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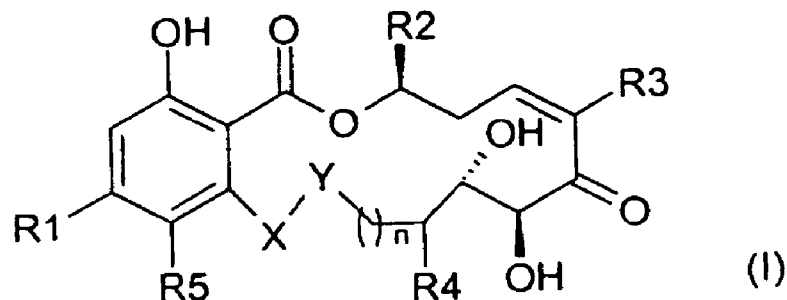
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(54) Title: IRREVERSIBLE INHIBITORS USEFUL FOR THE TREATMENT OF KINASE-RELATED PATHOLOGIES



(57) Abstract: The present invention provides new compounds (I) having a kinase inhibitory activity and useful for treating cancer.

## Irreversible inhibitors useful for the treatment of kinase-related pathologies.

### *Field of the Invention*

The present invention relates to new compounds useful as drug, in particular in the  
5 oncology field.

### *Background of the Invention*

Phosphorylation of proteins is one of the most prevalent cellular mechanisms for regulating protein function in a rapid and reversible fashion. There are approximately 518 protein kinases encoded in the human genome as well as a smaller set of protein phosphatases  
10 which can work synergistically or concurrently at various levels in cellular pathways, thus resulting in a tremendous combinatorial output. Virtually every signal transduction pathway implicates kinases and as much as 30% of all human proteins may be modified by them. A number of pathologies ranging from oncology to inflammation and neurodegenerative diseases can be attributed to a dysfunctional kinase. Accordingly, kinases have become one  
15 of the most intensively pursued classes of proteins for drug discovery, the vast majority being currently investigated for the treatment of cancer. To date, 11 kinase inhibitors have received FDA approval and there are approximately 30 distinct kinase targets being developed at the level of phase I clinical trials.

Four natural resorcylic acid lactones (RAL) containing a *cis*-enone moiety have been  
20 reported to be potent and irreversible kinase inhibitors (N. Winssinger, S. Barluenga, *Chem. Commun.* **2007**, 22-36; S. Barluenga, et al, *C. R. Chem.* **2008**, *11*, 1306-1317; T. Hofmann, K.-H. Altmann, *C. R. Chem.* **2008**, *11*, 1318-1335.) (L-783277 also named FR265082; LL-Z1640-2 also named 5Z-7-oxozeaenol and FR148083; radicicol A and hypothemycin). Two of these compounds (5Z-7-oxozeaenol and hypothemycin) have already been shown to be  
25 effective in animal models (J. Ninomiya-Tsuji, et al, *J. Biol. Chem.* **2003**, *278*, 18485-18490.; H. Tanaka, et al, *Jap. J. Cancer Res.* **1999**, *90*, 1139-1145.). As such, these *cis*-enone RAL represent a pharmacologically validated starting point for kinase inhibition. More recently, it has been shown by Santi and coworkers that hypothemycin irreversibly inactivates ERK2 by forming a covalent Michael adduct with the cys166 positioned in the ATP-binding pocket of  
30 this kinase (A. Schirmer, et al, *Proc. Natl. Acad. Sci. USA* **2006**, *103*, 4234-4239.). A

structure-bioinformatics analysis of the kinome revealed that 46 out of the 518 putative kinases contained a cysteine residue adequately positioned to participate in the Michael addition onto the *cis*-enone of hypothemycin. Nevertheless, it was clearly shown that a significant difference in the kinetics of inactivation existed amongst 16 out of the 46 tested  
5 kinases. Our profile of the related chloro-radicicol A against 127 kinases showed that this compound inhibited exclusively kinases from this subset of 46, however not all were inhibited (GSK3 $\beta$  for example was not inhibited) (P. Y. Dakas, et al, *Angew. Chem. Int. Ed. Engl.* **2007**, *46*, 6899-6902.). Furthermore it was shown by Nakajima and Miyake that there was a significant difference in activity (50 fold) between LL-Z1640-2 (FR148083) and L-783277  
10 (FR265082) in their inhibition of ERK2 suggesting that a good level of selectivity may be achieved within this subset of the kinome as L-783277 was reported to be a potent MEK inhibitor (4 nM) compared to LL-Z1640-2 (411 nM).

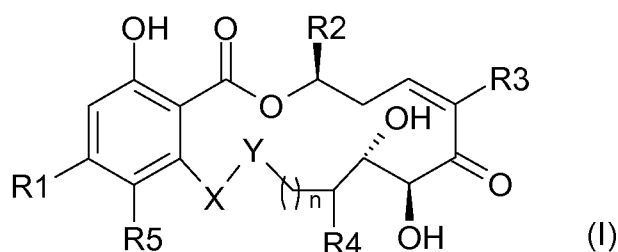
Analogues of radicicol A have been disclosed in WO 03/076424, WO 06/036941 and WO 08/149244.

15 However, there is still a strong need for kinase inhibitors that have improved potency and selectivity.

### ***Summary of the Invention***

The present invention provides new compounds with improved potency and selectivity as kinase inhibitors. In particular, they predominantly inhibit the vascular endothelial growth  
20 factor receptors (VEGFRs) and platelet-derived growth factor receptors (PDGFRs). Accordingly, they are useful as drug, in particular of therapeutic interest for treating cancer. A particular group of compounds disclosed herein present an advantage in comparison with recent approved small molecule inhibitors (sorafenib and sunitinib) targeting the VEGF receptors. While sunitinib did promote lung metastasis in an in vivo model, compounds of  
25 Formula II strongly inhibited lung metastasis. Therefore, compounds of Formula II are of a great therapeutic interest.

The present invention relates to a compound of formula I, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



n is 0 or 1;

R1 is selected from the group consisting of hydrogen, halogen, cyano, -OR, -NRR', -NRS(O)R', -NRS(O<sub>2</sub>)R', -SR, -S(O)R, -S(O<sub>2</sub>)R, -OC(O)R, -C(O)R, -C(O)OR, -NRC(O)R', -C(O)NRR', -OC(O)OR, aliphatic, heteroaliphatic, acyl, aryl, heteroaryl, alkylaryl, arylalkyl, alkylheteroaryl and heteroarylalkyl;

R2 is hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

R3 is hydrogen or halogen;

R4 is hydrogen or hydroxyl;

R5 is hydrogen, halogen, OR, NHR, NH-COR,

R and R', independently, are hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

X-Y is CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, or CH<sub>2</sub>-S, and,

provided that the compound is not a compound of Formula I with

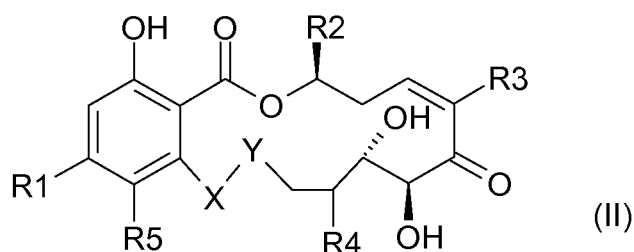
n is 0, R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, X-Y is O-CH<sub>2</sub> or CH<sub>2</sub>-S; or

n is 0, R1 is hydroxyl, R4 and R5 are hydrogen, R2 is methyl, R3 is fluorine, X-Y is O-CH<sub>2</sub>; or

n is 0, R1 is methoxy, R2 is methyl, R3 is fluorine, R4 and R5 are hydrogen, X-Y is CH<sub>2</sub>-S.

In a preferred embodiment, R1 is selected from the group consisting of -OR, -NHR and -SR.

In a particularly preferred embodiment, the invention relates to the compound having the formula II,



R1, R2, R3, R4, R5 and X-Y being as defined in compounds of Formula (I).

Preferably, the compound of Formula II has one or several of the following features:

- R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or

- R4 is hydrogen or hydroxyl, preferably hydrogen; and/or
- R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>; and/or

5 provided that the compound does not have concomitantly a hydrogen in R2 and O-CH<sub>2</sub> in X-Y.

More preferably, the compound of Formula II has one or several of the following features:

- R1 is hydroxyl or methoxy; and/or
- 10 - R2 is hydrogen or methyl; and/or
- R3 is hydrogen; and/or
- R4 is hydrogen; and/or
- R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>;

15 provided that the compound does not have concomitantly a hydrogen in R2 and O-CH<sub>2</sub> in X-Y.

Still more preferably, in the compound of Formula II, X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

In a particular embodiment of the compound of Formula II, R1 is hydroxyl or methoxy, R2, R3, R4 and R5 are hydrogen, and X-Y is CH=CH or CH<sub>2</sub>-CH<sub>2</sub>.

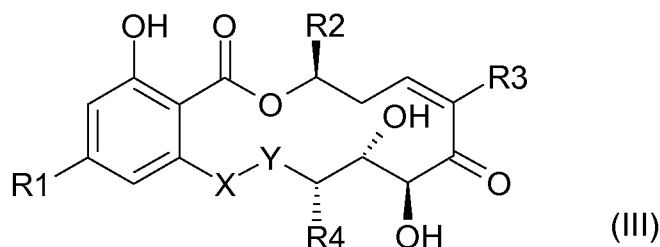
20 In another particular embodiment of the compound of Formula II, R1 is hydroxyl or methoxy, R2 is methyl, R3, R4 and R5 are hydrogen, and X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

Accordingly, the compound of Formula II may be selected from the group consisting of:

- 25 - a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, 30 R2 is methyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH=CH;

- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and,
- a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

5 In another particularly preferred embodiment, the invention relates to the compound having the formula III,



R1, R2, R3, R4 and X-Y being as defined in Formula (I).

