



(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:
09.09.2015 Bulletin 2015/37

(51) Int Cl.:
F23K 5/10 (2006.01)

(21) Application number: **14157773.4**

(22) Date of filing: **05.03.2014**

(84) Designated Contracting States:
AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO PL PT RO RS SE SI SK SM TR
Designated Extension States:
BA ME

(72) Inventors:
• **Stallmann, Olaf**
55270 Essenheim (DE)
• **Gerber, Siegfried Werner**
65817 Eppstein (DE)

(71) Applicant: **ALSTOM Technology Ltd**
5400 Baden (CH)

(74) Representative: **Alstom Technology Ltd**
CHTI Intellectual Property
Brown Boveri Strasse 7
5400 Baden (CH)

(54) **Method and system for supplying a fuel into a combustion chamber**

(57) The method for supplying a fuel into a combustion chamber (1) comprises mixing the fuel with supercritical pressure CO₂ forming a mixture and injecting the

mixture into a combustion chamber (1) having a pressure lower than the CO₂ critical pressure.

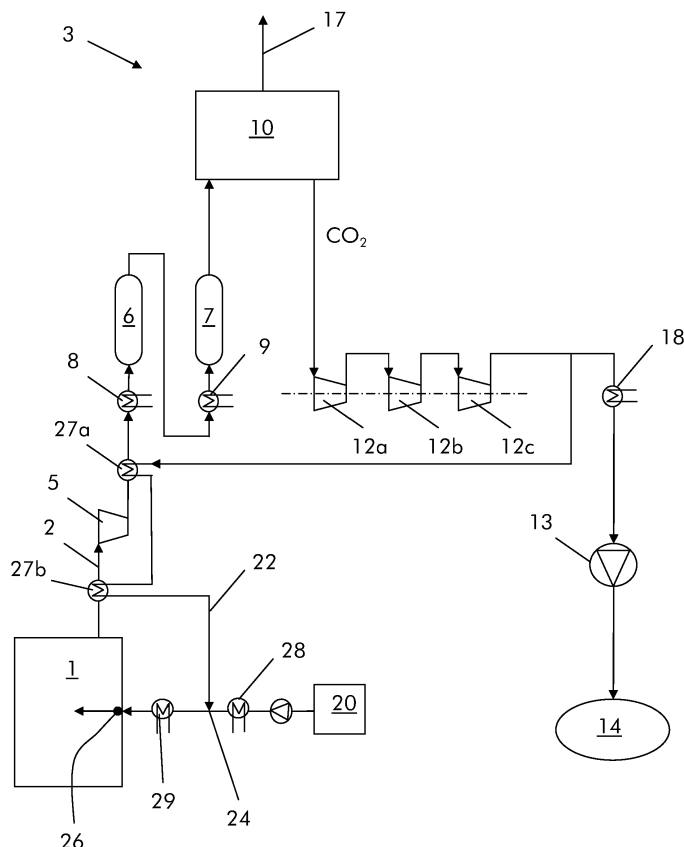


Fig. 1

Description

TECHNICAL FIELD

[0001] The present invention relates to a method and system for supplying a fuel into a combustion chamber. The combustion chamber is for example the combustion chamber of a boiler or furnace for industrial applications. The combustion chamber can be an oxy fuel combustion chamber, i.e. a combustion chamber that is supplied with a fuel and substantially pure oxygen, this is anyhow not needed and the combustion chamber can be supplied with a fuel and air. In addition, the combustion chamber can also be supplied with recirculated flue gas, but this is also not mandatory.

[0002] The fuel is a high viscosity fluid, i.e. a fluid that alone is not able to pass through the ducts and injectors of the combustion chamber, but needs appropriate helps for this. For example the fuel is heavy residue.

BACKGROUND

[0003] Crude oil undergoes a number of treatments in order to separate different products from it, such as for example, liquefied petroleum gas, gasoline, diesel oil, kerosene, etc.; the remaining of these treatments is the so called heavy residue, that is a high viscosity product that at atmospheric conditions becomes solid.

[0004] Traditionally, heavy residue is used in boilers as a fuel, for example in power plants.

[0005] In order to get a product that can be handled, such as for example injected and atomized through injectors into the combustion chamber of a boiler, the heavy residue is heated and mixed with kerosene and/or water in order to obtain a low viscosity fluid.

[0006] This known solution has some disadvantages.

[0007] Heavy residue must be heated up to a temperature very close to its coking temperature. At the coking temperature, the heavy residue forms solid coke deposits that accumulate in piping and injectors, blocking them. The coking temperature for the heavy residue can begin already at temperatures as low as 200-250°C depending on the originating crude feedstock. When the heavy residue is heated up close to the coking temperature, there is the risk that some fractions of the heavy residue start to coke due to an uneven mixture of the heavy residue.

[0008] In addition, the known solution requires the use of highly expensive fuel, such as kerosene, or high expensive fluid, such as water (in some countries water can be more expensive than oil).

SUMMARY

[0009] An aspect of the invention includes providing a method and system that avoid heating of the heavy residue or require a heating up to a temperature well far apart from the coking temperature, such that coking is prevented.

[0010] Another aspect of the invention includes providing a method and system that avoid or at least limit the use of expensive fuel (such as kerosene) or fluid (such as water) together with the heavy residue.

