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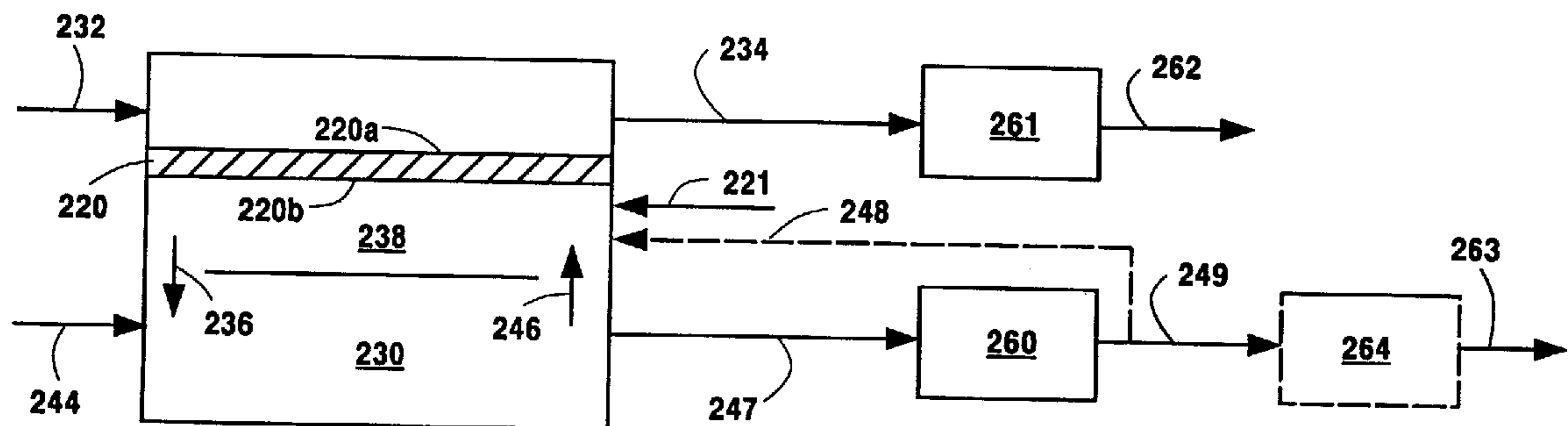
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(54) **PROCEDE DE COMBUSTION ENRICHEE UTILISANT DES
SYSTEMES DE CONDUCTEURS IONIQUES A
ELECTROLYTES SOLIDES**

(54) **PROCESS FOR ENRICHED COMBUSTION USING SOLID
ELECTROLYTE IONIC CONDUCTOR SYSTEMS**



(57) A combustion or partial oxidation process for using an oxidant with a low nitrogen concentration. An oxygen-containing gas is introduced into an ion transport module including an ion transport membrane having a retentate side with a first pressure and a permeate side with a second pressure to separate a purified oxygen gas stream on the permeate side and correspondingly depleting the oxygen on the retentate side to produce the oxygen-depleted gas stream. A purge gas stream containing a low concentration of nitrogen is passed to the permeate side to form an oxidant stream containing less than about 40% oxygen. The oxidant stream and a fuel are then introduced into a combustion or reaction chamber to produce heat and products of combustion or partial oxidation.



D-20320-1

- 44 -

PROCESS FOR ENRICHED COMBUSTION USING
SOLID ELECTROLYTE IONIC CONDUCTOR SYSTEMS

Abstract of the Invention

A combustion or partial oxidation process for using an oxidant with a low nitrogen concentration. An oxygen-containing gas is introduced into an ion transport module including an ion transport membrane having a retentate side with a first pressure and a permeate side with a second pressure to separate a purified oxygen gas stream on the permeate side and correspondingly depleting the oxygen on the retentate side to produce the oxygen-depleted gas stream. A purge gas stream containing a low concentration of nitrogen is passed to the permeate side to form an oxidant stream containing less than about 40% oxygen. The oxidant stream and a fuel are then introduced into a combustion or reaction chamber to produce heat and products of combustion or partial oxidation.

D-20320-1

- 1 -

PROCESS FOR ENRICHED COMBUSTION USING
SOLID ELECTROLYTE IONIC CONDUCTOR SYSTEMS

CROSS REFERENCE TO RELATED APPLICATIONS

5 This is a continuation-in-part application of U.S. patent application Serial No. 08/868,962, filed June 5, 1997.

FIELD OF THE INVENTION

10 The invention relates to the integration of oxygen enhanced combustion with oxygen separation processes that employ solid electrolyte ionic conductor membranes, and more particularly, to the integration of these processes to improve the economics, efficiency
15 and pollution-related problems of combustion processes using a dilute oxygen stream as the oxidant.

BACKGROUND OF THE INVENTION

20 Many different oxygen separation systems, for example, organic polymer membrane systems, have been used to separate selected gases from air and other gas mixtures. Air is a mixture of gases which may contain varying amounts of water vapor and, at sea level, has the following approximate composition by volume: oxygen
25 (20.9%), nitrogen (78%), argon (0.94%), with the balance consisting of other trace gases. An entirely different type of membrane, however, can be made from certain inorganic oxides. These solid electrolyte membranes are made from inorganic oxides, typified by
30 calcium- or yttrium-stabilized zirconium and analogous oxides having a fluorite or perovskite structure.

 Some of these solid oxides have the ability to conduct oxygen ions at elevated temperatures if an electric potential is applied across the membrane, that
35 is, they are electrically-driven or ionic conductors

D-20320-1

- 2 -

only. Recent research has led to the development of solid oxides which have the ability to conduct oxygen ions at elevated temperatures if a chemical driving potential is applied. These pressure-driven ionic
5 conductors or mixed conductors may be used as membranes for the extraction of oxygen from oxygen-containing gas streams if a sufficient partial oxygen pressure ratio is applied to provide the chemical driving potential. Since the selectivity of these materials for oxygen is
10 infinite and oxygen fluxes generally several orders of magnitude higher than for conventional membranes can be obtained, attractive opportunities are created for the production of oxygen using these ion transport membranes.

15 Although the potential for these oxide ceramic materials as gas separation membranes is great, there are certain problems in their use. The most obvious difficulty is that all of the known oxide ceramic materials exhibit appreciable oxygen ion conductivity
20 only at elevated temperatures. They usually must be operated well above 500°C, generally in the 600°C-900°C range. This limitation remains despite much research to find materials that work well at lower temperatures. Solid electrolyte ionic conductor technology is
25 described in more detail in Prasad et al., U.S. Patent No. 5,547,494, entitled *Staged Electrolyte Membrane*, which is hereby incorporated by reference to more fully describe the state of the art.

Combustion processes, however, usually operate at
30 high temperature and therefore there is the potential for efficiently integrating ion transport systems with oxygen enhanced combustion processes and the present invention involves novel schemes for the integration of ion transport systems with oxygen enhanced combustion
35 processes.

D-20320-1

- 3 -

Most conventional combustion processes use the most convenient and abundant source of oxygen, that is, air. The presence of nitrogen in air does not benefit the combustion process and, on the contrary, may create many problems. For example, nitrogen reacts with oxygen at combustion temperatures, forming nitrogen oxides (NO_x), an undesirable pollutant. In many instances, the products of combustion must be treated to reduce nitrogen oxide emissions below environmentally acceptable limits. Moreover, the presence of nitrogen increases the flue gas volume which in turn increases the heat losses in the flue gas and decreases the thermal efficiency of the combustion process. To minimize these problems, oxygen-enriched combustion (OEC) has been commercially practiced for many years. There are several benefits of oxygen-enriched combustion including reduced emissions (particularly nitrogen oxides), increased energy efficiency, reduced flue gas volume, cleaner and more stable combustion, and the potential for increased thermodynamic efficiency in downstream cycles. These benefits of OEC, however, must be weighed against the cost of the oxygen that has to be manufactured for this application. As a consequence, the market for OEC is greatly dependent on the cost of producing oxygen-enriched gas. It has been estimated that as much as 100,000 tons per day of oxygen would be required for the new markets in OEC if the cost of oxygen-enriched gas could be reduced to about \$15/ton. It appears that gas separation processes employing ion transport membranes have the promise of reaching that goal. OEC is discussed in detail in H. Kobayashi, Oxygen Enriched Combustion System Performance Study, Vol. 1: Technical and Economic Analysis (Report #DOE/ID/12597), 1986, and Vol. 2: Market Assessment (Report #DOE/ID/12597-3),

D-20320-1

- 4 -

1987, Union Carbide Company-Linde Division, Reports for the U.S. Dept. of Energy, Washington, D.C.).

Literature related to ion transport conductor technology for use in separating oxygen from gas stream
5 include:

Hegarty, U.S. Patent No. 4,545,787, entitled *Process for Producing By-Product Oxygen from Turbine Power Generation*, relates to a method of generating power from an compressed and heated air stream by
10 removing oxygen from the air stream, combusting a portion of the resultant air stream with a fuel stream, combining the combustion effluent with another portion of the resultant air stream, and expanding the final combustion product through a turbine to generate power.
15 Hegarty mentions the use of silver composite membranes and composite metal oxide solid electrolyte membranes for removing oxygen from the air stream.

Kang et al., U.S. Patent No. 5,516,359, entitled *Integrated High Temperature Method for Oxygen
20 Production*, relates to a process for separating oxygen from heated and compressed air using a solid electrolyte ionic conductor membrane where the nonpermeate product is heated further and passed through a turbine for power generation.

25 Mazanec et al., U.S. Patent No. 5,160,713, entitled, *Process for Separating Oxygen from an Oxygen-Containing Gas by Using a Bi-containing Mixed Metal Oxide Membrane*, discloses bismuth-containing materials that can be used as oxygen ion conductors.

30 Publications related to oxygen-enriched or enhanced combustion (OEC) include the above-mentioned U.S. Dept. of Energy reports authored by H. Kobayashi and H. Kobayashi, J.G. Boyle, J.G. Keller, J.B. Patton and R.C. Jain, *Technical and Economic Evaluation of
35 Oxygen Enriched Combustion Systems for Industrial*

D-20320-1

- 5 -

Furnace Applications, in Proceedings of the 1986 Symposium on Industrial Combustion Technologies, Chicago, IL, April 29-30, 1986, ed. M.A. Lukasiewics, American Society for Metals, Metals Park, OH, which
5 discusses the various technical and economic aspects of oxygen-enhanced combustion systems.

Oxygen-enriched combustion has been commercially practiced using oxygen manufactured by either cryogenic distillation or noncryogenic processes such as pressure
10 swing adsorption (PSA). All of these processes operate at or below 100°C and therefore are difficult to thermally integrate with combustion processes.

When the boiler of a steam power plant is fired with oxygen and fuel, the power required to separate
15 air in the state-of-the-art cryogenic plant is very significant and consumes about 16% of the total power generated from the single cycle steam boiler power plant. The compression of air required for air separation is the primary source of this power
20 requirement.

Oxygen is too expensive to use for most boiler applications. In a typical air-fuel fired boiler operation, air is fed at a pressure of several inches of H₂O into the combustion chamber that operates at
25 about the atmospheric pressure. Compressing air to a low pressure of even a few psig is considered too costly due to the increased power requirement for compression and a consequent loss of power generation efficiency.

30 One practical problem of using ceramic membranes is the lack of control resulting from leakage at ceramic joints and through cracks in the ceramic membrane tubes. Ceramic materials are susceptible in developing stress cracks when used at high
35 temperatures, and especially under changing temperature

D-20320-1

- 6 -

conditions. Therefore, it is highly desirable to develop a robust ceramic membrane system that can continue to operate efficiently and effectively despite ceramic membrane tube cracks due to thermal and
5 mechanical stress.

