

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
15 September 2005 (15.09.2005)

PCT

(10) International Publication Number
WO 2005/084798 A1

(51) International Patent Classification⁷: **B01J 29/04**,
29/87

TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA,
ZM, ZW.

(21) International Application Number:
PCT/US2005/006757

(84) Designated States (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(22) International Filing Date: 2 March 2005 (02.03.2005)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
10/790,948 2 March 2004 (02.03.2004) US

Declarations under Rule 4.17:

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations* AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW, *ARIPO patent* (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), *Eurasian patent* (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), *European patent* (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), *OAPI patent* (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii)) for all designations*

(71) Applicant (*for all designated States except US*): SAUDI BASIC INDUSTRIES CORPORATION [SA/SA]; P.O. Box 5101, Riyadh 11422 (SA).

(71) Applicants and

(72) Inventors: GHOSH, Ashim, Kumar [US/US]; 3418 E. Pine Brook Way, Houston, TX 77059 (US). HARVEY, Pamela [US/US]; 9033 Sharp Crest, Houston, TX 77036 (US).

(74) Agent: WHEELINGTON, Jimmy, Dale; SABIC Americas, Inc., SABIC Technology Center, 1600 Industrial Blvd., Sugar Land, TX 77478 (US).

Published:

- *with international search report*

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(81) Designated States (*unless otherwise indicated, for every kind of national protection available*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ,

(54) Title: SELECTIVE ZEOLITE CATALYST MODIFICATION

(57) Abstract: A method of modifying a zeolite catalyst to increase selectivity of the catalyst is achieved by dissolving alumina in a phosphorus-containing acid solution, and treating the zeolite catalyst with the dissolved alumina solution. A method of preparing an aromatic product, such as a xylene product, is also achieved by contacting the modified zeolite catalyst with an aromatic hydrocarbon, such as toluene, and an alkylating agent, such as methanol, under reaction conditions suitable for aromatic alkylation. For xylene products the aromatic hydrocarbon may be toluene and the reaction conditions may be suitable for at least one of toluene methylation and transalkylation.

WO 2005/084798 A1

SELECTIVE ZEOLITE CATALYST MODIFICATION

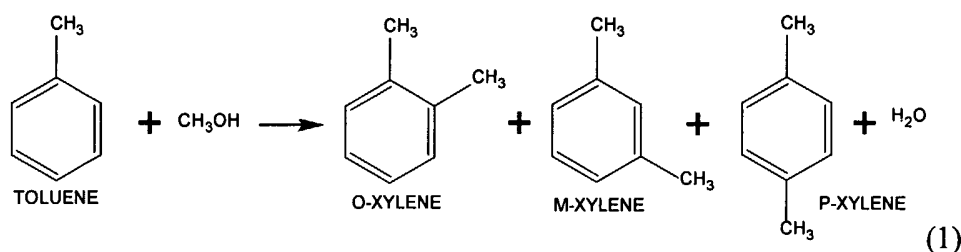
TECHNICAL FIELD

[0001] The invention relates generally to the alkylation of aromatic compounds and catalysts used for such reactions.

BACKGROUND

[0002] Para-xylene is a valuable substituted aromatic compound because of its great demand for its oxidation to terephthalic acid, a major component in forming polyester fibers and resins. It can be commercially produced from hydrotreating of naphtha (catalytic reforming), steam cracking of naphtha or gas oil, and toluene disproportionation.

[0003] Alkylation of toluene with methanol, which is also known as toluene methylation, has been used in laboratory studies to produce para-xylene. Toluene methylation has been known to occur over acidic catalyst, particularly over zeolite or zeolite-type catalyst. In particular, ZSM-5-type zeolite, zeolite Beta and silicaaluminophosphate (SAPO) catalysts have been used for this process. Generally, a thermodynamic equilibrium mixture of ortho (o)-, meta (m)- and para (p)-xylenes can be formed from the methylation of toluene, as is illustrated by the reaction below.



[0004] Thermodynamic equilibrium compositions of o-, m-, and p-xylenes may be around 25, 50 and 25 mole%, respectively, at a reaction temperature of about 500 °C. Such toluene methylation may occur over a wide range of temperatures, however. Para-xylene can be separated from mixed xylenes by a cycle of adsorption and isomerization. Byproducts such as C₉+ and other aromatic products can be produced by secondary alkylation of the xylene product.

[0005] A significantly higher amount of p-xylene can be obtained in toluene methylation reaction if the catalyst has shape selective properties. Shape selective properties can be obtained in modified zeolite catalysts by narrowing zeolite pore opening size, inactivation of the external surface of the zeolite or controlling zeolite acidity. Toluene methylation may occur over modified ZSM-5 or ZSM-5-type zeolite catalyst giving xylene products containing significantly greater amounts of p-xylene than the thermodynamic concentration.

[0006] In Kaeding, et al, *Selective Alkylation of Toluene with Methanol to Produce para-Xylene*, Journal of Catalysis, Vol. 67, pp. 159-174 (1981), a procedure of making a ZSM-5 catalyst by incorporating 5% phosphorus was described in which the catalyst was impregnated with a solution of diphenylphosphinous acid in toluene. The ZSM-5 catalyst thus modified showed toluene methylation activity with 84-90% para isomer in the xylene product. In another procedure, a catalyst was modified by incorporating 8.51% phosphorus from an aqueous phosphoric acid reagent. The catalyst showed p-xylene selectivity as high as 97%, however, the catalyst showed a decreasing activity within hours due to coke deposition.

