



(86) Date de dépôt PCT/PCT Filing Date: 2005/10/26
(87) Date publication PCT/PCT Publication Date: 2006/05/04
(85) Entrée phase nationale/National Entry: 2007/04/24
(86) N° demande PCT/PCT Application No.: US 2005/038700
(87) N° publication PCT/PCT Publication No.: 2006/047661
(30) Priorités/Priorities: 2004/10/26 (US60/622,993);
2004/10/26 (US60/622,851); 2004/10/26 (US60/622,963);
2004/10/26 (US60/622,931); 2004/10/26 (US60/623,117);
2004/10/26 (US60/622,912); 2004/10/26 (US60/622,994)

(51) Cl.Int./Int.Cl. *A61K 31/335* (2006.01),
C07F 9/40 (2006.01)
(71) Demandeur/Applicant:
GILEAD SCIENCES, INC., US
(72) Inventeurs/Inventors:
WATKINS, WILLIAM J., US;
CHO, AESOP, US
(74) Agent: ROBIC

(54) Titre : DERIVES DE PHOSPHONATE D'ACIDE MYCOPHENOLIQUE
(54) Title: PHOSPHONATE DERIVATIVES OF MYCOPHENOLIC ACID

(57) **Abrégé/Abstract:**

Phosphorus substituted mycophenolate oxime derivatives with anti-cancer, anti-viral, anti-inflammatory and anti-tissue/organ transplant rejection properties having use as therapeutics and for other industrial purposes are disclosed. The compositions inhibit tumor growth, viral growth, inflammation, and tissue/organ transplant rejection and/or are useful therapeutically for the treatment or prevention of cancer, viral infection, inflammation and tissue/organ transplant rejection, as well as in assays for the detection of cancer, viral infection, inflammation and tissue/organ transplant rejection.



(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
4 May 2006 (04.05.2006)

PCT

(10) International Publication Number
WO 2006/047661 A3

(51) International Patent Classification:

A61K 31/335 (2006.01) C07F 9/40 (2006.01)

(21) International Application Number:

PCT/US2005/038700

(22) International Filing Date: 26 October 2005 (26.10.2005)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

60/622,963	26 October 2004 (26.10.2004)	US
60/622,994	26 October 2004 (26.10.2004)	US
60/622,912	26 October 2004 (26.10.2004)	US
60/623,117	26 October 2004 (26.10.2004)	US
60/622,931	26 October 2004 (26.10.2004)	US
60/622,851	26 October 2004 (26.10.2004)	US
60/622,993	26 October 2004 (26.10.2004)	US

(71) Applicant (for all designated States except US): **GILEAD SCIENCES, INC.** [US/US]; 333 Lakeside Drive, Foster City, CA 94404 (US).

(71) Applicants and

(72) Inventors: **WATKINS, William, J.** [GB/US]; 626 Oneida Drive, Sunnyvale, CA 94087 (US). **CHO, Aesop** [US/US]; 1656 Notre Dame Drive, Mountain View, CA 94040 (US).

(74) Agents: **HARRIS, Robert, J.** et al.; Viksnins Harris & Padys PLLP, P.O. Box 111098, St. Paul, MN 55111-1098 (US).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

(88) Date of publication of the international search report:

24 August 2006

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: PHOSPHONATE DERIVATIVES OF MYCOPHENOLIC ACID

(57) Abstract: Phosphorus substituted mycophenolate oxime derivatives with anti-cancer, anti-viral, anti-inflammatory and anti-tissue/organ transplant rejection properties having use as therapeutics and for other industrial purposes are disclosed. The compositions inhibit tumor growth, viral growth, inflammation, and tissue/organ transplant rejection and/or are useful therapeutically for the treatment or prevention of cancer, viral infection, inflammation and tissue/organ transplant rejection, as well as in assays for the detection of cancer, viral infection, inflammation and tissue/organ transplant rejection.



WO 2006/047661 A3

PHOSPHONATE DERIVATIVES OF MYCOPHENOLIC ACID

5

RELATED APPLICATIONS

This patent document claims the benefit of priority of the following United States provisional patent applications, all of which were filed on October 26, 2004, and all of which are herein incorporated by reference: Serial No. 60/622,851; Serial No. 60/622,912; Serial No. 60/622,931; Serial No. 60/622,963; Serial No. 60/622,993; Serial No. 60/622,994; and Serial No. 60/623,117.

FIELD OF THE INVENTION

The invention relates generally to compounds with immunosuppressive, anti-inflammatory, anti-cancer and/or anti-viral activity.

15

BACKGROUND OF THE INVENTION

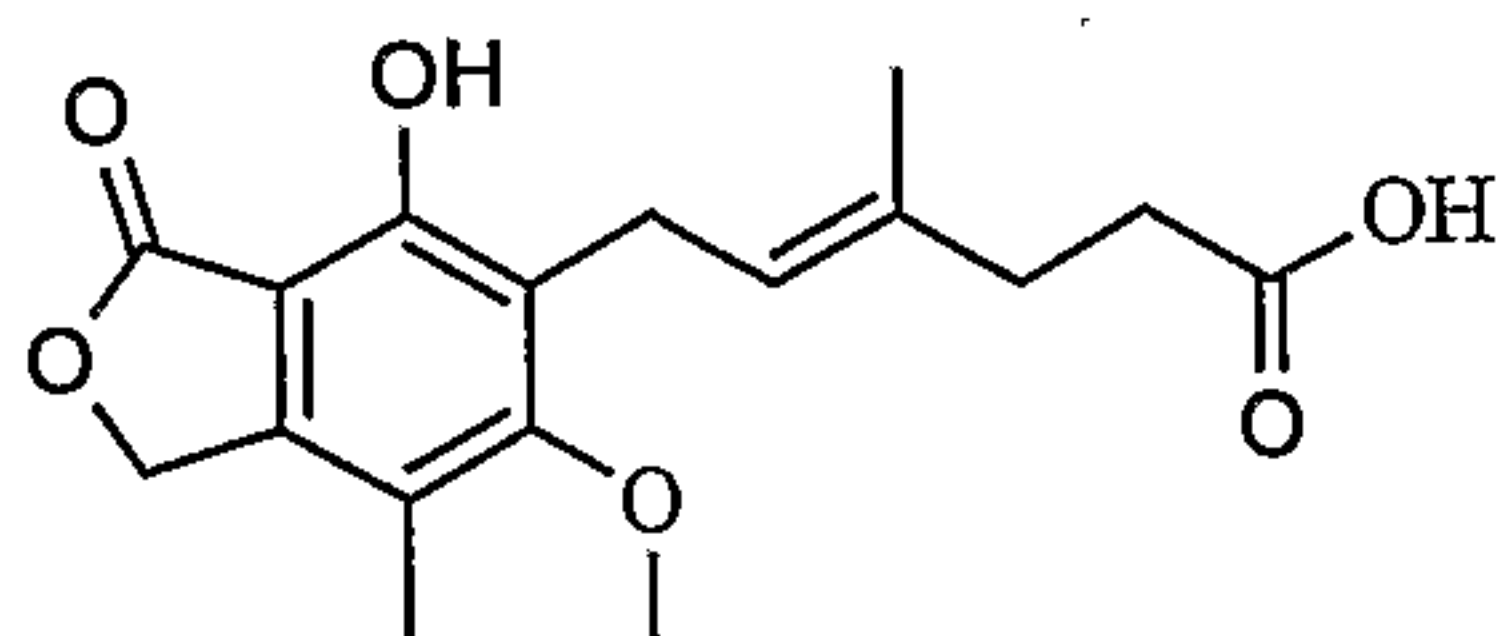
Improving the delivery of drugs and other agents to target cells and tissues has been the focus of considerable research for many years. Though many attempts have been made to develop effective methods for importing biologically active molecules into cells, both *in vivo* and *in vitro*, none has proved to be entirely satisfactory. Optimizing the association of the inhibitory drug with its intracellular target, while minimizing intercellular redistribution of the drug, *e.g.*, to neighboring cells, is often difficult or inefficient.

Most agents currently administered to a patient parenterally are not targeted, resulting in systemic delivery of the agent to cells and tissues of the body where it is unnecessary, and often undesirable. This may result in adverse drug side effects, and often limits the dose of a drug (*e.g.*, glucocorticoids and other anti-inflammatory drugs) that can be administered. By comparison, although oral administration of drugs is generally recognized as a convenient and economical method of administration, oral administration can result in either (a) uptake of the drug through the cellular and tissue barriers, *e.g.*, blood/brain, epithelial, cell membrane, resulting in undesirable systemic distribution, or (b) temporary residence of the drug within the gastrointestinal tract. Accordingly, a major goal has been to develop methods for specifically targeting agents to cells and tissues. Benefits of such treatment includes avoiding the general

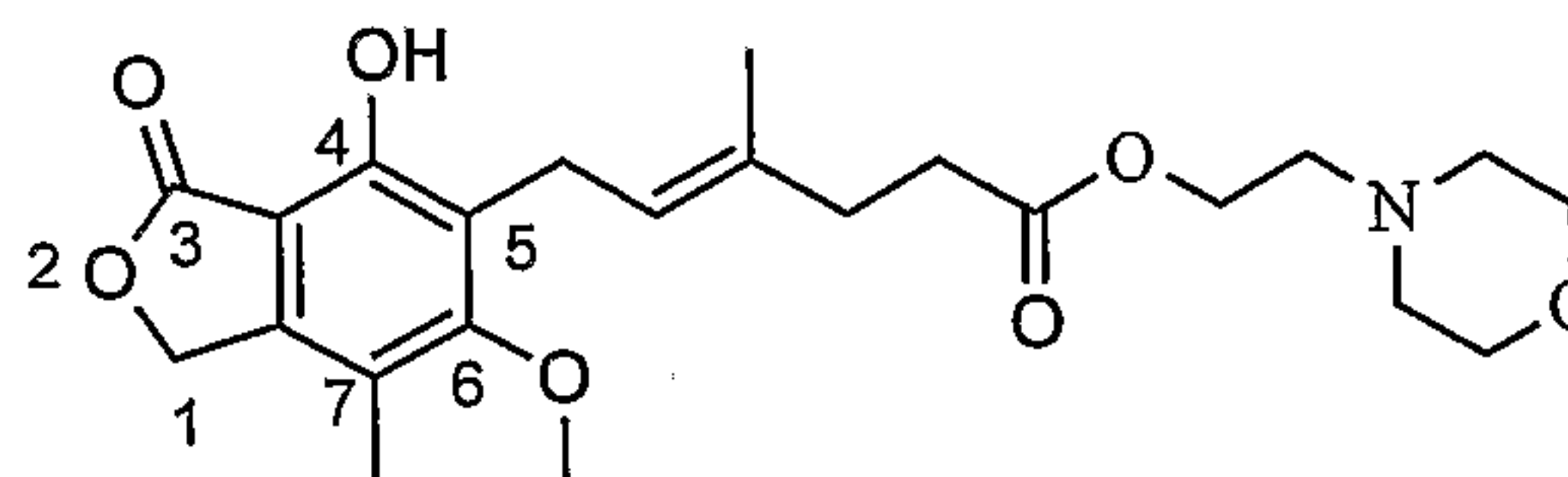
physiological effects of inappropriate delivery of such agents to other cells and tissues, such as uninfected cells.

Mycophenolic acid is a weakly-active antibiotic found in the fermentation broth of *Penicillium brevicompactum* (Allison *et al.* US4786637 and Nelson *et al.* US4753935). Mycophenolate mofetil, a prodrug of mycophenolic acid, is used for prevention of acute renal and cardiac allograft rejection in combination with cyclosporine (Sandoz' Sandimmune) and corticosteroids (The Pink Sheet, 1995, 57, 20). Mycophenolate mofetil is rapidly converted after oral or intravenous administration to its active metabolite, mycophenolic acid (Lee *et al. Pharm Res*, 1990, 7, 161). In fact, over 90% of the drug is cleared by glucuronidation and is excreted in the urine.

Mycophenolic acid is a noncompetitive, selective and reversible inhibitor of inosine monophosphate dehydrogenase (IMPDH), an important enzyme in the *de novo* synthesis of guanosine nucleotides in T and B lymphocytes (Brazelton, T. R., Morris, R. E. *Current Opinion in Immunology*, 1996, 8, 710; Fulton, B, Markham, A, *Drugs*, 1996, 51,2,278). IMPDH catalyzes the NAD-dependent oxidation of inosine 5'-monophosphate (IMP) to xanthate 5'-monophosphate. Rapidly proliferating cells and/or viral infections are heavily dependent on the availability of large nucleotide pools to meet their metabolic requirements. Compounds blocking this *de novo* biosynthesis pathway will act selectively on these cell types and leave the other ones substantially unaffected. Several classes of IMPDH inhibitors are now in use or under development, such as inhibitors that bind to the substrate site (ribavirin) or the NAD site (mycophenolic acid) (Goldstein, B. M., Colby, T. D. *Current Medicinal Chemistry*, 1999, 6, 7, 519).



Mycophenolic acid



Mycophenolate mofetil (MMF, RS-61443)

Mycophenolate mofetil (MMF) requires a 2-3 g/day dosing regimen in conjunction with other immunosuppressants. Although MMF is comparatively benign, gastrointestinal adverse effects have been observed (Platz *et al.*

Transplantation, 1991, 51, 1, 27). These adverse effects have been explained as the result of the combination therapy necessary for efficacy of the MMF drug. In some cases, adjustment of dosage of the immunosuppressants, particularly spreading the total dosage to 2 or more, may be sufficient (Behrand, M. *Drug Saf* 5 2001, 24, 9, 645). Reduction of the dose, increase of $t_{1/2}$, and improvement of efficacy by targeted delivery of this drug may be achieved by preparation of other prodrugs that have better intracellular trapping. The present invention describes preparation of phosphonate prodrugs of mycophenolic acid.

Organ transplant rejection is a major problem for transplant recipients. 10 Although anti-rejection drugs are commonly used, they are not always effective and often times are toxic over the long term.

Inflammation is also a major problem for many people. Effective anti-inflammatory drugs, with limited side effects, are in need.

Cancer is another major health problem worldwide. Although drugs 15 targeting tumors and cancerous cells are in wide use and have shown effectiveness, toxicity and side-effects have limited their usefulness.

Finally, viral infections are still another major public health problem worldwide. Although drugs targeting viruses are in wide use and have shown effectiveness, toxicity and development of resistant viral strains have limited 20 their usefulness.

Assay methods capable of determining the presence, absence or amounts of organ rejection, inflammation, cancer and viral infections are of practical utility in the search for inhibitors as well as for diagnosing the presence of organ rejection, inflammation, cancer and viral infections.

25 Inhibitors of viruses are useful to limit the establishment and progression of infection by a virus as well as in diagnostic assays for viral infections.

Inhibitors of tumor growth are useful to limit the establishment and progression of cancer, as well as in diagnostic assays for cancer.

Inflammation inhibitors are useful for inhibiting inflammation. Finally, 30 immunosuppressants are useful for suppressing the immune response to a foreign organ or tissue.

There is a need for anti-cancer, anti-viral, anti-inflammation and anti-rejection (*i.e.* immunosuppressive) therapeutic agents, *i.e.* drugs, having

improved anti-cancer, anti-viral, anti-inflammation or immunosuppressive properties, as well as pharmacokinetic properties, including enhanced activity against development of cancer, viral infection, inflammation and organ rejection, improved oral bioavailability, greater potency and extended effective half-life *in vivo*. Such inhibitors should be active against various cancers, virus strains, inflammation and/or tissue/organ rejection, have distinct resistance profiles, fewer side effects, less complicated dosing schedules, and be orally active. In particular, there is a need for a less onerous dosage regimen for such inhibitors, for example, one pill, once per day.

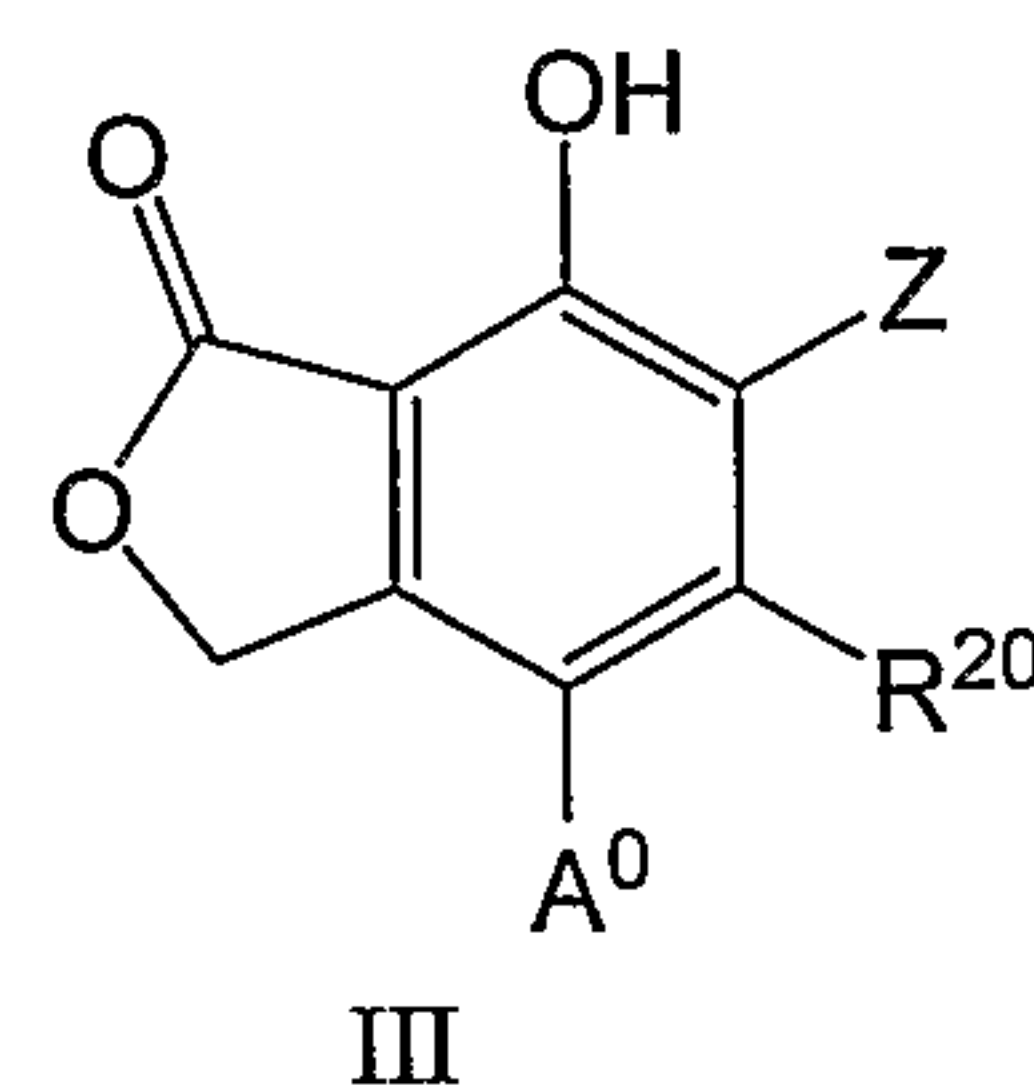
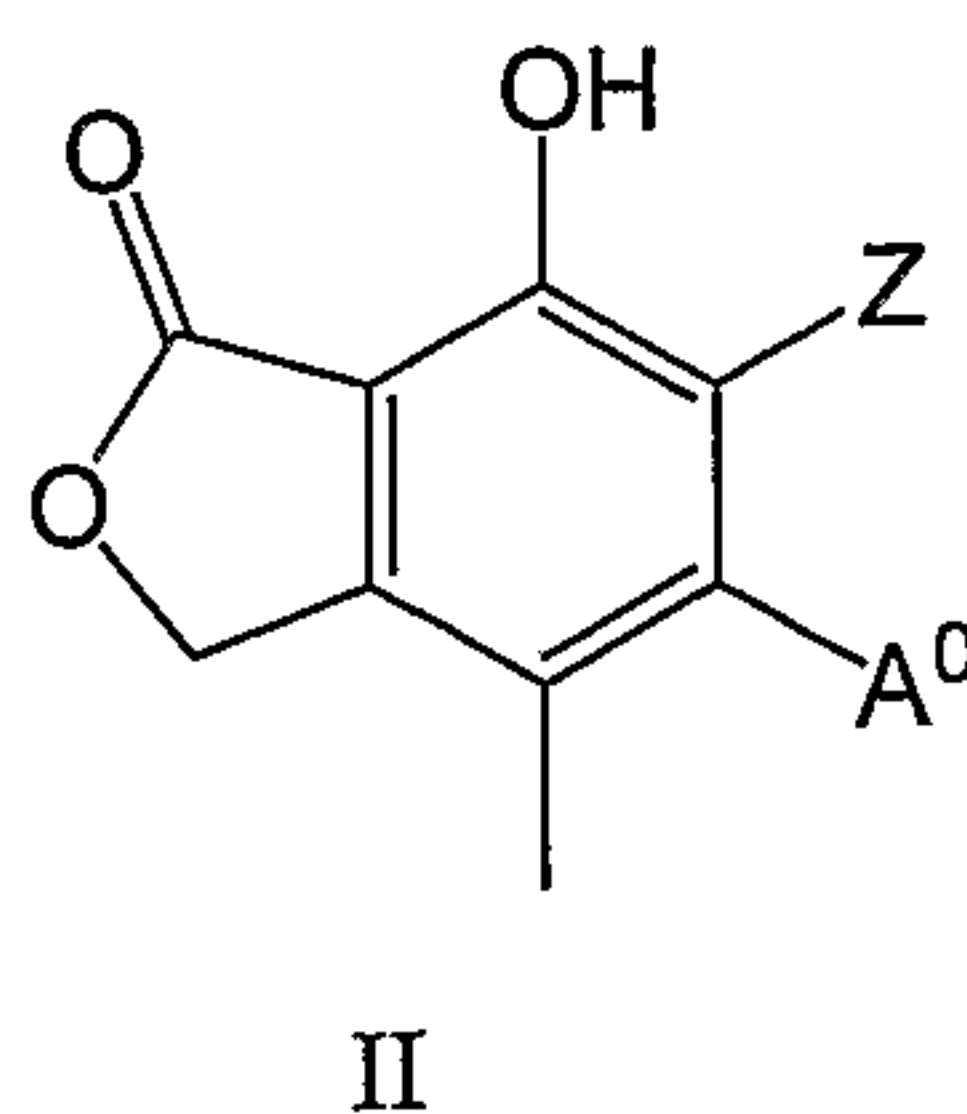
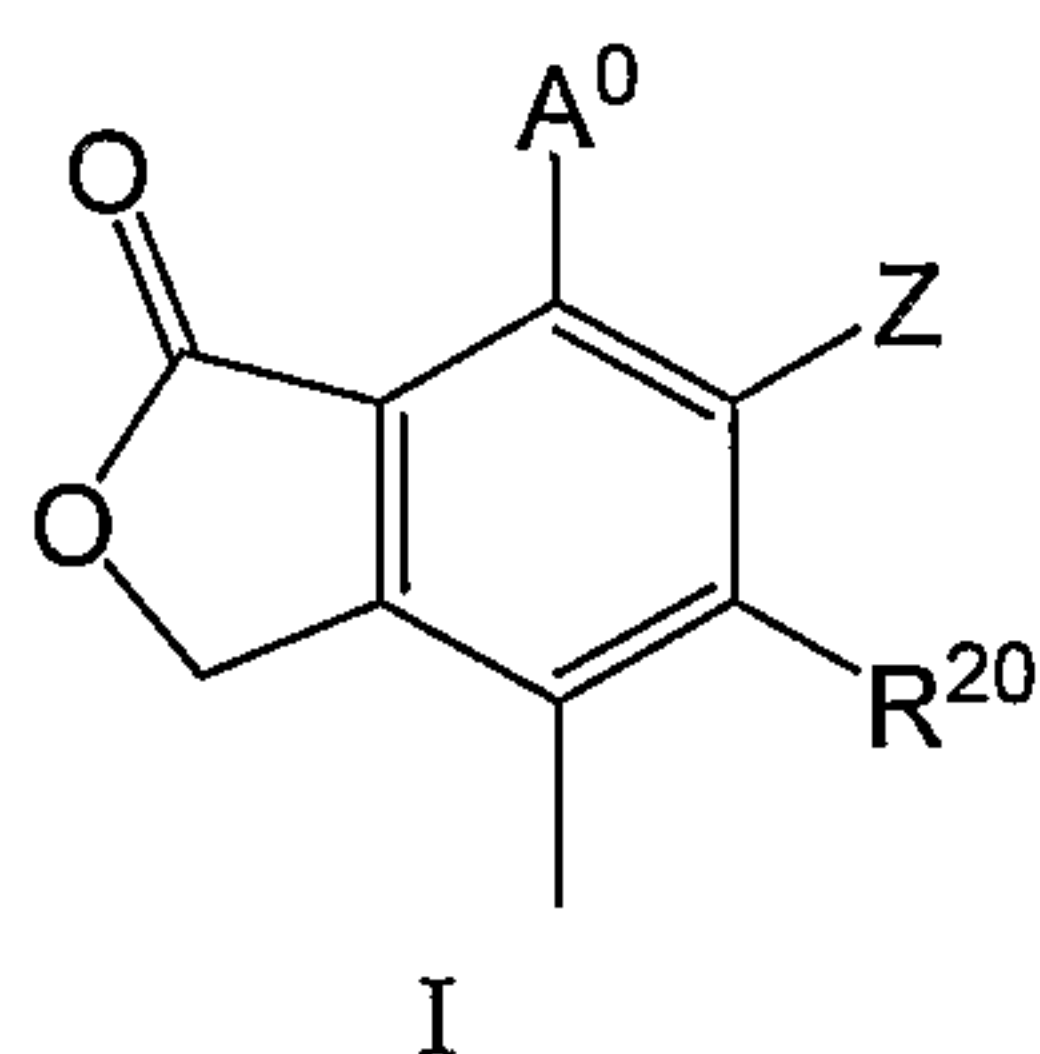
10 SUMMARY OF THE INVENTION

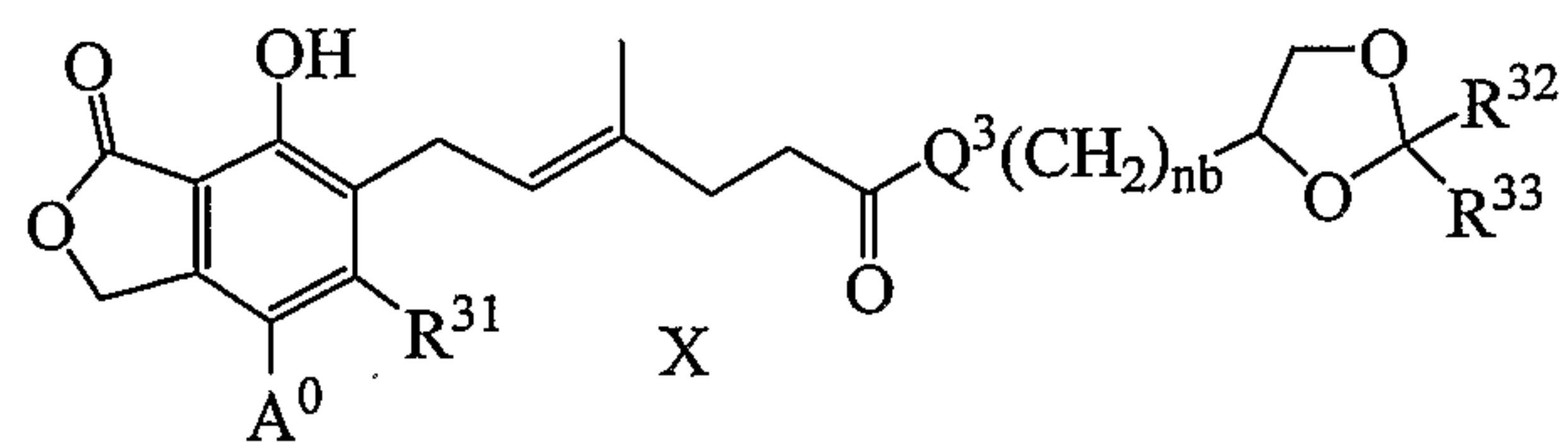
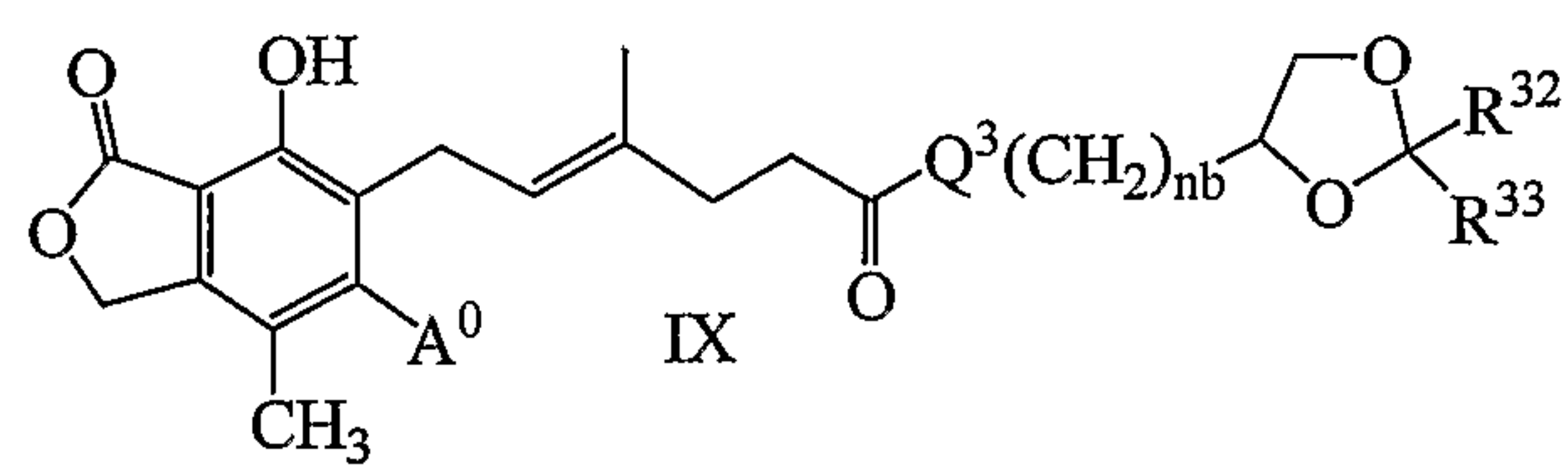
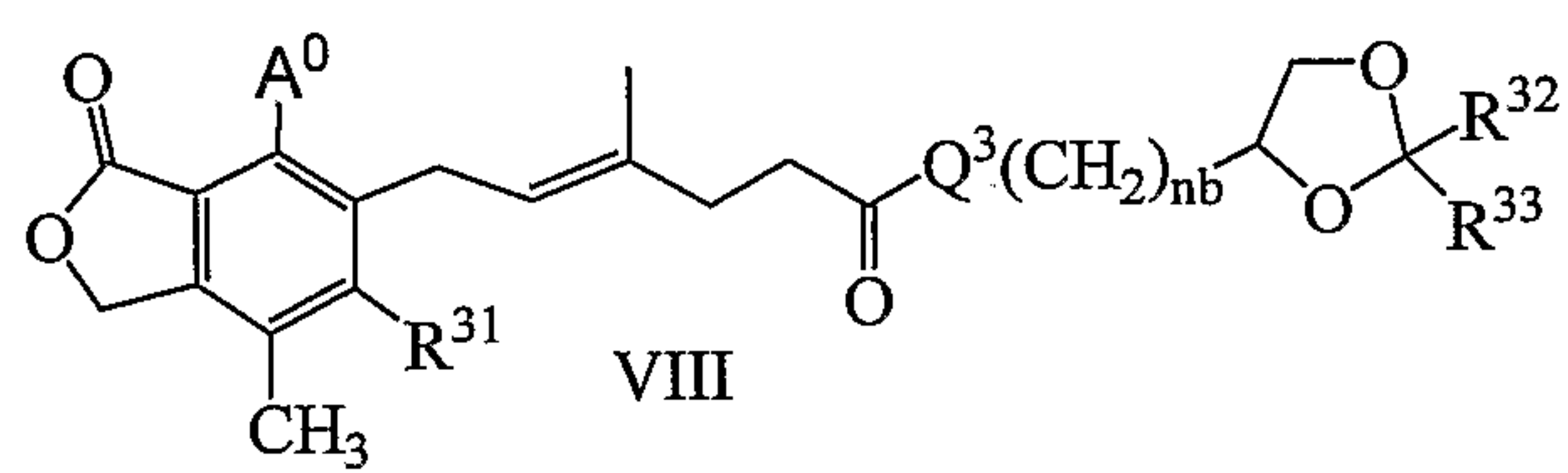
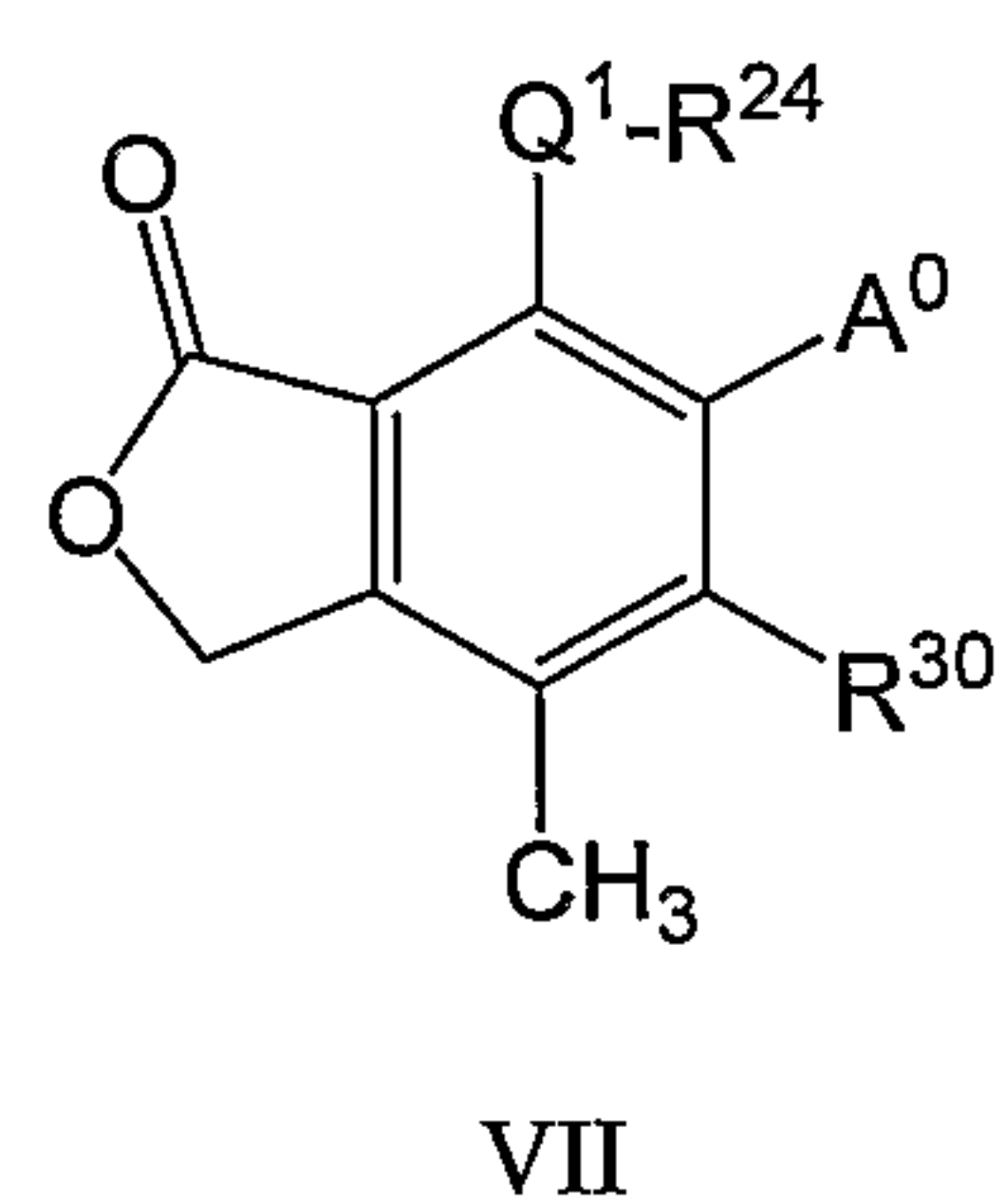
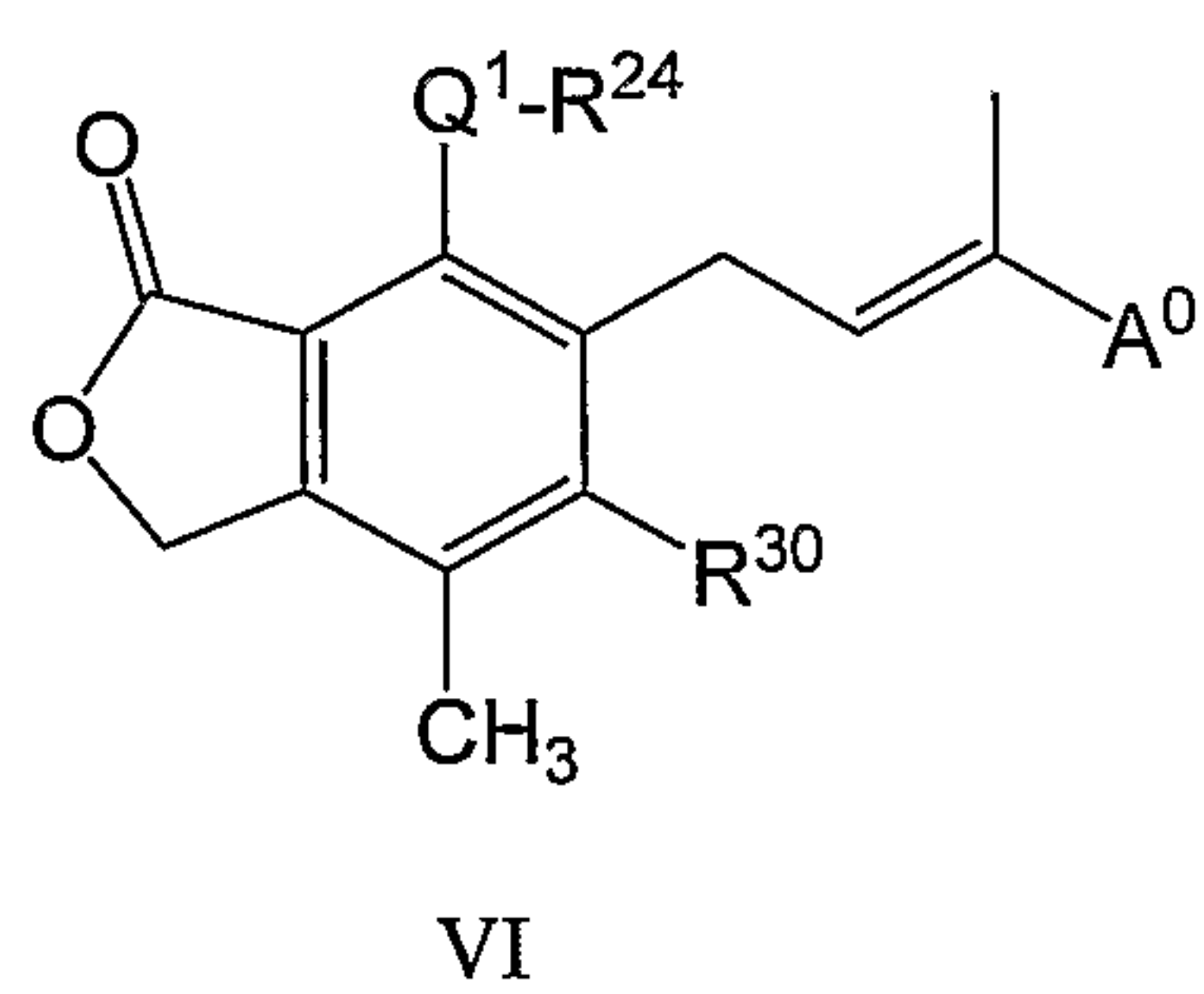
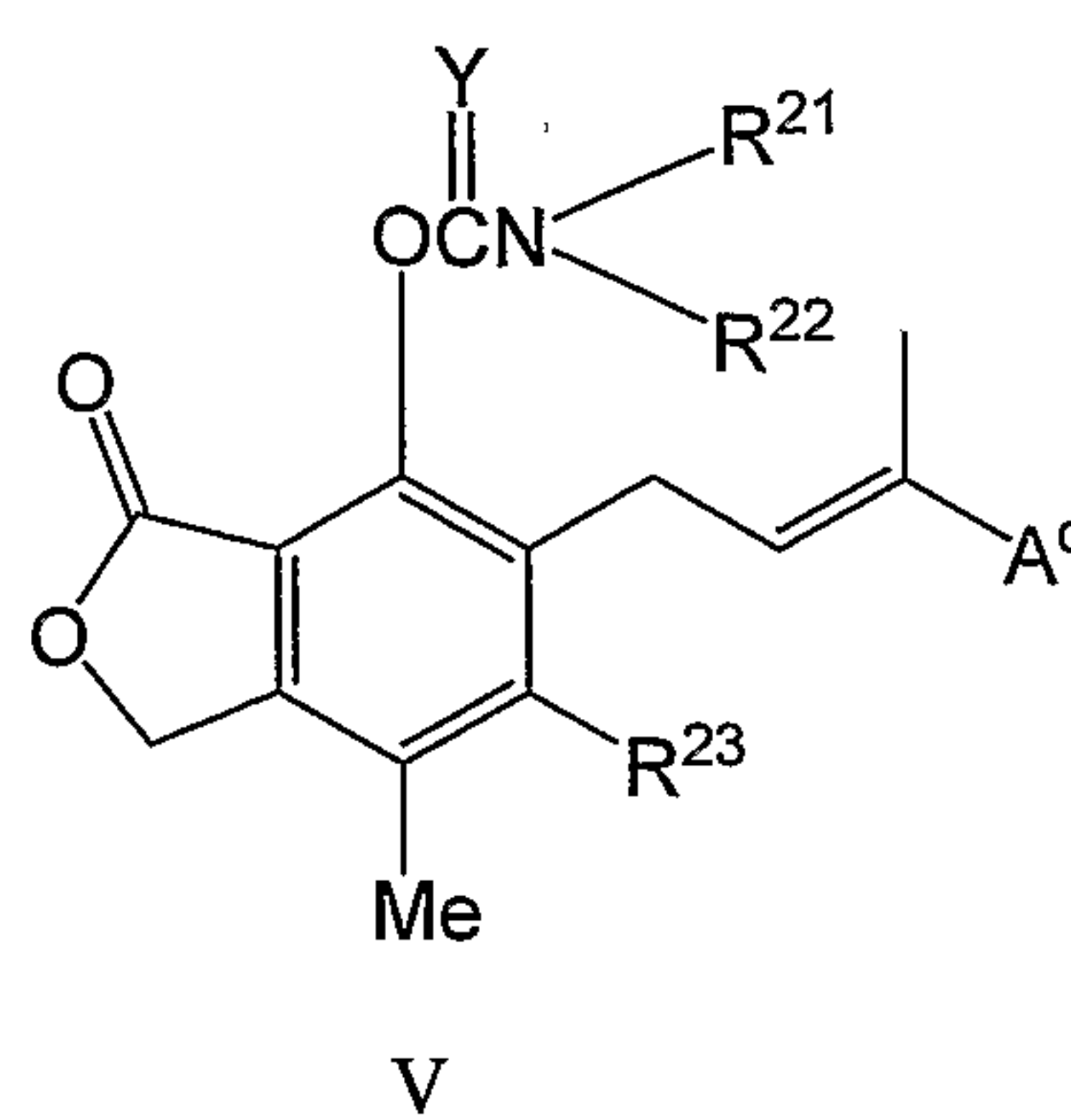
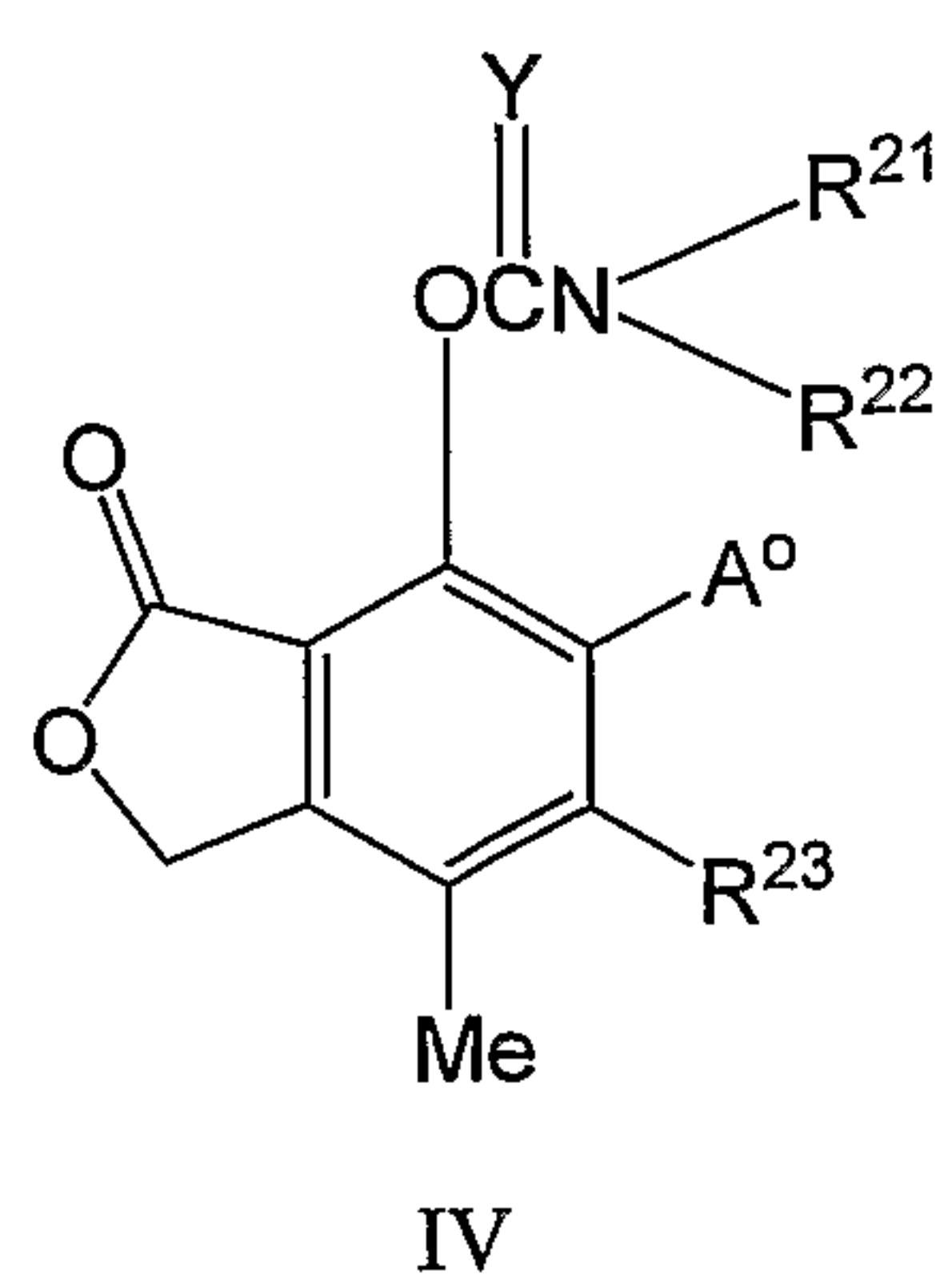
Intracellular targeting may be achieved by methods and compositions that allow accumulation or retention of biologically active agents inside cells. The present invention provides novel analogs of compounds that inhibit tumor growth, viral infection, inflammation, and tissue/organ rejection. Such novel compounds possess utilities of mycophenolic acid and its derivatives, and optionally provide cellular accumulation as set forth below.

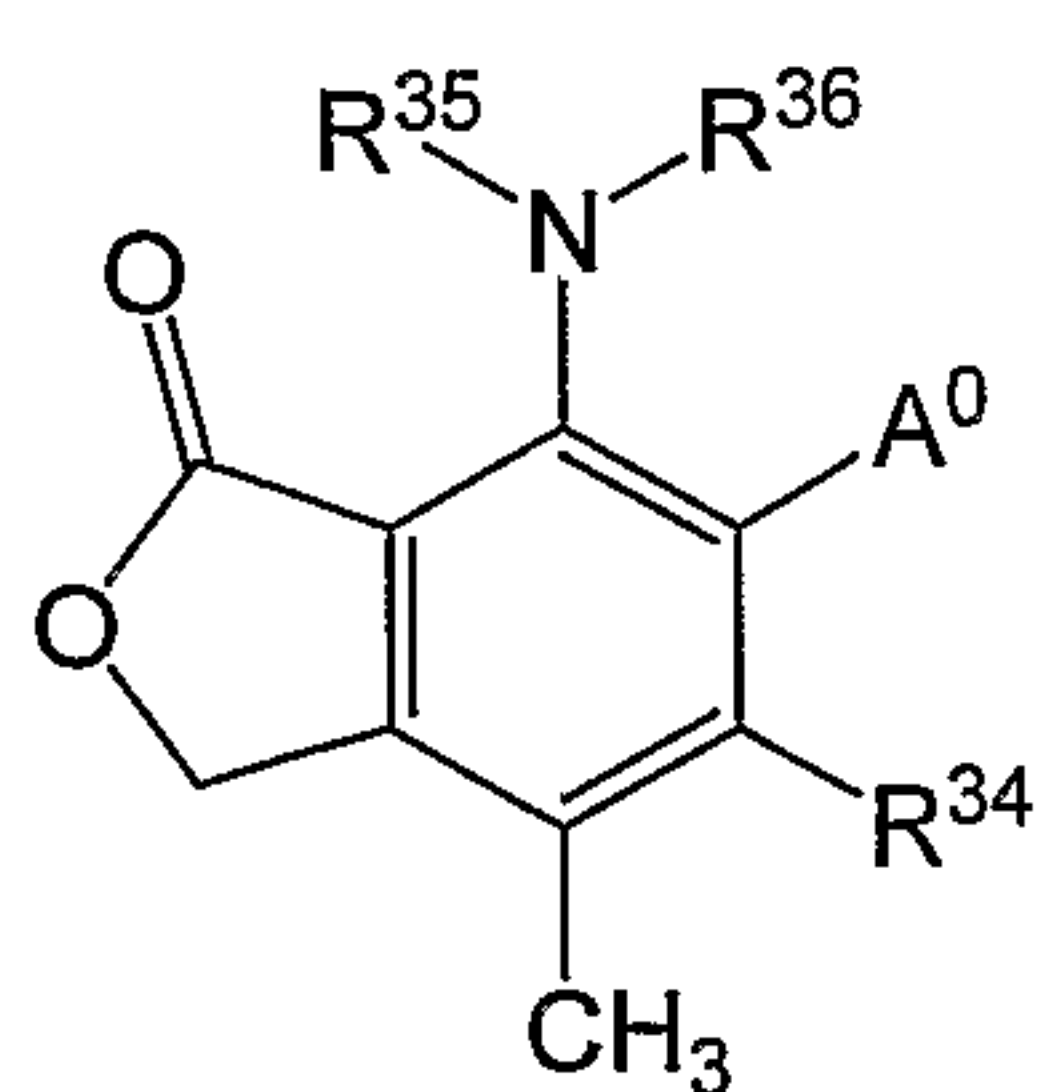
The present invention relates generally to the accumulation or retention of therapeutic compounds inside cells. The invention is more particularly related to attaining high concentrations of phosphonate-containing molecules in target cells. Such effective targeting may be applicable to a variety of therapeutic formulations and procedures.

Compositions of the invention include mycophenolic acid derivatives having at least one phosphonate group.

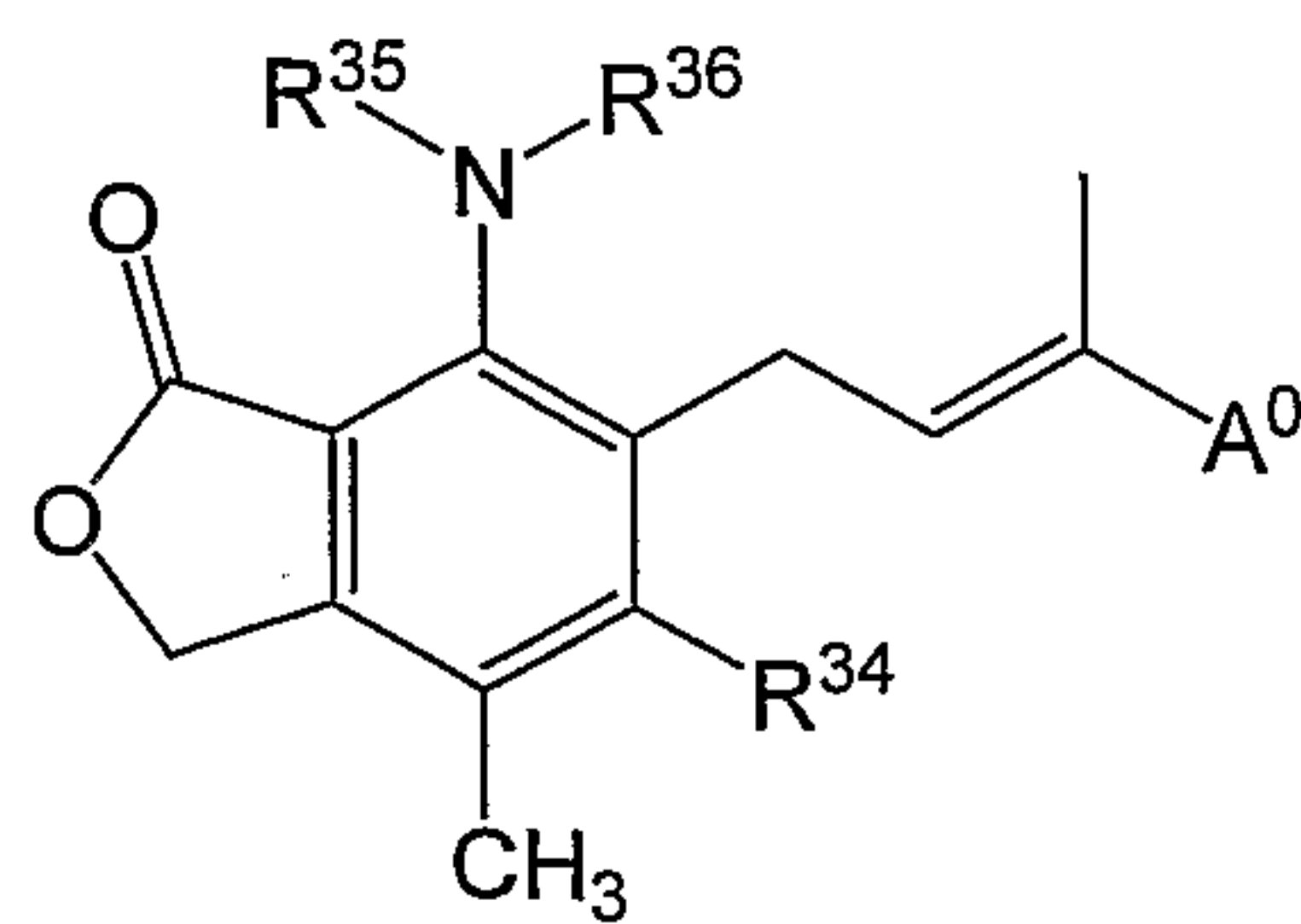
Accordingly, in one embodiment the invention provides a compound of any one of formulae I-XII:







XI

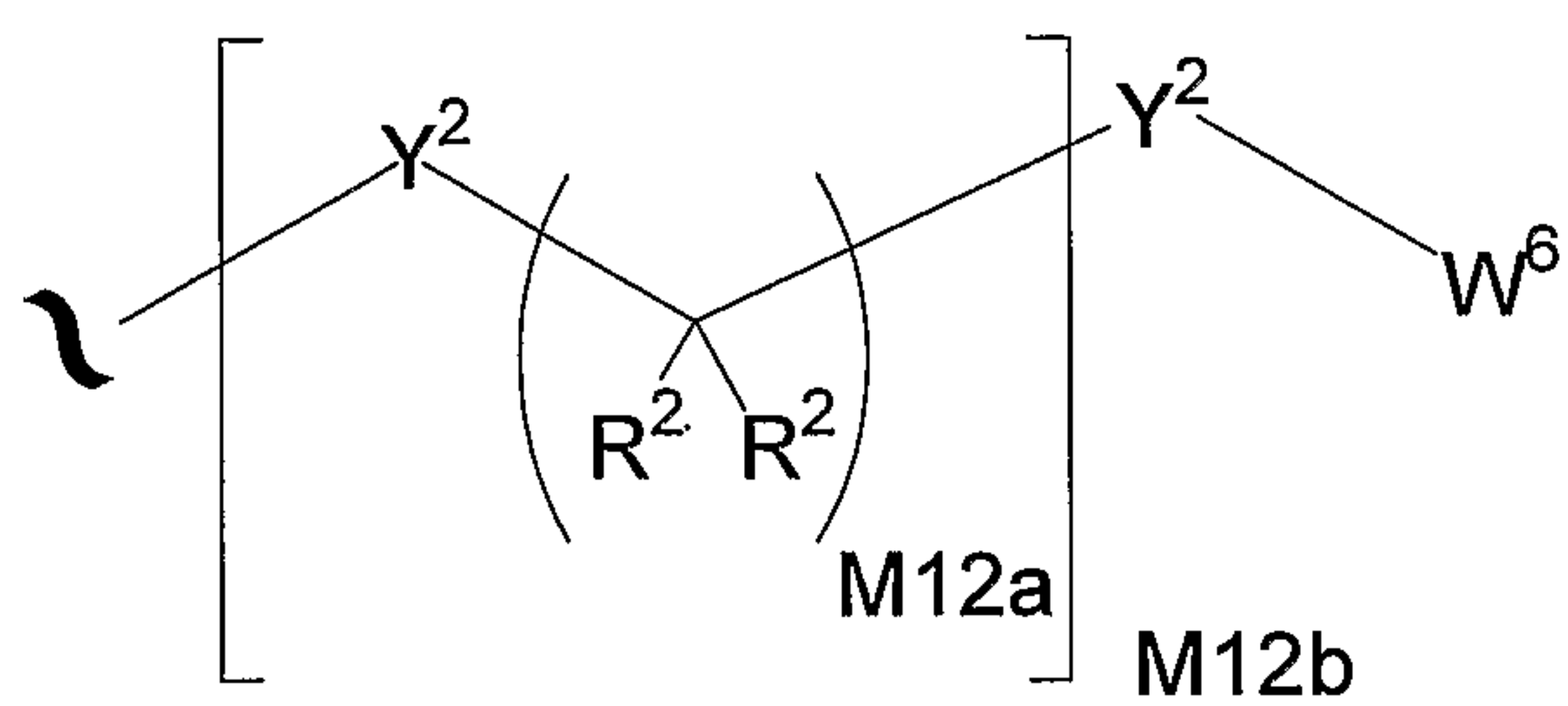


XII

wherein:

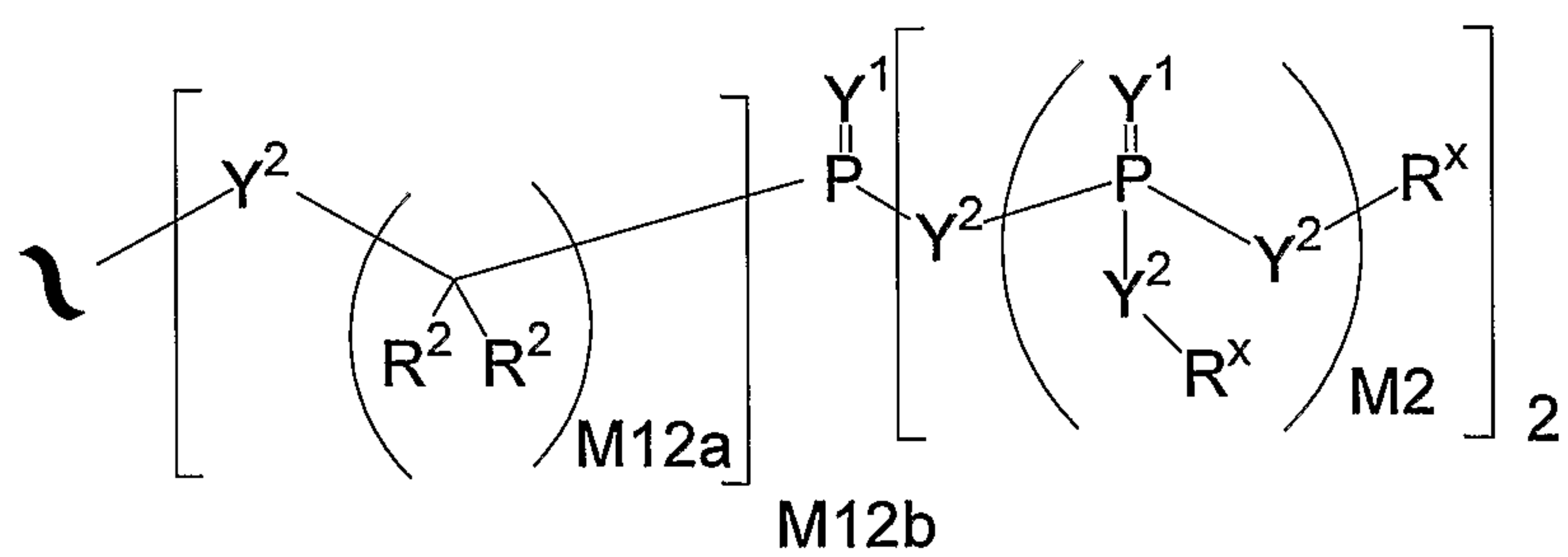
A^0 is A^1 ;

A^1 is:



5

A^3 is:



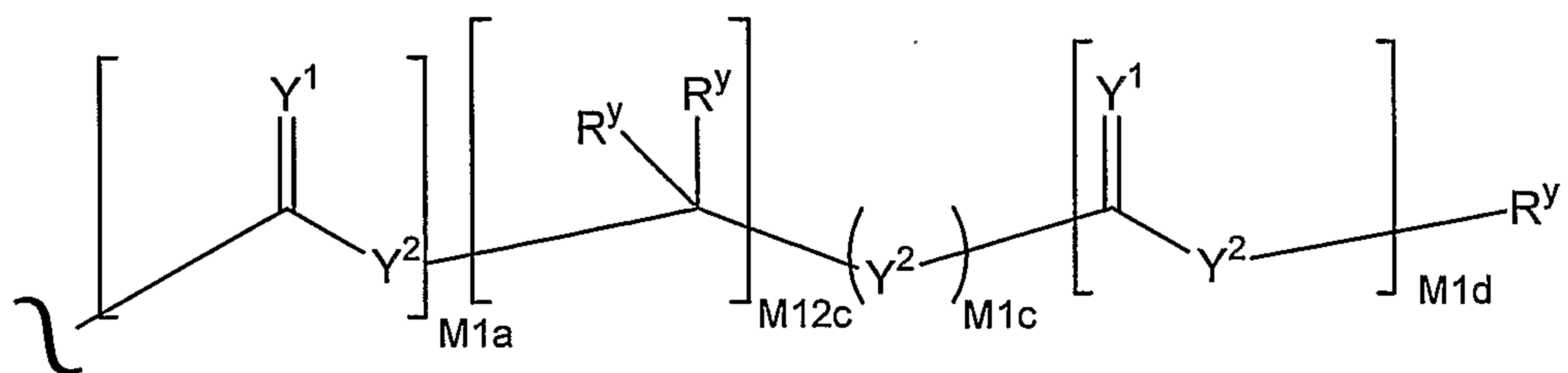
;

Y^1 is independently O, S, $N(R^x)$, $N(OR^x)$, or $N(N(R^x)(R^x))$;

Y^2 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, or

10 $-S(O)_{M2}-$; and when Y^2 joins two phosphorous atoms Y^2 can also be $C(R^2)(R^2)$;

R^x is independently H, R^2 , W^3 , a protecting group, or the formula:



;

R^y is independently H, W^3 , R^2 or a protecting group;

R^2 is independently H, R^3 or R^4 wherein each R^4 is independently

15 substituted with 0 to 3 R^3 groups;

R^3 is R^{3a} , R^{3b} , R^{3c} or R^{3d} , provided that when R^3 is bound to a heteroatom, then R^3 is R^{3c} or R^{3d} ;

R^{3a} is F, Cl, Br, I, -CN, N_3 or $-NO_2$;

R^{3b} is Y^1 ;

5 R^{3c} is $-R^x$, $-N(R^x)(R^x)$, $-SR^x$, $-S(O)R^x$, $-S(O)_2R^x$, $-S(O)(OR^x)$, $-S(O)_2(OR^x)$, $-OC(Y^1)R^x$, $-OC(Y^1)OR^x$, $-OC(Y^1)(N(R^x)(R^x))$, $-SC(Y^1)R^x$, $-SC(Y^1)OR^x$, $-SC(Y^1)(N(R^x)(R^x))$, $-N(R^x)C(Y^1)R^x$, $-N(R^x)C(Y^1)OR^x$, or $-N(R^x)C(Y^1)(N(R^x)(R^x))$;

R^{3d} is $-C(Y^1)R^x$, $-C(Y^1)OR^x$ or $-C(Y^1)(N(R^x)(R^x))$;

10 R^4 is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms, or alkynyl of 2 to 18 carbon atoms;

R^5 is R^4 wherein each R^4 is substituted with 0 to 3 R^3 groups;

W^3 is W^4 or W^5 ;

W^4 is R^5 , $-C(Y^1)R^5$, $-C(Y^1)W^5$, $-SO_2R^5$, or $-SO_2W^5$;

15 W^5 is carbocycle or heterocycle wherein W^5 is independently substituted with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

M2 is 0, 1 or 2;

M12a is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

20 M12b is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

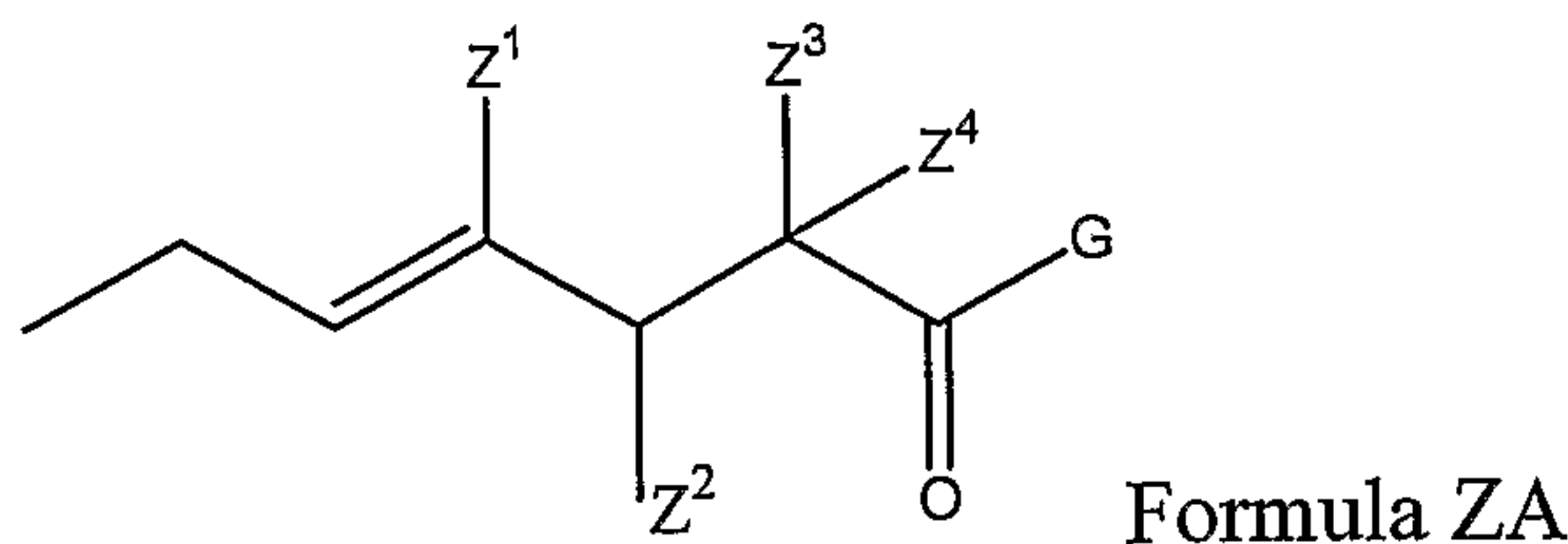
M1a, M1c, and M1d are independently 0 or 1;

M12c is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

R^{20} is ethyl, vinyl, or methoxy;

Z is a side chain selected from Formulae ZA, ZB, ZC, ZD, ZE, ZF, ZG,

25 and ZH:



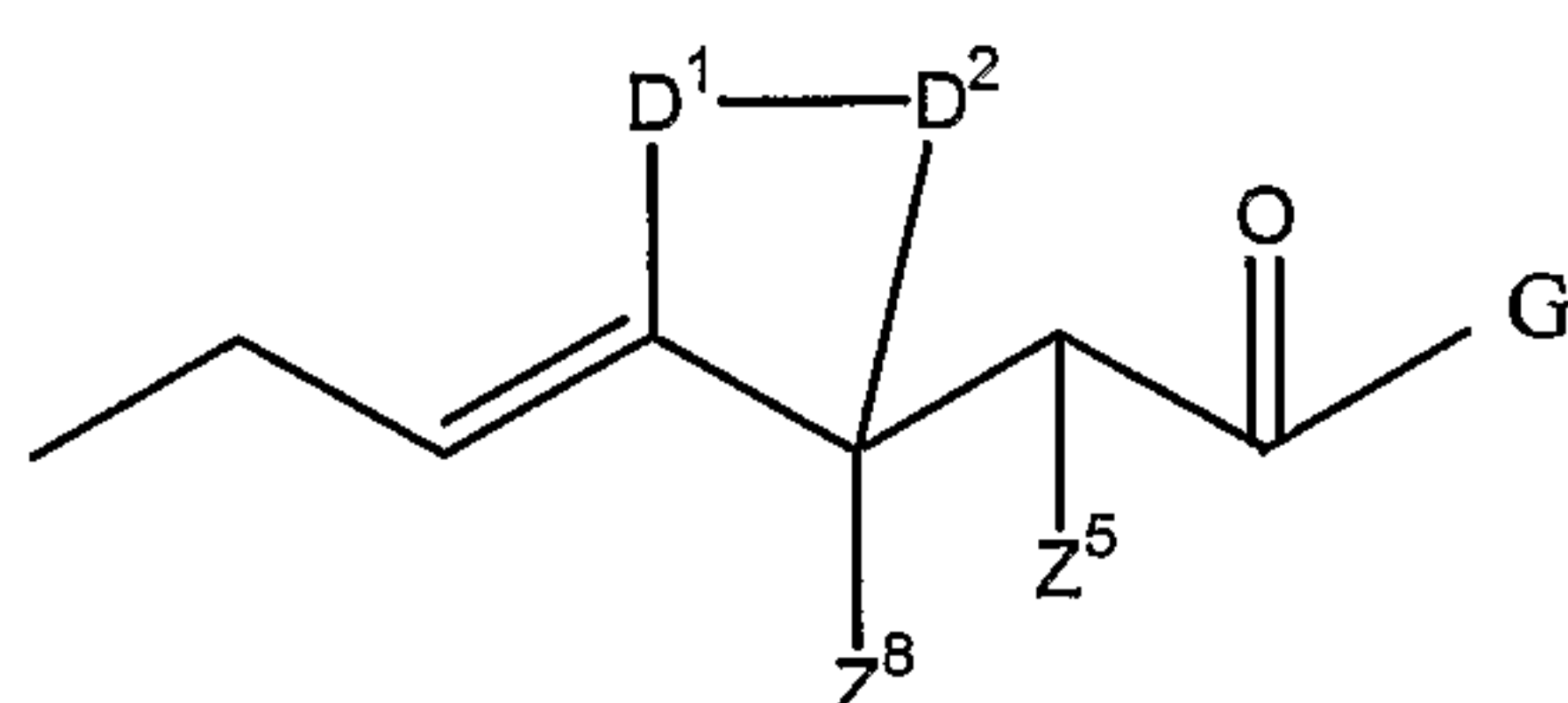
wherein:

Z^1 is hydrogen, alkyl, halo or CF_3 ;

Z^2 is hydrogen, alkyl, alkoxy, aryl, or $-CH_2Z^{13}$, where Z^{13} is halo, CN,

30 aryl or heterocycle;

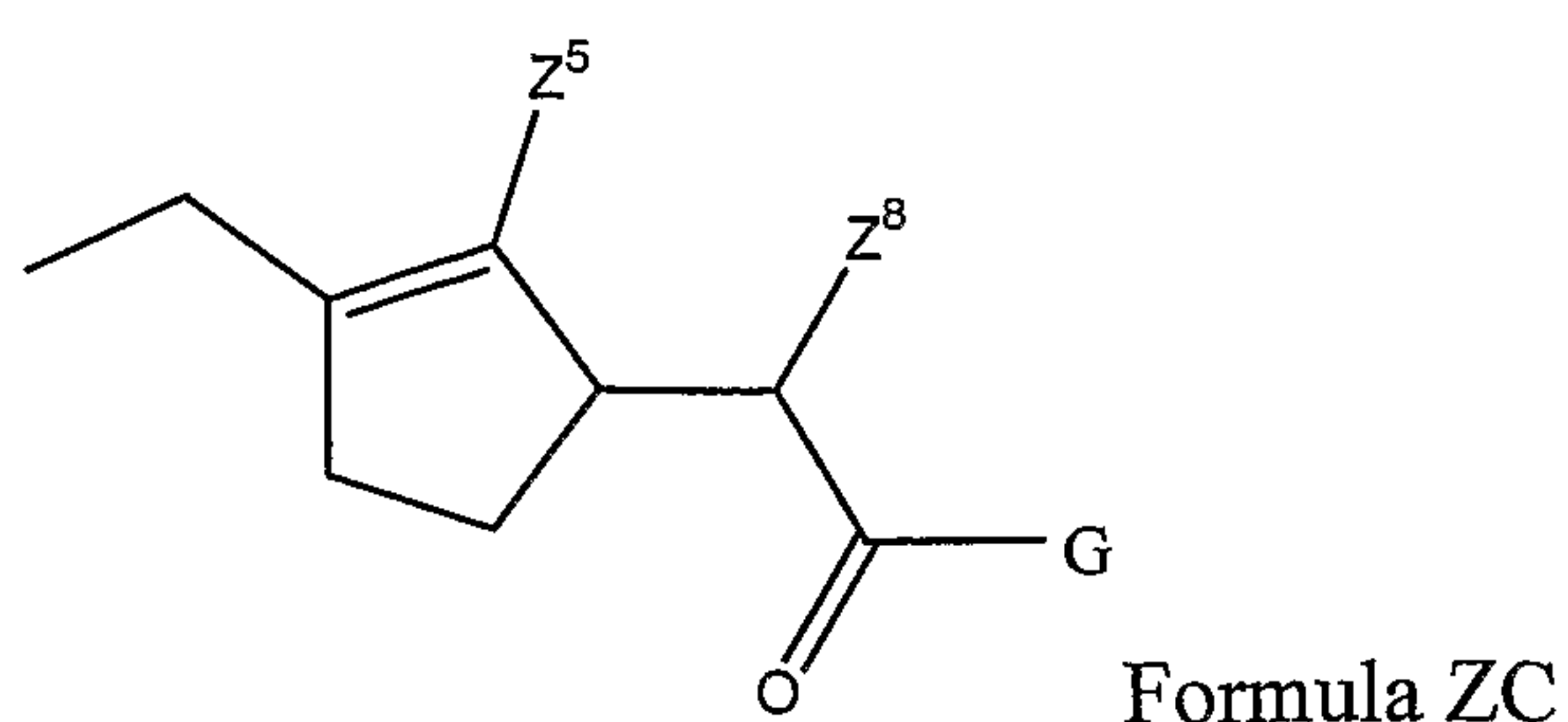
Z^3 is hydrogen, OH, SH, halo, alkyl, alkenyl, alkoxy, aryl,
 $-P(O)(OCH_3)_2$, $-P(O)(OH)(OCH_3)$, NHZ^{11} , or $-S(O)_m Z^{12}$,
 wherein Z^{11} is H, alkyl, acyl, or alkylsulfonyl;
 Z^{12} is alkyl; and
 5 m is 0, 1, or 2;
 Z^4 is hydrogen, OH, halo, alkyl, or aryl, provided that Z^4 is not OH or
 halo when Z^3 is OH, halo, $-P(O)(OCH_3)_2$, $-P(O)(OH)(OCH_3)$, NHZ^{11} , or SZ^{12} ;
 or Z^3 and Z^4 taken together with the carbon to which they are attached
 form cycloalkyl of three to five carbon atoms; and
 10 G is OH, alkoxy, thioalkyl, $-NG^1G^2$, $-O(CH_2)_n NG^1G^2$, or
 $-O(CH_2)_n N=G^3$, where
 n is an integer from 1 to 6,
 G^1 is hydrogen or alkyl,
 G^2 is hydrogen or alkyl, and
 15 $=G^3$ is alkylene of four to six carbon atoms, or alkylene of three to five
 carbon atoms plus one member that is $-O-$, $-S-$, or $-N(G^4)-$ where G^4 is hydrogen
 or alkyl;
 provided that when Z^1 is methyl, Z^2 , Z^3 , and Z^4 are not all H; or



Formula ZB

20

wherein:
 Z^5 is hydrogen or alkyl;
 Z^8 is hydrogen, alkyl, or forms a double bond with D^2 ;
 D^1 and D^2 together with their adjacent carbon atoms form an optionally
 25 substituted, saturated or unsaturated carbocyclic or heterocyclic ring of 3 to 7
 atoms; and
 G is as defined above; or

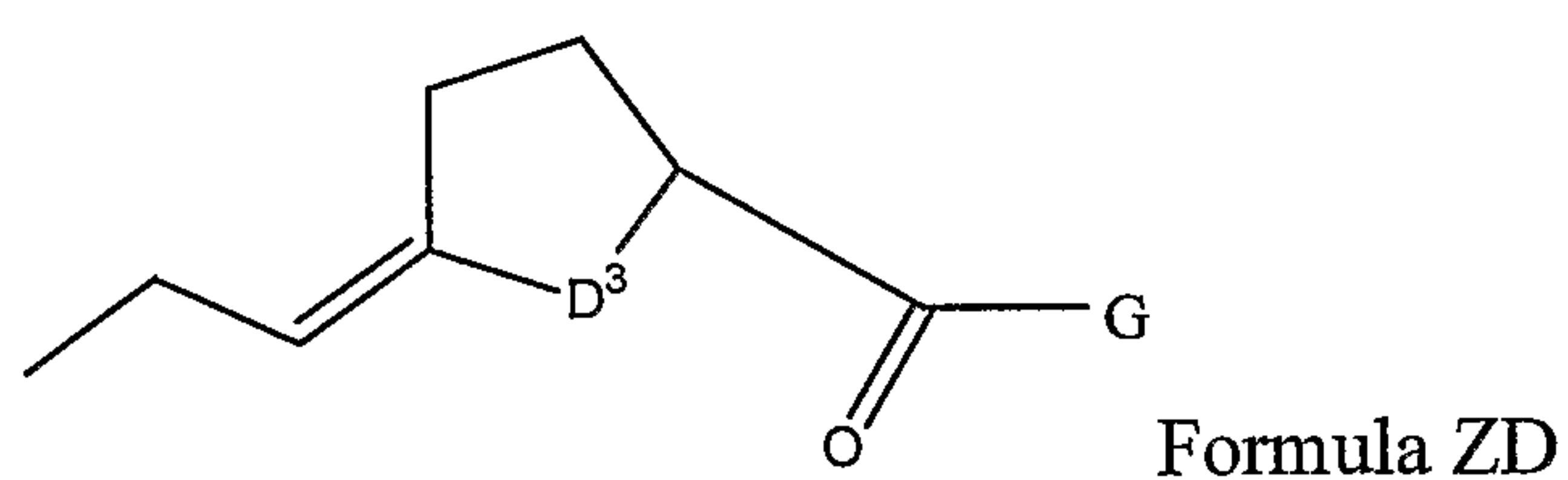


wherein:

Z⁸ is H or alkyl; and

Z⁵ and G are as defined above; or

5

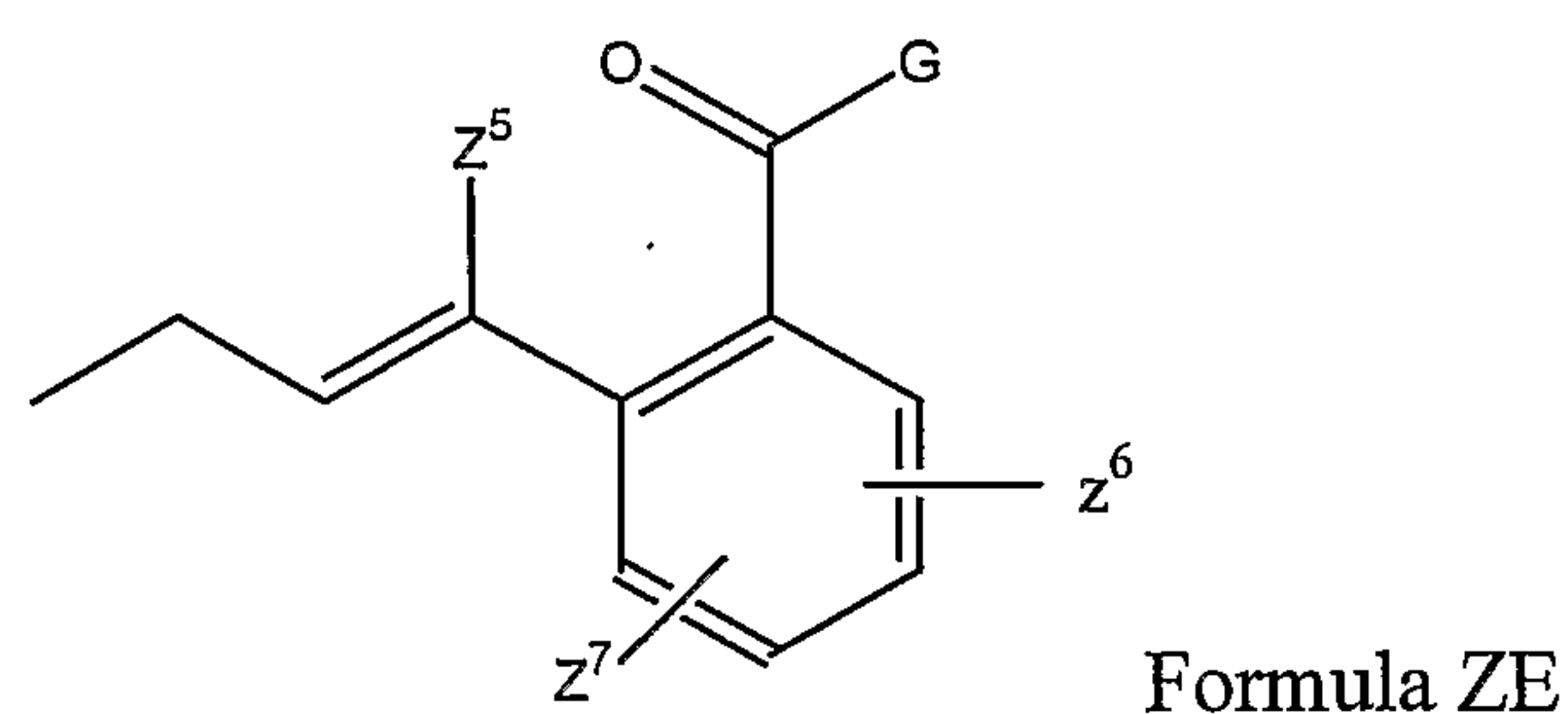


wherein:

D³ is -CH₂ - or -CH₂CH₂ -; and

G is as defined above; or

10



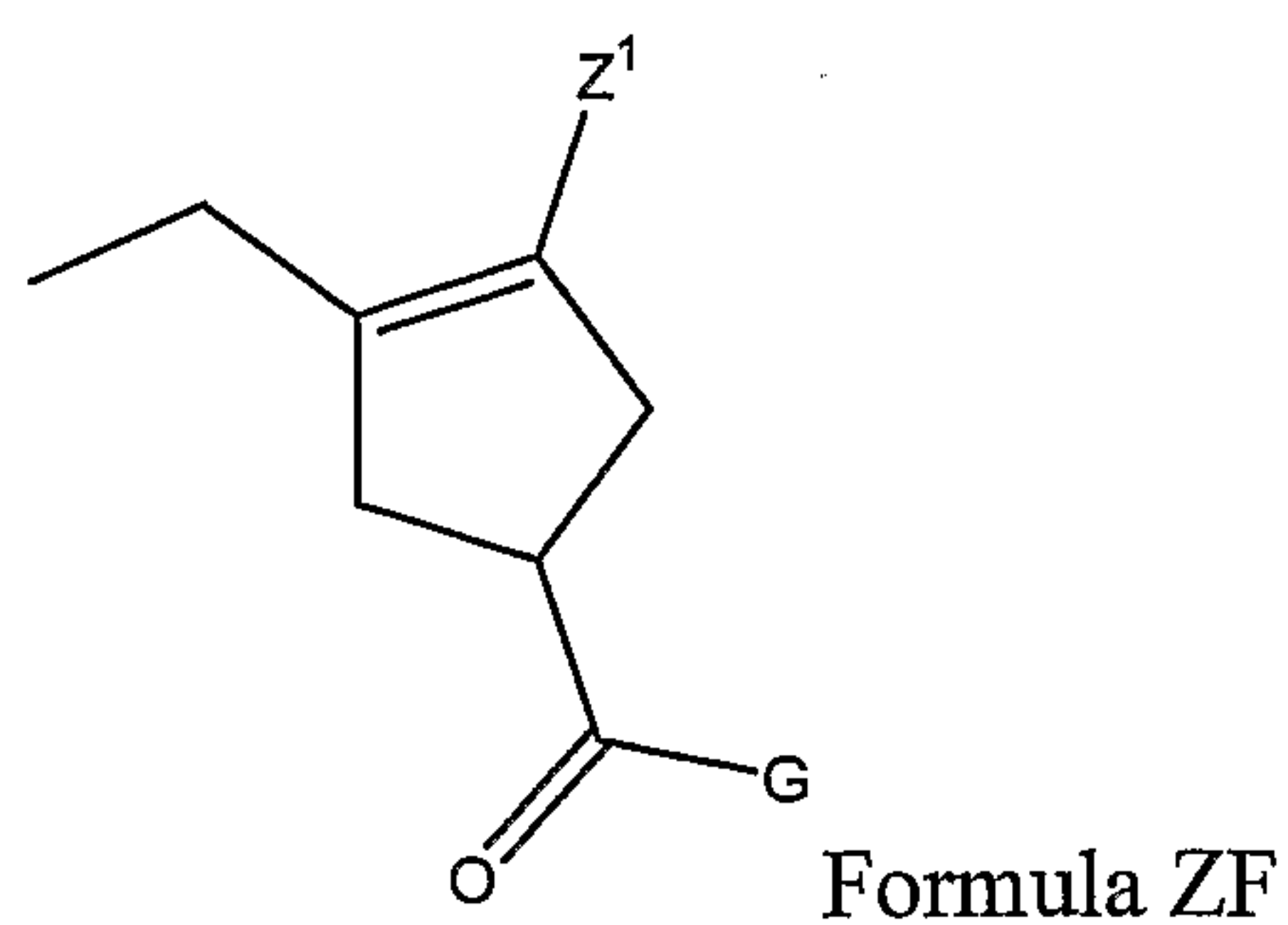
wherein:

Z⁶ is hydrogen, alkyl, alkoxy, -COOH, -NH₂, azido, or halo;

Z⁷ is hydrogen, alkyl, alkoxy, or halo; and

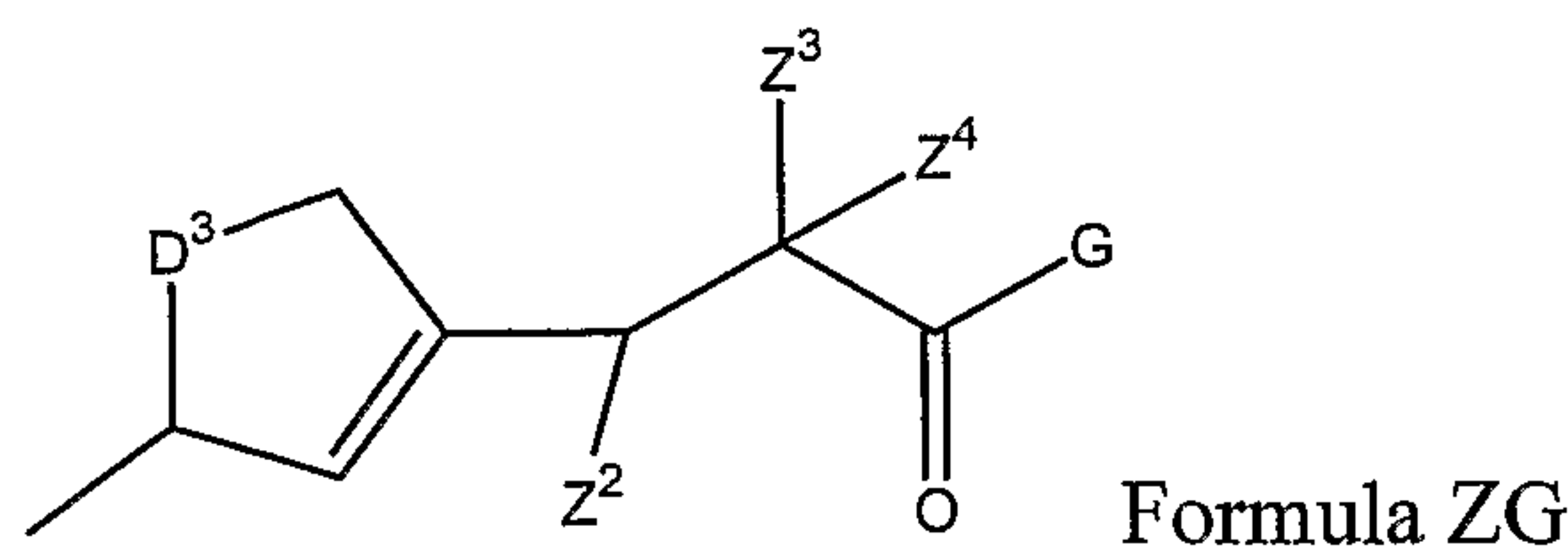
Z⁵ and G are as defined above; or

15



wherein:

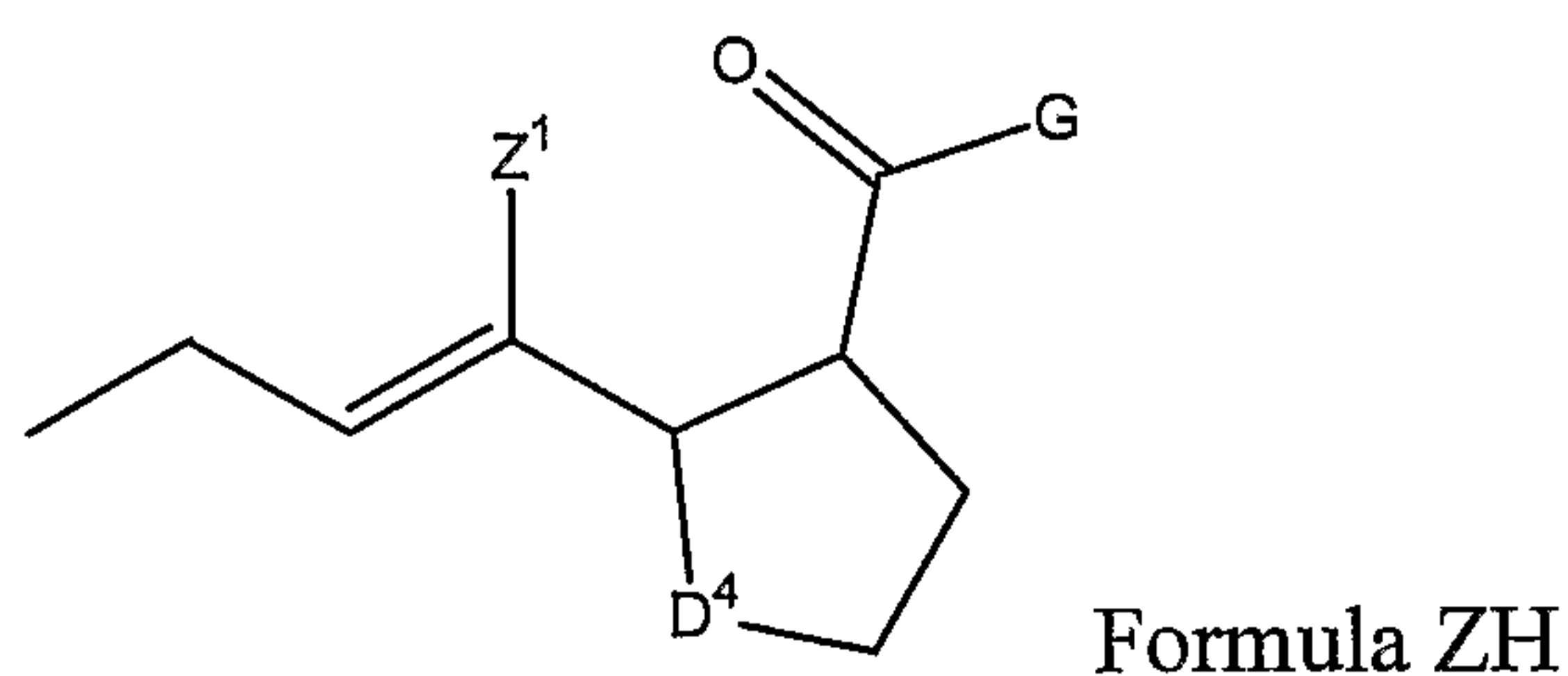
Z^1 and G are as defined above; or



5

wherein:

D^3 , Z^2 , Z^3 , Z^4 , and G are as defined above; or



wherein:

10 D^4 is $-\text{CH}_2-$, $-\text{CH}_2\text{CH}_2-$, $-\text{CF}_2\text{CH}_2-$, $-\text{CH}_2\text{CF}_2-$, $-\text{CF}_2\text{CF}_2-$,
 $-\text{CH}_2\text{CH}_2\text{CH}_2-$, $-\text{CF}_2\text{CH}_2\text{CH}_2-$, $-\text{CH}_2\text{CF}_2\text{CH}_2-$, $-\text{CH}_2\text{CH}_2\text{CF}_2-$,
 $-\text{CF}_2\text{CF}_2\text{CF}_2-$, $-\text{O}-$, $-\text{OCH}_2-$, $-\text{OCF}_2-$, $-\text{CH}_2\text{CHF}-$, $-\text{CHF}-\text{CHF}-$,
 $-\text{CHF}-\text{CH}_2\text{CH}_2-$, $-\text{CH}_2-\text{CHF}-\text{CH}_2-$, $-\text{CH}_2-\text{CH}_2-\text{CHF}-$, $-\text{CHF}-\text{CHF}-\text{CHF}-$,
 or $-\text{OCHF}-$; and

15 Z^1 and G are as defined above;

Y is O or S;

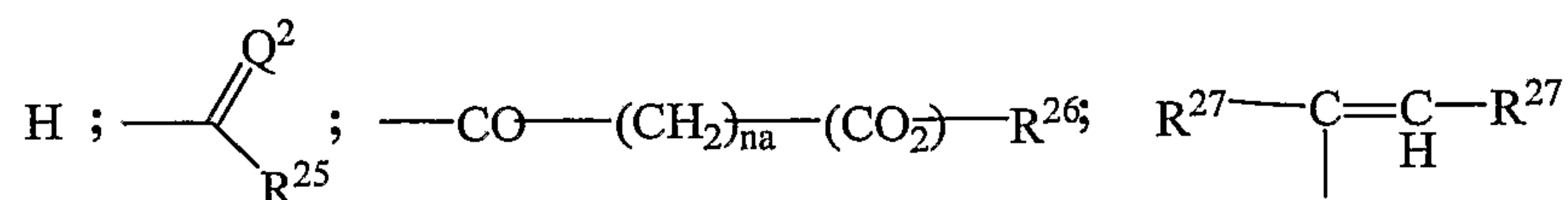
R^{21} is H or alkyl;

R^{22} is H, alkyl, aryl, or substituted aryl; and

R^{23} is methoxy, vinyl, or ethyl;

20 Q^1 is oxygen or sulfur;

R^{24} is selected from the following:



in which

Q^2 is oxygen or sulfur;

R^{25} is aryl, substituted aryl, arylalkyl, substituted arylalkyl, alkyl, substituted alkyl or $-NR^{28}R^{29}$, where R^{28} and R^{29} are independently H, alkyl, substituted alkyl, aryl, substituted aryl, or cycloalkyl;

na is an integer from 0-6;

5 R^{26} is H or alkyl;

each R^{27} is independently H or $-\text{CO}_2 R^{26}$; and

R^{30} is methoxy, ethyl, or vinyl;

provided Q^1 is not oxygen when R^{24} is H;

R^{31} is methoxy, ethyl, or vinyl;

10 R^{32} and R^{33} are each independently H, or alkyl;

each Q^3 is independently O or S; and

nb is an integer of 1-6;

R^{34} is methoxy, ethyl, or vinyl;

R^{35} is hydrogen or alkyl;

15 R^{36} is hydrogen, alkyl, $-\text{C}(\text{O})R^{39}$, $-\text{C}(\text{O})\text{NR}^{40}R^{41}$, $-\text{CO}_2R^{42}$, or $-\text{SO}_2R^{39}$;

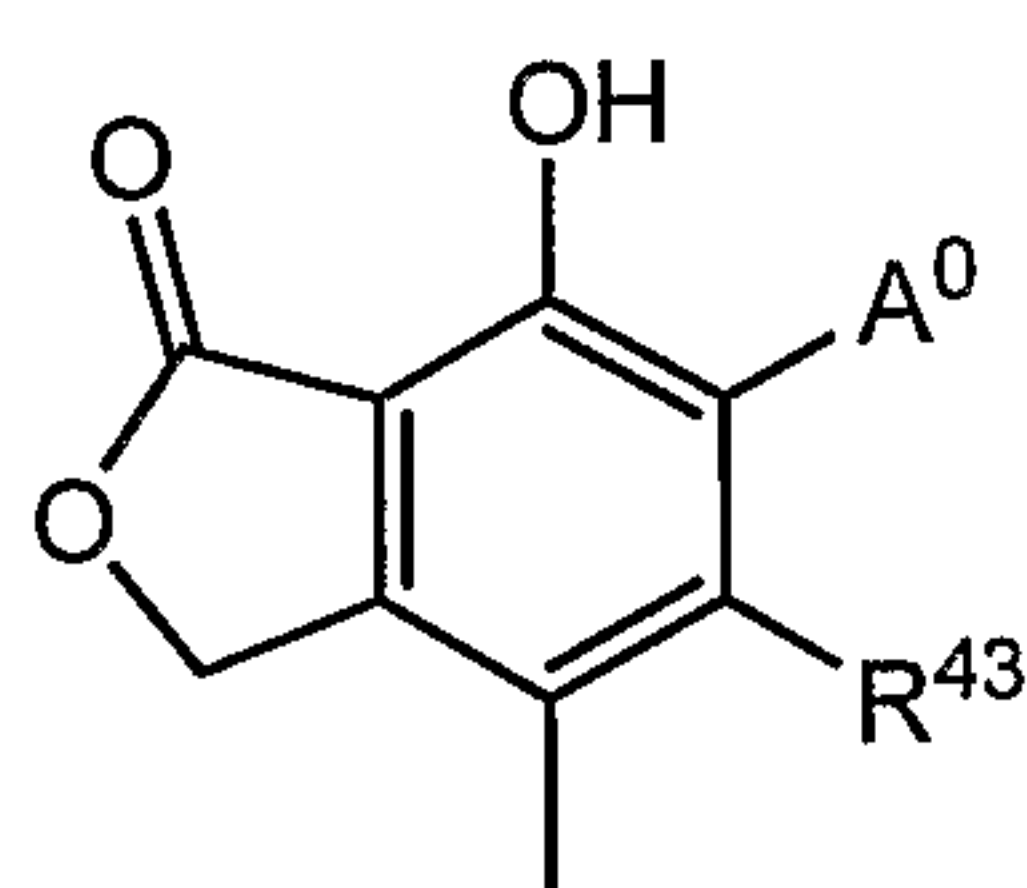
R^{39} is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl;

R^{40} is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl;

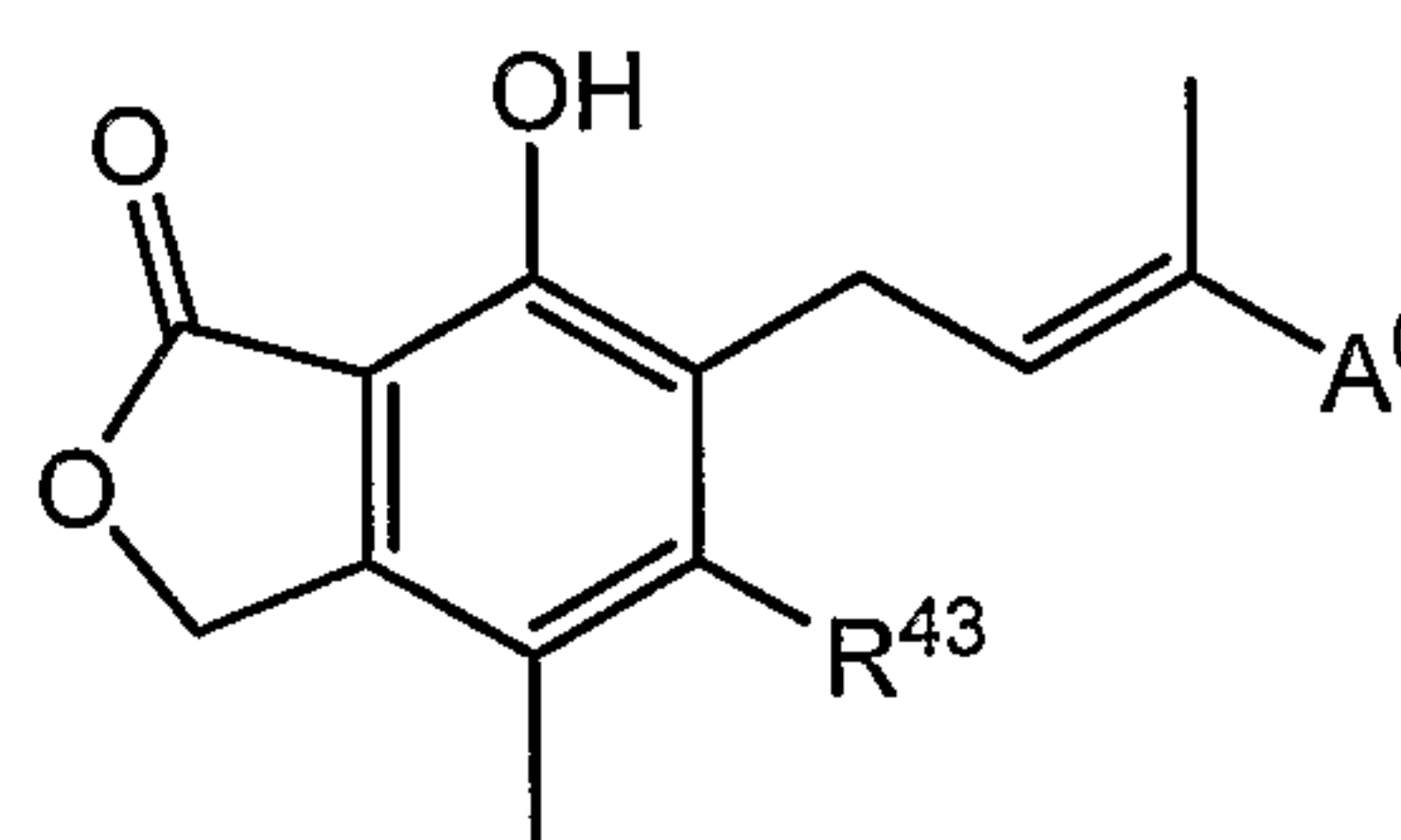
R^{41} is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl; and

R^{42} is alkyl, substituted alkyl, aryl, or substituted aryl.

20 One embodiment of the invention provides a compound of formula XIII or XIV:



XIII

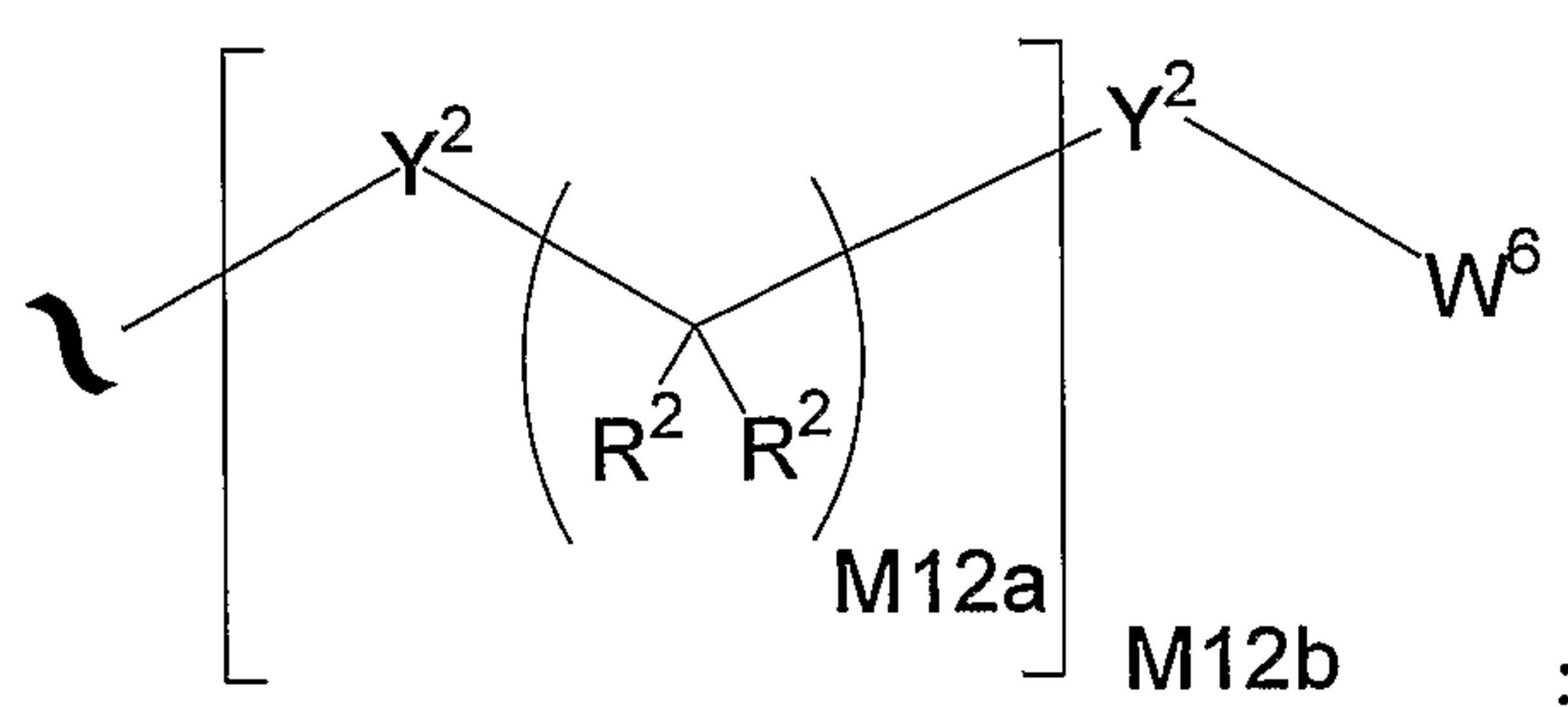


XIV

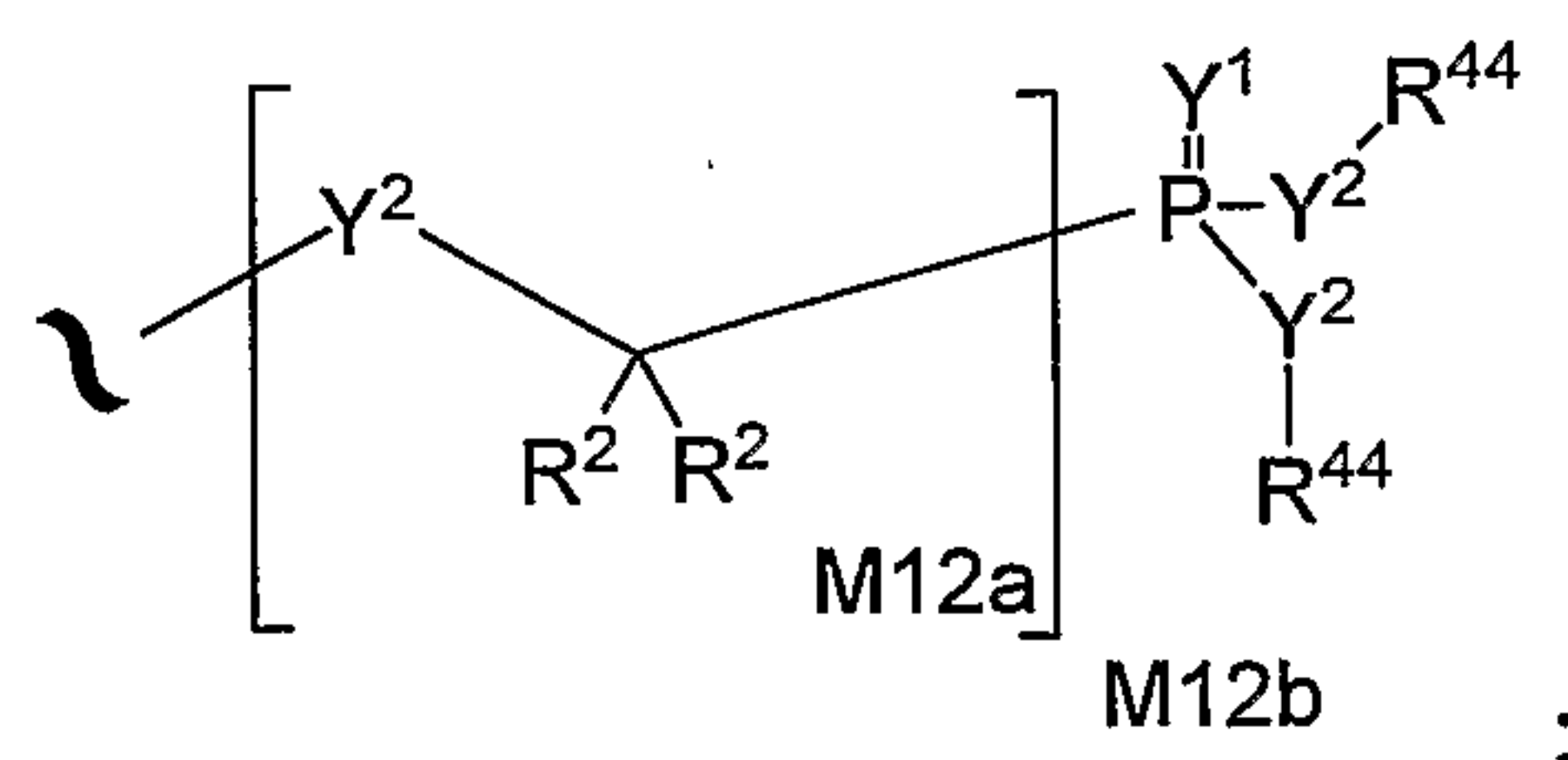
wherein:

A^0 is A^1 ;

25 A^1 is:



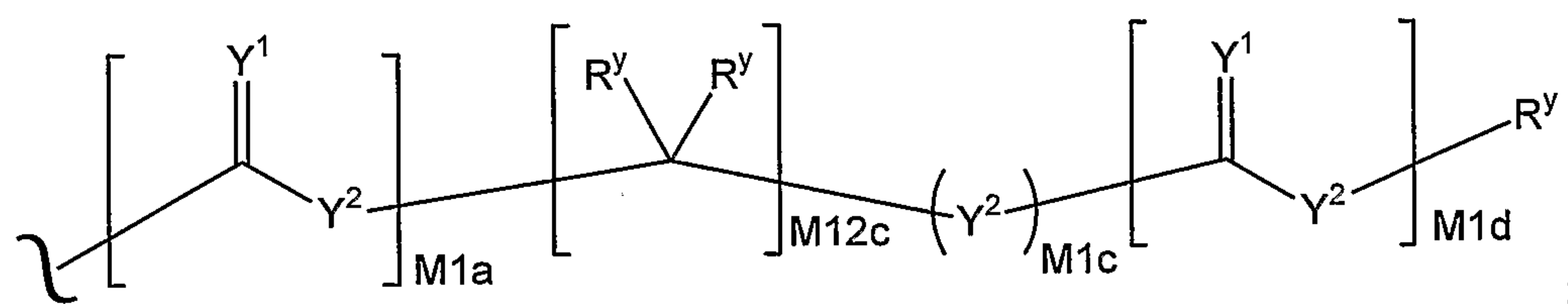
A³ is:



Y¹ is independently O, S, N(R^x), N(OR^x), or N(N(R^x)(R^x));

5 Y² is independently a bond, O, N(R^x), N(OR^x), N(N(R^x)(R^x)), or -S(O)_{M2}-; and when Y² joins two phosphorous atoms Y² can also be C(R²)(R²);

R^x is independently H, R², W³, a protecting group, or the formula:



R^y is independently H, W³, R² or a protecting group;

10 R² is independently H, R³ or R⁴ wherein each R⁴ is independently substituted with 0 to 3 R³ groups;

R³ is R^{3a}, R^{3b}, R^{3c} or R^{3d}, provided that when R³ is bound to a heteroatom, then R³ is R^{3c} or R^{3d};

R^{3a} is F, Cl, Br, I, -CN, N₃ or -NO₂;

15 R^{3b} is Y¹;

R^{3c} is -R^x, -N(R^x)(R^x), -SR^x, -S(O)R^x, -S(O)₂R^x, -S(O)(OR^x), -S(O)₂(OR^x), -OC(Y¹)R^x, -OC(Y¹)OR^x, -OC(Y¹)(N(R^x)(R^x)), -SC(Y¹)R^x, -SC(Y¹)OR^x, -SC(Y¹)(N(R^x)(R^x)), -N(R^x)C(Y¹)R^x, -N(R^x)C(Y¹)OR^x, or -N(R^x)C(Y¹)(N(R^x)(R^x));

20 R^{3d} is -C(Y¹)R^x, -C(Y¹)OR^x or -C(Y¹)(N(R^x)(R^x));

R⁴ is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms, or alkynyl of 2 to 18 carbon atoms;

R^5 is R^4 wherein each R^4 is substituted with 0 to 3 R^3 groups;

W^3 is W^4 or W^5 ;

W^4 is R^5 , $-C(Y^1)R^5$, $-C(Y^1)W^5$, $-SO_2R^5$, or $-SO_2W^5$;

W^5 is carbocycle or heterocycle wherein W^5 is independently substituted

5 with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

$M2$ is 0, 1 or 2;

$M12a$ is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

$M12b$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

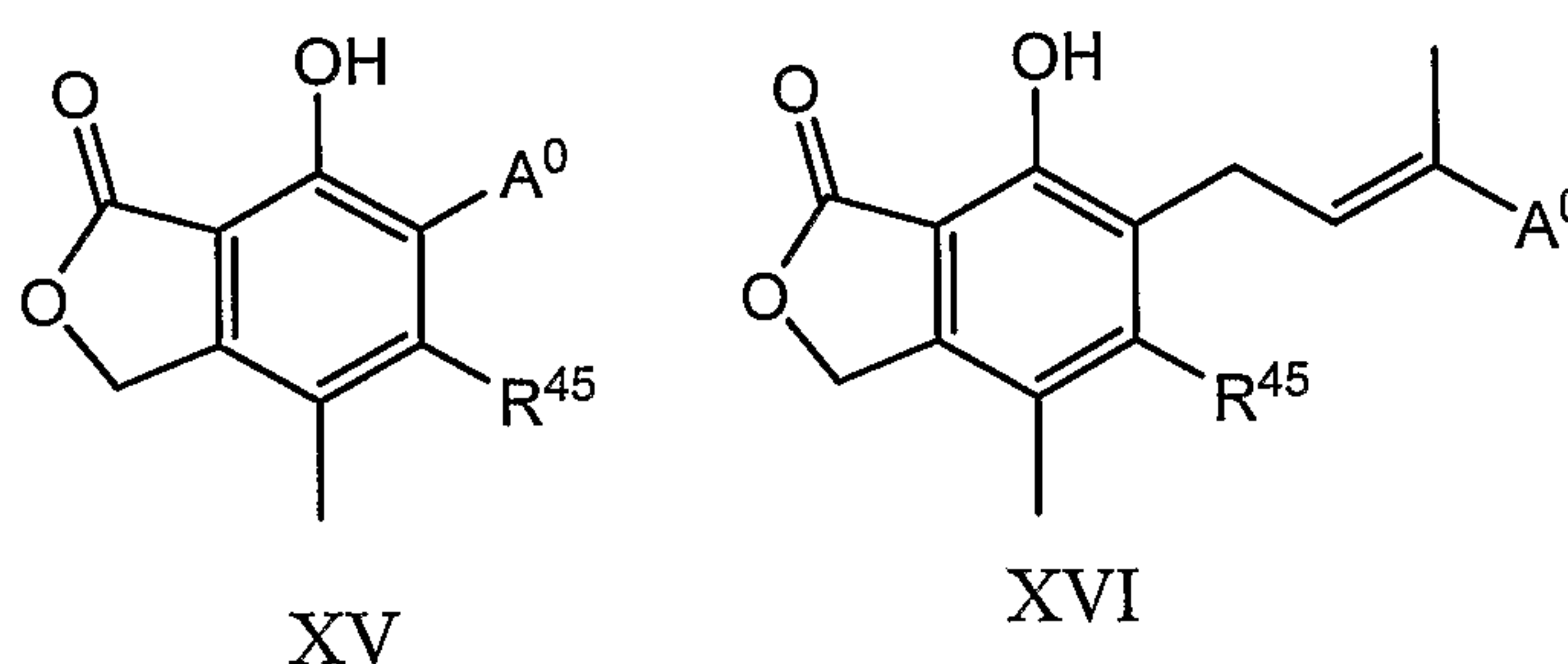
10 $M1a$, $M1c$, and $M1d$ are independently 0 or 1;

$M12c$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

R^{43} is ethyl, methoxy, or vinyl; and

R^{44} is R^X or $-N=C(R^4)(R^4)$, provided that at least one $-Y^2-R^{44}$ of A^3 is $-O-N=C(R^4)(R^4)$.

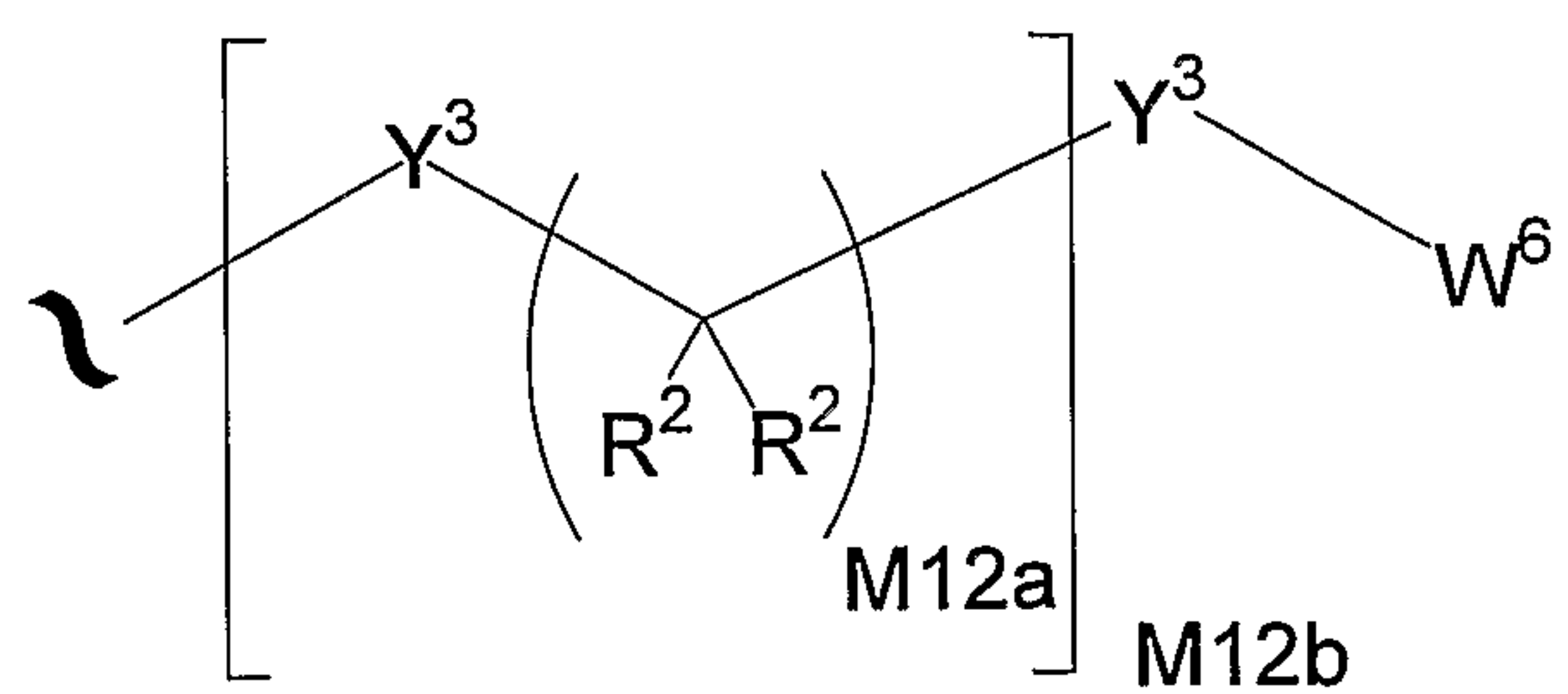
15 One embodiment of the invention provides a compound of formula XV or XVI:



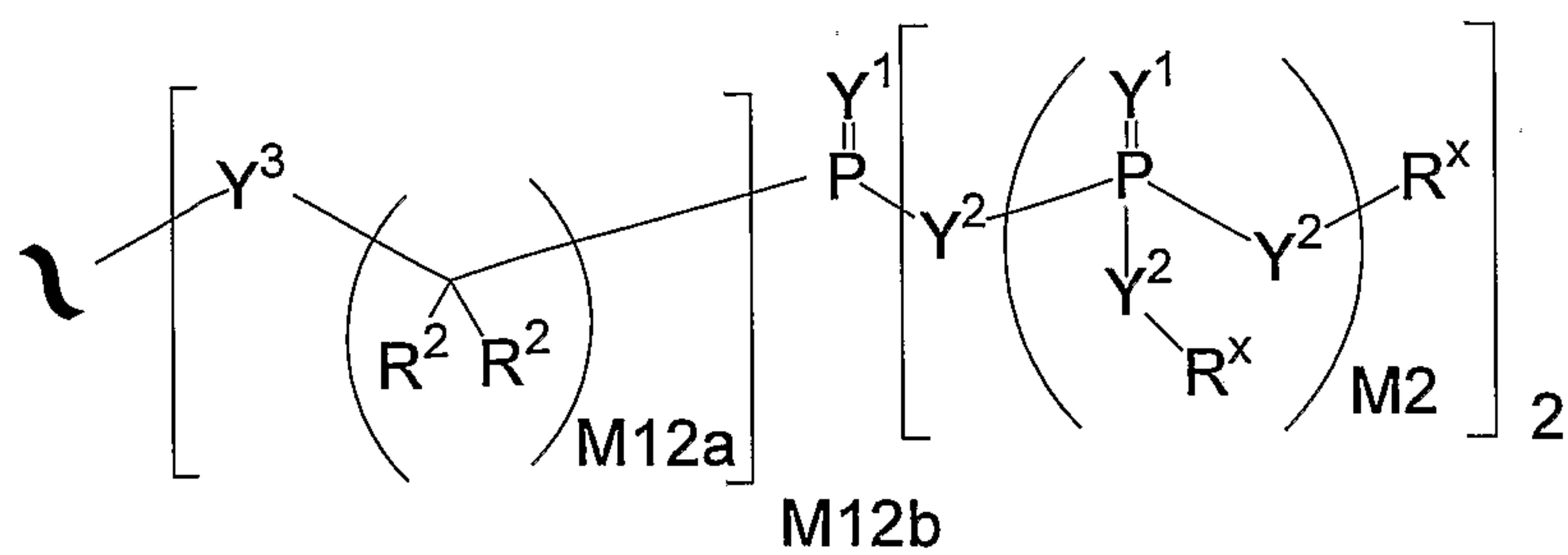
wherein:

A^0 is A^1 ;

20 A^1 is:



A^3 is:

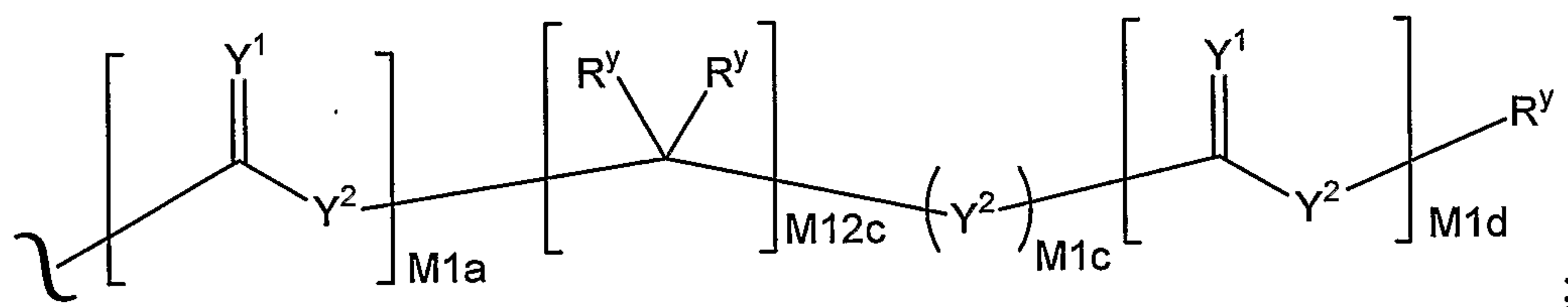


Y^1 is independently O, S, $N(R^x)$, $N(OR^x)$, or $N(N(R^x)(R^x))$;

Y^2 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, or $-S(O)_{M2}-$; and when Y^2 joins two phosphorous atoms Y^2 can also be $C(R^2)(R^2)$;

5 Y^3 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, $-S(O)_{M2}-$, or $-S(O)_{M2}-S(O)_{M2}-$, $-O(C=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)$, $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$; wherein A^1 comprises at least one Y^3 that is $-O(C=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)$, $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$;

10 R^x is independently H, R^2 , W^3 , a protecting group, or the formula:



R^y is independently H, W^3 , R^2 or a protecting group;

R^2 is independently H, R^3 or R^4 wherein each R^4 is independently substituted with 0 to 3 R^3 groups;

15 R^3 is R^{3a} , R^{3b} , R^{3c} or R^{3d} , provided that when R^3 is bound to a heteroatom, then R^3 is R^{3c} or R^{3d} ;

R^{3a} is F, Cl, Br, I, -CN, N_3 or $-NO_2$;

R^{3b} is Y^1 ;

20 R^{3c} is $-R^x$, $-N(R^x)(R^x)$, $-SR^x$, $-S(O)R^x$, $-S(O)_2R^x$, $-S(O)(OR^x)$, $-S(O)_2(OR^x)$, $-OC(Y^1)R^x$, $-OC(Y^1)OR^x$, $-OC(Y^1)(N(R^x)(R^x))$, $-SC(Y^1)R^x$, $-SC(Y^1)OR^x$, $-SC(Y^1)(N(R^x)(R^x))$, $-N(R^x)C(Y^1)R^x$, $-N(R^x)C(Y^1)OR^x$, or $-N(R^x)C(Y^1)(N(R^x)(R^x))$;

R^{3d} is $-C(Y^1)R^x$, $-C(Y^1)OR^x$ or $-C(Y^1)(N(R^x)(R^x))$;

25 R^4 is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms, or alkynyl of 2 to 18 carbon atoms;

R^5 is R^4 wherein each R^4 is substituted with 0 to 3 R^3 groups;

W^3 is W^4 or W^5 ;

W^4 is R^5 , $-C(Y^1)R^5$, $-C(Y^1)W^5$, $-SO_2R^5$, or $-SO_2W^5$;

5 W^5 is carbocycle or heterocycle wherein W^5 is independently substituted with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

M2 is 0, 1 or 2;

M12a is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

M12b is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

10 M1a, M1c, and M1d are independently 0 or 1;

M12c is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12; and

R^{45} is ethyl, methoxy, or vinyl.

The invention provides a pharmaceutical composition comprising an effective amount of a compound of the invention, or a pharmaceutically acceptable salt or solvate thereof, in combination with a pharmaceutically acceptable excipient.

This invention pertains to a method of increasing cellular accumulation and retention of drug compounds, thus improving their therapeutic and diagnostic value, comprising linking the compound to one or more (*e.g.* 1, 2, 3, or 4) phosphonate groups.

The invention also provides a method of inhibiting tumor growth, viral infection, inflammation, and/or tissue/organ transplant rejection, comprising administering to a mammal afflicted with one of the above disorders with an amount of a compound of the invention effective to inhibit the growth of said tumor cells, to inhibit growth of said virus, to inhibit inflammation, or to suppress the immune systems response to said transplanted tissue or organ.

The invention also provides a compound of the invention for use in medical therapy (preferably for use in treating cancer, *e.g.* solid tumors), as well as the use of a compound of the invention for the manufacture of a medicament useful for the treatment of cancer, *e.g.* solid tumors, for the treatment of viral infections, for the treatment of inflammation, or for immunosuppression.

The invention also provides processes and novel intermediates disclosed

herein which are useful for preparing compounds of the invention. Some of the compounds of the invention are useful to prepare other compound of the invention.

5 In another aspect of the invention, the tumor growth, viral infection, inflammation and/or tissue/organ transplant rejection is inhibited by a method comprising the step of treating a sample suspected of containing a tumor or a virus, or suspected of being inflamed or rejected, with a compound or composition of the invention.

DETAILED DESCRIPTION OF EXEMPLARY CLAIMS

10 Reference will now be made in detail to certain claims of the invention, examples of which are illustrated in the accompanying structures and formulas. While the invention will be described in conjunction with the enumerated claims, it will be understood that they are not intended to limit the invention to those claims. On the contrary, the invention is intended to cover all alternatives, 15 modifications, and equivalents, which may be included within the scope of the present invention as defined by the claims.

Definitions

Unless stated otherwise, the following terms and phrases as used herein are intended to have the following meanings:

20 When tradenames are used herein, applicants intend to independently include the tradename product and the active pharmaceutical ingredient(s) of the tradename product.

“Bioavailability” is the degree to which the pharmaceutically active agent becomes available to the target tissue after the agent's introduction into the body. 25 Enhancement of the bioavailability of a pharmaceutically active agent can provide a more efficient and effective treatment for patients because, for a given dose, more of the pharmaceutically active agent will be available at the targeted tissue sites.

The terms “phosphonate” and “phosphonate group” include functional 30 groups or moieties within a molecule that comprises a phosphorous that is 1) single-bonded to a carbon, 2) double-bonded to a heteroatom , 3) single-bonded to a heteroatom, and 4) single-bonded to another heteroatom, wherein each heteroatom can be the same or different. The terms “phosphonate” and

“phosphonate group” also include functional groups or moieties that comprise a phosphorous in the same oxidation state as the phosphorous described above, as well as functional groups or moieties that comprise a prodrug moiety that can separate from a compound so that the compound retains a phosphorous having the characteristics described above. For example, the terms “phosphonate” and “phosphonate group” include phosphonic acid, phosphonic monoester, phosphonic diester, phosphoramidate, and phosphonthioate functional groups. In one specific embodiment of the invention, the terms “phosphonate” and “phosphonate group” include functional groups or moieties within a molecule that comprises a phosphorous that is 1) single-bonded to a carbon, 2) double-bonded to an oxygen, 3) single-bonded to an oxygen, and 4) single-bonded to another oxygen, as well as functional groups or moieties that comprise a prodrug moiety that can separate from a compound so that the compound retains a phosphorous having such characteristics. In another specific embodiment of the invention, the terms “phosphonate” and “phosphonate group” include functional groups or moieties within a molecule that comprises a phosphorous that is 1) single-bonded to a carbon, 2) double-bonded to an oxygen, 3) single-bonded to an oxygen or nitrogen, and 4) single-bonded to another oxygen or nitrogen, as well as functional groups or moieties that comprise a prodrug moiety that can separate from a compound so that the compound retains a phosphorous having such characteristics.

The term “prodrug” as used herein refers to any compound that when administered to a biological system generates the drug substance, *i.e.* active ingredient, as a result of spontaneous chemical reaction(s), enzyme catalyzed chemical reaction(s), photolysis, and/or metabolic chemical reaction(s). A prodrug is thus a covalently modified analog or latent form of a therapeutically-active compound.

“Prodrug moiety” refers to a labile functional group that separates from the active inhibitory compound during metabolism, systemically, inside a cell, by hydrolysis, enzymatic cleavage, or by some other process (Bundgaard, Hans, “Design and Application of Prodrugs” in A Textbook of Drug Design and Development (1991), P. Krogsgaard-Larsen and H. Bundgaard, Eds. Harwood Academic Publishers, pp. 113-191). Enzymes that are capable of an enzymatic

activation mechanism with the phosphonate prodrug compounds of the invention include, but are not limited to, amidases, esterases, microbial enzymes, phospholipases, cholinesterases, and phosphatases. Prodrug moieties can serve to enhance solubility, absorption and lipophilicity to optimize drug delivery, bioavailability and efficacy. A prodrug moiety may include an active metabolite or drug itself.

Exemplary prodrug moieties include the hydrolytically sensitive or labile acyloxymethyl esters $-\text{CH}_2\text{OC}(=\text{O})\text{R}^9$ and acyloxymethyl carbonates $-\text{CH}_2\text{OC}(=\text{O})\text{OR}^9$ where R^9 is C_1 – C_6 alkyl, C_1 – C_6 substituted alkyl, C_6 – C_{20} aryl or C_6 – C_{20} substituted aryl. The acyloxyalkyl ester was first used as a prodrug strategy for carboxylic acids and then applied to phosphates and phosphonates by Farquhar *et al.* (1983) *J. Pharm. Sci.* 72: 324; also US Patent Nos. 4816570, 4968788, 5663159 and 5792756. Subsequently, the acyloxyalkyl ester was used to deliver phosphonic acids across cell membranes and to enhance oral bioavailability. A close variant of the acyloxyalkyl ester, the alkoxycarbonyloxyalkyl ester (carbonate), may also enhance oral bioavailability as a prodrug moiety in the compounds of the combinations of the invention. An exemplary acyloxymethyl ester is pivaloyloxymethoxy, (POM) $-\text{CH}_2\text{OC}(=\text{O})\text{C}(\text{CH}_3)_3$. An exemplary acyloxymethyl carbonate prodrug moiety is pivaloyloxymethylcarbonate (POC) $-\text{CH}_2\text{OC}(=\text{O})\text{OC}(\text{CH}_3)_3$.

The phosphonate group may be a phosphonate prodrug moiety. The prodrug moiety may be sensitive to hydrolysis, such as, but not limited to a pivaloyloxymethyl carbonate (POC) or POM group. Alternatively, the prodrug moiety may be sensitive to enzymatic potentiated cleavage, such as a lactate ester or a phosphoramidate-ester group.

Aryl esters of phosphorus groups, especially phenyl esters, are reported to enhance oral bioavailability (De Lombaert *et al.* (1994) *J. Med. Chem.* 37: 498). Phenyl esters containing a carboxylic ester ortho to the phosphate have also been described (Khamnei and Torrence, (1996) *J. Med. Chem.* 39:4109-4115). Benzyl esters are reported to generate the parent phosphonic acid. In some cases, substituents at the *ortho*-or *para*-position may accelerate the hydrolysis. Benzyl analogs with an acylated phenol or an alkylated phenol may

generate the phenolic compound through the action of enzymes, *e.g.*, esterases, oxidases, *etc.*, which in turn undergoes cleavage at the benzylic C–O bond to generate the phosphoric acid and the quinone methide intermediate. Examples of this class of prodrugs are described by Mitchell *et al.* (1992) *J. Chem. Soc. Perkin Trans. II* 2345; Glazier WO 91/19721. Still other benzylic prodrugs have been described containing a carboxylic ester-containing group attached to the benzylic methylene (Glazier WO 91/19721). Thio-containing prodrugs are reported to be useful for the intracellular delivery of phosphonate drugs. These proesters contain an ethylthio group in which the thiol group is either esterified with an acyl group or combined with another thiol group to form a disulfide. Deesterification or reduction of the disulfide generates the free thio intermediate that subsequently breaks down to the phosphoric acid and episulfide (Puech *et al.* (1993) *Antiviral Res.*, 22: 155-174; Benzaria *et al.* (1996) *J. Med. Chem.* 39: 4958). Cyclic phosphonate esters have also been described as prodrugs of phosphorus-containing compounds (Erion *et al.*, US Patent No. 6312662).

“Protecting group” refers to a moiety of a compound that masks or alters the properties of a functional group or the properties of the compound as a whole. Chemical protecting groups and strategies for protection/deprotection are well known in the art. *See e.g.*, Protective Groups in Organic Chemistry, Theodora W. Greene, John Wiley & Sons, Inc., New York, 1991. Protecting groups are often utilized to mask the reactivity of certain functional groups, to assist in the efficiency of desired chemical reactions, *e.g.*, making and breaking chemical bonds in an ordered and planned fashion. Protection of functional groups of a compound alters other physical properties besides the reactivity of the protected functional group, such as the polarity, lipophilicity (hydrophobicity), and other properties that can be measured by common analytical tools. Chemically protected intermediates may themselves be biologically active or inactive.

Protected compounds may also exhibit altered, and in some cases, optimized properties *in vitro* and *in vivo*, such as passage through cellular membranes and resistance to enzymatic degradation or sequestration. In this role, protected compounds with intended therapeutic effects may be referred to

as prodrugs. Another function of a protecting group is to convert the parental drug into a prodrug, whereby the parental drug is released upon conversion of the prodrug *in vivo*. Because active prodrugs may be absorbed more effectively than the parental drug, prodrugs may possess greater potency *in vivo* than the parental drug. Protecting groups are removed either *in vitro*, in the instance of chemical intermediates, or *in vivo*, in the case of prodrugs. With chemical intermediates, it is not particularly important that the resulting products after deprotection, *e.g.*, alcohols, be physiologically acceptable, although in general it is more desirable if the products are pharmacologically innocuous.

Any reference to any of the compounds of the invention also includes a reference to a physiologically acceptable salt thereof. Examples of physiologically acceptable salts of the compounds of the invention include salts derived from an appropriate base, such as an alkali metal (for example, sodium), an alkaline earth (for example, magnesium), ammonium and NX_4^+ (wherein X is $\text{C}_1\text{--C}_4$ alkyl). Physiologically acceptable salts of an hydrogen atom or an amino group include salts of organic carboxylic acids such as acetic, benzoic, lactic, fumaric, tartaric, maleic, malonic, malic, isethionic, lactobionic and succinic acids; organic sulfonic acids, such as methanesulfonic, ethanesulfonic, benzenesulfonic and p-toluenesulfonic acids; and inorganic acids, such as hydrochloric, sulfuric, phosphoric and sulfamic acids. Physiologically acceptable salts of a compound of an hydroxy group include the anion of said compound in combination with a suitable cation such as Na^+ and NX_4^+ (wherein X is independently selected from H or a $\text{C}_1\text{--C}_4$ alkyl group).

For therapeutic use, salts of active ingredients of the compounds of the invention will be physiologically acceptable, *i.e.* they will be salts derived from a physiologically acceptable acid or base. However, salts of acids or bases which are not physiologically acceptable may also find use, for example, in the preparation or purification of a physiologically acceptable compound. All salts, whether or not derived from a physiologically acceptable acid or base, are within the scope of the present invention.

“Alkyl” is $\text{C}_1\text{--C}_{18}$ hydrocarbon containing normal, secondary, tertiary or cyclic carbon atoms. Examples are methyl (Me, $-\text{CH}_3$), ethyl (Et, $-\text{CH}_2\text{CH}_3$), 1-

propyl (n-Pr, n-propyl, $-\text{CH}_2\text{CH}_2\text{CH}_3$), 2-propyl (i-Pr, i-propyl, $-\text{CH}(\text{CH}_3)_2$), 1-butyl (n-Bu, n-butyl, $-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 2-methyl-1-propyl (i-Bu, i-butyl, $-\text{CH}_2\text{CH}(\text{CH}_3)_2$), 2-butyl (s-Bu, s-butyl, $-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$), 2-methyl-2-propyl (t-Bu, t-butyl, $-\text{C}(\text{CH}_3)_3$), 1-pentyl (n-pentyl, $-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 2-pentyl ($-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_2\text{CH}_3$), 3-pentyl ($-\text{CH}(\text{CH}_2\text{CH}_3)_2$), 2-methyl-2-butyl ($-\text{C}(\text{CH}_3)_2\text{CH}_2\text{CH}_3$), 3-methyl-2-butyl ($-\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)_2$), 3-methyl-1-butyl ($-\text{CH}_2\text{CH}_2\text{CH}(\text{CH}_3)_2$), 2-methyl-1-butyl ($-\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$), 1-hexyl ($-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 2-hexyl ($-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$), 3-hexyl ($-\text{CH}(\text{CH}_2\text{CH}_3)(\text{CH}_2\text{CH}_2\text{CH}_3)$), 2-methyl-2-pentyl ($-\text{C}(\text{CH}_3)_2\text{CH}_2\text{CH}_2\text{CH}_3$), 3-methyl-2-pentyl ($-\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$), 4-methyl-2-pentyl ($-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}(\text{CH}_3)_2$), 3-methyl-3-pentyl ($-\text{C}(\text{CH}_3)(\text{CH}_2\text{CH}_3)_2$), 2-methyl-3-pentyl ($-\text{CH}(\text{CH}_2\text{CH}_3)\text{CH}(\text{CH}_3)_2$), 2,3-dimethyl-2-butyl ($-\text{C}(\text{CH}_3)_2\text{CH}(\text{CH}_3)_2$), 3,3-dimethyl-2-butyl ($-\text{CH}(\text{CH}_3)\text{C}(\text{CH}_3)_3$).

“Alkenyl” is $\text{C}_2\text{-C}_{18}$ hydrocarbon containing normal, secondary, tertiary or cyclic carbon atoms with at least one site of unsaturation, *i.e.* a carbon-carbon, sp^2 double bond. Examples include, but are not limited to, ethylene or vinyl ($-\text{CH}=\text{CH}_2$), allyl ($-\text{CH}_2\text{CH}=\text{CH}_2$), cyclopentenyl ($-\text{C}_5\text{H}_7$), and 5-hexenyl ($-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}=\text{CH}_2$).

“Alkynyl” is $\text{C}_2\text{-C}_{18}$ hydrocarbon containing normal, secondary, tertiary or cyclic carbon atoms with at least one site of unsaturation, *i.e.* a carbon-carbon, sp triple bond. Examples include, but are not limited to, acetylenic ($-\text{C}\equiv\text{CH}$) and propargyl ($-\text{CH}_2\text{C}\equiv\text{CH}$),

“Alkylene” refers to a saturated, branched or straight chain or cyclic hydrocarbon radical of 1-18 carbon atoms, and having two monovalent radical centers derived by the removal of two hydrogen atoms from the same or two different carbon atoms of a parent alkane. Typical alkylene radicals include, but are not limited to, methylene ($-\text{CH}_2-$), 1,2-ethyl ($-\text{CH}_2\text{CH}_2-$), 1,3-propyl ($-\text{CH}_2\text{CH}_2\text{CH}_2-$), 1,4-butyl ($-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2-$), and the like.

“Alkenylene” refers to an unsaturated, branched or straight chain or cyclic hydrocarbon radical of 2-18 carbon atoms, and having two monovalent radical centers derived by the removal of two hydrogen atoms from the same or two

different carbon atoms of a parent alkene. Typical alkenylene radicals include, but are not limited to, 1,2-ethylene (-CH=CH-).

“Alkynylene” refers to an unsaturated, branched or straight chain or cyclic hydrocarbon radical of 2-18 carbon atoms, and having two monovalent radical
5 centers derived by the removal of two hydrogen atoms from the same or two different carbon atoms of a parent alkyne. Typical alkynylene radicals include, but are not limited to, acetylene (-C≡C-), propargyl (-CH₂C≡C-), and 4-pentynyl (-CH₂CH₂CH₂C≡CH-).

“Aryl” means a monovalent aromatic hydrocarbon radical of 6-20 carbon
10 atoms derived by the removal of one hydrogen atom from a single carbon atom of a parent aromatic ring system. Typical aryl groups include, but are not limited to, radicals derived from benzene, substituted benzene, naphthalene, anthracene, biphenyl, and the like.

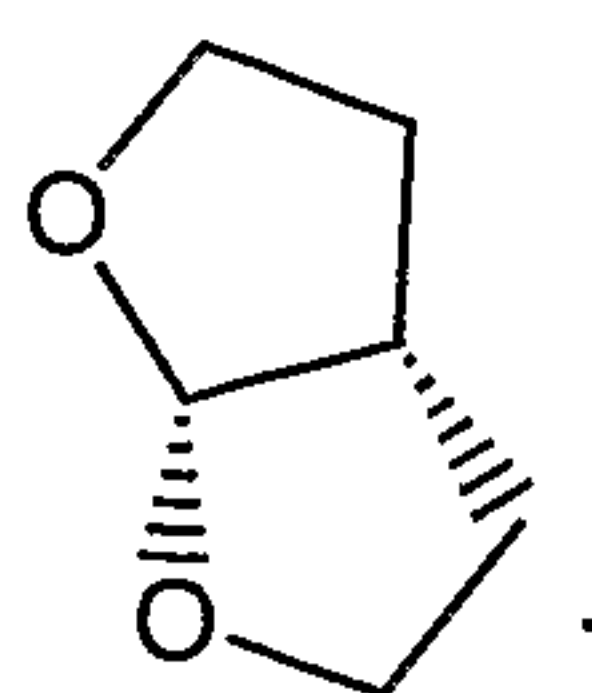
“Arylalkyl” refers to an acyclic alkyl radical in which one of the
15 hydrogen atoms bonded to a carbon atom, typically a terminal or sp³ carbon atom, is replaced with an aryl radical. Typical arylalkyl groups include, but are not limited to, benzyl, 2-phenylethan-1-yl, , naphthylmethyl, 2-naphthylethan-1-yl, naphthobenzyl, 2-naphthophenylethan-1-yl and the like. The arylalkyl group comprises 6 to 20 carbon atoms, *e.g.*, the alkyl moiety, including alkanyl,
20 alkenyl or alkynyl groups, of the arylalkyl group is 1 to 6 carbon atoms and the aryl moiety is 5 to 14 carbon atoms.

“Substituted alkyl,” “substituted aryl,” and “substituted arylalkyl” mean alkyl, aryl, and arylalkyl respectively, in which one or more hydrogen atoms are each independently replaced with a non-hydrogen substituent. Typical
25 substituents include, but are not limited to, -X, -R, -O⁻, -OR, -SR, -S⁻, -NR₂, -NR₃, =NR, -CX₃, -CN, -OCN, -SCN, -N=C=O, -NCS, -NO, -NO₂, =N₂, -N₃, NC(=O)R, -C(=O)R, -C(=O)NRR, -S(=O)₂O⁻, -S(=O)₂OH, -S(=O)₂R, -OS(=O)₂OR, -S(=O)₂NR, -S(=O)R, -OP(=O)O₂RR, -P(=O)O₂RR, -P(=O)(O⁻)₂, -P(=O)(OH)₂, -C(=O)R, -C(=O)X, -C(S)R, -C(O)OR, -C(O)O⁻, -C(S)OR,
30 -C(O)SR, -C(S)SR, -C(O)NRR, -C(S)NRR, -C(NR)NRR, where each X is independently a halogen: F, Cl, Br, or I; and each R is independently -H, alkyl,

aryl, heterocycle, protecting group or prodrug moiety. Alkylene, alkenylene, and alkynylene groups may also be similarly substituted.

“Heterocycle” as used herein includes by way of example and not limitation these heterocycles described in Paquette, Leo A.; Principles of
5 Modern Heterocyclic Chemistry (W.A. Benjamin, New York, 1968), particularly Chapters 1, 3, 4, 6, 7, and 9; The Chemistry of Heterocyclic Compounds, A Series of Monographs” (John Wiley & Sons, New York, 1950 to present), in particular Volumes 13, 14, 16, 19, and 28; and *J. Am. Chem. Soc.* (1960) 82:5566. In one specific embodiment of the invention “heterocycle” includes a
10 “carbocycle” as defined herein, wherein one or more (*e.g.* 1, 2, 3, or 4) carbon atoms have been replaced with a heteroatom (*e.g.* O, N, or S).

Examples of heterocycles include by way of example and not limitation pyridyl, dihydropyridyl, tetrahydropyridyl (piperidyl), thiazolyl, tetrahydrothiophenyl, sulfur oxidized tetrahydrothiophenyl, pyrimidinyl, furanyl,
15 thienyl, pyrrolyl, pyrazolyl, imidazolyl, tetrazolyl, benzofuranyl, thianaphthalenyl, indolyl, indolenyl, quinoliny, isoquinoliny, benzimidazolyl, piperidinyl, 4-piperidonyl, pyrrolidinyl, 2-pyrrolidonyl, pyrrolinyl, tetrahydrofuranyl, tetrahydroquinoliny, tetrahydroisoquinoliny, decahydroquinoliny, octahydroisoquinoliny, azocinyl, triazinyl, 6H-1,2,5-
20 thiadiazinyl, 2H,6H-1,5,2-dithiazinyl, thienyl, thianthrenyl, pyranyl, isobenzofuranyl, chromenyl, xanthenyl, phenoxathinyl, 2H-pyrrolyl, isothiazolyl, isoxazolyl, pyrazinyl, pyridazinyl, indoliziny, isoindolyl, 3H-indolyl, 1H-indazolyl, purinyl, 4H-quinoliziny, phthalazinyl, naphthyridinyl, quinoxalinyl, quinazolinyl, cinnolinyl, pteridinyl, 4aH-carbazolyl, carbazolyl, β -
25 carbolinyl, phenanthridinyl, acridinyl, pyrimidinyl, phenanthrolinyl, phenazinyl, phenothiazinyl, furazanyl, phenoxazinyl, isochromanyl, chromanyl, imidazolidinyl, imidazoliny, pyrazolidinyl, pyrazoliny, piperazinyl, indoliny, isoindoliny, quinuclidinyl, morpholiny, oxazolidinyl, benzotriazolyl, benzisoxazolyl, oxindolyl, benzoxazoliny, isatinoyl, and bis-tetrahydrofuranyl:



By way of example and not limitation, carbon bonded heterocycles are bonded at position 2, 3, 4, 5, or 6 of a pyridine, position 3, 4, 5, or 6 of a pyridazine, position 2, 4, 5, or 6 of a pyrimidine, position 2, 3, 5, or 6 of a pyrazine, position 2, 3, 4, or 5 of a furan, tetrahydrofuran, thiofuran, thiophene, pyrrole or tetrahydropyrrole, position 2, 4, or 5 of an oxazole, imidazole or thiazole, position 3, 4, or 5 of an isoxazole, pyrazole, or isothiazole, position 2 or 3 of an aziridine, position 2, 3, or 4 of an azetidine, position 2, 3, 4, 5, 6, 7, or 8 of a quinoline or position 1, 3, 4, 5, 6, 7, or 8 of an isoquinoline. Still more typically, carbon bonded heterocycles include 2-pyridyl, 3-pyridyl, 4-pyridyl, 5-pyridyl, 6-pyridyl, 3-pyridazinyl, 4-pyridazinyl, 5-pyridazinyl, 6-pyridazinyl, 2-pyrimidinyl, 4-pyrimidinyl, 5-pyrimidinyl, 6-pyrimidinyl, 2-pyrazinyl, 3-pyrazinyl, 5-pyrazinyl, 6-pyrazinyl, 2-thiazolyl, 4-thiazolyl, or 5-thiazolyl.

By way of example and not limitation, nitrogen bonded heterocycles are bonded at position 1 of an aziridine, azetidine, pyrrole, pyrrolidine, 2-pyrroline, 3-pyrroline, imidazole, imidazolidine, 2-imidazoline, 3-imidazoline, pyrazole, pyrazoline, 2-pyrazoline, 3-pyrazoline, piperidine, piperazine, indole, indoline, 1H-indazole, position 2 of a isoindole, or isoindoline, position 4 of a morpholine, and position 9 of a carbazole, or β -carboline. Still more typically, nitrogen bonded heterocycles include 1-aziridyl, 1-azetidyl, 1-pyrrolyl, 1-imidazolyl, 1-pyrazolyl, and 1-piperidinyl.

“Carbocycle” refers to a saturated, unsaturated or aromatic ring having 3 to 7 carbon atoms as a monocycle, 7 to 12 carbon atoms as a bicycle, and up to about 20 carbon atoms as a polycycle. Monocyclic carbocycles have 3 to 6 ring atoms, still more typically 5 or 6 ring atoms. Bicyclic carbocycles have 7 to 12 ring atoms, *e.g.*, arranged as a bicyclo [4,5], [5,5], [5,6] or [6,6] system, or 9 or 10 ring atoms arranged as a bicyclo [5,6] or [6,6] system. Examples of monocyclic carbocycles include cyclopropyl, cyclobutyl, cyclopentyl, 1-cyclopent-1-enyl, 1-cyclopent-2-enyl, 1-cyclopent-3-enyl, cyclohexyl, 1-cyclohex-1-enyl, 1-cyclohex-2-enyl, 1-cyclohex-3-enyl, phenyl, spiryl and naphthyl.

The term “chiral” refers to molecules which have the property of non-superimposability of the mirror image partner, while the term “achiral” refers to molecules which are superimposable on their mirror image partner.

The term "stereoisomers" refers to compounds which have identical chemical constitution, but differ with regard to the arrangement of the atoms or groups in space.

5 "Diastereomer" refers to a stereoisomer with two or more centers of chirality and whose molecules are not mirror images of one another. Diastereomers have different physical properties, *e.g.*, melting points, boiling points, spectral properties, and reactivities. Mixtures of diastereomers may separate under high resolution analytical procedures such as electrophoresis and chromatography.

10 "Enantiomers" refer to two stereoisomers of a compound which are non-superimposable mirror images of one another.

The term "treatment" or "treating," to the extent it relates to a disease or condition, includes preventing the disease or condition from occurring, inhibiting the disease or condition, eliminating the disease or condition, and/or relieving
15 one or more symptoms of the disease or condition.

Stereochemical definitions and conventions used herein generally follow S. P. Parker, Ed., McGraw-Hill Dictionary of Chemical Terms (1984) McGraw-Hill Book Company, New York; and Eliel, E. and Wilen, S., Stereochemistry of Organic Compounds (1994) John Wiley & Sons, Inc., New York. Many organic
20 compounds exist in optically active forms, *i.e.*, they have the ability to rotate the plane of plane-polarized light. In describing an optically active compound, the prefixes D and L or R and S are used to denote the absolute configuration of the molecule about its chiral center(s). The prefixes d and l or (+) and (-) are employed to designate the sign of rotation of plane-polarized light by the
25 compound, with (-) or l meaning that the compound is levorotatory. A compound prefixed with (+) or d is dextrorotatory. For a given chemical structure, these stereoisomers are identical except that they are mirror images of one another. A specific stereoisomer may also be referred to as an enantiomer, and a mixture of such isomers is often called an enantiomeric mixture. A 50:50
30 mixture of enantiomers is referred to as a racemic mixture or a racemate, which may occur where there has been no stereoselection or stereospecificity in a chemical reaction or process. The terms "racemic mixture" and "racemate" refer to an equimolar mixture of two enantiomeric species, devoid of optical activity.

Protecting Groups

In the context of the present invention, protecting groups include prodrug moieties and chemical protecting groups.

Protecting groups are available, commonly known and used, and are
5 optionally used to prevent side reactions with the protected group during synthetic procedures, *i.e.* routes or methods to prepare the compounds of the invention. For the most part the decision as to which groups to protect, when to do so, and the nature of the chemical protecting group "PG" will be dependent upon the chemistry of the reaction to be protected against (*e.g.*, acidic, basic,
10 oxidative, reductive or other conditions) and the intended direction of the synthesis. The PG groups do not need to be, and generally are not, the same if the compound is substituted with multiple PG. In general, PG will be used to protect functional groups such as carboxyl, hydroxyl, thio, or amino groups and to thus prevent side reactions or to otherwise facilitate the synthetic efficiency.
15 The order of deprotection to yield free, deprotected groups is dependent upon the intended direction of the synthesis and the reaction conditions to be encountered, and may occur in any order as determined by the artisan.

Various functional groups of the compounds of the invention may be protected. For example, protecting groups for -OH groups (whether hydroxyl,
20 carboxylic acid, phosphonic acid, or other functions) include "ether- or ester-forming groups". Ether- or ester-forming groups are capable of functioning as chemical protecting groups in the synthetic schemes set forth herein. However, some hydroxyl and thio protecting groups are neither ether- nor ester-forming groups, as will be understood by those skilled in the art, and are included with
25 amides, discussed below.