OBJECTS OF THE INVENTION

It is therefore an object of the invention to provide an oxygen enriched combustion or partial
10 oxidation process in which air is fed into the ceramic membrane module at near atmospheric pressure and require substantially less electric power than the present conventional practice.

It is another object of the invention to minimize
15 NO_x formation and thermal losses due to heating of nitrogen gas in the combustion process.

It is yet another object of the invention to recover nitrogen-rich gas stream from the ion transport membrane module to be used as a co-product.

20 It is another object of the invention to produce carbon dioxides-rich flue gas stream for recovery.

SUMMARY OF THE INVENTION

The invention comprises a combustion or partial
25 oxidation process for using an oxidant with a low nitrogen concentration. An oxygen-containing gas is introduced into an ion transport module including an ion transport membrane having a retentate side with a first pressure and a permeate side with a second
30 pressure to separate a purified oxygen gas stream on the permeate side and correspondingly depleting the oxygen on the retentate side to produce the oxygen-depleted gas stream. A purge gas stream containing a low concentration of nitrogen is passed to the permeate
35 side to form an oxidant stream containing less than

D-20320-1

- 7 -

about 80% oxygen. The oxidant stream and a fuel are then introduced into a combustion or reaction chamber to produce heat and products of combustion or partial oxidation.

5 In a preferred embodiment, air is used as the oxygen-containing gas. The ratio of the first pressure to the second pressure is less than 4.78, preferably between 0.5 to 4.0, more preferably between 0.8 to 2.0, and most preferably between 0.9 and 1.5. The oxidant
10 stream comprises between 1% to 40% oxygen, more preferably between 2% to 15% oxygen, and most preferably between 3% to 10% oxygen. The purge gas comprises less than 10% nitrogen, preferably less than 5% nitrogen. The temperature of the oxidant stream is
15 preferably above 500°C and the oxidant stream is introduced into the combustion or reaction chamber without substantial cooling.

BRIEF DESCRIPTION OF THE DRAWINGS

20 Other objects, features and advantages of the invention will occur to those skilled in the art from the following description of preferred embodiments and the accompanying drawings, in which:

Fig. 1 is a schematic diagram showing the
25 integration of ion transport oxygen production with oxygen-enriched combustion and a downstream process;

Fig. 2 is a schematic diagram showing the integration of ion transport oxygen production with oxygen-enriched combustion and a downstream process
30 similar to Fig. 1;

Fig. 3 is a schematic diagram similar to Fig. 2 where the combustor is integrated with the ion transport module;

Fig. 4 is a schematic diagram showing how the ion
35 transport process, combustor, and downstream process

D-20320-1

- 8 -

are integrated in a single module;

Fig. 5 is a schematic diagram showing an embodiment of the ion transport process, combustor, and downstream process integrated in a single module; and

5 Fig. 6 is a schematic diagram showing an embodiment of two separate ion transport membrane processes in which no flue gas recirculation is used in one of the processes.

10 DETAILED DESCRIPTION OF THE INVENTION

The invention will now be described in detail with reference to the figures in which like reference numerals are used to indicate like elements.

The present invention discloses process
15 configurations that enable the economically attractive integration of ion transport oxygen production with oxygen-enriched combustion (OEC). Although pressure-driven processes for ion transport membranes are preferred because of the simplicity of their design,
20 the concepts described herein are applicable to membrane systems utilizing either an ionic-conductor-only membrane having electrodes and an external circuit for electron return or a mixed conductor membrane.

Current commercial oxygen production processes
25 typically operate at temperatures below 100°C. Because of this low temperature, they do not gain significant efficiencies by integration with an OEC process. The elevated temperatures of operation (usually greater than 600°C) makes ion transport process intrinsically
30 well suited for integration with high temperature processes, such as combustion, that use oxygen. In addition, it will be shown that the combustion exhaust flue gases can be beneficially used to enhance ion transport membrane performance. Traditional oxygen
35 production processes (for example, PSA, TSA or

D-20320-1

- 9 -

membrane-based processes) cannot easily take advantage of the exhaust flue gases because of their high temperature when they leave the combustion chamber.

The essence of the current process configuration is an ion transport membrane that employs a solid oxygen ion conducting or mixed conducting membrane to separate oxygen from an oxygen-containing gas, typically, but not necessarily, air, and to utilize the oxygen separated in a downstream process including, but not limited to, oxygen-enriched combustion. To reduce the partial pressure of oxygen on the permeate side in the ion transport membrane, an oxygen-depleted gas (for example, waste gases from the combustion process or any downstream process) is used as a purge gas stream. Such purging greatly enhances the driving force across the ion transport membrane and effects a high oxygen flux and a lower membrane area requirement. These benefits are accrued even when the feed gas stream is at a relatively low pressure, thereby reducing the power requirements of the system to that of practical interest. Recirculation of the combustion exhaust gas is also beneficial because it provides a diluent stream that is important for controlling the temperature in the combustor and minimizing NO_x formation (for example, from nitrogen contained in infiltrating ambient air). Efficiency of this process could also be enhanced by adding fuel to the flue gas entering the oxygen separator. This further reduces the partial pressure oxygen on the permeate side, resulting in even higher oxygen fluxes in the ion transport separator. In some embodiments of the invention, the ion transport module can also function as the combustor, thereby eliminating the need for a separate combustor, unless the application requires a gas stream exiting the combustor at a temperature above 1100°C, the maximum

D-20320-1

- 10 -

operating temperature of many current ion transport membranes. It should be noted that the heat necessary to maintain the temperature of the ion transport module within the operating range can come from a variety of sources known to those skilled in the art, including, for example, heat generated in an afterburner and recirculated hot combustion product gases, among others.

In most mixed conductors, the electronic conductivity greatly exceeds the oxygen ion conductivity at the operating temperatures of interest, and the overall transport of oxygen from one side to the other is controlled by the oxygen ion conductivity. A number of potential mixed conductors have been identified in both the fluorite and perovskite crystal structures. The behavior of ion transport membranes has been extensively studied (for example, for fuel cells) and can be accurately modeled. Table 1 is a partial list of mixed conductors of interest for oxygen separation.

Table I

	Material composition
1.	$(\text{La}_{1-x}\text{Sr}_x)(\text{Co}_{1-y}\text{Fe}_y)\text{O}_{3-\delta}$ ($0 \leq x \leq 1$, $0 \leq y \leq 1$, δ from stoichiometry)
2.	$\text{SrMnO}_{3-\delta}$ $\text{SrMn}_{1-x}\text{Co}_x\text{O}_{3-\delta}$ ($0 \leq x \leq 1$, $0 \leq y \leq 1$, δ from stoichiometry) $\text{Sr}_{1-x}\text{Na}_x\text{MnO}_{3-\delta}$
3.	$\text{BaFe}_{0.5}\text{Co}_{0.5}\text{YO}_3$ SrCeO_3 $\text{YBa}_2\text{Cu}_3\text{O}_{7-\beta}$ ($0 \leq \beta \leq 1$, β from stoichiometry)
4.	$\text{La}_{0.2}\text{Ba}_{0.8}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{2.6}$; $\text{Pr}_{0.2}\text{Ba}_{0.8}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{2.6}$
5.	$\text{A}_x\text{A}'_x\text{A}''_x\text{B}_y\text{B}'_y\text{B}''_y\text{O}_{3-z}$ (x, x', x'', y, y', y'' all in 0-1 range) where: A, A', A'' = from groups 1, 2, 3 and f-block lanthanides B, B', B'' = from d-block transition metals

D-20320-1

- 11 -

6.	(a) Co-La-Bi type:	Cobalt oxide	15-75 mole %
		Lanthanum oxide	13-45 mole %
	(b) Co-Sr-Ce type:	Bismuth oxide	17-50 mole %
		Cobalt oxide	15-40 mole %
		Strontium oxide	40-55 mole %
	(c) Co-Sr-Bi type:	Cerium oxide	15-40 mole %
		Cobalt oxide	10-40 mole %
		Strontium oxide	5-50 mole %
	(d) Co-La-Ce type:	Bismuth oxide	35-70 mole %
		Cobalt oxide	10-40 mole %
		Lanthanum oxide	10-40 mole %
	(e) Co-La-Sr-Bi type:	Cerium oxide	30-70 mole %
		Cobalt oxide	15-70 mole %
		Lanthanum oxide	1-40 mole %
	(f) Co-La-Sr-Ce type:	Strontium oxide	1-40 mole %
		Bismuth oxide	25-50 mole %
		Cobalt oxide	10-40 mole %
		Lanthanum oxide	1-35 mole %
		Strontium oxide	1-35 mole %
		Cerium oxide	0-70 mole %
7.	$\text{Bi}_{2-x-y}\text{M}'_x\text{M}_y\text{O}_{3-\delta}$ ($0 \leq x \leq 1$, $0 \leq y \leq 1$, δ from stoichiometry) where: $\text{M}' = \text{Er, Y, Tm, Yb, Tb, Lu, Nd, Sm, Dy, Sr, Hf, Th, Ta, Nb, Pb, Sn, In, Ca, Sr, La}$ and mixtures thereof $\text{M} = \text{Mn, Fe, Co, Ni, Cu}$ and mixtures thereof		
8.	$\text{BaCe}_{1-x}\text{Gd}_x\text{O}_{3-x/2}$, where x equals from zero to about 1.		
9.	One of the materials of $\text{A}_s\text{A}'_t\text{B}_u\text{B}'_v\text{B}''_w\text{O}_x$ family whose composition is disclosed in U.S. Patent 5,306,411 (Mazanec <i>et al.</i>) as follows: A represents a lanthanide or Y, or a mixture thereof; A' represents an alkaline earth metal or a mixture thereof; B represents Fe; B' represents Cr or Ti, or a mixture thereof; B'' represents Mn, Co, V, Ni or Cu, or a mixture thereof; and s, t, u, v, w, and x are numbers such that: s/t equals from about 0.01 to about 100; u equals from about 0.01 to about 1; v equals from zero to about 1; w equals from zero to about 1; x equals a number that satisfies the valences of the A, A', B, B', B'' in the formula; and $0.9 < (s+t)/(u+v+w) < 1.1$		
10.	One of the materials of $\text{La}_{1-x}\text{Sr}_x\text{Cu}_{1-y}\text{M}_y\text{O}_{3-\delta}$ family, where: M represents Fe or Co; x equals from zero to about 1; y equals from zero to about 1; δ equals a number that satisfies the valences of La, Sr, Cu, and M in the formula.		
11.	One of the materials of $\text{Ce}_{1-x}\text{A}_x\text{O}_{2-\delta}$ family, where: A represents a lanthanide, Ru, or Y; or a mixture thereof; x equals from zero to about 1; y equals from zero to about 1; δ equals a number that satisfies the valences of Ce and A in the formula.		

