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DESCRIPTION

BACKGROUND

Cross-Reference to Related Applications

[0001] This application claims the benefit of priority to U.S. Provisional Application No. 61/479,281, filed April 26, 2011.

Field

[0002] Isotopically modified polyunsaturated fatty acids ("PUFAs") and other modified PUFAs for treating certain diseases, particularly Wet and Dry Age-related Macular Degeneration (AMD); Retinitis Pigmentosa (RP); Diabetic Retinopathy (DR); Cataracts; and Stargardt Disease (SD).

Description of the Related Art

[0003] Oxidative stress is implicated in a wide variety of diseases such as mitochondrial diseases, neurodegenerative diseases, inborn error's of metabolism, diabetes, diseases of the eye, kidney diseases, liver diseases, and cardiac diseases. Specifically, such diseases include but are not limited to Wet and Dry Age-related Macular Degeneration (AMD); Retinitis Pigmentosa (RP); Diabetic Retinopathy (DR); Cataracts; and Stargardt Disease (SD). See, for example, U.S. Patent Application Nos.: 20110046219 and 20080275005 and the references cited therein.

[0004] While the number of diseases associated with oxidative stress are numerous and diverse, it is well established that oxidative stress is caused by disturbances to the normal redox state within cells. An imbalance between routine production and detoxification of reactive oxygen species ("ROS") such as peroxides and free radicals can result in oxidative damage to cellular structures and machinery. Under normal conditions, potentially important sources of ROSs in aerobic organisms is the leakage of activated oxygen from mitochondria during normal oxidative respiration. Additionally, it is known that macrophages and enzymatic reactions also contribute to the generation of ROSs within cells. Because cells and their internal organelles are lipid membrane-bound, ROSs can readily contact membrane constituents and cause lipid oxidation. Ultimately, such oxidative damage can be relayed to other biomolecules within the cell, such as DNA and proteins, through direct and indirect contact with activated oxygen, oxidized membrane constituents, or other oxidized cellular

components. Thus, one can readily envision how oxidative damage may propagate throughout a cell given the mobility of internal constituents and the interconnectedness of cellular pathways.

[0005] Lipid-forming fatty acids are well-known as one of the major components of living cells. As such, they participate in numerous metabolic pathways, and play an important role in a variety of pathologies. Polyunsaturated Fatty Acids ("PUFAs") are an important subclass of fatty acids. An essential nutrient is a food component that directly, or *via* conversion, serves an essential biological function and which is not produced endogenously or in large enough amounts to cover the requirements. For homeothermic animals, the two rigorously essential PUFAs are linoleic (*cis,cis*-9,12-Octadecadienoic acid; (9Z,12Z)-9,12-Octadecadienoic acid; "LA"; 18:2;n-6) and alpha-linolenic (*cis,cis,cis*-9,12,15-Octadecatrienoic acid; (9Z,12Z,15Z)-9,12,15-Octadecatrienoic acid; "ALA"; 18:3;n-3) acids, formerly known as vitamin F (Cunnane SC. Progress in Lipid Research 2003; 42:544-568). LA, by further enzymatic desaturation and elongation, is converted into higher n-6 PUFAs such as arachidonic (AA; 20:4;n-6) acid; whereas ALA gives rise to a higher n-3 series, including, but not limited to, eicosapentaenoic acid (EPA; 20:5;n-3) and docosahexaenoic (DHA; 22:6;n-3) acid (Goyens PL. et al. Am. J. Clin. Nutr. 2006; 84:44-53). Because of the essential nature of certain PUFAs or PUFA precursors, there are many known instances of their deficiency and these are often linked to medical conditions. Furthermore, many PUFA supplements are available over-the-counter, with proven efficiency against certain ailments (See, for example, U.S. Patent No.: 7,271,315 and U.S. Patent No.: 7,381,558).

[0006] PUFAs endow mitochondrial membranes with appropriate fluidity necessary for optimal oxidative phosphorylation performance. PUFAs also play an important role in initiation and propagation of the oxidative stress. PUFAs react with ROS through a chain reaction that amplifies an original event (Sun M, Salomon RG, J. Am. Chem. Soc. 2004; 126:5699-5708). However, non-enzymatic formation of high levels of lipid hydroperoxides is known to result in several detrimental changes. Indeed, Coenzyme Q10 has been linked to increased PUFA toxicity via PUFA peroxidation and the toxicity of the resulting products (Do TQ et al, PNAS USA 1996; 93:7534-7539). Such oxidized products negatively affect the fluidity and permeability of their membranes; they lead to oxidation of membrane proteins; and they can be converted into a large number of highly reactive carbonyl compounds. The latter include reactive species such as acrolein, malonic dialdehyde, glyoxal, methylglyoxal, etc. (Negre-Salvayre A, et al. Brit. J. Pharmacol. 2008; 153:6-20). But the most prominent products of PUFA oxidation are alpha, beta-unsaturated aldehydes such as 4-hydroxynon-2-enal (4-HNE; formed from n-6 PUFAs like LA or AA), 4-hydroxyhex-2-enal (4-HHE; formed from n-3 PUFAs like ALA or DHA), and corresponding ketoaldehydes (Esterbauer H, et al. Free Rad. Biol. Med. 1991; 11:81-128; Long EK, Picklo MJ. Free Rad. Biol. Med. 2010; 49:1-8). These reactive carbonyls cross-link (bio)molecules through Michael addition or Schiff base formation pathways, and have been implicated in a large number of pathological processes (such as those introduced above), age-related and oxidative stress-related conditions, and aging. Importantly, in some cases, PUFAs appear to oxidize at specific sites because methylene groups of 1,4-diene systems (the bis-allylic position) are substantially less stable to ROS, and to enzymes such as cyclooxygenases and lipoxygenases, than allylic methylenes.

WO2011/053870, WO2008/143642, Townend et al ("Dietary Macronutrient Intake and Five-Year Incident Cataract: The Blue Mountains Eye Study", AMERICAN JOURNAL OF OPHTHALMOLOGY, ELSEVIER,AMSTERDAM, NL,vol. 143, no. 6, 22 May 2007 (2007-05-22), pp. 932-939), WO 00/21524 A1, WO 2010/010365 A1, WO 2009/097331 A1, Hill, S ("Isotope-reinforced polyunsaturated fatty acids protect yeast cells from oxidative stress.", FREE RADICAL BIOLOGY & MEDICINE, vol. 50, no. 1, 1 January 2011 (2011-01-01), pp 130-138), Shchepinov M. S. et al ("Isotope effect, essential diet components, and prospects of aging retardation.", RUSSIAN JOURNAL OF GENERAL CHEMISTRY, vol. 80, no. 7, 2010, pp 1514-1522), and US 2009/215896 A1 disclose the use of PUFAs.

[0007] We have now discovered that oxidation resistant PUFAs, PUFA mimetics, PUFA esters and/or fats containing oxidation resistant PUFAs and PUFA mimetics are useful for mitigating and/or treating oxidative retinal diseases.

SUMMARY

[0008] Provided herein is a polyunsaturated substance for use in the treatment of Wet or Dry Age-related Macular Degeneration (AMD), Retinitis Pigmentosa (RP), Diabetic Retinopathy (DR), Cataracts, or Stargardt Disease (SD), wherein the polyunsaturated substance comprises a fatty acid or fatty acid ester that is isotopically modified with deuterium at one or more positions, wherein the polyunsaturated substance is incorporated into the patient's body following administration.

[0009] In some embodiments, the fatty acid or fatty acid ester is stabilized at one or more bis-allylic positions. According to the present disclosure, the stabilization comprises at least one ^{13}C atom or at least one deuterium atom at a bis-allylic position. In some embodiments, the stabilization comprises at least two deuterium atoms at one or more bis-allylic position. According to the present disclosure, the stabilization utilizes an amount of isotopes that is above the naturally-occurring abundance level. In some embodiments, the stabilization utilizes an amount of isotopes that is significantly above the naturally-occurring abundance level of the isotope.

[0010] According to the present disclosure, the fatty acid or fatty acid ester has an isotopic purity of from about 20%-99%. Further according to the present disclosure, the isotopically stabilized fatty acids or fatty acid esters are administered to a patient along with non-stabilized fatty acids or fatty acid esters. The isotopically stabilized fatty acid or fatty acid esters comprise between 5% and 75%, between about 10% and 30%, or about 20% or more of the total amount of fatty acids or fatty acid esters administered to or ingested by the patient. In some embodiments, a cell or tissue of the patient maintains a sufficient concentration of the isotopically modified fatty acid or fatty acid ester to prevent autooxidation of the naturally occurring polyunsaturated fatty acid or fatty acid ester. According to the present disclosure, the stabilization utilizes an amount of isotope that is significantly above the naturally-occurring

abundance level of said isotope.

[0011] In some embodiments the polyunsaturated substance is an omega-3 fatty acid and/or an omega-6 fatty acid. In other embodiments, the the polyunsaturated substance is selected from the group consisting of 11,11-D2-linolenic acid, 14,14-D2-linolenic acid, 11,11,14,14-D4-linolenic acid, 11,11-D2-linoleic acid, 11-D-linolenic acid, 14-D-linolenic acid, 11,14-D2-linolenic acid, and 11-D-linoleic acid. According to the present disclosure, the the polyunsaturated substance may also be 14,14-D2-linoleic acid, 11,11,14,14-D4-linoleic acid, 14-D-linoleic acid, and 11,14-D2-linoleic acid. In other embodiments, the polyunsaturated substance is further stabilized at a pro-bis-allylic position. In some embodiments, the fatty acid is alpha linolenic acid, linoleic acid, gamma linolenic acid, dihomo gamma linolenic acid, or ester thereof. According to the present disclosure, the fatty acid is incorporated into the mitochondrial membrane. In some embodiments, the ester is a triglyceride, diglyceride, or monoglyceride.

[0012] Some embodiments further comprise co-administering an antioxidant. In some embodiments, the antioxidant is Coenzyme Q, idebenone, mitoquinone, mitoquinol. In other embodiments, the antioxidant is a mitochondrially-targeted antioxidant, vitamin E or vitamin C.

BRIEF DESCRIPTION OF THE DRAWINGS

[0013]

Figures 1A and 1B. (1A) ROS-driven oxidation of PUFAs; (1B) formation of toxic carbonyl compounds.

Figures 2A and 2B. ¹H- and ¹³C-NMR analysis of deuterated PUFAs described in Examples 1-4.

Figure 3. Sensitivity of coq null mutants to treatment with linolenic acid is abrogated by isotope-reinforcement. Yeast coq3, coq7 and coq9 null mutants were prepared in the W303 yeast genetic background (WT). Yeast strains were grown in YPD medium (1% Bacto-yeast extract, 2% Bacto-peptone, 2% dextrose) and harvested while in log phase growth (OD_{600nm}=0.1-1.0). Cells were washed twice with sterile water and resuspended in phosphate buffer (0.10 M sodium phosphate, pH 6.2, 0.2% dextrose) to an OD_{600nm}=0.2. Samples were removed and 1:5 serial dilutions starting at 0.20 OD/ml were plated on YPD plate medium, to provide a zero time untreated control (shown in top left panel). The designated fatty acids were added to 200 μM final concentration to 20 ml of yeast in phosphate buffer. At 2 h, 4 h, and 16 h samples were removed, 1:5 serial dilutions prepared, and spotted onto YPD plate medium. Pictures were taken after 2 days of growth at 30°C. This panel is representative of two independent assays, performed on different days.

Figure 4. Yeast coq mutants treated with isotope-reinforced D4-linolenic acid are resistant to PUFA-mediated cell killing. The fatty acid sensitive assay was performed as described in Figure 6-1, except that 100 μl aliquots were removed at 1, 2, and 4 h and, following dilution,

spread onto YPD plates. Pictures were taken after 2 to 2.5 days, and the number of colonies counted. Yeast strains include Wild type (circles), *atp2* (triangles), or *coq3* (squares); Fatty acid treatments include oleic C18:1 (solid line), linolenic, C18:3, n-3 (dashed line) or 11,11,14,14-D4-linolenic, C18:3, n-3, (dotted line).

Figure 5. Separation and detection of fatty acid methyl ester (FAME) standards by GC-MS. FAMES were prepared as described (Moss CW, Lambert MA, Merwin WH. Appl. Microbiol. 1974; 1, 80-85), and the indicated amounts of free fatty acids and 200 µg of C17:0 (an internal standard) were subjected to methylation and extraction. Samples analyses were performed on an Agilent 6890-6975 GC-MS with a DB-wax column (0.25 mm X 30 m X 0.25-µm film thickness) (Agilent, catalog 122-7031).

Figure 6. Uptake of exogenously supplied fatty acids by yeast. WT (W303) yeast were harvested at log phase and incubated in the presence of 200 µM of the designated fatty acid for either 0 or 4 h. Yeast cells were harvested, washed twice with sterile water and then subjected to alkaline methanolysis and saponification, and lipid extraction as described (Moss CW, Lambert MA, Merwin WH. Appl. Microbiol. 1974; 1, 80-85; (Shaw, 1953 Shaw, W. H. C.; Jefferies, J. P. Determination of ergosterol in yeast. Anal Chem 25:1130; 1953). Each designated fatty acid is given as µg per OD_{600nm} yeast, and was corrected for the recovery of the C17:0 internal standard.

Figure 7. Kinetics of O₂ consumption accompanied the oxidation of 0.71 M LA (plots 1 and 2) and 0.71 M D2-LA (plot 3) in chlorobenzene initiated by 40 mM AMVN at 37° C. Plot 2 - 0.23 mM HPMC was added to 0.71 M LA.

Figure 8. Dependence of the rate of oxidation of the mixture of LA and D2-LA in chlorobenzene solution on mixture composition. Conditions: [LA] + [11,11-d₂-LA] = 0.775 M; [AMVN] = 0.0217 M; 37° C. $R_{IN} = (1.10 \pm 0.08) \times 10^{-7}$ M/sec.

Figure 9. Isotope reinforcement at the bis-allylic position of polyunsaturated fatty acids attenuates lipid autoxidation. Wild-type, yeast Q-less *coq3*, or respiratory deficient *cor1* null mutants were incubated in the presence of 200 µM of LA and D2-LA at different ratios of PUFAs. Serial dilutions (1:5) starting at 0.2OD/ml were spotted on YPD solid plate medium. A zero-time untreated control is shown on the top left. Growth at 30° C.

Figure 10. Isotope reinforcement at the bis-allylic position of polyunsaturated fatty acids attenuates lipid autoxidation. Wild-type, yeast Q-less *coq3*, or respiratory deficient *cor1* null mutants were incubated in the presence of 200 µM of ALA and D4-LA at different ratios of PUFAs. Serial dilutions (1:5) starting at 0.2OD/ml were spotted on YPD solid plate medium. Growth at 30° C.

Figure 11. Chromatograms of the yeast extracts subjected to GC-MS analyses. The different traces represent the 0 and 4 h incubations, respectively. The peak area of Each FAME (C18:1, C18:3 and D4-linolenic) was divided by the peak area of the C17:0 standard, quantified with a calibration curve. The endogenous 16:0 and 16:1 change very little, while the exogenously

added fatty acids increased significantly.

Figure 12. Survival of H- and D-PUFA treated MVEC cells after acute intoxication by paraquat. For all cell types tested, D-PUFA had protective effect compared to controls, similar to that shown on Figure for MVEC cells.

Figure 13. Animal dosage studies of 1:1 D2-LA/D4-ALA indicating tissue enrichment with deuterium.

Figure 14. Animal dosage studies of 1:1 D2-LA/D4-ALA comparing any changes in fat distribution.

Figure 15. Animal dosage studies of 1:1 D2-LA/ALA indicating tissue enrichment with deuterium.

Figure 16. Control liver fat profile after 90-day animal dosage study.

Figure 17. Animal dosage studies of 1:1 D2-LA/D4-ALA indicating liver fat profile and enrichment with deuterium.

Figure 18. Liver fat profile after 90-day animal dosage study with D2-LA.

Figure 19. Animal dosage studies of 1:1 D2-LA/D4-ALA indicating brain fat profile and enrichment with deuterium.

Figure 20. Animal dosage studies of 1:1 D2-LA/ALA indicating brain fat profile and enrichment with deuterium.

Figure 21. Control brain fat profile after 90-day animal dosage study.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

[0014] As used herein, abbreviations are defined as follows:

α Lnn

Alpha-linolenic acid

4-HHE or HHE

4-Hydroxyhex-2-enal

4-HNE

4-Hydroxynon-2-enal (4-HNE; formed from n-6 PUFAs like LA or AA)

AA

Arachidonic (AA; 20:4;n-6) acid

AcOH

Acetic acid

ALA

	Alpha-linolenic acid
AMD	Age-related macular degeneration
AMVN	2,2'-Azobis(2,4-dimethylvaleronitrile)
D-	Deuterated
D1	Mono-deuterated
D2	Di-deuterated
D2-LA	Di-deuterated linoleic acid
D3	Tri-deuterated
D4	Tetra-deuterated
D5	Penta-deuterated
D6	Hexa-deuterated
DHA	Docosahexaenoic (22:6; n-3) acid
DMF	Dimethylformamide
DR	Diabetic Retinopathy
EPA	Eicosapentaenoic (20:5; n-3) acid
EtOAc	Ethyl acetate
EtOH	Ethanol
FAME	Fatty acid methyl ester
HPMC	6-Hydroxy-2,2,5,7,8-pentamethylbenzochroman
H-PUFA	Non-deuterated polyunsaturated fatty acid
IP	Intraperitoneal
IR	Infrared
KIE	

	Kinetic isotope effect
LA	Linoleic acid
LDL	Low-density lipoprotein
MDA	Malondialdehyde
MPTP	1-Methyl-4-phenyl-1,2,3,6-tetrahydropyridine
MVEC	Microvascular endothelium
NINCDS	Neurological and Communicative Disorders and Stroke
PUFA(s)	Polyunsaturated fatty acid(s)
R _{IN}	Rate of initiation
ROS	Reactive oxygen species
R _{OX}	Rate of oxidation
RP	Retinitis pigmentosa
RPE	Retinal pigment epithelium
SD	Stargardt Disease
SNOMED	Systematized Nomenclature of Medicine
SOD	Superoxide dismutase
TDMS	Toxicology Data Management System
TH	Tyrosine hydroxylase
THF	Tetrahydrofuran
TLC	Thin layer chromatography
V-SMOW	Vienna standard mean ocean water
WT	Wild type
YPD	

Medium containing 1% Bacto-yeast extract, 2% Bacto-peptone, 2% dextrose

Wet and Dry Age-related Macular Degeneration (AMD); Retinitis Pigmentosa (RP); Diabetic Retinopathy (DR); Cataracts; and Stargardt Disease (SD)

[0015] It has long been appreciated that oxidative stress plays a major role in many retinal degeneration conditions (Winkler BS et al, *Molecular Vision* 1999;5:32). Increased oxygen levels, exposure to light and high PUFA content lead to increased PUFA peroxidation in the eye tissues. AMD, the leading cause of blindness in elderly populations in the developed world, is accompanied by degeneration of retinal pigment epithelial cells (RPE). Although smoking, atherosclerosis, genetic factors and fibrinogen have been implicated in the pathogenesis of this disease, the exact cause of AMD remains unknown (Baskol G. Et al. *Ophthalmol.* 2006;220:12-16). Oxidative stress plays a major role in the pathogenesis of AMD (Beatty S, et al. *Survey Ophthalm.* 2000; 45:115-134). The retinal pigment epithelium (RPE) is essential for retinoid recycling and phagocytosis of photoreceptors. Methods of comparative proteomics have been applied to uncover global changes of RPE proteins that mediate oxidative stress-induced degeneration of RPE (Arnouk H et al, *J. Proteom.* 2011;74:254-261). Increased levels of PUFA peroxidation products such as HNE and HHE (Long EK, et al. *Free Rad. Biol. Med.* 2010; 49:1-8) as well as MDA (Singh SB et al, *J. Clin. Diagnostic Res.* 2010;4:2768-2769) have been reported in retinas, while increased level of oxidized phospholipids in photoreceptors and RPE is associated with aging and AMD (Suzuki M et al, *Molec. Vision* 2007;13:772-778). PUFA peroxidation products play a major role in formation of retinal pigment epithelial (RPE) lipofuscin, which itself, and mostly through its A2E component, can generate ROS upon irradiation with visible light and plays a major role in etiology of AMD (Katz ML, *Arch. Gerontol. Geriatr.* 2002; 34:359-370). In Stargardt disease (SD), lipofuscin content in the photoreceptor outer segment and RPE is increased due to ABCA4 mutations (Weng J et al, *PNAS* 2000;97:7154-7159). PUFA peroxidation products, including MDA, play such a prominent role in lens pathologies including formation of cataracts, that the PUFA peroxidation was proclaimed to be an initiating step in the human cataract pathogenesis (Borchman D. et al, *J. Lipid Res.* 2010; 51:2473-2488). Equally important is the role of PUFA peroxidation products in pathophysiology of diseases of human cornea, including pterygium and keratoconus (Shoham A, et al. *Free Rad. Biol. Med.* 2008; 45:1047-1055). Diabetic retinopathy is also associated with oxidative stress and PUFA peroxidation (Baynes JW, Thorpe SR. *Diabetes* 1999; 48:1-9). Peroxidation of linoleic acid and other PUFAs that make up LDL has been shown to affect exudative AMD (Javadzadeh a et al, *Molec. Vision* 2010;16:275-276).

[0016] Despite the important role the oxidative stress plays in these degenerative conditions, the success of the antioxidant therapies has so far been limited. This may be due to several reasons: (a) antioxidants are usually present in cells at high (virtually saturated) concentrations, and further supplementation leads to only marginal increases (Zimniak P *Ageing Res. Rev.* 2008; 7:281-300). The stochastic nature of ROS-inflicted damage is

therefore not sensitive to antioxidant therapies; (b) ROS themselves are important in cell signaling and other processes, including the requirement for low levels of ROS for hormetic (adaptive) upregulation of protective mechanisms; (c) some antioxidants (such as vitamin E) can become potent oxidants themselves, capable of initiating PUFA autoxidation (Bowry VW et al, JACS 1993;115:6029-6044). Moreover, vitamin A plays a major role in formation of RPE-associated lipofuscin fluorophore A2E and its derivatives, which are capable of generating ROS upon exposure to irradiation (Sparrow JR et al, Investigative Ophthalm. Vis. Sci. 1999;40:2988-2995); and (d) antioxidants are ineffective in neutralising the carbonyl compounds like HNE and HHE, because HNE and HHE, once formed, react in different ways compared to the free radical mechanism and so cannot be quenched by typical antioxidants.

[0017] In some aspects, identification of a subject who has or is susceptible to AMD, RP, DR, cataracts and/or SD may be determined by diagnostic tests known in the art such as fluorescein angiography or by identifying abnormalities in vascular processes. In addition, Optial Coherence Tomography diagnostics may be used to identify such subjects.

[0018] Some aspects of this invention arise from: (1) an understanding that while essential PUFAs are vital for proper functioning of lipid membranes, and in particular of the mitochondrial membranes, their inherent drawback, i.e., the propensity to be oxidized by ROS with detrimental outcome, is implicated in Wet and Dry Age-related Macular Degeneration (AMD); Retinitis Pigmentosa (RP); Diabetic Retinopathy (DR); Cataracts; and Stargardt Disease (SD); (2) antioxidants cannot prevent PUFA peroxidation due to stochastic nature of the process and the stability of PUFA peroxidation products (reactive carbonyls) to antioxidant treatment, and (3) the ROS-driven damage of oxidation-prone sites within PUFAs may be overcome by using an approach that makes them less amenable to such oxidations, without compromising any of their beneficial physical properties. Some aspects of this invention describe the use of the isotope effect to achieve this, only at sites in essential PUFAs and PUFA precursors that matter most for oxidation, while other aspects contemplate other sites in addition to those that matter most for oxidation.

[0019] Isotopically labeled substances should have minimal or non-existent effects on important biological processes. For example, the natural abundance of isotopes present in biological substrates implies that low levels of isotopically labeled compounds should have negligible effects on biological processes. Additionally, hydrogen atoms are incorporated into biological substrates from water, and it is known that the consumption of D₂O, or heavy water, does not pose a health threat to humans. See, e.g., "Physiological effect of heavy water." Elements and isotopes : formation, transformation, distribution. Dordrecht: Kluwer Acad. Publ. (2003) pp. 111-112 (indicating that a 70 kg person might drink 4.8 liters of heavy water without serious consequences). Moreover, many isotopically labeled compounds are approved by the U.S. Food & Drug Administration for diagnostic and treatment purposes.

[0020] It will be appreciated by those skilful in the art that the same effect can be achieved by protecting oxidation-prone positions within PUFAs using other chemical approaches. Certain PUFA mimetics, while possessing structural similarity with natural PUFAs, will nevertheless be

stable to ROS-driven oxidation due to structural reinforcement.

Compositions:

[0021] According to the present disclosure, an isotopically modified polyunsaturated fatty acid or a mimetic refers to a compound having structural similarity to a naturally occurring PUFA that is stabilized chemically or by reinforcement with one or more isotopes, for example ^{13}C and/or deuterium. Generally, if deuterium is used for reinforcement, one or both hydrogens on a methylene group may be reinforced.

[0022] Some aspects of this invention provide compounds that are analogues of essential PUFAs with either one, several, or all bis-allylic positions substituted with heavy isotopes. In some embodiments, the CH_2 groups, which will become the bis-allylic position in a PUFA upon enzymatic conversion, are substituted with one or two heavy isotopes. Such compounds are useful for the prevention or treatment of diseases in which PUFA oxidation is a factor or can contribute to disease progression.

[0023] The bis-allylic position generally refers to the position of the polyunsaturated fatty acid or mimetic thereof that corresponds to the methylene groups of 1,4-diene systems. The pro-bis-allylic position refers to the methylene group that becomes the bis-allylic position upon enzymatic desaturation.

[0024] In some embodiments, the chemical identity of PUFAs, i.e., the chemical structure without regard to the isotope substitutions or substitutions that mimic isotope substitutions, remains the same upon ingestion. For instance, the chemical identity of essential PUFAs, that is, PUFAs that mammals such as humans do not generally synthesize, may remain identical upon ingestion. In some cases, however, PUFAs may be further extended/desaturated in mammals, thus changing their chemical identity upon ingestion. Similarly with mimetics, the chemical identity may remain unchanged or may be subject to similar extension/desaturation. In some embodiments, PUFAs that are extended, and optionally desaturated, upon ingestion and further metabolism may be referred to as higher homologs.

[0025] According to the present disclosure, naturally-occurring abundance level refers to the level of isotopes, for example ^{13}C and/or deuterium that may be incorporated into PUFAs that would be relative to the natural abundance of the isotope in nature. For example, ^{13}C has a natural abundance of roughly 1% ^{13}C atoms in total carbon atoms. Thus, the relative percentage of carbon having greater than the natural abundance of ^{13}C in PUFAs may have greater than the natural abundance level of roughly 1% of its total carbon atoms reinforced with ^{13}C , such as 2%, but preferably about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100% of ^{13}C with respect to one or more carbon atoms in each PUFA molecule. According to the present disclosure, the

percentage of total carbon atoms reinforced with ^{13}C is at least 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%.

[0026] Regarding hydrogen, according to the present disclosure, deuterium has a natural abundance of roughly 0.0156% of all naturally occurring hydrogen in the oceans on earth. Thus, a PUFA having greater than the natural abundance of deuterium may have greater than this level or greater than the natural abundance level of roughly 0.0156% of its hydrogen atoms reinforced with deuterium, such as 0.02%, but preferably about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100% of deuterium with respect to one or more hydrogen atoms in each PUFA molecule. According to the present disclosure, the percentage of total hydrogen atoms reinforced with deuterium is at least 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%.

[0027] Further disclosed herein is a composition of PUFAs containing both isotopically modified PUFAs and isotopically unmodified PUFAs. The isotopic purity is a comparison between a) the relative number of molecules of isotopically modified PUFAs, and b) the total molecules of both isotopically modified PUFAs and PUFAs with no heavy atoms. In some embodiments, the isotopic purity refers to PUFAs that are otherwise the same except for the heavy atoms.

[0028] According to the present disclosure, isotopic purity refers to the percentage of molecules of an isotopically modified PUFAs in the composition relative to the total number of molecules of the isotopically modified PUFAs plus PUFAs with no heavy atoms. For example, the isotopic purity may be about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100% of the molecules of isotopically modified PUFAs relative to the total number of molecules of both the isotopically modified PUFAs plus PUFAs with no heavy atoms. According to the present disclosure, the isotopic purity is at least 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%. According to the present disclosure, isotopic purity of the PUFAs may be from about 10%-100%, 10%-95%, 10%-90%, 10%-85%, 10%-80%, 10%-75%, 10%-70%, 10%-65%, 10%-60%, 10%-55%, 10%-50%, 10%-45%, 10%-40%, 10%-35%, 10%-30%, 10%-25%, or 10%-20% of the total number of molecules of the PUFAs in the composition. According to the present disclosures, isotopic purity of the PUFAs may be from about 15%-100%, 15%-95%, 15%-90%, 15%-85%, 15%-80%, 15%-75%, 15%-70%, 15%-65%, 15%-60%, 15%-55%, 15%-50%, 15%-45%, 15%-40%, 15%-35%, 15%-30%, 15%-25%, or 15%-20% of the total number of molecules of the PUFAs in the composition. According to the present disclosure, isotopic purity of the PUFAs may be from about 20%-100%, 20%-95%, 20%-90%, 20%-85%, 20%-80%, 20%-75%, 20%-70%, 20%-65%, 20%-60%, 20%-55%, 20%-50%, 20%-45%, 20%-40%, 20%-35%, 20%-30%, or 20%-25% of the total number of molecules of the PUFAs in the composition. Two molecules of an isotopically modified PUFA out of a total of 100 total molecules of isotopically modified PUFAs plus PUFAs with no heavy atoms will have 2% isotopic purity, regardless of the number of heavy atoms the two isotopically modified molecules contain.

[0029] An isotopically modified PUFA molecule contains at least one deuterium atom, such as when one of the two hydrogens in a methylene group is replaced by deuterium, and thus may be referred to as a "D1" PUFA. Similarly, an isotopically modified PUFA molecule may contain two deuterium atoms, such as when the two hydrogens in a methylene group are both replaced by deuterium, and thus may be referred to as a "D2" PUFA. Similarly, an isotopically modified PUFA molecule may contain three deuterium atoms and may be referred to as a "D3" PUFA. Similarly, an isotopically modified PUFA molecule may contain four deuterium atoms and may be referred to as a "D4" PUFA. In some embodiments, an isotopically modified PUFA molecule may contain five deuterium atoms or six deuterium atoms and may be referred to as a "D5" or "D6" PUFA, respectively.

[0030] The number of heavy atoms in a molecule, or the isotopic load, may vary. For example, a molecule with a relatively low isotopic load may contain about 1, 2, 3, 4, 5, or 6 deuterium atoms. A molecule with a moderate isotopic load may contain about 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, or 20 deuterium atoms. In a molecule with a very high load, every hydrogen may be replaced with a deuterium. Thus, the isotopic load refers to the percentage of heavy atoms in each PUFA molecule. For example, the isotopic load may be about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100% of the number of the same type of atoms in comparison to a PUFA with no heavy atoms of the same type (e.g. hydrogen would be the "same type" as deuterium). In some embodiments, the isotopic load is at least 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%. Unintended side effects are expected to be reduced where there is high isotopic purity in a PUFA composition but low isotopic load in a given molecule. For example, the metabolic pathways will likely be less affected by using a PUFA composition with high isotopic purity but low isotopic load.

