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CA 2237443 C 2009/07/14

(11)(21) **2 237 443**

(12) **BREVET CANADIEN
CANADIAN PATENT**

(13) **C**

(22) Date de dépôt/Filing Date: 1998/05/12

(41) Mise à la disp. pub./Open to Public Insp.: 1998/11/24

(45) Date de délivrance/Issue Date: 2009/07/14

(30) Priorité/Priority: 1997/05/24 (DE197 21 728.1)

(51) Cl.Int./Int.Cl. *C09D 133/08* (2006.01),
B01F 17/00 (2006.01), *C08F 8/32* (2006.01),
C09D 17/00 (2006.01), *C09D 7/02* (2006.01)

(72) Inventeurs/Inventors:

HAUBENNESTEL, KARLHEINZ, DE;
PRITSCHINS, WOLFGANG, DE;
ORTH, ULRICH, DE

(73) Propriétaire/Owner:

BYK-CHEMIE GMBH, DE

(74) Agent: KIRBY EADES GALE BAKER

(54) Titre : AGENTS DISPERSANTS POUR PIGMENTS ET DILUANTS A BASE DE POLYMERES D'ESTERS
ALKYLIQUES DE L'ACIDE ACRYLIQUE

(54) Title: DISPERSING AGENTS FOR PIGMENTS OR EXTENDERS BASED ON ACRYLIC ACID ALKYL ESTER
POLYMERS

(57) Abrégé/Abstract:

The present invention relates to dispersing agents for pigments or extenders based on acrylic acid alkyl ester polymers, wherein at least part of the ester groups of the polymers is reacted to form acid amides, wherein the dispersing agent is obtainable by the aminolysis of ester groups of the polymers by means of amines, wherein, for the aminolysis, at least one amine is used from the group comprising a) amines of general formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 - 4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals which comprise 1 - 12, preferably 1 - 6 carbon atoms and which can be the same or different, and b) amines of general formula $\text{NH}_2 - \text{R}^1 - \text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2 - 4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which can comprise up to 2 nitrogen atoms or which can additionally comprise oxygen. The invention also relates to the use of the dispersing agents according to the invention for the production of pigment concentrates and coating media, and to a process for producing the dispersing agents.

Byk Chemie, Wesel

Abstract

5 The present invention relates to dispersing agents for pigments or extenders based on acrylic acid alkyl ester polymers, wherein at least part of the ester groups of the polymers is reacted to form acid amides, wherein the dispersing agent is obtainable by the aminolysis of ester groups of the polymers by means of amines, wherein, for the aminolysis, at least one amine is used from the group comprising a) amines of general
10 formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 - 4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals which comprise 1 - 12, preferably 1 - 6 carbon atoms and which can be the same or different, and b)
15 amines of general formula $\text{NH}_2 - \text{R}^1 - \text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2 - 4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which can comprise up to 2 nitrogen atoms or which can additionally comprise oxygen.

The invention also relates to the use of the dispersing agents according to the invention for the production of pigment concentrates and coating media, and to a process for producing the dispersing agents.

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Byk Chemie, Wesel

5 Dispersing agents for pigments or extenders based on acrylic acid alkyl ester polymers

This invention relates to dispersing agents for pigments or extenders based on acrylic acid alkyl ester polymers, wherein at least part of the ester groups of the polymers are reacted to form acid amides, and to a process for producing dispersing agents of this 10 type. The invention also relates to the use of the dispersing agents for the production of pigment concentrates and coatings.

According to the prior art, polyacrylic acid esters comprising acidic and basic groups, which can also be converted into salts, are used as dispersing agents. These are 15 produced by the polymerisation of corresponding monomeric acrylic acid esters, such as ethyl acrylate, n-butyl acrylate, 2-ethylhexyl acrylate, lauryl acrylate, 2-hydroxyethyl acrylate and alkoxylation products thereof, and of acrylic acid and dimethylaminoethyl acrylate, for example. For these polymers, other monomers comprising vinyl double bonds can also be used in conjunction during polymerisation, such as styrene, 20 vinylpyridine, vinylimidazole and alkyl vinyl ethers for example (EP 0 311 157, US-PS 3 980 602).

