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(54) Title: ELECTROPHOTOGRAPHIC MEDIA WITH CARBOXYLIC ACID POLYMER

(57) Abstract: The invention relates to an image receiving element comprising in order a support, at least one polyolefin resin coating and at least one toner receiver layer wherein said toner receiver layer comprises fuser-oil sorbent additive and a carboxylic acid containing thermoplastic polymer, wherein said sorbent additive is present in an amount greater than 10% by weight of said toner receiver layer.

ELECTROPHOTOGRAPHIC MEDIA WITH CARBOXYLIC ACID POLYMER

FIELD OF THE INVENTION

The invention relates to paper for electrophotographic printing. In a preferred form it relates to an imaging element comprising a toner receiver layer that provides photographic quality print using electrophotography and is capable of generating prints resistant to humidity and thermal blocking, is fuser-oil absorbent, glossable, writable and fingerprint and water resistant and has good toner adhesion.

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BACKGROUND OF THE INVENTION

The production of near photographic quality images using electrophotographic imaging technology is highly desirable. It is even more desirable to produce such images on substrates that render the print with the look and feel of a typical photographic print produced with silver halide imaging technology, such as the degree and uniformity of glossiness, stiffness and opacity, and high resolution and sharpness with corresponding low grain appearance. The advantages to producing photographic quality images on such substrates using digital electrophotography include improved environmental friendliness, ease of use, and versatility for customizing images, such as when text and images are combined.

U.S. Patent 5,846,637 describes a coated xerographic photographic paper comprised of (1) a cellulosic substrate; (2) a first antistatic coating layer in contact with one surface of the substrate; (3) a second toner receiver coating on the top of the antistatic layer, and comprised of a mixture of a binder polymer, a toner spreading agent, a lightfastness inducing agent, a biocide, and a filler; and (4) a third traction controlling coating in contact with the back side of the substrate comprised of a mixture of a polymer with a glass transition temperature of from between about -50°C to about 50°C, an antistatic agent, a lightfastness agent, a biocide and a pigment. This paper provides for the third layer on the backside of

the substrate to receive toner, but this is not sufficient for ensuring high image quality should the image be created on this third layer instead of the second layer on the other surface of the substrate.

European Patent Application 1,336,901 A1 describes an electrophotographic image receiving sheet with a toner image receiving layer containing a releasing agent and formed on a support sheet for use in a fixing belt type electrophotography. The support used in the examples had a paper base with polyethylene layers on either side, where the image side is glossy and the backside has a matte finish. No provision is made for receiving the toner image on the backside.

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US patent application 2003/0082354 A1 discloses an image receiving sheet for electrophotography comprising a base paper and a toner image receiving layer comprising a thermoplastic resin and less than 40 percent by mass based on the thermoplastic resin, of a reinforcing filler pigment. The thermoplastic layer is infiltrated to a depth of 1 to 50 percent of the thickness of the base paper. It is desirable that the toner image receiving layer is substantially free of any pigment or filler in order to prevent blister formation and roughening of the toner image.

electrophotographic image receiving sheet comprising a support and a toner image receiving layer containing a thermoplastic resin and a pigment disposed on the surface of the support wherein the surface of the support has a glossiness of 25 percent or more at 75° and a pigment content less than 40 percent by mass based on the mass of the thermoplastic resin. In this case also it is desirable that the toner image receiving layer be substantially free of any pigment or filler in order to prevent blister formation.

Toner particle size also plays a key role in determining image quality in electrophotography, smaller particles generally yielding better image quality. However, as the particles get smaller, the physics of the forces holding the particles to the photoconductor changes drastically, needing new methods to effectively transfer them from the photoconductor to the receiver.

Photographic quality prints can be produced with this process if very small toner particles are used. The drawback with small particles is the difficulty in transferring them onto plain paper. One solution to this problem is explained in U.S. Pat. No. 4,968,578, where the surface of the receiver sheets are coated with a thermoplastic layer.

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While a number of the aforementioned art discloses toner receiver layers that are derived from water based resins, the disadvantage is that such resins are susceptible to blocking under conditions of high humidity especially for prints that come into face to face contact. Further, thermoplastics, including those that comprise the toners, depending upon their glass transition temperature can block at high temperatures. When the thermoplastic is hydrophilic enough to pick up significant moisture from the atmosphere, blocking is exacerbated.

PROBLEM TO BE SOLVED BY THE INVENTION

There exists a need for improved paper for electrophotographic printing that can provide resistance to high humidity and thermal blocking especially for toner receiver layers derived from water based coatings, high gloss, where differential gloss, image relief, and residual surface fuser-oil are minimized and toner adhesion is maximized. Further it is desirable that such prints be fingerprint and spill resistant. There also exists a need for improved writability on the electrophotographic media after imaging and/or glossing, particularly on the backside and on portions of the front side intended for writing such as calendars.

SUMMARY OF THE INVENTION

It is an object of the invention to provide a paper for electrophotographic printing that produces near photoquality prints.

It is another object to provide electrophotographic prints that are glossable with no additional overcoat.

It is a further object to provide electrophotographic glossy prints that show minimum differential gloss and do not show fuser-oil on the surface.

It is a further object to provide photo quality electrophotographic prints that are resistant to blocking under conditions of high relative humidity and temperature.

These and other objects of the invention are accomplished by providing an image receiving element comprising in order a support, at least one polyolefin resin coating, and at least one toner receiver layer, wherein said toner receiver layer comprises a fuser-oil sorbent additive and a carboxylic acid containing thermoplastic polymer, wherein said additive is present in an amount greater than 10% by weight of said toner receiver layer.

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ADVANTAGEOUS EFFECT OF THE INVENTION

The invention has numerous advantages. The invention provides a paper for electrophotographic printing that can provide near photo quality, high gloss prints that are resistant to humidity and thermal blocking, where differential gloss, image relief, and residual surface fuser-oil are minimized and toner adhesion is maximized, exhibits fingerprint resistance and water resistance compared to commercially available clay coated papers and further exhibits improved writability on the media after imaging and/or glossing, particularly on the backside and on portions of the front side intended for writing. The paper also provides an excellent degree of whiteness. These and other advantages will be apparent from the detailed description below.

DETAILED DESCRIPTION OF THE INVENTION

The image receiving element of this invention comprises in order a support, at least one polyolefin resin coating, and at least one toner receiver layer, coated on both sides of the resin coated support wherein said toner receiver layer comprises a fuser-oil sorbent additive and a carboxylic acid containing thermoplastic polymer and said additive is present in an amount of greater than 10% by weight of said layer.

The term "support" as used herein refers to a base or a substrate material that is the primary part of an imaging element such as paper, polyester,

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vinyl, synthetic paper, fabric, or other suitable material for the viewing of images. The support for use in the present invention may be any support typically used in imaging applications. Typical supports may be fabrics, paper, and polymer sheets. The support may either be transparent or opaque, reflective or non-reflective. The term as used herein, "transparent" means the ability to pass radiation without significant deviation or absorption. Opaque supports include plain paper, coated paper, synthetic paper, low density foam core based support and low density foam core based paper. The support can also consist of microporous materials such as polyethylene polymer-containing material sold by PPG Industries, Inc., Pittsburgh, Pennsylvania under the trade name of Teslin®, Tyvek® synthetic paper (DuPont Corp.), impregnated paper such as Duraform®, and OPPalyte® films (Mobil Chemical Co.) and other composite films listed in U.S. Patent 5,244,861. Transparent supports include glass, cellulose derivatives, such as a cellulose ester. cellulose triacetate, cellulose diacetate, cellulose acetate propionate, cellulose acetate butyrate, polyesters, such as poly(ethylene terephthalate), poly(ethylene naphthalate), poly-1,4-cyclohexanedimethylene terephthalate, poly(butylene terephthalate), and copolymers thereof, polyimides, polyamides, polycarbonates, polystyrene, polyolefins, such as polyethylene or polypropylene, polysulfones, polyacrylates, polyether imides, and mixtures thereof. The papers listed above include a broad range of papers, from high end papers, such as photographic paper to low end papers, such as newsprint. The support used in the invention may have a thickness of from about 50 to about 500 µm, preferably from about 75 to 300 μm.

