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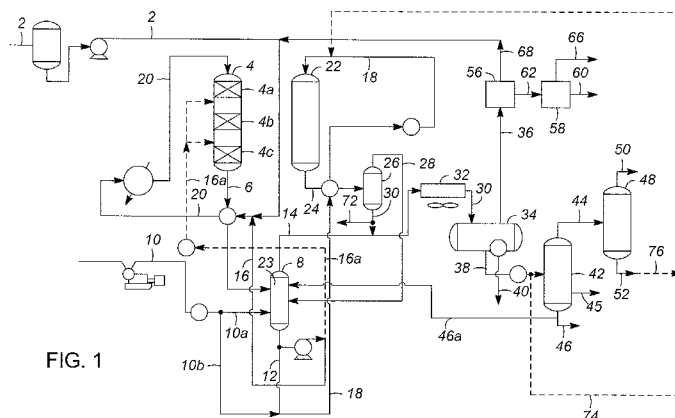
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(54) Title: PRODUCTION OF FUEL FROM CO-PROCESSING MULTIPLE RENEWABLE FEEDSTOCKS



(57) Abstract: A process for producing a fuel or fuel blending component from co-processing at least two different classes of renewable feedstocks, is presented. One feedstock comprises glycerides and free fatty acids in feedstocks such as plant and animal oils while the other feedstock comprises biomass derived pyrolysis oil. The source of the animal or plant oil and the biomass may be the same renewable source.

PRODUCTION OF FUEL FROM CO-PROCESSING
MULTIPLE RENEWABLE FEEDSTOCKS

CROSS-REFERENCE TO RELATED APPLICATION

- 5 **[0001]** This application claims priority from Provisional Application Serial No. 61/122,780 filed December 16, 2008, the contents of which are hereby incorporated by reference in its entirety.

BACKGROUND OF THE INVENTION

- 10 **[0002]** The process produces at least one fuel from at least two different types of renewable feedstocks including biomass derived pyrolysis oil and the triglycerides and free fatty acids found in plant and animal oils fats and greases. The two different types of renewable feedstocks are co-processed with at least one paraffin rich component produced from the triglycerides and free fatty acids found in plant and animal oils fats and greases and with at least one cyclic rich
- 15 component produced from a biomass derived pyrolysis oil. The effluent containing at least one paraffin rich fuel component and at least one aromatic rich fuel component is useful as at least one fuel or fuel blending component. The fuel, fuel additives, or blending components generated may include those in the gasoline boiling point range, the diesel boiling point range, and or the aviation boiling point range.
- 20 **[0003]** As the demand for gasoline, diesel fuel, and aviation fuel increases worldwide there is increasing interest in sources other than petroleum crude oil for producing these fuels. One such source is what has been termed renewable feedstocks. One type of renewable feedstocks include, but are not limited to, plant oils such as corn, rapeseed, canola, soybean and algal oils, animal fats such as inedible tallow, fish oils and various waste streams such as
- 25 yellow and brown greases and sewage sludge. The common feature of these feedstocks is that they are composed of glycerides and Free Fatty Acids (FFA). Both of these compounds contain aliphatic carbon chains having from 8 to 24 carbon atoms. The aliphatic carbon chains in the triglycerides or FFAs can also be mono-, di- or poly-unsaturated. Some of the glycerides from the renewable sources may be monoglycerides or diglycerides instead of or

in addition to the triglycerides. This type of feedstock is employed to generate the paraffinic portion of a fuel.

[0004] There are reports in the art disclosing the production of hydrocarbons from oils. For example, US 4,300,009 discloses the use of crystalline aluminosilicate zeolites to convert plant oils such as corn oil to hydrocarbons such as gasoline and chemicals such as para-xylene. US 4,992,605 discloses the production of hydrocarbon products in the diesel boiling point range by hydroprocessing vegetable oils such as canola or sunflower oil. Finally, US 2004/0230085 A1 discloses a process for treating a hydrocarbon component of biological origin by hydrodeoxygenation followed by isomerization.

[0005] The generation of the cyclic rich component of the fuel employs a process for obtaining a cyclic rich component from biomass. More particularly, this process relates to the treatment of pyrolysis oil produced from the pyrolysis of biomass to produce fuel or fuel blending or additive components. As discussed above, renewable energy sources are of increasing importance. They are a means of reducing dependence on petroleum oil and provide a substitute for fossil fuels. Also, renewable resources can provide for basic chemical constituents to be used in other industries, such as chemical monomers for the making of plastics. Biomass is a renewable resource that can provide some of the needs for sources of chemicals and fuels.

[0006] Biomass includes, but is not limited to, lignin, plant parts, fruits, vegetables, plant processing waste, wood chips, chaff, grain, grasses, corn, corn husks, weeds, aquatic plants, hay, paper, paper products, recycled paper and paper products, and any cellulose containing biological material or material of biological origin. Lignocellulosic biomass, or cellulosic biomass as used throughout the remainder of this document, consists of the three principle biopolymers cellulose, hemicellulose, and lignin. The ratio of these three components varies depending on the biomass source. Cellulosic biomass might also contain lipids, ash, and protein in varying amounts. The economics for converting biomass to fuels or chemicals depend on the ability to produce large amounts of biomass on marginal land, or in a water environment where there are few or no other significantly competing economic uses of that land or water environment. The economics can also depend on the disposal of biomass that would normally be placed in a landfill.

[0007] The growing, harvesting and processing of biomass in a water environment provides a space where there is plenty of sunlight and nutrients while not detracting from more productive alternate uses. Biomass is also generated in many everyday processes as a

waste product, such as waste material from crops. In addition, biomass contributes to the removal of carbon dioxide from the atmosphere as the biomass grows. The use of biomass can be one process for recycling atmospheric carbon while producing fuels and chemical precursors. Biomass when heated at short contact times in an environment with low or no oxygen, termed pyrolysis, will generate a liquid product known as pyrolysis oil. Synonyms for pyrolysis oil include bio-oil, pyrolysis liquids, bio-crude oil, wood liquids, wood oil, liquid smoke, wood distillates, pyroligneous acid, and liquid wood

[0008] The product of the biomass pyrolysis, the pyrolysis oil, contains what is known as pyrolysis oil non-aqueous phase. Pyrolysis oil non-aqueous phase is the water insoluble portion of the pyrolysis oil. The pyrolysis oil may be processed whole, or a portion of the aqueous phase may be removed to provide a pyrolysis oil enriched in pyrolysis oil non-aqueous phase which is co-processed along with the triglycerides through deoxygenation to produce the cyclic rich portion of the fuel or fuel blending component.

[0009] The process herein involves co-processing the pyrolysis oil feedstock and the triglyceride and FFA feedstock by hydrogenation, deoxygenation (decarboxylation, decarbonylation, and/or hydrodeoxygenation) in at least a first zone and hydroisomerization and hydrocracking in at least a second zone in order to generate a gasoline range product, a diesel range product, and or an aviation range product. Simply hydrogenating and deoxygenating the renewable glyceride or FFA feedstocks in a hydrogen environment in the presence of a hydrotreating catalyst results in straight chain paraffins having chain-lengths similar to, or slightly shorter than, the fatty acid composition of the feedstock. With many feedstocks, this approach results in a fuel meeting the general specification for a diesel fuel, but not the specifications for an aviation fuel. The selective hydrocracking reaction reduces the carbon chain length to allow selectivity to aviation fuel range paraffins while minimizing lower molecular weight products. Similarly, isomerization increases the concentration of branched paraffins and thereby improve cold flow properties such as cloud point or freeze point.

[0010] The pyrolysis oil and the triglycerides are co-processed to generate an effluent comprising at least one paraffin rich component and at least one cyclic rich component. The effluent is useful as a fuel or a fuel blending component. The relative amounts of the feedstocks may be controlled so that the resulting effluent meets specific requirements of a target fuel. Other additives or components may be blended with the effluent in order to meet

additional requirements of the target fuel. The target fuel may be in the boiling point ranges of gasoline, aviation, and diesel, and may be entirely derived from renewable sources. The target fuel is designed to power engines or devices that are currently distributed around the world without requiring upgrades to those engines. The target fuel may be blended with other components generated from renewable feedstocks to meet the specifications using entirely renewable feedstock derived blending components, or the target fuel may be blended with petroleum derived fuels or fuel blending components in any concentration.

SUMMARY OF THE INVENTION

[0011] A process for producing a fuel having both a paraffin rich component and a cyclic rich component where the paraffin rich component and the cyclic rich component are each produced from a renewable feedstock. The renewable feedstocks are co-processed together. The paraffin rich component is produced from the triglycerides and free fatty acids found in plant and animal oils, fats, and greases, and the cyclic rich component is produced from a biomass derived pyrolysis oil. The cyclic rich component has biomass derived pyrolysis oil as the renewable feedstock. The pyrolysis oil is derived from the pyrolysis of biomass. The pyrolysis oil may optionally be enriched in pyrolysis oil non-aqueous phase through the removal of at least a portion of the aqueous phase, but the process also allows the whole pyrolysis oil to be processed without removal of a portion of the aqueous phase. The triglycerides and free fatty acid feedstock and the biomass derived pyrolysis oil feedstock are co-processed through the zones of the process.

[0012] Both types of feedstocks are simultaneously treated in a common reaction zone by hydrogenating and deoxygenating the feedstock mixture at reaction conditions to provide a first reaction zone product comprising n-paraffins and cyclic hydrocarbons.

[0013] In one embodiment, the first reaction zone may have two stages, with the triglycerides and free fatty acids along with the whole pyrolysis oil or the pyrolysis oil non-aqueous phase enriched pyrolysis oil, all being treated in a first deoxygenation zone generating a partially deoxygenated stream. Water, gasses, and light ends are removed and the remainder of the partially deoxygenated stream is further treated in a second deoxygenation zone to produce a deoxygenated product stream. The deoxygenated product stream comprises n-paraffins and cyclic hydrocarbon compounds that when fractionated are useful fuels in the gasoline and naphtha/gasoline, aviation, and diesel boiling point ranges.

[0014] After the second deoxygenation zone, water light ends, and gasses may be removed from the effluent of the second deoxygenation zone. Hydrogen may be separated and recycled. In one embodiment the first and second deoxygenation zones are combined and housed within in a single reactor.

5 [0015] The carbon dioxide and water generated as byproducts in the first reaction zone may be selectively removed from the first reaction product in an optional integrated hot high pressure stripper using hydrogen as the stripping gas. A diesel range stream, an aviation range stream, a naphtha/gasoline range stream, a naphtha/gasoline and LPG range stream or any mixture thereof may be optionally used as a rectification agent in the selective hot high
10 pressure hydrogen stripper to decrease the amount of first reaction zone diesel and aviation range product carried in the overhead of the selective hot high pressure hydrogen stripper. Note that the naphtha/gasoline range stream is within the boiling point range for gasoline and therefore may also be considered a gasoline boiling point range stream.

[0016] The hydrogen stripped first reaction zone product may optionally be introduced to
15 a hydroisomerization and selective cracking reaction zone. The selective hydrocracking allows for aviation fuel range products, if desired, by preferentially cracking C1 to C6 fragments off the end of the longer chain paraffins and by minimizing the number of cracking events per molecule. The selective cracking also breaks apart the stronger carbon-oxygen linkages that remain from the pyrolysis oil type of feedstock. The desired product, at least
20 one paraffin rich component in the diesel boiling point range, the aviation boiling point range, and or the naphthene boiling range product, is recovered.

[0017] The effluent comprising at least one paraffin rich component and at least one cyclic rich component form at least one fuel or may be blended with other components to form additional fuels.

25 BRIEF DESCRIPTION OF THE DRAWINGS

[0018] FIG. 1 is a schematic of one embodiment a process. FIG. 1 shows the option where the combined feedstocks of whole pyrolysis oil and triglyceride and free fatty acids is processed through a single stage of deoxygenation.

[0019] FIG. 2 is a schematic of one embodiment of a process. FIG. 2 shows the option
30 where the combined feedstocks of whole pyrolysis oil and triglyceride and free fatty acids is processed through two stages of deoxygenation.

[0020] FIG. 3 is a schematic of one embodiment of a process. FIG. 3 shows the option where the pyrolysis oil feedstock is phase separated with the pyrolysis oil non-aqueous phase being combined with a triglyceride and free fatty acids feedstock to be co-processed through two stages of deoxygenation.