Preferably, the compound of Formula (III) has one or several of the following features:

- 10 - R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- 15 - R4 is hydrogen or hydroxyl; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

More preferably, the compound of Formula (III) has one or several of the following features:

- 20 - R1 is hydroxyl; and/or
- R2 is hydrogen or methyl, preferably methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

25 Still more preferably, in the compound of Formula III, R2 is methyl, ethyl and isopropyl, preferably methyl, and R4 is hydroxyl.

Accordingly, the compound of Formula III may be selected from the group consisting of:

- 30 - a compound of formula III, wherein R1 is hydroxyl, R3 and R4 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

- a compound of formula III, wherein R1 is hydroxyl, R3 is hydrogen, R2 is methyl, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R2, R3 and R4 are hydrogen, and X-Y is O-CH<sub>2</sub>;
- 5      - a compound of formula III, wherein R1 is hydroxyl, R2 and R3 are hydrogen, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>; and,
- a compound of formula III, wherein R1 is hydroxyl, R4 is hydrogen, R2 is methyl, R3 is fluorine, and X-Y is CH=CH.

The present also relates to any of the above-disclosed compounds as a drug. It further relates to a pharmaceutical composition comprising any of the above-disclosed compounds and a pharmaceutically acceptable carrier.

The present invention relates to any of the above-disclosed compounds for use in the treatment of cancer, to the use of any of the above-disclosed compounds for the preparation of a medicament for treating cancer, and to a method for treating cancer in a subject, comprising administering a therapeutic effective amount of any of the above-disclosed compounds to said subject.

In a more particular aspect, the present invention relates to any of the above-disclosed compounds of Formula (II) for use in the treatment of cancer while preventing metastasis occurrence, to the use of any of the above-disclosed compounds for the preparation of a medicament for treating cancer while preventing metastasis occurrence, and to a method for treating cancer while preventing metastasis occurrence in a subject, comprising administering a therapeutic effective amount of any of the above-disclosed compounds to said subject.

### ***Brief Description of the Drawings***

**Figure 1.** Structure of natural resorcylic acid lactones bearing a *cis*-enone.

25      **Figure 2.** General structure and synthetic planning of a library of *cis*-enone RAL.

**Figure 3.** Synthesis of R<sub>F</sub> tagged fragments 6a-d. *a*) PMBOC=NHCCl<sub>3</sub> (1.0 equiv), CSA (0.14 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 12 h, 72-86%; *b*) DIBAL-H (1.1 equiv), PhMe, -78°C, 1 h, 45-52%; *c*) CBr<sub>4</sub> (4.0 equiv), PPh<sub>3</sub> (8.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 45 min, 63%; *d*) *n*BuLi (2.0 equiv), THF, -78°C, 1 h, and 23°C, 1.5 h, 85%; *e*) HC<sub>2</sub>MgBr (1.5 equiv), THF, -78°C to 23°C, 12 h, 88%; *f*) EOM-Cl (3.0 equiv), *i*Pr<sub>2</sub>NEt (3.0 equiv), *n*-Bu<sub>4</sub>N<sup>+</sup>I (cat), CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 12 h, 89%; *g*) Cp<sub>2</sub>ZrCl<sub>2</sub> (0.25 equiv), AlMe<sub>3</sub> (3.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 19 h and reflux 5 days; *h*) I<sub>2</sub> (1.5 equiv), THF, -30°C, 15 min, 62%. CSA = camphorsulfonic acid, Cp =

cyclopentadienyl, DIBAL-H = diisobutylaluminium hydride, EOM = ethoxymethyl, PMB = *p*-methoxybenzyl, THF = tetrahydrofuran.

**Figure 4.** Synthesis of fragments 7. *a*) 2-methoxypropene (2.0 equiv), *p*TsA (0.04 equiv), CaSO<sub>4</sub> (0.25 equiv), DMF, 0°C, 3 h, 60% or 2,2-dimethoxypropane (3.5 equiv), *p*TsA (0.02 equiv), acetone, 23°C, 12 h, 90%; *b*) LiAlH<sub>4</sub> (1.4 equiv), THF, 0°C to 23°C, 2 h, 95%;  
 5 *c*) TBDPS-Cl (0.9 equiv), imidazole (1.5 equiv), DMF, 23°C, 2-12 h, 66-99%; *d*) SO<sub>3</sub>.Py complex (3.5 equiv), Et<sub>3</sub>N (4.9 equiv), CH<sub>2</sub>Cl<sub>2</sub>/DMSO 4/1, 0°C to 23°C, 0.5-1 h, 91-94%; *e*) BrPPh<sub>3</sub>CH<sub>3</sub> (3.0 equiv), NaHDMS (2.8 equiv), THF, -78°C to 23°C, 1-12 h, 72-86%; *f*) PivCl (2.0 equiv), Et<sub>3</sub>N (4.0 equiv), DMAP (0.2 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C to 23°C, 12 h, 93%; *g*) 9-BBN  
 10 (2.2 equiv), THF, 0°C to 23°C, 3.5 h, then 3N NaOH/H<sub>2</sub>O<sub>2</sub>, 0°C to 23°C, 1.5 h, 94%; *h*) NaOMe (3.0 equiv), MeOH, 23°C, 16 h, 83%; *i*) BnOH (7.2 equiv), 23°C, 10 h, 96%; *j*) EOM-Cl (8.0 equiv), *i*Pr<sub>2</sub>NEt (8.0 equiv), *n*Bu<sub>4</sub>NI (cat), CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 12 h, 95%; *k*) H<sub>2</sub>, Pd/C (5 mol%), MeOH, 23°C, 5 h, 90%; *l*) O<sub>3</sub>, PPh<sub>3</sub> (2.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, -78°C, 1 h, 90%. Bn = benzyl, 9-BBN = 9-borabicyclo[3.3.1]nonane, DMAP = 4-dimethylaminopyridine, DMF =  
 15 N,N-dimethylformamide, DMSO = dimethylsulfoxide, EOM = ethoxymethyl, NaHDMS = sodium bis(trimethylsilyl)amide, Piv = pivaloyl, *p*TsA = *p*-toluene sulfonic acid, py = pyridine, TBAI = tetrabutylammonium iodide, TBDPS = *tert*-butyldiphenylsilyl, THF = tetrahydrofuran.

**Figure 5.** Synthesis of compounds 5 (12 permutations) via fluororous mixture synthesis  
 20 using three pools. *a*) *n*BuLi (1.1 equiv), THF, -78°C, 10 min, then 7 (1.2 equiv), -78°C, 30 to 60 min, 80%; *b*) BzCl (2.5 equiv), pyridine (2.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C to 23°C, 4 h, 90%; *c*) TBAF (1.5 equiv), THF, 23°C, 3 h, 86-94%; *d*) H<sub>2</sub>, Pd/CaCO<sub>3</sub> (1.72 equiv), MeOH, 23°C, 45 min, 90-96%; *e*) PPh<sub>3</sub> (1.5 equiv), imidazole (2.5 equiv), I<sub>2</sub> (1.5 equiv), THF, 0°C to 23°C, 30 min, 70-93%; *f*) 6d (1.0 equiv), *t*BuLi (2.0 equiv), Et<sub>2</sub>O, -78°C, 20 min, then 7 (1.2 equiv),  
 25 pentane, -78°C to 0°C, 7 h, >80%. Bz = benzoyl, TBAF = tetrabutylammonium fluoride, THF = tetrahydrofuran.

**Figure 6.** Synthesis of fragments 4. *a*) TFA (20.0 equiv), TFAA (7.6 equiv), acetone, 23°C, 12 h, 54%; *b*) EOM-Cl (4.0 equiv), *i*Pr<sub>2</sub>NEt (4.0 equiv), *n*-Bu<sub>4</sub>N<sup>+</sup>I (cat), CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 12 h, quant.; *c*) TMSCH<sub>2</sub>CH<sub>2</sub>OH (4.0 equiv), NaHDMS (4.4 equiv), THF, 0°C to 23°C, 12 h,  
 30 69%; *d*) LDA (2.0 equiv), THF, -78°C, 30 min, (PhSe)<sub>2</sub> (0.9 equiv), THF, -78°C, 2 h, 75%. EOM = ethoxymethyl, LDA = lithium diisopropylamide, NaHDMS = sodium bis(trimethylsilyl)amide, TFA = trifluoro acetic acid, TFAA = trifluoro acetic anhydride, THF = tetrahydrofuran, TMSE = 2-trimethylsilylethyl.



**Figure 7.** Library synthesis. *a*) LDA (2.0 equiv), THF, -78°C, 10 min, then 5 (0.9 equiv), THF, -78°C, 2.0 h, 60%; *b*) LDA (2.0 equiv), THF/HMPA (10/1), -78°C, 10 min, then 5 (0.9 equiv), THF, -78°C, 30 min and then H<sub>2</sub>O<sub>2</sub> (2.0 equiv), THF, 23°C, 2 h, 64-85%; *c*) K<sub>2</sub>CO<sub>3</sub> (2.0 equiv), 5 (0.9 equiv), DMF, 100°C, 12 h, 94-100%; *d*) DDQ (1.2 equiv), CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O (2/1), 23°C, 2 h, 85-96%; *e*) TBAF (3.0 equiv), THF, 23°C, 2 h, quant.; *f*) R<sub>F</sub>-PPh<sub>3</sub> (2.0 equiv), R<sub>F</sub>-DIAD (2.0 equiv), PhMe, 23°C, 2 h, 50-85%; *g*) 1% NaOH in MeOH, reflux, 12 h, 80-90%; *h*) PS-SO<sub>3</sub>H (5.0 equiv), MeOH, 50°C, 2 h, >90%; *i*) PS-IBX (3.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>/few drops of DMSO, 23°C, 1-3 h (monitored by TLC), 50%; *j*) DMP (1.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, reflux, 4 h, 80-90%; *k*) 40% aq. HF in CH<sub>3</sub>CN (1/10), 23°C, 3-6 h, 50-70%.