The imaging supports of the invention can comprise any number of auxiliary layers, for example, functional layers. Such auxiliary layers may include tie layers or adhesion promoting layers, conveyance layers, barrier layers, splice providing layers, UV absorption layers, and waterproofing layers.

The polyolefin resin, preferably coated on the paper support can be any suitable melt extrusion coatable polyolefin material known in the art. Suitable polymers for the polyolefin resin coating include polyethylene, polypropylene, polymethylpentene, polystyrene, polybutylene, and mixtures thereof. Polyolefin

copolymers, including copolymers of polyethylene, polypropylene and polyolefins derived from hexene, butene, and octene are also useful. The polyolefin may also be copolymerized with one or more copolymers including polyesters, such as polyethylene terephthalate, polysulfones, polyurethanes, polyvinyls,

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polycarbonates, cellulose esters, such as cellulose acetate and cellulose propionate, and polyacrylates. Specific examples of copolymerizable monomers include vinyl stearate, vinyl acetate, acrylic acid, methyl acrylate, ethyl acrylate, acrylamide, methacrylic acid, methyl methacrylate, ethyl methacrylate, methacrylamide, butadiene, isoprene, and vinyl chloride.

Polyethylene is preferred for coated paper supports, as it is low in cost and has desirable coating properties. Preferred polyolefins are film forming and adhesive to paper. Usable polyethylenes may include high density polyethylene, low density polyethylene, linear low density polyethylene, and polyethylene blends. Polyethylene having a density in the range of from 0.910 g/cm³ to 0.980 g/cm³ is particularly preferred. The polyolefin resin, such as polypropylene, may also be laminated to one or both sides of the paper support and optionally biaxially oriented.

It is desirable to incorporate white pigments in the polyolefin resin layer of the support to give the required optical properties for the support. Any suitable white pigment may be incorporated in the polyolefin resin layers, such as, for example, zinc oxide, zinc sulfide, zirconium dioxide, white lead, lead sulfate, lead chloride, lead aluminate, lead phthalate, antimony trioxide, white bismuth, tin oxide, white manganese, white tungsten, and combinations thereof. The preferred pigment is titanium dioxide because of its high refractive index, which gives excellent optical properties at a reasonable cost. The pigment is used in any form that is conveniently dispersed within the polyolefin. The preferred pigment is anatase titanium dioxide. The most preferred pigment is rutile titanium dioxide because it has the highest refractive index at the lowest cost. The average pigment diameter of the rutile TiO_2 is most preferably in the range of 0.1 to 0.26 μ m. The pigments that are greater than 0.26 μ m are too yellow for an imaging element application and the pigments that are less than 0.1 μ m are not sufficiently opaque

when dispersed in polymers. Preferably, the white pigment should be employed in the range of from about 10 to about 50 percent by weight, based on the total weight of the polyolefin coating. Below 10 percent TiO₂, the imaging system will not be sufficiently opaque and will have inferior optical properties. Above 50 percent TiO₂, the polymer blend is not manufacturable.

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The surface of the TiO₂ can be treated with inorganic compounds such as aluminum hydroxide, alumina with a fluoride compound or fluoride ions, silica with a fluoride compound or fluoride ion, silicon hydroxide, silicon dioxide, boron oxide, boria-modified silica (as described in U.S. Pat. No. 4,781,761), phosphates, zinc oxide or, ZrO₂ and with organic treatments such as polyhydric alcohol, polyhydric amine, metal soap, alkyl titanate, polysiloxanes, or silanes. The organic and inorganic TiO₂ treatments can be used alone or in any combination. The amount of the surface treating agents is preferably in the range of 0.2 to 2.0% for the inorganic treatment and 0.1 to 1% for the organic treatment, relative to the weight of the titanium dioxide. At these levels of treatment, the TiO₂ disperses well in the polymer and does not interfere with the manufacture of the imaging support.

The polyolefin resins and TiO₂ and optional other additives may be mixed with each other in the presence of a dispersing agent. Examples of dispersing agents are metal salts of higher fatty acids such as sodium palmitate, sodium stearate, calcium palmitate, sodium laurate, calcium stearate, aluminum stearate, magnesium stearate, zirconium octylate, or zinc stearate higher fatty acids, higher fatty amide, and higher fatty acids. The preferred dispersing agent is sodium stearate and the most preferred dispersing agent is zinc stearate. Both of these dispersing agents give superior whiteness to the resin coated layer.

In addition, it may be necessary to use various additives such as colorants, brightening agents, antistatic agents, plasticizers, antioxidants, slip agents, or lubricants, and light stabilizers in the resin coated supports as well as biocides in the paper elements. These additives are added to improve, among other things, the dispersibility of fillers and/or colorants, as well as the thermal and color stability during processing and the manufacturability and the longevity of the

finished article. For example, the polyolefin coating may contain antioxidants such as 4,4'-butylidene-bis(6-tert-butyl-meta-cresol), di-lauryl-3,3'thiopropionate, N-butylated-p-aminophenol, 2,6-di-tert-butyl-p-cresol, 2,2-di-tert-butyl-4-methyl-phenol, N,N-disalicylidene-1,2-diaminopropane, tetra(2,4-tert-butylphenyl)-4,4'-diphenyl diphosphonite, octadecyl 3-(3',5'-di-tert-butyl-4'-hydroxyphenyl propionate), combinations of the above, and the like; heat stabilizers, such as higher aliphatic acid metal salts such as magnesium stearate, calcium stearate, zinc stearate, aluminum stearate, calcium palmitate, zirconium octylate, sodium laurate, and salts of benzoic acid such as sodium benzoate,
calcium benzoate, magnesium benzoate and zinc benzoate; light stabilizers such as hindered amine light stabilizers (HALS), of which a preferred example is poly{[6-[(1,1,3,3-tetramethylbutylamino}-1,3,5-triazine-4-piperidinyl)-imino]-1,6-hexanediyl[{2,2,6,6-tetramethyl-4-piperdinyl)imino]}(Chimassorb 944 LD/FL).

polyolefin structures, such as those achieved by multiple coatings, either sequential or via coextrusion. To minimize the number of resins required, a structure consisting of 1 to 3 layers on each side is preferred. In one embodiment of the present invention, at least one or all the layers can further comprise polypropylene. In a 3-layer structure, two of the three layers on each side may have substantially similar composition, preferably the two outer flange layers. The ratio of thickness of the center or bottom layer to an outer flange layer is in the range of 1 to 8 with 5 to 7 being most preferable. The polyolefin resin of the flange layers may contain, optionally, pigments and other addenda.

The coating of the paper base material with the polyolefin to form
the support preferably is by extrusion from a hot melt as is known in the art. The
support utilized in the invention may be practiced within a wide range of extrusion
temperatures, for example, from 150°C to 350°C, and speeds, for example, from
60 m/min. to 460 m/min., depending on the particular intended application of the
support. For many applications, preferred extrusion temperatures are from 300°C
to 330°C.

The extruded film is cast and quenched in a nip formed between a pressure roll and a chill roll. The process conditions are well known in the art and are such as to avoid air entrainment between the resin and the paper base. The surface of chill roll may be polished or matte or any other texture depending on the surface desired on the finished polyolefin coating.