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DETAILED DESCRIPTION OF THE INVENTION

[0021] The invention provides a process for generating a fuel or fuel blending component from a renewable feedstock comprising at least glycerides and a second renewable feedstock comprising at least a portion of a pyrolysis oil, where the two renewable feedstocks are co-processed together to provide the fuel or fuel blending component. The fuel may be one of
10 several types of hydrocarbon mixtures, a diesel boiling point range hydrocarbon mixture, an aviation boiling point range hydrocarbon mixture, and/or a naphtha/gasoline boiling point range mixture.

[0022] The term renewable feedstock is meant to include feedstocks other than those obtained from petroleum crude oil. Another term that has been used to describe members of
15 this class of feedstock is biorenewable fats and oils. The first class of renewable feedstocks that can be used to generate a paraffin rich portion of the fuel include any of those which comprise glycerides and free fatty acids (FFA). Most of the glycerides will be triglycerides, but monoglycerides and diglycerides may be present and processed as well. Examples of these renewable feedstocks include, but are not limited to, canola oil, corn oil, soy oils,
20 rapeseed oil, soybean oil, colza oil, tall oil, sunflower oil, hempseed oil, olive oil, linseed oil, coconut oil, castor oil, peanut oil, palm oil, mustard oil, cottonseed oil, jatropha oil, inedible tallow, yellow and brown greases, lard, train oil, fats in milk, fish oil, algal oil, sewage sludge, cuphea oil, camelina oil, jatropha oil, curcas oil, babassu oil, palm kernel oil, and the like. The glycerides and FFAs of the typical vegetable or animal fat contain aliphatic
25 hydrocarbon chains in their structure which have 8 to 24 carbon atoms with a majority of the fats and oils containing high concentrations of fatty acids with 16 and 18 carbon atoms. Mixtures or co-feeds of renewable feedstocks and petroleum-derived hydrocarbons may also be used as the first feedstock. Other feedstock components which may be combined into the first feedstock include spent motor oils and industrial lubricants, used paraffin waxes, liquids
30 derived from the gasification of coal, biomass, natural gas followed by a downstream liquefaction step such as Fischer-Tropsch technology, liquids derived from depolymerization,

thermal or chemical, of waste plastics such as polypropylene, high density polyethylene, and low density polyethylene; and other synthetic oils generated as byproducts from petrochemical and chemical processes. Mixtures of the above may also be used as co-feed components in the first feedstock.

5 **[0023]** The second class of renewable feedstock used herein is a feedstock originating from lignocellulose. In the U.S. and worldwide, there are huge amounts of lignocellulosic material, or biomass, which is not utilized, but is left to decay, often in a landfill, or just in an open field or forest. The material includes large amounts of wood waste products, and leaves and stalks of crops or other plant material that is regularly discarded and left to decay in
10 fields. The emergence of inedible lipid-bearing crops for the production of renewable diesel will also produce increased amounts of biomass post extraction, often known as meal. Growth of cellulosic ethanol will also produce large amounts of a lignin side product. Biomass includes, but is not limited to, lignin, plant parts, fruits, vegetables, plant processing waste, wood chips, chaff, grain, grasses, corn, corn husks, weeds, aquatic plants, hay, meal,
15 paper, paper products, recycled paper and paper products, and any cellulose containing biological material or material of biological origin. This biomass material can be pyrolyzed to make a pyrolysis oil, but due to poor thermal stability, the high water content of the pyrolysis oil, often greater than 25%, high total acid number often greater than 100, low heating value, and phase incompatibility with petroleum based materials, pyrolysis oil has found little use
20 other than as a source of specialty chemicals and as a low grade heating fuel.

25 **[0024]** The pyrolysis of the biomass to form the pyrolysis oil is achieved by any technique known in the art, see for example, Mohan, D.; Pittman, C. U.; Steele, P. H. Energy and Fuels, 2006, 20, 848-889. Once the pyrolysis oil is generated from the biomass, although optional, it is not necessary to separate the pyrolysis oil non-aqueous phase from the
30 pyrolysis oil before further processing, thereby eliminating a step previously employed in industry. The whole pyrolysis oil may be processed, without a portion of the aqueous phase being removed to enrich the pyrolysis oil in the pyrolysis oil non-aqueous phase. The pyrolytic lignin contains potentially high value products in the form of aromatic and naphthenic compounds having complex structures that comprises aromatic rings that are
30 linked by oxygen atoms or carbon atoms. These structures can be broken into smaller segments when decarboxylated, decarbonylated, or hydrodeoxygenated, while maintaining the aromatic ring structures. One desired product is at least one cyclic hydrocarbon-rich

stream. In one embodiment, this processing of the pyrolytic lignin may be accomplished in the presence of the rest of the pyrolysis oil with no separation of the pyrolytic lignin before processing required. The pyrolysis oil non-aqueous phase is primarily a pyrolysis product of the lignin portion of biomass. It can be separated from the rest of the whole pyrolysis oil during the pyrolysis process or through post-processing to produce an additional aqueous phase, which includes pyrolysis products primarily from the cellulose and hemicellulose portion of the biomass. The pyrolysis process can convert all components in the biomass feedstock into products useful as fuels or fuel components after full deoxygenation of the pyrolysis oil product. The water soluble components can also be transformed to naphthenes and aromatics under pyrolysis conditions due, for example, to a high concentration of phenols. Optionally, the pyrolysis oil may be separated and only a portion of the pyrolysis oil be introduced to the dehydrogenation and deoxygenation zone as the second feedstock.

[0025] The first renewable feedstock containing the glycerides and free fatty acids may contain a variety of impurities. For example, tall oil is a byproduct of the wood processing industry and tall oil contains esters and rosin acids in addition to FFAs. Rosin acids are cyclic carboxylic acids. The first renewable feedstock may also contain contaminants such as alkali metals, e.g. sodium and potassium, phosphorous as well as solids, water and detergents. An optional first step is to remove as much of these contaminants as possible before being combined with the second renewable feedstock. One possible pretreatment step involves contacting the first renewable feedstock with an ion-exchange resin in a pretreatment zone at pretreatment conditions. The ion-exchange resin is an acidic ion exchange resin such as AmberlystTM-15 and can be used as a bed in a reactor through which the first feedstock is flowed through, either upflow or downflow.

[0026] Another possible means for removing contaminants from the first renewable feedstock is a mild acid wash. This is carried out by contacting the first renewable feedstock with a solution of water mixed with an acid such as sulfuric, nitric, phosphoric, or hydrochloric in a reactor. The acid and feedstock can be contacted either in a batch or continuous process. Contacting is done with a dilute acid solution usually at ambient temperature and atmospheric pressure. If the contacting is done in a continuous manner, it is usually done in a counter current manner. Yet another possible means of removing metal contaminants from the first renewable feedstock is through the use of guard beds which are well known in the art. These can include alumina guard beds either with or without

demetallation catalysts such as nickel or cobalt. Filtration and solvent extraction techniques are other choices which may be employed. Hydroprocessing such as that described in USAN 11/770,826, hereby incorporated by reference, is another pretreatment technique which may be employed.

5 **[0027]** The first and second renewable feedstocks are flowed to a first reaction zone comprising one or more catalyst beds in one or more reactors. The term “feedstock” is meant to include feedstocks that have not been treated to remove contaminants as well as those feedstocks purified in a pretreatment zone. The first and second renewable feedstocks may be combined into a single stream, or each feedstock may be introduced independently to the first
10 reaction zone. The pyrolysis oil feedstock may have some water removed, or may be processed as the whole feedstock. The process will be described with reference to the embodiment where the feedstocks are combined into a single stream which is introduced to the first reaction zone. In the reaction first zone, the feedstocks are contacted with a hydrogenation or hydrotreating catalyst in the presence of hydrogen at hydrogenation
15 conditions to hydrogenate the reactive components such as olefinic or unsaturated portions of the hydrocarbons. Hydrogenation and hydrotreating catalysts are any of those well known in the art such as nickel or nickel/molybdenum dispersed on a high surface area support. Other hydrogenation catalysts include one or more noble metal catalytic elements dispersed on a high surface area support. Non-limiting examples of noble metals include Pt and/or Pd
20 dispersed on gamma-alumina or activated carbon. Hydrogenation conditions include a temperature of 40°C to 400°C and a pressure of 689 kPa absolute (100 psia) to 13,790 kPa absolute (2000 psia). In another embodiment the hydrogenation conditions include a temperature of 200°C to 300°C and a pressure of 1379 kPa absolute (200 psia) to 4826 kPa absolute (700 psia). Other operating conditions for the hydrogenation zone are well known in
25 the art.

[0028] The catalysts enumerated above are also capable of catalyzing decarboxylation, decarbonylation and/or hydrodeoxygenation of the feedstock to remove oxygen. Decarboxylation, decarbonylation, and hydrodeoxygenation are herein collectively referred to as deoxygenation reactions. Decarboxylation conditions include a relatively low pressure of
30 689 kPa (100 psia) to 13,790 kPa (2000 psia), a temperature of 200°C to 400°C and a liquid hourly space velocity of 0.5 to 10 hr⁻¹. In another embodiment the decarboxylation conditions include the same relatively low pressure of 689 kPa (100 psia) to 6895 kPa (1000

psia), a temperature of 288°C to 345°C and a liquid hourly space velocity of 1 to 4 hr⁻¹. Since hydrogenation is an exothermic reaction, as the feedstocks flow through the catalyst bed(s) the temperature increases and decarboxylation and hydrodeoxygenation will begin to occur. Thus, it is envisioned and is within the scope of this invention that all the reactions occur simultaneously in one reactor or in one bed. Alternatively, the conditions can be controlled such that hydrogenation primarily occurs in one bed and decarboxylation and/or hydrodeoxygenation occurs in a second bed. Of course if only one bed is used, then hydrogenation occurs primarily at the front of the bed, while decarboxylation/hydrodeoxygenation occurs mainly in the middle and bottom of the bed. Finally, desired hydrogenation can be carried out in one reactor, while decarboxylation, decarbonylation, and/or hydrodeoxygenation can be carried out in a separate reactor.

[0029] The first reaction zone performs catalytic decarboxylation, decarbonylation, and hydrodeoxygenation of oxygen polymers and single oxygenated molecules in the pyrolysis oil and triglycerides by breaking the oxygen linkages, and forming water and CO₂ from the oxygen and leaving smaller molecules. For example, the phenylpropyl ether linkages in the pyrolysis oil non-aqueous phase will be partially deoxygenated producing some aromatic rings, such as alkylbenzenes and polyalkylbenzenes. Very reactive oxygenates will be deoxygenated as well, including small molecular weight carboxylic acids therefore greatly increasing the thermal stability of the product. Pyrolysis oil components not derived from lignin, including cellulose, hemicellulose, free sugars, may yield products such as acetic acid, furfural, furan, levoglucosan, 5-hydroxymethylfurfural, hydroxyacetaldehyde, formaldehyde, and others such as those described in Mohan, D.; Pittman, C. U.; Steele, P. H. *Energy and Fuels*, 2006, 20, 848-889. Therefore, pyrolysis oil components not derived from lignin will also be partially or fully deoxygenated to produce a significant amount of light hydrocarbon fractions and water. The light hydrocarbon fractions may contain hydrocarbons with six or fewer carbon atoms. The paraffins resulting from the deoxygenation of the triglycerides may contain from 15 to 18 or from 8 to 18 carbon atoms depending upon the source of the renewable feedstock. The reactions of decarbonylation, decarboxylation and hydrodeoxygenation are collectively referred to as deoxygenation reactions. Hydrogenation of olefins also occur in this zone.

[0030] The reaction product from the hydrogenation and deoxygenation reactions will comprise both a liquid portion and a gaseous portion. The liquid portion comprises a hydrocarbon fraction comprising n-paraffins and having a large concentration of paraffins in

the 15 to 18 carbon number range resulting from the hydrogenation and deoxygenation of the triglyceride and free fatty acid feedstock; and a wide boiling range of predominately cyclic hydrocarbons resulting from the hydrogenation and deoxygenation of the pyrolysis oil feedstock. Different triglyceride and free fatty acid feedstocks will result in different distributions of components. A portion of this hydrocarbon fraction, after separation from the gaseous portion, may be used as the hydrocarbon recycle described below. Although this hydrocarbon fraction is useful as a diesel fuel or diesel fuel blending component, additional fuels, such as aviation fuels or aviation fuel blending components which typically have a concentration of paraffins in the range of 9 to 15 carbon atoms, may be produced with additional processing. Also, because the hydrocarbon fraction contains a large amount of n-paraffins, it may have poor cold flow properties. Many diesel and aviation fuels and blending components must have better cold flow properties and so the reaction product is further reacted under isomerization conditions to isomerize at least a portion of the n-paraffins to branched paraffins.