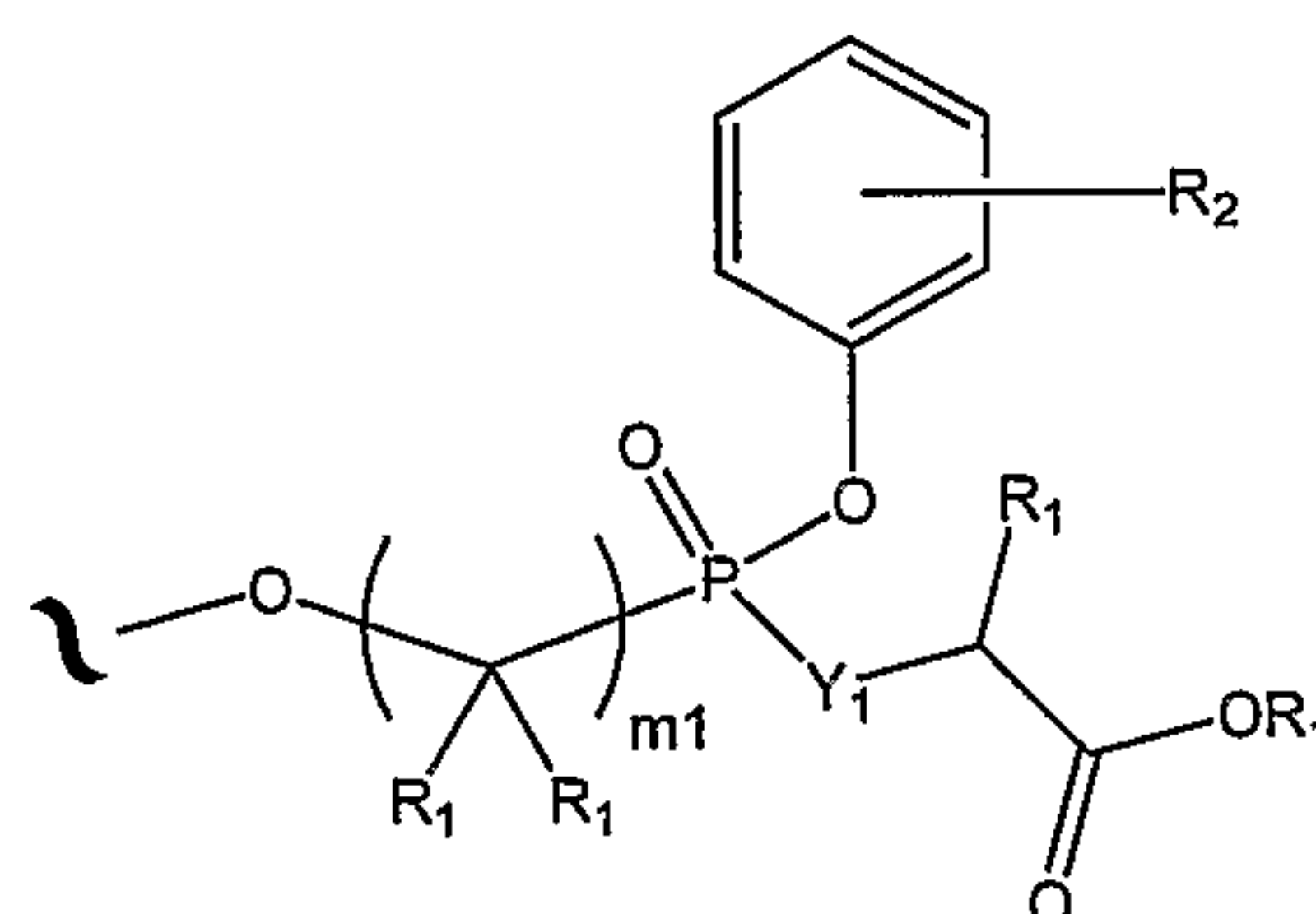
A very large number of hydroxyl protecting groups and amide-forming groups and corresponding chemical cleavage reactions are described in Protective Groups in Organic Synthesis, Theodora W. Greene (John Wiley & Sons, Inc., New York, 1991, ISBN 0-471-62301-6) ("Greene"). See also
30 Kocienski, Philip J.; Protecting Groups (Georg Thieme Verlag Stuttgart, New York, 1994). In particular Chapter 1, Protecting Groups: An Overview, pages 1-20, Chapter 2, Hydroxyl Protecting Groups, pages 21-94, Chapter 3, Diol Protecting Groups, pages 95-117, Chapter 4, Carboxyl Protecting Groups, pages

118-154, Chapter 5, Carbonyl Protecting Groups, pages 155-184. For protecting groups for carboxylic acid, phosphonic acid, phosphonate, sulfonic acid and other protecting groups for acids see Greene as set forth below. Such groups include by way of example and not limitation, esters, amides, hydrazides, and the like. Ether- and Ester-forming protecting groups

Ester-forming groups include: (1) phosphonate ester-forming groups, such as phosphoramidate esters, phosphorothioate esters, phosphonate esters, and phosphon-bis-amidates; (2) carboxyl ester-forming groups, and (3) sulphur ester-forming groups, such as sulphonate, sulfate, and sulfinat.

The phosphonate moieties of the compounds of the invention may or may not be prodrug moieties, *i.e.* they may or may be susceptible to hydrolytic or enzymatic cleavage or modification. Certain phosphonate moieties are stable under most or nearly all metabolic conditions. For example, a dialkylphosphonate, where the alkyl groups are two or more carbons, may have appreciable stability *in vivo* due to a slow rate of hydrolysis.

Within the context of phosphonate prodrug moieties, a large number of structurally-diverse prodrugs have been described for phosphonic acids (Freeman and Ross in Progress in Medicinal Chemistry 34: 112-147 (1997) and are included within the scope of the present invention. An exemplary phosphonate ester-forming group is the phenyl carbocycle in substructure A₃ having the formula:



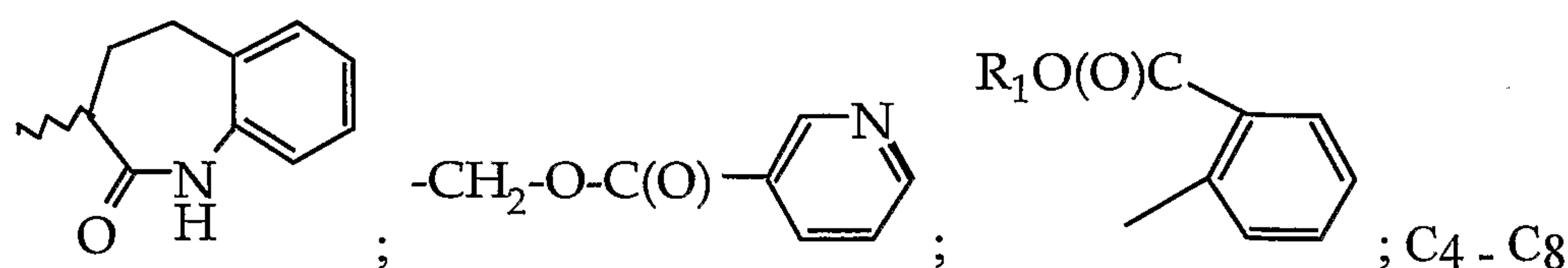
wherein R₁ may be H or C₁–C₁₂ alkyl; m₁ is 1, 2, 3, 4, 5, 6, 7 or 8, and the phenyl carbocycle is substituted with 0 to 3 R₂ groups. Where Y₁ is O, a lactate ester is formed, and where Y₁ is N(R₂), N(OR₂) or N(N(R₂)₂), a phosphoramidate ester results.

In its ester-forming role, a protecting group typically is bound to any acidic group such as, by way of example and not limitation, a –CO₂H or

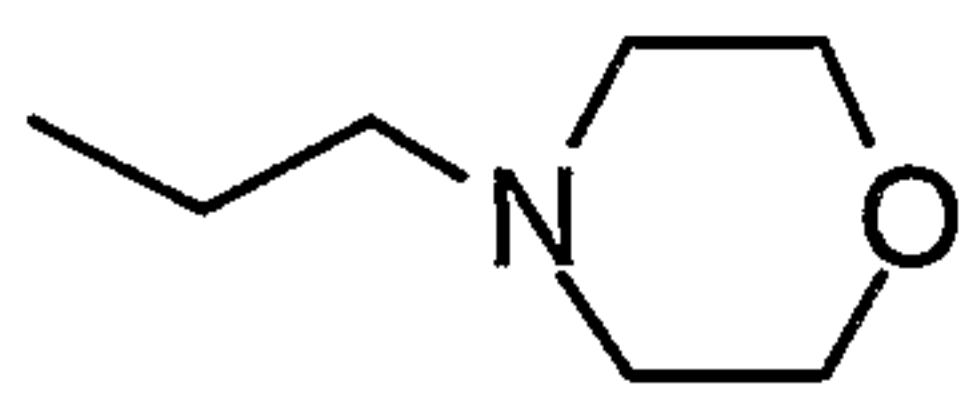
-C(S)OH group, thereby resulting in $-\text{CO}_2\text{R}^x$ where R^x is defined herein. Also, R^x for example includes the enumerated ester groups of WO 95/07920.

Examples of protecting groups include:

- $\text{C}_3\text{--C}_{12}$ heterocycle (described above) or aryl. These aromatic groups optionally are polycyclic or monocyclic. Examples include phenyl, spiryl, 2- and 3-pyrrolyl, 2- and 3-thienyl, 2- and 4-imidazolyl, 2-, 4- and 5-oxazolyl, 3- and 4-isoxazolyl, 2-, 4- and 5-thiazolyl, 3-, 4- and 5-isothiazolyl, 3- and 4-pyrazolyl, 1-, 2-, 3- and 4-pyridinyl, and 1-, 2-, 4- and 5-pyrimidinyl,
- $\text{C}_3\text{--C}_{12}$ heterocycle or aryl substituted with halo, R^1 , $\text{R}^1\text{-O-C}_1\text{--C}_{12}$ alkylene, $\text{C}_1\text{--C}_{12}$ alkoxy, CN, NO_2 , OH, carboxy, carboxyester, thiol, thioester, $\text{C}_1\text{--C}_{12}$ haloalkyl (1-6 halogen atoms), $\text{C}_2\text{--C}_{12}$ alkenyl or $\text{C}_2\text{--C}_{12}$ alkynyl. Such groups include 2-, 3- and 4-alkoxyphenyl ($\text{C}_1\text{--C}_{12}$ alkyl), 2-, 3- and 4-methoxyphenyl, 2-, 3- and 4-ethoxyphenyl, 2,3-, 2,4-, 2,5-, 2,6-, 3,4- and 3,5-diethoxyphenyl, 2- and 3-carboethoxy-4-hydroxyphenyl, 2- and 3-ethoxy-4-hydroxyphenyl, 2- and 3-ethoxy-5-hydroxyphenyl, 2- and 3-ethoxy-6-hydroxyphenyl, 2-, 3- and 4-O-acetylphenyl, 2-, 3- and 4-dimethylaminophenyl, 2-, 3- and 4-methylmercaptophenyl, 2-, 3- and 4-halophenyl (including 2-, 3- and 4-fluorophenyl and 2-, 3- and 4-chlorophenyl), 2,3-, 2,4-, 2,5-, 2,6-, 3,4- and 3,5-dimethylphenyl, 2,3-, 2,4-, 2,5-, 2,6-, 3,4- and 3,5-biscarboxyethylphenyl, 2,3-, 2,4-, 2,5-, 2,6-, 3,4- and 3,5-dimethoxyphenyl, 2,3-, 2,4-, 2,5-, 2,6-, 3,4- and 3,5-dihalophenyl (including 2,4-difluorophenyl and 3,5-difluorophenyl), 2-, 3- and 4-haloalkylphenyl (1 to 5 halogen atoms, $\text{C}_1\text{--C}_{12}$ alkyl including 4-trifluoromethylphenyl), 2-, 3- and 4-cyanophenyl, 2-, 3- and 4-nitrophenyl, 2-, 3- and 4-haloalkylbenzyl (1 to 5 halogen atoms, $\text{C}_1\text{--C}_{12}$ alkyl including 4-trifluoromethylbenzyl and 2-, 3- and 4-trichloromethylphenyl and 2-, 3- and 4-trichloromethylbenzyl), 4-N-methylpiperidinyl, 3-N-methylpiperidinyl, 1-ethylpiperazinyl, benzyl, alkylsalicylphenyl ($\text{C}_1\text{--C}_4$ alkyl, including 2-, 3- and 4-ethylsalicylphenyl), 2-, 3- and 4-acetylphenyl, 1,8-dihydroxynaphthyl ($-\text{C}_{10}\text{H}_6\text{-OH}$) and aryloxy ethyl [$\text{C}_6\text{--C}_9$ aryl (including phenoxy ethyl)], 2,2'-dihydroxybiphenyl, 2-, 3- and 4-N,N-dialkylaminophenol, $-\text{C}_6\text{H}_4\text{CH}_2\text{-N}(\text{CH}_3)_2$, trimethoxybenzyl, triethoxybenzyl, 2-alkyl pyridinyl ($\text{C}_1\text{--C}_4$ alkyl);



esters of 2-carboxyphenyl; and C₁-C₄ alkylene-C₃-C₆ aryl (including benzyl, -CH₂-pyrrolyl, -CH₂-thienyl, -CH₂-imidazolyl, -CH₂-oxazolyl, -CH₂-isoxazolyl, -CH₂-thiazolyl, -CH₂-isothiazolyl, -CH₂-pyrazolyl, -CH₂-pyridinyl and -CH₂-pyrimidinyl) substituted in the aryl moiety by 3 to 5 halogen atoms or 1 to 2 atoms or groups selected from halogen, C₁-C₁₂ alkoxy (including methoxy and ethoxy), cyano, nitro, OH, C₁-C₁₂ haloalkyl (1 to 6 halogen atoms; including -CH₂CCl₃), C₁-C₁₂ alkyl (including methyl and ethyl), C₂-C₁₂ alkenyl or C₂-C₁₂ alkynyl; alkoxy ethyl [C₁-C₆ alkyl including -CH₂-CH₂-O-CH₃ (methoxy ethyl)]; alkyl substituted by any of the groups set forth above for aryl, in particular OH or by 1 to 3 halo atoms (including -CH₃, -CH(CH₃)₂, -C(CH₃)₃, -CH₂CH₃, -(CH₂)₂CH₃, -(CH₂)₃CH₃, -(CH₂)₄CH₃, -(CH₂)₅CH₃, -CH₂CH₂F, -CH₂CH₂Cl, -CH₂CF₃, and -CH₂CCl₃);



; -N-2-propylmorpholino, 2,3-dihydro-6-hydroxyindene, sesamol, catechol monoester, -CH₂-C(O)-N(R¹)₂, -CH₂-S(O)(R¹), -CH₂-S(O)₂(R¹), -CH₂-CH(OC(O)CH₂R¹)-CH₂(OC(O)CH₂R¹), cholesteryl, enolpyruvate (HOOC-C(=CH₂)-), glycerol;

a 5 or 6 carbon monosaccharide, disaccharide or oligosaccharide (3 to 9 monosaccharide residues);

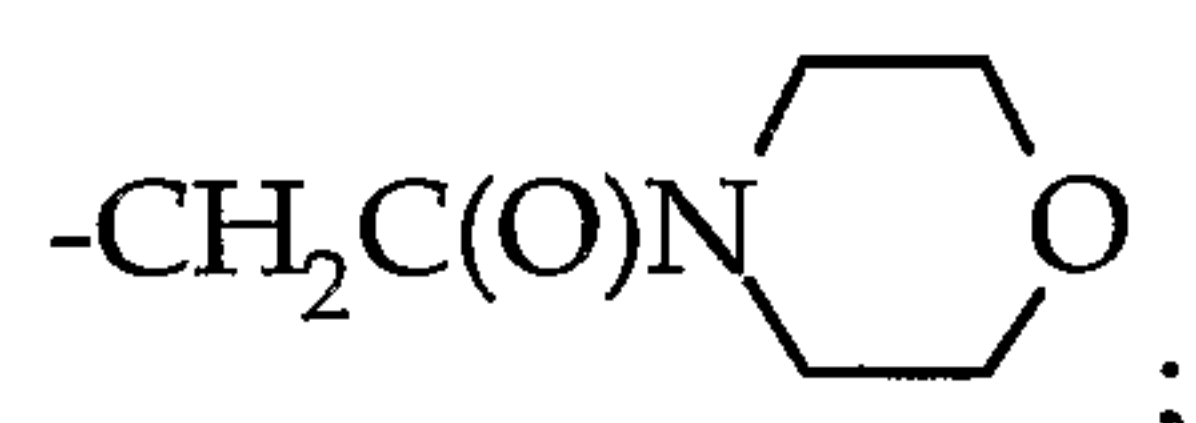
triglycerides such as α-D-β-diglycerides (wherein the fatty acids composing glyceride lipids generally are naturally occurring saturated or unsaturated C₆-26, C₆-18 or C₆-10 fatty acids such as linoleic, lauric, myristic, palmitic, stearic, oleic, palmitoleic, linolenic and the like fatty acids) linked to acyl of the parental compounds herein through a glyceryl oxygen of the triglyceride;

phospholipids linked to the carboxyl group through the phosphate of the phospholipid;

phthalidyl (shown in Fig. 1 of Clayton *et al.*, *Antimicrob. Agents Chemo.* (1974) 5(6):670-671;

cyclic carbonates such as (5-R_d-2-oxo-1,3-dioxolen-4-yl) methyl esters (Sakamoto *et al.*, *Chem. Pharm. Bull.* (1984) 32(6)2241-2248) where R_d is R₁,

5 R₄ or aryl; and



The hydroxyl groups of the compounds of this invention optionally are substituted with one of groups III, IV or V disclosed in WO 94/21604, or with isopropyl.

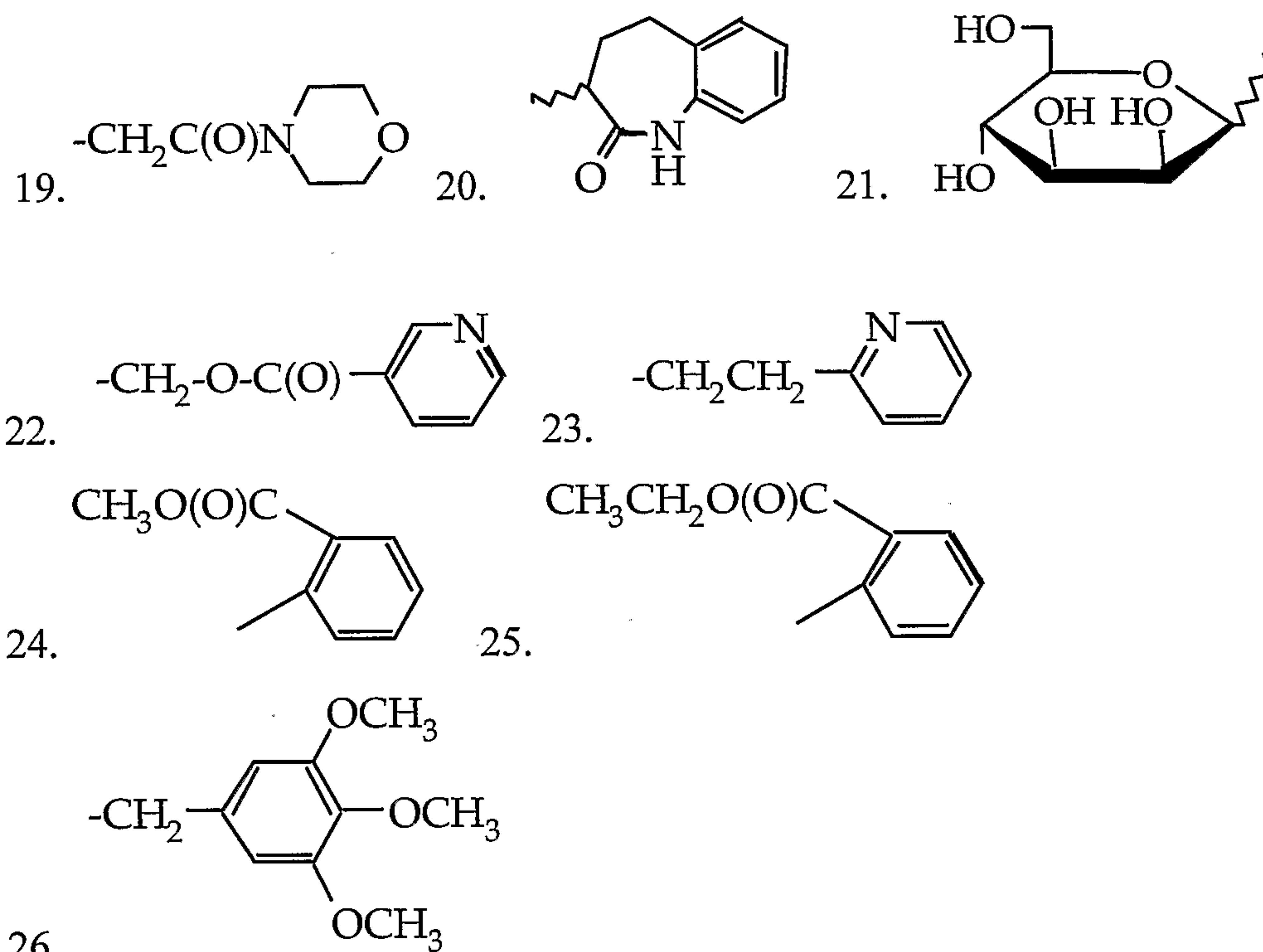
10 Table A lists examples of protecting group ester moieties that for example can be bonded via oxygen to -C(O)O- and -P(O)(O-)₂ groups. Several amidates also are shown, which are bound directly to -C(O)- or -P(O)₂. Esters of structures 1-5, 8-10 and 16, 17, 19-22 are synthesized by reacting the compound herein having a free hydroxyl with the corresponding halide (chloride or acyl
15 chloride and the like) and N,N-dicyclohexyl-N-morpholine carboxamidine (or another base such as DBU, triethylamine, CsCO₃, N,N-dimethylaniline and the like) in DMF (or other solvent such as acetonitrile or N-methylpyrrolidone). When the compound to be protected is a phosphonate, the esters of structures 5-7, 11, 12, 21, and 23-26 are synthesized by reaction of the alcohol or alkoxide salt
20 (or the corresponding amines in the case of compounds such as 13, 14 and 15) with the monochlorophosphonate or dichlorophosphonate (or another activated phosphonate).

TABLE A

25	1. -CH ₂ -C(O)-N(R ₁) ₂ *	10. -CH ₂ -O-C(O)-C(CH ₃) ₃
	2. -CH ₂ -S(O)(R ₁)	11. -CH ₂ -CCl ₃
	3. -CH ₂ -S(O) ₂ (R ₁)	12. -C ₆ H ₅
	4. -CH ₂ -O-C(O)-CH ₂ -C ₆ H ₅	13. -NH-CH ₂ -C(O)O-CH ₂ CH ₃
	5. 3-cholesteryl	14. -N(CH ₃)-CH ₂ -C(O)O-CH ₂ CH ₃
30	6. 3-pyridyl	15. -NHR ₁

7. N-ethylmorpholino
 8. $-\text{CH}_2-\text{O}-\text{C}(\text{O})-\text{C}_6\text{H}_5$
 9. $-\text{CH}_2-\text{O}-\text{C}(\text{O})-\text{CH}_2\text{CH}_3$
 16. $-\text{CH}_2-\text{O}-\text{C}(\text{O})-\text{C}_{10}\text{H}_{15}$
 17. $-\text{CH}_2-\text{O}-\text{C}(\text{O})-\text{CH}(\text{CH}_3)_2$
 18. $-\text{CH}_2-\text{C}\equiv\text{H}(\text{OC}(\text{O})\text{CH}_2\text{R}_1)-\text{CH}_2-$
 $-(\text{OC}(\text{O})\text{CH}_2\text{R}_1)^*$

5



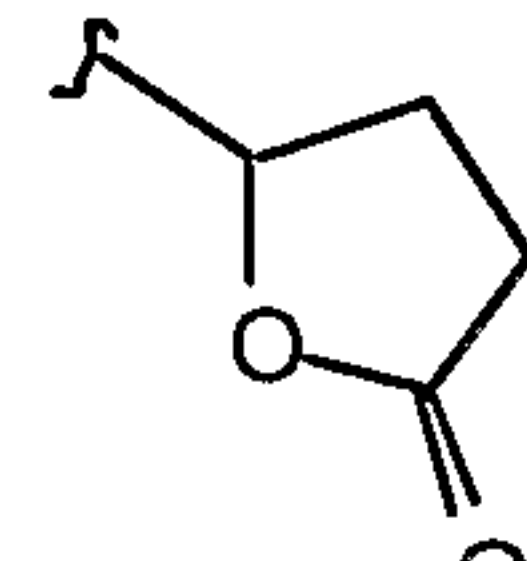
10

- chiral center is (R), (S) or racemate.

Other esters that are suitable for use herein are described in EP 632048.

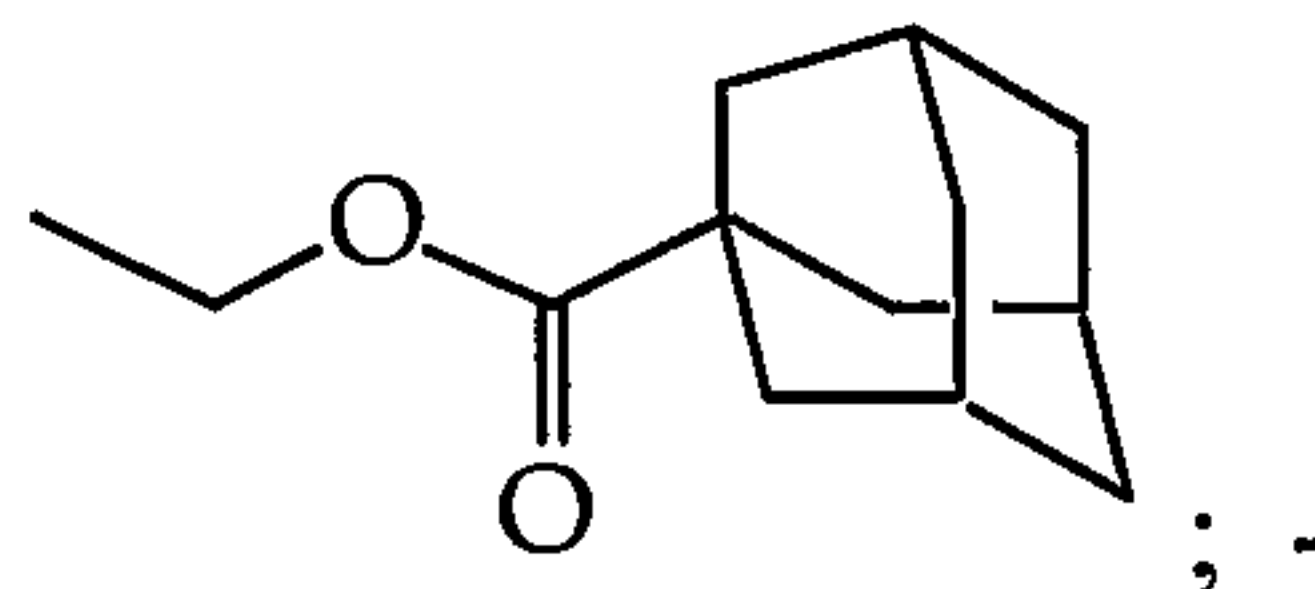
Protecting groups also includes "double ester" forming profunctionalities

15

such as $-\text{CH}_2\text{OC}(\text{O})\text{OCH}_3$,  $-\text{CH}_2\text{SCOCH}_3$, $-\text{CH}_2\text{OCON}(\text{CH}_3)_2$, or alkyl- or aryl-acyloxyalkyl groups of the structure $-\text{CH}(\text{R}^1 \text{ or } \text{W}^5)\text{O}((\text{CO})\text{R}^{37})$ or $-\text{CH}(\text{R}^1 \text{ or } \text{W}^5)((\text{CO})\text{OR}^{38})$ (linked to oxygen of the acidic group) wherein R^{37} and R^{38} are alkyl, aryl, or alkylaryl groups (see U.S. Patent No. 4968788). Frequently R^{37} and R^{38} are bulky groups such as branched alkyl, ortho-substituted aryl, meta-substituted aryl, or combinations thereof, including normal, secondary, iso- and tertiary alkyls of 1-6 carbon atoms. An example is

20

the pivaloyloxymethyl group. These are of particular use with prodrugs for oral administration. Examples of such useful protecting groups are alkylacyloxymethyl esters and their derivatives, including -



CH(CH₂CH₂OCH₃)OC(O)C(CH₃)₃,

- 5 CH₂OC(O)C₁₀H₁₅, -CH₂OC(O)C(CH₃)₃, -CH(CH₂OCH₃)OC(O)C(CH₃)₃,
 -CH(CH(CH₃)₂)OC(O)C(CH₃)₃, -CH₂OC(O)CH₂CH(CH₃)₂, -
 CH₂OC(O)C₆H₁₁, -CH₂OC(O)C₆H₅, -CH₂OC(O)C₁₀H₁₅, -
 CH₂OC(O)CH₂CH₃, -CH₂OC(O)CH(CH₃)₂, -CH₂OC(O)C(CH₃)₃ and -
 CH₂OC(O)CH₂C₆H₅.

- 10 In some claims the protected acidic group is an ester of the acidic group
 and is the residue of a hydroxyl-containing functionality. In other claims, an
 amino compound is used to protect the acid functionality. The residues of
 suitable hydroxyl or amino-containing functionalities are set forth above or are
 found in WO 95/07920. Of particular interest are the residues of amino acids,
 15 amino acid esters, polypeptides, or aryl alcohols. Typical amino acid,
 polypeptide and carboxyl-esterified amino acid residues are described on pages
 11-18 and related text of WO 95/07920 as groups L1 or L2. WO 95/07920
 expressly teaches the amidates of phosphonic acids, but it will be understood that
 such amidates are formed with any of the acid groups set forth herein and the
 20 amino acid residues set forth in WO 95/07920.

- Typical esters for protecting acidic functionalities are also described in
 WO 95/07920, again understanding that the same esters can be formed with the
 acidic groups herein as with the phosphonate of the '920 publication. Typical
 ester groups are defined at least on WO 95/07920 pages 89-93 (under R³¹ or
 25 R³⁵), the table on page 105, and pages 21-23 (as R). Of particular interest are
 esters of unsubstituted aryl such as phenyl or arylalkyl such benzyl, or hydroxy-,
 halo-, alkoxy-, carboxy- and/or alkylestercarboxy-substituted aryl or alkylaryl,
 especially phenyl, ortho-ethoxyphenyl, or C₁-C₄ alkylestercarboxyphenyl
 (salicylate C₁-C₁₂ alkylesters).

The protected acidic groups, particularly when using the esters or amides of WO 95/07920, are useful as prodrugs for oral administration. However, it is not essential that the acidic group be protected in order for the compounds of this invention to be effectively administered by the oral route. When the compounds of the invention having protected groups, in particular amino acid amidates or substituted and unsubstituted aryl esters are administered systemically or orally they are capable of hydrolytic cleavage *in vivo* to yield the free acid.

One or more of the acidic hydroxyls are protected. If more than one acidic hydroxyl is protected then the same or a different protecting group is employed, *e.g.*, the esters may be different or the same, or a mixed amidate and ester may be used.

Typical hydroxy protecting groups described in Greene (pages 14-118) include substituted methyl and alkyl ethers, substituted benzyl ethers, silyl ethers, esters including sulfonic acid esters, and carbonates. For example:

- Ethers (methyl, *t*-butyl, allyl);
- Substituted Methyl Ethers (Methoxymethyl, Methylthiomethyl, *t*-Butylthiomethyl, (Phenyldimethylsilyl)methoxymethyl, Benzyloxymethyl, *p*-Methoxybenzyloxymethyl, (4-Methoxyphenoxy)methyl, Guaiacolmethyl, *t*-Butoxymethyl, 4-Pentenylloxymethyl, Siloxymethyl, 2-Methoxyethoxymethyl, 2,2,2-Trichloroethoxymethyl, Bis(2-chloroethoxy)methyl, 2-(Trimethylsilyl)ethoxymethyl, Tetrahydropyranyl, 3-Bromotetrahydropyranyl, Tetrahydrothiopyranyl, 1-Methoxycyclohexyl, 4-Methoxytetrahydropyranyl, 4-Methoxytetrahydrothiopyranyl, 4-Methoxytetrahydrothiopyranyl *S,S*-Dioxido, 1-[(2-Chloro-4-methyl)phenyl]-4-methoxypiperidin-4-yl, 1,4-Dioxan-2-yl, Tetrahydrofuranyl, Tetrahydrothiofuranyl, 2,3,3a,4,5,6,7,7a-Octahydro-7,8,8-trimethyl-4,7-methanobenzofuran-2-yl));
- Substituted Ethyl Ethers (1-Ethoxyethyl, 1-(2-Chloroethoxy)ethyl, 1-Methyl-1-methoxyethyl, 1-Methyl-1-benzyloxyethyl, 1-Methyl-1-benzyloxy-2-fluoroethyl, 2,2,2-Trichloroethyl, 2-Trimethylsilylethyl, 2-(Phenylselenyl)ethyl,
- *p*-Chlorophenyl, *p*-Methoxyphenyl, 2,4-Dinitrophenyl, Benzyl);

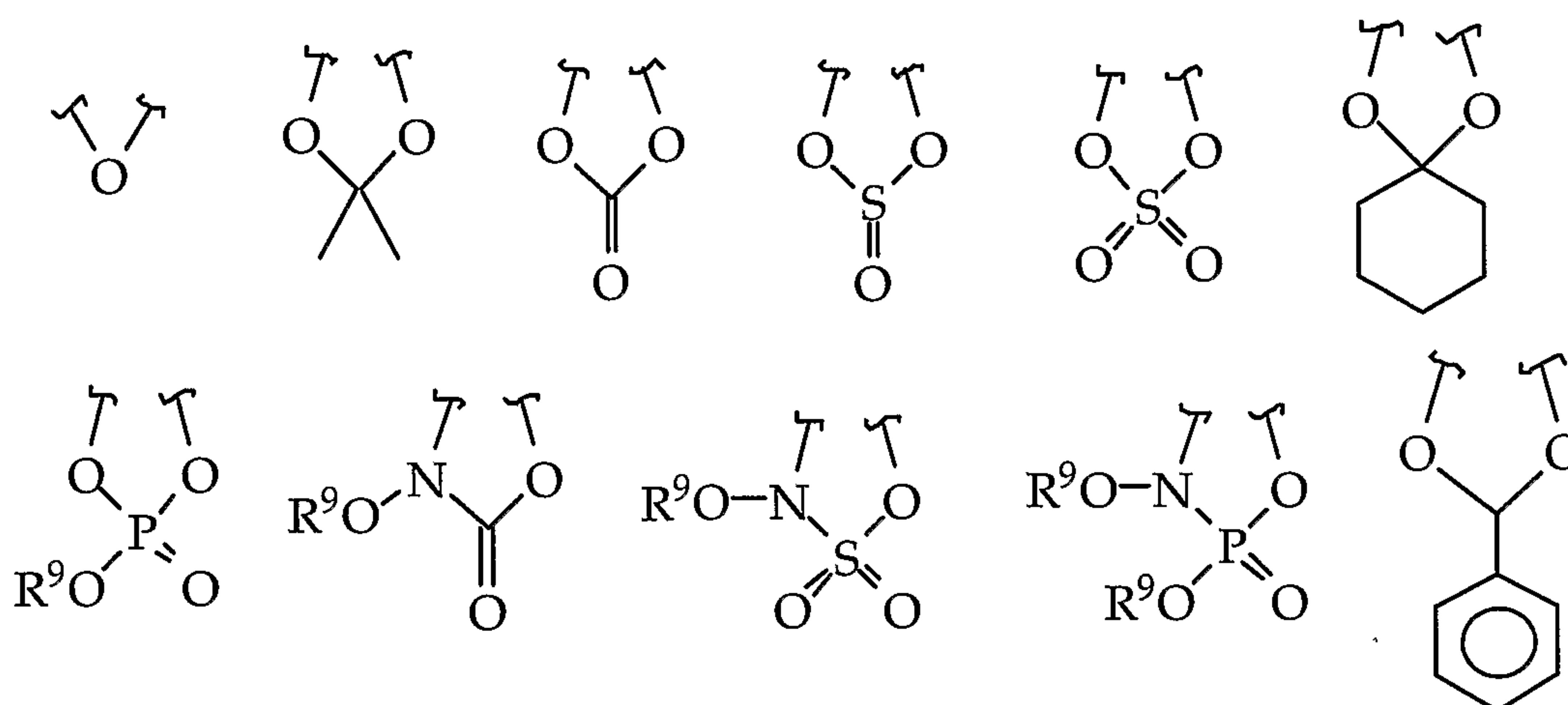
- Substituted Benzyl Ethers (*p*-Methoxybenzyl, 3,4-Dimethoxybenzyl, *o*-Nitrobenzyl, *p*-Nitrobenzyl, *p*-Halobenzyl, 2,6-Dichlorobenzyl, *p*-Cyanobenzyl, *p*-Phenylbenzyl, 2- and 4-Picolyl, 3-Methyl-2-picolyl *N*-Oxido, Diphenylmethyl, *p,p'*-Dinitrobenzhydryl, 5-Dibenzosuberyl, Triphenylmethyl, α -Naphthyldiphenylmethyl, *p*-methoxyphenyldiphenylmethyl, Di(*p*-methoxyphenyl)phenylmethyl, Tri(*p*-methoxyphenyl)methyl, 4-(4'-Bromophenacyloxy)phenyldiphenylmethyl, 4,4',4''-Tris(4,5-dichlorophthalimidophenyl)methyl, 4,4',4''-Tris(levulinoyloxyphenyl)methyl, 4,4',4''-Tris(benzoyloxyphenyl)methyl, 3-(Imidazol-1-ylmethyl)bis(4',4''-dimethoxyphenyl)methyl, 1,1-Bis(4-methoxyphenyl)-1'-pyrenylmethyl, 9-Anthryl, 9-(9-Phenyl)xanthenyl, 9-(9-Phenyl-10-oxo)anthryl, 1,3-Benzodithiolan-2-yl, Benzisothiazolyl *S,S*-Dioxido);
- Silyl Ethers (Trimethylsilyl, Triethylsilyl, Triisopropylsilyl, Dimethylisopropylsilyl, Diethylisopropylsilyl, Dimethylhexylsilyl, *t*-Butyldimethylsilyl, *t*-Butyldiphenylsilyl, Tribenzylsilyl, Tri-*p*-xylylsilyl, Triphenylsilyl, Diphenylmethylsilyl, *t*-Butylmethoxyphenylsilyl);
- Esters (Formate, Benzoylformate, Acetate, Chloroacetate, Dichloroacetate, Trichloroacetate, Trifluoroacetate, Methoxyacetate, Triphenylmethoxyacetate, Phenoxyacetate, *p*-Chlorophenoxyacetate, *p*-poly-Phenylacetate, 3-Phenylpropionate, 4-Oxopentanoate (Levulinate), 4,4-(Ethylenedithio)pentanoate, Pivaloate, Adamantoate, Crotonate, 4-Methoxycrotonate, Benzoate, *p*-Phenylbenzoate, 2,4,6-Trimethylbenzoate (Mesitoate));
- Carbonates (Methyl, 9-Fluorenylmethyl, Ethyl, 2,2,2-Trichloroethyl, 2-(Trimethylsilyl)ethyl, 2-(Phenylsulfonyl)ethyl, 2-(Triphenylphosphonio)ethyl, Isobutyl, Vinyl, Allyl, *p*-Nitrophenyl, Benzyl, *p*-Methoxybenzyl, 3,4-Dimethoxybenzyl, *o*-Nitrobenzyl, *p*-Nitrobenzyl, *S*-Benzyl Thiocarbonate, 4-Ethoxy-1-naphthyl, Methyl Dithiocarbonate);
- Groups With Assisted Cleavage (2-Iodobenzoate, 4-Azidobutyrate, 4-Nitro-4-methylpentanoate, *o*-(Dibromomethyl)benzoate, 2-Formylbenzenesulfonate, 2-(Methylthiomethoxy)ethyl Carbonate, 4-

- (Methylthiomethoxy)butyrate, 2-(Methylthiomethoxymethyl)benzoate);
 Miscellaneous Esters (2,6-Dichloro-4-methylphenoxyacetate, 2,6-Dichloro-
 4-(1,1,3,3 tetramethylbutyl)phenoxyacetate, 2,4-Bis(1,1-
 dimethylpropyl)phenoxyacetate, Chlorodiphenylacetate, Isobutyrate,
 5 Monosuccinate, (*E*)-2-Methyl-2-butenate (Tigloate), *o*-
 (Methoxycarbonyl)benzoate, *p*-poly-Benzoate, α -Naphthoate, Nitrate, Alkyl
N,N,N',N'-Tetramethylphosphorodiamidate, *N*-Phenylcarbamate, Borate,
 Dimethylphosphinothioyl, 2,4-Dinitrophenylsulfenate); and
- Sulfonates (Sulfate, Methanesulfonate (Mesylate), Benzylsulfonate,
 10 Tosylate).

Typical 1,2-diol protecting groups (thus, generally where two OH groups
 are taken together with the protecting functionality) are described in Greene at
 pages 118-142 and include Cyclic Acetals and Ketals (Methylene, Ethylidene, 1-
t-Butylethylidene, 1-Phenylethylidene, (4-Methoxyphenyl)ethylidene, 2,2,2-
 15 Trichloroethylidene, Acetonide (Isopropylidene), Cyclopentylidene,
 Cyclohexylidene, Cycloheptylidene, Benzylidene, *p*-Methoxybenzylidene, 2,4-
 Dimethoxybenzylidene, 3,4-Dimethoxybenzylidene, 2-Nitrobenzylidene); Cyclic
 Ortho Esters (Methoxymethylene, Ethoxymethylene, Dimethoxymethylene, 1-
 Methoxyethylidene, 1-Ethoxyethylidene, 1,2-Dimethoxyethylidene, α -
 20 Methoxybenzylidene, 1-(*N,N*-Dimethylamino)ethylidene Derivative, α -(*N,N*-
 Dimethylamino)benzylidene Derivative, 2-Oxacyclopentylidene); Silyl
 Derivatives (Di-*t*-butylsilylene Group, 1,3-(1,1,3,3-
 Tetraisopropylidisiloxanylidene), and Tetra-*t*-butoxydisiloxane-1,3-diylidene),
 Cyclic Carbonates, Cyclic Boronates, Ethyl Boronate and Phenyl Boronate.

25 More typically, 1,2-diol protecting groups include those shown in Table
 B, still more typically, epoxides, acetonides, cyclic ketals and aryl acetals.

Table B



wherein R⁹ is C₁-C₆ alkyl.

Amino protecting groups

5 Another set of protecting groups include any of the typical amino protecting groups described by Greene at pages 315-385. They include:

- Carbamates: (methyl and ethyl, 9-fluorenylmethyl, 9(2-sulfo)fluorenylmethyl, 9-(2,7-dibromo)fluorenylmethyl, 2,7-di-*t*-butyl-[9-(10,10-dioxo-10,10,10,10-tetrahydrothioxanthyl)]methyl, 4-methoxyphenacyl);
- Substituted Ethyl: (2,2,2-trichloroethyl, 2-trimethylsilylethyl, 2-phenylethyl, 1-(1-adamantyl)-1-methylethyl, 1,1-dimethyl-2-haloethyl, 1,1-dimethyl-2,2-dibromoethyl, 1,1-dimethyl-2,2,2-trichloroethyl, 1-methyl-1-(4-biphenyl)ethyl, 1-(3,5-di-*t*-butylphenyl)-1-methylethyl, 2-(2'- and 4'-pyridyl)ethyl, 2-(*N,N*-dicyclohexylcarboxamido)ethyl, *t*-butyl, 1-adamantyl, vinyl, allyl, 1-isopropylallyl, cinnamyl, 4-nitrocinnamyl, 8-quinolyl, *N*-hydroxypiperidinyll, alkylidithio, benzyl, *p*-methoxybenzyl, *p*-nitrobenzyl, *p*-bromobenzyl, *p*-chlorobenzyl, 2,4-dichlorobenzyl, 4-methylsulfinylbenzyl, 9-anthrylmethyl, diphenylmethyl);
- Groups With Assisted Cleavage: (2-methylthioethyl, 2-methylsulfonylethyl, 2-(*p*-toluenesulfonyl)ethyl, [2-(1,3-dithianyl)]methyl, 4-methylthiophenyl, 2,4-dimethylthiophenyl, 2-phosphonioethyl, 2-triphenylphosphonioisopropyl, 1,1-dimethyl-2-cyanoethyl, *m*-chloro-*p*-

acyloxybenzyl, *p*-(dihydroxyboryl)benzyl, 5-benzisoxazolylmethyl, 2-(trifluoromethyl)-6-chromonylmethyl);

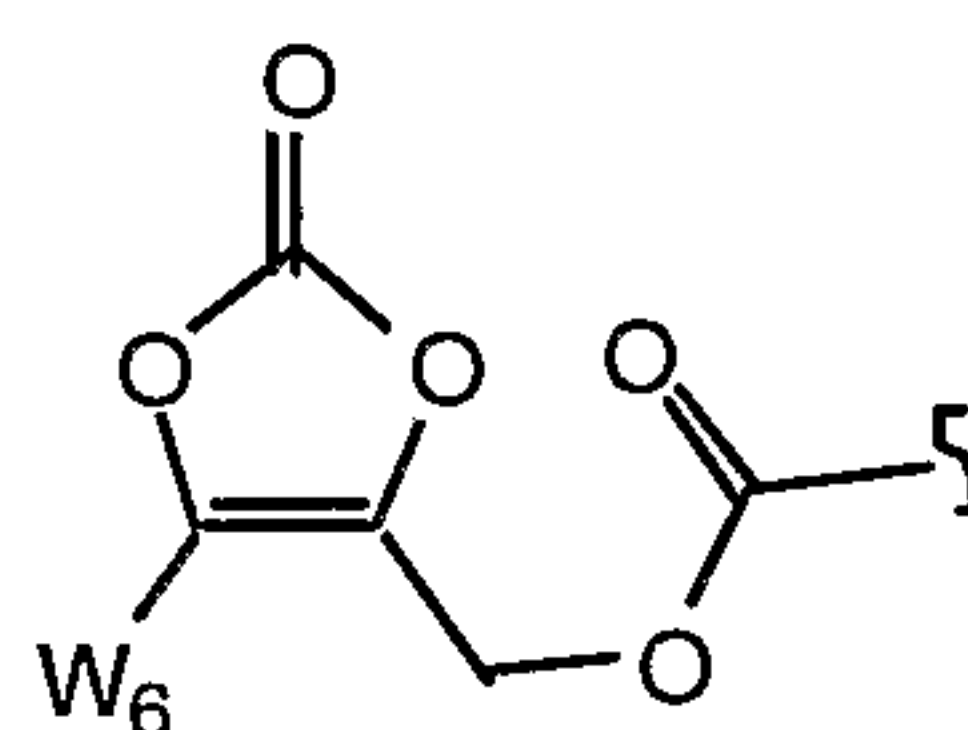
- Groups Capable of Photolytic Cleavage: (*m*-nitrophenyl, 3,5-dimethoxybenzyl, *o*-nitrobenzyl, 3,4-dimethoxy-6-nitrobenzyl, phenyl(*o*-nitrophenyl)methyl); Urea-Type Derivatives (phenothiazinyl-(10)-carbonyl, *N'*-*p*-toluenesulfonylaminocarbonyl, *N'*-phenylaminothiocarbonyl);
- Miscellaneous Carbamates: (*t*-amyl, *S*-benzyl thiocarbamate, *p*-cyanobenzyl, cyclobutyl, cyclohexyl, cyclopentyl, cyclopropylmethyl, *p*-decyloxybenzyl, diisopropylmethyl, 2,2-dimethoxycarbonylvinyl, *o*-(*N,N*-dimethylcarboxamido)benzyl, 1,1-dimethyl-3-(*N,N*-dimethylcarboxamido)propyl, 1,1-dimethylpropynyl, di(2-pyridyl)methyl, 2-furanylmethyl, 2-Iodoethyl, Isobornyl, Isobutyl, Isonicotinyl, *p*-(*p'*-Methoxyphenylazo)benzyl, 1-methylcyclobutyl, 1-methylcyclohexyl, 1-methyl-1-cyclopropylmethyl, 1-methyl-1-(3,5-dimethoxyphenyl)ethyl, 1-methyl-1-(*p*-phenylazophenyl)ethyl, 1-methyl-1-phenylethyl, 1-methyl-1-(4-pyridyl)ethyl, phenyl, *p*-(phenylazo)benzyl, 2,4,6-tri-*t*-butylphenyl, 4-(trimethylammonium)benzyl, 2,4,6-trimethylbenzyl);
- Amides: (*N*-formyl, *N*-acetyl, *N*-chloroacetyl, *N*-trichloroacetyl, *N*-trifluoroacetyl, *N*-phenylacetyl, *N*-3-phenylpropionyl, *N*-picolinoyl, *N*-3-pyridylcarboxamide, *N*-benzoylphenylalanyl, *N*-benzoyl, *N*-*p*-phenylbenzoyl);
- Amides With Assisted Cleavage: (*N*-*o*-nitrophenylacetyl, *N*-*o*-nitrophenoxyacetyl, *N*-acetoacetyl, (*N'*-dithiobenzyloxycarbonylamino)acetyl, *N*-3-(*p*-hydroxyphenyl)propionyl, *N*-3-(*o*-nitrophenyl)propionyl, *N*-2-methyl-2-(*o*-nitrophenoxy)propionyl, *N*-2-methyl-2-(*o*-phenylazophenoxy)propionyl, *N*-4-chlorobutyryl, *N*-3-methyl-3-nitrobutyryl, *N*-*o*-nitrocinnamoyl, *N*-acetylmethionine, *N*-*o*-nitrobenzoyl, *N*-*o*-(benzoyloxymethyl)benzoyl, 4,5-diphenyl-3-oxazolin-2-one);
- Cyclic Imide Derivatives: (*N*-phthalimide, *N*-dithiasuccinoyl, *N*-2,3-diphenylmaleoyl, *N*-2,5-dimethylpyrrolyl, *N*-1,1,4,4-tetramethyldisilylazacyclopentane adduct, 5-substituted 1,3-dimethyl-1,3,5-

triazacyclohexan-2-one, 5-substituted 1,3-dibenzyl-1,3,5-triazacyclohexan-2-one, 1-substituted 3,5-dinitro-4-pyridonyl);

- *N*-Alkyl and *N*-Aryl Amines: (*N*-methyl, *N*-allyl, *N*-[2-(trimethylsilyl)ethoxy]methyl, *N*-3-acetoxypropyl, *N*-(1-isopropyl-4-nitro-2-oxo-3-pyrrolin-3-yl), Quaternary Ammonium Salts, *N*-benzyl, *N*-di(4-methoxyphenyl)methyl, *N*-5-dibenzosuberyl, *N*-triphenylmethyl, *N*-(4-methoxyphenyl)diphenylmethyl, *N*-9-phenylfluorenyl, *N*-2,7-dichloro-9-fluorenylmethylene, *N*-ferrocenylmethyl, *N*-2-picolylamine *N'*-oxide);
- Imine Derivatives: (*N*-1,1-dimethylthiomethylene, *N*-benzylidene, *N*-*p*-methoxybenzylidene, *N*-diphenylmethylene, *N*-[(2-pyridyl)mesityl]methylene, *N*,(*N'*,*N'*-dimethylaminomethylene, *N*,*N'*-isopropylidene, *N*-*p*-nitrobenzylidene, *N*-salicylidene, *N*-5-chlorosalicylidene, *N*-(5-chloro-2-hydroxyphenyl)phenylmethylene, *N*-cyclohexylidene);
- Enamine Derivatives: (*N*-(5,5-dimethyl-3-oxo-1-cyclohexenyl));
- *N*-Metal Derivatives (*N*-borane derivatives, *N*-diphenylborinic acid derivatives, *N*-[phenyl(pentacarbonylchromium- or -tungsten)]carbenyl, *N*-copper or *N*-zinc chelate);
- *N*-*N* Derivatives: (*N*-nitro, *N*-nitroso, *N*-oxide);
- *N*-*P* Derivatives: (*N*-diphenylphosphinyl, *N*-dimethylthiophosphinyl, *N*-diphenylthiophosphinyl, *N*-dialkyl phosphoryl, *N*-dibenzyl phosphoryl, *N*-diphenyl phosphoryl);
- *N*-*Si* Derivatives, *N*-*S* Derivatives, and *N*-Sulfenyl Derivatives: (*N*-benzenesulfenyl, *N*-*o*-nitrobenzenesulfenyl, *N*-2,4-dinitrobenzenesulfenyl, *N*-pentachlorobenzenesulfenyl, *N*-2-nitro-4-methoxybenzenesulfenyl, *N*-triphenylmethylsulfenyl, *N*-3-nitropyridinesulfenyl); and *N*-sulfonyl Derivatives (*N*-*p*-toluenesulfonyl, *N*-benzenesulfonyl, *N*-2,3,6-trimethyl-4-methoxybenzenesulfonyl, *N*-2,4,6-trimethoxybenzenesulfonyl, *N*-2,6-dimethyl-4-methoxybenzenesulfonyl, *N*-pentamethylbenzenesulfonyl, *N*-2,3,5,6,-tetramethyl-4-methoxybenzenesulfonyl, *N*-4-methoxybenzenesulfonyl, *N*-2,4,6-trimethylbenzenesulfonyl, *N*-2,6-dimethoxy-4-methylbenzenesulfonyl, *N*-2,2,5,7,8-pentamethylchroman-6-sulfonyl, *N*-methanesulfonyl, *N*-β-trimethylsilyethanesulfonyl, *N*-9-

anthracenesulfonyl, *N*-4-(4',8'-dimethoxynaphthylmethyl)benzenesulfonyl, *N*-benzylsulfonyl, *N*-trifluoromethylsulfonyl, *N*-phenacylsulfonyl).

More typically, protected amino groups include carbamates and amides, still more typically, -NHC(O)R^1 or $\text{-N=CR}^1\text{N(R}^1)_2$. Another protecting group, also useful as a prodrug for amino or $\text{-NH(R}^5)$, is:



See for example Alexander, J. *et al.* (1996) *J. Med. Chem.* 39:480-486.

Amino acid and polypeptide protecting group and conjugates

An amino acid or polypeptide protecting group of a compound of the invention has the structure $\text{R}^{15}\text{NHCH(R}^{16})\text{C(O)-}$, where R^{15} is H, an amino acid or polypeptide residue, or R^5 , and R^{16} is defined below.

R^{16} is lower alkyl or lower alkyl ($\text{C}_1\text{-C}_6$) substituted with amino, carboxyl, amide, carboxyl ester, hydroxyl, $\text{C}_6\text{-C}_7$ aryl, guanidinyll, imidazolyl, indolyl, sulfhydryl, sulfoxide, and/or alkylphosphate. R^{10} also is taken together with the amino acid α N to form a proline residue ($\text{R}^{10} = \text{-CH}_2)_3\text{-}$). However, R^{10} is generally the side group of a naturally-occurring amino acid such as H, -CH_3 , $\text{-CH(CH}_3)_2$, $\text{-CH}_2\text{-CH(CH}_3)_2$, $\text{-CHCH}_3\text{-CH}_2\text{-CH}_3$, $\text{-CH}_2\text{-C}_6\text{H}_5$, $\text{-CH}_2\text{CH}_2\text{-S-CH}_3$, $\text{-CH}_2\text{OH}$, -CH(OH)-CH_3 , $\text{-CH}_2\text{-SH}$, $\text{-CH}_2\text{-C}_6\text{H}_4\text{OH}$, $\text{-CH}_2\text{-CO-NH}_2$, $\text{-CH}_2\text{-CH}_2\text{-CO-NH}_2$, $\text{-CH}_2\text{-COOH}$, $\text{-CH}_2\text{-CH}_2\text{-COOH}$, $\text{-(CH}_2)_4\text{-NH}_2$ and $\text{-(CH}_2)_3\text{-NH-C(NH}_2)_2$. R_{10} also includes 1-guanidinoprop-3-yl, benzyl, 4-hydroxybenzyl, imidazol-4-yl, indol-3-yl, methoxyphenyl and ethoxyphenyl.

Another set of protecting groups include the residue of an amino-containing compound, in particular an amino acid, a polypeptide, a protecting group, $\text{-NHSO}_2\text{R}$, NHC(O)R , -N(R)_2 , NH_2 or -NH(R)(H) , whereby for example a carboxylic acid is reacted, *i.e.* coupled, with the amine to form an amide, as in C(O)NR_2 . A phosphonic acid may be reacted with the amine to form a phosphoramidate, as in $\text{-P(O)(OR)(NR}_2)$.

In general, amino acids have the structure $R^{17}C(O)CH(R^{16})NH-$, where R^{17} is -OH, -OR, an amino acid or a polypeptide residue. Amino acids are low molecular weight compounds, on the order of less than about 1000 MW and which contain at least one amino or imino group and at least one carboxyl group.

5 Generally the amino acids will be found in nature, *i.e.*, can be detected in biological material such as bacteria or other microbes, plants, animals or man. Suitable amino acids typically are alpha amino acids, *i.e.* compounds characterized by one amino or imino nitrogen atom separated from the carbon atom of one carboxyl group by a single substituted or unsubstituted alpha carbon
10 atom. Of particular interest are hydrophobic residues such as mono-or di-alkyl or aryl amino acids, cycloalkylamino acids and the like. These residues contribute to cell permeability by increasing the partition coefficient of the parental drug. Typically, the residue does not contain a sulfhydryl or guanidino substituent.

15 Naturally-occurring amino acid residues are those residues found naturally in plants, animals or microbes, especially proteins thereof. Polypeptides most typically will be substantially composed of such naturally-occurring amino acid residues. These amino acids are glycine, alanine, valine, leucine, isoleucine, serine, threonine, cysteine, methionine, glutamic acid,
20 aspartic acid, lysine, hydroxylysine, arginine, histidine, phenylalanine, tyrosine, tryptophan, proline, asparagine, glutamine and hydroxyproline. Additionally, unnatural amino acids, for example, valanine, phenylglycine and homoarginine are also included. Commonly encountered amino acids that are not gene-encoded may also be used in the present invention. All of the amino acids used
25 in the present invention may be either the D- or L- optical isomer. In addition, other peptidomimetics are also useful in the present invention. For a general review, see Spatola, A. F., in Chemistry and Biochemistry of Amino Acids, Peptides and Proteins, B. Weinstein, eds., Marcel Dekker, New York, p. 267 (1983).

30 When protecting groups are single amino acid residues or polypeptides they optionally are substituted at R^3 of substituents A^1 or A^3 in a compound of the invention. These conjugates are produced by forming an amide bond between a carboxyl group of the amino acid (or C-terminal amino acid of a

polypeptide for example). Similarly, conjugates are formed between R^3 and an amino group of an amino acid or polypeptide. Generally, only one of any site in the parental molecule is amidated with an amino acid as described herein, although it is within the scope of this invention to introduce amino acids at more than one permitted site. Usually, a carboxyl group of R^3 is amidated with an amino acid. In general, the α -amino or α -carboxyl group of the amino acid or the terminal amino or carboxyl group of a polypeptide are bonded to the parental functionalities, *i.e.*, carboxyl or amino groups in the amino acid side chains generally are not used to form the amide bonds with the parental compound (although these groups may need to be protected during synthesis of the conjugates as described further below).

With respect to the carboxyl-containing side chains of amino acids or polypeptides it will be understood that the carboxyl group optionally will be blocked, *e.g.*, by R^1 , esterified with R^5 or amidated. Similarly, the amino side chains R^{16} optionally will be blocked with R^1 or substituted with R^5 .

Such ester or amide bonds with side chain amino or carboxyl groups, like the esters or amides with the parental molecule, optionally are hydrolyzable *in vivo* or *in vitro* under acidic (pH <3) or basic (pH >10) conditions. Alternatively, they are substantially stable in the gastrointestinal tract of humans but are hydrolyzed enzymatically in blood or in intracellular environments. The esters or amino acid or polypeptide amidates also are useful as intermediates for the preparation of the parental molecule containing free amino or carboxyl groups. The free acid or base of the parental compound, for example, is readily formed from the esters or amino acid or polypeptide conjugates of this invention by conventional hydrolysis procedures.

When an amino acid residue contains one or more chiral centers, any of the D, L, meso, threo or erythro (as appropriate) racemates, scalemates or mixtures thereof may be used. In general, if the intermediates are to be hydrolyzed non-enzymatically (as would be the case where the amides are used as chemical intermediates for the free acids or free amines), D isomers are useful. On the other hand, L isomers are more versatile since they can be susceptible to both non-enzymatic and enzymatic hydrolysis, and are more

efficiently transported by amino acid or dipeptidyl transport systems in the gastrointestinal tract.

Examples of suitable amino acids whose residues are represented by R^x or R^y include the following:

5 Glycine;

Aminopolycarboxylic acids, *e.g.*, aspartic acid, β -hydroxyaspartic acid, glutamic acid, β -hydroxyglutamic acid, β -methylasspartic acid, β -methylglutamic acid, β , β -dimethylasspartic acid, γ -hydroxyglutamic acid, β , γ -dihydroxyglutamic acid, β -phenylglutamic acid, γ -methyleneglutamic acid, 3-aminoadipic acid, 2-aminopimelic acid, 2-aminosuberlic acid and 2-aminosebacic acid;

Amino acid amides such as glutamine and asparagine;

Polyamino- or polybasic-monocarboxylic acids such as arginine, lysine, β -aminoalanine, γ -aminobutyric acid, ornithine, citrulline, homoarginine, homocitrulline, hydroxylysine, allohydroxylysine and diaminobutyric acid;

15 Other basic amino acid residues such as histidine;

Diaminodicarboxylic acids such as α , α' -diaminosuccinic acid, α , α' -diaminoglutaric acid, α , α' -diaminoadipic acid, α , α' -diaminopimelic acid, α , α' -diamino- β -hydroxypimelic acid, α , α' -diaminosuberlic acid, α , α' -diaminoazelaic acid, and α , α' -diaminosebacic acid;

20 Imino acids such as proline, hydroxyproline, allohydroxyproline, γ -methylproline, pipercolic acid, 5-hydroxypipercolic acid, and azetidine-2-carboxylic acid;

A mono- or di-alkyl (typically C_1 - C_8 branched or normal) amino acid such as alanine, valine, leucine, allylglycine, butyric acid, norvaline, norleucine, 25 heptyline, α -methylserine, α -amino- α -methyl- γ -hydroxyvaleric acid, α -amino- α -methyl- δ -hydroxyvaleric acid, α -amino- α -methyl- ϵ -hydroxycaproic acid, isovaline, α -methylglutamic acid, α -aminoisobutyric acid, α -aminodiethylacetic acid, α -aminodiisopropylacetic acid, α -aminodi-n-propylacetic acid, α -aminodiisobutylacetic acid, α -aminodi-n-butylacetic acid, α -aminoethylisopropylacetic acid, α -amino-n-propylacetic acid, α -aminodiisoamylacetic acid, α -methylasspartic acid, α -methylglutamic acid, 1-30

aminocyclopropane-1-carboxylic acid, isoleucine, alloisoleucine, *tert*-leucine, β -methyltryptophan and α -amino- β -ethyl- β -phenylpropionic acid;

β -phenylserinyl;

Aliphatic α -amino- β -hydroxy acids such as serine, β -hydroxyleucine, β -hydroxynorleucine, β -hydroxynorvaline, and α -amino- β -hydroxystearic acid;

α -Amino, α -, γ -, δ - or ϵ -hydroxy acids such as homoserine, δ -hydroxynorvaline, γ -hydroxynorvaline and ϵ -hydroxynorleucine residues; canavine and canaline; γ -hydroxyornithine;

2-hexosaminic acids such as D-glucosaminic acid or D-galactosaminic acid;

α -Amino- β -thiols such as penicillamine, β -thiolnorvaline or β -thiolbutyrine;

Other sulfur containing amino acid residues including cysteine; homocystine, β -phenylmethionine, methionine, S-allyl-L-cysteine sulfoxide, 2-thiolhistidine, cystathionine, and thiol ethers of cysteine or homocysteine;

Phenylalanine, tryptophan and ring-substituted α -amino acids such as the phenyl- or cyclohexylamino acids α -aminophenylacetic acid, α -aminocyclohexylacetic acid and α -amino- β -cyclohexylpropionic acid; phenylalanine analogues and derivatives comprising aryl, lower alkyl, hydroxy, guanidino, oxyalkylether, nitro, sulfur or halo-substituted phenyl (*e.g.*, tyrosine, methyltyrosine and *o*-chloro-, *p*-chloro-, 3,4-dichloro, *o*-, *m*- or *p*-methyl-, 2,4,6-trimethyl-, 2-ethoxy-5-nitro-, 2-hydroxy-5-nitro- and *p*-nitro-phenylalanine); furyl-, thienyl-, pyridyl-, pyrimidinyl-, purinyl- or naphthyl-alanines; and tryptophan analogues and derivatives including kynurenine, 3-hydroxykynurenine, 2-hydroxytryptophan and 4-carboxytryptophan;

α -Amino substituted amino acids including sarcosine (N-methylglycine), N-benzylglycine, N-methylalanine, N-benzylalanine, N-methylphenylalanine, N-benzylphenylalanine, N-methylvaline and N-benzylvaline; and

α -Hydroxy and substituted α -hydroxy amino acids including serine, threonine, allothreonine, phosphoserine and phosphothreonine.

Polypeptides are polymers of amino acids in which a carboxyl group of one amino acid monomer is bonded to an amino or imino group of the next amino acid monomer by an amide bond. Polypeptides include dipeptides, low

molecular weight polypeptides (about 1500-5000 MW) and proteins. Proteins optionally contain 3, 5, 10, 50, 75, 100 or more residues, and suitably are substantially sequence-homologous with human, animal, plant or microbial proteins. They include enzymes (*e.g.*, hydrogen peroxidase) as well as
5 immunogens such as KLH, or antibodies or proteins of any type against which one wishes to raise an immune response. The nature and identity of the polypeptide may vary widely.

The polypeptide amidates are useful as immunogens in raising antibodies against either the polypeptide (if it is not immunogenic in the animal to which it
10 is administered) or against the epitopes on the remainder of the compound of this invention.

Antibodies capable of binding to the parental non-peptidyl compound are used to separate the parental compound from mixtures, for example in diagnosis or manufacturing of the parental compound. The conjugates of parental
15 compound and polypeptide generally are more immunogenic than the polypeptides in closely homologous animals, and therefore make the polypeptide more immunogenic for facilitating raising antibodies against it. Accordingly, the polypeptide or protein may not need to be immunogenic in an animal typically used to raise antibodies, *e.g.*, rabbit, mouse, horse, or rat, but the final product
20 conjugate should be immunogenic in at least one of such animals. The polypeptide optionally contains a peptidolytic enzyme cleavage site at the peptide bond between the first and second residues adjacent to the acidic heteroatom. Such cleavage sites are flanked by enzymatic recognition structures, *e.g.*, a particular sequence of residues recognized by a peptidolytic
25 enzyme.

Peptidolytic enzymes for cleaving the polypeptide conjugates of this invention are well known, and in particular include carboxypeptidases. Carboxypeptidases digest polypeptides by removing C-terminal residues, and are specific in many instances for particular C-terminal sequences. Such enzymes
30 and their substrate requirements in general are well known. For example, a dipeptide (having a given pair of residues and a free carboxyl terminus) is covalently bonded through its α -amino group to the phosphorus or carbon atoms

of the compounds herein. In claims where W₁ is phosphonate it is expected that this peptide will be cleaved by the appropriate peptidolytic enzyme, leaving the carboxyl of the proximal amino acid residue to autocatalytically cleave the phosphonoamidate bond.

5 Suitable dipeptidyl groups (designated by their single letter code) are
 AA, AR, AN, AD, AC, AE, AQ, AG, AH, AI, AL, AK, AM, AF, AP, AS, AT,
 AW, AY, AV, RA, RR, RN, RD, RC, RE, RQ, RG, RH, RI, RL, RK, RM, RF,
 RP, RS, RT, RW, RY, RV, NA, NR, NN, ND, NC, NE, NQ, NG, NH, NI, NL,
 NK, NM, NF, NP, NS, NT, NW, NY, NV, DA, DR, DN, DD, DC, DE, DQ, DG,
 10 DH, DI, DL, DK, DM, DF, DP, DS, DT, DW, DY, DV, CA, CR, CN, CD, CC,
 CE, CQ, CG, CH, CI, CL, CK, CM, CF, CP, CS, CT, CW, CY, CV, EA, ER,
 EN, ED, EC, EE, EQ, EG, EH, EI, EL, EK, EM, EF, EP, ES, ET, EW, EY, EV,
 QA, QR, QN, QD, QC, QE, QQ, QG, QH, QI, QL, QK, QM, QF, QP, QS, QT,
 QW, QY, QV, GA, GR, GN, GD, GC, GE, GQ, GG, GH, GI, GL, GK, GM, GF,
 15 GP, GS, GT, GW, GY, GV, HA, HR, HN, HD, HC, HE, HQ, HG, HH, HI, HL,
 HK, HM, HF, HP, HS, HT, HW, HY, HV, IA, IR, IN, ID, IC, IE, IQ, IG, IH, II,
 IL, IK, IM, IF, IP, IS, IT, IW, IY, IV, LA, LR, LN, LD, LC, LE, LQ, LG, LH,
 LI, LL, LK, LM, LF, LP, LS, LT, LW, LY, LV, KA, KR, KN, KD, KC, KE,
 KQ, KG, KH, KI, KL, KK, KM, KF, KP, KS, KT, KW, KY, KV, MA, MR,
 20 MN, MD, MC, ME, MQ, MG, MH, MI, ML, MK, MM, MF, MP, MS, MT,
 MW, MY, MV, FA, FR, FN, FD, FC, FE, FQ, FG, FH, FI, FL, FK, FM, FF, FP,
 FS, FT, FW, FY, FV, PA, PR, PN, PD, PC, PE, PQ, PG, PH, PI, PL, PK, PM,
 PF, PP, PS, PT, PW, PY, PV, SA, SR, SN, SD, SC, SE, SQ, SG, SH, SI, SL, SK,
 SM, SF, SP, SS, ST, SW, SY, SV, TA, TR, TN, TD, TC, TE, TQ, TG, TH, TI,
 25 TL, TK, TM, TF, TP, TS, TT, TW, TY, TV, WA, WR, WN, WD, WC, WE,
 WQ, WG, WH, WI, WL, WK, WM, WF, WP, WS, WT, WW, WY, WV, YA,
 YR, YN, YD, YC, YE, YQ, YG, YH, YI, YL, YK, YM, YF, YP, YS, YT, YW,
 YY, YV, VA, VR, VN, VD, VC, VE, VQ, VG, VH, VI, VL, VK, VM, VF, VP,
 VS, VT, VW, VY and VV.

30 Tripeptide residues are also useful as protecting groups. When a phosphonate is to be protected, the sequence -X⁴-pro-X⁵- (where X⁴ is any amino acid residue and X⁵ is an amino acid residue, a carboxyl ester of proline, or

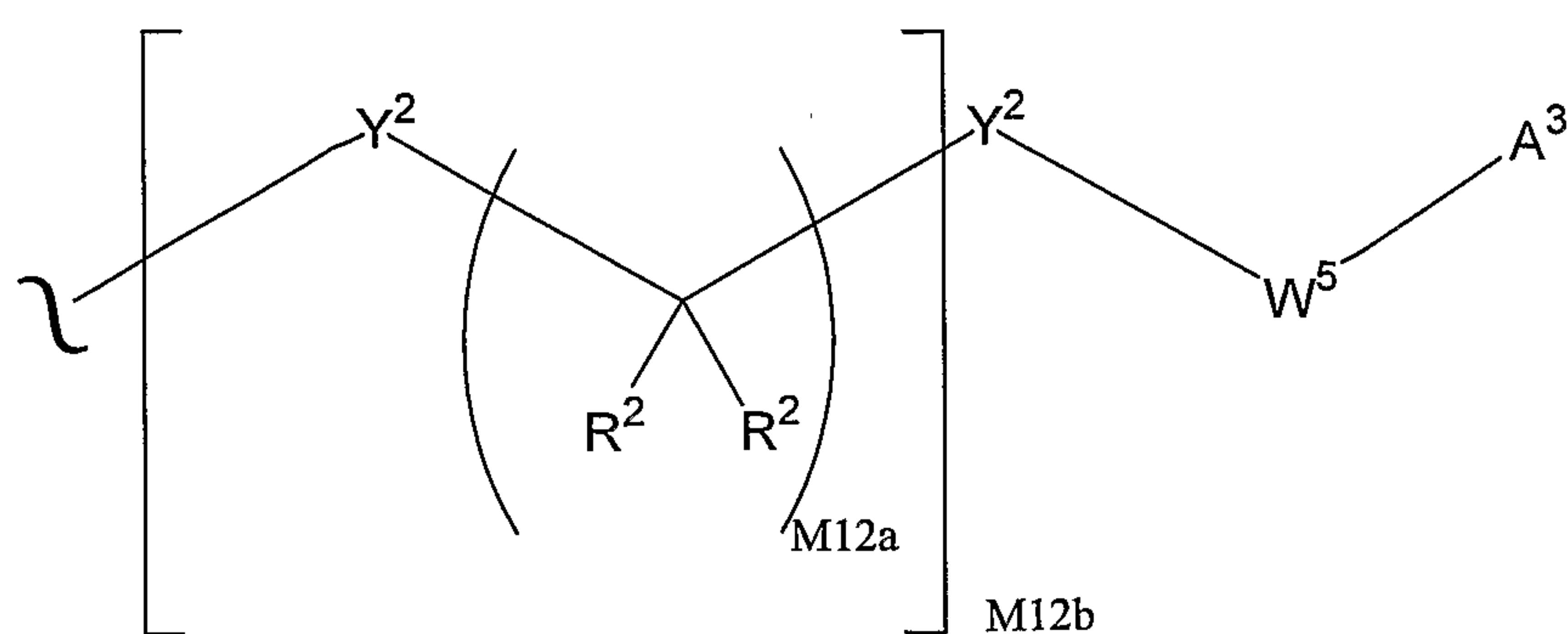
hydrogen) will be cleaved by luminal carboxypeptidase to yield X^4 with a free carboxyl, which in turn is expected to autocatalytically cleave the phosphonoamidate bond. The carboxy group of X^5 optionally is esterified with benzyl.

5 Dipeptide or tripeptide species can be selected on the basis of known transport properties and/or susceptibility to peptidases that can affect transport to intestinal mucosal or other cell types. Dipeptides and tripeptides lacking an α -amino group are transport substrates for the peptide transporter found in brush border membrane of intestinal mucosal cells (Bai, J.P.F., (1992) *Pharm Res.*
10 9:969-978). Transport competent peptides can thus be used to enhance bioavailability of the amidate compounds. Di- or tripeptides having one or more amino acids in the D configuration are also compatible with peptide transport and can be utilized in the amidate compounds of this invention. Amino acids in the D configuration can be used to reduce the susceptibility of a di- or tripeptide
15 to hydrolysis by proteases common to the brush border such as aminopeptidase N. In addition, di- or tripeptides alternatively are selected on the basis of their relative resistance to hydrolysis by proteases found in the lumen of the intestine. For example, tripeptides or polypeptides lacking asp and/or glu are poor substrates for aminopeptidase A, di- or tripeptides lacking amino acid residues
20 on the N-terminal side of hydrophobic amino acids (leu, tyr, phe, val, trp) are poor substrates for endopeptidase, and peptides lacking a pro residue at the penultimate position at a free carboxyl terminus are poor substrates for carboxypeptidase P. Similar considerations can also be applied to the selection of peptides that are either relatively resistant or relatively susceptible to
25 hydrolysis by cytosolic, renal, hepatic, serum or other peptidases. Such poorly cleaved polypeptide amidates are immunogens or are useful for bonding to proteins in order to prepare immunogens.

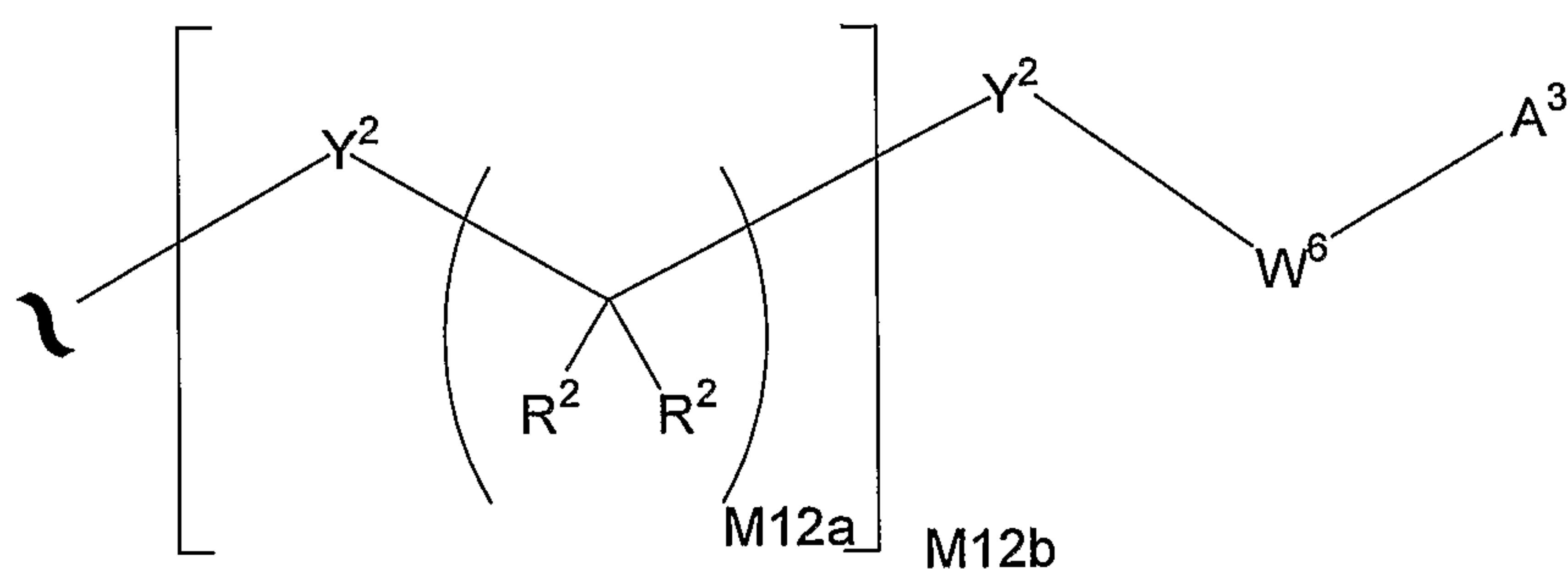
Specific Embodiments of the Invention

Specific values described for radicals, substituents, and ranges, as well as
30 specific embodiments of the invention described herein, are for illustration only; they do not exclude other defined values or other values within defined ranges.

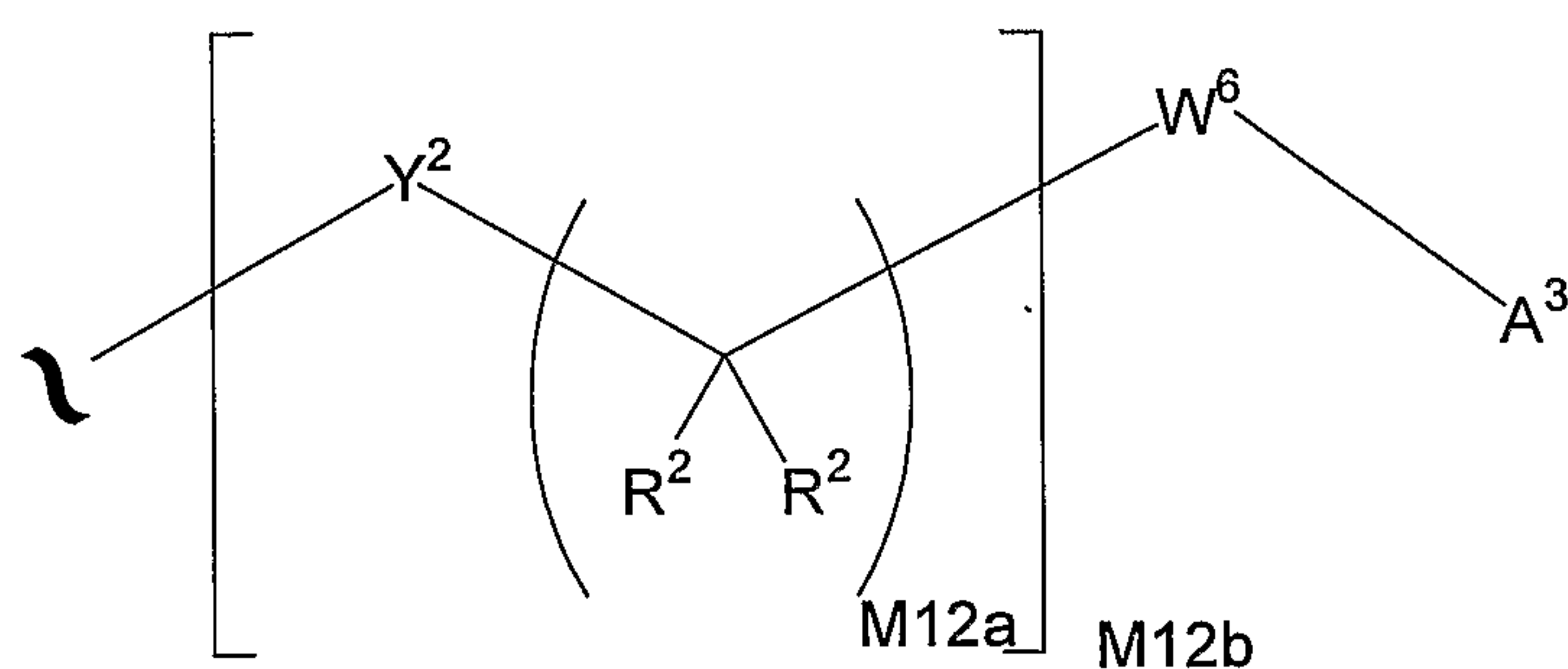
In some embodiments of the invention, A^1 is of the formula:



In some embodiments of the invention, A¹ is of the formula:

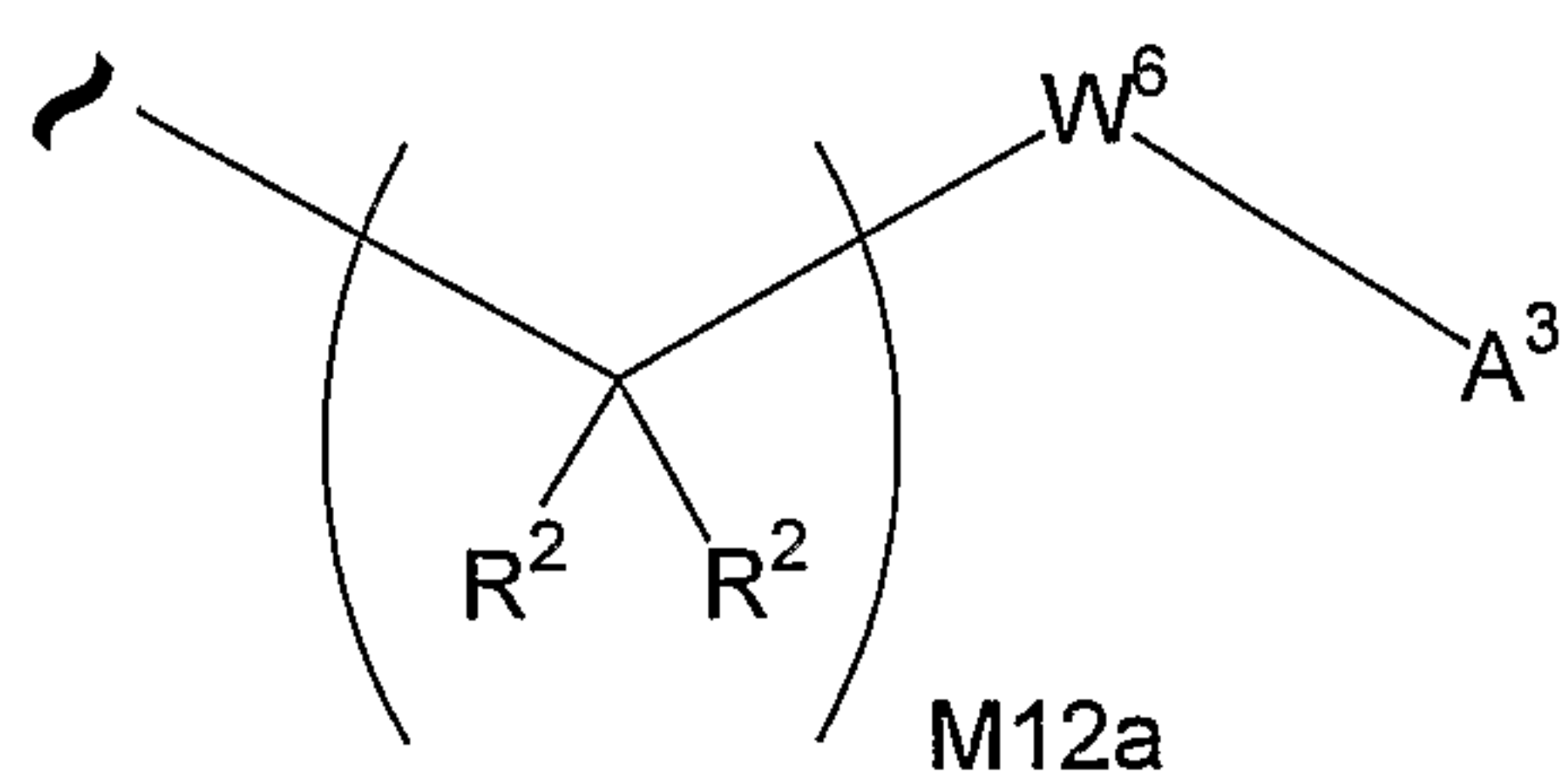


In some embodiments of the invention, A¹ is of the formula:

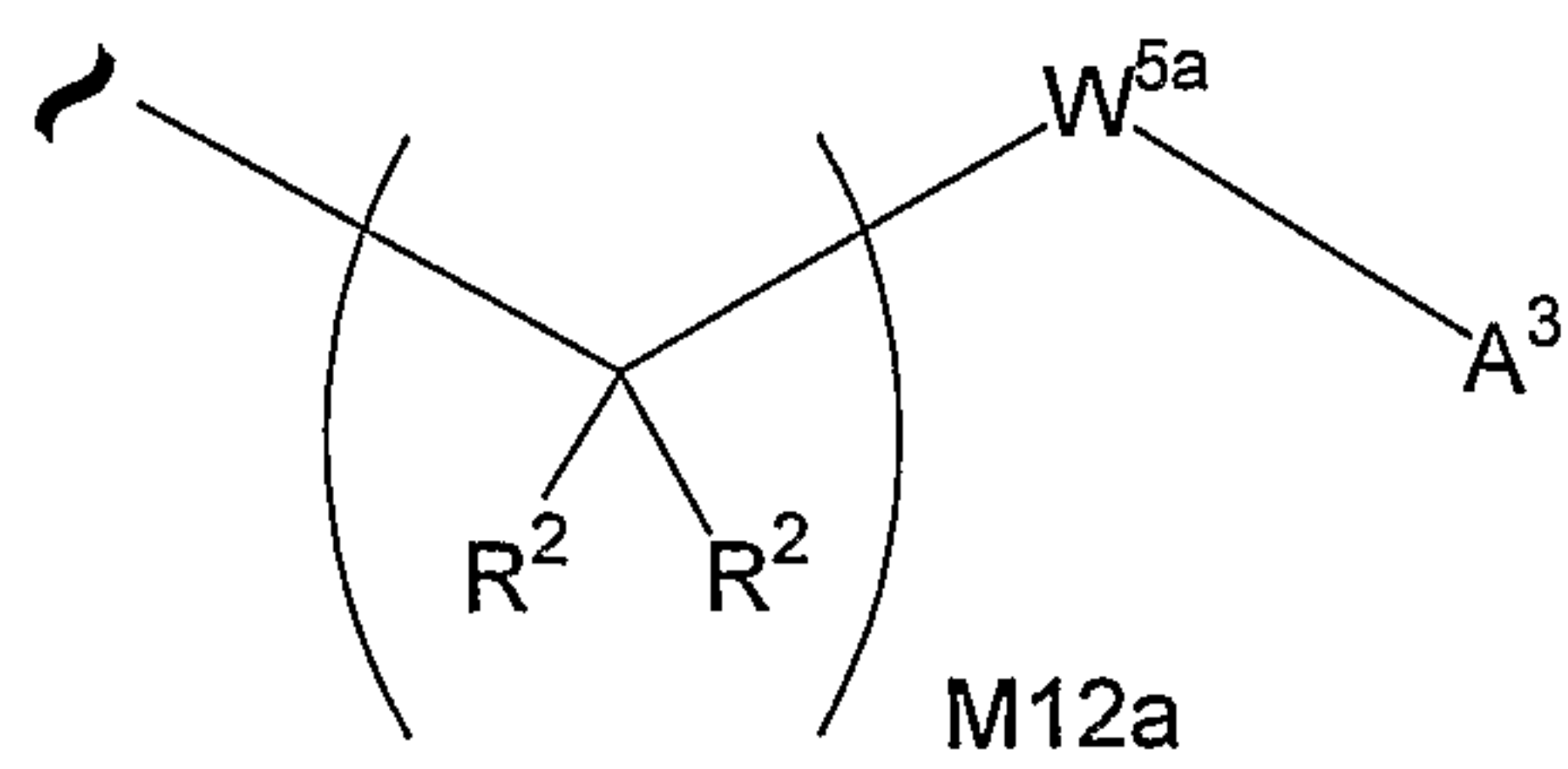


5

In some embodiments of the invention, A¹ is of the formula:



In some embodiments of the invention, A¹ is of the formula:

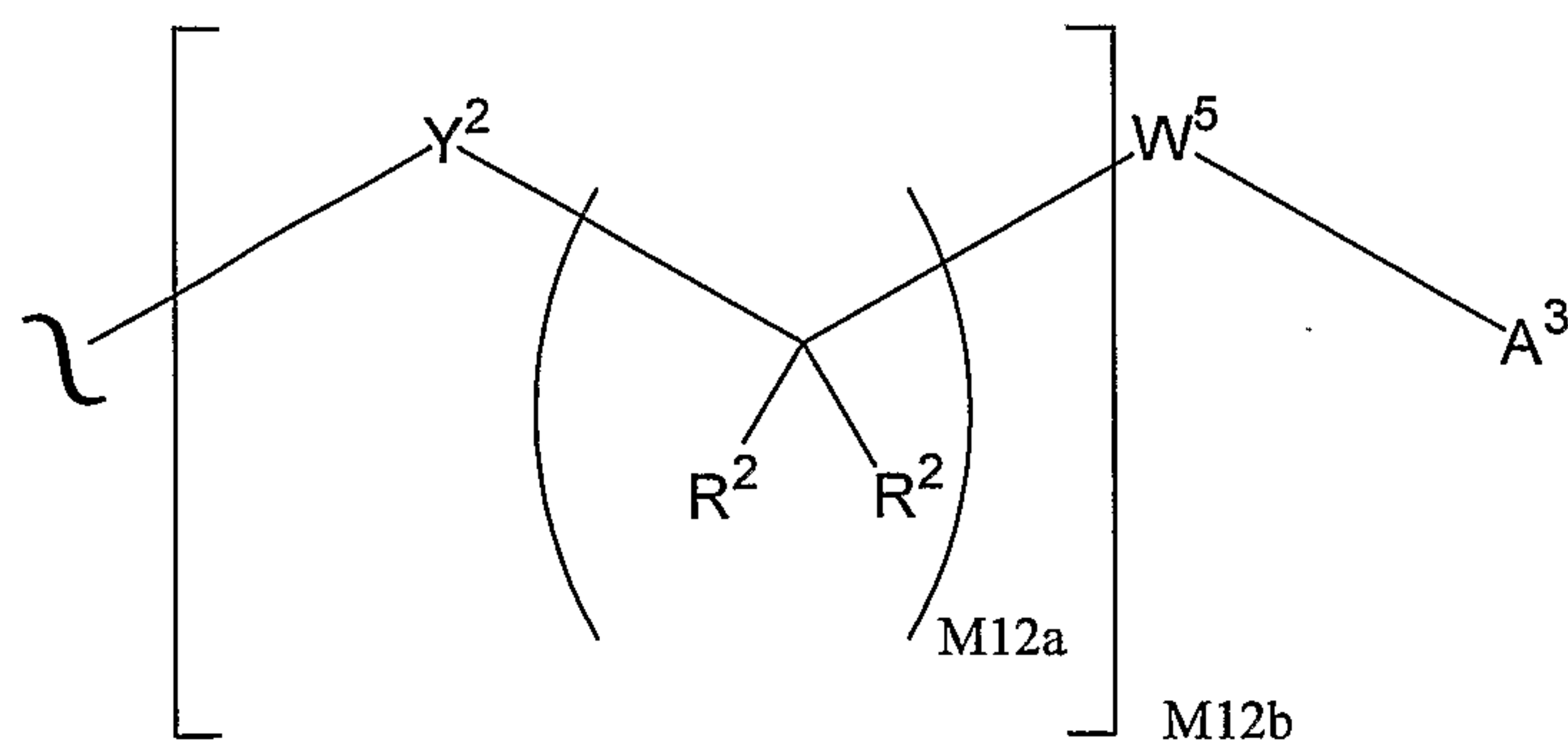


10

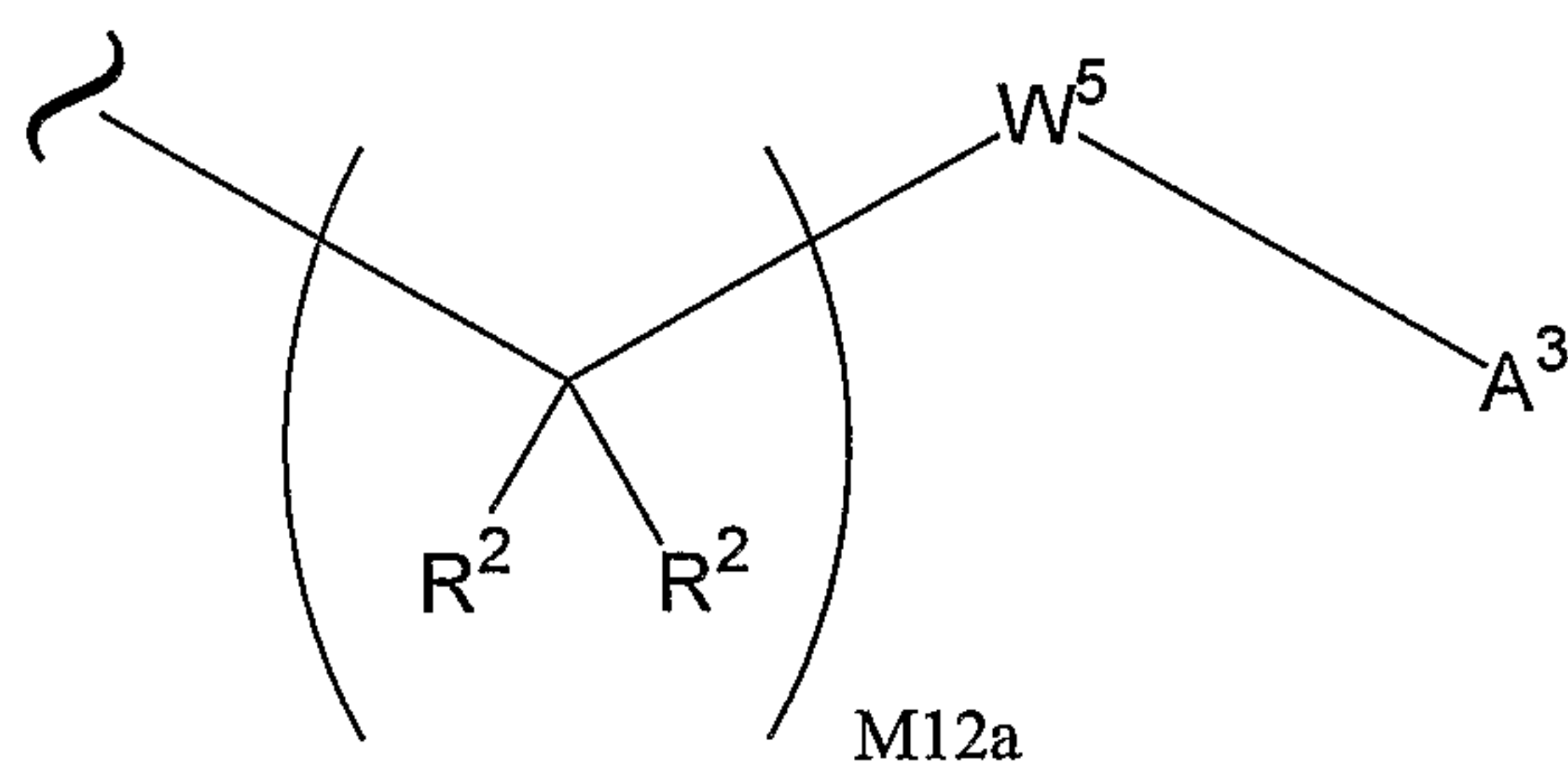
and W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently substituted with 0 or 1 R² groups.

In some embodiments of the invention, M12a is 1.

In some embodiments of the invention, A¹ is of the formula:

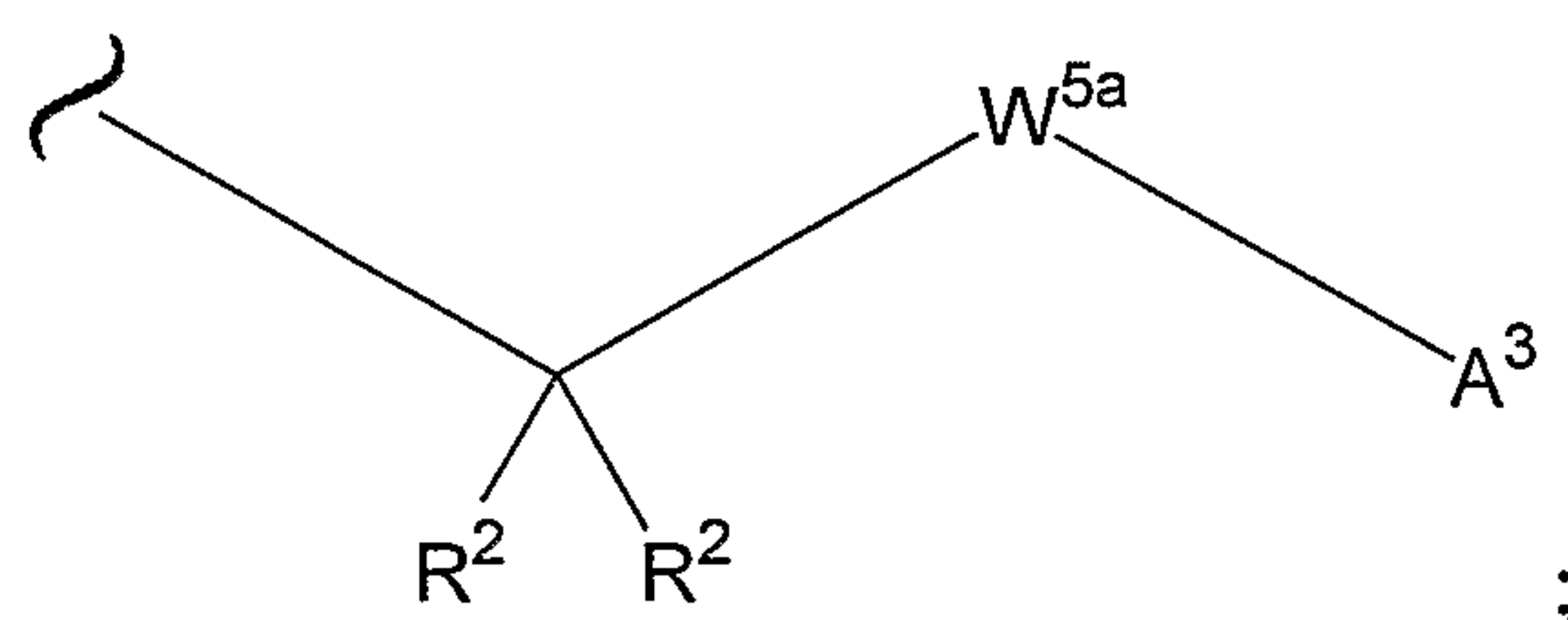


In some embodiments of the invention, A¹ is of the formula:



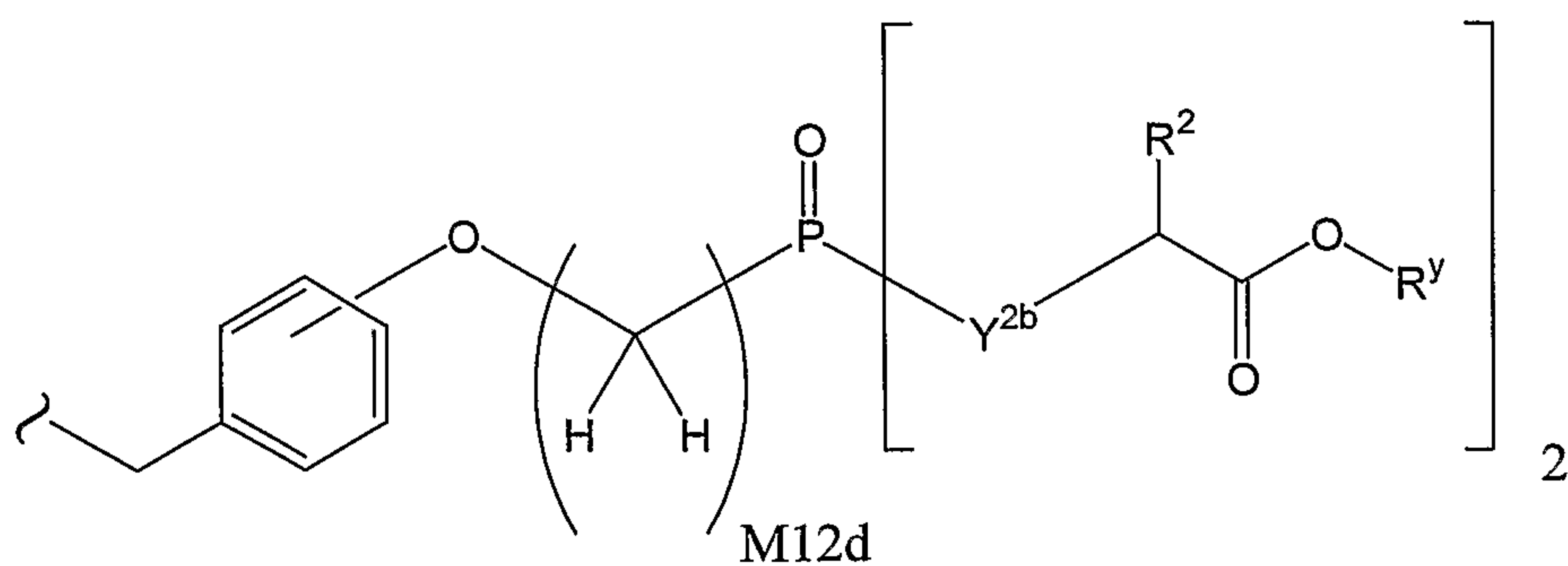
5

In some embodiments of the invention, A¹ is of the formula:



and W^{5a} is a carbocycle independently substituted with 0 or 1 R² groups.

In some embodiments of the invention, A¹ is of the formula:

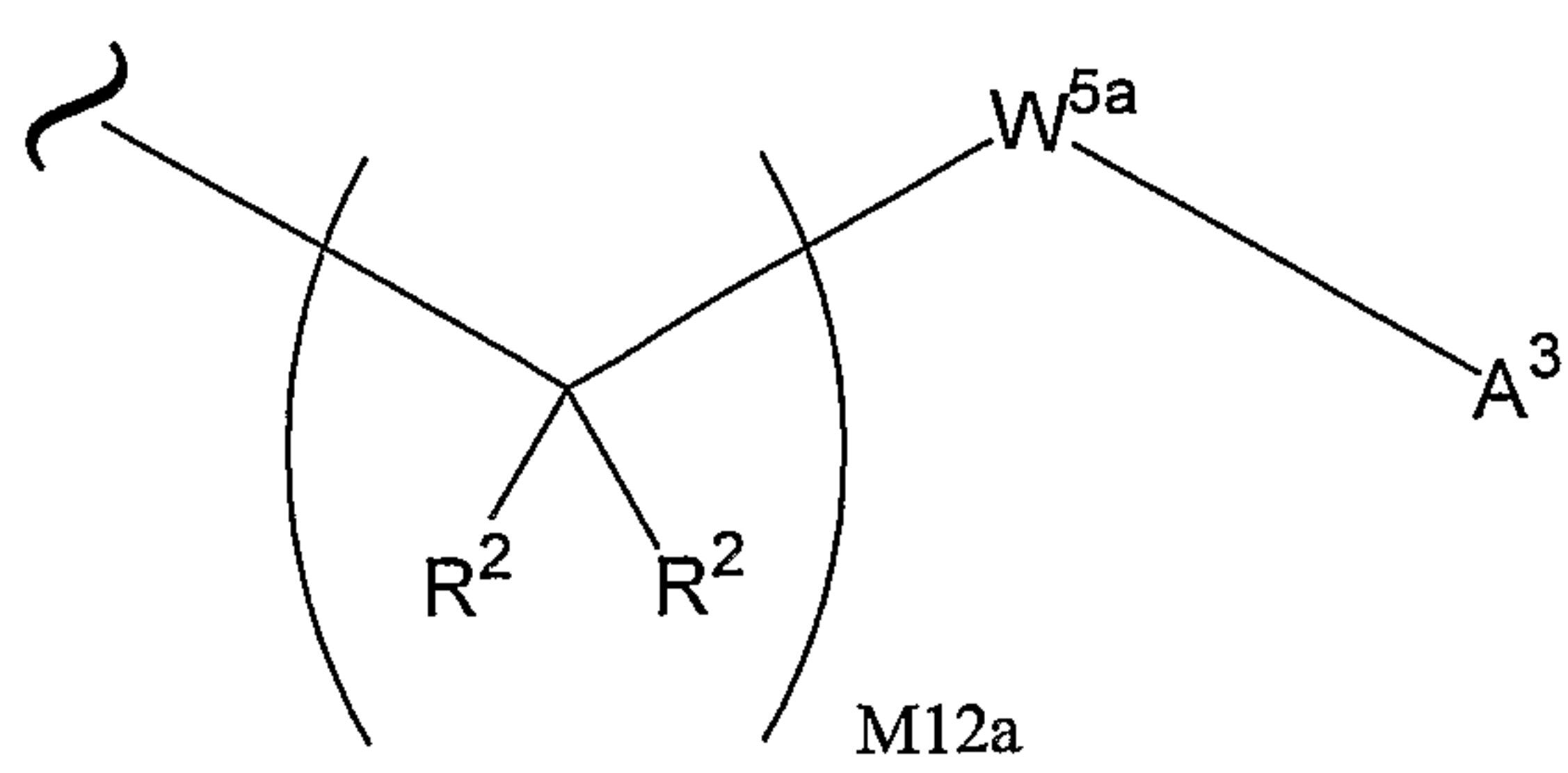


10

Y^{2b} is O or N(R²); and

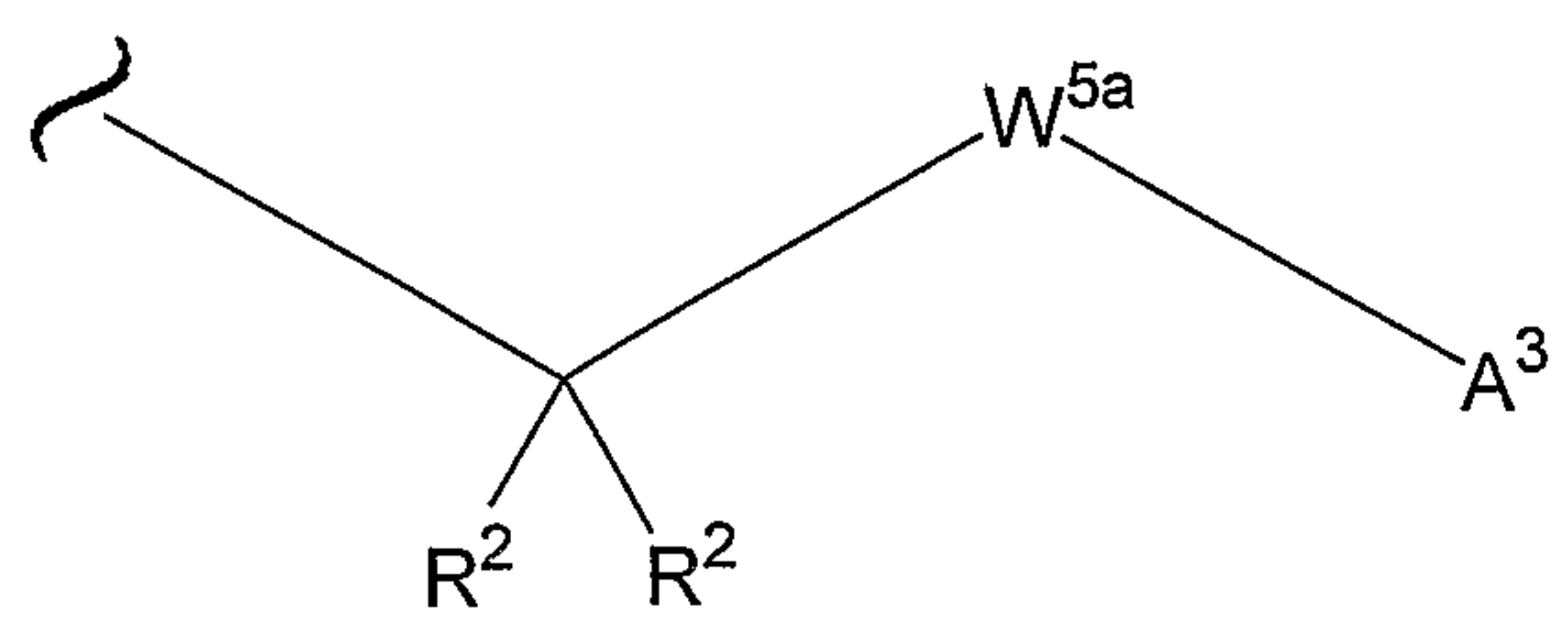
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

In some embodiments of the invention, A¹ is of the formula:



and W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

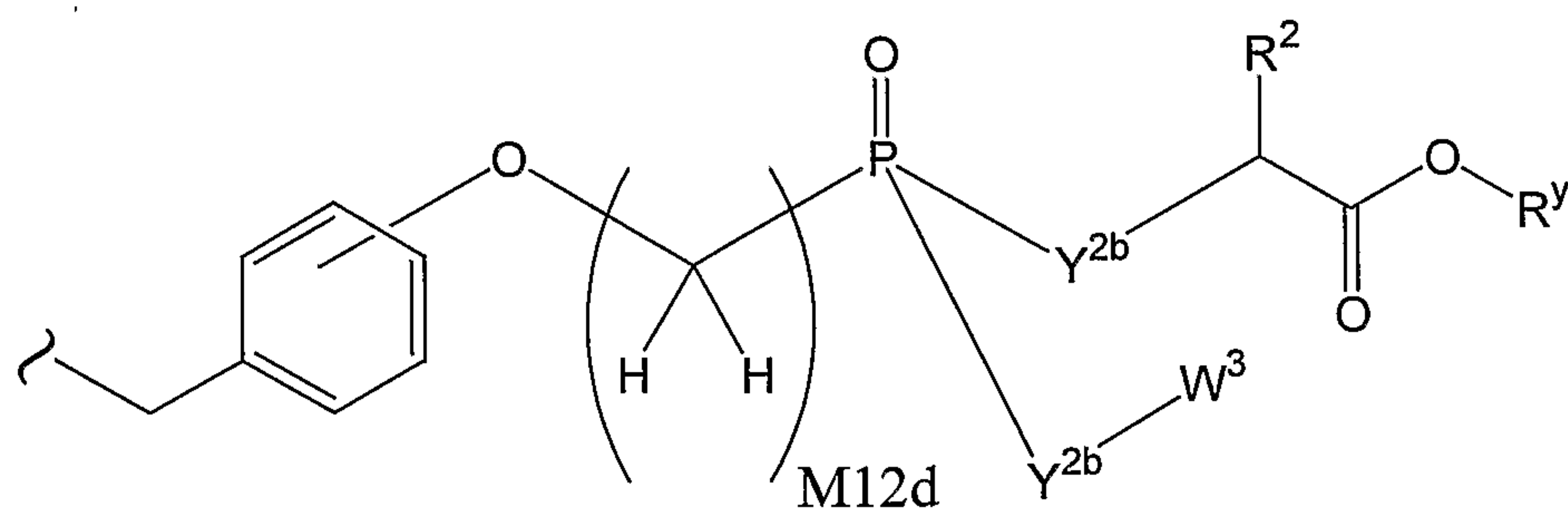
In some embodiments of the invention, A^1 is of the formula:



5

and W^{5a} is a carbocycle or heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^1 is of the formula:

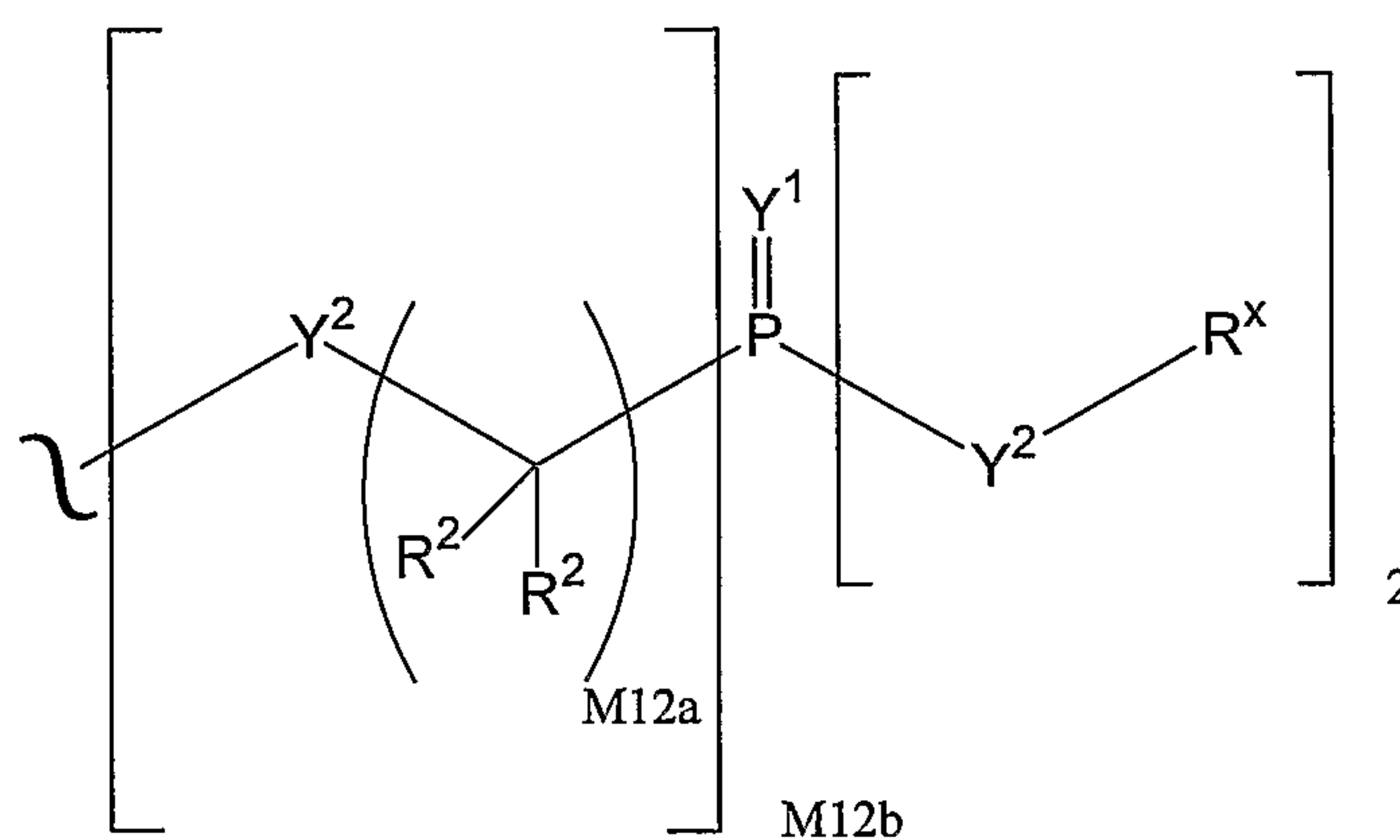


10

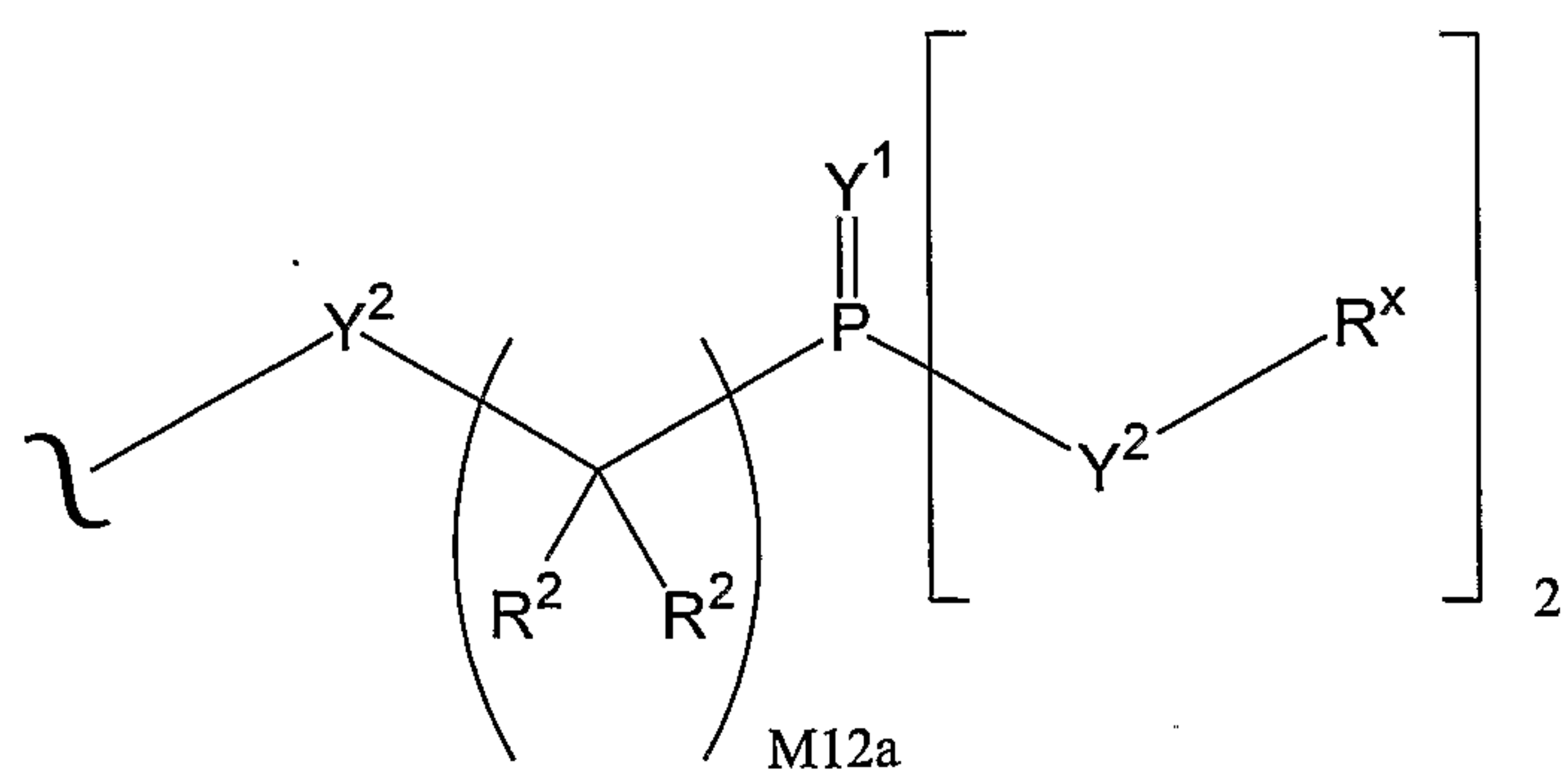
Y^{2b} is independently O or $N(R^2)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

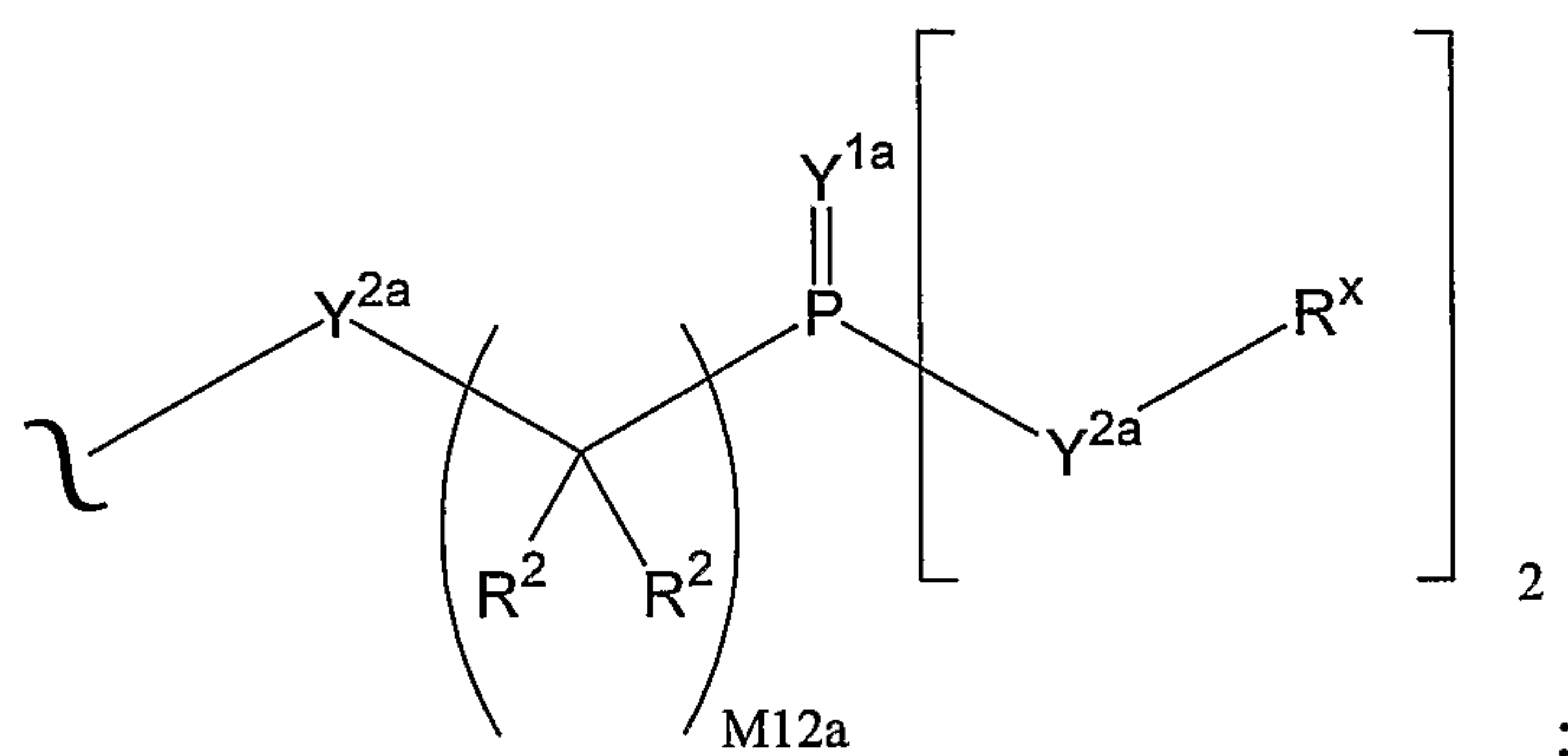
In some embodiments of the invention, A^3 is of the formula:



In some embodiments of the invention, A^3 is of the formula:



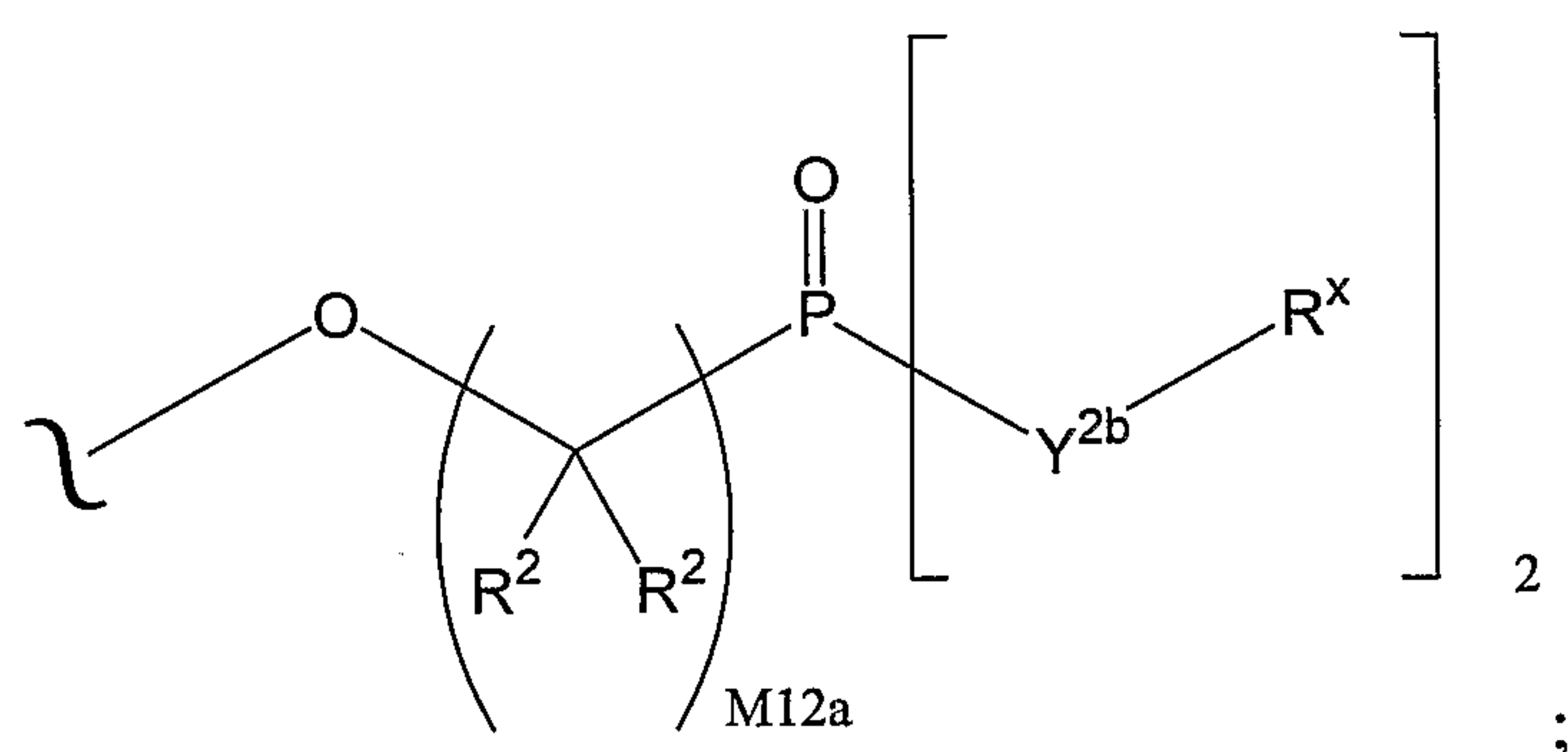
In some embodiments of the invention, A³ is of the formula:



Y^{1a} is O or S; and

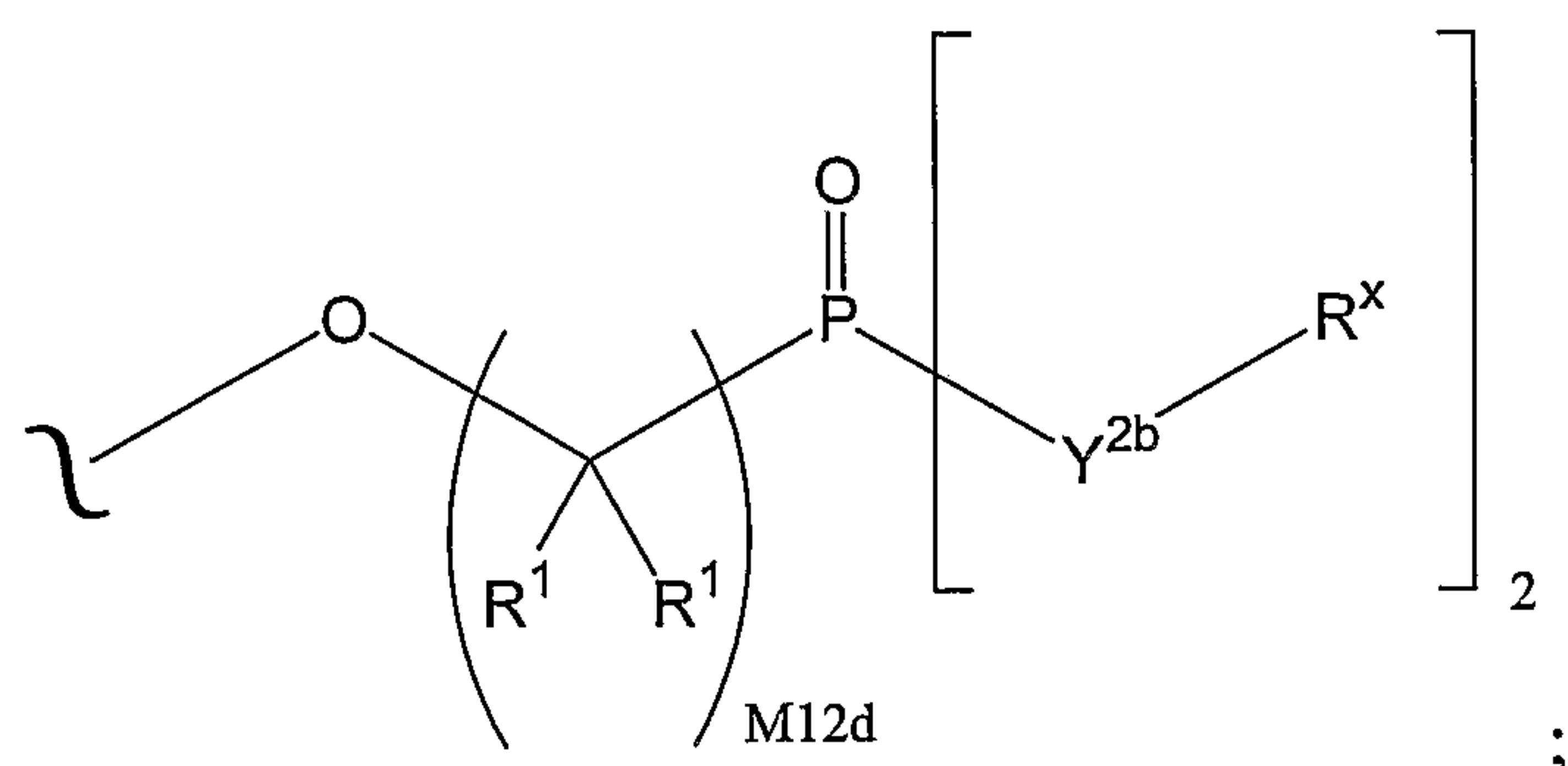
5 Y^{2a} is independently O, N(R^x) or S.

In some embodiments of the invention, A³ is of the formula:



and Y^{2b} is O or N(R^x).

In some embodiments of the invention, A³ is of the formula:

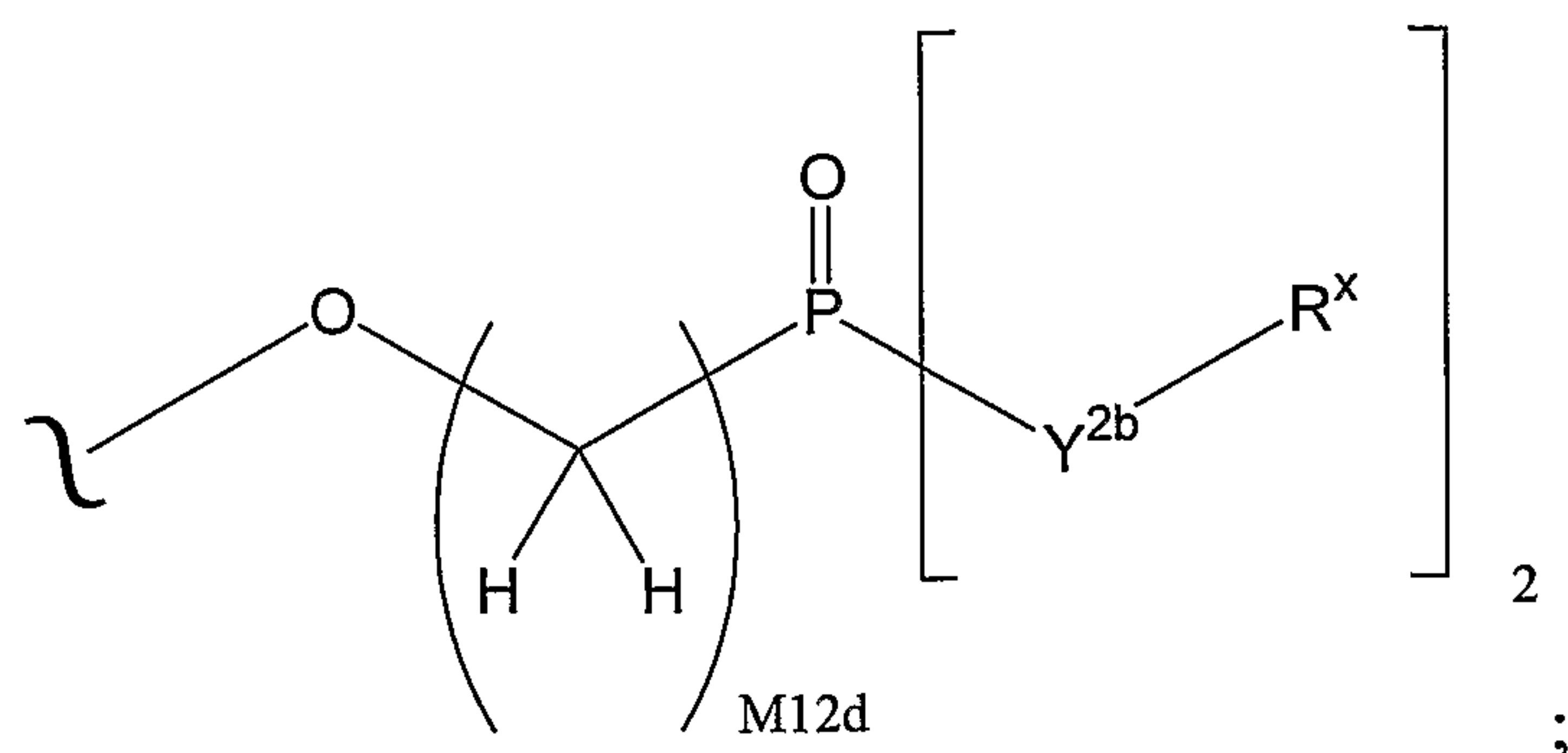


R¹ is independently H or alkyl of 1 to 18 carbon atoms;

Y^{2b} is O or $N(R^x)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

In some embodiments of the invention, A^3 is of the formula:



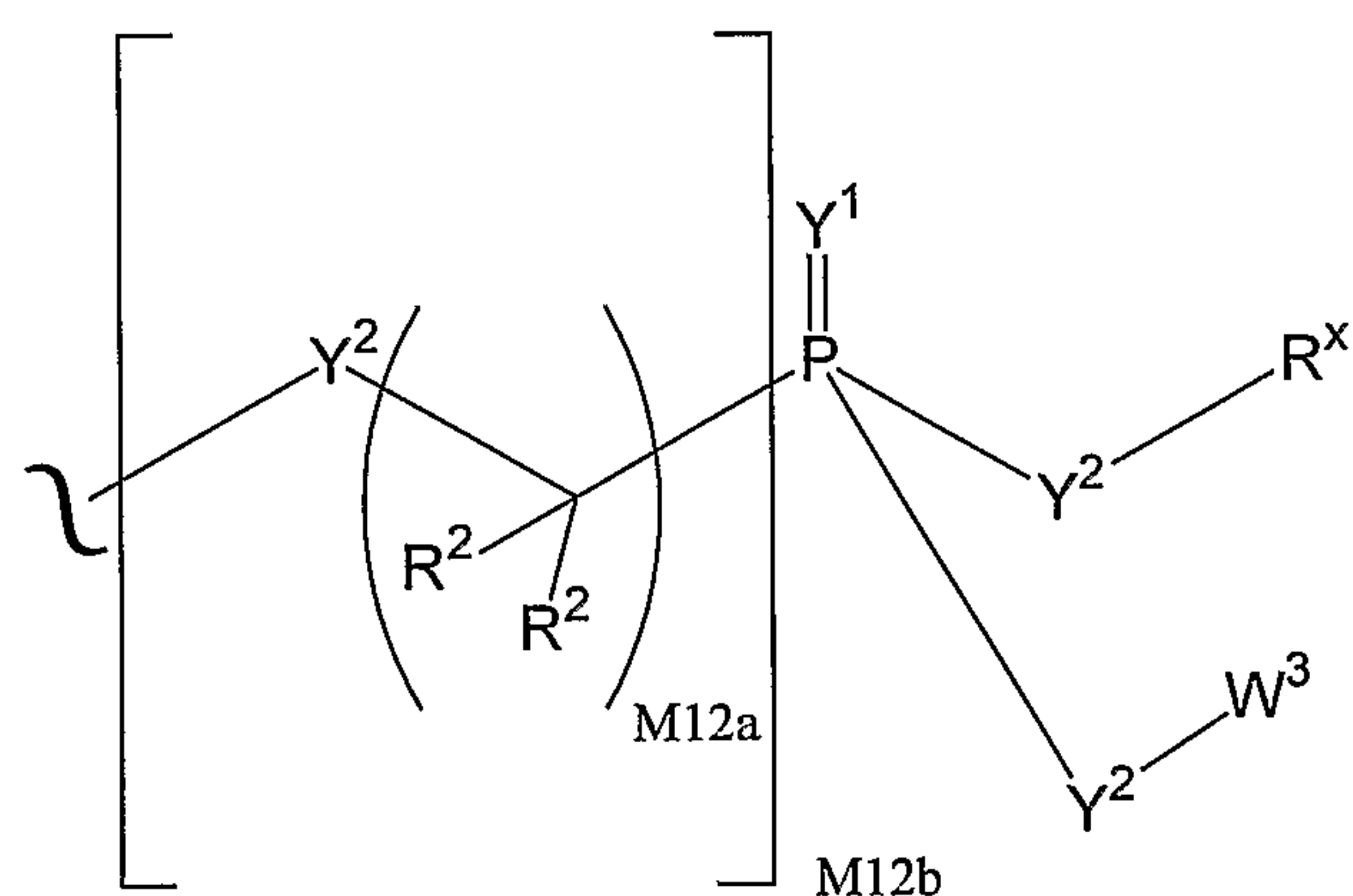
5

Y^{2b} is O or $N(R^x)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

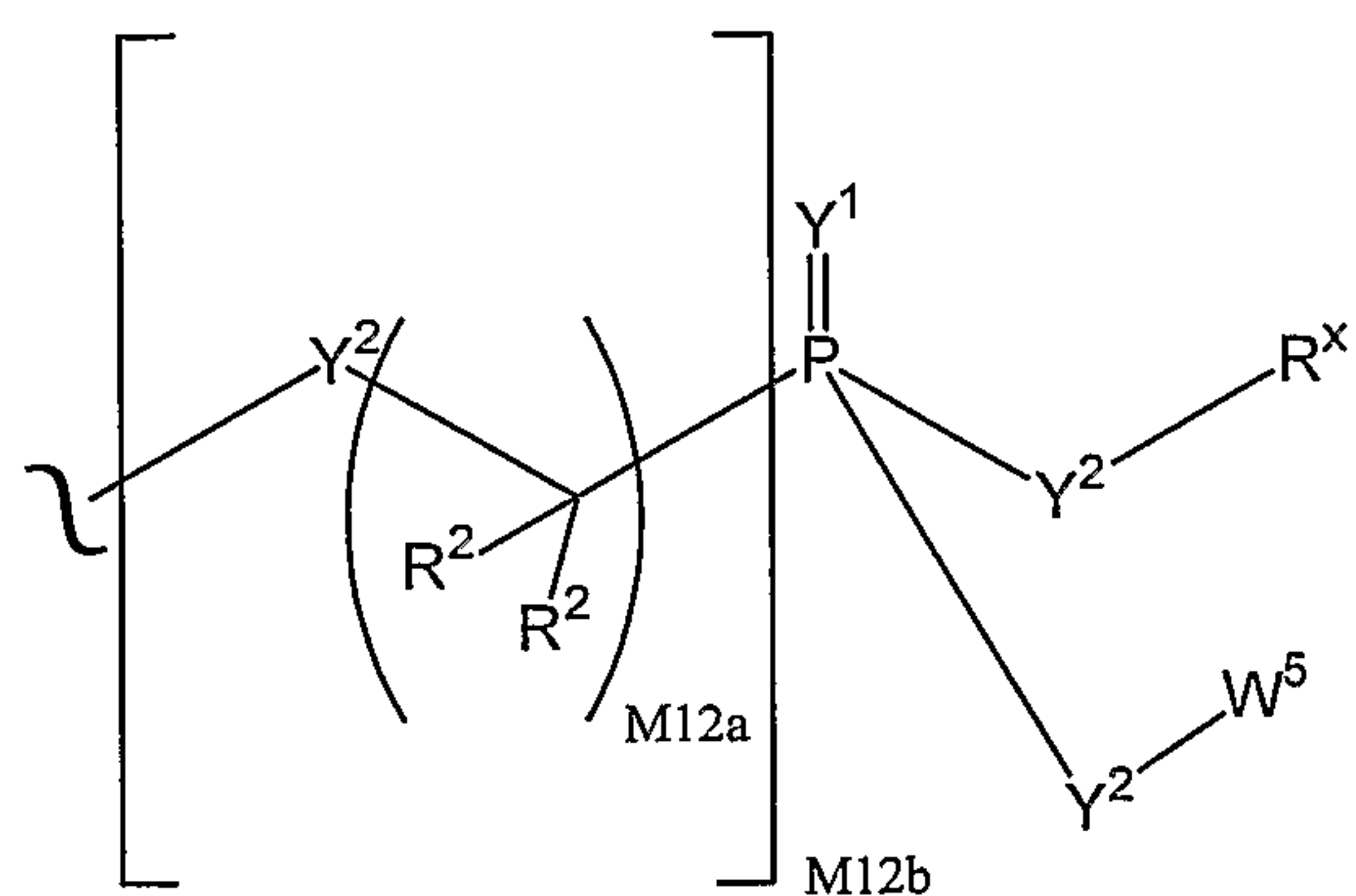
In some embodiments of the invention, M12d is 1.

In some embodiments of the invention, A^3 is of the formula:



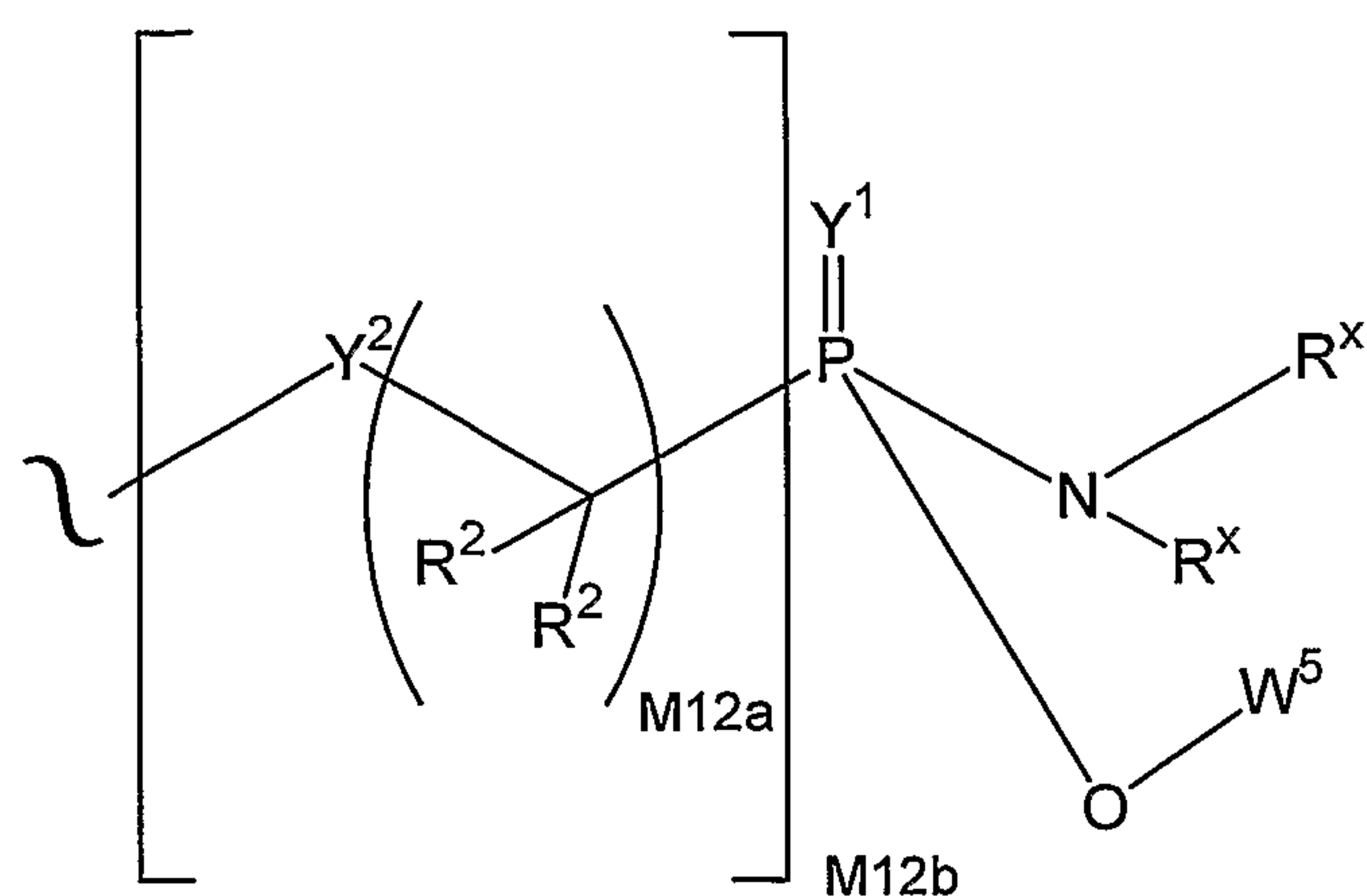
10

In some embodiments of the invention, A^3 is of the formula:



In some embodiments of the invention, W^5 is a carbocycle.

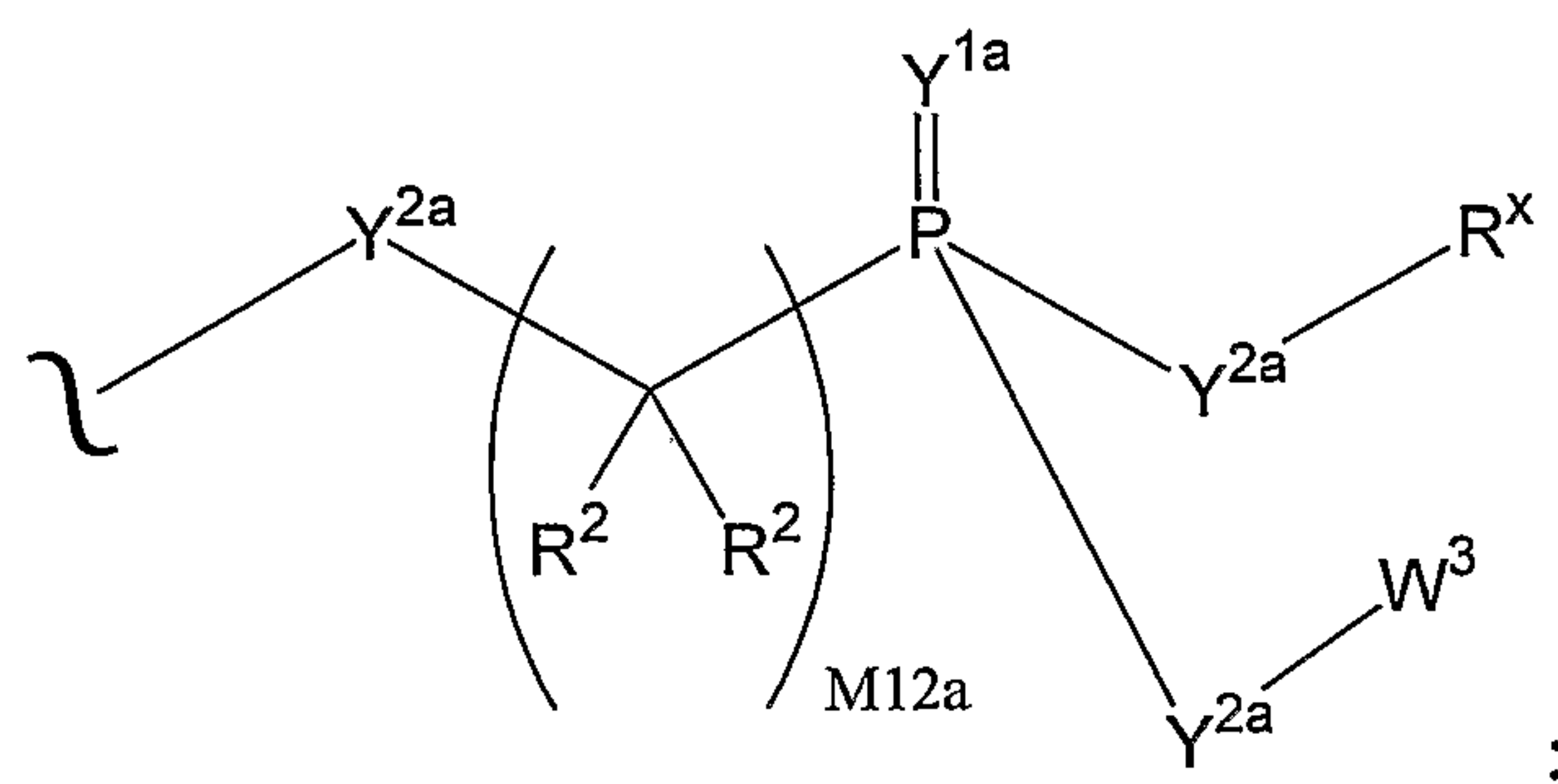
In some embodiments of the invention, A^3 is of the formula:



In some embodiments of the invention, W^5 is phenyl.

In some embodiments of the invention, M12b is 1.

In some embodiments of the invention, A^3 is of the formula:

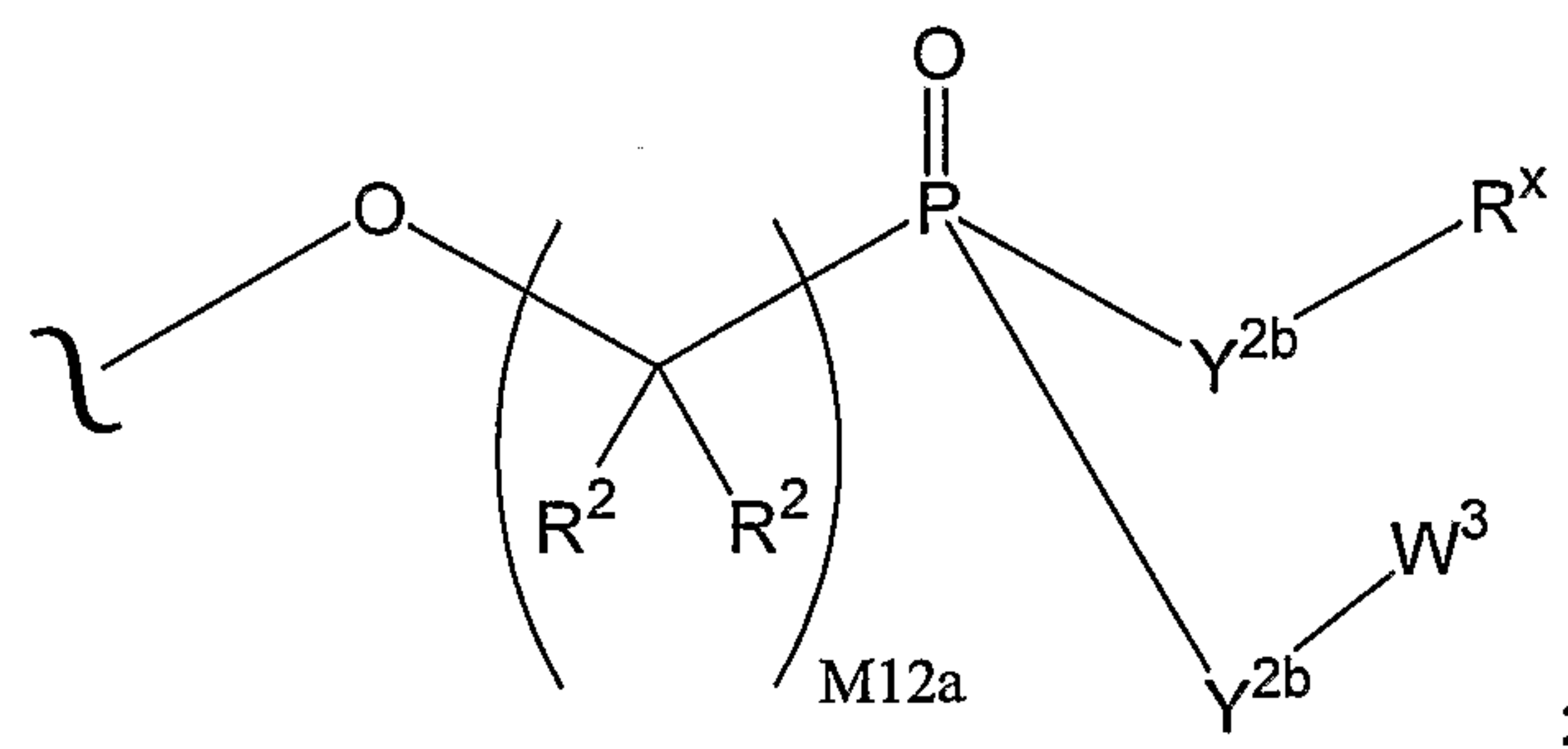


5

Y^{1a} is O or S; and

Y^{2a} is independently O, $N(R^x)$ or S.

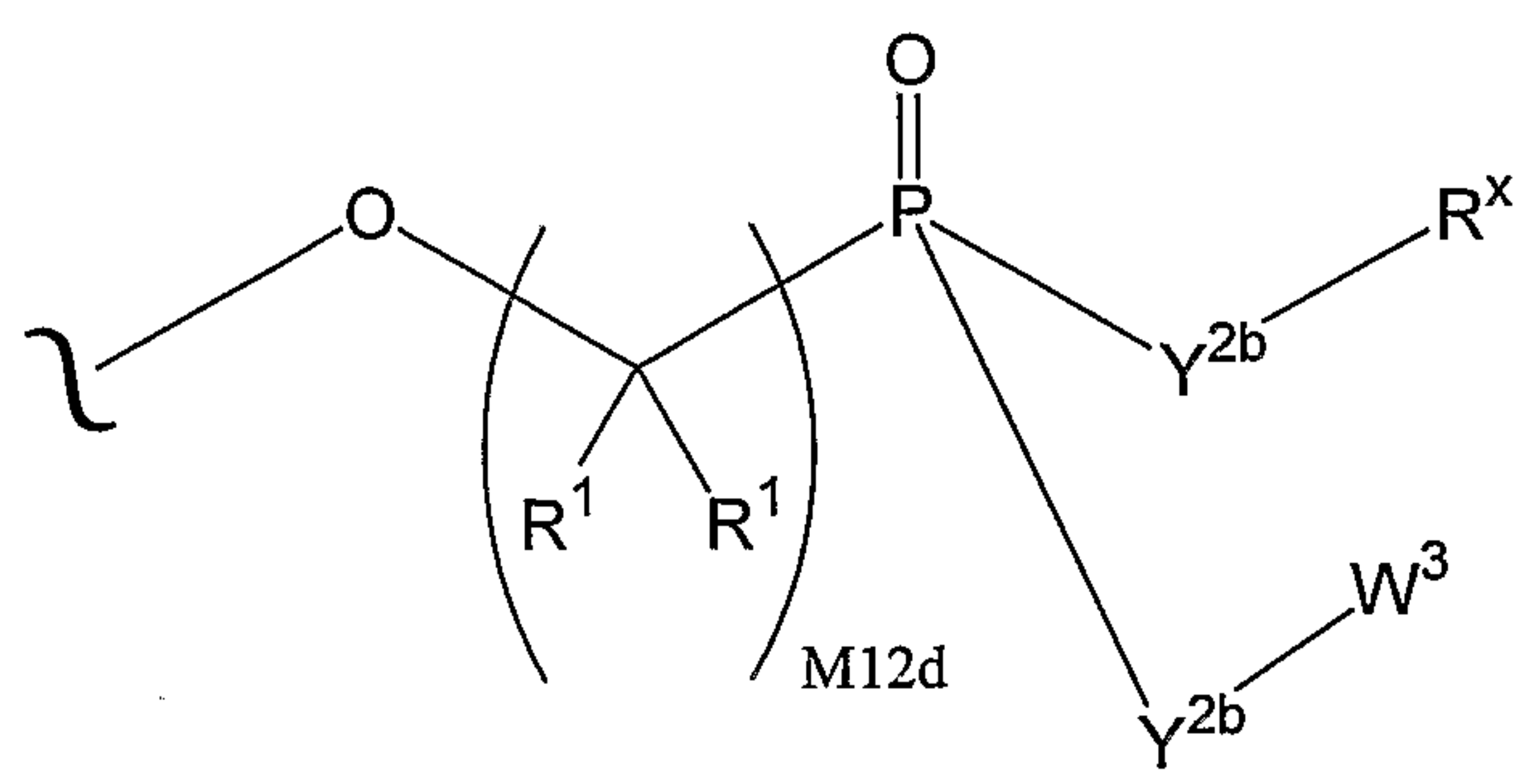
In some embodiments of the invention, A^3 is of the formula:



10

and Y^{2b} is independently O or $N(R^x)$.

