Abstract:

Various agents for improving adhesion of cold end coatings on glass surfaces are described. Also described are agents for improving bonding between cold end coatings and organic inks. And, methods and related detection solutions for detecting the presence of cold end coatings on a substrate such as a glass bottle are disclosed. Additionally, particular additives and primers are described that can be applied onto cold end coatings for improving adhesion thereto.
CROSS REFERENCES TO RELATED APPLICATIONS


FIELD OF THE INVENTION

[0002] The present invention relates to improving adhesion of cold end coatings (CECs) on glass surfaces, improving bonding between organic inks and CECs and glass surfaces, and detection methods for CECs.

BACKGROUND OF THE INVENTION

[0003] The beverage bottle industry applies a lubricious transparent cold end coating (CEC) onto formed glass bottles to prevent surface damage and caustic attack of the glass bottles. Screen printed organic inks are typically applied directly upon a CEC. Unfortunately, the CEC can interfere with bonding of the organic ink and adhesion at the glass surface. Therefore, various materials have been evaluated as cross-linkers to promote adhesion on CEC's, on glass surfaces, and for use in other additives. As a result, particular chemistries and procedures for application of CEC's at bottling plants have been recommended. The basic chemistry necessary for improvement of bonding between the organic ink and glass surface is the addition of a silane to a CEC composition. The adhesion improvement between the organic ink layer and the CEC is mainly based on the cross-linking function of the materials.
Quality issues may arise with the CEC that can significantly impact hot caustic durability performance in returnable glass bottles. A bottler or decorator typically experiences significant field failures with many commercially available inks. This is likely due to poor CEC quality in terms of adhesion on glass and bonding with the organic inks. As noted, the CEC is transparent, as it is made of various clear mediums containing fatty acids, polymers, etc. Thus, there is currently no easy way to ensure that the CEC is present on the glass surface, that it contains the necessary silane, that it is applied uniformly, and/or that the CEC solution or coated bottles have not aged or degraded to an extent such that the silane no longer has the requisite surface functionality.

Therefore, the objectives of the present invention include improving adhesion of CECs on glass surfaces, improving bonding with organic inks, and providing fast and easy test methods for detection of the invisible and functional CEC and its components.

SUMMARY OF THE INVENTION

The difficulties and drawbacks associated with currently known CECs are addressed by the present invention.

In one aspect, the present invention provides an additive for incorporating in a cold end coating (CEC) composition for improving adhesion thereto. The additive is one or more organo metalate agents.

In another aspect, the present invention provides a pretreatment composition for application onto a glass surface. The pretreatment composition serves to improve characteristics of a cold end coating (CEC) applied subsequently thereto. The pretreatment composition includes one or more silane agents.
In yet another aspect, the present invention provides a composition for detecting the presence of a cold end coating (CEC). The composition includes at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof.

In still another aspect, the present invention provides a primer composition for improving adhesion to a cold end coating (CEC). The primer composition comprises one or more silane agents in combination with one or more titanate agents.

And, in a further aspect, the present invention provides a method of detecting the presence of a cold end coating using a forensic indicator. The method comprises applying a composition comprising a forensic indicator onto a surface in question. The method also comprises reviewing the surface for development and visual indication of the forensic indicator.

In still a further aspect, the present invention provides a method of detecting the presence of a cold end coating. The method comprises applying a composition for detecting the presence of a cold end coating (CEC). The composition includes at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof onto a surface in question. The method also comprises directing UV light to the surface after applying the composition. And, the method comprises reviewing the surface for one or more visual indicators.

The present invention also provides a method for improving adhesion by an ink to a cold end coating (CEC). The method comprises providing a cold end
coating formulation. And, the method comprises incorporating one or more organo metalate agents into the formulation.

[0014] In still another aspect, the present invention provides a method for improving adhesion by an ink to a cold end coating (CEC). The method comprises providing a substrate having a cold end coating thereon. The method also comprises providing a primer composition including at least one silane and at least one titanate. And, the method comprises applying the primer composition to the substrate and the cold end coating.

[0015] In an additional aspect, the present invention provides a method for promoting bonding between an ink and a glass substrate. The method comprises providing a cold end coating composition. The method also comprises incorporating an effective amount of at least one silane agent into the cold end coating composition. The silane agent is selected from the group consisting of amino silanes, epoxy silanes, chlorosilanes, methoxy silanes, ethoxy silanes, alkoxy silanes, acyloxy silanes, dipodal silanes, and combinations thereof. Upon application of the cold end coating composition and silane agent incorporated therein onto the glass substrate and forming of a cold end coating, and upon depositing ink upon the coating, the bonding between the ink and the glass substrate is promoted.

