(19) World Intellectual Property Organization International Bureau





(43) International Publication Date 9 October 2003 (09.10.2003)

(10) International Publication Number WO 03/082760 A1

- (51) International Patent Classification7: C03C 17/34, 17/42, B32B 9/00
- (21) International Application Number: PCT/US03/00578
- **(22) International Filing Date:** 10 January 2003 (10.01.2003)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data: 60/366,528 25 March 2002 (25.03.2002)
- GUARDIAN INDUSTRIES CORP. (71) Applicant: [US/US]; 2300 Harmon Road, Auburn Hills, MI 48326-1714 (US).
- (72) Inventor: MURPHY, Nestor, P.; 1517 Steward Road, #342, Monroe, MI 48162 (US).
- (74) Agent: DAVIDSON, Bryan, H.; Nixon & Vanderhye P.C., 1100 North Glebe Road, Suite 800, Arlington, VA 22201-4714 (US).

- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ANTI-REFLECTIVE HYDROPHOBIC COATINGS AND METHODS

(57) Abstract: Substrates have anti-reflective hydrophobic surface coatings comprised of the reaction products of a vapor-deposited chlorosilyl group containing compound and a vapor-deposited alkylsilane. Most preferably the substrate is glass. In one preferred form of the invention, highly durable antireflective hydrophobic coatings may be provided by forming a silicon oxide anchor layer from a humidified reaction product of silicon tetrachloride, followed by the vapor-deposition of a chloroalkylsilane, preferably dimethyldichlorosilane (DMDCS). The layer thickensses of the anchor layer and the overlayer are such that the coating exhibits light reflectance of less than about 1.5% (more preferably less than about 1.0%) at a wavelength of about 525 nm (+/- 50 nm).

- 1 -

ANTI-REFLECTIVE HYDROPHOBIC COATINGS AND METHODS

CROSS-REFERENCE TO RELATED APPLICATION

This application is based on, and claims domestic priority benefits under 35 U.S.C. § 119(e) from, U.S. Provisional Patent Application Serial No. 60/366,528 filed on March 25, 2002, the entire content of which is expressly incorporated hereinto by reference.

5

10

15

20

FIELD OF THE INVENTION

The present invention relates generally to coated substrates and methods of coating the same. In preferred embodiments, the present invention relates to transparent substrates having an anti-reflective hydrophobic (water repellant) coating thereon.

BACKGROUND AND SUMMARY OF THE INVENTION

Glass is typically made of silicates that are melted to form a clear, transparent, solid material. The fundamental molecular structural unit of conventional glass is a SiO₄ tetrahedron. Ordinary float glass (named for its production process whereby a molten ribbon of glass is floated on molten metal to provide a smooth surface) includes additional amounts of soda (Na₂O), usually in the form of sodium carbonate or nitrate during the production process, lime (CaO) and other oxides (usually aluminum and magnesium oxides) to form a soda-lime-silica structure known colloquially as soda-lime glass. Other specialized glass can be prepared by the introduction of other additives and constituents.

It is sometimes highly desirable for conventional glass to have hydrophobic (water repellant) surface properties when employed in certain

- 2 -

end-use applications, such as for automotive window glass. Various proposals exist to impart hydrophobic (water-repellant) properties to glass substrates. For example, U.S. Patent Nos. 4,263,350, 4,274,856, 5,665,424 and 5,723,172 (the entire content of each being incorporated expressly hereinto by reference) disclose generally that glass surfaces can be coated with a vapor deposited layer of an chloroalkylsilane, such as dimethyldichlorosilane (DMDCS) so as to improve their hydrophobicity and/or release properties. Other proposals exist whereby a fluoroalkylsilane (FAS) coating may be employed to "cap" an underlayer on a glass substrate so as to improve coating durability. Please see in this regard, U.S. Patent Nos. 5,328,768, 5,372,851, 5,380,585 and 5,580,605 (the entire content of each being incorporated expressly hereinto by reference). In addition, International Application WO 00/25938 (the entire content of which is expressly incorporated hereinto by reference) discloses that a silicon film consisting of chains of siloxane groups each terminating in an end molecule which reacts with water to form an OH group, may be capped by further reaction of that OH group with trimethylchlorosilane to form trimethylchlorosiloxane.