[0031] In another embodiment, the combined feedstock containing both the triglycerides and free fatty acids and the pyrolysis oil is fully deoxygenated in two separate zones, a partial deoxygenation zone and a full deoxygenation zone. The partial deoxygenation zone may also be considered to be a hydrotreating zone and the full deoxygenation zone may be considered to be a hydrocracking zone. "Full" deoxygenation is meant to include deoxygenating at least 99 % to 100 % of available oxygenated hydrocarbons. For this embodiment, the zones will primarily be referred to herein as a partial deoxygenation zone and a full deoxygenation zone. In the partial deoxygenation zone, partial deoxygenation occurs at milder conditions than the full deoxygenation zone and uses a catalyst such as a hydrotreating catalyst. In general, the partial oxidation zone removes the most reactive and thermally instable oxygenates. The catalysts and conditions of the partial deoxygenation zone are selected so that the more reactive compounds are deoxygenated while minimizing the thermal polymerization well known to occur in pyrolysis oils subjected to high temperatures. The oxygen level of the combined feedstock, which typically ranges from 10 to 50 wt.% is reduced to a significantly lower level, from 5 to 20 wt. % in the partial deoxygenation zone. Water is reduced from the combined feedstock levels of from 10 to 30 wt. % to levels from 1 wt.% to 10 wt.%. The acidity is greatly reduced as well in the partial deoxygenation zone, from a Total Acid Number (TAN), as determined by ASTM

D664, level of 125 mg KOH/g to 200 mg KOH/g in the feedstock to a reduced level from 20 mg KOH/g to 100 mg KOH/g in the partial deoxygenation zone effluent.

[0032] The more thermally stable effluent from the partial deoxygenation zone can then be fully deoxygenated in the full deoxygenation zone. In the full deoxygenation zone a first hydroprocessing catalyst is employed with the option of more severe process conditions in order to catalyze the deoxygenation of less reactive oxygenates. Some hydrocracking of feedstock molecules will also occur to a higher extent than in the partial deoxygenation zone. In the full deoxygenation zone, oxygen content is reduced from 5 wt.% to 20 wt.% to much lower levels, from ppm concentrations to 0.5 wt.%. Water is also greatly reduced in the full deoxygenation zone, from 1 wt.% to 10 wt.% down to levels from 100 ppm to 1000 ppm. The acidity is greatly reduced from initial Total Acid Number (TAN), as determined by ASTM D664, levels of 20 to 100 mg KOH/g to lower levels from 0.01 to 4 mg KOH/g. The effluent of the full deoxygenation zone is a hydrocarbon mixture rich in paraffins, naphthenes and aromatics.

[0033] Optionally, when employing the embodiment where the hydrogenation and deoxygenation are conducted in two zones, partially deoxygenated effluent may be passed to a separation zone before being introduced to the full deoxygenation zone. Carbon oxides, possibly hydrogen sulfide, and C3 and lighter components are separated and removed in an overhead line and a partially deoxygenated product stream is removed from the separation zone. The separation zone may comprise a separator. Depending upon whether the separator is operated in a hot or cold mode, the water may be removed as a vapor (hot separator mode) or as a liquid (cold separator mode). The overhead comprises a large quantity of hydrogen and at least the carbon dioxide from the decarboxylation reaction. The carbon dioxide can be removed from the hydrogen by means well known in the art such as reaction with a hot carbonate solution, pressure swing absorption, etc. Also, absorption with an amine in processes such as described in co-pending applications USAN 12/193,172 and USAN 12/193,196 hereby incorporated by reference, may be employed. If desired, essentially pure carbon dioxide can be recovered by regenerating the spent absorption media. Therefore the overhead is passed through one or more scrubbers such as amine scrubbers to remove carbon dioxide and hydrogen sulfide. Depending upon the scrubber technology selected some portion of water may also be retained by the scrubber. The lighter hydrocarbons and gasses, possibly including a portion of water, are conducted to steam reforming zone. In one embodiment the light hydrocarbon fractions may

contain hydrocarbons with six or fewer carbon atoms. After purification, hydrogen generated in the steam reforming zone is conducted to combine with feedstock and partially deoxygenated product stream. The hydrogen may be recycled to combine with the feedstock as shown or may be introduced directly to the reaction zone where hydrogenation primarily occurs and/or to any subsequent reactor beds.

[0034] The partially deoxygenated product stream along with recycle hydrogen stream and optional hydrocarbon recycle, is passed to a second hydrodeoxygenation zone, where the remaining oxygen is removed. The full deoxygenation zone performs catalytic decarboxylation, decarbonylation, and hydrodeoxygenation of the remaining oxygen compounds that are more stable than those reacted in the first stage. The same or different hydroprocessing catalyst may be used as compared to the partial deoxygenation zone. In one embodiment, a more active catalyst and more severe process conditions are employed in the full deoxygenation zone as compared to the partial deoxygenation zone in order to catalyze full deoxygenation.

[0035] As mentioned above, the effluent of the hydrogenation and deoxygenation zone comprises a liquid hydrocarbon portion and a gaseous portion. The gaseous portion comprises hydrogen, carbon dioxide, carbon monoxide, water vapor, propane and perhaps sulfur components such as hydrogen sulfide or possibly a phosphorous component such as phosphine. In one embodiment, such as the single stage deoxygenation embodiment, the effluent from the deoxygenation zone is conducted to a hot high pressure hydrogen stripper. One purpose of the hot high pressure hydrogen stripper is to selectively separate at least a portion of the gaseous portion of the effluent from the liquid portion of the effluent. As hydrogen is an expensive resource, to conserve costs, the separated hydrogen is recycled to the first reaction zone containing the deoxygenation reactor(s). Also, failure to remove the water, carbon monoxide, and carbon dioxide from the effluent may result in poor catalyst performance in a downstream zone such as the optional isomerization zone. Water, carbon monoxide, carbon dioxide, any ammonia or hydrogen sulfide are selectively stripped in the hot high pressure hydrogen stripper using hydrogen. The hydrogen used for the stripping may be dry, and free of carbon oxides. The temperature may be controlled in a limited range to achieve the desired separation and the pressure may be maintained at approximately the same pressure as the two reaction zones to minimize both investment and operating costs. The hot high pressure hydrogen stripper may be operated at conditions ranging from a pressure of 689 kPa absolute (100 psia) to 13,790 kPa absolute (2000 psia), and a temperature of 40°C to

350°C. In another embodiment the hot high pressure hydrogen stripper may be operated at conditions ranging from a pressure of 1379 kPa absolute (200 psia) to 4826 kPa absolute (700 psia), or 2413 kPa absolute (350 psia) to 4882 kPa absolute (650 psia), and a temperature of 50°C to 350°C. The hot high pressure hydrogen stripper may be operated at essentially the same pressure as the reaction zone. By “essentially”, it is meant that the operating pressure of the hot high pressure hydrogen stripper is within 1034 kPa absolute (150 psia) of the operating pressure of the reaction zone. For example, in one embodiment the hot high pressure hydrogen stripper separation zone is no more than 1034 kPa absolute (150 psia) less than that of the reaction zone.

[0036] The effluent enters the hot high pressure stripper and at least a portion of the gaseous components, are carried with the hydrogen stripping gas and separated into an overhead stream. The remainder of the deoxygenation zone effluent stream is removed as hot high pressure hydrogen stripper bottoms and contains the liquid hydrocarbon fraction having components such as normal and isomerized hydrocarbons having from 8 to 24 carbon atoms, a wide range of cycloparaffins and aromatics with a boiling point range from 25 to 400°C, and some heavier cyclic hydrocarbons with a boiling point range from 400 to 600°C. A portion of this liquid hydrocarbon fraction in hot high pressure hydrogen stripper bottoms may be used as the hydrocarbon recycle described below.

[0037] In some embodiments, a portion of the lighter hydrocarbons generated in the deoxygenation zone may be also carried with the hydrogen in the hot high pressure hydrogen stripper and removed in the overhead stream. Any hydrocarbons removed in the overhead stream will effectively bypass the optional isomerization zone, discussed below. Some of the hydrocarbons bypassing the optional isomerization zone may be normal hydrocarbons which, due to bypassing the isomerization stage, will not be isomerized to branched hydrocarbons.

At least a portion of these normal hydrocarbons may ultimately end up in the diesel range product or the aviation range product, and depending upon the specifications required for the products, the normal hydrocarbons may have an undesired effect on the diesel range product and the aviation range product. For example, in applications where the diesel range product is required to meet specific cloud point specifications, or where the aviation range product is required to meet specific freeze point specifications, the normal hydrocarbons from the hot high pressure hydrogen stripper overhead may interfere with meeting the required specification.

[0038] Therefore, in some embodiments it is advantageous to take steps to prevent normal hydrocarbons from being removed in the hot high pressure hydrogen stripper overhead and bypassing the isomerization zone. For example, one or more, or a mixture of additional rectification agents may be optionally introduced into the hot high pressure hydrogen stripper to reduce the amount of hydrocarbons in the hot high pressure hydrogen stripper overhead stream. Suitable example of additional rectification agents include the diesel boiling point range product, the aviation boiling point range product, the naphtha/gasoline boiling range product, the mixture of naphtha/gasoline and LPG, or any combinations thereof. These streams may be recycled and introduced to the hot high pressure hydrogen stripper, at a location of the stripper that is above the deoxygenation zone effluent introduction location and in the rectification zone. The rectification zone, if present, may contain vapor liquid contacting devices such as trays or packing to increase the efficiency of the rectification. The optional rectification agent would operate to force an increased amount of the hydrocarbon product from the deoxygenation zone to travel downward in the hot high pressure hydrogen stripper and be removed in the hot high pressure hydrogen stripper bottoms stream instead of being carried with the stripping hydrogen gas into the hot high pressure hydrogen stripper overhead. Other rectification agents from independent sources may be used instead of, or in combination with, the diesel boiling point range product, the naphtha/gasoline product, and the naphtha/gasoline and LPG stream.

[0039] In another embodiment, a separator may be used to separate the gaseous portion of the effluent from the liquid portion of the effluent.

[0040] Hydrogen is a reactant in at least some of the reactions above, and a sufficient quantity of hydrogen must be in solution to most effectively take part in the catalytic reaction. Past processes have operated at high pressures in order to achieve a desired amount of hydrogen in solution and readily available for reaction. However, higher pressure operations are more costly to build and to operate as compared to their lower pressure counterparts. One advantage of an embodiment of the invention is the operating pressure may be in the range of 1379 kPa absolute (200 psia) to 4826 kPa absolute (700 psia) which is lower than that found in other previous operations. In another embodiment the operating pressure is in the range of 2413 kPa absolute (350 psia) to 4481 kPa absolute (650 psia), and in yet another embodiment operating pressure is in the range of 2758 kPa absolute (400 psia) to 4137 kPa absolute (600

psia). Furthermore, the rate of reaction is increased resulting in a greater amount of throughput of material through the reactor in a given period of time.

[0041] In one embodiment, the desired amount of hydrogen is kept in solution at lower pressures by employing a large recycle of hydrocarbon to the hydrogenation and

5 deoxygenation reaction zone. A hydrocarbon recycle may also be employed in order to control the temperature in the reaction zones since the reactions are exothermic reactions. Hydrogen has a greater solubility in the hydrocarbon product than it does in the feedstock. By utilizing a large hydrocarbon recycle the solubility of hydrogen in the combined liquid phase in the reaction zone is greatly increased and higher pressures are not needed to increase the
10 amount of hydrogen in solution. In one embodiment of the invention, the volume ratio of hydrocarbon recycle to feedstock is from 2:1 to 8:1. In another embodiment the ratio is in the range of 3:1 to 6:1 and in yet another embodiment the ratio is in the range of 4:1 to 5:1.