[0007] Unfortunately, there are a number of technical hurdles for toluene methylation to be commercially successful. These include fast catalyst deactivation, low methanol selectivity, and so on. Most, if not all, of the catalysts used for toluene methylation show fast catalyst deactivation. Typically, toluene conversion declines with time on stream due to rapid coke formation on the catalyst. The catalyst deactivation is one of the most difficult technical hurdles to overcome for commercial use of toluene methylation.

BRIEF DESCRIPTION OF THE DRAWINGS

[0008] For a more complete understanding of the present invention, reference is now made to the following descriptions taken in conjunction with the accompanying figures, in which:

[0009] FIGURE 1 is a plot of NH₃-TPD for alumina-PA modified ZSM-5 (E), PA modified ZSM-5 (L), and non-modified ZSM-5 (P) catalysts.

[0010] FIGURE 2 is a plot of para-xylene selectivity as a function of phosphoric acid usage for zeolite catalysts treated with phosphoric acid only and those modified in accordance with the invention; and

[0011] FIGURE 3 is a plot of total xylene conversion and para-xylene conversion over time for catalysts treated with phosphoric acid only and that modified in accordance with the invention.

DETAILED DESCRIPTION

[0012] The modification of ZSM-5-type zeolite catalysts with phosphoric acid (PA) has been shown to yield significantly greater amounts of p-xylene than the thermodynamic equilibrium value in toluene methylation with unmodified catalysts. It has been found, however, that increased selectivity for para-xylene beyond that achieved from phosphoric acid-only modified zeolites, given the same quantity of phosphoric acid usage, can be achieved by incorporating dissolved or "digested" alumina into such catalysts.

[0013] As used herein, the expression "ZSM-5-type" is meant to refer to those zeolites that are isostructurally the same as ZSM-5 zeolites. Additionally, the expressions "ZSM-5" and "ZSM-5-type" may also be used herein interchangeably to encompass one another and should not be construed in any limiting sense. The ZSM-5 zeolite catalysts and their preparation are described in U.S. Patent No. 3,702,886, which is herein incorporated by reference. In the present invention, the ZSM-5 zeolite catalyst may include those having a silica/alumina molar ratio of from 25 to 1000 prior to modification, more particularly a silica/alumina molar ratio of from about 30 to about 300 prior to modification.

[0014] Although specific reference has been made to ZSM-5-type zeolites for use in toluene methylation, for which the invention has particular application, the modification discussed herein may have application to other zeolites with pore diameters ranging from 5.0 Å to 7.0 Å, such as mordenite, omega, etc. Additionally, specific reference is made to the use of the modified catalyst in toluene methylation, for which the modified zeolite is particularly well suited. It will be apparent to those skilled in the art, however, that the catalyst may have application for use in other types of reactions, such as transalkylation and other aromatic alkylation reactions. In particular, the catalyst of the invention may have application to such reactions to provide increased selectivity for para-isomer in mixed dialkylated aromatic products.

[0015] As used herein, catalytic activity can be expressed as the % moles of toluene converted with respect to the moles of toluene fed and can be defined as:

$$\text{Mole\% Toluene Conversion} = (T_i - T_o / T_i) \times 100 \quad (2)$$

where, T_i is the number of moles of toluene fed and T_o is the number of moles toluene unreacted. As used herein, selectivity for total xylenes may be expressed as:

$$\text{Mole\% Total Xylene Selectivity} = (X_{\text{tx}}/T_i - T_o) \times 100 \quad (3)$$

where, X_{tx} is the number of moles of total (o-, m- or p-) xylenes in the product.

As used herein, selectivity for p-xylene may be expressed as:

$$\text{Mole\% p-Xylene Selectivity} = (X_p/X_{\text{tx}}) \times 100 \quad (4)$$

where, X_p is the number of moles of p-xylene.

[0016] The zeolite may be modified by treating the zeolite with alumina that has been pre-dissolved in a phosphorus (P) containing inorganic acid, such as phosphoric acid (H_3PO_4 , $\text{pKa}=2.12$ for step 1) or phosphorus acid (H_3PO_3 , $\text{pKa}=2.00$ for step 1). Although specific mention of phosphoric acid is made in the description that follows, it should be apparent that other phosphorus-containing acids could be used instead. The alumina, which is initially in a solid form, is dissolved in the acid solution. The phosphoric acid solution may be of sufficient concentration and quantity to dissolve all or substantially all of the alumina, which may be evident when the solution turns clear. Any remaining or undissolved alumina may be filtered or otherwise removed from the solution prior to use in modifying the zeolite catalyst, if necessary.

[0017] In dissolving the alumina, the phosphoric acid may be used in excess amounts, as is discussed further on. Examples of suitable amounts of alumina may be from about 0.005 g to about 0.10 g alumina per gram of zeolite powder. Suitable amounts of the phosphoric acid may be from about 0.05 g or more to about 0.5 g per gram of zeolite powder. The acid may be in a concentrated liquid form of at least 50% acid by weight of aqueous solution, such as 85% acid by weight aqueous solution.

[0018] To facilitate dissolving the alumina, it can be first added to water and stirred while heated to an elevated temperature. The phosphoric acid may then be added to the alumina. An example of a suitable temperature is from about 70 to about 100 °C. The dissolved alumina solution may contain an excess amount or residual phosphoric acid.

