



## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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<b>(21) International Application Number:</b> PCT/GB96/00518 <b>(22) International Filing Date:</b> 6 March 1996 (06.03.96) <b>(30) Priority Data:</b> 9504995.3                      11 March 1995 (11.03.95)                      GB <b>(71) Applicant (for all designated States except US):</b> ZENECA LIMITED [GB/GB]; 15 Stanhope Gate, London W1Y 6LN (GB). <b>(72) Inventors; and</b> <b>(75) Inventors/Applicants (for US only):</b> POPAT, Ajay, Haridas [GB/GB]; 66 Winchester Avenue, Great Sankey, Warrington WA5 1XX (GB). LAWSON, John, Robert [GB/GB]; 14 West Street, Middleton, Manchester M24 6BE (GB). <b>(74) Agents:</b> MAYALL, John; Zeneca Specialties, Intellectual Property Group, Hexagon House, P.O. Box 42, Blackley, Manchester M9 8ZS (GB) et al.		<b>(81) Designated States:</b> AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, US, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
<b>(54) Title:</b> LIQUID PHOTOCURABLE COMPOSITIONS		
<b>(57) Abstract</b>  This invention relates to a liquid composition comprising: a) 2 to 20 parts of a monomeric poly(meth)acrylate having a functionality of at least 3 and a MW of at least 650; b) 20 to 60 parts of a urethane(meth)acrylate having a functionality of 2 to 4 and a MW of 400 to 10,000; c) 20 to 80 parts of a monomeric or oligomeric di(meth)acrylate based on bisphenol A or bisphenol F; and d) 0.1 to 10 parts of a photoinitiator; wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100. The compositions are suitable for the sterolithographic production of three-dimensional articles such as models of mechanical and body parts and are particularly notable for their low curl and shrinkage coupled with fast rate of cure.		

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### Liquid photocurable compositions.

This invention relates to photocurable compositions, to a process for their polymerisation by means of actinic radiation, to a process for the production of three-dimensional articles from the compositions and to articles so prepared.

5 It is known that complicated three-dimensional articles can be produced from liquid, photocurable compositions by means of stereolithography. One such process is described in Hull's US Patent 4,575,330. Articles are built up in layers, each new curable layer of resin being firmly attached to the preceding pre-cured layer by preliminary  
10 curing by means of UV/VIS light. The overall construction of the three-dimensional article is normally controlled by computer. Photocurable compositions (e.g. liquid resins and resin mixtures) and their use as coating agents, adhesives and photoresists are known. However, the majority of such compositions are not suitable for the production of  
15 solidified three-dimensional articles by stereolithography because some are too viscous, whilst others are insufficiently light-sensitive, cure slowly or suffer from excessive shrinking or curling when they are cured.

Ideally photocurable compositions for stereolithography cure reasonably quickly and have a small volume shrinkage in the transition from  
20 the liquid to solid state. The so-called "curl factor" is often quoted as a measure of shrinkage induced deformation and a curl factor of 30% is considered by many as being the maximum acceptable, depending on what the article is, and curl factors of below 20% are preferred.

Canadian Patent application 2028541 suggests photocurable  
25 compositions for stereolithography. Whilst these compositions do work they tend to have a high curl distortion and high viscosity.

According to the present invention there is provided a liquid composition comprising:

30 a) 2 to 20 parts of a monomeric poly(meth)acrylate having a functionality of at least 3 and a MW of at least 650;

b) 20 to 60 parts of a urethane(meth)acrylate having a functionality of 2 to 4 and a MW of 400 to 10,000;

c) 20 to 80 parts of a monomeric or oligomeric di(meth)acrylate based on bisphenol A or bisphenol F; and

35 d) 0.1 to 10 parts of a photoinitiator;

wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100.