In some embodiments of the invention, A^3 is of the formula:



R^1 is independently H or alkyl of 1 to 18 carbon atoms;

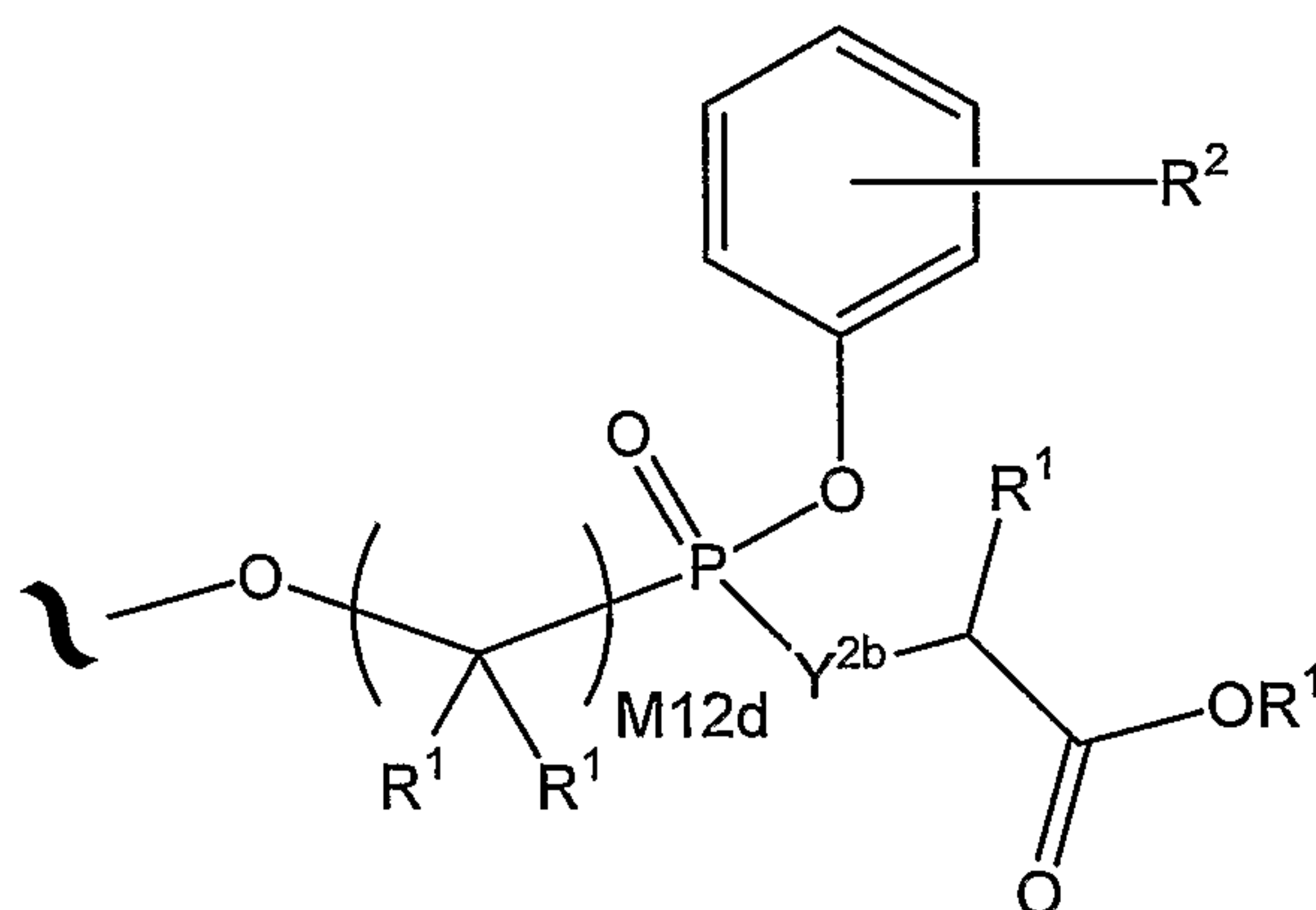
Y^{2b} is independently O or $N(R^x)$; and

5 M_{12d} is 1, 2, 3, 4, 5, 6, 7 or 8.

In some embodiments of the invention, R^1 is H.

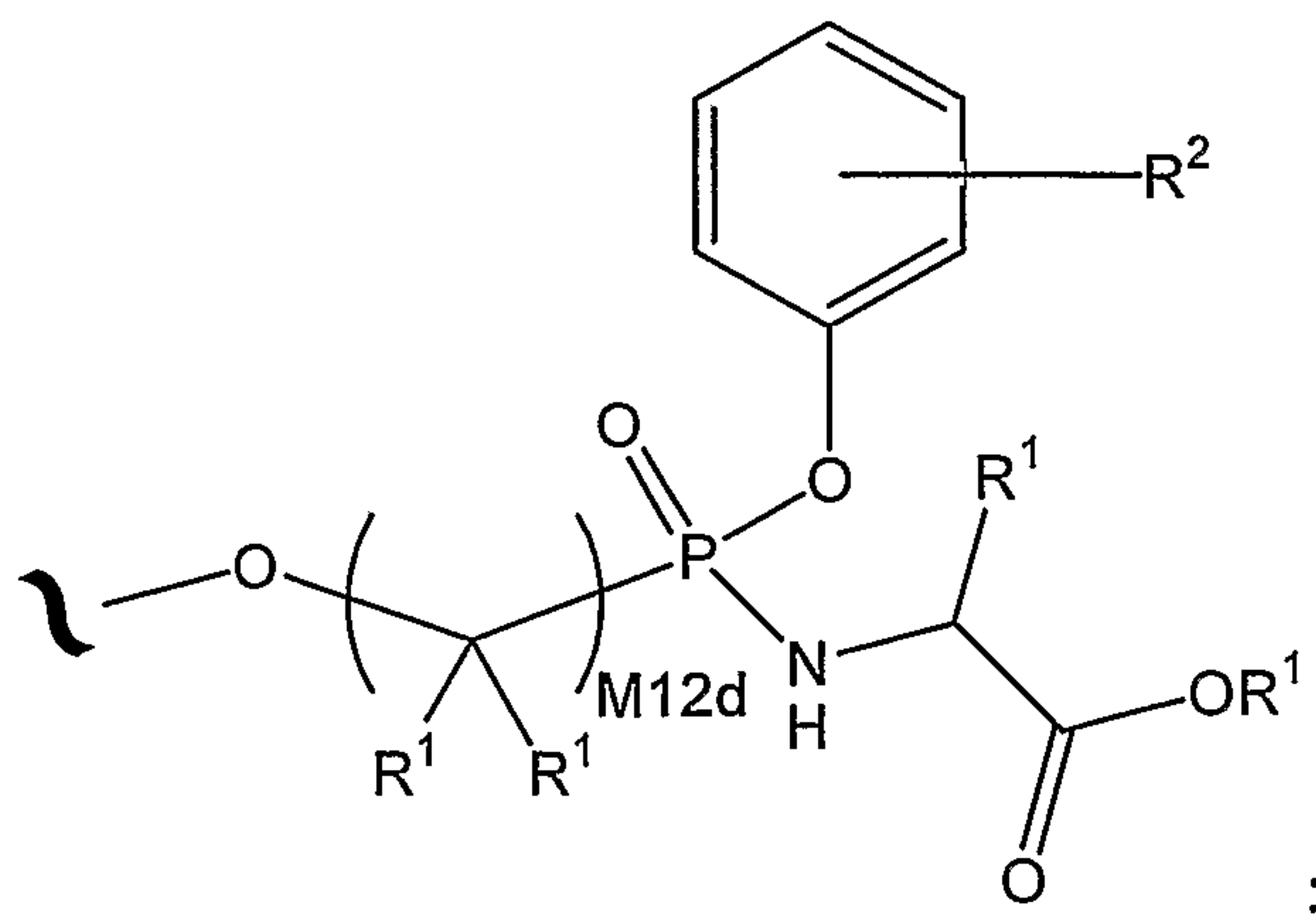
In some embodiments of the invention, M_{12d} is 1.

In some embodiments of the invention, A^3 is of the formula:



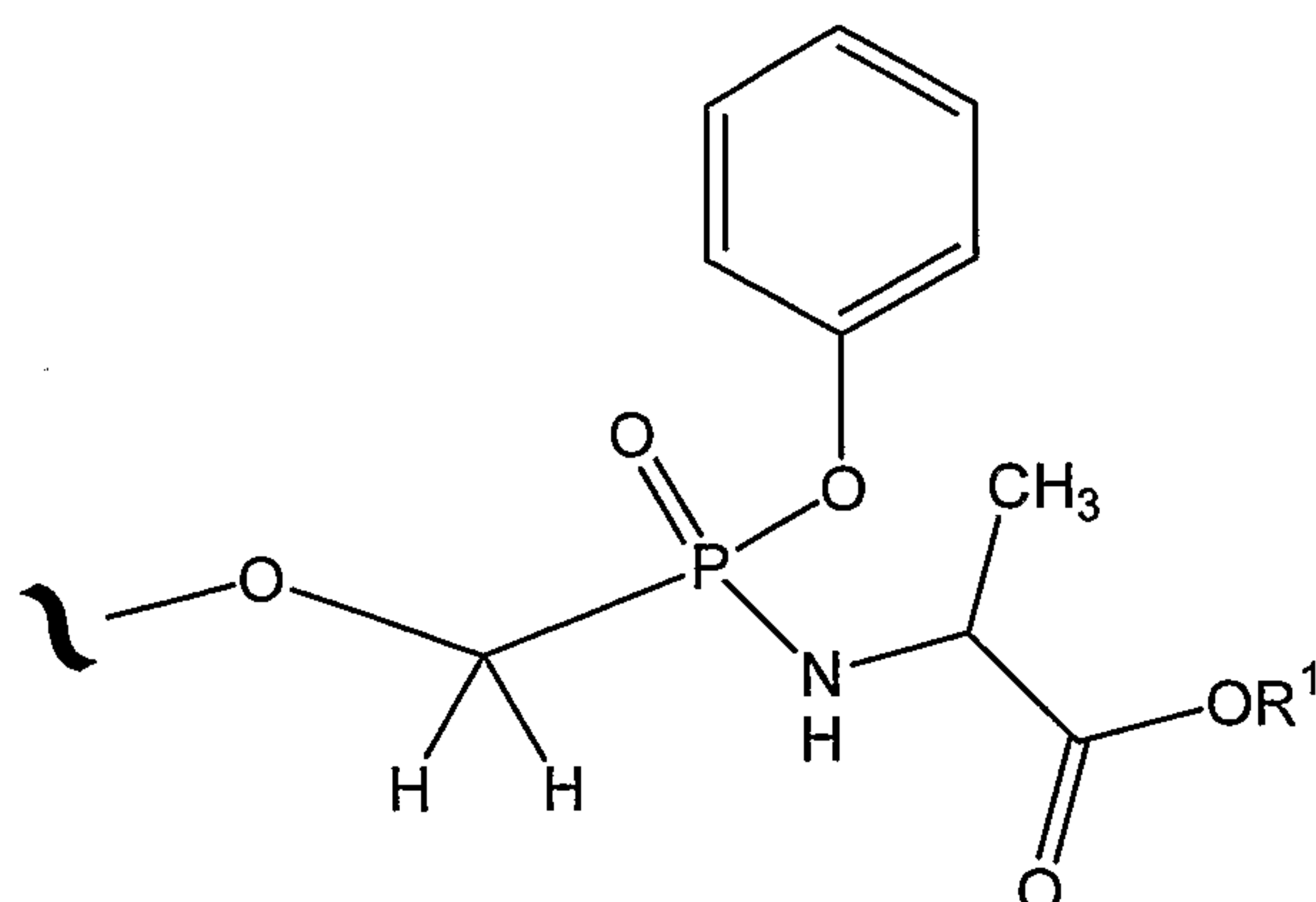
10 wherein R^1 is independently H or alkyl of 1 to 18 carbon atoms;
and the phenyl carbocycle is substituted with 0, 1, 2, or 3 R^2 groups.

In some embodiments of the invention, A^3 is of the formula:



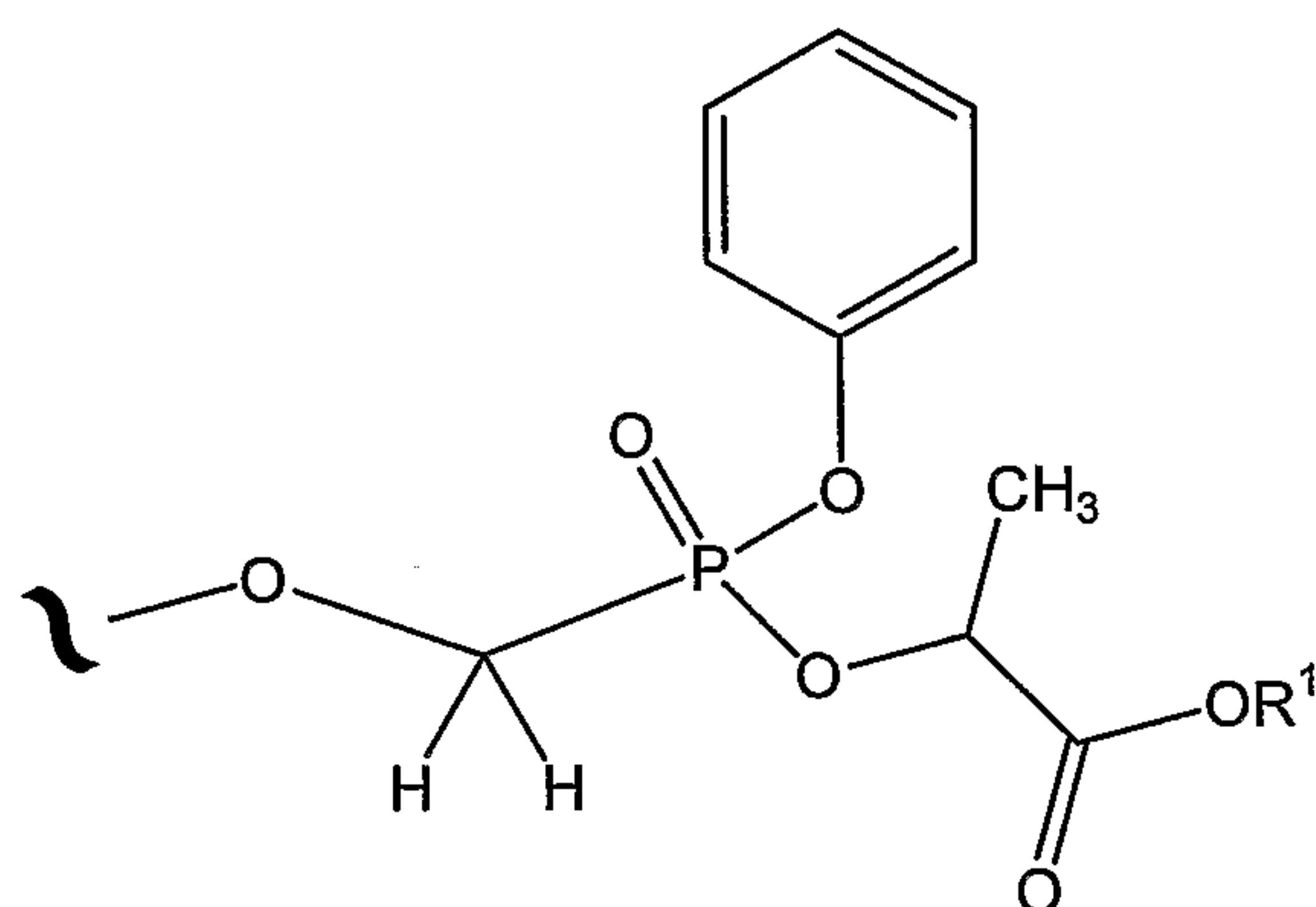
wherein R^1 is independently H or alkyl of 1 to 18 carbon atoms.

In some embodiments of the invention, A³ is of the formula:



and R¹ is H or alkyl of 1 to 18 carbon atoms.

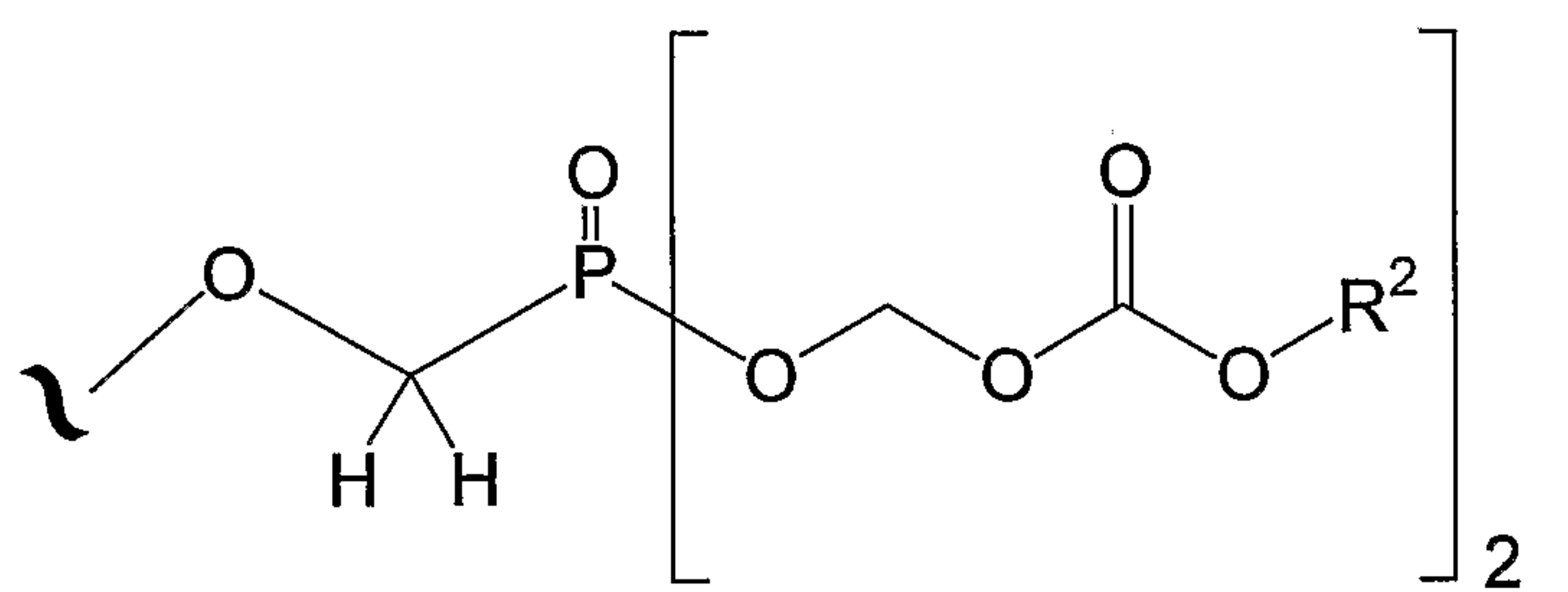
In some embodiments of the invention, A³ is of the formula:



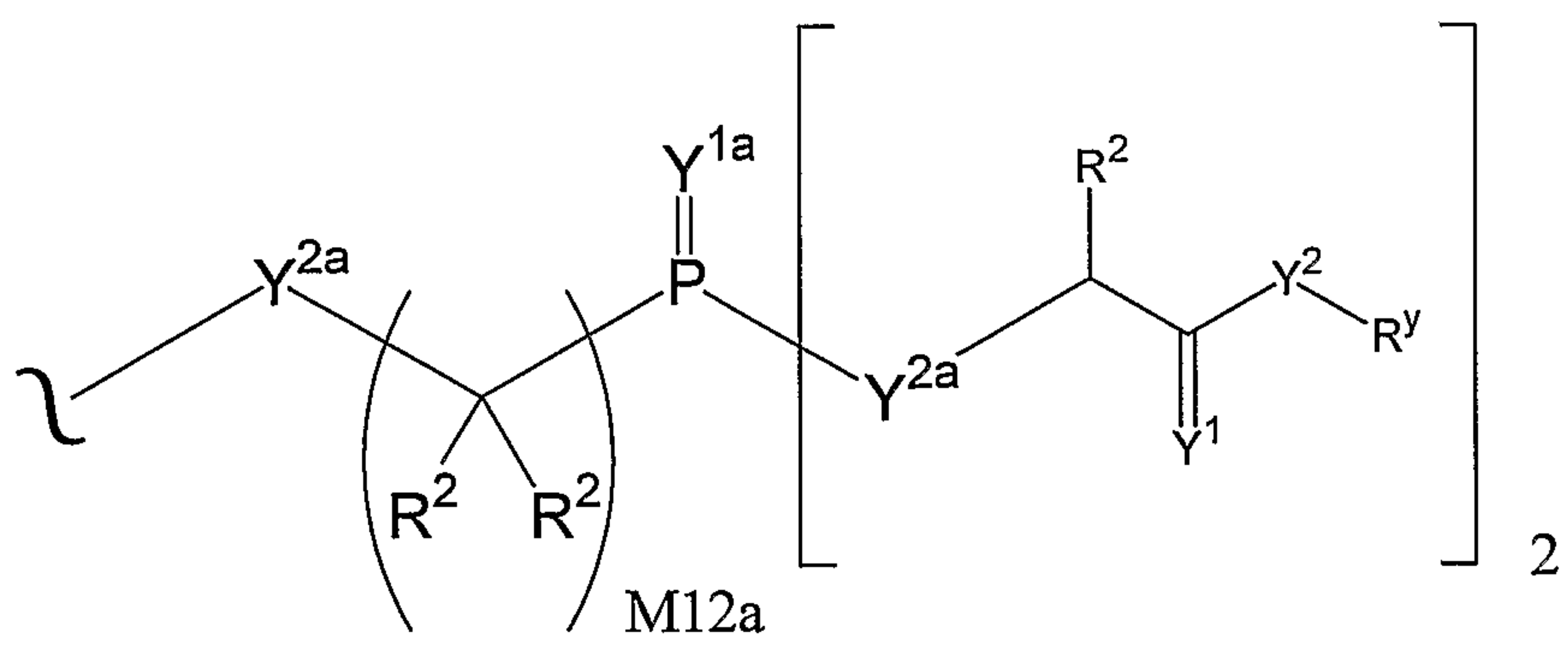
5

and R¹ is H or alkyl of 1 to 18 carbon atoms.

In some embodiments of the invention, A³ is of the formula:



In some embodiments of the invention, A³ is of the formula:

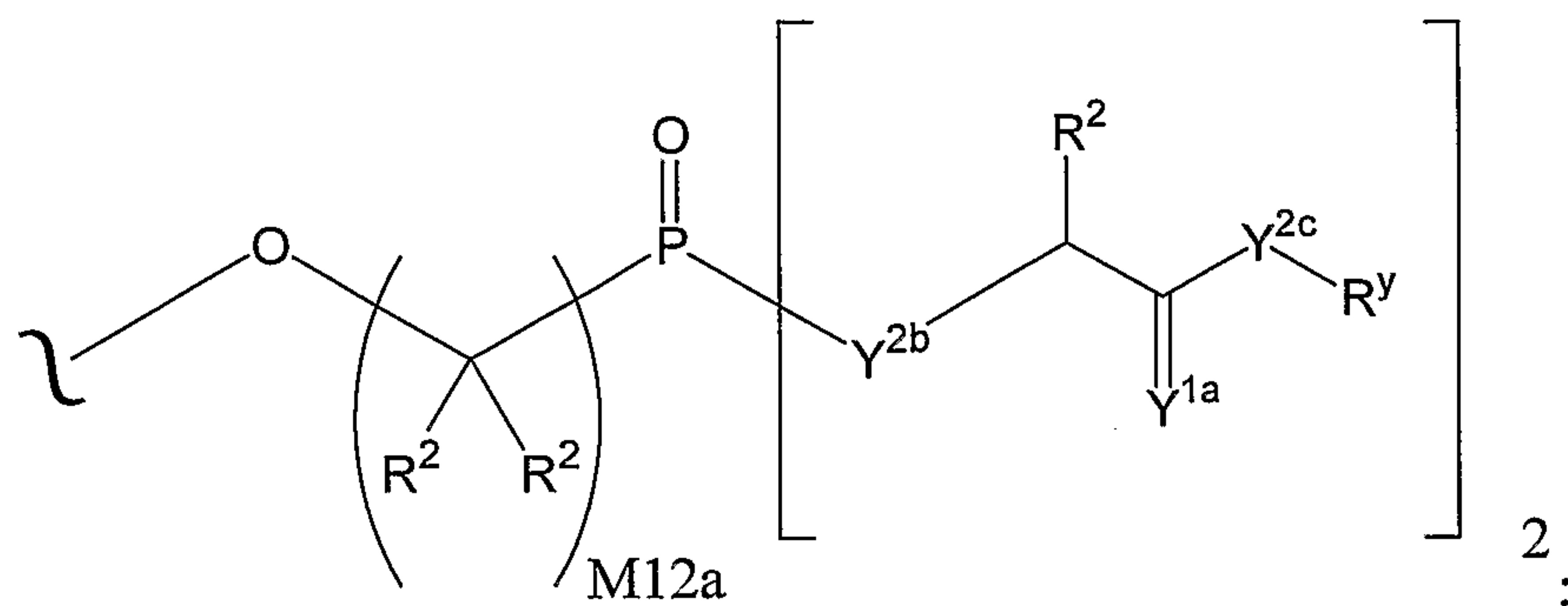


10

Y^{1a} is O or S; and

Y^{2a} is independently O, N(R²) or S.

In some embodiments of the invention, A³ is of the formula:



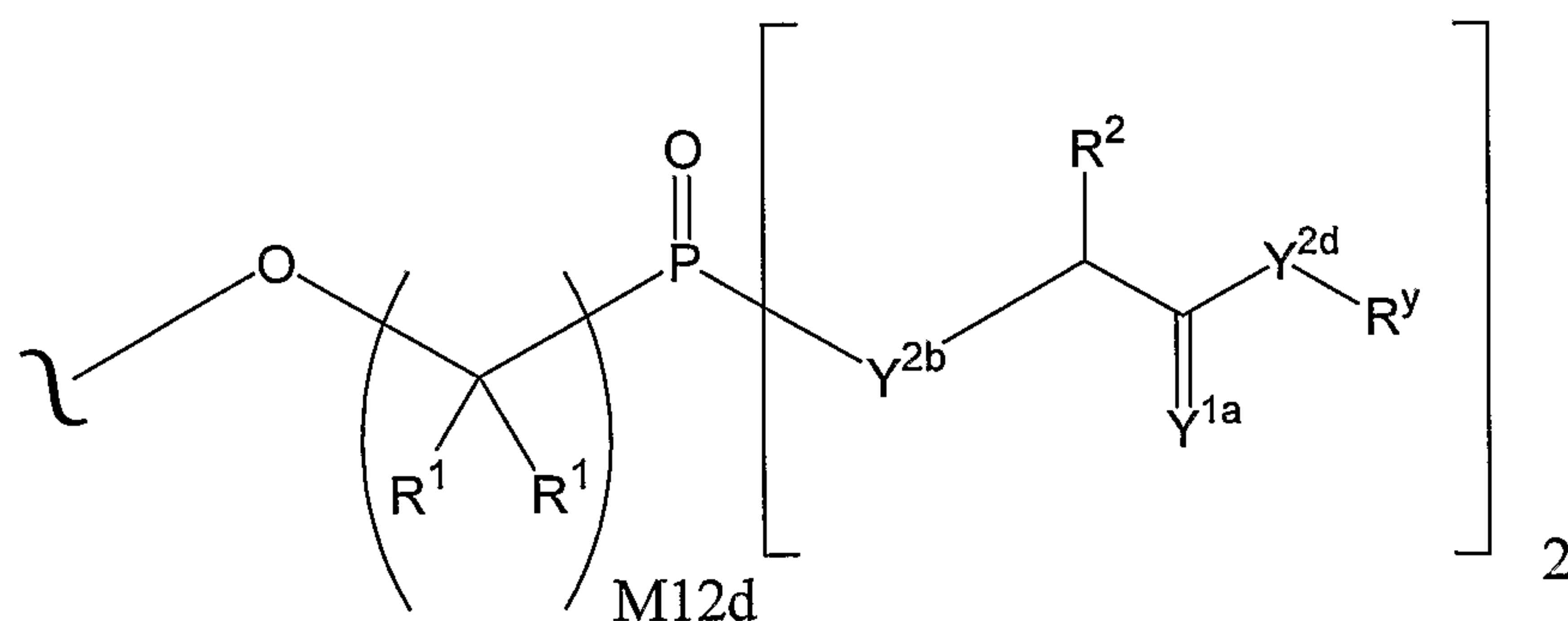
5

Y^{1a} is O or S;

Y^{2b} is O or N(R²); and

Y^{2c} is O, N(R^y) or S.

In some embodiments of the invention, A³ is of the formula:



10

R¹ is independently H or alkyl of 1 to 18 carbon atoms;

Y^{1a} is O or S;

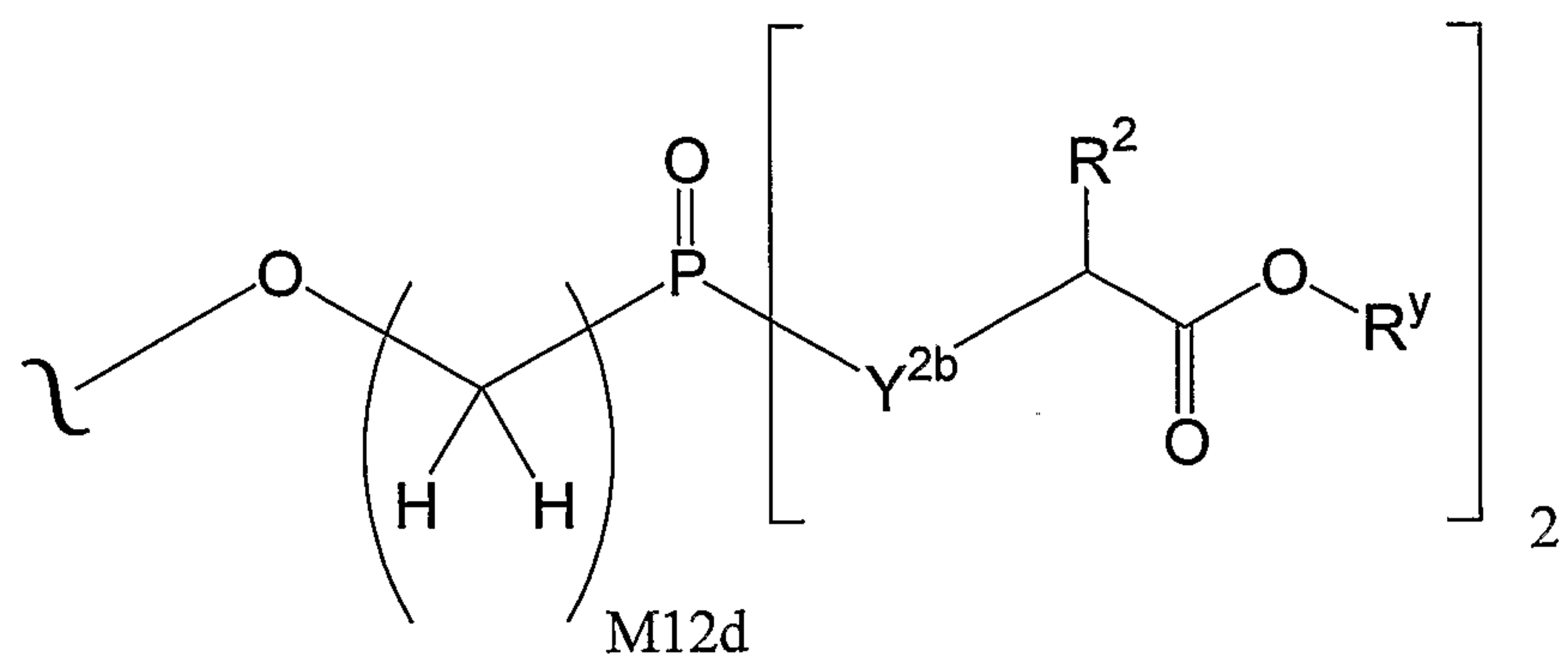
Y^{2b} is O or N(R²);

Y^{2d} is O or N(R^y); and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

15

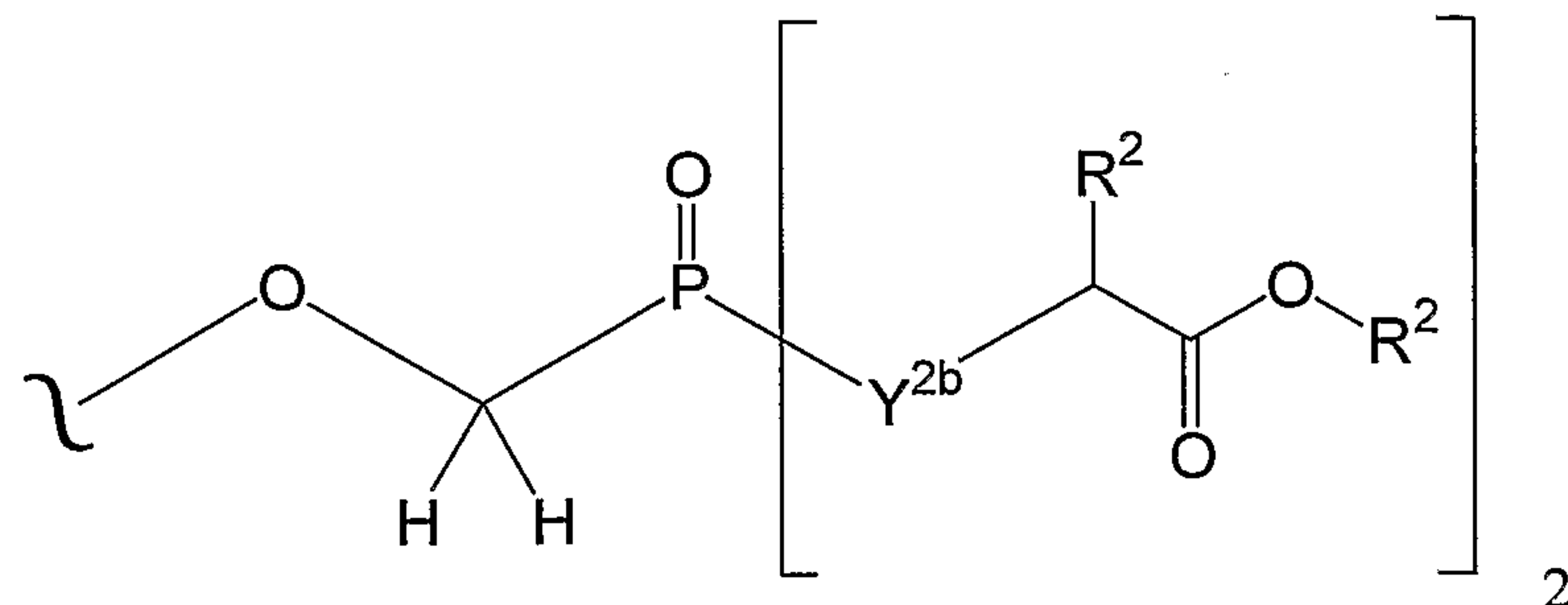
In some embodiments of the invention, A³ is of the formula:



Y^{2b} is O or $N(R^2)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

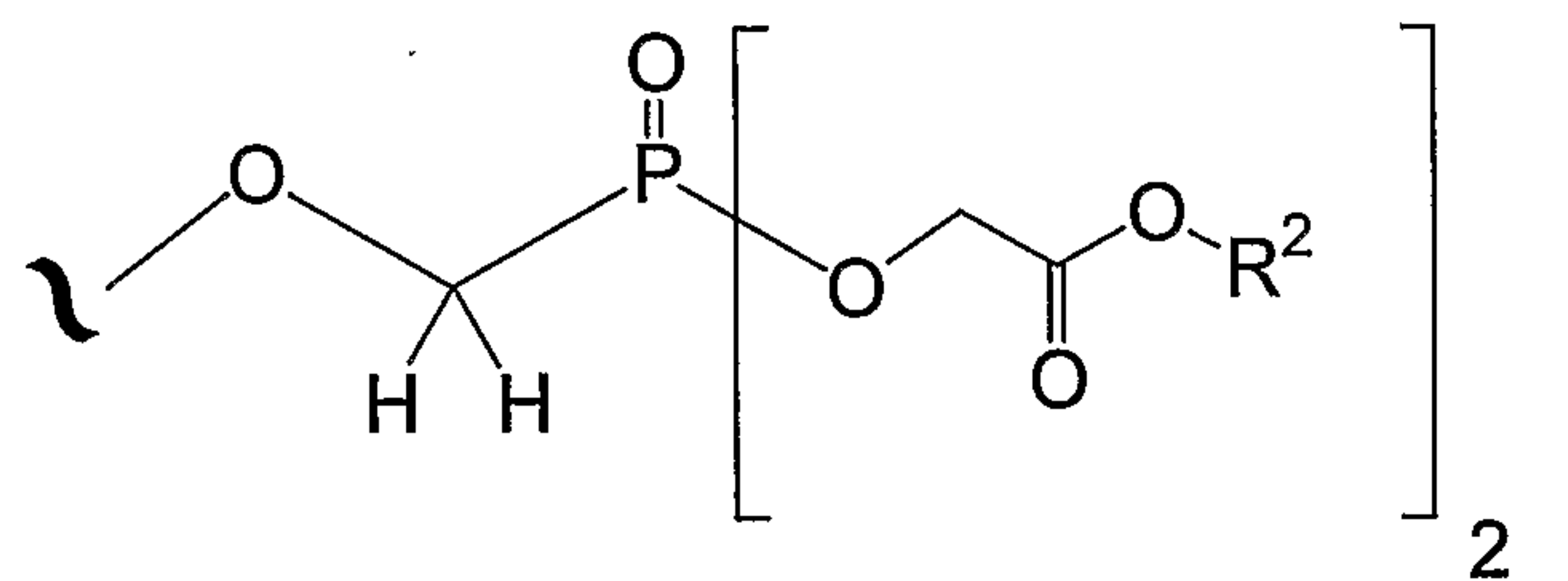
In some embodiments of the invention, A^3 is of the formula:



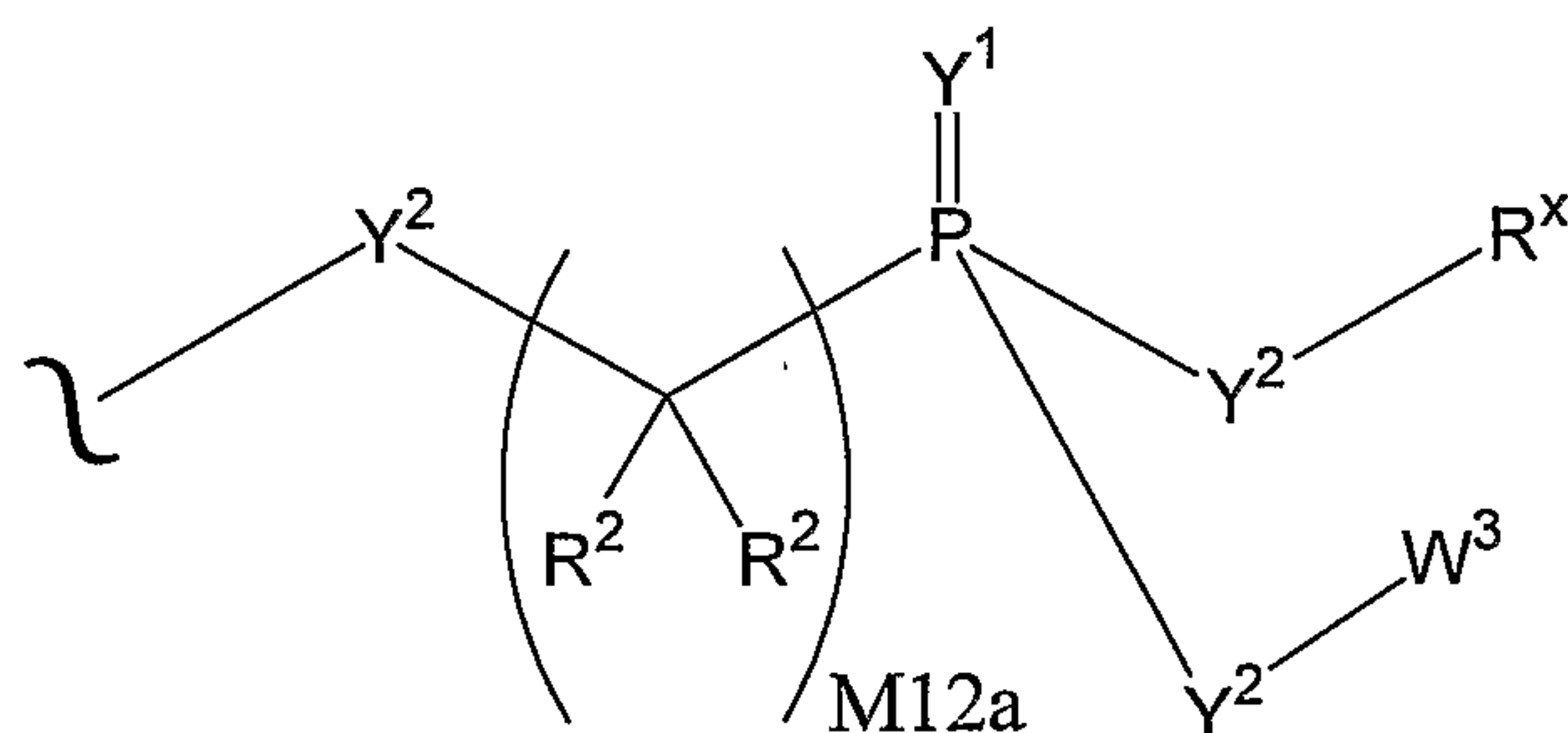
5

and Y^{2b} is O or $N(R^2)$.

In some embodiments of the invention, A^3 is of the formula:

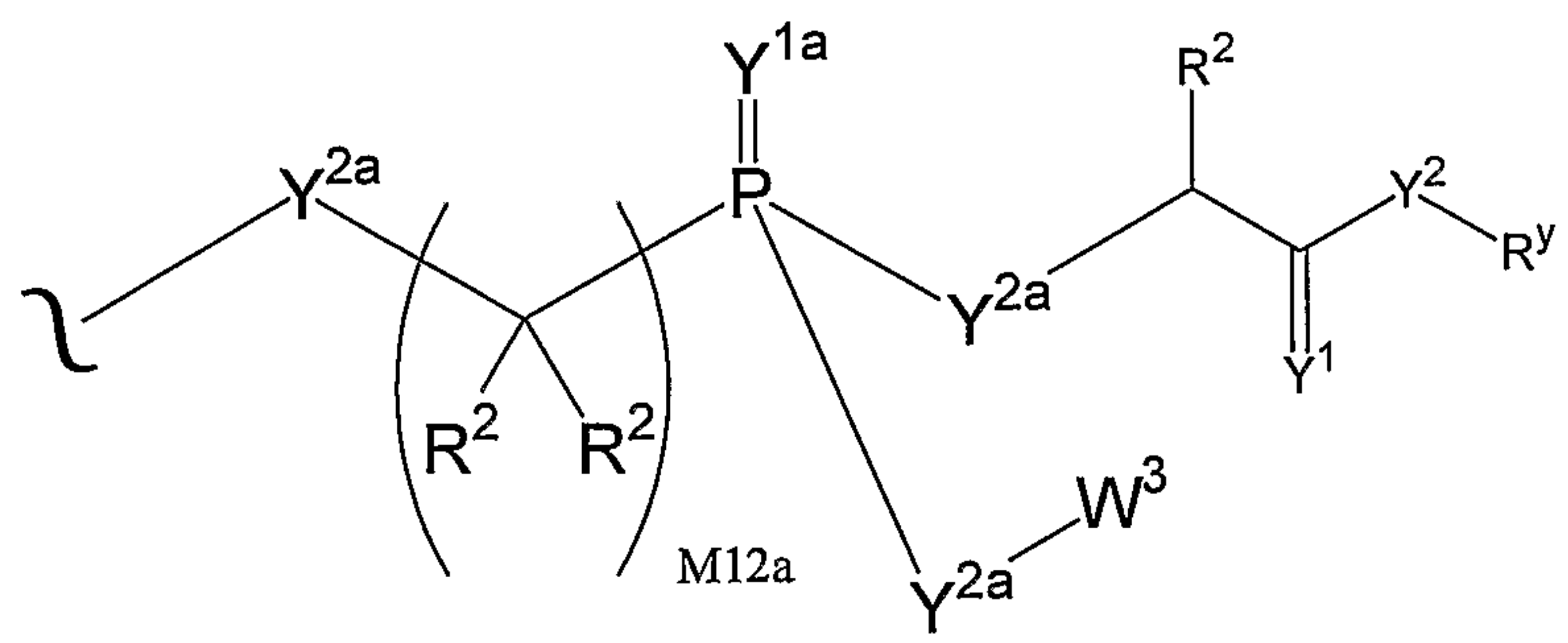


In some embodiments of the invention, A^3 is of the formula:



10

In some embodiments of the invention, A^3 is of the formula:

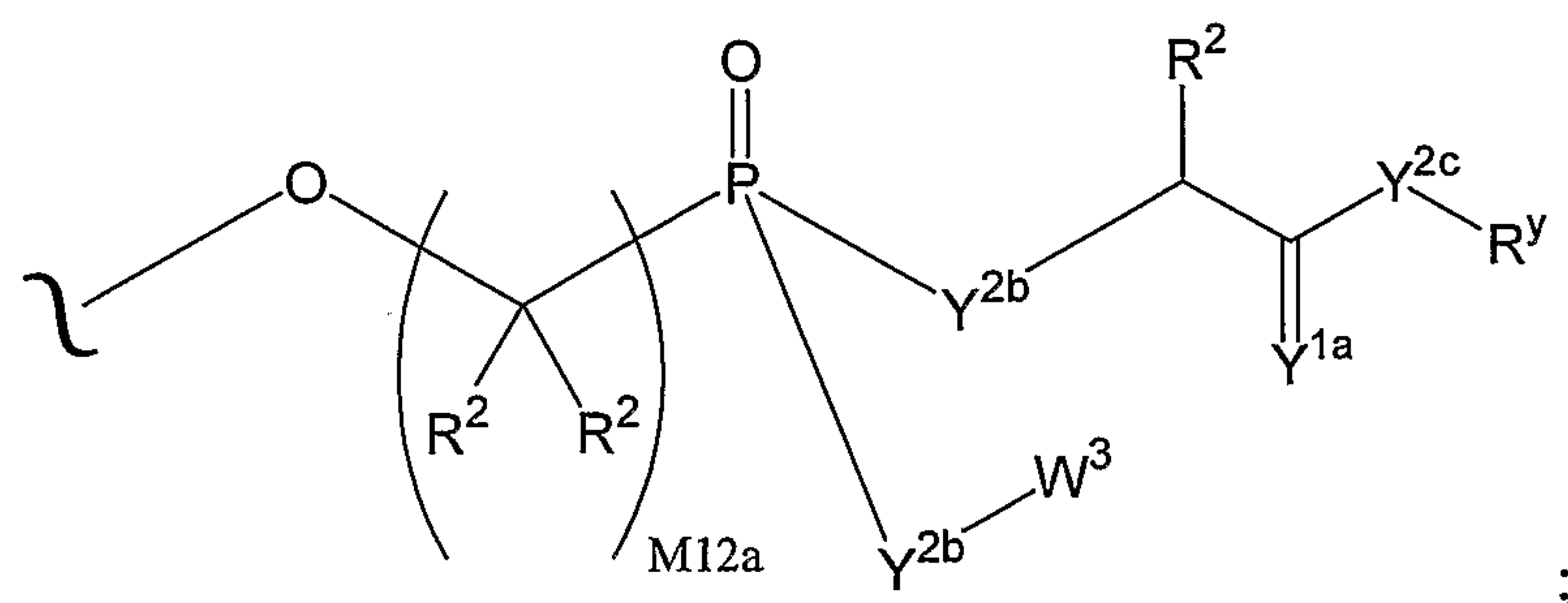


Y^{1a} is O or S; and

Y^{2a} is independently O, $N(R^2)$ or S.

15

In some embodiments of the invention, A^3 is of the formula:

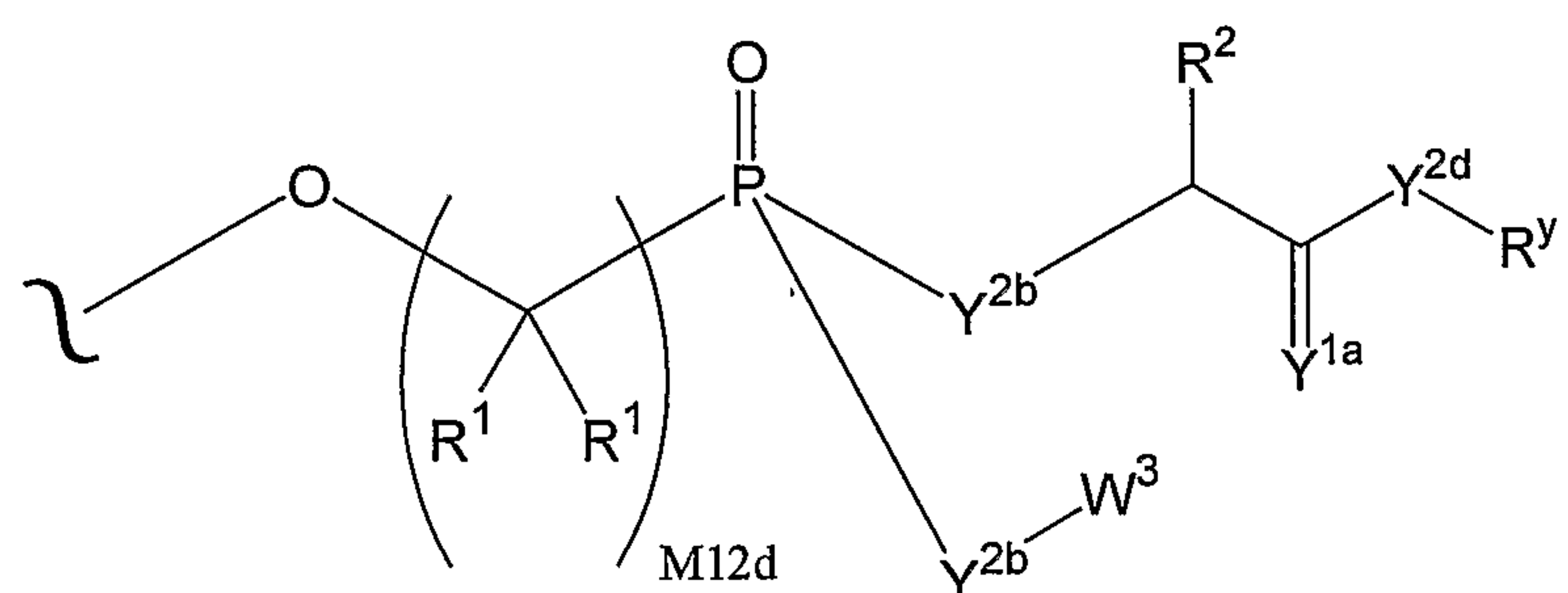


Y^{1a} is O or S;

Y^{2b} is independently O or $N(R^2)$; and

5 Y^{2c} is O, $N(R^y)$ or S.

In some embodiments of the invention, A^3 is of the formula:



R^1 is independently H or alkyl of 1 to 18 carbon atoms;

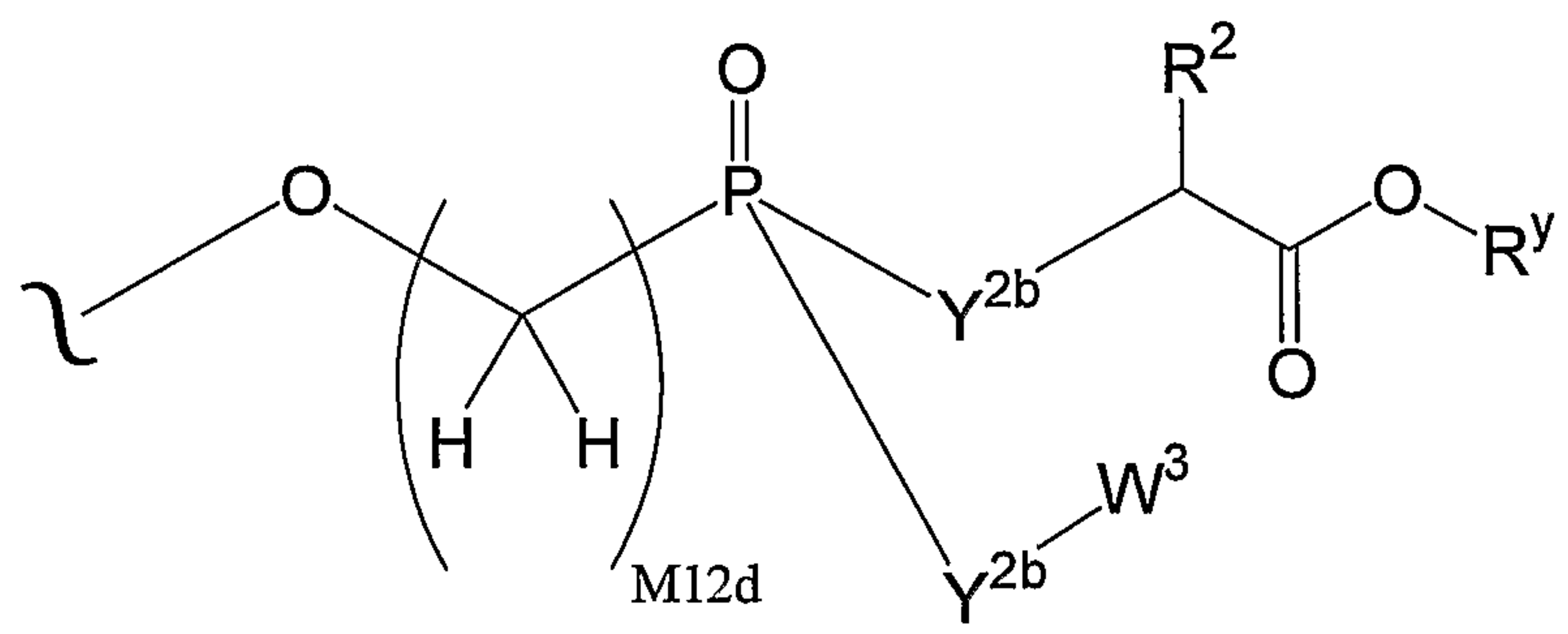
10 Y^{1a} is O or S;

Y^{2b} is independently O or $N(R^2)$;

Y^{2d} is O or $N(R^y)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

In some embodiments of the invention, A^3 is of the formula:

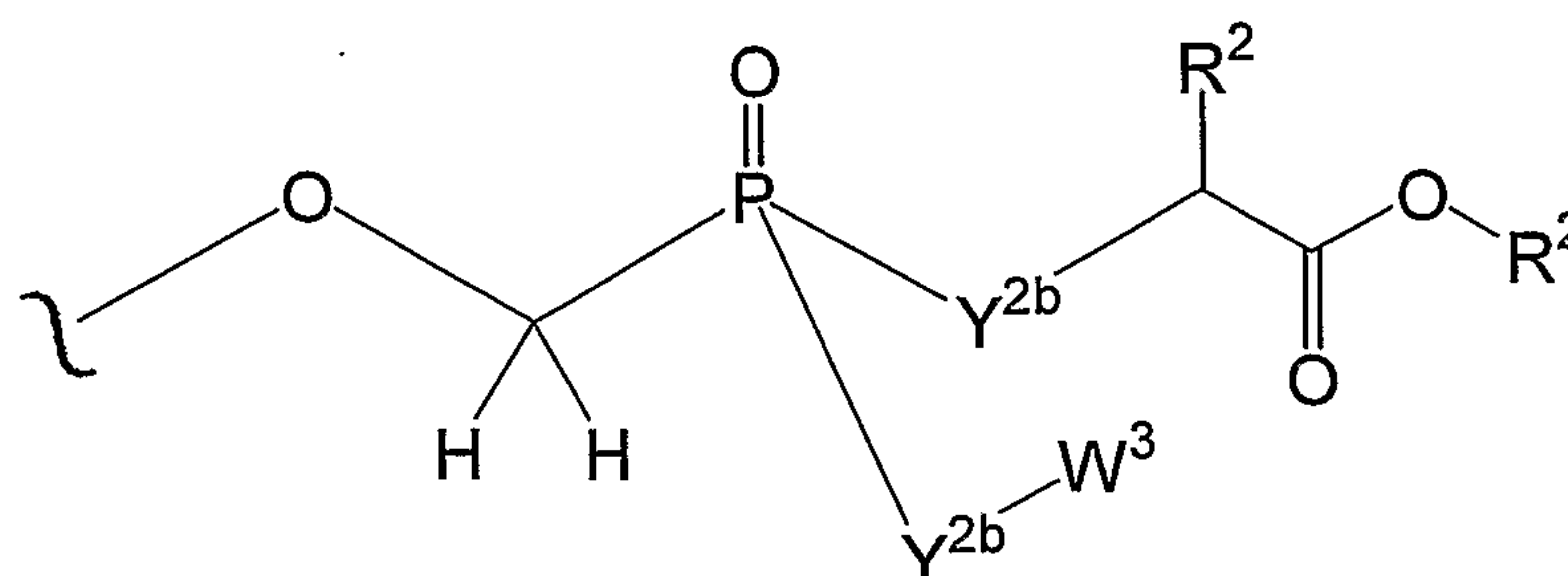


15

Y^{2b} is independently O or $N(R^2)$; and

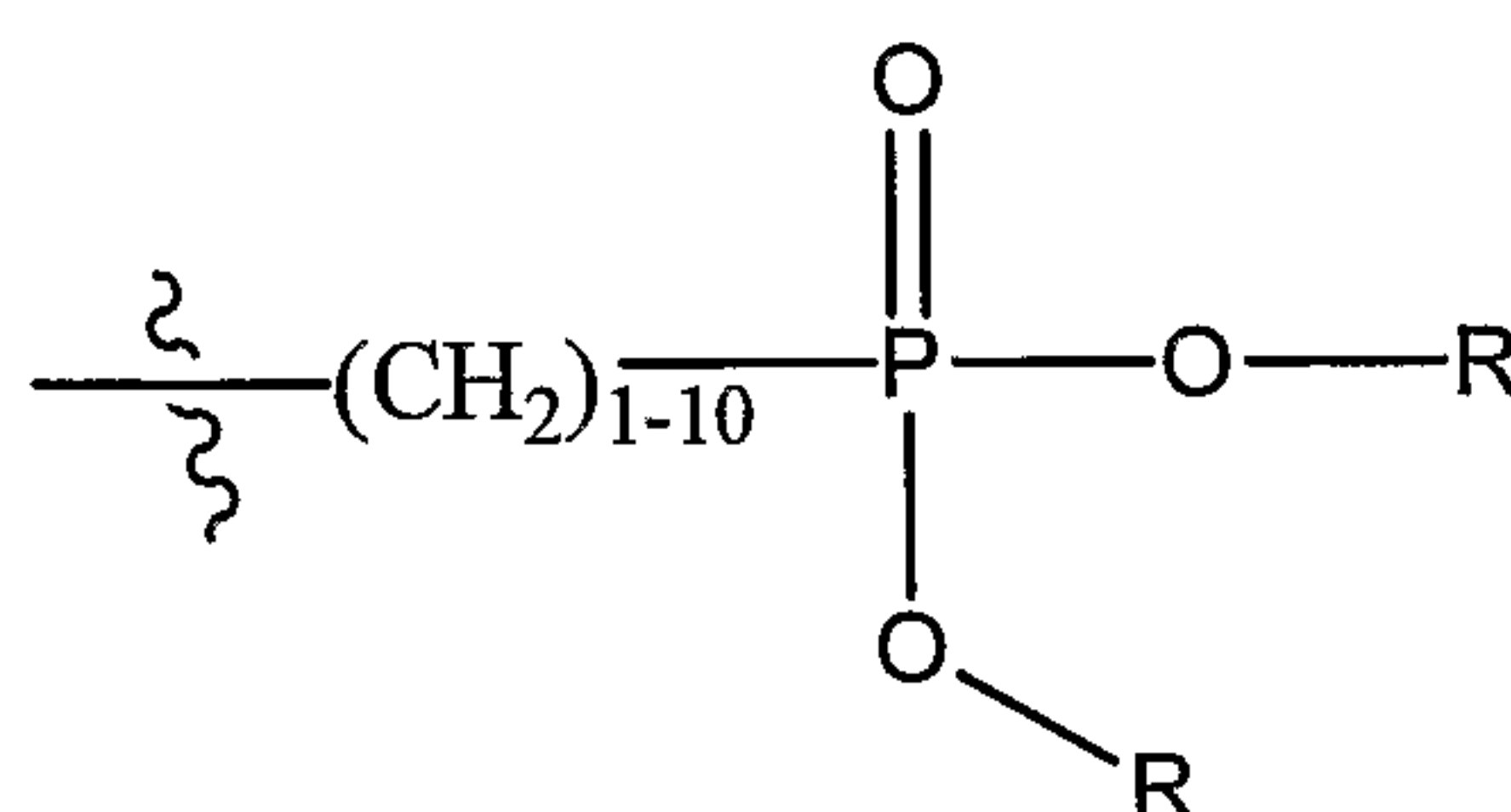
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

In some embodiments of the invention, A^3 is of the formula:



and Y^{2b} is independently O or $N(R^2)$.

In some embodiments of the invention, A^0 is of the formula:



5 wherein each R is independently (C_1-C_6) alkyl.

In some embodiments of the invention, R^{20} is methoxy.

In some embodiments of the invention, R^{20} is vinyl or ethyl.

In some embodiments of the invention, each alkyl has 1 to 6 carbon atoms and aryl is phenyl.

10 In some embodiments of the invention, Z is formula ZA; G is $-O(CH_2)_nN=G^3$; n is an integer from 2 to 4; and $=G^3$ is alkylene of four to six carbon atoms, or alkylene of three to five carbon atoms plus one member that is -O-, -S-, or $-N(G^4)-$; where G^4 is hydrogen or alkyl.

In some embodiments of the invention, $=G^3$ is alkylene of four to six
15 carbon atoms, or alkylene of three to five carbon atoms plus one member that is -O- or -S-; where G^4 is hydrogen or alkyl.

In some embodiments of the invention, G is not OH.

In some embodiments of the invention, G is not morpholinoethoxy.

In some embodiments of the invention, Z^3 is not $-P(O)(OCH_3)_2$, or -
20 $P(O)(OH)(OCH_3)$ and Z^4 is not $-P(O)(OCH_3)_2$, or $-P(O)(OH)(OCH_3)$.

In some embodiments of the invention, Z^3 is not $-P(O)(OH)_2$ and Z^4 is not $-P(O)(OH)_2$.

In some embodiments of the invention, R^{21} is (C_1-C_6) alkyl.

In some embodiments of the invention, R^{22} is (C_1-C_6) alkyl.

25 In some embodiments of the invention, R^{22} is substituted phenyl.

In some embodiments of the invention, phenyl is substituted with CO_2R^{46} , wherein R^{46} is H or alkyl.

In some embodiments of the invention, phenyl is substituted at the *para* position.

5 In some embodiments of the invention, R^{46} is H.

In some embodiments of the invention, R^{46} is alkyl.

In some embodiments of the invention, R^{46} is $(\text{C}_1\text{-C}_6)\text{alkyl}$.

In some embodiments of the invention, R^{23} is methoxy.

In some embodiments of the invention, R^{23} is ethyl or vinyl.

10 In some embodiments of the invention, A^0 comprises a phosphonate other than $\text{P}(\text{O})(\text{OH})(\text{OMe})$.

In some embodiments of the invention, A^0 comprises a phosphonate other than $\text{P}(\text{O})(\text{OMe})_2$.

15 In some embodiments of the invention, A^0 comprises a phosphonate other than $\text{P}(\text{O})(\text{OH})_2$.

In some embodiments of the invention, R^{30} is methoxy.

In some embodiments of the invention, R^{30} is ethyl.

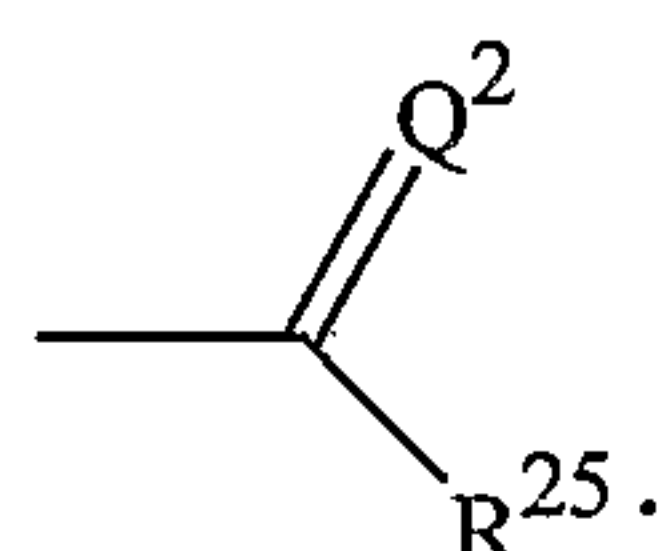
In some embodiments of the invention, R^{30} is vinyl.

20 In some embodiments of the invention, A^0 comprises a phosphonate other than $-\text{P}(=\text{O})(\text{OH})(\text{OMe})$ or $-\text{P}(=\text{O})(\text{OMe})(\text{OMe})$.

In some embodiments of the invention, A^0 comprises a phosphonate other than $-\text{P}(=\text{O})(\text{OH})(\text{OH})$.

25 In some embodiments of the invention, R^{25} is alkyl, haloalkyl or $-\text{NR}^{28}\text{R}^{29}$, where R^{28} and R^{29} are independently H, alkyl, haloalkyl, cycloalkyl, phenyl optionally monosubstituted with halogen, hydroxy, carboxy, chlorocarbonyl, nitro, cyano, phenyl, alkyl, acyl, alkoxy carbonyl, acylamino, dialkylamino or dialkylaminoethoxycarbonyl, phenyl optionally disubstituted with hydroxy, carboxy, nitro or alkyl, or benzyl optionally substituted with dialkylamino.

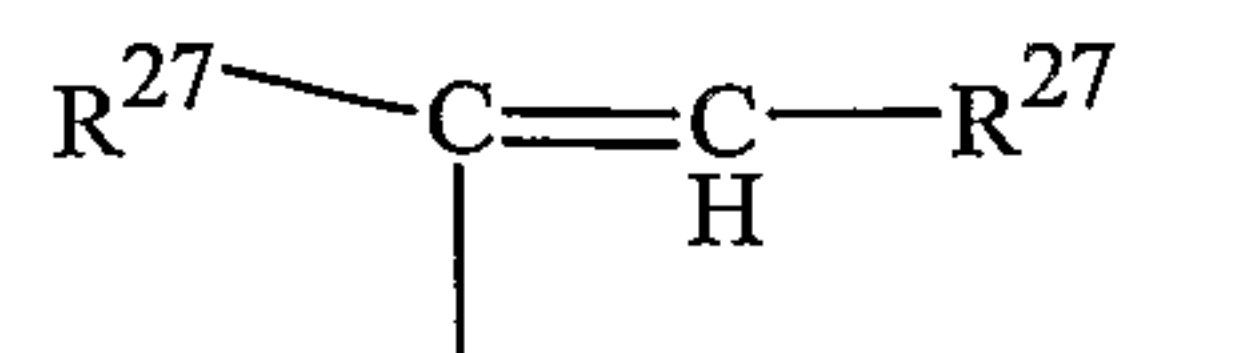
30 In some embodiments of the invention, R^{24} is



In some embodiments of the invention, R^{24} is



In some embodiments of the invention, R^{24} is



5 In some embodiments of the invention, Q^2 is oxygen.

In some embodiments of the invention, Q^2 is sulfur.

In some embodiments of the invention, na is 1, 2, 3, or 4.

In some embodiments of the invention, R^{31} is methoxy.

In some embodiments of the invention, R^{31} is ethyl or vinyl.

10 In some embodiments of the invention, R^{32} and R^{33} are each independently H.

In some embodiments of the invention, R^{32} and R^{33} are each independently alkyl having 1 to 6 carbon atoms.

15 In some embodiments of the invention, R^{32} and R^{33} are each methyl or ethyl.

In some embodiments of the invention, Q^3 is O.

In some embodiments of the invention, Q^3 is S.

In some embodiments of the invention, nb is 2, 3, or 4.

In some embodiments of the invention, nb is 1.

20 In some embodiments of the invention, R^{34} is methoxy.

In some embodiments of the invention, R^{34} is ethyl.

In some embodiments of the invention, R^{34} is vinyl.

In some embodiments of the invention, R^{36} is hydrogen, alkyl, $-C(O)R^{39}$, $-C(O)NR^{40}R^{41}$, $-CO_2R^{42}$, or $-SO_2R^{39}$.

25 In some embodiments of the invention, R^{36} is hydrogen, alkyl having 1-6 carbons, $-C(O)R^{39}$, $-C(O)NR^{40}R^{41}$, $-CO_2R^{42}$, or $-SO_2R^{39}$.

In some embodiments of the invention, R^{39} is hydrogen, alkyl having 1-6 carbons, halo substituted alkyl having 1-6 carbons, phenyl, or substituted phenyl.

30 In some embodiments of the invention, R^{40} is hydrogen, alkyl having 1-6 carbons, phenyl, or substituted phenyl.

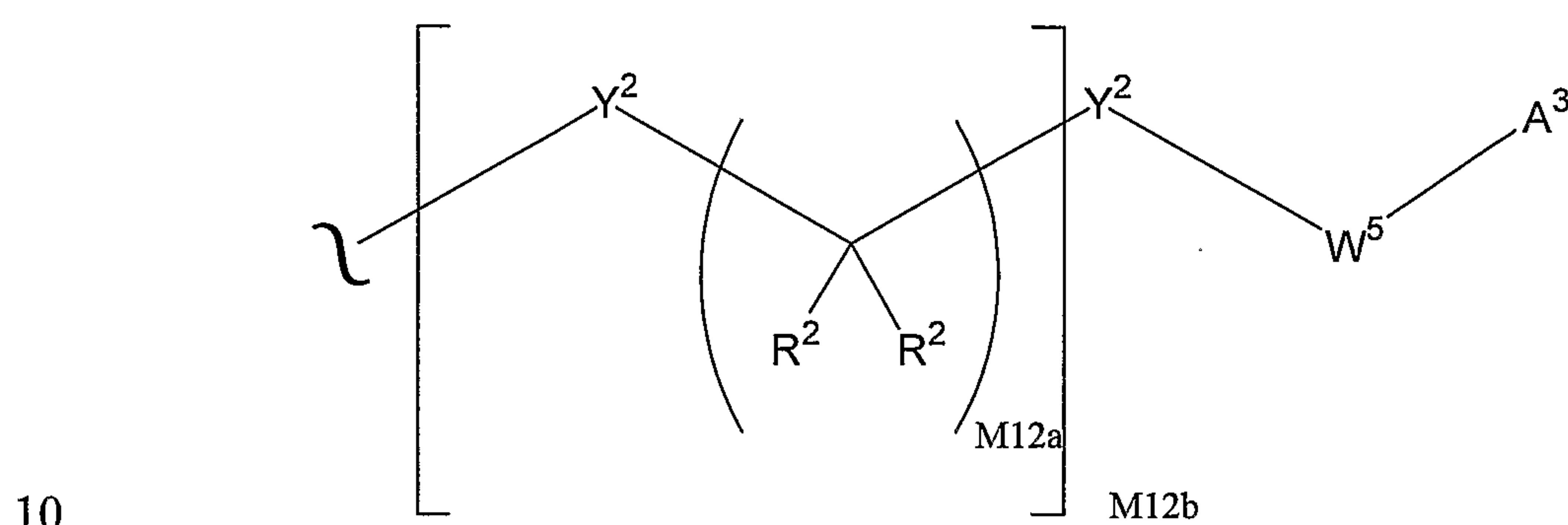
In some embodiments of the invention, R^{41} is hydrogen, alkyl having 1-6 carbons, phenyl, or substituted phenyl.

In some embodiments of the invention, R^{42} is alkyl having 1-6 carbons, phenyl, or substituted phenyl.

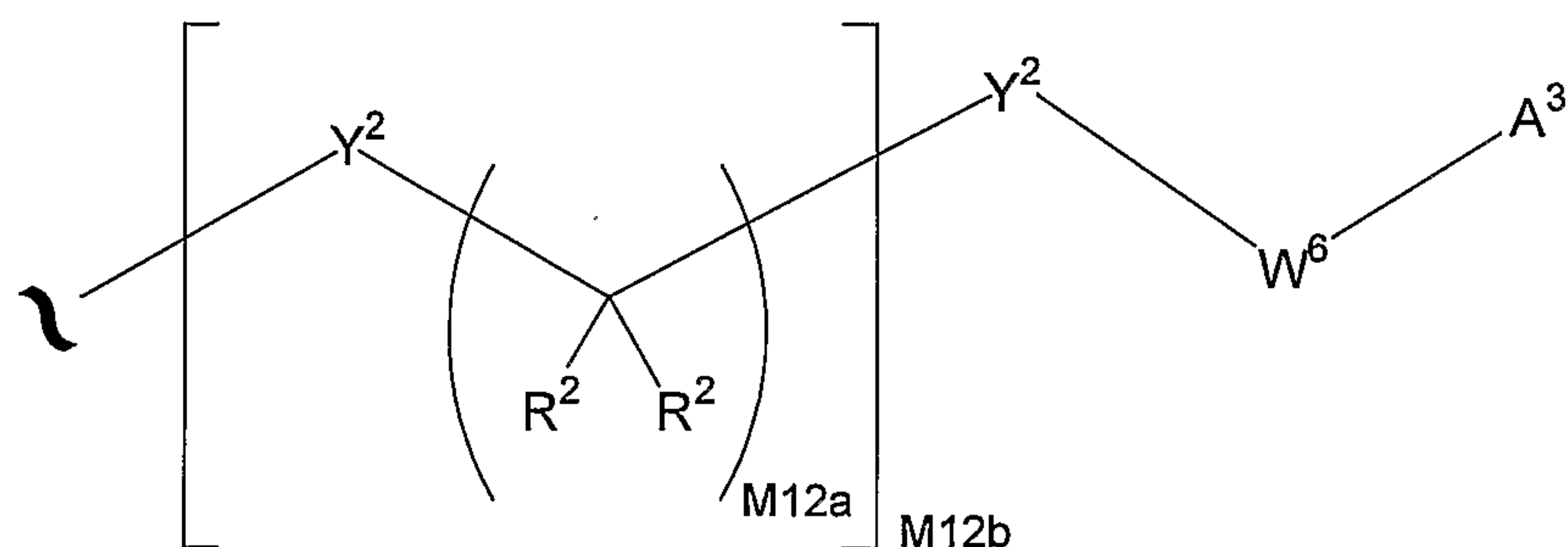
5 In some embodiments of the invention, A^0 comprises a phosphonate other than $-P(=O)(OH)(OMe)$ or $-P(=O)(OMe)(OMe)$.

In some embodiments of the invention, A^0 comprises a phosphonate other than $-P(=O)(OH)(OH)$.

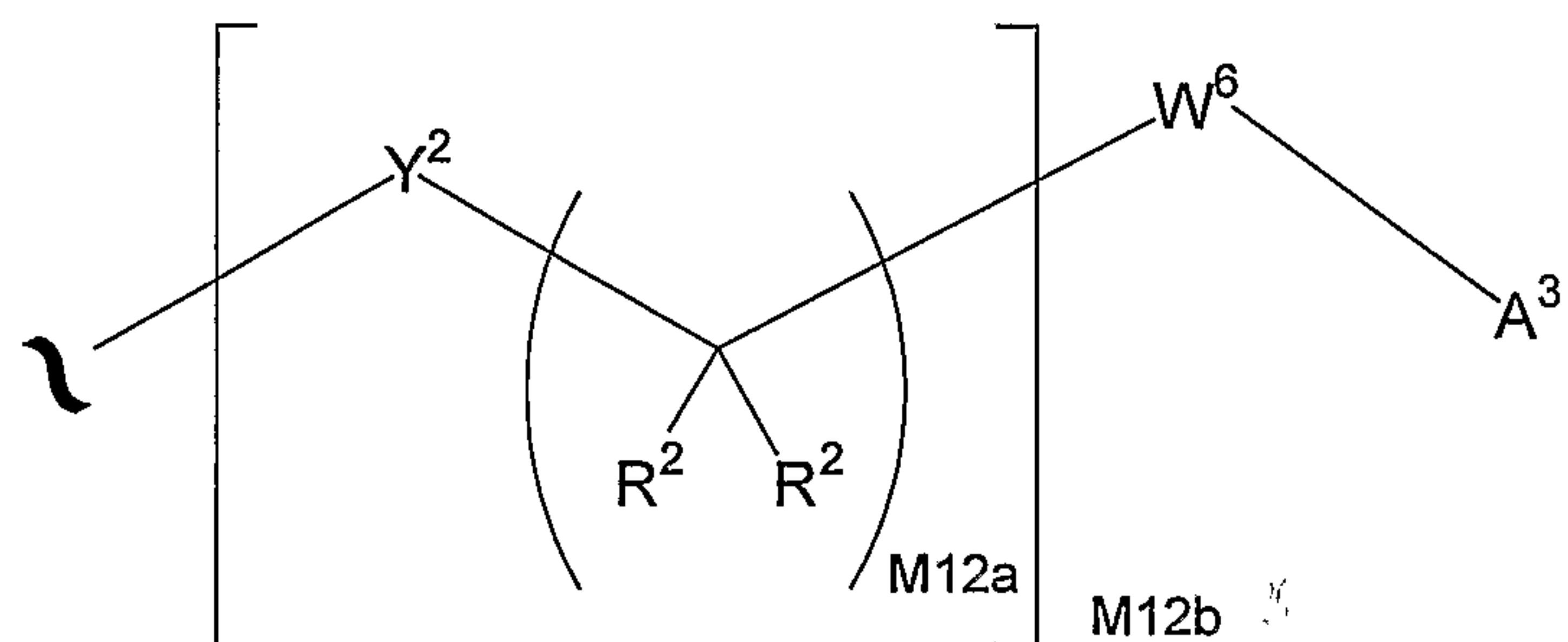
In some embodiments of the invention, A^1 is of the formula:



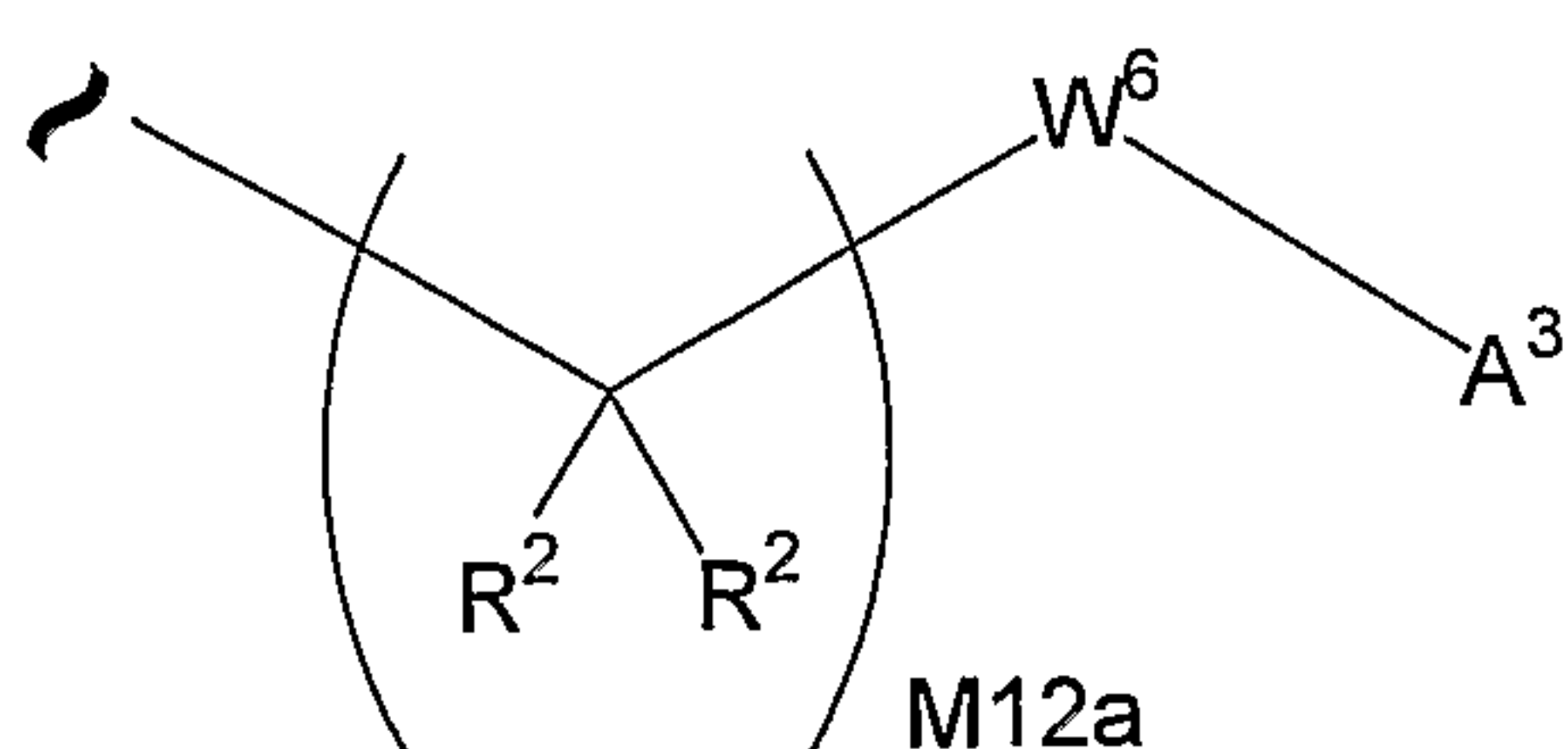
In some embodiments of the invention, A^1 is of the formula:



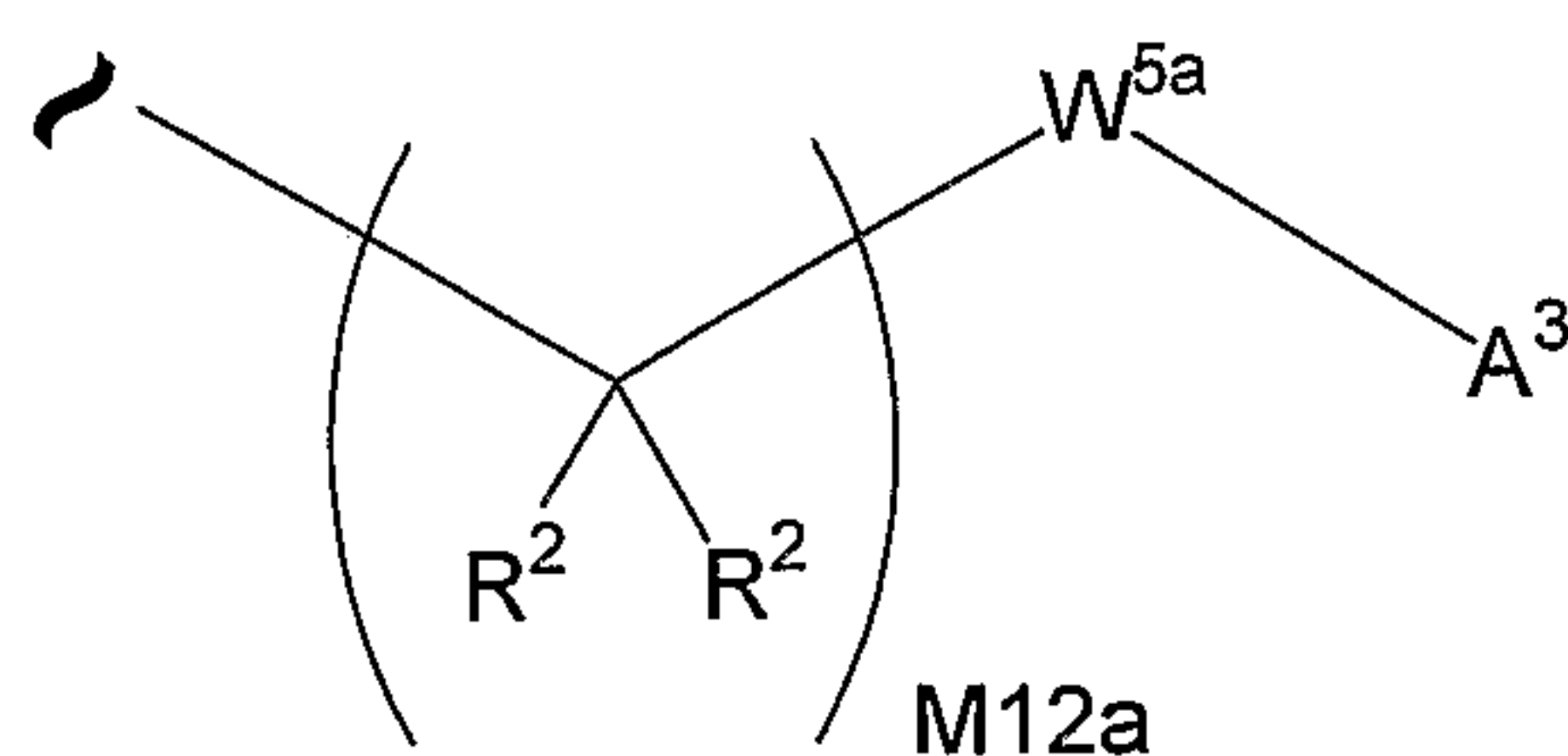
In some embodiments of the invention, A^1 is of the formula:



15 In some embodiments of the invention, A^1 is of the formula:



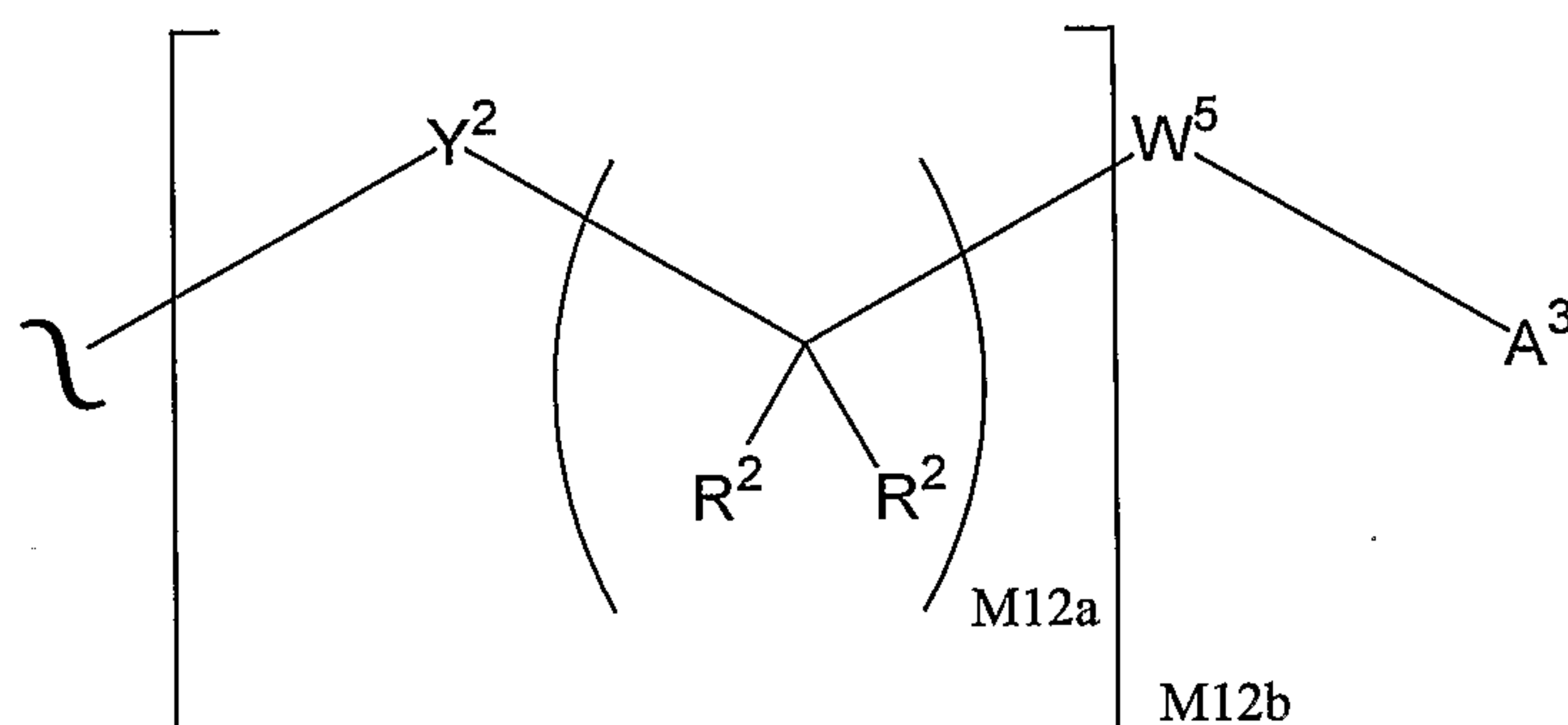
In some embodiments of the invention, A¹ is of the formula:



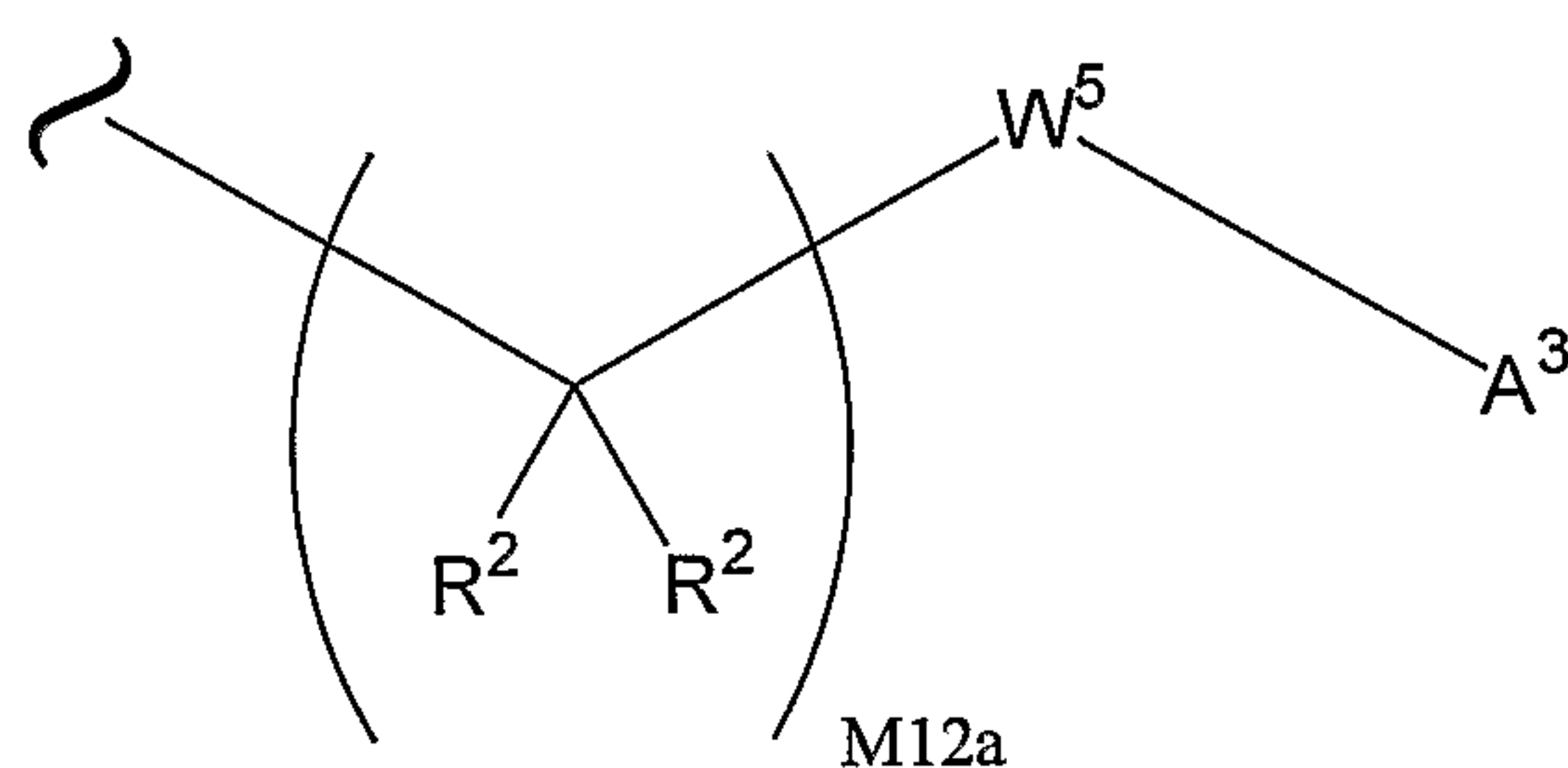
- 5 and W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently substituted with 0 or 1 R² groups.

In some embodiments of the invention, M12a is 1.

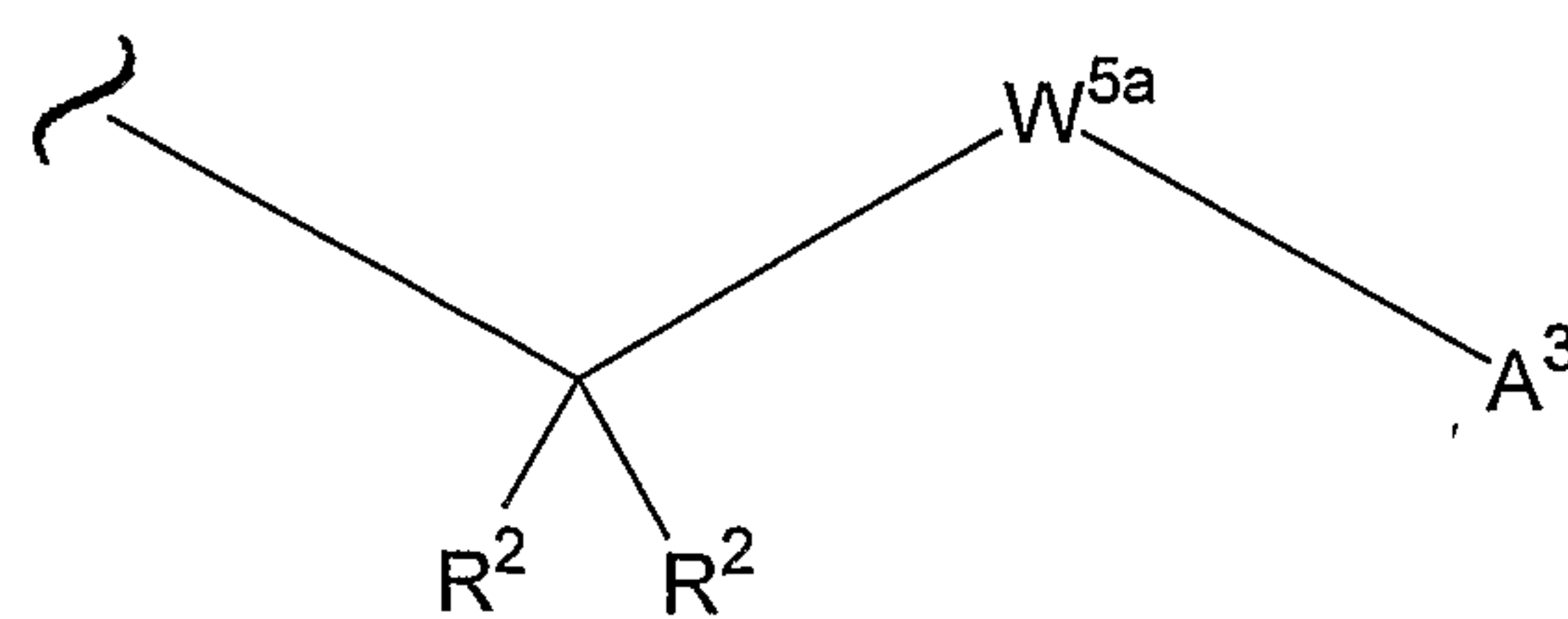
In some embodiments of the invention, A¹ is of the formula:



- 10 In some embodiments of the invention, A¹ is of the formula:

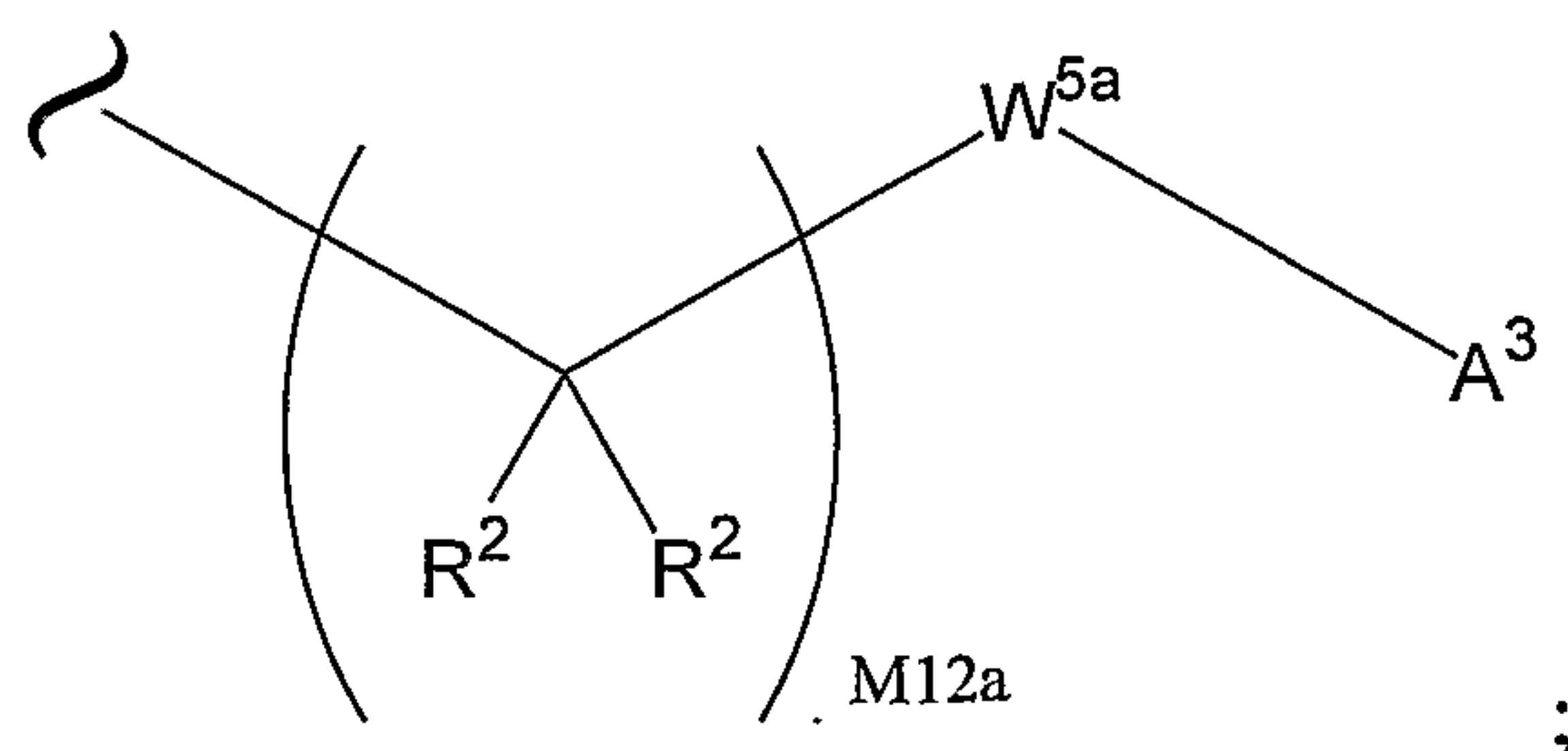


In some embodiments of the invention, A¹ is of the formula:



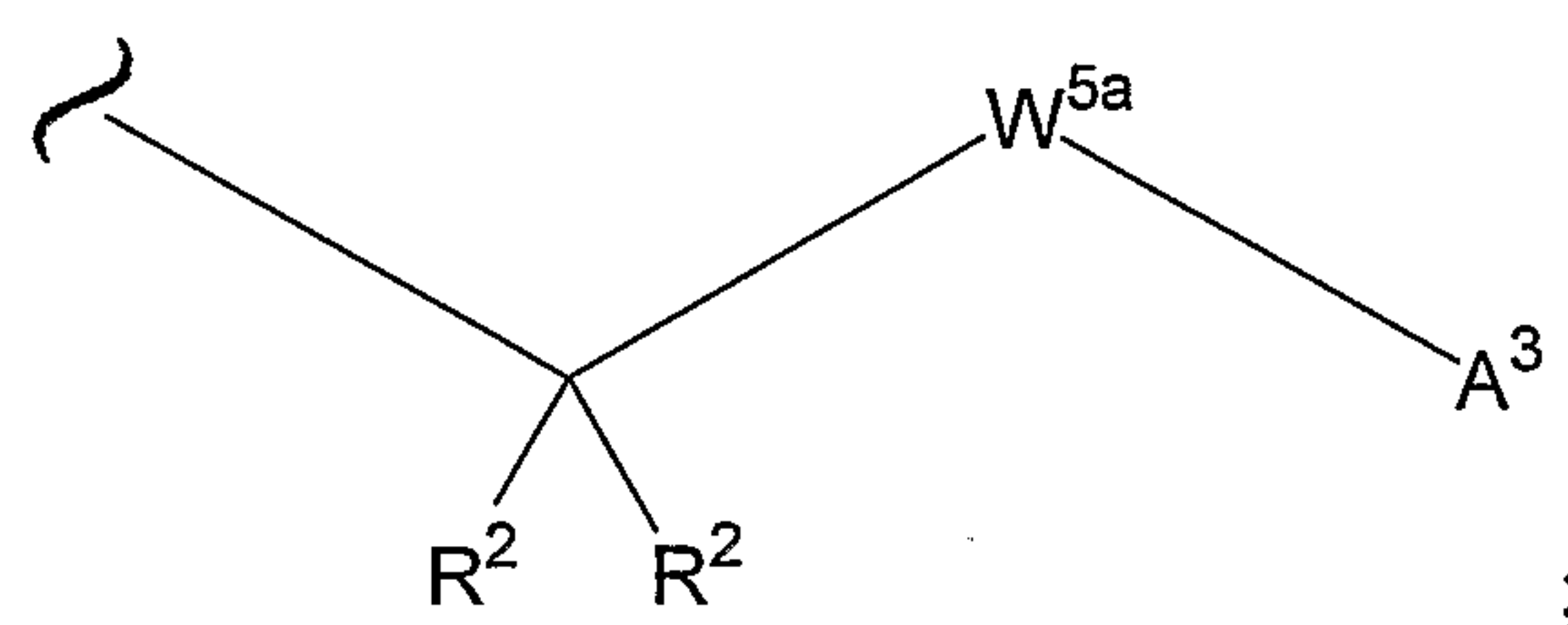
W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^1 is of the formula:



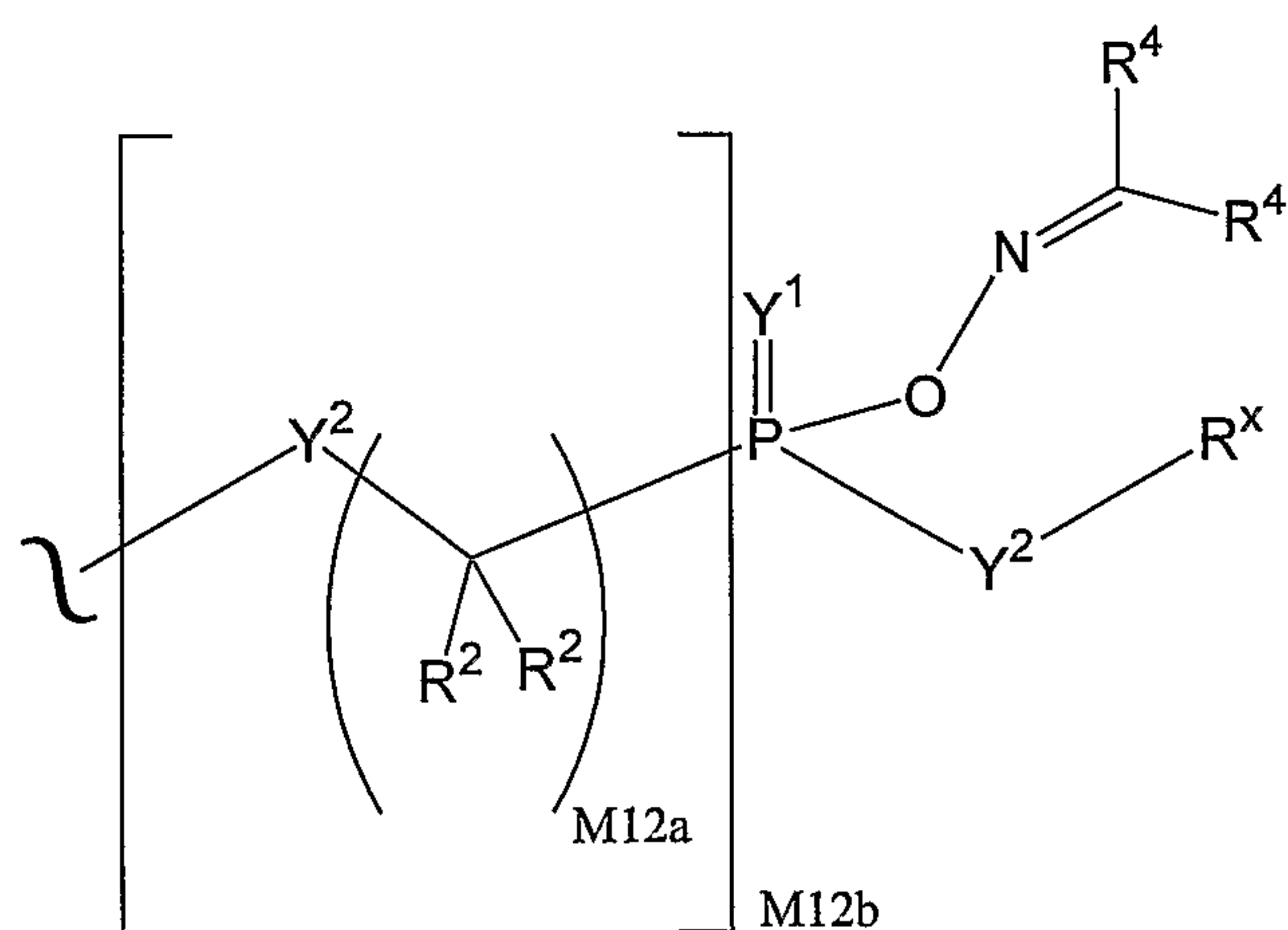
5 W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^1 is of the formula:

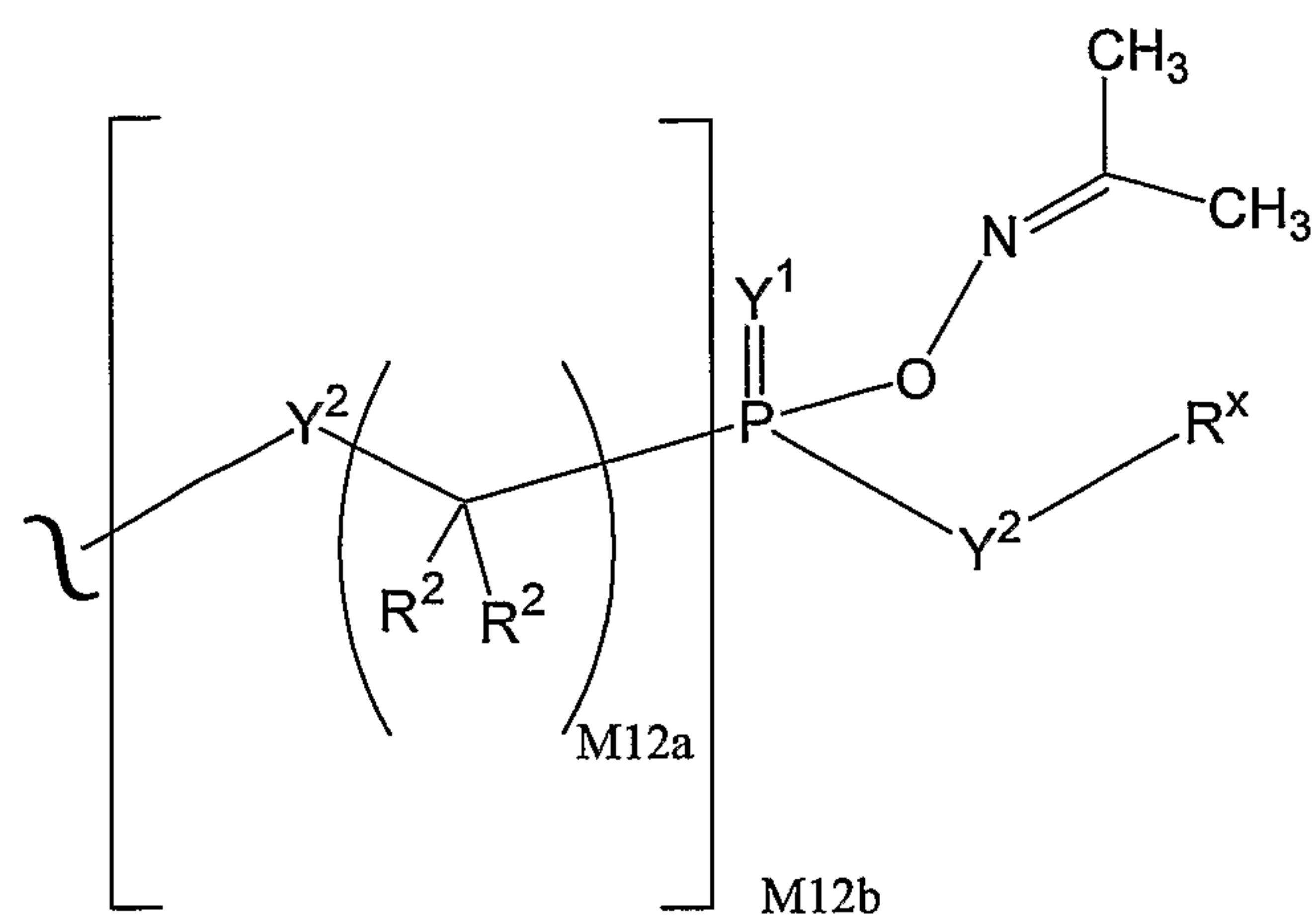


W^{5a} is a carbocycle or heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

10 In some embodiments of the invention, A^3 is of the formula:

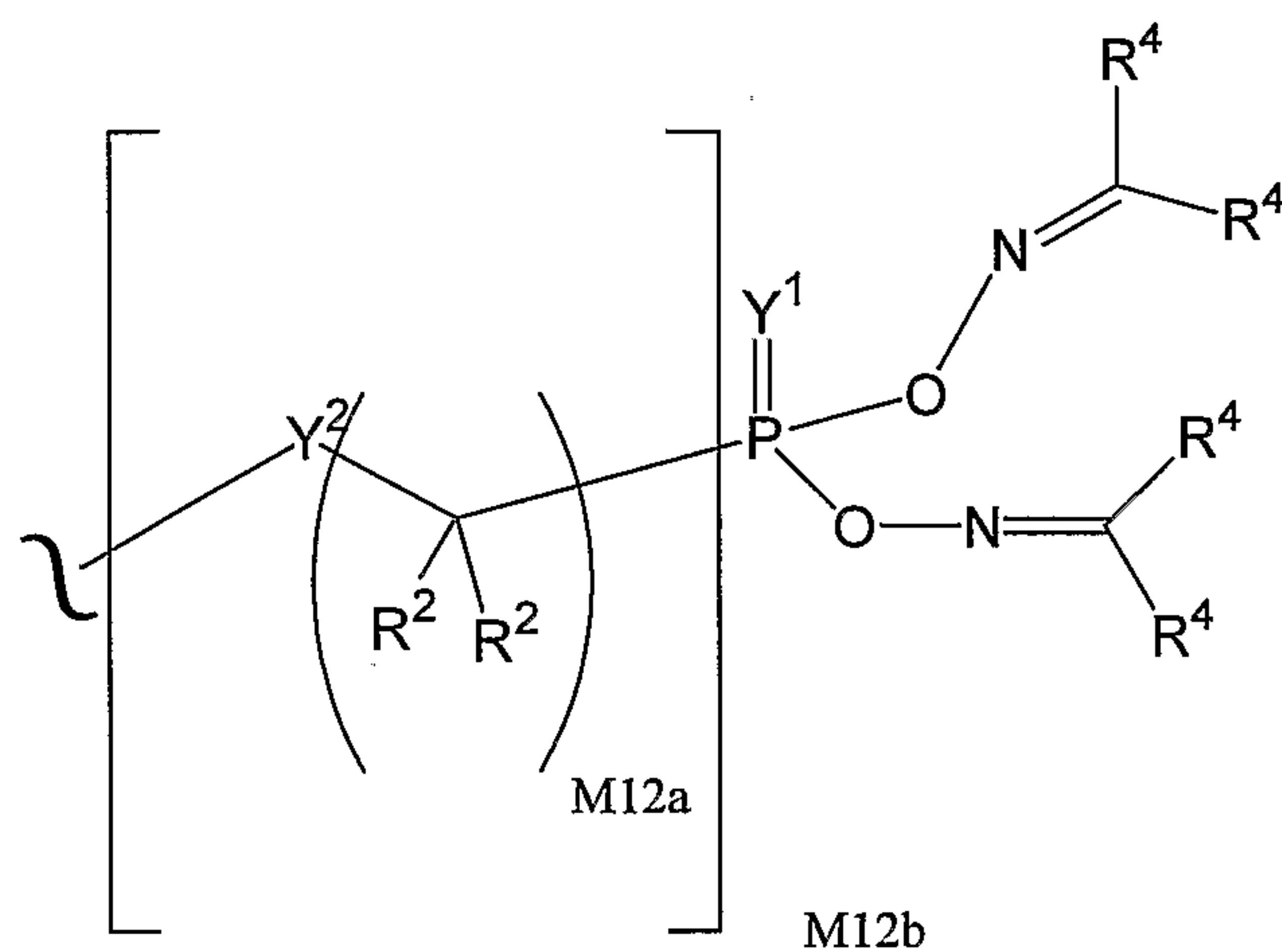


In some embodiments of the invention, A^3 is of the formula:



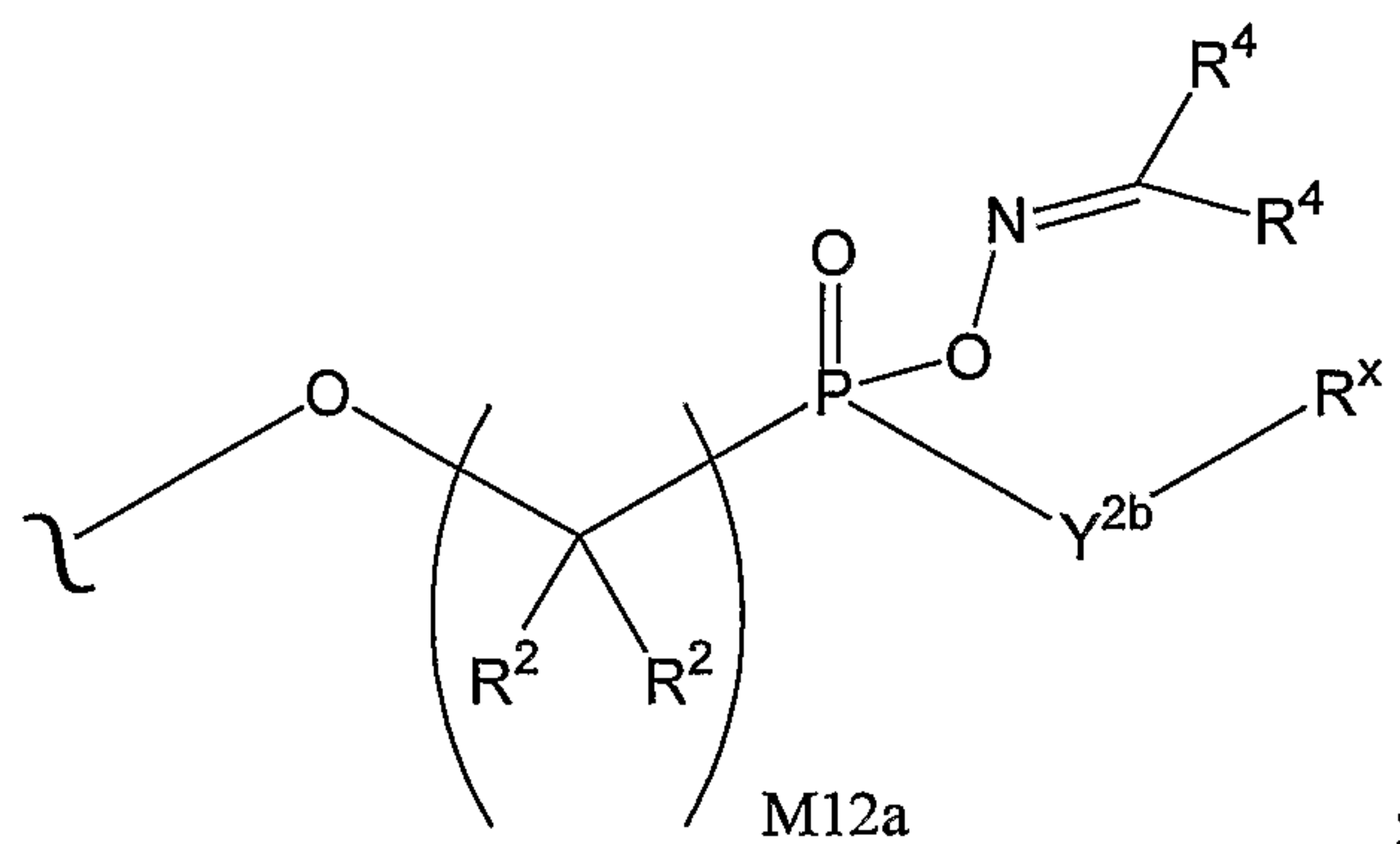
In some embodiments of the invention, Y^2 is independently O, $N(R^x)$ or S.

In some embodiments of the invention, A^3 is of the formula:



5

In some embodiments of the invention, A^3 is of the formula:

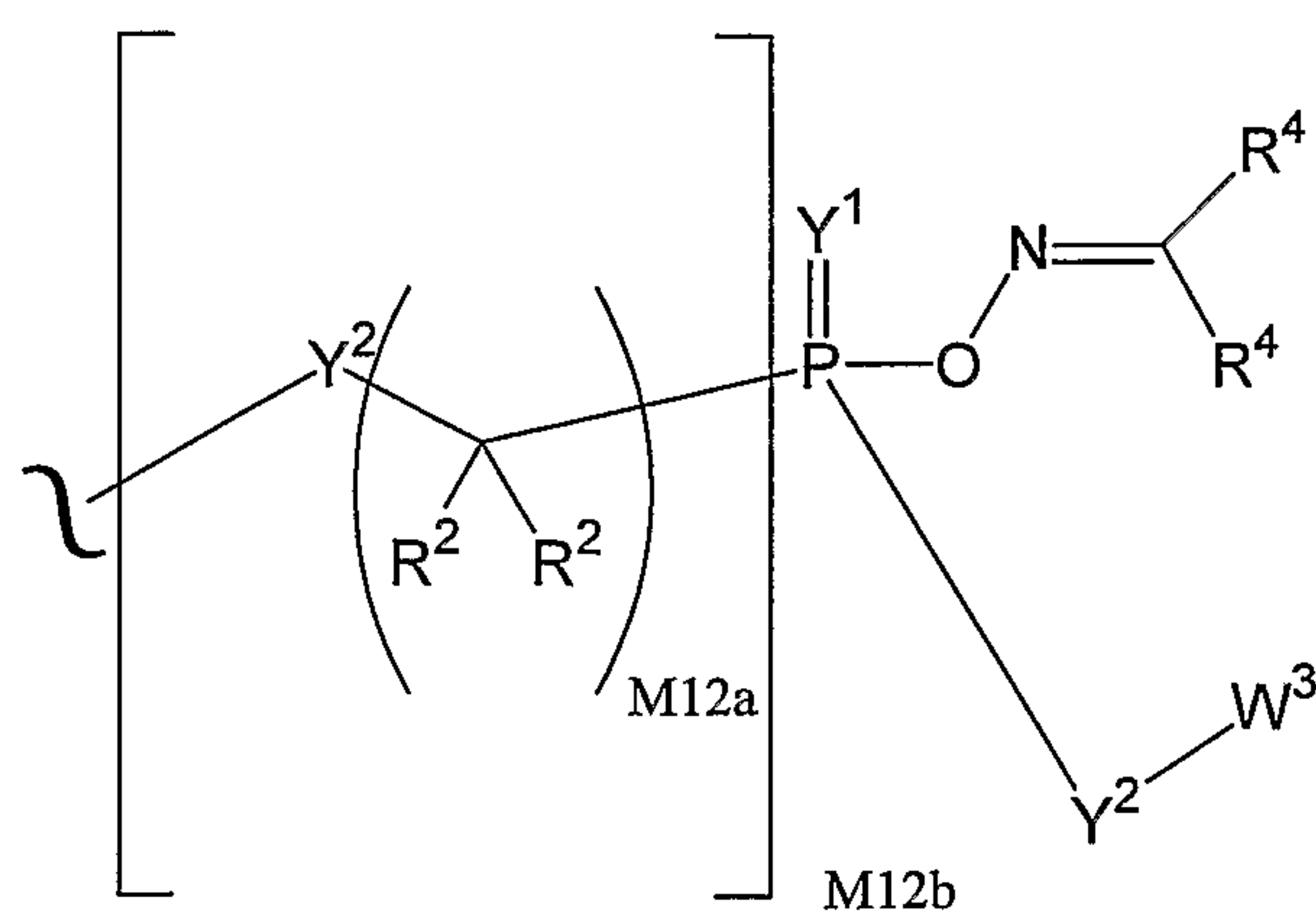


and Y^{2b} is O or $N(R^x)$.

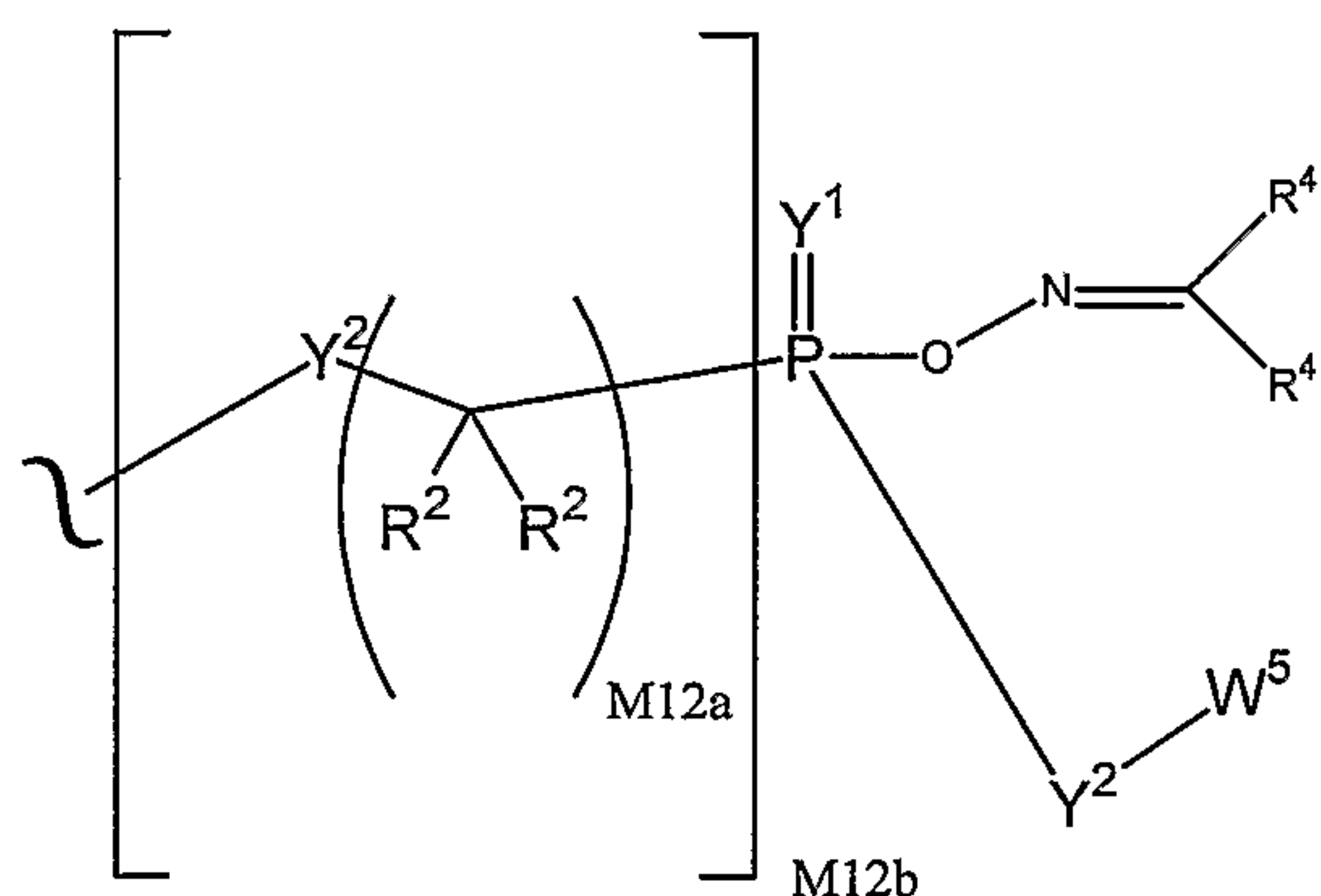
In some embodiments of the invention, M12a is 1, 2, 3, 4, 5, 6, 7 or 8.

10 In some embodiments of the invention, M12a is 1 and each R^2 is hydrogen.

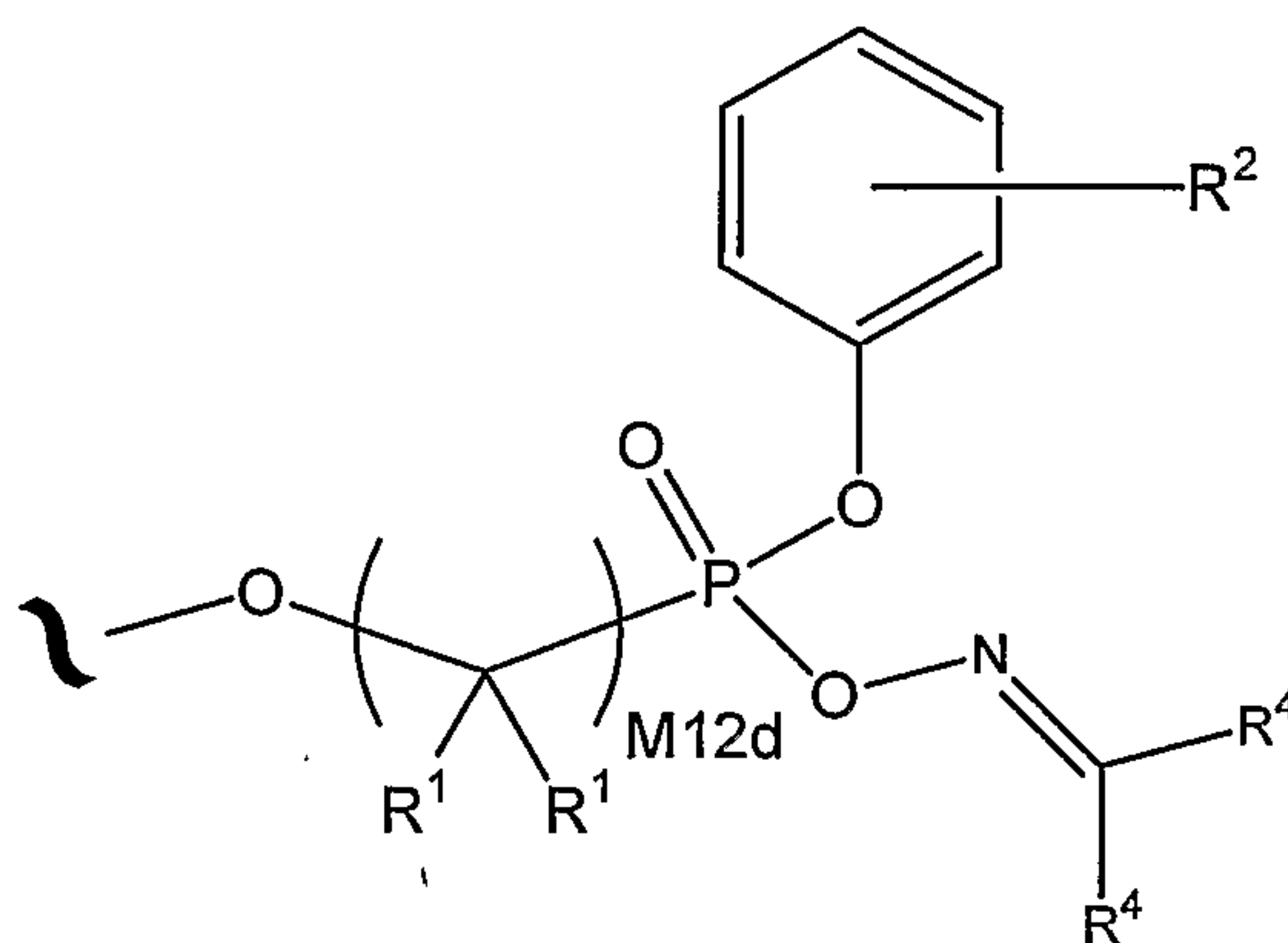
In some embodiments of the invention, A^3 is of the formula:



In some embodiments of the invention, A³ is of the formula:

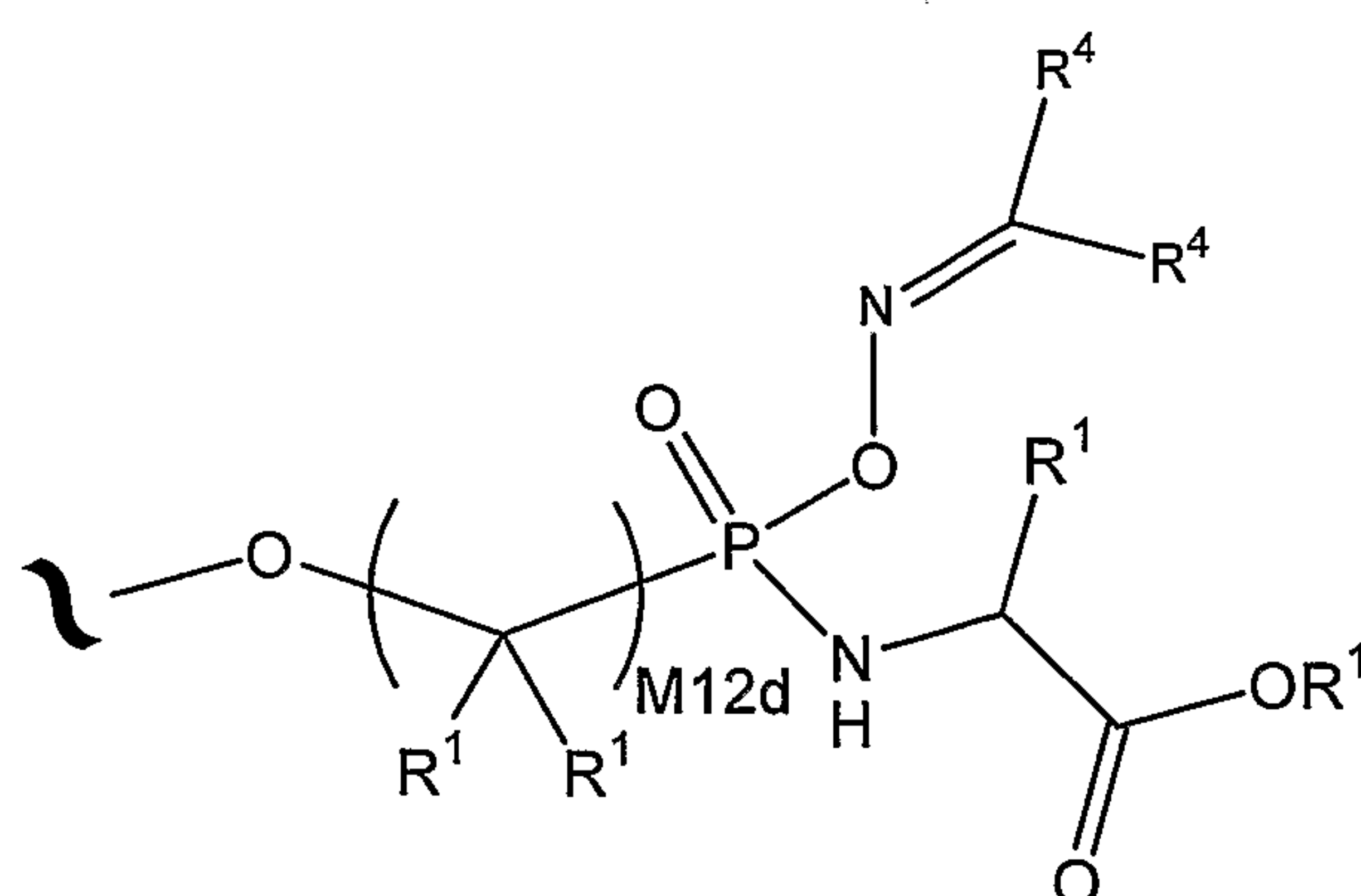


- 5 In some embodiments of the invention, W⁵ is a carbocycle.
In some embodiments of the invention, A³ is of the formula:



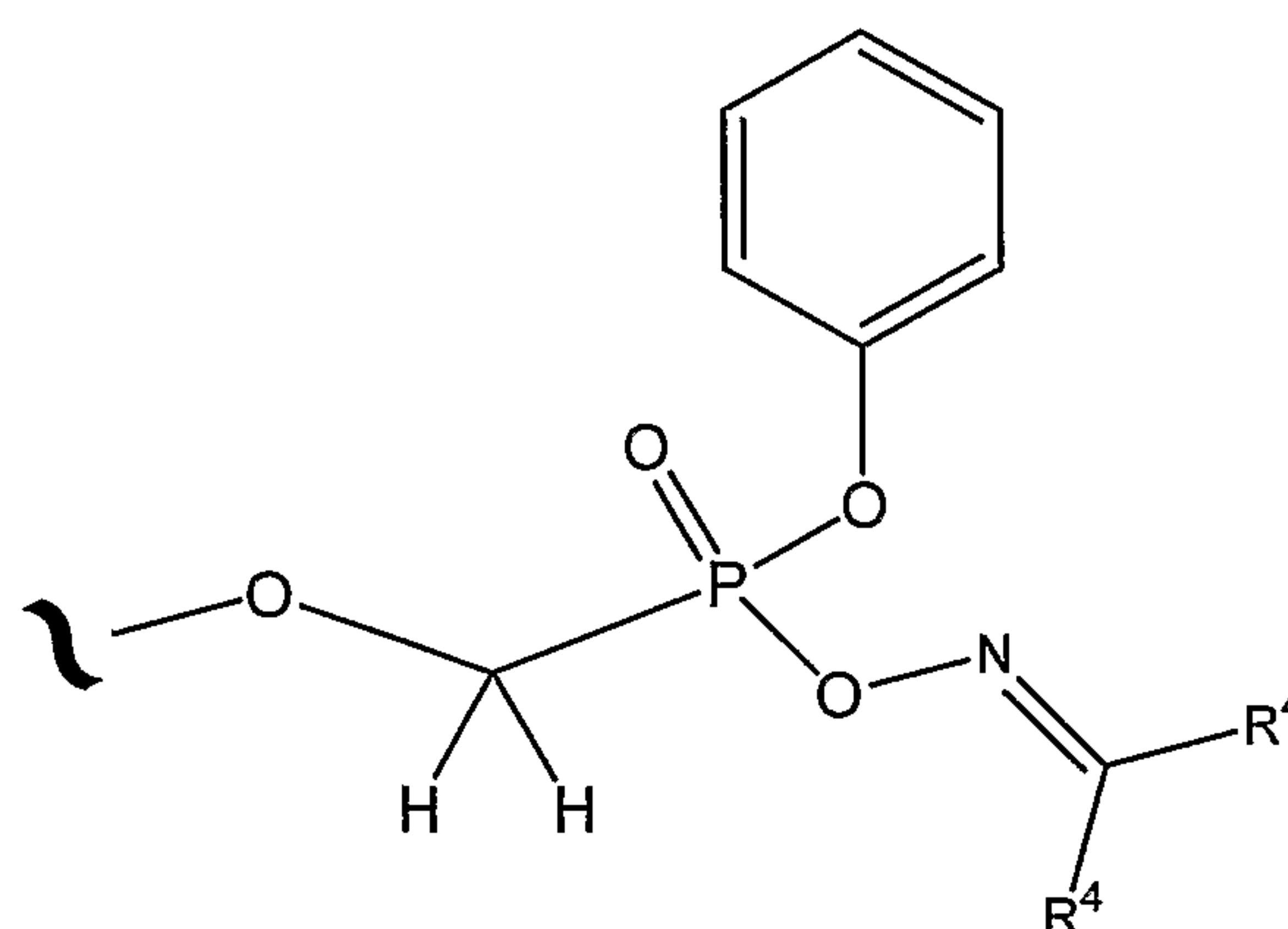
wherein R¹ is independently H or alkyl of 1 to 18 carbon atoms; and the phenyl carbocycle is substituted with 0, 1, 2, or 3 R² groups.

- 10 In some embodiments of the invention, A³ is of the formula:



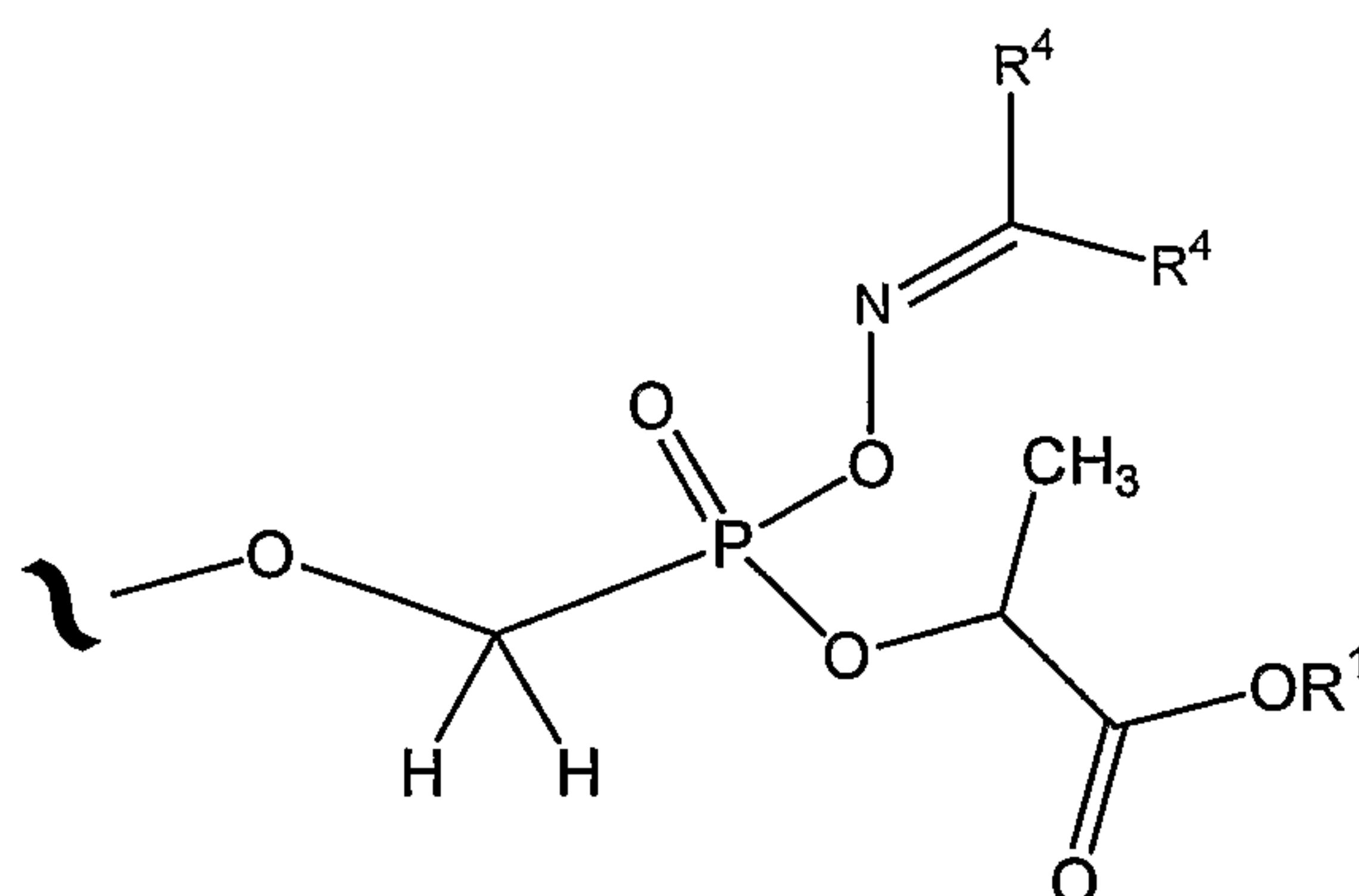
wherein R^1 is independently H or alkyl of 1 to 18 carbon atoms.

In some embodiments of the invention, A^3 is of the formula:



5

In some embodiments of the invention, A^3 is of the formula:



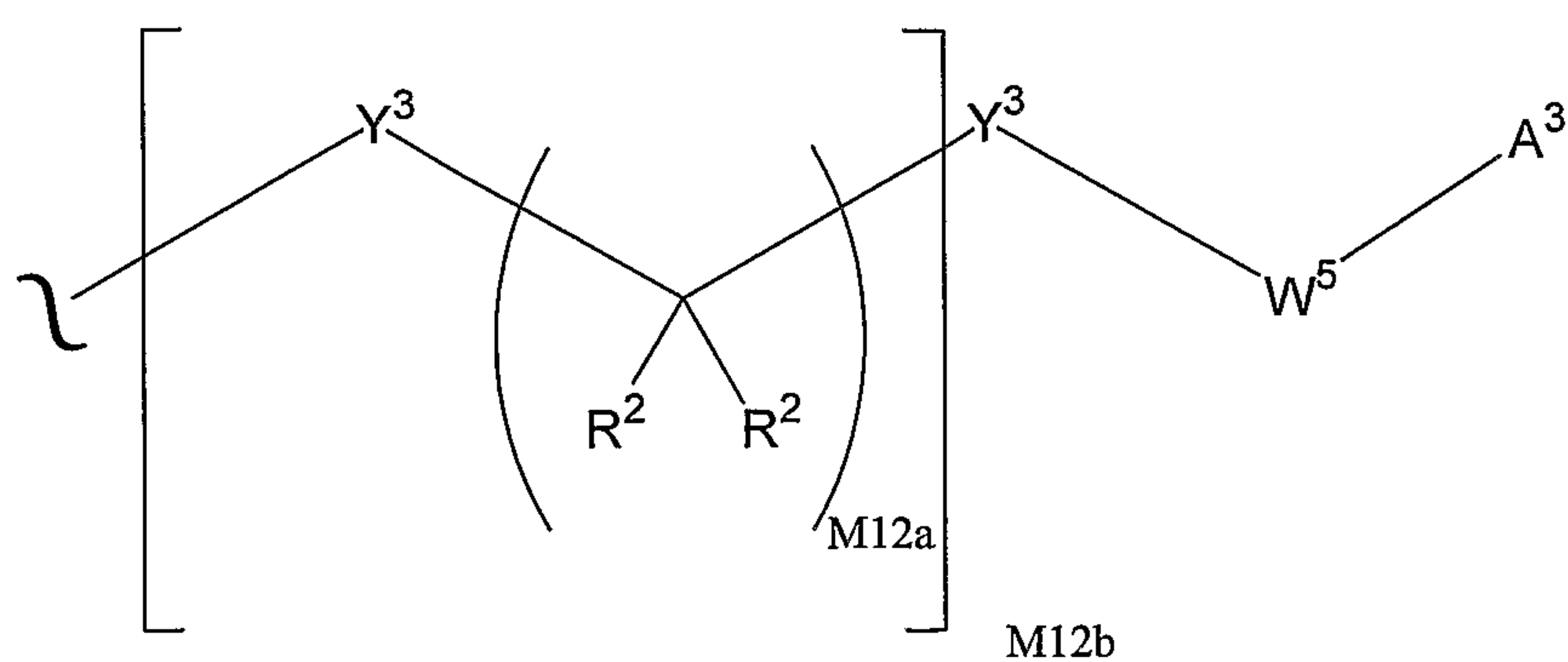
and R^1 is independently H or alkyl of 1 to 18 carbon atoms.

In some embodiments of the invention, R^{43} is ethyl or vinyl.

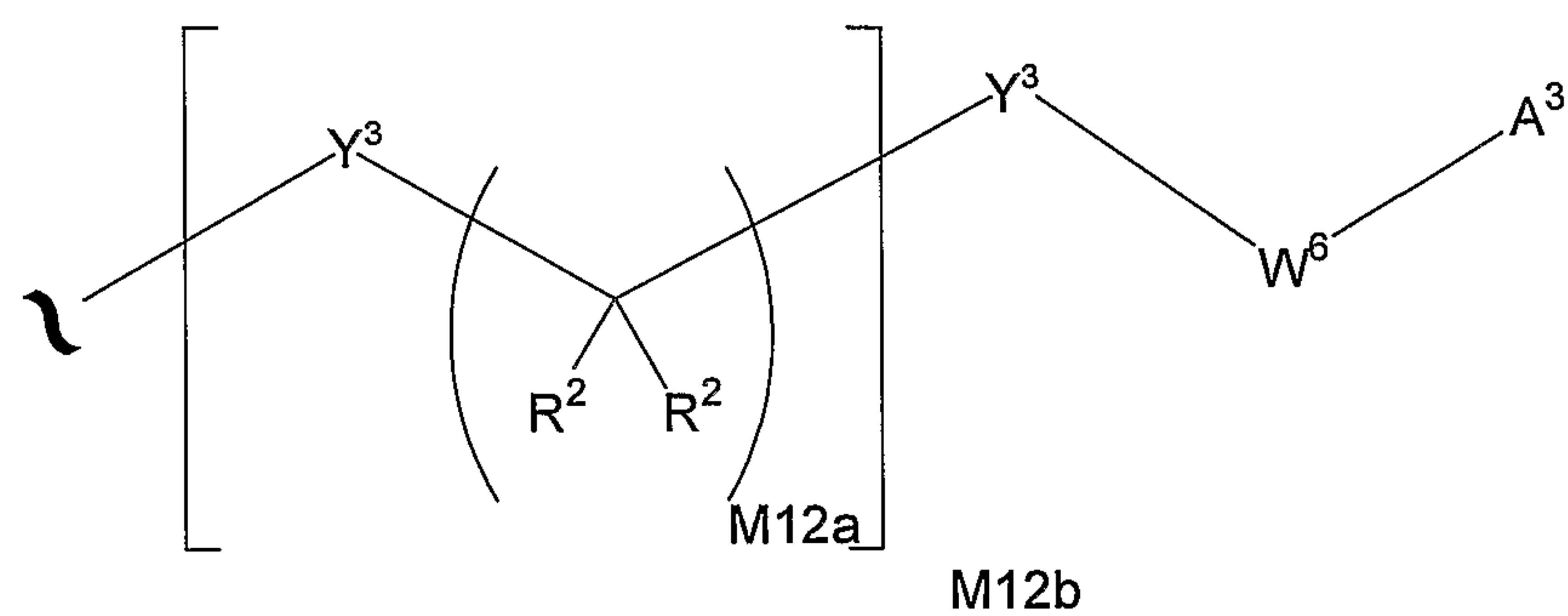
In some embodiments of the invention, R^{43} is methoxy.