The electrophotographic processes and their individual steps have been well described in detail in many books and publications. The processes incorporate the basic steps of creating an electrostatic image, including charging and exposing a photoconductor, developing that image with charged, colored particles (toner), optionally transferring the resulting developed image to a secondary intermediate substrate, such as a cylinder with a rubber-like soft-elastic surface or a rubber blanket, and then transferring onto a final substrate or receiver and fixing or fusing the image onto the receiver. In terms of environmental stability and extending image quality, the intermediate transfer method is more desirable. The final image receiving element of the invention has a toner receiver layer designed to receive the toner particles.

It is know to fix the toner pattern to the toner receiver layer, the toner on the receiving sheet is subjected to heat and pressure, for example, by passing the sheet through the nip of fusing rolls. Both the toner polymer and the thermoplastic polymer of the toner receiver layer are softened or fused sufficiently to adhere together under the pressure of the fusing rolls. When both the toner receiver layer and the toner soften and fuse, the toner can be at least partially embedded in the thermoplastic toner receiver layer. For self-fixing toners, residual liquid is removed from the paper by air-drying or heating. Upon evaporation of the solvent these toners form a film bonded to the paper. For heat-fusible toners, thermoplastic polymers are used as part of the particle. Heating both removes residual liquid and fixes the toner to paper. The fusing step can be accomplished by the application of heat and pressure to the final image. Fusing can provide increased color saturation, improved toner adhesion to the receiver, and modification of the image surface texture. A fusing device can be a cylinder or belt. The fusing device can have an elastomeric coating which provides a

conformable surface to enable improved heat transfer to the receiver. The fusing device can have a smooth or textured surface. The fusing step can be combined with the transfer step. In many fusing devices such as the fuser utilized with the toner receiver layer of the present invention, a fuser-oil, such silicone oil is applied at the nip to aid the release of the toner from the fuser roller.

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In forming toner images on conventional receiving sheets, the fusing and fixing of the toner to the sheet by the fusing rolls, creates gloss in the toned areas, i.e., in the so-called D max or black areas of the image. In the untoned areas, however, in the so-called D min or white areas, no gloss is formed. In accordance with the present invention, however, it is preferred that when the toner-bearing receiver sheet is subjected to heat and pressure in the fusing roll nip, the entire surface of the sheet develops a substantially uniform gloss. The resulting electrophotographic image has the look and feel of a silver halide photographic print.

15 In a preferred embodiment, a belt fusing apparatus as described in U.S. Patent 5,895,153 can be used to provide high gloss finish to the electrophotographically printed image receiving element of this invention. The belt fuser can be separate from or integral with the reproduction apparatus. In a preferred embodiment of the present invention, the belt fuser is a secondary step. 20 The toned image is at first fixed by passing the electrophotographically printed sheet through the nip of fusing rolls within the reproduction apparatus and then subjected to belt fusing to obtain a high uniform glossy finish. The belt fusing apparatus includes an input transport for delivering marking particle imagebearing receiver members to a fusing assembly. The fusing assembly comprises a 25 fusing belt entrained about a heated fusing roller and a steering roller, for movement in a predetermined direction about a closed loop path. The fusing belt is, for example, a thin metallic or heat resistant plastic belt. Metal belts can be electroformed nickel, stainless steel, aluminum, copper or other such metals, with the belt thickness being about 2 to 5 mils. Seamless plastic belts can be formed of 30 materials such as polyimide, polypropylene, or the like, with the belt thickness summarily being about 2 to 5 mils. Usually these fusing belts are coated with thin

hard coatings of releasing material such as silicone resins, fluoropolymers, or the like. The coatings are typically thin (1 to 10 microns), very smooth, and shiny. Such fusing belts could also be made with some textured surface to produce images of lower gloss or texture.

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The belt fuser can have a pressure roller located in nip relation with the heated fusing roller. A flow of air is directed at an area of the belt run upstream of the steering roller and adjacent to the steering roller to cool such area. The cooling action provides for a commensurate cooling of a receiver member, bearing a marking particle image, while such member is in contact with the fusing belt. The cooling action for the receiver member serves as the mechanism to substantially prevent offset of the marking particle image to the pressure roller.

The belt fusing apparatus can be mounted in operative association with a belt tracking control mechanism.

electrophotographically printed image receiving element of this invention by using calendering methods known in the art. Calendering is defined herein as a process in which pressure is applied to the imaged substrate, that has been preferably roller fused in the printing apparatus, by passing it between highly polished, metal rollers that are optionally heated, imparting a glossy, smooth surface finish to the substrate. The degree of pressure and heat controls the extent of gloss.

Calendering differs from roller fusing in that the latter does not necessarily use highly polished rollers, is always carried out at high temperatures and the nip pressures are lower than those experienced at the calendering nip.

The toner utilized with the image receiving element of this invention contains a polymer (a binder resin), a colorant and an optional releasing agent.

As the polymer, known binder resins are useable. Concretely, these binder resins include homopolymers and copolymers such as polyesters, styrenes, e.g. styrene and chlorostyrene; monoolefins, e.g. ethylene, propylene, butylene and isoprene; vinyl esters, e.g. vinyl acetate, vinyl propionate, vinyl benzoate and vinyl butyrate; α-methylene aliphatic monocarboxylic acid esters, e.g. methyl acrylate,

ethyl acrylate, butyl acrylate, dodecyl acrylate, octyl acrylate, phenyl acrylate, methyl methacrylate, ethyl methacrylate, butyl methacrylate and dodecyl methacrylate; vinyl ethers, e.g. vinyl methyl ether, vinyl ethyl ether and vinyl butyl ether; and vinyl ketones, e.g. vinyl methyl ketone, vinyl hexyl ketone and vinyl isopropenyl ketone. Particularly desirable binder resins include polystyrene resin, polyester resin, styrene/alkyl acrylate copolymers, styrene/alkyl methacrylate copolymers, styrene/acrylonitrile copolymer, styrene/butadiene copolymer, styrene/maleic anhydride copolymer, polyethylene resin and polypropylene resin. They further include polyurethane resin, epoxy resin, silicone resin, polyamide resin, modified rosin, paraffins and waxes. In these resins, styrene/acryl resins are particularly preferable.

As the colorants, known colorants can be used. The colorants include, for example, carbon black, Aniline Blue, Calcoil Blue, Chrome Yellow, Ultramarine Blue, Du Pont Oil Red, Quinoline Yellow, Methylene Blue Chloride, Phthalocyanine Blue, Malachite Green Oxalate, Lamp Black, Rose Bengal, C.I. Pigment Red 48:1, C.I. Pigment Red 122, C.I. Pigment Red 57:1, C.I. Pigment Yellow 97, C.I. Pigment Yellow 12, C.I. Pigment Yellow 17, C.I. Pigment Blue 15:1 and C.I. Pigment Blue 15:3. The colorant content is, for example, 2 to 8% by weight of the toner. When the colorant content is 2% or more by weight, a sufficient coloring power can be obtained, and when it is 8% or less by weight, good transparency can be obtained.

The toner utilized with the image receiving element of the present invention optionally contains a releasing agent. The releasing agents preferably used herein are waxes. Concretely, the releasing agents usable herein are low-molecular weight polyolefins such as polyethylene, polypropylene and polybutene; silicone resins which can be softened by heating; fatty acid amides such as oleamide, erucamide, ricinoleamide and stearamide; vegetable waxes such as carnauba wax, rice wax, candelilla wax, Japan wax and jojoba oil; animal waxes such as bees wax; mineral and petroleum waxes such as montan wax, ozocerite, ceresine, paraffin wax, microcrystalline wax and Fischer-Tropsch wax; and modified products thereof. When a wax containing a wax ester having a high

polarity, such as carnauba wax or candelilla wax, is used as the releasing agent, the amount of the wax exposed to the toner particle surface is inclined to be large. On the contrary, when a wax having a low polarity such as polyethylene wax or paraffin wax is used, the amount of the wax exposed to the toner particle surface is inclined to be small.

