



(51) International Patent Classification:

C01B 32/194 (2017.01) H01M 4/36 (2006.01)

C01B 32/156 (2017.01) H01G 11/36 (2013.01)

H01M 4/133 (2010.01) B82Y 40/00 (2011.01)

H01M 4/1393 (2010.01)

(21) International Application Number:

PCT/US2021/040375

(22) International Filing Date:

02 July 2021 (02.07.2021)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

63/048,504

06 July 2020 (06.07.2020)

US

(71) Applicant: LAWRENCE LIVERMORE NATIONAL

SECURITY, LLC [US/US]; 7000 East Avenue, P.O. Box 808, L-703, Livermore, California 94550 (US).

(72) Inventors: CAMPBELL, Patrick G.; c/o Lawrence Liv-

ermore National Laboratory, 7000 East Avenue, P.O. Box 808, L-703, Livermore, California 94550 (US). BIENER,

Monika M.; c/o Lawrence Livermore National Laboratory, 7000 East Avenue, P.O. Box 808, L-703, Livermore, California 94550 (US). HERNANDEZ, Maira Raquel Ceron; c/o Lawrence Livermore National Laboratory, 7000 East Avenue, P.O. Box 808, L-703, Livermore, California 94550 (US).

(74) Agent: KONSKI, Antoinette F. et al.; Foley & Lardner LLP, 3000 K Street, N.W., Suite 600, Washington, District of Columbia 20007-5109 (US).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DJ, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, IT, JO, JP, KE, KG, KH, KN, KP, KR, KW, KZ, LA, LC, LK, LR, LS, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC, SD, SE, SG, SK, SL, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, WS, ZA, ZM, ZW.

(54) Title: GRAPHENE MACRO-ASSEMBLY-FULLERENE COMPOSITE FOR ELECTRICAL ENERGY STORAGE

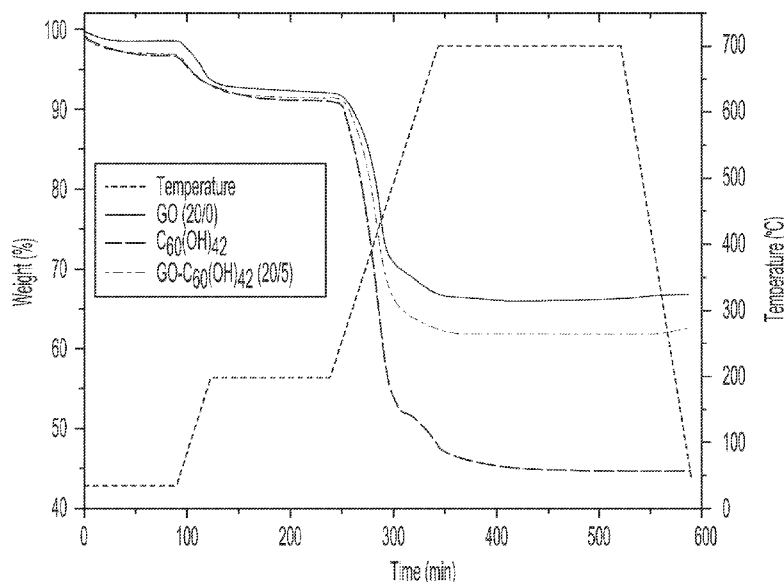


FIG. 11

(57) Abstract: Disclosed here is a method for producing a graphene macro-assembly (GMA)-fullerene composite, comprising providing a mixture of graphene oxide and water, adding a hydroxylated fullerene to the mixture, and forming a gel of the hydroxylated fullerene and the mixture. Also described are a GMA-fullerene composite produced, an electrode comprising the GMA-fullerene composite, and a supercapacitor comprising the electrode.



(84) **Designated States** (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, ST, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

Declarations under Rule 4.17:

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))*
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))*

Published:

- *with international search report (Art. 21(3))*
- *before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments (Rule 48.2(h))*

GRAPHENE MACRO-ASSEMBLY-FULLERENE COMPOSITE FOR ELECTRICAL ENERGY STORAGE

CROSS-REFERENCE TO RELATED APPLICATIONS

[0001] The present application claims the benefit of U.S. Provisional Application No. 63/048,504, filed July 6, 2020, which is incorporated by reference in its entirety herein.

FEDERAL FUNDING STATEMENT

[0002] This invention was made with Government support under Contract No. DE-AC52-07NA27344 awarded by the United States Department of Energy. The Government has certain rights in the invention.

BACKGROUND

[0003] Supercapacitors (also known as ultracapacitors or electrical double-layer capacitors) have the potential to replace Li ion batteries as the next-generation electrical energy storage technology in demanding applications due to their high power density and excellent cycling stability. Graphene-based supercapacitor electrodes are particularly promising because they feature high surface area, good electrical conductivity, and chemical inertness. Researchers at Lawrence Livermore National Laboratory have developed binder-free 3D mesoporous graphene macro-assemblies (GMAs) that have exceptionally high surface area ($\sim 1500 \text{ m}^2/\text{g}$) and excellent conductivity ($\sim 100 \text{ S/m}$) using abundant and low cost starting materials. These GMAs offer many advantages over traditional carbon-based supercapacitor electrodes such as deterministic control over pore morphology, increased conductivity, and the absence of conductive filler and binder materials. However, the interfacial capacitance of graphene-based electrodes is limited by the low density of states at the Fermi level to $\sim 10 \text{ mF/cm}^2$ (corresponding to 0.01 electron per carbon atom for the stability window of aqueous electrolytes). To replace Li-ion batteries in energy-demanding applications, these materials need improvements to their energy storage performance.

[0004] Fullerenes (also known as Bucky-balls) can store 10 times the energy per carbon as graphene (6 electrons per C_{60} molecule or 1 electron per 10 carbon atoms). Since the discovery of C_{60} , fullerenes have attracted pronounced attention due to their applications in medicinal chemistry (as MRI contrast agents, in tumor diagnosis and radio-immunotherapy), material science and photovoltaic solar cells, among others. Functionalization or chemical modification of fullerenes has been used to increase their solubility, allow their characterization

and explore their physical and chemical properties. Fullerenes possess highly reactive double bonds that allow the study of their reactivity using different types of reactions, such as oxidation reactions, transition metal complexation, hydrogenations, halogenations, radical additions, cycloadditions (1,3-dipolar, [2+2], [4+2], [3+2], [2+2+1]), addition of nucleophiles (Bingel additions), silylations and electrosynthesis.

SUMMARY

[0005] One aspect the invention described herein relates to a method for producing a GMA-fullerene composite. The method can include providing graphene oxide, adding a hydroxylated fullerene compound to the graphene oxide, and performing a gelation process using the graphene oxide and the hydroxylated fullerene.

[0006] In some embodiments, providing the graphene oxide includes providing a water suspension of graphene oxide.

[0007] In some embodiments, the water suspension of graphene oxide includes a mixture of at least about 7 mg of graphene oxide in 1 mL of water.

[0008] In some embodiments, the hydroxylated fullerene compound includes a C₆₀ hydroxylated fullerene, a C₇₀ hydroxylated fullerene, or a C₁₀₀ hydroxylated fullerene.

[0009] In some embodiments, adding the hydroxylated fullerene compound includes adding about 5mg of water soluble C₆₀(OH)₄₂ to the graphene oxide, the graphene oxide comprising a mixture of about 20 mg of graphene oxide in 1 mL of water.

[0010] In some embodiments, the method includes sonicating the hydroxylated fullerene and the graphene oxide.

[0011] In some embodiments, the method includes adding, to about 1 g of a suspension formed from the hydroxylated fullerene compound and the graphene oxide, about 211 μ L of ammonium hydroxide solution to form a mixture, wherein performing the gelation process includes performing the gelation process using the mixture.

[0012] In some embodiments, performing the gelation process includes performing the gelation process at about 80 °C for about 72 hours.

[0013] In some embodiments, performing the gelation process provides a graphene oxide-fullerene compound aerogel, and the method includes drying and carbonizing the graphene oxide-fullerene compound aerogel.

[0014] In some embodiments, carbonizing the graphene oxide-fullerene compound aerogel includes carbonizing the graphene oxide-fullerene compound aerogel at about 700 °C.

[0015] In some embodiments, the method includes performing at least one acetone wash of the graphene oxide-fullerene compound aerogel prior to drying the graphene oxide-fullerene compound aerogel.

[0016] In some embodiments, the GMA-fullerene composite comprises at least 25 weight percent of fullerene.

[0017] In some embodiments, performing the gelation process includes covalently bonding the hydroxylated fullerene compound with graphene sheets formed from the graphene oxide.

[0018] A further aspect of the invention relates to a GMA-fullerene composite produced as described herein.

[0019] An additional aspect of the invention relates to a supercapacitor that includes a GMA-fullerene composite produced as described herein.

[0020] These and other features, together with the organization and manner of operation thereof, will become apparent from the following detailed description when taken in conjunction with the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

[0021] FIG. 1: (A) a GMA-C₆₀ composite obtained by noncovalently functionalizing the GMA with 4.5 mM C₆₀; (B) CV of a GMA-C₆₀ with 13% loading of C₆₀ in MeCN/TBAP electrolyte (5 mV/s).

[0022] FIG. 2: (A) a GMA-PA-C₆₀ composite obtained by noncovalently functionalizing the GMA with 7 mM PA-C₆₀; (B) CV of a GMA-PA-C₆₀ composite with 50% loading of PA-C₆₀ in MeCN/TBAP electrolyte (5 mV/s).

[0023] FIG. 3: CV of a non-functionalized GMA in MeCN/TBAP electrolyte (5 mV/s).

[0024] FIG. 4: CV of a GMA-fullerene composite with a final loading of 138% in MeCN/TBAP electrolyte (10 mV/s). This GMA-fullerene composite was obtained by covalently functionalizing the GMA with diazonium-C₆₀.

[0025] FIG. 5: Examples of phenylamine-functionalized fullerene compounds for

functionalization of GMA.

[0026] FIG. 6: Additional examples of fullerene compounds for functionalization of GMA.

[0027] FIG. 7: CV in MeCN/TBAP electrolyte of a GMA-C₆₀ with 17 wt.% loading of C₆₀ (5 mV/s, cycle 1, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line).

[0028] FIG. 8: CV in MeCN/TBAP electrolyte of a GMA-C₆₀-1 composite with 44 wt.% loading of C₆₀-1 (5 mV/s, cycle 1, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line).

[0029] FIG. 9: CV in MeCN/TBAP electrolyte of a covalently functionalized GMA-fullerene composite with diazonium-C₆₀, final loading 138 wt.% (5 mV/s, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line).

[0030] FIG. 10: Examples of diazonium-functionalized fullerene compounds for functionalization of GMA.

[0031] FIG. 11: TGA of GO (blue line), C₆₀(OH)₄₂ (gray line), and GO-C₆₀(OH)₄₂ (green line).

[0032] FIG. 12: CV of GMA (black line) and C₆₀(OH)₄₂ carbonized at 700° C (red line, cycle 1; blue line, cycle 15).

[0033] FIGS. 13A-13C: SEMs uncarbonized and carbonized GMA.

DETAILED DESCRIPTION

[0034] Reference will now be made in detail to some specific embodiments of the invention contemplated by the inventors for carrying out the invention. Certain examples of these specific embodiments are illustrated in the accompanying drawings. While the invention is described in conjunction with these specific embodiments, it will be understood that it is not intended to limit the invention to the described embodiments. On the contrary, it is intended to cover alternatives, modifications, and equivalents as may be included within the spirit and scope of the invention as defined by the appended claims.

[0035] In the following description, numerous specific details are set forth in order to provide a thorough understanding of the present invention. Particular example embodiments of the present invention may be implemented without some or all of these specific details. In other instances, well known process operations have not been described in detail in order not

to unnecessarily obscure the present invention.

[0036] Various techniques and mechanisms of the present invention will sometimes be described in singular form for clarity. However, it should be noted that some embodiments include multiple iterations of a technique or multiple instantiations of a mechanism unless noted otherwise.

