(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau

(43) International Publication Date
14 July 2011 (14.07.2011)

PCT

(10) International Publication Number
WO 2011/085281 A1

(51) International Patent Classification:
C09D 11/00 (2006.01)

(21) International Application Number:
PCT/US2011/020622

(22) International Filing Date:
10 January 2011 (10.01.2011)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
61/293,868 11 January 2010 (11.01.2010) US

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(81) Designated States *(unless otherwise indicated, for every kind of national protection available)*: AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

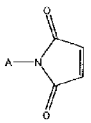
(84) Designated States *(unless otherwise indicated, for every kind of regional protection available)*: ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— *with international search report (Art. 21(3))*

WO 2011/085281 A1

(54) Title: COMPOSITIONS COMPRISING A REACTIVE MONOMER AND USES THEREOF



(I)

(57) Abstract: Disclosed herein are compositions such as ink or coating compositions that include at least one reactive monomer that includes a maleimide of the general formula (I) A — is or includes an aryl, an alkylaryl, or an arylalkyl. The reactive monomer may be or include 4-butylphenyl maleimide. The composition may also include a reactive co-solvent that may include at least one of a (meth)acryl monomer or prepolymer, a (meth)acryl ester of an epoxy type monomer or prepolymer, and a urethane type monomer or prepolymer.

COMPOSITIONS COMPRISING A REACTIVE MONOMER AND USES THEREOF

FIELD OF THE INVENTION

[0001] This application relates to compositions comprising a reactive monomer that is suitable for inks or ink coatings and processes for preparing the same, and more particularly, to compositions wherein the reactive monomers comprise a maleimide.

BACKGROUND OF THE INVENTION

[0002] Commercial printing processes are dominated by lithography, flexography, letterpress, screen printing and electrophotographic printing. Rapidly evolving technologies for sublimation/melt-type printing and ink-jet are becoming more commercially attractive processes. Increasing in influence of these printing technologies is curable ink systems.

[0003] A distinguishing feature of printing ink is its visual appearance. The color, transparency, intensity or density, and gloss often determine the suitability of the ink for a particular application. Another distinguishing feature of printing ink is its adhesion to surfaces, resistance to scratching and defacement, impact resistance, resistance to heat, resistance to solvents or other media, lightfastness, UV stability, and flexibility.

[0004] In many printing processes, once these challenges are met, the ink is then evaluated for suitability for color matching. Color matching often requires the use of one colored ink in concert with other different colored inks. In one example, International Commission on Illumination (CIE) color matching, provides for an increase in the color spectrum though a process of mixing primary colors (red, green, and blue) to produce secondary colors (cyan, magenta, yellow) and myriads of possibilities between them. For such a system to function properly, the ink must be truly compatible, not only in physical/chemical properties, but in color properties too.

[0005] There are many raw materials employed in the manufacturing of ink products. The four basic components of a printing ink are pigments and dyes, resins, solvents, and additives. These components can be broken down into further details covering potential ingredients such as pigments and dyes, oils, resins, solvents, plasticizers, waxes, driers, chelating agents, anti-oxidants, surfactants, deodorants and fragrances, defoaming agents, adhesion promoters, photo-initiators, reactive diluents, oligomers, inhibitors, and laking agents. Not all of these ingredients will be used for all inks and some ingredients are capable of serving more than one purpose.

[0006] Viscosity is a key element to the physical properties and commercial performance capabilities of an ink system. As indicated in the Kipphan's Handbook of Print Media: Technologies and Production Methods (Springer Verlag, New York, **2001**) and Leach and Pierce's Printing Ink Manual (Kluwer, Boston, **1999**) typical ranges of viscosity are presented below in Table 1.

Table 1: Typical Viscosity Ranges for Various Printing Processes

Printing Process	**Typical Viscosity Range (Pa*s)**
Lithography	2 to 30
Offset	40 to 100
Letterpress	50 to 150
Sublimation and Melt-Type printing	solid at room temperature and melts at elevated temperature
Electrophotographic	~0.1 to 10, for liquid toner Solid, for dry toner
Flexography	0.05 to 0.5
Gravure	0.01 to 0.2
Screen	1.5 to 2.0 or higher
Ink-jet	~0.001 to 0.1
Intaglio	9 to 25

[0007] Typical techniques for measuring the viscosity of an ink system include capillary viscometers, falling sphere viscometers, flow cups (i.e., Zahn, Shell and Ford), rotational viscometers, cone and plate viscometers (i.e., Haake, TA Instruments), controlled stress rheometers, falling bar viscometers and the like.

[0008] As printing speeds become faster and materials more specialized, certain aspects of the printing process have evolved. For example, in some printing press applications, it is not uncommon to employ substrates that are pre-treated, by providing a primer coating to enable adhesion to the surface or surface treating with corona or flame, thereby enabling good ink performance on the substrate despite the added cost in materials and/or production time.

[0009] Printing technologies are applied to many different surfaces. For example, polyester film, polyolefin film (PE and PP), polycarbonate, polyimide film, metals (i.e., aluminum, steel,

copper), glass, vinyl film, Tyvec, canvas, polyvinylidene chloride films, paper, polyurethane, ceramics, wood and the like.

[0010] Given the many requirements and challenges for reactive materials in ink applications, new reactive monomers and compositions containing them may be developed as discussed herein.

SUMMARY OF THE INVENTION

[0011] Compositions such as ink or coating compositions containing reactive monomers are disclosed herein. The reactive monomers include a maleimide and may be curable by exposure to various types of energy, and as such may include a curing initiator. The compositions include at least one reactive monomer having a maleimide of the general formula $H_2C_2(CO)_2N$-A, wherein A— is or includes an aryl, alkylaryl, or arylalkyl.

[0012] In one embodiment, the reactive monomer may be or include 4-butylphenyl maleimide. In another embodiment, the composition may include a reactive co-solvent that may include at least one of a (meth)acryl monomer or prepolymer, a (meth)acryl ester of an epoxy type monomer or prepolymer, and a urethane type monomer or prepolymer.

[0013] In another aspect, methods of printing or coating a substrate are disclosed that apply the ink or coating compositions described herein to a substrate. The substrate may be paper, metal, plastic, glass, textile, or combinations thereof.

DETAILED DESCRIPTION OF THE INVENTION

[0014] The term "reactive monomer" throughout the specification and the claims is defined as a material comprised of a carbon-carbon double bond. When activated as described herein, the carbon-carbon double bond reacts with other carbon-carbon double bonds to form a polymeric material.

[0015] The term "curing" throughout the specification and the claims refers to the process of polymerizing, e.g., converting a liquid to a solid, by exposure to appropriate energy source. The resulting cured product is incapable of demonstrating a molecular weight as determined by gel permeation chromatography (GPC).

The term "lithography" throughout the specification and the claims refers to a printing process that employs a flat printing plate. The printable area is usually ink-receptive while the non-printable area repels ink.

[0016] The term "flexography" throughout the specification and the claims refers to a relief printing process. This process employs rubber or photopolymer plates and an ink system. Typically, printing is achieved via transfer of the ink from an anilox roller.

[0017] The term "letterpress" throughout the specification and the claims refers to a relief printing process employing rubber or photopolymer plates and an ink system. Typically, printing is achieve via transfer of the ink from a roller and pressed into contact with the substrate.

[0018] The term "screen printing" throughout the specification and the claims refers to a stencil printing process. A rubber squeegee is employed to push ink through a stencil onto a substrate.