DDQ = 2,3-dichloro-5,6-dicyanobenzoquinone, DIAD = diisopropyl azodicarboxylate, DMF = N,N-dimethylformamide, DMP = Dess-Martin periodinane, DMSO = dimethylsulfoxide, EOM = ethoxymethyl, HMPA = hexamethylphosphoramide, IBX = 2-iodoxybenzoic acid, LDA = lithium diisopropylamide, PS = polystyrene, TBAF = tetrabutylammonium fluoride, THF = tetrahydrofuran.

**Figure 8.** *a*) DMDO (4.0 equiv), CH<sub>3</sub>CN, 0°C 1-2 h, 60-70%; *b*) CH<sub>2</sub>N<sub>2</sub> (5-10 equiv), Et<sub>2</sub>O, 23 °C, 3-12 h, 50-60%; *c*) RNH<sub>2</sub>OH.HCl (10.0 equiv), pyridine, 40 °C, 12 h, >50%. DMDO = dimethyldioxirane.

**Figure 9.** IC<sub>50</sub> of selected library members against a panel of 19 kinases.

**Figure 10.** Residual activity as a percentage of control for 31 of the 46 putative kinase inhibited by two a resorcyllide containing an alkane (1afh) or ether (1bgi) at the benzylic position.

**Figure 11.** Residual activity as a percentage of control for mutation of kinases.

**Figure 12.** Comparison of cellular efficacy between 1agh, 1aej and known approved drug on RENCA renal carcinoma (Sunitinib and Sorafenib).

**Figure 13.** Divergent synthesis of compounds 3 and 5.

**Figure 14.** Activity of different RALs in a RENCA cellular proliferation assay.

**Figure 15.** In vivo efficacy. Group 1: vehicle (n=12/10); Group 2: sunitinib, 20 mg, q1d (n=12); Group 3: 3, 20 mg, q1d (n=12/11); Group 4: 3, 20 mg, q2d (n=12); Group 5: 5, 20 mg, q1d (n=12/11). A: In vivo luciferase activity (Photons/s); B: Primary tumor volume (mm<sup>3</sup>); C: Primary tumor weight (g); . Data are displayed as means +/- SEM. The inset shows the individual data points together with their corresponding mean values.

**Figure 16.** Metases of to the lung. Group 1: vehicle (n=12/10); Group 2: sunitinib, 20 mg, q1d (n=12); Group 3: 3, 20 mg, q1d (n=12/11); Group 4: 3, 20 mg, q2d (n=12); Group 5: 5, 20 mg, q1d (n=12/11). A: Number of lung metastases Data are displayed as means +/-

SEM. The inset shows the individual data points together with their corresponding mean values. B: Lung luciferase activity (LU/ $\mu$ g protein). Data are displayed as means  $\pm$  SEM. The inset shows the individual data points together with their corresponding mean values.

**Figure 17.** Tumor vasculature histology (CD-31 staining) for vehicle group, group 3 (3, 20 mg, q1d); group 5: (5, 20 mg, q1d).

**Figure 18.** Key disconnections for the synthesis of 5-fluoro-*cis*-enone resorcylics.

**Figure 19.** Synthesis of fluoroenones 1 and 2. **A:** Synthesis of key intermediate 14 from ester 5. Reagents and conditions: a) TBDPSCl (1.1 equiv), imidazole (2.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 238C, 13 h, 90%; b) Dibal-H (1.1 equiv), PhMe, -788C, 1 h, 85%; c) 1. (EtO)<sub>2</sub>POCHFCO<sub>2</sub>Et, nBuLi, THF, 0 8C, 1 h; 2. 6 (1.0 equiv), 0!238C, 12 h, 83 %; d) Dibal-H (2.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0!238C, 85%; e) (COCl)<sub>2</sub> (1.5 equiv), DMSO (2.45 equiv), Et<sub>3</sub>N (4.0 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 1 h, 98%; f) 1. PMBO(CH<sub>2</sub>)<sub>2</sub>CCCH (2.1 equiv), nBuLi (2.0 equiv), THF, -788C, 1 h; 2. 7 (1.0 equiv), -788C, 1 h, 62%; g) H<sub>2</sub>, Pd/CaCO<sub>3</sub>, MeOH, 238C, 25 min, 97%; h) BzCl (1.5 equiv), pyridine (2.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0!238C, 14 h, 91%; i) DDQ (1.2 equiv), CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O, 238C, 3 h, 92%; j) PPh<sub>3</sub> (1.8 equiv), imidazole (3.0 equiv), I<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 238C, 1.5 h, 90%; k) OSO<sub>4</sub> (1 %), NMO·H<sub>2</sub>O (2.0 equiv), THF, 238C, 15 h, 78%; l) 2-methoxy propene (1.5 equiv), PPTS (0.1 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 238C, 1 h, 93%. Abbreviations: Bz, benzoyl; Dibal-H, diisobutylaluminium hydride; DDQ, 2,3-dichloro-5,6-dicyanobenzoquinone; NMO, N-methylmorpholine N-oxide; PPTS, pyridinium p-toluenesulfonate; TBDPS, tert-butyldiphenylsilyl. **B:** Synthesis of fluoroenones 1 and 2 from key intermediate 14. Reagents and conditions: a) 1. LDA (2.0 equiv), THF/HMPA, -788C, 10 min; 2. 14 (1.0 equiv), THF, -788C, 20 min, 80 %; 3. H<sub>2</sub>O<sub>2</sub> (2.0 equiv), THF, 238C, 3 h, 92%; b) K<sub>2</sub>CO<sub>3</sub> (2.0 equiv), 14 (1.0 equiv), DMF, 80 8C, 12 h, 99%; c) TBAF (10 equiv), THF, 238C, 48–56 h; d) PPh<sub>3</sub> (2.0 equiv), DIAD (2.0 equiv), PhMe, 238C, 2–4 h, 70–74% (two steps); e) 1% NaOH in MeOH, 508C, 2–2.5 h, 69– 76%; f) DMP (1.5 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 238C, 14 h, 80–88 %; g) 40% aq HF in CH<sub>3</sub>CN (1:10), 23 8C, 6.5 h, 65%; h) PS-SO<sub>3</sub>H (10 equiv), MeOH, 23 8C, 9 h, 70%. Abbreviations: DIAD, diisopropyl azodicarboxylate; LDA, lithiumdiisopropylamide; HMPA, hexamethylphosphoramide; DMF, N,N-dimethylformamide; DMP, Dess–Martin periodinane; TBAF, tetrabutylammoniumfluoride; PS, polystyrene supported.

**Figure 20.** Inhibition of VEGF-R2 and KIT.

**Figure 21.** Cellular inhibition of VEGFR2 by a) LL-Z1640-2 (IC<sub>50</sub>=6.5 nM), b) fluoroenone 1 (IC<sub>50</sub>=17 nM) and c) fluoroenone 2 (IC<sub>50</sub>=70 nM). Assays were performed in triplicate.

### ***Detailed Description of the Invention***

The present invention provides kinase inhibitors with improved potency and selectivity. The vascular endothelial growth factor receptors (VEGFRs) and platelet-derived growth factor receptors (PDGFRs) were predominantly the most inhibited kinases by the provided kinase inhibitors.

A broader profile against 359 kinases (excluding mutants) revealed a pattern similar to that of sunitinib (SU11248) which is generally considered a multitarget receptor kinase inhibitor of VEGFRs and PDGFRs. Compounds of Formula I were good inhibitors of VEGFRs in vitro, and inhibited tumor growth in vivo with comparable efficacy to sunitinib, an FDA-approved VEGFRs inhibitor.

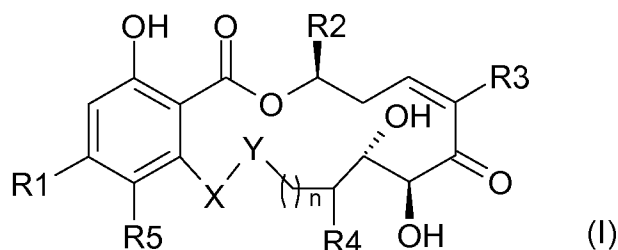
Therefore, compounds of Formula I, as well as compounds of Formula II, IIa, IIb and/or III and any compound specifically disclosed herein, are of therapeutic interest for treating cancer. Indeed, multiple strategies for inhibiting the VEGF pathway have been shown to hinder tumor growth and the recent approval of small molecule inhibitors (sorafenib and sunitinib) as well as neutralizing antibodies (bevacizumab) targeting the VEGF receptors have clinically validated this strategy.

However, enduring clinical response are rare and there is mounting evidence that antiangiogenic therapy, while delaying primary tumor growth, can concomitantly promote tumor adaptation and progression to stages of greater malignancy with heighten invasiveness and distant metastasis. It has recently been shown that short-term sunitinib treatment resulted in accelerated metastasis in a mice model. However, most preclinical studies focus on primary tumor growth with less attention on metastasis. While sunitinib and sorafenib are very effective inhibitors of the angiogenic pathways and primary tumor growth, the inventors sought to evaluate resorcylic acids, as the irreversible inhibition of the target kinases may provide different outcome. Several programs to develop irreversible kinase inhibitors have already progressed to the clinic.

In addition, while compounds of Formula III did promote lung metastasis in an in vivo model to a similar extent as sunitinib, compounds of Formula II strongly inhibited lung metastasis. Therefore, compounds of Formula II are of a great therapeutic interest.

### **Compounds**

The present invention relates to a compound of formula I, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



wherein,

n is 0 or 1;

R1 is selected from the group consisting of hydrogen, halogen, cyano, -OR, -NRR', -NRS(O)R', -NRS(O<sub>2</sub>)R', -SR, -S(O)R, -S(O<sub>2</sub>)R, -OC(O)R, -C(O)R, -C(O)OR, -NRC(O)R', -C(O)NRR', -OC(O)OR, aliphatic, heteroaliphatic, acyl, aryl, heteroaryl, alkylaryl, arylalkyl, alkylheteroaryl and heteroarylalkyl;

R2 is hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

R3 is hydrogen or halogen;

R4 is hydrogen or hydroxyl;

R5 is hydrogen, halogen, OR, NHR, NH-COR, R being hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

X-Y is CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, or CH<sub>2</sub>-S, and,

provided that the compound is not a compound of Formula I with

n is 0, R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, X-Y is O-CH<sub>2</sub> or CH<sub>2</sub>-S; or

n is 0, R1 is hydroxyl, R4 and R5 are hydrogen, R2 is methyl, R3 is fluorine, X-Y is O-CH<sub>2</sub>; or

n is 0, R1 is methoxy, R2 is methyl, R3 is fluorine, R4 and R5 are hydrogen, X-Y is CH<sub>2</sub>-S.

