NOVEL INHIBITORS FOR USE IN THE BLACK AND WHITE DEVELOPMENT OF COLOR REVERSAL FILM

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FIG. 1

FIG. 2

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The present invention relates to novel compounds useful in chemical reactions involved in the processing of color film. More particularly, the present invention relates to compounds capable of selectively preventing negative overdevelopment.

In multi-layer photographic elements there are usually three different sensitive emulsion layers coated on one side of a single support. These emulsion layers may contain color couplers, such as described in U.S. Patent 2,322,027, issued to E. E. Jelley and P. W. Vittum on June 15, 1943, each are capable of reacting with the development product of aromatic amine developing agents on photographic development to form colored images. One of the difficulties encountered in the development of such multi-layer elements, especially when the development is carried out at high temperatures, is the over-development of the top emulsion layer, normally the blue sensitive, yellow-image forming layer, before the bottom layer is fully developed. The overdevelopment, which is caused primarily by the excessive concentration of the developer in the top layer, results in improper image density in that layer as compared to the remaining layers. Although it had been known that certain compounds could be employed to inhibit the overdevelopment of the top layer of color photographic elements, these compounds suffered from the disadvantage, although diffusing slower than the developer, of penetrating beyond the top sensitive emulsion layer and adversely affecting the image density in the next layers.

It is therefore an object of the present invention to provide a novel class of compounds. It is another object of the present invention to provide a novel class of compounds which act as inhibitors in the development of the top layer of multi-layer photographic element without affecting the other layers in the element. A further object of the present invention is to provide a class of compounds which aid in the development of color images of uniform characteristics in all layers as compared to the remaining layers. A still further object is to provide a method for permitting high temperature development of multi-layer color elements. Other objects will be apparent hereinafter.

In accordance with the present invention it was discovered that 5-mercaptotetrazoles having the general formula

wherein M is a member of the class consisting of alkali metals and hydrogen and X is a radical having a formula selected from the group consisting of —SO₃R and —COR, wherein R is a hydrocarbyl radical of 1 to 22 carbon atoms, act as selective inhibitors in the development of multi-layer photographic elements. The hydrocarbyl radical "R" in the novel 5-mercaptotetrazoles may be an alkyl group, an aryl group, an aralkyl group, or an alkaralkyl group. The hydrocarbyl group may further contain substituents such as alkoxy, phenoxy, halogen, cyano, nitro, amino, substituted amino, sulfamyl, substituted sulfamyl, sulfonylphenyl, sulfonylethyl, fluosulfonyl, sulfonamidophenyl, sulfonamidoalkyl, carbosylic acid carboxylate carbonyl, carbamylphosphoryl, carbamylalkyl, carbonylalkyl, carbonylalkylamino, and similar groups. It is to be understood that the term "hydrocarbyl" as employed in the present invention is used as including the aforesaid and similar substituent groups. Many of the aforesaid functional substituents are useful in increasing the solubility of the novel mercaptotetrazole.

The location of the —NHₓ group on the phenyl ring of the 1-phenyl-5-mercaptotetrazole is not critical and may be ortho, para, or meta.

The novel compounds are produced by the reactions of the 1-(aminophenyl)-5-mercaptotetrazole hydrochloride with an acid chloride having the formula RO(O)Cl, where "R" has the above indicated meaning. The 1-(aminophenyl)-5-mercaptotetrazole is prepared by reaction of an aminocetanilide, (ortho-, meta-, or para-), with thiophenone to result in the formation of the corresponding acetylamidophenyl iso(thio)cyanate which is then reacted with an azide to form the mercaptotetrazole ring. This product is then hydrolyzed with concentrated hydrochloric acid to form the 1-(aminophenyl)-5-mercaptotetrazole hydrochloride. The formation of the mercaptotetrazole ring has been described by Stolle and Strittmatter in the Journal fuer Praktische Chemie, vol. 137, pages 60 to 64 and by Stolle and Fr. Henke-Stark in the same journal, vol. 124, pages 261-300. An alternate method for the production of the substituted phenyl-5-mercaptotetrazole is to react an acetylamidophenyl substituted thiosemicarbazide with nitrous acid as described by Freund and Hempel in Berichte der Deutschen Chemischen Gesellschaft, vol. 28 at page 77.

