Office de la Propriété Intellectuelle du Canada
Un organisme d'Industrie Canada

Canadian Intellectual Property Office
An agency of Industry Canada

CA 2628120 A1 2007/05/18

(21) 2 628 120

(12) DEMANDE DE BREVET CANADIEN
CANADIAN PATENT APPLICATION

(13) A1

(86) Date de dépôt PCT/PCT Filing Date: 2006/11/02
(87) Date publication PCT/PCT Publication Date: 2007/05/18
(85) Entrée phase nationale/National Entry: 2008/05/01
(86) N° demande PCT/PCT Application No.: US 2006/043082
(87) N° publication PCT/PCT Publication No.: 2007/056210
(30) Priorité/Priority: 2005/11/04 (US60/733,427)

(51) Cl.Int./Int.Cl. *A01N 27/00* (2006.01),
A61K 31/015 (2006.01)

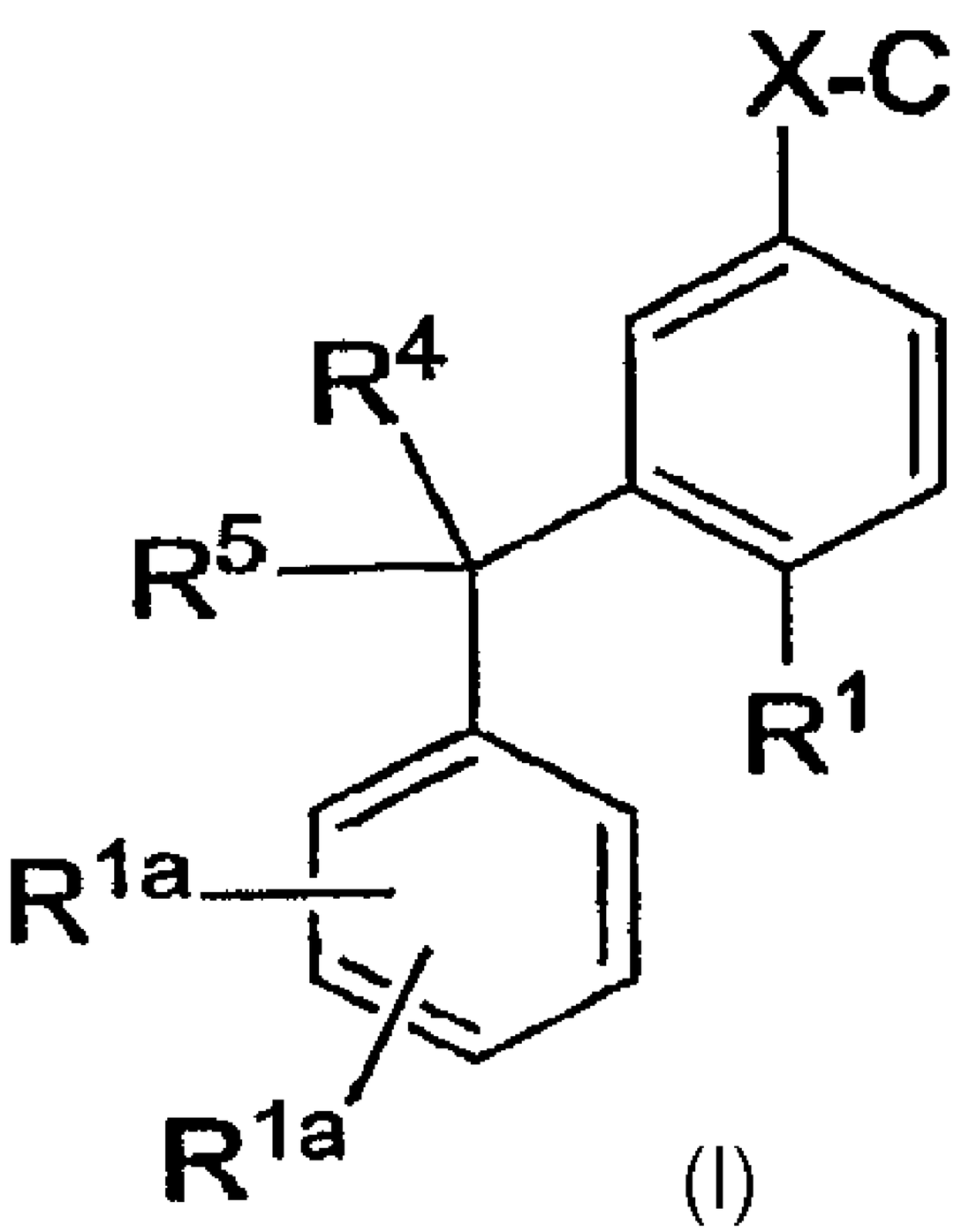
(71) Demandeur/Applicant:
MERCK & CO., INC., US

(72) Inventeurs/Inventors:
ARMSTRONG, HELEN M., US;
CHANG, LINDA L., US;
CHU, LIN, US;
SISCO, ROSEMARY, US;
OK, HYUN O., US;
XU, JINYOU, US;
UJJAINWALLA, FEROZE, US

(74) Agent: OGILVY RENAULT LLP/S.E.N.C.R.L.,S.R.L.

(54) Titre : DERIVES DE DIPHENYLMETHANE SERVANT D'INHIBITEURS DE LA BIOSYNTHESE DES LEUCOTRIENES

(54) Title: DIPHENYLMETHANE DERIVATIVES AS INHIBITORS OF LEUKOTRIENE BIOSYNTHESIS



(I)

(57) Abrégé/Abstract:
The instant invention provides compounds of Formula I which are 5-lipoxygenase activating protein inhibitors. Compounds of Formula I are useful as anti-atherosclerotic, anti-asthmatic, anti-allergic, anti-inflammatory, and cytoprotective agents.

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau

(43) International Publication Date
18 May 2007 (18.05.2007)

PCT

(10) International Publication Number
WO 2007/056210 A2

(51) International Patent Classification:
A61K 31/325 (2006.01)

(21) International Application Number:
PCT/US2006/043082

(22) International Filing Date:
2 November 2006 (02.11.2006)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
60/733,427 4 November 2005 (04.11.2005) US

(71) Applicant *(for all designated States except US)*: **MERCK & CO., INC.** [US/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US).

(72) Inventors; and

(75) Inventors/Applicants *(for US only)*: **ARMSTRONG, Helen, M.** [US/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **CHANG, Linda, L.** [CA/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **CHU, Lin** [US/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **SISCO, Rosemary** [US/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **OK, Hyun, O.** [US/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **XU, Jinyou** [CN/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US). **UJJAINWALLA, Feroze** [GB/US]; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US).

(74) Common Representative: **MERCK & CO., INC.**; 126 East Lincoln Avenue, Rahway, New Jersey 07065-0907 (US).

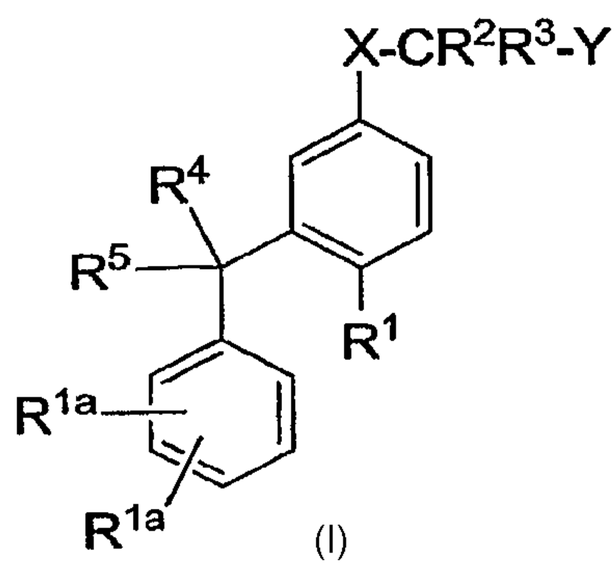
(81) Designated States *(unless otherwise indicated, for every kind of national protection available)*: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States *(unless otherwise indicated, for every kind of regional protection available)*: ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:
— *without international search report and to be republished upon receipt of that report*

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: DIPHENYLMETHANE DERIVATIVES AS INHIBITORS OF LEUKOTRIENE BIOSYNTHESIS



(I)

(57) Abstract: The instant invention provides compounds of Formula I which are 5-lipoxygenase activating protein inhibitors. Compounds of Formula I are useful as anti-atherosclerotic, anti-asthmatic, anti-allergic, anti-inflammatory, and cytoprotective agents.

WO 2007/056210 A2

TITLE OF THE INVENTION

DIPHENYLMETHANE DERIVATIVES AS INHIBITORS OF LEUKOTRIENE BIOSYNTHESIS

FIELD OF THE INVENTION

The instant invention involves compounds that inhibit 5-lipoxygenase activating protein (FLAP), compositions containing such compounds and methods of treatment using such compounds for the treatment and prevention of atherosclerosis and related diseases and conditions.

BACKGROUND OF THE INVENTION

Inhibition of leukotriene biosynthesis has been an active area of pharmaceutical research for many years. Leukotrienes are potent contractile and inflammatory mediators derived through the oxygenation of arachidonic acid by 5-lipoxygenase.

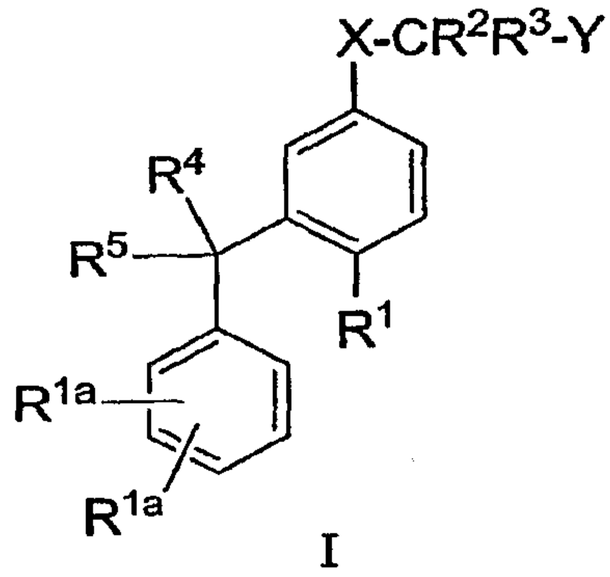
One class of leukotriene biosynthesis inhibitors are those known to act through inhibition of 5-lipoxygenase (5-LO). In general, 5-LO inhibitors have been sought for the treatment of allergic rhinitis, asthma and inflammatory conditions including arthritis. One example of a 5-LO inhibitor is the marketed drug zileuton, which is indicated for the treatment of asthma. More recently, it has been reported that 5-LO may be an important contributor to the atherogenic process; see Mehrabian, M. et al., Circulation Research, 2002 Jul 26, 91(2):120-126.

A new class of leukotriene biosynthesis inhibitors (now known as FLAP inhibitors) distinct from 5-LO inhibitors is described in Miller, D.K. et al., Nature, vol. 343, No. 6255, pp. 278-281, 18 Jan 1990. These compounds inhibit the formation of cellular leukotrienes but have no direct effect on soluble 5-LO activity. These compounds were used to identify and isolate the inner nuclear membrane 18,000 dalton protein 5-lipoxygenase-activating protein (FLAP). In cells, arachidonic acid is released from membrane phospholipids by the action of cytosolic phospholipase 2. This arachidonic acid is transferred to nuclear membrane bound 5-lipoxygenase by FLAP. The presence of FLAP in cells is essential for the synthesis of leukotrienes. Additionally, based on studies described in Helgadottir, A., et al., Nature Genetics, vol 36, no. 3 (March 2004) 233-239, it is believed that the gene encoding 5-lipoxygenase activating protein confers risk for myocardial infarction and stroke in humans.

Despite significant therapeutic advances in the treatment and prevention of atherosclerosis and ensuing atherosclerotic disease events, such as the improvements that have been achieved with HMG-CoA reductase inhibitors, further treatment options are clearly needed. The instant invention addresses that need by providing compounds, compositions and methods for the treatment or prevention of atherosclerosis as well as related conditions.

SUMMARY OF THE INVENTION

The instant invention relates to compounds of Formula I which are FLAP inhibitors, methods for their preparation, and methods and pharmaceutical formulations for using these compounds in mammals, especially humans. This invention provides compounds of structural Formula I:



I

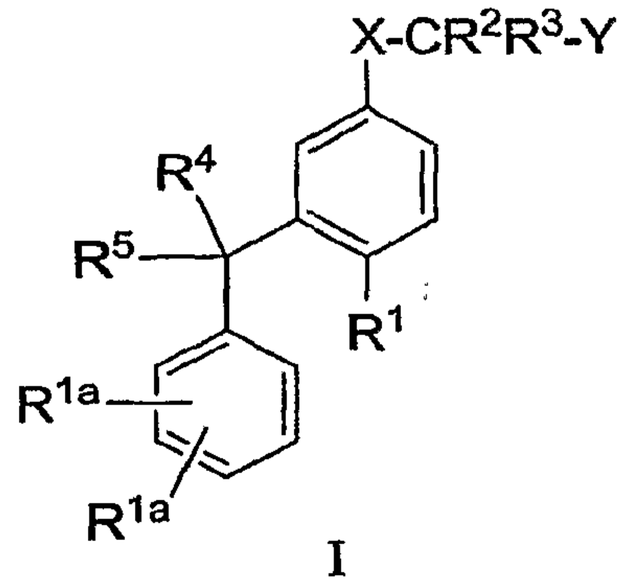
and pharmaceutically acceptable salts and solvates thereof. This invention also involves the use of compounds described herein to slow or halt atherogenesis. Therefore, one object of the instant invention is to provide a method for treating atherosclerosis, which includes halting or slowing the progression of atherosclerotic disease once it has become clinically evident, comprising administering a therapeutically effective amount of a compound of Formula I to a patient in need of such treatment. Another object is to provide methods for preventing or reducing the risk of developing atherosclerosis and atherosclerotic disease events, comprising administering a prophylactically effective amount of a compound of Formula I to a patient who is at risk of developing atherosclerosis or having an atherosclerotic disease event.

The compounds of Formula I are also useful as anti-asthmatic, anti-allergic, anti-inflammatory and cytoprotective agents. They are also useful in treating angina, cerebral spasm, glomerular nephritis, hepatitis, endotoxemia, uveitis, and allograft rejection. The instant invention provides methods of treatment comprising administering a therapeutically effective amount of a compound of Formula I to a patient in need of the above-described treatments.

A further object is to provide the use of FLAP inhibitors of Formula I in combination with other therapeutically effective agents, including other anti-atherosclerotic drugs. These and other objects will be evident from the description contained herein.

DETAILED DESCRIPTION OF THE INVENTION

The instant invention provides compounds represented by structural Formula I:



I

and pharmaceutically acceptable salts and solvates thereof wherein:

each R^{1a} is independently selected from the group consisting of -H, -F, -Cl, -Br, -C_{1-6}alkyl, -CN, -OH, C_{1-6}alkyl-OH, -OC_{1-6} alkyl, -fluoroC_{1-6} alkyl, -fluoroC_{1-6} alkoxy, -NH_2,
-NHC_{1-6}alkyl, -N(C_{1-6}alkyl)$_2$, -C_{1-6}alkyl-NH_2, -C_{1-6}alkyl-NHC_{1-6}alkyl, -C_{1-6}alkyl-N(C_{1-6}alkyl)$_2$, -NHC(O)C_{1-6}alkyl, -CO_2C_{1-6}alkyl, -C(O)NHC_{1-6}alkyl, and -C(O)N(C_{1-6}alkyl)$_2$;

R^1 is selected from the group consisting of:

a) Z^1,

b) -CO_2R^a, -C(O)NR^aR^b, -N(R^a)$_2$, -$NR^bSO_pR^a$, -$NR^bC(O)R^a$, -$NR^bC(O)NR^aR^b$, -$NR^bCO_2R^a$, -O(O)NR^aR^b, -OH and -CN,

c) -C_{1-6}alkyl, -C_{2-6} alkenyl, -C_{2-6}alkynyl, -OC_{1-6}alkyl, -OC_{2-6}alkenyl and –OC_{2-6}alkynyl, said groups being optionally substituted with R^6 and optionally substituted with R^7, wherein R^6 is selected from the group consisting of -CO_2R^a, -C(O)NR^aR^b, -N(R^a)$_2$, -$NR^bSO_pR^a$, -$NR^bC(O)R^a$, -$NR^bC(O)NR^aR^b$, -$NR^bCO_2R^a$, -OC(O)NR^aR^b, -C(O)$SO_pNR^aR^b$, -C(O)$NR^bNR^aR^b$, -S(O)$_pNR^aR^b$, -$SO_pNR^bC(O)R^a$, -S(O)$_pR^a$, –F , -CF_3, phenyl, Hetcy, and Z^1; and R^7 is selected from the group consisting of -F and –OH, and

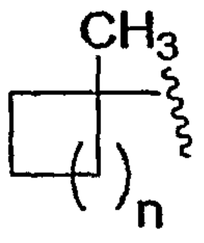
d) phenyl, optionally substituted with 1-2 members selected from the group consisting of: -F, -Cl, -C_{1-6}alkyl, -CN, -OH, -OC_{1-6} alkyl, -fluoroC_{1-6}alkyl, -fluoroC_{1-6}alkoxy, -NH_2, -NHC_{1-6}alkyl, -N(C_{1-6}alkyl)$_2$, -C_{1-6}alkyl-NH_2, -C_{1-6}alkyl-NHC_{1-6}alkyl, -C_{1-6}alkyl-N(C_{1-6}alkyl)$_2$, -C_{1-6}alkyl-CN, -NHC(O)C_{1-6}alkyl, -C(O)NHC_{1-6}alkyl, and -C(O)N(C_{1-6}alkyl)$_2$;

R^2 is selected from the group consisting of –H and -C_{1-6}alkyl optionally substituted with a group selected from –OH and –F;

R^3 is selected from the group consisting of –H and -C_{1-6}alkyl;

R^4 is selected from the group consisting of hydrogen, fluorine, hydroxy, C_{1-3} alkyl optionally substituted with one to five fluorines;

R^5 is selected from the group consisting of (a) C_{1-6} alkyl optionally substituted with one to five fluorines, (b) C_{3-6} cycloalkyl, and (c)



;

n is an integer selected from 0, 1, 2, and 3;

each "p" independently represents an integer selected from 0, 1 and 2;

each R^a is independently selected from the group consisting of

a) –H,

b) -C_{1-4}alkyl, -C_{2-4}alkenyl, and –C_{2-4}alkynyl, wherein each is optionally substituted with 1-2 members selected from the group consisting of: -OH, -OC_{1-4}alkyl, -CN, -NH_2, -NHC_{1-4}alkyl, and -$N(C_{1-4}alkyl)_2$, -F, and –CF_3,

c) phenyl and phenyl-C_{1-4}alkyl-, the phenyl moieties being optionally substituted with 1-2 members selected from the group consisting of -F, -Cl, -C_{1-4} alkyl, -CN, -OH, -OC_{1-4} alkyl, -fluoroC_{1-4}alkyl, -fluoroC_{1-4}alkoxy, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, -C_{1-4}alkyl-NH_2, -C_{1-4}alkyl-NHC_{1-4}alkyl, -C_{1-4}alkyl-$N(C_{1-4}alkyl)_2$, -C_{1-4}alkyl-CN, -$NHC(O)C_{1-4}$alkyl, -$C(O)NHC_{1-4}$alkyl, and -$C(O)N(C_{1-4}alkyl)_2$,

and the alkyl portion of phenyl-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of -OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, and 1-3 of fluoro,

d) Hetcy and Hetcy-C_{1-4}alkyl-, the Hetcy moieties being optionally substituted on carbon with 1-2 members selected from the group consisting of -F, -OH, -CO_2H, -C_{1-4}alkyl, -CO_2C_{1-4}alkyl, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, -$NHC(O)C_{1-4}$alkyl, oxo, -$C(O)NHC_{1-4}$alkyl and -$C(O)N(C_{1-4}alkyl)_2$; and optionally substituted on nitrogen when present with a group selected from -C_{1-4} alkyl, and -C_{1-4} acyl,

and the alkyl portion of Hetcy-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of –OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$ and 1-3 of fluoro,

e) Z^2 and Z^2-C_{1-4}alkyl- and the alkyl portion of Z^2-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of –OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$ and 1-3 of fluoro;

each R^b is independently selected from the group consisting of -H and -C_{1-4} alkyl optionally substituted with 1-2 members selected from the group consisting of NH_2, -OH, -F, -CN, and -CF_3;

X is selected from the group consisting of -O-, $S(O)_p$, NR^b, and -CHR^8-, wherein R^8 is selected from the group consisting of -H, -OH and -C_{1-6} alkyl optionally substituted with a group selected from -OH and -F;

Y is selected from the group consisting of:

a) a 9-membered unsaturated *ortho*-fused bicyclic ring system containing 2-3 heteroatoms selected from the group consisting of –N=, –NH–, –N(Me)–, –S– and –O–, and wherein the ring system is optionally substituted with 1-3 of fluoro,

b) a 10-membered aromatic *ortho*-fused bicyclic ring system containing 1-3 of –N= , wherein the ring system is optionally substituted with 1-3 of fluoro, and

c) pyridinyl substituted with a group selected from -C_{1-4}alkyl, -F, -CF_2H, and CF_3, and optionally having a second substituent which is -C_{1-4}alkyl;

Hetcy is selected from the group consisting of azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl, morpholinyl, tetraydrofuranyl, β–lactamyl, δ–lactamyl and γ-lactamyl;

Z^1 is selected from the group consisting of:

a) a 5-membered unsaturated heterocyclic ring containing 2-4 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4} alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro,

b) a 5-membered unsaturated heterocyclic ring containing 2-3 heteroatoms selected from one oxygen or one sulfur and 1-2 of nitrogen, wherein one nitrogen in the ring is optionally substituted with a group selected from C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from

-OH, -SH, =O, =S, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl, and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro,

c) a 6-membered unsaturated heterocyclic ring containing 1-2 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon atom in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro,

d) an 8-membered unsaturated ortho-fused bicyclic ring system containing 3-5 heteroatoms selected from one sulfur and 2-4 of nitrogen atoms wherein one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro, and

e) a 9-membered unsaturated *ortho*-fused bicyclic ring system containing 3-4 nitrogen atoms, wherein one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl, and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro; and

Z^2 is is selected from the group consisting of:

a) a 5-membered unsaturated heterocyclic ring containing 2-4 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, =O, =S, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl, and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro,

b) a 5-membered unsaturated heterocyclic ring containing 2-3 heteroatoms selected from one oxygen or one sulfur and 1-2 of nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, and C_{1-4}alkyl optionally substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro, and

c) a 6-membered unsaturated heterocyclic ring containing 1-2 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon atom in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro.

The invention is described herein in detail using the terms defined below unless otherwise specified. "Alkyl", as well as other groups having the prefix "alk", such as alkoxy, alkanoyl and the like, means carbon chains which may be linear, branched or cyclic, or combinations thereof, containing the indicated number of carbon atoms. "Non-cyclic alkyl"is a subset of alkyl and means linear and branched alkyl, and does not include cycloalkyl. If no number is specified, 1-10 carbon atoms are intended for linear or branched alkyl groups. Cycloalkyl, which must have a minimum of 3 carbons to form a carbocyclic ring, is a subset of alkyl and is also intended to be included within the meaning of "alkyl" when the specified number of carbon atoms for an alkyl group encompasses three or more carbon atoms, or when no number of carbon atoms is specified. As a result, each occurrence of the term "alkyl" independently represents the group consisting of (a) non-cyclic alkyl, (b) cycloalkyl and (c) a combination of non-cyclic alkyl with cycloalkyl. Therefore, it is understood that when "C_{1-3}alkyl" is recited, this encompasses linear and branched 1-3 carbon chains and cyclopropyl. Similarly, when "C_{1-4}alkyl" is recited, this encompasses linear and branched 1-4 carbon chains as well as cyclopropyl, -CH_2-cyclopropyl, -cyclopropyl-CH_3 and cyclobutyl. Similarly, when "C_{1-6} alkyl" is recited, this encompasses linear and branched 1-6 carbon chains and C_{3-6} cycloalkyl, as well as combinations of non-cyclic alkyl with C_{3-5}cycloalkyl which contain a total up to of six carbon atoms. Examples of alkyl groups include but are not limited to methyl, ethyl, propyl, isopropyl, butyl, *sec*- and *tert*-butyl, 1,1-dimethylbutyl, pentyl, isopentyl, hexyl, heptyl, octyl, nonyl and the like, as well as the cycloalkyl groups cyclopropyl, cyclobutyl, cyclopentyl, and cyclohexyl. Cyclopropyl and cyclobutyl are preferred cycloalkyl groups.

"Alkenyl" means carbon chains which contain at least one carbon-carbon double bond, and which may be linear or branched or combinations thereof, containing the indicated number of carbon atoms, and more particularly 3-6 carbons. Examples of alkenyl include vinyl, allyl, isopropenyl, pentenyl, hexenyl, heptenyl, 1-propenyl, 2-butenyl, 2-methyl-2-butenyl, and the like.

"Alkynyl" means carbon chains which contain at least one carbon-carbon triple bond, and which may be linear or branched or combinations thereof, containing the indicated number of carbon atoms, and more particularly 3-6 carbons. Examples of alkynyl include ethynyl, propargyl, 3-methyl-1-pentynyl, 2-heptynyl and the like.

"Acyl" refers to an alkyl group as defined above linked through a carbonyl group. A preferred example is acetyl, $CH_3C(O)$-.

"Aryl" (Ar) means mono- and bicyclic aromatic rings containing 6-12 carbon atoms. Examples of aryl include phenyl, naphthyl, indenyl and the like.

"Halogen" (Halo) includes fluoro, chloro, bromo and iodo, preferably -F and -Cl, more preferably -F.

The phrase "8-membered unsaturated ortho-fused bicyclic ring system" as used herein means a 5 membered ring fused to a 5-membered ring wherein the rings have two, and only two, adjacent atoms in common, i.e., they are *ortho*-fused. The phrase "9-membered unsaturated *ortho*-fused bicyclic ring system" as used herein means a 6 membered ring and a 5-membered ring *ortho*-fused together. The phrase "10-membered aromatic *ortho*-fused bicyclic ring system" as used herein means two 6-membered rings *ortho*-fused together. Said bicyclic ring systems are comprised of carbon atoms and the indicated number and kind of heterotaoms, and may be substituted as defined herein. The term"unsaturated" encompasses both aromatic rings as well as non-aromatic unsaturated rings.

"Hetcy" can be linked to a compound of structural Formula I via carbon or nitrogen in the Hetcy ring. Each of "Z^1" and "Z^2" can be linked to a compound of structural Formula I via carbon or nitrogen in the Z^1 or Z^2 ring or ring system, and is preferably linked via carbon. "Y" can be linked to a compound of structural Formula I via carbon or nitrogen in the Y ring or ring system, and is preferably linked via carbon.

The term "optionally substituted" means "unsubstituted or substituted," and therefore, the genus described herein encompasses compounds containing the specified optional substituent as well as compounds that do not contain the optional substituent. For example, the phrase "-C_{1-3}alkyl optionally substituted with a group selected from –OH and -F" encompasses unsubstituted -C_{1-3}alkyl, fluoro susbtituted -C_{1-3}alkyl and hydroxy substituted -C_{1-3}alkyl.

Reference to the compounds of this invention as those of "Formula I," "Formula Ia," "Formula Ib," or any other generic structural formulas depicted herein, is intended to encompass compounds falling within the scope of each of these structural formulas including pharmaceutically acceptable salts, esters and solvates thereof where such salts, esters and solvates are possible. The term "pharmaceutically acceptable salts" refers to salts prepared from pharmaceutically acceptable non-toxic

bases or acids including inorganic or organic bases and inorganic or organic acids. Salts derived from inorganic bases include aluminum, ammonium, calcium, copper, ferric, ferrous, lithium, magnesium, manganic salts, manganous, potassium, sodium, zinc, and the like. Particularly preferred are the ammonium, calcium, lithium, magnesium, potassium, and sodium salts. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines, and basic ion exchange resins, such as arginine, betaine, caffeine, choline, N,N'-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylaminoethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, hydrabamine, isopropylamine, lysine, methylglucamine, morpholine, piperazine, piperidine, polyamine resins, procaine, purines, theobromine, triethylamine, trimethylamine, tripropylamine, tromethamine, and the like. When the compound of the present invention is basic, salts may be prepared from pharmaceutically acceptable non-toxic acids, including inorganic and organic acids. Such acids include acetic, benzenesulfonic, benzoic, camphorsulfonic, citric, ethanesulfonic, formic, fumaric, gluconic, glutamic, hydrobromic, hydrochloric, isethionic, lactic, maleic, malic, mandelic, methanesulfonic, malonic, mucic, nitric, pamoic, pantothenic, phosphoric, propionic, succinic, sulfuric, tartaric, p-toluenesulfonic acid, trifluoroacetic acid, and the like; and particularly citric, fumaric, hydrobromic, hydrochloric, maleic, phosphoric, sulfuric, and tartaric acids.

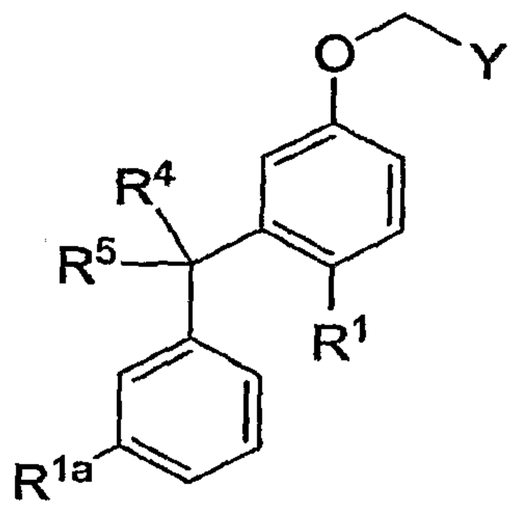
Also, in the case of a carboxylic acid (-COOH) or alcohol group being present in the compounds of the present invention, pharmaceutically acceptable esters of carboxylic acid derivatives, such as methyl, ethyl, or pivaloyloxymethyl, or acyl derivatives of alcohols, such as *O*-acetyl, *O*-pivaloyl, *O*-benzoyl, and *O*-aminoacyl can be employed. Included are those esters and acyl groups known in the art for modifying the solubility or hydrolysis characteristics for use as sustained-release or prodrug formulations.

The compounds of Formula I may contain one or more asymmetric centers, and can thus occur as racemates, racemic mixtures, single enantiomers, diastereoisomeric mixtures and individual diastereoisomers. The present invention in all its embodiments includes all such isomers, as well as salts, esters and solvates of such racemates, mixtures, enantiomers and diastereoisomers. Furthermore, some of the crystalline forms of compounds of the present invention may exist as polymorphs and as such are intended to be included in the present invention. In addition, some of the compounds of the instant invention may form solvates with water or common organic solvents. Such solvates and hydrates are likewise encompassed within the scope of this invention. Some of the compounds described herein contain olefinic double bonds. The invention includes both E and Z geometric isomers. Some of the compounds decribed herein may exist as tautomers, e.g., keto-enol tautomers. Individual tautomers as well as mixtures thereof are included in the present invention.

Compounds of structural Formula I may be separated into their individual diastereoisomers by, e.g., fractional crystallization from suitable solvents, e.g., methylene

chloride/hexanes or ethyl acetate/hexanes, or via chiral chromatography using an optically active stationary phase. Absolute stereochemistry may be determined by X-ray crystallography of crytalline products or crystalline intermediates which are derivatized, if necessary, with a reagent containing a stereogenic center of known configuration. Alternatively, any stereoisomer of a compound of the general Formula I may be obtained by stereospecific synthesis using optically pure starting materials or reagents of known absolute configuration.

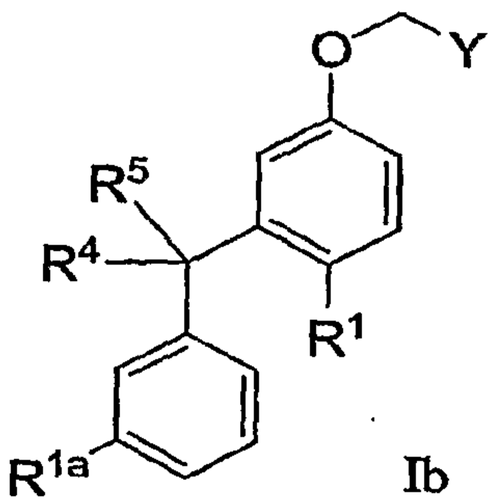
In an embodiment of this invention are compounds within the scope of Formula I having structural Formula Ia:



Ia

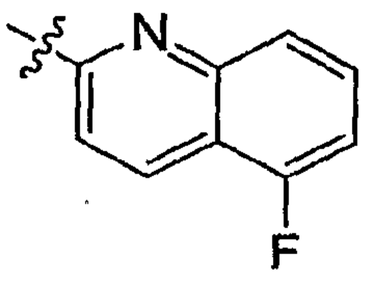
and pharmaceutically acceptable salts and solvates thereof wherein R^1, R^4, R^5, R^{1a}, and Y are as defined in Formula I.

In another embodiment of this invention are compounds within the scope of Formula I and Formula Ia, having structural Formula Ib:

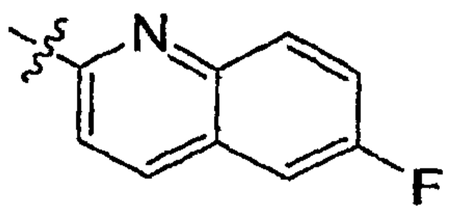


Ib

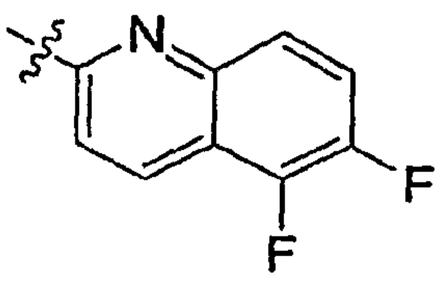
and the pharmaceutically acceptable salts and solvates thereof wherein R^1, R^4, R^5, and R^{1a} are as defined in Formula I and Y is selected from group consisting of:



,

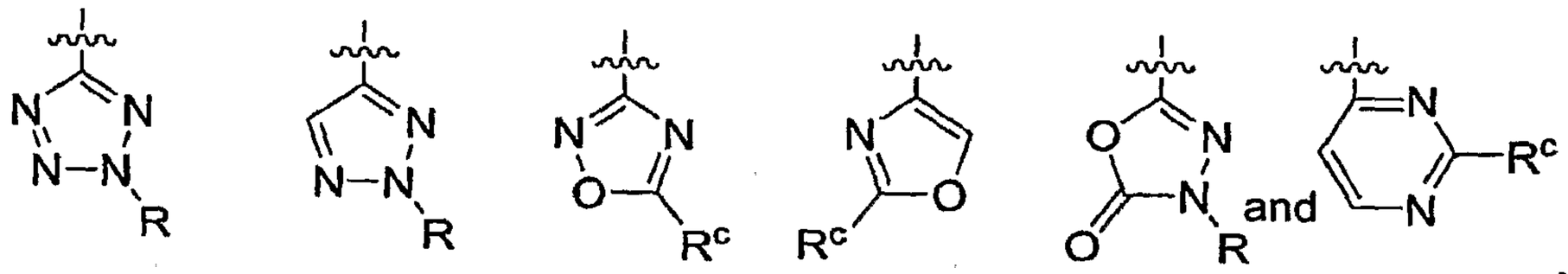


and

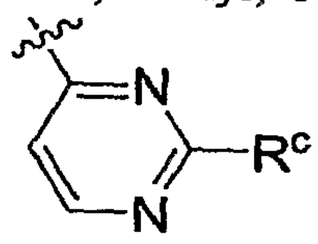
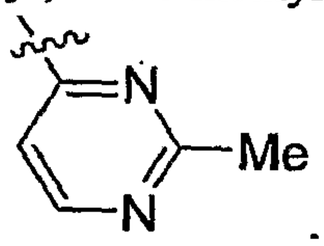


In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^{1a} is as defined above in Formula I. In a class of this embodiment, each R^{1a} is independently selected from –H and –F. In a subclass of this class, R^{1a} is –H.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^1 is as defined in Formula I. In a class of this embodiment, R^1 is selected from –COOH, -$COOR^a$, –C(O)-NR^aR^b, -OC(O)-NR^aR^b, -$CH_2C(O)$-NR^aR^b, and Z^1. In a subclass of this class, R^1 is selected from -C(O)-NR^aR^b, -OC(O)-NR^aR^b, particularly -OC(O)-N(H)-pyridin-3-yl and Z^1. In a further subclass, R^1 is selected from



wherein R is selected from –H and -C_{1-4}alkyl optionally substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and particularly R is selected from –H, methyl, ethyl, and -fluoroethyl; and R^c is selected from –H, -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, and -C_{1-4}alkyl optionally substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro, and particularly R^c is selected from –H, methyl, -NH_2, OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

Particularly, R^1 is  and more particularly it is  .

In another embodiment of this invention, R^2 is as defined above in Formula I. In a class of this embodiment, R^2 is –H.

In another embodiment of this invention, R^3 is as defined above in Formula I. In a class of this embodiment, R^3 is –H.

In another embodiment of this invention are compounds of Formula I, Ia and Ib wherein R^4 is hydrogen.

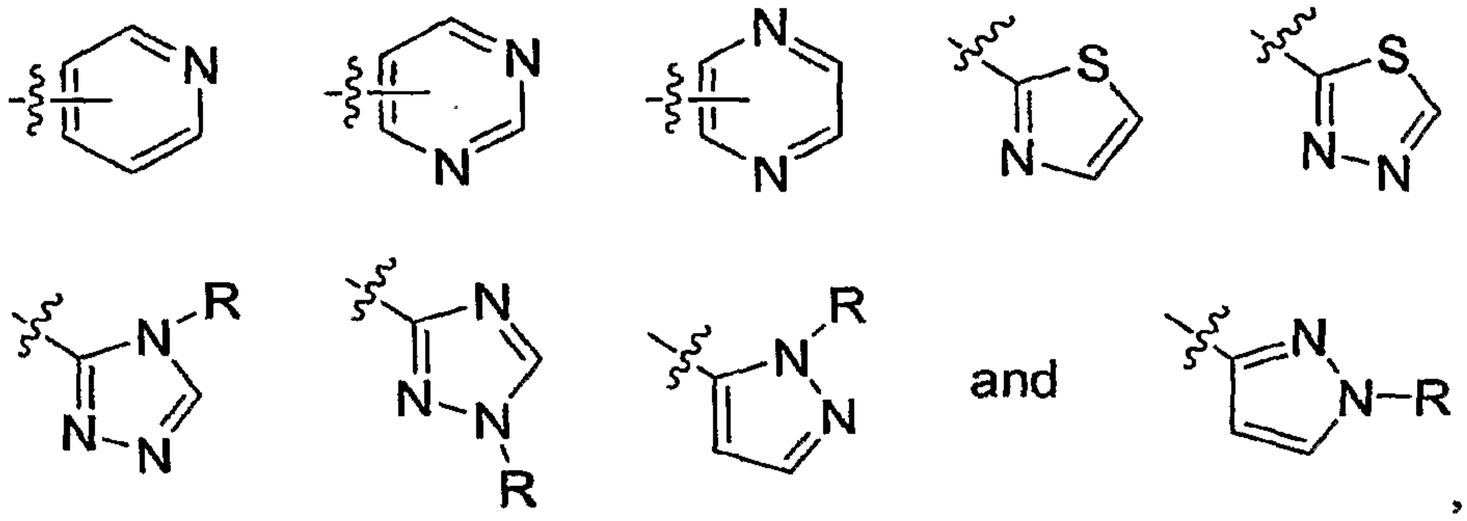
In another embodiment of this invention are compounds of Formula I, Ia and Ib wherein R^5 is C_{1-6} alkyl. In a class of this embodiment, R^5 is *t*-butyl. In a subclass of this class, R^4 is hydrogen.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^6 is as defined above in Formula I. In a class of this embodiment, R^6 is selected from –H, -$CONR^aR^b$, -$OCONR^aR^b$, –CO_2R^a, and Z^1.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^7 is as defined above in Formula I.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein "p" is an integer selected from 0, 1 and 2, and particularly p is 2.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^a is as defined above in Formula I. In a class of this embodiment, R^a is selected from –H and Z^2. In a subclass of this class, R^a is selected from pyridinyl, particularly pyridin-3-yl, pyrimidinyl, pyrazinyl, thiazolyl, thiadiazolyl, triazolyl and pyrazolyl. In a further subclass of this class, R^a is selected from



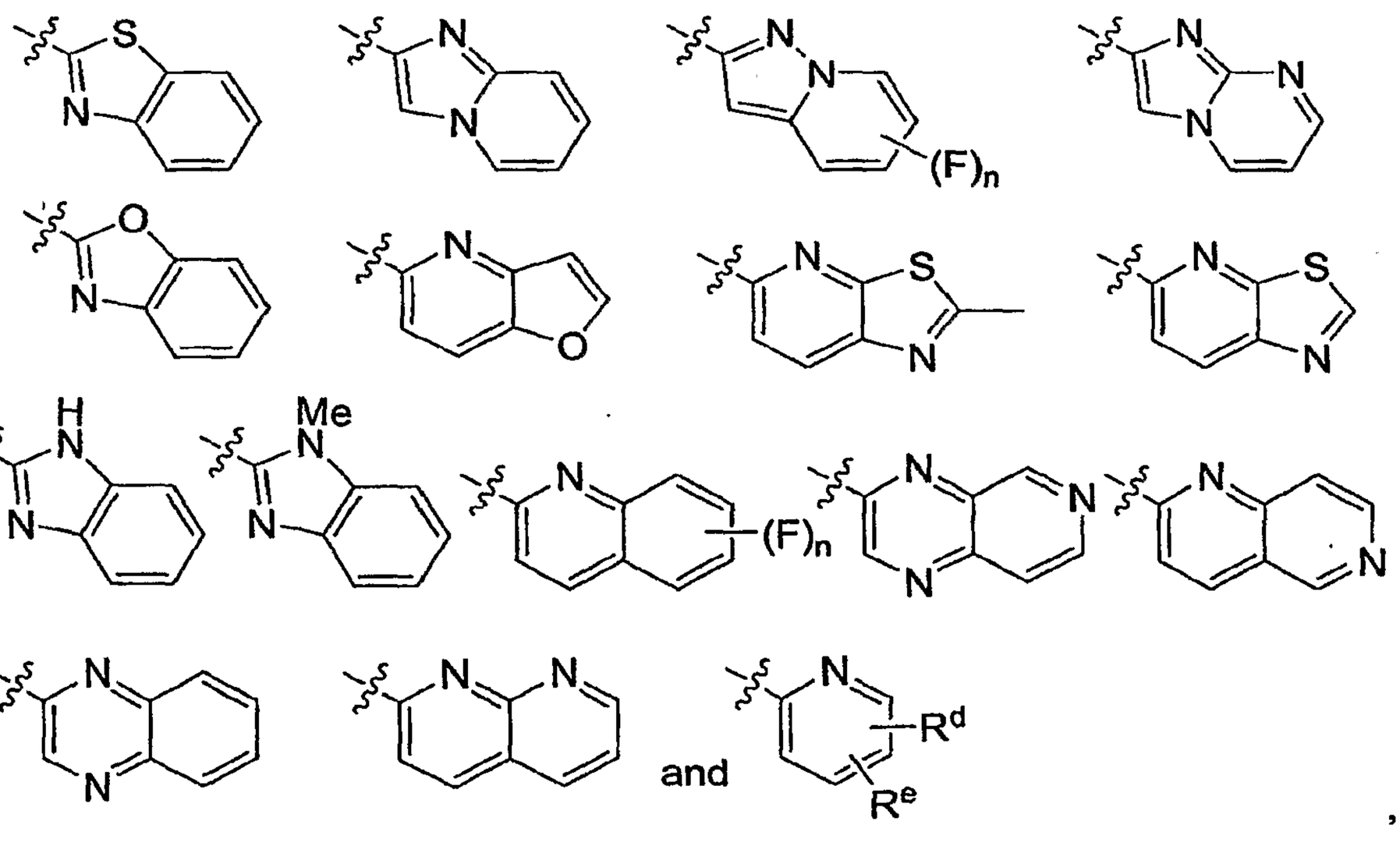
,

wherein R is as defined above.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein R^b is as defined above in Formula I. In a class of this embodiment, R^b is selected from –H, methyl, ethyl, propyl, and isopropyl. In a subclass of this class, R^b is –H or methyl.

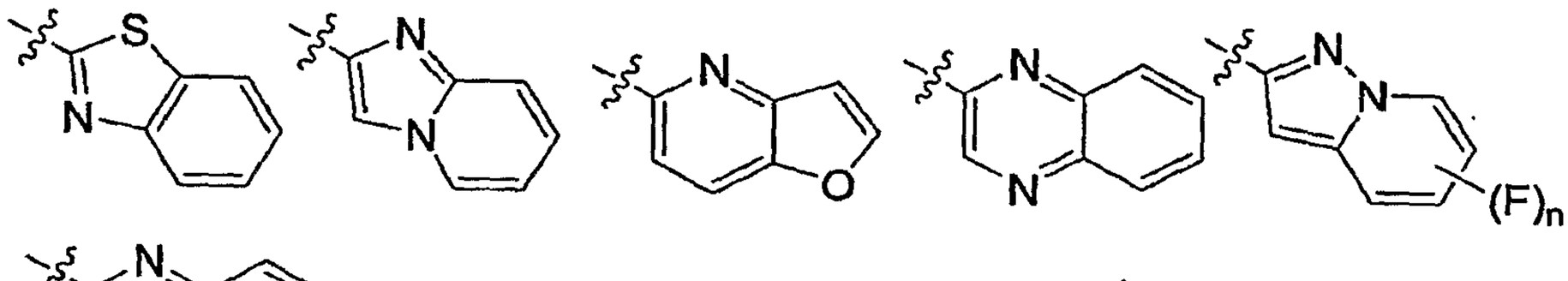
In another embodiment of this invention, X is as defined above in Formula I. In a class of this embodiment, X is –O-.

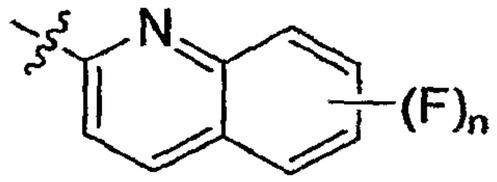
In another embodiment of this invention are compounds of Formula I and Ia wherein Y is as defined in Formula I. In a class of this embodiment, Y is selected from:



,

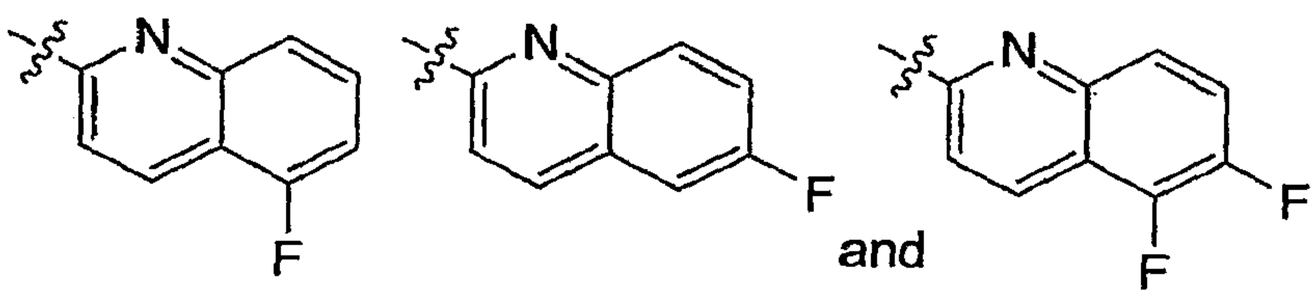
wherein R^d is selected from -C_{1-4}alkyl, -F, -CF_2H, and -CF_3; R^e is –H or –C_{1-4} alkyl; and n is an integer selected from zero, 1, 2 and 3. In a subclass of this class, Y is selected from:





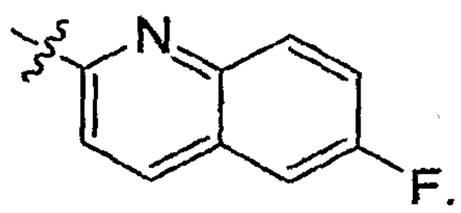
, particularly

n is an integer selected from 1, 2 and 3



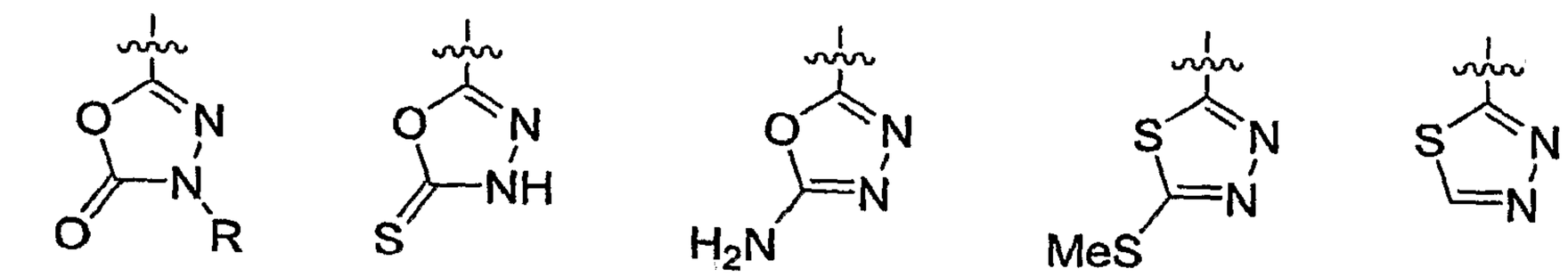
and

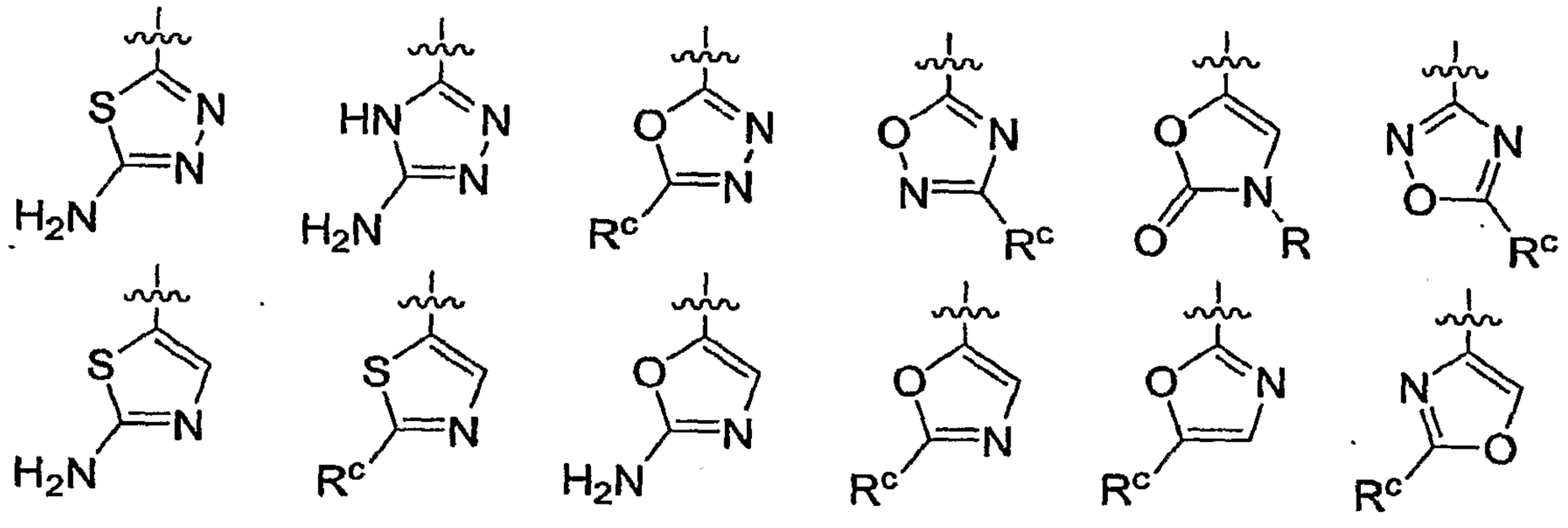
In yet a further subclass of this class, Y is

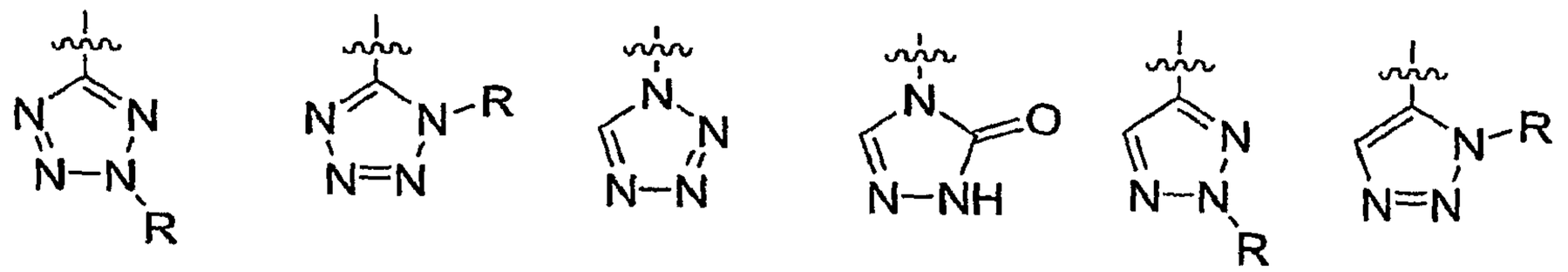


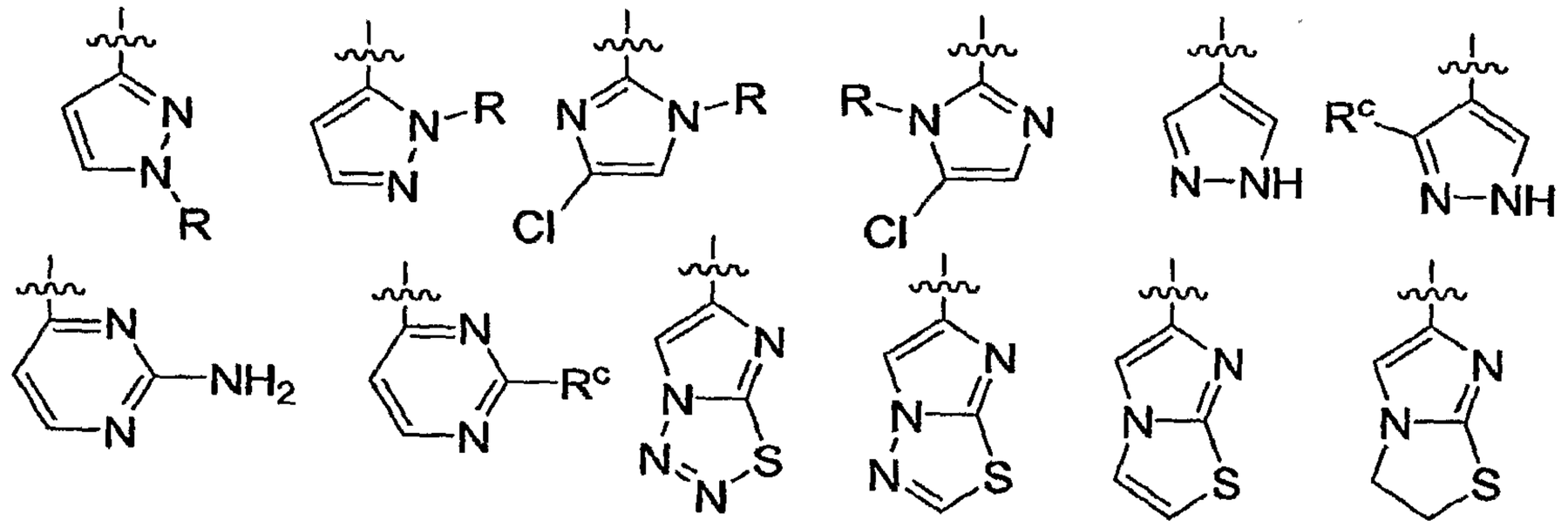
In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein Hetcy is as defined in Formula I. In a class of this embodiment, Hetcy is selected from pyrrolidinyl and piperidinyl, each member being optionally substituted as defined in Formula I.

