

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
3 September 2009 (03.09.2009)

(10) International Publication Number
WO 2009/108158 A1

(51) International Patent Classification:

A01N 59/16 (2006.01) *A01P 1/00* (2006.01)
A01N 37/02 (2006.01) *C08K 13/02* (2006.01)

(21) International Application Number:

PCT/US2008/004175

(22) International Filing Date:

31 March 2008 (31.03.2008)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

12/072,486 26 February 2008 (26.02.2008) US

(71) Applicant (for all designated States except US): **MO-MENTIVE PERFORMANCE MATERIALS INC.** [US/US]; 187 Danbury Road, Wilton, CT 06897 (US).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **ZHU, Aijun** [CN/US]; 13832 Piedmont Cove, Fort Wayne, IN 46845 (US). **WANG, Yi-Feng** [CA/US]; 21 Stablegate Drive, Clifton Park, NY 12065 (US).

(74) Agents: **DILWORTH, Peter, G.** et al.; Dilworth & Barrese, LLP, 33 Earle Ovington Boulevard, Uniondale, NY 11553 (US).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MT, NL, NO, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report (Art. 21(3))



WO 2009/108158 A1

(54) Title: SILICONE RUBBER EXHIBITING EFFECTIVE ANTIMICROBIAL ACTIVITY

(57) **Abstract:** A silicone rubber exhibiting antimicrobial efficacy contains: a) at least one silicone rubber derived from a curable silicone rubber-forming composition; b) at least one silver-containing antimicrobial agent incorporated in silicone rubber (a) in X weight percent amount; and, c) at least one carboxylic acid incorporated in silicone rubber (a) in Y weight percent amount, silver-containing antimicrobial agent (b) when incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein and carboxylic acid (c) when incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein imparting no significant antimicrobial activity to silicone rubber (a) but in the combination therein of up to X weight percent silver-containing antimicrobial agent (b) and up to Y weight percent carboxylic acid (c) imparting significant antimicrobial activity to silicone rubber (a), such activity being exhibited at the interface of an exposed surface of silicone rubber (a) and a microbe-populated aqueous material in contact therewith.

SILICONE RUBBER EXHIBITING EFFECTIVE ANTIMICROBIAL ACTIVITY

BACKGROUND OF THE INVENTION

[0001] This invention relates to a silicone rubber exhibiting effective antimicrobial activity in a microbe-populated aqueous environment to which a surface of the rubber is exposed.

[0002] Silver-containing antimicrobial agents have been incorporated in a variety of synthetic resins in order to provide products which exhibit antimicrobial characteristics. Although silver-based antimicrobial agents may provide suitable antimicrobial properties for some kinds of resins, it has been discovered that silicone rubbers containing silver-type antimicrobial agents often fail to exhibit significant antimicrobial activity. This lack of antimicrobial effectiveness may be due to there being relatively low amounts of silver available at the surface of the rubbers. Whatever the actual cause may be for the observed lack of efficacious antimicrobial activity in silicone rubbers containing known types of silver-based antimicrobial agents, it remains that a need exists for imparting effective levels of antimicrobial activity to silicone rubbers which are intended to utilize silver-containing antimicrobial agents that by themselves are inefficacious.

SUMMARY OF THE INVENTION

[0003] In accordance with the present invention, there is provided a silicone rubber exhibiting antimicrobial efficacy which comprises:

- a) at least one silicone rubber derived from a curable silicone rubber-forming composition;
- b) at least one silver-containing antimicrobial agent incorporated in silicone rubber (a) in X weight percent amount; and,
- c) at least one carboxylic acid incorporated in silicone rubber (a) in Y weight percent amount,
silver-containing antimicrobial agent (b) when incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein and carboxylic acid (c) when

incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein imparting no significant antimicrobial activity to silicone rubber (a) but in the combination therein of up to X weight percent silver-containing antimicrobial agent (b) and up to Y weight percent carboxylic acid (c) imparting significant antimicrobial activity to silicone rubber (a), such activity being exhibited at the interface of an exposed surface of silicone rubber (a) and a microbe-populated aqueous material in contact therewith.

[0004] Although not scientifically demonstrated at this time, it is believed that silver-containing antimicrobial substance (b), which lacks appreciable antimicrobial activity when present by itself in silicone rubber (a), exhibits a significant level of antimicrobial activity when carboxylic acid (c) is additionally present in the rubber due to the ability of the carboxylic acid to increase the availability of antimicrobial silver at the surface of the rubber, perhaps by facilitating or improving the migration of silver from within the rubber to its surface.