5

10

15

20

25

In commonly owned, U.S. Patent Application Serial No. 09/921,303, filed on February 1, 2001(the entire content of which is expressly incorporated hereinto by reference) there are disclosed coated substrates (preferably glass) and methods which exhibit improved hydrophobicity and durability. In some of the especially preferred embodiments of the present invention, coated substrates and methods are provided which include an SiO_x-containing anchor layer comprised of a controllably humidified vapor phase deposition of a chlorosilyl group containing compound (most preferably silicone tetrachloride), and a

- 3 -

hydrophobic capping layer chemically bonded to the SiO_x-containing anchor layer.

5

10

15

20

25

It has now been discovered that coated substrates of the type disclosed and claimed in the above-identified U.S. Application Serial No. 09/921,303 may also be rendered anti-reflective. Thus, according to one aspect of this invention, substrates may be provided with an anti-reflective hydrophobic surface coating comprised of the reaction products of a chlorosilyl group containing compound and an alkylsilane. Most preferably the substrate is glass. In one preferred form of the invention, highly durable hydrophobic coatings may be formed by forming a silicon oxide anchor layer from a humidified reaction product of silicon tetrachloride, followed by the vapor-deposition of a chloroalkylsilane, preferably dimethyldichlorosilane (DMDCS). The layer thicknesses of the anchor layer and the overlayer are such that the coating exhibits light reflectance of less than about 1.5% (more preferably less than about 1.0%) at a wavelength of about 525 nm (+/- about 50 nm).

These and other aspects and advantages will become more apparent after careful consideration is given to the following detailed description of the preferred exemplary embodiments thereof.

BRIEF DESCRIPTION OF THE ACCOMPANYING DRAWINGS

Reference will hereinafter be made to the accompanying drawings wherein like reference numerals throughout the various FIGURES denote like structural elements, and wherein;

FIGURE 1 is a schematic depiction of a coated glass substrate in accordance with the present invention; and

-4-

FIGURES 2-6 are theoretical plots of reflectance (%) versus wavelength (nm) of a coated substrate having a structure shown in FIGURE 1 at different SiO_x base layer and DMDCS overlayer thicknesses.

DETAILED DESCRIPTION OF THE INVENTION

5

Virtually any substrate that is self-supporting and has, or may be induced to have, active surface hydrogen atoms may be coated in accordance with the present invention. Thus, rigid or flexible substrates formed of glass, plastics, ceramics and the like may be coated in accordance with the present invention. Most preferably, the substrate is glass, with conventional soda-lime float glass being especially preferred.

10

15

In one particularly preferred embodiment of a coated substrate 10 in accordance with the present invention as shown in FIGURE 1, an anchor layer 12 comprised of a silicon oxide (SiO_x) is formed by vapordeposition of a silicon-oxide forming compound onto a glass substrate 14 in a controllably humidified environment. In especially preferred forms of the invention, the silicon oxide layer may be obtained by the reaction of a compound having a chlorosilyl group, such as silicon tetrachloride (SiCl₄), with the surface of the glass to form an underlayer containing SiO₄. Other silanes that form silicon oxide may optionally, or additionally, be employed, such as hexachlorodisiloxane. When using silicon tetrachloride, it has been found that diluting 1 part the silicone tetrachloride with 10 parts pentane is particularly effective.

20

Vapor-phase silicon tetrachloride is most preferably introduced into a closed chamber having a controlled interior humidity environment which results in chlorine-terminated silicon oxide chains to attach directly to the glass substrate surface. Rehydrating the chlorine-terminated silicon oxide

25

chains (e.g., by controlling the humidity in the chamber) will replace the terminal chlorine atoms with hydrogen atoms so that, upon the sequential introduction of vapor-phase chloroalkylsilanes with intermediate rehumidification of the chamber, such as dimethyldichlorosilane (DMDCS), a durable hydrophobic layer 16 over the SiO₄ anchor layer is obtained. Preferred alkylchloro silanes that may be used in accordance with the present invention are represented by the formula Cl_xSiR_y, where x is 1 to 3, y is 4 - x, and R is an alkyl group.