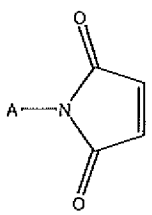
[0019] The term "ink-jet" throughout the specification and the claims refers to a computer controlled stream of ink droplets, ejected at high speed, onto a printing surface.

[0020] The term "electrophotographic" throughout the specification and the claims refers to a printing process whereby light is used to selectively discharge an electrostatic field, forming an electrostatically charged image. Toner of a proper charge is then transfer to the substrate and fused to the surface by heat or other process.

[0021] The compositions herein may be curable ink systems that include a reactive monomer. The curable ink system may have a polymerization process that involves reacting the reactive monomer with another reactive component of the composition or with the same or different additional reactive monomers. The polymerization process may be initiated by thermal effects or irradiation such as by exposure to α, γ, and x-rays, UV, E-beam, and the like.

[0022] Among the properties that can be beneficially impacted by the reactive monomers are solution viscosity, cure speed, adhesion, impact resistance, toughness, coating hardness, surface tension, wetting, foaming, tensile strength, solvency, dispersive properties, flexibility, chemical resistance, abrasion resistance, and penetration.

[0023] The reactive monomers may include a maleimide of the general formula $H_2C_2(CO)_2N$-A, which has the chemical structure shown in formula (1) below.



(1)

A— is or includes an aryl, alkylaryl, or an arylalkyl. **A—**, however, should not produce a compound that has a melting point that is unfavorable to formulation in ink or coating compositions. In that respect, the melting point of the maleimide of formula (1) may be less than about 150°C. In one embodiment, the melting point of the maleimide may be between about 10°C to 125°C.

[0024] The aryl may be a carbocyclic or heterocyclic aryl. Suitable carbocyclic aryls include phenyl, naphthyl, and biphenyl. Heterocyclic aryls may include one or more oxygen, nitrogen, and/or sulfur atoms in the ring, for example, the aryl may be a furanyl, a pyranyl, or a pyridyl. Non-limiting examples of such maleimides include N-phenyl maleimide, 1-(1-naphthalenyl)-1H-pyrrole-2,5-dione, 1-(2-naphthalenyl)-1H-pyrrole-2,5-dione, and 1-(biphenyl)-1H-pyrrole-2,5-dione.

[0025] In one embodiment, **A—** includes a phenyl. The phenyl may substituted with one or more halogens, alkyl or fluoroalkyl groups which may be straight chain, branched or cyclic, alkoxy groups, nitro groups, and carboxy groups. When one substituent is present on the phenyl, it may be at the 2 position, 3 position, or 4 position. When two substituents are present on the phenyl, the substituents may be at the 2 and 3 positions, 2 and 4 positions, 2 and 5 positions, 2 and 6 positions, 3 and 4 positions, or 3 and 5 positions. If one of the substituents is an alkyl or fluoroalkyl, the alkyl may contain 1-20 carbons, preferably 1-5 carbons.

[0026] Non-limiting examples of suitable maleimides having **A—** as a substituted phenyl are given below in Table 1.

Table 1

[0027] Structure Name

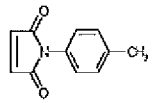
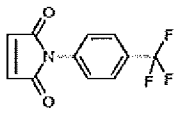
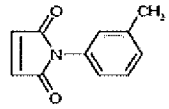
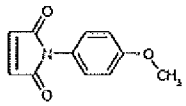
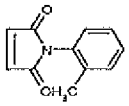
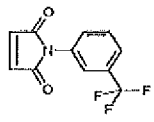
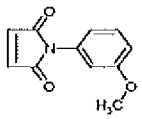
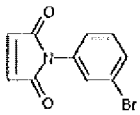
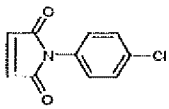
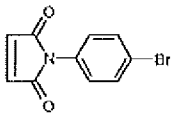
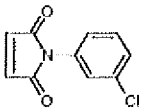
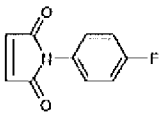
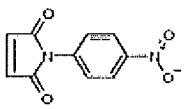
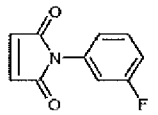
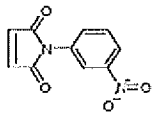
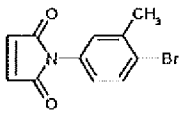
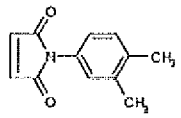
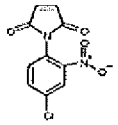
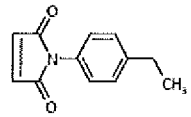
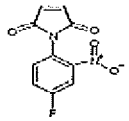
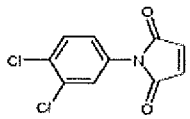
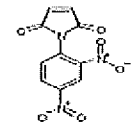
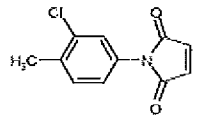
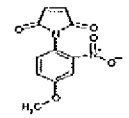
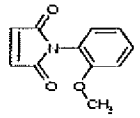
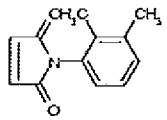
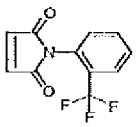
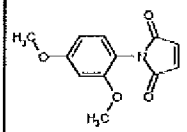
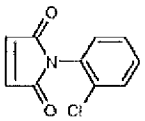
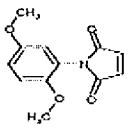
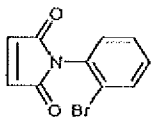
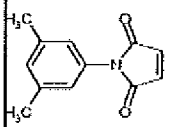
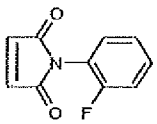
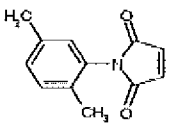
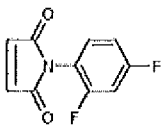
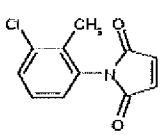
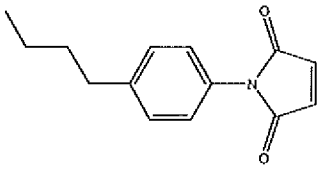
	1-(4-Methylphenyl)-1H-pyrrole-2,5-dione		1-(4-(Trifluoromethyl)phenyl)-1H-pyrrole-2,5-dione
	1-(3-Methylphenyl)-1H-pyrrole-2,5-dione		1-(4-Methoxyphenyl)-1H-pyrrole-2,5-dione
	1-(2-Methylphenyl)-1H-pyrrole-2,5-dione		1-(3-(Trifluoromethyl)phenyl)-1H-pyrrole-2,5-dione
	1-(3-Methoxyphenyl)-1H-pyrrole-2,5-dione		1-(3-Bromphenyl)-1H-pyrrole-2,5-dione
	1-(4-Chlorophenyl)-1H-pyrrole-2,5-dione		1-(4-Bromphenyl)-1H-pyrrole-2,5-dione
	1-(3-Chlorophenyl)-1H-pyrrole-2,5-dione		1-(4-Fluorophenyl)-1H-pyrrole-2,5-dione
	1-(4-Nitrophenyl)-1H-pyrrole-2,5-dione		1-(3-Fluorophenyl)-1H-pyrrole-2,5-dione
	1-(3-Nitrophenyl)-1H-pyrrole-2,5-dione		1-(4-Brom-3-methylphenyl)-1H-pyrrole-2,5-dione

Table 1 (Continued)

[0028]