10

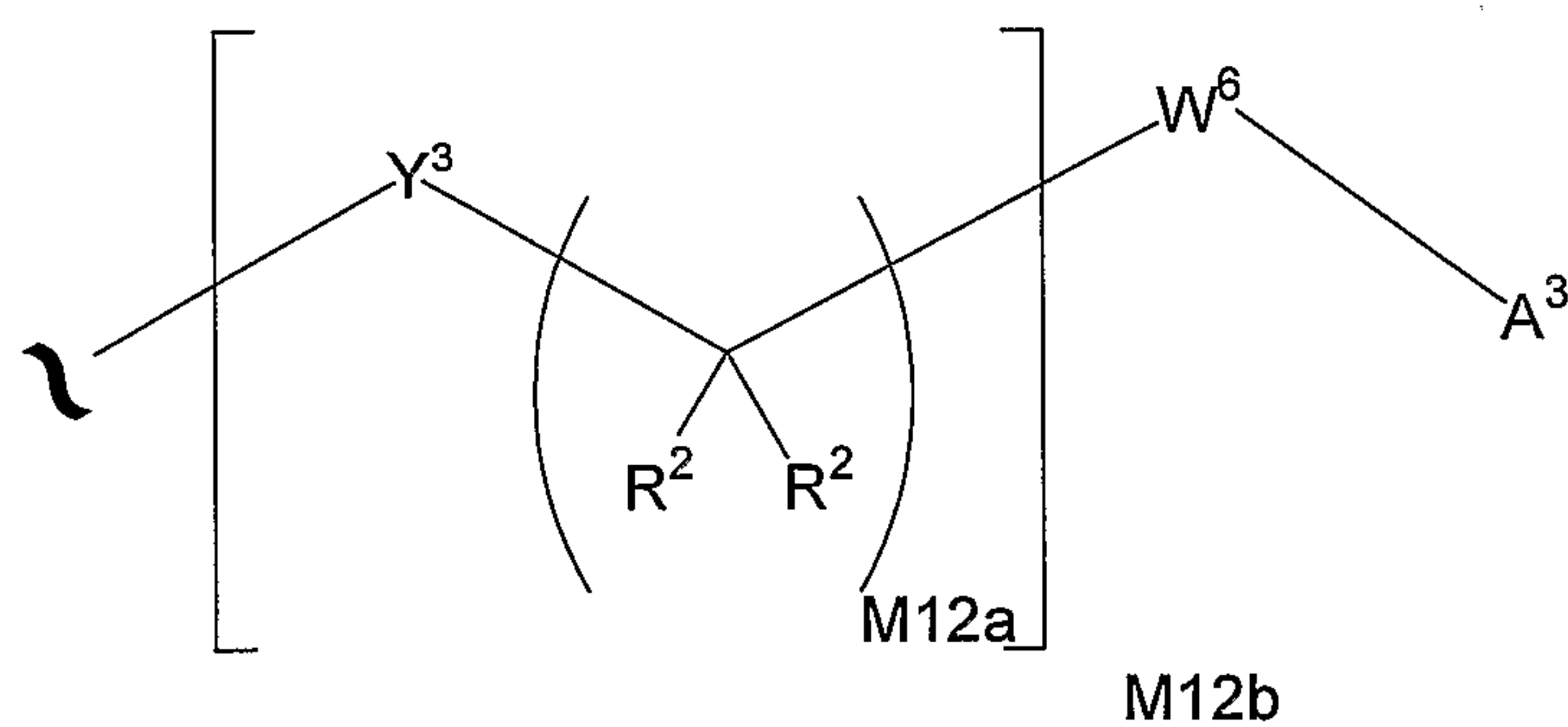
In some embodiments of the invention, A^1 is of the formula:



In some embodiments of the invention, A^1 is of the formula:

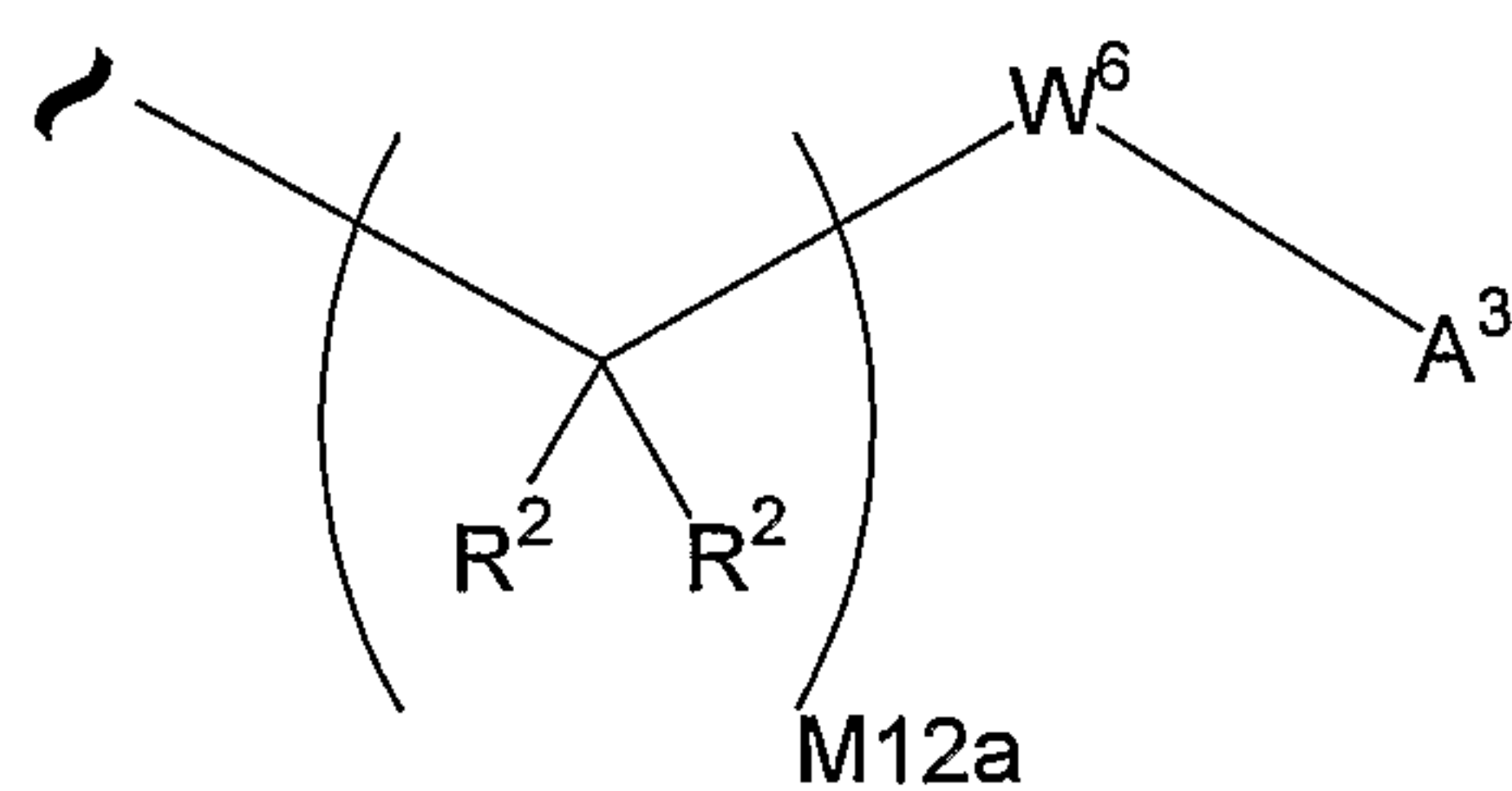


In some embodiments of the invention, A^1 is of the formula:

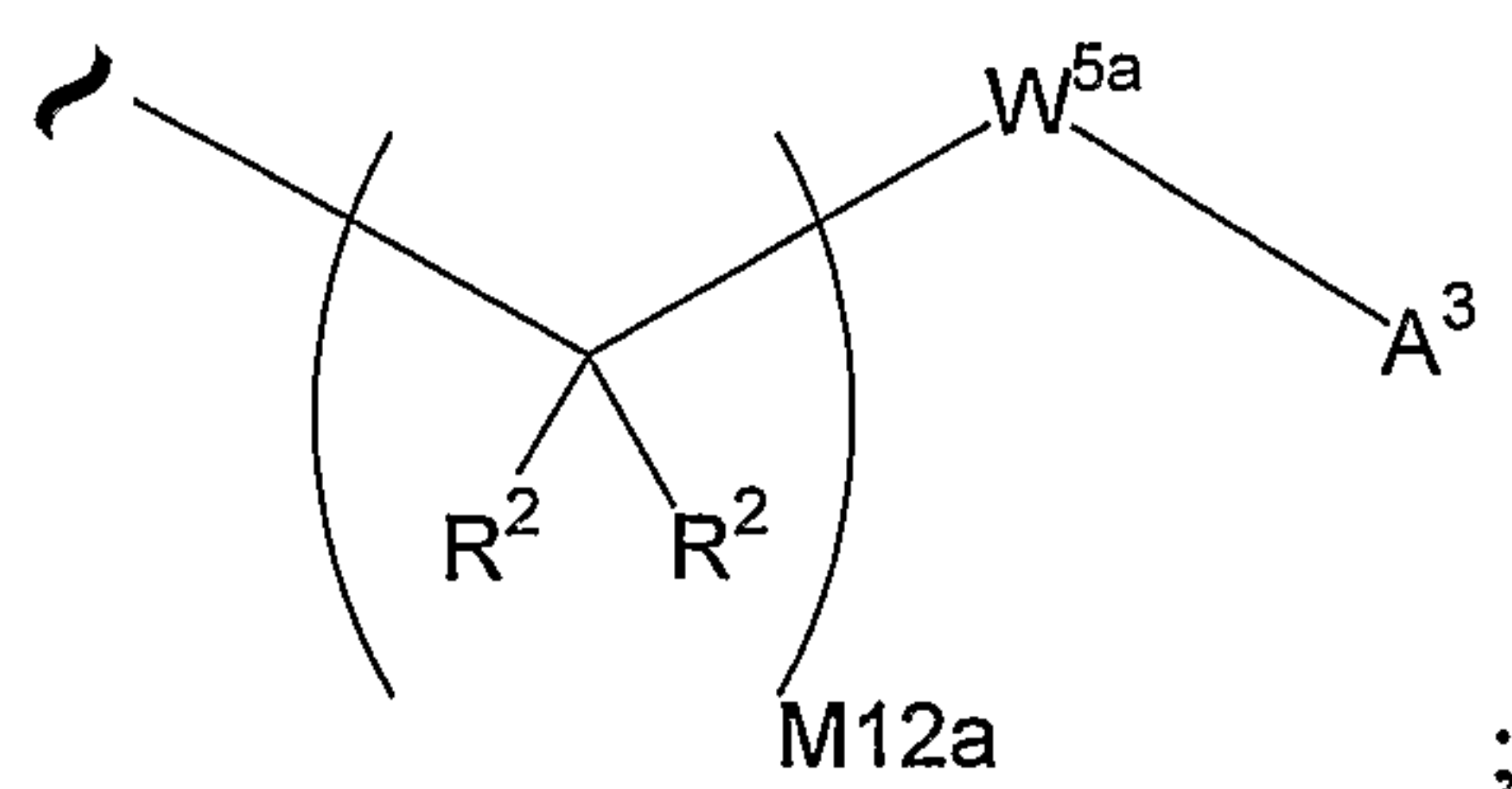


5

In some embodiments of the invention, A^1 is of the formula:



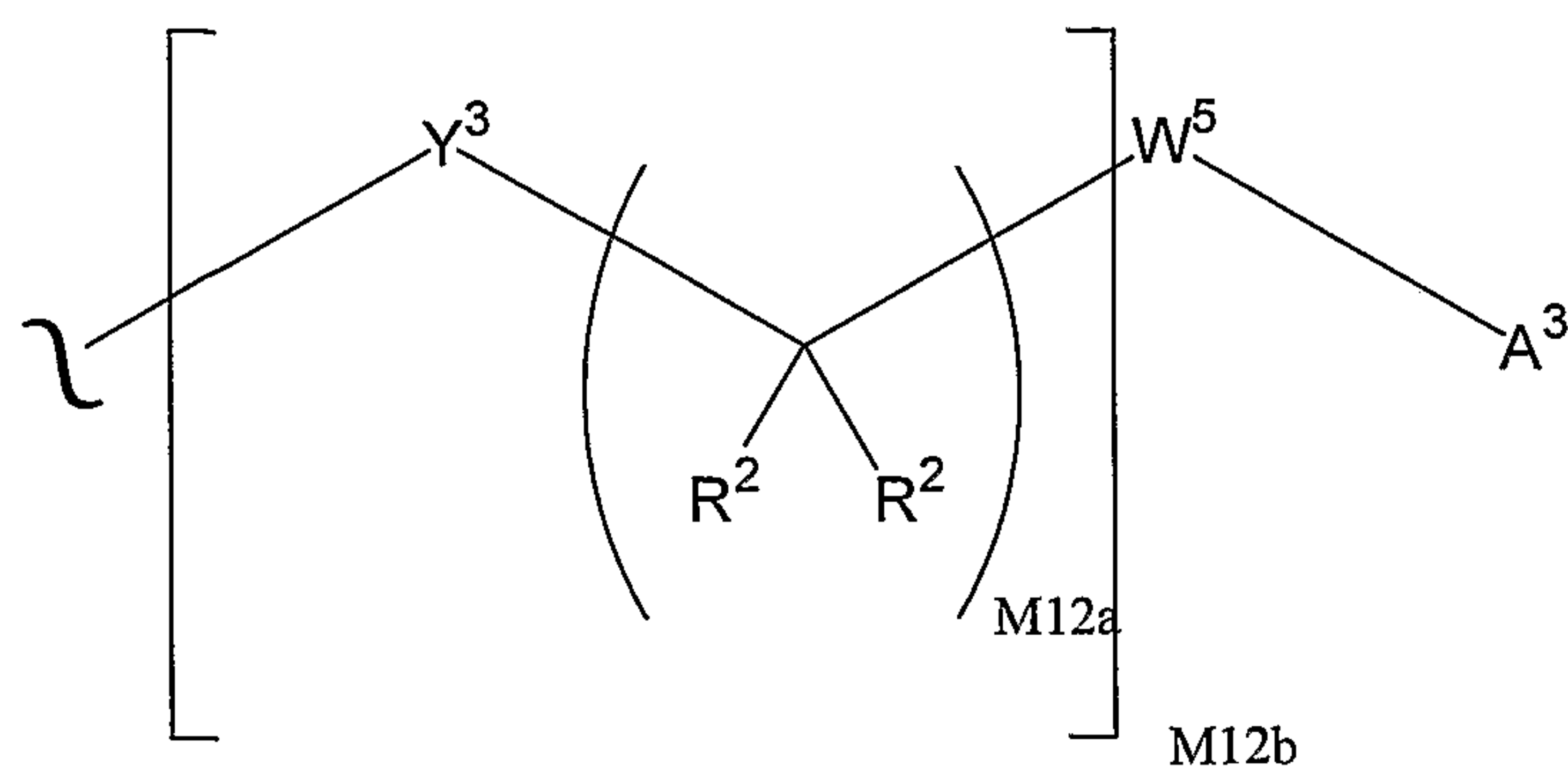
In some embodiments of the invention, A^1 is of the formula:



wherein W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

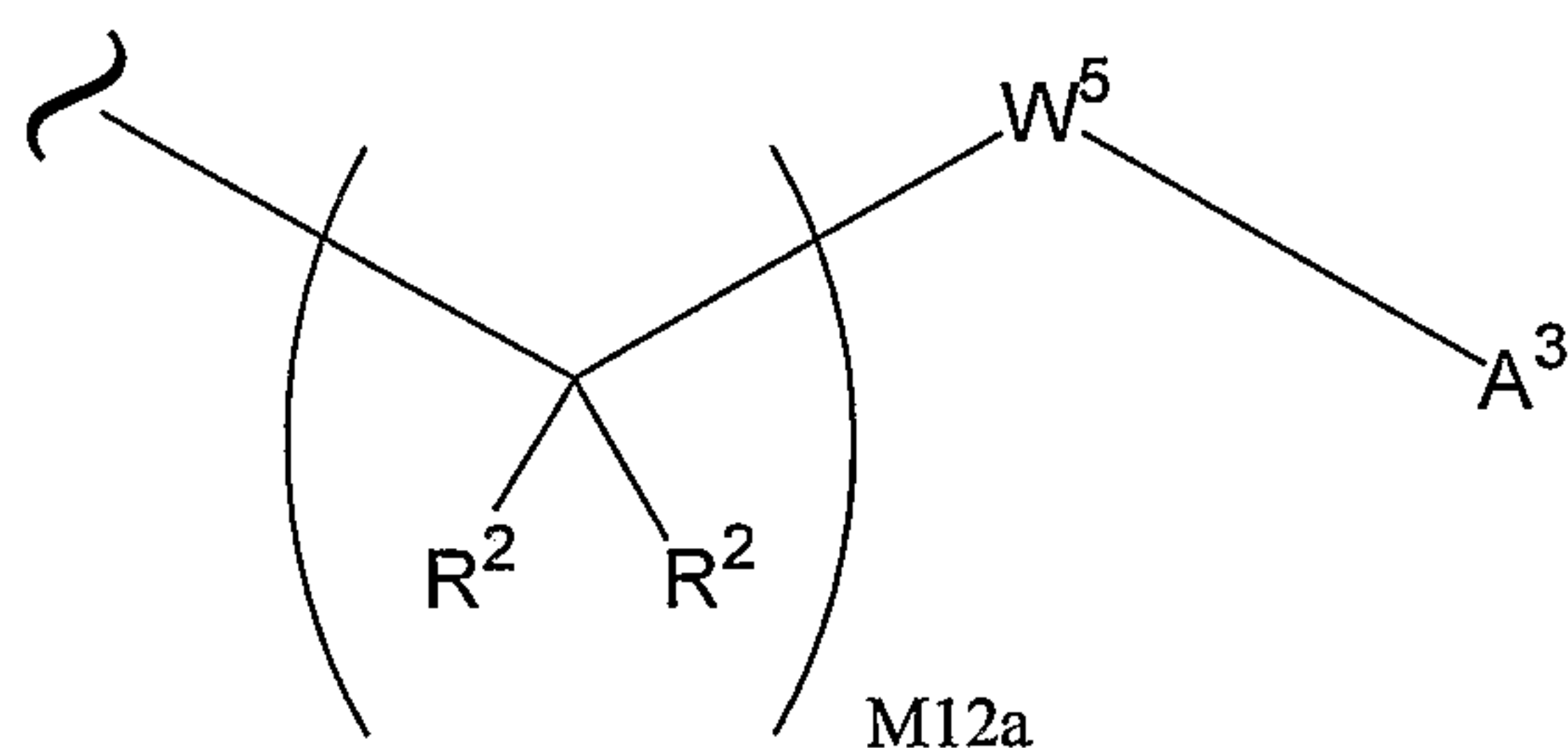
In some embodiments of the invention, M12a is 1.

In some embodiments of the invention, A^1 is of the formula:

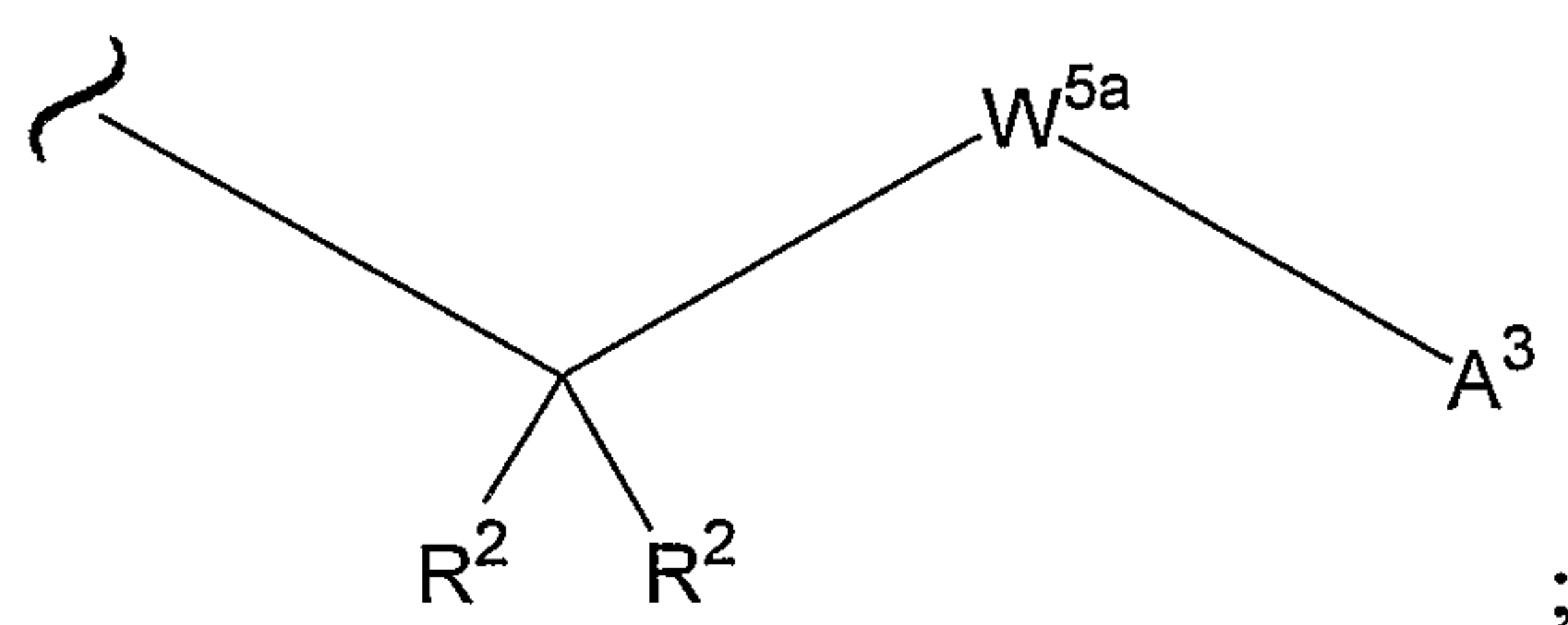


5

In some embodiments of the invention, A^1 is of the formula:



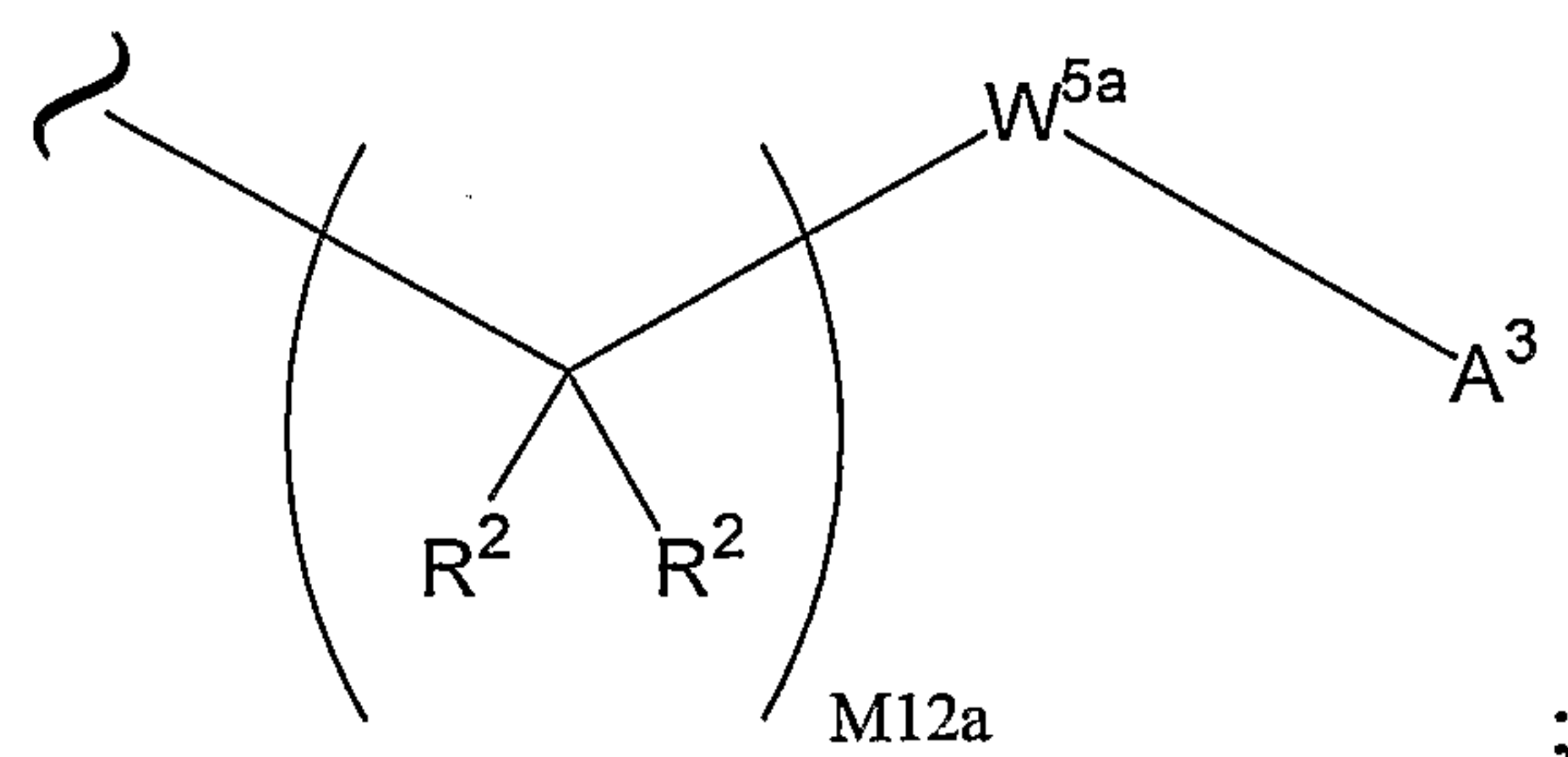
In some embodiments of the invention, A^1 is of the formula:



10

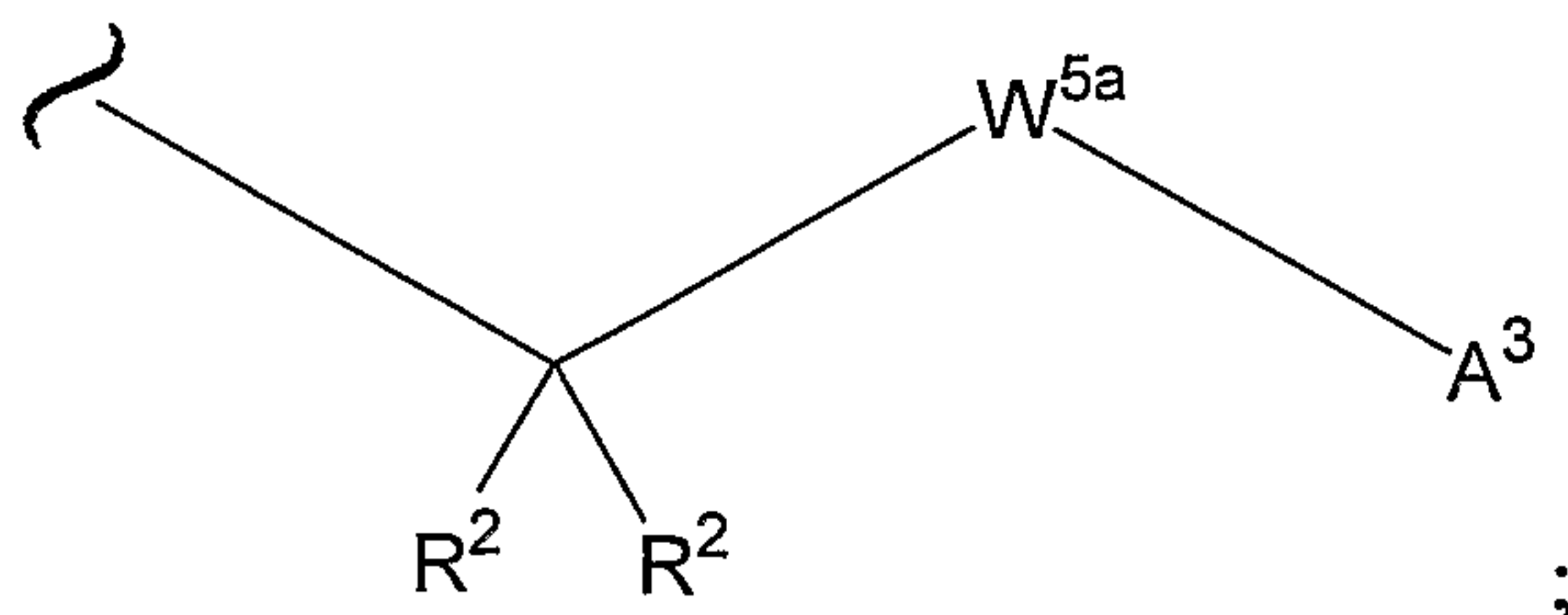
wherein W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^1 is of the formula:



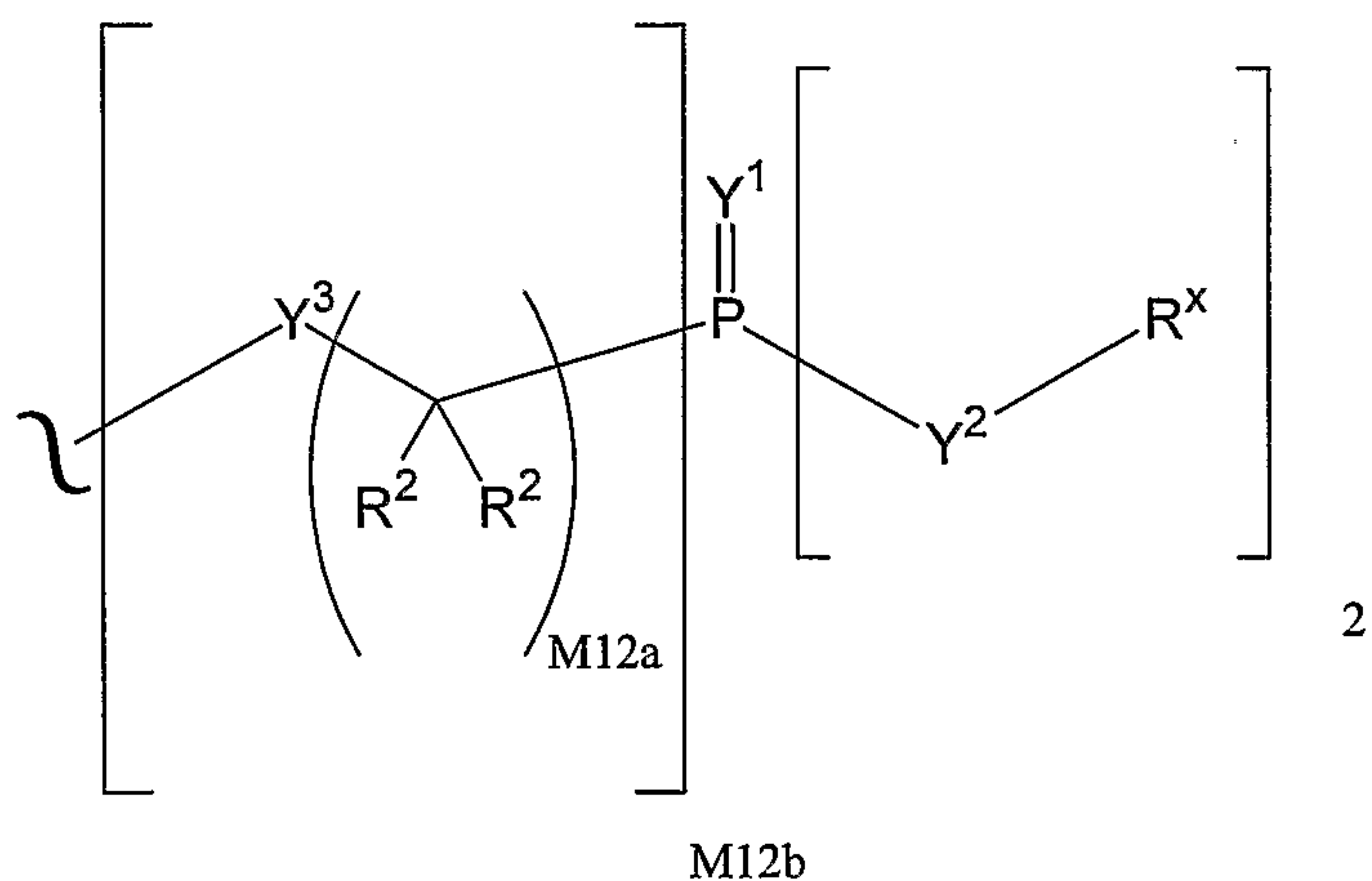
wherein W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^1 is of the formula:

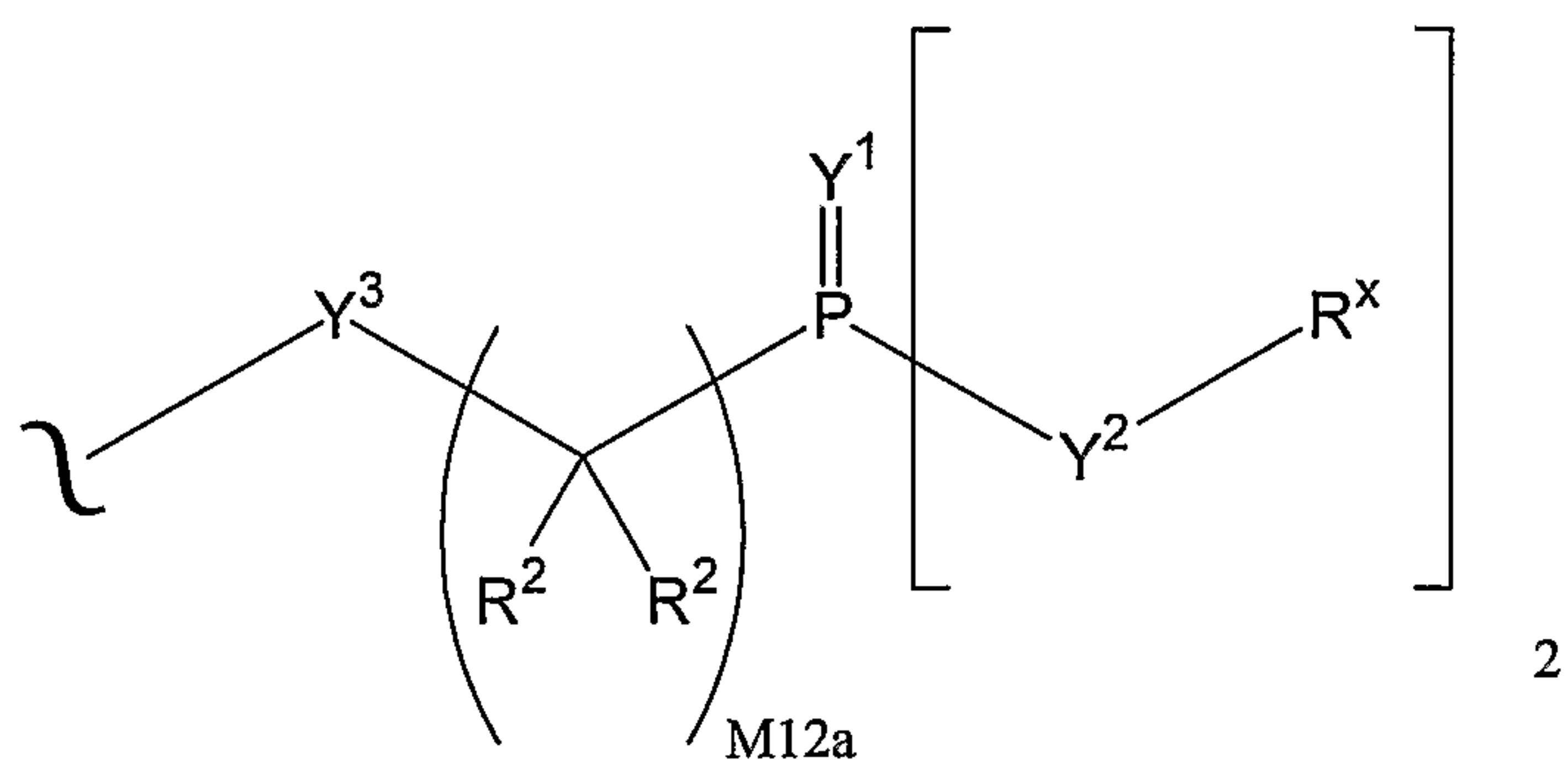


5 wherein W^{5a} is a carbocycle or heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

In some embodiments of the invention, A^3 is of the formula:

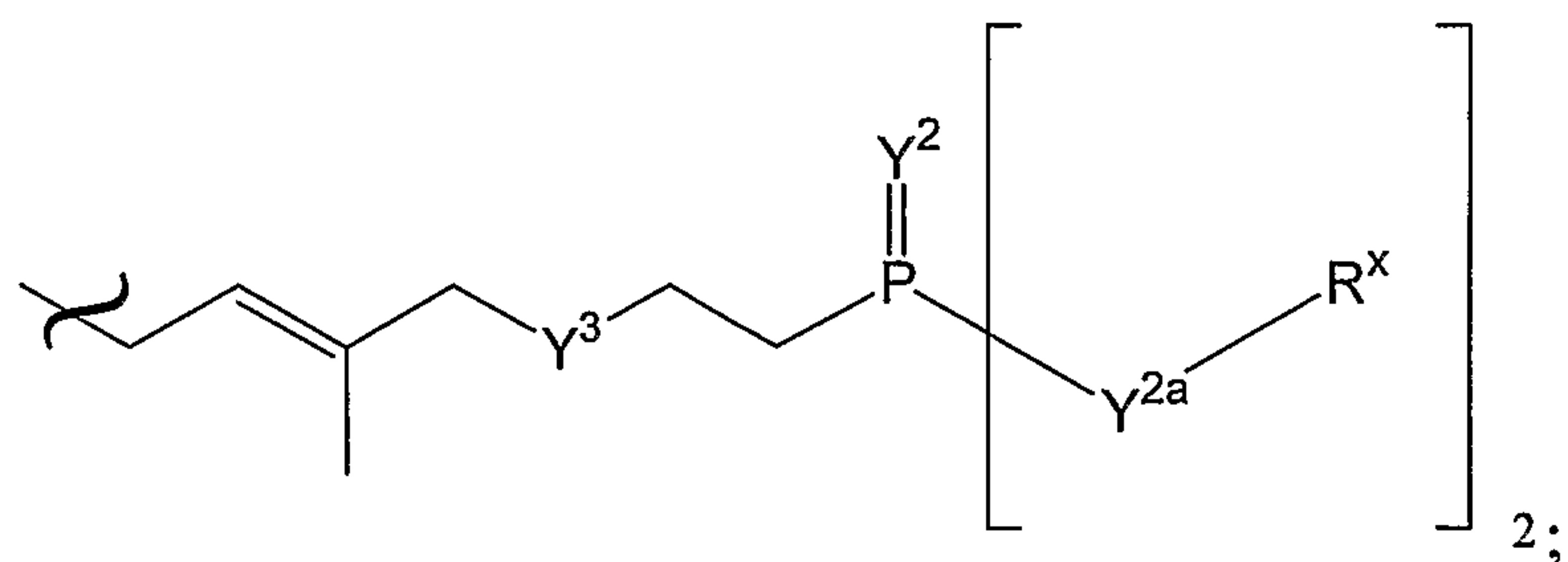


In some embodiments of the invention, A^3 is of the formula:



10

In some embodiments of the invention, A^0 is of the formula:



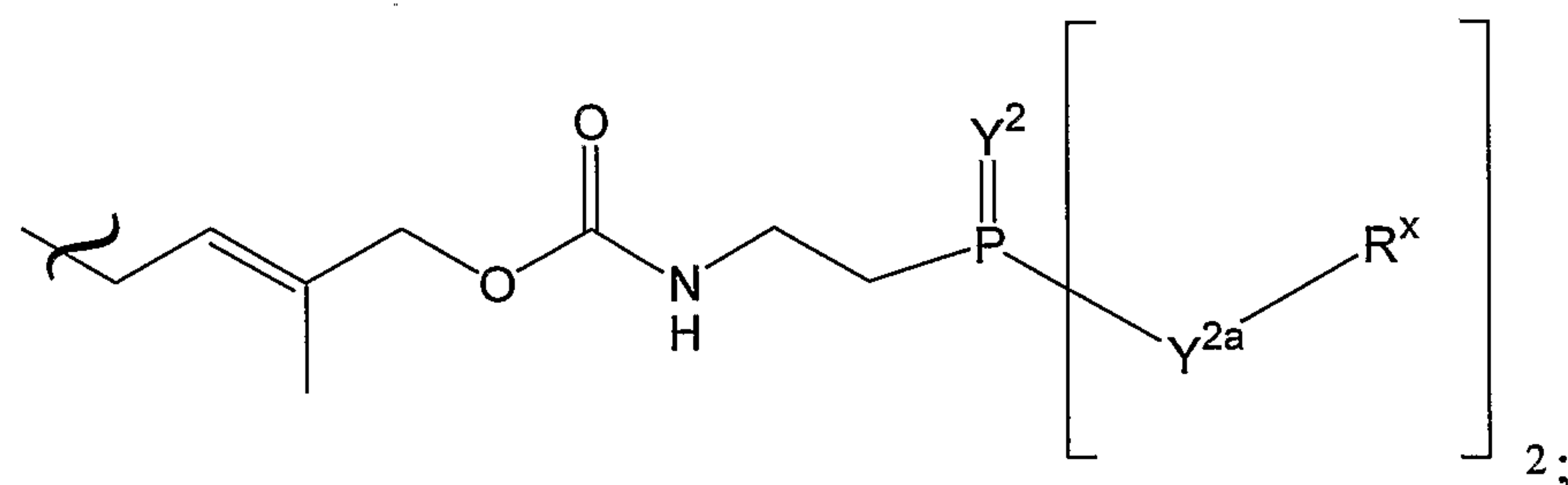
wherein:

Y^2 is O or S;

Y^{2a} is O, $N(R^x)$ or S; and

Y^3 is $-OC(=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)$,
 5 $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$.

In some embodiments of the invention, A^0 is of the formula:

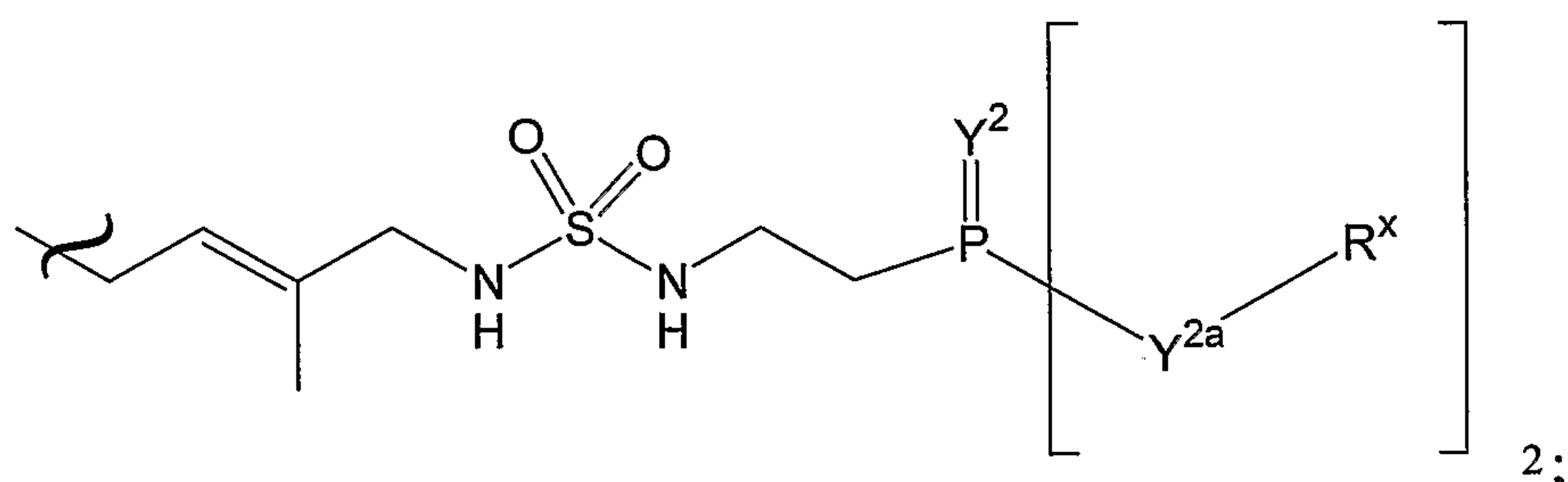


wherein:

Y^2 is O or S; and

10 Y^{2a} is O, $N(R^x)$ or S.

In some embodiments of the invention, A^0 is of the formula:

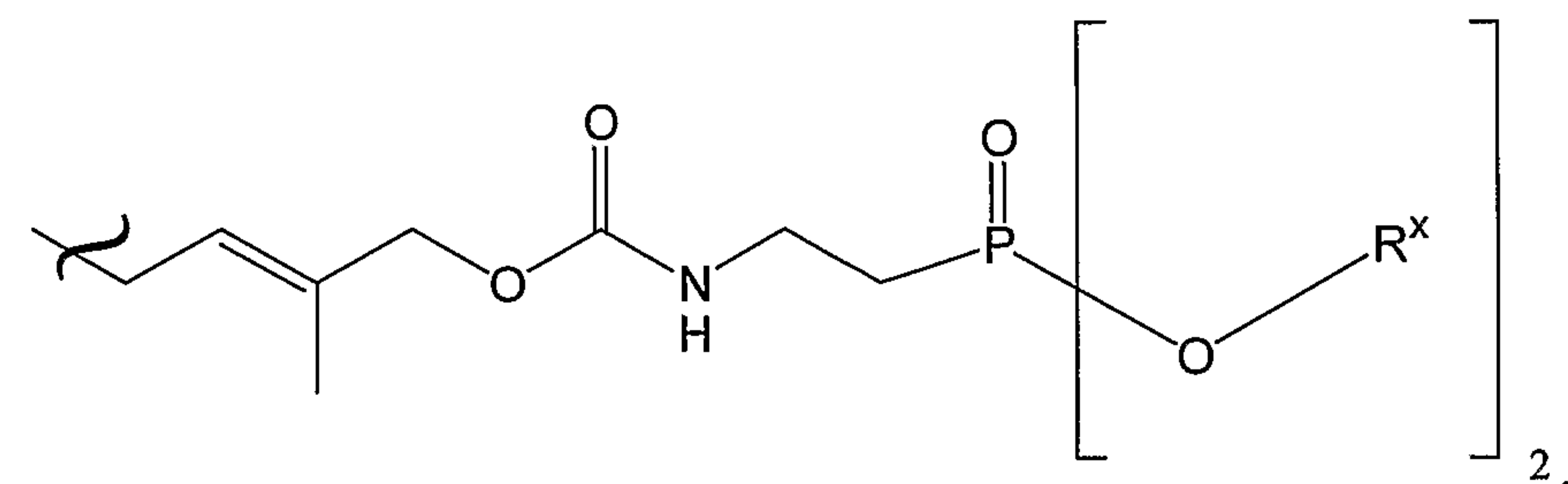


wherein:

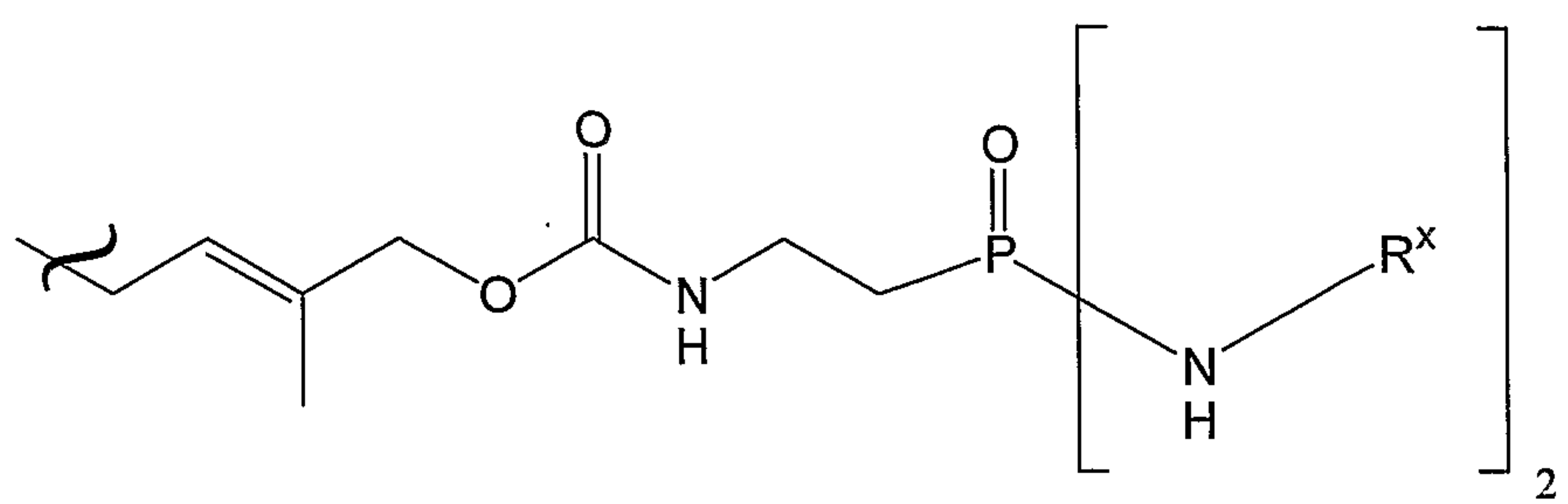
Y^2 is O or S; and

15 Y^{2a} is O, $N(R^x)$ or S.

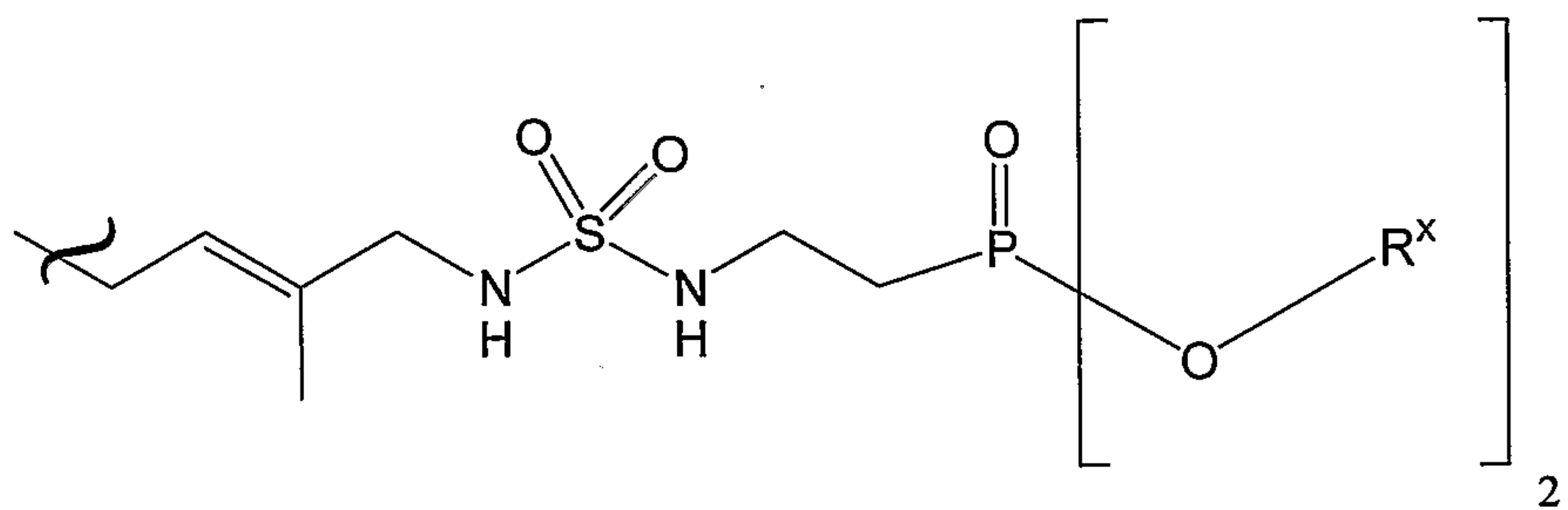
In some embodiments of the invention, A^0 is of the formula:



In some embodiments of the invention, A^0 is of the formula:



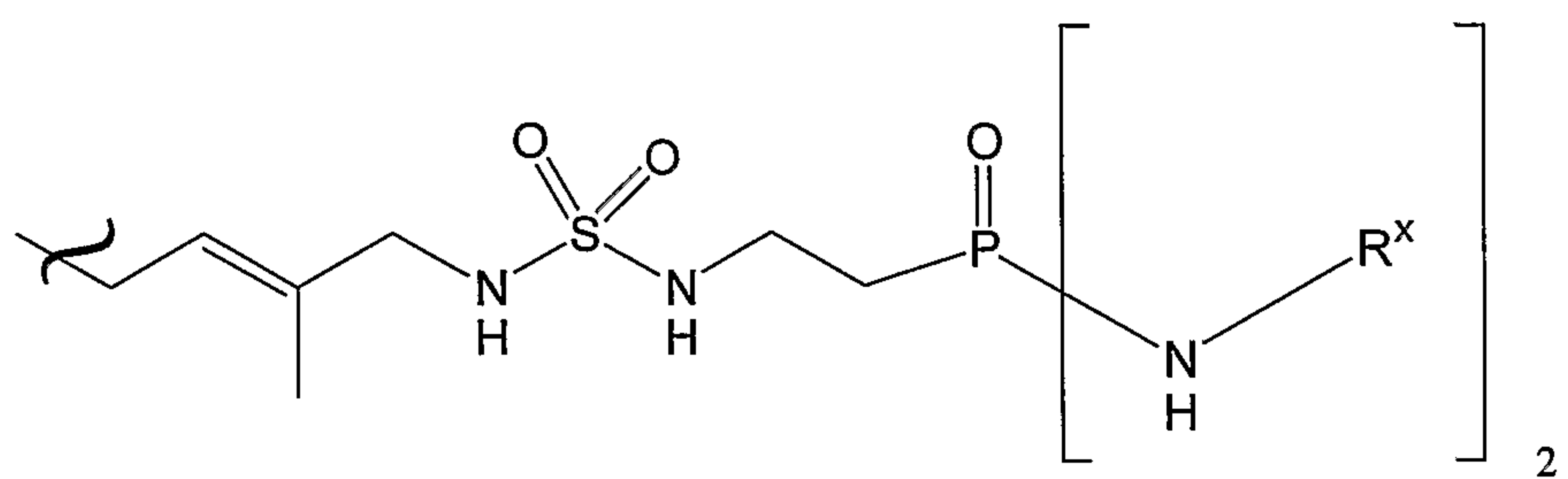
In some embodiments of the invention, A^0 is of the formula:



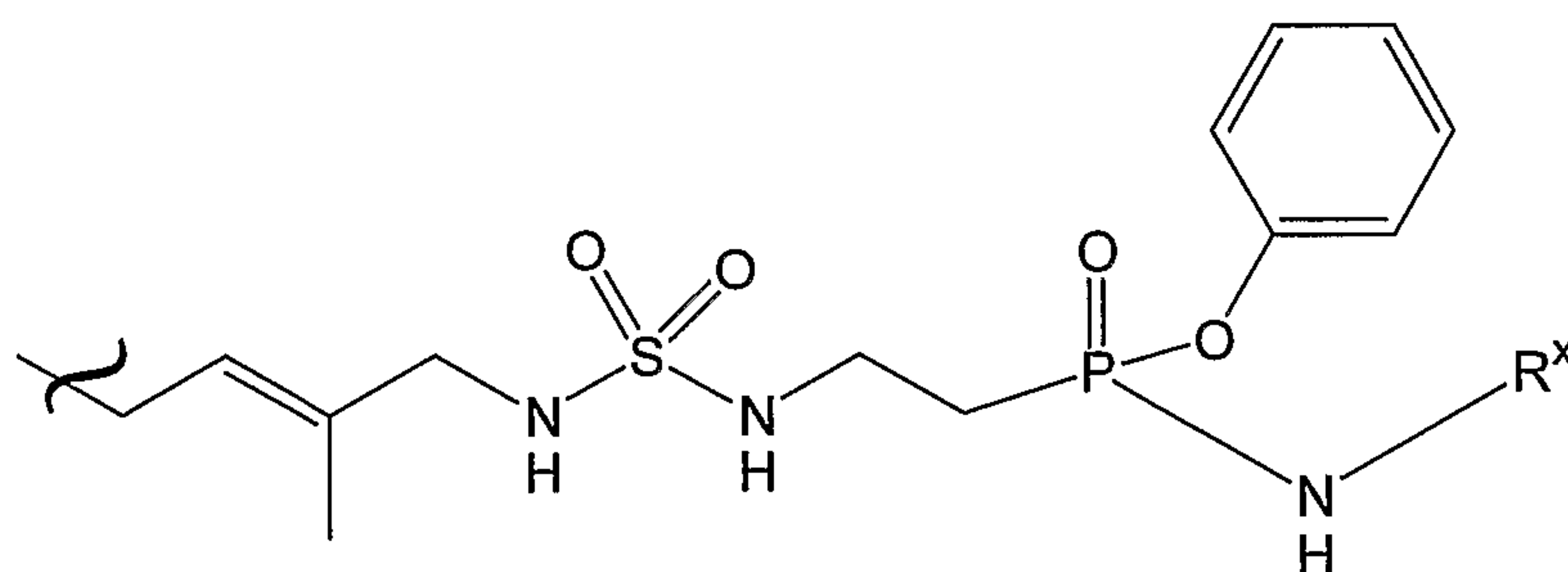
In some embodiments of the invention, R^x is hydrogen or alkyl.

5

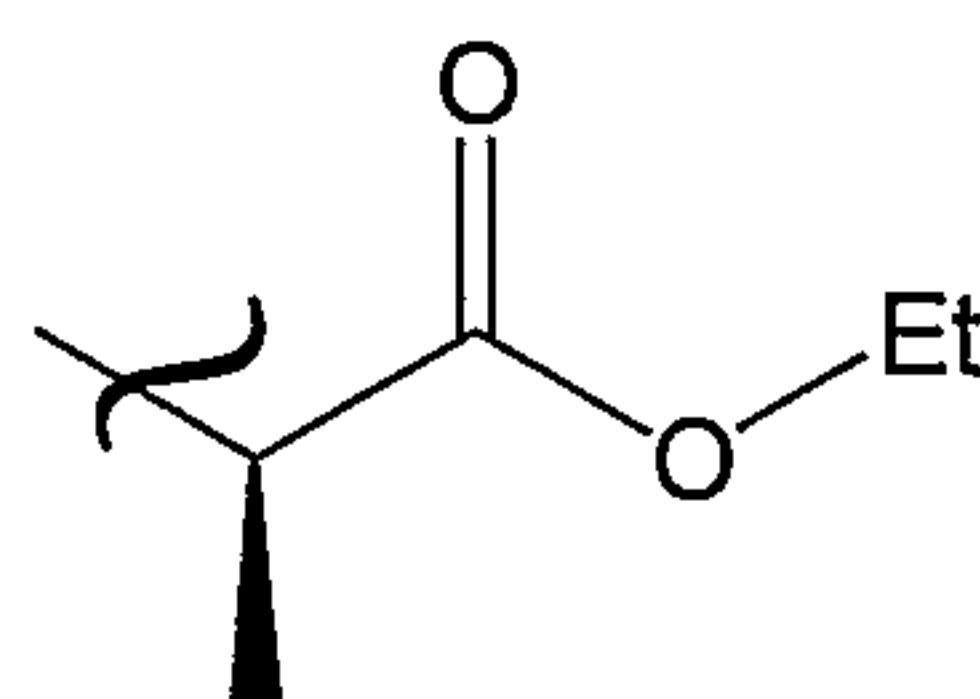
In some embodiments of the invention, A^0 is of the formula:



In some embodiments of the invention, A^0 is of the formula:

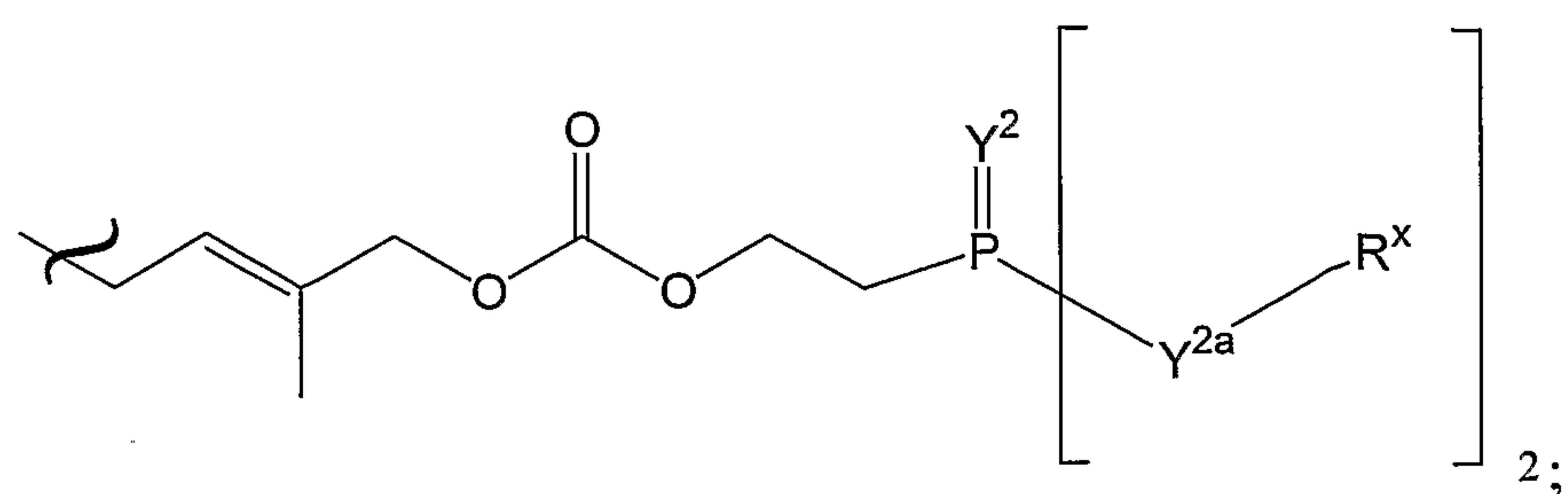


In some embodiments of the invention, R^x is of the formula:



10

In some embodiments of the invention, A^0 is of the formula:

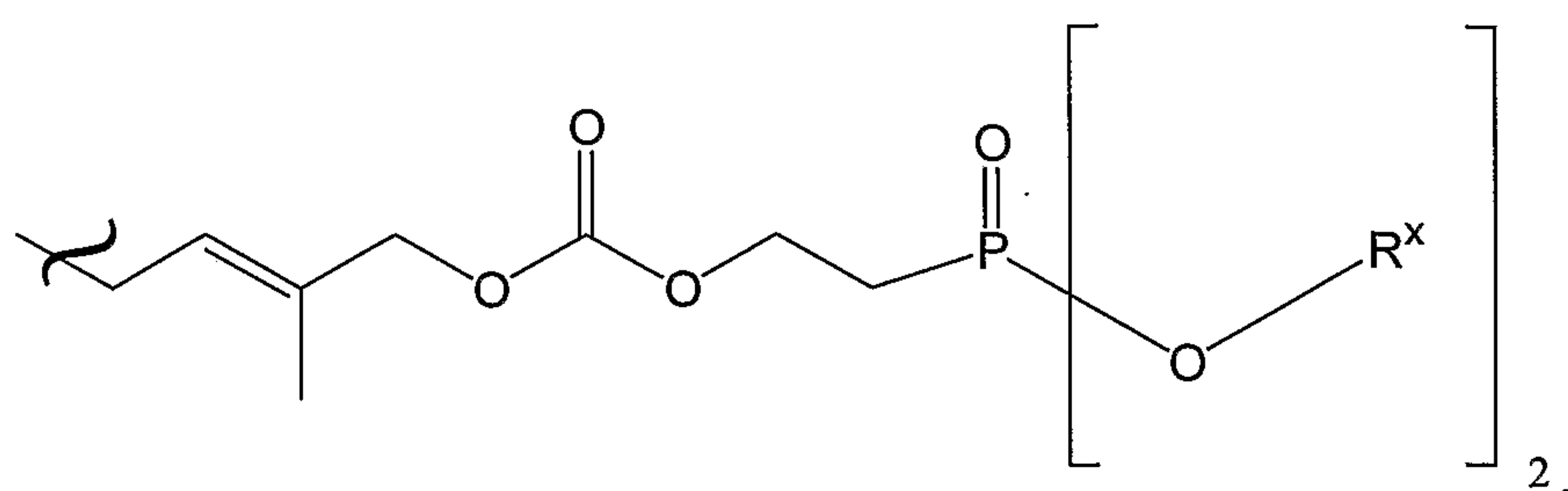


wherein:

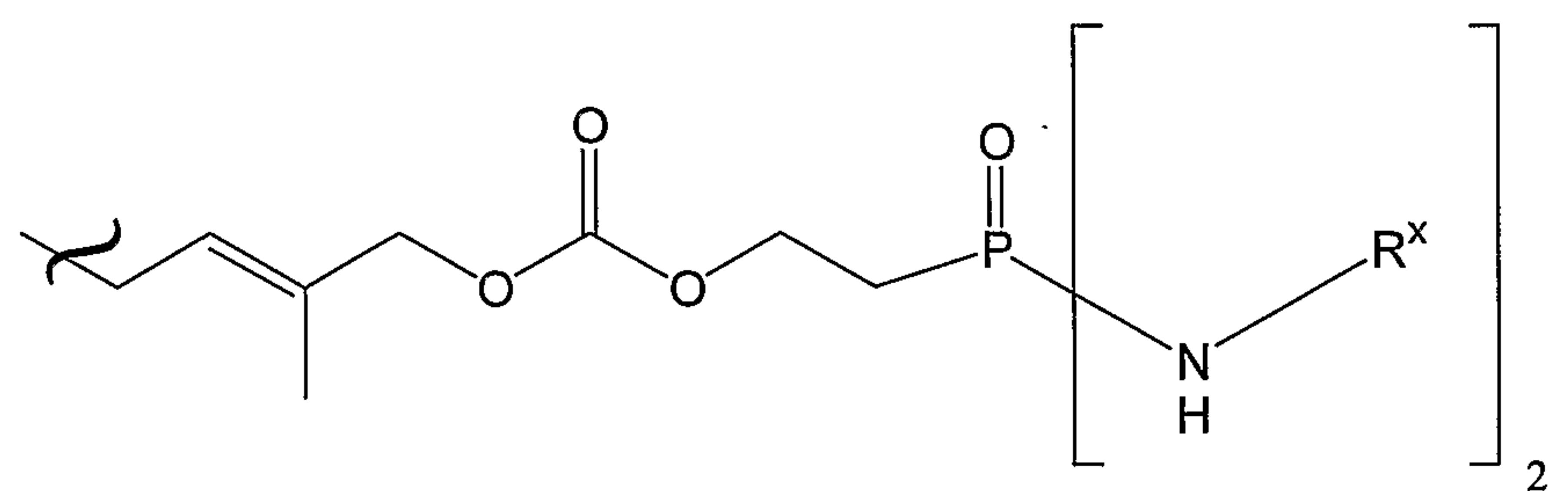
Y^2 is O or S; and

Y^{2a} is O, $\text{N}(\text{R}^x)$ or S.

5 In some embodiments of the invention, A^0 is of the formula:

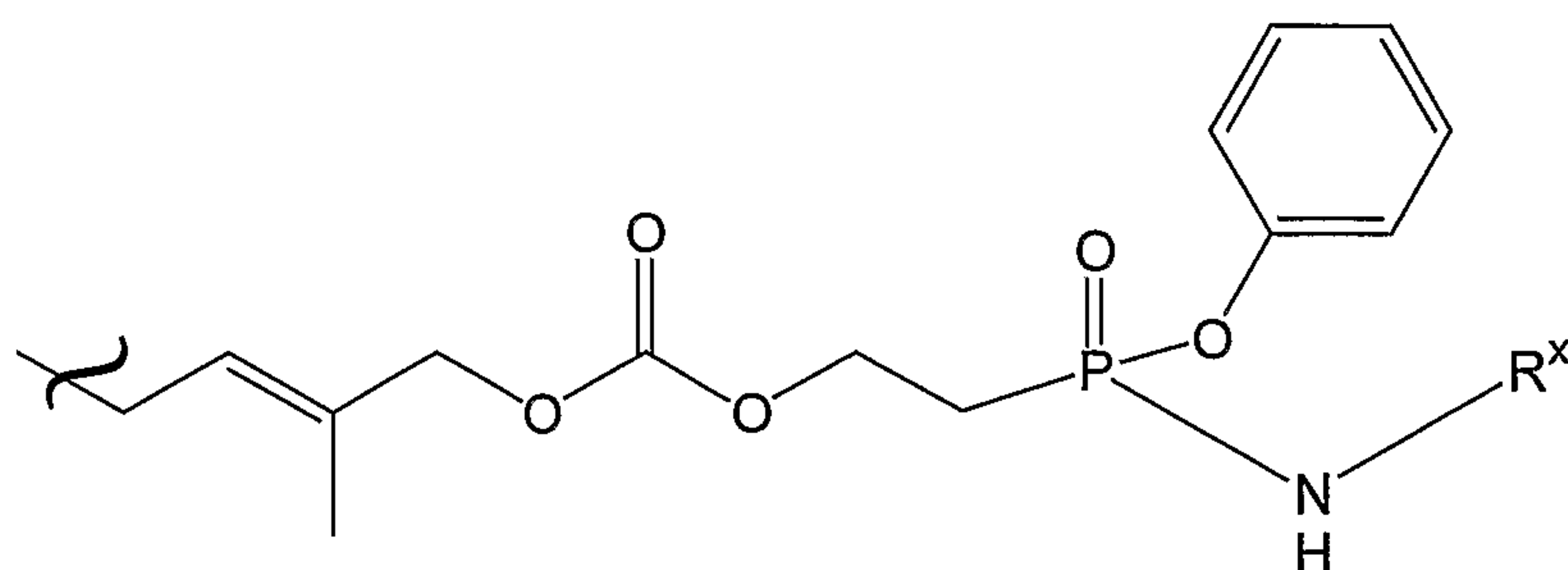


In some embodiments of the invention, A^0 is of the formula:

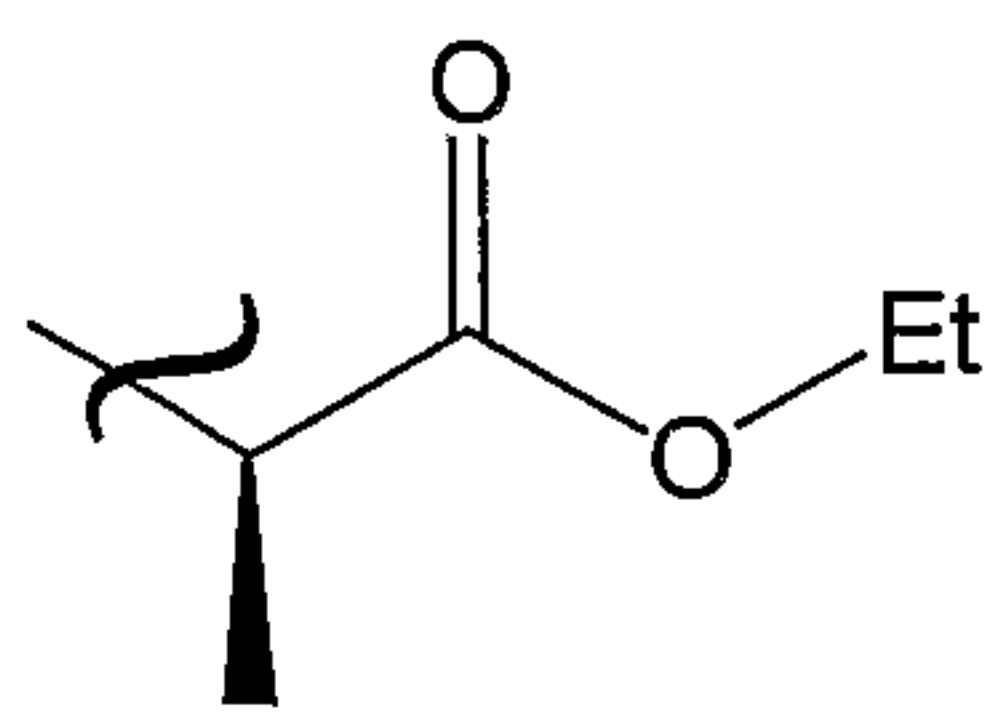


In some embodiments of the invention, R^x is hydrogen or alkyl.

10 In some embodiments of the invention, A^0 is of the formula:



In some embodiments of the invention, R^x is of the formula:



In some embodiments of the invention, the compound is isolated and purified.

One embodiment of the invention provides a method of inhibiting tumor
5 growth comprising the step of contacting a sample or subject suspected of containing a tumor with a compound of the invention. In some embodiments of the invention, the tumor is *in vivo*.

One embodiment of the invention provides a method for the treatment or prevention of the symptoms or effects of cancer in an animal which comprises
10 administering to said animal a formulation comprising a therapeutically effective amount of a compound of the invention. In some embodiments of the invention, the compound is formulated with a pharmaceutically acceptable carrier. In some embodiments of the invention, the formulation further comprises a second active ingredient.

15 One embodiment of the invention provides a method of inhibiting the activity of a virus comprising the step of contacting a sample or subject suspected of containing a virus with a compound of the invention. In some embodiments of the invention, the virus is *in vivo*.

One embodiment of the invention provides a method for the treatment or
20 prevention of the symptoms or effects of viral infection in an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound of the invention. In some embodiments of the invention, the compound is formulated with a pharmaceutically acceptable carrier. In some embodiments of the invention, the
25 formulation further comprises a second active ingredient.

One embodiment of the invention provides a method of inhibiting inflammation comprising the step of contacting a sample or subject suspected of being inflamed with a compound of the invention. In some embodiments of the invention, the inflammation is *in vivo*.

One embodiment of the invention provides a method for the treatment or prevention of the symptoms or effects of inflammation in an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound of the invention. In some
5 embodiments of the invention, the compound is formulated with a pharmaceutically acceptable carrier. In some embodiments of the invention, the formulation further comprises a second active ingredient.

One embodiment of the invention provides a method for the treatment or prevention of the symptoms or effects of tissue or organ transplant rejection in
10 an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound of the invention. In some embodiments of the invention, the compound is formulated with a pharmaceutically acceptable carrier. In some embodiments of the invention, the formulation further comprises a second active ingredient.

15 One embodiment of the invention provides a pharmaceutical composition comprising a pharmaceutical carrier and a compound of the invention. In some embodiments of the invention, the composition further comprises at least one additional active ingredient.

One embodiment of the invention provides a unit dosage form
20 comprising a compound of the invention and a pharmaceutically acceptable carrier. In some embodiments of the invention, the unit dosage form further comprises at least one additional active ingredient.

One embodiment of the invention provides a use of a compound of the invention to prepare a medicament for treatment of cancer.

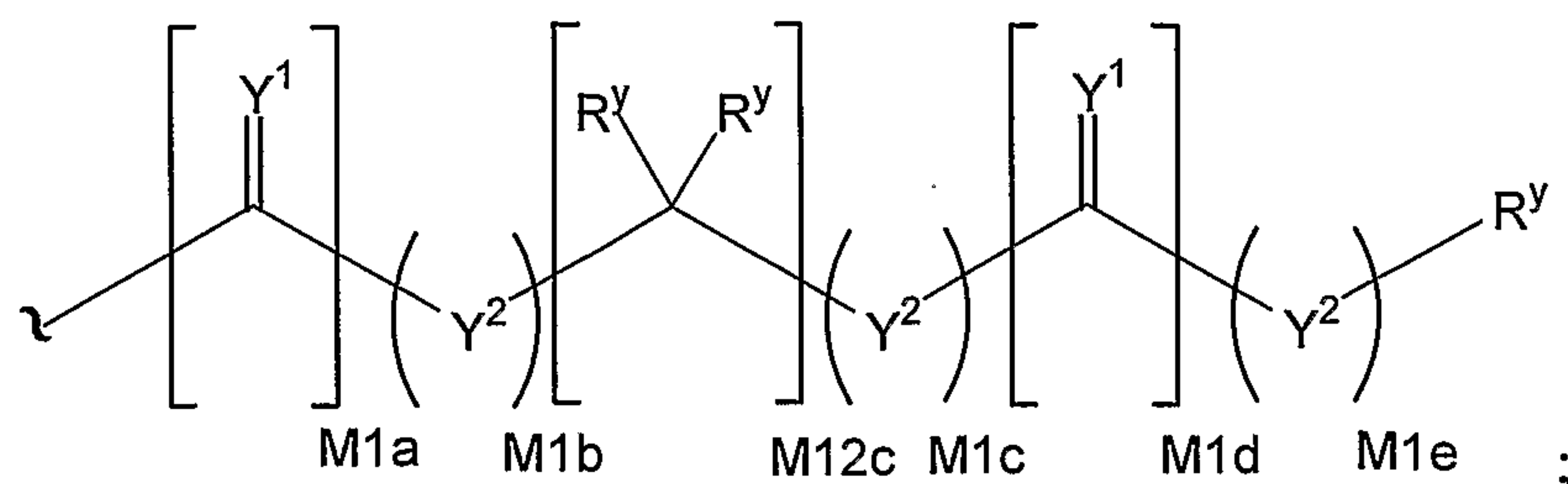
25 One embodiment of the invention provides a use of a compound of the invention to prepare a medicament for treatment of a virus infection.

One embodiment of the invention provides a use of a compound of the invention to prepare a medicament for treatment of inflammation.

One embodiment of the invention provides a use of a compound of the
30 invention to prepare a medicament for treatment of tissue or organ transplant rejection.

Certain embodiments of the invention also provide compounds described in the Examples herein.

In one specific embodiment of the invention R^x is a group of the formula:



wherein:

5 m1a, m1b, m1c, m1d and m1e are independently 0 or 1;

m12c is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

R^y is H, W^3 , R^2 or a protecting group;

provided that:

if m1a, m12c, and m1d are 0, then m1b, m1c and m1e are 0;

10 if m1a and m12c are 0 and m1d is not 0, then m1b and m1c are 0;

if m1a and m1d are 0 and m12c is not 0, then m1b and at least one of m1c and m1e are 0;

if m1a is 0 and m12c and m1d are not 0, then m1b is 0;

15 if m12c and m1d are 0 and m1a is not 0, then at least two of m1b, m1c and m1e are 0;

if m12c is 0 and m1a and m1d are not 0, then at least one of m1b and m1c are 0; and

if m1d is 0 and m1a and m12c are not 0, then at least one of m1c and m1e are 0.

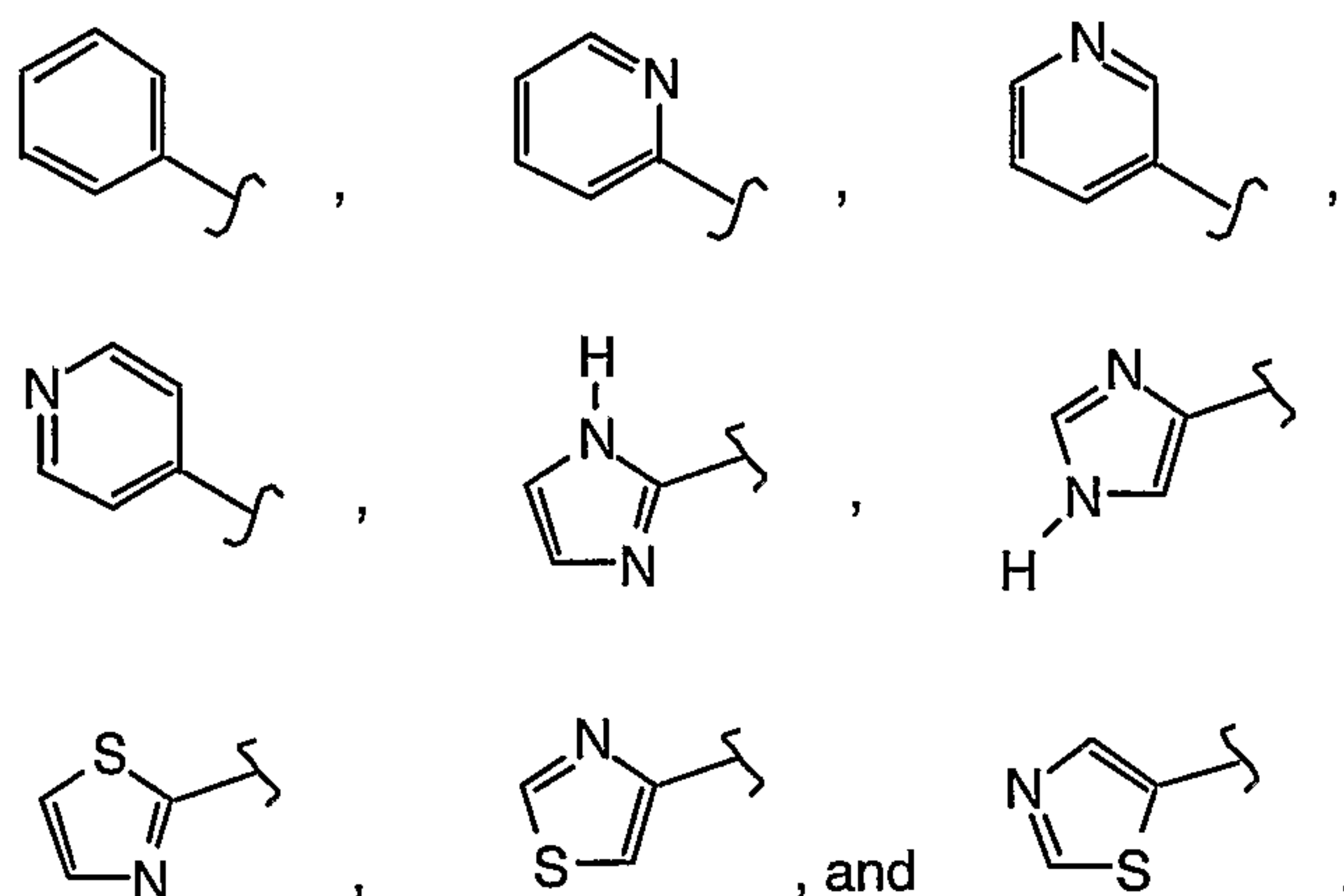
20 In another specific embodiment of the invention, R^{25} is phenyl optionally monosubstituted with halogen, hydroxy, carboxy, chlorocarbonyl, nitro, cyano, phenyl, alkyl, acyl, alkoxycarbonyl, acylamino, dialkylamino or dialkylaminoethoxycarbonyl, phenyl optionally disubstituted with hydroxy, carboxy, nitro or alkyl, or benzyl optionally substituted with dialkylamino.

25 In compounds of the invention W^5 carbocycles and W^5 heterocycles may be independently substituted with 0 to 3 R^2 groups. W^5 may be a saturated, unsaturated or aromatic ring comprising a mono- or bicyclic carbocycle or heterocycle. W^5 may have 3 to 10 ring atoms, *e.g.*, 3 to 7 ring atoms. The W^5 rings are saturated when containing 3 ring atoms, saturated or mono-unsaturated

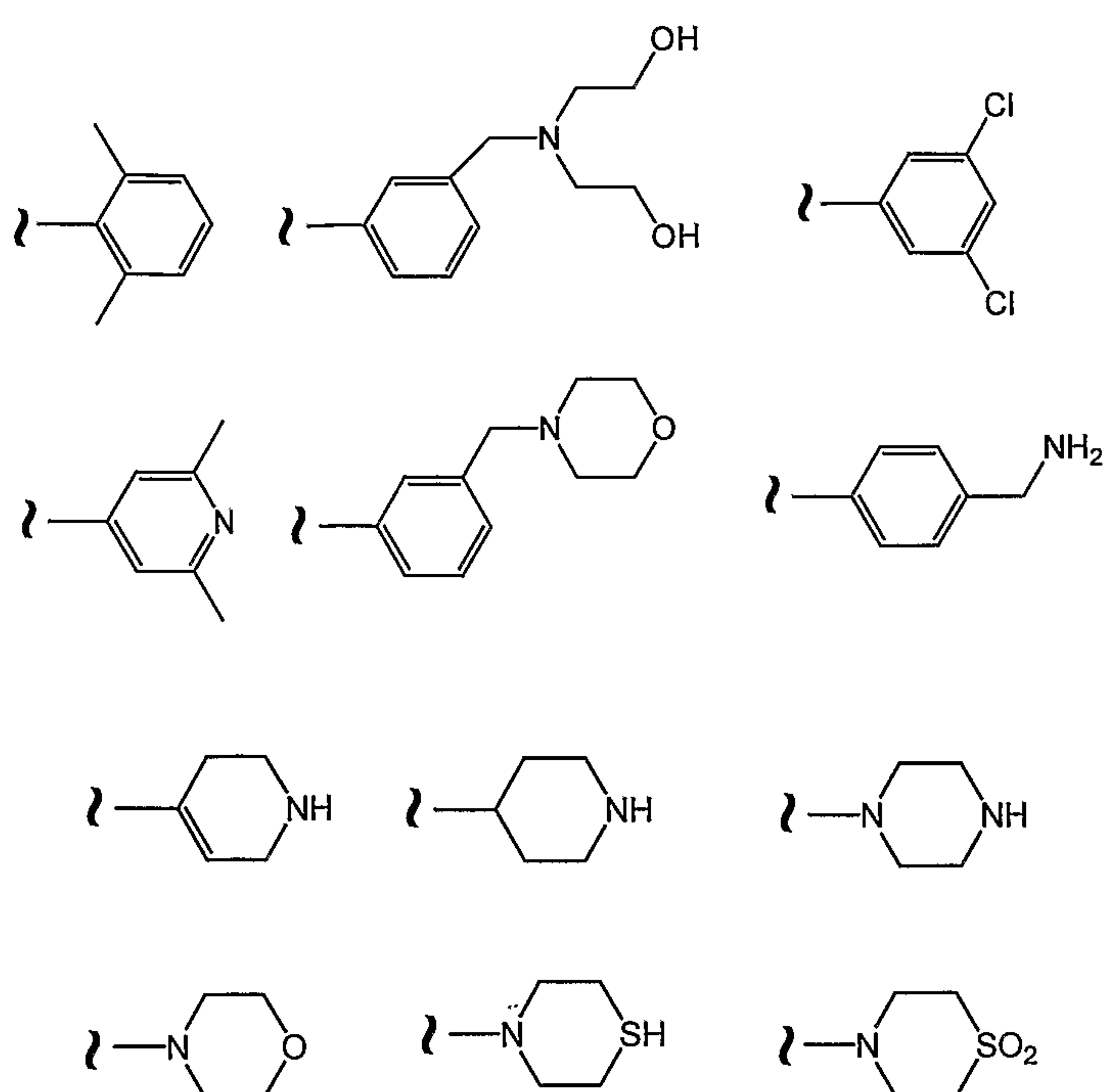
when containing 4 ring atoms, saturated, or mono- or di-unsaturated when containing 5 ring atoms, and saturated, mono- or di-unsaturated, or aromatic when containing 6 ring atoms.

A W^5 heterocycle may be a monocycle having 3 to 7 ring members (2 to 6 carbon atoms and 1 to 3 heteroatoms selected from N, O, P, and S) or a bicycle having 7 to 10 ring members (4 to 9 carbon atoms and 1 to 3 heteroatoms selected from N, O, P, and S). W^5 heterocyclic monocycles may have 3 to 6 ring atoms (2 to 5 carbon atoms and 1 to 2 heteroatoms selected from N, O, and S); or 5 or 6 ring atoms (3 to 5 carbon atoms and 1 to 2 heteroatoms selected from N and S). W^5 heterocyclic bicycles have 7 to 10 ring atoms (6 to 9 carbon atoms and 1 to 2 heteroatoms selected from N, O, and S) arranged as a bicyclo [4,5], [5,5], [5,6], or [6,6] system; or 9 to 10 ring atoms (8 to 9 carbon atoms and 1 to 2 hetero atoms selected from N and S) arranged as a bicyclo [5,6] or [6,6] system. The W^5 heterocycle may be bonded to Y^2 through a carbon, nitrogen, sulfur or other atom by a stable covalent bond.

W^5 heterocycles include for example, pyridyl, dihydropyridyl isomers, piperidine, pyridazinyl, pyrimidinyl, pyrazinyl, s-triazinyl, oxazolyl, imidazolyl, thiazolyl, isoxazolyl, pyrazolyl, isothiazolyl, furanyl, thiofuranyl, thienyl, and pyrrolyl. W^5 also includes, but is not limited to, examples such as:

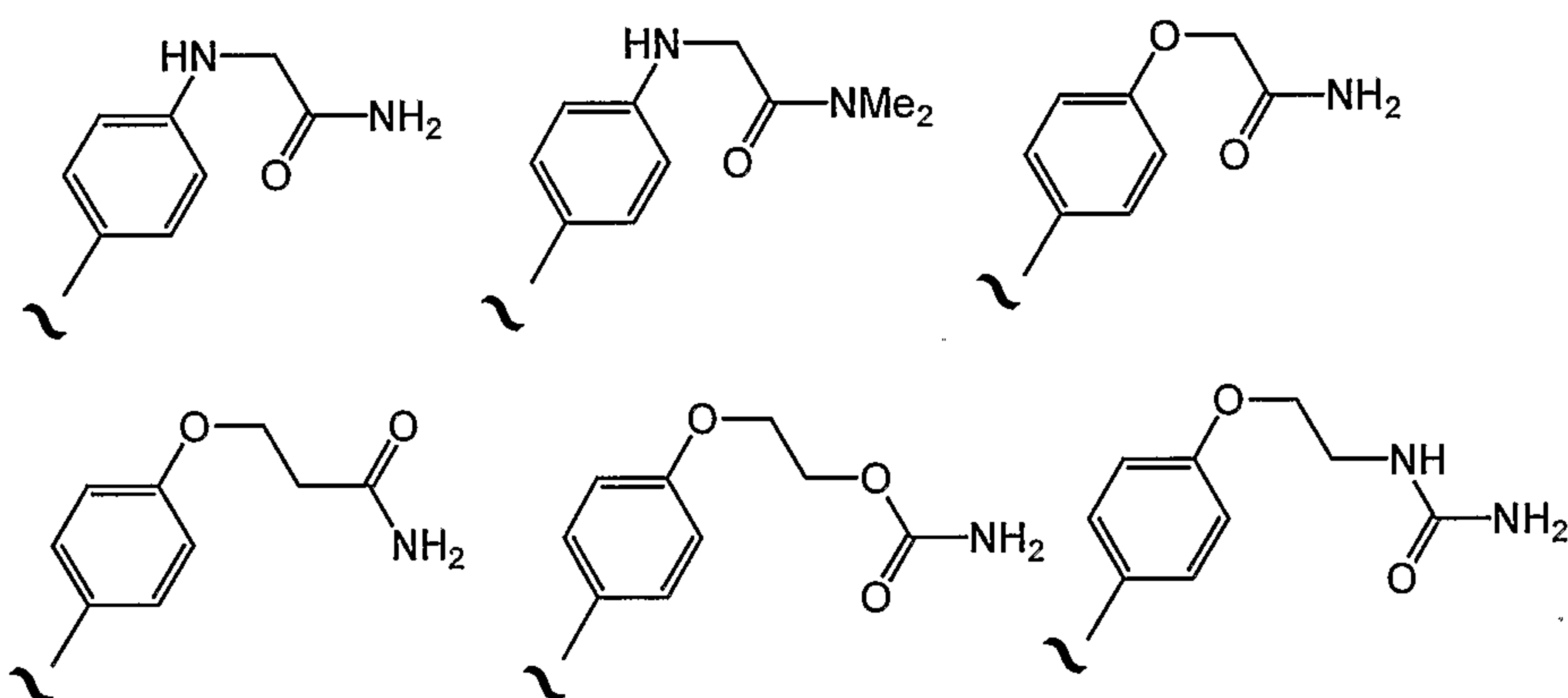


W^5 carbocycles and heterocycles may be independently substituted with 0 to 3 R^2 groups, as defined above. For example, substituted W^5 carbocycles include:



5

Examples of substituted phenyl carbocycles include:



Intracellular Targeting

The phosphonate group of the compounds of the invention may cleave *in vivo* in stages after they have reached the desired site of action, *i.e.* inside a cell.

- 10 One mechanism of action inside a cell may entail a first cleavage, *e.g.* by esterase, to provide a negatively-charged "locked-in" intermediate. Cleavage of a terminal ester grouping in a compound of the invention thus affords an unstable intermediate which releases a negatively charged "locked in" intermediate.

- 15 After passage inside a cell, intracellular enzymatic cleavage or modification of the phosphonate or prodrug compound may result in an

intracellular accumulation of the cleaved or modified compound by a “trapping” mechanism. The cleaved or modified compound may then be “locked-in” the cell by a significant change in charge, polarity, or other physical property change which decreases the rate at which the cleaved or modified compound can exit the cell, relative to the rate at which it entered as the phosphonate prodrug. Other mechanisms by which a therapeutic effect are achieved may be operative as well. Enzymes which are capable of an enzymatic activation mechanism with the phosphonate prodrug compounds of the invention include, but are not limited to, amidases, esterases, microbial enzymes, phospholipases, cholinesterases, and phosphatases.

From the foregoing, it will be apparent that many different drugs can be derivatized in accord with the present invention. Numerous such drugs are specifically mentioned herein. However, it should be understood that the discussion of drug families and their specific members for derivatization according to this invention is not intended to be exhaustive, but merely illustrative.

Specific Compounds of the Invention

Typically, compounds of the invention have a molecular weight of from about 400 amu to about 10,000 amu; in a specific embodiment of the invention, compounds have a molecular weight of less than about 5000 amu; in another specific embodiment of the invention, compounds have a molecular weight of less than about 2500 amu; in another specific embodiment of the invention, compounds have a molecular weight of less than about 1000 amu; in another specific embodiment of the invention, compounds have a molecular weight of less than about 800 amu; in another specific embodiment of the invention, compounds have a molecular weight of less than about 600 amu; and in another specific embodiment of the invention, compounds have a molecular weight of less than about 600 amu and a molecular weight of greater than about 400 amu.

The compounds of the invention also typically have a logD(polarity) less than about 5. In one embodiment the invention provides compounds having a logD less than about 4; in another one embodiment the invention provides compounds having a logD less than about 3; in another one embodiment the invention provides compounds having a logD greater than about -5; in another

one embodiment the invention provides compounds having a logD greater than about -3; and in another one embodiment the invention provides compounds having a logD greater than about 0 and less than about 3.

Selected substituents within the compounds of the invention are present
5 to a recursive degree. In this context, "recursive substituent" means that a substituent may recite another instance of itself. Because of the recursive nature of such substituents, theoretically, a large number may be present in any given claim. For example, R^x contains a R^y substituent. R^y can be R^2 , which in turn can be R^3 . If R^3 is selected to be R^{3c} , then a second instance of R^x can be
10 selected. One of ordinary skill in the art of medicinal chemistry understands that the total number of such substituents is reasonably limited by the desired properties of the compound intended. Such properties include, by of example and not limitation, physical properties such as molecular weight, solubility or log P, application properties such as activity against the intended target, and
15 practical properties such as ease of synthesis.

By way of example and not limitation, W^3 , R^y and R^3 are all recursive substituents in certain claims. Typically, each of these may independently occur 20, 19, 18, 17, 16, 15, 14, 13, 12, 11, 10, 9, 8, 7, 6, 5, 4, 3, 2, 1, or 0, times in a given claim. More typically, each of these may independently occur 12 or fewer
20 times in a given claim. More typically yet, W^3 will occur 0 to 8 times, R^y will occur 0 to 6 times and R^3 will occur 0 to 10 times in a given claim. Even more typically, W^3 will occur 0 to 6 times, R^y will occur 0 to 4 times and R^3 will occur 0 to 8 times in a given claim.

Recursive substituents are an intended aspect of the invention. One of
25 ordinary skill in the art of medicinal chemistry understands the versatility of such substituents. To the degree that recursive substituents are present in an claim of the invention, the total number will be determined as set forth above.

Whenever a compound described herein is substituted with more than one of the same designated group, *e.g.*, " R^1 " or " R^{6a} ", then it will be understood
30 that the groups may be the same or different, *i.e.*, each group is independently selected. Wavy lines indicate the site of covalent bond attachments to the adjoining groups, moieties, or atoms.

In one embodiment of the invention, the compound is in an isolated and purified form. Generally, the term "isolated and purified" means that the compound is substantially free from biological materials (*e.g.* blood, tissue, cells, *etc.*). In one specific embodiment of the invention, the term means that the compound or conjugate of the invention is at least about 50 wt.% free from biological materials; in another specific embodiment, the term means that the compound or conjugate of the invention is at least about 75 wt.% free from biological materials; in another specific embodiment, the term means that the compound or conjugate of the invention is at least about 90 wt.% free from biological materials; in another specific embodiment, the term means that the compound or conjugate of the invention is at least about 98 wt.% free from biological materials; and in another embodiment, the term means that the compound or conjugate of the invention is at least about 99 wt.% free from biological materials. In another specific embodiment, the invention provides a compound or conjugate of the invention that has been synthetically prepared (*e.g., ex vivo*).

In one specific embodiment of the invention the compound is not an anti-inflammatory compound; in another embodiment, the compound is not an anti-cancer agent; in another embodiment, the compound is not a PNP-inhibitor; in another embodiment, the compound is not a compound that is active against immune-mediated conditions; in another embodiment, the compound is not a compound that is active against metabolic diseases; in another embodiment, the compound is not an antiviral agent; in another embodiment, the compound is not a nucleoside; in another embodiment, the compound is not a kinase inhibitor; in another embodiment, the compound is not an antimetabolite; in another embodiment, the compound is not an IMPDH inhibitor; and in another embodiment, the compound is not an anti-infective agent.

Cellular Accumulation

In one embodiment, the invention provides compounds capable of accumulating in human PBMC (peripheral blood mononuclear cells). PBMC refer to blood cells having round lymphocytes and monocytes. Physiologically, PBMC are critical components of the mechanism against infection. PBMC may be isolated from heparinized whole blood of normal healthy donors or buffy

coats, by standard density gradient centrifugation and harvested from the interface, washed (*e.g.* phosphate-buffered saline) and stored in freezing medium. PBMC may be cultured in multi-well plates. At various times of culture, supernatant may be either removed for assessment, or cells may be
5 harvested and analyzed (Smith R. et al (2003) *Blood* 102(7):2532-2540). The compounds of this claim may further comprise a phosphonate or phosphonate prodrug. More typically, the phosphonate or phosphonate prodrug can have the structure A³ as described herein.

Typically, compounds of the invention demonstrate improved
10 intracellular half-life of the compounds or intracellular metabolites of the compounds in human PBMC when compared to analogs of the compounds not having the phosphonate or phosphonate prodrug. Typically, the half-life is improved by at least about 50%, more typically at least in the range 50-100%, still more typically at least about 100%, more typically yet greater than about
15 100%.

In one embodiment of the invention the intracellular half-life of a metabolite of the compound in human PBMCs is improved when compared to an analog of the compound not having the phosphonate or phosphonate prodrug. In such claims, the metabolite may be generated intracellularly, *e.g.* generated
20 within human PBMC. The metabolite may be a product of the cleavage of a phosphonate prodrug within human PBMCs. The phosphonate prodrug may be cleaved to form a metabolite having at least one negative charge at physiological pH. The phosphonate prodrug may be enzymatically cleaved within human PBMC to form a phosphonate having at least one active hydrogen atom of the
25 form P-OH.

Stereoisomers

The compounds of the invention may have chiral centers, *e.g.*, chiral carbon or phosphorus atoms. The compounds of the invention thus include racemic mixtures of all stereoisomers, including enantiomers, diastereomers, and
30 atropisomers. In addition, the compounds of the invention include enriched or resolved optical isomers at any or all asymmetric, chiral atoms. In other words, the chiral centers apparent from the depictions are provided as the chiral isomers or racemic mixtures. Both racemic and diastereomeric mixtures, as well as the

individual optical isomers isolated or synthesized, substantially free of their enantiomeric or diastereomeric partners, are all within the scope of the invention. The racemic mixtures are separated into their individual, substantially optically pure isomers through well-known techniques such as, for example, the
5 separation of diastereomeric salts formed with optically active adjuncts, *e.g.*, acids or bases followed by conversion back to the optically active substances. In most instances, the desired optical isomer is synthesized by means of stereospecific reactions, beginning with the appropriate stereoisomer of the desired starting material.

10 The compounds of the invention can also exist as tautomeric isomers in certain cases. All though only one delocalized resonance structure may be depicted, all such forms are contemplated within the scope of the invention. For example, ene-amine tautomers can exist for purine, pyrimidine, imidazole, guanidine, amidine, and tetrazole systems and all their possible tautomeric forms
15 are within the scope of the invention.

Salts and Hydrates

The compositions of this invention optionally comprise salts of the compounds herein, especially pharmaceutically acceptable non-toxic salts containing, for example, Na^+ , Li^+ , K^+ , Ca^{+2} and Mg^{+2} . Such salts may include
20 those derived by combination of appropriate cations such as alkali and alkaline earth metal ions or ammonium and quaternary amino ions with an acid anion moiety, typically a carboxylic acid. Monovalent salts are preferred if a water soluble salt is desired.

Metal salts typically are prepared by reacting the metal hydroxide with a
25 compound of this invention. Examples of metal salts which are prepared in this way are salts containing Li^+ , Na^+ , and K^+ . A less soluble metal salt can be precipitated from the solution of a more soluble salt by addition of the suitable metal compound.

In addition, salts may be formed from acid addition of certain organic
30 and inorganic acids, *e.g.*, HCl , HBr , H_2SO_4 , H_3PO_4 or organic sulfonic acids, to basic centers, typically amines, or to acidic groups. Finally, it is to be understood that the compositions herein comprise compounds of the invention in

their un-ionized, as well as zwitterionic form, and combinations with stoichiometric amounts of water as in hydrates.

Also included within the scope of this invention are the salts of the parental compounds with one or more amino acids. Any of the amino acids
5 described above are suitable, especially the naturally-occurring amino acids found as protein components, although the amino acid typically is one bearing a side chain with a basic or acidic group, *e.g.*, lysine, arginine or glutamic acid, or a neutral group such as glycine, serine, threonine, alanine, isoleucine, or leucine.

Another aspect of the invention relates to methods of inhibiting tumor
10 growth, viral infection, inflammation and tissue/organ transplant rejection comprising the step of treating a sample or subject suspected of needing such inhibition with a composition of the invention.

Compositions of the invention may act as inhibitors of tumor growth, viral infection, inflammation and tissue/organ transplant rejection, or as
15 intermediates for such inhibitors or have other utilities as described below. The inhibitors will bind to locations on the surface or in a cavity of a cell having a geometry unique to Mycophenolate like compounds. Compositions binding a cell may bind with varying degrees of reversibility. Those compounds binding substantially irreversibly are ideal candidates for use in this method of the
20 invention. Once labeled, the substantially irreversibly binding compositions are useful as probes for the detection of cancer, viruses, inflammation or tissue/organ transplant rejection. Accordingly, the invention relates to methods of detecting cancer, viruses, inflammation or tissue/organ transplant rejection in a sample or subject suspected of containing a tumor, containing a virus, being
25 inflamed or rejecting a tissue/organ transplant, comprising the steps of: treating such a sample or subject with a composition comprising a compound of the invention bound to a label; and observing the effect of the sample on the activity of the label. Suitable labels are well known in the diagnostics field and include stable free radicals, fluorophores, radioisotopes, enzymes, chemiluminescent
30 groups and chromogens. The compounds herein are labeled in conventional fashion using functional groups such as hydroxyl or amino.

Within the context of the invention samples suspected of containing a tumor, containing a virus, being inflamed or rejecting a tissue/organ transplant

include natural or man-made materials such as living organisms; tissue or cell cultures; biological samples such as biological material samples (blood, serum, urine, cerebrospinal fluid, tears, sputum, saliva, tissue samples, and the like); laboratory samples; food, water, or air samples; bioproduct samples such as
5 extracts of cells, particularly recombinant cells synthesizing a desired glycoprotein; and the like. Typically the sample will be suspected of containing an organism that induces cancer cell growth, frequently a pathogenic organism such as a tumor virus. Samples can be contained in any medium including water and organic solvent\water mixtures. Samples include living organisms such as
10 humans, and man made materials such as cell cultures.

The treating step of the invention comprises adding the composition of the invention to the sample or it comprises adding a precursor of the composition to the sample. The addition step comprises any method of administration as described above.

15 If desired, the anti-cancer, anti-virus, anti-inflammation, and/or anti-tissue/organ transplant rejection activity of a Mycophenolate-like compound after application of the composition can be observed by any method including direct and indirect methods of detecting such activity. Quantitative, qualitative, and semiquantitative methods of determining such activity are all contemplated.
20 Typically one of the screening methods described above are applied, however, any other method such as observation of the physiological properties of a living organism are also applicable.

However, in screening compounds capable of inhibiting some viruses it should be kept in mind that the results of enzyme assays may not correlate with
25 cell culture assays. Thus, a cell based assay should be the primary screening tool.

Screens for Anti-Cancer, Anti-Viral, Anti-Inflammatory, and Anti-Tissue/Organ Transplant Rejection Compounds

Compositions of the invention are screened for anti-cancer, anti-viral,
30 anti-inflammatory and anti-tissue/organ transplant rejection activity by any of the conventional techniques for evaluating enzyme activity. Within the context of the invention, typically compositions are first screened for inhibitory activity *in vitro* and compositions showing inhibitory activity are then screened for

activity *in vivo*. Compositions having *in vitro* K_i (inhibitory constants) of less than about 5×10^{-6} M, typically less than about 1×10^{-7} M and preferably less than about 5×10^{-8} M are preferred for *in vivo* use.

Useful *in vitro* screens have been described in detail and will not be elaborated here. However, the examples describe suitable *in vitro* assays.

Pharmaceutical Formulations

The compounds of this invention are formulated with conventional carriers and excipients, which will be selected in accord with ordinary practice. Tablets will contain excipients, glidants, fillers, binders and the like. Aqueous formulations are prepared in sterile form, and when intended for delivery by other than oral administration generally will be isotonic. All formulations will optionally contain excipients such as those set forth in the Handbook of Pharmaceutical Excipients (1986). Excipients include ascorbic acid and other antioxidants, chelating agents such as EDTA, carbohydrates such as dextrin, hydroxyalkylcellulose, hydroxyalkylmethylcellulose, stearic acid and the like. The pH of the formulations ranges from about 3 to about 11, but is ordinarily about 7 to 10.

While it is possible for the active ingredients to be administered alone it may be preferable to present them as pharmaceutical formulations. The formulations, both for veterinary and for human use, of the invention comprise at least one active ingredient, as above defined, together with one or more acceptable carriers therefor and optionally other therapeutic ingredients. The carrier(s) must be "acceptable" in the sense of being compatible with the other ingredients of the formulation and physiologically innocuous to the recipient thereof.

The formulations include those suitable for the foregoing administration routes. The formulations may conveniently be presented in unit dosage form and may be prepared by any of the methods well known in the art of pharmacy. Techniques and formulations generally are found in Remington's Pharmaceutical Sciences (Mack Publishing Co., Easton, PA). Such methods include the step of bringing into association the active ingredient with the carrier which constitutes one or more accessory ingredients. In general the formulations are prepared by

uniformly and intimately bringing into association the active ingredient with liquid carriers or finely divided solid carriers or both, and then, if necessary, shaping the product.

Formulations of the present invention suitable for oral administration
5 may be presented as discrete units such as capsules, cachets or tablets each containing a predetermined amount of the active ingredient; as a powder or granules; as a solution or a suspension in an aqueous or non-aqueous liquid; or as an oil-in-water liquid emulsion or a water-in-oil liquid emulsion. The active ingredient may also be administered as a bolus, electuary or paste.

10 A tablet is made by compression or molding, optionally with one or more accessory ingredients. Compressed tablets may be prepared by compressing in a suitable machine the active ingredient in a free-flowing form such as a powder or granules, optionally mixed with a binder, lubricant, inert diluent, preservative, surface active or dispersing agent. Molded tablets may be made by molding in a
15 suitable machine a mixture of the powdered active ingredient moistened with an inert liquid diluent. The tablets may optionally be coated or scored and optionally are formulated so as to provide slow or controlled release of the active ingredient therefrom.

For administration to the eye or other external tissues *e.g.*, mouth and
20 skin, the formulations are preferably applied as a topical ointment or cream containing the active ingredient(s) in an amount of, for example, 0.075 to 20% w/w (including active ingredient(s) in a range between 0.1% and 20% in increments of 0.1% w/w such as 0.6% w/w, 0.7% w/w, *etc.*), preferably 0.2 to 15% w/w and most preferably 0.5 to 10% w/w. When formulated in an
25 ointment, the active ingredients may be employed with either a paraffinic or a water-miscible ointment base. Alternatively, the active ingredients may be formulated in a cream with an oil-in-water cream base.

If desired, the aqueous phase of the cream base may include, for example, at least 30% w/w of a polyhydric alcohol, *i.e.* an alcohol having two or
30 more hydroxyl groups such as propylene glycol, butane 1,3-diol, mannitol, sorbitol, glycerol and polyethylene glycol (including PEG 400) and mixtures thereof. The topical formulations may desirably include a compound which enhances absorption or penetration of the active ingredient through the skin or

other affected areas. Examples of such dermal penetration enhancers include dimethyl sulphoxide and related analogs.

The oily phase of the emulsions of this invention may be constituted from known ingredients in a known manner. While the phase may comprise
5 merely an emulsifier (otherwise known as an emulgent), it desirably comprises a mixture of at least one emulsifier with a fat or an oil or with both a fat and an oil. Preferably, a hydrophilic emulsifier is included together with a lipophilic emulsifier which acts as a stabilizer. It is also preferred to include both an oil and a fat. Together, the emulsifier(s) with or without stabilizer(s) make up the
10 so-called emulsifying wax, and the wax together with the oil and fat make up the so-called emulsifying ointment base which forms the oily dispersed phase of the cream formulations.

Emulgents and emulsion stabilizers suitable for use in the formulation of the invention include Tween[®] 60, Span[®] 80, cetostearyl alcohol, benzyl
15 alcohol, myristyl alcohol, glyceryl mono-stearate and sodium lauryl sulfate.

The choice of suitable oils or fats for the formulation is based on achieving the desired cosmetic properties. The cream should preferably be a non-greasy, non-staining and washable product with suitable consistency to avoid leakage from tubes or other containers. Straight or branched chain, mono-
20 or dibasic alkyl esters such as di-isoadipate, isocetyl stearate, propylene glycol diester of coconut fatty acids, isopropyl myristate, decyl oleate, isopropyl palmitate, butyl stearate, 2-ethylhexyl palmitate or a blend of branched chain esters known as Crodamol CAP may be used, the last three being preferred esters. These may be used alone or in combination depending on the properties
25 required. Alternatively, high melting point lipids such as white soft paraffin and/or liquid paraffin or other mineral oils are used.

Pharmaceutical formulations according to the present invention comprise one or more compounds of the invention together with one or more pharmaceutically acceptable carriers or excipients and optionally other
30 therapeutic agents. Pharmaceutical formulations containing the active ingredient may be in any form suitable for the intended method of administration. When used for oral use for example, tablets, troches, lozenges, aqueous or oil

suspensions, dispersible powders or granules, emulsions, hard or soft capsules, syrups or elixirs may be prepared. Compositions intended for oral use may be prepared according to any method known to the art for the manufacture of pharmaceutical compositions and such compositions may contain one or more
5 agents including sweetening agents, flavoring agents, coloring agents and preserving agents, in order to provide a palatable preparation. Tablets containing the active ingredient in admixture with non-toxic pharmaceutically acceptable excipient which are suitable for manufacture of tablets are acceptable. These excipients may be, for example, inert diluents, such as calcium or sodium
10 carbonate, lactose, lactose monohydrate, croscarmellose sodium, povidone, calcium or sodium phosphate; granulating and disintegrating agents, such as maize starch, or alginic acid; binding agents, such as cellulose, microcrystalline cellulose, starch, gelatin or acacia; and lubricating agents, such as magnesium stearate, stearic acid or talc. Tablets may be uncoated or may be coated by
15 known techniques including microencapsulation to delay disintegration and adsorption in the gastrointestinal tract and thereby provide a sustained action over a longer period. For example, a time delay material such as glyceryl monostearate or glyceryl distearate alone or with a wax may be employed.

Formulations for oral use may be also presented as hard gelatin capsules
20 where the active ingredient is mixed with an inert solid diluent, for example calcium phosphate or kaolin, or as soft gelatin capsules wherein the active ingredient is mixed with water or an oil medium, such as peanut oil, liquid paraffin or olive oil.

Aqueous suspensions of the invention contain the active materials in
25 admixture with excipients suitable for the manufacture of aqueous suspensions. Such excipients include a suspending agent, such as sodium carboxymethylcellulose, methylcellulose, hydroxypropyl methylcellulose, sodium alginate, polyvinylpyrrolidone, gum tragacanth and gum acacia, and dispersing or wetting agents such as a naturally occurring phosphatide (*e.g.*,
30 lecithin), a condensation product of an alkylene oxide with a fatty acid (*e.g.*, polyoxyethylene stearate), a condensation product of ethylene oxide with a long chain aliphatic alcohol (*e.g.*, heptadecaethyleneoxycetanol), a condensation product of ethylene oxide with a partial ester derived from a fatty acid and a

hexitol anhydride (*e.g.*, polyoxyethylene sorbitan monooleate). The aqueous suspension may also contain one or more preservatives such as ethyl or n-propyl p-hydroxy-benzoate, one or more coloring agents, one or more flavoring agents and one or more sweetening agents, such as sucrose or saccharin.

5 Oil suspensions may be formulated by suspending the active ingredient in a vegetable oil, such as arachis oil, olive oil, sesame oil or coconut oil, or in a mineral oil such as liquid paraffin. The oral suspensions may contain a thickening agent, such as beeswax, hard paraffin or cetyl alcohol. Sweetening agents, such as those set forth above, and flavoring agents may be added to
10 provide a palatable oral preparation. These compositions may be preserved by the addition of an antioxidant such as ascorbic acid.

Dispersible powders and granules of the invention suitable for preparation of an aqueous suspension by the addition of water provide the active ingredient in admixture with a dispersing or wetting agent, a suspending agent,
15 and one or more preservatives. Suitable dispersing or wetting agents and suspending agents are exemplified by those disclosed above. Additional excipients, for example sweetening, flavoring and coloring agents, may also be present.

The pharmaceutical compositions of the invention may also be in the
20 form of oil-in-water emulsions. The oily phase may be a vegetable oil, such as olive oil or arachis oil, a mineral oil, such as liquid paraffin, or a mixture of these. Suitable emulsifying agents include naturally-occurring gums, such as gum acacia and gum tragacanth, naturally occurring phosphatides, such as soybean lecithin, esters or partial esters derived from fatty acids and hexitol
25 anhydrides, such as sorbitan monooleate, and condensation products of these partial esters with ethylene oxide, such as polyoxyethylene sorbitan monooleate. The emulsion may also contain sweetening and flavoring agents. Syrups and elixirs may be formulated with sweetening agents, such as glycerol, sorbitol or sucrose. Such formulations may also contain a demulcent, a preservative, a
30 flavoring or a coloring agent.

The pharmaceutical compositions of the invention may be in the form of a sterile injectable preparation, such as a sterile injectable aqueous or oleaginous suspension. This suspension may be formulated according to the known art using

those suitable dispersing or wetting agents and suspending agents which have been mentioned above. The sterile injectable preparation may also be a sterile injectable solution or suspension in a non-toxic parenterally acceptable diluent or solvent, such as a solution in 1,3-butane-diol or prepared as a lyophilized powder. Among the acceptable vehicles and solvents that may be employed are water, Ringer's solution and isotonic sodium chloride solution. In addition, sterile fixed oils may conventionally be employed as a solvent or suspending medium. For this purpose any bland fixed oil may be employed including synthetic mono- or diglycerides. In addition, fatty acids such as oleic acid may likewise be used in the preparation of injectables.

The amount of active ingredient that may be combined with the carrier material to produce a single dosage form will vary depending upon the host treated and the particular mode of administration. For example, a time-release formulation intended for oral administration to humans may contain approximately 1 to 1000 mg of active material compounded with an appropriate and convenient amount of carrier material which may vary from about 5 to about 95% of the total compositions (weight:weight). The pharmaceutical composition can be prepared to provide easily measurable amounts for administration. For example, an aqueous solution intended for intravenous infusion may contain from about 3 to 500 μg of the active ingredient per milliliter of solution in order that infusion of a suitable volume at a rate of about 30 mL/hr can occur.

Formulations suitable for administration to the eye include eye drops wherein the active ingredient is dissolved or suspended in a suitable carrier, especially an aqueous solvent for the active ingredient. The active ingredient is preferably present in such formulations in a concentration of 0.5 to 20%, advantageously 0.5 to 10% particularly about 1.5% w/w.

Formulations suitable for topical administration in the mouth include lozenges comprising the active ingredient in a flavored basis, usually sucrose and acacia or tragacanth; pastilles comprising the active ingredient in an inert basis such as gelatin and glycerin, or sucrose and acacia; and mouthwashes comprising the active ingredient in a suitable liquid carrier.

Formulations for rectal administration may be presented as a suppository with a suitable base comprising for example cocoa butter or a salicylate.

Formulations suitable for intrapulmonary or nasal administration have a particle size for example in the range of 0.1 to 500 microns (including particle sizes in a range between 0.1 and 500 microns in increments microns such as 0.5, 1, 30 microns, 35 microns, *etc.*), which is administered by rapid inhalation through the nasal passage or by inhalation through the mouth so as to reach the alveolar sacs. Suitable formulations include aqueous or oily solutions of the active ingredient. Formulations suitable for aerosol or dry powder administration may be prepared according to conventional methods and may be delivered with other therapeutic agents such as compounds heretofore used in the treatment or prophylaxis of a given condition.

Formulations suitable for vaginal administration may be presented as pessaries, tampons, creams, gels, pastes, foams or spray formulations containing in addition to the active ingredient such carriers as are known in the art to be appropriate.

Formulations suitable for parenteral administration include aqueous and non-aqueous sterile injection solutions which may contain anti-oxidants, buffers, bacteriostats and solutes which render the formulation isotonic with the blood of the intended recipient; and aqueous and non-aqueous sterile suspensions which may include suspending agents and thickening agents.

The formulations are presented in unit-dose or multi-dose containers, for example sealed ampoules and vials, and may be stored in a freeze-dried (lyophilized) condition requiring only the addition of the sterile liquid carrier, for example water for injection, immediately prior to use. Extemporaneous injection solutions and suspensions are prepared from sterile powders, granules and tablets of the kind previously described. Preferred unit dosage formulations are those containing a daily dose or unit daily sub-dose, as herein above recited, or an appropriate fraction thereof, of the active ingredient.

It should be understood that in addition to the ingredients particularly mentioned above the formulations of this invention may include other agents conventional in the art having regard to the type of formulation in question, for example those suitable for oral administration may include flavoring agents.

The invention further provides veterinary compositions comprising at least one active ingredient as above defined together with a veterinary carrier therefor.

Veterinary carriers are materials useful for the purpose of administering
5 the composition and may be solid, liquid or gaseous materials which are otherwise inert or acceptable in the veterinary art and are compatible with the active ingredient. These veterinary compositions may be administered orally, parenterally or by any other desired route.

Compounds of the invention can also be formulated to provide controlled
10 release of the active ingredient to allow less frequent dosing or to improve the pharmacokinetic or toxicity profile of the active ingredient. Accordingly, the invention also provided compositions comprising one or more compounds of the invention formulated for sustained or controlled release.

Effective dose of active ingredient depends at least on the nature of the
15 condition being treated, toxicity, whether the compound is being used prophylactically (lower doses), the method of delivery, and the pharmaceutical formulation, and will be determined by the clinician using conventional dose escalation studies. It can be expected to be from about 0.0001 to about 100 mg/kg body weight per day. Typically, from about 0.01 to about 10 mg/kg body
20 weight per day. More typically, from about .01 to about 5 mg/kg body weight per day. More typically, from about .05 to about 0.5 mg/kg body weight per day. For example, the daily candidate dose for an adult human of approximately 70 kg body weight will range from 1 mg to 1000 mg, preferably between 5 mg and 500 mg, and may take the form of single or multiple doses.

25 Routes of Administration

One or more compounds of the invention (herein referred to as the active ingredients) are administered by any route appropriate to the condition to be treated. Suitable routes include oral, rectal, nasal, topical (including buccal and sublingual), vaginal and parenteral (including subcutaneous, intramuscular,
30 intravenous, intradermal, intrathecal and epidural), and the like. It will be appreciated that the preferred route may vary with for example the condition of the recipient. An advantage of the compounds of this invention is that they are orally bioavailable and can be dosed orally.

Combination Therapy

Active ingredients of the invention are also used in combination with other active ingredients. Such combinations are selected based on the condition to be treated, cross-reactivities of ingredients and pharmaco-properties of the combination.

It is also possible to combine any compound of the invention with one or more other active ingredients in a unitary dosage form for simultaneous or sequential administration to a patient. The combination therapy may be administered as a simultaneous or sequential regimen. When administered sequentially, the combination may be administered in two or more administrations.

The combination therapy may provide “synergy” and “synergistic effect”, *i.e.* the effect achieved when the active ingredients used together is greater than the sum of the effects that results from using the compounds separately. A synergistic effect may be attained when the active ingredients are: (1) co-formulated and administered or delivered simultaneously in a combined formulation; (2) delivered by alternation or in parallel as separate formulations; or (3) by some other regimen. When delivered in alternation therapy, a synergistic effect may be attained when the compounds are administered or delivered sequentially, *e.g.*, in separate tablets, pills or capsules, or by different injections in separate syringes. In general, during alternation therapy, an effective dosage of each active ingredient is administered sequentially, *i.e.* serially, whereas in combination therapy, effective dosages of two or more active ingredients are administered together.

Metabolites of the Compounds of the Invention

Also falling within the scope of this invention are the *in vivo* metabolic products of the compounds described herein. Such products may result for example from the oxidation, reduction, hydrolysis, amidation, esterification and the like of the administered compound, primarily due to enzymatic processes. Accordingly, the invention includes compounds produced by a process comprising contacting a compound of this invention with a mammal for a period of time sufficient to yield a metabolic product thereof. Such products typically

are identified by preparing a radiolabelled (*e.g.*, C¹⁴ or H³) compound of the invention, administering it parenterally in a detectable dose (*e.g.*, greater than about 0.5 mg/kg) to an animal such as rat, mouse, guinea pig, monkey, or to man, allowing sufficient time for metabolism to occur (typically about 30
5 seconds to 30 hours) and isolating its conversion products from the urine, blood or other biological samples. These products are easily isolated since they are labeled (others are isolated by the use of antibodies capable of binding epitopes surviving in the metabolite). The metabolite structures are determined in conventional fashion, *e.g.*, by MS or NMR analysis. In general, analysis of
10 metabolites is done in the same way as conventional drug metabolism studies well-known to those skilled in the art. The conversion products, so long as they are not otherwise found *in vivo*, are useful in diagnostic assays for therapeutic dosing of the compounds of the invention even if they possess no therapeutic activity of their own.

15 Recipes and methods for determining stability of compounds in surrogate gastrointestinal secretions are known. Compounds are defined herein as stable in the gastrointestinal tract where less than about 50 mole percent of the protected groups are deprotected in surrogate intestinal or gastric juice upon incubation for 1 hour at 37 °C. Simply because the compounds are stable to the
20 gastrointestinal tract does not mean that they cannot be hydrolyzed *in vivo*. The phosphonate prodrugs of the invention typically will be stable in the digestive system but are substantially hydrolyzed to the parental drug in the digestive lumen, liver or other metabolic organ, or within cells in general.

Exemplary Methods of Making the Compounds of the Invention.

25 The invention also relates to methods of making the compositions of the invention. The compositions are prepared by any of the applicable techniques of organic synthesis. Many such techniques are well known in the art. However, many of the known techniques are elaborated in Compendium of Organic Synthetic Methods (John Wiley & Sons, New York), Vol. 1, Ian T. Harrison and
30 Shuyen Harrison, 1971; Vol. 2, Ian T. Harrison and Shuyen Harrison, 1974; Vol. 3, Louis S. Hegedus and Leroy Wade, 1977; Vol. 4, Leroy G. Wade, jr., 1980; Vol. 5, Leroy G. Wade, Jr., 1984; and Vol. 6, Michael B. Smith; as well as

March, J., Advanced Organic Chemistry, Third Edition, (John Wiley & Sons, New York, 1985), Comprehensive Organic Synthesis. Selectivity, Strategy & Efficiency in Modern Organic Chemistry. In 9 Volumes, Barry M. Trost, Editor-in-Chief (Pergamon Press, New York, 1993 printing).

5 A number of exemplary methods for the preparation of the compositions of the invention are provided below. These methods are intended to illustrate the nature of such preparations are not intended to limit the scope of applicable methods.

Schemes and Examples

10 General aspects of these exemplary methods are described below and in the Examples. Each of the products of the following processes is optionally separated, isolated, and/or purified prior to its use in subsequent processes.

 Generally, the reaction conditions such as temperature, reaction time, solvents, work-up procedures, and the like, will be those common in the art for
15 the particular reaction to be performed. The cited reference material, together with material cited therein, contains detailed descriptions of such conditions. Typically the temperatures will be -100 °C to 200 °C, solvents will be aprotic or protic, and reaction times will be 10 seconds to 10 days. Work-up typically consists of quenching any unreacted reagents followed by partition between a
20 water/organic layer system (extraction) and separating the layer containing the product.

 Oxidation and reduction reactions are typically carried out at temperatures near room temperature (about 20 °C), although for metal hydride reductions frequently the temperature is reduced to 0 °C to -100 °C, solvents are
25 typically aprotic for reductions and may be either protic or aprotic for oxidations. Reaction times are adjusted to achieve desired conversions.

 Condensation reactions are typically carried out at temperatures near room temperature, although for non-equilibrating, kinetically controlled condensations reduced temperatures (0 °C to -100 °C) are also common.
30 Solvents can be either protic (common in equilibrating reactions) or aprotic (common in kinetically controlled reactions).

Standard synthetic techniques such as azeotropic removal of reaction by-products and use of anhydrous reaction conditions (*e.g.*, inert gas environments) are common in the art and will be applied when applicable.

The terms “treated”, “treating”, “treatment”, and the like, when used in connection with a chemical synthetic operation, mean contacting, mixing, reacting, allowing to react, bringing into contact, and other terms common in the art for indicating that one or more chemical entities is treated in such a manner as to convert it to one or more other chemical entities. This means that “treating compound one with compound two” is synonymous with “allowing compound one to react with compound two”, “contacting compound one with compound two”, “reacting compound one with compound two”, and other expressions common in the art of organic synthesis for reasonably indicating that compound one was “treated”, “reacted”, “allowed to react”, *etc.*, with compound two. For example, treating indicates the reasonable and usual manner in which organic chemicals are allowed to react. Normal concentrations (0.01M to 10M, typically 0.1M to 1M), temperatures (-100 °C to 250 °C, typically -78 °C to 150 °C, more typically -78 °C to 100 °C, still more typically 0 °C to 100 °C), reaction vessels (typically glass, plastic, metal), solvents, pressures, atmospheres (typically air for oxygen and water insensitive reactions or nitrogen or argon for oxygen or water sensitive), *etc.*, are intended unless otherwise indicated. The knowledge of similar reactions known in the art of organic synthesis are used in selecting the conditions and apparatus for “treating” in a given process. In particular, one of ordinary skill in the art of organic synthesis selects conditions and apparatus reasonably expected to successfully carry out the chemical reactions of the described processes based on the knowledge in the art.

Modifications of each of the exemplary schemes and in the examples (hereafter “exemplary schemes”) leads to various analogs of the specific exemplary materials produce. The above-cited citations describing suitable methods of organic synthesis are applicable to such modifications.

In each of the exemplary schemes it may be advantageous to separate reaction products from one another and/or from starting materials. The desired products of each step or series of steps is separated and/or purified (hereinafter separated) to the desired degree of homogeneity by the techniques common in

the art. Typically such separations involve multiphase extraction, crystallization from a solvent or solvent mixture, distillation, sublimation, or chromatography. Chromatography can involve any number of methods including, for example: reverse-phase and normal phase; size exclusion; ion exchange; high, medium,
5 and low pressure liquid chromatography methods and apparatus; small scale analytical; simulated moving bed (SMB) and preparative thin or thick layer chromatography, as well as techniques of small scale thin layer and flash chromatography.

Another class of separation methods involves treatment of a mixture with
10 a reagent selected to bind to or render otherwise separable a desired product, unreacted starting material, reaction by product, or the like. Such reagents include adsorbents or absorbents such as activated carbon, molecular sieves, ion exchange media, or the like. Alternatively, the reagents can be acids in the case of a basic material, bases in the case of an acidic material, binding reagents such
15 as antibodies, binding proteins, selective chelators such as crown ethers, liquid/liquid ion extraction reagents (LIX), or the like.

Selection of appropriate methods of separation depends on the nature of the materials involved. For example, boiling point, and molecular weight in distillation and sublimation, presence or absence of polar functional groups in
20 chromatography, stability of materials in acidic and basic media in multiphase extraction, and the like. One skilled in the art will apply techniques most likely to achieve the desired separation.

A single stereoisomer, *e.g.*, an enantiomer, substantially free of its stereoisomer may be obtained by resolution of the racemic mixture using a
25 method such as formation of diastereomers using optically active resolving agents (Stereochemistry of Carbon Compounds, (1962) by E. L. Eliel, McGraw Hill; Lochmuller, C. H., (1975) *J. Chromatogr.*, 113:(3) 283-302). Racemic mixtures of chiral compounds of the invention can be separated and isolated by any suitable method, including: (1) formation of ionic, diastereomeric salts with
30 chiral compounds and separation by fractional crystallization or other methods, (2) formation of diastereomeric compounds with chiral derivatizing reagents, separation of the diastereomers, and conversion to the pure stereoisomers, and

(3) separation of the substantially pure or enriched stereoisomers directly under chiral conditions.

Under method (1), diastereomeric salts can be formed by reaction of enantiomerically pure chiral bases such as brucine, quinine, ephedrine, 5 strychnine, α -methyl- β -phenylethylamine (amphetamine), and the like with asymmetric compounds bearing acidic functionality, such as carboxylic acid and sulfonic acid. The diastereomeric salts may be induced to separate by fractional crystallization or ionic chromatography. For separation of the optical isomers of amino compounds, addition of chiral carboxylic or sulfonic acids, such as 10 camphorsulfonic acid, tartaric acid, mandelic acid, or lactic acid can result in formation of the diastereomeric salts.

Alternatively, by method (2), the substrate to be resolved is reacted with one enantiomer of a chiral compound to form a diastereomeric pair (Eliel, E. and Wilen, S. (1994) Stereochemistry of Organic Compounds, John Wiley & Sons, 15 Inc., p. 322). Diastereomeric compounds can be formed by reacting asymmetric compounds with enantiomerically pure chiral derivatizing reagents, such as menthyl derivatives, followed by separation of the diastereomers and hydrolysis to yield the free, enantiomerically enriched xanthene. A method of determining optical purity involves making chiral esters, such as a menthyl ester, *e.g.*, 20 (-)-menthyl chloroformate in the presence of base, or Mosher ester, α -methoxy- α -(trifluoromethyl)phenyl acetate (Jacob III. (1982) *J. Org. Chem.* 47:4165), of the racemic mixture, and analyzing the NMR spectrum for the presence of the two atropisomeric diastereomers. Stable diastereomers of atropisomeric compounds can be separated and isolated by normal- and reverse-phase 25 chromatography following methods for separation of atropisomeric naphthyl-isoquinolines (Hoye, T., WO 96/15111). By method (3), a racemic mixture of two enantiomers can be separated by chromatography using a chiral stationary phase (Chiral Liquid Chromatography (1989) W. J. Lough, Ed. Chapman and Hall, New York; Okamoto, (1990) *J. of Chromatogr.* 513:375-378). Enriched or 30 purified enantiomers can be distinguished by methods used to distinguish other chiral molecules with asymmetric carbon atoms, such as optical rotation and circular dichroism.

Examples General Section

A number of exemplary methods for the preparation of compounds of the invention are provided herein, for example, in the Examples hereinbelow. These methods are intended to illustrate the nature of such preparations and are not
5 intended to limit the scope of applicable methods. Certain compounds of the invention can be used as intermediates for the preparation of other compounds of the invention. For example, the interconversion of various phosphonate compounds of the invention is illustrated below.

Interconversions of the Phosphonates R-Link-P(O)(OR¹)₂, R-Link-P(O)(OR¹)(OH) and R-Link-P(O)(OH)₂.

The following schemes **32-38** described the preparation of phosphonate esters of the general structure R-link-P(O)(OR¹)₂, in which the groups R¹ may be the same or different. The R¹ groups attached to a phosphonate ester, or to precursors thereto, may be changed using established chemical transformations.
15 The interconversion reactions of phosphonates are illustrated in Scheme **S32**. The group R in Scheme **32** represents the substructure, *i.e.* the drug “scaffold, to which the substituent link-P(O)(OR¹)₂ is attached, either in the compounds of the invention, or in precursors thereto. At the point in the synthetic route of conducting a phosphonate interconversion, certain functional groups in R may be
20 protected. The methods employed for a given phosphonate transformation depend on the nature of the substituent R¹, and of the substrate to which the phosphonate group is attached. The preparation and hydrolysis of phosphonate esters is described in Organic Phosphorus Compounds, G. M. Kosolapoff, L. Maeir, eds, Wiley, 1976, p. 9ff.

25 In general, synthesis of phosphonate esters is achieved by coupling a nucleophile amine or alcohol with the corresponding activated phosphonate electrophilic precursor. For example, chlorophosphonate addition on to 5'-hydroxy of nucleoside is a well known method for preparation of nucleoside phosphate monoesters. The activated precursor can be prepared by several well
30 known methods. Chlorophosphonates useful for synthesis of the prodrugs are prepared from the substituted-1,3-propanediol (Wissner, et al, (1992) *J. Med Chem.* 35:1650). Chlorophosphonates are made by oxidation of the corresponding chlorophospholanes (Anderson, et al, (1984) *J. Org. Chem.*

49:1304) which are obtained by reaction of the substituted diol with phosphorus trichloride. Alternatively, the chlorophosphonate agent is made by treating substituted-1,3-diols with phosphorusoxychloride (Patois, et al, (1990) *J. Chem. Soc. Perkin Trans. I*, 1577). Chlorophosphonate species may also be generated in situ from corresponding cyclic phosphites (Silverburg, et al., (1996) *Tetrahedron lett.*, 37:771-774), which in turn can be either made from chlorophospholane or phosphoramidate intermediate. Phosphorofluoridate intermediate prepared either from pyrophosphate or phosphoric acid may also act as precursor in preparation of cyclic prodrugs (Watanabe et al., (1988) *Tetrahedron lett.*, 29:5763-66).

Phosphonate prodrugs of the present invention may also be prepared from the free acid by Mitsunobu reactions (Mitsunobu, (1981) *Synthesis*, 1; Campbell, (1992) *J. Org. Chem.* 57:6331), and other acid coupling reagents including, but not limited to, carbodiimides (Alexander, et al, (1994) *Collect. Czech. Chem. Commun.* 59:1853; Casara et al, (1992) *Bioorg. Med. Chem. Lett.* 2:145; Ohashi et al, (1988) *Tetrahedron Lett.*, 29:1189), and benzotriazolyloxytris-(dimethylamino)phosphonium salts (Campagne et al (1993) *Tetrahedron Lett.* 34:6743).

Aryl halides undergo Ni^{+2} catalyzed reaction with phosphite derivatives to give aryl phosphonate containing compounds (Balthazar, et al (1980) *J. Org. Chem.* 45:5425). Phosphonates may also be prepared from the chlorophosphonate in the presence of a palladium catalyst using aromatic triflates (Petrakis et al (1987) *J. Am. Chem. Soc.* 109:2831; Lu et al (1987) *Synthesis* 726). In another method, aryl phosphonate esters are prepared from aryl phosphates under anionic rearrangement conditions (Melvin (1981) *Tetrahedron Lett.* 22:3375; Casteel et al (1991) *Synthesis*, 691). N-Alkoxy aryl salts with alkali metal derivatives of cyclic alkyl phosphonate provide general synthesis for heteroaryl-2-phosphonate linkers (Redmore (1970) *J. Org. Chem.* 35:4114). These above mentioned methods can also be extended to compounds where the W^5 group is a heterocycle. Cyclic-1,3-propanyl prodrugs of phosphonates are also synthesized from phosphonic diacids and substituted propane-1,3-diols using a coupling reagent such as 1,3-dicyclohexylcarbodiimide (DCC) in presence of a base (e.g., pyridine). Other carbodiimide based coupling agents like 1,3-disopropylcarbodiimide or water

soluble reagent, 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI) can also be utilized for the synthesis of cyclic phosphonate prodrugs.

The conversion of a phosphonate diester **S32.1** into the corresponding phosphonate monoester **S32.2** (Scheme 32, Reaction 1) is accomplished by a number of methods. For example, the ester **S32.1** in which R¹ is an aralkyl group such as benzyl, is converted into the monoester compound **S32.2** by reaction with a tertiary organic base such as diazabicyclooctane (DABCO) or quinuclidine, as described in *J. Org. Chem.* (1995) 60:2946. The reaction is performed in an inert hydrocarbon solvent such as toluene or xylene, at about 110 °C. The conversion of the diester **S32.1** in which R¹ is an aryl group such as phenyl, or an alkenyl group such as allyl, into the monoester **S32.2** is effected by treatment of the ester **S32.1** with a base such as aqueous sodium hydroxide in acetonitrile or lithium hydroxide in aqueous tetrahydrofuran. Phosphonate diesters **S32.1** in which one of the groups R¹ is aralkyl, such as benzyl, and the other is alkyl, is converted into the monoesters **S32.2** in which R¹ is alkyl by hydrogenation, for example using a palladium on carbon catalyst. Phosphonate diesters in which both of the groups R¹ are alkenyl, such as allyl, is converted into the monoester **S32.2** in which R¹ is alkenyl, by treatment with chlorotris(triphenylphosphine)rhodium (Wilkinson's catalyst) in aqueous ethanol at reflux, optionally in the presence of diazabicyclooctane, for example by using the procedure described in *J. Org. Chem.* (1973) 38:3224, for the cleavage of allyl carboxylates.

The conversion of a phosphonate diester **S32.1** or a phosphonate monoester **S32.2** into the corresponding phosphonic acid **S32.3** (Scheme 32, Reactions 2 and 3) can be effected by reaction of the diester or the monoester with trimethylsilyl bromide, as described in *J. Chem. Soc., Chem. Comm.*, (1979) 739. The reaction is conducted in an inert solvent such as, for example, dichloromethane, optionally in the presence of a silylating agent such as bis(trimethylsilyl)trifluoroacetamide, at ambient temperature. A phosphonate monoester **S32.2** in which R¹ is aralkyl such as benzyl, is converted into the corresponding phosphonic acid **S32.3** by hydrogenation over a palladium catalyst, or by treatment with hydrogen chloride in an ethereal solvent such as dioxane. A phosphonate monoester **S32.2** in which R¹ is alkenyl such as, for

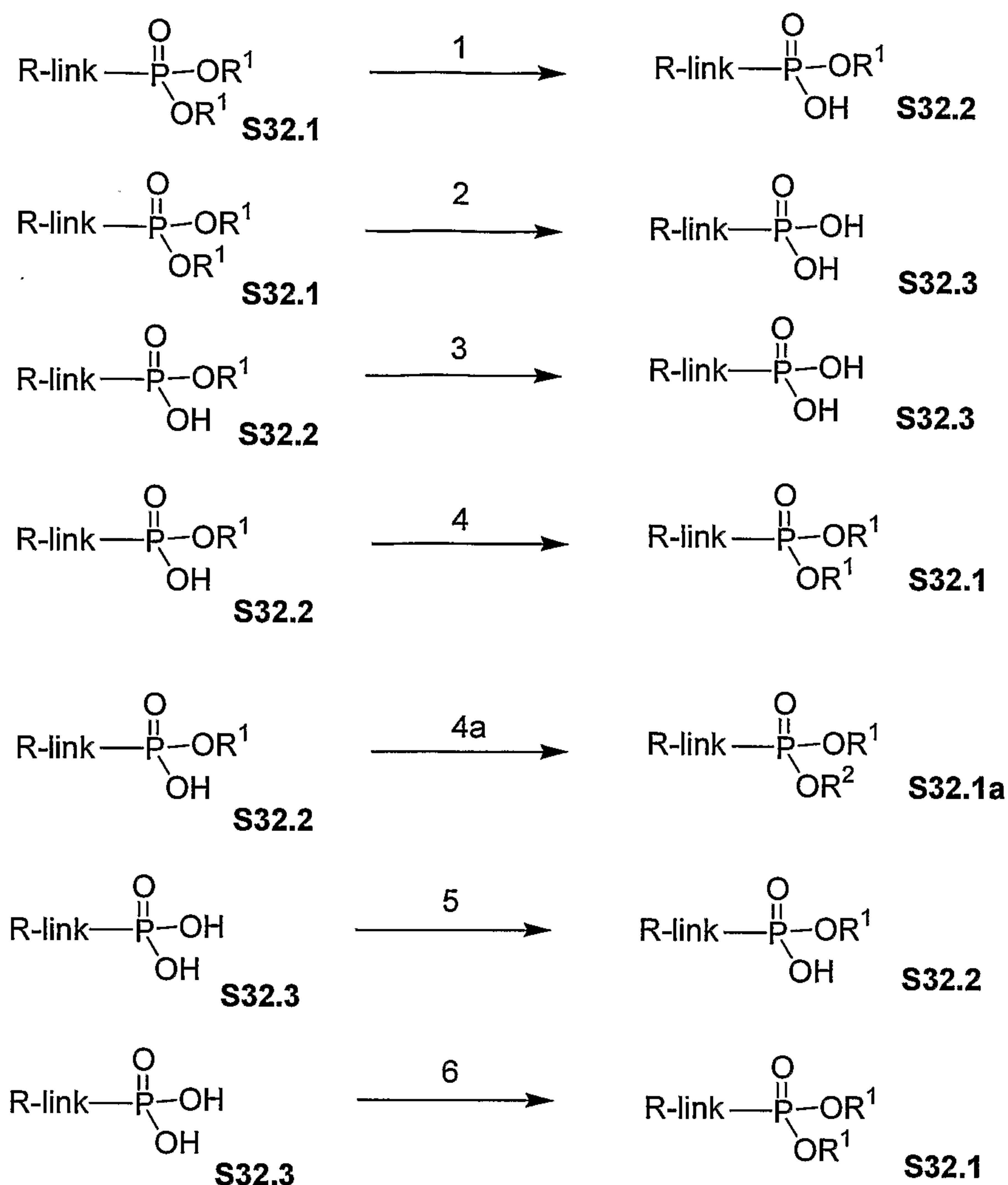
example, allyl, is converted into the phosphonic acid **S32.3** by reaction with Wilkinson's catalyst in an aqueous organic solvent, for example in 15% aqueous acetonitrile, or in aqueous ethanol, for example using the procedure described in *Helv. Chim. Acta.* (1985) 68:618. Palladium catalyzed hydrogenolysis of
5 phosphonate esters **S32.1** in which R^1 is benzyl is described in *J. Org. Chem.* (1959) 24:434. Platinum-catalyzed hydrogenolysis of phosphonate esters **S32.1** in which R^1 is phenyl is described in *J. Am. Chem. Soc.* (1956) 78:2336.

The conversion of a phosphonate monoester **S32.2** into a phosphonate diester **S32.1** (Scheme 32, Reaction 4) in which the newly introduced R^1 group is
10 alkyl, aralkyl, haloalkyl such as chloroethyl, or aralkyl is effected by a number of reactions in which the substrate **S32.2** is reacted with a hydroxy compound R^1OH , in the presence of a coupling agent. Typically, the second phosphonate ester group is different than the first introduced phosphonate ester group, *i.e.* R^1 is followed by the introduction of R^2 where each of R^1 and R^2 is alkyl, aralkyl,
15 haloalkyl such as chloroethyl, or aralkyl (Scheme 32, Reaction 4a) whereby **S32.2** is converted to **S32.1a**. Suitable coupling agents are those employed for the preparation of carboxylate esters, and include a carbodiimide such as dicyclohexylcarbodiimide, in which case the reaction is preferably conducted in a basic organic solvent such as pyridine, or (benzotriazol-1-
20 yloxy)tripyrrolidinophosphonium hexafluorophosphate (PYBOP, Sigma), in which case the reaction is performed in a polar solvent such as dimethylformamide, in the presence of a tertiary organic base such as diisopropylethylamine, or Aldrithiol-2 (Aldrich) in which case the reaction is conducted in a basic solvent such as pyridine, in the presence of a triaryl
25 phosphine such as triphenylphosphine. Alternatively, the conversion of the phosphonate monoester **S32.2** to the diester **S32.1** is effected by the use of the Mitsunobu reaction, as described above (Scheme 7). The substrate is reacted with the hydroxy compound R^1OH , in the presence of diethyl azodicarboxylate and a triarylphosphine such as triphenyl phosphine. Alternatively, the
30 phosphonate monoester **S32.2** is transformed into the phosphonate diester **S32.1**, in which the introduced R^1 group is alkenyl or aralkyl, by reaction of the monoester with the halide R^1Br , in which R^1 is as alkenyl or aralkyl. The alkylation reaction is conducted in a polar organic solvent such as

dimethylformamide or acetonitrile, in the presence of a base such as cesium carbonate. Alternatively, the phosphonate monoester is transformed into the phosphonate diester in a two step procedure. In the first step, the phosphonate monoester **S32.2** is transformed into the chloro analog $\text{RP}(\text{O})(\text{OR}^1)\text{Cl}$ by
5 reaction with thionyl chloride or oxalyl chloride and the like, as described in Organic Phosphorus Compounds, G. M. Kosolapoff, L. Maeir, eds, Wiley, 1976, p. 17, and the thus-obtained product $\text{RP}(\text{O})(\text{OR}^1)\text{Cl}$ is then reacted with the hydroxy compound R^1OH , in the presence of a base such as triethylamine, to afford the phosphonate diester **S32.1**.

10 A phosphonic acid $\text{R-link-P}(\text{O})(\text{OH})_2$ is transformed into a phosphonate monoester $\text{RP}(\text{O})(\text{OR}^1)(\text{OH})$ (Scheme 32, Reaction 5) by means of the methods described above of for the preparation of the phosphonate diester $\text{R-link-P}(\text{O})(\text{OR}^1)_2$ **S32.1**, except that only one molar proportion of the component R^1OH or R^1Br is employed. Dialkyl phosphonates may be prepared according to
15 the methods of: Quast et al (1974) *Synthesis* 490; Stowell et al (1990) *Tetrahedron Lett.* 3261; US 5663159.

A phosphonic acid $\text{R-link-P}(\text{O})(\text{OH})_2$ **S32.3** is transformed into a phosphonate diester $\text{R-link-P}(\text{O})(\text{OR}^1)_2$ **S32.1** (Scheme 32, Reaction 6) by a coupling reaction with the hydroxy compound R^1OH , in the presence of a
20 coupling agent such as Aldrithiol-2 (Aldrich) and triphenylphosphine. The reaction is conducted in a basic solvent such as pyridine. Alternatively, phosphonic acids **S32.3** are transformed into phosphonic esters **S32.1** in which R^1 is aryl, by means of a coupling reaction employing, for example, dicyclohexylcarbodiimide in pyridine at ca 70 °C. Alternatively, phosphonic
25 acids **S32.3** are transformed into phosphonic esters **S32.1** in which R^1 is alkenyl, by means of an alkylation reaction. The phosphonic acid is reacted with the alkenyl bromide R^1Br in a polar organic solvent such as acetonitrile solution at reflux temperature, the presence of a base such as cesium carbonate, to afford the phosphonic ester **S32.1**.

Scheme 32

Preparation of phosphonate carbamates.

Phosphonate esters may contain a carbamate linkage. The preparation of carbamates is described in Comprehensive Organic Functional Group Transformations, A. R. Katritzky, ed., Pergamon, 1995, Vol. 6, p. 416ff, and in Organic Functional Group Preparations, by S. R. Sandler and W. Karo, Academic Press, 1986, p. 260ff. The carbamoyl group may be formed by reaction of a hydroxy group according to the methods known in the art, including the teachings of Ellis, US 2002/0103378 A1 and Hajima, US 6018049.

Scheme 33 illustrates various methods by which the carbamate linkage is synthesized. As shown in Scheme 33, in the general reaction generating carbamates, an alcohol S33.1, is converted into the activated derivative S33.2 in which Lv is a leaving group such as halo, imidazolyl, benztriazolyl and the like, as described herein. The activated derivative S33.2 is then reacted with an amine

S33.3, to afford the carbamate product S33.4. Examples 1 – 7 in Scheme 33 depict methods by which the general reaction is effected. Examples 8 - 10 illustrate alternative methods for the preparation of carbamates.

Scheme 33, Example 1 illustrates the preparation of carbamates
5 employing a chloroformyl derivative of the alcohol S33.5. In this procedure, the alcohol S33.5 is reacted with phosgene, in an inert solvent such as toluene, at about 0 °C, as described in *Org. Syn. Coll. Vol. 3*, 167, 1965, or with an equivalent reagent such as trichloromethoxy chloroformate, as described in *Org. Syn. Coll. Vol. 6*, 715, 1988, to afford the chloroformate S33.6. The latter
10 compound is then reacted with the amine component S33.3, in the presence of an organic or inorganic base, to afford the carbamate S33.7. For example, the chloroformyl compound S33.6 is reacted with the amine S33.3 in a water-miscible solvent such as tetrahydrofuran, in the presence of aqueous sodium hydroxide, as described in *Org. Syn. Coll. Vol. 3*, 167, 1965, to yield the
15 carbamate S33.7. Alternatively, the reaction is performed in dichloromethane in the presence of an organic base such as diisopropylethylamine or dimethylaminopyridine.

Scheme 33, Example 2 depicts the reaction of the chloroformate compound S33.6 with imidazole to produce the imidazolide S33.8. The
20 imidazolide product is then reacted with the amine S33.3 to yield the carbamate S33.7. The preparation of the imidazolide is performed in an aprotic solvent such as dichloromethane at 0°, and the preparation of the carbamate is conducted in a similar solvent at ambient temperature, optionally in the presence of a base such as dimethylaminopyridine, as described in *J. Med. Chem.*, 1989, 32, 357.

25 Scheme 33 Example 3, depicts the reaction of the chloroformate S33.6 with an activated hydroxyl compound R"OH, to yield the mixed carbonate ester S33.10. The reaction is conducted in an inert organic solvent such as ether or dichloromethane, in the presence of a base such as dicyclohexylamine or triethylamine. The hydroxyl component R"OH is selected from the group of
30 compounds S33.19 - S33.24 shown in Scheme 33, and similar compounds. For example, if the component R"OH is hydroxybenztriazole S33.19, N-hydroxysuccinimide S33.20, or pentachlorophenol, S33.21, the mixed carbonate S33.10 is obtained by the reaction of the chloroformate with the hydroxyl

compound in an ethereal solvent in the presence of dicyclohexylamine, as described in *Can. J. Chem.*, 1982, 60, 976. A similar reaction in which the component R"OH is pentafluorophenol **S33.22** or 2-hydroxypyridine **S33.23** is performed in an ethereal solvent in the presence of triethylamine, as described in
5 *Syn.*, 1986, 303, and *Chem. Ber.* 118, 468, 1985.

Scheme 33 Example 4 illustrates the preparation of carbamates in which an alkyloxycarbonylimidazole **S33.8** is employed. In this procedure, an alcohol **S33.5** is reacted with an equimolar amount of carbonyl diimidazole **S33.11** to prepare the intermediate **S33.8**. The reaction is conducted in an aprotic organic
10 solvent such as dichloromethane or tetrahydrofuran. The acyloxyimidazole **S33.8** is then reacted with an equimolar amount of the amine R'NH₂ to afford the carbamate **S33.7**. The reaction is performed in an aprotic organic solvent such as dichloromethane, as described in *Tet. Lett.*, 42, 2001, 5227, to afford the carbamate **S33.7**.

15 Scheme 33, Example 5 illustrates the preparation of carbamates by means of an intermediate alkoxy carbonyl benztriazole **S33.13**. In this procedure, an alcohol ROH is reacted at ambient temperature with an equimolar amount of benztriazole carbonyl chloride **S33.12**, to afford the alkoxy carbonyl product **S33.13**. The reaction is performed in an organic solvent such as benzene or
20 toluene, in the presence of a tertiary organic amine such as triethylamine, as described in *Synthesis.*, 1977, 704. The product is then reacted with the amine R'NH₂ to afford the carbamate **S33.7**. The reaction is conducted in toluene or ethanol, at from ambient temperature to about 80 °C as described in *Synthesis.*, 1977, 704.

25 Scheme 33, Example 6 illustrates the preparation of carbamates in which a carbonate (R"O)₂CO, **S33.14**, is reacted with an alcohol **S33.5** to afford the intermediate alkyloxycarbonyl intermediate **S33.15**. The latter reagent is then reacted with the amine R'NH₂ to afford the carbamate **S33.7**. The procedure in which the reagent **S33.15** is derived from hydroxybenztriazole **S33.19** is
30 described in *Synthesis*, 1993, 908; the procedure in which the reagent **S33.15** is derived from N-hydroxysuccinimide **S33.20** is described in *Tet. Lett.*, 1992, 2781; the procedure in which the reagent **S33.15** is derived from 2-hydroxypyridine **S33.23** is described in *Tet. Lett.*, 1991, 4251; the procedure in

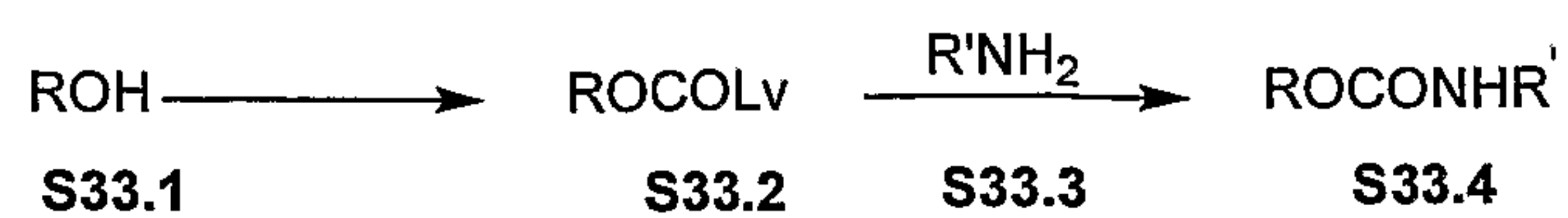
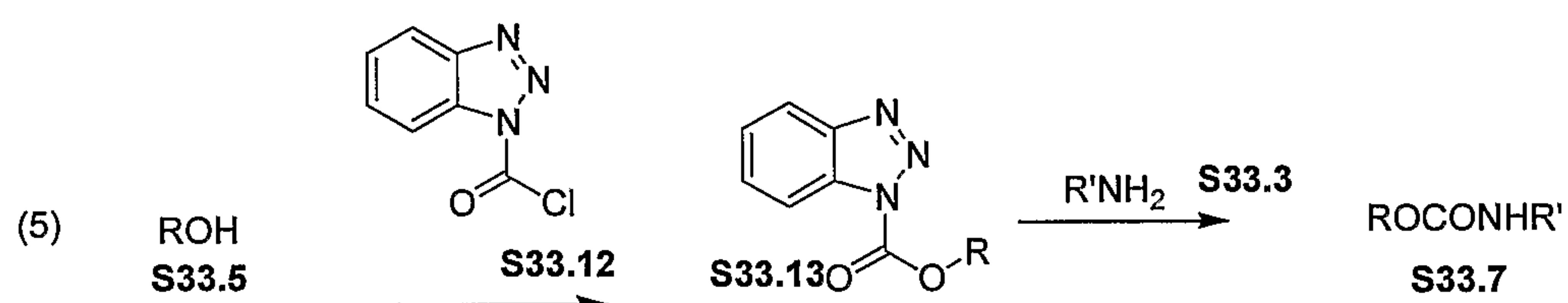
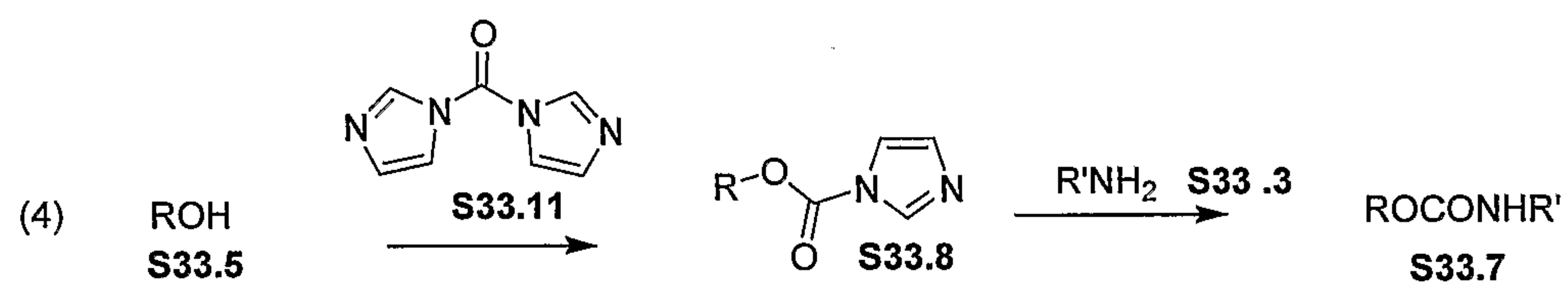
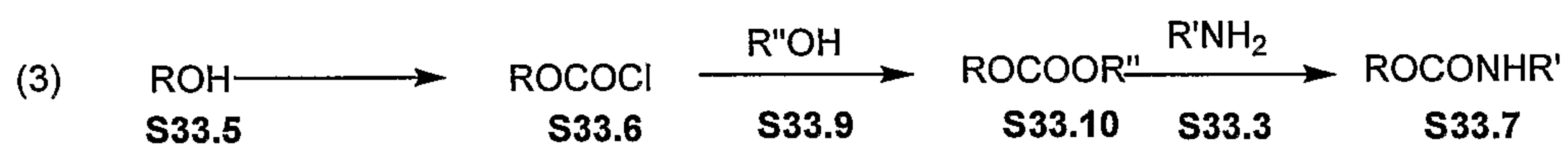
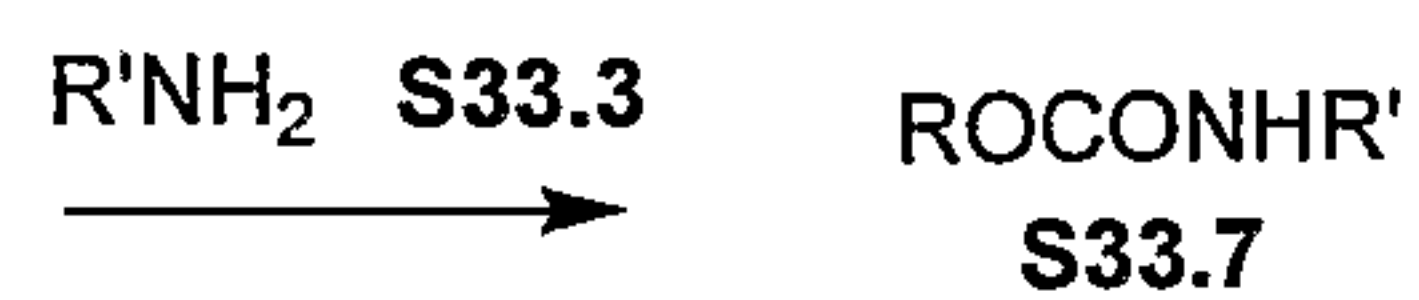
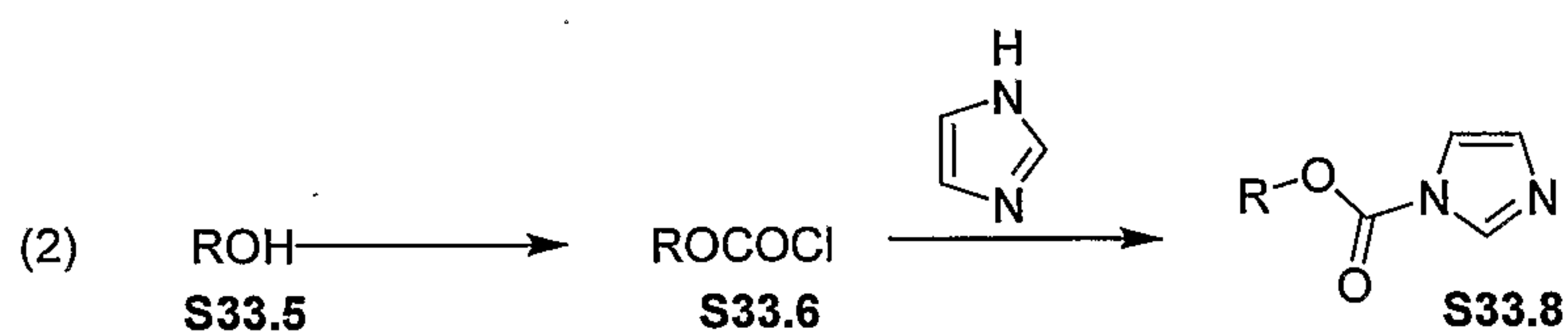
which the reagent **S33.15** is derived from 4-nitrophenol **S33.24** is described in *Synthesis*. 1993, 103. The reaction between equimolar amounts of the alcohol ROH and the carbonate **S33.14** is conducted in an inert organic solvent at ambient temperature.

