There is disclosed a method for manufacturing fatty acid alkyl esters from tall oil comprising the steps of: a) esterifying tall oil in at least one esterification reactor in the presence of an acidic catalyst and an C1 to C8 alcohol to form a crude product stream comprising fatty acid alkyl esters and H2O, b) separating H2O and alcohol from the crude product stream formed in step a) to form a dehydrated fatty acid alkyl ester product stream, and c) separating dehydrated fatty acid alkyl ester product stream from step b) into at least two product streams wherein one product stream is enriched in fatty acid alkyl esters and one product stream is enriched in resin acid compounds. There are disclosed fatty acid alkyl esters and resin acids manufactured by the method. Moreover there is disclosed a fuel composition and its use as an automotive fuel, said fuel composition comprises the fatty acid alkyl esters produced according to the present invention.
AUTOMOTIVE FUELS AND FINE CHEMICALS FROM CRUDE TALL OIL

Technical field

The present invention relates to a method for manufacturing fatty acid alkyl esters from crude tall oil (CTO) and other fatty acid rich raw materials. The present invention further relates to fatty acid alkyl ester and resin acids formed in said method. Moreover the present invention relates to automotive fuels comprising fatty acid alkyl esters manufactured according to the present method.

Background

In times of high cost, for fossil fuels and greenhouse gases emission considerations the interest for producing automotive fuels and chemicals from "green sources" such as wood has increased.

From both an environmental and economical point of view it is desired to replace conventional automotive fuels from crude oil with automotive fuels from renewable resources. Tall oil is a renewable raw material originating from wood. The tall oil comprises organic compounds that can be converted to combustion engine fuels such as diesel fuel.

In addition it is also possible to recover and valorise other valuable compounds in tall oil such as resin acids and sterols.

Tall oil is a major by-product of the alkaline kraft pulping process. The tall oil originates from the extractives in the wood raw material. In the pulping process rosin acids (RA) and fatty acids (FA), which occur as free acids or their esters, are saponified by the
alkaline cooking liquor to their corresponding sodium salts. These salts, or soaps, along with neutral organic components, often called unsaponifiables, are dissolved and suspended in the spent cooking liquor (black liquor). This liquor is later concentrated and the soaps and neutrals are separated as tall oil soap skimmings. Many pulp mills are recovering this soap and after acidulation a crude tall oil (CTO) is obtained for export or upgrade at the mill.

The tall oil recovered from a softwood kraft mill typically consist of approximately 35-60% fatty acids, including oleic, linoleic, linolenic and palmitic acids, 15-55% rosin acids, including abietic, dehydroabietic and neoabietic acids and 5-35% unsaponifiable and neutral material including sterols such as beta-sitosterol. Hardwoods also contain extractives including fatty acids and neutrals (beta-sitosterol, betulin) but no resin acids.

In addition tall oil contains a small fraction of contaminants from black liquor such as sulphur compounds (up to 1000 ppm as S), lignin components and fibers. The tall oil is normally exported from the pulp mill to central tall oil distillation plants.

The production of alternative green biofuels such as biodiesel has experienced a strong growth over the past ten years. Biodiesel is normally produced from vegetable oils over catalytic transesterification to yield fatty acid alkyl esters (FAAE) which esters may be used partly or fully as a component in biodiesel fuel. In addition to replacing fossil based oil raw material, fatty acid alkyl esters are efficient lubricants in low sulphur diesel fuel.
Normally tall oil fatty acids comprise a large portion of linoleic acids providing for good cold flow properties of a biodiesel. Oxidative stability may be a problem due to the allylic and bisallylic double bonds in linoleic and linolenic acids respectively. Another concern with tall oil as raw material for biodiesel is the rather high content of sulphur compounds in tall oil (500-1000 ppm). The maximum allowable content of sulphur in biodiesel according to European and US standards for biodiesel is 10 ppm.

Over the years many processes have been suggested for recovery of valuable products from tall oil. Relevant prior art patent and patent applications are given below.

US 4,992,605 discloses a process for producing a diesel fuel additive comprising treatment of a fatty acid with gaseous hydrogen under high pressure.

US 5,705,722 describes a process for production of a diesel fuel cetane number improver comprising contacting tall oil with hydrogen at high temperature.

