A new process for extracting and purifying glucosinolates from plant material, preferably broccoli sprouts or seeds is described. An alcoholic extract is adsorbed onto a basic resin and eluted with ammonia. Optionally, the alcoholic extract is passed through an ion-exchange column containing acidic resin prior to the adsorption/elution step.
PROCESS FOR EXTRACTION OF GLUCOSINOLATES
FROM BROCCOLI SEEDS

BRIEF DESCRIPTION OF THE INVENTION

This invention relates to a process for extracting glucosinolates from broccoli seeds, sprouts, or florets which comprises a step of adsorbing a broccoli seed extract on a basic resin, followed by elution of the adsorbed glucosinolates and collecting the glucosinolate-rich eluate.

BACKGROUND OF THE INVENTION

Broccoli seeds have a high amount of glucosinolates, including glucoraphanin, glucoiberin, glucoerucin. While the glucosinolates are not biologically active, cleavage by the enzyme myrosinase (present in many plant cells and in the gut microflora) results in the formation of active isothiocyanates. These isothiocyanates, including sulforaphane, have been shown to have numerous health-promoting properties, and in some experiments have even been shown to exert various anti-cancer effects.

Previously, various authors have developed extraction/purification schemes to obtain glucoraphanin from broccoli seeds. For example, West et al. 2002 J. ChromatogA 966:227-232 describes use of ion-pair and hydrophilic interaction chromatographies for purifying various glucosinolates. See also Toribio et al. 2007 J. ChromatogA 1170:44-51 which describes purification of sinablin and glucoraphanin using strong ion-exchange displacement centrifugal partition chromatography. However, both these techniques are directed to purification of compounds, and not merely extracting a glucoraphanin-containing extract from broccoli seeds which is economical and uses food-grade reagents.

It would be desirable to have a simple, robust glucosinate extraction method which is suitable for production of food-grade or nutraceutical-grade glucosinolates, especially glucoraphanin.
DETAILED DESCRIPTION ON THE INVENTION

A new process for the production of a glucosinolate-containing extract has been developed in accordance with this invention, which comprises the steps of:

a) extracting glucosinolate-containing plant parts or seeds, or an extract of glucosinolate-containing plant parts or seeds with an extraction medium that comprises a lower alcohol or ketone or an aqueous mixture of a lower alcohol or ketone to obtain a alcoholic or ketonic extract;

b) optionally completely or partially evaporating the extraction medium of step a);

c) optionally contacting the extract of step a) or step b) with a cation exchange resin;

d) adsorbing the extract from step a), step b), or step c) onto a basic resin; and

e) optionally eluting the resulting glucosinolate-containing extract.

Suitable solvents for step b) through e) include water, C1-4 alcohols, C3-4 ketones, and mixtures thereof.

The cation exchange resin is preferably in its acidic form, and more preferably a strong acidic ion exchanger is used.

If desired, an additional step may be performed on the glucosinolate-containing extract obtained from step e). In this embodiment of the invention, the volatiles of the glucosinolate-containing extract are evaporated. The result of this step is a solid extract containing glucosinolates, which includes glucoraphanin, glucerucin, and glucoiberin. The solid extract made by this process also forms another aspect of this invention.

The starting materials for this invention may be any glucosinolate-containing plant material, or any glucosinolate-containing extract. Preferably, the plant material is from the Brassicaceae family, such as broccoli, mustard, rapeseed, cauliflower, kohlrabi, cabbage, bok choy, turnip, radish, wasabi, horse radish and brussel sprouts.

The plant parts may be the sprouts or seeds, as it is known that sprouts and seeds often contain higher amounts of glucosinolates than mature plant leaves, but florets or heads may also be used. If desired, the plant parts may be first subjected to a pre-treatment step of
washing, or de-fatting (for seeds). In preferred embodiments, the plant is a broccoli plant, and seeds are extracted. In this case, the original starting material for the process may be either a broccoli seed extract or the broccoli seeds themselves.

