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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>6</sup> :</b> <b>C07F 9/6574 // C08K 5/527, 5/5333</b>	<b>A1</b>	<b>(11) International Publication Number:</b> <b>WO 96/35694</b> <b>(43) International Publication Date:</b> 14 November 1996 (14.11.96)
<b>(21) International Application Number:</b> PCT/EP96/01914 <b>(22) International Filing Date:</b> 3 May 1996 (03.05.96)  <b>(30) Priority Data:</b> 08/438,152                      9 May 1995 (09.05.95)                      US  <b>(71) Applicant:</b> AKZO NOBEL N.V. [NL/NL]; Velperweg 76, NL-6824 BM Arnhem (NL). <b>(72) Inventor:</b> TELSCHOW, Jeffrey, Earl; 416 Benedict Avenue, Tarrytown, NY 10591 (US). <b>(74) Agent:</b> SCHALKWIJK, Pieter, Cornelis; Akzo Nobel N.V., Patent Dept. (Dept. APTA), P.O. Box 9300, NL-6800 SB Arnhem (NL).		<b>(81) Designated States:</b> CA, CN, JP, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).  <b>Published</b> <i>With international search report.</i>
<b>(54) Title:</b> (PENTAERYTHRITOL PHOSPHATE ALCOHOL) (CYCLIC NEOPENTYLENE GLYCOL) PHOSPHITE AND PHOSPHONATE  <b>(57) Abstract</b>  (Pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphite and phosphonate are disclosed as flame retardant compounds. The first named compound is formed by the reaction of pentaerythritol phosphate alcohol with a neopentylene glycol halophosphite, such as neopentylene glycol chlorophosphite. The (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphonate is formed by heating (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphite.		

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(PENTAERYTHRITOL PHOSPHATE ALCOHOL) (CYCLIC NEOPENTYLENE GLYCOL)  
PHOSPHITE AND PHOSPHONATE

Background of the Invention

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Various derivatives of pentaerythritol phosphate are known as flame retardant additives for polymers such as polypropylene. A recent example is provided by U.S. Patent No. 4,801,625 to W.J. Parr et al. which describes ether, ester and carbonate derivatives of pentaerythritol phosphate. The carbonate version of such compounds can  
10 be advantageously prepared by the reaction of pentaerythritol phosphate alcohol with a dihydrocarbyl carbonate as described in U.S. Patent No. 5,235,085.

Summary of the Invention

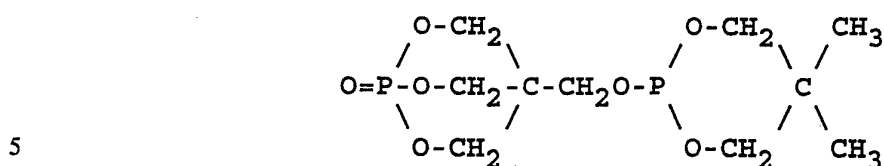
15 The invention relates to (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol)phosphite and (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphonate which are useful as flame retardant compounds. The first named compound is formed by the reaction of pentaerythritol phosphate alcohol ("PEPA") with a neopentylene glycol halophosphite, such as neopentylene glycol chlorophosphite. The  
20 (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphonate is formed by heating (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphite.

U.S. Patent No. 5,362,898 discloses certain bis(pentaerythritol phosphate alcohol) alkylphosphonate compounds, such as those containing from about one to about four carbon atoms in their alkyl moiety, with bis(pentaerythritol phosphate alcohol)  
25 methylphosphonate being a preferred species thereof.

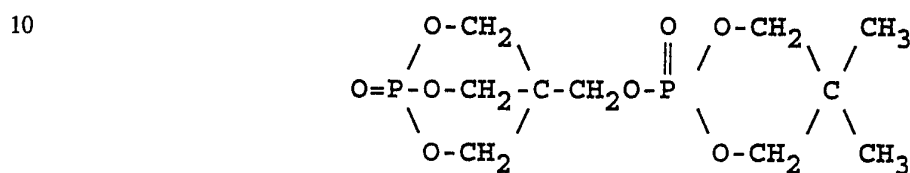
Description of the Preferred Embodiments

The present novel (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol)  
30 phosphite compound is of the formula:

