[54]	GLASS I	FOR TELEVISION DISPLAY	FO	REIGN I
•	CATHO	DE-RAY TUBES	1,123,857	8/1968
[72]	Inventor:	Coenraad Maria La Grouw, Emmasingel, Eindhoven, Netherlands	Primary Exam Assistant Exa	
[73]	Assignee:	U.S. Philips Corporation, New York, N.Y.	Attorney—Fi	
[22]	Filed:	Oct. 22, 1969	[57]	
[21]	Appl. No.:	868,611	Glass for en	
[30]	Fore	eign Application Priority Data	at most 0.5 r	
	Oct. 28, 19	68 Netherlands6815397	of 40 to 45 k weight:	. volt, a
[52] [51] [58]	Int. Cl		$egin{array}{l} SiO_2 \ Li_2O \ Na_2O \end{array}$	58-67 0-1 2-3
[56]	ŤT	References Cited NITED STATES PATENTS	K₂O CaO BaO	11-14 3-4.5 11-14
	5,303 10/19 2,298 1/19	•		2

PATENTS OR APPLICATIONS

Great Britain.....106/53

-James E. Poer –M. Bell Trifari

ABSTRACT

s of television display cathode-ray tubes, glass for color television, which transmits X-ray radiation at an acceleration voltage and which has a composition in percent by

SiO_2	58-67	PbO	2-7
Li_2O	0-1	MgO	0-3
Na_2O	2-3	$\mathrm{Al_2O_3}$	1-4
K_2O	11-14	$-\operatorname{As}_2\operatorname{O}_3 \pm \operatorname{Sb}_2\operatorname{O}_3$	0.3 - 0.7
CaO	3-4.5	CeO_2	0.05-0.6
BaO	11-14		

2 Claims, No Drawings

GLASS FOR TELEVISION DISPLAY CATHODE-RAY TUBES

The invention relates to glass for an envelope of a television display cathode-ray tube, particularly glass for the face-plate of the tube.

Particular requirements are imposed on glass for envelopes of cathode-ray tubes for the display of colored television images as compared with that for the display of monochrome television images. Such special glasses are known, for example, from the British Pat. specification No. 1,123,857 the composition of which in percent by weight lies within the following range of compositions:

SiO ₂	62-66	BaO	11-14
Li₂O	0–1	MgO	0-3
Na₂O	7–8.5	PbO	0-2
K ₂ O	6.5-9	Al_2O_3	1-4
CaO	2-4.5	As ₂ O ₃ +	• •
CeO,	0.05-0.3	Sb_2O_3	0.3-0.7

The special requirements which, as compared with glass for the envelopes for monochrome display, are imposed on glass for envelopes for color display, are connected with differences in the manufacture and in the use of these tubes. In the first place, the glass components of envelopes for color display unlike those for the monochrome display envelopes cannot be sealed by fusing them together but must be connected together with the aid of an enamel. This is connected with the fact that a shadow mask is provided in these tubes, which mask determines the path of the required three electron beams. Furthermore, an extremely fine grating-like pattern of three different luminescent substances corresponding to the apertures of the shadow mask is provided on the inner side of the screen. The requirements relative to the maximum permissible distortion of the glass are in this case much more stringent in connection therewith than for glass of envelopes for monochrome display. In addition, the temperature at which the tube must be heated during evacuation and sealing must be approximately 20° higher and the heat treatment is of a longer duration than for the tubes for monochrome display.

The glasses within the above-mentioned range are eminently satisfactory in a technological respect relative to the softening point, the quality and the thermal coefficient of expansion. For the acceleration voltages until recently used on the electron guns, the absorption of these glasses for the X-ray radiation generated during operation as a result of the electron bombardment on the glass and on the shadow mask is sufficiently great. This even applies when the tube is built in in a cabinet in direct vision construction, thus without a protective to take the total capture of take to take the total capture of take the total capture of take to take the total capture of take the total capture of take to take the transmitting at most 0.5 m K. volt; it was found that the acceleration voltage of ever glasses. The electric resis cm and at least 10¹⁵ ohm. while these values are 10⁸ mentioned known glasses.

The requirement up till now had been that the intensity of the transmitted X-ray radiation may be at most 0.5 milliroentgen per hour (mr/h) at a maximum thickness of 11 mm of the screen glass, an acceleration voltage of 27.5 k. volt and an 55 anode current of 300 μ A in a television display tube.

There is, however, a tendency to still further increase the margin of safety to X-ray radiation transmitted by television display tubes. There is a need of a kind of glass in which at most 0.5 mr/h is transmitted at an acceleration voltage of 35 k.volt. The above-described glasses then no longer have a sufficiently high absorption and do not satisfy the stricter safety requirements. For reasons of a technological nature, the thickness of the screen cannot be increased much further than 11 mm. To obtain a sufficiently high absorption while using a 65 glass within the above-mentioned range of compositions, the screen should be thicker by as much as 2.5 mm.

For a satisfactory processing of the glass and moulding face-plates thereof, it is necessary that the temperature dependence of the viscosity is not too great. In practice this means 70 that the temperature difference between the softening point, which is the temperature at which the viscosity of the glass is $10^{7.6}$ poises, and the annealing point, which is the temperature at which the viscosity of the glass is $10^{13.4}$ poises, must be at least 190° C.

In connection with the conventional manufacturing technique and the very stringent requirements which are imposed on the maximum permissible distortion of the glass components during manufacture of the tube, it is necessary that glass for a color display tube has an annealing point which is not lower than 485° C.