[0031] One will readily appreciate that when one of the two hydrogens of a methylene group is replaced with a deuterium atom, the resultant compound may possess a stereocenter. In some embodiments, it may be desirable to use racemic compounds. In other embodiments, it may be desirable to use enantiomerically pure compounds. In additional embodiments, it may be desirable to use diastereomerically pure compounds. In some embodiments, it may be desirable to use mixtures of compounds having enantiomeric excesses and/or diastereomeric excesses of about 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%. In other embodiments, the enantiomeric excesses and/or diastereomeric excesses is at least 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100%. In some embodiments, it may be preferable to utilize stereochemically pure enantiomers and/or diastereomers of embodiments - such as when contact with chiral molecules is being targeted for attenuating oxidative damage. However, in many circumstances, non-chiral molecules are being targeted for attenuating oxidative damage. In such circumstances, embodiments may be utilized without concern for their stereochemical purity. Moreover, in some embodiments, mixtures of enantiomers and diastereomers may be used even when the compounds are targeting chiral molecules for attenuating oxidative damage.

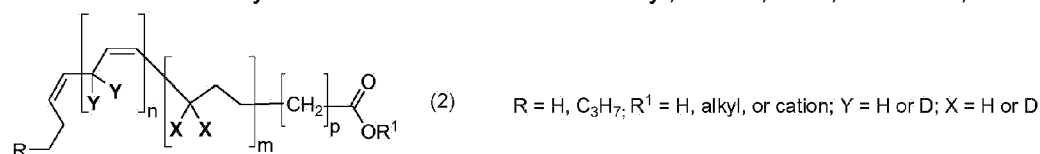
[0032] In some aspects, isotopically modified PUFAs impart an amount of heavy atoms in a particular tissue. Thus, in some aspects, the amount of heavy molecules will be a particular percentage of the same type of molecules in a tissue. For example, the number of heavy molecules may be about 1%-100% of the total amount of the same type of molecules. In some aspects, 10-50% the molecules are substituted with the same type of heavy molecules.

[0033] According to the present disclosure, a compound with the same chemical bonding structure as an essential PUFA but with a different isotopic composition at particular positions will have significantly and usefully different chemical properties from the unsubstituted compound. The particular positions with respect to oxidation, including oxidation by ROS, comprise bis-allylic positions of essential polyunsaturated fatty acids and their derivatives, as shown in **Figure 1**. The essential PUFAs isotopically reinforced at bis-allylic positions shown below will be more stable to the oxidation. Accordingly, disclosed herein are particular methods of using compounds of Formula (1) or salts thereof, whereas the sites can be further reinforced with carbon-13. R^1 = alkyl, H, or cation; m = 1-10; n = 1-5, where at each bis-allylic position, one or both Y atoms are deuterium atoms, for example,



11,11-Dideutero-cis,cis-9,12-Octadecadienoic acid (11,11-Dideutero-(9Z,12Z)-9,12-Octadecadienoic acid; D2-LA); and 11,11,14,14-Tetradeutero-cis,cis,cis-9,12,15-Octadecatrienoic acid (11,11,14,14-Tetradeutero-(9Z,12Z,15Z)-9,12,15-Octadecatrienoic acid; D4-ALA). According to the present disclosure, said positions, in addition to deuteration, can be further reinforced by carbon-13, each at levels of isotope abundance above the naturally-occurring abundance level. All other carbon-hydrogen bonds in the PUFA molecule may optionally contain deuterium and/or carbon-13 at, or above, the natural abundance level.

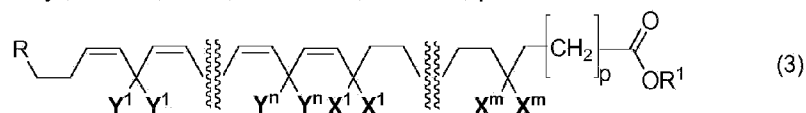
[0034] Essential PUFAs are biochemically converted into higher homologues by desaturation and elongation. Therefore, some sites which are not bis-allylic in the precursor PUFAs will become bis-allylic upon biochemical transformation. Such sites then become sensitive to oxidation, including oxidation by ROS. In a further embodiment, such pro-bis-allylic sites, in addition to existing bis-allylic sites are reinforced by isotope substitution as shown below. Accordingly, this aspect of the invention provides for the use of compounds of Formula (2) or salt thereof, where at each bis-allylic position, and at each pro-bis-allylic position, one or more of X or Y atoms may be deuterium atoms. R^1 = alkyl, cation, or H; m = 1-10; n = 1-5; p = 1-10.



[0035] Said positions, in addition to deuteration, can be further reinforced by carbon-13, each at levels of isotope abundance above the naturally-occurring abundance level. All other

carbon-hydrogen bonds in the PUFA molecule may contain optionally deuterium and/or carbon-13 at or above the natural abundance level.

[0036] Oxidation of PUFAs at different bis-allylic sites gives rise to different sets of oxidation products. For example, 4-HNE is formed from n-6 PUFAs whereas 4-HHE is formed from n-3 PUFAs (Negre-Salvayre A, et al. Brit. J. Pharmacol. 2008; 153:6-20). The products of such oxidation possess different regulatory, toxic, signaling, etc. properties. It is therefore desirable to control the relative extent of such oxidations. Accordingly, some aspects of the invention provide for the use of compounds of Formula (3), or salt thereof, differentially reinforced with heavy stable isotopes at selected bis-allylic or pro-bis-allylic positions, to control the relative yield of oxidation at different sites, as shown below, such that any of the pairs of Y^1 - Y^n and/or X^1 - X^m at the bis-allylic or pro-bis-allylic positions of PUFAs may contain deuterium atoms. R^1 = alkyl, cation, or H; $m = 1$ -10; $n = 1$ -6; $p = 1$ -10

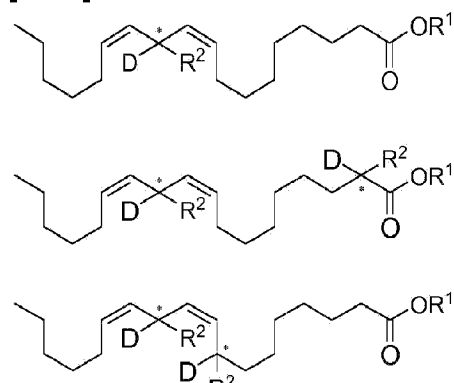


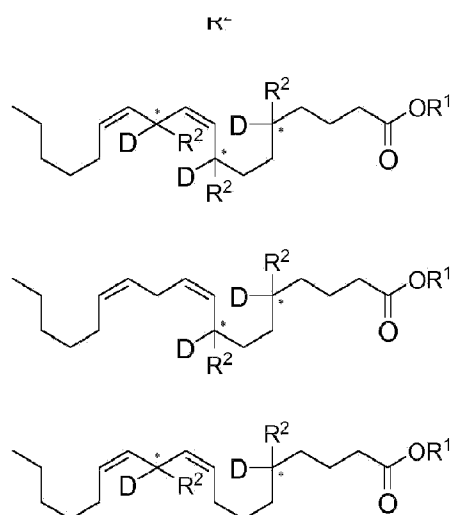
$R = H, C_3H_7$; $R^1 = H$, alkyl, or cation; $Y = H$ or D ; $X = H$ or D

[0037] Said positions, in addition to deuteration, can be further reinforced by carbon-13. All other carbon-hydrogen bonds in the PUFA molecule may contain deuterium at, or above the natural abundance level. It will be appreciated that the break lines in the structure shown above represents a PUFA with a varying number of double bonds, a varying number of total carbons, and a varying combination of isotope reinforced bis-allylic and pro-bis-allylic sites.

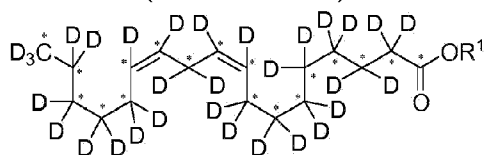
[0038] Exact structures of compounds illustrated above are shown below that provide for both isotope reinforced n-3 (omega-3) and n-6 (omega-6) essential polyunsaturated fatty acids, and the PUFAs made from them biochemically by desaturation/elongation. Any one of these compounds may be used to slow oxidation. In the following compounds, the PUFAs are isotopically reinforced at oxidation sensitive sites and/or sites that may become oxidation sensitive upon biochemical desaturation/elongation. R^1 may be H, alkyl, or cation; R^2 may be H or D; * represents either ^{12}C or ^{13}C .

[0039] D-Linoleic acids include:

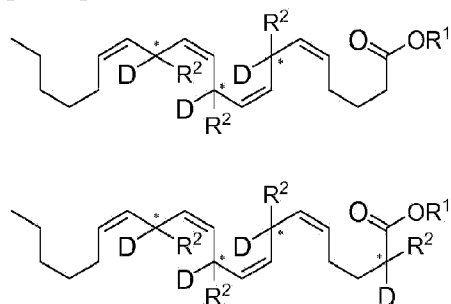




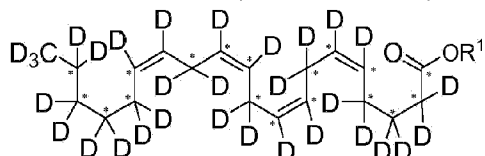
[0040] The per-deuterated linoleic acid below may be produced by microbiological methods, for example by growing in media containing deuterium (according to the invention) and/or carbon-13 (as a reference).



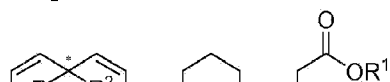
[0041] D-Arachidonic acids include:

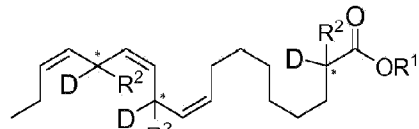
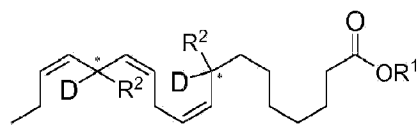
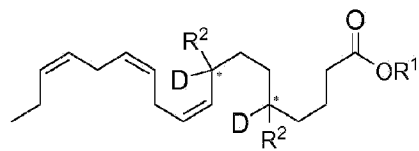
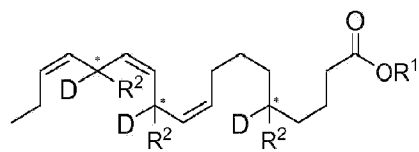
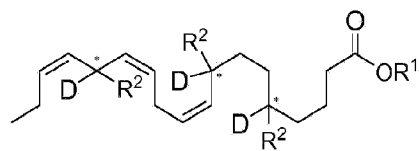
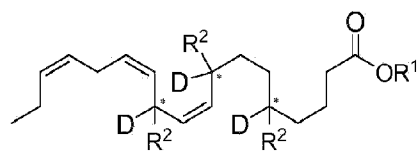
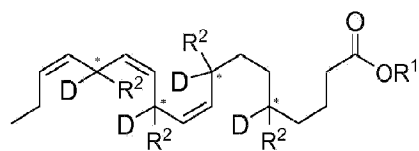
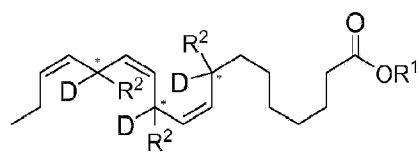
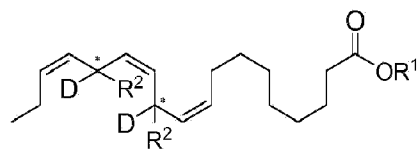
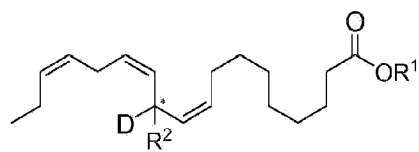
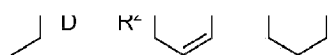


[0042] The per-deuterated arachidonic acid below may be produced by microbiological methods, such as by growing in media containing deuterium (according to the invention) and/or carbon-13 (as a reference).



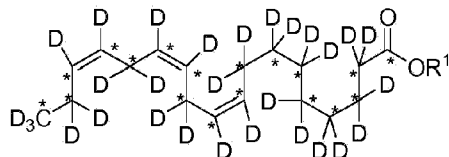
[0043] D-Linolenic acids include:





R⁴

[0044] Per-deuterated linolenic acid below may be produced by microbiological methods, such as growing in media containing deuterium (according to the invention) and/or carbon-13 (as a reference).



[0045] In some aspects of the invention, any PUFAs, whether essential or not, that are capable of being taken up from diet and used in the body, can be utilized. In the case of essential or non-essential PUFAs or precursors, the supplemented stabilized materials can compete with other dietary uptake and bio-manufacture to reduce the available disease-causing species concentrations.

[0046] In some aspects of the invention, the PUFAs isotopically reinforced at oxidation sensitive positions as described by way of the structures above are heavy isotope enriched at said positions as compared to the natural abundance of the appropriate isotope, deuterium and/or carbon-13.

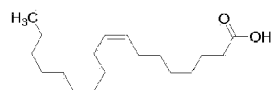
[0047] In some embodiments, the disclosed compounds are enriched to 99% isotope purity or more. In some embodiments, the heavy isotope enrichment at said positions is between 50%-99% deuterium.

[0048] In some embodiments, the modified fatty acids, when dosed via diet as drugs or supplements, may be dosed as non-toxic and pharmaceutically suitable esters of the parent fatty acid, such as an ethyl ester or glyceryl ester. This ester assists in tolerance of the drug in the gut, assists in digestion, and relies on the high levels of esterases in the intestines to de-esterify the ester into the active acid form of the drug which adsorbs. Hence, , the invention encompasses the esters of the modified fatty acids herein. Examples of this type of drug in the market, nutrition, and clinical trials literature, including Glaxo's Lovaza, (mixtures of omega 3 fatty acid esters, EPA, DHA, and alpha-linolenic acid), Abbott's Omacor (omega-3-fatty acid esters), and most fish oil supplements (DHA and EPA esters). In some aspects, incorporation of the fatty acid ester into tissues or cells refers to the incorporation of the modified parent PUFA as it would be used as a bodily constituent.

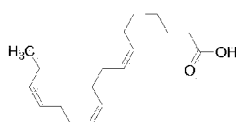
[0049] According to the present disclosure, stabilized compositions may mimic natural occurring fatty acids without changing their elemental composition. For example, the substituent may retain the chemical valence shell. Some embodiments include naturally occurring fatty acids or fatty acid esters, that are modified chemically to be effective at preventing specific disease mechanisms, but are modified in a way (such as isotopic substitution) that does not change the elemental composition of the material. For example,

deuterium is a form of the same element hydrogen. In some aspects, these compounds maintain elemental composition and are stabilized against oxidation. Some compounds that are stabilized against oxidation are stabilized at oxidation sensitive loci. Some compounds are stabilized against oxidation via heavy isotope substitution, then at bis-allylic carbon hydrogen bonds, etc.

[0050] According to the present disclosure, oxidation-prone bis-allylic sites of PUFAs can be protected against hydrogen abstraction by moving bis-allylic hydrogen-activating double bonds further apart, thus eliminating the bis-allylic positions while retaining certain PUFA fluidity as shown below. These PUFA mimetics have no bis-allylic positions.



Octadeca-8,12-dienoic acid

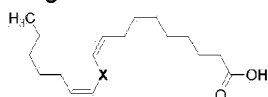


Octadeca-7,11,15-trienoic acid



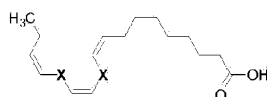
R = H, C₃H₇; R¹ = H; alkyl; n = 1-4; m = 1-12

[0051] According to the present disclosure, oxidation-prone bis-allylic sites of PUFAs can be protected against hydrogen abstraction by using heteroatoms with valence II, thus eliminating the bis-allylic hydrogens as shown below. These PUFA mimetics also have no bis-allylic hydrogens.



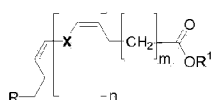
X = S: 10-Hept-1-enylsulfanyl-dec-9-enoic acid

X = O: 10-Hept-1-enyloxy-dec-9-enoic acid



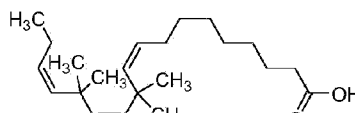
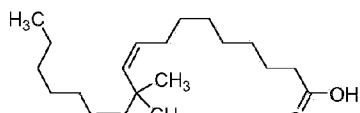
X = S: 10-(2-But-1-enylsulfanyl-vinylsulfanyl)-dec-9-enoic acid

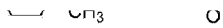
X = O: 10-(2-But-1-enyloxy-vinyloxy)-dec-9-enoic acid



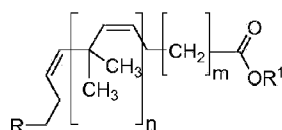
R = H, C₃H₇; R¹ = H; alkyl; X = O; S; n = 1-5; m = 1-12

[0052] Further according to the present disclosure, PUFA mimetics, i.e. compounds structurally similar to natural PUFAs but unable to get oxidized because of the structural differences, can be employed for the above mentioned purposes. Oxidation-prone bis-allylic sites of PUFAs can be protected against hydrogen abstraction by di-methylation or halogenation as shown below. The hydrogen atoms on the methyl groups may optionally be halogens, such as fluorine, or deuterium. These PUFA mimetics are dimethylated at bis-allylic sites.





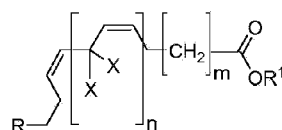
11,11-Dimethyl-octadeca-9,12-dienoic acid



R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12

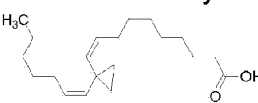


11,11,14,14-Tetramethyl-octadeca-9,12,15-trienoic acid

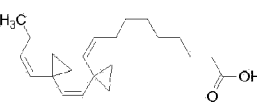


R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12; X = F, Cl, Br, or I

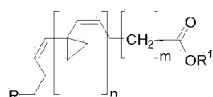
[0053] Further according to the present disclosure, oxidation-prone bis-allylic sites of PUFAs can be protected against hydrogen abstraction by alkylation as shown below. These PUFA mimetics are dialkylated at bis-allylic sites.



10-(1-Hept-1-enyl-cyclopropyl)-dec-9-enoic acid



10-{1-[2-(1-But-1-enyl-cyclopropyl)-vinyl]-cyclopropyl}-dec-9-enoic acid



R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12

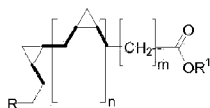
[0054] Further according to the present disclosure, cyclopropyl groups can be used instead of double bonds, thus rendering the acids certain fluidity while eliminating the bis-allylic sites as shown below. These PUFA mimetics have cyclopropyl groups instead of double bonds.



8-[2-(2-Pentyl-cyclopropylmethyl)-cyclopropyl]-octanoic acid

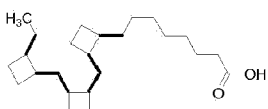
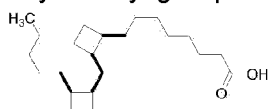


8-[2-[2-(2-Ethyl-cyclopropylmethyl)-cyclopropylmethyl]-cyclopropyl]-octanoic acid



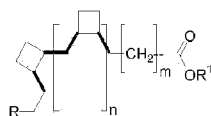
R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12

[0055] Further according to the present disclosure, 1,2-substituted cyclobutyl groups in appropriate conformation can be used instead of double bonds, thus rendering the acids certain fluidity while eliminating the bis-allylic sites as shown below. These PUFA mimetics have 1,2-cyclobutyl groups instead of double bonds.



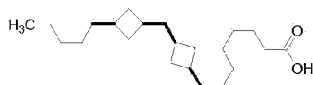
8-[2-(2-Pentyl-cyclobutylmethyl)-cyclobutyl]-octanoic acid

8-[2-[2-(2-Ethyl-cyclobutylmethyl)-cyclobutylmethyl]-cyclobutyl]-octanoic acid

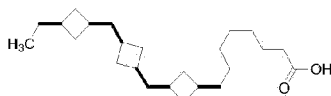


R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12

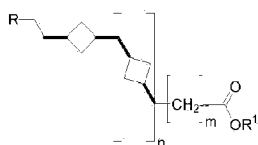
[0056] In a modification of the previous disclosure of mimetics with 1,2-cyclobutyl groups instead of double bonds, 1,3-substituted cyclobutyl groups in appropriate conformation can be used instead of double bonds, thus rendering the acids certain fluidity while eliminating the bis-allylic sites. The following PUFA mimetics have 1,3-cyclobutyl groups instead of double bonds.



8-[3-(3-Pentyl-cyclobutylmethyl)-cyclobutyl]-octanoic acid



8-[3-[3-(3-Ethyl-cyclobutylmethyl)-cyclobutylmethyl]-cyclobutyl]-octanoic acid



R = H, C₃H₇; R¹ = H; alkyl; n = 1-5; m = 1-12

[0057] It is a well known principle in medicinal chemistry that certain functional groups are isosteric and/or bioisosteric with certain other functional groups. Bioisosteres are substituents or groups with similar physical or chemical properties which produce broadly similar biological properties to a chemical compound. For example, well known isosteres and/or bioisosteres for hydrogen include halogens such as fluorine; isosteres and/or bioisosteres of alkenes include alkynes, phenyl rings, cyclopropyl rings, cyclobutyl rings, cyclopentyl rings, cyclohexyl rings, thioethers, and the like; isosteres and/or bioisosteres of carbonyls include sulfoxides, sulfones, thiocarbonyls, and the like; isosteres and/or bioisosteres of esters include amides, sulfonic acid esters, sulfonamides, sulfinyl acid esters, sulfinylamides, and the like. Consequently, PUFA mimetics also include compounds having isosteric and/or bioisosteric functional groups.

[0058] It is contemplated that it may be useful to formulate PUFAs as a pro-drug in the form of an ester, for use in the invention. A pro-drug is a pharmacological substance may itself have biological activity, but upon administration the pro-drug is metabolized into a form that also exerts biological activity. Many different types of pro-drugs are known and they can be classified into two major types based upon their cellular sites of metabolism. Type I pro-drugs are those that are metabolized intracellularly, while Type II are those that are metabolized extracellularly. It is well-known that carboxylic acids may be converted to esters and various other functional groups to enhance pharmacokinetics such as absorption, distribution, metabolism, and excretion. Esters are a well-known pro-drug form of carboxylic acids formed

by the condensation of an alcohol (or its chemical equivalent) with a carboxylic acid (or its chemical equivalent). According to the present disclosure, alcohols (or their chemical equivalent) for incorporation into esters of PUFAs include pharmaceutically acceptable alcohols or chemicals that upon metabolism yield pharmaceutically acceptable alcohols. Such alcohols include, but are not limited to, propylene glycol, ethanol, isopropanol, 2-(2-ethoxyethoxy)ethanol (Transcutol®, Gattefosse, Westwood, N.J. 07675), benzyl alcohol, glycerol, polyethylene glycol 200, polyethylene glycol 300, or polyethylene glycol 400; polyoxyethylene castor oil derivatives (for example, polyoxyethyleneglyceroltricitrate or polyoxyl 35 castor oil (Cremophor®EL, BASF Corp.), polyoxyethyleneglycerol oxystearate (Cremophor®RH 40 (polyethyleneglycol 40 hydrogenated castor oil) or Cremophor®RH 60 (polyethyleneglycol 60 hydrogenated castor oil), BASF Corp.)); saturated polyglycolized glycerides (for example, Gelucire® 35/10, Gelucire® 44/14, Gelucire® 46/07, Gelucire® 50/13 or Gelucire® 53/10, available from Gattefosse, Westwood, N.J. 07675); polyoxyethylene alkyl ethers (for example, cetomacrogol 1000); polyoxyethylene stearates (for example, PEG-6 stearate, PEG-8 stearate, polyoxyl 40 stearate NF, polyoxyethyl 50 stearate NF, PEG-12 stearate, PEG-20 stearate, PEG-100 stearate, PEG-12 distearate, PEG-32 distearate, or PEG-150 distearate); ethyl oleate, isopropyl palmitate, isopropyl myristate; dimethyl isosorbide; N-methylpyrrolidinone; paraffin; cholesterol; lecithin; suppository bases; pharmaceutically acceptable waxes (for example, carnauba wax, yellow wax, white wax, microcrystalline wax, or emulsifying wax); pharmaceutically acceptable silicon fluids; sorbitan fatty acid esters (including sorbitan laurate, sorbitan oleate, sorbitan palmitate, or sorbitan stearate); pharmaceutically acceptable saturated fats or pharmaceutically acceptable saturated oils (for example, hydrogenated castor oil (glyceryl-tris-12-hydroxystearate), cetyl esters wax (a mixture of primarily C14-C18 saturated esters of C14-C18 saturated fatty acids having a melting range of about 43°-47° C), or glyceryl monostearate).

[0059] In some embodiments, the fatty acid pro-drug is represented by the ester P-B, wherein the radical P is a PUFA and the radical B is a biologically acceptable molecule. Thus, cleavage of the ester P-B affords a PUFA and a biologically acceptable molecule. Such cleavage may be induced by acids, bases, oxidizing agents, and/or reducing agents. Examples of biologically acceptable molecules include, but are not limited to, nutritional materials, peptides, amino acids, proteins, carbohydrates (including monosaccharides, disaccharides, polysaccharides, glycosaminoglycans, and oligosaccharides), nucleotides, nucleosides, lipids (including mono-, di- and tri-substituted glycerols, glycerophospholipids, sphingolipids, and steroids).

[0060] According to the present disclosure, alcohols (or their chemical equivalent) for incorporation into pro-drugs of PUFAs include alcohols with 1 to 50 carbon atoms ("C₁₋₅₀ alcohols"), C₁₋₄₅ alcohols, C₁₋₄₀ alcohols, C₁₋₃₅ alcohols, C₁₋₃₀ alcohols, C₁₋₂₅ alcohols, C₁₋₂₀ alcohols, C₁₋₁₅ alcohols, C₁₋₁₀ alcohols, C₁₋₆ alcohols (whenever it appears herein, a numerical range such as "1-50" refers to each integer in the given range; e.g., "1 to 50 carbon atoms" means that the alkyl group may consist of 1 carbon atom, 2 carbon atoms, 3 carbon atoms, etc., up to and including 50 carbon atoms, although the present definition also covers the occurrence of the term "alkyl" where no numerical range is designated). Such alcohols may be branched, unbranched, saturated, unsaturated, polyunsaturated and/or include one or more

heteroatoms such as nitrogen, oxygen, sulfur, phosphorus, boron, silicone, fluorine, chlorine, bromine, or iodine. Exemplary alcohols include methyl, ethyl, propyl, iso-propyl, n-butyl, isobutyl, sec-butyl, tert-butyl, pentyl, hexyl, perfluoromethyl, perchloromethyl, perfluoro-tert-butyl, perchloro-tert-butyl, and benzyl alcohols as well as ether alcohols such as polyethylene glycols. In some embodiments, the alcohol contains a charged species. Such species may be anionic or cationic. According to the present disclosure, the species may be a positively charged phosphorus atom. Alternatively, the positively charged phosphorus atom may be a phosphonium cation. Alternatively the charged species may be a primary, secondary, tertiary, or quaternary ammonium cation.

[0061] According to the present disclosure, alcohols (or their chemical equivalent) for incorporation into pro-drugs of PUFAs include polyalcohols such as diols, triols, tetra-ols, pentaols, etc. Examples of polyalcohols include ethylene glycol, propylene glycol, 1,3-butylene glycol, polyethylene glycol, methylpropanediol, ethoxydiglycol, hexylene glycol, dipropylene glycol glycerol, and carbohydrates. Esters formed from polyalcohols and PUFAs may be mono-esters, di-esters, tri-esters, etc. According to the present disclosure, multiply esterified polyalcohols may be esterified with the same PUFAs. Alternatively, multiply esterified polyalcohols may be esterified with different PUFAs. Alternatively, the different PUFAs may be stabilized in the same manner. Alternatively, the different PUFAs may be stabilized in different manners (such as deuterium substitution in one PUFA and ^{13}C substitution in another PUFA). In some embodiments, one or more PUFAs is an omega-3 fatty acid and one or more PUFAs is an omega-6 fatty acid.

[0062] It is also contemplated that it may be useful to formulate PUFAs and/or PUFA mimetics and/or PUFA esters as salts for use in the present disclosure. For example, the use of salt formation as a means of tailoring the properties of pharmaceutical compounds is well known. See Stahl et al., *Handbook of pharmaceutical salts: Properties, selection and use* (2002) Weinheim/Zurich: Wiley-VCH/VHCA; Gould, Salt selection for basic drugs, *Int. J. Pharm.* (1986), 33:201-217. Salt formation can be used to increase or decrease solubility, to improve stability or toxicity, and to reduce hygroscopicity of a drug product.

[0063] Formulation of PUFAs and/or PUFA mimetics and/or PUFA pro-drugs as salts includes, but is not limited to, the use of basic inorganic salt forming agents, basic organic salt forming agents, and salt forming agents containing both acidic and basic functional groups. Various useful inorganic bases for forming salts include, but are not limited to, alkali metal salts such as salts of lithium, sodium, potassium rubidium, cesium, and francium, and alkaline earth metal salts such as beryllium, magnesium, calcium, strontium, barium, and radium, and metals such as aluminum. These inorganic bases may further include counterions such as carbonates, hydrogen carbonates, sulfates, hydrogen sulfates, sulfites, hydrogen sulfites, phosphates, hydrogen phosphates, dihydrogen phosphates, phosphites, hydrogen phosphites, hydroxides, oxides, sulfides, alkoxides such as methoxide, ethoxide, and t-butoxide, and the like. Various useful organic bases for forming salts include, but are not limited to, amino acids, basic amino acids such as arginine, lysine, ornithine and the like, ammonia, alkylamines such as methylamine, ethylamine, dimethylamine, diethylamine, trimethylamine, triethylamine and the

like, heterocyclic amines such as pyridine, picoline and the like, alkanolamines such as ethanolamine, diethanolamine, triethanolamine and the like, diethylaminoethanol, dimethylaminoethanol, N-methylglucamine, dicyclohexylamine, N,N'-dibenzylethylenediamine, ethylenediamine, piperazine, choline, trolamine, imidazole, diolamine, betaine, tromethamine, meglumine, chlorprocain, procaine, and the like.

[0064] Salt formulations of PUFAs and/or PUFA mimetics and/or PUFA pro-drugs include, but are not limited to, pharmaceutically acceptable basic inorganic salts, basic organic salts, and/or organic compounds having both acidic and basic functional groups. Pharmaceutically acceptable salts are well known in the art and include many of the above-recited inorganic and organic bases. Pharmaceutically acceptable salts further include salts and salt-forming agents found in drugs approved by the Food and Drug Administration and foreign regulatory agencies. Pharmaceutically acceptable organic cations for incorporation include, but are not limited to, benzathine, chlorprocaine, choline, diethanolamine, ethylenediamine, meglumine, procaine, benethamine, clemizole, diethylamine, piperazine, and tromethamine. Pharmaceutically acceptable metallic cations for incorporation include, but are not limited to, aluminum, calcium, lithium, magnesium, potassium, sodium, zinc, barium, and bismuth. Additional salt-forming agents include, but are not limited to, arginine, betaine, carnitine, diethylamine, L-glutamine, 2-(4-imidazolyl)ethylamine, isobutanolamine, lysine, N-methylpiperazine, morpholine, and theobromine.

[0065] Moreover, several lists of pharmaceutically approved counterions exists. See Bighley et al., Salt forms of drugs and absorption. 1996 In: Swarbrick J. et al. eds. Encyclopaedia of pharmaceutical technology, Vol. 13 New York: Marcel Dekker, Inc. pp 453-499; Gould, P.L., Int. J. Pharm. 1986, 33, 201-217; Berge, J. Pharm. Sci. 1977, 66, 1-19; Heinrich Stahl P., Wermuch C.G. (editors), Handbook of Pharmaceutical Salts, IUPAC, 2002; Stahl et al., Handbook of pharmaceutical salts: Properties, selection and use (2002) Weinheim/Zurich: Wiley-VCH/VHCA.