Acid chlorides which may be employed in the reaction with the mercaptotetrazole hydrochloride include the chlorides of the following acids: acetic, aceto-acetic, dichloroacetic, propionic, valeric, iso-valeric, caproic, caprylic, undecanonic stearic, benzoic, o-acetoxybenzoic, anthranilic, 3-amino-6-nitrobenzoic, p-tolyl acetic, 2,4-dinitrophenyl-acetic, methylsulfonic, o-toluene sulfonic, benzene sulfonic, p-bromobenzene sulfonic and o-formylbenzene sulfonic acid.

The compounds of the present invention are also employed in connection with the color processing of multi-layer photographic elements such as described in the Jelley and Vittum patent, U.S. 2,322,027. The sensitive elements in that type of photographic element comprise a support of cellulose esters, paper or a suitable base having coated thereon a red sensitive emulsion layer containing a cyanos-forming coupler, a green sensitive emulsion layer containing a magenta-forming coupler and a blue sensitive emulsion layer containing a yellow-forming coupler. The couplers may be of the composition and incorporated in the manner described in the aforesaid Jelley and Vittum patent.

The compounds of the present invention are however not limited in their use to the described multi-layer element. Thus, the photographic element may contain fewer or more sensitive emulsion layers and may also contain filter interlayers, subbing layers, backing layers and overcoating layers applied in the well known manner. One or more of the emulsion layers may also be free of incorporated couplers, the coupling agent or couplers for these layers being contained in the color developing solution.

One of the advantages of the compounds of the present invention when employed as developed agents is that they are useful in high temperature processing. Thus, photographic elements may be developed at temperatures of 140° to 160° F. when using the compounds of the present invention.

In one embodiment of the present invention the 1-(aminophenyl)-5-mercaptotetrazole is applied during processing to the image-exposed multi-layer color photo-
graphic element so that this compound is present in the top silver halide emulsion layer (but not in the middle or bottom layers) during the negative development step of the color process. This is advantageously accomplished by including the 5-mercaptotetrazole of Formula I in the prehardener solution, in the hydroxylamine sulfate bath following the prehardener or in the negative developer solution or in combinations of these. The compounds of Formula I are used over a wide range of concentrations, i.e., from about 0.001 g./l. to about 1 g./l.

The optimum amount will depend upon many factors, such as, the particular compound of Formula I, the processing solution compositions, developer activities, processing temperature, the characteristics of the multi-layer color element and others. The optimum amount of our compound of Formula I for use according to our invention can be readily determined by methods well known in the art.

The prehardener solution used according to our invention comprises formaldehyde, and a compound of Formula I. Succinaldehyde is used to advantage in our prehardener solutions.

The present invention is further illustrated by the following examples:

**Example I**

A solution of 18.7 g. (0.1 mole) of m-aminooacetanilide hydrochloride in 100 ml. of water was added to a suspension of 25 g. (0.22 mole) of thiophosgene in 100 ml. of water. The solid isothiocyanate was collected by filtration and air dried.

This solid, 19.5 g. (0.25 mole) of sodium azide and 50 ml. of ethanol were added to 500 ml. of water. The solution was refluxed for three hours, cooled and acidified with concentrated hydrochloric acid. The resulting solid 5-mercaptotetrazole was collected and dried.

This product was suspended in 500 ml. of a 50-50 water, concentrated hydrochloric acid mixture and boiled until complete solution was obtained. On cooling white needles separated out and were collected. The product was identified as the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride.

To a suspension of 2.3 g. (0.01 mole) of 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride in 20 ml. of acetonitrile, 1.6 g. (0.02 mole) of pyridine was added. To the resulting solution, 0.78 g. (0.01 mole) of acetyl chloride was added and the mixture was heated at 60°C. for one hour. The temperature was then raised to boiling and water was added until clouding occurred. On cooling the solid 1-(3-acetamidophenyl)-5-mercaptotetrazole, M.P. 201-2°C., precipitated out of the mixture and was isolated by filtration and then dried.