Finally it is of importance that a glass for a color television display tube has approximately the same coefficient of expansion as that of the known glasses (approximately 99×10^{-7} between 30° and 300° C.), so that a better match is obtained with the existing glasses and metal components which must be sealed on or in respectively.

In the kind of glass according to the present invention, a content of PbO is present with an approximately equal BaO content relative to the known glass. It is by no means surprising in itself that the absorption of X-ray radiation is increased as a result thereof. It was, however, not obvious that it was possible to maintain the physical properties of the glass at the same level by means of a few other modifications. Furthermore, it is also known (from U. K. Pat. specification No. 664,769) that no discoloration occurs on the glass of the faceplate due to electron bombardment, provided that the glass contains CeO₂ and provided that the glass contains no more than 1 percent of readily reducible oxides. However, the glass according to the invention which does not satisfy the last-mentioned requirement owing to its content of PbO, does not discolor under the influence of the electron bombardment.

The range of glass compositions according to the present invention is characterized by the following limits in percent by weight:

	SiO ₂	58-67	PbO	2-7
	Li₂Ō	0-1	MgO	0-3
	Na ₂ O	2-3	Al ₂ O ₂	1-4
35	K₂Ō	11-14	As ₂ O ₃ +	0.3-0.7
	CaO	3-4.5	CeO ₂	0.05-0.6
	BaO	11-14	2002	0.05 0.0

The softening point of these glasses lies between 690° and 710° C.; the annealing point between 485° and 510° C. and the thermal coefficient of expansion is approximately 97 to 100 × 10⁻⁷ between 30° and 300° C. The glasses according to the invention amply satisfy the above-mentioned requirement of transmitting at most 0.5 mr/h at an acceleration voltage of 35 K.volt; it was found that this amount was not yet reached at an acceleration voltage of even 44 to 45 k.volt when using these glasses. The electric resistance thereof is at least 10⁹⁻⁴ ohm. cm and at least 10⁷⁻⁵ ohm.cm at 250° and 350°, respectively, while these values are 10⁸⁻⁵ and 10⁸⁻⁷ ohm.cm for the abovementioned known glasses.

The following glass is an example of a glass suitable for the relevant purpose. It is obtained in a manner which is common practice in glass technology by melting the relevant oxides or compounds which are converted into the oxides.

SiO ₂ Li ₂ O Na ₂ O K ₂ O CaO BaO	59% by weight 0.4% by weight 2.4% by weight 12.7% by weight 3.6% by weight 12.3% by weight	PbO 6.3% by weight. Al ₂ O ₃ 2.2% by weight. Sb ₂ O ₃ 0.3% by weight. CeO ₂ 0.5% by weight. Softening point 705°C. annealing point 502°C. coefficient of expansion $(30-300^{\circ}\text{ C.}) 97 \times 10^{-7}$ $\log p 250^{\circ}\text{ C.} = 10.3$
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What is claimed is:

1. Glass for envelopes of television display cathode-ray tubes, particularly intended for the face-plate of the tube, consisting essentially of the following in percent by weight:

$ \begin{array}{ccc} O & SiO_2 \\ Li_2O & Na_2O \\ K_2O & \end{array} $	58-67 0-1 2-3 11-14	$\begin{array}{c} \text{PbO} \\ \text{MgO} \\ \text{Al}_2\text{O}_3 \\ \text{As}_2\text{O}_3 + \end{array}$	2-7 0-3 1-4 0.3-0.7
BaO 11-	14	$\mathrm{Sb_2O_3}$	

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12.3 6.3 2.2 0.3 0.5

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wherein said glass transmits less than 0.5 mr/h of X-ray radiation at an electron acceleration voltage up to 45 k.volts. through a thickness of said glass of 1 i mm. 2. Glass as claimed in claim 1, consisting essentially of the following in percent by weight:		SiO₂ Li₂O Na₂O K₂O CaO	59 0.4 2.4 12.7 3.6	* * * *	BaO PbO Al ₂ O ₃ Sb ₂ O ₃ CeO ₂
	10				
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	25				
	30				
	35				
	40				
	45				
	50 55				
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75

65

70

6636

PO-1050 (5/69)

UNITED STATES PATENT OFFICE CERTIFICATE OF CORRECTION

Patent	No. 3,663,246	Dated_	May	16,	1972	

Inventor(s) COENRAAD MARIA LA GROUW

It is certified that error appears in the above-identified patent and that said Letters Patent are hereby corrected as shown below:

In the Index, under "[30] Foreign Application Priority Data" the priority date should read -- Oct. 29, 1968 --.

Column 2, line 35, "As $_2$ 0 $_3$ +" should read -- As $_2$ 0 $_3$ + Sb $_2$ 0 $_3$ ---.

In Claim 1, on the line befor "Ba0 11-14" insert the following: -- Ca0 3-4.5 CeO_2 0.05-0.6 --.

Signed and sealed this 1st day of May 1973.

(SEAL) Attest:

EDWARD M. FLETCHER, JR. Attesting Officer

ROBERT GOTTSCHALK Commissioner of Fatents

Signed and sealed this

day of

, 1972.

PO-1050 (5/69)

UNITED STATES PATENT OFFICE CERTIFICATE OF CORRECTION

Patent	No.	3.	66	3.3	246
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Dated May 16, 1972

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