Structure	Name	Structure	Name
	1-(3,4-Dimethylphenyl)-1H-pyrrole-2,5-dione		1-(4-Chloro-2-nitrophenyl)-1H-pyrrole-2,5-dione
	1-(4-Ethylphenyl)-1H-pyrrole-2,5-dione		1-(4-Fluoro-2-nitrophenyl)-1H-pyrrole-2,5-dione
	1-(3,4-Dichlorphenyl)-1H-pyrrole-2,5-dione		1-(2,4-Dinitrophenyl)-1H-pyrrole-2,5-dione
	1-(3-Chlor-4-methylphenyl)-1H-pyrrole-2,5-dione		1-(4-Methoxy-2-nitrophenyl)-1H-pyrrole-2,5-dione
	1-(2-Methoxyphenyl)-1H-pyrrole-2,5-dione		1-(2,3-Dimethylphenyl)-1H-pyrrole-2,5-dione
	1-(2-(Trifluoromethyl)phenyl)-1H-pyrrole-2,5-dione		1-(2,4-dimethoxyphenyl)-1H-pyrrole-2,5-dione
	1-(2-Chlorophenyl)-1H-pyrrole-2,5-dione		1-(2,5-Dimethoxyphenyl)-1H-pyrrole-2,5-dione
	1-(2-Bromphenyl)-1H-pyrrole-2,5-dione		1-(3,5-Dimethylphenyl)-1H-pyrrole-2,5-dione
	1-(2-Fluorophenyl)-1H-pyrrole-2,5-dione		1-(2,5-Dimethylphenyl)-1H-pyrrole-2,5-dione
	1-(2,4-Difluorophenyl)-1H-pyrrole-2,5-dione		1-(3-chloro-2-methylphenyl)-1H-pyrrole-2,5-dione

[0029] As seen in Table 1, in some embodiments, **A**— is an alkylaryl. The alkylaryl may be a butylphenyl, which as shown in formula (2) below may have the butyl substituted at the 4 position of the phenyl.



(2)

[0030] In an alternate embodiment, **A**— may be an arylalkyl. Non-limiting examples of suitable arylalkyls are given below in Table 2.

Table 2

[0031]

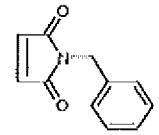
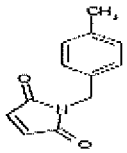
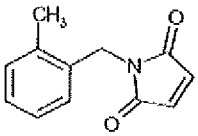
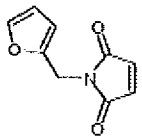
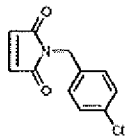
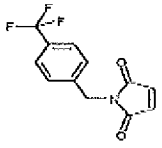
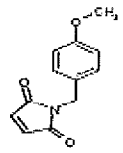
Structure	Name
	1-((Phenyl)methyl)-1H-pyrrole-2,5-dione N-BENZYLMALEIMIDE
	1-((4-methylphenyl)methyl)-1H-pyrrole-2,5-dione
	1-((2-methylphenyl)methyl)-1H-pyrrole-2,5-dione
	2-(2-Furanylmethyl))-1H-pyrrole-2,5-dione
	1-((4-Chlorophenyl)methyl)-1H-pyrrole-2,5-dione

Table 2 (Continued)

[0032]

Structure	Name
	1-((4-Trifluoromethylphenyl)methyl)-1H-pyrrole-2,5-dione



1-((4-methoxyphenyl)methyl)-1H-pyrrole-2,5-dione

[0033] In addition to the reactive monomers, the compositions disclosed herein may include reactive co-solvents. The reactive co-solvents may include (meth)acryl monomers or pre-polymers, a (meth)acryl ester of an epoxy type monomer or pre-polymer, and a urethane type monomers or pre-polymers. The reactive co-solvents include a carbon-carbon double bond capable of reacting with the polymerizable unit of the reactive monomers, which itself includes a carbon-carbon double. In one embodiment, a mixture of reactive co-solvents may be in the formulation with the reactive monomer.

[0034] Examples of reactive co-solvents include but are not limited to 2-hydroxy methyl methacrylate (HEMA), 2-hydroxy ethyl acrylate (HEA), 2-phenoxy ethyl acrylate (PHEA), 2-ethylhexyl-diglycol acrylate, 2-(2-ethoxyethoxy)ethyl acrylate (EOEOEA), lauryl acrylate (LA), Stearyl acrylate (SA), isobornyl acrylate (IBOA), acrylic acid-2-ethylhexyl ester, isodecyl acrylate, acryloyl morpholine (ACMO), cyclic trimethylol-propane formal acrylate (CTFA), 3-(Methacryloylamino)propyl]trimethylammonium chloride (MAPTAC), (3-Acrylamidopropyl)trimethylammonium chloride (APTAC), C8-C10 acrylate (ODA), isodecyl acrylate (ISODA), lauryl methacrylate (LM), stearyl methacrylate (SM), 2,2,2-Trifluoroethyl methacrylate, 2-Acrylamido-2-methyl-1-propanesulfonic acid, 2-Acrylamido-2-methyl-1-propanesulfonic acid sodium salt, [2-(Methacryloyloxy)ethyl]dimethyl-(3-sulfopropyl)ammonium hydroxide, [3-(Methacryloylamino)propyl]dimethyl(3-sulfopropyl)ammonium hydroxide inner salt, 1,6-hexanediol diacrylate (HDDA), dipropylene glycol diacrylate (DPGDA), tripropylene glycol diacrylate (TPGDA), 1,4-butanediol diacrylate (BDDA), Tripropylene glycol diacrylate (TPGDA), dipropyleneglycol diacrylate (DPGDA), Tripropylene glycol diacrylate (TRPGDA), 1,9-nonanediol diacrylate (NNDA), neopentyl glycol diacrylate (NPGDA), propoxylated neopentyl glycol diacrylate (NPG2PODA), polyethylene glycol (200) diacrylate (PEG(200)DA), polyethylene glycol (400) diacrylate (PEG(400)DA), polyethylene glycol (600) diacrylate (PEG(600)DA), ethoxylated bisphenol-A diacrylate (BPA2EODA), triethylene glycol diacrylate (TEGDA), triethylene glycol dimethacrylate (TEGDMA), glycerol propoxylated triacrylate (GPTA), diethylene glycol dimethacrylate (DEGDMA), ethoxylated bisphenol-A dimethacrylate (BPA10EODMA), trimethylolpropane triacrylate (TMPTA), pentaerythritol triacrylate (PET3A), ethoxylated tri-methylolpropane triacrylate (TMP3EOTA), propxylated tri-methylolpropane triacrylate (TMP3POTA),

propoxylated glyceryl triacrylate (GPTA), trimethylolpropane trimethylacrylate (TMPTMA), ethoxylated trimethylolpropane trimethacrylate (TMP3EOTMA), 2,2-dionol diacrylate, pentaerythritol tetraacrylate (PETA), neopentylglycol diacrylate hydroxypivalate, 2-acryloyloxyethylphthalic acid, 2-acryloyloxyethyl-2-hydroxyethylphthalic acid, dimethyloltricyclodecane diacrylate, 2-acryloyloxyethylsuccinic acid, nonylphenol ethylene oxide adduct acrylate, methoxy-polyethylene glycol acrylate, tetramethylolmethane triacrylate, dipentaerythritol hexaacrylate (DPHA), isocyanate-functional unsaturated acrylic ester resin, urethane diacrylates oligomers, urethane acrylates, modified urethane acrylates, polyester acrylates, modified bisphenol A diacrylate, phenoxy-polyethylene glycol acrylate, bisphenol A propylene oxide modified diacrylate, bisphenol A ethylene oxide adduct diacrylate, pentaerythritol triacrylate hexamethylenediisocyanate, urethane prepolymer, isoamyl acrylate, isomyristyl acrylate, isostearyl acrylate, carbitol acrylate, cyclohexyl acrylate, tetrahydrofurfuryl acrylate, 1,4-butane-diol-monoacrylate and/or diglycidyl ether of 1,4-butanediol, and the like.