[0016] As will be realized, the invention is capable of other and different embodiments and its several details are capable of modifications in various respects, all without departing from the invention. Accordingly, the description is to be regarded as illustrative and not restrictive.
DETAILED DESCRIPTION OF THE EMBODIMENTS

[0017] In accordance with the present invention, materials suitable for promoting bonding adhesion of functional CECs with glass surfaces and organic inks are disclosed. Various organo metalates have been discovered as potential cross-linkers for improved bonding between CEC’s and organic inks and inert surfaces such as glass.

[0018] Furthermore, a pretreatment composition for a glass surface is described containing one or more particular silane agents, which improves characteristics of a CEC applied onto the pretreatment layer.

[0019] Also described herein are reagents suitable for use as detectors of CECs and for use in a simple method for qualifying a CEC and its components on glass surfaces. Another aspect of the present invention is the utilization of visible and ultraviolet (UV) functional pigments or dyes in the CECs and/or its silane solution and/or composition for easy detection. These can include UV optical brighteners, food color dyes, and natural dyes from vegetables, algaes, and other stains. UV optical brightener dyes and pigments along with infrared (IR) active dyes and pigments can also be utilized to detect the presence and application uniformity of the CEC’s. The present invention includes use of a forensic detection solution to determine the presence and functionality of primary amine groups attached to silanes, one of the components that are used to improve interfacial bonding between a glass surface and the coating, in solutions and on the glass surface itself. Also, the present invention includes the use of an aqueous potassium permanganate (KMnO₄) solution to qualitatively identify the primary amine groups attached to the silane coupling agents.
In addition, various additives and primers are described which can be applied onto a CEC to improve adhesion of inks thereto.

Each of these aspects are described in greater detail as follows.

Organo Metalates and CEC’s

It has been surprisingly discovered that the caustic performance, i.e. resistance by a glass substrate to degradation from exposure to a caustic environment, is significantly improved by incorporating one or more organo metalates into a CEC, and cross-linking the resulting CEC. In a preferred embodiment according to the present invention, the materials suitable for cross-linking the CECs are organo metalates, preferably organo transition metalates, and most preferably organo titanates and zirconates. These agents can be used in combination with each other or with other components. Organically chelated organo metal oxides have at least two main functions, cross-linking and adhesion. These agents can cross-link polymers by forming stable bonds with the functional groups (eg. -OH) of polymeric resins. In a molecule of a particular organo metalate, the presence of two or more labile alkoxy groups provide the cross-linking activity. The cross-linking requires the presence of functional groups on the resin used in the organic ink and in the film (CEC) to which the ink adheres. The alkoxy groups react with the hydroxyl groups of both the resin (used in the ink) and the film (CEC) to form a cross-link, thereby bonding the ink to the polymer surface. Organo metalates can act as adhesion promoters even on unreactive substrates such as glass, metal, etc. and functional coatings.
Pretreatment Compositions and Silanes for CECs

[0023] It has also been discovered that adding one or more specific silane coupling agents to a cold end coating (CEC) composition significantly promotes bonding between an ink and the underlying glass surface. Thus, such silanes can serve as pretreatment agents for CEC's. A variety of silane agents can be used such as amino silanes, epoxy silanes, chlorosilanes, methoxy silanes, ethoxy silanes, alkoxy silanes, acyloxy silanes, and/or dipodal silanes. These agents can be used in any combination with each other or in combination with other agents. It is preferred to use glycidoxypropyltrialkoxy silane, and 7-aminopropyltrialkoxy silane either singly or in combination. The incorporation of these silane agents also improves the caustic performance of the CEC. Various related methods are provided in accordance with the invention.

Detection Methods, Compositions and Forensic Agents

[0024] The present invention also provides methods and compositions for detecting the presence of a CEC and in certain applications, assessing one or more characteristics of the CEC. Specifically, one or more reagents and/or additives are incorporated into a CEC, thereby facilitating performing a detection method as described herein.

[0025] Reagents suitable for use as detectors and for use in a relatively simple method for qualifying functional CECs and components thereof on a glass surface are also described herein. The reagents include UV functional optical brighteners, functional dyes and/or pigments, forensic class chemicals such as ninhydrin, and a solution of potassium permanganate having reactive functional groups or ions that react with components of functional coatings producing differential color change.
Any of these agents or components can be used singly or in combination with
themselves or with other agents or components. Examples of dyes include but are
not limited to food color dyes, natural dyes, infrared active dyes and the like. Optical
brightening agents (OBAs) are also included and are generally dyes that absorb light
in the UV region of the electromagnetic spectrum, and re-emit light in a specific wave
length.