However, proposals have also been made regarding how dispersing agents such as these can be produced by a transesterification reaction of polyacrylic acid esters, whereby the 25 alkyl group is replaced by longer chain, saturated or unsaturated alcohols, polyoxyalkylene monoalcohols or dialkylaminoalkanols (EP 0 595 129, DEP 3 906 702, DEP 3 842 201, DEP 3 842 202).

The advantage of transesterification is stated to be the considerably narrower molecular 30 weight distribution of polymers produced in this manner, compared with polymers

produced by the polymerisation of monomers. Moreover, it is claimed that these polymers are free from interfering and toxic monomer fractions.

These products according to the prior art have the disadvantage that defined monomers
5 which are particularly suitable for producing these polymers with a heterocyclic structure, for example, are not available commercially.

Secondly, it has been shown that the basic transesterification products described in EP 0 595 129 do not impart a satisfactory stability to dispersed pigments, i.e. the pigments
10 flocculate again after dispersion and do not exhibit the optimum transparency and gloss development.

It is an object of the present invention to provide polyacrylic acid compounds which do not have the above disadvantages or only to a substantially lesser extent and to enable
15 achieving a broad variety of polyacrylic acid compounds.

In this manner, acrylic acid ester/N-alkylacrylic acid amide copolymers can be obtained, the acrylic acid amides of which are not commercially available as
20 monomers.

Surprisingly, it has been shown that significant improvements in the dispersing and stabilising effect can be achieved for acrylate polymers such as these, which are produced by a polymer analogous reaction, if the corresponding amines are
25 used, at least in part, instead of alcohols during the reaction of the alkyl acrylates.

With suitable catalysts, such as paratoluenesulphonic acid or H_2SO_4 for example, aminolysis of the ester bond proceeds similarly to transesterification, with elimination of the alkanol from the polyalkyl acrylate.

In contrast, the aminolysis of acrylic acid/acrylic acid ester copolymers proceeds without further addition of a catalyst, since the carboxyl groups themselves are already catalytically active.

5 In accordance with one aspect of the present invention there is provided a dispersing agent for pigments or extenders comprising at least one acrylic acid alkyl ester polymer, wherein at least a portion of the ester groups of the at least one polymer are converted to acid amides. In preferred embodiments, the dispersing agent is obtained by reacting the at least a portion of ester groups with at least one amine,
10 wherein the at least one amine is selected from the group consisting of a) an amine of general formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals which comprise 1 – 12 carbon atoms and which are the same or different, and b) an amine of general formula $\text{NH}_2 - \text{R}^1 - \text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which comprises 1 or 2 nitrogen atoms, and which lacks oxygen, or
15 comprises 1 or 2 oxygen atoms.

The object of the present invention is achieved by the provision of dispersing agents
20 for pigments or extenders which are based on acrylic acid alkyl ester polymers, in which at least part of the ester groups of the polymer is reacted to form acid amides, wherein the dispersing agent is obtainable by the aminolysis of ester groups of the polymers by means of amines, wherein, for the aminolysis, at least one amine can be used from the group comprising a) amines of general formula $\text{NH}_2-\text{R}^1-\text{NR}^2\text{R}^3$,
25 wherein R^1 is a divalent alkylene radical comprising 2-4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals which comprise 1-12, preferably 1-6 carbon atoms and which can be the same or different, and b) amines of general formula $\text{NH}_2-\text{R}^1-\text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2-4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which can comprise up
30 to 2 nitrogen atoms or which can additionally comprise oxygen.

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In addition to these amines, one or more amines can also be used from the group comprising c) saturated or unsaturated aliphatic amines comprising 6-22 carbon atoms, d) alicyclic amines comprising up to 6 carbon atoms, e) aryl-substituted alkylamines and f) polyoxyalkylene amines $\text{NH}_2-\text{R}^1-\text{[O-R}^2-\text{]}_x\text{O-R}^3$, wherein R^1 is a 5 divalent alkylene radical comprising 2-3 carbon atoms, R^2 is a divalent alkylene radical comprising 2-4 carbon atoms and R^3 is an alkyl radical comprising 1-4 carbon atoms.