If one starts with plant parts, optionally washed and/or defatted and/or otherwise treated, then they are subjected to an extraction step. In this extraction, it is preferable to use an aqueous medium, or a lower alcohol or ketone wherein a lower alcohol or ketone is a C1 to C4 alcohol or ketone or mixtures thereof. This may be performed in the presence of charcoal or other similar material such as celite. See, for example Toribio et al., supra. The extract obtained can then be used in the next step. If desired, the extract can be subjected to further purification steps such as ultrafiltration. Also the extract may be optionally concentrated by a complete or partial evaporation of the solvent.

In the next step, the extract of plant parts (either as obtained above, or, optionally, a commercially available extract) is then extracted with an extraction medium which comprises a lower alcohol or lower ketone. The term "lower alcohol" or "lower ketone" means that the alcohol is a C1-C4 alcohol or the ketone is a C3-C4 ketone, or mixtures thereof; and is preferably an alcohol or ketone which is approved for use in food manufacturing, such as ethanol or acetone, although using technical ethanol may also be used. The alcohol or ketone are preferably ethanol or acetone, and may be in an aqueous solution such as at least about 40% alcohol or ketone; in a preferred embodiment, the lower alcohol or lower ketone is at least about 70% in an aqueous solution, and in a more preferred embodiment it is from about 70% to about 95%. In this step, temperature is not particularly critical. The extract may be filtered or decanted to separate the solubles from the insolubles.

The extract may be evaporated to remove the volatiles, followed by dissolving in an appropriate solvent such as water, lower alcohols, ketones, or a mixture thereof for the next step.

Optionally, the alcoholic or ketonic extract in the appropriate solvent or solvent mixture is subjected to a cationic ion-exchange column, preferably a strong acidic resin, in its acidic form such as DOWEX® 50W or AMBERLYST® 15 (both available from Sigma Aldrich). It is also preferred, in keeping with the intended use of the final product as a nutraceutical or
The extract which is obtained from either the alcoholic extraction step, or preferably the ion-exchange step, is then adsorbed onto a basic resin. The basic resin may be either a strongly or weakly basic resin, preferably a weakly basic resin. Examples of suitable resins include AMBERLITE® IRA-67, and LEWATIT® VPOC 1065, (both available from Sigma Aldrich). In keeping with the goal of manufacturing nutraceutical/food grade material, it is preferred that the resin be suitable from a regulatory view for this purpose. For this step, temperature is not particularly critical; ambient temperature is preferred.

The ion-exchange column so-prepared is then eluted with a base such as ammonia, diluted potassium hydroxide, sodium hydroxide, sodium carbonate, sodium hydrogen carbonate or the like, in water, lower alcohol (C1-C4) or acetone or in mixtures thereof. The preferred base is ammonia in water, lower alcohol (C1-C4) or mixtures of the solvents.

The resulting eluate (final extract) contains glucosinolates in a more purified form than the original starting material. While actual amounts of major products may vary from lot-to-lot, depending on the content of the starting plant material; a typical final extract will contain the following major glucosinolates: glucoraphanin, glucoperucin, and glucoiberin. This product forms yet another embodiment of this invention.

If desired, the final extract may be evaporated, freeze-dried or spray-dried using conventional means so that it is a solid extract. These processed solid extracts also form an aspect of this invention.

In one preferred embodiment of the invention, the extract after ion exchange treatment and elution with ammonia, contains the ammonium salt of glucoraphanin as a major product. In an optional, but preferred step, the ammonium salt is changed into an ammonium-free extract suitable for further processing. In this optional step, the extract which is acidic, is made more basic. This can be done by adding any conventional source of base, such as alkali- or earth alkali hydroxides, such as magnesium hydroxide, calcium hydroxide, sodium hydroxide,
potassium hydroxide. Preferably the base is suitable for use in food, such as sodium hydroxide.

Enough base is added so that the pH rises to above 7.0, more preferably to about pH 7 to pH 12, and even more preferably to pH 9 to 11.