[0026] Preferably, the optical brightener additives which are based on stilbenes,
coumarines or azolines, are utilized to detect the presence of CECs. These can be
used singly or in combination. More specifically, Keystone Keyfluor white CBS-X
(yellow) (CBS-X) and Cartax CXDP P (blue) (CXDPP) optical brightener/fluorescent
pigment are used in the present invention. Keystone Keyfluor white CBS-X is
believed to be an oxazole and is available from Keystone Aniline Corp. of Chicago,
IL. Cartax CXDP is available from Clariant GmbH of Augsburg Germany. Cartax
CXDP is believed to be a substituted benzo-oxazinone. Solutions of CBS-X and
CXDP P were used at 0.1% but other dyes/pigments can also be used and at
various levels. By using a UV lamp, a solution previously sprayed onto a warmed
glass to form a somewhat clear coating, fluoresces to bright blue color with CBS-X,
and neon yellowish green with CXDP P, indicating the presence of the coating (CEC)
on the glass.

[0027] Another embodiment of the present invention relates to a forensic
detection solution to determine the presence and functionality of primary amine
groups attached to silanes on a glass surface. The reagents are made as aqueous
and/or solvent solutions, and subsequently sprayed, dripped, or otherwise applied on
the surface of CECs for detection. A preferred forensic reagent or indicator utilized in
the present invention is ninhydrin (formula I). The reaction of ninhydrin with amines
is highly specific and the absorption characteristics of the formed chromophore follow Beer's law. Thus, reagents based on ninhydrin are an excellent choice for detection and quantification of amines and amino acids that may be present in a CEC.

![Chemical structures](image)

(I) NINHYDRIN

(II)

[0028] The ninhydrin reagent solution is preferably utilized as an aqueous alcoholic solution. A ninhydrin concentration range of 2-3% prepared in n-butanol which is acidified with glacial acetic acid is preferred, as this reagent in solvents is more sensitive towards reaction with amines. Ninhydrin reacts slowly at room temperature. Consequently, elevated temperatures are employed to reduce the conversion time to about one minute. The chromophore (formula II) formed as a result of the reaction of ninhydrin with amino groups, exhibits different colors. The reaction proceeds via a condensation reaction between carbonyl and amino groups, and depending on the nature of the amino group attached on the silane, such as primary or secondary, and nature and chain length of carbon between the amino group and the silane, the resulting color varies between purple, pink, violet, and blue. The method of detection is convenient and relatively fast, as it precisely results merely by change of color due to heating within a few minutes.

[0029] In addition to or instead of ninhydrin, one or more ninhydrin derivatives can be used. For example, the present invention includes but is not limited to the use of the following ninhydrin derivatives: benzoninhydrin, thieno[?]ninhydrin,
naptho / ninhydrin, bis-ninhydrin analogs, aryl and heteroaryl ninhydrin analogs, ninhydrin hemiketals, and combinations thereof. Benzoninhydrin is preferred for use with zinc salts. 5-methoxy/5-methylthio ninhydrin exhibits intense fluorescence. Thienoj/ninhydrin exhibits good chromogenic and fluorogenic properties. Bis-ninhydrin analogs can be utilized to provide polymeric Ruhemann's purple species. Aryl and heteroaryl ninhydrin analogs exhibit fluorogenic properties without metal ion treatment. Ninhydrin hemiketals may be preferred for particular applications owing to their ability to detect latent fingerprints or chemical moieties on thermal paper without background staining. Additional details associated with ninhydrin reagents are provided in US Patent 4,274,833.

[0030] In another embodiment of the present invention, an aqueous potassium permanganate solution is used to qualitatively detect the primary amine groups which are attached to the silane coupling agent(s). Thus, potassium permanganate is another forensic indicator, as described herein. A simple test with dilute permanganate solution yields significant differences among the silanes. This reagent indicates the presence of primary amines by changing its color from purple to brown as aliphatic primary amines are known to reduce purple Mn⁷⁺ ions to brown Mn²⁺³⁺. Thus, this reagent may serve as a suitable forensic indicator.

[0031] The present invention also includes the use of one or more other forensic agents in addition to, or instead of, the agents noted herein. For example, suitable forensic agents can also include 1,8-diazafluorene-9-one (DFO), 1,2-indanediones, and combinations thereof. 1,8-diazafluorene-9-one (DFO) reacts with alpha-amino acids to produce a red product. The red product is analogous to Ruhemann's purple product resulting from ninhydrin. In contrast to ninhydrin, DFO develops fingerprints or chemical moieties that are highly fluorescent without secondary treatment with
metal salts. Compounds or the class of 1,2-indanediones are more sensitive than DFO and react faster thereby typically revealing more information. In addition to being more sensitive, the cost of these reagents is generally less than that of DFO. Moreover, 1,2-indanediones typically do not require precise development conditions to prevent thermal decomposition.