**Example II**

Employing the procedure of Example I, 1-(3-caproamido)phenyl-5-mercaptotetrazole, M.P. 188-9°C. was prepared from the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride by reaction with capryloy chloride.

**Example III**

Employing the procedure of Example I, 1-(3-pelargonamido)-phenyl-5-mercaptotetrazole, M.P. 172-3°C. was prepared from the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride by reaction with pelargonyl chloride.

**Example IV**

Employing the procedure of Example I, 1-(3-alpha-(2,4-di-t-amlyphenoxo)acetamidophenyl)-5-mercaptotetrazole, M.P. 164-6°C. was prepared from the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride by reaction with alpha-(2,4-di-t-amlyphenoxo)-acyetyl chloride.

**Example V**

Employing the procedure of Example I, 1-(3-methyl sulfonamido)-phenyl-5-mercaptotetrazole, M.P. 163°C. (dec.), was prepared from the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride by reaction with methyl sulfonfyl chloride.

**Example VI**

Employing the procedure of Example I, 1-(3-p-toluene sulfonamido)phenyl-5-mercaptotetrazole, M.P. 140-3°C. was prepared from the 1-(3-aminophenyl)-5-mercaptotetrazole hydrochloride by reaction with p-toluene sulfonfyl chloride.

The foregoing examples have illustrated one of the preparative methods employed to produce the novel compounds of the present invention. It will be apparent that the same method can be employed using the other acid chlorides described hereinafore, as well as any other acid chloride whose use extends the scope of the definition of the novel compounds of the present invention.

The use of the compounds of the present invention is further illustrated in the following example:

**Example VII**

Two strips of a multi-layer color element comprising a support coated with a red-sensitive silver halide emulsion layer containing a cyan-forming coupler, a green-sensitive silver halide emulsion layer containing a magenta-forming coupler, a bleachable yellow filter layer and a blue-sensitive silver halide emulsion layer containing a yellow-forming coupler were given identical exposures to a neutral density step wedge on an Eastman 1B sensitometer, then given color processing as follows:

- **Solution:**
  - **Time in seconds**
  - **Prehardener**
  - **Negative developer**
  - **Stop bath**
  - **Re-exposure to #2 Photoflash (30 sec. front and 30 sec. back):**
    - **Color developer**
    - **Stop bath**
    - **Wash**
    - **Bleach**
    - **Fix**
    - **Wash**
    - **Stabilizer**
    - **Dry**

The prehardener solution was a formalin-succinaldehyde prehardener such as is described in Baden et al., U.S. Serial 321,323 filed November 4, 1963. A hydroxylamine sulfate neutralizer bath such as is described by Blackmer and Vogt in U.S. Serial 111,489 filed May 22, 1961, was used. A conventional silver halide solvent type hydroquinone—p-methyaminophenol sulfite developer was used for negative development. A conventional phenylenediamine color developer solution was used for color development. The stop bath was a conventional acetic acid type, the bleach a conventional alkali metal ferricyanide bromide type, the fixing bath a conventional alkali metal thiosulfate bath, and the stabilizer a solution of formalin and commercially available "Photo-Flo." The solutions were used at 160°F. in a drum processor. The two strips were given identical processing except for the prehardener solution. Strip No. 1 was given standard prehardener solution, while Strip No. 2 was given standard prehardener solution to which 0.02 g./l. of 1-(3-caproamido)phenyl-5-mercaptotetrazole has been added. Optical density measurements were made of the yellow dye, the cyan dye and the magenta dye in the image of each step of the neutral density tablet in each strip. The accompanying drawings, FIGS. 1 and 2, show graphs of these data. FIG. 1 shows for Strip 1 which was given standard prehardening, the density curves 1, 2 and 3 for yellow dye, cyan dye and magenta dye respectively. FIG. 2 shows for Strip 2 which was prehardened with the prehardener containing 1-(3-caproamidoaminophenyl)-5-mercapto-
tetrazole, the density curves 1a, 2a, and 3a for yellow dye, cyan dye and magenta dye respectively. FIG. 1 shows that the yellow dye densities are too low with respect to the cyan and magenta dye densities at almost every step and color reproduction is unsatisfactory. FIG. 2 shows that the yellow dye densities in Strip No. 2 are good and their relationship to the cyan and magenta dye densities will give good color reproduction. A comparison of FIG. 1 with FIG. 2 shows that curve 1a (yellow dye) is 50% higher at step 6, 78% higher at step 8, and 84% higher at step 12 than curve 1 at the same steps respectively, while curve 2 is substantially the same as curve 2a and curve 3 is substantially the same as curve 3a. This shows that the 1-(3-caproamidophenyl)-5-mercaptotetrazole in the prehardener reduced the negative development in the top yellow-forming layer but had virtually no effect on the negative development of the other image layers. One would expect that the substantial inhibiting effect produced by the 1-(3-caproamidophenyl)-5-mercaptotetrazole on the developing action of the relatively strong negative developer would be much greater on the relatively weak color developer in the yellow-forming layer thus resulting in the development of little or no yellow dye. It is therefore very unexpected that Strip No. 2 would have from 50 to 84% more yellow dye developed than Strip No. 1.