US 2,640,823 discloses a process for treating tall oil, wherein free fatty acids selectively are esterified with a lower alcohol. The resultant mixture is extracted with a selective polar solvent to separate rosin acids and the esters of fatty acids and unsaponifiable matter. The raffinate is distilled to obtain fatty acids in purified form. Also other components in the tall oil are extracted and recovered.

US 2,294,446 discloses a process for treatment of tall oil comprising addition of a small amount of for instance sulphuric acid. It is described that sulphonated products
resulting from the acid treatment act as catalysts. There
is shown a step with centrifugal separation of the mixture.
The mixture is then esterified using a lower alcohol such
as methanol.

WO 2004/080942 discloses a process for obtaining fatty acid
alkyl esters, rosin acids and sterols from tall oil. The
tall oil is esterified with a lower alcohol, and the
sterols are esterified with boric acid or transesterified
with a catalyst. The fatty acid esters and the rosin esters
are separated from the sterol esters. The fatty acid esters
and the rosin acids are also separated. Esterification can
be performed under acidic conditions for example using
methane sulphonic acid as catalyst.

WO 2004/074233 discloses a process of treating crude tall
oil (CTO) wherein the CTO is subjected to the steps; a)
reacting the free fatty acids in the CTO with low ex-
alkohols, b) separating the fatty acid alkyl esters from
the remaining CTO to produce a first stream of fatty acid
ester. Several subsequent steps of recovering other
components of the CTO are also disclosed.

GB 1264058 discloses a fuel composition comprising a small
amount of tall oil fatty acid.

Before the invention is disclosed and described in detail,
it is to be understood that this invention is not limited
to particular configurations, process steps and materials
disclosed herein as such configurations, process steps and
materials may vary somewhat. It is also to be understood
that the terminology employed herein is used for the
purpose of describing particular embodiments only and is
not intended to be limiting since the scope of the present
invention is limited only by the appended claims and equivalents thereof.

It should be noted that, as used in this specification and the appended claims, the singular forms "a", "an" and "the" include plural referents unless the context clearly dictates otherwise.

Summary of the present invention

The main objective of the present invention is to recover and upgrade crude tall oil to high value fine chemicals and automotive fuels. It is furthermore an objective to provide an esterification process with a higher yield of fatty acid alkyl ester than prior art. Yet another objective is to provide a method for producing an automotive biodiesel type of fuel with low sulphur content. A further objective is to provide a method for recovering fine chemicals from tall oil by selective esterification of fatty acids. Moreover there is provided a fatty acid alkyl ester, and a resin acid manufactured with the method according to the present invention. There is also provided a fuel composition comprising the fatty acid alkyl ester manufactured by the method of the present invention.

The present invention discloses an innovative sequence of reaction and separation steps enabling the production of a fatty acid alkyl ester from tall oil in high yield and with very low sulphur content. Furthermore other valuable fine chemicals such as resin acids and beta-sitosterol can be recovered in the sequential procedure described by the present invention. By continuous removal of water formed during esterification the yield of fatty acid alkyl ester is increased. Low sulphur content fatty acid alkyl ester
can advantageously be used as a biodiesel fuel component and by-products from the reaction can be upgraded to valuable fine chemicals. Addition of raw materials comprising monounsaturated/saturated fatty acids to the esterification reactor feed stream will increase the oxidative stability of a CTO based fatty acid alkyl ester/biodiesel.

The present invention thus provides a method for manufacturing fatty acid alkyl esters from tall oil comprising the steps of: a) esterifying tall oil in at least one esterification reactor in the presence of an acidic catalyst 'and a C1 to C8 alcohol to form a crude product 'stream comprising fatty acid alkyl esters and H2C15 and b) separating H2O and alcohol from the crude product stream formed in step a) to form a dehydrated fatty acid alkyl ester product stream, and c) separating dehydrated fatty acid alkyl ester_product_stream_1 into at least two product streams wherein one stream is enriched in fatty acid alkyl esters and one stream is enriched in resin acid compounds.

Further embodiments of the present invention are described in the following description and appended dependent claims.

Description of the drawing

Figure 1 shows one embodiment of the esterification method of the present invention.

