

628457

COMMONWEALTH of AUSTRALIA
Patents Act 1952

APPLICATION FOR A STANDARD PATENT

I/We

Nippon Paint Co., Ltd.

of

2-1-2, Oyodokita, Kita-ku, Osaka-shi, Osaka-fu, Japan

hereby apply for the grant of a Standard Patent for an invention entitled:

Curable composition

which is described in the accompanying complete specification.


Details of basic application(s):-

<u>Number</u>	<u>Convention Country</u>	<u>Date</u>
331572/1988	Japan	28 December 1988
220605/1989	Japan	28 August 1989

The address for service is care of DAVIES & COLLISON, Patent Attorneys, of 1 Little Collins Street, Melbourne, in the State of Victoria, Commonwealth of Australia.

DATED this TWENTY EIGHTH day of DECEMBER 1989

To: THE COMMISSIONER OF PATENTS


.....
a member of the firm of
DAVIES & COLLISON for
and on behalf of the
applicant(s)

Davies & Collison, Melbourne

COMMONWEALTH OF AUSTRALIA
PATENTS ACT 1952

DECLARATION IN SUPPORT OF CONVENTION OR
NON-CONVENTION APPLICATION FOR A PATENT

Insert title of invention.

In support of the Application made for a patent for an invention
entitled: "CURABLE COMPOSITION"

Insert full name(s) and address(es)
of declarant(s) being the appli-
cant(s) or person(s) authorized to
sign on behalf of an applicant
company.

I Kazuo Sasaki
~~we~~
of Nippon Paint Co., Ltd.
of 2-1-2, Oyodokita, Kita-ku,
Osaka-shi, Osaka-fu,
Japan.

Cross out whichever of paragraphs
1(a) or 1(b) does not apply

1(a) relates to application made
by individual(s)
1(b) relates to application made
by company; insert name of
applicant company.

do solemnly and sincerely declare as follows :-

1. (a) ~~I am the applicant for the patent~~
or (b) I am authorized by

Nippon Paint Co., Ltd.

Cross out whichever of paragraphs
2(a) or 2(b) does not apply

2(a) relates to application made
by inventor(s)
2(b) relates to application made
by company(s) or person(s) who
are not inventor(s); insert full
name(s) and address(es) of inven-
tors.

the applicant..... for the patent to make this declaration on ^{its} ~~his~~ behalf.

2. (a) ~~I am the actual inventor of the invention~~
or (b) 1. Hiroshi MIWA

- of 753-8, Azatatumikakiuchi, Minamino, Itami-shi, Hyogo.
- 2. Yoshitaka OKUDE
of 23-14, Nagisaminami-cho, Hirakata-shi, Osaka.
- 3. Masakazu WATANABE
of 2-3-4, Katsube, Toyonaka-shi, Osaka.
- 4. Sakuichi KONISHI
of 1335-14, Komyo-cho, Ikoma-shi, Nara.

All of Japan

~~is~~ are the actual inventor..... of the invention and the facts upon which the applicant.....
~~is~~ is entitled to make the application are as follows :-

State manner in which applicant(s)
derive title from inventor(s)

The actual inventors assigned the invention to
the said applicant.

Cross out paragraphs 3 and 4
for non-convention applications.
For convention applications,
insert basic country(s) followed
by date(s) and basic applicant(s).

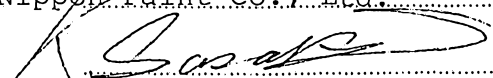
3. The basic application.S..... as defined by Section 141 of the Act ~~was~~ made
in Japan on the ..28.. December 1988.....
by Nippon Paint Co., Ltd.....
in Japan on the ..28.. August 1989.....
by Nippon Paint Co., Ltd.....
in on the
by.....

4. The basic application.S..... referred to in paragraph 3 of this Declaration ~~was~~ were
the first application.S..... made in a Convention country in respect of the invention the subject
of the application.

Insert place and date of signature.

Declared at Osaka-fu this 15th day of December 1989.

Signature of declarant(s) (no
attestation required)

Nippon Paint Co., Ltd.


Note: Initial all alterations.

Kazuo SASAKI (president)
DAVIES & COLLISON, MELBOURNE and CANBERRA.

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(19) AUSTRALIAN PATENT OFFICE (10) Acceptance No. 628457

(54) Title
CURABLE COMPOSITION

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(57) Claim

1. A curable composition comprising

(A) a compound having a carboxylic anhydride group wherein if said compound is a copolymer, the number average weight of the copolymer is between 500 and 40,000;

(B) an oxazolidine compound wherein said oxazolidine compound is present in the composition in such an amount that an amino group produced from the oxazolidine compound is 0.2 to 3 equivalents based on one equivalent of the carboxylic anhydride group; and

(C) a polymer containing an alkoxysilyl group formed from 20 to 100 percent of silane compounds and having a molecular weight of 500 to 40,000 wherein said polymer is present in the composition in such an amount that an alkoxysilyl group is 0.2 to 10 equivalent based on 1 mol of an oxazolidine group.

2. A curable composition according to claim 1 wherein said compound (A) is a polymeric compound or a monomeric compound.

628+57

COMMONWEALTH OF AUSTRALIA
PATENTS ACT 1952
COMPLETE SPECIFICATION

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COMPLETE SPECIFICATION FOR THE INVENTION ENTITLED:

Curable composition

The following statement is a full description of this invention, including the best method of performing it known to me/us:-

FIELD OF THE INVENTION

The present invention relates to a curable
5 composition superior in solvent resistance, chemical
resistance and durability, in particular to a curable
composition suitable for paint.

