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55133-3427 (US). STROBEL, Mark, A. [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US).

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(71) Applicant (for all designated States except US): **3M INNOVATIVE PROPERTIES COMPANY** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **PETT, Todd, G.** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US). **HEBRINK, Timothy, J.** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US). **WRIGHT, Robin, E.** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US). **DAVID, Moses, M.** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US). **MCCLURE, Donald, J.** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US). **JING, Naiyong** [US/US]; 3M Center, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US).

(74) Agents: **ALLEN, Gregory, D.** et al.; 3M Center Office of Intellectual Property Counsel, Post Office Box 33427, Saint Paul, Minnesota 55133-3427 (US).

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(54) Title: GLASS-LIKE POLYMERIC ANTIREFLECTIVE FILMS COATED WITH SILICA NANOPARTICLES, METHODS OF MAKING AND LIGHT ABSORBING DEVICES USING SAME

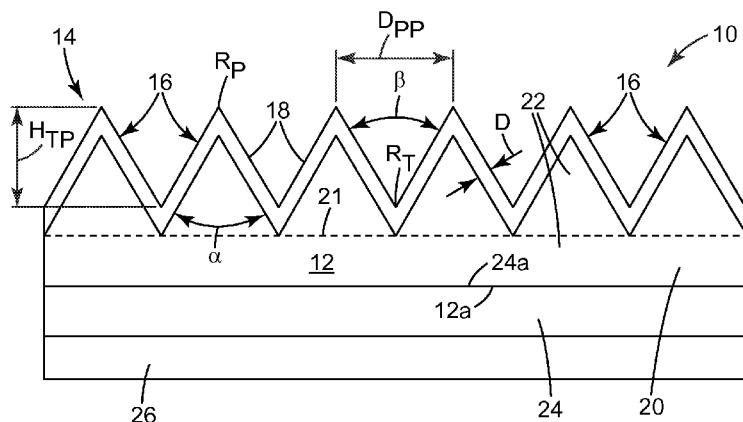


Fig. 1

(57) Abstract: A transparent anti-reflective structured film comprising a structured film substrate having a structured face, with anti-reflective structures defining a structured surface. The structured face is anti-reflective to light, with at least a substantial portion of the structured surface comprising a glass-like surface. At least the anti-reflective structures comprise a cross-linked silicone elastomeric material, and the glass-like surface comprises an SiO<sub>2</sub> stoichiometry. The glass-like surface is coated with a coating of at least one layer of agglomerates of silica nanoparticles, with the agglomerates comprising a three-dimensional porous network of silica nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles. A light energy absorbing device comprising the transparent anti-reflective structured film disposed so as to be between a source of light energy and a light energy receiving face of a light absorber, when light energy is being absorbed by the light absorber.

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**Declarations under Rule 4.17:**

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))* — *with international search report (Art. 21(3))*
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))*

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**GLASS-LIKE POLYMERIC ANTIREFLECTIVE FILMS COATED WITH SILICA NANOPARTICLES,  
METHODS OF MAKING AND LIGHT ABSORBING DEVICES USING SAME**

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**Cross Reference To Related Application**

This application claims the benefit of U.S. Provisional Patent Application No. 61/424707, filed December 20, 2010, the disclosure of which is incorporated by reference herein in its entirety,

10 This invention was made with Government support under Contract No. DE-EE0000587, awarded by DOE. The Government has certain rights in this invention.

15 The present invention pertains to transparent anti-reflective structured films, in particular, to transparent anti-reflective structured films comprising a cross-linked silicone elastomeric material coated with nanoparticles, and more particularly, to such films having anti-reflective structures with glass-like surfaces coated with silica nanoparticles, methods of making such films, and light absorbing devices comprising such films.

**BACKGROUND**

20 With the rising costs of conventional power generation based on burning fossil fuels (e.g., oil and coal based power plants), and the desire to reduce associated greenhouse gases, investment into non-conventional sources of power have increased. For example, the US Department of Energy has invested heavily into the research and development of solar power generation (e.g., solar energy based hot water and electricity generation). One such non-conventional source of power generation is the use of photovoltaic cells to convert solar light energy into electricity. Solar light energy has also been used to 25 directly or indirectly heat water for residential and commercial use. Along with this increased level of interest, there is a need for improving the efficiency at which such non-conventional solar energy technologies can absorb light energy and thereby increase the amount of solar energy available for use.

**SUMMARY OF THE INVENTION**

30 The present invention provides a way to improve the efficiency (i.e., increase the energy generating potential) of solar and other light energy absorbing technologies by enabling more useful light energy into the corresponding light absorbing element (e.g., photovoltaic cell).

35 Silicone elastomers are known for their stability under long-term ultra-violet light exposure, and they can be optically clear and tough. Unfortunately, silicone elastomers also have relatively tacky surfaces that tend to attract, pick-up and hold dirt and dust particles. Until now, this characteristic of picking-up and holding dirt and dust has made silicone elastomers an undesirable candidate for forming

the exposed surface of a light energy absorbing or conversion device such as, e.g., an optically transparent prismatic cover for a photovoltaic cell. The present invention is predicated, at least in part, on the discovery that this tackiness of silicone elastomeric surfaces can be significantly reduced, and their resistance to dirt and dust particle pick-up significantly increased, by treating the structured surface of 5 silicone elastomeric material so that it becomes a glass-like surface that comprises a SiO<sub>2</sub> stoichiometry. The resistance to dirt and dust particle pick-up can be further increased, while maintaining abrasion resistance, by coating the glass-like surface with inorganic nanoparticles

In one aspect of the present invention, a transparent anti-reflective structured film, sheet, web or the like is provided that comprises a structured film substrate comprising a structured face having anti-reflective structures defining a structured surface that is anti-reflective to light. At least the anti-reflective 10 structures comprise a cross-linked silicone elastomeric material, with all, most or at least a substantial portion of the structured surface comprising a glass-like surface, and the glass-like surface comprising an SiO<sub>2</sub> stoichiometry. The glass-like surface is coated with a coating of at least one layer of agglomerates of silica nanoparticles, with the agglomerates comprising a three-dimensional porous network of silica 15 nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles.

It is desirable for the glass-like surface to comprise more than about 50 molar percent SiO<sub>2</sub>. It can be desirable for the glass-like surface to comprise in the range of from more than about 50 molar percent SiO<sub>2</sub> to about 90 molar percent SiO<sub>2</sub>, or . It can also be desirable for the glass-like surface to comprise in the range of from at least about 60 molar percent SiO<sub>2</sub> to about 90 molar percent SiO<sub>2</sub>. 20 Preferably, the glass-like surface comprises at least about 60, 65, 70, 75, 80 or 85 molar percent SiO<sub>2</sub>. The glass-like surface may comprise at least some amount of SiOH. As used herein, any reference to SiOH includes not only SiOH but other constituents comprising Si and OH as well.

It is desirable for only an outer layer of each anti-reflective structure to be glass-like (i.e., for each 25 anti-reflective structure to have a glass-like surface). The anti-reflective structures can project out from a base portion or backing of the structured film substrate. The depth of the glass-like surface depends on the settings (e.g., intensity and/or duration) of the treatment (e.g., flame temperature and duration, UV radiation intensity and duration, oxygen plasma power and duration, etc.) used to form the glass-like surface on at least a significant portion of the anti-reflective structures.

In another aspect of the present invention, a method is provided for making a transparent anti-reflective structured film according to the present invention. The method comprises providing a 30 structured film substrate having a structured face comprising anti-reflective structures defining an anti-reflective structured surface that is anti-reflective to light, with at least the anti-reflective structures comprising a cross-linked silicone elastomeric material; and treating the anti-reflective structured surface so as to transform cross-linked silicone elastomeric material defining at least a substantial portion of the anti-reflective structured surface into a glass-like material comprising an SiO<sub>2</sub> stoichiometry, such that at 35 least a substantial portion of the anti-reflective structured surface comprises a glass-like surface having

the  $\text{SiO}_2$  stoichiometry; and coating the glass-like surface with a coating of at least one layer of agglomerates of silica nanoparticles, with the agglomerates comprising a three-dimensional porous network of silica nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles.

The step of providing a structured film substrate can comprise providing a silicone precursor material that is curable so as to form the cross-linked silicone elastomeric material; forming the silicone precursor material into the shape of the structured film substrate; and curing the silicone precursor material so as to form the structured film substrate. The step of coating the glass-like surface with at least one layer of agglomerates of silica nanoparticles can be performed using a conventional air-knife coating technique.

In an additional aspect of the present invention, a light energy absorbing device (e.g., solar hot water system, photovoltaic electric generating system, etc.) is provided that comprises a light absorber (e.g., solar hot water circulating tubes or other conduits, photovoltaic cell, etc.) and a transparent anti-reflective structured film. The light absorber has a light energy receiving face, and the transparent anti-reflective structured film is disposed so as to be between a source of light energy (e.g., the sun) and the light energy receiving face, at least while light energy from the source is being absorbed by the light absorber. Light energy absorbing devices (e.g., solar energy conversion devices) are used in a wide array of applications, both earth-bound applications and space-based applications. In some embodiments, the solar energy conversion device may be attached to a land-based, water-based, air-based and/or space-based vehicle, such as an automobile, a airplane, a train, a boat or a space satellite. Many of these environments can be very hostile to organic polymeric materials.

In a further aspect of the present invention, a method is provided for making a light energy absorbing device. This method comprises providing a transparent anti-reflective structured film according to the present invention, providing a light absorber having a light receiving face, and securing the anti-reflective structured film in relation to the light absorber so that light can pass through the anti-reflective structured film to the light receiving face of the light absorber.

As used herein and unless otherwise indicated, the term “film” is synonymous with a sheet, a web and like structures.

As used herein, the term “transparent” refers to the ability of a structure, e.g., the inventive film, to allow a desired bandwidth of light transmission therethrough. A structure can still be transparent, as that term is used herein, without also being considered clear. That is, a structure can be considered hazy and still be transparent as the term is used herein. It is desirable for a transparent structure according to the present invention to allow at least 85%, 91%, 92%, 93%, 94%, 95%, 96%, 97% or 98% light transmission therethrough. The present invention can be useful with a wide band of light wavelengths. For example, it can be desirable for the present invention to be transparent to the transmission of light within the wavelength band of from about 400 nm to about 2500 nm. This band generally corresponds to the band of visible light including near infrared (IR) light.

As used herein, the term “anti-reflective structures” refers to surface structures that change the angle of incidence of light such that the light enters the polymeric material beyond the critical angle and is internally transmitted.

As used herein, the term “glass-like surface” refers to the surface of a silicone elastomeric substrate (i.e., a substrate comprising a cross-linked silicone elastomeric material, a thermoplastic silicone elastomeric material, or both), where the surface comprises a silica ( $\text{SiO}_2$ ) stoichiometry and exhibits resistance to dirt and dust particle pick-up (i.e., dirt resistance) and/or abrasion resistance comparable or at least similar to that exhibited by a 100% glass surface. A “glass-like surface” can be hydrophilic, but may not need to be hydrophilic. In addition, a “glass-like surface” exhibits a degree of light transmission, after being subjected to the dirt pick-up test, the falling sand test or both tests, as described below, that is acceptable for its intended light energy absorbing application. As used herein, a portion of the anti-reflective structured surface is substantially glass-like, when that surface exhibits a desirable resistance to the Dirt Pick-Up Test, the Falling Sand Test, or both tests, as evidenced by the impact such testing has on the degree of light transmission and/or the change in light transmission exhibited by the glass-like surface after being so tested.

