Title: STRIPED LIQUID PERSONAL CLEANSING COMPOSITIONS CONTAINING A CLEANSING PHASE AND A SEPARATE BENEFIT PHASE COMPRISING A WATER IN OIL EMULSION

Abstract: Personal cleansing compositions, comprise (A) a cleansing phase containing a surfactant and water; and (B) a separate benefit phase comprising at least one water in oil emulsion; wherein the cleansing and benefit phases are packaged together and are in physical contact. The two phases are packaged in physical contact and remain separate and stable at ambient conditions for at least 180 days. These compositions and corresponding methods provide improved cosmetics, skin feel, and/or skin benefit efficacy.
STRIPED LIQUID PERSONAL CLEANSING COMPOSITIONS CONTAINING A
CLEANSING PHASE AND A SEPARATE BENEFIT PHASE COMPRISING A WATER
IN OIL EMULSION

FIELD OF THE INVENTION

The present invention relates to striped liquid personal cleansing compositions
comprising a cleansing phase and a separate benefit phase comprising a water in oil emulsion
wherein the two phases are packaged in physical contact while remaining stable for long periods
of time.

BACKGROUND OF THE INVENTION

Personal cleansing compositions that purport to provide skin-conditioning benefits are
known. Many of these compositions are aqueous systems comprising an emulsified conditioning
oil or other similar materials in combination with a lathering surfactant. Although many of these
products can provide both conditioning and cleansing benefits, there are often trade-offs
associated with their use. For instance, it is often difficult to formulate a product that deposits a
sufficient amount of skin conditioning agents on skin during use. In order to combat
emulsification of the skin conditioning agents by the cleansing surfactant, large amounts of the
skin conditioning agent are often added to the compositions. Unfortunately, raising the level of
skin conditioning agent in order to achieve increased deposition can negatively affect product
lather performance and stability.

One attempt at providing conditioning and cleansing benefits from a personal cleansing
product while maintaining stability has been the use of dual-chamber packaging. These packages
comprise separate cleansing compositions and conditioning compositions, and allow for the co-
dispensing of the two in a single or dual stream. The separate conditioning and cleansing
compositions thus remain physically separate and stable during prolonged storage and just prior to
application, but then mix during or after dispensing to provide conditioning and cleansing benefits
from a physically stable system. Although such dual-chamber delivery systems provide improved
conditioning benefits versus of conventional systems, it is often difficult to achieve consistent and
uniform performance because of the uneven dispensing ratio between the cleansing phase and the
conditioning phase from these dual-chamber packages. Additionally, these packaging systems
add considerable cost to the finished product.

Striped personal cleansing compositions are also known in the art. However, these
compositions do not contain a cleansing phase and a benefit phase and thus stability has not been
an issue for these products.
Accordingly, the need still remains for stable personal cleansing compositions that provide both cleansing and improved skin conditioning benefits. It has now been found that striped personal cleansing compositions comprising two phases in physical contact that remain stable for long periods of time can be formulated.

These striped personal cleansing compositions of the present invention comprise cleansing and benefit phases that are packaged in physical contact yet remain stable. These compositions provide improved deposition of conditioning agents on skin.

The compositions of the present invention further provide superior cosmetics via the striped appearance and improved skin feel during and after application. It has been found that such compositions can be formulated with sufficiently high levels of benefit agents without compromising product lather performance and stability. The superior lather performance of these compositions can be demonstrated via the lather volume method described herein.

It has also been found that the striped personal cleansing compositions herein can be formulated with selected skin active agents that provide improved chronic skin benefits to the skin. These compositions comprise a cleansing phase containing a cleansing surfactant and at least one additional benefit phase containing a skin active agent, wherein the cleansing and active phases are packaged in physical contact while remaining stable for long periods of time.

**SUMMARY OF THE INVENTION**

The present invention is directed to a striped personal cleansing composition comprising first stripe comprising a cleansing phase comprising a surfactant and water, and at least one additional stripe comprising a separate benefit phase comprising a water in oil emulsion.

**DETAILED DESCRIPTION**

The striped personal cleansing compositions of the present invention comprise a first stripe comprising a cleansing phase, and at least one separate additional stripe comprising a benefit phase. The benefit phase comprises a water in oil emulsion. These and other essential limitations of the compositions and methods of the present invention, as well as many of the optional ingredients suitable for use herein, are described in detail hereinafter.

By the term “stripe” as used herein, is meant that the cleansing phase and the benefit phase herein occupy separate but distinct physical spaces inside the package in which they are stored, but are in direct contact with one another (i.e., they are not separated by a barrier and they are not emulsified or mixed to any significant degree). In one preferred embodiment of the present invention, the cleansing phase and the benefit phase are present within the container as distinct layers or “stripes”. The stripes may be relatively uniform and even across the dimension of the package. Alternatively, the layers may be uneven, i.e. wavy, or may be nonuniform in dimension. The stripes do not need to necessarily extend across the entire dimension of the package. The
"stripe" can be various geometric shapes, various different colors or include glitter or pearlescence.

The term "ambient conditions" as used herein, refers to surrounding conditions at one (1) atmosphere of pressure, 50% relative humidity, and 25°C.

The term “stable” as used herein, unless otherwise specified, refers to compositions that maintain at least two “separate” phases when sitting in physical contact at ambient conditions for a period of at least about 180 days. By “separate” is meant that there is substantially no mixing of the phases, observable to the naked eye, prior to dispensing of the composition.

The term “personal cleansing composition” as used herein, refers to compositions intended for topical application to the skin or hair.

The phrase “substantially free of” as used herein, means that the composition comprises less than about 3%, preferably less than about 1%, more preferably less than about 0.5%, even more preferably less than about 0.25%, and most preferably less than about 0.1% of the stated ingredient.

All percentages, parts and ratios as used herein are by weight of the total composition, unless otherwise specified. All such weights as they pertain to listed ingredients are based on the active level and, therefore, do not include solvents or by-products that may be included in commercially available materials, unless otherwise specified.

All cited references are incorporated herein by reference in their entireties. Citation of any reference is not an admission regarding any determination as to its availability as prior art to the claimed invention.

The personal cleansing compositions and methods of the present invention can comprise, consist of, or consist essentially of, the essential elements and limitations of the invention described herein, as well as any additional or optional ingredients, components, or limitations described herein or otherwise useful in personal cleansing compositions intended for topical application to the hair or skin.

**Product Form**

The personal cleansing compositions of the present invention are typically in the form of a liquid. The term “liquid” as used herein means that the composition is generally flowable to some degree. “Liquids”, therefore, can include liquid, semi-liquid, cream, lotion or gel compositions intended for topical application to skin. These compositions typically exhibit a viscosity of equal to or greater than about 3,000 cps, but less than 1,000,000 cps. These compositions contain a cleansing phase and a benefit phase, both of which are described in greater detail hereinafter.
All of the product forms contemplated for purposes of defining the compositions and 
methods of the present invention are rinse-off formulations, by which is meant the product is 
applied topically to the skin or hair and then subsequently (i.e., within minutes) rinsed away with 
water, or otherwise wiped off using a substrate or other suitable removal means.

**Cleansing Phase**

The personal cleansing compositions of the present invention comprise an aqueous 
cleansing phase that contains a surfactant suitable for application to the skin or hair. Suitable 
surfactants for use herein include any known or otherwise effective cleansing surfactant which are 
suitable for application to the skin, and which are otherwise compatible with the other essential 
ingredients in the aqueous cleansing phase of the compositions. These cleansing surfactants 
include anionic, nonionic, cationic, zwitterionic or amphoteric surfactants, or combinations 
thereof. Other suitable surfactants are described in McCutcheon's, Emulsifiers and Detergents, 

The aqueous cleansing phase of the personal care compositions preferably comprises a 
cleansing surfactant at concentrations ranging from about 3% to about 60%, more preferably from 
about 4% to about 30%, even more preferably from about 5% to about 25%, by weight of the 
aqueous cleansing phase. The preferred pH range of the cleansing phase is from about 5 to about 
8.