[0066] It may be unnecessary to substitute all isotopically unmodified PUFAs, such as non-deuterated PUFAs, with isotopically modified PUFAs such as deuterated PUFAs. It is preferable to have sufficient isotopically modified PUFAs such as D-PUFAs in the membrane to prevent unmodified PUFAs such as H-PUFAs from sustaining a chain reaction of self-oxidation. During self-oxidation, when one PUFA oxidizes, and there is a non-oxidized PUFA in the vicinity, the non-oxidized PUFA can get oxidized by the oxidized PUFA. This may also be referred to as autooxidation. In some instances, if there is a low concentration, for example "dilute" H-PUFAs in the membrane with D-PUFAs, this oxidation cycle may be broken due to the distance separating H-PUFAs. According to the present disclosure, the concentration of isotopically modified PUFAs is present in a sufficient amount to maintain autooxidation chain reaction. To break the autooxidation chain reaction, for example, 1-60%, 5-50%, or 15-35% of the total molecules of the same type are in the membrane. This may be measured by IRMS (isotope ratio mass spectrometry).

[0067] A further aspect of the present disclosure provides a dietary, supplementary or pharmaceutical composition of the active compounds. According to the present disclosure, the

dietary, supplementary, or pharmaceutical composition may comprise a salt of the active compound.

[0068] Various useful inorganic bases for forming salts include, but are not limited to, alkali metal salts such as salts of lithium, sodium, potassium rubidium, cesium, and francium, and alkaline earth metal salts such as beryllium, magnesium, calcium, strontium, barium, and radium, and metals such as aluminum. These inorganic bases may further include counterions such as carbonates, hydrogen carbonates, sulfates, hydrogen sulfates, sulfites, hydrogen sulfites, phosphates, hydrogen phosphates, dihydrogen phosphates, phosphites, hydrogen phosphites, hydroxides, oxides, sulfides, alkoxides such as methoxide, ethoxide, and t-butoxide, and the like.

[0069] Various useful organic bases for forming salts include, but are not limited to, amino acids; basic amino acids such as arginine, lysine, ornithine and the like; ammonia; ammonium hydroxide; alkylamines such as methylamine, ethylamine, dimethylamine, diethylamine, trimethylamine, triethylamine and the like; heterocyclic amines such as pyridine, picoline and the like; alkanolamines such as ethanolamine, diethanolamine, triethanolamine and the like, diethylaminoethanol, dimethylaminoethanol; N-methylglucamine; dicyclohexylamine; N,N'-dibenzylethylenediamine; ethylenediamine; piperazine; choline; trolamine; imidazole; diolamine; betaine; tromethamine; meglumine; chloroprocain; procaine; and the like.

[0070] Salts of active compounds may include, but are not limited to, pharmaceutically acceptable salts. Pharmaceutically acceptable salts are well known in the art and include many of the above-listed salt-forming agents. Pharmaceutically acceptable salts further include salts and salt-forming agents of the type present in drugs approved by the Food and Drug Administration and foreign regulatory agencies.

[0071] Pharmaceutically acceptable organic cations for incorporation into a salt of an active compound include, but are not limited to, benzathine, chloroprocaine, choline, diethanolamine, ethylenediamine, meglumine, procaine, benethamine, clemizole, diethylamine, piperazine, and tromethamine.

[0072] Pharmaceutically acceptable metallic cations for incorporation into a salt of an active compound include, but are not limited to, aluminum, calcium, lithium, magnesium, potassium, sodium, zinc, barium, and bismuth.

[0073] Additional salt-forming agents having potential usefulness as forming salts include, but are not limited to, acetylaminocetic acid, N-acetyl-L-asparagine, N-acetylcystine, arginine, betaine, carnitine, L-glutamine, 2-(4-imidazolyl)ethylamine, isobutanolamine, lysine, N-methylpiperazine, and morpholine.

[0074] Moreover, several lists of pharmaceutically approved counterions exists. See Bighley et al., Salt forms of drugs and absorption. 1996 In: Swarbrick J. et al. eds. Encyclopaedia of pharmaceutical technology, Vol. 13 New York: Marcel Dekker, Inc. pp 453-499; Gould, P.L., Int.

J. Pharm. 1986, 33, 201-217; Berge, J. Pharm. Sci. 1977, 66, 1-19; Heinrich Stahl P., Wermuch C.G. (editors), Handbook of Pharmaceutical Salts, IUPAC, 2002; Stahl et al., Handbook of pharmaceutical salts: Properties, selection and use (2002) Weinheim/Zurich: Wiley-VCH/VHCA.

Co-administration

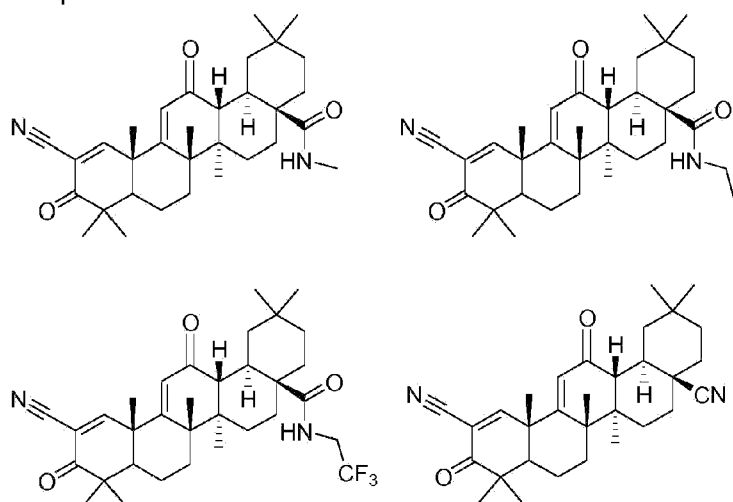
[0075] In some embodiments, the compounds disclosed herein are administered in combination. For example, in some embodiments, two, three, four, and/or five or more stabilized compounds are administered together. In some embodiments, stabilized compounds are administered in approximately similar amounts. In other embodiments, stabilized compounds are administered in differing amounts. For example, any one of two or more compounds in a mixture may represent about 1% to about 99% of a mixture, about 5% to about 95% of a mixture, about 10% to about 90% of a mixture, about 15% to about 85% of a mixture, about 20% to about 80% of a mixture, about 25% to about 75% of a mixture, about 30% to about 70% of a mixture, about 35% to about 65% of a mixture, about 40% to about 60% of a mixture, about 40% to about 60% of a mixture, about 45% to about 55% of a mixture, and/or about 50% of a mixture. In other embodiments, any one of two or more compounds in a mixture may represent about: 5%, 10%, 15%, 20%, 25%, 30%, 35%, 40%, 45%, 50%, 65%, 60%, 65%, 70%, 75%, 80%, 85%, 90%, 95%, or 100% of a mixture.

[0076] Although antioxidants cannot cancel the negative effects of PUFA peroxidation due to the stochastic nature of the process and the stability of PUFA peroxidation products (reactive carbonyls) to antioxidant treatment, co-administration of antioxidants with compositions resistant to oxidation, such as those described herein, may prove beneficial for treating oxidative stress-related disorders.

[0077] Certain antioxidants contemplated as useful for co-administration include the following: vitamins, such as vitamin C and vitamin E; glutathione, lipoic acid, uric acid, carotenes, lycopene, lutein, anthocyanins, oxalic acid, phytic acid, tannins, coenzyme Q, melatonin, tocopherols, tocotrienols, polyphenols including resveratrol, flavonoids, selenium, eugenol, idebenone, mitoquinone, mitoquinol, ubiquinone, Szeto-Schiller peptides, and mitochondrial-targeted antioxidants. When not explicitly mentioned, quinone derivatives of the aforementioned antioxidants are also contemplated as useful for co-administration within the present disclosure. According to one embodiment the antioxidant is Coenzyme Q, idebenone, mitoquinone, mitoquinol, vitamin C, or vitamin E.

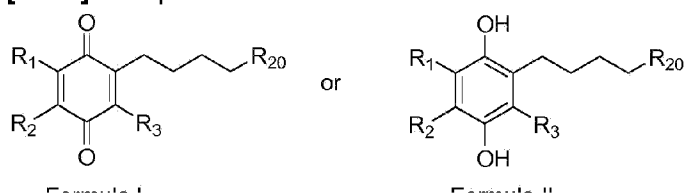
[0078] According to the present disclosure, stabilized compounds are administered with compounds that upregulate antioxidant genes. Further according to the present disclosure, stabilized compounds are administered with compounds that affect signaling pathways, such as the Keap1/Nrf2/ARE signaling pathway, thereby resulting in the production of anti-inflammatory and/or antioxidant proteins, such as heme oxygenase-1 (HO-1). According to the present disclosure, stabilized compounds are administered with antioxidant inflammation modulators. Antioxidant inflammation modulators suppress pro-oxidant and/or pro-

inflammatory transcription factors. According to the present disclosure, antioxidant inflammation modulators are activators of the transcription factor Nrf2. Nrf2 activation promotes the antioxidant, detoxification, and anti-inflammatory genes upregulation. Further according to the present disclosure, antioxidant inflammation modulators suppress NF- κ B. According to the present disclosure, antioxidant inflammation modulators suppress STAT3. Further according to the present disclosure, stabilized compounds are administered with compounds that affect histone deacetylase activity. According to the present disclosure, stabilized compounds are administered with compounds that bind to antioxidant response elements (ARE). Further according to the present disclosure, stabilized compounds are administered with bardoxolone methyl (2-cyano-3,12-dioxooleane-1,9(11)-dien-28-oic acid methyl ester) as the antioxidant inflammation modulator. According to the present disclosure, the antioxidant inflammation modulator is 2-cyano-3,12-dioxooleane-1,9(11)-dien-28-oic acid, or a pharmaceutically acceptable ester thereof. In other embodiments, the antioxidant inflammation modulator is an amide of 2-cyano-3,12-dioxooleane-1,9(11)-dien-28-oic acid. According to the present disclosure, the antioxidant inflammation modulator is a triterpenoid. In other embodiments, the antioxidant inflammation modulator is selected from the following compounds:



[0079] Additional antioxidants believed to be useful in co-administration therapies include those compounds disclosed in U.S. Patent Nos. 6,331,532; 7,179,928; 7,232,809; 7,888,334; 7,888,335; 7,432,305; 7,470,798; and 7,514,461; and U.S. Patent Application Nos. 20020052342; 20030069208; 20040106579; 20050043553; 20050245487; 20060229278; 20070238709; 20070270381; 20080161267; 20080275005; 20090258841; 20100029706; and 20110046219;. These compounds are mitochondrially-targeted compounds and include, but are not limited to:

[0080] Compounds of Formulas I or II



Formula I

Formula II

wherein R_1 and R_2 are independently selected from $-C_1-C_4$ alkyl, $-C_1-C_4$ haloalkyl, $-CN$, $-F$, $-Cl$, $-Br$, and $-I$; R_3 is selected from $-C_1-C_4$ alkyl, $-C_1-C_4$ haloalkyl, $-CN$, $-F$, $-Cl$, and $-I$, and R_{20} is independently selected from $-C_1-C_{20}$ alkyl, $-C_1-C_{20}$ alkenyl, $-C_1-C_{20}$ alkynyl, and $-C_1-C_{20}$ containing at least one double bond and at least one triple bond.

[0081] Compounds such as: 3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-2-yl)-propionic acid methyl ester; 3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chroman-2-yl)-propionic acid; 2,2-Dimethyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; 3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-propano-2H-benzo[h]chromen-2-yl)-propionic acid methyl ester; 2-Methyl-2-[3-(thiazol-2-ylsulfanyl)-propyl]-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; [3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-2-yl)-propyl]-phosphonic acid dimethyl ester; [3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-2-yl)-propyl]-phosphonic acid; 3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-2-yl)-propionic acid methyl ester; 4-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-2-yl)-butane-1-sulfonic acid dimethylamide; 2-(3-Hydroxy-propyl)-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; 2-(3-Chloro-propyl)-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; 2,2-Dimethyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; -(2-Chloro-ethyl)-2-methyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; 2-Methyl-2-thiazol-2-yl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-6-ol; 2,2-Dimethyl-3,4,7,8,9,10-hexahydro-7,10-ethano-2H-benzo[h]chromen-6-ol; 3-(6-Hydroxy-2-methyl-3,4,7,8,9,10-hexahydro-7,10-ethano-2H-benzo[h]chromen-2-yl)-propionic acid; 2-(3-Chloro-propyl)-2-methyl-3,4,7,8,9,10-hexahydro-7,10-ethano-2H-benzo[h]chromen-6-ol; 4-(6-Hydroxy-2,2-dimethyl-3,4,7,8,9,10-hexahydro-7,10-methano-2H-benzo[h]chromen-5-ylmethylene)-2-methyl-5-propyl-2,4-dihydropyrazol-3-one.

[0082] Compounds such as: 2,2,7,8-Tetramethyl-5-phenyl-chroman-6-ol; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-benzoic acid methyl ester; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-benzoic acid; 2,2,7,8-Tetramethyl-5-pyridin-4-yl-chroman-6-ol; 2,2,7,8-Tetramethyl-5-pyridin-3-yl-chroman-6-ol; 5-(4-Methanesulfonyl-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 5-(4-Dimethylamino-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 5-(4-Chloro-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-benzenesulfonamide; 5-(4-Methoxy-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; (6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-ylmethyl)-1-hydroxyurea; 2,2,7,8-Tetramethyl-5-(3-nitrophenyl)-chroman-6-ol; 2,2,7,8-Tetramethyl-5-(4-trifluoromethyl-phenyl)-chroman-6-ol; 5-(4-tert-Butyl-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 2,2,7,8-Tetramethyl-5-(3,4,5-trimethoxyphenyl)-chroman-6-ol; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-benzonitrile; 5-(2,5-Dimethoxy-3,4-dimethyl-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 5-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-benzene-1,2,3-triol; 5-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl)-2,3-dimethyl-benzene-1,4-diol; 5-(2-Chloro-phenyl)-2,2,7,8-tetramethyl-chroman-6-ol; 5-Furan-2-yl-2,2,7,8-tetramethyl-chroman-6-ol; 5-Allylsulfanylmethyl-2,2,8-trimethyl-7-(3-methylbutyl)-

chroman-6-ol; 5-Cyclopentylsulfanylmethyl-2,2,7,8-tetramethyl-chroman-6-ol; 5-Hexylsulfanylmethyl-2,2,7,8-tetramethyl-chroman-6-ol; 5-Allylsulfanylmethyl-2,2,7,8-tetramethyl-chroman-6-ol; 5-(4,6-Dimethyl-pyrimidin-2-ylsulfanylmethyl)-2,2,7,8-tetramethyl-chroman-6-ol; 1-[3-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl-methylsulfanyl)-2-methylpropionyl]-pyrrolidine-2-carboxylic acid; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-ylmethylene)-5-methyl-2-phenyl-2,4-dihydro-pyrazol-3-one; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl-methylene)-3-phenyl-4H-isoxazol-5-one; 4-[4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl-methylene)-3-methyl-5-oxo-4,5-dihydro-pyrazol-1-yl]-benzoic acid; 4-(6-Hydroxy-2,2,7,8-tetramethyl-chroman-5-yl-methylene)-2-methyl-5-propyl-2,4-dihydro-pyrazol-3-one; 5-Hydroxy-3-(6-hydroxy-2,2,7,8-tetramethyl-chroman-5-yl-methylene)-3H-benzofuran-2-one; 2,5,7,8-Tetramethyl-2-thiophen-2-yl-chroman-6-ol; 2-(2,5-Dimethyl-thiophen-3-yl)-2,5,7,8-tetramethyl-chroman-6-ol; 2-(2,5-Dimethyl-thiophen-3-yl)-2,7,8-trimethyl-chroman-6-ol; 8-Chloro-2-(2,5-dimethyl-thiophen-3-yl)-2,5,7-trimethyl-chroman-6-ol; 5-Chloro-2,7,8-trimethyl-2-thiophen-2-yl-chroman-6-ol; 5-[3-(6-Methoxymethoxy-2,7,8-trimethyl-chroman-2-yl)-propylidene]-thiazolidine-2,4-dione; 5-[3-(6-Hydroxy-2,7,8-trimethyl-chroman-2-yl)-propylidene]-thiazolidine-2,4-dione; 3-[6-Hydroxy-2,7,8-trimethyl-2-(4,8,12-trimethyl-tridecyl)-chroman-5-yl-methylsulfanyl]-2-methyl-propionic acid; 2,7,8-Trimethyl-5-(5-methyl-1H-benzimidazol-2-yl-sulfanylmethyl)-2-(4,8,12-trimethyl-tridecyl)-chroman-6-ol; 2-[6-Hydroxy-2,7,8-trimethyl-2-(4,8,12-trimethyl-tridecyl)-chroman-5-ylmethylsulfanyl]-ethanesulfonic acid; 5-(4,6-Dimethyl-pyrimidin-2-ylsulfanylmethyl)-2,7,8-trimethyl-2-(4,8,12-trimethyl-tridecyl)-chroman-6-ol; 4-[2-(4,8-Dimethyl-tridecyl)-6-hydroxy-2,7,8-trimethyl-chroman-5-ylmethylsulfanyl]-benzoic acid; 1-[3-[6-Hydroxy-2,7,8-trimethyl-2-(4,8,12-trimethyl-tridecyl)-chroman-5-ylmethylsulfanyl]-2-methyl-propionyl]-pyrrolidine-2-carboxylic acid; 2-(2,2-Dichlorovinyl)-2,5,7,8-tetramethyl-chroman-6-ol; 2-(2,2-Dibromo-vinyl)-2,5,7,8-tetramethyl-chroman-6-ol; 5-(5-Chloro-3-methyl-pent-2-enyl)-2,2,7,8-tetramethyl-chroman-6-ol; 5-Chloro-2-(2,5-dimethyl-thiophen-3-yl)-2,7,8-trimethyl-chroman-6-ol; 2-(3-Chloro-propyl)-5,7-dimethyl-2-thiophen-2-yl-chroman-6-ol; 5-Chloro-2-(2,5-dimethyl-thiazol-4-yl)-2,7,8-trimethyl-chroman-6-ol; 5-Chloro-2-(2,5-dimethyl-thiazol-4-yl)-2,7,8-trimethyl-2H-chromen-6-ol; and 5-Chloro-2-(2,5-dimethyl-thiazol-4-yl)-2,7,8-trimethyl-chroman-6-ol.

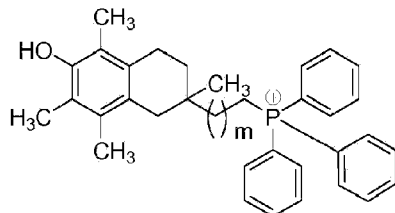
[0083] Compounds such as: dimebolin (2,8-dimethyl-5-(2-(6-methylpyridin-3-yl)ethyl)-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indole), 8-chloro-2-methyl-5-(2-(6-methylpyridin-3-yl)ethyl)-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indole, mebhydroline (5-benzyl-2-methyl-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indole), 2,8-dimethyl-1,3,4,4a,5,9b-hexahydro-1H-pyrido[4,3-b]indole, 8-fluoro-2-(3-(pyridin-3-yl)propyl)-2,3,4,5-tetrahydro-1H-pyrido[4,3-b]indole, and 8-methyl-1,3,4,4a,5,9b-tetrahydro-1H-pyrido[4,3-b]indole.

[0084] Compounds such as: 2-(3-hydroxy-3-methylbutyl)-3,5-dimethyl-6-(4-(trifluoromethyl)phenyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-6-(4-methoxyphenyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 4-(5-(3-hydroxy-3-methylbutyl)-2,4-dimethyl-3,6-dioxocyclohexa-1,4-dienyl)benzonitrile; 2-(3-hydroxy-3-methylbutyl)-3,5-dimethyl-6-(naphthalen-2-yl)cyclohexa-2,5-diene-1,4-dione; 2-(3,4-difluorophenyl)-6-(3-hydroxy-3-methylbutyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-fluorophenyl)-6-(3-hydroxy-3-methylbutyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-

chlorophenyl)-6-(3-hydroxy-3-methylbutyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2,3-dihydrobenzofuran-2-yl)-6-(3-hydroxy-3-methylbutyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-phenethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-phenylcyclohexa-2,5-diene-1,4-dione; 2-benzyl-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(3-phenylpropyl)cyclohexa-2,5-diene-1,4-dione; 2-(1-hydroxy-2-phenylethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-3-(4-methoxyphenyl)-5,6-dimethyl-cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(4-(trifluoromethyl)-phenyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(naphthalen-2-yl)cyclohexa-2,5-diene-1,4-dione; 2-(benzofuran-2-yl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-chlorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-ethylphenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(3-(trifluoromethyl)phenyl)-cyclohexa-2,5-diene-1,4-dione; 2-(4-tert-butylphenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-cyclohexa-2,5-diene-1,4-dione; 2-(4-fluorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-fluorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 4-(2-(3-hydroxy-3-methylbutyl)-4,5-dimethyl-3,6-dioxocyclohexa-1,4-dienyl)benzonitrile; 2-(3,4-difluorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-cyclohexa-2,5-diene-1,4-dione; 2-(2-fluorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-3-(3-methoxyphenyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-fluoro-2-methoxyphenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(benzo[d][1,3]dioxol-5-yl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2,4-difluorophenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-3-(4-methoxyphenyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3,5-bis(trifluoromethyl)phenyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(4-chlorophenyl)-6-(3-hydroxy-3-methylbutyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(thiazol-2-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(thiazol-5-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(pyridin-2-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(pyridazin-4-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(thiophen-2-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(thiophen-3-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(2-(furan-2-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(furan-3-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(1H-pyrazol-5-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(1H-pyrazol-4-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(1H-pyrazol-1-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(1H-imidazol-5-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(2-(1H-imidazol-2-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(oxazol-5-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(oxazol-2-

yl)ethyl)cyclohexa-2,5-diene-1,4-dione; 2-(3-hydroxy-3-methylbutyl)-5,6-dimethyl-3-(2-(oxazol-4-yl)ethyl)cyclohexa-2,5-diene-1,4-dione; and 2-(2-(1H-indol-3-yl)ethyl)-3-(3-hydroxy-3-methylbutyl)-5,6-dimethylcyclohexa-2,5-diene-1,4-dione.

[0085] Compounds such as:



wherein m is -C₁-C₂₀ alkyl, -C₁-C₂₀ alkenyl, -C₁-C₂₀ alkynyl, or -C₁-C₂₀ containing at least one double bond and at least one triple bond, and the counterion is a pharmaceutically acceptable anion.

[0086] Compounds such as: 3-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)propyl triphenylphosphonium salts; 4-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)butyl triphenylphosphonium salts; 5-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)pentyl triphenylphosphonium salts; 6-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)hexyl triphenylphosphonium salts; 7-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)heptyl triphenylphosphonium salts; 8-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)octyl triphenylphosphonium salts; 9-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)nonyl triphenylphosphonium salts; 10-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)decyl triphenylphosphonium salts; 11-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)undecyl triphenylphosphonium salts; 12-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)dodecyl triphenylphosphonium salts; 13-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)propyldecyl triphenylphosphonium salts; 14-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)butyldecyl triphenylphosphonium salts; 15-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)pentadecyl triphenylphosphonium salts; 16-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)hexadecyl triphenylphosphonium salts; 17-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)heptadecyl triphenylphosphonium salts; 18-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)octadecyl triphenylphosphonium salts; 19-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)nonadecyl triphenylphosphonium salts; 20-(4,5-dimethoxy-2-methyl-3,6-dioxo-1,4-cyclohexadien-1-yl)icosyl triphenylphosphonium salts; 3-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)propyl triphenylphosphonium salts; 4-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)butyl triphenylphosphonium salts; 5-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)pentyl triphenylphosphonium salts; 6-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)hexyl triphenylphosphonium salts; 7-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)heptyl triphenylphosphonium salts; 8-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)octyl triphenylphosphonium salts; 9-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)nonyl triphenylphosphonium salts; 10-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)decyl triphenylphosphonium salts; 11-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)undecyl triphenylphosphonium salts; 12-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)dodecyl triphenylphosphonium salts; 13-(4,5-dimethoxy-2-methyl-3,6-

dihydroxybenzyl)propyldecyl triphenylphosphonium salts; 14-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)butyldecyl triphenylphosphonium salts; 15-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)pentadecyl triphenylphosphonium salts; 16-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)hexadecyl triphenylphosphonium salts; 17-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)heptadecyl triphenylphosphonium salts; 18-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)octadecyl triphenylphosphonium salts; 19-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)nonadecyl triphenylphosphonium salts; 20-(4,5-dimethoxy-2-methyl-3,6-dihydroxyphenyl)icosyl triphenylphosphonium salts; wherein the counterion of the salt is a pharmaceutically acceptable anion such as bromide, methanesulfonate ethanesulfonate, propanesulfonate, benzenesulfonate, p-toluenesulfonate, or 2-naphthylene sulfonate.

[0087] Additionally, it is contemplated that coadministration of antioxidants could take the form of consuming foods known to have increased levels of beneficial antioxidants. Such foods include both regular foods and "superfoods" which contain antioxidants. These foods include fruits, vegetables, and other foodstuffs such as strawberries, blackcurrants, blackberries, oranges, blueberries, pomegranates, tea, coffee, olive oil, chocolate, cinnamon, herbs, red wine, grain cereals, eggs, meat, legumes, nuts, spinach, turnip, rhubarb, cacao beans, maize, beans, cabbage, and the like.

Delivery and Additional Formulations:

[0088] It is well known that triglycerides are the main constituents of vegetable oils and animal fats. It is also known that a triglyceride is an ester compound derived from glycerol and three fatty acids. Triglycerides are metabolized by enzymes such as lipases which hydrolyse ester bonds and release fatty acids and glycerol. Indeed, this metabolism releases fatty acids which can then be taken upon by cells via a fatty acid transporter protein. It is contemplated that PUFAs and PUFA mimetics that are useful in treating various diseases may be incorporated into fats such as triglycerides, diglycerides, and/or monoglycerides for administration to a patient.

[0089] The delivery of the PUFAs, PUFA mimetics, PUFA esters, and triglycerides containing PUFAs and/or PUFA mimetics could be through a modified diet. Alternatively, the PUFAs, PUFA mimetics, PUFA esters, and triglycerides containing PUFAs and/or PUFA mimetics can be administered as foods or food supplements, on their own or as complexes with 'carriers', including, but not limited to, complexes with albumin.

[0090] Other methods of delivering the reinforced PUFAs or their precursors, such as methods typically used for drug delivery and medication delivery, can also be employed. These methods include, but are not limited to, peroral delivery, topical delivery, transmucosal delivery such as nasal delivery, nasal delivery through cribriform plate, intravenous delivery, subcutaneous delivery, inhalation, or through eye drops.

[0091] Targeted delivery methods and sustained release methods, including, but not limited to,

the liposome delivery method, can also be employed.

[0092] It is contemplated that the isotopically modified compounds described herein may be administered over a course of time, in which the cells and tissues of the subject will contain increasing levels of isotopically modified compounds over the course of time in which the compounds are administered.

[0093] Compositions containing the active ingredient may be in a form suitable for oral use, for example, as tablets, troches, lozenges, aqueous or oily suspensions, oil-in-water emulsions, dispersible powders or granules, emulsions, hard or soft capsules, or syrups or elixirs. Such compositions may contain excipients such as bulking agents, solubilization agents, taste masking agents, stabilisers, colouring agents, preservatives and other agents known to those ordinarily skilled in the art of pharmaceutical formulation. In addition, oral forms may include food or food supplements containing the compounds described herein. According to the present disclosure supplements can be tailor-made so that one type of PUFA, such as omega-3 or omega-6 fatty acids can be added to food or used as a supplement depending on the dominant fat that the food or the subject's diet contains. Moreover, compositions can be tailor-made depending on the disease to be treated. For example, an LDL related condition may require more D-linoleic acid because cardiolipin, which is made of linoleic acid, is oxidized. In other embodiments, such as retinal disease and neurological/CNS conditions may require more omega-3 fatty acids such as D-linolenic acid, because D-omega-3 fatty acids are more relevant for treating these diseases. In some aspects of the disclosure, when the disease is associated with HNE, then D-omega-6 fatty acids should be prescribed, whereas for HHE, D-omega-3 fatty acids should be prescribed.

[0094] Compositions may also be suitable for delivery by topical application, as a spray, cream, ointment, lotion, or as a component or additive to a patch, bandage or wound dressing. In addition the compound can be delivered to the site of the disease by mechanical means, or targeted to the site of the disease through the use of systemic targeting technologies such as liposomes (with or without chemical modification that provides them with affinity for the diseased tissue), antibodies, aptamers, lectins, or chemical ligands such as albumin, with affinity for aspects of the diseased tissue that are less abundant or not present on normal tissue. In some aspects, topical application of cosmetics may include the use of a carrier which is an isotopically modified compound or mimetic described herein for delivering through skin such as by a patch. Eye disorders may be treated with eyedrops.

[0095] A pharmaceutical composition may also be in a form suitable for administration by injection. Such compositions may be in the form of a solution, a suspension or an emulsion. Such compositions may include stabilizing agents, antimicrobial agents or other materials to improve the function of the medicament. Some aspects of the invention also encompass dry or desiccated forms of the compounds which can readily be formed or reconstituted into a solution suspension or emulsion suitable for administration by injection, or for oral or topical use. Delivery by injection may be suitable for systemic delivery, and also local delivery such as injection into the eye for treating disorders relating to the eye.

Dosages

[0096] In the present disclosure, compounds are dosed at about 0.01 mg/kg to about 1000 mg/kg, about 0.1 mg/kg to about 100 mg/kg, and/or about 1 mg/kg to about 10 mg/kg. Further according to the present disclosure, compounds are dosed at about: 0.01, 0.1, 1.0, 5.0, 10, 25, 50, 75, 100, 150, 200, 300, 400, 500, and/or 1000 mg/kg.

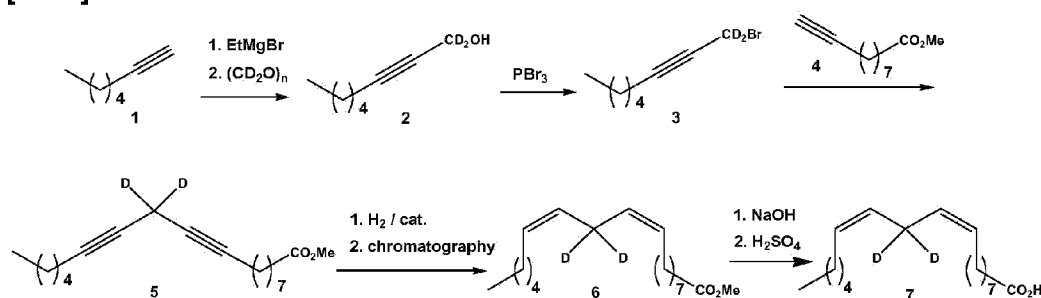
EXAMPLES

[0097] Experimental: MALDI-TOF mass-spectra were recorded on a PE-ABI Voyager Elite delayed extraction instrument. Spectra were acquired with an accelerating voltage of 25 KV and 100 ms delay in the positive ion mode. Unless otherwise specified, the ^1H NMR spectra were recorded on a Varian Gemini 200 MHz spectrometer. HPLC was carried out on a Waters system. Chemicals were from Sigma-Aldrich Chemical Company (USA), Avocado research chemicals (UK), Lancaster Synthesis Ltd (UK), and Acros Organics (Fisher Scientific, UK). Silica gel, TLC plates and solvents were from BDH/Merck. IR spectra were recorded with Vertex 70 spectrometer. ^1H and ^{13}C NMR spectra were obtained with a Bruker AC 400 instrument at 400 and 100 MHz respectively, in CDCl_3 (TMS at $\delta = 0.00$ or CHCl_3 at $\delta = 7.26$ for ^1H and CHCl_3 at $\delta = 77.0$ for ^{13}C as an internal standard).

