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Beaucamp et al.

[54] METHOD FOR THE DETERMINATION OF CHOLESTEROL

- [75] Inventors: Klaus Beaucamp; Hans Möllering; Gunter Lang; Wolfgang Gruber; Peter Roeschlau, all of Tutzing, Upper Bavaria, Germany
- [73] Assignee: Boehringer Mannheim GmbH, Mannheim-Waldhof, Germany
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[45] Dec. 9, 1975

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Primary Examiner—A. Louis Monacell Assistant Examiner—C. A. Fan Attorney, Agent, or Firm—Burgess, Dinklage & Sprung

[57] ABSTRACT

Total cholesterol or bound cholesterol in a sample is determined by treating the sample with cholesterol esterase, thereby releasing the bound cholesterol, and then determining the resulting total cholesterol by known methods; specifically preferred are cholesterol esterases from Candida rugosa ATCC 14830, Rhizopus spec. WS 90027 and Aspergillus spec. WS 90030.

18 Claims, No Drawings

METHOD FOR THE DETERMINATION OF CHOLESTEROL

The present invention relates to a method of determining cholesterol, and more specifically to a method ⁵ of determining either total cholesterol or bound cholesterol.

Cholesterol is present in biological matter, such as serum or the like, partially in free form and partially in bound form as ester. For the determination of either 10 bound cholesterol or total cholesterol it is necessary first to release the cholesterol that is present in bound form. This has been done hitherto through saponification under alkaline conditions, using alcoholic potash lye, for example. After the saponification, the released 15 cholesterol can then be determined either chemically or enzymatically by one of the known methods. The chemical determination may be performed, for example, by the Liebermann-Burchard method, and enzymatic determination may be performed by means of 20 cholesterol oxidase, cholesterol dehydrogenase or cholesterol dehydrase. Since the individual cholesterol esters as well as the free cholesterol are known to have different extinction coefficients in the chemical methods of determination, it is necessary to transform the $\ ^{25}$ cholesterol esters to free cholesterol by alkaline hydrolysis.

In any case, however, the alkaline saponification of the bound cholesterol is a troublesome and time-consuming step in the procedure. Furthermore, the relatively agressive reagents used may lead to a decomposition of the cholesterol. In order to prevent such decomposition and thus forestall falsification of the results of the analysis, a hydrolysis must be performed under relatively mild conditions, and this in turn undesirably increases the length of time required for the determination.

The alkaline liberation of the cholesterol is especially disadvantageous when the determination of the cholesterol is afterwards to be performed by the preferred enzymatic methods. Since the enzymes are inactivated, as it is known, in the strongly alkaline medium, the hydrolyzate must be acidified by the addition of acid to pH 5 to 8 before the enzymatic determination can be started. All this results in the fact that the determination of the ⁴⁵ total cholesterol or of the bound cholesterol still takes an undesirably long time and requires too much work.

The present invention provides a process for the determination of total cholesterol or bound cholesterol which substantially obviates the above-mentioned dis- ⁵⁰ advantages.

Essentially, the process of the invention comprises the determination of total cholesterol or of bound cholesterol by releasing the bound cholesterol using cholesterol esterase, followed by determination of the re- ⁵⁵ leased cholesterol by known methods.

It has been found that, using cholesterol esterase, a rapid and quantitative saponification of bound cholesterol can be performed. This process is especially advantageous whenever the subsequent determination of ⁶⁰ the released cholesterol is performed enzymatically, using cholesterol oxidase or cholesterol dehydrase, for example. In this case, the process of the invention makes possible the all-enzymatic determination of cholesterol and therefore a decided improvement of routine medical diagnosis plus an easy adaptation of the process for performance in automatic analysis apparatus. 2

It is already known that in the pancreas and liver a cholesterol esterase is present. It could not be concluded from this knowledge, however, that such an enzyme would be suitable for the rapid, complete saponification of cholesterol esters in the framework of a quantitative analysis process, because the cleavage rates determined were not quantitative, amounting to only 80% maximum (Biochimica et Biophysica Acta 270, (1972), 156–166). Furthermore, bound choles-

- terol is present in biological matter in the form of esters of widely different acid. For an enzymatic process to be useful in the framework of a process of analysis, it is required that all of the esters that may occur be cleaved quantitatively with approximately the same speed and
- ⁵ with the same reliability. On account of the known properties of these enzymes it is surprising that the cholesterol esterases are capable of cleaving quantitatively, within a very short time, all of the cholesterol esters that occur. This is especially surprising also because in the known cholesterol esterases there are considerable differences with regard to their activity against various cholesterol esters.