The aqueous cleansing phase of the personal care compositions preferably produces a 
Total Lather Volume of at least 350 ml, preferably greater than 400ml, even more preferably 
greater than 600ml, as described in the Lathering Volume Test. The aqueous cleansing phase of 
the personal care compositions preferably produces a Flash Lather Volume of at least 150 ml, 
preferably greater than 200ml, most preferably greater than 300ml as described in the Lathering 
Volume Test.

Anionic surfactants suitable for use in the cleansing phase include alkyl and alkyl ether 
sulfates. These materials have the respective formula ROSO$_3$M and RO(C$_2$H$_4$O)$_x$SO$_3$M, 
wherein R is alkyl or alkenyl of from about 8 to about 24 carbon atoms, x is 1 to 10, and M is a 
water-soluble cation such as ammonium, sodium, potassium and triethanolamine. The alkyl ether 
sulfates are typically made as condensation products of ethylene oxide and monohydric alcohols 
having from about 8 to about 24 carbon atoms. Preferably, R has from about 10 to about 18 
carbon atoms in both the alkyl and alkyl ether sulfates. The alcohols can be derived from fats, 
e.g., coconut oil or tallow, or can be synthetic. Lauryl alcohol and straight chain alcohols derived 
from coconut oil are preferred herein. Such alcohols are reacted with about 1 to about 10, 
preferably from about 3 to about 5, and more preferably with about 3, molar proportions of
ethylene oxide and the resulting mixture of molecular species having, for example, an average of 3 moles of ethylene oxide per mole of alcohol, is sulfated and neutralized.

Other suitable anionic surfactants include water-soluble salts of the organic, sulfuric acid reaction products of the general formula \([R^1\text{-}SO_3\text{-}M]\), wherein \(R^1\) is chosen from the group consisting of a straight or branched chain, saturated aliphatic hydrocarbon radical having from about 8 to about 24, preferably about 10 to about 18, carbon atoms; and \(M\) is a cation. Suitable examples are the salts of an organic sulfuric acid reaction product of a hydrocarbon of the methane series, including iso-, neo-, ineso-, and \(n\)-paraffins, having about 8 to about 24 carbon atoms, preferably about 10 to about 18 carbon atoms and a sulfonating agent, e.g., \(\text{SO}_3\), \(\text{H}_2\text{SO}_4\), oleum, obtained according to known sulfonation methods, including bleaching and hydrolysis. Preferred are alkali metal and ammonium sulfonated \(\text{C}_{10-18} n\)-paraffins.

Preferred anionic surfactants for use in the cleansing phase include ammonium lauryl sulfate, ammonium laureth sulfate, triethylamine lauryl sulfate, triethylamine laureth sulfate, triethanolamine lauryl sulfate, triethanolamine laureth sulfate, monoethanolamine lauryl sulfate, monoethanolamine laureth sulfate, diethanolamine lauryl sulfate, diethanolamine laureth sulfate, lauric monoglyceride sodium sulfate, sodium lauryl sulfate, sodium laureth sulfate, potassium laureth sulfate, sodium lauryl sarcosinate, sodium lauroyl sarcosinate, lauryl sarcosine, cocoyl sarcosine, ammonium cocoyl sulfate, ammonium lauroyl sulfate, sodium cocoyl sulfate, potassium cocoyl sulfate, potassium lauryl sulfate, sodium lauroyl sulfonate, sodium dodecyl benzene sulfonate, and combinations thereof.

Amphoteric surfactants suitable for use in the cleansing phase include those that are broadly described as derivatives of aliphatic secondary and tertiary amines in which the aliphatic radical can be straight or branched chain and wherein one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic water solubilizing group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate. Examples of compounds falling within this definition are sodium 3-dodecylaminopropionate, sodium 3-dodecylaminopropanoate sulfonate, sodium lauryl sarcosinate, \(N\)-alkyltaurines such as the one prepared by reacting dodecylamine with sodium isethionate according to the teaching of U.S. Patent 2,658,072, \(N\)-higher alkyl aspartic acids such as those produced according to the teaching of U.S. Patent 2,438,091, and the products described in U.S. Patent 2,528,378.

Zwitterionic surfactants suitable for use in the cleansing phase include those that are broadly described as derivatives of aliphatic quaternary ammonium, phosphonium, and sulfonium compounds, in which the aliphatic radicals can be straight or branched chain, and wherein one of
the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate. Such suitable zwitterionic surfactants can be represented by the formula:

\[ \text{R}^2 - Y^+ - \text{CH}_2 - \text{R}^4 - Z \]

wherein \( \text{R}^2 \) contains an alky1, alkenyl, or hydroxy alkyl radical of from about 8 to about 18 carbon atoms, from 0 to about 10 ethylene oxide moieties and from 0 to about 1 glyceryl moiety;

\( Y \) is selected from the group consisting of nitrogen, phosphorus, and sulfur atoms; \( \text{R}^3 \) is an alkyl or monohydroxyalkyl group containing about 1 to about 3 carbon atoms; \( X \) is 1 when \( Y \) is a sulfur atom, and 2 when \( Y \) is a nitrogen or phosphorus atom; \( \text{R}^4 \) is an alkylene or hydroxyalkylene of from about 1 to about 4 carbon atoms and \( Z \) is a radical selected from the group consisting of carboxylate, sulfonate, sulfate, phosphonate, and phosphate groups.

Other zwitterionic surfactants suitable for use in the cleansing phase include betaines, including high alkyl betaines such as coco dimethyl carboxymethyl betaine, cocoamidopropyl betaine, cocobetaine, lauryl amidopropyl betaine, oleyl betaine, lauryl dimethyl carboxymethyl betaine, lauryl dimethyl alphacarboxyethyl betaine, cetyl dimethyl carboxymethyl betaine, lauryl bis-(2-hydroxyethyl) carboxymethyl betaine, stearyl bis-(2-hydroxypropyl) carboxymethyl betaine, oleyl dimethyl gamma-carboxypropyl betaine, and lauryl bis-(2-hydroxypropyl)alpha-carboxyethyl betaine. The sulfobetaines may be represented by coco dimethyl sulfopropyl betaine, stearyl dimethyl sulfopropyl betaine, lauryl dimethyl sulfoethyl betaine, lauryl bis-(2-hydroxyethyl) sulfopropyl betaine and the like; amidobetaines and amidosulfobetaines, wherein the RCONH(CH$_2$)$_3$ radical is attached to the nitrogen atom of the betaine are also useful in this invention.

**Characteristics of Cleansing Phase Preferred For Stability**

**Lamellar Structurant**

The compositions of the present invention preferably comprise about 0.1% to 10% by wt. of a structurant agent in the cleansing phase which functions in the compositions to form a lamellar phase. It is believed the lamellar phase enhances the interfacial stability between the cleansing phase and the benefit phase.

Suitable structurant include fatty acids or ester derivatives thereof, a fatty alcohol, or trihydroxystearin, polycare 133. More preferably the structurant is selected from lauric acid or trihydroxystearin.
In an additional embodiment of the present invention the surfactant compositions for use in the cleansing phase exhibit Non-Newtonian shear thinning behavior (herein referred to as free flowing compositions). These cleansing compositions comprising water, at least one anionic surfactant, an electrolyte and at least one alkanolamide. It has been found that by employing a cleansing phase exhibiting Non-Newtonian shear thinning behavior, the stability of the resulting personal cleansing composition may be increased.

The alkanolamide if present has the general structure of:

\[
\begin{align*}
O & (R_1-O)_xH \\
\| & / \\
R-C-N & \backslash \\
& (R_2-O)_yH
\end{align*}
\]

wherein R is \(C_8\) to \(C_{24}\) or preferably in some embodiments \(C_8\) to \(C_{22}\) or in other embodiments \(C_8\) to \(C_{18}\) saturated or unsaturated straight chain or branched aliphatic group, \(R_1\) and \(R_2\) are the same or different \(C_{2-4}\) straight chain or branched aliphatic group, \(x = 0\) to \(10\); \(y = 1 - 10\) and wherein the sum of \(x\) and \(y\) is less than or equal to \(10\).

