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(54) Title: HAIR CLEANSING COMPOSITION

(57) Abstract: A shampoo composition is disclosed that has from about 5 to about 40 weight % of one or more detersive surfactants and an aqueous carrier comprising water and one or more polyols, wherein the weight ratio of one or more polyols to water is higher than or equal to about 0.4, and the water content is higher than or equal to about 9% by weight of the composition. The composition delivers thorough hair cleansing, good lather, good conditioning wet feel, clean rinse feel during use, leaving hair with very low residue on the hair surface. The use of the shampoo composition also provides high hair volume, smooth dry hair feel, and bouncy hair that is easy to style.

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HAIR CLEANSING COMPOSITION

FIELD OF THE INVENTION

The present invention relates to a deep cleansing hair care composition and method that provide good in-use experience and superior hair benefits when dry.

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BACKGROUND OF THE INVENTION

Typically, clarifying shampoos are employed to provide a high level of cleaning for hair, good lather and clean rinse feel. However, such shampoos are perceived to strip the natural moisturizers from the hair fibers, as they result in high wet hair friction that is translated into poor wet feel and high dry hair friction that is translated into poor, non-moisturized dry hair feel. The poor wet feel and high dry hair friction are particularly noticeable in the case of consumers that have chemically or physically damaged hair resulting from permanent or semi-permanent styling treatments, oxidative coloring treatments, thermal treatments, etc. To remedy the poor wet feel and the high dry hair friction, consumers often turn to traditional conditioning shampoos which can provide good wet and dry hair feel. However, many of these products do not possess strong cleansing ability, they generate lower volume of lather during use, they do not provide clean rinse feel and they can allow more soils, sebum and other residues to remain on the hair after shampooing. In addition, the conditioning agents themselves can contribute to the feeling of residue that is added on the hair surface during the shampoo and/or conditioner process that they may accumulate in each cycle. This can result in hair weigh-down with low volume that is perceived as greasy, and that may also be perceived by the consumer as lanky, non-bouncy and difficult to style. Thus, there is a need for the development of shampoo compositions that provide strong cleansing ability and at the same time contribute to high hair volume, good wet and dry hair feel, and bouncy, easy to style hair.

Described herein is a deep cleansing hair care composition that enables a thorough removal of soils such as sebum, conditioning, styling and other residues from hair surface in order to achieve clean hair with high hair volume, smooth dry hair feel and elastic hair fibers that impart bouncy hair. In addition, the hair care composition described herein results in an excellent in-use experience by providing rich lather, good wet conditioning feel, and a clean rinse feel. Additionally, the hair care composition is stable and uniform and can be clear, translucent or opaque.

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SUMMARY OF THE INVENTION

The invention relates to a hair care composition comprising: from about 5 to about 40 weight % of the hair care composition one or more detersive surfactants; wherein, the detersive surfactant comprises less than about 10 weight % of the hair care composition of a linear anionic surfactant selected from the group consisting of:

(1) alkyl sulfates

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$$R - O - S - O M$$

Alkyl sulfates

where R is a linear C₈-C₂₄ alkyl and M⁺ is monovalent cation;

(2) alkyl ether sulfates

$$\mathsf{R} - \mathsf{O} - \mathsf{CH}_2 - \mathsf{CH}_2 - \mathsf{O} + \mathsf{S} - \mathsf{O} \oplus \mathsf{M}$$

Alkyl ether sulfates

where R is a linear C₈-C₂₄ alkyl, n=1-2, and M⁺ is monovalent cation;

(3) and mixture thereof,

and a carrier comprising one or more polyols and water, wherein the carrier comprises from about 20 to about 80 weight % of the hair care composition of one or more polyols, and from about 9% to about 75% by weight of the hair care composition of water; and wherein the weight ratio of one or more polyols to water is 0.4 or higher.

DETAILED DESCRIPTION OF THE INVENTION

While the specification concludes with claims which particularly point out and distinctly claim the invention, it is believed that the provided invention will be better understood from the following description.

In all embodiments of the present invention, all percentages and ratios used herein are by weight of the total composition, unless otherwise designated. All measurements are understood to be made at ambient conditions, where "ambient conditions" means conditions at about 25 °C, under about one atmosphere of pressure, and at about 50% relative humidity (RH), unless otherwise designated. All numeric ranges are inclusive of narrower ranges; delineated upper and lower range limits are combinable to create further ranges not explicitly delineated. All numerical amounts are understood to be modified by the word "about" unless otherwise specifically indicated. All weights and % weights as they pertain to listed ingredients are based on the active level and do not include carriers or by-products that may be included in commercially available materials, unless otherwise specified. The number of significant digits conveys neither a limitation on the indicated amounts nor on the accuracy of the measurements. The term "molecular weight" or "M.Wt." as used herein refers to the weight average molecular weight unless otherwise stated. The weight average molecular weight may be measured by gel permeation chromatography. "QS" means sufficient quantity for 100%.

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DEFINITIONS

"Dermatologically acceptable" means that the compositions or components described are suitable for use in contact with human skin tissue without undue toxicity, incompatibility, instability, allergic response, and the like.

"Safe and effective amount" means an amount of a compound or composition sufficient to significantly induce a positive benefit.

"Soluble" means at least about $0.1~{\rm g}$ of solute dissolves in $100~{\rm ml}$ of solvent, at $25~{\rm ^{\circ}C}$ and $1~{\rm atm}$ of pressure.

The term "substantially free from" or "substantially free of" as used herein means less than about 1%, or less than about 0.8%, or less than about 0.5%, or less than about 0.3%, or about 0%, or from about 1% to 0%, or about 0.8% to 0%, or about 0.3% to 0% by total weight of the composition.

"Hair," as used herein, means mammalian hair including scalp hair, facial hair and body hair, particularly hair on the human head and scalp.

"Cosmetically acceptable," as used herein, means that the compositions, formulations or components described are suitable for use in contact with human keratinous tissue without undue toxicity, incompatibility, instability, allergic response, and the like. All compositions described

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herein which have the purpose of being directly applied to keratinous tissue are limited to those being cosmetically acceptable.

"Derivatives," as used herein, includes but is not limited to, amide, ether, ester, amino, carboxyl, acetyl, acid, salt and/or alcohol derivatives of a given compound.

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"Polymer," as used herein, means a chemical formed from the polymerisation of two or more monomers. The term "polymer" as used herein shall include all materials made by the polymerisation of monomers as well as natural polymers. Polymers made from only one type of monomer are called homopolymers. Polymers made from two or more different types of monomers are called copolymers. The distribution of the different monomers can be calculated statistically or block-wise – both possibilities are suitable for the present invention. Except if stated otherwise, the term "polymer" used herein includes any type of polymer including homopolymers and copolymers.

SHAMPOO COMPOSITION

Typical clarifying shampoos, which contain minimal conditioning agents, provide superior cleansing via removal of soils, sebum and other residues. However, they are perceived as hair stripping, with an in-use wet feel that does not provide the signal of conditioning. In addition, the hair after washing with clarifying shampoo is not smooth. The hair care compositions described herein which contain a carrier, a significant portion (about 20% to about 80%) of which is a watermiscible polyol, such as glycerin, are able to minimize undesirable hair stripping. These compositions not only provide superior removal of soils, but also show excellent lather quality and clean rinse feel. The polyols may contribute to superior cleansing by reducing the size of the surfactant micelle structure. This, in turn, can result in rapid transfer of the micelles on hair fiber surfaces, and rapid removal of soils from these surfaces. It is surprisingly observed that the lather created during the cleansing process of the inventive compositions is creamy and resilient, providing slippery, conditioning feel. The superior cleansing is associated with volume expansion of the dry hair. It is also surprising that the inventive compositions show larger hair expansion than traditional clarifying shampoos (see results in Table 1 for Ex 1 to Ex 3 versus Comparative Ex 1 and Comparative Ex 2). Additionally, the friction of dry hair that has been washed by the inventive compositions is below the expected level (resulting in hair smooth feel), as the corresponding surfaces are substantially free from oils, sebum and other hydrophobic materials. This may be the result of the plasticizing effect of the polyol (such as glycerin) on keratin protein and/or the expected

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higher water absorption of moisture in hair fibers because of the presence of residual amount of glycerin.

Described herein is a shampoo composition comprising a detersive surfactant, and an aqueous carrier comprising water and polyol, wherein the weight ratio of polyol to water is from about 0.4 or higher, wherein the polyol content is from about 20 to 80 weight % of the composition, and wherein the water content is from about 9% to about 75% by weight of the composition.

The hair care composition may comprise from about 4 to about 40 weight % of one or more detersive surfactant. The hair care composition may comprise from about 0 to about 10 weight percent of alkyl sulfates with linear alkyl group. An alkyl sulfate having a linear alkyl group is a sulfate with an alkyl group with a formula of CH3(CH2)n-, wherein each carbon is bound to two neighbors and two hydrogen atoms, except the terminal carbon which is bound to three hydrogen atoms

The hair care composition may further comprise a cationic polymer.

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The shampoo composition delivers consumer desired cleansing power (low residues on hair surface) in addition to good lather, good conditioning wet feel, clean rinse feel, high hair volume, smooth non-coated dry hair feel, and moisturized bouncy hair that is easy to style.

Water is a small molecule that moves in and out of damaged hair cuticles depending on the environment change. It can weaken hair fibers when wet, it can make hair fibers brittle when dry and it can cause hair color to fade by carrying color pigment out of hair. Polyols such as glycerin adds elasticity to damaged hair fibers and results in hair that is less sensitive to environment changes.

Shampoos with high levels of polyol such as glycerin can also affect surfactant packing. The micelles in the hair care composition are more segregated and therefore transport onto the hair surface quicker which can result in improved soil removal.

Typically, the viscosity of an aqueous solution is increased when the glycerin content is increased (Segur, J.B. and Oberstar, H., Ind. & Eng. Chem., 43: 2117-2120, 1951). However, the viscosity of the hair care composition of the present invention decrease when glycerin content is increased in the glycerin and water mixture carrier. Consumers traditionally prefer products with a viscosity of about 1000 cps to about 15,000 cps which result in enough viscosity so that they do not immediately run out of the hand after dispensing; however, too viscous products are difficult to spread onto the hair.