The acrylic acid alkyl ester polymers may contain, as a comonomer, one or more 10 monomers from the group comprising itaconic acid esters, maleic acid esters, methacrylic acid esters, (meth)acrylic acid, styrene, alkyl vinyl ethers, vinyl acetate or mixtures thereof, and preferably have a weight average molecular weight of 1,000-50,000. At lower molecular weights, the acrylic acid ester polymers

lose some efficiency; at higher molecular weights their processability decreases considerably due to higher viscosities. The acrylic acid alkyl ester polymers most preferably have a molecular weight of 2000 to 20,000.

5 The dispersing agents can also be used in the form of salts of the amino-functional polymers produced by aminolysis with a) and/or b). Phosphoric acid or phosphoric acid esters and/or sulphonic acids and/or carboxylic acids are suitable for salt formation.

Components a) and/or b) and optionally c) to f) are used in amounts such that 10 preferably 1 to 50 %, more preferably 5 to 40 % and most preferably 10 to 30 % of the groups which are capable of amidisation are reacted.

15 Aliphatic diamines comprising a tertiary amino group and a primary or secondary amino group, such as 2-(diethylamino)ethylamine, 3-(dimethylamino)propylamine, 3-(diethylamino)propylamine or 1-diethylamino-4-aminopentane for example, are used as component a) in the polyacrylates to be used according to the invention. A particularly preferred component a) is $\text{NH}_2\text{-(CH}_2\text{)}_3\text{-N(CH}_3\text{)}_2$ (dimethyl-aminopropylamine).

20 Heterocyclic amines comprising an additional exocyclic primary or secondary amino group, such as N-(3-aminopropyl)imidazole, N-(3-aminopropyl)morpholine or N-(2-aminoethyl)-piperidine for example, are used as components b). A particularly preferred component b) is N-(3 -aminopropyl)imidazole.

25 Components a) and b) can be used on their own or jointly in any desired ratio.

In addition, components c) - f) can be used on their own or in admixture for the polymer analogous reaction.

30 2-ethylhexylamine, oleylamine or stearylamine can be used as component c) for example; cyclohexylamine or dicyclohexylamine can be used as component d) for

example; benzylamine can be used as component e) for example; the polyoxyalkylene amines which are known by trade name Jeffamin® can be used as component f) for example.

5 In this respect, it does not matter whether the amines are reacted as a mixture or successively in several steps.

In addition, alcohols can also be incorporated by transesterification into the polymers according to the invention during aminolysis with the amines described in a) - f). This 10 is advantageous if solubility and/or compatibility problems arise at high degrees of conversion of the ester groups with the amines listed under a) - f) to form the corresponding amides. In situations such as these, the remaining (C₁-C₄) alkyl ester groups can be replaced by longer chain substituents, such as oleyl alcohol, stearyl alcohol or benzyl alcohol for example, by transesterification.

15

Aminolysis of the polyacrylic acid esters proceeds in the manner known in the art, in the presence of catalysts, such as p-toluenesulphonic acid or H₂SO₄ for example, at reaction temperatures of about 130 - 210°C, optionally in the presence of a suitable solvent. The alcohol which is evolved in the course of this procedure can be distilled 20 off, during the reaction or after the reaction is complete, according to choice, or can remain in the reaction mixture.

The dispersing agents according to the invention normally used in an amount of 0.5 to 100, preferably 10 to 75, most preferably 25 - 60 parts by weight, with respect to 100 25 parts by weight of the solid to be dispersed. However, this depends on the surface area of the solid to be dispersed which is to be covered. For example, carbon black requires higher amounts of dispersing agents than does TiO₂. 100 parts by weight can also be exceeded in this respect. The dispersing agents are generally deposited on the solids in the presence of organic solvents and/or water. However, they 30 can also be deposited directly on the solids to be dispersed.