The addition of a base such as sodium hydroxide to form a basic environment, such as from pH 9 to 11, allows the exchange of ammonium ions to sodium ions. The volatiles can then be separated from the resulting sodium glucosinolate by conventional means, such as by use of reduced pressure (i.e. removed using a partial vacuum), resulting an-ammonium free extract.

The following non-limiting Examples better illustrate the invention

**Example 1:**

An extraction was made following the procedure generally described in A.Toribio et al.; *J.Chromatogr.A* 1170(2007) 44-51, which is hereby incorporated by reference. 3 kg of broccoli seeds were stirred for 2 h in 20 liters of water at reflux temperature. The resulting warm solution was filtered. The filtrate was then agitated for 2 h with 150 g of charcoal. The suspension was filtered and concentrated at reduced pressure at 60 °C to 310 g residue with a glucoraphanin content of 12%.

**Example 2:**

4.Og of the residue of Example 1 was suspended in 40 ml of an ethanohwater (82%:18%) mixture and heated to reflux for approximately 30 min. The resulting suspension was filtered and the mother liquid was evaporated under reduced pressure (30 mbar) at 60 °C. An extract was obtained (2.2 g) with a purity of glucoraphanin of 16%.
**Example 3:**

2 g ion exchanger AMBERLITE® IRA-67 was stirred with 2.0 g extract of Example 1 in 10 ml water for 30 min. The liquid was removed by filtration and the loaded ion exchanger washed with water (5 ml). The loaded ion exchanger was stirred at ambient with 10 ml 5% aqueous ammonium hydroxide solution in methanol for 30 min. After filtration, the filtrate was evaporated under reduced pressure (30 mbar, 70 °C). The result was 97.3 mg of extract containing glucoraphanin with a purity of 54%.

**Example 4:**

Washing the loaded ion exchanger of Example 3 with 10 ml 5% aqueous ammonium hydroxide solution in water yielded, after evaporation 217 mg glucoraphanin with a purity of 19%.

**Example 5:**

2 g acidic ion exchanger AMBERLYST® 15 was stirred with 1 g extract of Example 2 in 10 ml water for 30 min at ambient temperature. The liquid was removed by filtration and the ion exchanger washed with water (5 ml). The eluate, containing the free glucoraphanin acid, was stirred with the weakly basic ion exchanger AMBERLYST® IRA-67 (2 g) for 30 min. at ambient temperature. The liquid was removed by filtration and the loaded ion exchanger washed with water (10 ml). The loaded ion exchanger was stirred with 10 ml of a solution of 5% aqueous ammonia in methanol for 30 min at ambient. After filtration, the eluate was evaporated under reduced pressure (30 mbar, 70 °C). Yield: 95.2 mg glucoraphanin with a purity of 69% (determined by HPLC).

**Example 6:**

Washing the loaded basic ion exchanger from Example 5 with 5% aqueous ammonium hydroxide solution in water yielded after evaporation, 385.3 mg glucoraphanin (purity 25%).
Example 7

Ig of the extract of Example 5 was stirred in 10 ml water. An aqueous solution of sodium hydroxide (1 N) was added until a pH of 10.5 was reached. The mixture was stirred for 30 min at room temperature and the volatiles were removed under reduced pressure (30 mbar, 70 °C). Yield: 980 mg glucoraphanin with a purity of 64% (determined by HPLC). 42% of the ammonium ions were exchanged to sodium ions.

***
What is Claimed is:

1. A process for obtaining a glucosinolate-containing extract from glucosinolate-containing plant parts or seeds or from an extract of glucosinolate-containing plant parts or seeds comprising the steps of:
   a) extracting the glucosinolate-containing plant parts or seeds, or the extract of glucosinolate-containing plant parts or seeds with an extraction medium that comprises a lower alcohol or ketone or a mixture thereof with water, to obtain a alcoholic or ketonic glucosinolate-containing extract; and
   b) optionally contacting the glucosinolate-containing extract of step a) with a cation ion-exchange resin; and
   c) adsorbing the glucosinolate-containing extract from either step a) or step b) onto a basic resin; and
   d) optionally eluting the resulting glucosinolate-containing extract.