In the particular embodiment described in figure 1 dehydrated and heat-treated tall oil (by which heat treatment volatile sulphur compounds is removed) is transferred through line 6 to a continuously operating...
stirred tank reactor CSTR (2). Palm fatty acid distillate comprising at least 70% free fatty acids is fed through line (7) to the CSTR (2). The proportion of palm fatty acids feed and tall oil feed to the CSTR is 1 to 1. Dry methanol is injected into the CSTR (2) through line (8). An acidic esterification catalyst PTSA (para toluene sulphonic acid) corresponding to a weight of 0.5% of total amount of fatty acids fed to the CSTR is injected into the CSTR (2) through line (9). A reaction mixture comprising fatty acid methyl ester and unreacted fatty acids, resin acids, methanol and H₂O is discharged from the CSTR (2) to a reactive distillation column (3) through line (13). Substantially all of the fatty acids fed to the reactive distillation column (3) through line (13) are converted to fatty acid methyl ester during the downward passage of fatty material in the reactive distillation column. Esterification catalyst is added to the reactive distillation column through JLLne_ (9). The temperature and pressure in the reactive distillation column (3) is selected so that unreacted methanol and H₂O is evaporated and discharged from the column through line (14). The steam (H₂O) and methanol mixture leaving the column through line (14) is cooled in cooler (19) and charged to a methanol stripper (4) wherein H₂O and methanol is separated into two streams. Methanol leaving the stripper (4) through line (15) is recycled and injected in gaseous form to the reactive distillation column (3). H₂O is leaving the methanol stripper (4) through line (11). Volatile sulphur compounds present in gaseous stream (14) can be removed by adsorption, fractionation or alkali scrubbing (not shown).

The crude fatty acid methyl ester rich product stream leaving the reactive distillation column (3) is transferred through line (18) to an evaporation/distillation column (5).
operating under vacuum (0.05 bar). A portion of the crude fatty acid methyl ester is recycled to the reactive distillation column after heating in heat exchanger (20). In the evaporation/distillation column (5) the crude fatty acid methyl ester rich product is divided in two final product streams. One product stream comprising substantially pure fatty acid methyl ester (FAME) is discharged from the upper part of the evaporation/distillation column through line (12). The FAME is further purified and exported for blending into a biodiesel automotive fuel. The second product stream comprising resin acids and neutral high boiling components is discharged from the bottom part of the evaporation/distillation column through line (16). Part of the bottom fraction is recycled and preheated in heater (15) before re-injection in the evaporation/distillation column (5). The resin acid and neutral component rich stream J16 can be used directly as a Joioijuel ox be treated _ _ for recovery of pure resin acids and beta-sitosterol.

Vacuum is provided for in the column (5) by line (17) connected to a vacuum pump system (1).

A small portion of undesired sulphur compounds are volatilised in the evaporation column (5) and removed through the vacuum system line (17). Any undesired sulphur left in the fatty acid alkyl ester product discharged through line 12 is removed by a caustic treatment (NaOH) reducing the total sulphur content of the product fatty acid alkyl ester to below 10 ppm.

It is to be understood that this invention is not limited to the particular embodiment shown above. The scope of the present invention is limited only by the appended claims and equivalents thereof.
Detailed description of the invention

The inventor of the present invention has discovered a new and efficient method for the production of pure fatty acid alkyl esters (fatty acid alkyl esters) in high yield from fatty acids present in tall oil. In addition a valuable by-product stream comprising resin acids is also recovered. Optionally a stream of neutral components may be recovered as yet another valuable by-product stream.

The feedstock material of the present invention is tall oil originating from crude tall oil soap traditionally recovered in alkaline pulp mills. The crude tall oil soap comprises fatty acid and resin acid soaps, neutral organic components and a small portion of entrained black liquor components (lignin, sulphur compounds and fibres). As a component in the crude tall oil soap could be removed by mechanical/physical separation using for example a decanter centrifuge. Mechanically/physically purified crude tall oil soap may thereafter be further purified in a step removing at least some of the neutral components by solvent extraction to form purified tall oil soap. Optionally the tall oil soap mixture is purified by solvent extraction prior to forming crude tall oil by lowering the pH of tall oil soap mixture by addition of acid.