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein Z^1 is as defined in Formula I. In a class of this embodiment, Z^1 is selected from:



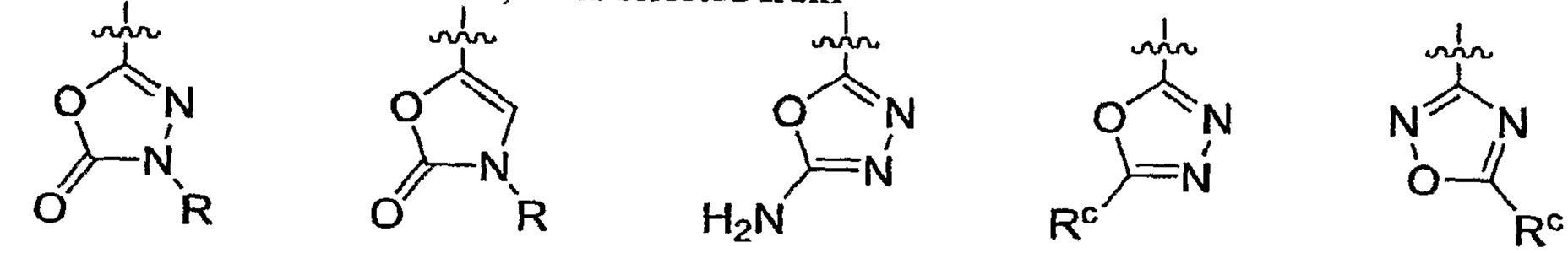


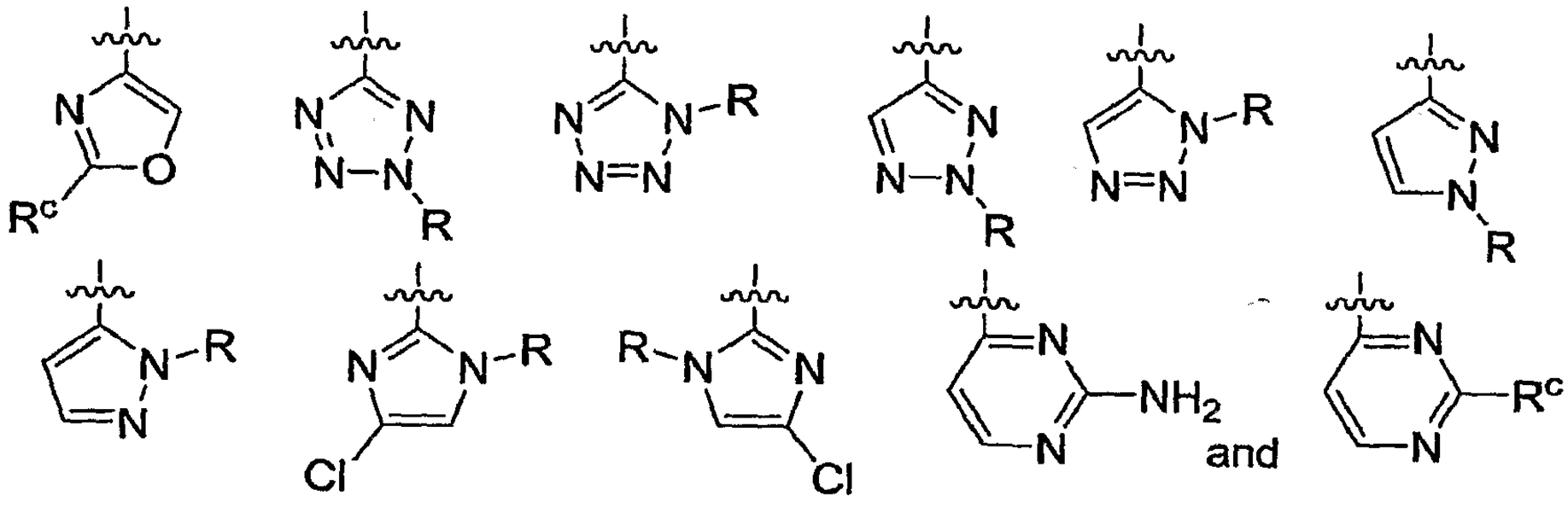




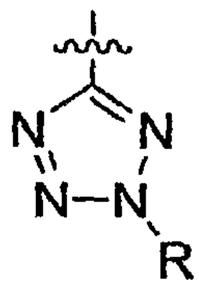
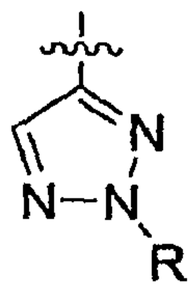
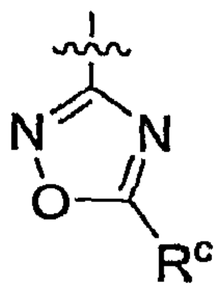
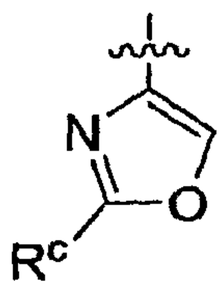
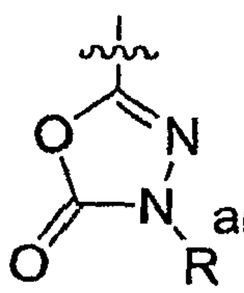
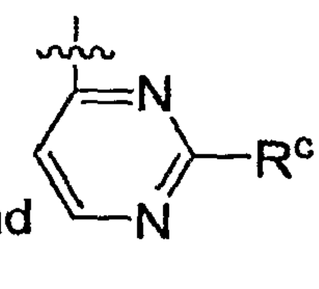
wherein R is selected from –H and -C_{1-4}alkyl optionally substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and particularly R is selected from –H, methyl, ethyl, and -fluoroethyl; and R^c is selected from –H, -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, and -C_{1-4}alkyl optionally substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro, and particularly R^c is selected from –H, methyl, -NH_2, -OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

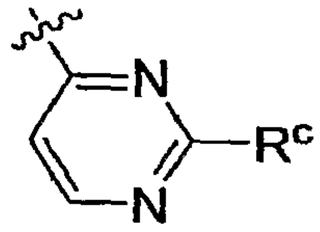
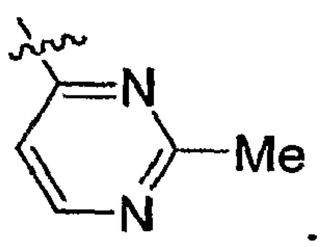
In a class of this embodiment, Z^1 is selected from



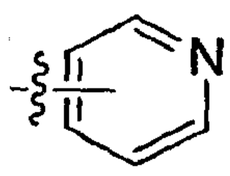
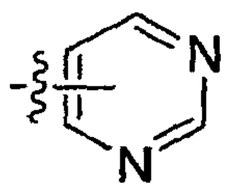
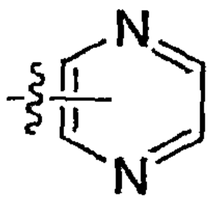
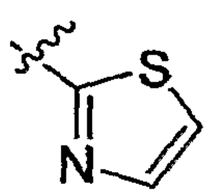
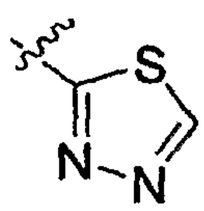
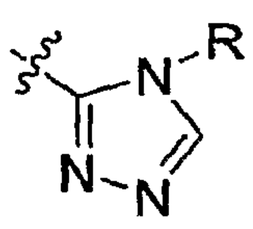
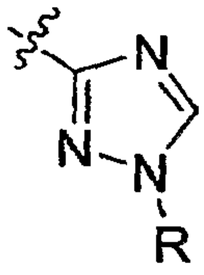
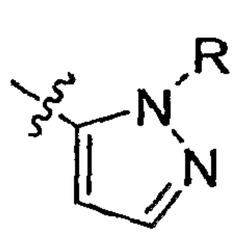
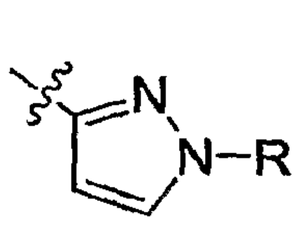


In a subclass of this class, Z^1 is selected from:






and


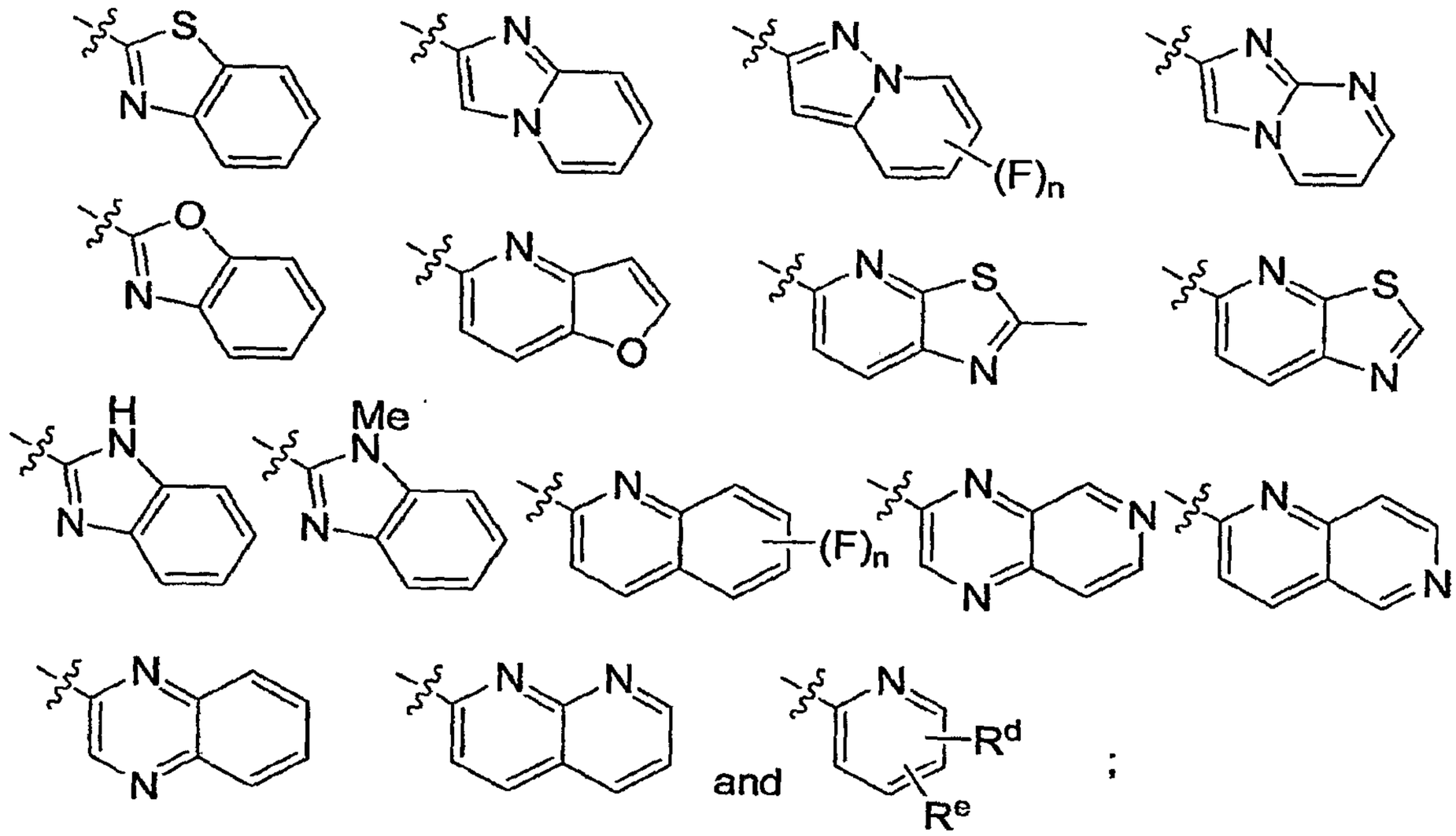
Particularly, Z^1 is  and more particularly it is  .

In another embodiment of this invention are compounds of Formula I, Ia and Ib, wherein Z^2 is as defined in Formula I. In a class of this embodiment, Z^2 is selected from pyridinyl, pyrimidinyl, pyrazinyl, thiazolyl, thiadiazolyl, triazolyl and pyrazolyl, each member being optionally substituted as defined in Formula I. In a subclass of this class, Z^2 is selected from









and


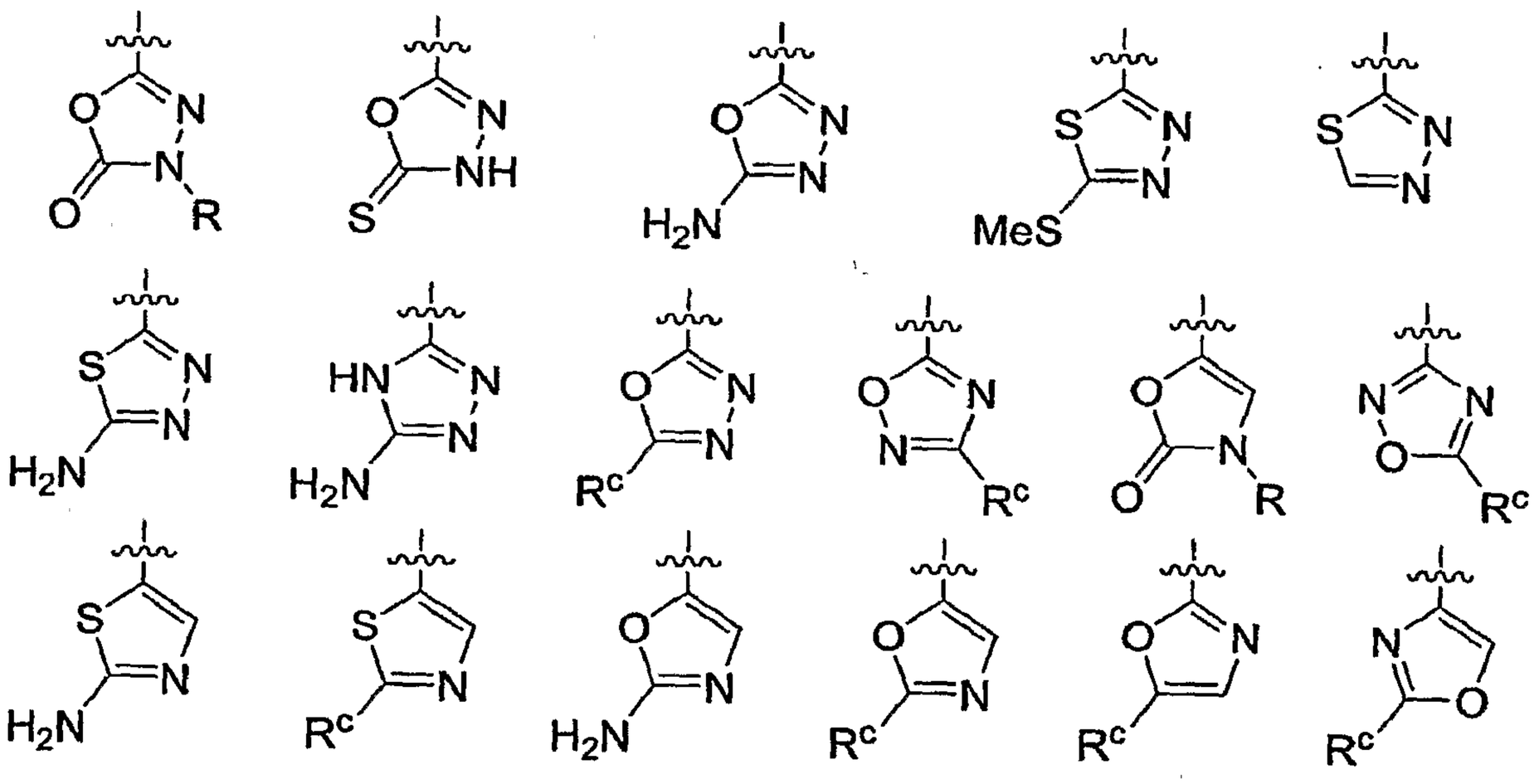
wherein R is as defined above.

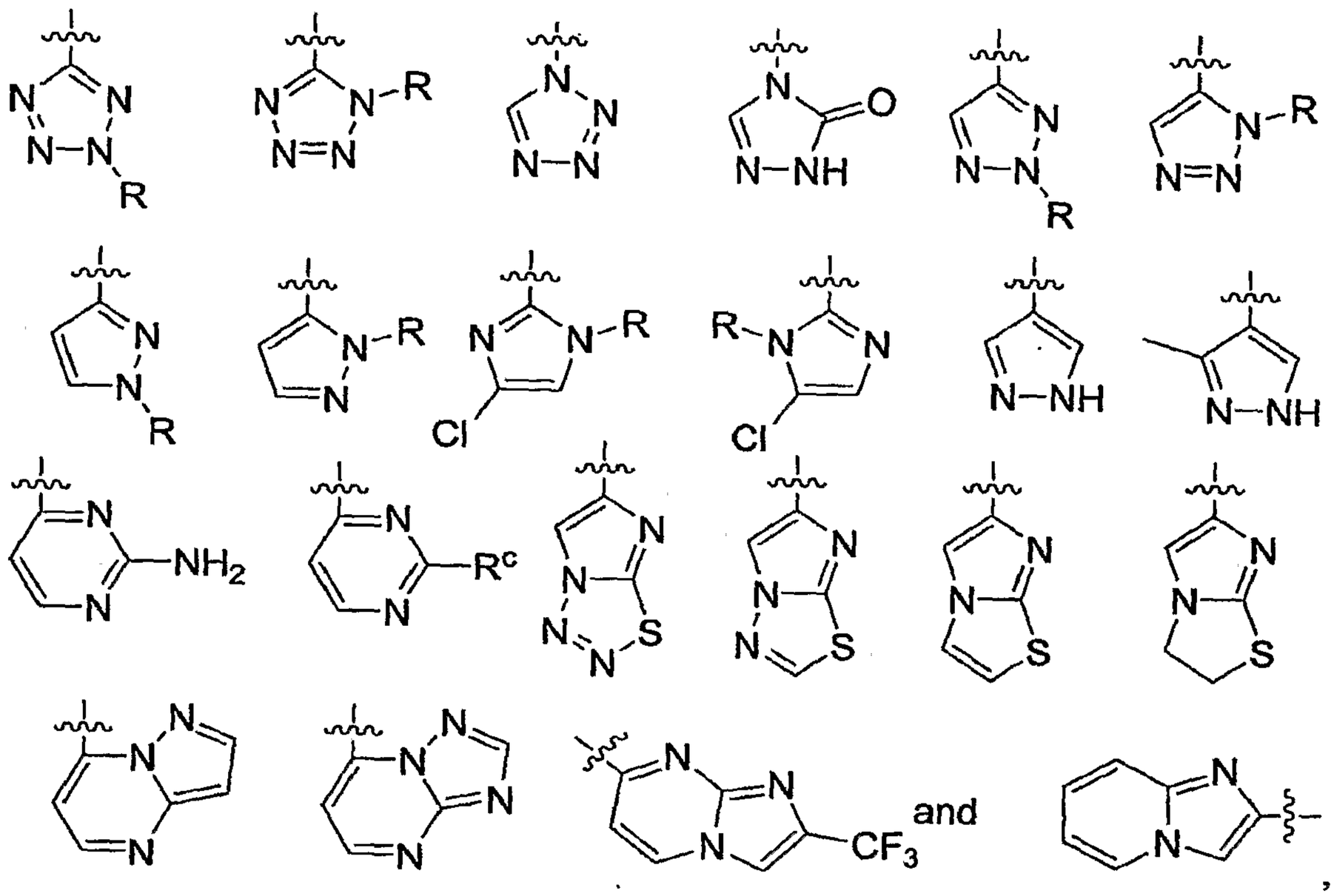
In a particular embodiment of this invention are compounds of Formula I wherein Y is selected from the group consisting of



wherein R^d is selected from $-C_{1-4}$ alkyl, -F, $-CF_2H$, and $-CF_3$; R^e is –H or $-C_{1-4}$ alkyl; and n is an integer selected from zero, 1, 2, and 3;

in a class thereof, R^1 is selected from –COOH, $-COOR^a$, $-C(O)-NR^aR^b$, $-OC(O)-NR^aR^b$, $-CH_2C(O)-NR^aR^b$ and Z^1; in a subclass thereof, X is –O-; in a further subclass thereof , Z^1 is selected from the group consisting of:





in a yet further subclass thereof, R^a is selected from –H and Z^2, and R^b is selected from –H, methyl, ethyl, propyl and isopropyl; in yet a further subclass thereof, Z^2 is selected from pyridinyl, pyrimidinyl, pyrazinyl, thiazolyl, thiadiazolyl, triazolyl and pyrazolyl; in a yet further subclass thereof, R^6 is selected from –H, -$CONR^aR^b$, -$OCONR^aR^b$, –CO_2R^a and Z^1; in a yet further subclass thereof, each R^{1a} is independently selected from –H and –F; in a yet further subclass thereof, R^2 is –H and R^3 is –H; and in a final subclass thereof, Hetcy is selected from pyrrolidinyl and piperidinyl.

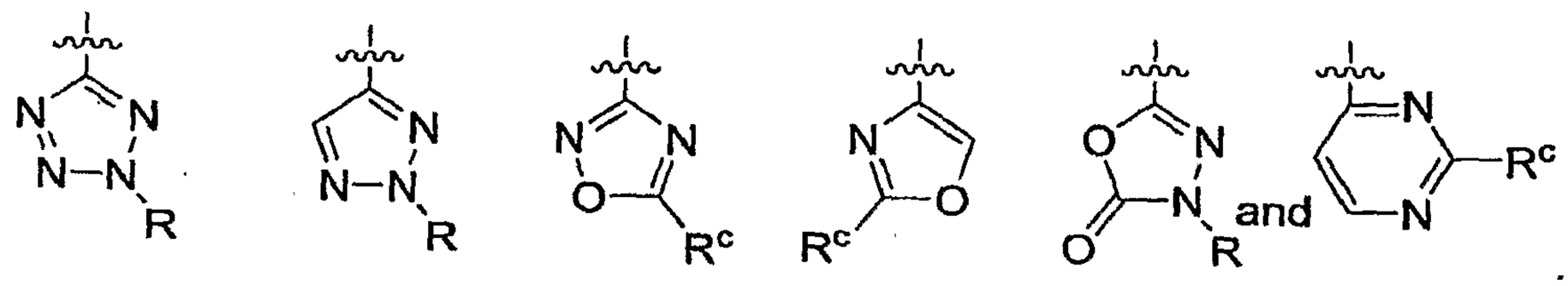
In a more particular embodiment are compounds of Formula Ia and Formula Ib wherein R^{1a} is selected from –H and –F. In a class of this embodiment R^1 is selected from -$OC(O)NR^aR^b$ and Z^1, wherein Z^1 is selected from:

(a) a 5-membered unsaturated heterocyclic ring containing 2-4 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, =O, =S, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN and 1-3 of fluoro,

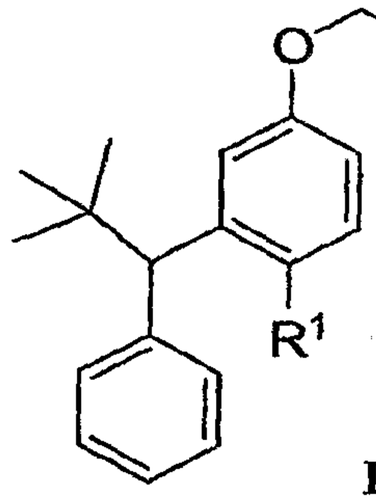
(b) a 5-membered unsaturated heterocyclic ring containing 2-3 heteroatoms selected from one oxygen or one sulfur and 1-2 of nitrogen, wherein one nitrogen in the ring is optionally substituted with a group selected from C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, and C_{1-4}alkyl optionally substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN and 1-3 of fluoro, and

(c) a 6-membered unsaturated heterocyclic ring containing 1-2 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN and 1-3 of fluoro, and one carbon atom in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN and 1-3 of fluoro.

In a subclass of this embodiment, R^1 is selected from:

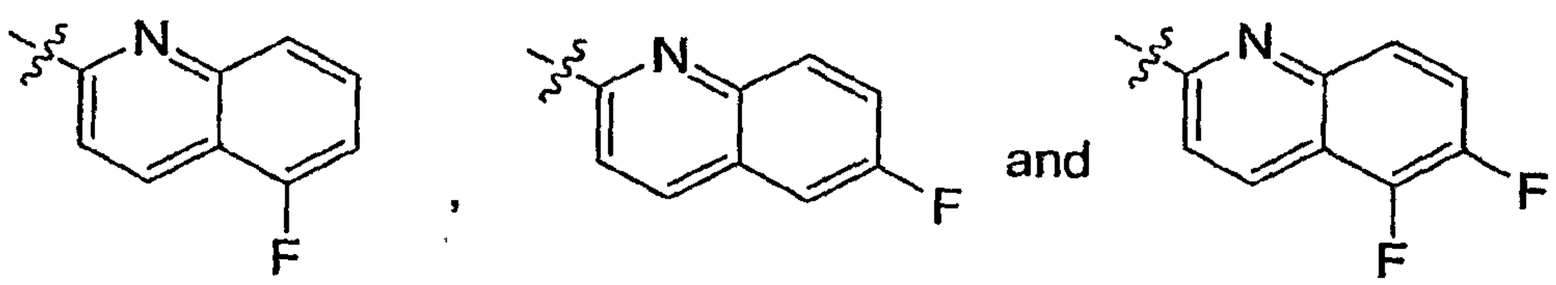


In a separate embodiment of compounds of the present invention are those of structural formula Id:

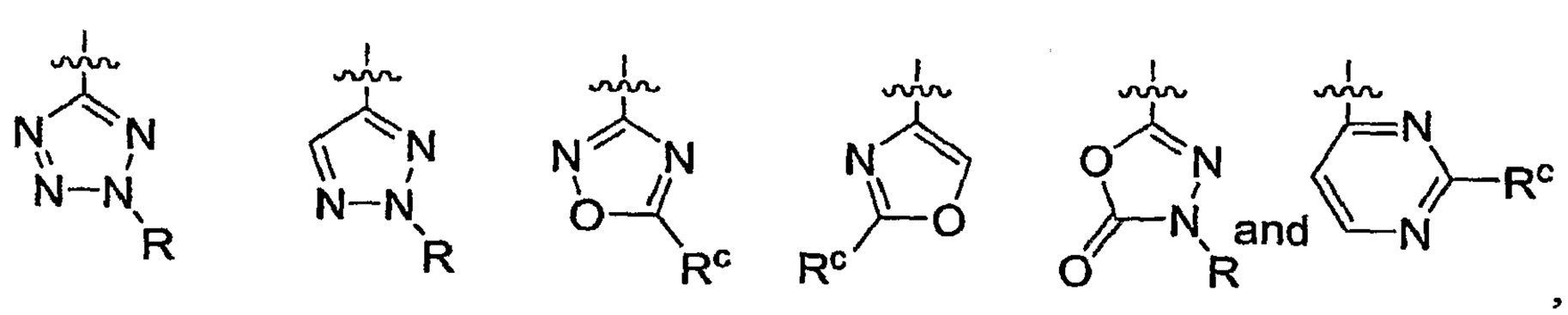


Id

wherein Y is selected from the group consisting of:

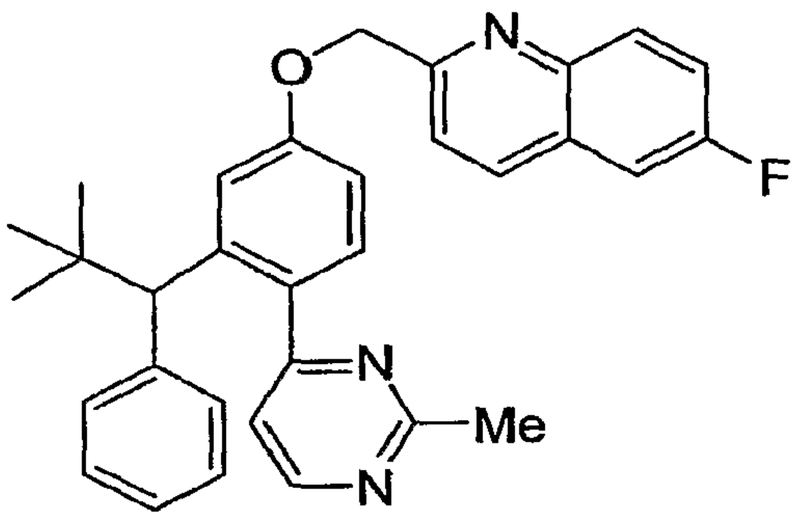


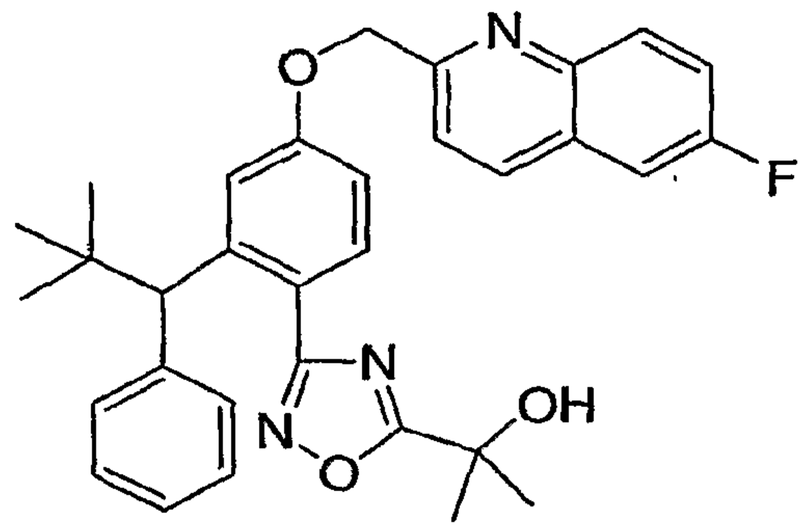
and R^1 is selected from the group consisting of:

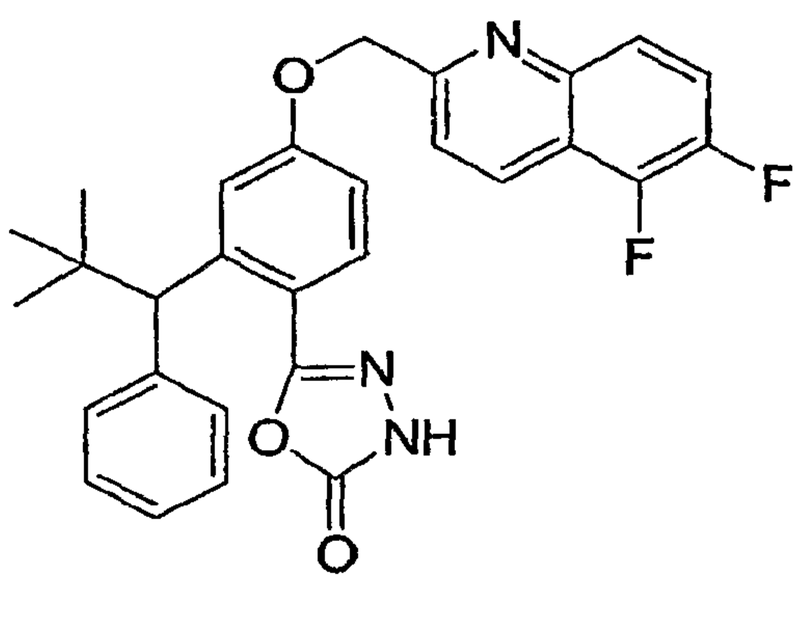


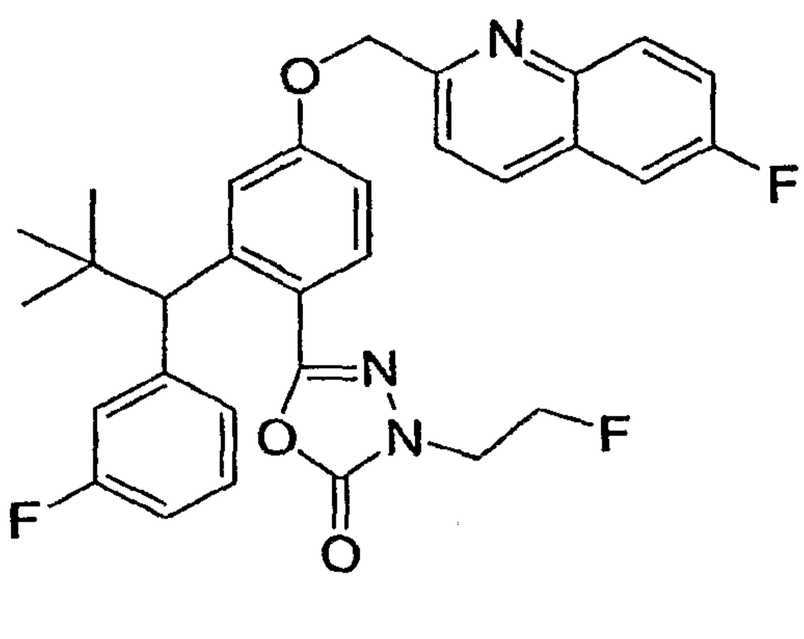
wherein R is selected from –H and -C_{1-4}alkyl optionally substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro; and R^c is selected from –H, methyl, -NH_2, OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

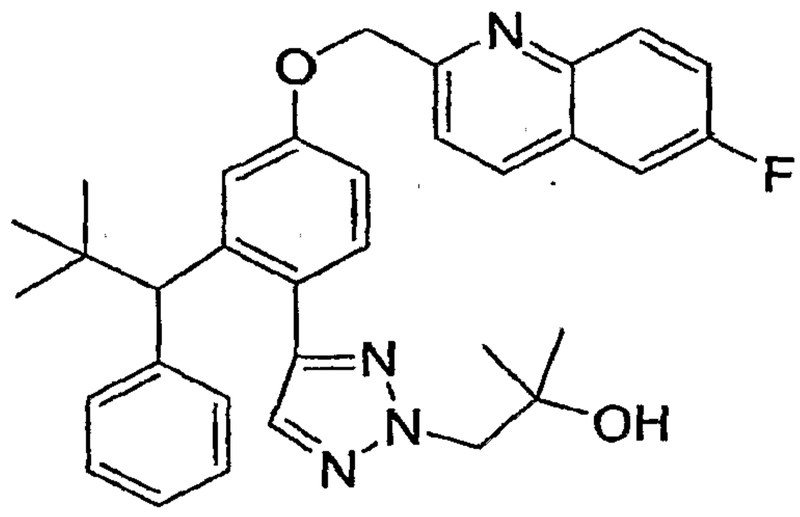
Illustrative, but nonlimiting, examples of compounds of the present invention that are useful as inhibitors of leukotriene biosynthesis are the following:

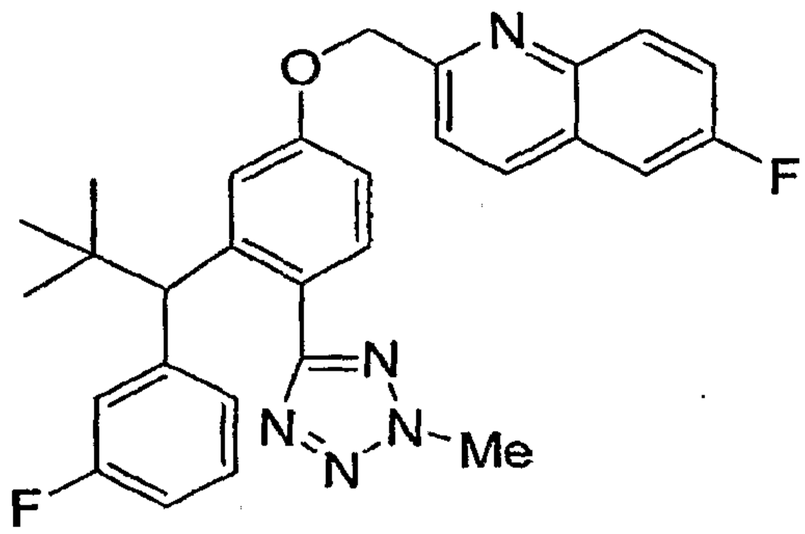


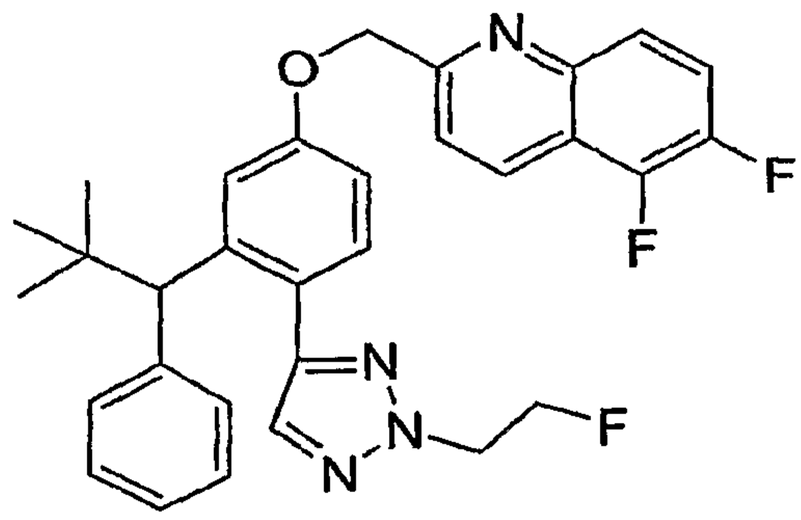


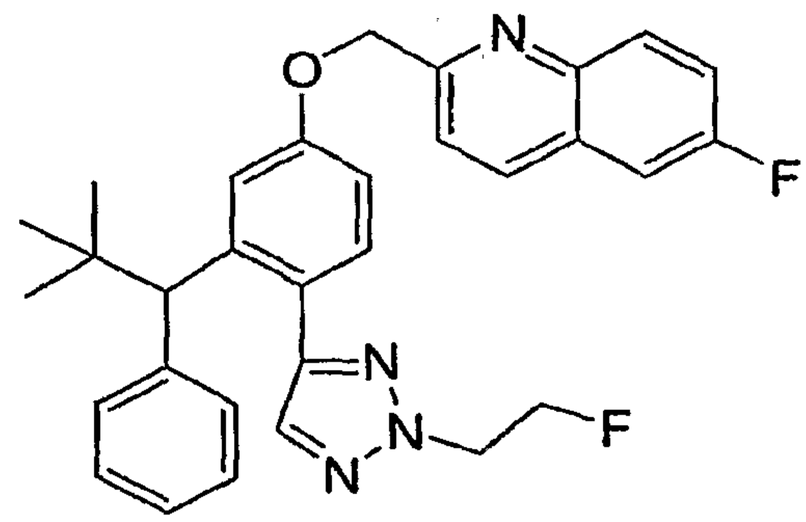


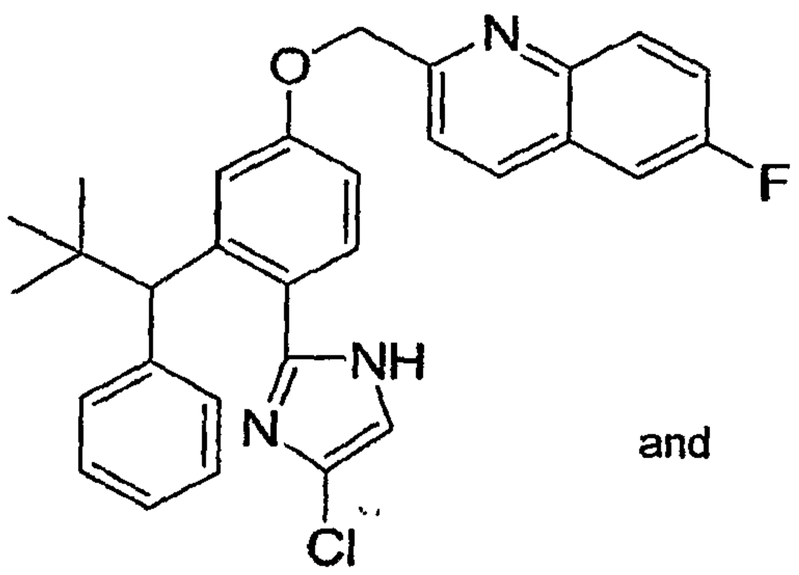




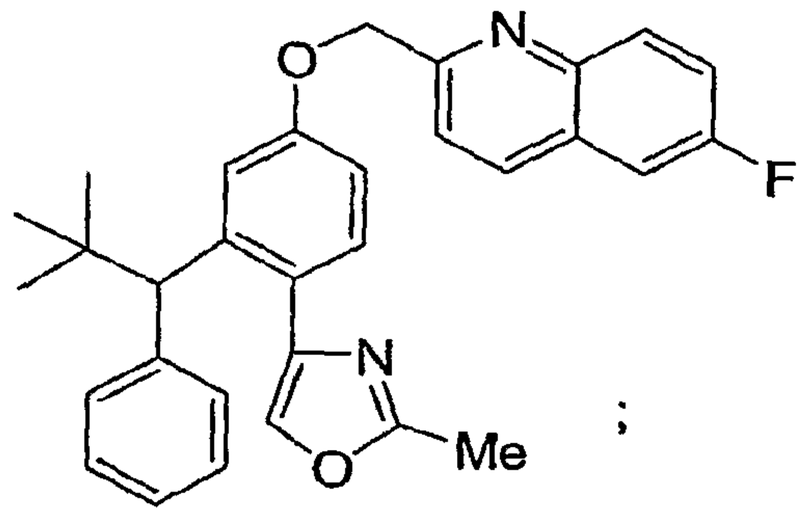








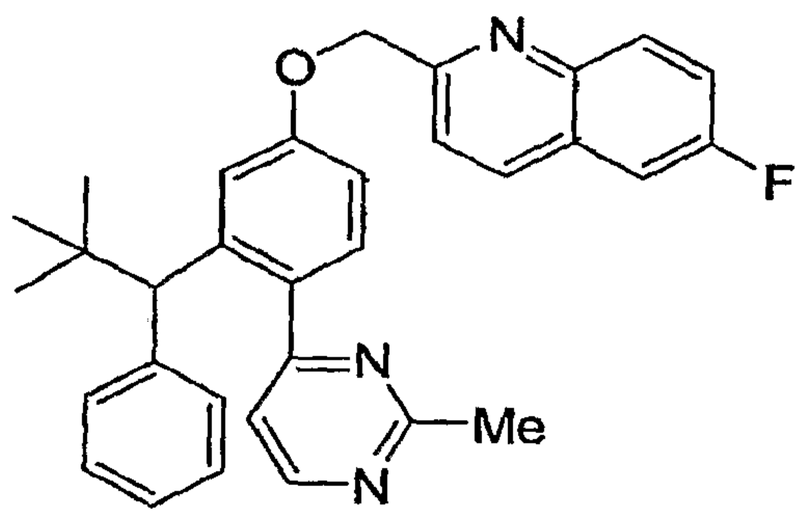
and

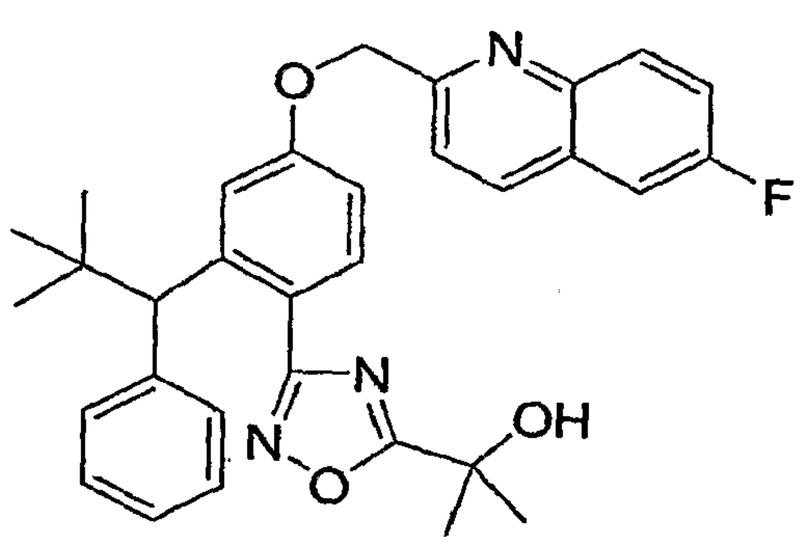


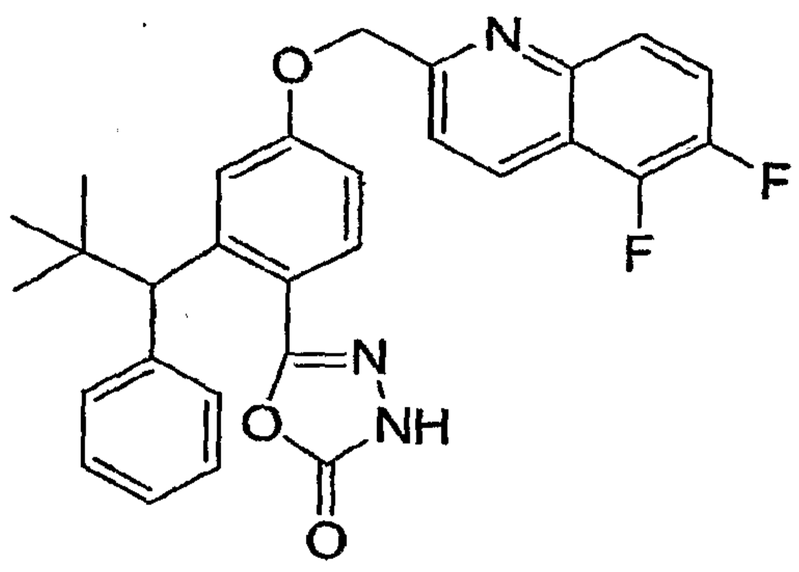
;

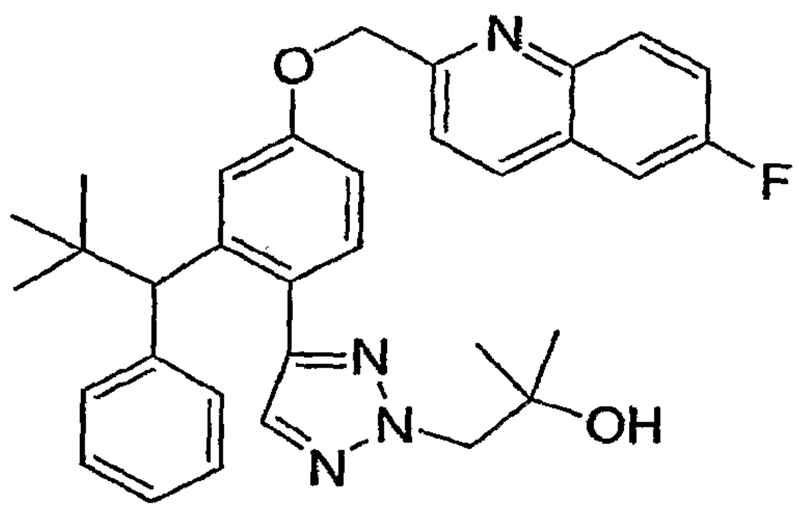
and the pharmaceutically acceptable salts and solvates thereof.

Further illustrative of the compounds of the present invention are those selected from the group consisting of:

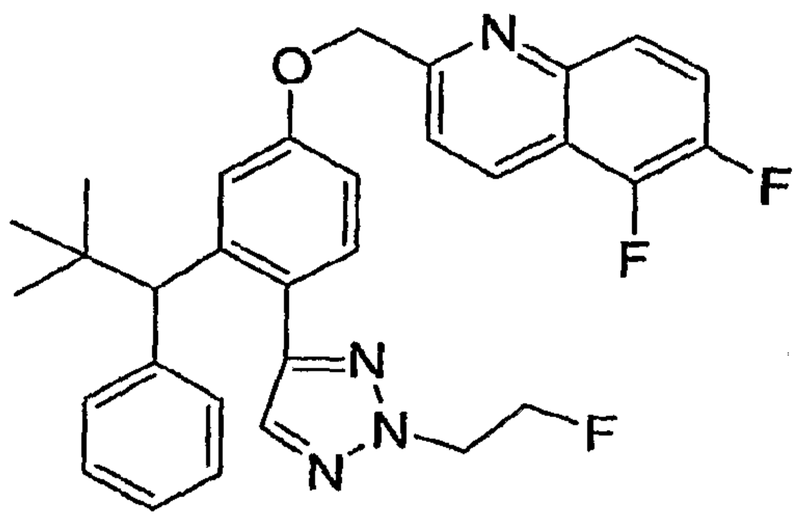








and



and the pharmaceutically acceptable salts and solvates thereof.

Examples of compounds that fall within the present invention include those shown in the examples contained herein, as well as salts and solvates thereof. When racemic mixtures are shown, the specific enantiomers are also included, as are the salts and solvates of the specific enantiomers.

The compounds of Formula I can be used for the treatment of atherosclerosis comprising administering a therapeutically effective amount of a compound of Formula I to a patient in need of such treatment. A further aspect of this invention involves a method for preventing or reducing the risk of developing atherosclerosis, comprising administering a prophylactically effective amount of a compound of Formula I to a patient in need of such treatment. Atherosclerosis is characterized by the deposition of atheromatous plaques containing cholesterol and lipids on the innermost layer of the walls of large and medium-sized arteries. Atherosclerosis encompasses vascular diseases and conditions that are recognized and understood by physicians practicing in the relevant fields of medicine. Atherosclerotic cardiovascular disease including restenosis following revascularization procedures, coronary heart disease (also known as coronary artery disease or ischemic heart disease), cerebrovascular disease including multi-infarct dementia, and peripheral vessel disease including erectile dysfunction, are all clinical manifestations of atherosclerosis and are therefore encompassed by the terms "atherosclerosis" and "atherosclerotic disease."

A FLAP inhibitor may be administered to prevent or reduce the risk of occurrence, or recurrence where the potential exists, of a coronary heart disease event, a cerebrovascular event, and/or intermittent claudication. Coronary heart disease events are intended to include CHD death, myocardial infarction (i.e., a heart attack), and coronary revascularization procedures. Cerebrovascular events are intended to include ischemic or hemorrhagic stroke (also known as cerebrovascular accidents) and transient ischemic attacks. Intermittent claudication is a clinical manifestation of peripheral vessel disease. The term "atherosclerotic disease event" as used herein is intended to encompass coronary heart disease events, cerebrovascular events, and intermittent claudication. It is intended that persons who have previously experienced one or more non-fatal atherosclerotic disease events are those for whom the potential for recurrence of such an event exists.

Accordingly, the instant invention also provides a method for preventing or reducing the risk of a first or subsequent occurrence of an atherosclerotic disease event comprising the administration of a prophylactically effective amount of a FLAP inhibitor to a patient at risk for such an event. The patient may already have atherosclerotic disease at the time of administration, or may be at risk for developing it.

The method of this invention particularly serves to prevent or slow new atherosclerotic lesion or plaque formation, and to prevent or slow progression of existing lesions or plaques, as well as to cause regression of existing lesions or plaques. Accordingly, one aspect of this invention involves a method for halting or slowing the progression of atherosclerosis, including halting or slowing atherosclerotic plaque progression, comprising administering a therapeutically effective amount of a

FLAP inhibitor to a patient in need of such treatment. This method also includes halting or slowing progression of atherosclerotic plaques existing at the time the instant treatment is begun (i.e., "existing atherosclerotic plaques"), as well as halting or slowing formation of new atherosclerotic plaques in patients with atherosclerosis.

Another aspect of this invention involves a method for regression of atherosclerosis, including regression of atherosclerotic plaques existing at the time the instant treatment is begun, comprising administering a therapeutically effective amount of a FLAP inhibitor to a patient in need of such treatment. Another aspect of this invention involves a method for preventing or reducing the risk of atherosclerotic plaque rupture comprising administering a prophylactically effective amount of a FLAP inhibitor to a patient in need of such treatment.