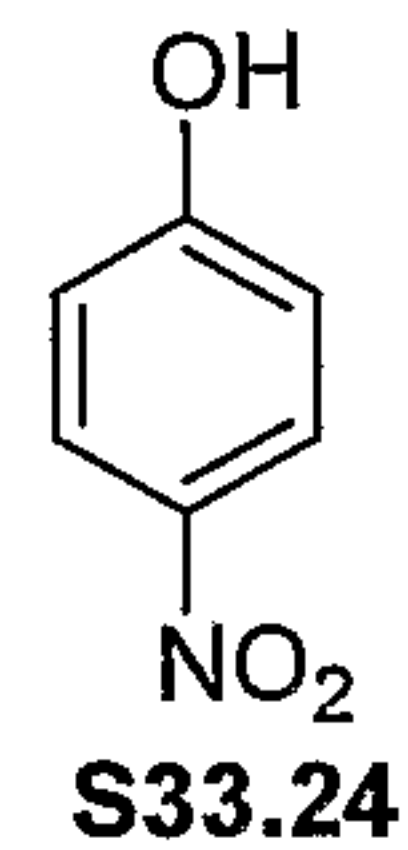
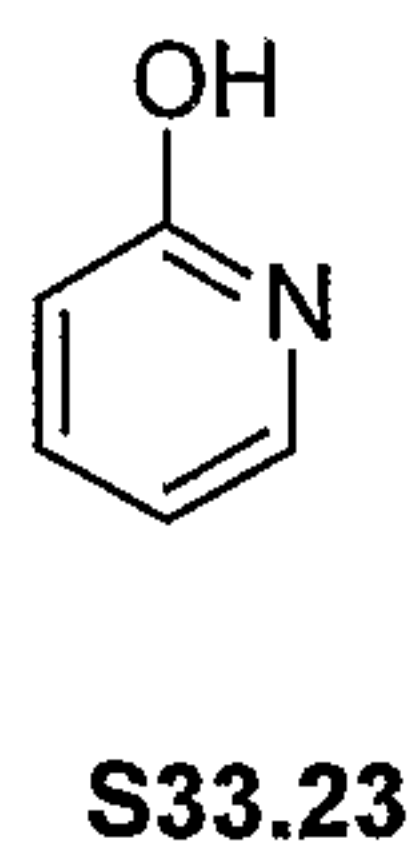
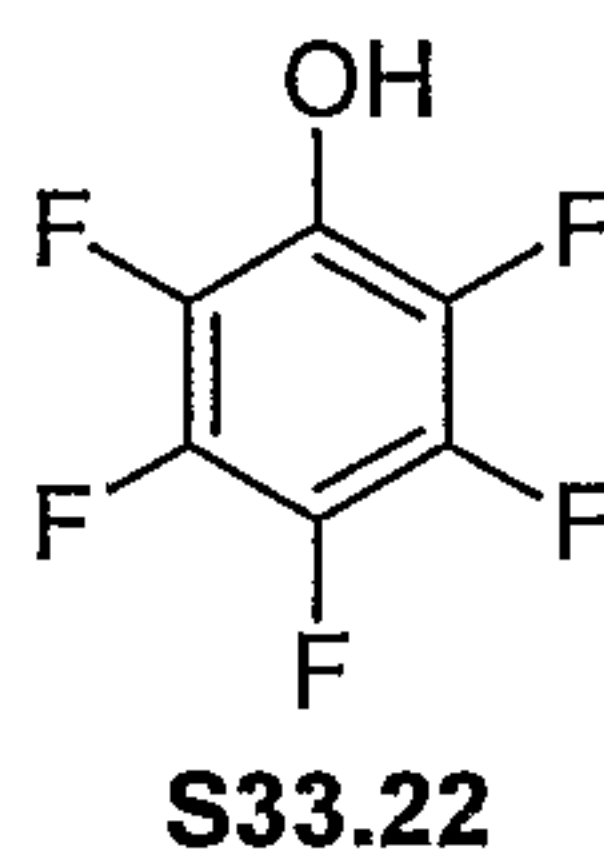
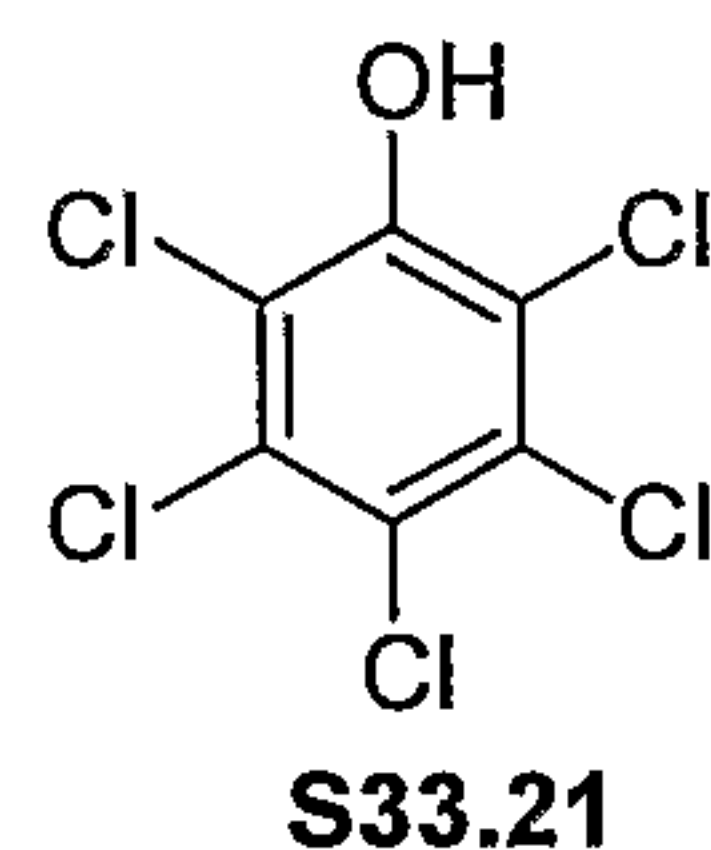
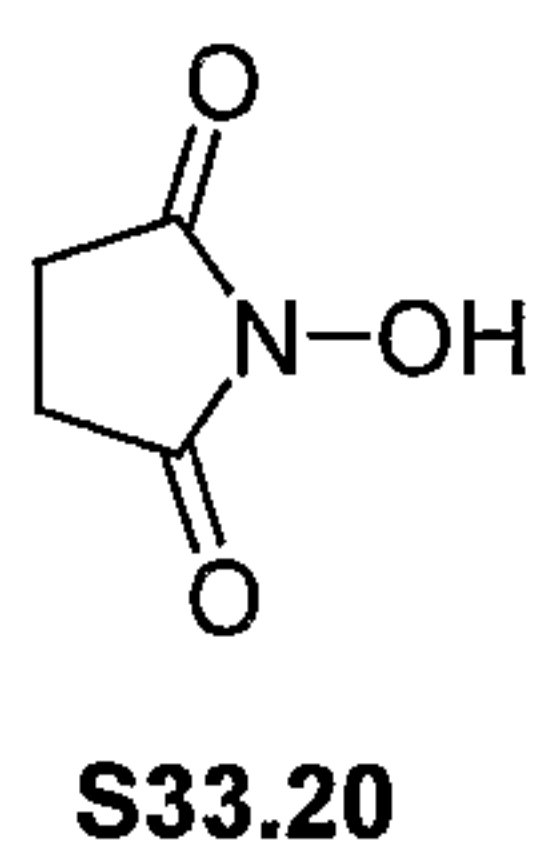
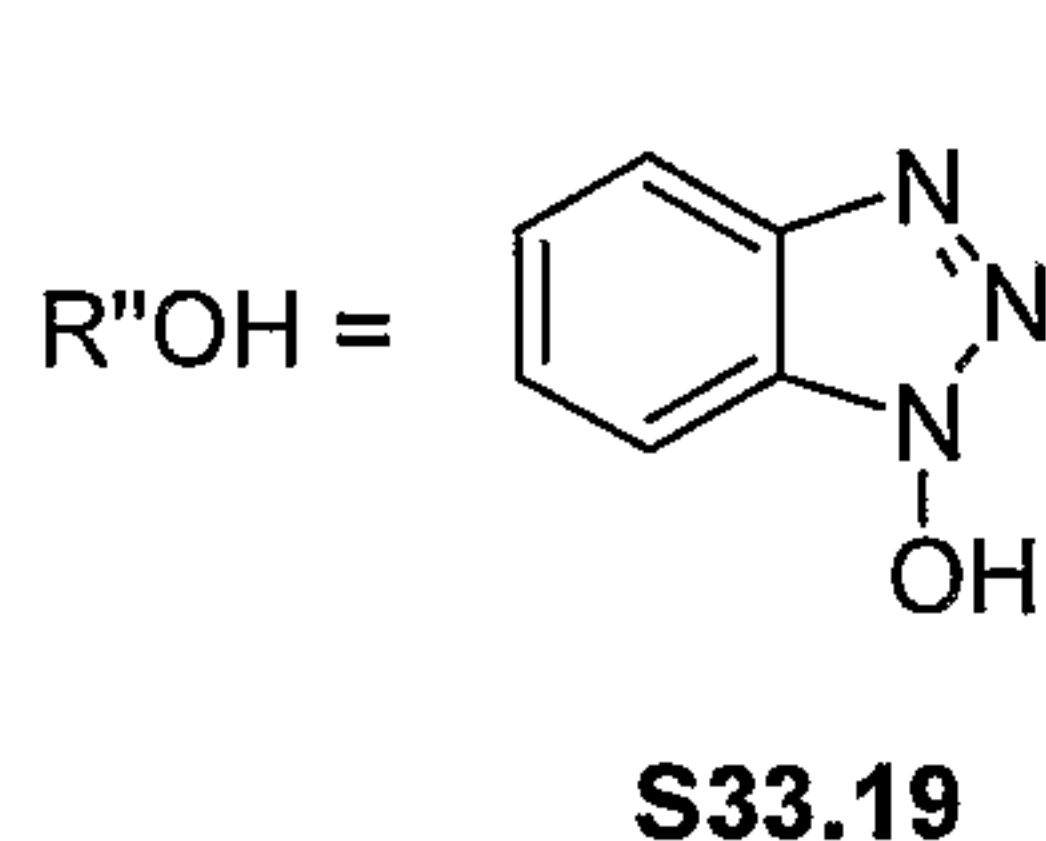
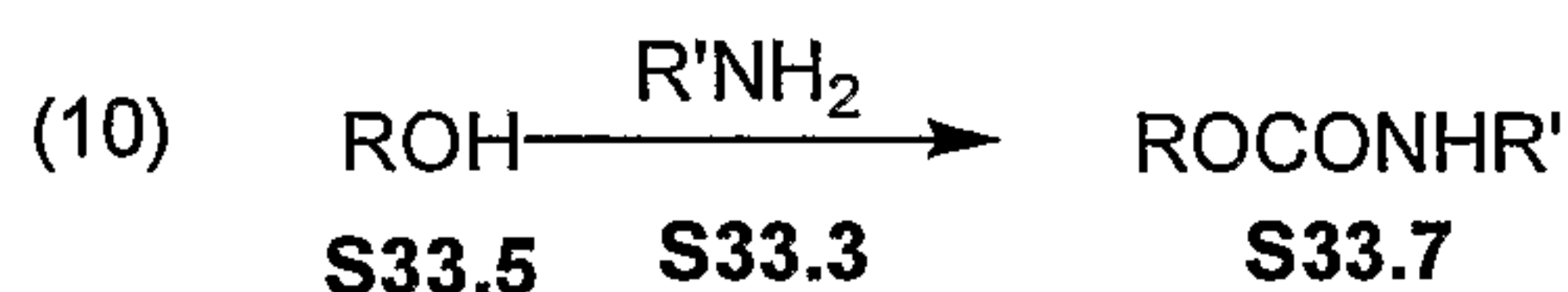
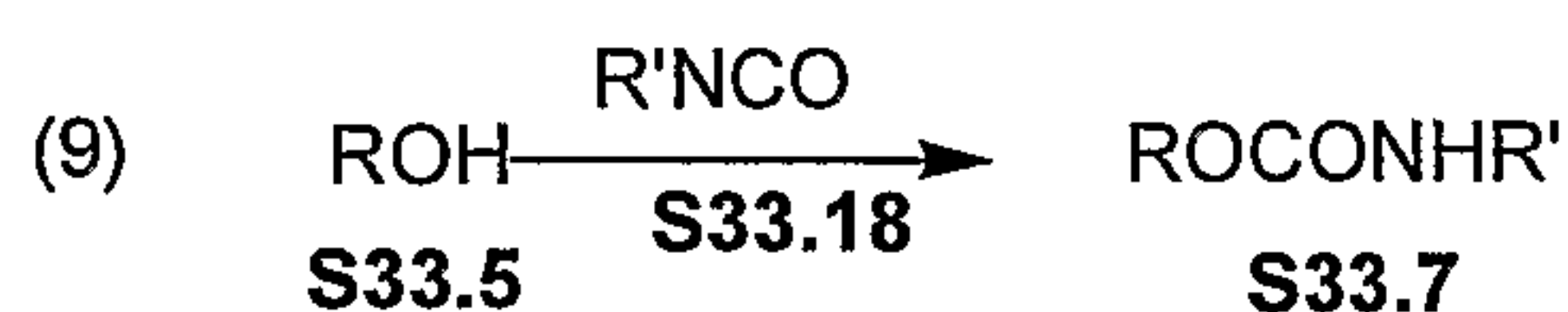
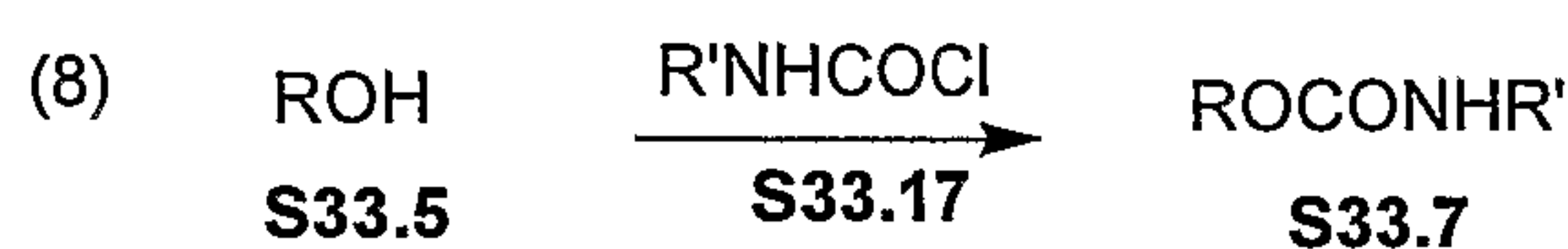
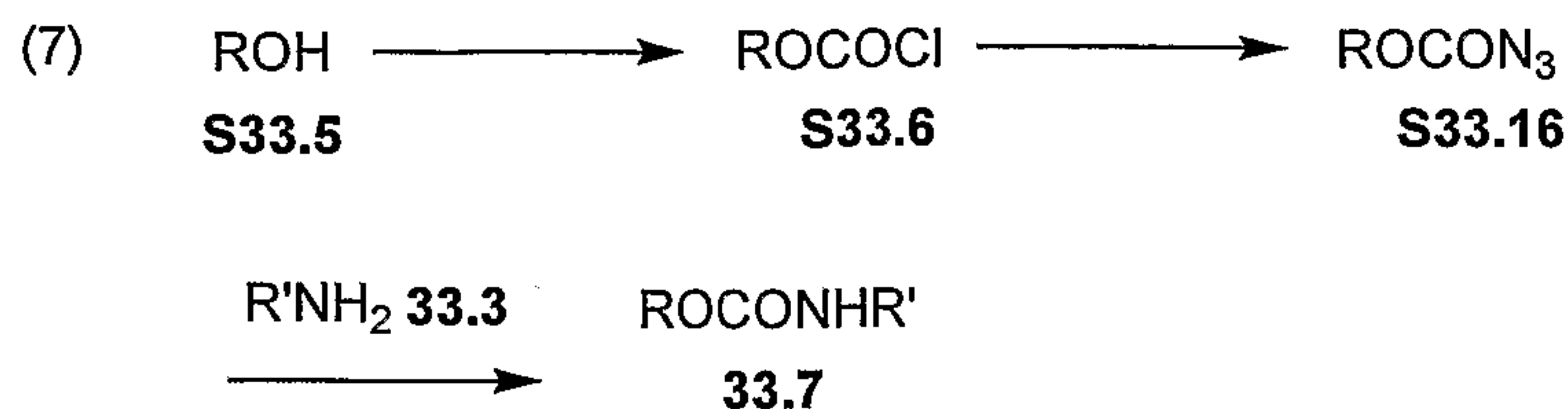
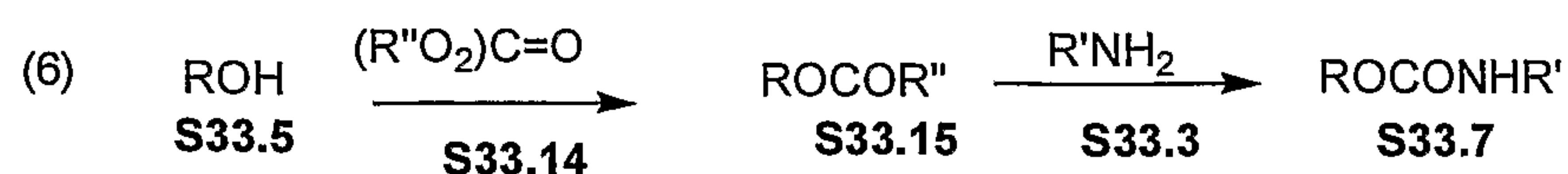
5 Scheme **33**, Example **7** illustrates the preparation of carbamates from alkoxy carbonyl azides **S33.16**. In this procedure, an alkyl chloroformate **S33.6** is reacted with an azide, for example sodium azide, to afford the alkoxy carbonyl azide **S33.16**. The latter compound is then reacted with an equimolar amount of the amine $R'NH_2$ to afford the carbamate **S33.7**. The reaction is conducted at
10 ambient temperature in a polar aprotic solvent such as dimethylsulfoxide, for example as described in *Synthesis*., 1982, 404.

 Scheme **33**, Example **8** illustrates the preparation of carbamates by means of the reaction between an alcohol ROH and the chloroformyl derivative of an amine **S33.17**. In this procedure, which is described in Synthetic Organic
15 Chemistry, R. B. Wagner, H. D. Zook, Wiley, 1953, p. 647, the reactants are combined at ambient temperature in an aprotic solvent such as acetonitrile, in the presence of a base such as triethylamine, to afford the carbamate **S33.7**.

 Scheme **33**, Example **9** illustrates the preparation of carbamates by means of the reaction between an alcohol ROH and an isocyanate **S33.18**. In this
20 procedure, which is described in Synthetic Organic Chemistry, R. B. Wagner, H. D. Zook, Wiley, 1953, p. 645, the reactants are combined at ambient temperature in an aprotic solvent such as ether or dichloromethane and the like, to afford the carbamate **S33.7**.

 Scheme **33**, Example **10** illustrates the preparation of carbamates by
25 means of the reaction between an alcohol ROH and an amine $R'NH_2$. In this procedure, which is described in *Chem. Lett.* 1972, 373, the reactants are combined at ambient temperature in an aprotic organic solvent such as tetrahydrofuran, in the presence of a tertiary base such as triethylamine, and selenium. Carbon monoxide is passed through the solution and the reaction
30 proceeds to afford the carbamate **S33.7**.

Scheme 33. Preparation of carbamates.**General reaction****Examples**



Preparation of Carboalkoxy-Substituted Phosphonate Bisamides,
Monoamides, Diesters and Monoesters.

A number of methods are available for the conversion of phosphonic
5 acids into amides and esters. In one group of methods, the phosphonic acid is
either converted into an isolated activated intermediate such as a phosphoryl
chloride, or the phosphonic acid is activated in situ for reaction with an amine or
a hydroxy compound.

The conversion of phosphonic acids into phosphoryl chlorides is accomplished by reaction with thionyl chloride, for example as described in *J. Gen. Chem. USSR*, 1983, 53, 480, *Zh. Obschei Khim.*, 1958, 28, 1063, or *J. Org. Chem.*, 1994, 59, 6144, or by reaction with oxalyl chloride, as described in *J. Am. Chem. Soc.*, 1994, 116, 3251, or *J. Org. Chem.*, 1994, 59, 6144, or by reaction with phosphorus pentachloride, as described in *J. Org. Chem.*, 2001, 66, 329, or in *J. Med. Chem.*, 1995, 38, 1372. The resultant phosphoryl chlorides are then reacted with amines or hydroxy compounds in the presence of a base to afford the amidate or ester products.

Phosphonic acids are converted into activated imidazolyl derivatives by reaction with carbonyl diimidazole, as described in *J. Chem. Soc., Chem. Comm.* (1991) 312, or *Nucleosides & Nucleotides* (2000) 19:1885. Activated sulfonyloxy derivatives are obtained by the reaction of phosphonic acids with trichloromethylsulfonyl chloride or with triisopropylbenzenesulfonyl chloride, as described in *Tet. Lett.* (1996) 7857, or *Bioorg. Med. Chem. Lett.* (1998) 8:663. The activated sulfonyloxy derivatives are then reacted with amines or hydroxy compounds to afford amidates or esters.

Alternatively, the phosphonic acid and the amine or hydroxy reactant are combined in the presence of a diimide coupling agent. The preparation of phosphonic amidates and esters by means of coupling reactions in the presence of dicyclohexyl carbodiimide is described, for example, in *J. Chem. Soc., Chem. Comm.* (1991) 312 or *Coll. Czech. Chem. Comm.* (1987) 52:2792. The use of ethyl dimethylaminopropyl carbodiimide for activation and coupling of phosphonic acids is described in *Tet. Lett.*, (2001) 42:8841, or *Nucleosides & Nucleotides* (2000) 19:1885.

A number of additional coupling reagents have been described for the preparation of amidates and esters from phosphonic acids. The agents include Aldrithiol-2, and PYBOP and BOP, as described in *J. Org. Chem.*, 1995, 60, 5214, and *J. Med. Chem.* (1997) 40:3842, mesitylene-2-sulfonyl-3-nitro-1,2,4-triazole (MSNT), as described in *J. Med. Chem.* (1996) 39:4958, diphenylphosphoryl azide, as described in *J. Org. Chem.* (1984) 49:1158, 1-(2,4,6-triisopropylbenzenesulfonyl-3-nitro-1,2,4-triazole (TPSNT) as described in *Bioorg. Med. Chem. Lett.* (1998) 8:1013,

bromotris(dimethylamino)phosphonium hexafluorophosphate (BroP), as described in *Tet. Lett.*, (1996) 37:3997, 2-chloro-5,5-dimethyl-2-oxo-1,3,2-dioxaphosphinane, as described in *Nucleosides Nucleotides* 1995, 14, 871, and diphenyl chlorophosphate, as described in *J. Med. Chem.*, 1988, 31, 1305.

5 Phosphonic acids are converted into amidates and esters by means of the Mitsunobu reaction, in which the phosphonic acid and the amine or hydroxy reactant are combined in the presence of a triaryl phosphine and a dialkyl azodicarboxylate. The procedure is described in *Org. Lett.*, 2001, 3, 643, or *J. Med. Chem.*, 1997, 40, 3842.

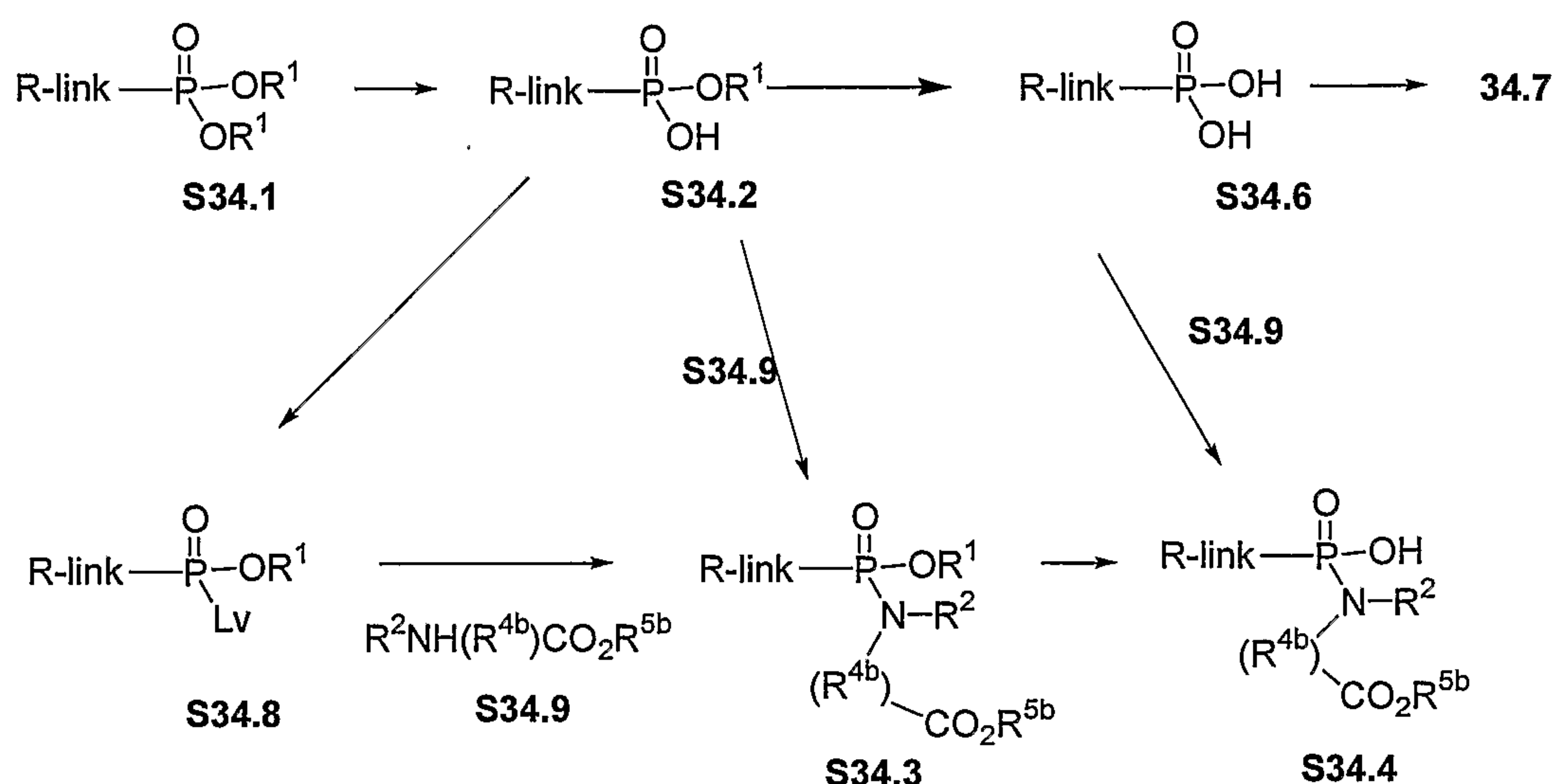
10 Phosphonic esters are also obtained by the reaction between phosphonic acids and halo compounds, in the presence of a suitable base. The method is described, for example, in *Anal. Chem.*, 1987, 59, 1056, or *J. Chem. Soc. Perkin Trans., I*, 1993, 19, 2303, or *J. Med. Chem.*, 1995, 38, 1372, or *Tet. Lett.*, 2002, 43, 1161.

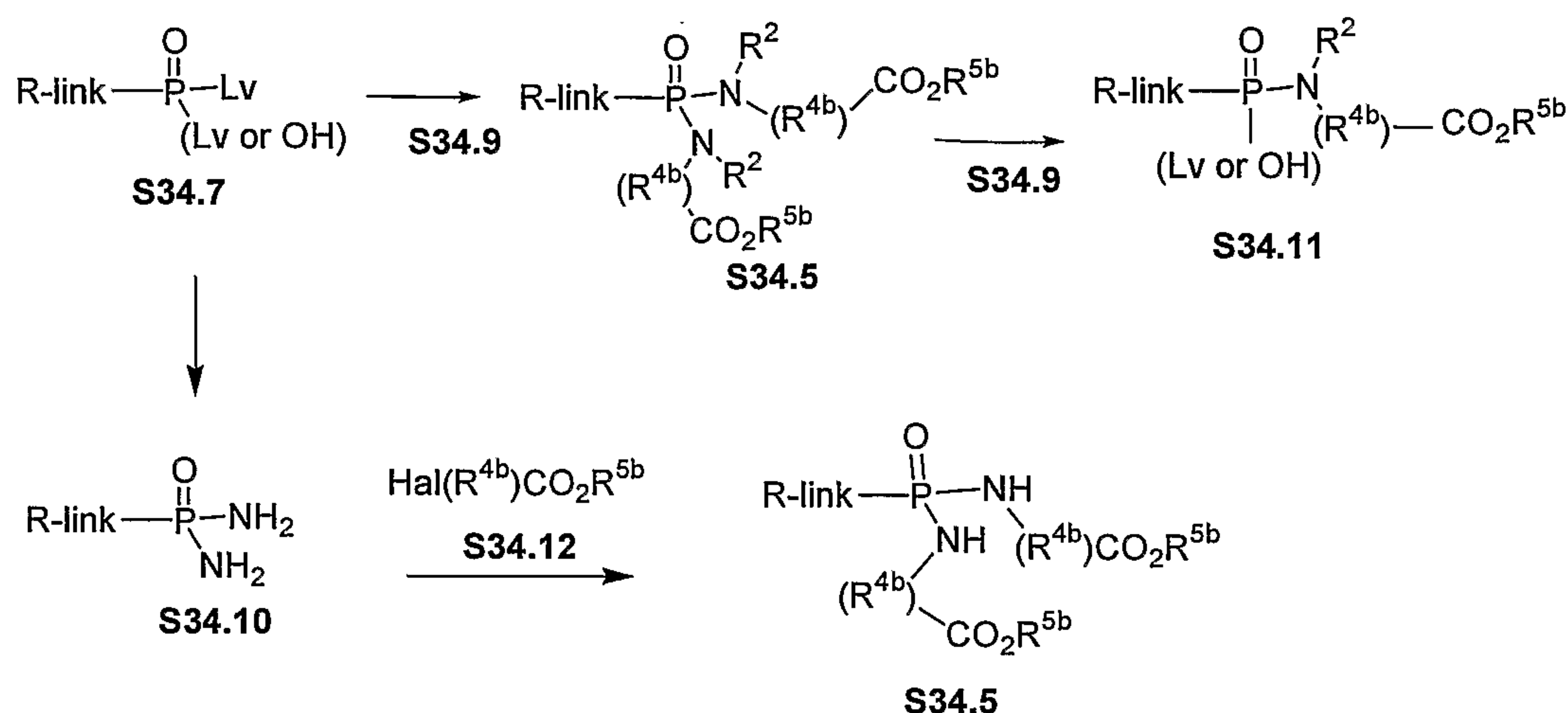
15 Schemes 34-37 illustrate the conversion of phosphonate esters and phosphonic acids into carboalkoxy-substituted phosphonbisamidates (Scheme 34), phosphonamidates (Scheme 35), phosphonate monoesters (Scheme 36) and phosphonate diesters, (Scheme 37). Scheme 38 illustrates synthesis of gem-dialkyl amino phosphonate reagents.

20 Scheme 34 illustrates various methods for the conversion of phosphonate diesters S34.1 into phosphonbisamidates S34.5. The diester S34.1, prepared as described previously, is hydrolyzed, either to the monoester S34.2 or to the phosphonic acid S34.6. The methods employed for these transformations are described above. The monoester S34.2 is converted into the monoamidate S34.3
25 by reaction with an aminoester S34.9, in which the group R^2 is H or alkyl; the group R^{4b} is a divalent alkylene moiety such as, for example, $CHCH_3$, $CHCH_2CH_3$, $CH(CH(CH_3)_2)$, $CH(CH_2Ph)$, and the like, or a side chain group present in natural or modified amino acids; and the group R^{5b} is (C_1-C_{12}) alkyl, such as methyl, ethyl, propyl, isopropyl, or isobutyl; (C_6-C_{20}) aryl, such as
30 phenyl or substituted phenyl; or (C_6-C_{20}) arylalkyl, such as benzyl or benzyldryl. The reactants are combined in the presence of a coupling agent such as a carbodiimide, for example dicyclohexyl carbodiimide, as described in

J. Am. Chem. Soc., (1957) 79:3575, optionally in the presence of an activating agent such as hydroxybenztriazole, to yield the amidate product **S34.3**. The amidate-forming reaction is also effected in the presence of coupling agents such as BOP, as described in *J. Org. Chem.* (1995) 60:5214, Aldrithiol, PYBOP and similar coupling agents used for the preparation of amides and esters.

Alternatively, the reactants **S34.2** and **S34.9** are transformed into the monoamidate **S34.3** by means of a Mitsunobu reaction. The preparation of amidates by means of the Mitsunobu reaction is described in *J. Med. Chem.* (1995) 38:2742. Equimolar amounts of the reactants are combined in an inert solvent such as tetrahydrofuran in the presence of a triaryl phosphine and a dialkyl azodicarboxylate. The thus-obtained monoamidate ester **S34.3** is then transformed into amidate phosphonic acid **S34.4**. The conditions used for the hydrolysis reaction depend on the nature of the R^1 group, as described previously. The phosphonic acid amidate **S34.4** is then reacted with an aminoester **S34.9**, as described above, to yield the bisamidate product **S34.5**, in which the amino substituents are the same or different. Alternatively, the phosphonic acid **S34.6** may be treated with two different amino ester reagents simultaneously, *i.e.* **S34.9** where R^2 , R^{4b} or R^{5b} are different. The resulting mixture of bisamidate products **S34.5** may then be separable, *e.g.* by chromatography.

Scheme 34



An example of this procedure is shown in Scheme 34, Example 1. In this procedure, a dibenzyl phosphonate **S34.14** is reacted with diazabicyclooctane (DABCO) in toluene at reflux, as described in *J. Org. Chem.*, 1995, 60, 2946, to afford the monobenzyl phosphonate **S34.15**. The product is then reacted with equimolar amounts of ethyl alaninate **S34.16** and dicyclohexyl carbodiimide in pyridine, to yield the amidate product **S34.17**. The benzyl group is then removed, for example by hydrogenolysis over a palladium catalyst, to give the monoacid product **S34.18** which may be unstable according to *J. Med. Chem.* (1997) 40(23):3842. This compound **S34.18** is then reacted in a Mitsunobu reaction with ethyl leucinate **S34.19**, triphenyl phosphine and diethylazodicarboxylate, as described in *J. Med. Chem.*, 1995, 38, 2742, to produce the bisamidate product **S34.20**.

Using the above procedures, but employing in place of ethyl leucinate **S34.19** or ethyl alaninate **S34.16**, different aminoesters **S34.9**, the corresponding products **S34.5** are obtained.

Alternatively, the phosphonic acid **S34.6** is converted into the bisamidate **S34.5** by use of the coupling reactions described above. The reaction is performed in one step, in which case the nitrogen-related substituents present in the product **S34.5** are the same, or in two steps, in which case the nitrogen-related substituents can be different.

An example of the method is shown in Scheme 34, Example 2. In this procedure, a phosphonic acid **S34.6** is reacted in pyridine solution with excess ethyl phenylalaninate **S34.21** and dicyclohexylcarbodiimide, for example as

described in *J. Chem. Soc., Chem. Comm.*, 1991, 1063, to give the bisamidate product **S34.22**.

Using the above procedures, but employing, in place of ethyl phenylalaninate, different aminoesters **S34.9**, the corresponding products **S34.5** are obtained.

As a further alternative, the phosphonic acid **S34.6** is converted into the mono or bis-activated derivative **S34.7**, in which Lv is a leaving group such as chloro, imidazolyl, triisopropylbenzenesulfonyloxy *etc.* The conversion of phosphonic acids into chlorides **S34.7** (Lv = Cl) is effected by reaction with thionyl chloride or oxalyl chloride and the like, as described in Organic Phosphorus Compounds, G. M. Kosolapoff, L. Maeir, eds, Wiley, 1976, p. 17. The conversion of phosphonic acids into monoimidazolides **S34.7** (Lv = imidazolyl) is described in *J. Med. Chem.*, 2002, 45, 1284 and in *J. Chem. Soc. Chem. Comm.*, 1991, 312. Alternatively, the phosphonic acid is activated by reaction with triisopropylbenzenesulfonyl chloride, as described in *Nucleosides and Nucleotides*, 2000, 10, 1885. The activated product is then reacted with the aminoester **S34.9**, in the presence of a base, to give the bisamidate **S34.5**. The reaction is performed in one step, in which case the nitrogen substituents present in the product **S34.5** are the same, or in two steps, via the intermediate **S34.11**, in which case the nitrogen substituents can be different.

Examples of these methods are shown in Scheme 34, Examples 3 and 5. In the procedure illustrated in Scheme 34, Example 3, a phosphonic acid **S34.6** is reacted with ten molar equivalents of thionyl chloride, as described in *Zh. Obschei Khim.*, 1958, 28, 1063, to give the dichloro compound **S34.23**. The product is then reacted at reflux temperature in a polar aprotic solvent such as acetonitrile, and in the presence of a base such as triethylamine, with butyl serinate **S34.24** to afford the bisamidate product **S34.25**.

Using the above procedures, but employing, in place of butyl serinate **S34.24**, different aminoesters **S34.9**, the corresponding products **S34.5** are obtained.

In the procedure illustrated in Scheme 34, Example 5, the phosphonic acid **S34.6** is reacted, as described in *J. Chem. Soc. Chem. Comm.*, 1991, 312, with carbonyl diimidazole to give the imidazolide **S34.32**. The product is then

reacted in acetonitrile solution at ambient temperature, with one molar equivalent of ethyl alaninate **S34.33** to yield the monodisplacement product **S34.34**. The latter compound is then reacted with carbonyl diimidazole to produce the activated intermediate **S34.35**, and the product is then reacted, under
5 the same conditions, with ethyl N-methylalaninate **S34.33a** to give the bisamidate product **S34.36**.

Using the above procedures, but employing, in place of ethyl alaninate **S34.33** or ethyl N-methylalaninate **S34.33a**, different aminoesters **S34.9**, the corresponding products **S34.5** are obtained.

10 The intermediate monoamidate **S34.3** is also prepared from the monoester **S34.2** by first converting the monoester into the activated derivative **S34.8** in which Lv is a leaving group such as halo, imidazolyl etc, using the procedures described above. The product **S34.8** is then reacted with an aminoester **S34.9** in the presence of a base such as pyridine, to give an
15 intermediate monoamidate product **S34.3**. The latter compound is then converted, by removal of the R¹ group and coupling of the product with the aminoester **S34.9**, as described above, into the bisamidate **S34.5**.

An example of this procedure, in which the phosphonic acid is activated by conversion to the chloro derivative **S34.26**, is shown in Scheme 34, Example
20 4. In this procedure, the phosphonic monobenzyl ester **S34.15** is reacted, in dichloromethane, with thionyl chloride, as described in *Tet. Letters.*, 1994, 35, 4097, to afford the phosphoryl chloride **S34.26**. The product is then reacted in acetonitrile solution at ambient temperature with one molar equivalent of ethyl 3-amino-2-methylpropionate **S34.27** to yield the monoamidate product **S34.28**.
25 The latter compound is hydrogenated in ethylacetate over a 5% palladium on carbon catalyst to produce the monoacid product **S34.29**. The product is subjected to a Mitsunobu coupling procedure, with equimolar amounts of butyl alaninate **S34.30**, triphenyl phosphine, diethylazodicarboxylate and triethylamine in tetrahydrofuran, to give the bisamidate product **S34.31**.

30 Using the above procedures, but employing, in place of ethyl 3-amino-2-methylpropionate **S34.27** or butyl alaninate **S34.30**, different aminoesters **S34.9**, the corresponding products **S34.5** are obtained.

The activated phosphonic acid derivative **S34.7** is also converted into the bisamidate **S34.5** via the diamino compound **S34.10**. The conversion of activated phosphonic acid derivatives such as phosphoryl chlorides into the corresponding amino analogs **S34.10**, by reaction with ammonia, is described in
5 Organic Phosphorus Compounds, G. M. Kosolapoff, L. Maeir, eds, Wiley, 1976. The bisamino compound **S34.10** is then reacted at elevated temperature with a haloester **S34.12** (Hal = halogen, *i.e.* F, Cl, Br, I), in a polar organic solvent such as dimethylformamide, in the presence of a base such as 4, 4-dimethylaminopyridine (DMAP) or potassium carbonate, to yield the bisamidate
10 **S34.5**. Alternatively, **S34.6** may be treated with two different amino ester reagents simulataneously, *i.e.* **S34.12** where R^{4b} or R^{5b} are different. The resulting mixture of bisamidate products **S34.5** may then be separable, *e.g.* by chromatography.

An example of this procedure is shown in Scheme 34, Example 6. In this
15 method, a dichlorophosphonate **S34.23** is reacted with ammonia to afford the diamide **S34.37**. The reaction is performed in aqueous, aqueous alcoholic or alcoholic solution, at reflux temperature. The resulting diamino compound is then reacted with two molar equivalents of ethyl 2-bromo-3-methylbutyrate **S34.38**, in a polar organic solvent such as N-methylpyrrolidinone at ca. 150 °C,
20 in the presence of a base such as potassium carbonate, and optionally in the presence of a catalytic amount of potassium iodide, to afford the bisamidate product **S34.39**.

Using the above procedures, but employing, in place of ethyl 2-bromo-3-methylbutyrate **S34.38**, different haloesters **S34.12** the corresponding products
25 **S34.5** are obtained.

The procedures shown in Scheme 34 are also applicable to the preparation of bisamidates in which the aminoester moiety incorporates different functional groups. Scheme 34, Example 7 illustrates the preparation of bisamidates derived from tyrosine. In this procedure, the monoimidazolidine
30 **S34.32** is reacted with propyl tyrosinate **S34.40**, as described in Example 5, to yield the monoamidate **S34.41**. The product is reacted with carbonyl diimidazole to give the imidazolidine **S34.42**, and this material is reacted with a further molar equivalent of propyl tyrosinate to produce the bisamidate product **S34.43**.

Using the above procedures, but employing, in place of propyl tyrosinate S34.40, different aminoesters S34.9, the corresponding products S34.5 are obtained. The aminoesters employed in the two stages of the above procedure can be the same or different, so that bisamidates with the same or different
5 amino substituents are prepared.

Scheme 35 illustrates methods for the preparation of phosphonate monoamidates.

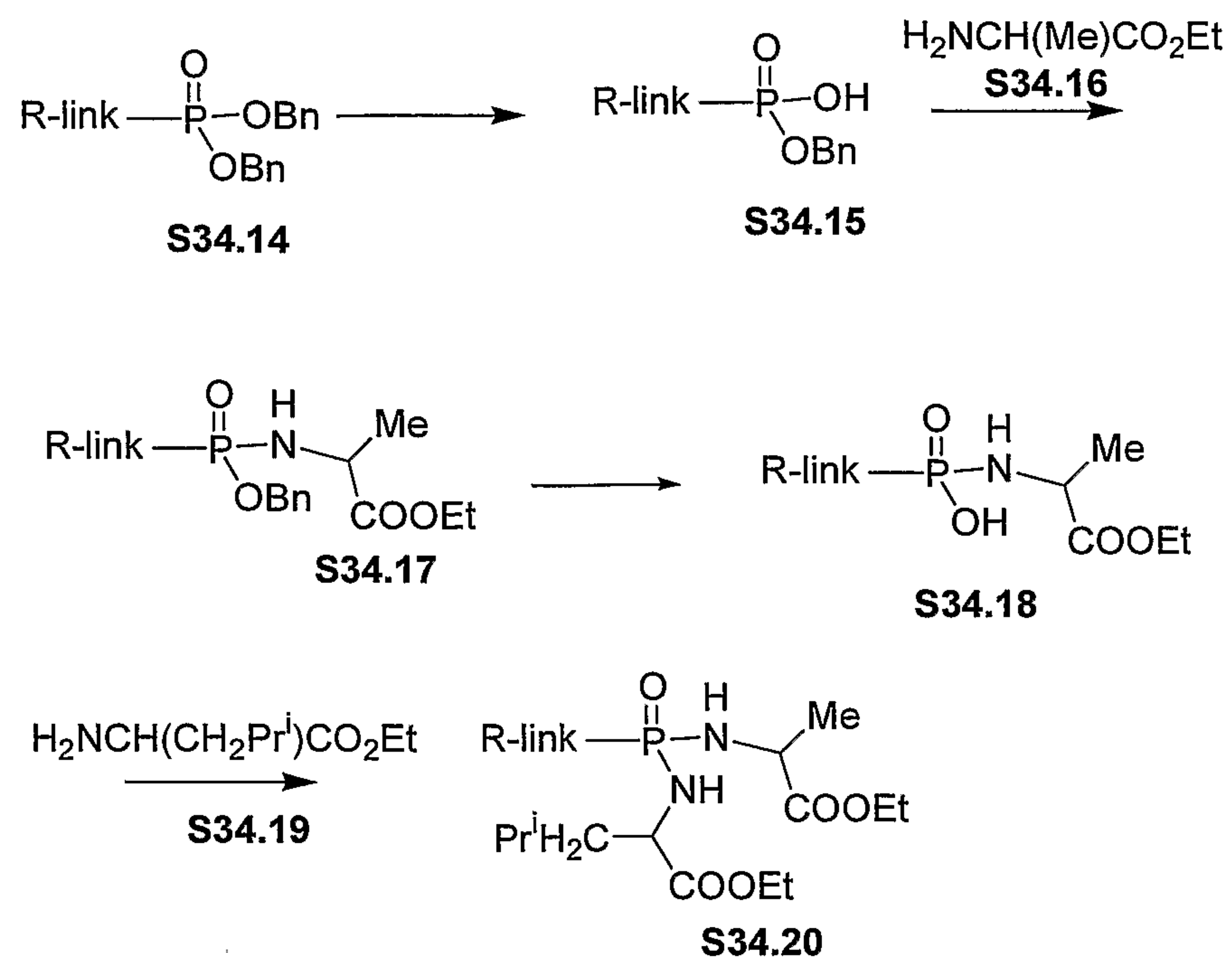
In one procedure, a phosphonate monoester S34.1 is converted, as described in Scheme 34, into the activated derivative S34.8. This compound is
10 then reacted, as described above, with an aminoester S34.9, in the presence of a base, to afford the monoamidate product S35.1.

The procedure is illustrated in Scheme 35, Example 1. In this method, a monophenyl phosphonate S35.7 is reacted with, for example, thionyl chloride, as described in *J. Gen. Chem. USSR.*, 1983, 32, 367, to give the chloro product
15 S35.8. The product is then reacted, as described in Scheme 34, with ethyl alaninate, to yield the amidate S35.10.

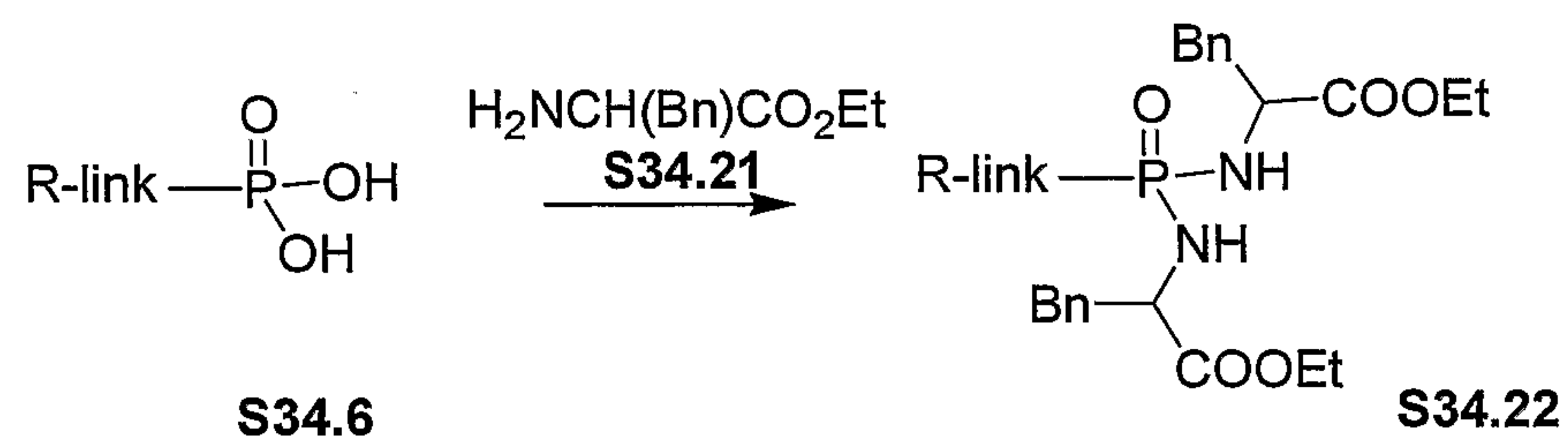
Using the above procedures, but employing, in place of ethyl alaninate S35.9, different aminoesters S34.9, the corresponding products S35.1 are obtained.

20 Alternatively, the phosphonate monoester S34.1 is coupled, as described in Scheme 34, with an aminoester S34.9 to produce the amidate S35.1. If necessary, the R¹ substituent is then altered, by initial cleavage to afford the phosphonic acid S35.2. The procedures for this transformation depend on the nature of the R¹ group, and are described above. The phosphonic acid is then
25 transformed into the ester amidate product S35.3, by reaction with the hydroxy compound R³OH, in which the group R³ is aryl, heterocycle, alkyl, cycloalkyl, haloalkyl etc, using the same coupling procedures (carbodiimide, Aldrichiol-2, PYBOP, Mitsunobu reaction etc) described in Scheme 34 for the coupling of amines and phosphonic acids.

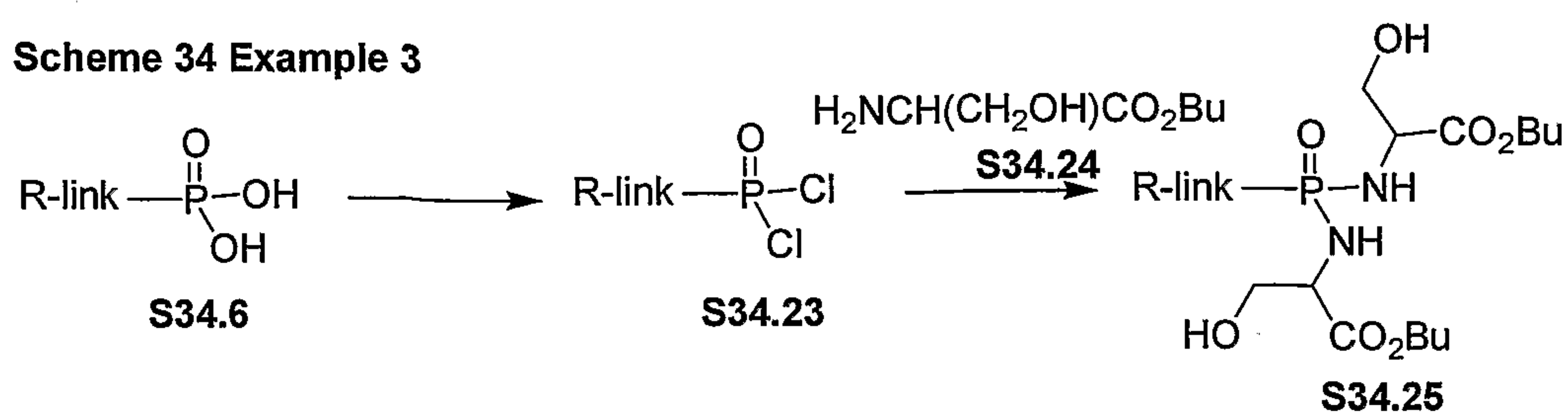
Scheme 34 Example 1



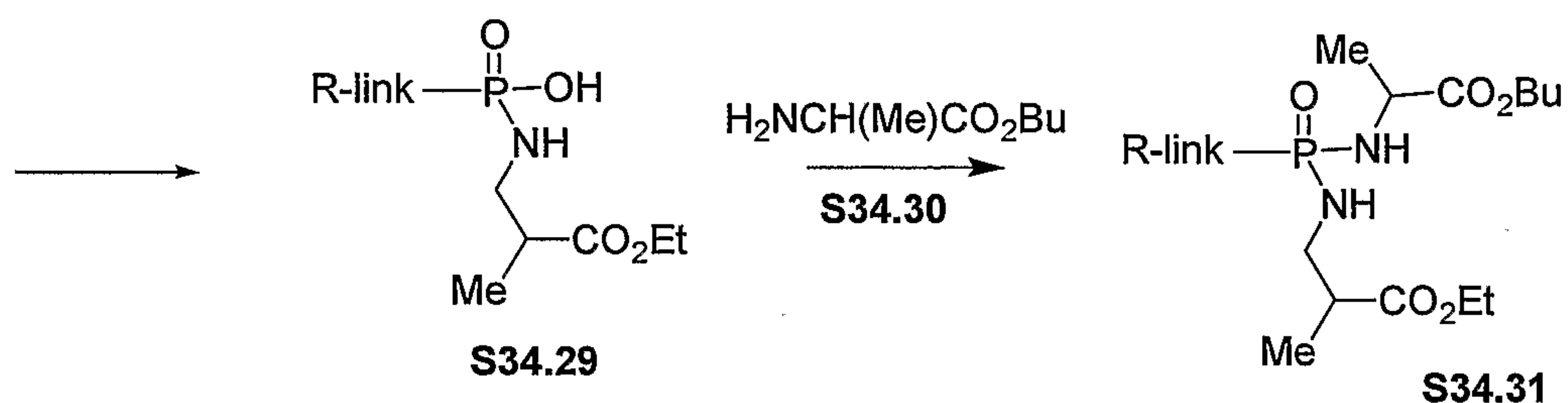
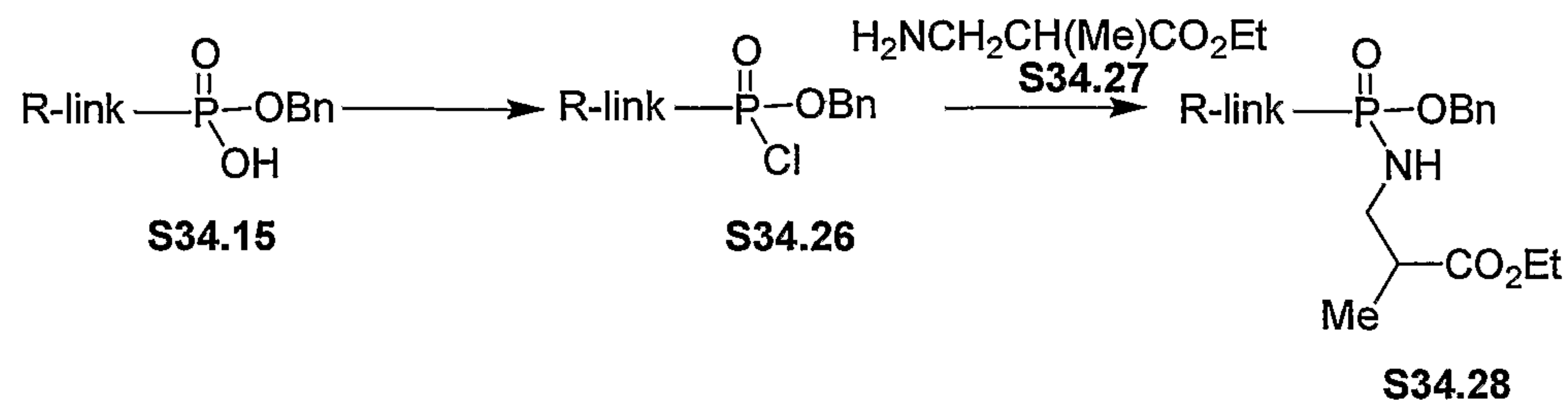
Scheme 34 Example 2



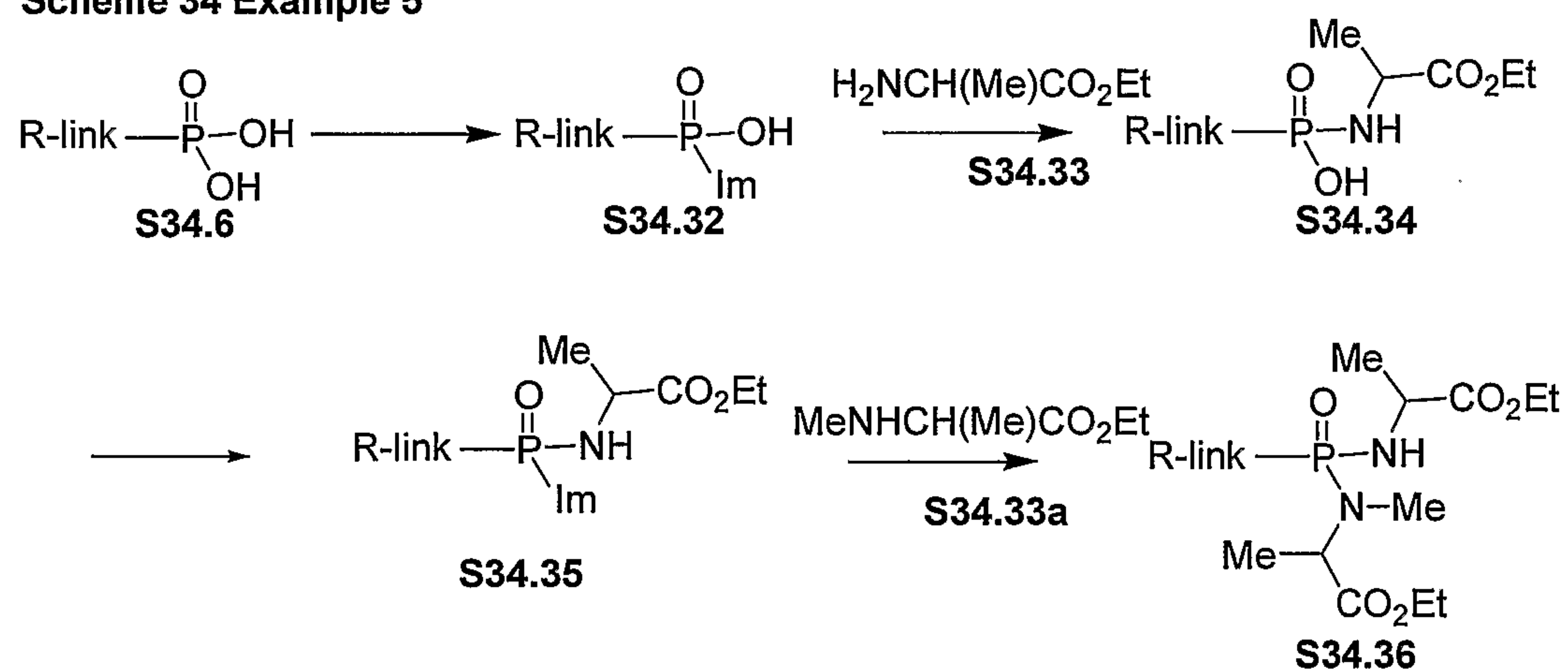
Scheme 34 Example 3



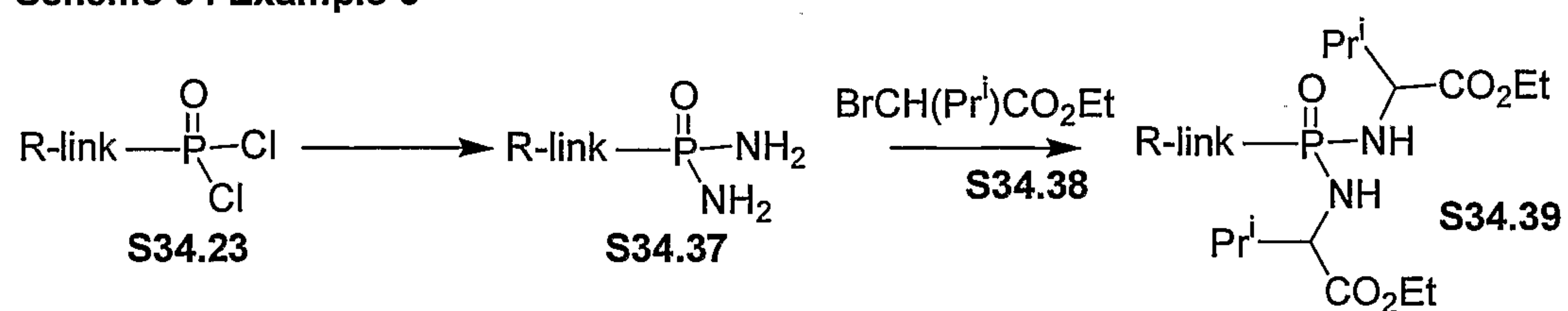
Scheme 34 Example 4



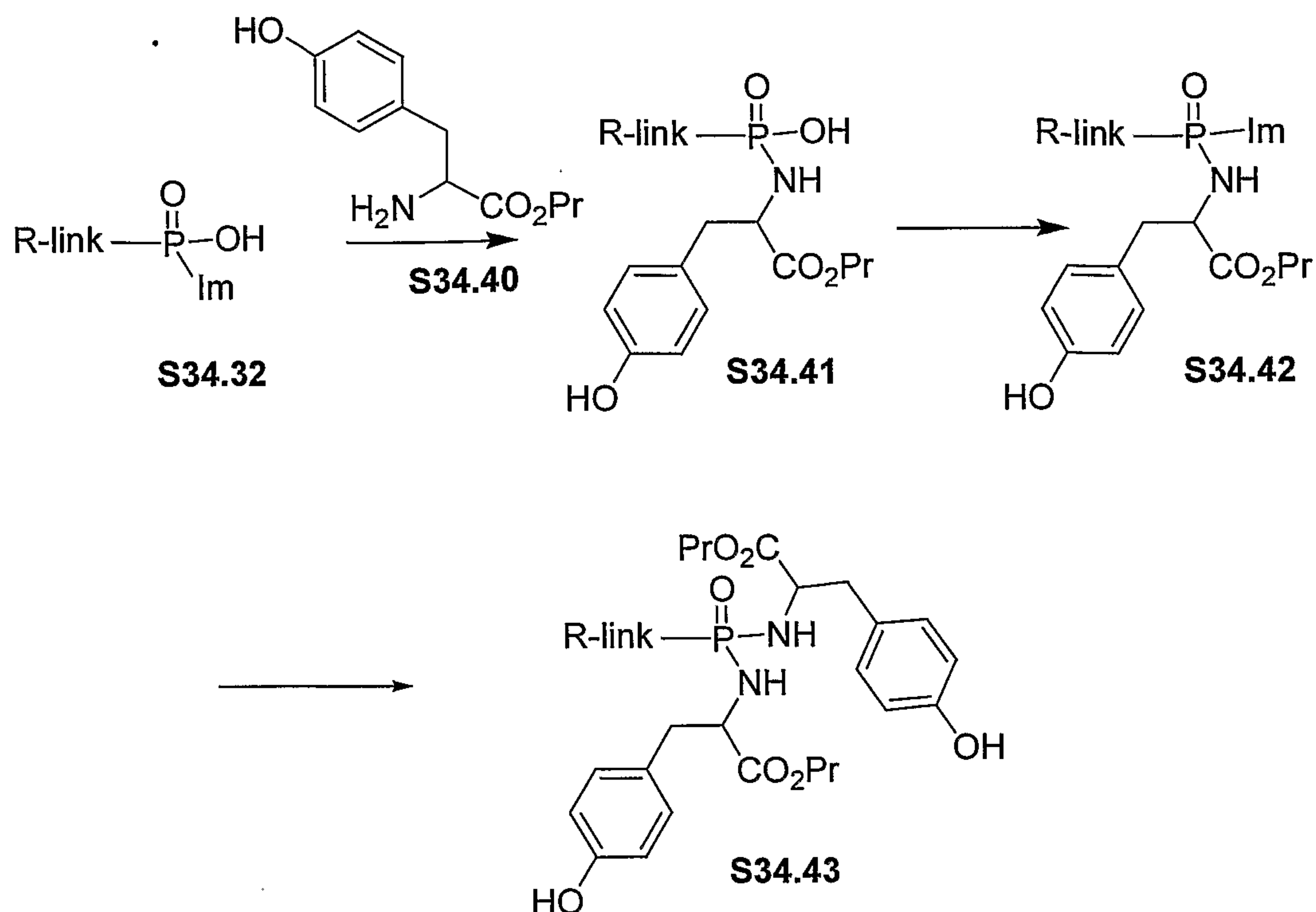
Scheme 34 Example 5



Scheme 34 Example 6



Scheme 34 Example 7



Examples of this method are shown in Scheme 35, Examples 1-3. In the sequence shown in Example 2, a monobenzyl phosphonate **S35.11** is transformed by reaction with ethyl alaninate, using one of the methods described above, into the monoamidate **S35.12**. The benzyl group is then removed by catalytic hydrogenation in ethylacetate solution over a 5% palladium on carbon catalyst, to afford the phosphonic acid amidate **S35.13**. The product is then reacted in dichloromethane solution at ambient temperature with equimolar amounts of 1-(dimethylaminopropyl)-3-ethylcarbodiimide and trifluoroethanol **S35.14**, for example as described in *Tet. Lett.*, 2001, 42, 8841, to yield the amidate ester **S35.15**.

In the sequence shown in Scheme 35, Example 3, the monoamidate **S35.13** is coupled, in tetrahydrofuran solution at ambient temperature, with equimolar amounts of dicyclohexyl carbodiimide and 4-hydroxy-N-methylpiperidine **S35.16**, to produce the amidate ester product **S35.17**.

Using the above procedures, but employing, in place of the ethyl alaninate product **S35.12** different monoacids **S35.2**, and in place of

trifluoroethanol **S35.14** or 4-hydroxy-N-methylpiperidine **S35.16**, different hydroxy compounds R^3OH , the corresponding products **S35.3** are obtained.

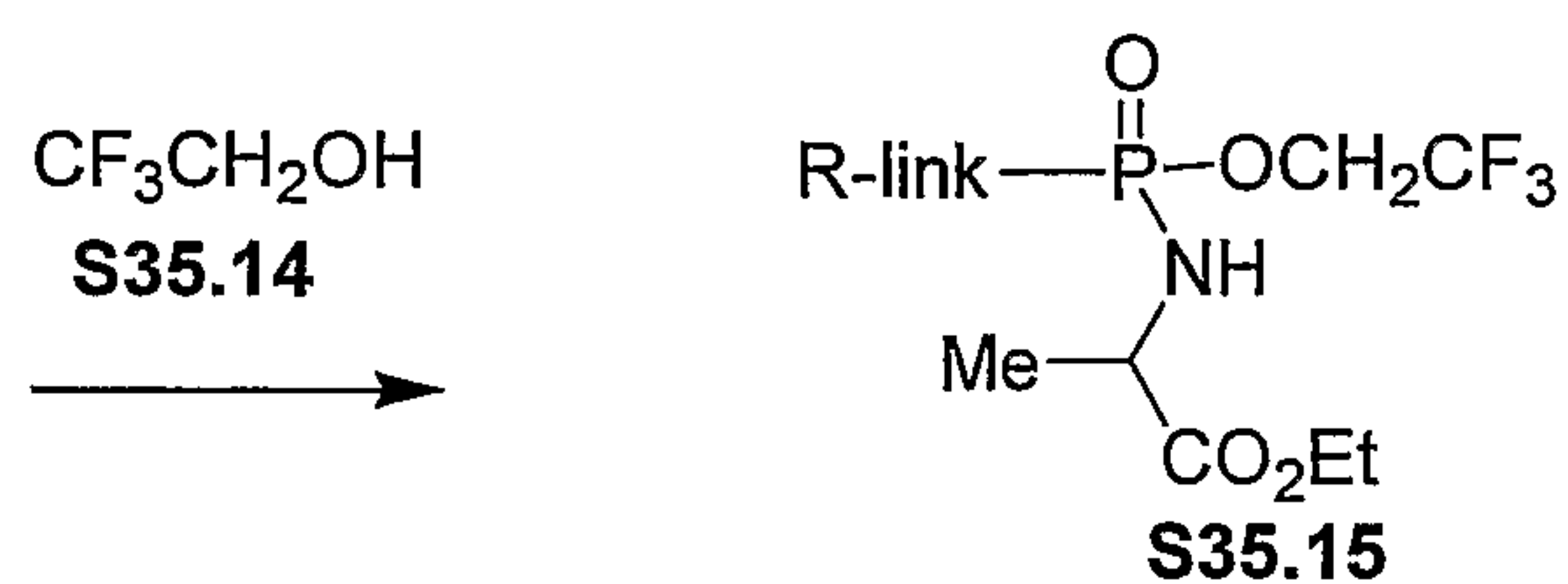
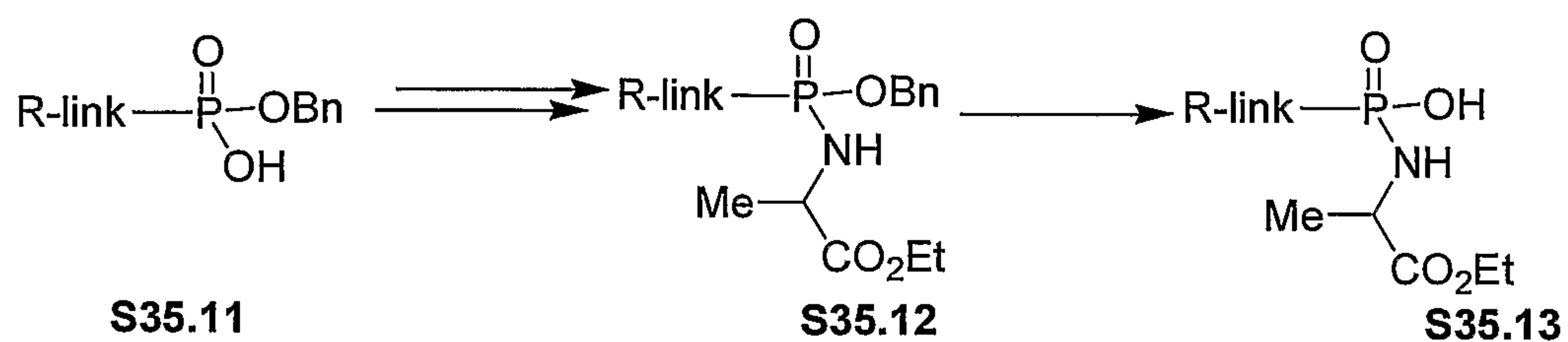
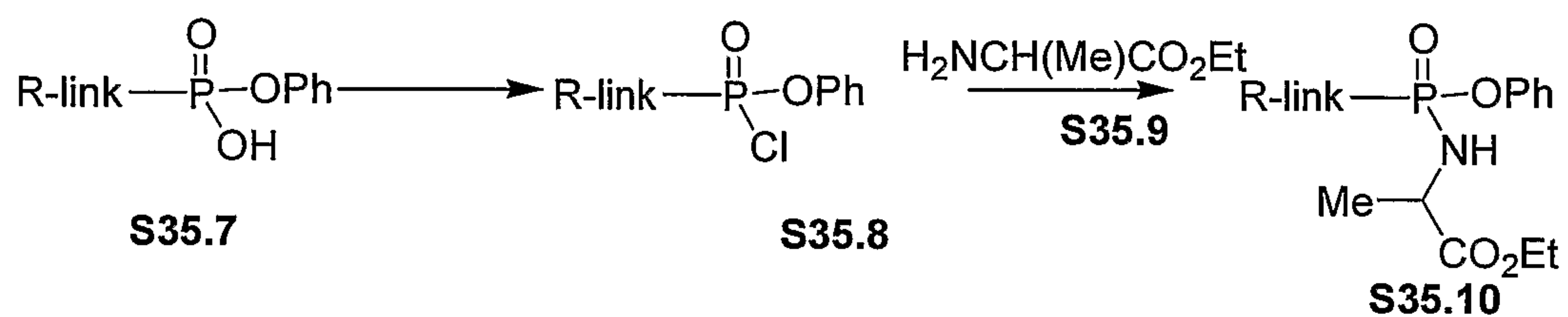
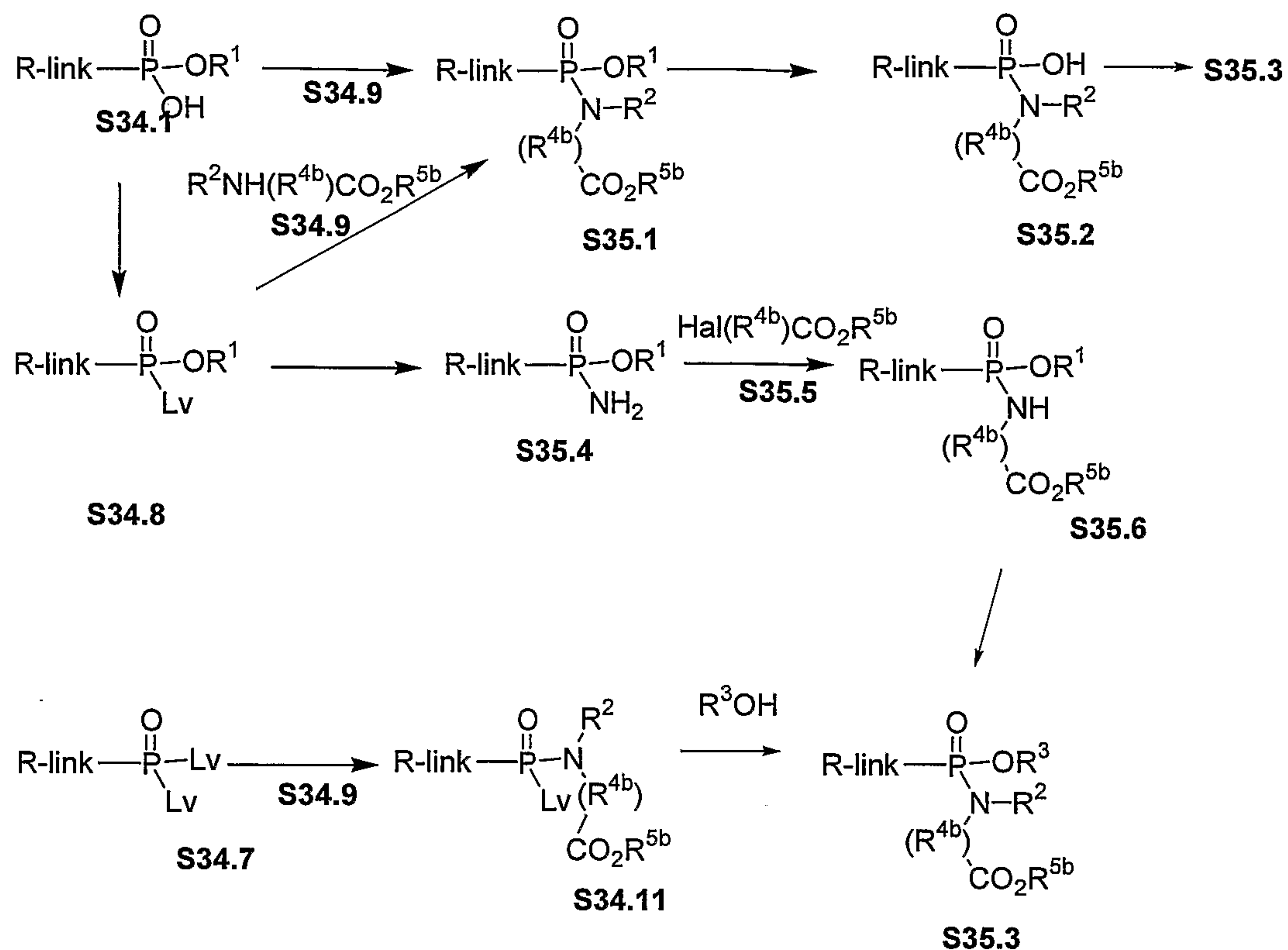
Alternatively, the activated phosphonate ester **S34.8** is reacted with ammonia to yield the amidate **S35.4**. The product is then reacted, as described in
5 Scheme 34, with a haloester **S35.5**, in the presence of a base, to produce the amidate product **S35.6**. If appropriate, the nature of the R^1 group is changed, using the procedures described above, to give the product **S35.3**. The method is illustrated in Scheme 35, Example 4. In this sequence, the monophenyl phosphoryl chloride **S35.18** is reacted, as described in Scheme 34, with
10 ammonia, to yield the amino product **S35.19**. This material is then reacted in N-methylpyrrolidinone solution at 170 °C with butyl 2-bromo-3-phenylpropionate **S35.20** and potassium carbonate, to afford the amidate product **S35.21**.

Using these procedures, but employing, in place of butyl 2-bromo-3-phenylpropionate **S35.20**, different haloesters **S35.5**, the corresponding products
15 **S35.6** are obtained.

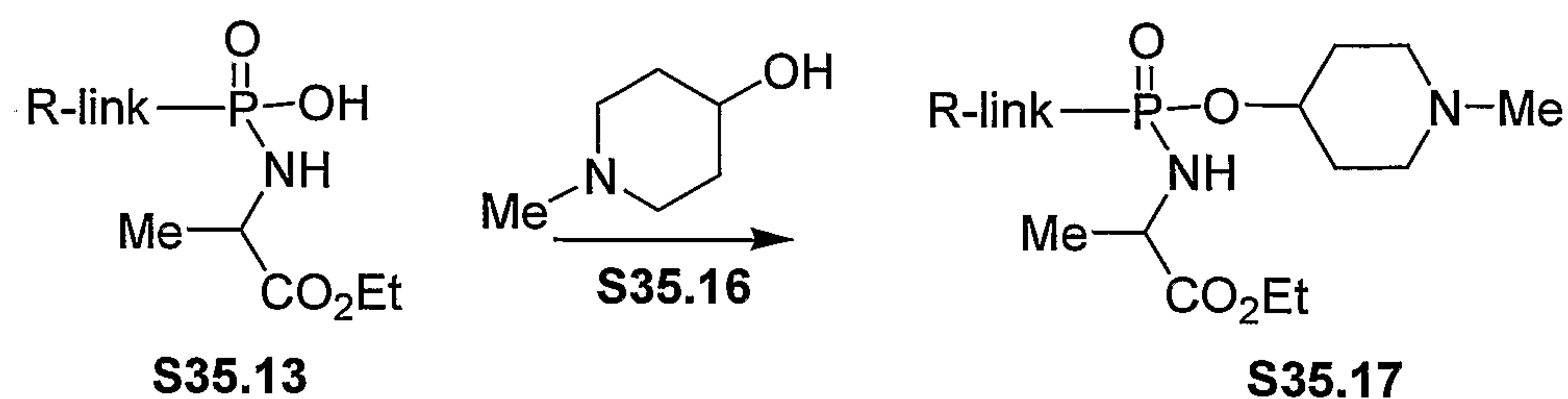
The monoamidate products **S35.3** are also prepared from the doubly activated phosphonate derivatives **S34.7**. In this procedure, examples of which are described in *Synlett.*, 1998, 1, 73, the intermediate **S34.7** is reacted with a limited amount of the aminoester **S34.9** to give the mono-displacement product
20 **S34.11**. The latter compound is then reacted with the hydroxy compound R^3OH in a polar organic solvent such as dimethylformamide, in the presence of a base such as diisopropylethylamine, to yield the monoamidate ester **S35.3**.

The method is illustrated in Scheme 35, Example 5. In this method, the phosphoryl dichloride **S35.22** is reacted in dichloromethane solution with one
25 molar equivalent of ethyl N-methyl tyrosinate **S35.23** and dimethylaminopyridine, to generate the monoamidate **S35.24**. The product is then reacted with phenol **S35.25** in dimethylformamide containing potassium carbonate, to yield the ester amidate product **S35.26**.

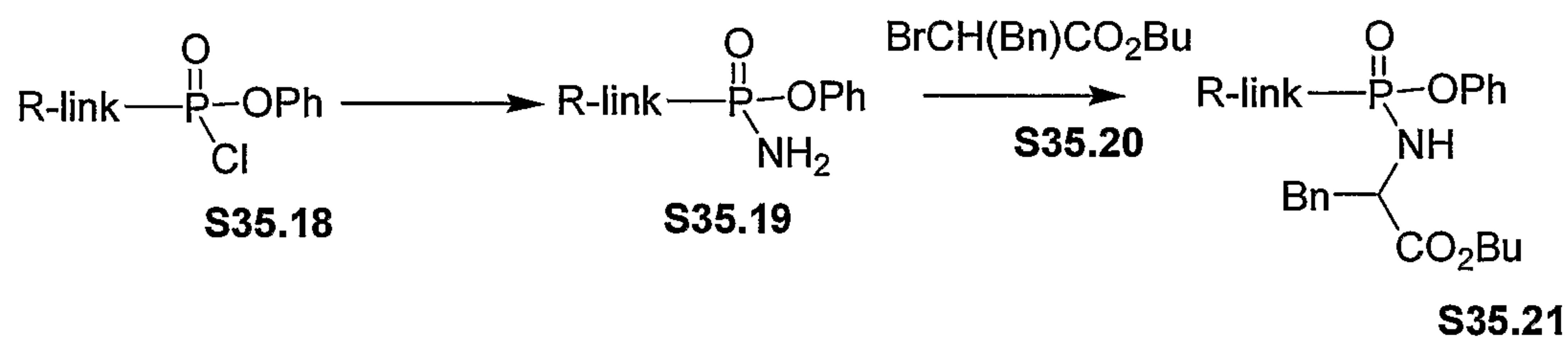
Using these procedures, but employing, in place of ethyl N-methyl tyrosinate **S35.23** or phenol **S35.25**, the aminoesters **S34.9** and/or the hydroxy
30 compounds R^3OH , the corresponding products **S35.3** are obtained.



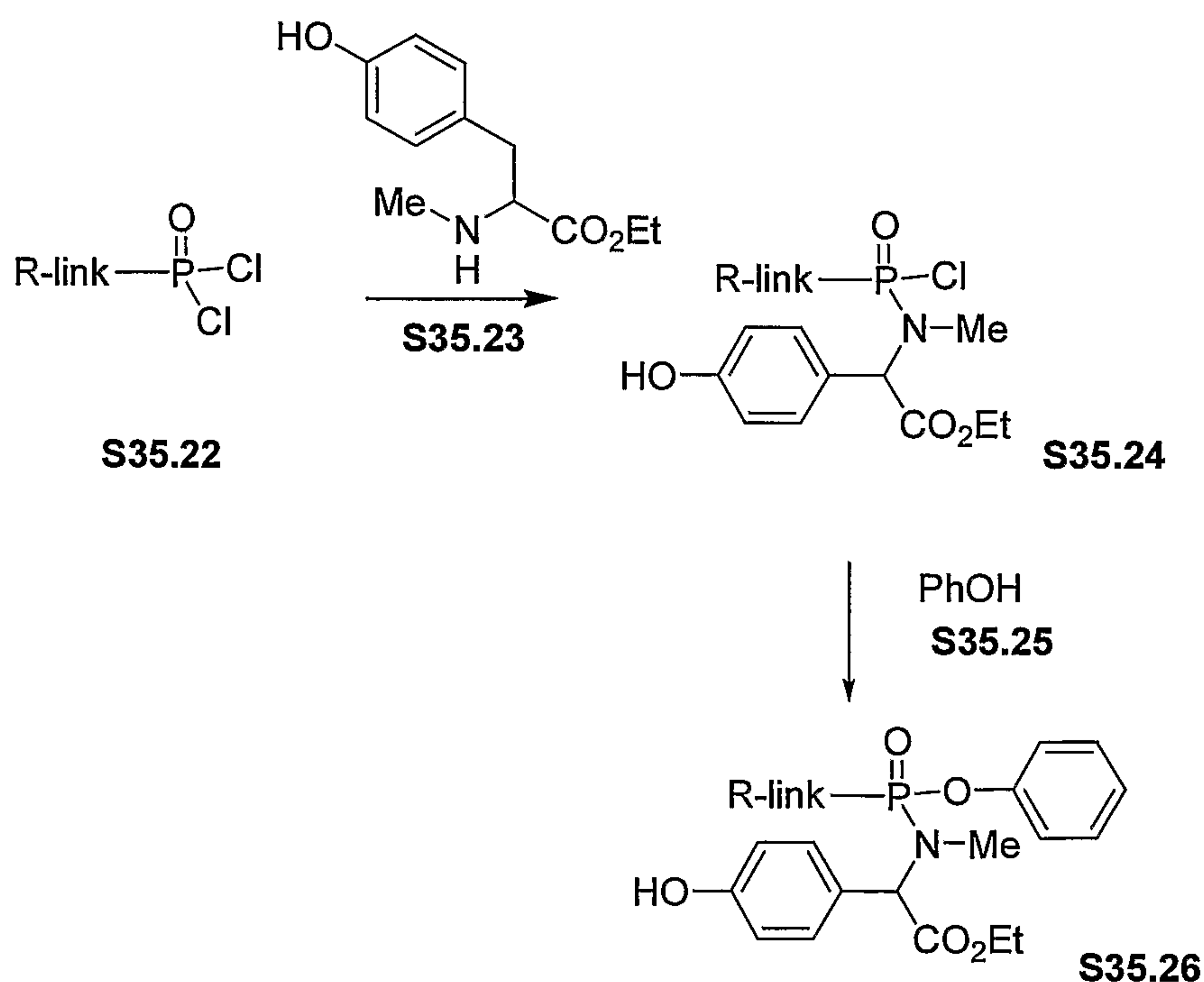
Scheme 35 Example 3



Scheme 35 Example 4



Scheme 35 Example 5



5 Scheme 36 illustrates methods for the preparation of carboalkoxy-substituted phosphonate diesters in which one of the ester groups incorporates a carboalkoxy substituent.

In one procedure, a phosphonate monoester **S34.1**, prepared as described above, is coupled, using one of the methods described above, with a hydroxyester **S36.1**, in which the groups R^{4b} and R^{5b} are as described in Scheme 34. For example, equimolar amounts of the reactants are coupled in the presence of a carbodiimide such as dicyclohexyl carbodiimide, as described in *Aust. J. Chem.*, 1963, 609, optionally in the presence of dimethylaminopyridine, as described in *Tet.*, 1999, 55, 12997. The reaction is conducted in an inert solvent at ambient temperature.

The procedure is illustrated in Scheme 36, Example 1. In this method, a monophenyl phosphonate **S36.9** is coupled, in dichloromethane solution in the presence of dicyclohexyl carbodiimide, with ethyl 3-hydroxy-2-methylpropionate **S36.10** to yield the phosphonate mixed diester **S36.11**.

Using this procedure, but employing, in place of ethyl 3-hydroxy-2-methylpropionate **S36.10**, different hydroxyesters **S33.1**, the corresponding products **S33.2** are obtained.

The conversion of a phosphonate monoester **S34.1** into a mixed diester **S36.2** is also accomplished by means of a Mitsunobu coupling reaction with the hydroxyester **S36.1**, as described in *Org. Lett.*, 2001, 643. In this method, the reactants **S34.1** and **S36.1** are combined in a polar solvent such as tetrahydrofuran, in the presence of a triarylphosphine and a dialkyl azodicarboxylate, to give the mixed diester **S36.2**. The R^1 substituent is varied by cleavage, using the methods described previously, to afford the monoacid product **S36.3**. The product is then coupled, for example using methods described above, with the hydroxy compound R^3OH , to give the diester product **S36.4**.

The procedure is illustrated in Scheme 36, Example 2. In this method, a monoallyl phosphonate **S36.12** is coupled in tetrahydrofuran solution, in the presence of triphenylphosphine and diethylazodicarboxylate, with ethyl lactate **S36.13** to give the mixed diester **S36.14**. The product is reacted with tris(triphenylphosphine) rhodium chloride (Wilkinson catalyst) in acetonitrile, as described previously, to remove the allyl group and produce the monoacid product **S36.15**. The latter compound is then coupled, in pyridine solution at

ambient temperature, in the presence of dicyclohexyl carbodiimide, with one molar equivalent of 3-hydroxypyridine **S36.16** to yield the mixed diester **S36.17**.

Using the above procedures, but employing, in place of the ethyl lactate **S36.13** or 3-hydroxypyridine, a different hydroxyester **S36.1** and/or a different
5 hydroxy compound R^3OH , the corresponding products **S36.4** are obtained.

The mixed diesters **S36.2** are also obtained from the monoesters **S34.1** via the intermediacy of the activated monoesters **S36.5**. In this procedure, the monoester **S34.1** is converted into the activated compound **S36.5** by reaction with, for example, phosphorus pentachloride, as described in *J. Org. Chem.*,
10 2001, 66, 329, or with thionyl chloride or oxalyl chloride ($Lv = Cl$), or with triisopropylbenzenesulfonyl chloride in pyridine, as described in *Nucleosides and Nucleotides*, 2000, 19, 1885, or with carbonyl diimidazole, as described in *J. Med. Chem.*, 2002, 45, 1284. The resultant activated monoester is then reacted with the hydroxyester **S36.1**, as described above, to yield the mixed diester
15 **S36.2**.

The procedure is illustrated in Scheme 36, Example 3. In this sequence, a monophenyl phosphonate **S36.9** is reacted, in acetonitrile solution at 70 °C, with ten equivalents of thionyl chloride, so as to produce the phosphoryl chloride **S36.19**. The product is then reacted with ethyl 4-carbamoyl-2-hydroxybutyrate
20 **S36.20** in dichloromethane containing triethylamine, to give the mixed diester **S36.21**.

Using the above procedures, but employing, in place of ethyl 4-carbamoyl-2-hydroxybutyrate **S36.20**, different hydroxyesters **S36.1**, the corresponding products **S36.2** are obtained.

25 The mixed phosphonate diesters are also obtained by an alternative route for incorporation of the R^3O group into intermediates **S36.3** in which the hydroxyester moiety is already incorporated. In this procedure, the monoacid intermediate **S36.3** is converted into the activated derivative **S36.6** in which Lv is a leaving group such as chloro, imidazole, and the like, as previously
30 described. The activated intermediate is then reacted with the hydroxy compound R^3OH , in the presence of a base, to yield the mixed diester product **S36.4**.

The method is illustrated in Scheme 36, Example 4. In this sequence, the phosphonate monoacid **S36.22** is reacted with trichloromethanesulfonyl chloride in tetrahydrofuran containing collidine, as described in *J. Med. Chem.*, 1995, 38, 4648, to produce the trichloromethanesulfonyloxy product **S36.23**. This
5 compound is reacted with 3-(morpholinomethyl)phenol **S36.24** in dichloromethane containing triethylamine, to yield the mixed diester product **S36.25**.

Using the above procedures, but employing, in place of with 3-(morpholinomethyl)phenol **S36.24**, different alcohols R^3OH , the corresponding
10 products **S36.4** are obtained.

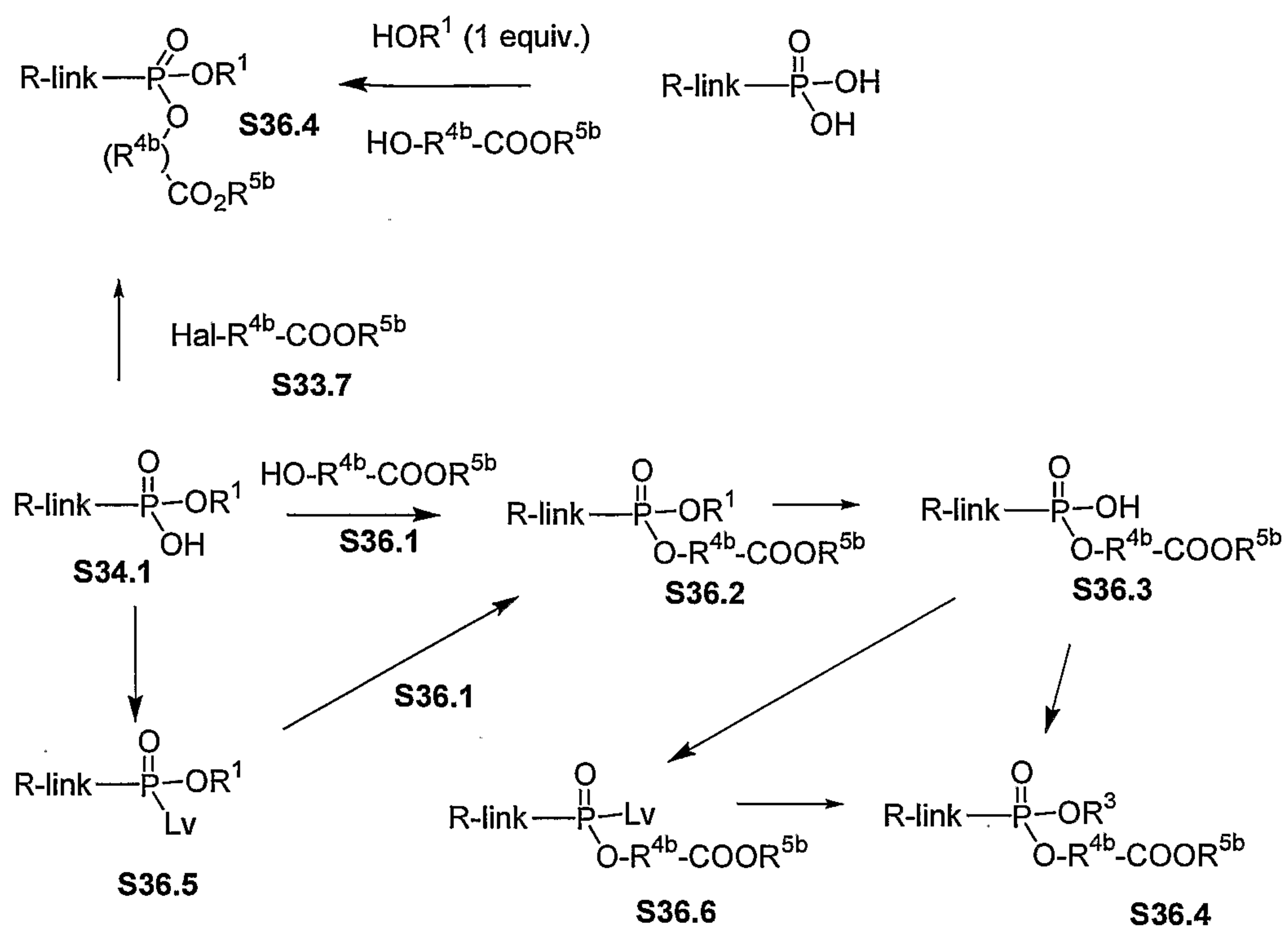
The phosphonate esters **S36.4** are also obtained by means of alkylation reactions performed on the monoesters **S34.1**. The reaction between the monoacid **S34.1** and the haloester **S36.7** is performed in a polar solvent in the presence of a base such as diisopropylethylamine, as described in *Anal. Chem.*,
15 1987, 59, 1056, or triethylamine, as described in *J. Med. Chem.*, 1995, 38, 1372, or in a non-polar solvent such as benzene, in the presence of 18-crown-6, as described in *Syn. Comm.*, 1995, 25, 3565.

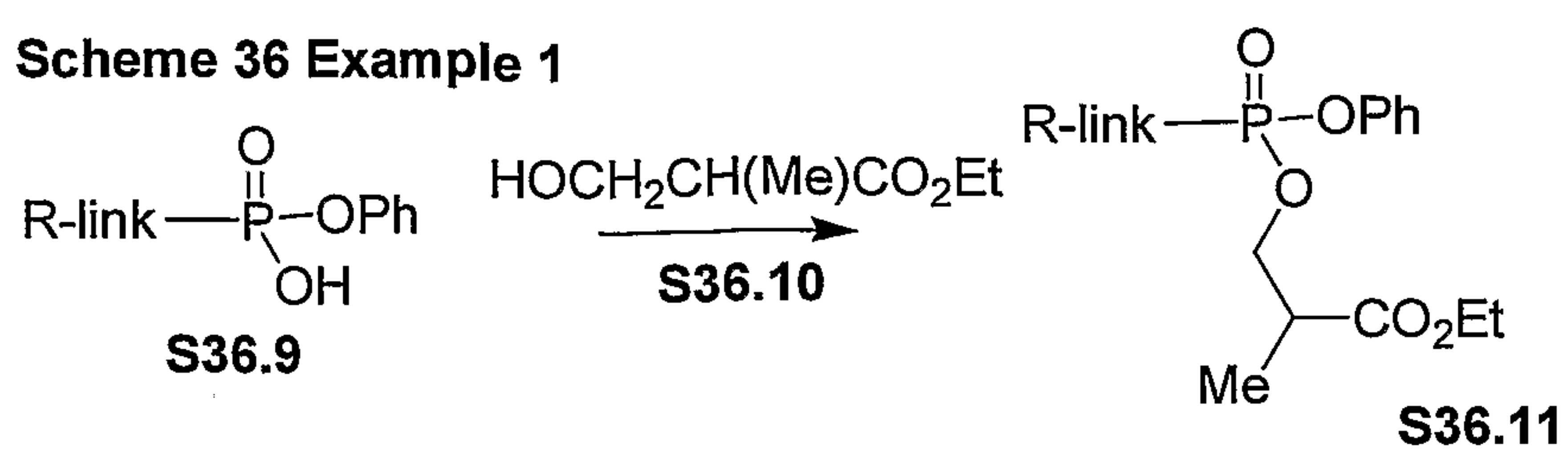
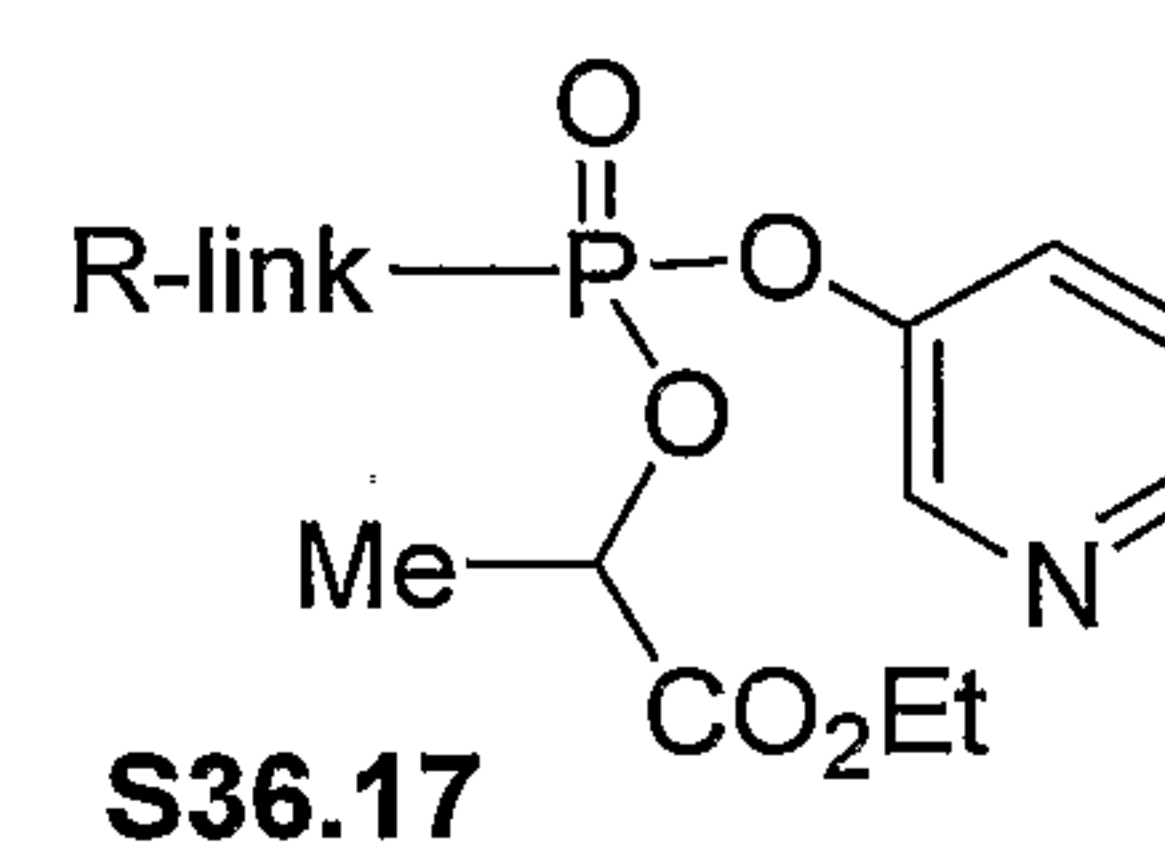
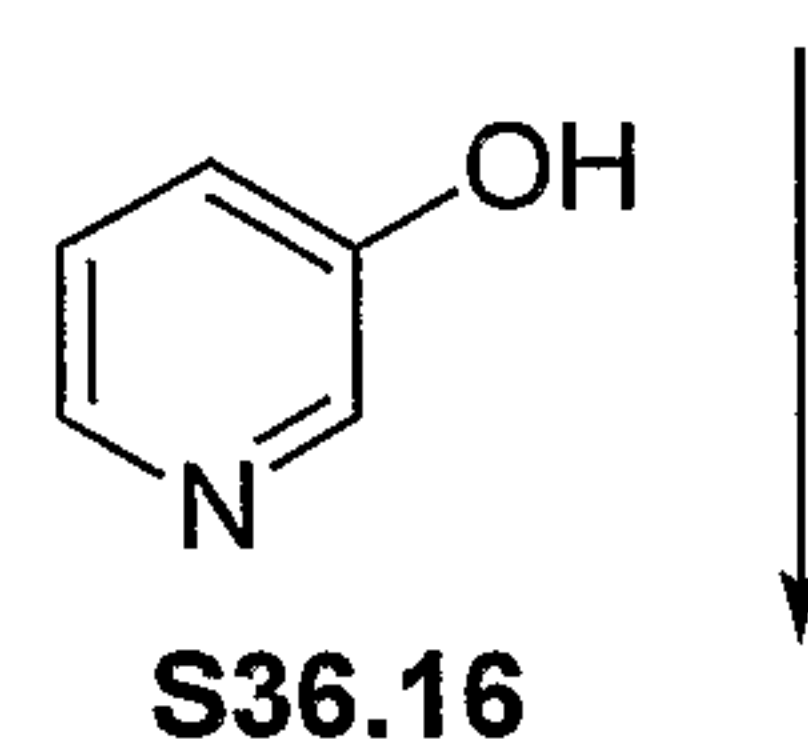
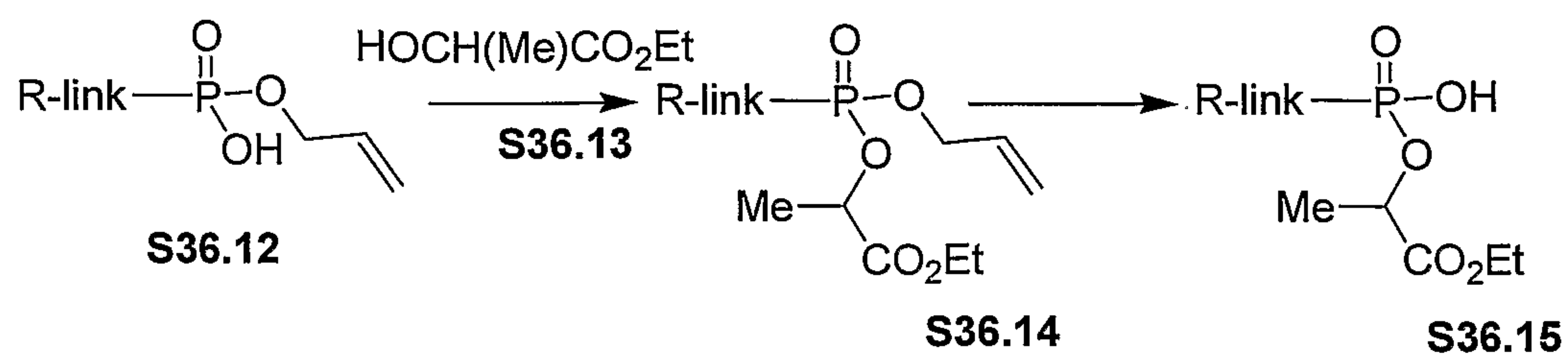
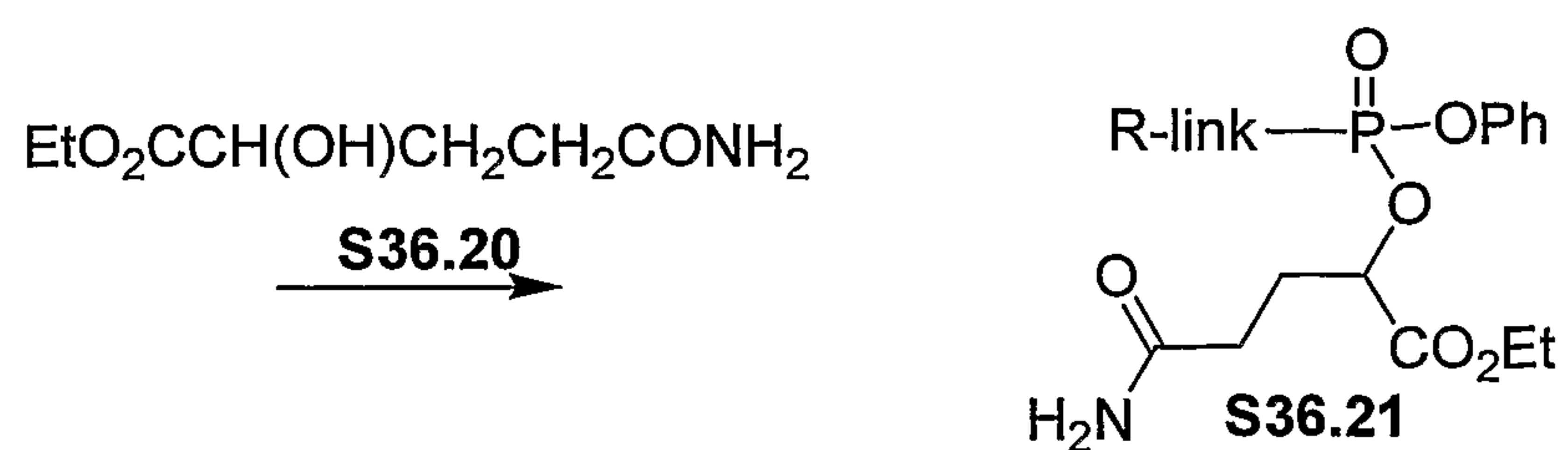
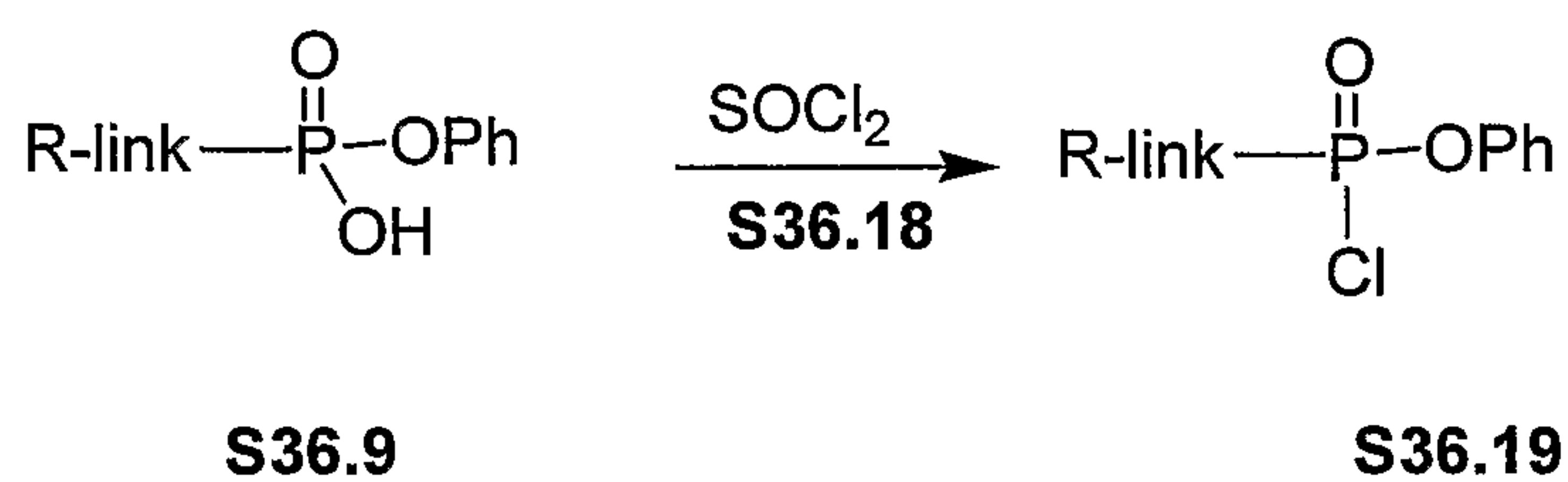
The method is illustrated in Scheme 36, Example 5. In this procedure, the monoacid **S36.26** is reacted with ethyl 2-bromo-3-phenylpropionate **S36.27** and
20 diisopropylethylamine in dimethylformamide at 80 °C to afford the mixed diester product **S36.28**.

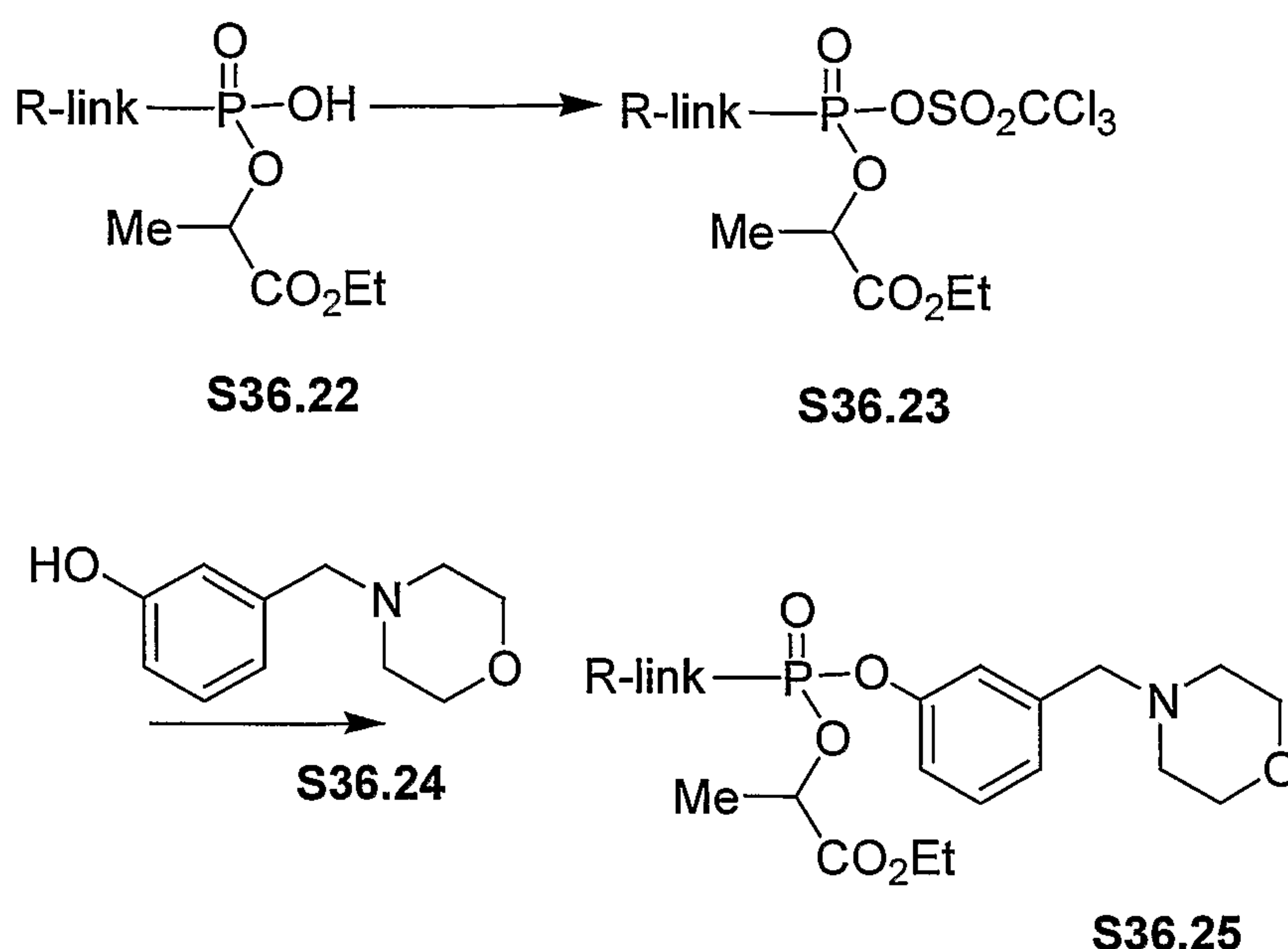
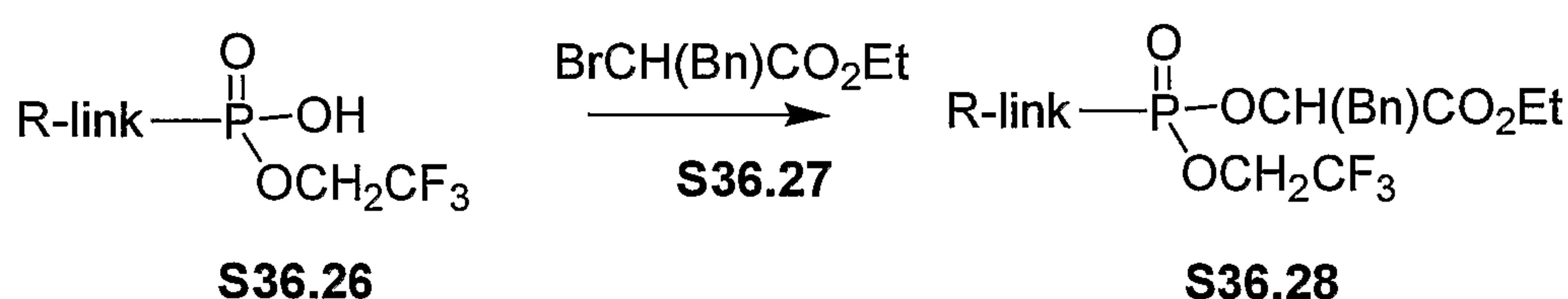
Using the above procedure, but employing, in place of ethyl 2-bromo-3-phenylpropionate **S36.27**, different haloesters **S36.7**, the corresponding products **S36.4** are obtained.

25

Scheme 36



Scheme 36 Example 1**Scheme 36 Example 2****Scheme 36 Example 3**

Scheme 36 Example 4**Scheme 36 Example 5**

Scheme 37 illustrates methods for the preparation of phosphonate diesters in which both the ester substituents incorporate carboalkoxy groups.

The compounds are prepared directly or indirectly from the phosphonic acids **S34.6**. In one alternative, the phosphonic acid is coupled with the hydroxyester **S37.2**, using the conditions described previously in Schemes 34-36, such as coupling reactions using dicyclohexyl carbodiimide or similar reagents, or under the conditions of the Mitsunobu reaction, to afford the diester product **S37.3** in which the ester substituents are identical.

This method is illustrated in Scheme 37, Example 1. In this procedure, the phosphonic acid **S34.6** is reacted with three molar equivalents of butyl lactate **S37.5** in the presence of Aldrithiol-2 and triphenyl phosphine in pyridine at ca. 70 °C, to afford the diester **S37.6**.

Using the above procedure, but employing, in place of butyl lactate **S37.5**, different hydroxyesters **S37.2**, the corresponding products **S37.3** are obtained.

Alternatively, the diesters **S37.3** are obtained by alkylation of the
5 phosphonic acid **S34.6** with a haloester **S37.1**. The alkylation reaction is performed as described in Scheme 36 for the preparation of the esters **S36.4**.

This method is illustrated in Scheme 37, Example 2. In this procedure, the phosphonic acid **S34.6** is reacted with excess ethyl 3-bromo-2-methylpropionate **S37.7** and diisopropylethylamine in dimethylformamide at ca.
10 80 °C, as described in *Anal. Chem.*, 1987, 59, 1056, to produce the diester **S37.8**.

Using the above procedure, but employing, in place of ethyl 3-bromo-2-methylpropionate **S37.7**, different haloesters **S37.1**, the corresponding products **S37.3** are obtained.

The diesters **S37.3** are also obtained by displacement reactions of
15 activated derivatives **S34.7** of the phosphonic acid with the hydroxyesters **S37.2**. The displacement reaction is performed in a polar solvent in the presence of a suitable base, as described in Scheme 36. The displacement reaction is performed in the presence of an excess of the hydroxyester, to afford the diester product **S37.3** in which the ester substituents are identical, or sequentially with
20 limited amounts of different hydroxyesters, to prepare diesters **S37.3** in which the ester substituents are different.

The methods are illustrated in Scheme 37, Examples 3 and 4. As shown in Example 3, the phosphoryl dichloride **S35.22** is reacted with three molar equivalents of ethyl 3-hydroxy-2-(hydroxymethyl)propionate **S37.9** in
25 tetrahydrofuran containing potassium carbonate, to obtain the diester product **S37.10**.

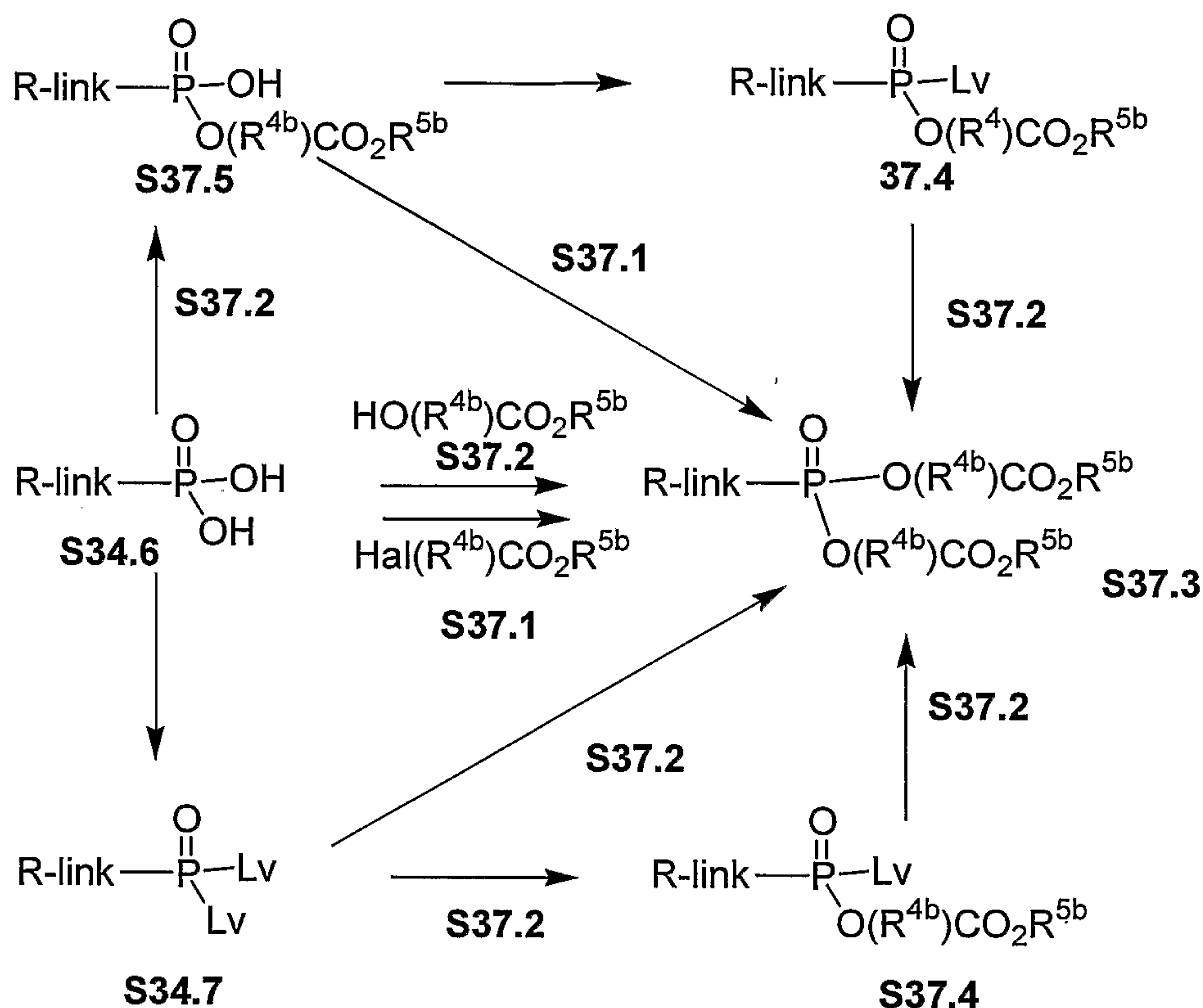
Using the above procedure, but employing, in place of ethyl 3-hydroxy-2-(hydroxymethyl)propionate **S37.9**, different hydroxyesters **S37.2**, the corresponding products **S37.3** are obtained.

30 Scheme 37, Example 4 depicts the displacement reaction between equimolar amounts of the phosphoryl dichloride **S35.22** and ethyl 2-methyl-3-hydroxypropionate **S37.11**, to yield the monoester product **S37.12**. The reaction is conducted in acetonitrile at 70° in the presence of diisopropylethylamine. The

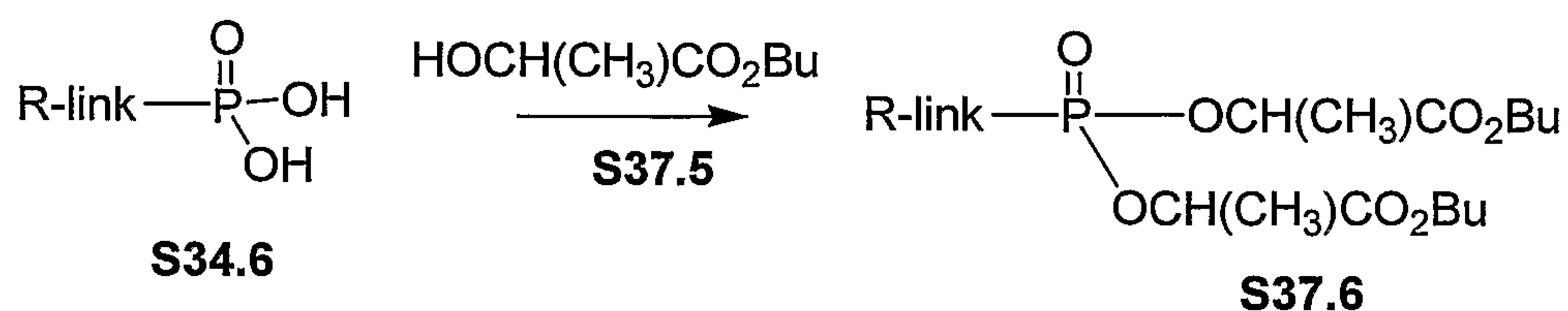
product **S37.12** is then reacted, under the same conditions, with one molar equivalent of ethyl lactate **S37.13**, to give the diester product **S37.14**.

Using the above procedures, but employing, in place of ethyl 2-methyl-3-hydroxypropionate **S37.11** and ethyl lactate **S37.13**, sequential reactions with
5 different hydroxyesters **S37.2**, the corresponding products **S37.3** are obtained.

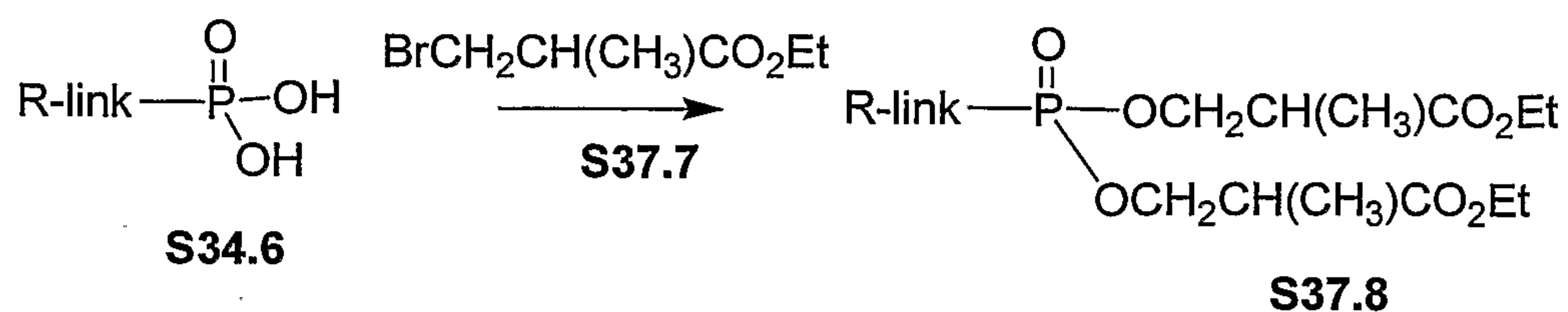
Scheme 37



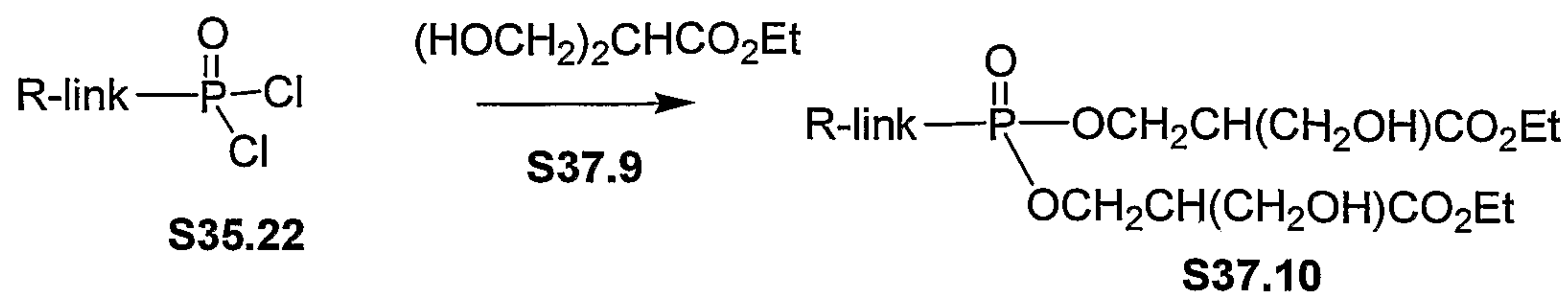
Scheme 37 Example 1



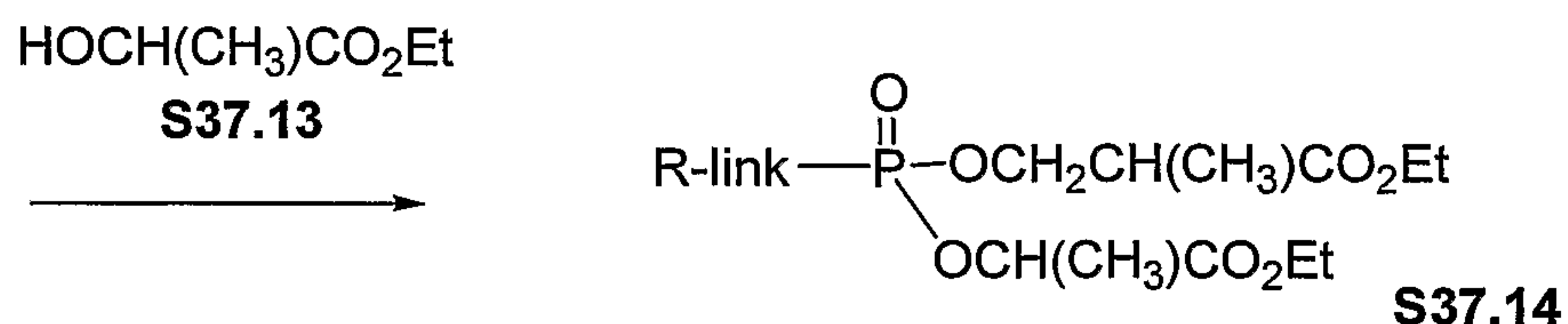
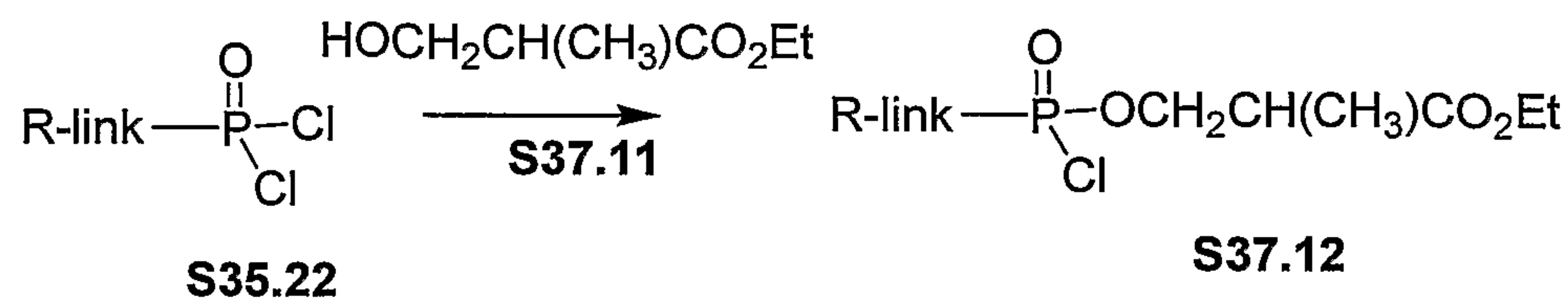
Scheme 37 Example 2



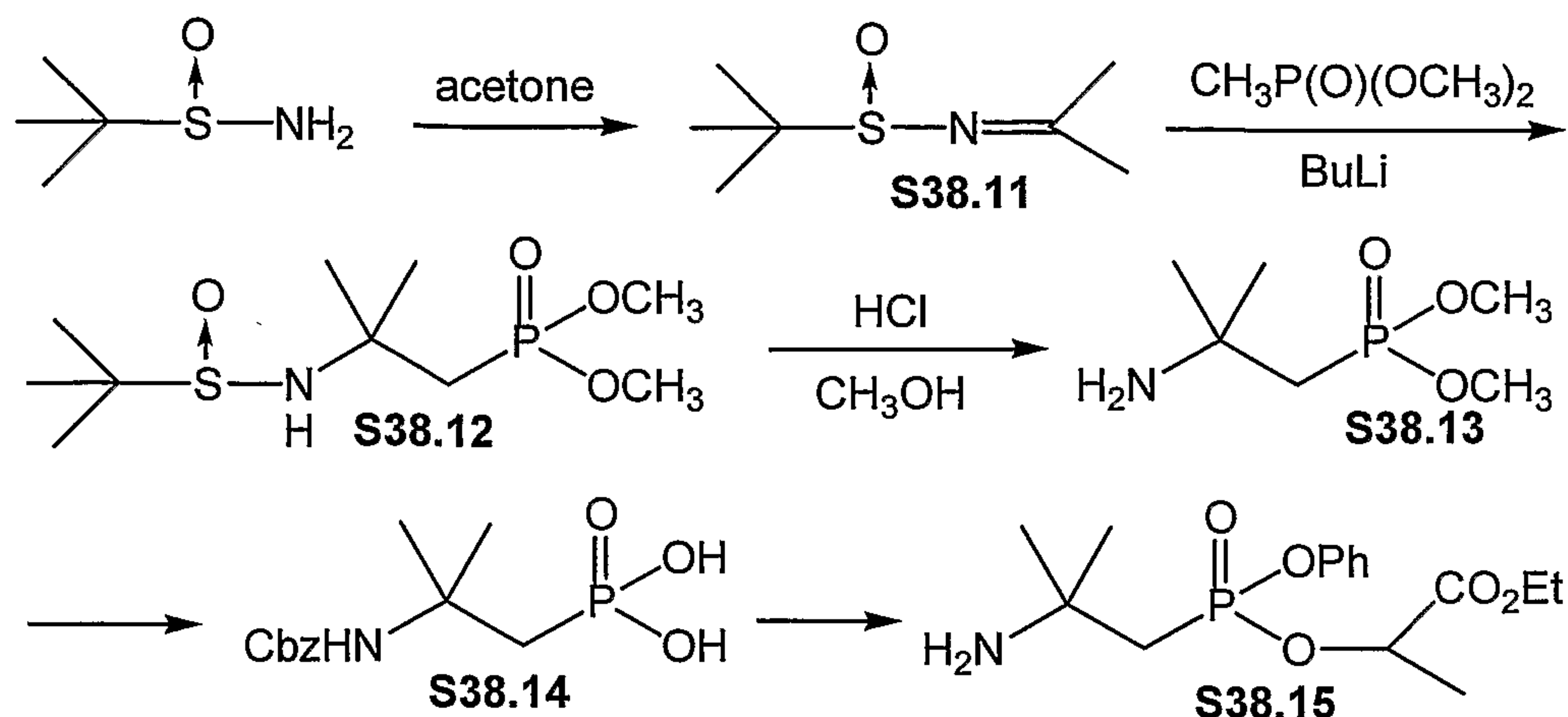
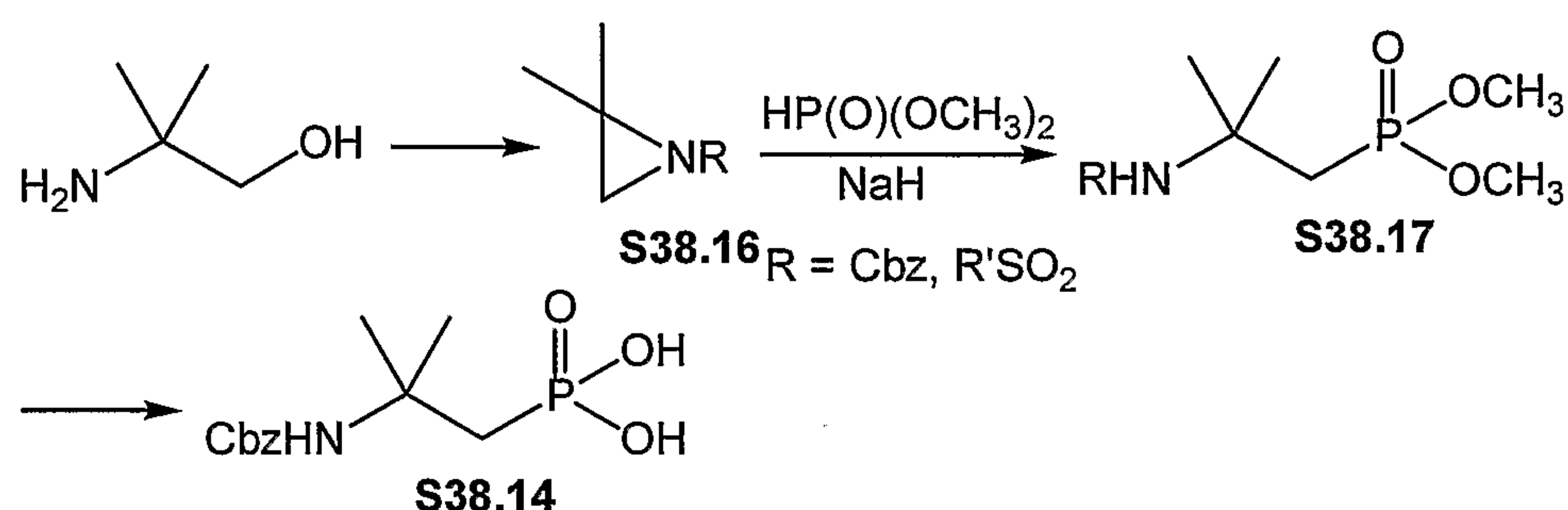
Scheme 37 Example 3



Scheme 37 Example 4

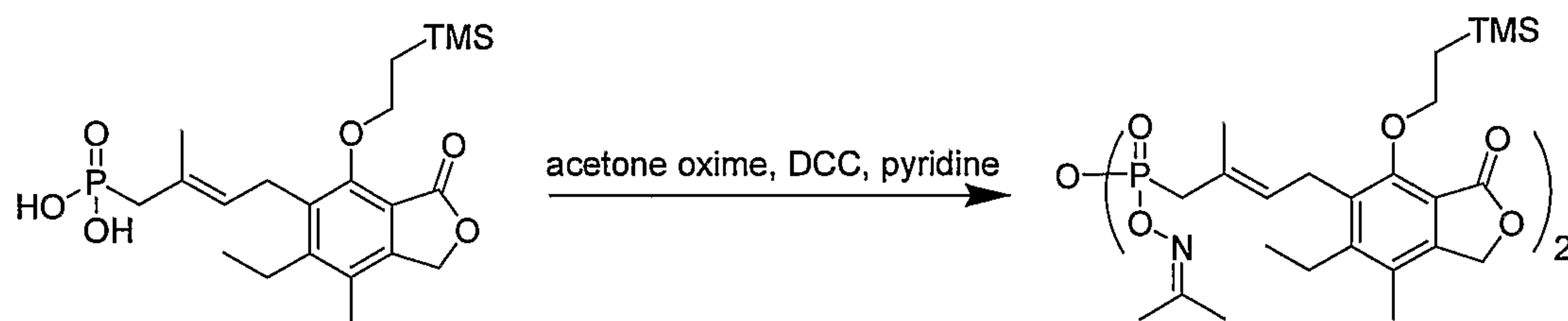


2,2-Dimethyl-2-aminoethylphosphonic acid intermediates can be prepared by the route in Scheme 5. Condensation of 2-methyl-2-propanesulfonamide with acetone give sulfinyl imine **S38.11** (*J. Org. Chem.* **1999**, 64, 12). Addition of dimethyl methylphosphonate lithium to **S38.11** afford **S38.12**. Acidic methanolysis of **S38.12** provide amine **S38.13**. Protection of amine with Cbz group and removal of methyl groups yield phosphonic acid **S38.14**, which can be converted to desired **S38.15** (Scheme 38a) using methods reported earlier on. An alternative synthesis of compound **S38.14** is also shown in Scheme 38b. Commercially available 2-amino-2-methyl-1-propanol is converted to aziridines **S38.16** according to literature methods (*J. Org. Chem.* **1992**, 57, 5813; *Syn. Lett.* **1997**, 8, 893). Aziridine opening with phosphite give **S38.17** (*Tetrahedron Lett.* **1980**, 21, 1623). Reprotection of **S38.17** affords **S38.14**.

Scheme 38a**Scheme 38b**

5

The invention will now be illustrated by the following non-limiting Examples.

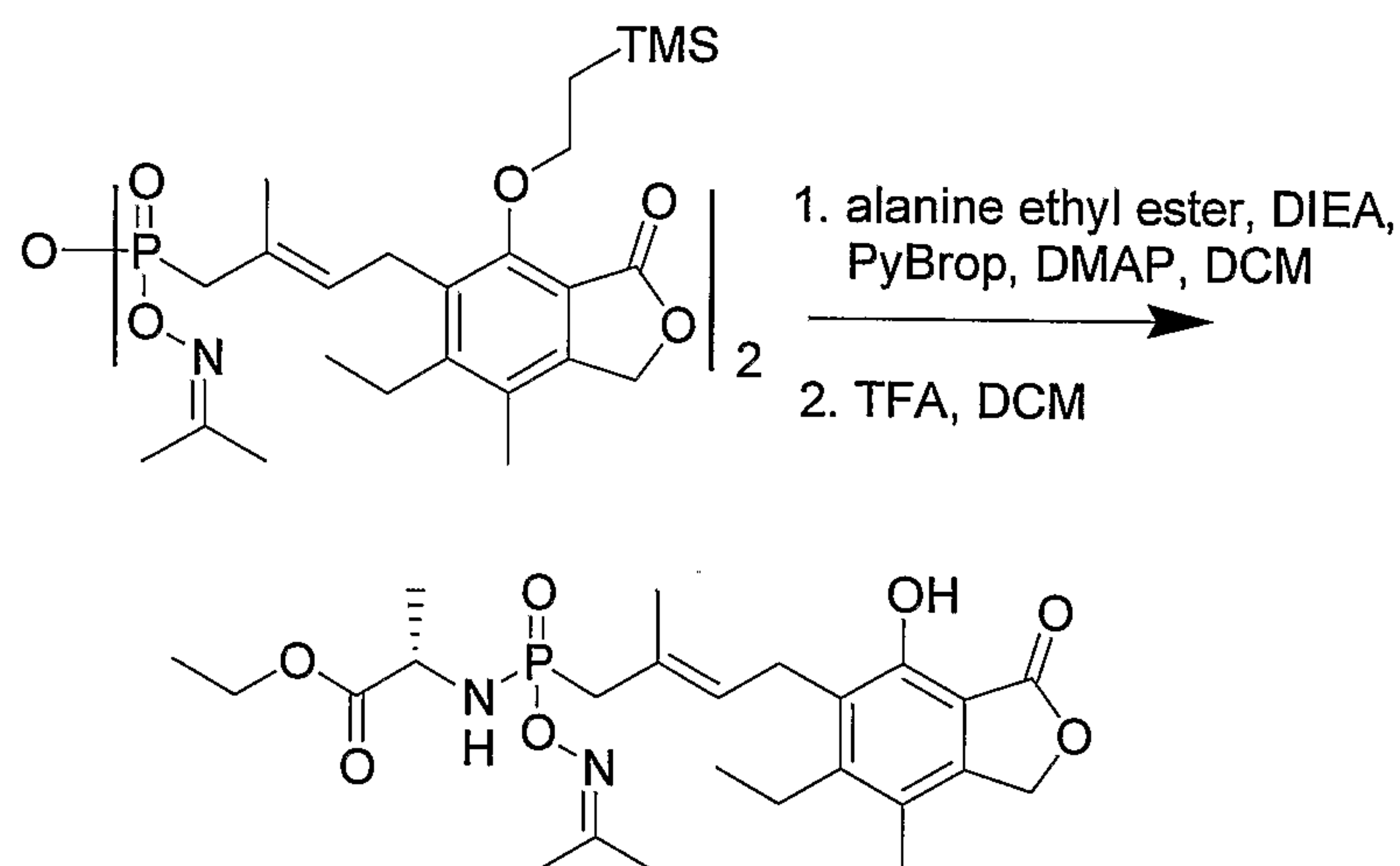
Example 1

10

{4-[6-Ethyl-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid propan-2-one oxime ester anhydride: A solution of {4-[6-ethyl-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid (450 mg, 1.02 mmol), acetone oxime (150mg, 2.04 mmol) and DCC (850 mg, 4.08 mmol) in pyridine (3 mL) was stirred at 70 °C for 1.5

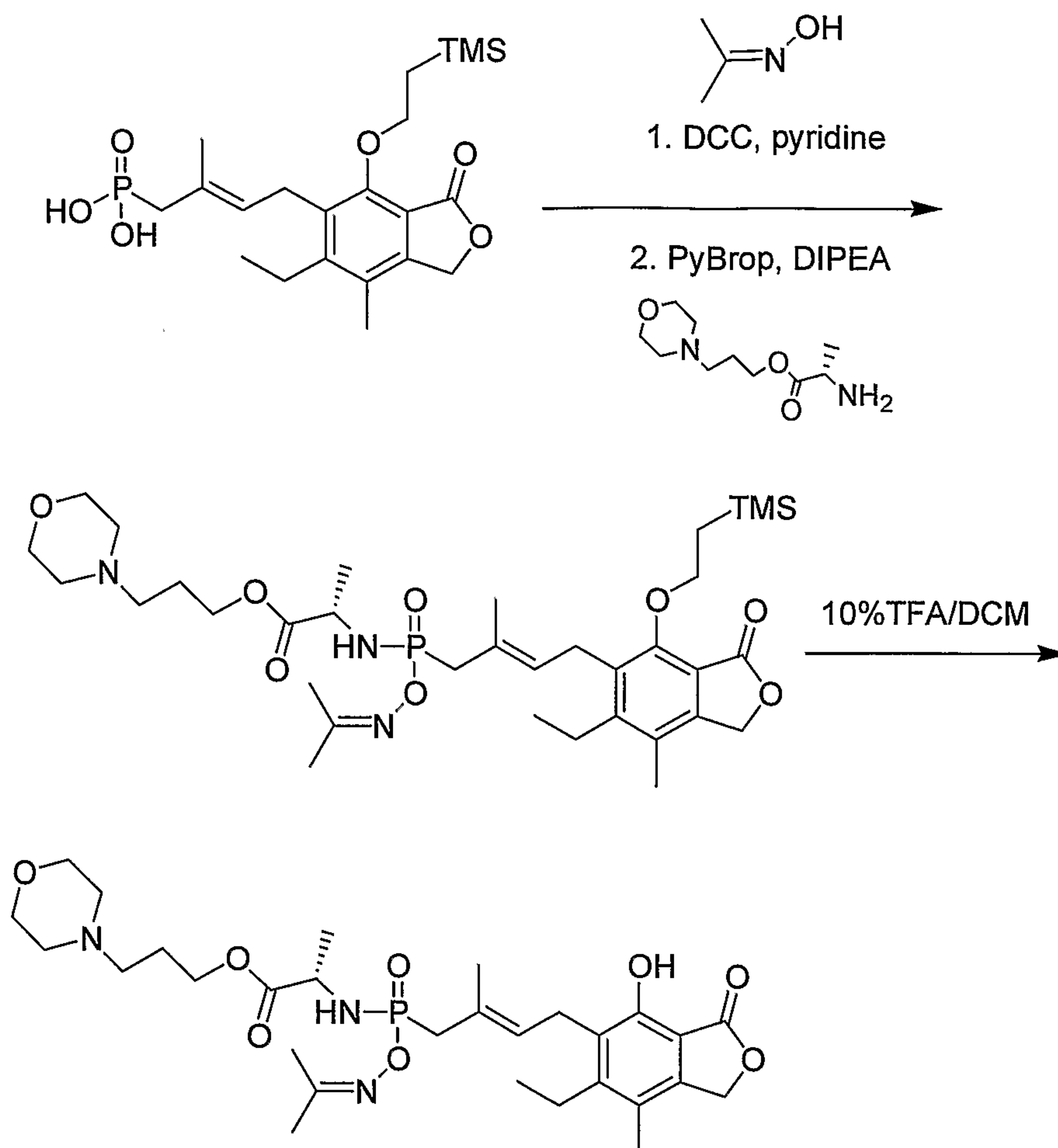
15

hr. The reaction was quenched by adding water (4 mL). The mixture was stirred at room temperature for 30 min. and concentrated. The residue was partitioned between CH₂Cl₂ and diluted aqueous HCl. The organic layer was concentrated and purified by RP HPLC, affording the desired product (396 mg, 57 %). MS
5 (*m/z*) 995 [M+Na]⁺.



2-(S)-{[4-(6-Ethyl-4-hydroxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-(isopropylideneamino-*N*-oxy)-phosphinoylamino}-propionic acid ethyl ester: this product was prepared using methods similar to
10 those described herein except using {4-[6-ethyl-7-methyl-3-oxo-4-(2-trimethylsilylanyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid propan-2-one oxime ester anhydride (JX-2260-159). ¹H NMR (300 MHz, CDCl₃) δ 1.14 (m, 3H), 1.22-1.36 (m, 6H), 1.89 (s, 3H), 1.93 (s, 3H), 1.96 (s+s, 3H), 2.17 (s, 3H), 2.70 (m, 4H), 3.46 (m, 3H), 4.19 (m, 3H), 5.21 (s,
15 2H), 5.25 (m, 1H), 7.69 (brs+brs, 1H) ppm; ³¹P (121.4 MHz, CDCl₃) δ 33.41, 33.71; MS (*m/z*) 495 [M+H]⁺.

Example 2



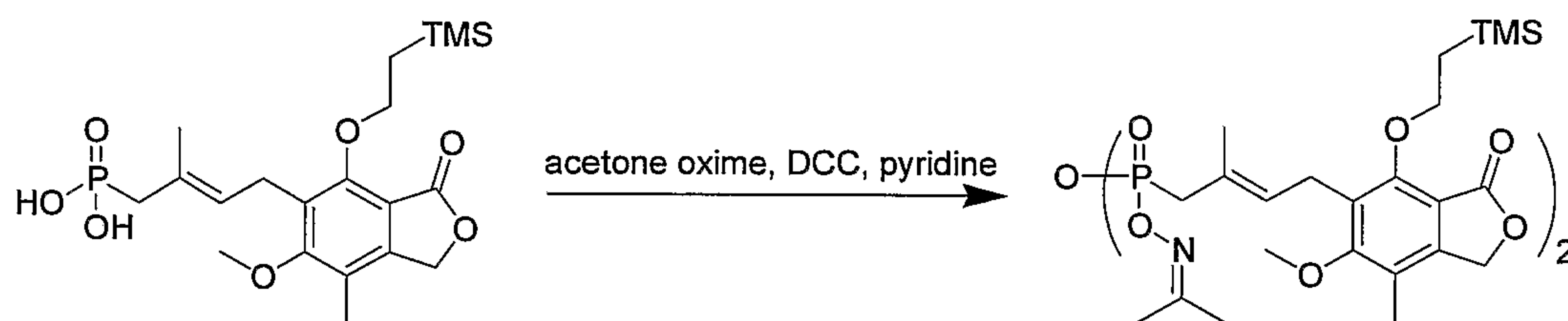
2-**{[4-(4-Hydroxy-6-ethyl-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-**
 5 **2-methyl-but-2-enyl]-(isopropylideneamino-N-oxy)-phosphinoylamino}-**
propionic acid 3-morpholin-4-yl-propyl ester: DCC (112 mg, 0.52 mmol) was
 added to a solution of {4-[6-Ethyl-7-methyl-3-oxo-4-(2-trimethylsilanyl-
 ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid
 (60 mg, 0.14 mmol) and acetone oxime (10 mg, 0.26 mmol) in pyridine (0.5
 10 mL). The mixture was heated to 70°C for 40 min then cooled to room
 temperature before quenching with water (2 mL). After removal of solvent, the
 residue was dissolved in DCM and washed with 0.5N HCl and brine. The
 organic layer was dried over sodium sulfate and concentrated to dryness. The
 crude material was purified by RP HPLC to afford 47 mg (71%) of the
 15 symmetrical anhydride.

This symmetrical anhydride material (23 mg, 0.024 mmol), methyl 2-amino-propionic acid 3-morpholin-4-yl-propyl ester (52 mg, 0.24 mmol) and DIEA (42 μ L, 0.24 mmol) were dissolved in DCM (0.5 mL). DMAP (3 mg, catalytic) was added, followed by PyBROP (45 mg, 0.096 mmol). Stirring at room temperature was continued overnight. The solvent was evaporated and the residue was purified by RP-HPLC (eluent: water/MeCN) to afford the desired product.

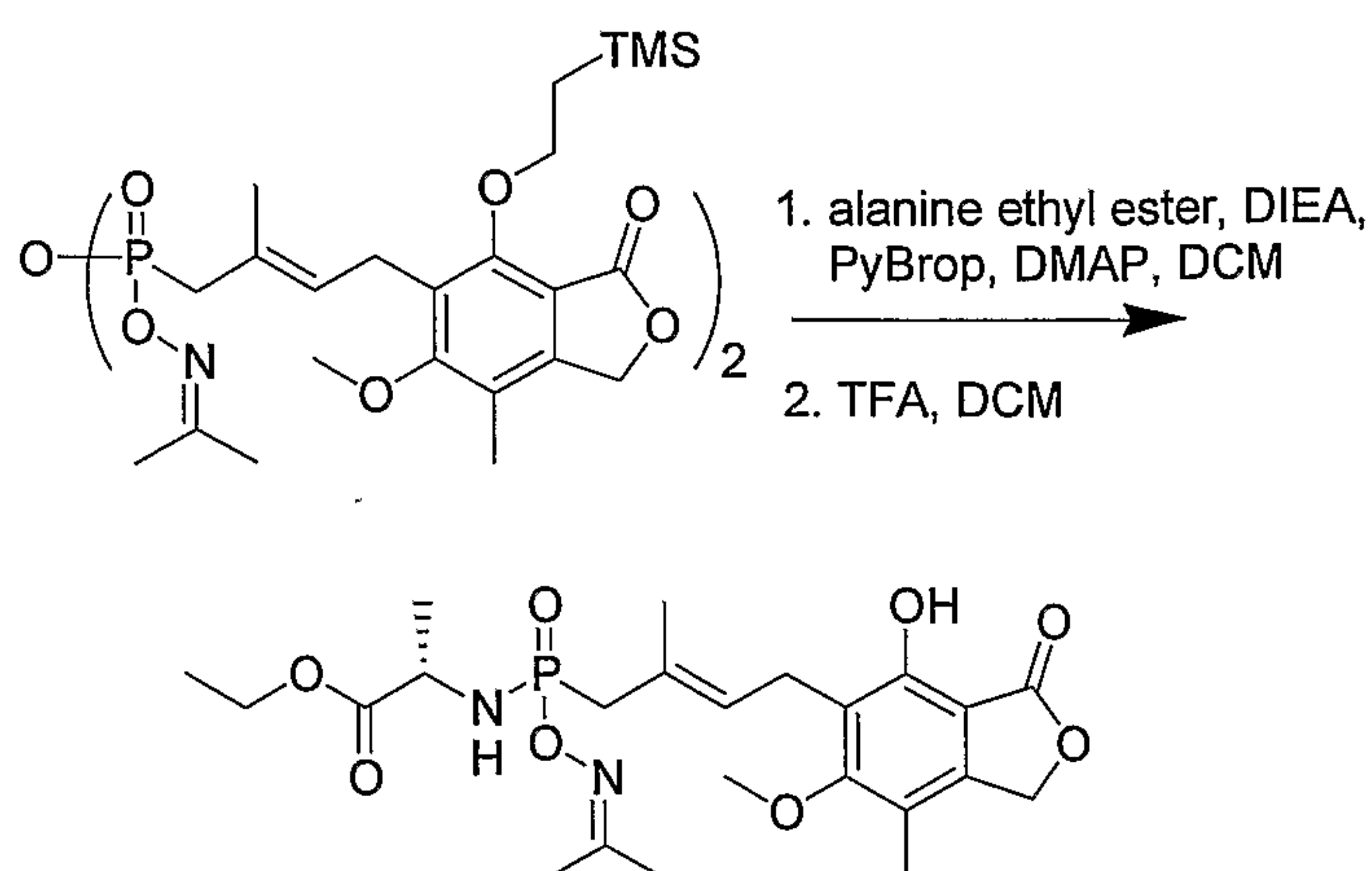
The product obtained was re-suspended in 10% TFA/ CH_2Cl_2 (1 mL) at 0°C for 2 hours. The reaction mixture was neutralized by adding pyridine (0.1 mL). The crude material was purified by RP HPLC using a C18 column with a gradient of H_2O , 0.1% TFA-acetonitrile, to provide 4.8 mg (34 %) of the desired product. ^{31}P (121.4 MHz, CD_3OD) δ 36.02, 36.28 ppm; MS (m/z) 522.9 $[\text{M}-\text{H}]^+$, 524.6 $[\text{M}+\text{H}]^+$.

15

Example 3



{4-[6-Methoxy-7-methyl-3-oxo-4-(2-trimethylsilanyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid propan-2-one oxime ester anhydride: A solution of {4-[6-methoxy-7-methyl-3-oxo-4-(2-trimethylsilanyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid (60 mg, 0.136 mmol), acetone oxime (20mg, 0.272 mmol) and DCC (113 mg, 0.544 mmol) in pyridine (0.4 mL) was stirred at 70 °C for 40 min. The reaction was quenched by adding water (2 mL), and concentrated. The residue was partitioned between CH_2Cl_2 and diluted aqueous HCl. The organic layer was concentrated and purified by RP HPLC, affording the desired product (57 mg) contaminated with a byproduct DCU which was used for the next reaction without further purification. MS (m/z) 999 $[\text{M}+\text{Na}]^+$.

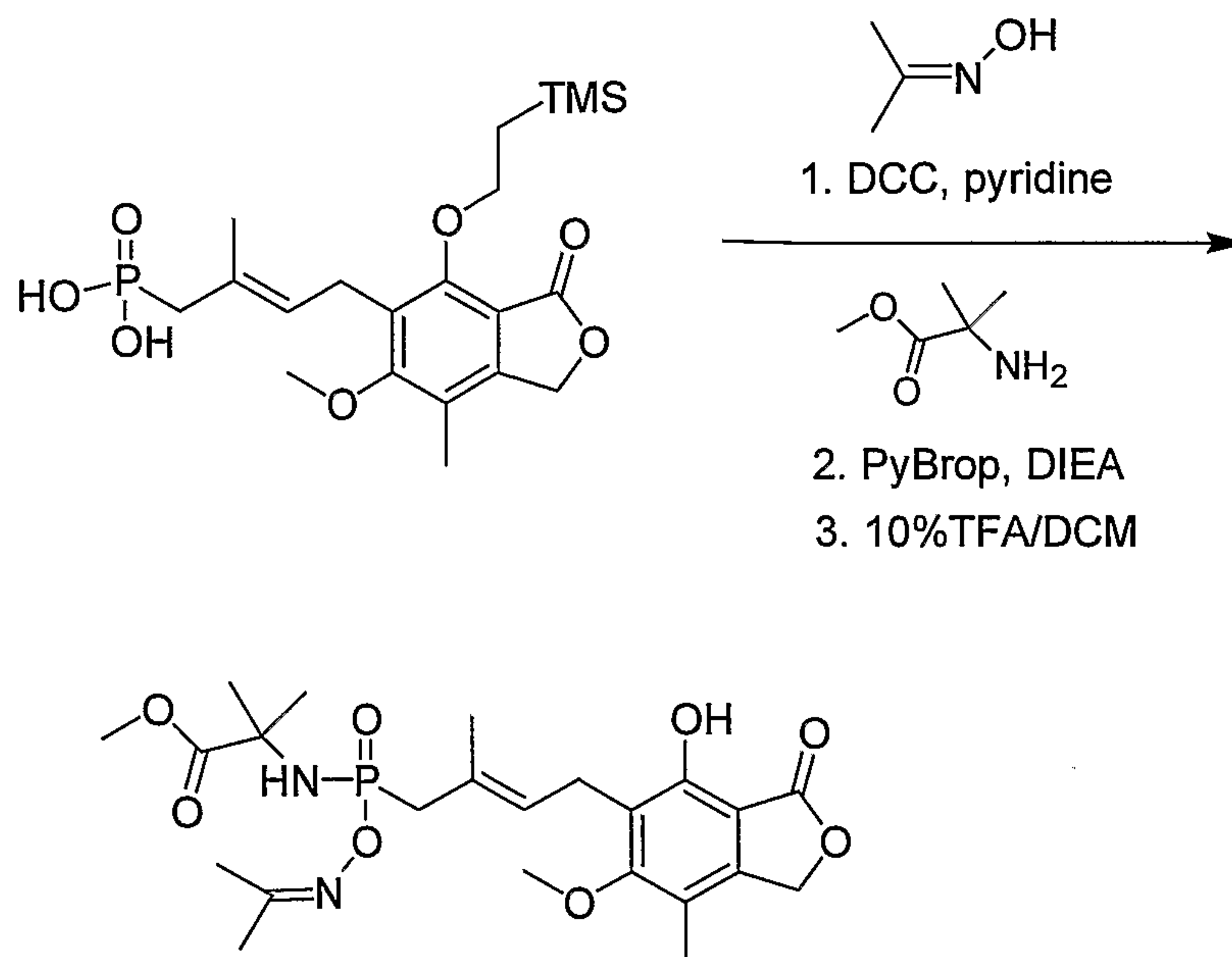


2-(S)-{[4-(4-Hydroxy-6-methoxy-7-methyl-3-oxo-1,3-dihydro-
 5 isobenzofuran-5-yl)-2-methyl-but-2-enyl]-(isopropylideneamino-N-oxy)-
 phosphinoylamino}-propionic acid ethyl ester: This product was prepared
 using methods similar to those described herein except using {4-[6-methoxy-7-
 methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-
 methyl-but-2-enyl}-phosphonic acid propan-2-one oxime ester anhydride (JX-
 10 2260-91). ¹H NMR (300 MHz, CDCl₃) δ 1.23- 1.38 (m, 6H), 1.90 (s, 3H), 1.95
 (m, 6H), 2.15 (s, 3H), 2.71 (m, 2H), 3.44 (m, 1H), 3.55 (m, 1H), 3.79 (s, 3H),
 4.15 (m, 2H), 5.21 (s, 2H), 5.38 (m, 1H), 5.64-6.10 (m, 1H), 7.75 (brs+brs, 1H)
 ppm; ³¹P (121.4 MHz, CDCl₃) δ 33.43, 33.80; MS (*m/z*) 497 [M+H]⁺.