5 **[0011]** These and further aspects are attained by providing a method and a system in accordance with the accompanying claims.

10 **[0012]** The described solution addresses heavy residue being fuels that have a viscosity of more than 150 cSt at a temperature of 100°C.

BRIEF DESCRIPTION OF THE DRAWINGS

15 **[0013]** Further characteristics and advantages will be more apparent from the description of a preferred but non-exclusive embodiment of the method and system, illustrated by way of non-limiting example in the accompanying drawings, in which:

20 Figure 1 shows an example of the system in a first embodiment;

Figure 2 shows an example of the system in a second embodiment;

25 Figure 3 shows an example of the system in a third embodiment;

Figure 4 shows an example of the system in a fourth embodiment;

Figure 5 shows an example of injection, and

30 Figure 6 shows a nozzle.

DETAILED DESCRIPTION OF EXEMPLARY EMBODIMENTS

35 **[0014]** In the following the system for supplying a fuel such as heavy residue (i.e. a fuel having a viscosity equal to or greater than 150 cSt at a temperature of 100°C) into a combustion chamber 1 is described first.

[0015] The combustion chamber is a part of a boiler for a power plant or an industrial furnace.

40 **[0016]** The combustion chamber 1 combusts a fuel and generates flue gas 2 that is supplied to a flue gas treatment system 3.

45 **[0017]** The flue gas treatment system 3 can be of any kind and, for example, it can include a compressor 5, a mercury removal unit 6 and a drier 7. Heat exchangers 8, 9 are preferably provided upstream of the mercury removal unit 6 and drier 7.

50 **[0018]** From the drier 7 the flue gas is supplied into a separation unit 10, where the CO₂ is separated from other gas. The CO₂ is thus supplied to a compressor 12a-c (for example a multi stage, intercooled compressor) where it is compressed up to or above the supercritical pressure, and then to a pump 13 (but this pump is not needed and it is used according to the design) to be further compressed for storage at 14.

55 **[0019]** The separation unit 10 can be of any type, for example figure 2 shows a scheme in which the separation unit includes two stages 15a, 15b of condensation by

cooling. In this example CO₂ that condenses 16a, 16b at each condensation stage 15a, 15b is also used as cooling medium. Likewise, gas 17 being the flue gas deprived of CO₂ is expanded and used as cooling medium at the condensation stages 15a, 15b; the gas 17 is then vented.

[0020] Between the compressor 12a-c and the pump 13 or storage 14 a heat exchanger 18 can be provided (it is not mandatory and is provided according to the specific design). The heat exchanger 18 is used for cooling the CO₂ compressed at the compressor 12a-c.

[0021] The system 1 comprises a fuel supply 20, a supercritical pressure CO₂ supply 22, a mixing system 24 for forming a mixture of fuel and supercritical pressure CO₂, injectors 26 connected to the mixing system 24 for injecting the mixture into the combustion chamber 1.

[0022] The mixing system can be defined by the converging pipes through which supercritical pressure CO₂ and fuel pass through, or by a dedicated mixer or tank.

[0023] Advantageously the combustion chamber has a pressure lower than the CO₂ critical pressure.

[0024] In addition, the system preferably has (but this is not needed and depends on the particular design and conditions) at least a heat exchanger for heating the mixture and/or the fuel and/or the supercritical pressure CO₂.

[0025] The heat exchangers can include:

- the heat exchangers 27a, 27b that allow heating of the supercritical pressure CO₂ before it is mixed with the fuel; in different embodiments both the heat exchangers 27a and 27b can be provided, or only one of them can be provided, or none of them can be provided;
- the heat exchanger 28 that allows heating of the fuel before it is mixed with the supercritical pressure CO₂; in different examples it can be provided or not;
- the heat exchanger 29 that allows heating of the mixture of fuel and supercritical pressure CO₂; in different examples it can be provided or not.

[0026] Naturally, all or none of the heat exchangers 27a, 27b, 28, 29 can be provided, in addition any combination of heat exchangers 27a, 27b, 28, 29 is possible.

[0027] The supercritical pressure CO₂ can be supplied from the line that forwards the CO₂ to the storage 14, from a position upstream the pump 13 or downstream the pump 13. For example the supercritical pressure CO₂ supply 22 can depart from a position between the compressor 12a-c and the heat exchanger 18 (figures 1 and 2) and/or from a position between heat exchanger 18 and the pump 13 and/or from a position between the pump 13 and the storage 14 (figure 14).

[0028] The operation of the system is apparent from that described and illustrated and is substantially the following.

[0029] At the combustion chamber 1 flue gas 2 is generated from the combustion of the fuel. The flue gas is compressed at the compressor 5 and then the flue gas

undergoes cooling at the heat exchanger 8, mercury removal at the mercury removal unit 6, cooling at the heat exchanger 9, water removal at the drier 7. Thus the flue gas is supplied into the separation unit 10 where CO₂ is separated from the other gas 17 (such as nitrogen, argon, etc.). The gas 17 is vented into the atmosphere and the CO₂ (that is still gas) is supplied to the compressor 12a-c (usually a multistage, intercooled compressor that is used to prepare the CO₂ for storage).

[0030] At the compressor 12a-c the CO₂ is compressed up to or above the critical pressure (critical pressure 72.9 bar or 7.39 MPa); then at the heat exchanger 18 the CO₂ compressed at or above the critical pressure is cooled.