The ability of the compounds of Formula I to inhibit biosynthesis of the leukotrienes makes them useful for preventing or reversing the symptoms induced by the leukotrienes in a human subject. This inhibition of the mammalian biosynthesis of leukotrienes indicates that the compounds and pharmaceutical compositions thereof are useful to treat, prevent, or ameliorate in mammals and especially in humans: 1) pulmonary disorders including diseases such as asthma, chronic bronchitis, and related obstructive airway diseases, 2) allergies and allergic reactions such as allergic rhinitis, contact dermatitis, allergic conjunctivitis, and the like, 3) inflammation such as arthritis or inflammatory bowel disease, 4) pain, 5) skin disorders such as atopic eczema, and the like, 6) cardiovascular disorders such as angina, formation of atherosclerotic plaques, myocardial ischemia, hypertension, platelet aggregation and the like, 7) renal insufficiency arising from ischaemia induced by immunological or chemical (cyclosporin) etiology and 8) migraine or cluster headache, 9) ocular conditions such as uveitis, 10) hepatitis resulting from chemical, immunological or infectious stimuli, 11) trauma or shock states such as burn injuries, endotoxemia and the like, 12) allograft rejection, 13) prevention of side effects associated with therapeutic administration of cytokines such as Interleukin II and tumor necrosis factor, 14) chronic lung diseases such as cystic fibrosis, bronchitis and other small- and large-airway diseases, 15) cholecystitis, 16) multiple sclerosis, and 17) proliferation of myoblastic leukemia cells.

Thus, the compounds of the present invention may also be used to treat or prevent mammalian (especially, human) disease states such as erosive gastritis; erosive esophagitis; diarrhea; cerebral spasm; premature labor; spontaneous abortion; dysmenorrhea; ischemia; noxious agent-induced damage or necrosis of hepatic, pancreatic, renal, or myocardial tissue; liver parenchymal damage caused by hepatoxic agents such as CCl_4 and D-galactosamine; ischemic renal failure; disease-induced hepatic damage; bile salt induced pancreatic or gastric damage; trauma- or stress-induced cell damage; and glycerol-induced renal failure. The compounds also act as inhibitors of tumor metastasis and exhibit cytoprotective action.

The FLAP inhibitors of this invention can also be administered for prevention, amelioration and treatment of glomerulonephritis (see Guasch A., Zayas C.F., Badr KF. (1999), "MK-591 acutely restores glomerular size selectivity and reduces proteinuria in human glomerulonephritis,"

Kidney Int., 56:261-267); and also for and prevention, amelioration and treatment of kidney damage resulting from diabetes complications (see Valdivielso JM, Montero A., Badr KF., Munger KA. (2003), "Inhibition of FLAP decreases proteinuria in diabetic rats," J. Nephrol., 16(1):85-940.)

In addition, the compounds of this invention can also be used for the treatment of chronic obstructive pulmonary disease (COPD). As described in S. Kilfeather, Chest, 2002, vol 121, 197, airway neutrophilia in COPD patients is believed to be a contributing source of inflammation and is associated with airway remodeling. The presence of neutrophils is mediated in part by LTB_4, and treatment with the instant compounds could be used to reduce neutrophilic inflammation in patients with COPD.

The cytoprotective activity of a compound may be observed in both animals and man by noting the increased resistance of the gastrointestinal mucosa to the noxious effects of strong irritants, for example, the ulcerogenic effects of aspirin or indomethacin. In addition to lessening the effect of non-steroidal anti-inflammatory drugs on the gastrointestinal tract, animal studies show that cytoprotective compounds will prevent gastric lesions induced by oral administration of strong acids, strong bases, ethanol, hypertonic saline solutions, and the like. Two assays can be used to measure cytoprotective ability. These assays are: (A) an ethanol-induced lesion assay and (B) an indomethacin-induced ulcer assay and are described in EP 140,684.

In particular, the compounds of the invention would be useful to reduce the gastric erosion caused by co-administration of a cyclooxygenase-2 selective inhibitor and low-dose aspirin. Cyclooxygenase-2 selective inhibitors are widely used as effective anti-inflammatory drugs with less potential for gastrointestinal complications as compared to traditional, non-selective non-steroidal anti-inflammatory drugs. However, the combined use of a cyclooxygenase-2 selective inhibitor with low-dose aspirin for cardio protection may compromise the gastrointestinal safety of this class of compounds. By virtue of its activity as a 5-lipoxygenase inhibitor, the compounds of the invention would be expected to be gastric protective in this regard. See Fiorucci, et al. FASEB J. 17:1171-1173, 2003.

Cyclooxygenase-2 selective inhibitors for use with the invention include but are not limited to etoricoxib (ARCOXIA™), celecoxib (CELEBREX®) and valdecoxib (BEXTRA™). A compound of this invention in combination with a cyclooxygenase-2 selective inhibitor could be administered in unit dosage form or separately to a patient on low-dose aspirin therapy. Alternatively, the cyclooxygenase-2 inhibitor could be administered in unit dosage form with low-dose aspirin, in which case a compound of this invention would be administered separately. All three active ingredients in unit dosage form is also encompassed. Conventional dosage amounts of the cyclooxygenase-2 selective inhibitor and aspirin (for cardio protection) may be utilized. For example, aspirin could be administered at 81 mg once daily.

In general, FLAP inhibitors can be identified as those compounds which have an IC_{50} in the "FLAP Binding Assay" that is less than or equal to 1 μM, and preferably 500 nM or less.

The term "patient" includes mammals, especially humans, who use the instant active agents for the prevention or treatment of a medical condition. Administering of the drug to the patient includes both self-administration and administration to the patient by another person. The patient may be

in need of treatment for an existing disease or medical condition, or may desire prophylactic treatment to prevent or reduce the risk of onset of atherosclerosis.

The term "therapeutically effective amount" is intended to mean that amount of a drug or pharmaceutical agent that will elicit the biological or medical response of a tissue, a system, animal or human that is being sought by a researcher, veterinarian, medical doctor or other clinician. The term "prophylactically effective amount" is intended to mean that amount of a pharmaceutical drug that will prevent or reduce the risk of occurrence of the biological or medical event that is sought to be prevented in a tissue, a system, animal or human by a researcher, veterinarian, medical doctor or other clinician.

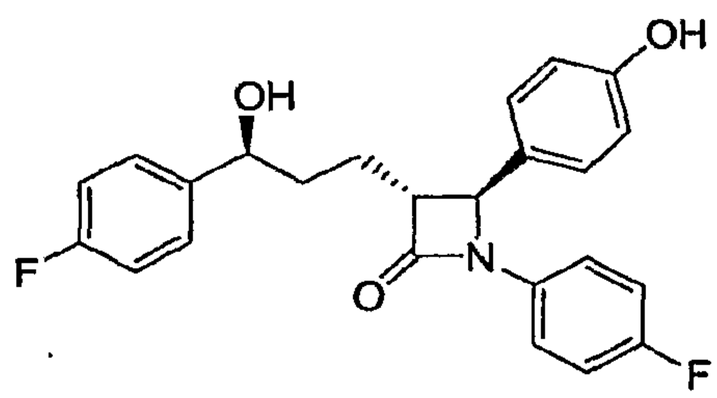
An effective amount of a FLAP inhibitor in the method of this invention is in the range of about 0.001 mg/kg to about 100 mg/kg of body weight per day, preferably 0.01 mg to about 10 mg per kg, and most preferably 0.1 to 1 mg per kg, in single or divided doses. A single daily dose is preferred but not necessary. On the other hand, it may be necessary to use dosages outside these limits in some cases. As examples, the daily dosage amount may be selected from, but not limited to 25 mg, 50 mg, 75 mg, 100 mg, 125 mg, 150 mg, 200 mg and 250 mg. It will be understood, however, that the specific dose level for any particular patient will depend upon a variety of factors including the age, body weight, general health, sex, diet, time of administration, route of administration, rate of excretion, drug combination and the severity of the patient's condition. A consideration of these factors is well within the purview of the ordinarily skilled clinician for the purpose of determining the therapeutically effective or prophylactically effective dosage amount needed to prevent, counter, or arrest the progress of the condition. It is expected that the FLAP inhibitor will administered chronically on a daily basis for a length of time appropriate to treat or prevent the medical condition relevant to the patient, including a course of therapy lasting months, years or the life of the patient.

In a broad embodiment, any suitable additional active agent or agents, including but not limited to anti-atherosclerotic agents, may be used in combination with the compound of Formula I in a single dosage formulation, or may be administered to the patient in a separate dosage formulation, which allows for concurrent or sequential administration of the active agents. One or more additional active agents may be administered with a compound of Formula I. The additional active agent or agents can be lipid modifying compounds or agents having other pharmaceutical activities, or agents that have both lipid-modifying effects and other pharmaceutical activities. Examples of additional active agents which may be employed include but are not limited to HMG-CoA reductase inhibitors, which include statins in their lactonized or dihydroxy open acid forms and pharmaceutically acceptable salts and esters thereof, including but not limited to lovastatin (see US Patent No. 4,342,767), simvastatin (see US Patent No. 4,444,784), dihydroxy open-acid simvastatin, particularly the ammonium or calcium salts thereof, pravastatin, particularly the sodium salt thereof (see US Patent No. 4,346,227), fluvastatin particularly the sodium salt thereof (see US Patent No. 5,354,772), atorvastatin, particularly the calcium salt thereof (see US Patent No. 5,273,995), pitavastatin also referred to as NK-104 (see PCT international publication number WO 97/23200) and rosuvastatin, also known as ZD-4522, (CRESTOR®; see US Patent No.

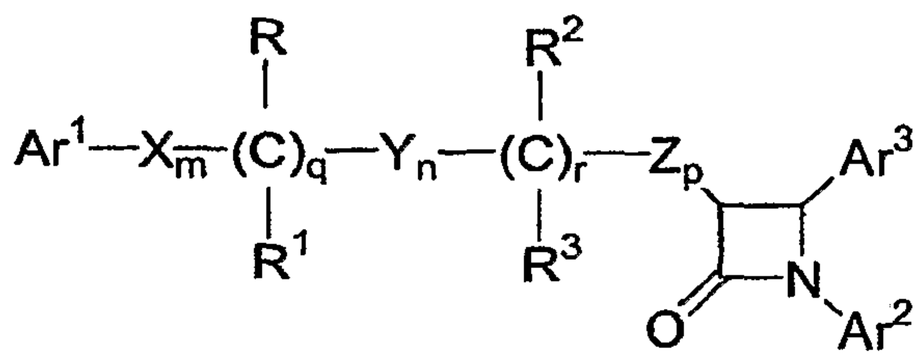
5,260,440, and Drugs of the Future, 1999, 24(5), pp. 511-513); 5-lipoxygenase inhibitors; cholesterol ester transfer protein (CETP) inhibitors, for example JTT-705 and torcetrapib, alsoknown as CP529,414; HMG-CoA synthase inhibitors; squalene epoxidase inhibitors; squalene synthetase inhibitors (also known as squalene synthase inhibitors), acyl-coenzyme A: cholesterol acyltransferase (ACAT) inhibitors including selective inhibitors of ACAT-1 or ACAT-2 as well as dual inhibitors of ACAT-1 and -2; microsomal triglyceride transfer protein (MTP) inhibitors; niacin; bile acid sequestrants; LDL (low density lipoprotein) receptor inducers; platelet aggregation inhibitors, for example glycoprotein IIb/IIIa fibrinogen receptor antagonists and aspirin; human peroxisome proliferator activated receptor gamma (PPARγ) agonists including the compounds commonly referred to as glitazones for example pioglitazone and rosiglitazone and, including those compounds included within the structural class known as thiazolidinediones as well as those PPARγ agonists outside the thiazolidine dione structural class; PPARα agonists such as clofibrate, fenofibrate including micronized fenofibrate, and gemfibrozil; PPAR dual α/γ agonists; vitamin B_6 (also known as pyridoxine) and the pharmaceutically acceptable salts thereof such as the HCl salt; vitamin B_{12} (also known as cyanocobalamin); folic acid or a pharmaceutically acceptable salt or ester thereof such as the sodium salt and the methylglucamine salt; anti-oxidant vitamins such as vitamin C and E and beta carotene; beta-blockers; angiotensin II antagonists such as losartan; angiotensin converting enzyme inhibitors such as enalapril and captopril; calcium channel blockers such as nifedipine and diltiazam; endothelian antagonists; agents that enhance ABCA1 gene expression; FXR and LXR ligands including both inhibitors and agonists; bisphosphonate compounds such as alendronate sodium; and cyclooxygenase-2 inhibitors such as celecoxib.

Still another type of agent that can be used in combination with the compounds of this invention are cholesterol absorption inhibitors. Cholesterol absorption inhibitors block the movement of cholesterol from the intestinal lumen into enterocytes of the small intestinal wall. This blockade is their primary mode of action in reducing serum cholesterol levels. These compounds are distinct from compounds which reduce serum cholesterol levels primarily by mechanisms of action such as acyl coenzyme A - cholesterol acyl transferase (ACAT) inhibition, inhibition of triglyceride synthesis, MTP inhibition, bile acid sequestration, and transcription modulation such as agonists or antagonists of nuclear hormones. Cholesterol absorption inhibitors are described in U.S. Patent 5,846,966, U.S. Patent 5,631,365, U.S. Patent 5,767,115, U.S. Patent 6,133,001, U.S. Patent 5,886,171, U.S. Patent 5,856,473, U.S. Patent 5,756,470, U.S. Patent 5,739,321, U.S. Patent 5,919,672, WO 00/63703, WO /0060107, WO 00/38725, WO 00/34240, WO 00/20623, WO 97/45406, WO 97/16424, WO 97/16455, and WO 95/08532.

An exemplary cholesterol absorption inhibitor is ezetimibe, also known as SCH-58235, which is 1-(4-fluorophenyl)-3(R)-[3(S)-(4-fluorophenyl)-3-hydroxypropyl)]-4(S)-(4-hydroxyphenyl)-2-azetidinone, described in U.S. Patent Nos. 5,767,115 and 5,846,966 and shown below as



Additional exemplary hydroxy-substituted azetidinone cholesterol absorption inhibitors are specifically described in U.S. Patent 5,767,115, column 39, lines 54-61 and column 40, lines 1-51, represented by the formula



as defined in column 2, lines 20-63. These and other cholesterol absorption inhibitors can be identified according to the assay of hypolipidemic compounds using the hyperlipidemic hamster described in U.S. Patent 5,767,115, column 19, lines 47-65, in which hamsters are fed a controlled cholesterol diet and dosed with test compounds for seven days. Plasma lipid analysis is conducted and data is reported as percent reduction of lipid versus control.

Therapeutically effective amounts of cholesterol absorption inhibitors include dosages of from about 0.01 mg/kg to about 30 mg/kg of body weight per day, preferably about 0.1 mg/kg to about 15 mg/kg. For an average body weight of 70 kg, the dosage level is therefore from about 0.7 mg to about 2100 mg of drug per day, e.g. 10, 20, 40, 100 or 200 mg per day, preferably given as a single daily dose or in divided doses two to six times a day, or in sustained release form. This dosage regimen may be adjusted to provide the optimal therapeutic response when the cholesterol absorption inhibitor is used in combination with a compound of the instant invention.

In the method of treatment of this invention, the FLAP inhibitors may be administered via any suitable route of administration such as orally, parenterally, or rectally in dosage unit formulations containing conventional non-toxic pharmaceutically acceptable carriers, adjuvants and vehicles. The term parenteral as used herein includes subcutaneous injections, intravenous, intramuscular, intrasternal injection or infusion techniques. Oral formulations are preferred.

For oral use, the pharmaceutical compositions of this invention containing the active ingredient may be in forms such as tablets, troches, lozenges, aqueous or oily suspensions, dispersible powders or granules, emulsions, hard or soft capsules, or syrups or elixirs. Compositions intended for oral use may be prepared according to any method known to the art for the manufacture of pharmaceutical compositions and such compositions may contain one or more agents selected from the

group consisting of sweetening agents, flavoring agents, coloring agents and preserving agents in order to provide pharmaceutically elegant and palatable preparations. Tablets contain the active ingredient in admixture with non-toxic pharmaceutically acceptable excipients, which are suitable for the manufacture of tablets. These excipients may be for example, inert diluents, such as calcium carbonate, sodium carbonate, lactose, calcium phosphate or sodium phosphate; granulating and disintegrating agents, for example, corn starch, or alginic acid; binding agents, for example starch, gelatin or acacia, and lubricating agents, for example, magnesium stearate, stearic acid or talc.

Oral immediate-release and time-controlled release dosage forms may be employed, as well as enterically coated oral dosage forms. Tablets may be uncoated or they may be coated by known techniques to delay disintegration and absorption in the gastrointestinal tract and thereby provide a sustained action over a longer period. For example, a time delay material such as glyceryl monostearate or glyceryl distearate may be employed. One example of a time-controlled release device is described in U.S. Patent No. 5,366,738. They may also be coated by the technique described in U.S. Patent Nos. 4,256,108; 4,166,452; and 4,265,874 to form osmotic therapeutic tablets for controlled release.

Formulations for oral use may also be presented as hard gelatin capsules wherein the active ingredient is mixed with an inert solid diluent, for example, calcium carbonate, calcium phosphate or kaolin, or as soft gelatin capsules wherein the active ingredients is mixed with water or miscible solvents such as propylene glycol, PEGs and ethanol, or an oil medium, for example peanut oil, liquid paraffin, or olive oil.

Aqueous suspensions contain the active material in admixture with excipients suitable for the manufacture of aqueous suspensions. Such excipients are suspending agents, for example sodium carboxymethylcellulose, methylcellulose, hydroxy-propylmethycellulose, sodium alginate, polyvinyl-pyrrolidone, gum tragacanth and gum acacia; dispersing or wetting agents may be a naturally-occurring phosphatide, for example lecithin, or condensation products of an alkylene oxide with fatty acids, for example polyoxyethylene stearate, or condensation products of ethylene oxide with long chain aliphatic alcohols, for example heptadecaethyleneoxycetanol, or condensation products of ethylene oxide with partial esters derived from fatty acids and a hexitol such as polyoxyethylene sorbitol monooleate, or condensation products of ethylene oxide with partial esters derived from fatty acids and hexitol anhydrides, for example polyethylene sorbitan monooleate. The aqueous suspensions may also contain one or more preservatives, for example ethyl, or n-propyl, p-hydroxybenzoate, one or more colouring agents, one or more flavouring agents, and one or more sweetening agents, such as sucrose, saccharin or aspartame.

Oily suspensions may be formulated by suspending the active ingredient in a vegetable oil, for example arachis oil, olive oil, sesame oil or coconut oil, or in mineral oil such as liquid paraffin. The oily suspensions may contain a thickening agent, for example beeswax, hard paraffin or cetyl alcohol. Sweetening agents such as those set forth above, and flavoring agents may be added to provide

a palatable oral preparation. These compositions may be preserved by the addition of an anti-oxidant such as ascorbic acid.

Dispersible powders and granules suitable for preparation of an aqueous suspension by the addition of water provide the active ingredient in admixture with a dispersing or wetting agent, suspending agent and one or more preservatives. Suitable dispersing or wetting agents and suspending agents are exemplified by those already mentioned above. Additional excipients, for example sweetening, flavoring and coloring agents, may also be present.

The pharmaceutical compositions of the invention may also be in the form of an oil-in-water emulsions. The oily phase may be a vegetable oil, for example olive oil or arachis oil, or a mineral oil, for example liquid paraffin or mixtures of these. Suitable emulsifying agents may be naturally-occurring phosphatides, for example soy bean, lecithin, and esters or partial esters derived from fatty acids and hexitol anhydrides, for example sorbitan monooleate, and condensation products of the said partial esters with ethylene oxide, for example polyoxyethylene sorbitan monooleate. The emulsions may also contain sweetening and flavouring agents.

Syrups and elixirs may be formulated with sweetening agents, for example glycerol, propylene glycol, sorbitol or sucrose. Such formulations may also contain a demulcent, a preservative and flavoring and coloring agents. The pharmaceutical compositions may be in the form of a sterile injectable aqueous or oleagenous suspension. This suspension may be formulated according to the known art using those suitable dispersing or wetting agents and suspending agents which have been mentioned above. The sterile injectable preparation may also be a sterile injectable solution or suspension in a non-toxic parenterally-acceptable diluent or solvent, for example as a solution in 1,3-butanediol. Among the acceptable vehicles and solvents that may be employed are water, Ringer's solution and isotonic sodium chloride solution. Cosolvents such as ethanol, propylene glycol or polyethylene glycols may also be used. In addition, sterile, fixed oils are conventionally employed as a solvent or suspending medium. For this purpose any bland fixed oil may be employed including synthetic mono- or diglycerides. In addition, fatty acids such as oleic acid find use in the preparation of injectables.

Compounds useful in the method of treatment of the invention may also be administered in the form of a suppository for rectal administration of the drug. These compositions can be prepared by mixing the drug with a suitable non-irritating excipient which is solid at ordinary temperatures but liquid at the rectal temperature and will therefore melt in the rectum to release the drug. Such materials are cocoa butter and polyethylene glycols.

The instant invention also encompasses a process for preparing a pharmaceutical composition comprising combining a compound of Formula I with a pharmaceutically acceptable carrier. Also encompassed is the pharmaceutical composition which is made by combining a compound of Formula I with a pharmaceutically acceptable carrier.

A therapeutically effective amount of a compound of Formula I can be used for the preparation of a medicament useful for treating or preventing any of the medical conditions described herein, in dosage amounts described herein. For example, a compound of Formula I can be used for the preparation of a medicament useful for the treatment of asthma, allergies and allergic conditions, inflammation, COPD or erosive gastritis. Additionally, the medicament may be useful for preventing or reducing the risk of developing atherosclerotic disease, halting or slowing the progression of atherosclerotic disease once it has become clinically manifest, and preventing or reducing the risk of a first or subsequent occurrence of an atherosclerotic disease event. The medicament comprised of a compound of Formula I may also be prepared with one or more additional active agents, such as those described herein.

The compounds of structural formula I of the present invention can be prepared according to the procedures of the following Schemes and Examples, using appropriate materials and are further exemplified by the specific examples which follow. Moreover, by utilizing the procedures described herein, one of ordinary skill in the art can readily prepare additional compounds of the present invention claimed herein. The compounds illustrated in the examples are not, however, to be construed as forming the only genus that is considered as the invention. The Examples further illustrate details for the preparation of the compounds of the present invention. Those skilled in the art will readily understand that known variations of the conditions and processes of the following preparative procedures can be used to prepare these compounds. The instant compounds are generally isolated in the form of their pharmaceutically acceptable salts, such as those described previously hereinabove. The free amine bases corresponding to the isolated salts can be generated by neutralization with a suitable base, such as aqueous sodium hydrogencarbonate, sodium carbonate, sodium hydroxide, or potassium hydroxide, and extraction of the liberated amine free base into an organic solvent followed by evaporation. The amine free base isolated in this manner can be further converted into another pharmaceutically acceptable salt by dissolution in an organic solvent followed by addition of the appropriate acid and subsequent evaporation, precipitation, or crystallization. All temperatures are degrees Celsius unless otherwise noted. Mass spectra (MS) were measured by electron-spray ion-mass spectroscopy.

The phrase "standard peptide coupling reaction conditions" means coupling a carboxylic acid with an amine using an acid activating agent such as HATU, EDC, and PyBOP in an inert solvent such as dichloromethane or DMF in the presence of a auxiliary nucleophile such as HOAT or HOBT.

The use of protecting groups for the amine and carboxylic acid functionalities to facilitate the desired reaction and minimize undesired reactions is well documented. Conditions required to add and remove protecting groups are found in standard textbooks such as Greene, T, and Wuts, P. G. M., *Protective Groups in Organic Synthesis*, John Wiley & Sons, Inc., New York, NY, 1999. CBZ and BOC are commonly used amino protecting groups in organic synthesis, and their removal conditions are known to those skilled in the art. For example, CBZ may be removed by catalytic hydrogenation in the presence of a noble metal or its oxide such as palladium on activated carbon in a protic solvent such as MeOH or

EtOH. In cases where catalytic hydrogenation is contraindicated due to the presence of other potentially reactive functionalities, removal of CBZ groups can also be achieved by treatment with a solution of hydrogen bromide in acetic acid or by treatment with a mixture of TFA and dimethylsulfide. Removal of BOC protecting groups is carried out with a strong acid, such as TFA, hydrochloric acid, or hydrogen chloride gas, in a solvent such as DCM, dioxane, MeOH, or EtOAc.

Some abbreviations used herein are as follows:

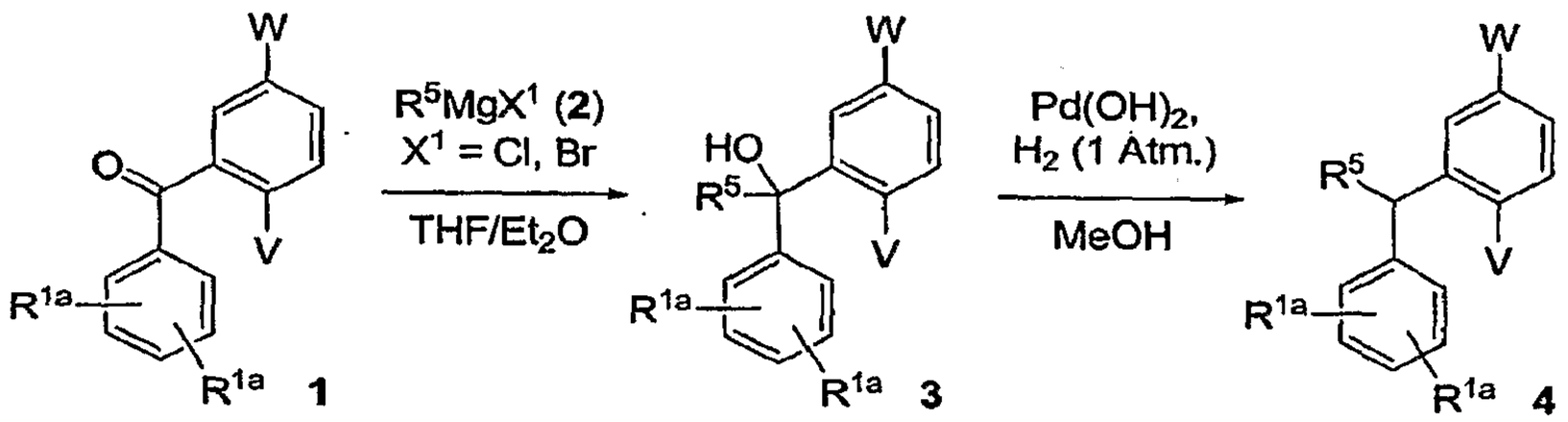
Ac is acetyl; aq. is aqueous; Ar is Aryl; 9-BBN is 9-Borabicyclo[3.3.1]nonane; BOC (Boc) is *tert*-butyloxycarbonyl; Bn is benzyl; Bu is butyl; celite is Celite® diatomaceous earth; CBZ (Cbz) is benzyloxycarbonyl; DCM is dichloromethane; DEAD is diethyl azodicarboxylate; Dess-Martin Periodinane is 1,1,1-tris(acetyloxy)-1,1-dihydro-1,2-benzodoxol-3-(1*H*)-one; DIAD is diisopropylazodicarboxylate; DIBAL-H is diisobutylaluminum hydride; DIPEA is *N,N*-diisopropylethylamine; DMA is *N,N*-dimethylacetamide; DMAP is 4-dimethylaminopyridine; DMF is *N,N*-dimethylformamide; dppf is 1,1'-bis(diphenylphosphino)ferrocene; EDC is 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide.HCl; equiv. is equivalent(s); ES is electron spray ion-mass spectroscopy; Et is ethyl; EtOAc is ethyl acetate; EtOH is ethanol; HATU is *O*-(7-azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate; HCl is hydrogen chloride; HAR is heteroaryl; HOAt is 1-hydroxy-7-azabenzotriazole; HOBt is 1-hydroxybenzotriazole hydrate; HPLC is high performance liquid chromatography; *i* is *iso;* LDA is lithium diisopropylamide; LG is leaving group; *m* is *meta*; Me is methyl; MeOH is methanol; m.p. is melting point; MS is mass spectrum; Ms is methanesulfonyl; NMM is *N*-methylmorpholine; NMO is *N*-methylmorpholine-*N*-oxide; NMP is *N*-methylpyrrolidine; NMR is nuclear magnetic resonance; nOe is nuclear Overhauser effect; *o* is *ortho*; OAc is acetoxy, *p* is *para;* PCC is pyridinium chlorochromate; Ph is phenyl; Pr is propyl; *p*-TSA is *para*-toluenesulfonic acid; PyBOP is benzotriazol-1-yloxytripyrrolidinephosphonium hexafluorophosphate; R^o, R^p, R^r, R^s, R^t, R^u, R^v, R^w, R^x, R^y and R^z are unspecified substituents such that the definition of Formula I of the present invention is satisfied; sat. is saturated; SFC is supercritical fluid chromatography; *t* is *tert*; ^{t}Bu is *tert*-butyl; Tf is trifluoromethanesulfonyl; TFA is trifluoroacetic acid; THF is tetrahydrofuran; TLC is thin layer chromatography; and TPAP is tetrapropylammonium perruthenate.

Reaction schemes A-Q illustrate the methods employed in the synthesis of the compounds of the present invention of structural formula I. All abbreviations are as defined above unless indicated otherwise.

Reaction scheme A illustrates the preferred method for synthesis of compounds of structural formula **4**, wherein either or both of the phenyl rings in starting material **1** may optionally have substituents represented by R^{1a}. In this method, a benzophenone of type **1** is treated with an organometallic reagent of type **2**, capable of transferring an alkyl group, and the product of the reaction is a compound of structural formula **3**. Preferred organometallic reagents for this transformation include organomagnesium (Grignard) or organolithium compounds. When Grignard reagents are employed as shown in scheme A, it is customary to conduct the reaction in a suitable ethereal solvent such as diethyl

ether, or THF or mixtures thereof, at temperatures between –78 °C and the boiling temperature of the solvent. In the case of an organolithium reagent, the reaction can be conducted in a variety of solvents such as diethyl ether or hexanes, at temperatures between –78 °C and room temperature. The Grignard and the organolithium reagents are often purchased commercially, but can be prepared synthetically according to known methods in organic synthesis. Removal of the tertiary hydroxyl group in **3** will depend upon the identity of the W and V substituents. If these substituents are unaffected by hydrogenation conditions, then the hydroxyl group may be removed by hydrogenolysis using a palladium-on-carbon catalyst in a solvent such as MeOH or EtOH and in the presence of hydrogen gas or a hydrogen donor such as formic acid. Occasionally it may be the case that either one or both of the W and V substituents are sensitive to hydrogenation conditions, and in these instances **3** is reacted with an organosilane such as triethylsilane in the presence of a protic acid like TFA or a Lewis acid like boron trifluoride. It is customary to conduct the reaction in an inert organic solvent like DCM or 1,2-dichloroethane at temperatures between 0 °C and boiling point of the solvent. Depending on the nature of the W and V substituents, compound **4** can then be transformed to other compounds of the present invention.

Scheme A

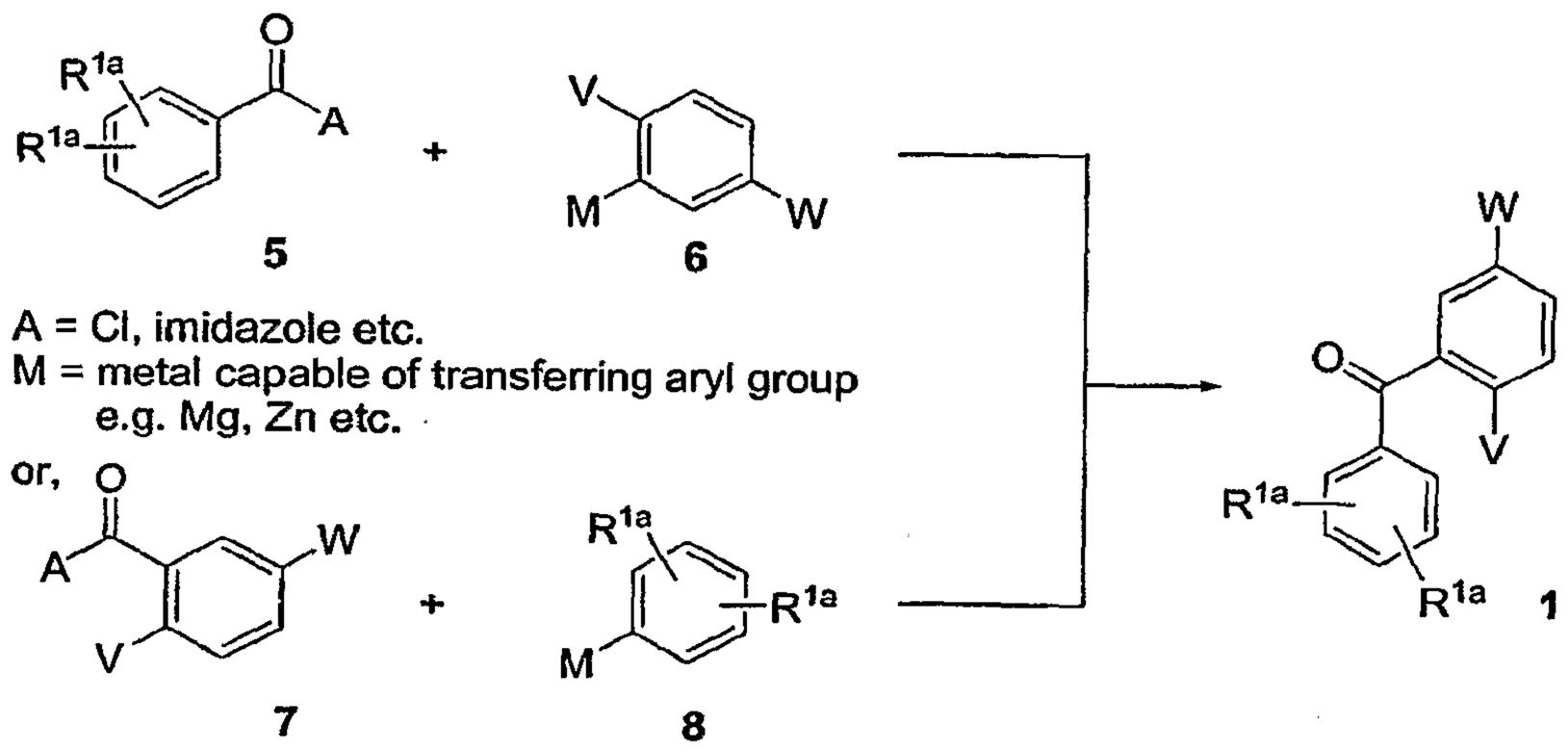


W = X-CR^2R^3-Y as defined in formula I or a group that can be converted to X-CR^2R^3-Y
V = R^1 as defined in formula I or a group that can be converted to R^1

Reaction scheme B illustrates the preferred method for the synthesis of diarylketones of type **1**. In this method, a benzoic acid derivative of type **5** is treated with an organometallic reagent of type **6**, capable of transferring an aryl group. Preferred organometallic reagents for effecting this transformation include organomagnesium (**6**, M = Mg) and organozinc (**6**, M = Zn) compounds. When organozinc compounds (**6**) are employed, it is preferable to employ benzoyl chlorides (**5**, A = Cl) as the second aromatic coupling fragment, and the reaction is referred to as a Negishi-type coupling. The organozinc reagents (**6**) are usually generated and used *in situ* by transmetalation of organomagnesium or organolithium reagents with zinc(II) chloride. However, other methods for the preparation of organozinc reagents are known to those skilled in the art of organic synthesis, and in many cases, they may also be commercially available. Typically, the Negishi-type coupling is conducted in the presence a suitable palladium catalyst such as dichlorobis(triphenylphosphine)palladium(II) or

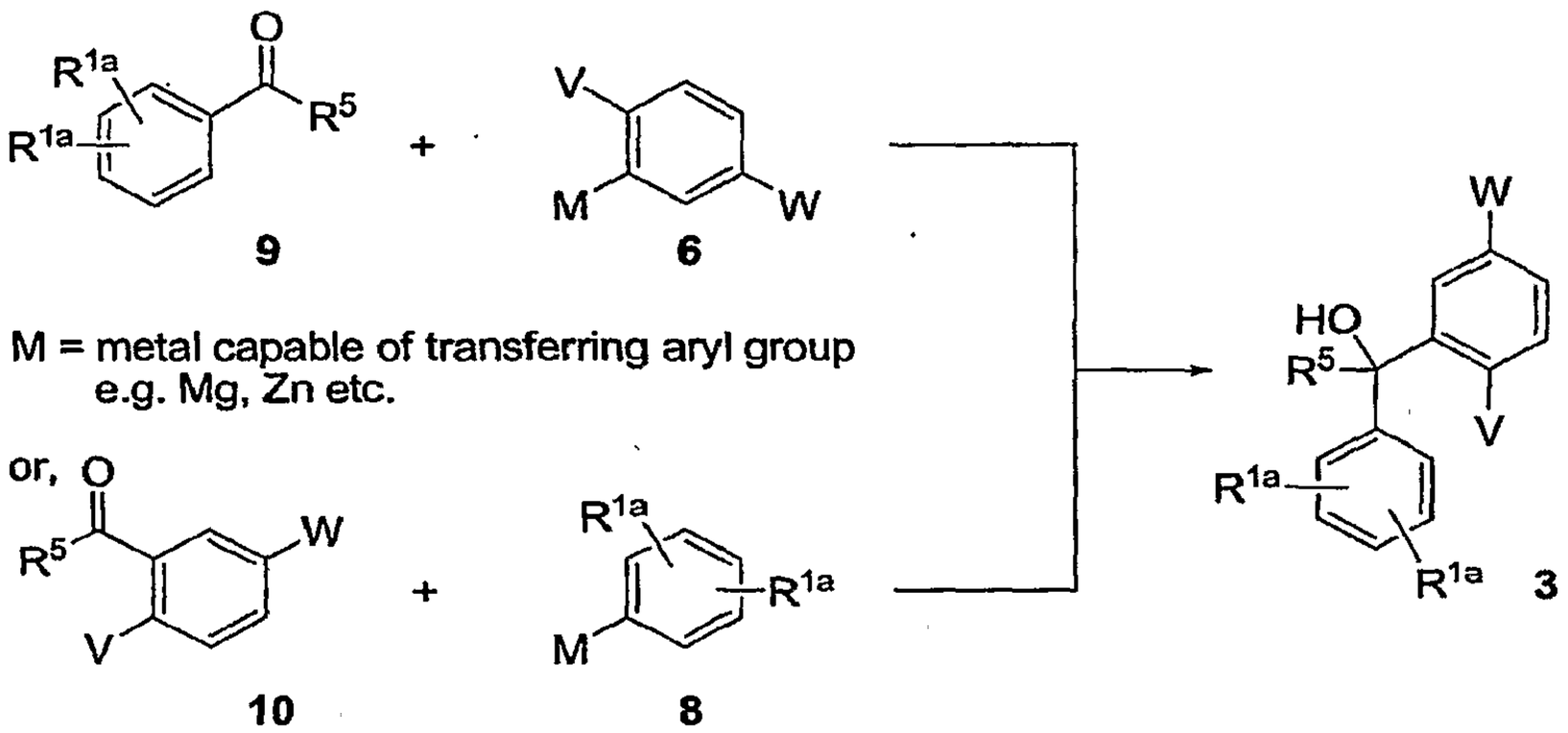
tetrakistriphenylphosphinepalladium(0) and an inert organic solvent such as THF, or DMA or the like. It is customary to perform the reaction at temperatures between 0° C and ambient temperature, for a period of 2-24 hours. If an organomagnesium reagent (**6**) is used as one of the aromatic coupling fragments, it is preferable to employ an activated carboxylic acid, such as an acyl imidazole derivative (**5**, A = imidazole), as the complementary reaction component. The acyl imidazole derivative (**5**) is usually generated and used *in situ* by treatment of the respective benzoic acid precursor (**5**, A = OH) with carbonyl diimidazole, in a dipolar aprotic solvent such as DCM or THF or mixtures thereof, at temperatures between 0° C and room temperature. The reaction between the intermediary acyl imidazole species (**5**) and the Grignard reagent (**6**) is usually performed at low temperature, such as -78° C, to avoid side reactions, and the product of the reaction is a benzophenone of structural formula **1**. In yet another variation of this method, **1** can also be prepared from the reaction of a benzoic acid derivative of type **7** and an organometallic reagent of type **8**, using the methods discussed in this passage.

Scheme B



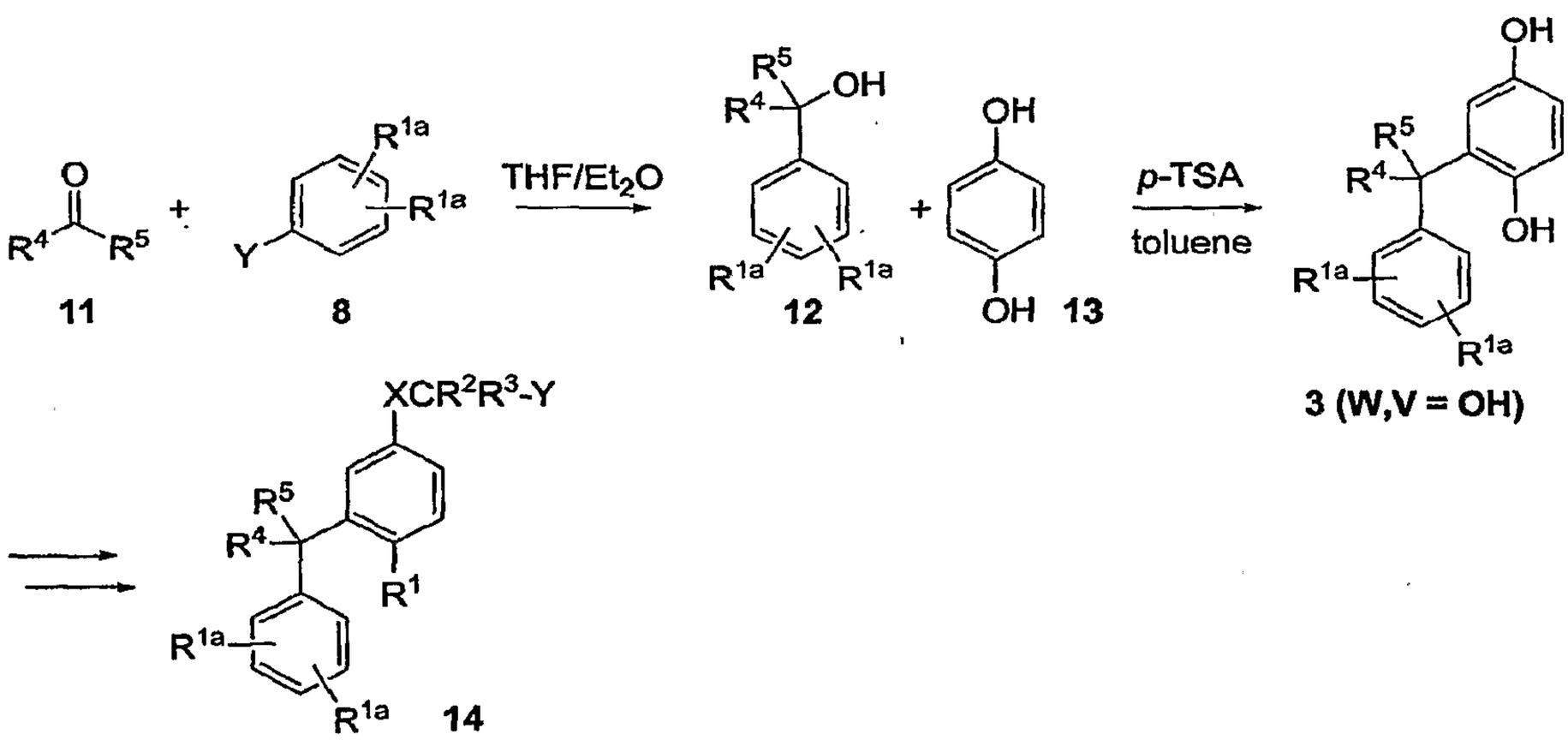
Reaction scheme C illustrates an alternative method for the synthesis of diarylmethanols of type **3**. In this method, an alkyl-aryl ketone of type **9** is treated with an organometallic reagent of type **6**, capable of transferring an aryl group. Preferred organometallic reagents for effecting this transformation include organomagnesium (Grignard) or organolithium compounds, and are used in a similar manner to that described in Scheme A. Alternatively, **3** can also be prepared from the reaction of an alkyl-aryl ketone of type **10** and an organometallic reagent of type **8**.

Scheme C



Reaction scheme D illustrates the preferred method for the generation of compounds of type **3** (W, V = OH). In this method, the aromatic halves are introduced using a combination of the aforementioned Grignard methodology and a Friedel-Crafts arylation strategy. Conditions for effecting the latter transformations are as described above. Compounds of type **3** (W, V = OH) can be elaborated in numerous ways, some examples of which are shown in the subsequent schemes, to furnish compounds of the present invention.

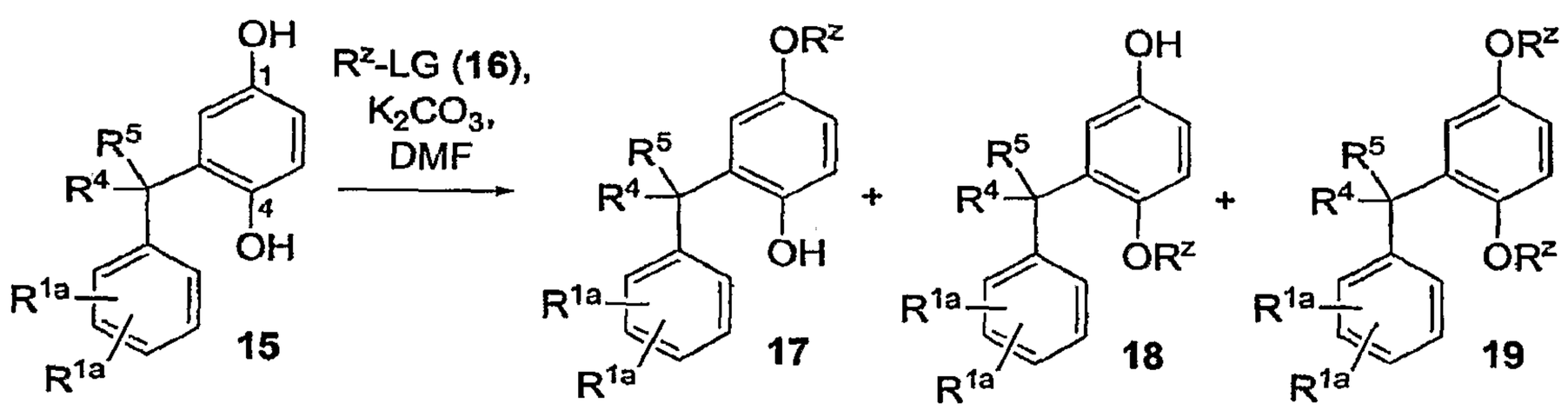
Scheme D



Reaction scheme E illustrates the synthesis of a compound of structural formula **17** in which it is desirable to first elaborate the more reactive hydroxyl group (1-position) of **15**. For example,

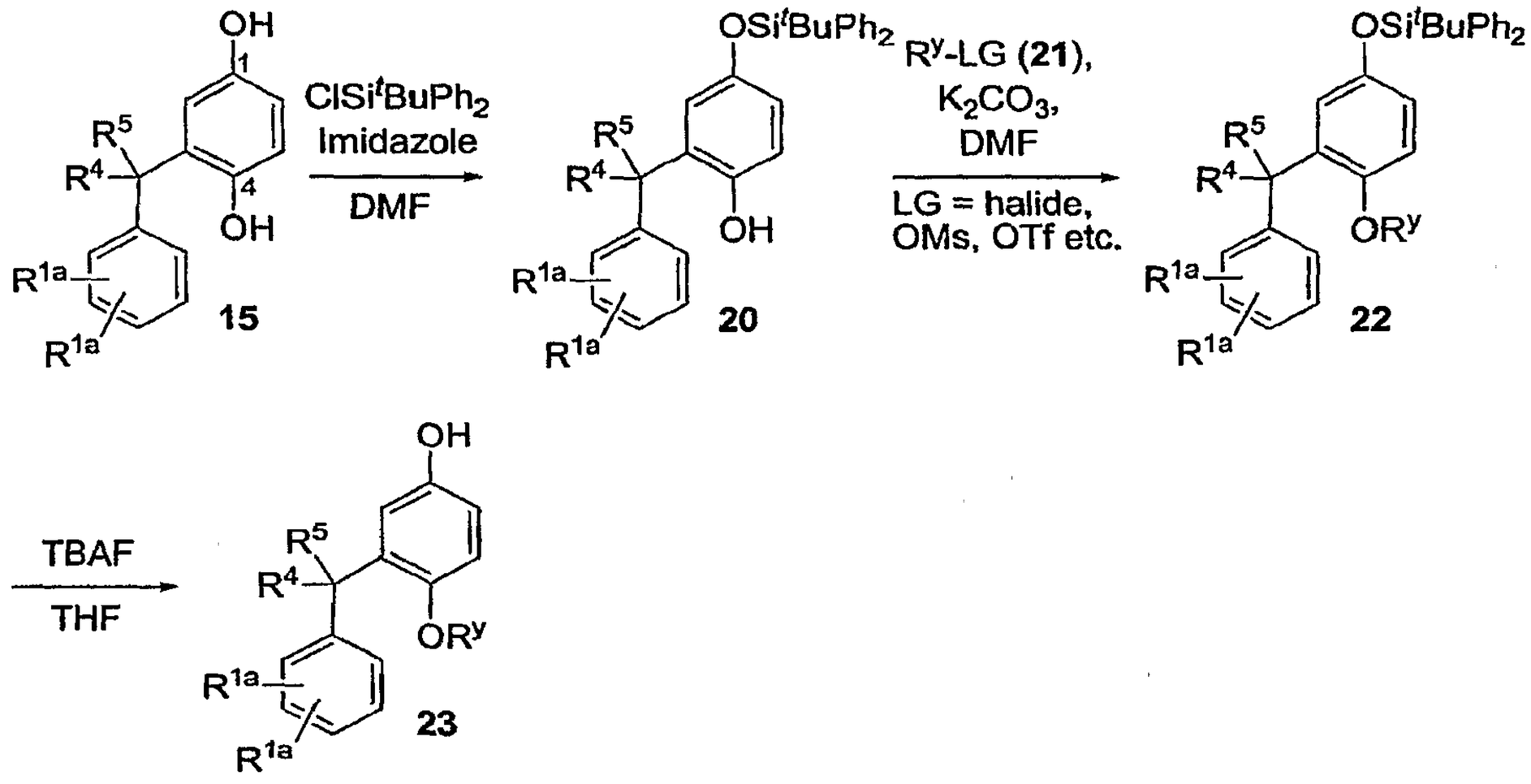
15 can be directly alkylated using an alkylating agent of type **16.** The reaction is conducted typically in the presence of a suitable base such as potassium carbonate or cesium carbonate, in a polar aprotic solvent such as DMF, in which the substituent LG of **16** is a good leaving group such as a halide, mesylate or triflate. The major products from the reaction are the *mono*-alkylated product of structural formula **17** and the *bis*-alkylated product of structural formula **19** which can be readily separated by flash chromatography. In some cases, a small amount of the regioisomeric *mono*-alkylated product **18** is observed.

Scheme E



Reaction scheme F illustrates a protecting group strategy for the synthesis of a compound of type **22** in which it is desirable to elaborate the less reactive hydroxyl group (4-position) of **15.** For example, the more reactive hydroxyl group (1-position) in **15** can be selectively protected with a range of groups known in organic synthesis, exemplified in this case by a silicon-based protecting group approach. In this method, **15** is treated with a suitable silylating agent such as chloro-*tert*-butyldiphenylsilane, in the presence of imidazole, in a solvent like DMF. The reaction is conducted typically at temperatures between 0 °C and room temperature, for periods of 12-24 hours. The product is a silyl ether of type **20**, which can be directly alkylated using the conditions described in the discussion for scheme E to afford a product of type **22.** The silicon protecting group can be removed by any of the appropriate desilylation methods such as treatment with TBAF in THF or hydrogen fluoride in pyridine and the product of this reaction is a phenol of type **23.**

Scheme F

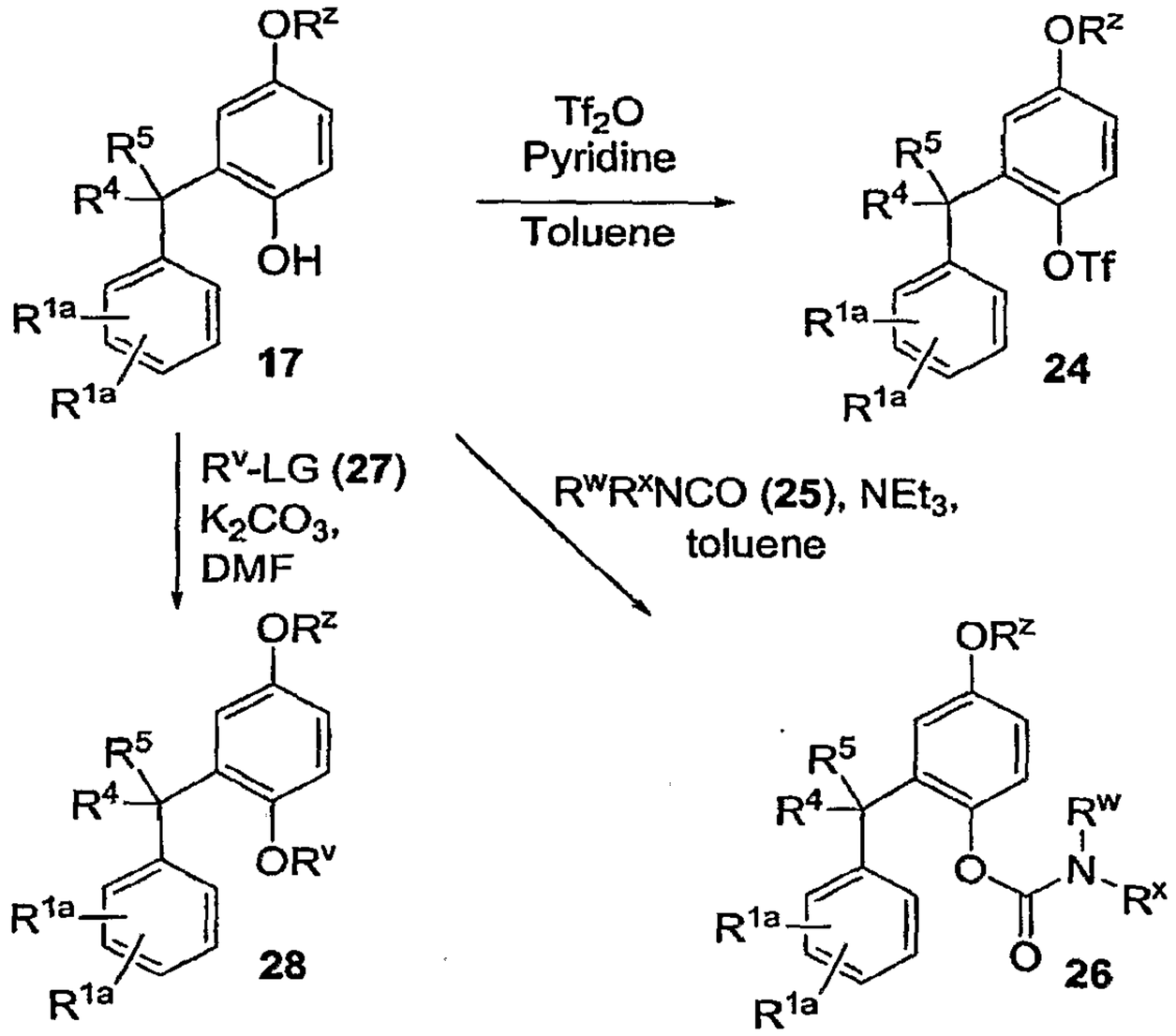


Reaction scheme G illustrates some of the preferred methods for the elaboration of **17**. For example, **17** can be treated with a triflating agent such as triflic anhydride or the like in the presence of a suitable base such as pyridine or triethylamine in an aprotic solvent like toluene. It is customary to conduct the reaction at temperatures between –78 °C and room temperature, for periods of 1-24 hours. The product of the reaction is a triflate of structural formula **24** which can be elaborated by a variety of synthetic methods known to those skilled in organic synthesis, three of which are outlined in schemes H, I, and J.