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Consumer testing indicates that the hair care compositions described herein deliver soft and moisturized feel on damaged hair. Additionally, consumer testing indicates the hair care compositions described herein deliver one or more consumer benefits including, but not limited to rich and creamy lather, deep clean, scalp comfort, volume and shine.

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A. DETERSIVE SURFACTANT

The shampoo composition comprises one or more detersive surfactants, which provides cleaning performance to the composition. The one or more detersive surfactants in turn may comprise anionic surfactants, amphoteric or zwitterionic surfactants, nonionic surfactants or mixtures thereof. Various examples and descriptions of detersive surfactants are set forth in U.S. Patent No. 6,649,155; U.S. Patent Application Publication No. 2008/0317698; and U.S. Patent Application Publication No. 2008/0206355, which are incorporated herein by reference in their entirety.

The concentration of the detersive surfactant component in the shampoo composition should be sufficient to provide the desired cleaning and lather performance. Generally, ranges are from about 5 wt% to about 40 wt%, from about 5 wt% to about 35 wt%, from about 8 wt% to about 35 wt%, from about 10 wt% to about 30 wt%, from about 15 wt% to about 25%, from about 10 wt% to about 20 wt% by weight of the composition.

The total surfactants can include, but are not limited to nonionic, amphoteric, branched anionic surfactants, linear anionic surfactants and combinations thereof.

1. Branched Anionic Surfactant

Suitable branched anionic surfactant, with a tail having an alkyl chain with 8 carbon atoms or higher, include, but are not limited to the following surfactants: sodium undecyl sulfate, sodium trideceth sulfate, sodium C8-13 alkyl sulfate, sodium C8-15 alkyl sulfate, sodium C8-18 alkyl sulfate, sodium C8-13 pareth sulfate, sodium C8-13 pareth-n sulfate, sodium C8-14 pareth-n sulfate, and combinations thereof. Other salts of all the aforementioned surfactants are useful, such as TEA, DEA, ammonia, potassium salts. Useful alkoxylates include the ethylene oxide, propylene oxide and EO/PO mixed alkoxylates. Phosphates, carboxylates and sulfonates prepared from branched alcohols are also useful anionic branched surfactants. Branched surfactants can be derived from synthetic alcohols such as the primary alcohols from the liquid hydrocarbons produced by Fischer-Tropsch condensed syngas, for example SafolTM 23 Alcohol available from

Sasol North America, Houston, Tex.; from synthetic alcohols such as NeodolTM 23 Alcohol available from Shell Chemicals, USA; from synthetically made alcohols such as those described in U.S. Pat. No. 6,335,312 issued to Coffindaffer, et al on Jan. 1, 2002. Suitable examples of alcohols are SafolTM 23 and NeodolTM 23. Suitable examples of alkoxylated alcohols are SafolTM 23-3 and NeodolTM 23-3. Sulfates can be prepared by conventional processes to high purity from a sulfur based SO3air stream process, chlorosulfonic acid process, sulfuric acid process, or Oleum process. Preparation via air stream in a falling film reactor is a preferred sulfation process. The anionic surfactant may also be STnS, wherein n can define average moles of ethoxylation. n can range from about 0 to about 3.5, from about 0.5 to about 3.5, from about 1.1 to about 3.

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2. Linear Anionic Surfactant

Suitable linear anionic detersive surfactant components for use in the composition herein include those which are known for use in hair care or other personal care shampoo compositions. The anionic detersive surfactant may be a combination of sodium lauryl sulfate and sodium laureth-n sulfate. Alternatively, the anionic detersive surfactant can be sodium laureth sulfate with an average of one mole ethoxylate. The concentration of the anionic surfactant component in the composition should be sufficient to provide the desired cleaning and lather performance.

Anionic surfactants suitable for use herein include alkyl sulfates and alkyl ether sulfates of the formula $ROSO_3M$ and $RO(C_2H_4O)_xSO_3M$, wherein R is alkyl or alkenyl of from about 8 to about 18 carbon atoms, x is 1 to 10, and M is a water-soluble cation such as ammonium, sodium, potassium, and triethanolamine cation or salts of the divalent magnesium ion with two anionic surfactant anions. The alkyl ether sulfates may be made as condensation products of ethylene oxide and monohydric alcohols having from about 8 to about 24 carbon atoms. The alcohols can be derived from fats such as coconut oil, palm oil, palm kernel oil, or tallow, or can be synthetic.

Table 1: Examples of Typical Alkyl Sulfates and Alky Ether Sulfates

Surfactant	Supplier	Activity	SLS	SLE1S	SLE2S	SLE3S	SLE>3S
Sodium Lauryl Sulfate	Stepan STEOL SLS	29% by weight	100	0	0	0	0
Sodium	Stepan	26% by	45.5	26.3	11.8	0.07	16.33

Laureth-1	STEOL SLES-	weight					
Sulfate	1						
Sodium Laureth-3 Sulfate	Stepan STEOL SLES- 3	28% by weight	18	16.7	12.6	12.4	40.30

Some non-limiting examples of linear anionic surfactants are:

Alkyl Sulfates

$$R - O - \begin{matrix} O \\ \parallel \\ S - O \end{matrix} \bigcirc \bigoplus_{M}$$

Alkyl sulfates

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where R is C_8 - C_{18} alkyl (linear, saturated or unsaturated) or mixtures thereof and M^+ is monovalent cation. Examples include sodium lauryl sulfate (where R is C_{12} alkyl and M^+ is Na^+), ammonium lauryl sulfate (where R is C_{12} alkyl and M^+ is NH_3^+), and sodium coco-sulfate (where R is coconut alkyl and M^+ is Na^+);

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Alkyl Ether Sulfates

$$\mathsf{R} - \mathsf{O} - \mathsf{CH}_2 - \mathsf{CH}_2 - \mathsf{O} + \mathsf{S} - \mathsf{O} \oplus \mathsf{M}$$

Alkyl ether sulfates

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where R is C_8 - C_{18} alkyl (linear, saturated or unsaturated) or mixtures thereof, n=1-12, and M^+ is monovalent cation. Examples include sodium laureth sulfate (where R is C_{12} alkyl and M^+ is Na^+ , n=1-

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3), ammonium laureth sulfate (where R is C_{12} alkyl, M^{+} is NH_{3}^{+} , n=1-3), and Sodium trideceth sulfate (where R is C_{13} alkyl, M^{+} is Na^{+} , and n=1-4);

Some non-limiting examples of sulfonate surfactants are:

5 Alkyl glyceryl ether sulfonates:

$$R \longrightarrow CH_2 \longrightarrow CH \longrightarrow CH_2 \longrightarrow S \longrightarrow O \longrightarrow M$$

$$OH \longrightarrow O$$

where $R = C_8 - C_{18}$ alkyl (linear, saturated or unsaturated) or mixtures thereof and M^+ = monovalent cation, such as Sodium Cocoglyceryl Ether Sulfonate (R = coco alkyl, $M^+ = Na^+$);

Alpha olefin sulfonates prepared by sulfonation of long chain alpha olefins. Alpha olefin sulfonates consist of mixtures of alkene sulfonates,

$$R-CH_2-CH=-CH-CH_2-S-O$$

where $R = C_4 - C_{18}$ alkyl or mixtures thereof and M^+ = monovalent cation;

15 Hydroxyalkyl sulfonates,

$$R-CH_2-CH-CH_2-CH_2-S-O$$
OH
OH

where $R = C_4 - C_{18}$ alkyl or mixtures thereof and $M^+ =$ monovalent cation. Examples include Sodium C12-14 Olefin Sulfonate ($R = C_8 - C_{10}$ alkyl, $M^+ = Na^+$) and Sodium C 14-16 Olefin Sulfonate ($R = C_{10} - C_{12}$ alkyl, $M^+ = Na^+$).

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Examples of additional anionic surfactants suitable for use herein include, but are not limited to, ammonium lauryl sulfate, ammonium laureth sulfate, triethylamine lauryl sulfate, triethylamine laureth

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sulfate, triethanolamine lauryl sulfate, triethanolamine laureth sulfate, monoethanolamine lauryl sulfate, diethanolamine laureth sulfate, lauric monoglyceride sodium sulfate, sodium lauryl sulfate, sodium laureth sulfate, potassium laureth sulfate, sodium lauryl sarcosinate, sodium lauryl sarcosinate, lauryl sarcosine, cocoyl sarcosine, ammonium cocoyl sulfate, ammonium lauroyl sulfate, sodium cocoyl sulfate, sodium lauroyl sulfate, potassium cocoyl sulfate, potassium lauryl sulfate, monoethanolamine cocoyl sulfate, sodium trideceth-1 sulfate, sulfate, sodium trideceth-2 sulfate, sodium trideceth-3 sulfate, sodium tridecyl sulfate, sodium methyl lauroyl taurate, sodium methyl cocoyl taurate, sodium lauroyl isethionate, sodium cocoyl isethionate, sodium laurethsulfosuccinate, sodium laurylsulfosuccinate, sodium tridecyl benzene sulfonate, sodium dodecyl benzene sulfonate, and mixtures thereof.

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Still other suitable anionic surfactants are the reaction products of fatty acids esterified with isethionic acid and neutralized with sodium hydroxide. Other similar anionic surfactants are described in U.S. Patent Nos. 2,486,921; 2,486,922; and 2,396,278, which are incorporated herein by reference in their entirety. If the hair care composition comprises an anionic linear alkyl sulfate/linear alkyl ether sulfate at a level exceeding about 10%, the hair care composition may further comprise from about 1% to about 15%, from about 2% to about 15%, from about 2% to about 10%, from about 2% to about 9%, from about 2% to about 8.5%, from about 2% to about 8% by weight of the hair care composition of an additional surfactant. The additional surfactant can be chosen from the group including amphoteric surfactant, zwitterionic surfactant, nonionic surfactant, and mixtures thereof. The ratio of anionic surfactant to additional surfactant is from about 0.3 to about 6; from about 0.3 to about 5.5, from about 0.3 to about 5. from about 0.5 to about 5.