D-20320-1

- 12 -

12.	One of the materials of $\text{Sr}_{1-x}\text{Bi}_x\text{FeO}_{3-\delta}$ family, where: A represents a lanthanide or Y, or a mixture thereof; x equals from zero to about 1; y equals from zero to about 1; δ equals a number that satisfies the valences of Ce and A in the formula.
13.	One of the materials of $\text{Sr}_x\text{Fe}_y\text{Co}_z\text{O}_w$ family, where: x equals from zero to about 1; y equals from zero to about 1; z equals from zero to about 1; w equals a number that satisfies the valences of Sr, Fe and Co in the formula.
14.	Dual phase mixed conductors (electronic/ionic): $(\text{Pd})_{0.5}/(\text{YSZ})_{0.5}$ $(\text{Pt})_{0.5}/(\text{YSZ})_{0.5}$ $(\text{B-MgLaCrO}_x)_{0.5}/(\text{YSZ})_{0.5}$ $(\text{In}_{90}\%\text{Pt}_{10}\%)_{0.6}/(\text{YSZ})_{0.5}$ $(\text{In}_{90}\%\text{Pt}_{10}\%)_{0.5}/(\text{YSZ})_{0.5}$ $(\text{In}_{95}\%\text{Pr}_{2.5}\%\text{Zr}_{2.5}\%)_{0.5}/(\text{YSZ})_{0.5}$ Any of the materials described in 1-13, to which a high temperature metallic phase (e.g., Pd, Pt, Ag, Au, Ti, Ta, W) is added.

Fig. 1 is a schematic diagram showing the integration of ion transport oxygen production with oxygen-enriched combustion. During operation, feed gas stream 1 containing elemental oxygen, usually air, is compressed to a relatively low pressure in blower or compressor 2 to produce compressed feed gas stream 3 which is heated in heat exchanger 33 against waste gas stream 31 and product nitrogen gas stream 37 to produce warmed feed gas stream 4. Gas stream 28 may be divided from warmed feed gas stream 4 and used in optional afterburner 26 to leave feed gas stream 5 which is optionally heated in heater 34 to produce hot feed gas stream 6. Hot feed gas stream 6 then enters the feed side of ion transport module 35 employing ion transport membrane 7 having a retentate side 7a and a permeate side 7b. A portion of the oxygen in hot feed gas stream 6 is removed in ion transport module 35 and the exiting gas stream 8 becomes nitrogen-enriched with respect to feed gas stream 1. The permeate side 7b of ion transport membrane 7 is purged using purge gas stream 9 containing products of combustion. Permeate

D-20320-1

- 13 -

gas stream 10 contains oxygen and this gas stream 10 is later mixed with fuel gas stream 11. Air, oxygen or oxygen enriched air stream 12 can be optionally added to gas stream 10.

5 Combustible gas stream 13, after passing through an optional blower (not shown), then enters combustor 14. Optionally or in addition to or in place of fuel gas stream 11, fuel gas stream 15 can be fed directly to combustor or furnace 14 to generate heat and
10 transfer heat to a furnace charge or to heat transfer surfaces. By operating combustor 14 close to stoichiometric or slightly fuel-rich condition, the oxygen concentration in exhaust gas stream 16 can be maintained at low levels. In this embodiment exhaust
15 gas stream 16 from combustor 14 is split into two portions, gas stream 17 and gas stream 18. Gas stream 18 is used optionally in downstream process 19 that requires heat input and the relatively cooler exhaust gas stream 20 from downstream process 19 may also be
20 split into two portions, exhaust gas stream 21 and exhaust gas stream 22. Fuel gas stream 25 may be added to exhaust gas stream 21 to produce gas stream 38.

Gas stream 38 can be added to gas stream 17 to produce gas stream 9 which enters ion transport module
25 35 and is used to purge the permeate side 17b of ion transport membrane 7. Although not shown herein, gas stream 17 or combustor 14 can be used to heat feed air 3 and/or warmed feed gas stream 5 by heat exchange to produce hot feed gas stream 6 rather than using heat
30 exchanger 33 and/or optional heater 34. Exhaust gas stream 22 optionally fed into an optional afterburner 26 where air stream 27 or gas stream 28 is optionally added to produce hot waste gas stream 29. Hot waste gas stream 29 may become gas stream 30 or gas stream
35 31. As mentioned above, gas stream 31 is used in heat

D-20320-1

- 14 -

exchanger **33** to heat compressed feed gas stream **3** to produce waste gas stream **32**. Gas stream **30** may be mixed with nitrogen-rich retentate gas stream **8** if nitrogen is not to be used as a co-product and if the temperature of exhaust gas stream **30** is adequately high. When Retentate gas stream **8** is at a higher pressure than exhaust gas stream **30** and it may be necessary to release the excess pressure of retentate gas stream **8** using expansion valve **23** to produce retentate gas stream **34** before it is mixed with gas stream **30**. When the pressure of stream **8** exceeds 2 atm, it is advantageous to replace expansion valve **23** with an expansion turbine to extract shaft work or to generate power. If retentate gas stream **24** is desired as a nitrogen-rich product gas stream, gas streams **36** and **30** are not mixed.

The use of an oxygen-depleted purge gas stream **9** in ion transport module **35** will greatly lower the oxygen partial pressure on the permeate side **7b** of ion transport membrane **7** and enable rapid oxygen transport through membrane **7**. Fuel gas streams **11**, **15**, and **25** can be introduced into the process configuration at any or all of the points shown in Fig. 1 to obtain the benefits of the invention; the use of at least one fuel gas stream is essential for the invention. For example, it may be desirable to add fuel gas stream **25** upstream of ion transport module **35** to greatly reduce the oxygen partial pressure on the permeate side **7b** of ion transport membrane **7**. This would also result in some heat generation in ion transport module **35** due to fuel combustion, thereby offsetting some of the heating requirements of the oxygen transport process. In this case, the exiting nitrogen-rich gas stream **8** from ion transport module **35** could be made hotter. This would make the heat transfer in heat exchanger **33** more

D-20320-1

- 15 -

efficient, thereby reducing the area required for the heat exchange and potentially eliminating the need for heater **34** upstream of ion transport module **35**. If enough fuel can be combusted in ion transport module **35** on the purge or permeate side **7b** of ion transport membrane **7**, it may altogether eliminate the need for a separate combustor **14**, that is, ion transport module **35** would also serve as the combustor (as described in Fig. 3). In such a situation, significant system simplification and cost reduction may result.

Reactive purge arrangements are disclosed in "Reactive Purge for Solid Electrolyte Membrane Gas Separation", U.S. Serial No. 08/567,699, filed December 5, 1995 and incorporated herein by reference. Preferred configuration for ion transport modules utilizing a reactive purge are disclosed in "Solid Electrolyte Ionic Conductor Reactor Design", U.S. Serial No. 08/848,204, filed April 29, 1997 and also incorporated herein by reference. Both applications are commonly owned with the present application.

It may be advantageous to operate combustor **14** with a slightly fuel-rich mixture because this will lead to partial oxidation of the fuel added to permeate gas stream **10**, resulting in an exhaust gas stream **16** containing hydrogen gas and carbon monoxide. As mentioned above, gas stream **17** is optionally used to purge the permeate side **7b** of ion transport membrane **7**. It should be noted that hydrogen gas is a highly reducing gas with a higher reactivity than many other gaseous fuels, and its presence in ion transport module **35** will result in an extremely low oxygen partial pressure on the purge side **7b** of ion transport membrane **7** and this will enable an even more rapid transport of oxygen through ion transport membrane **7**. Of course, similar results could be achieved by introducing

D-20320-1

- 16 -

hydrogen gas as fuel gas stream 25, however, it will not be cost-effective as the fuel-rich feed to combustor 14, since hydrogen gas is a relatively expensive fuel. The use of a fuel-rich feed to combustor 14 as described obviates the need for a use of a pre-produced hydrogen gas, since hydrogen gas is produced as a part of the process cycle. Running combustor 14 in a fuel-rich condition, however, could cause exhaust gas streams 18 and 22 to contain carbon monoxide and hydrogen gas, both of which can be simply vented to the atmosphere if the concentration is low. As mentioned above, it may be possible, however, to install afterburner 26 (perhaps catalytic) to which excess air 27 is added to burn off the carbon monoxide and hydrogen gas if their concentration is sufficiently high. Gas stream 28 of warmed feed gas stream 4 could also be added to afterburner 26 to provide for the afterburner requirements.

It is interesting to note that virtue of the recirculation of combustion products as purge gas stream 9, and because of the infinite selectivity of ion transport membrane 7 for oxygen, it is possible to limit the temperature rise of gas stream 13 in combustor 14 without the need for excess air and thereby exclude nitrogen from the combustion process, which substantially eliminates NO_x formation. This synergistic effect is a general principle of the invention and is a feature of many of the embodiments of the invention.

Typical ranges for operating parameters of the ion transport module used in the invention are as follows. Temperature: Typically in the 400-1000°C range, and preferably in the 400-800°C range. Pressure: The purge side pressure will be typically in 1-3 atm. range. The feed side pressure will be 0.8-3

D-20320-1

- 17 -

atm. if nitrogen is not a co-product, and 1-20 atm. if nitrogen is a co-product.

Oxygen ion conductivity (μ_i) of the Ion Transport

5 Membrane: Typically in the 0.01-100 S/cm range (1 S = 1/ohm).

Thickness of the Ion Transport Membrane: The ion transport membrane can be employed in the form of a dense film, or a thin film supported on a porous substrate. The thickness (t) of the ion transport
10 membrane/layer will be typically less than 5000 microns; preferably less than 1000 microns, and most preferably less than 100 microns.

Configuration: The ion transport membrane elements may be tubular or planar.

15 As mentioned above, asymmetric or composite ion transport membranes (that is, pressure-driven membranes) are used in the examples discussed herein. The following properties are based on typical values reported in the literature for such membranes as could
20 be used in the present invention.

Effective membrane thickness: 20 micron

Ionic conductivity, μ_i : 0.5 S/cm

Operating temperature: 800°C

25 Substrate porosity: 40%

Standard mathematical models have been employed to determine to determine the operating conditions for the process shown in Fig. 1, that is, the membrane area
30 requirement and the power and thermal energy inputs required at various points. This example, modeling a process using a configuration of Fig. 1, is for illustrative purposes only and no attempt has been made to optimize the process configuration. The major
35 reason why optimization is not attempted is that

D-20320-1

- 18 -

optimization is generally based on economic considerations and the commercial production of ion transport membrane systems is yet far from being mature, and no reliable cost estimates are presently available on such systems.

For the present example, looking at Fig. 1, fuel is added only to the process as fuel gas stream 11. In addition, optional gas stream 17 is not considered, that is, gas streams 16 and 18 are identical.

Furthermore, nitrogen is not sought as a co-product and retentate gas stream 36, obtained from retentate gas stream 8 after reducing the excess pressure of the retentate using release valve 23, is mixed with gas stream 30, taken from exhaust gas stream 29. In general, however, it is not effective to lower the pressure of retentate gas stream 8 or to add gas stream 30 to retentate gas stream 8 upstream of heat exchanger 33. Since exhaust gas stream 22 does not contain carbon monoxide and hydrogen gas, afterburner 26 is not installed.

Basis for the example: A downstream process requiring a heat input of 5 million BTU/hr.