The amount of alkanolamide when present in the composition is about 0.1% to about 10% by weight, and in some embodiments is preferably about 2% to about 5% by weight. Some preferred alkanolamides include Cocamide MEA (Coco monethanolamide) and Cocamide MIPA (Coco monoisopropanolamide). A co-surfactant from the classes of nonionic surfactant, amphoteric and/or zwitterionic surfactant or cationic surfactant may be optionally incorporated.

The electrolyte, if used, can be added per se to the composition or it can be formed in situ the counter-ions included in one of the raw materials. The electrolyte preferably includes an anion comprising phosphate, chloride, sulfate or citrate and a cation comprising sodium, ammonium, potassium, magnesium or mixtures thereof. Some preferred electrolytes are sodium or ammonium chloride or sodium or ammonium sulfate.

The electrolyte should be present in an amount, which facilitates formation of the free flowing composition. Generally, this amount is from about 0.1% by weight to about 15% by weight, preferably from about 1% to about 6% by weight of the cleansing phase, but may be varied if required.

**Optional Ingredients for use in the Cleansing Phase**

Other suitable optional ingredients which may be employed in the cleansing phase include humectants and solutes. A variety of humectants and solutes can be employed and can be present at a level of from about 0.1% to about 50%, preferably from about 0.5% to about 35%, and more preferably from about 2% to about 20% of the personal care composition.
Nonionic polyethylene/polypropylene glycol polymers are preferably used as skin conditioning agents. Polymers useful herein that are especially preferred are PEG-2M wherein x equals 2 and n has an average value of about 2,000 (PEG 2-M is also known as Polyox WSR® N-10 from Union Carbide and as PEG-2,000); PEG-5M wherein x equals 2 and n has an average value of about 5,000 (PEG 5-M is also known as Polyox WSR® 35 and Polyox WSR® N-80, both from Union Carbide and as PEG-5,000 and Polyethylene Glycol 200,000); PEG-7M wherein x equals 2 and n has an average value of about 7,000 (PEG 7-M is also known as Polyox WSR® (N-750 from Union Carbide); PEG-9M wherein x equals 2 and n has an average value of about 9,000 (PEG 9-M is also known as Polyox WSR® N-3333 from Union Carbide); PEG-14 M wherein x equals 2 and n has an average value of about 14,000 (PEG 14-M is also known as Polyox WSR-205 and Polyox WSR® N-3000 both from Union Carbide); and PEG-90M wherein x equals 2 and n has an average value of about 90,000 (PEG-90M is also known as Polyox WSR®-301 from Union Carbide.)

The striped personal cleansing compositions of the present invention may additionally comprise an organic cationic deposition polymer in the cleansing phase as a deposition aid for the benefit agents described hereinafter. Concentrations of the cationic deposition polymer preferably range from about 0.025% to about 3%, more preferably from about 0.05% to about 2%, even more preferably from about 0.1% to about 1%, by weight of the cleansing phase composition.

Suitable cationic deposition polymers for use in the striped personal cleansing composition of the present invention contain cationic nitrogen-containing moieties such as quaternary ammonium or cationic protonated amino moieties. The cationic protonated amines can be primary, secondary, or tertiary amines (preferably secondary or tertiary), depending upon the particular species and the selected pH of the personal cleansing composition. The average molecular weight of the cationic deposition polymer is between about 5,000 to about 10 million, preferably at least about 100,000, more preferably at least about 200,000, but preferably not more than about 2 million, more preferably not more than about 1.5 million. The polymers also have a cationic charge density ranging from about 0.2 meq/gm to about 5 meq/gm, preferably at least about 0.4 meq/gm, more preferably at least about 0.6 meq/gm, at the pH of intended use of the personal cleansing composition, which pH will generally range from about pH 4 to about pH 9, preferably between about pH 5 and about pH 8.

The charge density can be controlled and adjusted in accordance with techniques well known in the art. As used herein the "charge density" of the cationic polymers is defined as the number of cationic sites per polymer gram atomic weight (molecular weight), and can be expressed in terms of meq/gram of cationic charge. In general, adjustment of the proportions of
amine or quaternary ammonium moieties in the polymer, as well as pH of the personal cleansing composition in the case of the amines, will affect the charge density.

Any anionic counterions can be used in association with the cationic deposition polymers so long as the polymers remain soluble in water, in the personal cleansing composition, or in a coacervate phase of the personal cleansing composition, and so long as the counterions are physically and chemically compatible with the essential components of the personal cleansing composition or do not otherwise unduly impair product performance, stability or aesthetics. Nonlimiting examples of such counterions include halides (e.g., chlorine, fluorine, bromine, iodine), sulfate and methylsulfate.

Nonlimiting examples of cationic deposition polymers for use in the personal cleansing composition include polysaccharide polymers, such as cationic cellulose derivatives. Preferred cationic cellulose polymers are the salts of hydroxyethyl cellulose reacted with trimethyl ammonium substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 10 which are available from Amerchol Corp. (Edison, N.J., USA) in their Polymer KG, JR and LR series of polymers with the most preferred being KG-30M.

Other suitable cationic deposition polymers include cationic guar gum derivatives, such as guar hydroxypropyltrimonium chloride, specific examples of which include the Jaguar series (preferably Jaguar C-17) commercially available from Rhodia Inc., and N-Hance polymer series commercially available from Aqualon.

Other suitable cationic deposition polymers include synthetic cationic polymers. The cationic polymers suitable for use in the cleansing composition herein is water soluble or dispersible, non crosslinked, cationic polymers having a cationic charge density of from about 4 meq/gm to about 7 meq/gm, preferably from about 4 meq/gm to about 6 meq/gm, more preferably from about 4.2 meq/gm to about 5.5 meq/gm. The select polymers also must have an average molecular weight of from about 1,000 to about 1 million, preferably from about 10,000 to about 500,000, more preferably from about 75,000 to about 250,000.

The concentration of the cationic polymer in the cleansing composition ranges about 0.025% to about 5%, preferably from about 0.1% to about 3%, more preferably from about 0.2% to about 1%, by weight of the composition.

A non-limiting example of a commercially available synthetic cationic polymer for use in the cleansing compositions is polymethacrylamidopropyl trimonium chloride, available under the trade name Polycare 133, from Rhodia, Cranberry, N.J., U.S.A.

The cationic polymers herein are either soluble in the cleansing phase, or preferably are soluble in a complex coacervate phase in the striped personal cleansing composition formed by the cationic deposition polymer and the anionic surfactant component described hereinbefore.
Complex coacervates of the cationic deposition polymer can also be formed with other charged materials in the personal cleansing composition.

Coacervate formation is dependent upon a variety of criteria such as molecular weight, component concentration, and ratio of interacting ionic components, ionic strength (including, modification of ionic strength, for example, by addition of salts), charge density of the cationic and anionic components, pH, and temperature. Coacervate systems and the effect of these parameters have been described, for example, by J. Caelles, et al., "Anionic and Cationic Compounds in Mixed Systems", Cosmetics & Toiletries, Vol. 106, April 1991, pp 49-54, C. J. van Oss, "Coacervation, Complex-Coacervation and Flocculation", J. Dispersion Science and Technology, Vol. 9 (5,6), 1988-89, pp 561-573, and D. J. Burgess, "Practical Analysis of Complex Coacervate Systems", J. of Colloid anti Interface Science, Vol. 140, No. 1, November 1990, pp 227-238, which descriptions are incorporated herein by reference.

It is believed to be particularly advantageous for the cationic deposition polymer to be present in the personal cleansing composition in a coacervate phase, or to form a coacervate phase upon application or rinsing of the cleansing composition to or from the skin. Complex coacervates are believed to more readily deposit on the skin, which results in improved deposition of the benefit materials. Thus, in general, it is preferred that the cationic deposition polymer exists in the personal cleansing composition as a coacervate phase or form a coacervate phase upon dilution. If not already a coacervate in the personal cleansing composition, the cationic deposition polymer will preferably exist in a complex coacervate form in the cleansing composition upon dilution with water.

Techniques for analysis of formation of complex coacervates are known in the art. For example, centrifugation analyses of the personal cleansing compositions, at any chosen stage of dilution, can be utilized to identify whether a coacervate phase has formed.

