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(54) Title: **PROCESS FOR MAKING A HIGH DENSITY DETERGENT COMPOSITION IN A SINGLE MIXER/DENSIFIER WITH
SELECTED RECYCLE STREAMS**

(57) Abstract

A process for continuously preparing high density detergent composition is provided. The process comprises the steps of: (a) continuously charging a detergent surfactant paste and dry starting detergent material into a mixer/densifier for densification and build-up to obtain agglomerates; (b) feeding the agglomerates into a conditioning apparatus for improving the flow properties of the agglomerates and for separating the agglomerates into a first agglomerate mixture and a second agglomerate mixture; (d) recycling the first agglomerate mixture into the mixer/densifier for further agglomeration; (e) admixing adjunct detergent ingredients to the second agglomerate mixture so as to form the high density detergent composition.

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PROCESS FOR MAKING A HIGH DENSITY DETERGENT COMPOSITION IN A SINGLE
MIXER/DENSIFIER WITH SELECTED RECYCLE STREAMS

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FIELD OF THE INVENTION

10 The present invention generally relates to a process for producing a high density laundry detergent composition containing agglomerates. More particularly, the invention is directed to a continuous process during which a high density detergent composition is produced by feeding a surfactant paste and dry starting detergent material into a single mixer/densifier and then into conditioning and screening apparatus. The process includes optimally selected recycle stream
15 configurations so as to produce a high density detergent composition containing agglomerates with improved flow and particle size properties. Such improved properties enhance consumer acceptance of the detergent composition produced by the instant process.

BACKGROUND OF THE INVENTION

 Recently, there has been considerable interest within the detergent industry for laundry
20 detergents which are "compact" and therefore, have low dosage volumes. To facilitate production of these so-called low dosage detergents, many attempts have been made to produce high bulk density detergents, for example, with a density of 600 g/l or higher. The low dosage detergents are currently in high demand as they conserve resources and can be sold in small packages which are more convenient for consumers.

25 Generally, there are two primary types of processes by which detergent particles or powders can be prepared. The first type of process involves spray-drying an aqueous detergent slurry in a spray-drying tower to produce highly porous detergent particles. In the second type of process, the various detergent components are dry mixed after which they are agglomerated with a binder such as a nonionic or anionic surfactant. In both processes, the most important factors which govern the
30 density of the resulting detergent material are the density, porosity, particle size and surface area of the various starting materials and their respective chemical composition. These parameters, however, can only be varied within a limited range. Thus, a substantial bulk density increase can only be achieved by additional processing steps which lead to densification of the detergent material.

35 There have been many attempts in the art for providing processes which increase the density of detergent particles or powders. Particular attention has been given to densification of spray-dried particles by "post-tower" treatment. For example, one attempt involves a batch process

in which spray-dried or granulated detergent powders containing sodium tripolyphosphate and sodium sulfate are densified and spheronized in a Marumerizer®. This apparatus comprises a substantially horizontal, roughened, rotatable table positioned within and at the base of a substantially vertical, smooth walled cylinder. This process, however, is essentially a batch process and is therefore less suitable for the large scale production of detergent powders. More recently, other attempts have been made to provide a continuous processes for increasing the density of "post-tower" or spray dried detergent particles. Typically, such processes require a first apparatus which pulverizes or grinds the particles and a second apparatus which increases the density of the pulverized particles by agglomeration. These processes achieve the desired increase in density only by treating or densifying "post tower" or spray dried particles.

However, all of the aforementioned processes are directed primarily for densifying or otherwise processing spray dried granules. Currently, the relative amounts and types of materials subjected to spray drying processes in the production of detergent granules has been limited. For example, it has been difficult to attain high levels of surfactant in the resulting detergent composition, a feature which facilitates production of low dosage detergents. Thus, it would be desirable to have a process by which detergent compositions can be produced without having the limitations imposed by conventional spray drying techniques.

To that end, the art is also replete with disclosures of processes which entail agglomerating detergent compositions. For example, attempts have been made to agglomerate detergent builders by mixing zeolite and/or layered silicates in a mixer to form free flowing agglomerates. While such attempts suggest that their process can be used to produce detergent agglomerates, they do not provide a mechanism by which starting detergent materials in the form of pastes, liquids and dry materials can be effectively agglomerated into crisp, free flowing detergent agglomerates having a high density of at least 650 g/l. Moreover, such agglomeration processes have produced detergent agglomerates containing a wide range of particle sizes, for example "overs" and "fines" are typically produced. The "overs" or larger than desired agglomerate particles have a tendency to decrease the overall solubility of the detergent composition in the washing solution which leads to poor cleaning and the presence of insoluble "clumps" ultimately resulting in consumer dissatisfaction. The "fines" or smaller than desired agglomerate particles have a tendency to "gel" in the washing solution and also give the detergent product an undesirable sense of "dustiness." Further, past attempts to recycle such "overs" and "fines" has resulted in the exponential growth of additional undesirable over-sized and under-sized agglomerates since the "overs" typically provide a nucleation site or seed for the agglomeration of even larger particles, while recycling "fines" inhibits agglomeration leading to the production of more "fines" in the process.

Accordingly, there remains a need in the art for a process which produces a high density detergent composition containing agglomerates having improved flow and particle size properties.

Also, there remains a need for such a process which is more efficient and economical to facilitate large-scale production of low dosage or compact detergents.

BACKGROUND ART

The following references are directed to densifying spray-dried granules: Appel et al, U.S. Patent No. 5,133,924 (Lever); Bortolotti et al, U.S. Patent No. 5,160,657 (Lever); Johnson et al, British patent No. 1,517,713 (Unilever); and Curtis, European Patent Application 451,894. The following references are directed to producing detergents by agglomeration: Beerse et al, U.S. Patent No. 5,108,646 (Procter & Gamble); Hollingsworth et al, European Patent Application 351,937 (Unilever); and Swatling et al, U.S. Patent No. 5,205,958.