The solids which can be used are the organic and inorganic pigments which are known to one skilled in the art, and which are listed both in the Pigment Handbook, Vol. 1-3, John Wiley & Sons, New York 1988 and in Ullmann's Encyclopedia of Industrial Chemistry, 5th Edition, Vol. 20, pp. 213 *et seq.* (Phthalocyanines) and pp. 371 *et seq.* (Pigments, Organic). Carbon black, TiO_2 , iron oxide pigments, phthalocyanines and azo pigments can be cited as examples here. Moreover, mineral extenders, e.g. calcium carbonate or calcium oxide, and also flame retardants, such as aluminium or magnesium hydroxide for example, can be dispersed. Matting agents, such as hydrated silicas for example, can also be dispersed and stabilised in an outstanding manner. Other examples are described in EP 0 318 999, for example.

The dispersing agents according to the invention are particularly suitable for the production of pigment concentrates. For this purpose, the compounds according to the invention are introduced into an organic solvent and/or water, and the pigments to be dispersed are added with stirring. In addition, these pigment concentrates may contain binder vehicles and/or other adjuvant substances. These pigment concentrates can be incorporated in various binder vehicles, such as alkyd resins, polyester resins, acrylate resins, polyurethane resins or epoxy resins for example. However, pigments can also be dispersed directly, solvent free, in the polyacrylates according to the invention, and are then particularly suitable for the pigmentation of thermoplastic and thermosetting plastics formulations.

The dispersing agents according to the invention are particularly suitable for the production of coatings, wherein a binder vehicle, optionally a solvent, pigments and/or extenders, the dispersing agent and optionally adjuvant substances are dispersed together.

Production examples:

Production of polyacrylates (not according to the invention)

The polyacrylates to be used for aminolysis were produced by methods which are generally known, e.g. by radical polymerisation. More special production methods such 5 as anionic polymerisation or group transfer polymerisation could also be employed.

Production of acrylic acid ester/N-alkylacrylic acid amide copolymers by aminolysis (according to the invention)**10 Example 1:**

1600 g of a polyethyl acrylate with a weight average molecular weight of about 11,500 were mixed with 160 g diethylaminopropylamine. In addition, 3.6 g p-toluenesulphonic acid were added as a catalyst. The reaction mixture was heated to reflux (about 180°C) 15 under nitrogen. The boiling point fell to about 139°C, due to the ethanol evolved during the reaction. After about 12-15 hours the reaction was complete and the ethanol which was evolved was distilled off. The polymer obtained in this manner had a weight average molecular weight of 12,550 g/mole.

20 Example 2:

1450 g of a poly-n-butyl acrylate with a weight average molecular weight of about 12,000 were mixed with 177 g dimethylaminopropylamine. In addition, 0.5 g p-toluenesulphonic acid were added as a catalyst. The reaction mixture was heated to 25 reflux (about 180°C) under nitrogen. The boiling point fell to about 130°C, due to the n-butanol evolved during the reaction. After about 15-17 hours the reaction was complete and the n-butanol which was evolved was distilled off. The polymer obtained in this manner had a weight average molecular weight of 12,850 g/mole.

Example 3:

52.9 g of one of the styrene/acrylate/acrylic acid copolymers, which had an acid number of about 53 (mg/g KOH) (Joncryl 611/S.C.Johnson Polymer) and a weight average molecular weight of about 7700, were dissolved in 59.15 g PMA and treated with 6.25 g (0.05 moles) aminopropylimidazole. The reaction mixture was heated to reflux (about 145°C) under nitrogen. The reaction was complete after about 4-6 hours. The polymer obtained in this manner had a weight average molecular weight of 8250 g/mole. The acid number remained the same.

10 Example 4:

100 g of the polymer produced as in example 1 were mixed with 10.7 g (about 0.1 mole) benzylamine. The reaction mixture was heated to reflux (about 200°C) under nitrogen. After about 8 hours the reaction was complete and the ethanol which was 15 evolved was distilled off.