[0098] One of ordinary skill in the art will recognize that the below described syntheses can be readily modified to prepare additional oxidation-resistant compounds. For example, one will recognize the ester of one type of stabilized compound can be cleaved to afford the corresponding carboxylic acid. Likewise, carboxylic acids can be readily converted into additional derivatives, such as esters. Additionally, one will appreciate that by varying the identity of the isotopically labeled starting materials, isotopic variants of the below described compounds may be prepared. In the below described syntheses, paraformaldehyde- d_2 is used as an isotopically labeled starting material. One will readily appreciate that the same synthetic transformations can be used with paraformaldehyde- d_1 , formaldehyde- d_1 , paraformaldehyde- d_2 , formaldehyde- d_2 , and carbon-13 labeled variants of the aforementioned compounds. Formaldehyde- d_1 is a well-characterized compound and is readily available from known sources such as formic acid- d_1 , formic acid- d_2 , and/or dichloromethane- d_1 using generally known and understood synthetic transformations. Furthermore, radioactive analogues of the compounds described herein can be prepared using tritium-containing starting materials. These compounds would be useful for determining incorporation in the cells and tissues of animals.

Example 1. Synthesis of 11,11-D₂-linoleic acid

[0099]



[0100] **1,1-Dideutero-oct-2-yn-1-ol (2)** To a solution of ethylmagnesium bromide prepared from bromoethane (100 ml), 1,2-dibromoethane (1 ml) and magnesium turnings (31.2 g) in dry THF (800 ml), heptyn-1 ((1); 170 ml) was added dropwise over 30-60 min under argon. The reaction mixture was stirred for 1 h, and then deuteroparaform (30 g) was carefully added in one portion. The reaction mixture was gently refluxed for 2 h, chilled to -10°C, and then 5-7 ml of water was slowly added. The mixture was poured into 0.5 kg slurry of crushed ice and 40 ml concentrated sulphuric acid and washed with 0.5 L of hexane. The organic phase was separated, and the remaining aqueous phase was extracted with 5:1 hexane:ethyl acetate (3 x 300 ml). The combined organic fraction was washed with sat. NaCl (1 x 50 ml), sat. NaHCO₃, (1 x 50 ml), and dried over Na₂SO₄. The solvent was evaporated *in vacuo* to yield 119.3 g (99%) of colourless oil which was used without further purification. HRMS, *m/z* calculated for C₈H₁₂D₂O: 128.1168; found: 128.1173. ¹H NMR (CDCl₃, δ): 2.18 (t, J = 7.0, 2H), 1.57 (s, 1H), 1.47 (q, J = 7.0 Hz, 2H), 1.31 (m, 4H), 0.87 (t, J = 7.0 Hz, 3H).

[0101] **1,1-Dideutero-1-bromo-oct-2-yne (3)** To a solution of (2) (3.48 g; 27.2 mmol) and pyridine (19 ml) in dry diethyl ether (300 ml), 36 ml of PBr₃ in 35 ml diethyl ether was added dropwise with stirring over 30 min at -15°C under argon. The reaction mixture was allowed to gradually warm up to r.t. and then refluxed 3 h with stirring and 1 h without stirring. The reaction mixture was then cooled down to -10°C and 500 ml of cold water was added. When the residue dissolved, saturated NaCl (250 ml) and hexane (250 ml) were added, and the organic layer was separated. The aqueous fraction was washed with hexane (2 x 100 ml), and the combined organic fractions were washed with NaCl (2 x 100 ml) and dried over Na₂SO₄ in presence of traces of hydroquinone and triethylamine. The solvent was removed by distillation at atmospheric pressure followed by rotary evaporation. The residue was fractionated by vacuum distillation (3 mm Hg) to give 147.4 g (82 % counting per deutero-paraform) of pale yellow oil. B.p. 75°C. HRMS, *m/z* calculated for C₈H₁₁D₂Br: 190.0324; found: 189.0301, 191.0321. ¹H NMR (CDCl₃, δ): 2.23 (t, J = 7.0 Hz, 2H, CH₂), 1.50 (m, 2H, CH₂), 1.33 (m, 4H, CH₂), 0.89 (t, J = 6.9 Hz, 3H, CH₃),

[0102] **11,11-Dideutero-octadeca-9,12-diynoic acid methyl ester (5)** CuI (133 g) was quickly added to 400 ml of DMF (freshly distilled over CaH₂), followed by dry NaI (106 g),

K_2CO_3 (143 g). Dec-9-ynoic acid methyl ester ((4); 65 g) was then added in one portion, followed by bromide (3) (67 g). Additional 250 ml of DMF was used to rinse the reagents off the flask walls into the bulk of reaction mixture, which was then stirred for 12 h. 500 ml of saturated aqueous NH_4Cl was then added with stirring, followed in a few minutes by saturated aqueous $NaCl$ and then by a 5:1 mixture of hexane:EtOAc (300 ml). The mixture was further stirred for 15 min and then filtered through a fine mesh Schott glass filter. The residue was washed with hexane:EtOAc mix several times. The organic fraction was separated, and the aqueous phase was additionally extracted (3 x 200 ml). The combined organic fraction was dried (Na_2SO_4), traces of hydroquinone and diphenylamine were added, and the solvent was evaporated *in vacuo*. The residue was immediately distilled at 1 mm Hg, to give 79 g (77%) of a 165-175°C boiling fraction. HRMS, m/z calculated for $C_{19}H_{28}D_2O_2$: 292.2369; found: 292.2365. 1H NMR ($CDCl_3$, δ): 3.67 (s, 3H, OCH_3), 2.3 (t, $J = 7.5$ Hz, 2H, CH_2), 2.14 (t, $J = 7.0$ Hz, 4H, CH_2), 1.63 (m, 2H, CH_2), 1.47 (m, 4H, CH_2), 1.3 (m, 10H, CH_2), 0.88 (t, $J = 7.0$ Hz, 3H, CH_3).

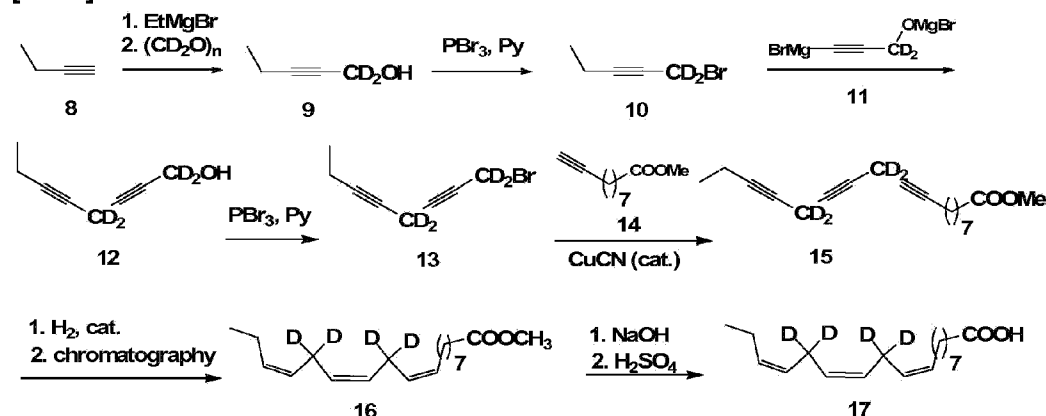
[0103] 11,11-Dideutero-*cis,cis*-octadeca-9,12-dienoic acid methyl ester (6) A suspension of nickel acetate tetrahydrate (31.5 g) in 96 % EtOH (400 ml) was heated with stirring to approx. 50-60°C until the salt dissolved. The flask was flushed with hydrogen, and then 130 ml of $NaBH_4$ solution, (prepared by a 15 min stirring of $NaBH_4$ suspension (7.2 g) in EtOH (170 ml) followed by filtering) was added dropwise over 20-30 min with stirring. In 15-20 min ethylenediamine (39 ml) was added in one portion, followed in 5 min by an addition of (5) (75 g) in EtOH (200 ml). The reaction mixture was very vigorously stirred under hydrogen (1 atm). The absorption of hydrogen stopped in about 2 h. To the reaction mixture, 900 ml of hexane and 55 ml of ice cold AcOH were added, followed by water (15 ml). Hexane (400 ml) was added, and the mixture was allowed to separate. Aqueous fractions were extracted by 5:1 mix of hexane:EtOAc. The completion of extraction was monitored by TLC. The combined organic phase was washed with diluted solution of H_2SO_4 , followed by saturated $NaHCO_3$ and saturated $NaCl$, and then dried over Na_2SO_4 . The solvent was removed at reduced pressure. Silica gel (Silica gel 60, Merck; 162 g) was added to a solution of silver nitrate (43 g) in anhydrous MeCN (360 ml), and the solvent removed on a rotavap. The obtained impregnated silica gel was dried for 3 h at 50°C (aspiration pump) and then 8 h on an oil pump. 30 g of this silica was used per gram of product. The reaction mixture was dissolved in a small volume of hexane and applied to the silver-modified silica gel, and pre-washed with a 1-3 % gradient of EtOAc. When the non-polar contaminants were washed off (control by TLC), the product was eluted with 10 % EtOAc and the solvent evaporated *in vacuo* to give 52 g of the title ester (6) as a colourless liquid. HRMS, m/z calculated for $C_{19}H_{32}D_2O_2$: 296.2682; found: 296.2676. IR (CCl_4): $\tilde{\nu} = 1740\text{ cm}^{-1}$. 1H NMR ($CDCl_3$, δ): 5.32 (m, 4H), 3.66 (s, 3H, OCH_3), 2.29 (t, $J = 7.5$ Hz, 2H, CH_2), 2.02 (m, 4H, CH_2), 1.60 (m, 2H, CH_2), 1.30 (m, 14H, CH_2), 0.88 (t, $J = 7.0$ Hz, 3H, CH_3).

[0104] 11,11-Dideutero-*cis,cis*-octadeca-9,12-dienoic acid (7) A solution of KOH (46 g) in water (115 ml) was added to a solution of ester (6) (46 g) in MeOH (60 ml). The reaction

mixture was stirred at 40-50°C for 2 h (control by TLC) and then diluted with 200 ml of water. Two thirds of the solvent were removed (rotavap). Diluted sulphuric acid was added to the residue to pH 2, followed by diethyl ether with a little pentane. The organic layer was separated and the aqueous layer washed with diethyl ether with a little pentane. The combined organic fractions were washed with saturated aqueous NaCl and then dried over Na₂SO₄. The solvent was evaporated to give 43 g of (7) (99%). IR (CCl₄): $\tilde{\nu}$ = 1741, 1711 cm⁻¹.

Example 2. Synthesis of 11,11,14,14-D₄-linolenic acid

[0105]



[0106] **1,1-Dideutero-pent-2-yn-1-ol (9)** But-1-yne (8) was slowly bubbled through a solution of ethylmagnesium bromide prepared from bromoethane (100 ml) and magnesium turnings (31.3 g) in dry THF (800 ml) on a bath (-5°C). Every now and then the bubbling was stopped and the cylinder with but-1-yne was weighed to measure the rate of consumption. The supply of alkyne was stopped shortly after a voluminous precipitate formed (the measured mass of alkyne consumed was 125 g). The reaction mixture was warmed up to r.t. over 30 min, and then stirred for 15 min. The mixture was then heated up to 30°C, at which point the precipitate dissolved, and then stirred at r.t. for another 30 min. Deuteroparaform (28 g) was added in one portion and the mixture was refluxed for 3 h, forming a clear solution. It was cooled down to r.t. and poured into a mixture of crushed ice (800 g) and 50 ml conc. H₂SO₄. Hexane (400 ml) was added and the organic layer was separated. The aqueous phase was saturated with NaCl and extracted with a 4:1 mixture of hexane:EtOAc (1 L). The completion of extraction process was monitored by TLC. The combined organic phases were washed with saturated NaCl, NaHCO₃ and again NaCl, and dried over Na₂SO₄. The solvent was removed by distillation at the atmospheric pressure (max vapour temperature 105°C). The residue (70.5 g; 94 %) was used without further purification. HRMS, *m/z* calculated for C₅H₆D₂O: 86.0699; found: 86.0751. ¹H NMR (CDCl₃, δ): 2.21 (q, J = 7.5 Hz, 2H, CH₂), 1.93 (br s, 1H, OH), 1.12 (t, J = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 87.7, 77.6, 13.7, 12.3 (signal of CD₂ is absent).

[0107] 1,1-Dideutero-1-bromo-pent-2-yne (10) To a solution of (9) (70.5 g) and pyridine (16.5 ml) in dry diethyl ether (280 ml), 32.3 ml of PBr₃ in 50 ml diethyl ether was added dropwise with stirring over 30 min at -10°C under argon. The reaction mixture was allowed to gradually warm up to r.t. over 1 h. A small amount of hydroquinone was added, and the mixture was then refluxed for 4.5 h. The reaction mixture was then cooled down to -10°C and 350 ml of cold water was added. When the residue dissolved, saturated NaCl (350 ml) and hexane (300 ml) were added, and the organic layer was separated. The aqueous fraction was washed with diethyl ether (2 x 150 ml), and the combined organic fractions were washed with NaCl (2 x 50 ml) and dried over Na₂SO₄ in presence of traces of hydroquinone and triethylamine. The solvent was removed at atmospheric pressure, and then the 147-155°C boiling fraction was distilled off. Alternatively, upon reaching 100°C, the distillation at atmospheric pressure was stopped and the product distilled off at 77-84°C (25 mm Hg). Yield: 107 g of clear liquid. HRMS, *m/z* calculated for C₅H₅D₂Br: 147.9855; found: 146.9814, 148.9835. IR (CCl₄): $\tilde{\nu}$ = 2251 cm⁻¹. ¹H NMR (CDCl₃, δ): 2.23 (q, J = 7.5 Hz, 2H, CH₂), 1.11 (t, J = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 89.3, 74.5, 13.4, 12.6 (signal of CD₂ is absent).

[0108] 1,1,4,4-Tetradeutero-octa-2,5-diyn-1-ol (12) Ethylmagnesium bromide, prepared from ethyl bromide (53 ml) and magnesium turnings (15.8 g) in 400 ml of dry THF, was added in small portions to 350 ml of dry THF, simultaneously with acetylene bubbling through this mixture (at approx. 25 L/h rate) with vigorous stirring. The Grignard reagent solution was fed to the mixture at approx. 10 ml per 2-5 min. When all ethylmagnesium bromide was added (after approx. 2.5 h), acetylene was bubbled through the system for another 15 min. Deuteroparaform (17.3 g) and CuCl (0.2 g) were added under argon, and the reaction mixture was refluxed without stirring for 2.5 h, until deuteroparaform dissolved, to yield a solution of (11). Ethylmagnesium bromide solution, prepared from 14.8 g magnesium and 50 ml ethyl bromide in 250 ml of dry THF, was added dropwise to the reaction mixture over 20 min. When the gas emanation ceased, a condenser was attached and 250 ml of solvent were distilled off. The reaction mixture was then cooled to 30°C, and CuCl (1.4 g) was added followed by a dropwise addition, over 15 min, of bromide (10) (69 g). The reaction mixture was then refluxed for 5 h, cooled slightly (a precipitate will form if cooling is too fast), and poured into a slurry of crushed ice (1-1.2 kg) and 40 ml concentrated H₂SO₄. The mixture was washed with hexane (600 ml). The organic fraction was separated, and the aqueous fraction was additionally extracted with 5:1 hexane:EtOAc (2 x 400 ml). The combined organic fraction was washed, with saturated NaCl, followed by saturated NaHCO₃ and NaCl. The bulk of the solvent was removed at atmospheric pressure in presence of traces of hydroquinone and triethylamine. The residue was flushed through 100 ml of silica gel (eluent: 7:1 hexane:EtOAc). The bulk of the solvent was removed at the atmospheric pressure, and the remainder on a rotavap. 49.5 g (85 %) of the title compound obtained was used without further purification. HRMS, *m/z* calculated for C₈H₆D₄O: 126.0979; found: 126.0899. IR(CCl₄): $\tilde{\nu}$ =3622 cm⁻¹. ¹H NMR(CDCl₃, δ):2.16(q,J=7.5Hz,2H, CH₂), 1.85 (br s, 1 H, OH), 1.11 (t, J = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 82.3, 80.4, 78.3, 72.6, 13.7, 12.2

[0109] **1,1,4,4-Tetradeutero-1-bromo-octa-2,5-diyne (13)** was synthesized as described for bromide (3); 2 ml of pyridine, 14 ml PBr₃ and 250 ml of diethyl ether was used for 54.2 g of alcohol (12). The product was purified by distillation at 4 mm Hg. Yield: 53 g (65 %) of (13); b.p. 100-110°C. HRMS, *m/z* calculated for C₈H₅D₄Br: 188.0135; found: 187.0136, 189.0143. IR (CCl₄): $\tilde{\nu}$ = 2255 cm⁻¹. ¹H NMR (CDCl₃, δ): 2.13 (q, J = 7.5 Hz, 2H, CH₂); 1.07 (t, J = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 82.5, 81.8, 75.0, 72.0, 13.6, 12.2.

[0110] **11,11,14,14-Tetradeutero-octadeca-8,12,15-triynoic acid methyl ester (15)** was synthesized in a way similar to that described for 11,11-dideutero-octadeca-9,12-diynoic acid methyl ester (5). CuI (97 g) was quickly added to 400 ml of DMF (freshly distilled over CaH₂), followed by dry NaI (77.5 g), K₂CO₃ (104.5 g). Dec-9-ynoic acid methyl ester ((14); 47.5 g) was then added in one portion, followed by bromide (13) (48.5 g). Additional 250 ml of DMF was used to rinse the reagents off the flask walls into the bulk of reaction mixture, which was then stirred for 12 h. 500 ml of saturated aqueous NH₄Cl was then added with stirring, followed in a few minutes by saturated aqueous NaCl (300 ml) followed by a 5:1 mixture of hexane:EtOAc (300 ml). The mixture was further stirred for 15 min and then filtered through a fine mesh Schott glass filter. The residue was washed with hexane:EtOAc mix several times. The organic fraction was separated, and the aqueous phase was additionally extracted (3 x 200 ml). The combined organic fraction was dried (Na₂SO₄), traces of hydroquinone and diphenylamine were added, and the solvent was evaporated *in vacuo*. The residue was immediately distilled at 1 mm Hg, to give 45.8 g (62%) of a 173-180°C boiling fraction. An additional crystallisation was carried out as follows. The ester (15) was dissolved in hexane (500 ml) and cooled down to -50°C. The crystals formed were washed in cold hexane. The yield of this step is 80 %. HRMS, *m/z* calculated for C₁₉H₂₂D₄O₂: 290.2180; found: 290.2200. ¹H NMR (CDCl₃, δ): 3.66 (s, 3H, OCH₃), 2.29 (t, J = 7.5 Hz, 2H, CH₂), 2.15 (m, 4H, CH₂), 1.61 (m, 2H, CH₂), 1.47 (m, 2H, CH₂), 1.30 (m, 6H, CH₂), 1.11 (t, J = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 174.1, 82.0, 80.6, 74.7, 74.6, 73.7, 73.0, 51.3, 33.9, 28.9, 28.6, 28.52, 28.49, 24.8, 18.5, 13.7, 12.2.

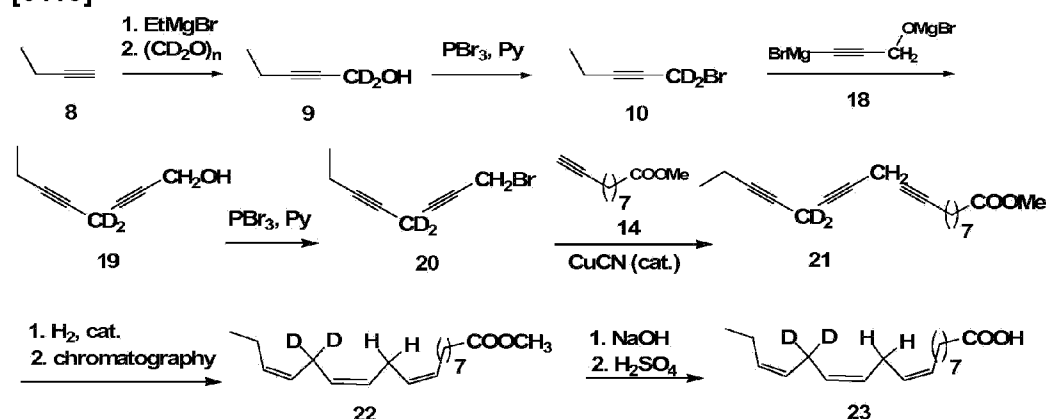
[0111] **11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid methyl ester (16)** was synthesized in a way similar to that described for 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid methyl ester ('6'). A suspension of nickel acetate tetrahydrate (42 g) in 96 % EtOH (400 ml) was heated with stirring to approx. 50-60°C until the salt dissolved. The flask was flushed with hydrogen, and then 130 ml of NaBH₄ solution, (prepared by a 15 min stirring of NaBH₄ suspension (7.2 g) in EtOH (170 ml) followed by filtering) was added dropwise over 20-30 min with stirring. In 15-20 min ethylenediamine (52 ml) was added in one portion, followed in 5 min by an addition of (15) (73 g) in EtOH (200 ml). The reaction mixture was very vigorously stirred under hydrogen (1 atm). The absorption of hydrogen stopped in about 2 h. To the reaction mixture, 900 ml of hexane and 55 ml of ice cold AcOH were added, followed by water (15 ml). Hexane (400 ml) was added, and the mixture was allowed to separate. Aqueous fractions were extracted by 5:1 mix of hexane:EtOAc. The completion of extraction was monitored by TLC. The combined organic phase was washed with diluted solution of H₂SO₄,

followed by saturated NaHCO_3 and saturated NaCl , and then dried over Na_2SO_4 . The solvent was removed at reduced pressure. Silica gel for purification was prepared as described for (6). 30 g of this silica was used per gram of product. The reaction mixture was dissolved in a small volume of hexane and applied to the silver-modified silica gel, and pre-washed with a 1-5 % gradient of EtOAc. When the non-polar contaminants were washed off (control by TLC), the product was eluted with 10 % EtOAc and the solvent evaporated in vacuo to give 42 g of the title ester (16) as a colourless liquid. HRMS, m/z calculated for $\text{C}_{19}\text{H}_{28}\text{D}_4\text{O}_2$: 296.2649; found: 296.2652. IR (CCl_4): $\tilde{\nu} = 1740\text{ cm}^{-1}$. ^1H NMR (CDCl_3 , δ): 5.4 (m, 6H, CH-double bond), 3.68 (s, 3H, OCH_3), 2.33 (t, $J = 7.5\text{ Hz}$, 2H, CH_2), 2.09 (m, 4H, CH_2), 1.62 (m, 2H, CH_2), 1.33 (m, 8H, CH_2), 0.97 (t, $J = 7.5\text{ Hz}$, 3H, CH_3). ^{13}C NMR (CDCl_3 , δ): 174.1, 131.9, 130.2, 128.2, 128.1, 127.7, 126.9, 51.3, 34.0, 29.5, 29.04, 29.02, 27.1, 25.5, 24.9, 20.5, 14.2.

[0112] **11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (17)** A solution of KOH (1.5 g, 27 mmol) in water (2.6 ml) was added to a solution of ester (16) (1.00 g, 3.4 mmol) in MeOH (15 ml). The reaction mixture was stirred at 40-50°C for 2 h (control by TLC) and then diluted with 20 ml of water. Two thirds of the solvent were removed (rotavap). Diluted sulfuric acid was added to the residue to pH 2, followed by diethyl ether with a little pentane (50 ml). The organic layer was separated and the aqueous layer washed with diethyl ether with a little pentane (3 x 30 ml). The combined organic fractions were washed with saturated aqueous NaCl and then dried over Na_2SO_4 . The solvent was evaporated to give 0.95 g of (17) (100%). IR (CCl_4): $\tilde{\nu} = 1741, 1711\text{ cm}^{-1}$.

Example 3. Synthesis of 14,14-D2-linolenic acid

[0113]



[0114] **4,4-Dideutero-octa-2,5-diyn-1-ol (19)** To a solution of ethylmagnesium bromide, prepared from ethyl bromide (9.2 ml, 123.4 mmol) and magnesium turnings (2.74 g, 112.8 mmol) in 40 ml of dry THF, on an ice bath with stirring, propargyl alcohol (3.16 g, 56.4 mmol) in THF (5 ml) was added dropwise over 10-15 min. The reaction mixture was allowed to warm up

to r.t. and stirred for another 2 h, with occasional warming to 40°C. To thus generated dianion, 0.13g of CuCl was added, followed by slow (over 15 min) addition of bromide (10) (6.9 g) in THF (20 ml). The reaction mixture was then stirred for 1 h at r.t. and then refluxed for 5 h. The reaction mixture was then refluxed for 5 h, cooled slightly (a precipitate will form if cooling is too fast), and poured into a slurry of crushed ice and 2.5 ml concentrated H₂SO₄. The mixture was washed with hexane (600 ml). The organic fraction was separated, and the aqueous fraction was additionally extracted with 5:1 hexane:EtOAc. The combined organic fraction was washed, with saturated NaCl, followed by saturated NaHCO₃ and NaCl, and dried over Na₂SO₄. The bulk of the solvent was removed at atmospheric pressure in presence of traces of hydroquinone and triethylamine. The product was purified by CC (hexane:EtOAc = 15:1) to give 3.45 g (59 %) of the product 19. HRMS, *m/z* calculated for C₈H₈D₂O: 124.0855; found: 124.0849. IR (CCl₄): $\tilde{\nu}$ = 3622 cm⁻¹. ¹H NMR (CDCl₃, δ): 4.21 (m, 2H, CH₂), 2.4 (m, 1H, OH), 2.16 (q, *J* = 7.5 Hz, 2H, CH₂), 1.11 (t, *J* = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 82.3, 80.4, 78.3, 72.6, 51.0, 13.7, 12.2.

[0115] **4,4-Dideutero-1-bromo-octa-2,5-diyne (20)** was synthesized as described for (3), except all solvent was removed on a rotavap. From 3.4 g (27 mmol) of (19), 3.9 g (75 %) of the bromide (20) was obtained, which was used without further purification. HRMS, *m/z* calculated for C₈H₇D₂Br: 186.0011; found: 185.0019, 187.0012. IR (CCl₄): $\tilde{\nu}$ = 2255 cm⁻¹. ¹H NMR (CDCl₃, δ): 3.88 (br s, 2H, CH₂), 2.13 (q, *J* = 7.5 Hz, 2H, CH₂), 1.07 (t, *J* = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 82.5, 81.8, 75.0, 72.0, 14.8, 13.6, 12.2.

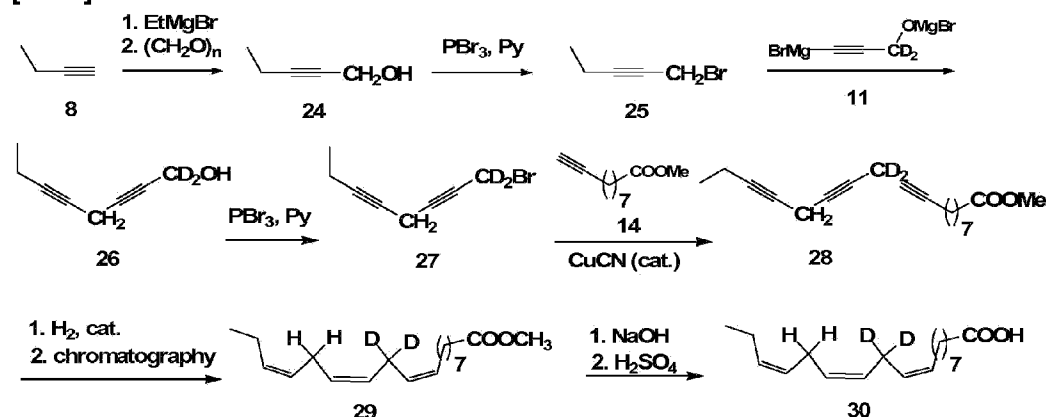
[0116] **14,14-Dideutero-octadeca-8,12,15-triynoic acid methyl ester (21)** was synthesized as described for (5). The product obtained from 9.7 g CuI, 7.8 g NaI, 10.5 g K₂CO₃, 4.85 g of bromide (20), 4.75 g of methyl ester (14) and 40 ml of anhydrous DMF, was purified by CC (25:1 hexane:EtOAc) to give 4.5 g (60%) of the title compound. HRMS, *m/z* calculated for C₁₉H₂₄D₂O₂: 288.2056; found: 288.2046. ¹H NMR (CDCl₃, δ): 3.66 (s, 3H, OCH₃), 3.12 (m, 2H, CH₂), 2.29 (t, *J* = 7.5 Hz, 2H, CH₂), 2.15 (m, 4H, CH₂), 1.61 (m, 2H, CH₂), 1.47 (m, 2H, CH₂), 1.30 (m, 6H, CH₂), 1.11 (t, *J* = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 174.1, 82.0, 80.6, 74.7, 74.6, 73.7, 73.0, 51.3, 33.9, 28.9, 28.6, 28.52, 28.49, 24.8, 18.5, 13.7, 12.2, 9.7.

[0117] **14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid methyl ester (22)** was synthesized as described for the linoleic acid derivative (6). For a reduction of 4.5 g of (21), 2.6 g of nickel acetate tetrahydrate and 3.2 ml ethylenediamine was used. The product was purified on AgNO₃-impregnated silica gel as described for (6). HRMS, *m/z* calculated for C₁₉H₃₀D₂O₂: 294.2526; found: 294.2529. IR (CCl₄): $\tilde{\nu}$ = 1740 cm⁻¹. ¹H NMR (CDCl₃, δ): 5.37 (m, 6H, CH-double bond), 3.68 (s, 3H, OCH₃), 2.82 (m, 2H, CH₂), 2.33 (t, *J* = 7.5 Hz, 2H, CH₂), 2.09 (m, 4H, CH₂), 1.62 (m, 2H, CH₂), 1.33 (m, 8H, CH₂), 0.97 (t, *J* = 7.5 Hz, 3H, CH₃). ¹³C NMR (CDCl₃, δ): 174.1, 131.9, 130.2, 128.2, 128.1, 127.7, 126.9, 51.3, 34.0, 29.5, 29.1, 29.04, 29.02, 27.1, 25.5, 24.9, 20.5, 14.2.

[0118] **14,14-Dideutero-*cis,cis,cis*-octadeca-8,12,15-trienoic acid (23)** To a solution of (22) (1 g, 3.4 mmol) in MeOH (15 ml), a solution of KOH (1.5 g, 27 mmol) in water (2.6 ml) was added in one portion. The reaction mixture was then processed as described for (7) to yield 0.94g (99 %) of the title acid. IR (CCl₄): $\tilde{\nu}$ = 1741, 1711 cm⁻¹.

Example 4. Synthesis of 11,11-D₂-linolenic acid

[0119]



[0120] **Pent-2-yn-1-ol (24)** Butyn-1 ((8); 10.4 g) was bubbled through an ice-cold solution prepared from bromoethane (11.2 ml) and magnesium turnings (3.6 g) in THF (100 ml). The reaction mixture was allowed to warm up to r.t. and then stirred for 15 min. The mixture was then heated up to 30°C, at which point all precipitate dissolved. The heating was removed and the mixture stirred for another 30 min, and then paraform (3 g) was added in one portion. The reaction mixture was refluxed for 3 h (all paraform dissolved), then cooled to r.t., poured into a mixture of crushed ice (80 g) and 8 ml conc. H₂SO₄, and extracted with diethyl ether. The organic phase was washed with saturated NaHCO₃ and NaCl, and dried over Na₂SO₄. The solvent was removed on a rotavap, and the residue (7.56 g; 90 %) was used without further purification. HRMS, *m/z* calculated for C₅H₈O: 84.0575; found: 84.0583.

