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[54] Title: ORAL COMPOSITIONS DIANE CUMMINS, of Wirral and FRANCISCUS JOHANNES GERRIT VAN DER CUDERAA, of Cheshire, both of Great Britain [75] Inventor (s): UNILEVER N.V., of Rotterdam, The "etherlands [78] Assignee (s): October 3, 1990 [22] Filed: 41306 Application Serial No: [21] FOREIGN APPLICATION PRIORITY DATA 8922434.9 [81] Number (s) October 5, 1989 [32] Date (s)

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ABSTRACT

Oral compositions such as dentifrices with an improved anti-plaque efficacy are obtained by inclusion therein of a mixture of a stanneus salt such as stanneus-fluoride or stanneuspyrophosphate and a sinc salt such as sinc citrate.

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ORAL COMPOSITIONS

Abstract of the disclosure

Unal compositions such as dentifices with an improved anti-plaque efficacy are obtained by inclusion therein of a mixture of a stannous, salt such as stannousfluoride or stannouspyrophosphate and a zinc salt such as zinc citrate.

The present invention relates to coal compositions such as dentifrices, mouthwashes, gels, subgingival rinse compositions, toothpastes, toothpowders, chewing gum, prophylactic pastes, lozenges, flosses, toothpicks which provide antiplaque benefits.

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In the prior art an abundancy of proposals has been made to obtain anti-plaque oral Many of these proposals compositions. (O hownver not resulted in a reasonably effective anti-plaque oral composition. One of the few really effective anti-plaque oral compositions is based upon the use of a zinc compound as an anti-plaque agent. This is more fully described in e.g. US Patent 4,022,880 (Vinson et al). 15 Another material which has been considered as anti-plaque agent is the stannous ion. This has e.g. been discussed in "Tooth Surface Interactions and Preventive Dentistry, IRL Press Ltd (London) 1981, pages 33-37, "The role of stannous 20 pyrophosphate in the plaque-inhibiting effect of dentifrices containing stannous fluoride" by Svatum and Rolla. Despite the many disclosures in the need for the anti-plaque area, improved anti-plaque products exists, which are 25

properly balanced with regard to efficacy and undesired possible adverse reaction in the mouth.

It has now been found that a stannous compound when used in combination with a zinc compound provides an improved anti-plaque efficacy. Consequently, in its broadest aspect the present invention relates to an oral composition with an improved anti-plaque activity, comprising a mixture of a stannous compound and a zinc compound as the anti-plaque active system.

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the present invention, can be any stannous compound with inorganic or organic counter ions. It can be a highly soluble stannous salt, or it can be a sparingly soluble stannous salt. Highly soluble stannous salt. Highly soluble stannous salts are e.g. stannous fluoride, stannous chloride, stannous chloride, stannous chloride, stannous fluoride, potassium stannous flouride, stannous flouride, stannous saltannous sulfate, stannous tartrate, stannous gluconate, disodium mono-stannous citrate etc. Of these highly soluble stannous salta stannous fluoride is the preferred stannous saltannous fluoride is the preferred stannous saltannous fluoride is the preferred stannous saltannous saltannous

Sparingly soluble stannous salts are e.g. 25 stannous pyrophosphate, stannõus metaphosphate, stannous exalate stannous phosphate, distannous citrate etc. Stannous pyrophosphate is a preferred sparingly soluble stannous salt. Mixtures of various highly soluble stannous salts may also be used, as well as mixtures of various sparingly soluble stannous salts and mixtures of highly and sparingly soluble stannous salts. A preferred mixture is the mixture of stannous fluoride and stannous pyrophosphats.

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Although highly soluble stammus salts can be used in the present invention, they tend to be not sufficiently stable upon storage. The stannous ions, dissolved in an aqueous solution tend to be converted therein to inert tin compounds, which do not provide for a reasonable anti-plaque activity. Therefore, if a highly soluble stammous salt is used, care should be taken to reduce the quantity of active dissolved stannous ions during storage of the oral composition, or to stabilize the stannous ions by other means.

When using a sparingly soluble stannous salt, care should be taken that there is a sufficient level of active dissolved stannous ions in the composition without giving rise to precipitation thereof as e.g. stannous oxide or stannous oxide

hydrate. One way of achieving this is by solubilising the stannous salt, e.g. the stannous pyrophosphate with a certain amount of an alkalimetal pyrophosphate, or an alkalimetal citrate, or a fluoride source.