5

10

15

20

25

The humidity during vapor-phase deposition of the silicon oxide anchor layer is important to achieve the desired end result of a durable hydrophobic coating on the substrate. In addition, controlled humidity during vapor phase deposition of the silicon oxide layer is important to achieve a coating with low haze characteristics. Thus, the humidity during vapor phase deposition of the silicon oxide anchor layer from silicon tetrachloride should be less than about 50% relative humidity, and advantageously less than about 45% relative. Preferably the relative humidity within the chamber is controlled to be about 40% or less. Thus, the silicon oxide layer will most preferably exhibit haze (non-specular light scattering) of less than about 3.0%, and typically less than about 2.0%. Advantageously, the haze of the silicon oxide layer will be less than about 1.5%, particularly less than about 0.8%.

The lower limit of relative humidity, and hence haze value, of the silicon oxide anchor layer is determined by the surface roughness that is desired. In this regard, it has been found that the greater the humidity, the greater the surface roughness of the resulting silicon oxide anchor layer and vice versa. Without wishing to be bound to any particular theory, it is believed that the surface roughness of the silicon oxide layer contributes materially to the durability of the hydrophobic coatings obtained according

-6-

to the invention as the peaks and valleys of the "rough" anchor layer provides physical pockets where the later applied chloroalkylsilane can be deposited. In addition, a "rough" anchor layer of silicon oxide may provide an increased surface area resulting in the chloroalkylsilane being more dense per unit area on the substrate thereby possibly improving durability properties of the resulting coating.

5

10

15

20

25

The coated substrates of the present invention will exhibit a tilt angle (30 μ L droplet size) of about 35° or less, and typically 30° or less. For some embodiments of the present invention, extremely low tilt angles of about 20° or less, or even about 10° or less, are obtainable. The coatings of the present invention are also highly durable. That is, the coated substrates of the present invention will exhibit a contact angle after 300 Taber abrasion cycles of greater than about 65°, and typically greater than about 70°. Even after 1000 Taber cycles, the coated substrates of the present invention will exhibit a contact angle of greater than about 60°, usually between about 65° to about 75°.

The coated substrates of the present invention can be conveniently produced using a closed reaction chamber configured to have an inlet opening for the chemical vapors, and a discharge opening to allow the chamber to be exhausted. The substrates are cleaned thoroughly and rinsed prior to being placed in the reaction chamber. The humidity within the chamber is controlled by the introduction of water vapor in dependence upon the chemical vapors being deposited. Thus, humidity within the reaction chamber of greater than about 10%, and less than about 80% are typically employed. The reaction chamber is most preferably maintained under ambient temperature (20°C-25°C) and atmospheric pressure (about 1.0 atmosphere) conditions during the vapor deposition of the underlayer and capping layer.

-7-

Important to the present invention are the respective thicknesses of the silicone oxide anchor layer and the vapor-deposited DMDCS overlayer. In this regard, the silicone oxide layer will most preferably be a layer obtained by vapor-depositing SiCl₄ and will exhibit an index of refraction of between about 1.42 to about 1.46 (typically about 1.44). The vapor-deposited DMDCS overlayer will typically exhibit an index of refraction of between about 1.28 to about 1.38 (typically about 1.35). The glass substrate will have an index of refraction of about 1.52 while air has an index of refraction of about 1.0. Thus, there exists according to the present invention a gradual decrease of the indices of refraction from the glass substrate to the air via the silicon oxide anchor layer and the DMDCS overlayer.

5

10

15

20

25

The relative thicknesses of the silicon oxide layer and the vapordeposited DMDCS layers are selected so that substantially minimal reflectance is obtained at a light wavelength of about 525 nm (+/- about 50 nm). The reflectance advantageously will be less than about 1.5%, more preferably less than about 1.0%, at light wavelengths of about 525 nm (+/- about 50 nm). That is, according to the present invention, the term "anti-reflective" is meant to refer to coatings having light reflectance of less than about 1.5%, and more preferably less than about 1.0%, at wavelengths of between about 475 nm to about 575 nm. In this regard, thicknesses of the silicon oxide anchor layer of between about 1600Å to about 2000Å may be employed in the practice of the present invention, while thicknesses of the vapor-deposited DMDCS layer may range from about 800Å to about 1200Å. In an especially preferred embodiment, the thickness of the silicon oxide layer is about 1800Å, while the thickness of the vapor-deposited DMDCS layer is about 1000Å. As shown in FIGURES 2-6, such combinations of layer thicknesses will provide for

-8-

minimal reflectance of less than about 1.5% at light wavelengths of about 525 nm (+/- about 50 nm).