As used herein, the term “silica or  $\text{SiO}_2$  stoichiometry” refers to a composition containing silicon and oxygen in proportions sufficiently close to the stoichiometry of silica (i.e., a 2 to 1 ratio of oxygen to silicon) that the composition exhibits at least some of the properties of silica glass. For example, a composition has a silica stoichiometry, when it has an oxygen to silicon ratio of at least 1.25 to 1.00, at least 1.5 to 1.0, at least 1.75 to 1.00 and preferably closer to or equal to an oxygen to silicon ratio of about 2.0 to 1.0.

The terms “comprises”, “comprising”, “including” and variations thereof do not have a limiting meaning where these terms appear in the description and claims.

The words “preferred” and “preferably” refer to embodiments of the invention that may afford certain benefits, under certain circumstances. However, other embodiments may also be preferred, under the same or other circumstances. Furthermore, the recitation of one or more preferred embodiments does not imply that other embodiments are not useful, and is not intended to exclude other embodiments from the scope of the invention.

As used herein, “a,” “an,” “the,” “at least one,” and “one or more” are used interchangeably, unless the content clearly dictates otherwise.

The term “and/or” means one or all of the listed elements or a combination of any two or more of the listed elements (e.g., using UV light, UVO, oxygen plasma, and/or heat to treat the anti-reflective structured surface means using UV light, UVO, oxygen plasma, heat or any combination of the three).

Also herein, the recitations of numerical ranges by endpoints include all numbers subsumed within that range (e.g., the range 1 to 5 includes 1, 1.5, 2, 2.75, 3, 3.80, 4, 4.6, 5, 5.3, etc.) and any range within that range.

The uses of anti-reflective structured films, as disclosed herein, have demonstrated reductions in the amount of light that is reflected and does not reach the light absorbing element(s) of the light energy absorbing device. For example, such anti-reflective structured films have enabled conventional photovoltaic solar modules to experience average power output increases in the range of from about 3% to about 7%. The present invention can help maintain the transmission of light through such anti-reflective structured films, during the life of the light energy absorbing device, by improving the resistance to dirt and dust particle pick-up (i.e., dirt resistance) and/or abrasion resistance of the exposed surface of the anti-reflective structured film. In this way, the present invention can help to reduce the amount of incident light reflecting off of the light exposed surface(s) of such light energy absorbing devices. In particular, by having a glass-like surface, the structured face of the present invention is easier to clean (i.e., is more resistant to dirt pick-up), and has relatively good mechanical durability (e.g., resistance to falling sand) compared to the same silicone elastomeric material without a glass-like surface, as well as compared to the same structured face made with other polymeric materials (e.g., polyurethanes). Dirt and dust particles that do accumulate on such a structured face can also be relatively easier to clean.

Light energy absorbing devices, and especially the structured face of the anti-reflective structured film, may be exposed to a variety of detrimental conditions from outside environments. For example, the structured face can be exposed to environmental elements such as rain, wind, hail, snow, ice, blowing sand, and the like which can damage the structured surface of the structured face. In addition, long term exposure to other environmental conditions such as heat and UV radiation exposure from the sun can also cause degradation of the structured face. For example, many polymeric organic materials are susceptible to breaking down upon repeated exposure to UV radiation. Weatherability for light energy absorbing devices such as, for example, a solar energy conversion device is generally measured in years, because it is desirable that the materials be able to function for years without deterioration or loss of performance. It is desirable for the materials to be able to withstand up to 20 years of outdoor exposure without significant loss of optical transmission or mechanical integrity. Typical polymeric organic materials are not able to withstand outdoor exposure without loss of optical transmission or mechanical integrity for extended periods of time, such as 20 years. In at least some embodiments, the structured face of the present invention is expected to exhibit dirt resistance and/or mechanical durability in the range of from at least about 5 years to at least about 20 years, and possibly longer (e.g., at least about 25 years). In addition, because it is made of a silicone material, the structured face can exhibit long term UV stability of at least about 15 years, about 20 years or even about 25 years.

These and other potential advantages of the invention are further shown and described in the drawings and detailed description of this invention, where like reference numerals are used to represent similar parts. It is to be understood, however, that the drawings and description are for illustration purposes only and should not be read in a manner that would unduly limit the scope of this invention.

BRIEF DESCRIPTION OF THE DRAWINGS

In the accompanying drawings:

FIG. 1 is a side edge view of a transparent anti-reflective structured film embodiment of the present invention;

5 FIG. 2 is a side edge view of an alternative transparent anti-reflective structured film embodiment of the present invention;

FIG. 3 is a side edge view of another transparent anti-reflective structured film embodiment of the present invention;

10 FIG. 4 is a side view of a light energy absorbing device embodiment having a transparent anti-reflective structured film disposed so as to increase the amount of light being absorbed by a light absorber; and

FIG. 5 is a side view of another light energy absorbing device embodiment showing the paths of reflection incident light can travel when so as to increase the amount of light absorbed by the light absorber.

15 DETAILED DESCRIPTION OF CERTAIN EMBODIMENTS

The description that follows more particularly exemplifies illustrative embodiments. In describing the following embodiments of the present invention, specific terminology is used for the sake of clarity. The invention, however, is not intended to be limited to the specific terms so selected, and each term so selected includes all technical equivalents that operate similarly. In addition, the same reference numbers are used to identify the same or similar elements of the different illustrated 20 embodiments.

Unless indicated to the contrary, the numerical parameters set forth in the foregoing specification and attached claims are approximations that can vary depending upon the desired properties sought to be obtained by those skilled in the art utilizing the teachings disclosed herein.

25 Referring to FIG. 1, an exemplary transparent anti-reflective structured film 10 comprises a structured film substrate 12 that has a major structured face 14 with anti-reflective structures, for example, in the form of prismatic riblets 16 that are anti-reflective to light (see FIG. 5). Each anti-reflective structure (e.g., riblet) 16 has a tip angle  $\alpha$ , a trough to peak height  $H_{TP}$ , and a structured surface 18 that is exposed. Two adjacent anti-reflective structures (e.g., riblets) 16 define a trough angle  $\beta$  and a peak to peak distance  $D_{PP}$  therebetween. The film 10 further comprises a base portion 20 from which the anti-reflective structures 16 extend. The base portion 20 can be an integrally formed part of the structures 16 as illustrated, or a separate layer as indicated by the dashed line 21. To improve their durability, it can be desirable to provide the riblets 16 with rounded peaks  $R_P$  and rounded troughs  $R_T$ .

30 The structured film substrate 12 comprises a cross-linked silicone elastomeric material. The silicone elastomeric material may be, for example, a two-part silicone rubber that can be thermally cured

or condensation cured (e.g., Momentive RTV615 Silicone), a polydimethyl siloxane that can be electron beam (e-beam) cured (e.g., DMS-S51 available from Gelest, located at Morrisville, PA), etc., or a combination thereof. Other specific examples of silicone materials that may be used to form the structured surface are other hydrosilylation cure silicones such as, e.g., Sylgard 184, available from Dow 5 Corning or Elastosil 601, available from Wacker Chemie AG); condensation cure silicones such as, e.g., 10 732 and 734 RTV silicones available from Dow Corning, and electron beam crosslinkable silicones such as, e.g., DMS-S42 available from Gelest or EL Polymer NA available from Wacker.

The structured face 14 is exposed to a treatment that transforms the cross-linked silicone elastomeric material defining all, most or at least a substantial portion of the surface 18 of each structure 10 16 into a glass-like material comprising a SiO<sub>2</sub> stoichiometry. Such treatments can include, for example, exposing the silicone material to one or a combination of vacuum-ultraviolet (VUV) light (e.g., a 172 nm Excimer VUV lamp treatment), vacuum-ultraviolet light and ozone (VUVO), a flame, oxygen plasma, and any other treatment that can convert silicon atoms in the silicone elastomeric material so as to exhibit 15 the SiO<sub>2</sub> stoichiometry. This glass-like material is formed to a depth D into each structure 16. The depth D of the glass-like material depends on the exposure intensity and/or duration of the treatment. Thus, the glass-like material can be formed to a depth D that leaves a core or remainder 22 of cross-linked silicone elastomeric material in each of the structures 16 or that leaves each structure 16 completely formed of the 20 glass-like material. With the later embodiment, there can be a remainder of the film substrate 12 (e.g., below dashed line 21) made of the untreated silicone elastomeric material. The closer the material forming the structured surface 18 exhibits a SiO<sub>2</sub> stoichiometry, the more likely the surface 18 will be 25 resistant to dirt and dust particle pick-up (indicated by the dirt pick-up test results), abrasion resistant (indicated by the falling sand test results), or both.

Not wishing to be bound by theory, it is believed that the glass-like surface is formed by breaking Si-C bonds on and near the exposed surface of the anti-reflective structures through absorption of high 25 energy photons or through thermal oxidation. After the breaking of such Si-C bonds, SiOH is produced, which condenses to form the SiO<sub>2</sub> stoichiometry.

The resistance to dirt and dust particle pick-up (indicated by the dirt pick-up test results) of the structured surface 18 can be further improved by coating the glass-like surface with a coating of at least one layer 23 of agglomerates of silica nanoparticles. The agglomerates comprise a three-dimensional 30 porous network of silica nanoparticles, and the silica nanoparticles are bonded to adjacent silica nanoparticles. It can be desirable for the silica nanoparticles to have an average size (i.e., a major diameter or dimension) in the range of from about 5 nanometers (nm) up to and including about 100 nm. It can be desirable for the silica nanoparticle coating to have a thickness up to about 1000 nm, and it can also be desirable for the silica nanoparticle coating to have a thickness in the range of from about 20 nm 35 up to and including about 100 nm. Applying coatings of greater than 100 nm can be problematic. For example, such thick coatings of nanoparticles can result in excessive agglomerating of the nanoparticles.

and/or can cause the coating to exhibit an undesirable coloring affect. Such a layer of silica nanoparticles may include any one or a combination of those disclosed in the PCT Publication No. WO2009/140482, Published US Patent Application No. US2001/036547, and US Patents Nos. US 3,986,997 and US 4,027,073, as well as PCT Application No. US2010/056773, US Provisional Patent Applications Nos. 5 61/390491 filed October 6, 2010 and 61/262423 filed November 18, 2009, which are all incorporated herein by reference in their entirety.