[0031] Depending on the availability of cooling media and specific design, the CO₂ can be either:

- compressed at the compressor 12a-c at or above the critical pressure and then condensed at the heat exchanger 18; the condensed CO₂ at a pressure above the critical pressure is then pumped via the pump 13,
- compressed to the final pressure at the compressor 12a-c; the final pressure is usually about 100 bar (in this case the pump 13 is not needed), cooling after compression at the heat exchanger 18 is possible at a temperature above or below the critical pressure,
- compressed at the compressor 12a-c at or above the critical pressure and cooled to a temperature below the critical temperature at the heat exchanger 18; the compressed CO₂ is then pumped via the pump 13,
- compressed at the compressor 12a-c at or above the critical pressure, no cooling is provided in this case or cooling at the heat exchanger 18 to a temperature above the critical temperature, such that the CO₂ is in supercritical state.

[0032] Naturally also other combinations of compression and cooling are possible.

[0033] The supercritical CO₂, i.e. CO₂ at or above the critical pressure and at or above the critical temperature (304.25 K) has a high density and can thus be pumped.

[0034] Likewise, the CO₂ at or above the critical pressure and below the critical temperature (304.25 K) has a high density and can be pumped. This region is sometimes called the dense phase region.

[0035] At the pump 13 the supercritical CO₂ or CO₂ in the dense phase region is further compressed for storage. Downstream the pump 13 the pressure of the supercritical CO₂ is 100 bar or more.

[0036] Supercritical pressure CO₂ (the temperature of the supercritical pressure CO₂ can be the supercritical temperature, or it can be above or below the supercritical temperature) is thus preferably supplied from a position downstream the compressor 12a-c and upstream or downstream the pump 13 (when provided) to the mixing system 24. Likewise, heavy residue is supplied from the

fuel supply 20 to the mixing system 24. At the mixing system 24 the heavy residue and supercritical pressure CO₂ form a mixture.

[0037] The supercritical pressure CO₂ is a very good solvent for heavy residue. For this reason, the mixture of heavy residue and supercritical pressure CO₂ can contain a large amount of CO₂, such that the viscosity of the mixture allows injection in the combustion chamber 1.

[0038] In addition, in order to adjust the viscosity, the mixture (the temperature of the mixture after mixing is typically below 60°C) can be heated at the heat exchanger 29. Alternatively or in addition to the heating at the heat exchanger 29, also the supercritical pressure CO₂ and/or the heavy residue can be heated at the heat exchangers 27a, 27b, 28, in such a way that the temperature of the mixture containing the heavy residue and the supercritical pressure CO₂ falls in the preferred temperature range.

[0039] The preferred range for the temperature of the mixture is between 50-160°C, with a more preferred range between 60-90°C; the temperature is thus well below the coking temperature for the heavy residue.

[0040] From the mixing system 24 the mixture is supplied to the injectors 26 and is injected into the combustion chamber 1.

[0041] Advantageously, after the injection in the combustion chamber 1, the pressure of the mixture (containing supercritical pressure CO₂ and heavy residue) suddenly drops. The pressure drop causes a sudden expansion of the CO₂ (up to 10 times or more). Since the CO₂ is mixed with the heavy residue, this expansion helps atomization of heavy residue and its dispersion through the combustion chamber 1. Atomization and dispersion through the combustion chamber 1 help a complete and clean combustion.

[0042] Figure 5 shows the injectors 26 with nozzles 30, jets of mixtures 31 and the expanding CO₂ that promotes heavy residue dispersion through the combustion chamber 1 and atomization.

[0043] The fuel such as heavy residue atomized and dispersed through the combustion chamber can thus combust with air or oxygen.

[0044] Figure 6 shows an example of the nozzle 30; in this figure the arrow indicates the flow through the nozzle 30. In addition figure 6 shows the convergent-divergent design of the nozzle 30 that helps preventing excessive erosion.

[0045] The present invention also refers to a method for supplying a fuel into a combustion chamber.

[0046] The method comprises mixing the fuel with supercritical pressure CO₂ forming a mixture and injecting the mixture into a combustion chamber having a pressure lower than the CO₂ critical pressure.

[0047] The mixture contains between 10-70% by weight of supercritical pressure CO₂ and preferably between 15-25% by weight of supercritical pressure CO₂.

[0048] The mixture and/or the fuel and/or the supercritical pressure CO₂ can be heated in order to obtain a

heated mixture, whose viscosity is preferably between 15-30 centipoise. This viscosity allows easy injection of the heavy residue through the injector 26.

[0049] The temperature of the heated mixture is between 50-160°C and preferably between 60-90°C.

[0050] The mixture and/or the fuel and/or the supercritical pressure CO₂ are heated by cooling the flue gas.

[0051] Naturally the features described may be independently provided from one another.

[0052] In practice the materials used and the dimensions can be chosen at will according to requirements and to the state of the art.