It is contemplated that other transparent layers may be employed in the practice of the present invention. Thus, a layer may be interposed between the glass substrate and the silicon oxide anchor layer, provided that it exhibits an index of refraction which is between the indices of refraction exhibited by the silicon oxide layer and the glass substrate. Furthermore, a transparent layer may be interposed between the DMDCS overlayer and the silicon oxide, provided it exhibits an index of refraction which is between the indices of refraction exhibited by the DMDCS overlayer and the silicon oxide layer. And, as noted above, a capping layer may be applied over the DMDCS overlayer provided that such capping layer exhibits an index of refraction that is intermediate the indices of refraction of the DMDCS overlayer and air (1.0).

15

20

5

10

Therefore, while the invention has been described in connection with what is presently considered to be the most practical and preferred embodiment, it is to be understood that the invention is not to be limited to the disclosed embodiment, but on the contrary, is intended to cover various modifications and equivalent arrangements included within the spirit and scope of the appended claims.

- 9 -

PCT/US03/00578

WHAT IS CLAIMED IS:

WO 03/082760

- 1. A substrate having an anti-reflective hydrophobic surface coating comprised of an anchor layer of a vapor-deposited chlorosilyl group containing compound, and an overlayer of a vapor-deposited alkylsilane, wherein the coating exhibits light reflectance of less than about 1.5% at a wavelength of about 525 nm (+/- 50 nm).
- 2. The substrate of claim 1, wherein said anchor layer includes a layer of vapor-deposited silicon tetrachloride, and wherein said overlayer includes a layer of vapor-deposited dimethyldichlorosilane (DMDCS).
- 3. The substrate of claim 2, wherein the anchor includes a layer of SiO₄ having an index of refraction of between about 1.42 to about 1.46.
- 4. The substrate of claim 1 or 2, wherein the overlayer has an index of refraction of between about 1.28 to about 1.38.
- 5. The substrate of claim 3, wherein the thickness of the silicon oxide layer is between about 1600Å to about 2000Å.
- 6. The substrate of claim 5, wherein the thickness of the DMDCS overlayer is between about 800Å to about 1200Å.
- 7. A glass substrate having an anti-reflective hydrophobic surface coating comprised of a silicon oxide anchor layer having a layer thickness of about between about 1600Å to about 2000Å, and a vapor-deposited dimethyldichlorosilane (DMDCS) layer having a layer thickness of

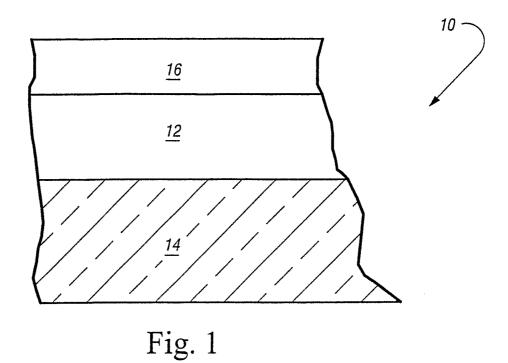
between about 800Å to about 1200Å, wherein the coating exhibits light reflectance of less than about 1.5% at a wavelength of between about 475 nm to about 575 nm.