[0019] Prior to adding the alumina-containing phosphoric acid solution, the zeolite powder may be combined with water to make an aqueous slurry or suspension. The slurry may be heated and stirred to facilitate catalyst preparation. An example of a

suitable temperature range for the zeolite slurry is from about 70 to about 100°C. The dissolved alumina may then be combined with the zeolite slurry. Alternatively, the dissolved alumina may be added to dry zeolite powder to form zeolite slurry.

[0020] Excess or residual phosphoric acid remaining in the dissolved alumina solution may facilitate modification of the zeolite to provide shape selective properties. Alternatively, additional phosphoric acid may be added to the slurry to modify or further facilitate modification of the zeolite to provide shape selective properties. In the case of ZSM-5-type zeolites, these catalysts can be modified to increase selectivity for p-xylene in toluene methylation reactions through modification of the zeolite with phosphoric acid. The incorporation of the alumina from the dissolved alumina solution, however, has been shown to further increase the selectivity of the zeolite beyond that achieved through modification solely with phosphoric acid, as is illustrated by the examples discussed further on.

[0021] The zeolite slurry, which now contains the dissolved alumina/acid solution, may then be heated until all liquids are evaporated. An example of a suitable temperature range is from 70 °C to 100 °C. The slurry may also be stirred or agitated during this step to ensure uniform treatment.

[0022] The alumina-PA-modified zeolite catalyst may be used unbound or be bound with a binder. Examples of suitable binders include such materials as alumina, clay, and silica. Those techniques used for preparing the bound catalyst are well known in the art. The catalyst, bound or unbound, may be calcined at a temperature between 400 °C and 570 °C in an environment containing oxygen, typically air.

[0023] The modified catalyst may be contacted with an appropriate feed under alkylation reaction conditions to carry out aromatic alkylation. Examples of alkylation reactions for which the invention has application include toluene alkylation with an alkylating agent such as methanol. Other alkylation reactions may include transalkylation, such as gas phase toluene disproportionation in the presence of hydrogen to produce benzene and mixed xylenes.

[0024] The reactor pressure for toluene methylation or other aromatic alkylation may vary, but typically ranges from about 10 to about 1000 psig.

[0025] The reaction may be carried in a variety of different reactors that are commonly used for carrying out aromatic alkylation reactions. Single or multi

reactors in series and/or parallel are suitable for carrying out the toluene methylation or other aromatic alkylation reactions.

[0026] In particular, the modified catalyst is useful in toluene methylation for preparing a xylene product from a feed of toluene and methanol that has increased selectivity for p-xylene. In such reactions, water may be introduced with the feed in an amount of at least 0.1 moles water per mole of toluene/methanol feed, as described in US. Patent Application Serial No. 10/675,780, filed September 30, 2003, which is herein incorporated by reference. A hydrogen cofeed is also used. The hydrogen may be used in an amount of at least 1.0 mole per mole of toluene/methanol feed.

[0027] When ZSM-5-type zeolite catalysts modified in accordance with the invention are used in toluene methylation, a xylene product having a p-xylene content of 80%, 85%, 90% or 95% or more by total moles of xylene may be obtained.

[0028] The following examples further illustrate the invention.

EXAMPLES

CATALYST PREPARATION

Catalysts A-H

[0029] To form each catalyst, a slurry of NH₄-ZSM-5 zeolite powder having a SiO₂/Al₂O₃ mole ratio of 280 in 50 ml of deionized water was prepared in a 400 ml beaker. The beaker was placed on a hot plate and the zeolite suspension was stirred using a magnetic stir bar. The temperature of the zeolite suspension (or slurry) was brought to around 80-85 °C. A solution of dissolved alumina in phosphoric acid was also prepared by adding an amount of alumina to 10 ml of deionized water. The alumina in water was then heated and stirred until the temperature was about 70 to 80 °C, at which point an amount of phosphoric acid (H₃PO₄) (85 wt% in aqueous) was slowly added to the slurry. A clear solution was eventually obtained indicating that all of the alumina had dissolved. The alumina/phosphoric acid solution was then added to the zeolite slurry. Heating and stirring of the slurry continued until substantially all liquids were evaporated. The zeolite was then dried at about 90 °C overnight and then was calcined in air at approximately 510 °C for 10 hrs. The modified and calcined zeolite was then crushed and sized using 20 and 40 mesh screens.

[0030] A series of catalysts A-H (see Table 1) were prepared using the above technique by varying the amount of alumina and the phosphoric acid with respect to the amount of starting NH_4 -ZSM-5 powder. The BET surface area and total pore volume (measured by N_2 adsorption) for catalysts A-H are presented in Table 1 below.

Comparative Catalysts I-O

[0031] For comparison purposes, ZSM-5 zeolite catalysts were treated with phosphoric acid, but without the dissolved alumina. The starting material was an NH_4 -ZSM-5 zeolite powder having a $\text{SiO}_2/\text{Al}_2\text{O}_3$ mole ratio of 280. For each catalyst, a slurry containing NH_4 -ZSM-5 zeolite and 100-150 ml of deionized water was prepared in a 400 ml beaker. The beaker was placed on a hot plate and the zeolite suspension was stirred using a magnetic stir bar. The temperature of the suspension was maintained around 90 °C. Phosphoric acid (85 wt% in aqueous) was added drop wise into the beaker. Heating was continued until all liquid was evaporated. The phosphoric-acid-only modified zeolite was dried at 90°C to 110 °C for at least four hours and then was calcined at 510 °C under air for 10 hours. The calcined zeolite was then crushed and sized using 20 and 40 mesh screens.