[0005] Whatever the actual mechanism involved, it remains that silver-containing antimicrobial agents that are known to be effective in a variety of synthetic resins but not in silicone rubber can be rendered efficacious when accompanied by a carboxylic acid.

[0006] The expressions "silicone rubber" and "silicone elastomer" are to be regarded herein as synonymous.

[0007] The expression "silver-containing antimicrobial agent" shall be understood herein to mean silver in any of its forms that exhibit antimicrobial activity in a microbe-populated aqueous environment and includes metallic silver, e.g., colloidal silver and nanosilver, silver compounds providing silver ions in aqueous media, complexes of silver, and the like.

[0008] The expression "carboxylic acid" shall be understood herein to include mono-, di-, tri-, etc., aliphatic, cycloaliphatic and aromatic carboxylic acids and the anhydrides thereof where such exist as well as mixtures of two or more of any of the foregoing.

[0009] Other than in the working examples or where otherwise indicated, all numbers expressing amounts of materials, reaction conditions, time durations, quantified properties of

materials, and so forth, stated in the specification and claims are to be understood as being modified in all instances by the term "about."

[0010] It will also be understood that any numerical range recited herein is intended to include all sub-ranges within that range and any combination of the various endpoints of such ranges or subranges.

[0011] It will be further understood that any compound, material or substance which is expressly or implicitly disclosed in the specification and/or recited in a claim as belonging to a group of structurally, compositionally and/or functionally related compounds, materials or substances includes individual representatives of the group and all combinations thereof.

DETAILED DESCRIPTION OF THE INVENTION

[0012] This invention encompasses a silicone rubber containing at least one silver-based antimicrobial agent that by itself exhibits little if any antimicrobial activity and at least one carboxylic acid in an amount sufficient to potentiate the antimicrobial activity of the silver-based antimicrobial agent.

[0013] This invention also encompasses a method of forming a silicone elastomer comprising the steps of providing a curable silicone rubber-forming composition, introducing therein at least one silver-containing antimicrobial agent in an amount of from 0.1 to 10, preferably from 0.2 to 5 and more preferably from 0.5 to 2, percent by weight, and at least one carboxylic acid in an amount of from 0.1 to 5, preferably from 0.2 to 3 and more preferably from 0.5 to 2, percent by weight, and subjecting the silicone rubber-forming composition to curing conditions to provide the silicone rubber.

[0014] Silicone rubber (a) can be obtained by curing any one of numerous known curable silicone rubber-forming compositions, e.g., liquid silicone rubber (LSR) compositions, room temperature vulcanizable silicone rubber (RTV) compositions and heat curable silicone elastomer (HCE) compositions.

[0015] A typical LSR composition is a multi-component combination of a vinyl-containing polydiorgansiloxane fluid, a hydrogen-containing polydiorganosiloxane fluid, an effective

amount of a platinum catalyst and a reinforcing filler such as a fumed silica and one or more additional additives. Typically, two-component LSR mixtures are mixed and charged into a preheated mold where they are rapidly cured to produce a silicone rubber article. A first component, or package, includes a vinyl-containing polydiorgansiloxane fluid, a silica filler and an effective amount of a platinum catalyst and a second component, or package, includes a hydrogen-containing polydiorganosiloxane fluid in combination with other ingredients including a vinyl-containing polyorganosiloxane fluid and a silica filler. Typically, the LSR composition is produced by kneading a polydiorganosiloxane, inorganic filler and additives by means of a kneading machine such as a Banbury mixer, a turbulizer, a change can mixer or a low intensity double arm dough mixer. In this process, polydiorganosiloxane, inorganic filler, treating agents and additives are batch mixed until the desired properties are obtained. The batch mixing process can take 12 to 30 hours per batch. After mixing, the LSR composition is stripped of volatiles and cooled. For additional details regarding LSRs and silicone rubbers obtained therefrom, reference may be made, *inter alia*, to U.S. Patent No. 6,444,154, the entire contents of which are incorporated by reference herein.