Crude tall oil is traditionally produced in pulp mills by acidulation of tall oil soap with sulphuric acid. The tall oil is often dried in a dehydrator to form a substantially dry tall oil prior to export from the mill.
Crude tall oil (CTO) recovered in accordance with the procedures described above is a raw material feed for the esterification reactor or reactors of the present invention. Other fatty acid containing material may also be fed to the esterification reactor or reactors. The objective of esterification is to form fatty acid alkyl esters (fatty acid alkyl esters) in high yield. The esterification plant comprises at least one reactor with a catalytic esterification stage wherein fatty acids present in the CTO are selectively esterified in the presence of a catalyst and a C1 to C8 alcohol such as methanol, ethanol or iso-propanol. Preferably the alcohol is methanol or ethanol.

The physical conditions in the reactor or reactors and the catalyst are preferably selected so that fatty acids in the tall oil are esterified in preference to resin acids. It is well known that fatty acids with primary carboxylic acid groups are esterified at milder conditions relative to resin acids and this fact is exploited for example in analytic procedures to quantify the portion of fatty acids in tall oil.

A problem with fatty acid esterification is the formation of $H_2O$, which drives the equilibrium esterification reaction backwards. Therefore $H_2O$ should be removed from the reaction mixture in order to obtain fatty acid alkyl esters in high yield. The initial step of a fatty acid esterification reaction is the protonation of the acid to give an oxonium ion, which undergoes an exchange reaction with an alcohol to give an intermediate reactant. The intermediate reactant in turn will lose a proton to become an ester. Each step in the process is reversible but in the presence of very large excess of the alcohol, the
equilibrium of the reaction is displaced so that esterification proceeds virtually to completion. However, in the presence of H₂O, which is a stronger electron donor than are aliphatic alcohols, formation of the intermediate is not favoured and esterification will not proceed fully.

Thus H₂O has to be removed from the reaction mixture to achieve a high esterification yield. The total yield of fatty acid alkyl ester calculated on fatty acids in the feed streams to the reactor or reactors is over 80 %, preferably over 90 % and may in some preferred embodiments be as high as 98 % provided that H₂O is continuously and efficiently removed from the reaction mixture. Alcohols with a lower boiling point than H₂O are in one embodiment removed 'together with the alcohol from the crude product stream' during the esterification reaction. The H₂O and alcohol vapours are collected and separated into a H₂O rich stream and an alcohol rich stream. The alcohol rich stream is recycled to the esterification reactor or reactors.

In one embodiment the sulphur compounds are separated from the crude product stream together with the separation of H₂O and alcohol.

The free fatty acids present in the CTO are precluding the use of alkaline catalysts for the esterification reaction. Alkaline catalysts are normally used in the state of the art production of biodiesel or fatty acid alkyl esters through transesterification of vegetable oils. Alkaline components will convert fatty acids into their soaps, which in turn will create emulsification problems. Formation of emulsions is undesirable by inhibiting mass transfer during esterification and lowers the reaction yield. Therefore the
The present invention discloses the use of an acidic catalyst which catalyst can be either homogeneous or heterogeneous.

While sulphuric acid or other strong mineral acids can be used as catalyst, organosulphonic acids are particularly suitable homogeneous acidic catalysts for use in the present invention. Spent acid catalyst that is not recycled within the process itself can conveniently be separated and recycled to a Kraft pulp mill liquor cycle. Particularly preferred organosulphonic acids include para-toluene sulphonic acid and methane acid. Heterogeneous fatty acid esterification catalysts are known and in particular solid resinous catalysts including organo acid-functionalised mesoporous silica catalysts can be used in the practise of the present invention. Other solid acidic heterogeneous catalysts that can be used include sulphated zirconium (SZ), tungstated zirconium (WZ), commercially available Nafion supported on silica (SAC-13). The acid catalyst preferably comprises at least one compound selected from the group consisting of sulphuric acid, organic acids such as para-toluene sulphonic acid and methane acid, solid resinous catalysts based on methane sulphonic acids, organosulphonic acid-functionalised mesoporous silica, sulphated zirconium, tungstated zirconium and mixtures thereof.