[0032] A series of catalysts I-O (see Table 1) were prepared by varying the amount of phosphoric acid with respect to the amount of starting NH_4 -ZSM-5 powder. BET surface area and total pore volume (measured by N_2 adsorption) for catalysts I-O are presented in Table 1.

Comparative Catalyst P

[0033] A non-modified ZSM-5 zeolite catalyst (Catalyst P) was also tested. The starting material was an NH_4 -ZSM-5 zeolite powder having a $\text{SiO}_2/\text{Al}_2\text{O}_3$ mole ratio of 280. The zeolite powder was calcined at 530 °C under air for 10 hr and then pressed and sized using 20 and 40 mesh screens for use in reactor for toluene methylation reaction.

EXAMPLE 1

[0034] The catalysts A-P, as referenced in Table 1 and prepared as described above, were used in toluene methylation reactions. The reactions were each carried out in a

fixed bed, continuous flow type reactor. In each case, the catalyst used was dried by slowly raising the catalyst bed temperature (about 5 °C/min) to 200 °C under hydrogen (H₂) flow for at least one hour. A premixed toluene and methanol feed (molar ratio 2/1) was added to the reactor at 200 °C and the catalyst bed inlet temperature was increased to about 500 °C. The liquid hourly space velocity (LHSV) based on toluene/methanol feed was maintained at about 31 hr⁻¹ and cofeed of H₂ gas was fed and maintained to provide a H₂/HC molar ratio of about 0.1. Water was added to the hydrocarbon (HC) feed and was vaporized prior to introduction to reactor. The H₂O/HC molar ratio was about 0.65 and reactor pressure was about 20 psig. The following results were obtained, as presented in Table 1 below.

TABLE 1

Catalyst #	Al ₂ O ₃ , g/g zeolite	H ₃ PO ₄ , g/g zeolite ^a	P, g/g zeolite ^a	SA, m ² /g	PV, ml/g	% Tol Conv	% Tot Xyl Sel	% PX in TX
A	0.014	0.110	0.035	328	0.192	26.0	93.4	84.5
B	0.007	0.110	0.035	305	0.188	21.4	93.6	87.8
C	0.014	0.167	0.053	298	0.181	18.3	95.6	93.2
D	0.022	0.216	0.068	268	0.167	13.2	94.9	91.9
			0.068	268	0.167	13.3	95.3	93.2
E	0.014	0.220	0.070	224	0.135	13.1	96.4	97.3
			0.070	224	0.135	15.5	95.5	94.8
F	0.022	0.259	0.082	213	0.128	17.0	95.3	93.8
H	0.014	0.220	0.070	223	0.135	17.5	95.1	91.6
I	0	0.076	0.024	331	0.208	28.2	92.9	80.7
J	0	0.139	0.044	299	0.182	27.0	93.4	86.6
K	0	0.278	0.088	229	0.159	22.9	94.9	92.6
L	0	0.278	0.088	183	0.138	19.6	94.2	92.5
M	0	0.545	0.123	156	0.113	18.9	95.3	91.7
N	0	0.390	0.123	156	0.113	18.9	95.3	91.2
O	0	0.390	0.172	164	0.120	17.2	94.0	89.1
P	0	0	0	375	0.244	33.4	89.3	30.8

^aBased on total amount of PA used during preparation.

[0035] A decrease in BET surface area (SA) and total pore volume (PV) of the zeolite catalysts results from the alumina-PA modification, as shown in Table 1. The total pore volume of the modified zeolite catalyst may be from 0.10 ml/g to 0.20 ml/g, more particularly from 0.12 to 0.18 ml/g. For example, the non-modified ZSM-5 catalyst (catalyst P in table 1) has SA and PV of 375 m²/g and 0.244 ml/g, respectively. The alumina-PA modified catalysts, A-H in Table 1, have a SA and PV in the range of 223-328 m²/g and 0.135-0.192 ml/g, respectively.

[0036] The acidity of the modified and non-modified zeolite catalysts were characterized by using ammonia temperature programmed desorption (NH₃-TPD) techniques. This method is well known in the art. For the NH₃-TPD analysis, the catalyst sample (0.2 g) was first dried at 500 °C for 3 hours under a He flow rate of 10 cc/min. The temperature was then reduced to 100 °C whereupon the catalyst was saturated with ammonia gas. After saturation with NH₃, the catalyst desorbed at 100 °C with He flow to desorb physisorbed NH₃ from the sample. NH₃-TPD was performed at a desorption temperature ramp of 18.4 °C/min under He flow rate of 16 cc/min. The desorbed NH₃ and water (if any) were monitored during the NH₃-TPD run. The non-modified catalyst (Catalyst P in Table 1) shows an NH₃-TPD peak of about 450 °C, showing the non-modified ZSM-5 catalyst had strong acid sites. The

alumina-PA modified catalysts show a broad NH_3 -TPD peak at about 300 °C, showing the modified catalysts possess weak to medium strength acid sites. Examples of NH_3 -TPD profiles for catalysts P, E and L are shown in Figure 1.