REFERENCE NUMBERS

[0053]

1	combustion chamber
2	flue gas
3	flue gas treatment system
5	compressor
6	mercury removal unit
7	dryer
8	heat exchanger
9	heat exchanger
10	separation unit
12a-c	compressor
13	pump
14	storage
15a, b	stage of condensation
16a, b	condensed CO ₂
17	gas
18	heat exchanger
20	fuel supply
22	supercritical pressure CO ₂ supply
24	mixing system
26	injector
27a, b	heat exchanger

- 28 heat exchanger
- 29 heat exchanger
- 30 nozzle
- 31 jet of mixture
- 32 CO₂

Claims

1. A method for supplying a fuel into a combustion chamber (1), **characterised by** mixing the fuel with supercritical pressure CO₂ forming a mixture, injecting the mixture into a combustion chamber (1) having a pressure lower than the CO₂ critical pressure. 20
2. The method of claim 1, **characterised in that** the supercritical pressure CO₂ contained in the mixture is in the supercritical state. 25
3. The method of claim 1, **characterised in that** the mixture contains between 10-70% by weight of supercritical pressure CO₂. 30
4. The method of claim 1, **characterised in that** the mixture contains between 15-25% by weight of supercritical pressure CO₂. 35
5. The method of claim 1, **characterised by** heating the mixture and/or the fuel and/or the supercritical pressure CO₂ in order to obtain a heated mixture. 40
6. The method of claim 5, **characterised in that** the mixture and/or the fuel and/or the supercritical pressure CO₂ are heated in order to obtain a heated mixture with a viscosity between 15-30 centipoise. 45
7. The method of claim 5, **characterised in that** the temperature of the heated mixture is between 50-160°C. 50
8. The method of claim 5, **characterised in that** the temperature of the heated mixture is between 60-90°C. 55
9. The method of claim 5, **characterised in that** flue gas (2) is discharged from the combustion chamber (1), wherein the mixture and/or the fuel and/or the supercritical pressure CO₂ are heated by cooling the flue gas. 55
10. The method of claim 1, **characterised in that** the fuel has a viscosity equal to or greater than 150 cSt

at a temperature of 100°C.

11. A system for supplying a fuel into a combustion chamber (1), **characterised by** comprising a fuel supply (20), a supercritical pressure CO₂ supply (22), a mixing system (24) for forming a mixture of fuel and supercritical pressure CO₂, at least an injector (26) connected to the mixing system (24), the at least an injector (26) for injecting the mixture into the combustion chamber (1), the combustion chamber (1) having a pressure lower than the CO₂ critical pressure. 5
12. The system of claim 11, **characterised by** further comprising at least a heat exchanger (27a, 27b, 28, 29) for heating the mixture and/or the fuel and/or the supercritical pressure CO₂. 10