[0042] Although the hydrocarbon fraction separated from the deoxygenation zone effluent may be useful as a fuel or fuel blending component, because it may comprise a large amount of
15 n-paraffins, it may have poor cold flow properties. Also, depending upon the feedstock, the amount of hydrocarbons suitable for aviation fuel or aviation fuel blending component may be small. Therefore the hydrocarbon fraction may optionally be contacted with an isomerization catalyst under isomerization conditions to at least partially isomerize the n-paraffins to branched paraffins and improve the cold flow properties of the liquid hydrocarbon fraction. The
20 isomerization catalysts and operating conditions may optionally be selected so that the isomerization catalyst also catalyzes selective hydrocracking of the paraffins. The selective hydrocracking also creates hydrocarbons in the aviation boiling point range. The effluent of the second reaction zone, the isomerization and selective hydrocracking zone, is enriched in branched-paraffins. Isomerization and selective hydrocracking can be carried out in a separate
25 bed of the same reactor, described above or the isomerization and selective hydrocracking can be carried out in a separate reactor. For ease of description, the following will address the embodiment where a second reactor is employed for the isomerization and selective hydrocracking reactions. The hydrogen stripped product of the deoxygenation reaction zone is contacted with an isomerization and selective hydrocracking catalyst in the presence of
30 hydrogen at isomerization and selective hydrocracking conditions to isomerize at least a portion of the normal paraffins to branched paraffins. Due to the presence of hydrogen, the reactions may be called hydroisomerization and hydrocracking.

[0043] The isomerization and selective hydrocracking of the hydrogenation and deoxygenation product can be accomplished in any manner known in the art or by using any suitable catalyst known in the art. One or more beds of catalyst may be used. It is preferred that the isomerization be operated in a co-current mode of operation. Fixed bed, trickle bed down flow or fixed bed liquid filled up-flow modes are both suitable. See also, for example, US 2004/0230085 A1 which is incorporated by reference in its entirety. Catalysts having an acid function and mild hydrogenation function are favorable for catalyzing both the isomerization reaction and the selective hydrocracking reaction. Suitable catalysts comprise a metal of Group VIII (IUPAC 8-10) of the Periodic Table and a support material. Suitable Group VIII metals include platinum and palladium, each of which may be used alone or in combination. The support material may be amorphous or crystalline or a combination of the two. Suitable support materials include amorphous alumina, amorphous silica-alumina, ferrierite, ALPO-31, SAPO-11, SAPO-31, SAPO-37, SAPO-41, SM-3, MgAPSO-31, FU-9, NU-10, NU-23, ZSM-12, ZSM-22, ZSM-23, ZSM-35, ZSM-48, ZSM-50, ZSM-57, MeAPO-11, MeAPO-31, MeAPO-41, MeAPSO-11, MeAPSO-31, MeAPSO-41, MeAPSO-46, ELAPO-11, ELAPO-31, ELAPO-41, ELAPSO-11, ELAPSO-31, ELAPSO-41, laumontite, cancrinite, offretite, hydrogen form of stillbite, magnesium or calcium form of mordenite, and magnesium or calcium form of partheite, each of which may be used alone or in combination. ALPO-31 is described in US 4,310,440. SAPO-11, SAPO-31, SAPO-37, and SAPO-41 are described in US 4,440,871. SM-3 is described in US 4,943,424; US 5,087,347; US 5,158,665; and US 5,208,005. MgAPSO is a MeAPSO, which is an acronym for a metal aluminumsilicophosphate molecular sieve, where the metal Me is magnesium (Mg). Suitable MeAPSO-31 catalysts include MgAPSO-31. MeAPSOs are described in US 4,793,984, and MgAPSOs are described in US 4,758,419. MgAPSO-31 is a preferred MgAPSO, where 31 means a MgAPSO having structure type 31. Many natural zeolites, such as ferrierite, that have an initially reduced pore size can be converted to forms suitable for selective hydrocracking and isomerization by removing associated alkali metal or alkaline earth metal by ammonium ion exchange and calcination to produce the substantially hydrogen form, as taught in US 4,795,623 and US 4,924,027. Further catalysts and conditions for skeletal isomerization are disclosed in US 5,510,306, US 5,082,956, and US 5,741,759.

[0044] The isomerization and selective hydrocracking catalyst may also comprise a modifier selected from the group consisting of lanthanum, cerium, praseodymium, neodymium,

phosphorus, samarium, gadolinium, terbium, and mixtures thereof, as described in US 5,716,897 and US 5,851,949. Other suitable support materials include ZSM-22, ZSM-23, and ZSM-35, which are described for use in dewaxing in US 5,246,566 and in the article entitled "New molecular sieve process for lube dewaxing by wax isomerization," written by S. J. Miller, in
5 Microporous Materials 2 (1994) 439-449. The teachings of US 4,310,440; US 4,440,871; US 4,793,984; US 4,758,419; US 4,943,424; US 5,087,347; US 5,158,665; US 5,208,005; US 5,246,566; US 5,716,897; and US 5,851,949 are hereby incorporated by reference.

[0045] US 5,444,032 and US 5,608,134 teach a suitable bifunctional catalyst which is constituted by an amorphous silica-alumina gel and one or more metals belonging to Group
10 VIIIA, and is effective in the hydroisomerization of long-chain normal paraffins containing more than 15 carbon atoms. An activated carbon catalyst support may also be used. US 5,981,419 and US 5,968,344 teach a suitable bifunctional catalyst which comprises: (a) a porous crystalline material isostructural with beta-zeolite selected from boro-silicate (BOR-B) and boro-alumino-silicate (Al-BOR-B) in which the molar $\text{SiO}_2:\text{Al}_2\text{O}_3$ ratio is higher than 300:1; (b) one
15 or more metal(s) belonging to Group VIIIA, selected from platinum and palladium, in an amount comprised within the range of from 0.05 to 5% by weight. Article V. Calemme et al., App. Catal. A: Gen., 190 (2000), 207 teaches yet another suitable catalyst.

[0046] The isomerization and selective hydrocracking catalyst may be any of those well known in the art such as those described and cited above. Isomerization and selective
20 cracking conditions include a temperature of 150°C to 360°C and a pressure of 1724 kPa absolute (250 psia) to 4726 kPa absolute (700 psia). In another embodiment the isomerization conditions include a temperature of 300°C to 360°C and a pressure of 3102 kPa absolute (450 psia) to 3792 kPa absolute (550 psia). Other operating conditions for the isomerization and selective hydrocracking zone are well known in the art. Some known isomerization catalysts,
25 when operated under more severe conditions, also provide the selective hydrocracking catalytic function.

[0047] The isomerization and selective cracking zone effluent is processed through one or more separation steps to obtain at least two purified hydrocarbon streams, one useful as a diesel fuel or a diesel fuel blending component and the second useful as aviation fuel or an
30 aviation fuel blending component. It is likely that the separation steps could be used to produce and additional purified hydrocarbon stream useful as a gasoline fuel or gasoline fuel blending component. Depending upon the application, various additives may be combined with

the diesel or aviation fuel composition generated in order to meet required specifications for different specific fuels. In particular, the aviation fuel produced would have the same physical properties and performance characteristics as aviation fuel qualified to meet at least one of: ASTM D 1655 Specification for Aviation Turbine Fuels Defense Stan 91--91 Turbine Fuel, Aviation Kerosene Type, Jet A-1 NATO code F-35, F-34, F-37 Aviation Fuel Quality Requirements for Jointly Operated Systems (Joint Checklist) A combination of ASTM and Def Stan requirements GOST 10227 Jet Fuel Specifications (Russia) Canadian CAN/CGSB-3.22 Aviation Turbine Fuel, Wide Cut Type Canadian CAN/CGSB-3.23 Aviation Turbine Fuel, Kerosene Type MIL-DTL-83133, JP-8, MIL-DTL-5624, JP-4, JP-5 QAV-1 (Brazil) Especificacao de Querosene de Aviacao No. 3 Jet Fuel (Chinese) according to GB6537 DCSEA 134A (France) Carbureacteur Pour Turbomachines D'Aviation, Type Kerosene Aviation Turbine Fuels of other countries, meeting the general grade requirements for Jet A, Jet A-1, Jet B, and TS-1 fuels as described in the IATA Guidance Material for Aviation Turbine Fuel Specifications. The aviation fuel is generally termed "jet fuel" herein and the term "jet fuel" is meant to encompass aviation fuel meeting the specifications above as well as to encompass aviation fuel used as a blending component of an aviation fuel meeting the specifications above. Additives may be added to the jet fuel in order to meet particular specifications. One particular type of jet fuel is JP-8, defined by Military Specification MIL-DTL-83133, which is a military grade type of highly refined kerosene based jet propellant specified by the United States Government. The fuel produced from glycerides or FFAs is very similar to SPK, also known as a synthetic paraffinic kerosene.

[0048] The specifications for different types of fuels are often expressed through acceptable ranges of chemical and physical requirements of the fuel. As stated above, aviation turbine fuels, a kerosene type fuel including JP-8, are specified by MIL-DTL-83133, JP-4, a blend of gasoline, kerosene and light distillates, is specified by MIL-DTL-5624 and JP-5 a kerosene type fuel with low volatility and high flash point is also specified by MIL-DTL-5624, with the written specification of each being periodically revised. Often a distillation range from 10 percent recovered to a final boiling point is used as a key parameter defining different types of fuels. The distillations ranges are typically measured by ASTM Test Method D 86 or D2887. Therefore, blending of different components in order to meet the specification is quite common. While the product of the present invention may meet fuel specifications, in some cases, blending of the product with other blending components may

be required to meet the desired set of fuel specifications. In other words, the aviation product of this invention is a composition which may be used with other components to form a fuel meeting at least one of the specifications for aviation fuel such as JP-8.

[0049] With the effluent stream of either the deoxygenation zone or the optional isomerization and selective hydrocracking zone comprising both a liquid component and a gaseous component, various portions of which may be recycled, multiple separation steps may be employed. For example, hydrogen may be first separated in an effluent separator with the separated hydrogen being removed in an overhead stream. Suitable operating conditions of the isomerization effluent separator include, for example, a temperature of 230°C and a pressure of 4100 kPa absolute (600 psia). If there is a low concentration of carbon oxides, or the carbon oxides are removed, the hydrogen may be recycled back to the hot high pressure hydrogen stripper for use both as a rectification gas and to combine with the remainder as a bottoms stream. Or, the hydrogen may be passed to the deoxygenation or isomerization reaction zones and thus the hydrogen becomes a component of the reaction zone feed streams in order to provide the necessary hydrogen partial pressures for the reactor. Different feedstocks will consume different amounts of hydrogen. The effluent separator allows flexibility for the process to operate even when larger amounts of hydrogen are consumed in the first reaction zone. Furthermore, at least a portion of the remainder or bottoms stream of the effluent separator may be recycled to the deoxygenation zone or the optional isomerization zone to perhaps increase the degree of isomerization.

[0050] The remainder of the effluent of either the isomerization and selective hydrocracking zone or the hydrogenation and deoxygenation zone, after the removal of hydrogen, still has liquid and gaseous components and is cooled, by techniques such as air cooling or water cooling and passed to a cold separator where the liquid component is separated from the gaseous component. Suitable operating conditions of the cold separator include, for example, a temperature of 20 to 60°C and a pressure of 3850 kPa absolute (560 psia). A water byproduct stream is also separated. At least a portion of the liquid component, after cooling and separating from the gaseous component, may be recycled back to the isomerization zone to increase the degree of isomerization. Prior to entering the cold separator, the remainder of the effluent may be combined with the hot high pressure hydrogen stripper overhead stream, and the resulting combined stream may be introduced into the cold separator.

[0051] The liquid component contains the hydrocarbons useful as diesel fuel or diesel fuel blending components, aviation fuel or aviation fuel blending components, gasoline fuel or gasoline fuel blending components, termed diesel boiling point range product, aviation boiling point range product, and gasoline boiling range product respectively, as well as amounts of LPG. The separated liquid component is further purified in a product distillation zone which separates lower boiling components and dissolved gases into an LPG and naphtha/gasoline boiling point range stream; an aviation boiling point range product; and a diesel boiling point range product. Suitable operating conditions of the product distillation zone include a temperature of from 20 to 200°C at the overhead and a pressure from 0 to 1379 kPa absolute (0 to 200 psia). The conditions of the distillation zone may be adjusted to control the relative amounts of hydrocarbon contained in the aviation boiling point range product stream, the diesel boiling point range product stream, and the LPG and naphtha/gasoline boiling point range stream.