Esterification reactions converting fatty acids in CTO in accordance with the procedures described above can be performed in batch or semi batch reactors. However it is preferred 'to conduct at least a portion of the esterification reactions in continuous reactors such as continuous stirred tank reactors (CSTR). A particularly preferred reactor system for performing the esterification
reactions is based on reactive distillation technology alone or in combination with at least one CSTR. An arrangement for refluxing excess alcohol can be used as an alternative to reactive distillation. In a preferred embodiment of the present invention the esterification reactions are performed using at least one continuous stirred tank reactor (CSTR) and at least one reactive distillation reactor in series.

Reactive distillation combines chemical reaction and distillation in one vessel. The combination of reaction and separation in one piece of equipment offers distinct advantages over conventional, sequential approaches. Especially for equilibrium limited reactions such as esterification and ester hydrolysis reactions, conversion can be increased far beyond chemical equilibrium conversion due to the continuous removal of reaction products from the react±DH zone.

Because of complex interactions between chemical reaction and separation, the performance of a reactive distillation column is influenced by several parameters, e.g. size and location of reactive and non-reactive column sections, reflux ratio, feed location, or throughput. A pre-reactor, for example a CSTR, may be installed to mix the reactants CTO and alcohol (optionally also catalyst) prior to charging the reactants to the reactive distillation column. The reaction temperature in the pre-reactor and reactive distillation column is selected for achieving optimum conversion of the fatty acids present in the CTO into alkyl esters minimizing the conversion of the more stable resin acids to esters. Such temperature, normally in the range from about 50 to about 250°C is also selected to minimize undesired side reactions such as dimerisation of the fatty
acids. The pressure in the reactive distillation column is from 1 - 50 bars, preferably in the range from 3-30 bars. Thus the tall oil esterification reactions are preferably carried out at a temperature in the range from about 50 to about 250°C, preferably in the range from 65 to 140°C.

When a homogeneous catalyst is used to catalyze the esterification reactions the column should preferably have structured packing internals. When a heterogeneous catalyst is used the catalyst is preferably immobilised by a column packing structure such as for example Katapak by Sulzer Chemtec.

Certain alcohols in the range C2-C8 including ethanol form azeotropes with H₂O, which complicates the recycling of H₂O-free alcohol to the esterification reaction zone. In applications using an azeotrope forming alcohol as a reactant, entrainers can be added to the reaction mixture to promote separation of excess alcohol and H₂O. Entrainers are well known for use as supporting additives in the art of azeotropic distillation. Without entrainer only the temperature and pressure could change the physical properties of the CTO alcohol mixture in the reactive distillation column. Changing temperature and/or pressure is not always possible, in particular for non-ideal mixtures such as CTO and alcohol. Furthermore such changes, may be incompatible with the reaction requirements. An important feature is that a reactive distillation process can be designed such that recycling of entrainer is realised internally, and not externally, as in conventional distillation processes. Suitable entrainer chemicals that have the appropriate solubility and azeotrooic form characteristics for use.
in the present CTO esterification process, includes alkanes such as hexane, alkenes such as 1-hexene and other organic compounds such as cyclohexadiene, propyl acetate, pentanone and di-propyl ether.

Neutral components present in the CTO feed do not react with alcohol under the conditions selected and neutrals are discharged from the esterification reactor or reactors with the high boiling temperature fraction (fatty acid alkyl ester /resin acid mixture). Spent homogeneous catalyst may also be present in discharged high boiling fraction.

A stoichiometric surplus charge of alcohol is normally used to drive the conversion of CTO to fatty acid alkyl esters to completion. The stoichiometric ratio of alcohol to fatty acids present in the feed is in the order of 1:1:1 to 3:1. Low boiling alcohol that has not reacted with the CTO will pass upwards through a reactive distillation column. A refluxing section is optionally installed in the upper part of the column. Unreacted alcohol can be separated from H₂O outside a reactive distillation column by pervaporation or stripping, such recovered alcohol is preferably recycled to a reactive distillation column or to a pre-reactor to the column. The point of charge for CTO, homogeneous catalyst and alcohol have to be selected for the specific alcohol, catalyst and CTO composition and such optimum charging points can easily be determined by the artisan skilled in the art.