Irrespective of the amount of the wax inclined to be exposed to the toner particle surface, waxes having a melting point in the range of 30 to 150° C. are preferred and those having a melting point in the range of 40 to 140° C. are more preferred.

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The wax is, for example, 0.1 to 10% by mass, and preferably 0.5 to 7% by mass, based on the toner.

The toner used with the image receiving element of the present invention may contain an additive. Fine powders of inorganic compounds and fine particles of organic compounds are used as the additive. Fine particles of the inorganic compounds are those of, for example, SiO_2 , TiO_2 , Al_2O_3 , CuO, ZnO, SnO_2 , Fe_2O_3 , MgO, BaO, CaO, K_2O , Na_2O , ZrO_2 , $CaO.SiO_2$, $K_2O.(TiO_2)_n$, Al_2O_3 . 2 SiO_2 , $CaCO_3$, $MgCO_3$, $BaSO_4$ and $MgSO_4$. The fine particles of organic compounds are those of fatty acids and derivatives thereof and metal salts thereof, and also those of resins such as fluoro resins, polyethylene resins and acrylic resins.

The average particle diameter of the toner used in the present invention is, for example, 3 to 15 micrometers, preferably 4 to 10 micrometers. The storage elastic modulus G' of the toner per se (determined at an angular frequency of 10 rad/sec) at 150°C. is preferably in the range of 10 to 200 Pa for good fusing.

The image receiving element of the present invention preferably comprises a toner receiver layer containing a polymer preferably coated on both surfaces of the above mentioned support coated with a polyolefin resin. The toner receiver layer as mentioned earlier has the function of receiving an image-forming toner from a developing drum or an intermediate transfer medium by (static) electricity, pressure, etc. in the transferring step and fixing the image by heat

and/or pressure, in the fixing step. Further, it also enables the entire surface of the element to develop a substantially uniform gloss after the invention fusing step, particularly after the belt fusing step. The resulting electrophotographic image has the look and feel of a silver halide photographic print. This is not possible on a commercially available standard paper since during the fusing step the thermoplastic is present only in the image areas leading to high differential gloss and difficulty in belt fusing due to differential adhesion forces of various areas of the print to the heated belt.

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The toner receiver layer utilized in the present invention suitably has a dry coverage of 5 to 50g/m², or 8 to 35g/m² in a preferred embodiment for achieving minimum differential gloss and image relief. The coverage of the toner receiver layer is such that it maintains the surface texture of the underlying polyolefin resin layer.

The toner receiver layer of the invention comprises a carboxylic acid containing thermoplastic polymer, alone or blended with other thermoplastic polymers that have a glass transition temperature or Tg that is close to that of the thermoplastic toner that is transferred to the toner receiver layer. Preferably, the Tg of the toner receiver layer is within 10°C of the Tg of the toner. Consequently, the toner and the toner receiver layer both melt when the toner is fixed to the receiving layer by heat and pressure. This contributes to the adhesion of the toner to the layer and to achieving of high gloss in both the toned (D max) and untoned (D min) areas of the image. High gloss and low differential gloss give the resultant prints a photo quality look and feel.

Materials useable for the toner receiver layer of the invention include a carboxylic acid containing thermoplastic polymer which is capable of being deformed at the fixing temperature and also capable of receiving the toner and providing uniform gloss after fusing. It is preferred that the Tg of the toner receiver layer be between 40 and 120°C preferably between 45 and 95°C. Tg values can be determined by techniques such as differential scanning calorimetry or differential thermal analysis, as disclosed in N. F. Mott and E. A. Davis, Electronic Processes in Non-Crystalline Material, Oxford University Press,

Belfast, 1971, at p. 192. The carboxylic acid containing thermoplastic polymers suitable for the toner receiver layer include, for example, vinyl polymers such as acrylic resins, polystyrene resins, polyester resins, polyurethane resins, polyamide resin, polyurea resin, polysulfone resin, and mixtures thereof. The carboxylic acid thermoplastic resins are preferably derived from acrylic polymers, styrenic polymers, styrene/acryl acid ester copolymers, styrene/methacrylic acid ester copolymers, polyurethanes, urethane-vinyl hybrid polymers and mixtures thereof. In many cases, since the above-mentioned resins and copolymers are used for forming the toner, the carboxylic acid containing thermoplastic polymer included in the toner image receiving layer preferably belongs to the same group as that of these resins and copolymers. The preferred thermoplastic resins are water soluble or dispersible to enable coating on to the polyethylene resin layer.

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Suitable carboxylic acid containing thermoplastic polymers useful as the toner receiver layer for the present invention include those obtained by polymerizing one or more ethylenically unsaturated monomers containing carboxylic acid groups include acrylic monomers such as acrylic acid, methacrylic acid, ethacrylic acid, itaconic acid, maleic acid, fumaric acid, citraconic acid, and styrene carboxylic acid such that the acid number of said polymer is between 5 and 70 and more preferably between 5 and 50. Acid number is an indication of the amount of fatty acid (carboxylic acid) groups contained in a polymer and is defined as the number of milligrams of potassium hydroxide required to neutralize the free acid in 1 gram of neat polymer. When the acid number is greater than 70 the toner receiver layer containing the polymer usually has a tendency to absorb enough moisture under conditions of high humidity and block when two imaged elements are placed in face-to-face contact under sufficient weight. Blocking is defined within the scope of this invention as visible deterioration of the surfaces upon separation of two 24 square inch imaged elements placed in face-to-face contact under a 100 gram weight for 24 hours at 38°C and 80% relative humidity. Suitable comonomers also include vinyl esters such as monoalkyl itaconate including monomethyl itaconate, monoethyl itaconate, and monobutyl itaconate,

monoalkyl maleate including monomethyl maleate, monoethyl maleate, and monobutyl maleate, alkyl esters of acrylic or methacrylic acid such as methyl methacrylate, ethyl methacrylate, butyl methacrylate, ethyl acrylate, butyl acrylate, hexyl acrylate, n-octyl acrylate, lauryl methacrylate, 2-ethylhexyl methacrylate, nonyl acrylate, benzyl methacrylate, the hydroxyalkyl esters of the same acids such as 2-hydroxyethyl acrylate, 2-hydroxyethyl methacrylate, and 2-hydroxypropyl methacrylate, the nitrile and amides of the same acids such as acrylonitrile, methacrylonitrile, and methacrylamide, vinyl acetate, vinyl propionate, vinylidene chloride, vinyl chloride, and vinyl aromatic compounds such as styrene, t-butyl styrene and vinyl toluene, dialkyl maleates, dialkyl itaconates, dialkyl methylenemalonates, ethylene, propylene and other alkylenes, isoprene, and butadiene. Suitable polyethylenically unsaturated monomers include butadiene, isoprene, allylmethacrylate, and diacrylates of alkyl diols such as butanediol diacrylate and hexanediol diacrylate, divinyl benzene and the like.

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In a particular embodiment, the carboxylic acid containing thermoplastic polymers of the toner receiver layer of the present invention is preferably derived from acrylic polymers that have the advantage of good adhesion, non-yellowing, are adjustable for high gloss and have a wide range of glass transition and minimum film forming temperatures. Other carboxylic acid containing polymers of choice include the vinyl-urethane hybrid polymers including acrylic-urethane polymers. In such vinyl hybrid polymers, the urethane component provides advantageous properties such as good film-formation, good chemical resistance, abrasion-resistance, toughness, elasticity and durability. The vinyl-urethane hybrid polymers (copolymers or interpenetrating networks) are very different from blends of the two. Such polymers are prepared by polymerizing vinyl addition monomers in the presence of a polyurethane prepolymer or a chain extended polyurethane as described in US Patent 5,695,920. Polymerization of the vinyl monomer in the presence of the polyurethane component of the vinylurethane hybrid polymer, causes the two polymers to reside in the same latex particle as an interpenetrating or semi-interpenetrating network or as a core shell particle resulting in improved resistance to water, organic solvents and

environmental conditions, improved tensile strength and modulus of elasticity. Toner receiver layers derived from carboxylic acid containing polymers in accordance with this invention is particularly advantageous due to superior physical properties including excellent resistance to water permeability, yellowing, exceptional resistance to high temperature and high humidity blocking and toughness necessary for providing resistance to scratches and abrasion.