Example 5:

100 g of the polymer produced as in example 1 were mixed with 26.7 g (about 0.1 mole) oleylamine. The reaction mixture was heated to reflux (about 200°C) under 20 nitrogen. After about 8 hours the reaction was complete and the ethanol which was evolved was distilled off.

Example 6:

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100 g of the polymer produced as in example 2 were mixed with 10.7 g (about 0.1 mole) benzylamine. The reaction mixture was heated to reflux (about 200°C) under nitrogen. After about 8 hours the reaction was complete and the n-butanol which was evolved was distilled off.

30

Example 7:

100 g of the polymer produced as in example 2 were mixed with 26.7 g (about 0.1 mole) oleylamine. The reaction mixture was heated to reflux (about 200°C) under 5 nitrogen. After about 8 hours the reaction was complete and the n-butanol which was evolved was distilled off.

Example 8:

10 100 g of the polymer produced as in example 1 were mixed with 62.3 g (about 0.1 mole) of a polyoxyalkylene amine (Pluriol® A520A/BASF). The reaction mixture was heated to reflux (about 200°C) under nitrogen. After about 5 hours the reaction was complete and the ethanol which was evolved was distilled off.

15 Example 9:

100 g of the polymer produced as in example 2 were mixed with 62.3 g (about 0.1 mole) of a polyoxyalkylene amine (Pluriol® A520A/BASF). The reaction mixture was heated to reflux (about 200°C) under nitrogen. After about 5 hours the reaction was 20 complete and the n-butanol which was evolved was distilled off.

Example 10:

98.71 g of the polymer produced as in example 1 were mixed with 34.55 g 25 methoxypolyethylene glycol (MW about 350). The reaction mixture was heated to 100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at 200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was stirred for a further 2 hours at 200°C. The ethanol which was evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the 30 batch was stirred for 1 hour at 200°C. The ethanol which was evolved was distilled off.

Example 11:

100.91 g of the polymer produced as in example 2 were mixed with 35.32 g
5 methoxypolyethylene glycol (MW about 350). The reaction mixture was heated to
100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch
was stirred for 2 hours at 200°C. A further 0.33 g tetraisopropyl orthotitanate were then
added and the batch was stirred for a further 2 hours at 200°C. The n-butanol which was
evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the
10 batch was stirred for 1 hour at 200°C. The n-butanol which was evolved was distilled
off.

Example 12:

15 94.80 g of the polymer produced as in example 1 were mixed with 20.15 g (about 0.1
mole) butyltriglycol. The reaction mixture was heated to 100°C under nitrogen. Then
0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at
200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was
stirred for a further 2 hours at 200°C. The ethanol which was evolved was distilled off.
20 0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour
at 200°C. The ethanol which was evolved was distilled off.

Example 13:

25 105.14 g of the polymer produced as in example 2 were mixed with 22.34 g
butyltriglycol. The reaction mixture was heated to 100°C under nitrogen. Then
0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at
200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was
stirred for a further 2 hours at 200°C. The n-butanol which was evolved was distilled off.

0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour at 200°C. The n-butanol which was evolved was distilled off.

Example 14:

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98.20 g of the polymer produced as in example 1 were mixed with 26.37 g (about 0.1 mole) oleyl alcohol. The reaction mixture was heated to 100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at 200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was 10 stirred for a further 2 hours at 200°C. The ethanol which was evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour at 200°C. The ethanol which was evolved was distilled off.

Example 15:

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93.50 g of the polymer produced as in example 2 were mixed with 25.10 g (about 0.1 mole) oleyl alcohol. The reaction mixture was heated to 100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at 200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was 20 stirred for a further 2 hours at 200°C. The n-butanol which was evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour at 200°C. The n-butanol which was evolved was distilled off.