BACKGROUND OF THE INVENTION

Melamine resins and polyisocyanate compounds have
10 been used as curing agents for thermosetting paints but they
showed problems in chemical resistance, toxicity to human body
and the like.

In order to solve these problems, the use of
blocked amines has been proposed, but it is difficult to
15 obtains curable compositions having superior properties.
For example, Japanese Kokai Publication (unexamined)
discloses a hydraulic composition which comprises acid
anhydride polymers and polyoxazolidines, but this
composition is insufficient in chemical resistance and
20 weather resistance and does not exhibit a superior finishing
appearance.

SUMMARY OF THE INVENTION

The present invention provides a curable
composition which is superior in chemical resistance and
25 weather resistance and appearance by the use of oxazolidine
compounds. Accordingly, the present invention provides a
~~curable composition which comprises (A) a compound having a~~



curable composition comprising

(A) a compound having a carboxylic anhydride group wherein if said compound is a copolymer, the number average weight of the copolymer is between 500 and 40,000;

(B) an oxazolidine compound wherein said oxazolidine compound is present in the composition in such an amount that an amino group produced from the oxazolidine compound is 0.2 to 3 equivalents based on one equivalent of the carboxylic anhydride group; and

(C) a polymer containing an alkoxysilyl group formed from 20 to 100 percent of silane compounds and having a molecular weight of 500 to 40,000 wherein said polymer is present in the composition in such an amount that an alkoxysilyl group is 0.2 to 10 equivalent based on 1 mol of an oxazolidine group.

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~~carboxylic anhydride group, (B) an oxazolidine compound and~~
~~(C) a polymer containing an alkoxysilyl group.~~ If an epoxy
compound (D) is further added to this composition, the solid
content of the composition is increased and other properties
5 are improved.

Component (A)

The compound having a carboxylic anhydride group
used in the present invention can be anyone which are used
in this field. Examples of the compounds are low molecular
10 weight acid anhydrides, such as maleic anhydride, phthalic
anhydride, hydrogenated phthalic anhydride, 4-methyl-
hydrogenated phthalic anhydride, pyromellitic anhydride,
triglyceride of pyromellitic anhydride, ethyleneglycol ester
of trimellitic anhydride or mixtures thereof; high molecular
15 weight acid anhydrides, such as copolymers of ethylenic acid
anhydrides (e.g. itaconic anhydride and maleic anhydride)
and other copolymerizable monomers; and the like. Examples
of the other copolymerizable monomers are styrene, α -
methylstyrene, esters of acrylic acid (for example methyl
20 acrylate, ethyl acrylate, butyl acrylate or 2-ethylhexyl
acrylate), esters of methacrylic acid (for example methyl
methacrylate, ethyl methacrylate, butyl methacrylate,
isobutyl methacrylate, t-butyl methacrylate, 2-ethylhexyl
methacrylate or lauryl methacrylate), acrylamide,
25 methacrylamide and the like. The copolymerization is
conducted by the known methods, for example the radical
polymerization method and the like. A number average



molecular weight of the copolymer is within the range of 500 to 40,000, preferably 1,000 to 20,000. If the number average molecular weight is larger than 40,000, a viscosity of the copolymers is increased, whereby a disadvantage occur in that a high solid thermosetting composition is hard to obtain. If the number average molecular weight is less than 500, the curing properties are insufficient. The molecular weight is determined by the GPC method. An acid value of the high molecular weight acid anhydride is preferably within the range of 30 to 500, preferably 50 to 350. If the compound having a carboxylic acid anhydride group (hereinafter referred to as "acid anhydride") is classified into monobasic-, dibasic- and tribasic compounds on the basis of number of acid anhydride groups, the use of polybasic acid anhydrides other than monobasic acid anhydrides is preferable because the enhanced curing properties can be obtained. The preferable acid anhydride include an ethyleneglycol ester of trimellitic anhydride, a triglyceride of trimellitic anhydride and a hydrogenated one thereof or a high molecular weight acid anhydride.

Component (B)

The oxazolidine compound employed in the present invention is obtained by a reaction between an alkanolamine compound containing a primary amine or secondary amine and a ketone or aldehyde. The "oxazolidine" according to the present invention may include an oxazolidine which has 5-member ring and a tetrahydro-oxazine which has 6-member



ring. This compound can be obtained by the dehydration reaction of polyfunctional alkanolamines and ketones or aldehydes. The polyfunctional alkanolamines are prepared by a reaction of an alkanolamine, such as mono- or

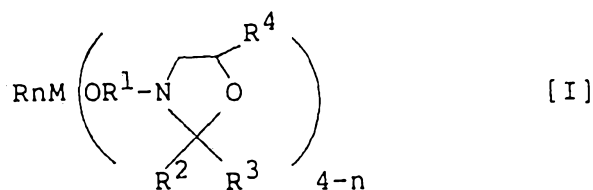
5 diethanolamine, mono- or dipropanolamine, mono- or diisopropanolamine and ¹N-methylethanolamine, and both-end primary amines (for example, ethylenediamine, hexamethylenediamine and isophorone diamine) and epoxy compounds (for example, ethylene oxide, butylene oxide and

10 the like). Examples of the ketones or aldehydes are acetone, methylethylketone, methylisobutylketone, cyclohexanone and isophorone, benzaldehyde, isobutylaldehyde, acetoaldehyde, propionaldehyde and n-butylaldehyde. In addition, the polyfunctional oxazolidine