[0052] The LPG and naphtha/gasoline stream may be further separated in a debutanizer or depropanizer in order to separate the LPG into an overhead stream, leaving the naphtha/gasoline in a bottoms stream. Suitable operating conditions of this unit include a temperature of from 20 to 200°C at the overhead and a pressure from 0 to 2758 kPa absolute (0 to 400 psia). The LPG may be sold as valuable product or may be used in other processes such as a feed to a hydrogen production facility. The gasoline boiling point range stream may be collected as product or may be used in other processes. Optionally, the conditions of the distillation zone may be adjusted so that the gasoline boiling point range hydrocarbons are removed as a side cut as opposed to being removed with the LPG thus avoiding a debutanizer or depropanizer.

[0053] The gaseous component separated in the product separator comprises mostly hydrogen and the carbon dioxide from the decarboxylation reaction. Other components such as carbon monoxide, propane, and hydrogen sulfide or other sulfur containing component may be present as well. It is desirable to recycle the hydrogen to the isomerization zone, but if the carbon dioxide was not removed, its concentration would quickly build up and effect the operation of the isomerization zone. The carbon dioxide can be removed from the hydrogen by means well known in the art such as reaction with a hot carbonate solution, pressure swing absorption, etc. Amine absorbers may be employed as taught in copending US applications USAN 12/193,176 and USAN 12/193,196 hereby incorporated by reference. If

desired, essentially pure carbon dioxide can be recovered by regenerating the spent absorption media.

[0054] Similarly, a sulfur containing component such as hydrogen sulfide may be present to maintain the sulfided state of the deoxygenation catalyst or to control the relative amounts of the decarboxylation reaction and the hydrogenation reaction that are both occurring in the deoxygenation zone. The amount of sulfur is generally controlled and so must be removed before the hydrogen is recycled. The sulfur components may be removed using techniques such as absorption with an amine or by caustic wash. Of course, depending upon the technique used, the carbon dioxide and sulfur containing components, and other components, may be removed in a single separation step such as a hydrogen selective membrane.

[0055] The hydrogen remaining after the removal of at least carbon dioxide may be recycled to the reaction zone where hydrogenation primarily occurs and/or to any subsequent beds or reactors. The recycle stream may be introduced to the inlet of the reaction zone and/or to any subsequent beds or reactors. One benefit of the hydrocarbon recycle is to control the temperature rise across the individual beds. However, as discussed above, the amount of hydrocarbon recycle may be determined based upon the desired hydrogen solubility in the reaction zone which is in excess of that used for temperature control. Increasing the hydrogen solubility in the reaction mixture allows for successful operation at lower pressures, and thus reduced cost.

[0056] As discussed above, at least a portion of the diesel boiling point range product; at least a portion of the aviation boiling point range product, at least a portion of the LPG and naphtha/gasoline stream; at least a portion of a naphtha/gasoline stream or an LPG stream generated by separating the LPG and naphtha/gasoline stream into an LPG stream and the naphtha/gasoline stream; or any combination thereof may be recycled to the optional rectification zone of the hot high pressure hydrogen stripper.

[0057] The following embodiments are presented in illustration of the process and are not intended as an undue limitation on the generally broad scope of the invention as set forth in the claims.

[0058] Turning to FIG. 1, the process for generating the paraffin rich component begins with a renewable feedstock stream 2 which may pass through an optional feed surge drum. The renewable feedstock 2 is a mixture of two different types of renewable feedstocks, one being primarily triglycerides and fatty acids and the other being pyrolysis oil. The feedstock stream is

combined with recycle gas stream 68 and recycle stream 16 to form combined feed stream 20, which is heat exchanged with reactor effluent and then introduced into deoxygenation reactor 4. The heat exchange may occur before or after the recycle is combined with the feed.

Deoxygenation reactor 4 may contain multiple beds shown in FIG. 2 as 4a, 4b and 4c.

5 Deoxygenation reactor 4 contains at least one catalyst capable of catalyzing decarboxylation and/or hydrodeoxygenation of the feedstock to remove oxygen. Deoxygenation reactor effluent stream 6 containing the products of the decarboxylation and/or hydrodeoxygenation reactions is removed from deoxygenation reactor 4 and heat exchanged with stream 20 containing feed to the deoxygenation reactor. Stream 6 comprises a liquid component containing largely normal
10 paraffin hydrocarbons in the diesel boiling point range and a wide range of cycloparaffins and aromatics and a gaseous component containing largely hydrogen, vaporous water, carbon monoxide, carbon dioxide and propane.

[0059] Deoxygenation reactor effluent stream 6 is then directed to hot high pressure hydrogen stripper 8. Make up hydrogen in line 10 is divided into two portions, stream 10a and
15 10b. Make up hydrogen in stream 10a is also introduced to hot high pressure hydrogen stripper 8. In hot high pressure hydrogen stripper 8, the gaseous component of deoxygenation reactor effluent 6 is selectively stripped from the liquid component of deoxygenation reactor effluent 6 using make-up hydrogen 10a and recycle hydrogen 28. The dissolved gaseous component comprising hydrogen, vaporous water, carbon monoxide, carbon dioxide hydrogen sulfide, and
20 at least a portion of the methane, ethane and propane, is selectively separated into hot high pressure hydrogen stripper overhead stream 14. The remaining liquid component of deoxygenation reactor effluent 6 comprising primarily normal paraffins having a carbon number from 8 to 24 and a wide range of cycloparaffins and aromatics with a boiling point range of 25°C to 400°C, and some heavier cyclic hydrocarbons with a boiling point range from 400°C to
25 600°C is removed as hot high pressure hydrogen stripper bottom 12.

[0060] A portion of hot high pressure hydrogen stripper bottoms forms recycle stream 16 and is combined with renewable feedstock stream 2 to create combined feed 20. Another portion of recycle stream 16, optional stream 16a, may be routed directly to deoxygenation reactor 4 and introduced at interstage locations such as between beds 4a and 4b and or between beds 4b and 4c
30 in order, or example, to aid in temperature control. The remainder of hot high pressure hydrogen stripper bottoms in stream 12 is combined with hydrogen stream 10b to form combined stream

18 which is routed to the optional isomerization and selective hydrocracking reactor 22. Stream 18 may be heat exchanged with isomerization reactor effluent 24.

[0061] The product of the optional isomerization and selective hydrocracker reactor containing a gaseous portion of hydrogen and light hydrocarbons and a branched-paraffin-enriched liquid portion is removed in line 24, and after optional heat exchange with stream 18, is introduced into hydrogen separator 26. The overhead stream 28 from hydrogen separator 26 contains primarily hydrogen which may be recycled back to hot high pressure hydrogen stripper 8. Bottom stream 30 from hydrogen separator 26 is air cooled using air cooler 32 and introduced into product separator 34. In product separator 34 the gaseous portion of the stream comprising hydrogen, carbon monoxide, hydrogen sulfide, carbon dioxide and propane are removed in stream 36 while the liquid hydrocarbon portion of the stream is removed in stream 38. A water byproduct stream 40 may also be removed from product separator 34. Stream 38 is introduced to product stripper 42 where components having higher relative volatilities are separated into stream 44, components within the boiling range of aviation fuel is removed in stream 45, with the remainder, the diesel range components, being withdrawn from product stripper 42 in line 46. Optionally, a portion of the diesel range components in line 46 are recycled in line 46a to hot high pressure hydrogen stripper 8 optional rectification zone 23 and used as an additional rectification agent. Stream 44 is introduced into fractionator 48 which operates to separate LPG into overhead 50 leaving a naphtha/gasoline bottoms 52. Any of optional lines 72, 74, or 76 may be used to recycle at least a portion of the isomerization zone effluent back to the isomerization zone to increase the amount of n-paraffins that are isomerized to branched paraffins.

[0062] The vapor stream 36 from product separator 34 contains the gaseous portion of the isomerization effluent which comprises at least hydrogen, carbon monoxide, hydrogen sulfide, carbon dioxide and propane and is directed to a system of amine absorbers to separate carbon dioxide and hydrogen sulfide from the vapor stream. Because of the cost of hydrogen, it is desirable to recycle the hydrogen to deoxygenation reactor 4, but it is not desirable to circulate the carbon dioxide or an excess of sulfur containing components. In order to separate sulfur containing components and carbon dioxide from the hydrogen, vapor stream 36 is passed through a system of at least two amine absorbers, also called scrubbers, starting with the first amine absorber zone 56. The amine chosen to be employed in first amine scrubber 56 is capable of selectively removing at least both the components of interest, carbon dioxide and the sulfur components such as hydrogen sulfide. Suitable amines are available from DOW and from

BASF, and in one embodiment the amines are a promoted or activated methyldiethanolamine (MDEA). See US 6,337,059, hereby incorporated by reference in its entirety. Suitable amines for the first amine absorber zone from DOW include the UCARSOL™ AP series solvents such as AP802, AP804, AP806, AP810 and AP814. The carbon dioxide and hydrogen sulfide are absorbed by the amine while the hydrogen passes through first amine scrubber zone and into line 68 to be recycled to the first reaction zone. The amine is regenerated and the carbon dioxide and hydrogen sulfide are released and removed in line 62. Within the first amine absorber zone, regenerated amine may be recycled for use again. The released carbon dioxide and hydrogen sulfide in line 62 are passed through second amine scrubber zone 58 which contains an amine selective to hydrogen sulfide, but not selective to carbon dioxide. Again, suitable amines are available from DOW and from BASF, and in one embodiment the amines are a promoted or activated MDEA. Suitable amines for the second amine absorber zone from DOW include the UCARSOL™ HS series solvents such as HS101, HS 102, HS103, HS104, HS115. Therefore the carbon dioxide passes through second amine scrubber zone 58 and into line 66. The amine may be regenerated which releases the hydrogen sulfide into line 60. Regenerated amine is then reused, and the hydrogen sulfide may be recycled to the deoxygenation reaction zone. Conditions for the first scrubber zone includes a temperature in the range of 30 to 60°C. The first absorber is operated at essentially the same pressure as the reaction zone. By “essentially” it is meant that the operating pressure of the first absorber is within 1034 kPa absolute (150 psia) of the operating pressure of the reaction zone. For example, the pressure of the first absorber is no more than 1034 kPa absolute (150 psia) less than that of the reaction zone. The second amine absorber zone is operated in a pressure range of from 138 kPa absolute (20 psia) to 241 kPa absolute (35 psia). Also, at least the first the absorber is operated at a temperature that is at least 1°C higher than that of the separator. Keeping the absorbers warmer than the separator operates to maintain any light hydrocarbons in the vapor phase and prevents the light hydrocarbons from condensing into the absorber solvent.

[0063] It is readily understood that instead of a portion of the diesel range components in line 46 being optionally recycled in line 46a to hot high pressure hydrogen stripper 8 optional rectification zone 23 and used as a rectification agent, a portion of naphtha/gasoline bottoms 52 is optionally recycled to hot high pressure hydrogen stripper 8 optional rectification zone 23 and used as a rectification agent. Similarly, instead of a portion of the diesel range components in line 46 being optionally recycled in line 46a to hot high pressure hydrogen stripper 8 optional

rectification zone 23 and used as a rectification agent, the diesel range components in line 46a and portion of naphtha/gasoline bottoms 52 combined to form a rectification agent stream which is optionally recycled to hot high pressure hydrogen stripper 8 optional rectification zone 23 and used as a rectification agent. In one embodiment, diesel range components in line 46 may be recycled to the isomerization and selective hydrocracking zone to increase the yield of aviation boiling point range product.

[0064] Minimizing the amount of normal paraffins that bypass the isomerization and selective hydrocracking zone helps to meet freeze point specifications for many aviation fuels without having to significantly lower the quantity of aviation fuel produced. Normal paraffins that bypass the isomerization and selective hydrocracking zone are not isomerized and the normal paraffins generally have higher freeze points than the corresponding isomerized paraffins.