Table II: Specifications of Important Gas Streams in the Integrated Process								
Gas Stream	Press. (psia)	Temp. (C)	Flow (1000 SCFH)	Composition (mole %)				
				O ₂	N ₂	CH ₄	CO ₂	H ₂ O
Feed 1	14.70	25.00	165.40	21.00	79.00	-	-	-
Feed 6 to ion transport module	20.00	800.00	165.40	21.00	79.00	-	-	-
Retentate 8	20.00	800.00	153.70	15.00	85.00	-	-	-
Purge 9	15.00	800.00	82.70	trace	-	-	33.30	66.70
Permeate 10	15.00	800.00	94.40	12.40	-	-	29.20	58.40
Fuel 11	15.00	25.00	5.80	-	-	100.00	-	-
Combustor input 13	15.00	745.00	100.20	11.70	-	5.80	27.50	55.00
Combustor exhaust 16	15.00	1660.00	100.20	trace	-	-	33.30	66.70

D-20320-1

- 19 -

Downstream process exhaust 20	15.00	800.00	100.20	trace	-	-	33.30	66.70
Waste 24	15.00	197.00	171.20	13.50	76.30	-	3.40	6.80

Table III: Specifications for Key Modules in the Integrated Process	
Ion transport membrane 7 area (m ²)	49.00
Feed blower or compressor 2 (efficiency = 75%) power (kW)	58.00
Feed heater 34 input (million BTU/hr)	0.50
Total amount of heat generated in combustor 14 (million BTU/hr)	5.30
Heat input requirement for downstream process 19 (million BTU/hr)	5.00

Fig. 2 is a schematic diagram similar to Fig. 1 showing a more efficient alternative using the catalytic afterburner installation. During operation, feed gas stream 41 containing elemental oxygen, usually air, is compressed to a relatively low pressure in blower or compressor 42 to produce compressed feed gas stream 43 which is heated in heat exchanger 73 against hot waste gas stream 40 and product nitrogen gas stream 64 to produce warmed feed gas stream 44. Gas stream 70 may be divided from warmed feed gas stream 44 and used in optional afterburner 69 to leave feed gas stream 74 which is optionally heated in heater 75 to produce hot feed gas stream 45. Hot feed gas stream 45 then enters the feed side of ion transport module 46 employing ion transport membrane 47 having a retentate side 47a and a permeate side 47b. A portion of the oxygen in hot feed gas stream 45 is removed in ion transport module 46 and the exiting gas stream 48 becomes nitrogen-enriched with respect to feed gas stream 41.

The permeate side 47b of ion transport membrane 47 is purged using purge gas stream 79 containing products of combustion. Permeate gas stream 50 contains oxygen and this gas stream 50 is later mixed with fuel gas stream 51. Air, oxygen or oxygen enriched air stream 52 can be optionally added to gas stream 50.

D-20320-1

- 20 -

Combustible gas stream **53**, after passing through an optional blower (not shown), then enters combustor or furnace **54**. Optionally or in addition to, or in place of fuel gas stream **51**, fuel gas stream **55** can be fed
5 directly to combustor **54**. In combustor or furnace **54** heat is transferred to a charge or heat transfer surfaces. By operating combustor **54** close to stoichiometric or slightly fuel-rich condition, the oxygen concentration in exhaust gas stream **56** can be
10 maintained at low levels.

Exhaust gas stream **56** from combustor **54** may be split into two portions, gas stream **57** and gas stream **58**. Gas stream **58** is used in downstream process **59** that requires heat input and the relatively cooler
15 exhaust gas stream **60** from downstream process **59** may also be split into two portions, exhaust gas stream **61** and exhaust gas stream **62**. Fuel gas stream **65** may be added to exhaust gas stream **61** to produce gas stream **78**. Gas stream **78** can be added to gas stream **57** to
20 produce gas stream **79** which enters ion transport module **46** and is used to purge the permeate side **47b** of ion transport membrane **47**.

Exhaust gas stream **62** may optionally be divided into two portions, hot waste gas stream **40** and gas
25 stream **77**. As mentioned above, hot waste gas stream **40** is used in heat exchanger **73** to heat compressed feed gas stream **43** to produce waste gas stream **74**. Gas stream **77** may be mixed with nitrogen-rich retentate gas stream **48** if nitrogen is not to be used as a co-product
30 and if the temperature of exhaust gas stream **77** is adequately high. The reason for this step is to remove any unreacted fuel in exhaust gas stream **62** by combustion in afterburner **69** and to also generate heat energy to improve the efficiency of heat exchanger **73**.
35 Retentate gas stream **48** is likely to be at a higher

D-20320-1

- 21 -

pressure than exhaust gas stream 77 and it may be necessary to release the excess pressure of retentate gas stream 48 using expansion valve 63 to produce retentate gas stream 76 before it is mixed with gas stream 77 to produce gas stream 80.

Gas stream 80 is fed into optional afterburner 69 where gas stream 70 is optionally added to produce hot waste gas stream 39. In this case one would need to make sure that stream 80 contains enough oxygen for the combustion to proceed to completion. As mentioned above, gas stream 70 taken from warmed feed gas stream 44 can be optionally added to afterburner 69 to ensure this. It should be noted that the flow rate of the combined stream is increased by blending the exhaust gases from ion transport module 46 and downstream process 59. This improves the capacity ratio in heat exchanger 73 and increases the heat transfer to compressed feed gas stream 43. Product gas stream 64 will contain oxygen (used in excess to ensure complete combustion) and products of combustion if afterburner 69 is used and product gas stream 64 is generally discarded as a waste stream.

As with the embodiment of the invention shown in Fig. 1, the use of an oxygen-depleted purge gas stream 79 in ion transport module 46 will greatly lower the oxygen partial pressure on the permeate side 47b of ion transport membrane 47 and enable rapid oxygen transport through membrane 47. Fuel gas streams 51, 55, and 65 can be introduced into the process configuration at any or all of the points shown in Fig. 2 to obtain the benefits of the invention and the use of at least one fuel gas stream is essential for the invention. As before, it may be desirable to add fuel gas stream 65 upstream of ion transport module 46 to greatly reduce the oxygen partial pressure on the permeate side 47b of

D-20320-1

- 22 -

ion transport membrane **47**. This would also result in some heat generation in ion transport module **46** due to fuel combustion, thereby offsetting some of the heating requirements of the oxygen transport process. In this case, the exiting nitrogen-rich gas stream **48** from ion transport module **46** could be made hotter and this would make the heat transfer in heat exchanger **73** more efficient, thereby reducing the area required for the heat exchange and potentially eliminating the need for heater **75** upstream of ion transport module **46**. If enough fuel can be combusted in ion transport module **46** on the purge or permeate side **47b** of ion transport membrane **47**, it may altogether eliminate the need for a separate combustor **54**, that is, ion transport module **46** would also serve as the combustor (as described in Fig. 3). In such a situation, significant system simplification and cost reduction may result.

As with the embodiment of the invention shown in Fig. 1, it may be advantageous to operate combustor **54** with a slightly fuel-rich mixture because this will lead to partial oxidation of the fuel added to permeate gas stream **50**, resulting in an exhaust gas stream **56** containing hydrogen gas and carbon monoxide. As mentioned above, gas stream **57** is optionally used to purge the permeate side **47b** of ion transport membrane **47** and the presence of hydrogen gas in ion transport module **46** will result in an extremely low oxygen partial pressure on the purge side **47b** of ion transport membrane **47** and this will enable an even more rapid transport of oxygen through ion transport membrane **47**. The use of a fuel-rich feed to combustor **54** produces hydrogen gas as a part of the process cycle. As mentioned above, it may be possible, to install afterburner **69** (perhaps catalytic) to burn off the carbon monoxide and hydrogen gas if their concentration

D-20320-1

- 23 -

is sufficiently high.

Fig. 3 is a schematic diagram showing another embodiment of the invention where the combustor is integrated with the ion transport module, that is, where the ion transport module itself serves as the combustor. During operation, feed gas stream **81** containing elemental oxygen, usually air, is compressed to a relatively low pressure in blower or compressor **82** to produce compressed feed gas stream **83** which is heated in heat exchanger **113** against hot waste gas stream **116** and product nitrogen gas stream **93** to produce warmed feed gas stream **95**. Gas stream **110** may be divided from warmed feed gas stream **95** and used in optional afterburner **109** to leave feed gas stream **84** which is optionally heated in heater **114** to produce hot feed gas stream **85**. Hot feed gas stream **85** then enters the feed side of ion transport-combustor module **86** employing ion transport membrane **87** having a retentate side **87a** and a permeate side **87b**. A portion of the oxygen in hot feed gas stream **85** is removed in ion transport-combustor module **86** and the exiting gas stream **88** becomes nitrogen-enriched with respect to feed gas stream **81**.

The permeate side **87b** of ion transport membrane **87** is purged using purge gas stream **89** containing products of combustion and fuel. Permeate gas stream **90** contains oxygen and air stream **92** can be optionally added to gas stream **90** to produce gas stream **98**. By operating ion transport module-combustor **86** close to stoichiometric or slightly fuel-rich condition, the oxygen concentration in exhaust gas stream **90** can be maintained at low levels. Gas stream **98** is used in downstream process **99** that requires heat input and the relatively cooler exhaust gas stream **100** from downstream process **99** is also be split into two

D-20320-1

- 24 -

portions, exhaust gas stream **101** and exhaust gas stream **102**. Fuel gas stream **105** preferably is added to exhaust gas stream **101** to produce gas stream **89** which enters ion transport-combustor module **86** and is used to
5 purge the permeate side **87b** of ion transport membrane **87**.

Exhaust gas stream **102** may optionally be divided into two portions, hot waste gas stream **116** and gas stream **115**. As mentioned above, hot waste gas stream
10 **116** is used in heat exchanger **113** to heat compressed feed gas stream **83** to produce waste gas stream **117**. Gas stream **115** may be mixed with nitrogen-rich retentate gas stream **88** if nitrogen is not to be used as a co-product and if the temperature of exhaust gas
15 stream **115** is adequately high. The reason for this step is to remove any unreacted fuel in exhaust gas stream **102** by combustion in afterburner **109** and to also generate heat energy to improve the efficiency of heat exchanger **113**. Retentate gas stream **88** is likely to be
20 at a higher pressure than exhaust gas stream **115** and it may be necessary to release the excess pressure of retentate gas stream **88** using expansion valve **103** to produce retentate gas stream **118** before it is mixed with gas stream **115** to produce gas stream **119**.

Gas stream **119** is fed into optional afterburner
25 **109** where gas stream **110** is optionally added to produce hot waste gas stream **93**. In this case one would need to make sure that stream **119** contains enough oxygen for the combustion to proceed to completion. As mentioned
30 above, gas stream **110** taken from warmed feed gas stream **95** can be optionally added to afterburner **109** to ensure this. It should be noted that the flow rate of the combined stream is increased by blending the exhaust gases from ion transport-combustor module **86** and
35 downstream process **99**. This improves the capacity

D-20320-1

- 25 -

ratio in heat exchanger **113** and increases the heat transfer to compressed feed gas stream **83**. Gas stream **94** will contain oxygen (used in excess to ensure complete combustion) and products of combustion if
5 afterburner **109** is used and gas stream **94** is generally discarded as a waste stream.

In the embodiment of Fig. 3, the heat of reaction generated in ion transport-combustor module **86** is removed from or consumed in the combustor in a
10 convective and/or radiative heat transfer process. For example, ion transport membrane **87** may be formed as tubes with reactive purge gas stream **89** flowing inside the tubes. Owing to the heat generated on the purge side **87b** of ion transport membrane **87** formed as tubes,
15 the tubes will be at a high temperature and will act as heating elements. The ion transport membrane **87** tubes will radiate to the retentate side **87a** or permeate side **87b** where a process such as glass melting or metal annealing could be carried out. Also, a part of the
20 heat generated in ion transport module **86** could be used to preheat compressed feed gas stream **85** and purge gas stream **89**, possibly obviating the need for heat exchanger **113** and heater **114**. Note that the furnace load will be placed on the permeate side **87b** of ion
25 transport membrane **87** (that is, the side with the oxidizing gas) in this case.