[0037] As shown in Table 1 and in Figure 2, the alumina/phosphoric acid modified catalyst showed higher selectivity for p-xylene compared to phosphoric acid modified catalyst based upon the total amount of phosphoric acid used. The catalysts, A-E, prepared after modification by alumina-PA showed increased p-xylene selectivity with increasing PA (Figure 1). As an example, catalyst E in Table 1 prepared by using 0.22g H_3PO_4 /g zeolite (0.07g P/g zeolite) showed 95-97% p-xylene concentration in the total xylenes. Whereas the catalysts modified by using phosphoric acid alone showed only as high as 93% p-xylene when used with 0.28 g H_3PO_4 /g zeolite (0.09 g P/g zeolite).

EXAMPLE 2

[0038] For comparison purposes, Catalyst E (alumina/PA-modified) and L (PA-modified) prepared using substantially the same amounts of phosphoric acid (0.22 g PA and 0.28 g PA, respectively, per gram of zeolite) were used in toluene methylation reactions. The reactions were each carried out in a fixed bed, continuous flow type reactor. In each case, the catalyst was dried by slowly raising the catalyst bed temperature (about 5°C/min) to 200 °C under hydrogen (H_2) flow for at least one hour. A premixed toluene and methanol feed (molar ratio 2/1) was added to the reactor at 200 °C and cofeed of H_2 gas was fed and maintained to provide a H_2 /HC molar ratio of about 7. Water was added to the hydrocarbon (HC) feed and was vaporized prior to introduction to reactor. The catalyst bed temperature was then increased to 500 °C. A liquid hourly space velocity (LHSV) (calculated based on toluene/methanol feed only) about 2 hr^{-1} was maintained. The H_2O /HC molar ratio was about 0.7 and reactor pressure was about 20 psig. The following results were obtained and are presented in Table 2 below and Figure 3.

TABLE 2

Catalyst E (Alumina/PA-Modified)				Catalyst L (PA-Modified)			
Run time, h	CBIT/°C	% Toluene conversion	% PX selectivity	Run time, h	CBIT/°C	% Toluene conversion	% PX selectivity
21.8	498	16.0	92.3	4.5	503	20.1	90.1
45.8	501	18.8	92.3	20.3	506	21.7	88.6
52.3	502	20.0	92.2	43.7	502	23.6	89.4
116.8	502	18.2	92.1	68.2	502	23.6	89.6
143.1	503	18.1	92.0	74.2	500	23.3	89.7
167.1	502	17.9	91.9	139.7	501	22.9	89.7
190.8	502	18.3	91.8	146.2	500	23.2	89.5
220.8	502	17.8	91.7	170.2	501	22.8	89.6
288.3	503	17.9	91.7	194.2	501	20.2	89.7
315.8	500	21.6	91.8	212.2	529	21.0	89.8
340.3	504	17.1	91.8	242.2	525	21.0	88.5
364.3	503	16.7	91.8	307.7	525	20.1	88.0
388.3	501	17.3	91.8	314.2	528	19.6	87.9
453.8	502	16.8	91.7	331.7	526	19.3	87.6
504.3 ^a	501	13.9	91.5	338.2	527	19.6	87.7
526.3	501	13.8	91.5				
549.8	501	14.7	91.5				
622.3	501	13.9	91.5				
628.3	502	14.3	91.5				

^aPower outage occurred between 455 and 500 hours run time resulting in complete shutdown, and the reactor restarted.

[0039] As can be seen from Table 2 and Figure 3, the alumina/PA-modified zeolite had higher p-xylene selectivity than did the PA-only modified zeolite. Both catalysts E and L showed almost the same deactivation profiles. Catalyst E (alumina/PA-modified) showed a small decrease in toluene conversion (0.19% decrease in toluene conversion per 24 hours) at constant catalyst bed inlet temperature around 500 °C up to about 450 hour run period, at which time a power outage occurred resulting in a complete shut-down of the reactor. When the reaction was restarted the catalyst showed a lower but steady toluene conversion during 500-630 run hours. Whereas the catalyst L (PA-modified) also showed small decrease in toluene conversion (0.33% decrease in toluene conversion per 24 hours). In case of catalyst L, catalyst

bed inlet temperature was around 500 °C during the first 194 hours and it was increased to about 525 °C and continued the run for about 340 hours.

[0040] While the invention has been shown in only some of its forms, it should be apparent to those skilled in the art that it is not so limited, but is susceptible to various changes and modifications without departing from the scope of the invention. Accordingly, it is appropriate that the appended claims be construed broadly and in a manner consistent with the scope of the invention.

CLAIMS

We claim:

1. A method of modifying a zeolite catalyst to increase selectivity of the catalyst for para-isomers in aromatic alkylation reactions, the method comprising dissolving alumina in a phosphorus-containing acid solution, and treating the zeolite with the dissolved alumina solution.
2. The method of claim 1, wherein:

the acid includes inorganic acids containing phosphorus.
3. The method of claim 1, wherein:

the acid includes at least one of phosphoric acid (H_3PO_4) and phosphorus acid (H_3PO_3) at a concentration of at least 50 by wt% of aqueous solution.
4. The method of claim 1, wherein:

dissolving the alumina includes dissolving the alumina in an excess of the acid.
5. The method of claim 1, wherein:

the alumina is incorporated into the zeolite in an amount of greater than 0.01 gram of alumina per gram of zeolite.
6. The method of claim 1, wherein:

the phosphorus-containing acid solution is used in an amount of at least 0.1 g of phosphorus-containing acid solution per gram of zeolite.