Amphoteric detersive surfactants suitable for use in the shampoo composition include those surfactants broadly described as derivatives of aliphatic secondary and tertiary amines in which the aliphatic radical can be straight or branched chain and wherein one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic group such as carboxy, sulfonate, sulfate, phosphate, or phosphonate. Exemplary amphoteric detersive surfactants for use in the present shampoo composition include cocoamphoacetate, cocoamphodiacetate, lauroamphodiacetate, and mixtures thereof.

Zwitterionic detersive surfactants suitable for use in the shampoo composition include those surfactants broadly described as derivatives of aliphatic quaternaryammonium, phosphonium, and

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sulfonium compounds, in which the aliphatic radicals can be straight or branched chain, and wherein one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic group such as carboxy, sulfonate, sulfate, phosphate or phosphonate. In another embodiment, zwitterionics such as betaines are selected.

Non limiting examples of other anionic, zwitterionic, amphoteric or optional additional surfactants suitable for use in the shampoo composition are described in McCutcheon's, Emulsifiers and Detergents, 1989 Annual, published by M. C. Publishing Co., and U.S. Patent Nos. 3,929,678, 2,658,072; 2,438,091; 2,528,378, which are incorporated herein by reference in their entirety. Non limiting examples of suitable zwitterionic or amphoteric surfactants are described in U.S. Patent Nos. 5,104,646 and 5,106,609, which are incorporated herein by reference in their entirety.

Non-ionic detersive surfactants suitable for use in the shampoo composition are selected from the group consisting of: Cocamide, Cocamide Methyl MEA, Cocamide DEA, Cocamide MEA, Cocamide MIPA, Lauramide DEA, Lauramide MEA, Lauramide MIPA, Myristamide DEA, Myristamide MEA, PEG-20 Cocamide MEA, PEG-2 Cocamide, PEG-3 Cocamide, PEG-4 Cocamide, PEG-5 Cocamide, PEG-6 Cocamide, PEG-7 Cocamide, PEG-3 Lauramide, PEG-5 Lauramide, PEG-3 Oleamide, PPG-2 Cocamide, PPG-2 Hydroxyethyl Cocamide, and mixtures thereof.

B. AQUEOUS CARRIER

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The shampoo composition comprises an aqueous carrier which is a mixture of water and a polyol. Suitable polyols include glycerin or ethylene glycol, propylene glycol, di-propylene glycol and mixtures thereof. The polyol may be glycerin. The water content of the composition is from about 9 % to about 75%, from about 10% to about 65%, from about 10% to about 55%, from about 10% to about 35% by weight of the composition. The polyol content is about 20% to about 80%, from about 25% to about 75%, or from about 30% to about 70%, from about 35% to about 70% from about 40% to about 70% by weight of the total composition. The weight ratio of polyol to water is from about 0.4 to about 7, from about 0.4 to about 6.5, from about 0.4 to about 5.5, and from about 0.5 to about 5.5.

Suitable polyols also include water-miscible organic solvent such as propylene glycol, dipropylene glycol, and ethylene glycol.

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The aqueous carrier may also include lower alcohols. The lower alcohols useful herein are monohydric alcohols having 1 to 6 carbons, in one aspect, ethanol and isopropanol.

In one aspect, the carrier may comprise other water-miscible or immiscible solvents with minimal or no significant concentrations of organic solvent, except as otherwise incidentally incorporated into the composition as minor ingredients of other components.

C. CATIONIC POLYMER

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The shampoo composition comprises one or more a cationic polymers. The hair care composition can comprises from about 0.05 weight % to about 2 weight % of one or more cationic polymers, from about 0.05 weight % to about 1.5 weight % of one or more cationic polymers, alternatively from about 0.05 weight % to about 1 weight % of one or more cationic polymers.

The polymer can include at least one of (a) a cationic guar polymer, (b) a cationic non-guar galactomannan polymer, (c) a cationic starch polymer, (d) a cationic copolymer of acrylamide monomers and cationic monomers, (e) a synthetic cationic polymer, (f) a cationic cellulose polymer or (g) a mixture of such polymers. The molecular weight of the cationic polymer can be from about 1,000 to about 10,000,000 and its charge density can be between about 0.1 meq/g to about 7 meq/g. The molecular weight of the cationic polymer can be from about 500,000 to about 900,000, from about 500,000 to about 800,000, and/or from about 250,000 to about 800,000.

(a) Cationic guar polymer

According to an embodiment of the present invention, the dispersion composition comprises a cationic guar polymer, which is a cationically substituted galactomannan (guar) gum derivatives. Guar gum for use in preparing these guar gum derivatives is typically obtained as a naturally occurring material from the seeds of the guar plant. The guar molecule itself is a straight chain mannan, which is branched at regular intervals with single membered galactose units on alternative mannose units. The mannose units are linked to each other by means of $\beta(1-4)$ glycosidic linkages. The galactose branching arises by way of an $\alpha(1-6)$ linkage. Cationic derivatives of the guar gums are obtained by reaction between the hydroxyl groups of the polygalactomannan and reactive quaternary ammonium compounds.

The cationic guar polymer may be formed from quaternary ammonium compounds.

Suitable cationic guar polymers include cationic guar gum derivatives, such as guar hydroxypropyltrimonium chloride. In an embodiment, the cationic guar polymer is a guar

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hydroxypropyltrimonium chloride. Specific examples of guar hydroxypropyltrimonium chlorides include the Jaguar® series commercially available from Rhone-Poulenc Incorporated, for example Jaguar[®] C-500, commercially available from Rhodia. Jaguar[®] C-500 has a charge density of 0.8 meq/g and a M.Wt. of 500,000 g/mole. Jaguar® C-17, which has a cationic charge density of about 0.6 meg/g and a M.Wt. of about 2.2 million g/mol and is available from Rhodia Company. Jaguar® C 13S which has a M.Wt. of 2.2 million g/mol and a cationic charge density of about 0.8 meg/g (available from Rhodia Company). Other suitable guar hydroxypropyltrimonium chloride are: guar hydroxypropyltrimonium chloride which has a charge density of about 1.1 meg/g and a M.Wt. of about 500,000 g/mole is available from ASI, a charge density of about 1.5 meg/g and a M.Wt. of about 500,000 g/mole is available from ASI. Other suitable guar hydroxypropyltrimonium chloride are: Hi-Care 1000, which has a charge density of about 0.7 meq/g and a M.Wt. of about 600,000 g/mole and is available from Rhodia; N-Hance 3269 and N-Hance 3270, which has a charge density of about 0.7 meq/g and a M.Wt. of about 425,000 g/mole and is available from ASI; N-Hance 3196, which has a charge density of about 0.8 and a M. Wt. Of about 1,100,000 g/ mole and is available from ASI. AquaCat CG518 has a charge density of about 0.9 meg/g and a M.Wt. of about 50,000 g/mole and is available from ASI. BF-13, which is a borate (boron) free guar of charge density of about 1.1 meg/g and M. W.t of about 800,000 and BF-17, which is a borate (boron) free guar of charge density of about 1.7 meq/g and M. W.t of about 800,000 both available from ASI.

(b) Cationic Non-Guar Galactomannan Polymers

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The dispersion compositions of the present invention comprise a galactomannan polymer derivative having a mannose to galactose ratio of between 5:1 and 1:1 on a monomer to monomer basis, the galactomannan polymer derivative selected from the group consisting of a cationic galactomannan polymer derivative and an amphoteric galactomannan polymer derivative having a net positive charge. As used herein, the term "cationic galactomannan" refers to a galactomannan polymer to which a cationic group is added. The term "amphoteric galactomannan" refers to a galactomannan polymer to which a cationic group and an anionic group are added such that the polymer has a net positive charge.

Galactomannan polymers are present in the endosperm of seeds of the Leguminosae family. Galactomannan polymers are made up of a combination of mannose monomers and galactose monomers. The galactomannan molecule is a straight chain mannan branched at regular intervals with single membered galactose units on specific mannose units. The mannose units are linked to

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each other by means of β (1-4) glycosidic linkages. The galactose branching arises by way of an α (1-6) linkage. The ratio of mannose monomers to galactose monomers varies according to the species of the plant and also is affected by climate. Non Guar Galactomannan polymer derivatives of the present invention have a ratio of mannose to galactose of greater than 2:1 on a monomer to monomer basis. Suitable ratios of mannose to galactose can be greater than about 3:1, and the ratio of mannose to galactose can be greater than about 4:1. Analysis of mannose to galactose ratios is well known in the art and is typically based on the measurement of the galactose content.

The gum for use in preparing the non-guar galactomannan polymer derivatives is typically obtained as naturally occurring material such as seeds or beans from plants. Examples of various non-guar galactomannan polymers include but are not limited to Tara gum (3 parts mannose/1 part galactose), Locust bean or Carob (4 parts mannose/1 part galactose), and Cassia gum (5 parts mannose/1 part galactose).

In one embodiment of the present invention, the galactomannan polymer derivative is a cationic derivative of the non-guar galactomannan polymer, which is obtained by reaction between the hydroxyl groups of the polygalactomannan polymer and reactive quaternary ammonium compounds. Suitable quaternary ammonium compounds for use in forming the cationic galactomannan polymer derivatives include those conforming to the general formulas 1-5, as defined above.

In another embodiment of the invention, the galactomannan polymer derivative is an amphoteric galactomannan polymer derivative having a net positive charge, obtained when the cationic galactomannan polymer derivative further comprises an anionic group.

In one embodiment of the invention the cationic non-guar galactomannan has a ratio of mannose to galactose is greater than about 4:1. The dispersion compositions of the present invention may comprise a galactomannan polymer derivative by weight of the composition. In one embodiment of the present invention, the shampoo compositions comprise from about 0.05% to about 2%, by weight of the composition, of a galactomannan polymer derivative.