[0037] **Introduction**

[0038] The article, “Mechanically Robust 3D Graphene Macroassembly with High Surface Area,” Worsley *et al.*, *Chem. Commun.*, 48:8428-8430 (2012), is incorporated by reference in its entirety.

[0039] The article, “Toward Macroscale, Isotropic Carbons with Graphene-Sheet-Like Electrical and Mechanical Properties,” Worsley *et al.*, *Adv. Funct. Mater.*, 24:4259-4264 (2014), is incorporated by reference in its entirety.

[0040] The article, “Synthesis and Characterization of Highly Crystalline Graphene Aerogels,” Worsley *et al.*, *ACS Nano*, 8:11013-11022 (2014), is incorporated by reference in its entirety.

[0041] US Patent No. 9,870,871 to Campbell *et al.*, “Graphene macro-assembly-fullerene composite for electrical energy storage,” is incorporated herein by reference in its entirety.

[0042] US Patent Publication No. 2018/0208467 to Ye *et al.*, “Three-dimensional deterministic graphene architectures formed using three-dimensional templates,” is incorporated by reference in its entirety.

[0043] US Patent Publication No. 2012/0034442 to Worsley *et al.*, “Graphene Aerogels,” is incorporated by reference in its entirety.

[0044] US Patent Publication No. 2014/0178289 to Worsley *et al.*, “High-Density 3D Graphene-Based Monolith and Related Materials, Methods, and Devices,” is incorporated by reference in its entirety.

[0045] US Patent Application No. 14/820411 to Worsley *et al.*, “Highly Crystalline Graphene Aerogels,” is incorporated by reference in its entirety.

[0046] **Method For Making GMA-Fullerene Composite**

[0047] Many embodiments of the invention described herein relates to a method for producing a GMA-fullerene composite, comprising providing a GMA comprising a three-

dimensional network of graphene sheets crosslinked by covalent carbon bonds, and incorporating at least 20 wt.% of at least one fullerene compound into the GMA based on the initial weight of the GMA to obtain the GMA-fullerene composite.

[0048] In some embodiments, the method comprises immersing GMA in a solution of the fullerene compound. In some embodiments, the solution comprises at least one organic solvent. Suitable solvents include, for example, CS₂, CH₂Cl₂, CS₂:CH₂Cl₂, and THF.

[0049] In some embodiments, the concentration of the fullerene compound in the solution is about 1-20 mM, or about 2-15 mM, or about 3-10 mM, or about 4-8 mM. In some embodiments, the concentration of the fullerene compound in the solution is about 0.5-8 mM.

[0050] In some embodiments, the method comprises heating the GMA immersed in a solution of the fullerene compound. In some embodiments, the method comprises heating the GMA immersed in a solution of the fullerene compound at a temperature of 50-100 °C. In some embodiments, the method comprises heating the GMA immersed in a solution of the fullerene compound for 12 to 96 hours.

[0051] In some embodiments, the method comprises *in-situ* synthesis of diazonium functionalized fullerene and covalent bonding thereof to GMA.

[0052] In some embodiments, the fullerene compound is covalently bound to the graphene sheets. In some embodiments, the method comprises reacting the GMA with least one fullerene compound represented by F^{*}-(R¹)_n, wherein: F^{*} comprises a fullerene having a surface comprising six-membered and five-membered rings, R¹ comprises a reactive group (e.g., diazonium) and a conjugated linker covalently connecting the reactive group to the fullerene, and n is at least one.

[0053] In some embodiments, the method comprises reacting the GMA with least one diazonium functionalized fullerene. In some embodiments, the diazonium functionalized fullerene is represented by F^{*}-(R¹)_n, wherein: F^{*} comprises a fullerene having a surface comprising six-membered and five-membered rings, R¹ comprises a diazonium group and a conjugated linker covalently connecting the diazonium group to the fullerene, and n is at least one.

[0054] In one embodiment, F^{*} comprises C₆₀. In another embodiment, F^{*} comprises C₇₀. In a further embodiment, F^{*} comprises C₈₄.

[0055] In one embodiment, n is 1. In another embodiment, n is 2. In a further

embodiment, n is 3 or more.

[0056] In one embodiment, R^1 is linked to the fullerene by one covalent bond. In another embodiment, R^1 is linked to the fullerene by two covalent bonds.

[0057] In one embodiment, R^1 comprises one reactive group (e.g., diazonium). In another embodiment, R^1 comprises two or more reactive groups.

[0058] In one embodiment, the diazonium group is represented by . When

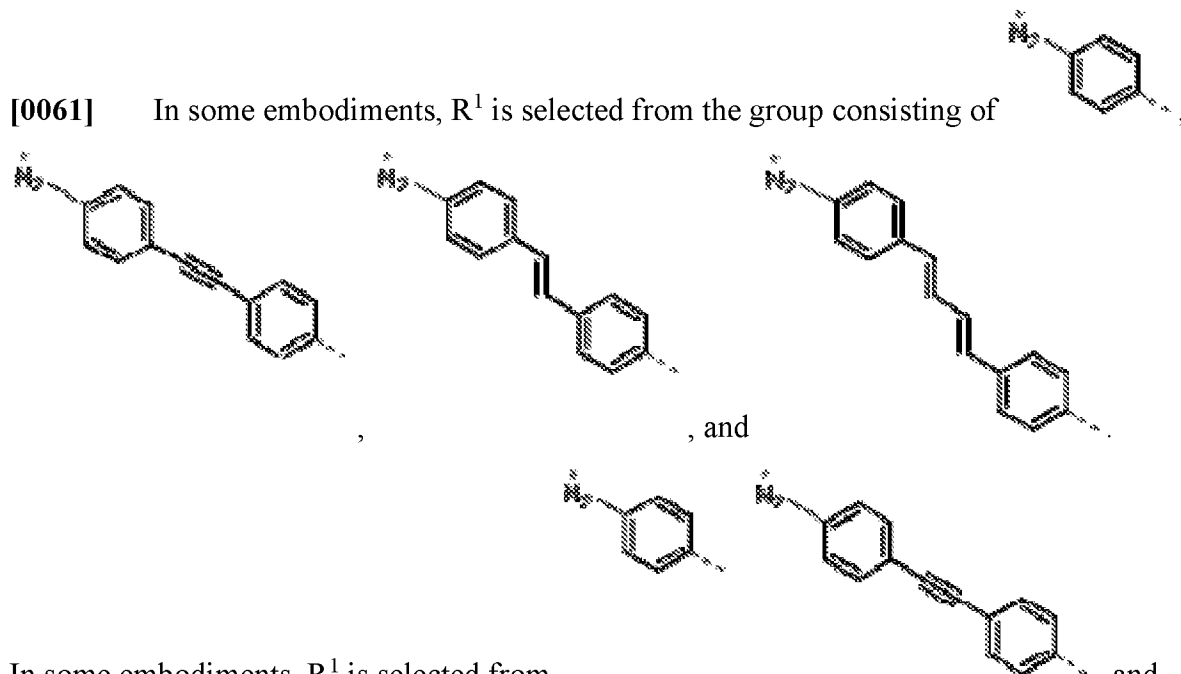
the counterion is Cl^- , the diazonium group can be represented by



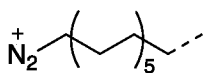
[0059] In one embodiment, the conjugated linker of R^1 comprises alternating single and multiple bonds. In another embodiment, the conjugated linker of R^1 comprises a conjugated hydrocarbon chain. In a further embodiment, the conjugated linker of R^1 comprises a conjugated hydrocarbon chain substituted with one or more heteroatoms (e.g., O, S and N).

[0060] In one embodiment, the conjugated linker of R^1 comprises at least one double bond or alkenylene bridge. In another embodiment, the conjugated linker of R^1 comprises at least one triple bond or alkynylene bridge. In a further embodiment, the conjugated linker of R^1 comprises at least one aromatic or heteroaromatic ring.

[0061] In some embodiments, R^1 is selected from the group consisting of



In some embodiments, R^1 is selected from , , and



[0062] In some embodiments, based on the initial weight of the GMA, at least 40 wt.%, or at least 60 wt.%, or at least 80 wt.%, or at least 100 wt.%, or at least 120 wt.% of the fullerene compound are incorporated into the GMA by covalently bonds.

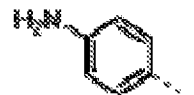
[0063] Alternatively, in some embodiments, the fullerene compound is noncovalently attached to the graphene sheets. In some embodiments, the method comprises incubating the GMA in a solution comprising at least one fullerene compound represented by $F^*-(R^2)_n$, wherein: F^* comprises a fullerene having a surface comprising six-membered and five-membered rings, R^2 comprises an aromatic or heteroaromatic group (e.g., phenylamine) and a conjugated linker covalently connecting the aromatic or heteroaromatic group to the fullerene, and n is at least one. In some embodiment, the aromatic or heteroaromatic group comprises 2, 3, 4 or more fused aromatic or heteroaromatic rings.

[0064] In some embodiments, the method comprises incubating the GMA in a solution comprising at least one phenylamine functionalized fullerene. In some embodiments, the phenylamine functionalized fullerene is represented by $F^*-(R^2)_n$, wherein: F^* comprises a fullerene having a surface comprising six-membered and five-membered rings, R^2 comprises a phenylamine group and a conjugated linker covalently connecting the phenylamine group to the fullerene, and n is at least one.

[0065] In one embodiment, n is 1. In another embodiment, n is 2. In a further embodiment, n is 3 or more.

[0066] In one embodiment, R^2 is linked to the fullerene by one covalent bond. In another embodiment, R^2 is linked to the fullerene by two covalent bonds.

[0067] In one embodiment, R^2 comprises one aromatic or heteroaromatic group (e.g., phenylamine). In another embodiment, R^2 comprises two or more aromatic or heteroaromatic groups.



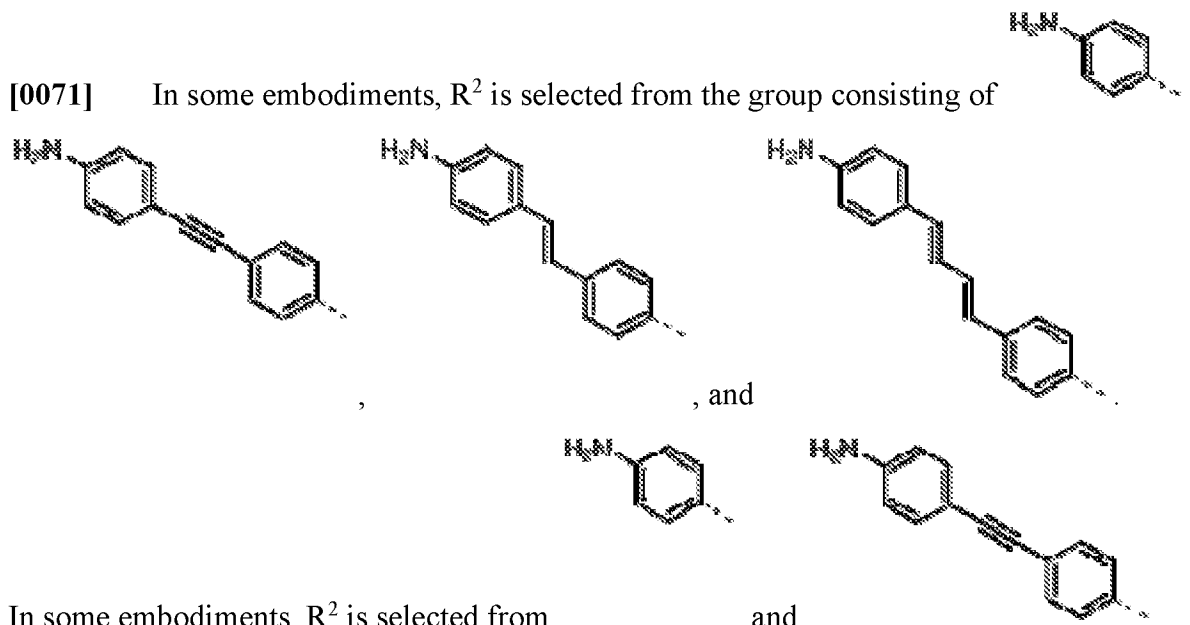
[0068] In one embodiment, the phenylamine group is represented by

[0069] In one embodiment, the conjugated linker of R^2 comprises alternating single and multiple bonds. In another embodiment, the conjugated linker of R^2 comprises a conjugated hydrocarbon chain. In a further embodiment, the conjugated linker of R^2 comprises a

conjugated hydrocarbon chain substituted with one or more heteroatoms (e.g., O, S and N).