**Benefit Phase (Water in Oil Emulsion)**

The benefit phase of the present invention comprises a water in oil emulsion comprising an oil, an emulsifier, water and preferably a density modifier. The oil phase is the continuous phase and the water phase is the discontinuous or "internal" phase.

**Oils**

The benefit phase of the present invention typically comprises from about 10% to about 99% of oil, more preferably 20 to about 95% oil, even more preferably from 50 to about 90% oil and most preferably from 60% to about 80%.

In general the higher the level of oil employed in the water in oil emulsion the more stable the personal cleansing composition employing the water in oil emulsion will be. Oils suitable for use herein include any natural and synthetic materials with an overall solubility parameter less
than about 12.5 \((\text{cal/cm}^3)^{0.5}\), preferably less than about 11.5 \((\text{cal/cm}^3)^{0.5}\). Solubility parameters for the oils described herein are determined by methods well known in the chemical arts for establishing the relative polar character of a material. A description of solubility parameters and means for determining them are described by C. D. Vaughn, “Solubility Effects in Product, Package, Penetration and Preservation” 103 Cosmetics and Toiletries 47-69, October 1988; and C. D. Vaughn, “Using Solubility Parameters in Cosmetics Formulation”, 36 J. Soc. Cosmetic Chemists 319-333, September/October, 1988.

The benefit agent for use in the benefit phase of the composition has a Vaughan Solubility Parameter (VSP) of from about 5 to about 10, preferably from about 6 to less than 10, more preferably from about 6 to about 9. Non-limiting examples of benefit agents having VSP values ranging from about 5 to about 10 include the following:

<table>
<thead>
<tr>
<th>Vaughan Solubility Parameters*</th>
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<tr>
<td>Cyclomethicone</td>
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<td>Squalene</td>
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<tr>
<td>Petrolatum</td>
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<tr>
<td>Isopropyl Palmitate</td>
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<tr>
<td>Isopropyl Myristate</td>
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<tr>
<td>Castor Oil</td>
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<td>Cholesterol</td>
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By “overall solubility parameter” is meant that it is possible to use oils with higher solubility parameters than 12.5 \((\text{cal/cm}^3)^{0.5}\) if they are blended with other oils to reduce the overall solubility parameter of the oil mixture to less than about 12.5 \((\text{cal/cm}^3)^{0.5}\). For example, a small portion of diethylene glycol (sol par = 13.61) could be blended with lanolin oil (sol par = 7.3) and a cosolublizing agent to create a mixture that has a solubility parameter of less than 12.5 \((\text{cal/cm}^3)^{0.5}\).

Suitable for use herein oils include but are not limited, to hydrocarbon oils and waxes, silicones, fatty acid derivatives, cholesterol, cholesterol derivatives, diglycerides, triglycerides, vegetable oils, vegetable oil derivatives, acetoglyceride esters, alkyl esters, alkenyl esters, lanolin and its derivatives, wax esters, beeswax derivatives, sterols and phospholipids, and combinations thereof.

Non-limiting examples of hydrocarbon oils and waxes suitable for use herein include petrolatum, mineral oil, micro-crystalline waxes, polyalkenes, paraffins, cerasin, ozokerite,
polyethylene, perhydrosqualene, poly alpha olefins, hydrogenated polyisobutenes and combinations thereof.

Non-limiting examples of silicone oils suitable for use herein include dimethicone copolyol, dimethyldiphenylsiloxane, diethyldiphenylsiloxane, mixed C1-C30 alkyl polysiloxanes, phenyl dimethicone, dimethiconol, and combinations thereof. Preferred are non-volatile silicones selected from dimethicone, dimethiconol, mixed C1-C30 alkyl polysiloxane, and combinations thereof. Nonlimiting examples of silicone oils useful herein are described in U.S. Patent No. 5,011,681 (Ciotti et al.).

Non-limiting examples of diglycerides and triglycerides suitable for use herein include castor oil, soy bean oil, derivatized soybean oils such as maleated soy bean oil, safflower oil, cotton seed oil, corn oil, walnut oil, peanut oil, olive oil, cod liver oil, almond oil, avocado oil, palm oil and sesame oil, vegetable oils, sunflower seed oil, and vegetable oil derivatives; coconut oil and derivatized coconut oil, cottonseed oil and derivatized cottonseed oil, jojoba oil, cocoa butter, and combinations thereof. In addition any of the above oils that have been partially or fully hydrogenated are also suitable.

Non-limiting examples of acetoglyceride esters suitable for use herein include acetylated monoglycerides.

Non-limiting examples of alkyl esters suitable for use herein include isopropyl esters of fatty acids and long chain esters of long chain fatty acids, e.g. SEFA (sucrose esters of fatty acids). Lauryl pyrrolidone carboxylic acid, pentaerthritol esters, aromatic mono, di or triesters, cetyl ricinoleate, non-limiting examples of which include isopropyl palmitate, isopropyl myristate, cetyl ricinoleate and stearyl ricinoleate. Other examples are: hexyl laurate, isohexyl laurate, myristyl myristate, isohexyl palmitate, decyl oleate, isodecyl oleate, hexadecyl stearate, decyl stearate, isopropyl isostearate, diisopropyl adipate, diisohexyl adipate, dihexyldecyl adipate, diisopropyl sebacate, acyl isononanoate lauril lactate, myristyl lactate, cetyl lactate, and combinations thereof.

Non-limiting examples of alkenyl esters suitable for use herein include oleyl myristate, oleyl stearate, oleyl oleate, and combinations thereof.

Non-limiting examples of lanolin and lanolin derivatives suitable for use herein include lanolin, lanolin oil, lanolin wax, lanolin alcohols, lanolin fatty acids, isopropyl lanolate, acetylated lanolin, acetylated lanolin alcohols, lanolin alcohol linoleate, lanolin alcohol riconoleate, hydroxylated lanolin, hydrogenated lanolin and combinations thereof.

Still other suitable oils include milk triglycerides (e.g., hydroxylated milk glyceride) and polyol fatty acid polyesters.
Still other suitable oils include wax esters, non-limiting examples of which include beeswax and beeswax derivatives, spermaceti, myristyl myristate, stearyl stearate, and combinations thereof. Also useful are vegetable waxes such as carnauba and candelilla waxes; sterols such as cholesterol, cholesterol fatty acid esters; and phospholipids such as lecithin and derivatives, sphingo lipids, ceramides, glycosphingo lipids, and combinations thereof.

**Low HLB Emulsifier**

The water-in-oil emulsion of the present invention also includes about 0.1% to about 20% of a low HLB emulsifier, more preferably from about 0.1% to about 10%, still more preferably from about 0.5% to about 9%, of one or more low HLB emulsifier. The low HLB emulsifier may function as an emulsifier.

Preferred low HLB emulsifiers are those having an HLB of from about 1 to about 10, more preferably from 1 to about 8. Suitable low HLB emulsifiers are those selected from saturated C14 to C30 fatty alcohols, saturated C16 to C30 fatty alcohols containing from about 1 to about 5 moles of ethylene oxide, saturated C16 to C30 diols, saturated C16 to C30 monoglycerol ethers, saturated C16 to C30 hydroxy fatty acids, C14 to C30 hydroxylated and nonhydroxylated saturated fatty acids, C14 to C30 saturated ethoxylated fatty acids, amines and alcohols containing from about 1 to about 5 moles of ethylene oxide diols, C14 to C30 saturated glyceryl mono esters with a monoglyceride content of at least 40%, C14 to C30 saturated polyglycerol esters having from about 1 to about 3 alkyl group and from about 2 to about 3 saturated glycerol units, C14 to C30 glyceryl mono ethers, C14 to C30 sorbitan mono/diesters, C14 to C30 saturated ethoxylated sorbitan mono/diesters with about 1 to about 5 moles of ethylene oxide, C14 to C30 saturated methyl glucoside esters, C14 to C30 saturated sucrose mono/diesters, C14 to C30 saturated ethoxylated methyl glucoside esters with about 1 to about 5 moles of ethylene oxide, C14 to C30 saturated polyglucosides having an average of between 1 to 2 glucose units and mixtures thereof, having a melting point of at least about 45°C.