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Examples of carboxylic acid containing thermoplastic polymers useful in the practice of this invention are the commercially available acrylic polymers NeoCryl A5090, A612, A1110, A1120, A6037, A6075, A6092, A625, A650 and A655 from Avecia. An example of a urethane polymer that is commercially available is Sancure 898 from Noveon and examples of acrylic-urethane hybrid polymers are NeoPac R-9000, R-9699 and R-9030 from Avecia.

Optionally, the carboxylic acid containing thermoplastic polymers in accordance with the invention may also be crosslinked using suitable crosslinking agents. Such an additive can improve the resistance of the toner receiver element to blocking under conditions of high temperature and humidity. Crosslinkers such as epoxy compounds, polyfunctional aziridines, methoxyalkyl melamines, triazines, polyisocyanates, carbodiimides, polyvalent metal cations, and the like may all be considered. If a crosslinker is added, care must be taken that excessive amounts are not used, as this will decrease the melt flow of the toner receiver layer causing possible lowering of gloss and toner adhesion while increasing differential gloss and image relief. The preferred crosslinker is a polyfunctional aziridine crosslinker such as CX100 from Avecia.

In a still further preferred embodiment, the toner receiver layer of the invention also additionally comprises, in addition to the carboxylic acid containing thermoplastic polymer, polyester resins that include polyester ionomers and are obtained by the condensation of a dicarboxylic acid component such as terephthalic acid, isophthalic acid, maleic acid, fumaric acid, phthalic acid, adipic acid, sebacic acid, azelaic acid, abietic acid, succinic acid, trimellitic acid or pyromellitic acid (such a dicarboxylic acid component may have a sulfonic acid group, carboxyl group or the like substituted thereof) with an alcohol component

such as ethylene glycol, diethylene glycol, propylene glycol, bisphenol A, diether derivative of bisphenol A (such as 2 ethylene oxide adduct of bisphenol A or 2 propylene oxide adduct of bisphenol A), bisphenol S, 2-ethylcyclohexyldimethanol, neopentyl glycol, cyclohexyldimethanol or glycerol (such an alcohol component may have a hydroxyl group or the like substituted thereof); polyacrylic acid ester resins or polymethacrylic acid ester resins such as polymethyl methacrylate, polybutyl methacrylate, polymethyl acrylate and polybutyl acrylate; polycarbonate resins; polyvinyl acetate resins; styrene acrylate resins; styrene/methacrylic acid ester copolymer resins and vinyltoluene acrylate resins.

Polyester ionomers are polyesters having ionic (or ionizable) functionalities. The term "polyester ionomer" of the toner receiver, as used herein, includes branched and unbranched, homo and copolymers, crosslinked or uncrosslinked polymers. The polyester ionomer is an essentially hydrophobic, substantially amorphous, thermoplastic polymer in which ionic groups or moieties are present in sufficient number to provide water dispersibility prior to coating. The polyester dispersions provide advantageous properties such as good film-formation, excellent fingerprint resistance, toughness, elasticity and durability. Furthermore, the polyesters exhibit tensile and flexural strength and resistance to various oils.

Procedures for the preparation of polyester ionomers for the toner receiver layer are described in U.S. Pat. Nos. 3,018,272; 3,563,942; 3,734,874; 3,779,993; 3,929,489; 4,307,174, 4,395,475, 5,939,355 and 3,929,489, the disclosures of which are incorporated herein by reference. The polyester ionomers useful in this invention comprise dicarboxylic acid recurring units typically derived from dicarboxylic acids or their functional equivalents and diol recurring units typically derived from diols. Generally, such polyesters are prepared by reacting one or more diols with one or more dicarboxylic acids or their functional equivalents (e.g. anhydrides, diesters or diacid halides), as described in detail in the cited patents. Such diols, dicarboxylic acids and their functional equivalents are sometimes referred to in the art as polymer precursors. It should be noted that,

as known in the art, carbonylimino groups can be used as linking groups rather than carbonyloxy groups. This modification is readily achieved by reacting one or more diamines or amino alcohols with one or more dicarboxylic acids or their functional equivalents. Mixtures of diols and diamines can be used if desired.

The polyesters for the toner receiver layer described herein (referred to as "ionomers" or "polyester ionomers") contain at least one ionic moiety, which can also be referred to as an ionic group, functionality, or radical. In a preferred embodiment of this invention, the recurring units containing ionic groups are present in the polyester ionomer in an amount of from about 1 to about 12 mole percent, based on the total moles of recurring units. Such ionic moieties can be provided by either ionic diol recurring units and/or ionic dicarboxylic acid recurring units, but preferably by the latter. Such ionic moieties can be anionic or cationic in nature, but preferably, they are anionic. Exemplary anionic ionic groups include carboxylic acid, sulfonic acid, and disulfonylimino and their salts and others known to a worker of ordinary skill in the art. Sulfonic acid ionic groups, or salts thereof, are preferred.

One type of ionic acid component has the structure

where M=H, Na, K or NH₄.

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Ionic dicarboxylic acid recurring units can be derived from 5-sodiosulfobenzene-1,3-dicarboxylic acid, 5-sodiosulfocyclohexane-1,3-dicarboxylic acid, 5-(4-sodiosulfophenoxy)benzene-1,3-dicarboxylic acid, 5-(4-sodiosulfophenoxy)cyclohexane-1,3-dicarboxylic acid, similar compounds and functional equivalents thereof and others described in U.K. Patent Specification

No. 1,470,059 (published Apr. 14, 1977). Other suitable polyester ionomers for protective overcoats in the imaged elements of the present invention are disclosed in U.S. Pat. Nos. 4,903,039 and 4,903,040, which are incorporated herein by reference.

Preferred polyester ionomers for use in the present invention include the EASTMAN AQ™ polymers manufactured by Eastman Chemical Company of Kingsport, Tenn.

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The polyester ionomers utilized in the toner receiver layer of this invention have a molecular weight (Mn) preferably above 10,000, more preferably above about 14,000 that disperse directly in water without the assistance of organic co-solvents, surfactants, or amines. As indicated above, this water dispersibility is attributable in large part to the presence of ionic substituents, for example, sulfonic acid moieties or salts thereof, for example, sodiosulfo moieties (SO₃Na) in the polymer. Properties and uses of these polymers are described in Publication No. GN-389B of Eastman Chemical Company, dated May 1990, the disclosure of which is incorporated herein by reference. Especially preferred is poly[1,4-cyclohexylenedimethylene-co-2,2'-oxydiethylene (46/54) isophthalate-co-5-sodiosulfo-1,3-benzenedicarboxylate (82/18)] (obtained as EASTMAN AQTM 55 polymer, Tg 55°C from Eastman Chemical Co.).

The commercially available salt forms of the polyester ionomer, including the aforementioned AQ® polymers, have been shown to be effective in the present invention. Mixtures of polyester ionomers can be used if desired. The polyester resins, such as the polyester ionomers are present in the toner receiver layer such that it comprises up to 75 weight percent of the thermoplastic polymer of the toner receiver layer.

