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Inventor: and

(72) Inventor; and

(71) Applicant: BURJANADZE, Sophia [GE/GE]; Digomi district 6/5, Apt. 25, 0159 Tbilisi (GE).

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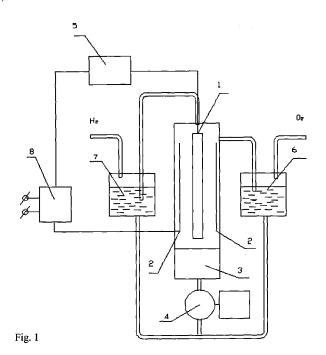
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(54) Title: DEVICE FOR HYDROGEN GENERATION BY CAVITATION ELECTROLISYS



(57) Abstract: The present invention relates to cavitation-electrolysis hydrogen generator. The cavitationelectrolysis hydrogen generator of the present invention comprises a filled with water cylidrical cavitator, locateded in it cathode and anode, conected by tube to cavitators bottom circulation water pump, conected to cathode and anode pulsed energy source, voltage modulator, hydrogen and oxigen tanks conected to pump and cavitators upper part, were enriched by hydrogen and oxigen water undergoes degassing and by conected to tank tube hydrogen uotput is obtained. According to the cavitation-electrolysis hydrogen generator of the present invention, there are advantages in that the efficiency of hydrogen generation can be improved by energy effective combination of cavitation and electrolysis and hydrogen production costs also is remarkably reduced due to device cheap materials construction.



DESCREPTION

Today hydrogen is obtained mainly from natural gas and despite the fact that this technology is energetically very competitive, development of this technology is not considered as a perspective direction. Firstly, natural gas is not renewable source of energy; secondly, this technology is characterized with greenhouse gases emission during the production of hydrogen. Therefore, production of hydrogen from natural gas cannot solve the problem of global warming. Same problems are presented during hydrogen production from coal.

Only completely clean technology of hydrogen production is electrolysis, but today its high energy consumption rate is a barrier for its commercialization. We propose solving exactly this problem in this invention with absolutely new cavitation-electrolysis method, which not only significantly decreases the energy consumption but also reduces cost of device.

The several types of electrolysis are currently known with their strong and weak points. Especial we would like to underline the two types of technologies: high temperature and high pressure electrolysis. Our completely new type cavitation-electrolysis method unites the strong sides of these technologies and doesn't include their weak points. For example, in the cavitation bubble very high temperatures and high pressures are located. Therefore we need not high pressure proof containers because high pressure is located in the cavitation bubble. As a result the cost of technology significantly decreases; the high pressure proof containers are costly technology. In the given technology for reaching high temperatures providing of additional heat is not required. As a result we have also significant increase in energy efficiency.

In widely-spread electrolyzers chemically very active catalysts are used, therefore electrodes are required to be covered by rare substances (for example platinum), which also makes this technology more expensive compared to our method which has no such requirements.

Really we have a new method of hydrogen production and a device, which by its main characteristics is cavitation physical-chemical reactor - electrolysis cell.

Most close to this invention is a cavitation hydrogen generator, which is represented by US patent (1), which used only hydraulic cavitation. In presented invention also effectively is used nonlinear electrolysis method, as a result efficiency of device increased compare to prototype. What about Meyer's nonlinear etecrolysis device, improved method of which is used in this invention, it without modification has production rate restriction. This problem is overcome in given invention.

Given method differs from existed electrolysis technologies mainly with using of cavitaion. Cavitation phenomenon is interesting by itself with diversity, contradictive and amazing discoveries.

Cavitation is defined as disruption of fluid by arising of negative pressure. The pressure decrease may be achieved by different methods: by fluid stream flow round the body, by solid body movement with high velocity through the fluid, by pushing fluid by plunger, by fast tearing away the plunger, by fluid turbulence. Also it is known acoustic ultrasonic cavitation. Therefore, mainly there are two types of cavitation in the fluid: acoustic and hydrodynamical. It is worth to mention separately the ultrasonic cavitation: sonoluminiscence and sonochemistry. The chemical effects of ultrasonic cavitation chemistry, sonochemistry are well investigated. During the treatment by ultrasound in water develops next reactions: H₂O -----))) H, OH, H₂, H₂O₂

Next thermal spliting reactions may develope in cavitaion bubble as a result of high temperatures:

 H_2O_2 ----- OH+OH; H+HO₂; H_2 +O₂;

As we saw, during the ultrasonic cavitation hydrogen is made and this fact is well known for science, but this process is charactered with high recombination rate. Thus, ultrsonic

cavitaion is not used for production of industrial hydrogen. In our technology hydrodynamical turbulent cavitator is used, which differe from sonochemistry and allows to treat huge mass of water and can obtain significant amount of hydorgen.