It can be desirable for the film 10, or any other transparent anti-reflective structured film according to the invention, to be used in combination with an optional transparent support backing 24. With such an embodiment, the support backing 24 has a major face 24a, and the structured film substrate 10 12 further comprises a major backing face 12a bonded to the major face 24a of the support backing 24 so as to form a transparent reinforced anti-reflective structured film. The support backing 24 can comprise a polymeric material or a glass or other transparent ceramic material. Exemplary polymeric materials may include at least one or a combination of a polymethyl(meth)acrylate (PMMA) film, polyvinylidene 15 fluoride (PVDF) film, polyethylene terephthalate (PET) film, primed PET film, polycarbonate film, cross-linked polyurethane film, acrylate film, ethylene tetrafluoroethylene (ETFE), fluorinated ethylene-propylene (FEP) film, or blends thereof. Ultra-violet light absorbers (such as, e.g., Tinuvin 1577 from Ciba Geigy, and Sukano TA11-10 MB03 PMMA-UVA and Sukano TA07-07 MV02 PET-UVA, both 20 available from Sukano Polymers Corporation, Duncan, SC) can be incorporated into PMMA and blends of PVDF and PMMA for improved outdoor durability. The other transparent ceramic material may be, e.g., quartz crystal, etc. Transparent nonwoven or woven fiber materials, or chopped transparent fibers, 25 may also be used to form the support backing 24. Such fiber materials can either be disposed in the silicone elastomeric material forming the structured film 10, disposed on the structured film 10, or both.

The transparent support backing 24 can also be chosen so as to dissipate static electricity. For example, the support backing can comprise one or more polymeric materials that enable the support 25 backing 24 to dissipate static electricity. In order to dissipate static electricity, the transparent support backing 24 may also comprise an inherently static dissipative polymer such as those available as STATRITE X5091 polyurethane or STATRITE M809 polymethyl metacrylate from Lubrizol Corp. Alternatively, static dissipative salts such as FC4400 available from 3M Company can be blended into the 30 polymer used to make the transparent support backing 24 (e.g., PVDF). In addition, or alternatively, the structured film substrate 12 can comprise such static dissipative salts.

Instead of, or in addition to the support backing 24, it can also be desirable for the film 10, or any other transparent anti-reflective structured film according to the invention, to be used in combination with an optional moisture barrier layer 26. In such an embodiment, the moisture barrier layer 26 can be formed, for example, by laminating, coating or otherwise bonding the moisture resistant barrier layer 26 35 indirectly through one or more intermediate layers (e.g., the support backing layer 24) or directly onto the major backing face 12a of the structured film substrate 12. Alternatively, the moisture barrier layer 26

can be formed by formulating the composition of the film 10 so as to exhibit moisture barrier properties (e.g., so as to inhibit moisture absorption, permeation, etc.).

The moisture barrier may be, for example, a barrier assembly or one or more of the barrier layers disclosed in International Patent Application No. PCT/US2009/062944, U.S. Patent Nos. 7,486,019 and 5 7,215,473, and Published U.S. Patent Application No. US 2006/0062937 A1, which are incorporated herein by reference in their entirety. A moisture barrier may be useful, because silicone has a high moisture vapor transmission rate and photovoltaic cells are typically moisture sensitive. Therefore, by being backed with a moisture barrier layer, a transparent anti-reflective structured film of the invention can be used directly on moisture sensitive photovoltaic cells (e.g., Copper/Indium/Gallium/Selenium or 10 CIGS photovoltaic cells).

Referring to FIG. 2, in another embodiment 10a of the transparent anti-reflective structured film of the invention, the major structured face 14 is exposed to a degree/duration of treatment that causes all of the silicone elastomeric material of each of the anti-reflective structures 16 to be transformed into a glass-like material comprising a  $\text{SiO}_2$  stoichiometry. With this embodiment, a remainder 22 of the film 15 substrate 12 remains the silicone elastomeric material. Dashed line 23 separates the glass-like material portion of substrate 12 from the silicone elastomeric material portion. Again, the structured surface 18 can be further improved by coating the glass-like surface with a coating of at least one layer 23 of agglomerates of silica nanoparticles.

Referring to FIG. 3, in an additional embodiment 10b of the transparent anti-reflective structured film of the invention, each of the anti-reflective structures 16 extend out from a separate base portion 20'. The separate base portion 20' can be one or more layers of a cross-linked silicone elastomeric material, or the separate base 20' can be one or more layers of a different material (e.g., less expensive material like PMMA, PVDF and PET). The separate base 20' is adhered or otherwise bonded to the anti-reflective structures 16 by any suitable means, depending on the compatibility between the silicone elastomeric 25 material and the different material. For example, the base portion 20' can have a major face 20a that is optionally coated with a primer or otherwise treated (e.g., a corona treatment) or prepared for receiving and bonding with a major backing face 16a of each of the silicone elastomeric anti-reflective structures 16. The anti-reflective structures 16 can be formed, for example, by using a tooling film (not shown) having a micro-replicated pattern formed in at least one of its major surfaces that matches the desired 30 pattern of anti-reflective structures 16.

A layer of the desired silicone elastomeric precursor material can be extruded, coated or otherwise applied onto the surface of the base portion face 20a. The micro-replicated major surface of the tooling film can then be brought into contact with the layer of silicone elastomeric precursor material so as to form the exposed surface of the applied silicone elastomeric precursor material into the shape of the 35 desired anti-reflective structures 16. Alternatively, the layer of silicone elastomeric precursor material can be extruded, coated or otherwise applied onto the micro-replicated major surface of the tooling film

and then the exposed back surface of the applied precursor material can be laminated or otherwise brought into contact so as to bond with the surface of the base portion face 20a. Alternatively, the layer of silicone elastomeric precursor material can be extruded, coated or otherwise applied directly between the micro-replicated major surface of the tooling film and the surface of the base portion face 20a, as all 5 three layers are laminated together such as, for example, by passing through a nip roll station. Once the formed precursor material is in contact with the surface of the base portion face 20a, the silicone elastomeric precursor material is initially cross-linked or cured, the tooling is removed, followed by subsequent treatment to produce the glass-like material in at least the surface 18 of the anti-reflective structures 16.

10 The anti-reflective structures can comprise at least one or a combination of prismatic, pyramidal, conical, hemispherical, parabolic, cylindrical, and columnar structures. The anti-reflective structures comprising prisms can have a prism tip angle  $\alpha$  of less than about 90 degrees, less than or equal to about 60 degrees, less than or equal to about 30 degrees, or in the range of from about 10 degrees up to about 90 degrees. Such anti-reflective prism structure can also exhibit a trough-to-trough or peak-to-peak pitch or 15 distance in the range of from about 2 microns to about 2 cm. The anti-reflective structures comprising prisms can also have a prism tip angle in the range of from about 15 degrees to about 75 degrees. The anti-reflective structures comprising prisms can also have a trough-to-trough or peak-to-peak pitch in the range of from about 10 microns to about 250 microns. In one embodiment of an anti-reflective structure 16 with improved durability, the riblets 16 have rounded peaks  $R_p$  and troughs  $R_t$  with a radius of about 5 20 microns, a trough angle  $\beta$  of about 53 degrees, a peak to peak pitch or distance  $D_{pp}$  of about 50 microns, and a trough to peak height  $H_{tp}$  of about 37.7307 microns.

It can be desirable for the anti-reflective structures to exhibit a refractive index that is less than about 1.55, and preferably a refractive index that is less than about 1.50. When the anti-reflective structures comprise prism structures (e.g., linear prism structures or riblets), it can be desirable for each of 25 the prisms to narrow from their base to a tip having an apex angle that is less than about 90 degrees, and preferably less than or equal to about 60 degrees. It can be desirable for such a prism structure to have a trough to peak height in the range of from about 10 microns to about 250 microns. It can also be desirable for such a prism structure to have a trough to peak height in the range of from about 25 microns to about 100 microns.

30 It can be desirable for a transparent anti-reflective structured film of the invention to exhibit at least about 85%, 90%, 91%, 92%, 93%, 94%, 95%, 96%, 97%, 98% or 99% light transmission, after the structured surface is exposed to the dirt pick-up test, the falling sand test, or a combination of both tests. These tests are described below. It can also be desirable for a transparent anti-reflective structured film of 35 the invention to exhibit a change in light transmission of less than 10%, 9%, 8%, 7%, 6%, 5%, 4%, 3%, 2% or 1%, after the structured surface is exposed to the dirt pick-up test, the falling sand test, or a combination of both tests.

A transparent anti-reflective structured film of the invention may also comprise inorganic particles, and preferably nanoparticles in the silicone elastomeric material of the anti-reflective structures. These particles may comprise any suitable inorganic material (e.g., silica, zirconia, titania, etc., or any combination thereof). Such particles may also be coated with a silane surface modification in order to 5 promote dispersion in silicone. Such particles may have a size in the range of up to and including about 2.0 microns. Silica particles can be up to the micron size, but it is preferable for particles made of other materials to be used in the nanometer sizes (i.e., in the range of from about 5 nm up to and including about 50 nm). Such particles, especially nanoparticles, may also be loaded into the silicone elastomeric material in the range of from 0 wt.% up to and including about 60 wt.%.

10 Referring to FIG. 4, any embodiment of a transparent anti-reflective structured film 10 of the invention can be used in a light energy absorbing device 30 such as, for example, a light source thermal energy absorbing device (e.g., a solar hot water system), a photovoltaic device or any other light energy absorbing device. Such a device 30 also comprises a light absorber 32 (e.g., a photovoltaic cell) having a light energy receiving face 32a, with the transparent anti-reflective structured film 10 being disposed 15 relative to the light absorber 32 so as to be between a source of light energy (e.g., the sun) and the light energy receiving face 32a. In this way, light energy from the source passes through the structured film 10 before being absorbed by the light absorber 32. The film 10 can be bonded, adhered, mechanically fastened or otherwise disposed in direct contact with the light energy receiving face 32a. Alternatively, if desired, one or more of a transparent support backing 24 or other intermediate layers can be disposed 20 between the film 10 and the light absorber 32.

Referring to FIG. 5, by using a transparent anti-reflective structured film 10 of the invention with a light absorber 32 of a light energy absorbing device 30, incident light (represented by arrows 40) striking the surfaces 18 of the anti-reflective structures 16 are likely to be reflected multiple times (represented by arrows 40<sub>R</sub>). Such multiple reflections of the light 40 increases the probability of light 40 25 being refracted into the light absorber 32, as well as of increasing the incident light acceptance angles. In this way, the use of such transparent anti-reflective structures can increase the efficiency and energy output of the device 30.

When the light absorbing device is a photovoltaic device, the light absorber is a photovoltaic module having at least one photovoltaic cell for converting solar or other light energy into electrical 30 energy. The anti-reflective structured film reduces surface reflection, which can increase the amount of light impinging on the photovoltaic cell resulting in increased production of electric energy. By using a transparent anti-reflective structured film of the invention in this manner, efficiencies in converting light energy to electrical energy may be improved by at least about 3% and possibly in the range of from about 5% up to and including about 10%. Because the transparent anti-reflective structures are in the form of a 35 film, the photovoltaic cell can be sufficiently flexible and pliant so as to be wound into a roll or folded without being damaged.

5 A light energy absorbing device of the invention can be made by mechanically attaching, adhesively bonding or otherwise securing the anti-reflective structured film to the light absorber so that light can pass through the anti-reflective structured film to the light receiving face of the light absorber (e.g., photovoltaic cell). The light absorber can be, for example, a solar hot water heater or other light generated thermal energy absorbing device, a photovoltaic cell for converting solar or other light energy into electrical energy or a combination thereof.