[0035] Additional examples of reactive co-solvents include methyl vinylether, ethyl vinylether, propyl vinylether, n-butyl vinylether, t-butyl vinylether, 2-ethylhexyl vinylether, n-nonyl vinylether, lauryl vinylether, cyclohexyl vinylether, cyclohexylmethyl vinylether, 4-methylcyclohexylmethyl vinylether, benzyl vinylether, dicyclopentenyl vinylether, 2-dicyclopentenoxyethyl vinyl ether, methoxyethyl vinylether, ethoxyethyl vinylether, butoxyethyl vinyl ether, methoxyethoxy vinylether, ethoxyethoxyethyl vinylether, methoxypolyethylene glycol vinylether, tetrahydrofurfuryl vinylether, dodecyl vinylether, diethylene glycol monovinylether, 2-hydroxyethyl vinylether, 2-hydroxypropyl vinylether, 4-hydroxybutyl vinylether, 4-hydroxymethylcyclohexylmethyl vinylether, polyethylene glycol vinylether, chloroethyl vinylether, chlorobutyl vinylether, phenylethyl vinylether, phenoxypolyethylene glycol vinylether, ethylene glycol divinylether, butylenes glycol divinylether, hexandiol divinylether, bisphenol A alkyleneoxide divinylethers, bisphenol F alkyleneoxide divinylethers, propyleneoxide adducts of trimethylolpropane trivinylether, triethylene glycol divinylether, cyclohexane dimethanol divinylether, N-vinyl-2-pyrrolidone (VP), N-vinyl caprolactam (VCap), N-vinyl imidazole (VI), n-vinyl amides, 4-vinyl pyridine, 2-vinyl pyridine, styrene, 5-vinyl-2-norbornene and the like.

[0036] Non-limiting examples of monofunctional epoxy compounds include phenyl glycidylether, p-tert-butylphenyl glycidylether, butyl glycidylether, 2-ethylhexyl glycidylether, allyl glycidylether, 1,2-butyleneoxide, 1,3-butadienemonooxide, 1,2-epoxydodecane, epichlorohydrin, 1,2-epoxydecane, styreneoxide, cyclohexeneoxide, 3-

methacryloyloxymethylcylcohexeneoxide, 3-acryloyloxymethylcylcohexeneoxide, 3-vinylcylcohexeneoxide, and the like.

[0037] Non-limiting examples of multifunctional epoxy compounds include 3,4-Epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate, 3-ethyl-3-((ethyloxetane-3-yl)methoxy) methyl)oxetane, bisphenol A diglycidylether, bisphenol F diglycidylether, bisphenol S diglycidylether, brominated bisphenol A diglycidylether, brominated bisphenol F diglycidylethers, brominated bisphenol S diglycidylether, epoxy novolak resins, hydrogenated bisphenol A diglycidylethers, hydrogenated bisphenol F diglycidylethers, hydrogenated bisphenol S diglycidylethers, 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexanecarboxylate, 2-(3,4-epoxycyclohexyl-5,5-spiro-3,4-epoxy) cyclohexane-meta-dioxane, bis(3,4-epoxycyclohexylmethyl) adipate, vinylcylcohexeneoxide, 4-vinylepoxycyclohexane, bis(3,4-epoxy-6- methylcyclohexylmethyl) adipate, 3,4-epoxy-6-methylcyclohexyl-3',4'-epoxy-6'-methylcyclohexane carboxylate, methylene-bis(3,4-epoxycyclohexane), dicyclopentadiene diepoxide, ethylene glycol di(3,4-epoxycyclohexylmethyl)ether, ethylene bis(3,4-epoxycyclohexanecarboxylate), epoxyhexahydrodioctyl phthalate, epoxyhexahydrodi-2-ethylhexyl phthalate, 1,4-butanediol diglycidylether, 1,6-hexanediol diglycidylether, glycerol triglycidylether, trimethylolpropane triglycidylether, polyethylene glycol diglycidylether, polypropylene glycol diglycidylether, 1,1,3-tetradecadienedioxide, limonenedioxide, 1,2,7,8-diepoxyoctane, 1,2,5,6-diepoxycyclooctane, and the like.

[0038] The compositions containing the reactive monomers and reactive co-solvents may be cured or cross-linked or polymerized by any appropriate known or yet to be developed method. The polymerization may be carried out by employing any method disclosed in *"Principles of Polymerization" 4th edition, 2004, Wiley by George Odian*, which is incorporated herein by reference in its entirety. Non-limiting exemplary methods include exposure to UV-radiation, UV-LED, laser beam, electron beam, and/or gamma irradiation or other high-energy source, and free-radical, cationic, anionic, or thermal polymerization, which may occur in the presence of suitable initiator(s) such as photoinitiators, free-radical initiators, anionic or cationic initiators, and thermal initiators. Suitable sources of radiation include, but are not limited to, mercury, xenon, halogen, and carbon arc lamps, sunlight, and radioactive sources.

[0039] A photoinitiator may be added to the compositions herein to initiate polymerization upon exposure of the composition to radiation. Suitable photoinitiators include those selected from the following non-limiting group of compounds: 2-hydroxy-2-methyl-1-phenylpropane-1-one, 1-hydroxycyclohexyl phenyl ketone, and 2-methyl-1-[4-(methylthio)phenyl]-2-

morphorinopropane-1-on; benzoins e.g. benzyl dimethyl ketal; benzophenones such as benzophenone, 4-phenylbenzophenone, and hydroxybenzophenone; thioxanthones such as isopropylthioxanthone and 2,4-diethylthioxanthone; acylphosphine oxides; and other special initiators such as methyl phenyl glyoxylate; bis[4-(di(4-(2-hydroxyethyl)phenyl)sulfonio)phenyl sulfide], a mixture of bis[4-diphenylsulfonio]phenyl)sulfide bis(hexafluoroantimonate and diphenyl-4-thiophenoxyphenylsulfonium hexafluoroantimonate, bis[4-(di(4-(2-hydroxyethyl)phenyl)sulfonio)phenyl sulfide], 5-2,4-cyclopentadiene-1-yl- [(1,2,3,4,5,6-.eta.)-(1-methylethyl-)benzene]-iron (1+)-hexafluorophosphate(1-)), 4-(2-hydroxytetradecanyloxy) diphenyliodonium hexafluoroantimonate, (4-hydroxynaphtyl) dimethylsulfonium hexafluoroantimonate), triphenylsulfonium hexafluorophosphate, triphenylsulfonium hexafluoroantimonate, 4-methoxyphenyldiphenylsulfonium hexafluoroantimonate, 4-methoxyphenyliodonium hexafluoroantimonate, bis(4-tert-butylphenyl)iodonium tetrafluoroborate, (bis(4-tert-butylphenyl)iodonium hexafluorophosphate), (bis(4-tert-phenyl)iodonium hexafluoroantimonate), (bis[4-(diphenylsulfonio)phenyl]sulfide bis(hexafluorophosphate)), Aryldiazonium salts, diaryliodonium salts, triaylsulfonium salts, triarylselenonium salts, dialkylphenacylsulfonium salts, triarylsulfoxonium salts, aryloxydiarylsulfonium salts, and the like for example, triphenylsulfonium hexaflurophosphate, methyidiphenylsulfonium hexafluorophosphate, dimethylphenylsulfonium hexaflurophosphate, diphenyinapththylsulfonium hexaflurophosphate, di(methoxynapththyl)methylsulfonium hexaflurophosphate, (4-octyloxyphenyl) phenyl iodonium hexafluoro antimonate, (4-octyloxyphenyl) diphenyl sulfonium hexafluoro antimonate, (4-decyloxyphenyl) phenyl iodonium hexafluoro antimonite, (4-dodecyloxyphenyl)diphenyl sulfonium hexafluoroantimonate. Particularly, employed photoinitaitors include 10-biphenyl-4-yl-2-isopropyl-9H-thixanthen-10-ium hexafurophosphate, 4,4'-dimethyl iodonium hexaflurophosphate, mixed triarylsulfonium hexaflurophosphate salts and reaction products of polyol and 10-(2-carboxymethoxy)-biphenyl-4yl-2-isopropyl-9-oxo-9H-thioxanthen-10-ium hexaflruophosphate. Further, these photoinitiators are used alone or in combination thereof. Alternatively, if essential, the photoinitiator may be used by mixing it with one or more photopolymerization accelerator, such as a benzoic acid (e.g., 4-dimethylaminobenzoic acid) or a tertiary amine, in any appropriate ratio. The photoinitiator may be added to the photopolymerizable composition in the range of about 0.1% to about 20% by weight.