In cases where no pre-reactor is used a major portion of the alcohol is charged in gaseous form to the lower section of a reaction distillation column. The CTO and catalyst (if homogeneously catalysed process) is preferably charged to the upper section of the column and the reactants are thus
passing counter currently with the gaseous alcohol through the column. A small portion of the alcohol may be entrained with the high boiling fatty acid alkyl ester /resin acid fraction to a sump recycler in the lower section of column. The presence of alcohol in the lower part of the column is advantageous in minimizing undesired side reactions.

Steam and excess alcohol is continuously evaporated or distilled off the esterification reaction mixture. Alcohol is recovered from the steam/alcohol mixture by stripping or pervaporation and is recycled to the alcohol storage or directly to the esterification reactor/reactors. The crude product stream rich in fatty acid alkyl esters also comprises resin acids and a portion of high boiling neutral components including sterols and squalene.

A sufficiently pure stream of fatty acid alkyl esters is obtained taking advantage of the considerably higher vapour pressure and lower boiling point of the fatty acid alkyl esters relative to resin acids and neutrals. Typically, a distillation column, a short path evaporator or any other type of efficient evaporator operated under vacuum can be used to separate the fatty acid alkyl esters from the higher boiling resin acids and neutrals. The vacuum pressure should be selected considering the feed oil composition, particularly fatty/resin acid ratio, and should be in the range of 1 bar down to 0.0005 bar, preferably 0.8 to 0.005 bar. During this unit operation of the process also undesired organosulphur compounds can be removed thereby forming a third process stream. This third process stream rich in sulphur is discharged from the distillation or evaporator separately from the fatty acid alkyl ester product stream. In one embodiment a third stream enriched in volatile sulphur compounds is separated...
from said crude product stream. Said separation is in one embodiment accomplished by utilising the difference in vapour pressure between sulphur compounds and fatty acid alkyl ester and resin acids.

While fatty acid alkyl esters are a main product from process, also the resin acids/neutral component mixture can be valorised and used for various purposes. The resin acids/neutral mixture can advantageously be further fractionated for recovery of fine chemicals. The neutrals can be separated from the resin acids by an aqueous alkaline wash dissolving the resin acids in the form of soaps. Neutrals are not dissolved in aqueous alkaline solutions. Additives such as polyelectrolytes and surfactants may be added to prevent emulsification, formation of micelles and other colloid structures disturbing the separation of neutrals and resin acid soaps in two phases. The resin acids can thereafter be obtained in reasonably pure form by acidulating the aqueous soap stream. Due to the high melting points of most resin acids the acidulation should be performed at a temperature above 70°C. The neutral rich stream can be removed and purified to obtain valuable fine chemicals. Non-limiting examples of components in the neutral rich stream include beta-sitosterol.

Reasonably pure resin acids may also be obtained by washing the product stream directly from the esterification stage with an aqueous alkaline washing liquid. The resin acids are solubilised in the washing liquid, while the fatty acid alkyl esters are combined with the neutrals to form a lipophilic phase. Again, as disclosed above, the resin acids can be obtained after acidulation of the alkaline washing liquid. Fatty acid alkyl esters can be purified
from neutrals by evaporation, taking advantage of the large
difference in vapour pressure and boiling points of
neutrals and fatty acid alkyl esters.

By the procedure described it is possible to manufacture
three valuable products from the CTO; one stream rich in
fatty acid alkyl esters, one stream rich in resin acids and
one stream rich in neutral components.

The FAAE rich stream can be further treated by at least one
method selected from purification, hydrogenation and
dimerisation. FAAEs can be used as a component in biodiesel
fuel or as a raw material in fine chemicals synthesises to
surfactants and lubricants.

The current demand for fatty acid alkyl esters with very
low sulphur content can be met by either stripping of
organosulphur compounds from tall oil or reaction mixture
streams as described above or by purification of the FAAE
rich product stream. Examples of such methods include
alkali treatment and selective sulphur adsorption. The
sulphur of the fatty acid alkyl esters is thereby decreased
to a level below about 300 ppm, preferably below about 50
ppm and most preferably to a level below about 10 ppm
sulphur.