An average molecular weight of the thermoplastic polymers used for forming the toner image receiving layer is preferably larger than that of the thermoplastic polymer used for forming the toner. However, the average molecular weight may change depending on the relationship between the thermodynamic properties of the toner polymer and that of the image-receiving layer polymer. For example, when the softening temperature of the image

receiving layer polymer is higher than that of the toner polymer, it is preferred that the average molecular weight of the thermoplastic polymer used for forming the toner image receiving layer be equal to or lower than that of the thermoplastic polymer used for forming the toner. It is also preferred to use a mixture of various thermoplastic polymers having the same compositions but different average molecular weights as the thermoplastic polymer for forming the toner image-receiving layer. It is preferred that the molecular weight distribution of the thermoplastic polymer used for forming the toner image receiving layer be broader than that of the thermoplastic polymer for forming the toner.

The carboxylic acid containing polymers for forming the toner receiver layer used in the present invention are preferably water-soluble polymers or a water-dispersed polymer such as the carboxylic acid containing thermoplastic polymers and the polyesterionomers described above. The water-soluble carboxylic acid containing polymer is not particularly limited as regards a desirable composition, bonding structure, molecular structure, molecular weight, molecular weight distribution and form so long as it is a polymer which is soluble in water. In this connection, the molecular structure of the water soluble carboxylic acid containing polymer is preferably a water-soluble group such as a sulfonic acid group, a hydroxyl group, a carboxylic acid group, an amino group, an amido group and an ether group. The water-soluble carboxylic acid containing polymers may also include, for example, vinylpyrrolidone/vinyl acetate copolymer, styrene/vinylpyrrolidone copolymer, styrene/maleic anhydride copolymer, water-soluble polyesters, water-soluble polyurethane, water-soluble nylon and water-soluble epoxy resin.

The water-dispersed carboxylic acid containing polymer is suitably selected from among a water-dispersed type resin such as water-dispersed carboxylic acid containing thermoplastic resins and the polyesterionomer resins described above, acrylic resins, water-dispersed polystyrene based latices and water- dispersed urethane resins; an emulsion such as acrylic resin emulsion, polyvinyl acetate emulsion, SBR (styrene/butadiene/rubber) emulsion; and a

copolymer, mixture or cation denaturant thereof. A combination of two or more of these resins is also usable.

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The toner receiver layer of the present invention also preferably contains a fuser-oil sorbent additive. Fuser-oil sorbent additives include adsorbents and absorbents and may be any suitable material. They have specific physical and chemical properties that allow them to capture the excess fuser-oil. Sorbent additives may be organic or inorganic and may be synthetic. Typical of such materials are clay, talc, glass wool, silica, peat moss, synthetic fibers such as nylon, plastic adsorbent microspheres and the like. The preferred material are clay and talc since they are readily available in a manner that can be easily formulated into coating dispersions for the toner receiver layer, can be obtained at a high brightness index and is inexpensive. Of these kaolin clay is optimum and is used in an amount greater than 10 weight percent of the toner receiver layer and preferably from 10 to 60 weight percent of the layer. The amount of sorbent additive such as clay in the layer can also be used to control the level of gloss in the imaged element, especially after belt fusing. The clay additive is also used to enable writablity after printing. This is especially useful in applications such as greeting cards and calendars where writability is important. The fuser-oil sorbent additive such as the clays usable herein preferably have a GE brightness index greater than 88% and include various modified and unmodified clays including nanoclays. Brightness is the percent of blue light reflected off a sample measured at an effective wavelength of 457nm. GE brightness is a directional brightness measurement utilizing essentially parallel beams of light to illuminate the paper surface at an angle of 45 degrees.

The clay materials suitable for the sorbent additive of this invention include phyllosilicates, e.g., montmorillonite, particularly sodium montmorillonite, magnesium montmorillonite, and/or calcium montmorillonite, nontronite, beidellite, volkonskoite, hectorite, saponite, sauconite, sobockite, stevensite, svinfordite, vermiculite, magadiite, kenyaite, talc, mica, kaolinite (kaolin or china clay), and mixtures thereof. Preferred clays are swellable so that other agents, usually organic ions or molecules, can intercalate or exfoliate the

layered material resulting in a desirable dispersion of the inorganic phase. The aforementioned clay can be natural or synthetic, for example, synthetic smectite clay. For this invention, the clay particles in the dispersed form preherfably have a particle size where greater then 90% of the particles are less than or equal to 2 micrometers. A preferred average particle size is between 0.1 and 2 micrometers for good oil absorption and uniform coating.

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The clay used in this invention can be an organoclay. Organoclays are produced by interacting the unfunctionalized clay with suitable intercalants. These intercalants are typically organic compounds, which are neutral or ionic. Useful neutral organic molecules include polar molecules such as amides, esters, lactams, nitriles, ureas, carbonates, phosphonates, sulfates, sulfonates, nitro compounds, and the like. The neutral organic intercalants can be monomeric, oligomeric or polymeric. Neutral organic molecules can cause intercalation in the layers of the clay through hydrogen bonding, without completely replacing the original charge balancing ions. Useful ionic compounds are cationic surfactants including onium species such as ammonium (primary, secondary, tertiary, and quaternary), phosphonium, or sulfonium derivatives of aliphatic, aromatic or arylaliphatic amines, phosphines and sulfides. Typically onium ions can cause intercalation in the layers through ion exchange with the metal cations of the preferred smectite clay. A number of commercial organoclays such as Kaogloss 90, hydrous aluminum silicate kaolin, obtained from Theile Kaolin Company and Cloisite 15A, a natural montmorillonite modified with a quaternary ammonium salt, available from clay vendors, such as Southern Clay Products and Nanocor, may be used in the practice of this invention.

The electrophotographic image receiving element utilized in the present invention preferably has a high degree of whiteness and brightness, when used for photo rich applications. Whiteness refers to the extent that paper diffusely reflects light of all wavelengths throughout the visible spectrum. Whiteness is an appearance term. As for the degree of whiteness, L* value, a measure of the lightness (100 for perfect white to 0 for black), in CIE 1976 (L*a*b*) color space is at least 80, preferably at least 85 and more preferably at least 90. The coloration

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of the white color is desirably as neutral as possible. As for the coloration of the white color, the value of $(a^*)^2 + (b^*)^2$ in L*a*b* space is preferably not higher than 50, more preferably not higher than 18 and most preferably not higher than 5. The chromaticity dimensions (a and b) give designations of color such that "a" measures redness when positive, gray when zero and greenness when negative, and "b" measures yellowness when positive, gray when zero and blueness when negative. The L*a*b* space is designed to have a uniform correspondence between geometric distances and perceptual distances between colors that are seen under the same reference illumination. The perceived difference between any two colors is proportional to the geometric distance in color space between their colors. Brightness is defined as the percentage reflectance of blue light only at a wavelength of 457 nm. Brightness is an arbitrarily defined, but carefully standardized, blue reflectance that is used throughout the pulp and paper industry for the control of mill processes and in certain types of research and development programs. Brightness is not whiteness. However, the brightness values of the pulps and pigments going into the paper provide an excellent measure of the maximum whiteness that can be achieved with proper tinting. The degree of whiteness for the electrophotographic image receiving sheet of the present invention is preferable greater than 90 in order to attain the look of photographic paper.

The materials useable for the toner receiver layer of the present invention may contain various additives in order to improve the stability of the output image and also the stability of the toner receiver layer itself. The additives for these purposes include various known antioxidants, agents for preventing aging, agents for preventing deterioration, agents for preventing deterioration caused by ozone, ultraviolet absorbers, light stabilizers, antiseptics and antifungal agents. The toner receiver layer of this invention may also contain other additives such as colorants, tinting agents, brightening agents, plasticizers, slip agents and the like.

The toner receiver layer may also contain matte particles to improve transport through the electrophotographic machine or improve writability

of the imaged element. The matte particles have an average diameter between 10 and 20 micrometers, preferably between 10 and 15 micrometers. The amount of matte particles incorporated in the toner receiver layer is dependent on the thickness of the toner receiver layer. Generally the matte has a dry coverage between 0.11 g/m² and 0.22 g/m², and preferably between 0.11 and 0.165 g/m² in order to obtain good writability and transport without compromising the surface gloss after imaging and fusing the toned image.