2. A process according to Claim 1, further comprising evaporating the volatiles from resulting glucosinolate-containing extract.

3. A process according to Claim 1 or 2, wherein the cation exchange resin of step b) is a strongly acidic resin.

4. A process according to any of Claims 1-3 wherein the alcohol of step a) is a C1-C4 alcohol.

5. A process according to Claim 4 wherein the alcohol is ethanol.

6. A process according to any of Claims 1-3 wherein the ketone of step a) is acetone.

7. A process according to Claim 4-6 wherein the C1-C4 alcohol or acetone is present in an aqueous solution at a concentration of at least about 40%.

8. A process according to Claim 5 wherein ethanol is present in an aqueous solution at about 70% to about 95%.
9. The process according to any of Claims 1-8 where the starting material is an extract of glucosinolate-containing plant parts or seeds.

10. The process of any of Claims 1-9 wherein the plant parts or seeds are selected from the group consisting of broccoli, mustard, rapeseed, cauliflower, kohlrabi, cabbage, bok choy, turnip, radish, wasabi, horse radish and brussel sprouts.

11. The process according to Claim 10 wherein the plant is broccoli.

12. The process according to Claim 10 wherein the plant part or seeds is broccoli seeds.

13. The process of any of Claims 1-12, further including a step of evaporation, freeze-drying, or spray-drying.

14. A process according to any of claims 1-13, further comprising adding sufficient base to the eluate of step d), to increase the pH to at least 7.0, and separating the resulting glucosinolate-containing salts.

15. A process according to Claim 14 wherein the glucosinolate-containing salts have alkali- or earth alkali cations.

16. A process according to any of Claims 14-15 wherein the separating comprises removing volatiles under reduced pressure.

17. The product produced by the process of any of Claims 1-16.

***
INTERNATIONAL SEARCH REPORT

A. CLASSIFICATION OF SUBJECT MATTER

INV. A61K36/31 B01D15/36 C07H15/14 C07H1/08 A61K31/7028 A61P35/00 A23L1/29

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

A61K BOID

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, EMBASE, BIOSIS, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

<table>
<thead>
<tr>
<th>Category*</th>
<th>Citation of document, with indication, where appropriate, of the relevant passages</th>
<th>Relevant to claim No.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Y</td>
<td>page, 45, left-hand column, paragraph 3 page 45, right-hand column, last paragraph - page 46, right-hand column, paragraph 1</td>
<td>1-17</td>
</tr>
</tbody>
</table>

Further documents are listed in the continuation of Box C. See patent family annex.

* Special categories of cited documents:

'A' document defining the general state of the art which is not considered to be of particular relevance

'E' earlier document published on or after the International filing date

'L' document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

'O' document referring to an oral disclosure, use, exhibition or other means

'P' document published prior to the international filing date but later than the priority date claimed

Date of the actual completion of the international search

27 October 2009

Date of mailing of the international search report

09/11/2009

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2
NL- 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Fax: (+31-70) 340-2016

Authorized officer

Mateo Rosel l, A

Form PCT/ISA/21 0 (second sheet) (April 2005)
DOCUMENTS CONSIDERED TO BE RELEVANT