Alternatively, **17** can be treated with an isocyanate of type **25** in the presence of a suitable base such as triethylamine, in an inert solvent like toluene (scheme G). Typically, the isocyanate reagent **25** can be purchased commercially or prepared synthetically and the product of the reaction is a carbamate of type **26**. In certain cases it may be preferable to generate **25** *in situ*, and this is typically accomplished from an appropriate precursor such as an acyl azide. In an alternative method, **17** can be treated with a suitable carbonyl equivalent such as phosgene, triphosgene or carbonyl diimidazole. After a short period of time, typically between 0.1-1 hour, a primary or secondary amine is added and the product of the reaction is a carbamate of structural formula **26**. The reaction sequence is conducted in a suitable inert organic solvent like DCM, at temperatures between 0 °C and room temperature, for periods of 1-24 hours.

In yet another example, **17** can be directly alkylated using the conditions described in the discussion of scheme D to afford a derivative of type **28**.

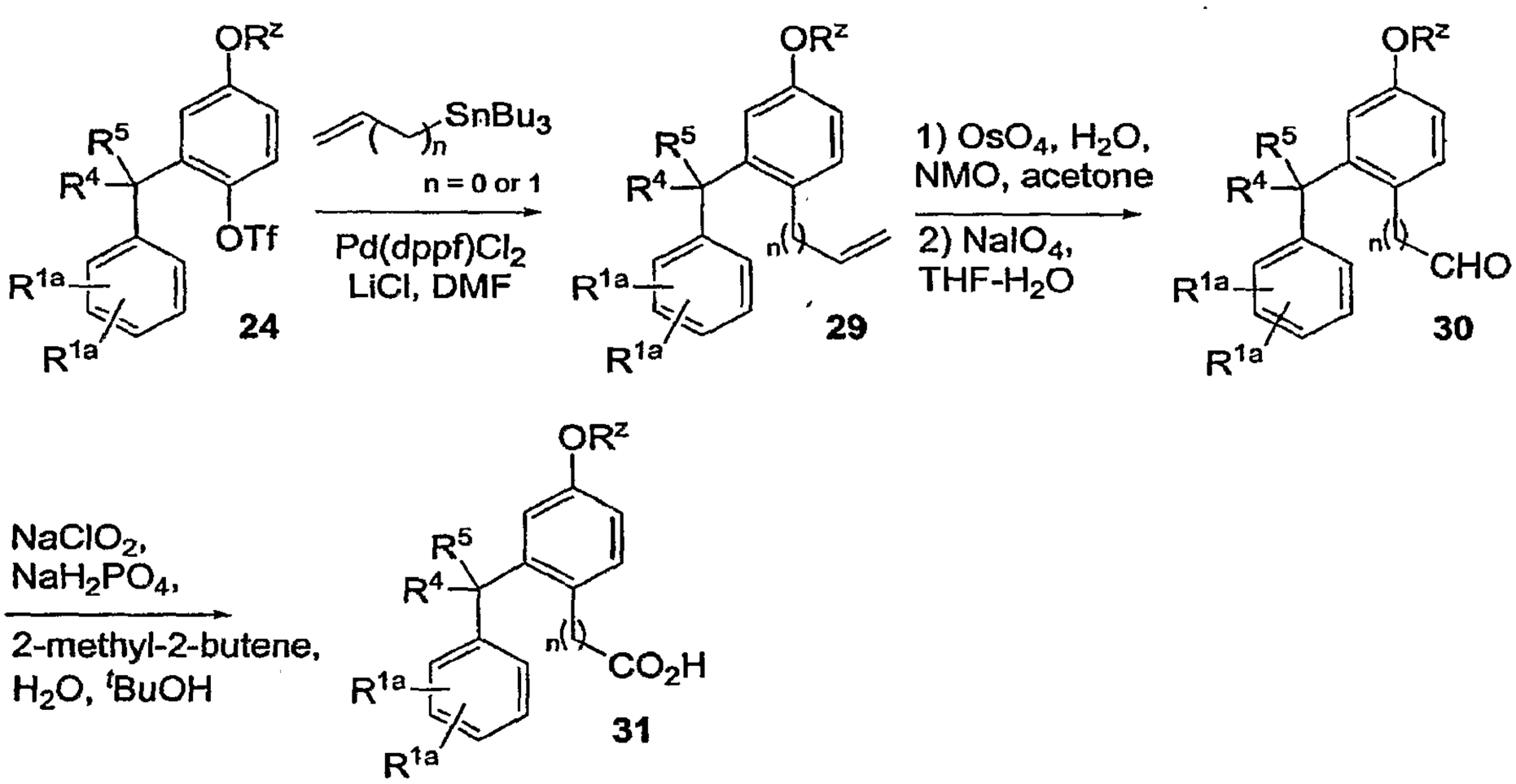
Scheme G



Reaction scheme H illustrates the preferred method of synthesis of compounds of structural formula **29**, **30** and **31**. In this method, **24** is treated with either allyltributylstannane or vinyltributylstannane in the presence of a suitable a palladium catalyst such as [1,1'-bis(diphenylphosphino)ferrocene]dichloropalladium(II), in an inert organic solvent like DMF or NMP. The reaction is usually conducted at elevated temperatures, typically between 50-120 °C, for periods of 2-24 hours. In certain cases, it may be essential to use an additive such as lithium chloride to promote the reaction. Often, the reaction times can be significantly reduced if the reaction is conducted under microwave irradiation. The product of the reaction is an alkene of structural formula **29** which can be synthetically elaborated, using a variety of methods known in organic synthesis. For example, **29** can be oxidatively cleaved to afford an aldehyde of type **30**, which can be further oxidized to a carboxylic acid derivative of structural formula **31**. A preferred method for the oxidative cleavage reaction is the two-step process shown in reaction scheme H. Alkene **29** is first oxidized to a vicinal diol using catalytic osmium tetraoxide in the presence of a stoichiometric reoxidant such as NMO, in a solvent system such as acetone-water. The intermediate vicinal diol which forms is generally not isolated, but is in turn subjected to cleavage with sodium periodate in a suitable mixed solvent system like THF-water to afford **30**. Both steps in the oxidative cleavage sequence are generally completed during periods of several minutes to a few hours, at temperatures between 0 °C and room temperature. Alternatively, the oxidative cleavage of **29** may also be accomplished using ozone, or by other methods known to those skilled in the

art. Aldehyde **30** can then be further oxidized to **31** using a buffered chlorite oxidation system. In this method, **30** is treated with sodium chlorite and monobasic sodium phosphate in the presence of a chlorine scavenger, such as 2-methyl-2-butene. The reaction is conducted typically in a solvent system like *n*-butanol-water, for periods of 1-6 hours, at temperatures between 0 °C and room temperature. In certain cases, **29** can be directly converted to **31** using the sodium periodate/ruthenium trichloride reagent system. Both **30** and **31** can be elaborated in numerous ways known in organic synthesis to furnish other compounds of the present invention.

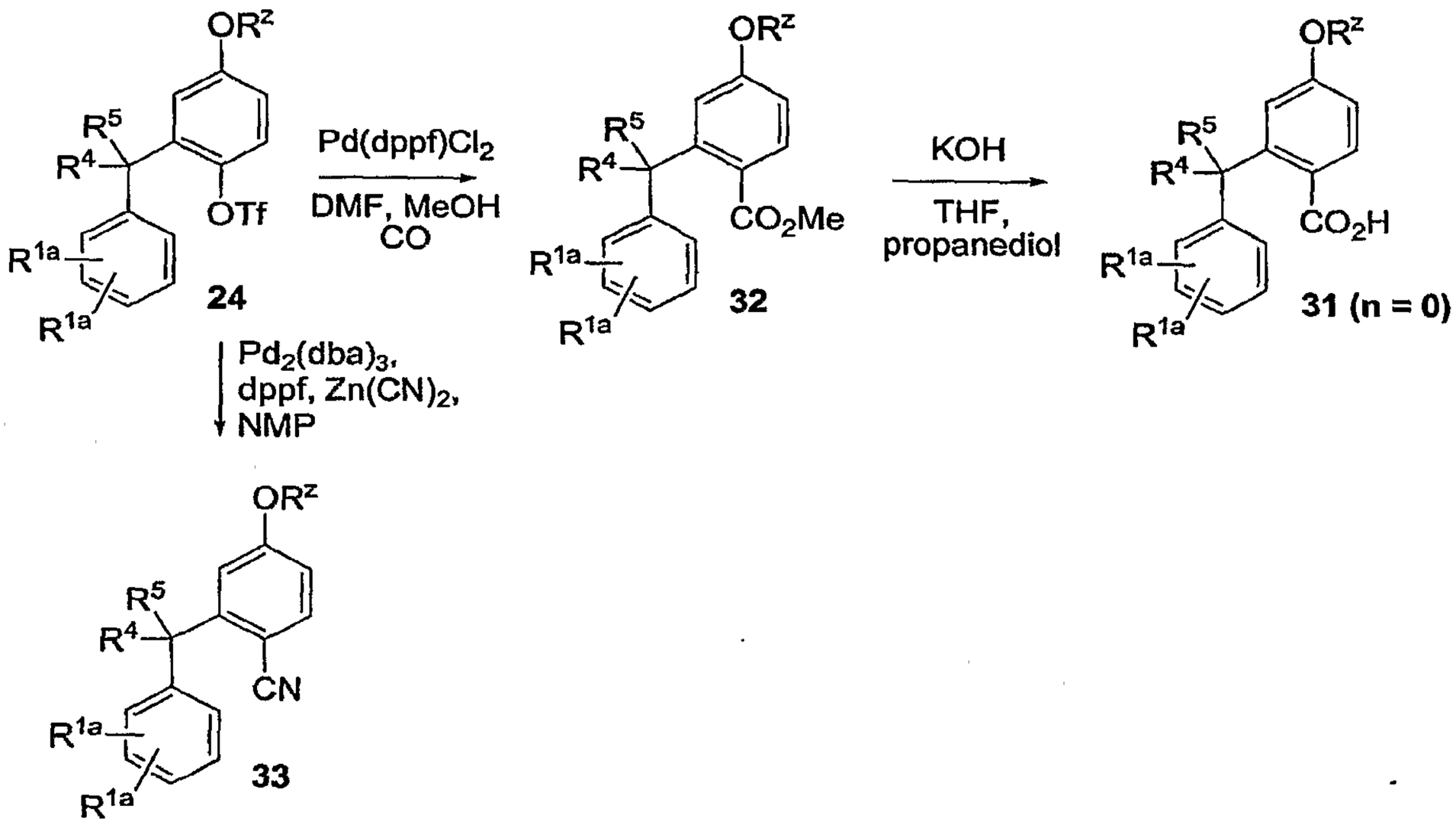
Scheme H



Reaction scheme I illustrates a preferred method of syntheses of compounds of structural formula **32**, **33**, and **34**. In this method, **24** is treated with MeOH in the presence of a suitable palladium catalyst such as [1,1'-bis(diphenylphosphino)ferrocene]dichloropalladium(II), in an inert organic solvent like DMF. The reaction is usually conducted at elevated temperature, typically between 50-100 °C, for periods of 6-24 hours, under an atmosphere of carbon monoxide. In certain cases it may be advisable to use elevated pressures of carbon monoxide or an additive such as lithium chloride to promote or accelerate the reaction. In specific instances, it may be preferable to perform the reaction under the influence of microwave irradiation. The product of the reaction is an ester of structural formula **32** which can be converted to **31** (n = 0) using a variety of hydrolytic methods known to those skilled in the art organic synthesis. A compound of type **24** can also be converted to a compound of structural formula **33**, again using organopalladium based methods. For example, **24** can be treated with a cyanide source, such as zinc cyanide, or potassium cyanide or the like, in the presence of a suitable palladium catalyst/ligand reagent system. It is customary to conduct the reaction in inert organic solvent, preferably a dipolar aprotic solvent, such as DMF and NMP at elevated reaction temperatures, typically between

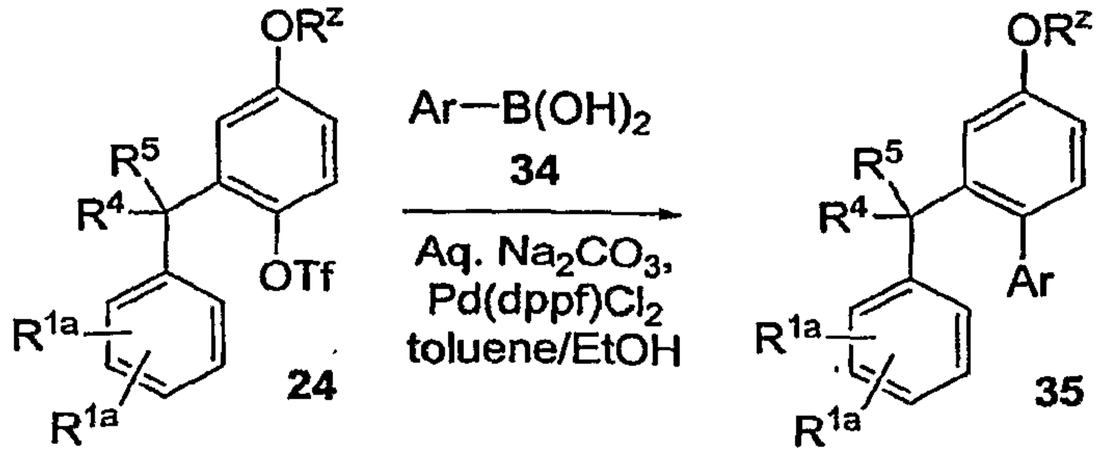
50-140 °C, for periods of 6-24 hours. The product of the reaction is a nitrile derivative of type **33**, which like **31** and **32**, can be elaborated to other compounds of the present invention.

Scheme I



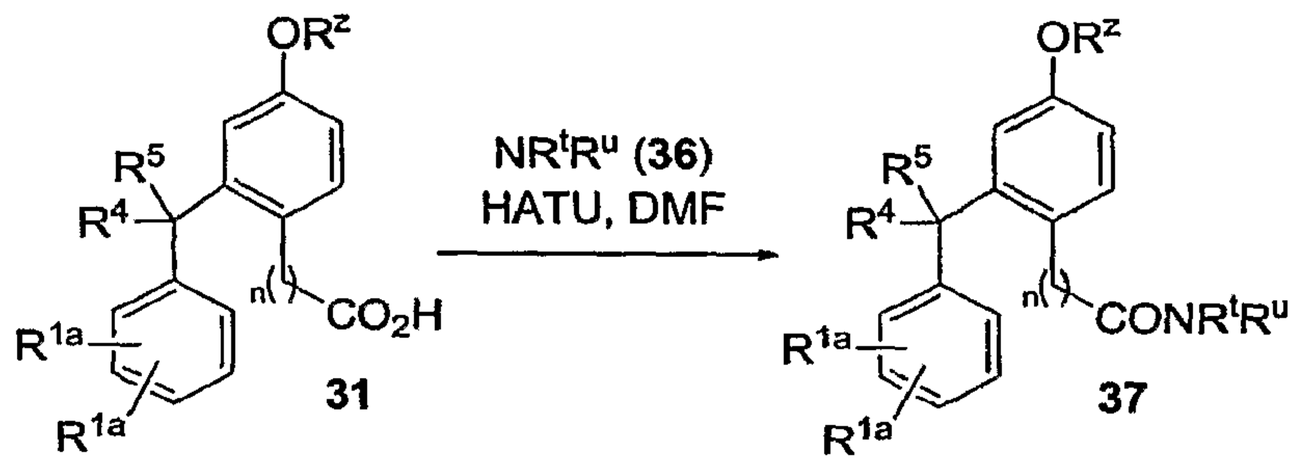
Reaction scheme J illustrates the preferred method of synthesis of compounds of structural formula **35**. In this method, commonly referred to as the Suzuki reaction, **24** is treated with an aryl- or heteroaryl-boronic acid of type **34** in the presence of a suitable palladium(0) catalyst such as [1,1'-bis(diphenylphosphino)ferrocene]-dichloropalladium(II) and aqueous sodium carbonate. The reaction is usually performed in a suitable combination of inert organic solvents such as toluene-EtOH, at about 80 °C, for a period of 6-24 hours and the product is a biaryl of structural formula **35**.

Scheme J



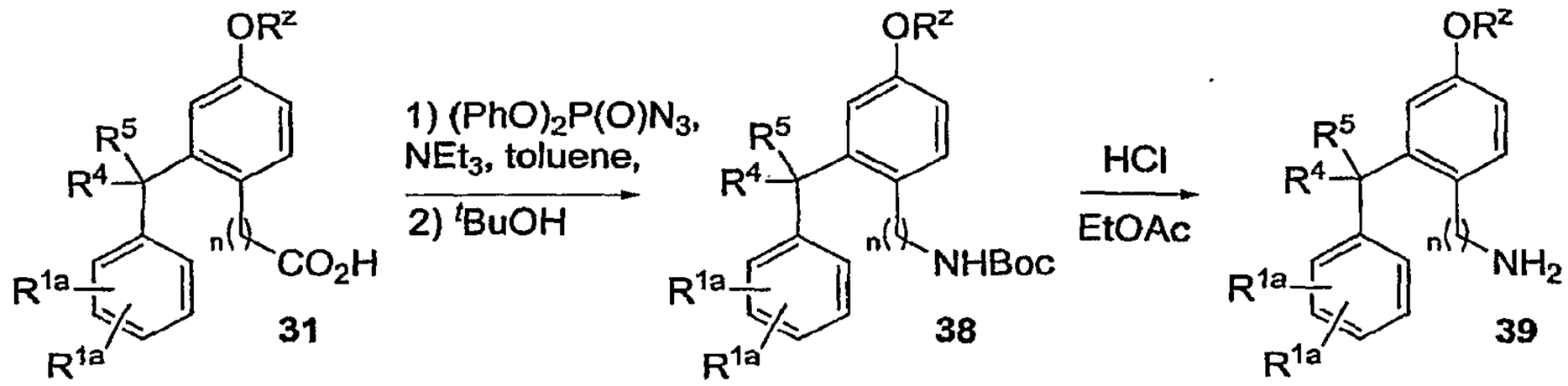
Reaction Scheme K illustrates the synthetic methodology in the most general case in which **31** is treated with an amine of type **36** to afford an amide of type **37**. The amide bond coupling reaction illustrated in reaction scheme K is conducted in an appropriate inert solvent such as DMF, DCM or the like and may be performed with a variety of reagents suitable for amide coupling reactions such as HATU, EDC or PyBOP. Preferred conditions for the amide bond coupling reaction shown in reaction Scheme K are known to those skilled in organic synthesis. Such modifications may include, but are not limited to, the use of basic reagents such as triethylamine, DIPEA, or NMM, or the addition of an additive such as HOAt or HOBt. Alternatively, **36** may be treated with an activated ester or acid chloride derivative of **31**, which also affords **37**. The amide bond coupling shown in reaction Scheme K is usually conducted at temperatures between 0 °C and room temperature, occasionally at elevated temperatures, and the coupling reaction is typically conducted for periods of 1 to 24 hours.

Scheme K



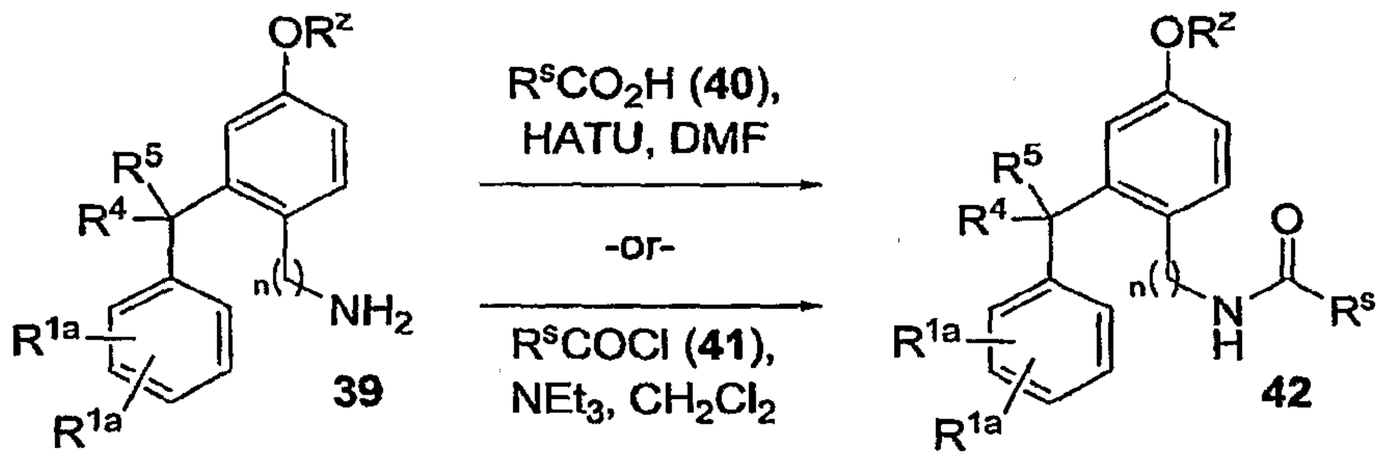
Reaction Scheme L illustrates a preferred method for the synthesis of a compound of type **39**. In this method, **31** is subjected to the Curtius reaction to afford the *N*-Boc protected amine of structural formula **38**. The reaction is performed by reacting **31** with diphenylphosphoryl azide in the presence of a tertiary amine such as triethylamine or DIPEA in a solvent such as toluene. The initial product is generally accepted to be the acyl azide, which is rearranged to the isocyanate in a thermal process analogous to the Wolff rearrangement of acyl carbenes. The rearrangment is conducted typically at the reflux temperature of the solvent, for instance 110 °C, and the rearrangement is usually completed in periods of 1-5 hours. The intermediate isocyanate which forms is generally not isolated, but is in turn subjected to *in situ* reaction with a suitable alcohol such as *tert*-butyl alcohol to afford carbamate **38**. The *N*-Boc group can be removed by a suitable deprotection method such as treatment with hydrogen chloride in EtOAc or TFA in DCM. The deprotection is conducted typically at temperatures between 0 °C and room temperature, and the reaction is usually complete in 0.5-3 hours. The product amine of structural formula **39** can be used as a coupling partner in reaction Scheme M or synthetically modified using a variety of methods known in organic synthesis to afford compounds of the present invention.

Scheme L



Reaction scheme M illustrates preferred methods for the syntheses of compounds of type **42**. For example, **39** can participate in amide bond coupling reactions with a carboxylic acid of type **40** to afford an amide structural formula **42**, using the reagents and conditions described for the generalized amide coupling protocol shown in reaction Scheme M. Alternatively, **39** may also be treated with an activated ester or acid chloride derivative of type **41**, which also affords **42**. Typical conditions for effecting such a transformation include treatment of **39** with acid chloride **41** in the presence of a tertiary amine base such as triethylamine. It is customary to perform the reaction in an inert organic solvent such as DMF or DCM, at temperatures between 0 °C and the reflux temperature of the solvent, frequently at room temperature and for periods of 1-24 hours.

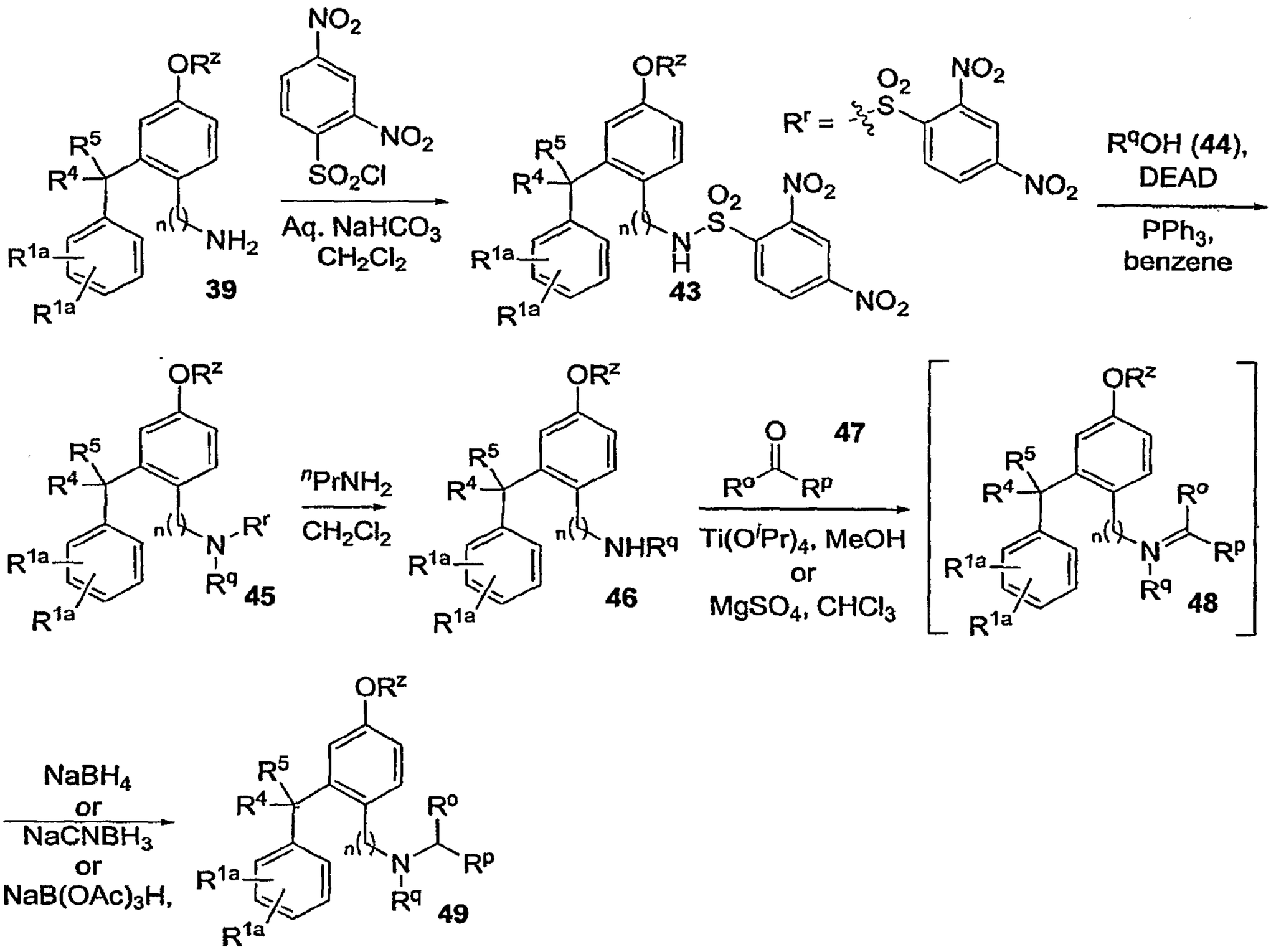
Scheme M



As shown in reaction scheme N, **39** can also be elaborated using the Fukuyama modification of the Mitsunobu reaction (Fukuyama, T.; Jow, C.-K.; Cheung, M. *Tetrahedron Lett.* **1995**, *36*, 6373-74). For example, **39** may be reacted with an arylsulfonyl chloride such as 2-nitrobenzenesulfonyl chloride, 4-nitrobenzenesulfonyl chloride or 2,4-dinitrobenzenesulfonyl chloride and a tertiary amine base such as 2,4,6-collidine or 2,6-lutidine in an inert organic solvent such as DCM. Alternatively, the reaction can also be performed under the classical Schötten-Baumann conditions as shown in scheme N, in which **39** and the arylsulfonyl chloride are allowed to react in aqueous alkaline solution. The product of this reaction is a sulfonamide of type **43**, which can be further modified by reaction with an alcohol of type **44** in the presence of triphenylphosphine and an activating agent such as

DEAD, DIAD, or the like. The reaction is performed in a suitable inert organic solvent such as benzene, toluene, THF or mixtures thereof, typically at room temperature, and the reaction is generally complete in 0.5-3 hours. The product of this reaction is a sulfonamide of type **45**, which can be desulfonylated in the presence of either a nucleophilic amine like *n*-propylamine, in a solvent such as DCM, or with a combination of mercaptoacetic acid and triethylamine in DCM. In either case, the reaction is conducted typically at room temperature, for periods of 5 minutes to 1 hour. When a 2- or 4-nitrobenzenesulfonyl derivative is employed, the cleavage of the sulfonamide is accomplished with either combinations of thiophenol and potassium carbonate in a solvent like DMF, or with mercaptoacetic acid and lithium hydroxide in DMF. In either case, the reaction is conducted at room temperature, for periods of 1-3 hours. The secondary amine product of type **46** can be modified further using a variety of methods known in organic synthesis to provide other compounds of the present invention. For example, **46** may be subjected to a reductive amination reaction with an aldehyde or ketone of type **47** to afford compounds of type **49**. Typical conditions for effecting such a reductive amination include preforming an imine **48** from aldehyde/ketone **47** and amine **46** followed by reduction of the intermediate imine with reagents capable of reducing carbon-nitrogen double bonds such as sodium borohydride, sodium cyanoborohydride or the like. Formation of the intermediate imine **48** may occur spontaneously in solution or it may be promoted with Lewis acid type reagents such as titanium (IV) isopropoxide or magnesium sulfate or the like. The formation of the imine is generally performed at temperatures between 0°C and the reflux temperature of the solvent, frequently at room temperature. The imine formation step is generally allowed to proceed to completion over a period of several hours to 1 day prior to the reduction step which minimizes the formation of alcohol by-products formed by simple reduction of the keto group in compounds of general formula **47**. The intermediate imine **48** may in some cases be isolated and purified; however, it is generally preferred to use it directly in the reduction step. The reduction of the imine **48** is typically conducted in an alcohol-based solvent such as MeOH or EtOH at temperatures between 0°C and room temperature, and the reduction is generally completed in a period of several hours or less.

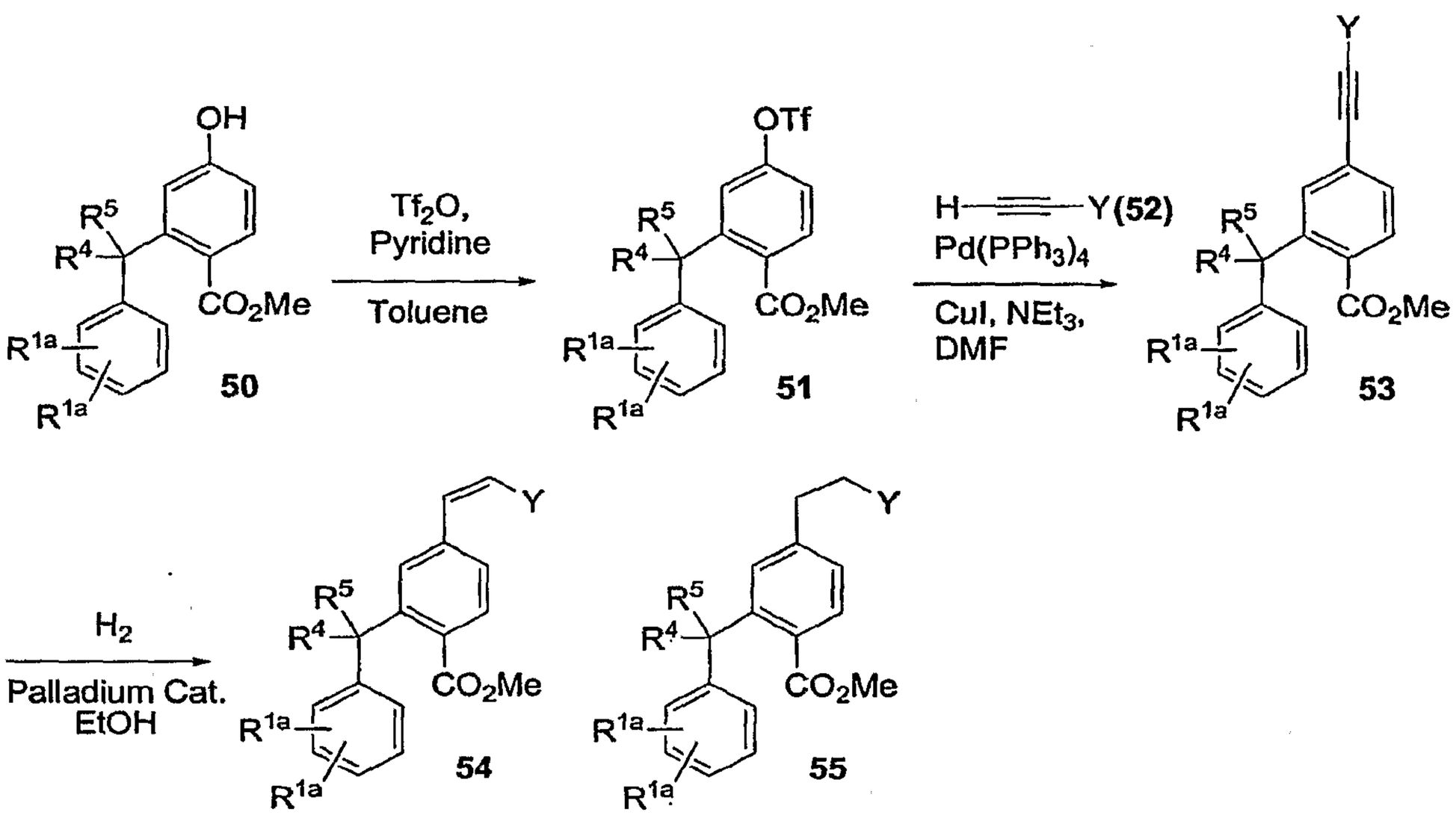
Scheme N



Reaction scheme O illustrates the preferred method of synthesis of compounds of structural formula **54** and **55**, in which group X (X-CR^2R^3-Y) of the present invention is a carbon atom. In this method, **50** is initially converted to triflate **51** using either the conditions described in scheme G, or variations thereof. Cross-coupling of **51** with a terminal alkyne of type **52**, in the presence of a suitable palladium catalyst, is referred to as the Sonogashira reaction. In the latter process, a copper(I) salt such as copper(I) iodide is also employed as co-catalyst, and the reaction is typically performed in the presence of an excess of amine base, such as triethylamine and diethylamine. The reaction is conducted in an inert organic solvent such as DMF, at temperatures ranging from ambient temperature to about 100 °C, for a period of 6-24 hours. The product of the reaction is an alkyne of type **53** which can then be converted into an alkene derivative of type **54** or a saturated alkane derivative of type **55**. If **54** is desired, preferred conditions for performing the partial reduction of **53** involve the use of a Lindlar catalyst reagent system under an atmospheric or elevated pressure of hydrogen. The reaction is usually conducted in an inert organic solvent, such as EtOH and EtOAc, or combinations thereof, and at room

temperature for a period of 3-15 hours. If **55** is desired, then the reduction of **53** is performed with any one of a variety of palladium-on-carbon catalysts, at either atmospheric or elevated pressure of hydrogen.

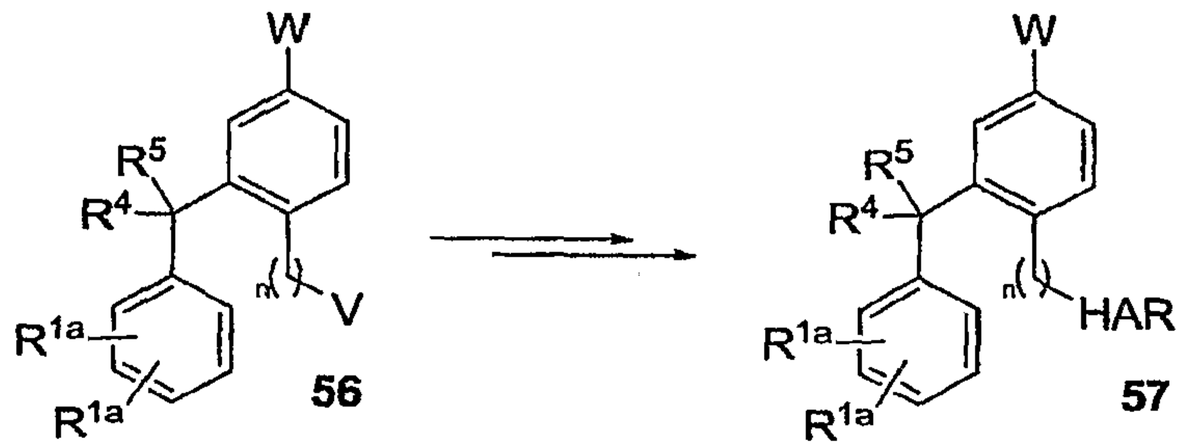
Scheme O



Scheme P illustrates that compounds of structural formula **56** can be elaborated to a variety of heterocyclic (HAR) derivatives of structural formula **57** using known methods in organic synthesis. Specific examples of such transformations are shown in the Examples section. Leading references for effecting such transformations include:

1) Joule, J.A; Mills, K and Smith, G.F. *Heterocyclic Chemistry*, Chapman & Hall, 1995, 3rd Edn., and references cited therein;
2) Katrittzky, A.R.; Rees, C.W. (Eds), *Comprehensive Heterocyclic Chemistry*: *The Structure, Reactions, Synthesis, and Uses of Heterocyclic Compounds*, Pergamon Press, Oxford, 1984, 8v, and references cited therein; and
3) *Comprehensive Heterocyclic Chemistry II*: *Review of the Literature 1982-1995*: *The Structure, Reactions, Synthesis and Uses of Heterocyclic Compounds*, Pergamon Press, New York, 2nd Edn., 1996, 11v, and references cited therein.

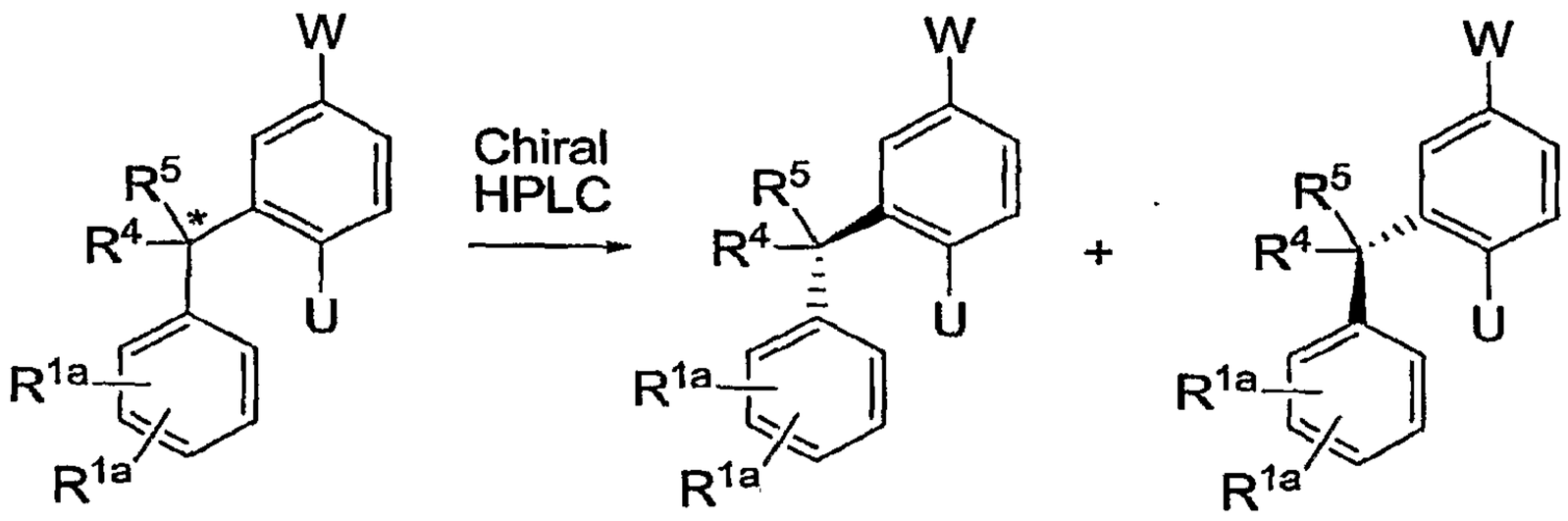
Scheme P



V = CO_2H, CO_2Me, CN etc.
W = XCR^2R^3Y or a group that can be converted to -XCR^2R^3Y

Scheme Q illustrates the preferred method for the resolution of a compound of structural formula **58** in which the asterisked carbon is a center of chirality. Generally, the latter, or intermediates en route to their preparation, may be resolved to afford enantiomerically pure compounds such as **59** and **60** by chiral stationary phase liquid chromatography techniques or other suitable methods known in organic synthesis. For example, in cases in which **58** features acidic or basic functionality, resolution of racemic mixtures can be achieved *via* crystallization of diastereoisomeric salts, derived from **58** and a chiral carboxylic acid (**58** contains basic functionality) or a chiral amine (**58** contains acidic functionality).

Scheme Q

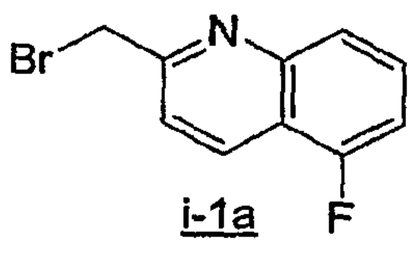


W = XCR^2R^3Y or a group that can be converted to XCR^2R^3Y;
U = R^1 or a group that can be converted into R^1

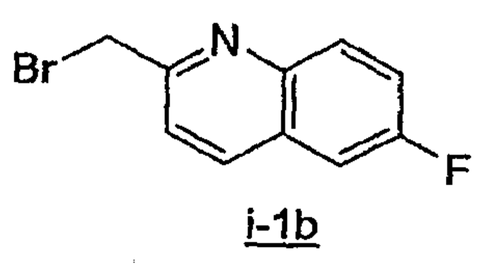
The following examples are provided to illustrate the invention and are not to be construed as limiting the scope of the invention in any manner.

Preparation of Intermediates:

2-(Bromomethyl)-5-fluoroquinoline (i-1a) and 2-(Bromomethyl)-6-fluoroquinoline (i-1b) were prepared according to the procedures described in *Bioorg. Med. Chem. Lett* **1998**, *8*, 965-970.

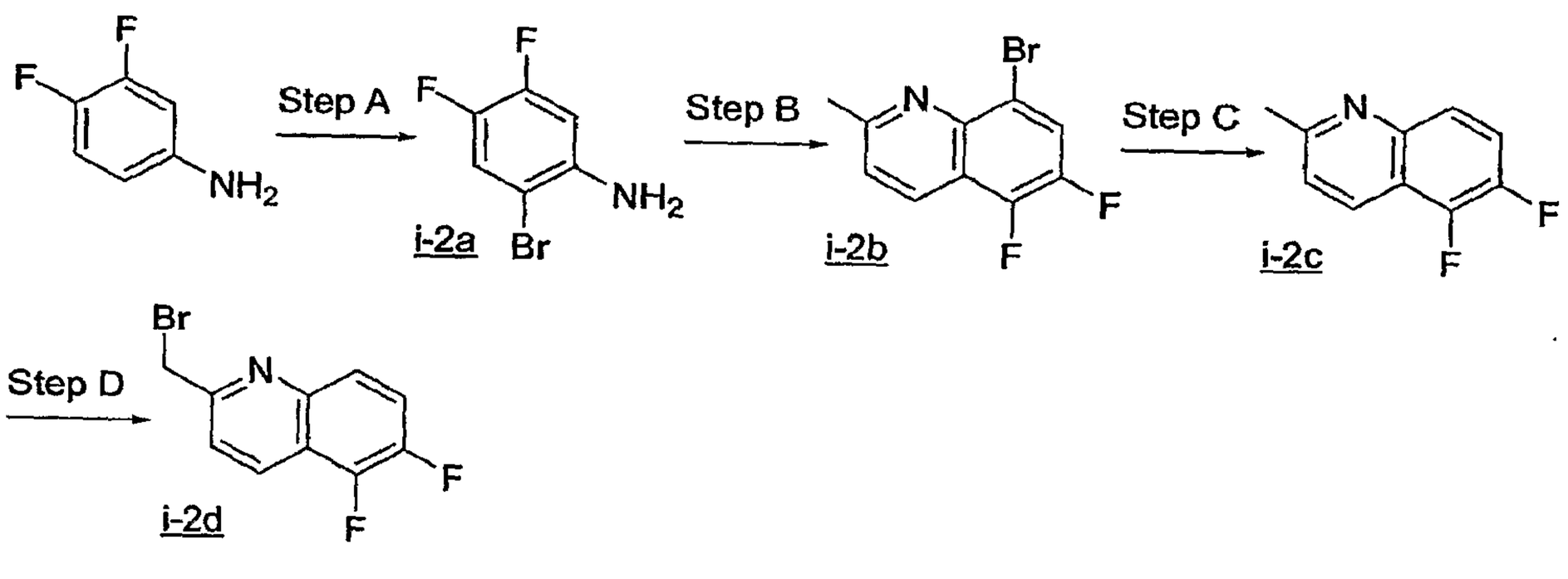


i-1a



i-1b

Scheme i-2



Preparation of 2-(bromomethyl)-5,6-difluoroquinoline (i-2d)

Step A: Preparation of (2-bromo-4,5-difluorophenyl)amine (i-2a)

Potassium carbonate (2.76 g, 20.0 mmol) was added to a stirred solution of 3,4-difluoroaniline (2.58 g, 20.0 mmol) in DCM (100 mL) at room temperature, and the resulting mixture was cooled to -15 °C. A solution of bromine (3.20 g, 1.02 mL, 20.0 mmol) in DCM (10 mL) was added dropwise *via* syringe. After 15 min, the reaction mixture was poured into ice/water and extracted three times with DCM. The combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by flash chromatography on silica gel (gradient elution; 10-20% EtOAc/Hexanes as eluent) to afford the title compound i-2a, *m/z* (ES) 210 $(MH)^+$.

Step B: Preparation of 8-bromo-5,6-difluoro-2-methylquinoline (i-2b)

A stirred suspension of i-2a (733 mg, 4.46 mmol) in 6N HCl (25 mL) was heated at 100 °C until the reaction mixture turned homogeneous. Toluene (6.0 mL) was added followed by dropwise addition of crotonaldehyde (740 mg, 8.92 mmol). After 3 h, the reaction mixture was cooled to room temperature and the separated aq. layer was neutralized cautiously with aq. 5N sodium hydroxide (ice cooling). The aq. phase was then extracted three times with EtOAc, the combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by flash chromatography on silica gel (gradient elution; 5-10% EtOAc/Hexanes as eluent) to afford the title compound i-2b, *m/z* (ES) 260 $(MH)^+$.

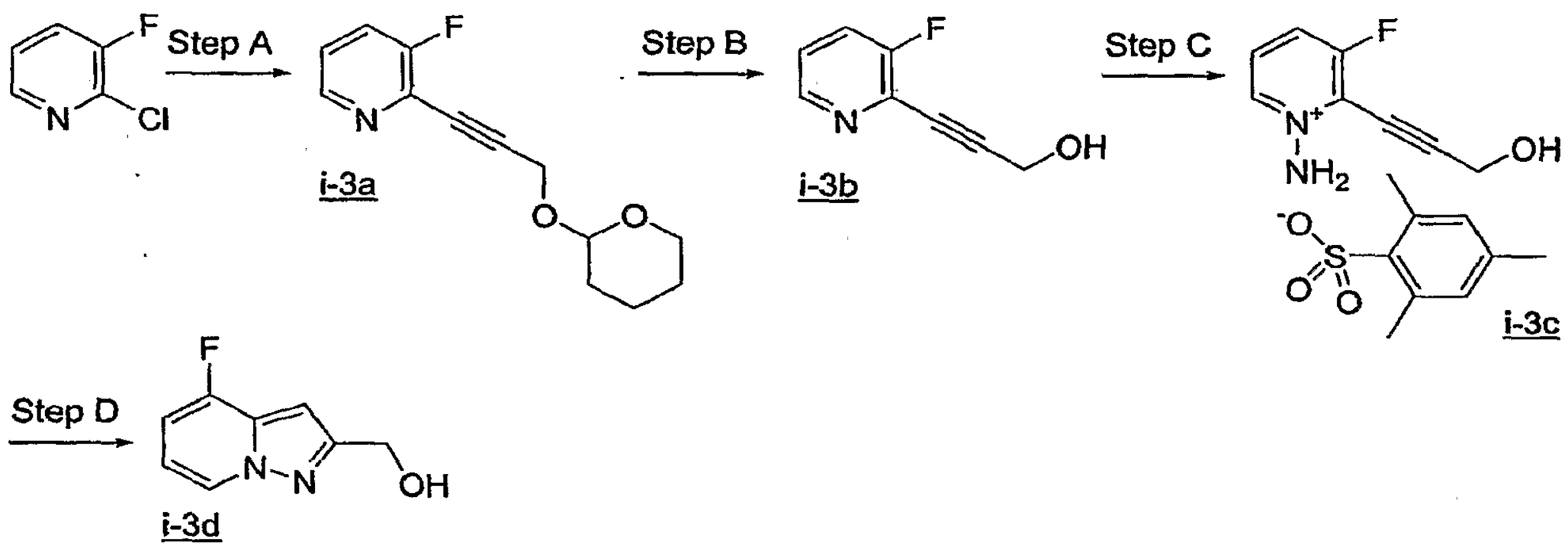
Step C: Preparation of 5,6-difluoro-2-methylquinoline (i-2c)

A mixture of i-2b (520 mg, 2.00 mmol), 2N sodium hydroxide (1.25 mL, 2.50 mmol) and palladium hydroxide on activated carbon (20%, 100 mg) was hydrogenated in EtOAc/MeOH (25 mL, 9:1) under atmospheric pressure (balloon) for 1h. The reaction mixture was filtered through Celite® and the filtrate concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 5-25% EtOAc/Hexanes as eluent) afforded the title compound i-2c, *m/z* (ES) 180 $(MH)^+$.

Step D: Preparation of 2-(bromomethyl)-5,6-difluoroquinoline (i-2d)

N-Bromosuccinimide (399 mg, 2.20 mmol) followed by benzoyl peroxide (50.0 mg) were added to a stirred solution of i-2c (300 mg, 1.68 mmol) in carbon terachloride (20 mL) at room temperature. The resulting mixture was heated to 76 °C and stirred for 3 h. After cooling to room temperature, the reaction mixture was filtered and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 5-15% EtOAc/Hexanes as eluent) provided the title compound i-2d, *m/z* (ES) 260 $(MH)^+$. ^1H NMR (500 MHz, $CDCl_3$): δ 4.71 (s, 2H), 7.60 (dd, J = 8.1 Hz, 9.6 Hz, 1H), 7.67 (d, J = 8.6 Hz, 1H), 7.88 (m, 1H), 8.45 (d, J = 8.6 Hz, 1H).

Scheme i-3



Preparation of (4-fluoropyrazolo[1,5-a]pyridin-2-yl)methanol (i-3d)

Step A: Preparation of 3-fluoro-2-[3-(tetrahydro-2*H*-pyran-2-yloxy)prop-1-yn-1-yl]pyridine (i-3a)

Tributyl[3-(tetrahydro-2H-pyran-2-yloxy)prop-1-yn-1-yl]stannane (13.8 g, 32.0 mmol, prepared according to Kyler, et al., *J. Org. Chem.*, **1987**, *52*, 4296-4298) and bis(triphenylphosphine)-palladium(II)chloride (4.92 g, 6.98 mmol) were added successively to a stirred solution of 2-chloro-3-fluoropyridine (6.32 g, 48.1 mmol) in dioxane (100 mL) at room temperature. The resulting mixture was

degassed with a gentle stream of nitrogen for 10 min, and then heated to 100 °C for approximately 6 h. After cooling to room temperature, the reaction mixture was quenched with sat. aq. potassium fluoride and diluted with EtOAc. After stirring vigorously for approximately 15 min, the precipitated solids were removed *via* filtration. The organic phase was separated from the filtrate, washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 10-60% EtOAc/Hexanes) gave the title compound i-3a, *m/z* (ES) 236 $(MH)^+$.