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The matte particles may comprise inorganic materials such as silica or alumina or polymeric materials. To maintain their shape after the fusing process, the polymeric matte particle should have a glass transition temperature greater than the fusing temperature or be crosslinked. The polymeric particles are generally made from vinyl monomers and / or divinyl monomers. Preferred polymeric particles are crosslinked acrylates or styrenic polymers. Most preferred are particles comprising poly(methyl methacrylate-co-divinylbenzene), poly(methyl methacrylate-co-ethylene glycol dimethacrylate) or poly(styrene-co-divinylbenzene).

Although the image receiving element of this invention is highly suitable for electrophotographic fusing methods that utilize a fuser-oil at the fuser nip, it may also be used with fuser-oil free systems.

The coating composition for the toner receiver layer of this invention can be applied by any of a number of well known techniques, such as dip coating, rod coating, blade coating, air knife coating, gravure coating and reverse roll coating, slot coating, extrusion coating, bead coating, curtain coating, spray coating and the like. Exemplary bead coating methods and apparatus are disclosed in U.S. Pat. No. 2,761,417 to Russell et al., U.S. Pat. No. 3,474,758 to Russell et al., U.S. Pat. No. 2,761,418 to Russell et al., U.S. Pat. No. 3,005,440 to Padday, and U.S. Pat. No. 3,920,862 to Damschroder et al. Exemplary curtain coating methods and apparatus are disclosed in U.S. Pat. No. 3,508,947 to Hughes, U.S. Pat. No. 3,632,374 to Greiller, and U.S. Pat. No. 4,830,887 to Reiter. Known coating methods are described in further detail in Research Disclosure No. 308119, Published December 1989, pages 1007 to 1008.

After applying the coated layers to the support, it may be dried by any of several methods including, but not limited to, heating belts, high temperature radiant sources and convective heating. Almost exclusively, however, the dryers in large scale coating machines primarily rely on convective heating. In convective heating, a heater is used to heat a gas, or mixture of gasses (e.g. – air), that is introduced into the dryer. This lowers the relative humidity of the gas (or gasses), which is then circulated by blowing it through the dryer sections. Several modes of circulation may be employed: co-current or counter-current to the direction of the web, or in a random fashion. Known drying methods are described in further detail in Research Disclosure No. 308119, Published December 1989, pages 1007 to 1008.

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In typical large scale coating machines using convective drying, the dryer setting can vary, depending on the length of the dryer and the load (amount of material to be dried). If the dwell time in drying zones is short and/or the load is heavy, a more rapid rate of solvent evaporation may be required. This can be achieved through higher temperatures and/or a large difference between the dry bulb temperature and wet bulb temperature of the supply gas (or gasses). For the aqueous coating composition of the toner receiver layer of this invention, it has been found that the coated layer uniformity was enhanced when the difference in dry bulb and wet bulb temperatures of the supply gas (or gasses) was no greater than 38°C, preferably no greater than 21°C, and most preferably no greater than 10°C. The specific sequence of drying conditions is based on the composition and wet coverage of the coated layer to prevent or minimize cracking and other drying defects.

While electrophotographic printing is preferred with the image receiving element of this invention, it may also be utilized with other printing modabilities such as inkjet, offset, and thermal.

The following examples illustrate the practice of this invention.

They are not intended to be exhaustive of all possible variations of the invention.

Parts and percentages are by weight unless otherwise indicated.

EXAMPLES

The polyester binder used in the following examples was a polyester ionomer, AQ55, purchased from Eastman Chemical Company.

Kaogloss 90, kaolin clay was obtained from Theile Kaolin Company as a 70 wt.%

dispersion in water. Aerosol® OT, dioctyl sodium sulfosuccinate, an anionic surfactant from Cytec Industries was used as the coating surfactant for the toner receiver layers coated from water. Carboxylic acid containing acrylic thermoplastic polymers, NeoCryl A 5090 (acid number 12), NeoCryl A612 (acid number 39), NeoCryl A6037 (acid number 53), and CX-100 (polyaziridine crosslinker) were purchased from Avecia. Carboxylic acid containing urethane thermoplastic polymer Sancure 898 (acid number 35) was purchased from Noveon.

A polyethylene resin melt containing 11.4wt% TiO2, 87.7wt% LDPE, and 0.9wt% of a mixture of colorants, optical brighteners and antioxidants, was extrusion coated on both sides of a 160 micrometer thick photographic paper support at 288-332°C. The surface finish of the resin coated paper was controlled by the finish on the chill roll used in the extrusion process. Polyethylene resin coated paper prepared thus was used for all the examples below.

Examples 1-7

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20 Preparation of the electrophotographic toner receiver layer (TRL)

A 32 weight percent aqueous solution of a mixture of NeoCryl A5090, AQ 55 and Kaogloss 90 in a the weight ratio specified in the Table 1 was coated on a corona discharge-treated, polyethylene resin coated paper described above to yield a dry coverage of 10.76 g/m² coating of Neocryl 5090 and AQ55.

Similar receiver sheets in accordance with the invention were prepared by substantially the same procedure and with the different toner receiver components and weight ratios, including varying levels of AQ55, as appropriate. In some cases the carboxylic acid polymer was crosslinked using CX100.

A control coating that did not include the carboxylic acid containing thermoplastic polymer was also prepared. The composition of the

prepared control coating and TRL coated media of the invention are given in TABLE 1.

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The TRL coated media and the control media were printed in the NexPress 2100 printer. The printed sheets were next belt fused using a 3 mil polyimide belt at 165°C, 6ips, 35psi(nip pressure). The printed sheets were next belt fused using a 3 mil polyimide belt at 165°C, 6ips, 35psi(nip pressure). The resultant prints were evaluated for humidity and thermal blocking. Humidity blocking was evaluated by first conditioning 4"x6" D min samples of the printed and belt fused sheets at 38°C and 80% RH for 24 hours, followed by placing the samples face to face and applying 1000gm weight on the samples for 24 hours up to one week in the same environment. The samples were then peeled apart and examined for surface distortion on a scale of 0-5 where 0 = no surface distortion and 5=severe surface distortion. Thermal blocking was measured by placing the samples face to face and applying 1000gm weight on the samples for up to 6 hours in a chamber at 71°C and 50% RH. The samples were then peeled apart and examined for image distortion on a scale of 0-5 where 0 = no image distortion and 5=severe distortion. Gloss measurements(60°) were made on the belt fused samples using a BYK Gardner Glossmeter in a Dmin(white) and Dmax(black area). Writability was assessed by writing on the surface of the printed sheets with a No. 2 pencil and/or a ballpoint pen before belt fusing.

Toner adhesion of the belt fused samples were measured using a commercial Microscratch Test Device developed and produced by CSM Instruments (Neuchatel, Switzerland). All samples were conditioned for 18 hours at 23°C/50% RH prior to testing. After this conditioning period, ramped load scratches were generated in the 2-300 gram load range in imaged areas using an angled Silicon Carbide cylinder with a 5 micrometer edge radius as the abrader. The cylinder was held at a fixed angle of 45 degrees relative to the surface of the sample in all cases in order to reduce downward penetration of the cylinder edge and enhance shear forces in interfacial regions. A scratch velocity of 10 mm/minute and a loading rate of 298 grams/minute were used in all cases. After scratch generation was complete, an optical microscope was used to examine

scratch morphologies and determine the load required to initiate color removal, which was used as a measure of toner layer durability (adhesive and cohesive) and interlayer adhesion within the media structure.

As can be seen from Table 1, the presence of the carboxylic acid containing thermoplastic polymer in the TRL imparts blocking resistance to the prints under high humidity and high temperature compared to the Control which blocked under the same conditions. The higher the content of the carboxylic acid containing polymer, the better the resistance.