[0070] In one embodiment, the conjugated linker of R^2 comprises at least one double bond or alkenylene bridge. In another embodiment, the conjugated linker of R^2 comprises at least one triple bond or alkynylene bridge. In a further embodiment, the conjugated linker of R^2 comprises at least one aromatic or heteroaromatic ring.

[0071] In some embodiments, R^2 is selected from the group consisting of



In some embodiments, R^2 is selected from and

[0072] In some embodiments, based on the initial weight of the GMA, at least 20 wt.%, or at least 40 wt.%, or at least 60 wt.%, or at least 80 wt.%, of the fullerene compound are incorporated into the GMA by noncovalently attachment (e.g., physisorption).

[0073] In some embodiment, the GMA is a graphene aerogel described in US Patent Publication No. 2012/0034442, which is incorporated by reference in its entirety. In some embodiment, the GMA is a high-density graphene monolith described in US Patent Publication No. 2014/0178289, which is incorporated by reference in its entirety. In some embodiment, the GMA is a highly crystalline grapehen aerogel described in US Patent Application No. 14/820411, which is incorporated by reference in its entirety.

[0074] The GMA can comprise, for example, a three-dimensional structure of graphene sheets interconnected or crosslinked by chemical bonds such as covalent carbon-carbon bonds. In some embodiments, 50% or more, or 70% or more, or 80% or more, or 90% or more of the covalent bonds interconnecting the graphene sheets are sp^2 carbon-carbon bonds. In some embodiments, 10% or less, or 5% or less, or 3% or less, or 1% or less of the graphene sheets are interconnected only by physical crosslinks. In some embodiments, 10%

or less, or 5% or less, or 3% or less, or 1% or less of the graphene sheets are interconnected only by metal crosslinks.

[0075] The GMA can be, for example, substantial free of graphene sheets with hydroxyl or epoxide functionalities. In some embodiments, 3% or less, or 1% or less, or 0.5% or less, or 0.1% or less of the carbon atoms in the GMA are connected to a hydroxyl or epoxide functionality. In some embodiments, the atomic oxygen content in the GMA is 10% or less, or 5% or less, or 3% or less, or 1% or less.

[0076] In some embodiments, the GMA consists essentially of covalently interconnected graphene sheets. In some embodiments, the GMA is not a macroporous foam. In some embodiments, the GMA is substantially free of a polymer coated on the internal surfaces of the GMA. In some embodiments, the GMA is substantially free of a metal or a metal compound coated on the internal surfaces of the GMA. In some embodiments, the GMA is substantially free of carbon nanoparticles.

[0077] GMA-Fullerene Composite

[0078] Many embodiments of the invention relate to a GMA-fullerene composite produced by the method described herein.

[0079] The weight of the fullerene component compared to the weight of the GMA component can be, for example, at least 20 wt.%, or at least 40 wt.%, or at least 60 wt.%, or at least 80 wt.%, or at least 100 wt.%, or at least 120 wt.%.

[0080] The GMA-fullerene composite can be a monolith having a thickness of, for example, at least 100 μm , or at least 1 mm, or at least 10 mm, or at least 100 mm, or about 10 μm to about 1 mm, or about 1 mm to about 100 mm.

[0081] The GMA-fullerene composite can have an electrical conductivity of, for example, at least 10 S/m, or at least 20 S/m, or at least 50 S/m, or at least 100 S/m, or at least 200 S/m, or at least 500 S/m, or about 10-1,000 S/m, or about 20-500 S/m, or about 50-200 S/m.

[0082] In some embodiments, the GMA-fullerene composite can have a BET surface area of, for example, at least 100 m^2/g , or at least 200 m^2/g , or at least 300 m^2/g , or at least 500 m^2/g , or at least 700 m^2/g , or about 100-1,500 m^2/g , or about 200-1,000 m^2/g .

[0083] In some embodiments, the GMA-fullerene composite can have a Young's modulus of, for example, at least 10 MPa, or at least 20 MPa, or at least 50 MPa, or at least 100

MPa, or at least 200 MPa, or at least 500 MPa, or about 10-1,000 MPa, or about 20-500 MPa.

[0084] In some embodiments, the GMA-fullerene composite can have a mesopore volume of, for example, at least 0.2 cm³/g, or at least 0.5 cm³/g, or at least 0.8 cm³/g, or at least 1 cm³/g, or at least 1.2 cm³/g, or at least 1.5 cm³/g, or about 0.2-5 cm³/g, or about 0.5-3 cm³/g.

[0085] In some embodiments, the GMA-fullerene composite comprises at least one fullerene compound covalently connected to at least one graphene sheet via at least one conjugated linker. In one embodiment, the conjugated linker comprises alternating single and multiple bonds. In another embodiment, the conjugated linker comprises a conjugated C₁-C₃₀, C₁-C₂₀, C₁-C₁₅, or C₁-C₁₀ hydrocarbon chain. In a further embodiment, the conjugated linker comprises a conjugated C₁-C₃₀, C₁-C₂₀, C₁-C₁₅, or C₁-C₁₀ hydrocarbon chain substituted with one or more heteroatoms (e.g., O, S and N). In one embodiment, the hydrocarbon chain comprises at least one double bond or alkenylene bridge. In another embodiment, the hydrocarbon chain comprises at least one triple bond or alkynylene bridge. In a further embodiment, the hydrocarbon chain comprises at least one aromatic or heteroaromatic ring.

[0086] Electrode and Supercapacitor

[0087] The GMA-fullerene composite described herein can be used in a variety of applications, including supercapacitors, battery electrodes, electrical energy storage, micro-batteries, hybrid capacitors, next-generation batteries, hybrid vehicles, and alternative energy storage.

[0088] The GMA-fullerene composite described herein is functionally advantageous in energy storage applications. In particular, C₆₀ can store up to 6 electrons, or 1 e⁻ / 10 carbon atoms (Echegoyen *et al.*, *Acc. Chem. Res.*, 1998, 31, 593–601), compared with 1 e⁻ / 100 carbon atoms for graphitic carbon materials (Berger *et al.*, *J. Phys. Chem. B*, 2004, 108(52):19912–19916; Sarma *et al.*, *Rev. Mod. Phys.*, 2011, 83(2):407–470; Wood *et al.*, *J. Phys. Chem. C*, 2014, 118(1):4–15), a 10-fold increase in electrical storage capacity. In principle a graphene aerogel functionalized with 50 wt% loading of fullerene could achieve 4 X greater energy storage compared with an unfunctionalized graphene aerogel. The measured capacity of an unfunctionalized graphene aerogel electrode is ~60 coulombs per gram (C/g) (Campbell *et al.*, *J. Mater. Chem. A*, 2014, 2:17764–17770). With fullerene functionalization (e.g., ~50 wt% loading), an electron storage capacity of at least about 100 C/g, or at least

about 150 C/g, or at least about 200 C/g, or at least about 250 C/g, or at least about 300 C/g, or at least about 350 C/g, or at least about 400 C/g, or about 100-500 C/g, or about 200-400 C/g, can be achieved. With higher fullerene loading (e.g., through the use of different length linkers), an electron storage capacity of up to about 600 C/g, or up to about 550 C/g, or up to about 500 C/g, or about 200-600 C/g, or about 300-500 C/g, or about 400-600 C/g, can be achieved. The theoretical maximum electron storage capacity for C₆₀ is 803 C/g – the equivalent of 222.5 mAh/g, which is approaching the capacity of lithium ion battery cathode materials (e.g., LiCoO₂ theoretical max is 274 mAh/g).

[0089] Accordingly, many embodiments of the invention described herein also relate to an electrode comprising the GMA-fullerene composite, as well as a supercapacitor comprising the electrode.

[0090] In one embodiment, the supercapacitor further comprises an organic liquid electrolyte in contact with the electrode. In another embodiment, the supercapacitor further comprises an ionic liquid electrolyte with the electrode. In an additional embodiment, the supercapacitor further comprises an aqueous electrolyte with the electrode.

[0091] Unlike anthraquinone and other redox-based charge-storage strategies that involve chemical reactions (proton-coupled electron transfer), fullerenes store charge in a purely electric double-layer capacitance (EDLC) mechanism that is an interfacial phenomenon and does not involve a chemical reaction, which means that charge/discharge rates can be faster and long-term stability will increase. Moreover, because the purely EDLC mechanism does not require protons, the GMA-fullerene composite described herein can be used in supercapacitors in combination with organic or ionic liquid electrolytes, which greatly increases the operational voltage window and thus the total energy stored ($E=1/2 CV^2$).

[0092] WORKING EXAMPLES

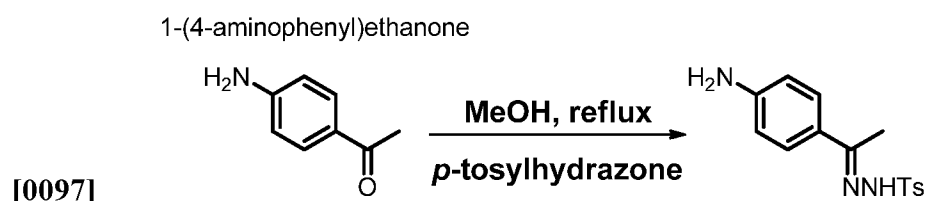
[0093] EXAMPLE 1 – Synthesis of GMA

[0094] Graphene macro-assemblies (GMA) were prepared in a similar manner to what were previously reported in Worsley *et al.*, *Chem. Commun.*, 48:8428-8430 (2012), which is incorporated herein by reference. Graphene oxide (GO, 1-2 layer, 300-800 nm diameter sheets) was purchased from Cheaptubes and used as received. GO was dispersed in Milli-Q H₂O (20 mg/mL) by ultrasonication for 24 h, and ammonium hydroxide catalyst (211 µL/g) was added to the resulting suspension. The GO suspension/catalyst mixture was cast into disk shaped molds, sealed, and placed in a 75 °C oven for 72 h for crosslinking/gelation. The

monolithic disks were washed in water, followed by acetone, and dried with supercritical CO₂. The disks were then carbonized at 1050 °C for 3 h under flowing N₂ to remove oxygen functionality (final O content <2 at.%). The resulting GMA disks are approximately 1 cm in diameter by 250 μm thickness, weigh ~1 mg, have density of ~0.07 g/cm³, and have a BET surface area of ~1300 m²/g.

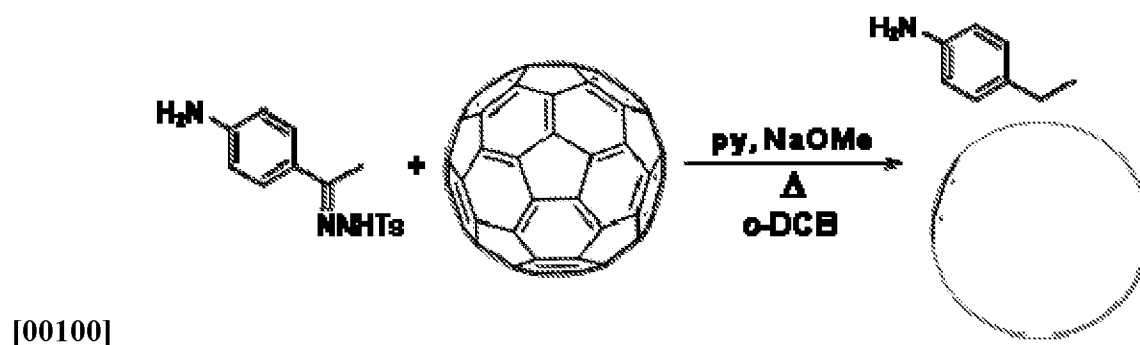
[0095] EXAMPLE 2 – Synthesis of 1-(4-aminophenyl)ethano-p-toluenesulfonyl hydrazone

[0096] A mixture of 1-(4-aminophenyl)ethano (1.0 g, 3.36 mmol) and p-toluenesulfonyl hydrazone (2.2 g, 11.88 mmol) in MeOH (20 mL) was stirred and refluxed for 2 days. The mixture was left without heating for 1 day and then cooled to -10 °C. The white powder was then filtered and washed with cold MeOH and dichloromethane, and dried under vacuum (65% yield).