Step B: Preparation of 3-(3-fluoropyridin-2-yl)prop-2-yn-1-ol (i-3b)

A stirred solution of i-3a (2.20 g, 9.35 mmol) in acetic acid /water (95 mL/15 mL) was heated at 40 °C for 8 h. After cooling to room temperature, the volatiles were removed *in vacuo* and the residue was partitioned between EtOAc and sat. aq. sodium bicarbonate. The organic phase was separated and the aq. phase was re-extracted three times with EtOAc. The combined organic extacts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 10-80% EtOAc/Hexanes) gave the title compound i-3b, *m/z* (ES) 134 $(MH)^+$-H_2O.

Step C: Preparation of 1-amino-3-fluoro-2-(3-hydroxyprop-1-yn-1-yl)pyridinium 2,4,6-trimethylbenzenesulfonate (i-3c)

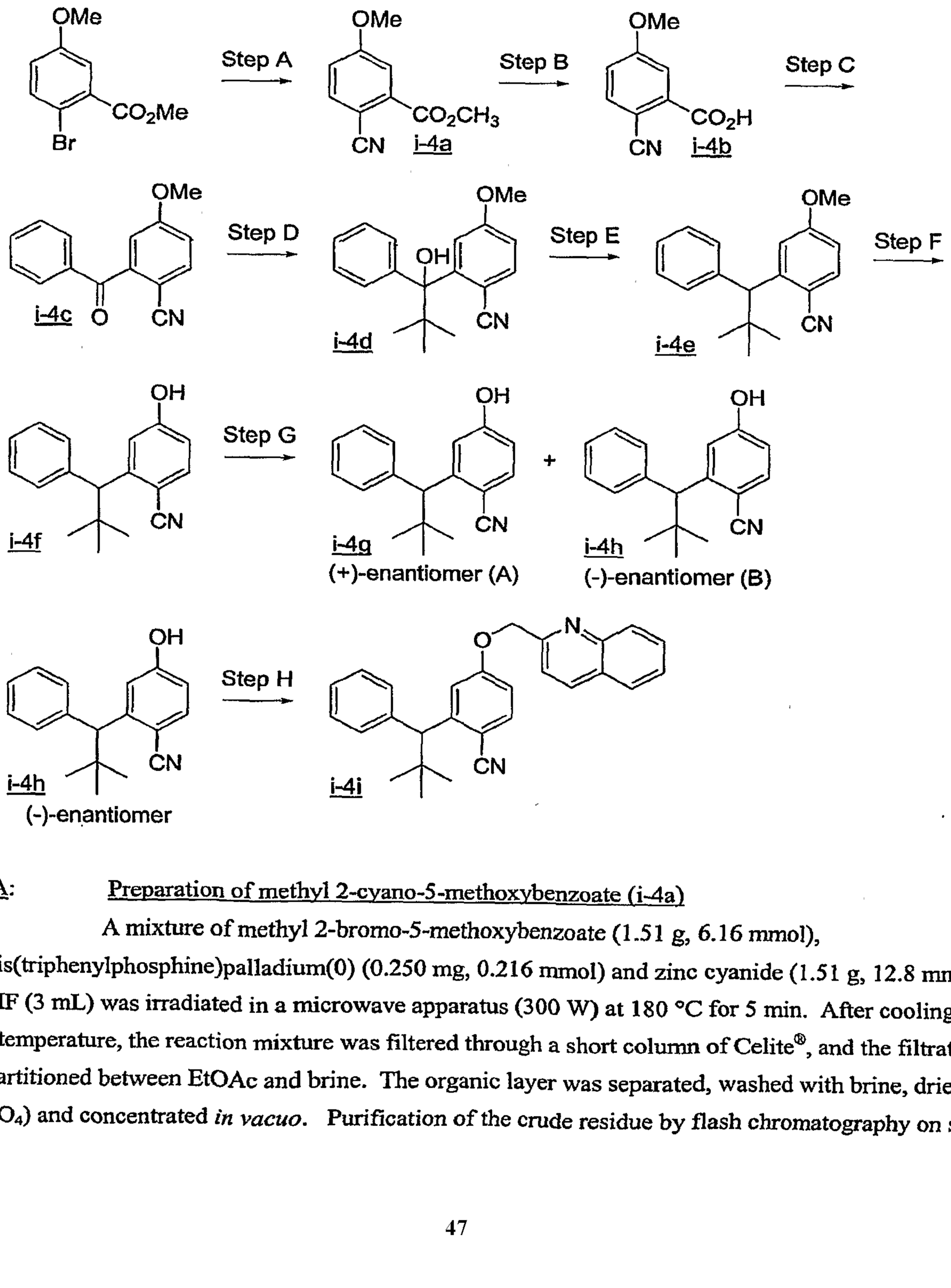
A solution of 2-[(aminooxy)sulfonyl]-1,3,5-trimethylbenzene (1.15 g, 5.30 mmol, prepared according to Tamura, et al., *Synthesis*, **1977**, 1-17) in DCM (15 mL) was added dropwise *via* syringe to a stirred solution of i-3b (536 mg, 3.55 mmol) in DCM (15 mL) at 0 °C. After 2 h, the reaction mixture was warmed to room temperature, aged for 10 min and then diluted with ether (30 mL). The precipitated crystals were collected *via* filtration and dried *in vacuo* to give the title compound i-9c. ^1H NMR (500 MHz, CD_3OD): δ 2.01 (s, 3H), 2.60 (s, 6H), 4.62 (s, 2H), 6.82 (s, 2H), 7.91 (m, 1H), 8.19 (m, 1H), 8.64 (d, J = 8.2 Hz, 1H).

Step D: Preparation of (4-fluoropyrazolo[1,5-*a*]pyridin-2-yl)methanol (i-3d)

Potassium carbonate (340 mg, 2.46 mmol) was added to a stirred solution of i-3c (450 mg, 1.23 mmol) in DMF (10 mL) at room temperature. After 18 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extacts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 20-60% EtOAc/Hexanes) gave the title compound i-3d. ^1H NMR (500 MHz, $CDCl_3$): 4.92 (s, 2H), 6.06 (m, 1H), 6.61 (s, 1H), 6.80 (dd, J =8.2Hz, 8.1 Hz, 1H), 8.64 (d, J = 8.2 Hz, 1H); *m/z* (ES) 149 $(MH)^+$-H_2O.

Following procedures similar to that described above for intermediate i-3d, (5-fluoropyrazolo[1,5-a]pyridin-2-yl)methanol, (6-fluoropyrazolo[1,5-a]pyridin-2-yl)methanol, and (7-fluoropyrazolo[1,5-a]pyridin-2-yl)methanol were prepared, starting from 2-bromo-4-fluoropyridine, 2-bromo-5-fluoropyridine, and 2-bromo-6-fluoropyridine, respectively.

Scheme i-4



Step A: Preparation of methyl 2-cyano-5-methoxybenzoate (i-4a)

A mixture of methyl 2-bromo-5-methoxybenzoate (1.51 g, 6.16 mmol), tetrakis(triphenylphosphine)palladium(0) (0.250 mg, 0.216 mmol) and zinc cyanide (1.51 g, 12.8 mmol) in DMF (3 mL) was irradiated in a microwave apparatus (300 W) at 180 °C for 5 min. After cooling to room temperature, the reaction mixture was filtered through a short column of Celite®, and the filtrate was partitioned between EtOAc and brine. The organic layer was separated, washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica

gel (gradient elution, 0-35% EtOAc/hexanes as eluent) afforded the title compound i-4a, *m/z* (ES) 192 $(MH)^+$.

Step B: Preparation of 2-cyano-5-methoxybenzoic acid (i-4b)

Lithium hydroxide (8.60 g, 358 mmol) was added to a stirred solution of i-4a (9.79 g, 51.2 mmol) in THF/water (1:1, 200 mL) at room temperature. After 5 h, the reaction mixture was poured into 1 N HCl and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo* to give the title compound i-4b, *m/z* (ES) 178 $(MH)^+$.

Step C: Preparation of 2-benzoyl-4-methoxybenzonitrile (i-4c)

1,1'-Carbonyldiimidazole (7.01 g, 43.2 mmol) was added to a stirred solution of i-4b (5.04 g, 28.5 mmol) in THF/DCM (1:1, 60 mL) at room temperature. After 2 h, the reaction mixture was cooled to -78 °C, and phenylmagnesium bromide (120 mL of a 1 M solution in THF, 0.120 mmol) was added dropwise *via* cannula. After 1.5 h, the reaction mixture was quenched with sat. aq. ammonium chloride and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-35% EtOAc/hexanes as eluent) afforded the title compound i-4c, *m/z* (ES) 238 $(MH)^+$.

Step D: Preparation of 2-(1-hydroxy-2,2-dimethyl-1-phenylpropyl)-4-methoxybenzonitrile (i-4d)

Tert-Butylmagnesium bromide (18.0 mL of a 1 M solution in THF, 18.0 mmol) was added to a stirred solution of i-4c (1.89 g, 7.99 mmol) in THF (30 mL) at 0 °C. After 3 h, the reaction mixture was quenched with sat. aq. ammonium chloride and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-50% EtOAc/hexanes as eluent) afforded the title compound i-4d, *m/z* (ES) 296 $(MH)^+$.

Step E: Preparation of 2-(2,2-dimethyl-1-phenylpropyl)-4-methoxybenzonitrile (i-4e)

Ammonium formate (5.17 g, 82.1 mmol) followed by palladium (430 mg of 10 wt. % on activated carbon) were added to a stirred solution of i-4d (1.02 g, 3.47 mmol) in acetic acid (10 mL). The reaction mixture was heated to 110 °C and stirred for 2 h. After cooling to room temperature, the reaction mixture was filtered through a short column of Celite® and concentrated *in vacuo*. The residue was dissolved in DMF (10 mL) and cyanuric chloride (0.911 g, 4.95 mmol) was added. After 2 h, the reaction mixture was quenched with sat. aq. sodium bicarbonate and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*.

Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-15% EtOAc/hexanes as eluent) afforded the title compound i-4e (0.809 g), *m/z* (ES) 280 $(MH)^+$.

Step F: Preparation of 2-(2,2-dimethyl-1-phenylpropyl)-4-hydroxybenzonitrile (i-4f)

Boron tribomide (15.0 mL of a 1 M solution in DCM, 15.0 mmol) was added to a stirred solution of i-4e (0.805 g, 2.91 mmol) in DCM (10 mL) at 0 °C. The reaction mixture was allowed to warm to room temperature and aged for approximately 12 h. The reaction mixture was poured into sat. aq. sodium bicarbonate and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-20% EtOAc/hexanes as eluent) afforded the title compound i-4f.

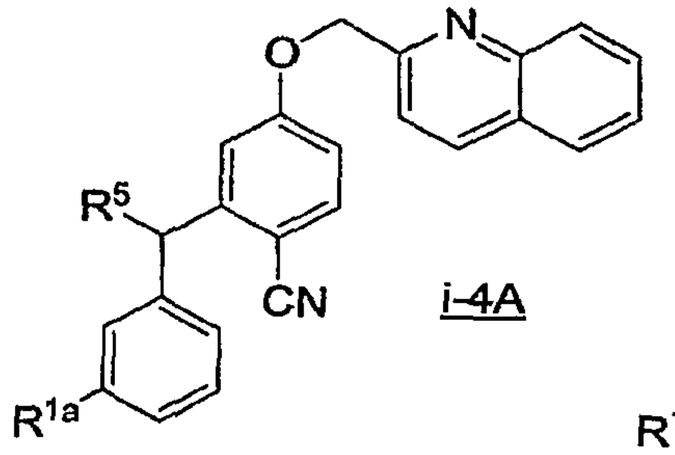
Step G: Preparation of (+)-2-(2,2-dimethyl-1-phenylpropyl)-4-hydroxybenzonitrile (i-4g) and (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-hydroxybenzonitrile (i-4h)

i-4f (1.80g) was resolved into its enantiomeric components by preparative chiral HPLC (ChiralCel™ *OJ* column, 30% isopropyl alcohol/heptane as eluent) to provide in order of elution:
i-4g (Enantiomer A): $[\alpha]_D^{20} = +2.4°$ (C = 1.00, MeOH); Retention time = 6.83' on analytical ChiralCel™ column (4.6 x 250 mm; 10 micron, flow rate = 0.5 mL/min, λ = 254 nM UV detection); ^{1}H-NMR (500 MHz, $CDCl_3$): δ 1.11 (s, 9H), 4.34 (s, 1H), 6.2-6.4 (br s, 1H), 6.77 (dd, *J* = 8.5 Hz, 2.2 Hz, 1H), 7.23-7.27 (m, 1H), 7.29-7.34 (m, 3H), 7.47 (m, 2H), 7.50 (d, *J* = 8.5 Hz, 1H).
i-4h (Enantiomer B): $[\alpha]_D^{20} = -2.4°$ (C = 0.20, MeOH)], Retension time = 9.74' on analytical ChiralCel™ column (4.6 x 250 mm; 10 micron, flow rate = 0.5 mL/min, λ = 254 nM UV detection); ^{1}H-NMR (500 MHz, $CDCl_3$): δ 1.11 (s, 9H), 4.34 (s, 1H), 6.2-6.4 (br s, 1H), 6.77 (dd, *J* = 8.5 Hz, 2.2 Hz, 1H), 7.23-7.27 (m, 1H), 7.29-7.34 (m, 3H), 7.47 (m, 2H), 7.50 (d, *J* = 8.5 Hz, 1H).

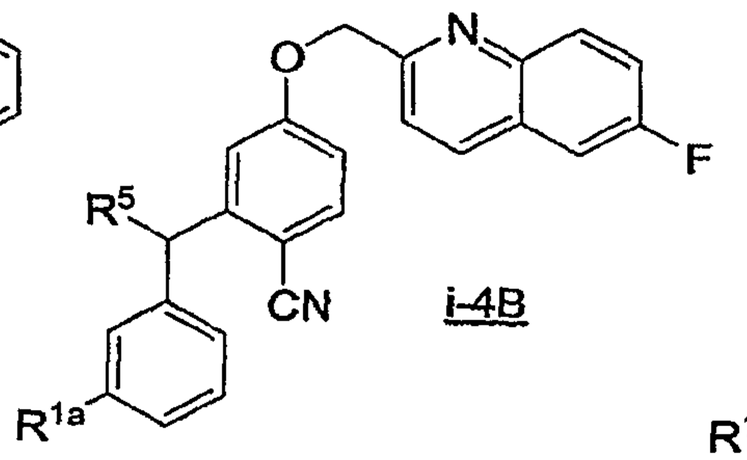
Step H: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzonitrile (i-4i)

Cesium carbonate (2.17 g, 8.21 mmol) followed by 2-(chloromethyl)quinoline hydrochloride (0.711 g, 3.32 mmol) were added to a stirred solution of i-4h (0.584 g, 2.20 mmol) in DMF (10 mL) at room temperature. After 12 h, the reaction mixture was poured into water (50 mL) and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-20% EtOAc/hexanes as eluent) afforded the title compound i-4i, *m/z* (ES) 407 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 1.04 (s, 9H), 4.31 (s, 1H), 5.50 (s, 2H), 6.97 (dd, *J* = 8.5 Hz, 2.5 Hz, 1H), 7.09-7.31 (m, 5H), 7.49 (d, *J* = 2.5 Hz, 1H), 7.53 (d, *J* = 8.5 Hz, 1H), 7.63-7.65 (m, 2H), 7.82 (t, *J* = 7.5 Hz, 1H), 7.90 (d, *J* = 8.0 Hz, 1H), 8.17 (d, *J* = 8 Hz, 1H), 8.23 (d, *J* = 8.5 Hz, 1H).

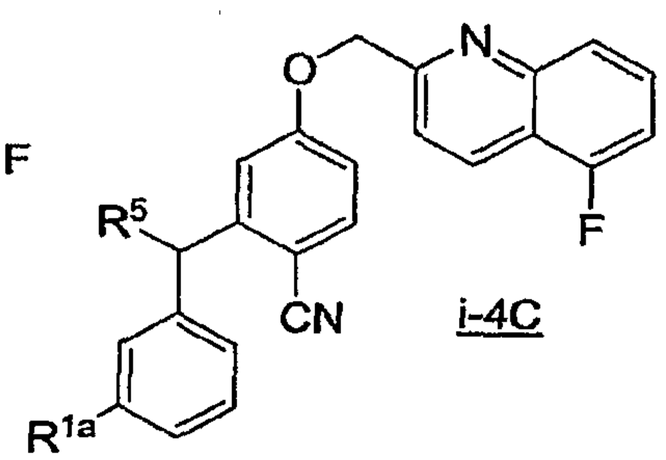
Following procedures similar to that described above for intermediate i-4i, the following additional intermediates can be prepared:



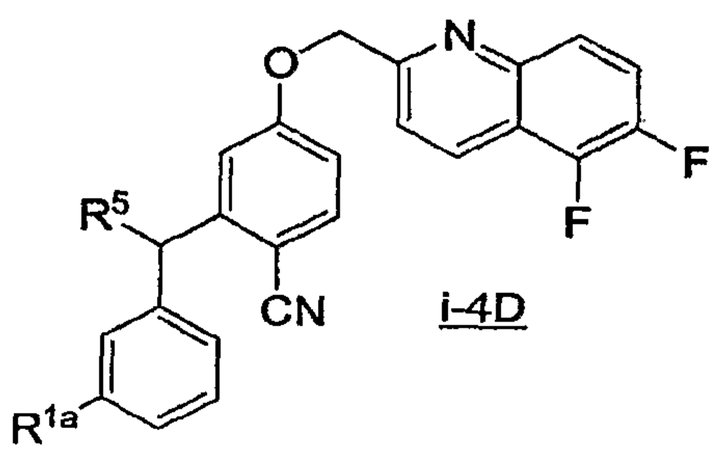
i-4A



i-4B

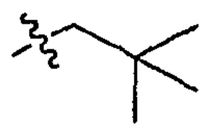
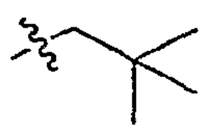


i-4C

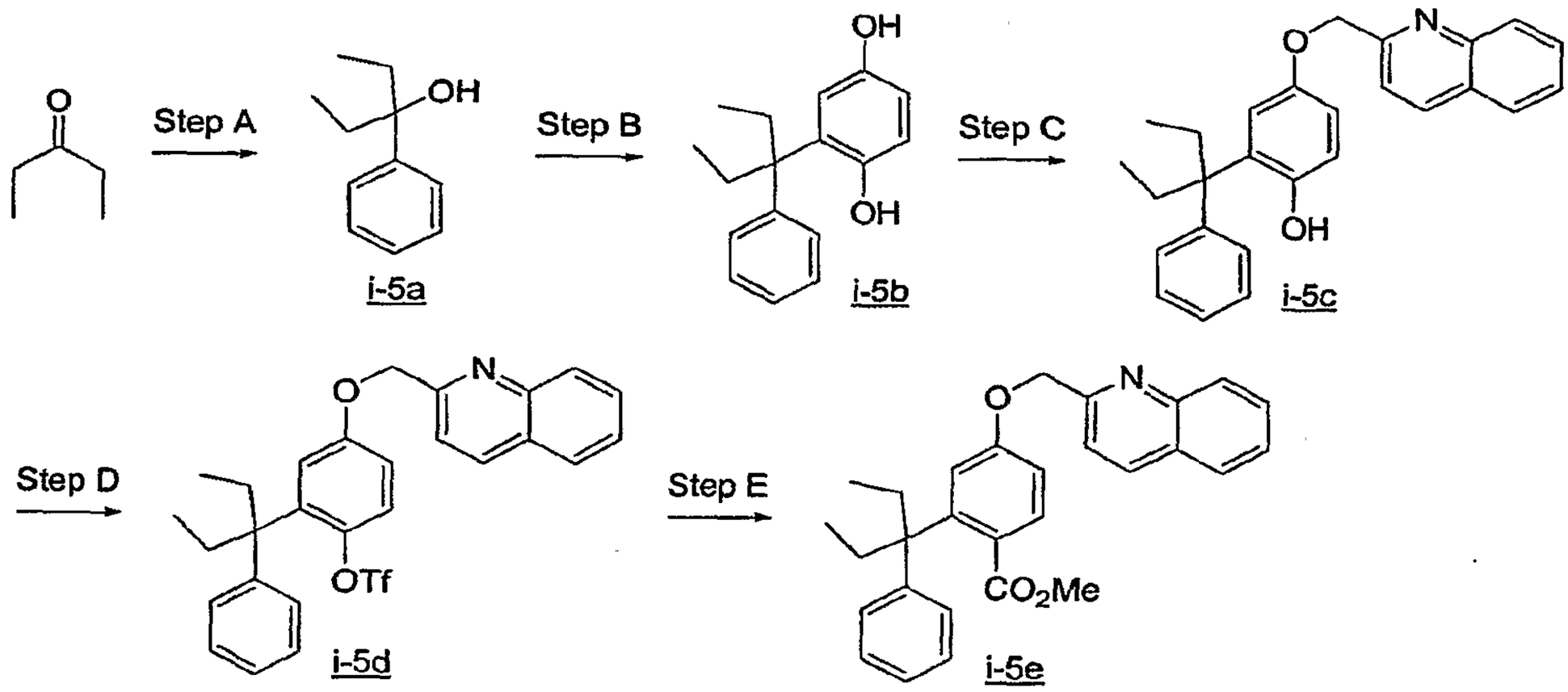


i-4D

Ex. #i-4A	Ex. #i-4B	Ex. #i-4C	Ex. #i-4D	R^5	R^{1a}
a	a	a	a	Me	H
b	b	b	b	Me	F
c	c	c	c	Et	H
d	d	d	d	Et	F
e	e	e	e	iso-Pr	H
f	f	f	f	iso-Pr	F
g	g	g	g	cyc-Pr	H
h	h	h	h	cyc-Pr	F
-	i	i	i	tert-Bu	H
j	j	j	j	tert-Bu	F
k	k	k	k	cyc-Bu	H

l	l	l	l	cyc-Bu	F
m	m	m	m		H
n	n	n	n		F

Scheme i-5



Methyl 2-(1-ethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzoate (i-5e)

Step A: Preparation of 3-phenylpentan-3-ol (i-5a)

3-Pentanone (5.00 g, 58.0 mmol) was added dropwise to a solution of phenylmagnesium bromide (29.0 mL of a 3 M solution in ether, 87.0 mmol) in diethyl ether (250 mL) at 0 °C. After completion of addition, the reaction mixture was allowed to warm to room temperature and aged for approximately 12 h. The reaction mixture was poured into sat. aq. ammonium chloride and extracted three times with EtOAc. The combined organic extracts were dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 0-20% EtOAc/hexanes) gave the title compound i-5a, ^1H NMR (500 MHz, $CDCl_3$): δ 0.8 (t, J = 7 Hz, 6H), 1.68 (s, 1H), 1.88 (m, 4H), 7.40 (m, 4H), 7.25 (m, 1H).

Step B: Preparation of 2-(1-ethyl-1-phenylpropyl)benzene-1,4-diol (i-5b)

A solution of hydroquinone (1.35 g, 12.3 mmol) and *p*-TSA monohydrate (116 mg, 0.61 mmol) in toluene (20 mL) was heated at 110 °C for 15 min with azeotropic removal of water using a Dean-Stark apparatus. A solution of i-5a (1.00 g, 6.09 mmol) in toluene (4 mL) was added to the above solution over a period of 6 h *via* syringe pump addition, and the resulting mixture was stirred at 110 °C for an additional 12 h. After cooling to room t the reaction mixture was poured into water

and extracted three times with EtOAc. The combined organic extracts were dried ($MgSO_4$) and concentrated *in vacuo*. During evaporation of the organic phase, the precipitated excess hydroquinone was removed by filtration. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 0-30% EtOAc/hexanes) gave the title compound i-5b, ^{1}H NMR (500 MHz, $CDCl_3$): δ 0.65 (t, J = 7.5 Hz, 6H), 2.05 (dq, J = 13.5, 7.6 Hz, 2H), 2.24 (dq, J = 13.5, 7.6 Hz, 2H), 4.01 (s, 1H), 4.58 (s, 1H), 6.66 (m, 2H) 7.00 (d, J = 2.9 Hz, 1H), 7.34 (m, 5H).

Step C: Preparation of 2-(1-ethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)phenol (i-5c).

2-(Chloromethyl)quinoline (268 mg, 1.51 mmol), potassium iodide (250 mg, 1.51 mmol) and potassium carbonate (321 mg, 2.32 mmol) were added sequentially to a stirred solution of i-5b (298 mg, 1.16 mmol) in DMF (1.7 mL) at room temperature. After 12 h, the reaction mixture was poured into water and the aq. phase was adjusted to pH7 by the addition of 1 N HCl. The aq. layer was extracted three times with EtOAc and the combined organic extracts were dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 0-30% EtOAc in hexanes as eluent) afforded the title compound i-5c, *m/z* (ES) 398 $(MH)^+$.

Step D: Preparation of 2-(1-ethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)phenyl trifluoromethanesulfonate (i-5d).

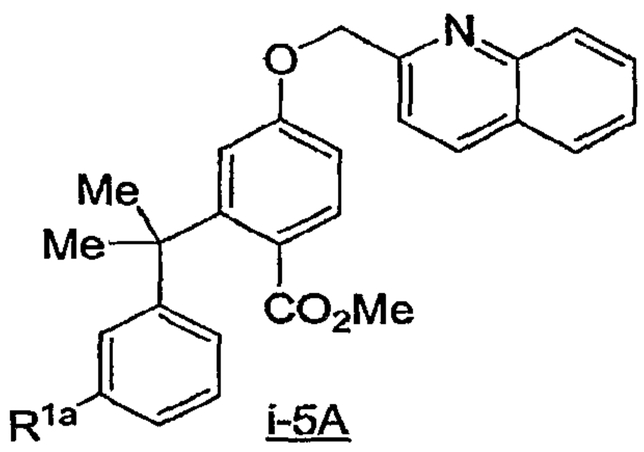
Sodium hydride (10.0 mg, 0.230 mmol) was added to a stirred solution of i-5c in THF (2.7 mL) at 0 °C. After 20 min, 2-[*N*,*N*-bis(trifluoromethylsulfonyl)amino]-5-chloropyridine (116 mg, 0.290 mmol) was added and the resulting mixture was warmed to room temperature. After 1 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution: 0-30% EtOAc/hexanes as eluent) furnished the title compound i-5d, *m/z* 530 $(MH)^+$.

Step E: Preparation of methyl 2-(1-ethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzoate (i-5e).

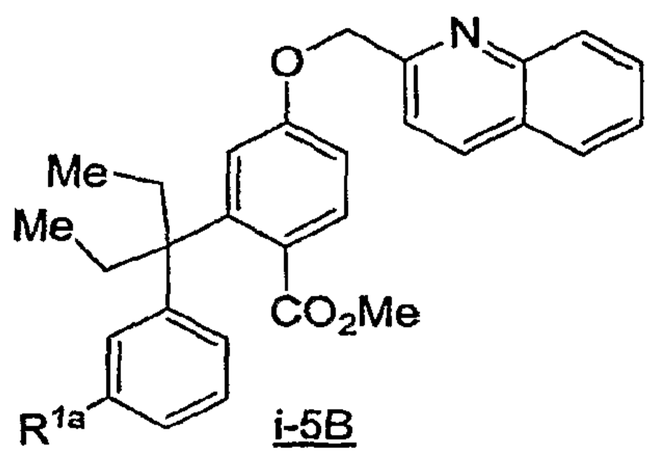
A stirred mixture of i-5d (70.0 mg, 0.130 mmol), palladium(II) acetate (4.70 mg, 0.02 mmol), 1,1'-bis(diphenylphosphino)ferrocene (16.0 mg, 0.030 mmol) and triethylamine (44 μL, 0.32 mmol) in MeOH/DMF (1:1, 2.0 mL) was purged with carbon monoxide for 10 min and then heated at 80 °C for approximately 18 h. The reaction mixture was filtered through a short column of Celite®, washing with EtOAc. The filtrate was concentrated *in vacuo* and purified by flash chromatography on silica gel (gradient elution: 0-30% EtOAc/hexanes) to yield the title compound i-5e, *m/z* 440 $(MH)^+$. ^{1}H NMR (500 MHz, $CDCl_3$): δ 0.60 (t, J = 7.6 Hz, 6H), 2.00 (dq, J = 13.0 Hz, 5.8 Hz, 2H), 2.47 (dq, J = 13.0 Hz,, 5.8 Hz, 2H), 3.1 (s, 3H), 5.5 (s, 2H), 6.9 (dd, J = 8.8 Hz, 2.5 Hz, 1H), 7.1 (d, J = 7.9 Hz, 2H),

7.15 (t, J = 6.6 Hz, 1H), 7.24 (m, 3H), 7.34 (d, J = 2.8 Hz, 1H), 7.61 (t, J = 7.2 Hz, 1H), 7.74 (d, J = 7.9 Hz, 1H), 7.79 (t, J = 7.9 Hz, 1H), 7.89 (d, J = 7.9 Hz, 1H), 8.14 (s, 1H), 8.26 (d, J = 7.9 Hz, 1H).

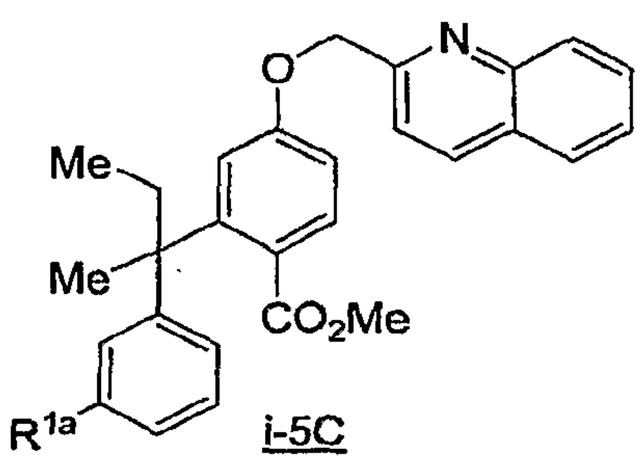
Following procedures similar to that described above for intermediates i-5e, the following additional intermediates can be prepared:



i-5A



i-5B



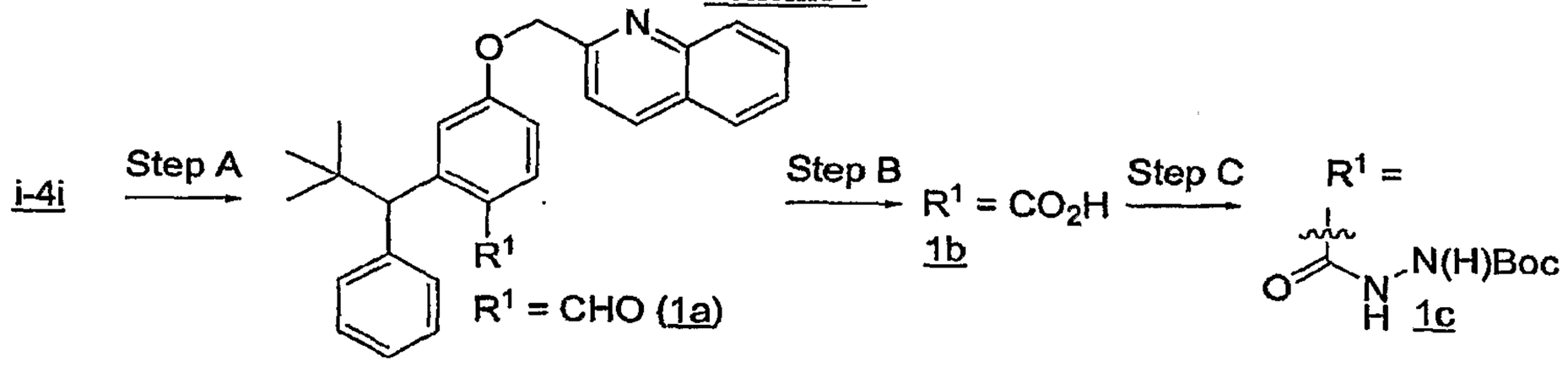
i-5C

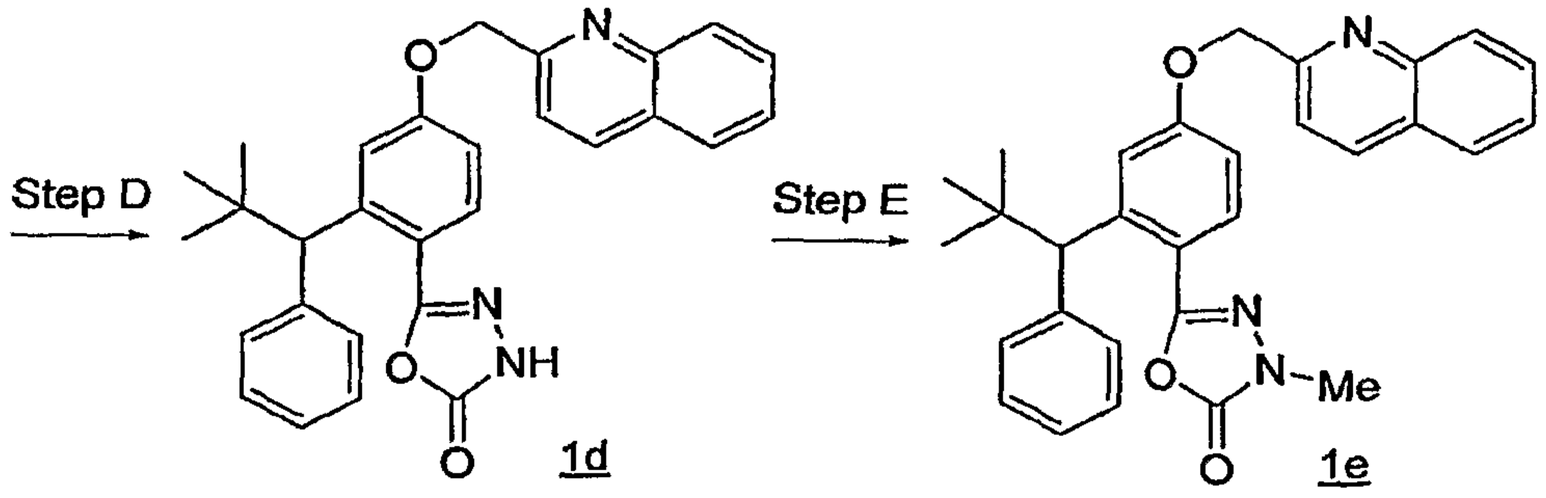
Ex. #i-5A	Ex. #i-5B	Ex. #i-5C	R^{1a}
a	-	a	H
b	b	b	F

In the Tables in the following Examples, compounds having mass spectral data were synthetically prepared.

Example 1

Scheme 1





Step A: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzaldehyde (1a)

DIBAL-H (9.0 mL of a 1 M solution in toluene, 9.00 mmol) was added to a solution of i-4i (900 mg, 2.20 mmol) in DCM (20 mL) at -78 °C. After 10 min, wet silica gel (excess) was added to quench the reaction. The resulting mixture was stirred at room temperature for 30 min, filtered and the residue washed with EtOAc. The filtrate was washed with water, brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-20% EtOAc/hexanes as eluent) afforded the title compound 1a, *m/z* (ES) 410 $(MH)^+$.

Step B: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzoic acid (1b)

2,3-Dimethyl-2-butene (1.30 mL of a 1 M solution in THF, 13.0 mmol), NaH_2PO_4 (956 mg, 7.96 mmol) and $NaClO_2$ (900 mg (80%), 7.96 mmol) were added sequentially to a stirred solution of 1a (538 mg, 1.13 mmol) in *t*-BuOH/water (20:8 mL) at 0 °C. After 1.5 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-5% MeOH/DCM as eluent) afforded the title compound 1b, *m/z* (ES) 426 $(MH)^+$.

Step C: Preparation of (-)-*tert*-Butyl-2-[2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)benzoyl]hydrazinecarboxylate (1c)

HATU (899 mg, 2.36 mmol), *t*-butyl carbazate (784 mg, 5.93 mmol) and DIPEA (1.05 mL, 5.93 mmol) were added sequentially to a stirred solution of 1b (501 mg, 1.18 mmol) in DMF (10 mL) at room temperature. After 12 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-50% EtOAc/hexanes as eluent) afforded the title compound 1c, *m/z* (ES) 540 $(MH)^+$.

Step D: Preparation of (-)-5-[2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)phenyl]-1,3,4-oxadiazol-2(3*H*)-one (1d)

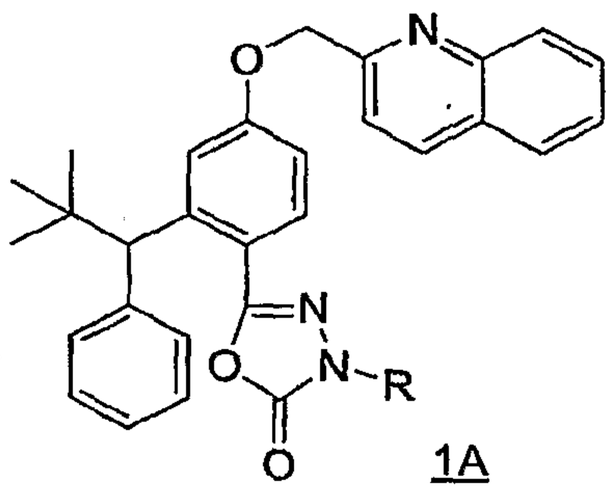
Trifluoroacetic acid (2.0 mL) was added to a stirred solution of 1c (606 mg, 1.21 mmol) in DCM (10 mL) at room temperature. After 3 h, the reaction mixture was concentrated *in vacuo*. The residue was dissolved in THF (15 mL), and Et_3N (2.0 mL, 14.6 mmol) and 1,1'-carbonyldiimidazole (606 mg, 3.74 mmol) were added. The reaction mixture was stirred at room temperature for 12 h, diluted with water and then extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-10% MeOH/DCM as eluent) afforded the title

compound 1d, *m/z* (ES) 466 $(MH)^+$. 1d.HCl ^{1}HNMR (500MHz, CD_3OD): δ 1.03 (s, 9H), 5.02 (s, 1H), 5.83 (s, 2H), 7.13-7.24 (m, 4H), 7.46 (d, *J* =7.5 Hz, 2H), 7.61 (d, *J* = 2.5 Hz, 1H), 7.71 (d, *J* = 8.5 Hz, 1H), 8.02 (t, *J* = 7.5 Hz, 1H), 8.21 (d, *J* = 7.5 Hz, 1H), 8.23 (t, *J* = 7.5 Hz, 1H), 8.39 (t, *J* = 8 Hz, 2H), 9.20 (d, *J* = 8.5 Hz, 1H).

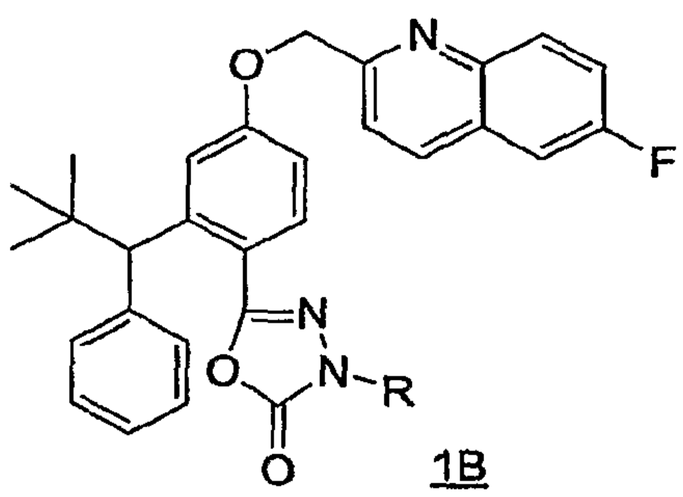
Step E: Preparation of (-)-5-[2-(2,2-dimethyl-1-phenylpropyl)-4-(quinolin-2-ylmethoxy)phenyl]-3-methyl-1,3,4-oxadiazol-2(3*H*)-one (1e)

Sodium hydride (60% in oil, 7.0 mg, 0.175 mmol) was added to a stirred solution of 1d (39.0 mg, 0.084 mmol) in DMF (2.0 mL) at 0 °C. After 10 min, methyl iodide (34.3 mg, 15 μL, 0.242 mmol) was added *via* syringe and the resulting mixture was aged for approximately 2 h. The reaction mixture was quenched with sat. aq. ammonium chloride, poured into water and extracted three times with EtOAc. The combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by preparative TLC on silica gel (30% EtOAc/hexanes as eluent) afforded 1e as an off-white foam, *m/z* (ES) 480 $(MH)^+$. 1e.HCl ^{1}HNMR (500MHz, CD_3OD): δ 1.14 (s, 9H), 3.53 (s, 3H), 5.01 (s, 1H), 5.84 (s, 2H), 7.15-7.24 (m, 4H), 7.47 (d, *J* = 7.5 Hz, 2H), 7.63 (d, *J* = 2.5 Hz, 1H), 7.72 (d, *J* = 8.5 Hz, 1H), 8.03 (t, *J* = 7.5 Hz, 1H), 8.22-8.24 (m, 2H), 8.39-8.41(m, 2H), 9.21 (d, *J* = 8.5 Hz, 1H).

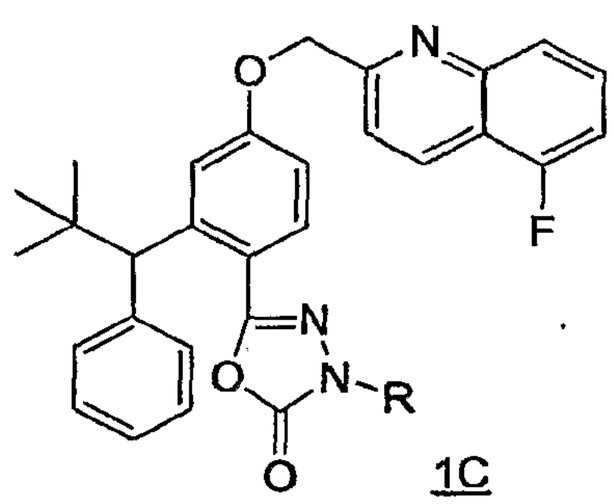
Following procedures similar to that described above for Example 1e, the following compounds can be prepared:



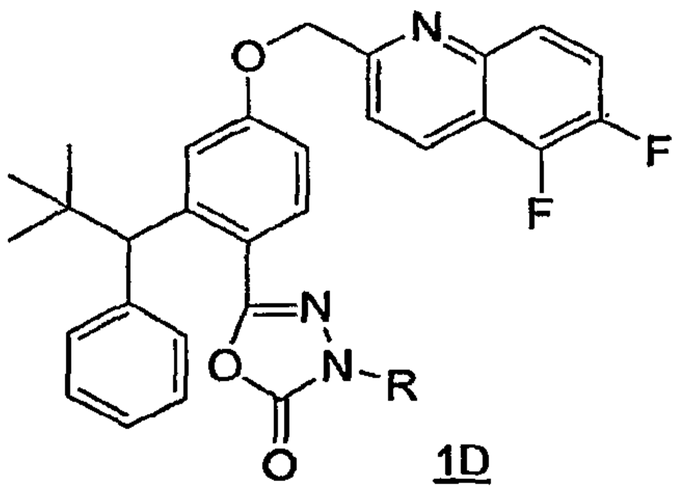
1A



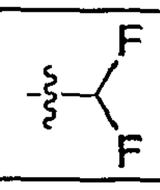
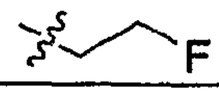
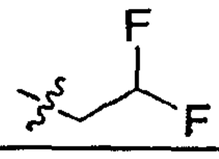
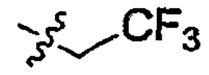
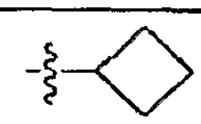
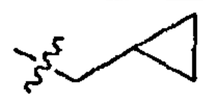
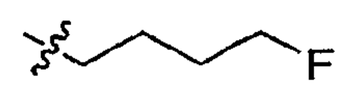
1B



1C



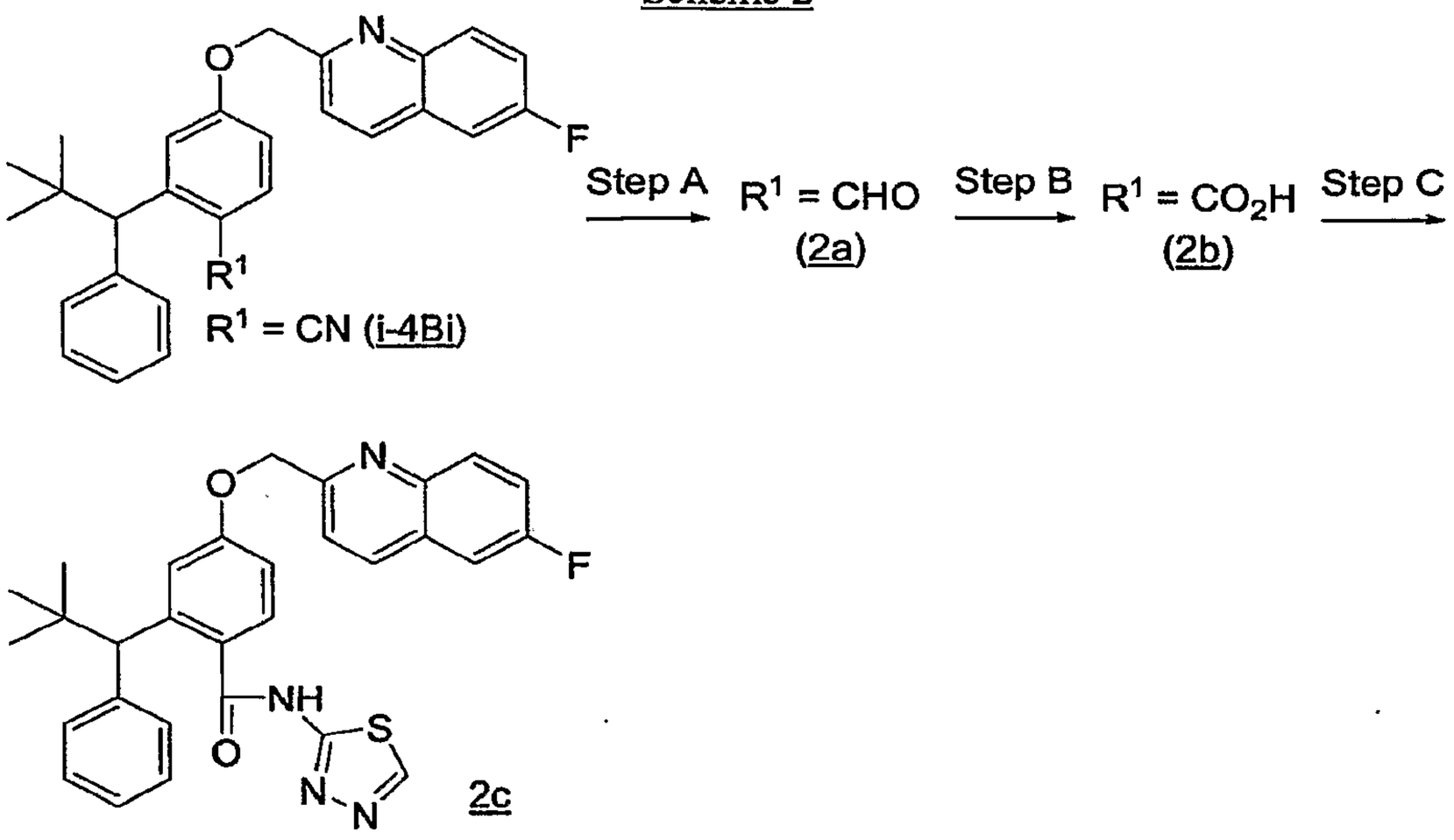
1D

Ex.#1A	Ex.#1B	Ex.#1C	Ex.#1D	R
-	a	a	a	H
-	b	b	b	Me
c	c	c	c	
d	d	d	d	Et
e	e	e	e	
f	f	f	f	
g	g	g	g	
h	h	h	h	Pr
i	i	i	i	iso-Pr
j	j	j	j	
k	k	k	k	
l	l	l	l	

Ex. #1Ba *m/z* (ES) 484 $(MH)^+$; Ex. #1Ca, *m/z* (ES) 484 $(MH)^+$; Ex. #1Da, *m/z* (ES) 503 $(MH)^+$; Ex. #1Bb, *m/z* (ES) 498 $(MH)^+$; Ex. #1Bc, *m/z* (ES) 534 $(MH)^+$; Ex. #1Bd, *m/z* (ES) 512 $(MH)^+$; Ex. #1Cd, *m/z* (ES) 512 $(MH)^+$; Ex. #1Dd, *m/z* (ES) 531 $(MH)^+$, Ex. #1Be, *m/z* (ES) 531 $(MH)^+$; Ex. #1Bf, *m/z* (ES) 548 $(MH)^+$; Ex. #1Bg, *m/z* (ES) 566 $(MH)^+$; Ex. #1Bj, *m/z* (ES) 538 $(MH)^+$; Ex. #1Bk, *m/z* (ES) 539 $(MH)^+$; Ex. #1Bl, *m/z* (ES) 558 $(MH)^+$.

Example 2

Scheme 2



Step A: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]benzaldehyde (2a)

Compound 2a can be prepared from intermediate i-4Bi following the procedure outlined in Scheme 1, Step A. Compound 2a: *m/z* (ES) 428 $(MH)^+$.

Step B: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]benzoic acid (2b)

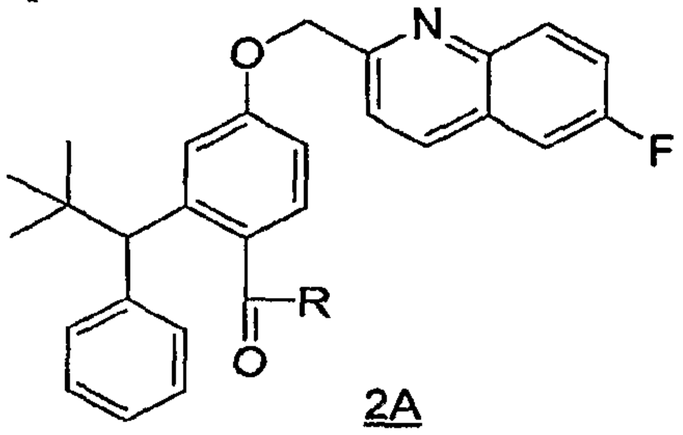
Compound 2b can be prepared from intermediate 2a following the procedure outlined in Scheme 1, Step B. Compound 2b: *m/z* (ES) 444 $(MH)^+$.

Step C: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]-*N*-1,3,4-thiadiazol-2-ylbenzamide (2c)

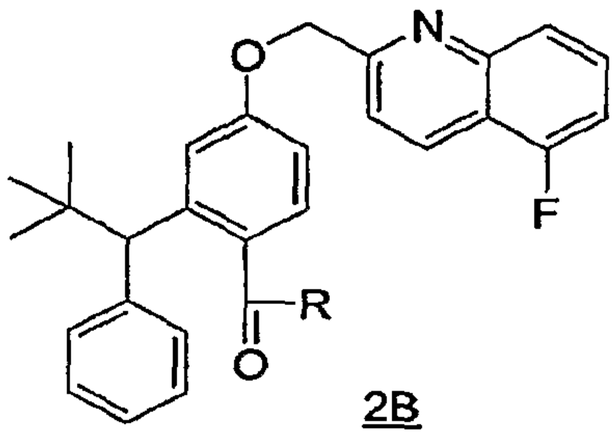
1,3,4-Thiadiazol-2-amine (17.2 mg, 0.170 mmol), HATU (55.9 mg, 0.147 mmol), DMAP (2.80 mg, 0.0230 mmol), and DIPEA (0.087 mL, 0.509 mmol) were added sequentially to a stirred solution of 2b (50.0 mg, 0.113 mmol) in DCM (1.5 mL) at room temperature. After 24 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with 5% aq. sodium bicarbonate (×3), water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 0-5% MeOH/DCM as eluent) followed by reversed phase preparative HPLC (gradient elution; 40-100% MeCN/H_2O) afforded the title compound 2c, *m/z* (ES) 527 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.94 (s, 9H), 4.65 (s, 1H), 5.43(s, 2H), 6.92 (dd, *J* = [illegible] 3 Hz, 1H), 7.15-7.15 (m, 3H), 7.30-7.38 (m,

2H), 7.44-7.48 (m, 3H), 7.51 (dt, J = 8.9 Hz, 2.7 Hz, 1H), 7.59 (d, J = 8.4 Hz, 1H), 7.64 (d, J = 8.4 Hz, 1H), 8.09 (dt, J = 9.1 Hz, 5.2 Hz, 1H), 8.12 (d, J = 8.4 Hz, 1H), 8.60 (s, 1H).