[0040] A In one embodiment, the reactive monomers and/or the reactive co-solvents include a polymerizable moiety that may be reacted through free-radical polymerization in the presence of a free-radical initiator. to the polymerizable moiety may be any chemical moiety which upon

exposure to an appropriate energy source (e,g. radiation or heat) decomposes into two independent uncharged fragments left with a highly reactive unpaired electron. Suitable free radical initiator for polymerization include, but are not limited to, various derivatives of peroxides, peresters and/or azo compounds. The free-radical initiator may be selected from dicumyl peroxide, dibenzoyl peroxide, 2-butanone peroxide, tert-butyl perbenzoate, di-tert-butyl peroxide, 2,5-bis(tert-butylperoxy)-2,5-dimethylhexane, bis(tert-butyl peroxyisopropyl)benzene, and tert-butyl hydroperoxide), diacyl peroxides, cumene hydroperoxide, dialkyl peroxides, hydroperoxides, ketone peroxides, monoperoxycarbonates, peroxydicarbonates, peroxyesters, and peroxyketals, including tertiary butyl perbenzoate, tertiary butyl peroctoate in diallyl phthalate, diacetyl peroxide in dimethyl phthalate, dibenzoyl peroxide, 1-hydroxy cyclohexyl-1-phenyl ketone, bis (2,4,6-trimethyl benzoyl)phenyl phosphine, benzoin ethyl ether, 2,2-dimethoxy-2-phenyl acetophenone, di(p-chlorobenzoyl) peroxide in dibutyl phthalate, di(2,4-dichlorobenzoyl) peroxide with dibutyl phthalate, dilauroyl peroxide, methyl ethyl ketone peroxide, cyclohexanone peroxide in dibutyl phthalate, 3,5-dihydroxy-3,4-dimethyl-1,2-dioxacyclopentane, t-butylperoxy(2-ethyl hexanoate), caprylyl peroxide, 2,5-dimethyl-2,5-di(benzoyl peroxy) hexane, 1-hydroxy cyclohexyl hydroperoxide-1, t-butyl peroxy (2-ethyl butyrate), 2,5-dimethyl-2,5-bis(t-butyl peroxy) hexane, cumyl hydroperoxide, diacetyl peroxide, t-butyl hydroperoxide, ditertiary butyl peroxide, 3,5-dihydroxy-3,5-dimethyl-1,2-oxacyclopentane, and 1,1-bis(t-butyl peroxy)-3,3,5-trimethyl cyclohexane and di-(4-t-butyl cyclohexyl) peroxydicarbonate, azo compounds such as azobisisobutyronitrile and azobiscyclohexanenitrile (e.g., 2,2'-azobis(2-methyl-propanenitrile), 2,2'-azobis(2-methylbutanenitrile), and 1,1'-azobis(cyclohexanecarbonitrile)) and the like including mixtures and combinations thereof.

[0041] Alternatively, the free-radical initiators disclosed above may be used for thermal based polymerization alone or as mixture thereof. Other suitable thermal initiators include 2,2'-azobis (2,4-dimethylpentanenitrile), 2,2'-azobis (2-methylpropanenitrile), 2,2'-azobis (2-methylbutanenitrile), peroxides such as benzoyl peroxide, and the like. In one embodiment, the thermal initiator is 2,2'-azobis(isobutyronitrile).

[0042] After curing, cross-linking, or polymerization of the composition, the structure (and presence) of the resulting polymer may be confirmed using spectral techniques known in the art. Non-limiting exemplary spectral techniques include ^{1}H-NMR, ^{13}C-NMR and FT-IR spectra.

[0043] In various embodiment, additives may be included in the described composition.

[0044] In one embodiment, a colorant may be included in the compositions. The colorant may be a pigment or dye. Combinations of pigments and dyes are also envisioned. Suitable pigments are described in Hunger's "Industrial Organic Pigments," Itoh's "Dictionary of Pigments," and Leach and Pierce's "Printing Ink Manual."

[0045] Examples of yellow colored, organic and inorganic, pigments include C.I. Pigment Yellow 1, C.I. Pigment Yellow 74, azo pigments such as C.I. Pigment 12 and C.I. Pigment Yellow 17 and the like.

[0046] Examples of black colored pigments include carbon black, titanium black, aniline black, and the like.

[0047] Examples of white colored pigments include basic lead carbonate, zinc oxide, barium sulfate, titanium oxide, silver white, strontium titanate, and the like.

[0048] Examples of red colored pigments include naphthol red (C.I. Pigment Red 2), C.I. Pigment Red 3, C.I. Pigment Red 176 and C.I. Pigment Red 23 and the like.

[0049] Examples of green colored pigments include phthalocyanine green (C.I. Pigment Green 7), C.I. Pigment Green 36, and C.I. Pigment Green 1 and the like.

[0050] Examples of blue colored pigments include phthalocyanine blue (C.I. Pigment Blue 15:3), C.I. Pigment Blue 15:6, and C.I. Pigment Blue 16 and the like.

[0051] In dispersing the pigment (or the dye???), dispersing machines and dispersants can be employed. Suitable dispersing machines include agitators, ball mills, bead mills, colloid mill, Cowles mixer, Henschel mixer, homogenizer, jet mill, John mill, kneader, pearl mill, roll mill, sand mill, STS mill, Tex mill, ultrasonic wave homogenizer, wet jet mill and the like. Suitable dispersants include carboxylic acids comprised of hydroxyl groups, long chain polyaminoamide salts with high molecular weight acid esters, high molecular weight polycarboxylic acid salts (Na and NH_4), alkyl pyrrolidones, the reactive monomer, high molecular weight copolymers, styrene acrylates, modified polyacrylates, polyvalent aliphatic carboxylic acids, naphthalenesulfonic acid/formalin condensates, polyoxyethylene alkylphosphoric esters, polyvinyl pyrrolidones, copolymers of vinyl pyrrolidone/vinyl acetates, alkylated polyvinyl pyrrolidones (alkylated with C4, C12, C20, C30, and the like), poly(maleic anhydride-co-methyl vinylether), poly(maleic anhydride-co-acrylic acid), copolymers of maleic anhydride, poly(tetrahydrofuran), Solsperse® dispersents (Zeneca), Zetasperse® Z-2100 and Z-2300 additives (Air Products), Surfynol® surfactants (104, 111, 121, 131, 136, 171, and 231(Air

Products)), Tamol™ dispersant (731 and 1124(Rohm and Haas)), Troysperse® 90W dispersant (Troy), AMP-95™ 2-amino-2-methyl-1-propanol solution (ANGUS Chemie GmbH), BYK®-346 additive (BYK), and the like. Examples of blue colored dyes include Acid Blue 1, Basic Blue 1 and C.I. Solvent Blue 7 and the like.

