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IRON AND ZINC BASED PHARMACEUTICAL FORMULATION FOR IRON
DEFICIENCY TREATMENT

5 Field of Invention

The present invention relates to a pharmaceutical formulation, particularly providing high iron absorption, for use in preventing and treating iron deficiency and iron deficiency anemia of various origins.

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Background of Invention

Iron, which is incorporated into the basic structure of hemoglobin, a protein providing the transport of oxygen in blood, bears great importance in binding the oxygen taken in the lungs to hemoglobin and transporting it to tissues, and therefore is an essential element with respect to proper tissue development in humans. Iron is not present in foodstuffs in adequate amounts, and if it is taken exogenously, it may cause various disorders, since it is absorbed difficultly. Anemia associated with iron deficiency, for instance, is one of the symptoms frequently encountered in blood diseases, and occurs when the blood hemoglobin level drops below a lower limit.

Iron deficiency is still the most-frequently occurring nutritional problem worldwide and iron deficiency anemia is present in around 30% of the world population. Iron deficiency is still a significant public health problem not only in underdeveloped countries, but in the developed countries' disadvantageous groups (infants, adolescents, pregnant women, and ethnical groups in inadequate socioeconomic conditions) as well. According to the report published by the World Health Organization (WHO), the occurrence rate of iron deficiency anemia in developing and developed countries is 36% and 8%, respectively.

Prevalence of iron deficiency anemia: It is the most common cause of anemia worldwide. Anemia is present in 30% of the world population and at least half of this rate is due to iron deficiency.

Infants and pregnant women compose the groups which are most susceptible to anemia. Anemia prevalence is determined to be as 43% in infants and 51% in pregnant women. Anemia prevalence is 37% in school-age children, 35% in nonpregnant women, and 18% in adult males. No certain prevalence figures are available for adolescents and elder people. The prevalence of adolescents is generally considered to be close to that of adult women.

Although anemia associated with iron deficiency has many causes, it is more frequently seen in societies with lower socioeconomic levels and with bad nutritional habits. The causes of anemia may comprise, *inter alia*, delayed commencement of dietary supplements, excessive caw milk consumption by babies, vegetarian nutrition, misapplied weight-loss diets, and eating disorders. In addition, acute or chronic blood loss, e.g. ulcerative hemorrhage or menstrual bleeding and parasitary infections increase the iron need, particularly during the first year of life and the adolescent period in which rapid growth occurs, resulting in possible anemia cases.

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The presence of high amounts of iron in the intestine during oral treatment triggers oxidative damage. Therefore, losses are encountered in the formation, destruction, and functions of intestinal mucosal cells. Zinc ensures the stability of intestinal cells and cell walls, thereby providing protection against the peroxidative damage of iron. The studies conducted reports that the presence of zinc in the medium reduces iron-dependent hydroxyl radical formation at intestinal mucosa and protects the cells against iron-dependent lipid peroxidative damage.

In zinc deficiency, however, atrophy, flattening, and blunting occur at intestinal villuses and in result, the absorption functions of intestines start deteriorating. No satisfactory outcomes can be achieved in iron deficiency treatments, which are conducted when zinc deficiency is prominent. As a matter of fact, since iron can

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not be absorbed adequately and since iron storages can not be filled entirely. frequent recurrences occur in anemia therapies.

In cases when iron is used together with zinc, it is known to maintain the iron:zinc rate between 1:1 and 2.5:1, in order to avoid the absorption of both elements from interfering each other negatively.

Folic acid cannot be produced in the human body, since PABA cannot be synthesized and the first glutamate cannot be added to the molecule. Folic acid plays role in the cellular DNA synthesis and is absolutely required during the formation stage of new cells. It is also required for erythrocytes to be regenerated during anemia. Deterioration occurs in the metabolism of cells which are rapidly destructed and regenerated in folic acid deficiency. The most-susceptible cells are normoblasts, leukocyte cells, platelet cells, intestinal epithelial cells with the fastest division and regeneration rates. The susceptibility of erythrocytes gives rise to megaloblastic anemia. In folic acid deficiency there may occur megaloblastic anemia, neural tube and other congenital birth defects, hyperhomosysteinemia.