Cholesterol esterases from microorganisms have proven to be especially suited for the process of the invention. They have proven superior to cholesterol esterases of other origin in regard to speed of cleavage and effectiveness and they are therefore preferred within the scope of the invention.

It has also been found that a number of microorganisms contain particularly active cholesterol esterases and can therefore be used directly within the scope of the invention without separation and purification of the cholesterol esterases. This is advantageous in avoiding the separation of certain esterases from the normally used mixture of a plurality of cholesterol esterases specific for various cholesterol esters. Such separation would greatly complicate the quantitative determination of all of the bound cholesterol.

In addition, the purification of the bound cholesterol esterases in lipoid membranes is difficult, and therefore results in a preparation which is less suitable, on account of its price, for use in routine diagnosis than a microorganism preparation which can be used without any enzyme purification.

Particularly advantageous results have been obtained in the process of the invention by the use of a cholesterol esterase derived from *Candida rugosa* (also referred to as Cylindracea) ATCC 14830 and WS 90031,

respectively, and from Aspergillus spec. WS 90030.
These two microorganisms may be used directly as such or in processed form, e.g., in the form of an acetone dry powder, within the scope of the invention. It is also, of course, possible to use a concentrated cholesterol esterase preparation made from these microorganisms, there being a special advantage in the fact that a certain concentration may in this case be achieved very simply. *Candida rugosa* is a microorganism that is produced on a large technical scale and is available commercially. The customary commercial form is an acetone dry powder stabilized with lactose, which has proven to be outstandingly suited for the invention. Similarly attractive properties have been found in:

	Actinomyces aureoverticillium	WS 90002
•	Actinom yces cyaneofuscatus	WS 90003
	Actinomyces griseomycini	WS 90004
	Actinomyces longisporus-fl.	WS 90005
	Actinomyces malachiticus	WS 90006
	Actinom yces roseolus	WS 90007

-continued		
Actinomyces toxytricini	WS 90008	
Actinomyces variabilis	WS 90009	
Streptomyces spec.	WS 90010	
Streptomyces autotrophicus	WS 90011	
Streptomyces canescens	WS 90012	
Streptomyces chartreusis	WS 90013	
Streptomyces michiganensis	WS 90014	
Streptomyces murinus	WS 90015	
Streptomyces hachijoensis	WS 90016	
Streptomyces caelestes	WS 90017	
Streptomyces tendae	WS 90018	,
Nocardia rubra	WS 90019	1
Candida mycoderma	WS 90020	
Candida albicans	WS 90021	
Candida albicans	WS 90022	
Candida albicans	WS 90023	
Candida spec.	WS 90024	
Cunninghamella elegans	WS 90025	
Mucor mucedo	WS 90026	
Rhizopus spec.	WS 90027	
Penicillium spec.	WS 90028	
Aspergillus spec.	WS 90029	

In addition to the preferred cholesterol esterases of 20 microbiological origin, however, cholesterol esterases of other origin may also be used in many cases.

As previously mentioned, an especially important advantage of the process of the invention consists in the fact that it makes possible an all-enzymatic determina- 25 tion of total cholesterol. It is important in this case that, with the preferred cholesterol esterase preparations made from microorganisms, a rapid and quantitative release of the cholesterol from its esters is possible. Especially with the preferred microorganisms mentioned 30 above, it is possible by the direct addition of same in a very small quantity, with the maintenance of the pH values and temperatures which are desirable in the subsequent enzymatic determination of cholesterol, to achieve within a few minutes a quantitative release of 35 the cholesterol, it having been found that the common carbohydrate-based stabilizing agents which are used for such microorganisms do not interfere with the cholesterol determination performed within the framework of the allenzymatic process.

