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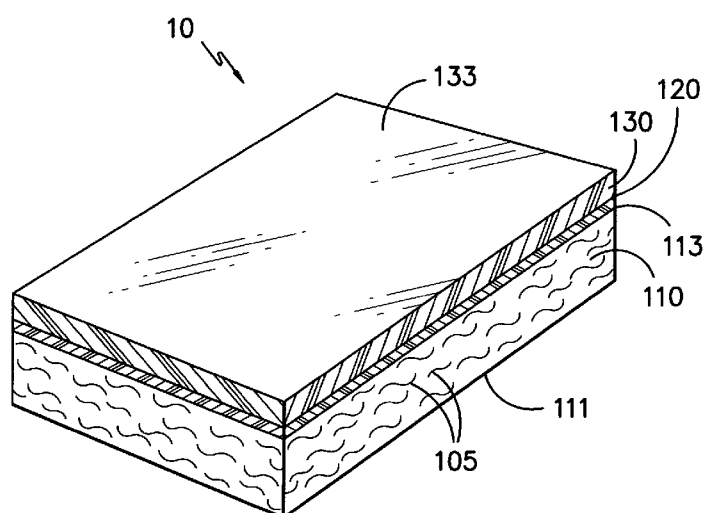


FIG. -1-

(57) Abstract: An odiferous organic compounds cover having an inner surface and an outer surface. The cover contains a thermoplastic non-woven layer having a first side and a second side and comprising thermoplastic fibers. The first side of the non-woven layer forms the inner side of the cover, has a contact angle with water greater than about 70°, and is non-wicking. The cover contains a UV barrier layer on the second side of the non-woven layer. The UV barrier layer has a thickness of between one to three times the diameter of the thermoplastic fibers of the non-woven layer and the UV barrier layer comprises discontinuities forming microchannels through the layer thickness. The cover also contains an odor mitigation layer on the UV barrier layer on the side of the barrier layer opposite the non-woven layer. The odor mitigation layer comprises a UV translucent polymer and anatase titanium dioxide.

COVER FOR CONTAINING AND ABATING ODIFEROUS ORGANIC EMANATIONS, COMPRISING A NONWOVEN LAYER AND TWO POROUS POLYMERIC LAYERS, AS WELL AS ANATASE AS PHOTOCATALYST, AND CONTAINMENT SYSTEM USING SUCH COVER

FIELD OF THE INVENTION

5 [0001] The present application is directed to covers for decaying organic materials. More particularly, the present application is directed towards a non-woven odiferous organic compound cover having additional polymeric layers and odor mitigating properties.

10 BACKGROUND

 [0002] Recent advances in the use of composting at waste treatment facilities has enabled the world to deal with their waste in a much more environmentally beneficial and cost effective manner. Composting allows the breakdown of organic materials back to environmentally beneficial byproducts in
15 a more time efficient and controlled manner. Additionally, composting can reduce the volume of material required to be land filled by up to 80%.

 [0003] Covering the composting pile with a textile type substance has shown many advantages. The cover allows the pile to generate a more uniform level of heat, keeps unwanted vermin and scavengers from entering the pile,
20 helps to maintain consolidation of the material, and allows for better moisture control of the compost pile. The use of plastic film has been used for many years and can in many states qualify the composting material for "in vessel" consideration. One issue with plastic covers, and even some textile covers, is the buildup of volatile organic compounds within the pile, as well as the inability
25 of the composting material to properly degenerate in an aerobic environment.

 [0004] Thus, there is a need for a cover which is capable of providing the positive aspects above, and also helps to control the emission of volatile substances. More particularly a cover which can control odorous fumes and pathogens developed during decomposition of the waste; and also provides
30 appropriate gaseous exchange and water vapor transmission to provide an

optimal aerobic environment. The optimal cover will provide operational reliability, product quality, minimum investment, and operating cost.

BRIEF SUMMARY OF THE INVENTION

5 [0005] The invention relates to an odiferous organic compounds cover having an inner surface and an outer surface. The cover contains a thermoplastic non-woven layer having a first side and a second side and comprising thermoplastic fibers. The first side of the non-woven layer forms the inner side of the cover, has a contact angle with water greater than about 70°,
10 and is non-wicking. The cover contains a UV barrier layer on the second side of the non-woven layer, where the UV barrier layer contains a UV blocking polymer which blocks at least 50% of UV(C) energy. The UV barrier layer has a thickness of between one to three times the diameter of the thermoplastic fibers of the non-woven layer and the UV barrier layer comprises discontinuities
15 forming microchannels through the layer thickness. The cover also contains an odor mitigation layer on the UV barrier layer on the side of the barrier layer opposite the non-woven layer. The odor mitigation layer comprises a UV translucent polymer which transmits at least about 60% of UV(C) energy and anatase titanium dioxide. The odor mitigation layer comprises discontinuities
20 forming microchannels through the layer thickness forms the outer surface of the cover. A system for composting waste using the cover is also disclosed.

BRIEF DESCRIPTION OF THE DRAWINGS

25 [0006] Fig. 1 illustrates a cross-section of one embodiment of the invention.

 [0007] Fig. 2 illustrates a cross-section of another embodiment of the invention.