SUMMARY OF THE INVENTION

The present invention meets the aforementioned needs in the art by providing a process which continuously produces a high density detergent composition directly from starting detergent ingredients. Consequently, the process achieves the desired high density detergent composition without unnecessary process parameters, such as the use of spray drying techniques and relatively high operating temperatures, all of which increase manufacturing costs. The process invention described herein also provides a detergent composition containing agglomerates having improved flow and particle size (i.e. more uniform) properties which ultimately results in a low dosage or compact detergent product having more acceptance by consumers. As used herein, the term "agglomerates" refers to particles formed by agglomerating starting detergent ingredients (liquid and/or particles) which typically have a smaller median particle size than the formed agglomerates. All percentages and ratios used herein are expressed as percentages by weight (anhydrous basis) unless otherwise indicated. All documents are incorporated herein by reference. All viscosities referenced herein are measured at 70°C ($\pm 5^\circ\text{C}$) and at shear rates of about 10 to 100 sec^{-1} .

In accordance with one aspect of the invention, a process for continuously preparing high density detergent composition is provided. The process comprises the steps of: (a) continuously charging a detergent surfactant paste and dry starting detergent material into a mixer/densifier for densification and build-up such that the finished agglomerates have a median particle size from about 300 microns to about 900 microns; (b) feeding the agglomerates into a conditioning apparatus for improving the flow properties of the agglomerates and for separating the agglomerates into a first agglomerate mixture and a second agglomerate mixture, wherein the first agglomerate mixture substantially has a particle size of less than about 150 microns and the second agglomerate mixture substantially has a particle size of at least about 150 microns; (d) recycling the first agglomerate mixture into the mixer/densifier for further agglomeration; (e) admixing adjunct detergent ingredients to the second agglomerate mixture so as to form the high density detergent composition.

In accordance with another aspect of the invention, another process for continuously preparing high density detergent composition is provided. This process comprises the steps of: (a) continuously charging a detergent surfactant paste and dry starting detergent material into a mixer/densifier for densification and build-up such that the agglomerates have a median particle size of from about 300 microns to about 900 microns; (b) screening the agglomerates so as to form a

first agglomerate mixture substantially having a particle size of less than about 6 mm and a second agglomerate mixture substantially having a particle size of less than about 6 mm; (c) feeding the first agglomerate mixture to a grinding apparatus and the second agglomerate mixture to a conditioning apparatus for improving the flow properties of the second agglomerate mixture and for separating the second agglomerate mixture into a third agglomerate mixture and a fourth agglomerate mixture, wherein the third agglomerate mixture substantially has a particle size of less than about 150 microns and the fourth agglomerate mixture substantially has a particle size of at least about 150 microns; (d) recycling the third agglomerate mixture into the high speed mixer/densifier for further agglomeration; (e) separating the fourth agglomerate mixture into a fifth agglomerate mixture and a sixth agglomerate mixture, wherein the fifth agglomerate mixture substantially has a particle size of at least about 900 microns and the sixth agglomerate mixture has a median particle size of from about 50 microns to about 1400 microns; (f) inputting the fifth agglomerate mixture into the grinding apparatus for grinding with the first agglomerate mixture to form a ground agglomerate mixture which is recycled into the conditioning apparatus; and (h) admixing adjunct detergent ingredients to the sixth agglomerate mixture so as to form the high density detergent composition. Another aspect of the invention is directed to a high density detergent composition made according to any one of the embodiments of the instant process.

Accordingly, it is an object of the invention to provide a process which produces a high density detergent composition containing agglomerates having improved flow and particle size properties. It is also an object of the invention to provide such a process which is more efficient and economical to facilitate large-scale production of low dosage or compact detergents. These and other objects, features and attendant advantages of the present invention will become apparent to those skilled in the art from a reading of the following detailed description of the preferred embodiment and the appended claims.

BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a flow diagram of a process in accordance with one embodiment of the invention in which undersized detergent agglomerates are recycled back into the mixer/densifier from the conditioning apparatus; and

Fig. 2 is a flow diagram of a process in accordance with another embodiment of the invention similar to Fig. 1 in which an additional recycling operation is included for purposes of further improving the properties of the resulting detergent product.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

Reference can be made to Figs. 1 and 2 for purposes of illustrating several embodiments of the process invention described herein. Fig. 1 illustrates a process 10 while Fig. 2 depicts a process 10' which is a modified version of process 10.

Process

Initially, the process 10 shown in Fig. 1 entails continuously charging a detergent surfactant paste 12 and dry starting detergent material 14 into a mixer/densifier 16 to obtain agglomerates 18. It should be understood that the surfactant paste 12 and dry starting detergent material 14 are densified and built-up in the mixer/densifier 16 so as to obtain the agglomerates 18.

5 The various ingredients which may be selected for the surfactant paste 12 and the dry starting detergent material 14 are described more fully hereinafter. However, it is preferable for the ratio of the surfactant paste to the dry detergent material to be from about 1:10 to about 10:1 and more preferably from about 1:4 to about 4:1. Preferably, the agglomerates 18 have a median particle size range of from about 300 microns to about 900 microns.

10 Typical apparatus used in process 10 for the mixer/densifier 16 include but is not limited to a Lödige Recycler CB-30, a Lödige Recycler KM-600 "Ploughshare," conventional twin-screw mixers, mixers commercially sold as Eirich, Schugi, O'Brien, and Drais mixers, and combinations of these and other mixers. The operating parameters will depend upon the particular mixer selected for operation as mixer/densifier 16. For example, high speed mixers and moderate speed mixers
15 will each require its own set of operating temperatures, residence times, rates of throughput, etc. However, the preferred mean residence time in the high speed mixer/densifier, e.g. Lödige Recycler CB-30, is from about 2 seconds to about 45 seconds, preferably from about 5 to 30 seconds, while the mean residence time in the moderate speed mixer/densifier, e.g. Lödige Recycler KM-600 "Ploughshare," is from about 0.5 minutes to about 15 minutes, preferably from about 1 to 10
20 minutes.