Example 16:

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104.52 g of the polymer produced as in example 1 were mixed with 21.95 g (about 0.1 mole) of an allyl polyether (Uniox PKA 5001/Nippon OIL & FATS Co., Ltd.). The reaction mixture was heated to 100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at 200°C. A further 0.33 g 30 tetraisopropyl orthotitanate were then added and the batch was stirred for a further 2

hours at 200°C. The ethanol which was evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour at 200°C. The ethanol which was evolved was distilled off.

5 Example 17:

98.45 g of the polymer produced as in example 2 were mixed with 20.67 g (about 0.1 mole) of an allyl polyether (Uniox PKA 5001/Nippon OIL & FATS Co., Ltd.). The reaction mixture was heated to 100°C under nitrogen. Then 0.33 g tetraisopropyl orthotitanate were added and the batch was stirred for 2 hours at 200°C. A further 0.33 g tetraisopropyl orthotitanate were then added and the batch was stirred for a further 2 hours at 200°C. The n-butanol which was evolved was distilled off. 0.17 g tetraisopropyl orthotitanate was then added and the batch was stirred for 1 hour at 200°C. The n-butanol which was evolved was distilled off.

15

Example 18:

1450 g of a polyethyl acrylate with a weight average molecular weight of about 49,000 were mixed with 14.3 g dimethylaminopropylamine. In addition, 0.7 g p-toluenesulphonic acid were added as a catalyst. The reaction mixture was heated to reflux (about 180°C) under nitrogen. The boiling point fell to about 130°C, due to the ethanol evolved during the reaction. After about 15-17 hours the reaction was complete and the ethanol which was evolved was distilled off. The polymer obtained in this manner had a weight average molecular weight of 49,500 g/mole.

25

Comparative example

Production of a comparison polymer by transesterification (not according to the invention).

30

73.8 g of a polyethyl acrylate were mixed with 20 g toluene, 53.6 g oleyl alcohol, 23.4 g diethyl ethanolamine and 0.7 g tetraisopropyl orthotitanate and were stirred for 2 hours under nitrogen and under reflux (about 150°C). A further 0.7 g tetraisopropyl orthotitanate were then added and the batch was again stirred for 2 hours under reflux 5 (the temperature fell to about 124°C). The ethanol which was evolved was distilled off with the toluene.

Examples of use

10 In order to assess the compounds according to the invention, pigment concentrates were prepared without additional binder vehicles, and their viscosity, which is a measure of the dispersibility of a solid, was determined.

15 The viscosities of the pigment pastes were measured using a Rheolab MC 10 (supplied by Physica/Pacer). The gloss was determined according to DIN 67530. The haze parameter was determined using the "Haze-Gloss" instrument supplied by the Byk-Gardner company. Delta E was determined according to DIN 5033 using the "TCS" instrument supplied by the Byk-Gardner company

20 For the preparation of the pigment pastes, the raw materials and the glass beads which were necessary for comminution were successively weighed into a 500 ml dispersing pot and were then dispersed for 40 minutes at 40°C, using a 40 mm Teflon disc.

25 The dispersed pastes were sieved through a paper sieve (mesh aperture 80 µm) and were introduced into glass bottles. Free-flowing pigment pastes were obtained, which had very good rheological properties.

These pigment pastes were then added to a two-component, isocyanate-crosslinking acrylate solution (Macrynal SM515; 70 % in butyl acetate/Desmodur N75; 75 % in

xylene/MPA (1:1)) and to a white lacquer, which was also based on Macrynal SM515/Desmodur N75. High-gloss lacquer films were obtained.