Vitamin C, a water soluble and potent antioxidant that cannot be synthesized by humans, is assistive in iron absorption. It plays roles in carnitine synthesis, the conversion of tryptophan to serotonin and thyroxin, the conversion of corticosteroids and aldosteron cholesterol to bile acids, immune system functions, preventing cancer, and in collagen synthesis.

An effort towards preventing and treating disorders associated with iron deficiency 25 is known from the application WO 2006024241. WO 2006024241 discloses a preparation in the form of tablet or aqueous suspension for preventing or treating iron deficiency, this preparation containing between 6 and 26% ferrous fumarate and between 74 and 94% haemoderivative in powder form, with an ionic iron/haem iron ratio of between 30 and 70% and an amino acid concentration of 30 between 20 and 30 µmol/100µmol of proteins.

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The patent JP 2006193428 discloses a prophylactic or ameliorative agent for iron deficiency anemia, comprising an iron compound containing various iron salts, e.g. ferrous fumarate, extracts of *Eleutherococcus Senticosus*, and rutin and/or a rutin derivative.

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Description of Invention

The object of the present invention is to provide a pharmaceutical formulation ensuring high iron absorption for efficiently preventing and treating iron deficiency and anemia associated with iron deficiency.

Another object of the present invention is to provide efficient iron, zinc, and folic acid, as essential agents, in proportions which will not interfere the absorption of each other, against malnutrition-associated macrocytes anemias, anemias due to hemorrhage, zinc deficiency co-occurring with iron deficiency, and zinc deficiency co-occurring with megaloblastic anemia.

A further object of the present invention is to provide efficient iron absorption against iron deficiencies during pregnancy, in infants and children, and against latent iron deficiency.

The pharmaceutical formulation according to the present invention preferably comprises 3 to 20 mg/ml iron, 1 to 20 mg/ml zinc, 1.5 to 60 mg/ml vitamin C, and 0.015 to 0.50 mg/ml folic acid. A more preferred pharmaceutical formulation according to the present invention comprises 5 to 12 mg/ml iron, 4 to 10 mg/ml zinc, 5 to 40 mg/ml vitamin C, and 0.08 to 0.40 mg/ml folic acid. The most preferred pharmaceutical formulation according to the present invention comprises 7 to 9 mg/ml iron, 5 to 7 mg/ml zinc, 10 to 20 mg/ml vitamin C, and 0.1 to 0.2 mg/ml folic acid. According to a preferred formulation of the present invention, the fructose amount may be kept between 30 mg/ml and 250 mg/ml, but this amount may be kept in more preferred formulations between 100 mg/ml and 200 mg/ml and in most preferred formulations between 120 mg/ml and 180 mg/ml.

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The formulation according to the present invention further comprises excipients, or auxiliary agents, such as sorbitol, sodium saccharin, fructose, sodium hydroxide, nipagin, sodium saccharin, neohesperidin dihidrochalcone, orange, lemon and mandarin essences citric acid, as well as deionized water.

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As an alternative to fructose, the formulation according to the present invention may contain mixed saccharide materials, such as sucrose, glucose, mannose, galactose, lactose, or a mixture thereof.

The proportions of fructose given above builds iron-fructose complexes with iron, thereby enhancing the dissolution rate and the absorption of iron.

The excipient citric acid is similarly supportive in iron absorption, but is also pH lowering, as well as maintaining the stability of vitamin C. According to a preferred formulation of the present invention, the citric acid amount may be kept between 1.5 mg/ml and 600 mg/ml, but this amount may be kept in more preferred formulations between 50 mg/ml and 300 mg/ml and in most preferred formulations between 90 mg/ml and 110 mg/ml.

As an alternative to citric acid for use in the formulation according to the present invention, other organic acids, carrying one or more carboxyl groups and organic side chains, e.g. tartaric acid, malic acid, may be included into the formulation.