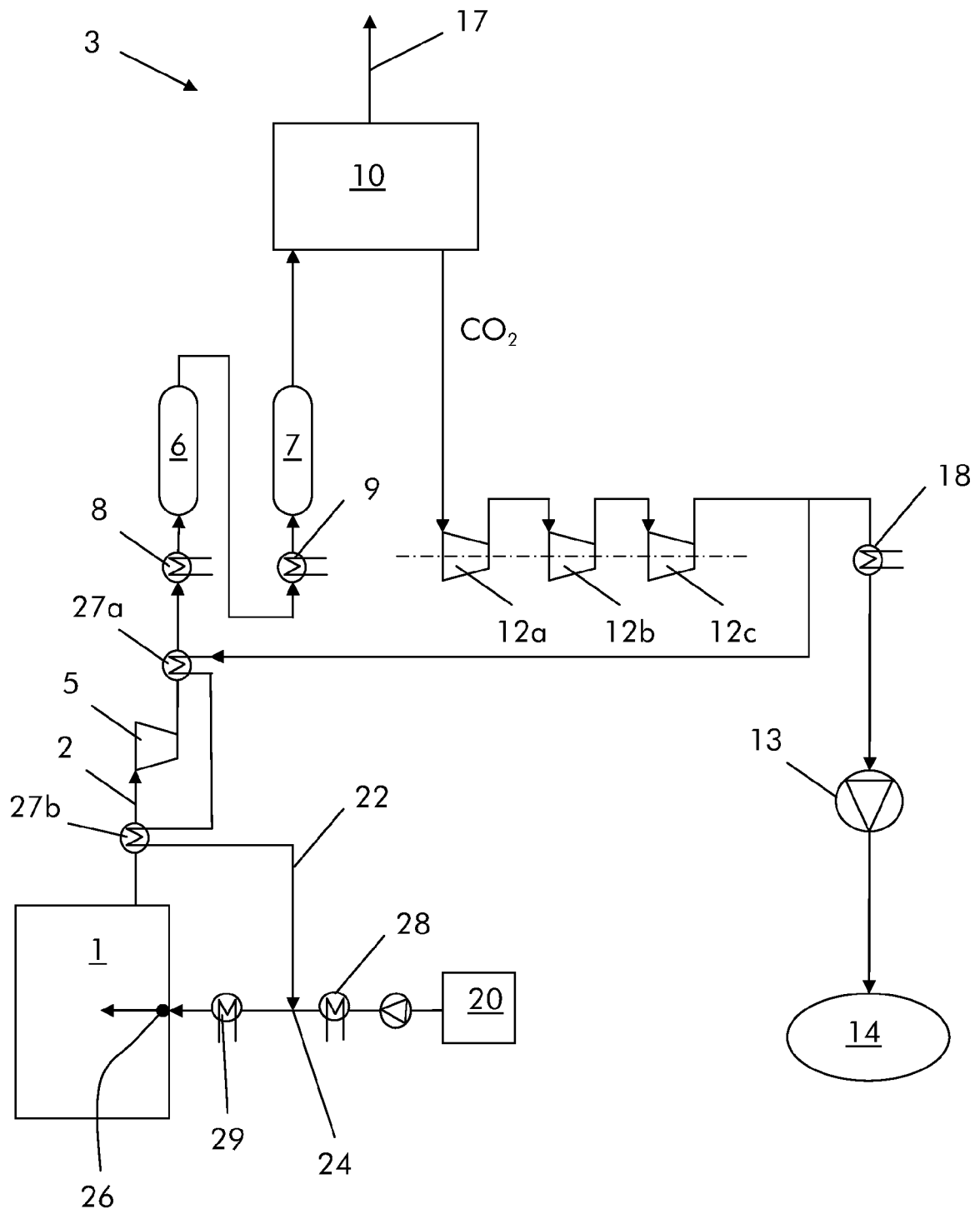


Fig. 1

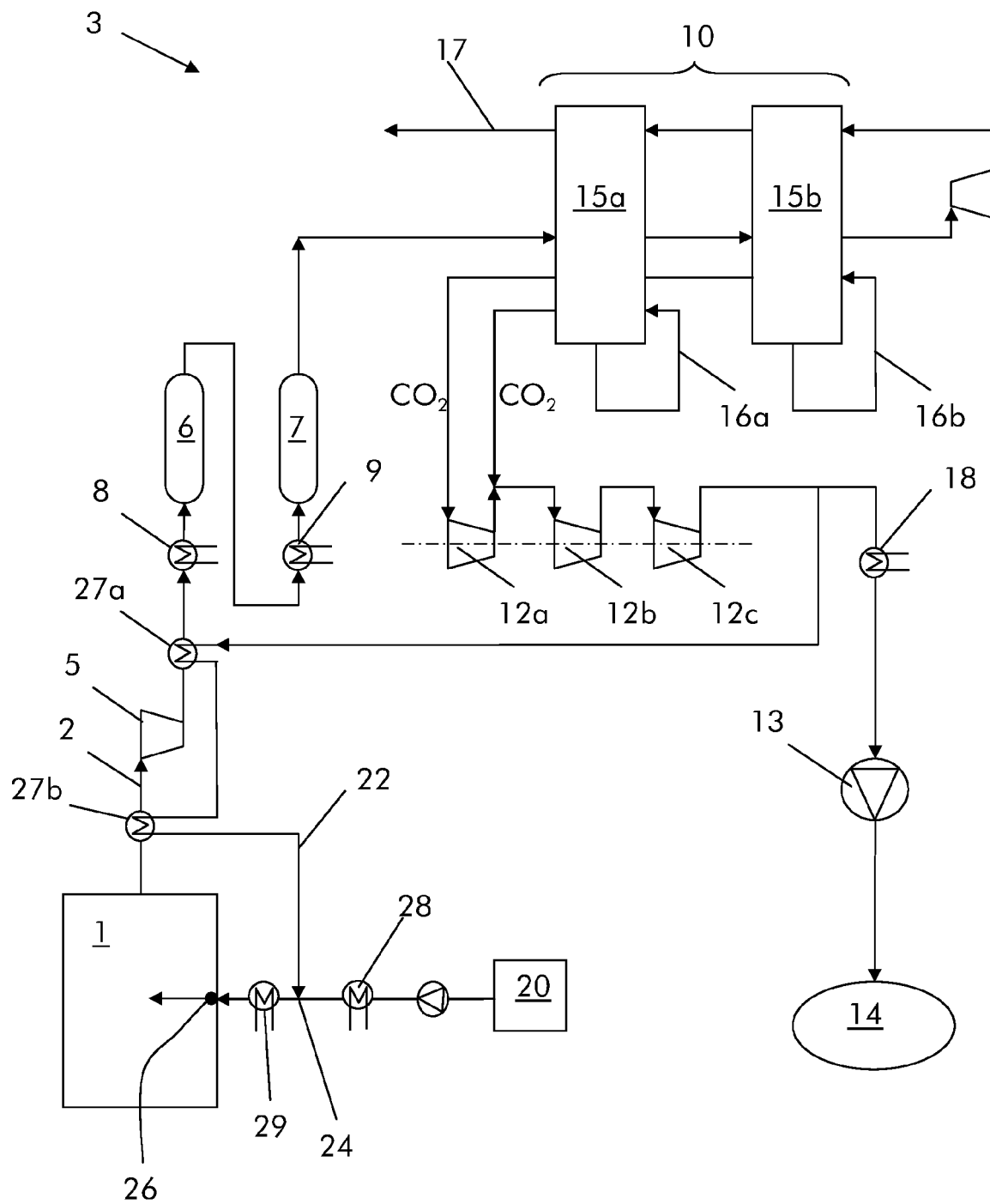