Formulations:

Pigment pastes: data in grams

	Spezial- schwarz 4	Bayferrox 130M	Novopermrot F3RK70	Heliogenblau L7101F
Dowanol PMA	58.0	21.5	61.5	67.0
Dispersing agent	12.0	8.5	8.5	13
Aerosil 200		1.00		
Spezialschwarz 4	30.0			
Bayferrox 130M		69.0		
Novopermrot F3RK70			30.0	
Heliogenblau L7101F				20.0
Glass beads	150	100	150	150
Total	100.0	100.0	100.0	100.0

Formulations:

Lacquer compositions: data in grams

5 Two-component clear lacquer

	Spezial- schwarz 4	Bayferrox 130M	Novopermrot F3RK70	Heliogenblau L7101F
Clear lacquer	12.40	11.50	11.60	10.70
Hardener solution (Desmodur N 75)	6.10	5.70	5.80	5.30
Pigment paste	1.50	2.80	2.60	4.00
Total	20.00	20.00	20.00	20.00

Two-component white lacquer

	Spezial- schwarz 4	Bayferrox 130M	Novopermrot F3RK70	Heliogenblau L7101F
Clear lacquer	15.30	14.90	14.70	14.70
Hardener solution (Desmodur N 75)	4.40	4.10	4.00	4.00
Pigment paste	0.30	1.00	1.30	1.30
Total	20.00	20.00	20.00	20.00

Results

Paste viscosity (in mPas) at D[sec⁻¹]

		D = 10[sec ⁻¹]	D = 100[sec ⁻¹]	D = 1000[sec ⁻¹]
Spezialschwarz 4	Comp. Example	36182	4327	535
	Example 1	14096	2204	2093
	Example 14	5140	988	243
	Example 15	10819	1572	308
Bayferrox 130M	Comp. Example	2747	984	421
	Example 1	2138	734	382
	Example 14	2435	762	416
	Example 15	1938	726	371
Novoperm F3RK70	Comp. Example	3216	1227	293
	Example 1	2425	987	208
	Example 14	2200	599	141
	Example 15	3095	872	206
Heliogenblau L7101F	Comp. Example	10307	2353	375
	Example 1	9360	2176	332
	Example 14	9993	1620	250
	Example 15	9450	1825	283

Gloss and haze in colouring lacquer; delta E in the white mixture

		Gloss	Haze	Delta E
Spezialschwarz 4	Comp. Example	12	374	22.83
	Example 11	89	6	0.7
	Example 13	88	7	0.5
	Example 14	92	7	1.14
	Example 15	65	62	5.84
Bayferrox 130M	Comp. Example	4	197	35.03
	Example 11	89	7	0.7
	Example 13	89	7	0.6
	Example 14	92	13	0.43
	Example 15	93	14	0.71
Novoperm F3RK70	Comp. Example	15	461	24.59
	Example 11	89	8	1.9
	Example 13	89	8	2.0
	Example 14	93	18	5.46
	Example 15	87	79	4.02
Heliogenblau L7101F	Comp. Example	19	512	13.39
	Example 11	89	10	8.5
	Example 13	89	10	8.3
	Example 14	92	23	2.47
	Example 15	92	14	0.91

Claims

1. A dispersing agent for pigments or extenders comprising at least one acrylic acid alkyl ester polymer, wherein at least a portion of the ester groups of the 5 at least one polymer are converted to acid amides.
2. The dispersing agent of claim 1, wherein the dispersing agent is obtained by reacting the at least a portion of ester groups with at least one amine, wherein the at least one amine is selected from the group consisting of a) an 10 amine of general formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals which comprise 1 – 12 carbon atoms and which are the same or different, and b) an amine of general formula $\text{NH}_2 - \text{R}^1 - \text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which 15 comprises 1 or 2 nitrogen atoms, and which lacks oxygen, or comprises 1 or 2 oxygen atoms.
3. The dispersing agent according to claim 2, wherein the at least one amine further comprises at least one amine selected from the group consisting of c) 20 a saturated or unsaturated aliphatic amine comprising 6 – 22 carbon atoms, d) an alicyclic amine comprising up to 6 carbon atoms, e) an aryl-substituted alkylamine, and f) a polyoxyalkylene amine $\text{NH}_2\text{-R}^1\text{-[O - R}^2\text{-)}_x\text{O - R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 – 3 carbon atoms, R^2 is a divalent alkylene radical comprising 2 – 4 carbon atoms and R^3 is an 25 alkyl radical comprising 1 – 4 carbon atoms.
4. The dispersing agent according to any one of claims 1 to 3, wherein the polymer contains, as a comonomer, at least one further monomer comprising vinyl double bonds.