DETAILED DESCRIPTION OF THE INVENTION

[0008] The invention is directed towards the containment of odiferous organic compounds that may create a human nuisance. The term "compost cover" is defined as a cover for decaying organic mater, the origin of which could be from composting or decaying materials, sewage treatment, cesspool, hog lagoons, manure dumps, portable toilets, trashcans, body bags, and the like.

[0009] Referring now to Figure 1, there is shown one embodiment of the odiferous organic compounds cover 10 (also referred to herein in the specification as the compost cover) including in order a non-woven layer 110, a UV barrier layer 120, and an odor mitigation layer 130. The non-woven layer 110 forms the inner surface 101 of the compost cover 10 adjacent the compost or other waste to be contained. The odor mitigation layer 130 forms the outer surface 133 of the cover 10 facing the surrounding environment. The cover 10 performs in a manner to provide controlled release of VOCs from the compost pile to the environment; and adequate oxygen levels from the environment to the compost pile. In one embodiment, the total thickness of the compost cover 10 is between about 200 to 1000 micrometers, more preferably 280 to 400 micrometers.

[0010] The non-woven layer 110 comprises thermoplastic fibers in a non-woven construction. The non-woven layer 110 contains thermoplastic fibers. Preferably, these fibers are synthetic fibers of a hydrophobic nature such as polypropylene or polypropylene co-polymer fibers. The fibers in the non-woven layer may be either staple or continuous. The non-woven layer has a first side 111 forming the inner surface of the compost cover 10 and a second side 113 in which the additional layers are placed. The term non-woven refers to structures incorporating a mass of fiber elements that are entangled and/or heat fused so as to provide a coordinated structure with a degree of internal coherency. Non-woven fabrics or webs may be formed from many processes such as for example, meltspun processes, hydroentangeling processes, mechanically entangled processes and the like.

[0011] In one embodiment, the thermoplastic fibers 105 of the non-woven layer 110 are mechanically entangled, and then a portion of the thermoplastic

fibers 105 are bonded to one another through calendaring the non-woven layer 110 on at least one side. Preferably, the non-woven layer 110 is calendared on the first side 111 (the side which forms the inner surface of the compost cover 10). Calendaring the side 111 of the non-woven layer 110 smoothes the surface of the layer increasing its strength, tear, and abrasion resistance and reduces the chance of the compost abrading or catching on the compost cover 10. In another embodiment, both sides of the non-woven layer 110 are calendared further enhancing the strength of the non-woven layer 110.

[0012] The inner surface 111 of the compost cover is hydrophobic with a contact angle greater than 70° , more preferably greater than 90° . The non-woven layer 110 is also non-wicking. Non-wicking is defined as having a water vapor transmission of less than $1000 \text{ g/m}^2/24$ hours as measured by ASTM E-96 standard. In a more preferred embodiment, the compost cover 10 has a water vapor transmission of less than $500 \text{ g/m}^2/24$ hours as measured by ASTM E-96 standard. This hydrophobic, non-wicking surface causes the water vapor from the compost to condense and form water droplets on the inner surface 111 of the compost cover 10 instead of diffusing through the cover. These water droplets form a film of water that actively contains the VOCs via dissolution. Through this mechanism, at least a portion of the odors emanating from the composting waste are blocked from the surrounding environment.

[0013] A UV barrier layer 120 is applied to the second side 113 of the non-woven layer 110. The UV barrier layer 120 contains a UV blocking polymer, where the UV blocking polymer blocks at least 50% of UV(C) energy. More preferably, the UV blocking polymer blocks at least about 70% of UV(C) energy. The UV barrier layer has a thickness of between one to three times the diameter of the thermoplastic fibers of the non-woven layer, more preferably about two times the diameter of these fibers. This allows some of the fibers from the non-woven layer 110 to poke through the UV barrier layer 120 creating discontinuities. These discontinuities form microchannels through the layer thickness and allow some water vapor and other gases such as carbon dioxide to diffuse through the compost cover 10. In one embodiment, the UV barrier layer 120 and the non-woven layer 110 together should have an air permeability

of less than about 500 cfm/ft². Preferably, the UV blocking polymer comprises silicone. It has been found that silicone provides good UV blocking, creates a layer with microchannels at the proper thicknesses, and adds to the strength and durability of the compost cover 10.

5 [0014] An odor mitigation layer 130 is located on the UV barrier layer 120 (on the side opposite to the non-woven layer 110) forming the outer surface 133 of the compost cover 10. The odor mitigation layer contains a UV translucent polymer transmits at least about 60% of UV(C) energy, anatase titanium dioxide, and optionally activated carbon. In a preferred embodiment, the UV translucent
10 polymer which transmits at least about 90% of UV(C) energy. In one embodiment, the UV translucent polymer is polyurethane. It has been found that polyurethane has good environmental stability and does not passivate the TiO₂.