The low HLB emulsifiers of the present invention are selected from stearic acid, palmitic acid, stearyl alcohol, cetyl alcohol, behenyl alcohol, stearic acid, palmitic acid, the polyethylene glycol ether of stearyl alcohol having an average of about 1 to about 5 ethylene oxide units, the polyethylene glycol ether of cetyl alcohol having an average of about 1 to about 5 ethylene oxide units, and mixtures thereof. More preferred low HLB emulsifiers of the present invention are selected from stearyl alcohol, cetyl alcohol, behenyl alcohol, the polyethylene glycol ether of stearyl alcohol having an average of about 2 ethylene oxide units (steareth-2), the polyethylene
glycol ether of cetyl alcohol having an average of about 2 ethylene oxide units, and mixtures thereof. Even more preferred low HLB emulsifiers are selected from stearic acid, palmitic acid, stearyl alcohol, cetyl alcohol, behenyl alcohol, steareth-2, and mixtures thereof.

**Density Modifiers**

To further improve stability under stress conditions such as high temperature and vibration, it is preferable to adjust the densities of the separate phases such that they are substantially equal. To achieve this, low density microspheres are added to the cleansing phase of the striped composition. The low density microspheres employed to reduce the overall density of the cleansing phase are particles having a density lower than 0.7 g/cm$^3$, preferably less than 0.2 g/cm$^3$, more preferably less than 0.1 g/cm$^3$, most preferably less than 0.05 g/cm$^3$. The low density microspheres generally have a diameter less than 200 μm, preferably less than 100 μm, most preferably less than 40 μm. Preferably, the density difference between the cleansing phase and the benefit phase is less than 0.15 g/cm$^3$, more preferably, the density difference is less than 0.10 g/cm$^3$, even more preferably, the density difference is less than 0.05 g/cm$^3$, most preferably, the density difference is less than 0.01 g/cm$^3$.

The microspheres are produced from any appropriate inorganic or organic material, compatible with a use on the skin, that is, nonirritating and nontoxic.


These microspheres may be produced from any nontoxic and non-irritant thermoplastic materials. Polymers or copolymers of acrylonitrile or of vinylidene chloride may be used, for example. It is possible to use, for example, a copolymer containing, by weight, from 0 to 60% of units derived from vinylidene chloride, from 20 to 90% of units derived from acrylonitrile and from 0 to 50% of units derived from an acrylic or styrene monomer, the sum of the percentages (by weight) being equal to 100. The acrylic monomer is, for example, a methyl or ethyl acrylate or methacrylate. The styrene monomer is, for example, alpha-methylstyrere or styrene. These microspheres can be in the dry or hydrated state.

The internal cavity of expanded hollow microspheres contains a gas, which can be a hydrocarbon such as isobutane or isopentane or alternatively air. Among hollow microspheres which can be used, special mention may be made of those marketed under the brand name EXPANCEL® (thermoplastic expandable microspheres) by the Akzo Nobel Company, especially those of DE (dry state) or WE (hydrated state) grade. Examples include: Expance® 091 DE 40 d30; Expancel® 091 DE 80 d30; Expancel® 051 DE 40 d60; Expancel® 091 WE 40 d24; Expancel® 053 DE 40 d20.
Representative microspheres derived from an inorganic material, include, for instance, "Qcel® Hollow Microspheres" and "EXTENDOSPHERESTM Ceramic Hollow Spheres", both available from the PQ Corporation. Examples are: Qcel® 300; Qcel® 6019; Qcel® 6042S.

Just as low density microspheres can be added to the cleansing phase of the present invention to improve vibrational stability, high density materials can be added to the benefit phase to increase its density having the same impact on stability.

**Aqueous Phase**

The benefit phase of the present invention typically comprises from about 1% to about 90% of an aqueous phase. The aqueous phase comprises a fluid selected from the group consisting of water, mono- and polyhydric alcohols (glycerin, propylene glycol, ethanol, isopropanol, etc.).

**Optional Ingredients**

The personal cleansing compositions of the present invention may further comprise other optional ingredients that may modify the physical, chemical, cosmetic or aesthetic characteristics of the compositions or serve as additional "active" components when deposited on the skin. The compositions may also further comprise optional inert ingredients. Many such optional ingredients are known for use in personal care compositions, and may also be used in the personal cleansing compositions herein, provided that such optional materials are compatible with the essential materials described herein, or do not otherwise unduly impair product performance.

Such optional ingredients are most typically those materials approved for use in cosmetics and that are described in reference books such as the CTFA Cosmetic Ingredient Handbook, Second Edition, The Cosmetic, Toiletries, and Fragrance Association, Inc. 1988, 1992. These optional materials can be used in any aspect of the compositions of the present invention, including either of the active or cleansing phases as described herein.

Optional ingredients for use in the cleansing phase of the compositions of the present invention can include any benefit phase material as described herein that is also compatible with the selected ingredients in the cleansing phase. Likewise, optional ingredients for use in the benefit phase of the compositions of the present invention can include any cleansing phase material described herein that is also compatible with the selected ingredients in the benefit phase.

Other optional ingredients for use in either phase of the composition, preferably the benefit phase, include silicone elastomer powders and fluids to provide any of a variety of product benefits, including improved product stability, application cosmetics, emolliency, conditioning, and so forth. The concentration of the silicone elastomers in the composition preferably ranges from about 0.1% to about 20%, more preferably from about 0.5% to about 10%, by weight of the composition. In this context, the weight percentages are based upon the weight of the silicone
elastomers material itself, excluding any silicone-containing fluid that typically accompanies such silicone elastomers materials in the formulation process. The silicone elastomers suitable for optional use herein include emulsifying and non-emulsifying silicone elastomers, non-limiting examples of which are described in U.S.S.N. 09/613,266 (assigned to The Procter & Gamble Company).

The separate benefit phase of the striped liquid personal cleansing compositions may optionally comprise the following skin benefit ingredients for enhanced delivery of these water in oil emulsion materials on skin. Non limiting examples of these optional ingredients include vitamins and derivatives thereof (e.g., ascorbic acid, vitamin E, tocopheryl acetate, and the like); sunscreens; thickening agents (e.g., polyol alkoxy ester, available as Crothix from Croda); preservatives for maintaining the anti microbial integrity of the cleansing compositions; anti-acne medicaments (resorcinol, salicylic acid, and the like); antioxidants; skin soothing and healing agents such as aloe vera extract, allantoin and the like; chelators and sequestrants; and agents suitable for aesthetic purposes such as fragrances, essential oils, skin sensates, pigments, pearlescent agents (e.g., mica and titanium dioxide), lakes, colorings, and the like (e.g., clove oil, menthol, camphor, eucalyptus oil, and eugenol).

Method of Use

The striped personal cleansing compositions of the present invention are preferably applied topically to the desired area of the skin or hair in an amount sufficient to provide effective delivery of the skin conditioning agent to the applied surface, or to otherwise provide effective skin conditioning benefits. The compositions can be applied directly to the skin or indirectly via the use of a cleansing puff, washcloth, sponge or other implement. The compositions are preferably diluted with water prior to, during, or after topical application, and then subsequently rinsed or wiped off of the applied surface, preferably rinsed off of the applied surface using water or a water-insoluble substrate in combination with water.

Method of Manufacture

The personal cleansing compositions of the present invention may be prepared by any known or otherwise effective technique, suitable for making and formulating the desired striped product form. It is effective to combine toothpaste-tube filling technology with a spinning stage design. Additionally, the present invention can be prepared by the method and apparatus as disclosed in U.S. patent 6,213,166, herein incorporated by reference. The method and apparatus allows two or more compositions to be filled with a spiral configuration into a single container. The method requires that at least two nozzles be employed to fill the container. The container is placed on a static mixer and spun as the composition is introduced into the container.
Alternatively, it is especially effective to combine at least two phases by first placing the separate compositions in separate storage tanks having a pump and a hose attached. The phases are then pumped in predetermined amounts into a single combining section. Next, the phases are moved from the combining sections into the blending sections and the phases are mixed in the blending section such that the single resulting product exhibits a distinct pattern of the phases. The next step involves pumping the product that was mixed in the blending section via a hose into a single nozzle, then placing the nozzle into a container and filing the container with the resulting product. Specific non-limiting examples of such methods as they are applied to specific embodiments of the present invention are described in the following examples.