As mentioned, a separated and concentrated cholesterol esterase, preferably one obtained from microorganism, may also be used for the process of the invention. A suitable concentration may be achieved by setting out from an acetone dry powder of the microor- 45 ganism or other biological material and subjecting it to a dialysis, a treatment with weakly basic anion exchanger and to an ammonium sulfate fractionation. In this manner, it is easy to achieve a 20-fold to 30-fold concentration of the cholesterol esterase. A prepara- 50 tion on a carbohydrate basis, modified with diethylaminoethanol groups, has proven to be an especially suitable weakly basic anion exchanger. In the ammonium sulfate fractionation, the fraction between 1.8 and 2.4 moles of ammonium sulfate is preferably obtained. The 55 ces erythropolis NCIB 9158 is preferably used. enzyme fraction thus obtained is then chromatographed, preferably on the above-named exchanger material.

Particularly good results are obtained within the scope of the invention with microorganisms which have 60 been cultivated in a nutrient medium containing cholesterol ester. In this case, the cholesterol ester or a mixture of cholesterol esters may be added during the cultivation as the sole source of carbon, or may be used together with another carbon source. Especially pre- 65 ferred is the use of microorganisms which are obtained in a multi-stage cultivating process, in which they are cultivated in the first stage on a suitable carbon sup-

plier, such as glycerin, and in the second stage on a cholesterol ester. A suitable cultivation process is described, for example, in German Published Specifications ("Offenlegungsschriften") Nos. 2,224,133 and 5 2,307,518.

The cholesterol esterase from Candida rugosa ATCC 14830 used preferentially in accordance with the invention has very good stability in the weakly acid region between pH 5 and 6.5. The optimum pH for the 10 enzyme is 7.5. One peculiarity of the enzymes is that the catalytic reaction takes place especially well when the salt content of the reaction medium is relatively high. Preferably, therefore, the process is performed in an 0.2 to 0.8 molar buffer solution. The pH may range 15 between 4.5 and 7.5, and will preferably range, as stated above, between pH 5 and 6.5. The effectiveness of the cholesterol esterase is preferably increased by the addition of surface active substances. Especially preferred is the addition of hydroxypolyethoxydodecane.

As previously mentioned, it is especially preferred that the process of the invention be performed allenzymatically, i.e., the cholesterol determination that follows is also performed enzymatically, preferably with the use of cholesterol oxidase. However, cholesterol dehydrase or dehydrogenase may also be used.

Determination with cholesterol oxidase is described, for example, in German Offenlegungsschrift No. 2,224,132. The process therein described may advantageously be combined with the process of the invention. In this case, it is possible in principle to measure the oxygen consumption, the H₂O₂ formation or the formation of cholestenone. The determination of the oxygen consumption may be performed, for example, by gas chromatography or polarometry, or by the polarization method. These methods of determination are in the prior art. The hydrogen peroxide that forms may be determined titrimetrically, potentiometrically, and colori-40 metrically, as well as enzymatically. Enzymatic determination is preferred, with the use of catalase or peroxidase, especially determination by catalase in the presence of beta diketones such as acetylacetone, low alcohols and buffer containing ammonium ions, or determination by peroxidase in the presence of a chromogen such as 2,2'-aminobenzothiazolinesulfonic acid. Cholestenone is determined by means of keto reagents such as 2,4-dinitrophenylhydrazine, or by photometry at 240 nm.

If the all-enzymatic determination of the total or bound cholesterol is performed with cholesterol oxidase, a cholesterol oxidase obtained from Nocardia erythropolis ATCC 17895, Nocardia erythropolis ATCC 4277, Nocardia formica ATCC 14811 or Proactinomy-

Additional subject matter of the invention is a reagent for the determination of cholesterol which consists of cholesterol esterase and a reagent for the determination of free cholesterol. Preferably, a reagent of this sort consists of a cholesterol esterase of microbiological origin, cholesterol, oxidase, a system for the determination of hydrogen peroxide, or a system for the determination of cholestenone. Quite especially preferred in this case is a reagent in which the cholesterol esterase is one of the microorganisms mentioned further above, especially in the form of an acetone dry powder or a protein fraction obtained therefrom having cholesterol esterase activity.