 [0015] When irradiated with light, titanium dioxide produces electrons with a powerful reductive action and holes with a powerful oxidative action, and
15 decomposes any molecular species with which it comes into contact by a redox action. This action of titanium dioxide, namely, its photocatalytic action, can be used to break down volatile organic compounds. A major advantage of this process is that it can be repeated without depleting the effective agent. Additionally, because the reaction products are harmless carbon dioxide and so
20 on, there are fewer restrictions on the reaction conditions, such as temperature, pH, gas atmosphere, and toxicity, than with such methods as biological treatment using microbes. Furthermore, even organic halogen compounds, organic phosphorus compounds, and other such compounds that are difficult to treat with methods such as biological treatment can be easily decomposed and
25 removed.

 [0016] In terms of performance as a photo catalyst, the crystal form of titanium dioxide is preferably anatase. Rutile, brookite, and amorphous forms are less desirable because the activity of the photo catalyst will be lower. A metal such as platinum, rhodium, ruthenium, palladium, silver, copper, or zinc may be
30 supported on the surface of the titanium dioxide, which further raises the oxidative decomposition rate of chemical substances. Preferably the TiO₂ has a

surface area of at least about 40 m²/gram. One type of anatase TiO₂ that is suitable for the application is Degussa P25.

[0017] In one embodiment, the odor mitigating layer 130 further comprises activated carbon. As used herein, "activated carbon" refers to highly porous carbon having a random or amorphous structure. Granules and pellets of activated carbon are well known, such as the products manufactured by Calgon Carbon, Inc. (Pittsburgh, Pa.), and can be used in the present invention. Activated carbon from any source can be used, such as that derived from bituminous coal or other forms of coal, or from pitch, coconut shells, corn husks, polyacrylonitrile (PAN) polymers, charred cellulosic fibers or materials, wood, and the like. Activated carbon particles can, for example, be formed directly by activation of coal or other materials, or by grinding carbonaceous material to a fine powder, agglomerating it with pitch or other adhesives, and then converting the agglomerate to activated carbon.

[0018] In one embodiment, the activated carbon is provided with functional groups to modify the surface properties of the product. For example, during the activation stage, the carbon can be exposed to hydrogen chloride to add chlorine groups, to oxygen or water vapor to add oxygen or hydroxyl groups, to ammonia to add amine groups, to hydrogen to add hydrogen, and so forth. Alternatively, a compound such as a non-gaseous molecule may be added to the carbon prior to activating it or prior to a post-treatment step, wherein the compound reacts at elevated temperature to add functional groups to the activated carbon.

[0019] Activated carbon in any form can also be impregnated with other materials to increase the adsorption of specific species. For example, activated carbon impregnated with citric acid can be used to increase the ability of activated carbon to adsorb ammonia. Impregnation with sodium hydroxide or other caustic compounds can be useful for removal of hydrogen sulfide. Impregnation with metals or metal ions such as copper sulfate and copper chloride is believed to be useful for removal of other sulfur compounds.

[0020] The odor mitigation layer 130, like the UV blocking layer 120, contains discontinuities, these discontinuities form microchannels through the layer thickness and allow water vapor and other gases such as carbon dioxide to diffuse through the compost cover 10. This allows for the release of gaseous buildup within the compost pile and maintenance of an aerobic condition of the compost pile. Preferably, the odor mitigation layer has a thickness of between about 2 and 12 mm. Additionally, the odor mitigation layer preferably contains about 0.05 to 4.0 %wt of TiO_2 and about 0.1 to 8 %wt activated carbon.

[0021] In one embodiment, the non-woven layer 110 further comprises a woven layer embedded within the non-woven layer as shown in Figure 2. There is shown in Figure 2 a compost cover 10 invention including in order a non-woven layer 110 containing non-woven fibers 105 and a woven layer 106, a UV barrier layer 120, and an odor mitigation layer 130. The non-woven layer 110 forms the inner surface 101 of the compost cover 10 adjacent the compost or other waste to be contained. The odor mitigation layer 130 forms the outer surface 133 of the compost cover 10 facing the surrounding environment.

[0022] The woven layer 106 in the non-woven layer 110 serves to enhance the tensile strength and durability to the compost cover 10. The woven layer 105 may be formed of any fibers and any materials suitable to the end use of the compost cover. For purposes of the present invention, fiber is defined as an elongated body, the length dimension of which is much greater than the dimensions of width and thickness. Accordingly, the term fiber as used herein includes a monofilament elongated body, a multifilament elongated body, ribbon, strip, and the like having regular or irregular cross sections. The term fibers includes a plurality of any one or combination of the above. By way of example only and not limitation, it is contemplated that the fibers in the woven layer 106 may be from natural or synthetic fibers such as polypropylene, polyester, glass (including fiberglass), nylon, and the like. Fibers such as glass and nylon are particularly beneficial in that they typically provide the greatest increase in tensile strength to the non-woven layer 110 at the lowest weight and cost.

[0023] One potentially preferred construction for the woven layer 106 in the non-woven layer 110 is a plain weave construction of ribbon elements of polypropylene. It is to be understood that the term "tape element" is meant to denote a relatively flat structure or tape having a width dimension substantially greater than a height dimension. Of course round yarn structures such as spun or monofilament yarns as well as other geometries may likewise be utilized if desired. In one contemplated woven construction, stabilized polypropylene ribbon fiber is woven at a density of about 24 ends per inch by about 22 picks per inch. The ribbon fiber forming the ends has a linear density of about 1000 denier. The ribbon fiber forming the picks has a linear density of about 600 denier.