10 A transparent anti-reflective structured film according to the present invention can be made by providing a transparent structured film substrate as described above and then treating the structured surface such that all, most or at least a substantial portion of the cross-linked silicone elastomeric material defining the structured surface is transformed into a glass-like material comprising an  $\text{SiO}_2$  stoichiometry. This glass-like surface is then coated with a coating comprising, consisting essentially of, or consisting of at least one layer of agglomerates of silica nanoparticles, with the agglomerates comprising a three-dimensional porous network of silica nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles.

15 The anti-reflective structured surface of the initial structured film substrate can be treated to form the glass-like surface, for example, by exposing the anti-reflective structured surface to any suitable treatment such as, for example, at least one or a combination of vacuum-ultraviolet (VUV) light (e.g., by 172 nm Excimer treatment), vacuum-ultraviolet light and ozone (VUVO), oxygen plasma, and heat (e.g., induction heating, a flame, etc.). In order to be suitable for use in a high volume manufacturing setting, 20 such treatments need to be performed relatively quickly. In particular, for example, high speed (i.e., high volume) web-based manufacturing processes would require the use of treatment times as short as, for example, less than or equal to about 5 minutes. At the same time, the surface of the silicone material needs to be treated for a sufficient period of time (e.g., 5 to 300 seconds of 172 nm Excimer lamp exposure) and at a suitable energy level (e.g., 10 to 50  $\text{mW/cm}^2$  in a nitrogen inert atmosphere of less than 25 50 ppm oxygen) to produce the desired level of conversion from silicon to the  $\text{SiO}_2$  stoichiometry. Depending on the settings (e.g., intensity and/or duration) of the treatment used to produce the glass-like surface, there may be a remaining portion of the structured film substrate that has not converted to the glass-like material. As seen, for example, in FIG. 2, the treatment settings may also be chosen so that all of the cross-linked silicone elastomeric material defining each of the anti-reflective structures is 30 transformed into the glass-like material comprising an  $\text{SiO}_2$  stoichiometry. Alternatively, the treatment settings may be chosen so that a core portion of each of the anti-reflective structures remains the silicone elastomeric material (see FIGS. 1, 3 and 4). To save on energy costs, it can be desirable to minimize the depth and degree to which the anti-reflective structured surface is converted into a the glass-like surface.

35 The transparent structured film substrate can be made by providing a silicone elastomeric precursor material that is curable so as to form the cross-linked silicone elastomeric material. This silicone elastomeric precursor material is formed into the shape of the structured film substrate using any

suitable forming technique. For example, appropriately sized-grooves can be formed in a substrate and then the substrate used as a mold surface on which the silicone elastomeric precursor material is coated so as to cast the major structured face with anti-reflective structures of the structured film substrate. Such a mold substrate can be made, for example, in accordance with the techniques and equipment disclosed in 5 U.S. Patent Publication No. US 2006/0234605, which is incorporated herein by reference in its entirety. While in this shape, the silicone elastomeric precursor material is cured so as to form the structured film substrate. Alternatively, the tool disclosed in U.S. Patent Publication No. US 2006/0234605 can be used to cast the appropriately sized-grooves in a polymeric mold substrate (e.g., in the form of a film) that is then used as the mold surface. Depending on the silicone elastomeric precursor material used, the curing 10 process can involve subjecting the precursor material to a cross-linking treatment (e.g., a thermal and/or radiation treatment). When the precursor material is a two-part self curing silicone elastomeric material, the curing process can involve maintaining the precursor material in contact with the mold surface for a long enough period, after the two parts are mixed, to allow cross-linking to occur.

The step of coating the glass-like surface with at least one layer of agglomerates of silica 15 nanoparticles can be performed in accordance with the techniques disclosed in, for example, the previously incorporated PCT Publication No. WO2009/140482, Published US Patent Application No. US2001/036547, and US Patents Nos. US 3,986,997 and US 4,027,073, as well as PCT Application No. US2010/056773, US Provisional Patent Applications Nos. 61/390491 filed October 6, 2010 and 61/262423 filed November 18, 2009. In order to obtain good wetting of the silica nanoparticle coating, it 20 can be desirable for the glass-like surface to exhibit hydrophilic properties. It may also be desirable to apply heat to the silica nanoparticle coating (e.g., with a flame, induction heat, etc.) so as to induce the nanoparticles to sinter together. It may be desirable to limit such a sintering step, otherwise excessive flattening of the silica nanoparticle agglomerates may result. While various coating techniques may be employed to apply the one or more layers of agglomerates of silica nanoparticles, it has been found 25 desirable to use conventional air-knife coating techniques to apply such a coating. Such air-knife coating processes help to insure a relatively thin coating of the silica nanoparticle agglomerates onto the glass-like anti-reflective structured surface.

The following Examples have been selected merely to further illustrate features, advantages, and/or other details of the invention. It is to be expressly understood, however, that while the Examples 30 serve this purpose, the particular ingredients and amounts used as well as other conditions and details are not to be construed in a manner that would unduly limit the scope of this invention.

EXAMPLESExample 1

RTV615 Part A and RTV615 Part B available from Momentive Performance Materials of Waterford, New York, were mixed at a 10:1 ratio and coated 100 microns thick onto each of four quartz glass slides. The silicone coated quartz glass slides were subsequently heated to 85°C for 30 minutes in a convection oven to cross-link/cure the thermally curable silicone elastomeric precursor material. These glass slides coated with cross-linked silicone (Samples 2-5) were then exposed to flame treatment as shown in Table 1. These flame treated silicone coated glass constructions were then analyzed by nano-indentation for Storage Modulus. Modulus changes in these flame treated silicone coated glass constructions are shown in Table 1.

Table 1

Flame Treated RTV615 Silicone			Nano-indenter Storage Modulus MegaPascals
Sample	Flame Conditions Temperature (°C)	Time (Seconds)	
1	0	0	12.3
2	2000	30	9.7
3	2000	30	8.6
4	2000	60	22.7
5	2000	60	19.5

15

Example 2

Fourteen 7.6 cm (3 inch) by 5.1 cm (2 inch) glass slides (available from VWR International, LLC.) were primed with a nano-silica based primer. The nano-silica primer consists of a 5 % by weight blend of a 70:30 ratio of a first colloidal silica (“NALCO 1115 COLLOIDAL SILICA”) and a second colloidal silica (“NALCO 1050 colloidal SILICA) in H<sub>2</sub>O, brought to a pH of 2.5-2.0 with HNO<sub>3</sub>. A thin (about 100 nanometers) even coating of the primer was applied to each glass slide by wiping the surface with a small wipe (obtained under the trade designation “KIMTECH” from Kimberly-Clark, Roswell, GA) dampened with the nano-silica primer solution. The primer was allowed to dry at room temperature. An addition cure silicone (Sylgard 184, available from Dow Corning, Midland, MI) was mixed at a ratio of 10:1 (part A to Part B) and was applied to the surface of the primed glass slides in a smooth coating at a thickness of 45 micrometers and allowed to fully cure over 48 hours. For Samples 6 and 7, the surface of the silicone was not treated, while Samples 8 through 19 were oxygen plasma treated using a commercial batch plasma system (Plasmatherm Model 3032) configured for reactive ion etching (RIE) with a 27-inch lower powered electrode and central gas pumping. The chamber is pumped by a roots

blower (Edwards Model EH1200) backed by a dry mechanical pump (Edwards Model iQDP80). RF power is delivered by a 3 kW, 13.56 Mhz solid-state generator (RFPP Model RF30H) through an impedance matching network. The system has a nominal base pressure of 5 mTorr. The flow rates of the gases are controlled by MKS flow controllers. Substrates for plasma treatment are placed on the lower 5 powered electrode. The silicone coated glass slides were placed on the powered electrode of the batch plasma apparatus. The plasma treatment was performed with an oxygen plasma by flowing oxygen gas (Oxygen Service Corporation, UHP Grade) at a flow rate of 500 standard cm<sup>3</sup>/min and plasma power of 3000 watts for 90 seconds. After the plasma treatment was completed, the chamber was vented to atmosphere and the samples removed.

10 Samples 6 through 17, along with one uncoated plain glass slide, were periodically subjected to the dirt pick-up test #1 described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated for each in the below Table 2 (Dirt Pick-up Test #1 Results). Transmission was tested using a Hazemeter. The tabulated data shows a significant increase in light 15 transmission for the oxygen plasma treated Samples 8 through 17 compared to untreated Samples 6 and 7. This difference in light transmission is caused by the additionally treated silicone elastomer surface (Samples 8 through 17) picking up and holding onto less dirt than the untreated Samples 6 and 7. The Table 2 data also shows that the treated Samples 6 through 17 exhibited light transparency comparable to that of the plain glass slides.

20 Table 2 (Dirt Pick-up Test #1 Results)

Sample	Time after O <sub>2</sub> Plasma (hr)	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
6	No Plasma Treatment	95.4	56.9	-38.5
7	No Plasma Treatment	95.2	62.1	-33.1
8	1	95.3	94.5	-0.8
9	1	95.2	94.4	-0.8
10	8	95.5	94.4	-1.1
11	8	95.6	94.4	-1.2
12	32	95.6	95	-0.6
13	32	95.6	95.1	-0.5
14	197	95.9	94.6	-1.3
15	197	95.5	93.9	-1.6
16	1131	95.7	95.4	-0.3
17	1131	95.8	95.1	-0.7
Plain Glass Slide		94.5	94.3	-0.2

25 Samples 18 and 19 were subjected to the falling sand test described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated in the below Table 3

(Falling Sand Test). The tabulated data shows that the treated Samples 18 and 19 exhibited light transparency comparable to that of the plain glass slides.

Table 3 (Falling Sand Test Results)

Sample	Time after O <sub>2</sub> Plasma (hr)	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
18	1131	95.8	93.2	-2.6
19	1131	95.7	93.3	-2.4
Plain Glass Slide		94.7	92.2	-2.5

5

Samples 18 and 19 were also subjected to the dirt test immediately following the falling sand test with initial transmission (T<sub>i</sub>) representing the transmission value after the falling sand test, final transmission (T<sub>f</sub>) representing the transmission value after the dirt test and the difference between the initial and final transmission (T<sub>d</sub>) being tabulated in the below Table 4 (Dirt Test #1 Results Immediately 10 Following Falling Sand Test). The tabulated data shows that the oxygen plasma treated silicone samples are within 2% of the performance of the plain glass slide.