By "C<sub>1</sub>-C<sub>5</sub> alkyl" is intended a saturated alkyl, in particular methyl, ethyl, propyl, isopropyl, butyl, isobutyl, *tert*-butyl, pentyl, and the other isomeric forms thereof.

Preferably, the C<sub>1</sub>-C<sub>5</sub> alkyl of Formula I is a C<sub>1</sub>-C<sub>3</sub> alkyl. By "C<sub>1</sub>-C<sub>3</sub> alkyl" is intended methyl, ethyl, propyl, and isopropyl.

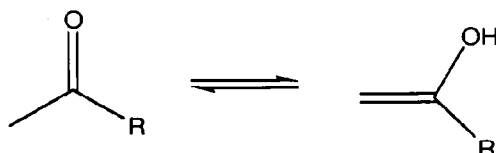
By "halogen" is intended an halogen selected from bromine, fluorine, iodine and chlorine, more preferably fluorine or chlorine.

By "aliphatic" is intended straight-chain, branched or cyclic C<sub>1</sub>-C<sub>12</sub> hydrocarbons which are completely saturated or which contain one or more units of unsaturation but which are not aromatic.

By "heteroaliphatic" as used herein is intended to have its customary meaning in the art and includes an aliphatic group substituted with one or more atoms other than carbon or

hydrogen in the aliphatic chain, nonlimiting examples of which are nitrogen, oxygen, sulfur, phosphorus, silicon or boron.

The term "tautomer" as used herein refers to alternate structures which are recognized in the art to be in equilibrium with the depicted structure. For example, the enol structure below is a tautomer of the ketone structure and recognized to be in equilibrium with the ketone structure.



As used herein, the term "solvate" or "pharmaceutically acceptable solvate," is a solvate formed from the association of one or more solvent molecules to one or more molecules of a compound of any one of formulae I, II, IIa, IIb, III or the compounds specifically disclosed herein. The term solvate includes hydrates (e.g., hemi-hydrate, monohydrate, dihydrate, trihydrate, tetrahydrate, and the like).

The term "aryl" as used herein is intended to have its customary meaning in the art, and includes any stable monocyclic, bicyclic, or tricyclic carbon ring, wherein at least one ring is aromatic, and especially phenyl, biphenyl, or naphthyl.

The term "aralkyl" or "arylalkyl" refers to an aryl group substituted with an alkyl substituent or linked to the molecule through an alkyl group as defined above.

The term "acyl" includes a group of the formula  $C(O)R_x$ , wherein  $R_x$  is an straight, branched, or cyclic alkyl.

The term "heterocycle", "heterocyclyl", or "heterocyclic" as used herein includes non-aromatic ring systems having four to fourteen members, preferably five to ten, in which one or more ring carbons, preferably one to four, are each replaced by a heteroatom.

Based on structure-activity relationship, it has been determined that R1 can include a high variety of groups, without changing the kinase inhibitory activity of the compound. Indeed, this substituent is pointing towards the solvent and only provides marginal changes in enzyme selectivity or affinity. On the other hand, a variety of substituent can be incorporated at that position to enhance the pharmacological properties at this position. For example, enzymatically stable substituent such as an alkyl ether may be used to reduce the metabolic clearance by inhibiting the glucuronidation and include a morpholino group to enhance aqueous solubility. Analogously, an aniline substituent provides a protonable group which can enhance aqueous solubility without affecting enzyme selectivity of affinity.

In a preferred embodiment, R1 is selected from the group consisting of –OR, –NHR and –SR. Preferably, R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl.

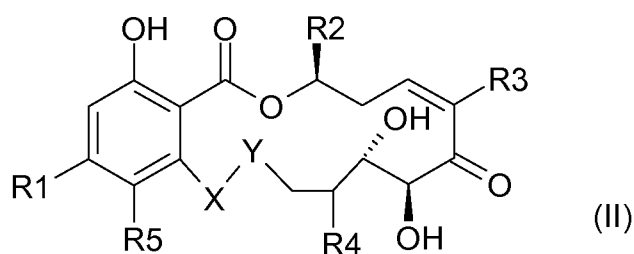
Preferably, R3 is hydrogen or fluorine.

5 Preferably, R5 is hydrogen or halogen. Preferably, R5 is hydrogen or chlorine. In particular, when n is 0, R5 is preferably hydrogen.

Preferably, the compound of Formula I has one or several (2, 3, 4, 5 or all) of the following features:

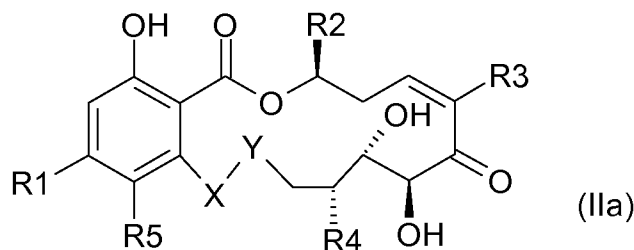
- n is 0 or 1; and/or
- 10 - R1 is –OR, –SR or –NHR and R is selected from hydrogen, methyl, ethyl and isopropyl, preferably R1 is –OR, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- 15 - R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl; and/or
- R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

20 In a first embodiment, the present invention relates to a compound of formula II, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



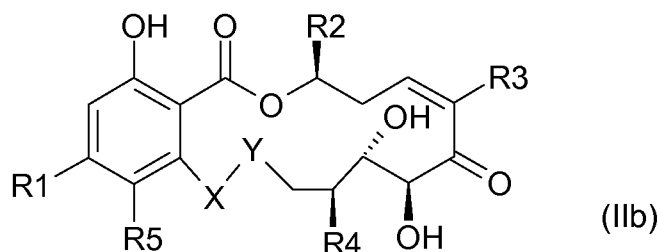
R1, R2, R3, R4, R5 and X-Y being as defined in Formula I.

25 More specifically, the present invention may relate to a compound of formula IIa, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



R1, R2, R3, R5 and X-Y being as defined in Formula I and R4 is hydroxy.

Alternatively, the present invention may relate to a compound of formula IIb, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



5 R1, R2, R3, R5 and X-Y being as defined in Formula I and R4 is hydroxy.

Preferably, the compound is of Formula II, IIa or IIb has one or several (2, 3, 4, 5 or all) of the following features:

- R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- 10 - R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl, preferably hydrogen; and/or
- R5 is hydrogen; and/or
- 15 - X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

In a particular preferred embodiment, the compound of Formula II does not have concomitantly a hydrogen for R2 and O-CH<sub>2</sub> for X-Y.

20 In a preferred embodiment, the compound of Formula II, IIa or IIb has one or several (2, 3, 4, 5 or all) of the following features:

- R1 is hydroxyl or methoxy; and/or
- R2 is hydrogen or methyl; and/or
- R3 is hydrogen; and/or
- R4 is hydrogen; and/or
- 25 - R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>;

provided that the compound of Formula II, IIa or IIb does not have concomitantly a hydrogen for R2 and O-CH<sub>2</sub> for X-Y.

30 In another particular preferred embodiment, the compound of Formula II, IIa or IIb presents CH<sub>2</sub>-CH<sub>2</sub> for X-Y.

In a particular embodiment, the compound of Formula II, IIa or IIb presents  $\text{CH}_2\text{-CH}_2$  for X-Y and has one or several of the following features:

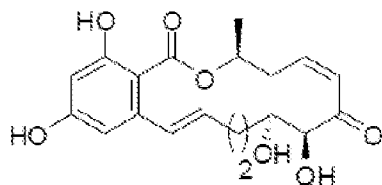
- R1 is hydroxyl or methoxy; and/or
- R2 is hydrogen or methyl; and/or
- 5     - R3 is hydrogen; and/or
- R4 is hydrogen; and/or
- R5 is hydrogen.

In another particular embodiment, the compound of Formula II, IIa or IIb presents R3, R4 and R5 being hydrogen, R1 being hydroxyl or methoxy, and R2 being hydrogen or methyl  
10     and X-Y being selected from  $\text{CH=CH}$ ,  $\text{CH}_2\text{-CH}_2$ , and  $\text{O-CH}_2$ . Preferably, when R2 is hydrogen, X-Y is  $\text{CH=CH}$  or  $\text{CH}_2\text{-CH}_2$ , more preferably is  $\text{CH}_2\text{-CH}_2$ . Alternatively, when R2 is methyl, X-Y is selected from  $\text{CH=CH}$ ,  $\text{CH}_2\text{-CH}_2$ , and  $\text{O-CH}_2$ . More preferably, R1 is hydroxyl.

In a particularly preferred embodiment, the compound of Formula II, IIa or IIb has all  
15     the above mentioned features. Accordingly, the compound of Formula II, IIa or IIb has R1 being hydroxyl or methoxy, R2 being hydrogen or methyl, R3, R4 and R5 being hydrogen, and X-Y being selected from  $\text{CH=CH}$ ,  $\text{CH}_2\text{-CH}_2$ ,  $\text{O-CH}_2$ , provided that the compound does not have concomitantly a hydrogen for R2 and  $\text{O-CH}_2$  for X-Y. Therefore, a particular embodiment of the invention relates to a compound of Formula II, IIa or IIb having R1 being  
20     hydroxyl or methoxy, R2, R3, R4 and R5 being hydrogen, and X-Y being  $\text{CH=CH}$  or  $\text{CH}_2\text{-CH}_2$ , more preferably being  $\text{CH}_2\text{-CH}_2$ . In an alternative embodiment, the invention relates to a compound of Formula II, IIa or IIb having R1 being hydroxyl or methoxy, R2 being methyl, R3, R4 and R5 being hydrogen, and X-Y being selected from  $\text{CH=CH}$ ,  $\text{CH}_2\text{-CH}_2$ , and  $\text{O-CH}_2$ .