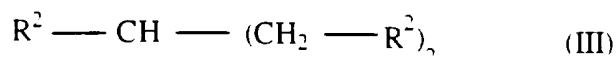
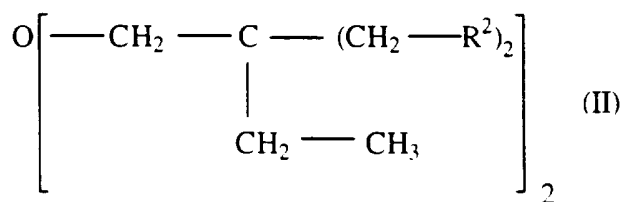
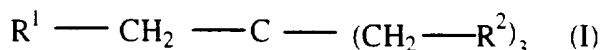
The liquid composition of the invention preferably comprises or consists essentially of 5 to 18 parts, more preferably 8 to 16 parts of  
40 component a); 20 to 50 parts, more preferably 25 to 45 parts of component b); 20 to 70 parts, more preferably 30 to 60 parts, especially 35 to 55

parts, more especially 40 to 50 parts of component c); and 1 to 9 parts, more preferably 2 to 8 parts, especially 3 to 7 parts of component d); wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100.

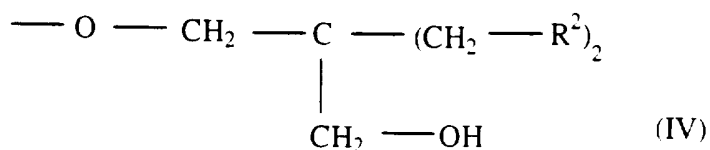
In a particularly preferred embodiment the liquid composition of the invention comprises or consists essentially of 8 to 16 parts of component a); 25 to 45 parts of component b); 40 to 50 parts of component c); and 3 to 7 parts of component d); wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100.

Component a) preferably has a MW (i.e. molecular weight) greater than 800, more preferably in the range 880-1200, especially in the range 900-1100.

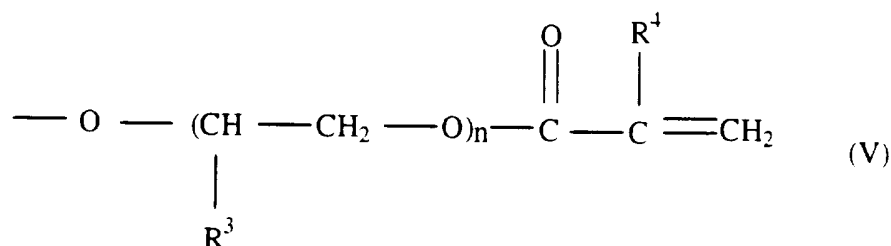
Examples of suitable compounds which may be used as component (a) are tri-, tetra- and penta-(meth)acrylates (including mixtures thereof), especially those of the formulae I, II, and III:



wherein R is a hydrogen atom, methyl, hydroxyl or a group of the formula IV:



wherein R<sup>2</sup> is a group of the formula V:



wherein n is from 3 to 8, preferably 4 to 8, and R<sup>3</sup> and R<sup>4</sup> are each independently hydrogen or methyl.

Compounds of the formulae I to III which are particularly preferred are those of the formula I in which R<sup>1</sup> is methyl or a group of the formula IV, R<sup>2</sup> is a group of the formula V wherein n is 4 or 5, and R<sup>3</sup> and R<sup>4</sup> are as hereinbefore defined.

As examples of compounds which can be used as component a) there may be mentioned highly propoxylated and ethoxylated 1,1,1-trimethylolpropane triacrylate or trimethacrylate and mixtures thereof. Compounds of this type are known and some are commercially available, for example SARTOMER products of Cray Valley Co. Limited, Newport, Wales and Sartomer Company supply such compounds under the product names SR-9021 and SR-9035.

One may use known urethane acrylates as component b) in compositions according to the invention and these can be prepared in a known manner, for example by reacting a hydroxyl-terminated polyurethane with acrylic acid or methacrylic acid to give the corresponding urethane(meth)acrylate, or by reacting an isocyanate-terminated prepolymer with hydroxyalkyl acrylates or methacrylates to give the urethane(meth)acrylate. Appropriate processes are disclosed in, for example, published European patent applications 114,982 and 133,908. The molecular weight of such acrylates is generally within the range from 400 to 10,000 preferably between 500 and 7,000.

Urethane (meth)acrylates are also available commercially under the trade marks NeoRad from Zeneca Resins, EBECRYL from UCB, Uvithane from Morton Thiokol and SR 9504, SR 9600, SR 9610, SR 9620, SR 9630, SR 9640 and SR 9650 from the Sartomer Company.