[0052] Examples of red colored dyes include Acid Red 18, Basic Red 1 and C.I. Solvent Red 8 and the like.

[0053] Examples of green colored dyes include Acid Green 1 and Basic Green 1 and the like.

[0054] Examples of black colored dyes include C.I. Solvent Black 5 and the like.

[0055] In one embodiment, an oil, such as a drying or non-drying oil, may be included in the compositions. Examples of drying oils include glycerides or triglycerides of fatty acids. Drying oils are characterized by the presence of unsaturated –CH=CH- groups. Additional examples of oils include oils of linseed, tung, oiticica, dehydrated castor, fish, and soya bean. Examples of non-drying oils include mineral, castor, and petroleum distillates.

[0056] In one embodiment, a resin, or high molecular weight polymer, may be included in the compositions. Examples of suitable resins include, acrylic polymers, polyvinylbutyral, polyurethanes, polyisocyanates, polyamides, polyesters, epoxies and polyepoxides, polyphenols, polycarbonates, polyvinylformal, shellac, vinylic, rubber based, waxes rosin, maleic resin and esters, manila copal, asphalts, starch and dextrin, gum Arabic, rosin modified phenolics, alkyds, terpenes, polystyrene, styrenic copolymers, styrene acrylates, silicone resins, alkylated urea formaldehyde resins, alkylated melamine formaldehyde resins, polyimides, poly(amide-imide) resins, chlorinated rubber, cyclized rubber, polyvinyl acetates, polyvinyl alcohols, alkylated polyvinyl alcohols, ketones resins, nitrocelluloses, ethyl cellulose, ethyl hydroxyethyl cellulose, cellulose acetate propionate, cellulose acetate butyrate, sodium carboxymethyl cellulose, polyethylene glycols and the like.

[0057] In one embodiment, a solvent may be included in the compositions. Suitable solvents can be identified in the *Industrial Solvents Handbook*, 4ed. edited by E.W. Flick (Noyes Daya Corp, Park Ridge, NJ, **1991**). Additional insight to solvent selection is also available in the *Polymer Handbook*, 4ed. edited by J. Brandrup, E.H. Immergut, and E.A. Grulke (John Wiley, New York, **1999**), which also describe *Solubility Parameters Values* by E.A. Grulke. These references are understood to be incorporated herein by reference in their entirety.

[0058] Examples of suitable solvents include hydrocarbon solvents (i.e., white spirit and paraffin oils, low and high boiling), aromatic hydrocarbons (toluene, xylene, paraffins, and naphthenes), alcohols (ethanol, n-propyl, isopropyl, n-butyl), alicyclic alcohols (cyclohexanol), glycols (monoethylene, monopropylene, hexylene, diethylene, dipropylene, triethylene), glycerin, ketones (acetone, butan-2-one, hexone, sexton, isophorone, diacetone alcohol), esters (ethyl acetate, isopropyl acetate, n-butyl acetate), n-methyl-2-pyrrolidone, γ-butyrolactone and the like.

[0059] In one embodiment, a plasticizer may be included in the composition. Examples of suitable plasticizers include abietates, adipates, alkyl pyrrolidones, alkylated caprolactams, benzoates, butyrates, citrates, epoxidized compounds, phthalates, polyester, polyol esters, ricinoleates, sebacates, stearates, and sulphonamides. Additional information regarding plasticizers can be found in the National Printing Ink Research Institute (NPIRI) "Raw Materials Data Handbook" (Volume 2). Specific examples include triethyl citrate, epoxidized soya bean oils, dimethyl phthalate, glyceryl triacetate, butyl ricinoleate, butyl stearate, n-octyl-2-pyrrolidone, n-dodecyl-2-pyrrolidone, n-cocoyl-2-pyrrolidone, n-hydrogenated tallowyl-2-pyrrolidone and the like.

[0060] In one embodiment, a wax may be included in the composition. Examples of suitable waxes include polyethylene, polytetrafluoroethylene, fatty acid amides (i.e., stearamide), petroleum (i.e., paraffins, slack, scale, jelly, microcrystalline, ceresin, montan, montan esters), beeswax, carnauba, shellac, Japan, candelilla, lanolin, alkylated polyvinyl pyrrolidones (alkylated with C4, C12, C20, C30, and the like), and the like.

[0061] In one embodiment, a drier may be included in the compositions. Examples of suitable driers include oil soluble soaps (formed from octoates, resonates, naphthenates, tallates, linoleates), cobalt, cobalt acetate, manganese, cerium, zirconium, lithium, calcium, zinc, lead acetate, manganese borate and the like.

[0062] In one embodiment, a chelating agent may be included in the compositions. Examples of suitable chelating agents include ethylenediaminetetra-acetic acid and sodium salts, nitrilotriacetic acid salts, sodium salts of diethylenetriamine-acetic acid, heptonates, alkanolamines, dimethyl glyoxime and the like.

[0063] In one embodiment, an anti-oxidant may be included in the compositions. Examples of suitable anti-oxidants include eugenol, hydroquinone, pyrocatechol, guaiacol, butylated

hydroxytoluene, butylated hydroxyanisole, methyl ethyl ketoxime, butylaldoxime, cyclohexanone oxime and the like.

[0064] In one embodiment, a surfactant may be included in the compositions.. Surfactants can also be employed in the presence of defoaming agents such as polydimethyl siloxanes and derivatives thereof. Examples of suitable surfactants include anionic (i.e., alkali metal soaps, ammonium and ammonium salts of long chain fatty acids), cationic (i.e., quaternary fatty ammonium halides, acetates, or suphates), non-ionic (i.e., polyethylene oxide chains attached to hydrocarbons), amphoteric and the like.

[0065] In one embodiment, deodorants and fragrances may be included in the compositions. Examples of suitable deodorants and fragrances include amyl and methyl salicylate, vanillin, citron, cedarwood, peppermint, lavender, carnation and the like.

[0066] In one embodiment, adhesion promoters may be included in the compositions. Examples of suitable adhesion promoters include titanium acetyl-acetonate, polyfunctional aziridines, polyethylene imines, chlorinated polyolefins, pentahydroxy(tetradecanoato) di-chromium, octadecanoato chromic chloride hydroxide, glycidoxy (epoxy) functional methoxy silane, β-(3,4-epoxycyclohexyl) ethyltriethoxysilane and the like.

[0067] In one embodiment, inhibitors may be included in the compositions. Examples of suitable inhibitors include hydroquinone, hydroquinone monomethyl ether, hydroquinone monopropyl ether, hydroquinone monobenzyl ether, amyl quinine, amyloxyhydroquinone, n-butylphenol, phenol, 4-methoxyphenol (MEHQ), phenothiazine, nitrobenzene and phenolic-thio compounds, alone or in combination thereof.