[0016] RTV silicone compositions may be formulated as one-component or two-component systems. A common form of one-component RTV silicone, curable by exposure to moisture, includes an organosiloxane polymer possessing terminal hydroxyl groups, a polyfunctional organosilicone cross-linking agent and a crosslinking catalyst, e.g., as disclosed in U.S. Patent Nos. 4,100,129, 4,593,085, 5,420,196, 5,932,650 and 6,737,494, the entire contents of which are incorporated by reference herein. Two-component RTV silicone systems typically comprise an "A" package containing a dihydroxy or silanol-terminated polydiorganosiloxane, a semi-reinforcing filler such as calcium carbonate or ground quartz, a reinforcing filler such as fumed silica and water and a "B" package containing a T or Q functional crosslinker and a condensation cure catalyst, e.g., as disclosed in U.S. Patent No. 4,490,500, the entire of contents of which are incorporated by reference herein.

[0017] HCE compositions generally consist of a diorganopolysiloxane gum, fluid, filler and, optionally, a curing agent such as an organic peroxide. For further details regarding HCE

silicone elastomers, reference may be made, *inter alia*, to U.S. Patent No. 6,245,875 and 6,750,279, the entire contents of which are incorporated by reference herein.

[0018] The curable silicone elastomer-forming composition can also contain one or more known and conventional additives, e.g., UV stabilizers, antioxidants, colorants, fillers, reinforcing agents, etc., incorporated in the usual amounts.

[0019] The silicone rubber of the invention is useful as, or in, many kinds of articles for which efficacious antimicrobial activity is a necessary or desirable property, e.g., medical and diagnostic instruments, catheters, wound care and medical dressing products, scar care management products, drug delivery systems, and the like.

[0020] Useful silver-containing antimicrobial agents include silver-containing inorganic compounds such as silver zirconium phosphates available from Milliken & Company under the tradename ALPHASAN.RTM. RC-2000, RC-5000 and RC-7000; silver-containing organic compounds such as silver carboxylates, e.g., silver benzoate, silver citrate and silver tartrate; silver-substituted zeolites available from Shingawa under the tradename ZEOMIC.RTM; silver-containing glasses available from Ishizuka Glass under the tradename IONPURE.RTM; AMP.RTM T558 and MICROFREE.RTM, both available from DuPont, and JMAC.RTM, available from Johnson Mathey; and, metallic silver materials such as colloidal silver, nanosilver and silver alloys.

[0021] Average particle size of the silver-containing antimicrobial agent can vary widely, e.g., from 5 to 30 nm in the case of nanosilver and from 5 microns to 100 microns for all other forms.

[0022] Carboxylic acid component (c) as previously defined can be selected from any of numerous known aliphatic, cycloaliphatic and aromatic carboxylic acids possessing one or more carboxylic acid groups, the anhydrides thereof where such exist, and mixtures thereof, provided the presence of a particular carboxylic acid, carboxylic acid anhydride or mixture of particular carboxylic acid(s) and/or carboxylic acid anhydride(s) in a given curable silicone rubber-forming composition does not adversely affect its cure or the properties of the cured rubber. In general, the carbon content of the useful carboxylic acids and carboxylic acid anhydrides will not exceed 14 carbon atoms when the carboxylic acids and anhydrides thereof are of the saturated aliphatic,

cycloaliphatic or aromatic variety and in the case of unsaturated aliphatic and cycloaliphatic carboxylic acids and carboxylic acid anhydrides where the presence of one or more olefinic bonds tends to lower the melting point and increase solubility, a carbon content that does not exceed 20 carbon atoms. Some useful carboxylic acids and carboxylic acid anhydrides that can be utilized herein include propanoic acid, 2-methyl propanoic acid, butanoic acid, pentanoic acid (valeric acid), hexanoic acid (caproic acid), 2-ethylhexanoic acid, heptanoic acid (enanthic acid), octanoic acid (caprylic acid), oleic acid, linoleic acid, linolenic acid, cyclohexanecarboxylic acid, cyclohexylacetic acid, cyclohexenecarboxylic acid, benzoic acid, benzeneacetic acid, propanedioic acid (malonic acid), butanedioic acid (succinic acid), hexanedioic acid (adipic acid), 2-butenedioc acid (maleic acid), acetic anhydride, maleic anhydride, and the like.

[0023] Comparative Examples 1-16 are illustrative of silicone rubbers containing various silver-type antimicrobial agents by themselves or the aliphatic carboxylic acid oleic acid by itself and as such are outside the scope of the invention. Examples 1-25 are illustrative of the antimicrobially efficacious silicone rubbers of this invention which contains both a silver-type antimicrobial agent and an aliphatic carboxylic acid.