[0065] In another embodiment, as shown in FIG. 2, feedstock 110 of whole pyrolysis oil and triglycerides and FFA enters partial deoxygenation zone 112 along with recycle hydrogen stream 154 and optional hydrocarbon recycle 156 where contact with a deoxygenation and hydrogenation catalyst at deoxygenation conditions generates partially deoxygenated stream 114. The deoxygenation zone 112 performs catalytic decarboxylation, decarbonylation, and hydrodeoxygenation of the triglycerides and of the oxygen polymers and single oxygenated molecules in the pyrolysis oil by breaking the oxygen linkages, and forming water and CO₂ from the oxygen and leaving smaller molecules as discussed above. Hydrogenation of olefins also occur in this zone. The catalysts and conditions of partial deoxygenation zone 112 are selected so that the more reactive compounds are deoxygenated. The effluent of partial deoxygenation zone is a partially deoxygenated stream 114 that has increased thermal stability as compared to the feed mixture of triglycerides, FFA and pyrolysis oil.

[0066] Partially deoxygenated stream 114 is passed to a separation zone 116. Carbon oxides, possibly hydrogen sulfide, and C3 and lighter components are separated and removed in overhead line 120 and a partially deoxygenated product stream 118 is removed from separation zone 116. Separation zone 116 may comprise a separator. Depending upon whether the separator is operated in a hot or cold mode, the water may be removed as a vapor in line 120 (hot separator mode) or as a liquid in line 122 (cold separator mode). Overhead line 120 comprises a large quantity of hydrogen and at least the carbon dioxide from the decarboxylation reaction. The carbon dioxide can be removed from the hydrogen by means well known in the art

such as reaction with a hot carbonate solution, pressure swing absorption, etc. Also, absorption with an amine in processes such as described in co-pending applications USAN 12/193,176 and USAN 12/193,196, hereby incorporated by reference, may be employed. If desired, essentially pure carbon dioxide can be recovered by regenerating the spent absorption media. Therefore

5 overhead line 120 is passed through one or more scrubbers 144 such as amine scrubbers to remove carbon dioxide in line 146 and hydrogen sulfide in line 148. Depending upon the scrubber technology selected some portion of water may also be retained by the scrubber. The lighter hydrocarbons and gasses, possibly including a portion of water, are conducted via line 150 to steam reforming zone 152. In one embodiment the light hydrocarbon fractions may
10 contain hydrocarbons with six or fewer carbon atoms. After purification, hydrogen generated in steam reforming zone 152 is conducted via line 154 to combine with feedstock 110 and partially deoxygenated product stream 118. The hydrogen may be recycled to combine with the feedstock as shown or may be introduced directly to the reaction zone where hydrogenation primarily occurs and/or to any subsequent reactor beds.

15 **[0067]** The partially deoxygenated product stream 118 along with recycle hydrogen stream 154 and optional hydrocarbon recycle 156, is passed to a full hydrodeoxygenation zone 124, where the remaining oxygen is removed. Full deoxygenation zone 124 performs catalytic decarboxylation, decarbonylation, and hydrodeoxygenation of the remaining oxygen compounds that are more stable than those reacted in the first stage. Therefore, a more active catalyst and
20 more severe process conditions are employed in full deoxygenation zone 124 as compared to partial deoxygenation zone 112 in order to catalyze full deoxygenation.

[0068] Full deoxygenation zone effluent 126 is introduced to phase separator 128. Carbon oxides, possibly hydrogen sulfide and C3 and lighter components are separated and removed in line 130 and liquid hydrocarbons are removed in line 132. Depending upon whether the
25 separator is operated in a hot or cold mode, the water may be removed as a vapor in line 130 (hot separator mode) or as a liquid in line 158 (cold separator mode). The overhead in line 130 comprises a large quantity of hydrogen and the carbon dioxide from the decarboxylation reaction. The carbon dioxide can be removed from the hydrogen by means well known in the art, reaction with a hot carbonate solution, pressure swing absorption, etc. Also, absorption with
30 an amine in processes such as described in co-pending applications USAN 12/193,176 and USAN 12/193,196, hereby incorporated by reference, may be employed. If desired, essentially pure carbon dioxide can be recovered by regenerating the spent absorption media. Therefore

line 130 is passed through one or more scrubbers 144 such as amine scrubbers to remove carbon dioxide in line 146 and hydrogen sulfide in line 148. Depending upon the scrubber technology selected some portion of water may also be retained by the scrubber. The lighter hydrocarbons and gasses, possibly including a portion of water, are conducted via line 150 to steam reforming zone 152. A liquid stream containing hydrocarbons is removed from separator 128 in line 132 and conducted to product fractionation zone 134. Product fractionation zone 134 is operated so that the lighter materials such as naphtha and LPG are removed in fractionation zone overhead stream 160. A portion of stream 160 may be optionally conducted in line 162 to the reforming zone 152. If desired, the naphtha and LPG may be further separated into an LPG stream and a naphtha stream (not shown). Product cut 136 contains the hydrocarbons in a boiling point range most beneficial to meeting the gasoline specifications. Product cut 138 removes hydrocarbons in a boiling range of aviation fuel or a blending component of aviation fuel. Bottoms stream 140 removes hydrocarbons that have a boiling point at least in the diesel boiling point range. A portion of bottoms stream 140 may be separated and recovered and used as fuel such as, for example, low sulfur heating oil fuel (not shown). It is likely that bottoms stream 140 may be acceptable for use as diesel or a diesel blending component. A portion of bottoms stream 140 is optionally recycled to partial deoxygenation zone 112 and/ or full deoxygenation reaction zone 124. A portion of a hydrocarbon stream may also be cooled down if necessary and used as cool quench liquid between beds of one of the deoxygenation zones, or between the first and the full deoxygenation zone to further control the heat of reaction and provide quench liquid for emergencies. The recycle stream may be introduced to the inlet of one or both of the reaction zones and/or to any subsequent beds or reactors. One benefit of the hydrocarbon recycle is to control the temperature rise across the individual beds. However, as discussed within, the amount of hydrocarbon recycle may be determined based upon the desired hydrogen solubility in the reaction zone. Increasing the hydrogen solubility in the reaction mixture allows for successful operation at lower pressures, and thus reduced cost. Operating with high recycle and maintaining high levels of hydrogen in the liquid phase helps dissipate hot spots at the catalyst surface and reduces the formation of undesirable heavy components which lead to coking and catalyst deactivation. Fractionation zone 134 may contain more than one fractionation column and thus the locations of the different streams separated may vary from that shown in the figures.

[0069] In another embodiment as shown in FIG. 3, a pyrolysis oil feed stream 210 is passed through phase separator 204 where it is separated into an aqueous phase and a pyrolysis oil non-aqueous phase. A portion or all of the pyrolysis oil non-aqueous phase is removed from separator 204 in stream 207 which is then combined with stream 206 to form combined stream 202. Optionally, some or all of the pyrolysis oil non-aqueous phase is removed via stream 208. Part of all of the aqueous phase is removed from separator 204 in stream 206 which is then combined with stream 207 to form combined stream 202.

Optionally, aqueous phase pyrolysis oil can be removed through line 205. Combined stream 202, which is a pyrolysis oil non-aqueous phase enriched pyrolysis oil, is mixed with a second feed stream 203 comprising triglycerides and FFAs to form combined feed 209 which passes into partial deoxygenation zone 112 where partial deoxygenation occurs along with hydrogenation of reactive functional groups as described above. The partially deoxygenated product stream 114 passes through separator 116 where CO, CO₂, H₂O, and H₂S are removed. Product stream 118 passes through full deoxygenation zone 124 where complete deoxygenation is catalyzed. Full deoxygenation zone product stream 126 passes through separator 128 where water, CO, CO₂, and H₂S are removed resulting in a liquid hydrocarbon stream 132. Liquid hydrocarbon stream 132 is passed through the fractionation zone 134 where it is separated into the desired streams as discussed above.

[0070] In another embodiment as shown in FIG. 4 optionally a pyrolysis oil feed stream 210 is passed through phase separator 204 where it is separated into an aqueous phase and a pyrolysis oil non-aqueous phase. A portion or all of pyrolysis oil non-aqueous phase is removed from separator 204 in stream 207 which is then combined with stream 206 to form combined stream 202. Optionally, some or all of the pyrolysis oil non-aqueous phase is removed via stream 208. Part of all of the aqueous phase is removed from separator 204 in stream 206 which is then combined with stream 207 to form combined stream 202.

Optionally, aqueous phase pyrolysis oil can be removed through line 205. Triglyceride and FFA stream 203 is added to combined stream 202 to form feed stream 209. Either feed stream 209 (for the embodiment using pyrolysis oil non-aqueous phase enriched pyrolysis oil), or pyrolysis oil feed stream 210 (for the embodiment using the whole pyrolysis oil) combined with Triglyceride and FFA stream 203 passes through deoxygenation zone 325 where contact with one or more catalysts fully deoxygenate the feed to produce a fully deoxygenated product stream 327. Deoxygenation zone 325 can employ a multifunctional

catalyst capable of deoxygenation and hydrogenation or a set of catalysts. For example, partial deoxygenation and hydrogenation can occur over the first catalyst in a first portion of zone 325 while full deoxygenation occurs in a more active catalyst in a second portion of zone 325. A stacked bed configuration may be advantageous because a less active catalyst in an upper zone will deoxygenate the most reactive oxygen compounds without generating exotherms that can promote the formation of thermal coke. The fully deoxygenated product stream 327 is fed to phase separator 128 where water, CO, CO₂, and H₂S are removed resulting in a liquid hydrocarbon stream 132. Liquid hydrocarbon stream 132 is passed through the fractionation zone 134 where it is separated into the desired streams as discussed above.

[0071] A portion of a hydrocarbon stream may also be cooled down if necessary and used as cool quench liquid between beds of one of the deoxygenation zones, or between the first and the full deoxygenation zone to further control the heat of reaction and provide quench liquid for emergencies. The recycle stream may be introduced to the inlet of one or both of the reaction zones and/or to any subsequent beds or reactors. One benefit of the hydrocarbon recycle is to control the temperature rise across the individual beds. However, as discussed within, the amount of hydrocarbon recycle may be determined based upon the desired hydrogen solubility in the reaction zone. Increasing the hydrogen solubility in the reaction mixture allows for successful operation at lower pressures, and thus reduced cost. Operating with high recycle and maintaining high levels of hydrogen in the liquid phase helps dissipate hot spots at the catalyst surface and reduces the formation of undesirable heavy components which lead to coking and catalyst deactivation. The fractionation zone may contain more than one fractionation column and thus the locations of the different streams separated may vary from that shown in the figures.

[0072] In another embodiment, the pyrolysis oil feed stream is separated to remove at least a portion of the aqueous phase thereby concentrating the amount of pyrolysis oil non-aqueous phase left in the pyrolysis oil and generating a pyrolysis oil non-aqueous phase-enriched pyrolysis oil. The separation may be accomplished by passing the pyrolysis oil through a phase separator where it is separated into an aqueous phase and a pyrolysis oil non-aqueous phase and removing at least a portion of the aqueous phase.

[0073] In another embodiment, both deoxygenation zones are housed in a single reactor. The deoxygenation zones may be combined through the use of a multifunctional catalyst

capable of deoxygenation and hydrogenation or a set of catalysts. Or a reactor housing two separate zones, such as a stacked bed reactor, may be employed. For example, partial deoxygenation and hydrogenation can occur over the first catalyst in a first portion of a reactor, a first zone, while full deoxygenation occurs with a more active catalyst in a second portion the reactor, a second zone. A stacked bed configuration may be advantageous because a less active catalyst in an upper zone will deoxygenate the most reactive oxygen compounds without generating exotherms that can promote the formation of thermal coke.

[0074] Hydrogen is needed for the deoxygenation and hydrogenation reactions above, and to be effective, a sufficient quantity of hydrogen must be in solution in the deoxygenation zone to most effectively take part in the catalytic reaction. If hydrogen is not available at the reaction site of the catalyst, the coke forms on the catalyst and deactivates the catalyst. High operating pressures may be used in order to achieve a desired amount of hydrogen in solution and readily available for reaction and to avoid coking reactions on the catalyst. However, higher pressure operations are more costly to build and to operate as compared to their lower pressure counterparts.