In a special and preferred embodiment, such a reagent consists of cholesterol oxidase, a cholesterol esterase preparation made from one of the above-mentioned microorganisms, catalase, acetyl acetone, methanol and aluminum ion-containing buffer, individually 5 or mixed. In still another preferred embodiment, the reagent consists of cholesterol oxidase, a preparation made of the above-mentioned microorganisms having cholesterol esterase activity, peroxidase, chromogen and buffer, individually or mixed. 2,2'-aminobenzo- 10 thiazolinesulfonic acid is preferred as the chromogen.

In still another preferred embodiment, the reagent of the invention consists of cholesterol oxidase, a cholesterol esterase preparation made from one of the named microorganisms, and a hydrazine derivative reacting 15 with keto groups with the formation of hydrazone, and in some cases a buffer. 2,4-dinitrophenylhydrazine is preferred as the hydrazine derivative.

The above-mentioned preferred reagent combinations may contain, in addition to the specified essential 20 components, commonly used solvents, stabilizers and-/or suface active substances. All these additive substances are known to persons skilled in the art and commonly used in detection systems for hydrogen peroxide and cholestenone.

Preferably, the above-mentioned reagent combinations will contain the essential components in the following rations:

- 1. 13 to 150 U of cholesterol oxidase, 0.05 to 0.5 mg of microorganism cholesterol esterase, 2×10^4 to 5 30 $\times\,10^{\rm 5}$ units of catalase, 0.05 to 0.2 ml of acetyl acetone and 2 to 10 ml of methanol in 100 ml of a pH 5 to 7 buffer containing ammonium ions, plus, if desired, 0.02 to 0.3 ml of a surface active agent, preferably hydroxypolyethoxydodecane.
- 2. 3 to 40 U of cholesterol oxidase, 0.05 to 0.5 mg of microorganism cholesterol esterase, and $2 \times$ ² to 1×10^4 U of peroxidase, 50 to 200 mg. of 2,2'amino-benzothiasolinesulfonic acid, and, if desired, $0.05\ to\ 0.5\ ml$ of surface active agent, preferably $\ 40$ hydroxypolyethoxydodecane, in 100 ml of pH 6 to 8 buffer.
- 3. 0.1 to 1 U of cholesterol oxidase, 0.05 to 0.5 mg of microorganism cholesterol esterase, 1 to 5 ml of a 1 mM solution of 2,4-dinitrophenylhydrazine, and, 45 if desired, 0.005 to 0.1 ml of surface active agent in 10 ml of pH 6 to 8 buffer.
- 4. 2 to 100 U of cholesterol oxidase, 0.05 to 0.5 mg of microorganism cholesterol esterase, and, if debly hydroxypolethoxydodecane), in 50 ml of pH 5 to 9 buffer, preferably 0.5 m of sodium phosphate pH 7.5 buffer.

With the process and reagent of the invention, an excholesterol determination with cholesterol oxidase in accordance with the invention, a quantitative cleavage of bound cholesterol is accomplished within one to three minutes with the addition of Candida rugosa powder in a quantity between 0.1 to 0.3 mg. The following examples are illustrative.

EXAMPLE 1

Using the method described in Example 1 of German⁶⁵ Offenlegungsschrift No. 2,224,132, the content of free cholesterol in serum was found to be 63 mg% (63 mg in 100 ml). For the determination of bound cholesterol, a

specimen of the same serum was treated for 30 minutes with alcoholic potash lye at 70°C. After neutralization and measurement of the cholesterol present, a total content of 181 mg% of cholesterol was found. From this appears that 118 mg of cholesterol were present in

bound form for every 100 ml.

The process was repeated with untreated serum, but at the beginning of the determination 0.3 mg% (with reference to the protein) of an acetone dry powder of Candida rugosa ATCC 14830 in commercial form was

added. After 3 minutes, polarographic determination showed a content of 183 mg% total cholesterol.

EXAMPLE 2

To concentrate the cholesterol esterase activity, commercial acetone dry powder of Candida rugosa ATCC 14830 was dissolved in potassium phosphate buffer pH 6.0 and dialyzed against the same buffer. After removal of the lactose contained in the solution as stabilizer, the specific cholesterol esterase activity was 0.3 U per mg of protein in the dialyzed solution.