The mixer/densifier 16 preferably imparts a requisite amount of energy to form the agglomerates 18. More particularly, the moderate speed mixer/densifier 20 imparts from about 5×10^{10} erg/kg to about 2×10^{12} erg/kg at a rate of from about 3×10^8 erg/kg-sec to about 3×10^9 erg/kg-sec to form agglomerates 18. The energy input and rate of input can be determined by
25 calculations from power readings to the mixer/densifier 16 with and without agglomerates, residence time of the agglomerates, and the mass of the agglomerates in the mixer/densifier 16. Such calculations are clearly within the scope of the skilled artisan.

Optionally, a coating agent can be added just before, in or after the mixer/densifier 16 to control or inhibit the degree of agglomeration. This optional step provides a means by which the
30 desired agglomerate particle size can be achieved. Preferably, the coating agent is selected from the group consisting of aluminosilicates, carbonates, silicates and mixtures thereof. Another optional step entails spraying a binder material into the mixer/densifier 16 so as to facilitate build-up agglomeration. Preferably, the binder is selected from the group consisting of water, anionic surfactants, nonionic surfactants, polyethylene glycol, polyvinyl pyrrolidone, polyacrylates, citric
35 acid and mixtures thereof.

Another step in the process 10 entails feeding the agglomerates 18 into a conditioning apparatus 20 which preferably includes one or more of a drying apparatus and a cooling apparatus

(not shown individually). The conditioning apparatus 20 in whatever form (fluid bed dryer, fluid bed cooler, airlift, etc.) is included for improving the flow properties of the agglomerates 18 and for separating them into a first agglomerate mixture 22 and a second agglomerate mixture 24.

Preferably, the agglomerate mixture 22 substantially has a particle size of less than about 150
5 microns and the agglomerate mixture 24 substantially has a particle size of at least about 150 microns. It should be understood by those skilled in the art that such separation process are not always perfect and agglomerate mixture 22 and/or 24 may contain agglomerate particles outside the recited range. The ultimate goal of process 10, however, is to substantially divide a major portion of the "fines" or undersized agglomerates 22 from the more desired sized agglomerates 24 which are
10 then sent to one or more finishing steps 26.

The agglomerate mixture 22 is recycled back into the mixer/densifier 16 for further agglomeration such that the agglomerates in mixture 22 are ultimately built-up to the desired particle size. Preferably, the finishing steps 26 will include admixing adjunct detergent ingredients to agglomerate mixture 24 so as to form a fully formulated high density detergent composition 28
15 which is ready for commercialization. In a preferred embodiment, the detergent composition 28 has a density of at least 650 g/l. Optionally, the finishing steps 26 includes admixing conventional spray-dried detergent particles to the agglomerate mixture 24 along with adjunct detergent ingredients to form detergent composition 28. In this case, detergent composition 28 preferably comprises from about 10% to about 40% by weight of the agglomerate mixture 24 and the balance
20 spray-dried detergent particles and adjunct ingredients.

Reference is now made to Fig. 2 which depicts process 10' for making a high density detergent composition in accordance with the invention. Similar to process 10, the process 10' comprises the steps of continuously charging a detergent surfactant paste 30 and dry starting detergent material 32 into a mixer/densifier 34 to obtain agglomerates 36 which preferably have a
25 median particle size from about 300 microns to about 900 microns. Thereafter, the agglomerates 36 are screened in screening apparatus 38 so as to form a first agglomerate mixture 40 substantially having a particle size of at least about 6 mm and a second agglomerate mixture 42 substantially having a particle size of less than about 6 mm. The agglomerate mixture 40 contains relatively wet oversized agglomerates and usually represents about 2 to 5% of the agglomerates 36 prior to
30 screening.

The agglomerate mixture 40 is fed to a grinding apparatus 44 while the agglomerate mixture 42 is fed to a conditioning apparatus 46 for improving the flow properties of the agglomerate mixture 42 and for separating it into a third agglomerate mixture 48 and a fourth agglomerate mixture 50. Preferably, the agglomerate mixture 48 substantially has a particle size of
35 less than about 150 microns and the agglomerate mixture 50 substantially has a particle size of at least 150 microns. The process 10' entails recycling the agglomerate mixture 48 back into the mixer/densifier 34 for further build-up agglomeration as described with respect to process 10 in Fig.

1. Thereafter, the agglomerate mixture 50 is separated via any known process/apparatus such as with conventional screening apparatus 52 or the like into a fifth agglomerate mixture 54 and a sixth agglomerate mixture 56. Preferably, the agglomerate mixture 54 has a particle size of at least 900 microns and the agglomerate mixture 56 has a median particle size of from about 50 microns to about 1400 microns.

The agglomerate mixture 54 which contains additional oversized particles is inputted into the grinding apparatus 44 for grinding with the agglomerate mixture 40 which also contains oversized agglomerate particles to form a ground agglomerate mixture 58. Continuous with the foregoing operations, the agglomerate mixture 58 is recycled back into the conditioning apparatus 46 which may include one or more fluid bed dryers and coolers as described previously. In such cases, the recycle stream of agglomerate mixture 58 can be sent to any one or a combination of such fluid bed dryers and coolers without departing from the scope of the invention. The agglomerate mixture 56 is then subjected to one or more finishing steps 60 as described previously. Preferably, the process 10' includes the step of admixing adjunct detergent ingredients to the agglomerate mixture 56 so as to form the high density detergent composition 62 which has a density of at least 650 g/l.