[0024] Antimicrobial efficacy was determined by the AATCC 100/Film Contact Method as modified in accordance with the description of this method which follows. The test method provides a quantitative procedure for the evaluation of the degree of antibacterial activity, in this case, test samples of cured silicone rubber samples, or specimens, containing a variety of silver-containing antimicrobial agents, carboxylic acids and mixtures of both.

[0025] Silicone rubber film of 2 mm thickness is inoculated with a known concentration of two test organisms, *Staphylococcus aureus* and *Escherichia coli*, and thereafter cut into 35 mm x 35 mm specimens. A test specimen is placed in a sterile 60 mm petri dish and "floated" in a 100 mm x 20 mm petri dish containing sterile water (for humidity purposes). The test specimen is then inoculated with 0.4 mL of the test organism and incubated for 24 hours after which the specimen is neutralized with Letheen broth and serial dilutions are performed. Percentage reduction and/or log reduction is reported along with the zero time count and 24 time count.

Test Procedures

[0026] 1. Inoculate SCD broth with the appropriate test organisms (*S. aureus* and *E. coli*). Incubate for 18-24 hours.

2. Standardize test organisms with a spectrophotometer to reach a population of 10^8 .

3. Dilute the test organisms with 0.2% Nutrient Broth and sterile saline to obtain an approximate population of $1-5 \times 10^5$.

4. Inoculate the test samples: two 35 mm x 35 mm test samples are inoculated with 0.4 mL of each test organism (one for the zero time and one for the 24 hour time).

5. Neutralize the zero time sample with Letheen Broth and perform serial dilutions. Plate with TSA agar and incubate 48 hours at 36-38°C.

6. Incubate the 24 hour "floated" test sample for 24 hours at 36-38°C.

7. After 24 hours, the 24 hour test sample is neutralized with Letheen Broth and serial dilutions are performed. These plates are also incubated 48 hours at 36-38°C.

8. Percent reduction and, optionally, log reduction, are reported.

COMPARATIVE EXAMPLES 1-4

[0027] These examples show that antimicrobial nanosilver agents, known to be effective by themselves when incorporated into various thermoplastic resins, are essentially inactive when incorporated in a conventional LSR.

[0028] Test specimens of a conventional first LSR (Silicone Rubber A) containing nanosilver particles were prepared with the antimicrobial effectiveness of each specimen being evaluated employing the aforescribed microbiological assay procedure. The results of the assays were as follows:

Table 1: Microbiological Test Results for Silicone Rubber A Containing Nanosilver

Comp. Ex.	Antimicrobial Nanosilver, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
1	0.5	<i>S. aureus</i>	1.2×10^5	1.6×10^5	no effect
2	0.5	<i>E. coli</i>	1.2×10^5	2.2×10^7	no effect
3	0.5	<i>S. aureus</i>	1.4×10^5	5.0×10^5	no effect
4	0.5	<i>E. coli</i>	1.2×10^5	7.5×10^7	no effect

* Percent reduction based on initial (0 time) number of bacteria.

COMPARATIVE EXAMPLES 5 AND 6

[0029] These examples demonstrate the ineffectiveness of the antimicrobial silver sodium hydrogen zirconium phosphate RC2000 (Milliken & Company) when incorporated by itself in a conventional second LSR (Silicone Rubber B). Using the aforescribed microbiological assay procedure, the results of the assays were as follows:

Table 2: Microbiological Test Results for Silicone Rubber B Containing Silver Sodium Hydrogen Zirconium Phosphate RC2000

Comp. Ex.	Antimicrobial Silver RC2000, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
5	20	<i>S. aureus</i>	1.2×10^5	1.3×10^6	no effect
6	20	<i>E. coli</i>	1.2×10^5	2.2×10^7	no effect

* Percent reduction based on initial (0 time) number of bacteria.

COMPARATIVE EXAMPLES 7 AND 8

[0030] These examples illustrate the effects of incorporating a silver ion-containing ceramic matrix carrier (SANITIZED AG, Switzerland) in Silicone Rubber A by itself. As the results in Table 3 show, while 1 wt.% loading of this antimicrobial silver-containing agent was ineffective against *S. aureus* and *E. coli*.

Table 3: Microbiological Test Results for Silicone Rubber A Incorporating Silver Ion-containing Ceramic Matrix

Comp. Ex.	Antimicrobial Silver Ion-Containing Ceramic Matrix, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
7	1	<i>S. aureus</i>	3.1×10^5	3.3×10^6	no effect
8	1	<i>E. coli</i>	1.5×10^5	1.4×10^6	no effect

* Percent reduction based on initial (0 time) number of bacteria.