Fig. 2

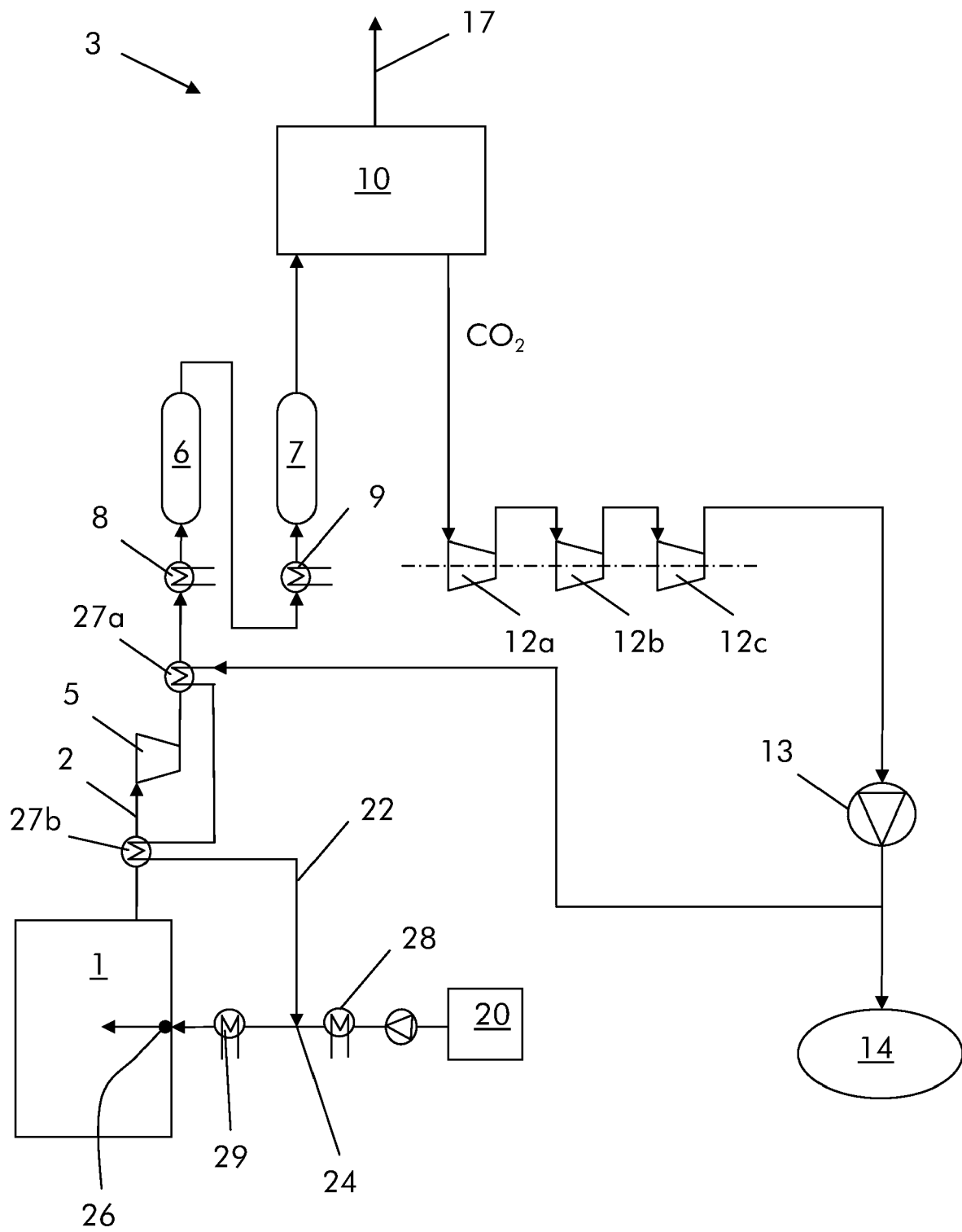


Fig. 3

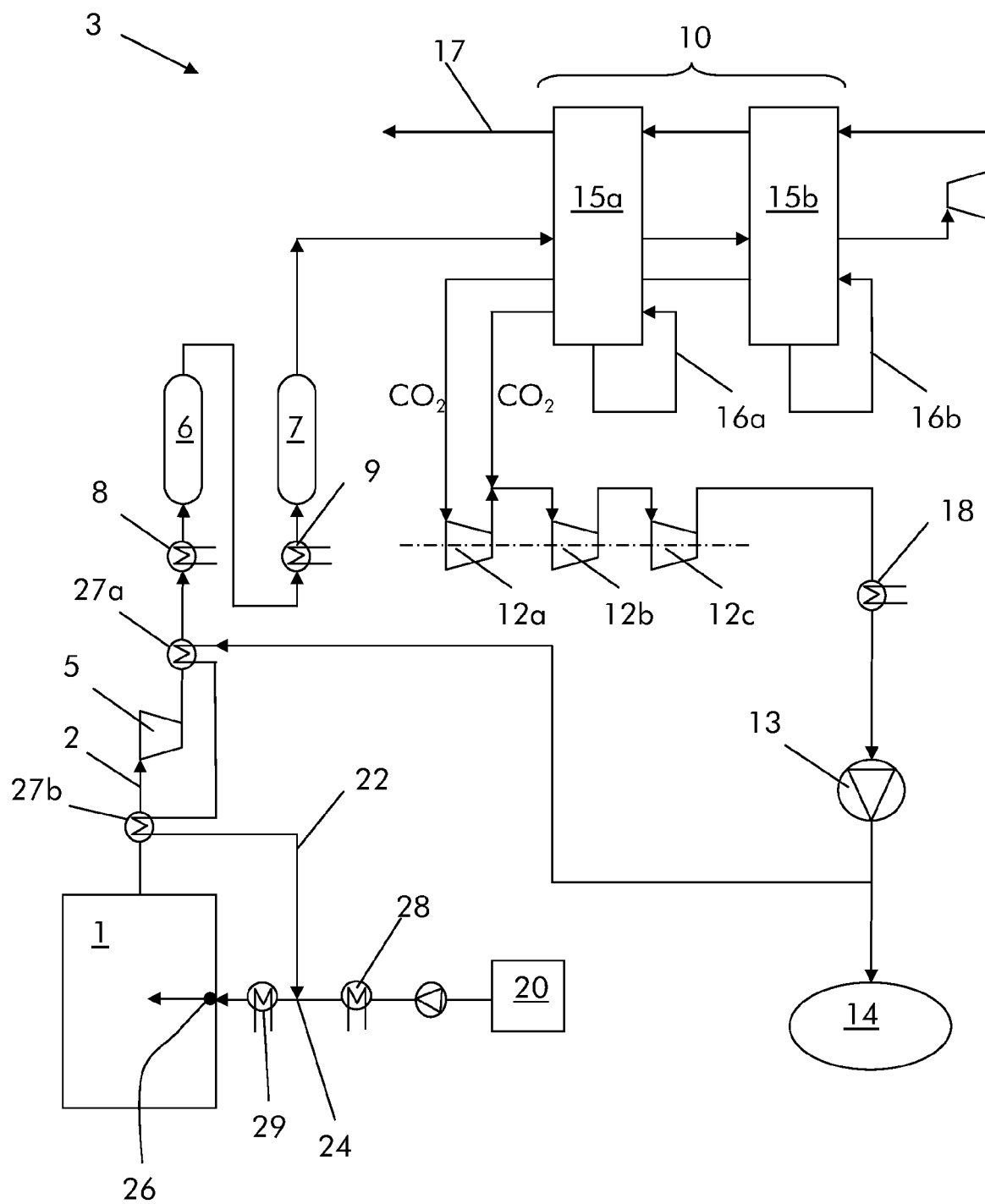