It is preferable to use urethane acrylates of MW 500-7000 derived

from aliphatic starting materials.

Di(meth)acrylates based on bisphenol A and bisphenol F which can be used in or as component c) include bisphenol A diacrylates and dimethacrylates and bisphenol F diacrylates and dimethacrylates and the diacrylates or dimethacrylates of alkoxyated, preferably ethoxylated or propoxylated, bisphenol A or F. The acrylates obtainable by reaction of bisphenol A or bisphenol F diglycidyl ether with (meth)acrylic acid are also suitable, especially mixtures consisting of ethoxylated bisphenol A diacrylate and ethoxylated bisphenol A dimethacrylate. Monomeric or oligomeric di(meth)acrylates of this type are also known and some are available commercially, for example from the Sartomer Company under the product name SR-348 for ethoxylated bisphenol A dimethacrylate and under the product name SR-349 for ethoxylated bisphenol A diacrylate. It is preferable to use the di(meth)acrylates of bisphenol A or F and of ethoxylated bisphenol A or of ethoxylated bisphenol F and mixtures thereof as the component c).

Preferably component c) has a MW of 300-1000.

Any type of photoinitiator which forms free radicals when irradiated suitably can be employed as the component d) in the mixtures according to the invention. Suitable classes of known photoinitiators include benzoin; benzoin ethers, for example benzoin methyl ether, ethyl ether and isopropyl ether, benzoin phenyl ether and benzoin acetate; acetophenones, e.g. acetophenone, 2,2-dimethoxyacetophenone and 1,1-dichloroacetophenone; benzil; benzil ketals, e.g. benzil dimethyl ketal and benzil diethyl ketal; anthraquinones, e.g. 2-methylantraquinone, 2-ethylantraquinone, 2-tert-butylantraquinone, 1-chloroanthraquinone and 2-amylantraquinone; triphenylphosphine; benzoylphosphine oxides, e.g. 2,4,6-trimethylbenzoyldiphenylphosphine oxide (Luzirin TPO); benzophenones, e.g. benzophenone and 4,4-bis-(N,N-dimethylamino)-benzophenone; thioxanthenes and xanthenes; acridine derivatives; phenazine derivatives; quinoxaline derivatives and 1-phenyl-1,2-propanedione-2-O-benzoyl oxime; 1-aminophenyl ketones and 1-hydroxyphenyl ketones, e.g. 1-hydroxycyclohexyl phenyl ketone, phenyl 1-hydroxyisopropyl ketone and 4-isopropylphenyl-1-hydroxyisopropyl ketone; all of which are known compounds.

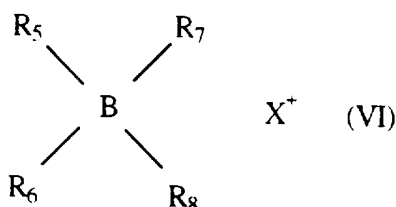
Photoinitiators which are particularly suitable for use when the actinic radiation is from a Hg-Cd laser are acetophenones, e.g. 2,2-dialkoxybenzophenones, and  $\alpha$ -hydroxyphenyl ketones, e.g. 1-hydroxycyclohexyl phenyl ketone and 2-hydroxyisopropyl phenyl ketone (=2-hydroxy-2,2-dimethylacetophenone).

A class of photoinitiators d) particularly suitable for argon ion lasers are benzil ketals, for example benzil dimethyl ketal, and especially an  $\alpha$ -hydroxyphenyl ketone, benzil dimethyl ketal or 2,4,6-trimethylbenzoyldiphenylphosphine oxide.

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Another class of suitable photoinitiators d) are the ionic dye-counter ion compounds which are capable of absorbing actinic radiation and producing free radicals which initiate polymerisation of the components.