15 compounds can be obtained by the reactions of an oxazolidine compound having a hydroxylic group, which are obtained by the dehydration of alkanolamines (such as diethanolamine and diisopropanolamine) by means of the above described ketones and aldehydes, with various kinds of reactive compounds or

20 acid chlorides. The reactive compounds include isocyanate compounds, for example, hexamethylene-diisocyanate, isophronediisocyanate, 1,3-(isocyanatemethyl)cyclohexane and the like. In addition, the acid chlorides include azipoyl dichloride, sebacoyl dichloride, decaloyl dichloride, terephthaloyl dichloride and the like. The oxazolidine

25 compound of the present invention also includes a compound represented by the following formula [I]:



5 wherein R is a hydrogen atom, an alkyl group, an aryl group, a vinyl group, a halogenated alkyl group or alkoxylic group containing 1 to 10 carbon atoms; n is an integer of 0 to 3; M is a silicon or titanium atom; R¹ is an alkyl group containing 2 to 5 carbon atoms; R² and R³ is a hydrogen
10 atom, an alkyl group or an aryl group containing 1 to 10 carbon atoms, respectively; and R⁴ is a hydrogen atom or a methyl group.

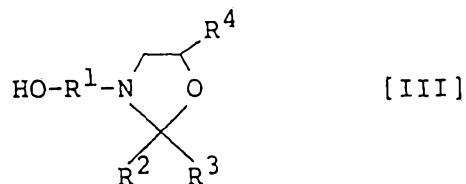
The oxazolidine compounds represented by the above described formula [I] can be obtained by the reaction

15 between compounds represented by the following formula [II]:

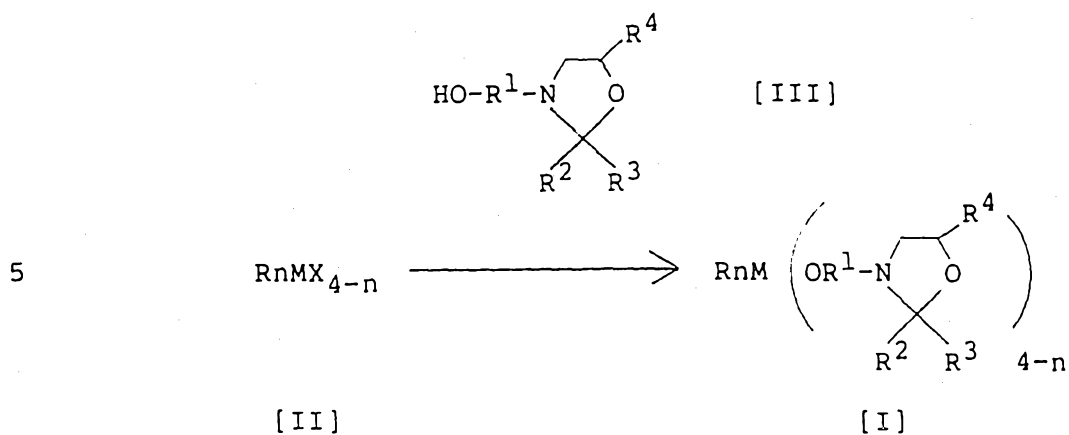


wherein R, n and M are same as in the formula [I]; X is a halogen atom,

and hydroxyoxazolidine compounds represented by the
20 following formula [III];



25 wherein R₁ to R₄ are same as in the formula [I]. This synthetic reaction is represented by the following equation:



10 wherein R, n, M, X and R¹ to R⁴ are same as in the formulae [I] and [II].

The compounds represented by the above described formula [II] include halogenated silane compounds or halogenated titanium compounds. The halogenated silane compounds include methyltrichlorosilane, 15 methyldichlorosilane, dimethyldichlorosilane, phenyltrichlorosilane, diphenyldichlorosilane, tetrachlorosilane, vinyltrichlorosilane, chloromethyltrichlorosilane, ethyltrichlorosilane, dimethoxymethylchlorosilane, diethoxydichlorosilane, 20 triethoxychlorosilane and the like. The halogenated titanium compounds include tetrachlorotitanium and the like.

The hydroxyoxazolidine compounds represented by the above described formula [III] are obtained by the usual oxazolidine-synthesizing reaction, as mentioned later. 25 Concrete examples of the hydroxyoxazolidine compounds [III] are 2-isopropyl-3-(2-hydroxypropyl)-5-methyloxazolidine,



2,2-dimethyl-3-(2-hydroxypropyl)-5-methyloxazolidine, 2-phenyl-3-(2-hydroxypropyl)-5-methyloxazolidine, 2-isopropyl-3-(2-hydroxyethyl)oxazolidine, 2-methyl-2-isobutyl-3-(2-hydroxyethyl-oxazolidine, 3-(2-hydroxypropyl)-5-methyloxazolidine and the like.

According to the method of the present invention, the above described compound [II] are reacted with the above described compound [III] for 1 to 5 hours at a reaction temperature of less than 30 °C in a nonpolar solvent in the presence of bases (e.g. pyridine and triethylamine) for trapping hydrochloric acid to obtain the compound [I]. There is the possibility that if the reaction temperature is higher than 30 °C, the formed compounds [I] are decomposed, so that the reaction temperature of less than 30 °C is preferable.