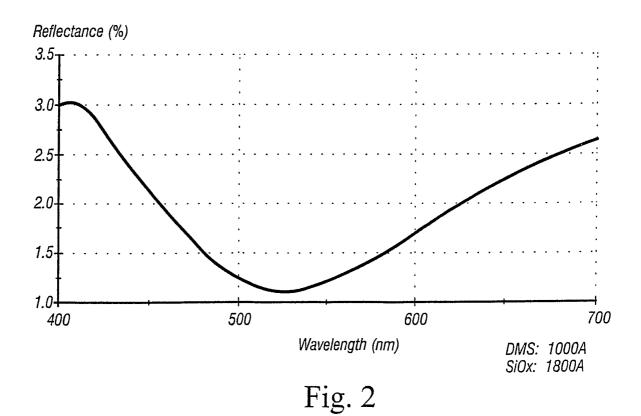
- 8. The substrate of claim 7, wherein said silicon oxide layer has a thickness of about 1800Å.
- 9. The substrate of claim 8, wherein said vapor-deposited DMDCS layer has a thickness of about 1000Å.
- 10. The substrate of claim 9, wherein said light reflectance is less than about 1.5% at a wavelength of about 525 nm.
- 11. A process for forming an anti-reflective hydrophobic coating on substrates comprising contacting a surface of the substrate to be coated with vapors of a chlorosilyl group containing compound, and an alkylsilane in a humid room temperature atmosphere, for a time sufficient such that the coating exhibits a light reflectance of less than about 1.5% at a wavelength of between about 475 nm to about 575 nm.
- 12. The process of claim 11, wherein the vapor of the chlorosilyl group containing compound and the vapor of the alkylsilane are brought sequentially into contact with the substrate
- 13. The process of claim 12, wherein the chlorosilyl group containing compound is silicon tetrachloride (SiCl₄), and wherein the chloroalkylsilane is dimethyldichlorosilane (DMDCS).

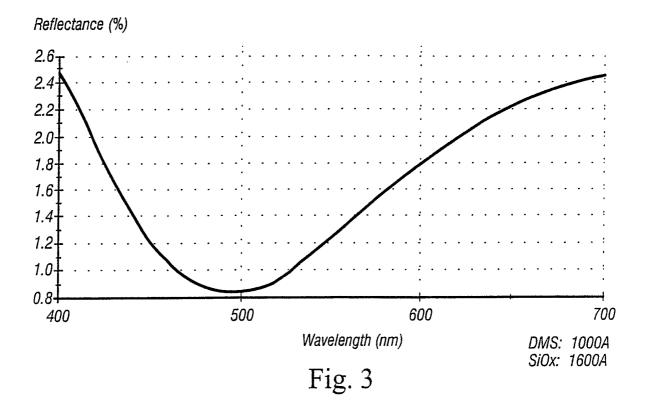
- 14. The process of claim 13, wherein said vapor-deposited SiCl₄ layer has thickness of between about 1600Å to about 2000Å.
- 15. The process of claim 14, wherein said vapor-deposited DMDCS layer has a thickness of between about 800Å to about 1200Å.
- 16. The process of any one of claims 15, which further comprises applying a capping layer onto the substrate.
- 17. The process of any one of any one of claims 11-14, which further comprises applying a capping layer onto the substrate.
- 18. A process for forming an antireflective hydrophobic coating on a glass substrate comprising the steps of:
 - (a) contacting a surface of the substrate to be coated with a vapor containing silicon tetrachloride for a time sufficient to form a silicon oxide layer on the glass surface; and then
 - (b) contacting the silicon oxide layer with a vapor containing dimethyldichlorosilane (DMDCS) for a time sufficient to form thereon a cross-linked overlayer of polydimethylsiloxane (PDMSO), and wherein
 - (c) steps (a) and (b) are practiced to obtain respective layer thicknesses so that the coating exhibits light reflectance of less than about 1.5% at a wavelength of between about 475 nm to about 575 nm.
- 19. The process of claim 18, wherein said vapor-deposited SiCl₄ layer has thickness of between about 1600Å to about 2000Å.

- 12 -

- 20. The process of claim 19, wherein said vapor-deposited DMDCS layer has a thickness of between about 800Å to about 1200Å.
- 21. A coated glass substrate made by the process of any one of claims 11-16 and 18-20.