15

20

25

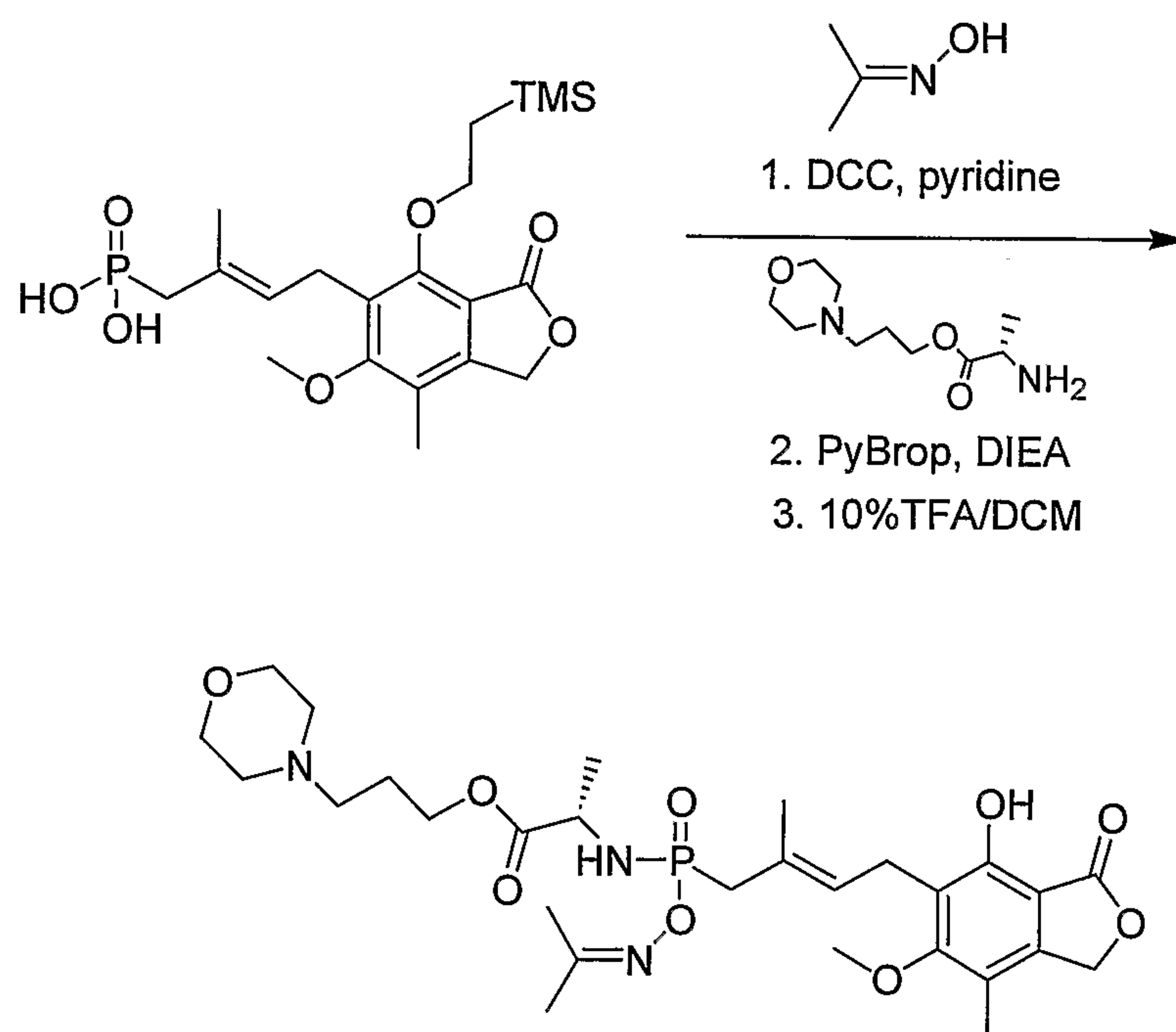
Example 4

2-**{[4-(4-Hydroxy-6-methoxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-(isopropylideneamino-*N*-oxy)-phosphinoylamino}-propionic acid 2-amino-2-methyl methyl ester**: DCC (113 mg, 0.544 mmol) was added to a solution of {4-[6-Methoxy-7-methyl-3-oxo-4-(2-trimethylsilanyloxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid (60 mg, 0.136 mmol) and acetone oxime (20 mg, 0.27 mmol) in pyridine (0.4 mL). The mixture was heated to 70°C for 40 min then cooled to room temperature before quenching with water (2 mL). After removal of solvent, the residue was dissolved in DCM and washed with 0.5N HCl and brine. The organic layer was dried over sodium sulfate and concentrated to dryness. The crude material was purified by RP HPLC to afford 62 mg (94%) of the symmetrical anhydride.

This symmetrical anhydride material (32 mg, 0.033 mmol), methyl aminoisobutyrate hydrochloride salt (94 mg, 0.62 mmol) and DIEA (162 μ L, 0.93 mmol) were dissolved in DCM (0.5 mL). DMAP (4 mg, catalytic) was added, followed by PyBROP (58 mg, 0.12 mmol). Stirring at room temperature was continued overnight. The solvent was evaporated and the residue was purified by RP-HPLC (eluent: water/MeCN) to afford the desired product.

The product obtained was re-suspended in a 10% TFA/ CH₂Cl₂ (1 mL) solution at 0°C for 2 hours. The reaction mixture was neutralized by adding pyridine (0.1 mL). The crude was purified by RP HPLC using a C18 column with a gradient of H₂O, 0.1% TFA-acetonitrile, to provide 9 mg (30 %) of the
 5 desired product. ¹H NMR (300 MHz, CD₃OD) δ 1.42 (s, 3H), 1.47 (s, 3H), 1.84 (s, 3H), 1.91 (s, 3H), 1.95 (s, 3H), 2.15 (s, 3H), 2.68-2.75 (m, 2H), 3.41 (m, 2H), 3.67 (s, 3H), 3.79 (s, 3H), 5.25 (s, 2H), 5.40 (m, 1H) ppm; ³¹P (121.4 MHz, CD₃OD) δ 35.72 ppm; MS (*m/z*) 496.8 [M+H]⁺.

10

Example 5

2-([4-(4-Hydroxy-6-methoxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-(isopropylideneamino-N-oxy)-phosphinoylamino)-propionic acid 3-morpholin-4-yl-propyl ester: DCC (113 mg, 0.544 mmol)
 15 was added to a solution of {4-[6-Methoxy-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl]-2-methyl-but-2-enyl}-phosphonic acid (60 mg, 0.136 mmol) and acetone oxime (20 mg, 0.27 mmol) in pyridine (0.4 mL). The solution mixture was heated to 70°C for 40 min then cooled to room
 20 temperature before quenching with water (2 mL). After removal of solvent, the residue was dissolved in DCM and washed with 0.5N HCl and brine. The

organic layer was dried over sodium sulfate and concentrated to dryness. The crude material was purified by RP HPLC to afford 62 mg (94%) of the symmetrical anhydride.

This symmetrical anhydride material (32 mg, 0.033 mmol), methyl 2-
5 Amino-propionic acid 3-morpholin-4-yl-propyl ester (71 mg, 0.33 mmol) and DIEA (57 μ L, 0.33 mmol) were dissolved in DCM (0.5 mL). DMAP (4 mg, catalytic) was added, followed by PyBROP (62 mg, 0.13 mmol). Stirring at room temperature was continued for overnight. The solvent was evaporated and the residue was purified by RP-HPLC (eluent: water/MeCN) to afford the
10 desired product.

The product obtained was re-suspended in a 10% TFA/ CH_2Cl_2 (1 mL) solution at 0°C for 2 hours. The reaction mixture was neutralized by adding pyridine (0.1 mL). The crude was purified by RP HPLC using a C18 column with a gradient of H_2O , 0.1% TFA-acetonitrile, to provide 9 mg (25 %) of the
15 desired product. ^1H NMR (300 MHz, CD_3OD) δ 1.29 (m, 3H), 1.80 (m, 2H), 1.84 (s, 3H), 1.91 (s, 3H), 1.95 (s, 3H), 2.15 (s, 3H), 2.43 (m, 6H), 2.68-2.75 (m, 2H), 3.41 (m, 2H), 3.68 (m, 4H), 3.79 (s, 3H), 3.91 (m, 1H), 4.14 (m, 2H), 5.25 (s, 2H), 5.40 (m, 1H) ppm; ^{31}P (121.4 MHz, CD_3OD) δ 36.05, 36.34 ppm; MS (m/z) 596.0 $[\text{M}+\text{H}]^+$.

20

Example 6

Preparation of Representative Compounds of The Invention

Representative compounds of the invention can be made according to the general route outlined below.

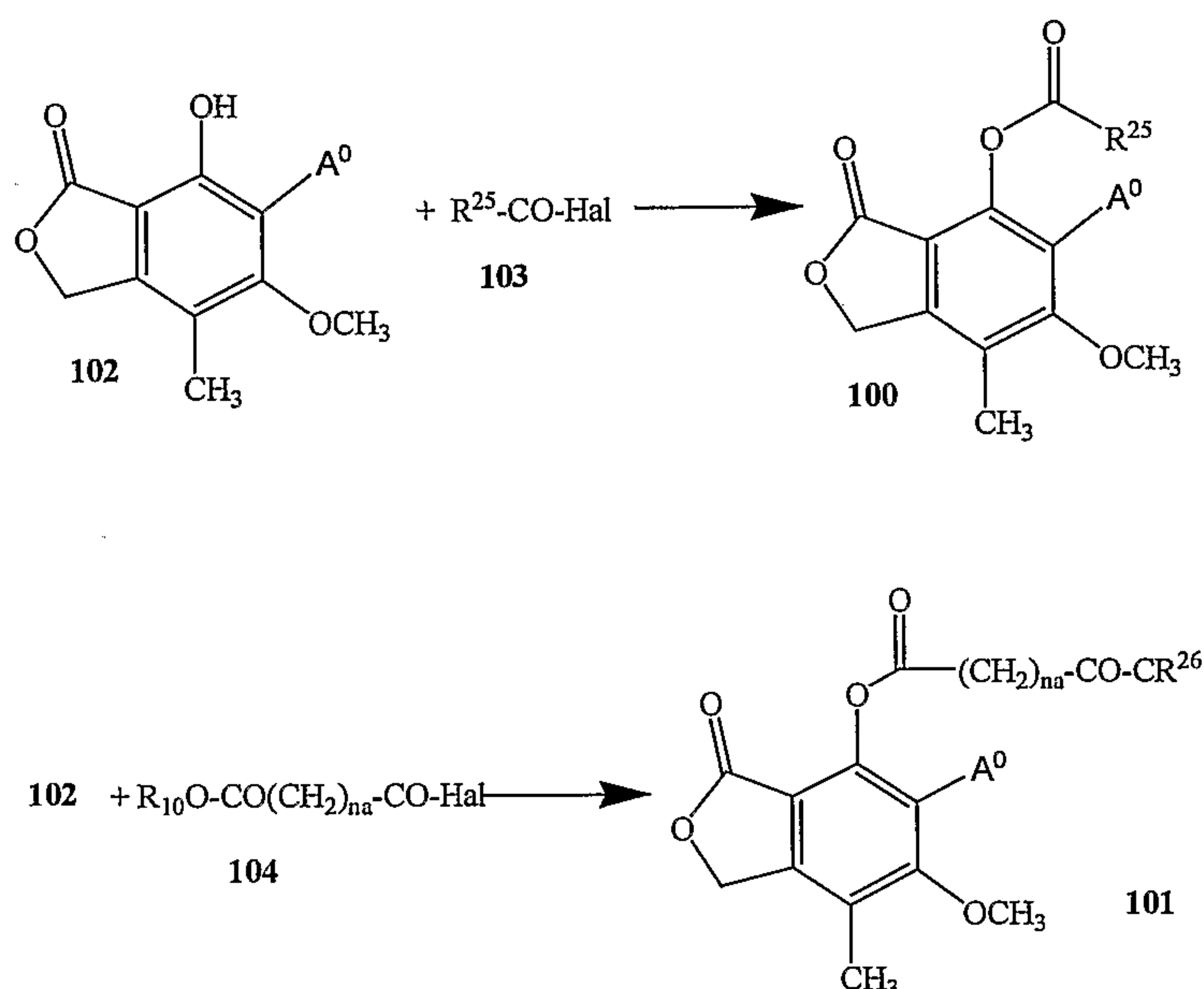
25 The novel compounds of the invention are prepared by the procedures detailed and illustrated below.

Syntheses are available in the published scientific and patent literature for many compounds related to those illustrated by Formulae VI and VII. For example, methods of preparing mycophenolic acid compounds and derivatives
30 thereof are described in U.S. Pat. Nos. 4959387, 4725622, 3705894, 3777020 and 3868454, Japanese Kokai Nos. 57/183776, 57/183777, and 48/86860, South African Application No. 68/4959, Great Britain Patent No. 1261060, Belgian

Patent No. 815330, and West German Patent No. 2237549.

Compounds of Formula 100 and Formula 101

The preparation of compounds of Formula 100 and Formula 101 is shown below.



5

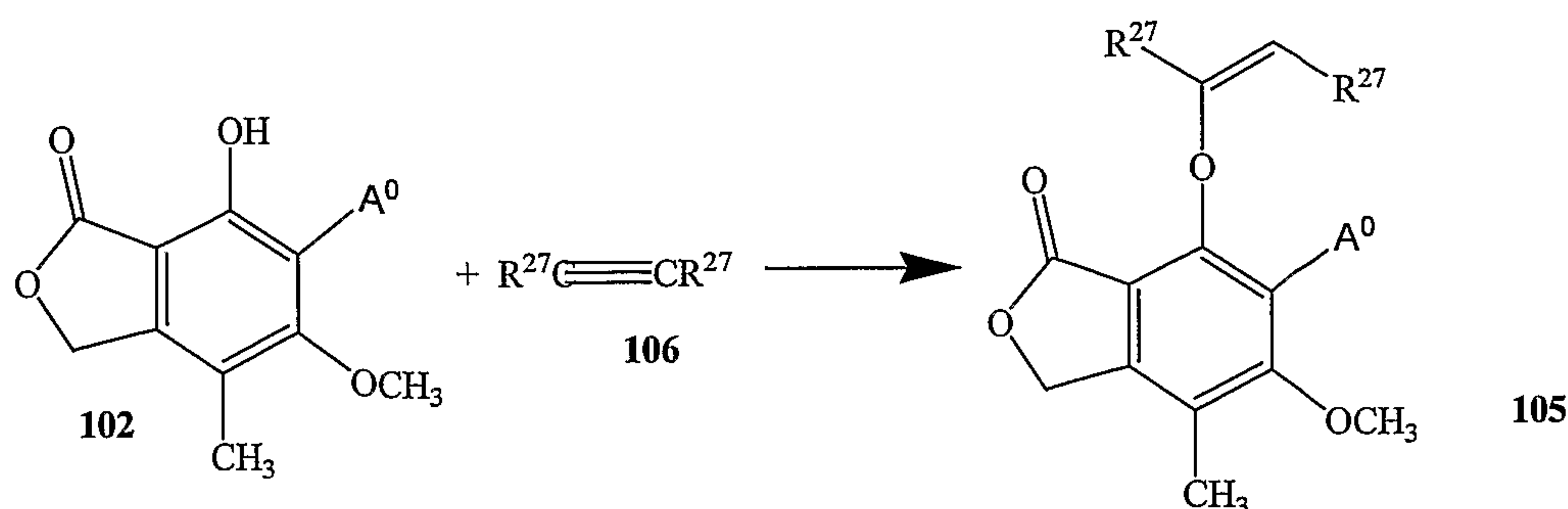
The compounds of Formula 100 are prepared directly from compounds of Formula 102 by reaction with an appropriately substituted acyl halide of Formula 103. To carry out this process, the compound of Formula 102 is dissolved in an inert organic solvent as defined above, for example, acetonitrile, and reacted with about 1 to 6 molar equivalents, or about 3 molar equivalents, of the appropriate compound of Formula 103, in the presence of about 1 to 6 molar equivalents, for example, about 3 molar equivalents, of an inorganic base or tertiary organic base such as pyridine. The reaction takes place at a temperature of about 0°C to about 25°C, for example, about 5°C, for about 1-10 hours, or about 3 hours. When the reaction is substantially complete, the product of Formula 100 is isolated by conventional means.

Similarly, substituting an acyl halide of Formula 104 for an acyl halide of Formula 103, the compounds of Formula 101 are prepared.

Compounds of Formula 105

Compounds of Formula 105 are prepared as shown below .

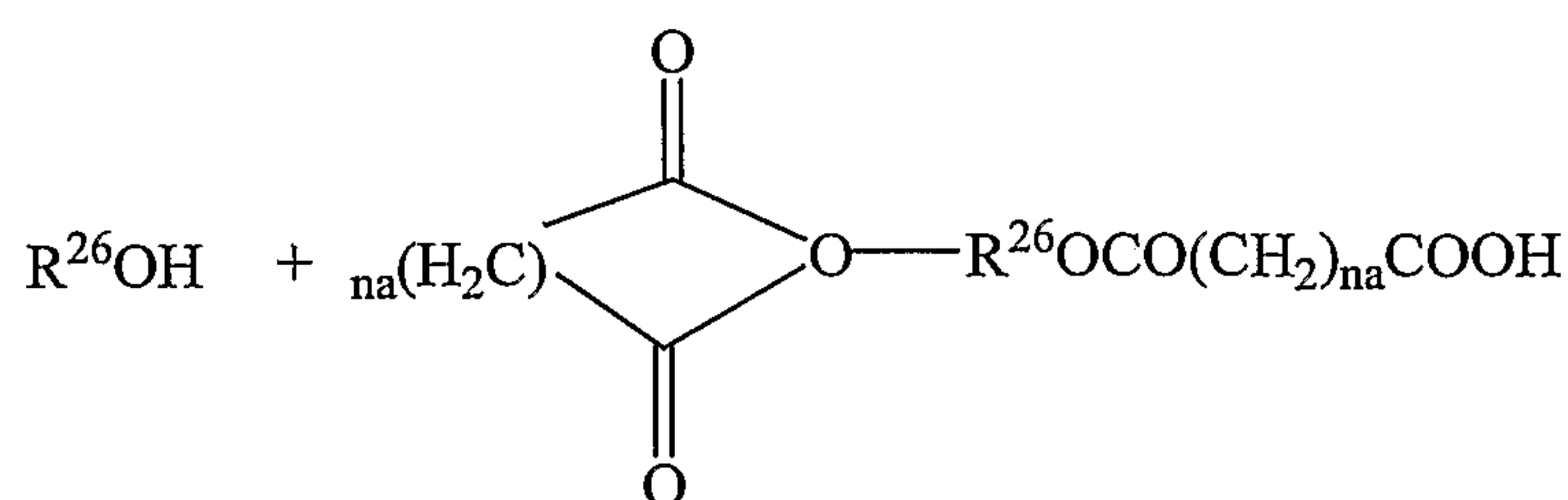
20



Compounds of Formula 105 are prepared directly from compounds of Formula 102 by reaction with an appropriately substituted acetylene of Formula 106. To carry out this process, the compound of Formula 102 is dissolved in an inert organic solvent, for example, tetrahydrofuran, and reacted with about 1 to 5 molar equivalents, or about 2 molar equivalents, of the appropriate compound of Formula 106 in the presence of about 1 to 5 molar equivalents, or about 2 molar equivalents, of a tertiary organic base or inorganic base, for example, pyridine. The reaction takes place at a temperature of about 0°C to about 50°C, or about 5°C, for about 1-48 hours, or about 16 hours. When the reaction is substantially complete, the product of Formula 105 is isolated by conventional means.

Starting Materials

The acyl halides of Formula 103 and Formula 104 are prepared from commercially available carboxylic acids or half esters of dicarboxylic acids respectively (*see* Synthetic Organic Chemistry, by Wagner and Zook, pp 546-547). The half esters of dicarboxylic acids, if not commercially available, can be prepared, for example, by the reaction of the appropriate alcohol and an anhydride formed from a dicarboxylic acid as shown in the reaction scheme below.



The reaction is discussed in more detail in Comprehensive Organic Chemistry, Vol. 2, by Barton and Ollis, p 687.

Acetylenes of Formula 106 are commercially available or are prepared from propiolic acid or acetylene dicarboxylic acid by conventional esterification procedures, which are discussed in more detail in Organic Functional Group Preparations, 2nd Edition, Vol 1, by Sandler and Karo, pp 289-309.

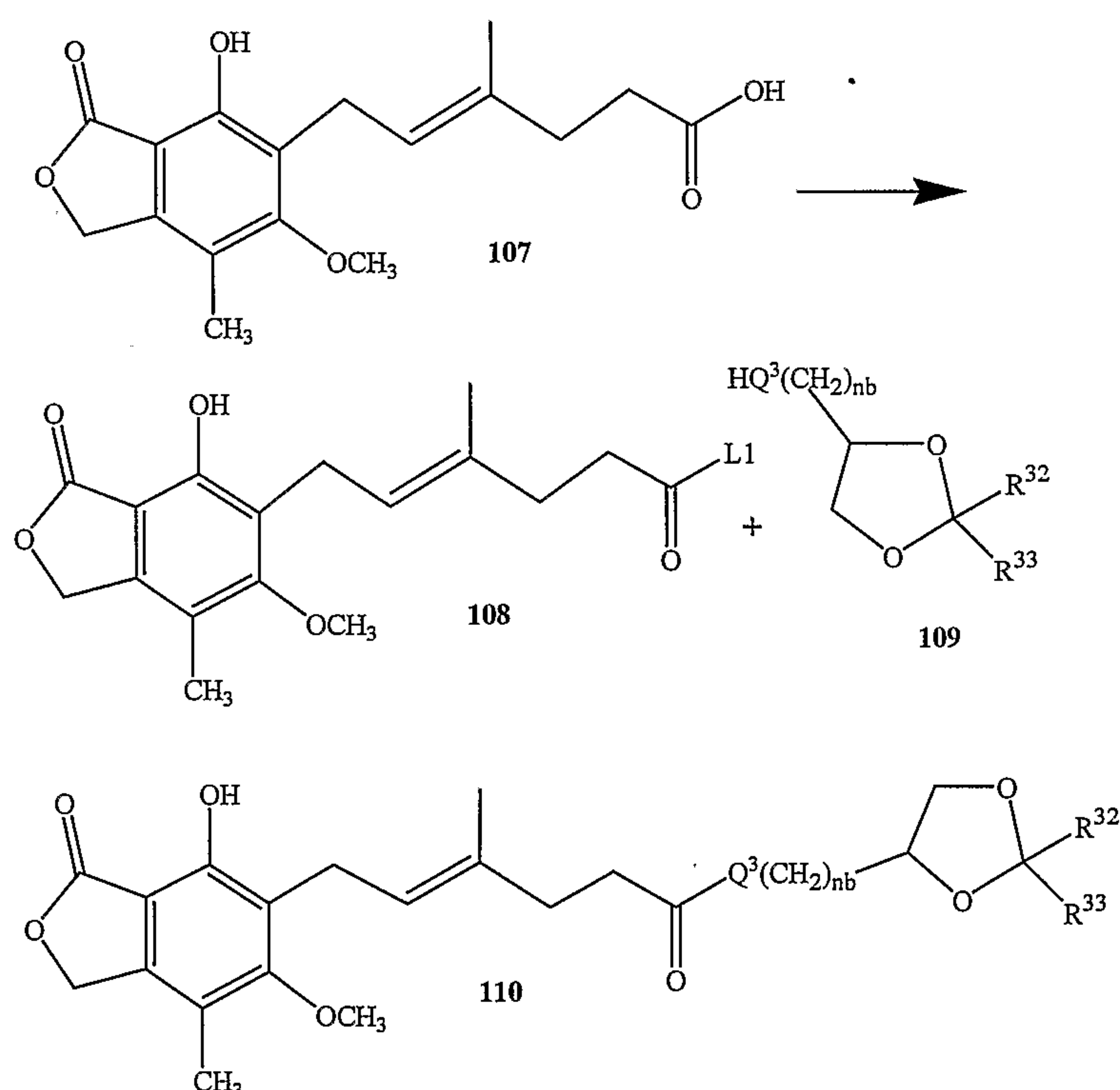
5

Example 7

Preparation of Representative Compounds of the Invention

Synthetic intermediates and processes that can be used to prepare mycophenolate derivatives of the invention are reported in U.S. Pat. No.

10 4,686,234. For example, representative compounds of the invention can be made according to the general route outlined below.



To prepare the compounds of the invention, the mycophenolic acid derivative of Formula 107 is first converted to an activated carbonyl derivative of Formula 108 in which L1 is a leaving group, such as halo, N-carbonylimidazole, alkoxy, acyloxy or the like, chosen to be capable of displacement by a compound of Formula 109 in the presence of a base. The compounds of Formula 108 are prepared by standard means available in the chemical arts. For example, the compound of Formula 108 where L1 is chloro is made by reaction with from 1.0 to 10 molar equivalents, for example, 4.0 molar

15

20

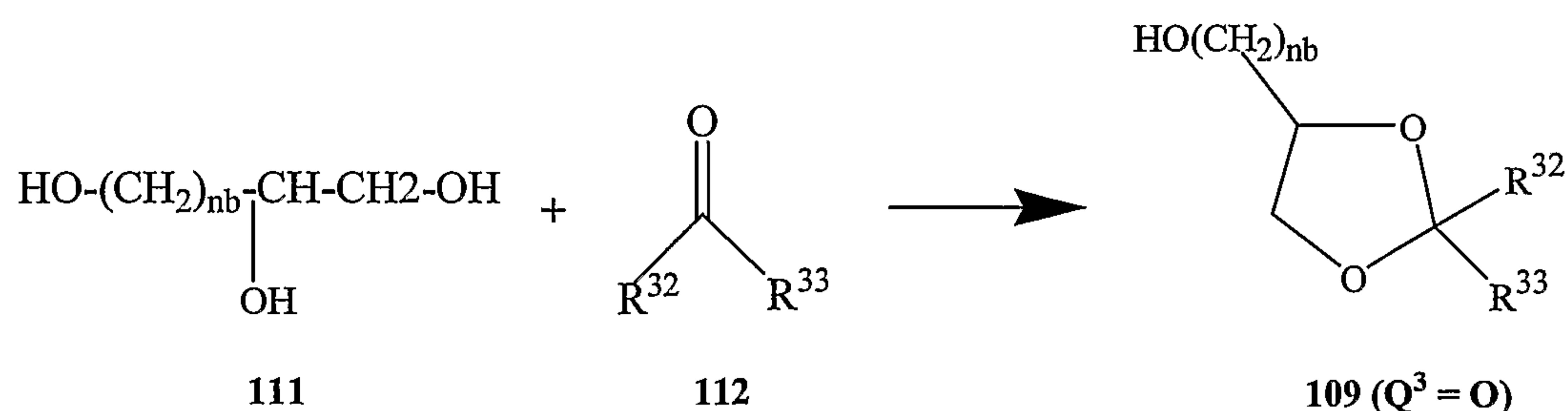
equivalents, of an inorganic acid halide such as phosphorus trichloride, phosphorus pentachloride or preferably thionyl chloride, optionally in the presence of a catalytic amount of an N,N-disubstituted amide, such as N,N-dimethylformamide in an inert organic solvent such as benzene, toluene, tetrahydrofuran, diethyl ether, chloroform or preferably dichloromethane. The reaction is conducted at a temperature of about 0°C to 90°C, for example, about 25°C, for about 1-12 hours, for example, about three hours.

The product of Formula 108 is then reacted with about 1-10 molar equivalents, for example, about 4.0 molar equivalents, of a compound of Formula 109 in an inert organic solvent, for example, dichloromethane. The reaction takes place in the presence of from 1-10 molar equivalents, for example, about 5 molar equivalents, of an inorganic base such as sodium carbonate, potassium bicarbonate or the like, or a tertiary organic base, such as triethylamine, N-methylpiperidine or preferably pyridine. The reaction is conducted at 0°C to 25°C, for example, about 5°C, for thirty minutes to six hours, for example, about one hour. The resulting product of Formula 110 is isolated and purified by conventional means.

The products of the reactions described herein can be isolated and purified by any suitable separation or purification procedure, such as, for example, filtration, extraction, crystallization, column chromatography, thin-layer chromatography, thick-layer chromatography, preparative low or high pressure liquid chromatography, or a combination of these procedures.

Preparation of Starting Materials

Some of the compounds of Formula 109 are commercially available. Alternatively, the compounds of Formula 109 may be prepared by reacting together the compounds of Formula 111 and 112 in the presence of an acid catalyst.

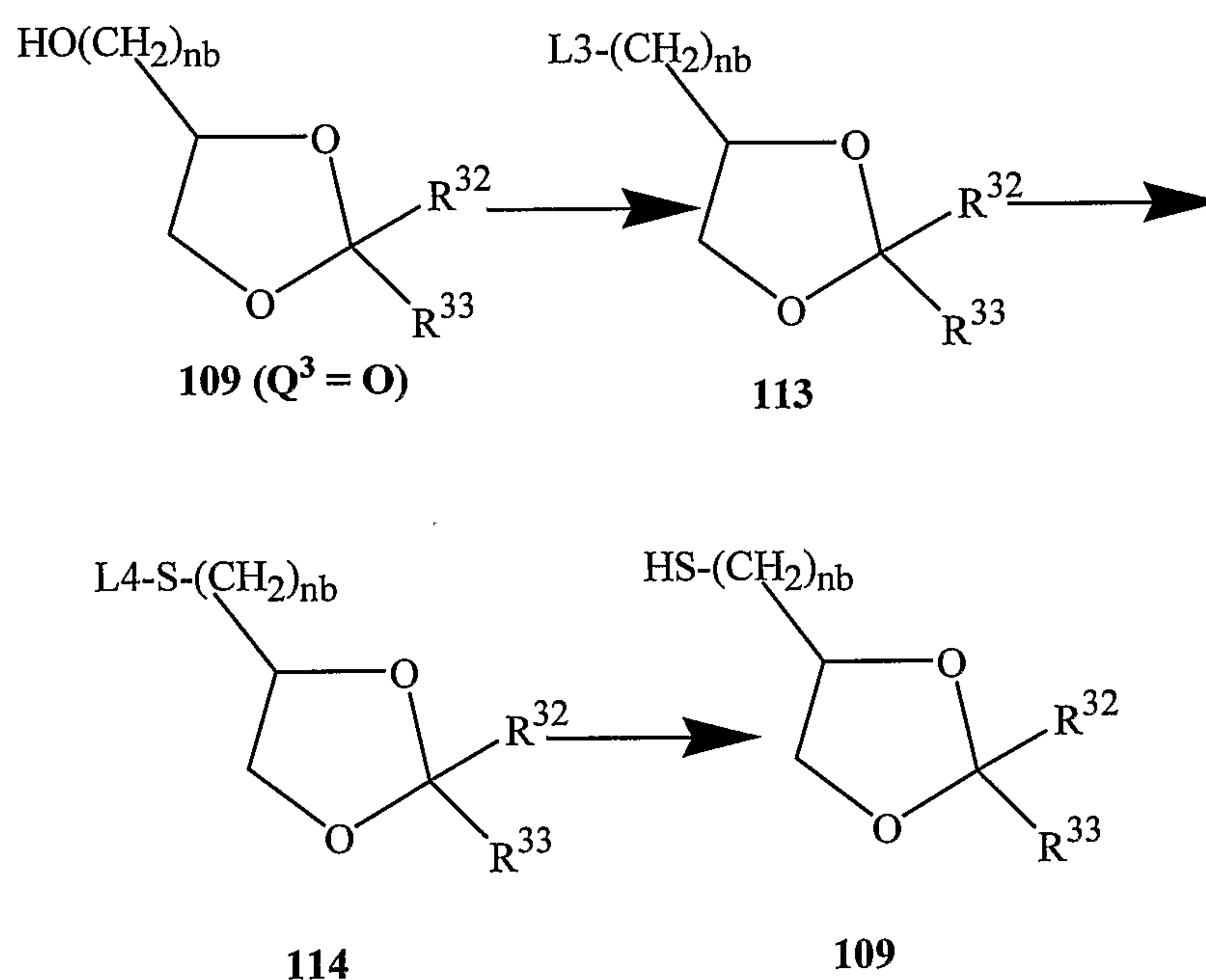


The reaction is discussed in greater detail in Organic Functional Group Preparation, 2nd Edition, Vol III, by Sandler and Karo, pp 4-17.

The triols of Formula 111 can be prepared from the appropriate olefinic alcohol, *e.g.*, allyl alcohol, 7-octen-1-ol, *etc.*, by reaction with aqueous sodium hypochlorite in the presence of osmium tetroxide, as described in the supplement to the 2nd edition of Rodds Chemistry of Carbon Compounds, Vol. 1.sup.E, p 145.

Compounds of Formula 109 where Q^3 is S are obtained from the corresponding compounds where Q^3 is O.

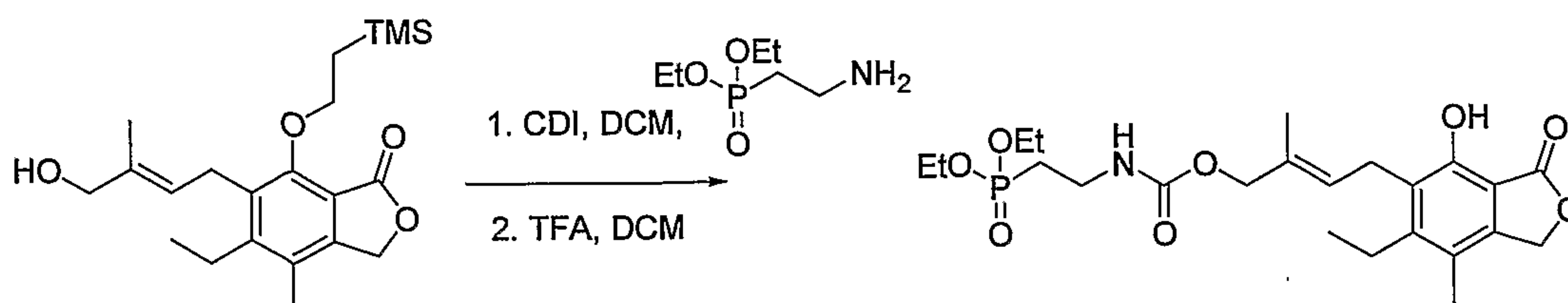
10



The compounds of Formula 109 where Q^3 is O are first converted to a compound 113 where L3 is a leaving group, by reaction with, for example, p-toluenesulfonyl chloride. The group L3 is then displaced with a sulfur nucleophile such as potassium thioacetate or thiourea to afford a compound of Formula 114 in which L4 is CH_3 , $CO-$ or $NH_2C(NH)-$. Hydrolysis of 114 then produces the compound of Formula 109 where Q^3 is S.

20

Example 8



5

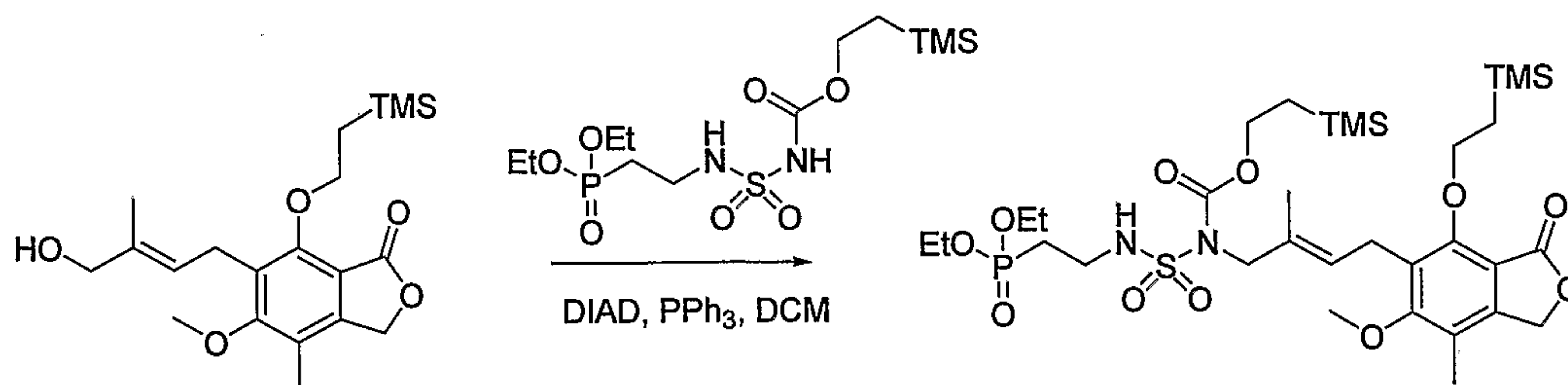
{2-[4-(6-Ethyl-4-hydroxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyloxycarbonylamino]-ethyl}-phosphonic acid diethyl

ester: To a solution of 5-ethyl-6-(4-hydroxy-3-methyl-but-2-enyl)-4-methyl-7-(2-trimethylsilyl-ethoxy)-3H-isobenzofuran-1-one (30 mg, 0.080 mmol) in CH₂Cl₂ (0.5 mL) was added 1,1'-carbonyl diimidazole (14 mg, 0.088 mmol) and the mixture was stirred at room temperature for 30 min. Diethyl aminoethylphosphonate (~100 mg) was then added and stirred was continued for 16 hr. The mixture was concentrated and purified by RP HPLC, affording {2-[4-(6-ethyl-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyloxycarbonylamino]-ethyl}-phosphonic acid diethyl ester. This intermediate was dissolved in CH₂Cl₂ (1 mL) and cooled to 0 °C. A solution of trifluoroacetic acid in CH₂Cl₂ (20%, 1 mL) was added and the mixture was stirred at the same temperature for 1 hr before adding pyridine (0.5 mL). The resulting mixture was concentrated and the residue was purified by RP HPLC to afford the desired product (25 mg, 65 %). ¹H NMR (300 MHz, CDCl₃) δ 1.14 (t, 3H, J= 7.5 Hz), 1.33 (t, 3H, J= 7.5 Hz), 1.93 (s, 3H), 1.98 (dt, 2H, J= 6.9, 17.7 Hz), 2.70 (q, 2H, J= 7.5 Hz), 3.47 (m, 4H), 4.12 (m, 4H), 4.44 (s, 2H), 5.22 (s, 2H), 5.31 (brt, 1H), 5.40 (t, 1H, J= 6.4 Hz), 7.68 (s, 1H) ppm; ³¹P (121.4 MHz, CDCl₃) δ 29.53; MS (*m/z*) 484 [M+H]⁺.

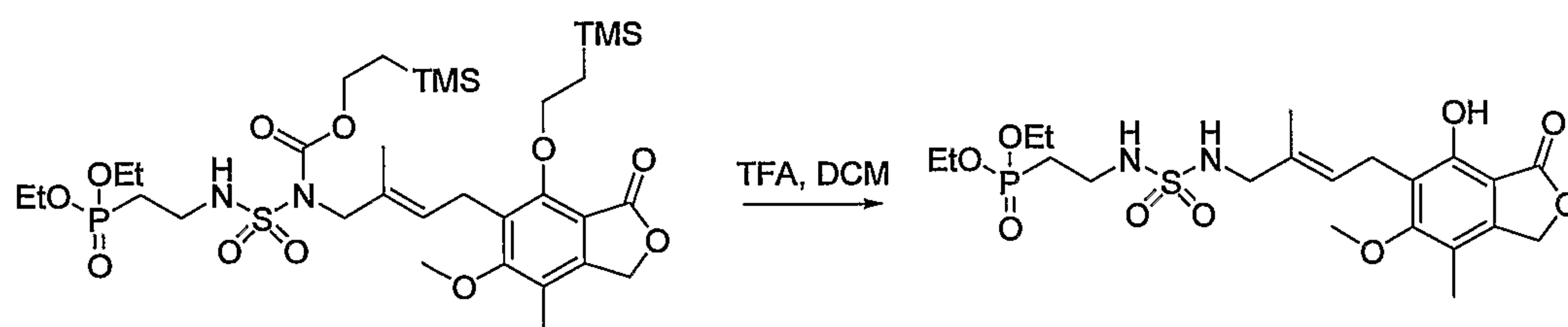
25

30

Example 9



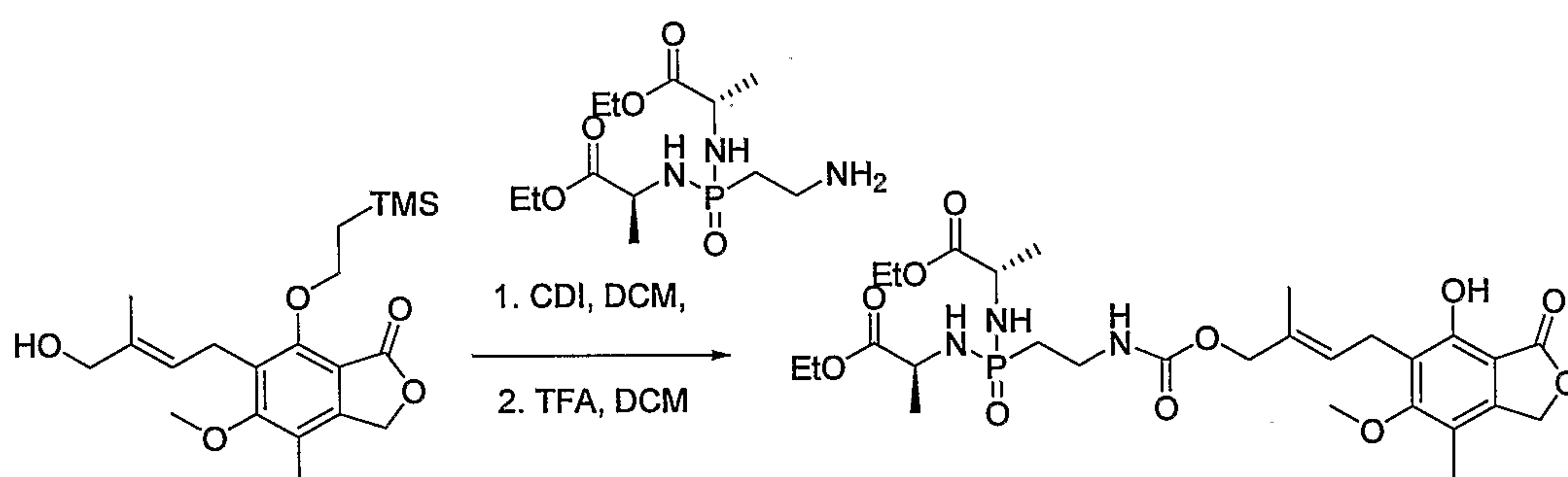
- 5 **{2-[4-(6-Methoxy-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-N-(2-trimethylsilyl-ethoxycarbonyl)aminosulfonylaminoethyl}-phosphonic acid diethyl ester:**
 To an ice-cold solution of triphenylphosphine (108mg, 0.41 mmol) in CH₂Cl₂ (0.5 mL) was slowly added diisopropyl azodicarboxylate (0.078 mL, 0.41
 10 mmol). To the resulting solution, a solution of 6-(4-hydroxy-3-methyl-but-2-enyl)-5-methoxy-4-methyl-7-(2-trimethylsilyl-ethoxy)-3H-isobenzofuran-1-one (102 mg, 0.27 mmol) and diethyl (N-(2-trimethylsilyl-ethoxy)carbonylamino)sulfonylaminoethyl phosphonate in CH₂Cl₂ (0.5 mL) was added. The mixture was then stirred at room temperature for 30 min.,
 15 concentrated and purified by RP HPLC to afford the desired product (131 gm, 64 %). ³¹P (121.4 MHz, CDCl₃) δ 28.20; MS (*m/z*) 765 [M+H]⁺.



- 20 **{2-[4-(4-Hydroxy-6-methoxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-aminosulfonylaminoethyl}-phosphonic acid diethyl ester:** To a cooled (-78 °C) solution of {2-[4-(6-methoxy-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-N-(2-trimethylsilyl-ethoxycarbonyl)amino-sulfonylaminoethyl}-
 25 phosphonic acid diethyl ester (JX-2260-151) (30 mg, 0.039 mmol) in CH₂Cl₂ (1 mL) was slowly added a solution of trifluoroacetic acid (40 %, 1 mL) in CH₂Cl₂.

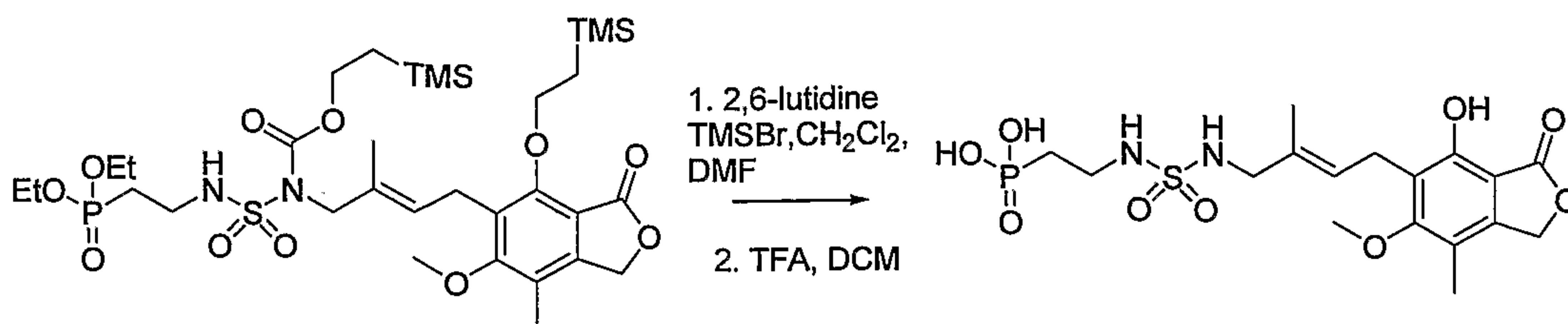
The mixture was stirred at 0 °C for 5 hr. Pyridine (1.5 mL) was added and the mixture was concentrated. The residue was purified by RP HPLC to afford the desired product (18 mg, 73 %) as the trifluoroacetic acid salt. ¹H NMR (300 MHz, CD₃OD) δ 1.33 (t, 3H, J= 7.5 Hz), 1.87 (s, 3H), 2.08 (dt, 2H, J= 8.1, 18.4 Hz), 2.16 (s, 3H), 3.16 (dt, 2H, J= 8.1, 10.4 Hz), 3.44 (m, 4H), 4.37 (s, 3H), 4.10 (m, 2H), 5.24 (s, 2H), 5.50 (t, 1H, J= 6.3 Hz) ppm; ³¹P (121.4 MHz, CD₃OD) δ 28.20; MS (*m/z*) 521 [M+H]⁺.

Example 10



2-(S)-{[4-(4-Hydroxy-6-methoxy-7-methyl-3-oxo-1,3-dihydroisobenzofuran-5-yl)-2-methylbut-2-enyloxycarbonylamino]-ethyl-[1-(S)-(ethoxycarbonyl)ethylamino]}-phosphinoylamino}-propionic acid ethyl ester: this product was prepared using methods similar to those described for preparation of {2-[4-(6-ethyl-4-hydroxy-7-methyl-3-oxo-1,3-dihydroisobenzofuran-5-yl)-2-methylbut-2-enyloxycarbonylamino]-ethyl}-phosphonic acid diethyl ester (JX-2260-141) except using 2-{aminoethyl-[1-(S)-(ethoxycarbonyl)ethylamino]}-phosphinoylamino}-propionic acid ethyl ester: ¹H NMR (300 MHz, CDCl₃) δ 1.28 (m, 6H), 1.41 (m, 6H), 1.82 (s, 3H), 1.93 (m, 2H), 2.16 (s, 3H), 3.12 (t, 3H), 3.43 (d, 2H), 3.36-3.56 (m, 2H), 3.77 (s, 3H), 4.01 (m, 2H), 4.16 (m, 4H), 4.44 (s, 2H), 5.21 (s, 2H), 5.53 (brt, 1H), 5.66 (brs, 1H), 7.73 (s, 1H) ppm; ³¹P (121.4 MHz, CDCl₃) δ 27.41; MS (*m/z*) 628 [M+H]⁺.

Example 11



5

{2-[4-(4-Hydroxy-1-6-methoxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-aminosulfonylaminoethyl}-phosphonic acid: To a

solution of {2-[4-(6-methoxy-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyl]-N-(2-trimethylsilyl-

10 ethoxycarbonyl)aminosulfonylaminoethyl}-phosphonic acid diethyl ester (JX-2260-151) (40 mg, 0.052 mmol) in CH₃CN (0.5 mL) and DMF (0.5 mL) were sequentially added 2,6-lutidine (0.073 mL, 0.624 mmol) and

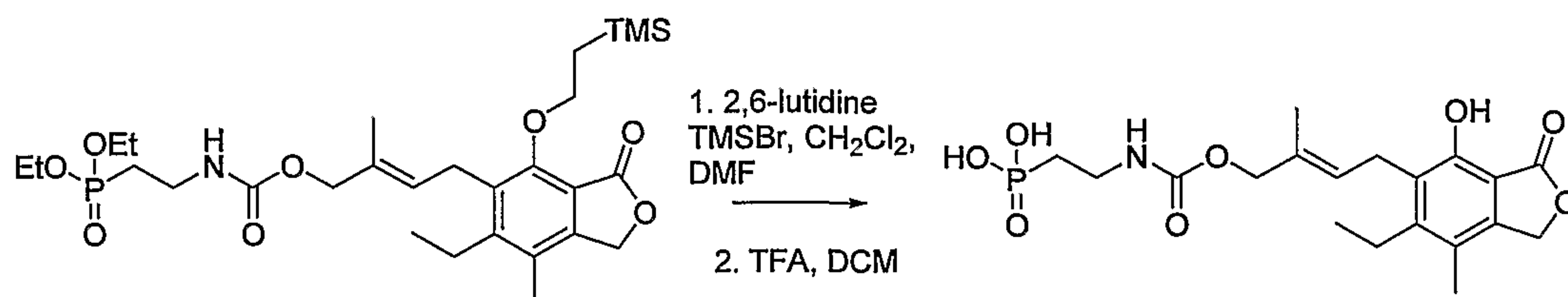
bromotrimethylsilane (0.067 mL, 0.52 mmol), and the mixture was stirred at room temperature for 3 days. MeOH (3 mL) was added and the mixture was

15 concentrated. The residue was purified by RP HPLC to afford the corresponding phosphonic acid. This intermediate was dissolved in CH₂Cl₂ containing 20 % trifluoroacetic acid (2.5 mL). After 6 hr at 0 °C, pyridine (1.5 mL) was added and the mixture was concentrated. The residue was purified by RP HPLC to afford the desired product (10 mg, 41 %). ¹H NMR (300 MHz, CD₃OD) δ 1.87

20 (s, 3H), 1.94 (m, 2H), 2.16 (s, 3H), 3.19 (q, 2H, J= 8.4 Hz), 3.44 (m, 4H), 3.79 (s, 3H), 5.26 (s, 2H), 5.50 (brt, 1H) ppm; ³¹P (121.4 MHz, CD₃OD) δ 24.24; MS (m/z) 565 [M+H]⁺.

Example 12

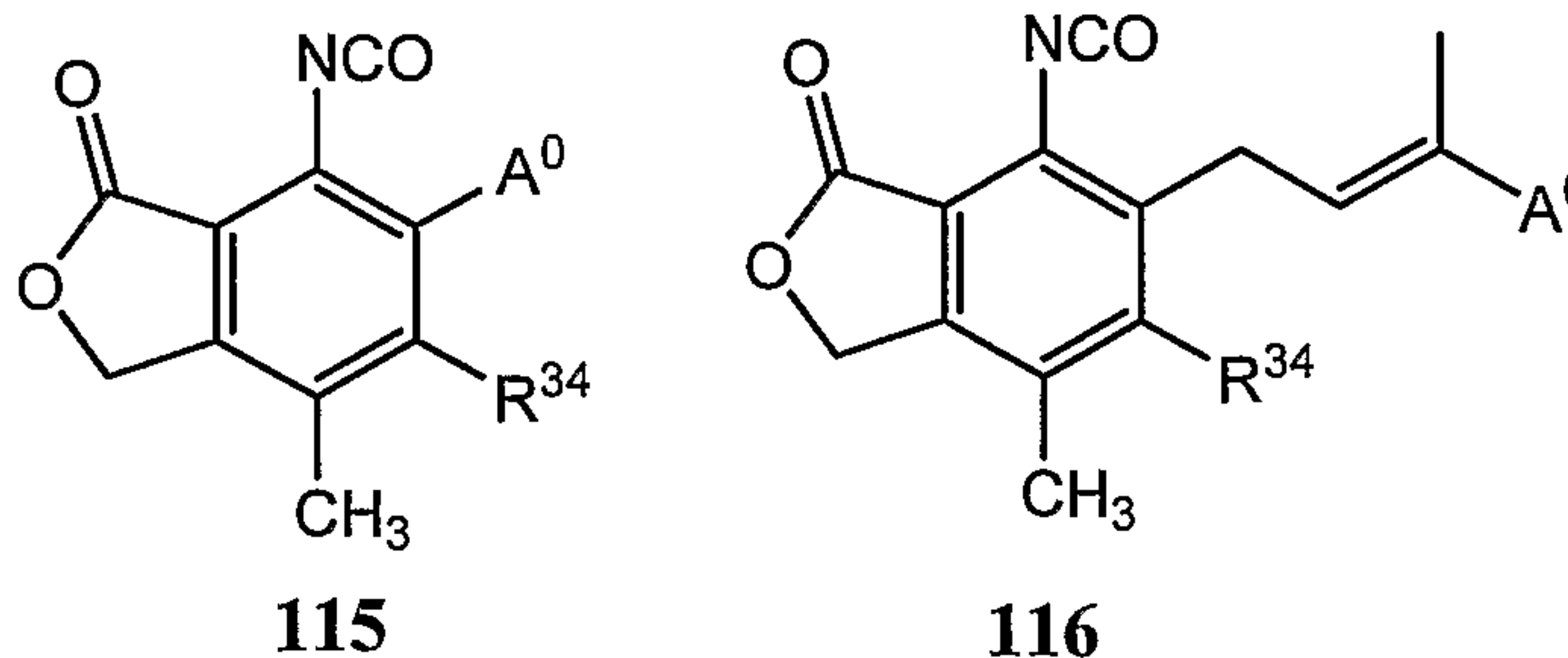
25



{2-[4-(6-Ethyl-4-hydroxy-7-methyl-3-oxo-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyloxycarbonylamino]-ethyl}-phosphonic acid: To a solution of {2-[4-(6-ethyl-7-methyl-3-oxo-4-(2-trimethylsilyl-ethoxy)-1,3-dihydro-isobenzofuran-5-yl)-2-methyl-but-2-enyloxycarbonylamino]-ethyl}-phosphonic acid diethyl ester (30 mg, 0.05 mmol) in CH₂Cl₂ (0.5 mL) and DMF (0.5 mL) were sequentially added 2,6-lutidine (0.21 mL, 1.73 mmol) and bromotrimethylsilane (0.171 mL, 1.32 mmol), and the mixture was stirred at room temperature for 18 hr. MeOH (3 mL) was added and the mixture was concentrated. The residue was purified by RP HPLC to afford the corresponding phosphonic acid. This intermediate was dissolved in CH₂Cl₂ containing 10 % trifluoroacetic acid (2 mL). After 1 hr at 0 °C, pyridine (0.3 mL) was added and the mixture was concentrated. The residue was purified by RP HPLC to afford the desired product (9 mg, 42 %). ¹H NMR (300 MHz, CD₃OD) δ 1.15 (t, 3H, J= 7.8 Hz), 1.84 (s, 3H), 1.81-1.92 (m, 2H), 2.20 (s, 3H), 2.74 (q, 2H, J= 7.8 Hz), 3.37 (m, 2H), 3.51 (d, 2H, J= 6.3 Hz), 4.41 (s, 2H), 5.27 (s, 2H), 5.39 (brt, 1H) ppm; ³¹P (121.4 MHz, CD₃OD) δ 24.20; MS (*m/z*) 428 [M+H]⁺.

Example 13

Compounds of Formula XI and XII can be prepared from the corresponding 4-isocyanato compounds of Formula 115 and 116:



The intermediate compounds of Formula **115** and **116** can be prepared from the corresponding alcohols as disclosed in U.S. Patent No. 5,380,879.

Example 14

The following illustrate representative pharmaceutical dosage forms, containing a compound of the invention ('Compound X'), for therapeutic or prophylactic use in humans.

5	<u>(i) Tablet 1</u>	<u>mg/tablet</u>
	Compound X=	100.0
	Lactose	77.5
	Povidone	15.0
	Croscarmellose sodium	12.0
10	Microcrystalline cellulose	92.5
	Magnesium stearate	<u>3.0</u>
		300.0
	<u>(ii) Tablet 2</u>	<u>mg/tablet</u>
15	Compound X=	20.0
	Microcrystalline cellulose	410.0
	Starch	50.0
	Sodium starch glycolate	15.0
	Magnesium stearate	<u>5.0</u>
20		500.0
	<u>(iii) Capsule</u>	<u>mg/capsule</u>
	Compound X=	10.0
	Colloidal silicon dioxide	1.5
25	Lactose	465.5
	Pregelatinized starch	120.0
	Magnesium stearate	<u>3.0</u>
		600.0
30	<u>(iv) Injection 1 (1 mg/mL)</u>	<u>mg/mL</u>
	Compound X= (free acid form)	1.0
	Dibasic sodium phosphate	12.0

	Monobasic sodium phosphate	0.7
	Sodium chloride	4.5
	1.0 N Sodium hydroxide solution	
	(pH adjustment to 7.0-7.5)	q.s.
5	Water for injection	q.s. ad 1 mL

	<u>(v) Injection 2 (10 mg/mL)</u>	<u>mg/mL</u>
	Compound X= (free acid form)	10.0
	Monobasic sodium phosphate	0.3
10	Dibasic sodium phosphate	1.1
	Polyethylene glycol 400	200.0
	0.1 N Sodium hydroxide solution	
	(pH adjustment to 7.0-7.5)	q.s.
	Water for injection	q.s. ad 1 mL

15

	<u>(vi) Aerosol</u>	<u>mg/can</u>
	>Compound X=	20.0
	Oleic acid	10.0
	Trichloromonofluoromethane	5,000.0
20	Dichlorodifluoromethane	10,000.0
	Dichlorotetrafluoroethane	5,000.0

The above formulations may be obtained by conventional procedures well known in the pharmaceutical art.

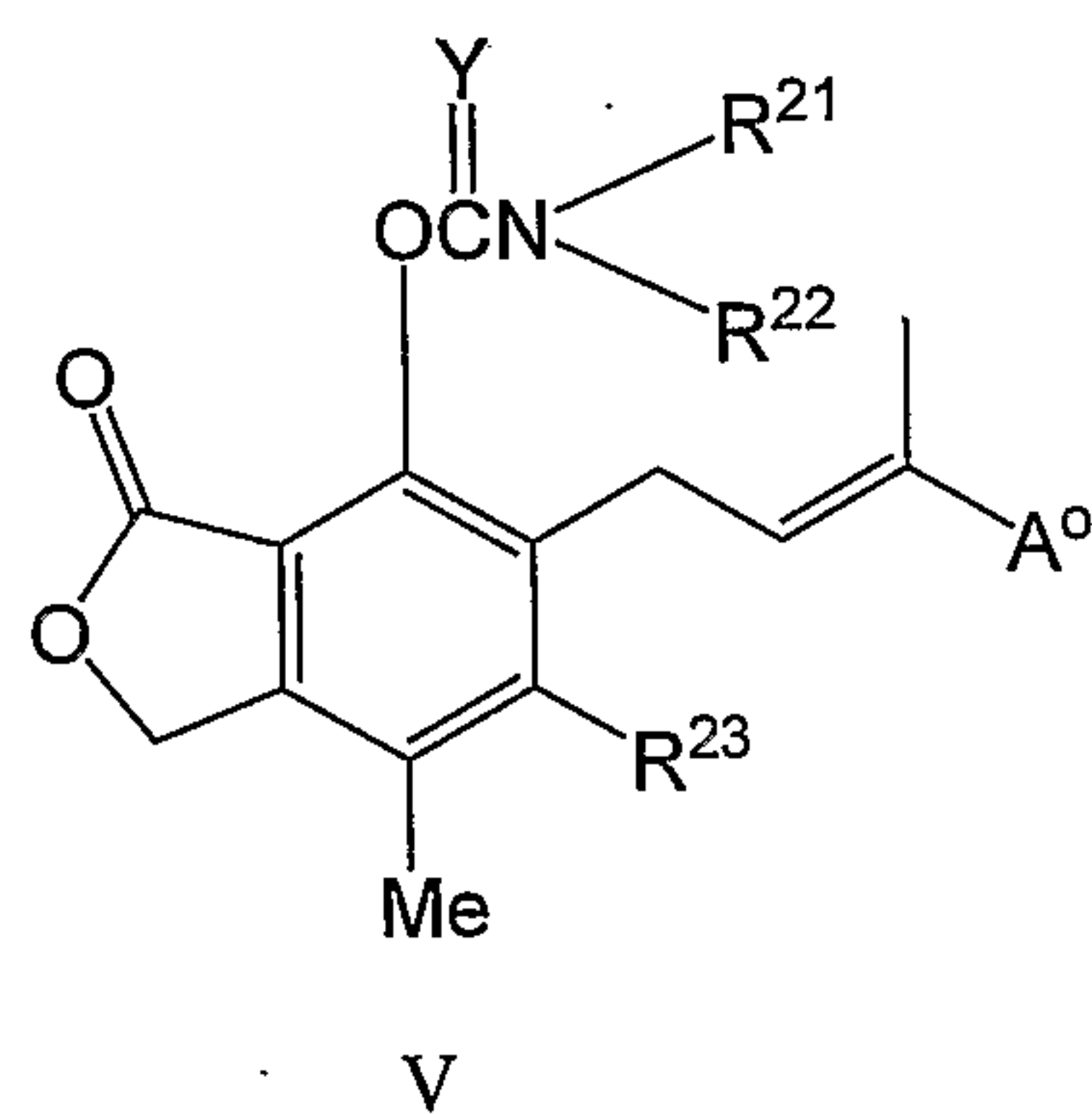
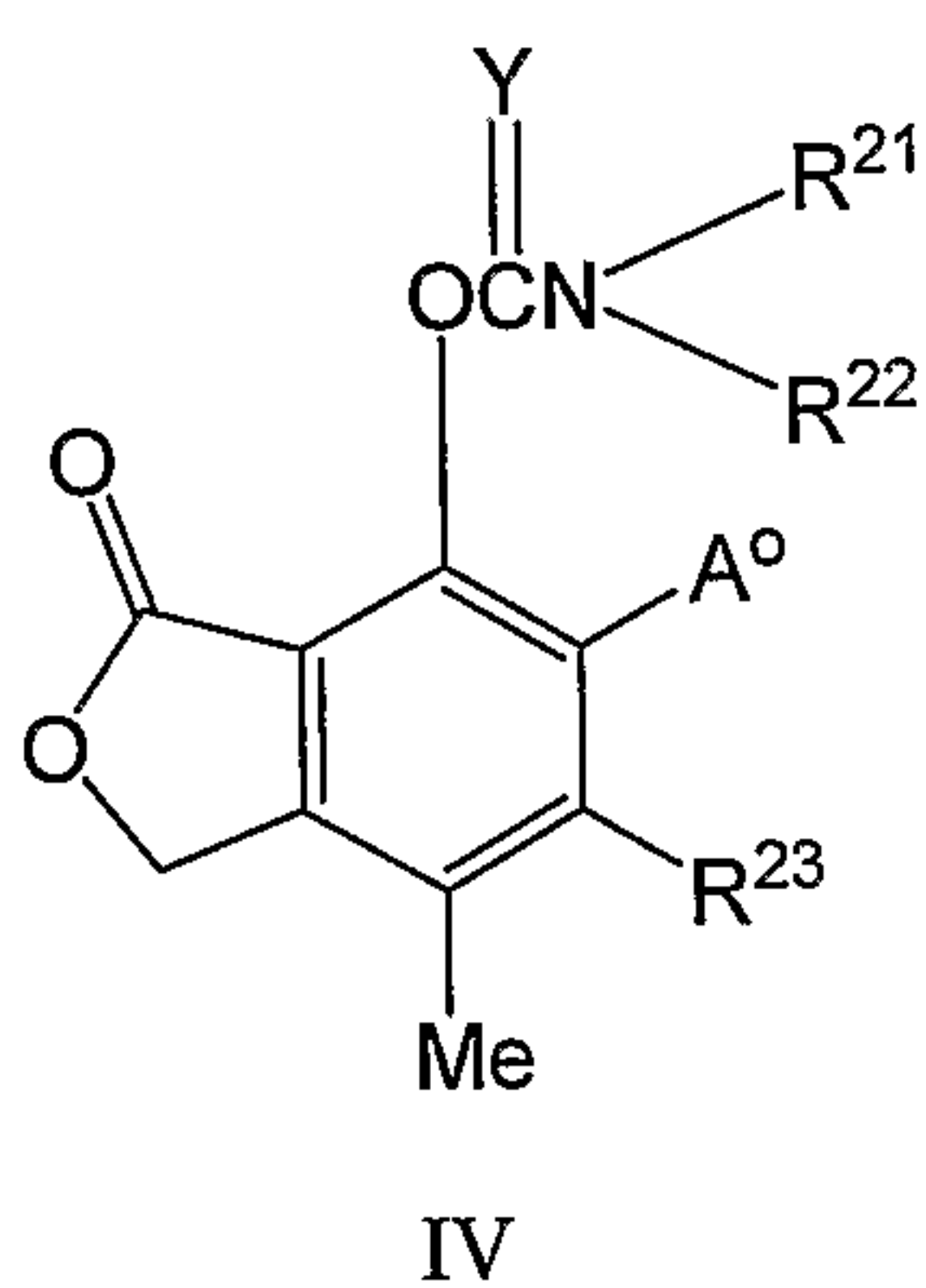
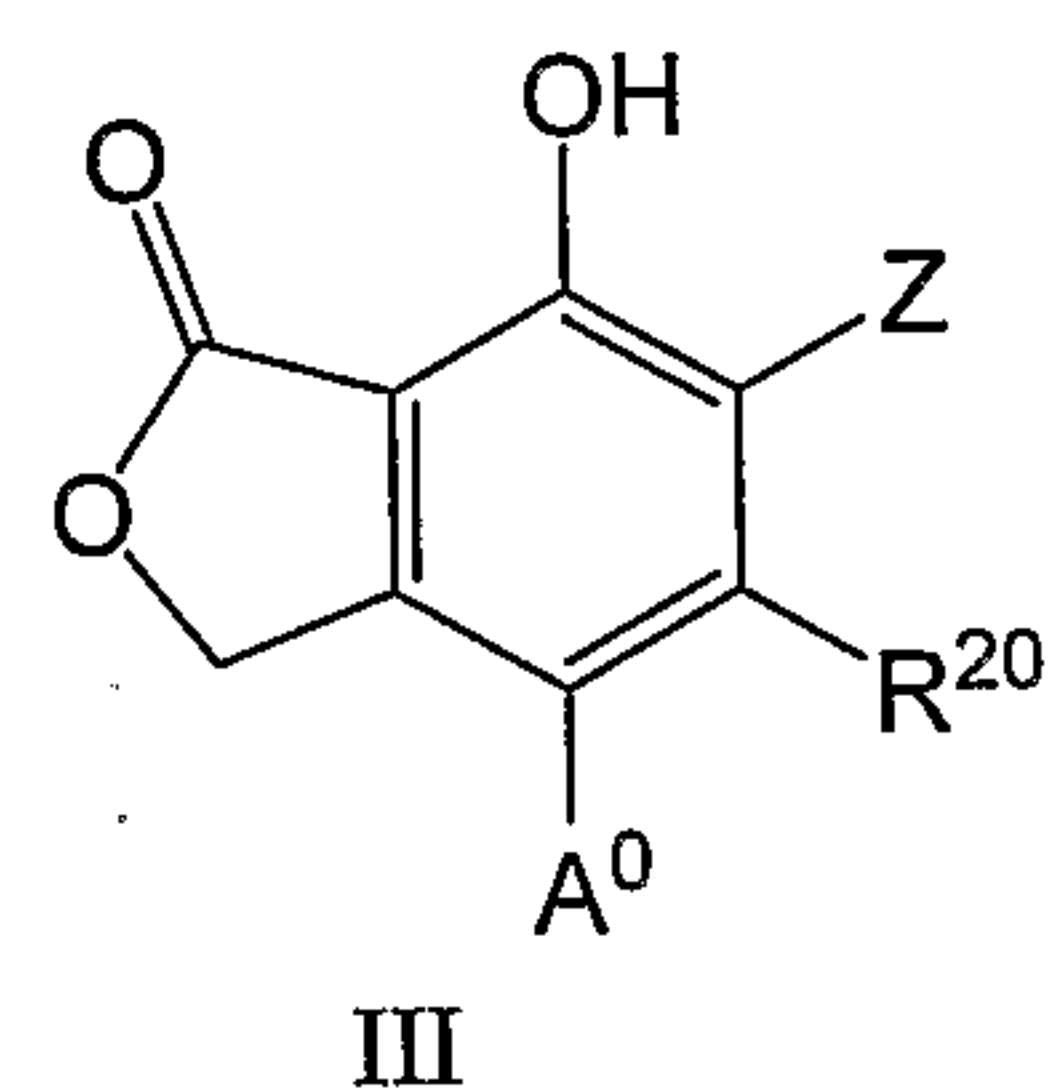
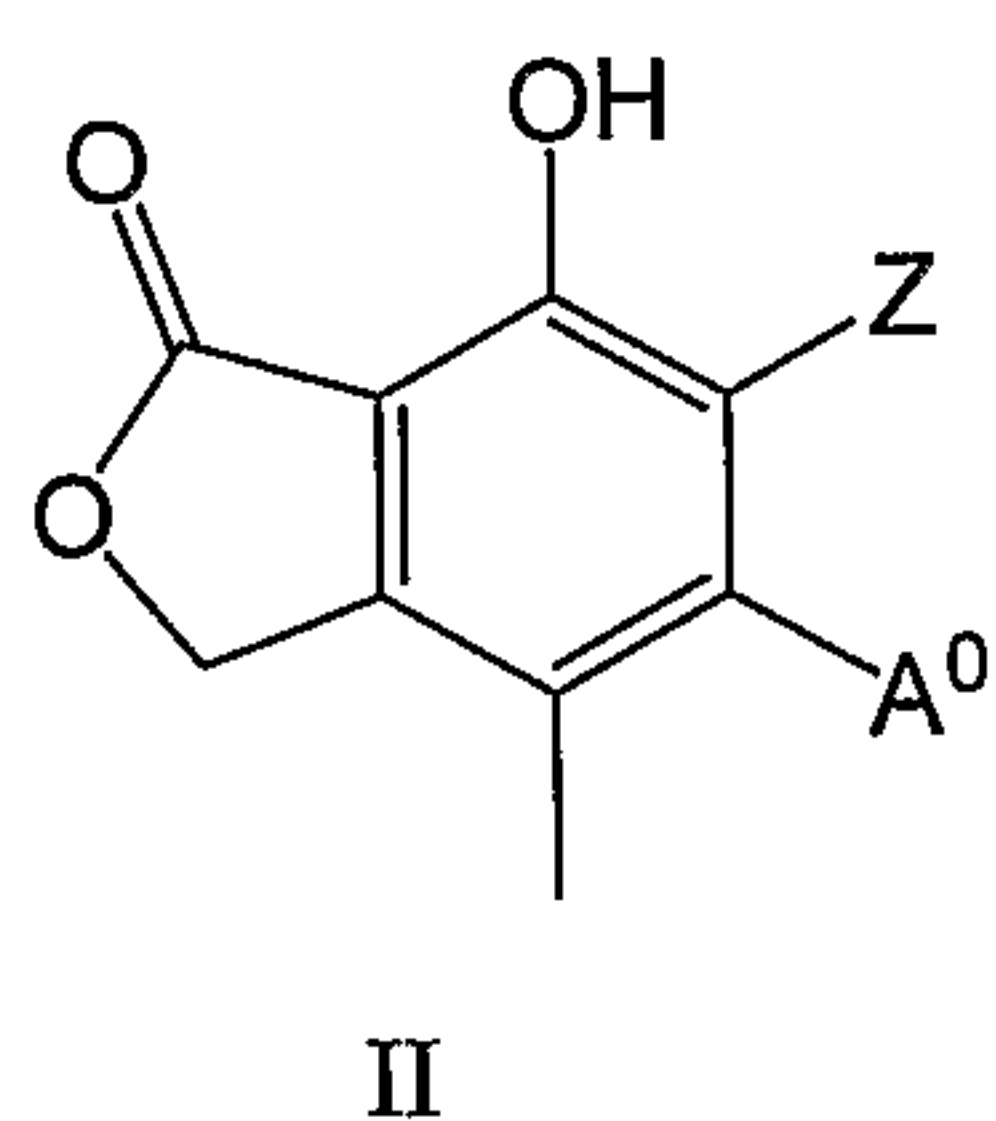
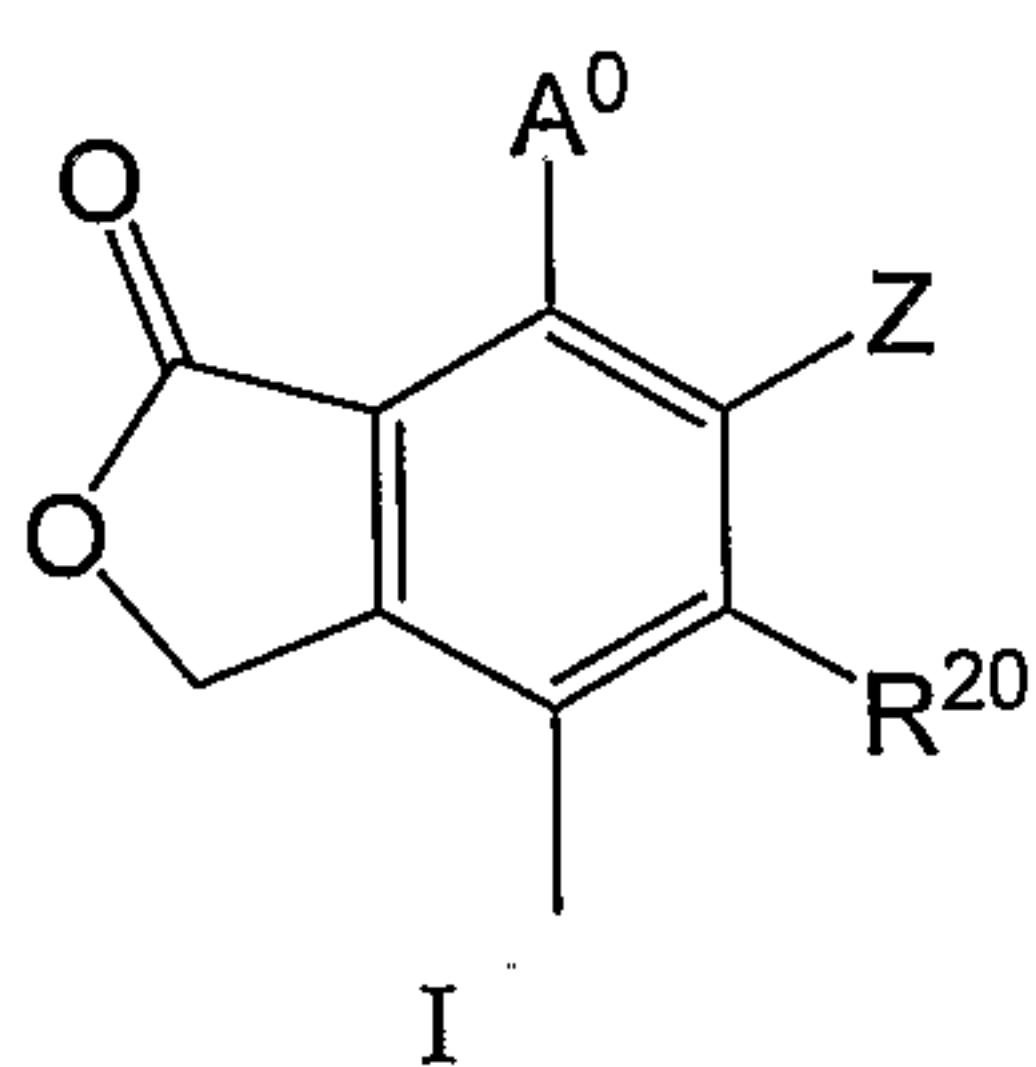
25 All literature and patent citations listed herein are hereby expressly incorporated by reference at the locations of their citation. Specifically cited sections or pages of the above cited works are incorporated by reference with specificity.

30 The invention has been described in detail sufficient to allow one of ordinary skill in the art to make and use the subject matter of the following embodiments. It is apparent that certain modifications of the methods and compositions of the following embodiments can be made within the scope and spirit of the invention.

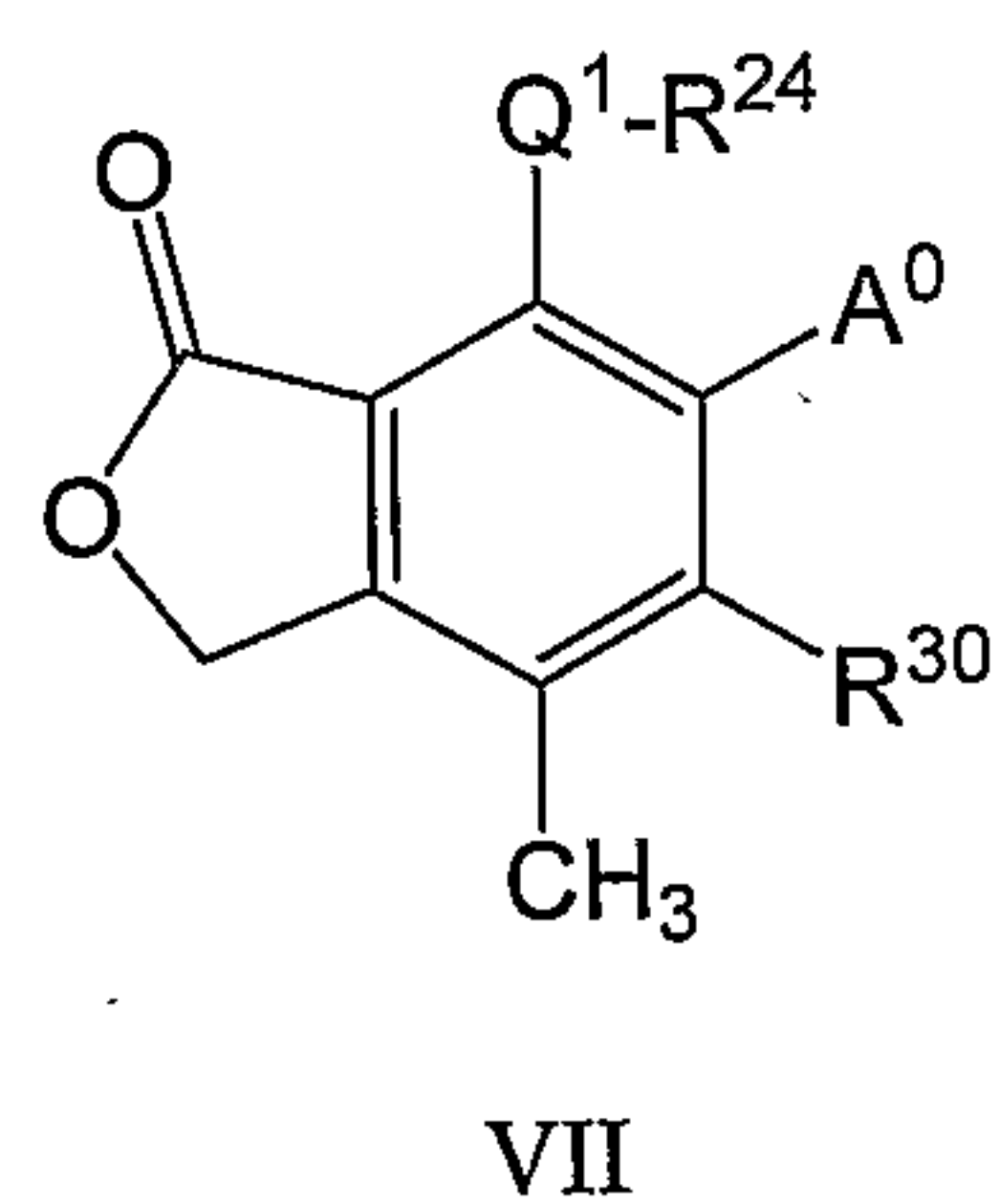
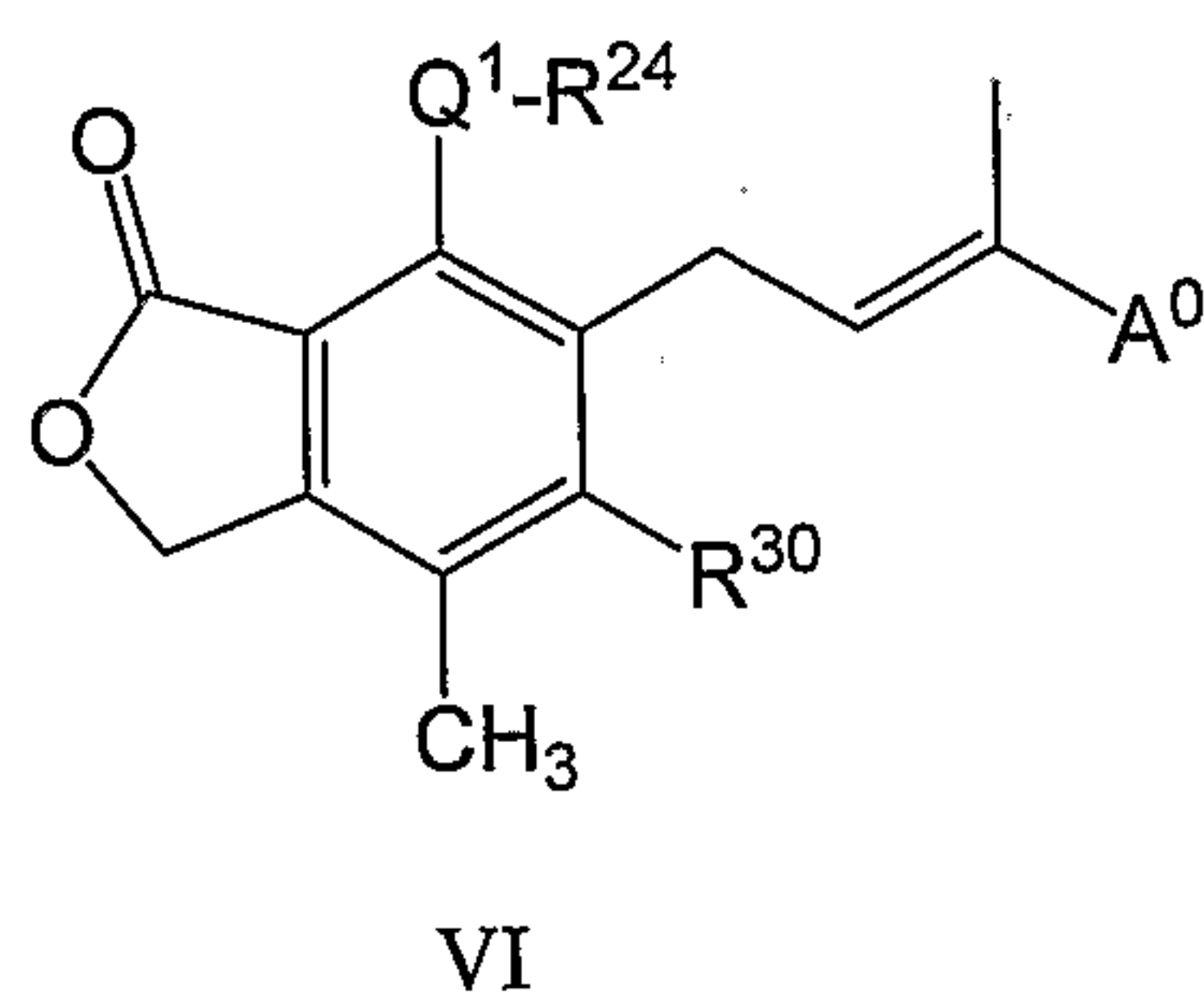
CLAIMS

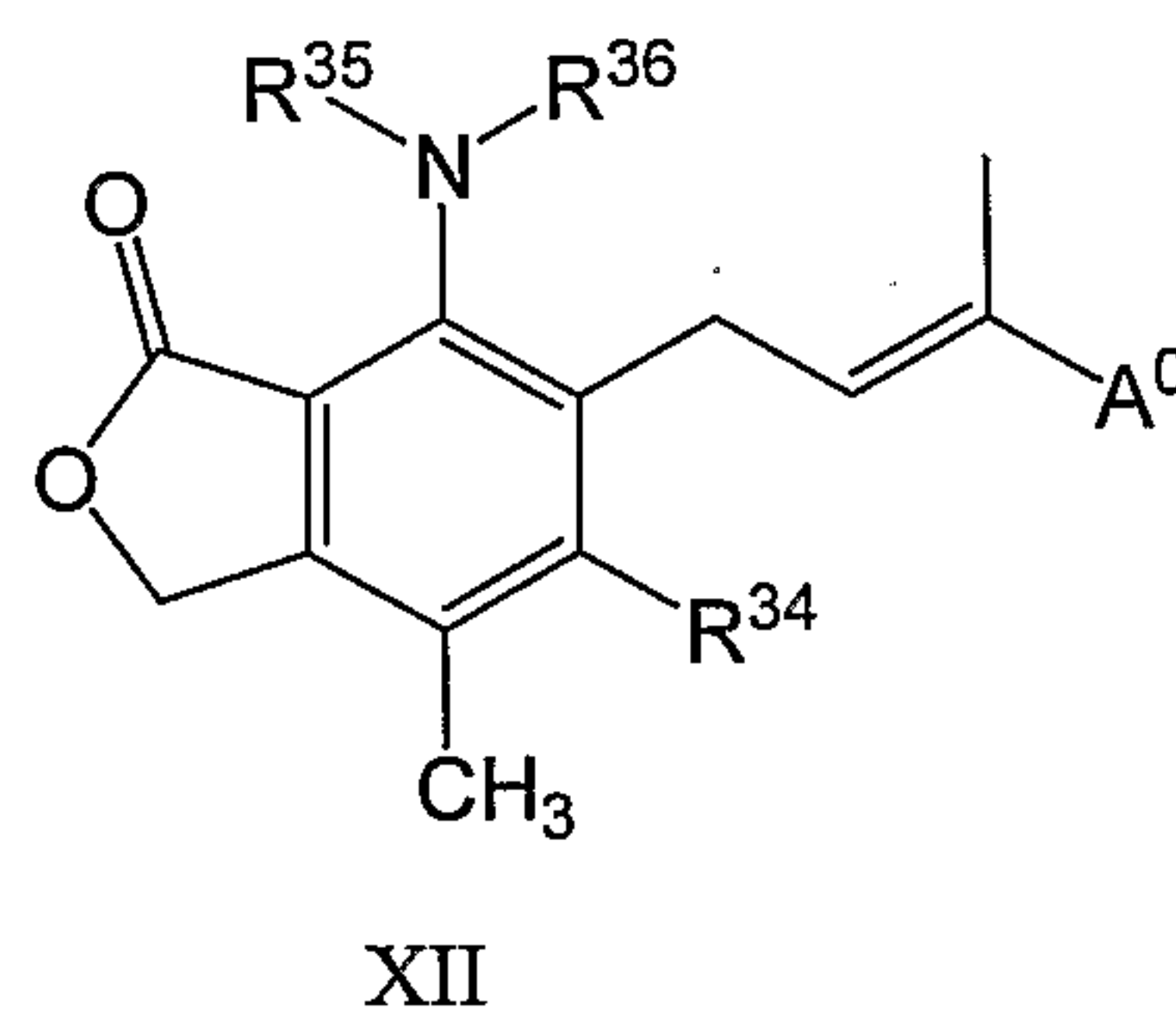
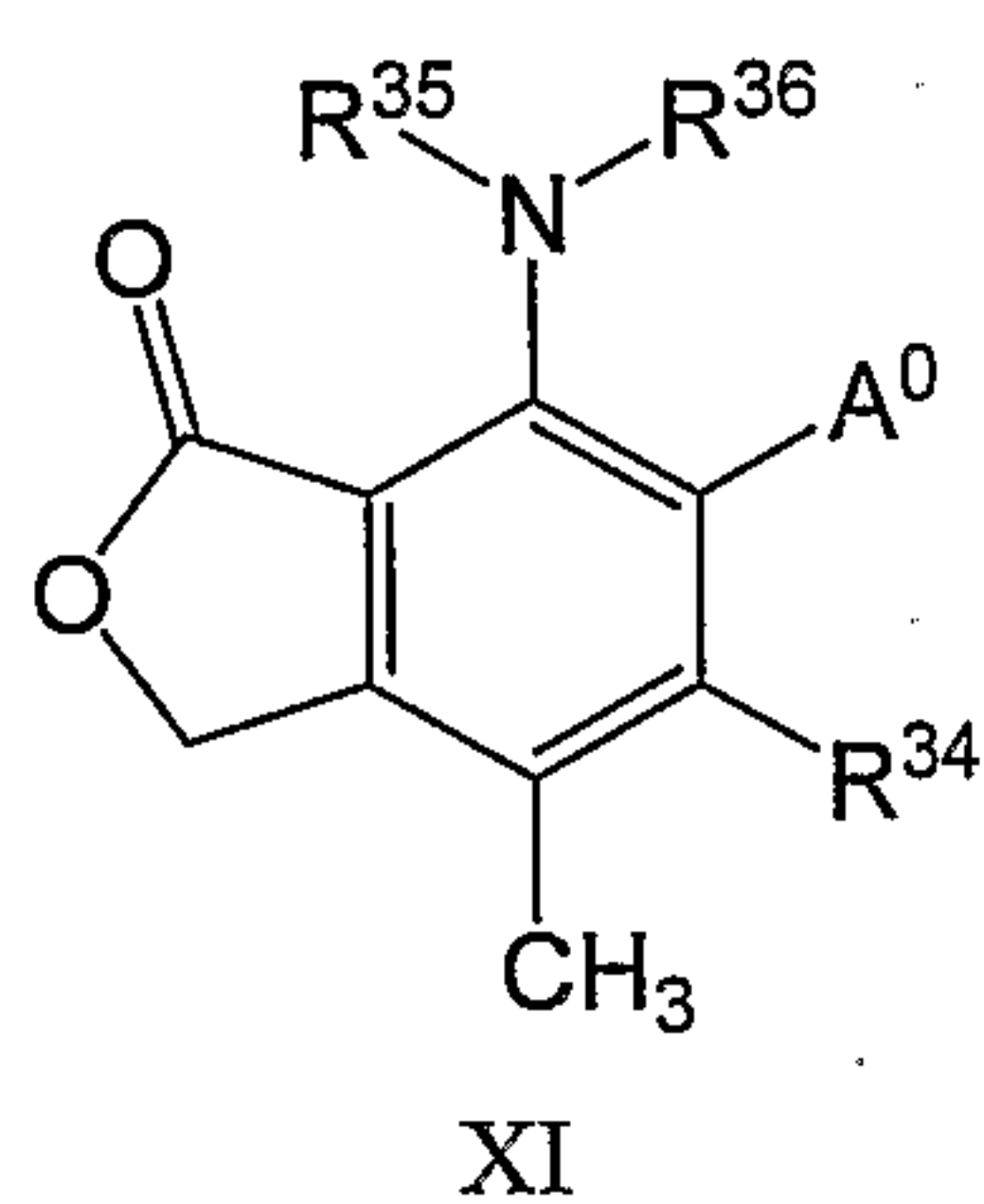
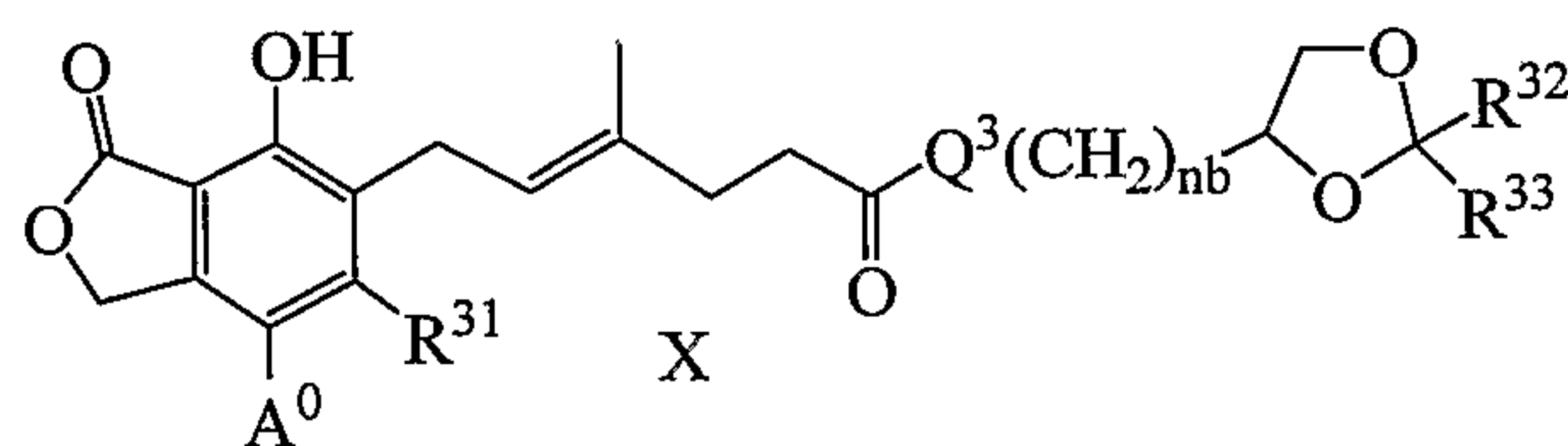
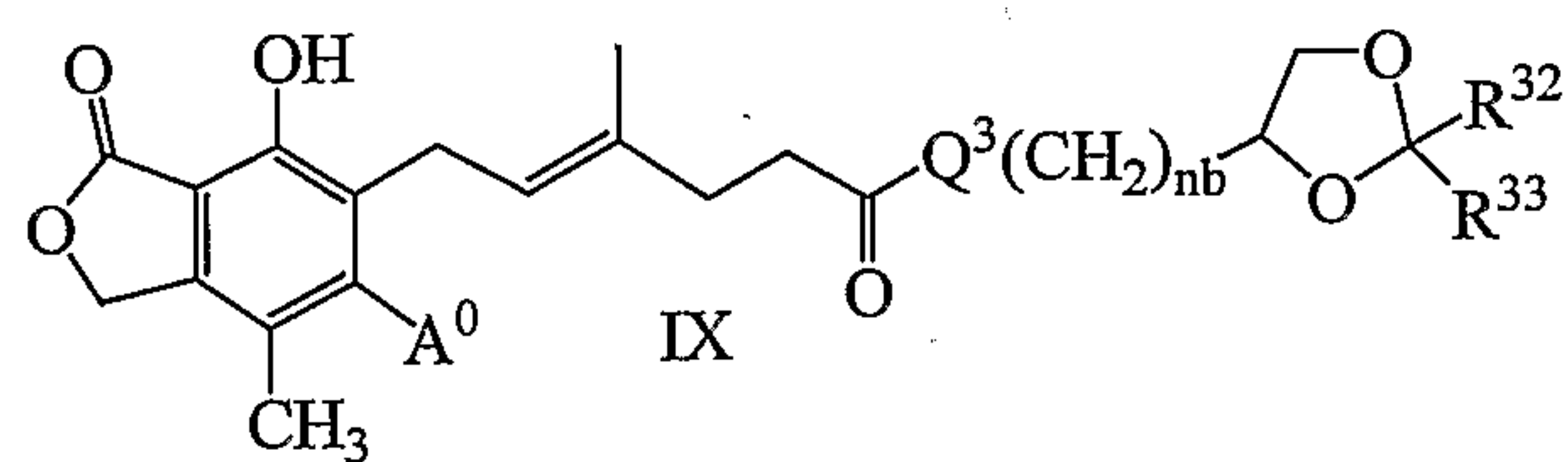
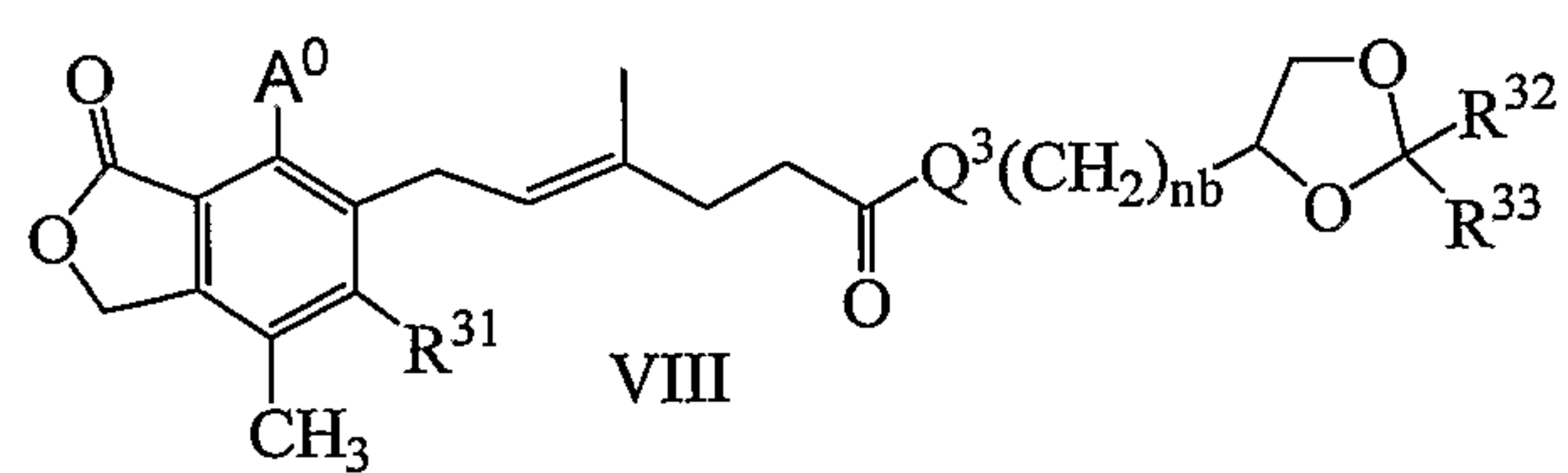
We claim:

- 5 1. A compound of any one of formulae I-XII:



10

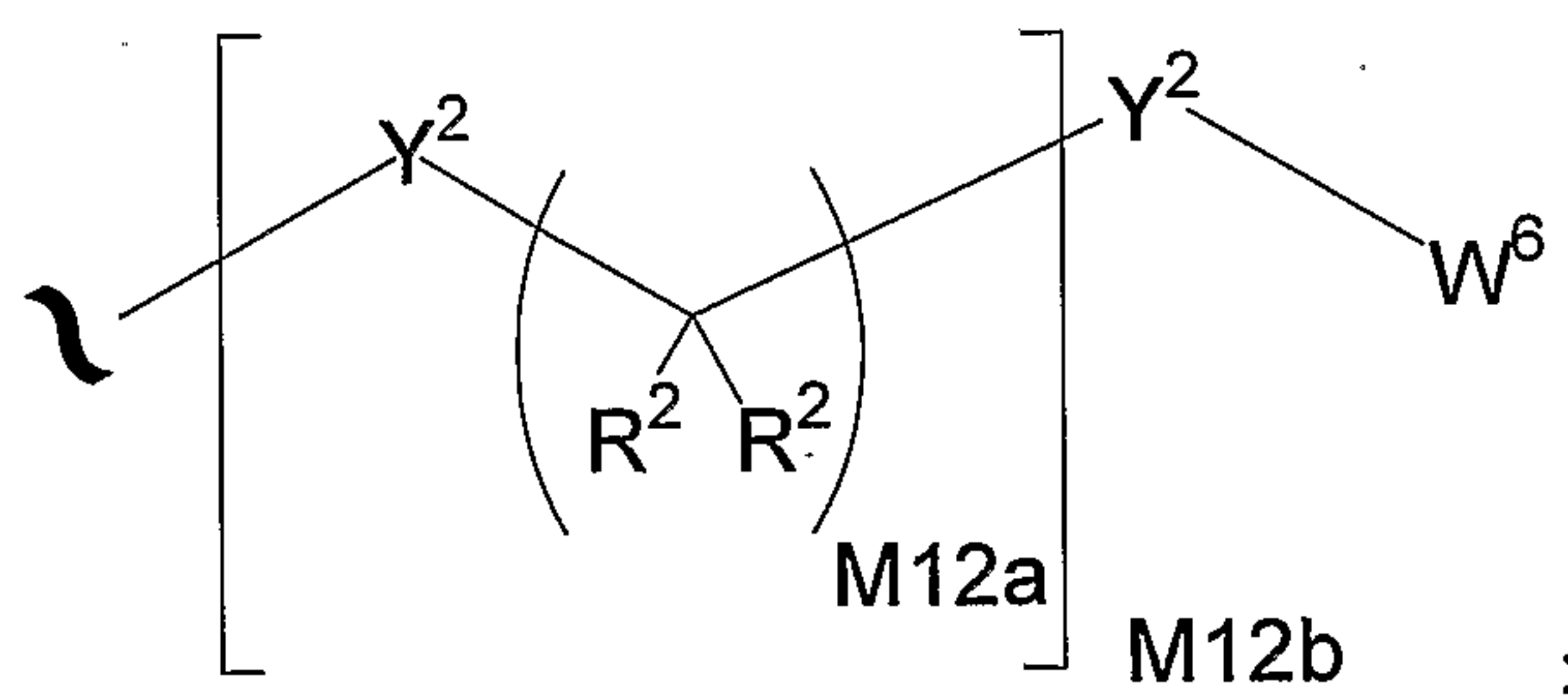




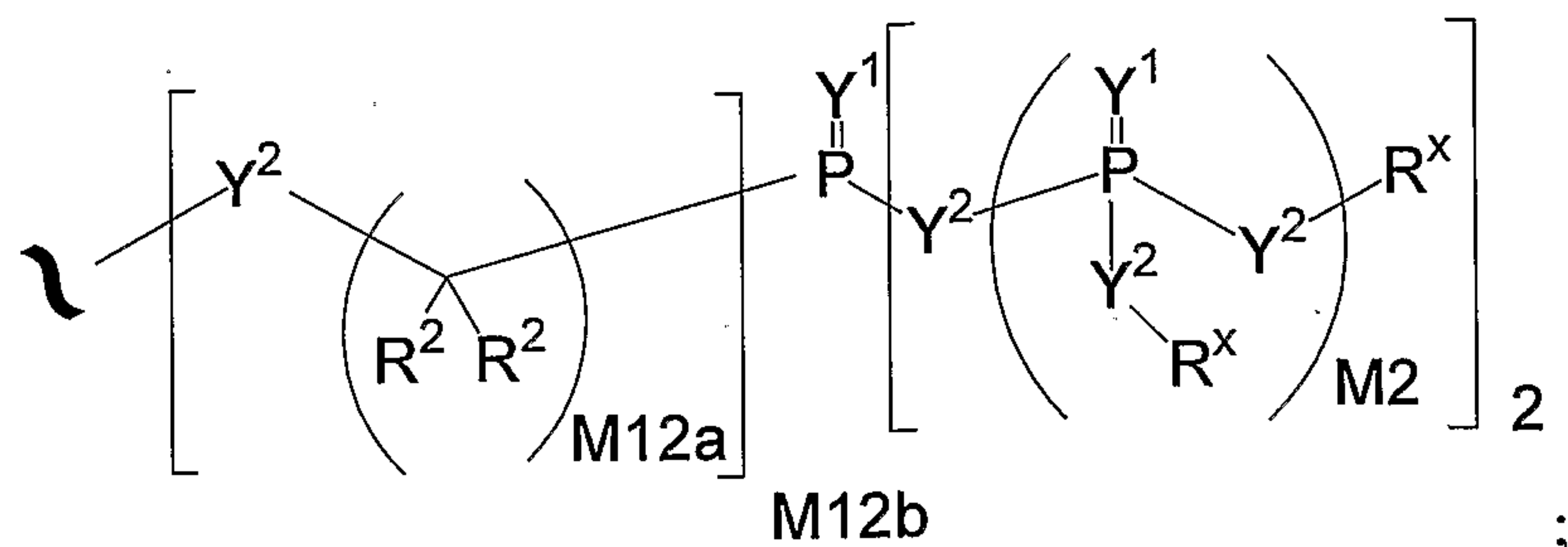
wherein:

A^0 is A^1 ;

A^1 is:



A³ is:

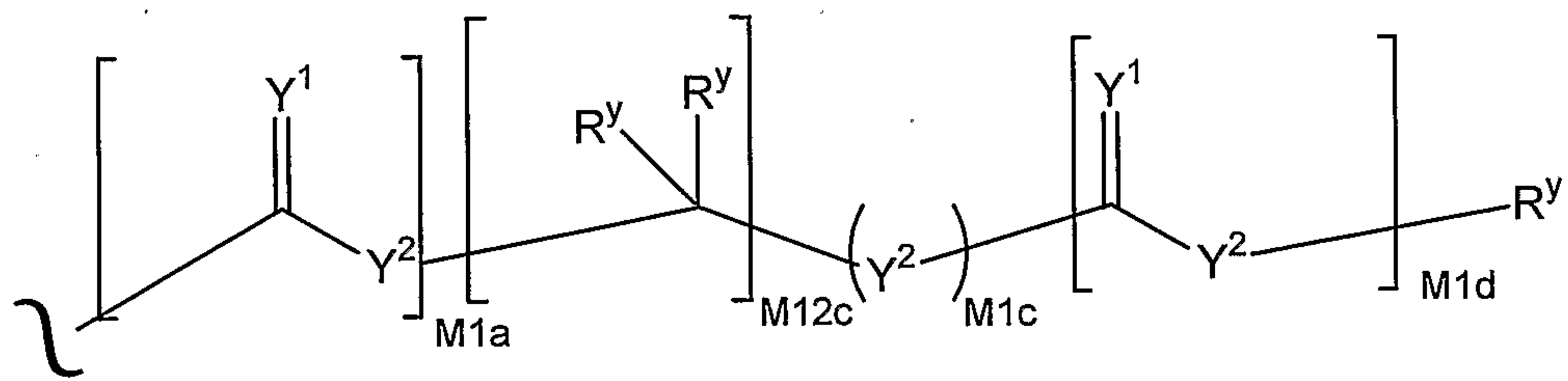


Y¹ is independently O, S, N(R^x), N(OR^x), or N(N(R^x)(R^x));

Y² is independently a bond, O, N(R^x), N(OR^x), N(N(R^x)(R^x)), or

5 -S(O)_{M2}-; and when Y² joins two phosphorous atoms Y² can also be C(R²)(R²);

R^x is independently H, R², W³, a protecting group, or the formula:



R^y is independently H, W³, R² or a protecting group;

R² is independently H, R³ or R⁴ wherein each R⁴ is independently

10 substituted with 0 to 3 R³ groups;

R³ is R^{3a}, R^{3b}, R^{3c} or R^{3d}, provided that when R³ is bound to a heteroatom, then R³ is R^{3c} or R^{3d};

R^{3a} is F, Cl, Br, I, -CN, N₃ or -NO₂;

R^{3b} is Y¹;

15 R^{3c} is -R^x, -N(R^x)(R^x), -SR^x, -S(O)R^x, -S(O)₂R^x, -S(O)(OR^x), -S(O)₂(OR^x), -OC(Y¹)R^x, -OC(Y¹)OR^x, -OC(Y¹)(N(R^x)(R^x)), -SC(Y¹)R^x, -SC(Y¹)OR^x, -SC(Y¹)(N(R^x)(R^x)), -N(R^x)C(Y¹)R^x, -N(R^x)C(Y¹)OR^x, or -N(R^x)C(Y¹)(N(R^x)(R^x));

R^{3d} is -C(Y¹)R^x, -C(Y¹)OR^x or -C(Y¹)(N(R^x)(R^x));

20 R⁴ is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms, or alkynyl of 2 to 18 carbon atoms;

R⁵ is R⁴ wherein each R⁴ is substituted with 0 to 3 R³ groups;

W³ is W⁴ or W⁵;

W⁴ is R⁵, -C(Y¹)R⁵, -C(Y¹)W⁵, -SO₂R⁵, or -SO₂W⁵;

W^5 is carbocycle or heterocycle wherein W^5 is independently substituted with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

$M2$ is 0, 1 or 2;

5 $M12a$ is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

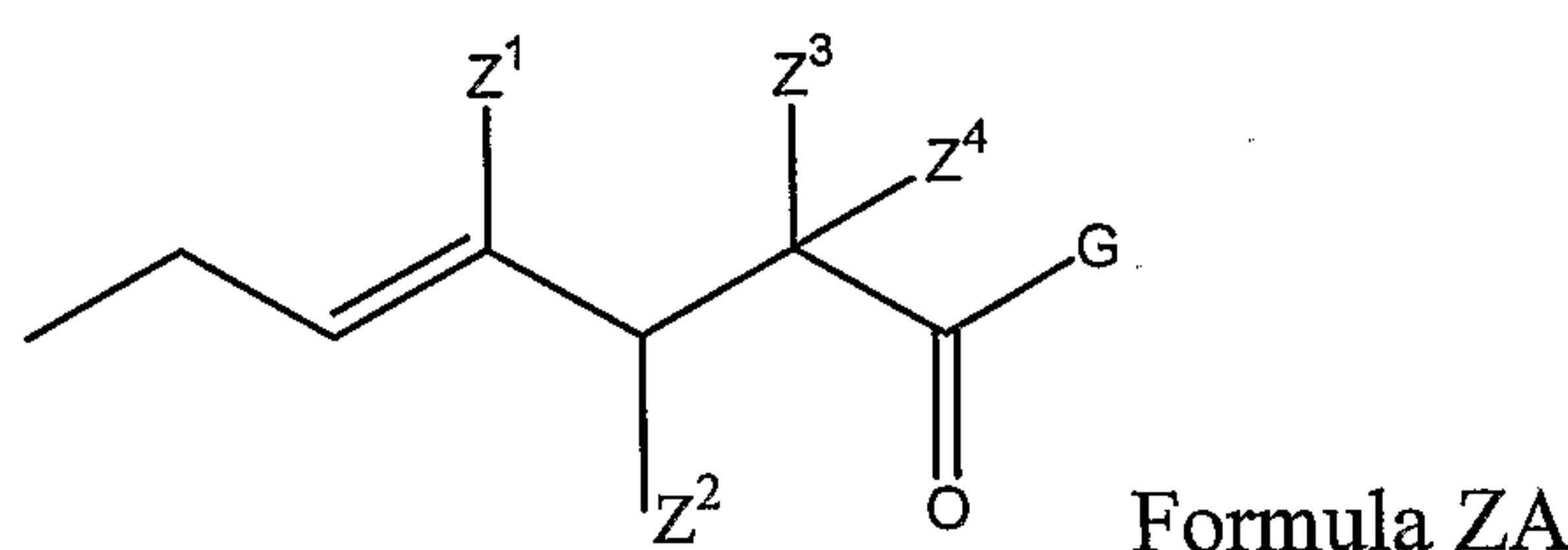
$M12b$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

$M1a$, $M1c$, and $M1d$ are independently 0 or 1;

$M12c$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

R^{20} is ethyl, vinyl, or methoxy;

10 Z is a side chain selected from Formulae ZA, ZB, ZC, ZD, ZE, ZF, ZG, and ZH:



wherein:

Z^1 is hydrogen, alkyl, halo or CF_3 ;

15 Z^2 is hydrogen, alkyl, alkoxy, aryl, or $-CH_2Z^{13}$, where Z^{13} is halo, CN, aryl or heterocycle;

Z^3 is hydrogen, OH, SH, halo, alkyl, alkenyl, alkoxy, aryl, $-P(O)(OCH_3)_2$, $-P(O)(OH)(OCH_3)$, NHZ^{11} , or $-S(O)_m Z^{12}$,

wherein Z^{11} is H, alkyl, acyl, or alkylsulfonyl;

20 Z^{12} is alkyl; and

m is 0, 1, or 2;

Z^4 is hydrogen, OH, halo, alkyl, or aryl, provided that Z^4 is not OH or halo when Z^3 is OH, halo, $-P(O)(OCH_3)_2$, $-P(O)(OH)(OCH_3)$, NHZ^{11} , or SZ^{12} ; or Z^3 and Z^4 taken together with the carbon to which they are attached

25 form cycloalkyl of three to five carbon atoms; and

G is OH, alkoxy, thioalkyl, $-NG^1G^2$, $-O(CH_2)_n NG^1G^2$, or $-O(CH_2)_n N=G^3$, where

n is an integer from 1 to 6,

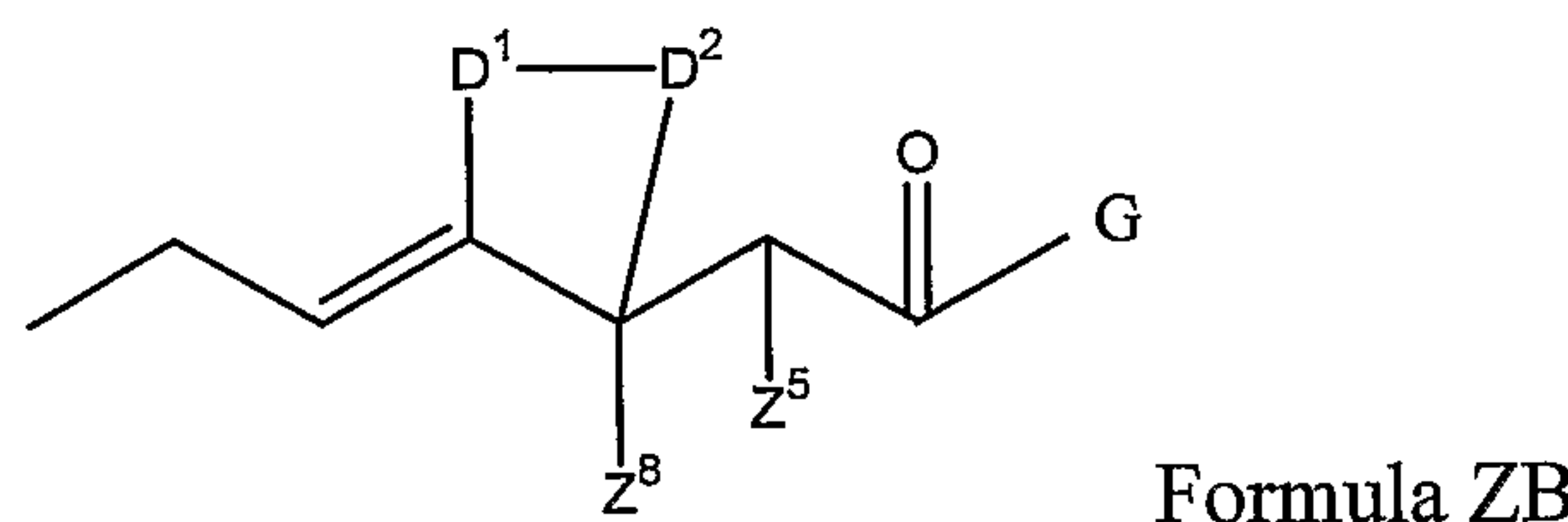
G^1 is hydrogen or alkyl,

30 G^2 is hydrogen or alkyl, and

$=G^3$ is alkylene of four to six carbon atoms, or alkylene of three to five carbon atoms plus one member that is -O-, -S-, or -N(G^4)- where G^4 is hydrogen or alkyl;

provided that when Z^1 is methyl, Z^2 , Z^3 , and Z^4 are not all H; or

5



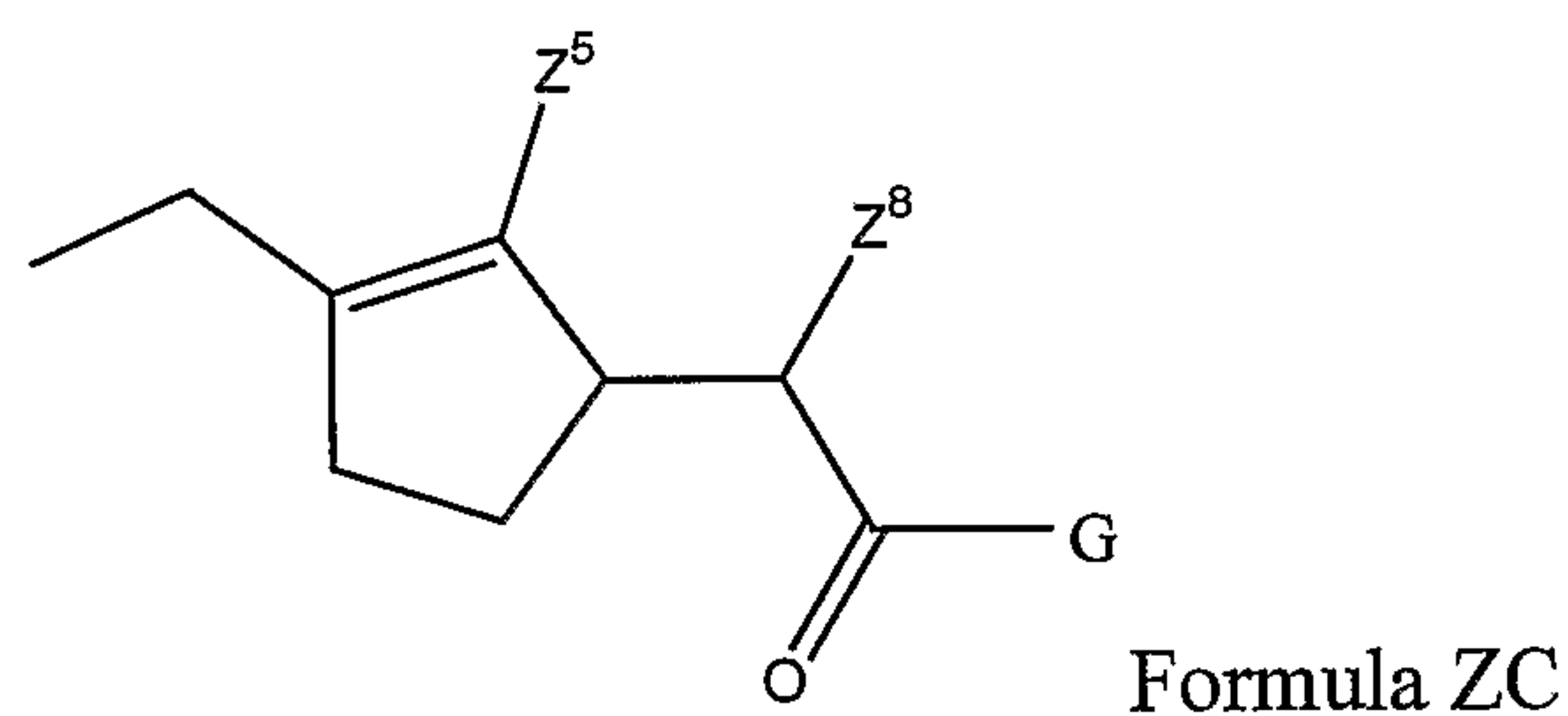
wherein:

Z^5 is hydrogen or alkyl;

Z^8 is hydrogen, alkyl, or forms a double bond with D^2 ;

10 D^1 and D^2 together with their adjacent carbon atoms form an optionally substituted, saturated or unsaturated carbocyclic or heterocyclic ring of 3 to 7 atoms; and

G is as defined above; or

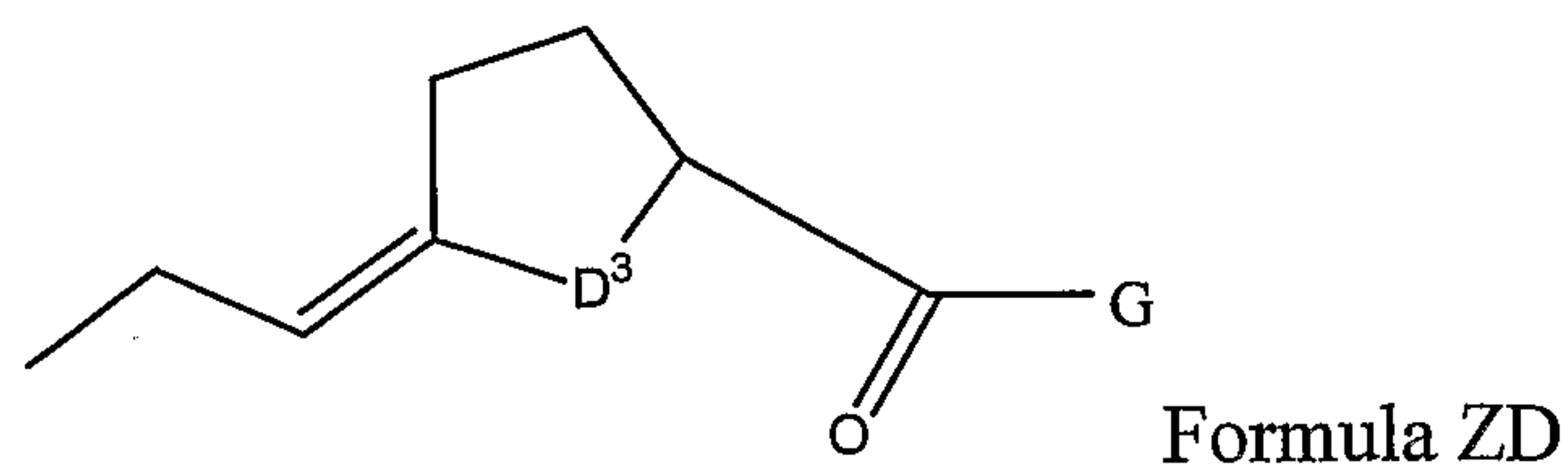


15

wherein:

Z^8 is H or alkyl; and

Z^5 and G are as defined above; or

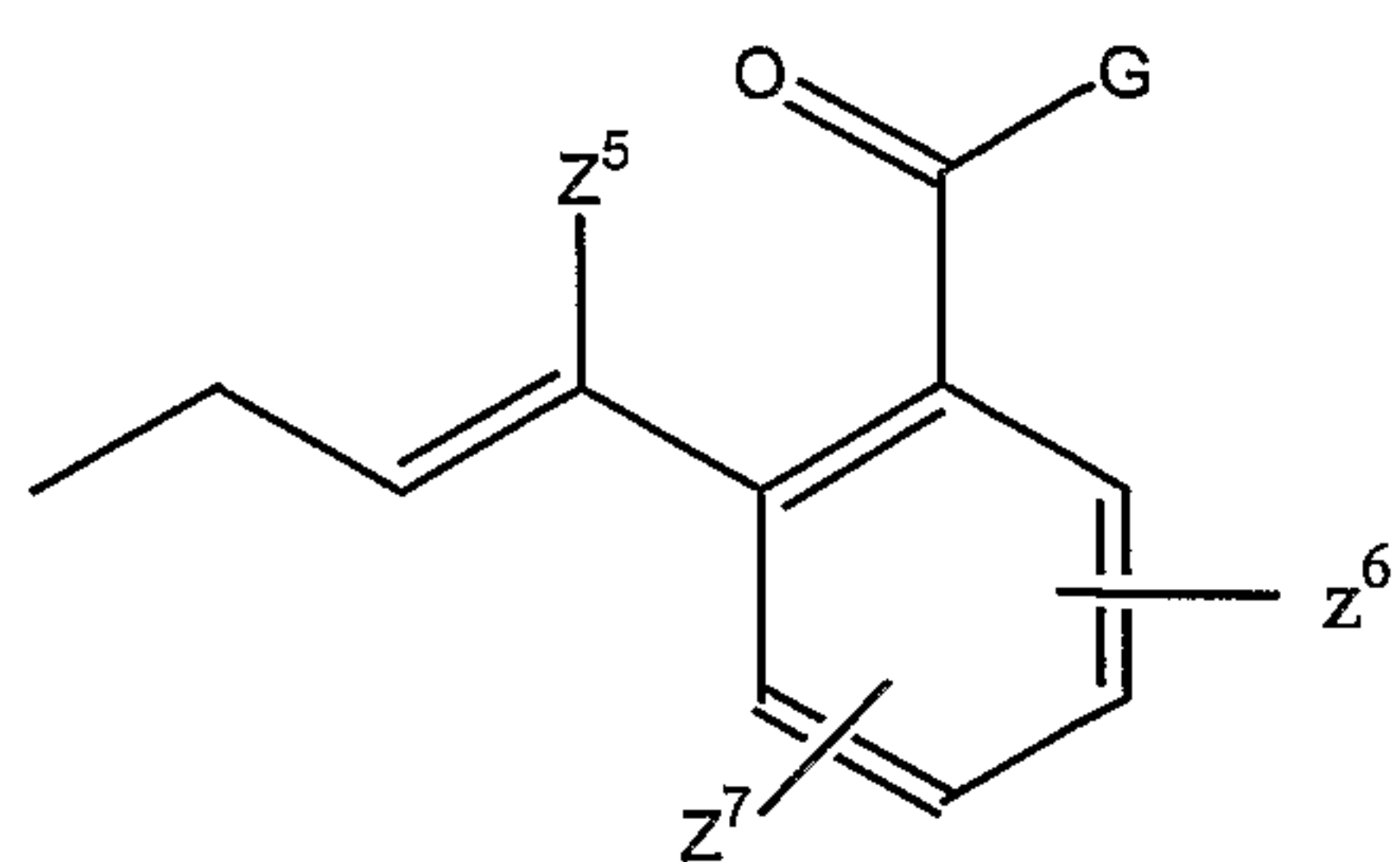


20

wherein:

D^3 is -CH₂- or -CH₂CH₂-; and

G is as defined above; or



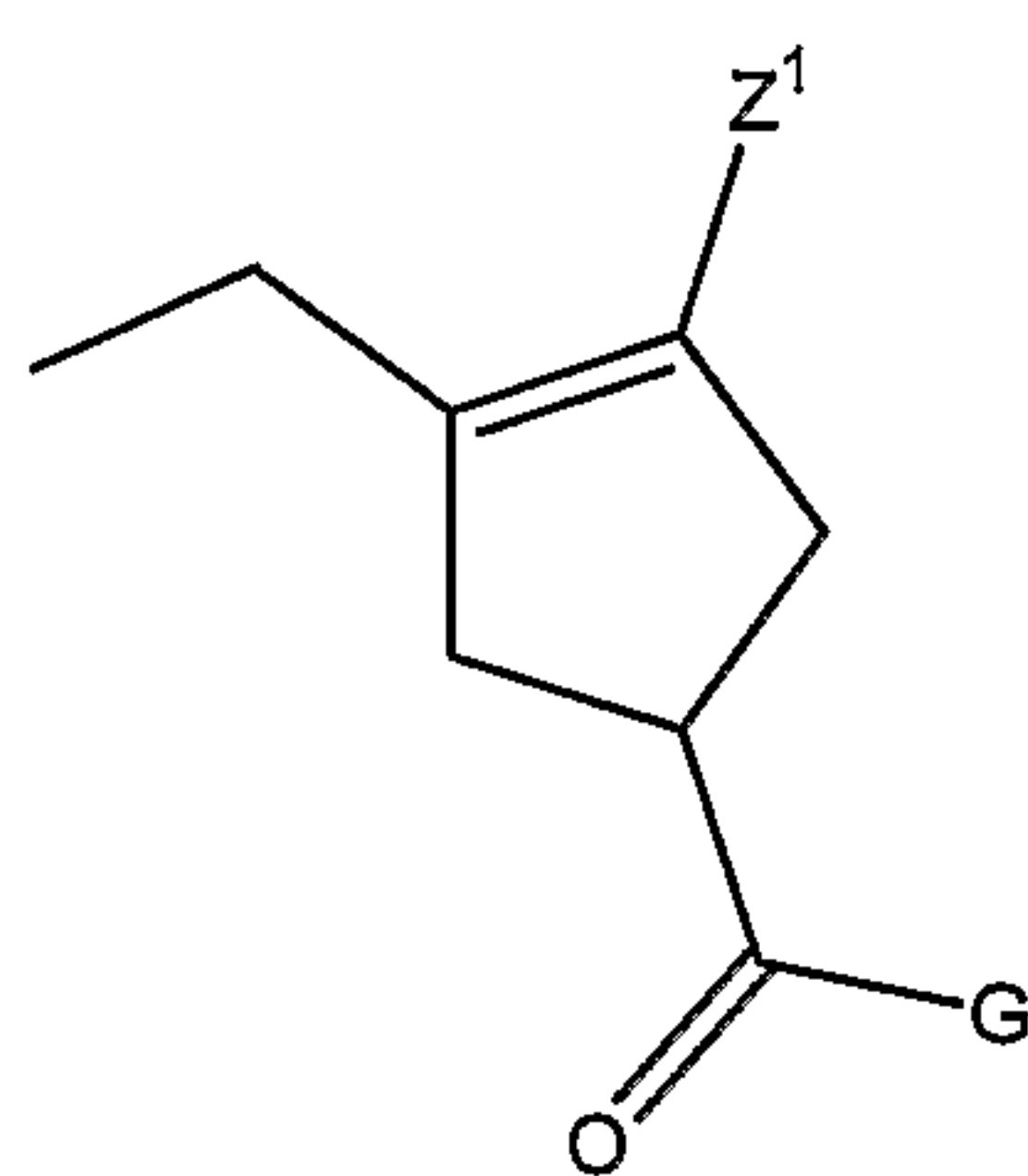
Formula ZE

wherein:

Z^6 is hydrogen, alkyl, alkoxy, $-\text{COOH}$, $-\text{NH}_2$, azido, or halo;

5 Z^7 is hydrogen, alkyl, alkoxy, or halo; and

Z^5 and G are as defined above; or

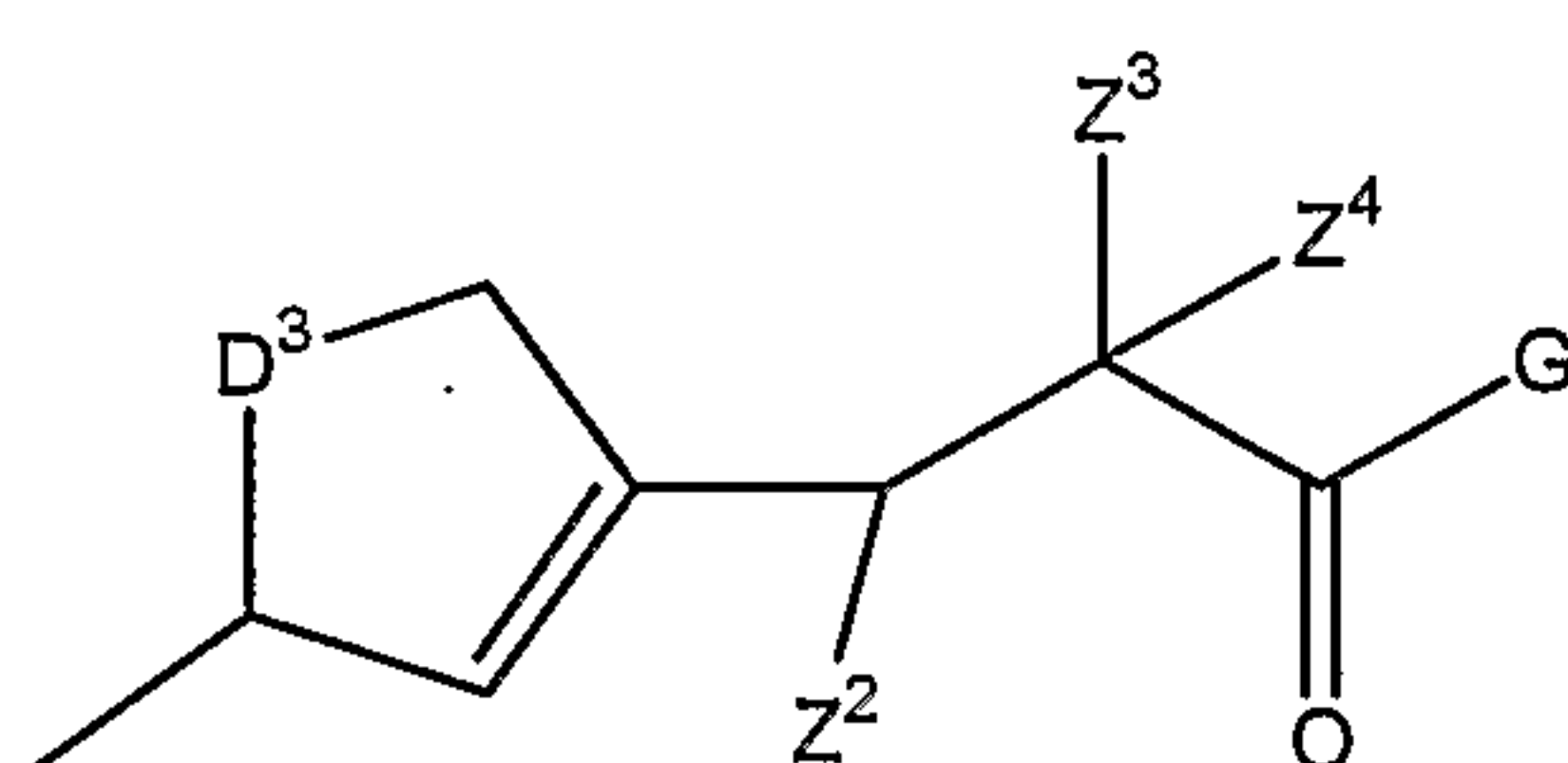


Formula ZF

wherein:

Z^1 and G are as defined above; or

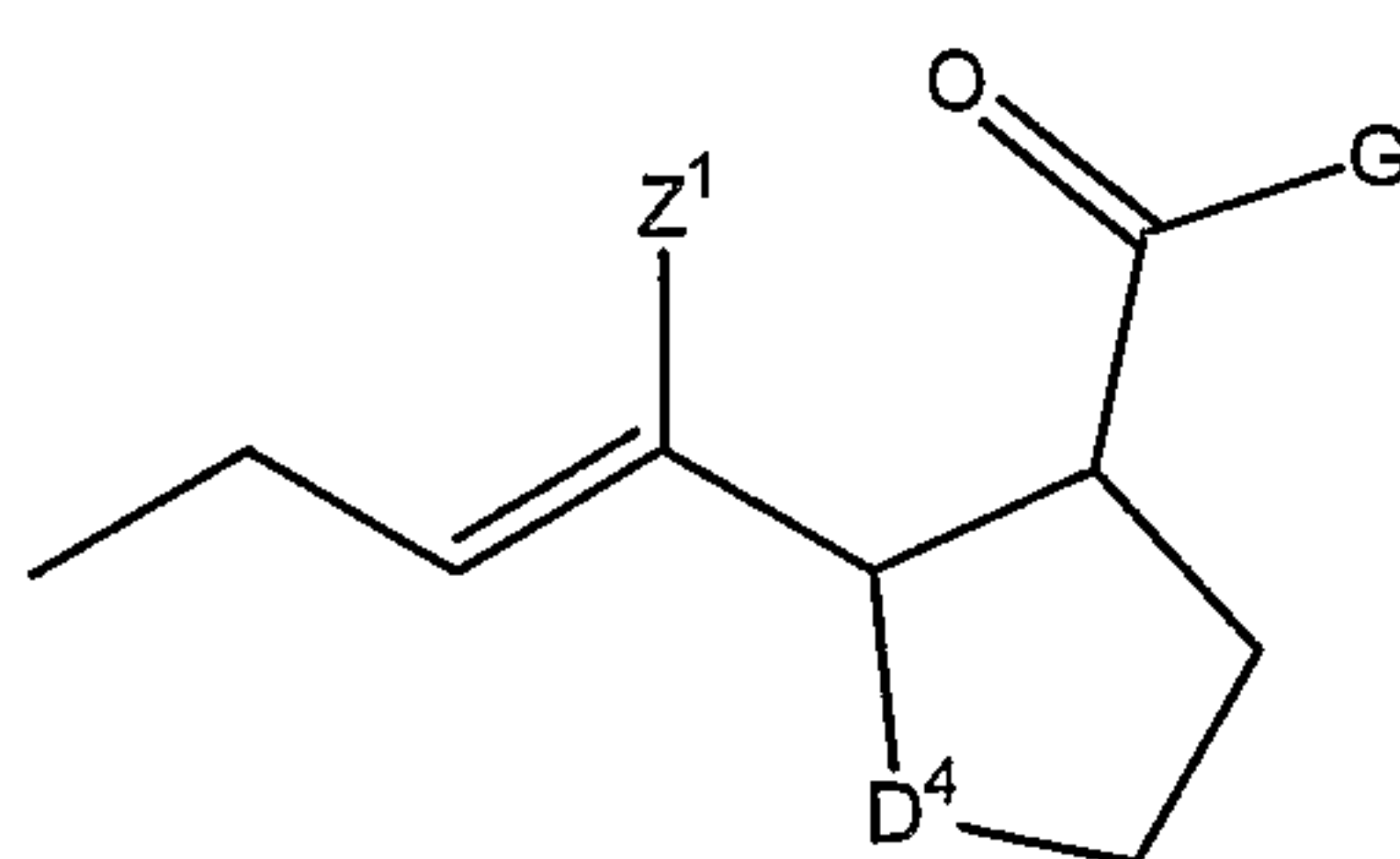
10



Formula ZG

wherein:

D^3 , Z^2 , Z^3 , Z^4 , and G are as defined above; or



Formula ZH

15

wherein:

D^4 is $-\text{CH}_2-$, $-\text{CH}_2\text{CH}_2-$, $-\text{CF}_2\text{CH}_2-$, $-\text{CH}_2\text{CF}_2-$, $-\text{CF}_2\text{CF}_2-$,

-CH₂CH₂CH₂-, -CF₂CH₂CH₂-, -CH₂CF₂CH₂-, -CH₂CH₂CF₂-,
 -CF₂CF₂CF₂-, -O-, -OCH₂-, -OCF₂-, -CH₂CHF-, -CHF-CHF-,
 -CHF-CH₂CH₂-, -CH₂-CHF-CH₂-, -CH₂-CH₂-CHF-, -CHF-CHF-CHF-,
 or -OCHF-; and

5 Z¹ and G are as defined above;

Y is O or S;

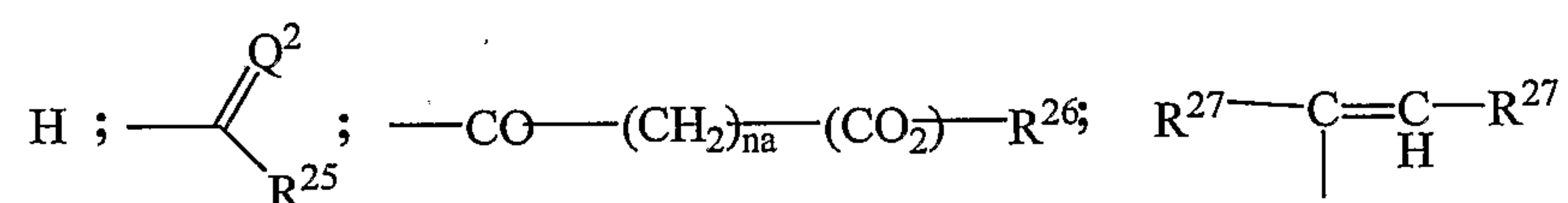
R²¹ is H or alkyl;

R²² is H, alkyl, aryl, or substituted aryl; and

R²³ is methoxy, vinyl, or ethyl;

10 Q¹ is oxygen or sulfur;

R²⁴ is selected from the following:



in which

Q² is oxygen or sulfur;

15 R²⁵ is aryl, substituted aryl, arylalkyl, substituted arylalkyl, alkyl,
 substituted alkyl or -NR²⁸R²⁹, where R²⁸ and R²⁹ are independently H, alkyl,
 substituted alkyl, aryl, substituted aryl, or cycloalkyl;

na is an integer from 0-6;

R²⁶ is H or alkyl;

20 each R²⁷ is independently H or -CO₂ R²⁶; and

R³⁰ is methoxy, ethyl, or vinyl;

provided Q¹ is not oxygen when R²⁴ is H;

R³¹ is methoxy, ethyl, or vinyl;

R³² and R³³ are each independently H, or alkyl;

25 each Q³ is independently O or S; and

nb is an integer of 1-6;

R³⁴ is methoxy, ethyl, or vinyl;

R³⁵ is hydrogen or alkyl;

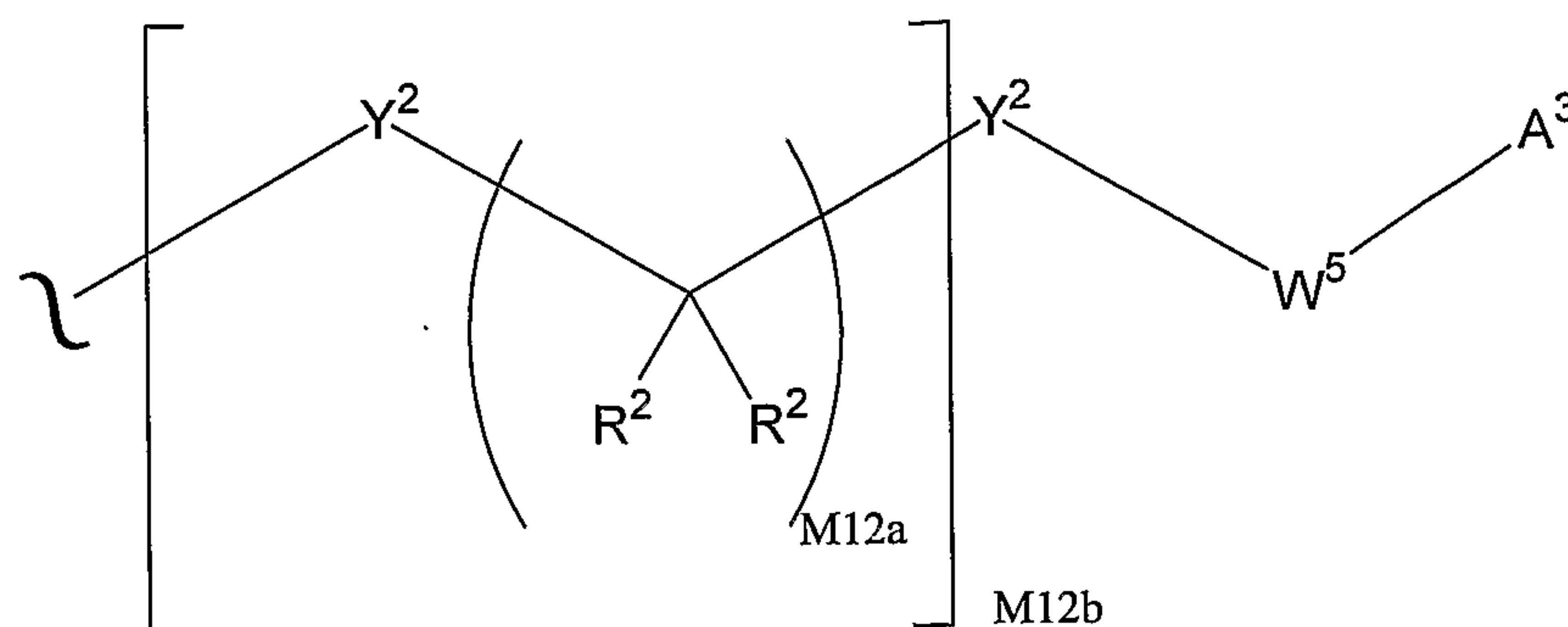
R³⁶ is hydrogen, alkyl, -C(O)R³⁹, -C(O)NR⁴⁰R⁴¹, -CO₂R⁴², or -SO₂R³⁹;

30 R³⁹ is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl;

R⁴⁰ is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl;

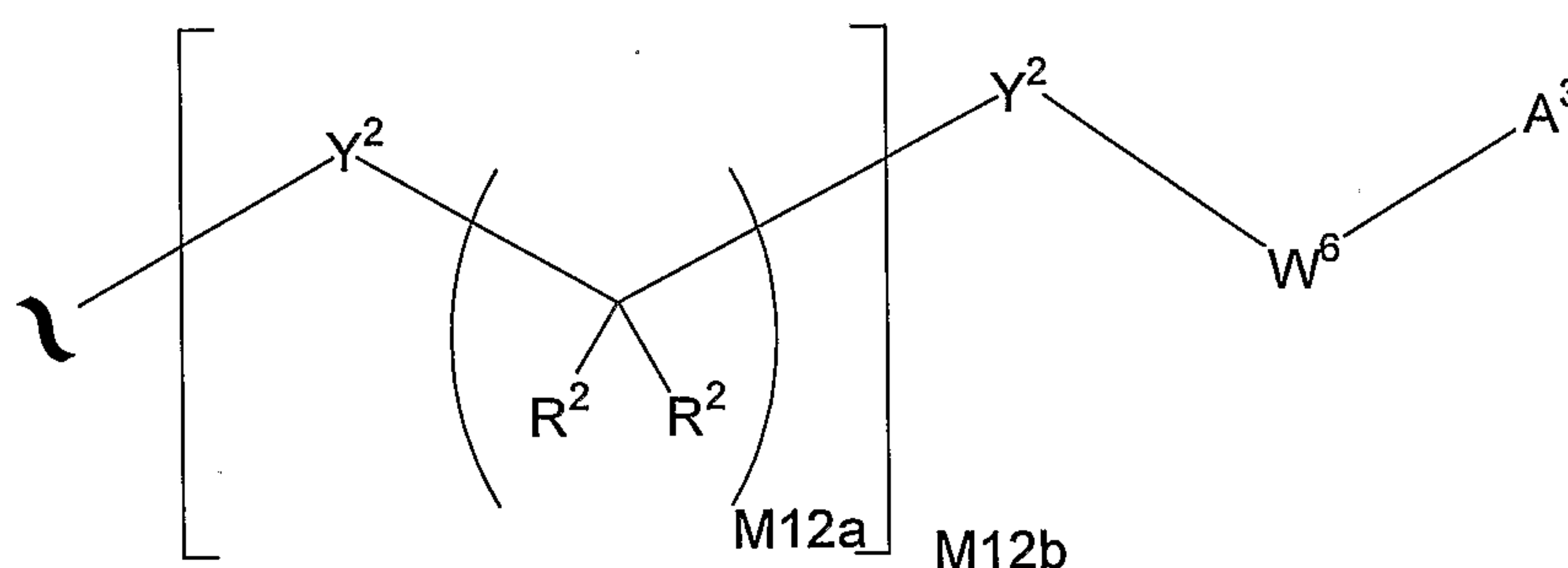
R^{41} is hydrogen, alkyl, substituted alkyl, aryl, or substituted aryl; and
 R^{42} is alkyl, substituted alkyl, aryl, or substituted aryl.

2. The compound of claim 1 wherein A^1 is of the formula:

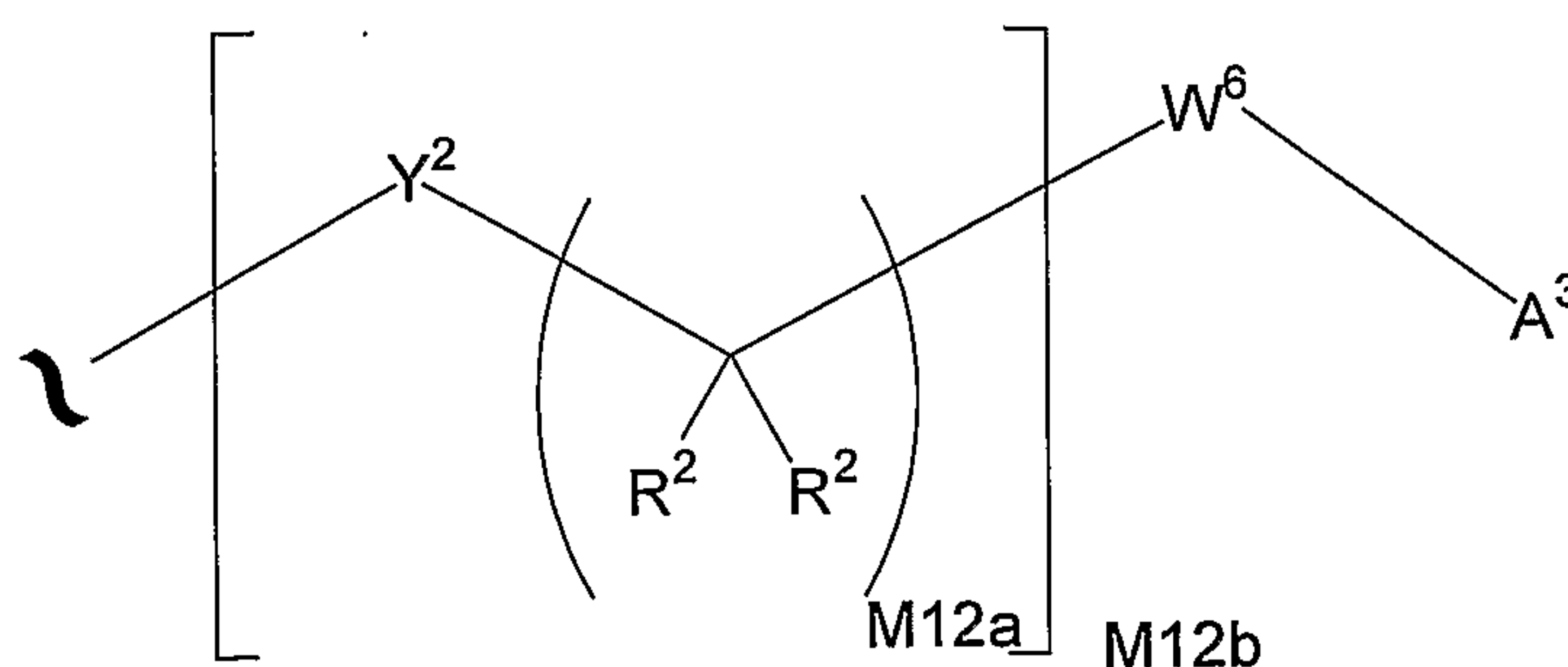


5

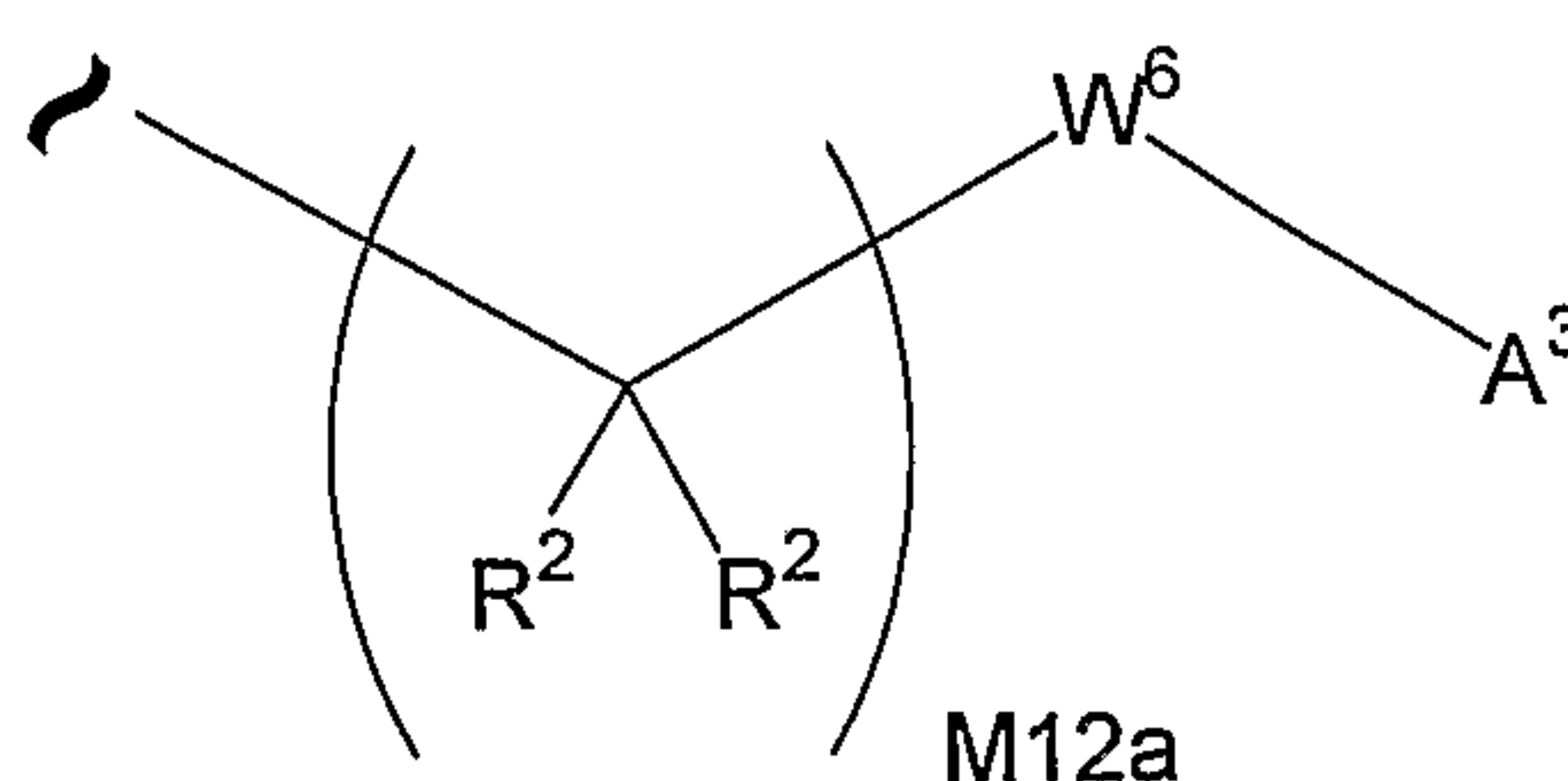
3. The compound of claim 1 wherein A^1 is of the formula:



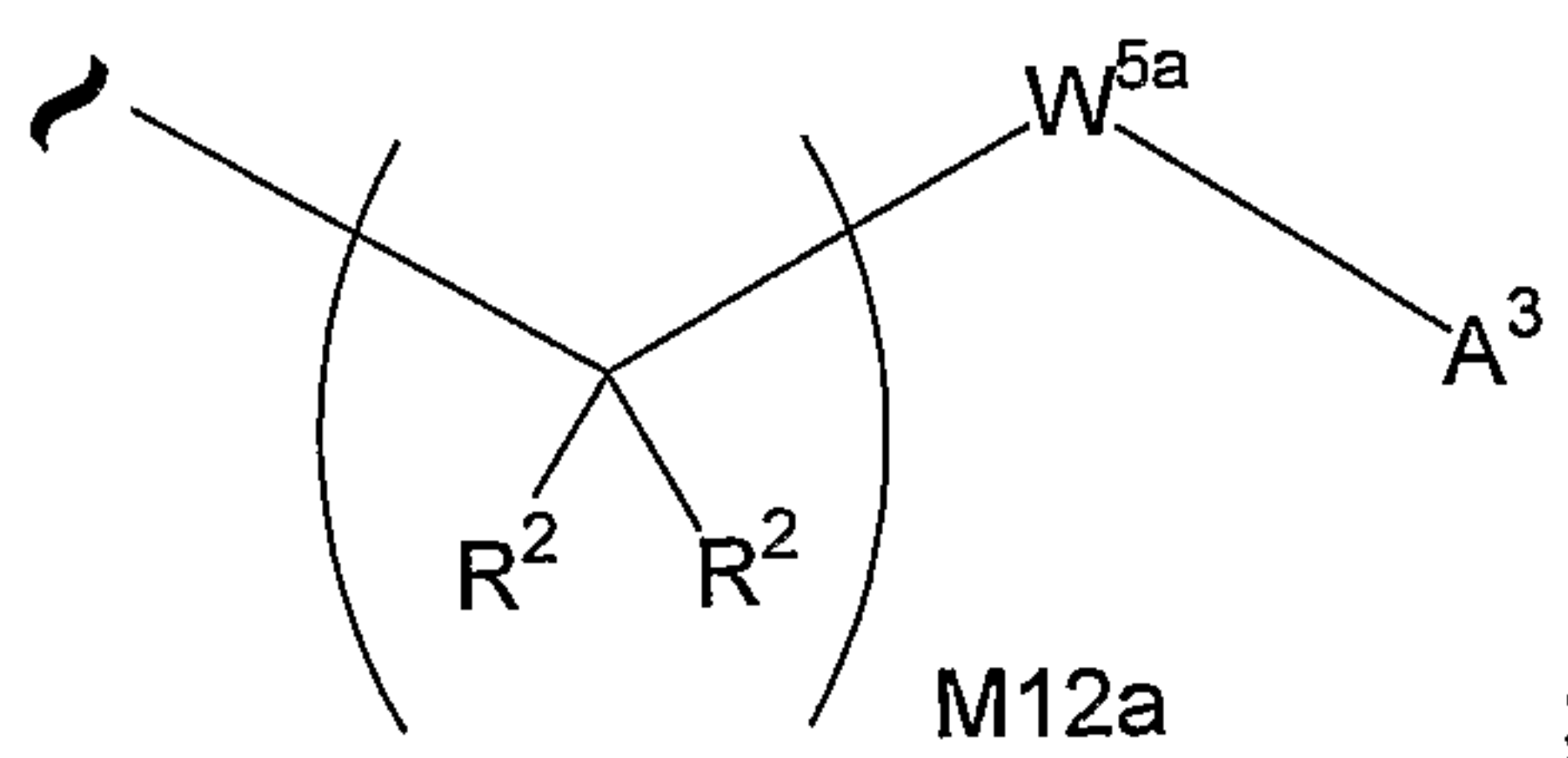
- 10 4. The compound of claim 1 wherein A^1 is of the formula:



5. The compound of claim 1 wherein A^1 is of the formula:



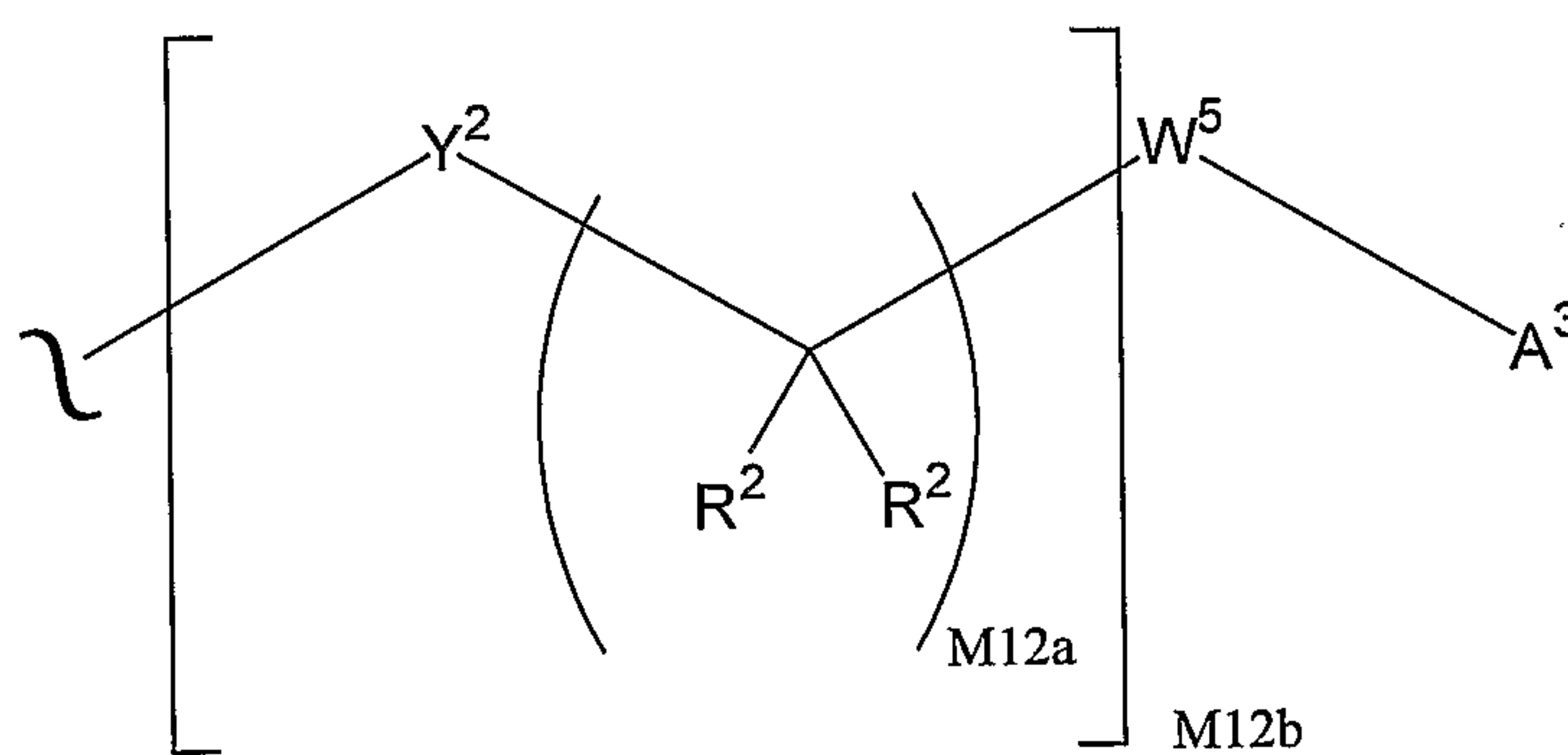
6. The compound of claim 1 wherein A^1 is of the formula:



and W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently
5 substituted with 0 or 1 R^2 groups.