An amount of the component (B) in the composition is 0.2 to 3 equivalent, preferably 0.5 to 2 equivalent, as measured as an amino group based on 1 equivalent of a carboxylic anhydride group in the component (A). If the amount of the component (B) is less than 0.2 equivalent, a good crosslinking degree is not obtained and thus the solvent resistance is insufficient. If the amount of the component (B) exceeds 3 equivalent, the weather resistance is reduced.

Component (C)

The polymers containing an alkoxysilyl group can be obtained by various kinds of method. A simple method is a

method (1) in which silane compounds having a polymerizable double bond and an alkoxylic group (e.g. 3-methacryloxypropyldimethoxymethylsilane and 3-methacryloxypropylmethoxysilane) are homopolymerized or
5 copolymerized with other polymerizable compounds. Examples of the other polymerizable compounds are styrene monomers, α -methylstyrene and esters of acrylic acid or methacrylic acid (e.g. methylmethacrylate, methylacrylate, ethylmethacrylate, ethylacrylate, butylmethacrylate,
10 butylacrylate, 2-ethylhexylmethacrylate and 2-ethylhexylmethacrylate). There also propose a method (2) in which polymers are modified with silane compounds. That is, allylmethacrylate (or allylacrylate) is copolymerized with the above described other polymerizable monomers, and then
15 silane compounds having a Si-H group, such as dimethoxysilane, trimethoxysilane, triethoxysilane and dimethoxymethylsilane, are introduced into the copolymers by a silyl-hydration reaction in the presence of platinum-containing catalysts; a method (3) in which compounds having
20 an unsaturated group and an isocyanate group are copolymerized with the above described other copolymerizable monomers, and then alkoxy silane compounds having -NH- (or -NH₂), such as 3-aminopropyltrimethoxysilane, 3-aminopropyldimethyl-ethoxysilane, 3-
25 allylaminopropyltrimethoxysilane and 3-aminopropyltriethoxysilane compounds, are introduced into the copolymer by reacting them with the isocyanate group in

the copolymers; or in which alkoxysilane compounds having a -SH group, such as mercaptomethyltrimethoxysilane, dimethoxy-3-mercaptopropylmethylsilane and 3-mercaptopropyltrimethoxysilane, are introduced in the copolymers by reacting them with the isocyanate group in the copolymers; and a method (4) in which for example glycidylmethacrylate (or methacrylic acid) is reacted with polymers containing a carboxylic group (or a glycidyl group) to introduce an unsaturated double bond into the polymers and the above described alkoxysilyl group containing -NH₂ group is introduced into the polymers by the Michel reaction. Of these methods, in particular the methods (1) and (2) are simple and advantageous.

An amount of the silyl compounds in the copolymer is not specially limited, but preferably 20 to 100 %. If it is less than 20 %, a problem occurs in curing properties. A molecular weight of this polymer is 500 to 40,000, preferably 1,000 to 20,000. If it is less than 500, a problem occurs in curing properties and has disadvantages in shedding and the like. If it exceeds 40,000, a high solid paint is hard to obtain and disadvantages occur in compatibility.

An amount of the alkoxysilyl group in the composition is 0.2 to 10 equivalents, preferably 0.5 to 5 equivalents, based on 1 mol of an oxazolidine group. If it is 0.2 equivalent or less, the curing properties are inferior and if it is $\frac{10}{k}$ equivalent or more, the water

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resistance is inferior.

Component (D)

Compounds having at least one oxysilane group are used as the epoxy compounds. Such the epoxy compounds include glycidyl ethers, such as ethyleneglycol diglycidyl ether, propylene glycol diglycidyl ether, 1,6-hexanediol glycidyl ether, trimethylolpropane triglycidyl ether, bisphenol A diglycidyl ether and hydrogenated bisphenol A-diglycidyl ether; glycidyl esters, such as diglycidyl ester of adipic acid, diglycidyl ester of hydrogenated phthalic acid, diglycidyl ester of phthalic acid, triglycidyl ester of trimellitic acid and the like; alicyclic epoxy compounds, such as 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane carboxylate, bis-(3,4-epoxycyclohexyl)adipate, vinylcyclohexene dioxide; and the like. Also, products on the market, for example ERL-4221 (manufactured by UCC Corporation), ERL-4299 (manufactured by UCC Corporation), Epolite 1600 (manufacture by Kyoisha Oils and Fats Co., Ltd.), Epolite 4000 (manufactured by Kyoisha Oils and Fats Co., Ltd.) and the like, may be used. The component (D) can be contained in the composition in such an amount that an epoxy equivalent is 0.2 to 3 equivalent, preferably 0.5 to 1.0 equivalent, based on 1 mol of a carboxylic acid anhydride group in the above described component (A). If it is less than 0.2 equivalent, the curing properties are poor and if it exceeds 3 equivalent, weather resistance is reduced.

Other components

The curable composition according to the present invention may be used as a coating composition. The above described components may be mixed to form a clear coating composition. In addition, they may be mixed together with pigments to form paints. The modified polymer of the acid anhydride group, for example when modified with diethylaminoethanol, contains an acid and an amine in a molecule, whereby being amphotericized and exhibiting a secondary effect that the dispersion of pigments is improved. Any conventional pigments may be used, for example, iron oxide, lead oxide, strontium chromate, carbon black, coal dust, titanium dioxide, talc, barium sulfate, cadmium yellow, cadmium red, chromium yellow, metallic pigments (such as, aluminum flake) and the like. Also organic pigments may also be used. That is to say, for example phthalocyanine blue and the like may be used. Also pearl mica and the like may be used. An amount of the pigments contained in the paint is generally represented by a ratio by weight of the pigments to nonvolatile components in the coating composition. According to the present invention, the ratio of the pigments to nonvolatile components in the coating composition may be as high as 2 : 1, in many cases 0.05 to 1 : 1.