<table>
<thead>
<tr>
<th>Category</th>
<th>Citation of document, with indication, where appropriate, of the relevant passages</th>
<th>Relevant to claim No.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Y</td>
<td>TRENERRY V C ET AL: &quot;The determination of glucoraphanin in broccoli seeds and...</td>
<td>1-17</td>
</tr>
<tr>
<td></td>
<td>ISSN: 0308-8146 [retrieved on 2006-01-01] abstract page 180, left-hand column, last paragraph - page 181, right-hand column, paragraph 3</td>
<td></td>
</tr>
<tr>
<td></td>
<td>page 181, right-hand column, paragraph 1-3 page 182, right-hand column, last paragraph - page 184, right-hand column, paragraph 3</td>
<td></td>
</tr>
<tr>
<td>Y</td>
<td>CHEN S ET AL: &quot;In vitro synthesis and purification of radioactively p-hydroxybenzylglucosinolate in Sinapis alba L.&quot;</td>
<td>1-17</td>
</tr>
<tr>
<td></td>
<td>ISSN: 0958-0344 page 175, left-hand column, paragraph 3 page 176, left-hand column, paragraph 1 - right-hand column, last paragraph</td>
<td></td>
</tr>
<tr>
<td>Y</td>
<td>ROCHFORT S ET AL: &quot;The isolation and purification of glucoraphanin from broccoli seeds by solid-phase extraction and preparative high performance liquid chromatography&quot;</td>
<td>1-17</td>
</tr>
<tr>
<td></td>
<td>ISSN: 0021-9673 [retrieved on 2006-07-07] abstract page 206; left-hand column, last paragraph - right-hand column, paragraph 1</td>
<td></td>
</tr>
<tr>
<td></td>
<td>page 207; left-hand column, last paragraph - page 209, right-hand column, paragraph 2</td>
<td></td>
</tr>
<tr>
<td>Y</td>
<td>WO 2004/089065 A (BEJO ZADEN B V [NL]; BARTEN PIET [NL]) 21 October 2004 (2004-10-21) page 6, lines 9-11; examples 1-7</td>
<td>1-17</td>
</tr>
<tr>
<td>Category</td>
<td>Citation of document, with indication, where appropriate, of the relevant passages</td>
<td>Relevant to claim No.</td>
</tr>
<tr>
<td>----------</td>
<td>--------------------------------------------------------------------------------</td>
<td>-----------------------</td>
</tr>
<tr>
<td>X</td>
<td>WEST L ET AL: &quot;Single column approach for the liquid chromatographic- separation of pol ar and non-pol ar glucosinolates from brocol i sprouts and seeds&quot; JOURNAL OF CHROMATOGRAPHY, ELSEVIER SCIENCE PUBLISHERS B.V. AMSTERDAM, NL, vol. 966, no. 1-2, 9 August 2002 (2002-08-09), pages 227-232, XP004372036 ISSN: 0021-9673 cited in the application abstract page 228, left-hand column, last paragraph - right-hand column, paragraph 2 page 229, left-hand column, last paragraph - page 231, right-hand column, paragraph 2; table 1</td>
<td>17</td>
</tr>
<tr>
<td>Y</td>
<td>TROYER J K ET AL: &quot;Analysis of glucosinolates from broccoli and other cruciferous vegetables by hydrophilic interaction liquid chromatography&quot; JOURNAL OF CHROMATOGRAPHY, ELSEVIER SCIENCE PUBLISHERS B.V. AMSTERDAM, NL, vol. 919, no. 2, 15 June 2001 (2001-06-15), pages 299-304, XP004248013 ISSN: 0021-9673 abstract page 299, right-hand column, last paragraph - page 300, right-hand column, paragraph 1 page 303, left-hand column, paragraph 3 - right-hand column, paragraph 2</td>
<td>1-17</td>
</tr>
</tbody>
</table>
## DOCUMENTS CONSIDERED TO BE RELEVANT

<table>
<thead>
<tr>
<th>Category</th>
<th>Citation of document, with indication, where appropriate, of the relevant passages</th>
<th>Relevant to claim No.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Patent document cited in search report</td>
<td>Publication date</td>
<td>Patent family member(s)</td>
</tr>
<tr>
<td>---------------------------------------</td>
<td>-----------------</td>
<td>-------------------------</td>
</tr>
<tr>
<td>WO 2004089065 A</td>
<td>21-10-2004</td>
<td>NL 1023179 C2</td>
</tr>
<tr>
<td></td>
<td></td>
<td>US 2007033675 A1</td>
</tr>
</tbody>
</table>