(c) Cationically Modified Starch Polymer

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The dispersion compositions of the present invention comprise water-soluble cationically modified starch polymers. As used herein, the term "cationically modified starch" refers to a starch to which a cationic group is added prior to degradation of the starch to a smaller molecular weight, or wherein a cationic group is added after modification of the starch to achieve a desired molecular

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weight. The definition of the term "cationically modified starch" also includes amphoterically modified starch. The term "amphoterically modified starch" refers to a starch hydrolysate to which a cationic group and an anionic group are added.

The dispersion compositions of the present invention comprise cationically modified starch polymers at a range of about 0.01% to about 10%, and/or from about 0.05% to about 5%, by weight of the composition.

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The cationically modified starch polymers disclosed herein have a percent of bound nitrogen of from about 0.5% to about 4%.

As used herein, the term "molecular weight" refers to the weight average molecular weight. The weight average molecular weight may be measured by gel permeation chromatography ("GPC") using a Waters 600E HPLC pump and Waters 717 auto-sampler equipped with a Polymer Laboratories PL Gel MIXED-A GPC column (Part Number 1110-6200, 600.times.7.5 mm, 20 um) at a column temperature of 55.degree. C. and at a flow rate of 1.0 ml/min (mobile phase consisting of Dimethylsulfoxide with 0.1% Lithium Bromide), and using a Wyatt DAWN EOS MALLS (multiangle laser light scattering detector) and Wyatt Optilab DSP (interferometric refractometer) detectors arranged in series (using a dn/dc of 0.066), all at detector temperatures of 50°C, with a method created by using a Polymer Laboratories narrow dispersed Polysaccharide standard (Mw=47,300), with an injection volume of 200 µl.

The dispersion compositions of the present invention include starch polymers that is chemically modified by the addition of amino and/or ammonium groups into the starch molecules. Non-limiting examples of these ammonium groups may include substituents such as hydroxypropyl trimmonium chloride, trimethylhydroxypropyl ammonium chloride, dimethylstearylhydroxypropyl ammonium chloride. See Solarek, D. B., Cationic Starches in Modified Starches: Properties and Uses, Wurzburg, O. B., Ed., CRC Press, Inc., Boca Raton, Fla. 1986, pp 113-125. The cationic groups may be added to the starch prior to degradation to a smaller molecular weight or the cationic groups may be added after such modification.

The cationically modified starch polymers of the present invention generally have a degree of substitution of a cationic group from about 0.1 to about 7. As used herein, the "degree of substitution" of the cationically modified starch polymers is an average measure of the number of hydroxyl groups on each anhydroglucose unit which is derivatized by substituent groups. Since each

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anhydroglucose unit has three potential hydroxyl groups available for substitution, the maximum possible degree of substitution is 3. The degree of substitution is expressed as the number of moles of substituent groups per mole of anhydroglucose unit, on a molar average basis. The degree of substitution may be determined using proton nuclear magnetic resonance spectroscopy (".sup.1H NMR") methods well known in the art. The source of starch before chemical modification can be chosen from a variety of sources such as tubers, legumes, cereal, and grains. Non-limiting examples of this source starch may include corn starch, wheat starch, rice starch, waxy corn starch, oat starch, cassaya starch, waxy barley, waxy rice starch, glutenous rice starch, sweet rice starch, amioca, potato starch, tapioca starch, oat starch, sago starch, sweet rice, or mixtures thereof.

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In one embodiment of the present invention, cationically modified starch polymers are selected from degraded cationic maize starch, cationic tapioca, cationic potato starch, and mixtures thereof. In another embodiment, cationically modified starch polymers are cationic corn starch and cationic tapioca.

The starch, prior to degradation or after modification to a smaller molecular weight, may comprise one or more additional modifications. For example, these modifications may include cross-linking, stabilization reactions, phosphorylations, and hydrolyzations. Stabilization reactions may include alkylation and esterification.

The cationically modified starch polymers in the present invention may be incorporated into the composition in the form of hydrolyzed starch (e.g., acid, enzyme, or alkaline degradation), oxidized starch (e.g., peroxide, peracid, hypochlorite, alkaline, or any other oxidizing agent), physically/mechanically degraded starch (e.g., via the thermo-mechanical energy input of the processing equipment), or combinations thereof.

An optimal form of the starch is one which is readily soluble in water and forms a substantially clear (% Transmittance.gtoreq.80 at 600 nm) solution in water. The transparency of the composition is measured by Ultra-Violet/Visible (UV/VIS) spectrophotometry, which determines the absorption or transmission of UV/VIS light by a sample, using a Gretag Macbeth Colorimeter Color i 5 according to the related instructions. A light wavelength of 600 nm has been shown to be adequate for characterizing the degree of clarity of cosmetic compositions.

Suitable cationically modified starch for use in compositions of the present invention is available from known starch suppliers. Also suitable for use in the present invention is nonionic modified starch that could be further derivatized to a cationically modified starch as is known in the

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art. Other suitable modified starch starting materials may be quaternized, as is known in the art, to produce the cationically modified starch polymer suitable for use in the invention.

(d) Cationic cellulose polymers

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Suitable cationic cellulose polymers are salts of hydroxyethyl cellulose reacted with trimethyl ammonium substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 10 and available from Dwo/ Amerchol Corp. (Edison, N.J., USA) in their Polymer LR, JR, and KG series of polymers. Other suitable types of cationic cellulose include the polymeric quaternary ammonium salts of hydroxyethyl cellulose reacted with lauryl dimethyl ammonium-substituted epoxide referred to in the industry (CTFA) as Polyquaternium 24. These materials are available from Dow/ Amerchol Corp. under the tradename Polymer LM-200. Other suitable types of cationic cellulose include the polymeric quaternary ammonium salts of hydroxyethyl cellulose reacted with lauryl dimethyl ammonium-substituted epoxide and trimethyl ammonium substituted epoxide referred to in the industry (CTFA) as Polyquaternium 67. These materials are available from Dow/ Amerchol Corp. under the tradename SoftCAT Polymer SL-5, SoftCAT Polymer SL-30, Polymer SL-60, Polymer SL-100, Polymer SK-L, Polymer SK-M, Polymer SK-MH, and Polymer SK-H.

D. Mechanical Foam Dispenser

The hair care composition can be delivered in a liquid or foam form. It may be delivered in a foam form via a mechanical foam dispenser. The mechanical foam dispenser described herein may be selected from the group consisting of squeeze foam dispensers, pump foam dispensers, other mechanical foam dispensers, and combinations thereof. In an embodiment, the mechanical foam dispenser is a squeeze foam dispenser. Non-limiting examples of suitable pump dispensers include those described in WO 2004/078903, WO 2004/078901, and WO 2005/078063 and may be supplied by Albea (60 Electric Ave., Thomaston, CT 06787 USA) or Rieke Packaging Systems (500 West Seventh St., Auburn, Indiana 46706).

The mechanical foam dispenser may comprise a reservoir for holding the hair care composition. The reservoir may be made out of any suitable material selected from the group consisting of plastic, metal, alloy, laminate, and combinations thereof. The reservoir may be a refillable reservoir such as a pour-in or screw-on reservoir, or the reservoir may be for one-time use. The reservoir may also be removable from the mechanical foam dispenser. Alternatively, the

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reservoir may be integrated with the mechanical foam dispenser. In an embodiment, there may be two or more reservoirs.

The reservoir may be comprised of a material selected from the group consisting of rigid materials, flexible materials, and combinations thereof. The reservoir may be comprised of a rigid material if it does not collapse under external atmospheric pressure when it is subject to an interior partial vacuum.

E. Aerosol Foam Dispenser

The hair care composition can be delivered in a liquid or foam form. It may be delivered in a foam form via an aerosol foam dispenser. The aerosol foam dispenser may comprise a reservoir for holding the hair care composition. The reservoir may be made out of any suitable material selected from the group consisting of plastic, metal, alloy, laminate, and combinations thereof. In an embodiment, the reservoir may be for one-time use. In an embodiment, the reservoir may be removable from the aerosol foam dispenser. Alternatively, the reservoir may be integrated with the aerosol foam dispenser. In an embodiment, there may be two or more reservoirs.

The reservoir may be comprised of a material selected from the group consisting of rigid materials, flexible materials, and combinations thereof. The reservoir may be comprised of a rigid material if it does not collapse under external atmospheric pressure when it is subject to an interior partial vacuum.

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F. Propellant

The hair care composition described herein may comprise from about 1% to about 10% propellant, alternatively from about 2% to about 9% propellant, and alternatively from about 3% to about 8% propellant, by weight of the hair care composition.

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The propellant may comprise one or more volatile materials, which in a gaseous state, may carry the other components of the hair care composition in particulate or droplet form. The propellant may have a boiling point within the range of from about -45° C. to about 5° C. The propellant may be liquefied when packaged in convention aerosol containers under pressure. The rapid boiling of the propellant upon leaving the aerosol foam dispenser may aid in the atomization of the other components of the hair care composition.

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Aerosol propellant which may be employed in the hair care composition may include the chemically-inert hydrocarbons such as propane, n-butane, isobutane, cyclopropane, and mixtures thereof, as well as halogenated hydrocarbons such as dichlorodifluoromethane, 1,1-dichloro-1,1,2,2-tetrafluoroethane, 1-chloro-1,1-difluoro-2,2-trifluoroethane, 1-chloro-1,1-difluoroethylene, 1,1-difluoroethane, dimethyl ether, monochlorodifluoromethane, trans-1,3,3,3-tetrafluoropropene, and mixtures thereof. The propellant may comprise hydrocarbons such as isobutane, propane, and butane, and these materials may be used for their low ozone reactivity and may be used as individual components where their vapor pressures at 21.1° C. range from about 1.17 Bar to about 7.45 Bar, alternatively from about 1.17 Bar to about 4.83 Bar, and alternatively from about 2.14 Bar to about 3.79 Bar. The propellant may be hydrofluoroolefins (HFOs).