While the method for producing fatty acid alkyl esters in
accordance with the description above is based on crude
tall oil as fatty acid feedstock also other fatty acid rich
raw materials can be added and used as feed to the
esterification reactor. Such fatty acid rich feedstocks
include for example palm oil fatty acid distillates or
other fatty acid rich materials containing oleic, palmitic
or stearic acids. A large fraction of triglycerides, if
present in such feedstocks, will be transesterified to fatty acid alkyl esters in the esterification reactors. In one embodiment of the present invention a stream comprising at least one fatty acid in addition to tall oil is added to the esterification step. Said fatty acid is preferably at least one fatty acid selected from the group consisting of palm fatty acid distillate, stearic, palmitic or oleic acid. An advantage of feeding saturated or monounsaturated fatty acid rich material such as palm fatty acids together with tall oil to the esterification reactor is that the oxidative stability of the product fatty acid alkyl esters can be improved. Furthermore, such addition of fatty acid rich feed to the esterification reactor or reactors enables the manufacturer to control the iodine number of the product fatty acid alkyl esters.

The procedure described herein for preparation of fatty acid alkyl esters can be combined with a crude tall oil soap neutrals purification process. Such processes are well known in the art and are often based on solvent extraction with the addition of demulsifiers to break crude tall oil soap H₂O emulsions. Besides the value of the neutral components as raw material for food ingredients and pharmaceuticals the purified tall oil material obtains a higher acid value. Acid value is a quality parameter for crude tall oil and a higher acid value tall oil feed is also advantageous in the method of the present invention. Neutral rich streams recovered from any of the process stages of the present invention can be further treated for example by evaporation, distillation or by crystallization to recover sterols and other organic fine chemicals originally present in the crude tall oil soap. In one embodiment the tall oil soap is purified in order to
increase the acid value of tall oil and to recover neutral components present in the tall oil soap mixture.

In order to prevent undesired esterification reactions between fatty acids in tall oil on one hand and fatty-alcohols and sterols also present in tall oil on the other hand, a C1-C8 alcohol can be added to the tall oil prior to performing the catalysed esterification step. Such addition has the advantage that an amount of the desired fatty acid alkyl ester is formed during for instance storage already before the catalysed esterification step is performed. Undesired esterification reactions between sterols and fatty acids are also inhibited.

In one embodiment volatile sulphur compounds are separated from tall oil by evaporation from tall oil prior to performing the esterification step.

The fatty acid alkyl ester product stream is in one embodiment, after the removal of H₂O and alcohol, divided into two separate process streams by evaporative separation at vacuum utilising the difference in boiling point between fatty acid alkyl esters and resin acids/neutral components.

In one embodiment a resin acid- and neutral component rich stream is separated into a resin acid rich stream and a neutral component rich stream by dissolution of resin acids as soaps in an alkaline aqueous mixture. In a preferred embodiment resin acids are recovered from resin acid soaps by treatment with an acid.

In yet another aspect the present invention, a biodiesel fuel composition comprising a fatty acid alkyl ester is provided. Such fuel composition is either essentially pure
fatty acid alkyl ester or a blended biodiesel fuel composition comprising in addition to fatty acid alkyl ester also fossil fuels and other optional additives. Typical examples of such blended fuels are normally called B5 or B20 denoting the percentage of biodiesel in standard diesel fuels.
Claims

1. A method for manufacturing fatty acid alkyl esters from tall oil comprising the steps of:

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a. esterifying tall oil in at least one esterification reactor in the presence of an acidic catalyst and a C1 to C8 alcohol to form a crude product stream comprising fatty acid alkyl esters and H₂O,

b. separating H₂O and alcohol from the crude product stream formed in step a) to form a dehydrated fatty acid alkyl ester product stream, and

c. separating dehydrated fatty acid alkyl ester product stream from step b) into at least two product streams wherein one product stream is enriched in fatty acid alkyl-esters and one product stream is enriched in resin acid compounds.

2. The method according to claim 1 wherein volatile sulphur compounds are separated from tall oil by evaporation from tall oil prior to esterification in step a)

3. The method according to claim 1 or 2 wherein volatile sulphur compounds are separated from the crude product stream together with the separation of H₂O and alcohol in step b)

4. The method according to any one of claims 1 to 3, wherein the stream enriched in fatty acid alkyl ester in step c) is further treated to remove sulphur compounds to a level below 10 ppm sulphur.
5. The method according to any one of claims 2 to 4, wherein the sulphur compounds are removed by at least one method selected from the group consisting of alkali treatment and adsorption.

