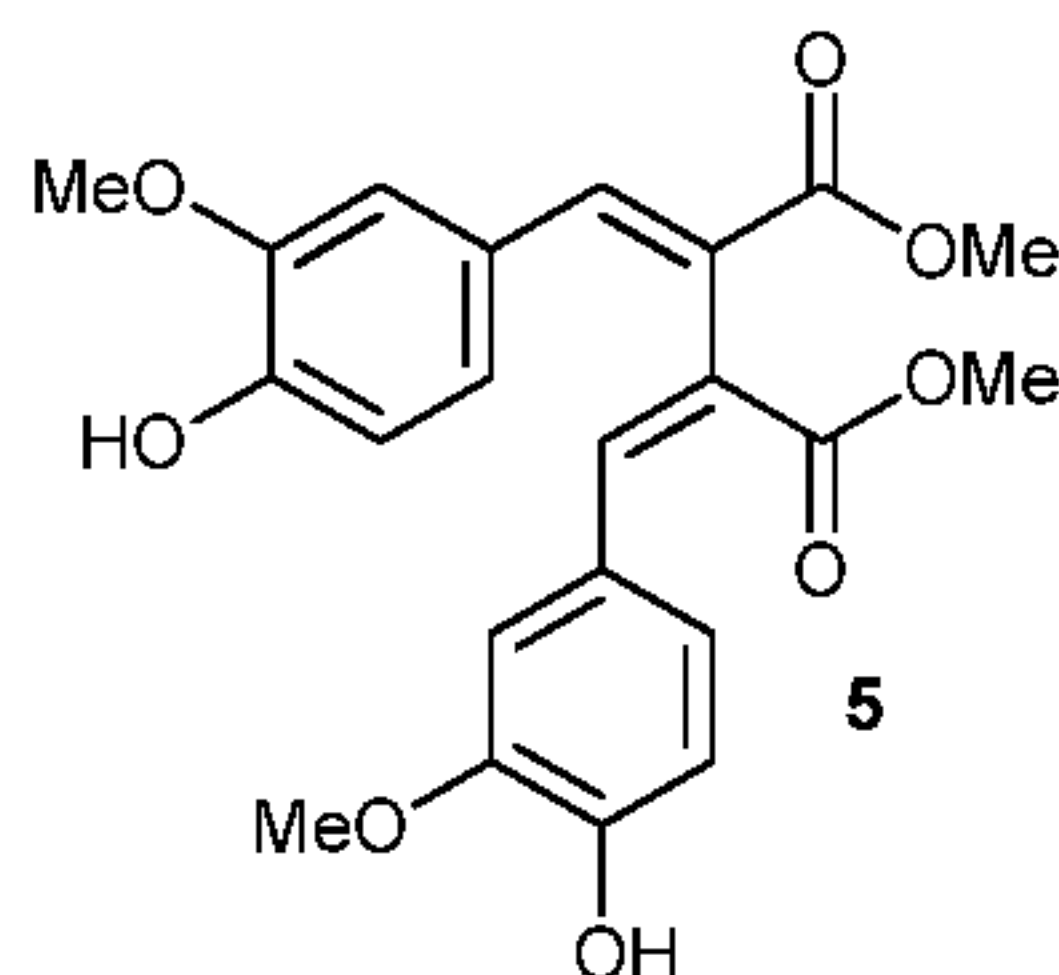
with a reducing agent to prepare the compound of formula (6). In certain embodiments, said reducing agent is lithium aluminum hydride (LAH) in THF.

- 15 [00117] In some embodiments, said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



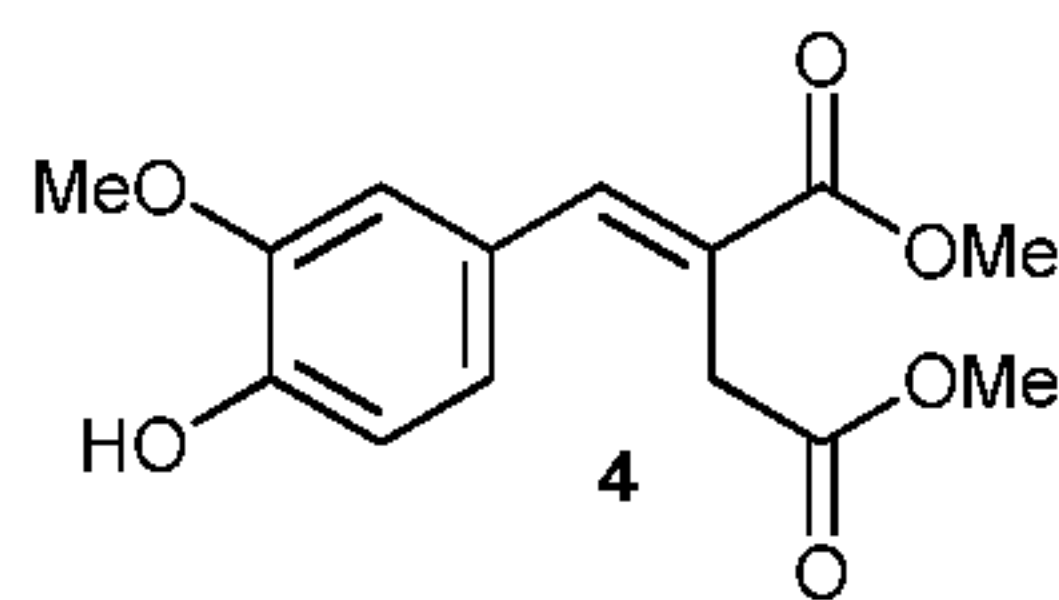
with a benzylating agent to prepare the compound of formula (S2). In certain embodiments, said benzylating agent is BnBr and NaH.

- 20 [00118] In some embodiments, said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1). In certain embodiments, said reducing agent is H₂ and Pd/C.

[00119] In some embodiments, said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5). In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[00120] In some embodiments, said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction. In some embodiments, said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires. In some embodiments, said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

[00121] The term “reacting” is meant to refer to the bringing together of the indicated reagents in such a way as to allow their molecular interaction and chemical transformation according to the thermodynamics and kinetics of the chemical system. Reacting can be facilitated, particularly for solid reagents, by using an appropriate solvent or mixture of solvents in which at least one of the reagents is at least partially soluble. Reacting is typically carried out for a suitable time and under conditions suitable to bring about the desired chemical transformation.

[00122] The processes described herein may include other suitable starting materials through the synthetic routes set forth above. In some embodiments, the processes described herein may also include additional steps before or after the steps described above, to add or remove suitable protecting groups. In addition, various synthetic steps may be performed in an alternate sequence or order to give the desired compounds.

[00123] In one aspect, the present invention provides a concise route for the preparation of both (*S,S*)-secoisolariciresinol diglucoside ((*S,S*)-SDG-1) and (*R,R*)-secoisolariciresinol diglucoside ((*R,R*)-SDG-2) starting from commercially available compounds, e.g., vanillin. The compounds ((*S,S*)-SDG-1) and ((*R,R*)-SDG-2) are bioactive ingredients in flaxseed. Both
5 compounds possess strong reducing power and high free radical scavenging activity for hydroxyl, peroxy, and DPPH (2,2-diphenyl-1-picrylhydrazyl) free radicals.

[00124] The processes of the present invention can yield substantially pure compound (*S,S*)-secoisolariciresinol diglucoside ((*S,S*)-SDG-1) and compound (*R,R*)-secoisolariciresinol diglucoside ((*R,R*)-SDG-2). For ((*S,S*)-SDG-1), by “substantially pure” is meant that
10 compound ((*S,S*)-SDG-1) is at least substantially separated from the environment in which it was formed or detected. Substantial purity can include compositions containing at least about 80.0%, or at least about 85.0%, or at least about 90.0%, or at least about 95.0%, or at least about 97.0%, or at least about 98.0%, or at least about 99.0%, or at least about 99.2%, or at least about 99.4%, or at least about 99.6%, or at least about 99.8%, or at least about 99.9%, or
15 even about 100% by weight of the compound. For ((*R,R*)-SDG-2), by “substantially pure” is meant that compound ((*R,R*)-SDG-2) is at least substantially separated from the environment in which it was formed or detected. Substantial purity can include compositions containing at least about 80.0%, or at least about 85.0%, or at least about 90.0%, or at least about 95.0%, or at least about 97.0%, or at least about 98.0%, or at least about 99.0%, or at least about 99.2%,
20 or at least about 99.4%, or at least about 99.6%, or at least about 99.8%, or at least about 99.9%, or even about 100% by weight of the compound.

[00125] A “pharmaceutical composition” is a formulation containing compounds in a form suitable for administration to a subject. In some embodiments, the pharmaceutical composition is in bulk or in unit dosage form. The unit dosage form is any of a variety of
25 forms, including, for example, a capsule, an IV bag, a tablet, a single pump on an aerosol inhaler, or a vial. The quantity of active ingredient (e.g., a formulation of a compound of the invention or a salt, solvate, polymorph, or prodrug thereof) in a unit dose of composition is an effective amount and is varied according to the particular treatment involved. One skilled in the art will appreciate that it is sometimes necessary to make routine variations to the dosage
30 depending on the age and condition of the patient. The dosage will also depend on the route of administration. A variety of routes are contemplated, including oral, pulmonary, rectal, parenteral, transdermal, subcutaneous, intravenous, intramuscular, intraperitoneal, inhalational, buccal, sublingual, intrapleural, intrathecal, intranasal, and the like. Dosage

forms for the topical or transdermal administration of a compound of the invention include powders, sprays, ointments, pastes, creams, lotions, gels, solutions, patches and inhalants. In some embodiments, the active compound is mixed under sterile conditions with a pharmaceutically acceptable carrier, and with any preservatives, buffers, or propellants that are required.

[00126] A “subject” includes mammals, e.g., humans, companion animals (e.g., dogs, cats, birds, and the like), farm animals (e.g., cows, sheep, pigs, horses, fowl, and the like) and laboratory animals (e.g., rats, mice, guinea pigs, birds, and the like). In some embodiments, the subject is human.

[00127] As used herein, the phrase “pharmaceutically acceptable” refers to those compounds, materials, compositions, carriers, and/or dosage forms which are, within the scope of sound medical judgment, suitable for use in contact with the tissues of human beings and animals without excessive toxicity, irritation, allergic response, or other problem or complication, commensurate with a reasonable benefit/risk ratio.

[00128] “Pharmaceutically acceptable excipient” means an excipient that is useful in preparing a pharmaceutical composition that is generally safe, non-toxic and neither biologically nor otherwise undesirable, and includes an excipient that is acceptable for veterinary use as well as human pharmaceutical use. A “pharmaceutically acceptable excipient” as used herein includes both one and more than one such excipient.

[00129] Embodiments of the invention also include compositions comprising (*S,S*)-secoisolariciresinol diglucoside ((*S,S*)-SDG-1). Preferably, these compositions are pharmaceutical compositions comprising (*S,S*)-secoisolariciresinol diglucoside ((*S,S*)-SDG-1) and at least one pharmaceutically acceptable excipient. In some embodiments, the compositions and pharmaceutical compositions may be prepared with substantially pure (*S,S*)-secoisolariciresinol diglucoside ((*S,S*)-SDG-1). In some embodiments, the compositions and pharmaceutical compositions have a diastereomeric excess (DE) of at least 90% DE, preferably at least 95% DE, more preferably at least 98% DE, and even more preferably at least 99% DE and most preferably about 100% DE. The compositions and pharmaceutical compositions may also be prepared as mixture of the diastereomeric forms of the compounds (e.g., as a racemic mixture or as a mixture with a ratio of 60:40, 70:30, 80:20 or 90:10 of ((*S,S*)-SDG-1) relative to ((*R,R*)-SDG-2)).

[00130] Embodiments of the invention also include compositions comprising (*R,R*)-secoisolariciresinol diglucoside ((*R,R*)-SDG-2). Preferably, these compositions are

pharmaceutical compositions comprising (*R,R*)-secoisolariciresinol diglucoside ((*R,R*)-SDG-2) and at least one pharmaceutically acceptable excipient. In some embodiments, the compositions and pharmaceutical compositions may be prepared with substantially pure (*R,R*)-secoisolariciresinol diglucoside ((*R,R*)-SDG-2). In some embodiments, the compositions and pharmaceutical compositions have a diastereomeric excess (DE) of at least 90% DE, preferably at least 95% DE, more preferably at least 98% DE, and even more preferably at least 99% DE and most preferably about 100% DE. The compositions and pharmaceutical compositions may also be prepared as mixture of the diastereomeric forms of the compounds (e.g., as a racemic mixture or as a mixture with a ratio of 60:40, 70:30, 80:20 or 90:10 of ((*R,R*)-SDG-2) relative to ((*S,S*)-SDG-1)).

[00131] Throughout the description, where compositions are described as having, including, or comprising specific components, it is contemplated that compositions also consist essentially of, or consist of, the recited components. Similarly, where processes are described as having, including, or comprising specific process steps, the processes also consist essentially of, or consist of, the recited processing steps. Further, it should be understood that the order of steps or order for performing certain actions are immaterial so long as the invention remains operable. Moreover, two or more steps or actions may be conducted simultaneously.

[00132] The following examples are presented in order to more fully illustrate the preferred embodiments of the invention. They should in no way be construed, however, as limiting the broad scope of the invention.

EXAMPLES

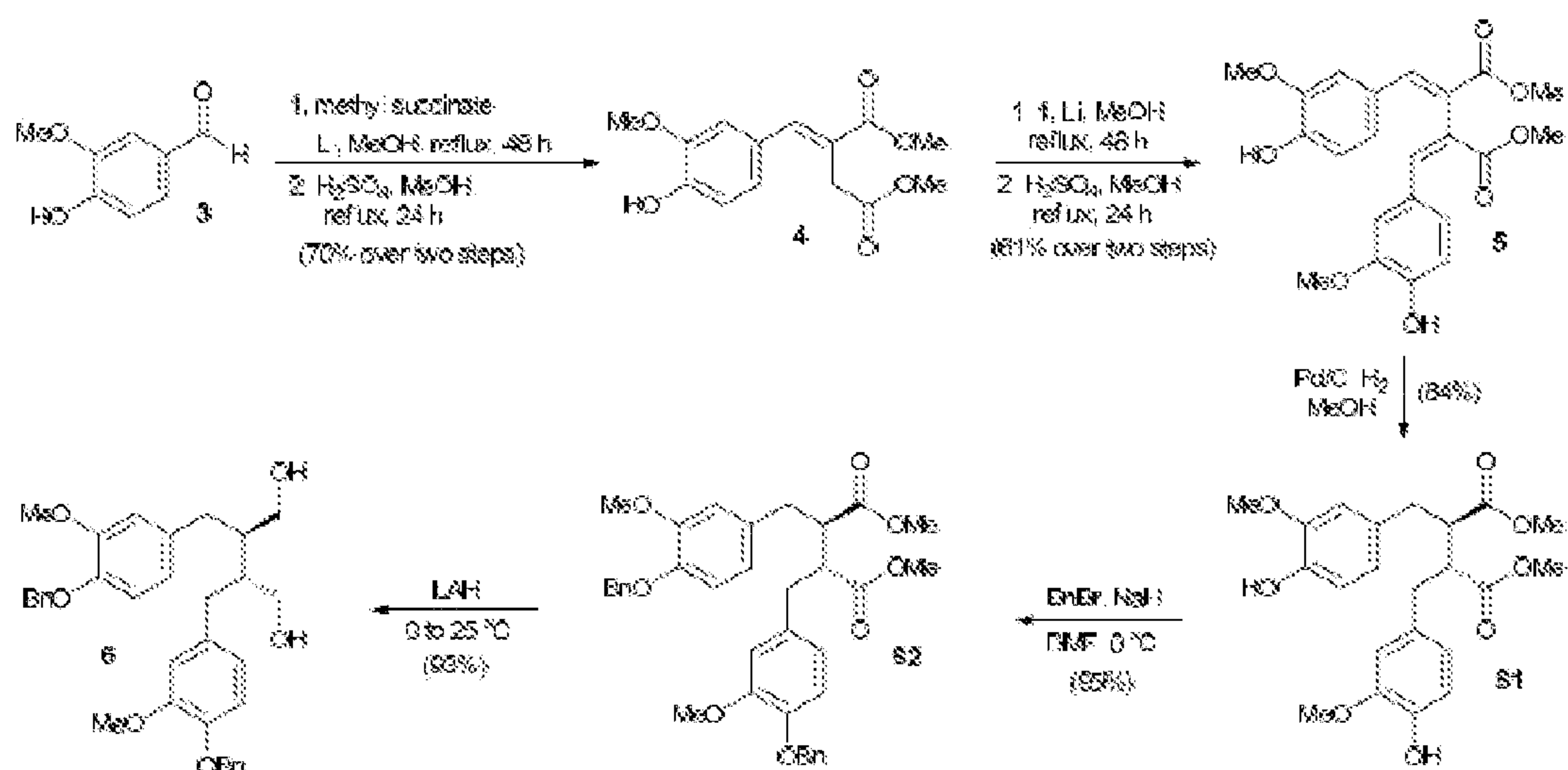
General Procedures for Chemical Reactions

[00133] All reactions were carried out under an argon atmosphere with dry solvents under anhydrous conditions, unless otherwise noted. Dry tetrahydrofuran (THF), dimethylformamide (DMF), and methylene chloride (CH₂Cl₂) were obtained by passing commercially available pre-dried, oxygen-free formulations through activated alumina columns. Yields refer to chromatographically and spectroscopically (¹H NMR) homogeneous materials, unless otherwise stated. Reagents were purchased at the highest commercial quality and used without further purification, unless otherwise stated. Reactions were monitored by thin-layer chromatography (TLC) carried out on 0.25 mm E. Merck silica gel plates (60F-254) using UV light as visualizing agent and an ethanolic solution of phosphomolybdic acid and cerium sulfate, and heat as developing agents. E. Merck silica gel (60, particle size 0.040–0.063 mm)

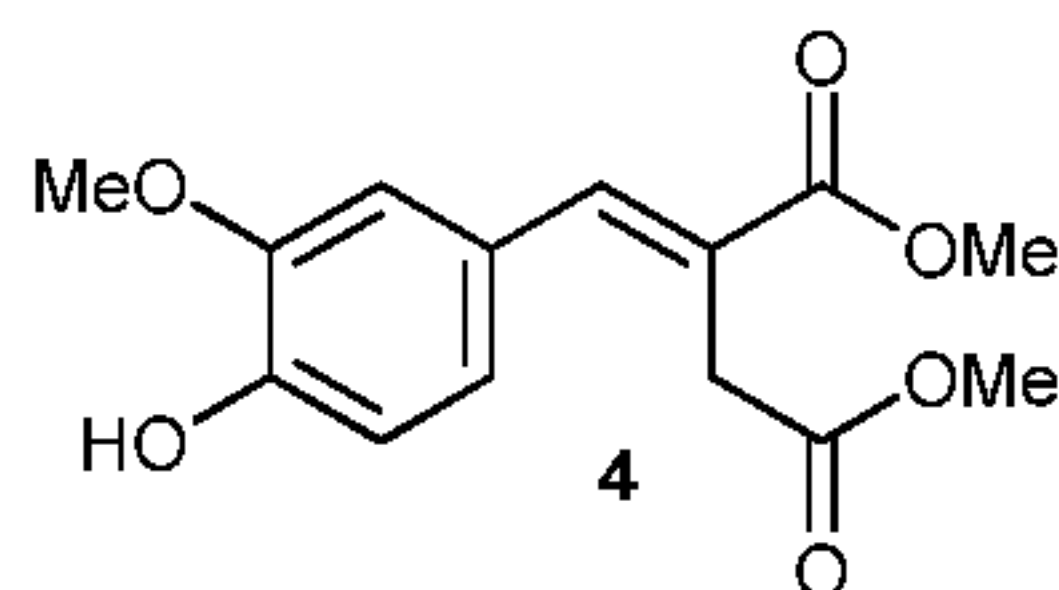
was used for flash column chromatography. Preparative thin-layer chromatography (PTLC) separations were carried out on 0.25 or 0.50 mm E. Merck silica gel plates (60F-254) unless otherwise noted. NMR spectra were recorded on a Bruker DRX-600 instrument and calibrated using residual undeuterated solvent as an internal reference. The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, m = multiplet. IR spectra were recorded on a Perkin-Elmer Spectrum 100 FT-IR spectrometer. High-resolution mass spectra (HR-MS) were recorded on a VG ZAB-ZSE mass spectrometer using ESI (electrospray ionization).

[00134] Scheme 1 shows a reaction route that was used in the synthesis of compound (6), an intermediate compound for the preparation of (*S,S*)-secoisolariciresinol diglucosides and (*R,R*)-secoisolariciresinol diglucosides.

[00135] **Scheme 1: Synthesis of compound (6)**



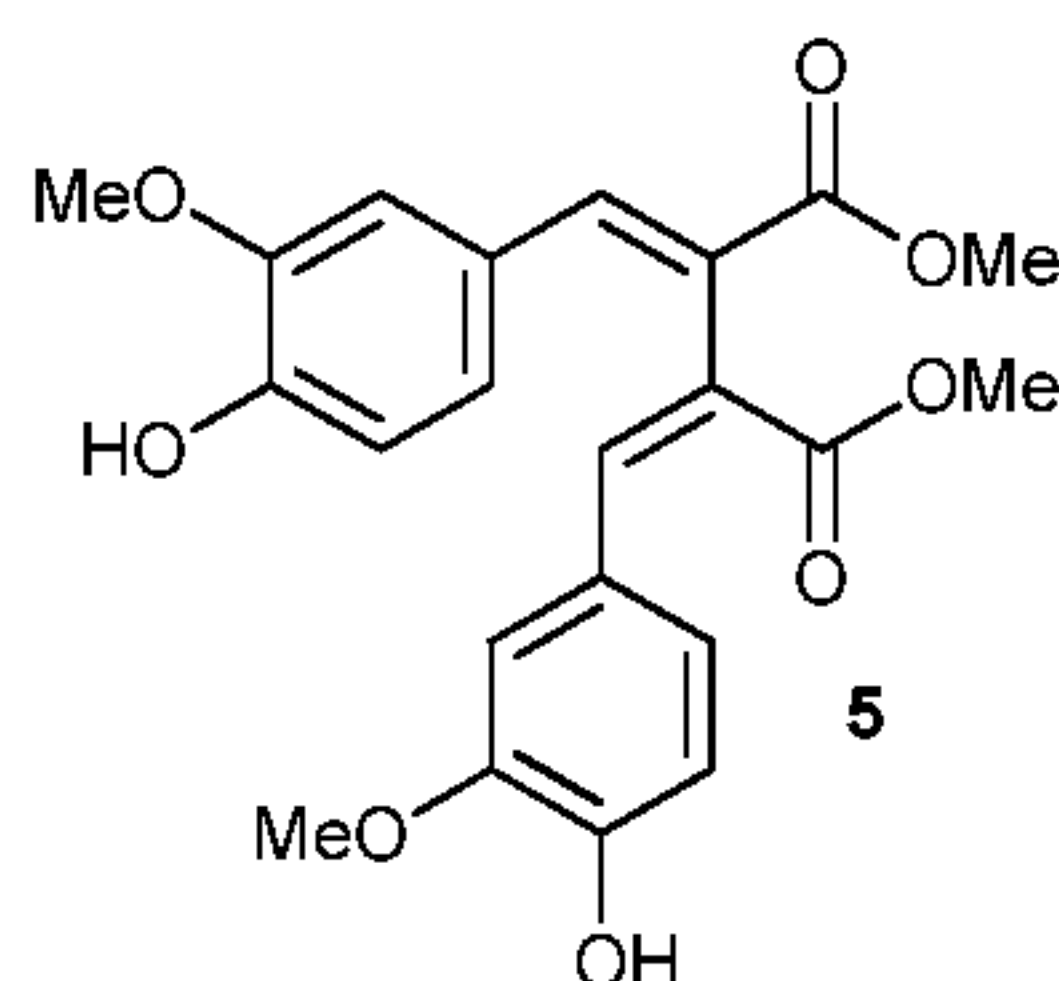
Example 1: Synthesis of Phenol 4



[00136] To a solution of vanillin **3** (14.0 g, 92.1 mmol, 1 equiv) and dimethyl succinate (12.13 mL, 92.1 mmol, 1 equiv) in MeOH (450 mL), lithium wire (1.79 g, 257.9 mmol, 2.8 equiv) was added slowly piecewise with an ice bath to control the exotherm. After the initial lithium had fully dissolved, more lithium (1.98 g, 285.5 mmol, 3.1 equiv) was added slowly piecewise and stirred until fully dissolved. The reaction mixture was then heated at reflux for 48 h.

After cooling the solution to room temperature, most of the methanol was removed by concentration on rotavap. EtOAc (1000 mL) was added and the solution was washed with a 2 M aq. HCl solution (700 mL), H₂O (3 × 1000 mL), then brine (200 mL). The organics layer was then dried (MgSO₄), filtered and concentrated. The crude was dissolved in MeOH (230 ml), H₂SO₄ (1 mL) was added, and the solution was heated at reflux overnight. Upon cooling the next morning, NaHCO₃ (3.0 g) was added to quench H₂SO₄ and the solution was mostly concentrated by rotavap. EtOAc (500 ml) was added and the solution was washed with H₂O (2 × 200 mL), then brine (100 mL). The organics layer was then dried (MgSO₄), filtered and concentrated to a brown oil. Dissolution in a small amount of CH₂Cl₂ and then flash column chromatography (silica, 1:9 → 2:8 → 3:7 → 4:6 → 5:5 ether:hexane) provided **4** (18.0 g, 64.2 mmol, 70% yield, unassigned olefin geometry) as an off-white solid. **4**: *R*_f = 0.17 (silica, ether:hexanes 1:1); IR (film): ν_{max} = 3422, 1702, 1514, 1434, 1258, 1195 1159, 1093, 1030, 924, 821 770, 729 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 7.79 (s, 1 H), 6.90–6.85 (m, 3 H), 6.19 (s, 1 H), 3.82 (s, 3 H), 3.78 (s, 3H), 3.69 (s, 3H), 3.57 (s, 2H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ = 171.94, 168.13, 146.80, 146.61, 142.38, 126.96, 123.39, 123.31, 114.76, 111.80, 55.85, 52.24, 52.20, 33.61 ppm; HRMS (ESI-TOF): calcd for C₁₄H₁₆O₆ [M + H⁺]: 280.102, found 281.1022.

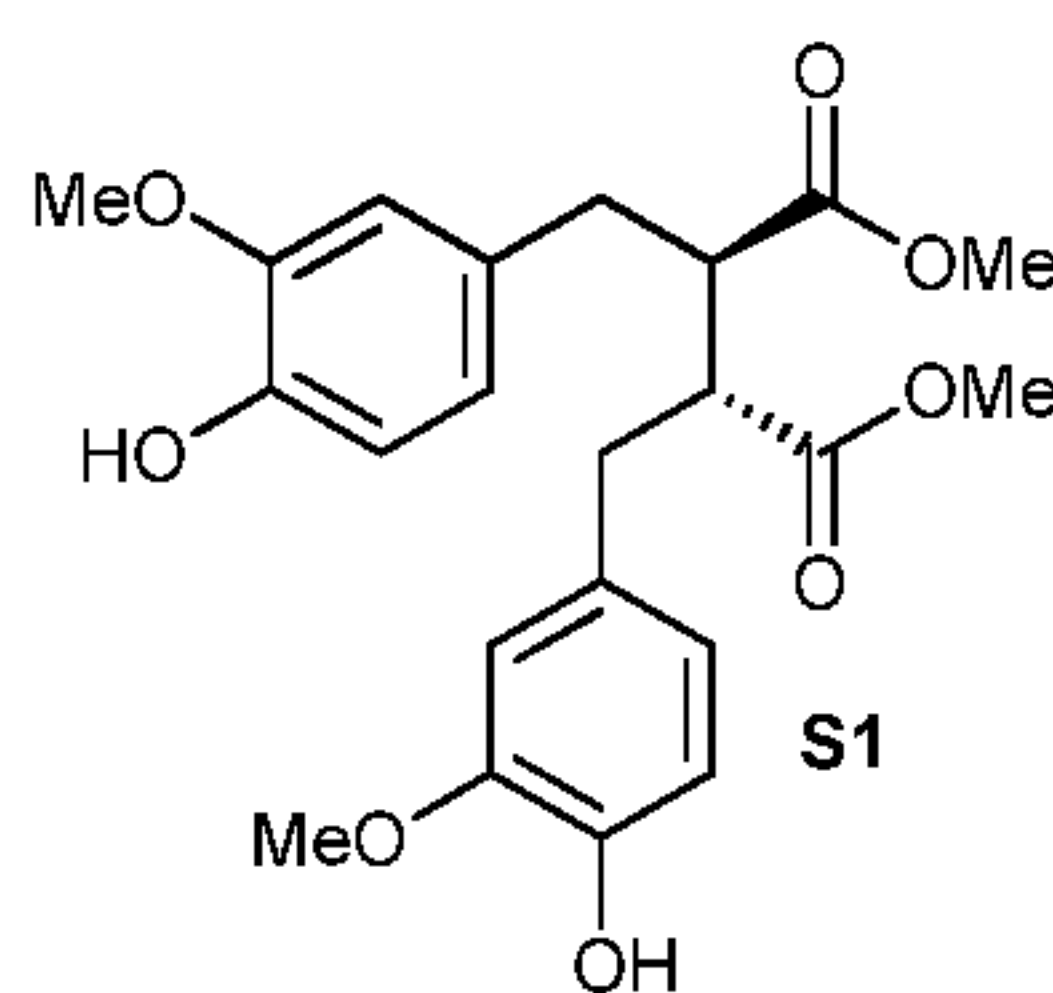
Example 2: Synthesis of Diester 5



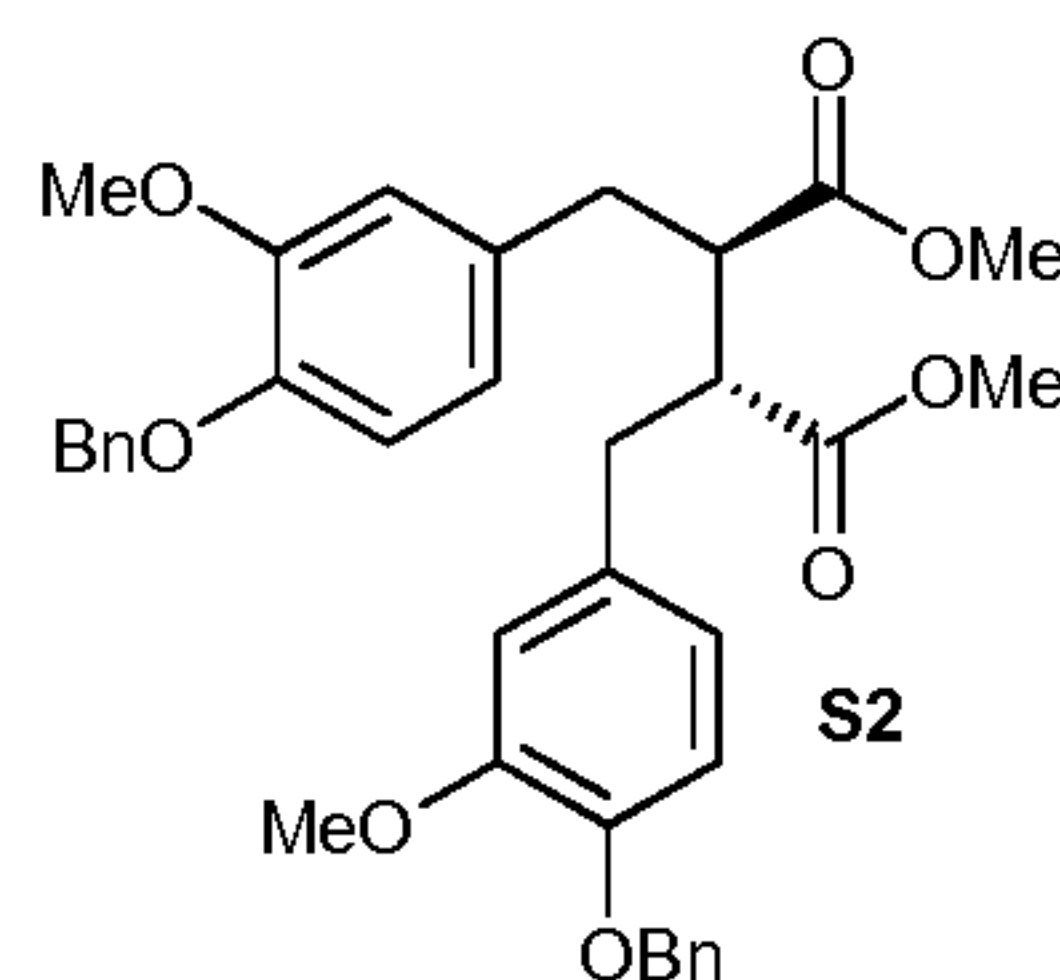
[00137] To a solution of diester **4** (15.00 g, 53.52 mmol, 1 equiv) and vanillin**3** (8.14 g, 53.52 mmol, 1 equiv) in MeOH (200 mL), lithium wire (2.6 g, 374.6 mmol, 7 equiv) was added slowly piecewise with an ice bath to control the exotherm and stirred until fully dissolved. The reaction mixture was then heated at reflux for 48 h. After cooling the solution to room temperature, most of the methanol was removed by concentration on rotavap. EtOAc (200 mL) was added, the solution was acidified with 2 M aq. HCl solution (500 mL), and then extracted with EtOAc (2 × 200 mL). The combined organics were washed with H₂O (2 × 200 mL), then brine (200 mL), and dried (MgSO₄), filtered and concentrated. The residue was dissolved in MeOH (300 mL), conc. H₂SO₄ (1 mL) was added, and the solution was heated at reflux overnight. Upon cooling the next morning, NaHCO₃ (3.0 g) was added to quench

H₂SO₄ and the solution was mostly concentrated by rotavap. EtOAc (400 ml) was added and the solution was washed with H₂O (2 × 200 mL), then brine (100 mL). The organics layer was then dried (MgSO₄), filtered and concentrated. Dissolution in a small amount of CH₂Cl₂ and then flash column chromatography (silica, 2:1:7 → 3:1:6 → 4:1:5 EtOAc:CH₂Cl₂:hexanes) provided **5** (13.6 g, 32.8 mmol, 61% yield, unassigned olefin geometry) as a yellow-orange solid. **5**: *R*_f = 0.28 (silica, EtOAc:hexanes 1:1); IR (film): $\nu_{\max}$ = 3388, 1696, 1588, 1509, 1431, 1209, 1157, 1029, 817, 733 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 7.86 (s, 2 H), 7.11 (d, *J* = 1.64 Hz, 2 H), 7.04 (dd, *J* = 8.48, 1.72 Hz, 2 H), 6.82 (d, *J* = 8.38 Hz, 2 H), 6.08 (s, 2 H), 3.7 (s, 6 H), 3.69 (s, 6 H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ = 168.00, 147.59, 146.54, 142.64, 127.10, 125.38, 124.06, 114.71, 111.52, 55.77, 52.51 ppm; HRMS (ESI-TOF): calcd for C₂₂H₂₂O₈ [M + H⁺]: 415.1387, found 415.1378.

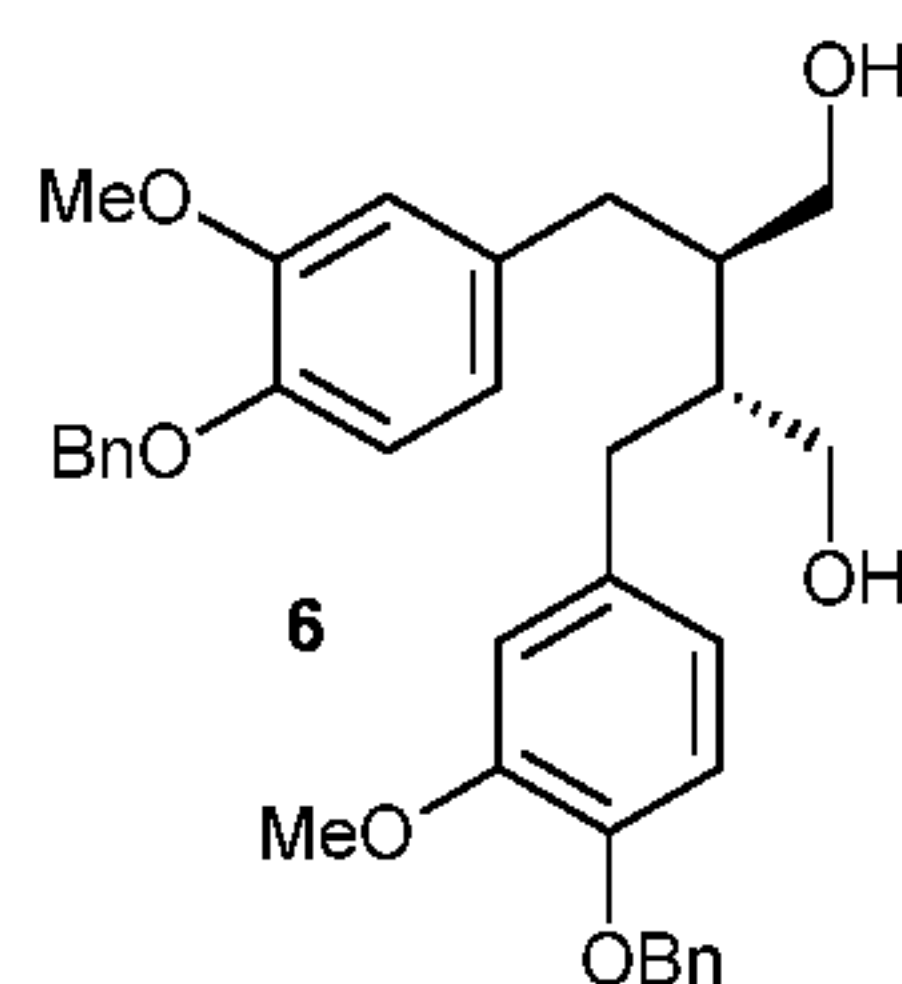
Example 3: Synthesis of Diester S1



[00138] A solution of **5** (5.00 g, 12.07 mmol, 1 equiv) in MeOH (120 mL) was saturated with an argon atmosphere by briefly exposing to vacuum and backfilling with argon several times. Palladium on carbon (10% Pd by weight, 0.500 g) was added and the solution was saturated with H₂ atmosphere by vacuum/backfill H₂. The solution was stirred overnight. The next morning the solution was put under an argon atmosphere, CH₂Cl₂ (400 mL) was added, and the solution was allowed to stir for 1 h. The mixture was then filtered through a pad of Celite (1.5 in, washing with MeOH and CH₂Cl₂), and the filtrate was concentrated. The resulting solid was dissolved in a small amount of CH₂Cl₂ and purified by flash column chromatography (2:8:1 → 3:7:1 → 4:6:1 EtOAc:hexanes:CH₂Cl₂) to give **S1** (4.24 g, 10.1 mmol, 84%) as an off-white solid. **S1**: *R*_f = 0.24 (silica, EtOAc:hexanes:CH₂Cl₂ 4:6:1); IR (film): $\nu_{\max}$ = 3440, 2951, 1726, 1514, 1432, 1267, 1198, 1151, 1121, 1029, 817, 797 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 6.78 (d, *J* = 8.09 Hz, 2 H), 6.58 (d, *J* = 8.09 Hz, 2 H), 6.45 (s, 2 H), 5.65 (s, 2 H), 3.75 (s, 6 H), 3.64 (s, 6 H), 3.00–2.94 (m, 2 H), 2.92–2.82 (m, 4 H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ = 174.03, 146.49, 144.21, 130.41, 121.88, 114.18, 111.26, 55.73, 51.86, 47.67, 35.38 ppm; HRMS (ESI-TOF): calcd for C₂₂H₂₆O₈ [M + H⁺]: 419.17, found 419.1700.

Example 4: Synthesis of Benzyl Ether S2

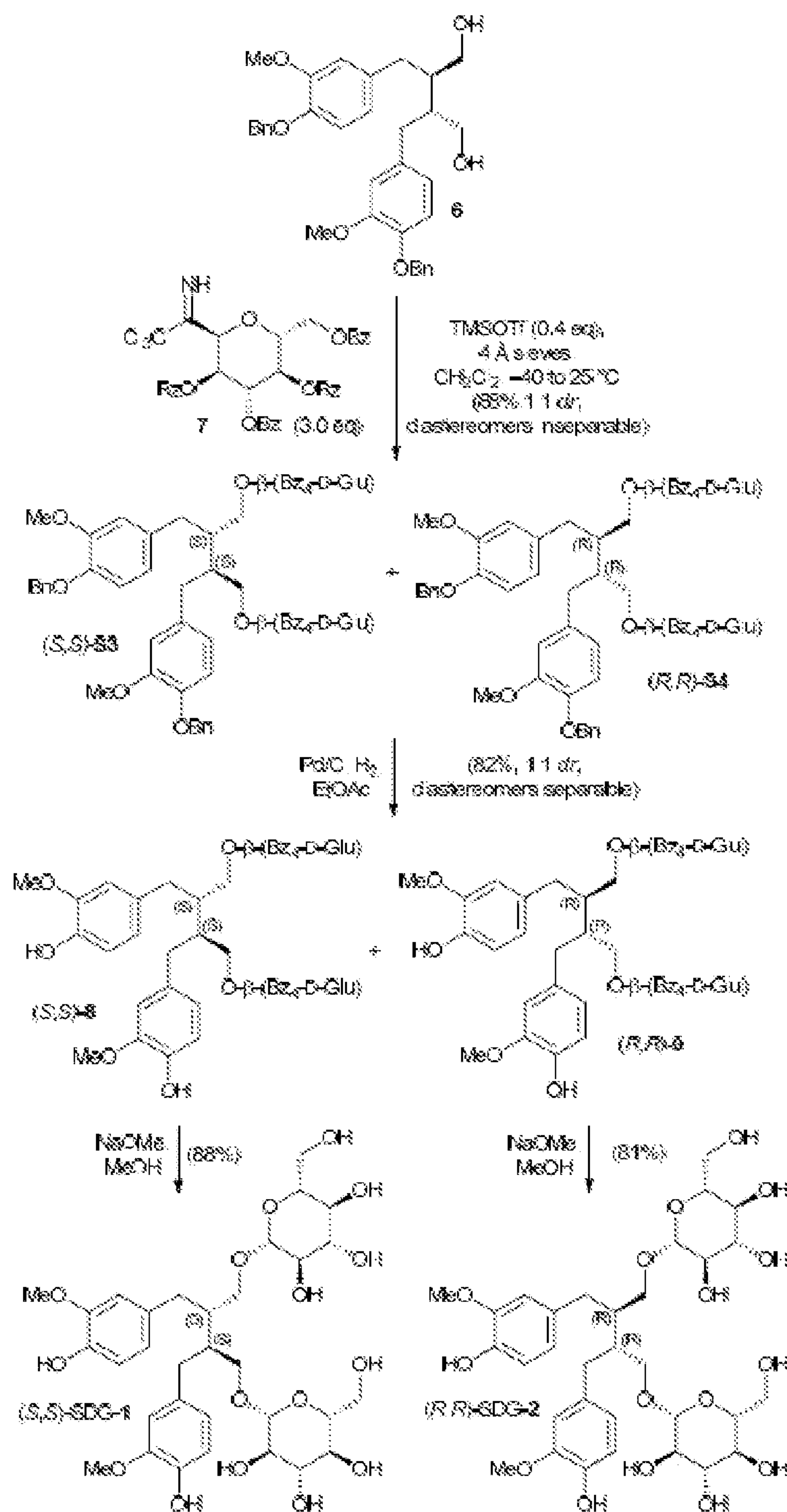
[00139] To a solution of **S1** (1.00 g, 2.39 mmol, 1 equiv) in DMF (24 mL) cooled to 0 °C with an ice bath, NaH (0.201 g, 5.02 mmol, 60% dispersion, 2.1 equiv) was added slowly and the solution was stirred at 0 °C for 1 h. BnBr (910 μ L, 7.65 mmol, 3.2 equiv) was added over 1 min, and the solution was stirred at 0 °C for 4 h. The reaction mixture was then poured into H₂O (300 mL) and EtOAc (100 mL). The organic layer was washed with H₂O (3 $\times$ 200 mL) and brine (1 $\times$ 50 mL), dried (MgSO₄), filtered and concentrated. The resulting solid was dissolved in a small amount of CH₂Cl₂ and purified by flash column chromatography (silica, 9:1:1 $\rightarrow$ 8:2:1 hexanes:EtOAc:CH₂Cl₂) to give **S2** (1.36 g, 2.27, 95% yield) as a white solid. **S2**: R_f = 0.23 (silica, EtOAc:hexanes 3:7); IR (film): $\nu_{\max}$ = 2949, 1730, 1512, 1453, 1253, 1225, 1138, 1023, 733 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 7.45–7.40 (m, 4 H), 7.38–7.33 (m, 4H), 7.31–7.27 (m, 2H), 6.74 (d, J = 8.12 Hz, 2 H), 6.59 (d, J = 1.69 Hz, 2 H), 6.53 (dd, J = 8.12, 1.69 Hz, 2 H), 5.12 (s, 4H), 3.80 (s, 6 H), 3.61 (s, 6 H), 3.03–2.85 (m, 6 H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ = 174.00, 149.58, 146.87, 137.37, 131.74, 128.65, 127.93, 127.37, 121.09, 114.05, 112.73, 71.14, 56.00, 51.91, 47.91, 35.32 ppm; HRMS (ESI-TOF): calcd for C₃₆H₃₈O₈ [M + H⁺]: 599.2639, found 599.2651.

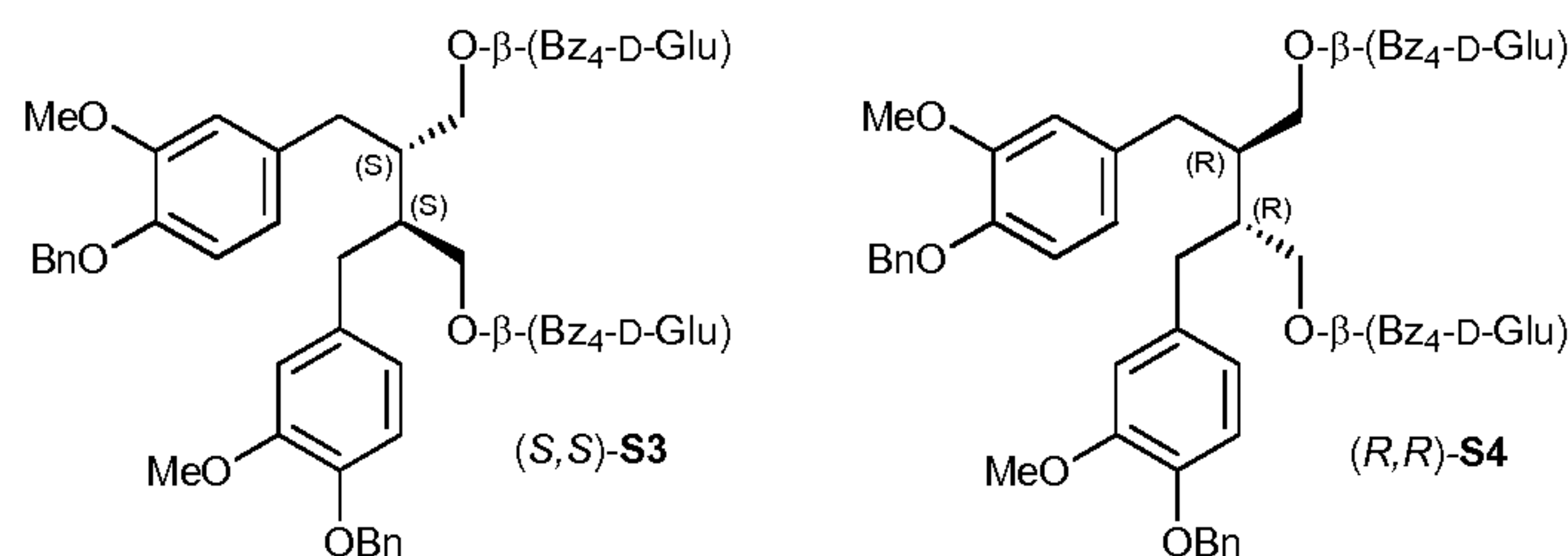
Example 5: Synthesis of Diol 6

[00140] To a solution of **S2** (0.341 g, 0.570 mmol, 1 equiv) in THF (6 mL) at 0 °C, lithium aluminum hydride (1 M soln in THF, 1.14 mL, 1.14 mmol, 2 equiv) was added dropwise. The solution was allowed to come to 25 °C overnight. The resulting mixture was poured into a flask containing H₂O (200 mL) and EtOAc (100 mL). Then 50 mL of an aq. sat. soln of Rochelle's salt was added and the mixture was stirred until the layers were readily separable. The organic layer was then washed with H₂O (2 $\times$ 100 mL) and brine (1 $\times$ 50 mL), and dried

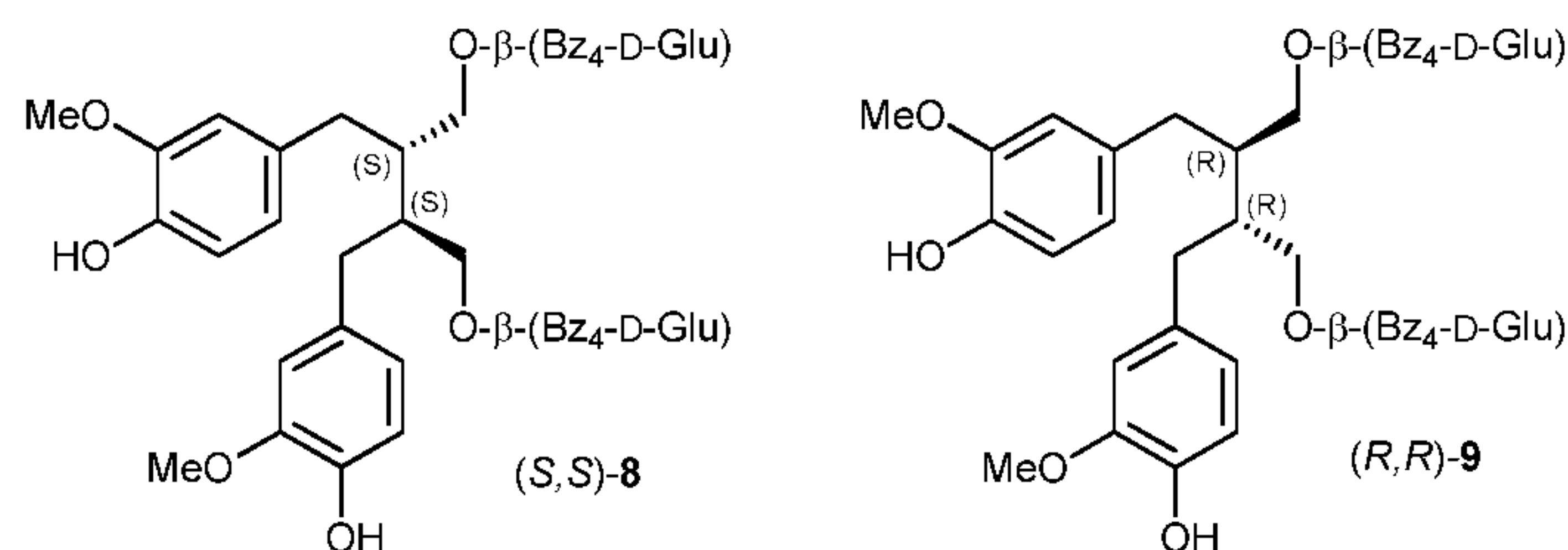
(MgSO₄), filtered and concentrated. The resulting solid was dissolved in a small amount of CH₂Cl₂ and purified by flash column chromatography (silica, 5:5:1 → 7:3:1 → 8:2:1 EtOAc:hexanes:CH₂Cl₂) to give diol **6** (0.286 g, 0.527 mmol, 93% yield) as a white solid. **6**: *R_f* = 0.2 (silica, 6:4 EtOAc:hexanes); IR (film): ν_{max} = 3287, 2933, 1510, 1453, 1259, 1223, 1137, 1009, 733, 695 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 7.47–7.42 (m, 4 H), 7.40–7.34 (m, 4 H), 7.33–7.28 (m, 2 H), 6.78 (d, *J* = 8.15 Hz, 2 H), 6.69 (s, 2 H), 6.62 (d, *J* = 8.15 Hz, 2 H), 5.11 (s, 4 H), 4.09 (s, 2 H), 3.82 (s, 6 H), 3.78 (d, *J* = 11.07 Hz, 2 H), 3.49 (d, *J* = 11.07 Hz, 2 H), 2.80–2.72 (m, 2 H), 2.69–2.62 (m, 2 H), 1.86 (s, 2 H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ 149.47, 146.38, 137.29, 133.87, 128.51, 127.80, 127.28, 121.02, 114.02, 112.79, 71.12, 60.19, 55.96, 43.79, 35.81 ppm; HRMS (ESI-TOF): calcd for C₃₄H₃₈O₆ [M + H⁺]: 543.2741, found 543.2741.

[00141] Scheme 2 shows a reaction route that was used in the synthesis of (*S,S*)-secoisolariciresinol diglucosides and (*R,R*)-secoisolariciresinol diglucosides, starting from compound (**6**).

Scheme 2: Synthesis of ((*S,S*)-SDG-1) and ((*R,R*)-SDG-2) from compound (6)

Example 6: Synthesis of Glycosidated Isomers (S,S)-S3 and (R,R)-S4

[00142] A flask was charged with diol **6** (0.408 g, 0.751 mmol, 1 equiv) and
 5 trichloroacetimidate **7**³ (1.67 g, 2.25 mmol, 3 equiv) and dried by benzene azeotrope (3 × 10 mL). Activated 4 Å molecular sieves (0.800 g) and CH₂Cl₂ (7.5 mL) were added and the solution was stirred for 1 h. After cooling to −40 °C, TMSOTf (54 μL, 0.300 mmol, 0.4 equiv) was added dropwise and the reaction mixture was allowed to warm to 25 °C overnight. The next morning, NEt₃ (200 μL) was added and the mixture was filtered through a silica pad
 10 (1 in, washing with EtOAc) and concentrated. The resulting crude was purified by flash column chromatography (silica, 9:1:1 → 8:2:1 → 7:2:2 hexanes:EtOAc:CH₂Cl₂) to give a 1:1 inseparable mixture of diastereomers (S,S)-**S3** and (R,R)-**S4** (1.128 g, 0.664 mmol, 88% yield). (S,S)-**S3**/(R,R)-**S4**: *R*_f = 0.22 (silica, 7:2:1 hexanes:EtOAc:CH₂Cl₂); IR (film): *v*_{max} = 2941, 1727, 1601, 1511, 1451, 1262, 1092, 1026, 708 cm^{−1}; ¹H NMR (600 MHz, CDCl₃) *δ* =
 15 8.04 (d, *J* = 7.70 Hz, 4H), 8.01 (d, *J* = 7.62 Hz, 4 H), 7.98–7.91 (m, 12 H), 7.87–7.81 (m, 12 H), 7.57–7.26 (m, 68 H), 6.61 (d, *J* = 8.26 Hz, 2 H), 6.49 (s, 2 H), 6.44 (d, *J* = 8.0 Hz, 2 H), 6.42–6.38 (m, 4 H), 6.23 (d, *J* = 8.14 Hz, 2 H), 5.93 (t, *J* = 10.07 Hz, 2 H), 5.81 (t, *J* = 9.68 Hz, 2 H), 5.72 (t, *J* = 10.07 Hz, 2 H), 5.67 (t, *J* = 9.68 Hz, 2 H), 5.56 (t, *J* = 7.72 Hz, 2 H), 5.45 (t, *J* = 8.24 Hz, 2 H), 5.08 (s, 8 H), 4.72 (dd, *J* = 12.19, 2.70 Hz, 2 H), 4.66 (d, *J* = 8.22
 20 Hz, 2 H), 4.61 (dd, *J* = 12.19, 3.08 Hz, 2 H), 4.49–4.42 (m, 4 H), 4.39 (d, *J* = 7.70 Hz, 2 H), 4.14–4.08 (m, 2 H), 3.98–3.93 (m, 2 H), 3.92–3.87 (m, 2 H), 3.74 (s, 6 H), 3.70 (s, 6 H), 3.65–3.61 (m, 2 H), 3.40–3.34 (m, 2 H), 3.24–3.18 (m, 2 H), 2.56–2.45 (m, 6 H), 2.41–2.34 (m, 2 H), 1.85 (s, 2 H), 1.73 (s, 2 H) ppm; ¹³C NMR (150 MHz, CDCl₃) *δ* = 166.20, 166.19, 165.93, 165.90, 165.29, 165.16, 164.98, 149.38, 149.35, 146.32, 146.20, 137.56, 133.73, 133.66, 133.56, 133.53, 133.45, 133.37, 133.34, 133.28, 129.93, 129.91, 129.86, 129.84, 129.76, 129.66, 129.64, 129.38, 129.30, 128.95, 128.92, 128.91, 128.60, 128.58, 128.55, 128.53, 128.42, 127.83, 127.81, 127.39, 127.32, 121.38, 121.19, 113.61, 112.70, 112.51, 101.31, 101.23, 73.06, 72.93, 72.12, 72.10, 72.06, 71.99, 71.05, 71.00, 69.89, 69.81, 69.61, 69.32, 63.07, 62.98, 55.94, 55.93, 41.00, 40.78, 35.35, 35.14 ppm; HRMS (ESI-TOF): calcd for
 25 C₁₀₂H₉₀O₂₄ [M + H⁺]: 1699.5895, found 1699.5917.