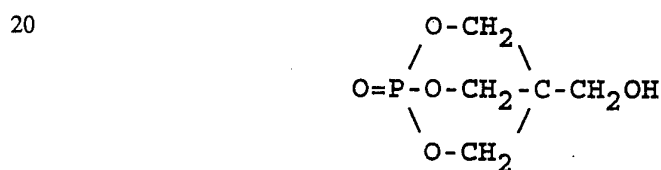
- 2 -



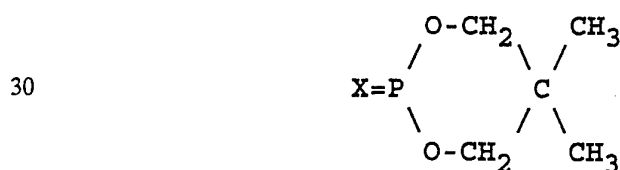
and the novel (pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphonate compound is of the formula:



One process for forming the above-described novel phosphite compound is by the reaction of a neopentylene glycol halophosphite with pentaerythritol phosphate alcohol which has the formula:



The neopentylene glycol halophosphite reagent is of the formula:



where X is a halogen atom, such as chlorine. The reaction can be conducted at room temperature in an appropriate organic solvent, such as acetonitrile using an acid acceptor, such as a trialkylamine.

Once the phosphite compound has been synthesized, it can be converted to the novel phosphonate compound of the present invention by heating, for example, in a high boiling organic solvent, such as an aryl phosphate solvent.

40 The present invention is further understood by reference to the Examples which follow.

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EXAMPLE 1

In a 250 mL, mechanically stirred, four-necked flask fitted with pot thermometer, condenser and dropping funnel were placed 18.0 grams (0.10 mole) of pentaerythritol phosphate alcohol (PEPA), 13.9 mL (10.1 grams, 0.10 mole) of triethylamine, and 75  
5 ml of dry acetonitrile. Then, 16.9 grams (0.10 mole) of neopentylene chlorophosphite (2-chloro-5,5-dimethyl-1,3,2-dioxaphosphorinane) was added dropwise from the funnel over twenty minutes, while maintaining the temperature of the stirred reaction mixture at 20-30°C under nitrogen. The mixture was allowed to stir for twenty-two hours at  
10 ambient temperature, the white solid which had formed was filtered and was dissolved in 300 mL of methylene chloride and was then extracted three times with 100 mL of water. On evaporation of the methylene chloride, 4.1 grams of a white powder with a melting point of 200-203°C remained. The solid exhibited two equally intense singlets by <sup>31</sup>P NMR (d<sub>6</sub>-DMSO) at +122.7 and -6.2 ppm, consistent with the desired PEPA-NPG  
15 phosphite product (5,5-dimethyl-2-(2-oxo-2,6,7-trioxa-1-phosphabicyclo[2.2.2]oct-4-yl-methoxy)-1,3,2-dioxaphosphorinane).

EXAMPLE 2

20 In a 100 mL, mechanically stirred, four-necked flask fitted with pot thermometer, were placed 11.4 grams (0.0367 mole) of the PEPA-NPG phosphite from Example 1, 20 mg of iodine, and 30 mL of Phosflex® 41P under nitrogen. The solution was heated for sixteen hours at 210°C to promote Arbuzov rearrangement. Analysis using <sup>31</sup>P NMR (d<sub>6</sub>-DMSO) revealed two resonances at +19.3 and -7.1 ppm, corresponding to the  
25 desired PEPA-NPG phosphonate product (5,5-dimethyl-2-(2-oxo-2,6,7-trioxa-1-phosphabicyclo[2.2.2]oct-4-ylmethyl)-1,3,2-dioxaphosphorinane, 2-oxide).

The foregoing Examples are intended to illustrate certain embodiments of the present invention and, for that reason, should not be construed in a limiting sense. The scope of  
30 protection sought is set forth in the claims which follow.

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Claims

1. (Pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphite.
2. (Pentaerythritol phosphate alcohol) (cyclic neopentylene glycol) phosphonate.

## INTERNATIONAL SEARCH REPORT

International Application No

PC./EP 96/01914

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C07F9/6574 //C08K5/527,C08K5/5333

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US,A,3 883 478 (GRESHAM JOHN T) 13 May 1975 cited in the application see the whole document ---	1,2
Y	US,A,5 362 898 (TELSCHOW JEFFREY E) 8 November 1994 cited in the application see the whole document ---	1,2
Y	FR,A,2 342 984 (BORG WARNER) 30 September 1977 see the whole document ---	1,2
Y	WO,A,92 16537 (ETHYL CORP) 1 October 1992 see the whole document ---	1,2
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☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

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P,Y	US,A,5 420 326 (TELSCHOW JEFFREY E) 30 May 1995 see the whole document -----	1,2



## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

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