[0024] The term "polypropylene" is intended to encompass any polymeric composition comprising propylene monomers, either alone or in mixture or copolymer with other randomly selected and oriented polyolefins, dienes, or other monomers (such as ethylene, butylene, and the like). Such a term also encompasses any different configuration and arrangement of the constituent monomers (such as syndiotactic, isotactic, and the like). Thus, the term as applied to fibers is intended to encompass actual long strands, tapes, threads, cut up segments and the like, of drawn polymer. The polypropylene may be of any standard melt flow (by testing); however, standard fiber grade polypropylene resins possess ranges of Melt Flow Indices between about 2 and 50.

[0025] In one embodiment, the polypropylene tape elements are stabilized. The term "stabilized polypropylene" refers to polypropylene to which one or more nucleating agents have been added. The terms "nucleating agent", and "nucleating agents" are intended to generally encompass, singularly or in combination, any additive to polypropylene that produces nucleation sites for polypropylene crystals from transition from its molten state to a solid, cooled structure. Hence, since the polypropylene composition (including nucleating agent additives) must be molten to eventually extrude the fiber itself, the nucleating agent provides such nucleation sites upon cooling of the polypropylene from its molten state. Such compounds provide the necessary nucleation sites prior to polypropylene recrystallization itself. Thus, any

compound that exhibits such a beneficial effect and property is included within this definition. Such nucleating agents more specifically include dibenzylidene sorbitol types, including, without limitation, dibenzylidene sorbitol (DBS), monomethyldibenzylidene sorbitol, such as 1,3:2,4-bis(p-methylbenzylidene) sorbitol (p-MDBS), dimethyl dibenzylidene sorbitol, such as 1,3:2,4-bis(3,4-dimethylbenzylidene) sorbitol (3,4-DMDBS); other compounds of this type include, again, without limitation, sodium benzoate, NA-11, and the like. Such a stabilized polypropylene is described in U. S. Patent Application Publication 2003/01341118 and US Patents 6,866,912 and 6,998,081, each of which are hereby fully incorporated by reference. The nucleation of the polypropylene fibers can be used to enhance fiber tensile strength as well as improve resistance to UV degradation.

[0026] In a preferred embodiment, the non-woven fibers 105 of the non-woven layer 110 are entangled through the woven layer 106 by needlepunching. By way of example only and not limitation, according to one contemplated practice, the non-woven layer 110 containing a woven layer 106 is a fused multi-component structure of a woven layer 106 and a non-woven fibers 105 needle punched through the woven layer 105. When the optional process of calendaring is applied to one or both sides a portion of the non-woven fibers 105 and the woven fibers fuse to form an enhanced stable non-woven layer 110. More details about materials and constructions of the non-woven layer containing a woven layer may be found in US Patent Publication 2003-0175475 and US Patent 6,866,912, both of which are herein incorporated by reference.

EXAMPLES

[0027] Various embodiments of the invention are shown by way of the Examples below, but the scope of the invention is not limited by the specific Examples provided herein.

Testing Methods

Air Flow Permeability

[0028] Air flow permeability was measured using were done using the ASTM D-737 test and the results are indicated using the units of cubic feet/minute/square foot of fabric (CFM/ft²).

VOC Diffusion – Static

5 [0029] Odiferous organic compounds cover samples were used to seal a vial containing methyl salicylate VOC. This sealed vial was then placed into a second vial and left to equilibrate for 20 minutes. The air in the second vial (outside of the first vial) was then samples and analyzed to determine the amount of VOC diffusing through the compost cover examples.

10 *VOC after exposure to Sunlight*

[0030] Narrow bottles with septum caps each containing different materials were spiked with the same concentration of propionaldehyde. These bottles were kept outside in the sun for 12 hours during a sunny day for UV exposure. The contents of these bottles were analyzed using manual
15 headspace.

[0031] The control bottle had no additional odor mitigating materials in the bottle. In the rest of the bottles, equivalent amounts of materials typically used in the compost cover were used.

Examples

20 *Example 1*

[0032] A thermoplastic non-woven layer was formed from staple polypropylene fibers having an average length of about 3 inches and an average denier of about 6. The fibers were entangled using needle punching. The non-woven layer was calendared on both sides using a heat of about 170°C (338°F).
25 The resultant non-woven layer had a thickness of 400 micrometers and a density of 5 oz/yd².

Example 2

[0033] The non-woven layer of Example 1 was coated with a UV barrier layer. The UV barrier layer was a silicone layer having a thickness of about 76

micrometers. The silicone used was Elastosil® LR 6293 available from the Wacker Silicone a division of Wacker-Chemie GmbH. The silicone layer had discontinuities forming microchannels through the layer thickness.

Example 3

5 [0034] The layers of Example 2 were coated with an odor mitigation layer, with the odor mitigation layer being coated on top of the silicone layer. The odor mitigation layer contained anatase TiO₂ and activated carbon in polyurethane. The polyurethane used was MILLITEX™ Resin MRX (a blocked isocyanate based cross-linking agent (with 35-45% solids) available from Milliken Chemical.