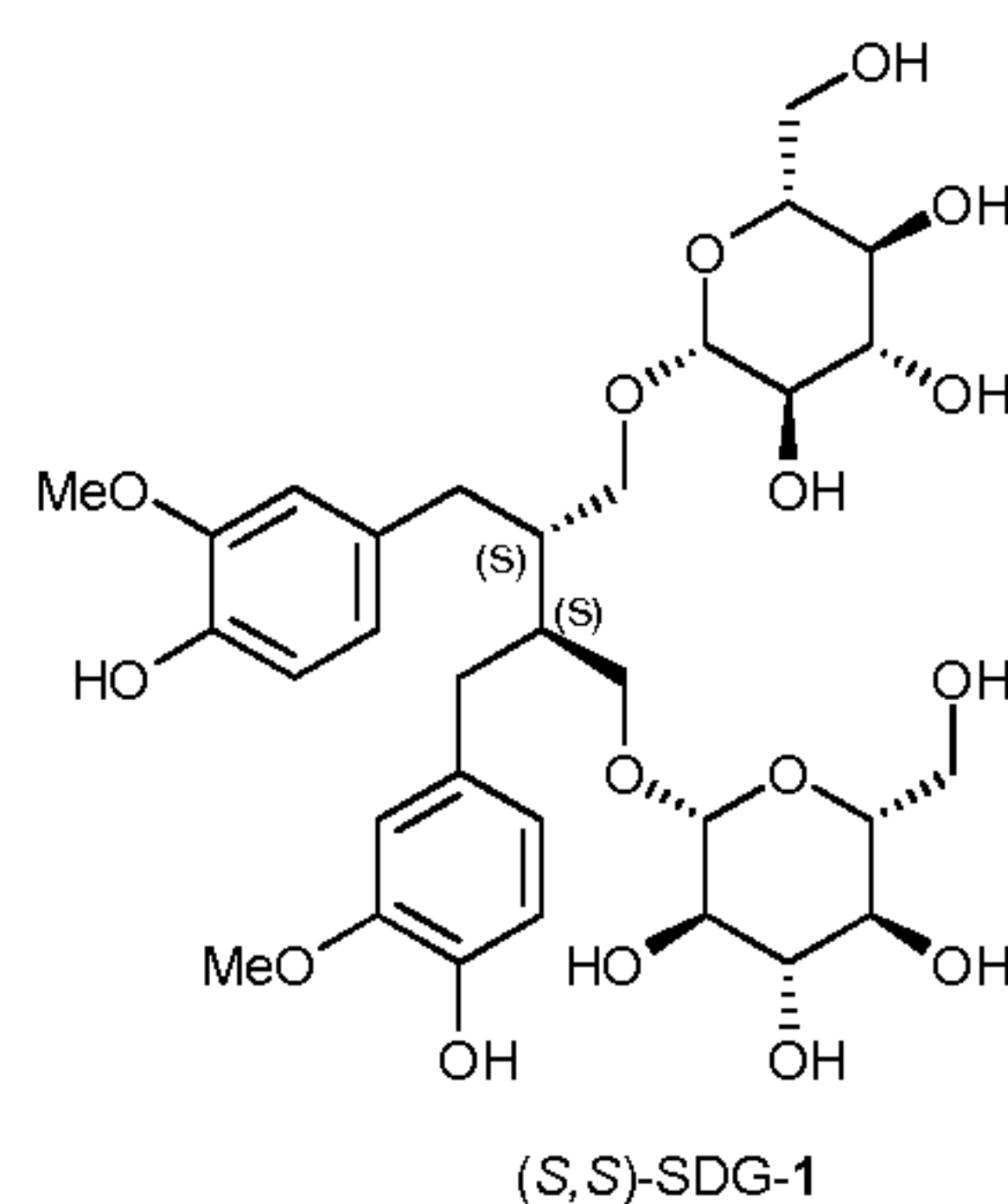
Example 7: Phenols(*S,S*)-8 and (*R,R*)-9

5 [00143] The 1:1 mixture of (*S,S*)-**S3** and (*R,R*)-**S4** (0.612 g, 0.360 mmol, 1 equiv) in EtOAc (3.6 mL) was saturated with an argon atmosphere by briefly exposing to vacuum and backfilling with argon several times. Palladium on carbon (10% Pd by weight, 0.120 g) was added and the solution was saturated with H₂ by vacuum/backfill H₂. After stirring at 25 °C for 36 h, the solution was put under an argon atmosphere, filtered through a pad of Celite (1.5
 10 in, washing with EtOAc and CH₂Cl₂), and the filtrate concentrated. The resulting solid was dissolved in a small amount of CH₂Cl₂ and purified by flash column chromatography (silica, 3:7 → 4:6 → 5:5 EtOAc:hexanes) to give the debenzylated glycoside (0.470 g, 0.309 mmol, 86% yield) as a 1:1 mixture of diastereomers. The diastereomers could be separated by preparative thin-layer chromatography (silica, 2 mm, multiple plates, 7:20 EtOAc:hexanes,
 15 >10 elution runs) to give (*S,S*)-**8** and (*R,R*)-**9** as off-white solids. (*S,S*)-**8**: *R*_f = 0.10 (4:6 EtOAc:hexanes); [*α*]_D³² = +1.2 (EtOAc, *c* = 4.3); IR (film): *v*_{max} = 3460, 2938, 1729, 1514, 1451, 1265, 1023, 1061, 1027, 709 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) *δ* = 8.03 (d, *J* = 8.42 Hz, 4 H), 7.92 (d, *J* = 8.22 Hz, 4 H), 7.83 (d, *J* = 7.48 Hz, 8 H), 7.58–7.46 (m, 6 H), 7.45–7.27 (m, 18 H), 6.62 (d, *J* = 7.90 Hz, 2 H), 6.43 (dd, *J* = 7.90, 1.16 Hz, 2 H), 6.38 (d, *J* = 1.16 Hz, 2
 20 H), 5.80 (t, *J* = 9.57 Hz, 2 H), 5.64 (t, *J* = 9.57 Hz, 2 H), 5.44 (t, *J* = 9.74 Hz, 2 H), 5.40 (s, 2 H), 4.62 (dd, *J* = 12.09, 2.98 Hz, 2 H), 4.44 (dd, *J* = 12.09, 5.17 Hz, 2 H), 4.41 (d, *J* = 8.03 Hz, 2 H), 3.96 (m, 2 H), 3.69 (s, 6 H), 3.64 (dd, *J* = 9.38, 2.82 Hz, 2 H), 3.21 (dd, *J* = 9.43, 4.13 Hz, 2 H), 2.48 (d, *J* = 6.8 Hz, 4 H), 1.71 (s, 2 H) ppm; ¹³C NMR (150 MHz, CDCl₃) *δ* = 166.26, 165.92, 165.35, 165.05, 146.27, 143.64, 133.60, 133.49, 133.38, 133.37, 132.41,
 25 129.95, 129.88, 129.85, 129.78, 129.67, 129.30, 128.94, 128.60, 128.57, 128.45, 122.01, 113.83, 111.49, 101.37, 72.96, 72.15, 72.00, 69.89, 69.54, 63.16, 55.82, 40.87, 35.26 ppm; HRMS (ESI-TOF): calcd for C₈₈H₇₈O₂₄ [*M* + H⁺]: 1519.4956, found 1519.4937. (*R,R*)-**9**: *R*_f = 0.10 (4:6 EtOAc:hexanes); [*α*]_D³² = +4.8 (EtOAc, *c* = 4.1); IR (film): *v*_{max} = 3457, 2942, 1726, 1262, 1026, 707 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) *δ* = 8.00 (d, *J* = 7.62 Hz, 4 H), 7.97 (d, *J* = 7.71 Hz, 4 H), 7.93 (d, *J* = 7.98 Hz, 4 H), 7.62 (d, *J* = 7.71 Hz, 4 H), 7.56–7.27 (m, 24 H),
 30

6.45 (d, $J = 7.95$ Hz, 2 H), 6.29 (s, 2 H), 6.24 (d, $J = 7.95$ Hz, 2 H), 5.91 (t, $J = 9.82$ Hz, 2 H), 5.70 (t, $J = 9.67$ Hz, 2 H), 5.54 (t, $J = 9.23$ Hz, 2 H), 5.40 (s, 2 H), 4.71 (dd, $J = 11.87, 3.29$ Hz, 2 H), 4.62 (d, $J = 7.92$ Hz, 2 H), 4.42 (dd, $J = 12.14, 4.72$ Hz, 2 H), 4.11–4.06 (m, 2 H), 3.85 (dd, $J = 9.53, 3.46$ Hz, 2 H), 3.65 (s, 6 H), 3.34 (dd, $J = 9.70, 4.68$ Hz, 2 H), 2.47–2.40 (m, 2 H), 2.37–2.30 (m, 2 H), 1.82 (s, 2 H) ppm; ^{13}C NMR (150 MHz, CDCl_3) $\delta = 166.23, 165.95, 165.34, 165.21, 146.33, 143.54, 133.56, 133.50, 133.36, 133.30, 132.44, 129.96, 129.92, 129.87, 129.65, 129.42, 128.98, 128.66, 128.55, 128.44, 121.98, 113.66, 111.12, 101.25, 73.07, 72.15, 72.08, 69.88, 69.78, 62.97, 55.78, 40.67, 35.50$ ppm; HRMS (ESI-TOF): calcd for $\text{C}_{36}\text{H}_{38}\text{O}_8[\text{M} + \text{H}^+]$: 1519.4956, found 1519.4947.

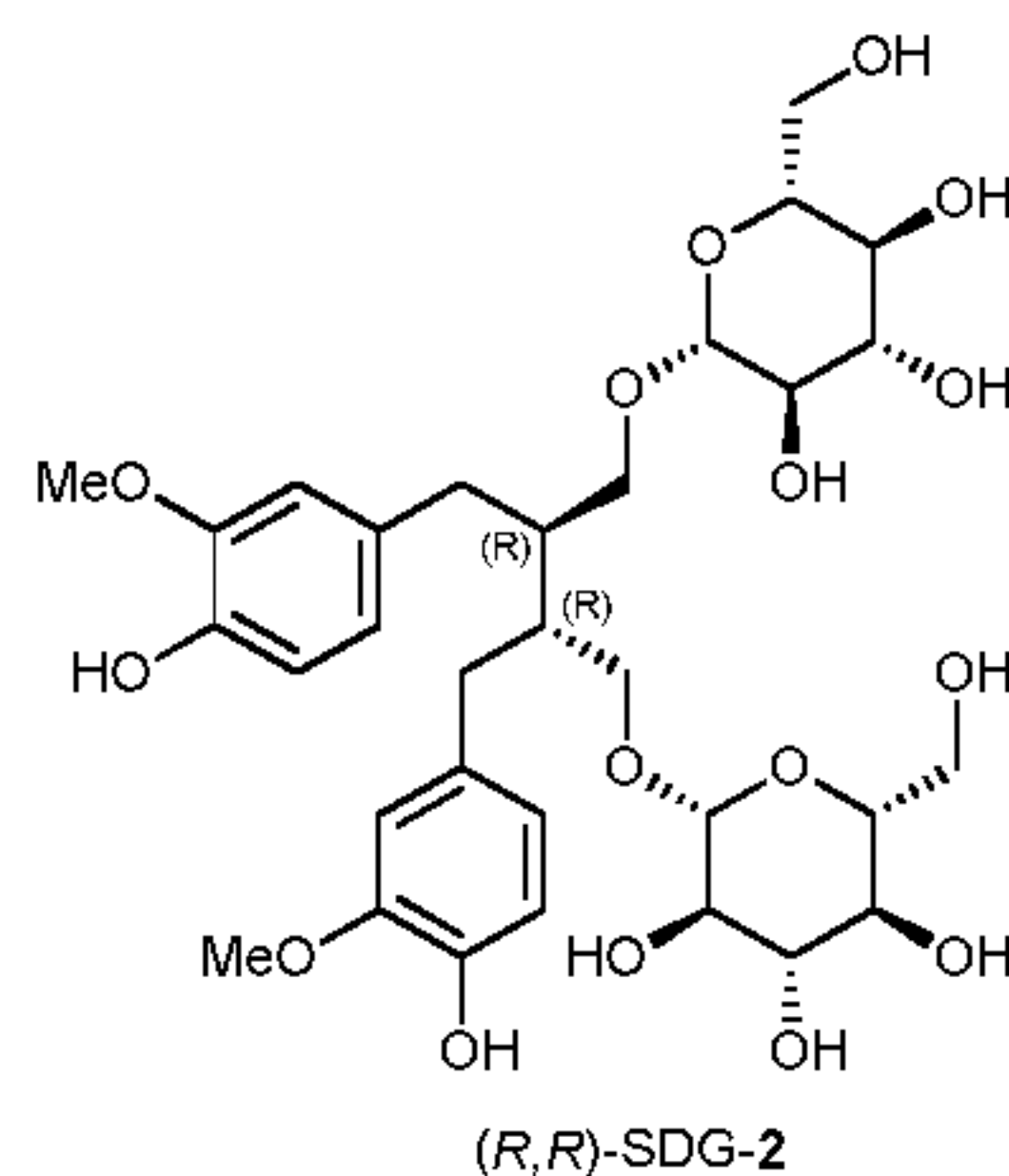
10

Example 8: Synthesis of Secoisolariciresinol Diglucoside(*S,S*)-SDG-1



[00144] To a flask containing dry (*S,S*)-**8** (0.043 g, 0.028 mmol, 1 equiv), a freshly prepared solution of NaOMe in MeOH (0.4 M, 2 mL, 28 equiv) was added and the solution was stirred for 60 h at 25 °C. The solution was then filtered through a pad of silica (0.5 in, washing with MeOH) and the filtrate was concentrated. The resulting solid was purified by preparative thin-layer chromatography (silica, 2 mm, 9:1 $\rightarrow$ 7:3 CH_2Cl_2 :MeOH, then 5:5 CH_2Cl_2 :MeOH half of plate length) and then passed through a small plug of reversed phase silica (100 Å, C_{18}) to provide (*S,S*)-SDG-1 (0.017 g, 0.025 mmol, 88% yield) as an off-white solid. (*S,S*)-SDG-1: $R_f = 0.57$ (silica, 1:1 CH_2Cl_2 :MeOH); $[\alpha]_D^{32} = -0.3$ (MeOH, $c = 1.2$); IR (film): $\nu_{\text{max}} = 3340, 2950, 1601, 1515, 1372, 1270, 1070, 1015, 798$ cm^{-1} ; ^1H NMR (600 MHz, CDCl_3) $\delta = 6.65$ (d, $J = 8.05$ Hz, 2 H), 6.59 (d, $J = 1.31$ Hz, 2 H), 6.56 (dd, $J = 8.05, 1.31$ Hz, 2 H), 4.24 (d, $J = 7.42$ Hz, 2 H), 4.08 (dd, $J = 10.09, 5.58$ Hz, 2 H), 3.85 (dd, $J = 12.00, 2.43$ Hz, 2 H), 3.73 (s, 6 H), 3.69 (dd, $J = 11.85, 5.55$ Hz, 2 H), 3.50–3.45 (m, 2 H), 3.38–3.28 (m, 4 H), 3.27–3.19 (m, 4 H), 2.69 (dd, $J = 13.82, 6.72$ Hz, 2 H), 2.61 (dd, $J = 13.82, 7.98$ Hz, 2 H), 2.12 (m, 2 H) ppm; ^{13}C NMR (150 MHz, CDCl_3) $\delta = 148.69, 145.35, 133.93, 122.89, 115.63, 113.51, 104.77, 78.16, 77.89, 75.25, 71.69, 71.18, 62.79, 56.25, 41.20, 35.60$ ppm; HRMS (ESI-TOF): calcd for $\text{C}_{32}\text{H}_{46}\text{O}_{16}[\text{M} + \text{H}^+]$: 687.2858, found 687.2856.

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Example 9: Synthesis of Secoisolariciresinol Diglucoside(*R,R*)-SDG-2

[00145] To a flask containing dry (*R,R*)-**9** (0.041 g, 0.027 mmol, 1 equiv), a freshly prepared solution of NaOMe in MeOH (0.4 M, 2 mL, 28 equiv) was added and the solution stirred for 60 h at 25 °C. The solution was then filtered through a pad of silica (0.5 in, washing with MeOH) and the filtrate concentrated. The resulting solid was purified by preparative thin-layer chromatography (silica, 2 mm, 9:1 → 7:3 CH₂Cl₂:MeOH, then 5:5 CH₂Cl₂:MeOH half of plate length) and then passed through a small plug of reversed phase silica (100 Å C₁₈, washing with MeOH) to provide (*R,R*)-SDG-**2** (0.015 g, 0.022 mmol, 81% yield) as an off-white solid. (*R,R*)-SDG-**2**: *R*_f = 0.50 (silica, 1:1 CH₂Cl₂:MeOH); [α]_D³² = -22.2 (MeOH, *c* = 1.0); IR (film): ν_{max} = 3336, 2949, 1651, 1409, 1014 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ = 6.66 (d, *J* = 8.06 Hz, 2 H), 6.64 (d, *J* = 1.63 Hz, 2 H), 6.59 (dd, *J* = 8.06, 1.63 Hz, 2 H), 4.21 (d, *J* = 7.82 Hz, 2 H), 3.91 (dd, *J* = 10.11, 5.69 Hz, 2 H), 3.87 (dd, *J* = 12.01, 2.01 Hz, 2 H), 3.75 (s, 6 H), 3.67 (dd, *J* = 12.02, 5.47 Hz, 2 H), 3.58 (dd, *J* = 9.90, 5.36 Hz, 2 H), 3.38–3.18 (m, 8 H), 2.76–2.62 (m, 4 H), 2.14–2.07 (m, 2 H) ppm; ¹³C NMR (150 MHz, CDCl₃) δ = 148.77, 145.37, 134.05, 122.83, 115.70, 113.56, 104.59, 78.19, 77.93, 75.19, 71.70, 70.62, 62.79, 56.32, 41.63, 35.62 ppm; HRMS (ESI-TOF): calcd for C₃₂H₄₆O₁₆ [M + H⁺]: 687.2858, found 687.2856.

Methods for Biological Assays**Assay Kits**

[00146] HORAC Assay kit (#TA30) was purchased from Oxford Biomedical Research. ORAC Assay kit (#STA345) was obtained from Cell Biolab, San Diego, CA. SDG standard was purchased from Chromadex Inc., San Diego, CA.

Reducing Power Activity Assay

[00147] The determination of reducing power was performed as described by Yen and Der (*J. Am. Oil Chem. Soc.* **1993**, 70, 383). Reducing power assay determines the reducing potential

of the test compound which reacts with potassium ferricyanide (Fe^{+3}) to form potassium ferrocyanide (Fe^{+2}), which subsequently reacts with ferric chloride to produce a ferric-ferrous complex that has maximum absorption at 700 nm. Various concentrations (1-500 μM) of test compounds were taken in sodium phosphate buffer (0.1 M, pH 6.6) in 96-well microplates and mixed with potassium ferricyanide (1%). Samples were incubated at 50°C and equal volume of 10% trichloroacetic acid was added. The upper layer was mixed with deionized water (1:1:2) and ferric chloride (0.1%). The absorbance was read at 700 nm on a Bio-Rad microplate reader (Bio-Rad, Hercules, CA). The increase in absorbance indicates increase in reducing power.

10 **Hydroxyl Radical Scavenging Potential (HORAC Assay)**

[00148] The ability of SDGs to scavenge hydroxyl radicals in a chemical system was evaluated using HORAC Assay kit (#TA30) obtained from Oxford Biomedical Research. Hydroxyl radicals were generated from hydrogen peroxide by Fenton reaction. Oxidation of fluorescein was measured on a fluorescence microplate reader. Antioxidants inhibit fluorescein oxidation. Gallic acid was used as a standard for calibration curve. Calculations used SDG concentration that fit the linear part of the calibration curve. SDG concentrations were used in the range of 8 μM -1mM. Antioxidant capacity against hydroxyl radicals was expressed as gallic acid equivalent (GAE).

Peroxyl Radical Scavenging Potential (ORAC ASSAY)

20 [00149] The ability of SDGs to scavenge peroxyl radicals in a chemical system was evaluated using an ORAC assay kit (#STA345) obtained from Cell Biolab (San Diego, CA). Peroxyl radicals were generated by AAPH (2,2'-azobis(2-amidinopropanedihydrochloride)). Oxidation of fluorescein was measured using a fluorescence microplate reader. Antioxidants inhibit fluorescein oxidation. Trolox was used as a standard for calibration curve. Calculations used SDG concentrations that fit the linear part of the calibration curve. SDG concentrations were used in the range of 8 μM -1mM. Antioxidant capacity against peroxyl radicals was expressed as Trolox equivalent (TE).

DPPH Radical Scavenging Assay.

30 [00150] The ability of the SDGs to scavenge DPPH radicals was assessed as described by Moree et al. (*Free Rad. Antiox.* **2011**, 1, 31) with minor modifications for use in microplates. Briefly, different concentrations of SDG isomers and other test compounds were incubated with 200 μL medium in 96-well microplates containing 0.1 M Tris buffer (pH 7.4) and 250 μM DPPH solution, and kept in the dark for 20 min. The absorbance was read at 517 nm

in a Bio-Rad microplate reader. Ascorbic acid and α -tocopherol were used as known antioxidants for comparison. The radical scavenging activity was measured as a decrease in the absorbance of DPPH and calculated using the following equation: Percentage inhibition = $[\text{O.D.}_{\text{control}} - \text{O.D.}_{\text{treated}} / \text{O.D.}_{\text{control}}] \times 100$.

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Example 10: Reducing Powers of (*S,S*)-SDG-1 and (*R,R*)-SDG-2

[00151] The reducing powers of synthetic (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, natural (*S,S*)-SDG-1, ascorbic acid, and α -tocopherol were determined by the reduction of $\text{K}_3\text{Fe}(\text{CN})_6$ in the presence of FeCl_3 , as measured by the absorbance of the resulting ferric-ferrous complex (Figure 2). The reducing power of synthetic (*S,S*)-SDG-1, synthetic (*R,R*)-SDG-2, and natural (*S,S*)-SDG-1 were significantly concentration dependent at higher concentrations; however, at all concentrations tested the SDGs had comparable or higher reducing power than known antioxidants ascorbic acid and α -tocopherol, with a notable increase in potency in the 200–500 μM range. A linear relationship between reducing power and substrate concentration was observed at lower concentrations (1–100 μM), allowing regression line equations to be established for the five compounds. This allowed the half maximal effective concentration (EC_{50}) for reducing power to be calculated (Figure 3). The EC_{50} (mean $\pm$ std. dev.) values for (*S,S*)-SDG-1 and (*R,R*)-SDG-2 were $292.17 \pm 27.71 \mu\text{M}$ and $331.94 \pm 21.21 \mu\text{M}$, respectively. These values were comparable to that of natural (*S,S*)-SDG-1 ($\text{EC}_{50} = 275.24 \pm 13.15 \mu\text{M}$) but approximately three-fold higher than that exhibited by ascorbic acid ($\text{EC}_{50} = 1129.32 \pm 88.79 \mu\text{M}$) and α -tocopherol ($\text{EC}_{50} = 944.62 \pm 148.00 \mu\text{M}$).

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Example 11: The Ability to Scavenge Hydroxyl and Peroxyl Radicals

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[00152] The ability of synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2 to scavenge hydroxyl and peroxyl radicals as manifested by their inhibition of the oxidation of fluorescein was assessed by the hydroxyl radical averting capacity (HORAC, gallic acid standard) and peroxyl radical absorbance capacity assays (ORAC, trolox standard), respectively (Table 1). Fluorescein oxidation by hydroxyl radicals was decreased by synthetic (*S,S*)-SDG-1 and synthetic (*R,R*)-SDG-2 in a concentration-dependent manner and was found to be two-fold higher than gallic acid. However, synthetic (*S,S*)-SDG-1 activity differed from natural (*S,S*)-SDG-1, likely due to trace impurities. Fluorescein oxidation by peroxyl radicals generated using 2,2'-azobis(2-amidinopropane) dihydrochloride (AAPH) was greatly reduced in the presence of synthetic

(*S,S*)-SDG-1, (*R,R*)-SDG-2 and natural (*S,S*)-SDG-1, with a two-fold increase in potency over the trolox standard (Table 1).

Table 1: Antioxidant Capacity of Synthetic and Natural SDGs

Entry	Antioxidant	Against Hydroxyl ^a Radicals (GAE) ^c	Against Peroxyl ^b Radicals (TE) ^d
1.	natural (<i>S,S</i>)-SDG-1	3.68±0.27	2.55±0.11
2.	synthetic (<i>R,R</i>)-SDG-2	1.96±0.27	2.20±0.10
3.	synthetic (<i>S,S</i>)-SDG-1	2.09±0.16	3.03±0.04

^a Determined by HORAC Assay; ^b Determined by ORAC Assay; ^c Gallic acid equivalent; ^d Trolox equivalents.

[00153] In Table 1, hydroxyl radicals were generated from hydrogen peroxide by Fenton reaction. Peroxyl radicals were generated by AAPH (2,2'-azobis(2-amidinopropane) dihydrochloride). Oxidation of fluorescein was measured. Calculations used SDG concentrations that fitted the linear part of the calibration curve. The results are presented as mean ± standard deviation (n = 3).

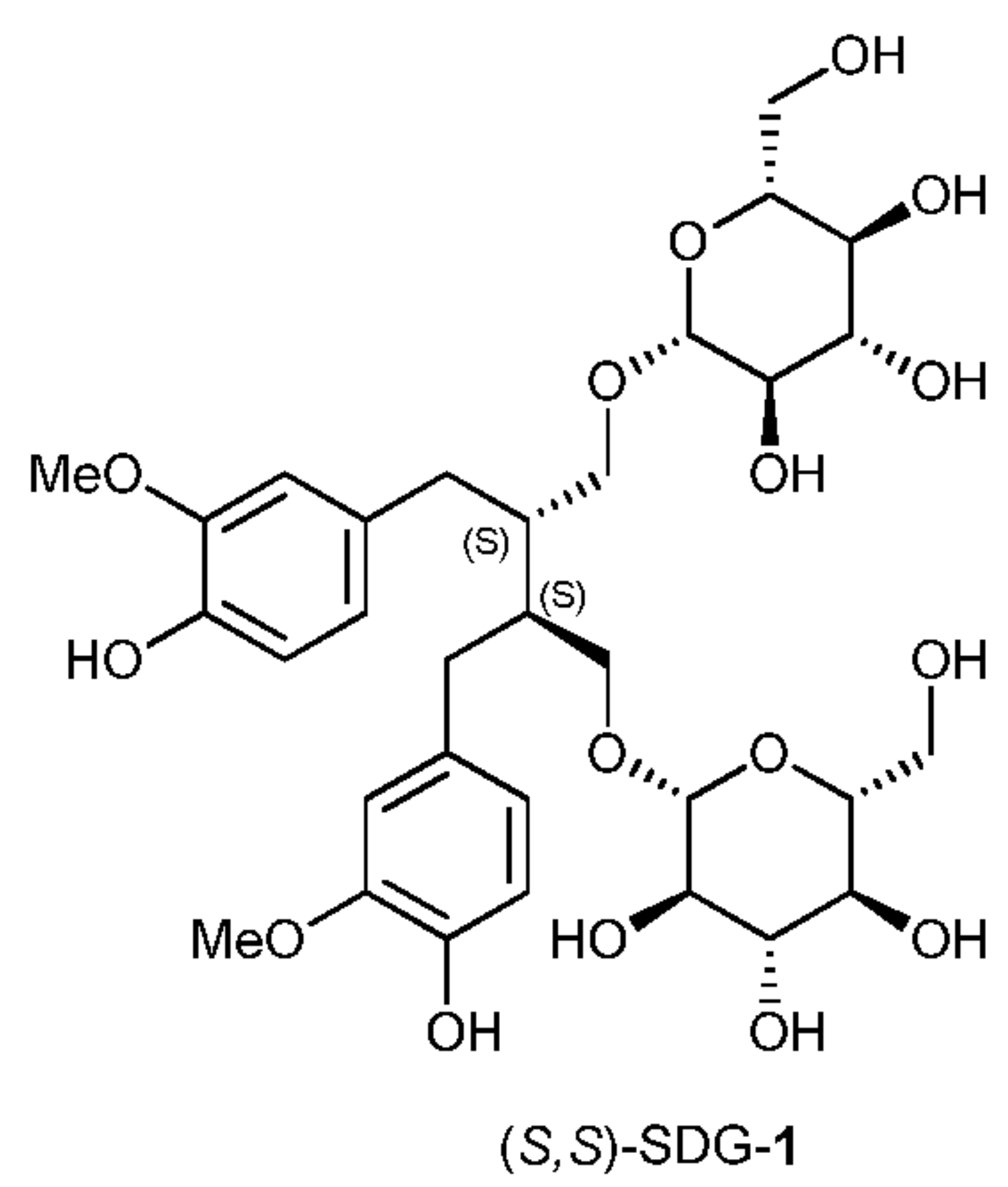
Example 12: Free Radical Scavenging Activities of (*S,S*)-SDG-1 and (*R,R*)-SDG-2

[00154] The free radical scavenging activities of synthetic (*S,S*)-SDG-1 and (*R,R*)-SDG-2 were determined using a 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging assay and was compared to those of natural (*S,S*)-SDG-1, ascorbic acid, and α-tocopherol (Figure 4). At low (5-25 μM) and mid-concentration (50–100 μM) ranges, the SDGs exhibited similar scavenging potentials; however at higher concentrations (250–500 μM), the inhibition by synthetic (*S,S*)-SDG-1 was significantly lower than those exerted by (*R,R*)-SDG-2 and natural (*S,S*)-SDG-1. Establishing regression lines for the potentials at low- and mid- concentration ranges (5–100 μM) allowed the free radical EC₅₀ scavenging activity of these compounds to be determined. As shown in Figure 5, synthetic (*R,R*)-SDG-2 (123.63 ± 8.67 μM) and synthetic (*S,S*)-SDG-1 (157.54 ± 21.3 μM) were not significantly different. These values were similar to those exhibited by natural (*S,S*)-SDG-1 (83.94± 2.80 μM) and α-tocopherol (132.81 ± 12.57 μM) but considerably lower than that shown by ascorbic acid (439.56 ± 11.81 μM). These results are comparable to those reported for SDG (Moree, S.; Khanum, S.A.; Rajesha, J. *Free Rad. Antiox.* **2011**, 1, 31).

[00155] It will be appreciated by those skilled in the art that changes could be made to the embodiments described above without departing from the broad inventive concept thereof. It is understood, therefore, that this invention is not limited to the particular embodiments disclosed, but it is intended to cover modifications that are within the spirit and scope of the
5 invention, as defined by the appended claims.

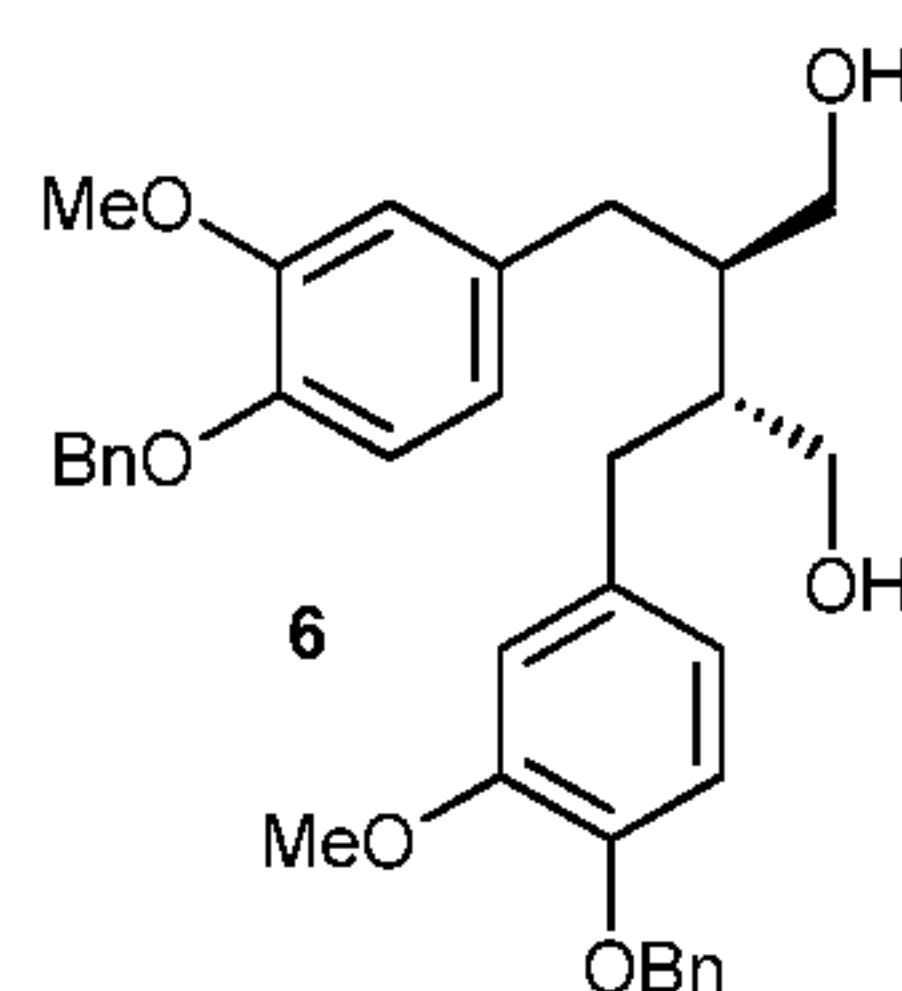
WHAT IS CLAIMED IS:

1. A process for preparing a compound of formula ((*S,S*)-SDG-1)

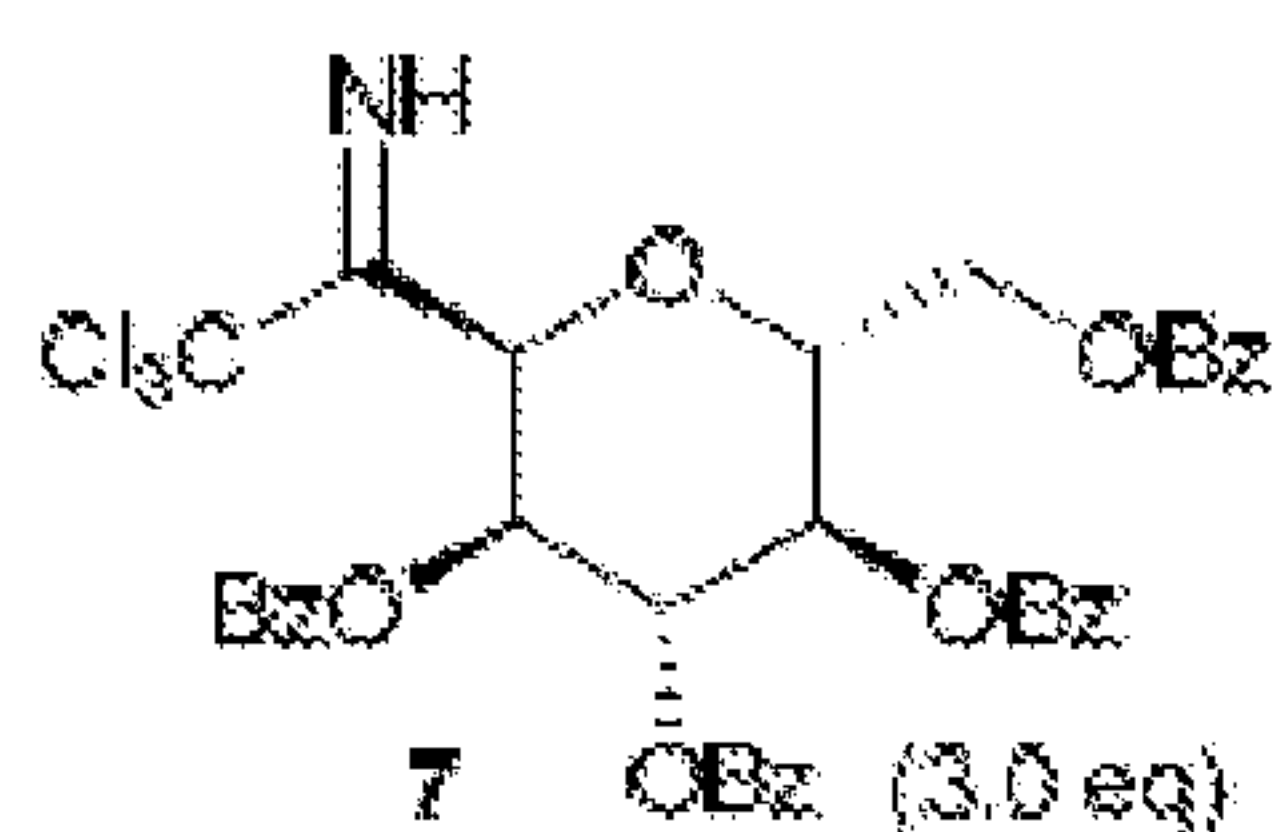


the process comprising:

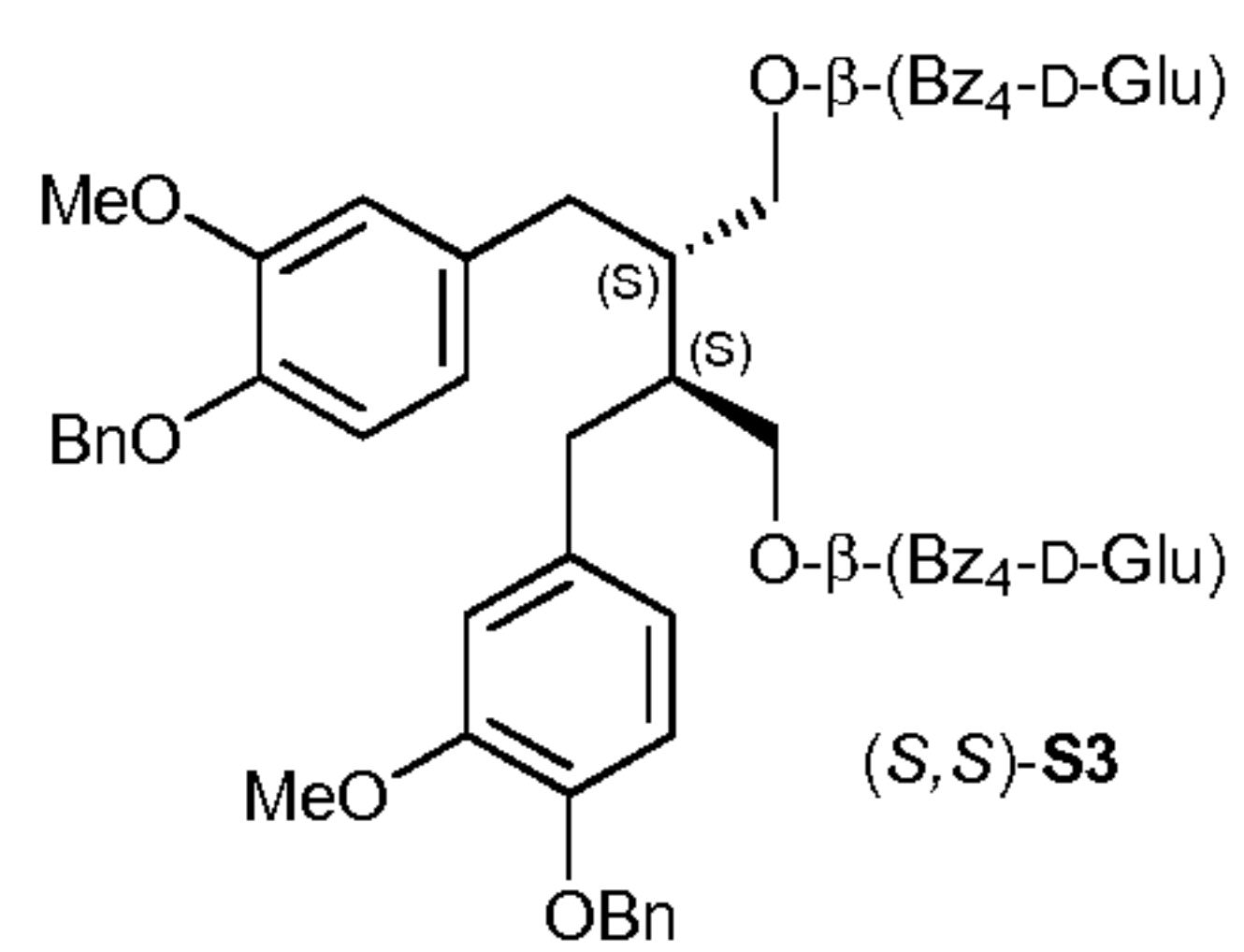
- (a) reacting a compound of formula (6)



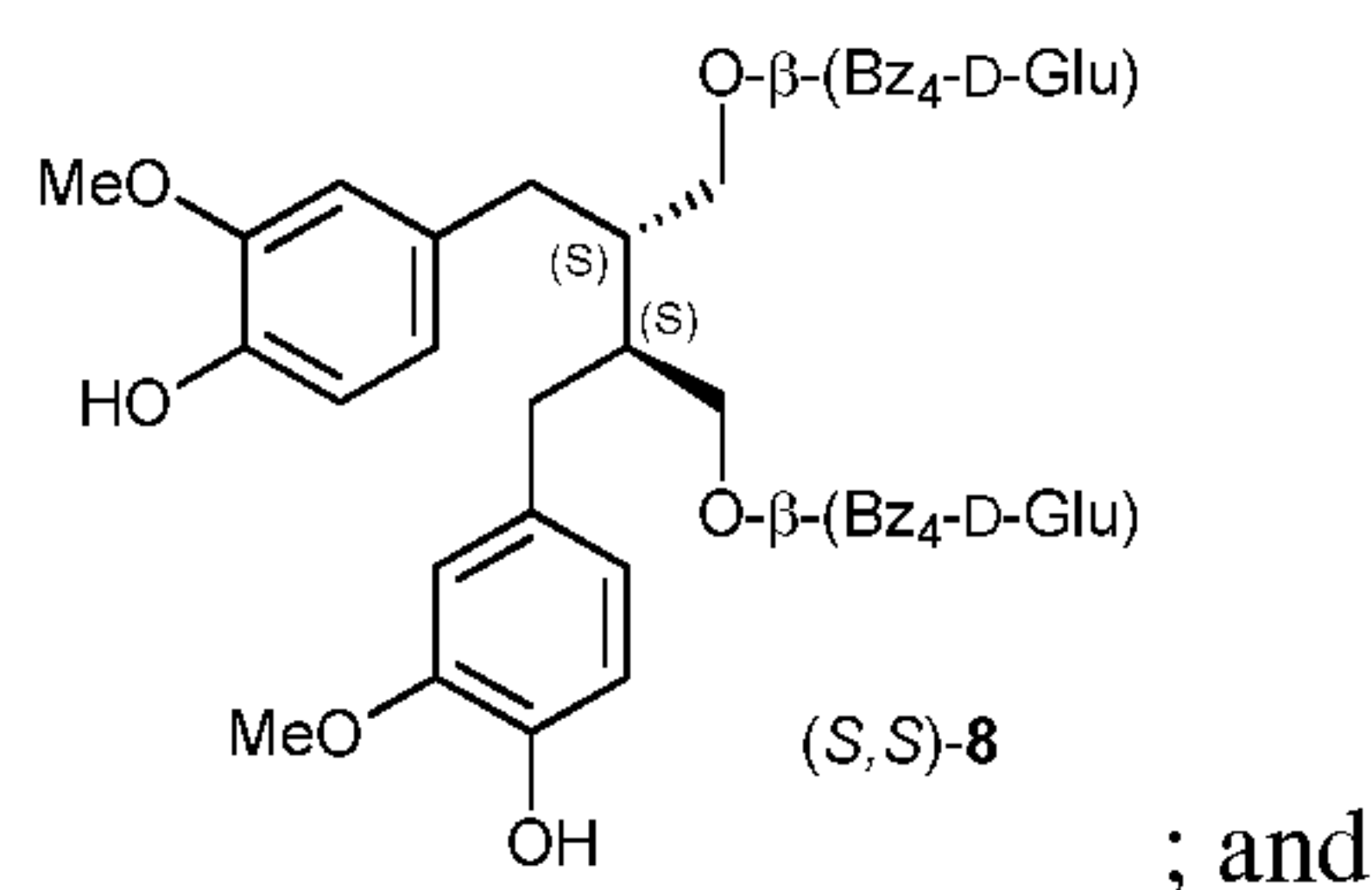
with a compound of formula (7)



to prepare a compound of formula ((*S,S*)-S3)

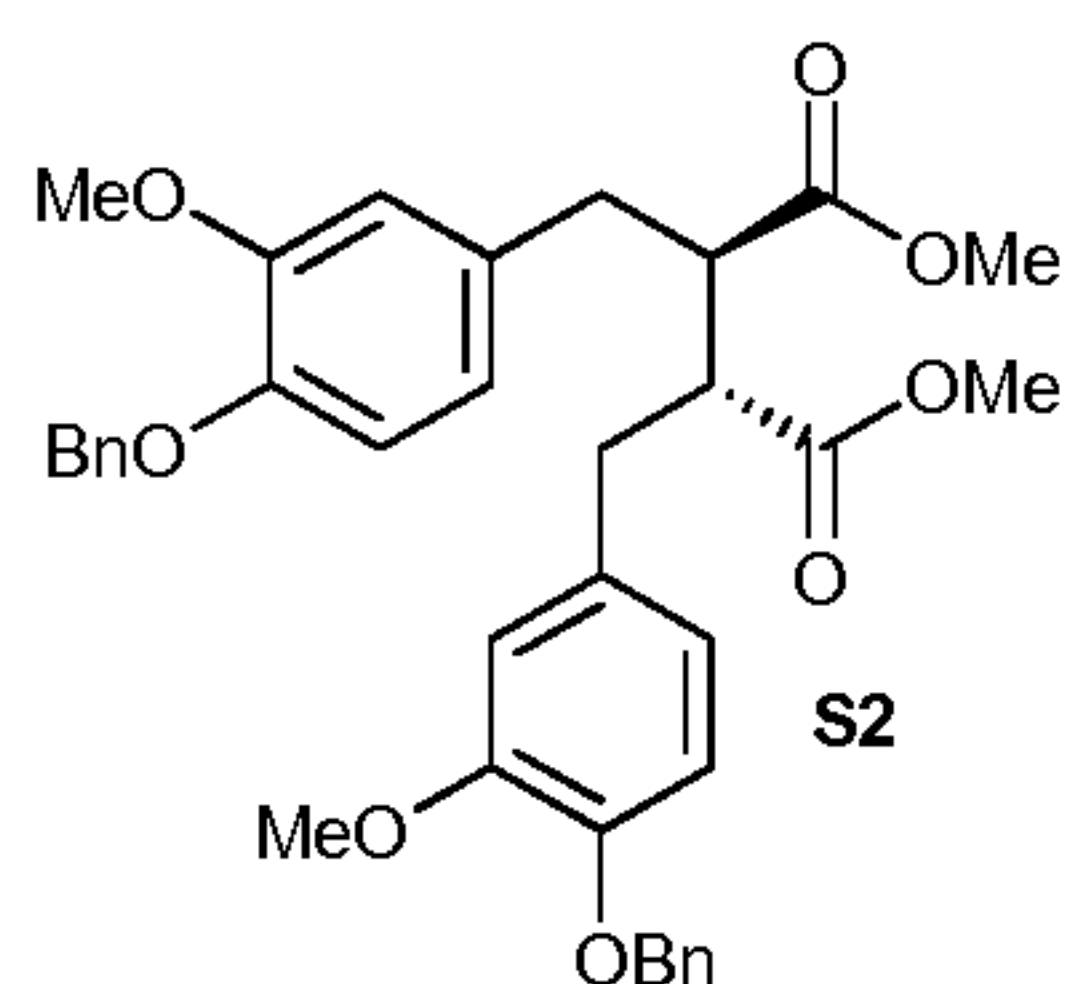


- (b) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare a compound of formula ((*S,S*)-8)



(c) deprotecting the compound of formula ((S,S)-8) to prepare the compound of formula ((S,S)-SDG-1).

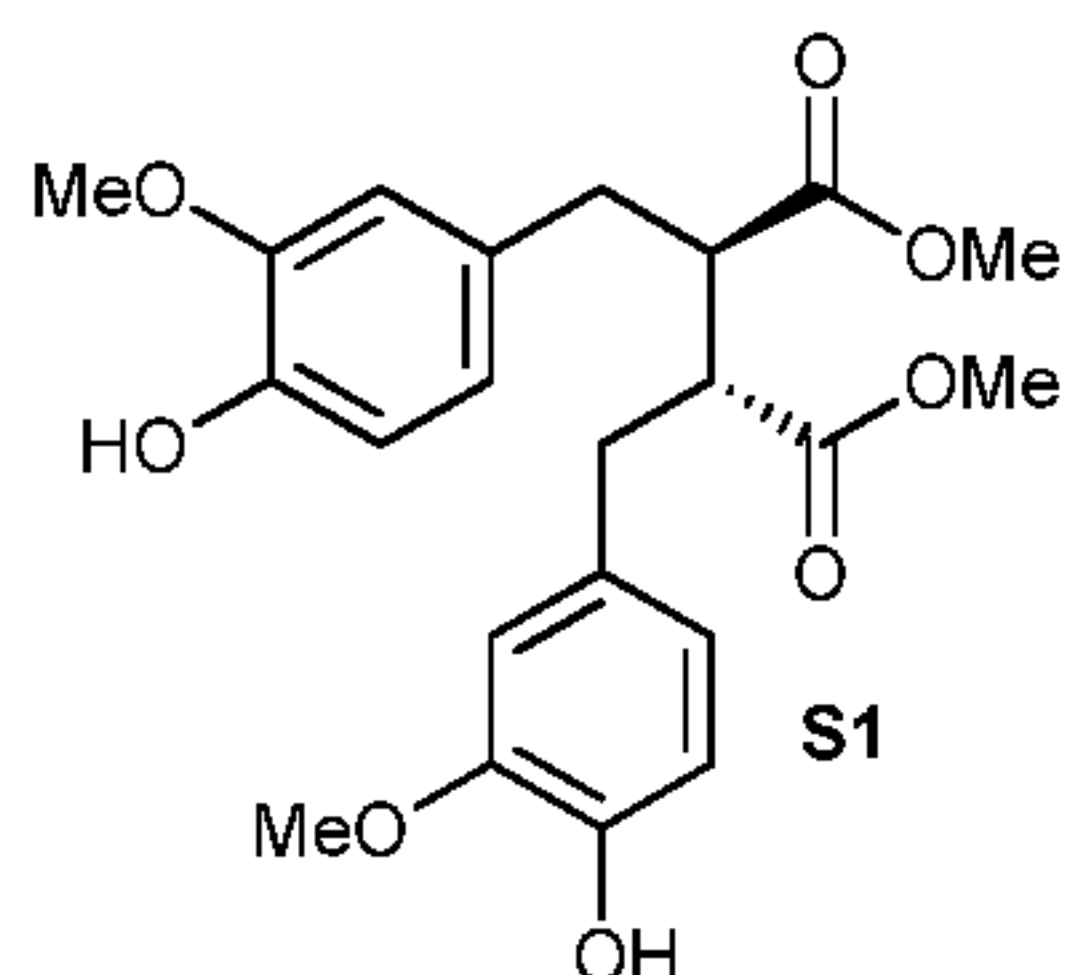
2. The process of claim 1, wherein said reacting is carried out in the presence of a Lewis acid.
3. The process of claim 2, wherein said Lewis acid is TMSOTf.
4. The process of claim 3, wherein said reacting is carried out in the presence of activated molecular sieves.
5. The process of claim 1, wherein said cleaving is carried out in the presence of H₂ and Pd/C in MeOH.
6. The process of claim 1, wherein said separation procedure is carried out using preparative thin layer chromatography.
7. The process of claim 1, wherein said deprotecting is carried out in a solution of NaOMe and MeOH.
8. The process of claim 1, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

9. The process of claim 8, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

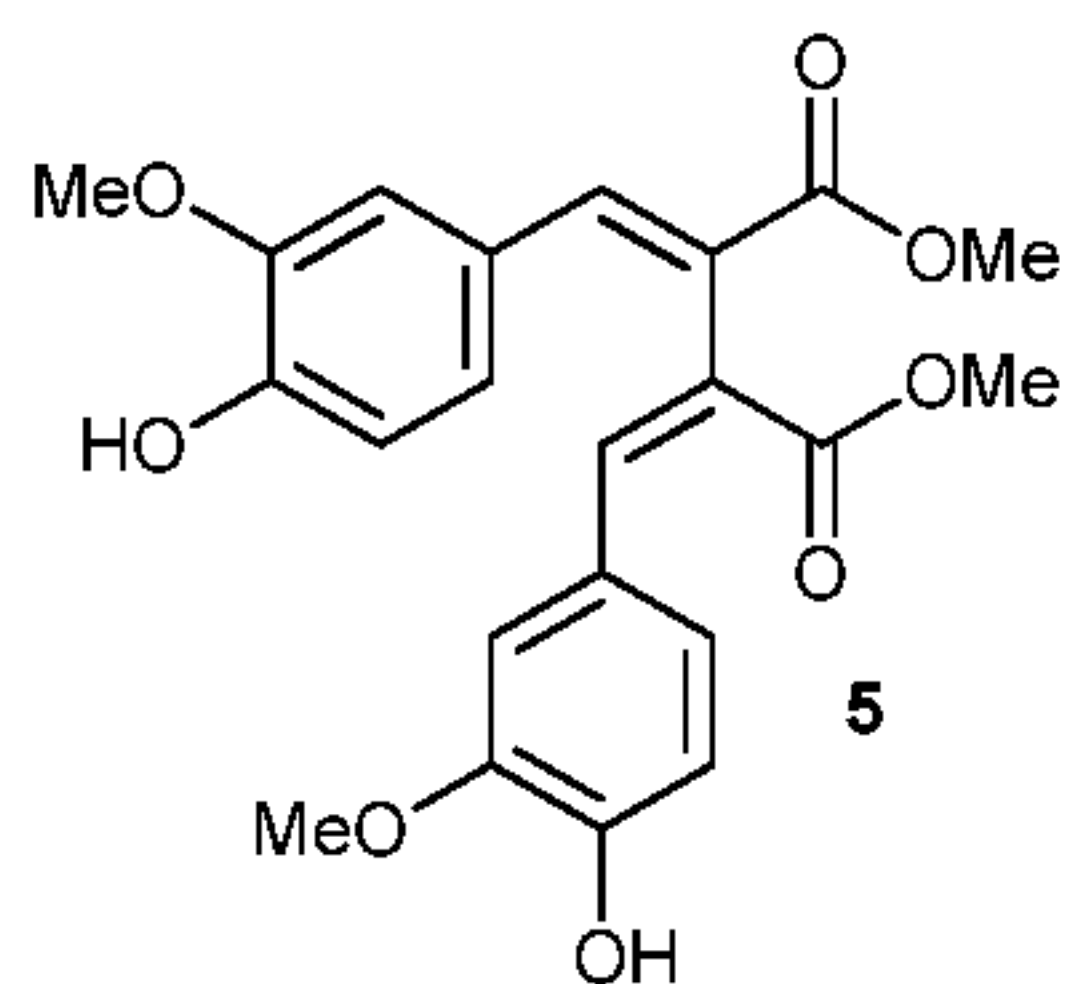
10. The process of claim 8, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

11. The process of claim 10, wherein said benzylating agent is BnBr and NaH.

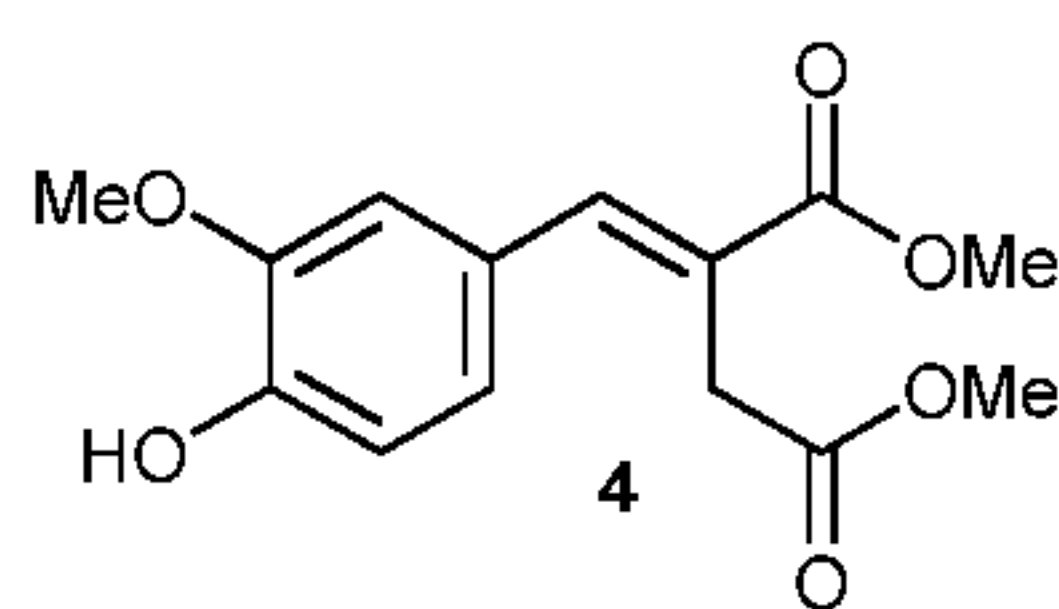
12. The process of claim 10, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

13. The process of claim 12, wherein said reducing agent is H₂ and Pd/C.

14. The process of claim 12, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

15. The process of claim 14, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium.

16. The process of claim 15, wherein said lithium is in the form of lithium wires.

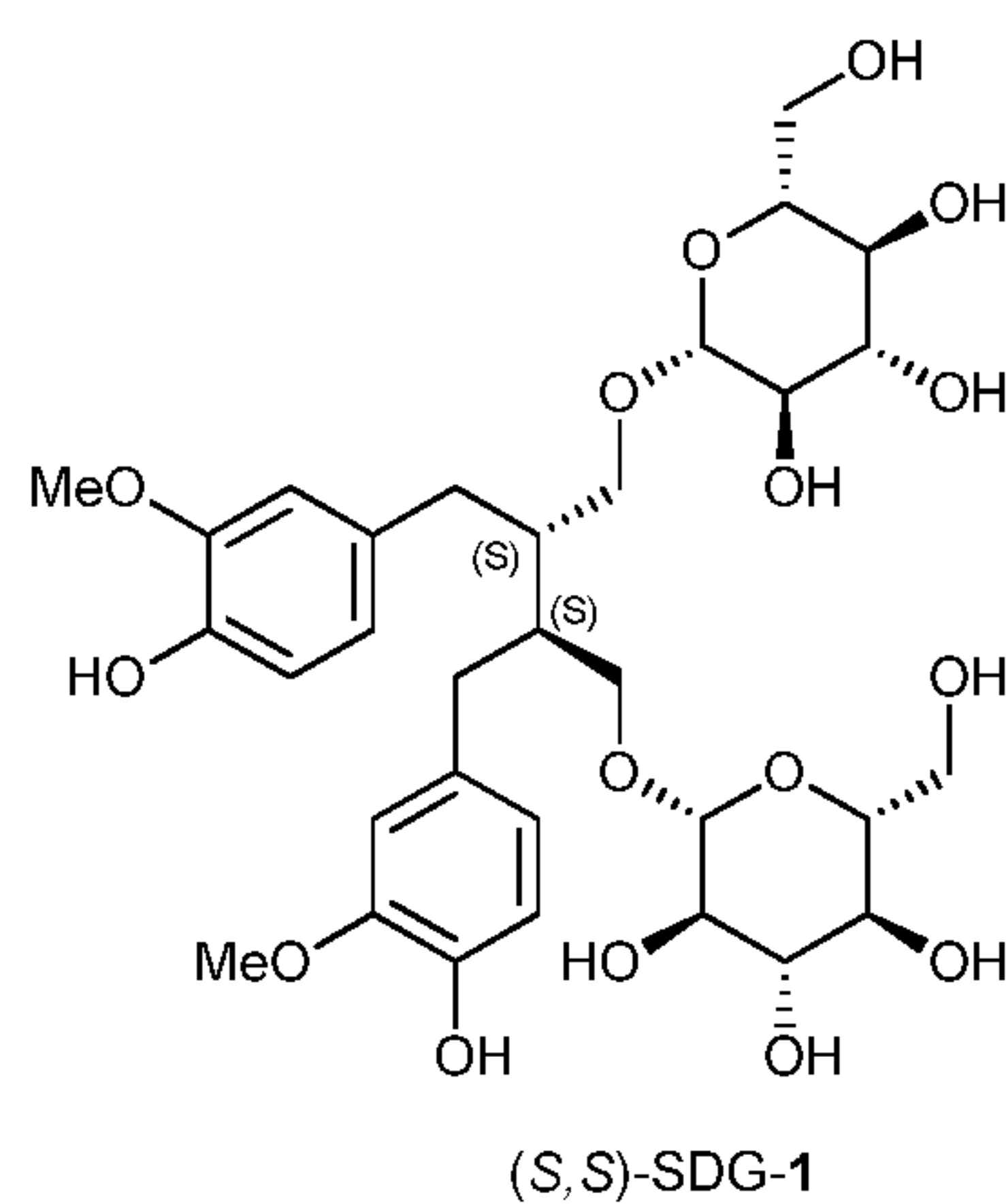
17. The process of claim 14, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