TABLE 1

Example #	Carboxylic acid polymer in TRL	Ratio of AQ to carboxylic acid polymer	Gloss 60° Dmin (Dmax)	Humidity Blocking 1 week	Thermal Blocking
Control	None	100:0	76(83)	5	5
1	NeoCryl A5090	25:75	75(80)	0	0
2	NeoCryl A612	25:75	80(82)	0	0
3	NeoCryl A6037	25:75	72(77)	0	0
4	NeoCryl A6037	50:50	66(82)	2	0
5	Sancure 898	25:75	66(79)	0	0
6	NeoCryl A 5090, 1% CX 100	0:100	64(71)	0	0
7	NeoCryl A 5090, 3% CX 100	0:100	66(69)	0	0

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The Examples of the invention showed little differential gloss after belt fusing and the overall gloss was much higher than the commercial clay coated papers which are not glossable using a belt fuser. Lustro Laser paper has a Dmin and Dmax 60° gloss value of 33.6 and 34.5 respectively and Stora Enzo 4CC Art

paper had 60° gloss values of 24.6 and 45.6 in the Dmin and Dmax areas respectively compared to the much higher gloss values attainable with the TRLs of the invention due to the belt fusability of prints on the papers of this invention.

Adhesion of toner as measured by onset of failure in adhesion test

(Table 2) described above showed that the presence of the carboxylic acid
containing polymer did not significantly compromise the toner adhesion to the
TRL and the adhesion numbers were well above what was typically seen with
commercially available papers such as Lustro Laser where the numbers varied
from 35 – 80g for scrape adhesion onset.

Table 2

Example #	Carboxylic acid polymer in TRL	Ratio of AQ to carboxylic acid polymer	Scrape Adhesion
			Onset (g)
Control	None	100:0	198
1	NeoCryl A5090	25:75	152.4
5	Sancure 898	25:75	110
6	NeoCryl A 5090, 1% CX 100	0:100	166.1
7	NeoCryl A 5090, 3% CX 100	0:100	150.3

CLAIMS:

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1. An image receiving element comprising in order a support, at least one polyolefin resin coating and at least one toner receiver layer wherein said toner receiver layer comprises fuser-oil sorbent additive and a carboxylic acid containing thermoplastic polymer, wherein said sorbent additive is present in an amount greater than 10% by weight of said toner receiver layer.

- 2. The image receiving element of claim 1 wherein said sorbent additive is clay.
- The image receiving element of claim 2 wherein said clay comprises montmorillonite, particularly sodium montmorillonite, magnesium montmorillonite, and/or calcium montmorillonite, nontronite, beidellite,
 volkonskoite, hectorite, saponite, sauconite, sobockite, stevensite, svinfordite, vermiculite, magadiite, kenyaite, talc, mica, kaolinite (kaolin or china clay), organic clays such as Cloisite 15A and mixtures thereof.
- 4. The image receiving element of claim 1 wherein said sorbent additive has a GE brightness greater than 88%.
 - 5. The image receiving element of claim 1 wherein said sorbent additive is present in an amount of between 10 and 60 weight percent of said toner receiver layer.

- 6. The image receiving element of claim 1 wherein said image receiving element further comprises a carboxylic acid containing thermoplastic polymer dispersible in water.
- 7. The image receiving element of claim 1 wherein said toner receiver layer further comprises a polyester.

8. The image receiving element of claim 1 wherein said toner receiver layer further comprises a polyester ionomer.

- 9. The image receiving element of claim 1 wherein said
 5 carboxylic acid containing thermoplastic polymer comprises a low acid number polymer.
 - 10. The image receiving element of claim 9 wherein said acid number is between 5 and 70.
- 11. The image receiving element of claim 1 wherein said carboxylic acid containing thermoplastic polymer comprises a crosslinked low acid number polymer.
- 12. The image receiving element of claim 1 wherein said15 carboxylic acid containing thermoplastic polymer comprises an acrylic polymer.
 - 13. The image receiving element of claim 1 wherein said carboxylic acid containing thermoplastic polymer comprises a urethane polymer.
- 20 14. The image receiving element of claim 1 wherein said sorbent additive has an average particle size of between 0.1 and 2 micrometers.
 - 15. The image receiving element of claim 1 wherein said at least one toner receiver layer has a dry coverage of between 8 and 35g/m².
 - 16. The image receiving element of claim 1 wherein said resin coating comprises at least one layer of polyethylene and titanium dioxide.
- 17. The image receiving element of claim 1 wherein said support comprises paper.

18. The image receiving element of claim 1 wherein said carboxylic acid containing thermoplastic polymer comprises a polymer having carboxylic acid groups derived from ethylenically unsaturated monomers selected from the group consisting of acrylic acid, methacrylic acid, itaconic acid, maleic acid, fumaric acid, styrene carboxylic acid, and mixtures thereof.

19. The image receiving element of claim 1 wherein said carboxylic acid containing thermoplastic polymer comprises urethane-vinyl copolymer.

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- 20. The image receiving element of claim 1 wherein said toner receiver layer further comprises a polyester or polyester ionomer dispersible in water.
- 15 21. The image receiving element of claim 20 wherein said polyester or polyester ionomer is present in an amount of up to 75 weight percent of the thermoplastic polymer.
- 22. The image receiving element of claim 20 wherein said polyester or polyester ionomer is present in an amount of between 0 and 50 weight percent of said thermoplastic polymer.
- 23. An imaged element comprising an image receiving element comprising in order a support, at least one polyolefin resin coating and at least one toner receiver layer wherein said toner receiver layer comprises fuser-oil sorbent additive and a carboxylic acid containing thermoplastic polymer, wherein said sorbent additive is present in an amount greater than 10% by weight of said receiver layer, wherein said at least one toner receiver layer has an image thereon formed from toner comprising pigment and bisphenol A polyester.

24. The imaged element of claim 23 wherein said element has a 60° gloss greater than 60 in non-imaged areas after belt fusing or calendering of said element.

- 5 25. The imaged element of claim 23 wherein said element has pencil writability.
 - 26. The imaged element of claim 23 wherein said element has ballpoint pen writability.

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27. The imaged element of claim 23 wherein when said imaged element is placed in face-to-face contact with another of said imaged element and a 1000 gm per 24 square inches load is applied for 24 hours at 38°C and 80% relative humidity, the imaged elements will separate without deterioration of the images of said imaged elements.

INTERNATIONAL SEARCH REPORT

International application No
/US2005/041569

A. CLASSIFICATION OF SUBJECT MATTER
INV. G03G7/00 B41M5/00 According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) GO3G B41M Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, PAJ, WPI Data C. DOCUMENTS CONSIDERED TO BE RELEVANT Category' Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. US 5 827 627 A (CLECKNER ET AL) χ 1 - 2227 October 1998 (1998-10-27) column 4, line 28 - line 52 column 6, line 28 - line 36 Y 23-27 column 7, line 50 - line 55; claims 1-13; example 3 US 5 789 123 A (CLECKNER ET AL) χ 1-22 4 August 1998 (1998-08-04) column 4, line 22 - column 5, line 56 23-27 column 8, line 30 - line 43; claims 1-15: examples 1-3Υ US 4 489 150 A (CHANG ET AL) 23-27 18 December 1984 (1984-12-18) column 1, line 35 - line 48; claims 1-7 X X See patent family annex. Further documents are listed in the continuation of Box C. Special categories of cited documents: *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the *A* document defining the general state of the art which is not considered to be of particular relevance invention "E" earlier document but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone filing date *L¹ document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled "O" document referring to an oral disclosure, use, exhibition or *P* document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 07/04/2006 30 March 2006 Authorized officer Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31–70) 340–2040, Tx. 31 651 epo ni, Fax: (+31–70) 340–3016 Bolger, W

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