The coating composition of the curable composition may generally contain a liquid diluent. The liquid diluent is a solvent or nonsolvent which is removed after applying

and which reduces the viscosity so that it is coated in a controlled thickness by a general coating method, such as spray coating and the like. The liquid diluent also assists wetting properties of a substrate, miscibility, package stability, coalescent properties and film-forming properties. Suitable examples of the diluents are an aromatic hydrocarbon, such as toluene and xylene; a ketone, such as methyl ethyl ketone and methyl isobutyl ketone; an secondary or tertiary alcohol, such as isopropyl alcohol and sec-butyl alcohol; a monoether of glycol, such as ethyleneglycol monoether and diethyleneglycol monoether; a monoether glycol acetate, such as 2-ethoxyethyl acetate; and a suitable mixture thereof. The diluent may be present in the coating composition in an amount of up to 60 % by weight, generally 20 to 55 % by weight based on the total amount of the diluent and the nonvolatile content of the coating composition.

Beside the above components, an additive, such as filler, a plasticizer, an antioxidation agent, a ultraviolet absorber, a flow controlling agent, a surfactant and the like can be added, if desirable. An amount of the additive is varied widely, generally up to 10 % by weight based on the nonvolatile content of the coating composition.

The coating composition may be applied on a substrate by spray coating, blushing, dipping, roll coating, flow coating and the like. The substrate to be employed includes wood, metal, glass, fabric, plastics, foam or

various primer-coated substrate. The coating composition is very suitable for plastics or metal, such as steel and aluminum. A thickness of film can be varied, but generally is 0.5 to 3 mil. After the coating composition has been applied, it is cured. Curing can be carried at a temperature of 100 to 200 °C, preferably 120 to 180 °C for 10 to 30 minutes.

The curable composition according to the present invention may be used for molding and other applications.

According to the present invention, the polymer having an alkoxysilyl group is added to the compound having an acid anhydride group and the oxazolidine compound to give the improved solvent resistance, chemical resistance and durability to the curable composition. In addition, if the composition is coated, the good flatness is obtained. It seems that it is the reason why such the superior effects are obtained that a hydroxylic group formed by a hydrolysis of oxazolidine is reacted with the polymer containing an alkoxysilyl group and the acid anhydride group to form crosslinks. In the case where the epoxy compound (D) is contained, it seems that this epoxy compound further contributes to the crosslink.

EXAMPLES

The present invention will be illustrated by the following examples which, however, are not construed as limiting the present invention to their details.

Production Example 1

120 parts by weight of butyl acetate was charged in a reaction vessel having a capacity of 1 l provided with a thermometer, a stirrer, a condenser and a nitrogen-introducing pipe, and heated heated up to 100 °C. A
5 solution containing 21 parts by weight of n-butyl acrylate, 95 parts by weight of n-butyl methacrylate, 34 parts by weight of 2-ethylhexyl methacrylate, 45 parts by weight of itaconic anhydride, 60 parts by weight by dioxane and 20 parts by weight of t-butylperoxyhexanoate was added dropwise
10 for 3 hours and stirring was continued for 2 hours to obtain an acrylic resin solution having a nonvolatile content of 51 % and a number average molecular weight of 2,500.

Production Examples 2 to 5

Synthesis of Polymers II to V Containing a
15 Carboxylic Anhydride Group

Polymers containing a carboxylic anhydride group were obtained in the same manner as in Production Example 1 using ingredients shown in Table 1.

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Table 1

		Production Examples			
		2	3	4	5
5	Butyl acetate	100	100	100	-
	Xylene	20	-	20	120
10	Butylene	10	-	50	-
	Methyl methacrylate	-	-	35	-
	Butyl acrylate	80	80	30	165
	2-Ethylhexyl methacrylate	45	45	15	-
	Ethyl methacrylate	15	-	-	-
	Ethyl acrylate	-	15	25	-
	Itaconic anhydride	45	55	-	-
	Maleic anhydride	-	-	30	30
	Dioxane	50	70	50	50
	t-Butylperoxy-2-ethylhexanoate	7	-	-	7
15	AIBN *	-	1.5	8	-
Nonvolatile content		52	52	52	52
Number average molecular weight		7,000	15,000	4,000	7,000

* AIBN: azobisisobutyronitril

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Production Example 6

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76 g of isobutylaldehyde was added dropwise to 133 g of diisopropanolamine and 70 g of toluene charged in a reaction vessel provided with a stirrer, a condenser having a water separator and a thermometer for 1 hour with cooling by means of ice. The reaction was continued for 5 hours with separating and removing water under reflux. Subsequently, the water separator was removed and 84 g of



1,6-hexane diisocyanate and 20 g of xylene were added dropwise to the above described reaction product at room temperature for 1 hour followed by continuing the reaction for 8 hours at 70 to 80 °C to obtain a blocked amine compound I which was blocked in the form of an oxazolidine group.

Production Example 7 to 9

Blocked amine compounds II to IV were obtained under the same conditions as in Production Example 6 using ingredients shown in Table 2.