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In general, the stannous salt is used in such an amount in the oral composition, that there is an effective amount of active dissolved stannous ions available in the composition to achieve an anti-plaque efficacy. For the highly reliable stannous salts this amount will generally range from 0.01-10%, preferably from 0.02-5 and particularly preferably from 0.1-3% by weight of the oral composition. As regards the sparingly soluble stannous salts these ranges are 0.05-10, preferably 0.1-5 and particularly 0.1-5 by weight of the oral composition.

The zinc compound, suitable for use in the present invention can be any highly soluble or sparingly soluble zinc compound having inorganic or organic counter ions. Suitable examples of such zinc salts are enumerated in US Patent 1,022,880 (Vinson et al), which is hereby incorporated by way of reference. () preferred zinc salt is zinc citrate trihydrate. In general,

the amount of zinc salt used in the present invention ranges from 0.05-5% (calculated as zinc ion). preferably from 0.1-4% and particularly preferably from 0.1-3% by weight of the oral composition.

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The oral composition of the present invention may contain an orally acceptable medium which contains usual additional ingredients in conventional amounts, depending upon the final 10 form of the composition, i.e. a dentifrice, a mouthwash, a gel and the like. Thus, as dentifrice it will usually comprise an abrasive cleaning agent in an amount of from 3-75% by weight. Suitable abrasive cleaning agents are milled or unmilled particulate aluminas; silica 1.5 xerogels, hydrogels and aerogels and precipitated particulate silicas; calciumpyrophosphate; insoluble sodium metaphosphate; calcium carbonate; particulate orthophosphate: dicalcium hydroxyapatite and so on. 20

Furthermore, the dentifrice may contain a liquid phase comprising water and a humectant in an amount of 10-79% by weight. Typical humectants are glycerol, sorbitol, polyethyleneglycol, polypropylene glycol, propylene glycol,

hydrogenated partially hydrolyzed polysaccharides and so on.

Rinders or Thickening agents such as sodium carboxymethylcellulose, hydroxypthylcellulose, bydroxypropylcellulose, xanthan gums, trish moss, qum tragacenth, finely-divided silicas and hectorites may also be included in the dentifrice in an amount of 0.5 - t0% by weight. Another conventional ingredient in a dentifrice is an organic surfactant such as a soap, on anionic, nonionic, cationic, ampholytic and/or a switterionic synthetic detergent surfactant in an amount of 0.2-5% by weight.

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When the composition is in the form of a mouthwash, it will usually contain an alcohol, a solubilizer, and when in the form of a gel it will usually contain a thickening agent.

Various other optional ingredients may be included in the compositions of the invention, such as flavouring agents, sweetening agents such as sodium saccharinate, whitening agents such as titanium dioxide or zinc oxide, preservatives, vitamins such as vitamin C and E, other antiplaque agents such as copper salts, sanguinarine, allantoin, p-aminobenzoic acid, derivatives,

hexetidine, chlorhexidine, 3-(4-propylhoptyl)-4-(2-hydroxyethyl)-morpholine, anti-bacterial agents such as Triclosan (2',4,4'-trichloro-2hydroxy-diphenyl ether), anti-calculus agents such as di- and/or tetra- alkalimetalpyrophosphates, pH adjusting agents, colouring agents, anti-caries agents such as casein, rasein digests, sodium trimetaphosphate, sodium fluoride and monospdiumfluorophosphate, anti-staining compounds such as silicone polymers, anti-inflammatory agents such as substituted salicylamilides, plant extracts, desensitizing agents for sensitive teeth such as polassium nitrate and polassium citrate, polymers polyvinylmethylether-maleic anhydriide copolymers and so on.

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The compositions of the present invention not only provide for an improved anti-plaque efficacy, but also have an anti-gingivitis and an anti-calculus benefit. The mixture of the stannous salt and the zinc salt can also be used in the manufacture of a medicament against gingivitis. The mixture also has an improved anti-microbial effect on the oral flora. The stannous salt and zinc salt may be used in the same phase of the oral composition, or they may each be present in a



separate phase, e.g. one of them may be present in the stripe phase of a so-called striped toothpaste and the other one may be present in the main phase of such a striped toothpaste. When a fluoride cource is also present in the composition, this may also be present in the phase, separate from the stannous salt containing phase.

The oral compositions of the present invention can be formulated to any desirable to pH-value. It is preferred that the compositions have a pH of between 3.5 and 5.5.

The present invention will now be further illustrated by the following Examples.