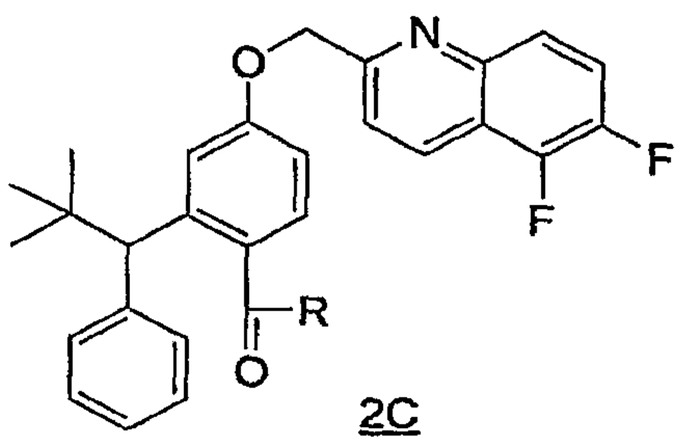
Following procedures similar to that described above for Example 2c, the following compounds can be prepared:



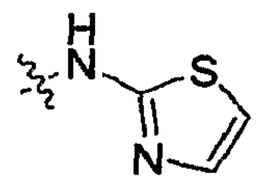
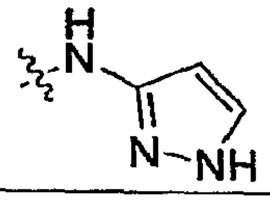
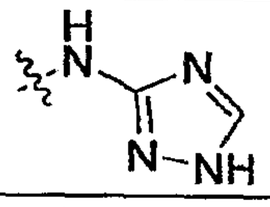
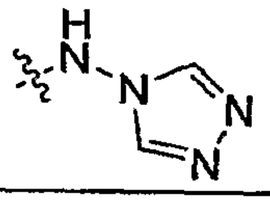
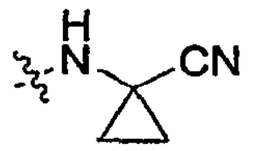
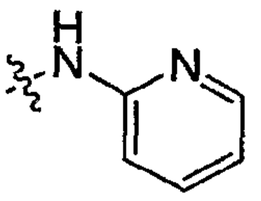
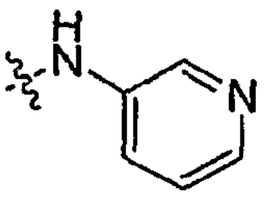
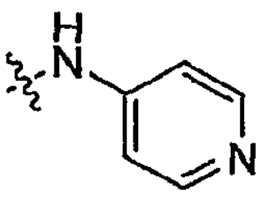
2A

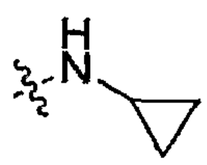
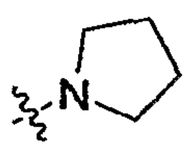
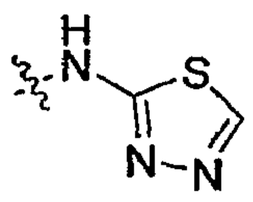


2B



2C

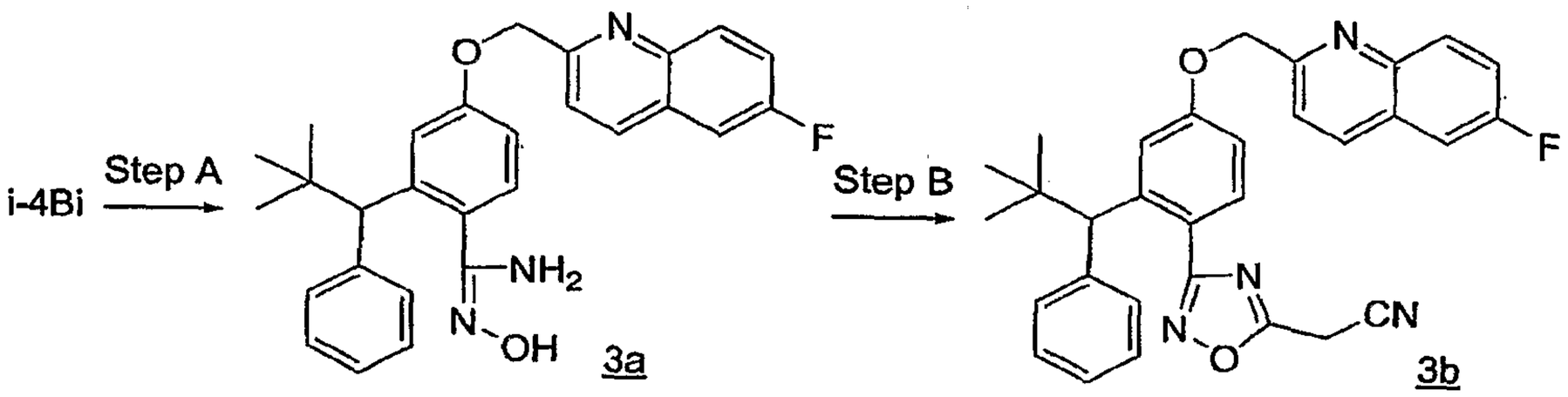
Ex. #2A	Ex. #2B	Ex. #2C	R
-	a	a	
b	b	b	
c	c	c	
d	d	d	
e	e	e	
f	f	f	
g	g	g	
h	h	h	

i	i	i	-N(H)Me
j	j	j	$-NMe_2$
k	k	k	-N(H)Et
l	l	l	$-NEt_2$
m	m	m	
n	n	n	
-	o	o	

Ex. #2Ab, *m/z* (ES) 509 $(MH)^+$; Ex. #2Ac, *m/z* (ES) 510 $(MH)^+$; Ex. #2Ad, *m/z* (ES) 510 $(MH)^+$; Ex. #2Ae, *m/z* (ES) 509 $(MH)^+$.

Example 3

Scheme 3



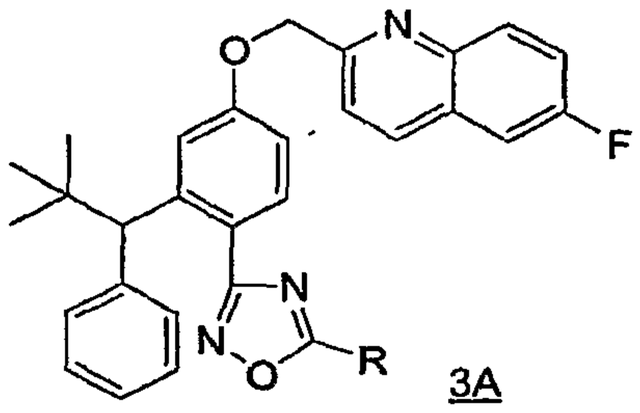
Step A: Preparation of (-)-2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]-*N*'-hydroxybenzenecarboximidamide (3a)

A thick-walled tube was charged with a solution of i-4Bi (400 mg, 0.943 mmol) in anhydrous EtOH (3.0 mL). Hydroxylamine (156 mg, 4.72 mmol, 312 μL of a 50% weight solution in water) was added and the resulting mixture was sealed and stirred at 120 °C for approximately 12 h. After cooling to room temperature, the reaction mixture was concentrated *in vacuo* and the crude residue was purified by flash chromatography on silica gel (gradient elution; 5-75% EtOAc/DCM as eluent) to provide the title compound 3a, *m/z* (ES) 458 $(MH)^+$.

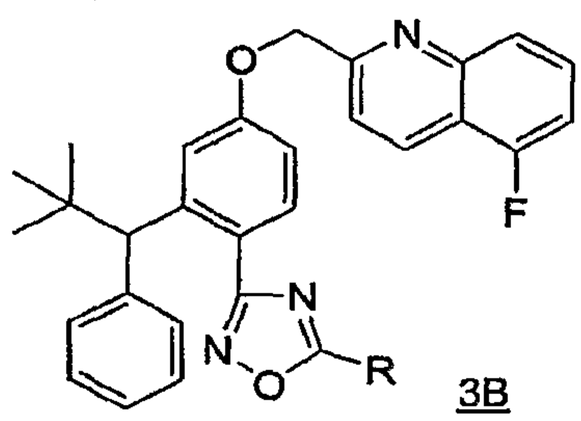
Step B: Preparation of (-)-(3-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}-1,2,4-oxadiazol-5-yl)acetonitrile (3b).

A solution of cyanoacetic acid (64.0 mg, 0.750 mmol) and dicyclohexylcarbodiimide (77.0 mg, 0.375 mmol) in DCM (0.75 mL) was stirred at room temperature for approximately 12 h. The reaction mixture was concentrated *in vacuo,* and the residue was taken up in anhydrous ether. The precipitated dicyclohexylurea was removed *via* filtration, and the filtrate was concentrated to dryness. The residue was dissolved in anhydrous pyridine (0.3 mL), and to this solution was added 3a (67.0 mg, 0.146 mmol). The resulting mixture was heated to 100 °C and aged for about 2 h. After cooling to room temperature, the reaction mixture was concentrated *in vacuo* and the crude product was purified by flash chromatography on silica gel (gradient elution; 0-20% EtOAc/DCM as eluent) providing the title compound 3b, *m/z* (ES) 508 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.95 (s, 9H), 4.13 (s, 2H), 4.97 (s, 1H), 5.48(s, 2H), 6.99 (dd, J = 8.7 Hz, 2.5 Hz, 1H), 7.10-7.15 (m, 3H), 7.34-7.38 (m, 2H), 7.49 (dd, , J = 8.7 Hz, 2.8 Hz, 1H), 7.51 (d, J = 2.6 Hz, 1H), 7.57 (dt, J = 8.9 Hz, 2.8 Hz, 1H), 7.69 (d, J = 8.4 Hz, 1H), 7.76 (d, J = 8.7 Hz, 1H), 8.13-8.19 (m, 2H).

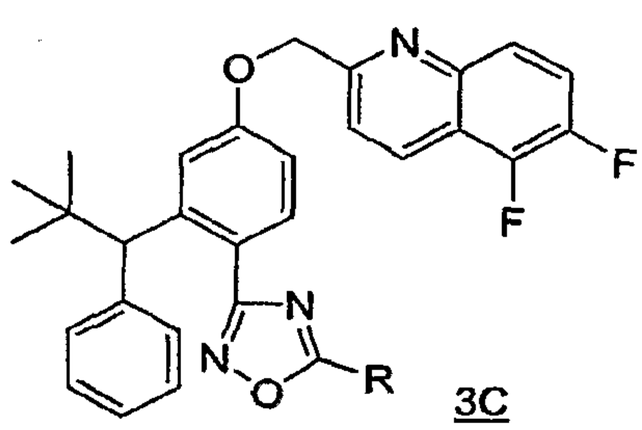
Following procedures similar to that described above for Example 3b, the following compounds can be prepared:



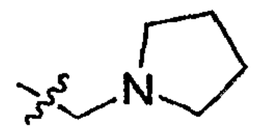
3A

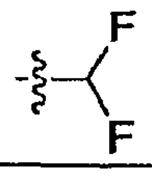
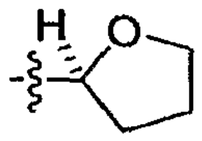
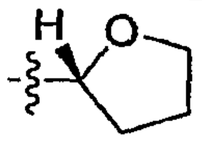
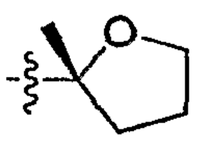
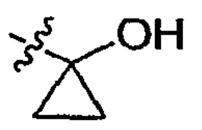
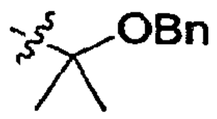
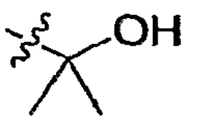


3B



3C

Ex. #3A	Ex. #3B	Ex. #3C	R
a	a	a	Me
b	b	b	CH_2F
c[a]	c[a]	c[a]	CH_2OBn
d[b]	d[b]	d[b]	CH_2OH
e	e	e	
f	f	f	CH_2CF_3

g	g	g	
h	h	h	
i	i	i	
j[c]	j[c]	j[c]	
k[b]	k[b]	k[b]	
l[d]	l[d]	l[d]	
m[b]	m[b]	m[b]	
-	n	n	CH_2CN

Ex. #3Aa, *m/z* (ES) 482 $(MH)^+$; Ex. #3Ab, *m/z* (ES) 500 $(MH)^+$; Ex. #3Ac, *m/z* (ES) 589 $(MH)^+$; Ex. #3Ad, *m/z* (ES) 498 $(MH)^+$; Ex. #3Ae, *m/z* (ES) 551 $(MH)^+$; Ex. #3Af, *m/z* (ES) 550 $(MH)^+$; Ex. #3Ag, *m/z* (ES) 518 $(MH)^+$; Ex. #3Ah, *m/z* (ES) 538 $(MH)^+$; Ex. #3Ai, *m/z* (ES) 538 $(MH)^+$; Ex. #3Aj, *m/z* (ES) 552 $(MH)^+$; Ex. #3Ak, *m/z* (ES) 524 $(MH)^+$; Ex. #3Al, *m/z* (ES) 616 $(MH)^+$; Ex. #3Am, *m/z* (ES) 526 $(MH)^+$.

[a]This compound was obtained by using the corresponding acyl chloride in Step B rather than the carboxylic acid.

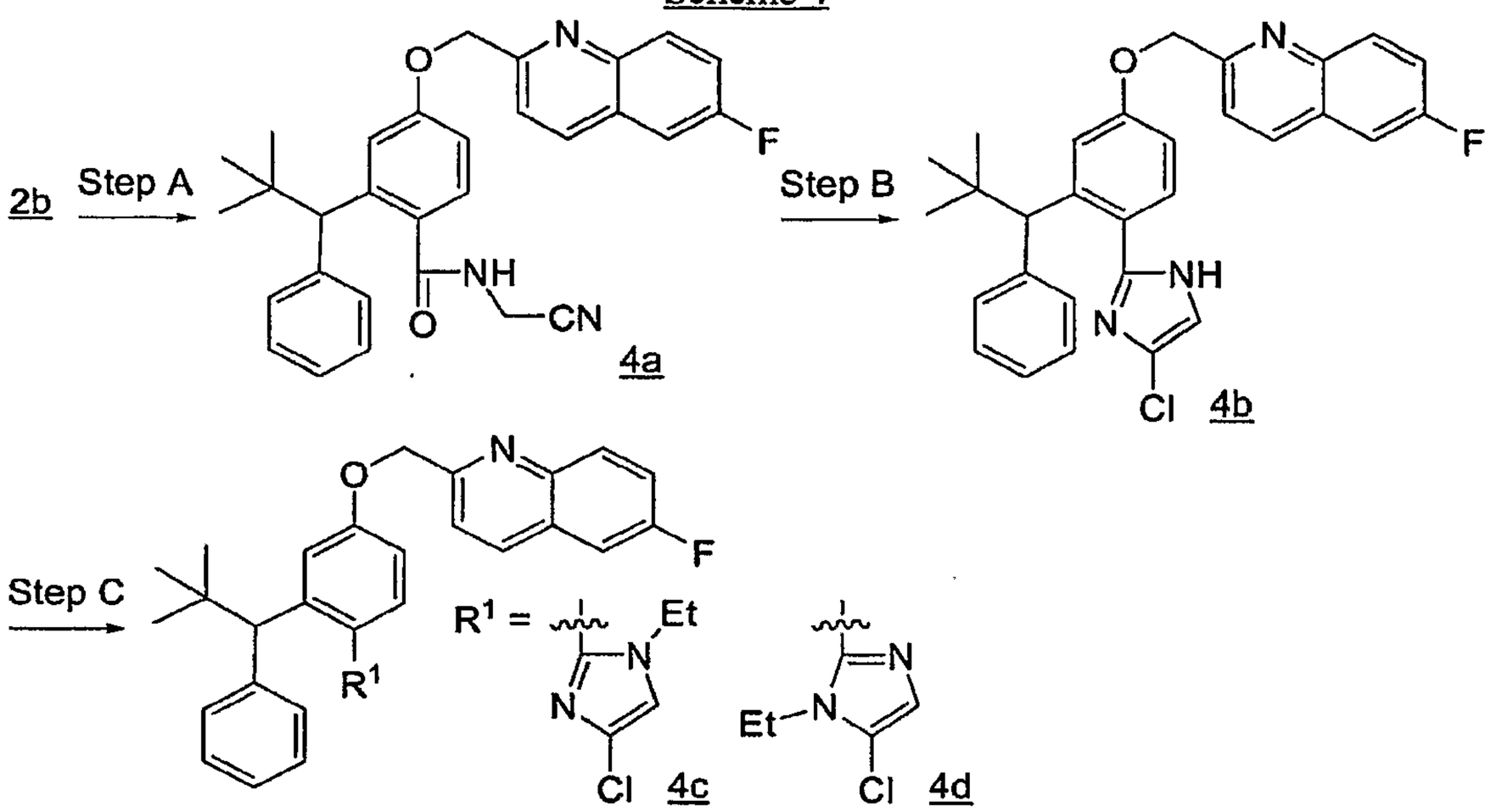
[b]The alcohol functionality was masked as a benzyl ether in Step B. Deprotection of the benzyl ether was achieved with methanesulfonic acid (according to *J. Am. Chem. Soc.*, **1996**, *118*, 4560) to provide the desired alcohol product.

[c]The acid required for Step B was prepared according to US Patent No. 6420418.

[d]The acid required for Step B was prepared by benzylating *t*-butyl 2-hydroxybutyrate (*Tetrahedron. Asym.* **2001**, *12*, 271), and subsequent ester hydrolysis (*Bioorg. Med. Chem. Lett.* **2002**, *12*,159).

Example 4

Scheme 4



Step A: Preparation of (-)-*N*-(cyanomethyl)-2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]benzamide (4a)

Aminoacetonitrile hydrochloride (21.0 mg, 0.372 mmol.), triethylamine (51.0 mg, 71 μL, 0.509 mmol), HATU (41.0 mg, 0.372 mmol), and DMAP (10.0 mg, 0.0080 mmol) were added successively to a stirred solution of 2b (150 mg, 0.339 mmol) in DCM/DMF (9:1, 1.0 mL) at room temperature. After 12h (reaction complete in typically 2 h), the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by flash chromatography on silica gel (gradient elution; 0-50% EtOAc/hexanes as eluent) to afford the title compound 4a, *m/z* (ES) 482$(MH)^+$.

Step B: Preparation of (-)-2-{[4-(4-chloro-1*H*-imidazol-2-yl)-3-(2,2-dimethyl-1-phenylpropyl)phenoxy]methyl}-6-fluoroquinoline (4b)

Triphenylphosphine (109 mg, 0.42 mmol) was added to a stirred solution of 4a (80.0 mg, 0.167 mmol) in acetonitrile (0.50 mL) at room temperature. Upon dissolution, carbon tetrachloride (64.0 mg, 0.416 mmol) was added dropwise *via* syringe. The resulting mixture was heated to 50 °C and aged for approximately 12 h. After cooling to room temperature, the volatiles were removed *in vacuo*. The residue was taken up in DCM, sat. aq. sodium bicarbonate (2.0 mL) was added, and the resulting biphasic mixture was stirred vigorously for approximately 15 min at room temperature. The organic phase was separated and the aq. phase was extracted twice with EtOAc. The combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by flash chromatography on silica gel lution; 0-50% EtOAc/hexanes as eluent) to

afford the title compound 4b, *m/z* (ES) 500 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.90 (s, 9H), 4.17 (s, 1H), 5.47 (s, 2H), 6.86-6.92 (m, 2H), 7.10-7.15 (m, 3H), 7.17-7.27 (m, 3H), 7.46-7.58 (m, 3H), 7.71 (d, J = 8.5 Hz, 1H), 8.12 (m, 1H), 8.17 (d, J = 8.5 Hz, 1H).

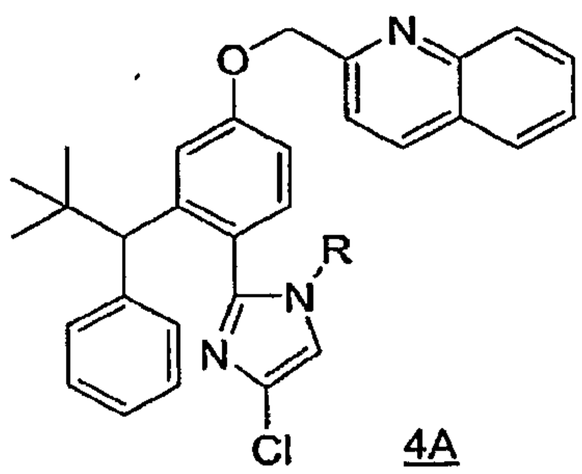
Step C: Preparation of (-)-2-{[4-(4-chloro-1-ethyl-1*H*-imidazol-2-yl)-3-(2,2-dimethyl-1-phenylpropyl)phenoxy]methyl}-6-fluoroquinoline (4c) and (-)-2-{[4-(5-chloro-1-ethyl-1*H*-imidazol-2-yl)-3-(2,2-dimethyl-1-phenylpropyl)phenoxy]methyl}-6-fluoroquinoline (4d)

Freshly ground anhydrous potassium carbonate (11.0 mg, 0.082 mmol) was added to a stirred solution of 4b (24.0 mg, 0.0480 mmol) in DMF (0.20 mL) at room temperature. After 10 min, ethyl iodide (9.70 mg, 0.0624 mmol) was added *via* syringe and the resulting mixture was stirred at room temperature overnight. The reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with water (×3), brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by flash chromatography on silica gel (gradient elution; 0-20% EtOAc/hexanes as eluent) to afford, in order of elution, 4c and 4d.

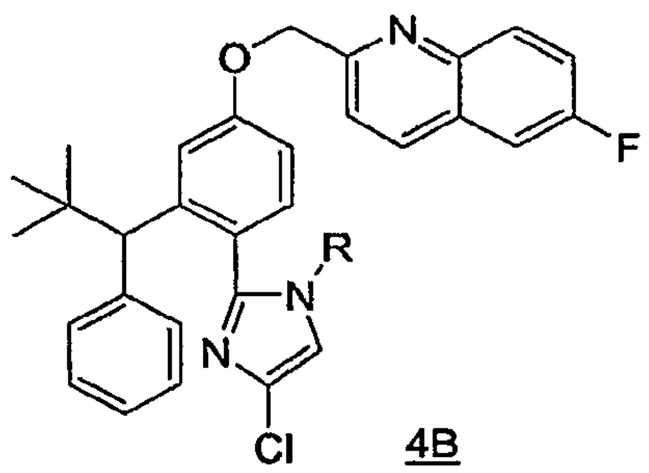
4c, *m/z* (ES) 528 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.80-1.05 (m, 12H), 2.80-3.00 (br, 2H), 3.85 (s, 1H), 5.48 (s, 2H), 6.81 (s, 1H), 6.91-6.98 (m, 3H), 7.05-7.12 (m, 4H), 7.50 (dd, J = 8.6 Hz, 2.7 Hz, 1H), 7.55 (dt, , J = 8.9 Hz, 2.8 Hz, 1H), 7.61 (d, J = 2.0 Hz, 1H), 7.54-7.58 (m, 1H), 8.12-8.19 (m, 1H), 8.19-8.24 (m, 1H).

4d, *m/z* (ES) 528 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.82-0.95 (m, 3H), 0.95 (s, 9H), 2.58-2.78 (br, 1H), 2.98-3.06 (br, 1H), 3.80 (s, 1H), 5.51 (s, 2H), 6.86-6.91 (m, 2H), 6.95 (dd, J = 8.5 Hz, 2.5 Hz, 1H), 7.05-7.13 (m, 4H), 7.15 (d, J = 8.5 Hz, 1H), 7.51 (dd, J = 8.7 Hz, 2.8 Hz, 1H), 7.55 (dt, , J = 8.8 Hz, 2.8 Hz, 1H), 7.62 (d, J = 2.3 Hz, 1H), 7.76 (d, J = 8.7 Hz, 1H), 8.13 (dd, J = 9.1 Hz, 5.2 Hz, 1H), 8.21 (d, J = 8.5 Hz, 1H).

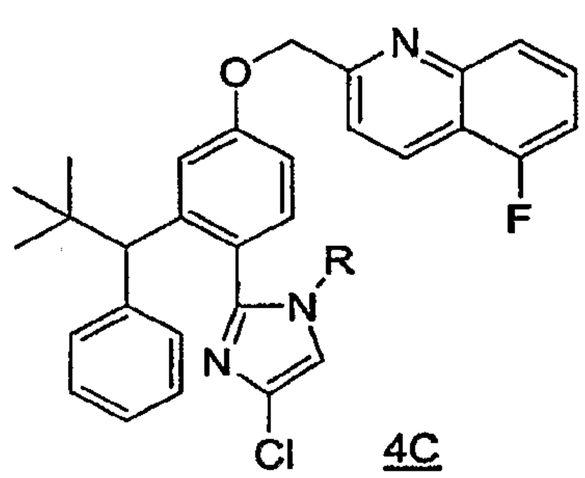
Following procedures similar to that described above for Examples 4b-d, the following compounds can be prepared:



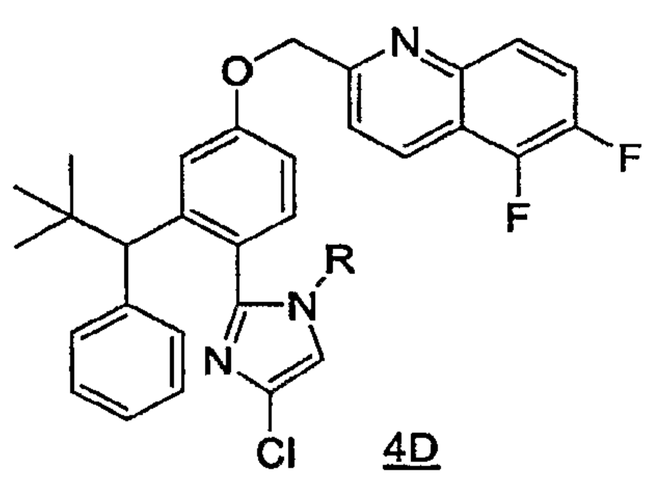
4A



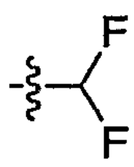
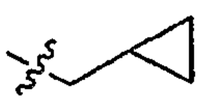
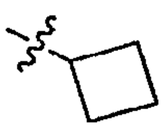
4B



4C



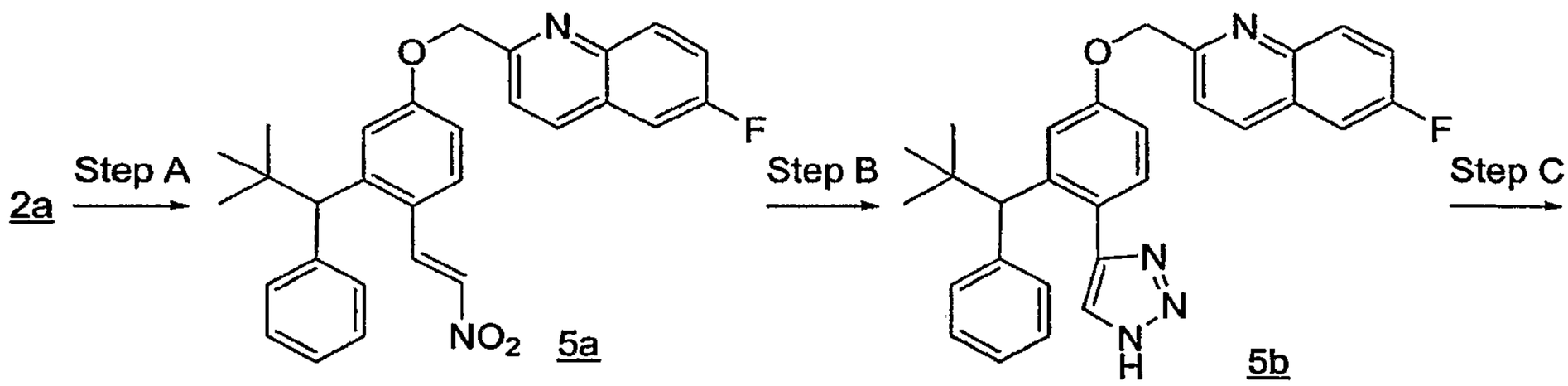
4D

Ex. #4A	Ex. #4B	Ex. #4C	Ex. #4D	R
a	-	a	a	H
b	b	b	b	Me
c	-	c	c	Et
d	d	d	d	
e	e	e	e	
f	f	f	f	
g	g	g	g	CH_2CH_2F

Ex. #4Aa, *m/z* (ES) 482 $(MH)^+$; Ex. #4Ab, *m/z* (ES) 496 $(MH)^+$; Ex. #4Bb, *m/z* (ES) 514 $(MH)^+$; Ex. #4Bd, *m/z* (ES) 551 $(MH)^+$; Ex. #4Be, *m/z* (ES) 554 $(MH)^+$; Ex. #4Bg, *m/z* (ES) 547 $(MH)^+$.

5

Scheme 5



Step A: Preparation of (-)-2-({3-(2,2-dimethyl-1-phenylpropyl)-4-[(*E*)-2-nitrovinyl]phenoxy}methyl)-6-fluoroquinoline (5a)

A microwave tube was charged with nitromethane (0.575 g, 9.45 mmol, 0.51 mL), ammonium acetate (38.0 mg, 0.50 mmole) and 2a (0.80 g, 1.89 mmol). The resulting mixture was irradiated in a microwave apparatus (300W) at 100 °C for 15 min. After cooling to room temperature, the reaction mixture was filtered, and the residue was washed copiously with EtOAc. The filtrate was evaporated *in vacuo*, and the residue was purified by flash chromatography on silica gel (gradient elution; 0-20% EtOAc/hexanes as eluent) to afford 5a, *m/z* (ES) 472 $(MH)^+$.

Step B: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(1*H*-1,2,3-triazol-4-yl)phenoxy]methyl}-6-fluoroquinoline (5b)

Sodium azide (82.0 mg, 1.26 mmol) was added to a stirred solution of 5a (200 mg, 0.423 mmol) in DMSO (0.5 mL) at room temperature and the resulting mixture was stirred at 50 °C for approximately 12 h. The reaction mixture was cooled to room temperature, poured into water, and extracted three times with EtOAc. The combined organic extracts were washed with water (×3), brine, dried ($MgSO_4$), and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 0-50% EtOAc/hexanes as eluent) afforded the title compound 5b, *m/z* (ES) 468 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.89 (s, 9H), 4.03 (s, 1H), 5.67 (s, 2H), 6.90-6.96 (m, 1H), 7.12-7.20 (m, 5H), 7.21-7.27 (m, 2H), 7.56 (d, , J = 2.3 Hz, 1H), 7.58-7.64 (m, , 2H), 7.66-7.74 (m, 1H), 7.94 (d, , J = 8.4 Hz, 1H), 8.42 (dd, J = 9.2 Hz, 4.8 Hz, 1H), 8.45 (d, J = 8.4 Hz, 1H).

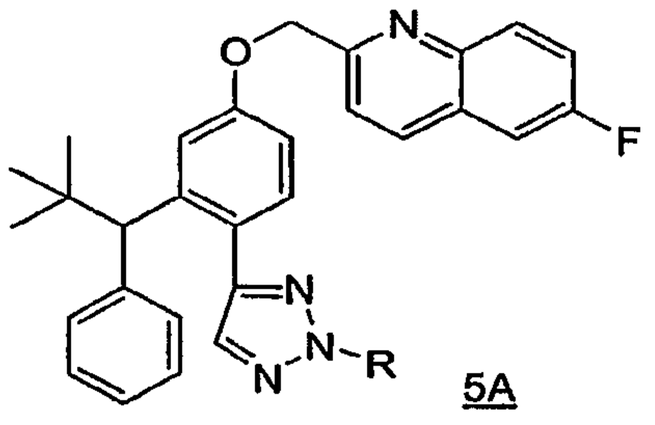
Step C: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(2-methyl-2*H*-1,2,3-triazol-4-yl)phenoxy]methyl}-6-fluoroquinoline (5c), (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(1-methyl-1*H*-1,2,3-triazol-5-yl)phenoxy]methyl}-6-fluoroquinoline (5d), and (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(1-methyl-1*H*-1,2,3-triazol-4-yl)phenoxy]methyl}-6-fluoroquinoline (5e)

Freshly ground anhydrous potassium carbonate (25.0 mg, 0.182 mmol) was added to a stirred solution of 5b (50.0 mg, 0.107 mmol) in DMF (0.5 mL) at room temperature. After approximately 1 h, methyl iodide (20.0 mg, 9 μL, 0.140 mmol) was added *via* syringe, and the resulting mixture was aged at room temperature for about 12 h. The reaction mixture was poured into water, adjusted to pH 5 with aq. citric acid, and extracted three times with EtOAc. The combined organic extracts were washed repeatedly with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 0-20% EtOAc/hexanes as eluent) afforded, in order of elution, 5c (17 mg) and a mixture of 5d/5e (17 mg, 4:1).

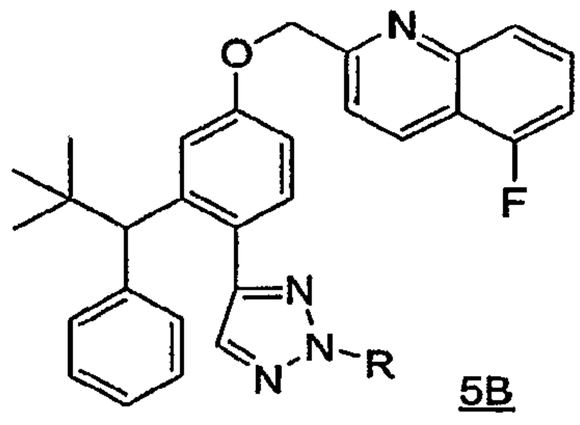
5c.HCl, *m/z* (ES) 481 $(MH)^+$. ^{1}H-NMR (500 MHz, CD_3OD): δ 0.90 (s, 9H), 4.25 (s, 3H), 4.39 (s, 1H), 5.72 (s, 2H), 7.08 (dd, J = 8.5 Hz, 2.5 Hz, 1H), 7.10-7.18 (m, 3H), 7.28-7.33 (m, 3H), 7.52 (s, 1H), 7.57 (d, , J = 2.6 Hz, 1H), 7.96 (dt, J = 8.9 Hz, 2.7 Hz, 1H), 8.02 (dd, J = 8.4 Hz, 2.7 Hz, 1H), 8.12 (d, J = 8.7 Hz, 1H), 8.38 (dd, J = 9.2 Hz, 4.6 Hz, 1H), 8.97 (d, J = 8.7 Hz, 1H).

5d(major)/5e(minor).HCl salt, *m/z* (ES) 481 $(MH)^+$. ^{1}H-NMR (500 MHz, CD_3OD): δ 0.98 (s, 9H, major), 1.03 (s, 9H, minor), 3.64 (s, 1H, major), 4.02 (s, 1H, minor), 4.08 (s, 3H, minor), 4.33 (s, 1H, major), 5.86 (s, 2H, major), 5.92 (s, 2H, minor), 6.90-7.20 (br s, IH, major + minor), 7.14-7.28 (m, 4H, major + minor), 7.28-7.42 (m, 2H, major + minor), 7.76 (d, J = 2.5 Hz, 1H, major), 7.77 (d, J = 2.5 Hz, 1H, minor), 7.93 (d, J = 2.0 Hz, 1H, minor), 8.05-8.16 (m, 2H, major + minor), 8.18 (s, 1H, major), 8.28 (d, J = 8.7 Hz, 1H, major), 8.32 (d, J = 8.7 Hz, 1H, minor), 8.52 (dd, J = 9.6 Hz, 4.5 Hz, 1H, major), 8.55 (dd, J = 9.6 Hz, 4.5 Hz, 1H, minor), 9.18 (d, J = 8.7 Hz, 1H, major), 9.20 (d, J = 8.7 Hz, 1H, minor).

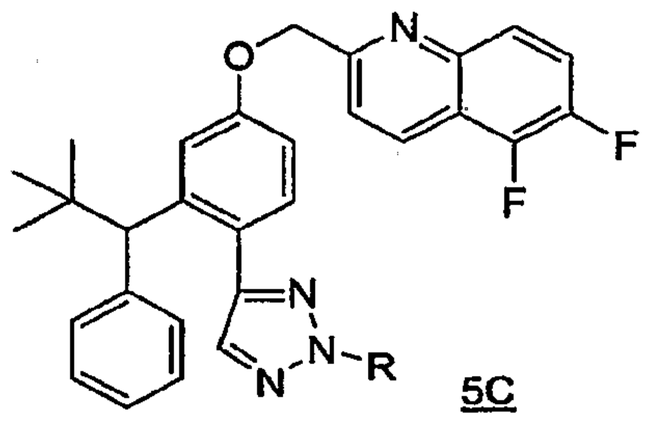
Following procedures similar to that described above for Example 5c, the following compounds can be prepared:



5A

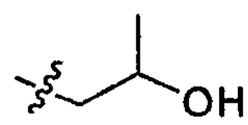
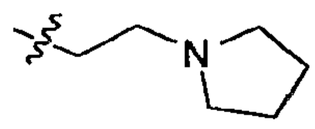


5B



5C

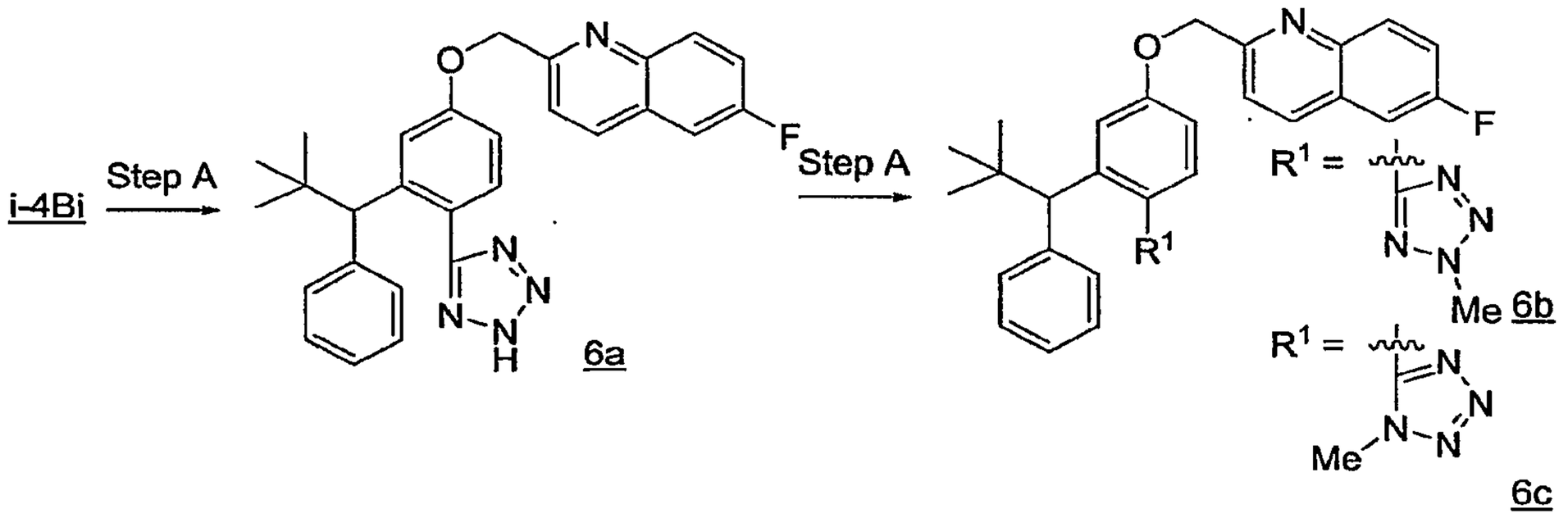
Ex. #5A	Ex. #5B	Ex. #5C	R

-	a	a	H
-	b	b	Me
c	c	c	Et
d	d	d	CH_2CH_2F
e[1]	e[1]	e[1]	CH_2CH_2OH
f[1]	f[1]	f[1]	
g	g	g	
h	h	h	Pr
i	i	i	CH_2CN
j	j	j	CH_2CH_2CN
k	k	k	$CH_2CH(Me)CN$
l	l	l	

Ex. #5Ac, *m/z* (ES) 495 $(MH)^+$; Ex. #5Ad, *m/z* (ES) 514 $(MH)^+$; Ex. #5Ae, *m/z* (ES) 511 $(MH)^+$; Ex. #5Af, *m/z* (ES) 525 $(MH)^+$; Ex. #5Ag, *m/z* (ES) 539 $(MH)^+$; Ex. #5Ah, *m/z* (ES) 509 $(MH)^+$; Ex. #5Ai, *m/z* (ES) 506 $(MH)^+$; Ex. #5Aj, *m/z* (ES) 520 $(MH)^+$; Ex. #5Ak, *m/z* (ES) 533 $(MH)^+$. Ex. #5Al, *m/z* (ES) 564 $(MH)^+$.

Example 6

Scheme 6



Step A: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(2*H*-tetrazol-5-yl)phenoxy]methyl}-6-fluoroquinoline (6a)

Trimethyltin azide (1.54 g, 7.48 mmol) was added to a stirred solution of i-4Bi (200 mg, 0.47 mmol) in toluene (10 mL) at room temperature. The resulting mixture was heated to 120 °C and aged for approximately 3 d. After cooling to room temperature, the reaction mixture was concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-10% MeOH/DCM as eluent) afforded the title compound 6a, *m/z* (ES) 468 $(MH)^+$.

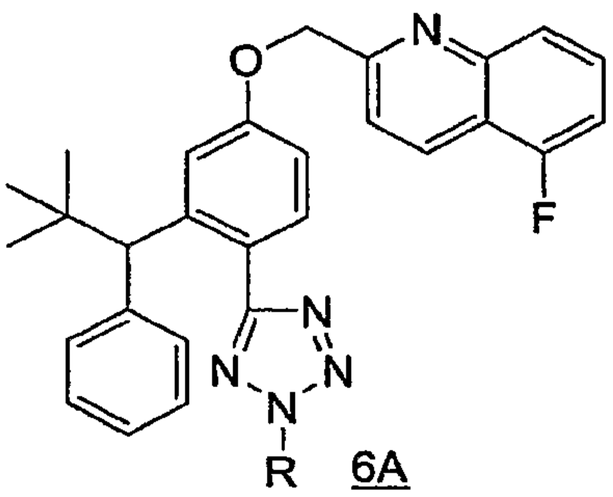
Step B: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(2-methyl-2*H*-tetrazol-5-yl)phenoxy]methyl}-6-fluoroquinoline (6b) and (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(1-methyl-1*H*-tetrazol-5-yl)phenoxy]methyl}-6-fluoroquinoline (6c)

Potassium carbonate (150 mg, 1.09 mmol) followed by methyl iodide (0.16 mL, 11.2 mmol) were added to a stirred solution of 6a (0.310 g, 0.688 mmol) in DMF (5 mL) at room temperature. After 1.5 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-30% EtOAc/hexanes as eluent) afforded, in order of elution, the title compound 6b and the title compound 6c.

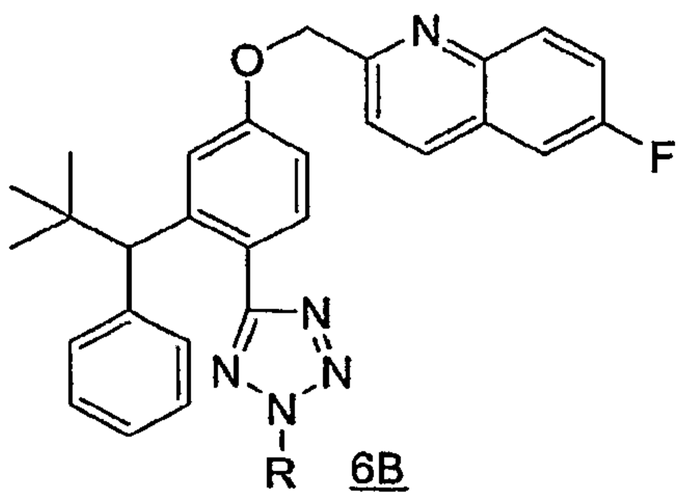
6b.HCl, *m/z* (ES) 482 $(MH)^+$. ^{1}HNMR (500MHz, CD_3OD): δ 0.94 (s, 9H), 4.48 (s, 3H), 4.88 (s, 1H), 5.75 (s, 2H), 7.10-7.19 (m, 4H), 7.42 (d, J =7.5 Hz, 2H), 7.57 (d, J = 2.5 Hz, 1H,), 7.68 (d, J = 8.5 Hz, 1H), 7.99-8.06 (m, 2H), 8.16 (d, J = 8.5 Hz, 1H), 8.41 (dd, J = 4.5, 9.0 Hz, 1H), 9.03 (d, J = 9.0 Hz, 1H).

6c.HCl, *m/z* (ES) 482 $(MH)^+$. ^{1}HNMR (500MHz, CD_3OD): δ 1.02 (s, 9H), 3.22 (s, 3H), 3.81 (s, 1H), 5.83 (s, 2H), 6.94-6.92 (m, 2H), 7.17 (m, 3H), 7.25 (dd, J = 2.5 Hz, 8.5 Hz,1H), 7.34 (m, 1H), 7.84 (d, J = 2.5 Hz, 1H), 7.99-8.07 (m, 2H), 8.21 (d, J = 8.5 Hz, 1H), 8.43 (dd, J = 5.0, 8.5 Hz, 1H), 9.05 (d, J = 8.5 Hz, 1H).

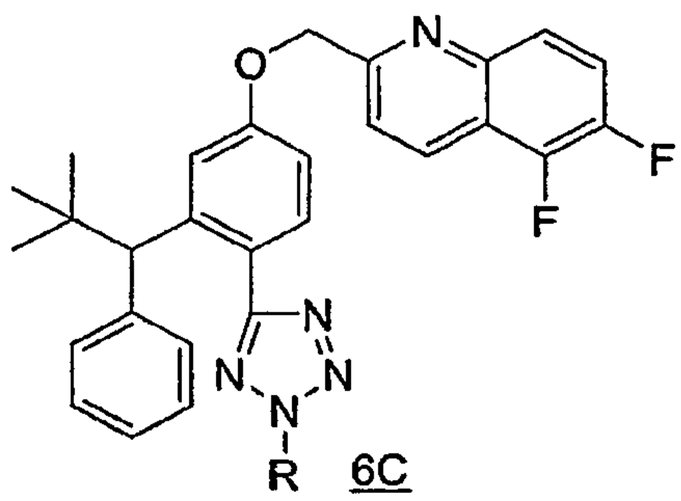
Following procedures similar to that described above for Example 6b, the following compounds can be prepared:



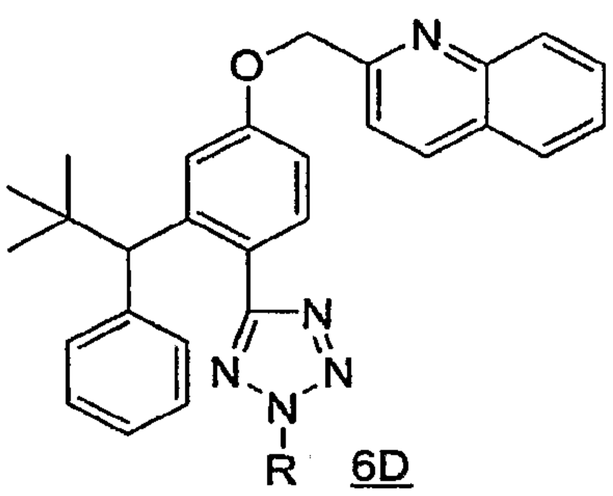
6A



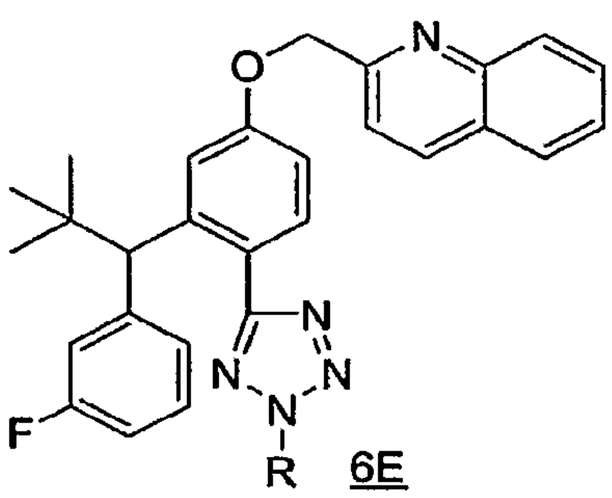
6B



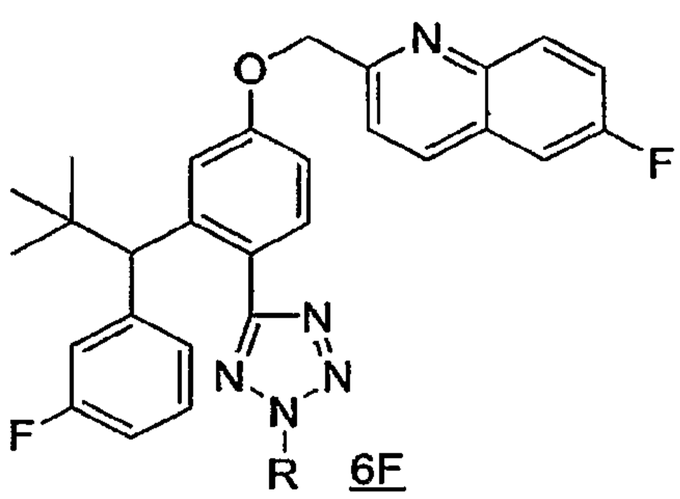
6C



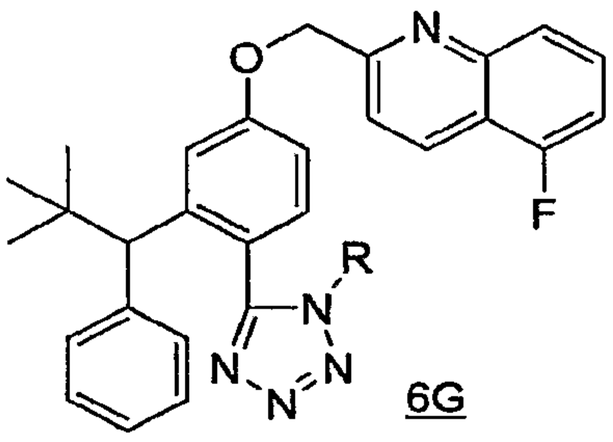
6D



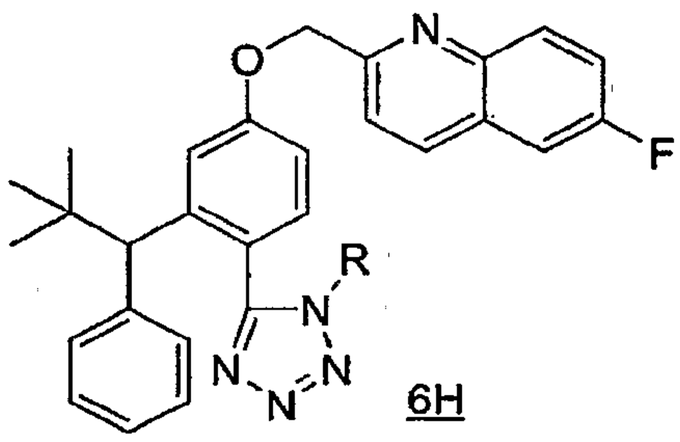
6E



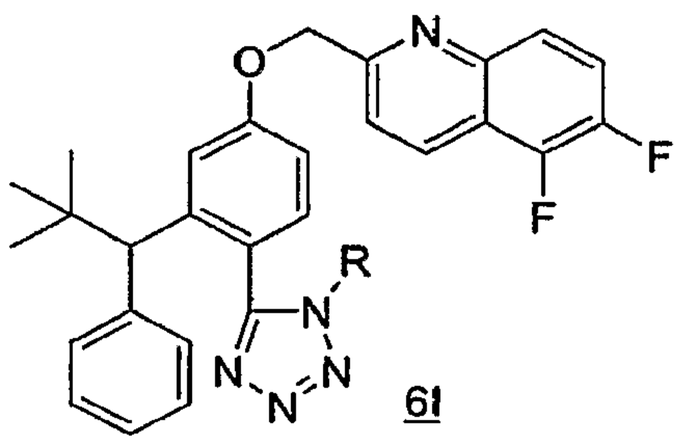
6F



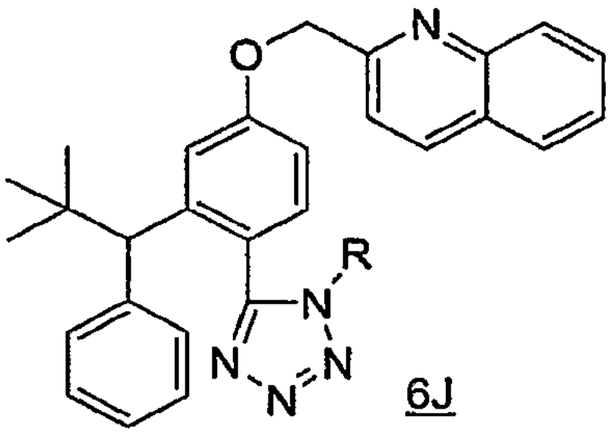
6G



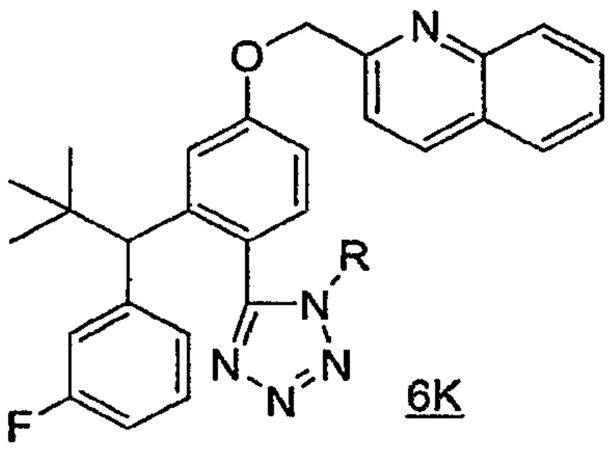
6H



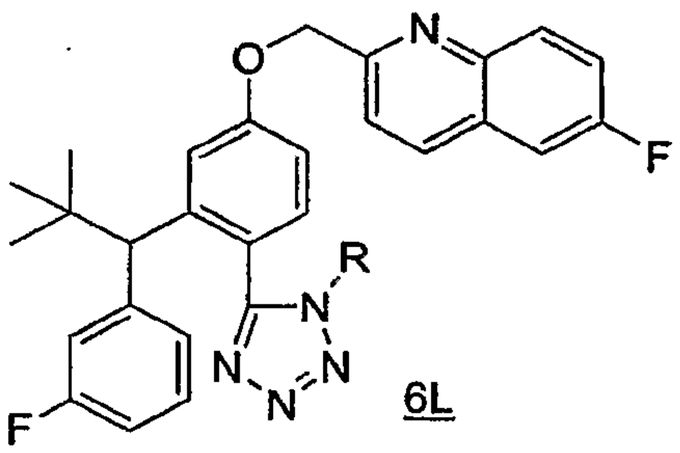
6I



6J

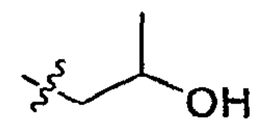
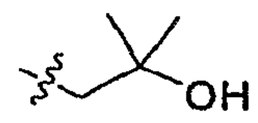
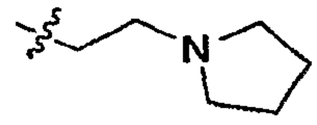


6K



6L

Ex. #6A	Ex. #6B	Ex. #6C	Ex. #6D	Ex. #6E	Ex. #6F	Ex. #6G	Ex. #6H	Ex. #6I	Ex. #6J	Ex. #6K	Ex. #6L	R
a	-	a	a	a	a	a	-	a	a	a	a	H
b	-	b	b	b	b	b	-	b	b	b	b	Me
c	c	c	c	c	c	c	c	c	c	c	c	Et

d	d	d	d	d	d	d	d	d	d	d	d	CH_2CH_2F
e	e	e	e	e	e	e	e	e	e	e	e	CH_2CH_2OH
f	f	f	f	f	f	f	f	f	f	f	f	
g	g	g	g	g	g	g	g	g	g	g	g	
h	h	h	h	h	h	h	h	h	h	h	h	CH_2CN
i	i	i	i	i	i	i	i	i	i	i	i	CH_2CH_2CN
j	j	j	j	j	j	j	j	j	j	j	j	$CH_2CH(Me)CN$
k	k	k	k	k	k	k	k	k	k	k	k	
l	l	l	l	l	l	l	l	l	l	l	l	CHF_2
m	m	m	m	m	m	m	m	m	m	m	m	CH_2CHF_2

Ex. #6Da (tautomer with #6Ja), *m/z* (ES) 450 $(MH)^+$;

Ex. #6Db, *m/z* (ES) 464 $(MH)^+$; Ex. #6Jb, *m/z* (ES) 464 $(MH)^+$;

Ex. #6Fa (tautomer with #6La), *m/z* (ES) 486 $(MH)^+$;

Ex. #6Fb, *m/z* (ES) 500 $(MH)^+$; Ex. #6Lb, *m/z* (ES) 500 $(MH)^+$;

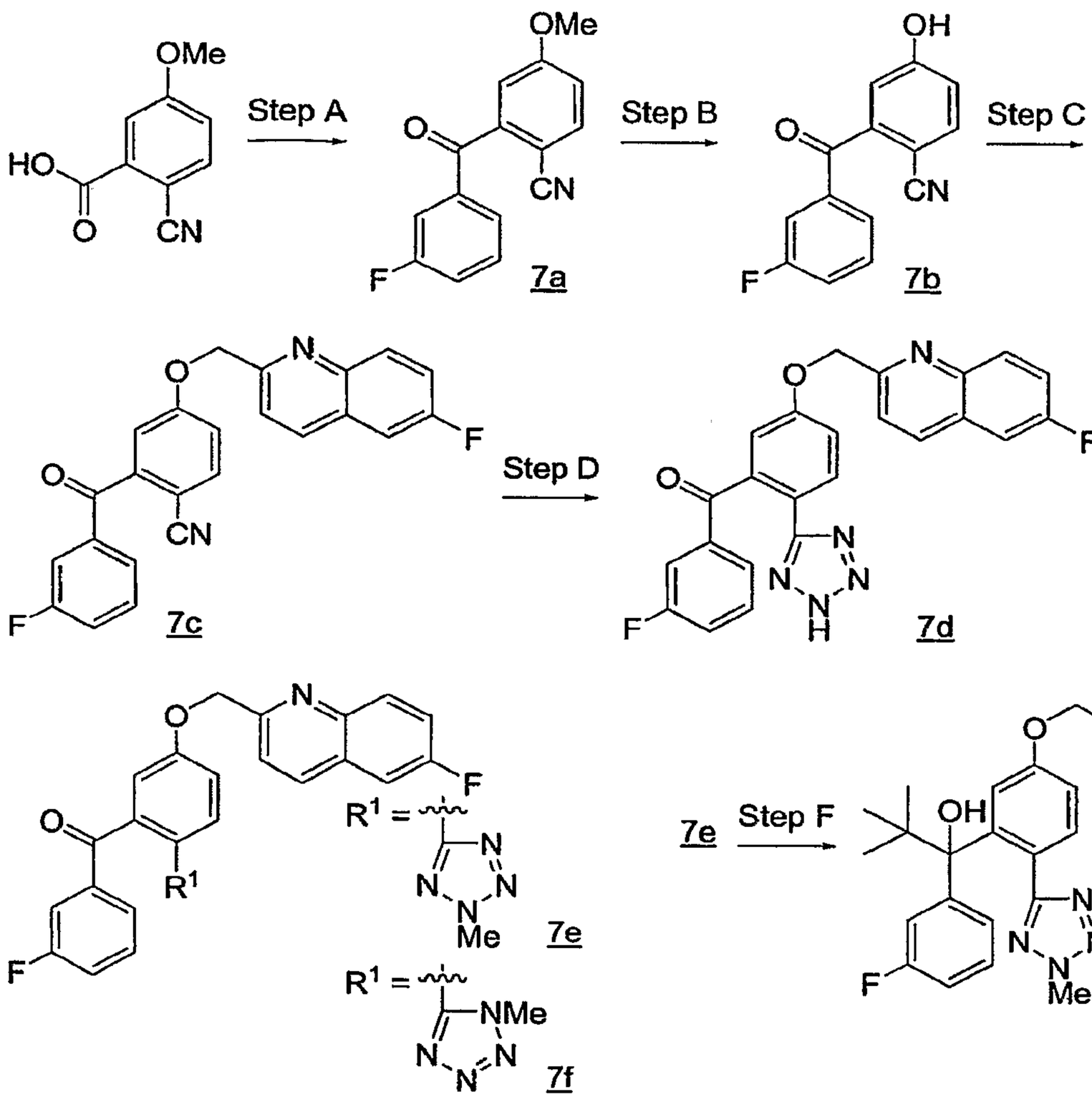
Ex. #6Fd, *m/z* (ES) 532 $(MH)^+$;

Ex. #6Fl, *m/z* (ES) 536 $(MH)^+$; Ex. #6Ll, *m/z* (ES) 536 $(MH)^+$;

Ex. #6Fm, *m/z* (ES) 550 $(MH)^+$

Example 7

Scheme 7



Step A: Preparation of 2-(3-fluorobenzoyl)-4-methoxybenzonitrile (7a)

1,1'-Carbonyldiimidazole (5.00 g, 30.8 mmol) was added to a stirred solution of 2-cyano-5-methoxybenzoic acid (4.00 g, 22.6 mmol) in THF/DCM (3:2, 50 mL) at room temperature. The resulting mixture was aged for approximately 2 h, and then cooled to -78 °C. 3-fluorophenylmagnesium bromide (100 mL of a 1 M solution in THF, 100 mmol) was added slowly *via* cannula, and after completion of addition, the resulting mixture was aged for about 1.5 h. The reaction mixture was quenched with sat. aq. ammonium chloride and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-35% EtOAc/hexanes as eluent) afforded the title compound 7a, *m/z* (ES) 256 $(MH)^+$.

