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- (71) **Applicant: RANBAXY LABORATORIES LIMITED**
[IN/IN]; Head Office: 12th Floor, Devika Tower, 06 Nehru
Place, New Delhi, Delhi 110019 (IN).
- (72) **Inventors: REDDY, P., Ramesh;** Village - C.C. Palli,
Mandal-Galiveedu, Kadapa, Andhra Pradesh 516267 (IN).
SHIRSATH, Krishnarao, Tukaram; Village - Pashte,
Taluka - Shindkheda, House No. 405, Village - Pashte,
Taluka -, Shindkheda, Dhule 425 405, Maharashtra (IN).
RAI, Bishwa, Prakash; Jagdishpur, Sohauly, Azamgarh
276301, Uttar Pradesh (IN). **SINGH, Shailendra, K.;** U-
18/9, FF, Phase - III, DLF City, Gurgaon, Haryana 122002
(IN). **TIWARI, Neera;** K-10/19, First Floor, Phase - II,
QLF, Qutab Enclave, Gurgaon 122001, Haryana (IN).
PRASAD, Mohan; D-50, Greenwoods City, Sector 46,
Gurgaon 122003, Haryana (IN). **ARORA, Sudershan,**

Kumar; A-3/803 Sahara Grace, Behind Sahara Mall, M.G.
Road, Gurgaon 122001, Haryana (IN).

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(54) **Title:** PROCESS FOR THE PURIFICATION OF BIAPENEM

(57) **Abstract:** The present invention relates to a process for the purification of biapenem.

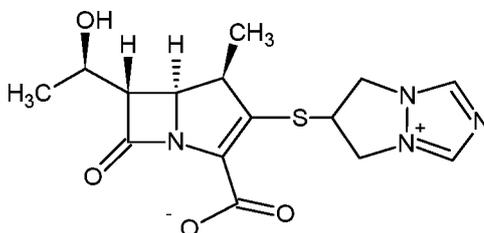
PROCESS FOR THE PURIFICATION OF BIAPENEM

Field of the Invention

The present invention relates to a process for the purification of biapenem.

Background of the Invention

5 Biapenem is chemically known as 6-[[2(4R,5S,6S)-carboxy-6-[(1R)-hydroxyethyl]-4-methyl-7-oxo-1-azabicyclo[3.2.0]hept-2-en-3-yl]thio]6,7-dihydro-5H-pyrazolo[1,2-a][1,2,4]triazol-4-ium inner salt, and is represented by Formula 1. It is indicated for the treatment of bacterial infection and sepsis.



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Formula 1

U.S. Patent No. 4,866,171, in Example 6, discloses the purification of biapenem using chromatography and/or lyophilization techniques. This patent also describes a process for the conversion of amorphous biapenem into a crystalline form by dissolving the amorphous biapenem in water while heating, followed by cooling, then washing the
15 obtained crystals with a 50% aqueous ethanol solution.

U.S. Patent No. 5,241,073 describes a process for the purification of biapenem involving column chromatography and crystallization with ethanol.

U.S. Patent No. 5,286,856 describes a process for the crystallization of biapenem from an aqueous solution, comprising maintaining the temperature of the aqueous solution
20 from eutectic temperature (-10°C to -2°C) to a temperature lower than 0°C, followed by lyophilization.

The *Journal of Organic Chemistry*, 63(23):8145-8149 (1998) describes the purification of biapenem involving resin chromatography.

The present invention provides an alternate process for the purification of
25 biapenem that avoids making use of tedious techniques like chromatography and lyophilization. At the same time, it results in a high yield and high purity of the final

product. Advantageously, the crystalline biapenem of this invention can be directly isolated from the reaction mixture. Further, the process of the present invention involves fewer steps, is easily scalable, and industrially advantageous.

Summary of the Invention

5 The present invention provides a process for the purification of biapenem comprising the steps of:

- (a) treating an aqueous solution of biapenem with charcoal;
- (b) filtering the mixture obtained in step (a);
- (c) adjusting the pH of the filtrate obtained in step (b) to 4.5 to 5.5; and
- 10 (d) adding an antisolvent to the solution obtained in step (c).

The present invention also provides a process for the purification of biapenem which comprises a step of crystallizing biapenem by adding an antisolvent to an aqueous solution of biapenem at a pH range of 4.5 to 5.5.

Detailed Description of the Invention

15 The term "about", as used herein, refers to any value which lies within the range defined by a variation of up to $\pm 10\%$ of the value.

The term "base", as employed herein, is meant to comprise sodium hydroxide, potassium hydroxide, magnesium hydroxide, ammonia solution, dipotassium hydrogen orthophosphate, magnesium carbonate, sodium carbonate, potassium carbonate, pyridine, 20 trimethylamine, triethylamine, diisopropylethylamine, N-methyl morpholine, and the like.