The invention relates in particular to one or several of the following compounds of  
25     Formula II, IIa or IIb:

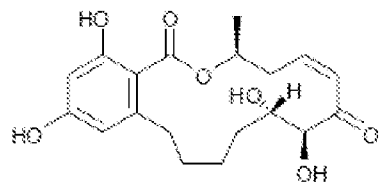
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is  $\text{CH=CH}$ ;



(also called herein 1aej or RAL 5)

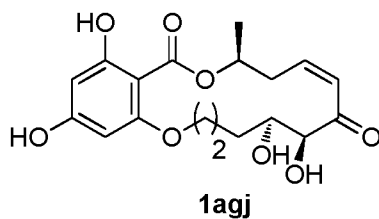
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen,  
30     R2 is methyl, and X-Y is  $\text{CH}_2\text{-CH}_2$ ;



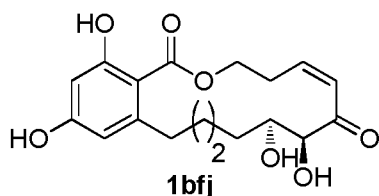


(also called herein 1afj or RAL 1)

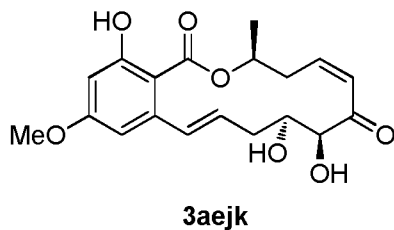
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is O-CH<sub>2</sub>;



- 5
- a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

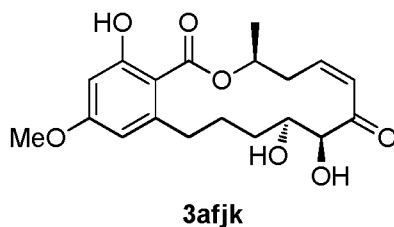


- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, and R2 is methyl, and X-Y is CH=CH;



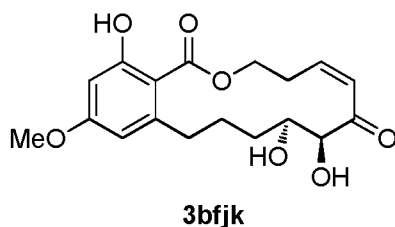
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- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, and R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

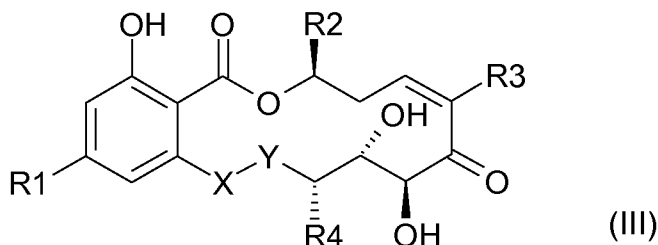


- a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

15



In a second preferred embodiment, the present invention relates to a compound of formula III, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



5        R1, R2, R3, R4, and X-Y being as defined in Formula I.

Preferably, the compound is of Formula III with R2 being selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen and methyl, still more preferably being a methyl.

10        In particular, the compound of Formula III may have one or several (2, 3, 4, 5 or all) of the following features:

- R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen and methyl, still more preferably is a methyl; and/or
- 15        - R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

20        More preferably, the compound is of Formula III may have one or several (2, 3, 4, 5 or all) of the following features:

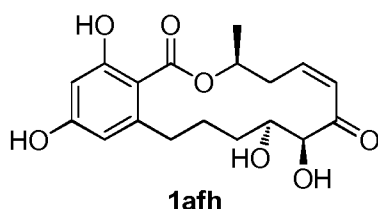
- R1 is hydroxyl; and/or
- R2 is hydrogen or methyl, preferably methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl; and/or
- 25        - X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>.

In a preferred embodiment, compounds of Formula III with R2 being methyl, ethyl and isopropyl, preferably methyl, and with R4 being hydroxyl are contemplated as they present a significant gain of activity.

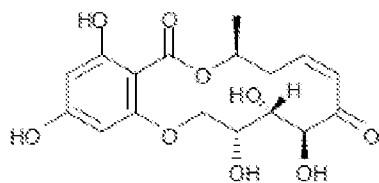
In a particular embodiment of the compounds of Formula III, when R4 is H, then X-Y is preferably not O-CH<sub>2</sub> or CH<sub>2</sub>-S. Preferably, X-Y is CH=CH or CH<sub>2</sub>-CH<sub>2</sub>.

The invention relates in particular to one or several of the following compounds of Formula III:

- a compound of formula III, wherein R1 is hydroxyl, R3 and R4 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

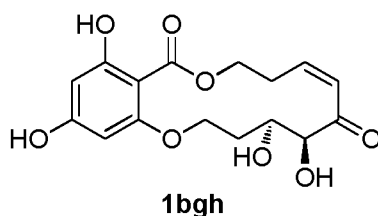


- a compound of formula III, wherein R1 is hydroxyl, R3 is hydrogen, R2 is methyl, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;

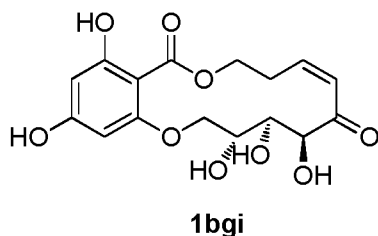


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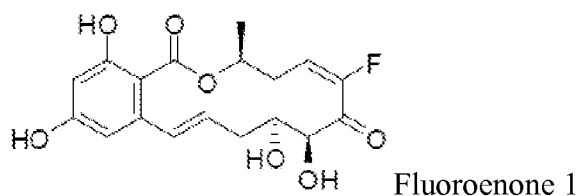
- a compound of formula III, wherein R1 is hydroxyl, R2, R3 and R4 are hydrogen, and X-Y is O-CH<sub>2</sub>;



- a compound of formula III, wherein R1 is hydroxyl, R2 and R3 are hydrogen, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;



- a compound of formula III, wherein R1 is hydroxyl, R4 is hydrogen, R2 is methyl, R3 is fluorine, and X-Y is CH=CH;



The term "pharmaceutically acceptable salt" is used throughout the specification to describe any pharmaceutically acceptable form (such as a salt, an ester, a phosphate ester, salt of an ester or a related group) of a compound which, upon administration to a patient, provides the compound described in the specification. In cases where compounds are sufficiently basic or acidic to form stable nontoxic acid or base salts, administration of the compounds as salts may be appropriate. The term pharmaceutically acceptable salts or complexes refers to salts or complexes that retain the desired biological activity of the compounds of the present invention and exhibit minimal undesired toxicological effects.

No limiting examples of such salts are (a) acid addition salts formed with inorganic acids such as sulfate, nitrate, bicarbonate, and carbonate salts (for example, hydrochloric acid, hydrobromic acid, sulfuric acid, phosphoric acid, nitric acid, and the like), and salts formed with organic acids including tosylate, methanesulfonate, acetate, citrate, malonate, tartarate, succinate, benzoate, ascorbate,  $\alpha$ -ketoglutarate, and  $\alpha$ -glycerophosphate salts, such as acetic acid, oxalic acid, tartaric acid, succinic acid, malic acid, ascorbic acid, benzoic acid, tannic acid, pantoic acid, alginic acid, polyglutamic acid, naphthalenesulfonic acid, naphthalenedisulfonic acid, and polygalacturonic acid, (b) base addition salts formed with metal cations such as zinc, calcium, bismuth, barium, magnesium, aluminum, copper, cobalt, nickel, cadmium, sodium, potassium, lithium and the like, or with a cation formed from ammonia, N,N-dibenzylethylenediamine, D-glucosamine, tetraethylammonium, or ethylenediamine, or (c) combinations of (a) and (b), e.g., a zinc tannate salt or the like. Also included in this definition are pharmaceutically acceptable quaternary salts known by those skilled in the art, which specifically include the quaternary ammonium salt of the formula - NR<sup>+</sup>A<sup>-</sup>, wherein R is as defined above and A is a counterion, including chloride, bromide, iodide, -O-alkyl, toluenesulfonate, methylsulfonate, sulfonate, phosphate, or carboxylate (such as benzoate, succinate, acetate, glycolate, maleate, malate, citrate, tartrate, ascorbate, benzoate, cinnamate, mandelate, benzyloate, and diphenylacetate).

Pharmaceutically acceptable salts may be obtained using standard procedures well known in the art, for example by reacting a sufficiently basic compound such as an amine with a suitable acid affording a physiologically acceptable anion.

Compounds of the present invention have chiral centers and may exist in and be isolated in optically active and racemic forms. The present invention encompasses any racemic, optically-active, diastereomeric, polymorphic, or stereoisomeric form, or mixtures thereof, of a compound of the invention, which possess the useful properties described herein.

5 It is understood that based on the number of asymmetric centers, a total number of  $2^n$  possible stereochemical isomers is possible. The present invention includes all possible stereochemical configurations of the compounds.

In some embodiments, the stereochemistry of the compounds of the invention will retain the natural stereochemistry of the natural resorcylic acid lactone. In other embodiments,  
10 the stereochemical configuration will be different than that found in radicicol A. It will be understood that the stereochemical configuration of any substituent at an asymmetric carbon of the compounds can be in the R or S configuration independent of other substituents at other asymmetric centers in the compound.

In one embodiment, the compounds are prepared in optically active form by  
15 asymmetric synthesis using the processes described herein or synthetic transformations known to those skilled in the art.