Cavitation bubble collapse produces intense local heating (5000K), high pressures (~1000 atm.), and enormous heating and cooling rates (>100k/sec), providing a unique interaction of energy and matter. These extreme conditions affect hydrogen bonding in water and facilitate destruction of water clusters, reducing energy consumption for the electrolysis.

One of the objectives of invention is development of turbulent cavitation theoretical model by mathematical modeling. Another objective of project is theoretical investigation of hydrogen bonding in water and experimental investigation of its effects on water macro properties. We think that hydrogen bonding and cluster formation in water affects electrolysis efficiency.

Developed during cavitaion phenomenon can be considered as plasma. It is also known that plasma chemical methods of hydrogen production are one of the most energy effective. Plamsa chemical processes which are obtained by nonuniforme plasma highfreuquency treatment of water are very energy effective.

In our device cavitation and electrolysis in other words plasma-chemical and electrochemical effective combination is used and achived maximal energy efficiency.

Expected result of invention is development of completely new cheap hydrogen technology by cavitation electrolysis method. It is expected that obtained by this technology electrolyzer will be 10 times cheaper of existed prototypes. The energy efficiency also will increase.

Really we have a new hydrogen production method and device, which by its main characteristics is cavitations physical-chemical reactor - electrolysis cell. We developed absolutely new approach for electrolysis: cavitations electrolysis method. Our approach is based on the investigation of fundamental properties of water, investigation of hydrogen bonding in water and its effects on water properties, investigation of cavitations effects on water fundamental properties and its usage for electrolysis.

The device for production of hydrogen by cavitation electrolysis method is shown on Fig. 1. Numbers indicate:

1,2 - electrodes, cilidrical cathode and anode;

3 - cavitator;

4- water pump;

5 - modulator;

6, 7 – oxigen and hydrogen tanks;

8 – energy source.

Electrodes 1, 2 – anode and cathode are made from stainless steel.

Cavitator includs water input and autput parts.

Water pump 4 is chosen by hydrogen production rates demand.

Modulator 5 includes resonance circuit elements.

Energy source 8 supplaies pulsed voltage.

Cavitation electrolysis device operates on next priciple:

By means of water pump (4) water is supplaied to cavitator (3), in which it undergoes turbulent hydravlic cavitation. Water continues cavitating and turbulent ratation, goes up and appear between coaxial, made from stainless steel cylidrical electrods (1, 2)(cathode and anode). Electrodes are supplaied by special impulse high frequency water own frequency modulate current. On the first eleqctrode – cathode hydrogen is generated, second electrode (anode) – oxigen is generated. Gas acumulation on electrodes by generation from water by cavitaton and electrolysis is more stimulated also by water rotation centrifugal effect. The distance between electrodes should be sufficient that water enriched by oxigena and hydrogen should not mixed up. Electrodes with supplaied water play role of condesator in resonance circute which is regulated acording water own frequencies. Device size is defined by production rates. Acodrdingly is made modulator which is switched on in circute between energys source and cathode. Only main ussue which should be taken into acoount is water own frequency.

As a result a device has not upper and lower production limits.

After passing space throgh electrodes enriched by hydrogen and oxigen water flow by tubes in two diferent tanks and undergoes degassing. After this water by tubes located at the bottom of tanks throgh the pump returns to cyclic system. Obtained hydrogen and oxigen lead out by tubes which are located on hydrogen and oxigen tanks.

Device does not need purificated water and can work on tap water. Device also can work on sea water, with adjusment for sea water frequency. Also it should be taken into account on this case utilization of aditional products obtained during sea water lelctrolysis.