The optional steps discussed with respect to the process 10 are equally applicable with respect to process 10'. By way of example, a coating agent can be added just before, in or after the mixer/densifier 34 to control or inhibit the degree of agglomeration. It has been found that adding a coating agent to the agglomerate mixture 50 or 56, i.e. before or after between the screening apparatus 52, yields a detergent composition with surprisingly improved flow properties. As mentioned previously, the coating agent is preferably selected from the group consisting of aluminosilicates, carbonates, silicates and mixtures thereof. The other optional steps such as spraying a binder material into the mixer/densifier 34 are useful in process 10' for purposes of facilitating build-up agglomeration. The residence times, energy input parameters, surfactant paste characteristics and ratios with starting dry detergent ingredients are all also preferably incorporated into the process 10'.

Detergent Surfactant Paste

The detergent surfactant paste used in the processes 10 and 10' is preferably in the form of an aqueous viscous paste, although forms are also contemplated by the invention. This so-called viscous surfactant paste has a viscosity of from about 5,000 cps to about 100,000 cps, more preferably from about 10,000 cps to about 80,000 cps, and contains at least about 10% water, more preferably at least about 20% water. The viscosity is measured at 70°C and at shear rates of about 10 to 100 sec.⁻¹. Furthermore, the surfactant paste, if used, preferably comprises a detergent surfactant in the amounts specified previously and the balance water and other conventional detergent ingredients.

The surfactant itself, in the viscous surfactant paste, is preferably selected from anionic, nonionic, zwitterionic, ampholytic and cationic classes and compatible mixtures thereof. Detergent surfactants useful herein are described in U.S. Patent 3,664,961, Norris, issued May 23, 1972, and in U.S. Patent 3,919,678, Laughlin et al., issued December 30, 1975. Useful cationic surfactants
 5 also include those described in U.S. Patent 4,222,905, Cockrell, issued September 16, 1980, and in U.S. Patent 4,239,659, Murphy, issued December 16, 1980, both of which are also incorporated herein by reference. Of the surfactants, anionics and nonionics are preferred and anionics are most preferred.

Nonlimiting examples of the preferred anionic surfactants useful in the surfactant paste
 10 include the conventional C_{11} - C_{18} alkyl benzene sulfonates ("LAS"), primary, branched-chain and random C_{10} - C_{20} alkyl sulfates ("AS"), the C_{10} - C_{18} secondary (2,3) alkyl sulfates of the formula $CH_3(CH_2)_x(CHOSO_3^- M^+) CH_3$ and $CH_3(CH_2)_y(CHOSO_3^- M^+) CH_2CH_3$ where x and (y + 1) are integers of at least about 7, preferably at least about 9, and M is a water-solubilizing cation, especially sodium, unsaturated sulfates such as oleyl sulfate, and the C_{10} - C_{18} alkyl alkoxy sulfates
 15 ("AE_xS"; especially EO 1-7 ethoxy sulfates).

Optionally, other exemplary surfactants useful in the paste of the invention include and C_{10} - C_{18} alkyl alkoxy carboxylates (especially the EO 1-5 ethoxycarboxylates), the C_{10} - C_{18} glycerol ethers, the C_{10} - C_{18} alkyl polyglycosides and their corresponding sulfated polyglycosides, and C_{12} - C_{18} alpha-sulfonated fatty acid esters. If desired, the conventional nonionic and amphoteric
 20 surfactants such as the C_{12} - C_{18} alkyl ethoxylates ("AE") including the so-called narrow peaked alkyl ethoxylates and C_6 - C_{12} alkyl phenol alkoxyates (especially ethoxylates and mixed ethoxy/propoxy), C_{12} - C_{18} betaines and sulfobetaines ("sultaines"), C_{10} - C_{18} amine oxides, and the like, can also be included in the overall compositions. The C_{10} - C_{18} N-alkyl polyhydroxy fatty acid amides can also be used. Typical examples include the C_{12} - C_{18} N-methylglucamides. See WO
 25 92/06154. Other sugar-derived surfactants include the N-alkoxy polyhydroxy fatty acid amides, such as C_{10} - C_{18} N-(3-methoxypropyl) glucamide. The N-propyl through N-hexyl C_{12} - C_{18} glucamides can be used for low sudsing. C_{10} - C_{20} conventional soaps may also be used. If high sudsing is desired, the branched-chain C_{10} - C_{16} soaps may be used. Mixtures of anionic and nonionic surfactants are especially useful. Other conventional useful surfactants are listed in
 30 standard texts.

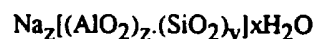
Dry Detergent Material

The starting dry detergent material of the processes 10 and 10' preferably comprises a detergency builder selected from the group consisting of aluminosilicates, crystalline layered silicates and mixtures thereof, and carbonate, preferably sodium carbonate. The aluminosilicates or
 35 aluminosilicate ion exchange materials used herein as a detergent builder preferably have both a high calcium ion exchange capacity and a high exchange rate. Without intending to be limited by theory, it is believed that such high calcium ion exchange rate and capacity are a function of several interrelated

factors which derive from the method by which the aluminosilicate ion exchange material is produced. In that regard, the aluminosilicate ion exchange materials used herein are preferably produced in accordance with Corkill et al, U.S. Patent No. 4,605,509 (Procter & Gamble), the disclosure of which is incorporated herein by reference.