[0121] **1-Bromo-pent-2-yne (25)** To a solution of (24) (11.7 g) and pyridine (2.66 ml) in dry diethyl ether (34 ml), 5.2 ml of PBr₃ in 5 ml diethyl ether was added dropwise with stirring over 30 min at -10°C under argon. The reaction mixture was allowed to gradually warm up to r.t. over 1 h. A catalytic amount of hydroquinone was added, and the mixture was then refluxed for 4.5 h. The reaction mixture was then cooled down to -10°C and 35 ml of cold water was added. When the residue dissolved, saturated NaCl (35 ml) and diethyl ether (30 ml) were added, and the organic layer was separated. The aqueous fraction was washed with diethyl ether (2 x 15 ml), and the combined organic fractions were washed with NaCl (2 x 400 ml) and dried over MgSO₄. The solvent was removed at atmospheric pressure, and then under reduced pressure (25 mm Hg), the 60-90°C fraction was collected. Yield: 11.1 g (84 %). HRMS, *m/z* calculated for C₅H₇Br: 145.9731; found: 144.9750, 146.9757.

[0122] **1,1-Dideutero-octa-2,5-diyne-1-ol (26)** was synthesized as described for (12) with 87 % yield. HRMS, m/z calculated for $C_8H_8D_2O$: 124.0855; found: 124.0868. IR (CCl_4): $\tilde{\nu}$ = 3622 cm^{-1} . 1H NMR ($CDCl_3$, δ): 2.65 (m, 2H, CH_2), 2.4 (m, 1H, OH), 2.1 (q, 2H, CH_2), 1.09 (t, 3H, CH_3).

[0123] **1,1-Dideutero-1-bromo-octa-2,5-diyne (27)** was synthesized as described for (3), except all solvent was removed on a rotavap. The product was purified by distillation at reduced pressure. Yield: 86 % (b.p. 100-105°C at 4 mm Hg). HRMS, m/z calculated for $C_8H_7D_2Br$: 186.0011; found: 184.9948, 187.9999. IR (CCl_4): $\tilde{\nu}$ = 2255 cm^{-1} . 1H NMR ($CDCl_3$, δ): 2.66 (m, 2H, CH_2), 2.1 (q, 2H, CH_2), 1.09 (t, 3H, CH_3).

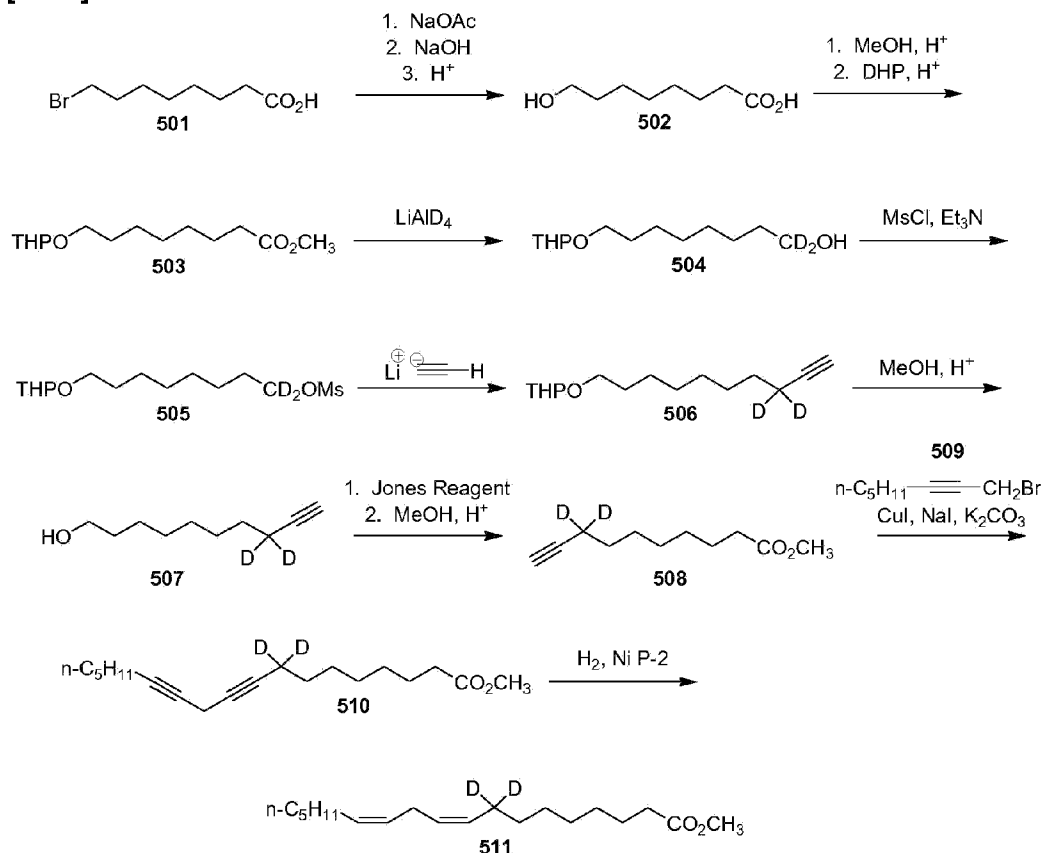
[0124] **11,11-Dideutero-octadeca-8,12,15-triynoic acid methyl ester (28)** was synthesized as described for (5). The product obtained from 7.1 g CuI, 5.66 g NaI, 7.65 g K_2CO_3 , 3.55 g of bromide (27), 3.47 g of methyl ester (14) and 30 ml of anhydrous DMF, was purified by CC (25:1 hexane:EtOAc) to give 3.7 g of the title compound. HRMS, m/z calculated for $C_{19}H_{24}D_2O_2$: 288.2056; found: 288.2069. 1H NMR ($CDCl_3$, δ): 3.7 (s, 3H, OCH_3), 3.15 (br. s, 2H, CH_2), 2.35 (m, 2H, CH_2), 2.17 (m, 4H, CH_2), 1.61 (m, 2H, CH_2), 1.48 (m, 2H, CH_2), 1.35 (m, 6H, CH_2), 1.11 (t, 3H, CH_3).

[0125] **11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid methyl ester (29)** was synthesized as described for the linoleic acid derivative (6). For a reduction of 3.7 g of (28), 2.16 g of nickel acetate tetrahydrate and 2.62 ml ethylenediamine was used. The product was purified on $AgNO_3$ -impregnated silica gel as described for (6) to give 1.5 g. HRMS, m/z calculated for $C_{19}H_{30}D_2O_2$: 294.2526; found: 294.2402. IR (CCl_4): $\tilde{\nu}$ = 1740 cm^{-1} . 1H NMR ($CDCl_3$, δ): 5.37 (m, 6H, CH-double bond), 3.6 (s, 3H, OCH_3), 2.82 (m, 2H, CH_2), 2.33 (t, o = 7.5 Hz, 2H, CH_2), 2.09 (m 4H, CH_2), 1.62 (m, 2H, CH_2), 1.33 (m, 8H, CH_2), 0.97 (t, J = 7.5 Hz, 3H, CH_3). ^{13}C NMR ($CDCl_3$, δ): 174.1, 131.9, 130.2, 128.2, 128.1, 127.7, 126.9, 51.3, 34.0, 29.5, 29.1, 29.04, 29.02, 27.1, 25.5, 24.9, 20.5, 14.2.

[0126] **11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (30)** To a solution of (29) (1.5 g, 5.1 mmol) in MeOH (7.5 ml), a solution of KOH (1.5 g, 27 mmol) in water (3 ml) was added in one portion. The reaction mixture was then processed as described for (17) to yield 0.9 g of the title acid. IR (CCl_4): $\tilde{\nu}$ = 1741, 1711 cm^{-1} . 1H NMR ($CDCl_3$, δ): 11.2 (br s, 1 H, COOH), 5.37 (m, 6H, CH-double bond), 2.83 (m, 2H, CH_2), 2.35 (t, J = 7.5 Hz, 2H, CH_2), 2.06 (m 4H, CH_2), 1.63 (m, 2H, CH_2), 1.32 (m, 8H, CH_2), 0.97 (t, J = 7.5 Hz, 3H, CH_3). ^{13}C NMR ($CDCl_3$, δ): 180.4, 131.9, 130.2, 128.3, 128.1, 127.6, 127.1, 34.1, 29.5, 29.1, 29.03, 28.98, 27.2, 25.5, 24.6, 20.5, 14.2.

Example 5. Synthesis of 8,8-D₂-Linoleic Acid Methyl Ester

[0127]



[0128] 8-Hydroxyoctanoic acid (502). A solution of 8-bromocaprylic acid (501, 37.5 g, 168 mmol), anhydrous sodium acetate (60.0 g, 732 mmol) and sodium iodide (1.0 g, 6.7 mmol) in DMF (200 ml) was stirred at 110-120°C for 8 h. The reaction mixture was cooled to r.t., a solution of potassium hydroxide (28 g, 0.5 mol) in water (150 ml), was added, and the mixture was stirred at 100°C for another hour. The reaction mixture was cooled to r.t. and poured into slurry of ice and concentrated sulfuric acid (45 ml). The solution obtained was saturated with NaCl and extracted (9 x 150 ml) with a mixture of EtOAc and petroleum ether (1:1). Combined organic fractions were washed twice with saturated NaCl and dried over Na₂SO₄. The solvent was evaporated to give 26.5 g (98%) of the product which was used without further purification. A small amount of the product was further purified by CC on silica (eluent: petroleum ether:EtOAc = 2:1) and characterized. ¹H NMR (400 MHz, CDCl₃) δ 1.27-1.39 (m, 6H), 1.50-1.68 (m, 4H), 2.32 (t, 2H, J = 7.5 Hz), 3.62 (t, 2H, J = 6.5 Hz), 6.87 (br. s., 2H).

[0129] Methyl 8-(tetrahydro-2H-pyran-2-yloxy)octanoate (503). 8-Hydroxyoctanoic acid (502; 26.3 g, 164 mmol) was dissolved in methanol (500 ml) containing acetyl chloride (3.5 ml). The reaction mixture was refluxed for 5 h and the solvent removed *in vacuo*. To the residue dissolved in CH₂Cl₂ (200 ml), 3,4-dihydro-2H-pyran (29 ml, 318 mmol) was added, and the reaction mixture was refluxed for 20 min. Upon addition of 5 ml of triethylamine, the solvent

was removed *in vacuo*, and the residue was dissolved in petroleum ether (100 ml) and washed with water. The organic layer was flush-purified on a small silica column (silica, 100 ml; eluent: from petroleum ether to petroleum ether:EtOAc = 20:1). The work-up yielded 38.2 g (90%) of the product which was used without further purification. A small amount of the product was further purified by CC on silica (eluent: petroleum ether: EtOAc = 15:1) and characterized. IR (CCl₄): $\tilde{\nu}$ = 1741 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.20-1.36 (m, 6H), 1.40-1.82 (m, 10H), 2.23 (t, 2H, J = 7.5 Hz), 3.30 (dt, 1 H, J = 9.5 Hz, 6.5 Hz), 3.39-3.46 (m, 1H), 3.59 (s, 3H), 3.65 (dt, 1 H, J = 9.5 Hz, 7.0 Hz), 3.76-3.83 (m, 1H), 4.47-4.52 (m, 1H).

[0130] [1,1-D₂]-8-(tetrahydro-2H-pyran-2-yloxy)octan-1-ol (504). To a stirred solution of ester (503) (37.5 g, 145 mmol) in diethyl ether (100 ml) in an ice bath, a suspension of LiAlD₄ (4.0 g, 95 mmol) in diethyl ether (300 ml) was added drop wise over 1 h. To the cold reaction mixture, water (4 ml), 15% NaOH (4 ml) and water (12 ml) were added with stirring. The precipitate was filtered and washed with ethyl ether. Evaporation *in vacuo* gave 33.5 g (99%) of the product. A small amount of the product was further purified by CC on silica (eluent: petroleum ether: EtOAc = 10:1) and characterized. IR (CCl₄): $\tilde{\nu}$ = 3638, 3499 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.22-1.33 (m, 8H), 1.42-1.56 (m, 8H), 1.61-1.69 (m, 1H), 1.71-1.80 (m, 1H), 2.38 (br. s., 1H), 3.31 (dt, 1 H, J = 9.5 Hz, 6.5 Hz), 3.40-3.46 (m, 1H), 3.66 (dt, 1 H, J = 9.5 Hz, 7.0 Hz), 3.76-3.84 (m, 1H), 4.49-4.53 (m, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 19.5, 25.3, 25.5, 26.0, 29.2, 29.3, 29.5, 30.6, 32.4, 62.1, 67.5, 98.7.

[0131] [1,1-D₂]-8-(tetrahydro-2H-pyran-2-yloxy)octyl methanesulfonate (505). To a solution of alcohol (504) (33.4 g, 144 mmol) and triethylamine (45 ml, 323 mmol) in diethyl ether (300 ml) at 0°C, a solution of MsCl (14.2 ml, 183 mmol) in diethyl ether (100 ml) was added drop wise over 1 h with stirring. The reaction mixture was warmed up to r.t. and treated with water. The organic phase, combined with washings (2 x 50 ml) of the aqueous phase with Et₂O, was washed twice with saturated NaCl, dried over Na₂SO₄, and decanted. This was flush-purified on a small silica column (silica, 100 ml; petroleum ether:EtOAc = 10:1). The work-up yielded 43.7 g (98%) of methanesulfonate (505). IR (CCl₄): $\tilde{\nu}$ = 1739 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.26-1.41 (m, 8H), 1.44-1.59 (m, 6H), 1.63-1.84 (m, 4H), 2.97 (s, 3H), 3.32 (dt, 1 H, J = 9.5 Hz, 6.5 Hz), 3.42-3.50 (m, 1H), 3.69 (dt, 1 H, J = 9.5 Hz, 7.0 Hz), 3.78-3.86 (m, 1H), 4.52-4.56 (m, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 19.6, 25.2, 25.4, 26.0, 28.7, 28.8, 29.1, 29.5, 30.7, 37.2, 62.3, 67.4, 98.8.

[0132] 2-([8,8-D₂]-dec-9-yne-1-yloxy)tetrahydro-2H-pyran (506). Methanesulfonate (505) (43.5 g, 140 mmol) in DMSO (100 ml) was added dropwise with stirring over 1 h to a suspension of a ethylenediamine - lithium acetylenide complex (70 g, 0.76 mol) in DMSO (200 ml), and then the mixture was stirred for 90 min. Reaction mixture was poured on ice, extracted (Et₂O, 3 x 150 ml), dried over Na₂SO₄ and evaporated. This was flush-purified on a small silica column (silica, 100 ml; petroleum ether). Removal of solvent (rotavap) gave 25.3 g (75%) of the product. A small amount of the product was further purified by CC on silica

(eluent: petroleum ether: EtOAc = 25:1) and characterized. IR (CCl₄): $\tilde{\nu}$ = 3314 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.21-1.38 (m, 8H), 1.42-1.57 (m, 8H), 1.62-1.70 (m, 1H), 1.73-1.83 (m, 1H), 1.89 (s, 1H), 3.32 (d.t., 1 H, J = 9.5 Hz, 6.5 Hz), 3.42-3.50 (m, 1H), 3.68 (d.t., 1 H, J = 9.5 Hz, 7.0 Hz) 3.78-3.86 (m, 1H), 4.51-4.54 (m, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 19.6, 25.4, 26.1, 28.1, 28.5, 28.9, 29.2, 29.6, 30.6, 30.7, 62.1, 67.5, 68.0, 98.7.

[0133] [8,8-D₂]-dec-9-yne-1-ol (507). Ether (506) (25 g, 104 mmol) was dissolved in methanol (300 ml) containing pyridinium *para*-toluenesulfonate (0.2 g). Reaction mixture was refluxed for 3 h, quenched with Et₃N (1 ml), the solvent removed *in vacuo*, the residue dissolved in petroleum ether and filtered through a small amount of silica gel. The solvent was evaporated to give 15.4 g (95 %) of the product. A small amount of the product was further purified by CC on silica (eluent: petroleum ether: EtOAc = 15:1) and characterized. IR (CCl₄): $\tilde{\nu}$ = 3638, 3508, 3314 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.22-1.40 (m, 8H), 1.42-1.56 (m, 4H), 1.91 (s, 1H), 2.29 (br. s., 1H), 3.59 (t, J = 6.5 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 25.6, 28.1, 28.5, 29.0, 29.2, 32.6, 62.8, 68.1, 84.6.

[0134] [8,8-D₂]-methyl dec-9-ynoate (508). To a solution of chromium trioxide (24 g, 0.24 mol) and concentrated sulfuric acid (21 ml) in water (100 ml) in a two-neck round bottom flask on water bath at 30°C with stirring, a solution of alcohol (507) (15.5 g, 99 mmol) in acetone (150 ml) was added dropwise over 90 min. Upon addition, the reaction mixture was stirred for another 15 min, and the excess of oxidizer was quenched with isopropyl alcohol. The mixture was poured into cold water and extracted with diethyl ether (5 x 50 ml). Combined organic fractions were washed with saturated NaCl, dried over Na₂SO₄, filtered, and the solvent removed *in vacuo*. The residue was dissolved in methanol (200 ml) and upon addition of concentrated sulfuric acid (1 ml) refluxed for 90 min. The acid was quenched with triethylamine (6.5 ml, 47 mmol), the solvent removed *in vacuo*, and the residue purified by CC on silica (eluent: petroleum ether: EtOAc = 50:1) to give 12.6 g (69 % counting per alcohol (507)) of ester (508). and characterized. IR (CCl₄): $\tilde{\nu}$ = 3314, 1740 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 1.19-1.38 (m, 6H), 1.41-1.48 (m, 2H), 1.51-1.61 (m, 2H), 1.88 (s, 1H), 2.25 (t, J = 7.5 Hz, 2H), 3.60 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 24.7, 28.0, 28.3, 28.6, 28.8, 33.9, 51.3, 68.1, 84.4, 174.0.

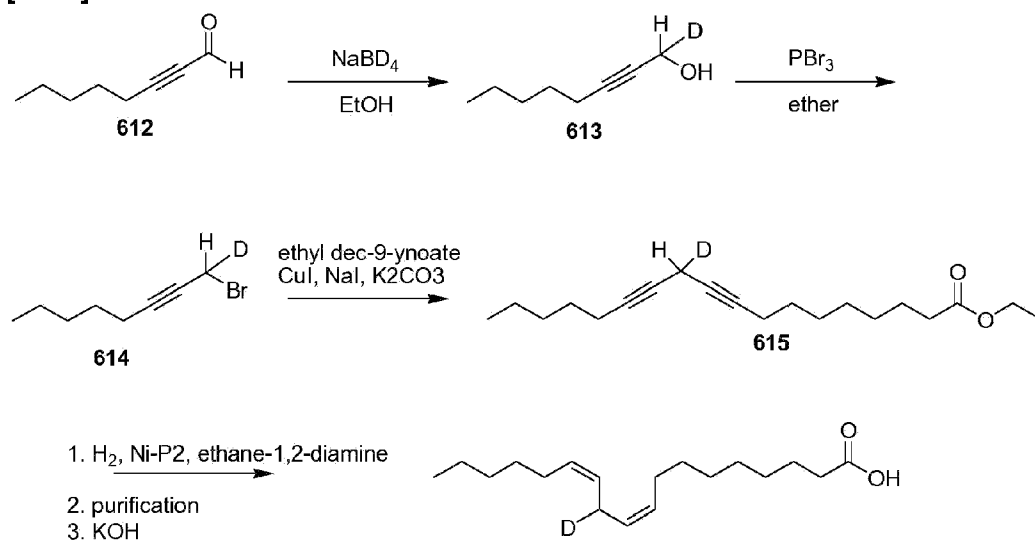
[0135] [8,8-D₂]-methyl octadeca-9,12-diynoate (510). To DMF (20 ml) were added with stirring CuI (3.9 g, 20 mmol), followed by NaI (3.1 g, 21 mmol), K₂CO₃ (4.2 g, 30 mmol), ester (508) (1.9 g, 10.3 mmol), and bromide (509) (2.04 g, 10.8 mmol, synthesized as described in [2]). The reaction mixture was stirred at r.t. for 12 h. Saturated aqueous ammonium chloride (20 ml) was added to the mixture, followed by saturated NaCl (15 ml). The precipitate and the aqueous phase were washed with petroleum ether. The combined organic fractions were washed with saturated sodium chloride, dried over Na₂SO₄ and evaporated *in vacuo*. The residue was purified by CC on silica (eluent: petroleum ether: EtOAc = 50:1) to give 2.47 g (82

%) of the product. ^1H NMR (400 MHz, CDCl_3) δ 0.86 (t, J = 7.0 Hz, 3H), 1.22-1.36 (m, 10H), 1.40-1.50 (m, 4H), 1.55-1.64 (m, 2H), 2.09-2.15 (m, 2H), 2.28 (t, J = 7.5 Hz, 2H), 3.09 (t, J = 2.5 Hz, 2H), 3.64 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 9.6, 13.9, 18.6, 22.1, 24.8, 28.3, 28.4, 28.5, 28.7, 28.9, 31.0, 34.0, 51.4, 74.4, 74.5, 80.2, 80.4, 174.2.

[0136] [8,8- D_2]-octadeca-9,12-dienoate (511). A suspension of finely ground $\text{Ni}(\text{Ac})_2 \times 4\text{H}_2\text{O}$ (0.8 g, 3.2 mmol) in 96 % ethanol (25 ml) was heated with stirring to 50-60°C until the salt was fully dissolved. The system was flushed with hydrogen, and then a solution of NaBH_4 (3.4 ml; obtained by 15 min stirring of NaBH_4 suspension (0.53 g, 14 mmol) in ethanol (12 ml) followed by filtering through a fine filter) was added over 10 min. Evolvement of hydrogen was observed. In 15-20 min, ethylenediamine (1.65 ml, 25 mmol) was added to the reaction mixture in one portion with stirring, followed by the solution of (510) (2.4 g, 8.2 mmol) in ethanol (10 ml). The reaction mixture was vigorously stirred under hydrogen until there was no further absorption of hydrogen, and then treated with acetic acid (2.3 ml), water (10 ml), and extracted with petroleum ether:EtOAc (5:1). Combined organic fractions were washed with 10 % sulfuric acid (10 ml), then with saturated sodium chloride, dried over Na_2SO_4 , and the solvent was removed *in vacuo*. The residue was purified by CC on silica (eluent: petroleum ether: EtOAc = 50:1) to give 2.33 g (96 %) of the product. The product was then purified again by CC on silica impregnated with 20 % AgNO_3 (eluent: petroleum ether to petroleum ether: EtOAc = 2:1). 1.75 g (72 %) of the product was obtained (97 % purity by GC). ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, J = 7.0 Hz, 3H), 1.20-1.40 (m, 14H), 1.55-1.66 (m, 2H), 1.97-2.09 (m, 2H), 2.29 (t, J = 7.5 Hz, 2H), 2.72-2.79 (m, 2H), 3.66 (s, 3H), 5.28-5.41 (m, 4H). ^{13}C NMR (100 MHz, CDCl_3) δ 14.0, 22.5, 24.9, 25.6, 27.2, 29.00, 29.08, 29.13, 29.3, 29.4, 31.5, 34.1, 51.4, 127.9, 128.0, 129.9, 130.2, 174.2.

Example 6. Synthesis of 11-D-Linoleic Acid

[0137]



[0138] oct-2-yn-1-ol (13). To a solution of oct-2-ynal [See Corey, E.J.; Schmidt, G. *Tetrahedron Lett.* 1979, 20, 399; Meyer, M. P.; Klinman, J. P. *Tetrahedron Lett.* 2008, 49, 3600] (**(612)**; 1.00 g, 8.1 mmol) in ethanol (15 ml) cooled to 0°C, 0.11 g (2.6 mmol) of NaBD₄ was added in portions over 5 min. Upon addition, the solution was stirred for another 30 min, diluted with water (100 ml), and then extracted with Et₂O (4 x 20 ml). The combined organic fractions were washed with saturated NaCl, dried (Na₂SO₄), and the solvent was removed at reduced pressure. Alcohol **613** (0.85 g, 83%) was purified by column chromatography (silica gel, petroleum ether:EtOAc (15:1)). ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, J = 7.0 Hz, 3H, CH₃), 1.32 (m, 4H, CH₂), 1.49 (quint, J = 7.0 Hz, 2H, CH₂), 1.81 (br s, 1H, OH), 2.19 (td, J = 7.0 Hz, 2.0 Hz, 2H, CH₂), 4.22 (m, 1H, CHD).

[0139] 1-bromooct-2-yne (614) was synthesized as described in [See Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138.]. ¹H NMR (400 MHz, CDCl₃) δ 0.89 (t, J = 7.0 Hz, 3H, CH₃), 1.32 (m, 4H, CH₂), 1.50 (quint, J = 7.0 Hz, 2H, CH₂), 2.22 (td, J = 7.0 Hz, 2.0 Hz, 2H, CH₂), 3.91 (m, 1H, CHD).

[0140] [11-²H]-ethyl octadeca-9,12-diynoate (615). was synthesized as described [See Meyer, M. P.; Klinman, J. P. *Tetrahedron Lett.* 2008, 49, 3600; Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138]. CuI (2 g, 10.5 mmol), NaI (1.58 g, 10.5 mmol), K₂CO₃ (2.1 g, 15 mmol), ethyl dec-9-ynoate (1.02 g, 5.2 mmol) and bromide **614** (1.03 g, 5.4 mmol) were added to DMF (10 ml) with stirring. The reaction mixture was stirred at RT for 12 h, then NH₄Cl (10 ml) and NaCl (8 ml) were added and the stirring continued for another 5 min. The precipitate was separated and washed with petroleum ether. Organic layers were separated, and the aqueous layer was extracted with petroleum ether. The combined organic fractions were washed with saturated NaCl, dried (Na₂SO₄), and the solvent was removed at reduced pressure. Column chromatography (silica gel, petroleum ether:EtOAc (15:1)) yielded 1.29 g (81 %) of the product. ¹H NMR (400 MHz, CDCl₃) δ 0.89 (t, J = 7.0 Hz, 3H, CH₃), 1.25 (t, J = 7.0 Hz, 3H, CH₃CH₂O), 1.31 (m, 10H, CH₂), 1.49 (m, 4H, CH₂), 1.61 (m, 2H, CH₂), 2.15 (td, J = 7.0 Hz, 2.0 Hz, 2H, CH₂ in propargylic position), 2.28 (t, J = 7.5 Hz, 2H, CH₂COOEt), 3.10 (m, 1H, CHD), 4.12 (q, J = 7.0 Hz, 2H, OCH₂CH₃). ¹³C NMR (100 MHz, CDCl₃) δ 9.6 (t, J = 19.0 Hz), 13.9, 14.1, 18.56, 18.57, 22.1, 24.8, 28.4, 28.6, 28.7, 28.9, 28.9, 31.0, 34.2, 60.0, 74.3, 74.5, 80.2, 80.3, 173.7.

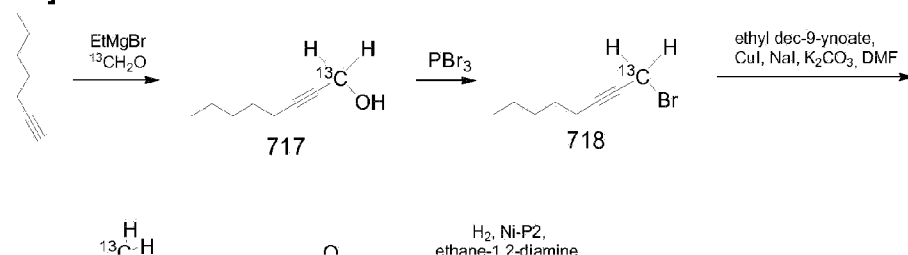
[0141] [11-²H]-linoleic acid (616) A suspension of trituated nickel acetate tetrahydrate (0.4 g, 1.6 mmol) in 96% ethanol (12 ml) was heated at 50-60°C with stirring until the salt dissolved. The system was flushed with hydrogen, and then 1.7 ml of NaBH₄ (obtained by 15-min stirring

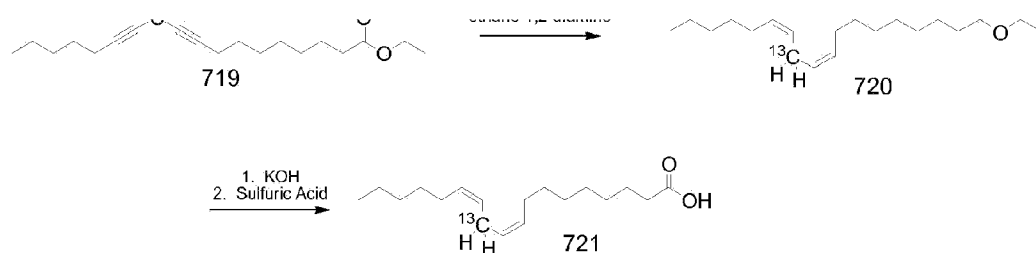
of a NaBH_4 suspension (0.27 g, 14 mmol) in ethanol (6 ml) followed by filtering) was added over 10 min, with some gas bubbles evolving. In 15-20 min, ethylenediamine (0.8 ml, 12 mmol) was added in one portion with stirring, followed in 5 min by a solution of diyne **615** (1.2 g, 3.9 mmol) in ethanol (5 ml). The reaction mixture was stirred vigorously until there was no more absorption of hydrogen, and then treated with acetic acid (1.2 ml), water (10 ml) and extracted with a mixture of petroleum ether and EtOAc (5:1). The combined organic fractions were washed with 10% sulphuric acid (5 ml) and then with saturated NaCl, dried (Na_2SO_4), and the solvent was removed at reduced pressure. Column chromatography (silica gel, petroleum ether:EtOAc (50:1)) yielded 1.14 g (94 %) of the product. The product was additionally purified [3] on a silver nitrate-impregnated silica (20% AgNO_3), with petroleum ether:EtOAc (2:1) as eluent to give 0.73 g (60 %) of the linoleic acid ethyl ester (> 96% purity by GC; GC-MS: MW 309 (GC-MS for a control non-deuterated linoleic acid ethyl ester: MW 308). ^1H NMR (400 MHz, CDCl_3) δ 0.89 (t, J = 7.0 Hz, 3H, CH_3), 1.25 (t, J = 7.0 Hz, 3H, $\text{CH}_3\text{CH}_2\text{O}$), 1.30 (m, 14H, CH_2), 1.61 (m, 2H, CH_2), 2.04 (m, 2H), 2.28 (t, J = 7.5 Hz, 2H, CH_2COOEt), 2.74 (m, 1H, CHD), 4.12 (q, J = 7.0 Hz, 2H, OCH_2CH_3), 5.34 (m, 4H, $\text{CH}=\text{CH}$). ^{13}C NMR (100 MHz, CDCl_3) δ 14.1, 14.2, 22.6, 25.0, 25.3 (t, J = 19.5 Hz), 27.17, 27.19, 29.08, 29.09, 29.14, 29.3, 29.6, 31.5, 34.4, 60.1, 127.8, 128.0, 130.0, 130.2, 173.9.

[0142] To obtain the free $[11\text{-}^2\text{H}]$ -linoleic acid (**616**), to the solution of the linoleic acid ethyl ester (0.704 g, 2.3 mmol) in ethanol (10 ml) a solution of KOH (0.4 g, 7.1 mmol) in water (0.8 ml) was added. The mixture was stirred at 50°C for 10 min and then diluted with water (20 ml), treated with 10 % solution of sulphuric acid (5 ml) and extracted with Et_2O (4 x 20 ml). The combined organic fractions were washed with saturated NaCl, dried over Na_2SO_4 , and the solvent was removed at reduced pressure. The residue was flushed through a small volume of silica gel (2 ml; eluent: petroleum ether:EtOAc (2:1)) and the solvent removed *in vacuo* to yield 0.629 g (98 %) of the indicated acid **616**. ^1H NMR (400 MHz, CDCl_3) δ 0.88 (t, J = 7.0 Hz, 3H, CH_3), 1.30 (m, 14H, CH_2), 1.60 (m, 2H, CH_2), 2.03 (m, 4H, CH_2), 2.33 (t, J = 7.5 Hz, 2H, CH_2COOEt), 2.74 (m, 1H, CHD), 5.32 (m, 4H, $\text{CH}=\text{CH}$), 11.6 (br s, 1H, COOH). ^{13}C NMR (100 MHz, CDCl_3) δ 14.1, 22.6, 24.6, 25.3 (t, J = 19.0 Hz), 27.16, 27.18, 29.00, 29.05, 29.12, 29.3, 29.6, 31.5, 34.0, 127.8, 128.0, 130.0, 130.2, 180.1.