6. The method according to any one of claims 1 to 5 wherein a third stream enriched in sulphur compounds is separated during step c), said separation accomplished by utilising the difference in vapour pressure between sulphur compounds and fatty acid alkyl ester and resin acids.

7. The method according to any one of claims 1 to 6, wherein an amount of a C1 to C8 alcohol is added to the tall oil prior to performing step a) of claim 1.

8. The method according to any one of claims 1 to 7 wherein said tall oil is purified by solvent extraction of tall oil soap prior to forming said tall oil.

9. The method according to any one of claims 1 to 8, wherein said C1 to C8 alcohol is methanol or ethanol.

10. The method according to any one of claims 1 to 9, wherein said at least one esterification reactor is a continuous stirred tank reactor (CSTR).

11. The method according to any one of claims 1 to 10, wherein said at least one esterification reactor is selected from the group consisting of a reactive distillation reactor and a reactor arrangement for refluxing excess alcohol.
12. The method according to any one of claims 1 to 11, wherein the esterification of tall oil is performed using at least one continuous stirred tank reactor (CSTR) and at least one reactive distillation reactor in series.

13. The method according to any one of claims 1 to 12, wherein step a) is carried out at a temperature in the range from about 50 to about 250 °C.

14. The method according to any one of claims 1 to 13, wherein step c) is carried out at a pressure in the range from about 1 bar to about 0.0005 bar.

15. The method according to any one of claims 1 to 14, wherein the H₂O and alcohol which is removed from the reaction mixture in step b) is separated for removal of H₂O and wherein the alcohol is recycled to the at least one esterification reactor.

16. The method according to any one of claims 1 to 15 wherein an entrainer is added to the reaction mixture to promote separation of excess alcohol and H₂O.

17. The method according to any one of claims 1 to 16, wherein a stream comprising at least one fatty acid in addition to tall oil is added to be present in step a) of claim 1.

18. The method according to claim 17, wherein said fatty acid is at least one fatty acid selected from the group consisting of palm fatty acid distillate, stearic, palmitic or oleic acid.
19. The method according to any one of claims 1 to 18, wherein said acid catalyst comprises at least one compound selected from the group consisting of sulphuric acid, organic acids such as para-toluene sulphinic acid and methane sulphinic acid, solid resinous catalysts based on sulphinic acids, organosulphinic acid-functionalised mesoporous silica, sulphated zirconium, tungstated zirconium and mixtures thereof.

20. The method according to any one of claims 1 to 19, wherein the dehydrated fatty acid alkyl ester product stream from step b) is divided into two separate process streams by evaporative distillation at vacuum utilising the difference in boiling point between fatty acid alkyl esters and resin acids/neutral components.

21. The method according to any one of claims 1 to 20 wherein a resin acid and neutral component stream is separated into a resin acid rich stream and a neutral component rich stream by dissolution of resin acids as soaps in an alkaline aqueous mixture.

22. The method according to claim 21 wherein resin acids are recovered from resin acid soaps by treatment with an acid.

23. A fatty acid alkyl ester manufactured according to the method of any one of claims 1 to 22.

24. A resin acid manufactured according to the method of any one of claims 1 to 22.
25. A fuel composition comprising a fatty acid alkyl ester manufactured according to the method of any one of claims 1 to 22.

26. Use of the fuel composition of claim 25 as an automotive fuel.
A. CLASSIFICATION OF SUBJECT MATTER

B. SEARCHED DOCUMENTS

C. DOCUMENTS CONSIDERED TO BE RELEVANT

D. Further documents are listed in the continuation of Box C.

E. See patent family annex.
**INTERNATIONAL SEARCH REPORT**

International patent classification (IPC)

*C11C 3/04* (2006.01)  
*C07C 68/08* (2006.01)  
*C07C 69/003* (2006.01)  
*C10L 1/02* (2006.01)  
*C10L 1/18* (2006.01)  
*C11B 13/00* (2006.01)

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Cited literature, if any, will be enclosed in paper form.
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