18. The process of claim 14, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

19. The process of claim 18, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

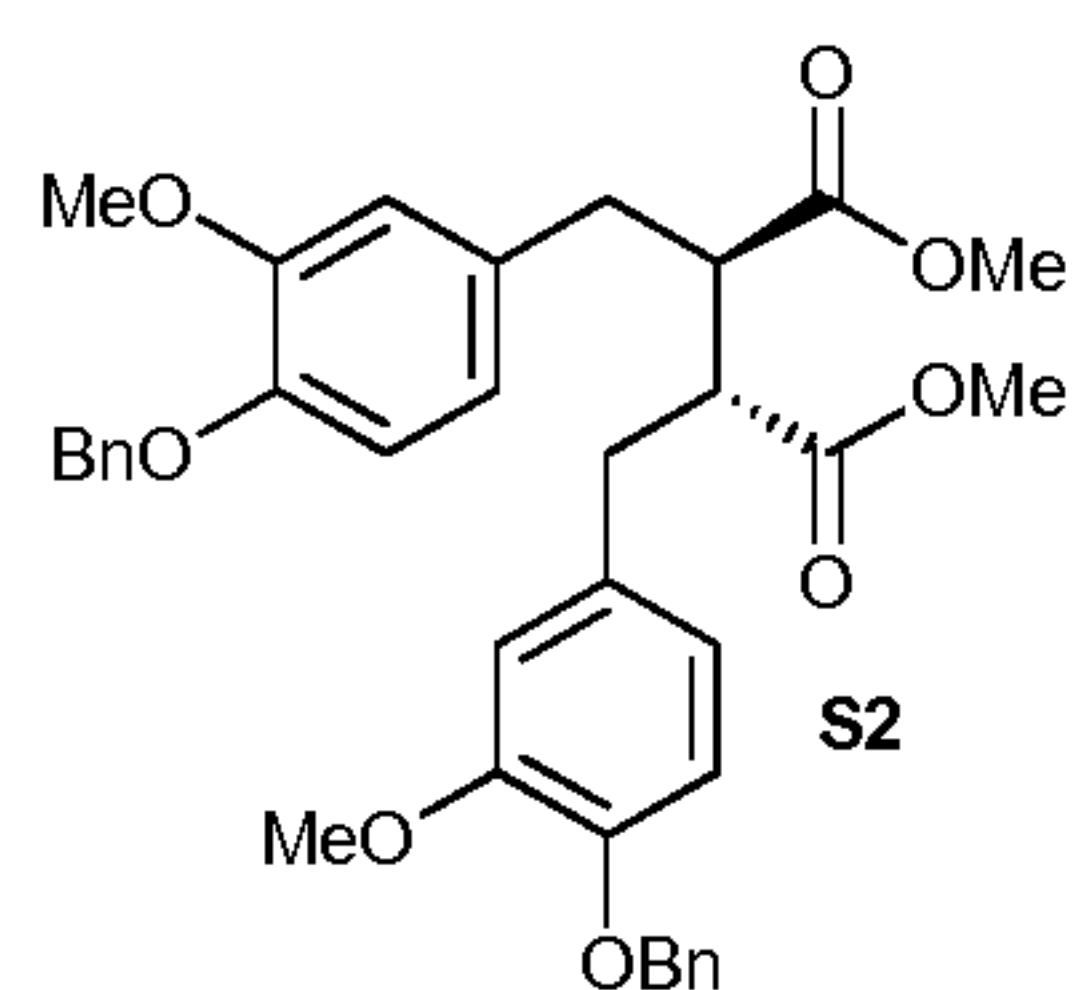
20. The process of claim 18, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

21. A process for preparing a compound of formula ((S,S)-SDG-1)

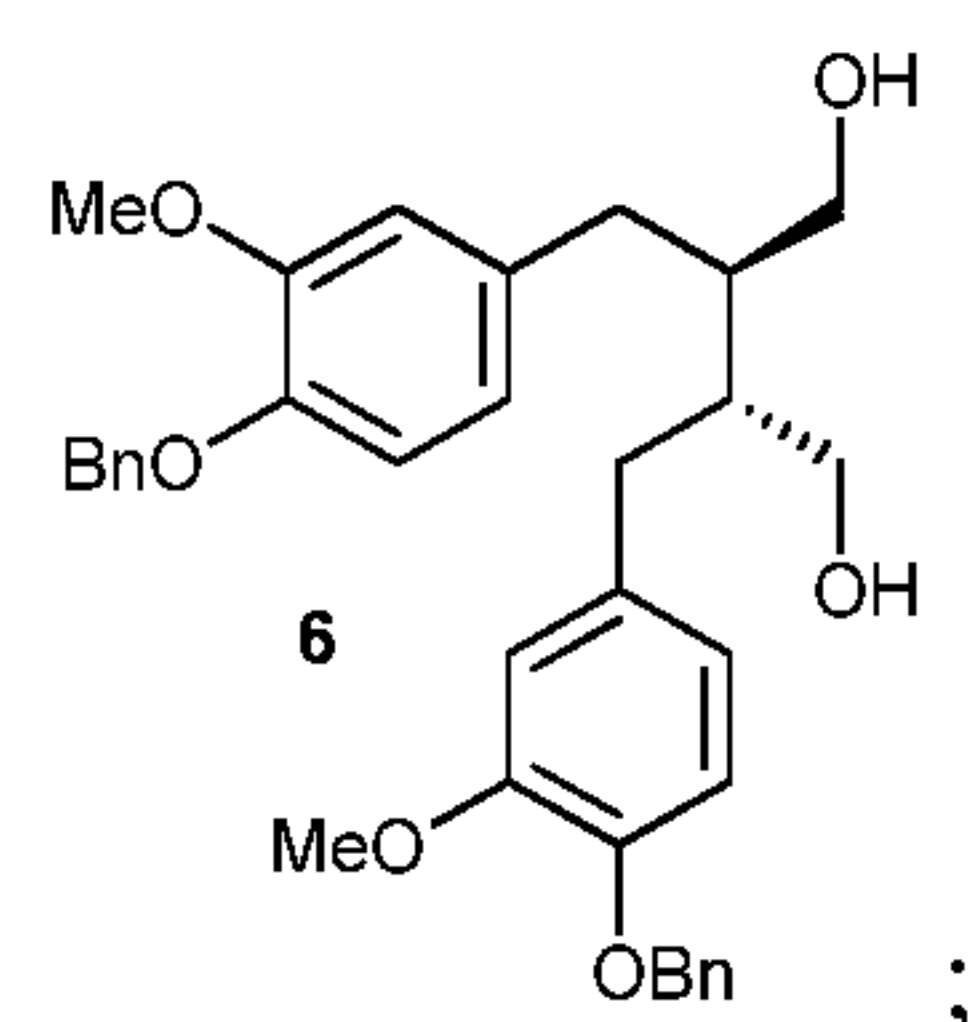


the process comprising:

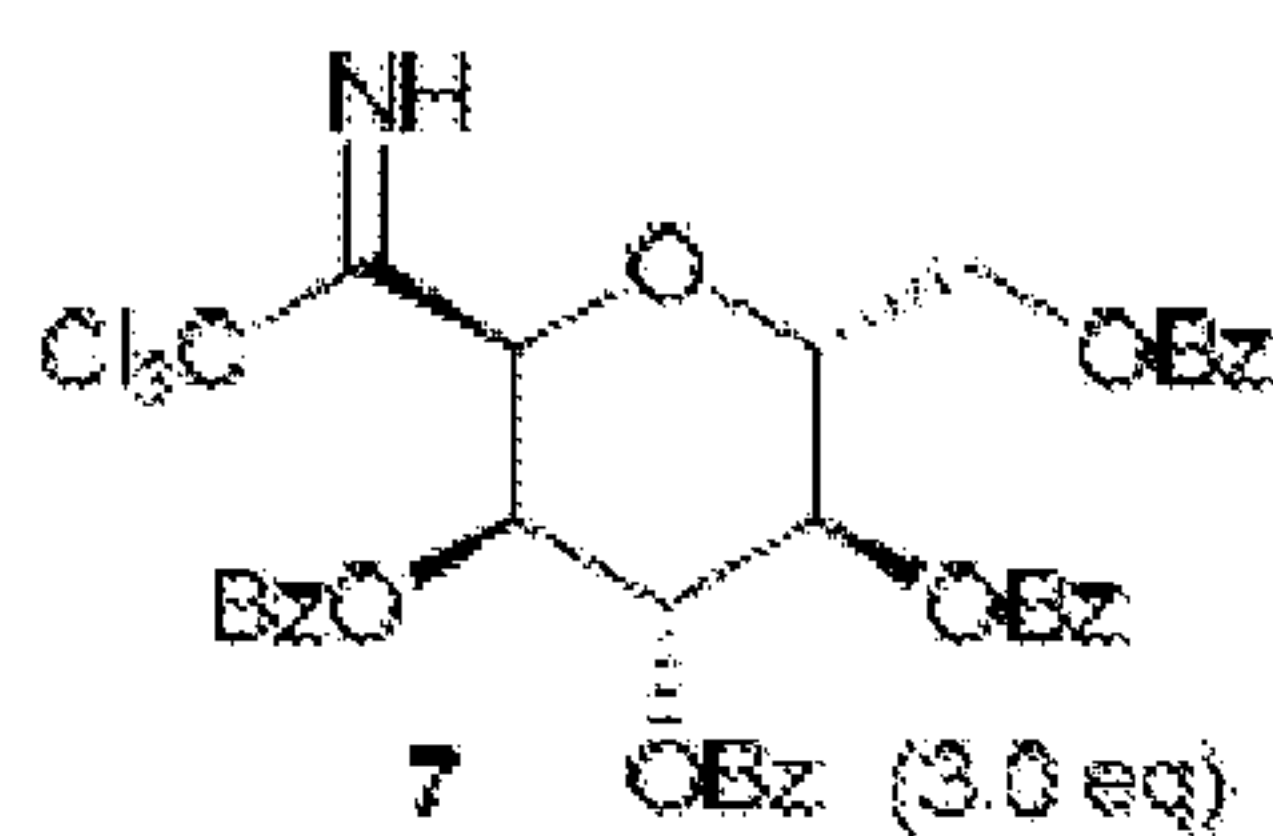
(a) reacting a compound of formula (S2)



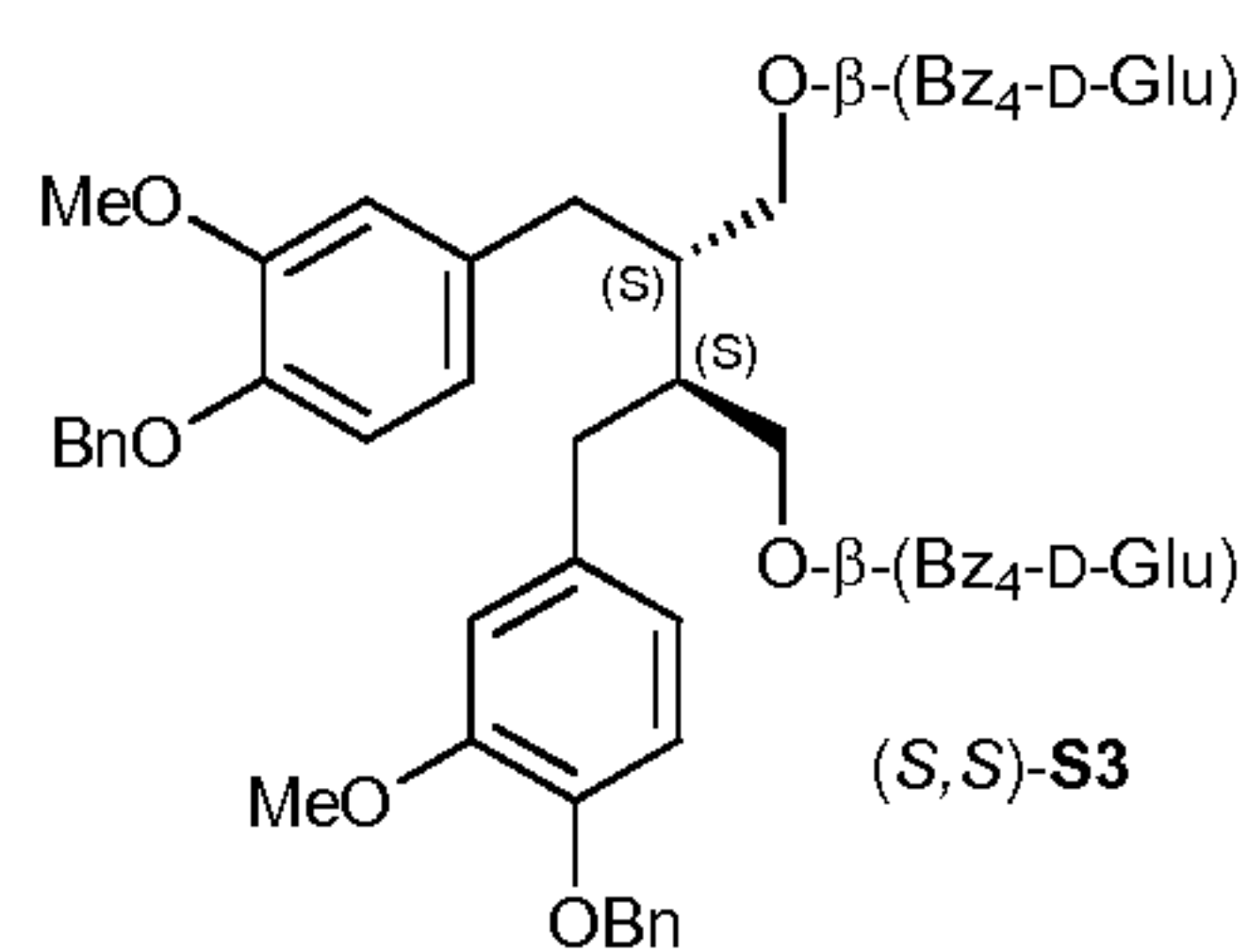
with a reducing agent to prepare a compound of formula (6)



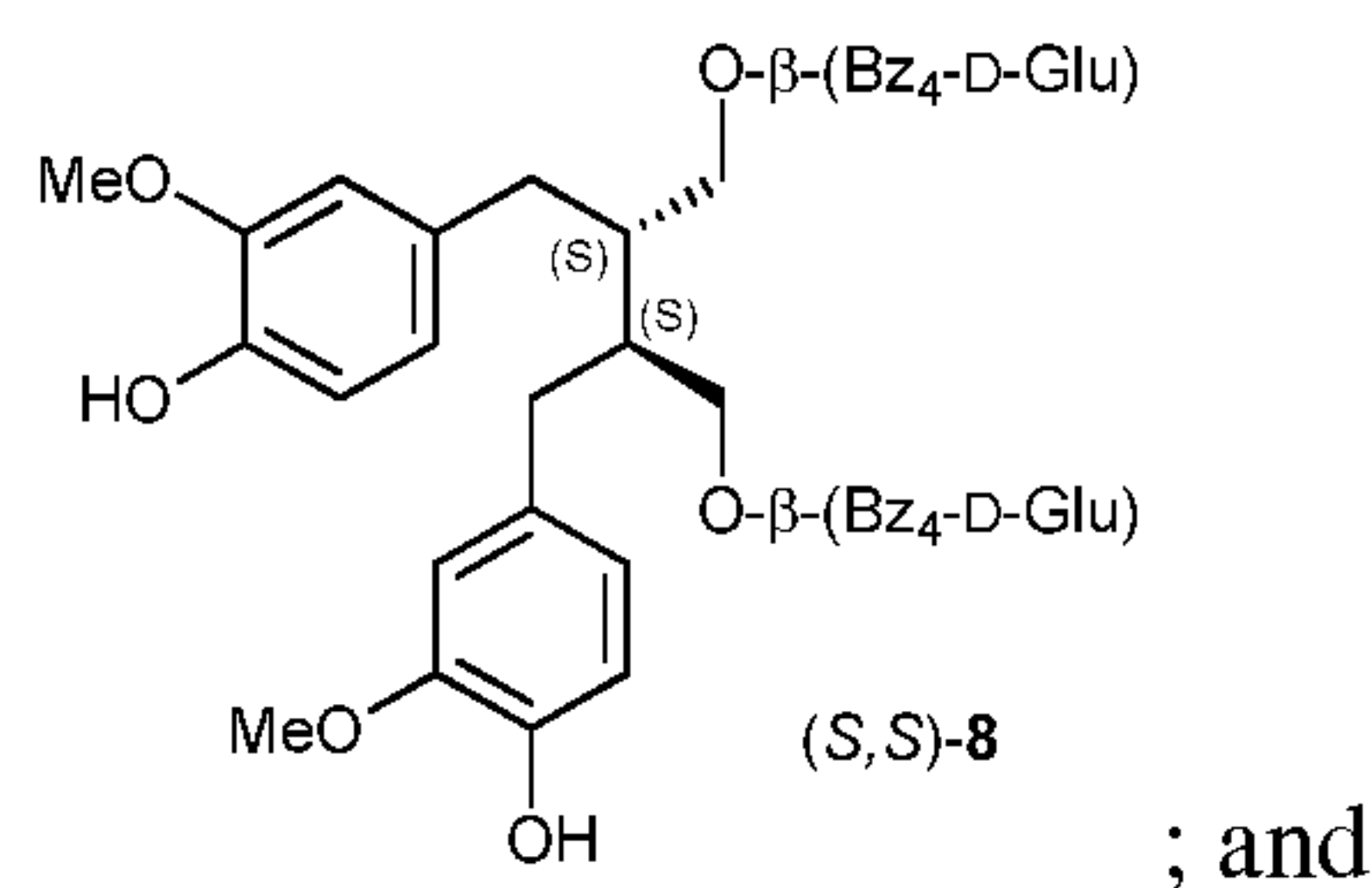
(b) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((S,S)-S3)



(c) cleaving benzyl ethers of the compound of formula ((*S,S*)-S3), followed by a separation procedure to prepare a compound of formula ((*S,S*)-8)



(d) deprotecting the compound of formula ((*S,S*)-8) to prepare the compound of formula ((*S,S*)-SDG-1).

22. The process of claim 21, wherein said reacting of step (a) is carried out in the presence of TMSOTf.

23. The process of claim 22, wherein said reacting of step (a) is carried out in the presence of activated molecular sieves.

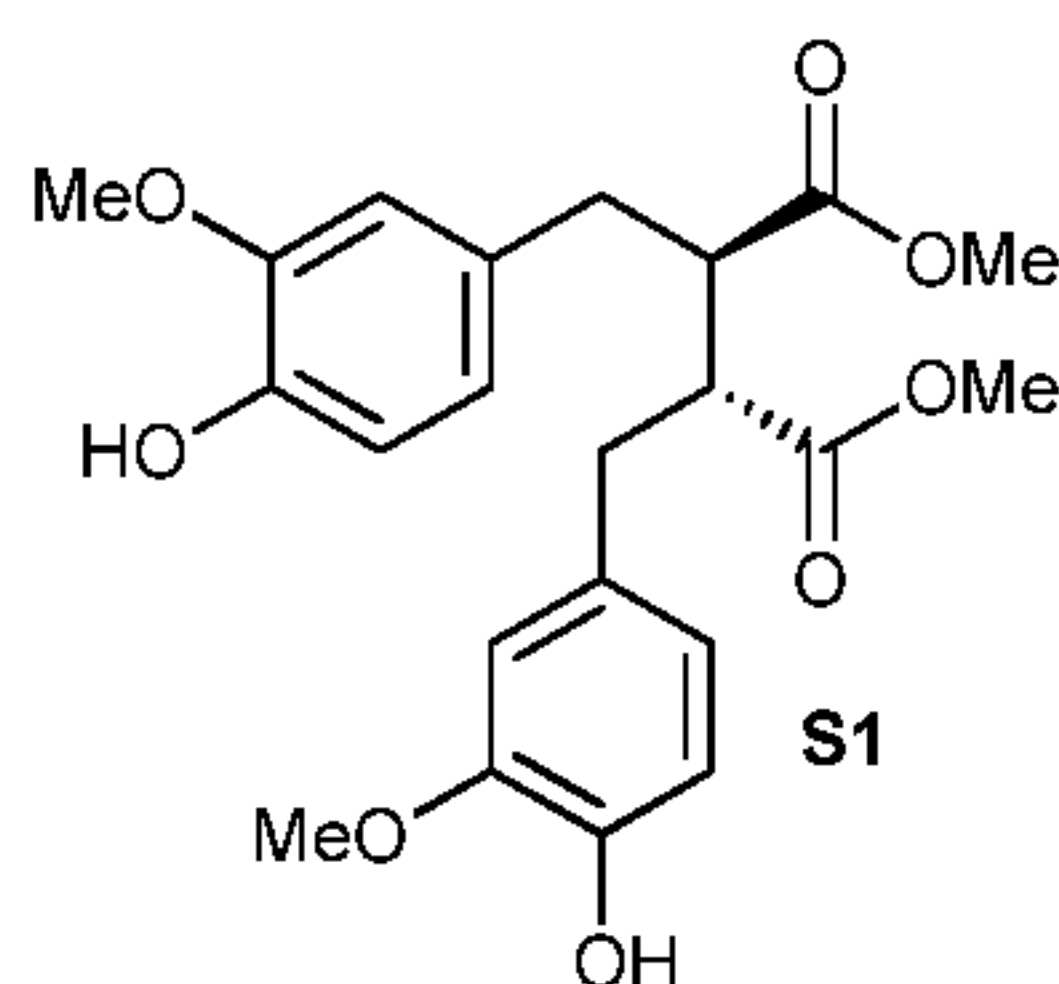
24. The process of claim 21, wherein said cleaving is carried out in the presence H₂ and Pd/C in MeOH.

25. The process of claim 21, wherein said separation procedure is carried out using preparative thin layer chromatography.

26. The process of claim 21, wherein said deprotecting is carried out in a solution of NaOMe and MeOH.

27. The process of claim 21, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

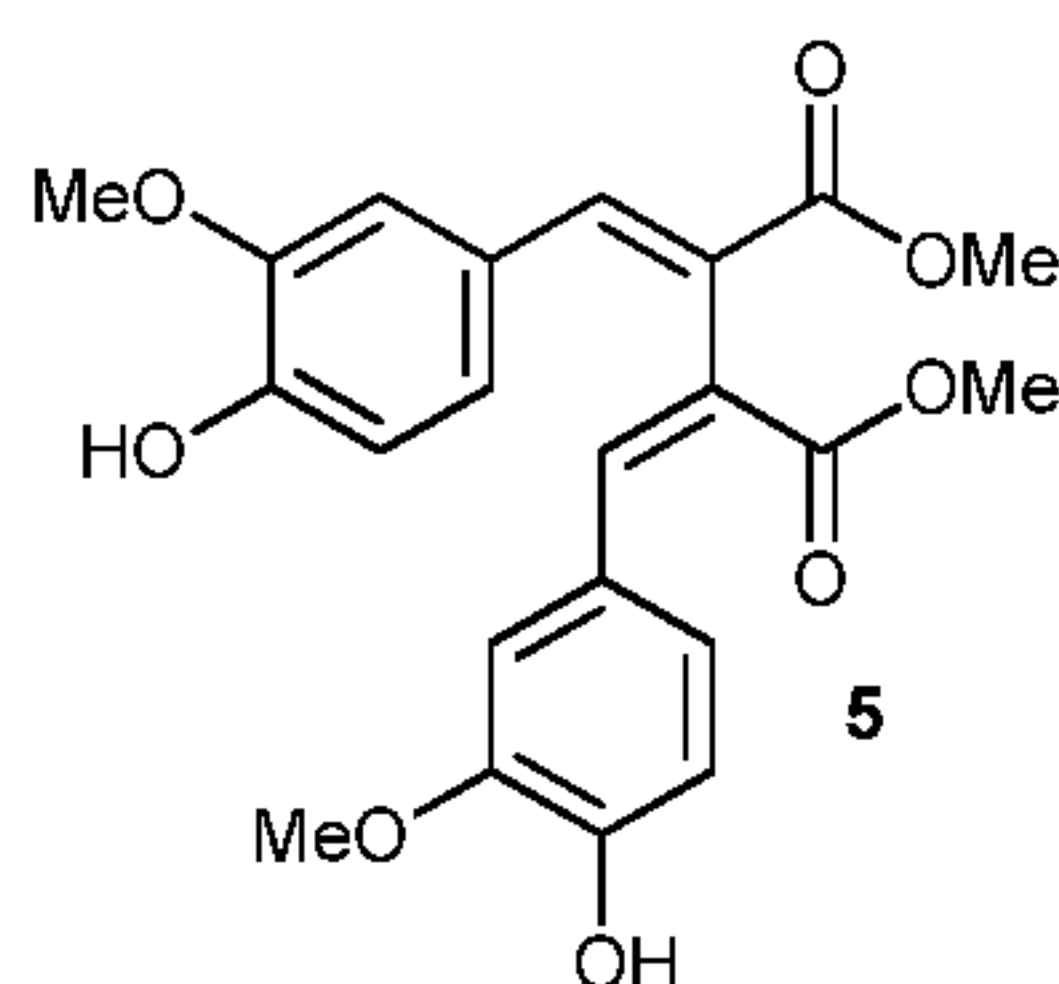
28. The process of claim 21, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to form the compound of formula (S2).

29. The process of claim 28, wherein said benzylating agent is BnBr and NaH.

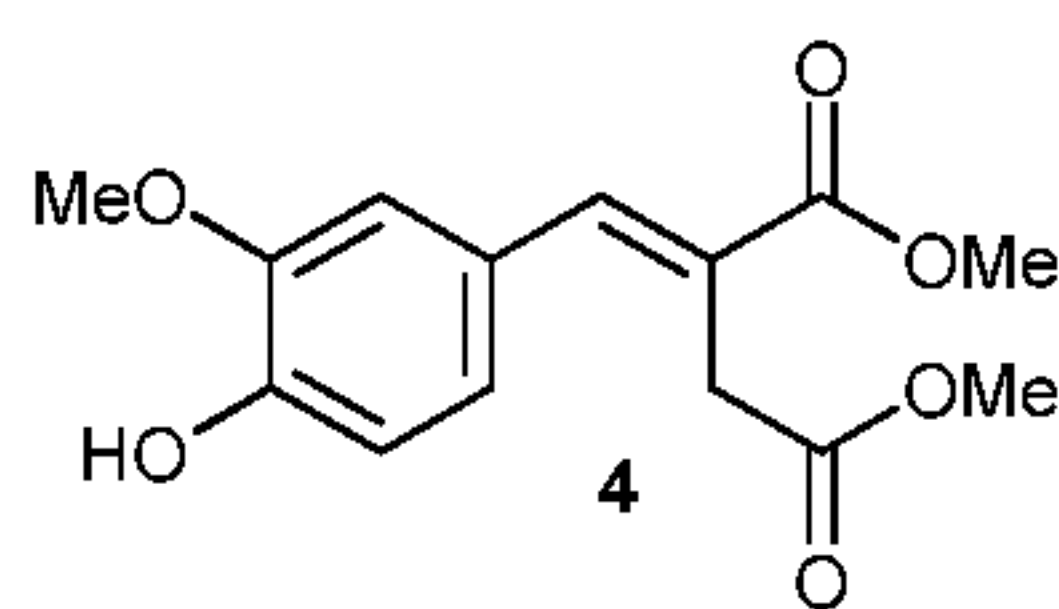
30. The process of claim 28, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

31. The process of claim 30, wherein said reducing agent is H₂ and Pd/C.

32. The process of claim 30, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

33. The process of claim 32, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

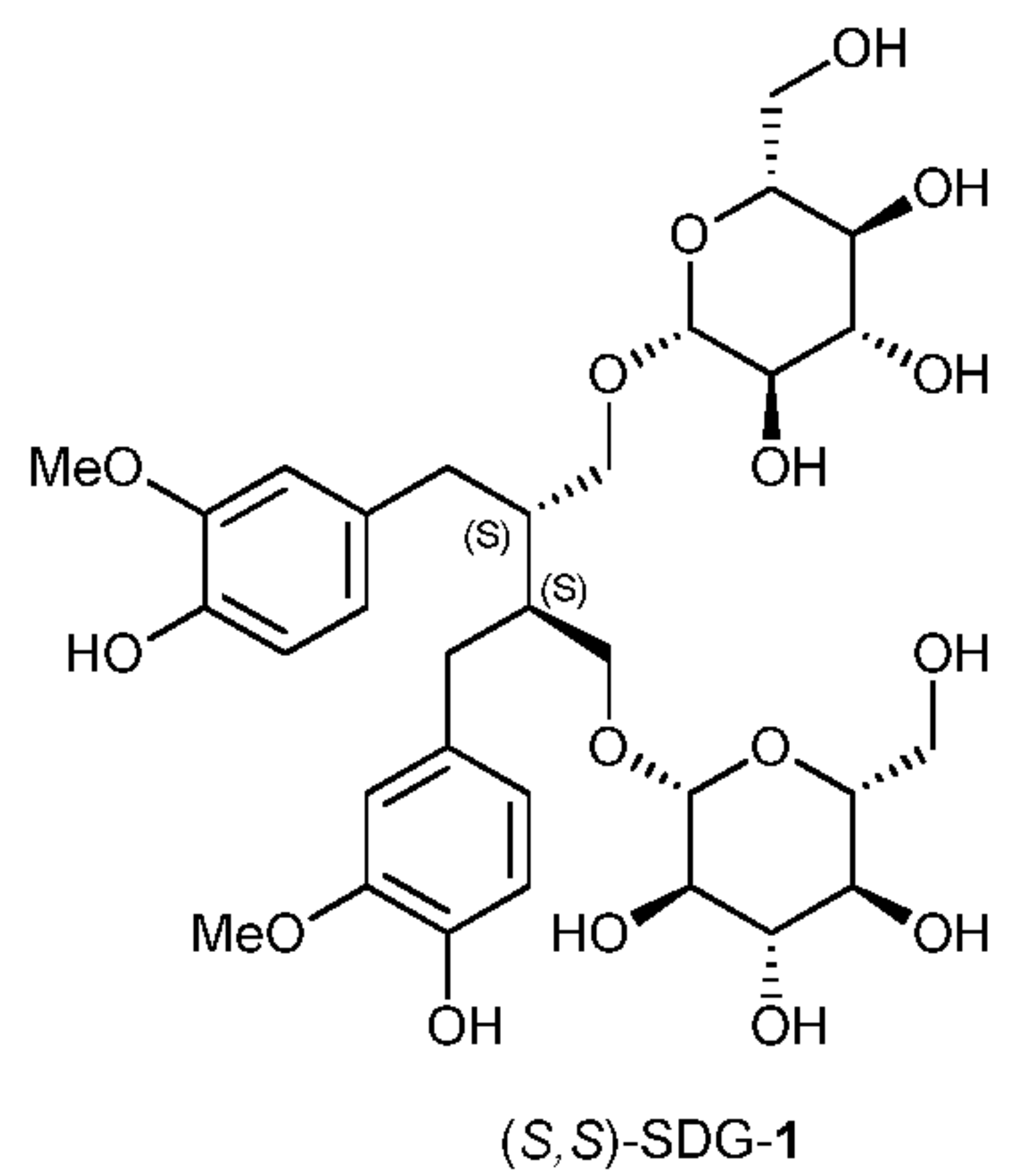
34. The process of claim 32, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

35. The process of claim 32, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

36. The process of claim 34, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

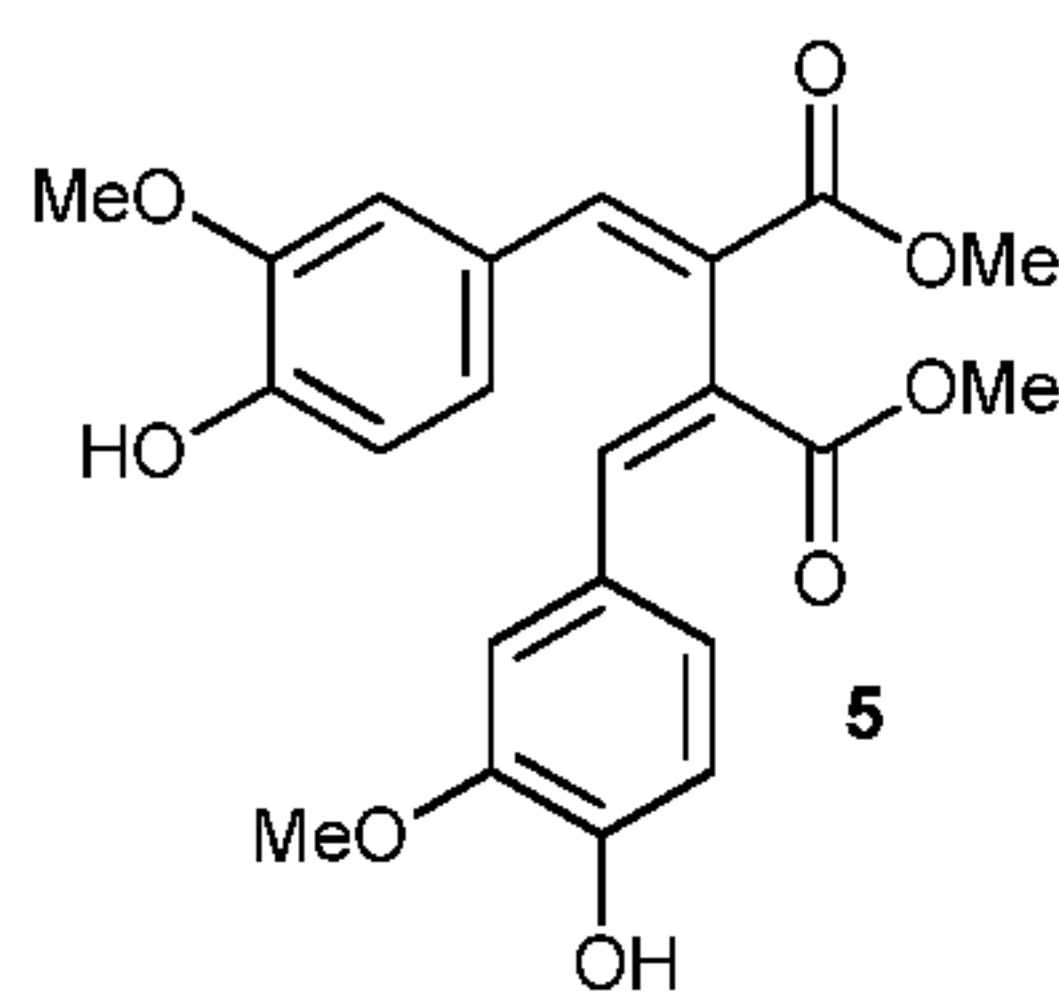
37. The process of claim 34, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

38. A process for preparing a compound of formula ((S,S)-SDG-1)

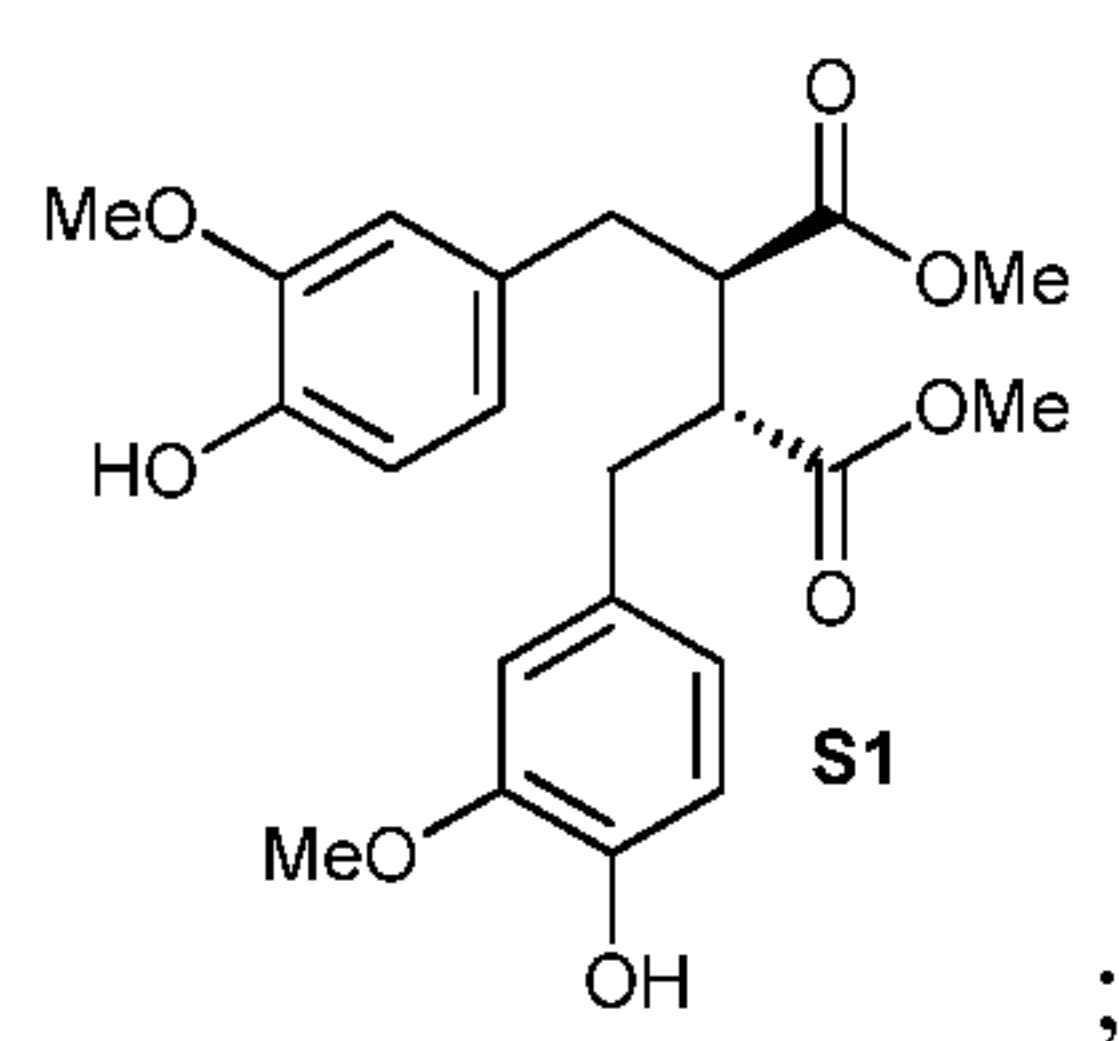


the process comprising:

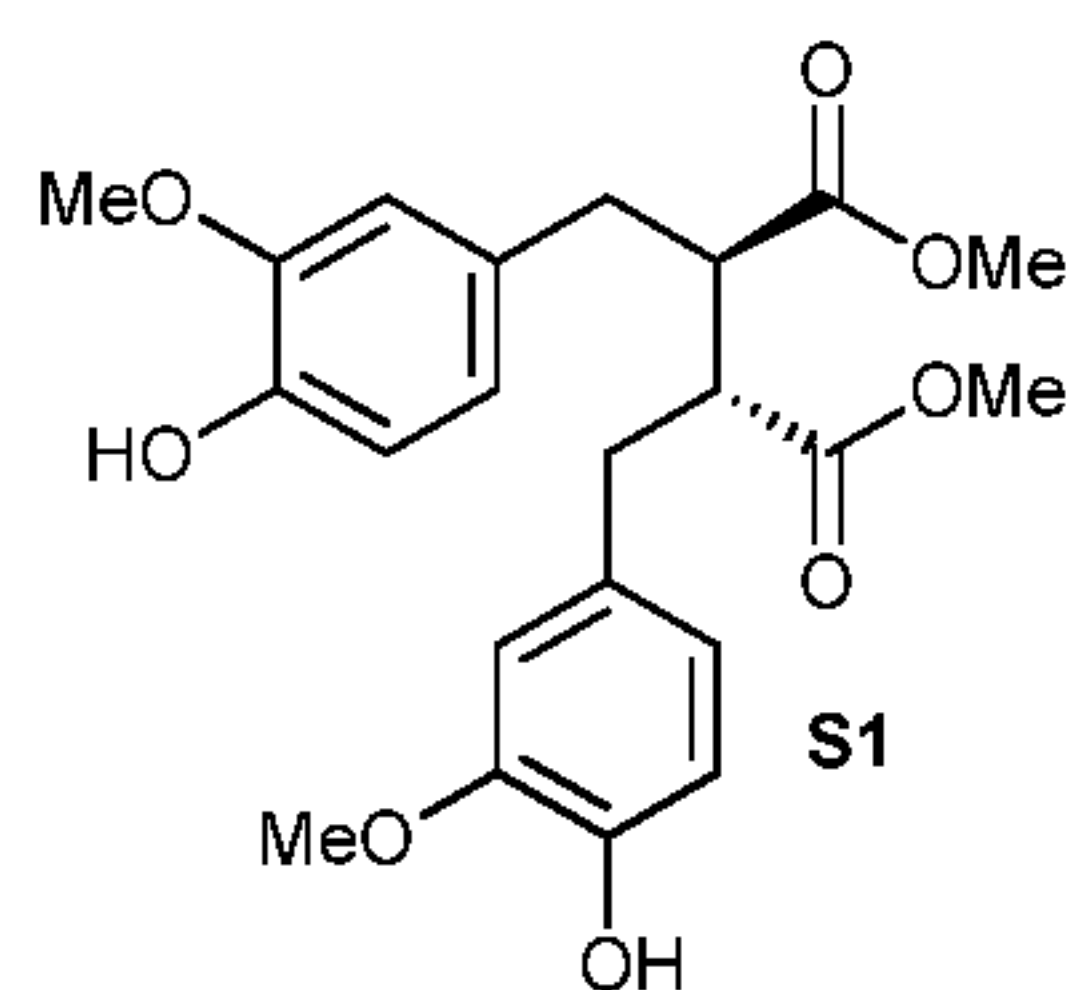
(a) reacting a compound of formula (5)



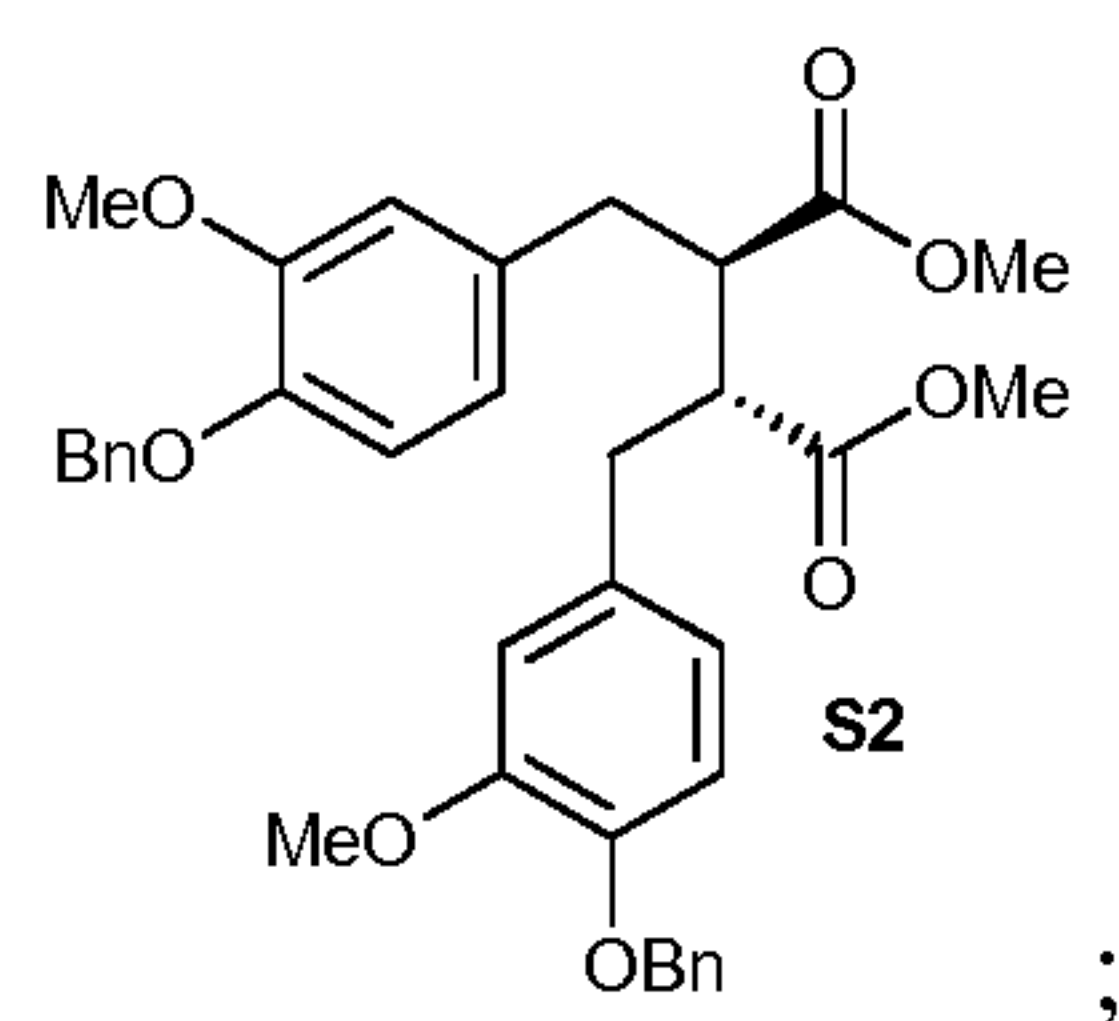
with a reducing agent to prepare a compound of formula (S1)



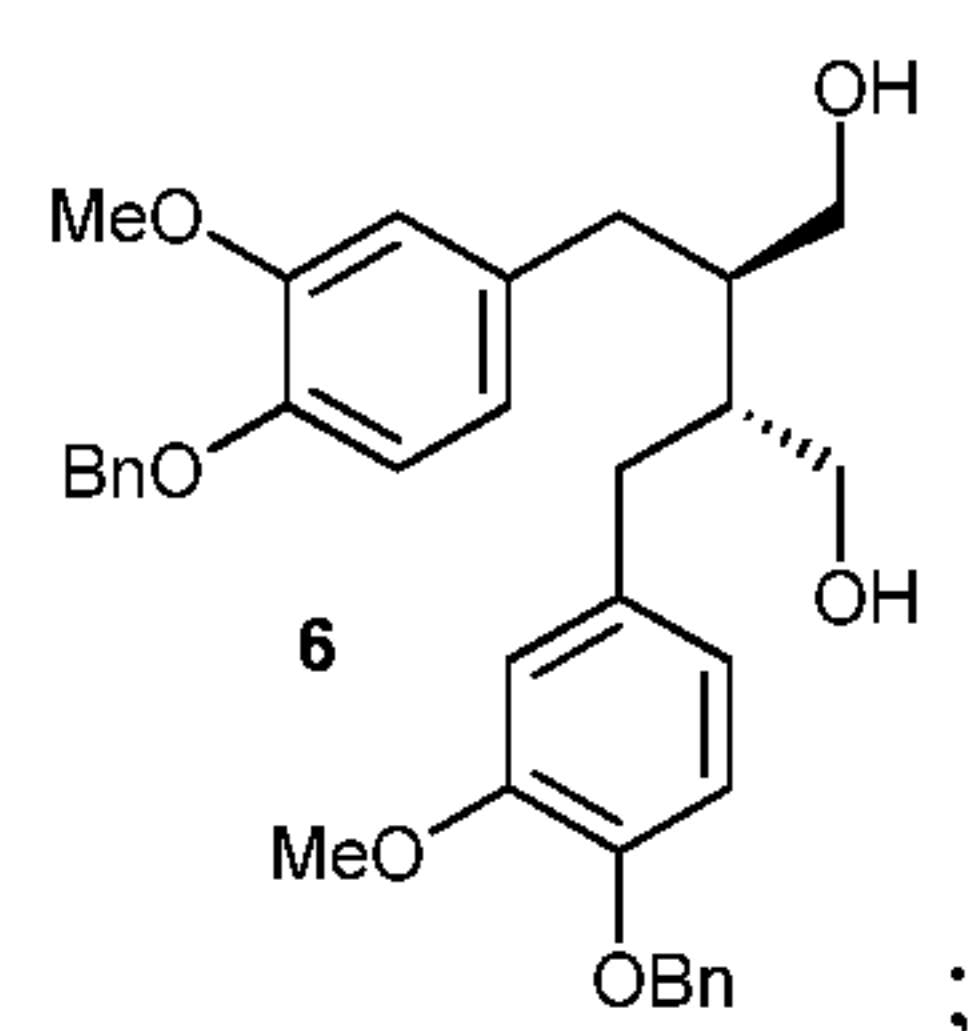
(b) reacting the compound of formula (S1)



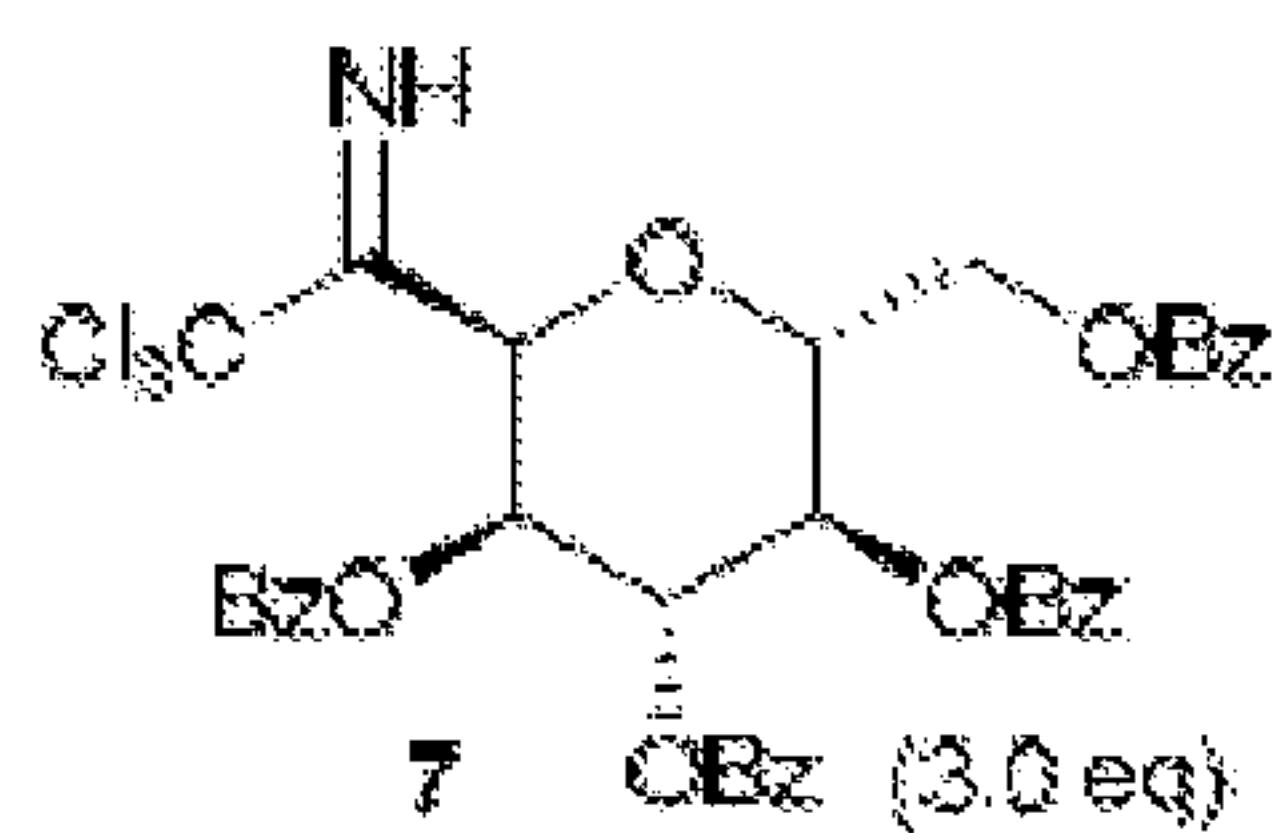
with a benzylating agent to prepare a compound of formula (S2)



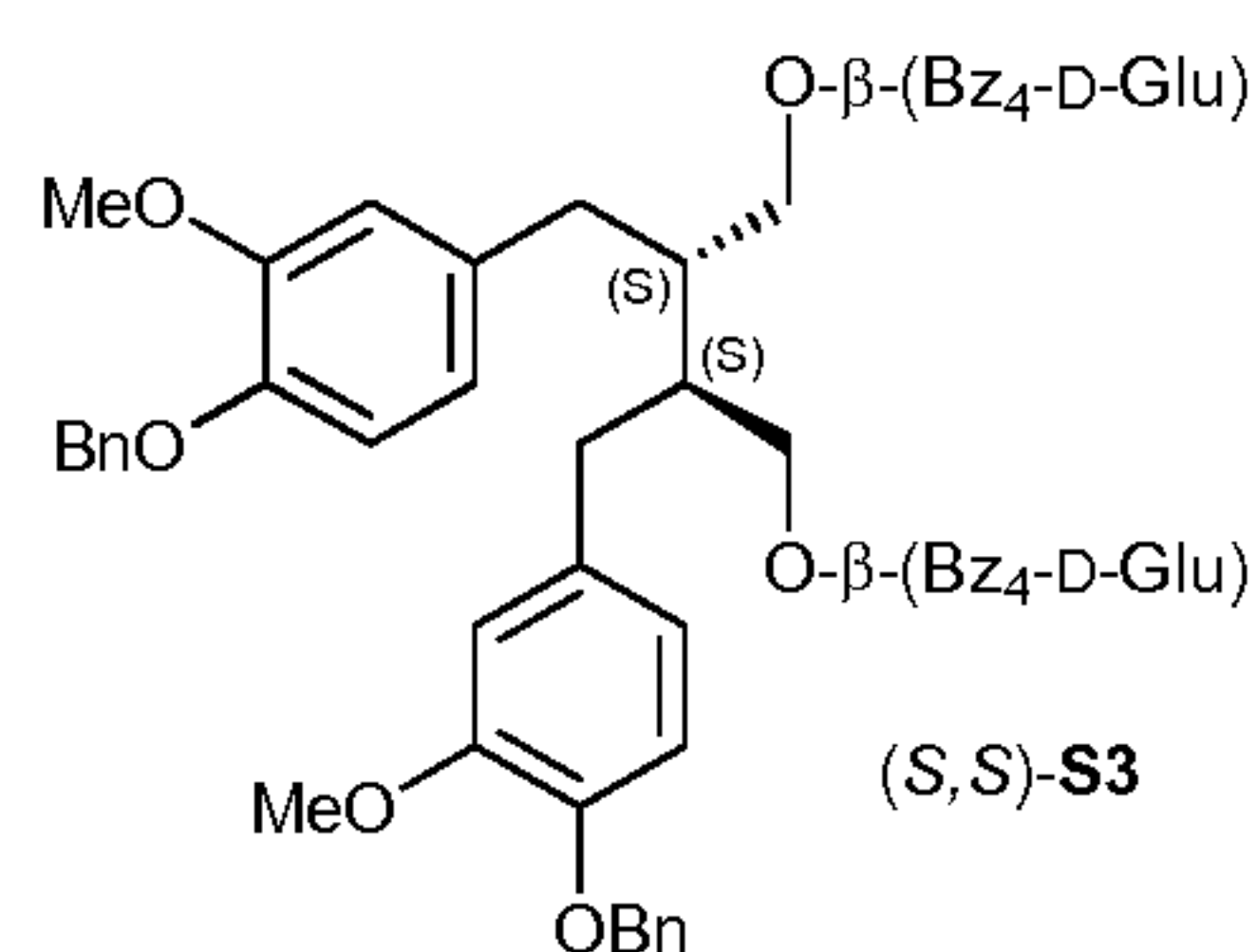
(c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6)



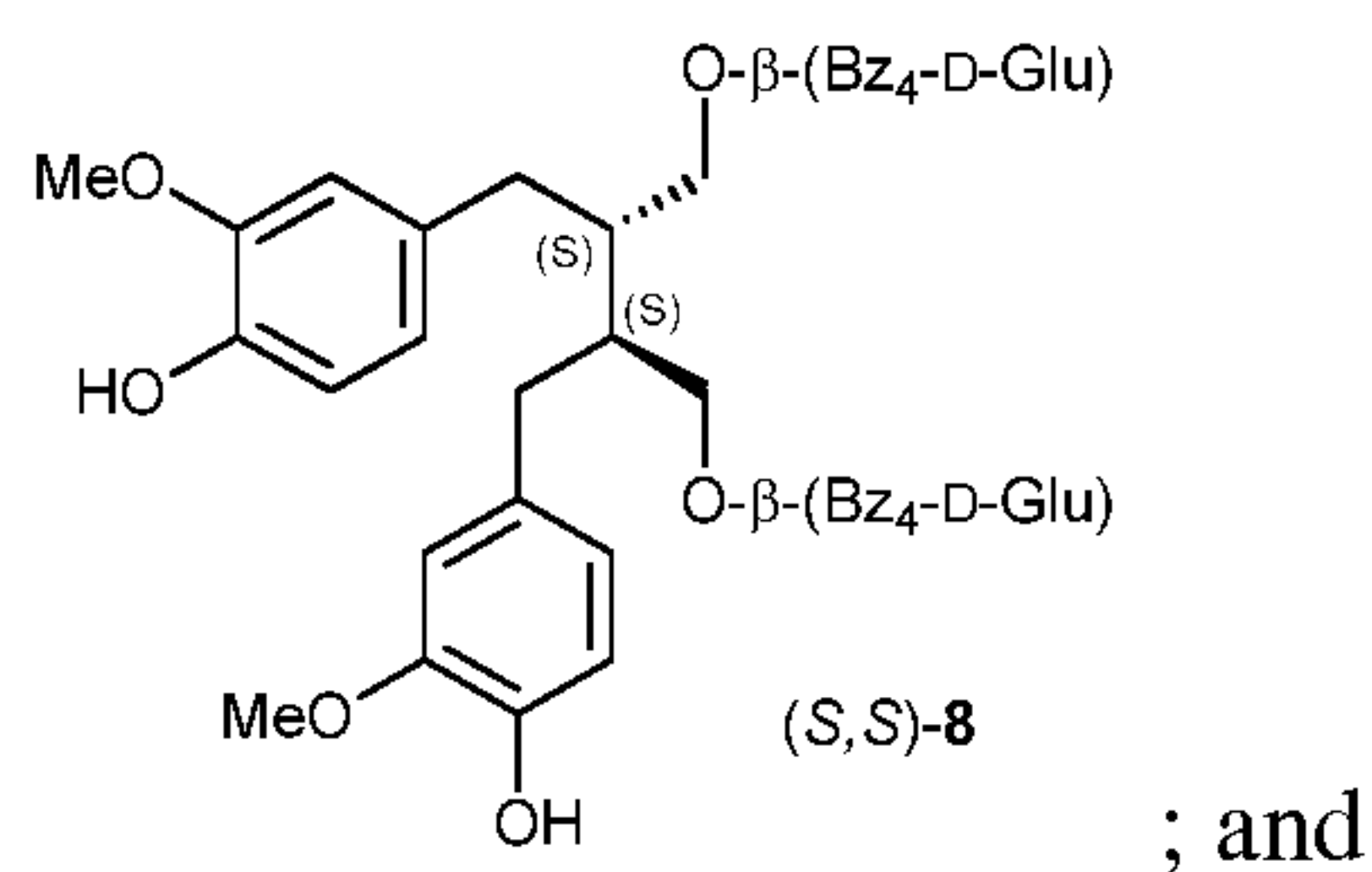
(d) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((S,S)-S3)



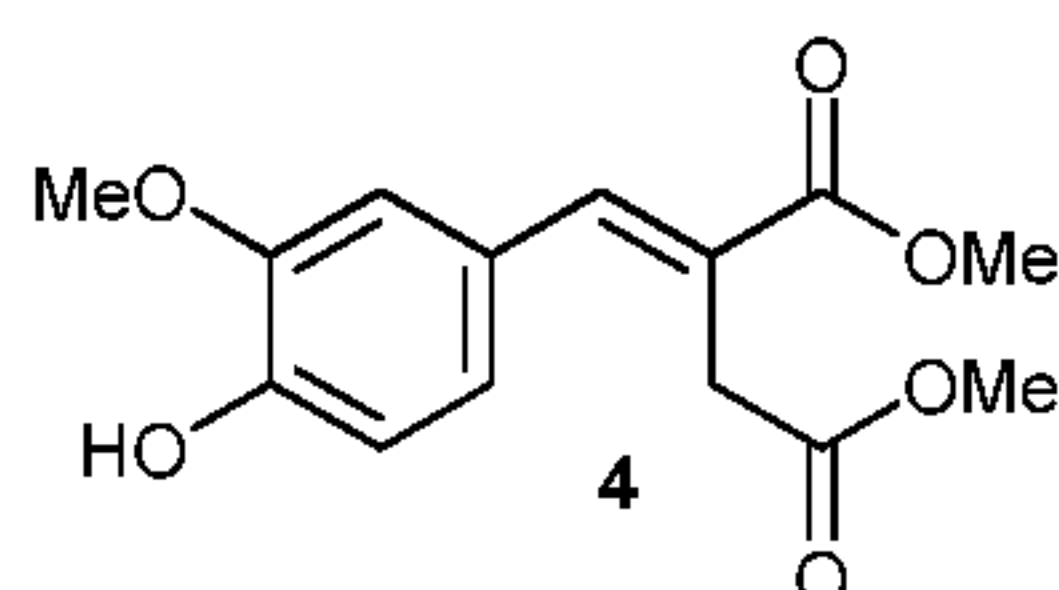
(e) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare a compound of formula ((S,S)-8)



; and

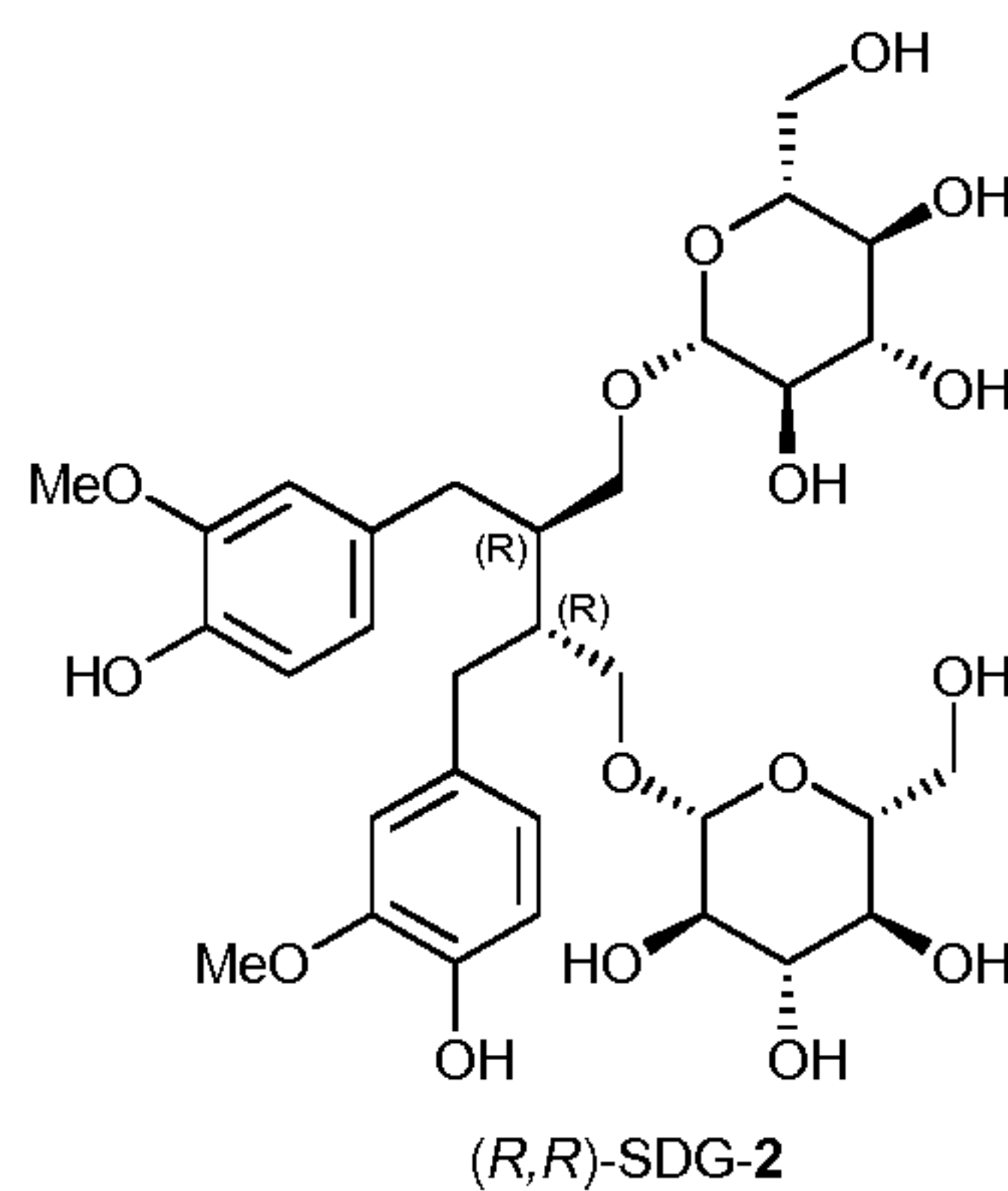
(f) deprotecting the compound of formula ((S,S)-8) to prepare the compound of formula ((S,S)-SDG-1).