Similar results are obtained by repeating Example VII but using other 5-mercaptotetrazoles of Formula I in the prehardener solution in place of 1-(3-caproamidophenyl)-5-mercaptotetrazole. Alternatively the 5-mercaptotetrazoles of Formula I are used to advantage in the hydroxylamine sulfate bath or in the negative developer solution to inhibit the negative development of only the top layer of the multilayer photographic element so that the top layer is not overdeveloped relative to the middle and bottom image-forming layers by the high temperature negative developer solution.

It will be understood that the modifications and examples included in the present specification are illustrative only and that the invention described is to be taken as limited only by the scope of the appended claims.

What is claimed is:

1. A prehardener solution for use in the processing of a multi-layer photographic color element, said prehardener solution comprising formaldehyde and a mercaptotetrazole having the formula:

   \[
   \begin{array}{c}
   N \equiv N \\
   \left. \begin{array}{c}
   (5-M) \\
   M \equiv C \equiv N \equiv X - Y \\
   \end{array} \right\} \\
   \end{array}
   \]

   where M is a member of the class consisting of alkali metals and hydrogen, and X is a radical having a formula selected from the class consisting of \(-SO_2R\) and \(-COR\) wherein R is a hydrocarbyl radical of 1 to 22 carbon atoms.

2. The method of selectively controlling the overdevelopment of a multi-layer photographic color element by a negative developer in a high temperature color process which comprises negative development of said element in the presence of a compound having the formula:

   \[
   \begin{array}{c}
   N \equiv N \\
   \left. \begin{array}{c}
   (5-M) \\
   M \equiv C \equiv N \equiv X - Y \\
   \end{array} \right\} \\
   \end{array}
   \]

   where M is a member of the class consisting of alkali metals and hydrogen, and X is a radical having a formula selected from the class consisting of \(-SO_2R\) and \(-COR\) wherein R is a hydrocarbyl radical of 1 to 22 carbon atoms.

3. The method of selectively controlling the overdevelopment of a multi-layer photographic color element by a negative developer in a high temperature process which comprises using a prehardener solution comprising formaldehyde, succinaldehyde and a compound having the formula:

   \[
   \begin{array}{c}
   N \equiv N \\
   \left. \begin{array}{c}
   (5-M) \\
   M \equiv C \equiv N \equiv X - Y \\
   \end{array} \right\} \\
   \end{array}
   \]

   where M is a member of the class consisting of alkali metals and hydrogen, and X is a radical having a formula selected from the class consisting of \(-SO_2R\) and \(-COR\) wherein R is a hydrocarbyl radical of 1 to 22 carbon atoms.

4. The prehardener solution of claim 1 in which the mercaptotetrazole is 1-(3-caproamidophenyl)-5-mercaptotetrazole.

No references cited.

NORMAN G. TORCHIN, Primary Examiner.
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