Additives and Primers for Improving Adhesion

Another aspect of the present invention relates to a primer composition that if applied onto a CEC, significantly improves adhesion of inks subsequently applied thereto. Preferably, the primer composition includes one or more silanes and one or more titanates. Use of the primer also promotes caustic performance of the resulting coated substrate or bottle. In this embodiment of the invention, the polymer resin itself, i.e. the CEC, serves as an adhesive. The resin in this embodiment can include conventional adhesive components such as epoxies and polyesters.

EXAMPLES

Example 1: Caustic Performance of Organo Titanate Cross-linked CEC

A HTP white ink formulation, referred to herein as "Ink A" comprised of EPON Resin 1001, an epoxy resin solids solution, and EPON Resin 828, a general purpose epoxy resin, both available from Hexion Specialty Chemicals in Columbus, Ohio, with the addition of 3% organo titanate, VERTEC P12, available from Johnson Matthey Catalysts of London, UK, was prepared and designated as sample 2 and a similar formulation however using 3% organo titanate, VERTEC 1A10 was prepared and designated as Sample 1. Testings were performed on a T5 stearate coating and
on a silane/T5 coating. "T5 stearate" is a stearate coating available under the
designation TEGOGLAS from Arkema, Inc. of Philadelphia, PA.

[0034] Caustic protocols: Tests started with a 2-3% NaOH aqueous solution at
65-70°C, containing a washing additive, and used a several minute soak.

[0035] There were no failures, so testing was made more severe by increasing
soak time, temperature, and concentration. Tests were taken to failure.

[0036] Results indicated that the titanates exhibited equal or superior
performance than that of the T5 coating alone and the silane/T5 coating.

Example 2: Pretreatment Cold End Coating for Glass Adhesion in Caustic

[0037] Adding silane coupling reagents such as SILQUEST A-187
(glycidoxypropyltrialkoxy silane), and SILQUEST A-1 100 (-/aminopropyltrialkoxy
silane), etc., both available from Momentive Performance Materials of Albany, NY, to
the cold end coating (CEC) promotes ink bonding to the glass surface. The best
performance was achieved by adding 0.2% SILQUEST A-1 100 into a 0.2% stearate
solution (modified Tego-5 available from Evonik Degussa GmbH of Parsippany, NJ).
This formed Sample 3. EPON Resin 1001 F, EPON Resin 828, and similar epoxy
resins that comprise formulations such as Ink A, were used and a white epoxy HTP
ink was screen printed onto glass and cured at 400°F (204°C) for 30 minutes. Glass
panels were then subjected to a caustic bath containing 2-5% NaOH and 0.25% of
an additive (BW61, an anti-rust additive from Johnson Diversey, Inc., of Sturtevant,
WI, or similar), at 65-80°C. The panels were soaked for several minutes followed by
a rinse, and then the cycle was repeated. Samples lasted beyond 30 cycles without
delamination. A varied caustic test utilizing up to a 5 hour soak instead of short
minute cycles was also used to determine the coatings' resistance to caustic washings. Details of the various pretreatments are described as follows.

[0038] SILQUEST A-1100 was mixed 50/50 with ethanol and pH adjusted with acetic acid to a neutral pH. This solution was then added to a concentration of 0.2% into a 0.2% T5 solution and sprayed onto warmed (approximately 95°C) glass panels.

[0039] SILQUEST A-187 was mixed 50/50 with ethanol and pH adjusted with acetic acid to a neutral pH. This solution was then added to a concentration of 0.2% into the 0.2% T5 solution and sprayed onto warmed (approximately 95°C) glass panels.

[0040] T5 CEC (stearate) was used to form a 0.2% solution in deionized (DI) water. The solution was then sprayed onto warmed (approximately 95°C) panels.

[0041] In test 1, 30 cycles as per the caustic protocol previously noted were performed. The failure cycle is listed below in Table 1.

<table>
<thead>
<tr>
<th>2 panels of each condition:</th>
<th>Ink A (no. of cycles)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tinside - no pretreatment</td>
<td>30</td>
</tr>
<tr>
<td>Airside - no pretreatment</td>
<td>24</td>
</tr>
<tr>
<td>Tinside – A-187</td>
<td>PASS</td>
</tr>
<tr>
<td>Airside – A-187</td>
<td>PASS</td>
</tr>
<tr>
<td>Tinside - A-1100</td>
<td>PASS</td>
</tr>
<tr>
<td>Airside - A-1100</td>
<td>PASS</td>
</tr>
</tbody>
</table>

[0042] In test 2, 30 cycles per the caustic protocol were performed and the failure cycle is listed below in Table 2.
This data demonstrates that T5 CEC provides 16-24 cycles prior to failure, no CEC yields 24-30 cycles prior to failure, and silane pretreatments provide more than 30 cycles before failure.