39. The process of claim 38, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



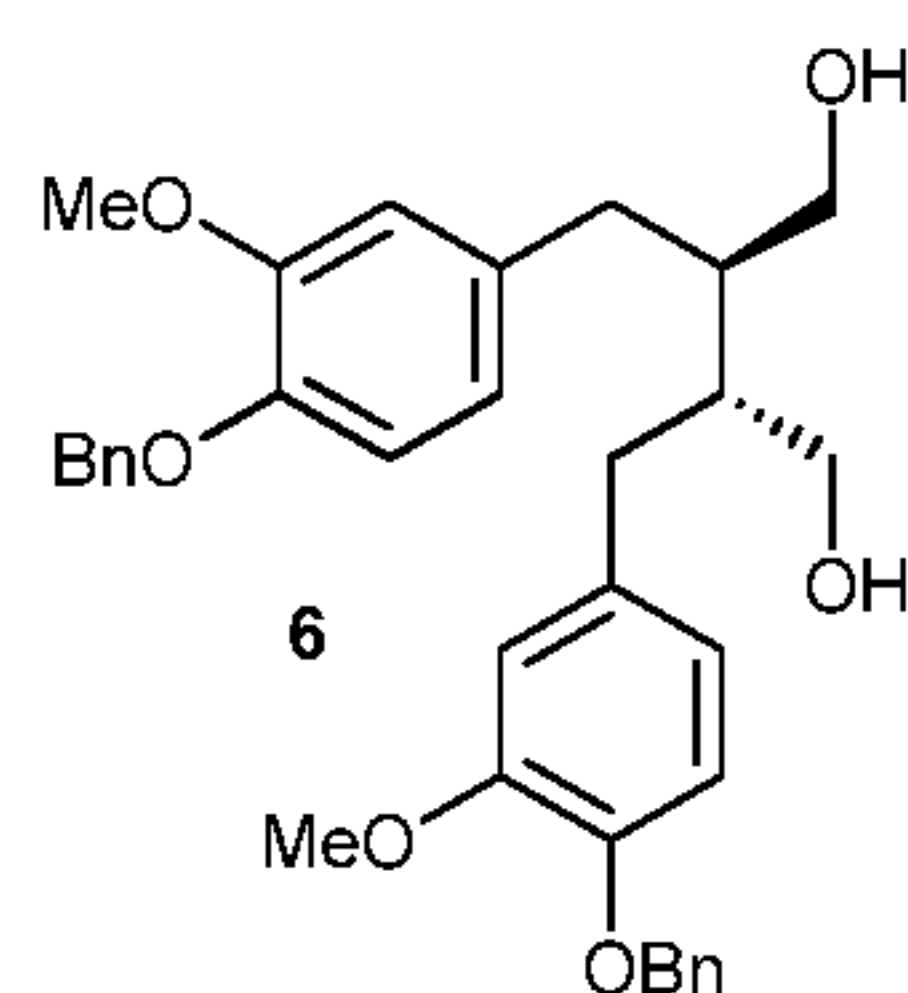
with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

40. A process for preparing a compound of formula ((R,R)-SDG-2)

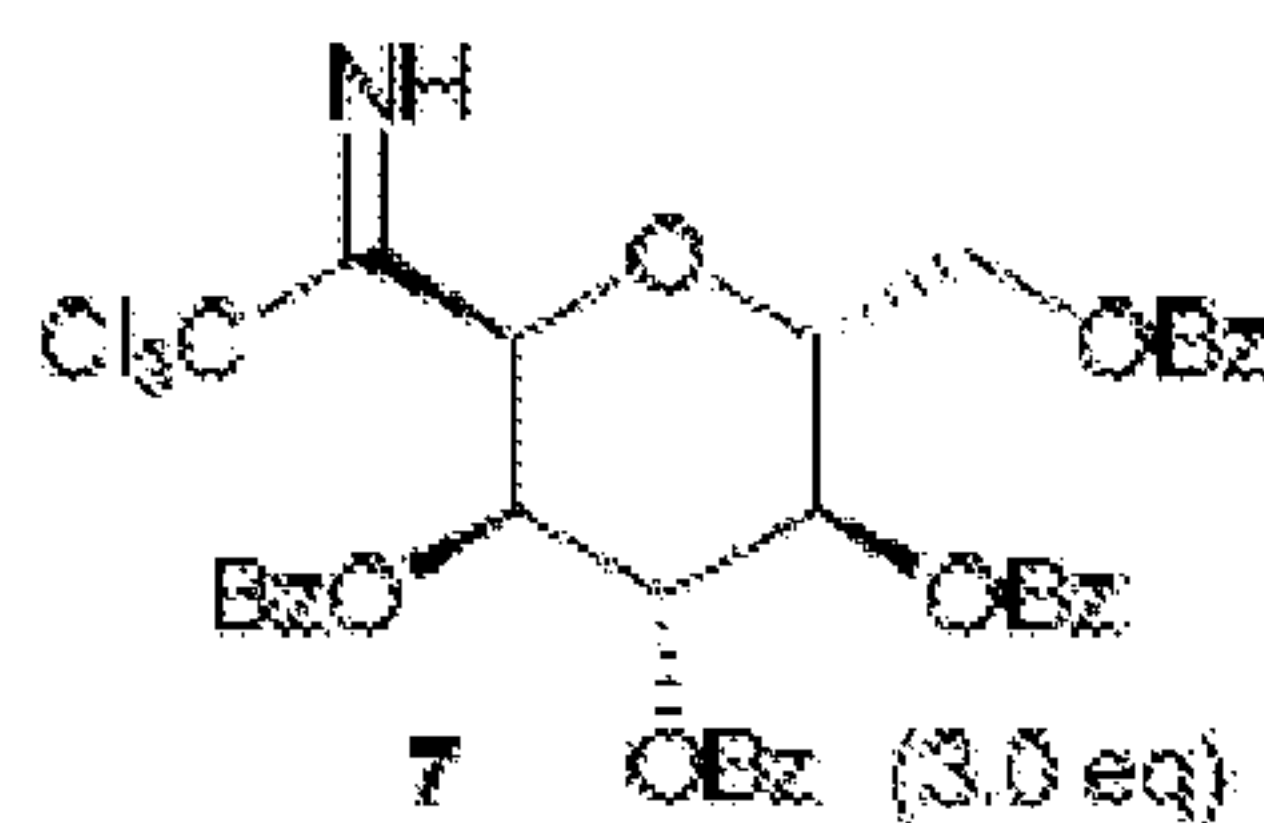


the process comprising:

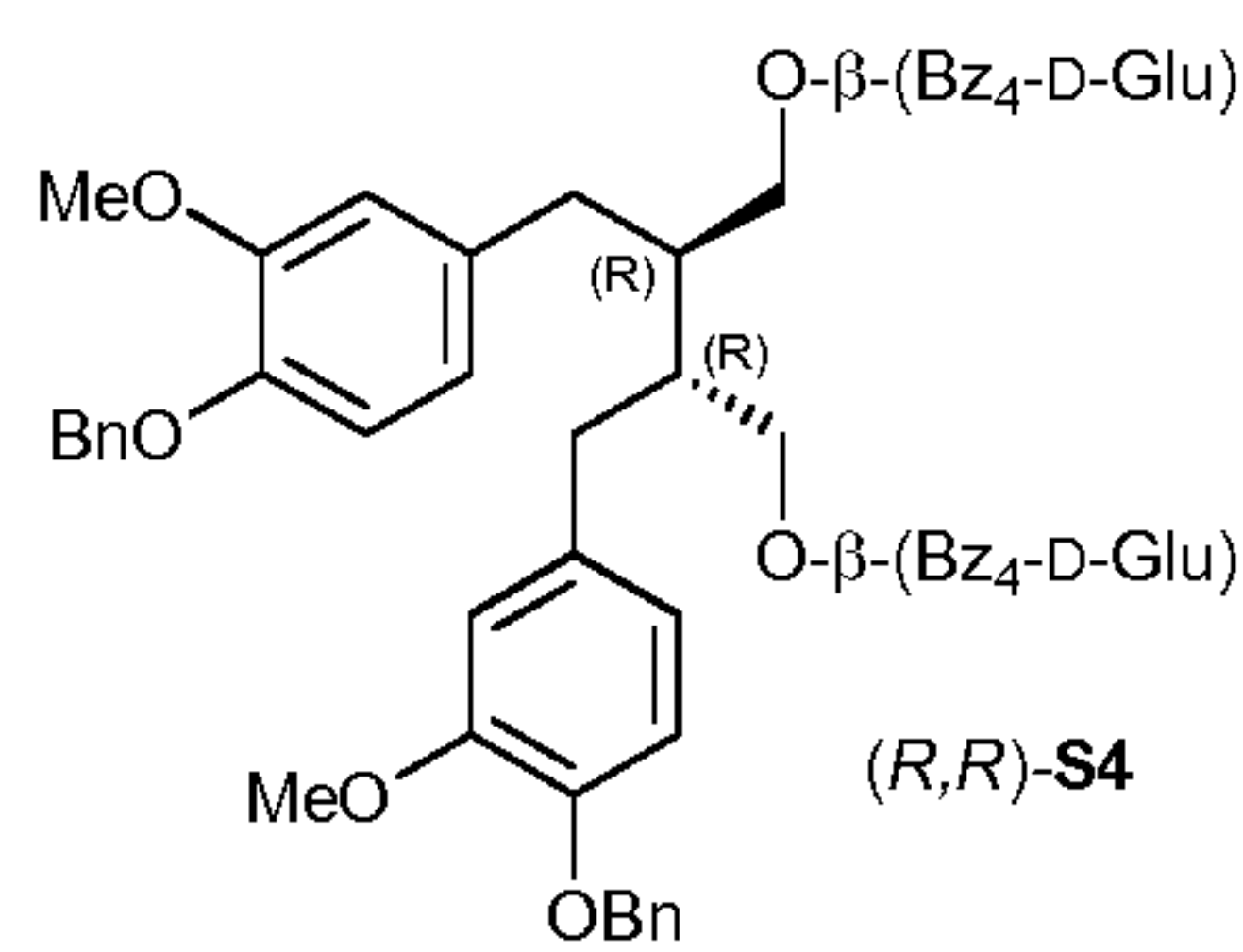
(a) reacting a compound of formula (6)



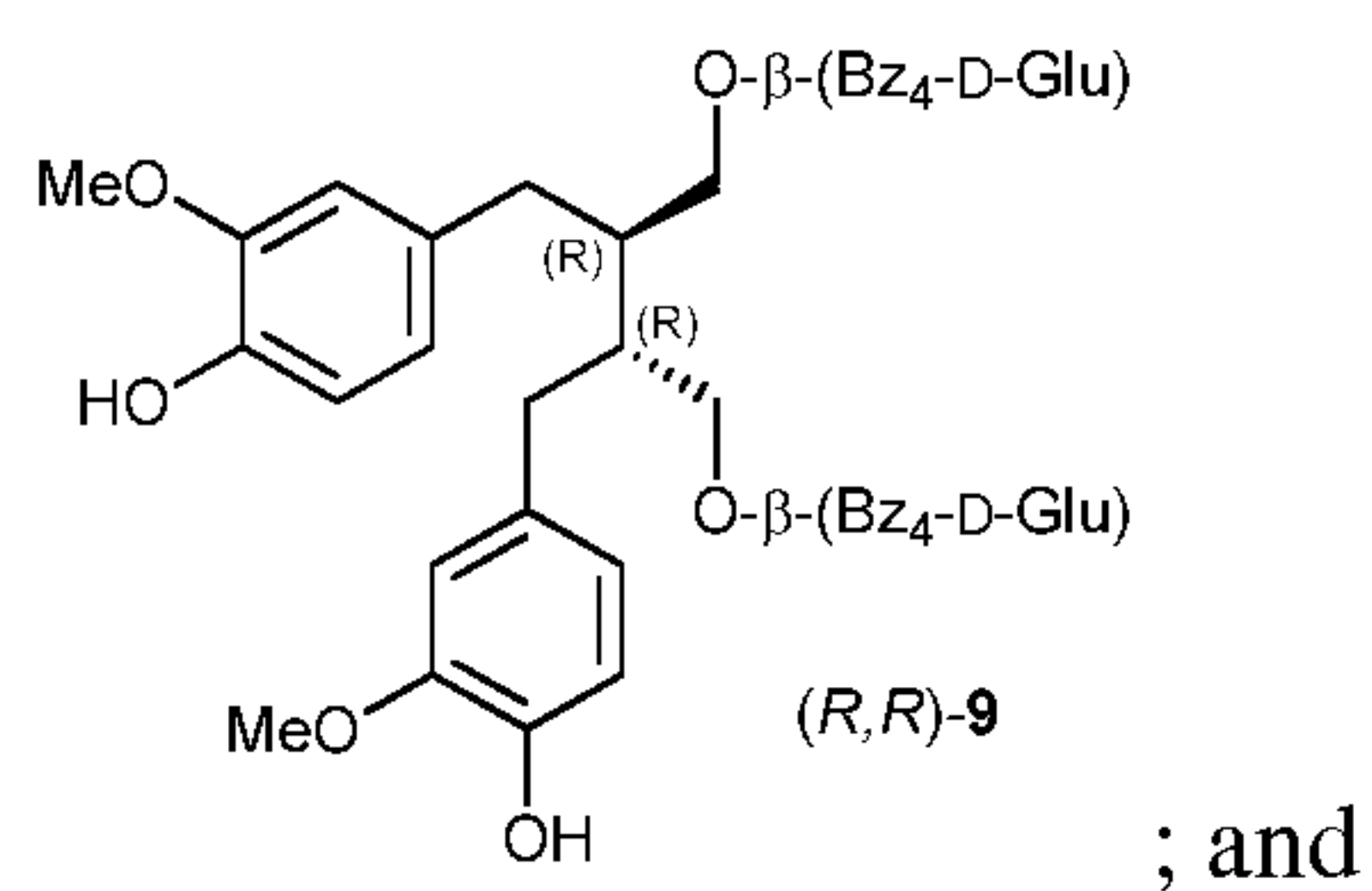
with a compound of formula (7)



to prepare a compound of formula ((*R,R*)-S4)



(b) cleaving benzyl ethers of the compound of formula ((*R,R*)-S4), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9)



(c) deprotecting the compound of formula ((*R,R*)-9) to prepare the compound of formula ((*R,R*)-SDG-2).

41. The process of claim 40, wherein said reacting is carried out in the presence of TMSOTf.

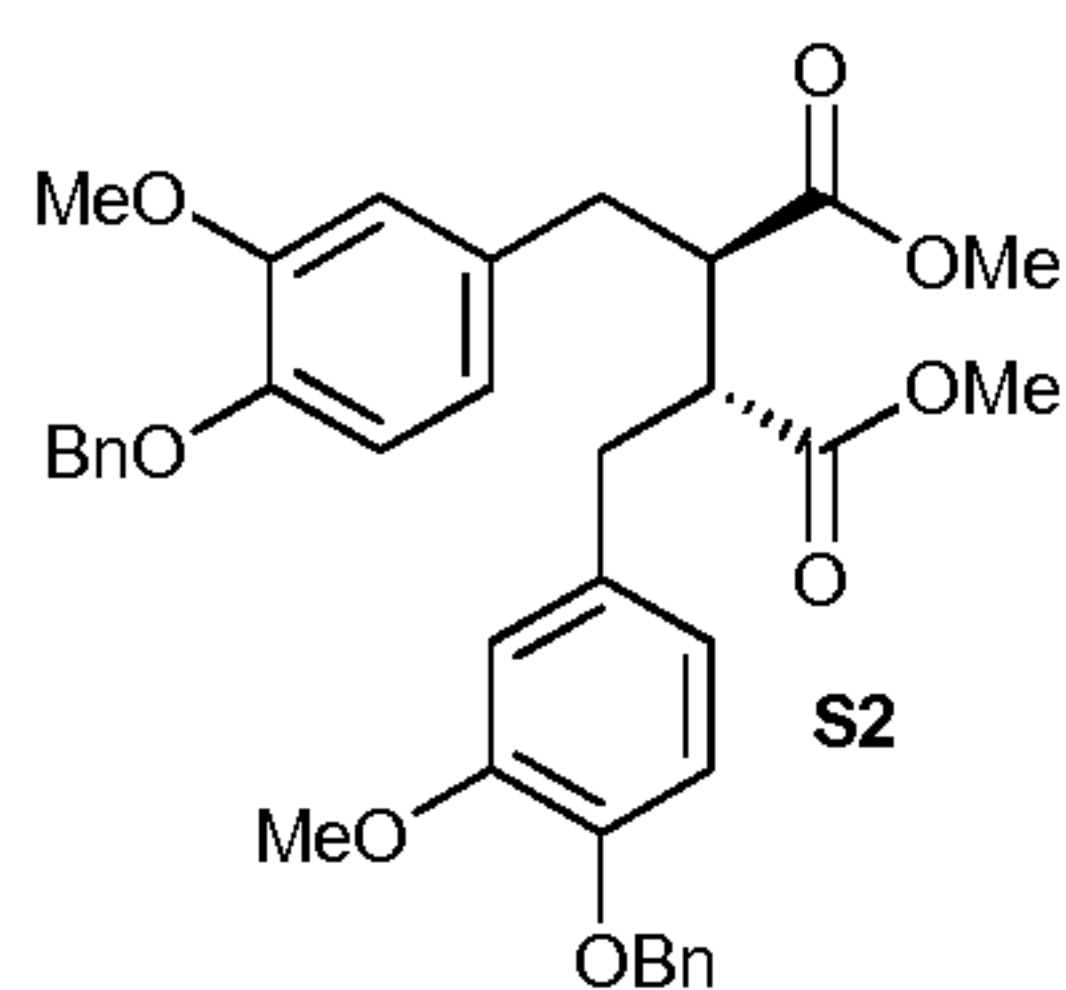
42. The process of claim 40, wherein said reacting is carried out in the presence of activated molecular sieves.

43. The process of claim 40, wherein said cleaving is carried out in the presence of H₂ and Pd/C in MeOH.

44. The process of claim 40, wherein said separation procedure is carried out using preparative thin layer chromatography.

45. The process of claim 40, wherein said deprotecting is carried out in a solution of NaOMe and MeOH.

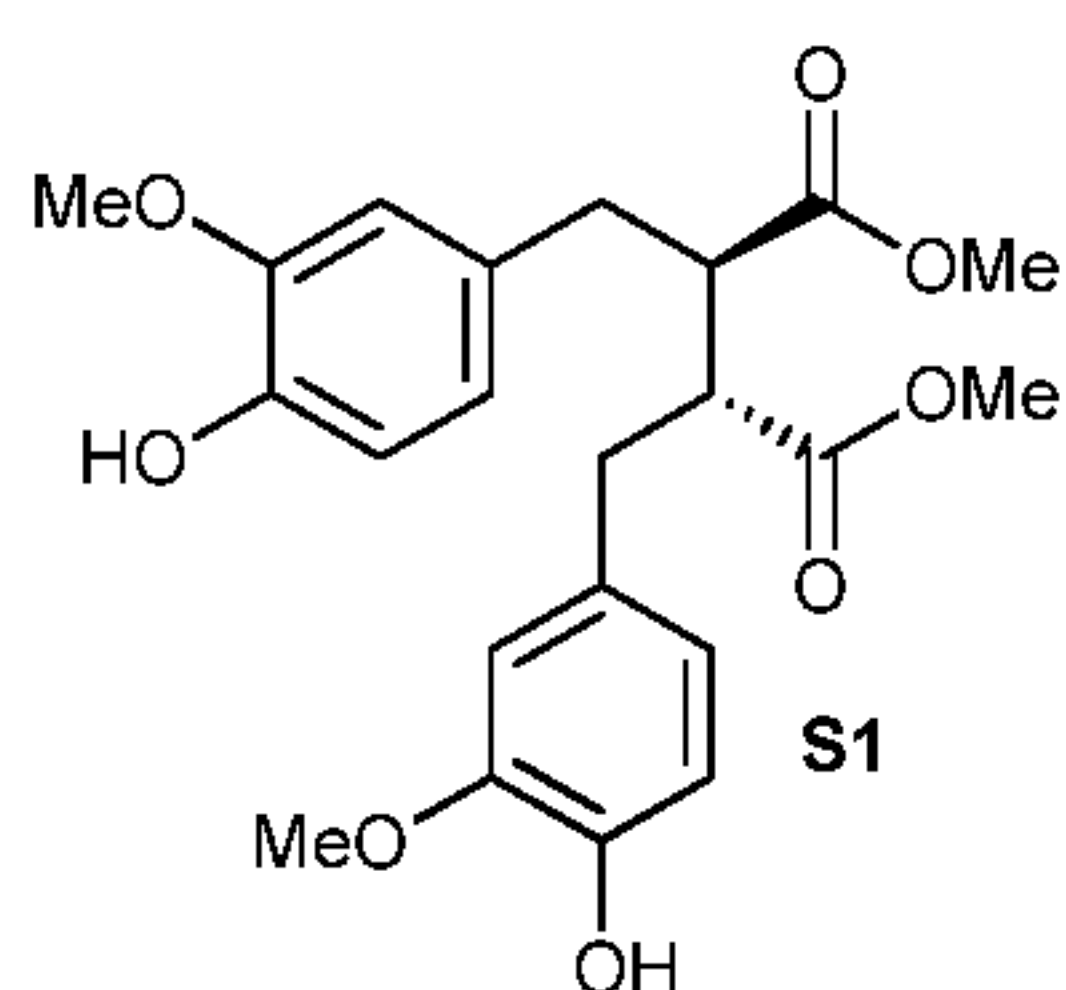
46. The process of claim 40, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

47. The process of claim 46, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

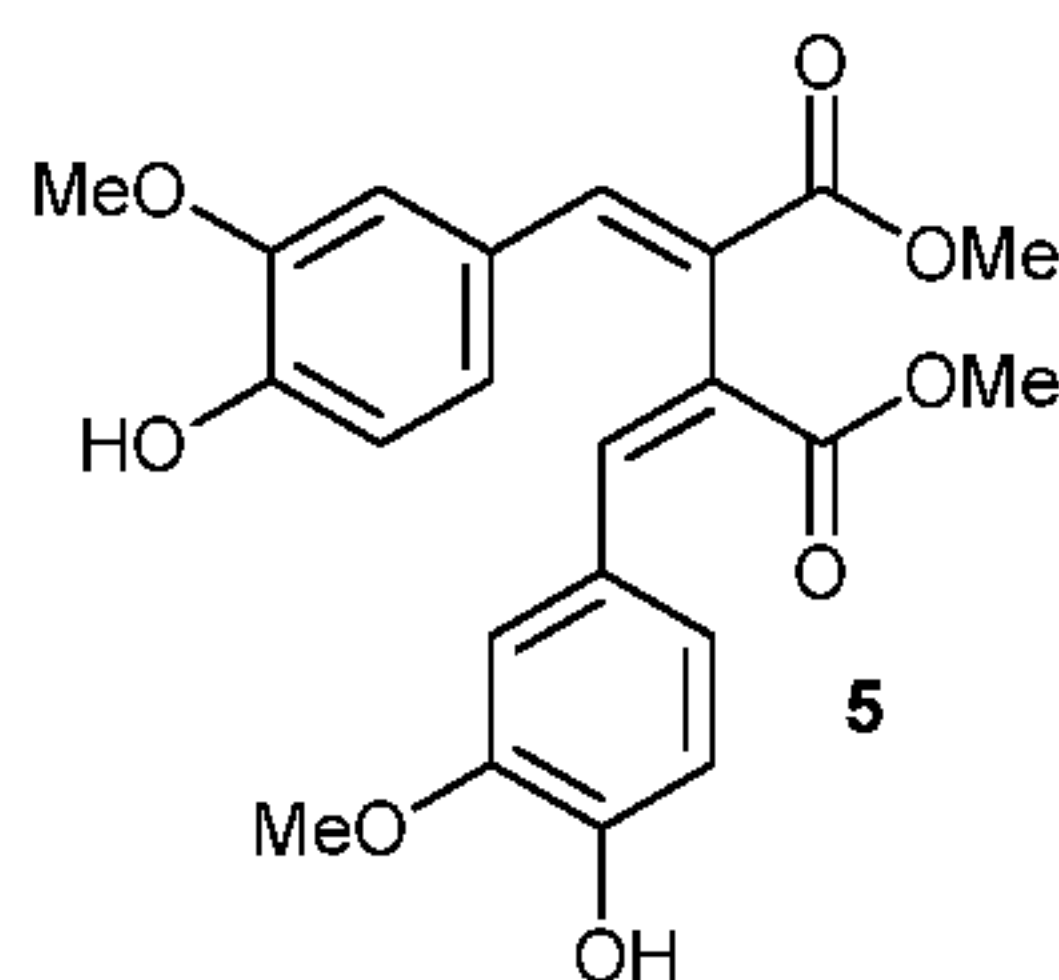
48. The process of claim 46, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

49. The process of claim 48, wherein said benzylating agent is BnBr and NaH.

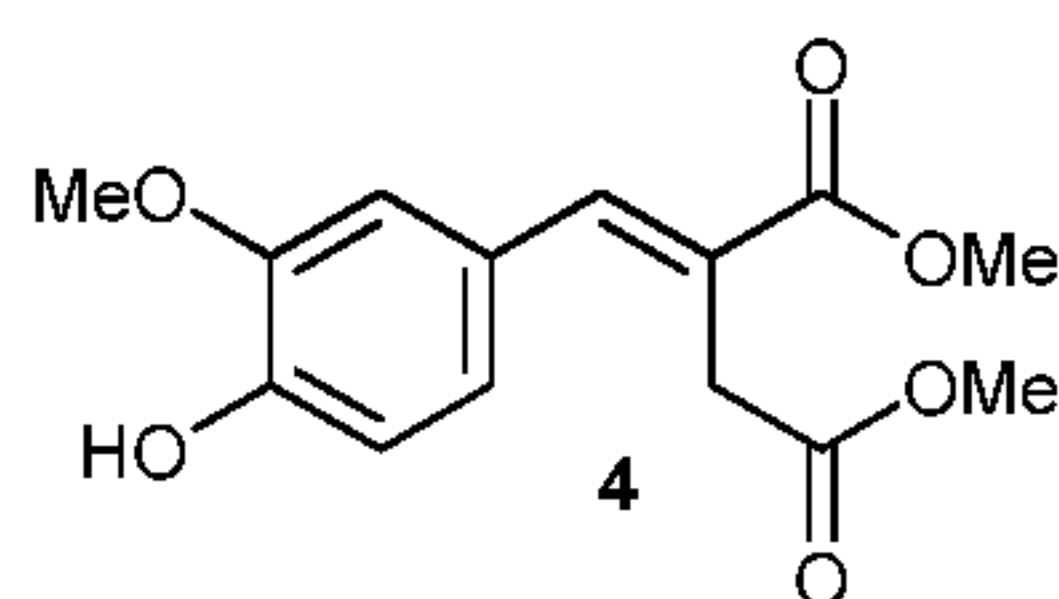
50. The process of claim 48, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

51. The process of claim 50, wherein said reducing agent is H_2 and Pd/C.

52. The process of claim 50, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

53. The process of claim 52, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium.

54. The process of claim 52, wherein said lithium is in the form of lithium wires.

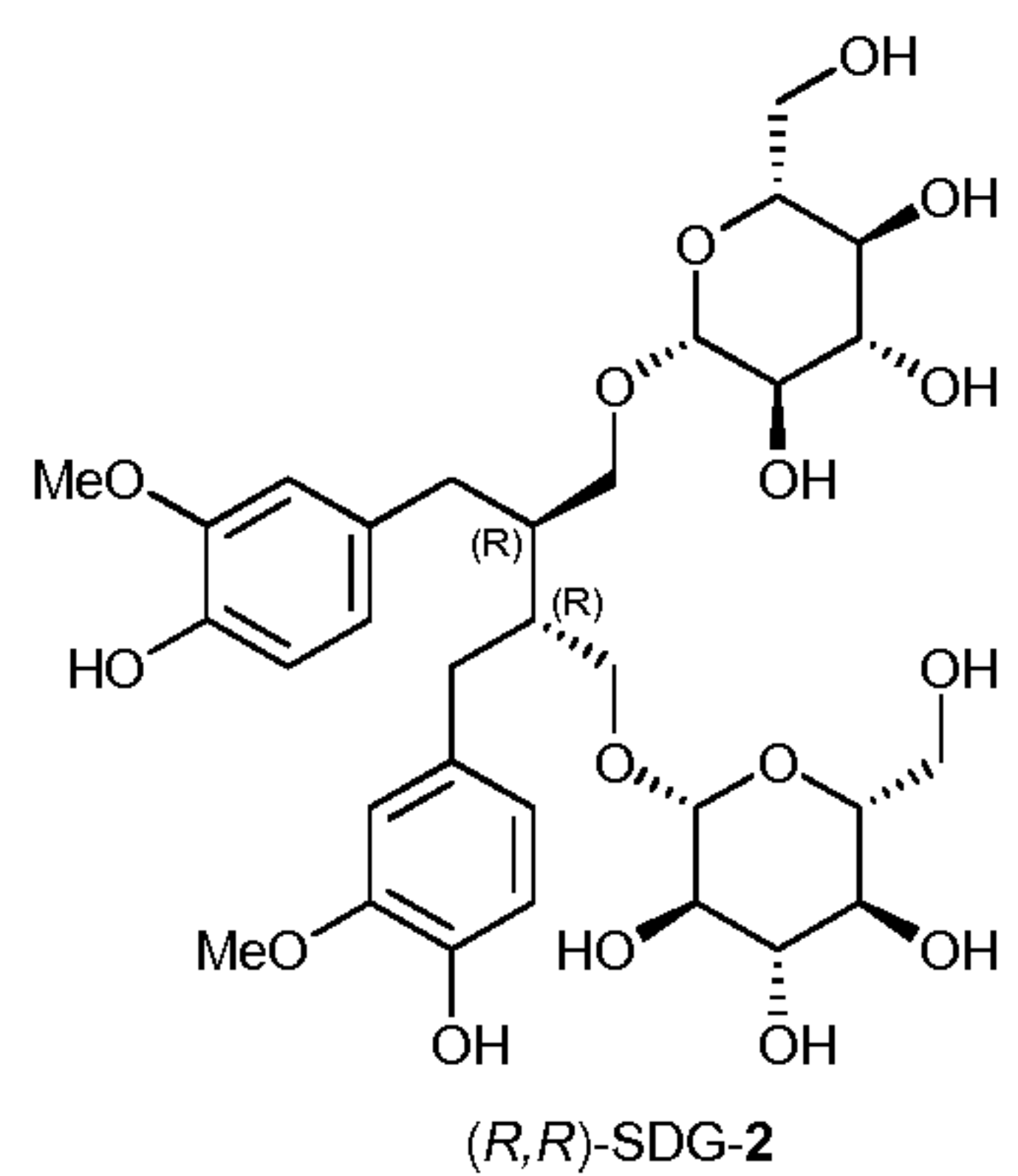
55. The process of claim 52, wherein said esterification reaction is carried out with MeOH in the presence of H_2SO_4 .

56. The process of claim 52, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

57. The process of claim 56, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

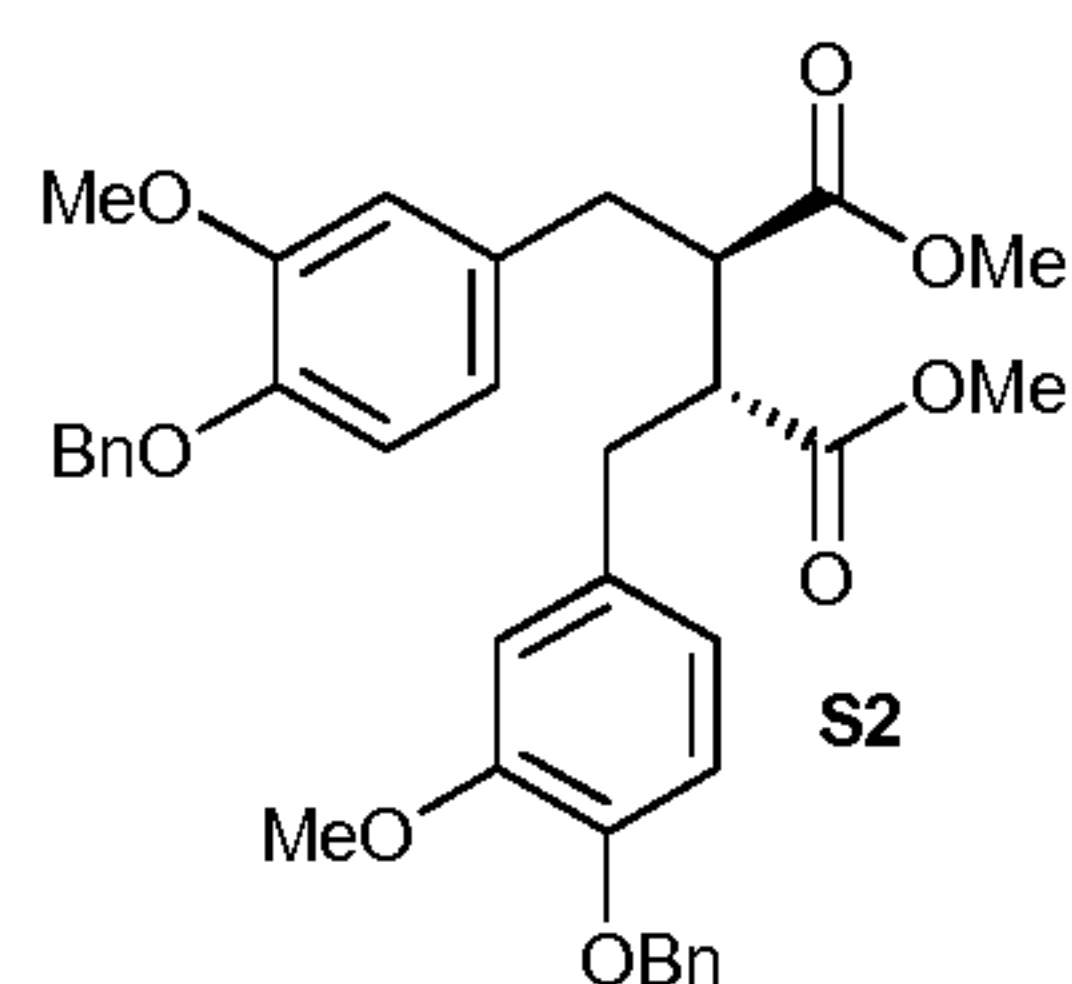
58. The process of claim 56, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

59. A process for preparing a compound of formula ((*R,R*)-SDG-2)

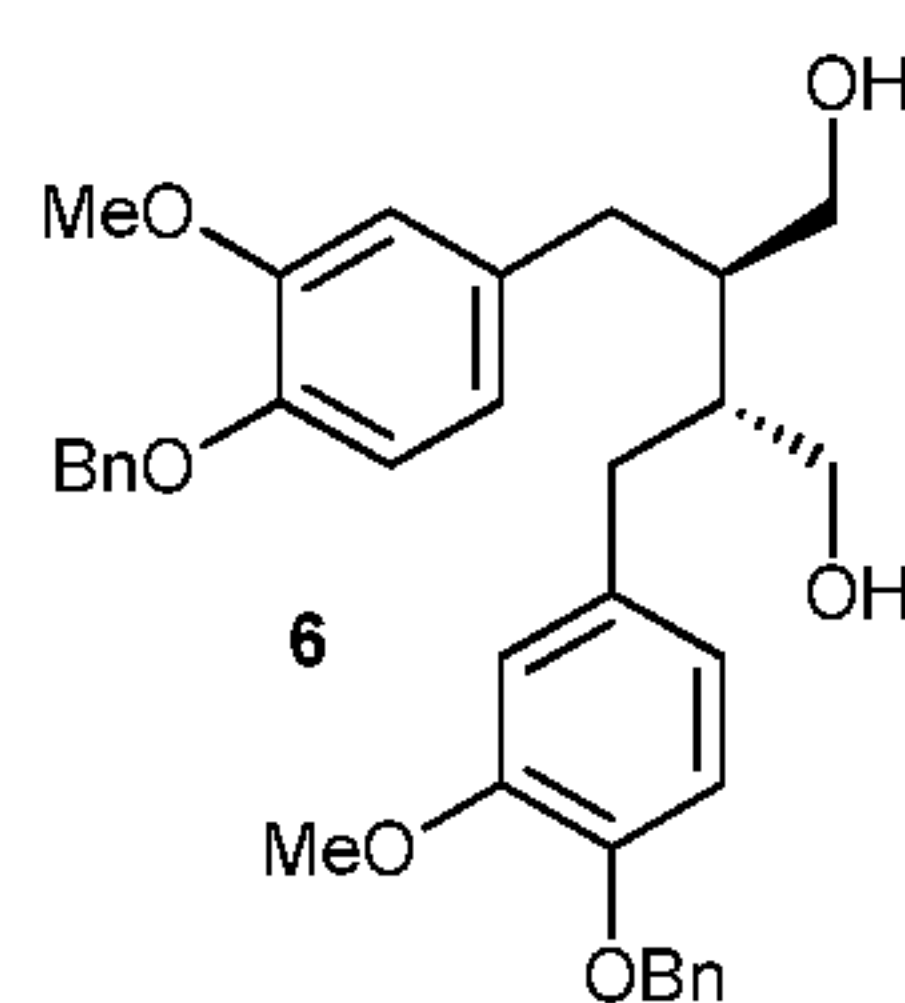


the process comprising:

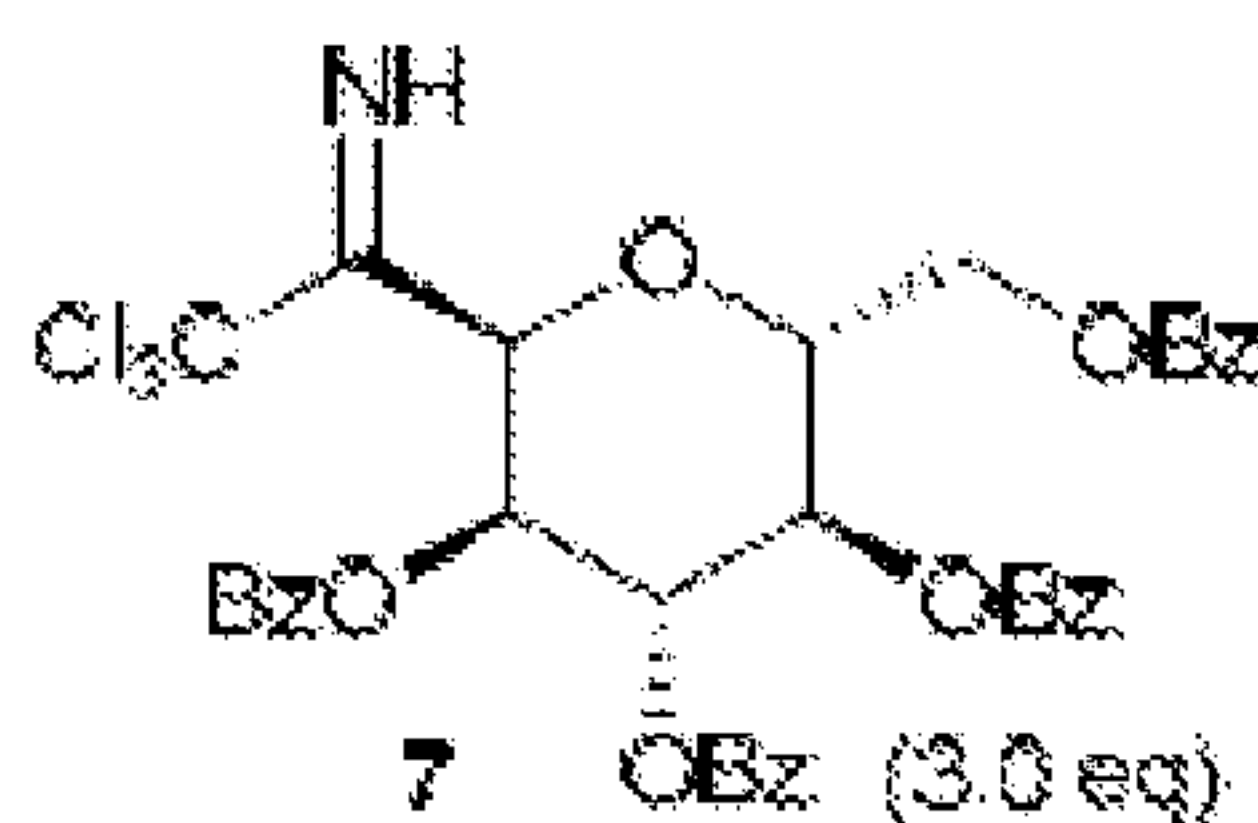
(a) reacting a compound of formula (S2)



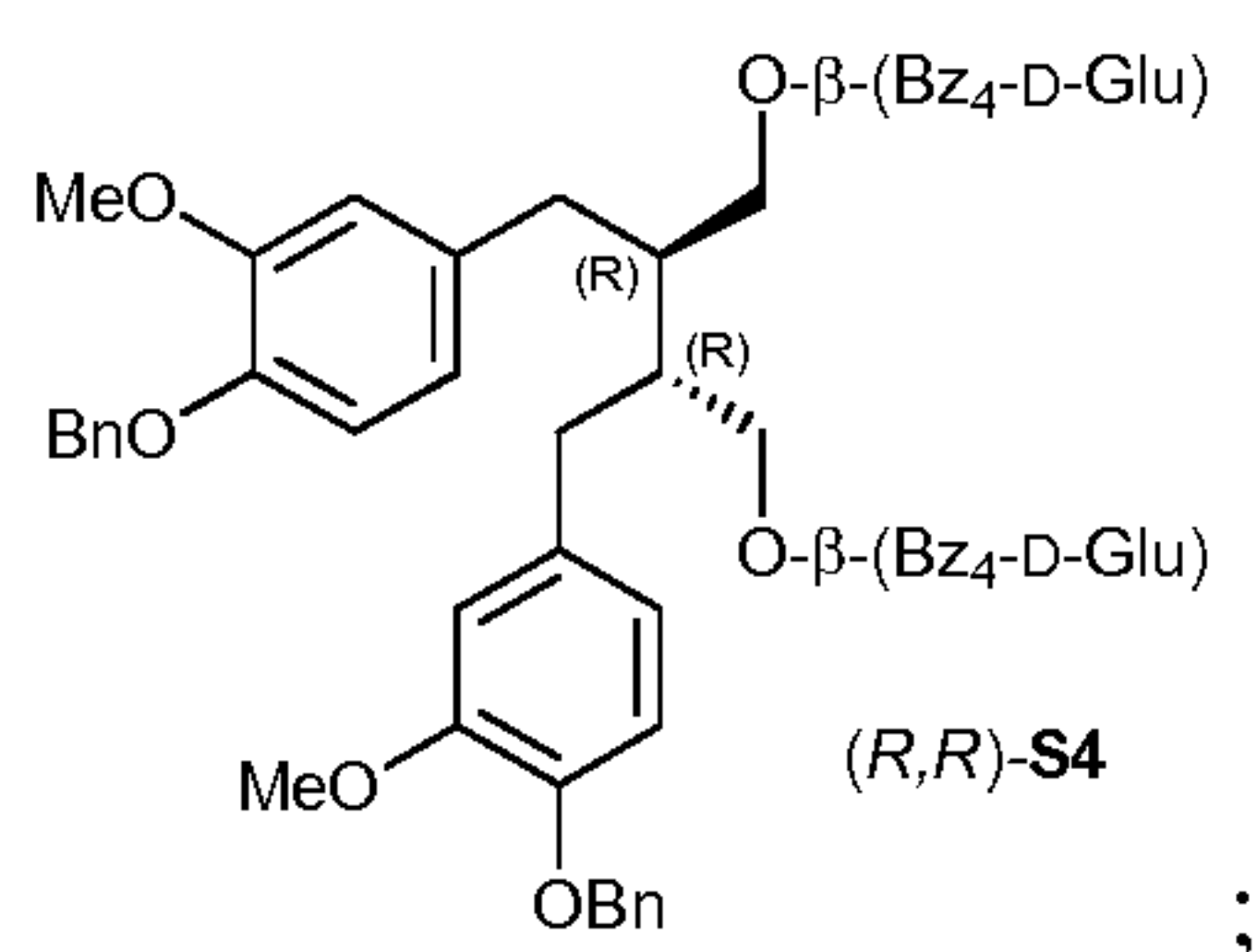
with a reducing agent to prepare a compound of formula (6)



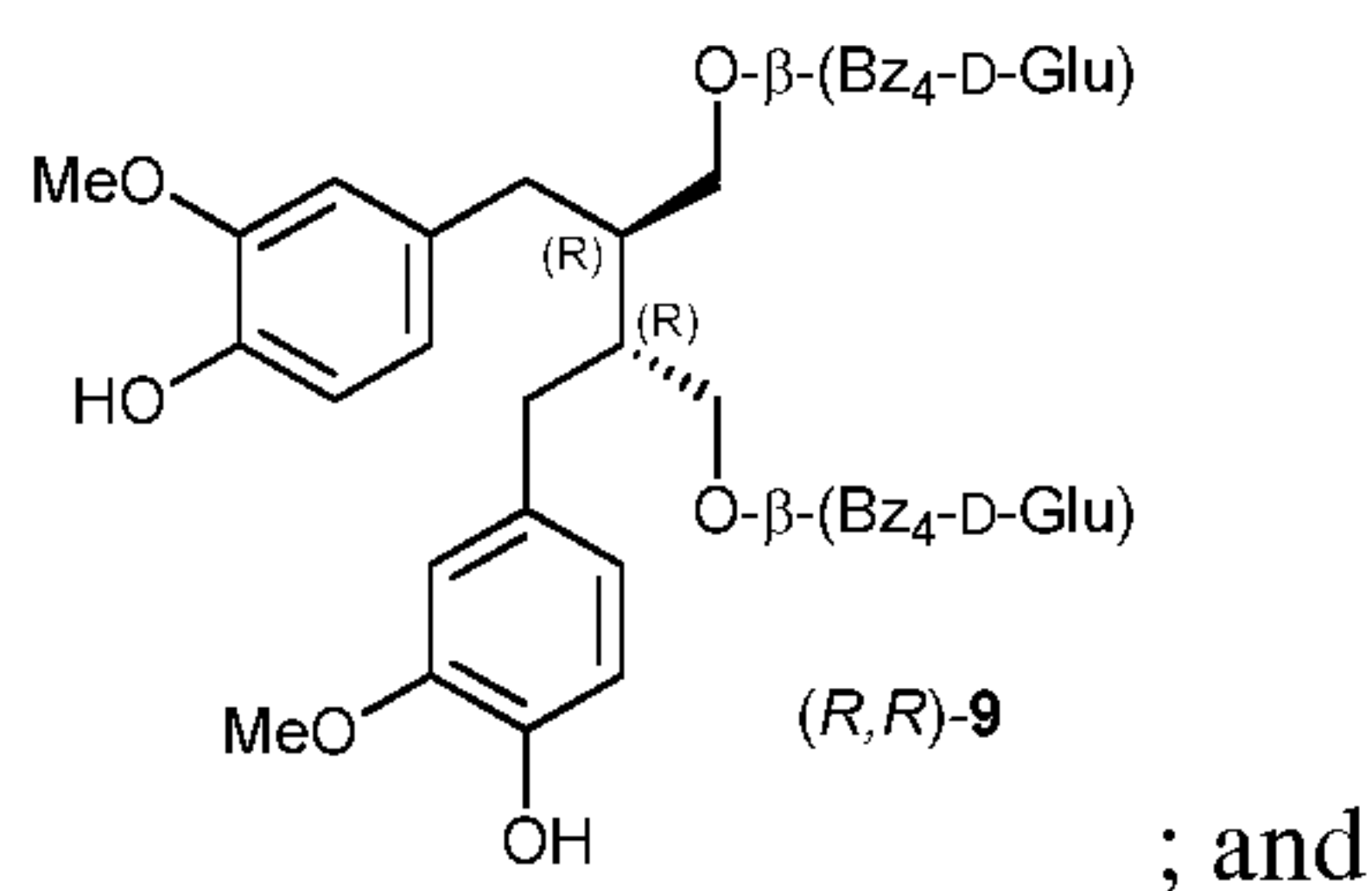
(b) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((*R,R*)-S4)



(c) cleaving benzyl ethers of the compound of formula ((R,R)-S4), followed by a separation procedure to prepare a compound of formula ((R,R)-9)



(d) deprotecting the compound of formula ((R,R)-9) to prepare the compound of formula ((R,R)-SDG-2).

60. The process of claim 59, wherein said reacting of step (a) is carried out in the presence of TMSOTf.

61. The process of claim 59, wherein said reacting of step (a) is carried out in the presence of activated molecular sieves.

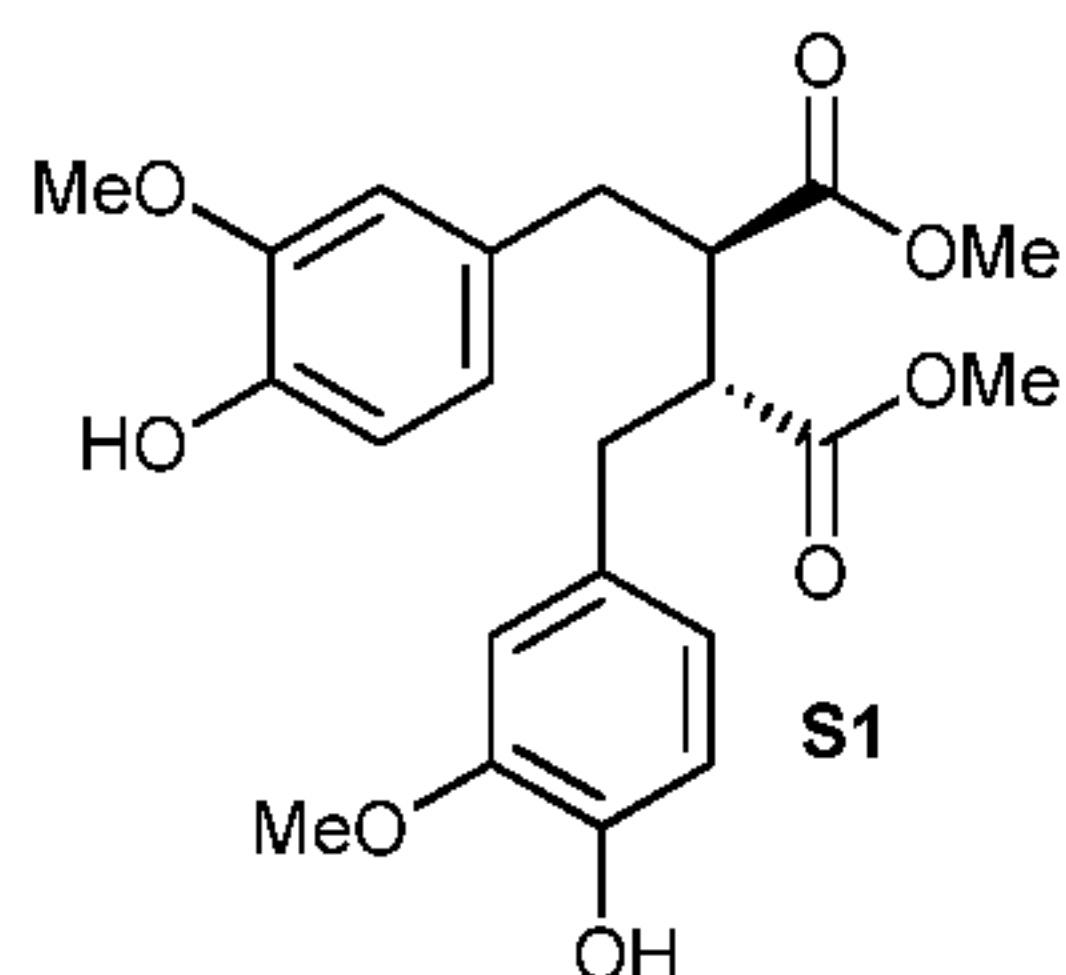
62. The process of claim 59, wherein said cleaving is carried out in the presence H₂ and Pd/C in MeOH.

63. The process of claim 59, wherein said separation procedure is carried out using preparative thin layer chromatography.

64. The process of claim 59, wherein said deprotecting is carried out in a solution of NaOMe and MeOH.

65. The process of claim 59, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

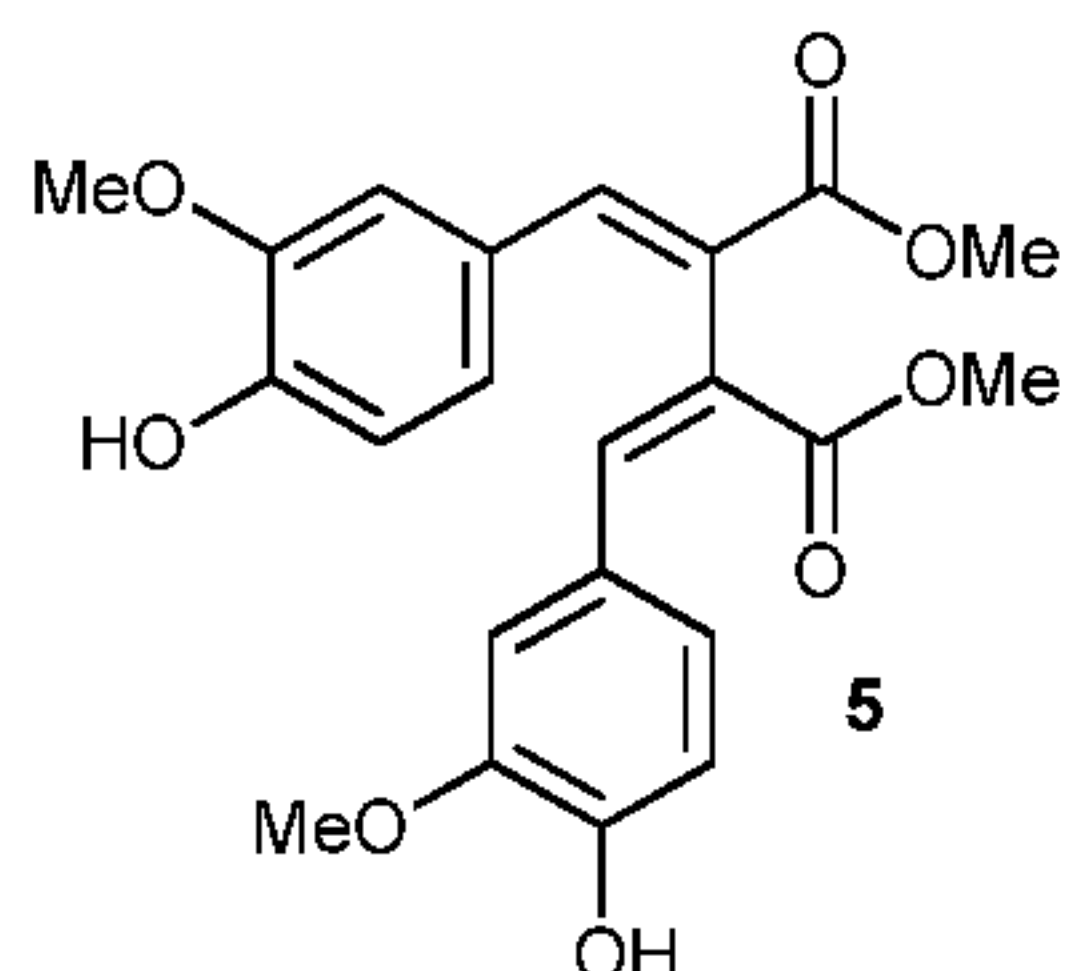
66. The process of claim 59, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to form the compound of formula (S2).