[0075] The desired amount of hydrogen may be kept in solution at lower pressures by employing a large recycle of hydrocarbon. An added benefit is the control of the temperature in the deoxygenation zone(s) since the deoxygenation reactions are exothermic reactions. However, the range of recycle to feedstock ratios used herein is set based on the need to control the level of hydrogen in the liquid phase and therefore reduce the deactivation rate of the catalyst. The amount of recycle is determined not on temperature control requirements, but instead, based upon hydrogen solubility requirements. Hydrogen has a greater solubility in the hydrocarbon product than it does in the pyrolysis oil feedstock or the portion of the pyrolysis oil feedstock after separation. By utilizing a large hydrocarbon recycle the solubility of hydrogen in the liquid phase in the reaction zone is greatly increased and higher pressures are not needed to increase the amount of hydrogen in solution and avoid catalyst deactivation at low pressures. The hydrocarbon recycle may be a portion of the stream in any of lines 132, 140, 138, or 136, or any combination thereof, and the hydrocarbon recycle is directed to deoxygenation zone 112 or 325. The figure shows optional hydrocarbon recycle 156 as a portion of diesel boiling point range component 140. However it is understood that in other embodiments portions different streams or combinations of stream such as the product stream 132 or any of fractionation zone streams 138, 136, 160 may be used as the hydrocarbon recycle.

Suitable volume ratios of hydrocarbon recycle to pyrolysis oil feedstock is from 2:1 to 8:1. In another embodiment the ratio is in the range of 3:1 to 6:1 and in yet another embodiment the ratio is in the range of 4:1 to 5:1.

[0076] Furthermore, the rate of reaction in the deoxygenation zone is increased with the hydrocarbon recycle resulting in a greater amount of throughput of material through the reactor in a given period of time. Lower operating pressures provide an additional advantage in increasing the decarboxylation reaction while reducing the hydrodeoxygenation reaction. The result is a reduction in the amount of hydrogen required to remove oxygen from the feedstock component and produce a finished product. Hydrogen can be a costly component of the feed and reduction of the hydrogen requirements is beneficial from an economic standpoint.

[0077] In another embodiment, mixtures or co-feeds of the pyrolysis oil, triglycerides and FFA and other renewable feedstocks or petroleum derived hydrocarbons may also be used as the feedstock to the deoxygenation zone. The mixture of the pyrolysis oil and another renewable feedstock or a petroleum derived hydrocarbon is selected to result in greater hydrogen solubility. Other feedstock components which may be used as a co-feed component in combination with the pyrolysis oil from the above listed biomass materials, include spent motor oil and industrial lubricants, used paraffin waxes, liquids derived from gasification of coal, biomass, or natural gas followed by a downstream liquefaction step such as Fischer-Tropsch technology; liquids derived from depolymerization, thermal or chemical, of waste plastics such as polypropylene, high density polyethylene, and low density polyethylene; and other synthetic oils generated as byproducts from petrochemical and chemical processes. One advantage of using a co-feed component is the transformation of what has been considered to be a waste product from a petroleum based or other process into a valuable co-feed component to the current process.

[0078] The partial deoxygenation zone is operated at a pressure from 3.4 MPa (500 psia) to 14 MPa (3000 psia), and preferably is operated at a pressure from 3.4 MPa (500 psia) to 12 MPa (1800 psia). The partial deoxygenation zone is operated at a temperature from 100°C to 400°C with one embodiment being from 300°C to 375°C. The partial deoxygenation zone is operated at a space velocity from 0.1 LHSV h⁻¹ to 1.5 LHSV h⁻¹ based on the combined feedstock; this space velocity range does not include any contribution from a recycle stream. In one embodiment the space velocity is from 0.25 to 1.0 LHSV h⁻¹. The hydrogen to liquid

hydrocarbon feed ratio is at 5000 to 20000 scf/bbl with one embodiment being from 10,000 to 15,000 scf/bbl. The catalyst in the partial deoxygenation zone is any hydrogenation and hydrotreating catalysts well known in the art such as nickel or nickel/molybdenum dispersed on a high surface area support. Other hydrogenation catalysts include one or more noble metal catalytic elements dispersed on a high surface area support. Non-limiting examples of noble metals include Pt and/or Pd dispersed on gamma-alumina or activated carbon. Another example includes the catalysts disclosed in US 6,841,085, hereby incorporated by reference.

[0079] In the full deoxygenation zone, the conditions are more severe and the catalyst more active compared to that of the partial deoxygenation zone. The catalyst is any hydrocracking catalyst, having a hydrocracking function, that is well known in the art such as nickel or nickel/molybdenum dispersed on a high surface area support. Another example is a combined zeolitic and amorphous silica-alumina catalyst with a metal deposited on the catalyst. The catalyst includes at least one metal selected from nickel (Ni), chromium (Cr), molybdenum (Mo), tungsten (W), cobalt (Co), rhodium (Rh), iridium (Ir), ruthenium (Ru), and rhenium (Re). In one embodiment, the catalyst includes a mixture of the metals Ni and Mo on the catalyst. The catalyst is preferably a large pore catalyst that provides sufficient pore size for allowing larger molecules into the pores for cracking to smaller molecular constituents. The metal content deposited on the catalysts used are deposited in amounts ranging from 0.1 wt. % to 20 wt. %, with specific embodiments having values for the metals including, but not limited to, nickel in a range from 0.5 wt. % to 10 wt. %, tungsten in a range from 5 wt. % to 20 wt. %, and molybdenum in a range from 5 wt. % to 20 wt. %. The metals can also be deposited in combinations on the catalysts with example combinations being Ni with W, and Ni with Mo. Zeolites used for the catalysts include, but are not limited to, beta zeolite, Y-zeolite, MFI type zeolites, mordenite, silicalite, SM3, and faujasite. The catalysts are capable of catalyzing decarboxylation, decarbonylation and/or hydrodeoxygenation of the feedstock to remove oxygen as well as hydrogenation to saturate olefins. Cracking may also occur. Decarboxylation, decarbonylation, and hydrodeoxygenation are herein collectively referred to as deoxygenation reactions.

[0080] The full deoxygenation zone conditions include a relatively low pressure of 3447 kPa (500 psia) to 13,790 kPa (2000 psia), a temperature of 300°C to 500°C and a liquid hourly space velocity of 0.1 to 3 hr⁻¹ based on fresh feed not recycle. In another embodiment the deoxygenation conditions include the same pressure of 6890 kPa (1000 psia) to 6895 kPa (1700

psia), a temperature of 350°C to 450°C and a liquid hourly space velocity of 0.15 to 0.40 hr⁻¹. It is envisioned and is within the scope of this invention that all the reactions are occurring simultaneously within a zone.

CLAIMS:

1. A process for producing a fuel from a first renewable feedstocks comprising at least glycerides and a second renewable feedstock comprising at least pyrolysis oil, said process comprising:
 - 5 a) treating the first and the second renewable feedstocks concurrently in a first reaction zone by catalytically hydrogenating and catalytically deoxygenating the glycerides and components of the pyrolysis oil using at least one catalyst at reaction conditions in the presence of hydrogen to provide an effluent stream comprising hydrogen, water, carbon oxides, and hydrocarbons; and
 - 10 b) separating at least hydrogen, carbon dioxide, and water, from the effluent stream and collecting at least a portion of the remainder for use as a fuel or fuel blending component.
2. The process of Claim 1 further comprising separating the remainder into a gasoline boiling point range component, an aviation boiling point range component, and a diesel
- 15 boiling point range component.
3. The process of Claim 1 further comprising introducing at least a portion of the hydrocarbons generated in the first reaction zone to a second reaction zone to contact an isomerization and selective hydrocracking catalyst at isomerization and selective hydrocracking conditions to selectively hydrocrack at least a portion of the hydrocarbons
- 20 and to isomerize at least a portion of the hydrocarbons, to generate the effluent stream.
4. The process of Claim 3 further comprising selectively separating, in a hot high pressure hydrogen stripper, a gaseous stream comprising at least a portion of the hydrogen, water, and carbon oxides from the hydrocarbons to generate the hydrocarbons passed to the second reaction zone.
- 25 5. The process of Claim 4 further comprising recycling, to a rectification zone in the hot high pressure hydrogen stripper, a portion of the remainder.
6. The process of Claim 4 wherein the second reaction zone effluent further comprises hydrogen and at least a portion of the hydrogen is separated from the effluent stream and recycled to the hot high pressure hydrogen stripper.
- 30 7. The process of Claim 1 further comprising recycling a portion of the hydrocarbons from the first reaction zone back to the first reaction zone at a volume ratio of recycle to renewable feedstock in the range of 2:1 to 8:1.

8. The process of Claim 3 wherein the first and second reaction zones are operated at conditions including a temperature of 40°C to 400°C and a pressure of 689 kPa absolute (100 psia) to 13,790 kPa absolute (2000 psia).
- 5 9. The process of Claim 4 wherein the hot high pressure hydrogen stripper is operated at a temperature of 40°C to 300°C and a pressure of 689 kPa absolute (100 psia) to 13,790 kPa absolute (2000 psia).
10. The process of Claim 4 wherein the hot high pressure hydrogen stripper is operated at a pressure that is within 1034 kPa absolute (150 psia) that of the first reaction zone and the second reaction zone is operated at a pressure at least 345 kPa absolute (50 psia) greater
10 than that of the first reaction zone.