Example 3: Addition of UV Active Indicator to the Pretreatment/CEC (with Caustic Durability) and Detection Method

0.01 % of the brightener additive CBS-X, 0.01 % of the brightener additive CXDPP, or other dyes/pigments solutions can be used to detect the CECs. The solution was sprayed onto warmed glass to form a somewhat clear coating. By detection with a UV lamp, the pretreatment on the glass exhibited a blue color with CBS-X, and a neon yellowish green with CXDPP indicating the presence of a CEC on the glass. The previously noted Sample 3 white HTP ink was screen printed onto the glass and cured at 400°F (204°C) for 30 minutes. Glass panels were then subjected to a caustic bath of 2-5 % NaOH containing 0.25% BW61, at 65-80°C for a several minute soak. The panels were then rinsed, and the process repeated. Samples lasted beyond 30 cycles. In addition, the fluorescence was still present under the ink label after 30 wash cycles by UV lamp detection.

Caustic Testing

Testing was based on the caustic protocol given in Example 2.
Several minute cycles at 65-80°C in 2-5% NaOH solution bath with the additive 0.25% Johnson Diversey BW61 were performed. The results of this investigation are set forth below in Table 3.

Table 3: Test 3

<table>
<thead>
<tr>
<th>Panels tested</th>
<th>No. of Panels</th>
<th>No. of Caustic Cycles</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sample 3 T-5 only</td>
<td>4 Tinside</td>
<td>&gt; 30 with no failures</td>
</tr>
<tr>
<td>Sample 3 Pretreatment Solution</td>
<td>4 Tinside</td>
<td>&gt; 30 with no failures</td>
</tr>
<tr>
<td>Sample 3 Tinside</td>
<td>4 w/ Cartax</td>
<td>&gt; 30 with no failures</td>
</tr>
<tr>
<td></td>
<td>Pigment</td>
<td></td>
</tr>
<tr>
<td>Sample 3 Tinside</td>
<td>4 w/ Keystone</td>
<td>&gt; 30 with no failures</td>
</tr>
<tr>
<td></td>
<td>Pigment</td>
<td></td>
</tr>
</tbody>
</table>

The results of caustic testing set forth above in Table 3 meet caustic protocol qualifications and demonstrates that the OBA's/fluorescent pigments do not affect the caustic durability.

Example 4: Detection of Silane in the CECs by Use of Ninhydrin

A 2% ninhydrin solution was formed in n-butanol by gently dissolving for about 30 minutes. The resulting green-yellow solution was acidified with glacial acetic acid. The final composition of acetic acid was approximately 2% by volume. Testing was performed on both glass panels and bulk solutions using old and fresh silane formulations with and without stearate ingredient. Testing samples were optionally heated on a hotplate or placed in a preheated oven after applying the ninhydrin solution.

A 2% ninhydrin solution was placed on the glass having a pretreatment CEC containing SILQUEST A-1 100 on the glass surface. When the glass was
heated, a reaction occurred and the yellowish-clear drop changed to a purplish color. This indicates that the pretreatment CEC is still on the glass or bottle and is active.

To check activation of the pretreatment solution itself, a drop of 0.2 - 2.0% pretreatment CEC was placed on the glass, and then a drop of the 2% ninhydrin solution was placed on top of the pretreatment drop. When heated, a change in color from yellowish-clear to purple-brown occurs when an amine is detected. This can be performed upon wet on wet coatings or wet on dried coatings.

Example 5: Use of Potassium Permanganate KMnO₄

The following test was used to detect the presence of silane, such as SILQUEST A1100, on a pretreated glass surface using KMnO₄. This reagent will indicate the presence of a primary amine group attached on the silane by changing its color from purple to brown.

A 0.2 N KMnO₄ solution was prepared by dissolving an appropriate amount of crystals of KMnO₄ in distilled water. Testings were performed on both glass panels and bottles using old and fresh silane formulations with or without a stearate ingredient. To speed up the reaction of amino group on the permanganate ion, the test samples were heated on a hot plate or in a preheated oven. The following are the relevant observations. Color change from pink to brown was noticed with fresh silane formula on glass panel and in bulk solution indicating the presence of silane in the CEC formulation. A similar change in color was observed when three or six months old bulk formulation was used. No color change was seen when only stearate solution was used on both glass panel and bulk.
Silane pretreatment does not adhere to PE/CEC and does not pass 26 cycles of caustic test. However, primers made with a combination of silane and titanate can improve the cure of adhesion to polyethylene (PE) coated surfaces.