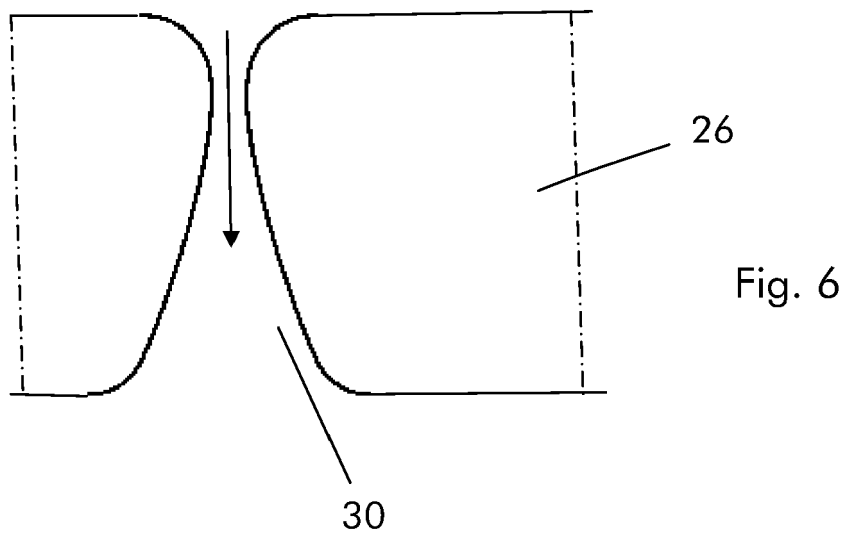
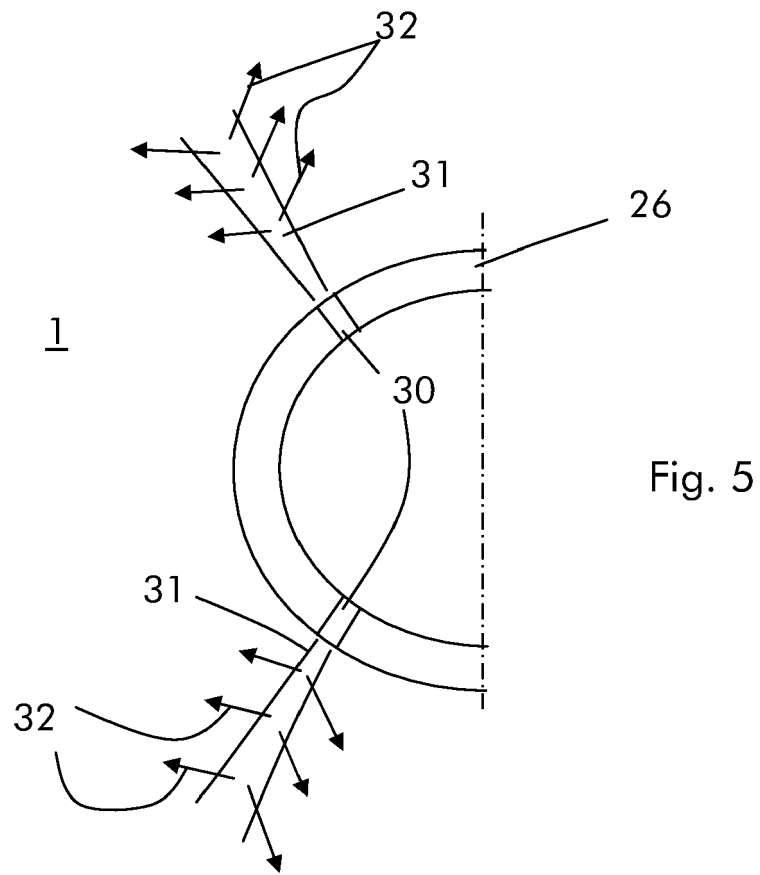