67. The process of claim 66, wherein said benzylating agent is BnBr and NaH.

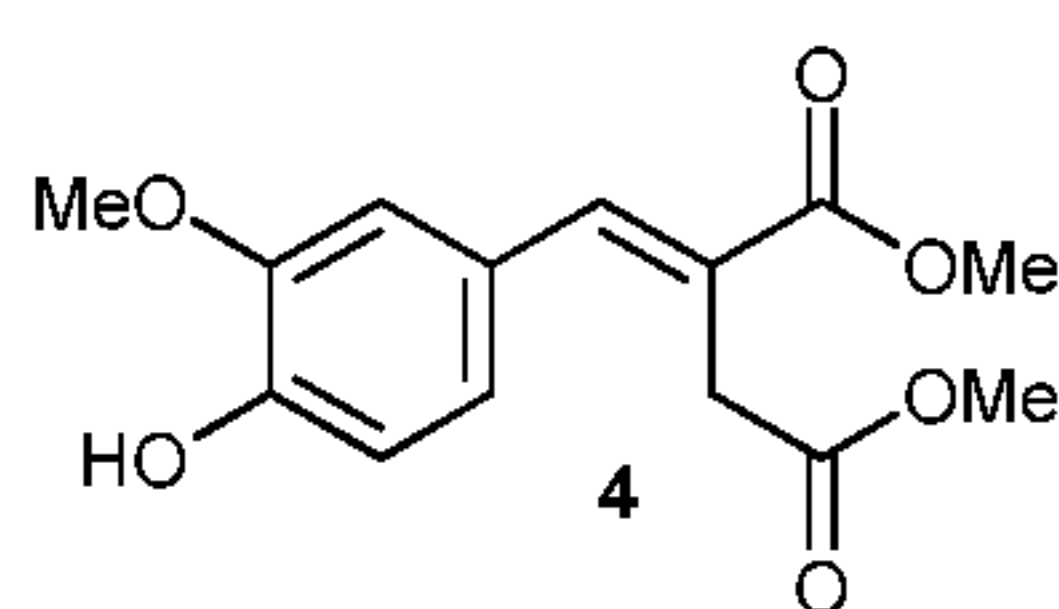
68. The process of claim 66, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

69. The process of claim 68, wherein said reducing agent is H₂ and Pd/C.

70. The process of claim 68, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

71. The process of claim 70, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

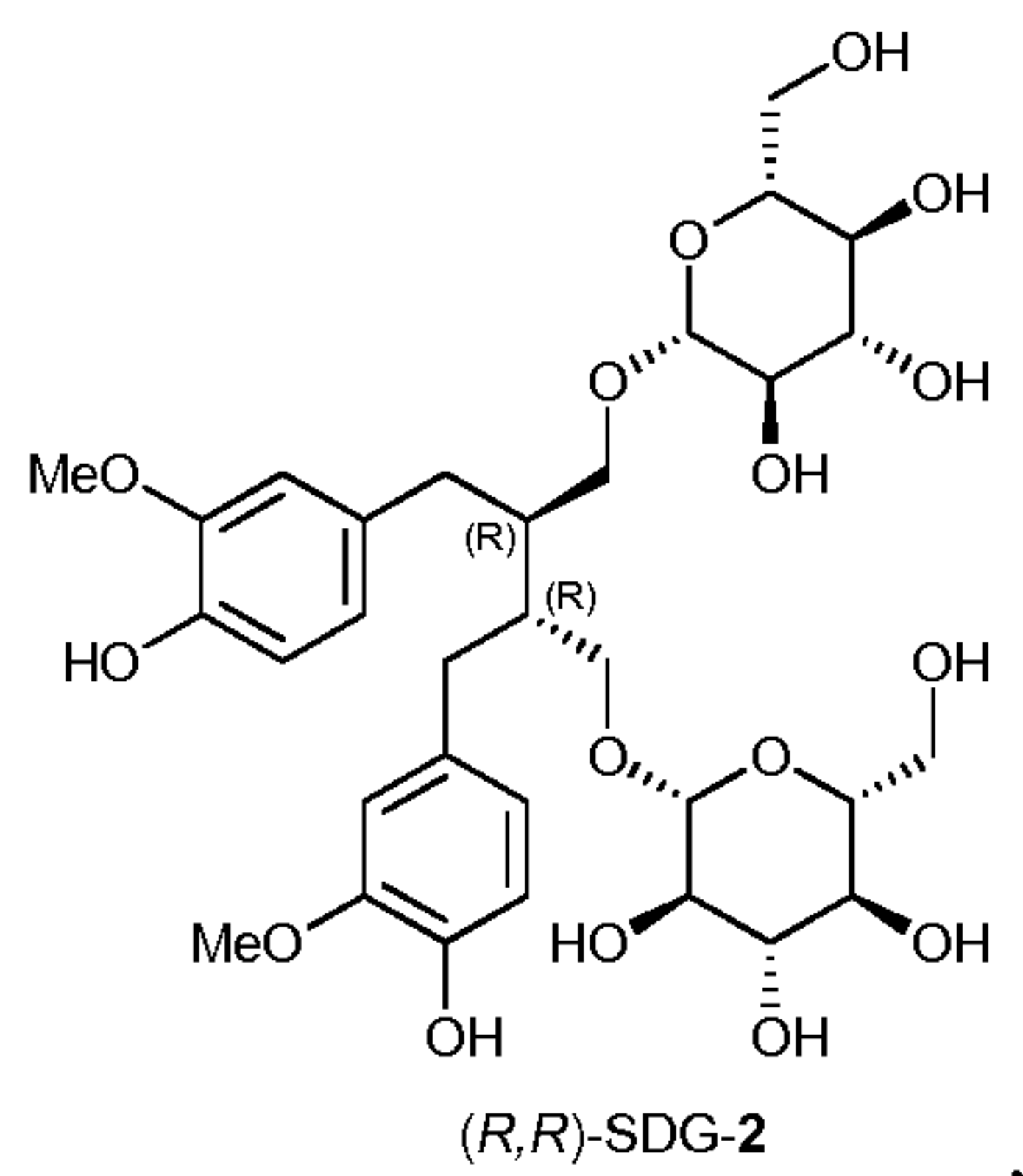
72. The process of claim 70, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

73. The process of claim 70, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

74. The process of claim 73, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

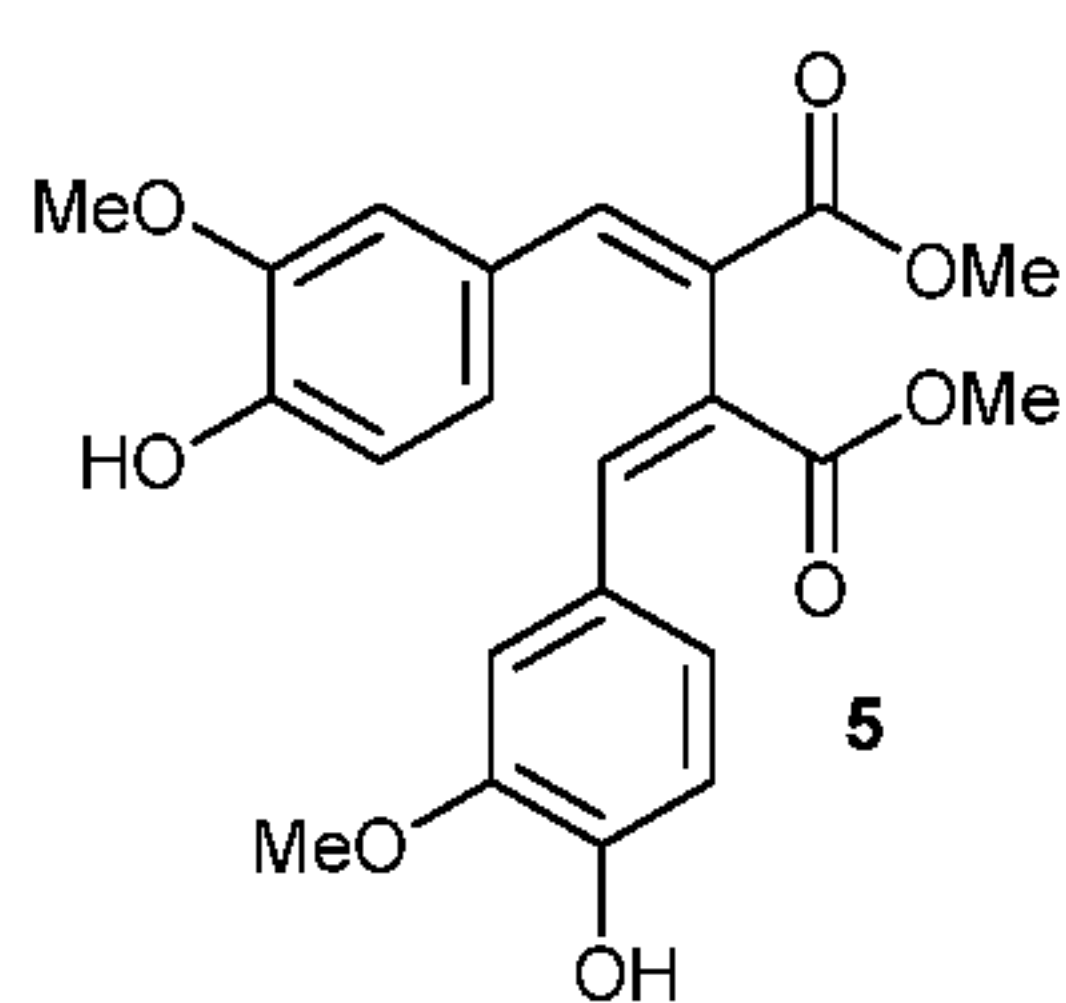
75. The process of claim 73, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

76. A process for preparing a compound of formula ((*R,R*)-SDG-2)

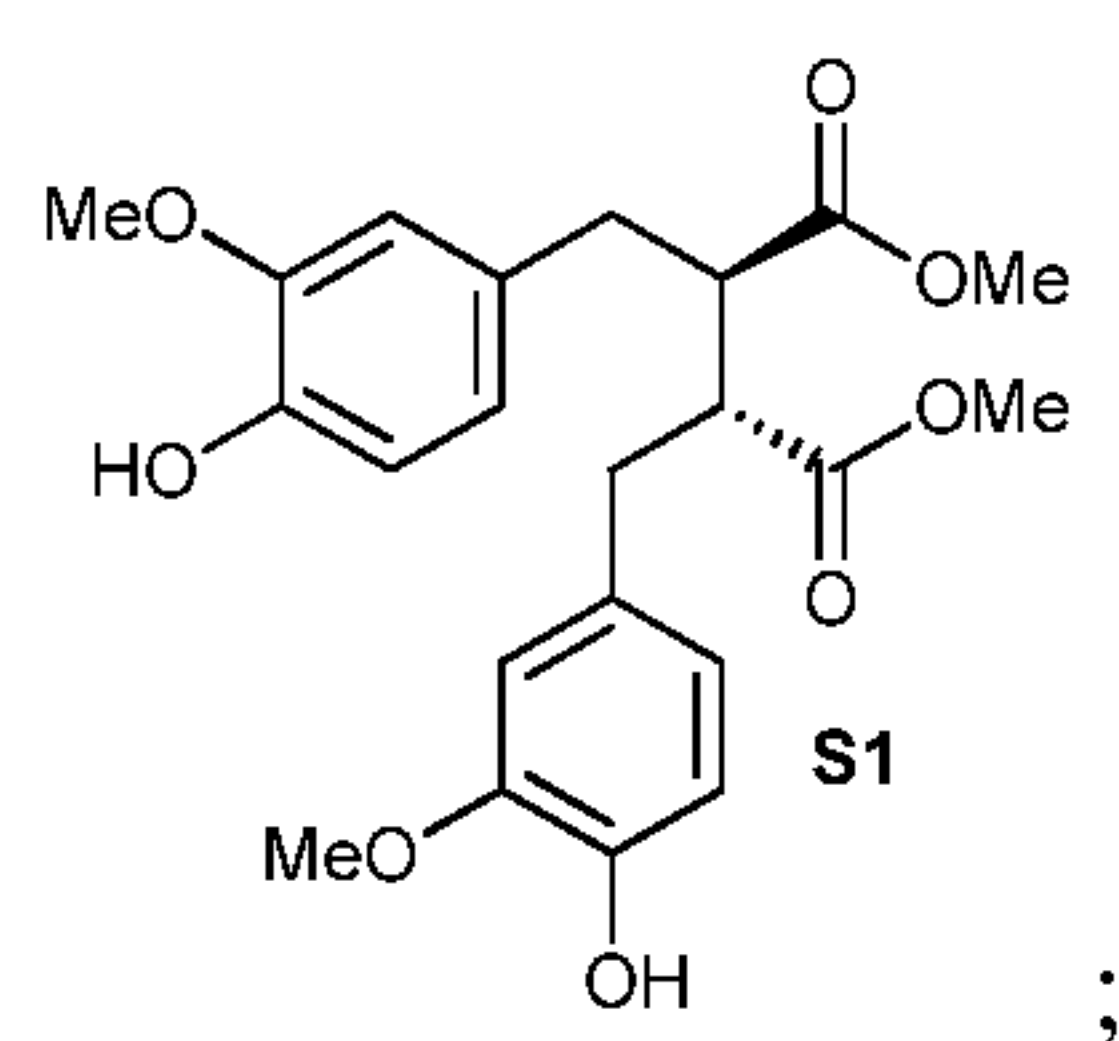


the process comprising:

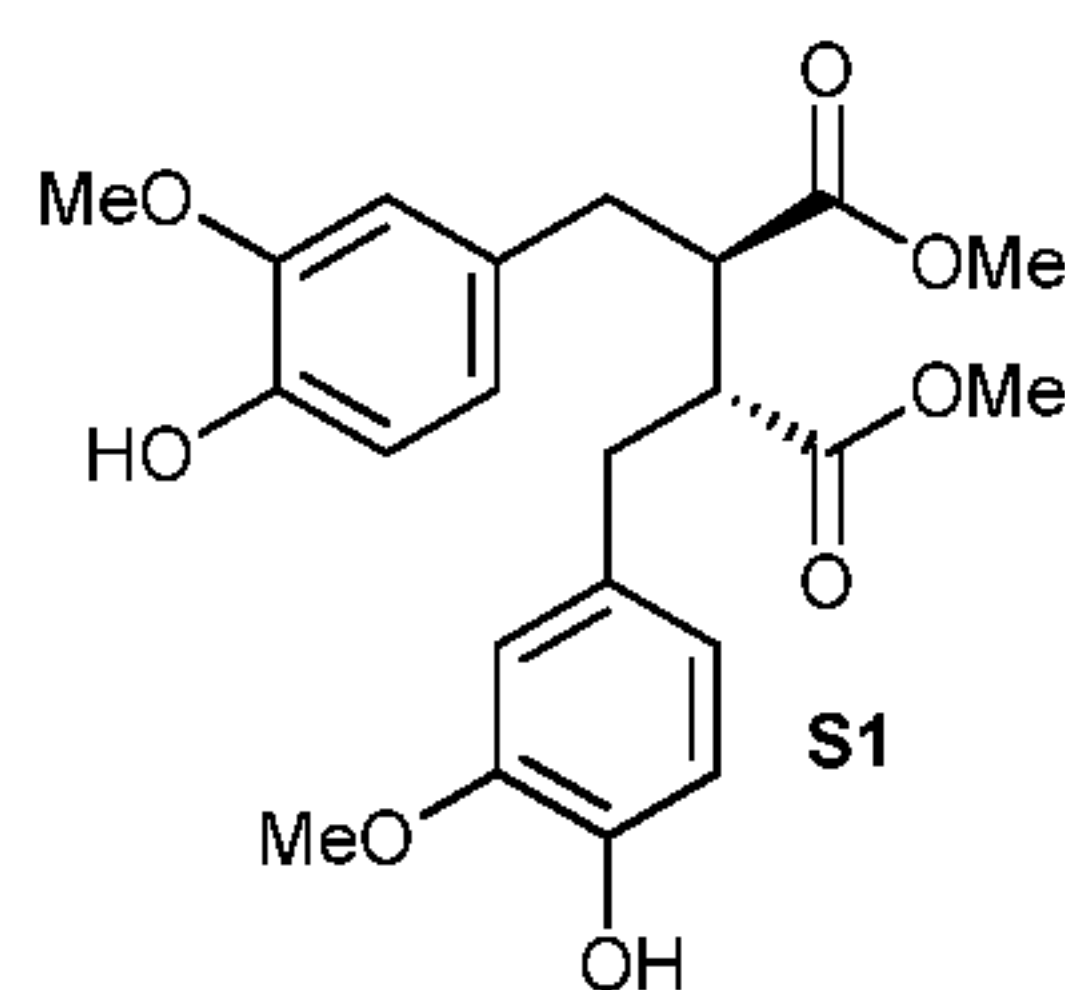
(a) reacting a compound of formula (5)



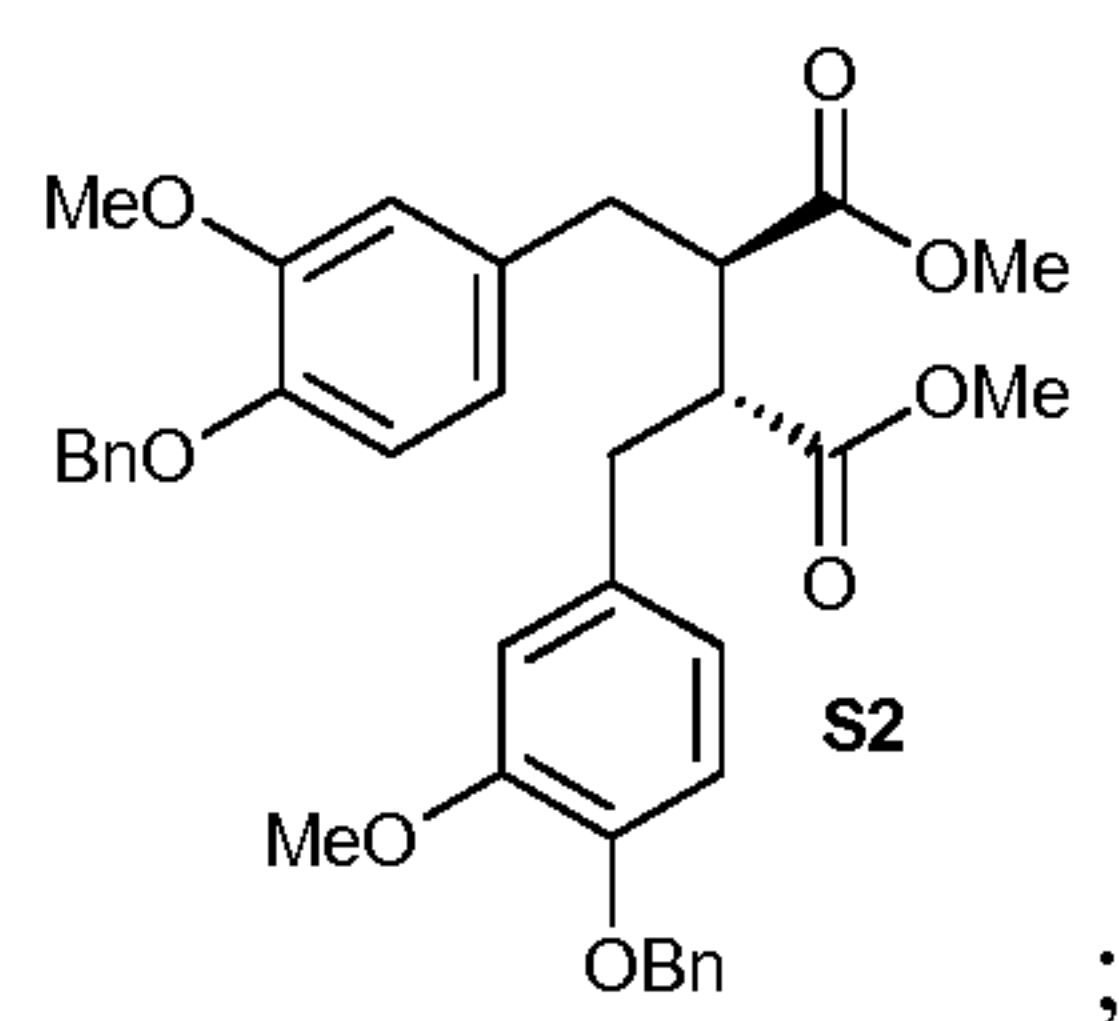
with a reducing agent to prepare a compound of formula (S1)



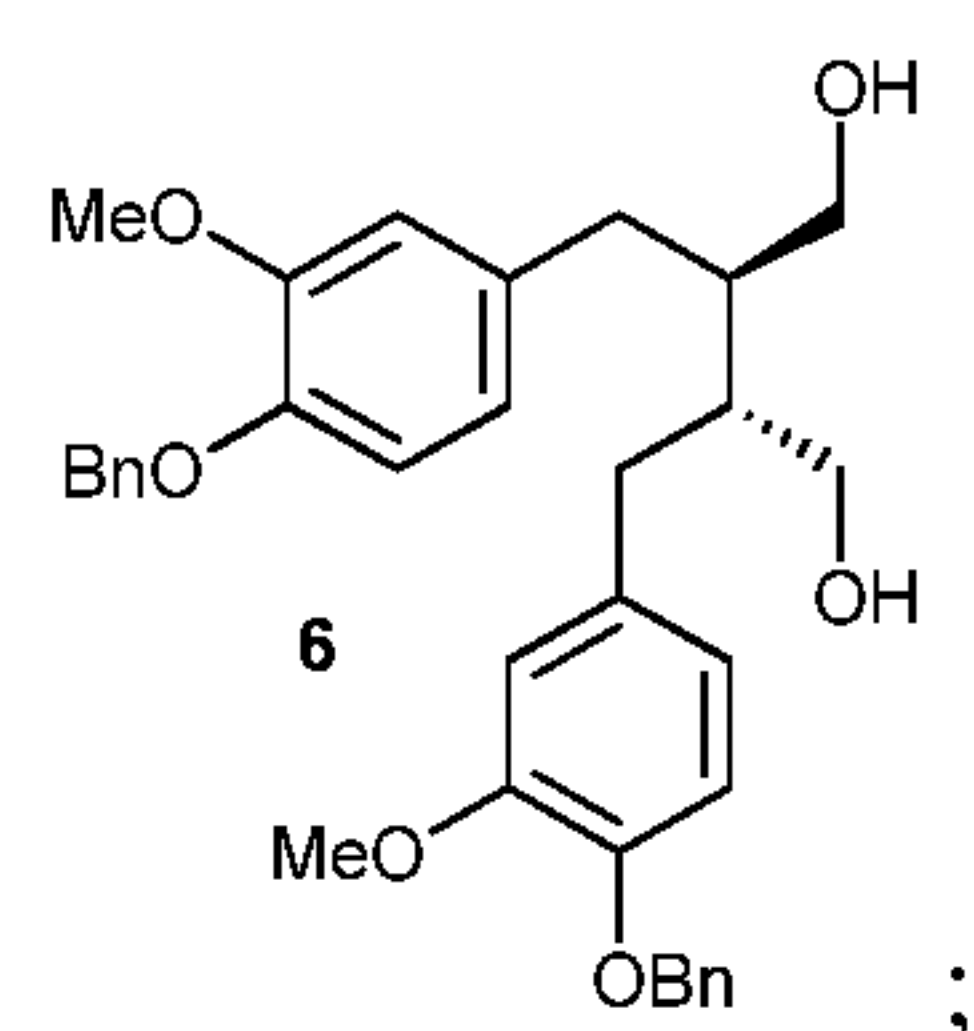
(b) reacting the compound of formula (S1)



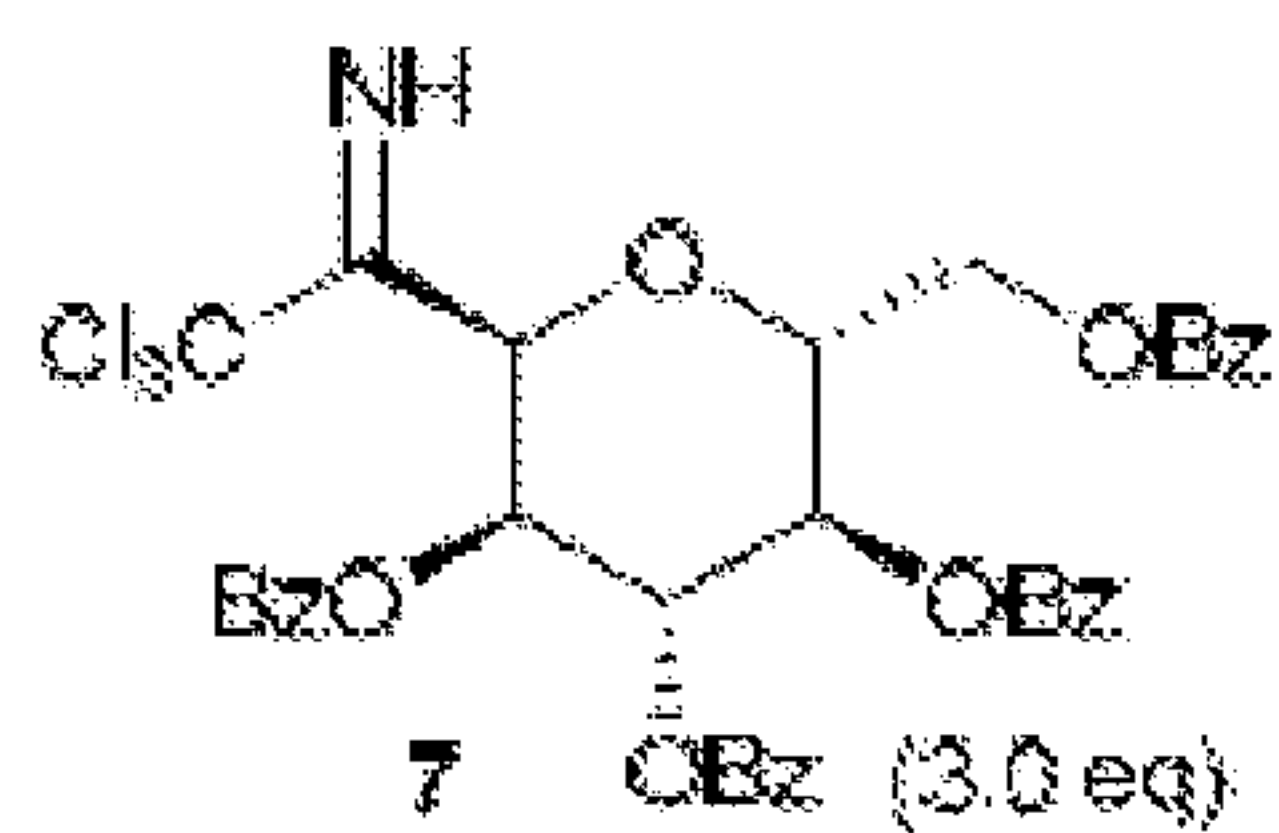
with a benzylating agent to prepare a compound of formula (S2)



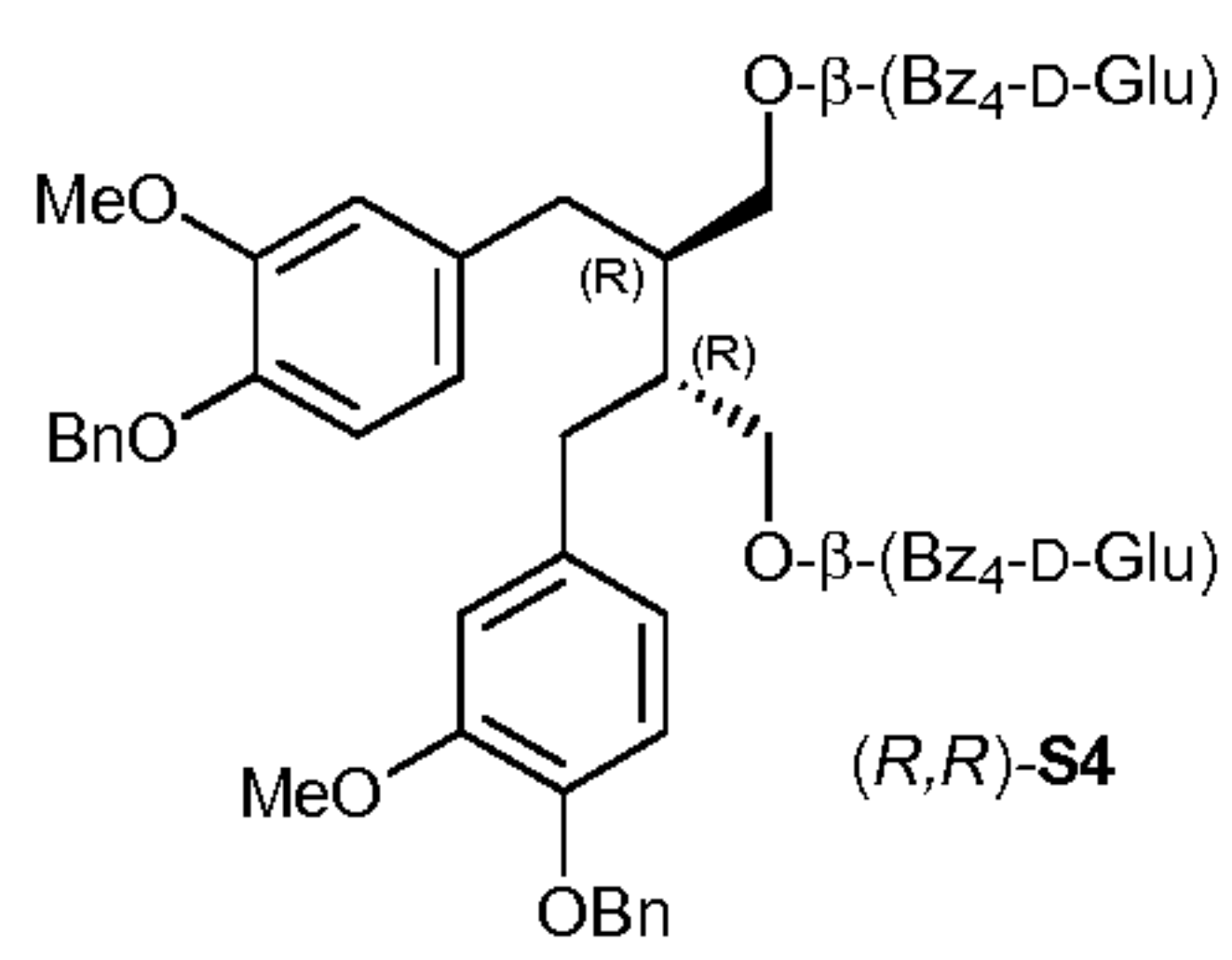
(c) reacting the compound of formula (S2) with a reducing agent to prepare a compound of formula (6)



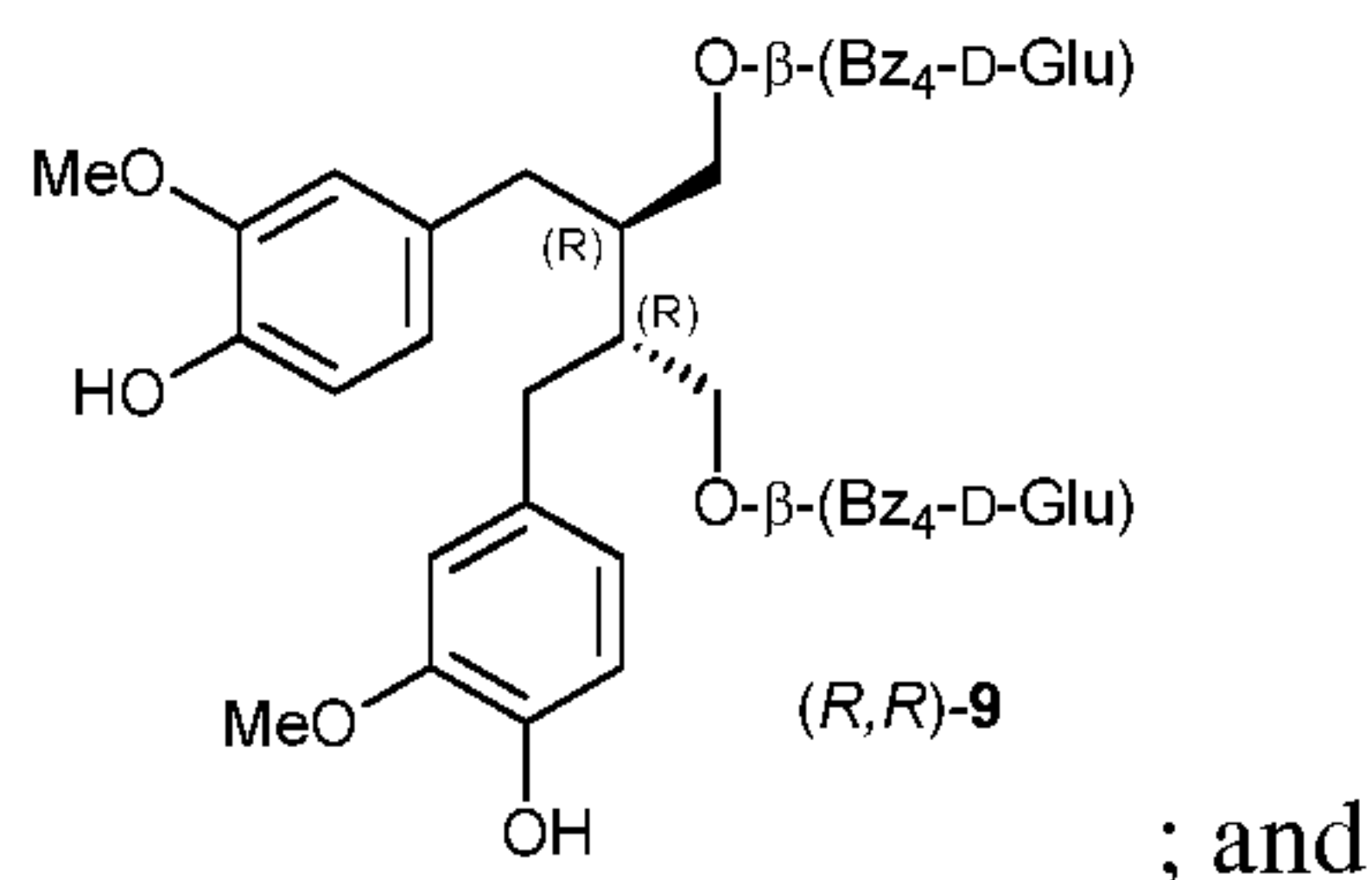
(d) reacting the compound of formula (6) with a compound of formula (7)



to prepare a compound of formula ((R,R)-S4)

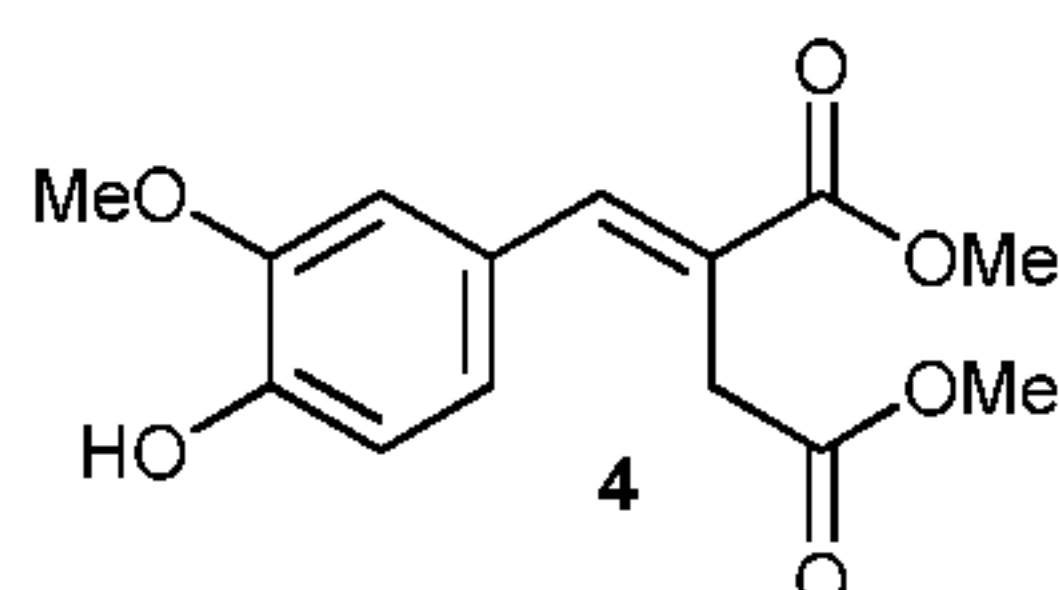


(e) cleaving benzyl ethers of the compound of formula ((*R,R*)-S3), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9)



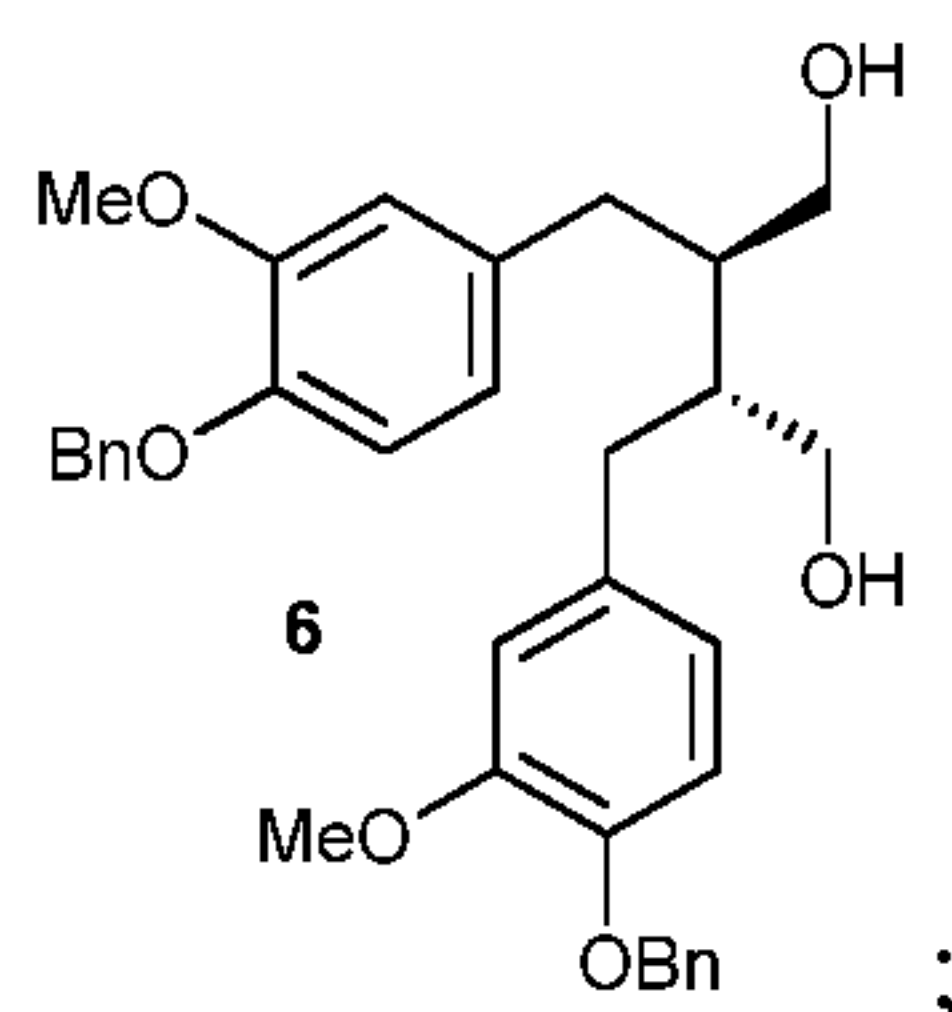
(f) deprotecting the compound of formula ((*R,R*)-9) to prepare the compound of formula ((*R,R*)-SDG-2).

77. The process of claim 76, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)

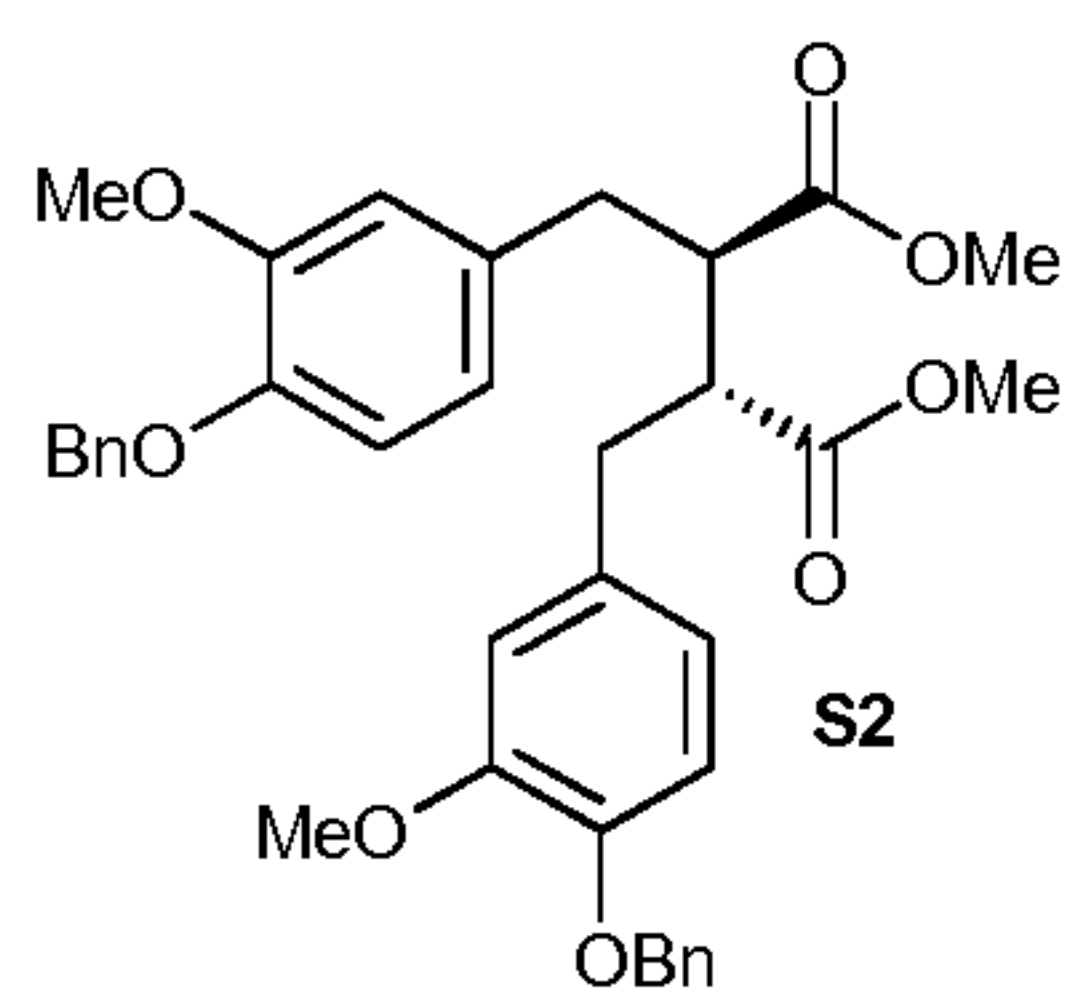


with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

78. A process for preparing a compound of formula (6)



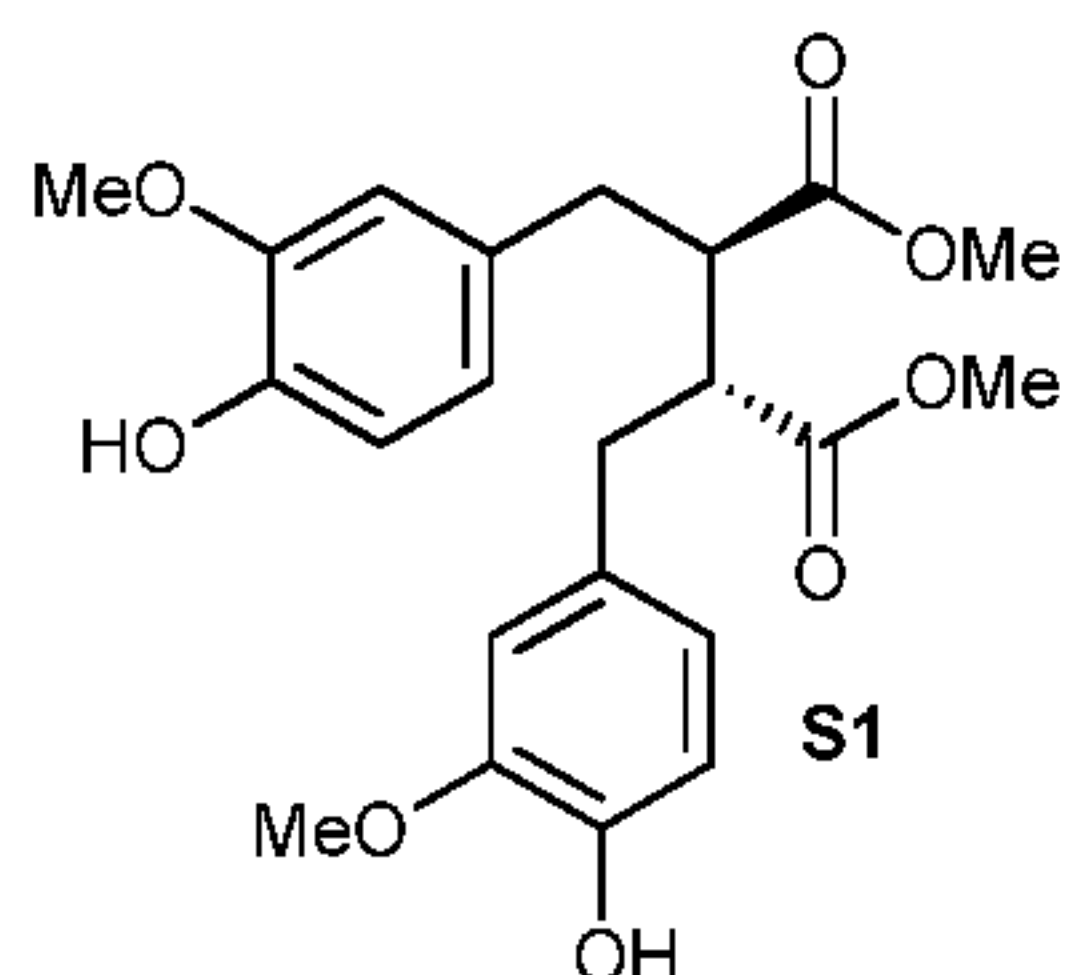
the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

79. The process of claim 78, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

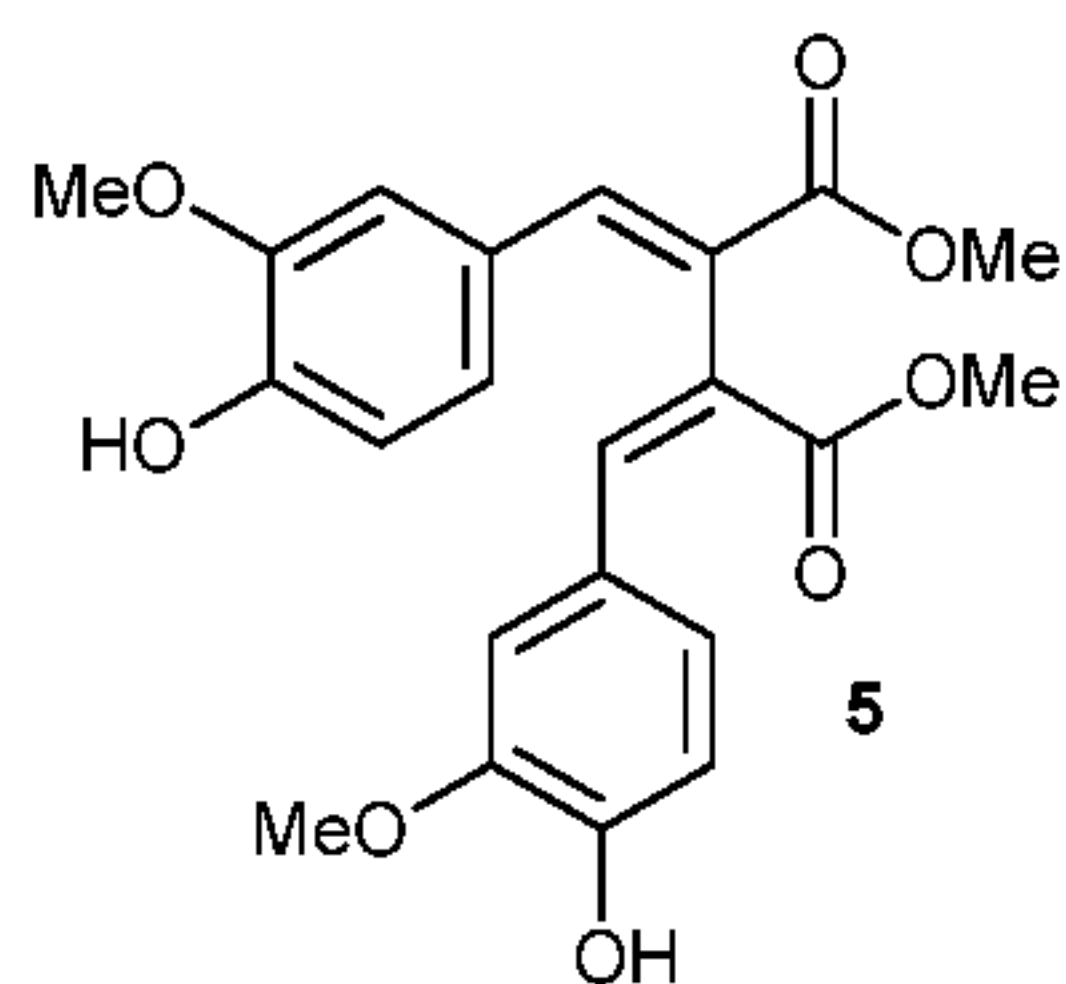
80. The process of claim 78, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

81. The process of claim 80, wherein said benzylating agent is BnBr and NaH.

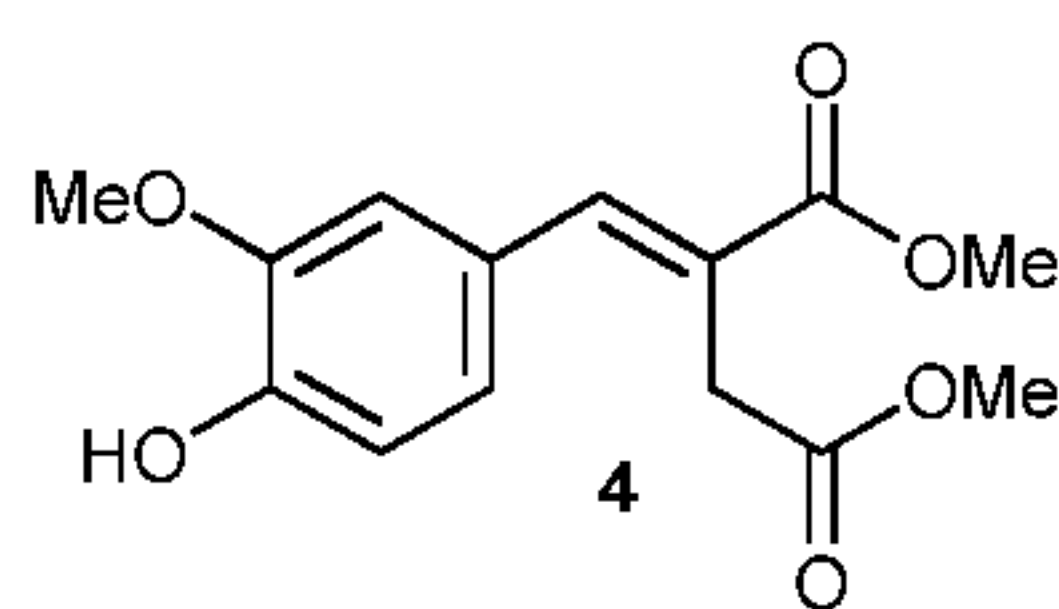
82. The process of claim 80, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

83. The process of claim 82, wherein said reducing agent is H₂ and Pd/C.

84. The process of claim 82, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

85. The process of claim 84, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

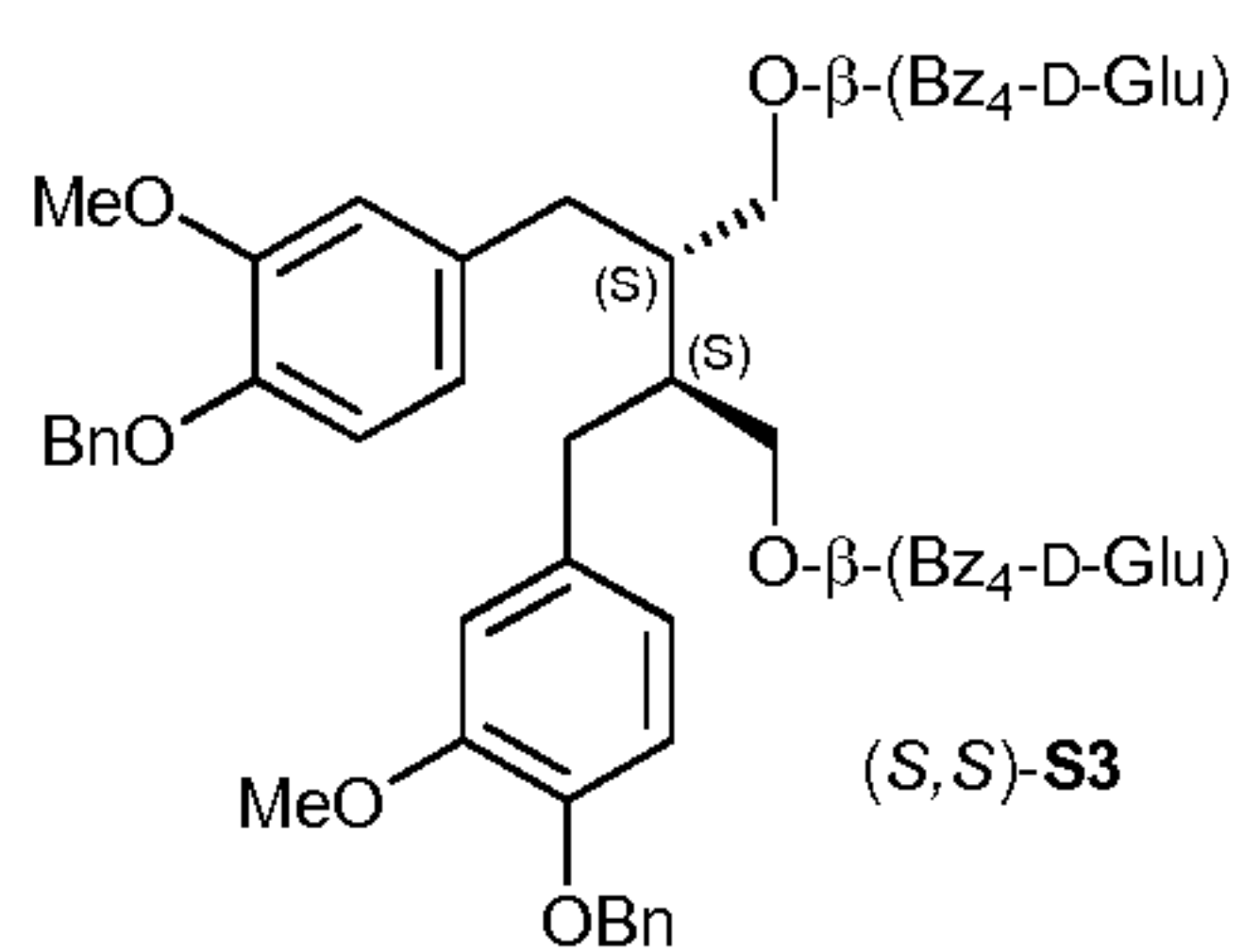
86. The process of claim 84, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

87. The process of claim 84, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

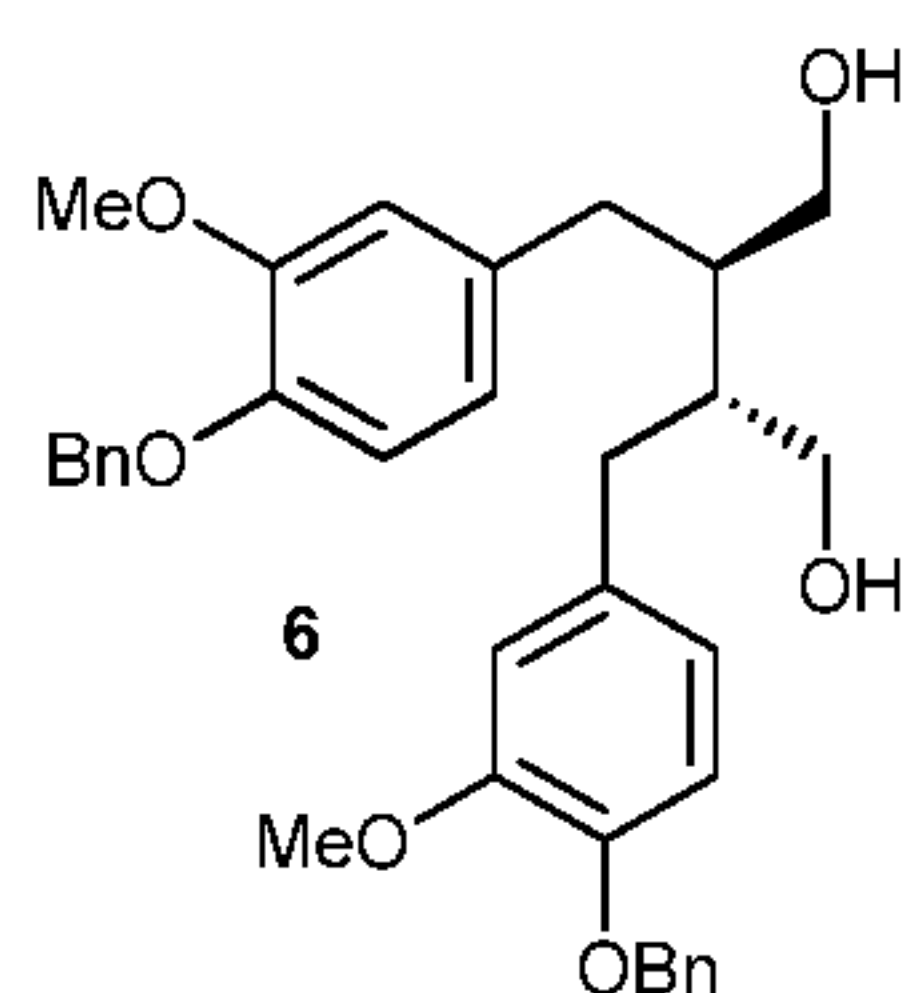
88. The process of claim 87, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

89. The process of claim 87, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

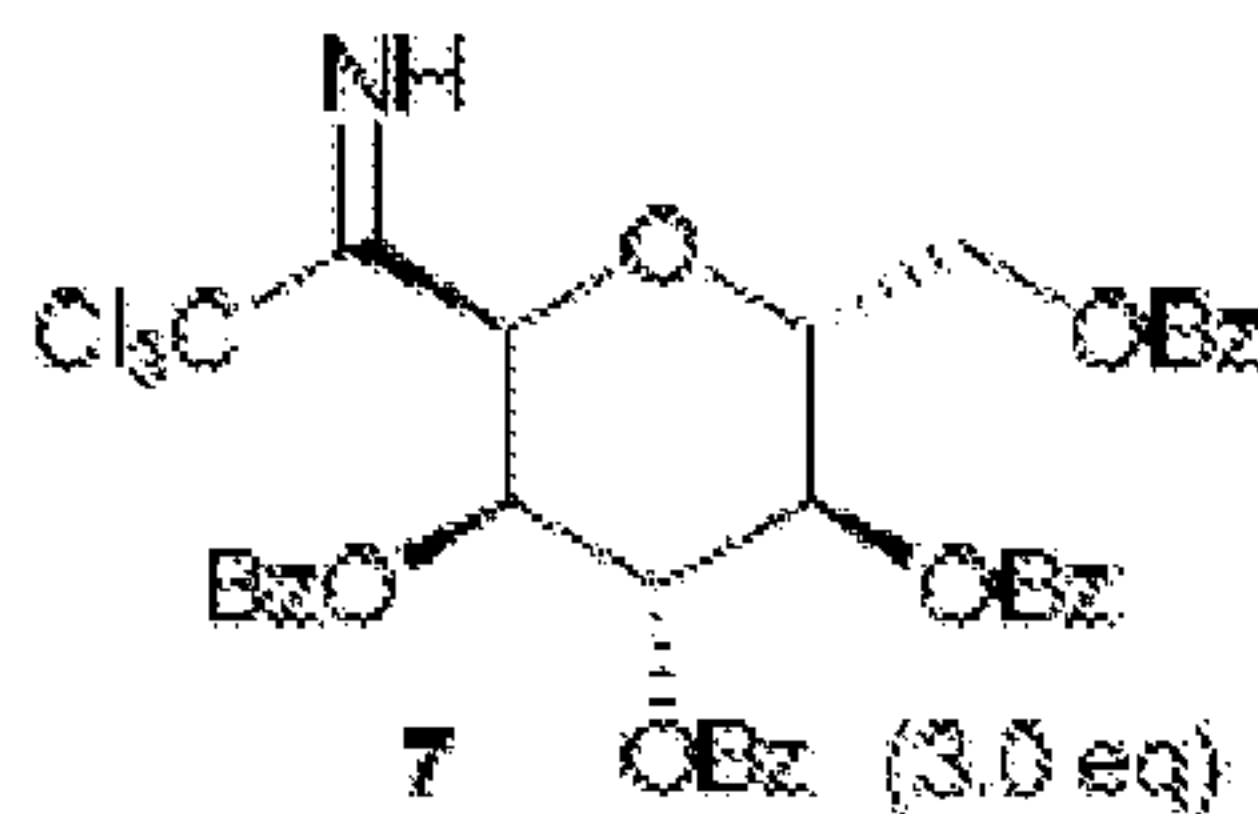
90. A process for preparing a compound of formula ((S,S)-S3)



the process comprising reacting a compound of formula (6)



with a compound of formula (7)

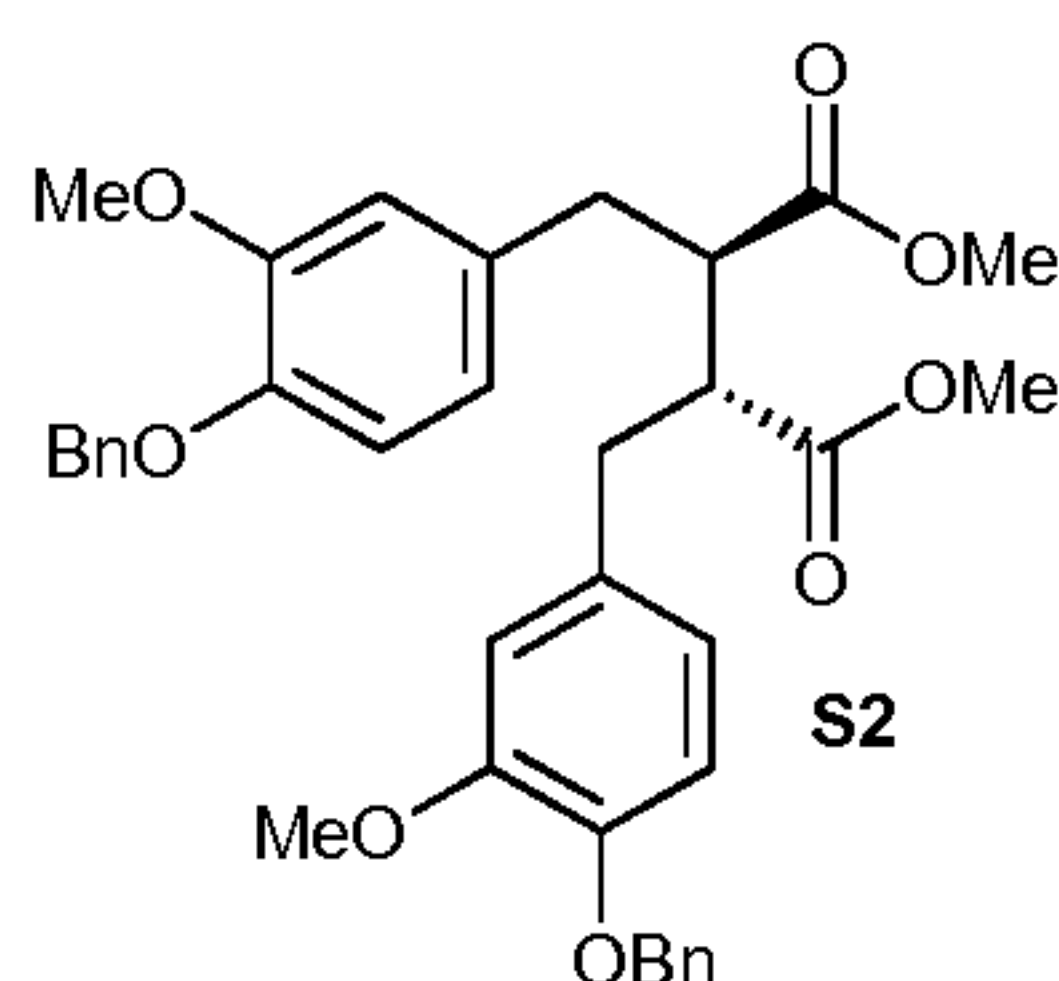


to prepare the compound of formula ((S,S)-S3).