The mixtures according to the invention containing ionic dye-counter ion compounds can be cured in a fairly variable manner in this way with visible light having a wavelength of 400-700nm. Ionic dye-counter ion compounds and their mode of action are known, for example from EP-A-0,223,587 and US Patents 4,751,102, 4,772,530 and 4,772,541. Examples of suitable ionic dye-counter ion compounds which may be mentioned are the anionic dye-iodonium ion complexes, the anionic dye-pyrylium ion complexes and especially, the cationic dye-borate anion compounds of the formula VI:



wherein  $X^+$  is a cationic dye and  $R_5$ ,  $R_6$ ,  $R_7$  and  $R_8$  are each independently alkyl, aryl, alkaryl, allyl, aralkyl, alkenyl or alkynyl group, an alicyclic group or a saturated or unsaturated heterocyclic group.

The photoinitiators are added in effective amounts, i.e. in amounts of about 0.1 to about 10 parts by weight, relative to the total amount of the components a) to d). If the mixtures according to the invention are used for stereolithographic processes in which laser radiation is used, it is preferred that the absorption capacity of the compositions is so adjusted by means of the type and concentration of the photoinitiator that the depth of curing at normal laser speed is approximately 0.1 to 2.5mm.

Compositions according to the invention can contain a plurality of photoinitiators which have a different radiation sensitivity at different wavelengths. This achieves, for example, better utilisation of a UV/VIS light source which radiates emission lines of different wavelengths. It is advantageous in this case if the various photoinitiators are so chosen and employed in such a concentration that a uniform optical absorption is produced in the case of the emission lines used.

As will be understood from the foregoing description and from the Examples, the specified amounts of components a) to d) refer to the total number of parts by weight of each of the defined component types. For example a mixture of 18 parts of an ethoxylated bisphenol A diacrylate and 18 parts of ethoxylated bisphenol A dimethacrylate constitute 36 parts in total and as such satisfy the definition for component C).

To the compositions of the invention there may be added small amounts

(0 to 4 percent by weight, preferably 0 percent) of the diacrylate and dimethacrylate esters of aliphatic or cycloaliphatic diols. Examples of such diols are 1,3-butylene glycol, 1,4-butanediol, neopentyl glycol, 1,6-hexanediol, diethylene glycol, triethylene glycol, tetraethylene glycol, 5 polyethylene glycol 400, polyethylene glycol 600, tripropylene glycol, ethoxylated or propoxylated neopentyl glycol, 1,4-dihydroxymethylcyclohexane, 2,2-bis-(4-hydroxycyclohexyl)-propane or bis-(4-hydroxycyclohexyl)-methane. However, it is preferred that the composition is free from di(meth)acrylate esters of aliphatic and 10 cycloaliphatic diols.

If desired one may add customary additives to the compositions, for example stabilisers, e.g. UV stabilisers, polymerisation inhibitors, mould release agents, wetting agents, flow control agents, infra-red absorbers, sensitisers, anti-sedimentation agents, surface-active agents, dyes, 15 pigments and fillers.

Compositions according to the invention can be prepared in a known manner, for example by premixing individual components and subsequently mixing these premixes or by mixing all the components by means of customary devices, such as stirred vessels, in the absence of light and, if 20 appropriate, at a slightly elevated temperature.

Liquid compositions according to the invention can be polymerised by irradiation with actinic light, for example, by means of electron or X-ray beams or UV or VIS light, e.g. by means of radiation within the wavelength range from 280 to 650nm. Laser radiation from HeCd, argon ions, or 25 nitrogen ions and also metal vapour and NdYAG lasers of multiplied frequency are particularly suitable. It is known to those skilled in the art that a suitable photoinitiator must be selected and, if appropriate, sensitised for each light source selected.

The invention also relates to a process for polymerising liquid 30 compositions according to the invention by irradiating them with actinic light.

Liquid compositions according to the invention preferably have a viscosity of 300 to 3000 mPa.s at 30°C, more preferably 400 to 1300 mPa.s and especially 500 to 1200 mPa.s. Compositions according to the invention 35 have surprisingly low viscosity and this enables fast processing times in stereolithography. The low viscosity is achieved without adversely affecting physical or mechanical properties of articles produced.