Examples of methods to obtain optically active materials are known in the art, and include at least the following: i) physical separation of crystals; ii) simultaneous crystallization; iii) enzymatic resolutions; IV) enzymatic asymmetric synthesis; v) chemical  
20 asymmetric synthesis; vi) diastereomer separations; vii) first- and second-order asymmetric transformations; viii) kinetic resolutions; ix) enantiospecific synthesis from non-racemic precursors; x) chiral liquid chromatography; xi) chiral gas chromatography; xii) extraction with chiral solvents; or xiii) transport across chiral membranes.

Compounds as disclosed herein may be prepared by the one skilled in the art based on  
25 its general knowledge, on the teaching of WO 2008/002497 and on the teaching of the examples of the present application.

### Formulations

The pharmaceutical compositions as disclosed herein may further comprise an additive. The additive may be selected from an anti-oxidant, a buffer, a bacteriostat, a liquid  
30 carrier, a solute, a suspending agent, a thickening agent, a flavoring agent, a gelatin, glycerin, a binder, a lubricant, an inert diluent, a preservative, a surface active agent, a dispersing agent, a biodegradable polymer, or any combination thereof.

The pharmaceutical compositions as disclosed herein may comprise a pharmaceutically acceptable carrier that is suitable for oral, parenteral, inhalation, topical, or intradermal administration. Optionally, compounds can be administered subcutaneously, intravenously, intramuscularly, parenterally, orally, submucosally, by inhalation, transdermally via a slow release patch, or topically, in an effective dosage range to treat cancer.

The pharmaceutical compositions may further comprise an additional active ingredient. Preferably, the additional active ingredient is an antitumoral drug.

### Uses

The present invention relates to a compound of Formula (I), (II), (IIa), (Iib) or (III) or any particular compound disclosed herein as a drug. In particular, the compound is for use for treating cancer. It also relates to the use of the compound for the preparation of a medicament for treating cancer. It finally relates to a method for treating cancer in a subject, comprising administering a therapeutic effective amount of the compound.

In a specific embodiment, the compound is selected from the group consisting of:

- a compound of formula III, wherein R1 is hydroxyl, R3 and R4 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R3 is hydrogen, R2 is methyl, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R2, R3 and R4 are hydrogen, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R2 and R3 are hydrogen, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R4 is hydrogen, R2 is methyl, R3 is fluorine, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and,
- 5      - a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

Preferably, the compound is selected from the group consisting of:

- a compound of formula III, wherein R1 is hydroxyl, R3 and R4 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- 10      - a compound of formula III, wherein R1 is hydroxyl, R3 is hydrogen, R2 is methyl, R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- 15      - a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, and R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and
- 20      - a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

In a particularly preferred embodiment, the present invention relates to any of the above-disclosed compounds of Formula (II) for use in the treatment of cancer while preventing metastasis occurrence, to the use of any of the above-disclosed compounds for the preparation of a medicament for treating cancer while preventing metastasis occurrence, and to a method for treating cancer while preventing metastasis occurrence in a subject, comprising administering a therapeutic effective amount of any of the above-disclosed compounds to said subject.

In a specific embodiment, the compound is selected from the group consisting of:

- 30      - a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;

- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- 5     - a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and,
- a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

More preferably, the compound is selected from the group consisting of:

- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- 15     - a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, and R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and
- 20     - a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

Within the context of the invention, the term treatment denotes curative, symptomatic, and preventive treatment. Pharmaceutical compositions and preparations of the invention can be used in humans with existing cancer or tumor, including at early or late stages of progression of the cancer. The pharmaceutical compositions and preparations of the invention will not necessarily cure the patient who has the cancer but will delay or slow the progression or prevent further progression of the disease, ameliorating thereby the patients' condition. In particular, the pharmaceutical compositions and preparations of the invention reduce the development of tumors, reduce tumor burden, produce tumor regression in a mammalian host and/or prevent metastasis occurrence and cancer relapse. In treating the cancer, the pharmaceutical composition of the invention is administered in a therapeutically effective amount.

By "effective amount" it is meant the quantity of the pharmaceutical composition of the invention which prevents, removes or reduces the deleterious effects of cancer in



mammals, including humans. It is understood that the administered dose may be adapted by those skilled in the art according to the patient, the pathology, the mode of administration, etc.

An effective dose can be readily determined by the use of conventional techniques and by observing results obtained under analogous circumstances. In determining the effective  
5 dose, a number of factors are considered including, but not limited to: the species of patient; its size, age, and general health; the specific disease involved; the degree of involvement or the severity of the disease; the response of the individual patient; the particular compound administered; the mode of administration; the bioavailability characteristics of the preparation administered; the dose regimen selected; and the use of concomitant medication.

10 The cancer may be a solid cancer or a hematopoietic cancer. The terms "cancer", "cancerous", or "malignant" refer to or describe the physiological condition in mammals that is typically characterized by unregulated cell growth. Examples of cancer include, for example, leukemia, lymphoma, blastoma, carcinoma and sarcoma. More particular examples of such cancers include chronic myeloid leukemia, acute lymphoblastic leukemia,  
15 Philadelphia chromosome positive acute lymphoblastic leukemia (Ph<sup>+</sup> ALL), squamous cell carcinoma, small-cell lung cancer, non-small cell lung cancer, glioma, gastrointestinal cancer, renal cancer, ovarian cancer, liver cancer, colorectal cancer, endometrial cancer, kidney cancer, prostate cancer, thyroid cancer, neuroblastoma, pancreatic cancer, glioblastoma multiforme, cervical cancer, stomach cancer, bladder cancer, hepatoma, breast cancer, colon  
20 carcinoma, and head and neck cancer, gastric cancer, germ cell tumor, pediatric sarcoma, sinonasal natural killer, multiple myeloma, acute myelogenous leukemia (AML), chronic lymphocytic leukemia, mastocytosis and any symptom associated with mastocytosis.

The compounds of the invention may be also administered in specific, measured amounts in the form of an aqueous suspension by use of a pump spray bottle. The aqueous  
25 suspension compositions of the present invention may be prepared by admixing the compounds with water and other pharmaceutically acceptable excipients. The aqueous suspension compositions according to the present invention may contain, inter alia, water, auxiliaries and/or one or more of the excipients, such as: suspending agents, e.g., microcrystalline cellulose, sodium carboxymethylcellulose, hydroxpropyl-methyl cellulose;  
30 humectants, e.g. glycenn and propylene glycol, acids, bases or buffer substances for adjusting the pH, e g , citric acid, sodium citrate, phosphoric acid, sodium phosphate as well as mixtures of citrate and phosphate buffers; surfactants, e.g. Polysorbate 80, and antimicrobial preservatives, e.g., benzalkonium chloride, phenylethyl alcohol and potassium sorbate.

Typical systemic dosages for all of the herein described conditions are those ranging from 0.01 mg/kg to 1500 mg/kg of body weight per day as a single daily dose or divided daily doses. Preferred dosages for the described conditions range from 0.5-1500 mg per day. A more particularly preferred dosage for the desired conditions ranges from 5-750 mg per day.

5 Typical dosages can also range from 0.01 to 1500, 0.02 to 1000, 0.2 to 500, 0.02 to 200, 0.05 to 100, 0.05 to 50, 0.075 to 50, 0.1 to 50, 0.5 to 50, 1 to 50, 2 to 50, 5 to 50, 10 to 50, 25 to 50, 25 to 75, 25 to 100, 100 to 150, or 150 or more mg/kg/day, as a single daily dose or divided daily doses. In one embodiment, the compounds are given in doses of between about 1 to about 5, about 5 to about 10, about 10 to about 25 or about 25 to about 50 mg/kg. Typical  
10 dosages for topical application are those ranging from 0.001 to 100% by weight of the active compound

The compounds are conveniently administered in units of any suitable dosage form, including but not limited to one containing from about 7 to 3000 mg, from about 70 to 1400 mg, or from about 25 to 1000 mg of active ingredient per unit dosage form. For example, an  
15 oral dosage of from about 50 to 1000 mg is usually convenient, including in one or multiple dosage forms of 50, 100, 200, 250, 300, 400, 500, 600, 700, 800, 900 or 1000 mgs. Lower dosages may be preferable, for example, from about 10-100 or 1-50 mgs. Also contemplated are doses of 0.1-50 mg, 0.1-20 mgs, or 0.1-10 mgs. Furthermore, lower doses may be utilized in the case of administration by a non-oral route, as for example, by injection or inhalation.

20 Pharmaceutically acceptable carriers that may be used in these pharmaceutical compositions are generally known in the art. They include, but are not limited to, ion exchangers, alumina, aluminum stearate, lecithin, serum proteins, such as human serum albumin, buffer substances such as phosphates, glycine, sorbic acid, potassium sorbate, partial glyceride mixtures of saturated vegetable fatty acids, water, solvents, salts or electrolytes,  
25 such as protamine sulfate, disodium hydrogen phosphate, potassium hydrogen phosphate, sodium chloride, zinc salts, silicates, colloidal silica, magnesium trisilicate, polyvinyl pyrrolidone, cellulose-based substances, polyethylene glycol, sodium carboxymethylcellulose, polyacrylates, waxes, oils, polyethylene-polyoxypropylene-block polymers, polyethylene glycol and wool fat. Pharmaceutically accepted vehicles can contain mixtures of more than  
30 one excipient in which the components and the ratios can be selected to optimize desired characteristics of the formulation including but not limited to shelf-life, stability, drug load, site of delivery, dissolution rate, self-emulsification, control of release rate and site of release, and metabolism.