A primer consisting of 3 glycidoxy-propyl-tri-methoxy and titanium acetyl-acetonate having a ratio of 2:1 of silane to titanate was demonstrated to promote adhesion to PE and improve caustic resistance. Other titanates such as modified tetra-isopropyl titanate, tetra-butyl titanate, and titanium phosphate improved curing of organic coatings or inks.

A set of bottles were treated with a silane/titanate primer and air dried to form a continuous layer. An indicator of 4,4,-bis(2 benzoxazolyl) stilbene was incorporated to show that the primer was evenly applied. The primer and indicator were prepared as per the following formula: 5 parts of indicator were mixed with 60 parts of di-ethylene glycol mono butyl ether. In a separate container, 22 parts of 3 glycidoxy-propyl-tri-methoxy silane were added to 11 parts of 5 parts of chelated titanate such as titanium acetyl-cetonate, modified tetra-butyl titanate, tetra-isopropyl titanate or titanium phosphate. Then 5 grams of the indicator was added. This mixture was diluted with 120 parts of deionized (DI) water.

The primer was applied by spray to bottles and allowed to air dry. The coated bottles were kept at room temperature until they were decorated with an epoxy based organic ink, i.e. previously described Ink A. Following the decoration, the bottles were baked for 30 minutes at 400°F (204°C). The bottles were observed under a UV light and the indicator showed good coverage by the silane/titanate primer.
The bottles were immersed into a caustic bath for 7 minutes. The bottles were then removed, rinsed with tap water followed by DI water and inspected for damage, such as removal of coating, eroding of the edges of letter(s) or discoloration. If no damage was observed then the bottles were returned to the warm caustic bath and the 7 minute cycle was repeated until failure or 26 cycles were achieved.

Bottles coated with silane/titanate primer passed 24 and 30 cycles, whereas bottles decorated with Ink A directly over PE failed after 16 and 21 cycles. The caustic bath consisted of 2-5% NaOH, 0.25% BW-61 surfactant and was held at 65-70°C for the duration of the test.

Example 7: Determination of Optimum Ratio of Silane/Titanate for Promoting Adhesion to PE/CEC Coated Bottles

Four primers were prepared by using different ratios of silane to titanate. The ratios were (i) 2 parts of silane to 1 part of titanate, and (ii) 1 part of silane to 2 parts of titanate. Both ratios were duplicated with two different indicators that are visible under UV light.

The indicators were dispersed in diethylene glycol mono butyl ether (5:60) and 5 grams were added to each of the silane/titanate solutions. Deionized (DI) water was used to dilute the primer. The bottles, i.e. PE/CEC coated, were washed with a soak solution and rinsed with tap water and dried in an oven before coating. The primers were sprayed on a set of four bottles and air dried before decoration.

The bottles were decorated with two inks, Ink A (white) and a similar ink, referred to herein as Ink B, however having a different color (red) and these bottles were baked at 30 minutes at 400°F (204X).
The bottles were immersed in a proprietary caustic batch and held for 1 hour at 65 to 70°C. After one hour, bottles with 1 part of silane to 2 parts of titanate showed some adhesion loss while bottles with 2 parts of silane and 1 part of titanate were clearly less affected by the exposure.

Both indicators were visible under the organic ink under a UV light.

Many other benefits will no doubt become apparent from future application and development of this technology.

All patents, published applications, and articles noted herein are hereby incorporated by reference in their entirety.

It will be understood that any one or more feature or component of one embodiment described herein can be combined with one or more other features or components of another embodiment. Thus, the present invention includes any and all combinations of components or features of the embodiments described herein.

As described hereinabove, the present invention solves many problems associated with previous methods and approaches. However, it will be appreciated that various changes in the details, materials and arrangements of components or steps, which have been herein described and illustrated in order to explain the nature of the invention, may be made by those skilled in the art without departing from the principle and scope of the invention, as expressed in the appended claims.
WHAT IS CLAIMED IS:

1. An additive for incorporating in a cold end coating (CEC) composition for improving adhesion thereto, the additive being one or more organo metalate agents.

2. The additive of claim 1 wherein the organo metalates are organo transition metalates.

3. The additive of claim 2 wherein the organo transition metalates are selected from the group consisting of organo titanates, organo zirconates, and combinations thereof.

4. A pretreatment composition for application onto a glass surface, the pretreatment composition serving to improve characteristics of a cold end coating (CEC) applied subsequently thereto, the pretreatment composition including one or more silane agents.

5. The pretreatment composition of claim 4 wherein the one or more silane agents are selected from the group consisting of amino silanes, epoxy silanes, chlorosilanes, methoxy silanes, ethoxy silanes, alkoxy silanes, acyloxy silanes, dipodal silanes, and combinations thereof.

6. The pretreatment composition of claim 5 wherein the silane agent is an alkoxy silane.
7 The pretreatment composition of claim 6 wherein the alkoxy silane is selected from the group consisting of glycidoxypropyltrialkoxy silane, γ-aminopropyltrialkoxy silane, and combinations thereof.