The invention also relates to a process for the production of a three-dimensional solidified article from the liquid compositions according 40 to the invention, preferably by stereolithography. Preferably this entails (a) the surface of a layer of the liquid composition according to the invention being irradiated either as the whole surface or in a predetermined pattern, by means of a UV/VIS (i.e. ultraviolet or visible)





light source, such that a layer is solidified in a desired layer thickness in the irradiated areas; (b) then a new layer of a composition according to the invention is formed on the solidified layer, and this is also irradiated either as the whole surface or in a predetermined pattern; and (c) by repeating steps (a) and (b) a three-dimensional article composed of several solidified layers adhered to one another is obtained.

The number of times steps (a) and (b) are repeated depends on the thickness of the resultant solidified layers and the size of the article. Thus steps (a) and (b) could be repeated 10 times if each solidified layer was 1mm deep and the article 1cm high and 500 times if each solidified layer was 0.5mm deep and the article 25cm high. Preferably the solidified layers each independently have a depth of 0.1 to 1mm, more preferably 0.2 to 0.6mm, especially 0.25 to 0.4mm. The solidified layers do not need to all be of the same depth. Repetition of steps (a) and (b) from 10 to 10,000, preferably 20 to 2000 times, forms one aspect of the invention.

The process for forming a three-dimensional article preferably uses a stereolithography apparatus, for example the SLA 250 or 500 supplied by 3D-Systems or the Stereos 300, 400 and 600 supplied by EOS.

There is no particular limit on what the three dimensional article can be, for example one may use the process to form ornamental and industrial articles and models of plant and animal parts. Industrial articles include mechanical parts, especially those used in automobiles, and models and prototypes thereof. Animal parts include bones, organs, tissues and combinations thereof. Examples of bones include joints (e.g. ball and socket joints such as the hip and shoulder, hinge joints such as the knee and elbow) the skull, jaw, spine, ribs, collarbone, shoulder blade, humerus, radius, ulna, teeth, finger and hand bones, breast bone, femur, tibia and fibula. Examples of organs include the liver, heart, lungs, kidneys, bladder, brain, eyes, intestines, pancreas and reproductive organs. Examples of tissue include muscle and cartilage.

As desired the three dimensional article can be a model which is the same size, smaller or larger than the original article. The low curl distortion and high rate of cure mean that such articles can be prepared quickly and accurately.

It is preferable to use a laser beam as the radiation source in this process.

Compositions according to the invention may also be used as coating agents; clear and hard coatings can be obtained on wood, paper, metal, ceramics or other surfaces. The coating thickness can be varied between wide limits, for example from 1 micrometer to 1mm. Relief images for printed circuits or printing plates can be produced direct from compositions according to the invention by irradiating the mixtures, for example by means of a computer-controlled laser beam of suitable wavelength.



or using a photomask and a corresponding light source.

The curl factor is determined on test specimens produced by stereolithographic processes, the deformation of a self-supporting part of the test specimen being determined by shrinkage. The curl factor is the ratio of the height of a deformed, fixed segment of the test specimen to the height of the non-deformed segment.

The invention is further illustrated by the following examples in which all parts and percentages are by weight unless stated otherwise.

Viscosities were measured at 30°C on a Brookfield viscometer RVT-11 using spindle 27 at 100 rpm.

The Tensile moduli, strength at break and elongation at break were measured using tensile testing bars prepared to ISO 527 and tested on an INSTRON 1122 fitted with a 1KN load cell.

Surface tension was measured using a torsion balance.

Linear shrinkage (L.S.) on cure was calculated by measuring the densities of the resins before and after curing. Linear shrinkage (%) was calculated by the equation:

$$L.S. = \left[ 1 - \sqrt[3]{\frac{\text{uncured density}}{\text{cured density}}} \right] \times 100\%$$

Processing parameters  $E_c$  and  $D_p$  mean respectively critical cure energy ( $\text{mJcm}^{-2}$ ) and penetration depth (mils) and were measured using a 3D systems SLA250 stereolithography apparatus.

The rate of cure was measured using the Real Time Infra-Red technique described by A. K. Davies in Radiation Curing Polymers 2, special publication No 89, Royal Society of Chemistry, ISBN 0-85186-377-9.

The following abbreviations are used in the Examples:-

SR-348: Ethoxylated bisphenol A dimethacrylate of MW 452, supplied by Sartomer.

