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TITLE: "RADIOPAQUE MEDIUM"

TECHNICAL FIELD

This invention relates to a radiopaque medium of particular use in the radiographic examination of the small bowel or intestine. That intestine is of relatively small diameter but of substantial length and as a result its radiographic examination has hitherto presented difficulties, especially when a "double contrast" examination is required.

BACKGROUND ART

A double contrast examination is one in which the wall of the bowel is lined with radiopaque material while the lumen is filled with radiolucent material, so enabling the radiologist to more clearly visualise the surface conditions of each wall. A double contrast study differs from a single contrast study in which the bowel is filled with radiopaque media and in which substantially only the bowel silhouette is rendered visible.

Hitherto attempts to achieve a double contrast study have involved infusing a relatively small quantity of radiopaque media into the duodenum for onward



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transmission into the small bowel immediately followed by the infusion of a large volume of water or aqueous solution intended to flush the radiopaque material through the bowel while distending the bowel. This technique has severe limitations because the radiopaque material rapidly becomes dispersed in the radiolucent aqueous material and thus in practice the only area available for double contrast examination, and then only briefly, is at the downstream front of the moving aqueous component.

With the foregoing in mind, the present invention was devised to provide a better medium for double contrast radiographic study, by enteroclysis, of the small bowel and which overcomes or at least ameliorates the deficiencies of the prior known media.

DISCLOSURE OF THE INVENTION

According to one aspect the invention consists in a two component medium for use in double contrast radiography comprising:

a first part which contains a radiopaque substance, a second part which is substantially radiolucent, said first and said second part after contact forming a compound or complex which inhibits migration of the radiopaque substance from the first part to the second part.

The compound or complex formed may result in a matrix of limited solubility which binds the radiopaque substance within the first part. The compound or complex



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formed may constitute a zone or barrier at the interface of the first and second part which hinders or prevents diffusion or migration of the radiopaque substance into the radiolucent part.

The compound or complex formed at the interface of the two parts may be insoluble or may be a compound or complex which is only slowly soluble in the first and second part and which thus acts as a diffusion barrier. In other embodiments the compound or complex may form a micellular structure or a liquid crystal structure which acts as a barrier to diffusion of the radiopaque substance into the radiolucent part.

In yet other embodiments the compound or complex may form a zone of higher viscosity than the first part and which inhibits migration of the radiopaque substance into the second part.

In a highly preferred embodiment a water soluble polymer in the radiopaque part interacts or reacts with a water soluble or dispersible polymer in the radiolucent part to form a micellular water dispersible macromolecular complex which prevents migration of the radiopaque substance into the radiolucent part.

In practice, the first part and then the second part are infused via a catheter. The formation of a diffusion barrier at the moving interface of the two components in the bowel restricts the radiopaque substance held within the radiopaque part from dispersing into the second infused part as the second part flushes the first infused



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part through the bowel whilst distending the same. The end result is a distended bowel substantially full of radiolucent media but with the walls coated with a thin layer of radiopaque material trapped adjacent the wall by a barrier comprising for example a water dispersible matrix of limited solubility.

BEST MODE FOR CARRYING OUT THE INVENTION

By way of example, particular embodiments of the invention are described in more detail hereinafter.

A first embodiment according to the invention comprises a two component medium of which the first part has a composition in the weight percentages as follows:-

EXAMPLE 1 - FIRST PART

COMPONENT	% W/W
Water	55.73
Vee Gum (Smectite Clay)	0.25
Sodium Carboxy Methyl Cellulose	0.61
Sodium Citrate	0.19
Sodium Hexa Meta Phosphate	0.09
Potassium Sorbate	0.12
Methyl Para Hydroxy Benzoate	0.04
Propyl Para Hydroxy Benzoate	0.02
Peppermint Oil	0.01
Simethicone Emulsion	0.08
Barium Sulphate Powder	Balance

Vee Gum is obtainable from Vanderbuilt Corp. U.S.A.

The Sodium Carboxy Methyl Cellulose has a viscosity of from 10-20 centipoise in 1% solution in water at 25°C.



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The second part has a composition in the weight percentages shown as follows:-

EXAMPLE 1 - SECOND PART

COMPONENT	% W/W
Water	97.79
PVP K 30 (Polyvinyl Pyrrolidone)	2.15
Methyl Para Hydroxy Benzoate	0.04
Propyl Para Hydroxy Benzoate	0.02

Both of the above-mentioned components may include N sulphuric acid or 30% potassium hydroxide as needed to adjust the pH to within the range of 6 - 6.5 inclusive.