The aerosol foam dispenser may be of the bag on valve type wherein the container comprises an inner bag and an outer container, which encloses the inner bag, while the inner bag has a valve mechanism attached which is movable between an open position and a closed position. The outer container may be formed from metal or plastic or the like, and any of the propellants described herein can be filled in a space between the outer container and the inner bag. The inner bag may be flexible, and can be made from a single material or from a composite material including plastic, which may comprise at least a polymeric layer and a layer which acts as a gas barrier, e.g., made from metal, such as Aluminum. The inner material of the bag may be inert to the contents of the composition, and the inner material may also be impenetrable by the contents of the composition in the bag. The inner bag may comprise a layer of a material which is essentially impermeable to the propellant inside of the bag. The inner bag may comprise a layer of a material which is essentially impermeable to the propellant outside of the bag which generally is not intended to be mixed with the composition in the inner bag during storage. In an embodiment where the propellant is inside the bag, it may be known as a foaming agent.

G. ADDITIONAL COMPONENTS

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The shampoo compositions of the present invention may optionally comprise one or more additional components known for use in hair care or personal care products, provided that the additional components are physically and chemically compatible with the essential components described herein, or do not otherwise unduly impair product stability, aesthetics or performance. Such additional components are most typically those described in reference books such as the CTFA

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Cosmetic Ingredient Handbook, Second Edition, The Cosmetic, Toiletries, and Fragrance Association, Inc. 1988, 1992. Individual concentrations of such additional components may range from about 0.001 wt% to about 10 wt% by weight of the hair care compositions.

Non-limiting examples of additional components for use in the hair care compositions include conditioning agents, anti-dandruff agents, particles, suspending agents, paraffinic hydrocarbons, propellants, viscosity modifiers, dyes, non-volatile solvents or diluents (water-soluble and water-insoluble), pearlescent aids, foam boosters, additional surfactants or nonionic cosurfactants, pediculocides, pH adjusting agents, perfumes, preservatives, proteins, skin active agents, sunscreens, UV absorbers, and vitamins.

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1. Conditioning Agent

The hair care compositions may comprise one or more conditioning agents. Conditioning agents include materials that are used to give a particular conditioning benefit to hair. The conditioning agents useful in the hair care compositions of the present invention typically comprise a water-insoluble, water-dispersible, non-volatile, liquid that forms emulsified, liquid particles. Suitable conditioning agents for use in the hair care composition are those conditioning agents characterized generally as silicones, organic conditioning oils or combinations thereof, or those conditioning agents which otherwise form liquid, dispersed particles in the aqueous surfactant matrix. The conditioning agent may contain one or more quaternary ammonium salt in its molecular structure. The conditioning agent may be a dimethiconol micro-emulsion.

One or more conditioning agents are present from about 0.01 wt% to about 10 wt%, from about 0.1 wt% to about 8 wt%, and from about 0.2 wt% to about 4 wt%, from about 0.5 to about 1.5% by weight of the composition.

25 Silicone Conditioning Agent

The compositions of the present invention may contain one or more silicone conditioning agents. Examples of the silicones include dimethicones, dimethiconols, cyclic silicones, methylphenyl polysiloxane, and modified silicones with various functional groups such as amino groups, quaternary ammonium salt groups, aliphatic groups, alcohol groups, carboxylic acid groups, ether groups, epoxy groups, sugar or polysaccharide groups, fluorine-modified alkyl groups, alkoxy groups, or combinations of such groups. Such silicones may be soluble or insoluble in the aqueous

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(or non-aqueous) product carrier. In the case of insoluble liquid silicones, the polymer can be in an emulsified form with droplet size of about 10 nm to about 30 micrometers

Suitable silicone conditioning agents include durable silicone materials such as cross-linkable silicone compounds containing different functional groups including siloxanes or silsequioxanes with terminal hydroxyl or alkoxy function groups. Non-limiting examples include Wacker Belsil ADM 8301E and Belsil ADM 6300E. Other suitable durable conditioning compounds include cross-linkable silicones such as MQ-resin, amino fluids and mixture thereof. Non-limiting examples include Wacker ADM 8500E, Dow Corning DX AP6087, Momentive Silform flexible resins, SR1000 MQ-resin and mxture thereof. Such compounds can cross-link upon drying on hair surface or after exposing to heat treatment to impart durable conditioning over multiple washing cycles.

Organic Conditioning Materials

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The conditioning agent of the compositions of the present invention may also comprise at least one organic conditioning material such as oil or wax, either alone or in combination with other conditioning agents, such as the silicones described above. The organic material can be nonpolymeric, oligomeric or polymeric. It may be in the form of oil or wax and may be added in the formulation neat or in a pre-emulsified form. Some non-limiting examples of organic conditioning materials include, but are not limited to: i) hydrocarbon oils; ii) polyolefins, iii) fatty esters, iv) fluorinated conditioning compounds, v) fatty alcohols, vi) alkyl glucosides and alkyl glucoside derivatives; vii) quaternary ammonium compounds; viii) polyethylene glycols and polypropylene glycols having a molecular weight of up to about 2,000,000 including those with CTFA names PEG-20 200, PEG-400, PEG-600, PEG-1000, PEG-2M, PEG-7M, PEG-14M, PEG-45M and mixtures thereof.

2. Rheology Modifier

In one embodiment the hair care product may include one or more rheology modifiers to adjust the rheological characteristics of the composition for better feel, in-use properties and the suspending stability of the composition. For example, the rheological properties are adjusted so that the composition remains uniform during its storage and transportation and it does not drip undesirably onto other areas of the body, clothing or home furnishings during its use. Any suitable rheology modifier can be used. In an embodiment, the leave-on treatment may comprise from about

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0.01% to about 3% of a rheology modifier, alternatively from about 0.1% to about 1% of a rheology modifier,

3. Benefit Agents

In an embodiment, the hair care composition further comprises one or more additional benefit agents. The benefit agents comprise a material selected from the group consisting of anti-fungal agents, anti-itch agents, anti-bacterial agents, anti-microbial agents, moisturization agents, anti-oxidants, chelants, vitamins, lipid soluble vitamins, perfumes, brighteners, enzymes, sensates, attractants, dyes, pigments, bleaches, and mixtures thereof.

According to an embodiment, the shampoo composition comprises an anti-dandruff active, which may be an anti-dandruff active particulate. The anti-dandruff active can be selected from the group consisting of: pyridinethione salts; azoles, such as an imidazole such as ketoconazole, econazole, climbazole and elubiol; selenium sulphide; coal tar, particulate sulfur; keratolytic agents such as salicylic acid; and mixtures thereof. In an embodiment, the anti-dandruff particulate is a pyridinethione salt.

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TEST METHODS

A. Cone/Plate Viscosity Method:

The viscosities of the examples are measured by a Cone/Plate Controlled Stress Brookfield Rheometer R/S Plus, by Brookfield Engineering Laboratories, Stoughton, MA. The cone used (Spindle C-75-1) has a diameter of 75 mm and 1° angle. The viscosity is determined using a steady state flow experiment at constant shear rate of 2 s⁻¹ and at temperature of 26.5 °C. The sample size is 2.5ml and the total measurement reading time is 3 minutes. The instrument cannot accurately report the viscosity of a liquid that is below 100cps. The viscosity is reported as less than 100cps.

B. Lather Rheology Method

Lather is generated in a vessel by adding 1) 180 ml of water (having a hardness of 7 grain per gallon) and 2) 20 ml of shampoo. A blade is placed in the center of the vessel. The mixture is blended at a speed of 1200 rpm for 1 minute.

Lather rheology is characterized using an AR2000 Rheometer from TA Instruments. A 60mm acrylate plastic top rotating plate is attached to the rheometer for later rheology measurement. A small amount of lather is applied on the bottom of the plate of the AR 2000 Rheometer using a spatulas. The top plastic plate is lowered with a gap of 1000 microns in between the two plates. The

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excess lather around the plates is removed. Oscillation stress sweeps are run to generate the elastic modulus or storage modulus (G'), the viscous modulus or loss modulus (G") and Tan Delta (the ratio of G"/G') of the later. The oscillation frequency is set at 1 Hz, oscillation stress is varied from 0.1 Pa to 25 Pa, and the testing temperature is 25°C. Elastic modulus (G') of the lather is the average G' value at oscillation stress range from 0.1 Pa to 0.2 pa.

C. Hair Wet Feel Friction method:

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A switch of 4 grams general population hair at 8 inches length is used for the measurement. Water temperature is set at 100°F, (hardness is 7 grain per gallon), and flow rate is 1.6 liter per minute. Amount of 0.4 ml of a liquid shampoo is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length, using a syringe. The hair switch is then 1st lathered for 30 seconds, rinse with water for 30 seconds, and 2nd lathered for 30 seconds. Water flow rate is then reduced to 0.2 liter per minute. The hair switch is sandwiched with a clamp under 1200 gram of force and pulled through the entire length while the water is running at the low flow rate. The pull time is 30 second. Friction is measured with a friction analyzer (such as Instron or MTS tensile measurement) with a load cell of 5kg. Repeat the pull under rinse for total of 21 times. Total 21 Friction values are collected. Hair Wet Feel Friction (F_{wet}) of shampoo is the final rinse friction which is the average friction of the last 7 friction measurements.

D. Hair Clean Rinse Feel method

A switch of 4 grams general population hair at 8 inches length is used for the measurement. Water temperature is set at 100°F, hardness is 7 grain per gallon, and flow rate is 1.6 liter per minute. Amount of 0.4 ml of a liquid shampoo is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length, using a syringe. The hair switch is then 1st lathered for 30 seconds, rinse with water for 30 seconds, and 2nd lathered for 30 seconds. Water flow rate is then reduced to 0.2 liter per minute. The hair switch is sandwiched with a clamp under 1200 gram of force and pulled through the entire length while the water is running at the low flow rate. The pull time is 30 second. Friction is measured with a Friction analyzer with a load cell of 5kg. Repeat the pull under rinse for total of 21 times. Total 21 Friction values are collected. Hair Wet Feel Friction (F_{wet}) of shampoo is the final rinse friction which is the average friction of the last 7 friction measurements. Then, the water is shut off.