SR-349: Ethoxylated bisphenol A diacrylate of MW 424 supplied by Sartomer.

SR-9035:  $\text{CH}_2\text{CH}_2\text{C} - [\text{CH}_2\text{O}(\text{CH}_2\text{CH}_2\text{O})_n\text{COCH}=\text{CH}_2]$  wherein  $n = 5$  of MW 956 supplied by Sartomer.

NR-2720: A urethane acrylate of formula  $(\text{H}_2\text{C}=\text{CHCO}_2\text{CH}_2\text{CH}_2(\text{OCOC}_6\text{H}_4)_n\text{OCONH}-\text{C}_6\text{H}_4-)_m-\text{CH}_3$  wherein  $n = 2$  of MW 950 supplied by Zeneca Resins US.

DAR1173: U.V. Photoinitiator of formula  $\text{Ph-CO-C}(\text{CH}_3)_2\text{OH}$ .



IRG: Irgacure 651, a U.V. Photoinitiator from Ciba Geigy.  
EB-270: Aliphatic urethane acrylate of MW approximately 1500 supplied by UCB Chemicals.  
DC190: a silicone type surfactant supplied by Dow Corning Ltd.

5 Examples 1 to 6

Liquid compositions according to the invention were prepared by stirring together at room temperature the number of parts by weight of components indicated in Table 1 below. After stirring for a few hours the homogenous mixtures were transferred into separate bottles for storage.

10 Table 1

Example No.	SR-349	SR-348	NR-2720	SR-9035	DAR 1173	DC190	IRG
1	8.0	42.0	34.5	10.0	5.5	---	---
2	---	45.0	34.5	15.0	5.5	---	---
15 3	8.0	42.0	34.5	10.0	5.5	+0.2	---
4	---	45.0	34.5	15	5.5	+0.2	---
5	20	35	29	10.5	5.5	---	---
6	20.5	35.9	29.8	10.8	---	---	3

20 For comparison commercially available liquid compositions were obtained from Ciba-Geigy (XB5143) and DuPont (SOMOS 3110).

Testing

The viscosity, tensile modulus, strength at break, elongation and surface tension of the exemplified compositions and the two commercially available resins XB5143 and SOMOS 3110 were measured as described above.

25 The results are shown in Table 2 below:-

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Table 2

Example No	Viscosity @ 30°C mPa.s	Tensile Modulus MPa	Strength @ break MPa	elongation %	surface tension Dyne cm <sup>-1</sup>
1	970	880	34	8	44
2	830	620	28	11	44
3	960	850	33	8	27
4	825	630	28	12	30
5	810	840	33	8	45
6	1100	953	44	8	44
XB 5143*	1900	600	24	8	40
SOMOS 3110*	735	960	43	9	41

15 \* Comparative

The linear shrinkage on cure, shore hardness, Ec, Dp, rate of cure and curl factor were measured as described above. The results are shown in Table 3 below:-

Table 3

Example No	Linear shrinkage on cure %	Shore hardness "D"	E. mJ cm <sup>2</sup> (He-Cd)	D. mils (He-Cd)	Rate of cure secs	Curl Factor
1	1.93	74	--	--	60	10%
2	1.88	70	6.1	5.1	53	8%
3	1.95	75	--	--	58	--
4	1.90	70	--	--	54	--
5	1.77	75	5.8	4.9	53	10%
6	---	75	7.5	6.2	55	10%
XB 5143*	2.03	73	4.5	6.0	62	15%
SOMOS 3110*	2.12	80	2.5	5.1	150	32%

\* Comparative

Examples 7 to 19

Further liquid compositions may be prepared having the formulations given in Table 4 below:

Table 4

Example No.	SR-349	SR-348	NR-2720	SR-9035	DAR 1173	IRG	EB-270
7	15	46	24	9.5	5.5	-	-
8	10	40	25	19.5	5.5	-	-
9	19	44	21	11	5	-	-
10	4.5	40	40	10	5.5	-	-
11	-	42	42	10.5	5.5	-	-
12	-	49.9	30.2	14.4	5.6	-	-
13	-	45.5	34.5	15.0	5.0	-	-
14	55.0	-	29	10.5	5.5	-	-
15	55.0	-	29	10.5	-	5.5	-
16	56.6	-	29.8	10.8	-	2.8	-
17	20.5	35.9	29.8	10.8	-	3.0	-
18	10.9	36.6	-	12.5	5.5	-	34.5
19	-	50	-	19.5	5.5	-	25.0