Table 4 (Dirt Test #1 Results Immediately Following Falling Sand Test)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
18	93.2	91.7	-1.5
19	93.3	91.9	-1.4
Glass Slide	92.2	92.3	0.1

15

### Example 3

Two 30.5 cm (12 inch) by 15.2 cm (6 inch) sheets of polyethylene terephthalate (PET) (obtained under the trade designation “MELINEX 618” from DuPont Teijin Films, Chester, VA) were primed with a nano-silica based primer. The nano-silica primer consists of a 5 % by weight blend of a 70:30 ratio of a first colloidal silica (“NALCO 1115 COLLOIDAL SILICA”) and a second colloidal silica (“NALCO 20 1050 colloidal SILICA) in H<sub>2</sub>O, brought to a pH of 2.5-2.0 with HNO<sub>3</sub>. A thin (about 100 nanometers) even coating of the primer was applied to each glass slide by wiping the surface with a small wipe (obtained under the trade designation “KIMTECH” from Kimberly-Clark, Roswell, GA) dampened with the nano-silica primer solution. The primer was allowed to dry at room temperature. A silanol terminated 25 polydimethylsiloxane fluid (DMS-S51, available from Gelest Inc., Morrisville, PA) was coated on to the primed PET film at a thickness of 100 micrometers using a notch bar coater. The PET film with silicone coating was taped to a carrier web and passed through an electron beam at an acceleration voltage of 190 keV and a dose of 10 MRads. The E-beam unit was a broadband curtain type electron beam processor (obtained from PCT Engineered Systems, LLC). Six pieces (Samples 20, 21, 22, 23, 24 and 25) were cut

from one of the silicone coated film, Samples 20, 21, 22 and 23 had the surface of the silicone coating exposed to 120 seconds of an 172 nm Excimer Lamp (Model UER20-172, available from Ushio America, Inc., Cypress, CA) at an irradiance of 10 mW/cm<sup>2</sup> with nitrogen purging. Samples 24 and 25 did not receive an Excimer Lamp treatment. Six pieces (Samples 26, 27, 28, 29, 30 and 31) were cut from the other silicone coated film which received an extra e-beam dose of four passes at 120 keV and a dose of 15 MRad/pass to further cross-link the surface. Samples 26, 27, 28 and 29 were then exposed 120 seconds of an 172 nm Excimer Lamp at an irradiance of 10 mW/cm<sup>2</sup>. Samples 30 and 31 did not receive an Excimer Lamp treatment.

Samples 20, 21, 24, 26, 27 and 30 were subjected to the dirt pick-up test #1 described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated for each in the below Table 5 (Dirt Pick-up Test #1 Results). The tabulated data shows a significant increase in light transmission for the additionally treated Samples 20, 21, 26 and 27 compared to the untreated Samples 24 and 30. This difference in light transmission is caused by the additionally treated silicone elastomer surface (Samples 20, 21, 26 and 27) picking up and holding onto less dirt than the untreated Samples 24 and 30. The Table 5 data also shows that the treated Samples 20, 21, 26 and 27 exhibited light transparency comparable to that of the plain glass slide from Table 5. The table also shows that the extra electron beam cross linking in Sample 30 results in less loss of transmission even without the 172 nm Excimer lamp treatment compared to Sample 24 which received neither electron beam nor 172 nm Excimer lamp treatment.

Table 5 (Dirt Test #1 Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
20	95.9	95.2	-0.7
21	95.6	95.2	-0.4
24	96.5	55.5	-41
26	96	94.5	-1.5
27	95.9	93.3	-2.6
30	96.3	82.6	-13.7

Samples 22, 23, 25, 28, 29 and 31 were subjected to the falling sand test described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated in the below Table 6 (Falling Sand Test). The tabulated data shows an increase in light transmission (i.e., Falling Sand Test results) for the Samples 28 and 29 that were additionally electron beam cross-linked before the 172 nm Excimer lamp treatment compared to Samples 22 and 23 that were not additionally electron beam cross-linked. The table also shows that extra electron beam cross linking without 172 nm Excimer lamp treatment in Sample 31 results in less loss of transmission compared to Sample 25 which did not receive

either extra electron beam cross linking or 172 nm Excimer lamp treatment as well as Samples 22 and 23 which received 172 nm Excimer lamp treatment only and Samples 28 and 29 which received both extra electron beam cross linking and 172 nm Excimer lamp treatment..

5 Table 6 (Falling Sand Test Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
22	96	84.5	-11.5
23	95.8	88	-7.8
25	96.2	92.5	-3.7
28	95.9	89.6	-6.3
29	95.9	90.2	-5.7
31	96.4	95.2	-1.2

**Example 4**

Two 30.5 cm (12 inch) by 15.2 cm (6 inch) sheets of polyethylene terephthalate (PET) (obtained under the trade designation “MELINEX 618” from DuPont Teijin Films, Chester, VA) were primed with a nano-silica based primer. The nano-silica primer consists of a 5 % by weight blend of a 70:30 ratio of a first colloidal silica (“NALCO 1115 COLLOIDAL SILICA”) and a second colloidal silica (“NALCO 1050 colloidal SILICA) in H<sub>2</sub>O, brought to a pH of 2.5-2.0 with HNO<sub>3</sub>. A thin (about 100 nanometers) even coating of the primer was applied to each glass slide by wiping the surface with a small wipe (obtained under the trade designation “KIMTECH” from Kimberly-Clark, Roswell, GA) dampened with the nano-silica primer solution. The primer was allowed to dry at room temperature. A thermal cure silicone elastomer (Sylgard 184, available from Dow Corning, Midland, MI) was coated between the primed PET and a polypropylene tooling film comprising anti-reflection surface structure. The PET film with silicone coating and polypropylene tooling film were placed in an oven at 75° C for an 90 minutes to cure the silicone coating. After removing the films from then oven the tooling film was removed resulting in a surface structured anti-reflection silicone coating attached to the PET film. In order to provide extra cross-linking for improved durability at the surface of the silicone surface structures, one sheet of the coated film was then taped to a carrier web and passed through an electron beam at an acceleration voltage of 120 keV and 15 MRads/pass for four passes. The E-beam unit was a broadband curtain type electron beam processor (obtained from PCT Engineered Systems, LLC).

Four pieces (Samples 32, 33, 34, and 35) were cut from the silicone coated film that did not receive the extra electron beam treatment, Samples 32 and 33 had the surface of the silicone coating exposed to 300 seconds of an 172 nm Excimer Lamp (Ushio Model UER20-172) at an irradiance of approximately 10 mW/cm<sup>2</sup> with nitrogen purging. Samples 34 and 35 did not receive an Excimer Lamp treatment. Four pieces (Samples 36, 37, 38 and 39) were cut from the other silicone coated film which received an extra e-beam treatment. Samples 37 and 38 were then exposed 300 seconds of an 172 nm

Excimer Lamp at an irradiance of approximately 10 mW/cm<sup>2</sup>. Samples 38 and 39 did not receive an Excimer Lamp treatment.

Samples 32, 34, 36 and 38 were subjected to the dirt pick-up test #1 described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated for each in the below Table 7 (Dirt Pick-up Test #1 Results). The tabulated data shows a significant increase in light transmission for the 172 nm Excimer lamp treated Samples 32 and 36 compared to the untreated Samples 34 and 38. This difference in light transmission is caused by the 172 nm Excimer lamp treated silicone elastomer surface (Samples 32 and 36) picking up and holding onto less dirt than the untreated Samples 34 and 38. The Table 7 data also shows that Sample 36 which received additional electron beam cross linking before the 172 nm Excimer lamp treatment produces better dirt pick-up test results than Sample 32 which received only the 172 nm Excimer lamp treatment.

Table 7 (Dirt Pickup Test #1 Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
32	101	97.6	-3.4
34	101	73.2	-27.8
36	101	99.3	-1.7
38	101	93.6	-7.4

Samples 33, 35, 37 and 39 were subjected to the falling sand test described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated in the below Table 8 (Falling Sand Test). The tabulated data shows an increase in light transmission (i.e., Falling Sand Test results) for the Samples 33 and 37 that received the 172 nm Excimer lamp treatment compared to Samples 35 and 39 that were not treated with the 172 nm Excimer lamp. An increase is also seen in light transmission (i.e., Falling Sand Test results) for the Samples 37 and 39 that were additionally electron beam cross-linked before the 172 nm Excimer lamp treatment compared to Samples 33 and 35 that were not additionally electron beam cross-linked.

Table 8 (Falling Sand Test Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
33	101	95.7	-5.3
35	101	90.4	-10.6
37	101	96.4	-4.6
39	101	94.4	-6.6

Samples 33, 35, 37 and 39 were also subjected to the dirt test #1 immediately following the falling sand test with initial transmission (Ti) representing the transmission value after the falling sand

test, final transmission (T<sub>f</sub>) representing the transmission value after the dirt test and the difference between the initial and final transmission (T<sub>d</sub>) being tabulated in the below Table 9 (Dirt Test #1 Results Immediately Following Falling Sand Test). The tabulated data shows that Samples 37 and 39, which received the extra electron beam surface treatment, maintained the sample's dirt resistance as seen by less loss of transmission after the dirt test compared to Sample 33 and 35 which did not receive the extra electron beam surface treatment.

Table 9 (Dirt Test #1 Results Immediately Following Falling Sand Test)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
33	95.7	86.6	-9.1
35	90.4	70.5	-19.9
37	96.4	94.2	-2.2
39	94.4	91.3	-3.1

10 **Example 5**

Silanol terminated polydimethyl siloxane (PDMS-S51 from Gelest) was coated at a thickness of 25 microns onto a sheet of polyethylene terephthalate (PET) (obtained under the trade designation "MELINEX 618" from DuPont Teijin Films, Chester, VA) that was primed with SS4120 (available from Mometive Performance Materials, Waterford, NY). The PET film with silicone coating was taped to a carrier web and passed through an electron beam at an acceleration voltage of 160 keV and a dose of 12 MRads. The E-beam unit was a broadband curtain type electron beam processor (obtained from PCT Engineered Systems, LLC). The silicone coated PET film was then cut into six pieces. Two of these silicone coated PET pieces (Samples 41 and 44) were floated on a thin film of water on top of 6.4 mm thick plate of aluminum and then exposed to a flame treatment by exposing the coated surface to an approximately 2000 °C flame for 20 seconds. Another two of the silicone coated PET pieces (Samples 42 and 45) were exposed to an additional electron beam treatment of 5 passes, at 140 kV and 12 Mrad/pass, to further cross link the surface of the silicone, and then were floated on a thin film of water on top of 6.4 mm thick plate of aluminum and exposed to an approximately 2000 °C flame for 20 seconds. The last silicone coated PET piece (Sample 40 and 43) was not additionally treated.

25 Samples 41 and 42 were subjected to the Dirt Pick-Up Test #2 described below, with the initial light transmission (T<sub>i</sub>) before being tested, the final light transmission (T<sub>f</sub>) after being tested, and the difference between the initial and final light transmissions (T<sub>d</sub>) being tabulated for each in the below Table 10 (Dirt Pick-up Test #2 Results). The tabulated data shows a significant increase in light transmission for the additionally treated Samples 41 and 42 compared to the untreated Sample 40. This 30 difference in light transmission is caused by the additionally treated silicone elastomer surface (Samples 41 and 42) picking up and holding onto less dirt than the untreated Sample 40. The Table 10 data also

shows that additional cross linking before the flame treatment (i.e., Sample 42) produces better Dirt Pick-Up Test results.