Table 2

		Production Examples		
		7	8	9
	Toluene	90	90	90
15	Diethanolamine	105	-	-
	N-methylisopropanolamine	-	89	-
	Diisopropanolamine	-	-	133
	Cyclohexanone	98	-	-
	Isobutylaldehyde	-	72	72
20	1,6-Hexane diisocyanate	168	-	-
	TAKENATE XD-170HN*	-	-	183
	Blocked amine compound	II	III	IV

* Polyisocyanate manufactured by Takeda Chemical Industries, Co., Ltd.

Production Example 10

Synthesis of Blocked Amine Compound V

72 g isobutylaldehyde was added dropwise to 70 g of toluene and 75 g of 3-amino-1-propanol charged in the same apparatus as in Production Example 6 at room temperature within 1 hour. The reaction was continued for further 5
5 hours with separating and removing water under reflux to obtain tetrahydro-1,3-oxazone compound. Subsequently, the water separator was removed and 99 g of ethyleneglycol dimethacrylate and 20 g of xylene were added to the above described reaction product followed by the reaction for 48
10 hours at 60 to 70 °C to obtain a blocked amine compound V.

Production Example 11

Synthesis of a polymer I Containing an Alkoxysilyl Group

200 g of xylene was charged in a reaction vessel
15 similar to that in Production Example 1 and heated up to 120 °C. A solution containing 150 g of 3-methcryloxypropyltrimethoxysilane, 20 g of n-butyl acrylate, 30 g of methyl methacrylate and 15 g of t-butylperoxyhexanoate was added dropwise to the reaction
20 vessel for 3 hours followed by continuing the reaction for further 2 hours to obtain a polymer I containing an alkoxysilyl group (nonvolatile content = 52 % and number average molecular weight = 2,000).

Production Example 12

25 Synthesis of a Polymer II Containing an Alkoxysilyl Group

A polymer containing an alkoxysilyl group II

(nonvolatile content = 51 % and number average molecular weight = 12,000) was obtained from 200 g of xylene, 100 g of 3-methacryloxypropyldiethoxymethylsilane, 25 g of n-butyl acrylate, 50 g of styrene, 25 g of methyl methacrylate and 3
5 g of t-butylperoxyhexanoate in the same manner as in Production Example 11.

Production Example 13

Synthesis of a Polymer III Containing an Alkoxysilyl Group

10 200 g of xylene was charged in the same reaction vessel as in Production Example 11 and heated up to 120 °C. A solution containing 80 g of isocyanatoethyl methacrylate, 50 g of n-butyl acrylate, 40 g of styrene, 30
15 g of ethyl methacrylate and 10 g of t-butylperoxyhexanoate was added dropwise to the reaction vessel for 3 hours followed by continuing the reaction for further 2 hours to obtain a polymer containing isocyanate groups. 115 g of 3-aminopropyltriethoxysilane was added thereto dropwise for 1
20 reaction for further 5 hours at room temperature to obtain a polymer III containing an alkoxysilyl group (nonvolatile content = 61 % and number average molecular weight = 4,000).

Example 1

25 A clear paint was prepared from the following ingredients and the resulting coated film was evaluated in performance.

	Compound I containing a carboxylic acid anhydride	130 g
	Blocked amine compound I	25 g
	Compound I containing an alkoxyethyl group	1.5 g
	Tinuvin 900	1.5 g
5	Ilganox 1010	1.0 g

The obtained clear paint was diluted with a thinner comprising butyl acetate and xylene in a ratio of 1 : 1 until a paint viscosity was obtained. A phosphated steel panel was coated with Powertop U-30 and Orga P-2
10 manufactured by Nippon Paint Co., Ltd. and then coated with a metallic base Superlac H-90 manufactured by Nippon Paint Co., Ltd. The resultant panel was then coated with the above mentioned clear paint in a wet-on-wet manner and baked
15 for 30 minutes at 140 °C. The results of the evaluation of the coated film are shown in Table 4.

Example 2 to 6

Clear paints were obtained from the ingredients shown in Table 3 in the same manner as in Example 1. The results of the evaluation of the coated film are shown in
20 Table 4.

Comparative Example 1

A clear paint was obtained from the ingredients shown in Table 3. The results of the evaluation of the coated film are shown in Table 4.
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Table 3

	Polymer containing carboxylic anhydride	Blocked amine compound	Polymer containing alkoxyethyl group	Additives
Example 2	Pro. Ex. 2, 130 g	Pro. Ex. 7, 35 g	Pro. Ex. 11, 60 g	Seasope 103 1.0 g Sanol LS-292 0.8 g
Example 3	Pro. Ex. 3, 110 g	Pro. Ex. 8, 30 g	Pro. Ex. 12, 80 g	Tinuvin 900 1.0 g Sanol LS-292 1.0 g
Example 4	Pro. Ex. 4, 150 g	Pro. Ex. 9, 23 g	Pro. Ex. 11, 30 g	Tinuvin 900 1.0 g Sanol LS-292 1.0 g
Example 5	Pro. Ex. 5, 145 g	Pro. Ex. 9, 20 g	Pro. Ex. 11, 40 g	Tinuvin 900 1.0 g Sanol LS-292 1.0 g
Example 6	Pro. Ex. 1, 130 g	Pro. Ex. 10, 45 g	Pro. Ex. 11, 40 g	Tinuvin 900 1.2 g Sanol LS-292 0.8 g
Comparative Example 1	Pro. Ex. 5, 145 g	Pro. Ex. 9, 20 g	-	Tinuvin 900 1.0 g Sanol LS-292 1.0 g



Table 4

	Chemical resistance		Solvent resistance	
	Acid	Alkali	XL rubbing	MEK spot
Example 1	o	o	o	o
Example 2	o	o	o	o
Example 3	o	o	o	o
Example 4	o	o	o	o
Example 5	o	o	o	o
Example 6	o	o	o	o
Comparative Example 1	x	o	o	Δ

Acid resistance: 6 drops of 6 % -sulfurous acid are dropped on the coated film and the change after the lapse of 24 hours is observed.