- 5 Preferably, the aluminosilicate ion exchange material is in "sodium" form since the potassium and hydrogen forms of the instant aluminosilicate do not exhibit the as high of an exchange rate and capacity as provided by the sodium form. Additionally, the aluminosilicate ion exchange material preferably is in over dried form so as to facilitate production of crisp detergent agglomerates as described herein. The aluminosilicate ion exchange materials used herein preferably have particle size
10 diameters which optimize their effectiveness as detergent builders. The term "particle size diameter" as used herein represents the average particle size diameter of a given aluminosilicate ion exchange material as determined by conventional analytical techniques, such as microscopic determination and scanning electron microscope (SEM). The preferred particle size diameter of the aluminosilicate is from about 0.1 micron to about 10 microns, more preferably from about 0.5 microns to about 9 microns.
15 Most preferably, the particle size diameter is from about 1 microns to about 8 microns.

Preferably, the aluminosilicate ion exchange material has the formula



wherein z and y are integers of at least 6, the molar ratio of z to y is from about 1 to about 5 and x is from about 10 to about 264. More preferably, the aluminosilicate has the formula



wherein x is from about 20 to about 30, preferably about 27. These preferred aluminosilicates are available commercially, for example under designations Zeolite A, Zeolite B and Zeolite X.

- Alternatively, naturally-occurring or synthetically derived aluminosilicate ion exchange materials suitable for use herein can be made as described in Krummel et al, U.S. Patent No. 3,985,669, the
25 disclosure of which is incorporated herein by reference.

- The aluminosilicates used herein are further characterized by their ion exchange capacity which is at least about 200 mg equivalent of CaCO_3 hardness/gram, calculated on an anhydrous basis, and which is preferably in a range from about 300 to 352 mg equivalent of CaCO_3 hardness/gram. Additionally, the instant aluminosilicate ion exchange materials are still further characterized by their
30 calcium ion exchange rate which is at least about 2 grains Ca^{++} /gallon/minute/-gram/gallon, and more preferably in a range from about 2 grains Ca^{++} /gallon/minute/-gram/gallon to about 6 grains Ca^{++} /gallon/minute/-gram/gallon.

Adjunct Detergent Ingredients

- The starting dry detergent material in the present process can include additional detergent
35 ingredients and/or, any number of additional ingredients can be incorporated in the detergent composition during subsequent steps of the present process. These adjunct ingredients include other detergency builders, bleaches, bleach activators, suds boosters or suds suppressors, anti-tarnish and

anticorrosion agents, soil suspending agents, soil release agents, germicides, pH adjusting agents, non-builder alkalinity sources, chelating agents, smectite clays, enzymes, enzyme-stabilizing agents and perfumes. See U.S. Patent 3,936,537, issued February 3, 1976 to Baskerville, Jr. et al., incorporated herein by reference.

5 Other builders can be generally selected from the various water-soluble, alkali metal, ammonium or substituted ammonium phosphates, polyphosphates, phosphonates, polyphosphonates, carbonates, borates, polyhydroxy sulfonates, polyacetates, carboxylates, and polycarboxylates. Preferred are the alkali metal, especially sodium, salts of the above. Preferred for use herein are the phosphates, carbonates, C₁₀₋₁₈ fatty acids, polycarboxylates, and mixtures
10 thereof. More preferred are sodium tripolyphosphate, tetrasodium pyrophosphate, citrate, tartrate mono- and di-succinates, and mixtures thereof (see below).

In comparison with amorphous sodium silicates, crystalline layered sodium silicates exhibit a clearly increased calcium and magnesium ion exchange capacity. In addition, the layered sodium silicates prefer magnesium ions over calcium ions, a feature necessary to insure that substantially all of
15 the "hardness" is removed from the wash water. These crystalline layered sodium silicates, however, are generally more expensive than amorphous silicates as well as other builders. Accordingly, in order to provide an economically feasible laundry detergent, the proportion of crystalline layered sodium silicates used must be determined judiciously.

The crystalline layered sodium silicates suitable for use herein preferably have the formula
20
$$\text{NaMSi}_x\text{O}_{2x+1}\cdot y\text{H}_2\text{O}$$
 wherein M is sodium or hydrogen, x is from about 1.9 to about 4 and y is from about 0 to about 20. More preferably, the crystalline layered sodium silicate has the formula

$$\text{NaMSi}_2\text{O}_5\cdot y\text{H}_2\text{O}$$
 wherein M is sodium or hydrogen, and y is from about 0 to about 20. These and other crystalline
25 layered sodium silicates are discussed in Corkill et al, U.S. Patent No. 4,605,509, previously incorporated herein by reference.

Specific examples of inorganic phosphate builders are sodium and potassium tripolyphosphate, pyrophosphate, polymeric metaphosphate having a degree of polymerization of from about 6 to 21, and orthophosphates. Examples of polyphosphonate builders are the sodium
30 and potassium salts of ethylene diphosphonic acid, the sodium and potassium salts of ethane 1-hydroxy-1, 1-diphosphonic acid and the sodium and potassium salts of ethane, 1,1,2-triphosphonic acid. Other phosphorus builder compounds are disclosed in U.S. Patents 3,159,581; 3,213,030; 3,422,021; 3,422,137; 3,400,176 and 3,400,148, all of which are incorporated herein by reference.

35 Examples of nonphosphorus, inorganic builders are tetraborate decahydrate and silicates having a weight ratio of SiO₂ to alkali metal oxide of from about 0.5 to about 4.0, preferably from about 1.0 to about 2.4. Water-soluble, nonphosphorus organic builders useful herein include the

various alkali metal, ammonium and substituted ammonium polyacetates, carboxylates, polycarboxylates and polyhydroxy sulfonates. Examples of polyacetate and polycarboxylate builders are the sodium, potassium, lithium, ammonium and substituted ammonium salts of ethylene diamine tetraacetic acid, nitrilotriacetic acid, oxydisuccinic acid, mellitic acid, benzene polycarboxylic acids, and citric acid.