1. US, 6719817B1, 13.04.2004

CLAIMS

1. Hydrogen generation cavitation eleqtrolysis device comprises filled with water cylidrical cavitator, locateded in it cathode and anode, which are made from stainless steel and have coaxial cylindrical form, conected by tube to cavitator circulation water pump, conected to cathode and anode pulsed energy source, switched on in the circute between energy source and cathode modulator, conected to cavitators upper part by tube hydrogen tank, connected to cavitator upper part by tube oxigen tank, connected to these tanks bottom parts by tubes mention above water pump, conected to hydrogen and oxigen tanks gas ouput tubes.

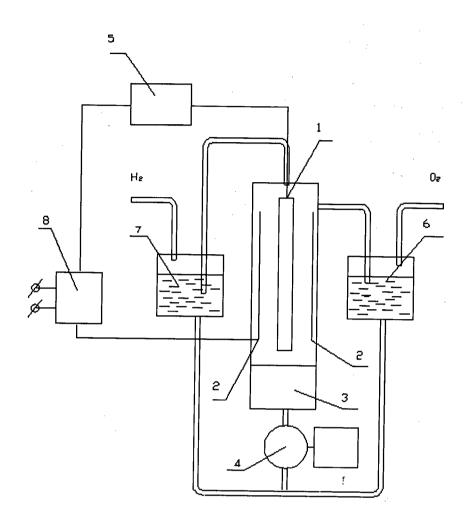


Fig. 1

INTERNATIONAL SEARCH REPORT

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A. CLASSIFICATION OF SUBJECT MATTER INV. C25B1/04 C25B11/02 C25B15/08 C25B9/00 ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

 $\begin{tabular}{ll} Minimum documentation searched (classification system followed by classification symbols) \\ C25B \end{tabular}$

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Х	US 2006/060464 A1 (CHANG CHAK M T [AU]) 23 March 2006 (2006-03-23) claims 53-58,63,65,66 paragraphs [0039], [0048], [0066], [0067], [0073] - [0075], [0079], [0105], [0106], [0116], [0117], [0267] - [0269], [0274]	1
X	US 2012/058405 A1 (KIRCHOFF JAMES A [US] ET AL) 8 March 2012 (2012-03-08) claims 1,18 figures 4-6,12 paragraphs [0011], [0012], [0018], [0020], [0026], [0054], [0069] - [0072], [0075], [0077], [0078] - [0080], [0088], [0123], [0124]	1

Further documents are listed in the continuation of Box C.	X See patent family annex.	
"A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family	
Date of the actual completion of the international search	Date of mailing of the international search report	
12 February 2014	19/02/2014	
Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer Perednis, Dainius	

INTERNATIONAL SEARCH REPORT

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C(Continua	ntion). DOCUMENTS CONSIDERED TO BE RELEVANT	1	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	
Υ	US 2009/159461 A1 (MCCUTCHEN WILMOT H [US] ET AL) 25 June 2009 (2009-06-25) paragraphs [0027] - [0033], [0037], [0150]	1	
Υ	US 6 719 817 B1 (MARIN DANIEL J [US]) 13 April 2004 (2004-04-13) cited in the application claims 1-4 figures 2-8 column 1, lines 6-8 column 2, lines 39-46 column 4, lines 56-61	1	
Y	US 4 184 931 A (INOUE KIYOSHI [JP]) 22 January 1980 (1980-01-22) claims 1,3 column 2, lines 50-63 column 3, lines 41-53		

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/GE2013/000009

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 2006060464 <i>I</i>	1 23-03-2006	AU 2003221634 A1 AU 2009200889 A1 CA 2483753 A1 CN 1653865 A GB 2405255 A GB 2424757 A JP 2005529455 A US 2006060464 A1 WO 03096767 A1	11-11-2003 26-03-2009 20-11-2003 10-08-2005 23-02-2005 04-10-2006 29-09-2005 23-03-2006 20-11-2003
US 2012058405 /	1 08-03-2012	US 2012058405 A1 WO 2013003499 A2	08-03-2012 03-01-2013
US 2009159461 A	1 25-06-2009	CN 102015544 A EP 2257498 A2 US 2009159461 A1 US 2012318671 A1 WO 2009111582 A2	13-04-2011 08-12-2010 25-06-2009 20-12-2012 11-09-2009
US 6719817 E	1 13-04-2004	NONE	
US 4184931 /	22-01-1980	DE 2810528 A1 FR 2383247 A1 GB 1552311 A US 4184931 A	14-09-1978 06-10-1978 12-09-1979 22-01-1980