7. The method of claim 1, wherein:

the zeolite catalyst is a ZSM-5-type zeolite catalyst with a silica to alumina ratio of from 25 to 1000.
8. The method of claim 1, wherein:

the treated ZSM-5 zeolite has total pore volume ranging from 0.10 ml/g to 0.20 ml/g.
9. The method of claim 1, wherein:

the modified catalyst has acid sites showing an ammonia desorption (NH₃-TPD) peak at 250-350 °C.
10. A method of preparing an alkyl aromatic product comprising:

dissolving alumina in an inorganic phosphorus-containing acid;

treating a zeolite catalyst with the alumina-containing acid solution; and

contacting the treated zeolite catalyst with an aromatic hydrocarbon and an alkylating agent under reaction conditions suitable for aromatic alkylation.
11. The method of claim 10, wherein:

the aromatic hydrocarbon is toluene.
12. The method of claim 11, wherein:

the alkylating agent is methanol.

13. The method of claim 10, wherein:
- the zeolite catalyst is a ZSM-5-type zeolite catalyst.
14. The method of claim 10, wherein:
- the alumina is incorporated into the zeolite in an amount at least 0.01 gram alumina per gram of zeolite.
15. The method of claim 10, wherein:
- the modified catalyst has from about 0.03 to about 0.08 gram of phosphorus per gram of zeolite.
16. A method of preparing a xylene product comprising:
- dissolving alumina in a phosphoric acid solution;
- treating a ZSM-5-type zeolite catalyst with the alumina-containing phosphoric acid solution; and
- contacting the treated zeolite catalyst with an aromatic feed of toluene under reaction conditions suitable for at least one of toluene methylation and transalkylation.
17. The method of claim 16, wherein:
- the phosphoric acid solution used in an amount of from about 0.2g to 0.3 g per gram of zeolite.

18. The method of claim 16, wherein:

the alumina is incorporated into the zeolite in an amount of from about 0.01 g/g zeolite to about 0.02 g/g zeolite.
19. The method of claim 16, wherein:

dissolving the alumina includes dissolving the alumina in an excess of phosphoric acid.
20. The method of claim 16, wherein:

the catalyst exhibits greater than 85% para-xylene selectivity of mixed xylenes when used in toluene methylation.
21. The method of claim 16, wherein:

the catalyst exhibits a toluene conversion over a period greater than 500 hours with a loss of conversion of less than 0.5% per day at a substantially constant catalyst bed temperature.
22. The method of claim 16, wherein:

the toluene methylation feed contains water in an amount of at least 0.1 moles per mole of toluene/methanol.
23. The method of claim 16, wherein:

the toluene methylation feed contains hydrogen in an amount of at least 1.0 mole per mole of toluene/methanol.

24. The method of claim 16, wherein:

the reaction conditions include a reactor pressure of at least 10 psig.