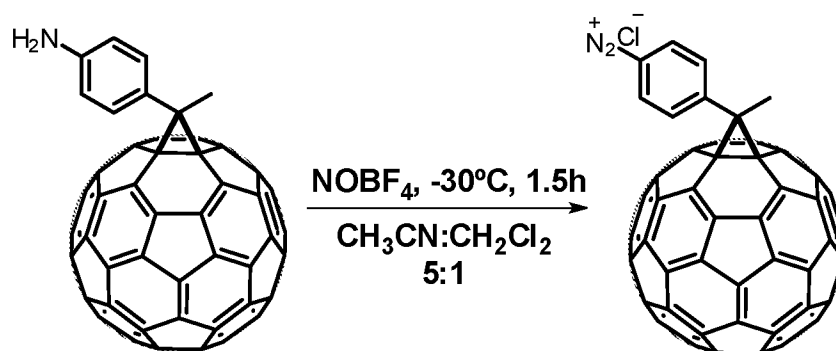
[0098] EXAMPLE 3 – Synthesis of 1-(4-aminophenyl)ethano-C₆₀ (PA-C₆₀)

[0099] Diazo addend was prepared in-situ by dissolving 1-(4-aminophenyl)ethano-p-toluenesulfonyl hydrazone (122.7 mg, 0.208 mmol) in 1.2 mL of anhydrous pyridine under N₂ atmosphere. NaOMe (56.3 mg, 1.042 mmol) was added, and the mixture was stirred for 30 min. A solution of 75.0 mg of C₆₀ (0.1042 mmol) in 7 mL of *o*-DCB (aka, 1,2-dichlorobenzene) was added and stirred at 110 °C for 6 h. The solvent from the reaction mixture was removed under nitrogen and the crude product was purified by silica gel column using initially CS₂ as the eluent to collect the unreacted [60]fullerene, followed by CS₂:CH₂Cl₂ 1:1 to collect the monoadduct (41%).



[00101] EXAMPLE 4 – Synthesis of 1-(4-diazoniumphenyl)ethano-C₆₀ tetrafluoroborate

[00102] Diazonium salt was prepared by dissolving 1-(4-aminophenyl)ethano-C₆₀ in a 5:1 mixture of CH₃CN:CH₂Cl₂ under N₂ atmosphere, the solution was cold down to -30 °C and nitrosonium tetrafluoroborate (NOBF₄) was added. The mixture was stirred for 1.5 h. The solvent from the reaction mixture was removed under reduced pressure and the crude product was purified by washing the crude mixture with CS₂:CH₂Cl₂ 7:3 to remove the unreacted monoadduct and THF to remove the excess of diazonium salt.



[00103]

[00104] EXAMPLE 5 – Noncovalent Functionalize of GMA with C₆₀

[00105] GMA was non-covalently functionalized with C₆₀ via physisorption. The following loading percentages based on the initial weight of the GMA were achieved.

[00106] 4.5 mM C₆₀ in CS₂, 3 days = 16% loading

[00107] 4 mM C₆₀ in CS₂, 15 h = 13% loading

[00108] 2 mM C₆₀ in CS₂, 15 h = 5% loading

[00109] FIG. 1(A) shows a GMA-C₆₀ composite obtained by noncovalently functionalizing the GMA with 4.5 mM C₆₀, and FIG. 1(B) shows CV of a GMA-C₆₀ composite with 13% loading of C₆₀ in MeCN/TBAP electrolyte (5 mV/s).

[00110] EXAMPLE 6 – Noncovalent Functionalize of GMA with PA-C₆₀

[00111] GMA was non-covalently functionalized with PA-C₆₀ via physisorption. PA-C₆₀ is more soluble than unmodified C₆₀. The following loading percentages based on the initial weight of the GMA were achieved.

[00112] 7 mM PA-C₆₀ in CS₂:CH₂Cl₂ 7:3, 3 days = 87% loading

[00113] 6 mM PA-C₆₀ in CS₂:CH₂Cl₂ 7:3, 17 h = 50% loading

[00114] 4 mM PA-C₆₀ in CS₂:CH₂Cl₂ 7:3, 14 h = 26% loading

[00115] 2.1 mM PA-C₆₀ in CS₂:CH₂Cl₂ 7:3, 14 h = 16% loading

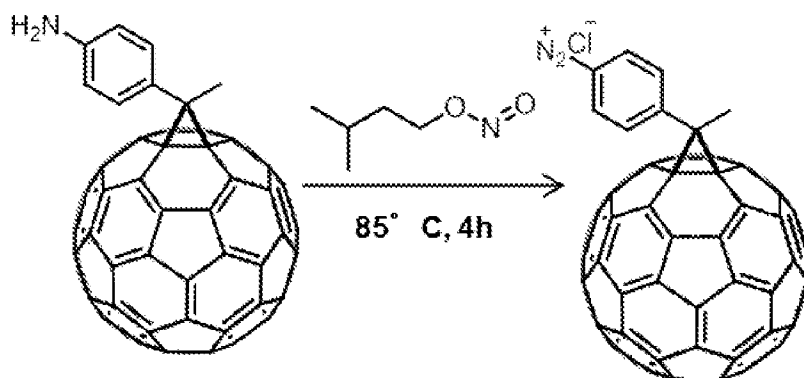
[00116] FIG 2(A) shows a GMA-PA-C₆₀ composite obtained by noncovalently functionalizing the GMA with 7 mM PA-C₆₀, and FIG 2(B) shows CV of a GMA-PA-C₆₀ composite with 50% loading of PA-C₆₀ in MeCN/TBAP electrolyte (5 mV/s).

[00117] EXAMPLE 7 – Covalent Functionalization of GMA with Diazonium-C₆₀

[00118] GMA was added to a 1 mM solution of diazonium-C₆₀ in THF, and heat at 60 °C for 2 days. A final loading of 67% based on the initial weight of the GMA was achieved.

[00119] EXAMPLE 8 – *In-situ* Synthesis of Diazonium-C₆₀ and Covalent Functionalization of GMA

[00120] To a 4 mM solution of PA-C₆₀ in *o*-DCB:MeCN 4:1, 2 drops of isopentyl nitrite were added and a precipitate was observed immediately. GMA was then added and heated at 85°C over the weekend. After the weekend the precipitate and solution were removed and GMA was washed several times with CS₂, CH₂Cl₂ and THF to remove not chemically bonded fullerene. GMA was then left in THF overnight and dried in the oven at 80° C for 2 days. A final loading of 138% based on the initial weight of the GMA was achieved.



[00121]

[00122] FIG. 4 shows CV of a GMA-fullerene composite with 138% loading of diazonium-C₆₀ in MeCN/TBAP electrolyte (10 mV/s), which is evidently functionally superior compared to CV of a non-functionalized GMA in MeCN/TBAP electrolyte (5 mV/s) as shown in FIG. 3.

[00123] EXAMPLE 9 – Noncovalent Functionalize of GMA with C₆₀

[00124] GMA was non-covalently functionalized with C₆₀ via physisorption. The following loading percentages based on the initial weight of the GMA were achieved.

[00125] 2-6 mM C₆₀ in CS₂, 15 h = 15% ± 4 loading

[00126] FIG. 7 shows CV in MeCN/TBAP electrolyte of a GMA-C₆₀ with 17 wt.% loading of C₆₀ (5 mV/s, cycle 1, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line).

[00127] EXAMPLE 10 – Noncovalent Functionalize of GMA with C₆₀-1

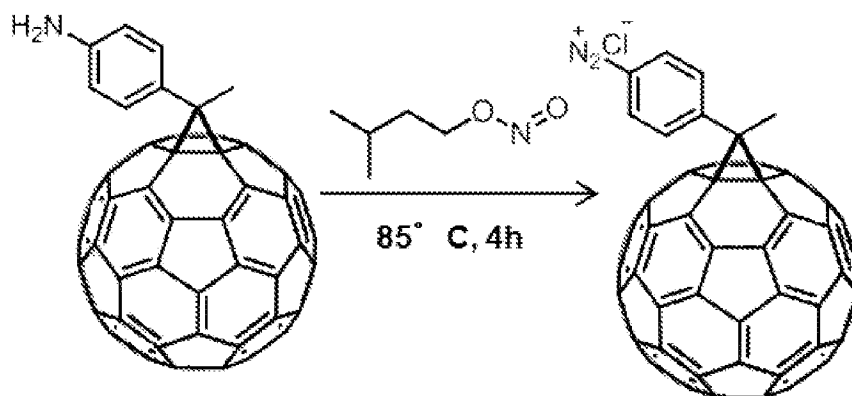
[00128] GMA was non-covalently functionalized with C₆₀-1 via physisorption. C₆₀-1 is more soluble than unmodified C₆₀. The following loading percentages based on the initial weight of the GMA were achieved.

[00129] 2-6 mM C₆₀-1 in CS₂, 15 h = 43 ± 3% loading

[00130] FIG. 8 shows CV in MeCN/TBAP electrolyte of a GMA-C₆₀-1 composite with 44 wt.% loading of C₆₀-1 (5 mV/s, cycle 1, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line).

[00131] EXAMPLE 11 – *In-situ* Synthesis of Diazonium-C₆₀ and Covalent Functionalization of GMA

[00132] To a 4 mM solution of C₆₀-1 in *o*-DCB:MeCN 4:1, 2 drops of isopentyl nitrite were added and a precipitate was observed immediately. GMA was then added and heated at 85°C over the weekend. After the weekend the precipitate and solution were removed and GMA was washed several times with CS₂, CH₂Cl₂ and THF to remove not chemically bonded fullerene. GMA was then left in THF overnight and dried in the oven at 80° C for 2 days. A final loading of 138% based on the initial weight of the GMA was achieved.



[00133]

[00134] FIG. 9 shows CV in MeCN/TBAP electrolyte of a covalently functionalized GMA-fullerene composite with diazonium-C₆₀, final loading 138 wt.% (5 mV/s, black line) and non-functionalized GMA (5 mV/s, cycle 1, red line). FIG. 10 shows examples of diazonium-functionalized.

[00135] Methods for Forming Covalently Attached Fullerene-GMA Materials Using Crosslinking of Hydroxylated-Fullerene and Graphene Oxide Starting Materials

[00136] The GMA-fullerene composites described herein can be formed by incorporating a hydroxylated fullerene into the sol-gel synthesis of GMA. This can increase the likelihood of the fullerene bonding with the graphene oxide to achieve the composites described herein. For example, the need for alignment between the fullerene and the graphene oxide in order to bond the fullerene and the graphene oxide can be reduced, which can enable more favorable reaction kinetics. Factors such as more effective bonding of the fullerene with the graphene (e.g., graphene sheets) during the sol-gel synthesis and a greater ratio of fullerene to graphene (e.g., higher concentrations of fullerenes) can enable improvements in capacitance and other performance factors of the GMA-fullerene composite, including for using the GMA-fullerene composite to form a supercapacitor (e.g., form an electrode of a supercapacitor).

[00137] In some embodiments, a water soluble hydroxylated fullerene is added to a water suspension of graphene oxide (e.g., before gelation is performed to produce the GMA). The hydroxylated fullerene can covalently bond with the graphene oxide (e.g., with graphene sheets) during a gelation process to produce a GO-fullerene aerogel. The GO-fullerene aerogel can be dried and carbonized to form the GMA-fullerene composite. This can incorporate at least twenty five weight percent of the fullerene into the GMA.