It is also possible to integrate ion transport-combustor module with an internal circulation of the flue (furnace) gas. If the furnace and the ion
30 transport-combustor module operate at about the same temperature (for example, between 800°-1200°C), then the ion transport-combustor module can be placed directly inside the furnace provided that the furnace atmosphere is "clean", that is, it does not contain any
35 species detrimental to the ion transport membrane. One

D-20320-1

- 26 -

way of implementing this idea is shown in Fig. 4 in which the ion transport process, combustor, and downstream process are all integrated in a single unit. Feed stream **132** such as heated air is directed against cathode side **120a** of membrane **120** to produce hot, oxygen-depleted retentate **134** such as nitrogen. Downstream process **130** (for example, a furnace load) is shown on the permeate or anode side **120b** of ion transport membrane **120**. In this configuration, fuel gas stream **121** is fed close to the surface of permeate side **120b**, thus sweeping away and/or efficiently consuming oxygen transported across ion transport membrane **120**. The combustion products in hot zone **138** could be recirculated in the furnace against anode side **120b** by natural or forced convection; for the construction shown in Fig. 4, combustion products stream **146**, preferably obtained from furnace **130** as shown in phantom by stream **146a**, and the fuel gas stream **121** are optionally fed through porous fuel distributor layer **122** adjacent to permeate side **120b** of ion transport membrane **120**. Preferably, distributor layer **122** defines at least one passage or chamber to more uniformly distribute fuel across the membrane **120**.

Reacted permeate **136** containing oxygen and combustion products is directed to furnace **130** through hot zone **138**. Preferably, a portion of hot nitrogen **140** is directed through valve **142** to provide an inerting atmosphere over the furnace **130**. Additional fuel **144** may be added to furnace **130** as desired.

In another construction, the ion transport membrane **120** is part of a separate module that is external to the furnace **130**. In either of the external or integrated constructions, a two-stage ion transport system can be established in which the anode side of the first stage is purged by the retentate stream from

D-20320-1

- 27 -

the first stage to produce a diluted oxygen permeate stream while the anode side of the second stage is reactively purged to produce a fuel-rich permeate stream. The two permeate streams are used in a furnace
5 for combustion with or without use of the hot nitrogen retentate streams in the furnace atmosphere.

When the peak furnace temperature is much higher than ion transport operating temperature, a furnace zone with the "right" temperature may be selected for
10 ion transport operation (for example, preheat section of a continuous reheat furnace), or a special chamber with proper heat sinks to control the temperature can be created. For example, in boiler applications or petroleum heaters, it would be feasible to use the
15 furnace heat loads (that is, water or oil tubes) to create a zone of an optimum temperature for the ion transport module. A large amount of flue gas is circulated through this zone to continuously purge oxygen and keep the oxygen concentration low. Low
20 oxygen concentration and high furnace gas circulation provide a synergy with the dilute oxygen combustion method.

There are many advantages of the integrated processes of the invention. For example, the oxygen
25 for the OEC can be extracted from a low pressure feed gas stream by using the exhaust gas stream for purging and this should result in low power requirement for the oxygen separation process.

Because only oxygen passes through the ion
30 transport membrane, no nitrogen is added to the purge gas stream exiting the ion transport module. Even if air is introduced into the combustion mixture, either intentionally (for example, optional gas stream 12) or by leakage, the fraction of nitrogen in the combustion
35 mixture will be small. This should minimize or

D-20320-1

- 28 -

eliminate NO_x formation in the combustor.

Further, by suitably blending of the exhaust gases taken before and after the downstream process, it is possible to control the purge inlet temperature to that
5 desired in the ion transport process. This may eliminate the need for preheating the purge gas independently.

In addition, if the combustion of all the fuel can be carried out in the ion transport module, the
10 separate combustor unit can be eliminated. This would yield significant system simplification and cost savings. Furthermore, if enough oxygen is removed from the feed gas stream in the ion transport module, then the nitrogen-rich stream retentate from the ion
15 transport module can be used as a product. This may be most attractive if some fuel is added, for example, fuel gas stream 11. If nitrogen is desired as a co-product, it may be advantageous to compress the feed gas stream to the pressure required for product
20 nitrogen supply. However, in this case, the retentate gas stream from the ion transport module may not be mixed with the exhaust gas stream from the downstream process. In this case, either a separate heat exchanger can be installed to recover heat from the
25 exhaust gas stream, or no heat recovery may be attempted since generally exhaust gas stream will be much smaller and colder as compared to the retentate gas stream.

Moreover, the use of the purge gas stream
30 decreases the oxygen concentration on the permeate side of the ion transport membrane. The reduced oxygen concentration makes the design of the ion transport module and of the downstream components (for example, combustor) on the purge side considerably easier from a
35 materials standpoint. In the absence of a purge

D-20320-1

- 29 -

stream, essentially pure oxygen would be produced on the permeate side of the ion transport membrane. Safe handling of such a high purity oxygen stream poses a significant challenge, especially at elevated
5 temperature.

In addition, the concentration of the oxygen in the purge exhaust can be easily controlled by a number of techniques: for example, varying the feed gas stream flowrate, varying the purge gas stream flowrate
10 (increased recycle of the combustion products), changing the ion transport module operating temperature, or varying the membrane area of the ion transport stage. These techniques are also effective in controlling the total amount of oxygen separated and
15 could be used for load tracking purposes.

Lastly, the use of the ion transport separator would eliminate the need for a stand-alone oxygen generator (for example, PSA) or an oxygen supply system (for example, liquid tank and vaporizer). This is
20 expected to yield substantial reduction in capital cost and in the cost of the oxygen produced.

An embodiment particularly directed to this invention is shown in Fig. 5. The feed stream or air
232 is preheated to a suitable temperature of about
25 500°C to 1000°C. The feed stream or air preferably contains a relatively high oxygen concentration of typically greater than 15% oxygen. The feed stream or air 232 is then introduced to the retentate side 220a of ion transport membrane 220 through permeate side
30 120b to the permeate chamber 238, which is connected to the main combustion chamber or furnace 230. The temperature of the main chamber is determined by the requirement of the heating process and can be higher or lower than the temperature of the permeate chamber 238.
35 A significant portion of flue gas generated in furnace

D-20320-1

- 30 -

230 is recirculated as stream 246, to permeate chamber 238. Stream 246 causes the oxygen concentration in the permeate side 220b of the ion transport membrane 220 to be low and increases oxygen flux through oxygen transport membrane 220. As an option, fuel stream 221 can be introduced to permeate chamber 238 to reduce the oxygen concentration of recirculated stream 246 and/or to reactively purge the permeate side 220b and also to provide heat to raise the air temperature on the retentate side. Permeate chamber 238 is optionally segmented into a two stage ion transport membrane system using a reactive purge in the first stage and a flue gas purge in the second stage (not shown). Fuel stream 221 is preferably injected at high velocity to induce proper circulation of hot flue gas stream 246 from furnace 230 by the aspiration effect of high velocity jets. Other means of hot flue gas recirculation including, but not limited to, a circulation fan and an eductor using compressed flue gas or steam can be used. In addition to or in place of internal recirculation stream 246, external recirculation stream 248 may be used. In a preferred embodiment flue gas stream 247 from furnace 230 is cooled in heat exchanger 260 and cooled side stream 248 is compressed and injected into first chamber 238 to induce proper circulation of the hot flue gas stream 246. The process can be used for many industrial furnaces such as metal heat treating, steel reheating, glass melting and aluminum melting furnaces, but it is particularly suitable for boilers to generate steam and for process fluid heating furnaces such as petroleum heaters, cracking furnaces and steam-methane reformer furnaces. In the following description of the invention, a process to generate steam is described. However, it is not intended to be limiting to the

D-20320-1

- 31 -

application of this invention. The purge gas preferably contains a low concentration of nitrogen, which through transport to the permeate side of the ion transport membrane forms oxidant stream **236** containing
5 between 1% to 40% oxygen, preferably between 2% to 15% oxygen, and most preferably between 3% to 10% oxygen.

The ion transport membrane **220** effectively forms two sides of the reactive area having a same or different pressure, the retentate side **220a** having a
10 first pressure and a permeate side **220b** having a second pressure. The ratio of the first pressure to the second pressure is less than 4.78, preferably between 0.5 to 4.0, more preferably between 0.8 to 2.0, and most preferably between 0.9 to 1.5. In the
15 conventional process of separating oxygen from air using a mixed conductor ion transport membrane system, feed air has to be compressed to a minimum pressure at 4.78 atm in order to achieve an oxygen partial pressure of 1 atm in the retentate side. Thus, a theoretical
20 minimum pressure ratio of 4.78 is required to produce pure oxygen at 1 atm in the permeate side. In order to produce a high oxygen flux through the membrane, a practical system requires pressure ratios of 10 to 30 so as to produce oxygen partial pressure ratios of
25 about 2 to 6. Large electric power is required to compress feed air to 10 to 20 atm. Low pressure ratios are advantageous to reduce the power required for compressing feed air. In an idealized system both the first and second pressures are equal and atmospheric
30 which minimize the power requirement and potential gas leakage between the retentate side **220a** and the permeate side **220b**. In such a process, the only pressure requirement for feed air and purge gas streams is to overcome normal pressure drops associated with
35 passing gas flows through the membrane module and the

D-20320-1

- 32 -

furnace. It should be noted that this invention allows permeation of oxygen through the ion transport membrane even when the pressure ratio is less than 1, i.e., the pressure of the retentate side is lower than that of the permeate side. It is because the permeation of oxygen is driven by the ratio of the oxygen partial pressure of the retentate side to that of the permeate side. By keeping the oxygen partial pressure low in the permeate side, it is possible to achieve an oxygen partial pressure ratio of greater than 2 while maintaining the total pressure ratio near or below 1. Under such a condition, gas leakage caused by imperfect seals of ceramic membrane tubes or by cracks in ceramic membrane tubes flow from the permeate side to the retentate side and prevent the infiltration of nitrogen into furnace **230**. In order to reduce NO_x formation and to maintain a high concentration of CO₂ in the flue gas, it is preferable to keep the nitrogen concentration in furnace **230** below 10%.

In permeate chamber **238**, fuel jets are optionally used to recirculate hot flue gas from the furnace **230** into the permeate chamber **238**. Steam jets or other gas jets, such as compressed cooled flue gas stream **248**, can be used to provide additional purge gas and to facilitate the circulation of flue gas through the jet pump effect. The fuel jet reacts with the oxygen contained in the recirculated flue gas and oxygen permeated from the ion transport membrane tubes, reducing the oxygen concentration in permeate chamber **238** and generating heat. Boiler tubes, steam super heater tubes and/or steam reheat tubes may be optionally placed in the permeate chamber **238** as heat sinks to control the chamber temperature. The temperature of the membrane tube in the permeate chamber **238** is preferably maintained at an optimum

D-20320-1

- 33 -

membrane temperature, typically between 600°C and 1000°C, by controlling the flow rate of feed air **232**, the flow rate of fuel **221** injected into this permeate chamber **238** and the amount of recirculated hot flue gas **246**. The average oxygen concentration in the oxidant mixture **236**, which consists of combustion products of fuel **221**, oxygen transported through the ion transport membrane **220**, and recirculated flue gas **246**, at the exit of the permeate chamber **238** may be controlled from 1% to 40% oxygen, preferably 2% to 10%, and most preferably between 3 to 10% oxygen. This mixture is circulated to the combustion chamber **230**, preferably without cooling. The air flow through the retentate side of the ion transport membrane tubes and the fuel/flue gas flow in the permeate side can be arranged in a counter current fashion to maximize the oxygen flux through the ion transport membrane tubes. Lower oxygen concentration in the permeate side **220b** is generally preferred to increase the oxygen permeation rate of the ion transport membrane. However, a large volume of flue gas has to be recirculated to keep the oxygen concentration low. Thus, the above mentioned range is preferred.