The term "antisolvent", as employed herein, is meant to comprise a solvent which is capable of crystallizing out biapenem from an aqueous solution of biapenem. Some of the non-limiting examples of "antisolvent" are acetone, ethyl methyl ketone, methanol, ethanol, propanol, isopropanol, and the like.

25 The present invention can be explained by way of the following aspects.

A first aspect of the present invention provides a process for the purification of biapenem comprising the steps of:

- (a) treating an aqueous solution of biapenem with charcoal;
- (b) filtering the mixture obtained in step (a);
- 30 (c) adjusting the pH of the filtrate obtained in step (b) to 4.5 to 5.5; and

(d) adding an antisolvent to the solution obtained in step (c).

According to one embodiment of this aspect, the aqueous solution of biapenem can be prepared by dissolving biapenem in water.

In another embodiment, the filtration of the mixture obtained in step (a) can be
5 performed through a hyflo bed and/or a filter paper.

In another embodiment, the pH of the filtrate obtained in step (b) can be adjusted to a range of 4.5 to 5.5 by the addition of a base.

In another embodiment, an antisolvent is added to the solution obtained in step (c) to crystallize out biapenem.

10 In another embodiment, the antisolvent is acetone.

Accordingly, an aqueous solution of biapenem is prepared by dissolving biapenem in water while heating at a temperature of about 60°C to about 70°C, followed by cooling to a temperature of about 25°C to about 35°C within 10 minutes. To this aqueous solution of biapenem, activated charcoal (enoantichromos carbon) is added and the resultant
15 mixture is filtered. The filtration can be performed through a hyflo bed and/or a filter paper. The pH of the filtrate is then adjusted to a range of 4.5 to 5.5 by adding a base. An antisolvent is added to the resultant solution to crystallize out biapenem, which is then filtered, washed, and dried under vacuum to obtain the purified biapenem.

The term "purified", as employed herein, refers to a purity of greater than 99% as
20 determined by HPLC.

This aspect of the present invention provides biapenem with greater than 80% yield and greater than 99% HPLC purity.

A second aspect of the present invention provides a process for the purification of biapenem which comprises a step of crystallizing biapenem by adding an antisolvent to an
25 aqueous solution of biapenem at a pH range of 4.5 to 5.5.

According to one embodiment of this aspect, an aqueous solution of biapenem can be prepared by dissolving biapenem in water.

In another embodiment, the pH of the solution can be adjusted to a range of 4.5 to 5.5 by the addition of a base.

30 In another embodiment, the antisolvent is acetone.

Accordingly, an aqueous solution of biapenem is prepared by dissolving biapenem in water while heating at a temperature of about 60°C to about 70°C, followed by cooling to a temperature of about 25°C to about 35°C within 10 minutes. This solution is treated with activated charcoal (enoantichromos carbon) and filtered. The pH of the filtrate is
5 adjusted to a range of 4.5 to 5.5 by adding a base. An antisolvent is added to the resultant solution to crystallize out biapenem, which is then filtered, washed, and dried under vacuum to obtain the purified biapenem.

This aspect of the present invention provides biapenem with greater than 80% yield and greater than 99% HPLC purity.

10 The biapenem to be purified (starting material) can be obtained using known methods, for example, the processes described in U.S. Patent Nos. 4,866,171 or 5,241,073.

While the present invention has been described in terms of its specific aspects, certain modifications and equivalents will be apparent to those skilled in the art, and are intended within the scope of the present invention.

15 EXAMPLES

Example 1: Purification of Biapenem

Biapenem (12 g) was added into water (300 mL) at 65°C, stirred for 5 minutes, and cooled to 30°C within 10 minutes. Enoantichromos carbon (0.6 g) was added to the reaction mixture and stirred for 10 minutes to 15 minutes at 25°C to 30°C. The reaction
20 mixture was filtered through a hyflo bed and washed with water (36 mL). The filtrate obtained was passed through a 0.45 micron filter, and its pH was adjusted to 5.5 using 5% aqueous sodium hydroxide solution at 10°C to 15°C. Acetone (336 mL) was added to the reaction mixture at 5°C to 10°C. The resultant slurry was stirred for 3 hours at 5°C to 10°C, filtered, and the obtained solid was washed with acetone (60 mL). The solid was
25 dried under reduced pressure (720 mmHg) at 30°C to 35°C to obtain the title product as white crystals.