COMPARATIVE EXAMPLES 9-12

[0031] Comparative Examples 7 and 8 were substantially repeated but with Silicone Rubber B and with different loadings of the silver ion-containing ceramic matrix. The test results are set forth in Table 4 as follows:

Table 4: Microbiological Test Results for Silicone Rubber B Incorporating Silver Ion-containing Ceramic Matrix

Comp. Ex.	Antimicrobial Silver Ion-containing Ceramic Matrix, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
9	2	<i>S. aureus</i>	1.9×10^5	2.5×10^5	No reduction
10	2	<i>E. coli</i>	2.0×10^5	1.9×10^7	No reduction
11	3	<i>S. aureus</i>	1.7×10^5	4.6×10^5	No reduction
12	3	<i>E. coli</i>	1.5×10^5	2.7×10^7	No reduction

* Percent reduction based on initial (0 time) number of bacteria.

EXAMPLE 1

[0032] This example illustrates a silicone rubber, specifically, LSR 2050, exhibiting high surface-available antimicrobial silver in accordance with the invention.

[0033] Following the general microbiological assay procedures described above, a specimen of Silicone Rubber A containing nanosilver and oleic acid was evaluated with the following results:

Table 5: Microbiological Test Results for Silicone Rubber A Containing Nanosilver

Example	Antimicrobial Nanosilver + Carboxylic Acid	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
1	0.5 wt.% nanosilver and 0.5 wt.% oleic acid	<i>S. aureus</i>	1.5 x 10 ⁵	6.6 x 10 ³	95.60%

* Percent reduction based on initial (0 time) number of bacteria.

COMPARATIVE EXAMPLES 13 AND 14; EXAMPLES 2 AND 3

[0034] Comparative Examples 13 and 14 show that the aliphatic carboxylic acid oleic acid is not effective as an antimicrobial agent when present by itself in Silicone Rubber A but as shown in Examples 2 and 3, in combination with silver sodium hydrogen zirconium phosphate RC 2000 as the antimicrobial silver-containing agent, provides a high level of antimicrobial effectiveness. Employing the standard microbiological test procedures described above, the tests results were as follows:

Table 6: Microbiological Test Results for Silicone Rubber A Containing Oleic Acid Alone and In Combination With Silver Sodium Hydrogen Zirconium Phosphate RC 2000

Example	Antimicrobial Silver RC2000, wt.%	Oleic Acid, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*	Log Reduction**
Comp. Ex. 13	-	0.8	<i>S. aureus</i>	1.5 x 10 ⁵	2.3 x 10 ⁶	No effect	No effect
Comp. Ex. 14	-	0.8	<i>E. coli</i>	1.2 x 10 ⁵	3.6 x 10 ⁷	No effect	No effect
Ex. 2	5	0.8	<i>S. aureus</i>	1.9 x 10 ⁵	1.3 x 10 ²	99.91%	4.2
Ex. 3	5	0.8	<i>E. coli</i>	1.5 x 10 ⁵	<1.0 x 10 ¹	>99.99%	>6.6

* Percent reduction based on initial (0 time) number of bacteria.

** Log reduction is based on the control as the reference.

COMPARATIVE EXAMPLES 15 AND 16; EXAMPLES 4 AND 5

[0035] Comparative Examples 13 and 14 and Examples 2 and 3 were substantially repeated but with the amounts of silver ion-containing ceramic matrix indicated in Table 7. As reported in the table, the test results were as follows:

Table 7: Microbiological Test Results for Silicone Rubber A Containing Oleic Acid Alone and In Combination With and Without Silver Ion-containing Ceramic Matrix

Example	Antimicrobial Silver Ion-containing Ceramic Matrix, wt.%	Oleic Acid, wt.%	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*	Log Reduction**
Comp. Ex. 15	-	0.8	<i>S. aureus</i>	1.5×10^5	2.3×10^6	No effect	No effect
Comp. Ex. 16	-	0.8	<i>E. coli</i>	1.2×10^5	3.6×10^7	No effect	No effect
Ex. 4	1	0.8	<i>S. aureus</i>	1.8×10^5	$<1.0 \times 10^1$	>99.99%	5.4
Ex. 5	1	0.8	<i>E. coli</i>	1.8×10^5	$<1.0 \times 10^1$	>99.99%	>6.6

* Percent reduction based on initial (0 time) number of bacteria.