Example 7. Synthesis of $[11\text{-}^{13}\text{C}]$ -Linoleic Acid (reference)

[0143]





[0144] [1-¹³C]-oct-2-yn-1-ol (717). The title compound has been synthesized according to the earlier described protocols (Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138) using ¹³C-paraform, and used without further purification. ¹H NMR (CDCl₃, δ): 4.22 (

д, J = 148 Hz, 2H), 2.18 (td, J₁ = 7.0, J₂ = 1 Hz, 2H), 1.91 (br s, 1H), 1.47 (quint, J = 7.0 Hz, 2H), 1.31 (m, 4H), 0.87 (t, J = 7.0 Hz, 3H).

[0145] [1-¹³C]-1-bromooct-2-yne (718) was synthesized as described in (Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138). Yield: 82% starting from ¹³C-paraform (per two steps). ¹H NMR (CDCl₃, δ): 3.93 (dt, J₁ = 158 Hz, J₂ = 2 Hz, 2H), 2.23 (m, 2H), 1.50 (m, 2H), 1.33 (m, 4H), 0.89 (t, J = 7 Hz, 3H).

[0146] [11-¹³C]-ethyl octadeca-9,12-diynoate (719). was synthesized as previously described (See Meyer, M. P.; Klinman, J. P. *Tetrahedron Lett.* 2008, 49, 3600; Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138). Yield: 93%. ¹H NMR (CDCl₃, δ): 4.10 (q, J = 7 Hz, 2H), 3.1 (dm, J = 134 Hz, 2H), 2.27 (t, J = 7.5 Hz, 2H), 2.13 (m, 4H), 1.60 (m, 2H), 1.47 (m, 4H), 1.3 (m, 10H), 1.24 (t, J = 7 Hz, 3H), 0.88 (t, J = 7.0 Hz, 3H).

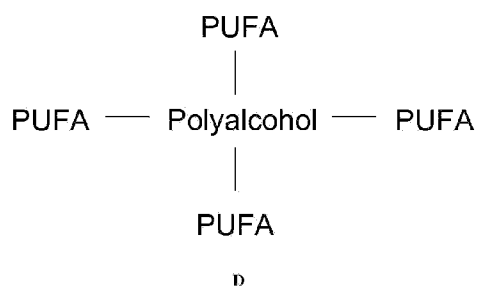
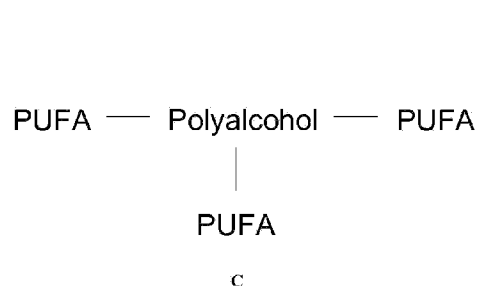
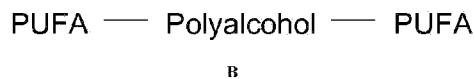
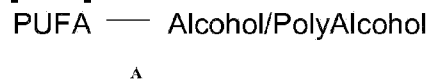
[0147] [11-¹³C]-linoleic acid ethyl ester (720) was synthesized as previously described (See Meyer, M. P.; Klinman, J. P. *Tetrahedron Lett.* 2008, 49, 3600; Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50 (1), 130-138). Yield: 56%. ¹H NMR (CDCl₃, δ): 5.34 (m, 4H), 4.12 (q, J = 7 Hz, 2H), 2.77 (dm, J = 126 Hz, 2H), 2.28 (t, J = 7.5 Hz, 2H), 2.04 (m, 4H), 1.61 (m, 2H), 1.30 (m, 14H), 1.25 (t, J = 7 Hz, 3H), 0.88 (t, J = 7.0 Hz, 3H).

[0148] [11-¹³C]-linoleic acid (721) was synthesized as previously described (See Meyer, M. P.; Klinman, J. P. *Tetrahedron Lett.* 2008, 49, 3600; Hill, Sh.; Hirano, K.; Shmanai, V. V.; Marbois, B. N.; Vidovic, D.; Bekish, A. V.; Kay, B.; Tse, V.; Fine, J.; Clarke, C. F.; Shchepinov, M. S. *Free Radic. Biol. Med.*, 2011, 50(1), 130-138); yield 98%. ¹H NMR (CDCl₃, δ): 10.5 (br s,

1H), 5.34 (m, 4H), 2.77 (dm, J = 126 Hz), 2.33 (t, J = 7.5 Hz, 2H), 2.03 (m, 4H), 1.60 (m, 2H), 1.30 (m, 14H), 0.88 (t, J = 7.0 Hz, 3H).

Example 8. General Preparation of Esters A-D

[0149]



[0150] General Procedure for Compound A. Thionyl chloride (2 equivalents) is slowly added to a solution of PUFA (1 equivalent) in CHCl_3 . The reaction mixture is heated to reflux for 1 hr, then it is allowed to cool to room temperature and the solvent is evaporated under reduced pressure to afford the carboxylic acid chloride derivative of the PUFA. The carboxylic acid chloride derivative is then dissolved in anhydrous pyridine and the alcohol (1 equivalent) dissolved in pyridine is slowly added (Note that the order of addition is reversed when the alcohol is a polyalcohol). Upon complete addition, the reaction mixture is allowed to stir at room temperature for 24 hr. The solvent is then removed under reduced pressure and the crude product is purified by column chromatography to afford Compound A.

[0151] 11,11 -Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**30**); 14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**23**); 11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**17**); and 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid (**7**) are each subjected to the above described procedure with the following alcohols: ethanol, glycerol, propylene glycol; glucose; 2-(2-ethoxyethoxy)ethanol; and estradiol to afford products corresponding to the general formula of Compound A.

[0152] General Procedure for Compound B. Thionyl chloride (2 equivalents) is slowly added to a solution of PUFA (1 equivalent) in CHCl_3 . The reaction mixture is heated to reflux for 1 hr, then it is allowed to cool to room temperature and the solvent is evaporated under reduced pressure to afford the carboxylic acid chloride derivative of the PUFA. The carboxylic acid chloride derivative is then dissolved in anhydrous pyridine and the alcohol (Compound A, 1 equivalent) dissolved in pyridine is slowly added. Upon complete addition, the reaction mixture is allowed to stir at room temperature for 24 hr. The solvent is then removed under reduced

pressure and the crude product is purified by column chromatography to afford Compound B.

[0153] The Compound A products that form from the condensation of 11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**30**); 14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**23**); 11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**17**); and 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid (**7**) with glycerol, propylene glycol; glucose; and estradiol are treated according to the above-described general procedure with 11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**30**); 14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**23**); 11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**17**); and 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid (**7**) as the PUFAs to afford products corresponding to the general formula of Compound B.

[0154] General Procedure for Compound C. Thionyl chloride (2 equivalents) is slowly added to a solution of PUFA (1 equivalent) in CHCl_3 . The reaction mixture is heated to reflux for 1 hr, then it is allowed to cool to room temperature and the solvent is evaporated under reduced pressure to afford the carboxylic acid chloride derivative of the PUFA. The carboxylic acid chloride derivative is then dissolved in anhydrous pyridine and the alcohol (Compound B, 1 equivalent) dissolved in pyridine is slowly added. Upon complete addition, the reaction mixture is allowed to stir at room temperature for 24 hr. The solvent is then removed under reduced pressure and the crude product is purified by column chromatography to afford Compound C.

[0155] The Compound B products that form from the condensation of Compound A products with glycerol and glucose are treated according to the above-described general procedure with 11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**30**); 14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**23**); 11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**17**); and 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid (**7**) as the PUFAs to afford products corresponding to the general formula of Compound C.

[0156] General Procedure for Compound D. Thionyl chloride (2 equivalents) is slowly added to a solution of PUFA (1 equivalent) in CHCl_3 . The reaction mixture is heated to reflux for 1 hr, then it is allowed to cool to room temperature and the solvent is evaporated under reduced pressure to afford the carboxylic acid chloride derivative of the PUFA. The carboxylic acid chloride derivative (4 equivalents) is then dissolved in anhydrous pyridine and the alcohol (1 equivalent) dissolved in pyridine is slowly added. Upon complete addition, the reaction mixture is allowed to stir at room temperature for 24 hr. The solvent is then removed under reduced pressure and the crude product is purified by column chromatography to afford Compound D.

[0157] The Compound C products that form from the condensation of Compound B products with glucose are treated according to the above-described general procedure with 11,11-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**30**); 14,14-Dideutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**23**); 11,11,14,14-Tetradeutero-cis,cis,cis-octadeca-8,12,15-trienoic acid (**17**); and 11,11-Dideutero-cis,cis-octadeca-9,12-dienoic acid (**7**) as the PUFAs to afford products corresponding to the general formula of Compound D.

Example 9. ^1H - and ^{13}C -NMR analysis of deuterated PUFAs described in Examples 1-4 (Figure 2).

[0158] Characteristic areas of ^1H and ^{13}C spectra, all values in ppm. (*Panel A*) Deuteration of Lin acid at pos. 11 is confirmed by the disappearance of peaks in ^1H and ^{13}C NMR spectra. Disappearance of the peak at δ_{H} 2.764 is expected due to absence of H atoms (^1H NMR). Disappearance of the peak at δ_{C} 25.5 in is due to combination of Nuclear Overhauser Effect, and splitting of this particular carbon atom into a quintet by two D atoms in the deuterated form of Lin acid. (*Panel B*) The ^1H NMR spectrum shows that the H atoms at C11 and C14 positions of site-specifically deuterated αLn coincide (δ_{H} 2.801) thus deuteration at either site (11,11- H_2 , 14,14- D_2 or 11,11- D_2 , 14,14- H_2) leads to a 50% decrease in integration of this peak, while deuteration of both sites (11,11,14,14- D_4) leads to the complete disappearance of the peak at δ_{H} 2.801. However, ^{13}C NMR experiments can clearly distinguish between the three deuterated forms, as the observed peaks for C11 and C14 positions are separated by a small but detectable difference. Thus, deuteration at either C11 or C14 positions leads to disappearance of the peak at δ_{C} 25.68 or δ_{C} 25.60, respectively, while deuteration at both sites leads to disappearance of the two corresponding peaks.

Example 10. Isotope Reinforcement Can Shut Down PUFA peroxidation

[0159] Q-less yeast (coq mutants) provide an ideal system to assess in vivo autoxidation of fatty acids. Coenzyme Q (ubiquinone or Q) serves as a small lipophilic antioxidant as well as an electron shuttle in the respiratory chain of the mitochondrial inner membrane. Ten *S. cerevisiae* genes (COQ1-COQ10) are required for coenzyme Q biosynthesis and function, and the deletion of any results in respiratory deficiency (Tran UC, Clarke CF. Mitochondrion 2007;7S,S62). It was shown that the coq yeast mutants are exquisitely sensitive to autoxidation products of PUFAs (Do TQ et al, PNAS USA 1996;93:7534-7539; Poon WW, Do TQ, Marbois BN, Clarke CF. Mol. Aspects Med. 1997;18,s121). Although *S. cerevisiae* do not produce PUFAs (Paltauf F, Daum G. Meth. Enzymol. 1992;209:514-522), they are able to utilize PUFAs when provided exogenously, allowing their content to be manipulated (Paltauf F, Daum G. Meth. Enzymol. 1992;209:514-522). Less than 1% of Q-less (coq2, coq3, and coq5) yeast mutants is viable following a four hour treatment with linolenic acid (Do TQ et al, PNAS USA 1996;93:7534-7539; Poon WW, Do TQ, Marbois BN, Clarke CF. Mol. Aspects Med. 1997;18,s121). In contrast, 70% of wild-type (the parental genetic background is strain W303-1B) cells subjected to this treatment remain viable. The Q-less yeast are also hypersensitive to other PUFAs that readily autoxidize (such as arachidonic acid), but behave the same as the wild-type parental strain to treatment with the monounsaturated oleic acid (Do TQ et al, PNAS USA 1996;93:7534-7539). The hypersensitivity of the Q-less yeast mutants is not a secondary effect of the inability to respire, because cor1 or atp2 mutant yeast (lacking either the bc1

complex or the ATP synthase, respectively) show wild-type resistance to PUFA treatment (Do TQ et al, PNAS USA 1996;93:7534-7539; Poon WW, Do TQ, Marbois BN, Clarke CF. Mol. Aspects Med. 1997;18,s121).

[0160] A plate dilution assay can be used to assess PUFA sensitivity. This assay can be performed by spotting serial five-fold dilutions of aliquots onto YPD plate media (**Fig. 3**). The sensitivity of the different strains can be observed by visual inspection of the density of cells in each spot.

[0161] Treatment with linolenic acid causes a dramatic loss of viability of the *coq* null mutants. In stark contrast, *coq* mutants treated with the D4-linolenic acid were not killed, and retained viabilities similar to yeast treated with oleic acid. Quantitative colony counting revealed that the viability of cells treated with oleic and D4-linolenic was similar (**Fig. 4**), while the viability of the *coq* mutants was reduced more than 100-fold following treatment with the standard linolenic acid for 4h. These results indicate that isotope-reinforced linolenic acid is much more resistant to autoxidation than is the standard linolenic acid, as evidenced by the resistance of the hypersensitive *coq* mutants to cell killing.

Example 11. GC-MS Can Detect Fatty Acids and PUFAs in Yeast Cells

[0162] Yeast do not synthesize PUFAs, however they do incorporate exogenously supplied linoleic and linolenic acids (Avery SV, et al. Applied Environ. Microbiol. 1996; 62,3960; Howlett NG, et al. Applied Environ. Microbiol. 1997; 63,2971). Therefore, it seems likely that yeast would also incorporate exogenously supplied D4-linolenic acid. However, it is possible that the differential sensitivity to linolenic and D4-linolenic might be attributed to differences in integration into the cell rather than autoxidation. To test whether this is the case, the extent of uptake of this fatty acid was monitored. First the conditions of separation of fatty acid methyl esters (FAME) of C18:1, C18:3, D4-18:3 and C17:0 (to be used as an internal standard) were determined. The GC-MS chromatogram shown in **Fig. 5** establishes both separation and sensitivity of detection of these fatty acid methyl ester standards.

[0163] Wild-type yeast were harvested during log phase growth and incubated in the presence of exogenously added fatty acid (for 0 or 4 h) in the presence of phosphate buffer plus 0.20% dextrose, as described for the fatty acid sensitivity assay. Cells were harvested, washed twice with 10 ml sterile water, and the yeast cell pellets were then processed by alkaline methanolysis as described above. The fatty acids are detected as methylesters (FAMEs) following GC-MS with C17:0 added as an internal standard (**Fig. 6**). The amounts of 18:3 and D4 detected after 4 h incubation were extrapolated from the calibration curve. These results indicate yeast avidly incorporate both linolenic and D4-linolenic acid during the 4 h incubation period. Based on these results, it is obvious that the enhanced resistance of the *coq* mutant yeast to treatment with D4-C18:3 is not due to lack of uptake.

[0164] D2-linolenic, 11, 11-D2-linolenic acid and 14, 14-D2-linolenic acid, were also used on

this yeast model and rendered comparable protection.

Example 12. Kinetic isotope effect in non-enzymatic oxidation of D2-LA in a chain reaction format.

[0165] The kinetics of oxygen consumption during the oxidation of LA and D2-LA was studied with a glass capillary microvolumeter (Figure 7). The rate of oxidation, R_{OX} , was measured as a slope of $[O_2]$ traces. The rate of initiation, R_{IN} , was determined by the inhibitor method with HPMC ("6-hydroxy-2,2,5,7,8-pentamethylbenzochroman") as a reference inhibitor. R_{IN} was calculated from the induction period of inhibited oxidation, t_{IND} : $R_{IN} = 2 \cdot [HPMC]/t_{IND}$. The rate of oxidation of 0.71 M LA (Fig. 7) was found to be 6.1×10^{-6} M/s. When the process was inhibited by 0.23 mM chain-breaking antioxidant HPMC, the duration of the induction period, t_{IND} , was about 48 min, with the R_{IN} value of around 0.16×10^{-6} M/s. The length of the kinetic chain calculated from these data was: $\nu = R_{OX}/R_{IN} = 38 \pm 3$. Based on this data, the calculated oxidizability of LA was $0.0215 \pm 0.008 \text{ M}^{-0.5}\text{s}^{-0.5}$ ($n = 5$) [Cosgrave J.P, et. al. Lipids, 1987, 22, 299-304]. For D2-LA, the reduction of R_{OX} to 0.18×10^{-6} M/s was observed (Figure 7). In contrast to LA, addition of HPMC did not result in the decrease in R_{OX} and the appearance of any detectable induction period (data not shown). The latter precludes a direct determination of R_{IN} . For a R_{IN} value for D2-LA oxidation being comparable to that of LA it follows that D2-LA oxidation was not a chain process ($\nu = 0.18 \times 10^{-6}/0.16 \times 10^{-6} \approx 1.1$). An estimated kinetic isotope effect ("KIE"), from comparison of R_{OX} for LA and D2-LA, was around $6.1 \times 10^{-6}/0.18 \times 10^{-6} \approx 35$. A similar KIE was determined during the oxidation of LA and 11,11-d₂-LA in Triton X-100 aqueous micelles (data not shown). For comparative purposes, the theoretical KIE is 6.9 at 25° C. See Carpenter, "Determination of Organic Reaction Mechanisms" (John Wiley & Sons, 1984), p. 89.

Example 13. Small Amounts of D2-LA Protect LA Against Peroxidation.

[0166] To simulate the likely *in vivo* conditions, the kinetics of the oxidation of the mixtures of D2-LA and LA were studied (Figure 8). In the experiments, the concentration of LA plus 11,11-d₂-LA was 0.775 M; the concentration of AMVN was 0.0217 M; and the reactions were carried out at 37° C. The results afforded an R_{IN} of $1.10 \pm 0.08 \times 10^{-7}$ M/sec. Additionally, the rate of oxidation of the mixtures was found to be non-additive and much lower than the additive value of R_{OX} for the individual compounds. Surprisingly, D2-LA essentially 'protects' the non-deuterated LA against autoxidation. A qualitatively similar effect was also observed during the oxidation of the mixture of 11,11-D2-LA with non-deuterated methyl linoleate (data not shown). These results suggest that even a partial replacement of non-deuterated LA by D2-LA may substantially slow down PUFA peroxidation.

Example 14. Small amounts of D2-LA Protect LA Against Peroxidation *In Vivo*.

[0167] The results described in Example 13 were reproduced *in vivo* using Q-less *coq3* yeast strains and different ratios of LA to D2-LA (Figure 9). Wild-type, yeast Q-less *coq3*, or respiratory deficient *cor1* null mutants were incubated in the presence of 200 μ M of LA and D2-LA at different ratios of PUFAs, as indicated in Fig. 9. Serial dilutions (1:5) starting at 0.20D/ml were spotted on YPD solid plate medium. Additionally, a zero-time untreated control was utilized and the results are shown on the top left of Fig. 9. Growth was at 30° C. The results indicate that approximately 10-15% of D2-LA was a sufficiently minimal amount to cancel the toxicity of LA. A similar incubation with the mono-deuterated PUFA, 11,11-D,H-LA, afforded no detectable loss in cell viability after 3 hours of treatment (data not shown). These results suggest that both D2-LA and 11,11-D,H-LA were resistant to lipid peroxidation.

[0168] Wild-type yeast cells were treated as described above except the yeast were treated with 200 μ M of the designated fatty acid for 2 hours, washed with sterile water, and were either not treated (triangles) or treated with 50 μ M CuSO₄ (squares) at room temperature. After 60 min of copper treatment cells were treated with 8 μ M C11-Bodipy 581/591 for 30 min at room temperature. Four 100 μ l aliquots were plated in a 96-well plate and the fluorescence was measured. Wild-type yeast cells treated with copper in the absence or presence of PUFA have significantly higher levels of lipid peroxidation as compared to yeast not treated with copper. However, copper-stressed wild-type yeast cells treated with 11,11-D₂-LA have lower levels of lipid peroxidation similar to yeast not treated with PUFA. Mono-deuterated 11,11-D,H-LA offered similar protection.

Example 15. Small Amounts of D4-ALA Protect ALA Against Peroxidation *In Vivo*.

[0169] The experimental protocol described for Example 14 was also reproduced *in vivo* using Q-less *coq3* yeast strains (Figure 10) and different ratios of ALA to D4-ALA. Wild-type, yeast Q-less *coq3*, or respiratory deficient *cor1* null mutants were incubated in the presence of 200 μ M of ALA and D4-Lnn (Linolenic acid) at different ratios of PUFAs, as indicated in Fig. 10. Serial dilutions (1:5) starting at 0.2OD/ml were spotted on YPD solid plate medium. Growth was at 30° C. The results indicate that approximately 15-20% of D2-Lnn was a sufficiently minimal amount to cancel the toxicity of ALA. Moreover, results indicate that the content of PUFA taken up by yeast cells roughly reflects the ratios added and suggests that yeast cells do not discriminate among the PUFAs provided.

Example 16. D-PUFA mitigates oxidative stress and increases survival in retinal cells implicated in AMD and Diabetic Retinopathy pathology

[0170] Several cell types, including microvascular endothelium (MVEC), retinal pigment epithelium (RPE) and retinal neurons (retinal ganglion cells) were tested for survival in cell culture. Cells were kept in the medium containing either hydrogenated (control) or deuterated D2-linoleic (ω -6; LA) and D4-linolenic (ω -3; ALA) acids (20 μ M; ratio of ω -6 to ω -3: 1:1 or 2:1) for 72 hrs. The incorporation of PUFAs into cells was monitored by GC (**Figure 11**). PUFAs were shown to be readily taken up by cells according to the Table 1, showing incorporation of PUFAs into MVECs.

Table 1

		Area unlabelled	Area labelled	ratio
control	linoleate	78392976	4556042	0.058
	linolenate	1488866	149411	0.100
PUFA	linoleate	96026830	5525295	0.058
	linolenate	2347729	113468	0.048
Deuterated PUFA	linoleate	34957060	2599969	0.074
	linolenate	747128	134824	0.180

[0171] The cells were then treated with paraquat (PQ; 500 μ M), a common oxidative stress-generating compound. For survival measurement, cells were counted using haemocytometer and trypan blue exclusion method. **Figure 12** shows the survival of H- and D-PUFA treated MVEC cells after acute intoxication by paraquat. For all cell types tested, D-PUFA had protective effect compared to controls, similar to that shown in Figure 8 for MVEC cells.

Example 17. Toxicology studies of mice supplemented with D-PUFA reveal no anomalies in major blood biomarkers.

[0172] With a more protracted dosing paradigm (i.e. 3 weeks of dietary replacement), chemical analysis of blood serum of H-PUFA- and D-PUFA-supplemented mice (performed at UC Davis) revealed no difference in major biomarkers of renal function, liver function, blood lipids, etc for H-PUFA/D-PUFA saline treated mice. In this example, D-PUFA is a 2:1 mixture of D2-linoleic acid: D4-linolenic acid.

[0173] Tested parameters included measurements of triglycerides; total protein; total bilirubin; phosphorus; free fatty acids; HDL; glucose; creatine; cholesterol; calcium; blood urea nitrogen; alkaline phosphatase; albumin; aspartate aminotransferase; and others in Table 2.

Table 2

Triglyceride mg/l
Total Protein g/l
Total Bilirubin mg
Phosphorus mg
Non-saturated Fatty Ac
High Density Lipoprote
Glucose mg/dl
Creatinine mg/dl
Cholesterol mg/l
Calcium mg/dl
Blood Urea Nitrogen
Alkaline Phosphatase
Albumin g/dl
Aspartate Aminotransferase
Alanine Aminotransferase
Sample volume
Mouse ID #

		ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L	ase U/L
4	100	273.0	3008.7	3.09	81.7	19.1	7.96	148.3	0.189	160.2	104.49	1.08	13.07	0.185	5.32	38.9
5	110	5726.7	8478.9	3.42	31.1	25.4	7.40	185.1	0.356	355.6	134.37	1.07	18.59	0.275	6.56	57.9
7	100	156.0	1470.6	2.82	35.1	18.9	7.64	151.2	0.154	174.6	107.39	1.11	10.14	0.192	5.26	82.7
10	60	518.4	4653.0	3.02	QNS	20.1	6.78	184.0	0.151	136.5	138.15	1.06	QNS	0.272	6.07	46.1
11	70	144.0	1635.3	3.63	72.7	20.3	8.75	170.8	0.179	107.9	139.86	1.18	9.33	0.162	5.72	33.5
13	14	3518.1	15669.0	QNS	<0.1	31.5	QNS	166.5	1.126	178.4	135.09	0.99	QNS	QNS	QNS	31.5
14	75	216.9	2107.8	3.03	42.4	24.4	7.46	173.6	0.170	93.3	47.76	1.06	10.41	0.235	6.07	43.8
25	75	589.5	4707.0	3.20	18.8	18.0	5.97	193.4	0.126	164.5	147.96	1.01	18.39	0.269	6.74	41.0
27	100	727.2	6015.6	2.63	<0.1	36.2	5.71	166.7	1.453	88.3	98.46	0.87	24.57	0.301	6.26	26.9
28	100	468.9	4018.5	2.93	49.3	21.2	6.90	164.4	0.232	224.9	50.54	1.02	14.16	0.231	5.87	49.6
29	29	1898.1	12510.0	QNS	QNS	24.9	QNS	208.8	0.111	QNS	77.58	0.20	QNS	QNS	QNS	27.9
30	100	2963.7	5971.2	3.38	50.3	18.2	6.25	174.7	0.225	227.4	131.04	1.17	21.42	0.349	6.28	46.7
Mean D-PUFA	76	1508	5289	3.17	52.6	22.6	7.67	158.5	0.332	172.1	115.30	1.06	12.31	0.220	5.83	47.8
SD D-PUFA	33	2225	5189	0.30	23.0	4.6	0.66	14.5	0.357	87.0	33.21	0.06	3.76	0.048	0.50	17.7
Mean H-PUFA	81	1329	6524	3.04	39.5	23.7	6.22	181.6	0.429	176.3	101.12	0.85	19.64	0.288	6.29	38
SD D-PUFA	31	1078	3428	0.33	17.9	8	0.51	19.0	0.575	65.5	39.40	0.38	4.44	0.050	0.36	11

Example 18. Histopathologic Studies

[0174] Microscopic changes were coded by the most specific topographic and morphologic diagnosis, and the Systematized Nomenclature of Medicine (SNOMED) and the National Toxicology Program's Toxicology Data Management System (TDMS) terminology manuals were used as guidelines. Data were recorded in Labcat® Histopathology module 4.30. A four-step grading system (minimal, mild, moderate, and marked) was used to define gradable changes.

[0175] C57BL6 male mice were dosed orally in the diet with PUFAs on Study Days 1 through 14, and were necropsied on Study Day 15. Group 1 consisted of 4 mice and received hydrogenated PUFAs. Group 2 consisted of 5 mice and received deuterated PUFAs (D2-LA and D4-ALA) On Study Day 8, all mice received intraperitoneal (IP) saline. Complete sets of protocol-specified tissues [liver (3 to 7 sections), lungs with bronchi (2 to 5 lobes), spleen, heart, and kidneys] from all submitted mice were examined histopathologically. No difference was observed between the H-PUFA and D-PUFA groups.

Example 19. Evaluation of Tissue-specific Deuteration

[0176] WT mice were housed at 12 animals (males separate from females) per cage and fed for 90 days ad libitum (typically, 5-6 g/day) on the AIN 93 diet, as pellets, with 6% total fat.

Approximately 10% of that total fat was made up of 1:1 mixture of D2-LA/D4-ALA (group 1), D2-LA/ALA (group 2), or LA/ALA (control group). The animals were sacrificed, organs harvested and stored at low temperature prior to analysis without the use of preservation agents. Lipid fractions were separated, pre-treated and analyzed by LC-MS according to standard protocols using LA, D2-LA, ALA and D4-ALA as controls.

[0177] Dosage studies of 1:1 D2-LA/D4-ALA indicated that tissues became highly enriched in deuterium, with about 40% of the total fat being deuterated (**Fig. 13**). Moreover, these studies indicated that fat distribution remained relatively unchanged by the tested dosage (**Fig. 14**). After dosage studies of 1:1 D2-LA/ALA, it was determined that about 27% of the total fat was deuterated (**Fig. 15**).

[0178] Specific organs, such as the liver and brain, were also evaluated (**Figs. 16-21**). While the liver had a different fat profile than previous tissues studied (**Fig. 16**), 90 day dosage studies with D2-LA/D4-ALA demonstrated that tissues became highly enriched in deuterium, with about 40% of the total fat being deuterated (**Fig. 17**). Moreover, the liver study indicated that fat distribution remained relatively unchanged by the tested dosage (**Fig. 16-17**). Additionally, 90 day dosage studies with D2-LA only illustrated a similar fat profile as previous studies, along with about 32% total fat being deuterated (**Fig. 18**). Consequently, fat profiles and deuteration profiles in the liver were maintained regardless of the administered deuterated component. Like the liver, the brain also had a different fat profile than previous tissues studied (**Figs. 19-21**). 90 day dosage studies with D2-LA/D4-ALA demonstrated that tissues became highly enriched in deuterium, with about 30% of the total fat being deuterated (**Fig. 19**). Moreover, the brain study indicated that fat distribution remained relatively unchanged by the tested dosage (**Figs. 19-21**). Additionally, 90 day dosage studies with D2-LA/ALA illustrated a similar fat profile as previous studies, along with about 23% total fat being deuterated (**Fig. 20**). Consequently, fat profiles and deuteration profiles in the brain were maintained regardless of the administered deuterated component.

Example 20: Testing PUFA Incorporation into Eye Tissues and Cells

[0179] Isotope ratio Mass-spectrometry can be used to confirm incorporation of D-PUFA into the phospholipid membranes of eye tissues. When delivering D2-LA and D4-ALA through dietary supplementation, incorporation into animal tissues can be monitored using an isotope ratio mass-spectrometry technique that will allow for measurement of the total increase in deuterium composition in lipid membranes, thus reporting on incorporation of D2-LA, D4-ALA, and any other PUFA derived from these compounds. Using this method, a substantial uptake of D-PUFA into mouse eye tissue can be detected. For example, mice are supplemented with D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg of D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg of LA, ALA, and 1:1 combinations of both LA and ALA) as the only PUFA source for 6 days, exposed acutely to a known oxidant or saline vehicle and continued on the same diet for an additional 6 days. Eyes are removed and dissected, and homogenate samples from saline-treated mice and test

compound-treated mice are analyzed for deuterium content as described for Example 7 above. D2-LA and D4-ALA are expected to be found in the tissues and cells analyzed.

Example 21: Testing for Efficacy Against Age-related Macular Degeneration

[0180] Retinal pigment epithelium (RPE) cells from human donor eyes are cultured according to known methods and exposed to D-PUFA (0.01, 0.1, 1.0, 10.0, 100, and 1000 μ M D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, 100, and 1000 μ M LA, ALA, and 1:1 combinations of both LA and ALA). At several time points (24, 28, and 72 hours), digital images are taken and analyzed with respect to lipofuscin and pigmentation according to known methods. D2-LA and D4-ALA are expected to prevent formation of lipofuscin.

[0181] The treatment-dependent ability of RPE cells to phagocytose, which is a crucial function, can also be evaluated. Retinal pigment epithelium (RPE) cells from human donor eyes are cultured according to known methods and exposed to D-PUFA (0.01, 0.1, 1.0, 10.0, 100, and 1000 μ M D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, 100, and 1000 μ M LA, ALA, and 1:1 combinations of both LA and ALA). After approximately three weeks, fluorescent latex beads are added to the cell cultures for approximately four hours. The cells could be washed to remove non-phagocytosed beads and fixed. Fluorescent images of the cells are taken and analyzed for lipofuscin content and the number of phagocytosed beads. D2-LA and D4-ALA are expected to prevent formation of lipofuscin yet allow for the cells to retain their ability to phagocytose.