10 The anatase TiO₂ had a surface area of about 40 m²/g and was obtained from DeGussa® P25 from Evonik Degussa in Germany. The activated carbon had an average particle size of 20 micrometers and was obtained as the GX230 grade supplied by PICA USA Inc. The polyurethane layer had approximately 0.8%wt TiO₂ and approximately 1.5%wt activated carbon. The odor mitigation layer

15 comprises discontinuities forming microchannels through the layer thickness. The polyurethane layer had approximately 1.2%wt TiO₂ and approximately 2.3%wt activated carbon.

Example 4

20 [0035] A thermoplastic non-woven layer was formed from staple polyester fibers having an average length of about 3 inches and an average denier of about 6. A polypropylene tape woven layer was laid onto the polypropylene fibers. The tape elements have a thickness of about 50 micrometers, a width of about 2 millimeters, and were woven in a plain weave. The fibers were entangled with each other and the woven tape layer by needle punching the

25 fibers through the woven layer. The non-woven layer was calendared on one side. The resultant non-woven layer had a weight of about 5.5 oz/yd².

Example 5

30 [0036] The non-woven layer of Example 4 was coated with a UV barrier layer. The UV barrier layer was a silicone layer having a thickness of about 67 micrometers. The silicone used was Elastosil® LR 6293 available from the

Wacker Silicone a division of Wacker-Chemie GmbH. The silicone layer had discontinuities forming microchannels through the layer thickness.

Example 6

[0037] The layers of Example 5 were coated with an odor mitigation layer, with the odor mitigation layer being coated on top of the silicone layer. The odor mitigation layer contained anatase TiO₂ and activated carbon in polyurethane. The polyurethane used was MILLITEX™ Resin MRX (a blocked isocyanate based cross-linking agent (with 35-45% solids) available from Milliken Chemical. The anatase TiO₂ had a surface area of about 40 m²/g and was obtained from DeGussa® P25 from Evonik Degussa in Germany. The activated carbon had an average particle size of 20 micrometers and was obtained as the GX230 grade supplied by PICA USA Inc. The polyurethane layer had approximately 1.2%wt TiO₂ and approximately 2.3%wt activated carbon. The odor mitigation layer comprises discontinuities forming microchannels through the layer thickness.

Results

[0038] The Examples were tested for performance comparing the compost cover of the invention to control samples and subcomponents of the compost cover.

	Air Flow Permeability (CFM/sqft)	VOC diffusion – Static
Ex. 1	98.4	248
Ex. 2	2.77	151
Ex. 3	10.08	73
Ex. 4	6.96	264
Ex. 5	7.50	128
Ex. 6	6.17	122

Table 1 – Air Permeability and VOC diffusion of examples

[0039] As one can observe from Table 1, the static VOC significantly improves (lower numbers mean less VOC permeates through the sample) by coating the non-woven layers (examples 1 and 4) with silicone (examples 2 and 5) and additionally the polyurethane with TiO₂ and activated carbon).

Samples	VOC content after exposure to Sunlight
Control (no added odor mitigating materials)	1365
TiO ₂ nanoparticle powder (no binder)	0
TiO ₂ nanoparticle powder in polyurethane	675
TiO ₂ nanoparticle powder and activated carbon in polyurethane	0

Table 2 – VOC mitigation of odor mitigating layer materials

[0040] Testing the components of the compost cover to mitigate odor in a UV setting as above proves that the anatase TiO₂ does reduce VOCs. When the TiO₂ is placed in a binder, in this case polyurethane, it is less effective: This is believed to occur because some of the surface of the TiO₂ is passivated by the binder system. Adding activated carbon to the mixture of TiO₂ and binder is very effective at reducing VOCs in the presence of UV light.

	Contact Angle (degrees)	Std. Dev
Ex. 2	101.9	5.3
Ex. 3	81.1	20.0
Ex. 4	Wets immediately (~ 0.0)	
Ex. 5	72.2	2.3
Ex. 6	56.1	37.6

Table 3 – Contact angle measurements on inner surface of examples

[0041] As one can see from the contact angle measurement data, water does not substantially wet out on the inner surface of the invention examples (3 and 6) trapping water droplets on the inner surface which in turn help control the odor of the composting material under the compost cover. It is believed that the addition of silicone to one outer side of the non-woven layer (comparing examples 4 and 5) causes the inner layer of the non-woven layer to become more hydrophobic.

[0042] All references, including publications, patent applications, and patents, cited herein are hereby incorporated by reference to the same extent as if each reference were individually and specifically indicated to be incorporated by reference and were set forth in its entirety herein.

5 [0043] The use of the terms “a” and “an” and “the” and similar referents in the context of describing the invention (especially in the context of the following claims) are to be construed to cover both the singular and the plural, unless otherwise indicated herein or clearly contradicted by context. The terms “comprising,” “having,” “including,” and “containing” are to be construed as open-
10 ended terms (i.e., meaning “including, but not limited to,”) unless otherwise noted. Recitation of ranges of values herein are merely intended to serve as a shorthand method of referring individually to each separate value falling within the range, unless otherwise indicated herein, and each separate value is incorporated into the specification as if it were individually recited herein. All
15 methods described herein can be performed in any suitable order unless otherwise indicated herein or otherwise clearly contradicted by context. The use of any and all examples, or exemplary language (e.g., “such as”) provided herein, is intended merely to better illuminate the invention and does not pose a limitation on the scope of the invention unless otherwise claimed. No language
20 in the specification should be construed as indicating any non-claimed element as essential to the practice of the invention.