[00138] The hydroxylated fullerene can be a C_{60} hydroxylated fullerene, such as $C_{60}(OH)_{42}$. The hydroxylated fullerene can be a C_{70} or C_{100} hydroxylated fullerene. In some embodiments, increasing the number of OH groups (e.g., more than 42 OH groups for C_{60} ; fewer than 42 may also be used but may limit solubility) of the hydroxylated fullerene can increase the water solubility of the hydroxylated fullerene, enabling greater numbers of fullerenes to be bonded to the graphene. The number of carbons in the hydroxylated fullerene can affect the shape of the fullerene and the solubility of the fullerene in water. For example, C_{100} hydroxylated fullerene can have a lower solubility than C_{60} hydroxylated fullerene. The method for forming the GMA-fullerene composite can be performed at relatively higher temperatures as the solubility of the hydroxylated fullerene increases.

[00139] In some embodiments, a water suspension of graphene oxide is provided. A hydroxylated fullerene is added to the water suspension of graphene oxide. The water suspension can be a mixture of at least about 7 mg of GO in 1 mL of water. The water

suspension can be a mixture of at least about 20 mg of GO in 1 mL of water. The hydroxylated fullerene can be water soluble. The hydroxylated fullerene can include a C₆₀, C₇₀, or C₁₀₀ hydroxylated fullerene. The hydroxylated fullerene and water suspension of graphene oxide can be sonicated. A gelation process can be performed, during which the hydroxylated fullerene can covalently bond with sheets of graphene. In some embodiments, at least twenty five weight percent of fullerene is incorporated into the GMA that is formed in the sol-gel process. Catalysts may be used during various steps of the process for producing the GMA-fullerene composite.

[00140] EXAMPLE 12 – Synthesis of GMA-fullerene composite

[00141] 5mg of water soluble hydroxylated fullerene C₆₀(OH)₄₂ was sonicated overnight with a mixture of 20mg of GO in 1mL of water. To 1g of the suspension 211 μL of ammonium hydroxide solution was added and the mixture was gelled at 80° C for 72 hours. A washing procedure was performed analogous to that for GO aerogels, by washing in water followed by washing in acetone. Some unreacted C₆₀(OH)₄₂ may dissolve and wash away (e.g., in the acetone wash). GO-C₆₀(OH)₄₂ aerogels were supercritically dried and carbonized at 700° C.

[00142] FIG. 11 shows the results of thermal gravimetric analysis (TGA) of GO, C₆₀(OH)₄₂, and GO-C₆₀(OH)₄₂. The TGA indicated that not all of the C₆₀(OH)₄₂ in the GO-C₆₀(OH)₄₂ composite decomposed during carbonization, demonstrating that the fullerene can remain available for electron storage capacity at the 700° C temperature at which carbonization is performed.

[00143] FIG. 12 shows CV of GMA (black line) and C₆₀(OH)₄₂ carbonized at 700° C (red line, cycle 1; blue line, cycle 15). The CV was performed in acetonitrile with 0.1 M TBAP electrolyte.

[00144] FIG. 13A shows scanning electron microscopy (SEM) of uncarbonized GMA. FIGS. 13B and 13C show SEM of carbonized GMA indicating clustering of fullerenes after carbonization (which may indicate changes in the structure and capacitance of the GMA-fullerene composite not identified using CV).

[00145] As used herein, the singular terms “a,” “an,” and “the” include plural referents unless the context clearly dictates otherwise. Thus, for example, reference to a compound can include multiple compounds unless the context clearly dictates otherwise.

[00146] As used herein, the terms “substantially,” “substantial,” and “about” are used to

describe and account for small variations. When used in conjunction with an event or circumstance, the terms can refer to instances in which the event or circumstance occurs precisely as well as instances in which the event or circumstance occurs to a close approximation. For example, the terms can refer to less than or equal to $\pm 10\%$, such as less than or equal to $\pm 5\%$, less than or equal to $\pm 4\%$, less than or equal to $\pm 3\%$, less than or equal to $\pm 2\%$, less than or equal to $\pm 1\%$, less than or equal to $\pm 0.5\%$, less than or equal to $\pm 0.1\%$, or less than or equal to $\pm 0.05\%$.

[00147] Additionally, amounts, ratios, and other numerical values are sometimes presented herein in a range format. It is to be understood that such range format is used for convenience and brevity and should be understood flexibly to include numerical values explicitly specified as limits of a range, but also to include all individual numerical values or sub-ranges encompassed within that range as if each numerical value and sub-range is explicitly specified. For example, a ratio in the range of about 1 to about 200 should be understood to include the explicitly recited limits of about 1 and about 200, but also to include individual ratios such as about 2, about 3, and about 4, and sub-ranges such as about 10 to about 50, about 20 to about 100, and so forth.

[00148] In the foregoing description, it will be readily apparent to one skilled in the art that varying substitutions and modifications may be made to the invention disclosed herein without departing from the scope and spirit of the invention. The invention illustratively described herein suitably may be practiced in the absence of any element or elements, limitation or limitations, which is not specifically disclosed herein. The terms and expressions which have been employed are used as terms of description and not of limitation, and there is no intention that in the use of such terms and expressions of excluding any equivalents of the features shown and described or portions thereof, but it is recognized that various modifications are possible within the scope of the invention. Thus, it should be understood that although the present invention has been illustrated by specific embodiments and optional features, modification and/or variation of the concepts herein disclosed may be resorted to by those skilled in the art, and that such modifications and variations are considered to be within the scopes of this invention.

WHAT IS CLAIMED IS:

1. A method for producing a graphene macro-assembly (GMA)-fullerene composite, comprising:
 - providing graphene oxide;
 - adding a hydroxylated fullerene compound to the graphene oxide; and
 - performing a gelation process using the graphene oxide and the hydroxylated fullerene.
2. The method of claim 1, wherein providing the graphene oxide comprises providing a water suspension of graphene oxide.
3. The method of claim 2, wherein the water suspension of graphene oxide comprises a mixture of at least about 7 mg of graphene oxide in 1 mL of water.
4. The method of any of the preceding claims, wherein the hydroxylated fullerene compound comprises a C₆₀ hydroxylated fullerene, a C₇₀ hydroxylated fullerene, or a C₁₀₀ hydroxylated fullerene.
5. The method of any of the preceding claims, wherein adding the hydroxylated fullerene compound comprises adding about 5mg of water soluble C₆₀(OH)₄₂ to the graphene oxide, the graphene oxide comprising a mixture of about 20 mg of graphene oxide in 1 mL of water.
6. The method of any of the preceding claims, further comprising sonicating the hydroxylated fullerene and the graphene oxide.
7. The method of any of the preceding claims, further comprising adding, to about 1 g of a suspension formed from the hydroxylated fullerene compound and the graphene oxide, about 211 μ L of ammonium hydroxide solution to form a mixture, wherein performing the gelation process comprises performing the gelation process using the mixture.
8. The method of any of the preceding claims, wherein performing the gelation process comprises performing the gelation process at about 80 °C for about 72 hours.

9. The method of any of the preceding claims, wherein performing the gelation process provides a graphene oxide-fullerene compound aerogel, the method further comprising drying and carbonizing the graphene oxide-fullerene compound aerogel.

10. The method of claim 9, wherein carbonizing the graphene oxide-fullerene compound aerogel comprising carbonizing the graphene oxide-fullerene compound aerogel at about 700 °C.

11. The method of claim 9, further comprising performing at least one acetone wash of the graphene oxide-fullerene compound aerogel prior to drying the graphene oxide-fullerene compound aerogel.

12. The method of any of the preceding claims, wherein the GMA-fullerene composite comprises at least 25 weight percent of fullerene.

13. The method of any of the preceding claims, wherein performing the gelation process comprises covalently bonding the hydroxylated fullerene compound with graphene sheets formed from the graphene oxide.

14. A composition produced using the method of any of claims 1 through 13.

15. A supercapacitor comprising an electrode comprising the GMA-fullerene composite produced using the method of any of claims 1 through 13.