Step B: Preparation of 2-(3-fluorobenzoyl)-4-hydroxybenzonitrile (7b)

Boron tribromide (10.0 mL of a 1 M solution in DCM, 10.0 mmol) was added to a stirred solution of 7a (0.514 g, 2.01 mmol) in DCM (10 mL) at 0 °C. After allowing to warm to room temperature, the reaction mixture was aged for approximately 12 h. The reaction mixture was then poured into 1 N sodium bicarbonate, and extra mes with EtOAc. The combined organic

extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-20% EtOAc/hexanes as eluent) afforded the title compound 7b, *m/z* (ES) 242 $(MH)^+$.

Step C: Preparation of 2-(3-fluorobenzoyl)-4-[(6-fluoroquinolin-2-yl)methoxy]benzonitrile (7c)

Cesium carbonate (1.63 g, 5.00 mmol) followed by 2-(bromo-methyl)-6-fluoroquinoline (0.600 g, 2.50 mmol) were added to a solution of 7b (0.388g, 1.61 mmol) in DMF (10 mL) at room temperature. After 12 h, the reaction mixture was poured into water (50 mL) and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated in vacuo. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-20% EtOAc/hexanes as eluent) afforded the title compound 7c, *m/z* (ES) 401 $(MH)^+$.

Step D: Preparation of (3-fluorophenyl)[5-[(6-fluoroquinolin-2-yl)methoxy]-2-(2*H*-tetrazol-5-yl)phenyl]methanone (7d)

Trimethyltin azide (2.12g, 10.3 mmol) was added to a stirred solution of 7c (0.514 g, 1.28 mmol) in toluene (10 mL) at room temperature. The resulting mixture was heated to 110 °C and aged for 5 d. After cooling to room temperature, the reaction mixture was concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-10% MeOH/DCM as eluent) afforded the title compound 7d, *m/z* (ES) 444 $(MH)^+$.

Step E: Preparation of (3-fluorophenyl)[5-[(6-fluoroquinolin-2-yl)methoxy]-2-(2-methyl-2*H*-tetrazol-5-yl)phenyl]methanone (7e) and (3-fluorophenyl)[5-[(6-fluoroquinolin-2-yl)methoxy]-2-(1-methyl-1*H*-tetrazol-5-yl)phenyl]methanone (7f)

Potassium carbonate (0.603 g, 4.36 mmol) followed by methyl iodide (0.80 ml, 12.8 mmol)were added to a stirred solution of 7d (0.523 g, 1.18 mmol) in DMF (10 mL) at room temperature. After 1.5 h, the reaction mixture was poured into water and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-30% EtOAc/hexanes as eluent) afforded, in order of elution, the title compound 7e and the title compound 7f.

7e: *m/z* (ES) 458 $(MH)^+$.

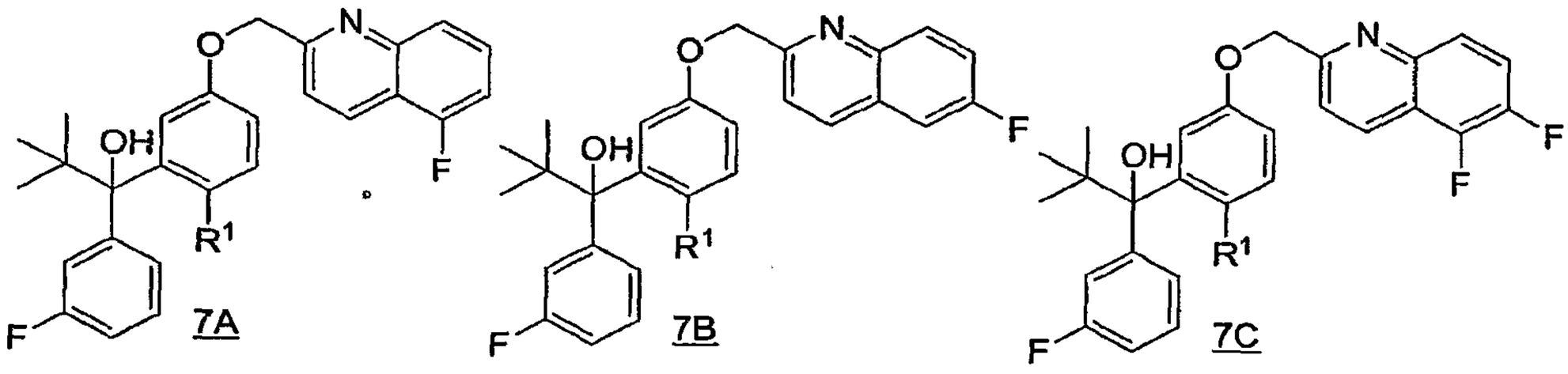
7f: *m/z* (ES) 458 $(MH)^+$.

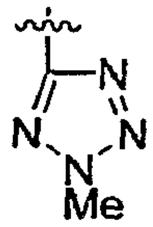
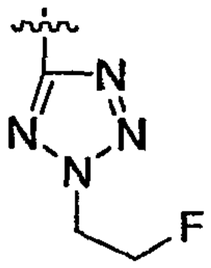
Step F: Preparation of 1-(3-fluorophenyl)-1-[5-[(6-fluoroquinolin-2-yl)methoxy]-2-(2-methyl-2*H*-tetrazol-5-yl)phenyl]-2,2-dimethylpropan-1-ol (7g)

Tert-Butyl magnesium bromide (2.0 mL of a 1 M solution in THF, 2.00 mmol) was added to a stirred solution of 7e (0.193 g, 0.422 mmol) in THF (8.0 mL) at 0 °C. After 3 h, the reaction mixture was quenched with sat. aq. ammonium chloride, poured into water and extracted three times with

EtOAc. The combined organic extracts were washed with brine, dried (Na_2SO_4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution, 0-50% EtOAc/hexanes as eluent) afforded the title compound 7g, *m/z* (ES) 516 $(MH)^+$. 7g.HCl ^{1}HNMR (500MHz, CD_3OD): 1.22 (bs, 9H). 4.16 (s, 3H), 5.81 (s, 2H), 6.62-6.59 (m, 1H), 6.69 (d, J = 8 Hz, 1H), 6.76-6.73 (m, 1H), 6.99-6.94 (m, 1H), 7.23 (dd, J = 2.5, 8.5 Hz, 1H), 7.43 (d, J = 8.5 Hz, 1H), 7.96 (d, J = 2.5 Hz, 1H), 8.06-8.00 (m, 1H), 8.07 (dd, J = 2.5, 8.5 Hz, 1H), 8.24 (d, J = 9 Hz, 1H), 8.43 (dd, J = 4.5, 9 Hz, 1H), 9.10 (d, J = 8.5 Hz, 1H).

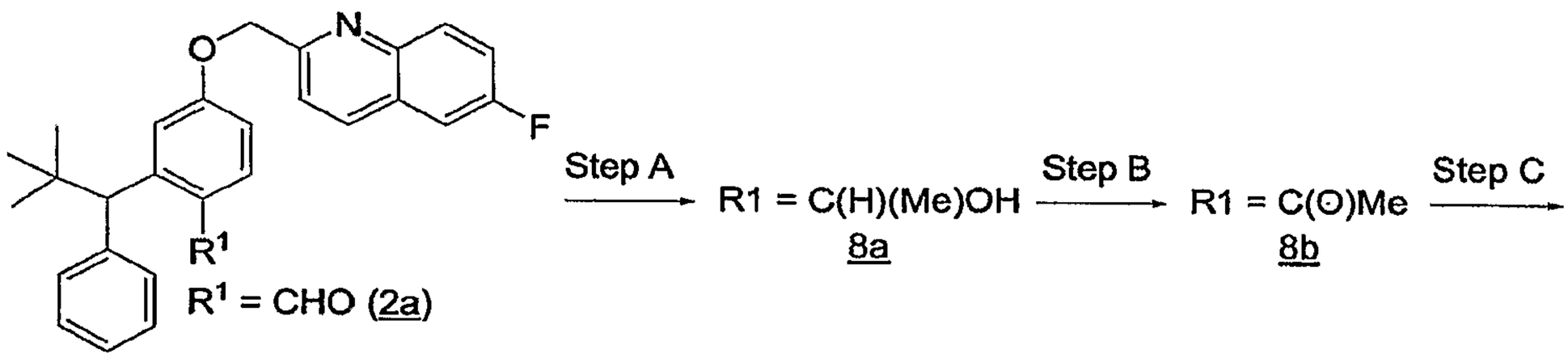
Following procedures similar to that described above for Example 7g, the following compounds can be prepared.

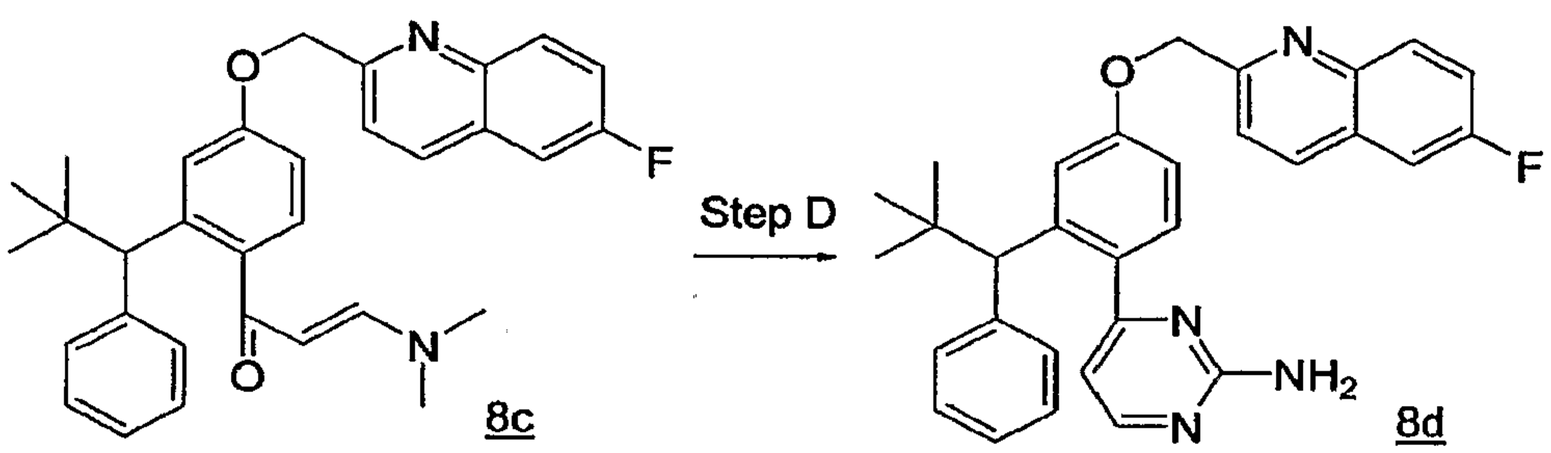


Ex. #7A	EX. #7B	EX. #7C	R^1
a	-	a	
b	b	b	

Example 8

Scheme 8





Step A: Preparation of (-)-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}ethanol (8a)

Methylmagnesium bromide (3.0 mL of a 1.4 M solution in ether, 4.20 mmol) was added to a stirred solution of 2a (1.23 g, 2.88 mmol) in THF (40 mL) at 0 °C. After 2 h, the reaction mixture was quenched with sat. aq. ammonium chloride, then poured into water and extracted three times with EtOAc. The combined organic extracts were washed with water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 10-35% EtOAc/Hexanes as eluent) afforded the title compound 8a as an equimolar mixture of diastereoisomeric alcohols, *m/z* (ES) 444 $(MH)^+$.

Step B: Preparation of (-)-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}ethanone (8b)

Manganese(IV) oxide (1.70 g, 19.2 mmol), followed by celite (2.00 g) were added to a stirred solution of 8a (850 mg, 1.92 mmol) in toluene (60 mL) at room temperature. The reaction mixture was heated to approximately 110 °C and aged for about 20 h. After cooling to room temperature, the reaction mixture was filtered and the residue washed copiously with EtOAc (30 mL). The filtrate was concentrated *in vacuo* and the crude residue was purified by flash chromatography on silica gel (gradient elution; 10-25% EtOAc/Hexanes as eluent) to provide the title compound 8b, *m/z* (ES) 442 $(MH)^+$.

Step C: Preparation of (-)-(2*E*)-3-(dimethylamino)-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}prop-2-en-1-one (8c)

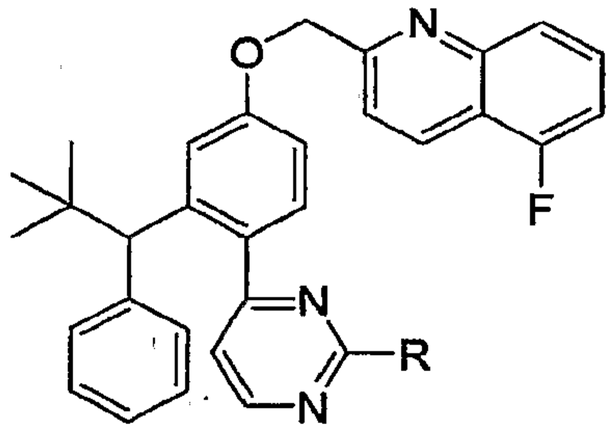
A thick-walled pressure tube was charged with 8b (250 mg, 0.569 mmol) and *N,N*-dimethylformamide diethyl acetal (3.0 mL). The resulting mixture was irradiated in a microwave apparatus (300W) at 140 °C for approximately r cooling to room temperature, the reaction

mixture was concentrated *in vacuo* and the crude residue was purified by flash chromatography on silica gel (gradient elution; 40-80% EtOAc/Hexanes as eluent) to afford the title compound 8c, *m/z* (ES) 498 $(MH)^+$.

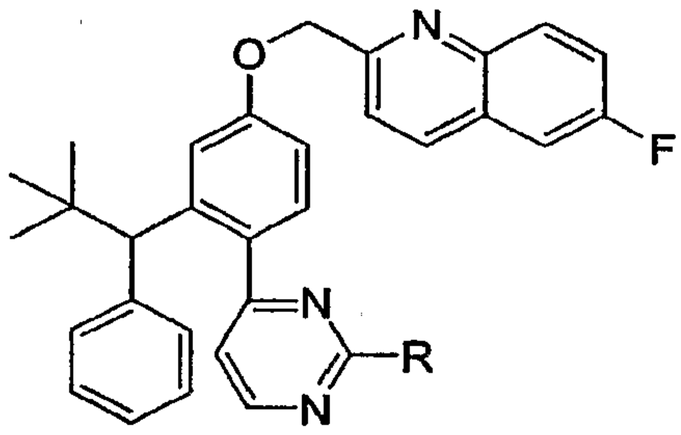
Step D: Preparation of (-)-4-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}pyrimidin-2-amine (8d)

Guanidine hydrochloride (23.6 mg, 0.248 mmol), followed by sodium methoxide (0.60 mL of a 0.5 M solution in MeOH, 0.300 mmol) were added to a stirred solution of 8c (62.0 mg, 0.124 mmol) in EtOH (1.0 mL) at room temperature. The resulting mixture was sealed, heated to 78 °C and aged for approximately 18 h. After cooling to room temperature, the reaction mixture was concentrated *in vacuo* and the residue was partitioned between EtOAc and water. The organic phase was separated, and the aq. layer was re-extracted with EtOAc. The combined organic extracts were washed with brine, dried ($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by preparative TLC (silica gel, 70 % EtOAc/Hexanes as eluent) to provide the title compound 8d, *m/z* (ES) 493 $(MH)^+$. ^1H NMR (500 MHz, $CDCl_3$): δ 0.84 (s. 9H), 4.12 (s, 1H), 5.08 (bs, N-H, 2H), 5.48 (s, 2H), 6.54 (d, J = 5.2 Hz, 1H), 6.95 (dd, J = 8 Hz, 2 Hz, 1H)), 7.08-7.12 (m, 3H), 7.19-7.23 (m, 3H), 7.48 (dd, J = 8.7 Hz , 2.7 Hz, 1H), 7.52 (d, J = 2.6 Hz, 1H), 7.55 (m, 1H), 7.70 (d, J = 8.5 Hz, 1H), 8.12 (dd, J = 6.9 Hz, 4.8 Hz, 1H), 8.15 (d, J = 8.9 Hz, 1H), 8.30 (d, J = 5.1 Hz, 1H).

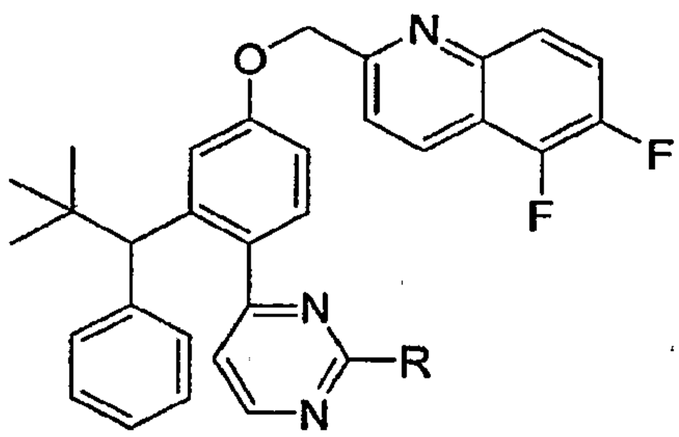
Following procedures similar to that described above for Example 8d, the follwing compounds can be prepared.



8A



8B



8C

Ex. #8A	Ex. #8B	Ex. #8C	R
a	-	a	NH_2
b	b	b	Me
c	c	c	CF_3

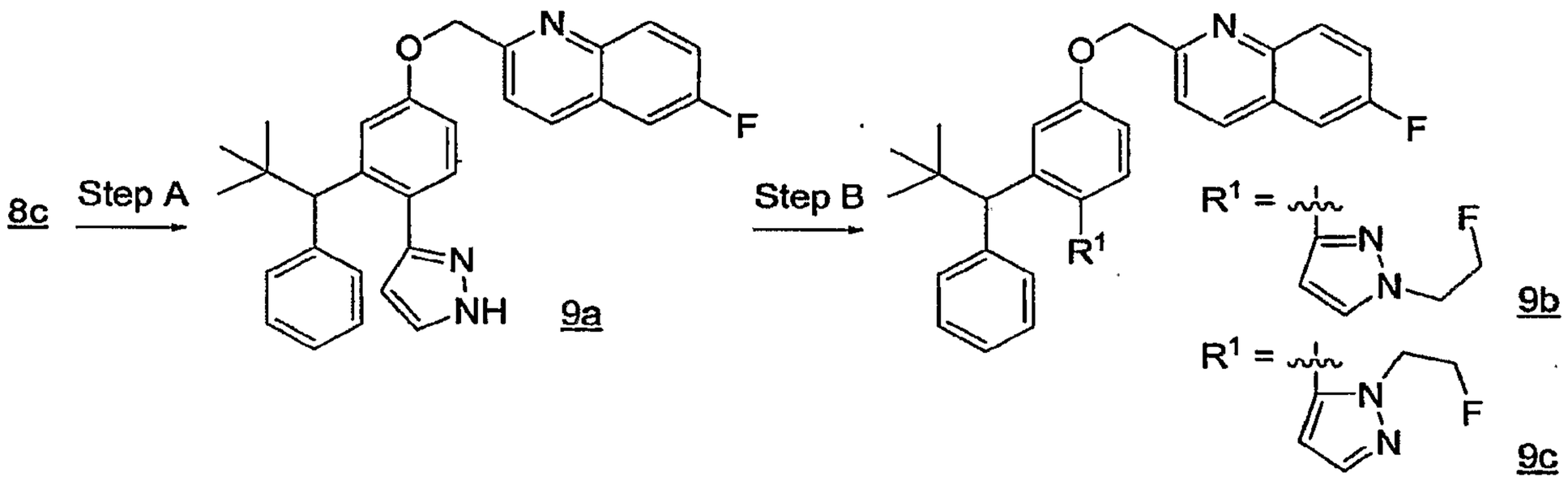
d	d	d	Et
e	e	e	Cyc-Pr

Ex. #8Ab, *m/z* (ES) 492 $(MH)^+$;

Ex. #8Bb, *m/z* (ES) 492 $(MH)^+$; Ex. #8Bc, *m/z* (ES) 546 $(MH)^+$; Ex. #8Be, *m/z* (ES) 518 $(MH)^+$.

Example 9

Scheme 9



Step A; Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(1*H*-pyrazol-3-yl)phenoxy]methyl}-6-fluoroquinoline (9a)

Anhydrous hydrazine (0.020 mL, excess) was added to a stirred solution of 8c in EtOH (1.0 mL) and the resulting mixture was heated in an oil bath at 110 °C for approximately 18 h. After cooling to room temperature, the volitiles are removed *in vacuo*. The crude residue was purified by preparative TLC (silica gel, 60 % EtOAc/Hexanes as eluent) to provide the title compound 9a, *m/z* (ES) 466 $(MH)^+$. ^1H NMR (500 MHz, $CDCl_3$): δ 0.91 (s. 9H), 3.99 (s, 1H), 5.49 (s, 2H), 6.29 (d, J = 1.6 Hz, 1H), 6.94 (dd, J = 8.5 Hz, 2.5 Hz, 1H)), 7.09-7.21 (m, 7H), 7.49 (dd, J = 8.9 Hz, 2.7 Hz, 1H), 7.55 (m, 2H), 7.64 (s,1H), 7.73 (d, J = 8.7 Hz, 1H), 8.14 (dd, J = 7.1 Hz, 5.0 Hz, 1H), 8.18 (d, J = 8.4 Hz, 1H).

Step B: Preparation of (-)-2-({3-(2,2-dimethyl-1-phenylpropyl)-4-[1-(2-fluoroethyl)-1*H*-pyrazol-3-yl]phenoxy}methyl)-6-fluoroquinoline (9b) and (-)-2-({3-(2,2-dimethyl-1-phenylpropyl)-4-[1-(2-fluoroethyl)-1*H*-pyrazol-5-yl]phenoxy}methyl)-6-fluoroquinoline (9c)

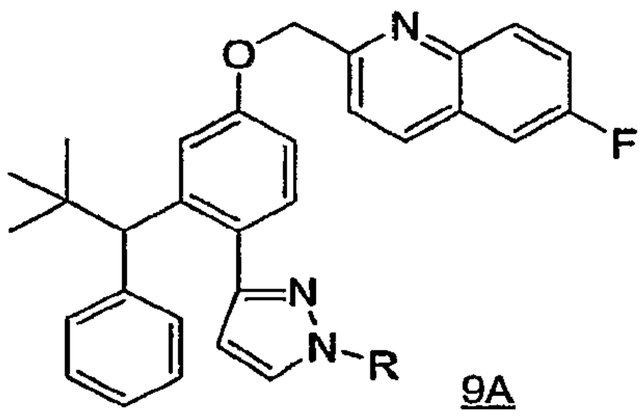
Sodium hydride (60% in oil, 7.00 mg, 0.162 mmol) was added to a stirred solution of 9a (55.0 mg, 0.118 mmol) in DMF (2 mL) at 0 °C. After 10 min, 1-bromo-2-fluoroethane (18.9 mg, 0.150 mmol) was added *via* syringe. The resulting mixture was warmed to room temperature and aged for approximately for 18 h. The reaction mixture was quenched with sat. aq. ammonium chloride and then extracted three times with EtOAc. The combined organic extracts were washed with water, brine, dried

($MgSO_4$) and concentrated *in vacuo*. The crude residue was purified by preparative TLC (silica gel, 30 % EtOAc/hexanes as eluent) to afford the title compounds 9b and 9c.

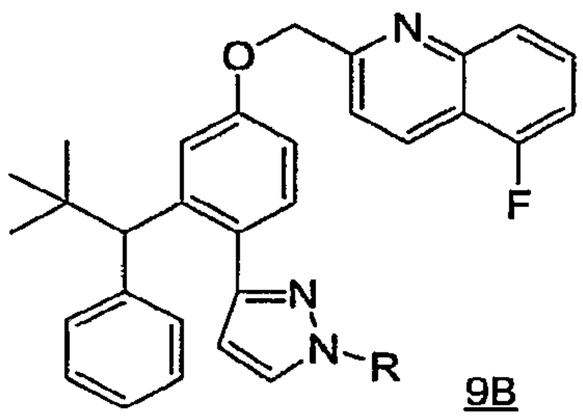
9b (slower moving band on TLC), *m/z* (ES) 512 $(MH)^+$. ^{1}HNMR (500MHz, $CDCl_3$): δ 0.91 (s. 9H), 4.40 (s, 1H), 4.48 (dt, *J* = 16.9 Hz, 4.7 Hz, 2H), 4.83 (dt, *J* = 37 Hz, 5.3 Hz, 2H), 5.47 (s, 2H), 6.20 (d, *J* = 2.1 Hz, 1H), 6.92 (dd, *J* = 8.5 Hz, 2.5 Hz, 1H)), 7.07-7.29 (m, 6H), 7.49 (dd, *J* = 8.9 Hz , 2.7 Hz, 1H), 7.51 (m, 2H), 7.53 (dt, J = 8.8Hz, 2.7 Hz, 1H), 7.77 (d, *J* = 8.4 Hz, 1H), 8.14 (m, 2H).

9c (faster moving band on TLC), *m/z* (ES) 512 $(MH)^+$. 9c (major rotomer/minor rotomer) ^{1}HNMR (500MHz, $CDCl_3$): δ 0.83 (s. 9H, minor), 0.93 (s, 9H major), 3.08-3.23 (m, 2H, major), 3.68 (s, 1H, major), 3.71 (s, 1H, minor), 4.23 (m, 2H, major), 4.44-4.56 (m, 2H, minor), 4.79-5.02 (m, 2H, minor) 5.51(s, 4H, major + minor), 5.88 (d, *J* = 1.4 Hz, 1H, minor), 6.23 (d, *J* = 1.6 Hz, 1H, major), 6.88 (m, 2H, major), 6.97 (m, 2H, minor), 7.04-7.13 (m, 10H, major + minor), 7.50-7.66 (m, 8H, major + minor), 7.75-7.80 (m, 2H, major + minor), 8.15 (m, 2H, major + minor), 8.18 (m, 2H, major + minor).

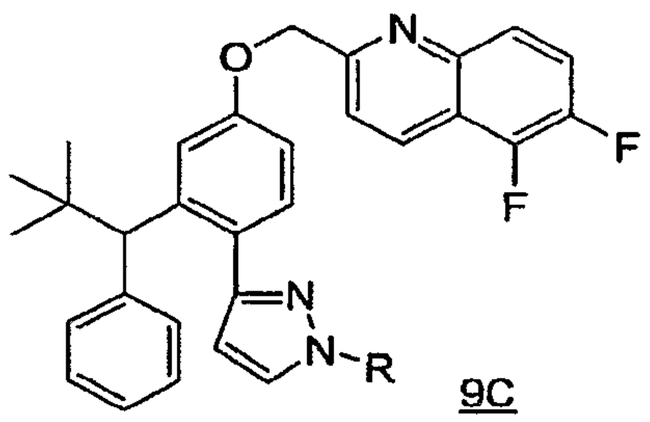
Following procedures similar to that described above for Example 9b, the following compounds can be prepared.



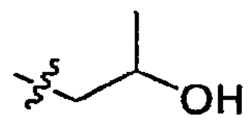
9A



9B



9C

Ex. #9A	Ex. #9B	Ex. #9C	R
-	a	a	H
b	b	b	Me
c	c	c	Et
-	d	d	CH_2CH_2F
e[1]	e[1]	e[1]	CH_2CH_2OH
f[1]	f[1]	f[1]	
g[1]	g[1]	g[1]	
h	h	h	CH_2CN

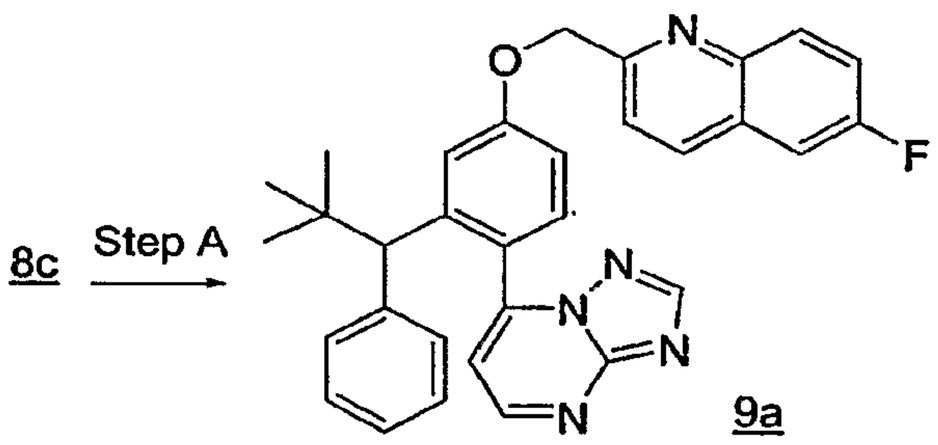
i	i	i	CH_2CH_2CN

[1]The alcohol functionality is masked as a benzyl ether in Step B. Deprotection of the benzyl ether is achieved with methanesulfonic acid (according to *J. Am. Chem. Soc.*, **1996**, *118*, 4560) to provide the desired alcohol product.

Ex. #9Ab, *m/z* (ES) 480 $(MH)^+$.

Example 10

Scheme 10

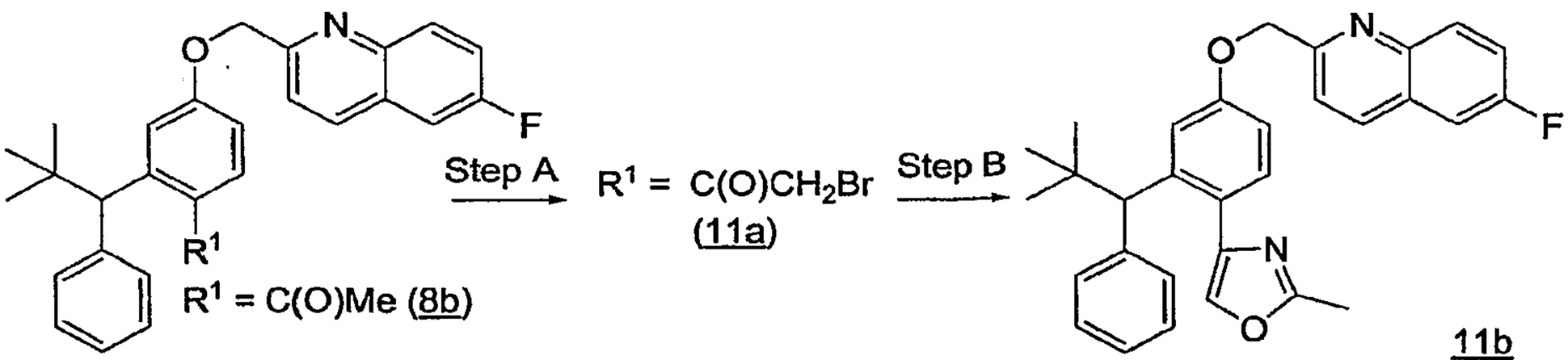


Step A: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-[1,2,4]triazolo[1,5-*a*]pyrimidin-7-ylphenoxy]methyl}-6-fluoroquinoline (9a)

1*H*-1,2,4-Triazol-5-amine (8.40 mg, 0.100 mmol) was added to a stirred solution of 8c (26.0 mg, 0.0524 mmol) in acetic acid (0.50 mL) at room temperature. The resulting mixture was heated to 117 °C and stirred for approximately 16 h. After cooling to room temperature, the volitiles were removed *in vacuo* and the residue was partitioned between EtOAc and sat. aq. sodium bicarbonate. The organic phase was separated and the aq. phase was re-extracted with EtOAc. The combined organic extracts were washed with water, brine, dried (MgSO4) and concentrated *in vacuo*. Purification of the crude residue by preparative TLC (silica gel, 40 % EtOAc/hexanes as eluent) afforded 9a, *m/z* (ES) 519 $(MH)^+$. ^{1}HNMR (500MHz, $CDCl_3$): δ 0.91 (s. 9H), 3.48 (s, 1H), 5.54 (s, 2H), 6.50 (d, J = 4.2 Hz, 1H), 6.86 (d, J = 7.5 Hz, 2H), 7.07 (m, 3H), 7.13-7.27 (m, 2H), 7.52 (dd, J = 8.7 Hz, 2.7Hz, 1H), 7.56 (dt, J = 8.8Hz, 2.8 Hz, 1H), 7.72 (bs,1H), 7.77 (d, J = 8.5 Hz, 1H), 8.14 (dd, J = 9.1 Hz, 5.2 Hz, 1H) 8.22 (d, J = 8.5 Hz, 1H), 8.55 (bs, 1H), 8.73 (bs, 1H).

Example 11

Scheme 11



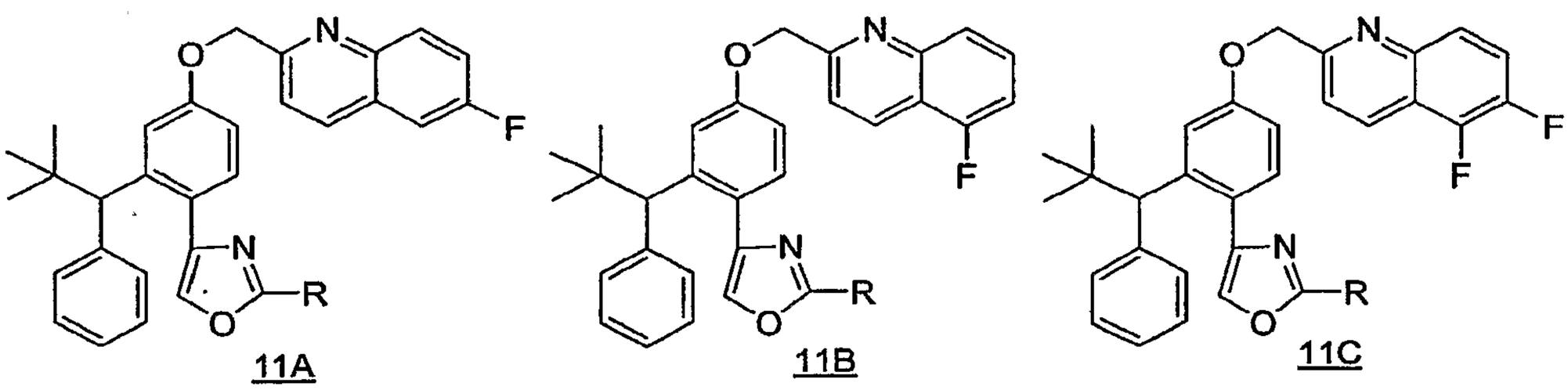
Step A: Preparation of (-)-2-bromo-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}ethanone (11a)

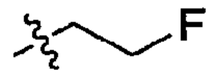
Pyrrolidinone hydrotribromide (248 mg, 0.500 mmol) was added to a stirred solution of 8b (216 mg, 0.470 mmol) in THF (10 mL) at room temperature and the resulting mixture was stirred at 40 °C for approximately 2 h. After cooling to room temperature, the reaction mixture was poured into sat. aq. sodium bicarbonate and extracted three times with EtOAc. The combined organic extracts were washed with brine, dried ($MgSO_4$) and concentrated in *vacuo*. Purification of the crude residue by flash chromatography on silica gel (gradient elution; 10-20% EtOAc/Hexanes as eluent) provided the title compound 11a, m/z (ES) 522 $(MH)^+$.

Step B: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(2-methyl-1,3-oxazol-4-yl)phenoxy]methyl}-6-fluoroquinoline (11b)

A stirred mixture of acetamide (100 mg, excess) and 11a (35 mg, 0.0674 mmol) was heated to 170 °C and aged for about 2 h. After cooling to room temperature, the reaction mixture was diluted with water and extracted three times with EtOAc. The combined organic extracts were washed with sat. aq. sodium bicarbonate, water, brine, dried ($MgSO_4$) and concentrated *in vacuo*. Purification of the crude residue by preparative TLC (silica gel, 30 % EtOAc/Hexanes as eluent) afforded the title compound 11b, *m/z* (ES) 481 $(MH)^+$. ^{1}H-NMR (500 MHz, $CDCl_3$): δ 0.94 (s, 9H), 2.58 (s, 3H), 4.30 (s, 1H), 5.47(s, 2H), 6.93 (dd, J = 8.4 Hz, 2.5 Hz, 1H), 7.10-7.12 (m, 3H), 7.24-7.31 (m, 3H), 7.37 9s, 1H), 7.48-7.49 (m, 3H), 7.55 (dt, J = 8.9 Hz, 3.0 Hz, 1H), 7.72 (d, J = 8.5 Hz, 1H), 8.13-8.17 (m, 2H).

Following procedures similar to that described above for Example 11b, the following compounds can be prepared:



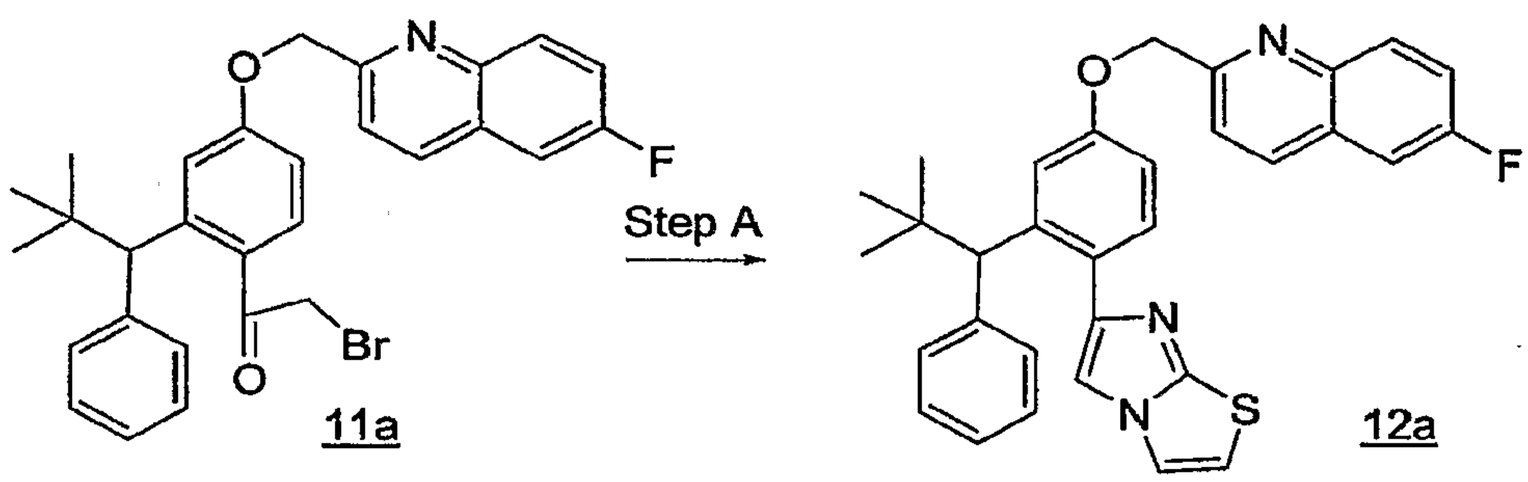
Ex. #11A	Ex. #11B	Ex. #11C	R
a	a	a	H
-	b	b	CH_3
c	c	c	CH_2CH_3
d	d	d	
e	e	e	NH_2

[1]Formamide can be used in Step B

[2]Urea can be used in Step B

Example 12

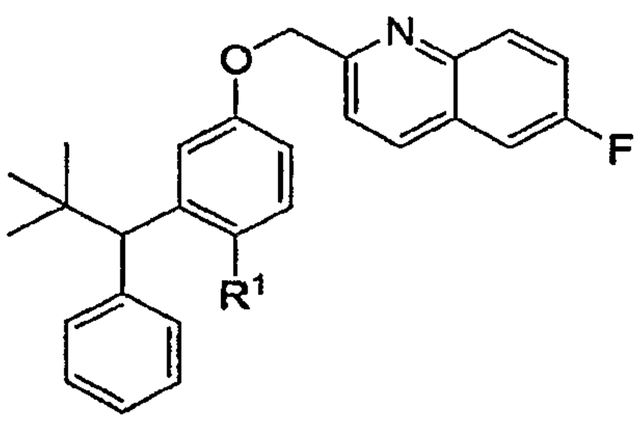
Scheme 12



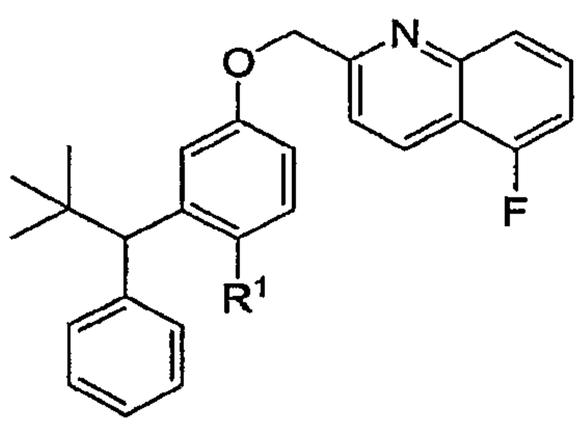
Step A: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-imidazo[2,1-b][1,3]thiazol-6-ylphenoxy]methyl}-6-fluoroquinoline (12a)

1,3-Thiazol-2-amine (10 mg, 0.100 mmol) was added to a stirred solution of 11a (45 mg, 0.096 mmol) in EtOH (2 mL) at room temperature. The reaction mixture was sealed, heated to 78 °C and aged for approximately 16 h. After cooling to room temperature, the volitiles were removed *in vacuo* and the residue was partitioned between EtOAc and sat. aq. sodium bicarbonate. The organic phase was separated and the aqueous phase was extracted with EtOAc. The combined organic extracts were washed with water, brine, dried (MgSO4) and concentrated *in vacuo*. Purification of the crude residue by preparative TLC (silica gel, 5 % MeOH/DCM as eluent) afforded 12a, *m/z* (ES) 522 $(MH)^+$. ^{1}HNMR (500MHz, $CDCl_3$): δ 0.92 (s. 9H), 4.53 (s, 1H) 5.48 (s, 2H), 6.85 (d, J = 4.6 Hz, 1H), 6.93 (d, J = 2.7 Hz, 1H)), 7.09-7.12 (m, 3H), 7.26 (s, 1H), 7.27 (m, 2H), 7.35 (d, J = 8.5 Hz, 1H), 7.45 (d, J = 4.6 Hz, 1H), 7.48-7.50 (m, 2H), 7.55 (dt, J = 8.9 Hz, 3.0 Hz, 1H), 7.74 (d, J = 8.7 Hz, 1H), 8.13 (dd, J = 8.8 Hz, 4.7 Hz, 1H) 8.16 (d, J = 8.7 Hz, 1H).

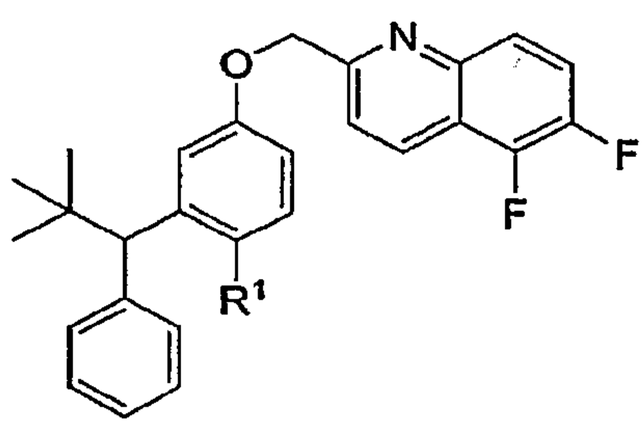
Following procedures similar to that described above for Example 12a, the following compounds can be prepared:



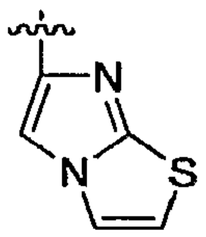
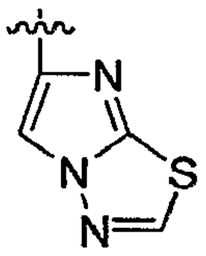
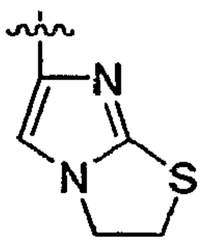
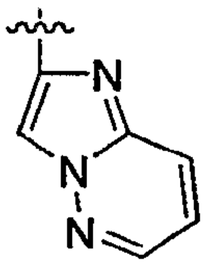
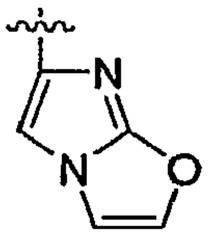
12A



12B



12C

Ex. #12A	EX. #12B	EX. #12C	R^1
-	a	a	
b[1]	b	b[1]	
c[2]	c[2]	c[2]	
d[3]	d[3]	d[3]	
e[4]	e[4]	e[4]	

[1]1,3,4-thiadiazol-2-amine was used in Step A.

[2]4,5-dihydro-1,3-thiazol-2-amine was used in Step A.

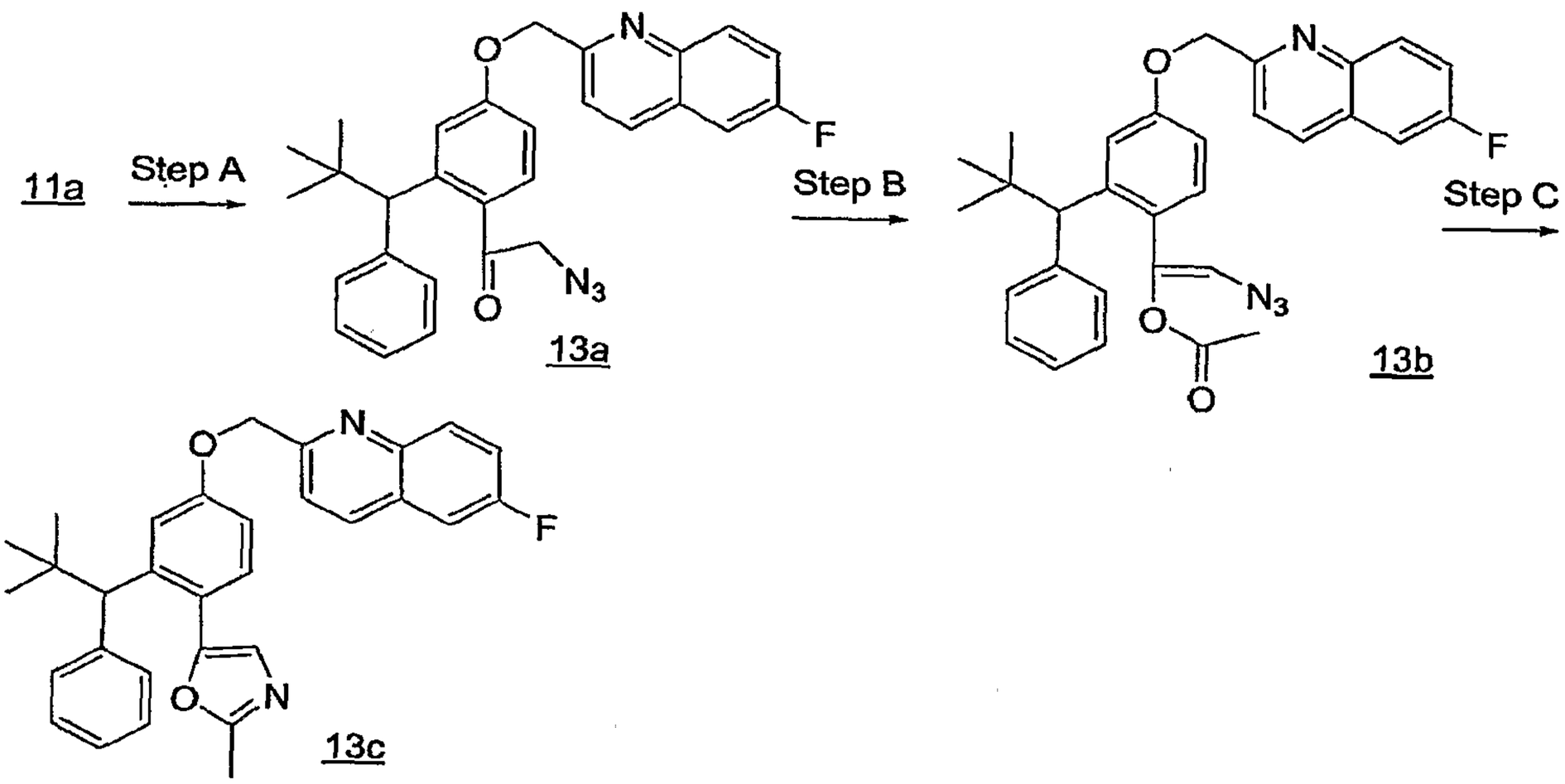
[3]Pyridazin-3-amine can be used in Step A.