[0044] Preferred embodiments of this invention are described herein, including the best mode known to the inventors for carrying out the invention. Variations of those preferred embodiments may become apparent to those of
25 ordinary skill in the art upon reading the foregoing description. The inventors expect skilled artisans to employ such variations as appropriate, and the inventors intend for the invention to be practiced otherwise than as specifically described herein. Accordingly, this invention includes all modifications and equivalents of the subject matter recited in the claims appended hereto as permitted by applicable law. Moreover, any combination of the above-described
30 elements in all possible variations thereof is encompassed by the invention unless otherwise indicated herein or otherwise clearly contradicted by context.

WHAT IS CLAIMED IS:

1. An odoriferous organic compound cover having an inner surface and an outer surface comprising:

5 a thermoplastic non-woven layer having a first side and a second side and comprising thermoplastic fibers, wherein the first side of the non-woven layer forms the inner side of the compost cover, and wherein the first side of the non-woven layer has a contact angle with water greater than about 70° and is non-wicking, non-wicking being defined as having a water vapor transmission of less
10 than 1000 g/m²/24 hours as measured by ASTM E-96 standard;

a UV barrier layer on the second side of the non-woven layer comprising a UV blocking polymer, wherein the UV blocking polymer blocks at least 50% of UV(C) energy, wherein the UV barrier layer has a thickness of between one to three times the diameter of the thermoplastic fibers of the non-woven layer, and
15 wherein the UV barrier layer comprises discontinuities forming microchannels though the layer thickness; and,

an odor mitigation layer on the UV barrier layer on the side of the barrier layer opposite the non-woven layer, wherein the odor mitigation layer comprises a UV translucent polymer, wherein the UV translucent polymer transmits at least
20 about 60% of UV(C) energy and anatase titanium dioxide, wherein the odor mitigation layer comprises discontinuities forming microchannels though the layer thickness, and wherein the odor mitigation layer forms the outer surface of the compost cover.

25 2. The odiferous organic compounds cover of claim 1, wherein the UV translucent polymer transmits at least about 90% of UV(C) energy.

3. The odiferous organic compounds cover of claim 1, wherein the first side of the non-woven layer is calendared and at least a portion of the thermoplastic
30 fibers are bonded to one another.

4. The odiferous organic compounds cover of claim 3, wherein the second side of the non-woven layer is calendared.

5 5. The odiferous organic compounds cover of claim 1, wherein the thermoplastic fibers comprise polypropylene fibers.

6. The odiferous organic compounds cover of claim 1, wherein the non-woven layer further comprises a woven layer embedded by the non-woven fibers.

10 7. The odiferous organic compounds cover of claim 1, wherein the first side of the non-woven layer has a contact angle with water greater than about 90°.

8. The odiferous organic compounds cover of claim 1, wherein the odor mitigation layer further comprises activated carbon.

15 9. The odiferous organic compounds cover of claim 1, wherein the UV blocking polymer comprises silicone.

20 10. The odiferous organic compounds cover of claim 1, wherein the UV translucent polymer comprises polyurethane.

11. An odiferous organic compounds containment system comprising:

a) decaying organic material;

25 b) a odiferous organic compounds cover oriented such that at least a portion of the decaying organic material is covered by the odiferous organic compounds cover, wherein the odiferous organic compounds cover comprises:

30 i) a thermoplastic non-woven layer having a first side and a second side and comprising thermoplastic fibers, wherein the first side of the non-woven layer forms the inner side of the cover, and wherein the first side of the non-woven layer has a contact angle with water greater than about 70° and is non-wicking, non-wicking being defined as having a water vapor transmission of less than 1000 g/m²/24 hours as measured by ASTM E-96 standard;

ii) a UV barrier layer on the second side of the non-woven layer comprising a UV blocking polymer, wherein the UV blocking polymer blocks at least 50% of UV(C) energy, wherein the UV barrier layer has a thickness of between one to three times the diameter of the thermoplastic fibers of the non-woven layer, and wherein the UV barrier layer comprises discontinuities forming microchannels through the layer thickness; and,

iii) an odor mitigation layer on the UV barrier layer on the side of the barrier layer opposite the non-woven layer, wherein the odor mitigation layer comprises a UV translucent polymer wherein the UV translucent polymer transmits at least about 60% of UV(C) energy and anatase titanium dioxide, wherein the odor mitigation layer comprises discontinuities forming microchannels through the layer thickness, and wherein the odor mitigation layer forms the outer surface of the cover.

12. The odiferous organic compounds containment system of claim 11, wherein the non-woven layer further comprises a woven layer embedded by the non-woven fibers.

13. The odiferous organic compounds containment system of claim 11, wherein the odor mitigation layer further comprises activated carbon.