In furnace **230**, fuel **244** is injected and mixed with the hot dilute oxygen stream **236** from permeate chamber **238** for combustion to generate steam in the conventional fashion for the radiant section of a steam boiler. The dilute oxygen combustion process as described in H.M. Ryan, et al., Development of the Dilute Oxygen Combustion Burner for High Temperature Furnace Applications, Proceedings of the Fourth International Conference on Technologies and Combustion or Clean Environment, July 7-10, 1997 and U.S. Patent No. 5,076,779 disclose the preferred combustion process to achieve low NO_x emissions and high thermal

D-20320-1

- 34 -

efficiency.

Preferably 50% to 98% of the flue gas generated in furnace **230** is recirculated to permeate chamber **238** and the balance to convective heat transfer unit **260** for
5 generating, super heating and reheating of steam and for feed water heating. Flue gas stream **249** is optionally passed through a condensing heat exchanger **264** to recover the latent heat of water vapor in the flue gas. The full recovery of the latent heat of
10 water vapor in flue gas provides a theoretical energy improvement of about 10% for natural gas combustion, in addition to the efficiency improvement as a result of reduced sensible heat in cooled flue gas stream **263**. Oxygen combustion provides a unique advantage over air
15 combustion for the recovery of latent heat due to the high concentration of water vapor in flue gas. For example, the concentration of water vapor in the flue gas from natural gas-oxygen combustion is about 65% and the dew point is about 88°C. By cooling the flue gas
20 to about 49°C the concentration of water vapor is reduced to 13% by volume. About 93% of water vapor can be condensed. By comparison the concentration of water vapor in the flue gas from natural gas-air combustion is about 18% and the dew point is about 58°C. By
25 cooling the flue gas to about 49°C, about 41% of water vapor can be condensed. Further cooling to 32°C is required to condense about 77% of water vapor in natural gas-air combustion. Accordingly, an unexpectedly large fraction of water vapor contained in
30 flue gas can be condensed at relatively high temperature when natural gas or a fuel with a high hydrogen to carbon atomic ratio is combusted with oxygen. By condensing water vapor the flue gas volume is further reduced and the carbon dioxide rich stream
35 is produced for potential recovery and production of

D-20320-1

- 35 -

carbon dioxide.

Hot nitrogen-rich stream **234** from retentate chamber is passed through a convective heat transfer unit **261** to heat feed water and optionally provide heat for steam generation, super heating or reheating. Alternatively this hot stream **234** can be used to preheat feed air, as shown in Figs. 1 to 3. A high purity nitrogen stream can be produced using a two stage ion transport membrane system with a reactive purge in the first stage and a flue gas purge in the second stage.

Heating of feed air stream **232** can also be done indirectly in a heat exchanger (not shown) with hot flue gas stream **247** and/or by placing air heating tubes in furnace **230** and/or in permeate chamber **238**. In addition, an in-line combustor (not shown) can be used to heat air to a higher temperature. Although in-line heating results in a reduction in the oxygen concentration of feed air stream **232** and hence reduce the driving force for oxygen flux, the simplicity of the direct in-line heating system offers a significant capital cost reduction over the indirect heating system.

Table IV and V show an example of specifications of an integrated ion transport membrane-combustion process depicted in Fig. 5.

D-20320-1

- 36 -

Table IV: Specifications of Important Gas Streams in the Integrated Process								
Gas Stream	Press. (psia)	Temp. (C)	Flow (1000 SCFH)	Composition (mole %)				
				O ₂	N ₂	CH ₄	CO ₂	H ₂ O
Feed 232 to ion transport module	17.70	800.00	44.4	21.00	79.00	-	-	-
Retentate 234	17.70	800.00	32.6	Trace	100.0	-	-	-
Purge 246	14.70	800.00	200.0	2.00	-	-	32.60	65.40
Oxidant 253	14.70	800.00	211.8	7.43	-	-	30.83	61.74
Fuel 244	25.00	25.00	3.800	-	-	100.0	-	-
Fuel 221	25.00	25.00	2.00	-	-	100.0	-	-
Combustor exhaust 247	14.70	1200.00	17.60	2.00	-	-	32.60	65.40
Downstream process exhaust 249	14.70	200.0	17.60	2.00	-	-	32.60	65.40
Cooled Exhaust 262	14.70	49.0.00	7.00	5.03	-	-	81.97	13.00

Table V: Specifications for Key Modules in the Integrated Process	
Feed blower (not shown) (efficiency = 75%) power (kW)	9.30
Total amount of fuel input 221 plus 244 (million BTU/hr)	5.80

5 In the above example, a large amount of hot flue gas is recirculated to keep the oxygen concentration in the permeate side of the membrane at 7.43%, which helps to increase the oxygen flux through the membrane without

10 compressing air feed stream to a high pressure. By producing high purity nitrogen with a reactive purge step in ion transport membrane 220, substantially all of oxygen contained in feed air stream 232 permeates through ion transport membrane 220 and used for

15 combustion with fuel streams 221 and 244. The flow rate of air feed is 44,290 SCFH as compared 165,400 SCFH in the previous example shown in Table II. The large reduction in the air feed rate reduces the amount of heat required to heat feed air and the power required

D-20320-1

- 37 -

in feed air blower. The blower power required in this example is 9.3kw and 11,740 SCFH of oxygen is separated by the ion transport membrane, corresponding to 0.079kw of power per 100 SCFH of oxygen. By comparison, the
5 previous example consumes 58kw of power to produce 11,600 SCFH of oxygen, corresponding to 0.5kw of power per 100 SCFH. It should be noted that the current state-of-the-art air separation technologies consume about 0.8kw of power to produce 100 SCFH industrial
10 grade oxygen at 90% to 95% oxygen purity and at 1 atmosphere. Thus, the integrated ion transport membrane-combustion processes of this invention reduce the power consumption to separate air for oxygen combustion by as much as 90%.

15 Another embodiment of this invention provides for a process where no flue gas recirculation is used in a reactively purged ion transport membrane chamber of a two-membrane chamber combustion process depicted in Fig. 6. Certain common features in both Figs 5 and 6
20 are not repeated herein, and the common features may be referred to in the description of the Fig. 5 embodiment.

In Fig. 6, hot nitrogen-rich stream **334** from the first retentate chamber is introduced into the second
25 retentate side **372a** of the ion transport membrane **372** in the second retentate chamber **370**. Oxygen permeates through the second permeate side **372b** of the ion transport membrane **372** into the second permeate chamber **350**, in which reactively purged high oxygen flux is
30 achieved even when nitrogen feed stream **334** contains low concentration of oxygen. Fuel stream **355** is injected into the second permeate chamber **350** to react with the oxygen flux from the second permeate side **372b**

D-20320-1

- 38 -

of the ion transport membrane **370**, thereby forming hot syngas stream **373**, a gas stream containing combustible gases such as hydrogen and carbon monoxide.

Optionally, recycled flue gas or steam **356** may is
5 injected into the second permeate chamber **350** to modulate the temperature.

Hot syngas stream **373** is introduced into furnace **330** and mixed with hot oxidant stream **336** from the first permeate chamber **338**, which provides the oxygen
10 for combustion. Optionally, additional air or oxygen stream **357** is injected into furnace **330** to combust the hot syngas stream **373** and to provide heat to the process.

Nitrogen-rich stream **334** is enriched as it passes
15 across the second retentate side **372a**, where oxygen is removed as it passes through second ion transport membrane **372** into second permeate chamber **350**, and a stream of high purity nitrogen **365** is formed. High purity nitrogen stream **365** then passes through
20 convective heat transfer unit **361** to recover cool nitrogen stream **362** for use or further processing.

It should be noted that a number of modifications of the process are possible within the spirit of the process configuration discussed above. For example, it
25 may be advantageous to use the exhaust gas from the downstream process to heat the feed gas stream. It is also possible to add some air to the purge gas stream exiting the ion transport module. This may be particularly desirable for start-up operations or for
30 load tracking purposes. In addition, although the processes described here are for mixed conductor pressure-driven ion transport membranes, it is obvious

D-20320-1

- 39 -

that the inventive concept is also applicable to primarily ionic conductors that are operated in the pressure- or electrically-driven mode with an external current return. Lastly, although a countercurrent
5 oxygen separation process is depicted in Fig. 1, the same process can also be carried out in a concurrent or a crossflow mode.

As mentioned above, the terms "solid electrolyte ionic conductor", "solid electrolyte", "ionic
10 conductor", and "ion transport membrane" are generally used herein to designate either an ionic-type (electrically-driven) system or a mixed conductor-type (pressure-driven) system unless otherwise specified.

The term "nitrogen" as used herein will usually
15 mean oxygen-depleted gas, that is, oxygen-depleted relative to the feed gas. As discussed above, the ion transport membrane only allows oxygen permeation. Therefore, the composition of the retentate will depend on the composition of the feed gas. The feed gas will
20 be depleted of oxygen but will retain nitrogen and any other gases (for example, argon) present in the feed gas. The meaning of the term will be clear to one of skill in the art in the context of the use of the term in light of the invention as disclosed herein.

25 As used herein the term "elemental oxygen" means any oxygen that is uncombined with any other element in the Periodic Table. While typically in diatomic form, elemental oxygen includes single oxygen atoms, triatomic ozone, and other forms uncombined with other
30 elements.

The term "high purity" refers to a product stream which contains less than five percent by volume of undesired gases. Preferably the product is at least 98.0% pure, more preferably 99.9% pure, and most
35 preferably at least 99.99% pure, where "pure" indicates

D-20320-1

- 40 -

an absence of undesired gases.

"Pressure-swing adsorption" or "PSA" systems refers to systems using adsorption materials which are selective for a gas, typically nitrogen, to separate
5 that gas from other gases. Such materials include rate-selective PSA materials, which are usually carbon-containing and provide high pressure nitrogen and low pressure oxygen, and equilibrium-selective PSA
10 materials, which are usually lithium-containing and provide low pressure nitrogen and high pressure oxygen.

Specific features of the invention are shown in one or more of the drawings for convenience only, as each feature may be combined with other features in accordance with the invention. In addition, various
15 changes and modifications may be made to the examples given without departing from the spirit of the invention. Such modifications may include the use of pressure-swing and thermal-swing adsorption beds or other methods of bulk oxygen separation to provide the
20 function of the polymeric membranes discussed above. Alternative embodiments will be recognized by those skilled in the art and they are intended to be included within the scope of the claims.

D-20320-1

- 41 -

The embodiments of the invention in which an exclusive property or privilege is claimed are defined as follows:

1. A combustion process using an oxidant with a low nitrogen concentration comprising:

5 (a) introducing an oxygen-containing gas to an ion transport module including an ion transport membrane having a retentate side with a first pressure and a permeate side with a second pressure forming a permeate chamber to separate a purified oxygen gas stream on the
10 permeate side and correspondingly depleting the oxygen on the retentate side to produce an oxygen-depleted gas stream;

(b) passing a purge gas stream containing a low concentration of nitrogen to the permeate side to form
15 an oxidant stream containing less than about 40% oxygen; and

(c) introducing said oxidant stream and a fuel into a combustion chamber to produce heat and products of combustion.
20

2. The combustion process of claim 1 wherein said oxygen-containing gas is air.

3. The combustion process of claim 1 wherein the
25 ratio of said first pressure to said second pressure is less than 4.78.

4. The combustion process of claim 1 wherein said oxidant stream comprises between 1% to 40% oxygen.
30

5. The combustion process of claim 1 wherein said purge gas comprises less than 10% nitrogen.

6. The combustion process of claim 1 wherein the
35 oxygen partial pressure ratio of said retentate side to

D-20320-1

- 42 -

said permeate side is greater than 2 and the ratio of the total pressure on said retentate side to said permeate side is less than 1.