Yield: 84%

HPLC Purity: 99.87%

Example 2: Purification of Biapenem

Biapenem (18 g) was added into water (450 mL) at 65°C, stirred for 5 minutes, and cooled to 30°C within 10 minutes. Enoantichromos carbon (0.9 g) was added to the reaction mixture and stirred for 30 minutes at 25°C to 30°C. The reaction mixture was
5 filtered through a hyflo bed and washed with water (54 mL). The filtrate obtained was passed through a 0.45 micron filter and its pH was adjusted to 4.9 using 5% aqueous sodium hydroxide solution at 10°C to 15°C. Acetone (504 mL) was added to the reaction mixture at 10°C to 15°C. The resultant slurry was stirred for 3 hours at 5°C to 10°C,
10 filtered, and the obtained solid was washed with acetone (90 mL). The solid was dried under reduced pressure (720 mmHg) at 35°C to 40°C to obtain the title product as white crystals.

Yield: 81.77%

HPLC Purity: 99.80%

We Claim:

- 1 1. A process for the purification of biapenem comprising the steps of:
 - 2 (a) treating an aqueous solution of biapenem with charcoal;
 - 3 (b) filtering the mixture obtained in step (a);
 - 4 (c) adjusting the pH of the filtrate obtained in step (b) to 4.5 to 5.5; and
 - 5 (d) adding an antisolvent to the solution obtained in step (c).
- 1 2. The process of claim 1, wherein the filtration is performed through a hyflo bed
2 and/or a filter paper.
- 1 3. The process of claim 1, wherein the pH is adjusted to 4.5 to 5.5 by the addition of a
2 base.
- 1 4. The process of claim 1, wherein an antisolvent is added to crystallize out
2 biapenem.
- 1 5. The process of claim 1 or claim 4, wherein the antisolvent is acetone.
- 1 6. A process for the purification of biapenem which comprises a step of crystallizing
2 biapenem by adding an antisolvent to an aqueous solution of biapenem at a pH range of
3 4.5 to 5.5.
- 1 7. The process of claim 6, wherein the pH is adjusted to 4.5 to 5.5 by the addition of a
2 base.
- 1 8. The process of claim 6, wherein the antisolvent is acetone.
- 1 9. The process of claim 6, wherein the aqueous solution of biapenem is formed by
2 treating an aqueous solution of biapenem with charcoal and filtering the resulting mixture.

INTERNATIONAL SEARCH REPORT

International application No
PCT/IB2013/061157

A. CLASSIFICATION OF SUBJECT MATTER
INV. C07D477/20
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal , WPI Data, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WO 02/057266 AI (MERCK & CO INC [US] ; WILLIAMS JOHN M [US] ; SKERLJ RENATO [US]) 25 July 2002 (2002-07-25) page 13, line 2 - line 11; example 1 -----	1-9
Y	WO 2009/047604 AI (ORCHID CHEMICALS & PHARM LTD [IN] ; UDAYAMPALAYAM PALANISAMY SENTH [IN]) 16 April 2009 (2009-04-16) page 6, line 34 - page 7, line 5 -----	1-9
Y	CN 102 268 025 A (HAINAN MEILAN SMITH KLINE PHARMACEUTICAL CO., LTD.) 7 December 2011 (2011-12-07) abstract & machine translation of CN102268025A provided by Google -----	1-9

Further documents are listed in the continuation of Box C.

See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

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"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

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Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Timmermans, Michel

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/IB2013/061157

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 02057266	A1	25-07-2002	AR 035728 A1 07-07-2004
			AT 278691 T 15-10-2004
			AU 2002234240 B2 27-04-2006
			BR 0206372 A 23-12-2003
			CA 2433874 A1 25-07-2002
			CN 1486318 A 31-03-2004
			CZ 20031911 A3 15-10-2003
			DE 60201498 D1 11-11-2004
			DE 60201498 T2 17-11-2005
			DK 1353923 T3 31-01-2005
			EG 23993 A 10-03-2008
			EP 1353923 A1 22-10-2003
			ES 2227423 T3 01-04-2005
			HR P20030566 A2 30-06-2005
			HU 0302764 A2 28-11-2003
			IL 156507 A 30-06-2010
			JP 4098626 B2 11-06-2008
			JP 2004518672 A 24-06-2004
			KR 20040007438 A 24-01-2004
			MX PA03006322 A 06-10-2003
			NZ 527191 A 30-04-2004
			PL 362912 A1 02-11-2004
			PT 1353923 E 31-01-2005
			SI 1353923 T1 28-02-2005
			SK 8972003 A3 07-10-2003
			TW 1318627 B 21-12-2009
UA 74870 C2 15-09-2003			
US 2004063931 A1 01-04-2004			
WO 02057266 A1 25-07-2002			
YU P55003 A 25-05-2006			

WO 2009047604	A1	16-04-2009	EP 2209787 A1 28-07-2010
			US 2010311984 A1 09-12-2010
			WO 2009047604 A1 16-04-2009

CN 102268025	A	07-12-2011	NONE