14. The odiferous organic compounds containment system of claim 11, wherein the UV blocking polymer comprises silicone.

15. The odiferous organic compounds containment system of claim 11, wherein the UV translucent polymer comprises polyurethane.

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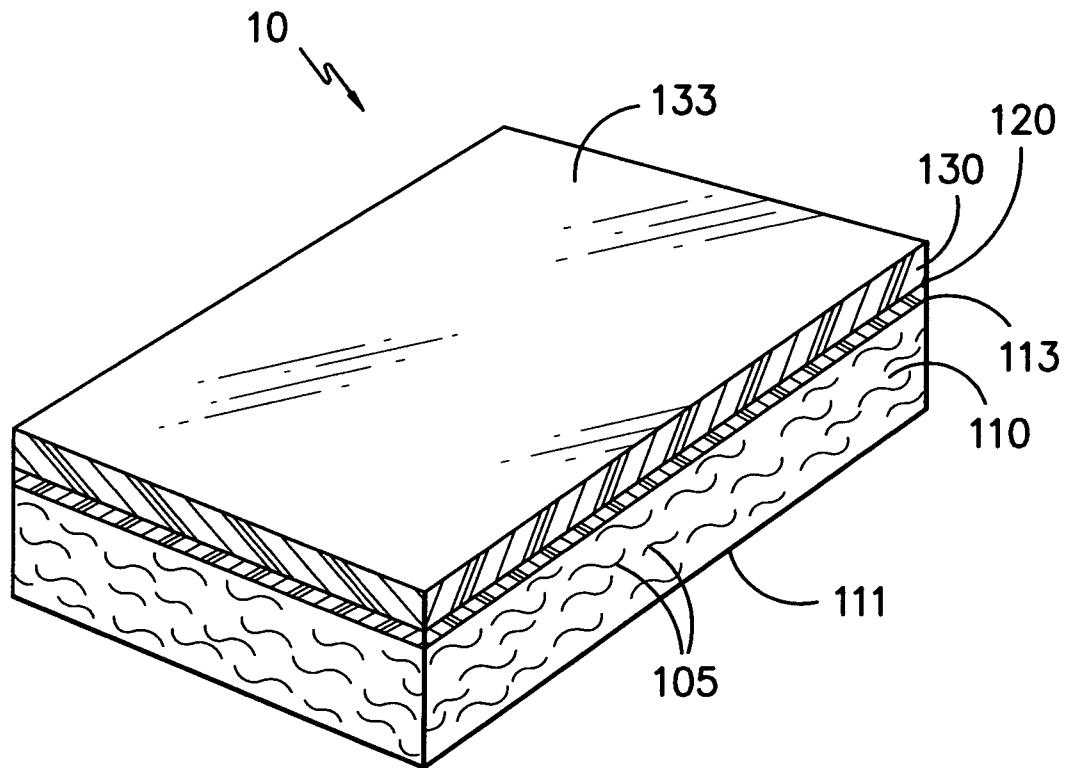
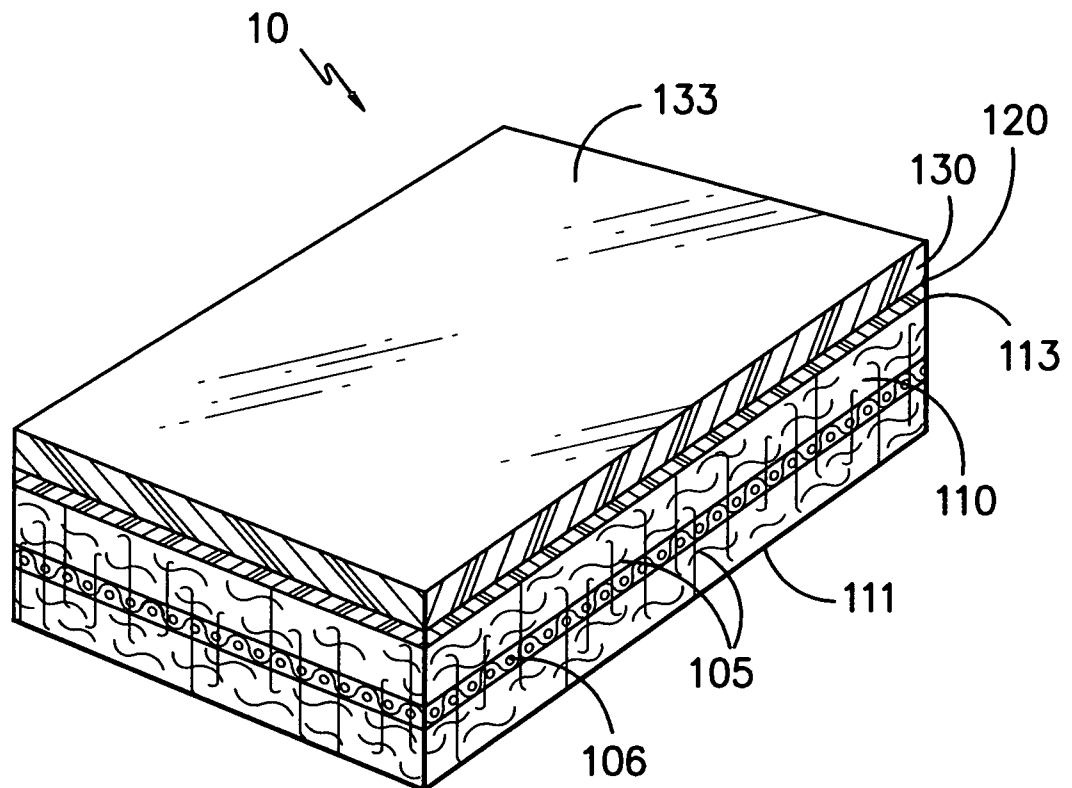