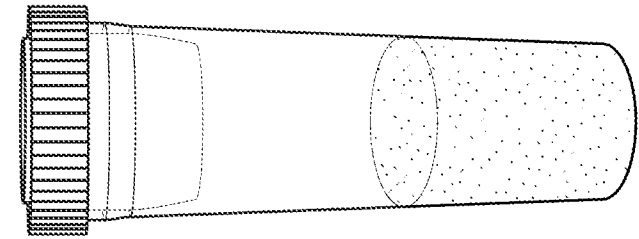


FIG. 1A

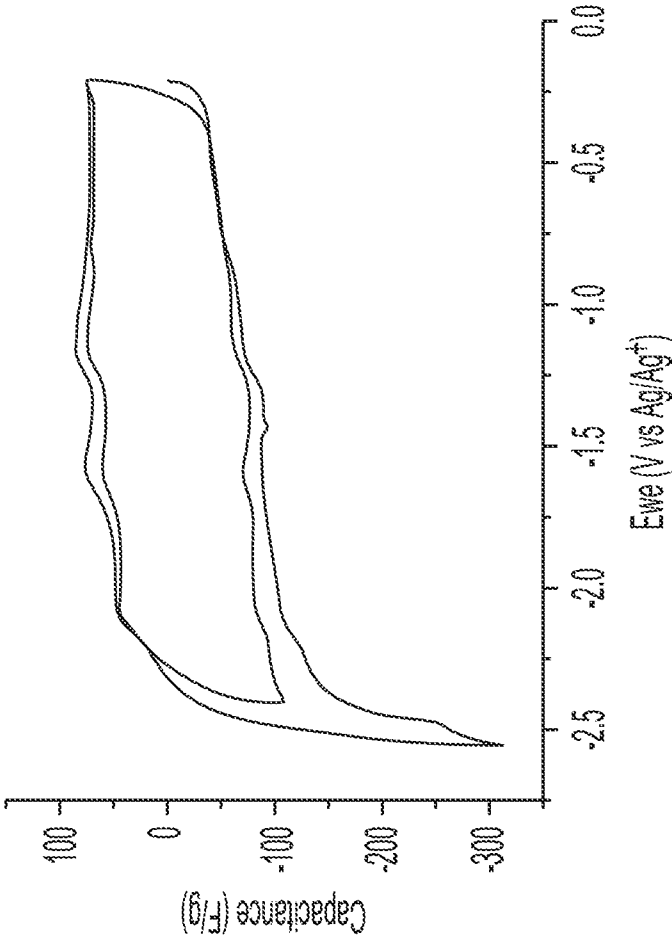


FIG. 1B

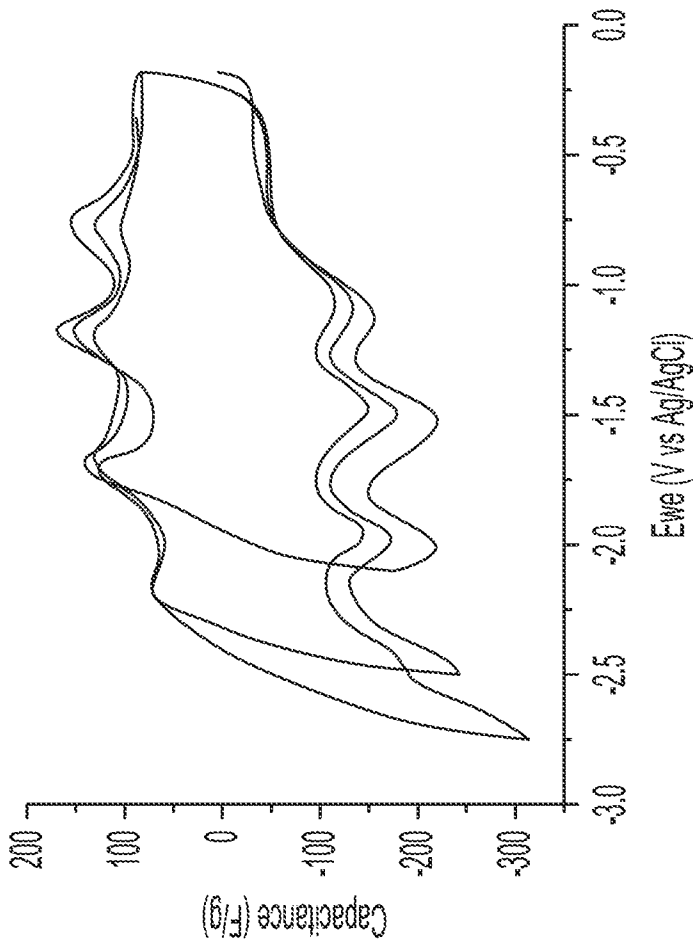


FIG. 2B

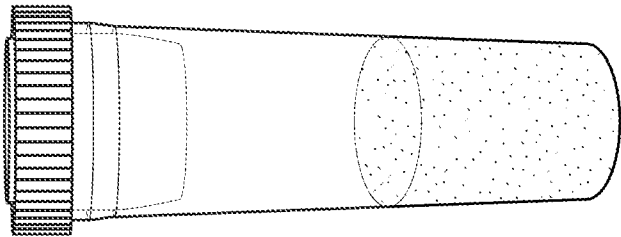


FIG. 2A

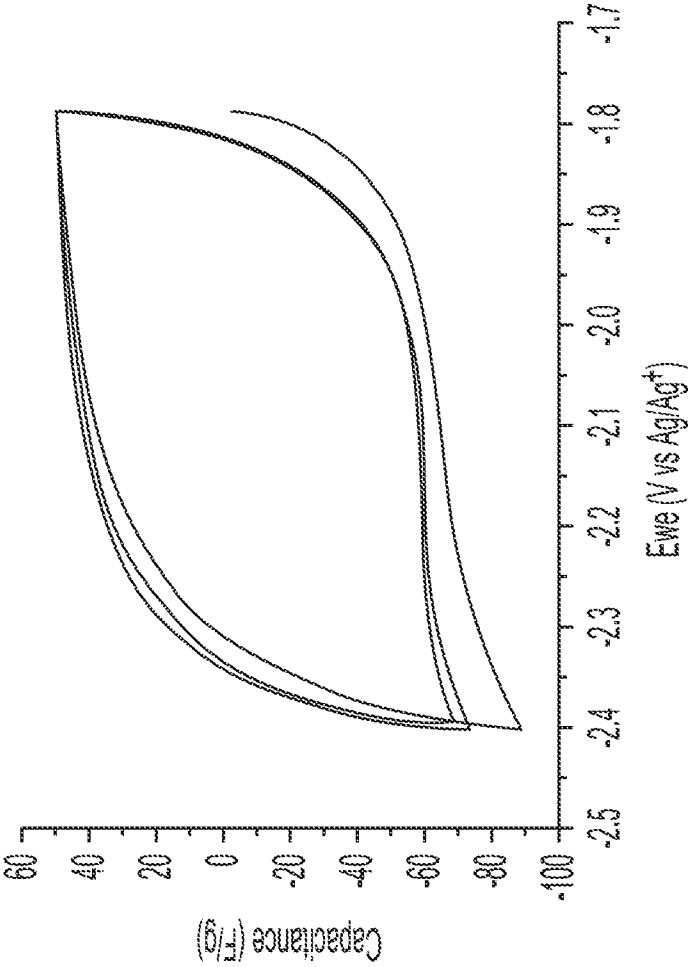


FIG. 3

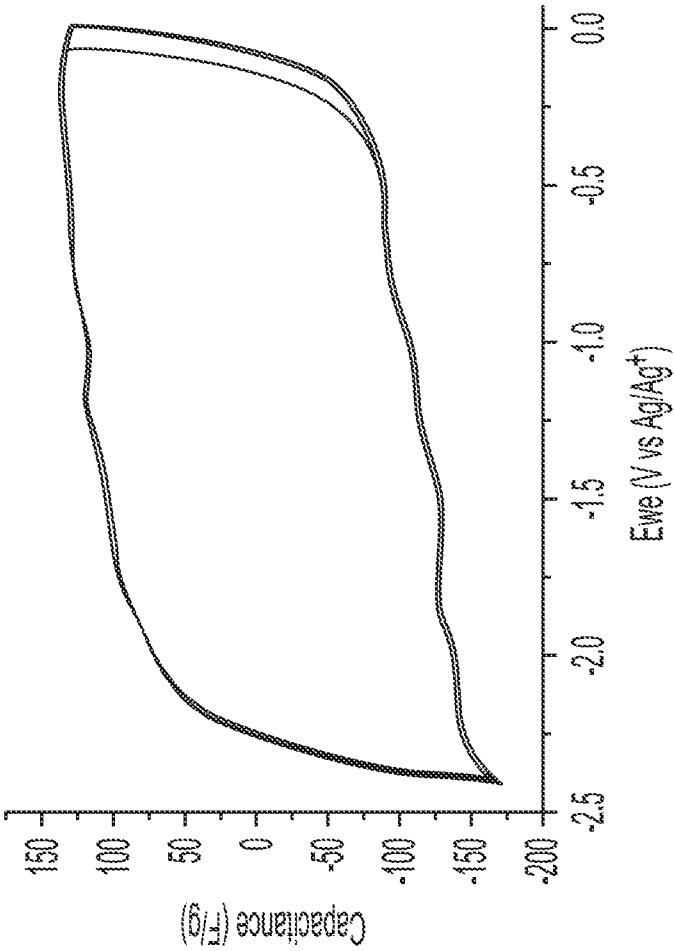


FIG. 4

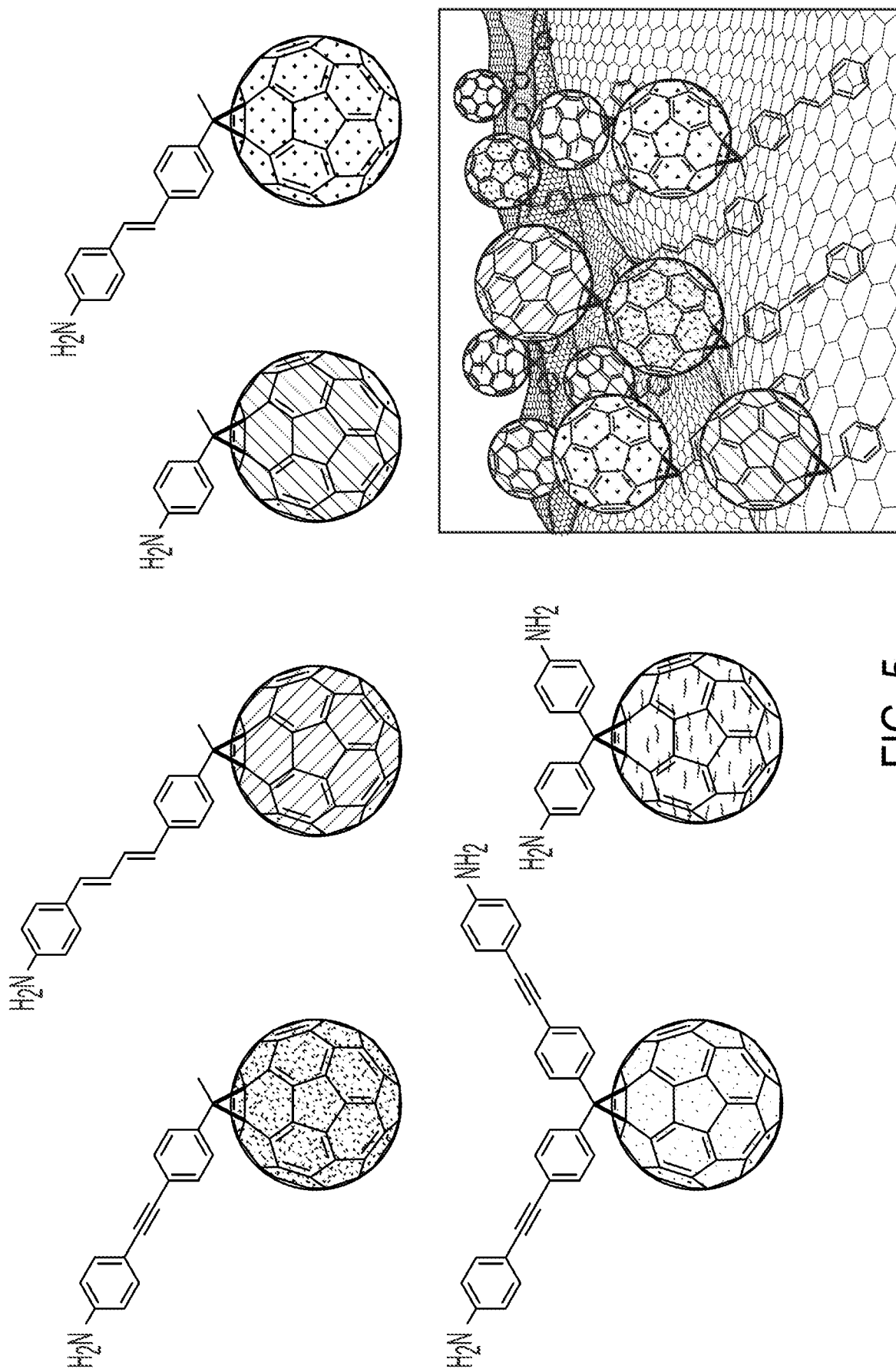


FIG. 5

6/13

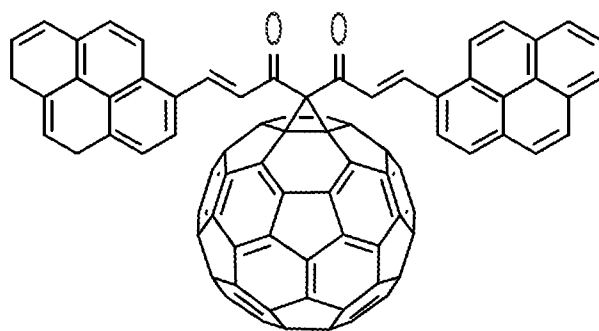
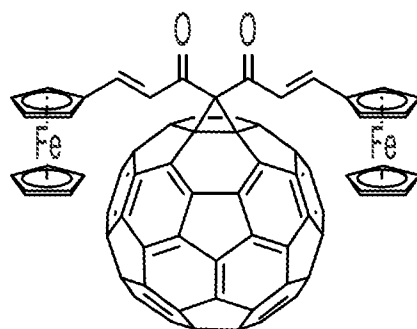
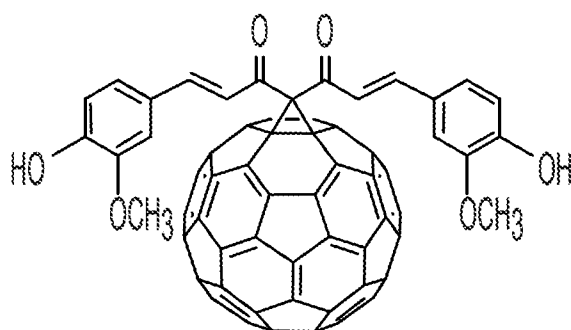
Curcumin-Pyrene-C₆₀Curcumin-Ferrocene-C₆₀Curcumin-C₆₀