The solution thus obtained was stirred together with an ion exchanger on a dextran basis modified with diethylaminoethanol groups; the exchanger was sepa-25 rated and eluted with 0.2 M of pH 6.0 phosphate buffer. A specific cholesterol esterase activity of 1.2 U/mg was found in the eluate.

The solution thus obtained was subjected to an ammonium sulfate fractionation. The protein fraction that precipitated between 1.8 and 2.4 M of ammonium sulfate was separated, and had a specific cholesterol esterase activity of 2.5 U/mg.

The product obtained was again dissolved in pH 6.0 35 phosphate buffer, dialyzed against the same buffer until salt-free, and then chromatographed on a column filled with the same anion exchanger as above. Elution was again performed with 0.2 M of pH 6.0 phosphate buffer. A specific activity of 7 U per mg of protein was found in the fraction having cholesterol esterase activity.

The concentrated cholesterol esterase preparation thus obtained was used in the cholesterol determination as described in Example 1, except that the amount used was only 0.001 mg with reference to protein. The results were the same as in Example 1.

The cholesterol esterase from Candida rugosa can be further purified by the conventional methods of enzyme refinement. Instead of the concentration procesired, 0.1 to 2.0 ml of surface active agent (prefera- 50 dures cited above, other conventional biochemical refinement procedures may be used, such as precipitation or fractionation with polyethyleneimine, organic solvents or salts, by chromatography through molecular sieve materials or weak anion exchanger with functremely rapid and complete saponification of bound 55 tional groups other than diethylaminoethanol groups, by protamine sulfate precipitation and the like.

EXAMPLE 3

To 0.5 ml of serum in the one case and cholesterol ATCC 14830 or Aspergillus sp. WS 90030 acetone dry 60 standard in the other, 1.0 ml of 0.5 M potassium phosphate pH 7.5 buffer containing 0.4% hydroxypolethoxydodecane, and 2.5 U of cholesterol esterase from Example 2 were added. This reaction mixture was incubated for 40 minutes at 37°C. Then 0.25 ml of this solution was added to 3 ml of cholesterol reagent containing two parts acetic acid, three parts acetic acid anhydride and one part sulfuric acid (Liebermann-Burchardt reagent).

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By using a standard as a reference magnitude, 170 mg% total cholesterol was found in a typical specimen. Comparative determination after saponification of the cholesterol ester with alcoholic potash lye gave 165 mg%.

EXAMPLE 4

10 ml of 0.5 M potassium phosphate buffer containing 0.4% hydroxypolyethoxydodecane, and 0.2 U of the cholesterol esterase of Example 2, were added to 100.02 ml of serum. The reaction solution was incubated for 60 minutes 37°C. Then the extinction (E_1) at 240 nm was read in a suitable spectral photometer and the reaction was started with 0.1 U of sterol dehydrase obtained from Brevibacterium sterolicum. After fifteen 15 minutes, the extinction (E2) was again read. The concentration of the Δ^4 cholestenone and hence of the cholesterol was found from the difference between the first and second reading on the basis of the molar extinction coefficient for Δ^4 cholestenone at 240 nm. Measure- ²⁰ ment of a typical specimen gave 183 mg% total cholesterol.

Comparative determination with a cholesterol oxidase from Nocardia erythropolis instead of sterol dehydrase gave 181 mg% cholesterol.

EXAMPLE 5

10 g of diammonium hydrogen phosphate was dissolved in 100 ml of water and adjusted to pH 7.0 with 85% phosphoric acid. Then 10⁵ units of catalase were ³⁰ added. The solution thus obtained was added to a mixture of 0.2 ml of acetyl acetone, 10 ml of methanol and 0.1 g of hydroxypolyethoxydodecane to produce a volume of 100 ml. To this solution, 2.5 units of cholesterol esterase from Rhizopus spec. (WS 90027) were added. 35 5.0 ml of the solution thus obtained was mixed with 0.02 ml of serum in the one case and 0.02 ml of a cholesterol standard solution containing 200 mg% cholesterol in the other. To aliquots of the serum-containing 40 specimen and of the cholesterol standard-containing specimen 0.1 unit of cholesterol oxidase was added and the mixtures were incubated for 60 minutes at 37°C. Then the dye that was formed was measured photometrically at 405 nm on the basis of the specimen zero value.