FIG. -1-

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*FIG. -2-*

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2009/004564

A. CLASSIFICATION OF SUBJECT MATTER

INV. B32B5/22 B32B27/08 B32B27/12 B32B27/20 B32B27/40
C05F17/02 D06M15/643

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

B32B C05F D06M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	DE 42 31 414 A1 (PLOUCQUET C F GMBH [DE]) 24 March 1994 (1994-03-24) column 1, line 16 - column 2, line 37 -----	1-15
Y	US 2008/011034 A1 (HOCHREIN KILIAN P [DE] ET AL) 17 January 2008 (2008-01-17) paragraphs [0002], [0007], [0009] - [0016] -----	1-15
Y	DE 298 16 749 U1 (SANDLER HELMUT HELSA WERKE [DE]) 4 February 1999 (1999-02-04) page 2, paragraph 3 - page 4, paragraph 1; claims 1-10 -----	1-15
Y	US 6 592 858 B1 (HONDA HIDENOBU [JP] ET AL) 15 July 2003 (2003-07-15) column 1, line 17 - line 36 ----- -/--	1-15

☒ Further documents are listed in the continuation of Box C.

☒ See patent family annex.

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Date of the actual completion of the international search

13 November 2009

Date of mailing of the international search report

03/12/2009

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PCT/US2009/004564

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5 872 072 A (MOURI MOTOYA [JP] ET AL) 16 February 1999 (1999-02-16) column 1, line 12 - line 16 column 2, line 7 - line 48 -----	1-15
T	ZHANG C H ET AL: "Preparation and characterization of hydrophilic modification of polypropylene non-woven fabric by dip-coating PVA (polyvinyl alcohol)" SEPARATION AND PURIFICATION TECHNOLOGY, ELSEVIER SCIENCE, AMSTERDAM, NL, vol. 61, no. 3, 15 July 2008 (2008-07-15), pages 276-286, XP022689164 ISSN: 1383-5866 [retrieved on 2007-11-09] page 276, column 2, paragraph 3; table 2 -----	1-15
T	VINOD M P ET AL: "SILICON BASED LIGHT EMITTING GELS" APPLIED PHYSICS LETTERS, AIP, AMERICAN INSTITUTE OF PHYSICS, MELVILLE, NY, US, vol. 68, no. 1, 1 January 1996 (1996-01-01), pages 81-83, XP000548797 ISSN: 0003-6951 page 82, column 1, paragraph 2 -----	1-15
T	R. THIRUVENKATACHARI, S. VIGNESWARAN & I.S.MOON: "A review on UV/TiO2 photocatalytic oxidation process" KOREAN J. CHEM ENG., vol. 25, no. 1, January 2008 (2008-01), pages 64-72, XP002555240 page 65, column 1, paragraph 2 - column 2, paragraph 4 -----	1-15
T	YACOBI B G ET AL: "Adhesive bonding in microelectronics and photonics" JOURNAL OF APPLIED PHYSICS, AMERICAN INSTITUTE OF PHYSICS, NEW YORK, US, vol. 91, no. 10, 15 May 2002 (2002-05-15), pages 6227-6262, XP012054553 ISSN: 0021-8979 page 6239, column 2, line 4 - line 6 -----	1-15

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/US2009/004564

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
DE 4231414	A1	24-03-1994	AT 402713 B NL 9301411 A	25-08-1997 18-04-1994
US 2008011034	A1	17-01-2008	AT 334816 T AU 778669 B2 AU 7548300 A CA 2384870 A1 DE 60029804 T2 EP 1216145 A1 ES 2269181 T3 WO 0121394 A1 JP 2003509212 T NZ 517950 A	15-08-2006 16-12-2004 24-04-2001 29-03-2001 08-02-2007 26-06-2002 01-04-2007 29-03-2001 11-03-2003 30-05-2003
DE 29816749	U1	04-02-1999	NONE	
US 6592858	B1	15-07-2003	AT 267907 T CA 2365600 A1 CN 1345390 A DE 60011079 D1 DE 60011079 T2 EP 1188854 A1 ES 2220391 T3 WO 0155498 A1	15-06-2004 02-08-2001 17-04-2002 01-07-2004 16-06-2005 20-03-2002 16-12-2004 02-08-2001
US 5872072	A	16-02-1999	AT 187095 T CN 1133203 A DE 69513625 D1 DE 69513625 T2 EP 0719579 A1	15-12-1999 16-10-1996 05-01-2000 20-04-2000 03-07-1996