[0068] In one embodiment, laking agents may be included in the compositions. Examples of suitable laking agents include tannic acid and derivatives, shellac, maleic acids and the like.

[0069] In one embodiment, silica may be included in the compositions. Examples of suitable silicas include fumed, precipitated, gel, colloidal and the like.

[0070] In another embodiment, a stabilizer to inhibit the premature cross-linking may be included in the compositions. The stabilizers may include, but are not limited to hydroquinone, hydroquinone monomethyl ether, hydroquinone monopropyl ether, hydroquinone monobenzyl ether, amyl quinine, amyloxyhydroquinone, n-butylphenol, phenol, 4-methoxyphenol (MEHQ), phenothiazine, nitrobenzene and phenolic-thio compounds, alone or in combination thereof.

[0071] In one aspect, the compositions may also include one or more additives in conventional quantities which may provide enhanced or altered properties to the compositions. These additives may be, but are not limited to, slip modifiers, thixotropic agents, laponites, flow or rheology control agents, UV-light absorbers, fungicides, bactericides, organic/inorganic filler particles (i.e., clays, kaolins), leveling agents, antistatic agents, viscosity modifier, therapeutic and/or preventive medicaments, and other ingredients apparent to those skilled in the art.

[0072] The compositions disclosed herein may include the reactive monomer as about 1 to about 50 % by weight of the composition and the co-solvent as about 4 to about 80 % by weight of the composition. Additionally, the compositions may include an initiator such as a photoinitiator or a polymerization initiator such as a free-radical initiator, a cationic initiator, an anionic initiator, or a thermal initiator as about 1 to about 10 % by weight of the composition. If a colorant is included in the composition it may be present as about 1 to about 50% by weight of the composition. Other additives or components may be present in the composition as about 0.1 to about 60% by weight of the composition.

[0073] According to one important embodiment of the invention, the reactive solution comprising polymerizable component produced by curing through any of the above said method and further formulated as composition to employ in various applications such as industrial, personal care, household and pharmaceuticals. Exemplary and non-limiting applications of the proposed compositions are essentially in the field of coating-UV curable, newspaper inks, packaging inks, lithographic inks, offset inks, gravure inks and plates, flexographic inks and plates, screen inks, ink-jet inks, RFID devices, adhesive inter-layers, adhesion promoters, substrate penetrants, varnishes, labels, food wrappers, labels and colors for toys, labels and colors for pencils, labels and colors for comics, inks for postal application, inks for monetary application, inks for official government documents, over print varnish, visual identification, security inks, packaging, shrink wraps, container sleeves, metal inks and coatings, anti-fog surfaces. Wherein, the composition is produced as solid, liquid or powder or in a solution form. These formulation may be applied to various surfaces as applicable to their intended use, for example, to metal such as steel, iron, copper, brass, gold, silver, and aluminum; to plastic such as vinyl, polyolefins such as polyethylene and polypropylene, Tyvec, polyester, PVDC, and nylon; to glass; and to a textile.

[0074] The compositions disclosed herein may be characterized as being conductive, metallic, pearlescent, fluorescent, and/or as exhibiting or having a thermal transition or phase change.

[0075] In one embodiment, the compositions may optionally include carbon nanostructures such nano-onions, horns, tubes, rods, wires, cones, dots, whiskers, filaments, nano-diamond, and graphene sheets. In another embodiment, the compositions disclosed herein may optionally include quantum dots.

[0076] The present invention is illustrated in detail by way of the below given examples. The examples are given herein for illustration of the invention and are not intended to be limiting thereof.

Example 1: UV Curing Ink Base

Material	Supplier	Mass (g)
N-Vinyl Pyrrolidone	ISP	9.875
4-butylphenyl maleimide	Reactive monomer	9.875
Irgacure® 184	Ciba	3.0
PETA-4 (SR295)	Sartomer	76.75
DABCO® DC193 surfactant	Dow Corning	0.5
Total		100

Example 2: Offset Lithographic Ink

[0077] Based on Leach and Pierce's Printing Ink Manual (Kluwer, Boston, 1999) (the contents of which are hereby incorporated by reference) as a guide for the production of a standard dry offset lithographic ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	w/w%
Pigment	18.0
Acrylate prepolymer	30.0
Modifying hard resin	25.0
4-butylphenyl maleimide (Reactive monomer)	16.0
Photo-initiator and amine synergist	9.0
Polyethylene wax	1.0
Silicone fluid	1.0

[0078] U.S. Patent No. 7,232,851 and WO 2003/014239 (the contents of which are hereby incorporated by reference) may also be used as a guide for the production of a lithographic inks.

Example 3: Flexographic Ink

[0079] Based on U.S. Patent No. 7,291,658 B2 (the contents of which are hereby incorporated by reference) as a guide for the production of a standard white flexographic ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	Description	Parts (w/w)
Pigment	R-706	40.0
Ashland D-30R	Ashland resin for grinding pigments	17.5
Ashland F-126R	Ashland resin	28.0
4-butylphenyl maleimide	Reactive Monomer	10.0
LG-37	Reactive defoaming agent	1.0
BYK® 019	Silicone defoamer	0.5
ViaCure® LX	UCB photoinitiator vehicle for light ink applications	3

Example 4: Letterpress Ink

[0080] Based on U.S. Patent No. 6,620,227 (B1) (the contents of which are hereby incorporated by reference) as a guide for the production of a CF (coated front) UV ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	Description	Parts (w/w)
KC 98-1410 UV from Kohl & Madden Ink	UV curable ink base	25.17
Bis-(3-allyl-4-hydroxy phenyl) sulfone	Acidic color developer	50.34
4-butylphenyl maleimide	Reactive monomer and Reactive solvent	23.49
Darocure® 4265 from Ciba	Photo-initiator	1.00

Example 5: Screen Printing Ink

[0081] Based on U.S. Patent No. 7,291,658 B2 (the contents of which are hereby incorporated by reference) as a guide for the production of a standard white screen-ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	Description	Parts (w/w)
Pigment	R-706	41.2
ViaScreen 515	UCB Vehicle for screen printing applications	30.1

Ebecryl 110	2-Phenoxyethyl acrylate	18.8
4-butylphenyl maleimide	Reactive Monomer	7.21
LG-37	Reactive defoaming agent	1.0
BYK® 019	Silicone defoamer	0.5
ViaCure® LX	UCB photoinitiator vehicle for ink applications	6

[0082] Additional teachings can be found in U.S. Patent No. 5,395,863 (the contents of which are hereby incorporated by reference).

Example 6: Screen Printing Ink

[0083] Based on U.S. Patent No. 4,418,138 A (the contents of which are hereby incorporated by reference) as a guide for the production of a standard black screen-ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	Parts (w/w)
Catofor 06	1.0
2-(carboxymethoxy)thioxanthone	0.5
Ethanol	5.0
Polyethylene glycol 200 diacrylate	10.5
4-butylphenyl maleimide (reactive monomer)	4.5
Uvecryl P101	2.0
20% Gohsenol KP08 solution	50.0
Anthrasol Blu-Black IRD	0.5
Polyethylene glycol 200	1.0

Example 7: Ink-jet Printing Ink

[0084] Based on WO 2007/036692 (A1) (the contents of which are hereby incorporated by reference) as a guide for the production of a standard UV inkjet ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	**Description**	**Parts (w/w)**
Polyethylene glycol 200 diacrylate	oligomer	9
Ethoxylated (20) trimethylolpropane	monomer	4.5

triacrylate		
4-butylphenyl maleimide	Reactive monomer	24.8
Water		44.9
Cab-O-Jet® 300	Carbon black pigment with carboxylate surface treatment for water application	12.5
Irgacure® 2959	Photo-initiator	4
FC4430	fluorosurfactant	0.2

Example 8: Electrophotographic Printing Ink

[0085] Based on U.S. Patent No. 5,332,644 A (the contents of which are hereby incorporated by reference) as a guide for the production of a standard electrophotographic coating, a modified formulation employing the reactive monomer was designed and is presented below.