5 7. A combustion process using an oxidant with a low nitrogen concentration comprising:

 (a) introducing an oxygen-containing gas to an ion transport module including a first and second ion transport membranes, said first ion transport membrane
10 having a first retentate side with a first retentate pressure and a first permeate side with a first permeate pressure forming a first permeate chamber to separate a first purified oxygen gas stream on the first permeate side and correspondingly depleting the
15 oxygen on the first retentate side to produce a first oxygen-depleted gas stream;

 (b) introducing said first oxygen-depleted gas stream to said second ion transport membrane having a second retentate side with a second retentate pressure
20 and a second permeate side with a second permeate pressure forming a second permeate chamber to separate a second purified oxygen gas stream on the second permeate side and correspondingly depleting the oxygen on the second retentate side to produce a second
25 oxygen-depleted gas stream;

 (c) passing a first purge gas stream containing a low concentration of nitrogen to said first permeate side to form an oxidant stream containing less than about 40% oxygen;

30 (d) passing a second purge gas stream containing a fuel and a low concentration of nitrogen to said second permeate side to form a syngas stream; and

 (e) introducing said oxidant stream, said syngas stream into a combustion chamber to produce heat and
35 products of combustion.

D-20320-1

- 43 -

8. The combustion process of claim 7 further comprising recovering nitrogen gas.

5 9. The combustion process of claim 7 wherein said oxygen-containing gas is air.

10 10. The combustion process of claim 7 wherein the ratio of said first retentate pressure to said first permeate pressure is less than 4.78.