The first part includes barium sulphate powder as a radiopaque substance. The second part is radiolucent. When the first and the second part are brought into contact the sodium carboxy methyl cellulose of the first part reacts with the polyvinyl pyrrolidone of the second part to form a water dispersible micellular complex or matrix of limited solubility at the interface of the first and second parts.

Migration of the barium sulphate from the first part into the second part is thereby prevented or at least inhibited.

The sodium hexa meta phosphate and sodium citrate act as a suspending agent for the barium sulphate and assist to buffer the pH of the system. Potassium sorbate, and methyl and propyl para hydroxy benzoates act as a preservative system. Peppermint oil act as a



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odourant. The use of suspending agents, preservatives, defoamers, flavourings and the like is conventional and other agents may be added or substituted.

The quantity of Carboxy Methyl Cellulose in the composition of Example 1 first part is selected having regard to a viscosity. The viscosity of the first part must be sufficiently low for administration of the first part via a catheter and sufficiently high to maintain suspension of the barium sulphate. The relative quantities of other ingredients is not critical and may be varied to an extent which may readily determined by experiment.

Other water soluble or water dispersible reactants may be substituted for the sodium carboxy methyl cellulose used in the composition of Example I - first part. For example other cellulose polymers such as alkyl hydroxy celluloses, alkyl hydroxy alkyl celluloses and other cellulose esters. Other reactants which may be substituted for the carboxy methyl cellulose are polysaccharides, vinyl ether - maleic anhydride co-polymer, polyvinyl alcohol, and copolymers or mixtures of the foregoing.

Advantageously the first part may contain a mixture of the foregoing reactants for example, sodium carboxy methyl cellulose with a minor proportion of vinyl ether - maleic anhydride copolymer. The finely divided barium sulphate may be incorporated therein. For preference the amount of vinyl ether - maleic anhydride copolymer is



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less than 3% by weight of the first part.

The second part may have a polyvinyl pyrrolidones, or a mixture of polyvinyl pyrrolidones of differing molecular weight, for example PVP K 90 with a lesser quantity of PVP K 30 wherein designations K30 and K90 are believed to be indicative of molecular weights of approximately 40,000 and 360,000 respectively. It will be understood that other constituents may be substituted for the polyvinyl pyrrolidones depending on the selected co-reactant of the first part.

Thus while carboxy methyl cellulose in the first part reacts with polyvinyl pyrrolidone in the second part to give a micellular complex which is sparingly soluble or slowly dispersible, use of vinyl ether - maleic anhydride polymer in the first part with polyvinyl alcohol in the second part forms an insoluble compound or complex. An insoluble compound may also be formed by including for example a carboxy methyl cellulose in the first part and a soluble trivalent foodgrade metallic salt in the second part.

If a drop of water is placed adjacent a drop of a barium sulphate suspension in water on a microscope slide and the drops are brought into contact, the droplets coalesce and the barium sulphate disperses throughout the coalesced liquid within a few minutes.

If a drop taken from a composition according to Example 1 - First Part is placed on the slide adjacent a drop taken from a composition according to Example I -



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Second Part then when the drops are brought into contact the barium sulphate suspension does not disperse into the second part over a period of many hours.

If the first part contains vinyl ether - maleic anhydride and the second contains polyvinyl pyrrolidone, then (if no barium sulphate is present to hide the reaction) an insoluble compound is visible at the boundary of the two drops.

It is of course highly preferred to select foodgrade components and preferably to select a combination of co-reactants which produce a water dispersible product.

When put to use about 200 cc for an average adult of the first part composition may be infused initially to be followed promptly by the infusion of about from 1 to 3 litres of the second part, the quantity depending on the disclosed volume of the particular patients small bowel, the second part being infused at as quick a rate as the patient can tolerate.

For preference, but not essentially, both parts also include various pharmaceutically acceptable additives such as flavouring or odourizing agents, suspending agents, preservatives, defoaming agents and emulsifiers.

Radiopaque substances other than barium sulphate may of course be utilized.

As will be apparent to those skilled in the art from the teaching hereof, the composition of the first and second part may be varied considerably without departing from the inventive concept disclosed herein.