Formulations can be prepared by a variety of techniques known in the art. Examples of formulation techniques can be found in literature publications and in texts such as "Water-insoluble drug formulation", edited by Rong Liu, 2000, Interpharm Press

If administered intravenously, carriers can be physiological saline, bacteriostatic water, Cremophor EL™ (BASF, Parsippany, NJ) or phosphate buffered saline (PBS). Sterile injectable forms of the compositions of this invention may be aqueous or oleaginous suspension. These suspensions may be formulated according to techniques known in the art using suitable dispersing or wetting agents and suspending agents. The sterile injectable preparation may also be a sterile injectable solution or suspension in a non-toxic parenterally-acceptable diluent or solvent, for example as a solution in 1,3-butanediol. Among the acceptable vehicles and solvents that may be employed are water, Ringer's solution and isotonic sodium chloride solution. In addition, sterile, fixed oils are conventionally employed as a solvent or suspending medium. For this purpose, any bland fixed oil may be employed including synthetic mono- or di-glycerides. Fatty acids, such as oleic acid and its glyceride derivatives are useful in the preparation of injectables, as are natural pharmaceutically-acceptable oils, such as olive oil or castor oil, especially in their polyoxyethylated versions. These oil solutions or suspensions may also contain a long-chain alcohol diluent or dispersant, such as carboxymethyl cellulose or similar dispersing agents which are commonly used in the formulation of pharmaceutically acceptable dosage forms including emulsions and suspensions. Other commonly used surfactants, such as Tweens, Spans and other surface-active emulsifying agents or bioavailability enhancers which are commonly used in the manufacture of pharmaceutically acceptable solid, liquid, or other dosage forms may also be used for the purposes of formulation.

One mode of administration of the active compound for systemic delivery is oral. Oral compositions will generally include an inert diluent or an edible carrier. They may be enclosed in gelatin capsules or compressed into tablets. For the purpose of oral therapeutic administration, the active compound can be incorporated with excipients and used in the form of tablets, troches, or capsules. Pharmaceutically compatible binding agents, and/or adjuvant materials can be included as part of the composition. The tablets, pills, capsules, troches and the like can contain any of the following ingredients, or compounds of a similar nature: a binder such as microcrystalline cellulose, gum tragacanth or gelatin; an excipient such as starch or lactose, a disintegrating agent such as alginic acid, Primogel, or corn starch; a lubricant such as magnesium stearate or Sterotes; a glidant such as colloidal silicon dioxide; a

sweetening agent such as sucrose or saccharin; or a flavoring agent such as peppermint, methyl salicylate, or orange flavoring.

When the dosage unit form is a capsule, it can contain, in addition to material of the above type, a liquid carrier such as a fatty oil. In addition, dosage unit forms can contain various other materials which modify the physical form of the dosage unit, for example, coatings of sugar, shellac, or other enteric agents. The compound or its salts can be administered as a component of an elixir, suspension, syrup, wafer, chewing gum or the like. A syrup may contain, in addition to the active compounds, sucrose as a sweetening agent and certain preservatives, dyes and colorings and flavors.

In a preferred embodiment, the active compounds are prepared with carriers that will protect the compound against rapid elimination from the body, such as a controlled release formulation, including implants and microencapsulated delivery systems. Biodegradable, biocompatible polymers can be used, such as ethylene vinyl acetate, polyanhydrides, polyglycolic acid, collagen, polyorthoesters, and polylactic acid. Methods for preparation of such formulations will be apparent to those skilled in the art. The materials can also be obtained commercially from Alza Corporation and Nova Pharmaceuticals, Inc. Liposomal suspensions (including liposomes targeted to infected cells with monoclonal antibodies to viral antigens) are also preferred as pharmaceutically acceptable carriers. These may be prepared according to methods known to those skilled in the art, for example, as described in U S Patent No. 4,522,811 (which is incorporated herein by reference in its entirety). For example, liposome formulations may be prepared by dissolving appropriate lipid(s) (such as stearyl phosphatidyl ethanolamine, stearyl phosphatidyl choline, arachidoyl phosphatidyl choline, and cholesterol) in an inorganic solvent that is then evaporated, leaving behind a thin film of dried lipid on the surface of the container. An aqueous solution of the compound is then introduced into the container. The container is then swirled by hand to free lipid material from the sides of the container and to disperse lipid aggregates, thereby forming the liposomal suspension.

Suitable vehicles or carriers for topical application can be prepared by conventional techniques, such as lotions, suspensions, ointments, creams, gels, tinctures, sprays, powders, pastes, slow-release transdermal patches, suppositories for application to rectal, vaginal, nasal or oral mucosa. In addition to the other materials listed above for systemic administration, thickening agents, emollients, and stabilizers can be used to prepare topical compositions. Examples of thickening agents include petrolatum, beeswax, xanthan gum, or polyethylene,

humectants such as sorbitol, emollients such as mineral oil, lanolin and its derivatives, or squalene.

The pharmaceutical composition as disclosed herein may be used in combination with chemotherapy, radiotherapy, hormonal therapy, immunotherapy, or monoclonal antibody  
5 therapy.

Chemotherapy refers to a cancer therapeutic treatment using chemical or biochemical substances, in particular using one or several antineoplastic agents.

Radiotherapy includes, but is not limited to,  $\gamma$ -rays, X-rays, and/or the directed delivery of radioisotopes to tumor cells. Other radiotherapies include microwaves and UV-  
10 irradiation. Other approaches to radiation therapy are also contemplated in the present invention.

The term “immunotherapy” refers to a cancer therapeutic treatment using the immune system to reject cancer. The therapeutic treatment stimulates the patient's immune system to attack the malignant tumor cells. It includes immunization of the patient with tumoral  
15 antigens (eg. by administering a cancer vaccine), in which case the patient's own immune system is trained to recognize tumor cells as targets to be destroyed, or administration of molecules stimulating the immune system such as cytokines, or administration of therapeutic antibodies as drugs, in which case the patient's immune system is recruited to destroy tumor cells by the therapeutic antibodies. In particular, antibodies are directed against specific  
20 antigens such as the unusual antigens that are presented on the surfaces of tumors. As illustrating example, one can cite Trastuzumab or Herceptin antibody which is directed against HER2 and approved by FDA for treating breast cancer.

The term “monoclonal antibody therapy” refers to any antibody that functions to deplete tumor cells in a patient. In particular, therapeutic antibodies specifically bind to  
25 antigens present on the surface of the tumor cells, e.g. tumor specific antigens present predominantly or exclusively on tumor cells. Alternatively, therapeutic antibodies may also prevent tumor growth by blocking specific cell receptors.

The term “hormone therapy” or “hormonal therapy” refers to a cancer treatment having for purpose to block, add or remove hormones. For instance, in breast cancer, the  
30 female hormones estrogen and progesterone can promote the growth of some breast cancer cells. So in these patients, hormone therapy is given to block estrogen and a non-exhaustive list commonly used drugs includes: Tamoxifen, Fareston, Arimidex, Aromasin, Femara, Zoladex/Lupron, Megace, and Halotestin.

## Examples

### Example 1

A library of resorcylic acid lactones (RAL) containing a *cis*-enone moiety targeting kinases bearing a cysteine residue within the ATP-binding pocket was prepared and evaluated against a panel of 19 kinases thus providing important structure-activity trends. Two new analogues were then profiled for their selectivity against a panel of 402 kinases providing the broadest evaluation of this pharmacophores' selectivity.

We report here a library of over 50 analogues based on this resorcylic acid scaffold and the activity of a representative subset of 31 members was evaluated against a panel of 19 kinases. Two analogues were further evaluated against a panel of 402 kinases which includes 359 distinct kinases and 43 mutants.

### Results and Discussion

The synthetic planning of the library was based on the previously developed chemistry (P. Y. Dakas, et al, *Angew. Chem. Int. Ed. Engl.* **2007**, *46*, 6899-6902; P. Y. Dakas, et al, *Chemistry*. **2009** *15*, 11490-7) leveraged on the use of fluoruous tags to facilitate isolation of reaction product and to carry mixtures of products through a common synthetic pathway (D. P. Curran, *Handbook of Fluorous Chemistry* **2004**, 101-127; Z. Luo, et al, *Science* **2001**, *291*, 1766-1769; W. Zhang, D. P. Curran, *Tetrahedron* **2006**, *62*, 11837-11865). Thus a library of the general structure **1** (Figure 2) was envisioned to emanate from the coupling of the key fragment **5** bearing a fluoruous tag encoding its structure and the different aromatic moieties **4** bearing either the selenoether or methyl or hydroxyl at position Y to obtain respectively an alkene, an alkane or phenolic ether at position X in the library **1**. While we have shown that this coupling chemistry could be carried out on solid phase using a thioether linker at position Y of intermediate **4** to access the alkane and alkene functionality (P. Y. Dakas, et al, *Chemistry*. **2009** *15*, 11490-7), this strategy precluded the formation of phenolic ethers which were deemed interesting and thus motivated the exclusive use of the fluoruous isolation technology for the purpose of the present library. Fragment **5** was foreseen to come from the coupling of fragment **6** and **7**, fragment **6** being either an alkyne which would be reduced to the *cis* alkene with Lindlar catalyst or a *cis*-vinyl iodide. Of course, the general library **1** could be further elaborated by epoxidation of the benzylic alkene to obtain hypothemycin analogues **2** or by methylation of the more acidic phenol and/or oxime formation on the ketone (**3**). Our choice of the fragments R<sup>1</sup>, R<sup>2</sup> and aryl substitutions was based on our preliminary structure