FIGURE 1

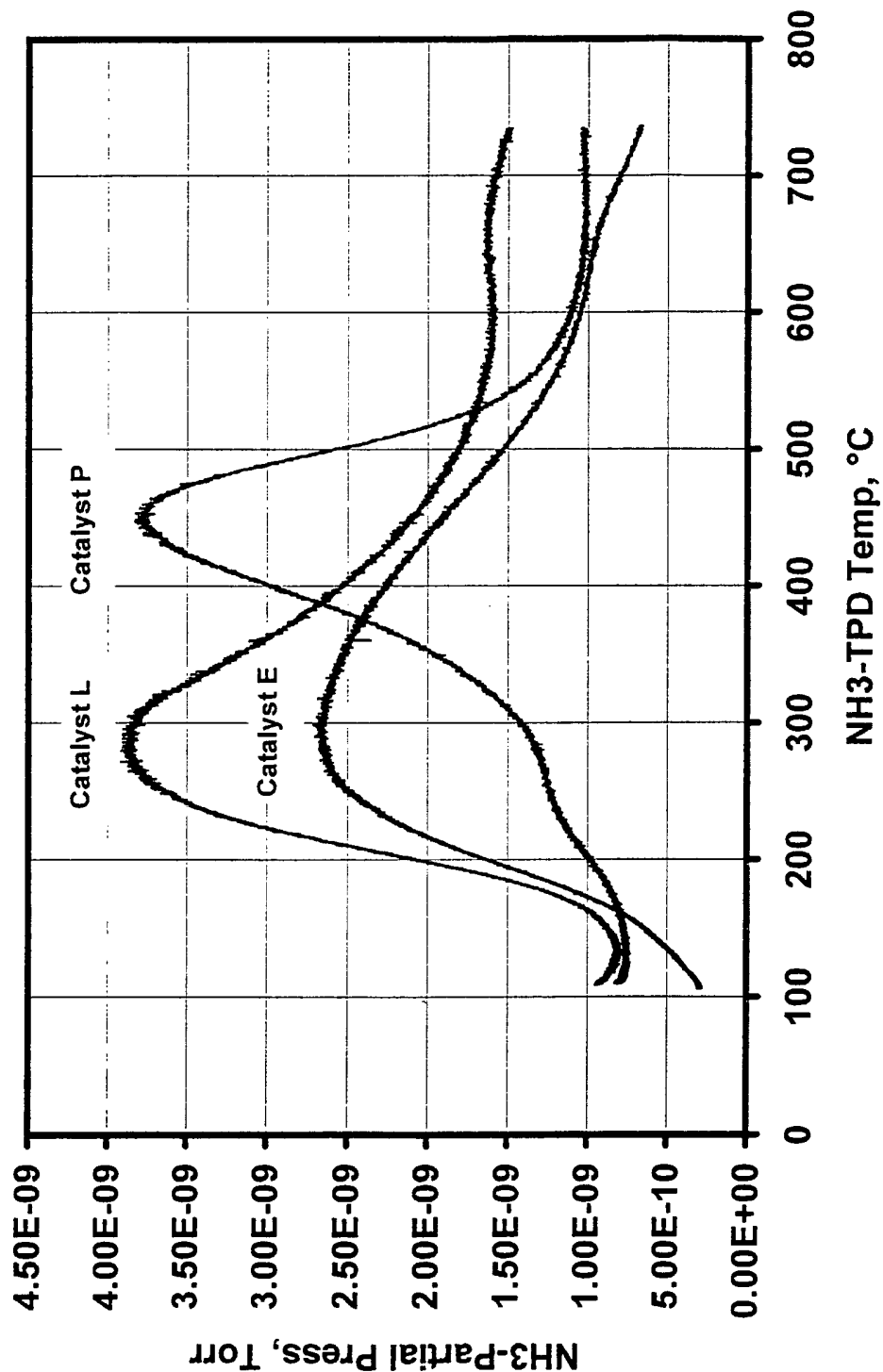


FIGURE 2

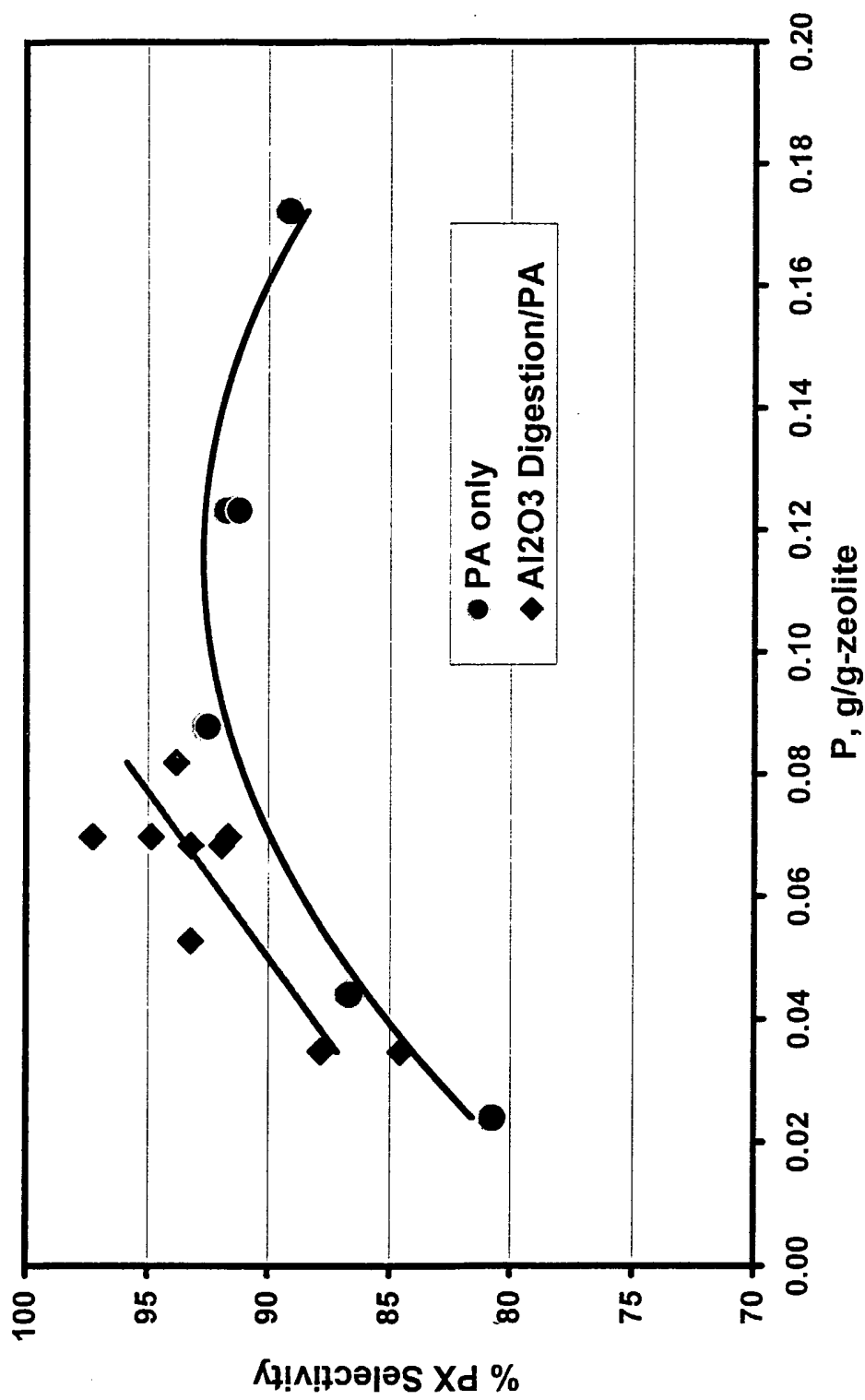
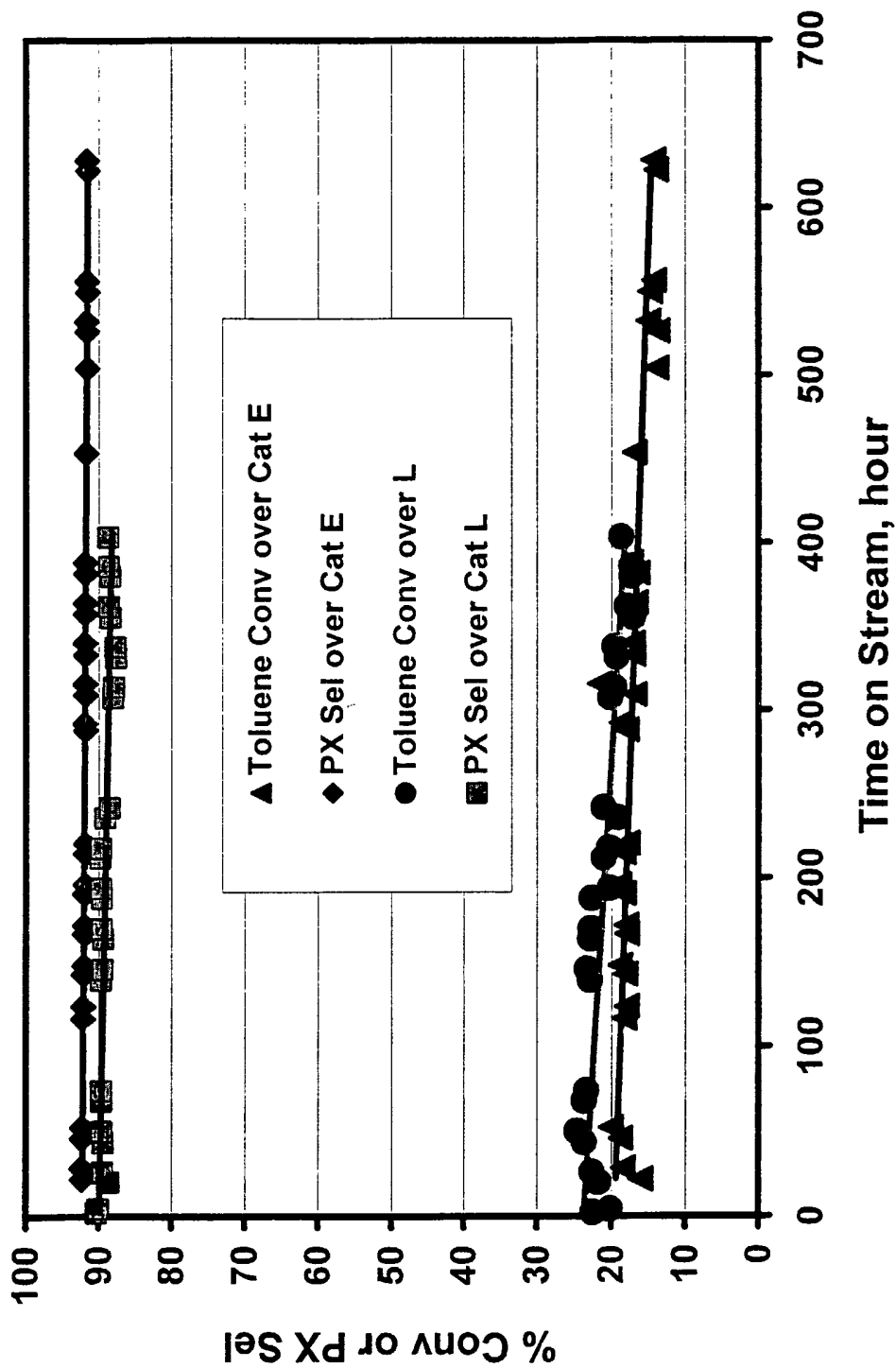