CLAIMS

1. According to the present invention there is provided a liquid composition comprising:

a) 2 to 20 parts of a monomeric poly(meth)acrylate having a functionality of at least 3 and a MW of at least 650;

b) 20 to 60 parts of a urethane(meth)acrylate having a functionality of 2 to 4 and a MW of 400 to 10,000;

10 c) 20 to 80 parts of a monomeric or oligomeric di(meth)acrylate based on bisphenol A or bisphenol F; and

d) 0.1 to 10 parts of a photoinitiator;

wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100.

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2. A liquid composition according to Claim 1 wherein component a) has a MW in the range 880-1200.

3. A liquid composition according to anyone of the preceding claims wherein component a) is present in an amount of 5 to 18 parts.

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4. A liquid composition according to anyone of the preceding claims wherein component b) is present in an amount of 20 to 50 parts.

5. A liquid composition according to anyone of the preceding claims wherein component c) is present in an amount of 35 to 55 parts.

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6. A liquid composition according to anyone of the preceding claims wherein component d) is present in an amount of 2 to 8 parts.

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7. A liquid composition according to Claim 1 which comprises or consists essentially of 8 to 16 parts of component a); 25 to 45 parts of component b); 40 to 50 parts of component c); and 3 to 7 parts of component d); wherein all parts are by weight and the total number of parts of a) + b) + c) + d) add up to 100.

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8. A liquid composition according to any one of the preceding claims wherein the monomeric or oligomeric di(meth)acrylate based on bisphenol A or bisphenol F is a mixture consisting of ethoxylated bisphenol A diacrylate and ethoxylated bisphenol A dimethacrylate.

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9. A liquid composition according to any one of the preceding claims having a viscosity of 300 to 3000 mPa.s at 30°C.

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10. Use of a liquid composition according to any one of claims 1 to 8 for the preparation of a three-dimensional solidified article.

5 11. A process for the preparation of a three-dimensional solidified article from a liquid composition by stereolithography characterised in that the liquid composition is as defined in any one of claims 1 to 9.

12. A process for the production of a three-dimensional solidified article wherein:

- 10 (a) the surface of a layer of a liquid composition according to any one of Claims 1 to 9 is irradiated either as a whole surface or in a predetermined pattern, by means of an ultraviolet or visible light source, such that a layer is solidified in a desired layer thickness in the irradiated areas;
- 15 (b) then a new layer of the composition is formed on the solidified layer, and this is also irradiated either as a whole surface or in a predetermined pattern; and
- (c) by repeating steps (a) and (b), a three dimensional article composed of several solidified layers adhered to one another is obtained.

20 13. A three-dimensional solidified article prepared from a liquid composition according to any one of claims 1 to 9 or by a process according to claim 11 or 12.

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# INTERNATIONAL SEARCH REPORT

national Application No  
PCT/GB 96/00518

A. CLASSIFICATION OF SUBJECT MATTER  
IPC 6 G03F7/027 G03C9/08

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 G03F G03C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP,A,0 425 441 (CIBA-GEIGY AG) 2 May 1991 cited in the application see page 6, line 24 - line 25 see page 7; example 4; table 1 ---	1-13
A	EP,A,0 554 215 (CIBA-GEIGY AG) 4 August 1993 see page 11; table 1 -----	1-13

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents :

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- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
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- \*&\* document member of the same patent family

Date of the actual completion of the international search

21 June 1996

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19.07.96

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**INTERNATIONAL SEARCH REPORT**  
Information on patent family members

International Application No  
PCT/GB 96/00518

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0425441	02-05-91	CA-A- 2028541 DE-D- 59010008 JP-A- 3160013	28-04-91 08-02-96 10-07-91
EP-A-0554215	04-08-93	CA-A- 2088031 JP-A- 5286897	28-07-93 02-11-93