If the personal cleansing compositions contain stripes of varying colors it may be desirable to package these compositions in a transparent package such that the consumer can view the pattern through the package. Because of the viscosity of the subject compositions it may also be desirable to include instructions to the consumer to store the package upside down, on its cap to facilitate dispensing.

**Analytical Methods**

**Lather Volume**

Lather volume of a striped liquid personal cleansing composition is measured using a graduated cylinder and a tumbling apparatus. A 1,000 ml graduated cylinder is chosen which is marked in 10 ml increments and has a height of 14.5 inches at the 1,000 ml mark from the inside of its base (for example, Pyrex No. 2982). Distilled water (100 grams at 23°C) is added to the graduated cylinder. The cylinder is clamped in a rotating device, which clamps the cylinder with an axis of rotation that transects the center of the graduated cylinder. One gram of the total personal cleansing composition (0.5g of the cleansing phase and 0.5g of the benefit phase when measuring the total product, or 1 g of the cleansing phase when the measuring the cleansing phase only) is added into the graduated cylinder and the cylinder is capped. The cylinder is rotated at a rate of 10 revolutions in about 20 seconds, and stopped in a vertical position to complete the first rotation sequence. A timer is set to allow 30 seconds for the lather thus generated to drain. After 30 seconds of such drainage, the first lather volume is measured to the nearest 10 ml mark by recording the lather height in ml up from the base (including any water that has drained to the bottom on top of which the lather is floating).

If the top surface of the lather is uneven, the lowest height at which it is possible to see halfway across the graduated cylinder is the first lather volume (ml). If the lather is so coarse that a single or only a few foam cells ("bubbles") reach across the entire cylinder, the height at which at least 10 foam cells are required to fill the space is the first lather volume, also in ml up from the base. Foam cells larger than one inch in any dimension, no matter where they occur, are
designated as unfilled air instead of lather. Foam that collects on the top of the graduated cylinder but does not drain is also incorporated in the measurement if the foam on the top is in its own continuous layer, by adding the ml of foam collected there using a ruler to measure thickness of the layer, to the ml of foam measured up from the base. The maximum foam height is 1,000 ml (even if the total foam height exceeds the 1,000 ml mark on the graduated cylinder). One minute after the first rotation is completed, a second rotation sequence is commenced which is identical in speed and duration to the first rotation sequence. The second lather volume is recorded in the same manner as the first, after the same 30 seconds of drainage time. A third sequence is completed and the third lather volume is measured in the same manner, with the same pause between each for drainage and taking the measurement.

The lather result after each sequence is added together and the Total Lather Volume determined as the sum of the three measurements, in ml. The Flash Lather Volume is the result after the first rotation sequence only, in ml, i.e., the first lather volume. Compositions according to the present invention perform significantly better in this test than similar compositions in conventional emulsion form.

**Viscosity of the Liquid Personal Cleansing Composition**

The Wells-Brookfield Cone/Plate Model DV-II+ Viscometer can be used to determine the viscosity of the liquid personal cleansing compositions herein. The determination is performed at 25°C with the 2.4cm⁰ cone measuring system with a gap of 0.013mm between the two small pins on the respective cone and plate. The measurement is performed by injecting 0.5ml of the sample, to be analyzed, between the cone and plate and, then, rotating the cone at a set speed of 1 rpm. The resistance to the rotation of the cone produces a torque that is proportional to the shear stress of the liquid sample. The amount of torque is read 2 minutes after loading the sample and computed by the viscometer into absolute centipoise units (mPa*s) based on the geometric constant of the cone, the rate of rotation, and the stress related torque.

**Yield Point of Liquid Personal Cleansing Composition**

The Carrimed CSL 100 Controlled Stress Rheometer can be used to determine the yield point of the liquid personal cleansing compositions. For purpose herein, the yield point is the amount of stress required to produce a strain of 1% on the liquid personal cleansing composition. The determination is performed at 77°F with the 4cm ⁰ cone measuring system set with a 51 micron gap. The determination is performed via the programmed application of a shear stress (typically from about 0.06 dynes/sq. centimeter to about 500 dynes/sq. centimeter) over time interval of 5 minutes. It is this amount of stress that results in a deformation of the sample, a shear stress vs. strain curve can be created. From this curve, the yield point of the liquid personal cleansing composition can be calculated.
EXAMPLES

The following examples further describe and demonstrate embodiments within the scope of the present invention. The examples are given solely for the purpose of illustration and are not to be construed as limitations of the present invention, as many variations thereof are possible without departing from the spirit and scope of the invention. All exemplified amounts are concentrations by weight of the total composition, i.e., wt/wt percentages, unless otherwise specified.

Each of the exemplified compositions provides improved deposition or effectiveness of the skin conditioning agents or optional ingredients delivered from each prepared composition. Examples 1-3.

The following examples described in Table 1 are non-limiting examples of the personal cleaning compositions herein.

Table 1: Cleansing Phase and Benefit phase Compositions

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Example 1</th>
<th>Example 2</th>
<th>Example 3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ammonium Laureth-3 Sulfate</td>
<td>3.0</td>
<td>3.0</td>
<td>3.0</td>
</tr>
<tr>
<td>Sodium Lauroamphoacetate (Miranol L-32 Ultra from Rhodia)</td>
<td>16.7</td>
<td>16.7</td>
<td>16.7</td>
</tr>
<tr>
<td>Ammonium Lauryl Sulfate</td>
<td>1.0</td>
<td>1.0</td>
<td>1.0</td>
</tr>
<tr>
<td>Lauric Acid</td>
<td>0.9</td>
<td>0.9</td>
<td>0.9</td>
</tr>
<tr>
<td>Trihydroxystearin (Thixcin R)</td>
<td>2.0</td>
<td>2.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Guar Hydroxypropyltrimonium Chloride (N-Hance 3196 from Aqualon)</td>
<td>0.17</td>
<td>0.75</td>
<td>0.75</td>
</tr>
<tr>
<td>Guar Hydroxypropyltrimonium Chloride (Jaguar C-17 from Rhodia)</td>
<td>0.58</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Polysquaternium 10 (UCARE polymer JR-30M from Amerchol)</td>
<td>0.45</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Polymethacrylamidopropyltrimonium Chloride (Polycare 133 from Rhodia)</td>
<td>-</td>
<td>0.24</td>
<td>-</td>
</tr>
<tr>
<td>Polysquaternium-39 (Merquart Plus 3300 from Calgon )</td>
<td>-</td>
<td>0.81</td>
<td>-</td>
</tr>
<tr>
<td>PEG 90M (Polyox WSR 301 from Union Carbide)</td>
<td>0.25</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Ingredient</td>
<td>0.45</td>
<td>2.45</td>
<td>2.45</td>
</tr>
<tr>
<td>------------</td>
<td>------</td>
<td>------</td>
<td>------</td>
</tr>
<tr>
<td>PEG-14M (Polyox WSR N-3000 H from Union Carbide)</td>
<td>-</td>
<td>1.0</td>
<td>4.0</td>
</tr>
<tr>
<td>Linoleamidopropyl PG-Dimonium Chloride Phosphate Dimethicone (Monasil PLN from Uniqema)</td>
<td>-</td>
<td>1.0</td>
<td>4.0</td>
</tr>
<tr>
<td>Glycerin</td>
<td>1.4</td>
<td>4.9</td>
<td>4.9</td>
</tr>
<tr>
<td>Sodium Chloride</td>
<td>0.3</td>
<td>0.3</td>
<td>0.3</td>
</tr>
<tr>
<td>Sodium Benzoate</td>
<td>0.25</td>
<td>0.25</td>
<td>0.25</td>
</tr>
<tr>
<td>Disodium EDTA</td>
<td>0.13</td>
<td>0.13</td>
<td>0.13</td>
</tr>
<tr>
<td>Glydant</td>
<td>0.37</td>
<td>0.37</td>
<td>0.37</td>
</tr>
<tr>
<td>Citric Acid</td>
<td>1.6</td>
<td>0.95</td>
<td>0.95</td>
</tr>
<tr>
<td>Titanium Dioxide</td>
<td>0.5</td>
<td>0.5</td>
<td>0.5</td>
</tr>
<tr>
<td>Perfume</td>
<td>0.5</td>
<td>0.5</td>
<td>0.5</td>
</tr>
<tr>
<td>Water</td>
<td>Q.S.</td>
<td>Q.S.</td>
<td>Q.S.</td>
</tr>
</tbody>
</table>