The hair switch is still sandwiched with the clamp under 1200 gram of force and pulled through the entire length. The pull time is 30 second. Friction is measured with the Friction analyzer

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with the load cell of 5kg. Repeat the pull for total of 10 times. Hair Rinse Feel Friction (F_{rinse}) of shampoo is the final rinse friction which is the average friction of the last 5 friction measurements. Hair Clean Rinse Feel (F_{clean}) is the magnitude of friction reduction from Hair Wet Feel Friction (F_{wet}) to Hair Rinse Feel Friction (F_{rinse}).

 $F_{\text{clean}} = F_{\text{wet}} - F_{\text{rinse}}$

E. Hair Switch Soil Removal Method

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A switch of 4 grams general population hair at 8 inches length is used for the measurement. It's weighted and recorded as W0. Amount of 2ml of coconut oil (Crodamol GTCC-LQ-(MV), supplied from Croda) is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length, using a syringe. The hair switch is then rubbed for 30 second to soak all of the oil into the hair switch. The soiled hair switch is weighted and recorded as W1. The Soil applied on the hair switch is calculated as W1-W0 and recorded as Wsoil applied. Water is then turned on. The temperature is set at 100oF, hardness is 7 grain per gallon, and flow rate is 1.6 liter per minute. The soiled hair switch is wetted with water for 1 second. Amount of 0.4 ml of a liquid shampoo is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length, using a syringe. The hair switch is lathered for 1 minute, rinsed with water for 1 minute, and squeezed to let the excess water out. The hair switch is hung on a rack and dried overnight. The cleaned and air dried hair switch is weighted and recorded as W2. The weight of oil residue left on the hair switch is calculated W2-W0 and recorded as Wsoil residue.

The Weight Percent Soil Residue left on the hair switch is calculated as % Soil Residue left on hair switch = $W_{\text{soil residue}} / W_{\text{soil applied}} \times 100$

The Weight Percent of Soil Removed from hair switch is calculated as:

% Soil Removed from hair switch = $(W_{\text{soil applied}} - W_{\text{soil residue}}) / W_{\text{soil applied}} \times 100$

F. Hair Switch Expansion Method

A switch of 4 grams general population hair at 8 inches length is used for the measurement. It's weighted and recorded as W0. Amount of 2ml of coconut oil (Crodamol GTCC-LQ-(MV), supplied from Croda) is applied on the hair switch in a zigzag pattern uniformly to cover the entire

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hair length, using a syringe. The hair switch is then rubbed for 30 second to soak all of the oil into the hair switch. The soiled hair switch is then hung in front of a lighted white board with back lights from behind. Pictures are then taken. The images are analyzed using Image Pro 7 Analyzer. Bulk area of the soiled hair switch is reported as Asoiled in number of pixels. Water is then turned on. The temperature is set at 100°F, hardness is 7 grain per gallon, and flow rate is 1.6 liter per minute. The soiled hair switch is wetted with water for 1 second. Amount of 0.4 ml of a liquid shampoo is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length, using a syringe. The hair switch is lathered for 1 minute, rinsed with water for 1 minute, and squeezed to let the excess water out. The hair switch is hung on a rack and dried overnight. The cleaned and air dried hair switch is hung in front of the lighted white board with back lights from behind. Pictures are then taken. The images are analyzed using Image Pro 7 Analyzer. Bulk area of the soiled hair switch is reported as Acleaned in number of pixels.

The Area expansion of hair switch is calculated as:

Area Expansion of Hair Switch = $A_{cleaned} - A_{soiled}$

The Percent expansion of hair switch is calculated as:

% Expansion of Hair Switch = $(A_{cleaned} - A_{soiled}) / D_{soiled} \times 100$

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G. Hair Flexibility Method

A switch of 4 grams general population hair at 8 inches length is used for the measurement. Water temperature is set at 100°F, hardness is 7 grain per gallon, and flow rate is 1.6 liter per minute. An amount of 0.4 ml of a shampoo is applied on the hair switch in a zigzag pattern uniformly to cover the entire hair length using a syringe. The hair switch is lathered for 1 minute, rinsed with water for 1 minute, and squeezed to let the excess water out. The hair switch is hung on a rack and dried overnight. Three replicates of hair switches are washed and dried for each shampoo.

Hair Flexibility of the dried hair switches is assessed by six panelists. Panelists are asked to grade the hair switches on a 0 to 10 scale (0 = low, 10 = high). Panelists' responses are then averaged to assign a hair flexibility score for the shampoo example.

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EXAMPLES

METHOD OF MAKING

The following examples illustrate embodiments of the invention described herein. The exemplified hair care compositions may be made by conventional formulation and mixing techniques or by mixing together one or more polyols (e.g. glycerin, propylene glycol etc.), water and surfactants along with any solids that need to be melted at an elevated temperature, e.g. about 75°C. The ingredients are mixed thoroughly at the elevated temperature and then cooled to ambient temperature. Additional ingredients, including electrolytes, polymers, silicone emulsions, preservatives and fragrances may be added to the cooled product. It will be appreciated that other modifications of the hair care compositions, and/or conditioner compositions within the skill of those in the formulation art can be undertaken without departing from the spirit and scope of this invention. All parts, percentages, and ratios herein are by weight unless otherwise specified. Some components may come from suppliers as dilute solutions. The amount stated reflects the weight percent of the active material, unless otherwise specified.

The following are non-limiting examples of Hair Care compositions described herein.

Table 1 Cleansing shampoo compositions

	Comp. Ex. 1	Comp. Ex. 2	Ex. 1	Ex. 2	Ex. 3
Appearance	Clear, stable	Clear, stable	Clear, stable	Clear, stable	Clear, stable
Stability	Yes	Yes	Yes	Yes	Yes
Viscosity (cP)	5	6	16	43	321
Glycerin	0	10	30	50	70
Water	83.8	73.8	53.8	33.8	13.8
Sodium Laureth Sulfate (SLE1S-70% active) ¹	8	8	8	8	8
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	7	7	7	7	7
pH adjusters to adjust pH 5.7 ± 0.3					
Glycerin/water ratio	0.00	0.14	0.56	1.48	5.07
Total Detersive Surfactant	15	15	15	15	15
% Soil Residual left	5.05%	3.64%	1.96%	2.42%	1.77%

Table 2 Cleansing shampoo compositions

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Ev 1	Ev. 5	Ev 6	Ev. 7	Ev Q	Ev 0	Ev
EX. 4	Ex. 3	EX. O	EX. /	Ex. o	EX. 9	EX.

							10
Appearance	Clear, stable	Clear, stable	Clear, stable	Clear, stable	Clear, stable	Clear, stable	Clear, stable
Stability	Yes	Yes	Yes	Yes	Yes	Yes	Yes
Viscosity (cP)	212	69	6466	2114	735	162	790
Glycerin	80	70	50	70	72	50	68.9
Water	10.8	20.8	35.8	15.8	8.81	28.8	9.89
Sodium Laureth Sulfate (SLE1S-70% active) ¹	8	8	8	8	8	8	8
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰			5	5	10	12	12
pH adjusters to adjust pH 5.7 ± 0.3							
Glycerin/water ratio	7.41	3.37	1.40	4.43	8.17	1.74	6.97
Total Detersive Surfactant	8	8	13	13	18	20	20

Table 3 Cleansing shampoo compositions

	Ex. 11	Ex. 12	Ex. 13	Ex. 14	Ex. 15	Ex. 16
Appearance	Clear, stable	Clear, stable				
Stability	Yes	Yes	Yes	Yes	Yes	Yes
Viscosity (cP)	1410	903	782	1051	1039	1542
Glycerin	53.53	30	46.7	46.7	46.7	46.7
Water	15.27	38.8	23.3	22.7	23.3	22.7
Sodium Laureth Sulfate (SLE1S-70% active) ¹	8	8				
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	22	22	26	26		
Sodium Tridecyl Ether Sulfate (ST3S-65 active) ¹¹					26	26
Cocoamdopropyl Betaine (CAPB 30% active) ⁴			4		4	
LAPB (35%) ⁷				4.7		4.7
pH adjusters to adjust pH 5.7 ± 0.3						
Glycerin/water ratio	3.51	0.77	2	2.1	2	2.1
Total Detersive Surfactant	30	30	30	31	30	31

Table 4 Cleansing shampoo compositions

Ex.	Ev. 18	Fv 10	Ex. 20	Ev. 21	Ex.
17	Ex. 10	EX. 19	EX. 20	EX. 21	22

Glycerin	30	50	70	-	-	-		
Propylene Glycol	-	-	_	30	50	70		
Water	53.8	33.8	13.8	53.8	33.8	13.8		
Sodium Laureth								
Sulfate (SLE1S-	5	5	5	5	5	5		
70% active) ¹								
Sodium Tridecyl								
Ether Sulfate	10	10	10	10	10	10		
(ST2S-65 active) ¹⁰								
Perfume	1.2	1.2	1.2	1.2	1.2	1.2		
Glycerin/water ratio	0.56	1.48	5.07	-	-	-		
Propylene				0.56	1.48	5.07		
Glycol/water ratio	-	-	-	0.30	1.40	3.07		
Total Detersive	15	15	15	15	15	15		
Surfactant	13	15	13	13	13	13		
pH adjusters to adjust pH 5.7 ± 0.3								

Table 5 Cleansing shampoo compositions

	Ex. 23	Ex. 240	Ex. 25	Ex. 26	Ex. 27	Ex. 28
Glycerin	50	50	50	50	70	70
Water	33.7	33.2	33.1	32.1	13.6	13.6
Sodium Laureth Sulfate (SLE1S-70% active) ¹	5	5	5	5	5	5
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	10	10	10	10	10	10
Guar, Hydroxylpropyl Trimonium Chloride, Jaguar C-500 ³	0.1	0.1	0.2	0.2	ı	-
Silicone Quaternium ⁴	-	0.5	0.5	1.5	-	-
Polyquaternium-10, Ucare LR400 ⁵	-	-	-	-	0.2	-
Polyquaternium-67, Softcat	-	-	-	-	-	0.2