8. A composition for detecting the presence of a cold end coating (CEC), the composition including at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof.

9. The composition of claim 8 wherein the composition includes optical brighteners.

10. The composition of claim 9 wherein the optical brighteners are selected from the group consisting of stilbenes, coumarines, azolines, and combinations thereof.

11. The composition of claim 8 wherein the composition includes forensic indicators.

12. The composition of claim 11 wherein the forensic indicator is selected from the group consisting of ninhydrin, ninhydrin derivatives, potassium permanganate, 1,8-diazafluorene-9-one (DFO), 1,2-indanediones, and combinations thereof.
13. The composition of claim 12 wherein the indicator is ninhydrin.

14. The composition of claim 12 wherein the indicator is a ninhydrin derivative.

15. The composition of claim 14 wherein the ninhydrin derivative is selected from the group consisting of benzoninhydrin, thienof/ninhydrin, napthof/ninhydrin, bis-ninhydrin analogs, aryl and heteroaryl ninhydrin analogs, ninhydrin hemiketals, and combinations thereof.

16. The composition of claim 12 wherein the indicator is potassium permanganate.

17. The composition of claim 12 wherein the indicator is 1,8-diazafluorene-9-one (DFO).

18. The composition of claim 12 wherein the indicator is a 1,2-indanedione.

19. A primer composition for improving adhesion to a cold end coating (CEC), the composition comprising one or more silane agents and one or more titanate agents.

20. A method of detecting the presence of a cold end coating using a forensic indicator, the method comprising:
applying a composition comprising a forensic indicator onto a surface in question;
reviewing the surface for development and visual indication of the forensic indicator.

21. The method of claim 20 wherein the forensic indicator is selected from the group consisting of ninhydrin, ninhydrin derivatives, potassium permanganate, 1,8-diazafluorene-9-one (DFO), 1,2-indanediones, and combinations thereof.

22. The method of claim 21 wherein the indicator is ninhydrin.

23. The method of claim 21 wherein the indicator is a ninhydrin derivative.

24. The method of claim 23 wherein the ninhydrin derivative is selected from the group consisting of benzoninhydrin, thieno[ /] ninhydrin, napthof/ ]ninhydrin, bis-ninhydrin analogs, aryl and heteroaryl ninhydrin analogs, ninhydrin hemiketals, and combinations thereof.

25. The method of claim 21 wherein the indicator is potassium permanganate.

26. The method of claim 21 wherein the indicator is 1,8-diazofluorene-9-one (DFO).

27. The method of claim 21 wherein the indicator is a 1,2-indanedione.
28. A method of detecting the presence of a cold end coating, the method comprising:

applying a composition for detecting the presence of a cold end coating (CEC), the composition including at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof onto a surface in question;

directing UV light to the surface after applying the composition;
reviewing the surface for one or more visual indicators.

29. The method of claim 28 wherein the composition includes optical brighteners.

30. The method of claim 29 wherein the optical brighteners are selected from the group consisting of stilbenes, coumarines, azolines, and combinations thereof.

31. The method of claim 28 wherein the composition includes forensic indicators.

32. The method of claim 31 wherein the forensic indicator is selected from the group consisting of ninhydrin, ninhydrin derivatives, potassium permanganate, 1,8-diazafluorene-9-one (DF90), 1,2-indanediones, and combinations thereof.
33. The method of claim 32 wherein the indicator is ninhydrin.

34. The method of claim 32 wherein the indicator is a ninhydrin derivative.

35. The method of claim 34 wherein the ninhydrin derivative is selected from the group consisting of benzoninhydrin, thienoj/ninhydrin, napthoj/ninhydrin, bis-ninhydrin analogs, aryl and heteroaryl ninhydrin analogs, ninhydrin hemiketals, and combinations thereof.

36. The method of claim 32 wherein the indicator is potassium permanganate.

37. The method of claim 32 wherein the indicator is 1,8-diazafuorene-9-one (DFO).

38. The method of claim 32 wherein the indicator is a 1,2-indanedione.

39. A method for improving adhesion by an ink to a cold end coating (CEC), the method comprising:
   providing a cold end coating formulation;
   incorporating one or more organo metalate agents into the formulation.

40. The method of claim 39 wherein the organo metalates are organo transition metalates.
41. The method of claim 40 wherein the organo transition metalates are selected from the group consisting of organo titanates, organo zirconates, and combinations thereof.

42. A method for improving adhesion by an ink to a cold end coating (CEC), the method comprising:
   providing a substrate having a cold end coating thereon;
   providing a primer composition including at least one silane and at least one titanate;
   applying the primer composition to the substrate and the cold end coating.