5. The dispersing agent according to claim 4, wherein the comonomer is a monomer selected from the group consisting of itaconic acid esters, maleic acid esters, methacrylic acid esters, (meth)acrylic acid, styrene, alkyl vinyl ethers, vinyl acetate and mixtures thereof of monomers from this group.
- 10 6. The dispersing agent according to any one of claims 1 to 5, wherein the alkyl radical of the acrylic acid ester contains 1 – 4 carbon atoms.
7. The dispersing agent according to any one of claims 1 to 6, wherein a catalyst is used for aminolysis
- 15 8. The dispersing agent according to any one of claims 1 to 7, wherein the acrylic acid alkyl ester polymers have an average molecular weight of 1,000 - 50,000.
9. The dispersing agent according to claim 8, wherein the acrylic acid alkyl ester polymers have an average molecular weight of 2,000 - 20,000.
- 20 10. The dispersing agent according to any one of claims 1 to 9, wherein the product obtained by aminolysis with an amine of general formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$ and/or $\text{NH}_2 - \text{R}^1 - \text{Z}$ is reacted to form a salt.
11. The dispersing agent according to any one of claims 1 to 10, wherein R^2 and R^3 comprise 1 - 6 carbon atoms.
- 25 12. Use of a dispersing agent as claimed in any one of claims 1 to 11 for the production of pigment concentrates, wherein the dispersing agent is homogenized together with the pigments and/or extenders to be dispersed.
- 30 13. Use of claim 12, wherein the dispersing agent is homogenized in the presence of organic solvents and/or water.

14. Use of claim 12 or 13, wherein the dispersing agent is homogenized in the presence of binder vehicles.
- 5 15. Use of claim 12, 13 or 14, wherein the dispersing agent is homogenized in the presence of customary lacquer adjuvant substances.
- 10 16. Use of a dispersing agent as claimed in any one of claims 1 to 11 for the production of a coating medium, wherein a binder vehicle, pigments and/or extenders, and the dispersing agent are dispersed together.
17. Use of claim 16, wherein a solvent is also present.
- 15 18. Use of claim 16 or 17, wherein at least one adjuvant substance is also present.
19. A process for producing a dispersing agent as claimed in any one of claims 1 to 11, comprising a step of aminolysing an acrylic acid ester polymer with at least one amine, wherein at least a portion of the ester groups of the polymer are reacted with at least one amine to form acid amides, wherein said at least one amine is selected from the group consisting of: a) an amine of general formula $\text{NH}_2 - \text{R}^1 - \text{NR}^2\text{R}^3$ wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and R^2 and R^3 are aliphatic and/or alicyclic alkyl radicals comprising 1 – 12 carbon atoms which can be the same or different, and b) an amine of general formula $\text{NH}_2\text{R}^1 - \text{Z}$, wherein R^1 is a divalent alkylene radical comprising 2 – 4 carbon atoms and Z is a 5- or 6-membered N-containing heterocycle which can comprise up to 2 nitrogen atoms and which lacks oxygen or comprises 1 or 2 oxygen atoms.
- 25 20. The process of claim 19, wherein said at least one amine further comprises at least one amine selected from the group consisting of: c) a saturated or unsaturated aliphatic amine comprising 6 – 22 carbon atoms, d) an alicyclic

amine comprising up to 6 carbon atoms, e) an aryl-substituted alkylamine and f) a polyoxyalkylene amine $\text{NH}_2\text{-R}^1\text{-}[\text{O-R}^2\text{-}]_x\text{-O-R}^3$, wherein R^1 is a divalent alkylene radical comprising 2 – 3 carbon atoms, R^2 is a divalent alkylene radical comprising 2 – 4 carbon atoms and R^3 is an alkyl radical comprising 1 – 4 carbon atoms.

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21. The process for producing a dispersing agent according to claim 19 or 20, wherein R^2 and R^3 comprise 1 – 6 carbon atoms.