$ SX1300^6 $						
Perfume	1.2	1.2	1.2	1.2	1.2	1.2
Glycerin/water ratio	1.48	1.51	1.51	1.56	5.15	5.15
Total Detersive Surfactant	15	15	15	15	15	15
pH adjusters to adjust pH 5.7 \pm 0.3	- 1				1	1

Table 6: Cleansing shampoo compositions

	Ex. 29	Ex. 30	Ex. 31
Glycerin	71.9	69.5	68.1
Water	13.1	14.5	14.9
Sodium Laureth Sulfate (SLE1S-70% active) ¹	3	4	4
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	12	12	13
Glycerin/water ratio	5.5	4.8	4.6
Total Detersive Surfactant	15	16	17
pH adjusters to adjust pH 5.7 ± 0.3	1		1

Table 7 Cleansing shampoo compositions

	Ex.	Ex. 33
	32	LX. 33
Glycerin	71.4	73.6
Water	12.6	11.4
Lauramidopropyl Betaine (LAPB 35% active) ⁷	4	3.5
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	12	11.5
Glycerin/water ratio	5.7	6.4
Total Detersive Surfactant	16	15

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pH adjusters to adjust pH 5.7 ± 0.3

Table 8 Cleansing shampoo compositions

	Ex 34	Ex 35	Ex 36	Ex 37
Glycerin	42.9	36.2	45.7	39.0
Water	25.1	29.8	24.3	29.0
Sodium Laureth Sulfate (SLE1S-70% active) ¹	6	8	6	8
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	26	26	24	24
Glycerin/water ratio	1.7	1.2	1.9	1.3
Total Detersive Surfactant	32	34	30	32
pH adjusters to adjust pH 5.7 ± 0.3				

Table 9 Cleansing shampoo compositions

	Ex 38	Ex 39	Ex 40	Ex 41	Ex 42
Glycerin	39.5	46.7	46.7	46.7	46.7
Water	26.5	23.3	22.7	23.3	22.7
Sodium Laureth Sulfate (SLE3S-70% active) 9	5	-	-	-	-
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	20	26	26	-	-
Sodium Tridecyl Ether Sulfate (ST3S-65 active) ¹¹	-	-	-	26	26
Cocoamdopropyl Betaine (CAPB 30% active) ²	4	4	-	4	-
LAPB (35%) ⁷	-	-	4.7	-	4.7
Lauryl Hydroxysultaine (LHS 42.5% active) ¹²	4	-	-	-	-
Glycerin/water ratio	1.5	2.0	2.1	2.0	2.1

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Total Detersive Surfactant	34	30	31	30	31

- 1. Sodium Laureth (1 molar ethylene oxide) sulfate at 70% active, supplier: Stephan Co
- 5 2. Tegobetaine F-B, 30% active, supplier: Goldschmidt Chemical
 - 3. Jaguar C500, MW of 500,000, CD of 0.8, from Rhodia
 - 4. Silicone quaternium micro-emulsion, 30% active, Abil ME 45, from Evonik
 - 5. Polyquaternium-10, Ucare LR400, from Dow Chemical
 - 6. Polyquaternium-67, Softcat SX1300, from Dow Chemical
 - 7. LAPB (Mackam DAB), at 35% active level, supplier: Rhodia
 - 8. LAPB (Mackam 1200), at 84% active level, supplier: Rhodia
 - 9. Sodium Laureth (3 molar ethylene oxide) sulfate at 70% active, supplier: Stephan Co
 - 10. Sodium Tridecyl Ether Sulfate (2 molar ethylene oxide), Stepan ST2S-65 (Steol-TD 402 65) 65% active, supplier: Stephan Co
- 11. Sodium Tridecyl Ether Sulfate (3 molar ethylene oxide), Stepan ST3S-65 (Steol-TD 403 65) 65% active, supplier: Stephan Co
 - 12. LHS (Mackam LHS) at 42.5% active level, supplier: Rhodia

FOAMED COMPOSITION EXAMPLES

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The hair care composition can be delivered in a liquid or foam form. It may be delivered in a foam form via a mechanical foam dispenser when the hair care composition has a viscosity less than about 500 cps. It may be further delivered in a foam form via an aerosol foam dispenser with about 1 to 10 weight percent of a propellant when the hair care composition has a viscosity less than about 3000 cps. The viscosity values are measured before adding in the propellants.

25 Table 10 Cleansing foam shampoo composition via Mechanical Pump Foamer

	Ex. A	Ex. B	Ex. C	Ex. D	Ex. E	Ex. F
Appearance	Clear,	Clear,	Clear,	Clear,	Clear,	Clear,
	stable	stable	stable	stable	stable	stable

Stability	Yes	Yes	Yes	Yes	Yes	Yes
Viscosity (cP)	16	43	321	212	69	162
Glycerin	30	50	70	80	70	50
Water	53.8	33.8	13.8	10.8	20.8	28.8
Sodium Laureth Sulfate (SLE1S-70% active)	8	8	8	8	8	8
Sodium Tridecyl Ether Sulfate (ST2S-65 active) ¹⁰	7	7	7			12
Glycerin/water ratio	0.56	1.48	5.07	7.41	3.37	1.74
Total Detersive Surfactant	15	15	15	8	8	20

Table 13 Cleansing foam shampoo composition via Aerosol Foam Former

	Ex. G	Ex. H	Ex. I	Ex. J	Ex. K	Ex. L	Ex. M	Ex. N	Ex. O
Appearance	Clear,	Clear,	Clea	Clea	Clear,	Clear,	Clear,	Clear,	Clear,
	stable	stable	r,	r,	stable	stable	stable	stable	stable
			stabl	stabl					
			e	e					
Stability	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes
Bulk Viscosity (cps)	1410	903	782	1051	1039	1542	735	162	790
Glycerin	53.53	30	46.7	46.7	46.7	46.7	72	50	68.9
Water	15.27	38.8	23.3	22.7	23.3	22.7	8.81	28.8	9.89
Sodium Laureth Sulfate	8	8					8	8	8
(SLE1S-70% active)									
Sodium Tridecyl Ether	22	22	26	26			10	12	12
Sulfate (ST2S-65									
active) ¹⁰									
Sodium Tridecyl Ether					26	26			
Sulfate (ST3S-65									
active) ¹¹									

Cocoamdopropyl			4		4				
Betaine (CAPB 30%									
active) ²									
LAPB (35%) ⁷				4.7		4.7			
Propellant HFO 16	5.5	5.5	5.5	5.5	5.5	5.5	5.5	5.5	5.5
Glycerin/water ratio	3.51	0.77	2	2.1	2	2.1	8.17	1.74	6.97
Total Detersive	30	30	30	31	30	31	18	20	20
Surfactant									
pH adjusters to adjust pH	5.7 ± 0.	.3	•	•		•	•		

Table 14 Cleansing foam shampoo composition via Aerosol Foam Former

	Ex. P	Ex. Q	Ex. R	Ex. S	Ex. T	Ex. U	Ex.	Ex.	Ex. X
							V	W	
Appearance	Clear,	Clear,	Clear,	Clear,	Clear,	Clear,	Clear	Clear,	Clear,
	stable	stable	stable	stable	stable	stable	,	stable	stable
							stabl		
							e		
Stability	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes
Bulk Viscosity (cps)	1410	903	782	1051	1039	1542	735	162	790
Glycerin	53.53	30	46.7	46.7	46.7	46.7	72	50	68.9
Water	15.27	38.8	23.3	22.7	23.3	22.7	8.81	28.8	9.89
Sodium Laureth	8	8					8	8	8
Sulfate (SLE1S-70%									
active)									
Sodium Tridecyl Ether	22	22	26	26			10	12	12
Sulfate (ST2S-65									
active) ¹⁰									
Sodium Tridecyl Ether					26	26			
Sulfate (ST3S-65									

active) ¹¹									
Cocoamdopropyl			4		4				
Betaine (CAPB 30%									
active) ²									
LAPB (35%) ⁷				4.7		4.7			
Propellant A46 15	4	4	4	4	4	4	4	4	4
Glycerin/water ratio	3.51	0.77	2	2.1	2	2.1	8.17	1.74	6.97
Total Detersive	30	30	30	31	30	31	18	20	20
Surfactant									
pH adjusters to adjust pl	H 5.7 ± ().3							

15. Aeron A-Blends, A46 (Isobutane/Propane = 84.85/15.15) from Diversified CPC International

16. Hydrofluoroolefins (HFO-1234ze) from Honeywell

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Every document cited herein, including any cross referenced or related patent or application and any patent application or patent to which this application claims priority or benefit thereof, is hereby incorporated herein by reference in its entirety unless expressly excluded or otherwise limited. The citation of any document is not an admission that it is prior art with respect to any invention disclosed or claimed herein or that it alone, or in any combination with any other reference or references, teaches, suggests or discloses any such invention. Further, to the extent that any meaning or definition of a term in this document conflicts with any meaning or definition of the same term in a document incorporated by reference, the meaning or definition assigned to that term in this document shall govern.

While particular embodiments of the present invention have been illustrated and described, it would be obvious to those skilled in the art that various other changes and modifications can be made without departing from the spirit and scope of the invention. It is therefore intended to cover

in the appended claims all such changes and modifications that are within the scope of this invention.