[0182] The effects of lipofuscin in vivo, i.e. in the living eye of experimental rats after an intravitreal injection of the compounds can also be investigated. D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA) are injected intravitreally in half-year old Wistar rats. The rats are then be evaluated using electroretinography for functional testing and by counting lipofuscin particles. D2-LA and D4-ALA are expected to decrease lipofuscin content.

[0183] An animal model of dry AMD (See Justilien V, Pang, J P, et al, SOD2 Knockdown Mouse Model of Early AMD, Invest Ophthalmol V is Sci, 2007, 48 (10), 4407-4420) is known which uses mice that have had a ribozyme that targets the protective enzyme manganese superoxide dismutase (MnSOD) injected beneath the retina. The ribozyme (AAV-Rz 432) method would allow for site specific somatic knockdown of SOD 2 expression in normal adult tissue so that the lesion is only in the injected eye and the rest of the body is unaffected. Thus the animal's other eye can act as a control. This protocol has been shown to induce in the eyes of the mice after a single subretinal injection of the ribozyme many of the abnormalities seen in the eyes of patients with AMD, with the most severe changes seen 120 days post injection. The surrogate markers for protection from reactive oxygen and nitrogen species damage can be examined by electroretinography (ERG) and morphometry.

[0184] For example, full field ERG analysis may be used to assess the loss of rod and cone function and measuring a-wave and b-wave amplitudes in dark adapted and light adapted mice. Digital fundus imaging may be used to monitor atrophic changes in the retinal pigment epithelium (RPE). Additionally, light microscopy may be used to qualitatively assess damage to the photoreceptors and RPE, to look for evidence of choroidal neovascularization and to measure changes in the thickness of Bruch's membrane. Retinal degeneration is determinable morphometrically by measuring the thickness of the outer nuclear layer and rod outer segments (See Goto Y, Peachey N S, Ripps H, Naash M I. Functional abnormalities in transgenic mice expressing a mutant rhodopsin gene, *Invest Ophthalmol Vis Sci* 1995, 36, 62-71; Faktorovich E G, Steinberg R H, Yasumura D, Matthes M T, LaVail M M. Photoreceptor degeneration in inherited dystrophy delayed by the basic fibroblast growth factor, *Nature* 1990, 347, 83-86).

[0185] Accordingly, D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA) are administered to the mouse model of dry AMD described in Justilien V, Pang, J P, et al, SOD2 Knockdown Mouse Model of Early AMD, *Invest Ophthalmol V is Sci*, 2007, 48 (10), 4407-4420). The surrogate markers for protection from reactive oxygen and nitrogen species damage are examined by electroretinography (ERG) and morphometry. D2-LA and D4-ALA are expected to protect against damage caused by reactive oxygen and nitrogen species.

[0186] As described above, new blood vessel growth and leakage are thought to facilitate wet AMD disease progression. As a representative model, artificial corneal burns can be induced in rat eyes to determine the effects of vehicle solution, non-stabilized test compounds, or test compounds at various concentrations on corneal neovascularization. More specifically, topical administration of vehicle solution, non-stabilized test compounds, or test compounds at various concentrations could be administered twice a day to rats in which corneal burns had been artificially induced by application of silver nitrate (70%) and/or potassium nitrate (30%).

[0187] Accordingly, D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA) are daily administered to rats having artificial corneal burns. Over a three week period, the rats are examined for new blood vessel growth and leakage. At the end of the trial, the rats are euthanized, their eyes are dissected, and blood vessel growth and leakage are further examined. D2-LA and D4-ALA are expected to decrease the amount of new vessel growth and leakage.

Example 22: Testing for Efficacy Against Diabetic Retinopathy

[0188] Diabetes mellitus occurs in spontaneously diabetic Torii (SDT) rats at about 20 weeks of age. Accordingly, blood glucose levels can be confirmed at about 25 weeks of age, and rats

showing an increase in blood glucose level (at least 300 mg/dl) can be used as a model for determining a compound's effectiveness at preventing and/or mitigating the diabetic retinopathy. For example, male SDT rats that develop diabetes mellitus are divided into two groups, that is, 1) a control group and 2) a test compound treatment group. D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA) are mixed with the rats' usual feed controlled for LA and ALA and fed daily. At 36 weeks of age and at 52 to 58 weeks of age, the rats are anesthetized and their eyeballs are excised and used in histopathological examination.

[0189] The histopathological examination is carried out in the following manner. The rat is anesthetized with ether and by intraperitoneal administration of Nembutal, and then the eyeballs are excised. The excised eyeballs are placed in a mixed solution (1:1:2) consisting of 4% glutaraldehyde, 10% neutral formalin and a phosphate buffer (pH 7.2, 0.3 mol/l). After 60 minutes, the eyeball is cut into halves under a stereoscopic microscope and then stored overnight at 4° C. The next morning, the eyeballs are embedded in a usual manner into paraffin to prepare a transverse section containing a bundle of optic nerves. The section is then stained with hematoxylin-eosin and analyzed according to known methods. D2-LA and D4-ALA are expected to prevent the development of retinopathy.

[0190] Fluorescein fundus angiography is also carried out by opening the rats' chests under the same anesthesia, injecting fluorescein dextran (Sigma, 50 mg/1 ml PBS) into the heart, and after 5 minutes, excising the eyeballs. The retina is then separated from the excised eyeball, spread on a slide glass to prepare a flat mounted specimen of the retina, and observed with a stereoscopic fluorescent microscope and photographed to evaluate retinopathy.

[0191] If retinal fold (retinal detachment) accompanied by thickening of the retina around the optic nerve head is recognized in the pathological examination, or when fluorescence dye leakage accompanied by retinal vascular tortuosity and/or caliber variation around the optic nerve head is recognized in the fluorescein fundus photography, it is assumed that severe retinopathy occurred in the SDT rat.

Example 23: Testing for Efficacy Against Stargardt Disease

[0192] ABCR^{-/-} mice is an animal model of human recessive Stargardt's Disease. As in humans with recessive Stargardt's disease, ABCR^{-/-} mice accumulate large lipofuscin deposits in the RPE cells of their eyes and eventually experience delayed dark adaptation. Lipofuscin accumulation and vision loss observed in this mouse model is also thought to be relevant to age-related macular degeneration (AMD) and other macular dystrophies.

[0193] ABCR^{-/-} mice are raised on a diet of D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0,

10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA). After 1, 3, 5, and 7 weeks, mice are sacrificed, their eyes are dissected, and the retinas and eyecups are pooled and homogenated with ethanol. The homogenate is centrifuged, the supernatant is drawn off, and the supernatant is analyzed for A2E-lipofuscin by HPLC. D2-LA and D4-ALA are expected to impede lipofuscin formation.

Example 24: Testing for Efficacy Against Retinitis Pigmentosa

[0194] Because α -phosphodiesterase-related mutations in RP-patients play an important part in degeneration, the rd-1 mouse is a relevant and usable model for determining the efficacy of the compounds disclosed herein. In untreated rd-1 mice the mutation brings about apoptosis of rod receptor cells which begins on about postnatal day 10 (PN 10), and eventually leads to night blindness on about PN18. At PN18 only cone photoreceptor cells remain, which will also eventually die. For comparative purposes, normal wild-type mice which are genetically the same as the rd-1 mouse could be used.

[0195] Test animals could be treated in accordance with the European Community guideline (86/609/EEC). Mice of the CH3 line and homozygous retinal degeneration 1 (rd1/rd1) are used. On the third day following birth (PN3) test animals are treated with D-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg D2-LA, D4-ALA, and 1:1 combinations of both D2-LA and D4-ALA) or H-PUFA (0.01, 0.1, 1.0, 10.0, and 100 mg/kg LA, ALA, and 1:1 combinations of both LA and ALA) for two weeks. Animals are sacrificed one day after completion of the treatment period, whereafter their eyes are removed and analysed for the number of photoreceptors in on each side of the optic nerve using the so-called "Terminal dUTP nick-end labelling-assay (TUNEL)." Commercially available cell death detection kits are used to quantify the assay. D2-LA and D4-ALA are expected to protect against photoreceptor cell death.

[0196] Experiments are also performed on an organ culture of retina explants from rd-1 and control mice under serum-free conditions, as described in Caffè A R, Ahuja P, Holmqvist B, Azadi S, Forsell J, Holmqvist I, Soderpalm A K, van Veen T (2001) "Mouse retina explants after long-term culture in serum-free medium.", J Chem Neuroanat 22: 263-273. D2-LA and D4-ALA are expected to protect against cell death.

REFERENCES CITED IN THE DESCRIPTION

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P a t e n t k r a v

- 5 **1.** Polyumættet stof til anvendelse til behandling af våd eller tør aldersrelateret makuladegeneration (AMD), retinitis pigmentosa (RP), diabetisk retinopati (DR), katarakter eller stargardt-sygdom (SD), hvor det polyumættede stof omfatter en fedtsyre eller fedtsyreester, der er isotopisk modificeret med deuterium ved en eller flere positioner;
hvor det polyumættede stof er inkorporeret i en patients krop efter indgivelse;
og
- 10 hvor mængden af den isotopisk modificerede fedtsyre eller fedtsyreester omfatter mellem fra 5 % til 75 % af den samlede mængde fedtsyrer eller fedtsyreestere, der indgives til eller indtages af patienten.
- 15 **2.** Polyumættet stof til anvendelse ifølge krav 1, hvor fedtsyren eller fedtsyreesteren er isotopisk modificeret med mindst et deuteriumatom ved en bis-allylisk position, hvor det mindst ene deuteriumatom forekommer i et niveau, der er signifikant over det naturligt forekommende abundansniveau for deuterium.
- 20 **3.** Polyumættet stof til anvendelse ifølge krav 2, hvor den isotopisk modificerede fedtsyre eller fedtsyreester omfatter mellem 20 % og 50 % af den samlede mængde af fedtsyrer eller fedtsyreestere, der indgives til patienten.
- 25 **4.** Polyumættet stof til anvendelse ifølge krav 2, hvor en celle eller væv af patienten bibeholder en tilstrækkelig koncentration af den isotopisk modificerede fedtsyre eller fedtsyreester til at forhindre autooxidation af den naturligt forekommende polyumættede fedtsyre eller fedtsyreester.
- 30 **5.** Polyumættet stof til anvendelse ifølge krav 2, hvor det polyumættede stof er en omega-3-fedtsyre, omega-3-fedtsyreester, omega-6-fedtsyre eller omega-6-fedtsyreester.
- 6.** Polyumættet stof til anvendelse ifølge krav 5, hvor det polyumættede stof er udvalgt fra gruppen bestående af 11,11-D₂-linolensyre, 14,14-D₂-linolensyre,

11,11,14, 14-D4-linolensyre, 11,11-D2-linolsyre, 11-D-linolensyre, 14-D-linolensyre, 11,14-D2-linolensyre og 11-D-linolsyre.

5 **7.** Polyumættet stof til anvendelse ifølge krav 5, hvor det polyumættede stof er yderligere isotopisk modificeret ved en pro-bis-allylisk position.

8. Polyumættet stof til anvendelse ifølge krav 2, hvor det polyumættede stof er en fedtsyreester.

10 **9.** Polyumættet stof til anvendelse ifølge krav 8, hvor esteren er et triglycerid, diglycerid eller monoglycerid.

10. Polyumættet stof til anvendelse ifølge krav 1, hvor anvendelsen endvidere omfatter indgivelse af en antioxidant.

15 **11.** Polyumættet stof til anvendelse ifølge krav 10, hvor antioxidanten er coenzym Q, idebenon, mitoquinon, mitoquinol, vitamin C eller vitamin E.

20 **12.** Polyumættet stof til anvendelse ifølge krav 1, hvor fedtsyren eller fedtsyreesteren er linolsyre, alpha-linolensyre, gamma-linolensyre, dihomogamma-linolensyre eller ester deraf.

13. Polyumættet stof til anvendelse ifølge krav 1, hvor den isotopisk modificerede fedtsyre eller fedtsyreester er en arachidonsyre eller ester deraf.

25 **14.** Polyumættet stof til anvendelse ifølge krav 1, hvor den isotopisk modificerede fedtsyre eller fedtsyreester er en docosatetraensyre eller ester deraf.

30 **15.** Polyumættet stof til anvendelse ifølge krav 1, hvor den isotopisk modificerede fedtsyre eller fedtsyreester er eicosapentaensyre eller ester deraf.

DRAWINGS

FIG. 1A

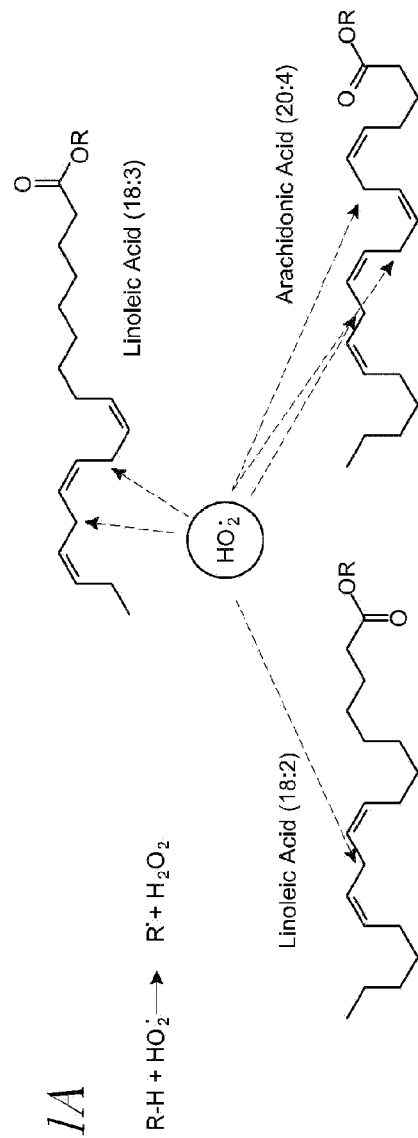


FIG. 1B

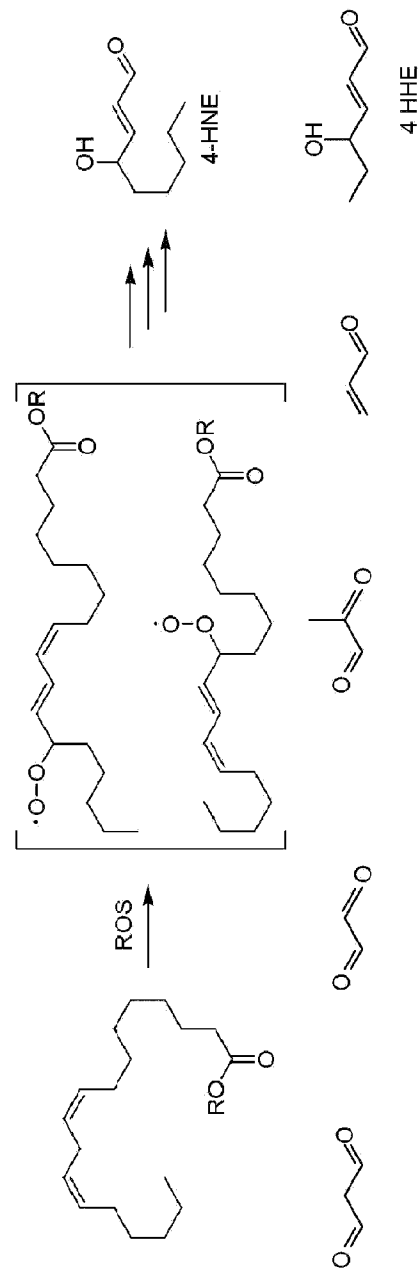


FIG. 2A

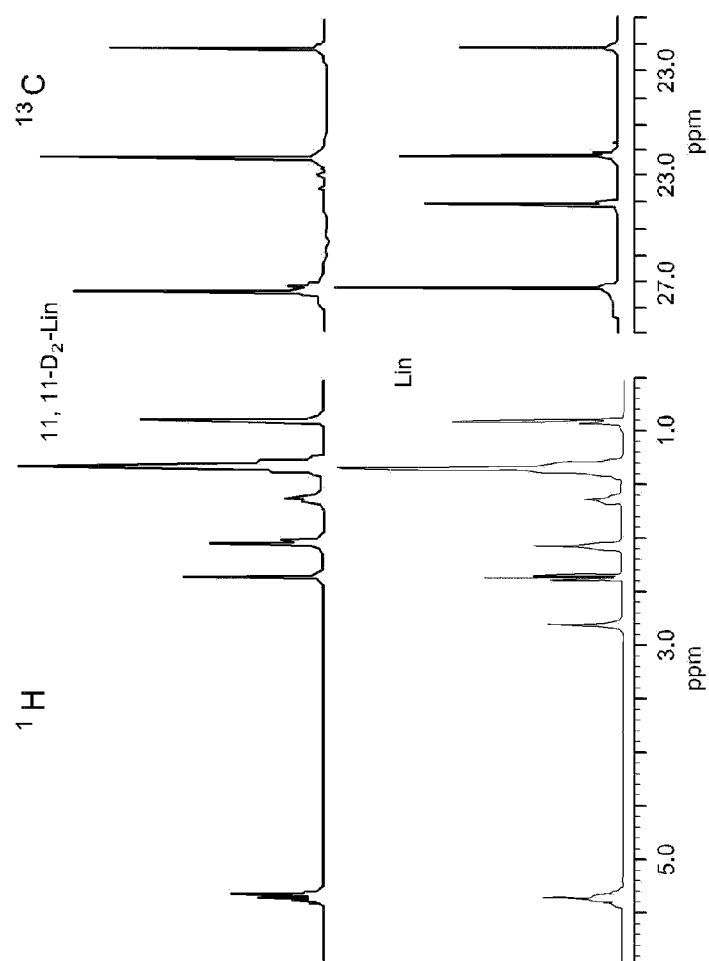
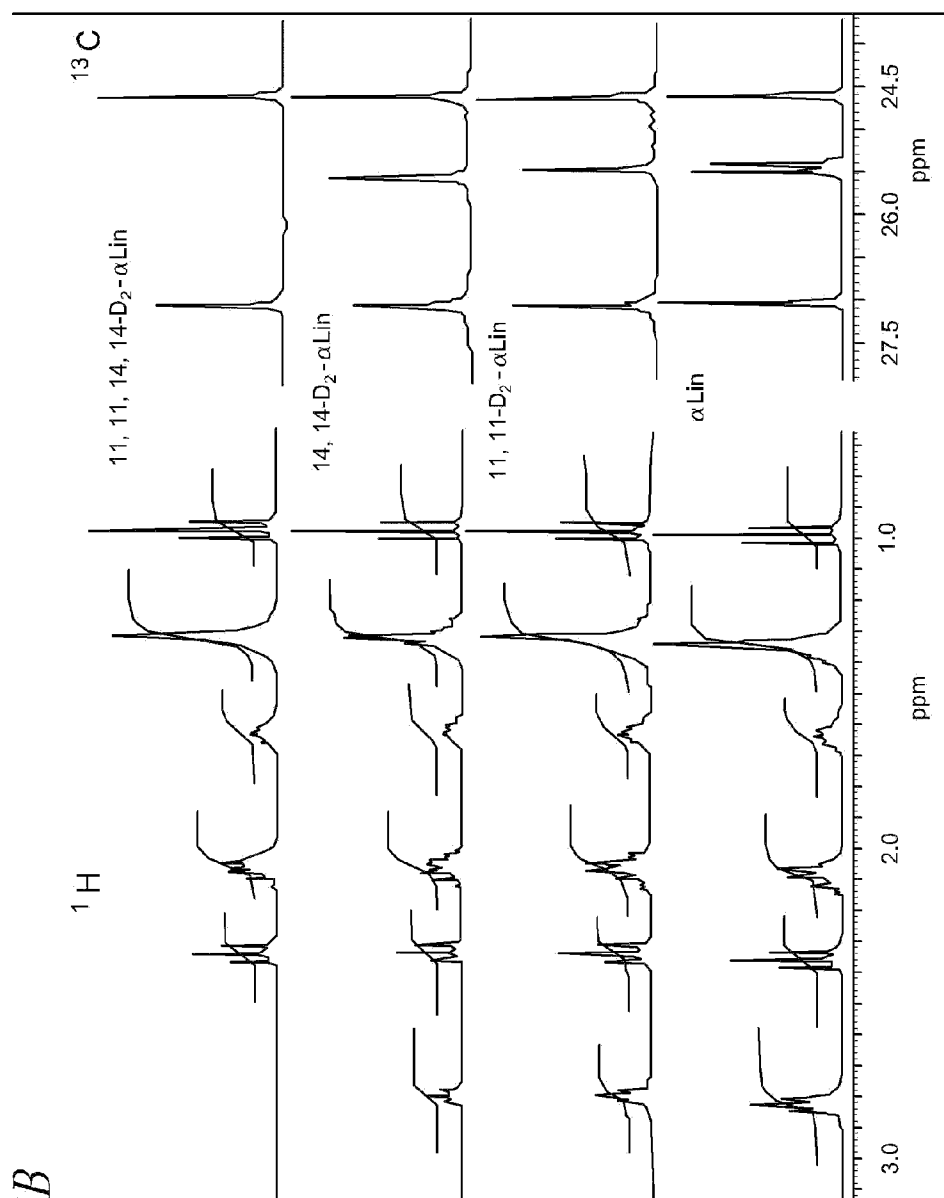


FIG. 2B



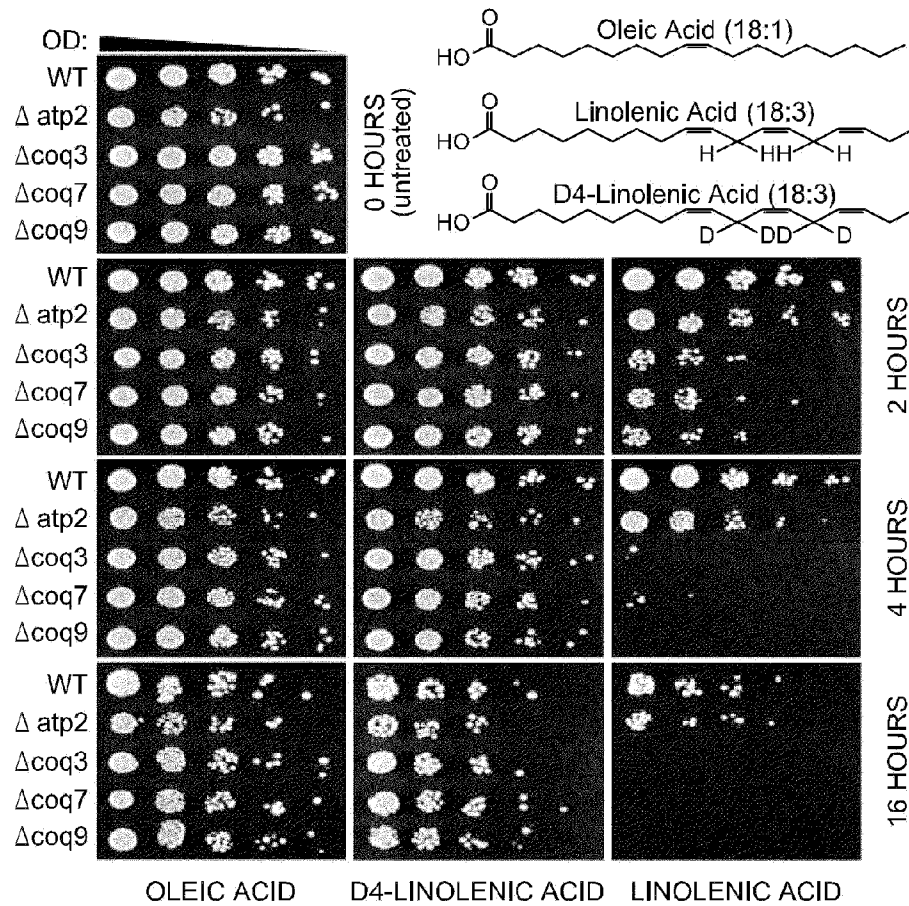


FIG. 3

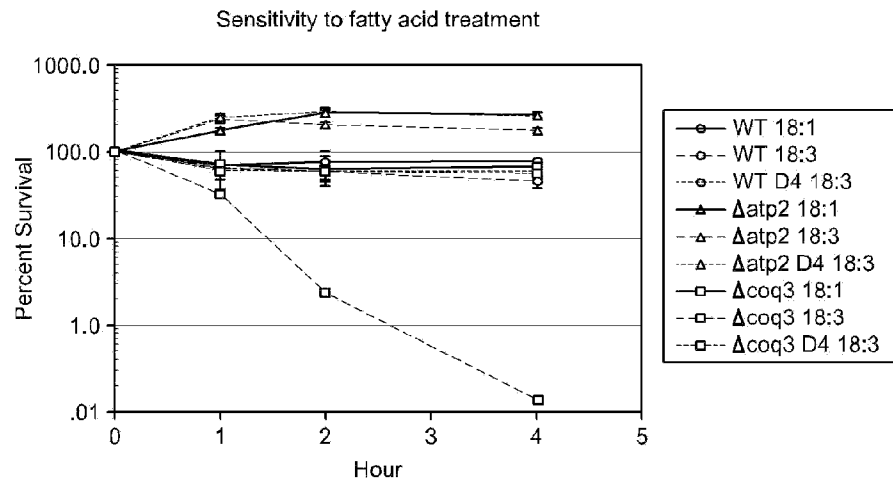
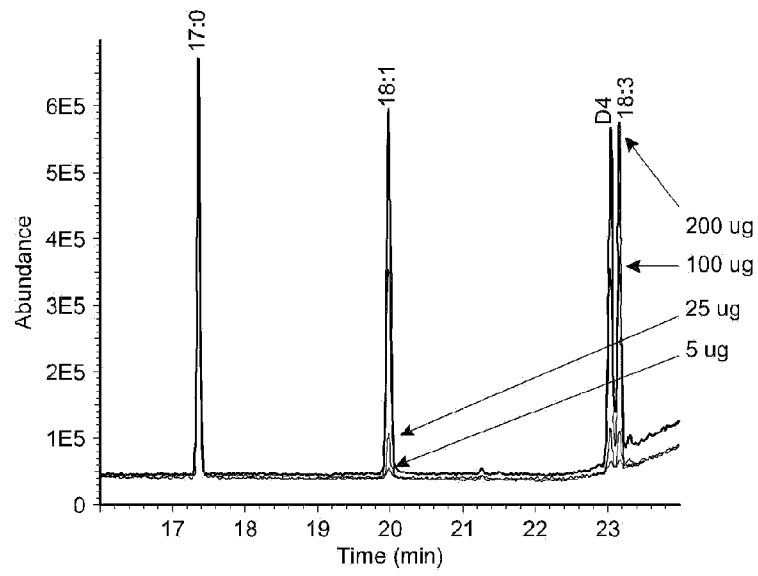
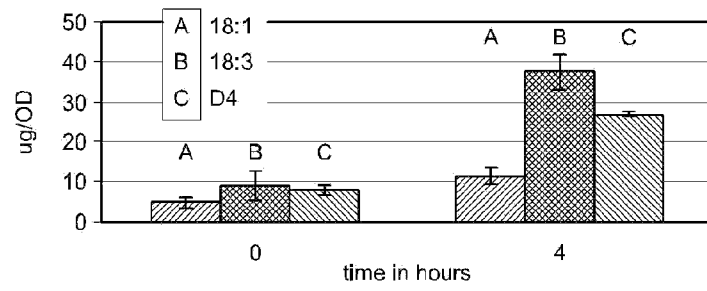
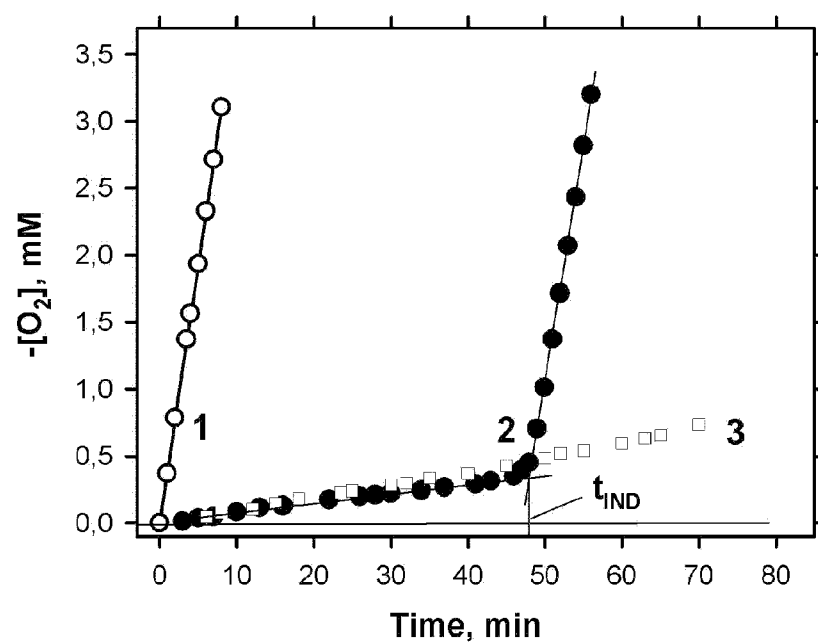