FIG. 6

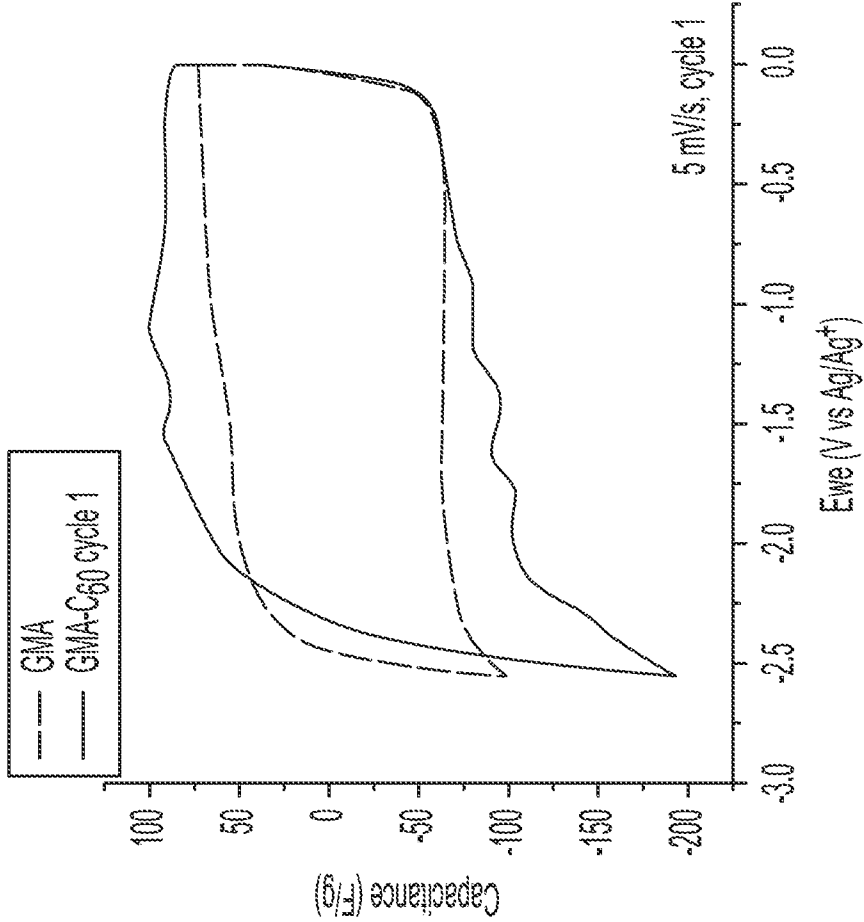


FIG. 7

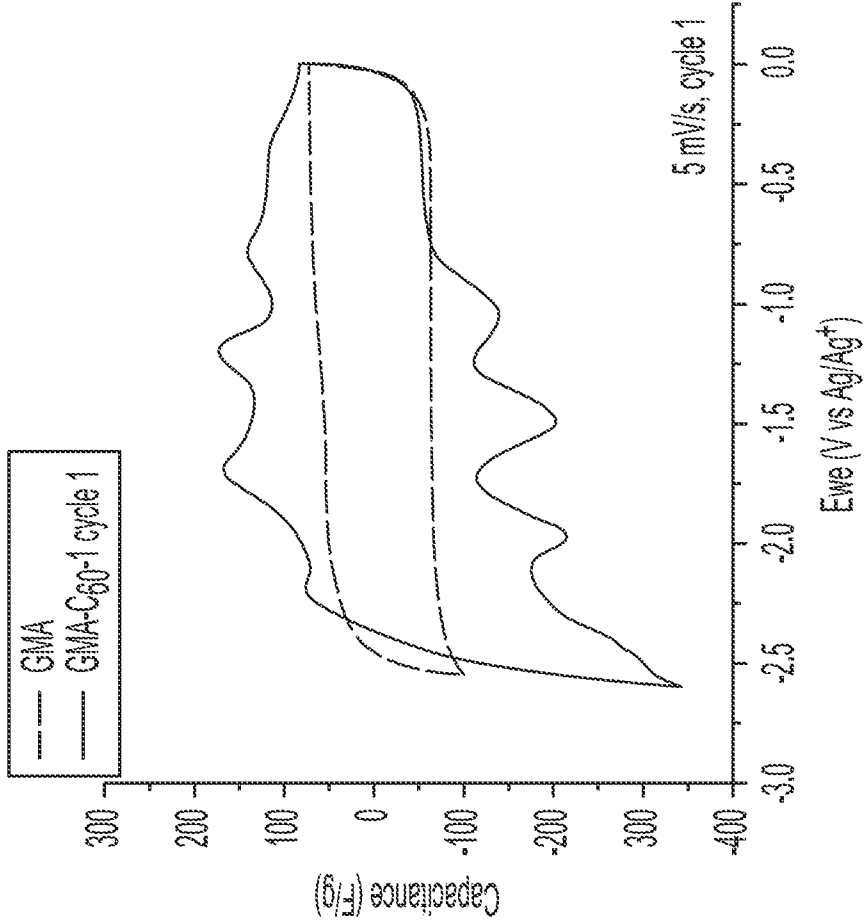


FIG. 8

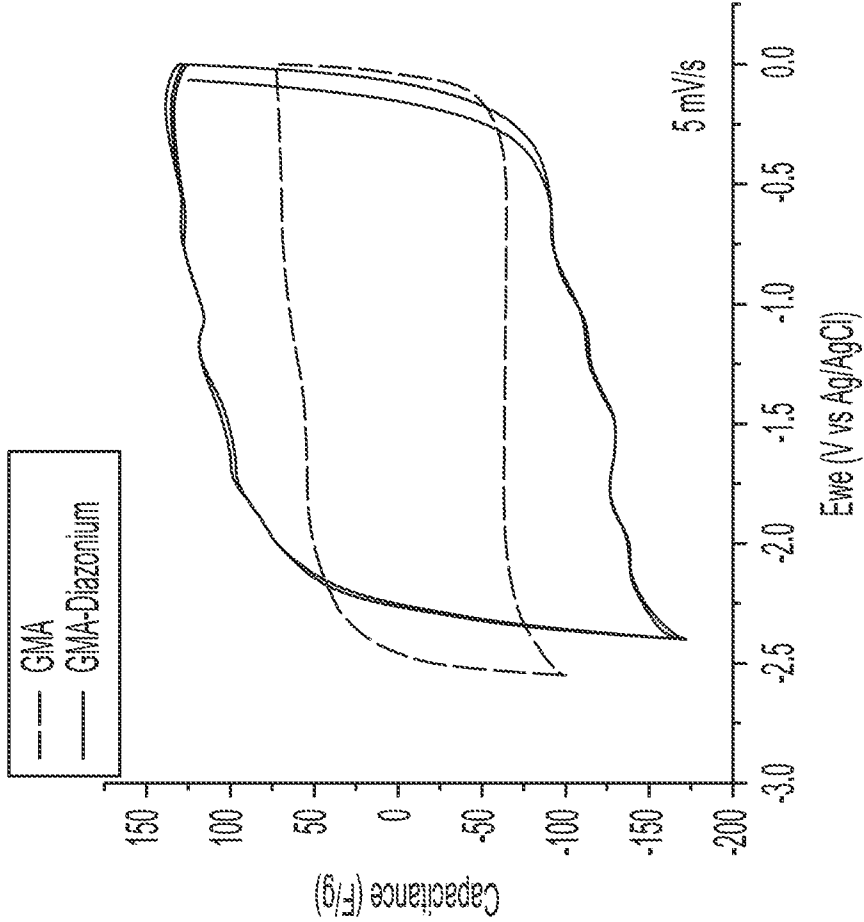


FIG. 9

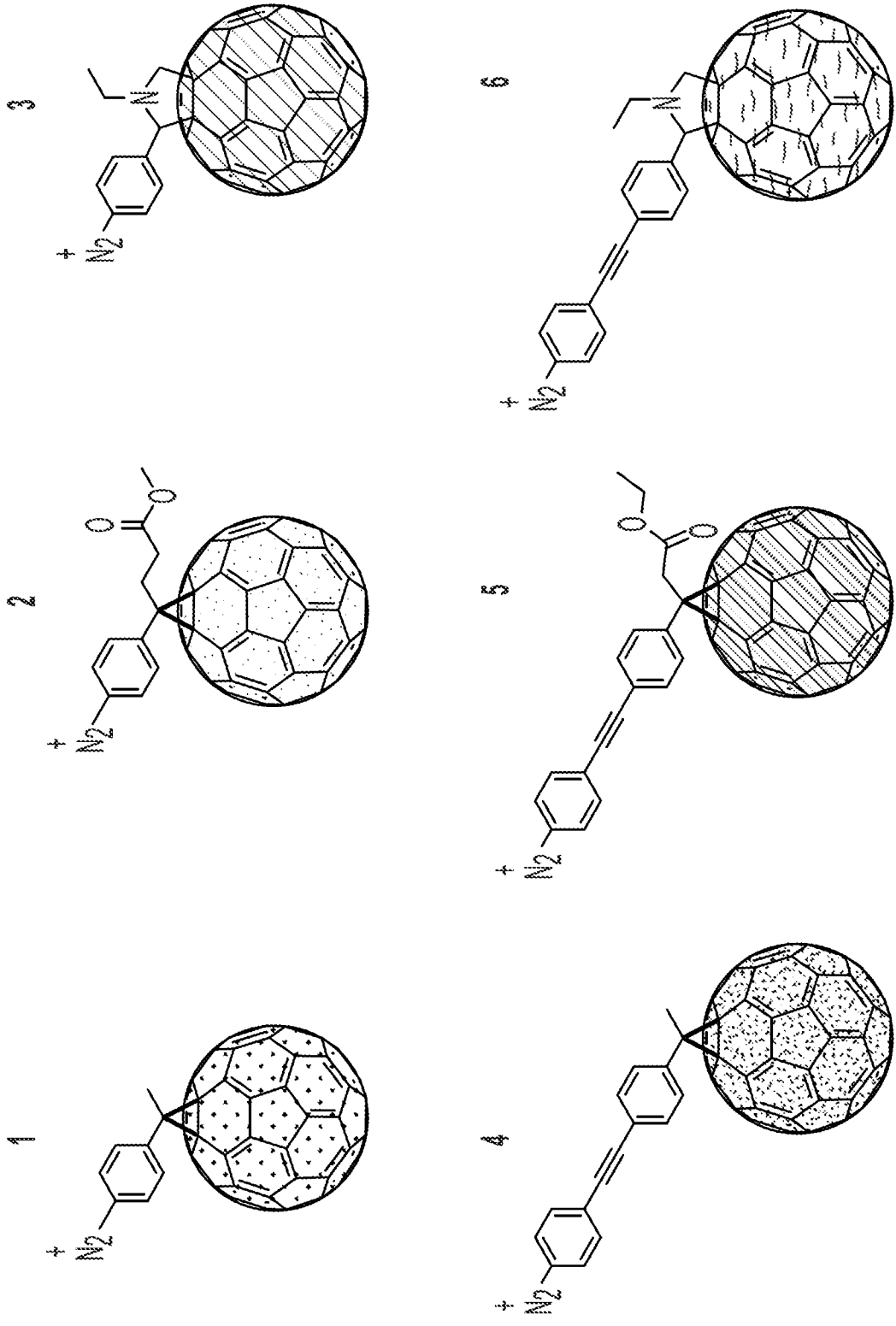


FIG. 10

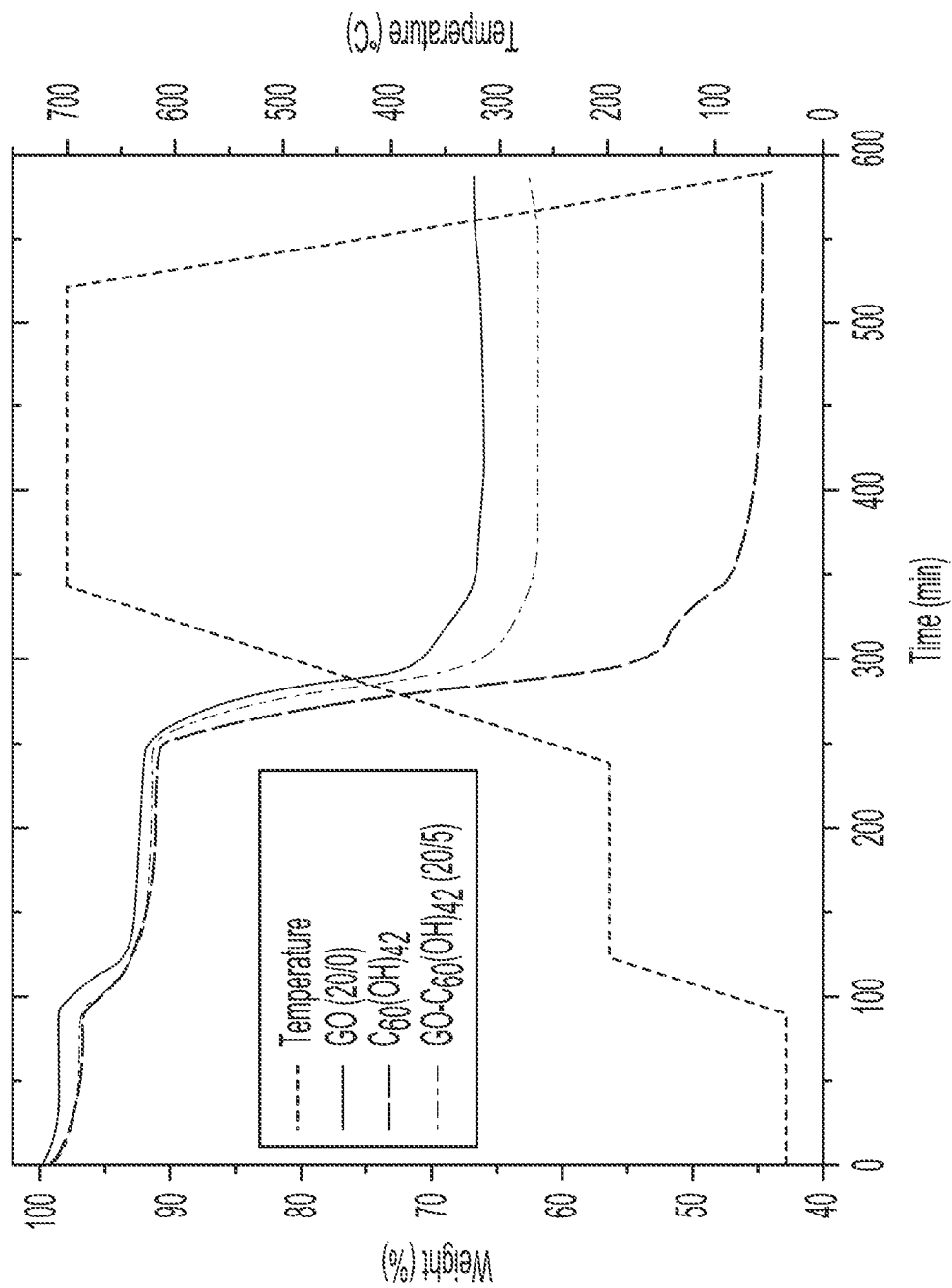


FIG. 11

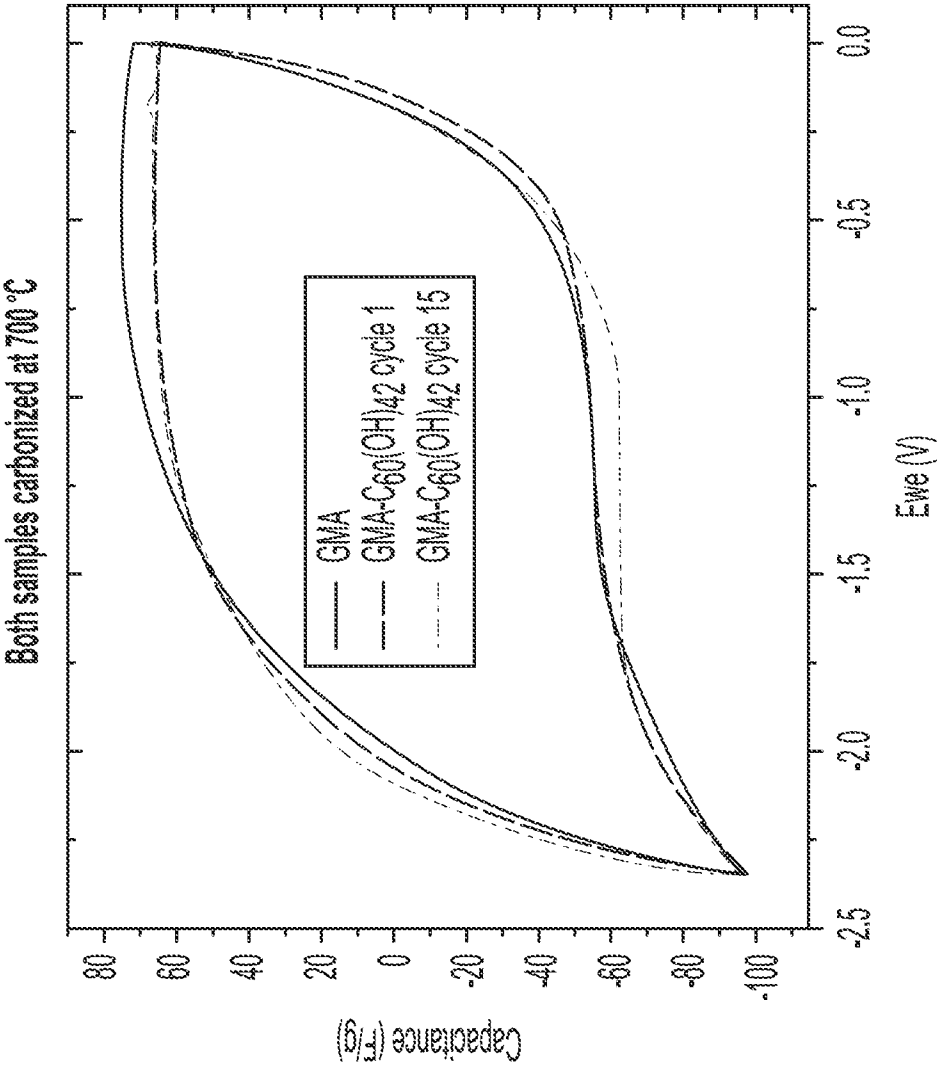


FIG. 12

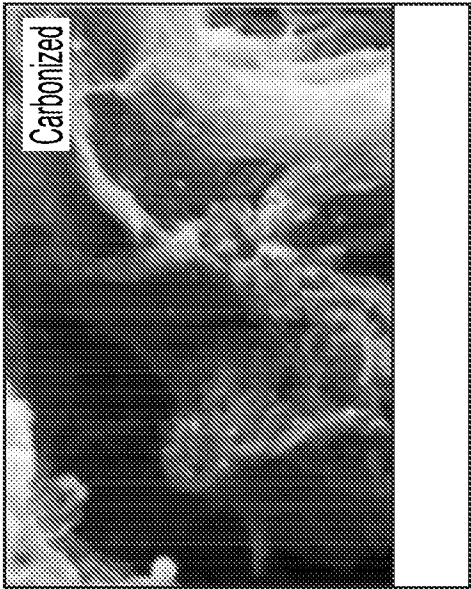


FIG. 13C

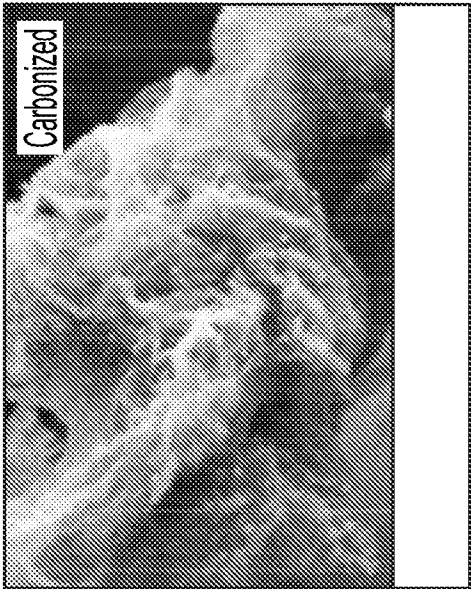


FIG. 13B

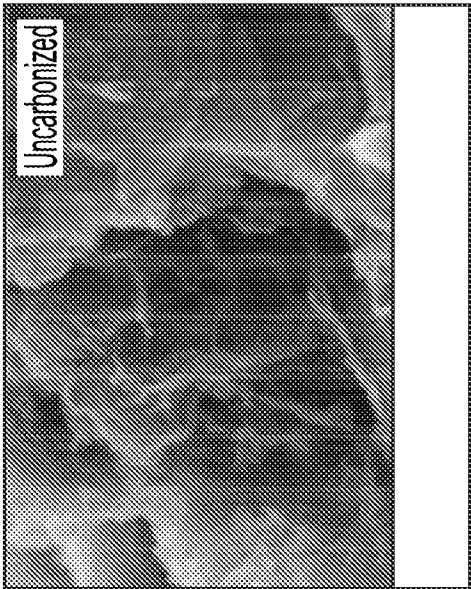


FIG. 13A

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US2021/040375

A. CLASSIFICATION OF SUBJECT MATTER

C01B 32/194(2017.01)i; **C01B 32/156**(2017.01)i; **H01M 4/133**(2010.01)i; **H01M 4/1393**(2010.01)i; **H01M 4/36**(2006.01)i; **H01G 11/36**(2013.01)i; B82Y 40/00(2011.01)n

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C01B 32/194(2017.01); C07C 1/32(2006.01); C07C 13/62(2006.01); H01G 11/24(2013.01); H01G 11/32(2013.01)

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Korean utility models and applications for utility models
Japanese utility models and applications for utility models

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

eKOMPASS(KIPO internal) & Keywords: graphene macro-assembly-fullerene composite, graphene oxide, hydroxylated fullerene, gelation process

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
DA	US 9870871 B1 (LAWRENCE LIVERMORE NATIONAL SECURITY, LLC) 16 January 2018 (2018-01-16) See example 1; claim 1.	1-4
A	GAN, TIAN et al., 'Facile synthesis of water-soluble fullerene-graphene oxide composites for electrodeposition of phosphotungstic acid-based electrocatalysts', Electrochimica Acta, 2013, vol. 111, pages 738-745 See the entire document.	1-4
A	ZHANG, XIAOYAN et al., 'Synthesis and characterization of a graphene-C60 hybrid material', Carbon, 2008, vol. 47, pages 313-347 See the entire document.	1-4