** Log reduction is based on the control as the reference.

EXAMPLES 6-25

[0036] Table 8 sets forth the results of microbiological testing of samples of Silicone Rubber B containing combinations of the silver ion-containing ceramic matrix of the previous examples ("Ag") with each of several different aliphatic carboxylic acids.

Examples	Aliphatic Acid + Ag Combination	Organism	Initial Contact Time (CFU/Sample)	24 Hour Contact Time (CFU/Sample)	Percent Reduction*
6, 7	0.41 wt.% 2-ethylhexanoic acid, 1 wt.% Ag additive	<i>S. aureus</i>	1.7×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.8×10^5	$<1.0 \times 10^1$	>99.99%
8,9	0.41 wt.% caprylic acid, 1 wt.% Ag additive	<i>S. aureus</i>	2.1×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	2.0×10^5	$<1.0 \times 10^1$	>99.99%
10, 11	0.57 wt.% lauric acid, 1 wt.% Ag additive	<i>S. aureus</i>	2.1×10^5	$<2.2 \times 10^4$	89.52%
		<i>E. coli</i>	1.8×10^5	$<1.0 \times 10^1$	>99.99%
12, 13	0.80 wt.% oleic acid, 1 wt.% Ag additive	<i>S. aureus</i>	1.9×10^5	$<1.2 \times 10^2$	99.94%
		<i>E. coli</i>	1.6×10^5	$<4.0 \times 10^1$	99.97%
14, 15	0.80 wt.% linoleic acid, 1 wt.% Ag additive	<i>S. aureus</i>	1.9×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.6×10^5	$<1.0 \times 10^1$	>99.99%
16, 17	0.41 wt.% 2-ethylhexanoic acid, 2 wt. % Ag additive	<i>S. aureus</i>	1.9×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.5×10^5	$<1.0 \times 10^1$	>99.99%
18, 19	0.41 wt.% caprylic acid, 2 wt.% Ag additive	<i>S. aureus</i>	1.6×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.2×10^5	$<1.0 \times 10^1$	>99.99%
20, 21	0.57 wt.%, lauric acid, 2 wt.% Ag additive	<i>S. aureus</i>	1.7×10^5	$<8.0 \times 10^1$	99.95%
		<i>E. coli</i>	1.6×10^5	$<2.0 \times 10^1$	99.99%
22, 23	0.80 wt.%, oleic acid, 2 wt.% Ag additive	<i>S. aureus</i>	1.7×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.7×10^5	$<2.5 \times 10^1$	99.99%
24, 25	0.80 wt.%, linoleic acid, 2 wt.% Ag additive	<i>S. aureus</i>	1.8×10^5	$<1.0 \times 10^1$	>99.99%
		<i>E. coli</i>	1.7×10^5	$<1.0 \times 10^1$	>99.99%

[0037] As the foregoing experimental results show, the combination of silver-based antimicrobial agents with aliphatic carboxylic acids greatly improves the antimicrobial efficacy of silicone rubbers in which they are incorporated, in some cases from no efficacy to high log reduction. The experimental results also demonstrate that the differences in the composition of the LSR have little if any impact on antimicrobial efficacy.

[0038] While the invention has been described with reference to a number of exemplary embodiments, it will be understood by those skilled in the art that various changes can be made and equivalents can be substituted for elements thereof without departing from the scope of the invention. In addition, many modifications can be made to adapt a particular situation or material to the teachings of the invention without departing from essential scope thereof. Therefore, it is intended that the invention not be limited to any particular exemplary embodiment disclosed herein.

Claims:

1. A silicone rubber exhibiting antimicrobial efficacy which comprises:
 - a) at least one silicone rubber derived from a curable silicone rubber-forming composition;
 - b) at least one silver-containing antimicrobial agent incorporated in silicone rubber (a) in X weight percent amount; and,
 - c) at least one carboxylic acid incorporated in silicone rubber (a) in Y weight percent amount,
silver-containing antimicrobial agent (b) when incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein and carboxylic acid (c) when incorporated by itself in silicone rubber (a) in an amount of X + Y weight percent therein imparting no significant antimicrobial activity to silicone rubber (a) but in the combination therein of up to X weight percent silver-containing antimicrobial agent (b) and up to Y weight percent carboxylic acid (c) imparting significant antimicrobial activity to silicone rubber (a), such activity being exhibited at the interface of an exposed surface of silicone rubber (a) and a microbe-populated aqueous material in contact therewith.
2. The silicone rubber of Claim 1 wherein X is a weight percent range selected from the group consisting of 0.1 to 10, 0.2 to 5 and 0.5 to 2 and Y is a weight percent range selected from the group consisting of 0.1 to 5, 0.2 to 3 and 0.5 to 2.
3. The silicone rubber of Claim 1 wherein the curable silicone rubber-forming composition from which silicone rubber (a) is derived is an LSR, RTV or HCE.
4. The silicone rubber of Claim 1 wherein silver-containing antimicrobial agent (b) is at least one member selected from the group consisting of inorganic silver compound, organic silver compound, silver ion-containing glass, silver ion-containing zeolite, silver ion-containing ceramic matrix, colloidal silver, nanosilver and silver alloy.