[4]1,3-oxazol-2-amine can be used in Step A.

Ex. #12Ab, *m/z* (ES) 523 $(MH)^+$, Ex. #12Ac, *m/z* (ES) 524 $(MH)^+$.

Example 13

Scheme 13



Step A: Preparation of (-)-2-azido-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}ethanone (13a)

Sodium azide (3.3 equiv.) is added to a stirred solution of 11a (1 equiv.) in DMF (0.1 M) at 0 °C. After allowing to warm to room temperature, the reaction mixture is aged until the reaction is deemed complete. The reaction mixture is poured into water and extract three times with EtOAc. The combined organic extracts are washed with water, brine, dried (MgSO4) and concentrated *in vacuo*. Purification of the crude residue by flash chromatography provides the title compound 13a.

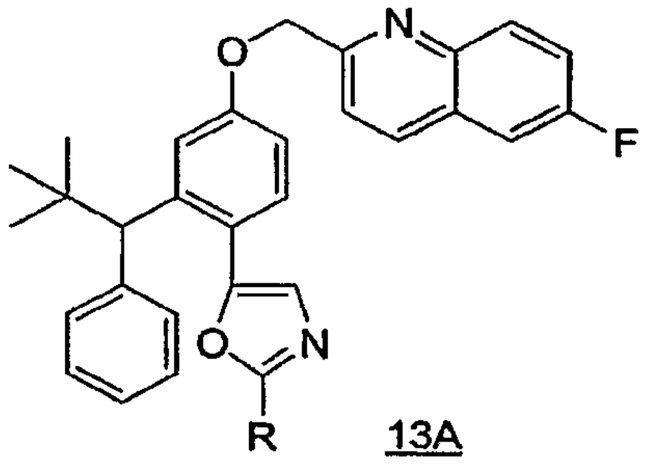
Step B: Preparation of (-)-(Z)-2-azido-1-{2-(2,2-dimethyl-1-phenylpropyl)-4-[(6-fluoroquinolin-2-yl)methoxy]phenyl}vinyl acetate (13b).

Lithium diisopropylamide (1.2 equiv.) is added to a stirred solution of 13a (1 equiv.) in THF (0.05 M) at -78 °C. After 5 min, acetic anhydride (1.2 equiv.) is added and the resulting mixture is stirred at -78 °C until the reaction is deemed complete. The reaction mixture is quenched with sat. aq. ammonium chloride and extracted three times with EtOAc. The combined organic extracts are washed with water, brine, dried (MgSO4) and concentrate *in vacuo*. Purification of the crude residue by flash chromatography affords the title compound 13b.

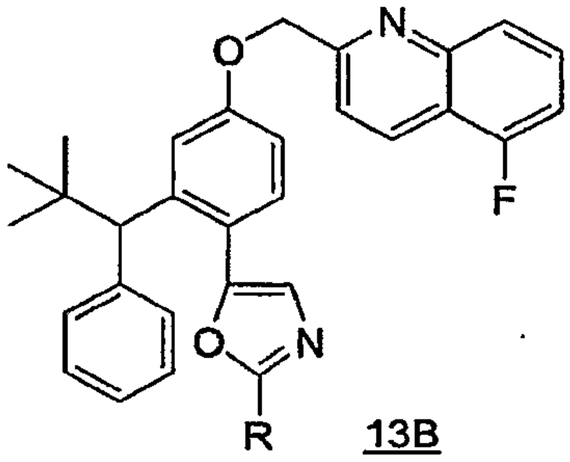
Step C: Preparation of (-)-2-{[3-(2,2-dimethyl-1-phenylpropyl)-4-(2-methyl-1,3-oxazol-5-yl)phenoxy]methyl}-6-fluoroquinoline (13c).

Triethylphosphite (1.7 equiv.) is added dropwise to a stirred solution of 13b (1 equiv.) in cyclohexane (0.05 M) at room temperature. The resulting mixture is heated to 80 °C and aged until the reaction is deemed complete. After cooling to room temperature, the reaction mixture is concentrated *in vacuo* and the crude residue is purified by flasl raphy to furnish the title compound 13c.

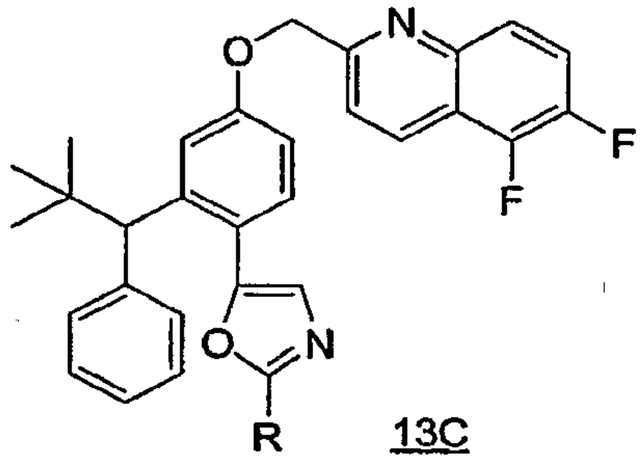
Following procedures similar to that described above for Example 13c, the following compounds can be prepared:



13A



13B



13C

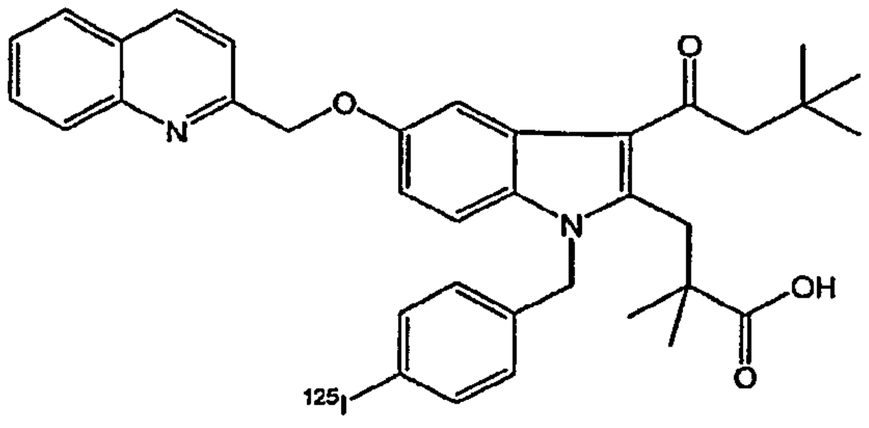
Ex. #13A	Ex. #13B	Ex. #13C	R
-	a	a	Me
b	b	b	Et
c[1]	c[1]	c[1]	CH_2OH
d[2]	d[2]	d[2]	CH_2F
e[3]	e[3]	e[3]	

[1]The alcohol functionality can be masked as a benzyl ether in Step B. Deprotection of the benzyl ether is achieved with methanesulfonic acid (according to *J. Am. Chem. Soc.*, **1996**, *118*, 4560) to provide the desired alcohol product.

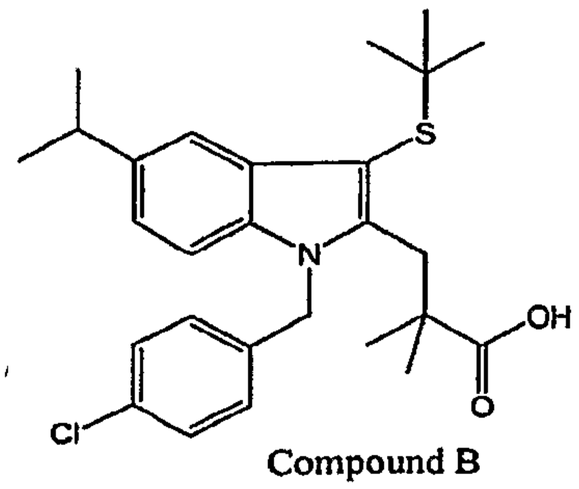
[2]Fluoroacetic anhydride derived from commercially available fluoracetic acid is used in Step B.

[3]3-Fluoropropanoic anhydride derived from commercially available 3-fluoropropanoic acid is used in Step B.

FLAP Binding Assay



Compound A



Compound B

A 100,000 x g pellet from human leukocyte 10,000 x g supernatants (1) is the source of FLAP. The 100,000 x g pellet membranes were resuspended in Tris-Tween assay buffer (100 mM Tris HCl pH 7.4, 140 mM NaCl, 2 mM EDTA, 0.5 mM dithiothreitol, 5% glycerol, 0.05% Tween 20) to yield a final protein concentration of 50 μg to 150 μg/ml. Aliquots (100 μl) of membrane suspension were added to 12 mm x 75 mm polypropylene tubes containing 100 μl Tris-Tween assay buffer, 30,000 cpm of Compound A in 5 μl MeOH:assay buffer (1:1), and 2 μl dimethyl sulfoxide or competitor (i.e., the compound to be tested) in dimethyl sulfoxide. Compound B (10 μM final concentration) was used to determine non-specific binding. After a 20 minute incubation at room temperature, tube contents were diluted to 4 ml with cold 0.1 M Tris HCl pH 7.4, 0.05% Tween 20 wash buffer and the membranes were collected by filtration of GFB filters presoaked in the wash buffer. Tubes and filters were rinsed with 2 x 4 ml aliquots of cold wash buffer. Filters were transferred to 12 mm x 3.5 mm polystyrene tubes for determination of radioactivity by gamma-scintillation counting.

Specific binding is defined as total binding minus non-specific binding. Total binding was Compound A bound to membranes in the absence of competitor; non-specific binding was Compound A bound in the presence of 10 uM Compound B. Preparation of Compound A is described in reference 1, below. The IC_{50} values were obtained by computer analysis (see reference 2, below) of the experimental data. Representative tested compounds of the invention were determined to have an IC_{50} < 1 uM, and preferred compounds had IC_{50} < 200 nM.

REFERENCES:

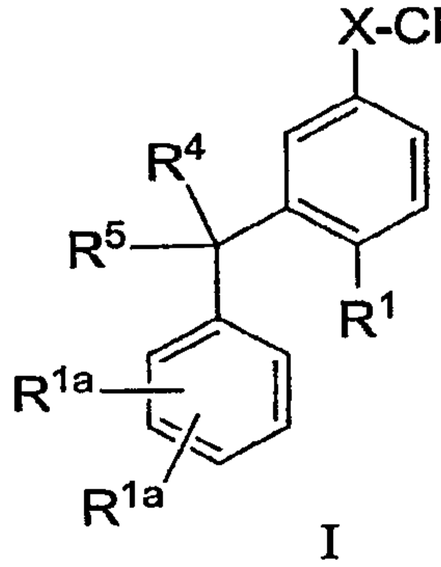
1. Charleson, S., Prasti, P., Leger, S., Gillard, J.W, Vickers, P.J., Mancini, J.A., Charleson, P., Guay, J., Ford-Hutchinson, A.W., and Evans, J.F. (1992) Characterization of a 5-lipoxygenase-activating protein binding assay: correlation of affinity for 5-lipoxygenase-activating protein with leukotriene synthesis inhibition. Mol Pharmacol 41:873-879.

2. Kinetic, EBDA, Ligand, Lowry: A collection of Radioligand Binding Analysis Programs by G.A. McPherson. Elsevier-BIOSOFT.

While the invention has been described with reference to certain particular embodiments thereof, numerous alternative embodiments will be apparent to those skilled in the art from the teachings described herein. All patents, patent applications and publications cited herein are incorporated by reference in their entirety.

WHAT IS CLAIMED IS:

1. A compound represented by Formula I:



I

or a pharmaceutically acceptable salt and/or solvate thereof, wherein:

each R^{1a} is independently selected from the group consisting of -H, -F, -Cl, -Br, -C_{1-6}alkyl, -CN, -OH, C_{1-6}alkyl-OH, -OC_{1-6} alkyl, -fluoroC_{1-6} alkyl, -fluoroC_{1-6} alkoxy, -NH_2, -NHC_{1-6}alkyl, -$N(C_{1-6}alkyl)_2$, -C_{1-6}alkyl-NH_2, -C_{1-6}alkyl-NHC_{1-6}alkyl, -C_{1-6}alkyl-$N(C_{1-6}alkyl)_2$, -$NHC(O)C_{1-6}$alkyl, -CO_2C_{1-6}alkyl, -$C(O)NHC_{1-6}$alkyl, and -$C(O)N(C_{1-6}alkyl)_2$;

R^1 is selected from the group consisting of:

a) Z^1,

b) -CO_2R^a, -$C(O)NR^aR^b$, -$N(R^a)_2$, -$NR^bSO_pR^a$, -$NR^bC(O)R^a$, -$NR^bC(O)NR^aR^b$, -$NR^bCO_2R^a$, -$OC(O)NR^aR^b$, -OH and -CN,

c) -C_{1-6}alkyl, -C_{2-6} alkenyl, -C_{2-6}alkynyl, -OC_{1-6}alkyl, -OC_{2-6}alkenyl and –OC_{2-6}alkynyl, said groups being optionally substituted with R^6 and optionally substituted with R^7,

wherein R^6 is selected from the group consisting of -CO_2R^a, -$C(O)NR^aR^b$, -$N(R^a)_2$, -$NR^bSO_pR^a$, -$NR^bC(O)R^a$, -$NR^bC(O)NR^aR^b$, -$NR^bCO_2R^a$, -$OC(O)NR^aR^b$, -$C(O)SO_pNR^aR^b$, -$C(O)NR^bNR^aR^b$, -$S(O)_pNR^aR^b$, -$SO_pNR^bC(O)R^a$, -$S(O)_pR^a$, –F , -CF_3, phenyl, Hetcy, and Z^1; and R^7 is selected from the group consisting of -F and –OH, and

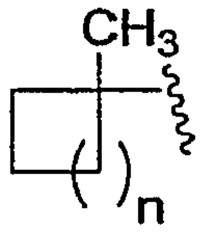
d) phenyl, optionally substituted with 1-2 members selected from the group consisting of: -F, -Cl, -C_{1-6}alkyl, -CN, -OH, -OC_{1-6} alkyl, -fluoroC_{1-6}alkyl, -fluoroC_{1-6}alkoxy, -NH_2, -NHC_{1-6}alkyl, -$N(C_{1-6}alkyl)_2$, -C_{1-6}alkyl-NH_2, -C_{1-6}alkyl-NHC_{1-6}alkyl, -C_{1-6}alkyl-$N(C_{1-6}alkyl)_2$, -C_{1-6}alkyl-CN, -$NHC(O)C_{1-6}$alkyl, -$C(O)NHC_{1-6}$alkyl, and -$C(O)N(C_{1-6}alkyl)_2$;

R^2 is selected from the group consisting of –H and -C_{1-6}alkyl optionally substituted with a group selected from –OH and –F;

R^3 is selected from the group consisting of –H and -C_{1-6}alkyl;

R^4 is selected from the group consisting of hydrogen, fluorine, hydroxy, C_{1-3} alkyl optionally substituted with one to five fluorines;

R^5 is selected from the group consisting of (a) C_{1-6} alkyl optionally substituted with one to five fluorines, (b) C_{3-6} cycloalkyl, and (c)



;

n is an integer selected from 0, 1, 2, and 3;

each "p" independently represents an integer selected from 0, 1 and 2;

each R^a is independently selected from the group consisting of

a) –H,

b) -C_{1-4}alkyl, -C_{2-4}alkenyl, and –C_{2-4}alkynyl, wherein each is optionally substituted with 1-2 members selected from the group consisting of: -OH, -OC_{1-4}alkyl, -CN, -NH_2, -NHC_{1-4}alkyl, and -$N(C_{1-4}alkyl)_2$, -F, and –CF_3,

c) phenyl and phenyl-C_{1-4}alkyl-, the phenyl moieties being optionally substituted with 1-2 members selected from the group consisting of -F, -Cl, -C_{1-4} alkyl, -CN, -OH, -OC_{1-4} alkyl, -fluoroC_{1-4}alkyl, -fluoroC_{1-4}alkoxy, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, -C_{1-4}alkyl-NH_2, -C_{1-4}alkyl-NHC_{1-4}alkyl, -C_{1-4}alkyl-$N(C_{1-4}alkyl)_2$, -C_{1-4}alkyl-CN, -$NHC(O)C_{1-4}$alkyl, -$C(O)NHC_{1-4}$alkyl, and -$C(O)N(C_{1-4}alkyl)_2$,

and the alkyl portion of phenyl-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of -OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, and 1-3 of fluoro,

d) Hetcy and Hetcy-C_{1-4}alkyl-, the Hetcy moieties being optionally substituted on carbon with 1-2 members selected from the group consisting of -F, -OH, -CO_2H, -C_{1-4}alkyl, -CO_2C_{1-4}alkyl, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$, -$NHC(O)C_{1-4}$alkyl, oxo, -$C(O)NHC_{1-4}$alkyl and -$C(O)N(C_{1-4}alkyl)_2$; and optionally substituted on nitrogen when present with a group selected from -C_{1-4} alkyl, and -C_{1-4} acyl,

and the alkyl portion of Hetcy-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of –OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$ and 1-3 of fluoro,

e) Z^2 and Z^2-C_{1-4}alkyl- and the alkyl portion of Z^2-C_{1-4}alkyl- being optionally substituted with a member selected from the group consisting of –OH, -CN, -OC_{1-4}alkyl, -NH_2, -NHC_{1-4}alkyl, -$N(C_{1-4}alkyl)_2$ and 1-3 of fluoro;

each R^b is independently selected from the group consisting of -H and -C_{1-4} alkyl optionally substituted with 1-2 members selected from the group consisting of NH_2, -OH, -F, -CN, and -CF_3;

X is selected from the group consisting of -O-, $S(O)_p$, NR^b, and $-CHR^8-$, wherein R^8 is selected from the group consisting of -H, -OH and $-C_{1-6}$ alkyl optionally substituted with a group selected from -OH and -F;

Y is selected from the group consisting of:

a) a 9-membered unsaturated *ortho*-fused bicyclic ring system containing 2-3 heteroatoms selected from the group consisting of –N=, –NH–, –N(Me)–, –S– and –O–, and wherein the ring system is optionally substituted with 1-3 of fluoro,

b) a 10-membered aromatic *ortho*-fused bicyclic ring system containing 1-3 of –N= , wherein the ring system is optionally substituted with 1-3 of fluoro, and

c) pyridinyl substituted with a group selected from $-C_{1-4}$alkyl, -F, $-CF_2H$, and CF_3, and optionally having a second substituent which is $-C_{1-4}$alkyl;

Hetcy is selected from the group consisting of azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl, morpholinyl, tetraydrofuranyl, β–lactamyl, δ–lactamyl and γ-lactamyl;

Z^1 is selected from the group consisting of:

a) a 5-membered unsaturated heterocyclic ring containing 2-4 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from $-C_{1-4}$alkyl and $-C_{1-4}$alkyl substituted with a group selected from $-NH_2$, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, $-NH_2$, $-CF_3$, -Cl, $-C_{1-4}$ alkyl and $-C_{1-4}$alkyl substituted with a group selected from $-NH_2$, -OH, $-OC_{1-4}$alkyl, -CN, and 1-3 of fluoro,

b) a 5-membered unsaturated heterocyclic ring containing 2-3 heteroatoms selected from one oxygen or one sulfur and 1-2 of nitrogen, wherein one nitrogen in the ring is optionally substituted with a group selected from C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from $–NH_2$, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from

-OH, -SH, =O, =S, -SMe, $-NH_2$, $-CF_3$, -Cl, $-C_{1-4}$alkyl, and C_{1-4}alkyl substituted with a group selected from $-NH_2$, -OH, $-OC_{1-4}$alkyl, -CN, and 1-3 of fluoro,

c) a 6-membered unsaturated heterocyclic ring containing 1-2 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from $-C_{1-4}$alkyl and $-C_{1-4}$alkyl substituted with a group selected from $–NH_2$, -OH, -CN, and 1-3 of fluoro, and one carbon atom in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, $-NH_2$, $-CF_3$, -Cl, $-C_{1-4}$alkyl and $-C_{1-4}$alkyl substituted with a group selected from $-NH_2$, -OH, $-OC_{1-4}$alkyl, -CN, and 1-3 of fluoro,

d) an 8-membered unsaturated ortho-fused bicyclic ring system containing 3-5 heteroatoms selected from one sulfur and 2-4 of nitrogen atoms wherein one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, $-NH_2$, $-CF_3$, -Cl, $-C_{1-4}$alkyl and C_{1-4}alkyl substituted with a group selected from $-NH_2$, -OH, $-OC_{1-4}$alkyl, -CN, and 1-3 of fluoro, and

e) a 9-membered unsaturated *ortho*-fused bicyclic ring system containing 3-4 nitrogen atoms, wherein one carbon in the ring is optionally substituted with a group selected from -OH, -SH,

-SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl, and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro; and

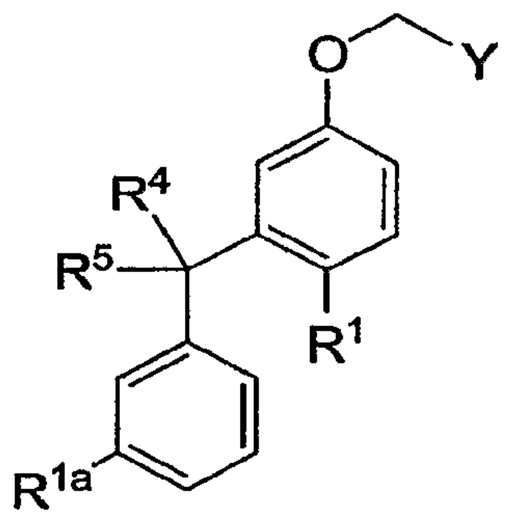
Z^2 is is selected from the group consisting of:

a) a 5-membered unsaturated heterocyclic ring containing 2-4 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, =O, =S, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl, and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro,

b) a 5-membered unsaturated heterocyclic ring containing 2-3 heteroatoms selected from one oxygen or one sulfur and 1-2 of nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from C_{1-4}alkyl and C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, and C_{1-4}alkyl optionally substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro, and

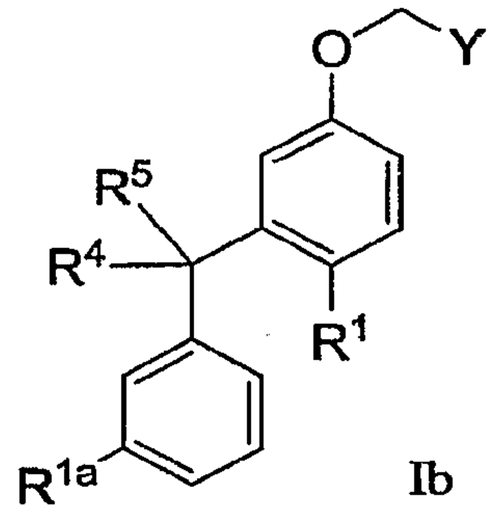
c) a 6-membered unsaturated heterocyclic ring containing 1-2 nitrogen atoms, wherein one nitrogen in the ring is optionally substituted with a group selected from -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro, and one carbon atom in the ring is optionally substituted with a group selected from -OH, -SH, -SMe, -NH_2, -CF_3, -Cl, -C_{1-4}alkyl and -C_{1-4}alkyl substituted with a group selected from -NH_2, -OH, -OC_{1-4}alkyl, -CN, and 1-3 of fluoro.

2. The compound of Claim 1 having structural Formula Ia:



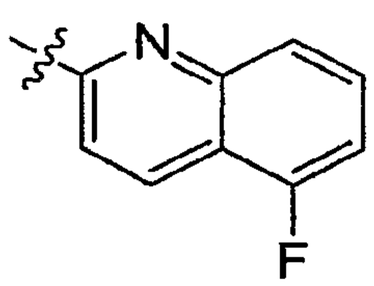
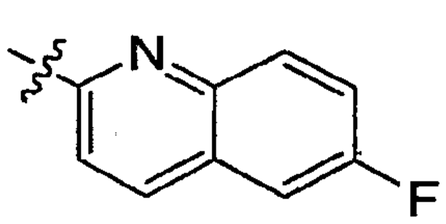
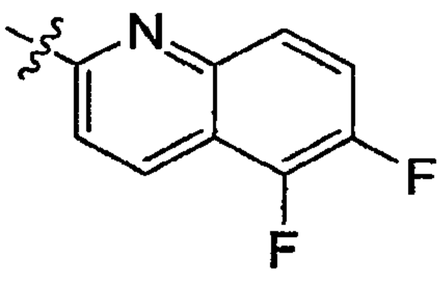
Ia

3. The compound of Claim 1 having structural Formula Ib:



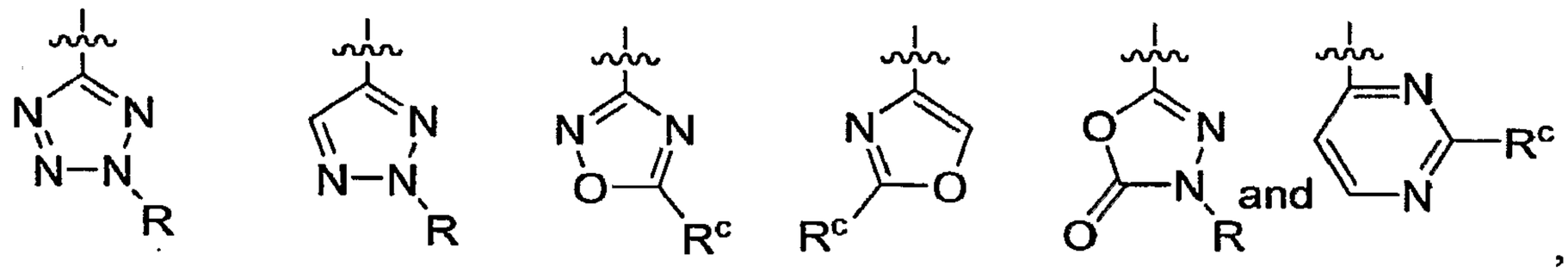
Ib

wherein Y is selected from group consisting of:


,

and

.

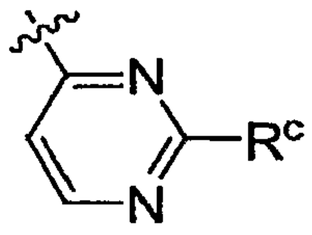
4. The compound of Claim 1 wherein R^1 is selected from the group consisting of -COOH, -$COOR^a$, -C(O)-NR^aR^b, -OC(O)-NR^aR^b, -$CH_2C(O)$-NR^aR^b, and Z^1.

5. The compound of Claim 4 wherein R^1 is selected from the group consisting of:
R^1 is selected from the tgroup consisting of



wherein R is selected from –H and -C_{1-4}alkyl optionally substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro; and R^c is selected from –H, methyl, -NH_2, OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

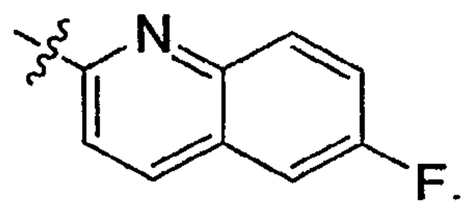
6. The compound of Claim 5 wherein R^1 is



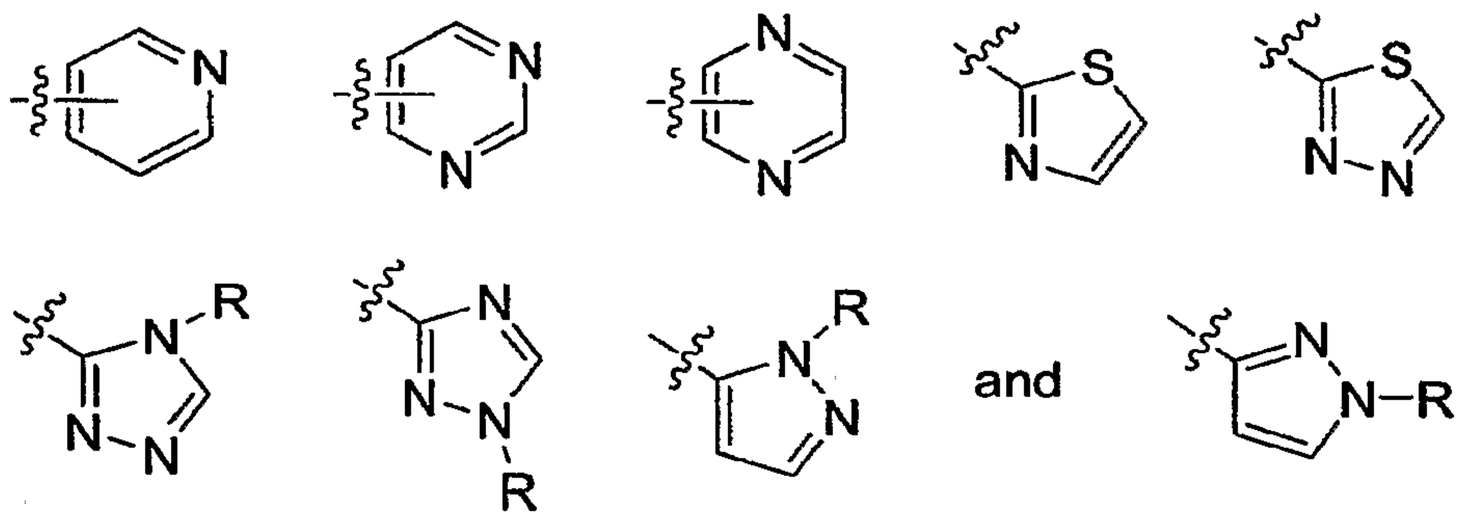
wherein R^c is selected from –H, methyl, -NH_2, -OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

7. The compound of Claim 1 wherein R^5 is *t*-butyl and R^4 is hydrogen.

8. The compound of Claim 1 wherein Y is

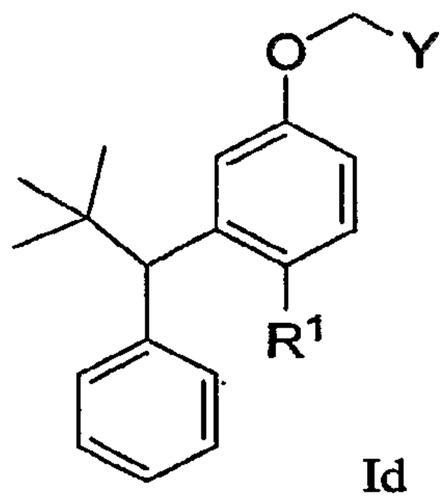


9. The compound of Claim 1 wherein Z^2 is selected from



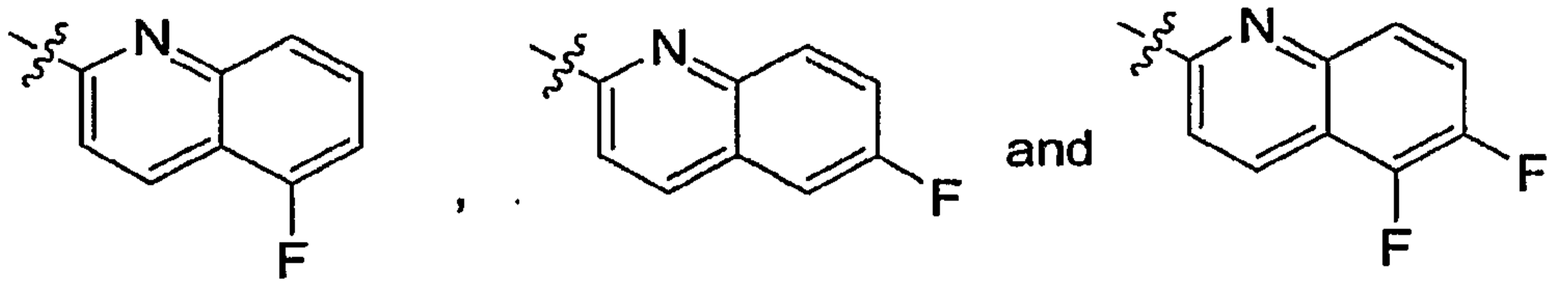
wherein R is selected from –H, methyl, ethyl, and –fluoroethyl.

10. The compound of Claim 1 of structural formula Id:

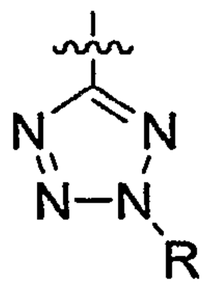
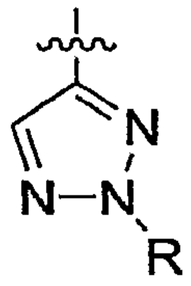
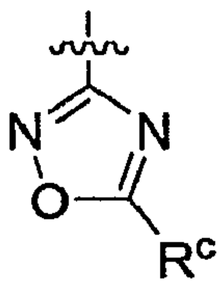
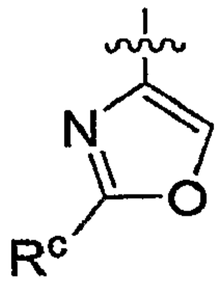
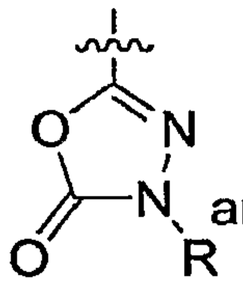
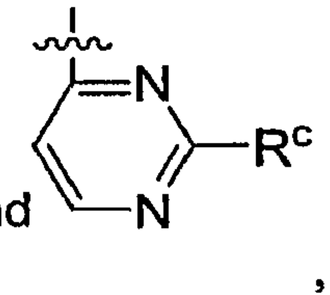


Id

wherein Y is selected from the group consisting of Y is selected from group consisting of:

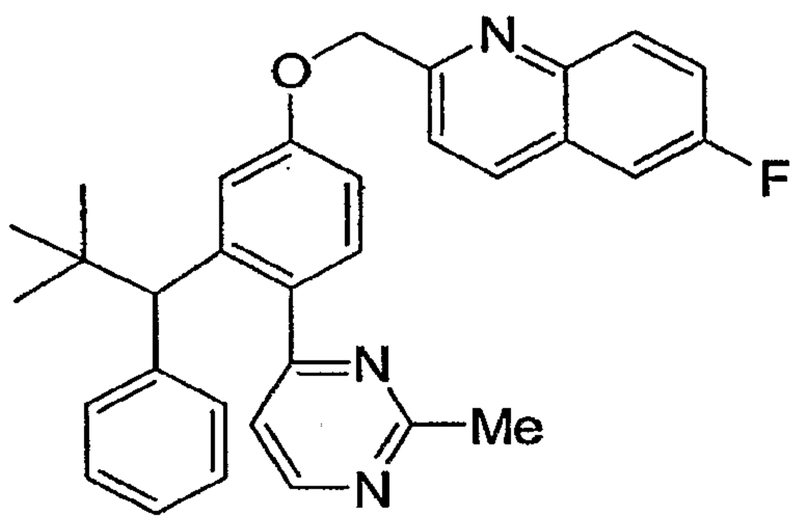
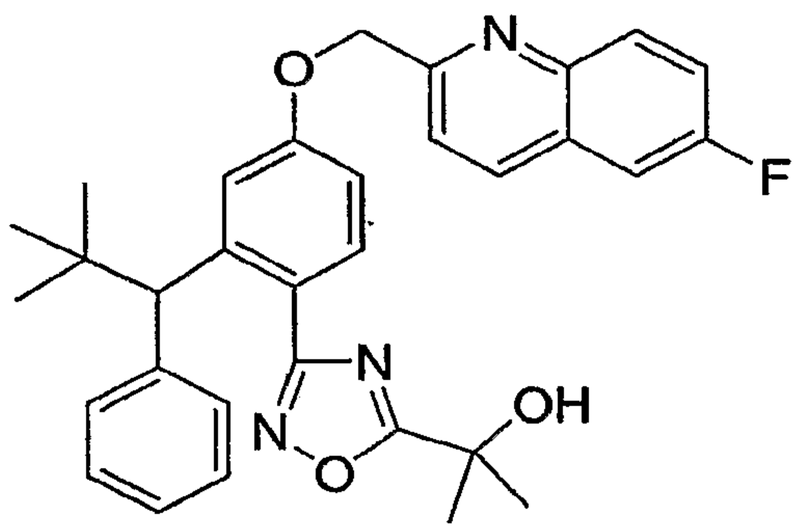
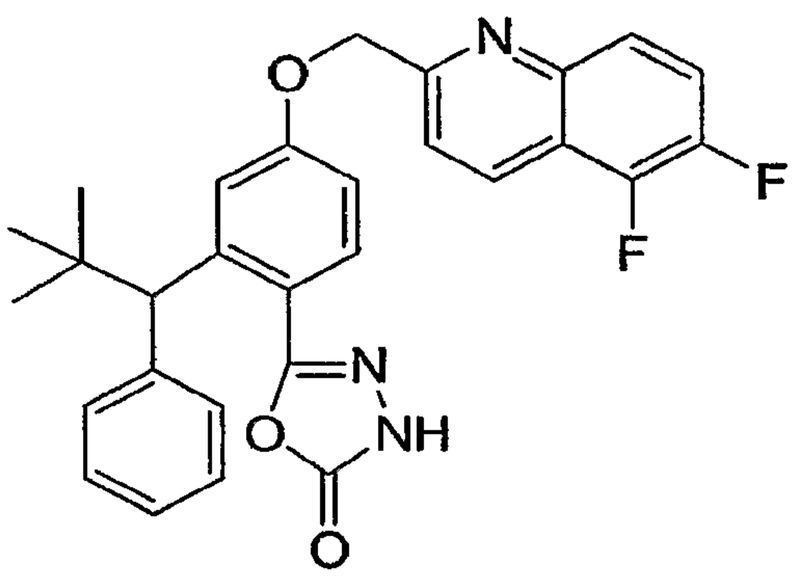
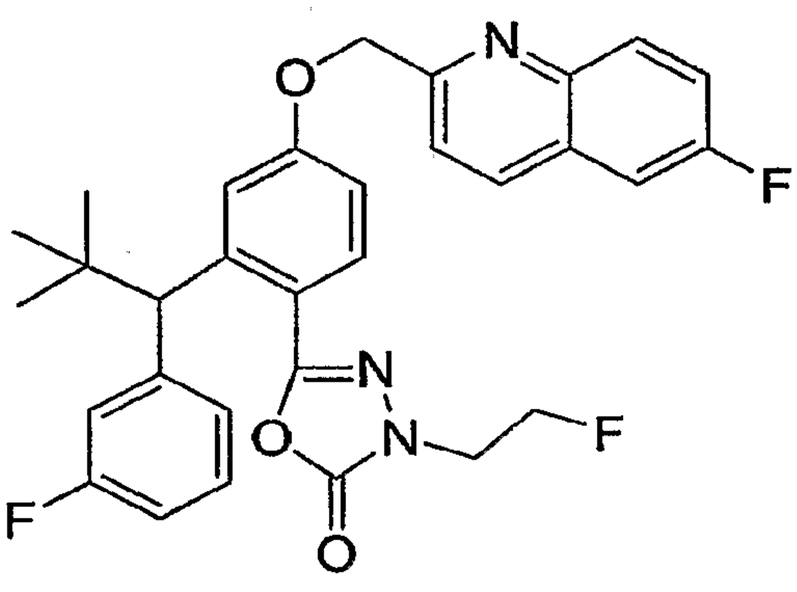


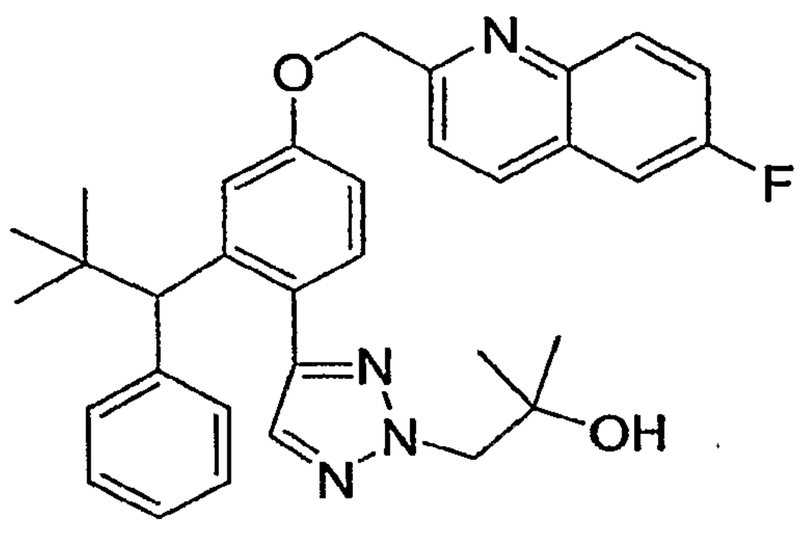
and R^1 is selected from the group consisting of:

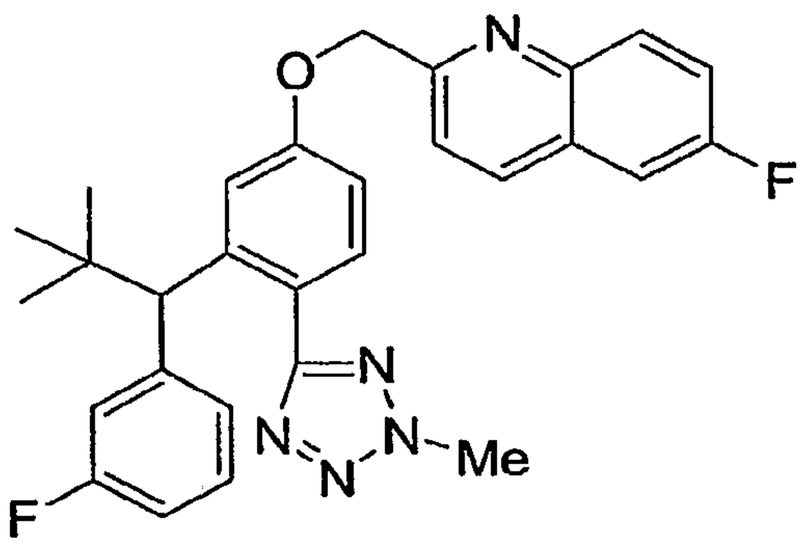





and

,

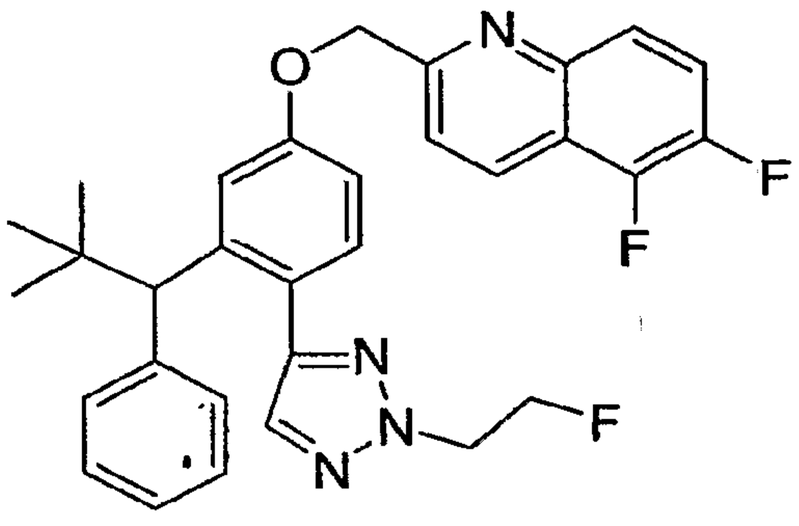
wherein R is selected from –H and -C_{1-4}alkyl optionally substituted with a group selected from –NH_2, -OH, -CN, and 1-3 of fluoro; and R^c is selected from –H, methyl, -NH_2, OH, -hydroxymethyl, fluoroethyl, and 1-methyl-1-hydroxyethyl.

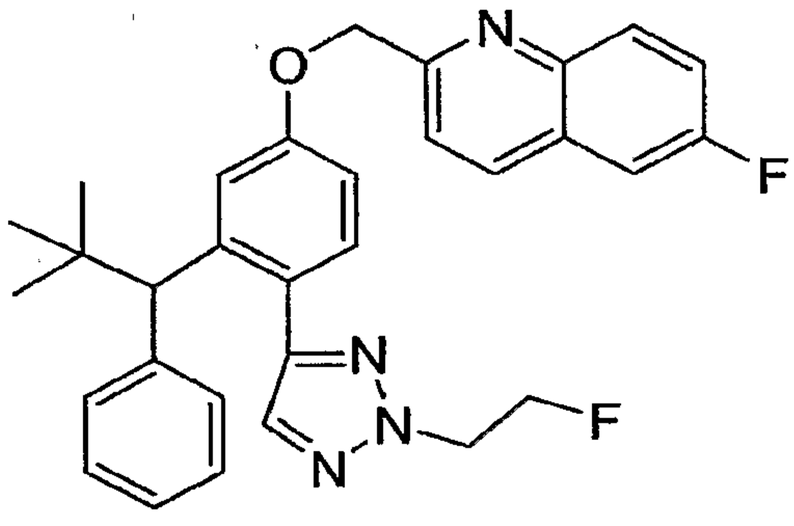
11. A compound which is selected from the group consisting of:

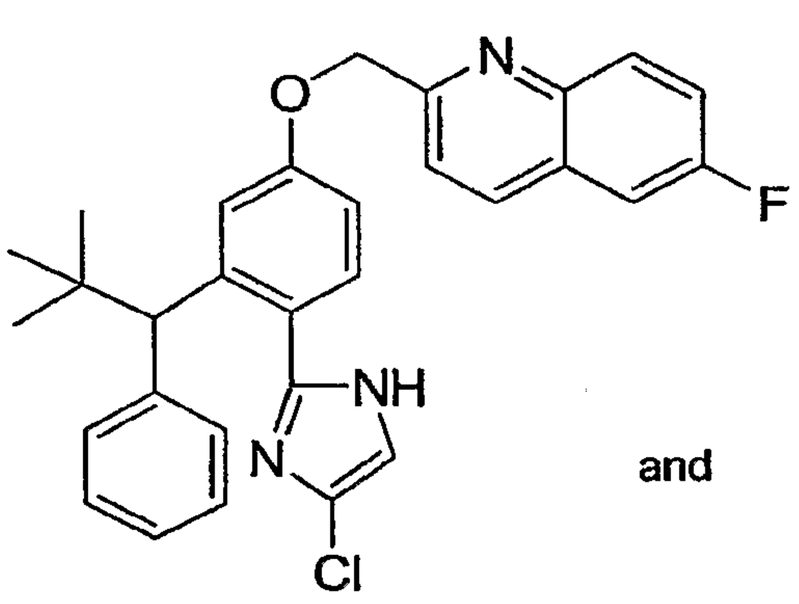







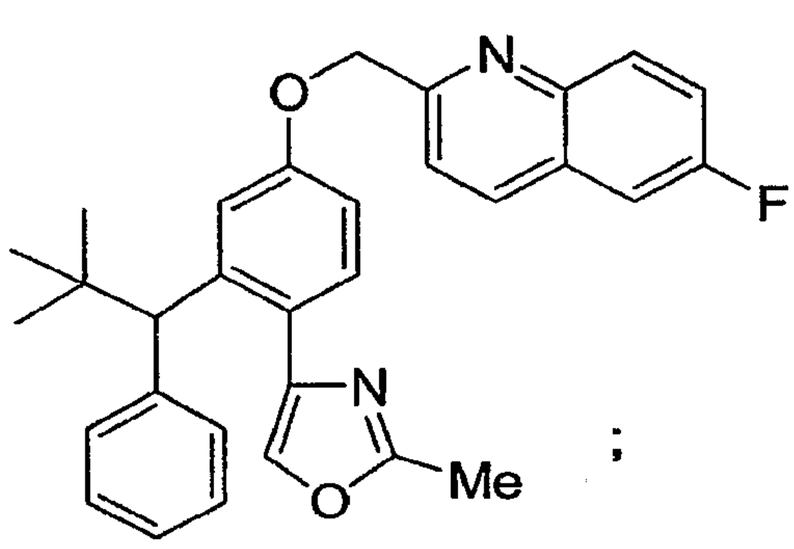








and



;

or a pharmaceutically acceptable salt and solvate thereof.

12. A pharmaceutical composition comprised of a therapeutically effective amount of a compound of Claim 1 and a pharmaceutically acceptable carrier.

13. A method for treating a leukotriene-mediated medical condition comprising administering a therapeutically effective amount of a compound of Claim 1 to a patient in need of such treatment.

14. A method for treating an inflammatory condition comprising administering a therapeutically effective amount of a compound of Claim 1 to a patient in need of such treatment.

15. A method for treating atherosclerosis comprising administering a therapeutically effective amount of a compound of Claim 1 to a patient in need of such treatment.

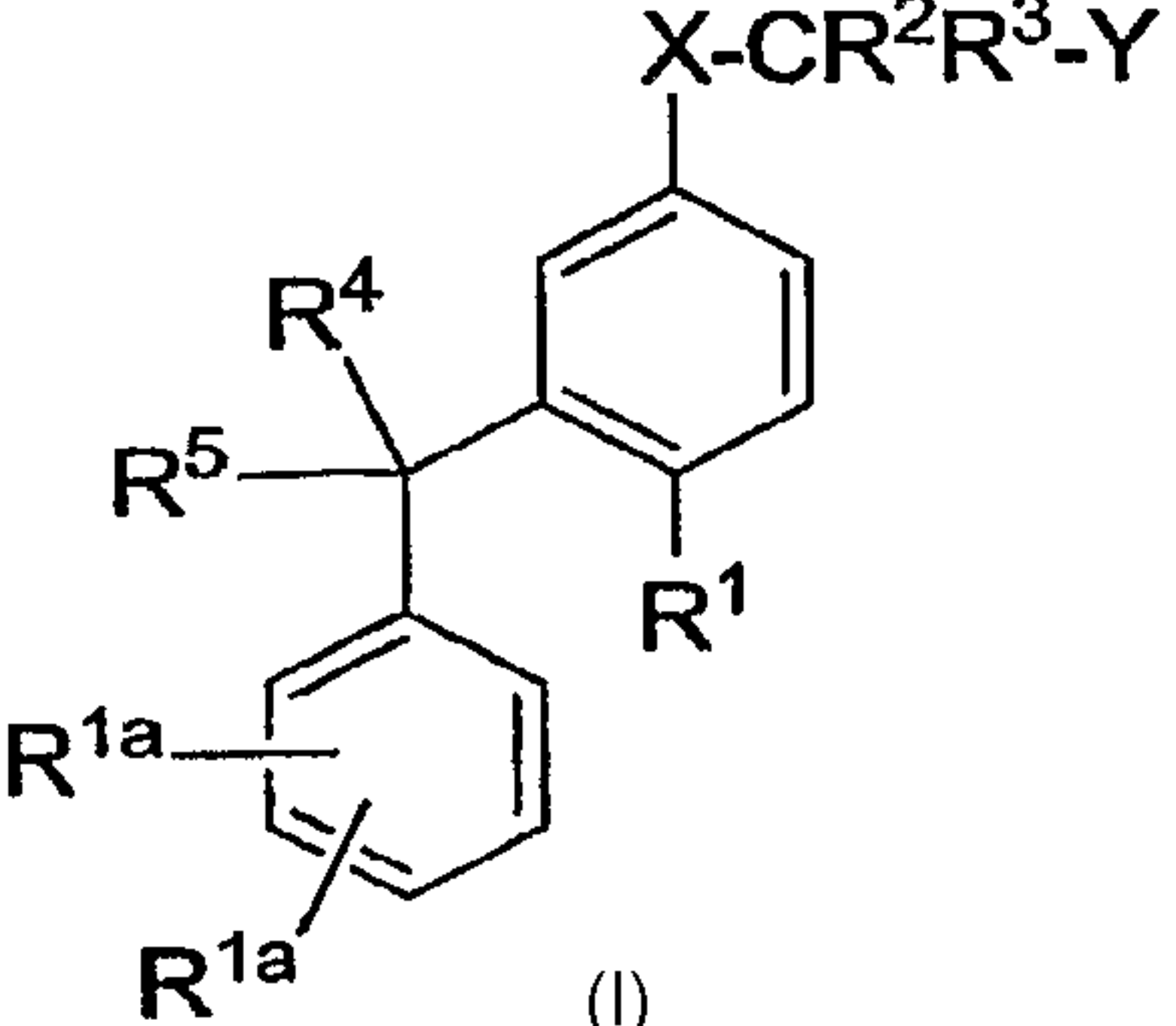
16. The method of Claim 15 for halting or slowing atherosclerotic plaque progression.

17. The method of Claim 15 for effecting regression of atherosclerotic plaque.

18. The method of Claim 15 for preventing or reducing the risk of atherosclerotic plaque rupture in a patient having atherosclerotic plaque.

19. A method for preventing or reducing the risk of an atherosclerotic disease event comprising administering a prophylactically effective amount of a compound of Claim 1 to a patient at risk for having an atherosclerotic disease event.

20. The method of treating atherosclerosis of Claim 15 further comprising administering to the patient a compound selected from the group consisting of an HMG-CoA reductase inhibitor, a cholesterol absorption inhibitor, a CETP inhibitor, a PPARγ agonist, a PPARα agonist, a PPAR dual α/γ agonist, and combinations thereof.



(I)