Polymeric polycarboxylate builders are set forth in U.S. Patent 3,308,067, Diehl, issued March 7, 1967, the disclosure of which is incorporated herein by reference. Such materials include the water-soluble salts of homo- and copolymers of aliphatic carboxylic acids such as maleic acid, itaconic acid, mesaconic acid, fumaric acid, aconitic acid, citraconic acid and methylene malonic acid. Some of these materials are useful as the water-soluble anionic polymer as hereinafter described, but only if in intimate admixture with the non-soap anionic surfactant.

Other suitable polycarboxylates for use herein are the polyacetal carboxylates described in U.S. Patent 4,144,226, issued March 13, 1979 to Crutchfield et al, and U.S. Patent 4,246,495, issued March 27, 1979 to Crutchfield et al, both of which are incorporated herein by reference. These polyacetal carboxylates can be prepared by bringing together under polymerization conditions an ester of glyoxylic acid and a polymerization initiator. The resulting polyacetal carboxylate ester is then attached to chemically stable end groups to stabilize the polyacetal carboxylate against rapid depolymerization in alkaline solution, converted to the corresponding salt, and added to a detergent composition. Particularly preferred polycarboxylate builders are the ether carboxylate builder compositions comprising a combination of tartrate monosuccinate and tartrate disuccinate described in U.S. Patent 4,663,071, Bush et al., issued May 5, 1987, the disclosure of which is incorporated herein by reference.

Bleaching agents and activators are described in U.S. Patent 4,412,934, Chung et al., issued November 1, 1983, and in U.S. Patent 4,483,781, Hartman, issued November 20, 1984, both of which are incorporated herein by reference. Chelating agents are also described in U.S. Patent 4,663,071, Bush et al., from Column 17, line 54 through Column 18, line 68, incorporated herein by reference. Suds modifiers are also optional ingredients and are described in U.S. Patents 3,933,672, issued January 20, 1976 to Bartoletta et al., and 4,136,045, issued January 23, 1979 to Gault et al., both incorporated herein by reference.

Suitable smectite clays for use herein are described in U.S. Patent 4,762,645, Tucker et al, issued August 9, 1988, Column 6, line 3 through Column 7, line 24, incorporated herein by reference. Suitable additional detergency builders for use herein are enumerated in the Baskerville patent, Column 13, line 54 through Column 16, line 16, and in U.S. Patent 4,663,071, Bush et al, issued May 5, 1987, both incorporated herein by reference.

In order to make the present invention more readily understood, reference is made to the following examples, which are intended to be illustrative only and not intended to be limiting in scope.

EXAMPLE I

This Example illustrates the process of the invention which produces free flowing, crisp, high density detergent composition. Two feed streams of various detergent starting ingredients are continuously fed, at a rate of 2800 kg/hr, into a Lödige Recycler KM-600 mixer/densifier, one of which comprises a surfactant paste containing surfactant and water and the other stream containing starting dry detergent material containing aluminosilicate and sodium carbonate. The rotational speed of the shaft in the Lödige KM-600 mixer/densifier is about 120 rpm and the mean residence time is about 10 minutes. The resulting detergent agglomerates are then fed to conditioning apparatus including a fluid bed dryer and then to a fluid bed cooler, the mean residence time being about 10 minutes and 15 minutes, respectively. The undersized or "fine" agglomerate particles (less than about 150 microns) from the fluid bed dryer and cooler are recycled back into the Lödige KM-600 mixer/densifying. A coating agent, aluminosilicate, is fed immediately after the Lödige KM-600 mixer/densifier but before the fluid bed dryer to enhance the flowability of the agglomerates. The detergent agglomerates exiting the fluid bed cooler are screened, after which adjunct detergent ingredients are admixed therewith to result in a fully formulated detergent product having a uniform particle size distribution. The composition of the detergent agglomerates exiting the fluid bed cooler is set forth in Table I below:

TABLE I

Component	% Weight
C ₁₄₋₁₅ alkyl sulfate/alkyl ethoxy sulfate	30.0
Aluminosilicate	37.8
Sodium carbonate	19.1
Misc. (water, perfume, etc.)	<u>13.1</u>
	100.0

The density of the agglomerates in Table I is 750 g/l and the median particle size is 475 microns.

Adjunct liquid detergent ingredients including perfumes, brighteners and enzymes are sprayed onto or admixed to the agglomerates/particles described above in the finishing step to result in a fully formulated finished detergent composition. The relative proportions of the overall finished detergent composition produced by the process of instant process is presented in Table II below:

TABLE II

<u>Component</u>	<u>(% weight)</u>
C ₁₄₋₁₅ alkyl sulfate/C ₁₄₋₁₅ alkyl ethoxy sulfate/C ₁₂ linear alkylbenzene sulfonate	21.6
Polyacrylate (MW=4500)	2.5
Polyethylene glycol (MW=4000)	1.7

Sodium Sulfate	6.9
Aluminosilicate	25.6
Sodium carbonate	17.9
Protease enzyme	0.3
Cellulase enzyme	0.4
Lipase enzyme	0.3
Minors (water, perfume, etc.)	<u>22.8</u>
	100.0

The density of the detergent composition in Table II is 660 g/l.

EXAMPLE II

This Example illustrates another process in accordance with the invention in which the steps described in Example I are performed in addition to the following steps: (1) screening the agglomerates exiting the Lödige KM-600 such that the oversized particles (at least about 4 mm) are sent to a grinder; (2) screening the oversized agglomerate particles (at least about 1180 microns) exiting the fluid bed cooler and sending those oversized particles to the grinder, as well; and (3) inputting the ground oversized agglomerate particles back into the fluid bed dryer and/or fluid bed cooler. Additionally, a coating agent, aluminosilicate, is added between the fluid bed cooler and the finishing (admixing and/or spraying adjunct ingredients) steps. The composition of the detergent agglomerates exiting the fluid bed cooler is set forth in Table III below:

TABLE III

Component	% Weight
C ₁₄₋₁₅ alkyl sulfate/alkyl ethoxy sulfate	30.0
Aluminosilicate	37.8
Sodium carbonate	19.1
Misc. (water, perfume, etc.)	<u>13.1</u>
	100.0

The density of the agglomerates in Table I is 750 g/l and the median particle size is 425 microns.