91. The process of claim 90, wherein said reacting is carried out in the presence of TMSOTf.

92. The process of claim 90, wherein said reacting is carried out in the presence of activated molecular sieves.

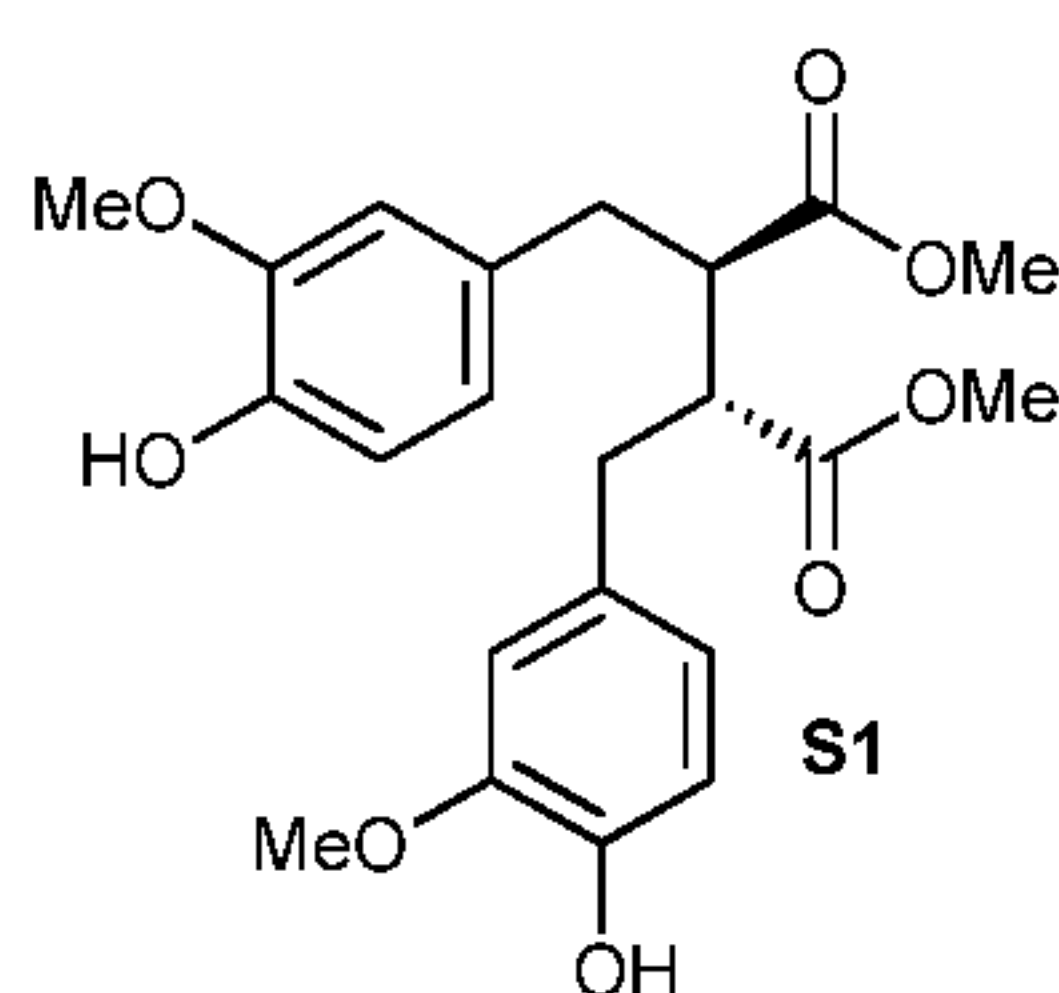
93. The process of claim 90, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

94. The process of claim 93, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

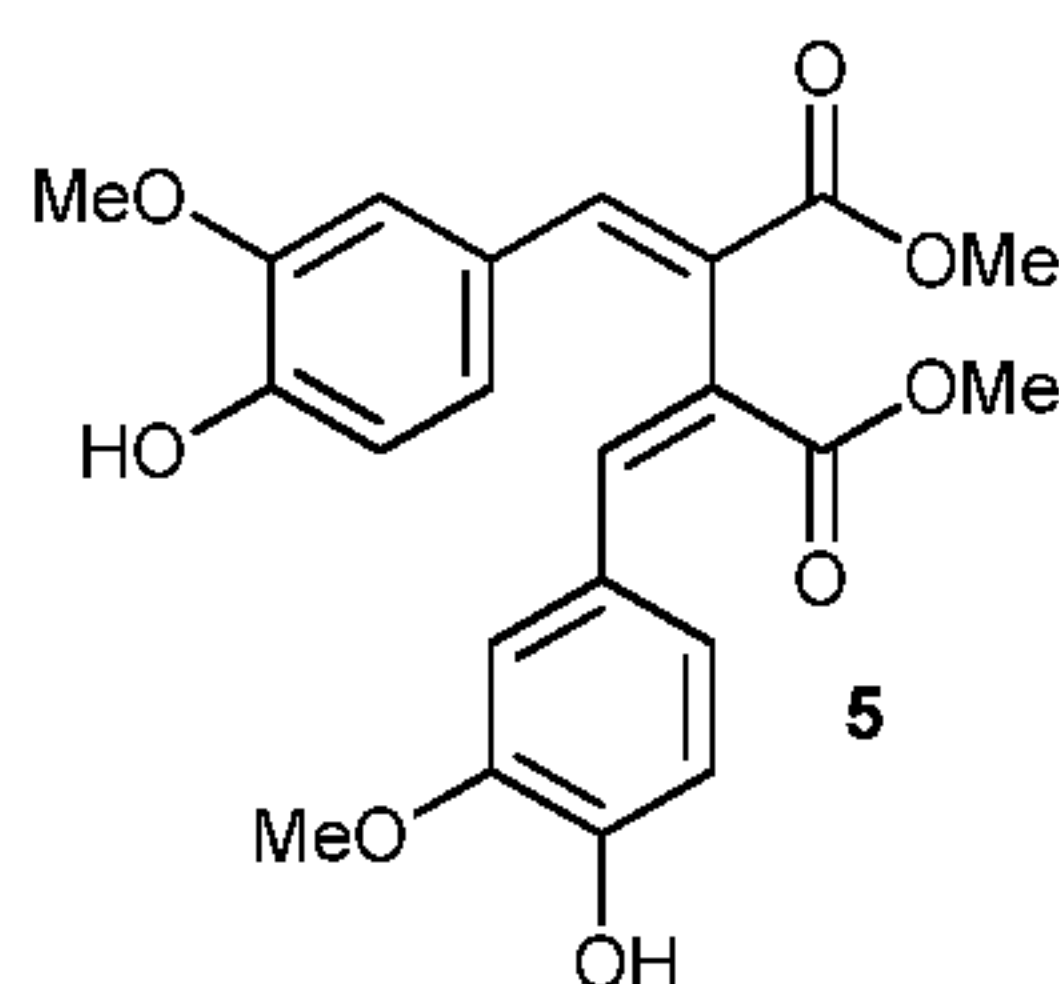
95. The process of claim 93, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

96. The process of claim 95, wherein said benzylating agent is BnBr and NaH.

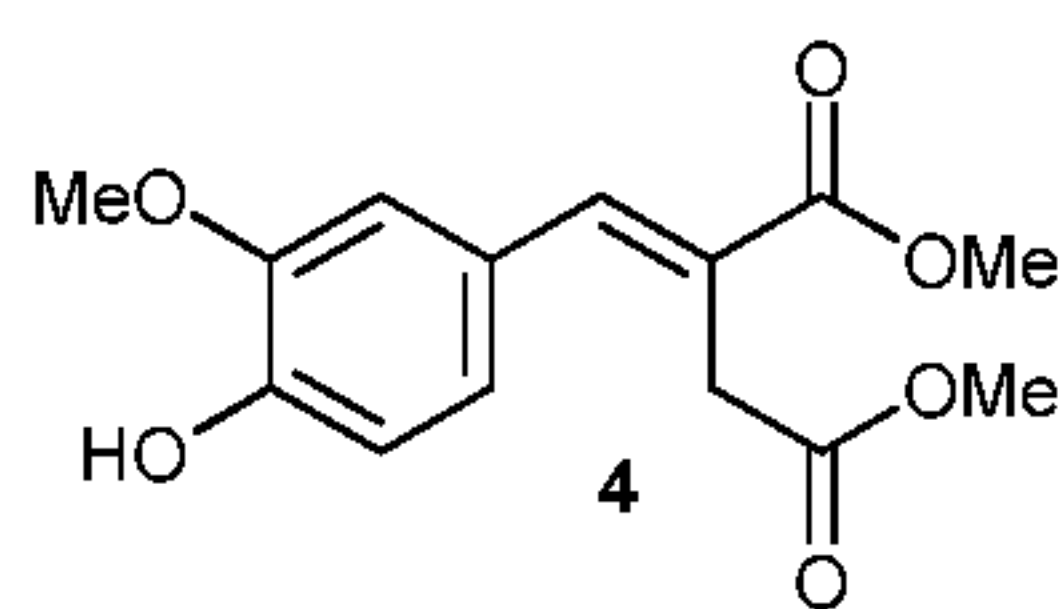
97. The process of claim 95, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

98. The process of claim 97, wherein said reducing agent is H₂ and Pd/C.

99. The process of claim 97, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

100. The process of claim 99, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

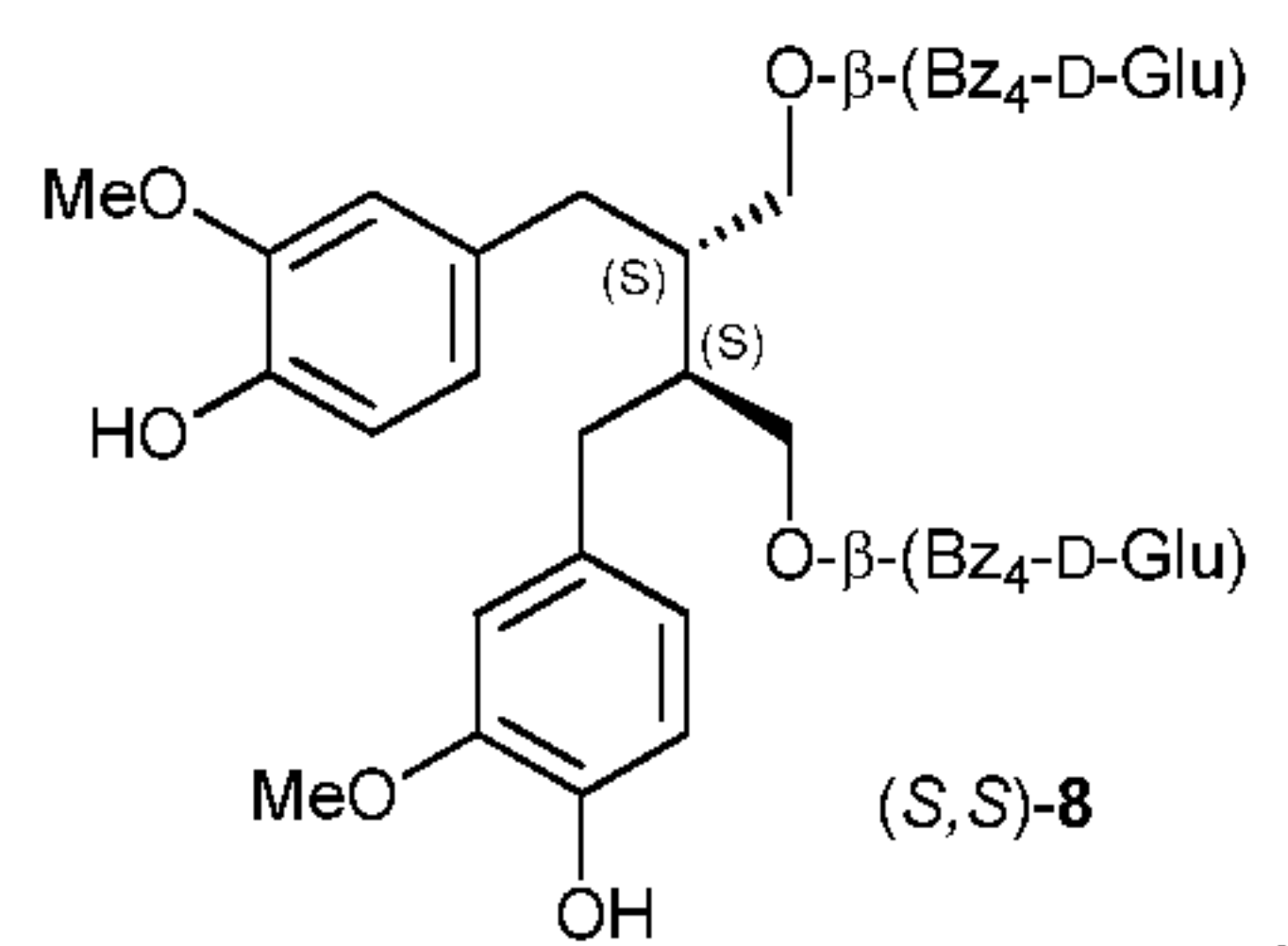
101. The process of claim 99, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

102. The process of claim 99, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

103. The process of claim 102, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

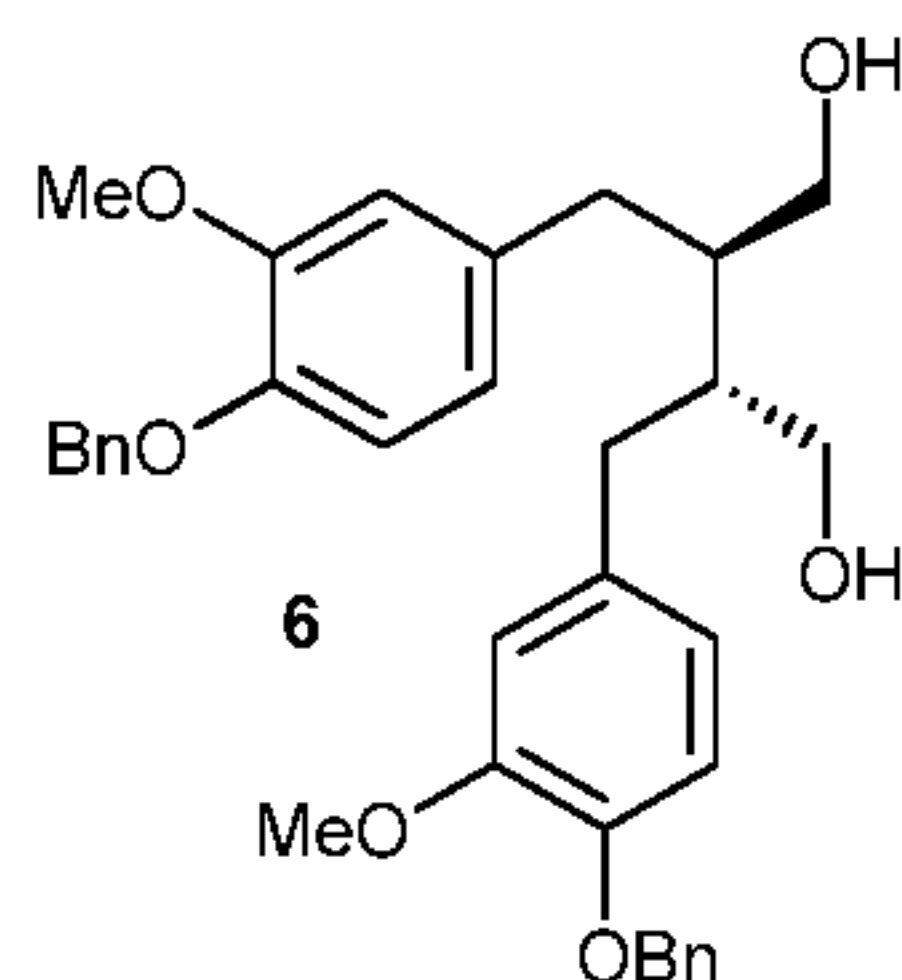
104. The process of claim 102, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

105. A process for preparing a compound of formula ((S,S)-8)

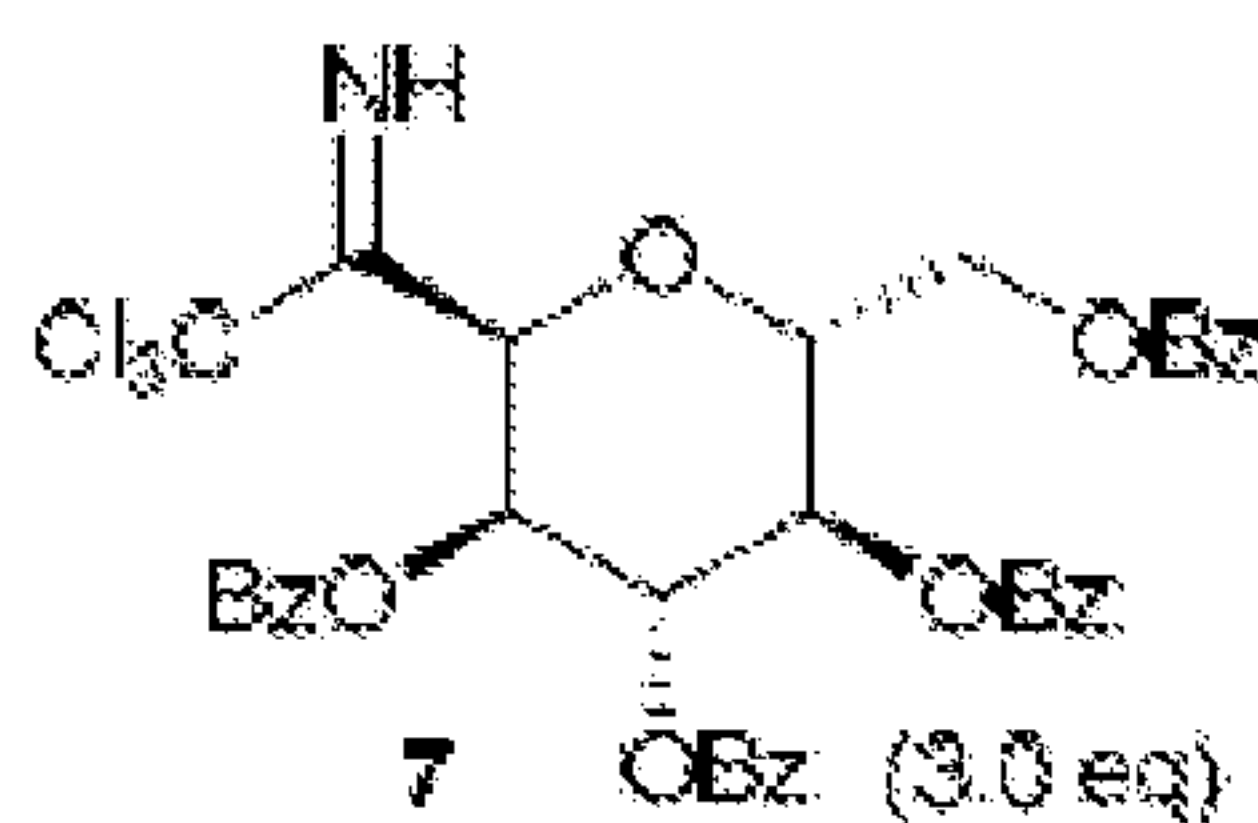


the process comprising:

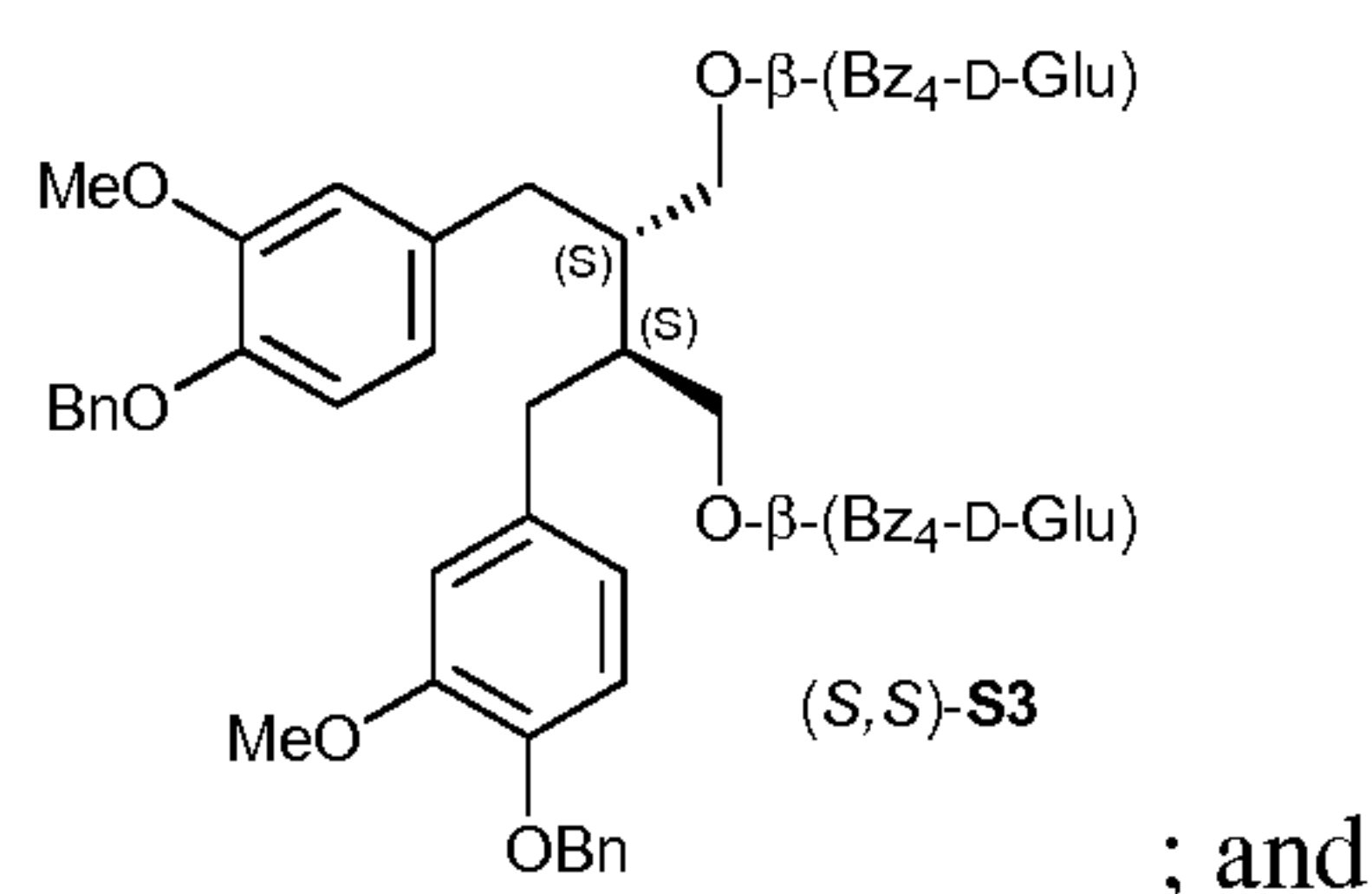
(a) reacting a compound of formula (6)



with a compound of formula (7)



to prepare a compound of formula ((S,S)-S3)



(b) cleaving benzyl ethers of the compound of formula ((S,S)-S3), followed by a separation procedure to prepare the compound of formula ((S,S)-8).

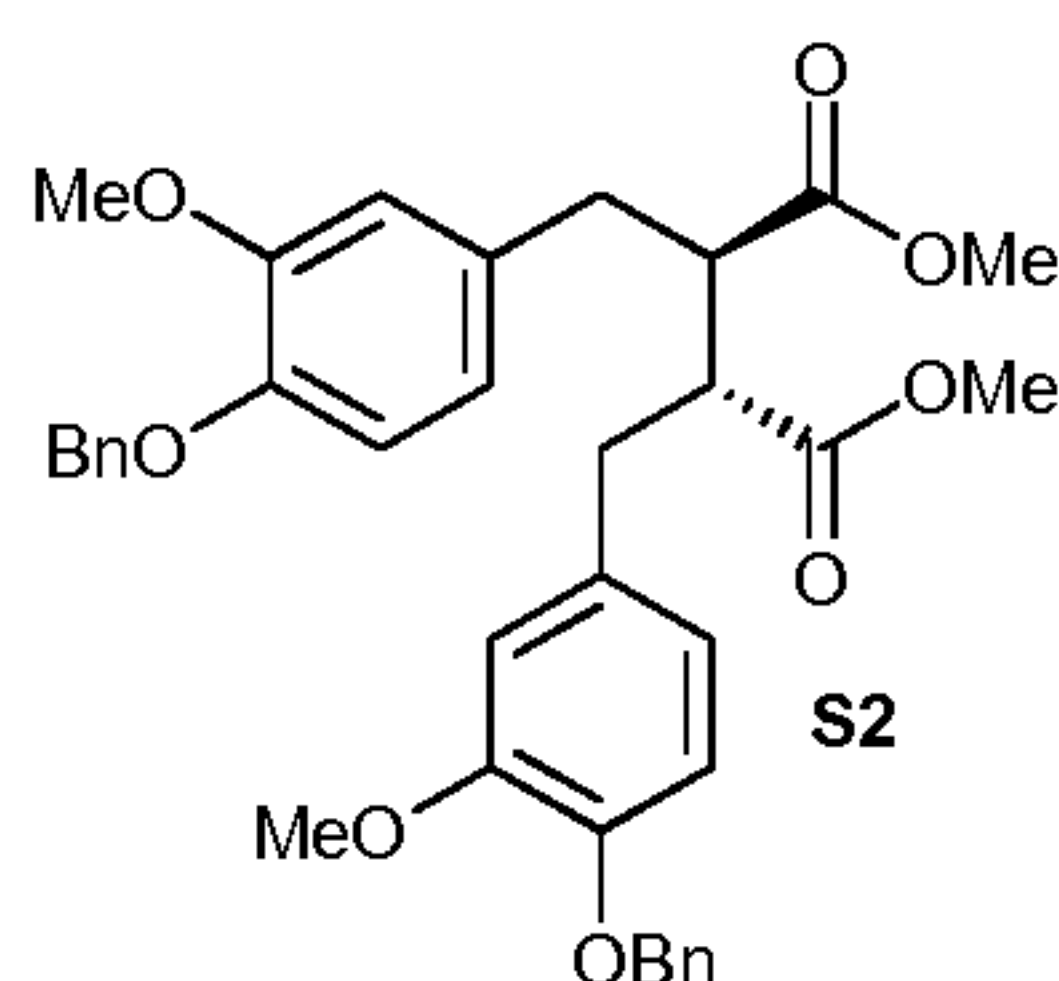
106. The process of claim 105, wherein said reacting is carried out in the presence of TMSOTf.

107. The process of claim 105, wherein said reacting is carried out in the presence of activated molecular sieves.

108. The process of claim 105, wherein said cleaving is carried out in the presence of H₂ and Pd/C in MeOH.

109. The process of claim 105, wherein said separation procedure is carried out using preparative thin layer chromatography.

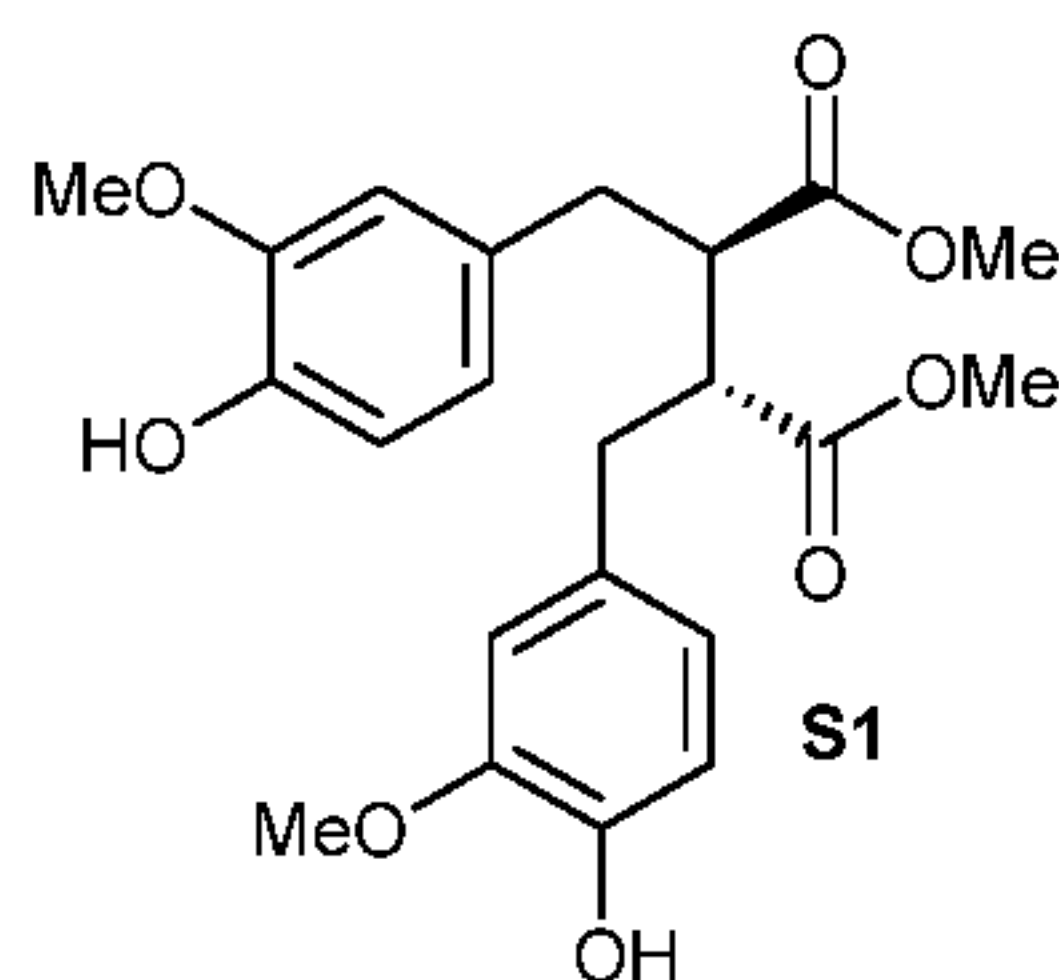
110. The process of claim 105, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

111. The process of claim 110, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

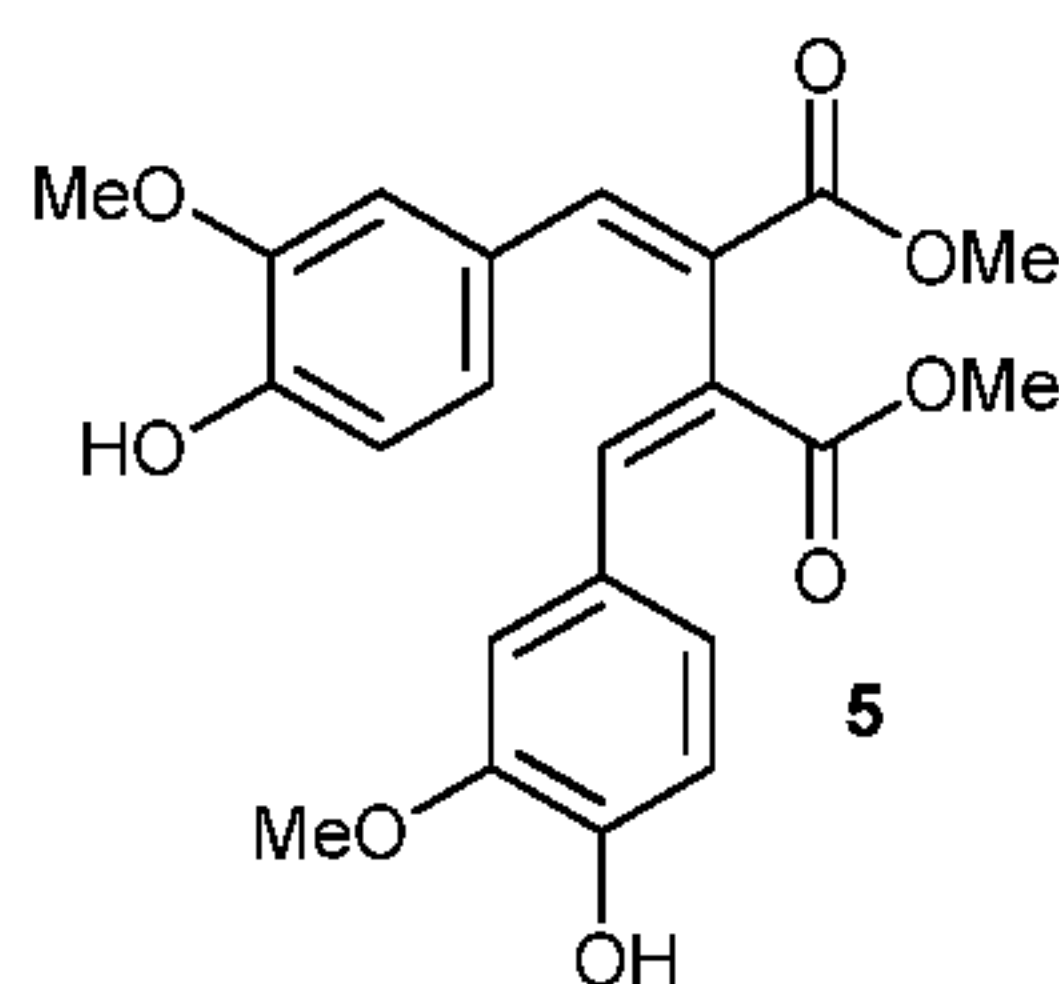
112. The process of claim 110, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

113. The process of claim 112, wherein said benzylating agent is BnBr and NaH.

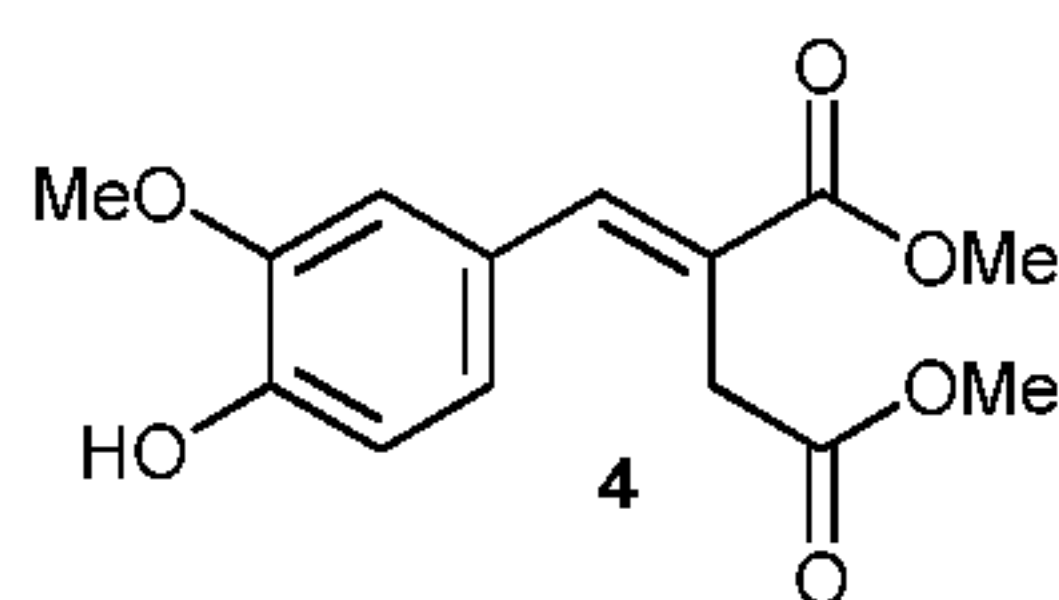
114. The process of claim 112, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

115. The process of claim 114, wherein said reducing agent is H₂ and Pd/C.

116. The process of claim 114, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

117. The process of claim 116, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

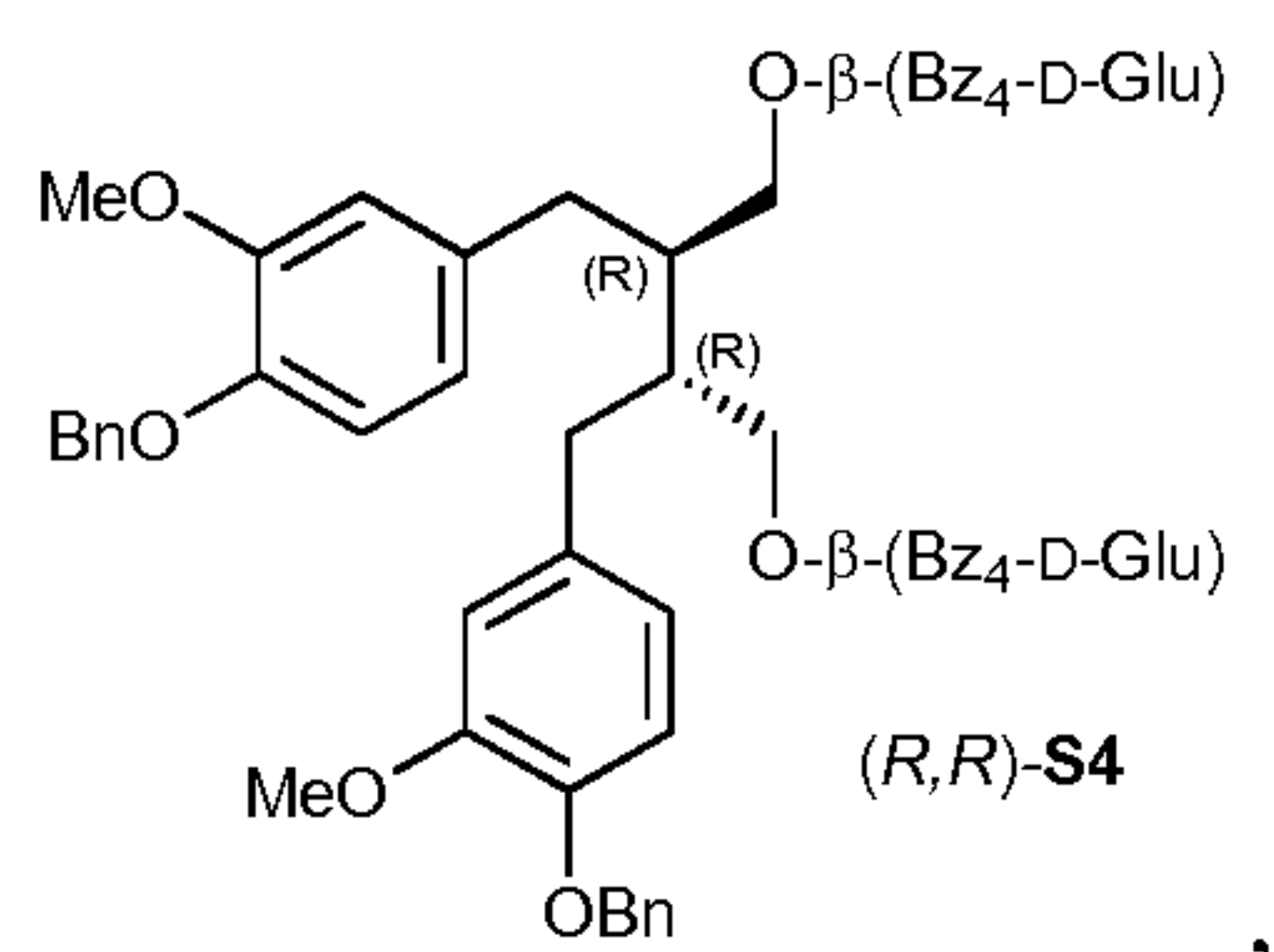
118. The process of claim 116, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

119. The process of claim 116, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

120. The process of claim 119, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

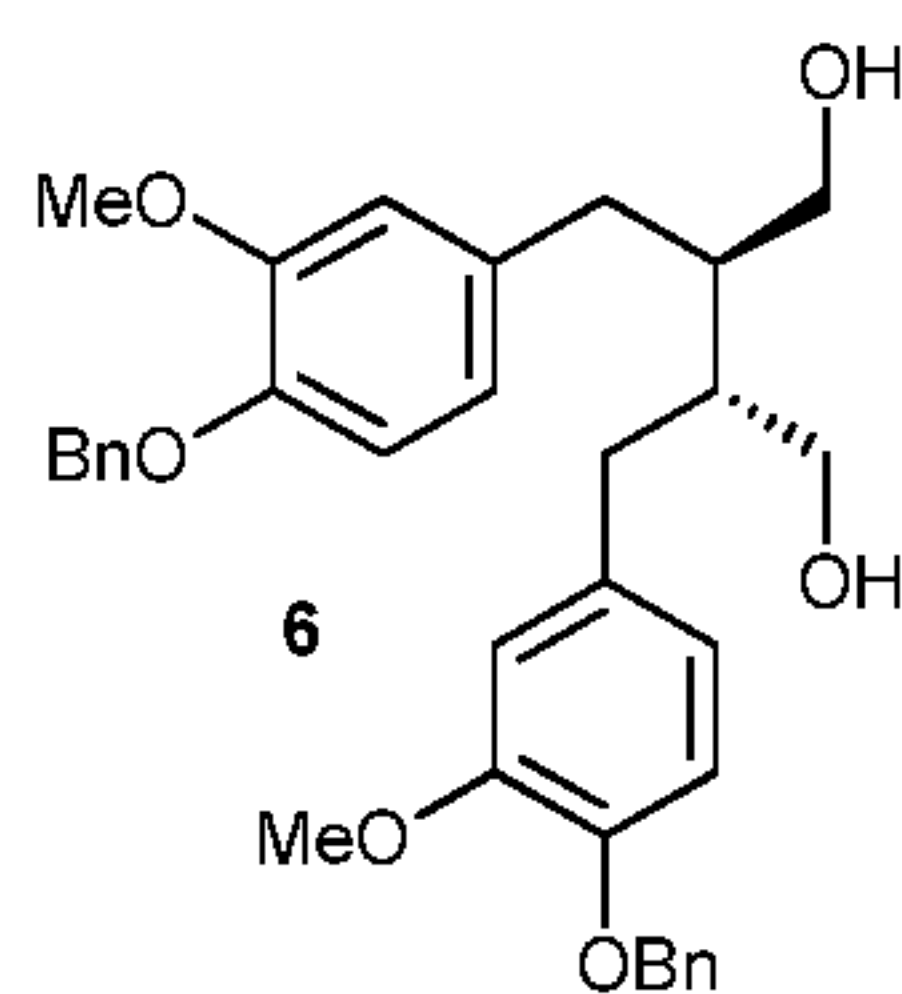
121. The process of claim 117, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

122. A process for preparing a compound of formula ((*R,R*)-S4)

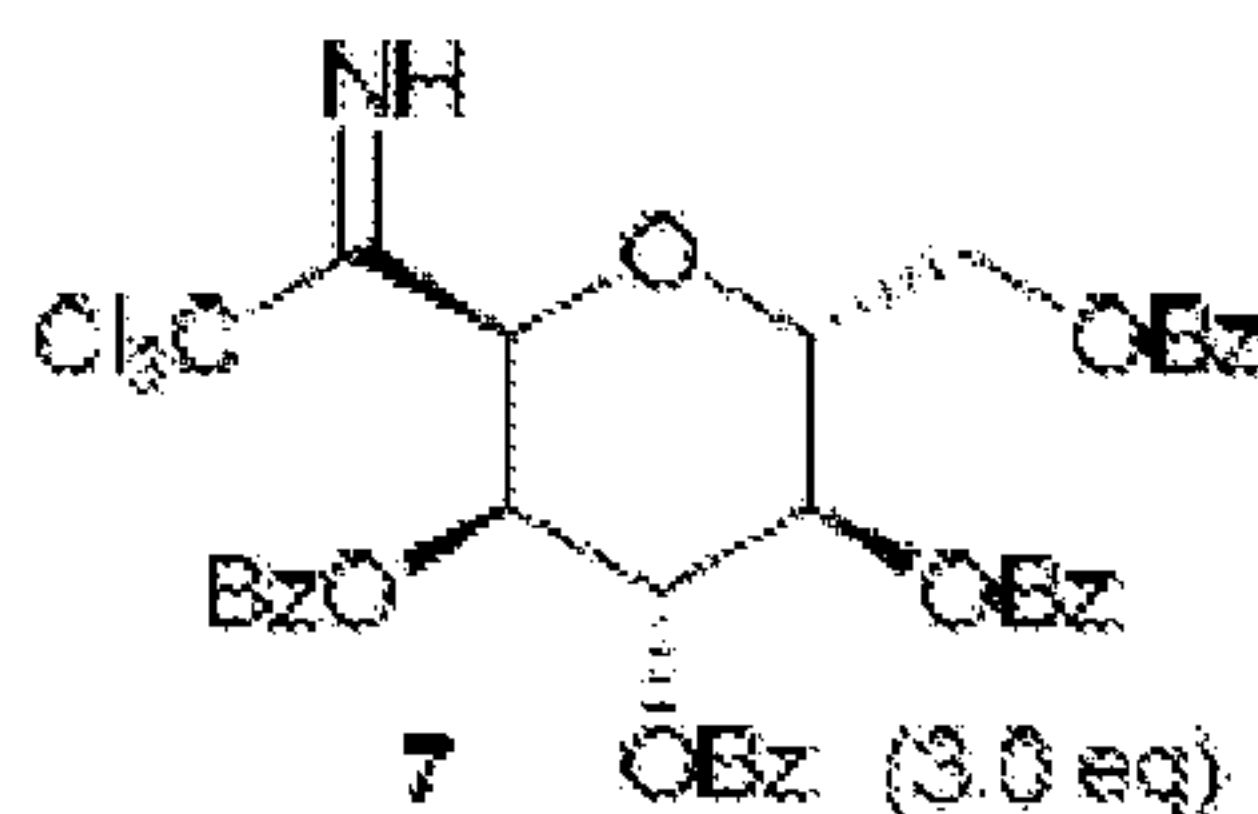


the process comprising:

(a) reacting a compound of formula (6)



with a compound of formula (7)

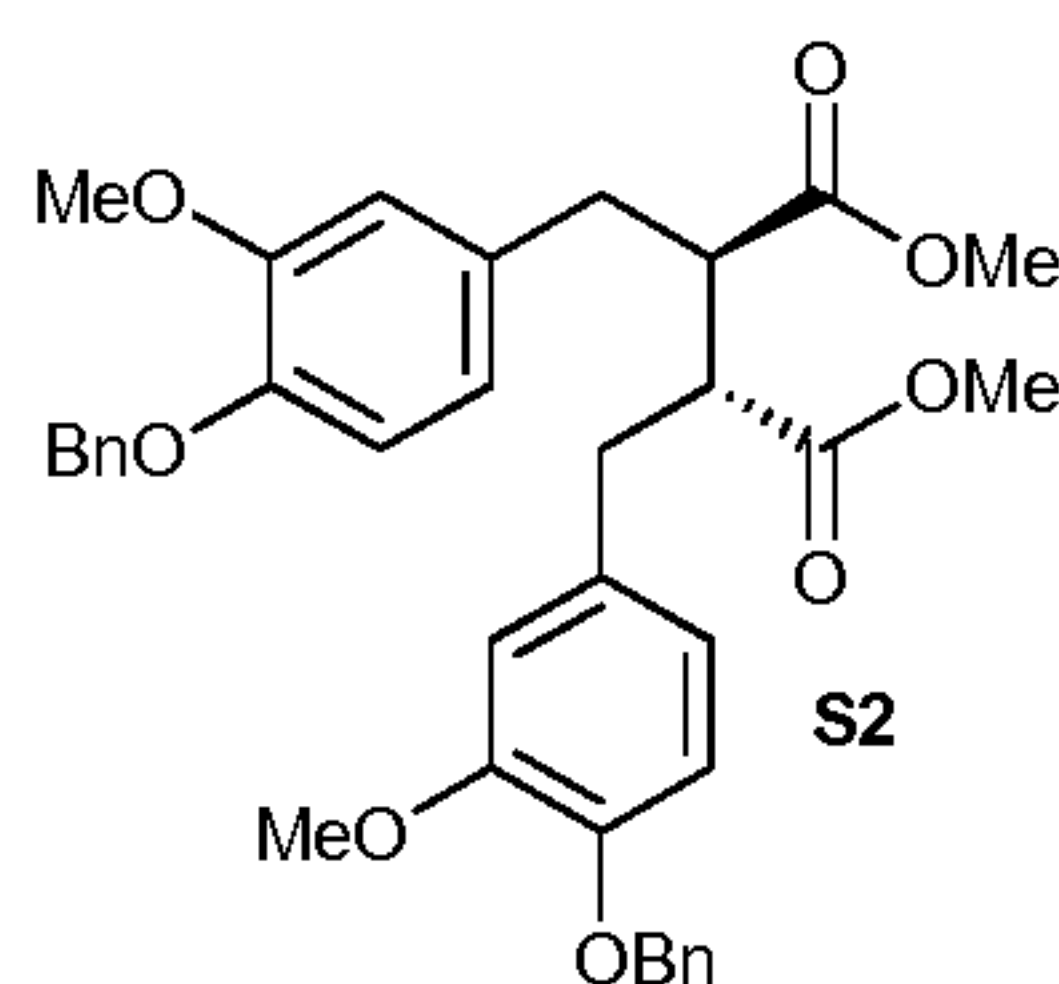


to prepare a compound of formula ((*R,R*)-S4).