**II. Benefit Composition**

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>80</th>
<th>80</th>
<th>80</th>
</tr>
</thead>
<tbody>
<tr>
<td>Petrolatum</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PEG-30 Dipolyhydroxystearate (Arlacel P135)</td>
<td>1</td>
<td>1</td>
<td>1</td>
</tr>
<tr>
<td>Water</td>
<td>19</td>
<td>19</td>
<td>19</td>
</tr>
</tbody>
</table>

The cleansing phase and benefit phase compositions described above can be prepared by conventional formulation and mixing techniques. The cleansing composition 1 can be prepared by first creating the following premixes: citric acid in water premix at 1:3 ratio, Guar polymer premix with Jaguar C-17 and N-Hance 3196 in water at 1:10 ratio, UCARE premix with JR-30M in water at about 1:30 ratio, and Polyox premix with PEG-90M and PEG-14M in Glycerin at about 1:2 ratio. Then, the following ingredients will be added into the main mixing vessel: ammonium lauryl sulfate, ammonium laureth-3 sulfate, citric acid premix, Miranol L-32 ultra, sodium chloride, sodium benzoate, disodium EDTA, lauric acid, Thixcin R, Guar premix, UCARE premix, Polyox Premix, and the rest of water. Then one will heat the vessel with agitation until it reaches 190°F (88°C). Let it mix for about 10 min. Cool the batch with a cold water bath with slow agitation until it reaches 110°F (43°C). Add the following ingredients: Glydant, perfume, Titanium Dioxide. Mix until a homogeneous solution forms.

The cleansing composition 2 can be prepared by first creating the following premixes: citric acid in water premix at 1:3 ratio, Guar polymer premix with N-Hance 3196 in water at 1:10
ratio, and Polyox premix with PEG-14M in Glycerin at about 1:2 ratio. Then, the following ingredients will be added into the main mixing vessel: ammonium lauryl sulfate, ammonium laureth-3 sulfate, citric acid premix, Miranol L-32 ultra, sodium chloride, sodium benzoate, disodium EDTA, lauric acid, Thixcin R, Guar premix, Polyox Premix, Polycare 133, Merquat Plus 3300, Monasil PLN, and the rest of water. Then, the vessel will be heated with agitation until it reaches 190°F (88°C). Let it mix for about 10 min. Next, the batch will be cooled with a cold water bath with slow agitation until it reaches 110°F (43°C). Finally, the following ingredients will be added: Glydant, perfume, Titanium Dioxide and mixed until a homogeneous solution forms.

The cleansing composition 3 can be prepared by first creating the following premixes: citric acid in water premix at 1:3 ratio, Guar polymer premix with N-Hance 3196 in water at 1:10 ratio, and Polyox premix with PEG-14M in Glycerin at about 1:2 ratio. Then, the following ingredients will be added into the main mixing vessel: ammonium lauryl sulfate, ammonium laureth-3 sulfate, citric acid premix, Miranol L-32 ultra, sodium chloride, sodium benzoate, disodium EDTA, lauric acid, Thixcin R, Guar premix, Polyox Premix, Monasil PLN, and the rest of water. Then the vessel will be heated with agitation until it reaches 190°F (88°C). The vessel will be mixed for about 10 min. Next, the batch will be cooled with a cold water bath with slow agitation until it reaches 110°F (43°C). Finally, the following ingredients will be added: Glydant, perfume, Titanium Dioxide and mixed until a homogeneous solution forms.

Benefit phase

The benefit phase can be prepared by adding Petrolatum into the main mixing vessel. Then, the vessel will be heated to 185°F and add Arlacel P135. Then, slowly add water with agitation. Keep agitating until homogeneous.

The cleansing and benefit phases are packaged into a single container by first placing the separate compositions in separate storage tanks having a pump and a hose attached. The phases are then pumped in predetermined amounts into a single combining section. Next, the phases are moved from the combining sections into the blending sections and the phases are mixed in the blending section such that the single resulting product exhibits a distinct pattern of the phases. The next step involves pumping the product that was mixed in the blending section via a hose into a single nozzle, then placing the nozzle into a container and filing the container with the resulting product. The sample stage spins the bottle during the filling process to create a striped appearance.

Examples 4-6
The following examples described in Table 2 are non-limiting examples of the personal cleaning compositions herein. Table 2: Cleansing Phase and Benefit phase Compositions

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Example 4</th>
<th>Example 5</th>
<th>Example 6</th>
</tr>
</thead>
<tbody>
<tr>
<td>Miracare SLB-365 (from Rhodia)</td>
<td>47.4</td>
<td>47.4</td>
<td>47.4</td>
</tr>
<tr>
<td>(Sodium Trideceth Sulfate, Sodium Lauramphoacetate, Cocamide MEA)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Guar Hydroxypropyltrimonium Chloride</td>
<td></td>
<td>-</td>
<td>0.7</td>
</tr>
<tr>
<td>(N-Hance 3196 from Aqualon)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PEG 90M (Polyox WSR 301 from Dow Chemical)</td>
<td>-</td>
<td>-</td>
<td>0.2</td>
</tr>
<tr>
<td>Cocamide MEA</td>
<td>3.0</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Polycare 133</td>
<td>-</td>
<td>-</td>
<td>0.4</td>
</tr>
<tr>
<td>Lauric Acid</td>
<td>-</td>
<td>2.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Sodium Chloride</td>
<td>3.5</td>
<td>3.5</td>
<td>3.5</td>
</tr>
<tr>
<td>Disodium EDTA</td>
<td>0.05</td>
<td>0.05</td>
<td>0.05</td>
</tr>
<tr>
<td>Glydant</td>
<td>0.67</td>
<td>0.67</td>
<td>0.67</td>
</tr>
<tr>
<td>Citric Acid</td>
<td>0.4</td>
<td>0.4</td>
<td>0.4</td>
</tr>
<tr>
<td>Perfume</td>
<td>2.0</td>
<td>2.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Water</td>
<td>Q.S.</td>
<td>Q.S.</td>
<td>Q.S.</td>
</tr>
<tr>
<td>(pH)</td>
<td>(6.0)</td>
<td>(6.0)</td>
<td>(6.0)</td>
</tr>
</tbody>
</table>

II. Benefit phase Composition

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Example 4</th>
<th>Example 5</th>
<th>Example 6</th>
</tr>
</thead>
<tbody>
<tr>
<td>Petrolatum</td>
<td></td>
<td>80</td>
<td>80</td>
</tr>
<tr>
<td>PEG-30 Dipolyhydroxystearate (Arlacel P135)</td>
<td>1</td>
<td>1</td>
<td>1</td>
</tr>
<tr>
<td>Water</td>
<td></td>
<td>19</td>
<td>19</td>
</tr>
</tbody>
</table>

The compositions described above can be prepared by conventional formulation and mixing techniques. The cleansing phase composition can be prepared by first adding citric acid into water at 1:3 ratios to form a citric acid premix. The following ingredients will then be added into the main mixing vessel in the following sequence: water, Miracare SLB-354, sodium chloride, sodium benzoate, Disodium EDTA, glydant. The main mixing vessel will start to be agitated. In a separate mixing vessel, disperse polymers (Polyquaterium 10, Jaguar C-17, or N-Hance 3196) in water at 1:10 ratio will form a polymer premix. The completely dispersed
polymer premix will be added into the main mixing vessel with continuous agitation. Polyox WSR 301 will be dispersed in water and then added to the main mixing vessel. Then, the rest of the water and perfume will be added into the batch. The batch will be kept agitating until a homogenous solution forms.

Benefit phase

The benefit phase can be prepared by adding Petrolatum into the main mixing vessel. Then, the vessel will be heated to 185F and add Arlacel P135. Then, slowly add water with agitation. Keep agitating until homogeneous.