5. The silicone rubber of Claim 1 wherein silver-containing antimicrobial agent (a) is at least one of nanosilver, silver ion-containing ceramic matrix and silver sodium hydrogen zirconium phosphate.

6. The silicone rubber of Claim 1 wherein carboxylic acid (c) is at least one member selected from the group consisting of aliphatic carboxylic acid, cycloaliphatic carboxylic acid, aromatic carboxylic acid and anhydrides thereof.

7. The silicone rubber of Claim 4 wherein carboxylic acid (c) is at least one member selected from the group consisting of aliphatic carboxylic acid, cycloaliphatic carboxylic acid, aromatic carboxylic acid and anhydrides thereof.

8. The silicone rubber of Claim 6 wherein carboxylic acid (c) is at least one aliphatic carboxylic acid selected from the group consisting of propanoic acid, 2-methylpropanoic acid, hexanoic acid, heptanoic acid, octanoic acid, lauric acid, loeic acid, linoleic acid, linolenic acid, cyclohexanecarboxylic acid, cyclohexylacetic acid, cyclohexenecarboxylic acid, benzoic acid, benzeneacetic acid, propanedioic acid, butanedioic acid, hexanedioic acid, 2-butenedioic acid, acetic anhydride and maleic anhydride.

9. The silicone rubber of Claim 7 wherein carboxylic acid (c) is at least one aliphatic carboxylic acid selected from the group consisting of propanoic acid, 2-methylpropanoic acid, hexanoic acid, heptanoic acid, octanoic acid, lauric acid, loeic acid, linoleic acid, linolenic acid, cyclohexanecarboxylic acid, cyclohexylacetic acid, cyclohexenecarboxylic acid, benzoic acid, benzeneacetic acid, propanedioic acid, butanedioic acid, hexanedioic acid, 2-butenedioic acid, acetic anhydride and maleic anhydride.

10. The silicone rubber of Claim 1 wherein the curable silicone rubber-forming composition from which silicone rubber (a) is derived is an LSR, RTV or HCE, silver-containing antimicrobial agent (b) is at least one member selected from the group consisting of inorganic silver compound, organic silver compound, silver ion-containing glass, silver ion-containing zeolite, silver ion-containing ceramic matrix, colloidal silver, nanosilver and silver alloy, carboxylic acid (c) is at least one member selected from the group consisting of aliphatic carboxylic acid, cycloaliphatic carboxylic acid, aromatic carboxylic acid and anhydrides thereof,

X is a weight percent range selected from the group consisting of 0.1 to 10, 0.2 to 5 and 0.5 to 2 and Y is a weight percent range selected from the group consisting of 0.1 to 5, 0.2 to 3 and 0.5 to 2.

11. An article made from, or having as a component thereof, the silicone rubber of Claim 1.

12. An article made from, or having as a component thereof, the silicone rubber of Claim 2.

13. An article made from, or having as a component thereof, the silicone rubber of Claim 3.

14. An article made from, or having as a component thereof, the silicone rubber of Claim 4.

15. An article made from, or having as a component thereof, the silicone rubber of Claim 6.

16. An article made from, or having as a component thereof, the silicone rubber of Claim 10.

17. A process of imparting efficacious antimicrobial activity to a silicone rubber (a) derived from a curable silicone rubber-forming composition which comprises:

a) incorporating in the curable silicone rubber-forming composition X weight percent of at least one silver-containing antimicrobial agent (b) and Y weight percent of at least one carboxylic acid (c); and,

b) curing the curable silicone rubber-forming composition to provide silicone rubber (a),

silver-containing antimicrobial agent (b) when incorporated by itself in the rubber-forming composition in an amount of X + Y weight percent therein and carboxylic acid (c) when incorporated by itself in the rubber-forming composition in an amount of X + Y weight percent therein imparting no significant antimicrobial activity to silicone rubbers (a) obtained

from the curing of the rubber-forming compositions but in the combination of up to X weight percent silver-containing antimicrobial agent (b) and up to Y weight percent carboxylic acid in the rubber-forming composition, imparting significant antimicrobial activity to silicone rubber (a) obtained therefrom, such activity being exhibited at the interface of an exposed surface of silicone rubber (a) and a microbe-populated aqueous material in contact therewith.