123. The process of claim 122, wherein said reacting is carried out in the presence of TMSOTf.

124. The process of claim 122, wherein said reacting is carried out in the presence of activated molecular sieves.

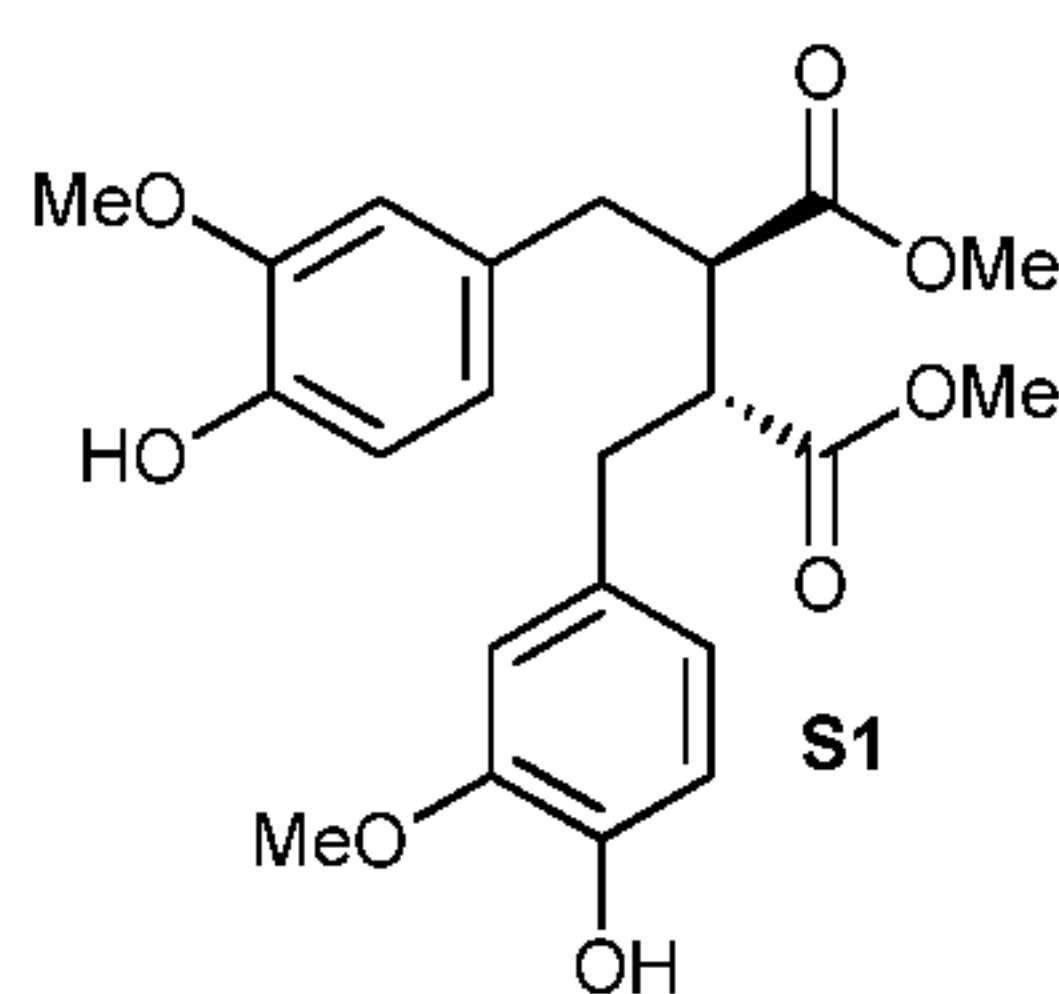
125. The process of claim 122, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

126. The process of claim 125, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

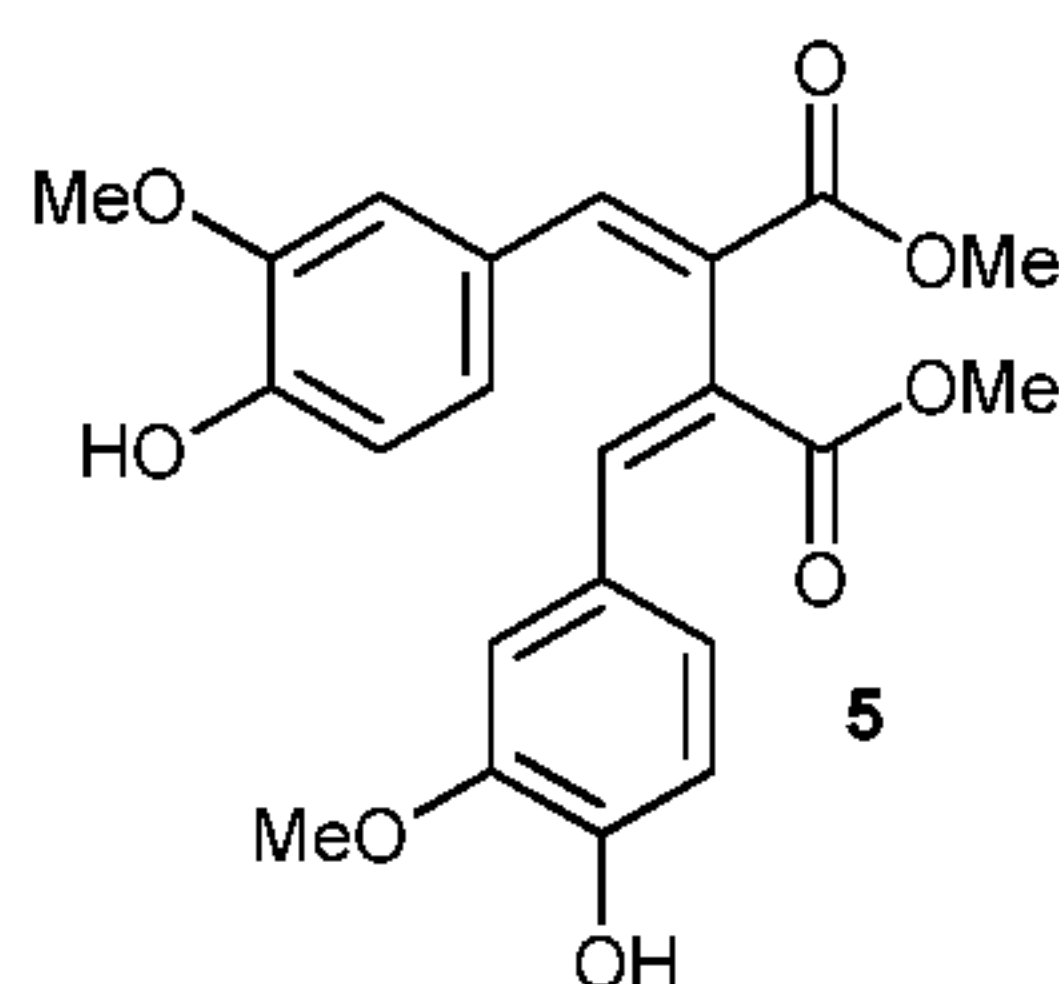
127. The process of claim 125, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

128. The process of claim 127, wherein said benzylating agent is BnBr and NaH.

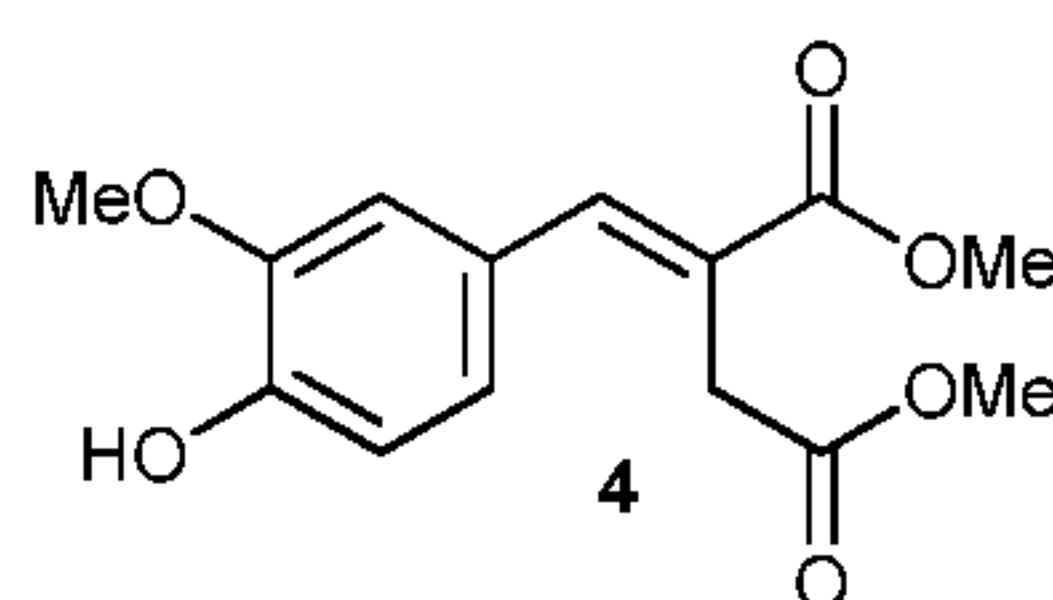
129. The process of claim 127, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

130. The process of claim 129, wherein said reducing agent is H_2 and Pd/C.

131. The process of claim 129, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

132. The process of claim 131, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

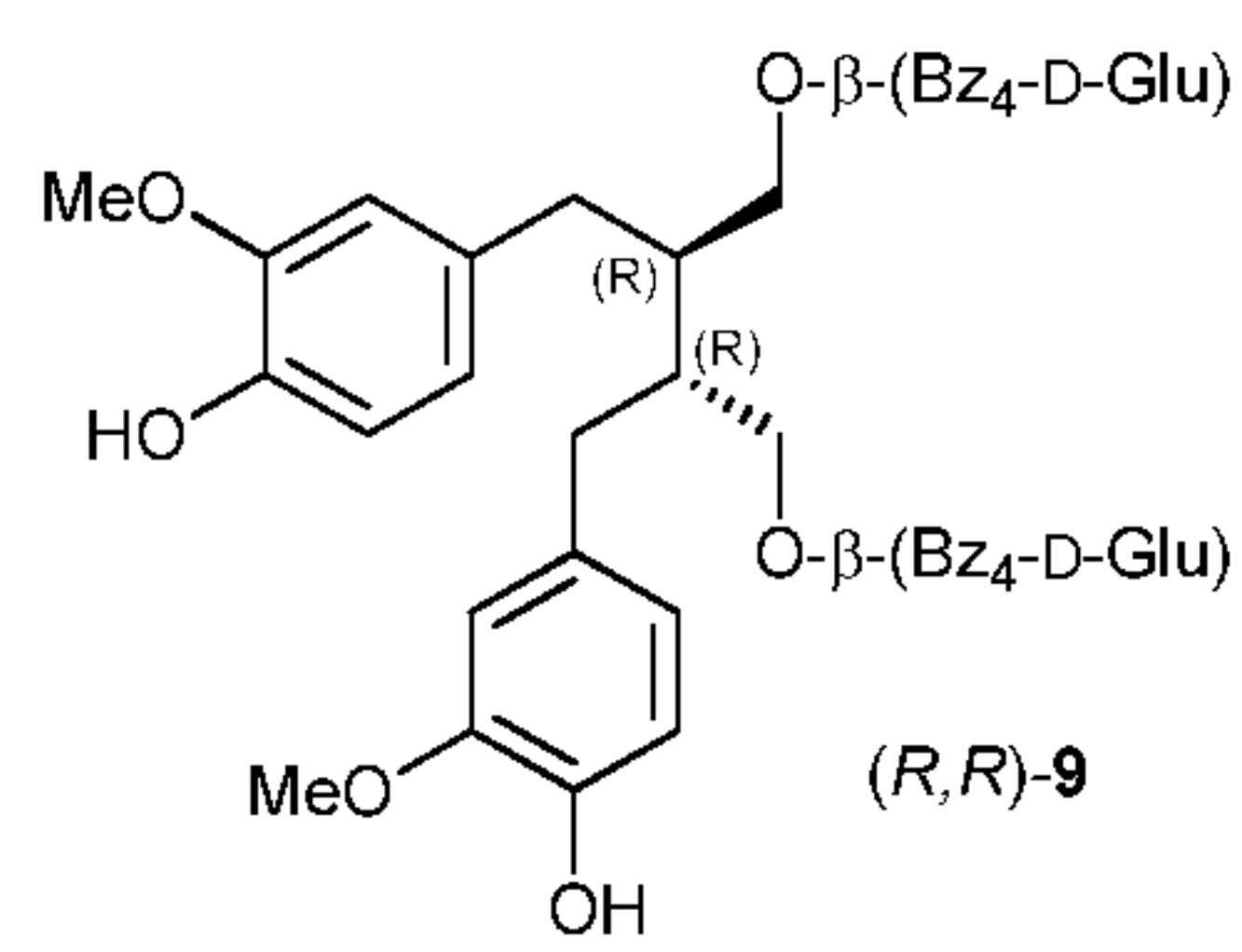
133. The process of claim 131, wherein said esterification reaction is carried out with MeOH in the presence of H_2SO_4 .

134. The process of claim 131, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

135. The process of claim 134, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

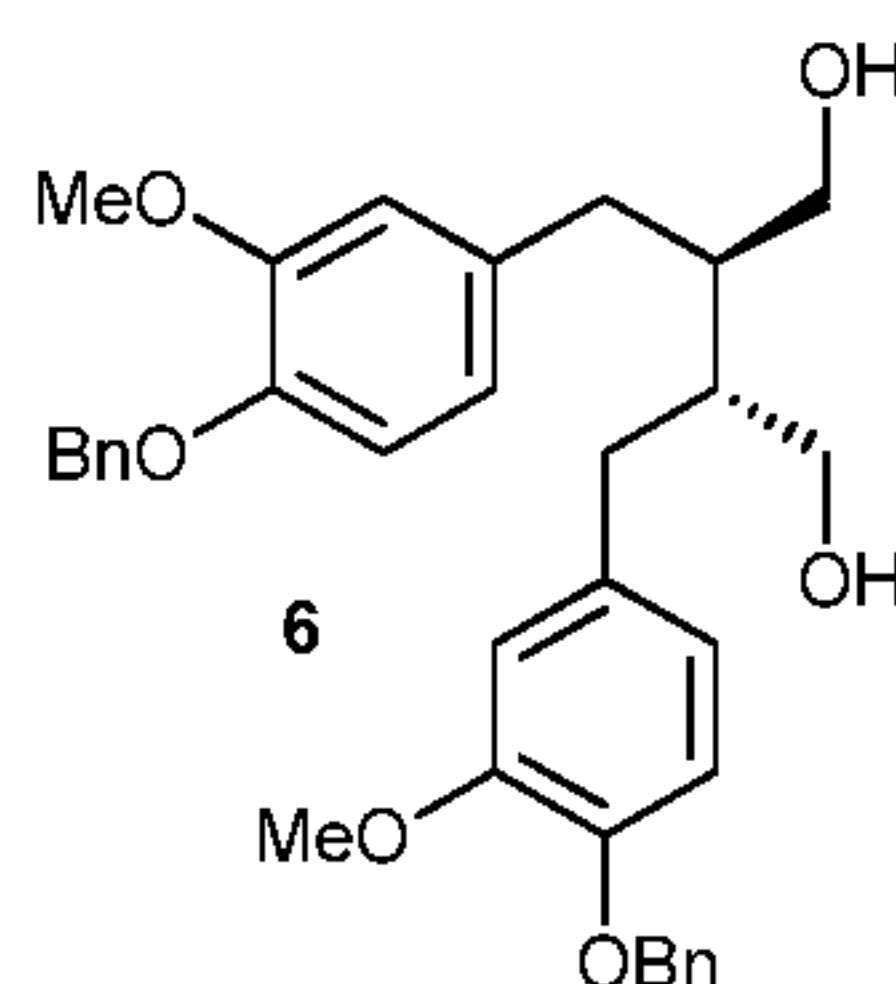
136. The process of claim 134, wherein said esterification reaction is carried out with MeOH in the presence of H_2SO_4 .

137. A process for preparing a compound of formula ((*R,R*)-9)

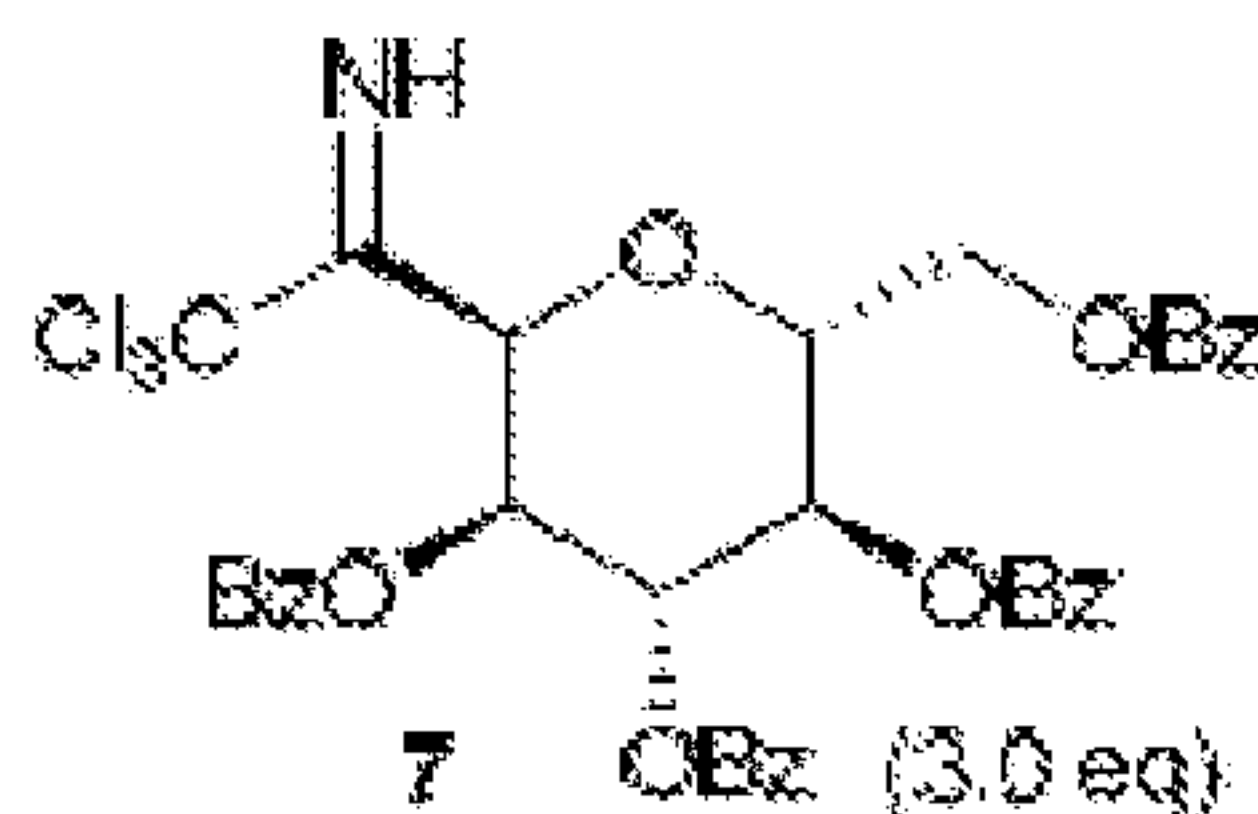


the process comprising:

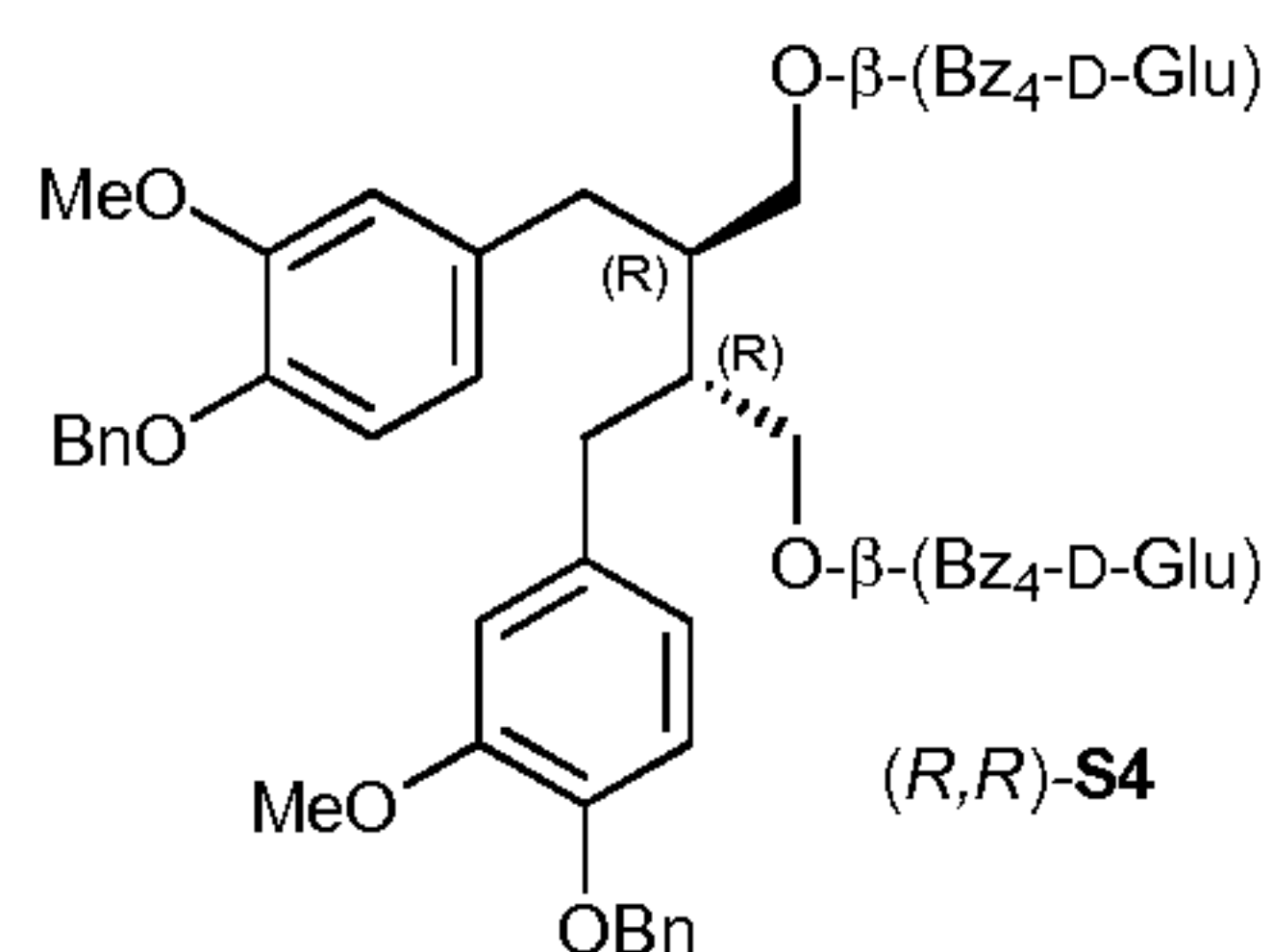
- (a) reacting a compound of formula (6)



with a compound of formula (7)



to prepare a compound of formula ((*R,R*)-S4)



- (b) cleaving benzyl ethers of the compound of formula ((*R,R*)-S4), followed by a separation procedure to prepare a compound of formula ((*R,R*)-9)

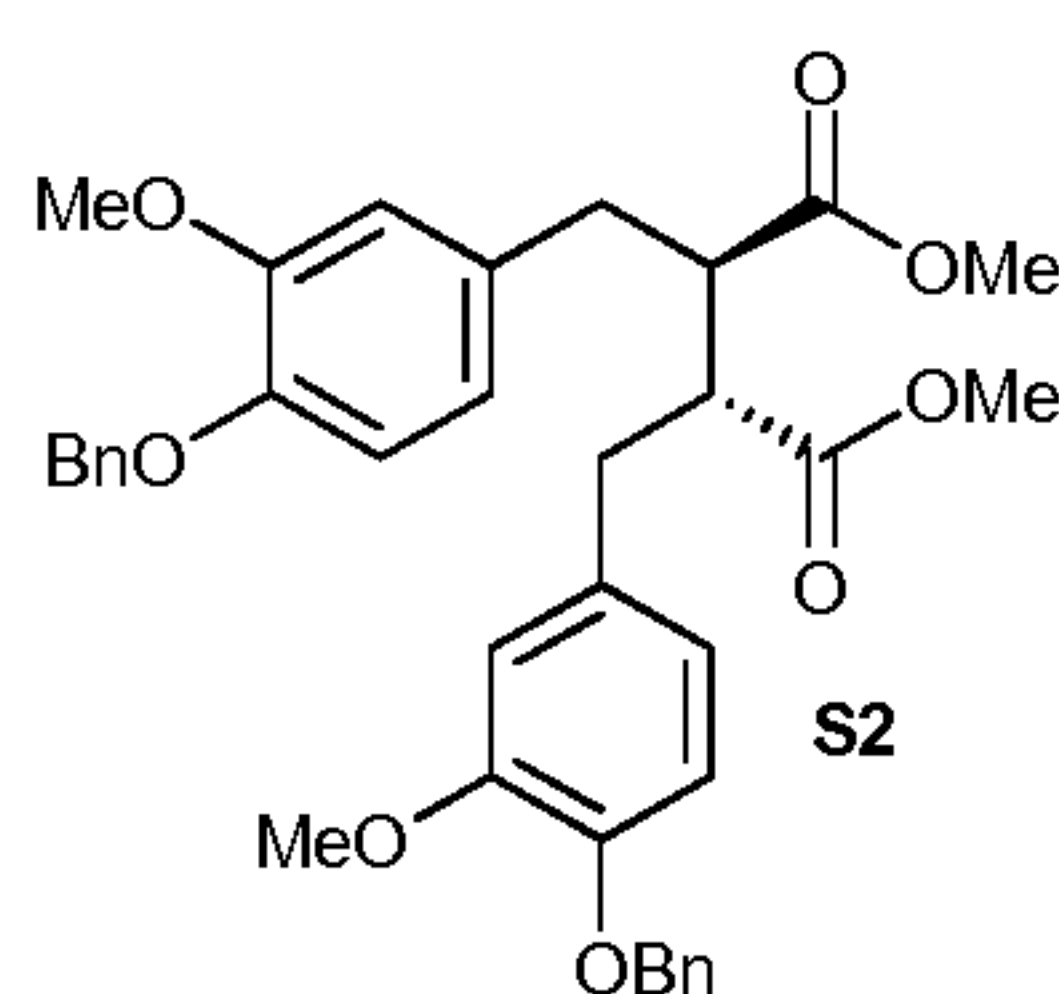
138. The process of claim 137, wherein said reacting is carried out in the presence of TMSOTf.

139. The process of claim 137, wherein said reacting is carried out in the presence of activated molecular sieves.

140. The process of claim 137, wherein said cleaving is carried out in the presence of H₂ and Pd/C in MeOH.

141. The process of claim 137, wherein said separation procedure is carried out using preparative thin layer chromatography.

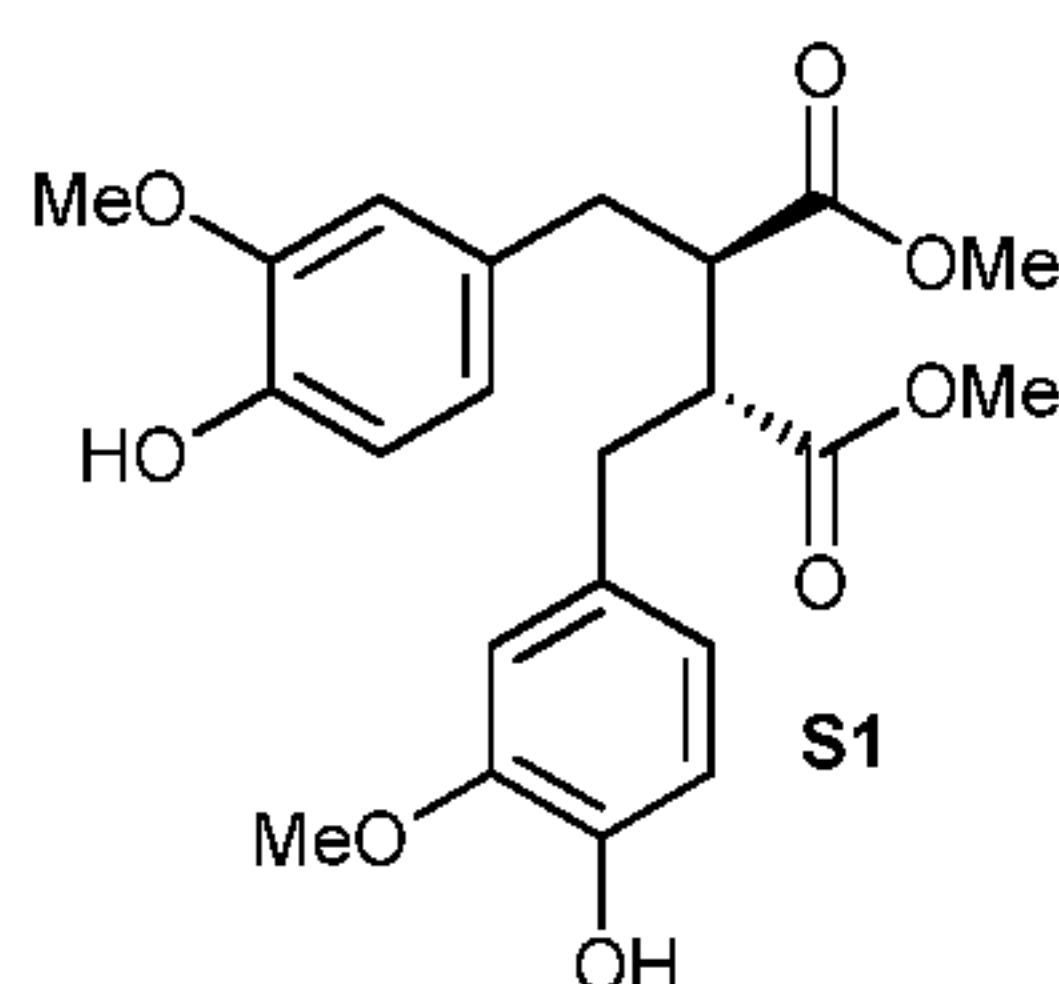
142. The process of claim 137, wherein said compound of formula (6) is prepared by the process comprising reacting a compound of formula (S2)



with a reducing agent to prepare the compound of formula (6).

143. The process of claim 142, wherein said reducing agent is lithium aluminum hydride (LAH) in THF.

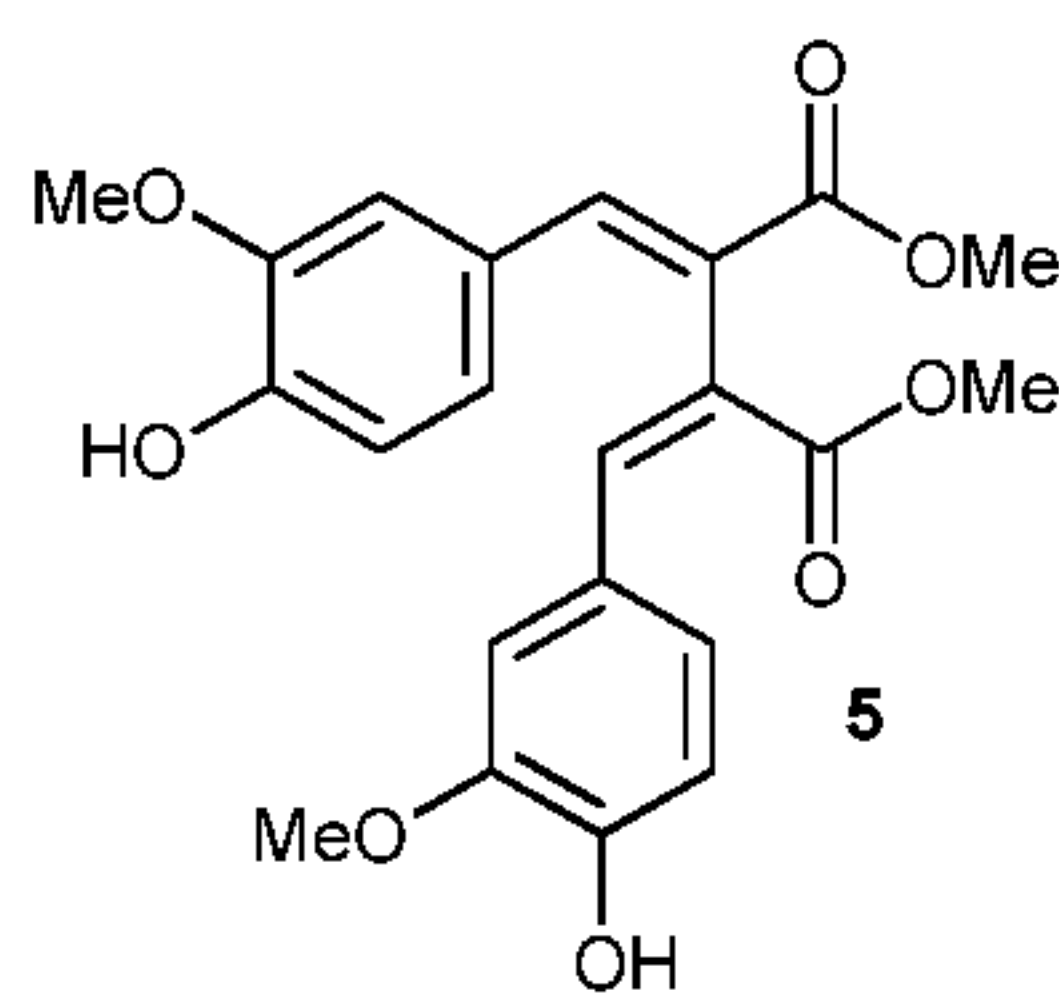
144. The process of claim 142, wherein said compound of formula (S2) is prepared by the process comprising reacting a compound of formula (S1)



with a benzylating agent to prepare the compound of formula (S2).

145. The process of claim 144, wherein said benzylating agent is BnBr and NaH.

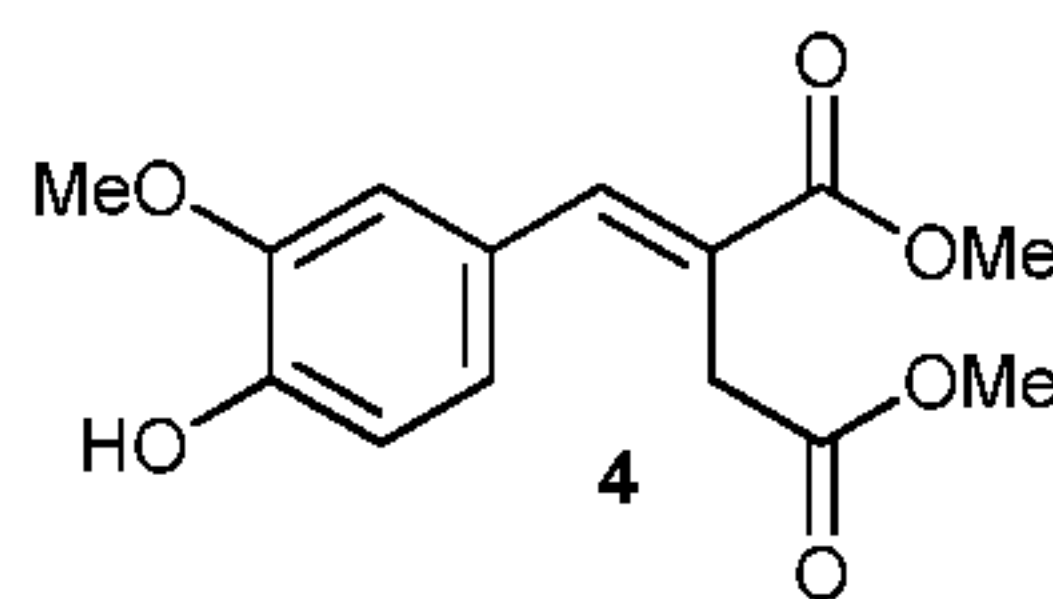
146. The process of claim 144, wherein said compound of formula (S1) is prepared by the process comprising reacting a compound of formula (5)



with a reducing agent to prepare the compound of formula (S1).

147. The process of claim 146, wherein said reducing agent is H₂ and Pd/C.

148. The process of claim 146, wherein said compound of formula (5) is prepared by the process comprising reacting a compound of formula (4)



with vanillin by a Stobbe condensation reaction, followed by an esterification reaction to prepare the compound of formula (5).

149. The process of claim 148, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

150. The process of claim 148, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

151. The process of claim 148, wherein said compound of formula (4) is prepared by the process comprising reacting vanillin with methyl succinate via a Stobbe condensation reaction, followed by an esterification reaction.

152. The process of claim 151, wherein said Stobbe condensation reaction is carried out in MeOH and in the presence of lithium wires.

153. The process of claim 151, wherein said esterification reaction is carried out with MeOH in the presence of H₂SO₄.

1 of 15

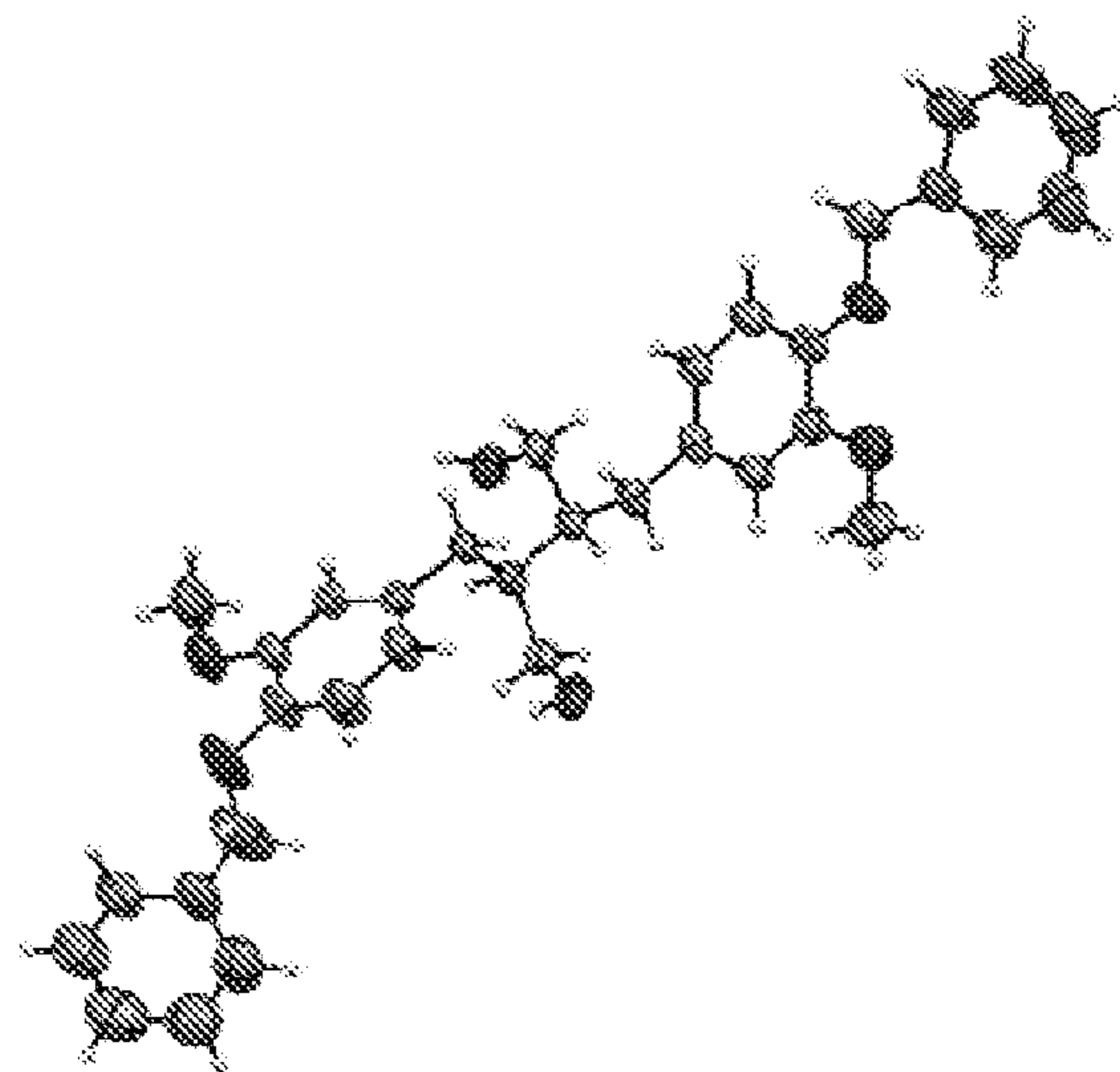


Figure 1

2 of 15

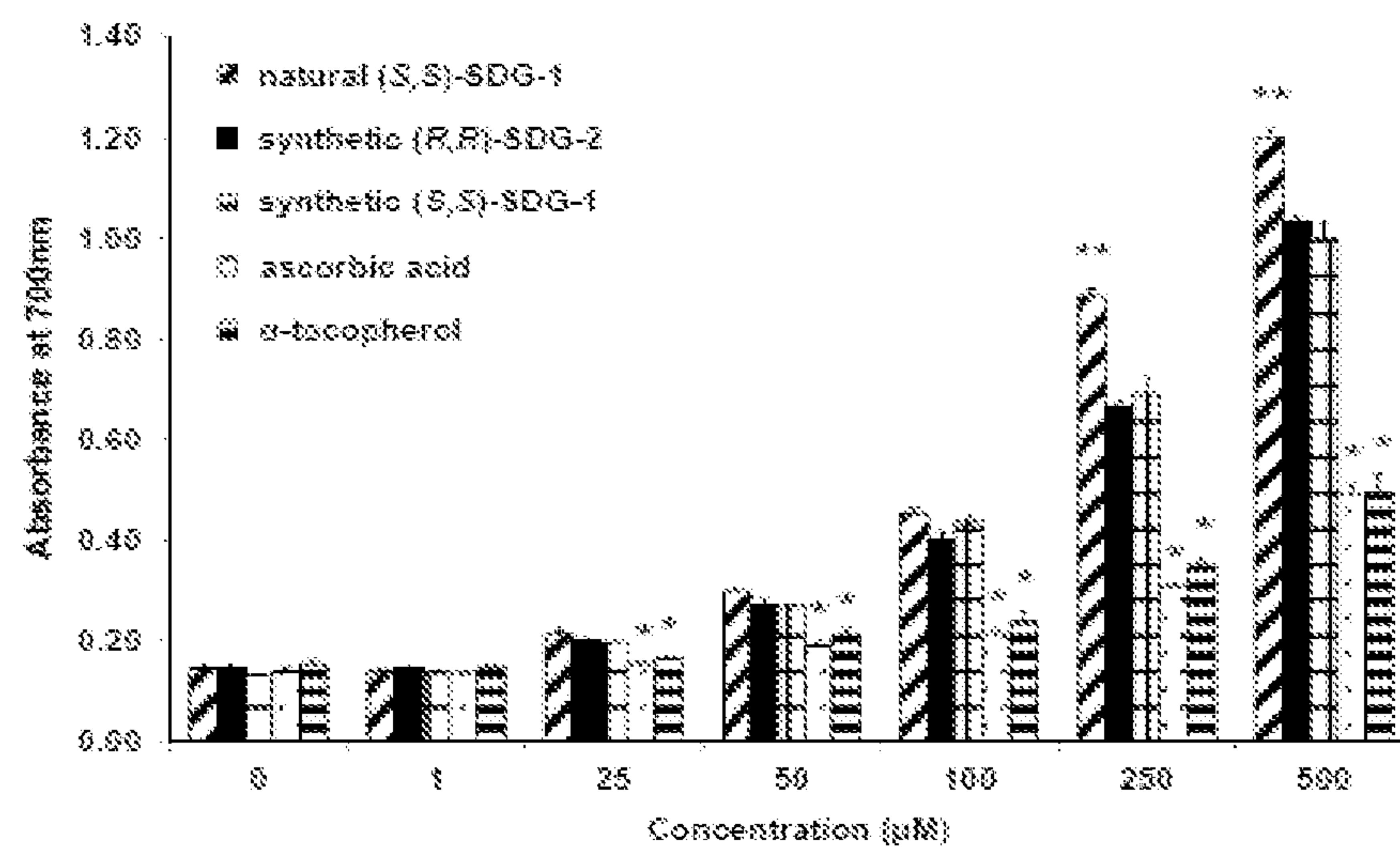


Figure 2

3 of 15

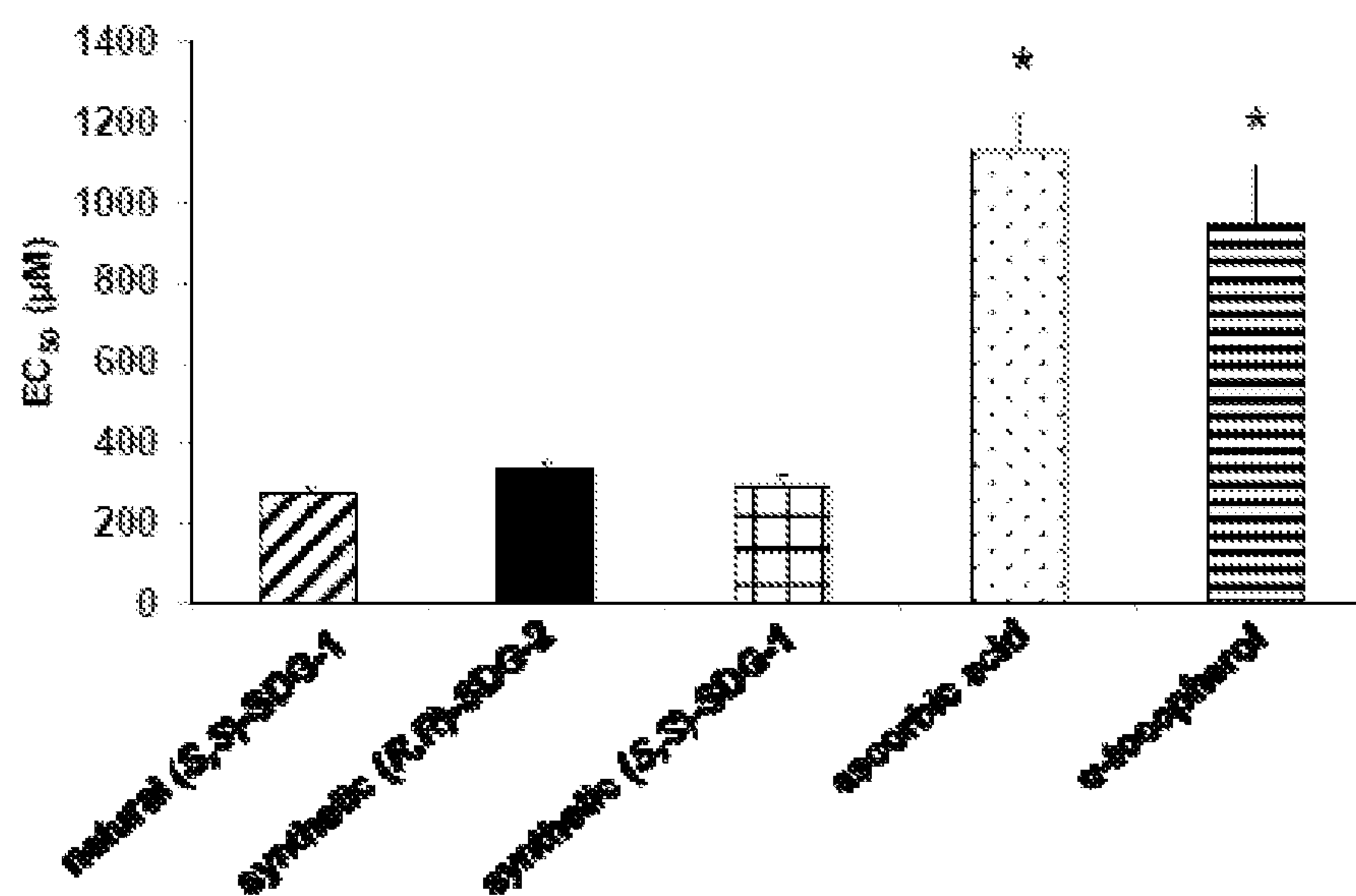


Figure 3

4 of 15

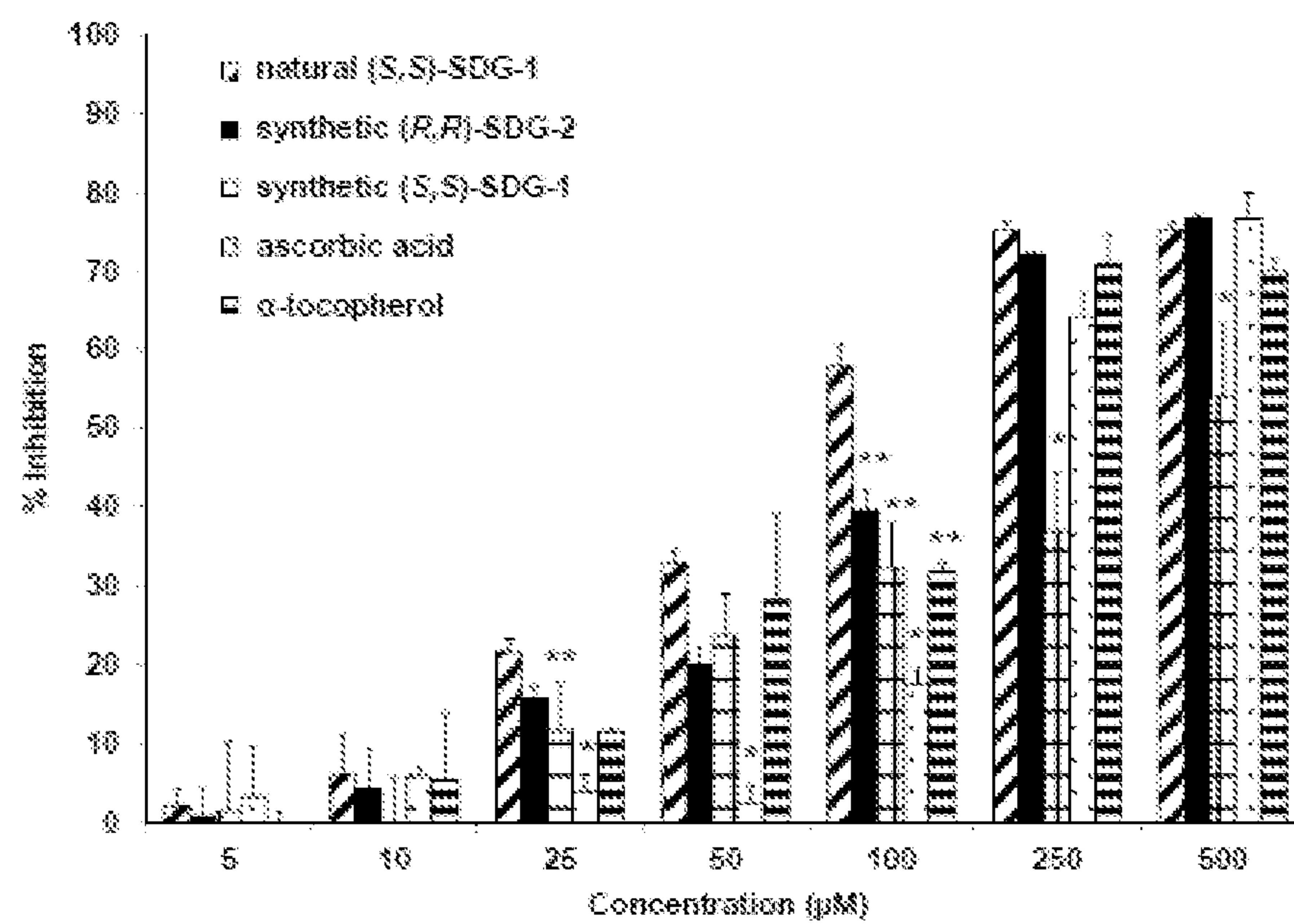


Figure 4

5 of 15

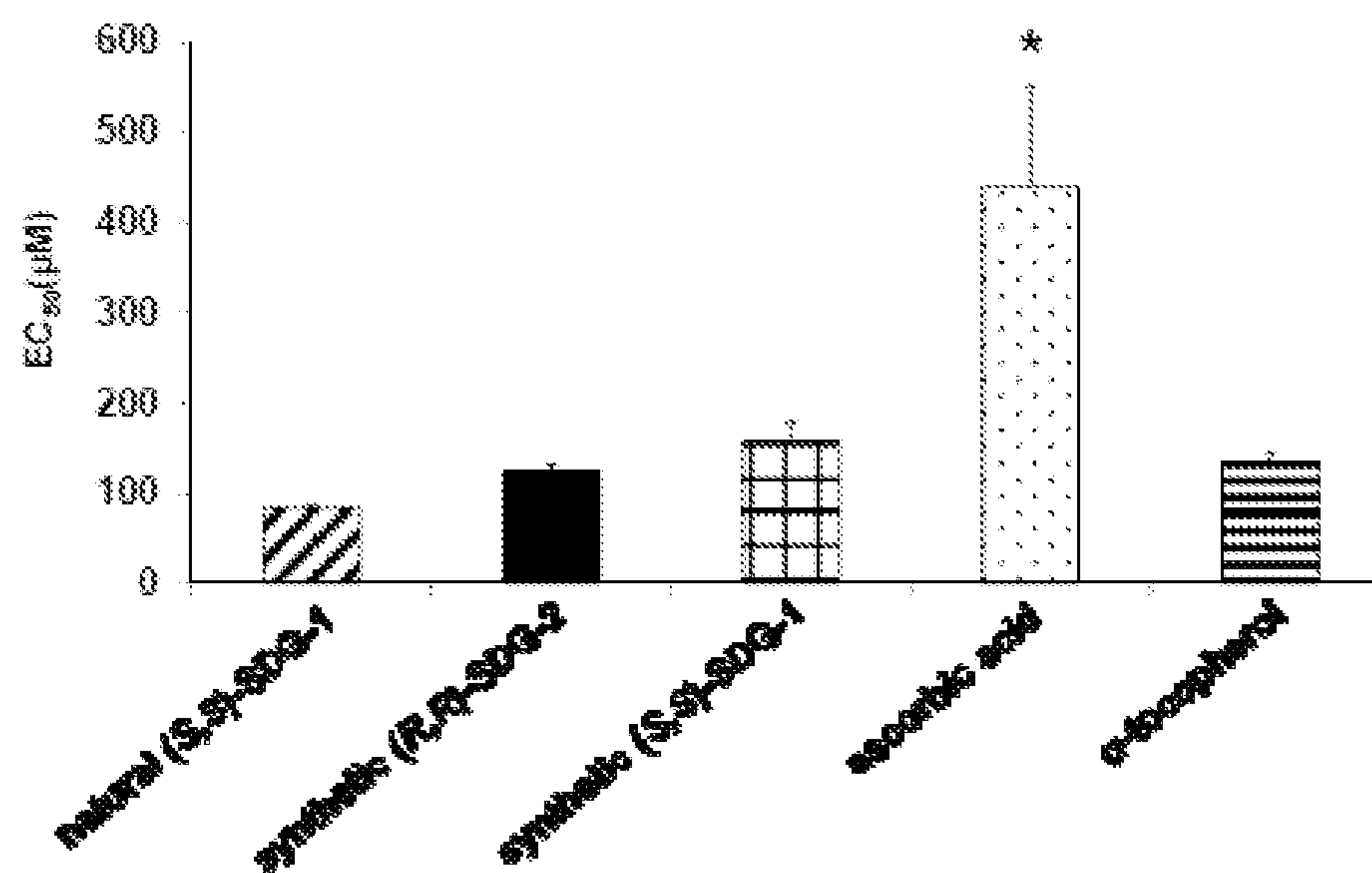


Figure 5

6 of 15

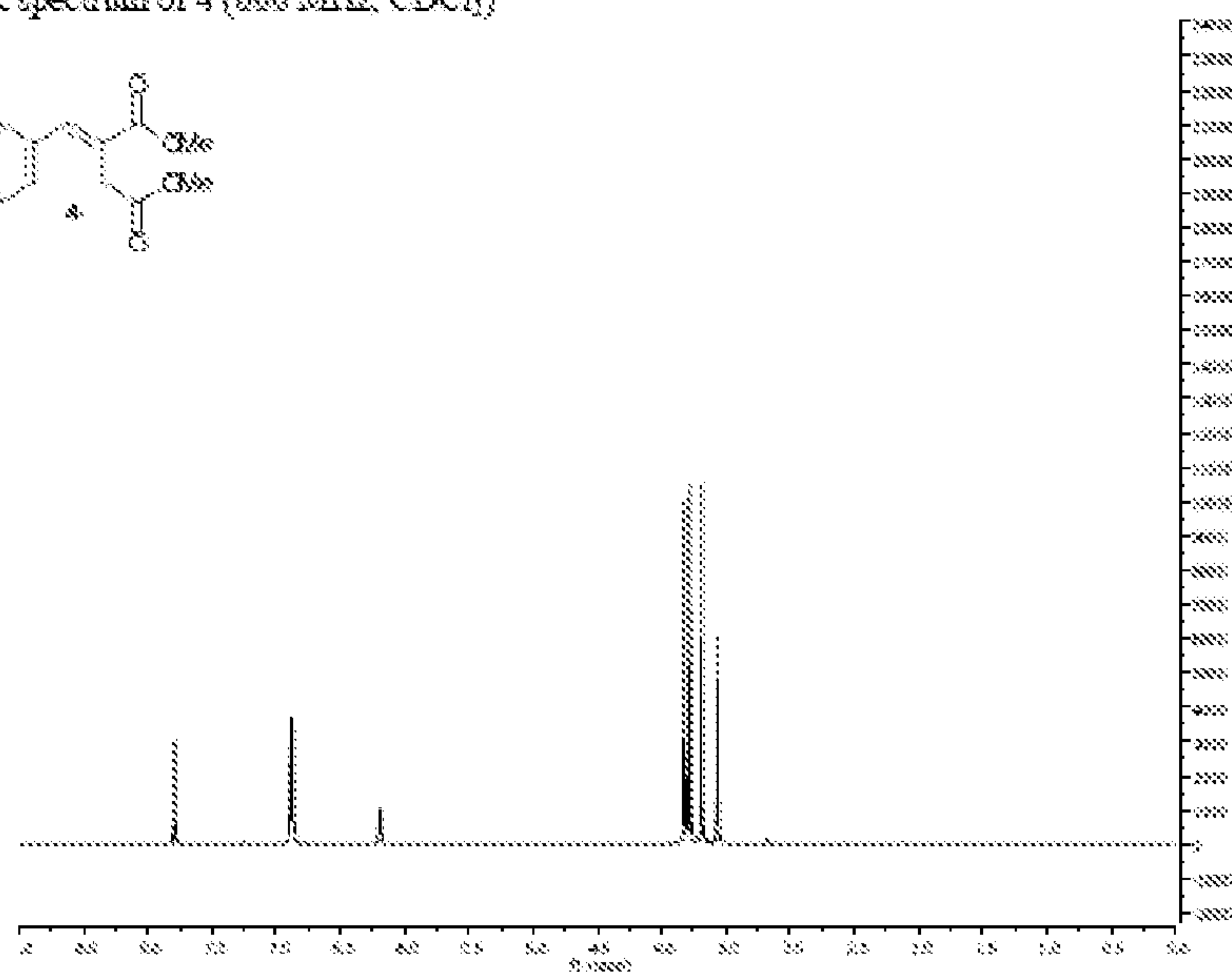
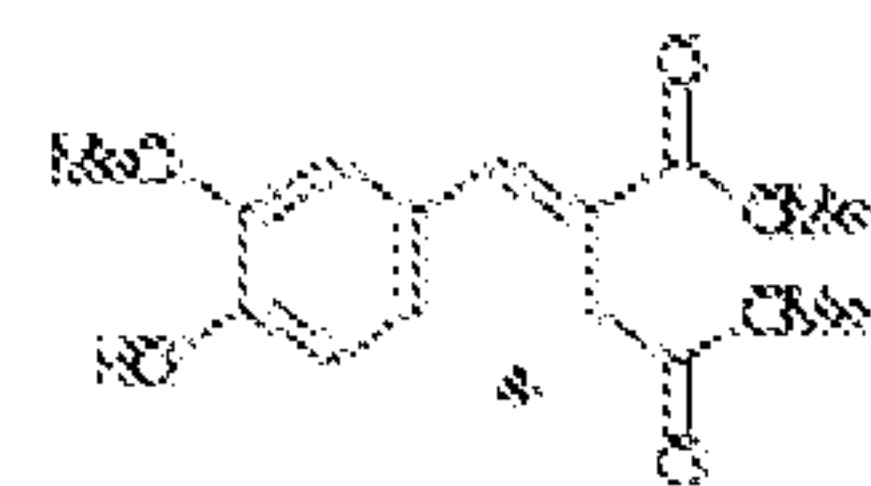
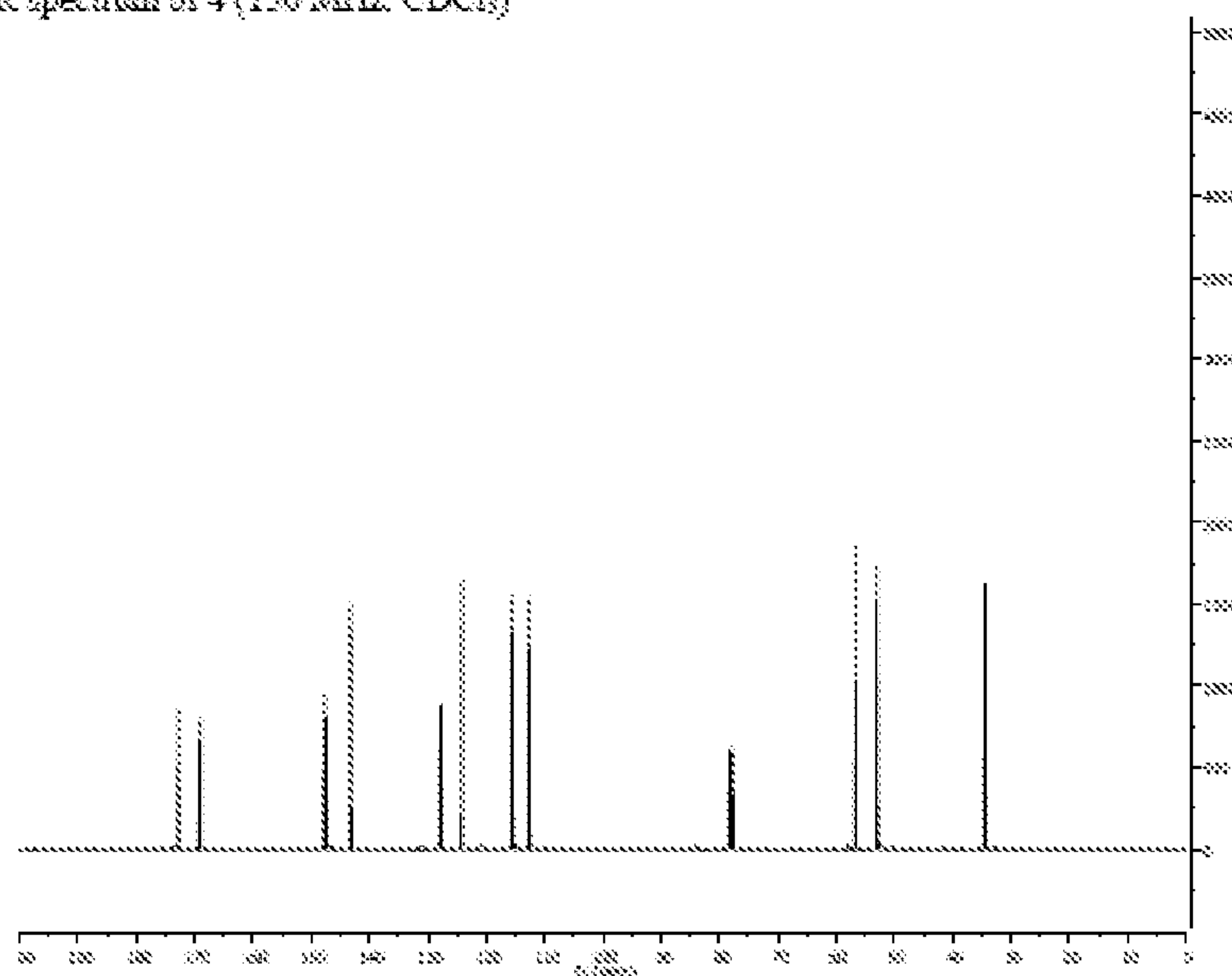
¹H NMR spectrum of 4 (600 MHz, CDCl₃)¹³C NMR spectrum of 4 (120 MHz, CDCl₃)