Alkali resistance: 6 drops of a 6.1 N-aqueous solution of NaOH are dropped on the coated film and the change after the lapse of 24 hours is observed.

XL rubbing: A gauze impregnated with xylene is reciprocated on the coated film 8 times and the change is observed.

MEK spot: 6 drops of methylethylketone are dropped on the coated film and the change after the lapse of 3 minutes is observed.

o: No change.

Δ: Shaded-off gloss, change of color.

x: Separation or dissolution.

Production Example 14

Synthesis of Oxazolidine Compound I

76 g of isobutylaldehyde was added dropwise to 133

g of diisopropanolamine and 70 g of toluene charged in a reaction vessel provided with a stirrer, a condenser having a water separator and a thermometer within 1 hour with cooling by means of ice. The reaction was continued for
5 further 5 hours with separating and removing water under reflux and the water separator was removed. Subsequently, 84 g of 1,6-hexanediisocyanate and 20 g of xylene were added dropwise to the above described reaction product within 1 hour at room temperature followed by conducting the reaction
10 for 8 hours at 70 to 80 °C to obtain an oxazolidineamine compound I.

Production Example 15 to 17

Synthesis of Oxazolidine Compounds II to IV

Oxazolidine compounds II to IV were obtained from
15 materials shown in Table 5 under the same conditions as in Production Example 7.

20

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Table 5

	Production Example No.			
	15	16	17	
5	Toluene	90	90	90
	Diethanolamine	105	-	-
	N-methylisopropanolamine	-	89	-
	Diisopropanolamine	-	-	133
	Cyclohexanone	98	-	-
	Isobutylaldehyde	-	72	72
10	1,6-Hexane diisocyanate	168	-	-
	Takenate XD-170HN*	-	-	183
	Oxazolidine compound	II	III	IV

* Polyisocyanate manufactured by Takeda Chemical Industries. Co., Ltd.

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Production Example 18

Synthesis of an Oxazolidine Compound V

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75 g of isobutylaldehyde was added dropwise to 200 g of benzene and 133 g of diisopropanolamine charged in the same reaction vessel as in Production Example 14 for 90 minutes at room temperature. The reaction was continued for further 5 hours with separating and removing water under reflux followed by removing benzene under a reduced pressure to obtain a hydroxyoxazolidine compound.

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After removing the water separator, 1,000 g of hexane and 103 g of triethylamine were added and then 64.6 g of dichlorodimethylsilane was added dropwise for 2 hours at

0 °C followed by stirring for further 2 hours at room temperature to deposit solid substances from the reaction solution and filtrate the resulting solid substances. The filtrate was concentrated to obtain an oxazolidine compound

5 V.

Production Example 19

Synthesis of an Oxazolidine Compound VI

An oxazolidine compound VI was obtained in the same manner as in Production Example 18 excepting that 42.5 g of
10 tetrachlorosilane was used in place of 64.6 g of dichlorodimethylsilane.

Production Example 20

Synthesis of an Oxazolidine Compound VII

15 An oxazolidine compound VII was obtained in the same manner as in Production Example 18 excepting that 134 g of dichloride dodecanate was used in place of 64.6 g of dichlorodimethylsilane.

Example 7

20 A clear paint was prepared in the following composition and the evaluation of the performance of the obtained coated film was conducted.

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	Compound I containing a carboxylic anhydride	130 g
	Oxazolidine compound V	13 g
	Compound I containing an alkoxysilyl group	65 g
5	3,4-Epoxy cyclohexylmethyl-3,4-epoxy cyclohexane carboxylate	7 g
	Tinuvin 900	1.5 g
	Ilganox 1010	1.0 g

10 A phosphated steel plate was coated as generally described in Example 1 with the exception that the above described clear paint is employed. The results of the evaluation of the coated film are shown in Table 7.

Examples 8 to 16

15 Clear paints were obtained in the compositions shown in Table 6 in the same manner as in Example 7. The results of the evaluation of the coated film are shown in Table 7.

Comparative Example 2

20 A clear paint was obtained in the composition shown in Table 6. The result of the evaluation of the coated film is shown in Table 7.