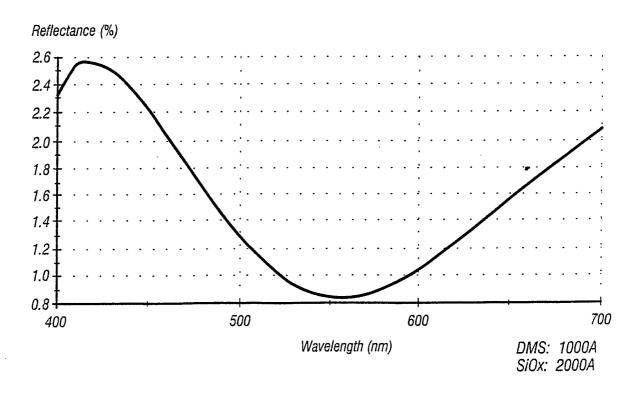
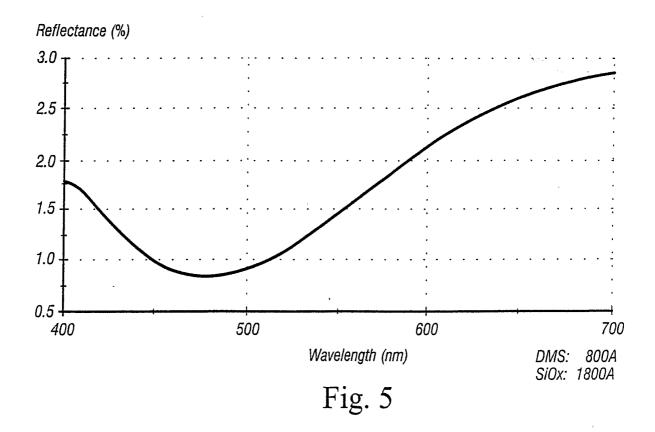
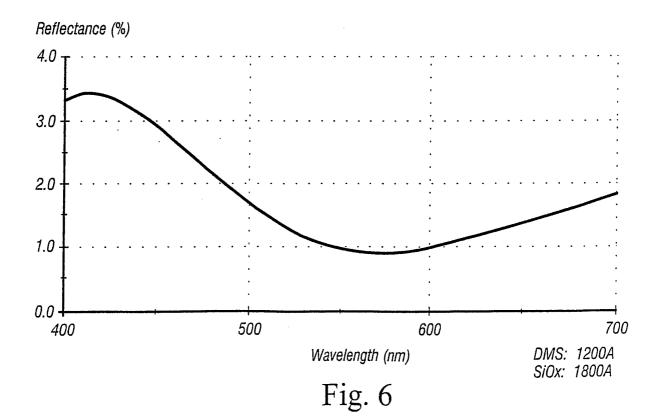


Fig. 4





IMPERNATIONAL SEARCH REPORT

Intentional Application No
PCT/US 03/00578

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C03C17/34 C03C C03C17/42 B32B9/00 According to International Patent Classification (IPC) or to both national classification and IPC Minimum documentation searched (classification system followed by classification symbols) IPC 7 C03C B32B Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) INSPEC, EPO-Internal C. DOCUMENTS CONSIDERED TO BE RELEVANT Relevant to claim No. Citation of document, with indication, where appropriate, of the relevant passages Υ DATABASE INSPEC 'Online! 1 - 21INSTITUTE OF ELECTRICAL ENGINEERS, STEVENAGE, GB; REULOS R: "A new energy tensor" Database accession no. 386216 XP002250670 abstract & COMPTES RENDUS DES SEANCES DE LA SOCIETE DE PHYSIQUE ET D'HISTORIE NATURELLE DE GENEVE, JAN.-APRIL 1971, SWITZERLAND, vol. 6, no. 1, pages 47-60, Further documents are listed in the continuation of box C. Patent family members are listed in annex. Χ Special categories of cited documents: "T" later document published after the international filing date or priority date and not in conflict with the application but "A" document defining the general state of the art which is not cited to understand the principle or theory underlying the considered to be of particular relevance invention "E" earlier document but published on or after the international "X" document of particular relevance; the claimed invention filing date cannot be considered novel or cannot be considered to "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such docu-"O" document referring to an oral disclosure, use, exhibition or ments, such combination being obvious to a person skilled other means document published prior to the international filing date but later than the priority date claimed in the art. "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 25/08/2003 19 August 2003 Authorized officer Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Maurer, R Fax: (+31-70) 340-3016