The cleansing and benefit phases are packaged into a single container by first placing the separate compositions in separate storage tanks having a pump and a hose attached. The phases are then pumped in predetermined amounts into a single combining section. Next, the phases are moved from the combining sections into the blending sections and the phases are mixed in the blending section such that the single resulting product exhibits a distinct pattern of the phases. The next step involves pumping the product that was mixed in the blending section via a hose into a single nozzle, then placing the nozzle into a container and filing the container with the resulting product. The sample stage spins the bottle during filling process to create a striped appearance.

Examples 7-9

The following examples described in Table 3 are non-limiting examples of the personal cleaning compositions herein. Table 3: Cleansing Phase and Benefit phase Compositions

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Example 7</th>
<th>Example 8</th>
<th>Example 9</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>wt%</td>
<td>wt%</td>
<td>wt%</td>
</tr>
<tr>
<td><strong>I. Cleansing Phase Composition</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Miracare SLB-365 (from Rhodia)</td>
<td>47.4</td>
<td>47.4</td>
<td>47.4</td>
</tr>
<tr>
<td>(Sodium Trideceth Sulfate, Sodium Lauramphoacetate, Cocamide MEA)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Sodium Chloride</td>
<td>3.5</td>
<td>3.5</td>
<td>3.5</td>
</tr>
<tr>
<td>Disodium EDTA</td>
<td>0.05</td>
<td>0.05</td>
<td>0.05</td>
</tr>
<tr>
<td>Glydant</td>
<td>0.67</td>
<td>0.67</td>
<td>0.67</td>
</tr>
<tr>
<td>Citric Acid</td>
<td>0.4</td>
<td>0.4</td>
<td>0.4</td>
</tr>
<tr>
<td>Perfume</td>
<td>2.0</td>
<td>2.0</td>
<td>2.0</td>
</tr>
<tr>
<td>Water</td>
<td>Q.S.</td>
<td>Q.S.</td>
<td>Q.S.</td>
</tr>
<tr>
<td>(pH)</td>
<td>(6.0)</td>
<td>(6.0)</td>
<td>(6.0)</td>
</tr>
<tr>
<td><strong>II. Benefit phase Composition</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Petrolatum</td>
<td>80</td>
<td>80</td>
<td>80</td>
</tr>
<tr>
<td>PEG-30 Dipolyhydroxystearate (Arlacel P135)</td>
<td>1</td>
<td>1</td>
<td>1</td>
</tr>
<tr>
<td>Water</td>
<td>19</td>
<td>19</td>
<td>19</td>
</tr>
</tbody>
</table>

The compositions described above can be prepared by conventional formulation and mixing techniques. The cleansing phase composition can be prepared by first adding citric acid into water at 1:3 ratio to form a citric acid premix. The following ingredients will be added into the main mixing vessel in the following sequence: water, Miracare SLB-354, sodium chloride, sodium benzoate, Disodium EDTA, glydant. The main mixing vessel will start to be agitated. Then, perfume will be added into the batch. The batch will be kept agitating until a homogenous solution forms.

Benefit phase

The benefit phase can be prepared by adding Petrolatum into the main mixing vessel. Then, the vessel will be heated to 185°F and add Arlacel P135. Then, slowly add water with agitation. Keep agitating until homogeneous.

The cleansing and benefit phases are packaged into a single container by first placing the separate compositions in separate storage tanks having a pump and a hose attached. The phases are then pumped in predetermined amounts into a single combining section. Next, the phases are moved from the combining sections into the blending sections and the phases are mixed in the blending section such that the single resulting product exhibits a distinct pattern of the phases. The next step involves pumping the product that was mixed in the blending section via a hose into a single nozzle, then placing the nozzle into a container and filing the container with the resulting product. The sample stage spins the bottle during the filling process to create a striped appearance.
WHAT IS CLAIMED IS:

1. A striped personal cleansing composition comprising:
   (a) a first stripe comprising a cleansing phase comprising a surfactant and water; and
   (b) at least one additional stripe comprising a benefit phase comprising a water in oil
      emulsion;

      wherein the cleansing phase and the benefit phase are in physical contact with one
      another and maintain stability.

2. A striped personal cleansing composition according to Claim 1, wherein the cleansing
   phase comprises:
      (i) at least one anionic surfactant;
      (ii) at least one electrolyte;
      (iii) at least one alkanolamide;
      (iv) water; and

      wherein the cleansing phase is non-Newtonian shear thinning, and has a viscosity of equal
      to or greater than 3000 cps.

3. A striped personal cleansing composition according to any one of the preceding claims,
   wherein the surfactant comprises from 3% to 60% by weight of the aqueous cleansing phase.

4. A striped personal cleansing composition according to Claim 2, wherein the electrolyte
   comprises:
      i) an anion selected from the group consisting of phosphate, chloride, sulfate, citrate
         and mixtures thereof, and
      ii) a cation selected from the group consisting of sodium, ammonium, potassium,
         magnesium and mixtures thereof; and wherein the electrolyte is present from 0.1%
         to 15% by weight of the cleansing phase.

5. A striped personal cleansing composition according to Claim 1 wherein said benefit phase
   comprises an emulsifier having HLB below 10 and wherein said emulsifier is selected from PEG-
   30 dipolyhydroxystearate, dimethicone copolyol, and mixtures thereof.

6. A striped personal cleansing composition according to any one of the preceding claims,
   wherein said benefit phase comprises from 10% to 90% of an oil.
7. A striped personal cleansing composition according to any one of the preceding claims, which comprises a density modifier; preferably wherein the density modifier is a hollow microsphere.

8. A striped personal cleansing composition according to any one of the preceding claims, further comprising a cationic deposition polymer in said cleansing phase; preferably wherein cationic deposition polymer is selected from the group of cationic cellulosic derivatives, cationic guar derivatives, cationic synthetic polymers, and mixtures thereof.

9. A striped personal cleansing composition according to any one of the preceding claims, wherein the cleansing phase additionally comprises a lamellar structurant; preferably wherein the lamellar structurant is selected from fatty acids, fatty esters, trihydroxystearin, fatty alcohols, and mixture thereof.

10. A striped personal cleansing composition according to any one of the preceding claims, wherein at least one phase contains a colorant; preferably wherein the cleansing and benefit phases visually form a pattern within the package; more preferably wherein the pattern is selected from the group consisting of striped, marbled, geometric, and mixtures thereof; even more preferably wherein the composition is packaged in a transparent container.

11. A striped personal cleansing composition according to any one of the preceding claims, wherein the composition is packaged in a container with instructions to store said container on the lid.

12. A striped personal cleansing composition according to any one of the preceding claims, wherein the composition comprises skin care actives, wherein the skin care actives are selected from the group consisting of vitamins and derivatives thereof; sunscreens; thickening agents; preservatives; anti-acne medicaments; antioxidants; skin soothing and healing; chelators and sequestrants; fragrances, essential oils, skin sensates, pigments, pearlescent agents, lakes, colorings, and mixtures thereof.

13. A striped personal cleansing composition comprising:
   a) a first stripe comprising a cleansing phase comprising from 1% to 50% by weight of the cleansing phase of a surfactant selected from the group consisting of anionic
surfactant, non-ionic surfactant, zwitterionic surfactant, cationic surfactant, soap and mixtures thereof;

wherein the cleansing phase is non-Newtonian shear thinning, has a viscosity of equal to or greater than 3,000 cps and a yield value of at least 0.1 Pa; and

b) at least one additional stripe comprising a benefit phase comprising a water in oil emulsion comprising an oil, an emulsifier and water;

wherein the ratio between the cleansing phase and the benefit phase is from 1:9 to 99:1; and wherein the cleansing phase and benefit phase are present as stripes wherein the stripe size is at least 0.1 mm in width and at least 1 mm in length; and wherein the cleansing phase and the benefit phase are in physical contact with one another and maintain stability.

14. A method of delivering skin conditioning benefits to the skin or hair, said method comprising the steps of:

a) dispensing an effective amount of a composition according to claim 1 onto an implement selected from the group consisting of a cleansing puff, washcloth, sponge and human hand;

b) topically applying said composition to the skin or hair using said implement; and

c) removing said composition from the skin or hair by rinsing with water.