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Table 6

	Compound containing a carboxylic anhydride group	Oxazolidine compound	Polymer containing an alkoxyethyl group	Epoxy compound	Additive
Example 8	Pro. Ex. 1, 130 parts	Pro. Ex. 14, 30 parts	Pro. Ex. 11, 60 parts	*1 7 parts	Seasope 103, 1.0 parts Sanol LS-292, 0.8 parts
Example 9	Pro. Ex. 2, 130 parts	Pro. Ex. 15, 20 parts	Pro. Ex. 12, 50 parts	*1 8 parts	Tinuvin 900, 1.0 parts Sanol LS-292, 1.0 parts
Example 10	Pro. Ex. 3, 110 parts	Pro. Ex. 16, 30 parts	Pro. Ex. 13, 40 parts	*2 10 parts	Tinuvin 900, 1.0 parts Sanol LS-292, 1.0 parts
Example 11	Pro. Ex. 4, 150 parts	Pro. Ex. 17, 20 parts	Pro. Ex. 11, 90 parts	*3 6 parts	Tinuvin 900, 1.2 parts Ilganox 1010, 1.0 parts
Example 12	Pro. Ex. 1, 130 parts	Pro. Ex. 18, 45 part	Pro. Ex. 11, 60 parts	*4 8 parts	Tinuvin 900, 1.2 parts Ilganox 1010, 1.0 parts
Example 13	Pro. Ex. 3, 110 parts	Pro. Ex. 19, 25 parts	Pro. Ex. 12, 70 parts	*4 10 parts	Tinuvin 900, 1.0 parts Ilganox 1010, 1.0 parts
Example 14	Pro. Ex. 5, 145 parts	Pro. Ex. 20, 35 parts	Pro. Ex. 11, 50 parts	*1 5 parts	Tinuvin 900, 1.0 parts Ilganox 1010, 1.0 parts
Example 15	Pro. Ex. 1, 130 parts	Pro. Ex. 19, 26 parts	Pro. Ex. 11, 50 parts	*1 10 parts	Tinuvin 900, 1.0 parts Ilganox 1010, 1.0 parts
Example 16	Pro. Ex. 1, 130 parts	Pro. Ex. 19, 8 parts	Pro. Ex. 11, 70 parts	*1 12 parts	Tinuvin 900, 1.0 parts Ilganox 1010, 1.0 parts
Comparative Example 2	Pro. Ex. 5, 145 parts	Pro. Ex. 17, 20 parts	-	-	Tinuvin 900, 1.0 parts Ilganox 1010, 1.0 parts



*1: 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexane
 carboxylate

*2: 1,6-hexanediol-diglycidylether

*3: ethyleneglycoldiglycidylether

5 *4: bis(3,4-epoxycyclohexyl)adipate

Table 7

	Chemical resistance		Solvent resistance		Water-Proofness
	Acid	Alkali	XL rubbing	MEK spot	
10 Example 7	o	o	o	o	o
Example 8	o	o	o	o	o
Example 9	o	o	o	o	o
Example 10	o	o	o	o	o
Example 11	o	o	o	o	o
Example 12	o	o	o	o	o
Example 13	o	o	o	o	o
Example 14	o	o	o	o	o
Example 15	o	o	o	o	o
Example 16	o	o	o	o	o
15 Comparative Example 2	x	o	o	Δ	Δ

Acid resistance: 6 drops of a 6 %- aqueous solution of sulfurous acid are dropped on the coated film and a change after the lapse of 24 hours is observed.

Alkali resistance: 6 drops of a 6.1 N-aqueous solution of NaOH are dropped on the coated film and a change after the lapse of 24 hours is observed.

25 XL rubbing: A gauze impregnated with xylene is reciprocated on the coated film 8 times and a change is observed.

MEK spot: 6 drops of methylethylketnone are dropped on the coated film and a change after the lapse of 3 minutes is observed.

5 Water-proofness A change after the immersion in water for 10 days at 50 °C is observed.

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THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:-

1. A curable composition comprising

(A) a compound having a carboxylic anhydride group wherein if said compound is a copolymer, the number average weight of the copolymer is between 500 and 40,000;

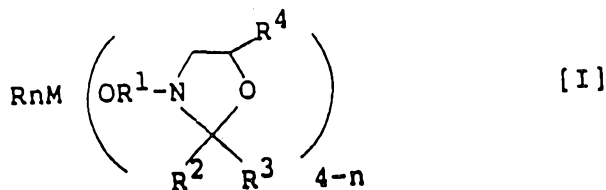
(B) an oxazolidine compound wherein said oxazolidine compound is present in the composition in such an amount that an amino group produced from the oxazolidine compound is 0.2 to 3 equivalents based on one equivalent of the carboxylic anhydride group; and

(C) a polymer containing an alkoxysilyl group formed from 20 to 100 percent of silane compounds and having a molecular weight of 500 to 40,000 wherein said polymer is present in the composition in such an amount that an alkoxysilyl group is 0.2 to 10 equivalent based on 1 mol of an oxazolidine group.

2. A curable composition according to claim 1 wherein said compound (A) is a polymeric compound or a monomeric compound.

3. A curable composition according to claim 1 or claim 2 wherein said compound (A) is an ester of trimellitic anhydride and ethylene glycol, or an ester of glycerol and trimellitic anhydride.

4. A curable composition according to any one of claims 1 to 3 wherein said oxazolidine compound (B) is represented by the following formula [I]:



wherein R is a hydrogen atom or an alkyl group, aryl group, vinyl group, halogenated alkyl group or alkoxylic group containing 1 to 10 carbon atoms; n is an integer of 0 to 3; M is a silicon or titanium atom; R¹ is an alkyl group containing 2 to 5 carbon atoms; R² and R³ is a hydrogen atom or an alkyl group or aryl group containing 1 to 10 carbon atoms, respectively; and R⁴ is a hydrogen atom or a methyl group.

5. A curable composition according to any one of claims 1 to 4 which further contains (D) an epoxy compound wherein said epoxy compound (D) is present in the composition in such an amount that an epoxy group is 0.2 to 3 equivalent based on one equivalent of a carboxylic anhydride group.

6. A curable composition according to claim 1, substantially as hereinbefore described with reference to the Examples.

DATED this 1st day of July, 1992.

NIPPON PAINT CO., LTD.

By Its Patent Attorneys

DAVIES COLLISON CAVE