A	ALSULAM, IBRAHIM K. et al., 'High-Yield Continuous-Flow Synthesis of Spheroidal C60@Graphene Composites as Supercapacitors', ASC OMEGA, 2019, vol. 4, pages 19279-19286 See the entire document.	1-4
A	US 2017-0194105 A1 (ZHAMU, ARUNA et al.) 06 July 2017 (2017-07-06) See the entire document.	1-4



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance
"D" document cited by the applicant in the international application
"E" earlier application or patent but published on or after the international filing date
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
"O" document referring to an oral disclosure, use, exhibition or other means
"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

21 December 2021

Date of mailing of the international search report

21 December 2021

Name and mailing address of the ISA/KR

**Korean Intellectual Property Office
189 Cheongsa-ro, Seo-gu, Daejeon
35208, Republic of Korea**

Facsimile No. +82-42-481-8578

Authorized officer

HAN, Inho

Telephone No. +82-42-481-3362

INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.

PCT/US2021/040375

Patent document cited in search report			Publication date (day/month/year)	Patent family member(s)			Publication date (day/month/year)
US	9870871	B1	16 January 2018	US	10633255	B2	28 April 2020
				US	2018-0118574	A1	03 May 2018
US	2017-0194105	A1	06 July 2017	CN	108431915	A	21 August 2018
				JP	2019-506736	A	07 March 2019
				KR	10-2018-0099735	A	05 September 2018
				US	9905373	B2	27 February 2018
				WO	2017-120131	A1	13 July 2017

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US2021/040375

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. ☒ Claims Nos.: **10-11**
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:

Claims 10-11 are not clear because they refer to multiple dependent claims, which do not comply with PCT Rule 6.4(a).

3. ☒ Claims Nos.: **5-9, 12-15**
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).