Example 1

the effectiveness of the dentifrice
compositions of this invention in inhibiting the growth of plaque on the teeth was determined by following a standard procedure for the measurement of plaque growth. The methodology of measuring plaque growth is that according to Harrap as described in J. Clin. Periodontol., 1974, 1,
166-174 which gives a procedure for assessing the amount of plaque on the teeth adjacent to the gingival margin. The procedure is as follows:

25 During the late afternoon, each subject



paste (having a composition as given hereinafter) for an unspecified period of time to remove as much plaque as possible. This is immediately followed by brushing for one minute with 1.5g of the allocated test paste. Residual paste is removed by rinsing the mouth with water and any remaining plaque disclosed by painting the teeth with an aqueous solution of Erythresin (0.5% w/w) using a soft camel hair brush. Excess dye is removed by rinsing with water and the amount of plaque assessed and recorded for each of 16 teeth (numbers 3 to 6 for each quadrant). The recorded plaque is designated P_Q.

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No further oral hygiene is permitted for 18 hours after which time each subject rinses his/her mouth with water to remove food debris and viscous saliva. Plaque assessment is then carried out as before and recorded (P_{18}) . The values of $P_{18}^{-P_0}$ for each tooth are averaged to give a $P_{18}^{-P_0}$ value per mouth. The mean of the values obtained for the subjects in the test is the plaque growth value. Panets of at least 12 subjects are used. The plaque growth value for a foothpaste without active ingredients is usually if the range 22 to

26. The plaque growth inhibition (PGI) is then computed for each test treatment by expressing the percentage inhibition compared to placebo:

 $\frac{\text{POT}}{\text{pT}} \approx \frac{\text{PO}}{\text{pT}} \text{pT} + \frac{\text{POY}}{\text{pT}} \text{For 100Y}$

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The composition of the simple, non-active toothpasts referred to above was the following:-

	Ingredient	%
	Alumina frihydrate	50.00
ŧo.	@lycerin	27.00
	Hydroxyethylcellulose	0.95
	Fitanium dioxide	0.50
	Water	

The following compositions were assessed as to their PG1 in accordance with the above test protocol.

Composition (in % by weight)

	; · · · ·	A	В	C	D	F	F
	Silica xerogel	14.0	14.0	14.0	14.0	14.0	14.0
	Silica aerogel	9.0	9.0	9.0	9.0	9.0	9.0
5	Sorbitol syrup(70%)	45.0	45.0	45.0	45.0	45.0	45.0
	Polyethyleneglycol (MW 1500)	5.0	5.0	5. 0	5.0	5.0	5.0
	Xanthan gum	0.6	0.6	0.6	0.6	0.6	0.6
	Saccharin	0.23	0, 23	0.23	0.23	0.23	0. 23
10	Benzoic acid	0.19	0.19	0.19	0.19	0.19	0.19
	Titanium dioxide	1.0	1.0	1.0	1.0	1.0	1.0
	Sodium lauryl- sulphate	1.5	1.5	1.5	1.5	1.5	1.5
	Flavour	1.0	1.0	1.0	1.0	1.0	ì.o
15	Sodium fluoride	-	_	ü	0.33	-	0. 33
	Monosodium fluorophosphate	1.1	1.1	1.1	-	1.1	•
	Stannouspyro- phosphate	1.0	2. 0	-	1.0	1.0	1.0
20	Zinc citrate trihydrate	-	-	0. 5	-	0.5	0. 5
	Water	to 100	to 100	to 100	to 100	to 100	to 100
	PGI-values	26	16	14/9	0	•	-
25		25 24	· •	- 3 0	- 50	32 0; 37; 41	34
		26	<u>-</u>	-	0	42;38	-

Example 2

The following formulations were fested as to their plaque growth inhibition effect in the manner as described in Example 1.

5	Composition (% by weight)					
		6	Н	ð		
	Alumina	54.25	54.75	55.25		
	Sorbital (70%)	27	27	27		
	Xanthan gum	0.88	0.83	0.88		
10	litanium dioxide	0.5	0.5	0.5		
	Sodium laurylsulphate	1.49	1.5	1.5		
	Saccharin	0.23	0.23	0.23		
	Benzoic acid	0.19	0,17	0.19		
	Flavour	1.0	1.0	1.0		
15	Stannous pyrophosphate	e 10	t.0	1.0		
	Monosodium fluorophosphate	1.1	1.1	1.1		
	Zinc citrate trihydrate		agent	0.5		
	Water	to 100	to 100	to 100		
	PGI value (mean)	O	23	33		

20 Again the anti-plaque efficacy of the composition of the invention (J) was superior to that of the comparative formulations (B, H).

Example 3,

the fullowing formulations were made, and their F6) values determined in the manner as

described in Example 1.