The agglomerates also surprisingly have a more narrow particle size distribution, wherein more than 90% of the agglomerates have a particle size between about 150 microns to about 1180 microns. This result unexpectedly matches the desired particle size distribution (i.e. all agglomerates less than about 1180 microns) more closely.

Adjunct liquid detergent ingredients including perfumes, brighteners and enzymes are sprayed onto or admixed to the agglomerates/particles described above in the finishing step to result in a fully formulated finished detergent composition. The relative proportions of the overall finished detergent composition produced by the process of instant process is presented in Table IV below:

TABLE IV

<u>Component</u>	<u>(% weight)</u>
<u>B</u>	
C ₁₄₋₁₅ alkyl sulfate/C ₁₄₋₁₅ alkyl ethoxy sulfate/C ₁₂ linear alkylbenzene sulfonate	21.6
Polyacrylate (MW=4500)	2.5
Polyethylene glycol (MW=4000)	1.7
Sodium Sulfate	6.9
Aluminosilicate	25.6
Sodium carbonate	17.9
Protease enzyme	0.3
Cellulase enzyme	0.4
Lipase enzyme	0.3
Minors (water, perfume, etc.)	<u>22.8</u>
	100.0

The density of the detergent composition in Table IV is 660 g/l.

Having thus described the invention in detail, it will be clear to those skilled in the art that various changes may be made without departing from the scope of the invention and the invention is

5 not to be considered limited to what is described in the specification.

WHAT IS CLAIMED IS:

1. A process for continuously preparing high density detergent composition characterized by the steps of:
 - (a) continuously charging a detergent surfactant paste and dry starting detergent material into a mixer/densifier for densification and build-up such that agglomerates having a median particle size from 300 microns to 900 microns are formed;
 - (b) feeding said agglomerates into a conditioning apparatus for improving the flow properties of said agglomerates and for separating said agglomerates into a first agglomerate mixture and a second agglomerate mixture, wherein said first agglomerate mixture substantially has a particle size of less than 150 microns and said second agglomerate mixture substantially has a particle size of at least 150 microns;
 - (c) recycling said first agglomerate mixture into said mixer/densifier for further agglomeration;
 - (e) admixing adjunct detergent ingredients to said second agglomerate mixture so as to form said high density detergent composition.
2. A process according to claim 1 wherein said conditioning apparatus is characterized by a fluid bed dryer and a fluid bed cooler.
3. A process according to claims 1-2 wherein the ratio of said surfactant paste to said dry detergent material is from 1:10 to 10:1
4. A process according to claims 1-3 wherein said dry starting material is characterized by a builder selected from the group consisting of aluminosilicates, crystalline layered silicates, and mixtures thereof and sodium carbonate.
5. A process according to claims 1-4 wherein the density of said detergent composition is at least 650 g/l.
6. A process according to claims 1-5 further characterized by the step of adding a coating agent after said mixer/densifier, wherein said coating agent is selected from the group consisting of aluminosilicates, carbonates, silicates and mixtures thereof.
7. A process according to claims 1-6 wherein said mixer/densifier is a high speed mixer/densifier and the mean residence time of said agglomerates in said high speed mixer/densifier is in a range of from 2 seconds to 45 seconds.

8. A process according to claims 1-7 wherein said mixer/densifier is a moderate speed mixer/densifier and the mean residence time of said agglomerates in said moderate speed mixer/densifier is in a range of from 0.5 minutes to 15 minutes.
9. A process according to claims 1-8 further characterized by the step of spraying a binder material into said mixer/densifier.
10. A process according to claims 1-9 wherein said binder is selected from the group consisting of water, anionic surfactants, nonionic surfactants, polyethylene glycol, polyvinyl pyrrolidone, polyacrylates, citric acid and mixtures thereof.