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THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:-

1. A two component medium for use in double contrast radiography comprising:
 - a first part which contains a radiopaque substance,
 - a second part which is substantially radiolucent,
 - said first and said second part after contact forming a compound or complex which inhibits migration of the radiopaque substance from the first part to the second part.
2. A two component medium according to claim 1 wherein the compound or complex is insoluble in water.
3. A two component medium according to claim 1 wherein the compound or complex is slowly soluble in water.
4. A two component medium according to claim 1 wherein the compound or complex has a higher viscosity than the initial viscosity of the first part.
5. A two component medium according to claim 1 wherein the compound or complex formed is macromolecular.
6. A two component medium according to claim 1 wherein the compound or complex is a macromolecular micellular complex.
7. A two component medium according to claim 1 wherein the compound or complex formed is a water dispersible matrix of limited solubility.
8. A two component medium for use in double contrast radiography comprising a first part which contains a



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radiopaque substance and a first reactant, and a second part which is substantially translucent and contains a second reactant said first and said second reactant on contact forming a compound or complex which inhibits migration of the radiopaque substance into the second part.

9. A two component medium according to claim 8 wherein the second reactant is polyvinyl pyrrolidone.

10. A two component medium according to claim 9 wherein the first reactant is selected from the group comprising alkyl hydroxy cellulose, alkyl hydroxy alkyl cellulose, cellulose polymers, cellulose esters, polysacharides and salt thereof.

11. A two component medium according to claim 9 wherein the first reactant is a vinyl ether - maleic anhydride polymer.

12. A two component medum according to claim 9 wherein the first reactant is a polyvinyl alcohol.

13. A two component medium according to claim 8 wherein the second reactant is polyvinyl alcohol.

14. A two component medium according to claim 8 wherein the first reactant is a carboxy methyl cellulose and the second reactant is polyvinyl pyrrolidone.

15. A composition according to claim 1 or claim 8 wherein the first part is an aqueous medium and wherein the radiopaque substance is finely divided barium sulphate.



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16. A composition according to claim 1 or claim 8 wherein the viscosity of the first part and the viscosity of the second part prior to reaction is sufficiently low for administration by enteroclysis.

17. A method for performing double contrast radiography comprising the steps of infusing in sequence into an intestine or bowel a first part which contains a radiographic substance and a second part which is substantially radiolucent, said parts being adapted on contact to form a compound or complex which slows or prevents migration of the radiopaque substance into the radiolucent part.

18. A method according to claim 17 wherein the first and second parts are infused by enteroclysis.



INTERNATIONAL SEARCH REPORT

International Application No PCT/AU 84/00101

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ³ According to International Patent Classification (IPC) or to both National Classification and IPC Int. Cl. ³ A61K 49/04	
II. FIELDS SEARCHED Minimum Documentation Searched ⁴	
Classification System	Classification Symbols
IPC US Cl.	A61K 49/04, 29/02, 27/08 424/4
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁵ .	
AU; IPC as above; Australian Classification 87.160	
III. DOCUMENTS CONSIDERED TO BE RELEVANT ¹⁴	
Category ⁶	Citation of Document, ¹⁴ with indication, where appropriate, of the relevant passages ¹⁷
A	US,A, 3784681 (FISCHLER) 8 January 1974 (08.01.74)
A	US,A, 3832457 (SUGIMOTO et.al.) 27 August 1974 (27.08.74)
A	US,A, 3862301 (CHERNISH et.al.) 21 January 1975 (21.01.75)
A	US,A, 4020152 (HEITZ) 26 April 1977 (26.04.77)
A	US,A, 4038379 (ELINOV et.al.) 26 July 1977 (26.07.77)
A	US,A, 4069306 (ROTHMAN) 17 January 1978 (17.01.78)
A	GB,A, 1392832 (SAKAI CHEMICAL INDUSTRY COMPANY LIMITED) 30 April 1975 (30.04.75)
A	GB,A, 1400985 (PHARMACIA AKTIEBOLAG) 23 July 1975 (23.07.75)
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IV. CERTIFICATION	
Date of the Actual Completion of the International Search ⁸ 10 September 1984 (10.09.84)	Date of Mailing of this International Search Report ⁹ (13.09.84) 13 SEPTEMBER 1984
International Searching Authority ¹ Australian Patent Office	Signature of Authorized Officer ¹⁰

ANNEX TO THE INTERNATIONAL SEARCH REPORT ON
INTERNATIONAL APPLICATION NO. PCT/AU 84/00101

This Annex lists the known "A" publication level patent family members relating to the patent documents cited in the above-mentioned international search report. The Australian Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent Document Cited in Search Report		Patent Family Members			
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US	4069306	CA 1046404 GB 1493974	DE 2510221 JP 50140622	FR 2263754 SE 7403402	
US	3832457	DE 2030690 GB 1315391	DE 2065532	FR 2053000	
US	3784681	AU 36769/71 DE 2162110 IT 1050660	CA 970280 FR 2118172 NL 7117052	CH 564351 GB 1331190	

END OF ANNEX