IMPERNATIONAL SEARCH REPORT

| | | FC1/U3 U3/UU3/6 | | |
|-------------|---|-----------------------|--|--|
| C.(Continua | ation) DOCUMENTS CONSIDERED TO BE RELEVANT | | | |
| Category ° | Citation of document, with indication, where appropriate, of the relevant passages | Relevant to claim No. | | |
| Y | DATABASE INSPEC 'Online! INSTITUTE OF ELECTRICAL ENGINEERS, STEVENAGE, GB; DEWAN E M: "The magnetic field as a mechanism to preserve relativistic momentum" Database accession no. 386217 XP002250671 abstract & AMERICAN JOURNAL OF PHYSICS, MAY 1972, USA, vol. 40, no. 5, pages 755-759, ISSN: 0002-9505 | 1-21 | | |
| Υ | US 4 274 856 A (BEESTRICE WILLIAM R ET AL) 23 June 1981 (1981-06-23) column 1, line 24-39; claims 1-6 | 1-21 | | |
| Y | US 4 263 350 A (VALIMONT JAMES L) 21 April 1981 (1981-04-21) column 1, line 34 -column 2, line 21; claims 1-7 | 1-21 | | |
| Y | EP 0 498 339 A (MATSUSHITA ELECTRIC IND CO LTD) 12 August 1992 (1992-08-12) page 2, line 36 -page 3, line 34; claims 1-12 | 1-21 | | |
| Υ | US 5 372 851 A (OGAWA KAZUFUMI ET AL) 13 December 1994 (1994-12-13) column 5, line 45 -column 6, line 52; claims 1-19; examples 1-13 | 1–21 | | |
| Y | EP 0 476 510 A (NISSAN MOTOR) 25 March 1992 (1992-03-25) page 2, line 1 -page 3, line 3; claims 1-5; examples 1,2 | 1-21 | | |
| | | | | |
| | | | | |

IMMERNATIONAL SEARCH REPORT

Information on patent family members

In Stional Application No
PCT/US 03/00578

| Patent document cited in search report | | Publication date | | Patent family member(s) | Publication date |
|--|---|---------------------|--------|-------------------------|------------------|
| US 4274856 | Α | 23-06-1981 | NONE | | |
| US 4263350 | Α | 21-04-1981 | NONE | | |
| EP 0498339 | Α | 12-08-1992 | JP | 2500151 B2 | 29-05-1996 |
| | | | JP | 4255344 A | 10-09-1992 |
| | | | JP | 2500816 B2 | 29-05-1996 |
| | | | JP | 4328136 A | 17-11-1992 |
| | | | JP | 3008507 B2 | 14-02-2000 |
| | | | JP | 4250158 A | 07-09-1992 |
| | | | CA | 2060026 A1 | 29-07-1992 |
| | | | CA | 2060294 A1 | 07-08-1992 |
| | | | DE | 69205849 D1 | 14-12-1995 |
| | | | DE | 69205849 T2 | 04-04-1996 |
| | | | DE | 69231787 D1 | 23-05-2001 |
| | | | DE | 69231787 T2 | 02-08-2001 |
| | | | EP | 0497204 A2 | 05-08-1992 |
| | | | EP | 0498339 A1 | 12-08-1992 |
| | | | KR | 9503700 B1 | 17-04-1995 |
| | | | KR | 9510641 B1 | 21-09-1995 |
| | | | US | 5443511 A | 22-08-1995 |
| | | | US | 5578340 A | 26-11-1996 |
| | | | US | 5500250 A | 19-03-1996 |
| | | | US | 5372888 A | 13-12-1994 |
| US 5372851 | Α | 13-12-1994 | JP | 2603017 B2 | 23-04-1997 |
| | | | JP | 5161844 A | 29-06-1993 |
| | | | JP | 5168913 A | 02-07-1993 |
| | | | DE | 69224888 D1 | 30-04-1998 |
| | | | DE | 69224888 T2 | 23-07-1998 |
| | | | EP | 0547550 A1 | 23-06-1993 |
| | | | KR | 9701519 B1 | 11-02-1997 |
| EP 0476510 | Α | 25-03-1992 | JP | 4124047 A | 24-04-1992 |
| | | | EP | 0476510 A1 | 25-03-1992 |