Figure 6

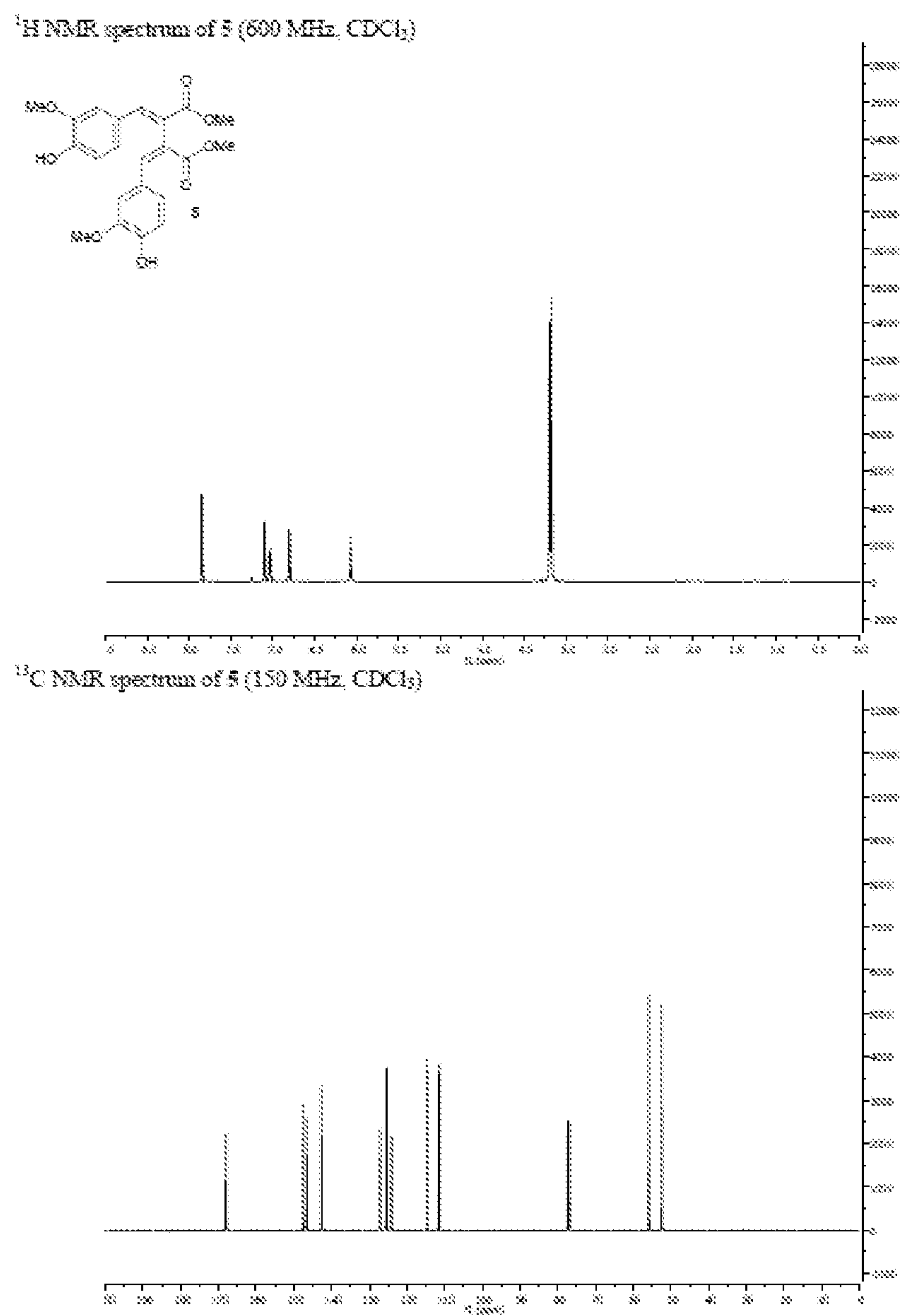


Figure 7

8 of 15

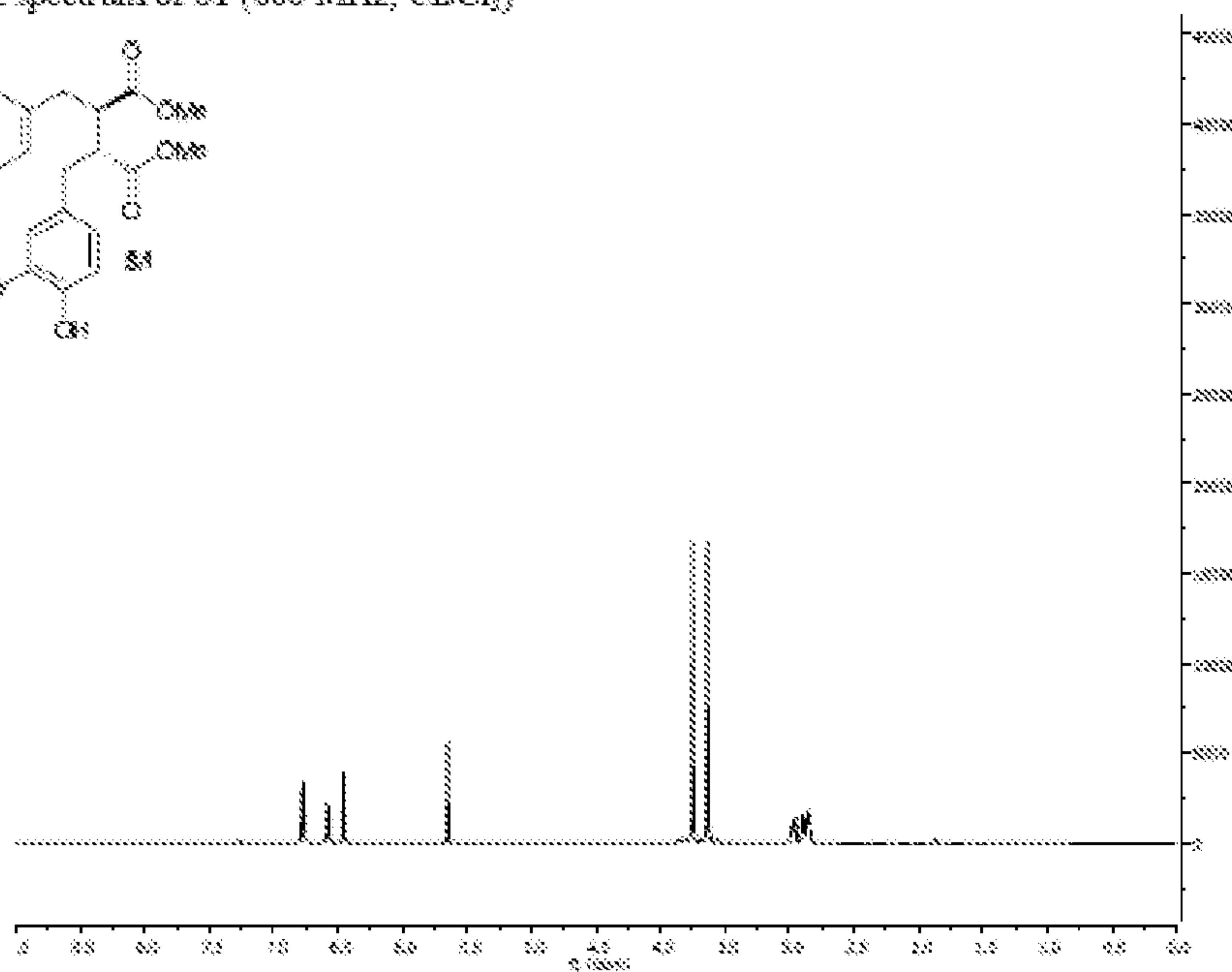
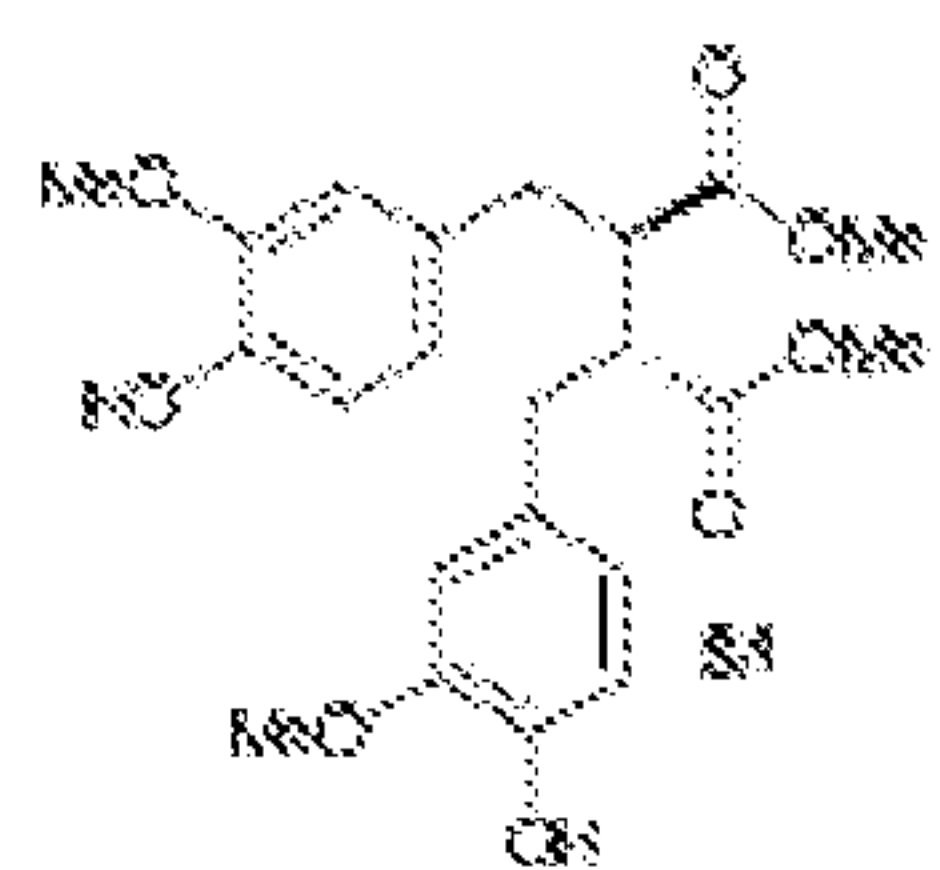
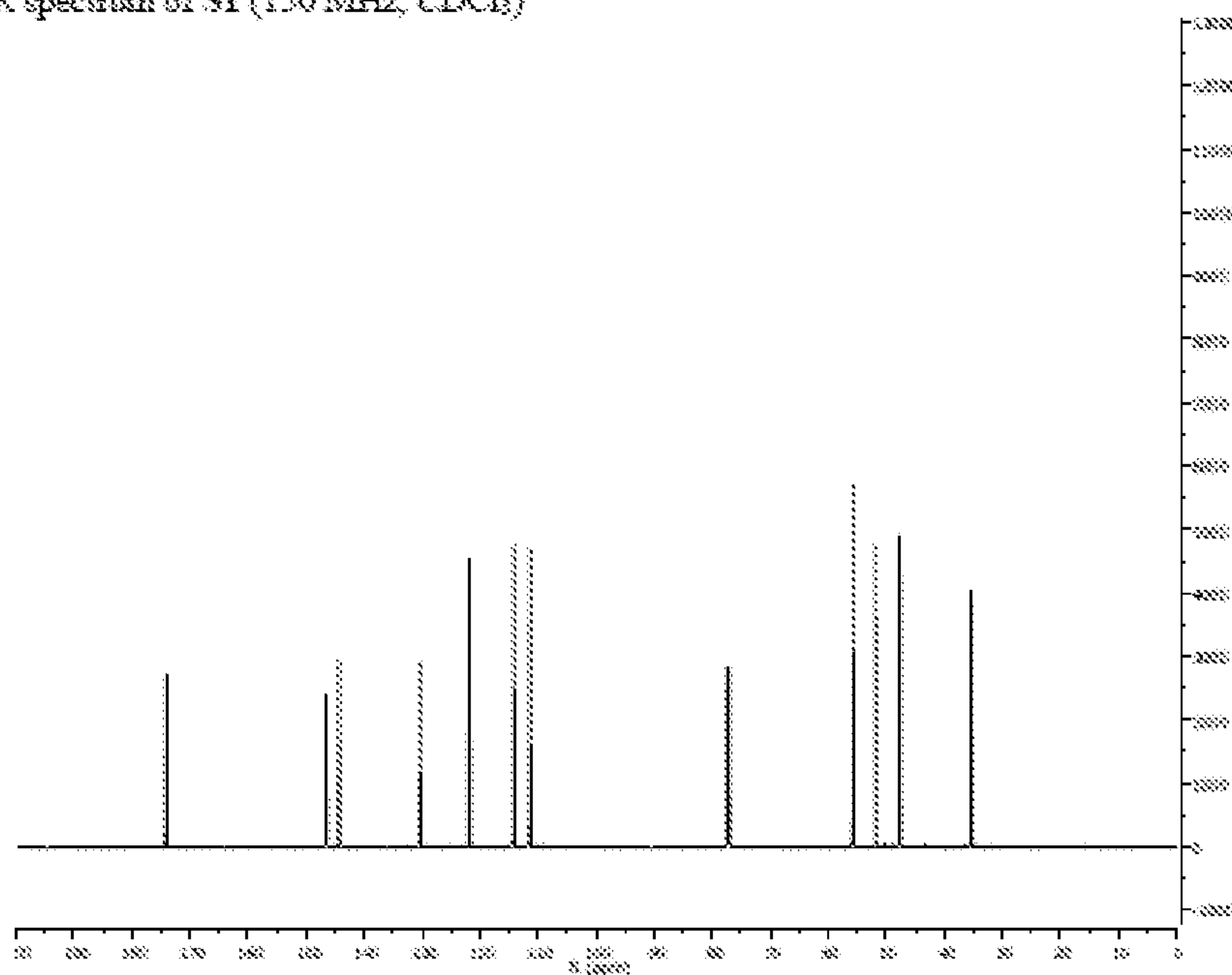
¹H NMR spectrum of S1 (600 MHz, CDCl₃)¹³C NMR spectrum of S1 (150 MHz, CDCl₃)

Figure 8

9 of 15

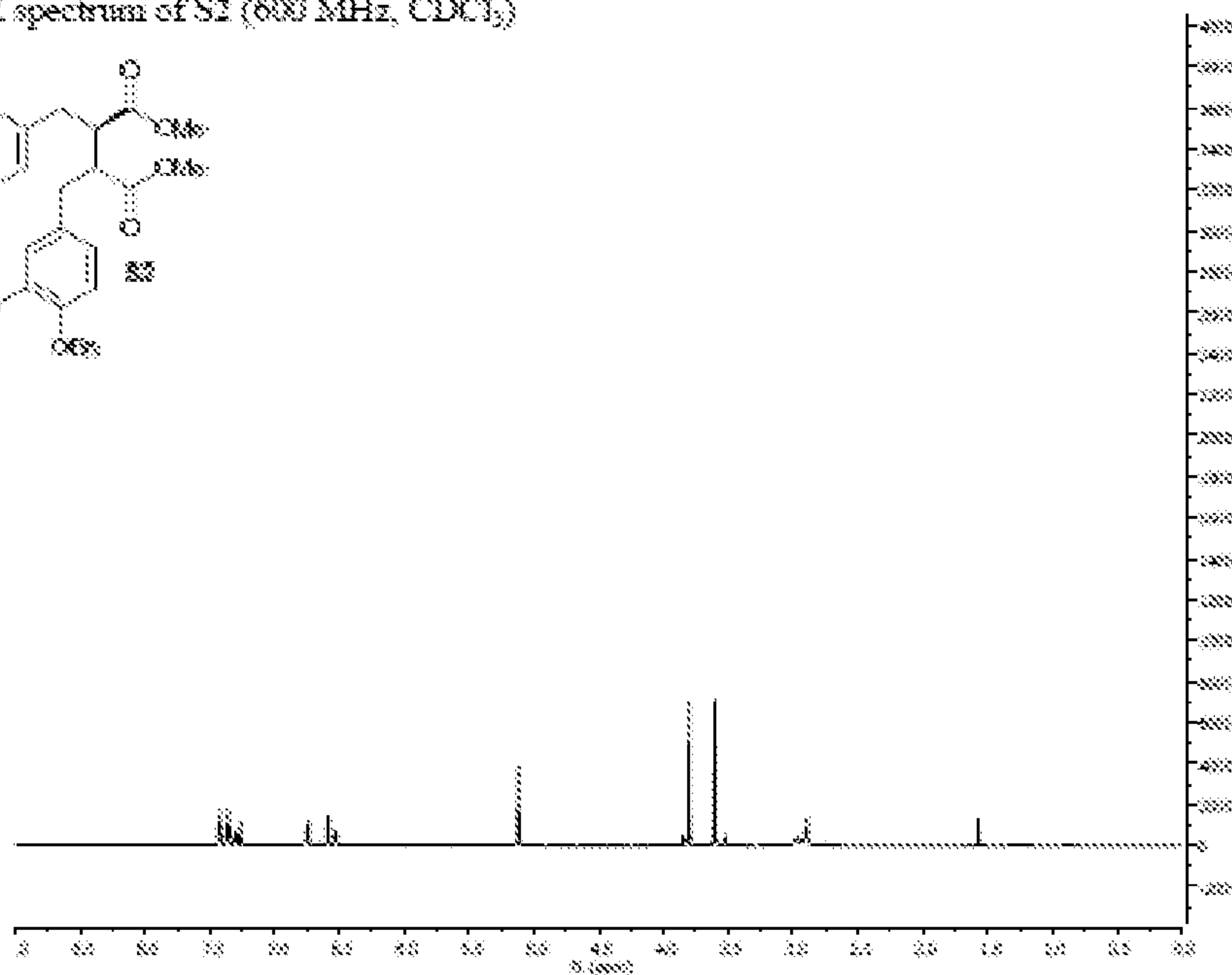
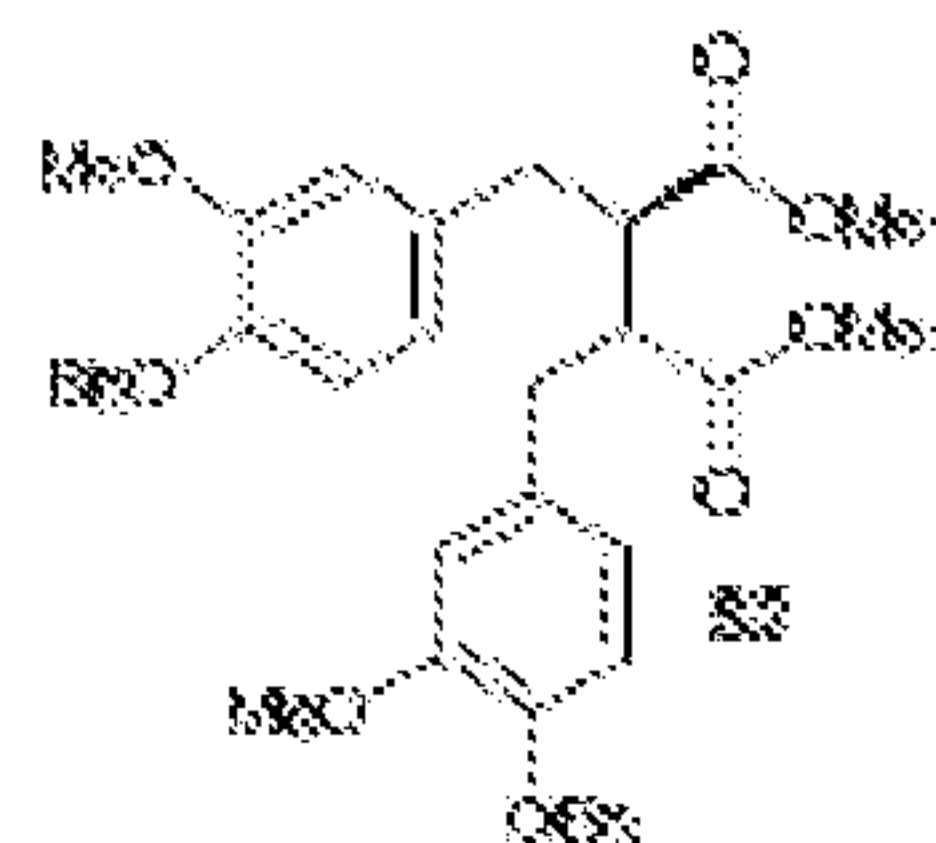
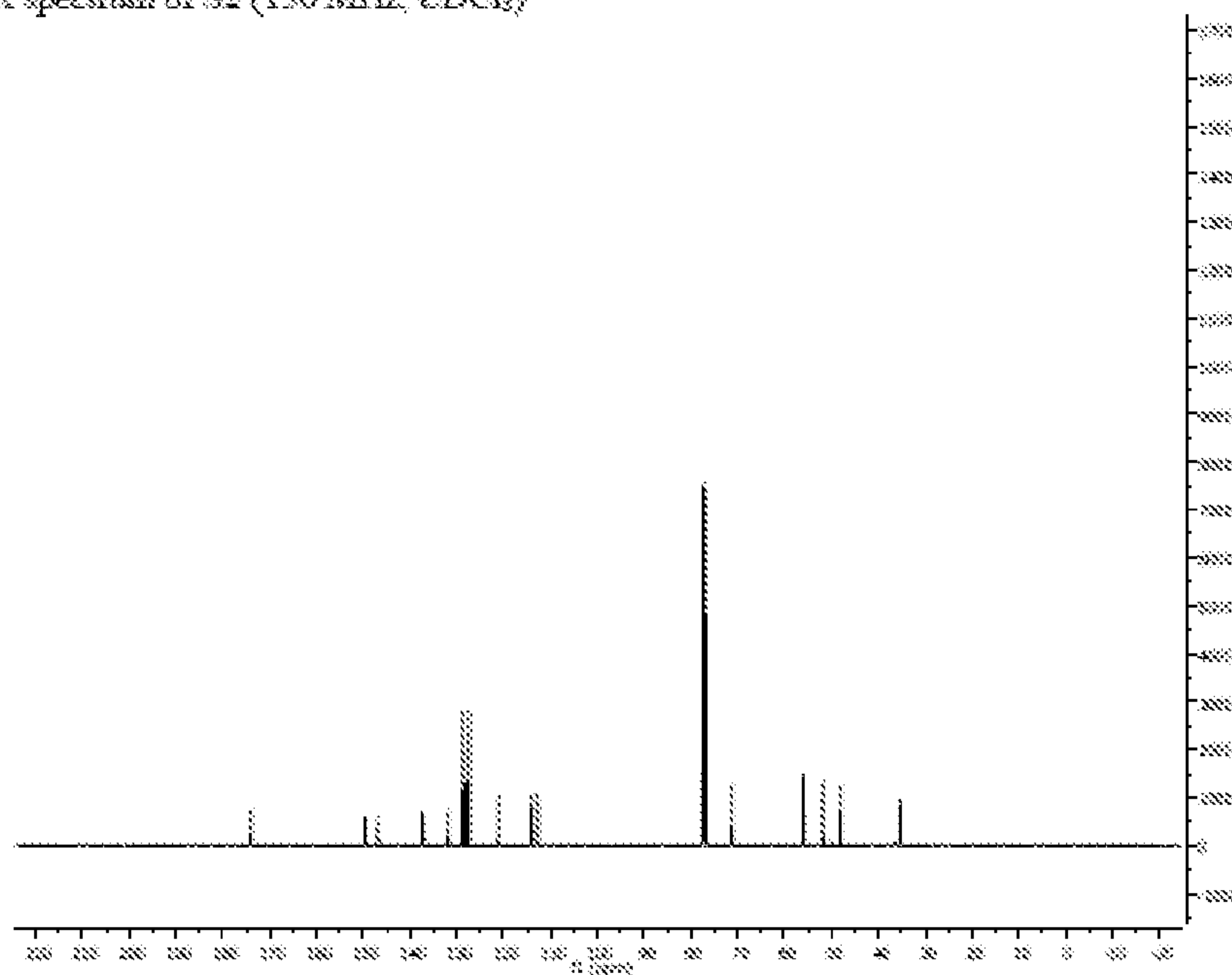
¹H NMR spectrum of S2 (600 MHz, CDCl₃)¹³C NMR spectrum of S2 (150 MHz, CDCl₃)

Figure 9

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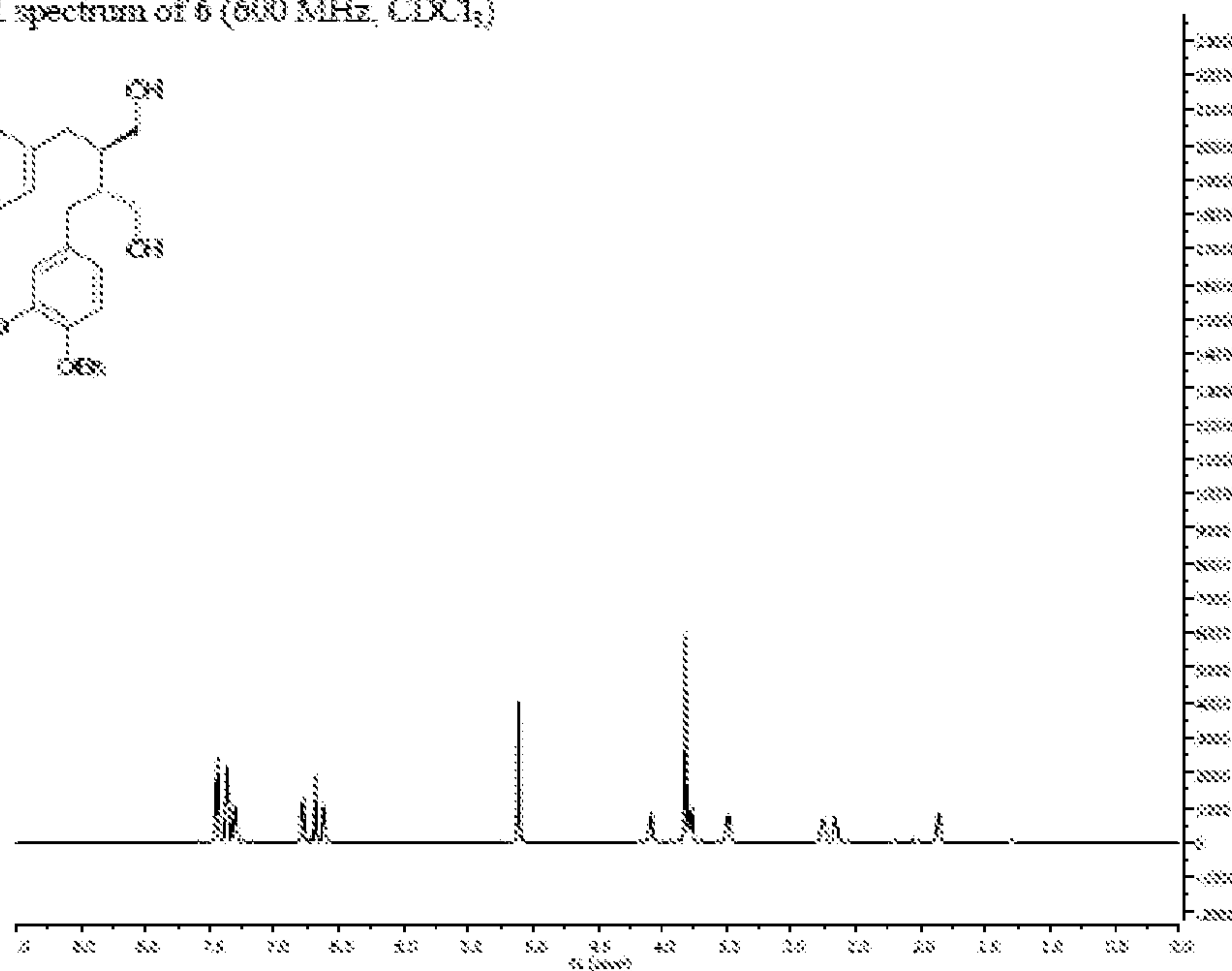
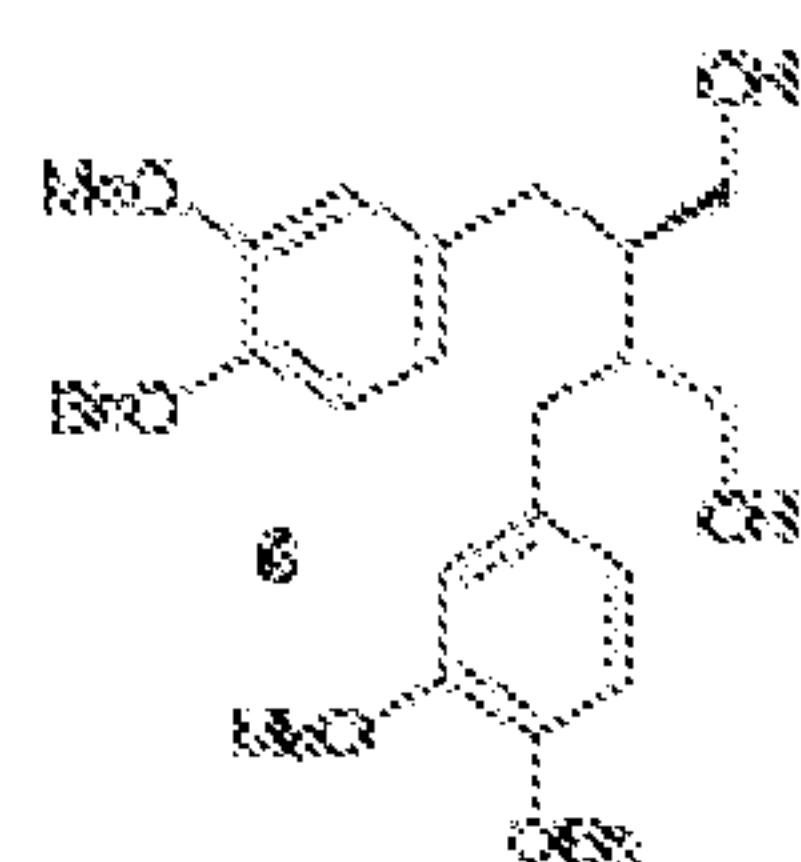
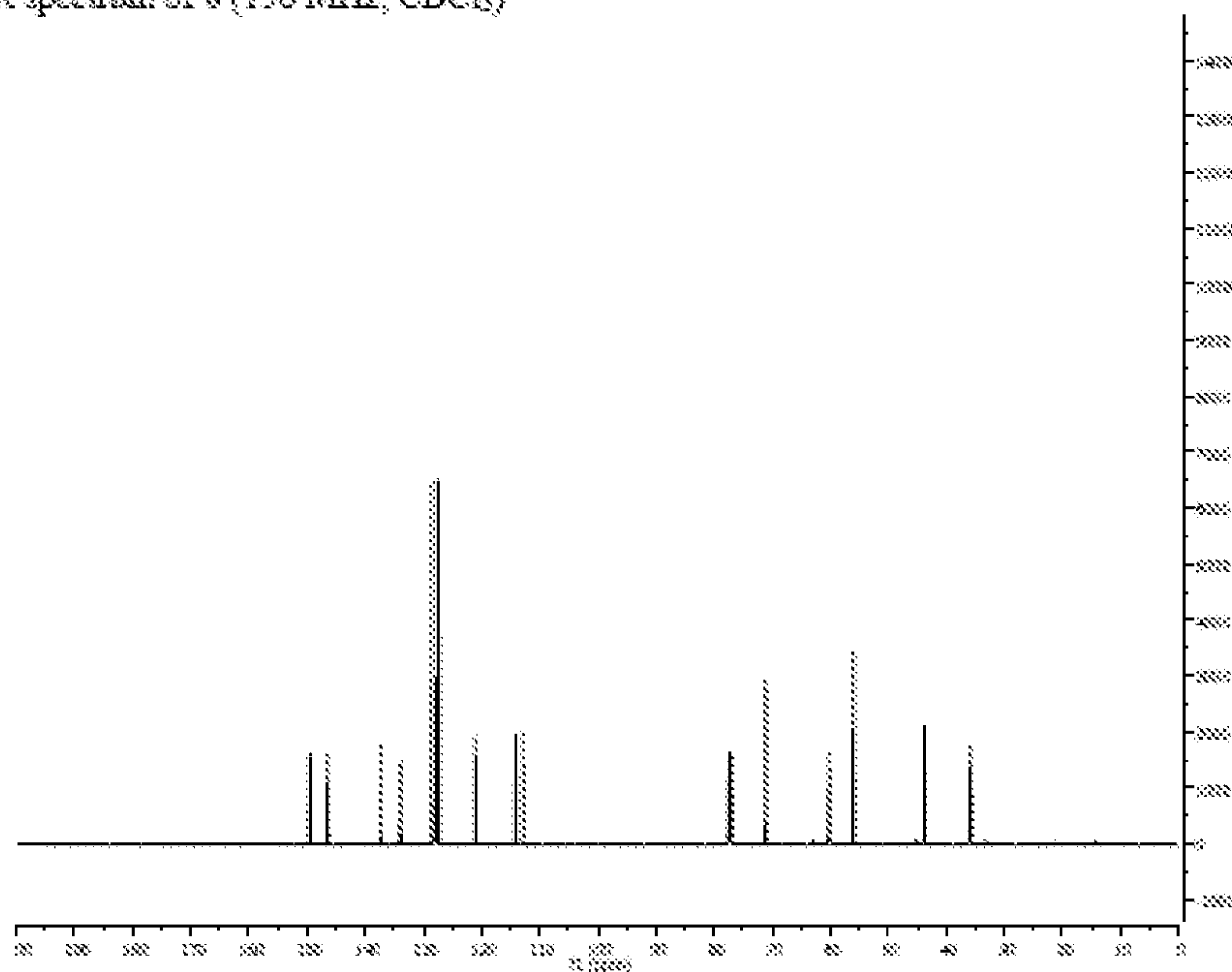
¹H NMR spectrum of 6 (600 MHz, CDCl₃)¹³C NMR spectrum of 6 (150 MHz, CDCl₃)

Figure 10

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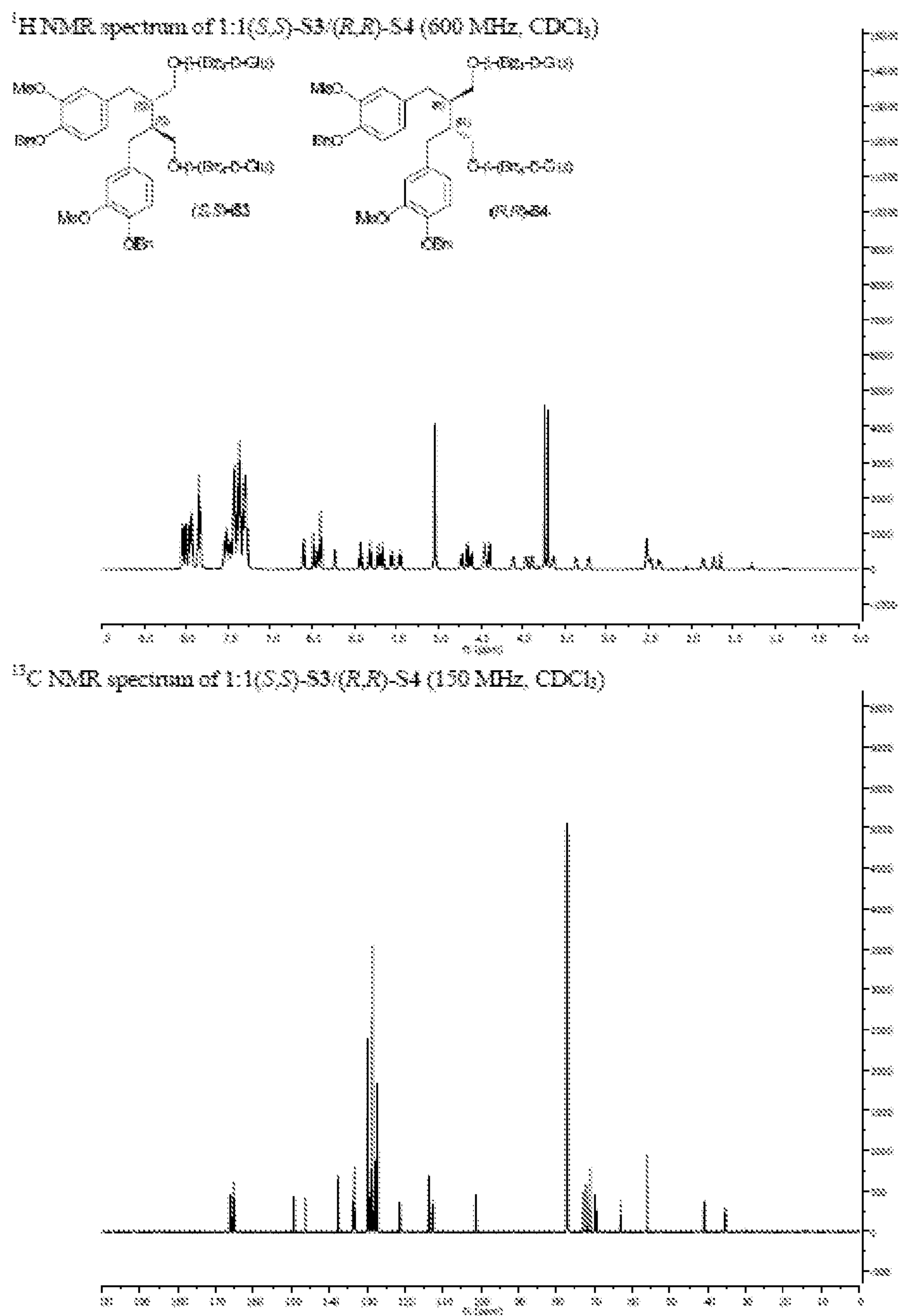


Figure 11

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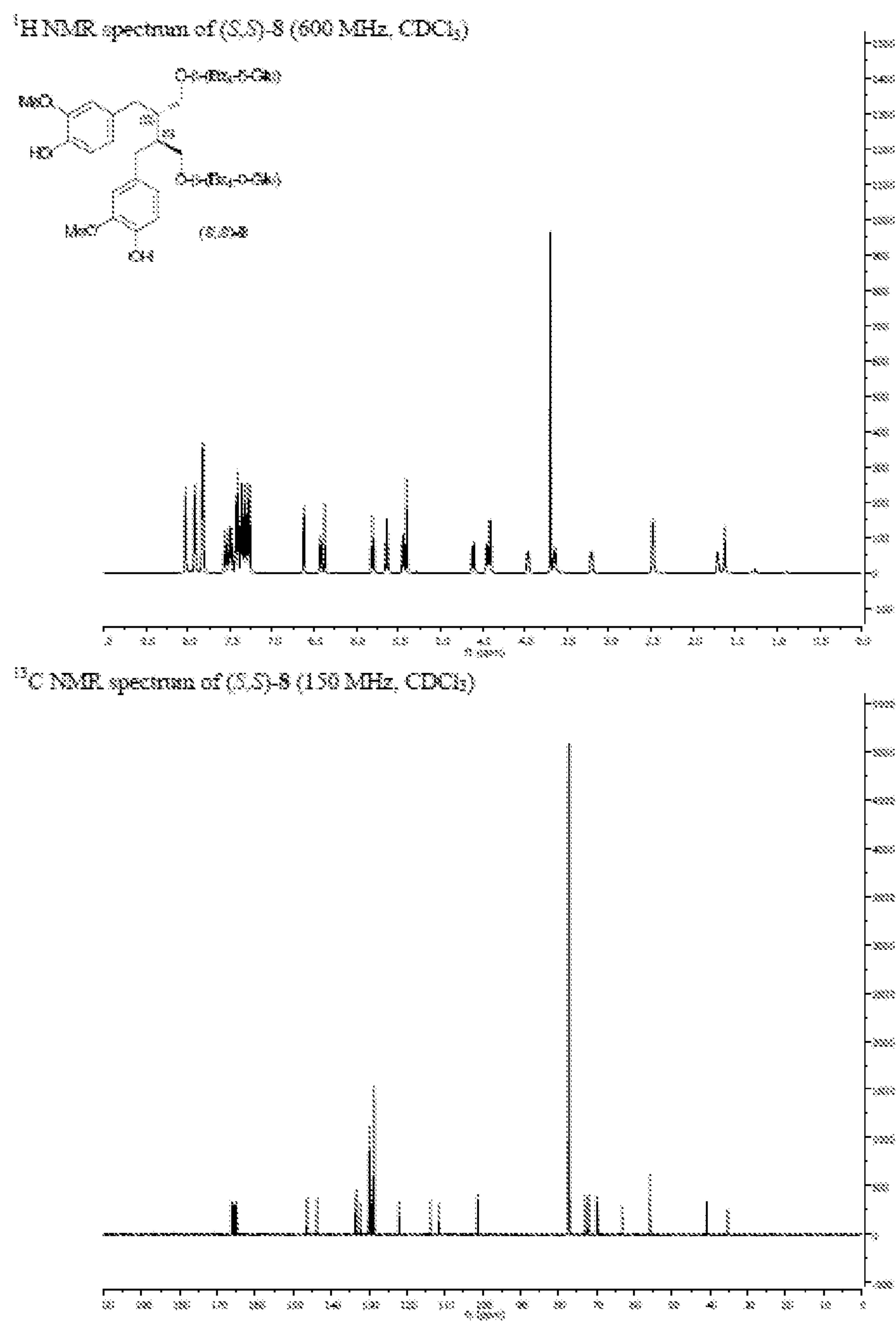


Figure 12

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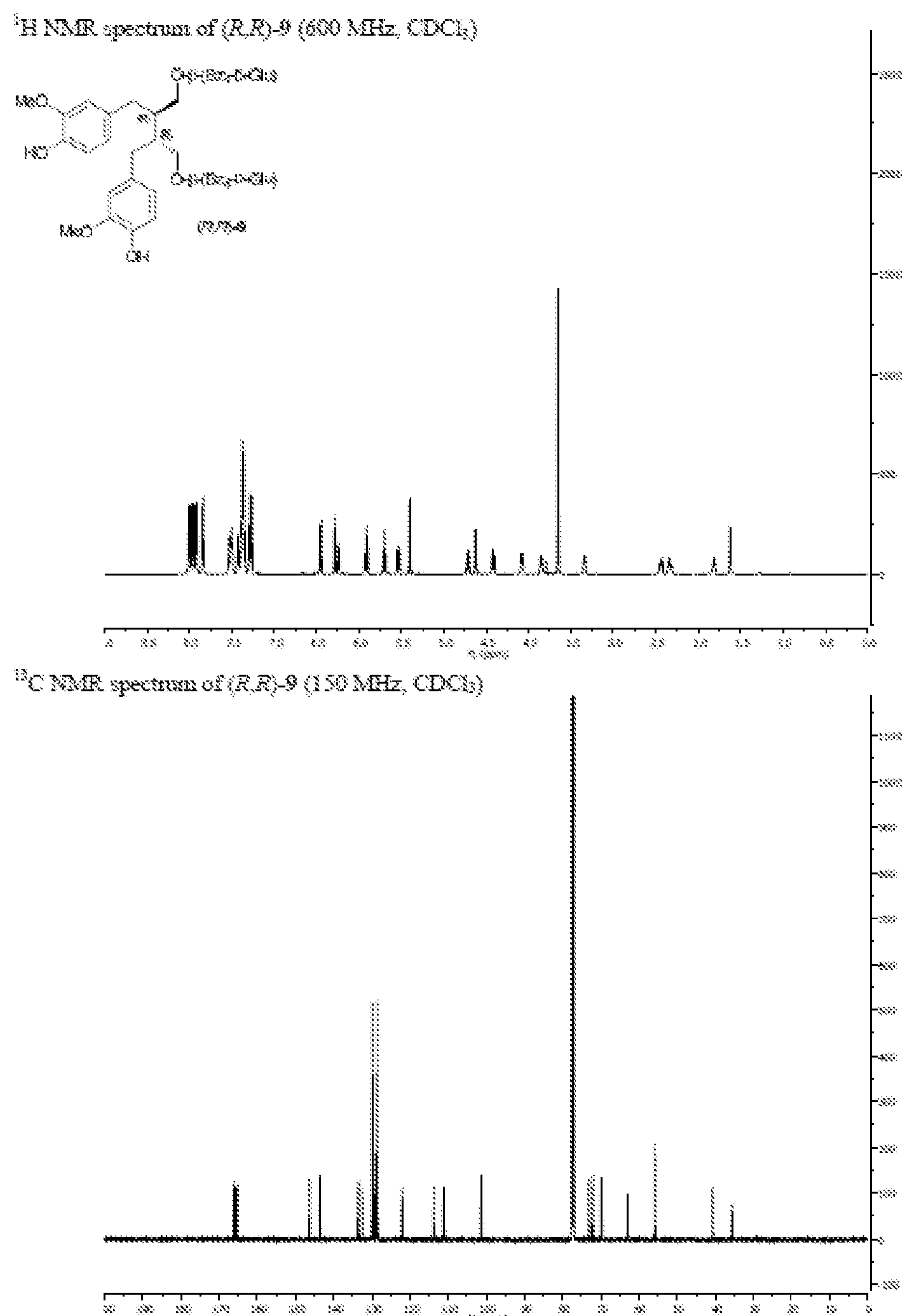


Figure 13

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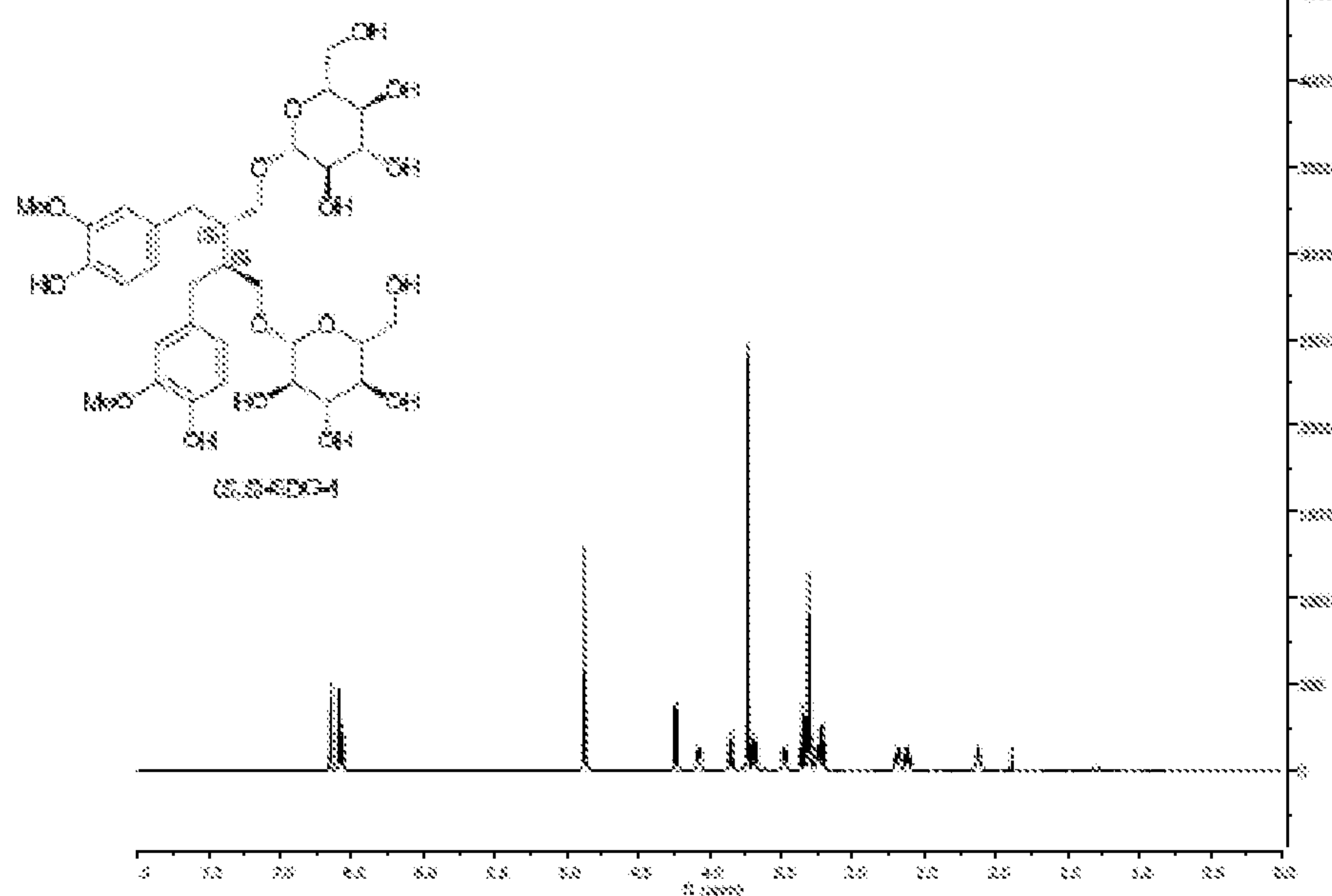
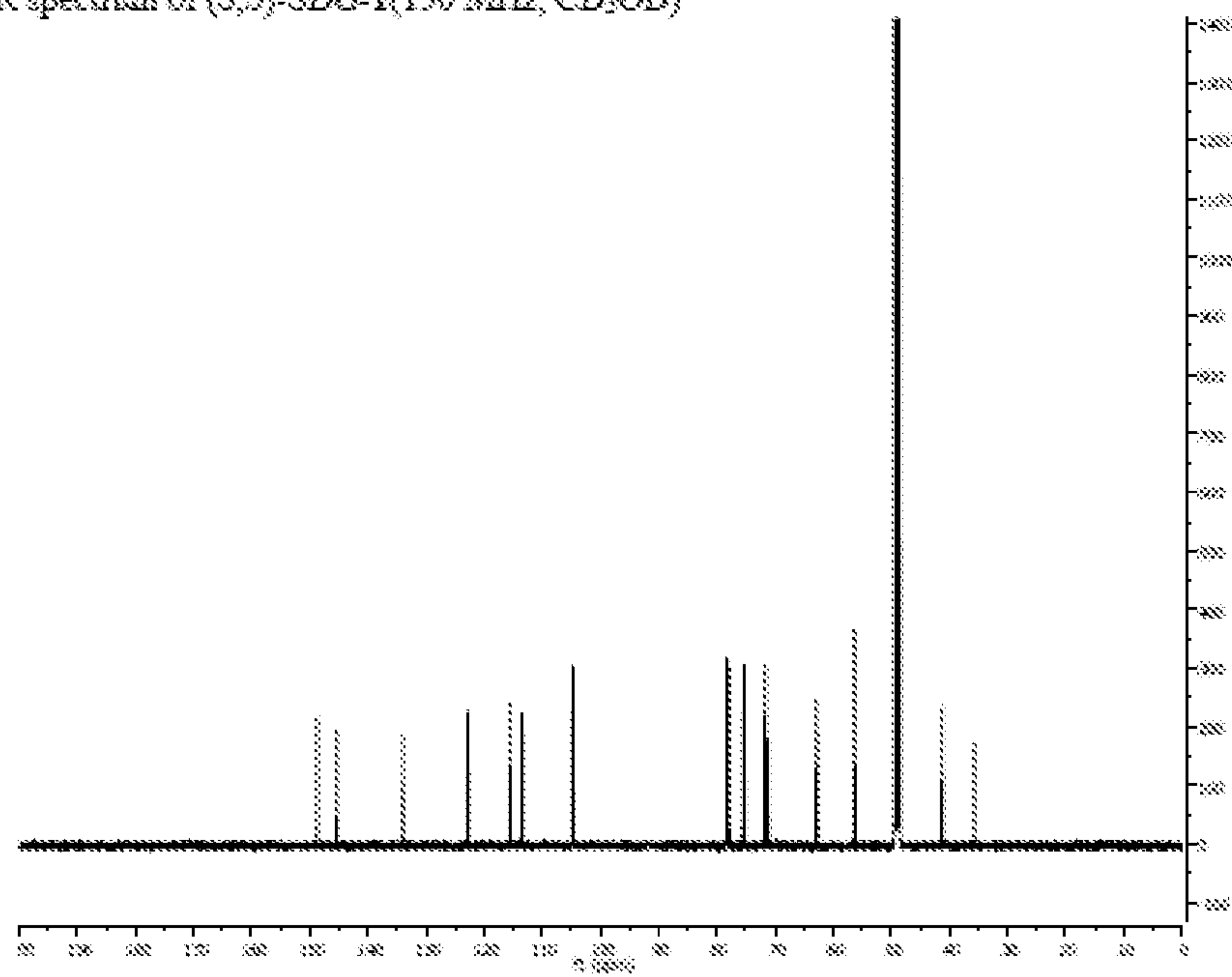
¹H NMR spectrum of (*S,S*)-SDG-1 (600 MHz, CD₃OD)¹³C NMR spectrum of (*S,S*)-SDG-1 (150 MHz, CD₃OD)

Figure 14

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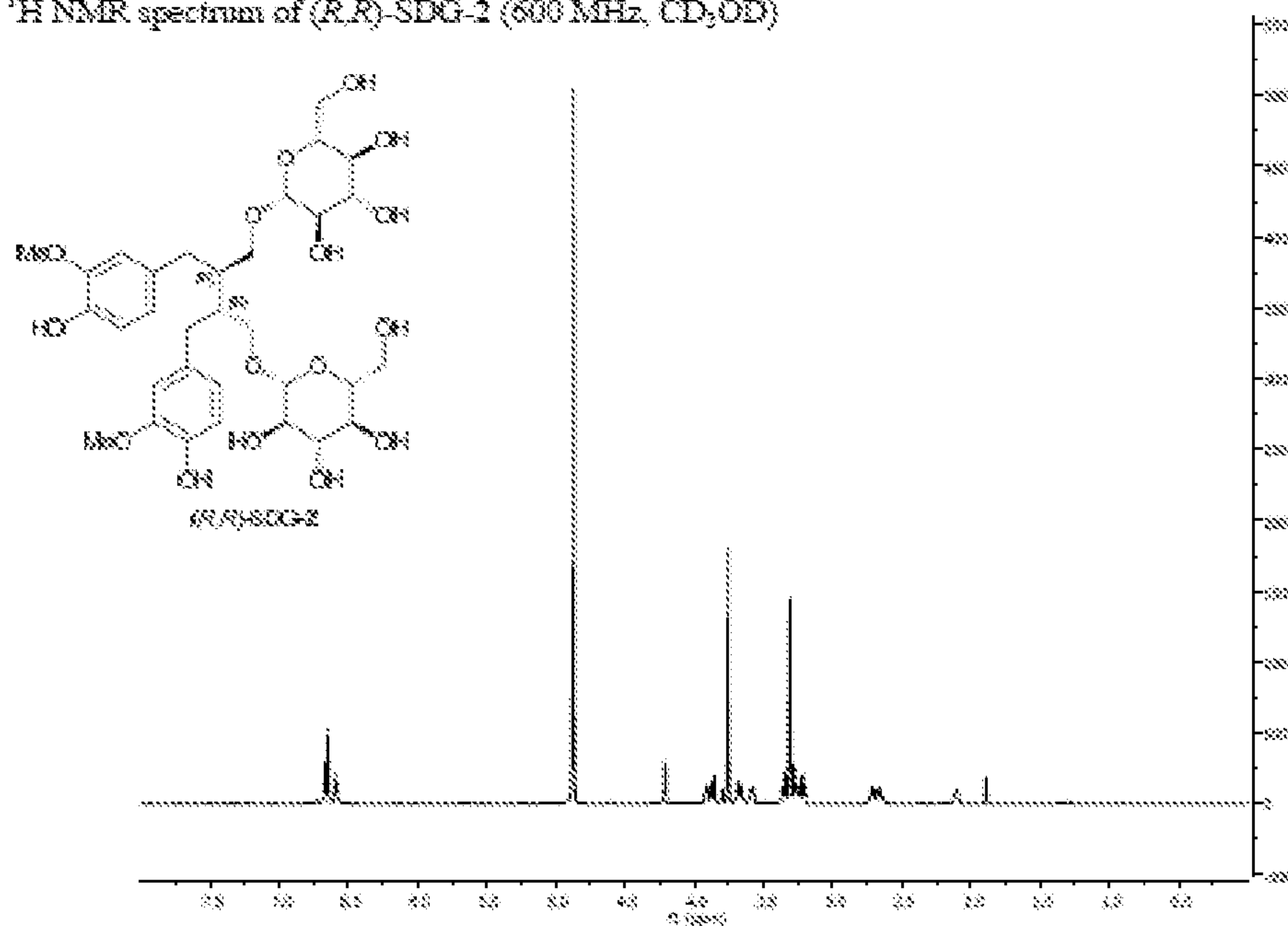
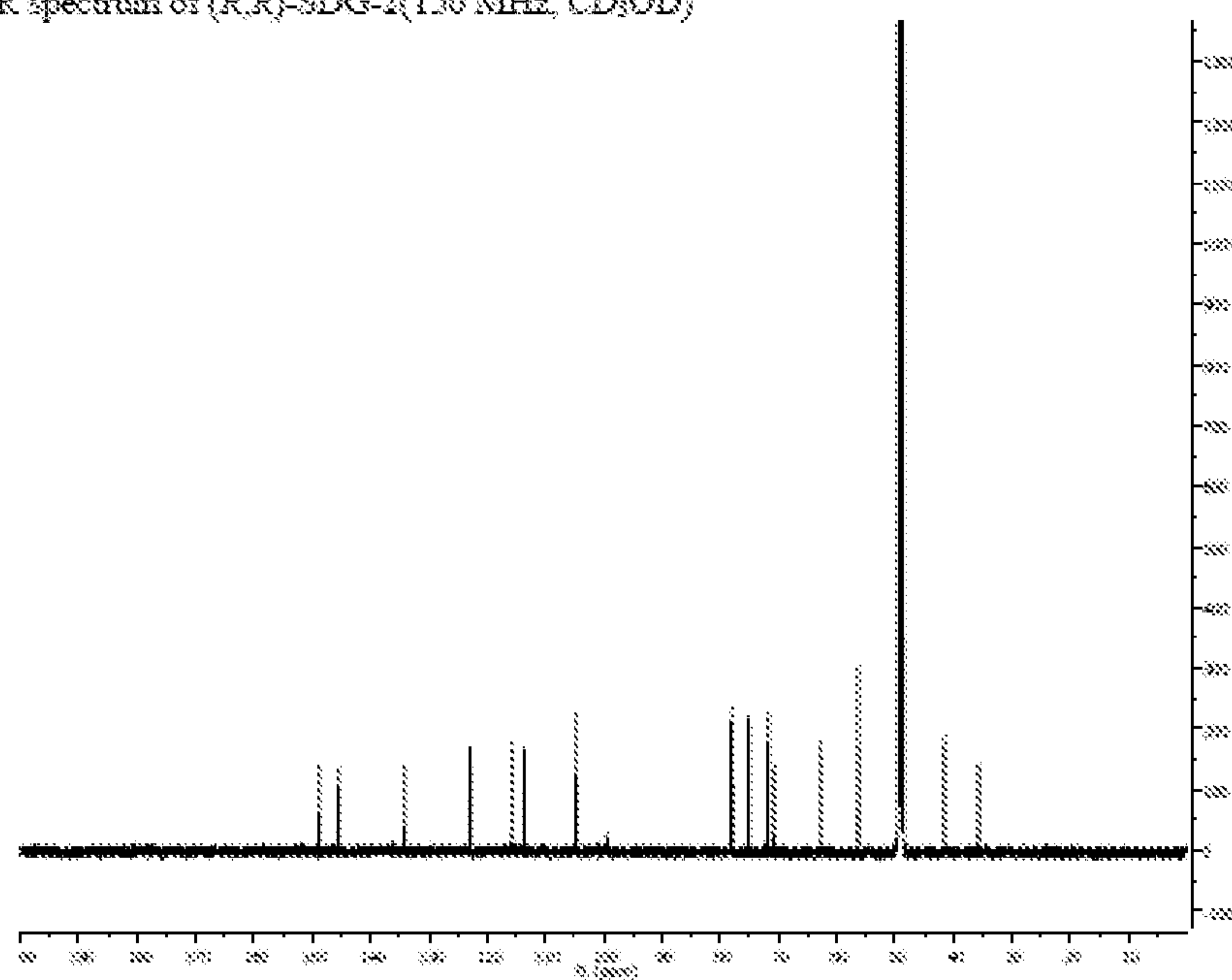
¹H NMR spectrum of (R,R)-SDG-2 (600 MHz, CD₃OD)¹³C NMR spectrum of (R,R)-SDG-2 (150 MHz, CD₃OD)

Figure 15

Figure 3