Composition (% by weight)

		K	I.
	Silica xeregel	10.50	10.50
5	Silina serogel	10.00	10.00
	Sarbital (70%)	67.87	67.95
-	Polyethyleneglycol (MW 1500)	50	5.0
	Ethenol	1.8	1.8
	Sodium laurylsulphate	1.47	1.47
10	Flavour	0.77	0.27
	Sódium carboxymethylcellulose	0.3	0.3
	Sodium saccharin	0.3	0.3
	Colouring agent	0.15	0.15
	Sodium benzoato	0.08	0.08
15	Flavour enhancer	0.4	0.4
	Sadium bydroxide (50% solution)	0.25	***
	Stannous fluoride	0.46	
	Zinc citrate trihydrate	0.50	0.50
•	Standous pyrophosphate	<u> </u>	1.00
20	Sodium fluoride	~ .	0.25
	Water	to 100	to 100
	PGI-value (muan)	30	23

Example 4,

The reduction in plaque and gingivitis of three formulations was investigated in two 21 days



experimental gingivitis studies, in the manner as described by C.A. Saxton, "The effect of dentifrice containing zinc citrate and Triclosan in developing gingivitis", Journal of Periodontal Eesearch 24 (1989) page 75. The formulations tested in study I were formulations E and C, and in study II formulation F was tested as well as the following formulation M.

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H (in % by weight)

£O.	Gilica merogel	(60,03)
	Silica aeroget	8.50
	Sorbital (70%)	45.00
	Palyathyleneglycol (MW 1500)	5.00
÷	Sodium laurylsulphate	1.5
15	Titanium dioxide	1.0
	Sodium carboxymethylcellulose	0.9
	Saccharin	0.2
	Flavour	1.0
	Monosodium fluorophosphate	0.80
20	Zinc citrate trihydrate	0.50
	Stannouspyrophosphate	1.00
	friclosan	0.50
	Water	to 100.

The results of the studies were as follows:

Plaque reduction Gingivitis reduction
Formulation

		Study 1	Study II	Study 1	Study II
55	c	9%		19%	- 100
	E	19%	27%	43%	47%
	71	·,	39%		62%

(- = not tested in the study)

These data show a clearly superior anti-10 plague and anti-gingivitis efficacy of compositions according to the present invention.

Емамріе 5

The relative anti-plaque activities of toothpaste formulations N, O and P were assessed using a 48-bour plaque screening model. 1.5 Formulation N was similar to formulation K of Example 3, save that it did not contain zinc citrate; Formulation O was identical to Formulation K and Formulation P was similar to Formulation L of Example 3, save that it contained 20 0.46% stannous fluoride instead of 0.25% sodium fluoride. Studies were conducted in a double blind manner, with neither examiner nor panelists having knowledge of the product identity. Panelists were required to meet certain entrance

criteria in order to be included in the study.

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Formulation

Panelists received a full mouth supragingival prophylaxis and scaling. Panelists were then instructed to refrain from all onal hygiene measures, except use of assigned test products, for the next 48 hours. Treatments were performed twice a day, in the morning (supervised) and in the evening, for two days. The following day the panelists used a disclosing solution and were then examined for plaque on the Ramford teeth using the DMPI (= Distal Mesial plaque index) plaque scoring system.

Panelists used 15 milliliters of a 25% toothpaste slurry for each treatment. Treatment slurries

% plaque growth inhibition

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The following results were obtained:

20 N 24.3

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CLAIMS

1. An oral composition with improved anti-plaque efficacy, comprising, in an orally acceptable medium, an anti-plaque active system which contains a mixture of 0.01-10% by weight, based on the total composition, of a stannous salt and 0.05-5% by weight (catculated as zinc ion), based on the total composition, of a zinc salt.

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- 2. A composition according to claim t, wherein the anti-plaque active system contains from 0.02 to 5% by weight of a highly soluble standous salt.
 - 3. A composition according to claim 1, wherein the stannous salt is stannous fluoride.
- 4. A composition according to claim 1,

 15 wherein the anti-plaque active system contains

 from 0.1 to 5% by weight of a sparingly soluble

 stanoous salt.
- 5. A composition according to claim 1,
 wherein the stannous salt is stannous
 20 pyrophosphate.
 - 6. A composition according to claim 1, wherein the anti-plaque active system contains a mixture of stannous fluoride and stannous pyrophosphate.
- 25 7. A composition according to claim 1,

wherein the auti-plaque active system contains 0.1-3% by weight (calculated as zime ion) of the zinc salt.

8. A composition according to claim 1,
 5 wherein the zinc salt is zinc citrate trihydrate.

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