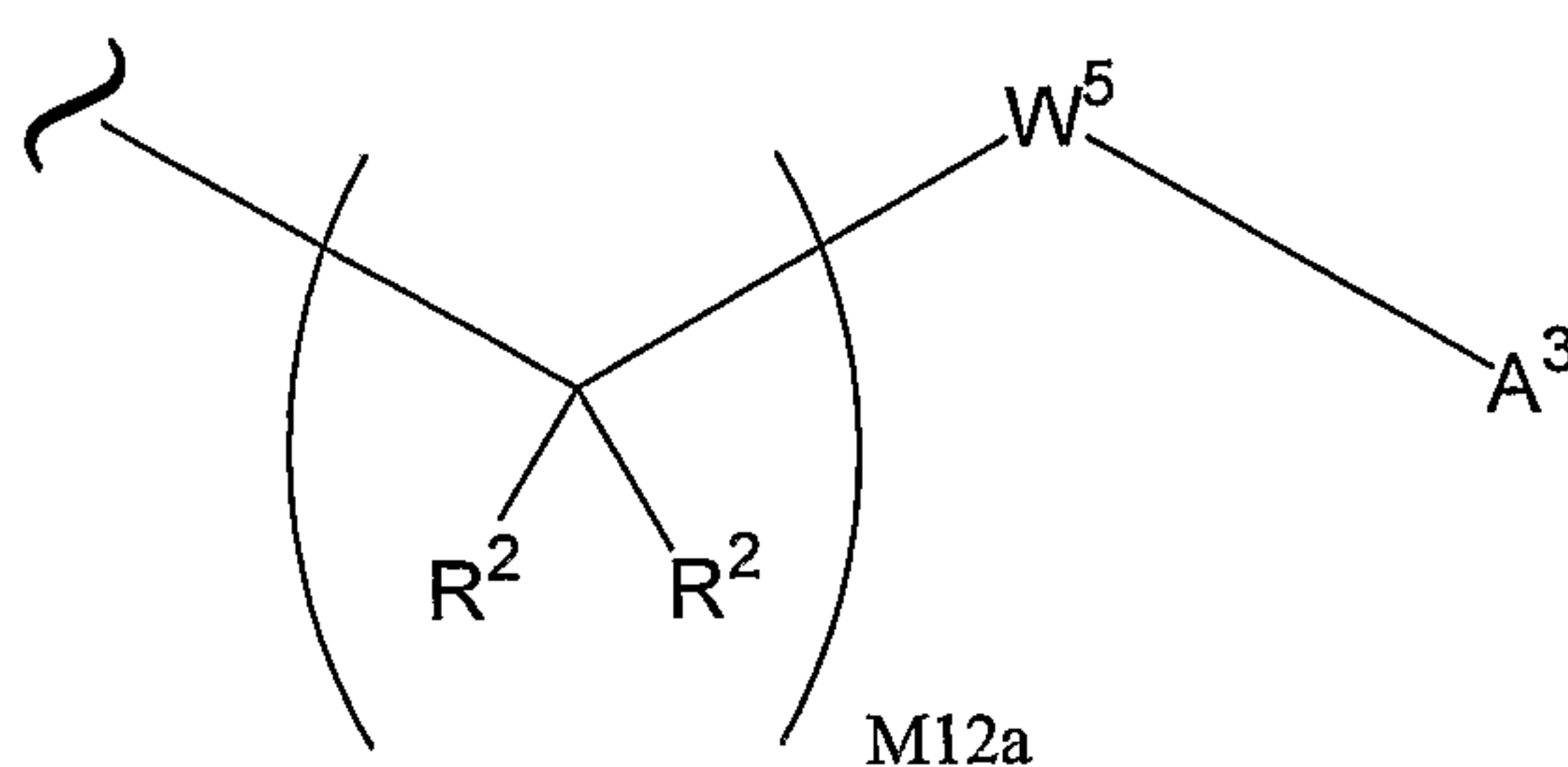
7. The compound of claim 1 wherein M12a is 1.

8. The compound of claim 1 wherein A^1 is of the formula:



10

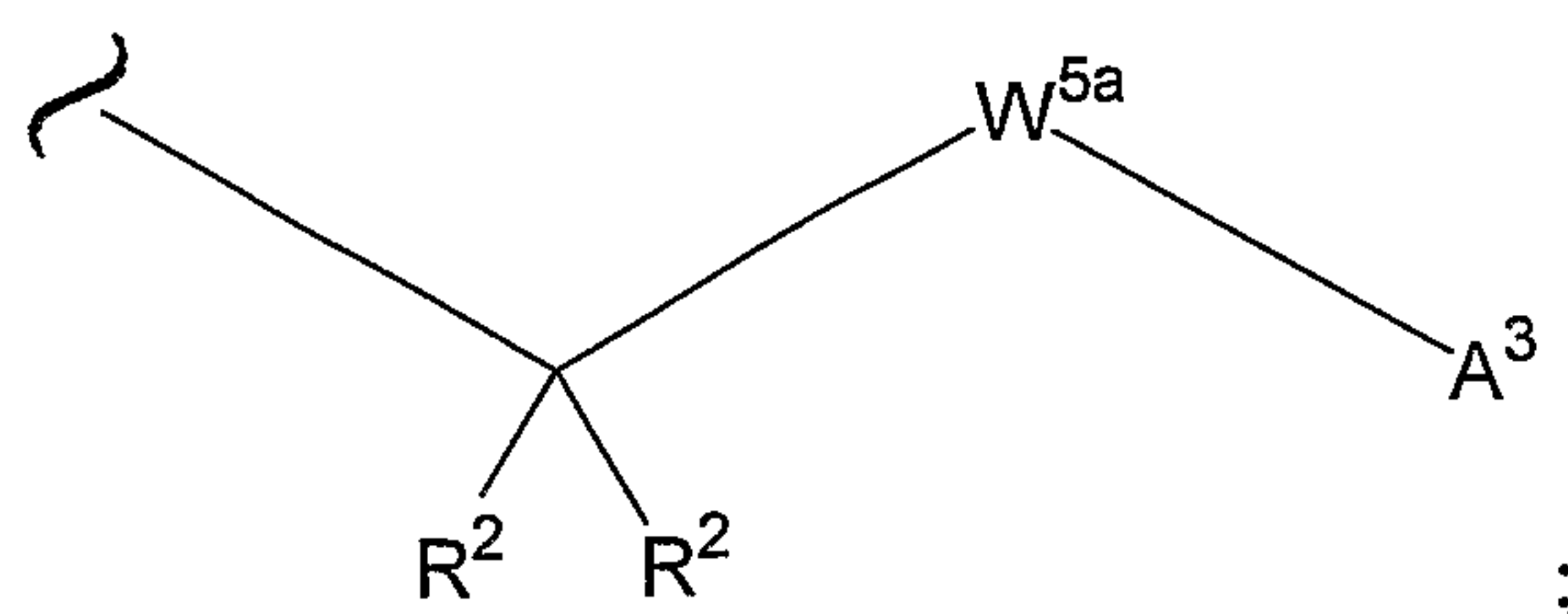
9. The compound of claim 1 wherein A^1 is of the formula:



15

20

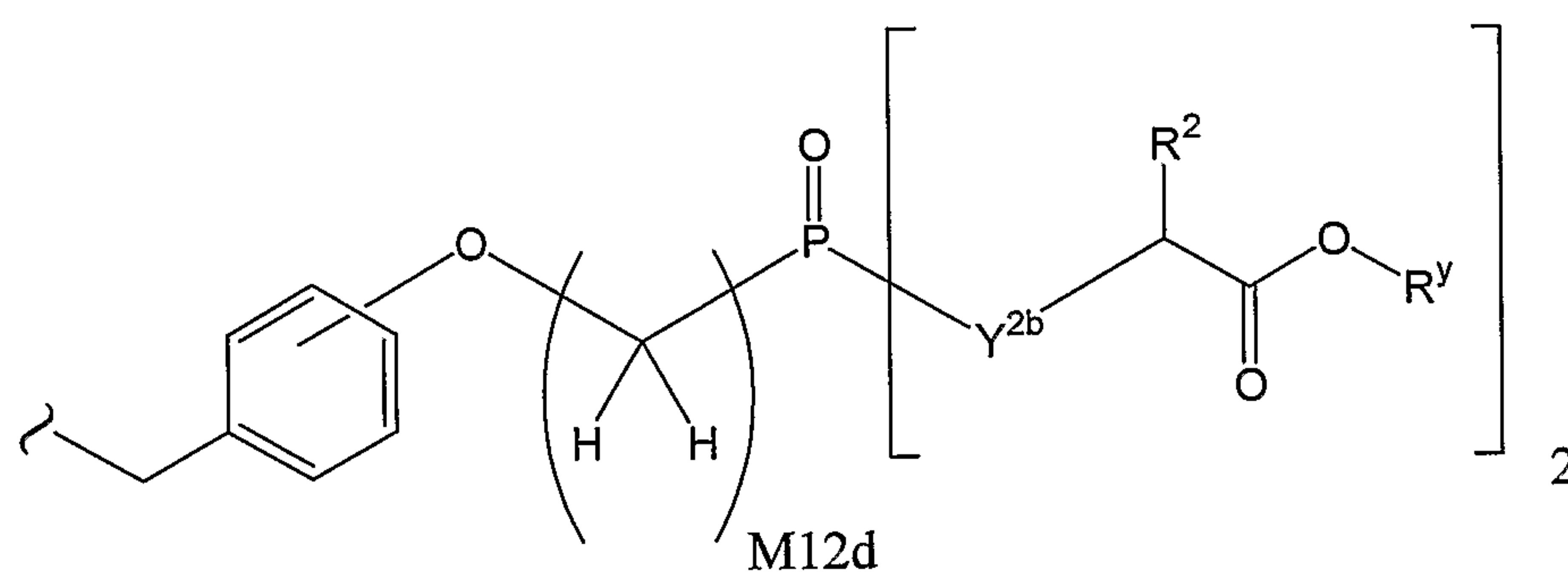
10. The compound of claim 1 wherein A^1 is of the formula:



and W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

5

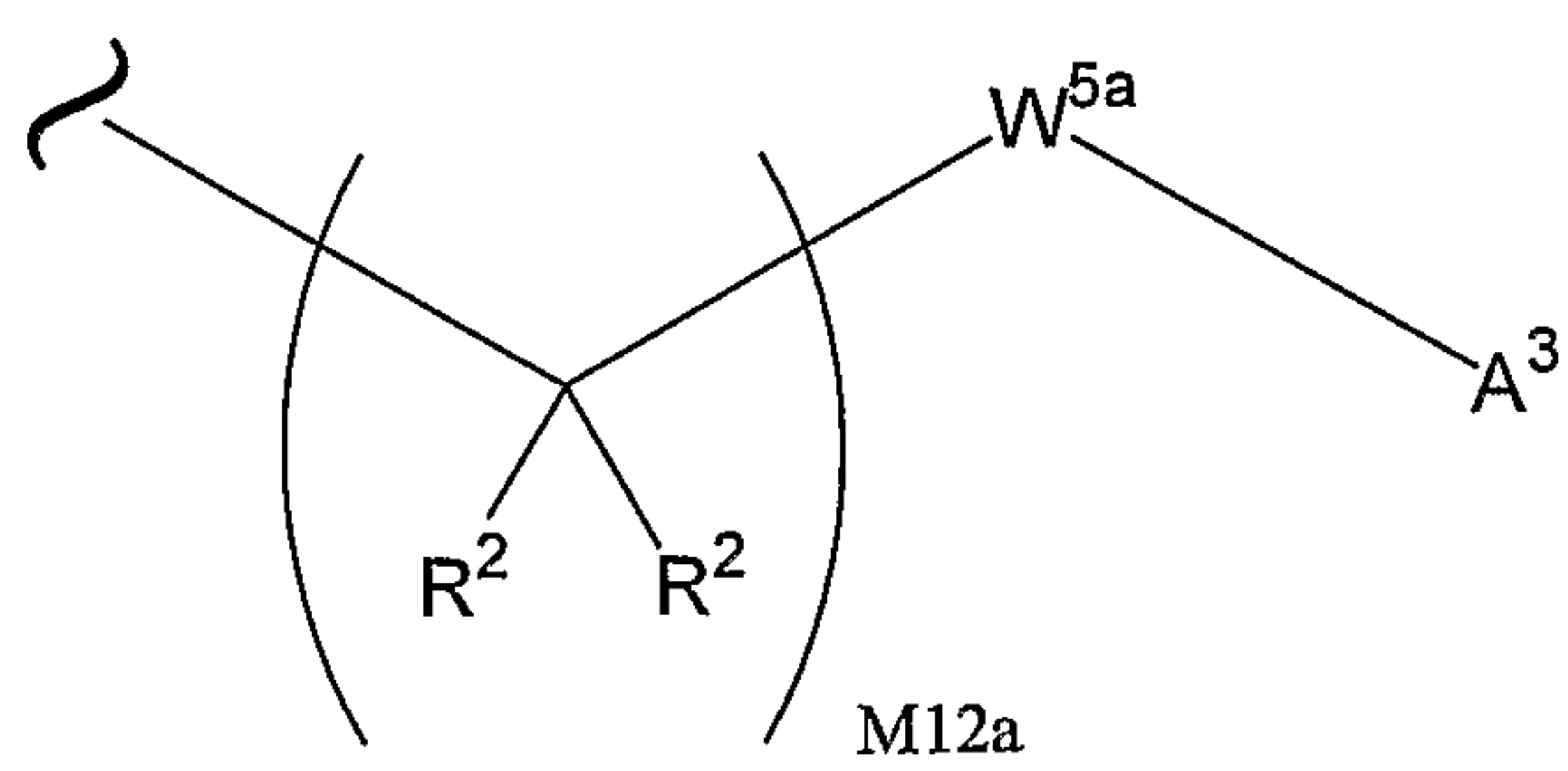
11. The compound of claim 1 wherein A^1 is of the formula:



Y^{2b} is O or $N(R^2)$; and

10 $M12d$ is 1, 2, 3, 4, 5, 6, 7 or 8.

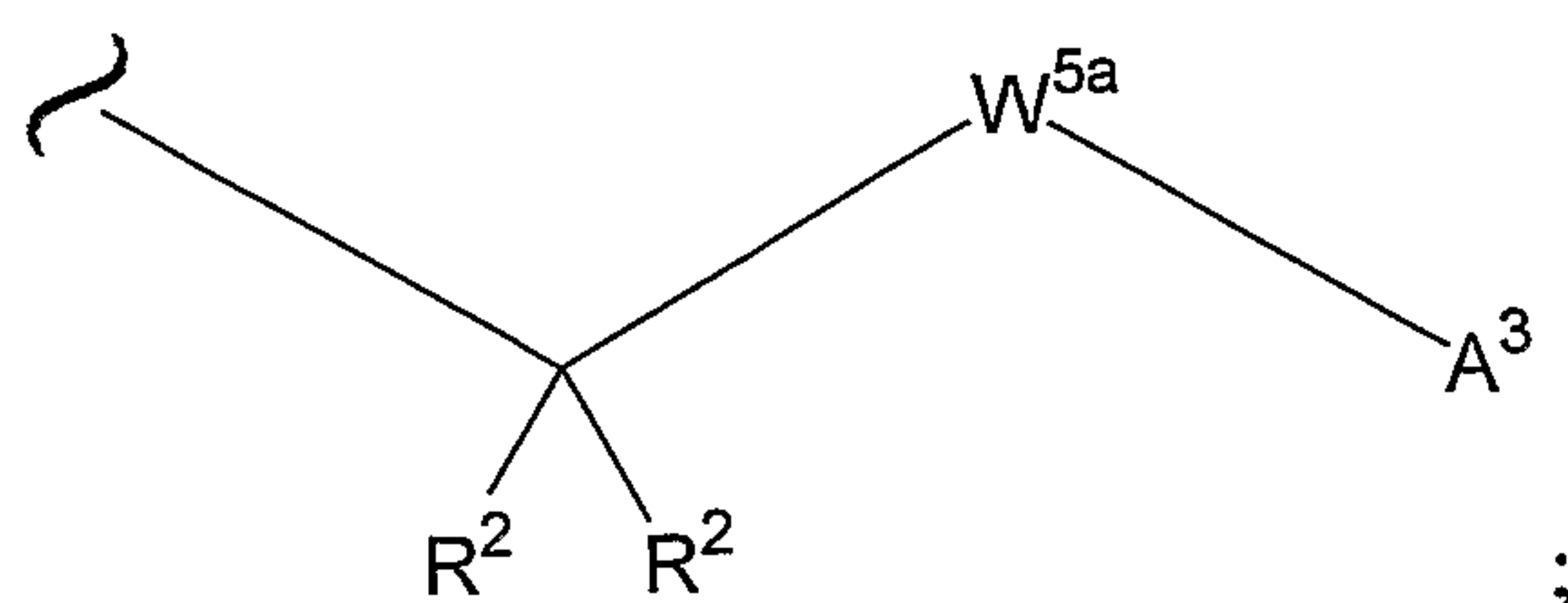
12. The compound of claim 1 wherein A^1 is of the formula:



15 and W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

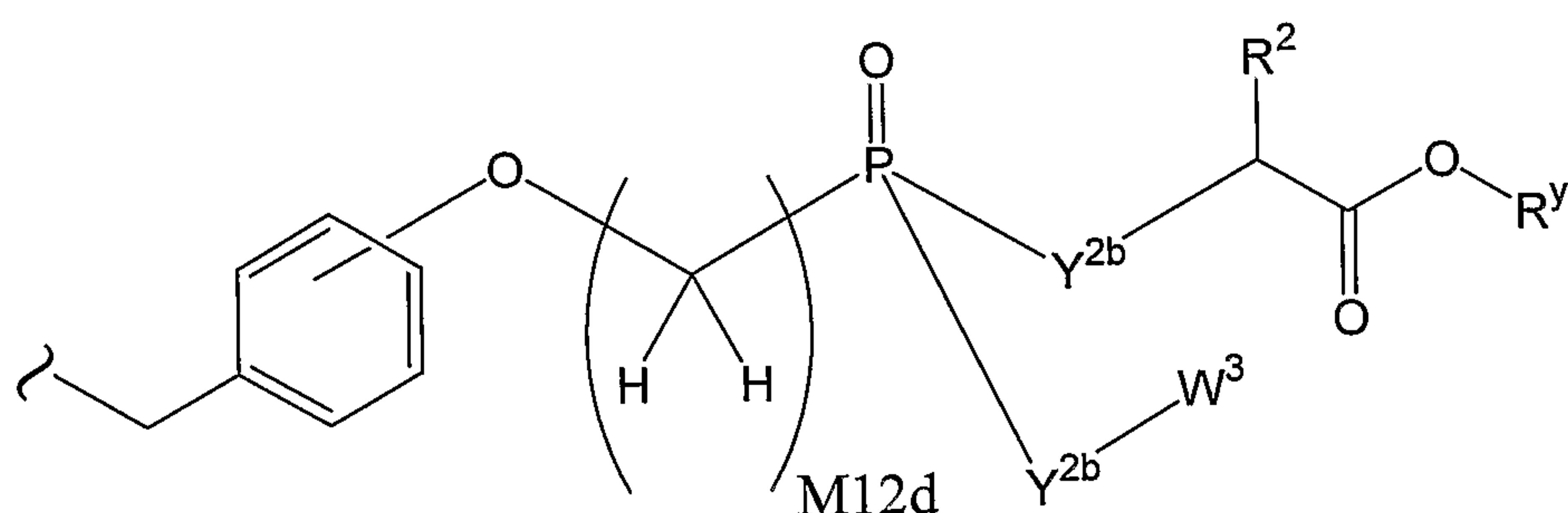
20

13. The compound of claim 1 wherein A^1 is of the formula:



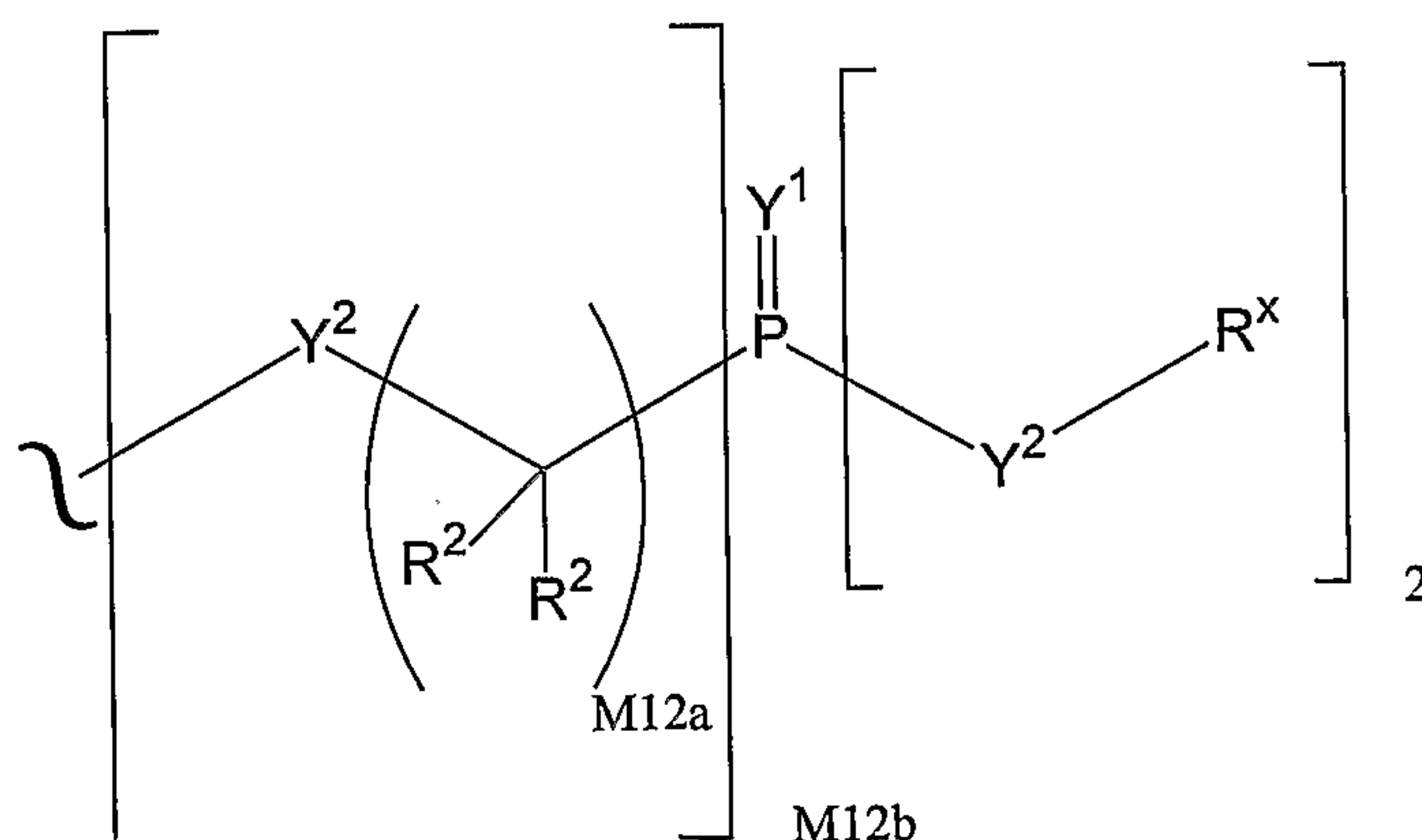
- and W^{5a} is a carbocycle or heterocycle where W^{5a} is independently
5 substituted with 0 or 1 R^2 groups.

14. The compound of claim 1 wherein A^1 is of the formula:



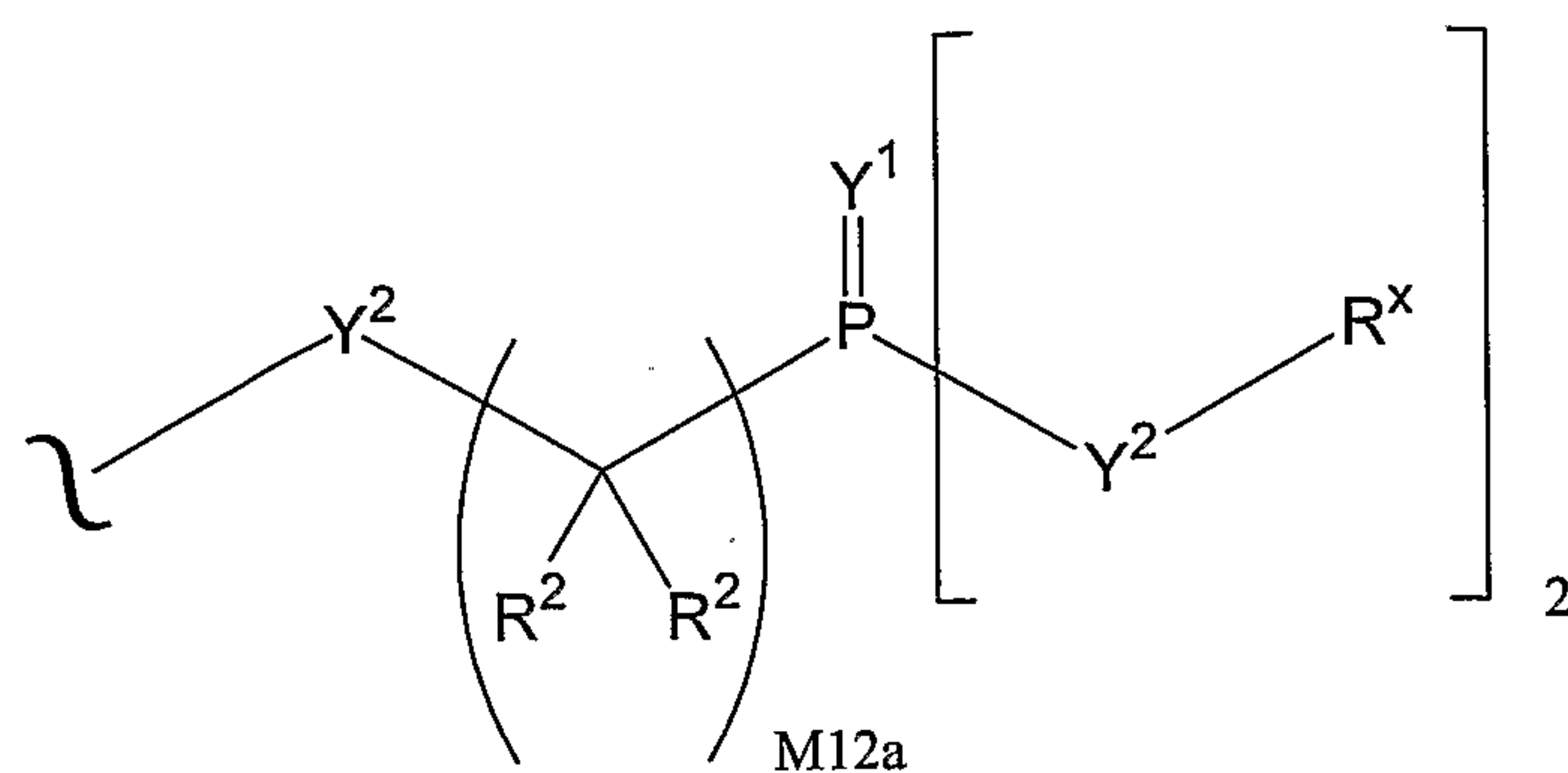
- 10 Y^{2b} is independently O or $N(R^2)$; and
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

15. The compound of any one of claims 1-14 wherein A^3 is of the formula:

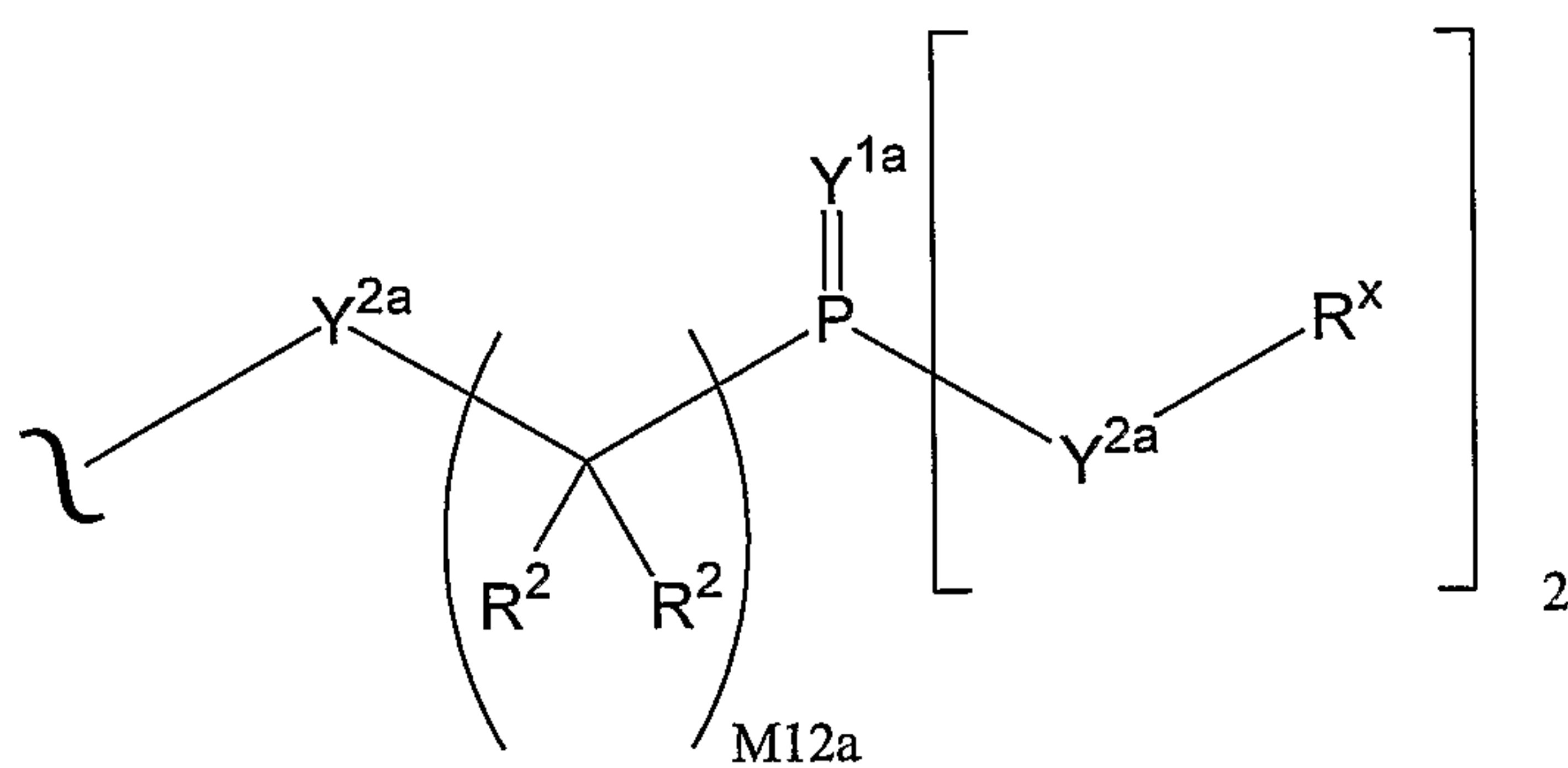


15

16. The compound of any one of claims 1-14 wherein A³ is of the formula:



17. The compound of any one of claims 1-14 wherein A³ is of the formula:

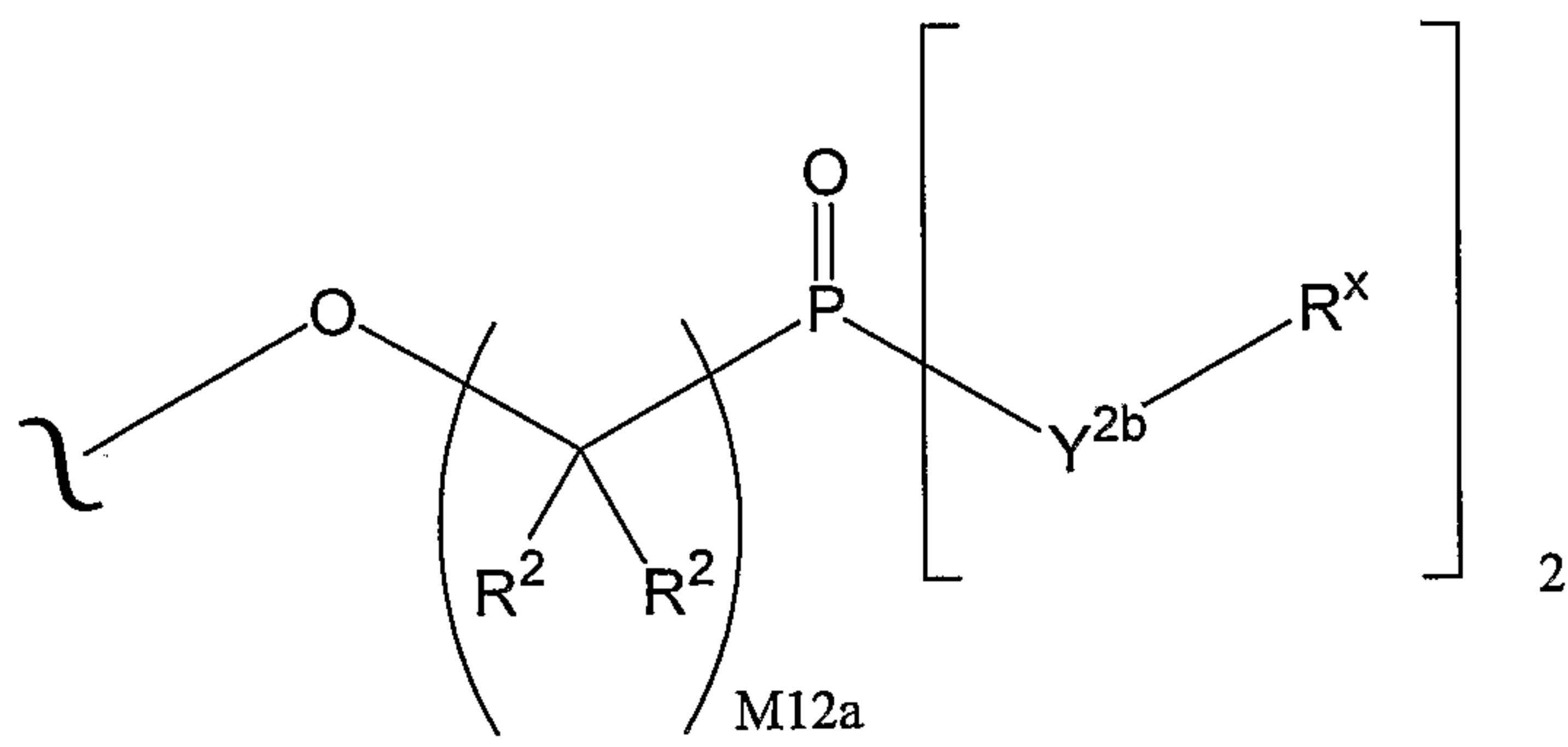


5

Y^{1a} is O or S; and

Y^{2a} is independently O, N(R^x) or S.

18. The compound of any one of claims 1-14 wherein A³ is of the formula:

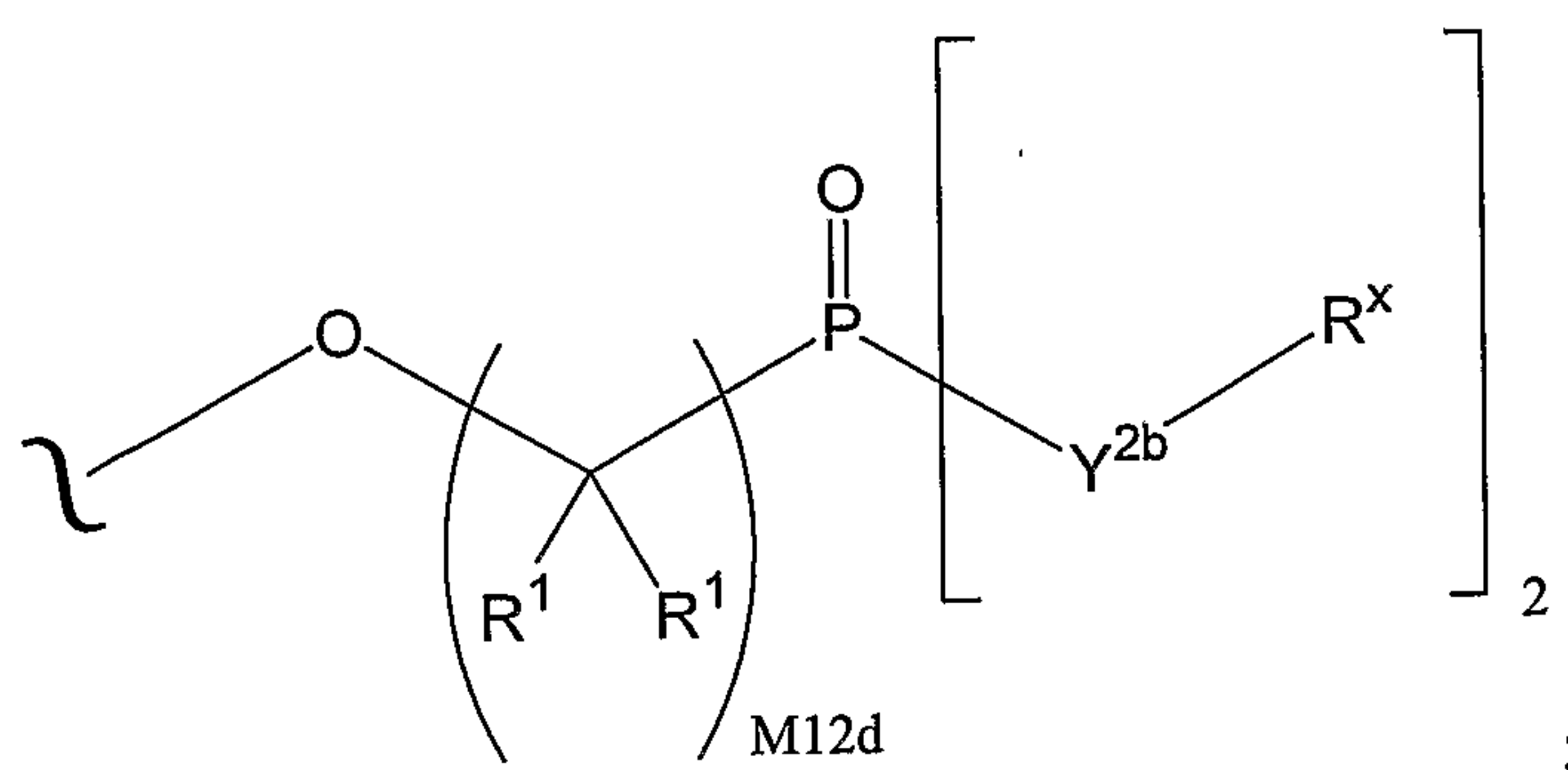


10

and Y^{2b} is O or N(R^x).

15

19. The compound of any one of claims 1-14 wherein A³ is of the formula:

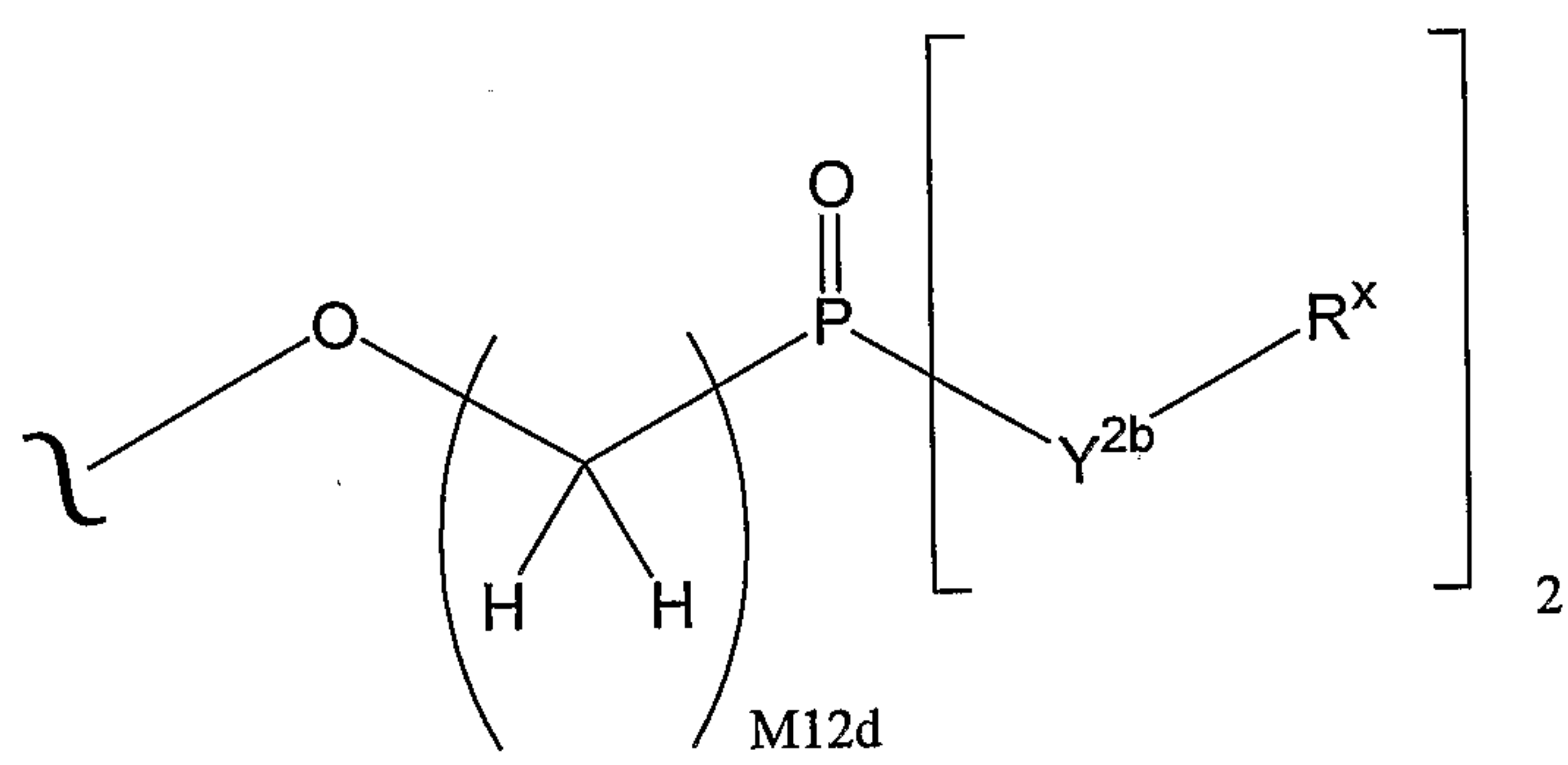


R¹ is independently H or alkyl of 1 to 18 carbon atoms;

5 Y^{2b} is O or N(R^x); and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

20. The compound of any one of claims 1-14 wherein A³ is of the formula:



10 Y^{2b} is O or N(R^x); and

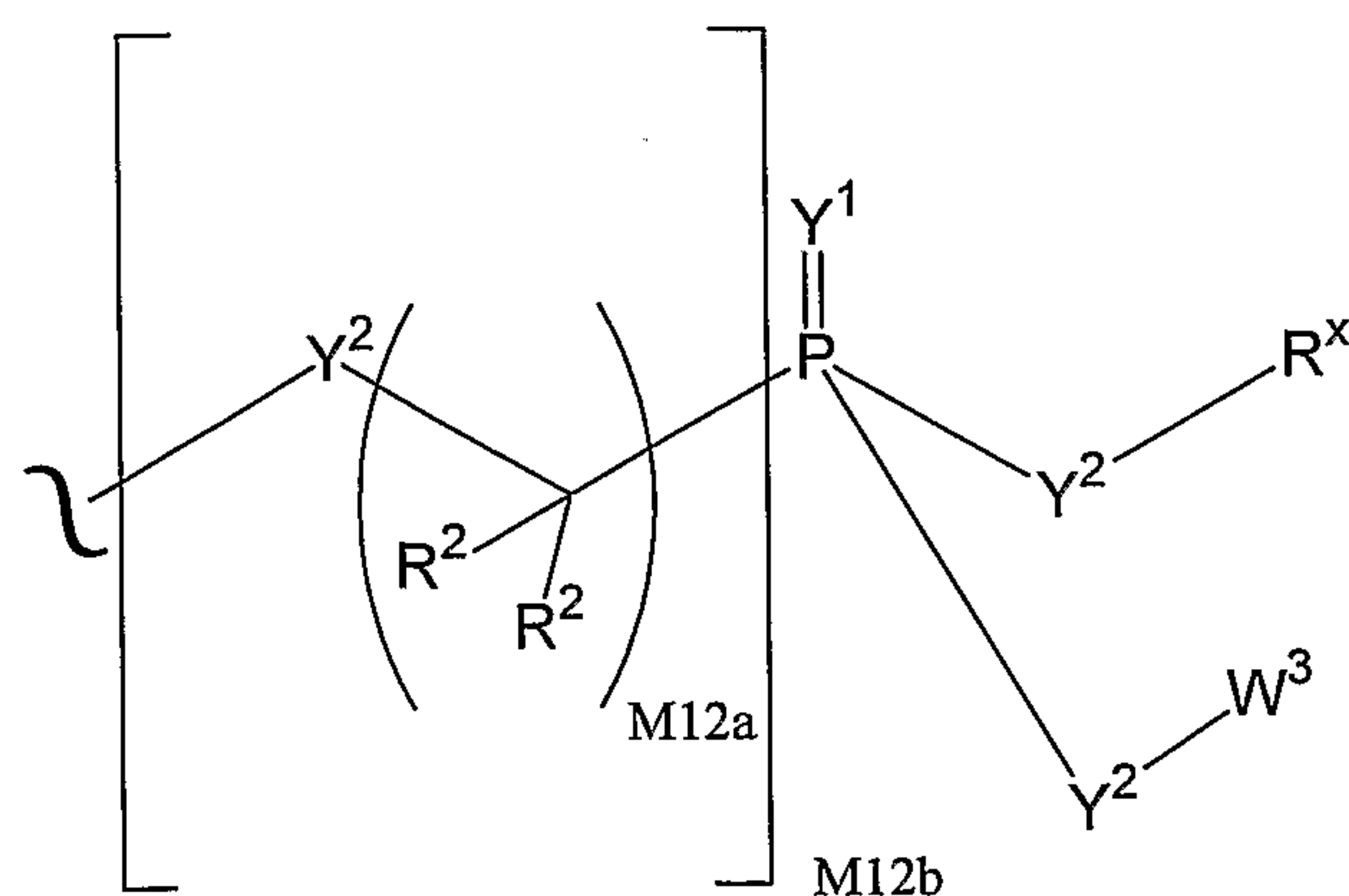
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

21. The compound of claim 20 wherein M12d is 1.

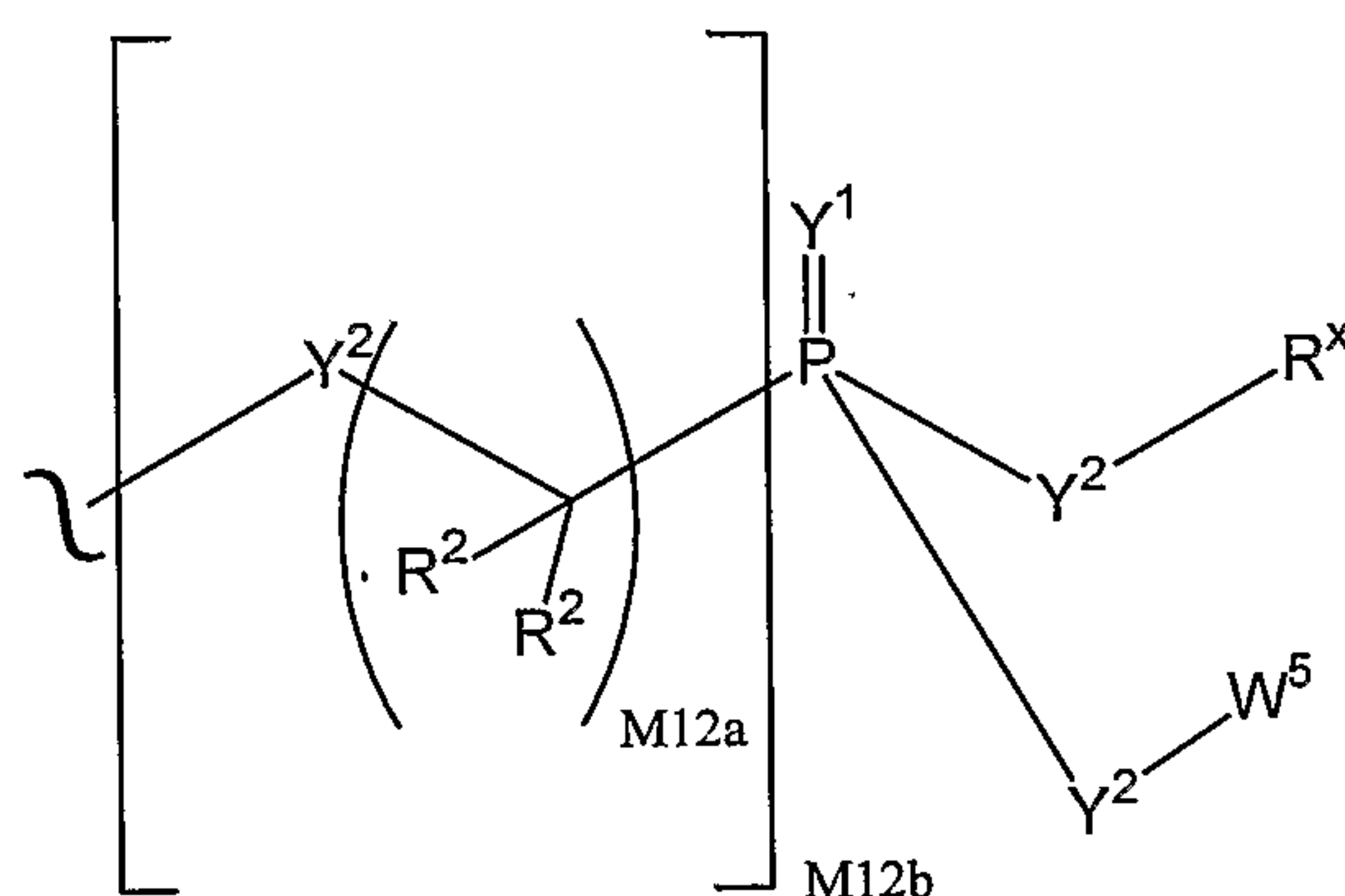
15

20

22. The compound of any one of claims 1-14 wherein A^3 is of the formula:



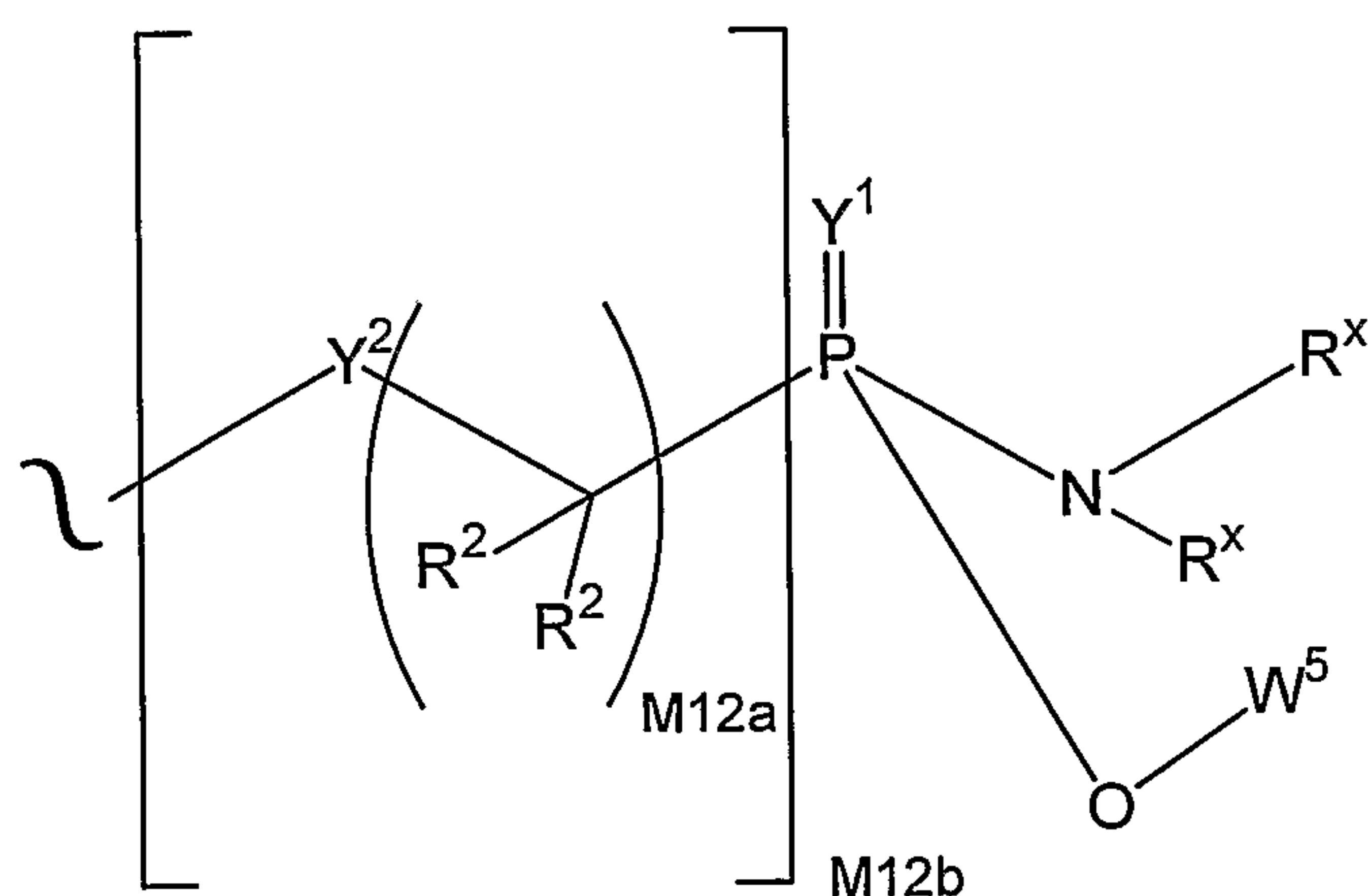
23. The compound of any one of claims 1-14 wherein A^3 is of the formula:



5

24. The compound of claim 23 wherein W^5 is a carbocycle.

25. The compound of any one of claims 1-14 wherein A^3 is of the formula:

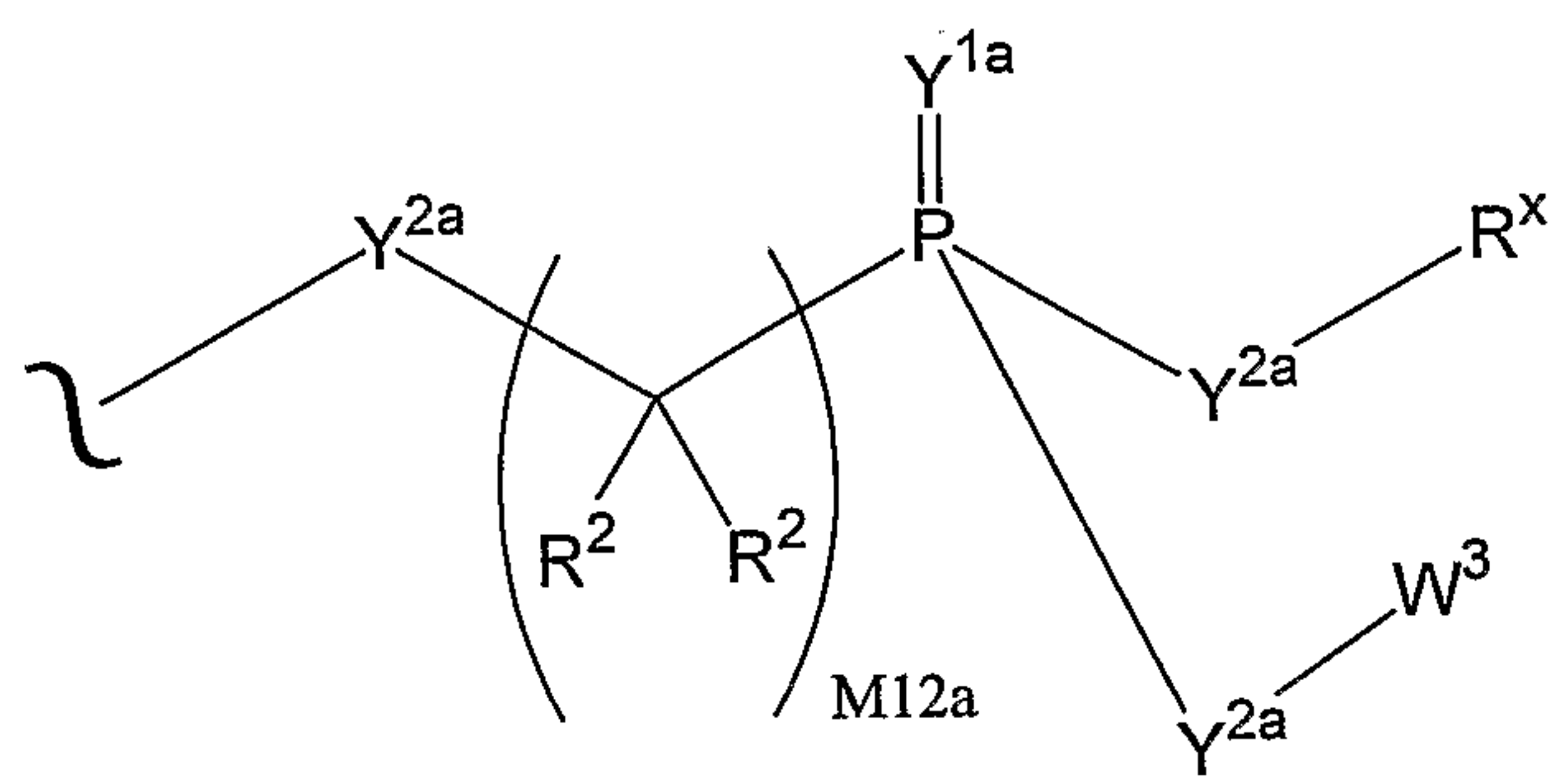


10

26. The compound of claim 25 wherein W^5 is phenyl.

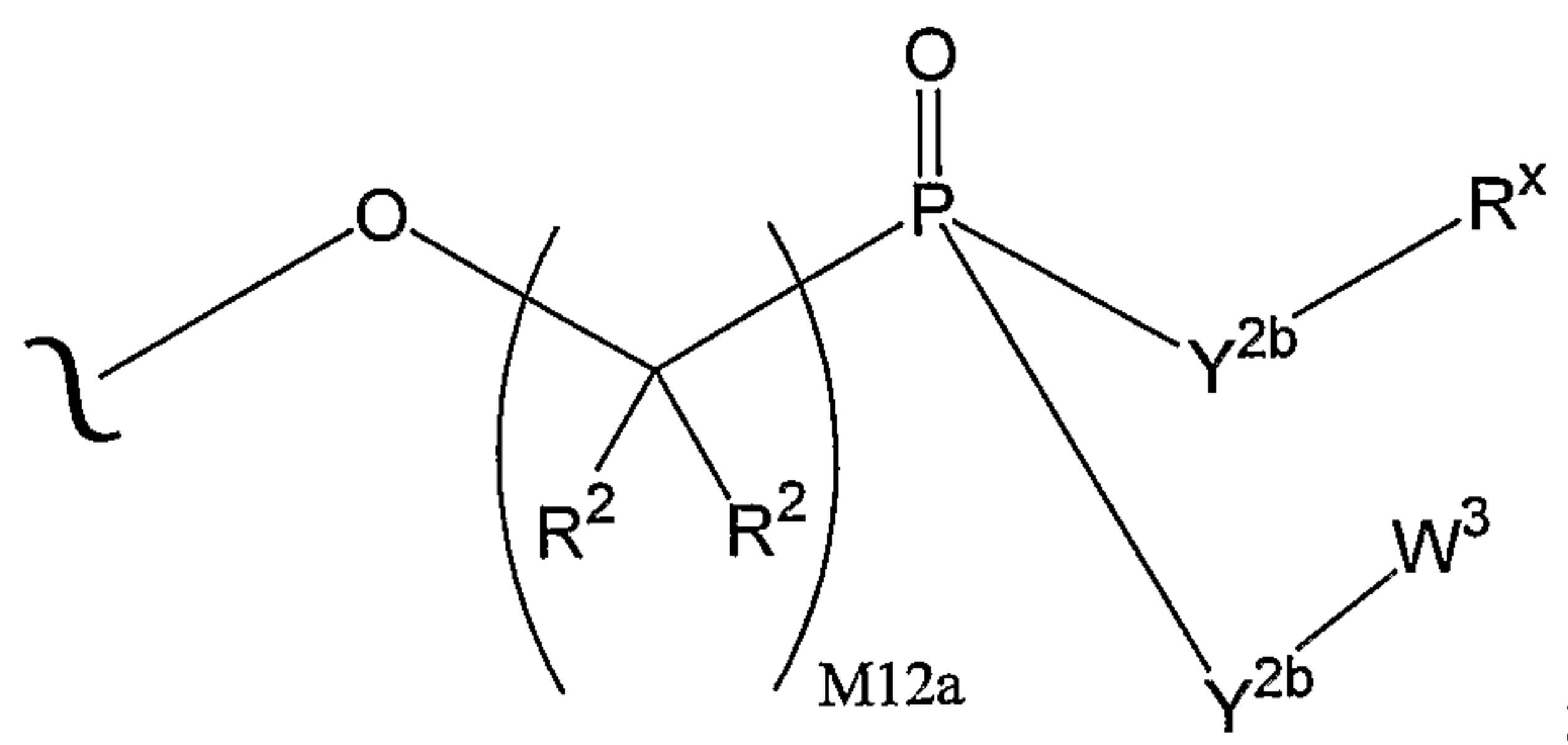
27. The compound of claim 26 wherein M12b is 1.

28. The compound of any one of claims 1-14 wherein A³ is of the formula:



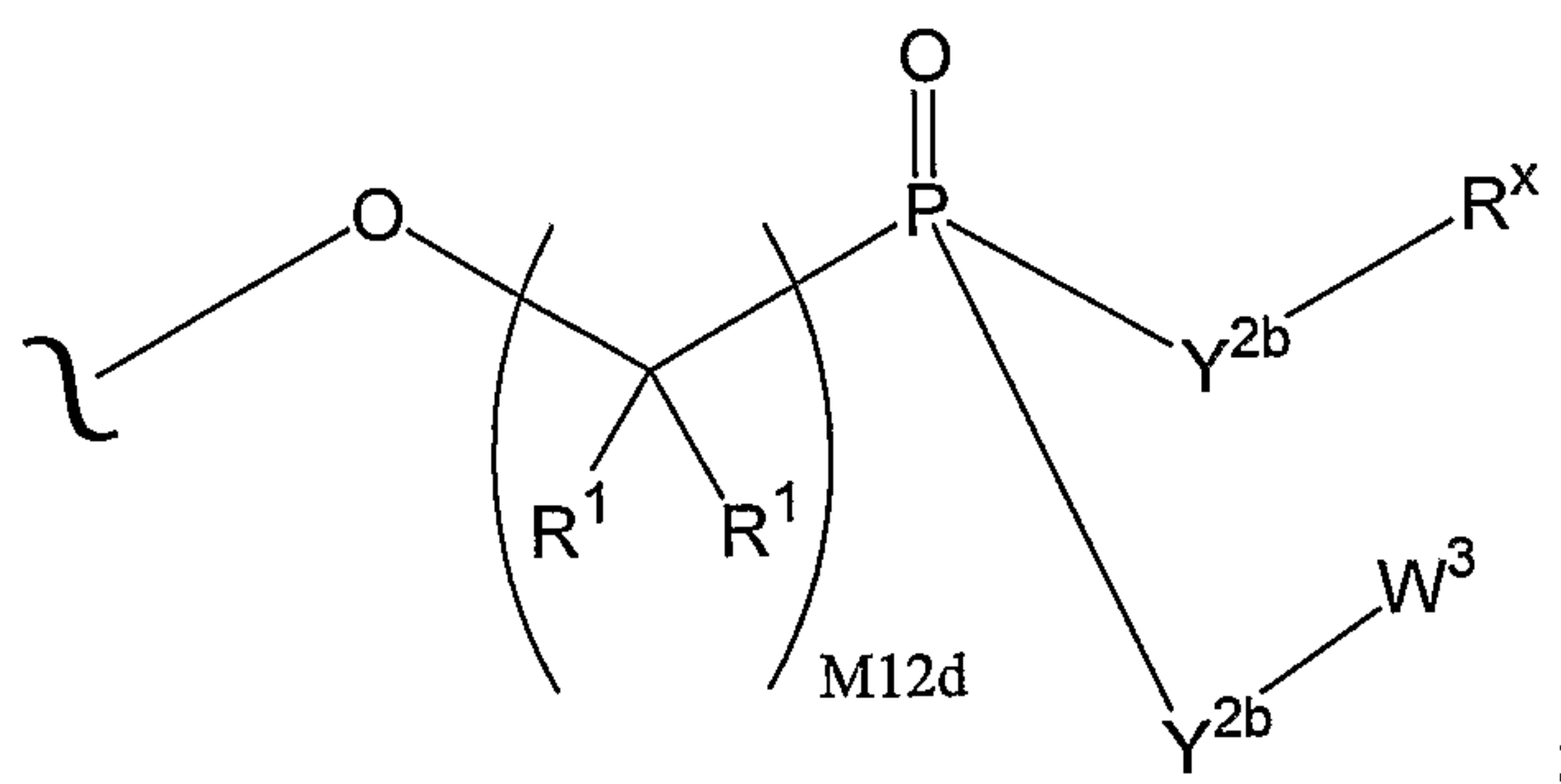
5 Y^{1a} is O or S; and
Y^{2a} is independently O, N(R^x) or S.

29. The compound of any one of claims 1-14 wherein A³ is of the formula:



10 and Y^{2b} is independently O or N(R^x).

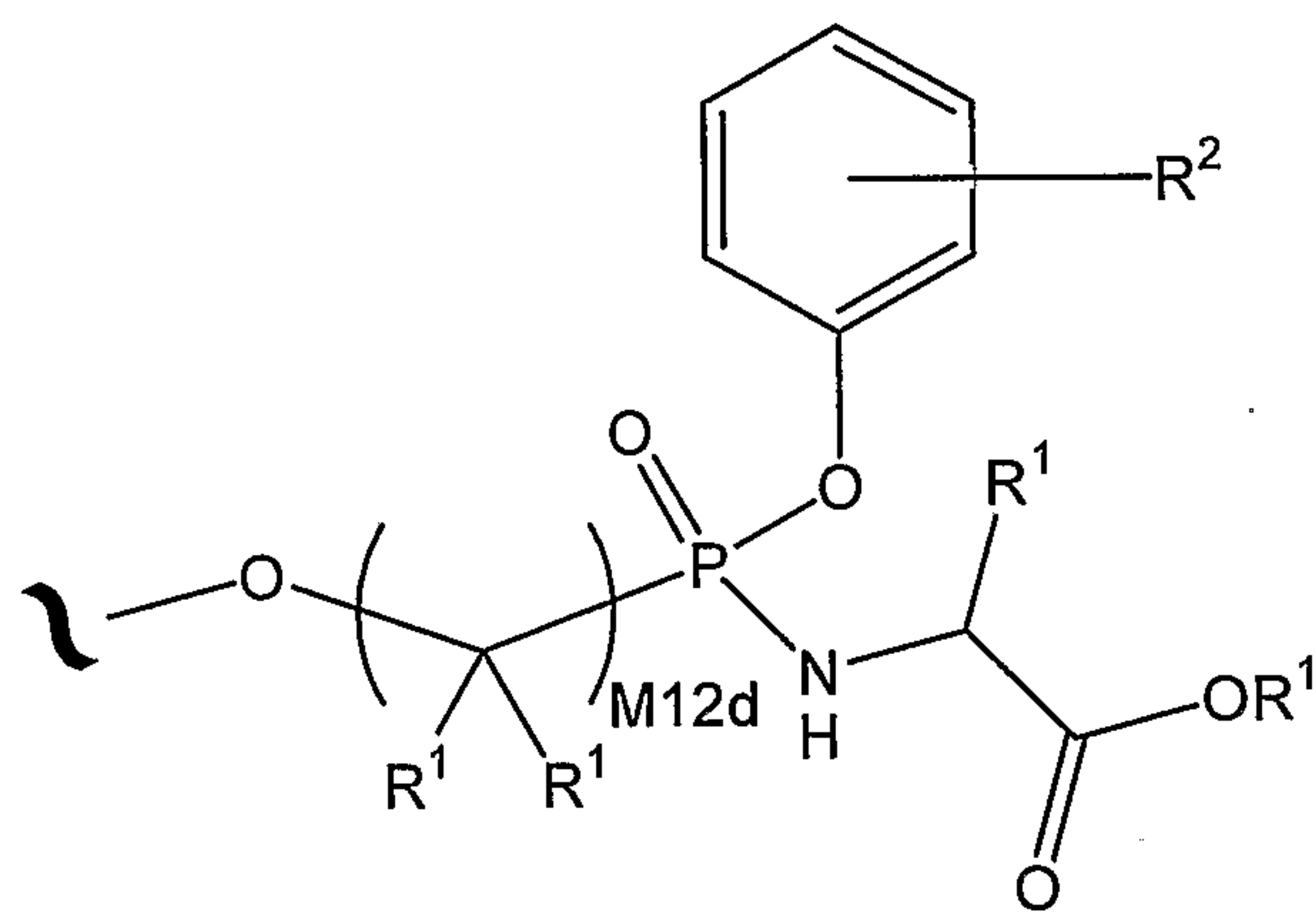
30. The compound of any one of claims 1-14 wherein A³ is of the formula:



15 R¹ is independently H or alkyl of 1 to 18 carbon atoms;
Y^{2b} is independently O or N(R^x); and
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

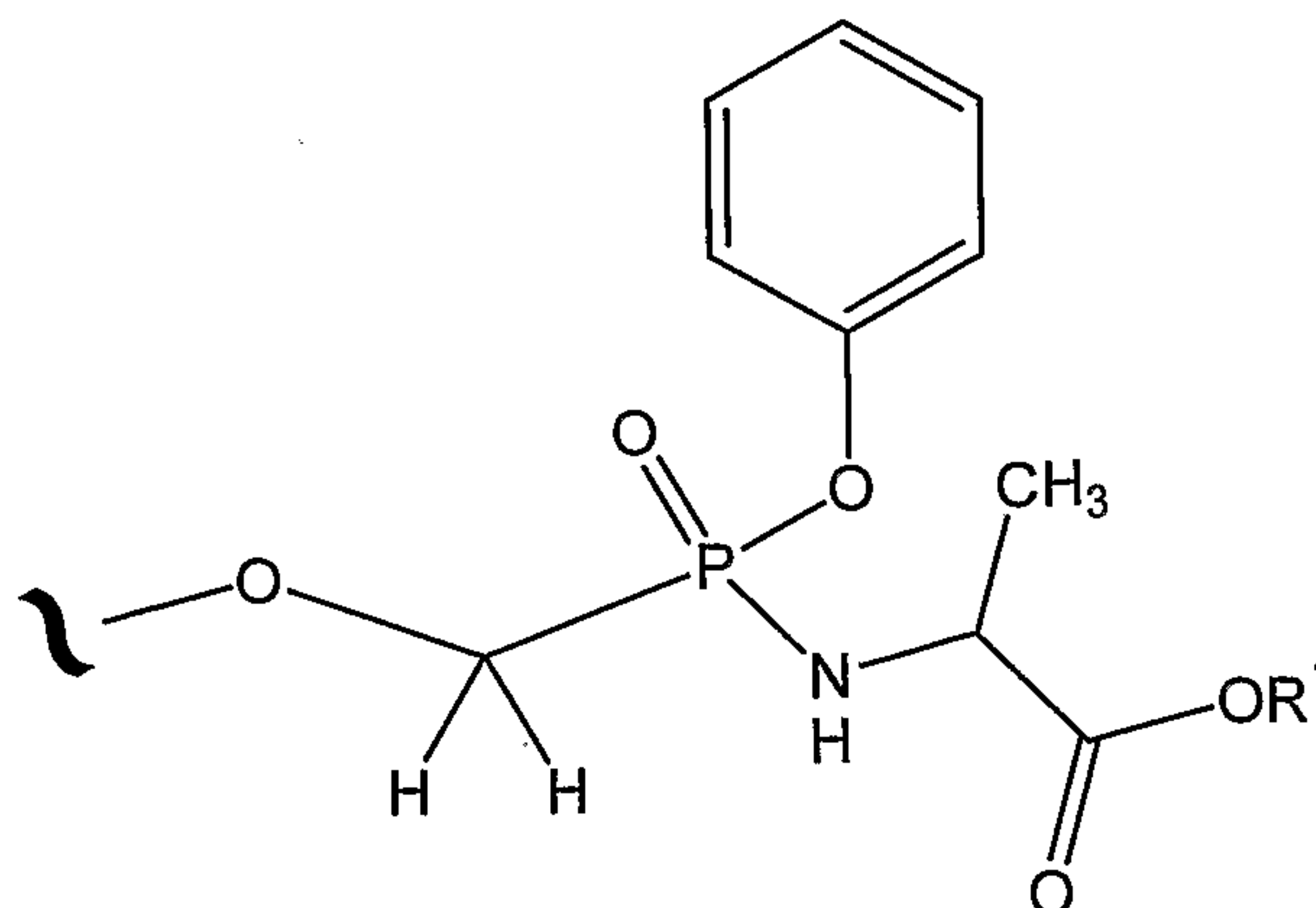
-

10 34. The compound of any one of claims 1-14 wherein A³ is of the formula:



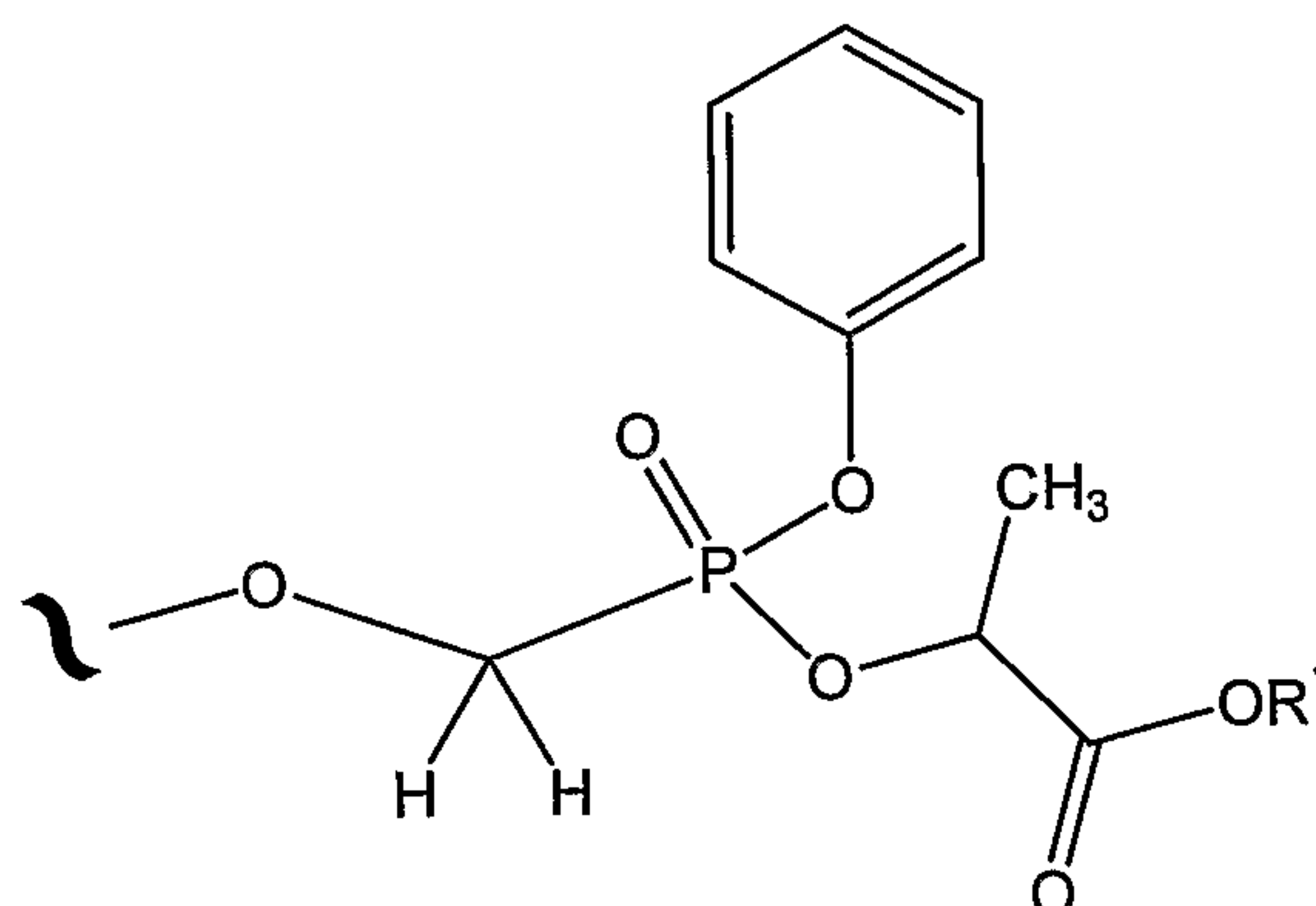
15

35. The compound of any one of claims 1-14 wherein A³ is of the formula:



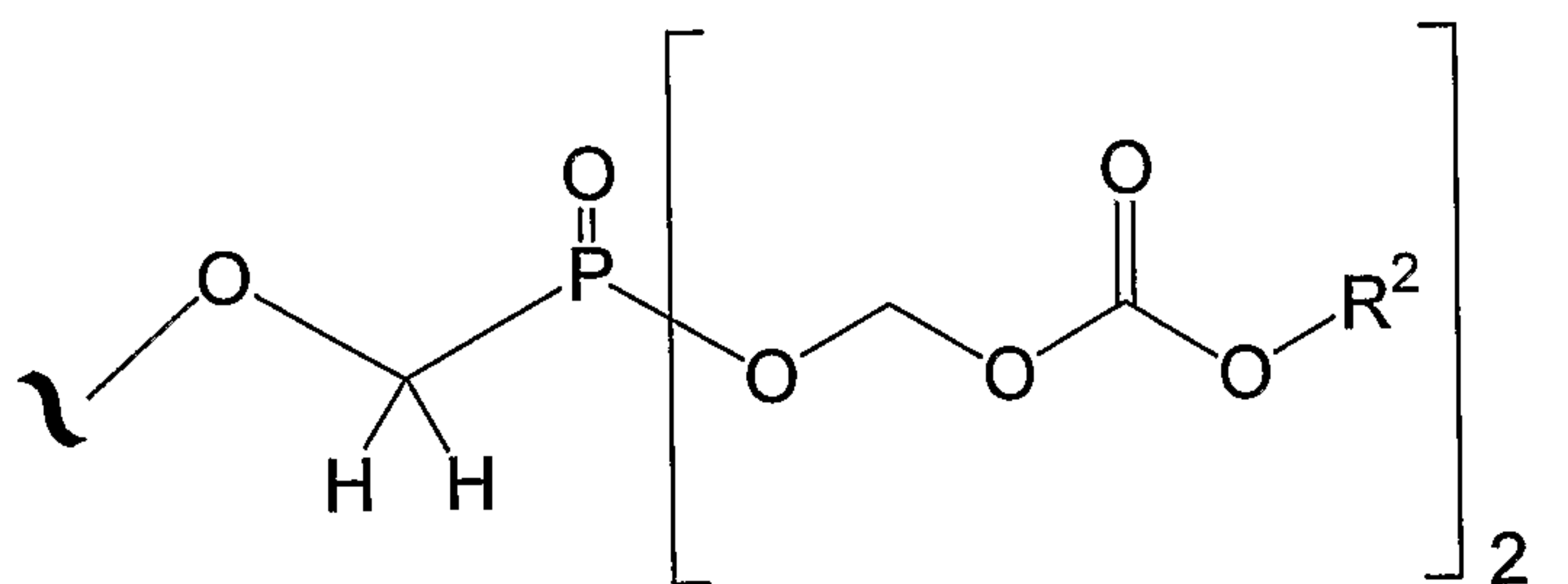
and R¹ is H or alkyl of 1 to 18 carbon atoms.

- 5 36. The compound of any one of claims 1-14 wherein A³ is of the formula:



and R¹ is H or alkyl of 1 to 18 carbon atoms.

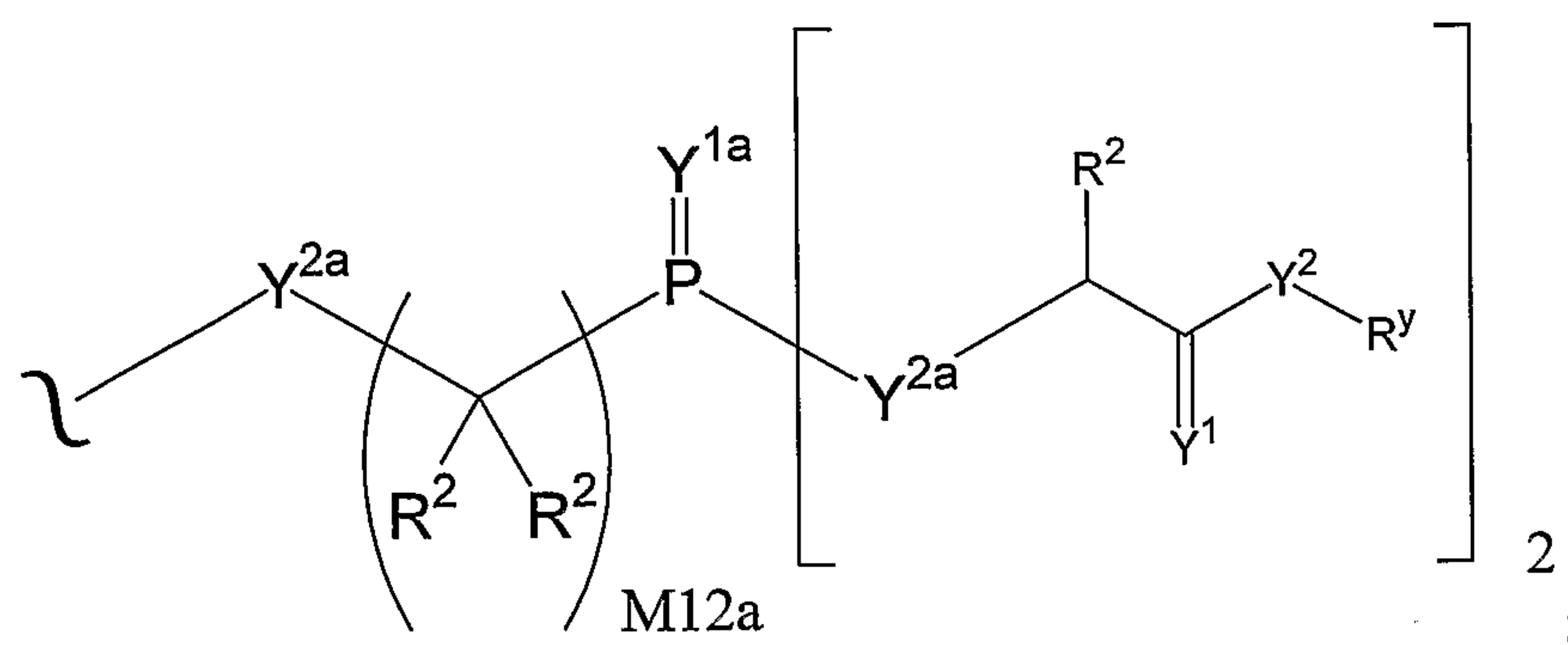
37. The compound of any one of claims 1-14 wherein A³ is of the formula:



10

15

38. The compound of any one of claims 1-14 wherein A^3 is of the formula:

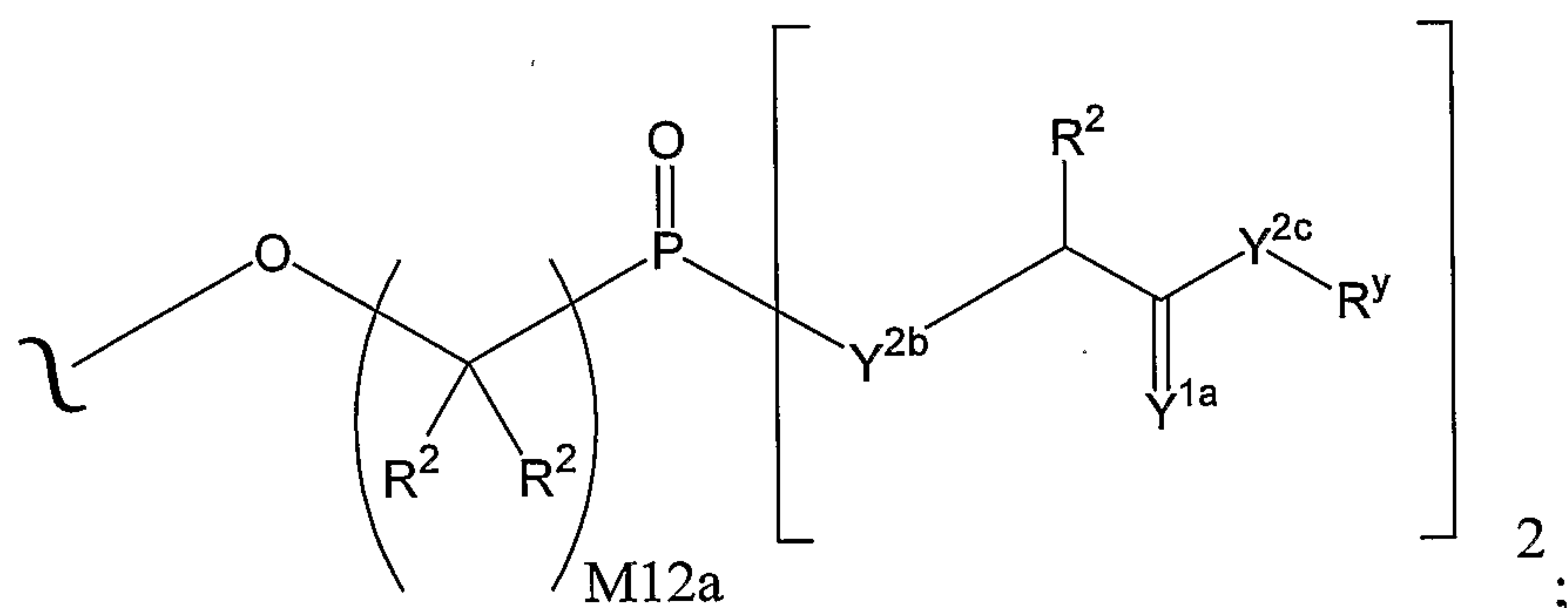


Y^{1a} is O or S; and

Y^{2a} is independently O, $N(R^2)$ or S.

5

39. The compound of any one of claims 1-14 wherein A^3 is of the formula:



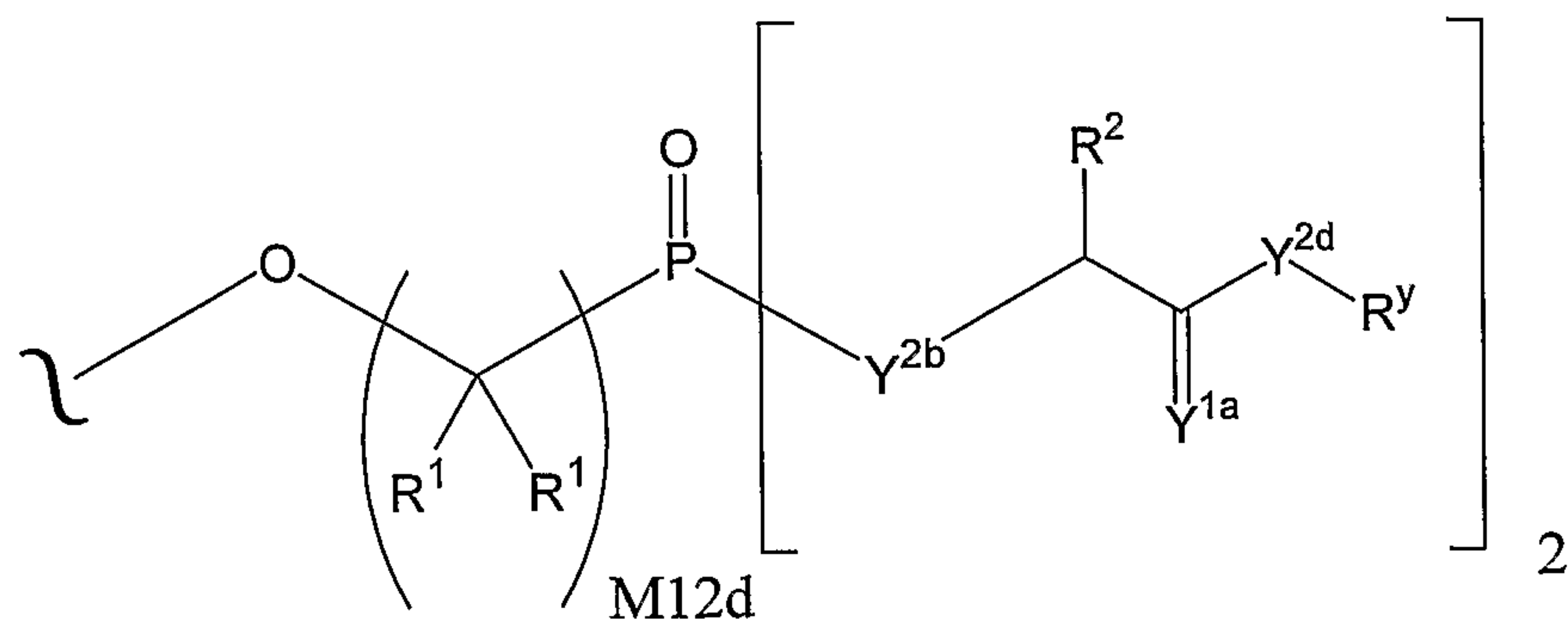
Y^{1a} is O or S;

Y^{2b} is O or $N(R^2)$; and

Y^{2c} is O, $N(R^y)$ or S.

10

40. The compound of any one of claims 1-14 wherein A^3 is of the formula:



R^1 is independently H or alkyl of 1 to 18 carbon atoms;

Y^{1a} is O or S;

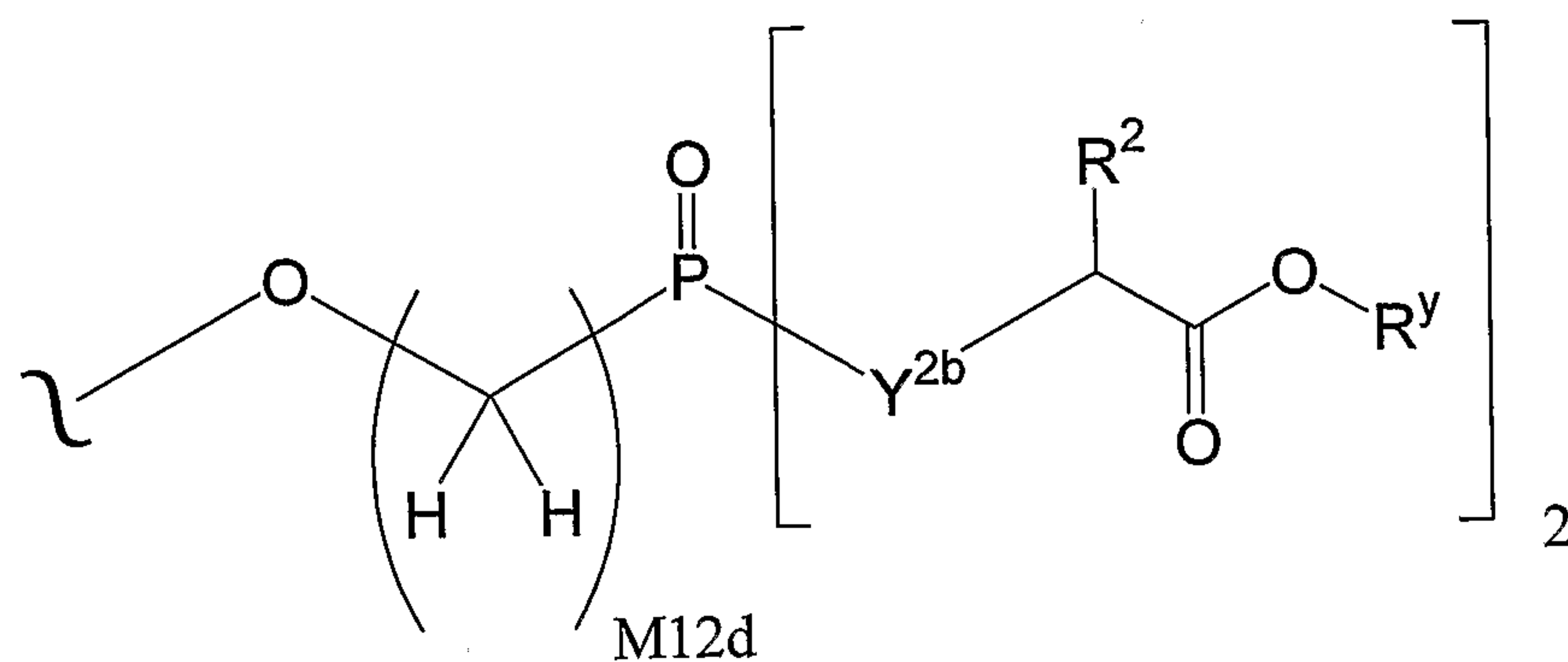
Y^{2b} is O or $N(R^2)$;

15

Y^{2d} is O or $N(R^y)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

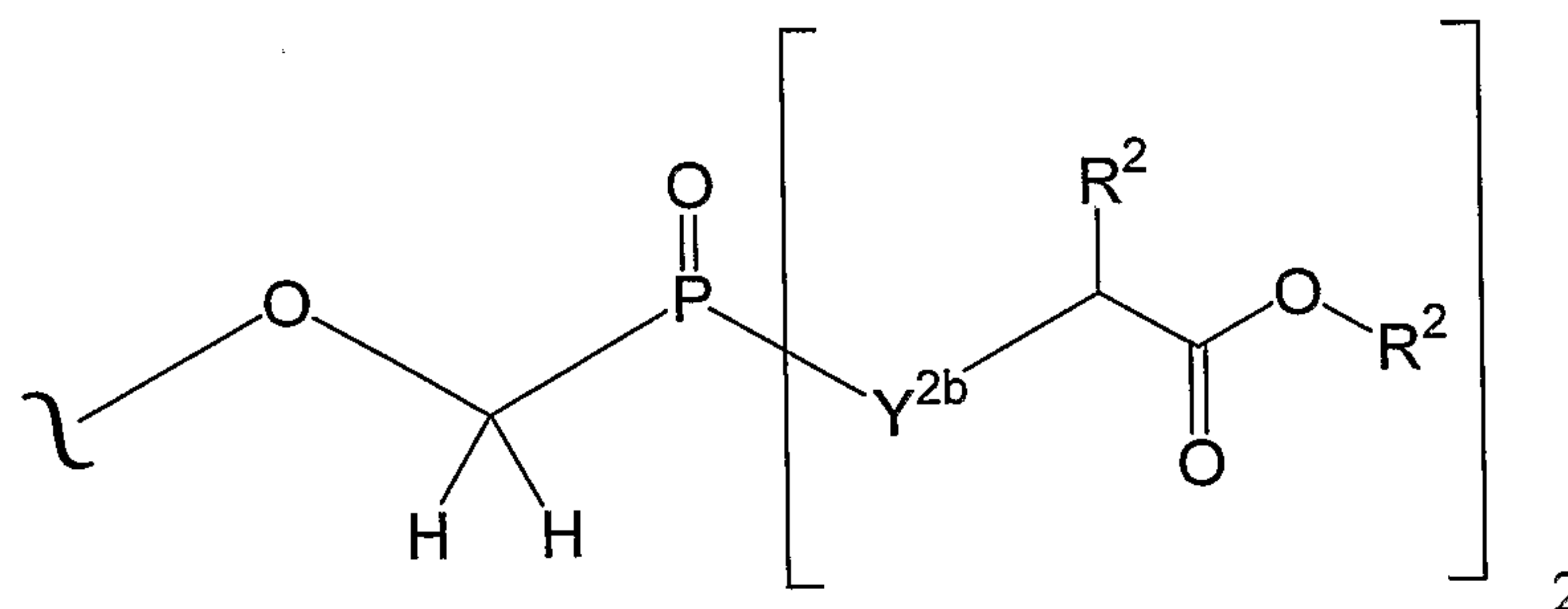
41. The compound of any one of claims 1-14 wherein A^3 is of the formula:



Y^{2b} is O or $N(R^2)$; and

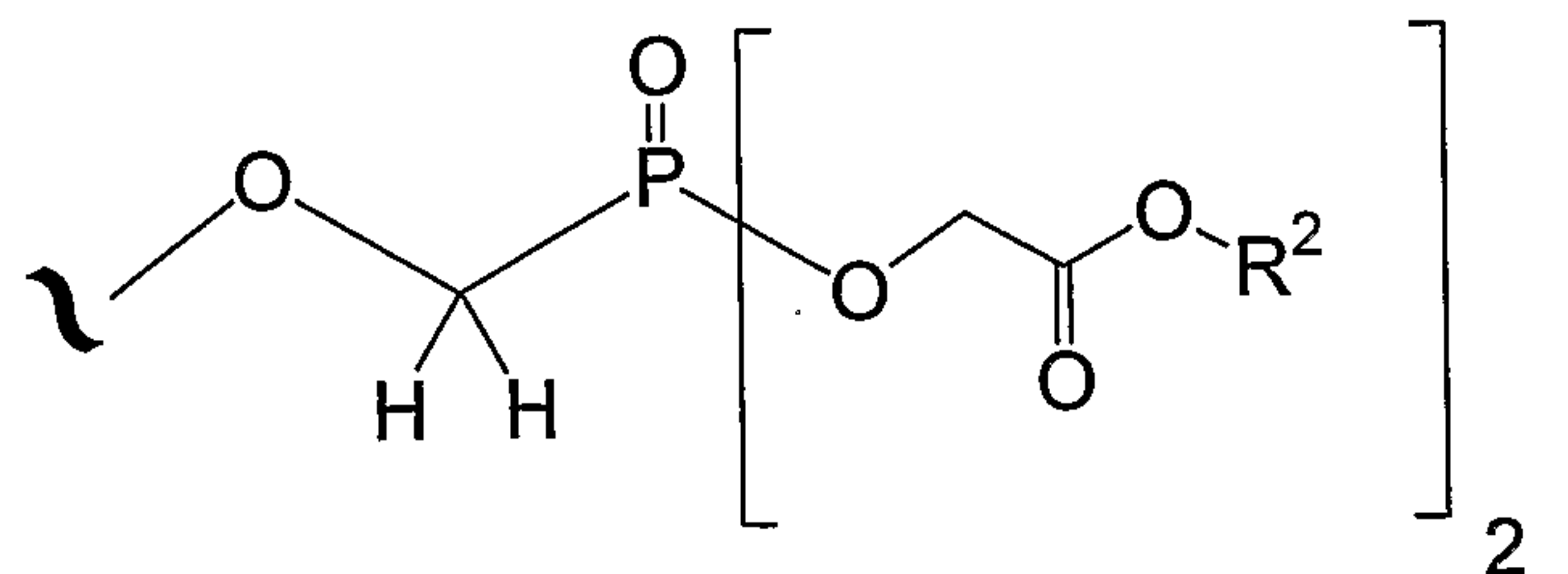
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

42. The compound of any one of claims 1-14 wherein A^3 is of the formula:

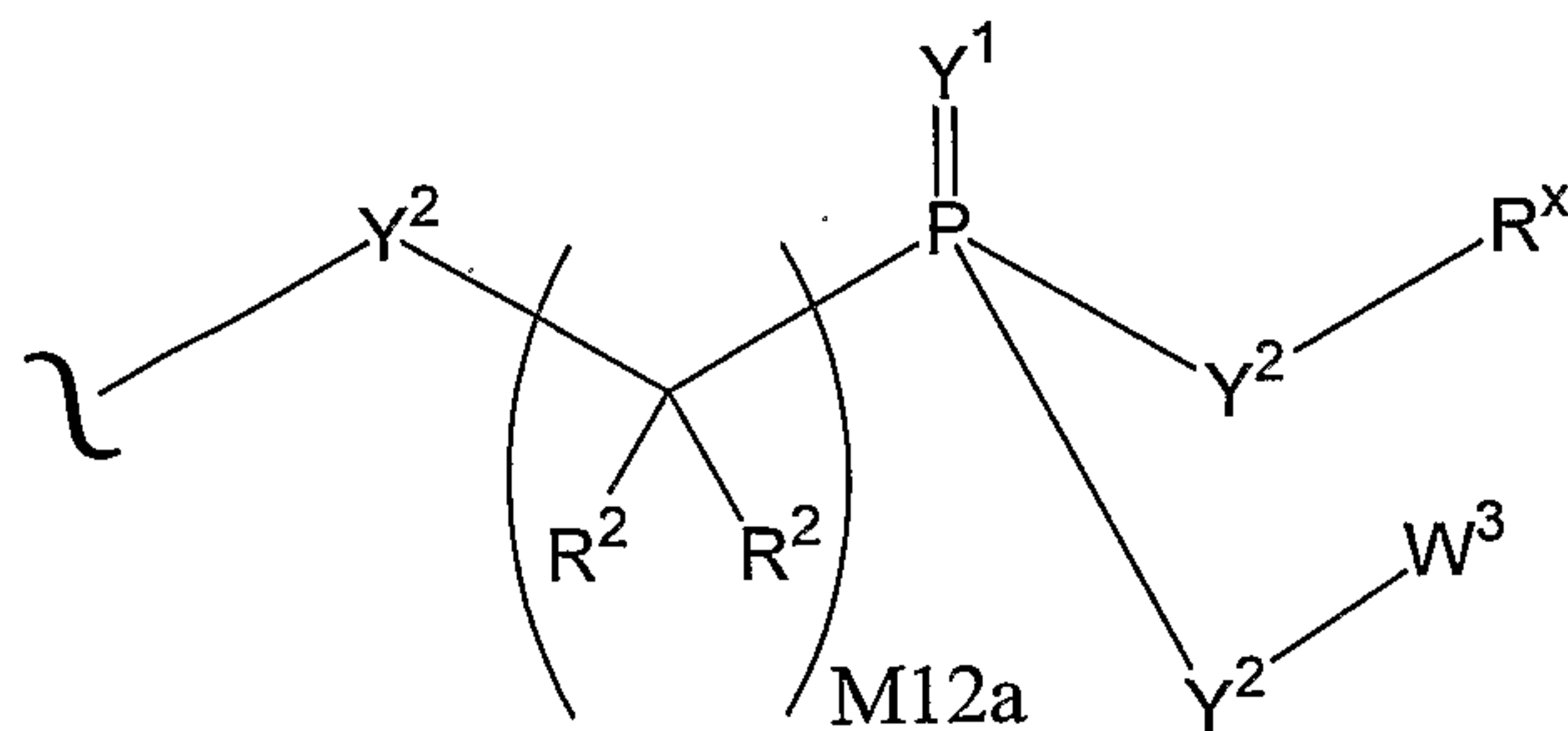


and Y^{2b} is O or $N(R^2)$.

43. The compound of any one of claims 1-14 wherein A^3 is of the formula:

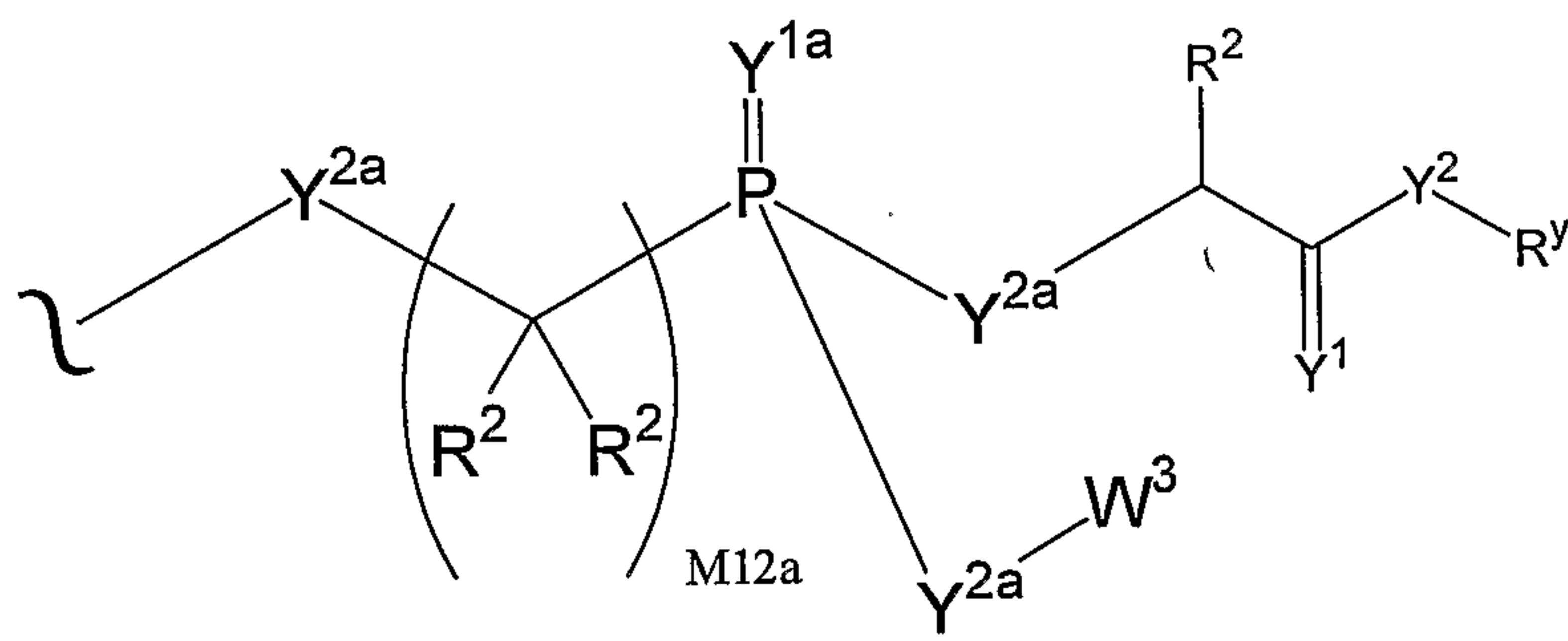


44. The compound of any one of claims 1-14 wherein A^3 is of the formula:



45. The compound of any one of claims 1-14 wherein A^3 is of the formula:

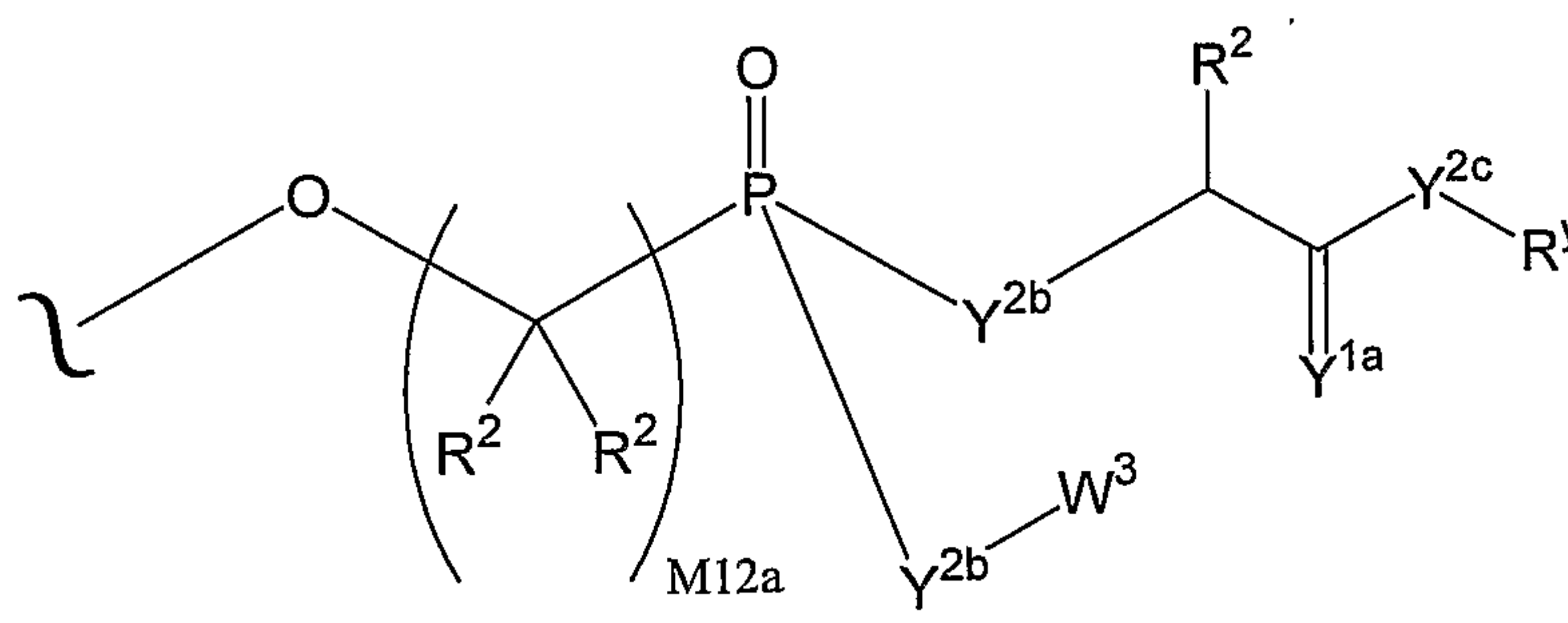
5



Y^{1a} is O or S; and

Y^{2a} is independently O, $N(R^2)$ or S.

- 10 46. The compound of any one of claims 1-14 wherein A^3 is of the formula:



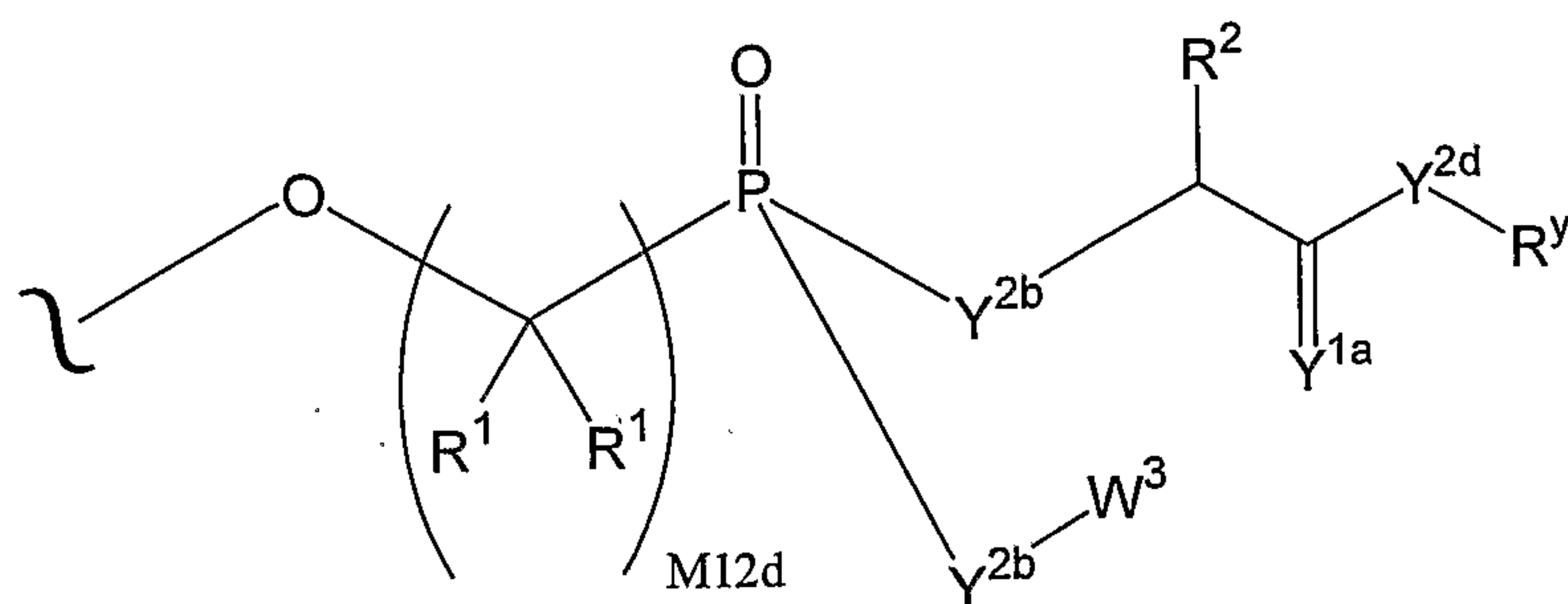
Y^{1a} is O or S;

Y^{2b} is independently O or $N(R^2)$; and

Y^{2c} is O, $N(R^y)$ or S.

15

47. The compound of any one of claims 1-14 wherein A^3 is of the formula:



R^1 is independently H or alkyl of 1 to 18 carbon atoms;

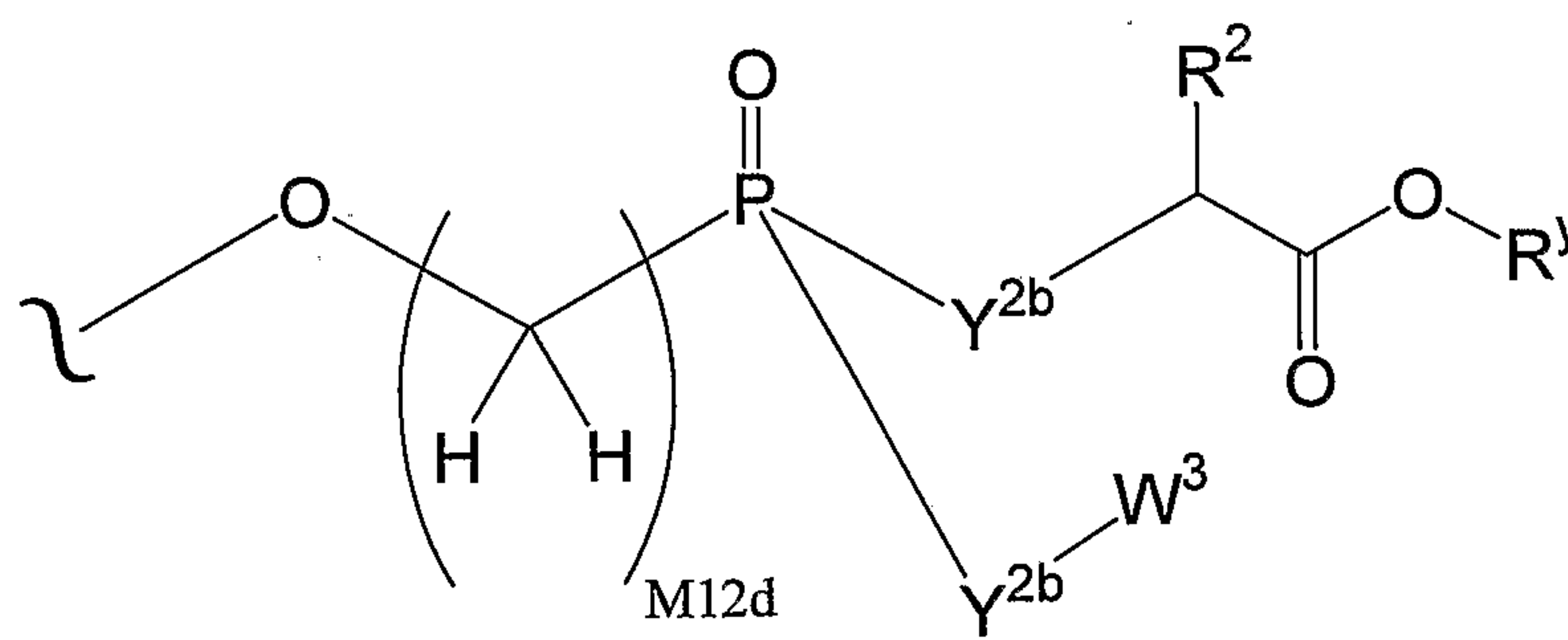
5 Y^{1a} is O or S;

Y^{2b} is independently O or $N(R^2)$;

Y^{2d} is O or $N(R^y)$; and

M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

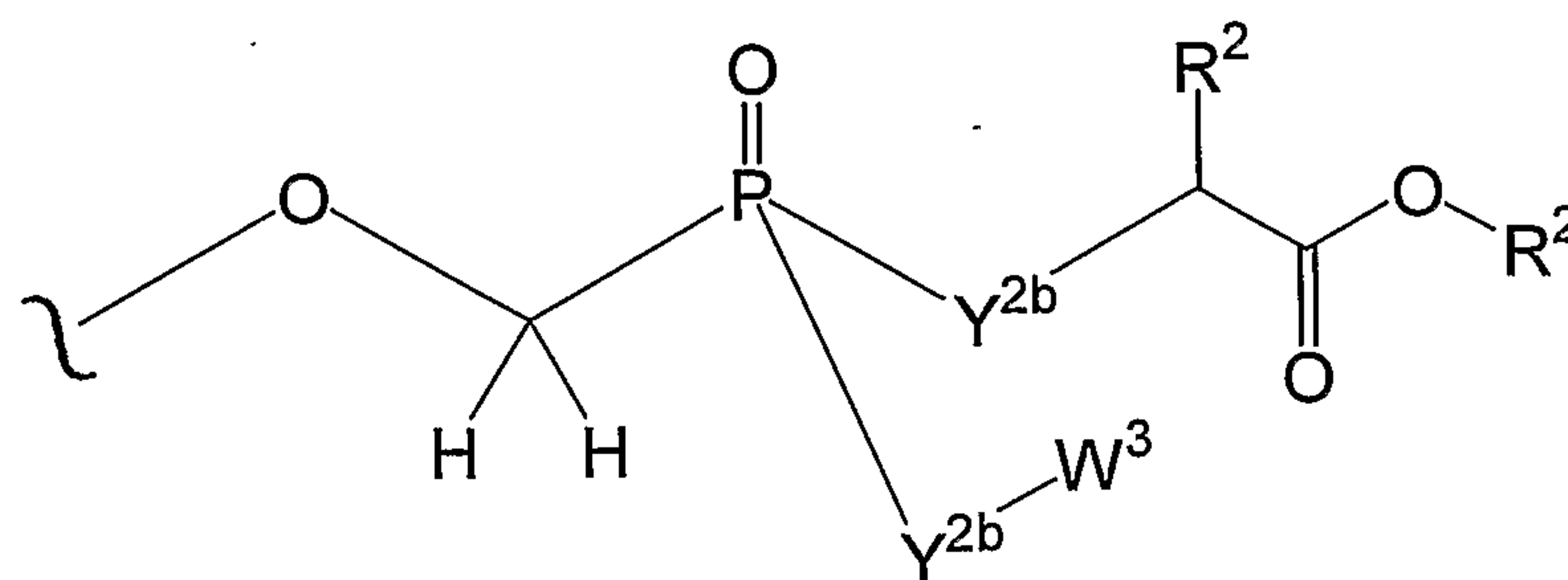
- 10 48. The compound of any one of claims 1-14 wherein A^3 is of the formula:



Y^{2b} is independently O or $N(R^2)$; and

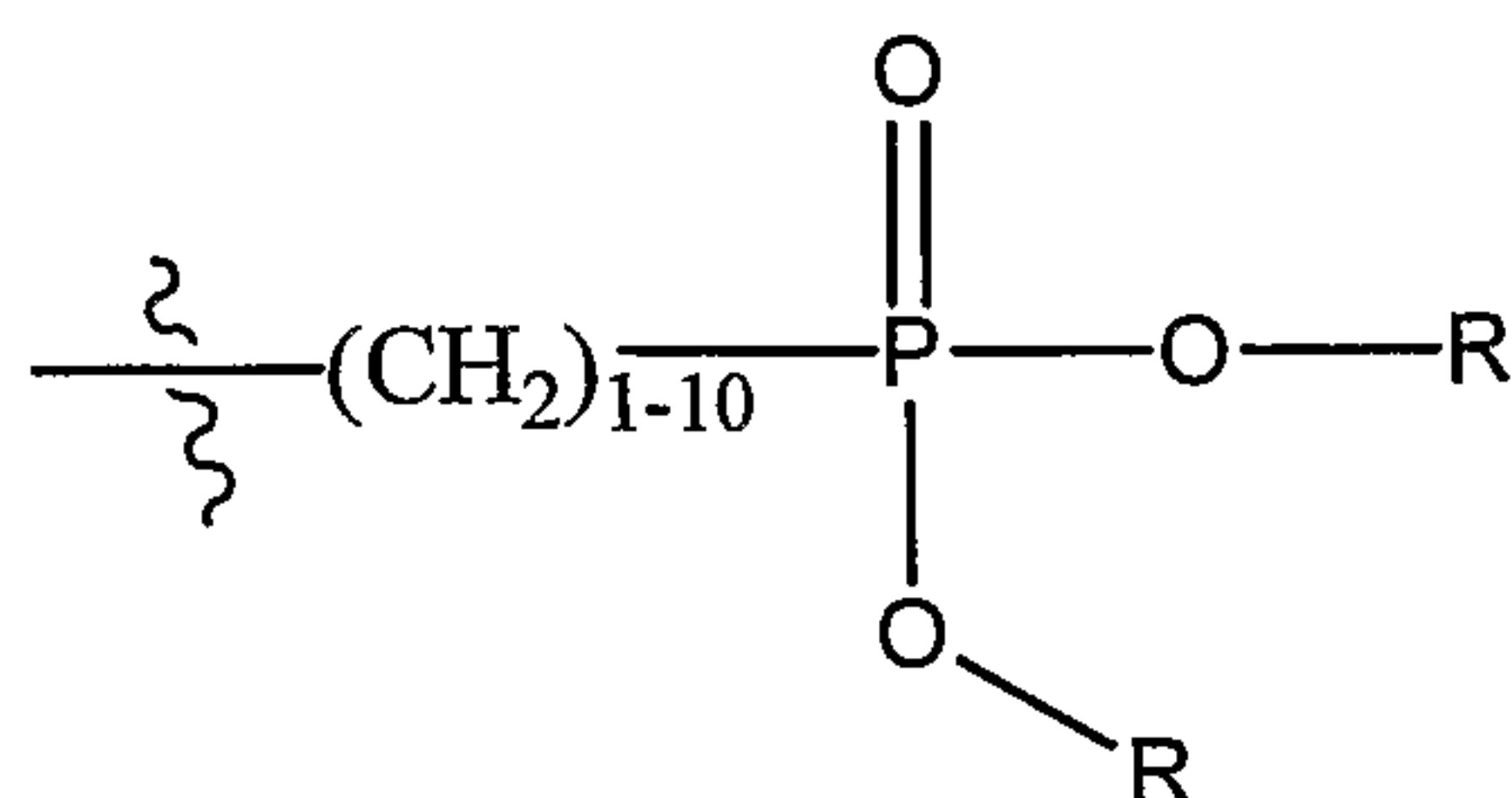
M12d is 1, 2, 3, 4, 5, 6, 7 or 8.

- 15 49. The compound of any one of claims 1-14 wherein A^3 is of the formula:



and Y^{2b} is independently O or $N(R^2)$.

50. The compound of claim 1 wherein A⁰ is of the formula:



wherein each R is independently (C₁-C₆)alkyl.

51. The compound of any one of claims 1-50 wherein R²⁰ is methoxy.

52. The compound of any one of claims 1-50 wherein R²⁰ is vinyl or ethyl.

53. The compound of any one of claims 1-52 wherein each alkyl has 1 to 6 carbon atoms and aryl is phenyl.

54. The compound of any one of claims 1-53 wherein Z is formula ZA; G is -O(CH₂)_nN=G³; n is an integer from 2 to 4; and =G³ is alkylene of four to six carbon atoms, or alkylene of three to five carbon atoms plus one member that is -O-, -S-, or -N(G⁴)-; where G⁴ is hydrogen or alkyl.

55. The compound of claim 54 wherein =G³ is alkylene of four to six carbon atoms, or alkylene of three to five carbon atoms plus one member that is -O- or -S-; where G⁴ is hydrogen or alkyl.

56. The compound of any one of claims 1-53 wherein G is not OH.

57. The compound of any one of claims 1-56 wherein G is not morpholinoethoxy.

58. The compound of any one of claims 1-57 wherein Z³ is not -P(O)(OCH₃)₂, or -P(O)(OH)(OCH₃) and Z⁴ is not -P(O)(OCH₃)₂, or -P(O)(OH)(OCH₃).

59. The compound of any one of claims 1-58 wherein Z^3 is not $-P(O)(OH)_2$ and Z^4 is not $-P(O)(OH)_2$.
60. The compound of any one of claims 1-50 wherein R^{21} is (C_1-C_6) alkyl.
- 5 61. The compound of any one of claims 1-50 or 60 wherein R^{22} is (C_1-C_6) alkyl.
62. The compound of any one of claims 1-50 or 60 wherein R^{22} is substituted
10 phenyl.
63. The compound of claim 62 wherein phenyl is substituted with CO_2R^{46} , wherein R^{46} is H or alkyl.
- 15 64. The compound of claim 62 or 63 wherein phenyl is substituted at the *para* position.
65. The compound of claim 63 or 64 wherein R^{46} is H.
- 20 66. The compound of claim 63 or 64 wherein R^{46} is alkyl.
67. The compound of claim 63 or 64 wherein R^{46} is (C_1-C_6) alkyl.
68. The compound of any one of claims 1-50 or 60-67 wherein R^{23} is
25 methoxy.
69. The compound of any one of claims 1-50 or 60-67 wherein R^{23} is ethyl or vinyl.
- 30 70. The compound of any one of claims 1-69 wherein A^0 comprises a phosphonate other than $P(O)(OH)(OMe)$.

71. The compound of any one of claims 1-70 wherein A⁰ comprises a phosphonate other than P(O)(OMe)₂.

72. The compound of any one of claims 1-71 wherein A⁰ comprises a phosphonate other than P(O)(OH)₂.

73. The compound of any one of claims 1-50 wherein R³⁰ is methoxy.

74. The compound of any one of claims 1-50 wherein R³⁰ is ethyl.

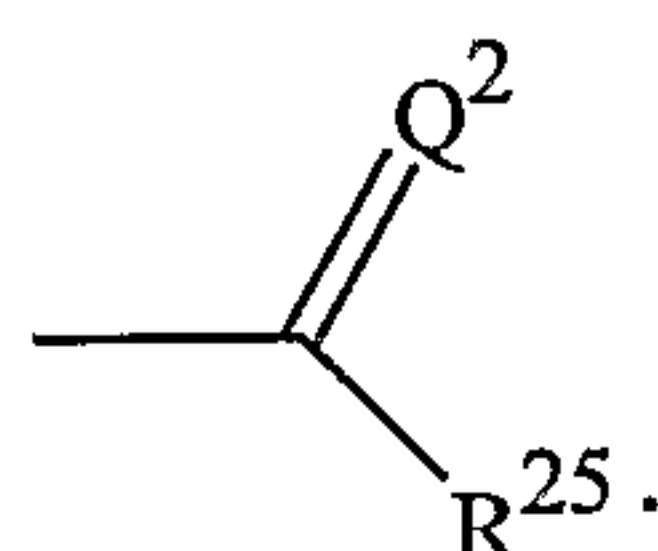
75. The compound of any one of claims 1-50 wherein R³⁰ is vinyl.

76. The compound of any one of claims 1-75 wherein A⁰ comprises a phosphonate other than -P(=O)(OH)(OMe) or -P(=O)(OMe)(OMe).

77. The compound of any one of claims 1-75 wherein A⁰ comprises a phosphonate other than -P(=O)(OH)(OH).

78. The compound of any one of claims 1-50 or 73-77 wherein R²⁵ is alkyl, haloalkyl or -NR²⁸R²⁹, where R²⁸ and R²⁹ are independently H, alkyl, haloalkyl, cycloalkyl, phenyl optionally monosubstituted with halogen, hydroxy, carboxy, chlorocarbonyl, nitro, cyano, phenyl, alkyl, acyl, alkoxycarbonyl, acylamino, dialkylamino or dialkylaminoethoxycarbonyl, phenyl optionally disubstituted with hydroxy, carboxy, nitro or alkyl, or benzyl optionally substituted with dialkylamino.

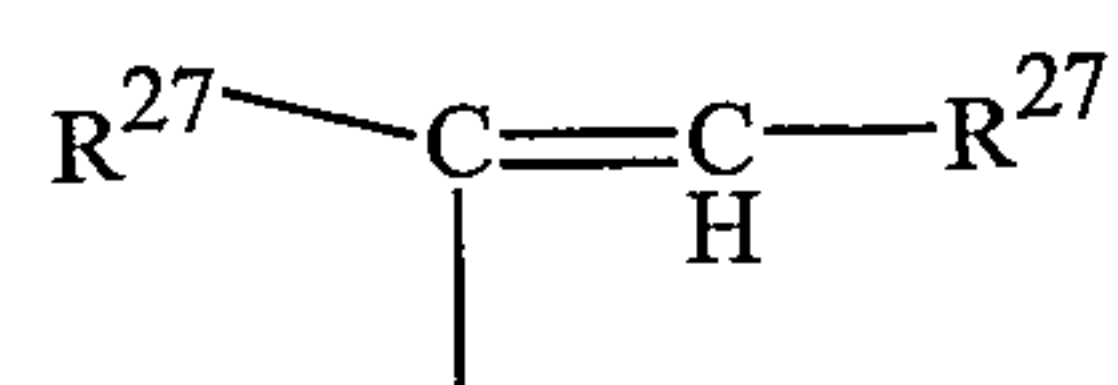
79. The compound of any one of claims 1-50 or 73-77 wherein R²⁴ is



80. The compound of any one of claims 1-50 or 73-77 wherein R²⁴ is



81. The compound of any one of claims 1-50 or 73-77 wherein R^{24} is



82. The compound of any one of claims 1-50 or 73-81 wherein Q^2 is oxygen.

5

83. The compound of any one of claims 1-50 or 73-81 wherein Q^2 is sulfur.

84. The compound of any one of claims 1-50 or 73-77 or 80 wherein n_a is 1, 2, 3, or 4.

10

85. The compound of any one of claims 1-50 wherein R^{31} is methoxy.

86. The compound of any one of claims 1-50 wherein R^{31} is ethyl or vinyl.

- 15 87. The compound of any one of claims 1-50 or 85-86 wherein R^{32} and R^{33} are each independently H.

88. The compound of any one of claims 1-50 or 85-86 wherein R^{32} and R^{33} are each independently alkyl having 1 to 6 carbon atoms.

20

89. The compound of any one of claims 1-50 or 85-86 wherein R^{32} and R^{33} are each methyl or ethyl.

90. The compound of any one of claims 1-50 or 85-89 wherein Q^3 is O.

25

91. The compound of any one of claims 1-50 or 85-89 wherein Q^3 is S.

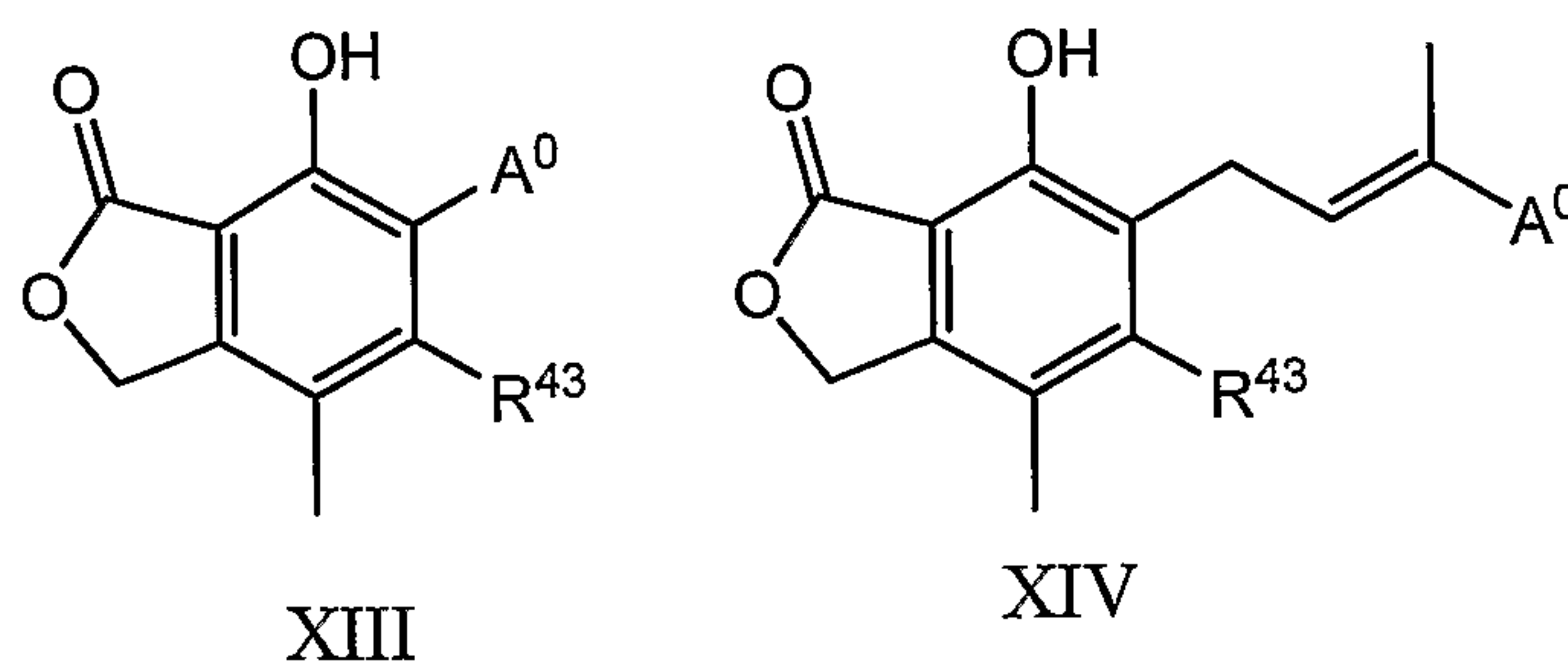
92. The compound of any one of claims 1-50 or 85-91 wherein n_b is 2, 3, or 4.

30

93. The compound of any one of claims 1-50 or 85-91 wherein n_b is 1.

94. The compound of any one of claims 1-50 wherein R^{34} is methoxy.
95. The compound of any one of claims 1-50 wherein R^{34} is ethyl.
- 5 96. The compound of any one of claims 1-50 wherein R^{34} is vinyl.
97. The compound of any one of claims 1-50 or 94-96 wherein R^{36} is hydrogen, alkyl, $-C(O)R^{39}$, $-C(O)NR^{40}R^{41}$, $-CO_2R^{42}$, or $-SO_2R^{39}$.
- 10 98. The compound of any one of claims 1-50 or 94-96 wherein R^{36} is hydrogen, alkyl having 1-6 carbons, $-C(O)R^{39}$, $-C(O)NR^{40}R^{41}$, $-CO_2R^{42}$, or $-SO_2R^{39}$.
- 15 99. The compound of any one of claims 1-50 or 94-98 wherein R^{39} is hydrogen, alkyl having 1-6 carbons, halo substituted alkyl having 1-6 carbons, phenyl, or substituted phenyl.
100. The compound of any one of claims 1-50 or 94-99 wherein R^{40} is hydrogen, alkyl having 1-6 carbons, phenyl, or substituted phenyl.
- 20 101. The compound of any one of claims 1-50 or 94-100 wherein R^{41} is hydrogen, alkyl having 1-6 carbons, phenyl, or substituted phenyl.
102. The compound of any one of claims 1-50 or 94-101 wherein R^{42} is alkyl
25. having 1-6 carbons, phenyl, or substituted phenyl.
103. The compound of any one of claims 1-102 wherein A^0 comprises a phosphonate other than $-P(=O)(OH)(OMe)$ or $-P(=O)(OMe)(OMe)$.
- 30 104. The compound of any one of claims 1-103 wherein A^0 comprises a phosphonate other than $-P(=O)(OH)(OH)$.

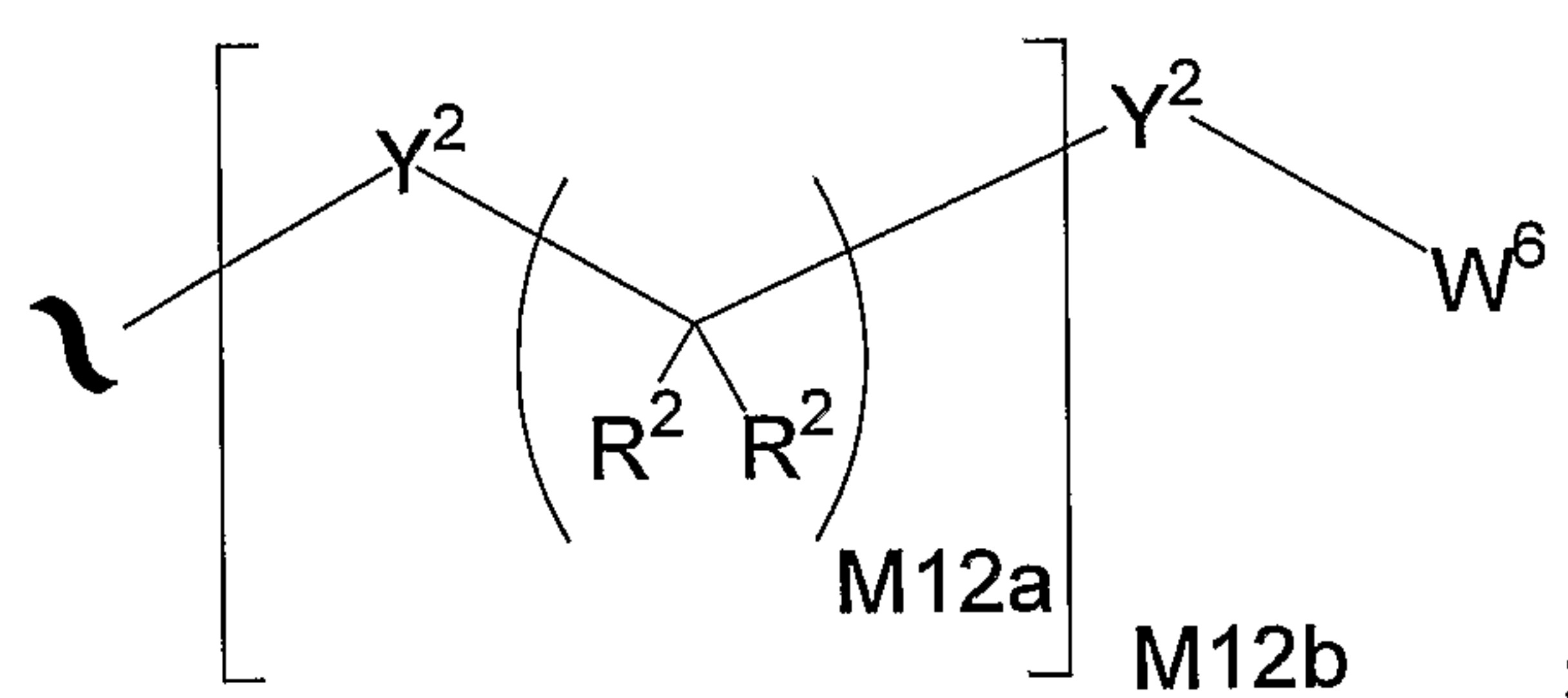
105. A compound of formula XIII or XIV:



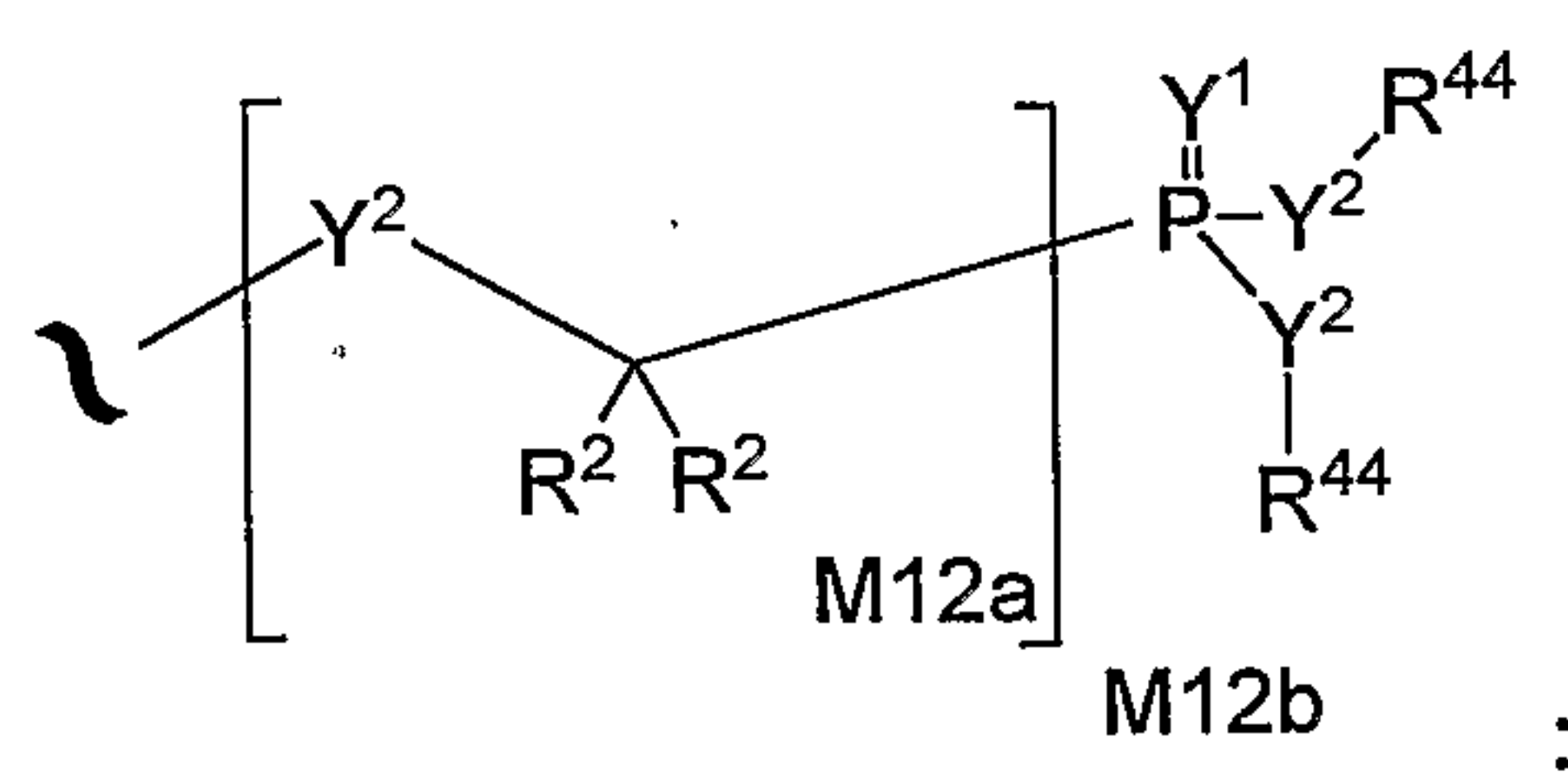
wherein:

A^0 is A^1 ;

5 A^1 is:



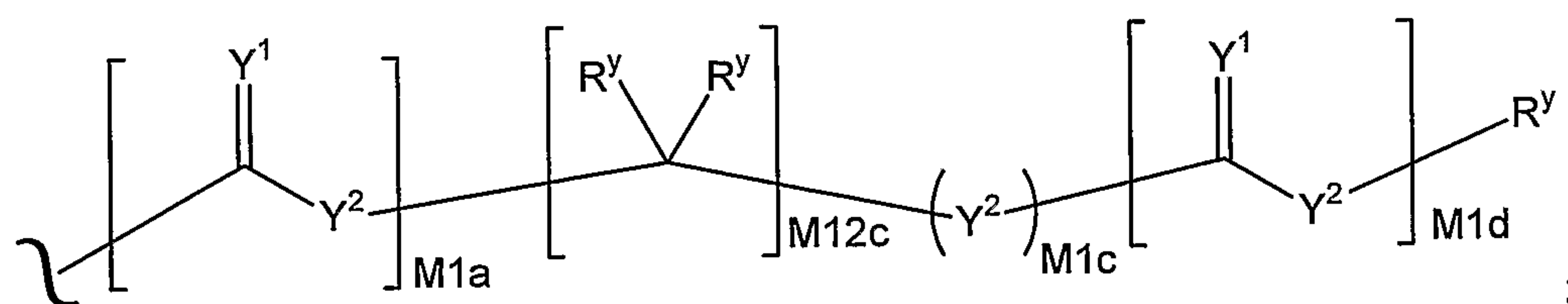
A^3 is:



Y^1 is independently O, S, $N(R^x)$, $N(OR^x)$, or $N(N(R^x)(R^x))$;

10 Y^2 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, or $-S(O)_{M2}-$; and when Y^2 joins two phosphorous atoms Y^2 can also be $C(R^2)(R^2)$;

R^x is independently H, R^2 , W^3 , a protecting group, or the formula:



R^y is independently H, W^3 , R^2 or a protecting group;

15 R^2 is independently H, R^3 or R^4 wherein each R^4 is independently substituted with 0 to 3 R^3 groups;

R^3 is R^{3a} , R^{3b} , R^{3c} or R^{3d} , provided that when R^3 is bound to a heteroatom, then R^3 is R^{3c} or R^{3d} ;

R^{3a} is F, Cl, Br, I, -CN, N_3 or $-NO_2$;

R^{3b} is Y^1 ;

5 R^{3c} is $-R^x$, $-N(R^x)(R^x)$, $-SR^x$, $-S(O)R^x$, $-S(O)_2R^x$, $-S(O)(OR^x)$, $-S(O)_2(OR^x)$, $-OC(Y^1)R^x$, $-OC(Y^1)OR^x$, $-OC(Y^1)(N(R^x)(R^x))$, $-SC(Y^1)R^x$, $-SC(Y^1)OR^x$, $-SC(Y^1)(N(R^x)(R^x))$, $-N(R^x)C(Y^1)R^x$, $-N(R^x)C(Y^1)OR^x$, or $-N(R^x)C(Y^1)(N(R^x)(R^x))$;

R^{3d} is $-C(Y^1)R^x$, $-C(Y^1)OR^x$ or $-C(Y^1)(N(R^x)(R^x))$;

10 R^4 is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms, or alkynyl of 2 to 18 carbon atoms;

R^5 is R^4 wherein each R^4 is substituted with 0 to 3 R^3 groups;

W^3 is W^4 or W^5 ;

W^4 is R^5 , $-C(Y^1)R^5$, $-C(Y^1)W^5$, $-SO_2R^5$, or $-SO_2W^5$;

15 W^5 is carbocycle or heterocycle wherein W^5 is independently substituted with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

M_2 is 0, 1 or 2;

M_{12a} is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

20 M_{12b} is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

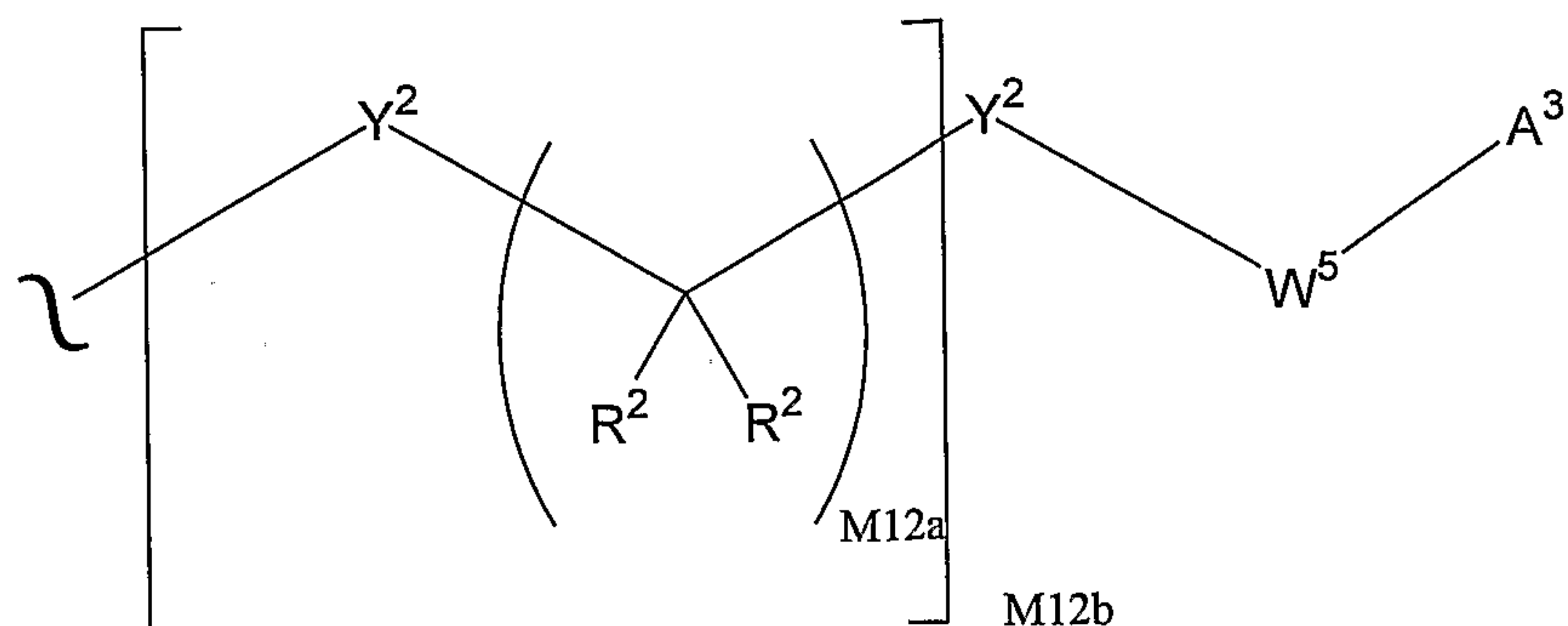
M_{1a} , M_{1c} , and M_{1d} are independently 0 or 1;

M_{12c} is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

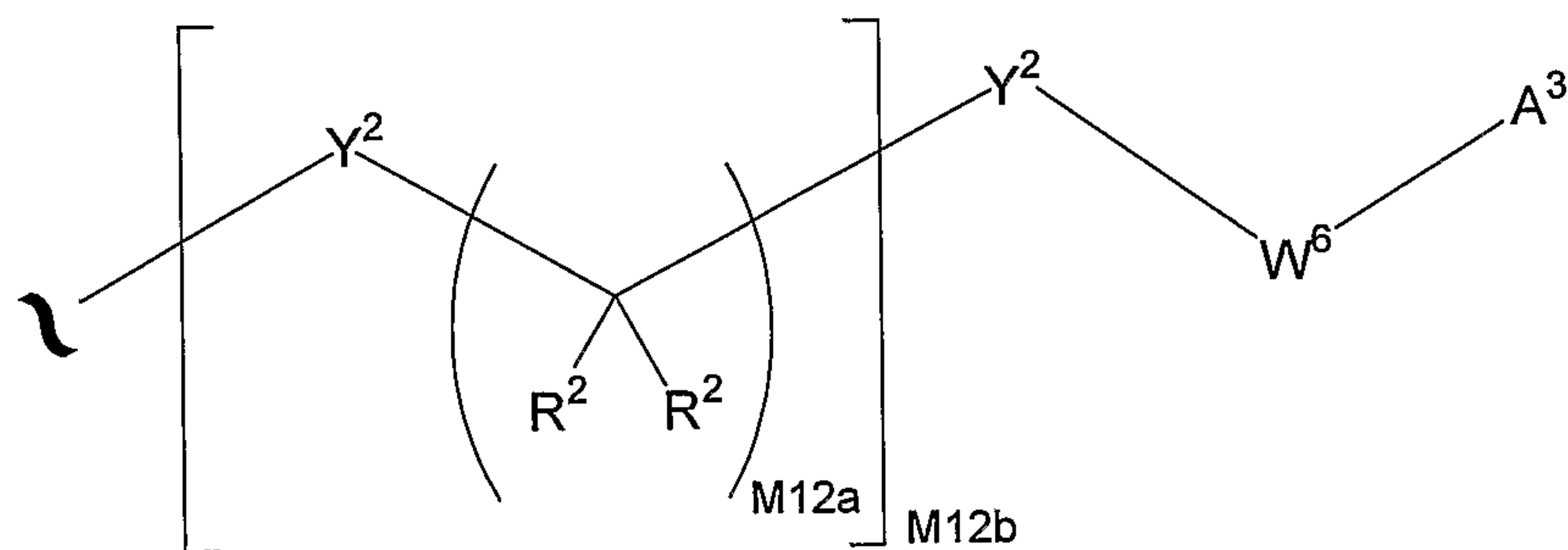
R^{43} is ethyl, methoxy, or vinyl; and

25 R^{44} is R^x or $-N=C(R^4)(R^4)$, provided that at least one $-Y^2-R^{44}$ of A^3 is $-O-N=C(R^4)(R^4)$.

106. The compound of claim 105 wherein A^1 is of the formula:

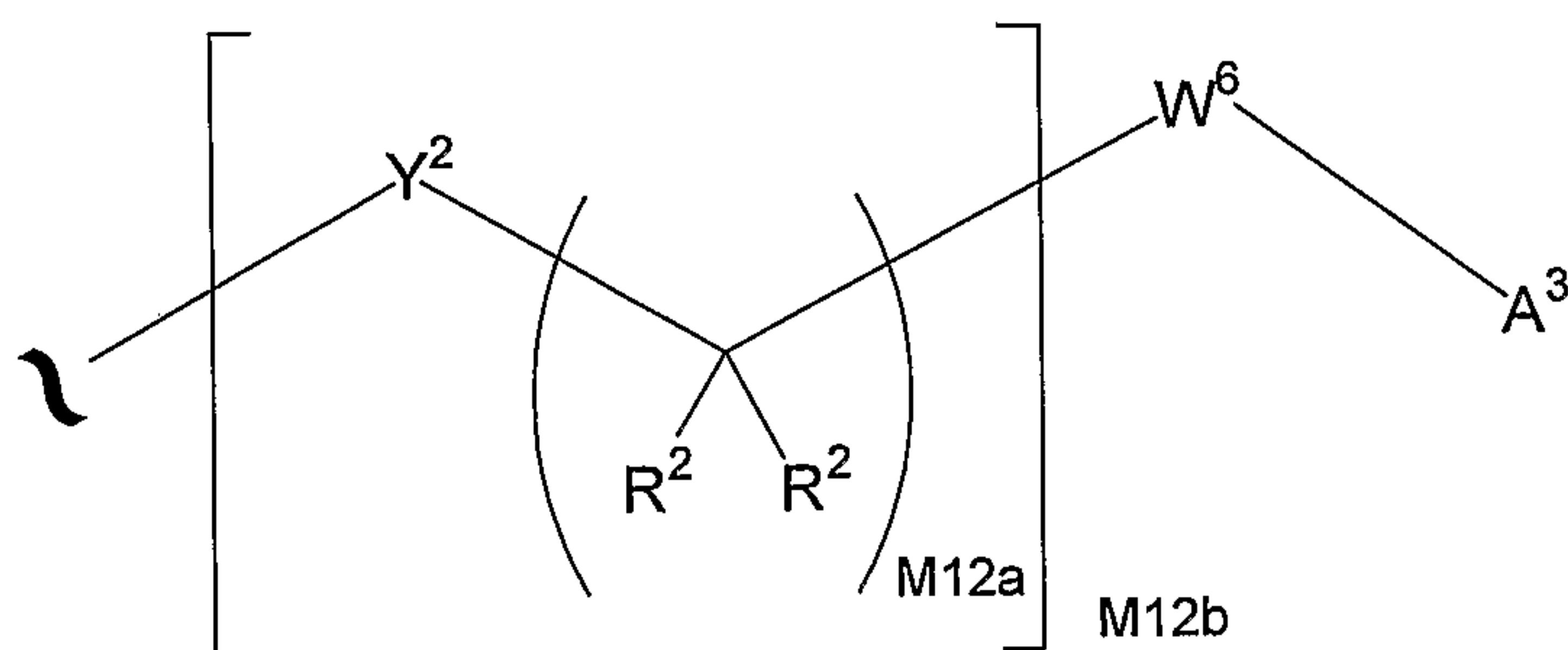


107. The compound of claim 105 wherein A^1 is of the formula:

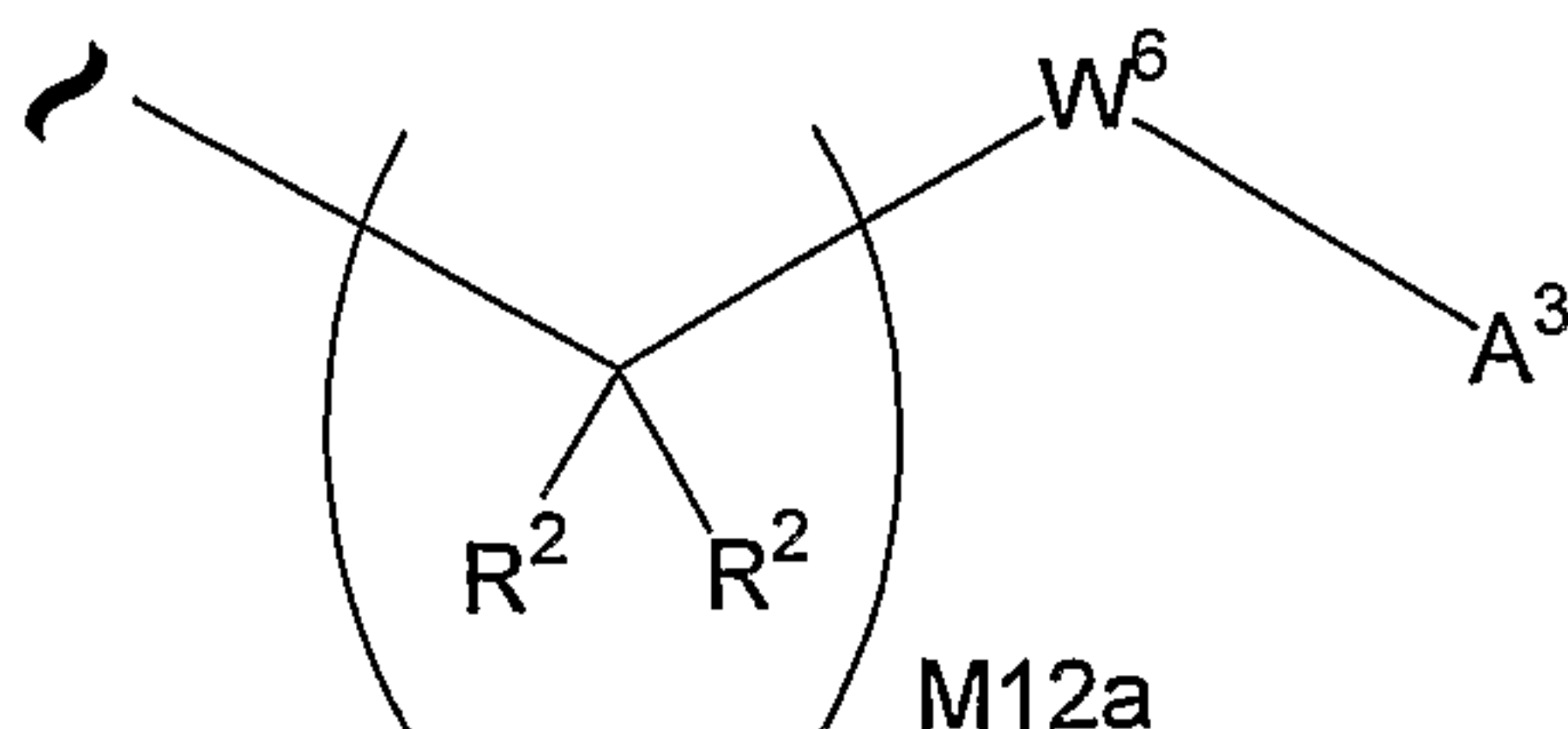


5

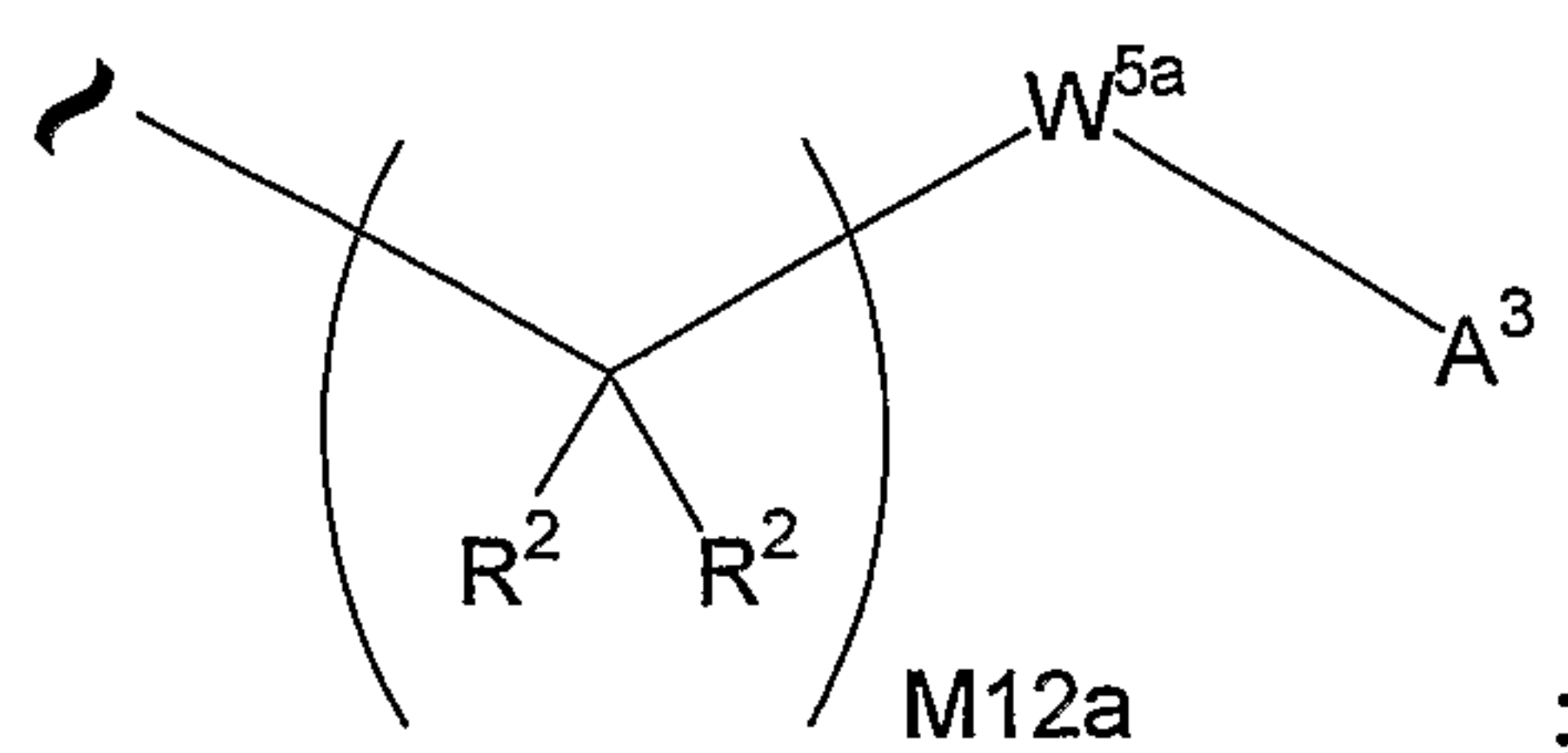
108. The compound of claim 105 wherein A^1 is of the formula:



10 109. The compound of claim 105 wherein A^1 is of the formula:



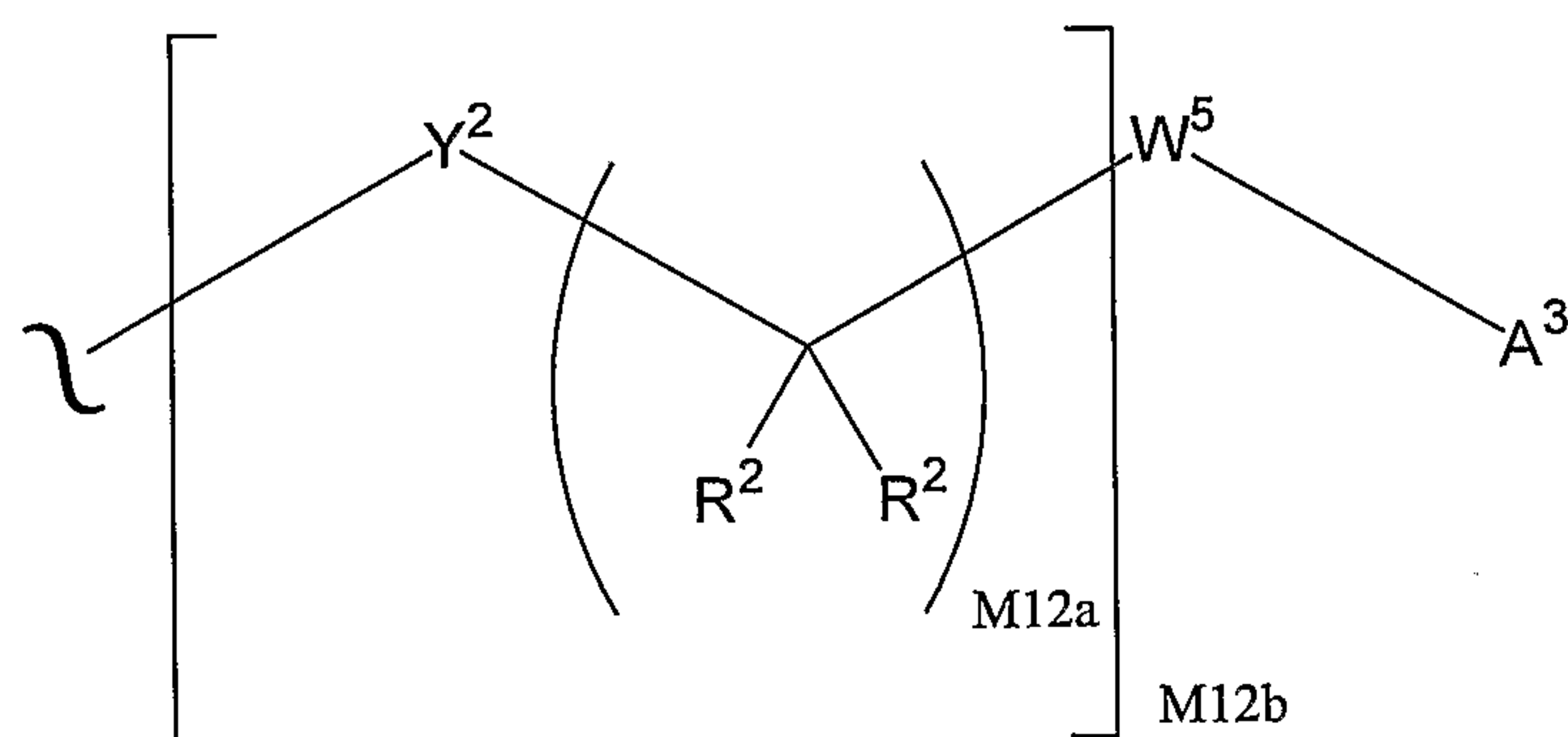
110. The compound of claim 105 wherein A^1 is of the formula:



and W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently
5 substituted with 0 or 1 R^2 groups.