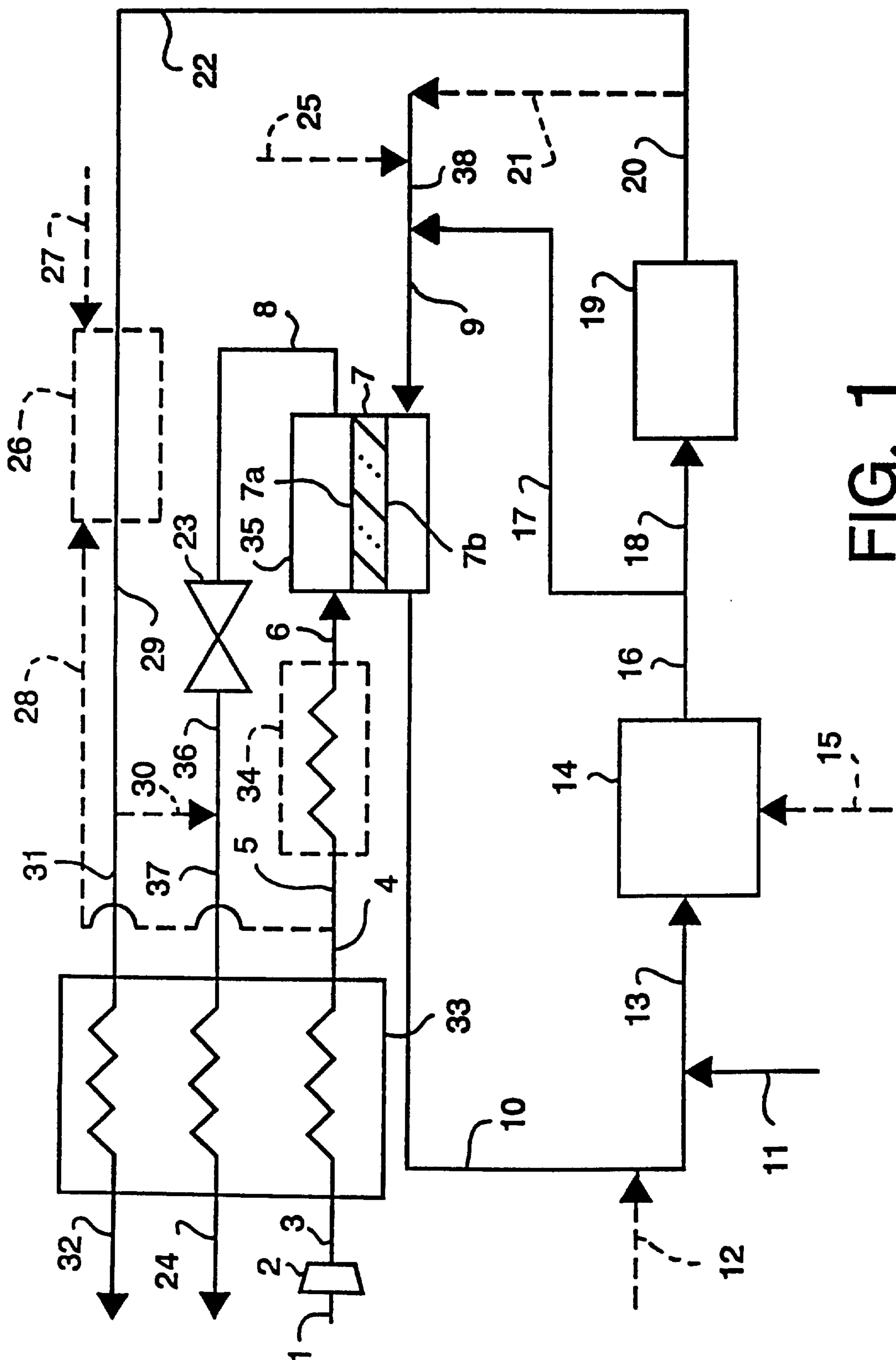


FIG. 1

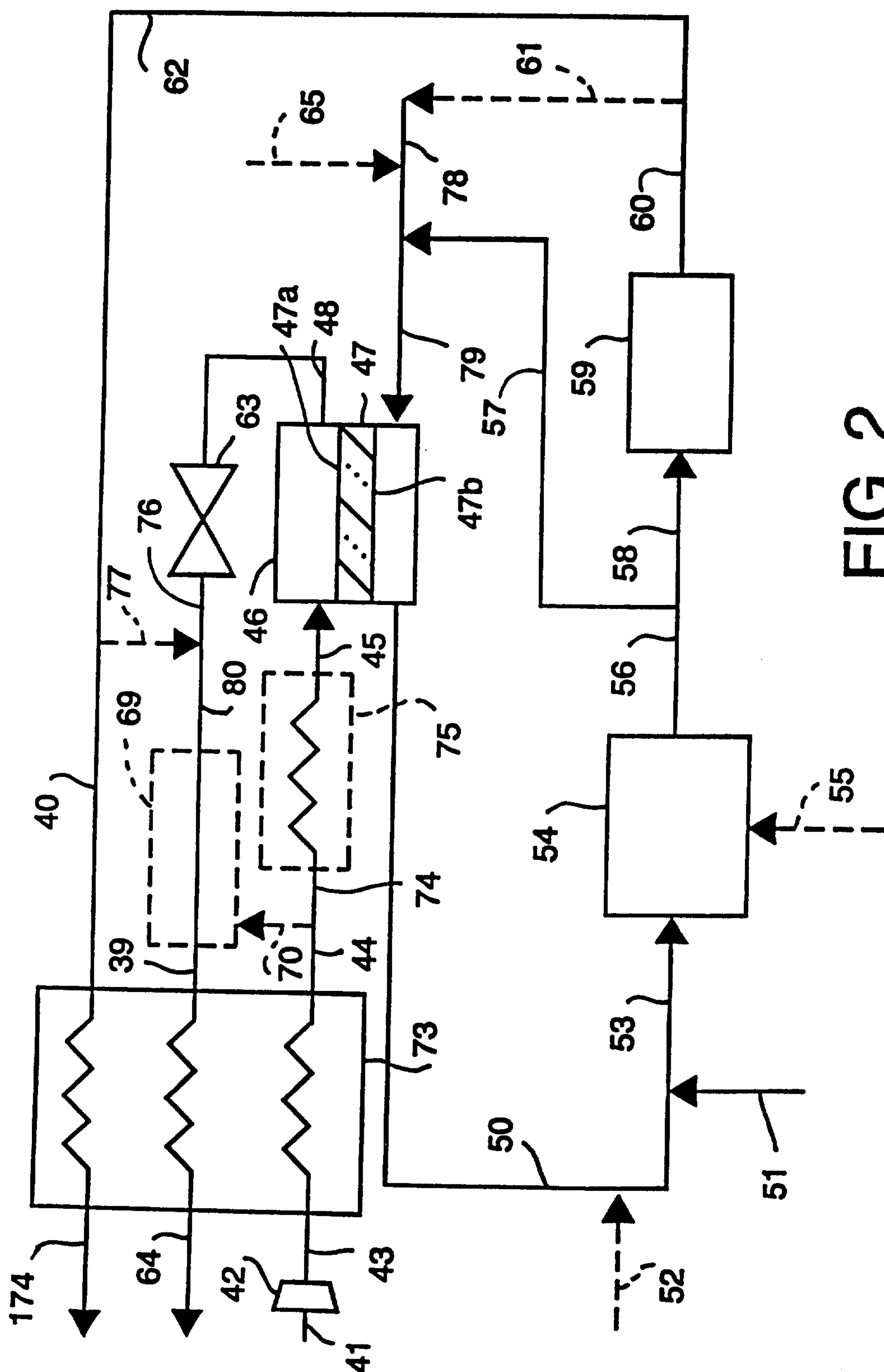


FIG. 2

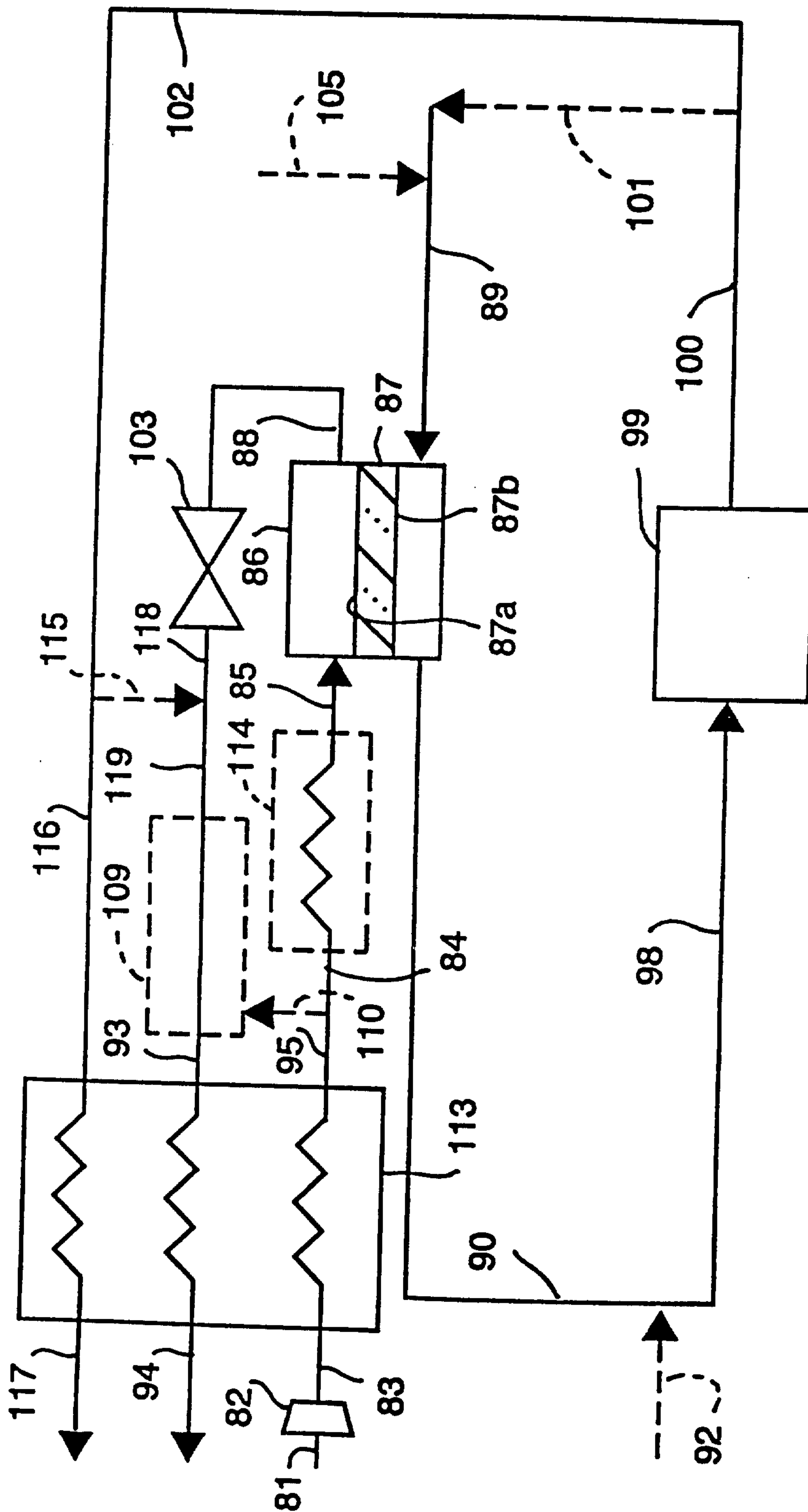
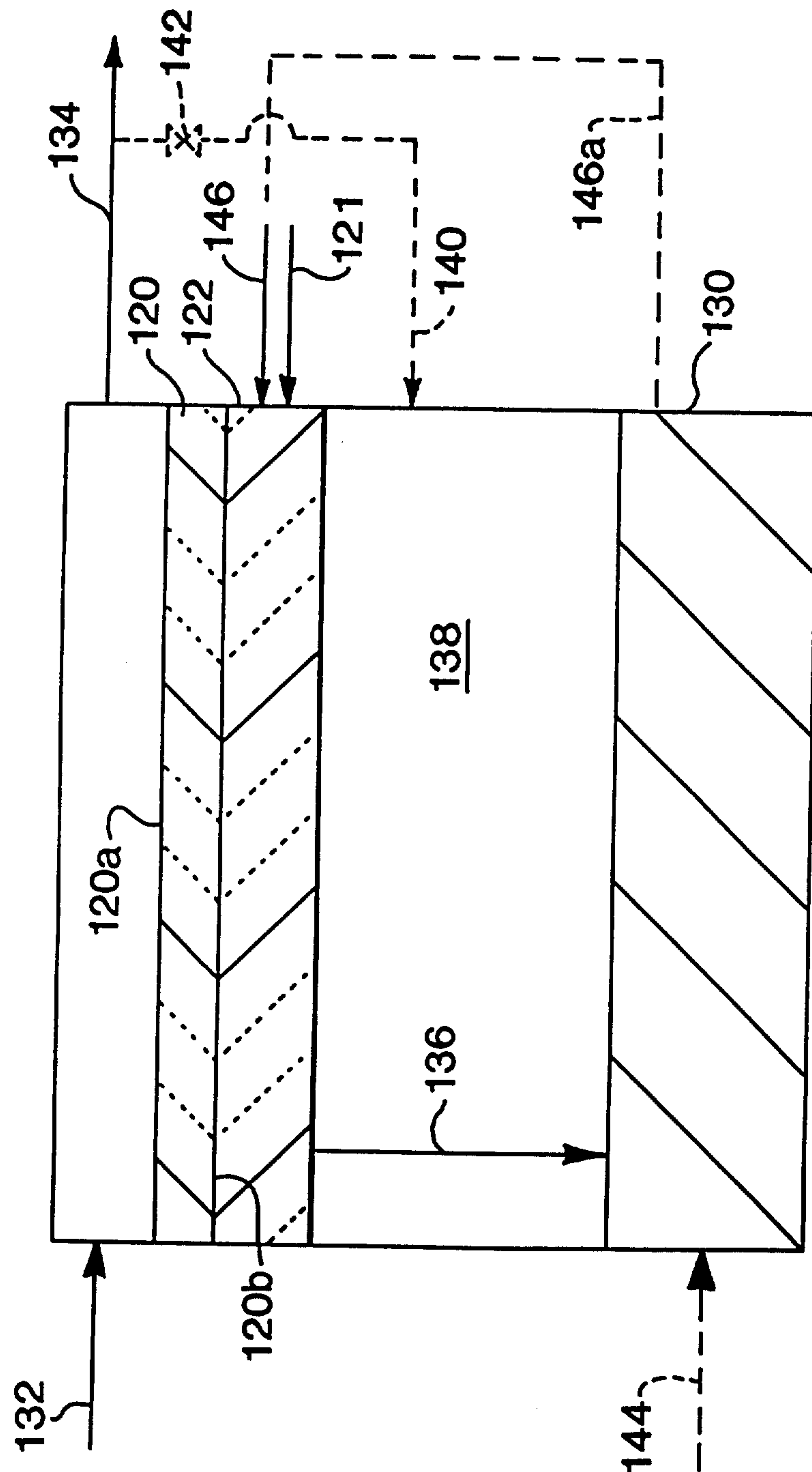


FIG. 3.

FIG. 4



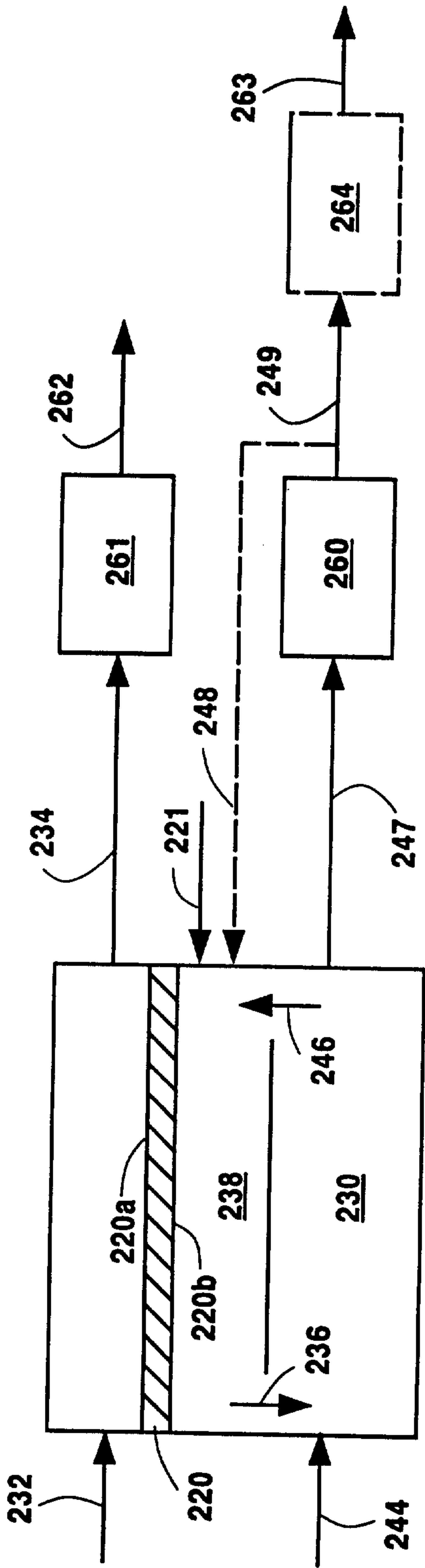


FIG. 5

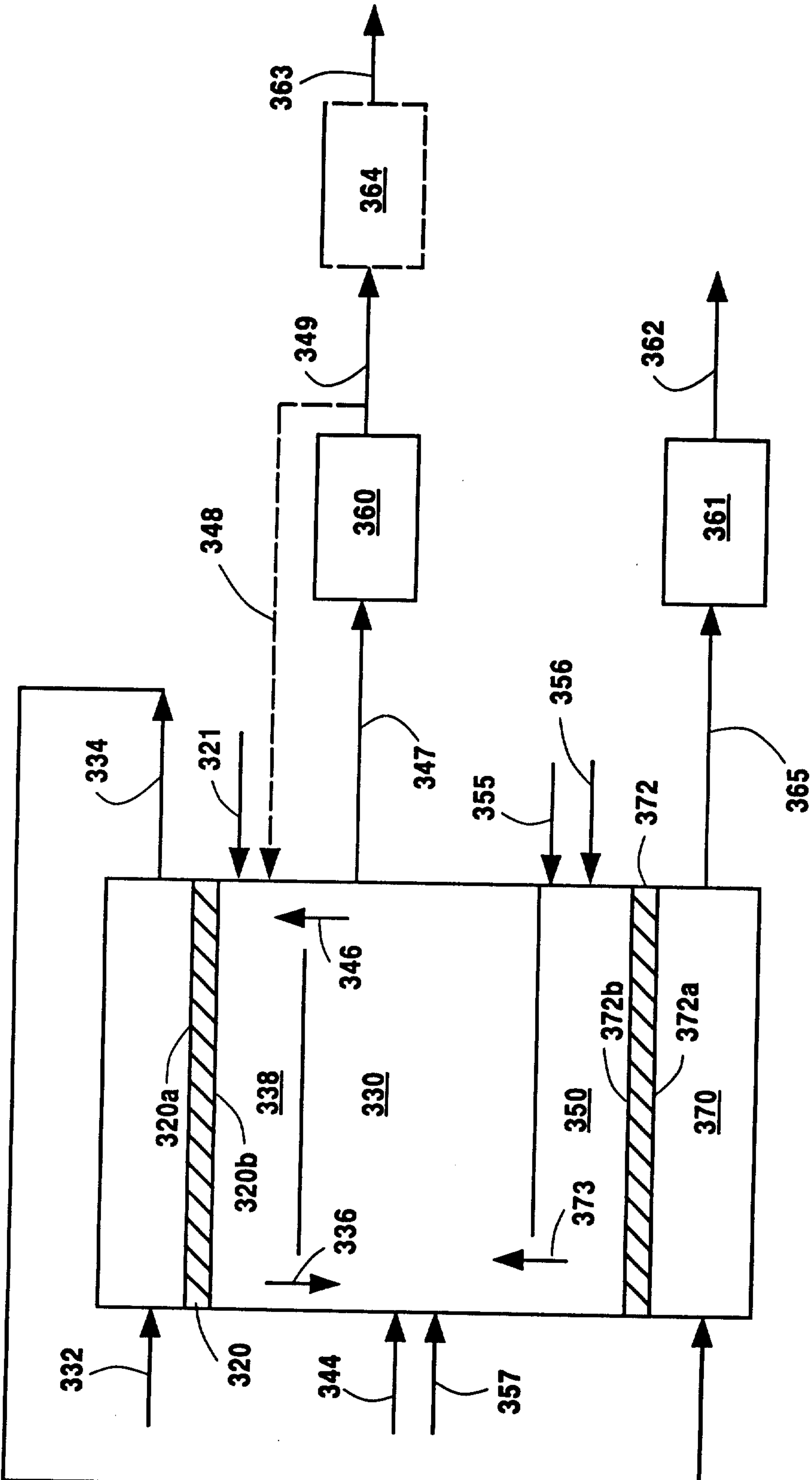


FIG. 6