18. The process of Claim 17 wherein X is a weight percent range selected from the group consisting of 0.1 to 10, 0.2 to 5 and 0.5 to 2 and Y is a weight percent range selected from the group consisting of 0.1 to 5, 0.2 to 3 and 0.5 to 2.

19. The process of Claim 17 wherein the curable silicone rubber-forming composition from which silicone rubber (a) is derived is an LSR, RTV or HCE.

20. The silicone rubber obtained by the process of Claim 17.

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2008/004175

A. CLASSIFICATION OF SUBJECT MATTER
INV. A01N59/16 A01N37/02 A01P1/00
ADD. C08K13/02

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched. (classification system followed by classification symbols)
A01N

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DATABASE WPI Section Ch, Week 199621 Thomson Scientific, London, GB; Class A26, AN 1996-205725 XP002499565 & JP 08 073743 A (DOW CORNING TORAY SILICONE) 19 March 1996 (1996-03-19) abstract -----	1-20
Y	US 2005/070956 A1 (ROUSSEAU ROBERT A [US]) 31 March 2005 (2005-03-31) page 2, paragraph 15 page 3, paragraphs 18,19 -----	1-20
Y	GB 2 414 184 A (HIMEDICA LTD [GB]) 23 November 2005 (2005-11-23) page 5, line 32 - page 6, line 25 -----	1-20
	-/-	

Further documents are listed in the continuation of Box C.

See patent family annex.

* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search Date of mailing of the international search report

14 October 2008

24/10/2008

Name and mailing address of the ISA/
European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Galley, Carl

INTERNATIONAL SEARCH REPORT

International application No PCT/US2008/004175

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WO 2006/031965 A (DU PONT [US]; CHEN JOHN CHU [US]; URADNISHECK JULIUS [US]) 23 March 2006 (2006-03-23) page 15, line 29 – page 17, line 9 page 32, lines 17-19 page 15, lines 16-18	1-20
Y	WO 02/083031 A (HELIX MEDICAL INC; SEDER EDMUND V [US]; NELSON JESSE N [US]) 24 October 2002 (2002-10-24) page 6, line 12 – page 7, line 3 page 7, lines 13-20	1-20
Y	WO 2007/127236 A (ACRYMED INC [US]; MCMAKEN JACK D [US]; GIBBINS BRUCE L [US]) 8 November 2007 (2007-11-08) pages 2-3, paragraph 17-21 page 3, paragraph 28	1-20
Y	JO ET AL: "Combined treatment with silver ions and organic acid enhances growth-inhibition of <i>Escherichia coli</i> 0157:H7" FOOD CONTROL, BUTTERWORTH, LONDON, GB, vol. 18, no. 10, 10 April 2007 (2007-04-10), pages 1235-1240, XP022025102 ISSN: 0956-7135 the whole document	1-20
Y	GB 2 402 880 A (JOHNSON & JOHNSON MEDICAL LTD [GB]) 22 December 2004 (2004-12-22) page 3, line 25 – page 4, line 24	1-20
Y	DATABASE WPI Section Ch, Week 200553 Thomson Scientific, London, GB; Class A60, AN 2005-518786 XP002499567 & KR 2005 012 316 A (PARK TAE SOO) 2 February 2005 (2005-02-02) abstract	1-20

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/US2008/004175

Patent document cited in search report		Publication date		Patent family member(s)		Publication date
JP 8073743	A	19-03-1996	JP	3403252 B2		06-05-2003
US 2005070956	A1	31-03-2005	AT	398967 T		15-07-2008
			EP	1677685 A1		12-07-2006
			WO	2005032378 A1		14-04-2005
GB 2414184	A	23-11-2005	NONE			
WO 2006031965	A	23-03-2006	AR	053968 A1		30-05-2007
WO 02083031	A	24-10-2002	US	2005171602 A1		04-08-2005
			US	2002193879 A1		19-12-2002
			US	2005020844 A1		27-01-2005
WO 2007127236	A	08-11-2007	US	2007293800 A1		20-12-2007
GB 2402880	A	22-12-2004	NONE			
KR 2005012316	A		NONE			