The iron employed in the pharmaceutical formulation according to the present invention is divalent iron (Fe (II)) based on its high absorption rate, but this fact does not exclude the use of trivalent iron (Fe (II)) from the scope of the present invention. Iron and zinc are included in a formulation according to the present invention preferably in the form of ferrous fumarate (C₄H₂FeO₄) iron salt and zinc sulfate (ZnSO₄) zinc salt, respectively.

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The iron salt may also be in the form of ferrous gluconate, ferrous succinate, ferrous glutamate, ferrous lactate, ferrous citrate, ferrous tartrate, ferrous pyrophosphate. According to a preferred embodiment according to the present

invention, the source of iron may be in the form of iron-EDTA, iron ammonium ortophosphate, iron II ammoniumsulfate iron complexes.

In another formulation according present invention, the source of iron may be in the form of iron III protein succinate, iron III polymaltose, iron III Sodium-EDTA, carbonyl iron, iron chloride.

As for the zinc salt, it may be in the form of zinc sulfate mono or heptahydrate, zinc acetate, zinc carbonate, zinc chloride, zinc gluconate, zinc picolinate.

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The formulation according to the present invention is preferably in the form of a syrup, but it can also be prepared in any oral liquid forms, including the aqueous suspension form. Thus, the oral administration thereof may be facilitated as compared to the tablet form, especially for children. Said "liquid form" further comprises effervescent tablet compositions, which are convenient for oral administration once they are dissolved in water.

The constituents making up an exemplary composition according to the present invention are listed as follows:

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| Constituent | Mass per 5 ml syrup (mg) |
|------------------|--------------------------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |
| Vitamin C | 50 |

Example 2:

| Constituent | Mass per 5 ml syrup (mg) |
|------------------|--------------------------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |

| Vitamin C | 50 |
|-----------|-----|
| Fructose | 750 |

Example 3:

| inc sulfate 66 | |
|------------------|--------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |
| Vitamin C | 50 |
| Fructose | 750 |
| Citric acid | 500 mg |

5 Example 4:

| Constituent | Mass per 5 ml syrup (mg) |
|------------------|--------------------------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |
| Vitamin C | 50 |
| Sorbitol | 1500 |
| Propylene glycol | 250 |
| Ethyl alcohol | 250 |
| Nipagin M sodium | 5 |

Example 5:

| Constituent | Mass per 5 ml syrup (mg) |
|------------------|--------------------------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |
| Vitamin C | 50 |

| Sorbitol | 1500 |
|-------------------------------|--------------------------------------|
| Neohesperidin dihydrochalcone | 5 |
| Citric acid | 500 |
| Propylene glycol | 250 |
| Ethyl alcohol | 250 |
| Nipagin M sodium | 5 |
| Sodium saccharine | 5 |
| Sunset yellow | 0.03 |
| Orange essence | 2 |
| Lemon essence | 2 |
| Mandarin essence | 2 |
| Grapefruit essence | 2 |
| Deionized water | quantum sufficit to complete to 5 ml |

Example 6:

| Constituent | Mass per 5 ml syrup (mg) |
|--------------------|--------------------------------------|
| Ferrous fumarate | 121 |
| Zinc sulfate | 66 |
| Folic acid | 0.2 |
| Vitamin C | 50 |
| Sorbitol | 1500 |
| Propylene glycol | 250 |
| Ethyl alcohol | 250 |
| Nipagin M sodium | 5 |
| Sodium saccharine | 5 . |
| Sunset yellow | 0.03 |
| Orange essence | 2 |
| Lemon essence | 2 |
| Mandarin essence | 2 |
| Grapefruit essence | 2 |
| Deionized water | quantum sufficit to complete to 5 ml |

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The proportion between iron and zinc in a formulation according to the present invention is preferably 1 to 2.5; the proportion between vitamin C and iron is preferably 0.5 to 3; and the proportion between folic acid and iron is preferably between 0.005 to 0.025. In another preferred formulation according to the present invention, the proportion between iron and fructose is preferably 0.04 to 0.1 and the proportion between vitamin C and citric acid is 1 to 10.