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CLAIMS

What is claimed is:

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- 1. A hair care composition comprising:
 - a) from about 5 to about 40 weight % of the hair care composition of one or more detersive surfactants wherein less than about 10 weight% of the hair care composition comprises a linear anionic surfactant selected from the group consisting of:
 - (1) alkyl sulfates

$$\begin{array}{c} \circ \\ \circ \\ \circ \\ \circ \end{array} = \begin{array}{c} \circ \\ \circ \\ \circ \end{array}$$

Alkyl sulfates

where R is a linear C_8 - C_{24} alkyl and M^+ is monovalent cation;

(2) alkyl ether sulfates

$$\mathbf{R} - \mathbf{O} + \mathbf{CH}_2 - \mathbf{CH}_2 - \mathbf{O} + \mathbf{S} - \mathbf{O} + \mathbf{M}$$

Alkyl ether sulfates

where R is a linear C_8 - C_{24} alkyl, n=1-2, and M^+ is monovalent cation;

- (3) and mixture thereof,
- b) a carrier comprising one or more polyols and water, wherein the carrier comprises from about 20 to about 80 weight % of the hair care composition of one or more polyols, and from about 9% to about 75% by weight of the hair care composition of water; and wherein the weight ratio of one or more polyols to water is about 0.4 or higher
- 2. The hair care composition according to claim 1, wherein one of the detersive surfactant is a branched anionic surfactant selected from the group consisting of :

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(1) alkyl sulfates

$$R \longrightarrow 0 \longrightarrow S \longrightarrow 0 \longrightarrow M$$

Alkyl sulfates

where R is a branched C₈-C₂₄ alkyl and M⁺ is monovalent cation;

(2) alkyl ether sulfates

$$\mathbf{R} - \mathbf{O} + \mathbf{CH}_2 - \mathbf{CH}_2 - \mathbf{O} + \mathbf{S} - \mathbf{O} + \mathbf{M}$$

Alkyl ether sulfates

where R is a branched C₈-C₂₄ alkyl, n=1-2, and M⁺ is monovalent cation;

(3) and mixture thereof,

- 3. The hair care composition according to any of the preceding claims, wherein one of the detersive surfactant is selected from the group consisting of:
- i) alkyl betaines

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$$\begin{array}{c|c} \operatorname{CH_3} & \operatorname{O} \\ | \oplus & \operatorname{II} & \ominus \\ \operatorname{CH_2-C-O} \\ | & \operatorname{CH_3} \end{array}$$

Alkyl betaines

where R is C_8 - C_{24} alkyl (saturated or unsaturated) or mixtures thereof. Examples include coco-betaine (where R is coco alkyl), lauryl betaine (where R is lauryl, $C_{12}H_{25}$), and oleyl betaine (where R is oleyl, $C_{18}H_{35}$);

ii) alkyl hydroxysultains

$$\begin{array}{c} CH_3 \\ |_{\bigoplus} \\ R - N - CH_2 - CH - CH_2 - S - O \\ |_{CH_3} \\ OH \end{array} \quad \begin{array}{c} O \\ |_{\bigoplus} \\ O \\ O \end{array}$$

Alkyl hydroxysultaines

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where R is C₈-C₂₄ alkyl (saturated or unsaturated) or mixture thereof. Examples include lauryl hydroxysultaine (where R is lauryl, C₁₂H₂₅) and coco-hydroxysultaine (where R is coco alkyl);

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iii) alkyl amphoacetates

$$\begin{array}{c|c} & \bigcirc & \bigcirc & \bigcirc & \bigcirc \\ \mathbb{R} - \mathbb{C} - \mathbb{N} \mathbb{H} - \mathbb{C} \mathbb{H}_2 - \mathbb{C} \mathbb{H}_2 - \mathbb{N} - \mathbb{C} \mathbb{H}_2 - \mathbb{C} - \mathbb{O} & \mathbb{M} \\ & & \mathbb{C} \mathbb{H}_2 - \mathbb{C} \mathbb{H}_2 - \mathbb{O} \mathbb{H} \end{array}$$

Alkyl amphoacetates

where R is C_6 - C_{24} alkyl (saturated or unsaturated) or mixtures thereof and M^+ is monovalent cation. Examples include sodium lauroamphoacetate (where R is lauryl and M^+ is Na^+), sodium cocoamphoacetate (where R is coco acyl and M^+ is Na^+),

15 and

- iv) mixtures thereof.
- 4. The hair care composition according to any of the preceding claims, wherein the one or more polyols is selected from the group consisting of glycerin, ethylene glycol, propylene glycol, dipropylene glycol, ethylene glycol and mixtures thereof., preferably wherein the one or more polyol is glycerin.
- 5. The hair care composition according to any of the preceding claims, comprising from about 10 to about 28 wt% of the hair care composition of an anionic surfactant, preferably from about 11 to about 26 wt% of the hair care composition of an anionic surfactant, preferably from about 11 to

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about 24 wt% of the hair care composition of an anionic surfactant, preferably from about 10 to about 20 wt% of the hair care composition of an anionic surfactant, even more preferably from about 10 to about 15 wt% of the hair care composition of an anionic surfactant.

- 5 6. The hair care composition according to any of the preceding claims, comprising from about 2% to about 15 wt % of the hair care composition of zwitterionic surfactant.
 - 7 The hair care composition according to any of the preceding claims, wherein the detersive surfactant is one of more nonionic surfactants.

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- 8. The hair care composition according to any of the preceding claims, wherein the hair care composition further contains about 0.05 to about 5 weight percent of a silicone conditioning agent, preferably wherein the silicone conditioning agent contains one of more quaternary ammonium salt in its molecular structure, even more preferably, wherein the silicone conditioning agent is dimethiconol micro-emulsion.
- 9. The hair care composition according to any of the preceding claims, wherein the hair care composition further comprises from about 0.05 to about 2 weight % of the hair care composition of one or more cationic polymers, preferably wherein the cationic polymers are selected from the group consisting of guar hydroxylpropyltrimonium chloride, Polyquaternium-6, Polyquaternium-7, Polyquaternium-10, Polyquaternium-39, Polyquaterinum-67, and mixtures thereof.
- 10. The hair care composition according to any of the preceding claims, wherein the guar hydroxylpropyltrimonium chloride has a weight average molecular weight of from about 50,000 to about 1.0 million g/mol, preferably wherein the guar hydroxylpropyltrimonium chloride has a weight average molecular weight of from about 100,000 to about 900,000 g/mol.
- 11. The hair care composition of according to any of the preceding claims, wherein the hair care composition is delivered in a foam form via a mechanical foam dispenser.

- 12. The hair care composition of any of claims 1-14 wherein the composition further comprises from about 2% to about 12% by weight of a propellant and is dispensed via an aerosol foam dispenser.
- 13. A method of cleansing hair using a hair care composition comprising:
 - a) applying a hair care composition to a soiled hair, wherein the hair care composition comprises:
 - 1) from about 5 to about 40 weight % of the hair care composition of one or more detersive surfactants wherein less than about 10 weight% of the hair care composition comprises a linear anionic surfactant selected from the group consisting of:
 - aa) alkyl sulfates

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$$R - O - S - O M$$

Alkyl sulfates

where R is a linear C_8 - C_{24} alkyl and M^+ is monovalent cation; bb) alkyl ether sulfates

$$\mathsf{R} - \mathsf{O} - \mathsf{CH}_2 - \mathsf{CH}_2 - \mathsf{O} + \mathsf{S} - \mathsf{O} \oplus \mathsf{M}$$

Alkyl ether sulfates

where R is a linear C_8 - C_{24} alkyl, n=1-2, and M^+ is monovalent cation; cc) and mixture thereof,

2) a carrier comprising water and one or more polyols, wherein the weight ratio of one or more polyols to water is higher than or equal to about 0.4, the one or more polyols content is from about 20 to about 80 weight % of the composition, and the water content is higher than or equal to 9% by weight of the composition;

- b) rinsing the hair with water and air dry,
- c) measuring the % oil residue left on the hair, wherein the % oil residue left on hair is about 0 to about 3 weight % of the initial oil amount as measured by the hair oil removal method.
- The method according to any of the preceding claims, wherein the % oil residue left on hair is about 0 to about 1.6 weight % of the initial oil amount as measured by the hair switch oil removal method, preferably wherein the % oil residue left on hair is about 0 to about 1.5 weight % of the initial oil amount as measured by the hair switch oil removal method, more preferably wherein the % oil residue left on hair is about 0 to about 1.4 weight % of the initial oil amount as measured by the hair switch oil removal method, even more preferably wherein the % oil residue left on hair is about 0 to about 1.3 weight % of the initial oil amount as measured by the hair switch oil removal method.

International application No PCT/US2016/066752

A. CLASSIFICATION OF SUBJECT MATTER INV. A61K8/34 A61K8 A61K8/44

A61K8/46

A61Q5/02

A61Q5/12

ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

A61K A61Q

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, WPI Data, CHEM ABS Data

C. DOCUM	ENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 2 042 216 A1 (ACO HUD NORDIC AB [SE]) 1 April 2009 (2009-04-01)	1,4,7, 11-14
Υ	examples 1,2	2,3
Х	JP 2012 001597 A (SHISEIDO CO LTD) 5 January 2012 (2012-01-05) examples 1-18	1,4,6,7, 13,14
X	US 2001/056049 A1 (ARONSON MICHAEL PAUL [US] ET AL) 27 December 2001 (2001-12-27) examples 13-24, 27-32, 34	1,4-7,9, 10
X	WO 02/092050 A2 (UNILEVER PLC [GB]; UNILEVER NV [NL]; LEVER HINDUSTAN LTD [IN]) 21 November 2002 (2002-11-21) claim 1; examples E-H, K-M, P, Q	1,4-9

Χ	Further documents are listed in the continuation of Box C.

Χ See patent family annex.

- * Special categories of cited documents :
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- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
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22/02/2017

Date of the actual completion of the international search Date of mailing of the international search report

10 February 2017

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Authorized officer

Lenzen, Achim

International application No
PCT/US2016/066752

		PC1/US2U10/U00/52
C(Continua	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Х	WO 01/42409 A1 (UNILEVER PLC [GB]; UNILEVER NV [NL]; LEVER HINDUSTAN LTD [IN]) 14 June 2001 (2001-06-14) page 5, line 3 - line 5; examples 1,2	1,4,5, 13,14
X	US 2006/183662 A1 (CROTTY BRIAN A [US] ET AL) 17 August 2006 (2006-08-17) claim 19; examples 1-6	1,4-7,9, 13,14
Υ	DE 43 15 396 A1 (HENKEL KGAA [DE]) 10 November 1994 (1994-11-10) claim 1	2
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