43. A method for promoting bonding between an ink and a glass substrate, the method comprising:
   providing a cold end coating composition;
   incorporating an effective amount of at least one silane agent into the cold end coating composition, the silane agent selected from the group consisting of amino silanes, epoxy silanes, chlorosilanes, methoxy silanes, ethoxy silanes, alkoxy silanes, acyloxy silanes, dipodal silanes, and combinations thereof;
   whereby upon application of the cold end coating composition and silane agent incorporated therein onto the glass substrate and forming of a cold end coating, and upon depositing ink upon the coating, the bonding between the ink and the glass substrate is promoted.
INTERNATIONAL SEARCH REPORT

International application No. PCT/US 1 1/21958

A. CLASSIFICATION OF SUBJECT MATTER

USPC - 427/469; 118/624; 118/625; 118/630

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

USPC - 427/469

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

USPC - 118/624; 118/625; 118/630 (see search terms below)

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

PubMed (PGPB,USPT,USOC,EPAB,JPAB); Google

Search Terms Used: cold end coating, organo titanate, silane, primer, adhesion

C. DOCUMENTS CONSIDERED TO BE RELEVANT

<table>
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<tr>
<th>Category</th>
<th>Citation of document, with indication, where appropriate, of the relevant passages</th>
<th>Relevant to claim No.</th>
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<tbody>
<tr>
<td>A</td>
<td>US 5,213,617 A (Blizzard) 25 May 1993 (25.05.1993), entire disclosure</td>
<td>1-3, 39-41</td>
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</table>

Further documents are listed in the continuation of Box C.

* Special categories of cited documents:
  "A" document containing general technical information
  "D" document which may throw doubts on the priority claim(s) or which is cited to establish the publication date of another invention
  "C" document containing technical information and/or amounting to an admission of novelty
  "F" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
  "O" document published prior to the international filing date but later than the priority date claimed

Date of the actual completion of the international search 05 May 2011 (05.05.2011)

Date of mailing of the international search report

20 MAY 2011

Authorized officer: Lee W. Young

PCT Helpdesk: 571-272-4300
PCT OIP: 571-272-7774

Form PCT/ISA/210 (second sheet) (July 2009)
INTERNATIONAL SEARCH REPORT

<table>
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<th>Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)</th>
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<td>This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:</td>
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<tr>
<td></td>
<td>1. Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:</td>
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<tr>
<td></td>
<td>2. Claims Nos.: because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:</td>
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<tr>
<td></td>
<td>3. Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).</td>
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<tr>
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<td>This International Searching Authority found multiple inventions in this international application, as follows:</td>
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<td></td>
<td>The following claim groups were found:</td>
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<td></td>
<td>Group i: Claims 1-3, 39-41</td>
</tr>
<tr>
<td></td>
<td>Group ii: Claims 4-7, 19, 42, 43</td>
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<td></td>
<td>Group iii: Claims 8-18, 20-38</td>
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<td>This application contains the following inventions or groups of inventions which are not so linked as to form a single general inventive concept under PCT Rule 13.1. In order for all inventions to be examined, the appropriate additional examination fees must be paid.</td>
</tr>
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<td>- see extra sheet</td>
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</table>

|            | 1. □ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims. |
|            | 2. □ As all searchable claims could be searched without effort justifying additional fees, this Authority did not invite payment of additional fees. |
|            | 3. □ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.: |
|            | 4. □ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.: |

| Remark on Protest | □ The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee. |
|                   | □ The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation. |
|                   | □ No protest accompanied the payment of additional search fees. |

Form PCT/ISA/210 (continuation of first sheet (2)) (July 2009)
Continuation of Box No. III - Observations where unity of invention is lacking

Group I is directed to an additive for incorporating in a cold end coating (CEC) composition for improving adhesion thereto, the additive being one or more organo metalate agents.

Group II is directed to a pretreatment composition for application onto a glass surface, the pretreatment composition serving to improve characteristics of a cold end coating (CEC) applied subsequently thereto, the pretreatment composition including one or more silane agents.

Group III is directed to a composition for detecting the presence of a cold end coating (CEC), the composition including at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof.

The inventions listed as Groups I-III do not relate to a single general inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons:

Group I includes the technical feature of an additive being one or more organo metalate agents, not found in groups II and III.

Group II includes the technical feature of a pretreatment composition including one or more silane agents, not found in groups I and III.

Group III includes the technical feature of at least one agent selected from the group consisting of UV optical brighteners, food color dyes, natural dyes, infrared active dyes and/or pigments, forensic indicators, and combinations thereof, not found in groups I and II.

Groups I-III therefore lack unity under PCT Rule 13 because they do not share a same or corresponding special technical feature.