Table 10 (Dirt Pick-up Test #2 Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
40	95.4	87.3	-8.1
41	95.2	92.5	-2.7
42	95.1	93	-2.1

5

Samples 43, 44 and 45 were subjected to the falling sand test described below, with the initial light transmission (Ti) before being tested, the final light transmission (Tf) after being tested, and the difference between the initial and final light transmissions (Td) being tabulated in the below Table 11 (Falling Sand Test). The tabulated data shows a significant increase in light transmission for the 10 additionally treated Sample 45 compared to the untreated Sample 43. This difference in light transmission is caused by the additionally treated silicone elastomeric surface (Sample 45) picking up and holding onto less dirt than the untreated Sample 43. The tabulated data also shows an increase in light transmission (i.e., Falling Sand Test results) for the Sample 45 that was additionally cross-linked before the flame treatment compared to Sample 44 that was not additionally cross-linked.

15

Table 11 (Falling Sand Test Results)

Sample	T <sub>i</sub>	T <sub>f</sub>	T <sub>d</sub>
43	96.5	92.4	-4.1
44	96.2	91.5	-4.7
45	95.2	92.1	-3.1

#### TEST METHODS

##### **Dirt Pick-up Test #1**

20 Coating soil resistance is tested using an apparatus constructed from a 95 mm square plastic petri-dish (trade name Falcon 35112; available from Becton Dickinson Labware) with a 5 cm hole drilled through bottom half of a petri-dish. A 5 cm by 8 cm coated sample is attached with adhesive tape on the outside of the petri-dish covering the 5 cm hole so that the coated surface of the sample is facing the inside of the petri dish and will be exposed directly to the test dirt. 50 g of Arizona Test Dirt (0-600 micrometer distribution; available from Powder Technology, Inc., Burnsville, MN) is placed into the 25 bottom half of the petri dish covering the coated samples. The two halves of the petri dish are combined securely and shaken lightly in side-to-side cycles so that the dirt tumbles back and forth over the surface of the sample. The sample is shaken for 60 cycles at a rate of 1 cycle per second. The sample is then removed from the testing apparatus and gently tapped to remove and loosely attached dirt. The

transmittance of the coated sample is measured before and after the dirt test using a Haze Gard Plus available from BYK-Gardner.

### **Dirt Pick-up Test #2**

5 As used herein, this dirt pick-up test involves tumbling a sample of the transparent anti-reflective structured film inside a 1 gallon Nalgen jar with 100 grams of fine/dusty Arizona dirt. A 1.5"x2.5" sample is attached to a larger 3"x5" piece of 10 mil PET. The sample and dirt tumble due to baffles on the inside of the Nalgen jar, which is laid horizontally on motorized rollers. After two minutes of tumbling the sample is blown off with canned air to remove excess dirt so that only dirt that is bound to  
10 the surface remains. The transmittance of the coated sample is measured before and after the dirt test using a Haze Gard Plus available from BYK-Gardner.

### **Falling Sand Test**

15 Coating abrasion resistance is tested using a Falling Sand Abrasion Tester (trade name HP-1160) available from Humboldt MFG. Co. A 5 cm by 8 cm coated sample is attached with adhesive tape to the testing platform centered underneath the outlet of the falling sand tube. 1000 g of ASTM C778 silica sand, available from U.S. Silica Company, is loaded into the hopper that feeds the falling sand tube. The gate is opened and the sand begins to fall a distance of 100 cm through the falling sand tube and impinges on the surface of the coated sample. Water is run over the abraded surface for 5 seconds and then the  
20 surface is lightly wiped using a damp KimWipe (Kimberly-Clark). The sample is then lightly wiped with a dry KimWipe to dry the sample. The transmittance of the coated samples is measured before and after the falling sand test using a Haze Gard Plus available from BYK-Gardner.

### **Exemplary Embodiments of the Present Invention**

#### **25 Anti-Reflective Film Embodiment 1**

A transparent anti-reflective structured film, sheet, web or the like comprising:

30 a structured film substrate having a major structured face and a major backing face, with the structured face comprising anti-reflective structures defining a structured surface and being anti-reflective to light, at least a substantial portion, most, or all of the structured surface comprising a glass-like surface, at least the anti-reflective structures comprising a cross-linked silicone elastomeric material, and the glass-like surface comprising an SiO<sub>2</sub> stoichiometry.

#### **Film Embodiment 2**

35 The film according to film embodiment 1, wherein the glass-like surface comprises the SiO<sub>2</sub> stoichiometry to a depth of at least about 1, 5, 10, 15, 20, 25, 30, 35, 40, 45, 50, 55, 60, 65, 70, 75, 80, 85, 90, 95, or 100 nanometers or even deeper into each of the anti-reflective structures.

Film Embodiment 3

The film according to film embodiment 1, wherein the glass-like surface comprises SiO<sub>2</sub> stoichiometry to a depth of up to about 100, 110, 120, 130, 140, or 150 nanometers or even deeper (e.g., as deep as 1, 2, 3, 4 or 5 microns) into each of the anti-reflective structures.

5 Film Embodiment 4

The film according to any one of film embodiments 1 to 3, wherein the glass-like surface comprises a minimum amount of at least about 10 molar % carbon atoms or at least about 20 molar % carbon atoms, up to about 40 molar % carbon atoms. It is believed that the relatively high amount of carbon atoms in the glass-like surface is indicative of the relatively short periods of time used in treating 10 the silicone material to produce the SiO<sub>2</sub> stoichiometry according to the present invention.

Film Embodiment 5

The film according to any one of film embodiments 1 to 4, wherein the anti-reflective structures comprise at least one or a combination of prismatic, pyramidal, conical, parabolic, hemispherical, cylindrical, and columnar structures.

15 Film Embodiment 6

The film according to any one of film embodiments 1 to 5, wherein the anti-reflective structures comprise prisms having a prism tip angle of less than about 90 degrees, less than or equal to about 60 degrees, or in the range of from about 10 degrees up to about 90 degrees and a pitch in the range of from about 2 microns to about 2 cm.

20 Film Embodiment 7

The film according to any one of film embodiments 1 to 6, wherein the anti-reflective structures comprise prisms having a prism tip angle in the range of from about 15 degrees to about 75 degrees and a pitch in the range of from about 10 microns to about 250 microns.

Film Embodiment 825 Film Embodiment 7

The film according to any one of film embodiments 1 to 7, wherein the anti-reflective structures comprise prisms having a trough to peak height in the range of from about 10 microns to about 250 microns.

Film Embodiment 930 Film Embodiment 8

The film according to any one of film embodiments 1 to 8, wherein the film exhibits at least about 80%, 81%, 82%, 83%, 84%, 85%, 90%, 91%, 92%, 93%, 94%, 95%, 96%, 97%, 98% or 99% light transmission, after the structured surface is exposed to the Dirt Pick-Up Test, the Falling Sand Test, or both tests.

Film Embodiment 1035 Film Embodiment 9

The film according to any one of film embodiments 1 to 9, wherein the film exhibits a change in light transmission of less than 10%, 9%, 8%, 7%, 6%, 5%, 4%, 3%, 2% or 1%, after the structured surface is exposed to the Dirt Pick-Up Test, the Falling Sand Test, or both tests.

Film Embodiment 11

The film according to any one of film embodiments 1 to 10, wherein the structured surface exhibits a storage modulus of at least about 20 MPa, and the remainder of each anti-reflective structure exhibits a lower storage modulus than that exhibited by the structured surface.

5 Film Embodiment 12

The film according to any one of film embodiments 1 to 11 further comprising inorganic nanoparticles (e.g., of silica, zirconia, titania, etc.) in the cross-linked silicone elastomeric material of at least the anti-reflective structures. Such particles may have a size in the range of up to and including about 2.0 microns. Silica particles can be up to the micron size, but it is preferable for particles made of other materials to be used in the nanometer sizes (i.e., in the range of from about 5 nm up to and including about 50 nm). Such particles, especially nanoparticles, may also be loaded into the silicone elastomeric material in the range of from about 0 wt.% up to and including about 60 wt.%.

10 Film Embodiment 13

The film according to any one of film embodiments 1 to 12, wherein the structured film substrate further comprises a base portion from which the anti-reflective structures extend. When the base portion and anti-reflective structures are both made of silicone elastomeric material, each of the anti-reflective structures can have about the same silicone elastomer cross-link density and the base portion can have a lower silicone elastomer cross-link density than that of each of the anti-reflective structures.

15 Film Embodiment 14

20 The film according to any one of film embodiments 1 to 13 in combination with a transparent support backing having a major face, wherein the structured film substrate further comprises a backing face (e.g., a major backing face) bonded to the major face of the support backing so as to form a reinforced anti-reflective structured film. The anti-reflective structures can form an exposed surface of the reinforced anti-reflective structured film.

25 Film Embodiment 15

The film according to film embodiment 14, wherein the transparent support backing dissipates static electricity.

Film Embodiment 16

30 The film according to any one of film embodiments 1 to 15 in combination with a barrier layer, wherein the structured film substrate further comprises a backing face (e.g., a major backing face), and the barrier layer is bonded to the backing face of the structured film substrate.

Film Embodiment 17

The film according to film embodiment 16, wherein the barrier layer is a moisture barrier.

Film Embodiment 18

The film according to any one of film embodiments 1 to 17, wherein each of the anti-reflective structures exhibits a silicone elastomer cross-link density gradient, with a silicone elastomer cross-link density that is higher closer to the glass-like surface and lower further away from the glass-like surface.

5 Film Embodiment 19

The film according to any one of film embodiments 1 to 18, wherein each of the anti-reflective structures has a core portion and a remainder portion in addition to its glass-like surface, with the cross-linked silicone elastomeric material forming the core portion having a cross-link density that is lower than the cross-linked silicone elastomeric material forming the remainder portion. The cross-linked silicone elastomeric material forming the core portion can have a substantially uniform cross-link density that is lower than the cross-linked silicone elastomeric material forming the remainder portion

10

Film Embodiment 20

The film according to any one of film embodiments 1 to 19, wherein the coating has a dry thickness in the range of from about 50 Å to about 10,000 Å. It can be desirable for this coating to have a thickness in the range of from about 50 Å to about 2500 Å.

15

Film Embodiment 21

The film according to any one of the film embodiments 1 to 20, wherein the silica nanoparticles have an average particle diameter of about 40 nanometers or less. The silica nanoparticles can also be spherical.

20

Film Embodiment 22

The film according to any one of film embodiments 1 to 21, wherein the coating comprises:

- a) from about 60 wt.% (i.e., percent by weight) to about 95 wt.% of the agglomerates of silica nanoparticles;
- b) from about 0.1 wt.% to about 20 wt.% of at least one or more tetraalkoxysilanes;
- c) optionally from about 0 wt.% to about 5 wt.% of a surfactant; and
- d) optionally from about 0 wt.% to about 5 wt.% of a wetting agent.

Film Embodiment 23

The film according to any one of film embodiments 1 to 20, wherein the coating comprises:

- a) from about 0.5wt.% to about 99 wt.% water;
- b) from about 0.1 wt.% to about 20 wt.% of the silica nanoparticles having an average particle diameter of 40 nm or less;
- c) optionally from about 0 wt.% to about 20 wt.% of the silica nanoparticles having an average particle diameter of greater than about 50 nm, wherein the sum of b) and c) is from about 0.1 wt.% to about 20 wt.%;
- d) a sufficient amount of an acid having a pKa of < 3.5 to reduce the pH to less than 5; and

35

5 e) optionally from about 0 wt% to about 20 wt. % of a tetraalkoxysilane, relative to the amount of the silica nanoparticles.