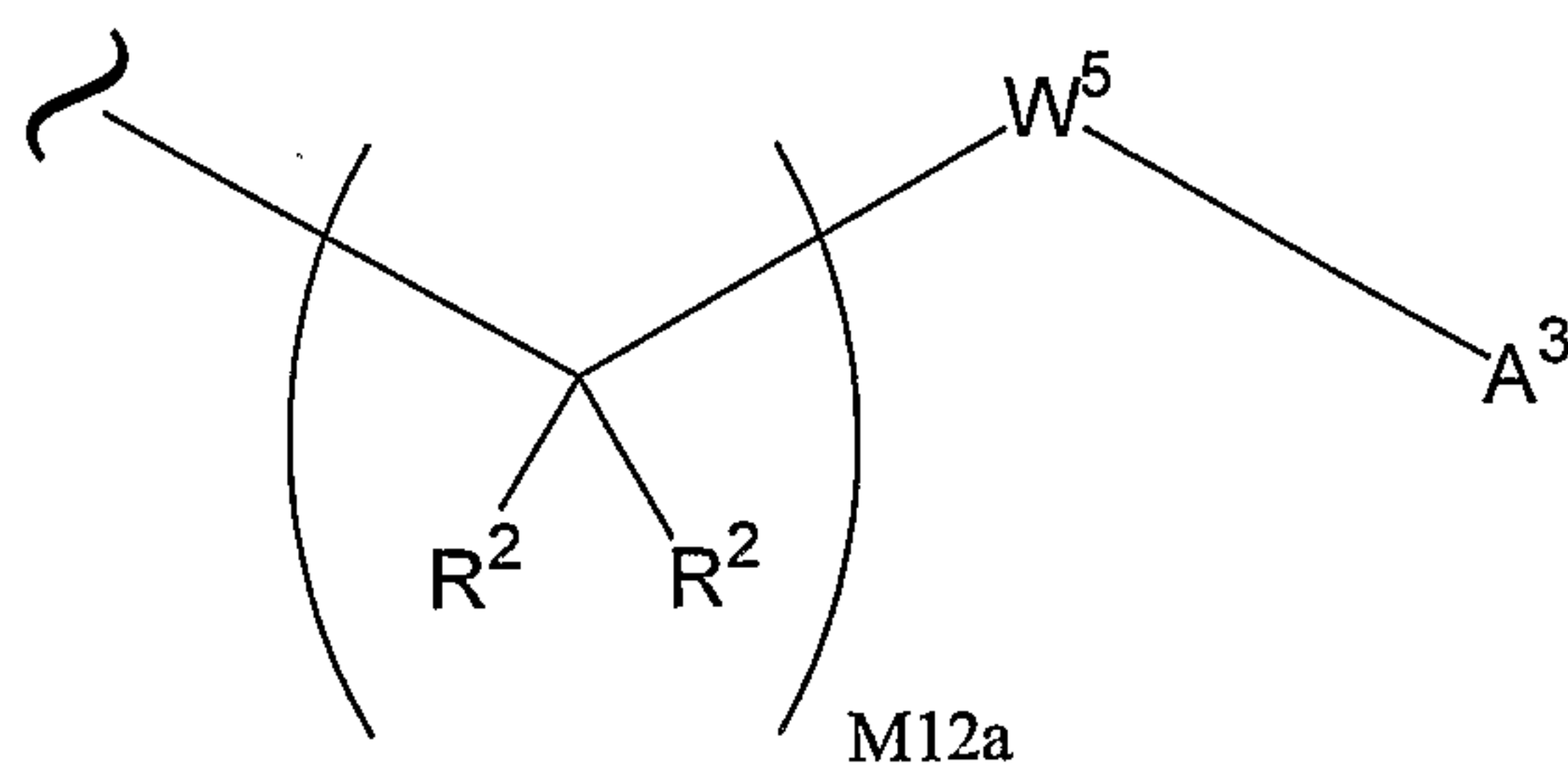
111. The compound of claim 105 wherein M12a is 1.

112. The compound of claim 105 wherein A^1 is of the formula:

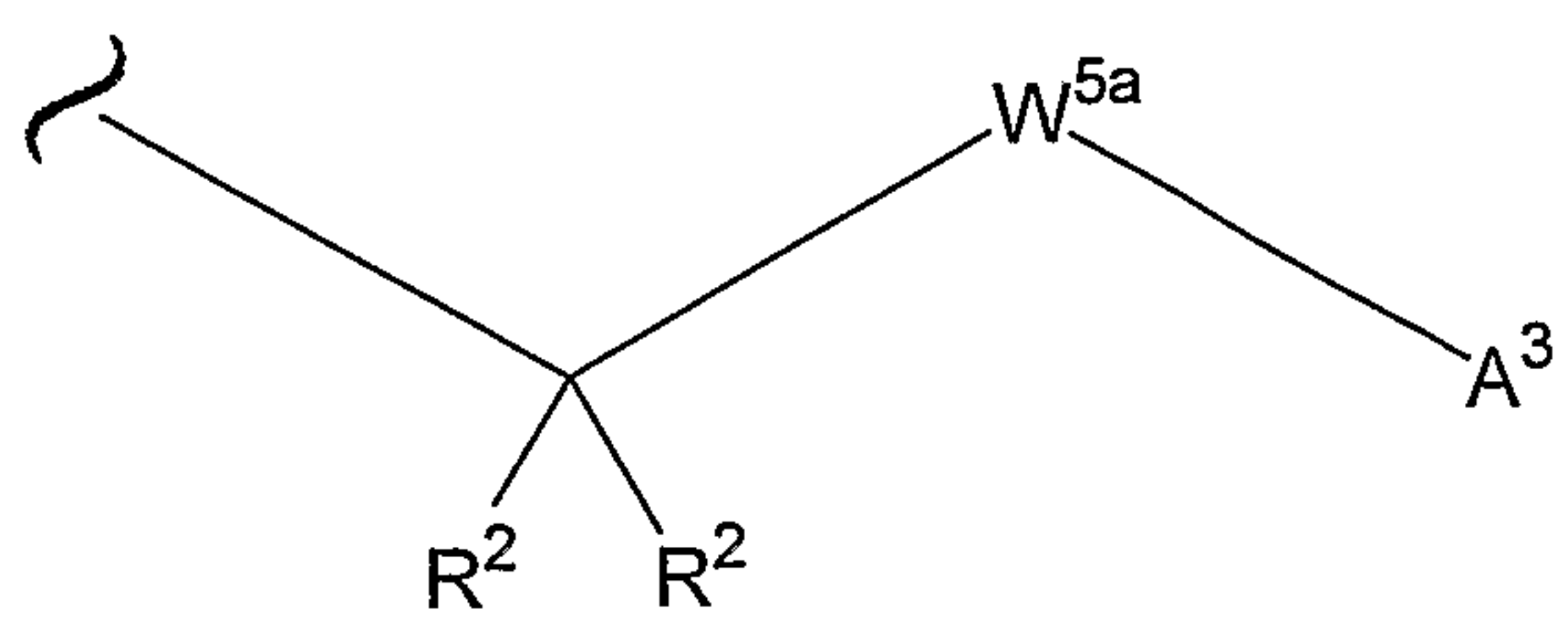


10

113. The compound of claim 105 wherein A^1 is of the formula:



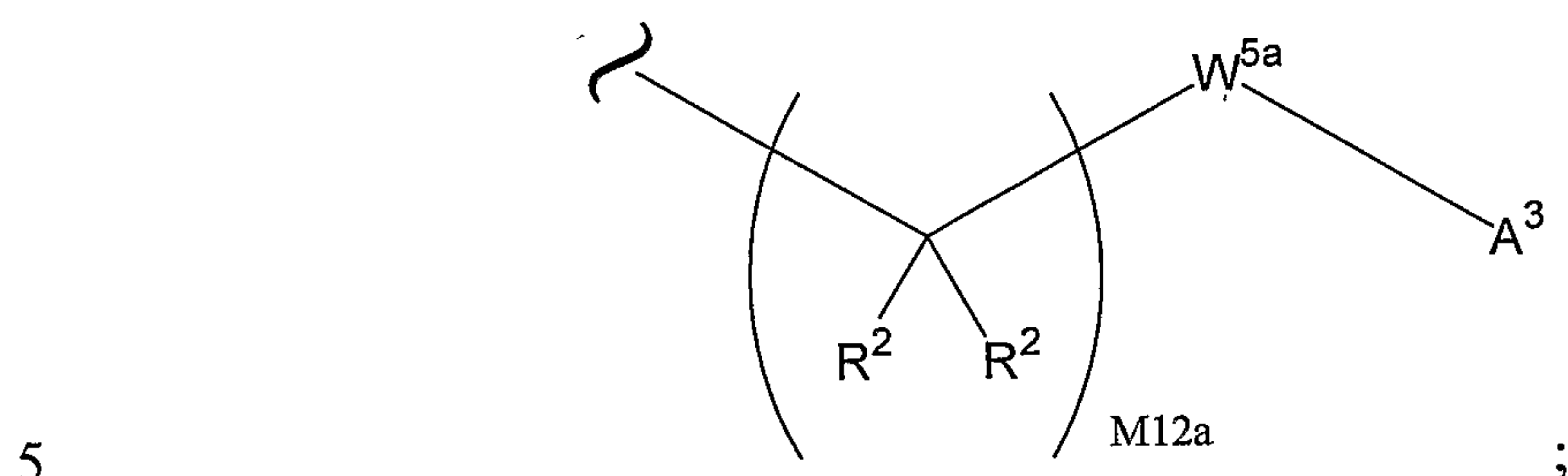
114. The compound of claim 105 wherein A^1 is of the formula:



15

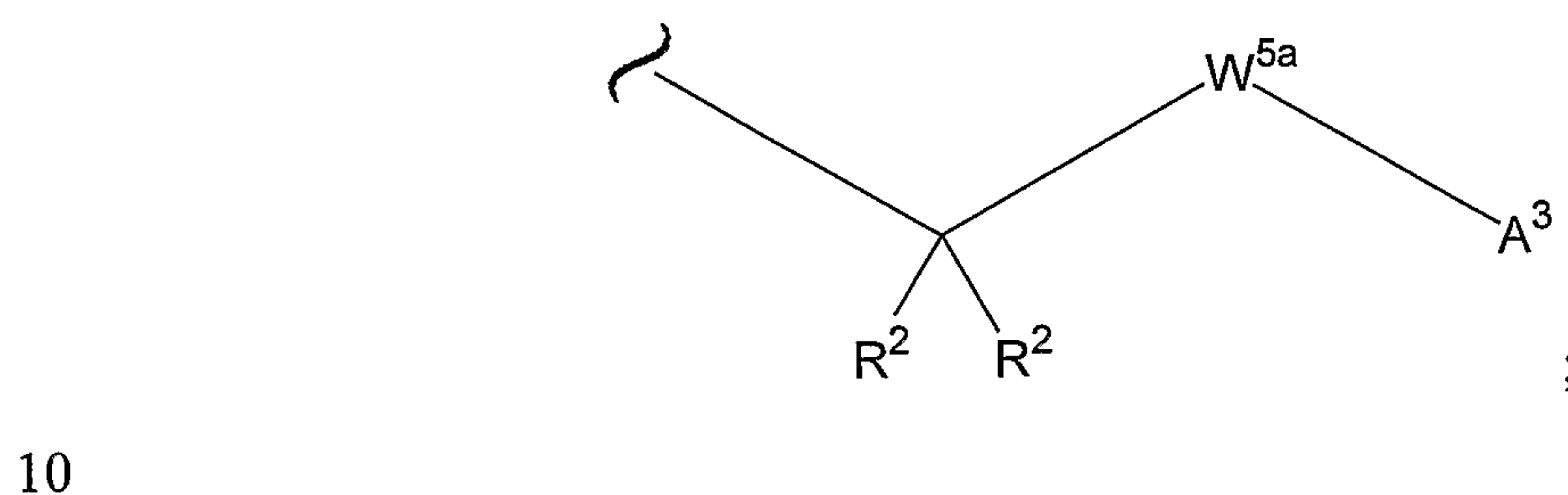
W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

115. The compound of claim 105 wherein A^1 is of the formula:



W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

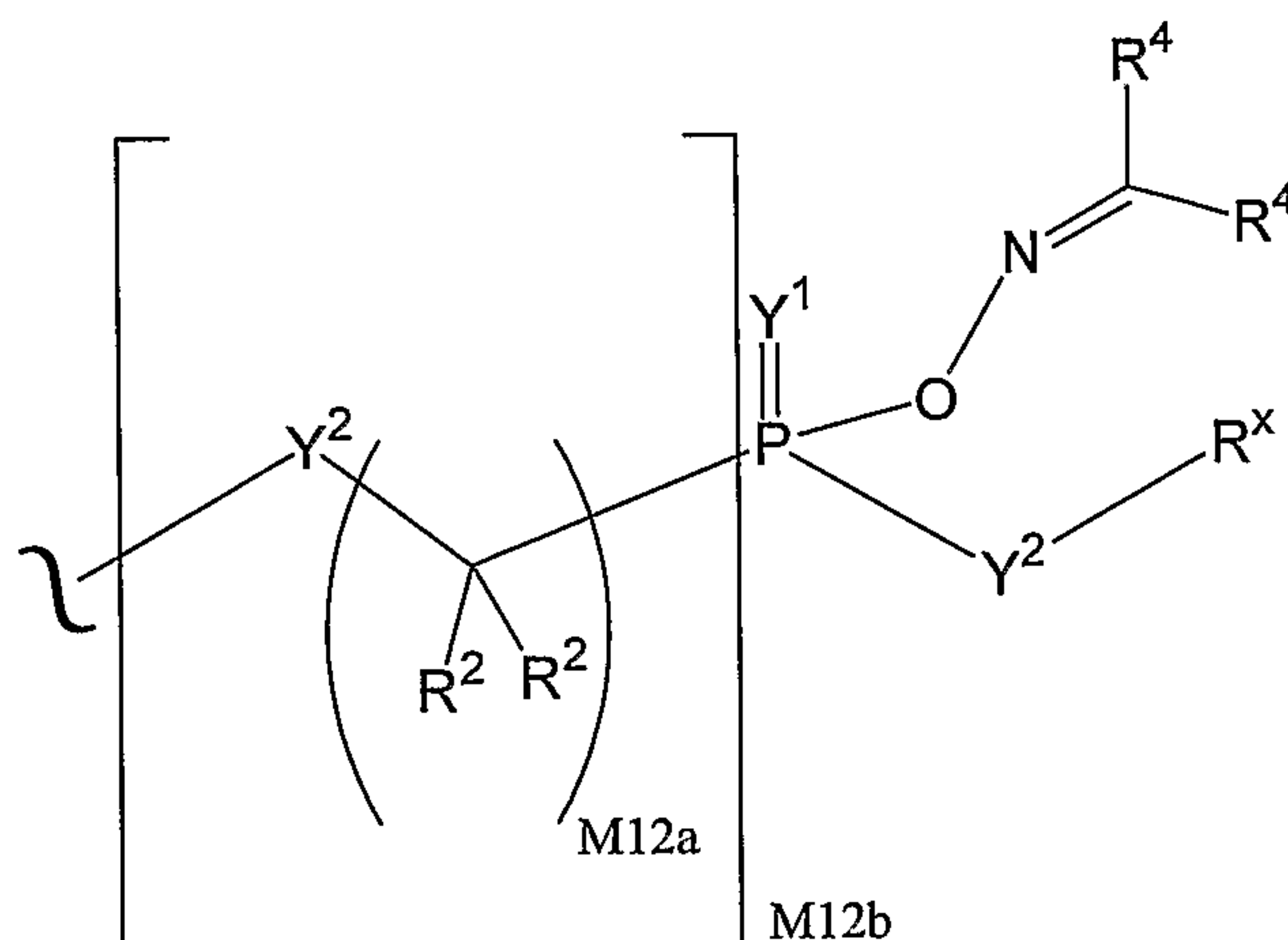
116. The compound of claim 105 wherein A^1 is of the formula:



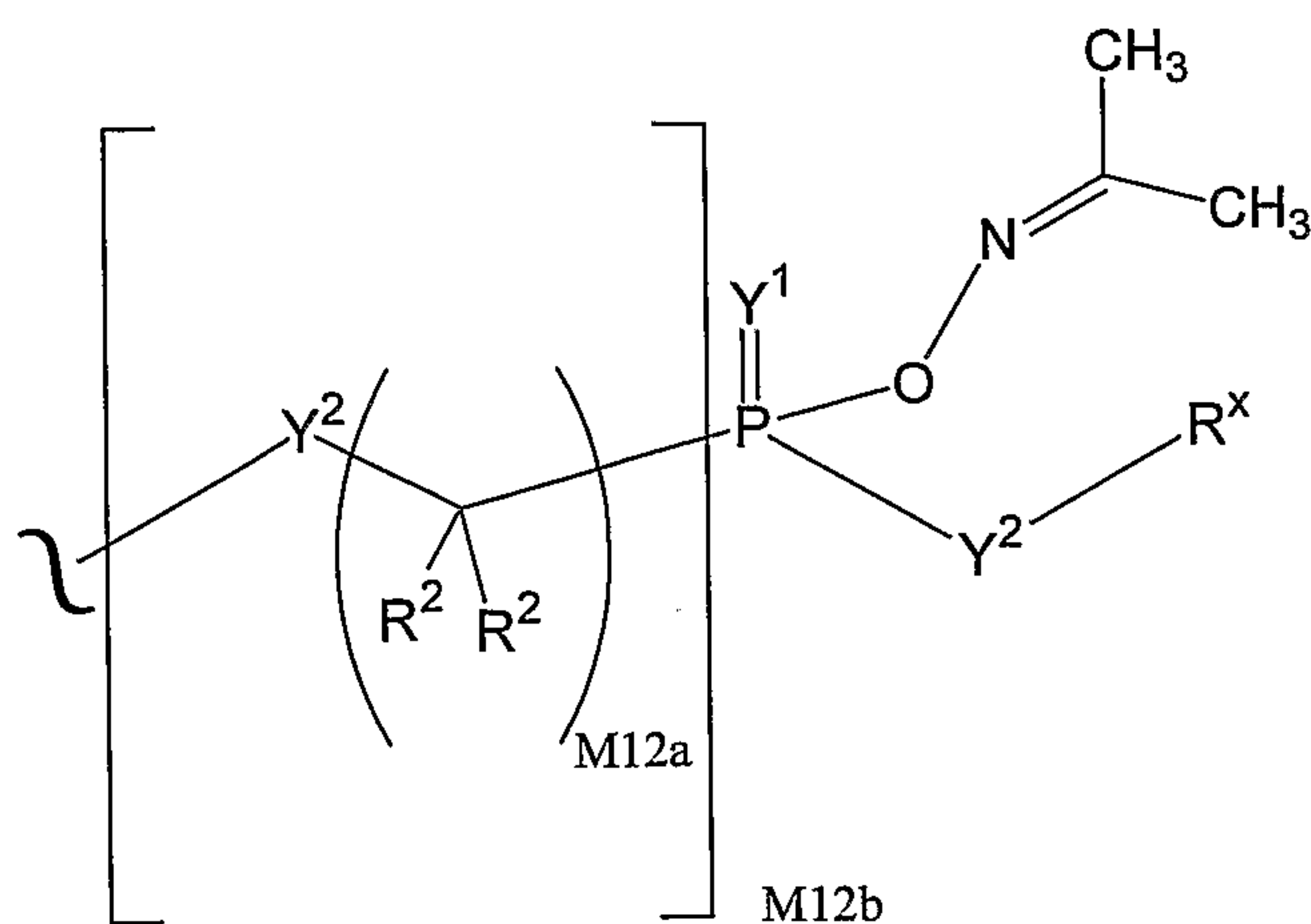
W^{5a} is a carbocycle or heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

117. The compound of any one of claims 105-116 wherein A^3 is of the formula:

15

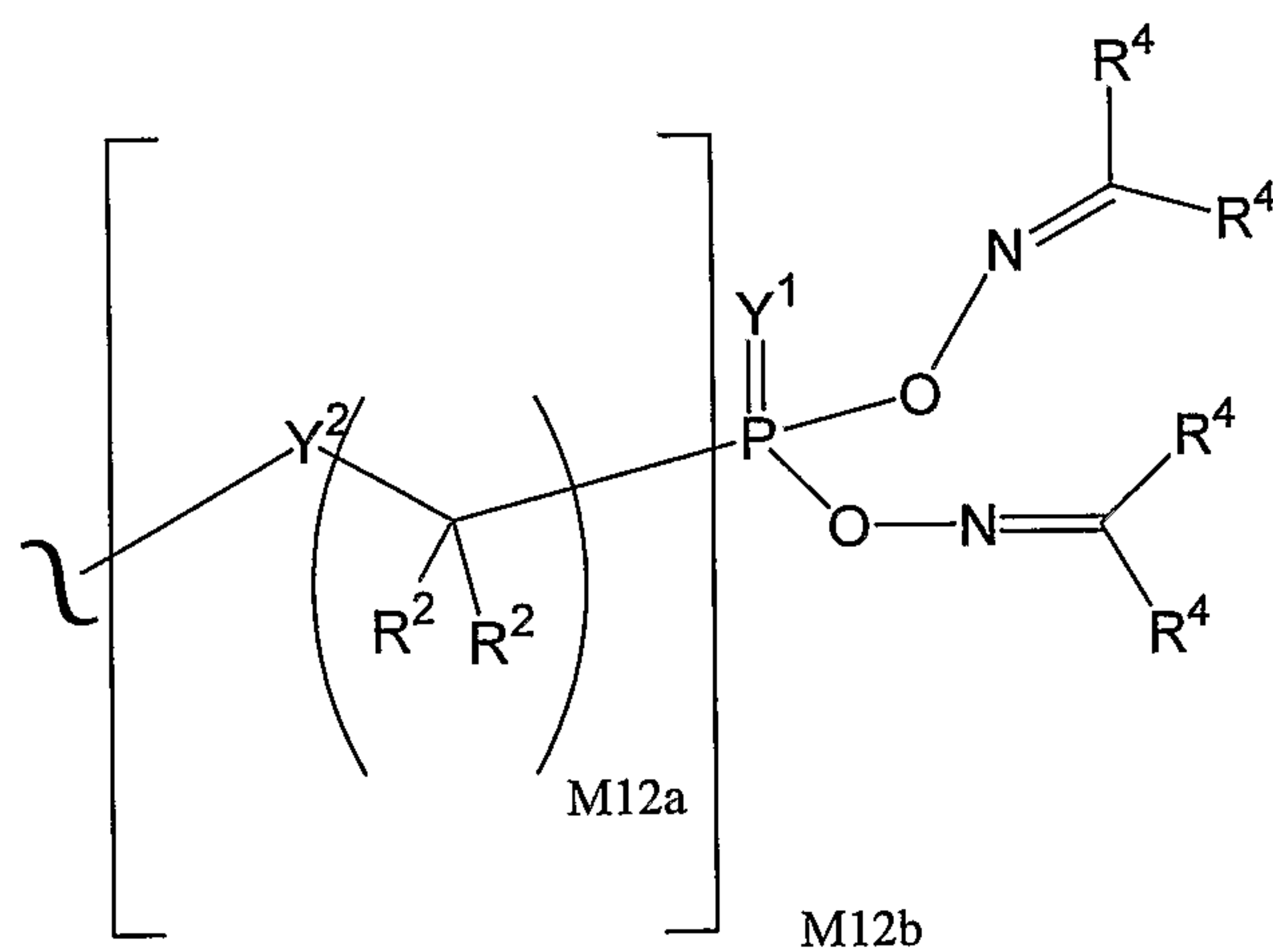


118. The compound of any one of claims 105-116 wherein A^3 is of the formula:



5 119. The compound of claim 117 or 118 wherein Y^2 is independently O, $N(R^x)$ or S.

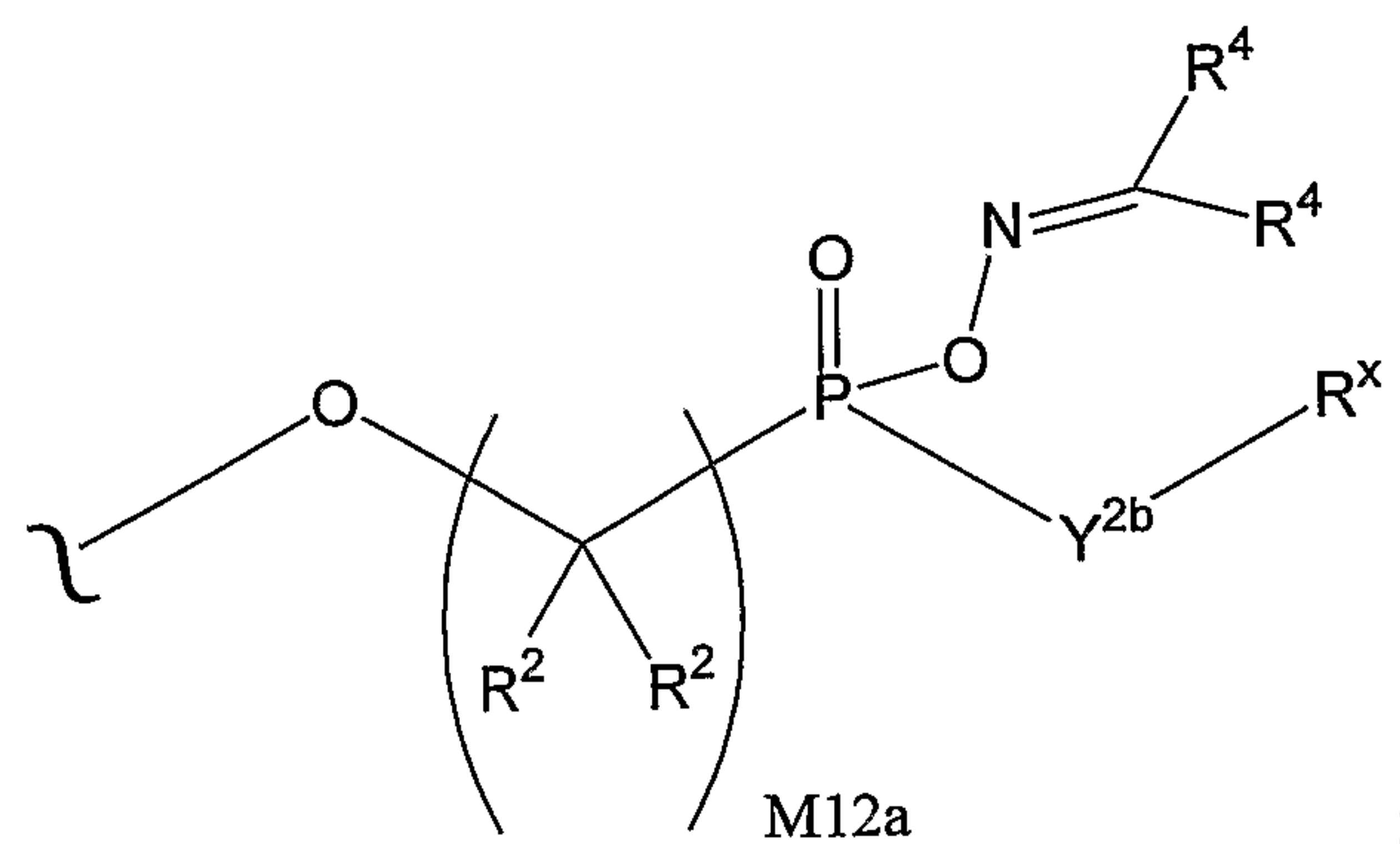
120. The compound of any one of claims 105-116 wherein A^3 is of the formula:



10

15

121. The compound of any one of claims 105-116 wherein A^3 is of the formula:



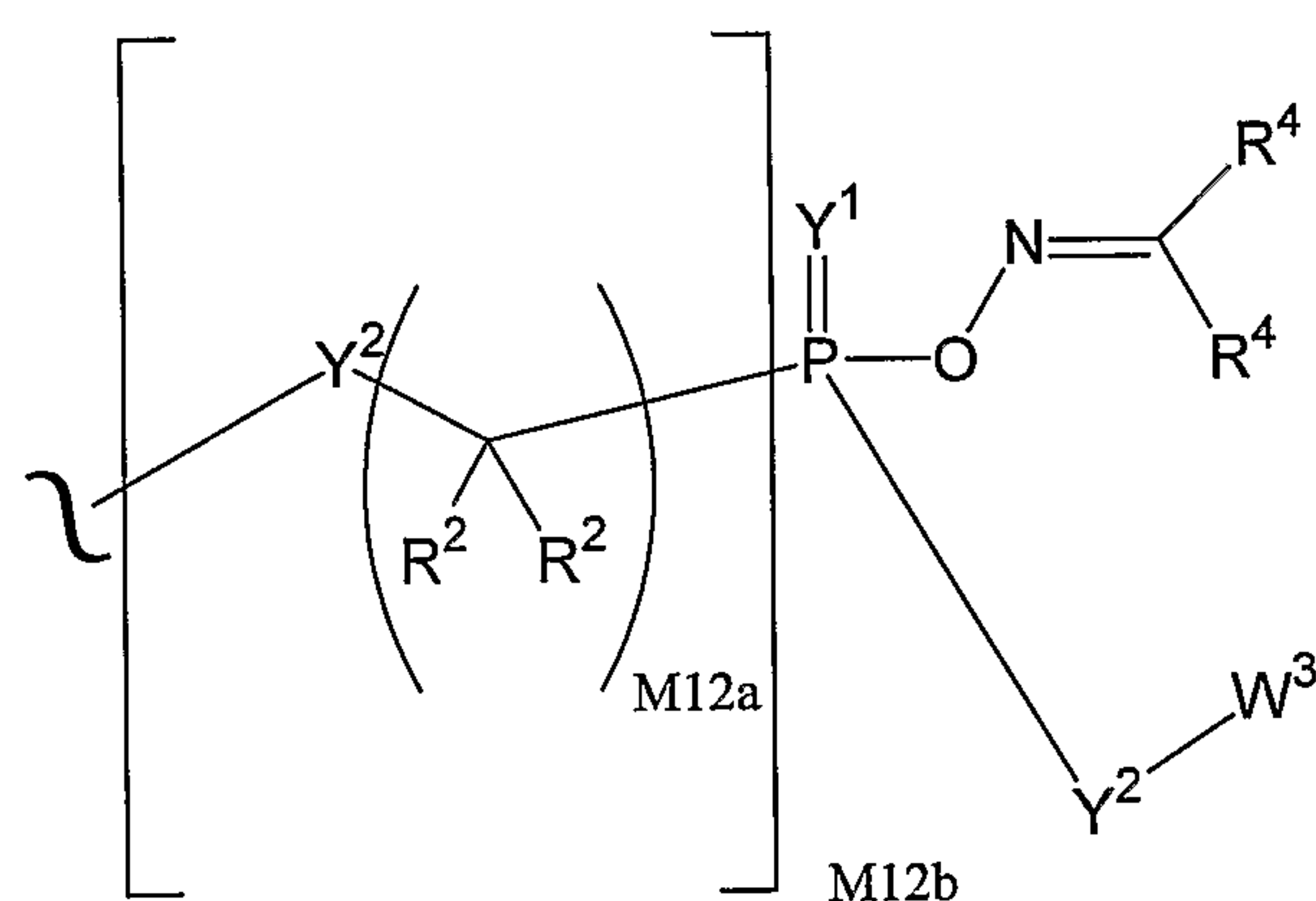
and Y^{2b} is O or $N(R^x)$.

5

122. The compound of claim 121 wherein M12a is 1, 2, 3, 4, 5, 6, 7 or 8.

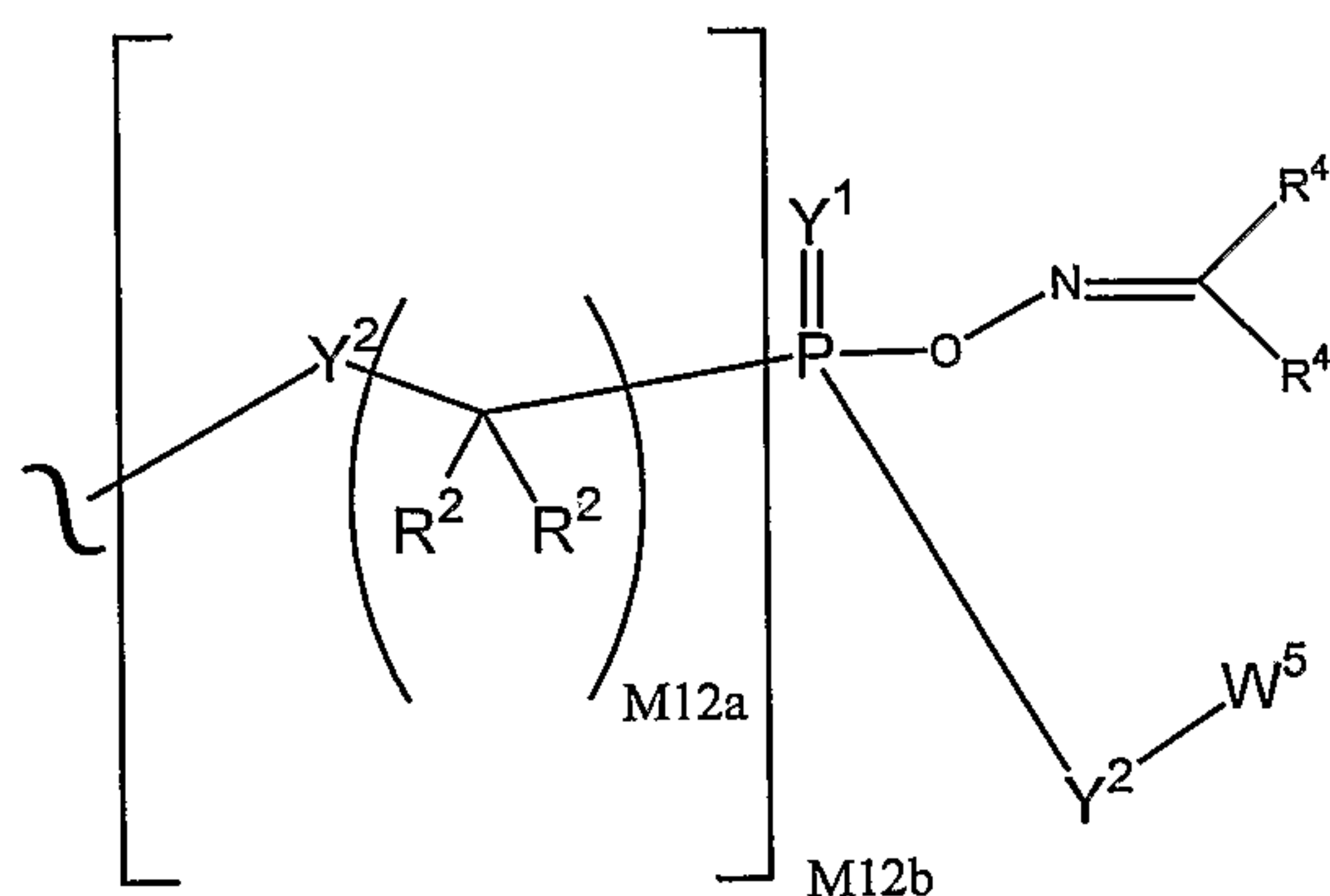
123. The compound of claim 122 wherein M12a is 1 and each R^2 is hydrogen.

10 124. The compound of any one of claims 105-116 wherein A^3 is of the formula:



15

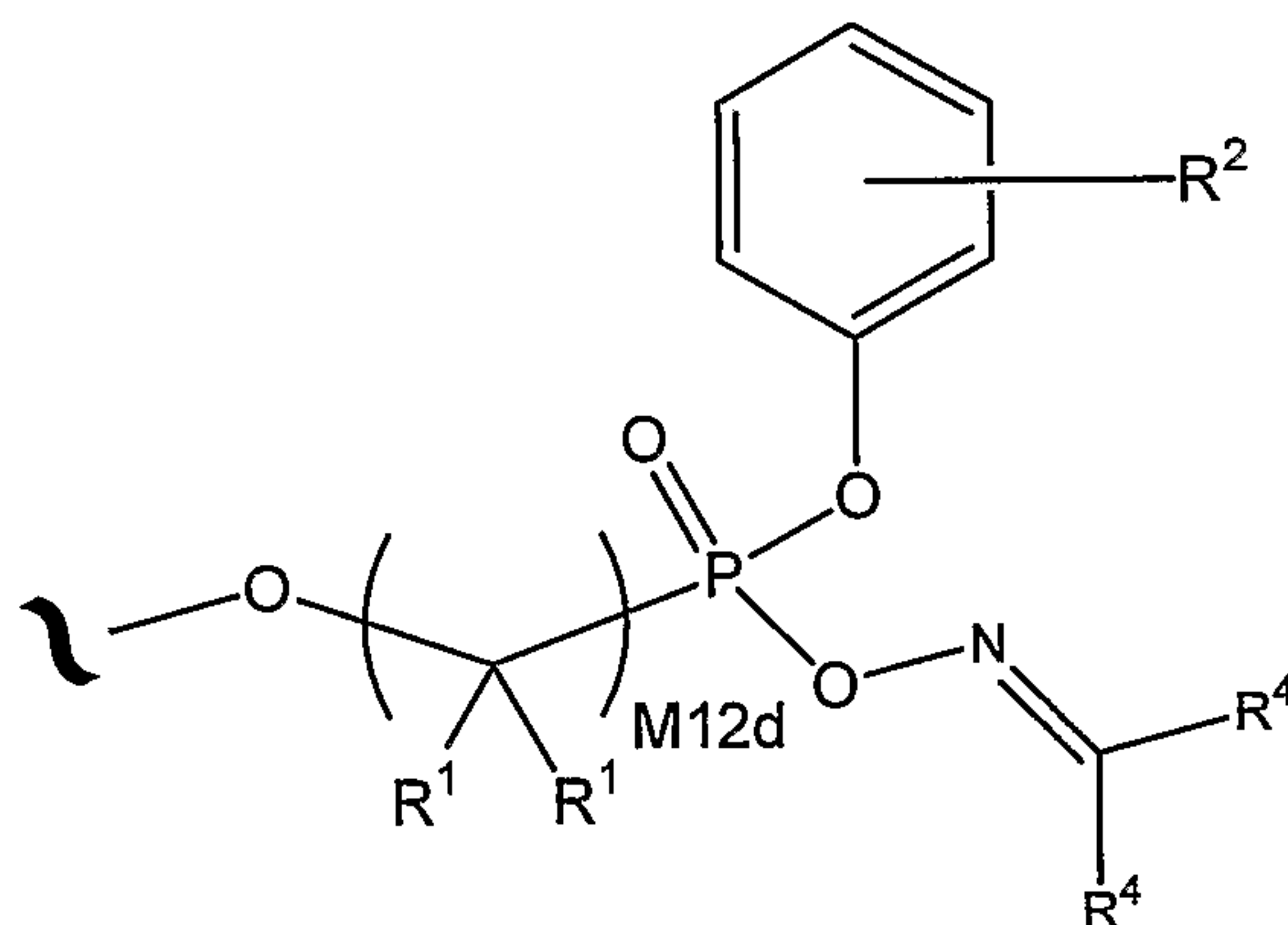
125. The compound of any one of claims 105-116 wherein A³ is of the formula:



5

126. The compound of claim 125 wherein W⁵ is a carbocycle.

127. The compound of any one of claims 105-116 wherein A³ is of the formula:



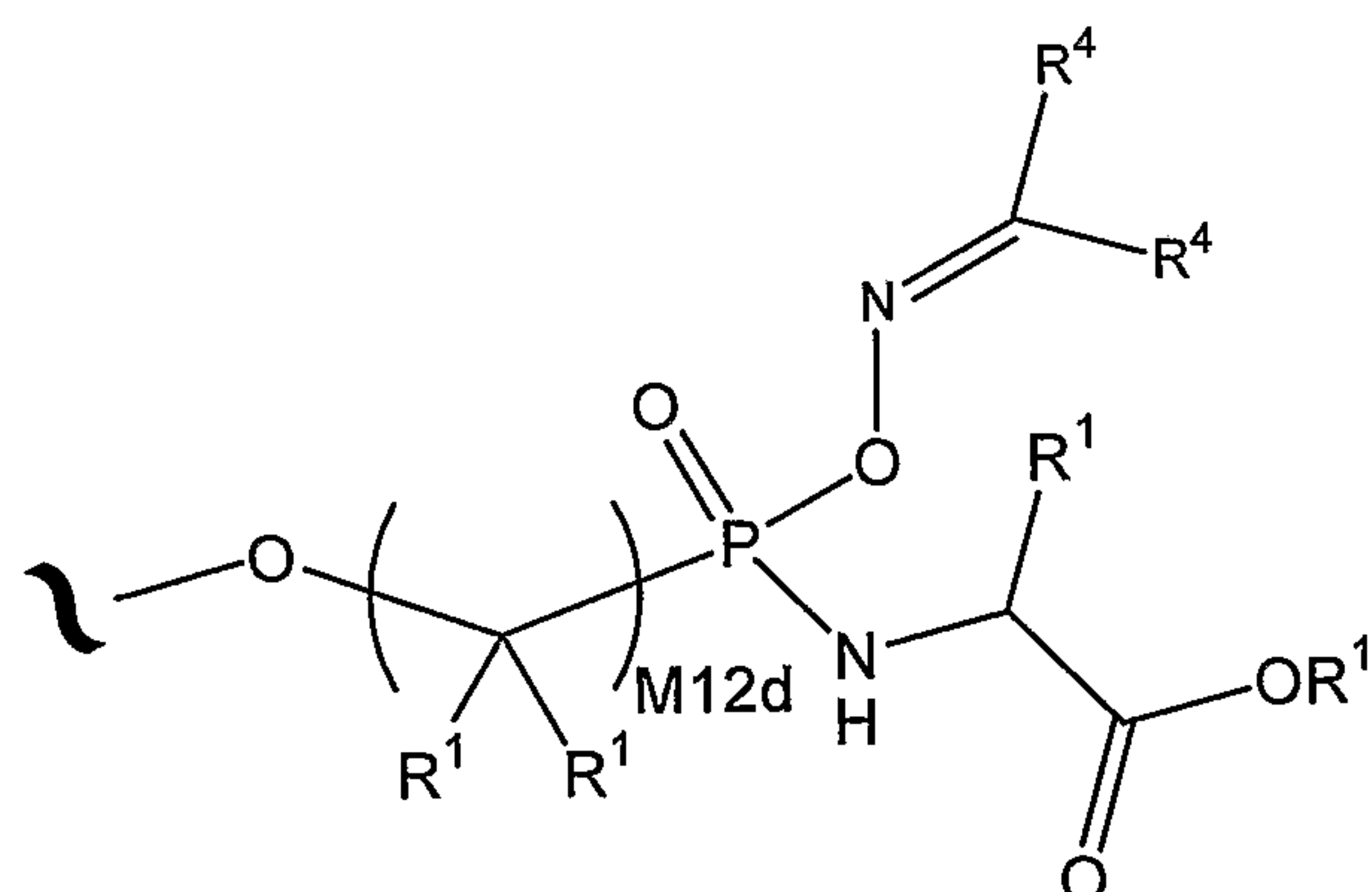
10

wherein R¹ is independently H or alkyl of 1 to 18 carbon atoms; and the phenyl carbocycle is substituted with 0, 1, 2, or 3 R² groups.

15

20

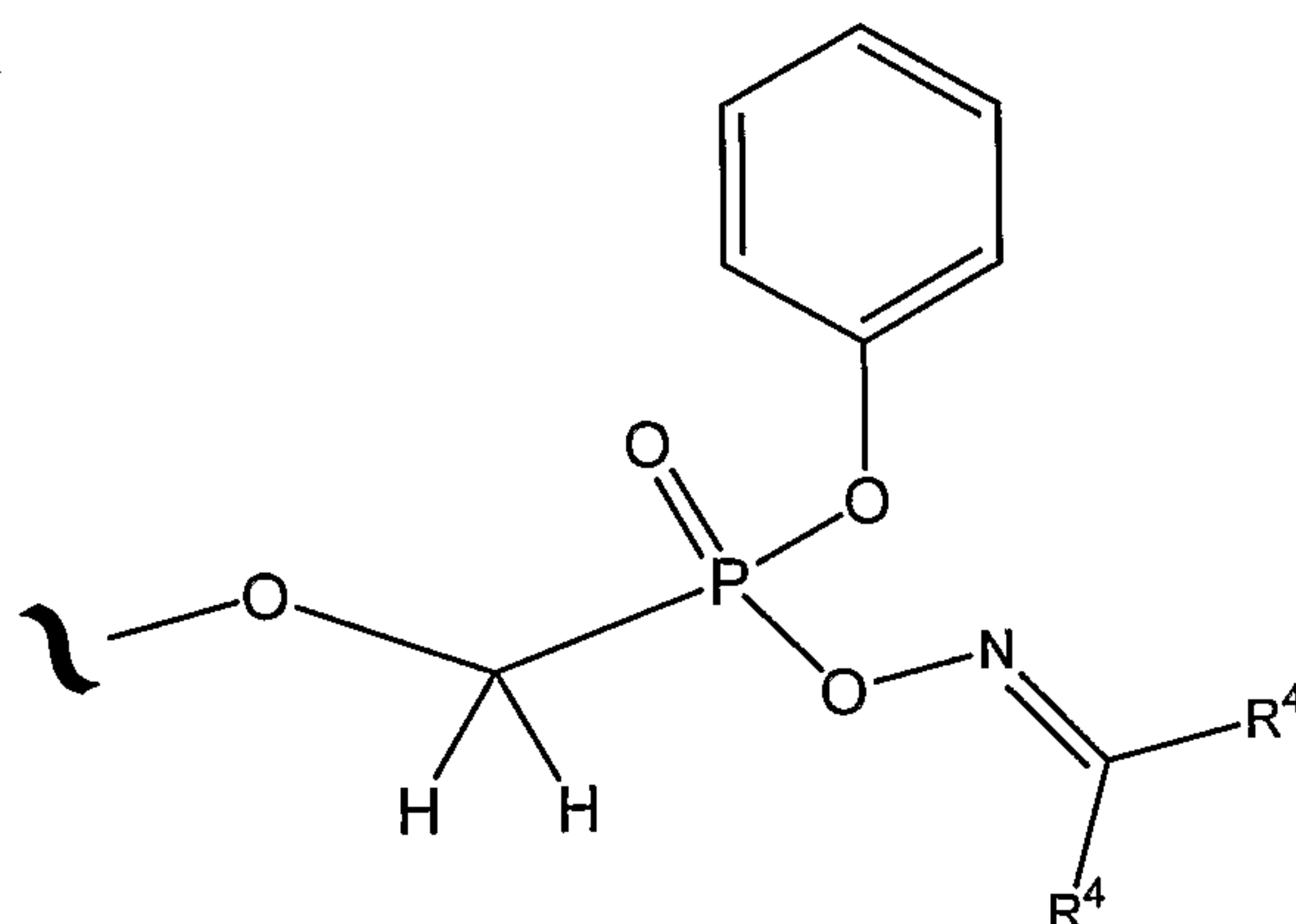
128. The compound of any one of claims 105-116 wherein A³ is of the formula:



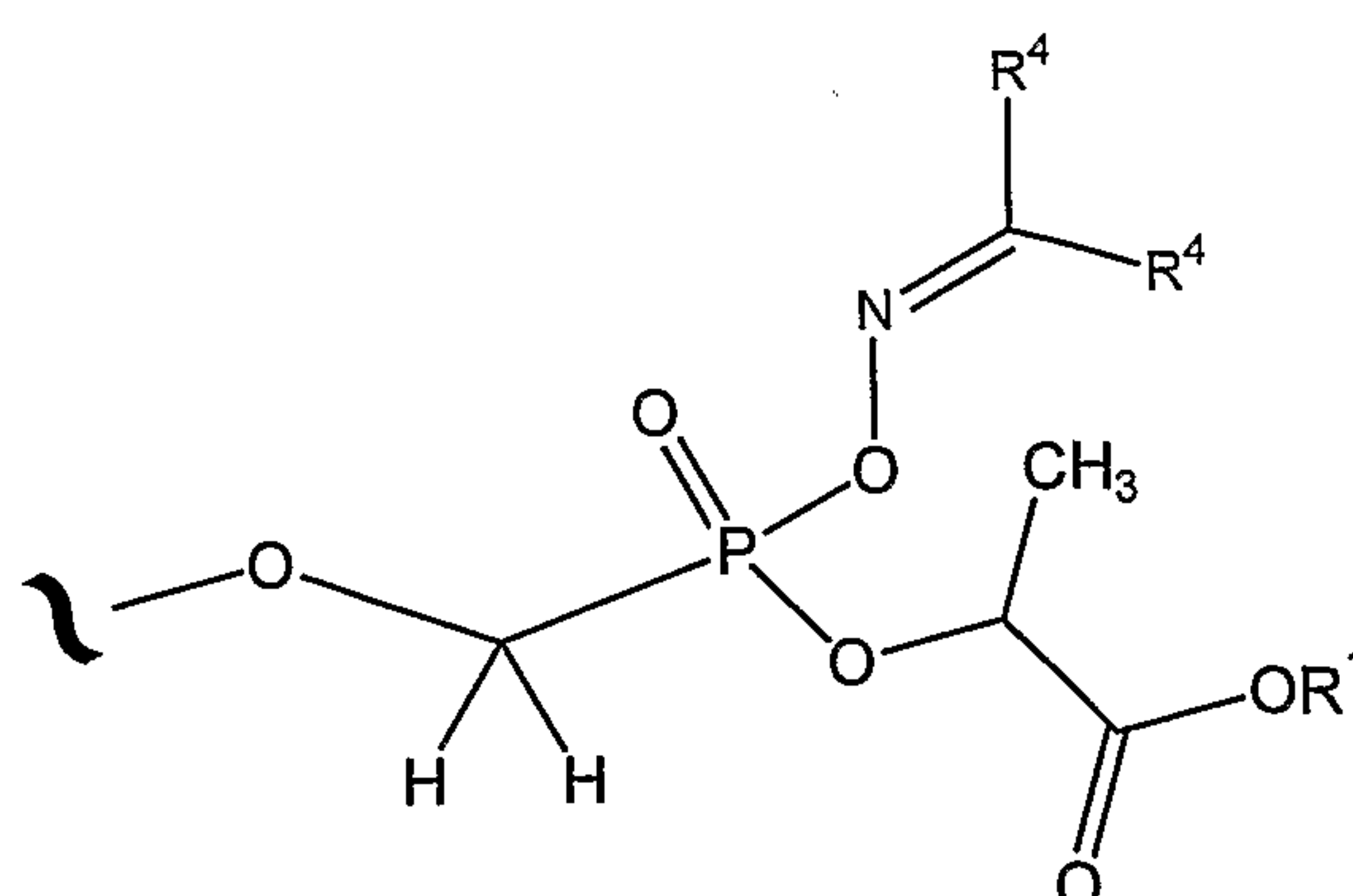
wherein R¹ is independently H or alkyl of 1 to 18 carbon atoms.

5

129. The compound of any one of claims 105-116 wherein A³ is of the formula:



10 130. The compound of any one of claims 105-116 wherein A³ is of the formula:



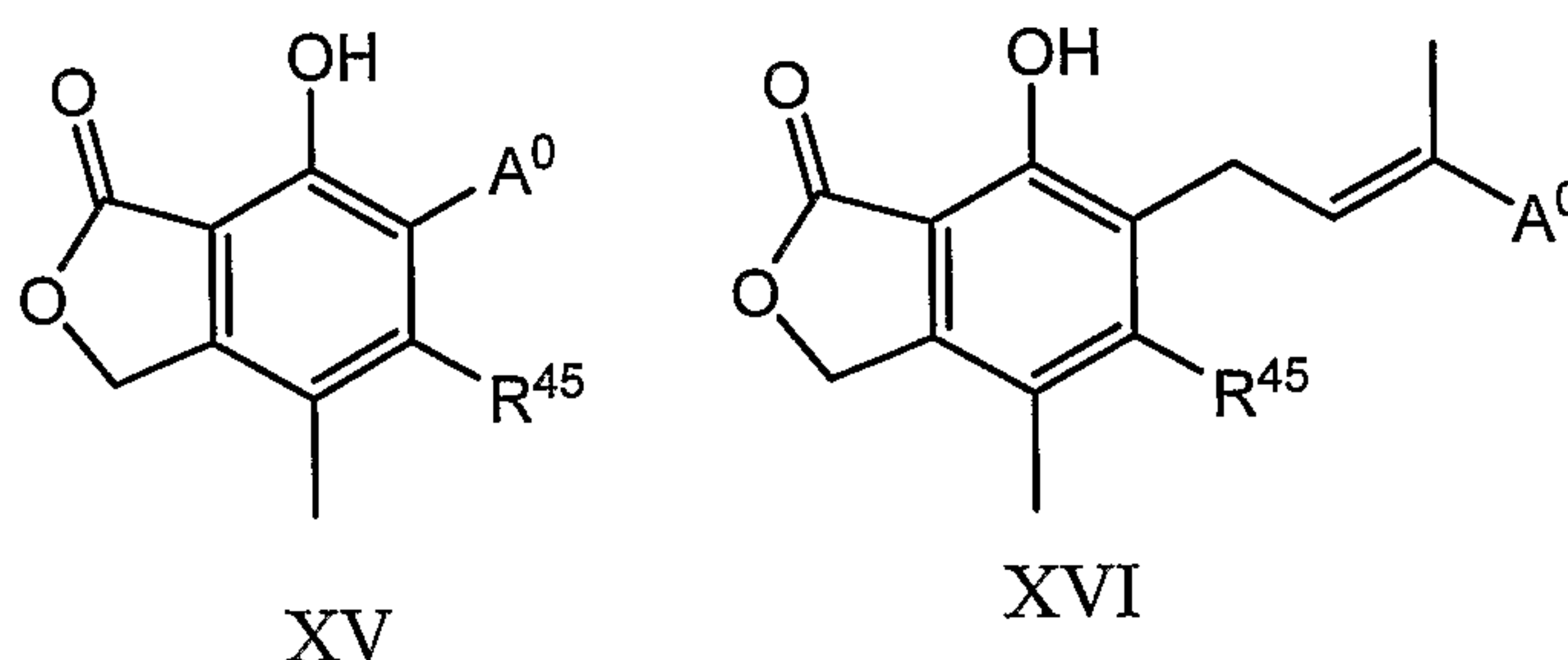
and R¹ is independently H or alkyl of 1 to 18 carbon atoms.

131. The compound of any one of claims 105-130 wherein R^{43} is ethyl or vinyl.

132. The compound of any one of claims 105-130 wherein R^{43} is methoxy.

5

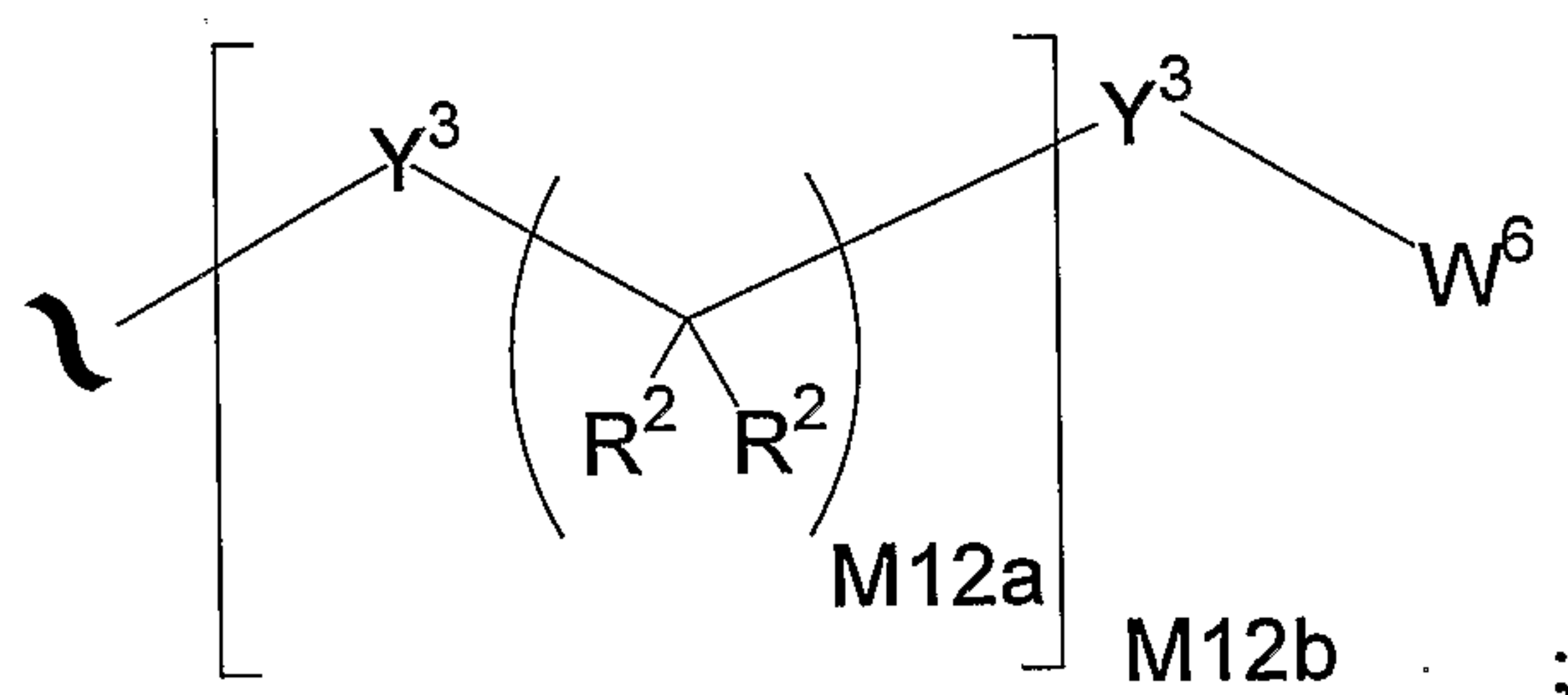
133. A compound of formula XV or XVI:



wherein:

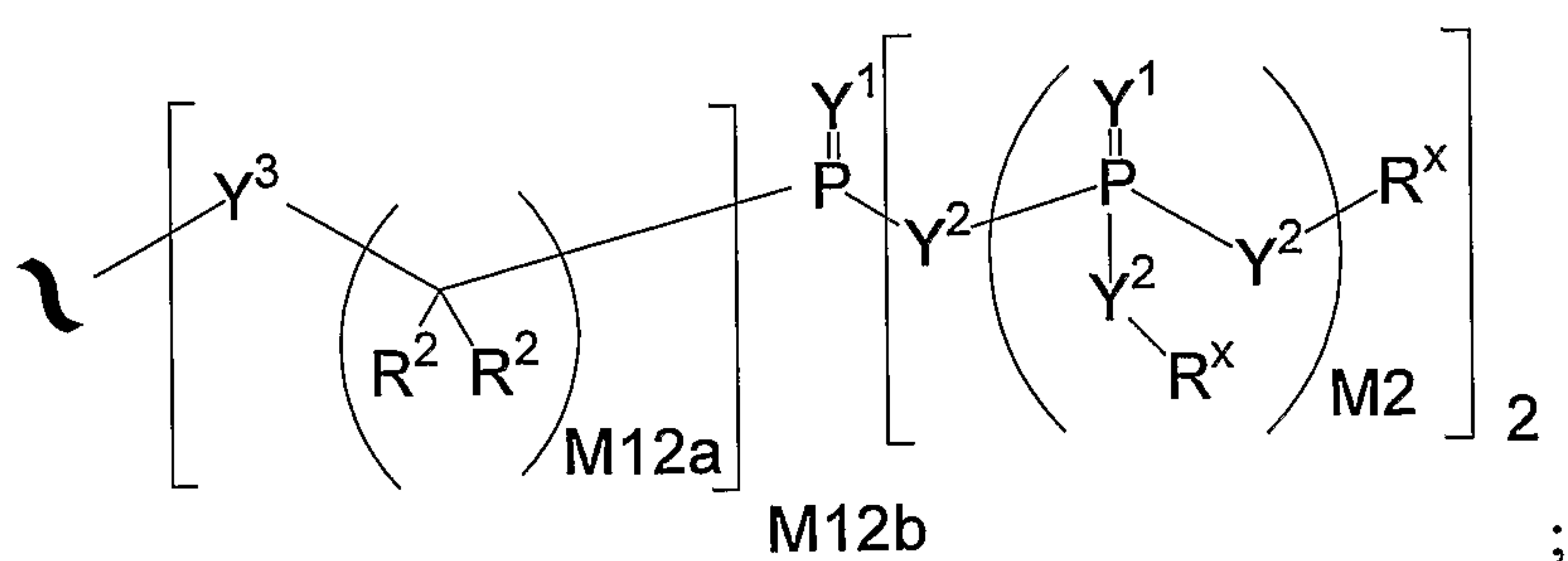
A^0 is A^1 ;

A^1 is:



10

A^3 is:



Y^1 is independently O, S, $N(R^x)$, $N(OR^x)$, or $N(N(R^x)(R^x))$;

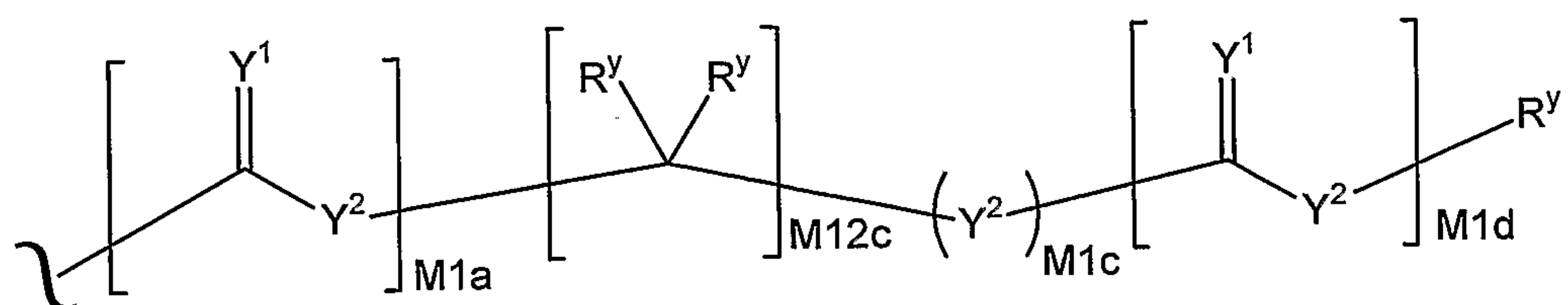
Y^2 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, or

15 $-S(O)_{M2}-$; and when Y^2 joins two phosphorous atoms Y^2 can also be $C(R^2)(R^2)$;

Y^3 is independently a bond, O, $N(R^x)$, $N(OR^x)$, $N(N(R^x)(R^x))$, $-S(O)_{M2}-$, or $-S(O)_{M2}-S(O)_{M2}-$, $-O(C=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)$, $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$; wherein A^1 comprises at least one

Y^3 that is $-O(C=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)-$,
 $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$;

R^x is independently H, R^2 , W^3 , a protecting group, or the formula:



5 R^y is independently H, W^3 , R^2 or a protecting group;

R^2 is independently H, R^3 or R^4 wherein each R^4 is independently
substituted with 0 to 3 R^3 groups;

R^3 is R^{3a} , R^{3b} , R^{3c} or R^{3d} , provided that when R^3 is bound to a
heteroatom, then R^3 is R^{3c} or R^{3d} ;

10 R^{3a} is F, Cl, Br, I, $-CN$, N_3 or $-NO_2$;

R^{3b} is Y^1 ;

R^{3c} is $-R^x$, $-N(R^x)(R^x)$, $-SR^x$, $-S(O)R^x$, $-S(O)_2R^x$, $-S(O)(OR^x)$,
 $-S(O)_2(OR^x)$, $-OC(Y^1)R^x$, $-OC(Y^1)OR^x$, $-OC(Y^1)(N(R^x)(R^x))$, $-SC(Y^1)R^x$,
 $-SC(Y^1)OR^x$, $-SC(Y^1)(N(R^x)(R^x))$, $-N(R^x)C(Y^1)R^x$, $-N(R^x)C(Y^1)OR^x$, or

15 $-N(R^x)C(Y^1)(N(R^x)(R^x))$;

R^{3d} is $-C(Y^1)R^x$, $-C(Y^1)OR^x$ or $-C(Y^1)(N(R^x)(R^x))$;

R^4 is an alkyl of 1 to 18 carbon atoms, alkenyl of 2 to 18 carbon atoms,
or alkynyl of 2 to 18 carbon atoms;

R^5 is R^4 wherein each R^4 is substituted with 0 to 3 R^3 groups;

20 W^3 is W^4 or W^5 ;

W^4 is R^5 , $-C(Y^1)R^5$, $-C(Y^1)W^5$, $-SO_2R^5$, or $-SO_2W^5$;

W^5 is carbocycle or heterocycle wherein W^5 is independently substituted
with 0 to 3 R^2 groups;

W^6 is W^3 independently substituted with 1, 2, or 3 A^3 groups;

25 $M2$ is 0, 1 or 2;

$M12a$ is 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

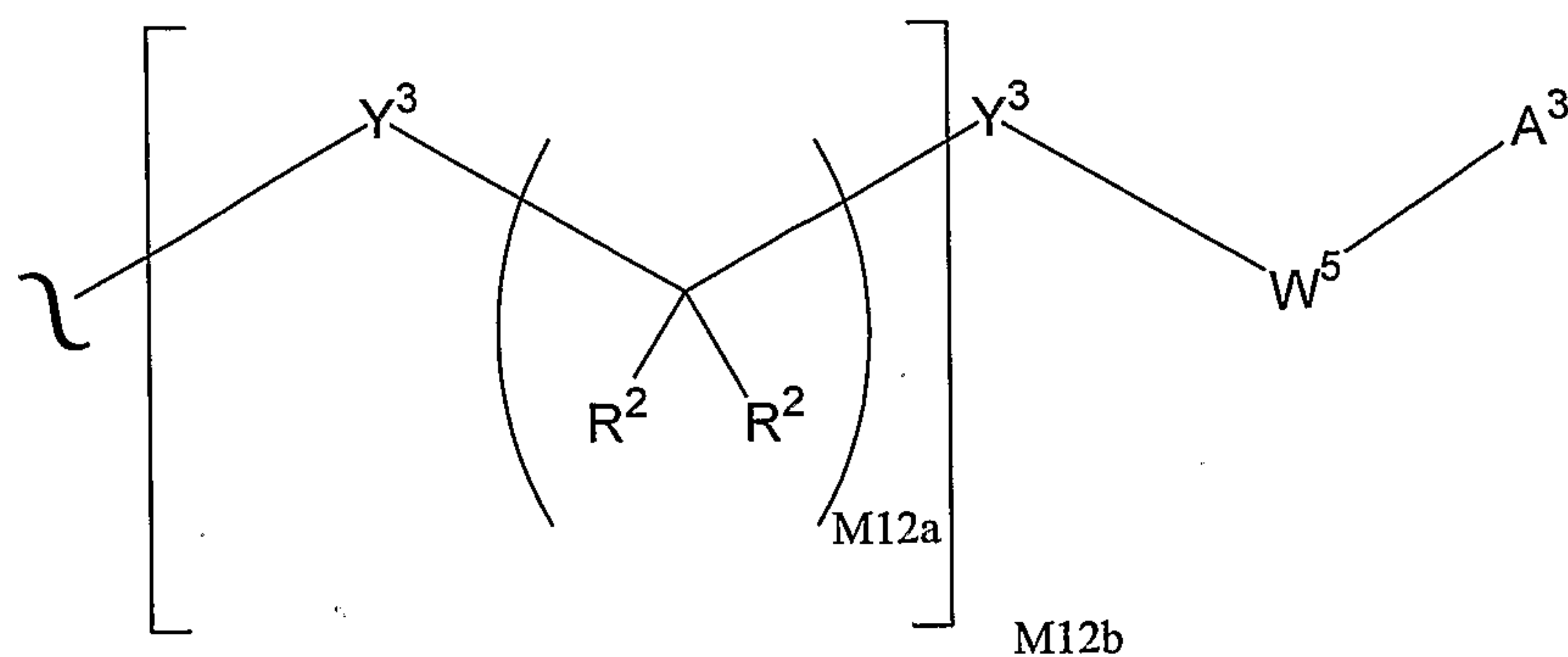
$M12b$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12;

$M1a$, $M1c$, and $M1d$ are independently 0 or 1;

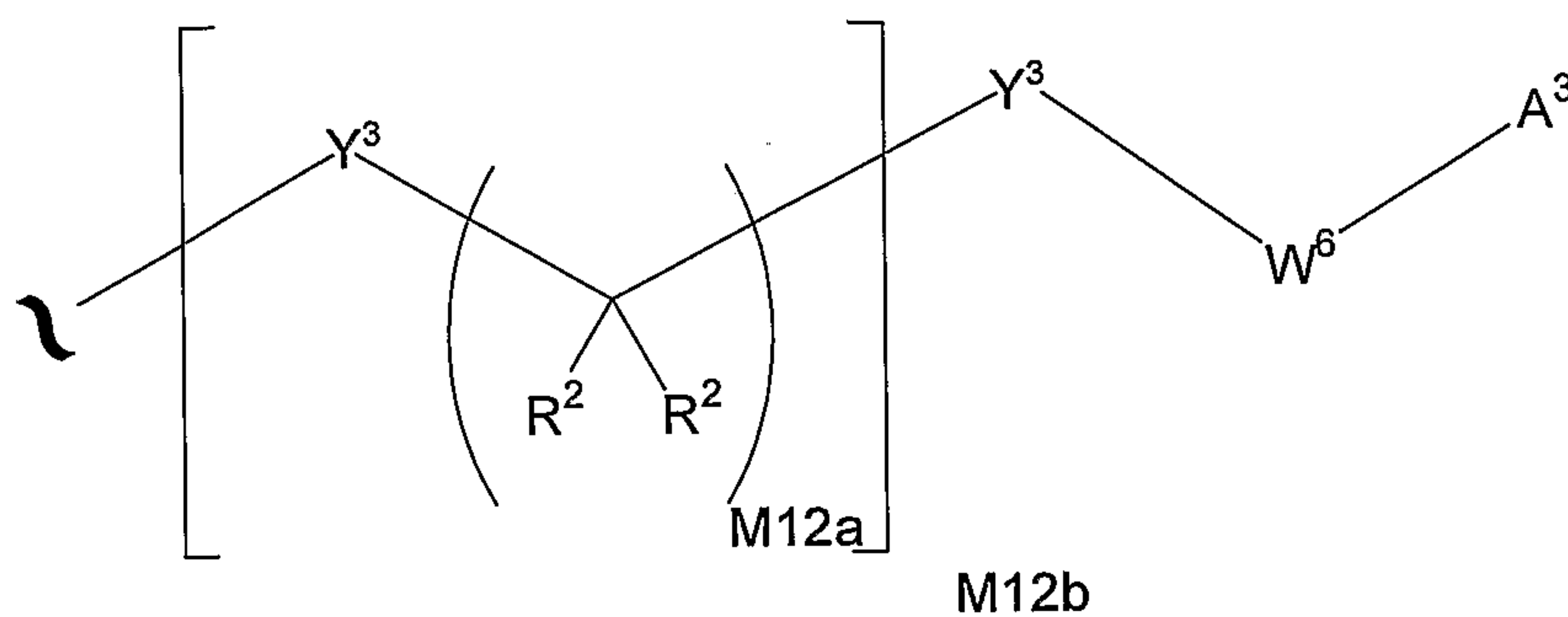
$M12c$ is 0, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11 or 12; and

30 R^{45} is ethyl, methoxy, or vinyl.

134. The compound of claim 133 wherein A^1 is of the formula:

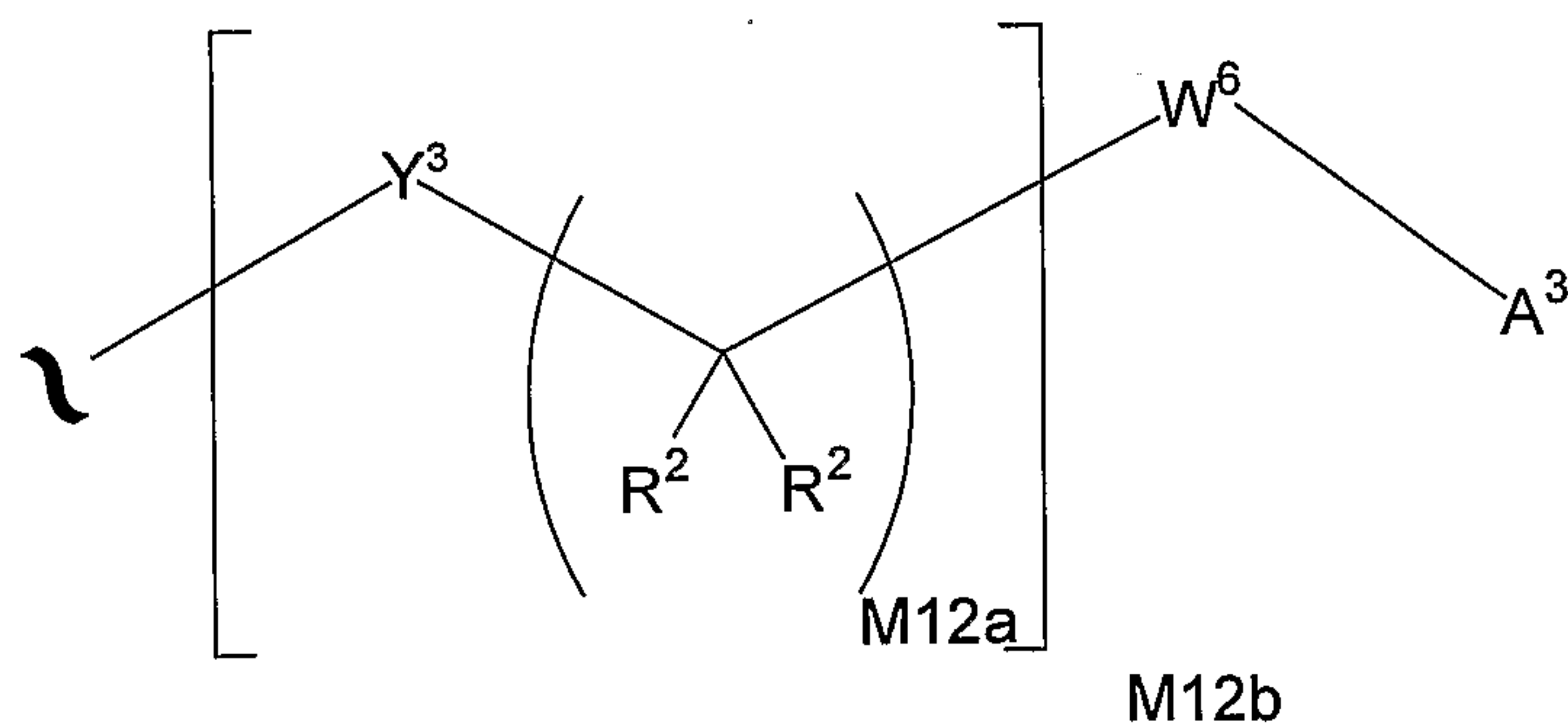


135. The compound of claim 133 wherein A^1 is of the formula:

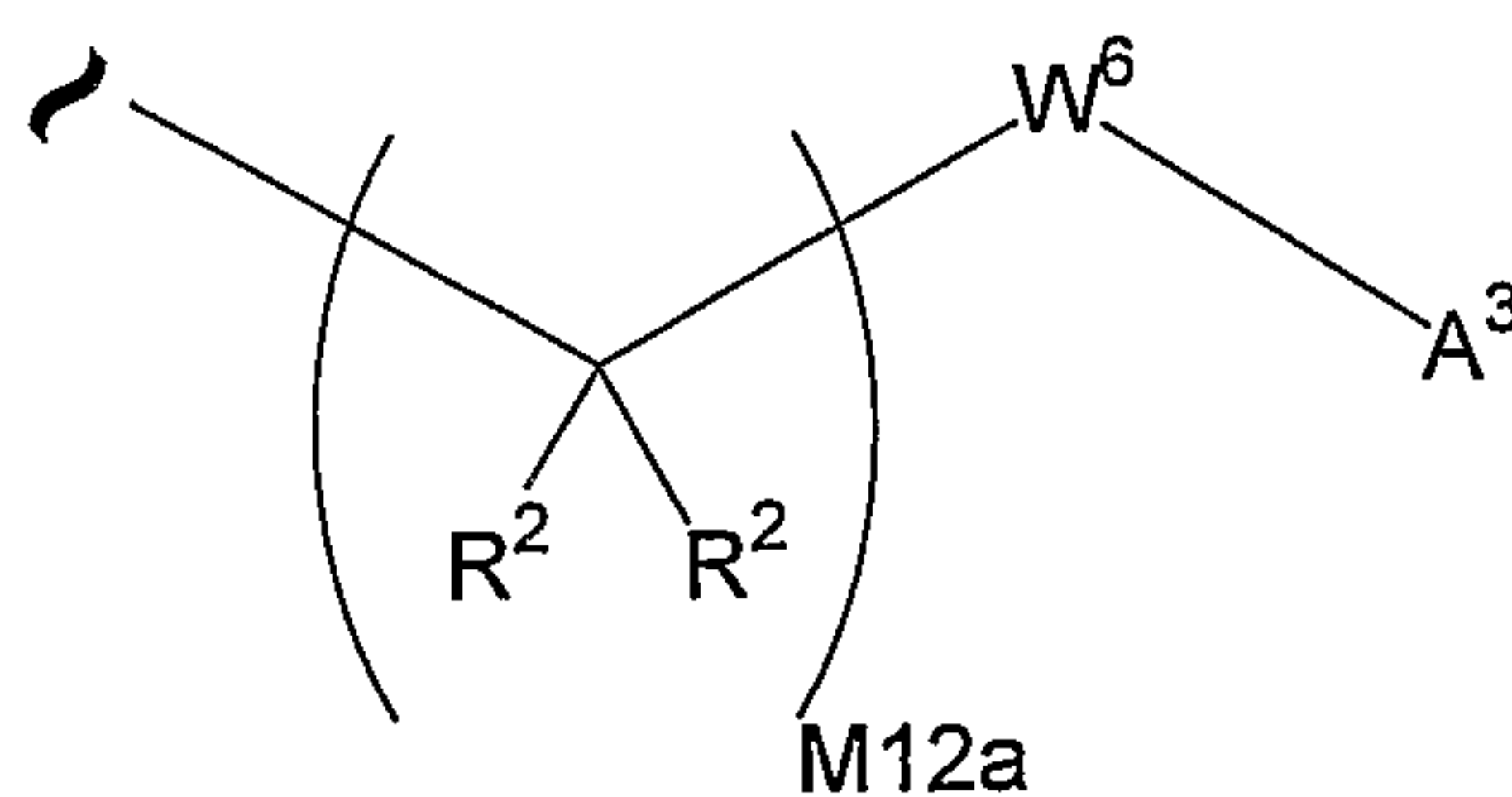


5

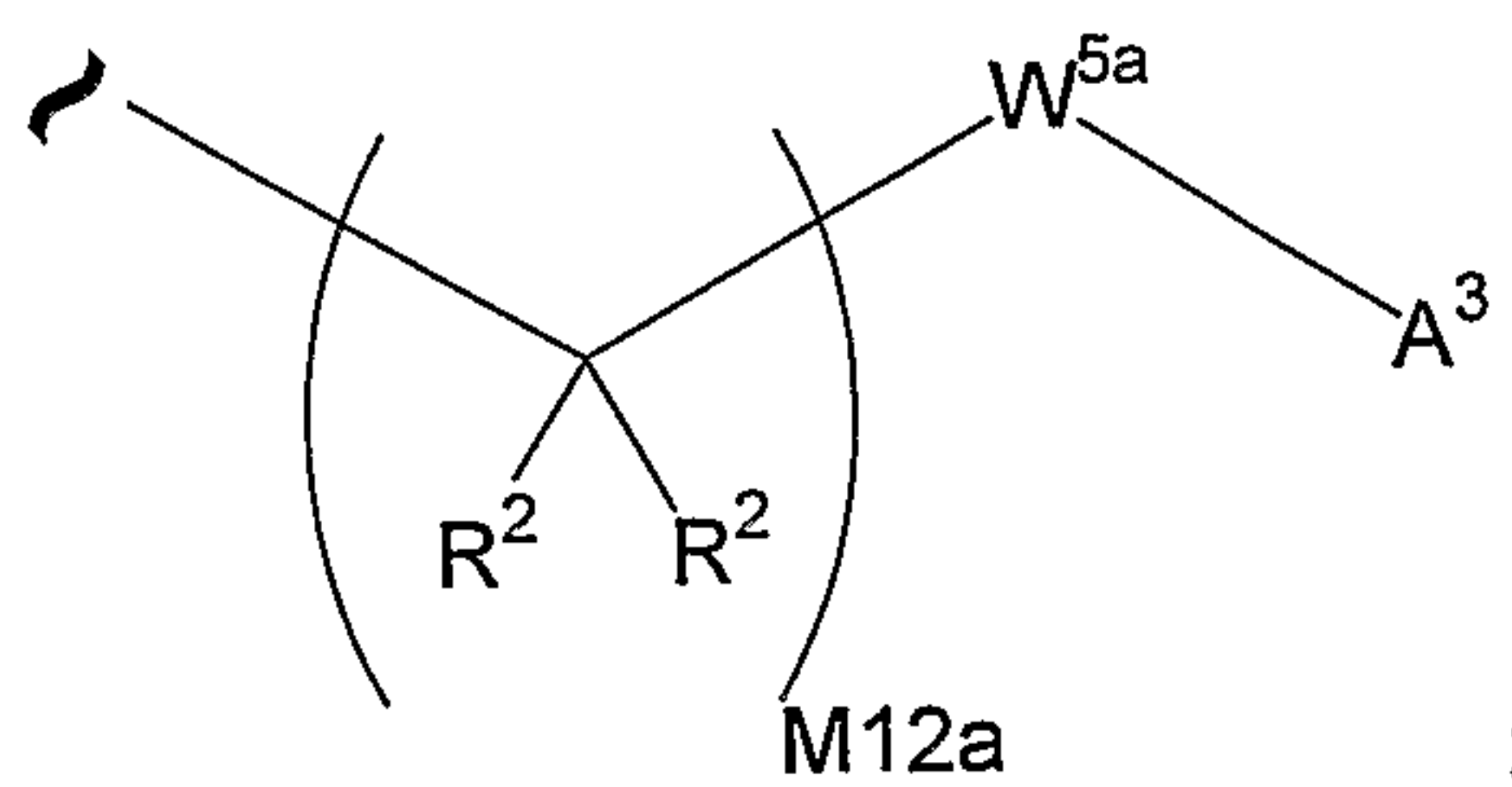
136. The compound of claim 133 wherein A^1 is of the formula:



10 137. The compound of claim 133 wherein A^1 is of the formula:



138. The compound of claim 133 wherein A^1 is of the formula:

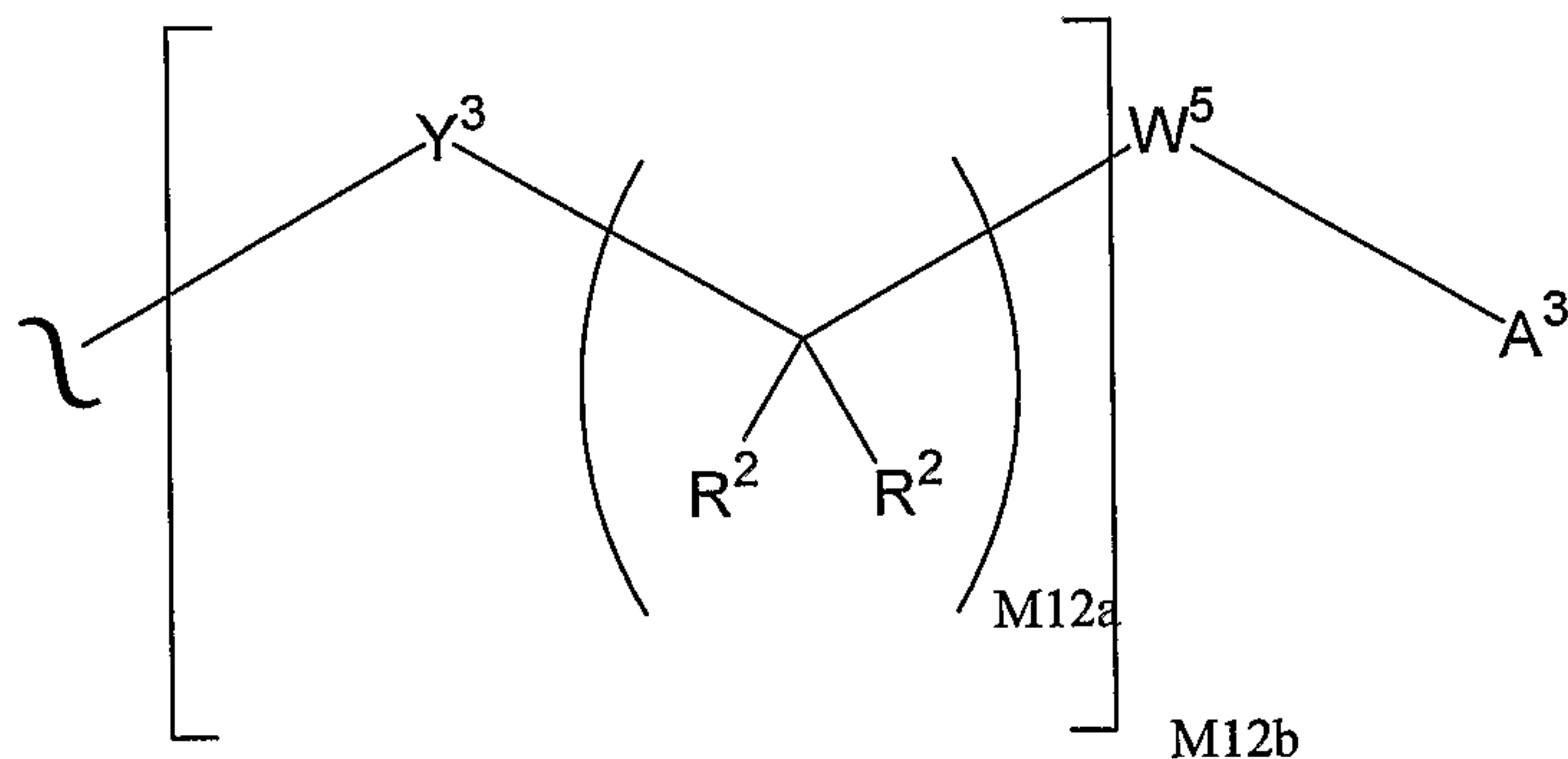


wherein W^{5a} is a carbocycle or a heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

5

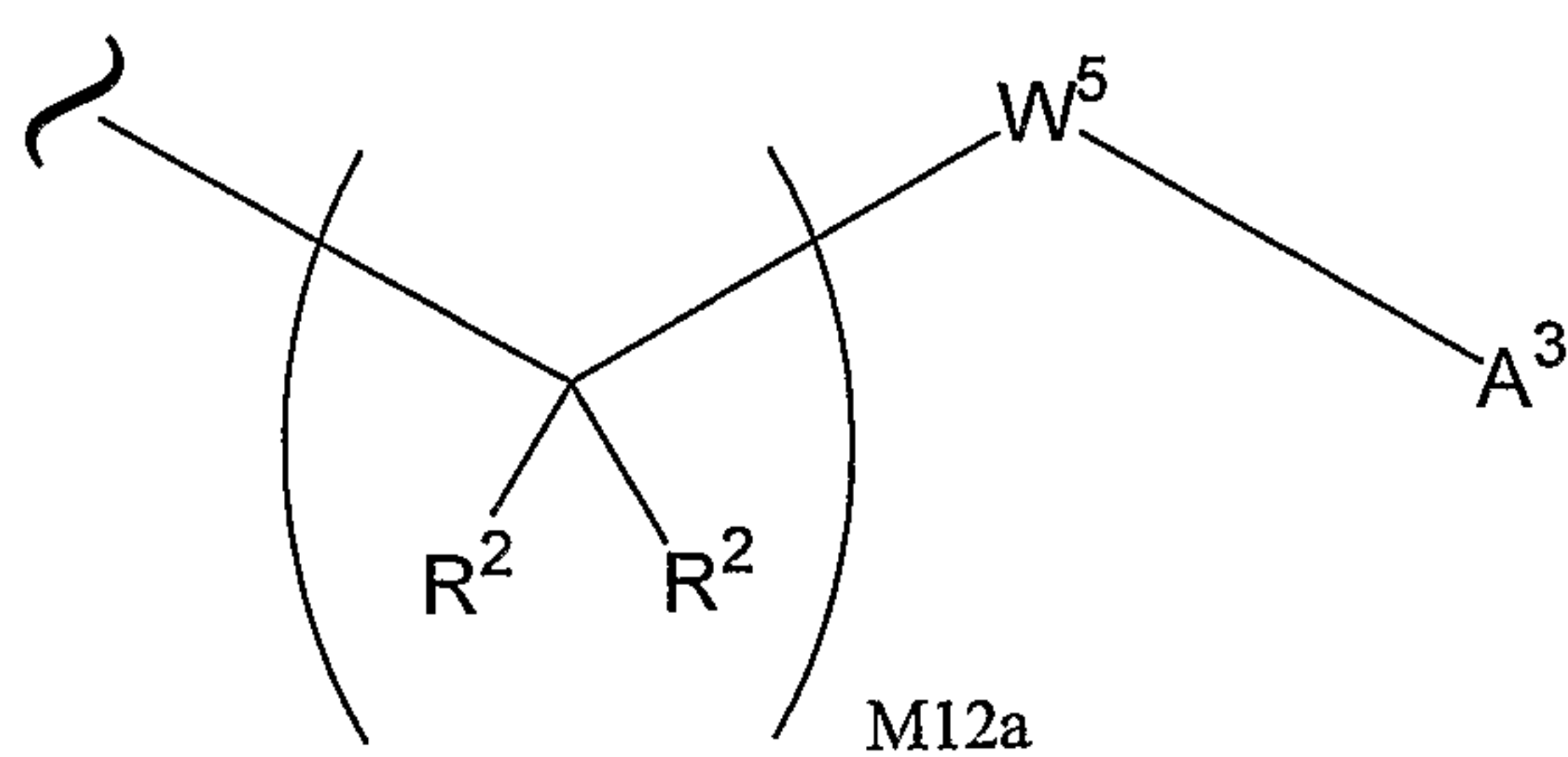
139. The compound of claim 133 wherein M12a is 1.

140. The compound of claim 133 wherein A^1 is of the formula:

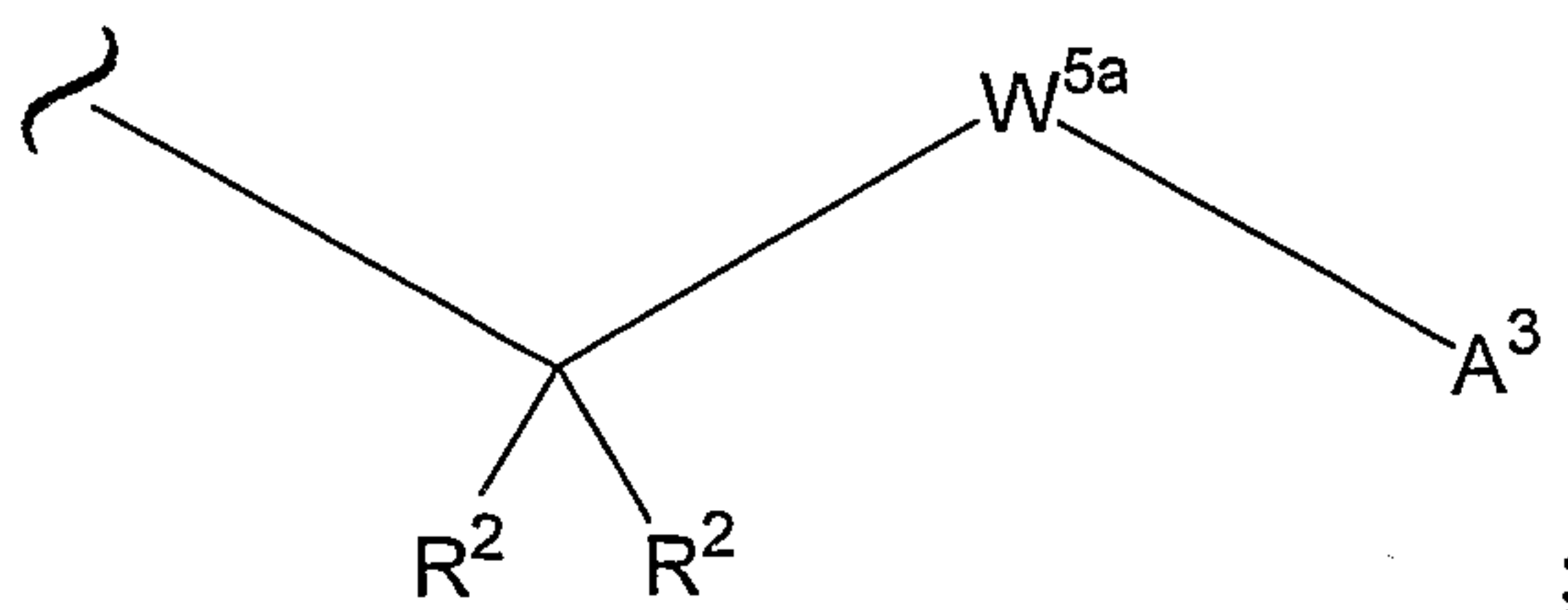


10

141. The compound of claim 133 wherein A^1 is of the formula:



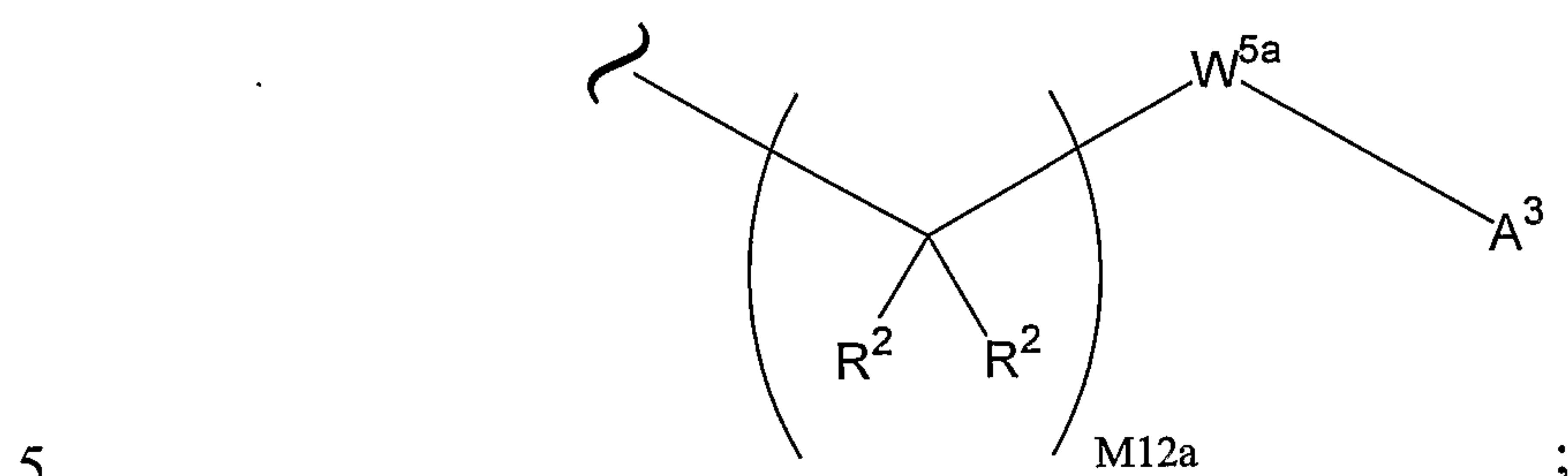
142. The compound of claim 133 wherein A^1 is of the formula:



15

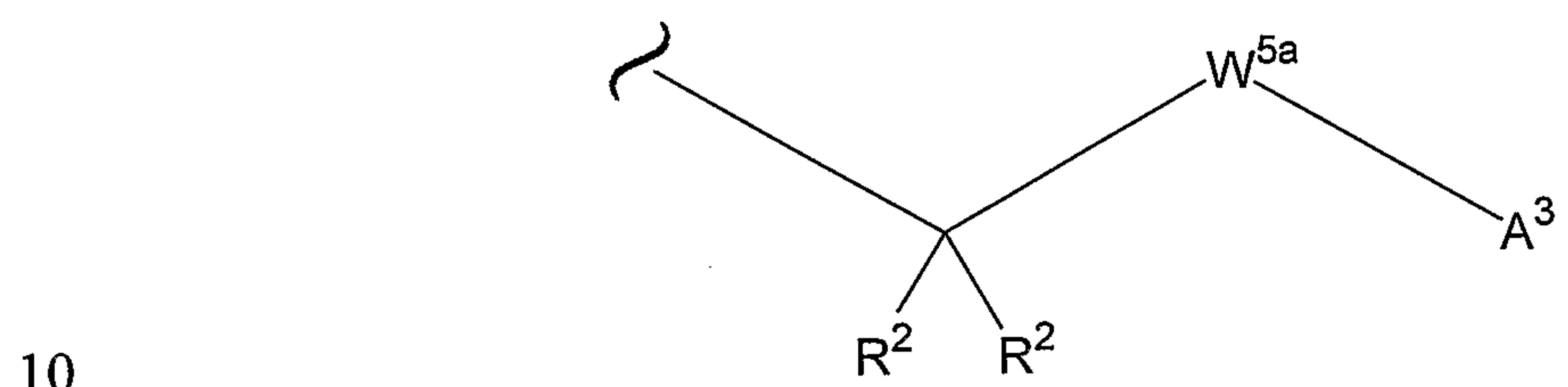
wherein W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

143. The compound of claim 133 wherein A^1 is of the formula:



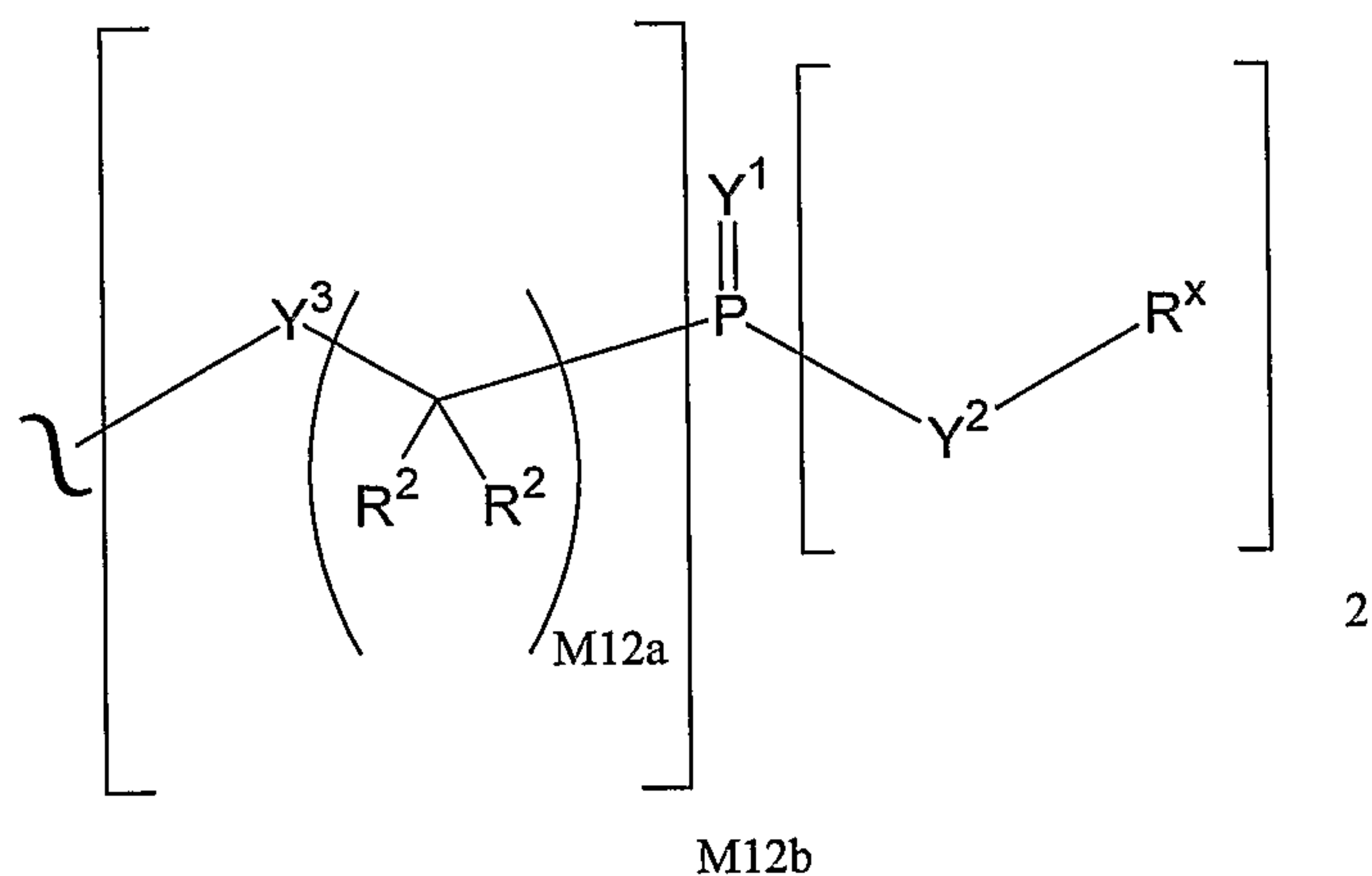
wherein W^{5a} is a carbocycle independently substituted with 0 or 1 R^2 groups.

144. The compound of claim 133 wherein A^1 is of the formula:

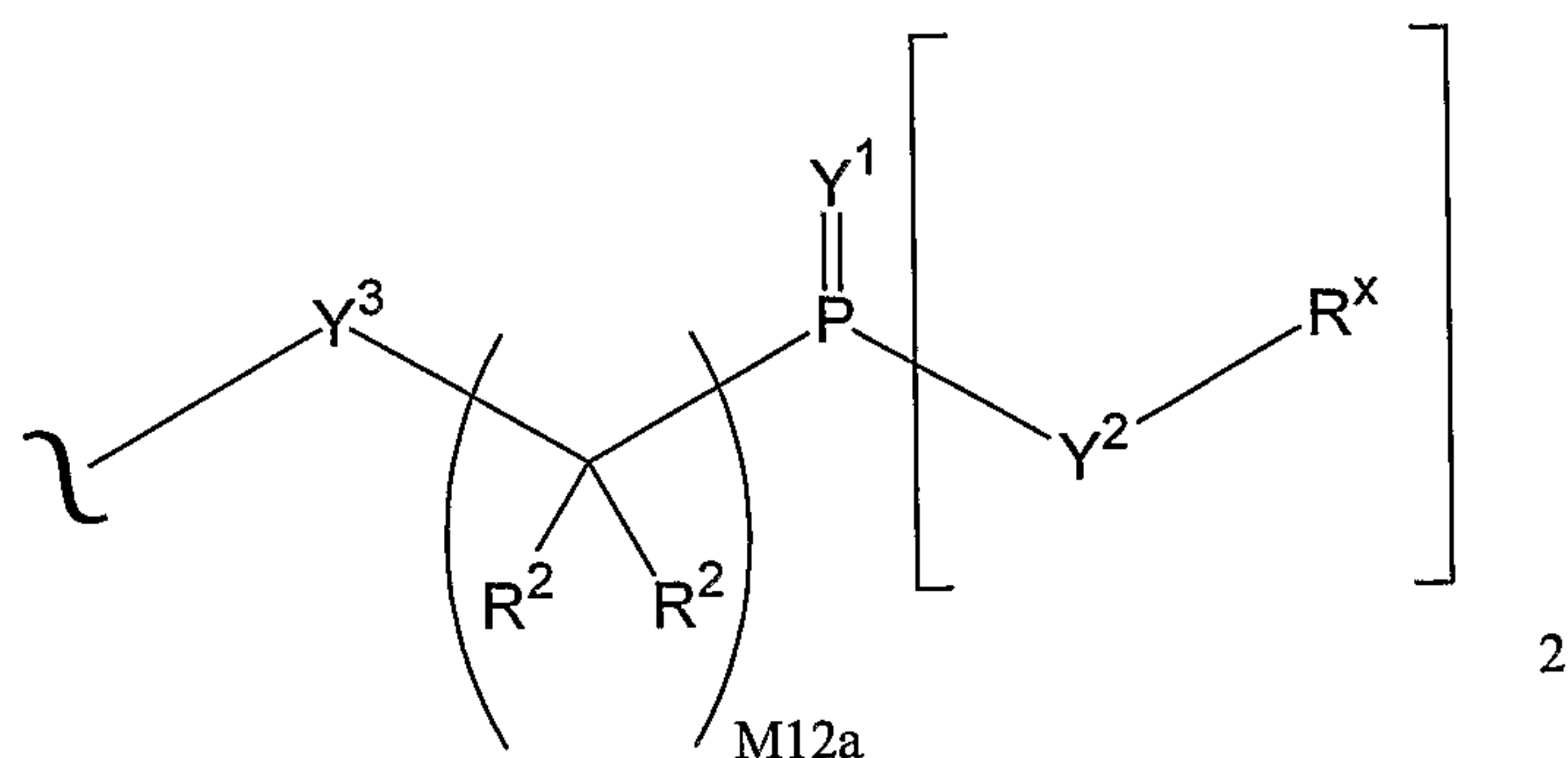


wherein W^{5a} is a carbocycle or heterocycle where W^{5a} is independently substituted with 0 or 1 R^2 groups.

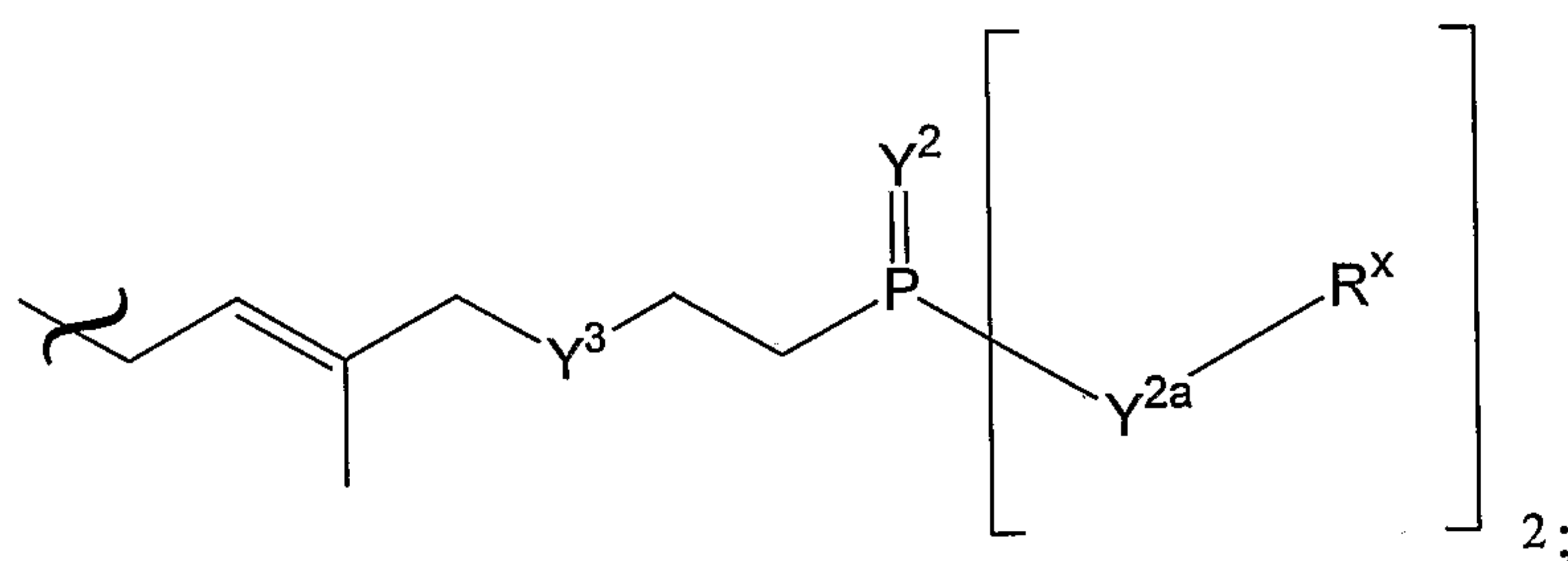
15 145. The compound of any one of claims 133-144 wherein A^3 is of the formula:



146. The compound of any one of claims 133-144 wherein A^3 is of the formula:



147. The compound of claim 133 wherein A^0 is of the formula:



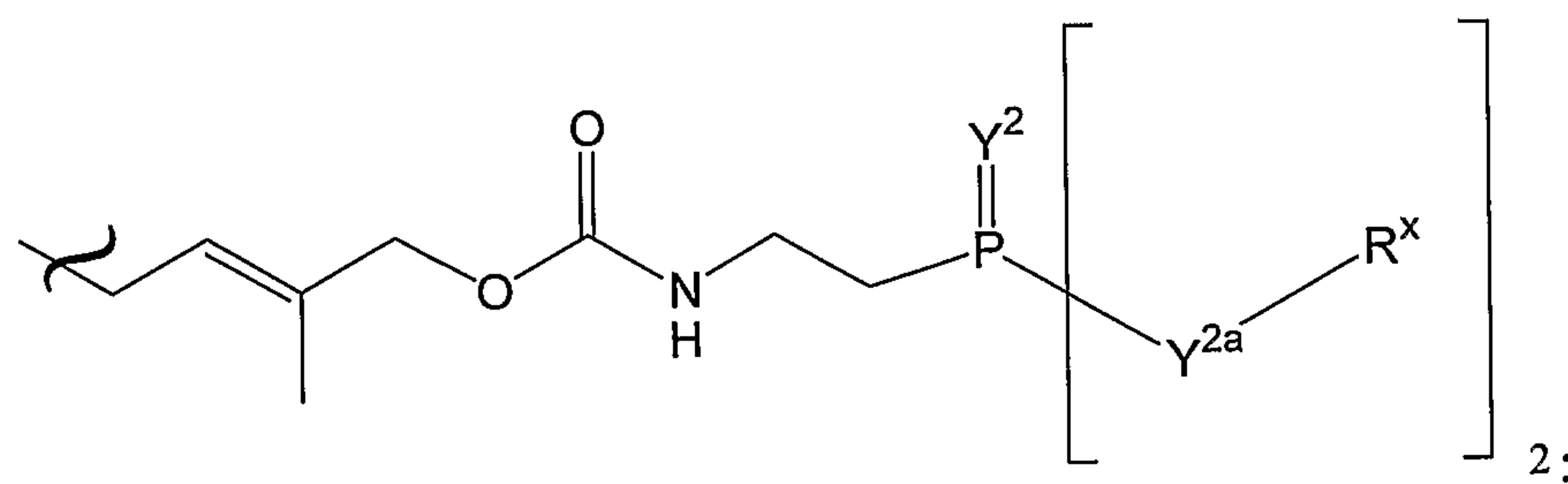
wherein:

Y^2 is O or S;

Y^{2a} is O, $N(R^x)$ or S; and

Y^3 is $-OC(=O)O-$, $-N(R^x)-C(=O)-O-$, $-N(R^x)-S(O)_2-N(R^x)-$, $-O-C(=O)-N(R^x)-$, or $-N(R^x)-C(=O)-N(R^x)-$.

148. The compound of claim 133 wherein A^0 is of the formula:

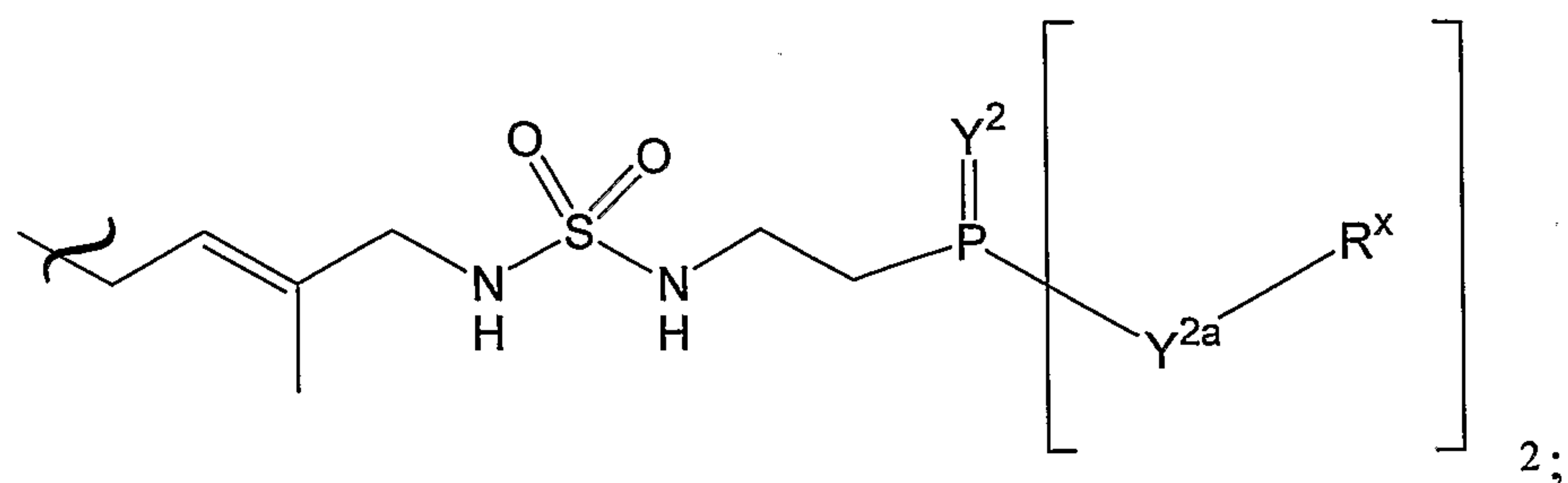


wherein:

Y^2 is O or S; and

Y^{2a} is O, $N(R^x)$ or S.

149. The compound of claim 133 wherein A^0 is of the formula:

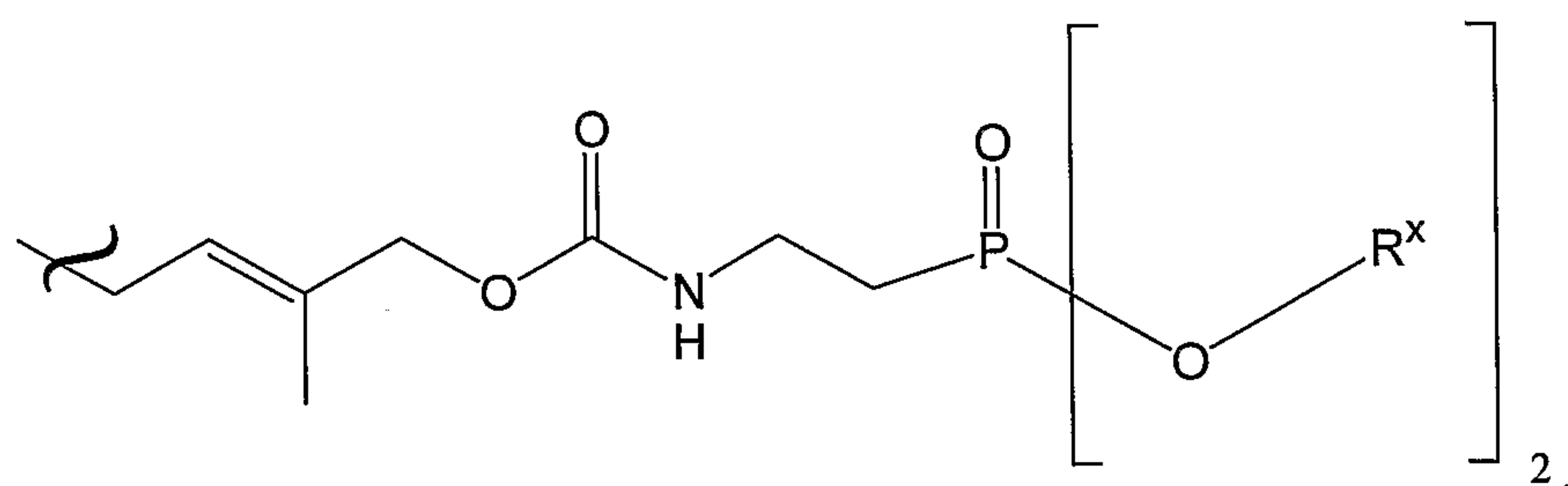


wherein:

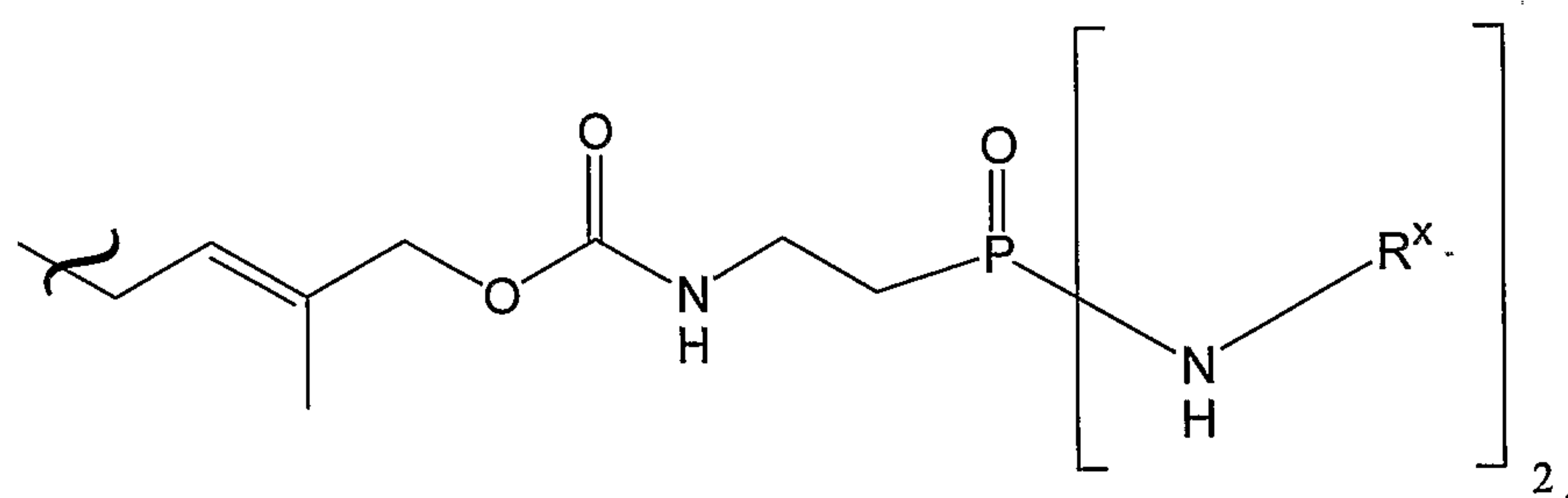
Y^2 is O or S; and

5 Y^{2a} is O, $\text{N}(\text{R}^x)$ or S.

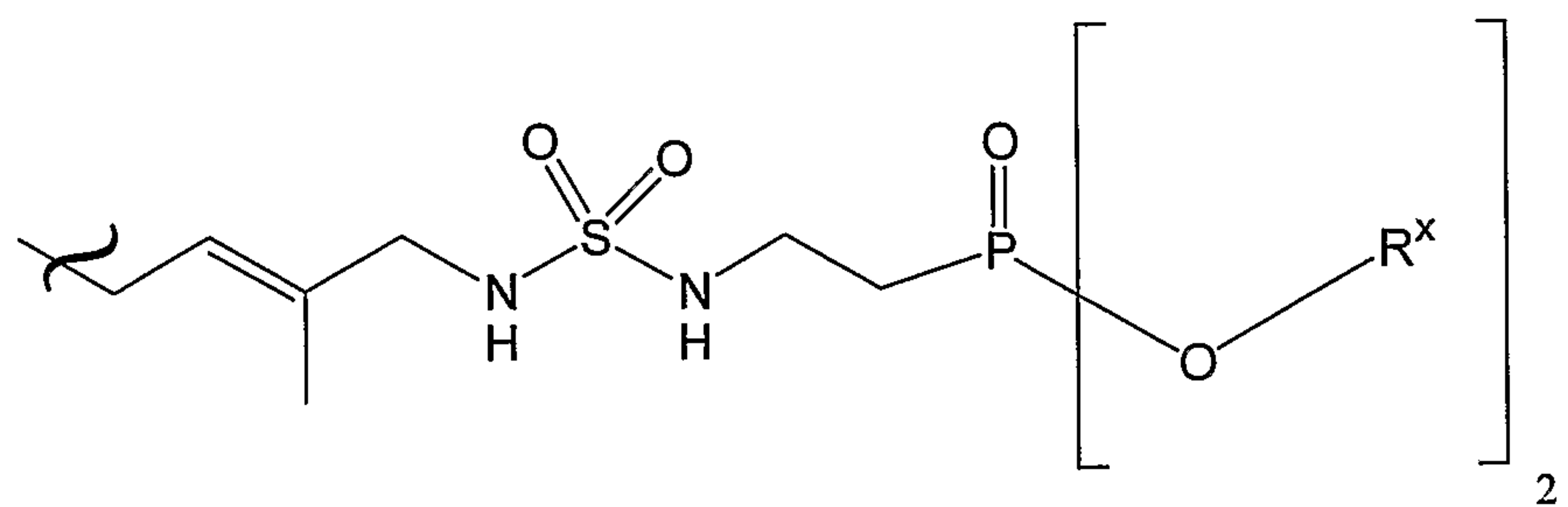
150. The compound of claim 133 wherein A^0 is of the formula:



10 151. The compound of claim 133 wherein A^0 is of the formula:

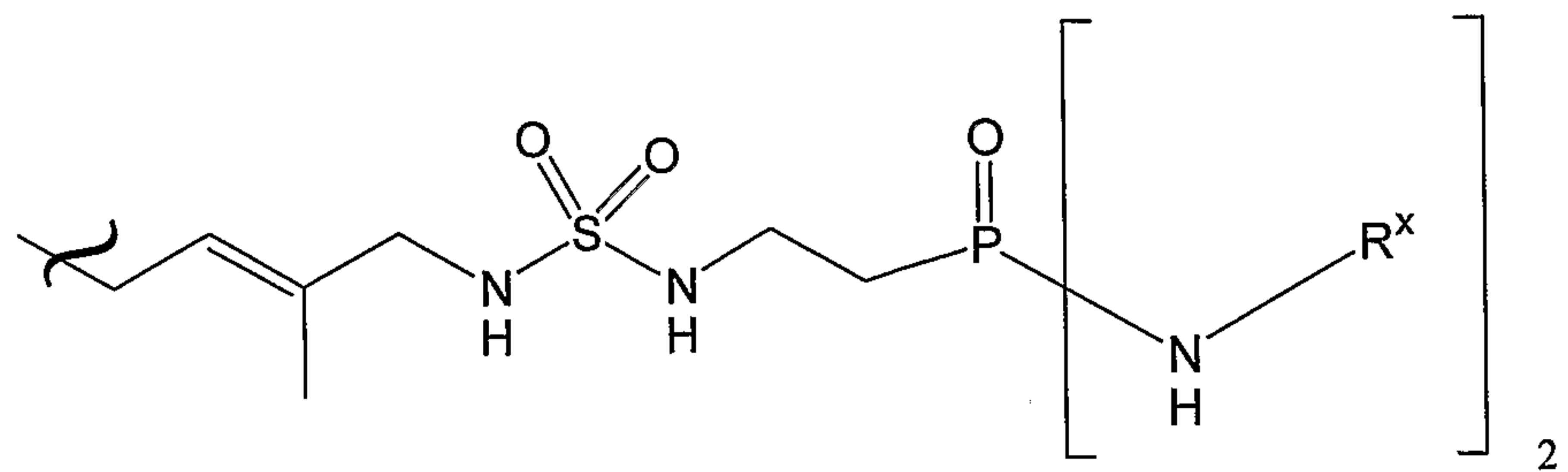


152. The compound of claim 133 wherein A^0 is of the formula:



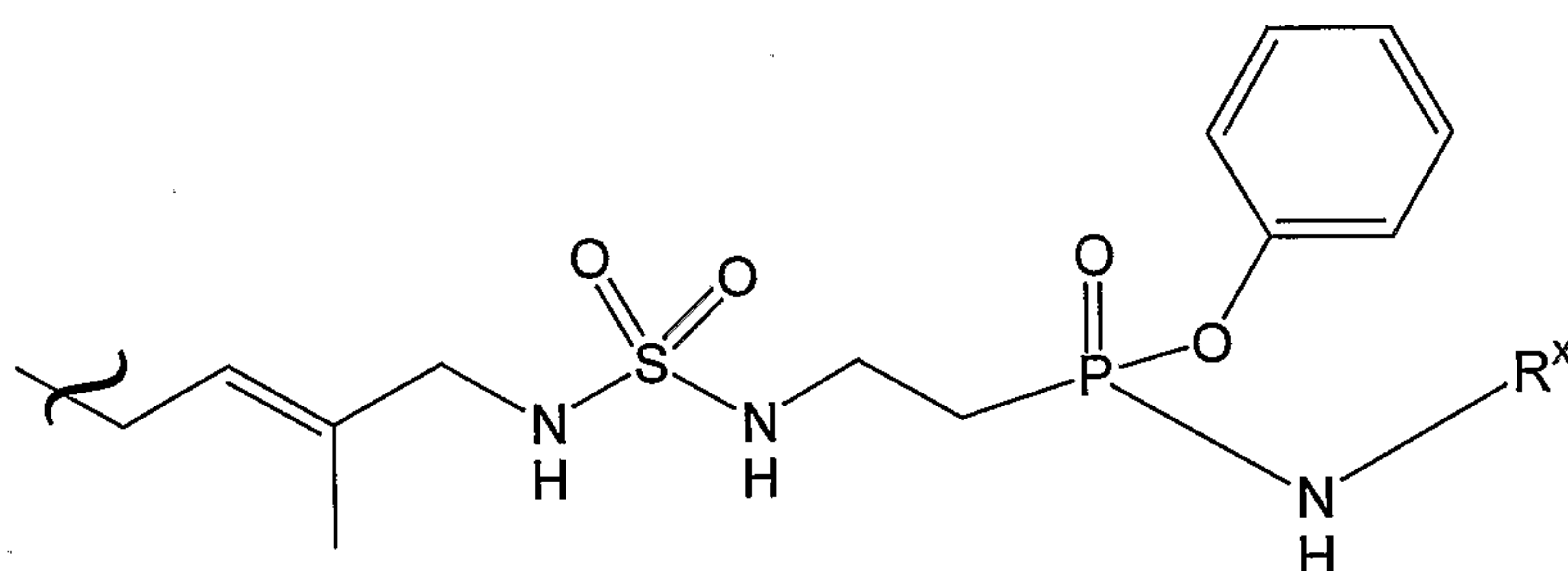
153. The compound of claim 151 or 152 wherein R^x is hydrogen or alkyl.

154. The compound of claim 133 wherein A^0 is of the formula:



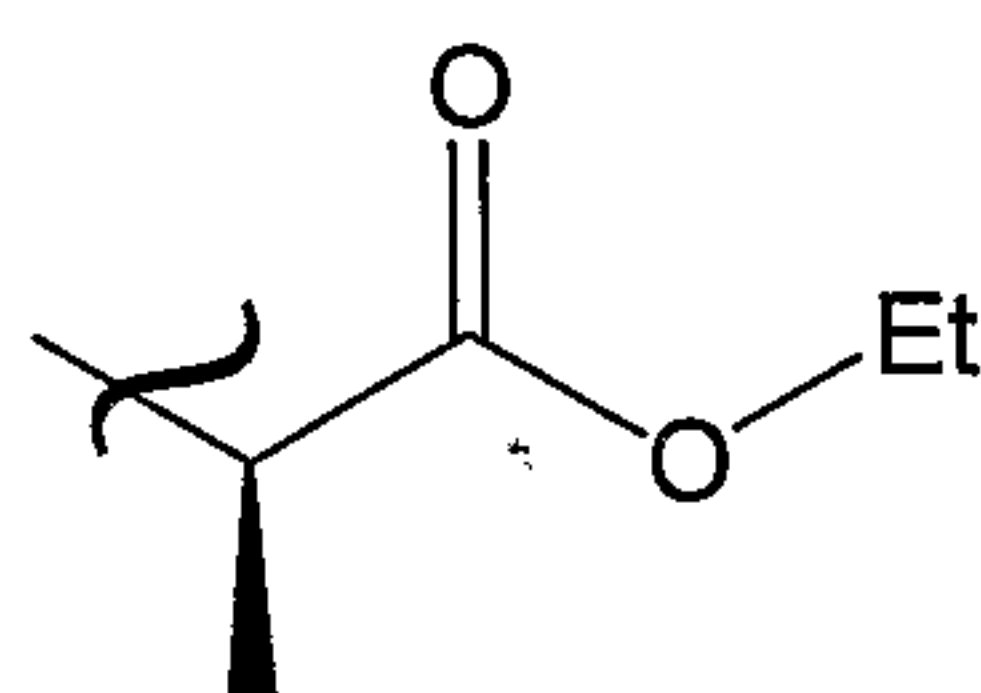
5

155. The compound of claim 133 wherein A^0 is of the formula:

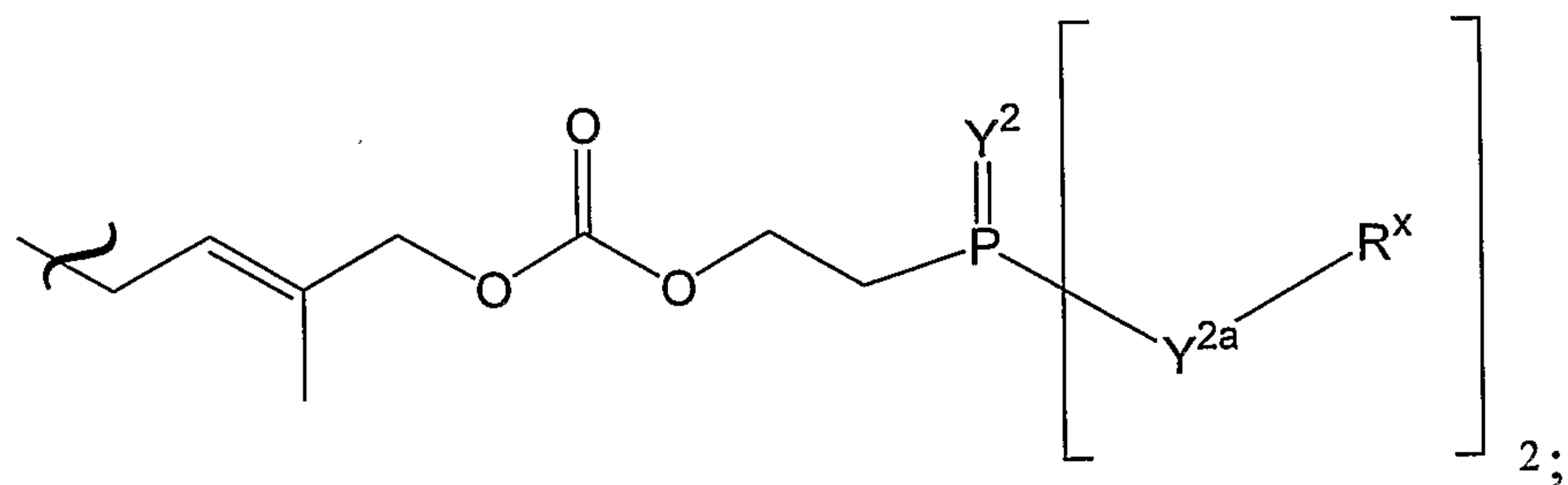


156. The compound of claim 154 or 155 wherein R^x is of the formula:

10



157. The compound of claim 133 wherein A^0 is of the formula:



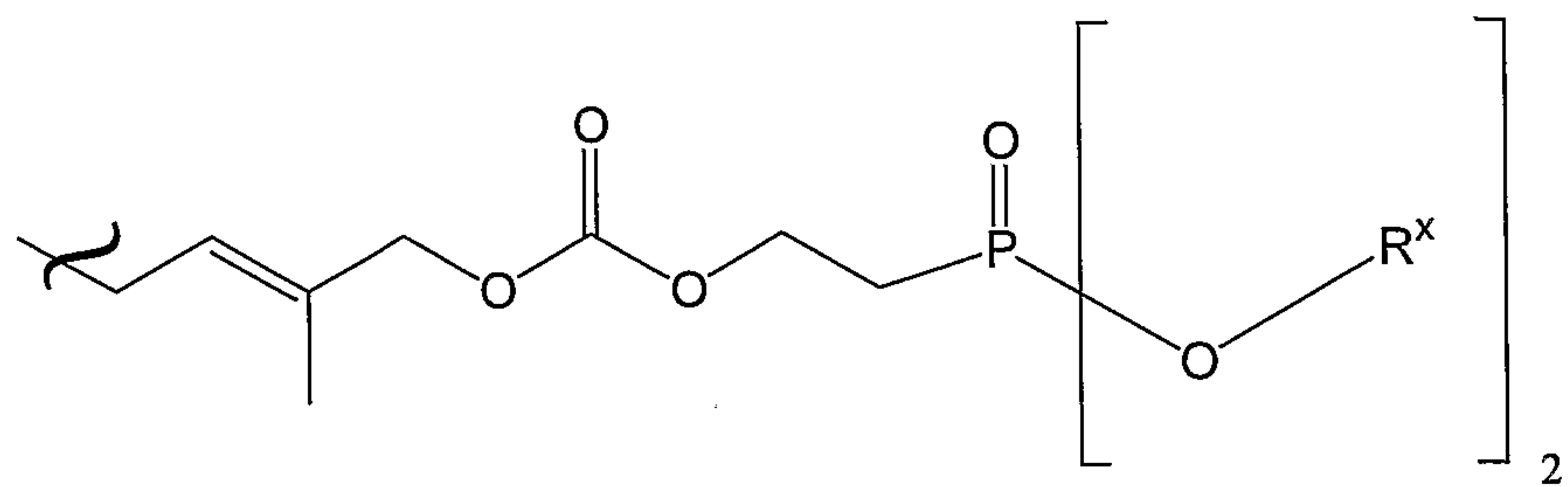
15

wherein:

Y^2 is O or S; and

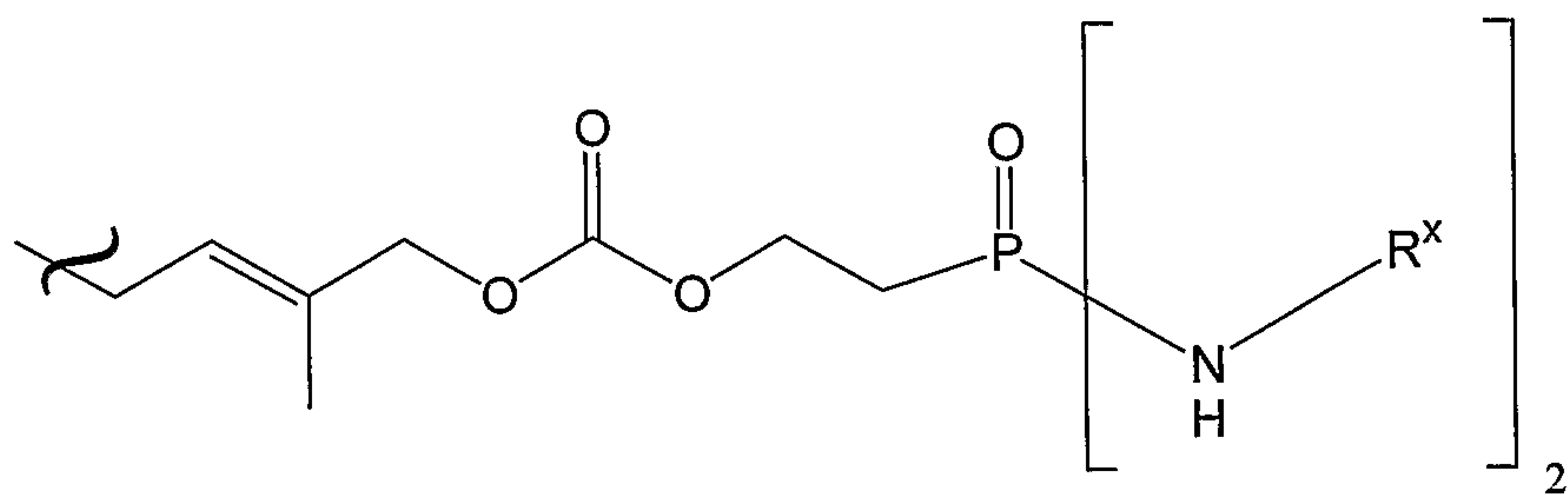
Y^{2a} is O, N(R^x) or S.

158. The compound of claim 133 wherein A⁰ is of the formula:



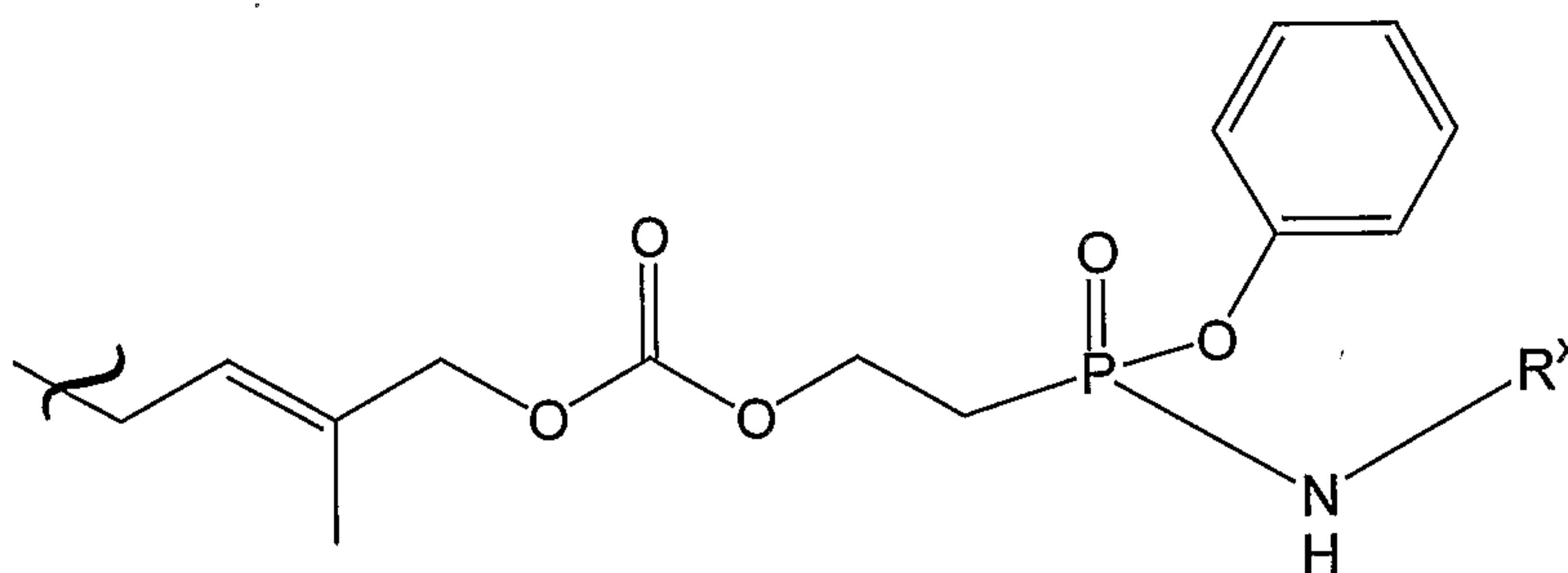
5

159. The compound of claim 133 wherein A⁰ is of the formula:



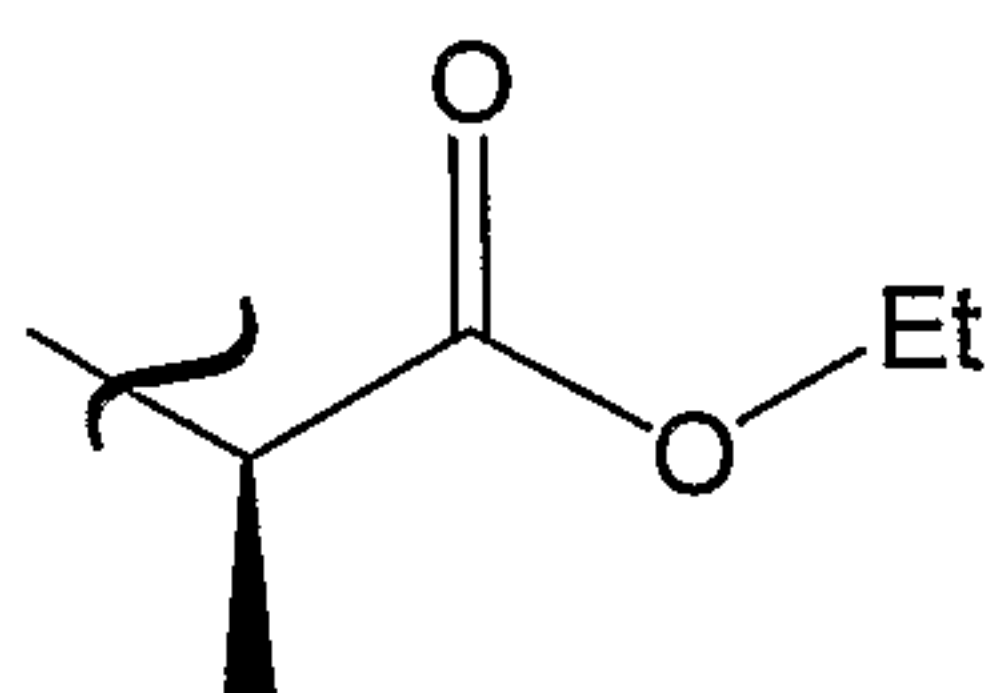
10 160. The compound of claim 159 wherein R^x is hydrogen or alkyl.

161. The compound of claim 133 wherein A⁰ is of the formula:



15

162. The compound of claim 159 or 161 wherein R^x is of the formula:



163. The compound of any one of claims 1-163 which is isolated and purified.

5

164. A method of inhibiting tumor growth comprising the step of contacting a sample or subject suspected of containing a tumor with a compound as described in any one of claims 1-163.

10 165. The method of claim 164 wherein the tumor is *in vivo*.

166. A method for the treatment or prevention of the symptoms or effects of cancer in an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound as described in
15 any one of claims 1-163.

167. The method of claim 165 wherein the compound is formulated with a pharmaceutically acceptable carrier.

20 168. The method of claim 166 or 167 wherein the formulation further comprises a second active ingredient.

169. A method of inhibiting the activity of a virus comprising the step of contacting a sample or subject suspected of containing a virus with a compound
25 as described in any one of claims 1-163.

170. The method of claim 169 wherein the virus is *in vivo*.

171. A method for the treatment or prevention of the symptoms or effects of viral infection in an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound as described in any one of claims 1-163.

5

172. The method of claim 171 wherein the compound is formulated with a pharmaceutically acceptable carrier.

173. The method of claim 171 or 172 wherein the formulation further
10 comprises a second active ingredient.

174. A method of inhibiting inflammation comprising the step of contacting a sample or subject suspected of being inflamed with a a compound as described in any one of claims 1-163.

15

175. The method of claim 174 wherein the inflammation is *in vivo*.

176. A method for the treatment or prevention of the symptoms or effects of inflammation in an animal which comprises administering to said animal a
20 formulation comprising a therapeutically effective amount of a a compound as described in any one of claims 1-163.

177. The method of claim 176 wherein the compound is formulated with a pharmaceutically acceptable carrier.

25

178. The method of claim 176 or 177 wherein the formulation further comprises a second active ingredient.

179. A method for the treatment or prevention of the symptoms or effects of
30 tissue or organ transplant rejection in an animal which comprises administering to said animal a formulation comprising a therapeutically effective amount of a compound as described in any one of claims 1-163.

180. The method of claim 179 wherein the compound is formulated with a pharmaceutically acceptable carrier.

181. The method of claim 179 or 180 wherein the formulation further
5 comprises a second active ingredient.

182. A pharmaceutical composition comprising a pharmaceutical carrier and a compound as described in any one of claims 1-163.

10 183. The composition of claim 182 which further comprises at least one additional active ingredient.

184. A unit dosage form comprising a compound as described in any one of claims 1-163 and a pharmaceutically acceptable carrier.

15

185. The unit dosage form of claim 184 which further comprises at least one additional active ingredient.

186. The use of a compound a compound as described in any one of claims 1-
20 163 to prepare a medicament for treatment of cancer.

187. The use of a compound as described in any one of claims 1-163 to prepare a medicament for treatment of a virus infection.

25 188. The use of a compound as described in any one of claims 1-163 to prepare a medicament for treatment of inflammation.

189. The use of a compound as described in any one of claims 1-163 to prepare a medicament for treatment of tissue or organ transplant rejection.

30

190. A compound described in the Examples herein.