Using a standard as a reference magnitude, the cholesterol content of the serum-containing specimen amounted to 154 mg% total cholesterol. The control determination performed with cholesterol esterase from Candida rugosa ATCC 14830 instead of choles- 50 cholesterol' esterase obtained from a microorganism terol esterase from Rhizopus spec. (WS 90027) gave the same value.

It will be understood that the specification and examples are illustrative but not limitative of the present invention and that other embodiments within the spirit 55 and scope of the invention will suggest themselves to those skilled in the art.

What is claimed is:

1. Method of determining total cholesterol or bound cholesterol in a sample, which method comprises treating said sample with cholesterol esterase obtained from a micro-organism, thereby releasing the bound cholesterol, and then determining the resulting cholesterol content of said sample using a standard determinatiion.

2. Method as claimed in claim 1 wherein said sample 65 contains only bound cholesterol and the said determination determines the amount of said bound cholesterol.

3. Method as claimed in claim 1 wherein said sample contains free cholesterol and bound cholesterol and the said determination determines total cholesterol.

4. Method as claimed in claim 1 wherein said microorganism is Candida rugosa ATCC 14830.

5. Method as claimed in claim 1 wherein said microorganism is Rhizopus spec. WS 90027.

6. Method as claimed in claim 1 wherein said microorganism is Aspergillus spec. WS 90030.

7. Method as claimed in claim 1 wherein said microorganism is

15	Actinomyces aureoverticillium Actinomyces cyaneofuscatus Actinomyces griseomycini Actinomyces longispurus-fl. Actinomyces malachiticus Actinomyces roseolus	WS 90002 WS 90003 WS 90004 WS 90005 WS 90006 WS 90007 WS 90008
20	Streptomyces canescens Streptomyces chartreusis Streptomyces michiganensis	WS 90009 WS 90010 WS 90011 WS 90012 WS 90013 WS 90014 WS 90015
25	Streptomyces murinus Streptomyces hachijoensis Streptomyces calestis Streptomyces tendae Nocardia rubra Candida mycoderma Candida albicans	WS 90016 WS 90017 WS 90018 WS 90019 WS 90020 WS 90021
30	Candida albicans Candida albicans Candida spec. Cunninghamella elegans Mucor mucedo Penicillium spec. Aspergillus spec.	WS 90022 WS 90023 WS 90024 WS 90025 WS 90026 WS 90028 or WS 90029.

8. Method as claimed in claim 1 wherein said determination is an enzymatic determination.

9. Method as claimed in claim 8 wherein the enzyme used for said enzymatic determination is cholesterol oxidase.

10. Method as claimed in claim 9 wherein said cholesterol oxidase is from Nocardia erythropolis ATCC 17895, Nocardia erythropolis ATCC 4277, Nocardia formica 14811 or Proactinomyces erythropolis NCIB 9158.

11. Method as claimed in claim 1 wherein said micro-45 organism has been cultivated on a cholesterol ester containing nutrient medium.

12. Reagent composition for the determination of cholesterol in a sample, with composition comprises and a system for the determination of free cholesterol.

13. Reagent composition as claimed in claim 12 wherein said composition comprises

a cholesterol esterase from a microorganism,

cholesterol oxidase,

a system for the determination of hydrogen peroxide.

14. Reagent composition as claimed in claim 13 wherein said microorganism is Candida rugoas ATCC 14830, Rhizopus spec. WS 90027 or Aspergillus spec. 60 WS 90030.

15. Reagent composition as claimed in claim 12 comprising

cholesterol esterase,

cholesterol oxidase,

catalase.

- acetyl acetone,
- methanol.

and a buffer containing ammonium ions.

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16. Reagent composition as claimed in claim 14 containing

peroxidase,

a chromogen,

a buffer,

a system for the determination of H₂Oh₂.

17. Reagent composition as claimed in claim 12 wherein said composition comprises

a cholesterol esterase from a microorganism, cholesterol oxidase,

a system for the determination of cholestenone. **18.** Reagent composition as claimed in claim **17** wherein said system for the determination of cholestenone is a hydrazine derivative reacting with keto groups, to result in the formation of hydrazone. * * * *

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