FIG. 4

*FIG. 5*

*FIG. 6*

*FIG. 7*

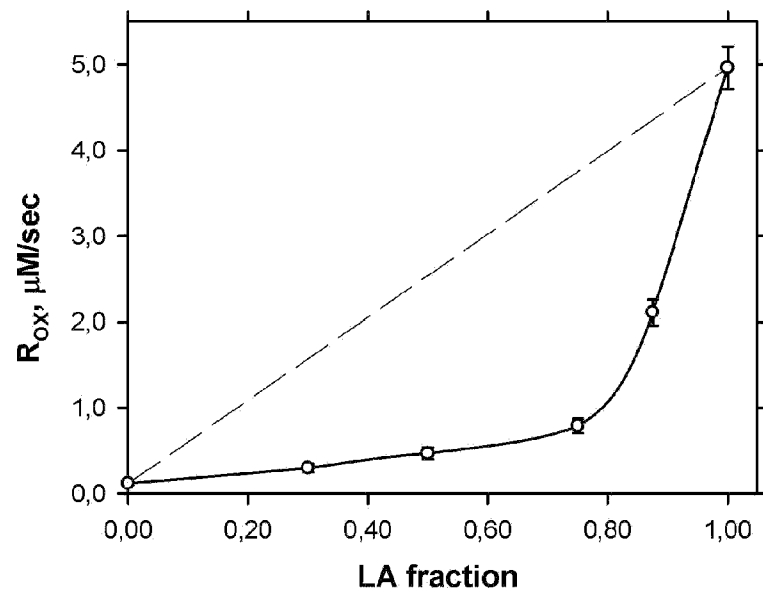
*FIG. 8*

FIG. 9

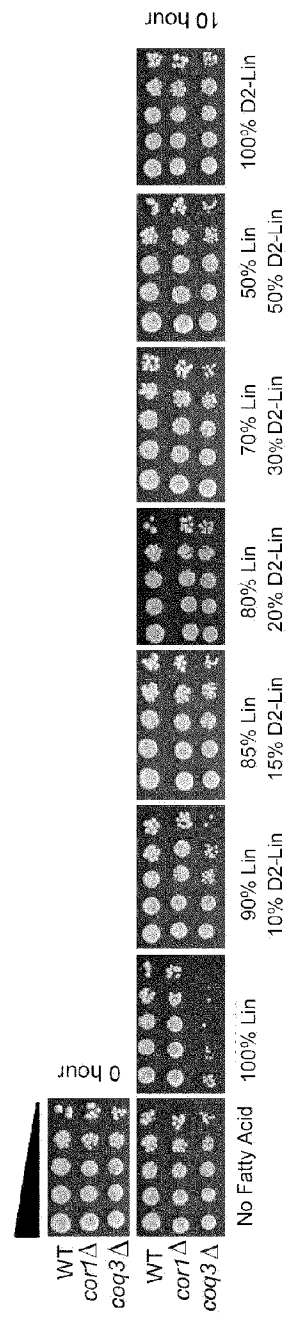


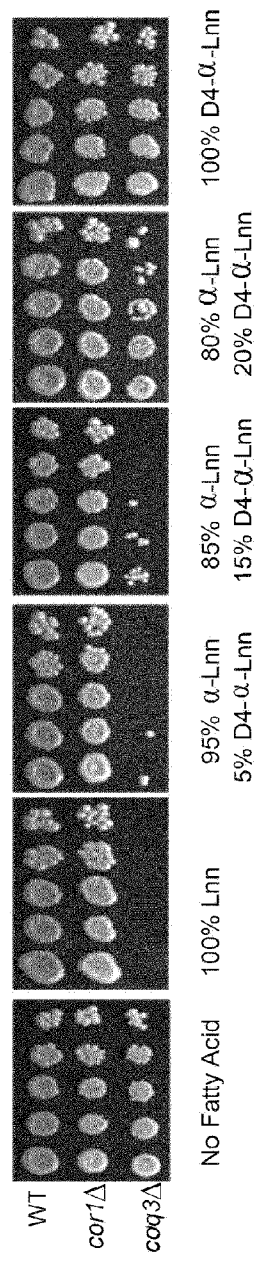
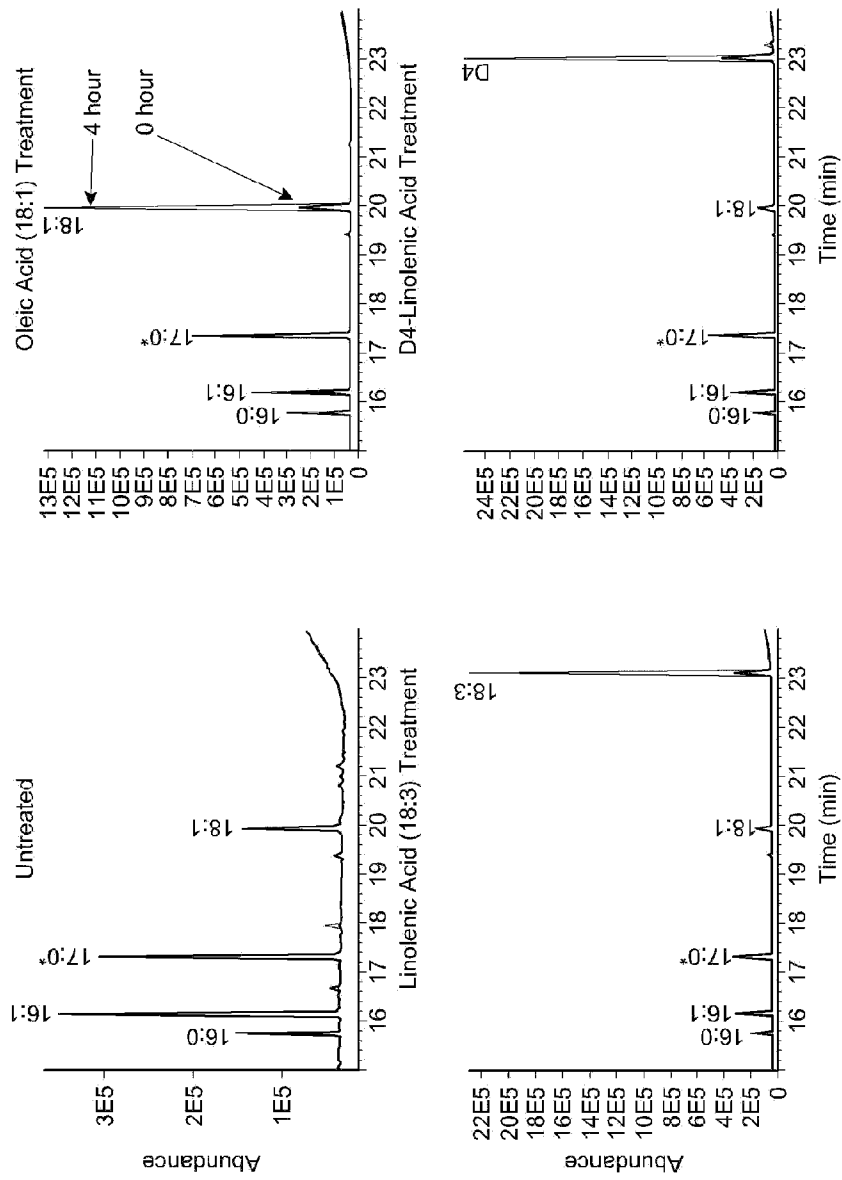
FIG. 10

FIG. 11



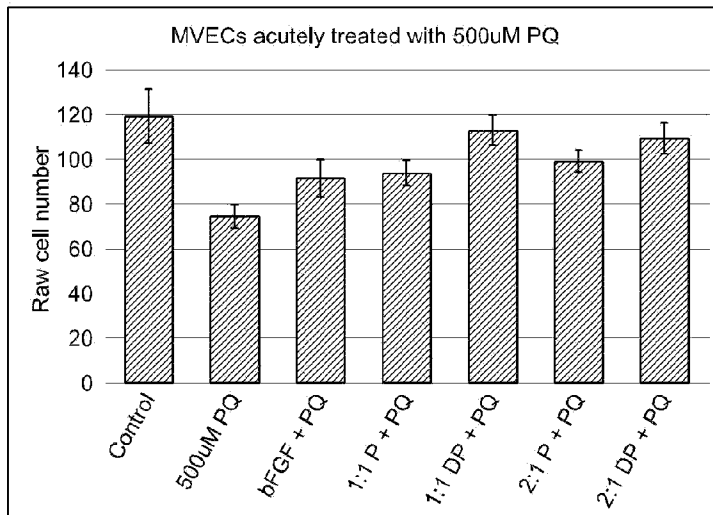
*FIG. 12*

FIG. 13

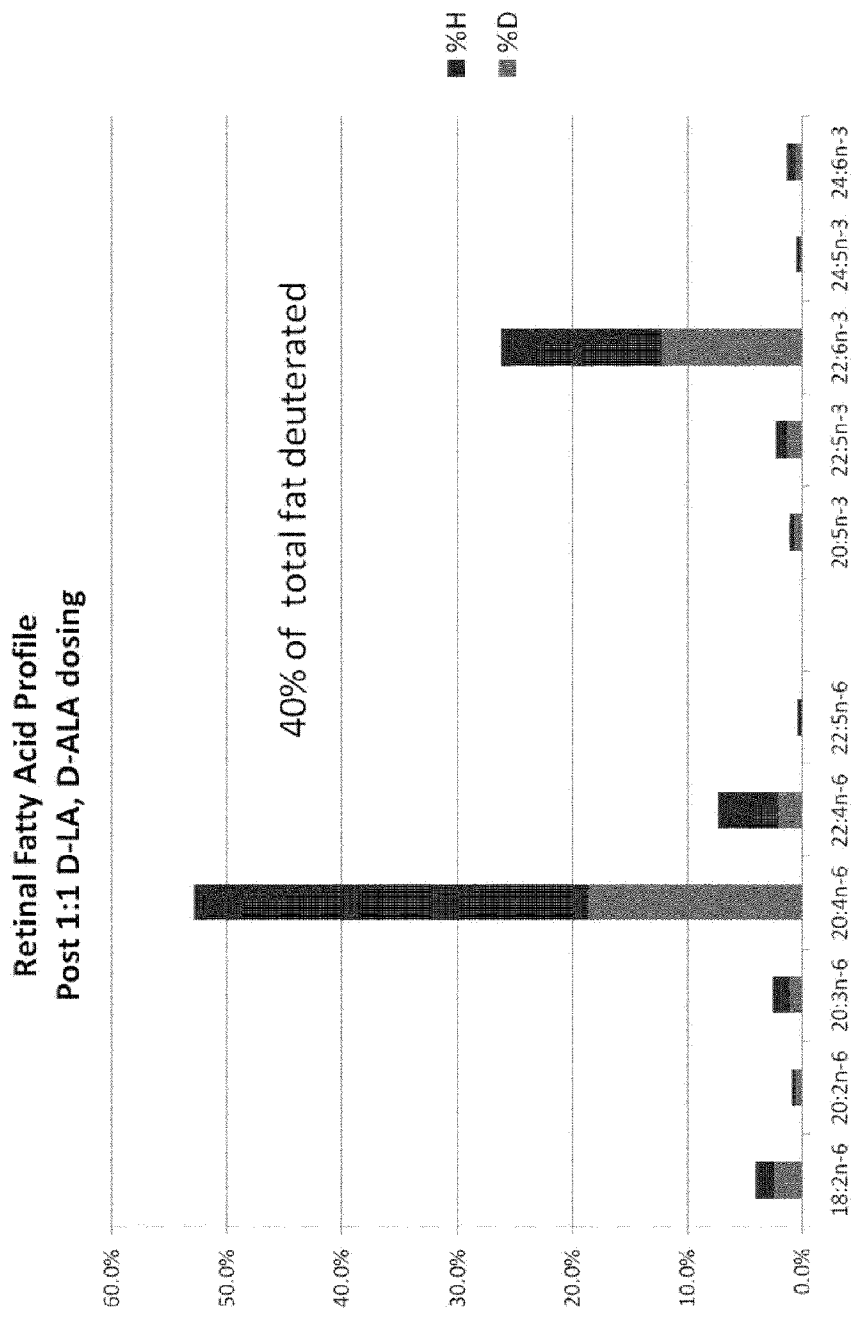


FIG. 14

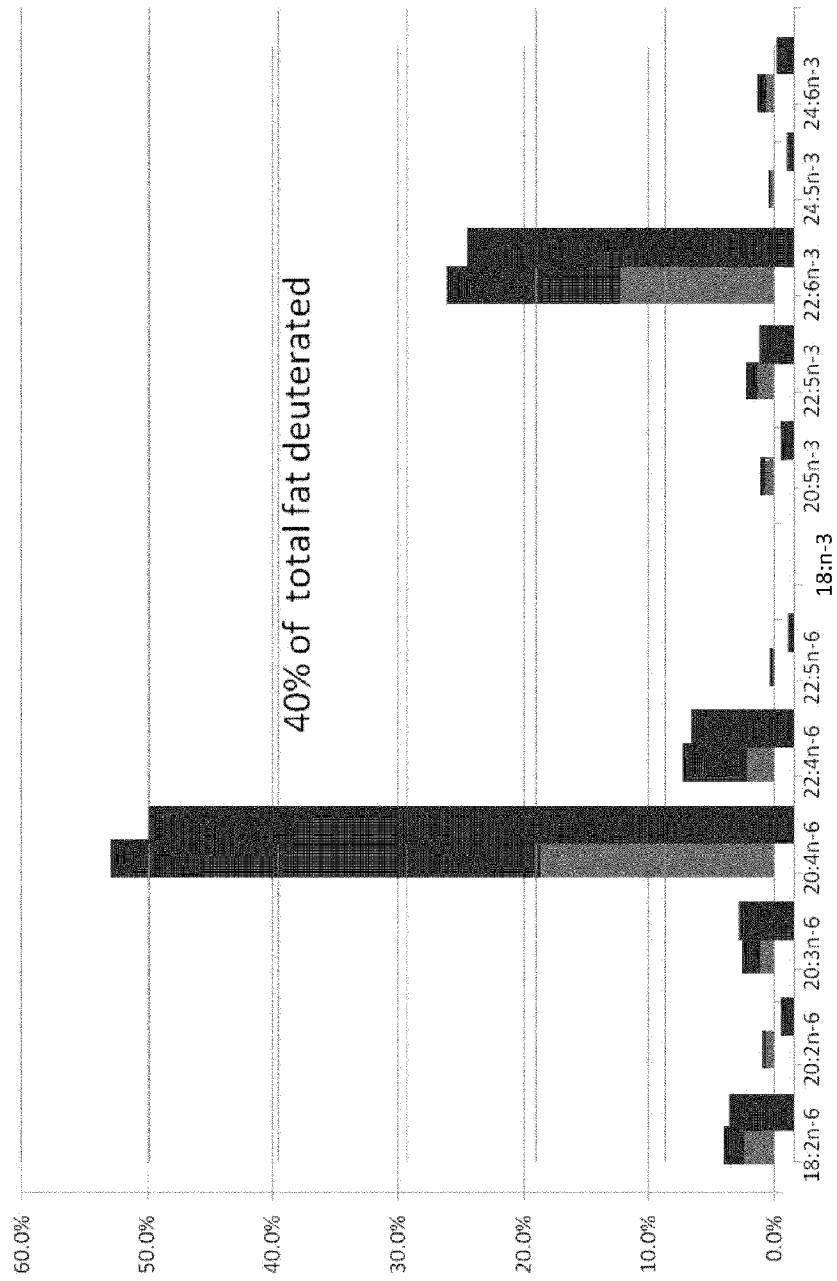


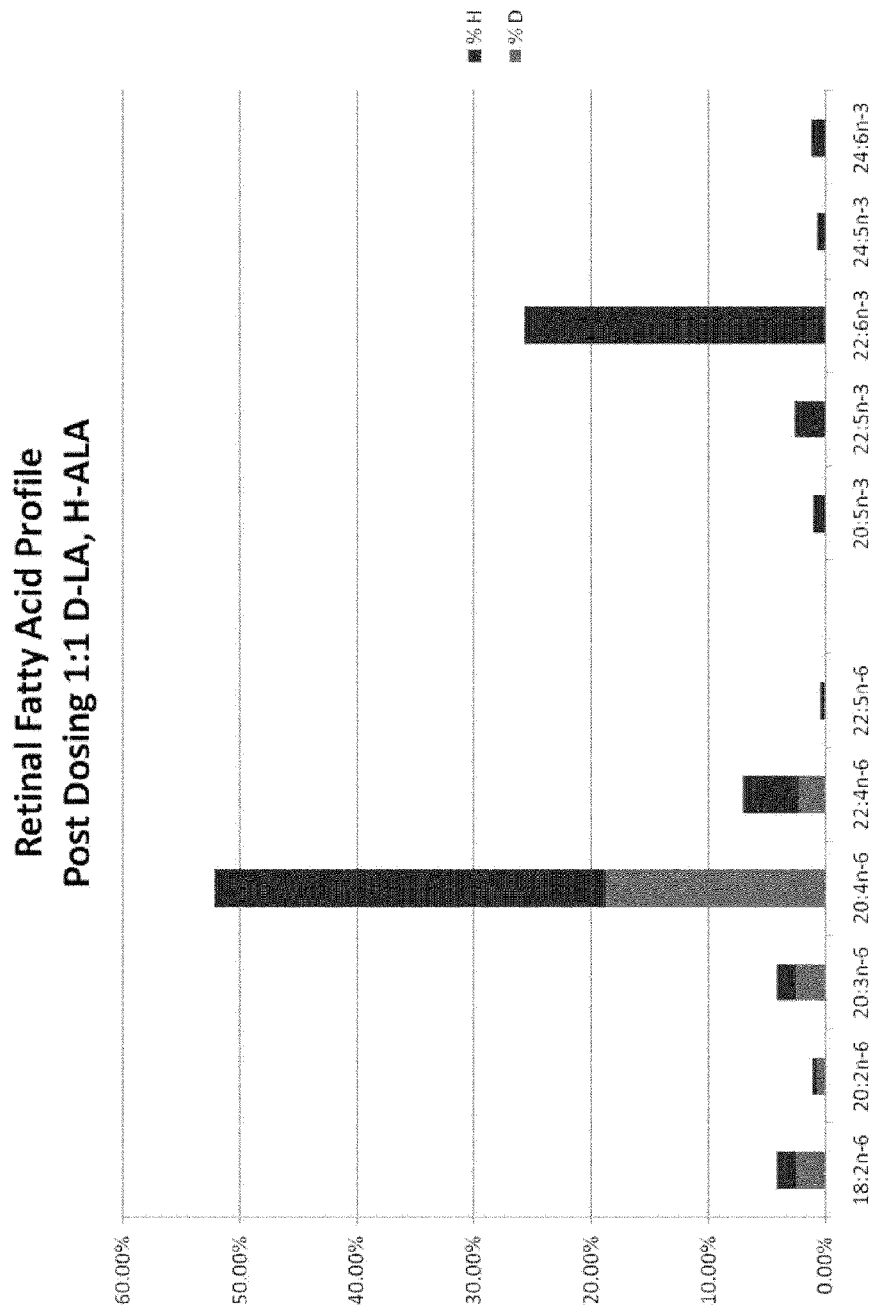
FIG. 15

FIG. 16

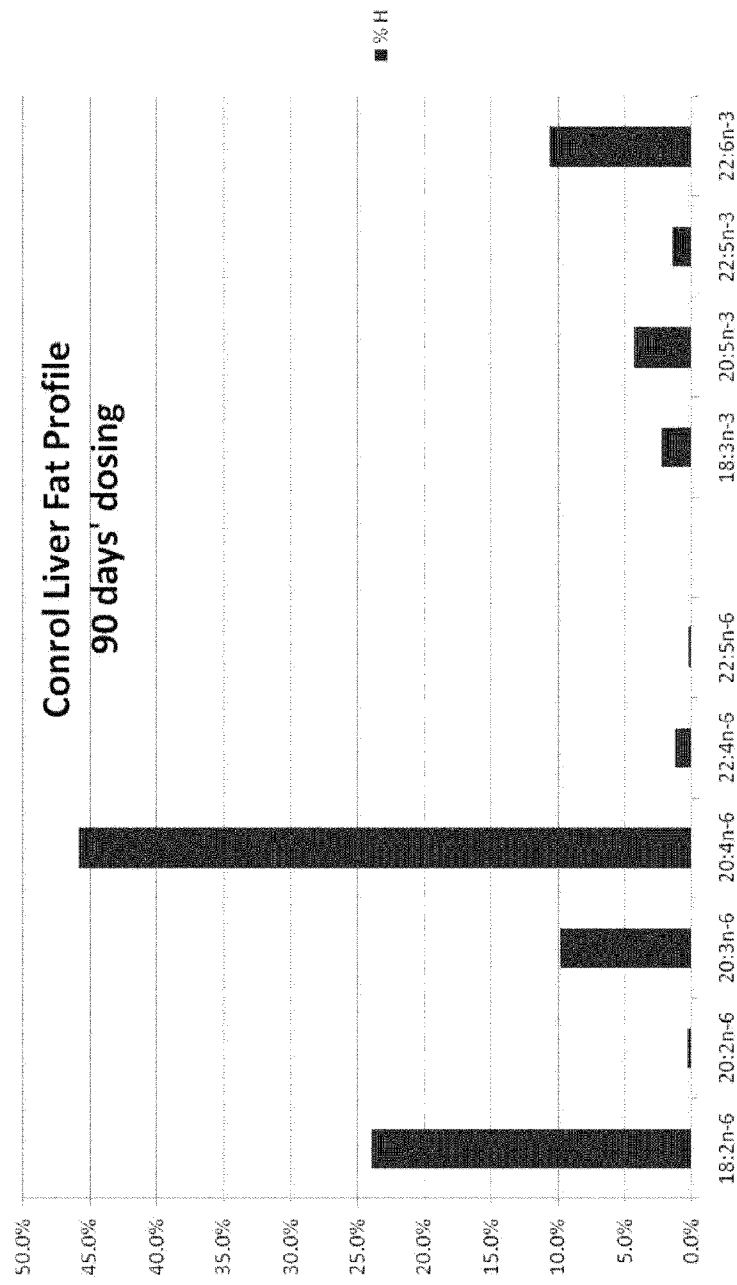


FIG. 17

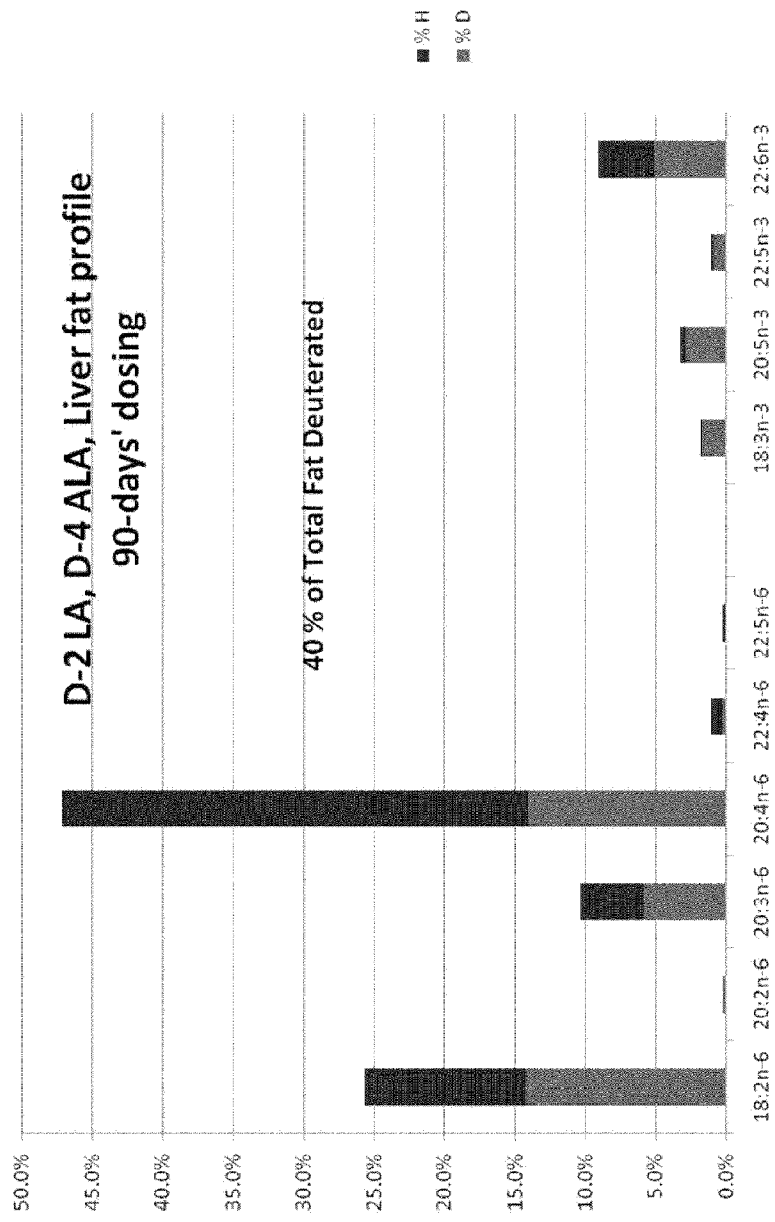


FIG. 18

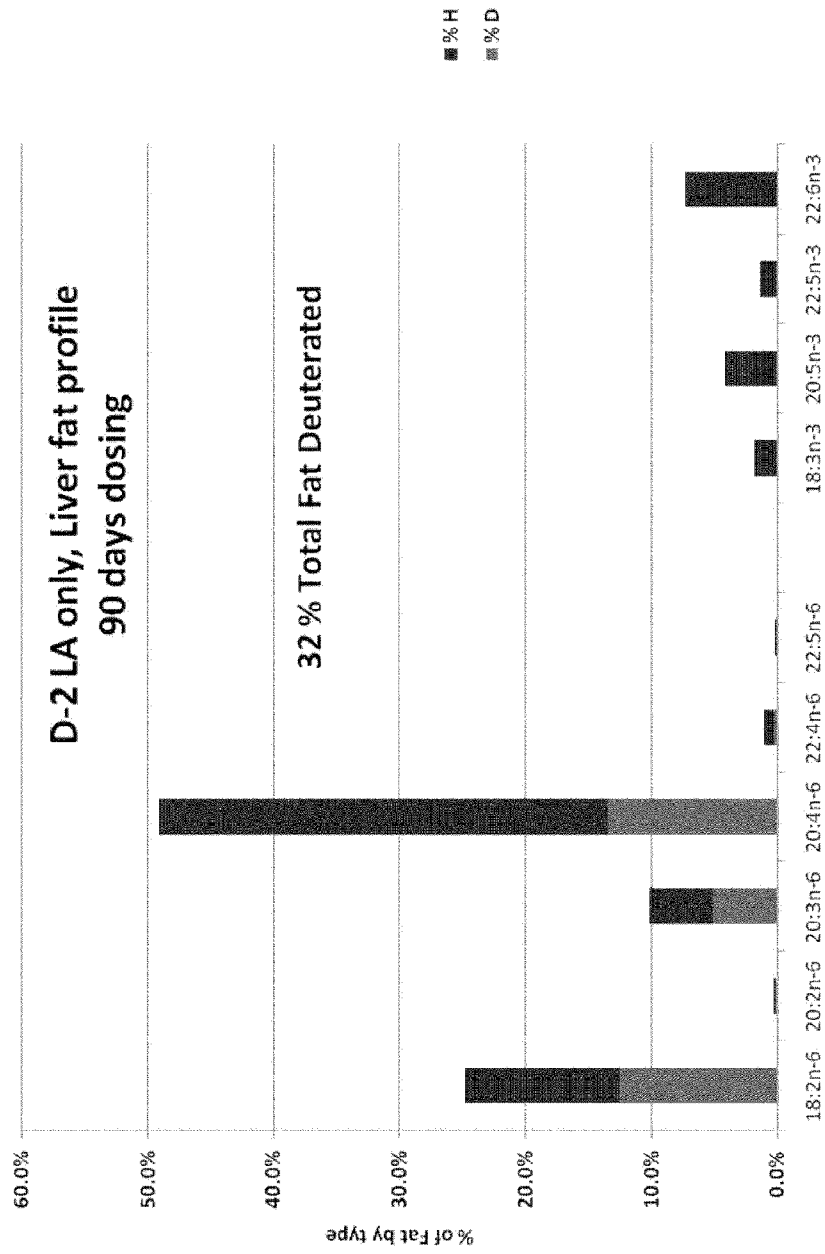


FIG. 19

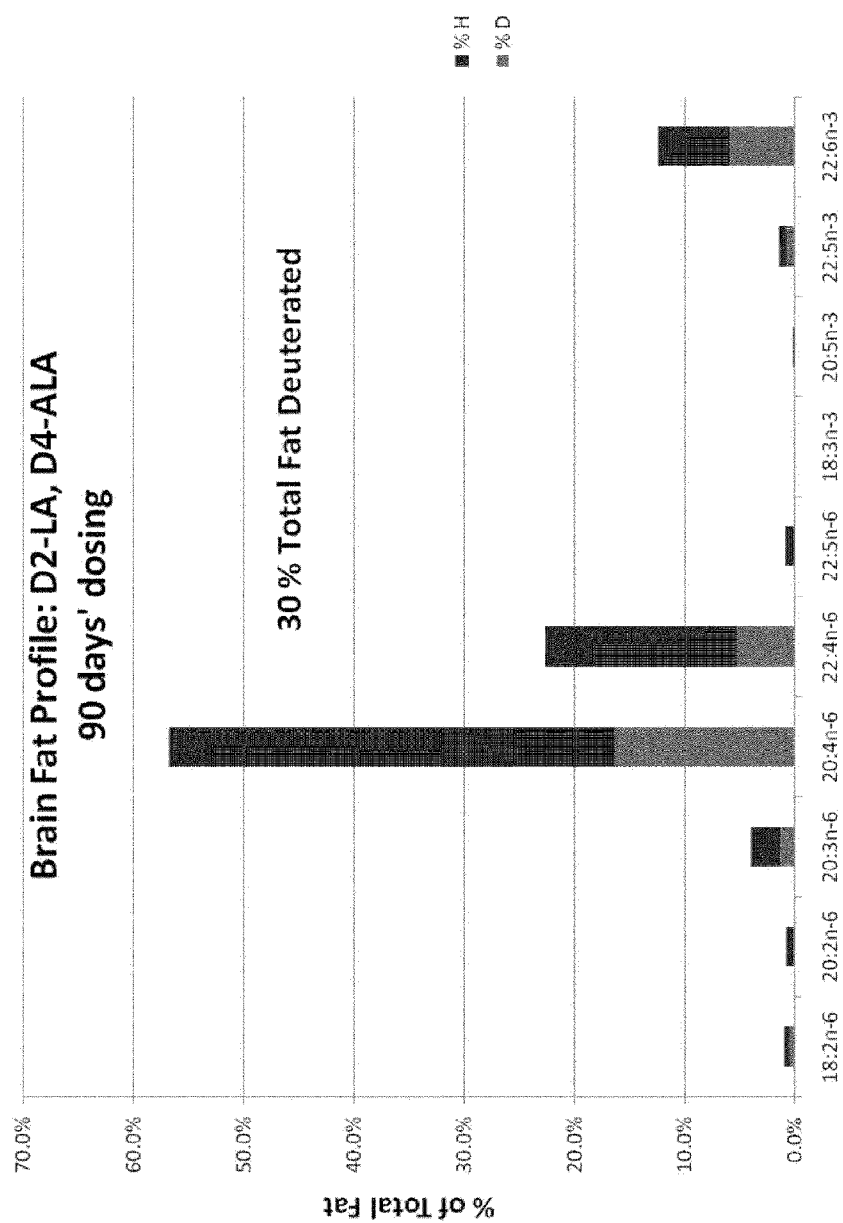


FIG. 20

Brain Fat Profile: D2-LA, H4-ALA
90 days' dosing

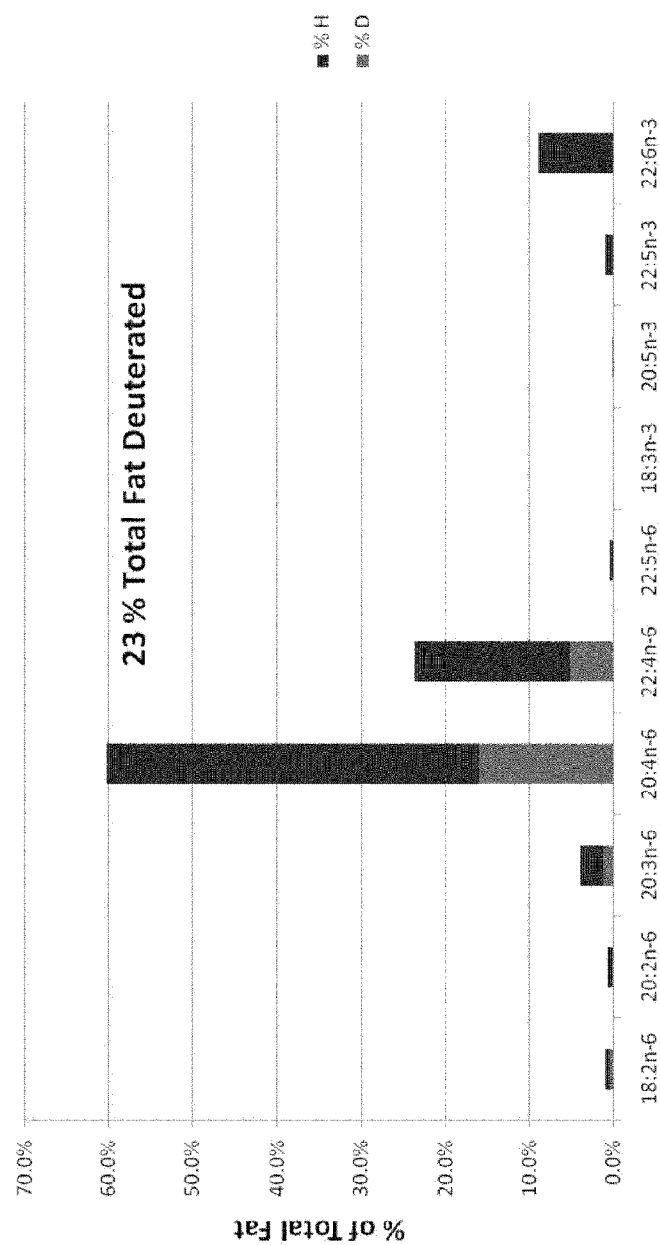


FIG. 21