#### Light Energy Absorbing Device Embodiment 1

10 A light energy absorbing device comprising: a light absorber having a light energy receiving face; and a transparent anti-reflective structured film, according to any one of film embodiments 1 to 23, disposed so as to be between a source of light energy and the light energy receiving face, while light energy from the source is being absorbed by the light absorber. A light energy absorbing device can include but is not limited to, for example, a thermal energy absorbing device for absorbing thermal energy from a light source (e.g., the sun), a photovoltaic device that converts light into electrical energy, or any other light energy absorbing device.

#### Device Embodiment 2

15 The device according to device embodiment 1, wherein the light absorbing device is a photovoltaic device, the light absorber comprises a photovoltaic module having at least one photovoltaic cell, and the anti-reflective structured film reduces surface reflections so as to improve the electrical power output of the photovoltaic module (i.e., to improve the efficiency of the photovoltaic module in converting light energy into electrical energy) by at least about 3%. It is desirable for the electrical power output of the photovoltaic module to be improved by at least about 5% and preferably in the range of from about 5% to about 10%, or even more.

#### Device Embodiment 3

20 The device according to device embodiment 1 or 2, wherein the light absorber comprises a photovoltaic cell, and the light absorbing device is a photovoltaic device that is flexible and pliant enough to be wound into a roll or folded without being damaged.

#### Device Embodiment 4

25 The device according to device embodiment 1 or 2, wherein the light absorbing device includes a rigid photovoltaic module.

#### Device Embodiment 5

30 The device according to device embodiment 1, wherein the light absorbing device includes a solar thermal panel.

#### Device Embodiment 6

The device according to any one of the device embodiments 1, 2, 4 and 5, wherein the structured film substrate is laminated to, is a coating on, or is otherwise disposed over a glass substrate.

#### Embodiment 1 of a Method of Making a Transparent Anti-reflective Film

35 A method of making a transparent anti-reflective structured film, the method comprising: providing a transparent structured film substrate having a major structured face and a major backing face,

with the structured face comprising anti-reflective structures defining an anti-reflective structured surface that is anti-reflective to light, and at least the anti-reflective structures comprising a cross-linked silicone elastomeric material; and treating all, most or at least a substantial portion of the anti-reflective structured surface so as to transform cross-linked silicone elastomeric material at the anti-reflective structured surface into a glass-like material comprising an  $\text{SiO}_2$  stoichiometry, such that all, most, or at least a substantial portion of the anti-reflective structured surface comprises a glass-like surface having the  $\text{SiO}_2$  stoichiometry; and coating the glass-like surface with a coating of at least one layer of agglomerates of silica nanoparticles, with the agglomerates comprising a three-dimensional porous network of silica nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles.

10 Embodiment 2 of a Method of Making a Film

The method according to the film making method embodiment 1, wherein the treating comprises exposing the anti-reflective structured surface to at least one or a combination of vacuum-ultraviolet (VUV) light (e.g., by 172 nm Excimer treatment), vacuum-ultraviolet light and ozone (VUVO), oxygen plasma, and heat (e.g., induction heating, a flame, etc.).

15 The surface of the silicone material needs to be treated for a sufficient period of time (e.g., 5 to 300 seconds of 172 nm Excimer lamp exposure) and at a suitable energy level (e.g., 10-50 mW/cm<sup>2</sup> in a nitrogen inert atmosphere of less than 50 ppm oxygen) to produce the desired level of conversion from silicon to the  $\text{SiO}_2$  stoichiometry.

Embodiment 3 of a Method of Making a Film

20 The method according to the film making method embodiment 1 or 2 further comprising: exposing the anti-reflective structured surface to e-beam radiation so as to cause further cross-linking of the cross-linked silicone elastomeric material of at least the structured surface, the e-beam radiation exposure being performed before the treating.

Embodiment 4 of a Method of Making a Film

25 The method according to any one of the film making method embodiments 1 to 3, wherein the step of providing a transparent structured film substrate comprises:

providing silicone precursor material (e.g., a molten thermoplastic or cross-linkable thermoset silicone elastomer resin) that is curable so as to form the cross-linked silicone elastomeric material; forming the silicone precursor material into the shape of the structured film substrate; and 30 curing the silicone precursor material so as to form the transparent structured film substrate.

Embodiment 1 of Method of Making a Light Energy Absorbing Device

A method of making a light energy absorbing device such as, for example, a light source (e.g., solar) thermal energy absorbing device, a photovoltaic device or any other light energy absorbing device, the method comprising:

5 providing a transparent anti-reflective structured film according to any one of embodiments 1 to 23; providing a light absorber (e.g., a solar hot water heater or other thermal energy absorbing device, a photovoltaic module having at least one photovoltaic cell for converting solar or other light energy into electrical energy, etc.) having a light receiving face; and mechanically attaching, adhesively bonding or otherwise securing the anti-reflective structured film in relation to the light absorber so that light can pass 10 through the anti-reflective structured film to the light receiving face of the light absorber.

Embodiment 2 of a Method of Making a Light Energy Absorbing Device

A method of making a light energy absorbing device such as, for example, a light source (e.g., solar) thermal energy absorbing device, a photovoltaic device or any other light energy absorbing device, the method comprising:

15 making a transparent anti-reflective structured film according to the method of any one of the methods of making a film embodiments 1 to 4; providing a light absorber (e.g., a solar hot water heater or other thermal energy absorbing device, a photovoltaic module having at least one photovoltaic cell for converting solar or other light energy into electrical energy) having a light energy receiving face; and mechanically attaching, adhesively bonding or otherwise securing the anti-reflective structured film in 20 relation to the light absorber so that light can pass through the anti-reflective structured film to the light energy receiving face of the light absorber.

This invention may take on various modifications and alterations without departing from its spirit and scope. Accordingly, this invention is not limited to the above-described but is to be controlled by the limitations set forth in the following claims and any equivalents thereof.

25 This invention may be suitably practiced in the absence of any element not specifically disclosed herein.

All patents and patent applications cited above, including those in the Background section, are incorporated by reference into this document in total.

What is claimed is:

1. A transparent anti-reflective structured film comprising:

5 a structured film substrate having a structured face, with said structured face comprising anti-reflective structures defining a structured surface and being anti-reflective to light, at least a substantial portion of said structured surface comprising a glass-like surface, at least said anti-reflective structures comprising a cross-linked silicone elastomeric material, and said glass-like surface comprising an SiO<sub>2</sub> stoichiometry and being coated with a coating of at least one layer of agglomerates of silica nanoparticles, with said agglomerates comprising a three-dimensional porous network of silica nanoparticles, and the 10 silica nanoparticles being bonded to adjacent silica nanoparticles.

2. The film according to claim 1, wherein said glass-like surface comprises said SiO<sub>2</sub> stoichiometry to a depth of at least about 5 nanometers into each of said anti-reflective structures.

15 3. The film according to claim 1 or 2, wherein said glass-like surface comprises said SiO<sub>2</sub> stoichiometry to a depth in the range of from at least about 10 nanometers to about 100 nanometers into each of said anti-reflective structures.

20 4. The film according to any one of claims 1 to 3, wherein said glass-like surface comprises a minimum amount of at least about 10 molar % carbon atoms.

5. The film according to any one of claims 1 to 3, wherein said glass-like surface comprises in the range of from a minimum amount of at least about 10 molar % carbon atoms up to about 40 molar % carbon atoms.

25 6. The film according to any one of claims 1 to 5, wherein said anti-reflective structures comprise prisms having a prism tip angle in the range of from about 15 degrees to about 75 degrees and a pitch in the range of from about 10 microns to about 250 microns.

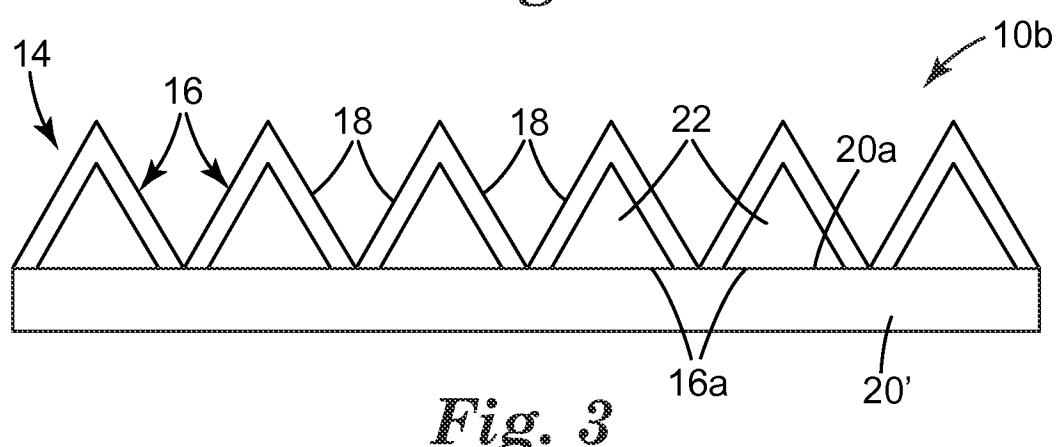
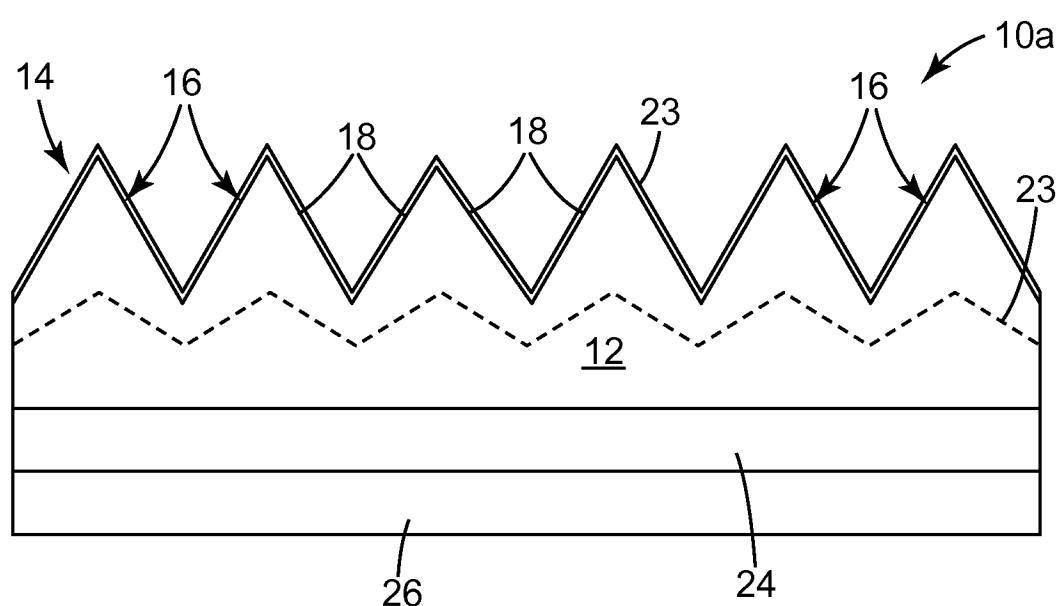
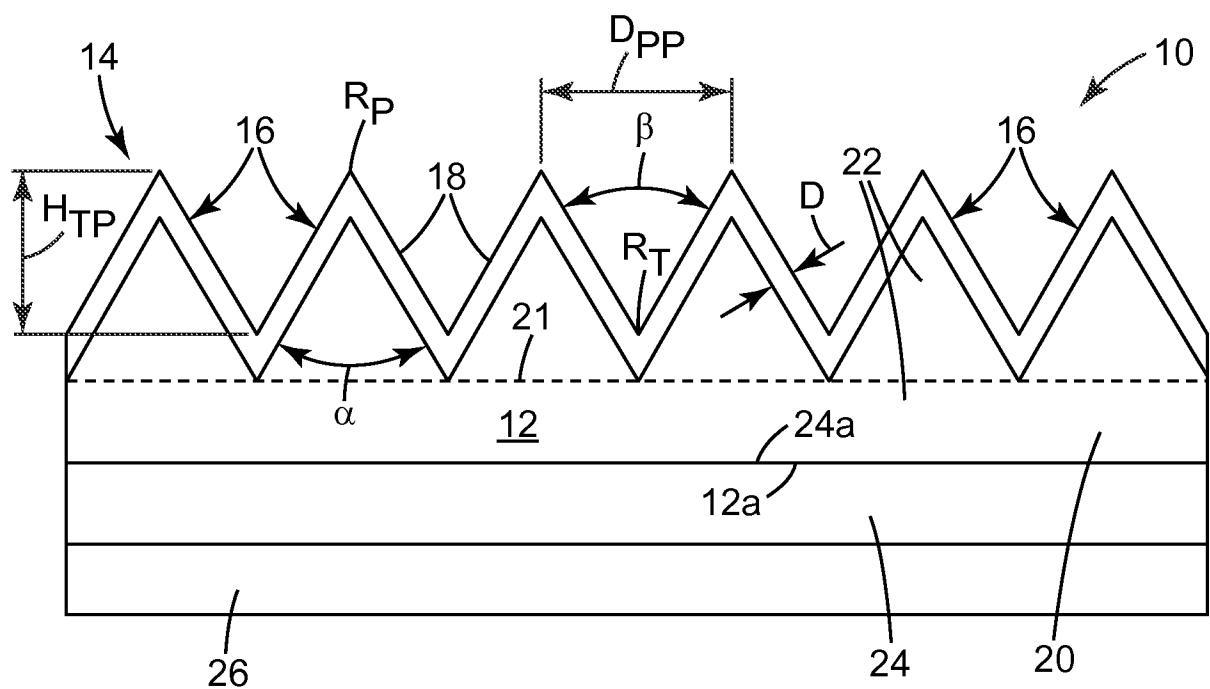
30 7. The film according to any one of claims 1 to 6, wherein said anti-reflective structures comprise prisms having a trough to peak height in the range of from about 10 microns to about 250 microns.