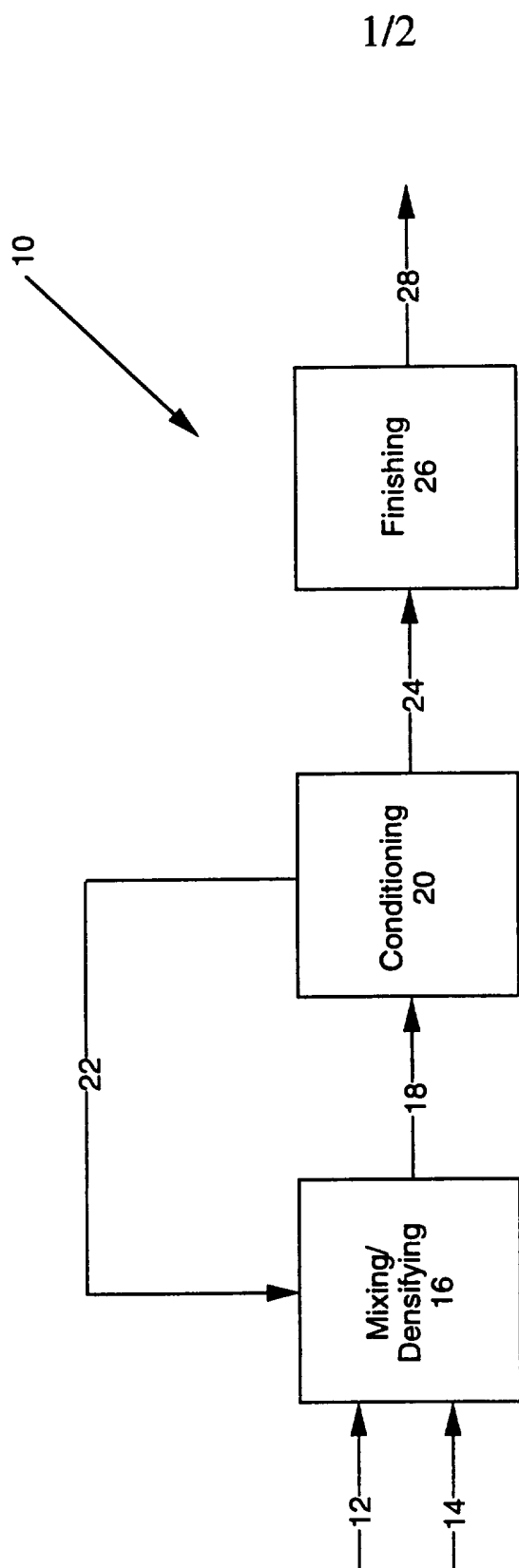


Fig. 1

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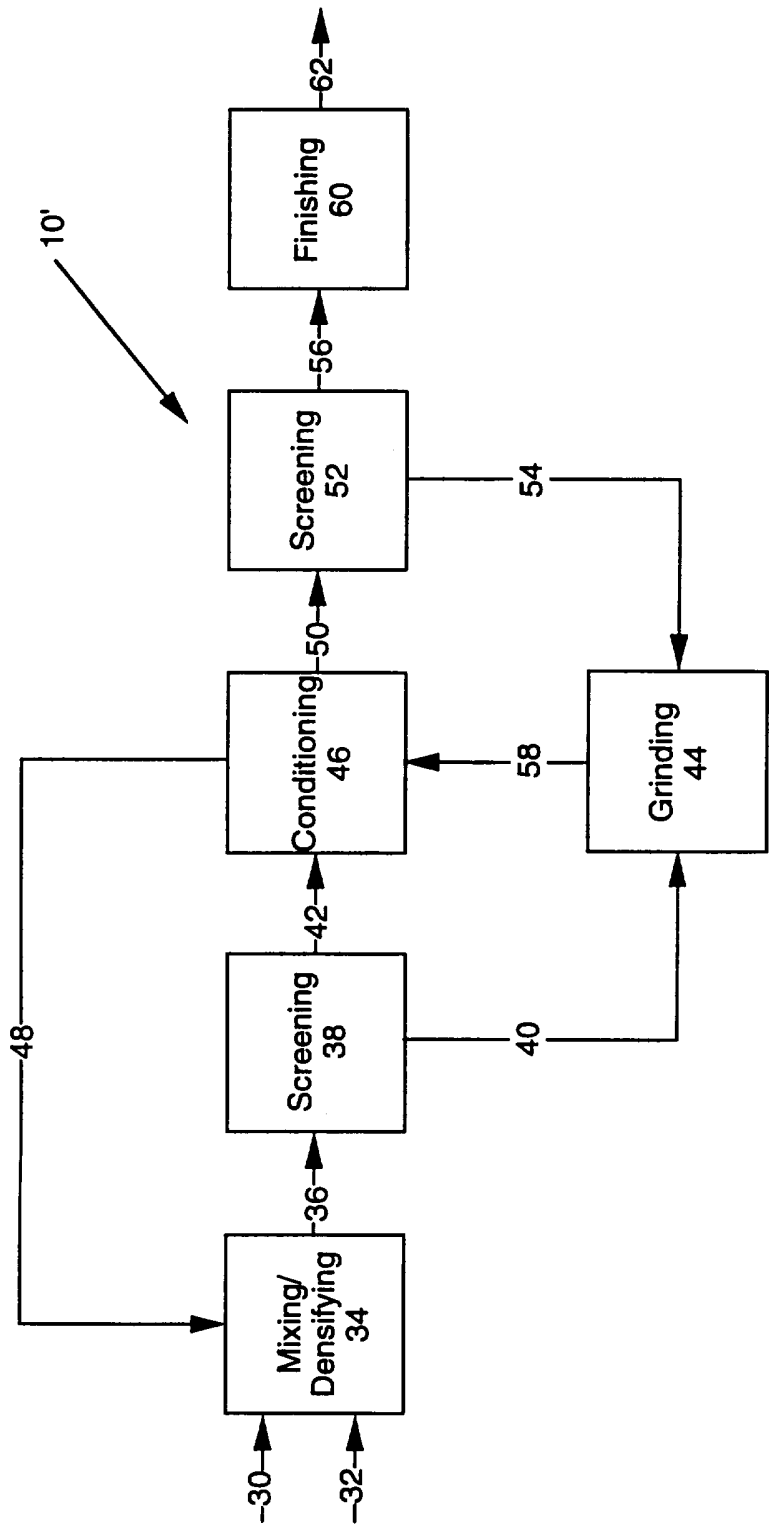


Fig. 2

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 95/11264

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C11D17/06 C11D11/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C11D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO,A,93 25378 (PROCTER & GAMBLE) 23 December 1993 see claims 1-6; example 1 ---	1,4,5,7
A	EP,A,0 513 824 (KAO CORP) 19 November 1992 see page 3, line 53 - page 4, line 35; claims 1,2,11 ---	1,4,5
A	US,A,4 970 017 (NAKAMURA MASAYOSHI ET AL) 13 November 1990 see claims; examples ---	1,4-6
A	EP,A,0 544 492 (UNILEVER) 2 June 1993 see page 7, line 30 - line 57; claims 1,12,13 --- -/--	1,4,5

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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- * & * document member of the same patent family

Date of the actual completion of the international search

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INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US 95/11264

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

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A	US,A,5 332 519 (MAZZOLA LOUIS) 26 July 1994 see column 6, line 63 - column 7, line 10; claims 1-6; examples -----	1,5

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