FIGURE 3



INTERNATIONAL SEARCH REPORT

International application No.

PCT/US05/06757

A. CLASSIFICATION OF SUBJECT MATTER
 IPC(7) : B01J 29/04, 29/87
 US CL : 502/60
 According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED
 Minimum documentation searched (classification system followed by classification symbols)
 U.S. : 502/60, 63, 64, 71, 77, 85

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
 Please See Continuation Sheet

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 3,702,886 (ARGAUER et al.) 14 November 1972 (14.11.1972), see entire document.	1-24
A	US 3,965,207 (WEINSTEIN) 22 June 1976 (22.06.1976), see entire document.	1-24
A	US 4,278,827 (CHU et al.) 14 July 1981 (14.07.1981), see entire document.	1-24
A	US 5,210,356 A (SHAMSHOUM et al.) 11 May 1993 (11.05.1993), see entire document.	1-24
A	US 5,227,558 A (SHAMSHOUM et al.) 13 July 1993 (13.07.1993), see entire document.	1-24

Further documents are listed in the continuation of Box C. See patent family annex.

* Special categories of cited documents:	
"A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E" earlier application or patent published on or after the international filing date	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"O" document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search 20 April 2005 (20.04.2005)	Date of mailing of the international search report 28 APR 2005
---	--

Name and mailing address of the ISA/US Mail Stop PCT, Attn: ISA/US Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450 Facsimile No. (703) 305-3230	Authorized officer Jean Vollano <i>J. Roberts for</i> Telephone No. (571) 272-1600
---	--

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US05/06757

Continuation of B. FIELDS SEARCHED Item 3:

WEST:

Search terms: selectivation, activation, alumina, "aluminum oxide", phosphorus, phosphoric