Raw Material	**Parts (w/w)**
Benzimidazole perylene (BZP)	9.25
4-butylphenyl maleimide (reactive monomer)	36.3
Styrene	54.45

Example 9: Intaglio Printing Ink

[0086] Based on U.S. Patent No. 6,787,583 (B2) (the contents of which are hereby incorporated by reference) as a guide for the production of a intaglio UV ink, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	**Parts (w/w)**
Ebecryl 2002	46.6
Montan Wax	4.0
4-butylphenyl maleimide (Reactive Monomer)	1.5
Emulsifier	1.5
UV Stabilizer	2.0
Igraliet® Red 8B	8.0
$CaCO_3$	30.0
Esacure ITX	2.6

Irgacure® 369	3.8

Example 10: Printing Plate

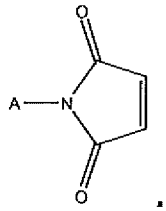
[0087] Based on U.S. Patent No. 4,011,084 (A) (the contents of which are hereby incorporated by reference) as a guide for the production of a UV curable printing plate, a modified formulation employing the reactive monomer was designed and is presented below:

Raw Material	**Parts (w/w)**
Polyurethane/ester diacrylate	75
4-butylphenyl maleimide (Reactive monomer)	25
Benzoin methyl ether	1
Potassium salt of N-nitrosocyclohexylhydroxylamine	0.05

[0088] While the foregoing written description of the invention enables one of ordinary skill to make and use the compositions described herein, those of ordinary skill will understand and appreciate the existence of variations, combinations, and equivalents of the specific embodiment, method, and examples herein. The invention should therefore not be limited by the above described embodiment, method, and examples.

[0089] What is claimed:

1. An ink composition comprising at least one reactive monomer that includes a maleimide of the general formula (1)



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2. The composition of claim 1 wherein A— is or includes an aryl, an alkylaryl, or an arylalkyl.

3. The composition of claim 2 wherein A— includes phenyl or a phenyl substituted with one or more halogens, alkyl or fluoroalkyl groups which may be straight chain, branched or cyclic, alkoxy groups, nitro groups, and carboxy groups.

4. The composition of claim 3 wherein the phenyl is substituted with an alkyl having one to twenty carbons.

5. The composition of claim 1 wherein A— is 4-butylphenyl.

6. The composition of claim 1, further comprising a reactive co-solvent.

7. The composition of claim 6, wherein the reactive co-solvent includes at least one of a (meth)acryl monomer or prepolymer, a (meth)acryl ester of an epoxy type monomer or prepolymer, and a urethane type monomer or prepolymer.

8. The composition of claim 7, wherein the reactive co-solvent is 2-phenoxy ethyl acrylate, polyethylene glycol (200) diacrylate, pentaerythritol tetraacrylate, a urethane diacrylate oligomer, a urethane acrylate, or styrene.

9. The composition of claim 1 further comprising a photoinitiator, a free-radical initiator, a cationic initiator, an anionic initiator, or a thermal initiator.

10. The composition of claim 9 further comprising at least one of a colorant, a dispersant, oil, a resin, a solvent, a plasticizer, a wax, a drier, a chelating agent, an anti-oxidant, a surfactant, a deodorant, a fragrance, an adhesion promoter, an inhibitor, a laking agent, silica, a stabilizer and a defoaming agent.

11. The composition of claim 10 comprising a colorant that includes a carbon pigment.

12. The composition of claim 1, wherein the ink composition is a lithographic ink, a flexographic ink, a gravure ink, a letterpress ink, a screening printing ink, ink-jet printing ink, an electrophotographic ink, an intaglio printing ink, or a collotype printing ink.

13. A method of printing or coating a substrate comprising:

applying an ink or coating composition to a substrate wherein the ink or coating composition comprises at least one reactive monomer comprising a maleimide group.

14. The method of claim 13, wherein the ink or coating composition further comprises a reactive co-solvent, wherein the reactive co-solvent includes at least one of a (meth)acryl monomer or prepolymer, a (meth)acryl ester of an epoxy type monomer or prepolymer, and a urethane type monomer or prepolymer.

15. The method of claim 14, wherein the reactive co-solvent is 2-phenoxy ethyl acrylate, polyethylene glycol (200) diacrylate, pentaerythritol tetraacrylate, a urethane diacrylate oligomer, a urethane acrylate, or styrene.

16. The method of claim 13, wherein the reactive monomer includes 4-butylphenyl maleimide.

17. The method of claim 13, wherein the ink or coating composition further comprises a curing initiator, wherein the curing initiator is a photoinitiator, a free-radical initiator, a cationic initiator, an anionic initiator, or a thermal initiator.

18. The method of claim 13 further comprising at least one of a colorant, a dispersant, oil, a resin, a solvent, a plasticizer, a wax, a drier, a chelating agent, an anti-oxidant, a surfactant, a deodorant, a fragrance, an adhesion promoter, an inhibitor, a laking agent , silica, a stabilizer and a defoaming agent.

19. The method of claim 13, wherein the composition is a lithographic ink, a flexographic ink, a gravure ink, a letterpress ink, a screening printing ink, ink-jet printing ink, an electrophotographic ink, an intaglio printing ink, or a collotype printing ink.

20. The method of any one of claims 13 - 19 wherein the substrate is selected from the group consisting of paper, metal, plastic, glass, textile and combinations thereof.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US 11/20622

A. CLASSIFICATION OF SUBJECT MATTER
IPC(8) - C09D 11/00 (2011.01)
USPC - 523/160,161
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC(8) - C09D 11/00 (2011.01)
USPC - 523/160,161

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
IPC(8) - C09D 11/02 (2011.01)
USPC - 106/31.13, 31.28

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
PubWest (PGPB,USPT,USOC,EPAB,JPAB); Google
Search Terms Used:
bismaleimide, 4-butylphenylmaleimide, phenylmaleimide ink, carbon pigment, phenylmaleimide styrene copolymer, wood coating

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X --- Y	US 6,555,593 B1 (HOYLE et al.) 29 April 2003 (29.04.2003), col 6, 7, 13	1-10, 12-20 --- 11
Y	US 2009/0090271 A1 (WYNANTS et al.) 09 April 2009 (09.04.2009), para [0077]	11

☐ Further documents are listed in the continuation of Box C. ☐

* Special categories of cited documents:
"A" document defining the general state of the art which is not considered to be of particular relevance
"E" earlier application or patent but published on or after the international filing date
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
"O" document referring to an oral disclosure, use, exhibition or other means
"P" document published prior to the international filing date but later than the priority date claimed
"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"&" document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
12 February 2011 (12.02.2011)	08 MAR 2011

Name and mailing address of the ISA/US	Authorized officer:
Mail Stop PCT, Attn: ISA/US, Commissioner for Patents P.O. Box 1450, Alexandria, Virginia 22313-1450 Facsimile No. 571-273-3201	Lee W. Young PCT Helpdesk: 571-272-4300 PCT OSP: 571-272-7774