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CLAIMS

- 1. A pharmaceutical formulation for use in preventing and treating iron deficiency and iron deficiency anemia, characterized by comprising 3 to 12 mg/ml iron, 1 to 12 mg/ml zinc, and 1.5 to 36 mg/ml vitamin C, as active agent.
- 2. A formulation according to Claim 1, characterized by further comprising 0.015 to 0.3 mg/ml folic acid.
- 3. A formulation according to claims 1 and 2, wherein the amount of iron is between 5 to 10 mg/ml, of zinc between 4 to 9 mg/ml, of vitamin C between 5 to 30 mg/ml, and of folic acid between 0.08 to 0.25 mg/ml.
- 4. A formulation according to claims 1 and 2, wherein the amount of iron is between 7 to 9 mg/ml, of zinc between 5 to 7 mg/ml, of vitamin C between 10 to 20 mg/ml, and of folic acid between 0.1 to 0.2 mg/ml.
- 5. A formulation according to any of the preceding claims, further comprising fructose, preferably in an amount between 30 to 300 mg/ml, more preferably in an amount between 100 to 200 mg/ml, and most preferably in an amount between 120 to 180 mg/ml.
 - 6. A formulation according to any of the preceding claims, further comprising citric acid, preferably in an amount between 1.5 to 600 mg/ml, more preferably in an amount between 50 to 300 mg/ml, and most preferably in an amount between 90 to 110 mg/ml.
- 7. A formulation according to any of the preceding claims, characterized in that iron is included into the formulation in the form of an iron salt, selected from the group consisting of ferrous fumarate, ferrous gluconate, ferrous succinate, ferrous glutamate, ferrous lactate, ferrous citrate, ferrous tartrate, ferrous pyrophosphate.

8. A formulation according to any of the preceding claims, characterized in that iron is included into the formulation in the form of an iron complex, selected from the group consisting of iron-EDTA, iron ammonium ortophosphate, iron Il amonniumsulfate.

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- 9. A formulation according to any of the preceding claims, characterized in that iron is selected from the group consisting of iron III protein succinate, iron III polymaltose, iron III Sodium-EDTA, carbonyl iron, iron chloride.
- 10. A formulation according to any of the preceding claims, characterized in that 10 zinc is included into the formulation in the form of a zinc salt, selected from the group consisting of zinc sulfate mono or heptahydrate, zinc acetate, zinc carbonate, zinc chloride, zinc gluconate, zinc picolinate.
- 11. A formulation according to any of the preceding claims, further comprising 15 excipient or auxiliary agent selected from the group consisting of sorbitol, propylene glycol, ethyl alcohol, sodium saccharin, fructose, sodium hydroxide, nipagin, sodium saccharin, neohesperidin dihidrochalcone, orange, lemon and mandarin essences, subset yellow, as well as deionized water. 20

12. A formulation according to any of the preceding claims, characterized in that said formulation is in oral liquid form, including syrup and aqueous suspension forms.

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- 13. A formulation according to Claim 12, characterized in that said oral liquid form also comprises liquid compositions, which result when effervescent tablets are dissolved in water for oral administration.
- 14. A formulation according to any of the preceding claims, characterized in that 30 the proportion of iron: zinc in said formulation is preferably 1 to 2.5; the proportion of vitamin C: iron is preferably 0.5 to 3; and the proportion of folic acid: iron is preferably between 0.005 to 0.025.

15. A formulation according to any of the preceding claims, characterized in that the proportion of iron : fructose in said formulation is between 0.04 and 0.1.

INTERNATIONAL SEARCH REPORT

International application No PCT/TR2009/000047

A. CLASSIFICATION OF SUBJECT MATTER INV. A61K31/375 A61K31/505

A61K33/26

A61K33/30

A61P7/06

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

A61K A61P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

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| X Further documents are listed in the continuation of Box C. | X See patent family annex. |
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Information on patent family members

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