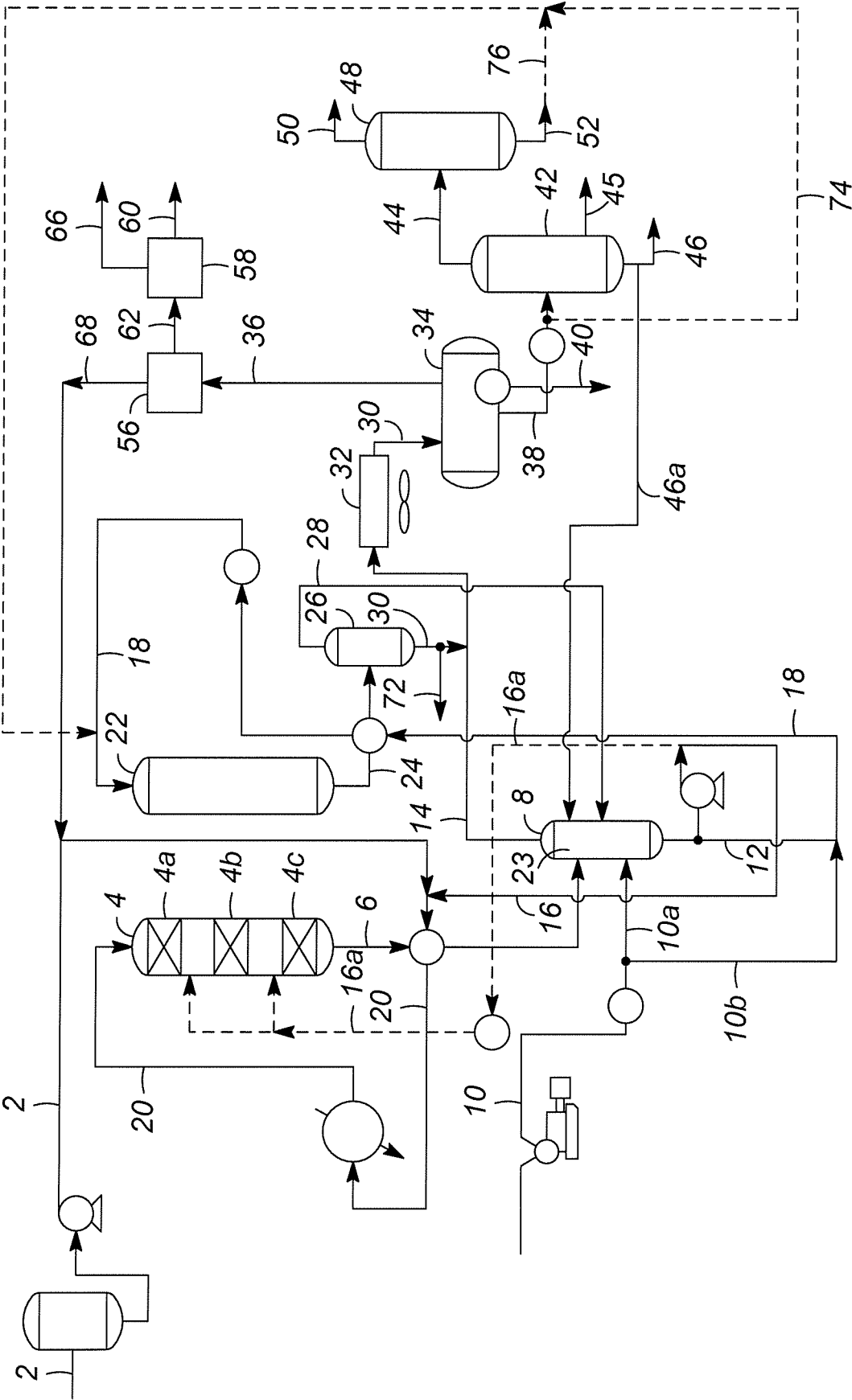


FIG. 1

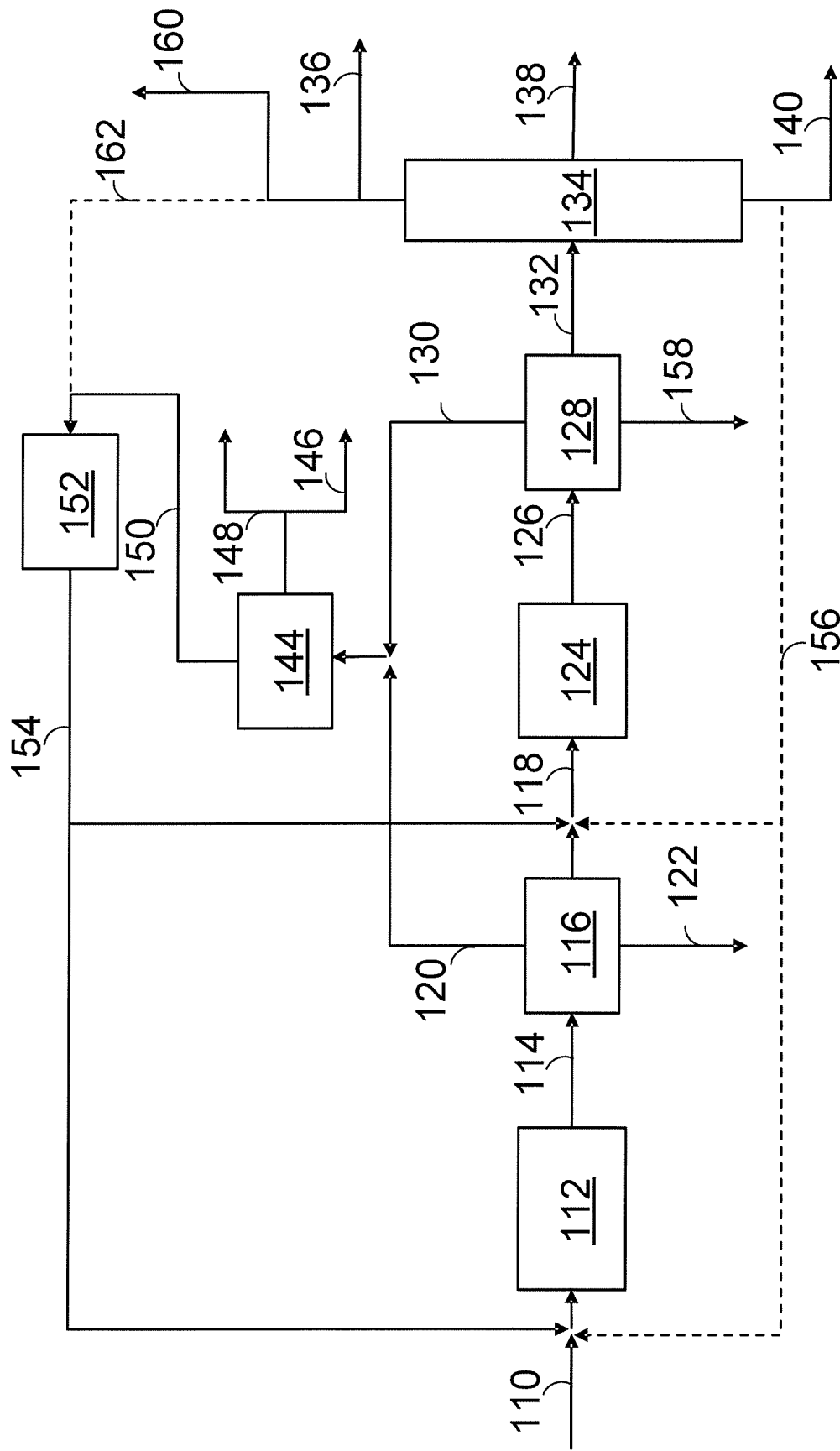


FIG. 2

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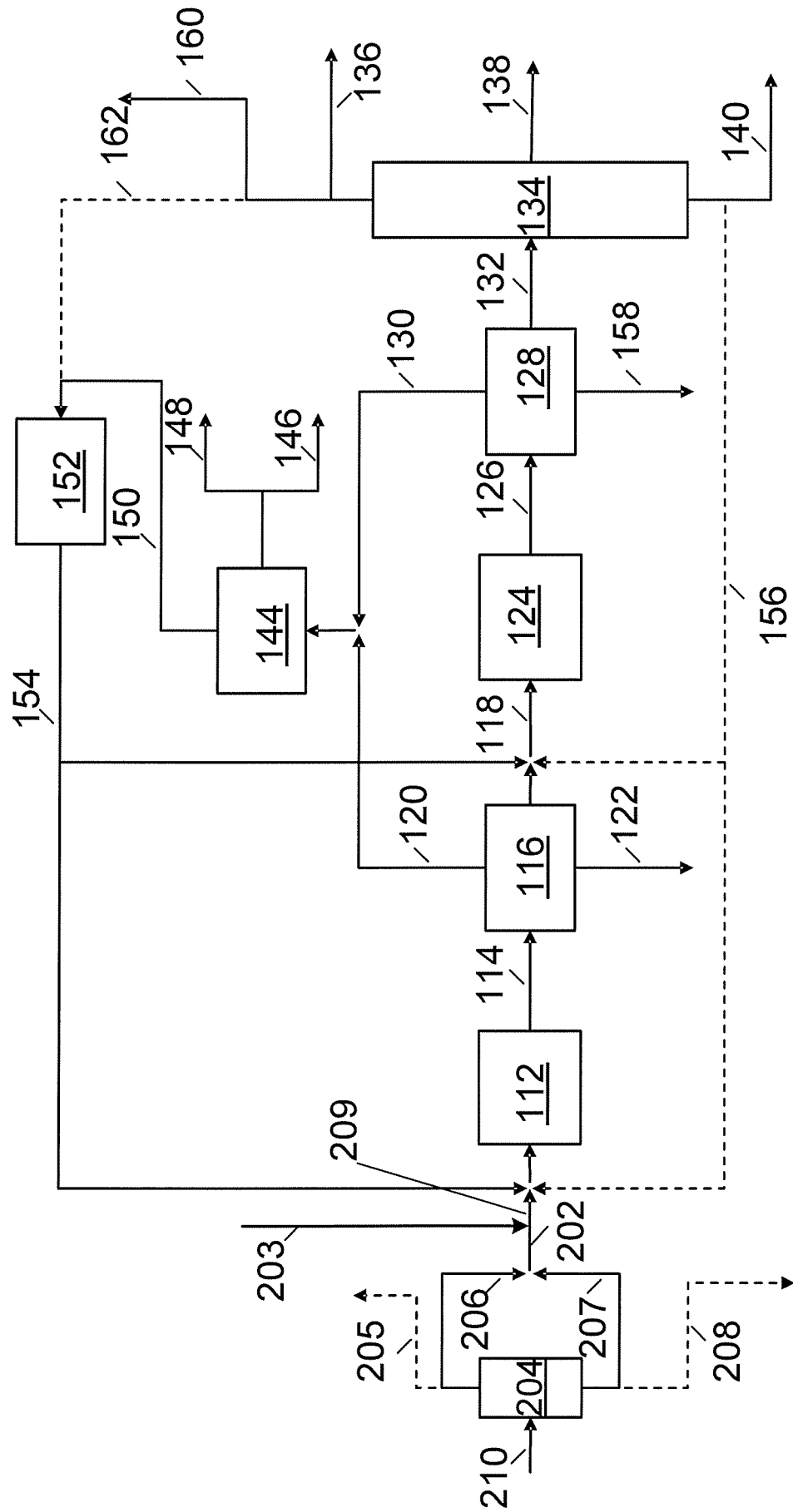


FIG. 3

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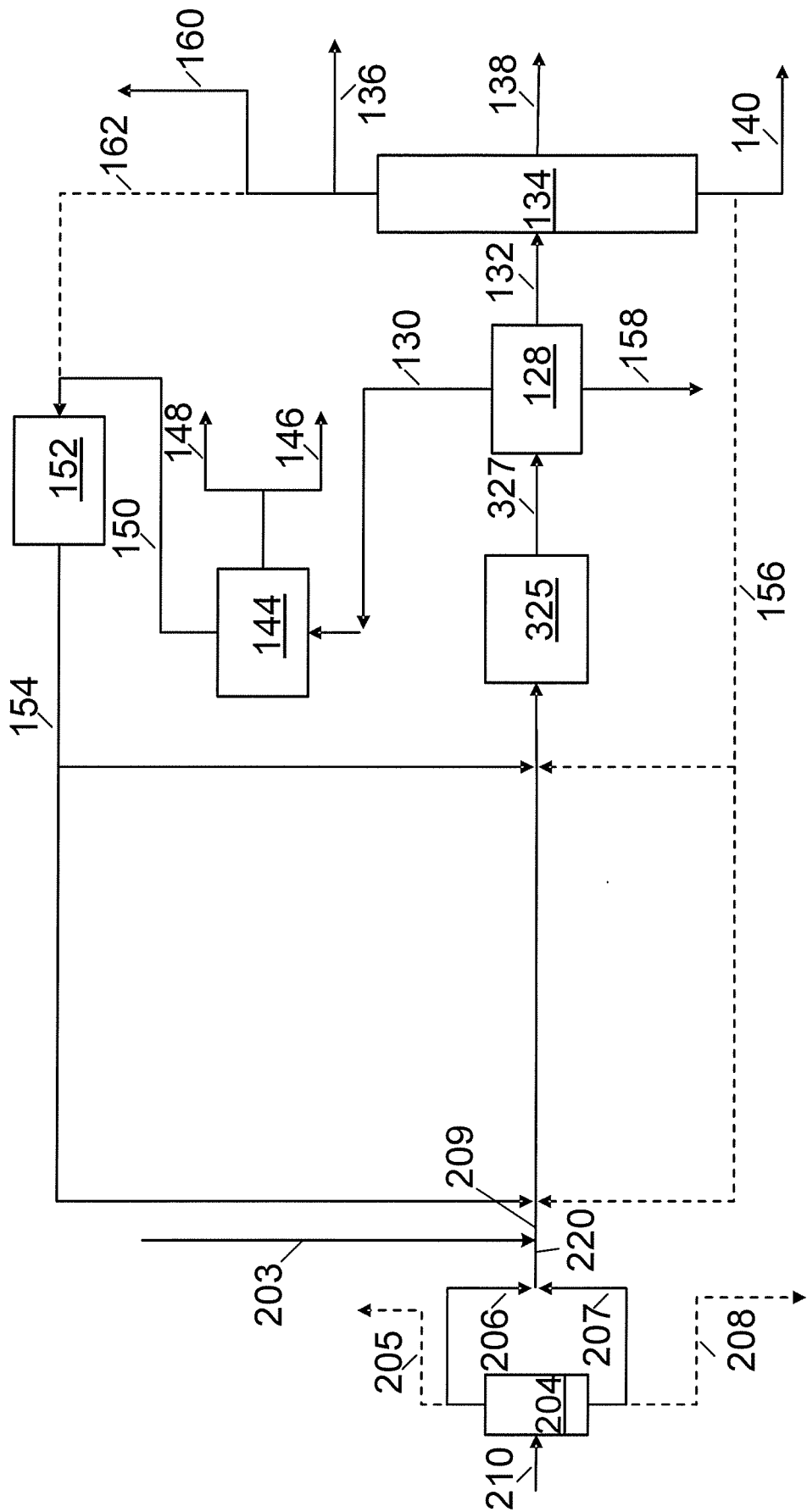


FIG. 4