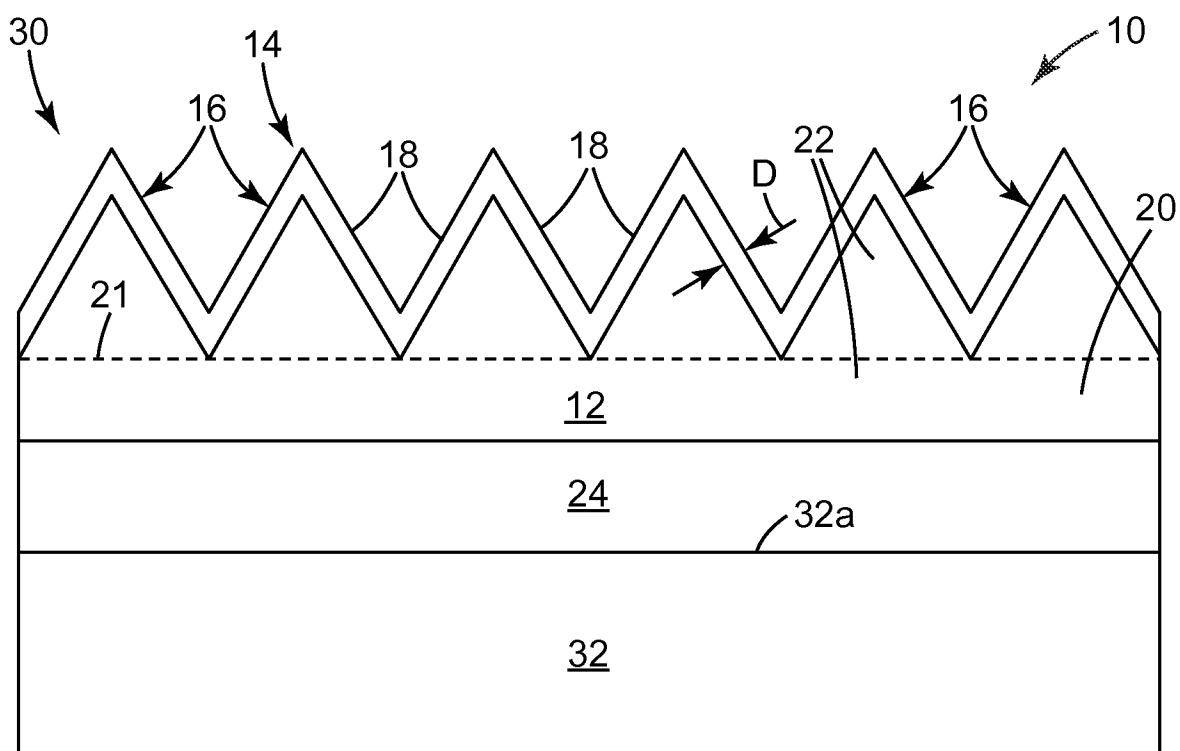
35 8. The film according to any one of claims 1 to 7, wherein said film exhibits a change in light transmission of less than 8%, after said structured surface is exposed to at least one of the Dirt Pick-Up Test and the Falling Sand Test.

9. A light energy absorbing device comprising:  
a light absorber having a light energy receiving face; and  
a transparent anti-reflective structured film, according to any one of claims 1 to 19, disposed so as  
to be between a source of light energy and said light energy receiving face, while light energy from the  
5 source is being absorbed by said light absorber.

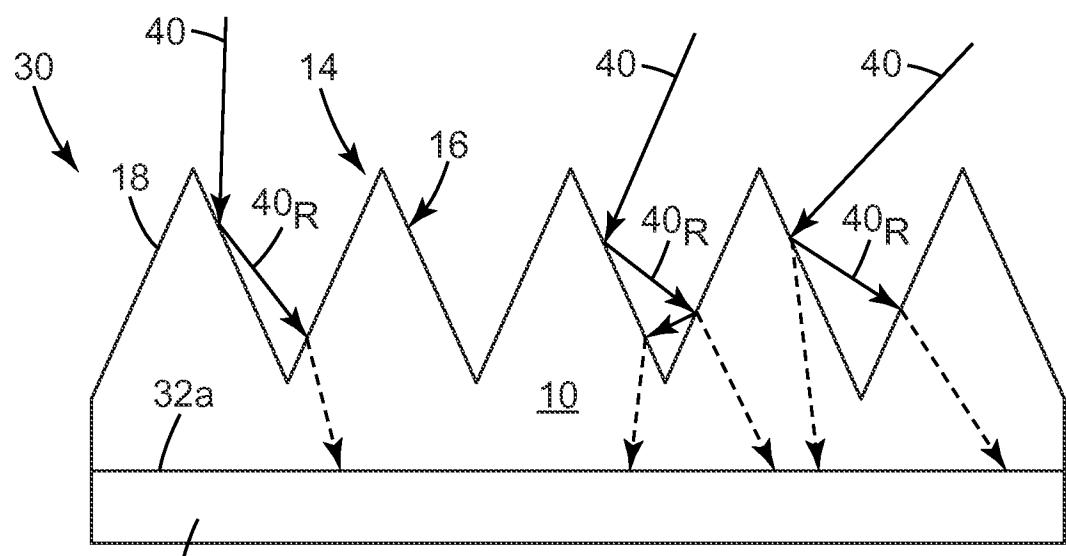
10. A method of making a transparent anti-reflective structured film, said method comprising:  
providing a structured film substrate having a structured face comprising anti-reflective structures  
defining an anti-reflective structured surface that is anti-reflective to light, with at least the anti-reflective  
10 structures comprising a cross-linked silicone elastomeric material; and  
treating the anti-reflective structured surface so as to transform cross-linked silicone elastomeric  
material defining at least a substantial portion of the anti-reflective structured surface into a glass-like  
material comprising an  $\text{SiO}_2$  stoichiometry, such that at least a substantial portion of the anti-reflective  
structured surface comprises a glass-like surface having the  $\text{SiO}_2$  stoichiometry; and  
15 coating the glass-like surface with a coating of at least one layer of agglomerates of silica  
nanoparticles, with said agglomerates comprising a three-dimensional porous network of silica  
nanoparticles, and the silica nanoparticles being bonded to adjacent silica nanoparticles.

1/2





*Fig. 4*



*Fig. 5*

# INTERNATIONAL SEARCH REPORT

International application No

PCT/US2011/064613

**A. CLASSIFICATION OF SUBJECT MATTER**  
 INV. G02B1/11  
 ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)  
 G02B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, INSPEC, WPI Data

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 2009/145049 A1 (INCTEC INC [JP]; SATO KAZUYA [JP]; MATSUMOTO TSUKASA [JP]; WATANABE YU) 3 December 2009 (2009-12-03) abstract; figures - & US 2011/157704 A1 (SATO KAZUYA [JP] ET AL) 30 June 2011 (2011-06-30) abstract; figures ----- MURAHARA M ET AL: "Water-resistant coating on optical material by photo-oxidation of silicone oil", PROCEEDINGS OF SPIE, SPIE, US, vol. 5991, 1 January 2005 (2005-01-01), pages 599118/1-599118/9, XP009108857, ISSN: 0277-786X, DOI: 10.1117/12.638773 the whole document ----- -/--	1-10
A		1-10

Further documents are listed in the continuation of Box C.

See patent family annex.

\* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

Date of mailing of the international search report

20 March 2012

10/04/2012

Name and mailing address of the ISA/  
 European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040,  
 Fax: (+31-70) 340-3016

Authorized officer

Ward, Seamus

## INTERNATIONAL SEARCH REPORT

International application No

PCT/US2011/064613

## C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	MATIENZO L J ET AL: "Transformation of poly(dimethylsiloxane) into thin surface films of SiO <sub>x</sub> by UV/ozone treatment. Part II: segregation and modification of doped polymer blends", JOURNAL OF MATERIALS SCIENCE, KLUWER ACADEMIC PUBLISHERS, BO, vol. 41, no. 19, 2 September 2006 (2006-09-02), pages 6374-6384, XP019435433, ISSN: 1573-4803, DOI: 10.1007/S10853-006-0713-4 the whole document -----	1-10

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Information on patent family members

International application No

PCT/US2011/064613

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