Fig. 4





EUROPEAN SEARCH REPORT

Application Number
EP 14 15 7773

5

10

15

20

25

30

35

40

45

50

55

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)
X	EP 0 506 069 A1 (UNION CARBIDE CHEM PLASTIC [US]) 30 September 1992 (1992-09-30) * page 1, line 3 - line 8 * * page 2, line 14 - line 27 * * page 3, line 39 - line 54 * * page 4, line 5 - line 18 * * page 4, line 28 - line 40 * * page 5, line 11 * * page 8, line 34 - line 42 * * page 10, line 30 - line 41 * * page 15, line 17 - line 20 * * page 15, line 57 - line 58 * * page 16, line 38 - line 42 * * page 17, line 24 - line 31 * -----	1-12	INV. F23K5/10
Y	WO 2012/159189 A1 (CANADA NATURAL RESOURCES [CA]; ZANGANEH KOUROSH ETEMADI [CA]; PEARSON) 29 November 2012 (2012-11-29) * page 3, line 7 - page 4, line 15 * * page 4, line 26 - line 28 * * page 6, line 14 - line 16 * * page 7, line 5 - line 10 * * page 10, line 27 - page 11, line 5 * -----	1-12	TECHNICAL FIELDS SEARCHED (IPC) F23K F23G
Y	EP 1 736 652 A2 (UNITED TECHNOLOGIES CORP [US]) 27 December 2006 (2006-12-27) * paragraphs [0003], [0019]; figure 2 * -----	1-12	
A	EP 0 807 458 A2 (KARLSRUHE FORSCHZENT [DE]) 19 November 1997 (1997-11-19) * figure 1 * * page 2, line 12 - line 16 * * page 3, line 1 - line 6 * ----- -/--	1,11	
The present search report has been drawn up for all claims			
Place of search The Hague		Date of completion of the search 14 August 2014	Examiner Mougey, Maurice
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

EPO FORM 1503 03.82 (P04C01)



EUROPEAN SEARCH REPORT

Application Number
EP 14 15 7773

5

10

15

20

25

30

35

40

45

50

55

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)
A	DE 10 2010 026792 A1 (MESSER GROUP GMBH [DE]) 12 January 2012 (2012-01-12) * paragraphs [0007], [0009], [0012], [0016]; figure 1 *	1,11	
A	WO 2007/063388 A2 (AIR LIQUIDE [FR]; VARAGANI RAJANI K [US]; PRANDA PAVOL [US]) 7 June 2007 (2007-06-07) * page 7, line 22 - page 8, line 16 * * page 9, line 16 - line 30 *	1,11	
			TECHNICAL FIELDS SEARCHED (IPC)
The present search report has been drawn up for all claims			
Place of search The Hague		Date of completion of the search 14 August 2014	Examiner Mougey, Maurice
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

EPO FORM 1503 03.82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 14 15 7773

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

14-08-2014

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0506069 A1	30-09-1992	AT 130864 T	15-12-1995
		CA 2064297 A1	30-09-1992
		DE 69206296 D1	11-01-1996
		DE 69206296 T2	18-04-1996
		DK 0506069 T3	27-12-1995
		EP 0506069 A1	30-09-1992
		ES 2080365 T3	01-02-1996
		GR 3018254 T3	29-02-1996
		JP 2651974 B2	10-09-1997
		JP H05302701 A	16-11-1993
		US 5170727 A	15-12-1992

WO 2012159189 A1	29-11-2012	AU 2011369049 A1	12-12-2013
		CA 2836814 A1	29-11-2012
		CN 103547861 A	29-01-2014
		EP 2715233 A1	09-04-2014
		JP 2014515448 A	30-06-2014
		KR 20140015575 A	06-02-2014
		US 2014080073 A1	20-03-2014
		WO 2012159189 A1	29-11-2012

EP 1736652 A2	27-12-2006	CA 2545896 A1	22-12-2006
		CN 1899661 A	24-01-2007
		EP 1736652 A2	27-12-2006
		JP 2007003184 A	11-01-2007
		KR 20060134796 A	28-12-2006
		US 2007006591 A1	11-01-2007

EP 0807458 A2	19-11-1997	AT 210489 T	15-12-2001
		DE 19619559 A1	27-11-1997
		EP 0807458 A2	19-11-1997

DE 102010026792 A1	12-01-2012	NONE	

WO 2007063388 A2	07-06-2007	AT 481603 T	15-10-2010
		AU 2006321346 A1	07-06-2007
		CA 2632527 A1	07-06-2007
		EP 2005063 A2	24-12-2008
		EP 2184336 A1	12-05-2010
		JP 5199110 B2	15-05-2013
		JP 2009517520 A	30-04-2009
		KR 20080092347 A	15-10-2008
		US 2007144415 A1	28-06-2007
		WO 2007063388 A2	07-06-2007

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82