activity data coupled to reported structure activity obtained by semi-synthesis of hypothemycin (B. R. Hearn, et al., *ChemMedChem* **2007**, 2, 1598-1600) and available structural information (M. Ohori, et al., *Biochem. Biophys. Res. Commun.* **2007**, 353, 633-637; G. Rastelli, et al., *J. Struct. Biol.* **2008**, 164, 18-23). Clearly, the nature of the functionality at the benzylic position appeared to be important in dictating the selectivity based on the afore mentioned difference in activity between LL-Z1640-2 bearing an alkene and L-783277 bearing an alkane for ERK2 (M. Ohori, et al., *Biochem. Biophys. Res. Commun.* **2007**, 353, 633-637) and MEK1 (A. Zhao, et al., *J. Antibio.* **1999**, 52, 1086-1094; J. Ninomiya-Tsuji, et al., *J. Biol. Chem.* **2003**, 278, 18485-18490). While this modification at the benzylic position may seem rather modest, preliminary modeling experiments have suggested fairly different conformational landscape for both compounds. We thus felt important to include both of these functionalities in the library. While the epoxide at the benzylic position imparts a similar conformation to an alkene, difference in dipoles may be significant. Alternatively, it appeared that a phenolic ether at that position should also provide significant difference providing a conformational profile more similar to the alkane but with different dipole moments. While structural information could rule out certain modifications, it should be noted that the crystal structures reported (M. Ohori, et al., *Biochem. Biophys. Res. Commun.* **2007**, 353, 633-637; G. Rastelli, et al., *J. Struct. Biol.* **2008**, 164, 18-23) are for the Michael adduct and the initial recognition event between the protein and the *cis*-enone may involve different macrocyclic and/ or kinase conformations. The modifications which were envisioned for the library are shown in Figure 2 with four different fragments for the ester moiety R<sup>1</sup> (**a-d**), three different functionalities at position X (**e-g**) and three different fragments for the lower part of the macrocycle R<sup>2</sup> (**h-j**). The chiral methyl substituent on the ester points towards the surface of  $\beta$ -pleated sheets. Our prior investigation had shown that the *R*-chirality at that center dramatically attenuated activity, nevertheless, compounds lacking that methyl group (fragment **b**) may broaden specificity. An extra hydroxyl group at the adjacent position (fragment **c**) was deemed interesting as it would probe whether additional interactions may be achieved and whether the chiral substituent imparts an important conformational bias on the macrocycle. The same reasoning was applied to the addition of a hydroxyl group on the lower part of the macrocycle (R<sup>2</sup>, fragment **i**). Modification of the alkene (fragment **d**) was seen as crucial in modulating the rate of conjugate addition (if the conjugate addition is catalyzed by a hydrogen bond to the carbonyl, the greater stability of a tertiary cation *vs.* a secondary cation may favored the reaction). Furthermore, having a methyl substituent at that position could alleviate the issues related to *cis/trans* isomerization. It is

known that the *trans*-enones are significantly less active than the *cis*-enone (A. Zhao, et al., *J. Antibio.* **1999**, 52, 1086-1094). The homoallylic diol appeared to be important for biological activity and it was already known that methylation of either led to significant reduction in activity (B. R. Hearn, et al., *ChemMedChem* **2007**, 2, 1598-1600). This functionality was thus kept constant in fragments **h-j**. Our preliminary investigations had shown that compounds lacking the diol were indeed inactive as well as acyclic analogues. However, a larger macrocycle appeared interesting in providing subtle changes in conformation profiles and could be probed with fragment **j**. While each of these modifications can be rationalized as providing a potential benefit, the combination of several modifications may provide unanticipated synergistic benefits.

As shown in Figure 3, fragments **6a-d** were obtained in one to five steps via well established chemistry. The first step to obtain **6a-c** involved a protection of the alcohols **8**, **10** and **11** with the trichloroacetimidate of the fluororous-PMB (**9**) bearing different length of the fluororous tag encoding the structure of the starting alcohol. For **10** and **11**, the alcohol protection was followed by a Dibal reduction and a Corey-Fuchs reaction or a Grignard addition of acetylene (2:1 dr) followed by EOM protection to afford **6a** and **6c** respectively. Compound **6d** was prepared from the racemic 4-hydroxypentyne using a known procedure (S. Ma, E.-i. Negishi, *J. Org. Chem.* **1997**, 62, 784-785.) to access the *Z*-vinyl iodide **13** which was protected with the fluororous tagged PMB. The product of each reaction was isolated by flash chromatography on fluororous silica gel. In general, a 10 to 20 fold ratio of silica to crude product weight was used for the isolation. The elutions were carried out systematically using a three step gradient (7:3 MeOH:H<sub>2</sub>O; 8:2 MeOH:H<sub>2</sub>O and pure MeOH) and the product was collected from the MeOH fraction without further attempt to optimize individual isolation or recovered product from mixed fractions.

The synthesis of fragments **7h-j** (Figure 4) was leveraged on the naturally abundant chirality of deoxyribose and lyxose using well established methodology. Thus, D-deoxyribose was selectively protected with an acetonide (J. Barbat, et al., *Carbohydr. Res.* **1983**, 116, 312-316) and reduced with LiAlH<sub>4</sub> to obtain diol **14** which was selectively protected on the less hindered alcohol with TBDPS-Cl and oxidized to aldehyde **7h**. Starting with the same selective protection of ribose but engaging the resulting lactol in a Wittig olefination (X. Geng, S. J. Danishefsky, *Org. Lett.* **2004**, 6, 413-416) rather than a reduction afforded alkene **15** after pivaloylation of the primary alcohol. A sequence involving a hydroboration, silyl protection, pivaloyl deprotection and oxidation afforded fragment **7j** which has an additional carbon compared to **7h**. The third fragment emanated from D-lyxose by protection of the



anomeric position followed by selective acetonide protection of the *cis*-diol, and EOM-protection of the remaining hydroxyl group. Deprotection of the anomeric position and conversion of the lactol to an alkene allowed protection of the unique hydroxyl group with TBDPS-Cl followed by ozonolysis of the alkene to obtain fragment **7i**. While a number of  
5 alternative strategies can be envisioned for each of these fragments, these procedures were found to be inexpensive, reliable and scalable.

The synthesis of key fragments **5** is shown in Figure 5. Starting with a mixture of three fluororous tagged alkynes **6** which were deprotonated with *n*Bu-Li and added to three vessels containing the different aldehydes **7** afforded three pools of **17**, each containing three  
10 compounds labeled with a unique fluororous tag. Benzoyl protection of the alcohol followed by desilylation, hydrogenation over Lindlar catalyst and iodination afforded the three pools of three compounds **5a-c**. All reactions were monitored by LC-MS and showed greater than 90% conversion. While the isolated yield of the product following fluororous isolation was not optimized, acceptable yields were generally obtained. The vinyl iodide **6d** was transmetalated  
15 with *t*BuLi and similarly added to three different pools of aldehydes **7**. A similar sequence of benzoylation, desilylation, and iodination afforded three pools of **5d**.

The aromatic fragments were obtained as shown in Figure 6. The fragment **4g** was obtained starting from 2,4,6-trihydroxybenzoic acid **18** by protection of the acid and *ortho* phenol with an acetonide (R. G. Dushin, S. J. Danishefsky, *J. Am. Chem. Soc.* **1992**, *114*, 655-  
20 659) followed by protection of the two remaining phenols with EOM-Cl to obtain **19** which was treated with the alkoxide of trimethylsilyl ethanol to yield **4g**. Compound **4f** was converted to **4e** by deprotonation with LDA and reaction with diphenyl diselenide.

The library synthesis is shown in Figure 7. Thus each aromatic fragment **4** was coupled to 4 pools of fragment **5**. Analysis of the reaction mixtures by LC-MS indicated good  
25 to excellent conversion for all reactions, with the phenol couplings being systematically the highest yielding followed by the selenoether and alkyl couplings. Amongst the different pools of electrophiles, the pool containing fragment *i* with an additional EOM-protected hydroxyl adjacent to the iodide being displaced gave the lowest coupling efficiency. However, the desired coupled product was obtained in all cases. Each pool was demixed to resolve the  
30 individual components which were detagged (DDQ) and deprotected (TBAF) to be engaged in a macrolactonization. We had previously observed that these reactions could be performed at fairly high concentration (0.1 M) without dimerization or oligomerization. Treatment of each compound with fluororous-tagged PPh<sub>3</sub> and fluororous-tagged DIAD afford high yield in the lactonization (>85%) except for compounds containing the fragment *c* bearing a sterically

more demanding alcohol for which 50% conversion was achieved. The allylic benzoate was removed in excellent yield for all macrocycles. We had previously noted that a selective allylic oxidation could be achieved in some cases but was unfortunately not general and depended on the relative geometry of the allylic alcohol (P. Y. Dakas, et al, *Chemistry*. **2009** 15, 11490-7). In the cases of the macrocycles bearing an ether (X = O), the selective oxidation of the fully deprotected macrocycle was reliably achieved with polymer-bound IBX. On the other hand, for the oxidation of the macrocycles containing an alkane or alkene at the benzylic position, the oxidation was performed prior to deprotection using Dess-Martin periodinane in refluxing CH<sub>2</sub>Cl<sub>2</sub> (the oxidation of one of the diastereoisomer is slower than the deprotected macrocycle and requires more stringent conditions). Removal of the acetonide and EOM using aqueous HF afforded the desired macrocycles in moderate to good yields. It should be noted that the reaction was never allowed to reach completion as prolonged reaction time led to product decomposition. Macrocycles **1ef** bearing an EOM group on the *para*-phenol were also isolated from most reactions. All compounds were purified by preparative TLC. The macrocycles **1e** could be further elaborated by epoxidation of the benzylic position using DMDO (P. Y. Dakas, et al, *Chemistry*. **2009** 15, 11490-7). However, the lability of the benzylic epoxide makes isolation of the product challenging and pharmacologically undesirable. Macrocycles could be also further derivatized by selective methylation of the *para*-phenol with diazomethane. While the reaction was quite selective for the methylation of the *para*-phenol (**3k** and **1**, Figure 8), to ensure completion of the reaction, these were performed with a large excess of diazomethane leading to the isolation of 10-40% of the bis-methylated product (not shown). The ketone could be converted to a methyl or benzyl oxime in high yield under standard conditions. In total, 51 macrocycles were isolated from these efforts.

The IC<sub>50</sub> of a subset of the library (28, containing at least one example of each modifications) was assayed against a panel of kinases (19) as representative of kinases bearing the adequately positioned cysteine residue (VGFR-R1-3, PDGF-R $\alpha$ , FLT3, MEK1, KIT, GSK $\alpha$ , MAP KAPK5) kinases bearing a cysteine residue at a different position within the ATP binding pocket (EGFR-3, JNK3, NEK2, NIK, SRC, ZAP70) and kinases which do not bear a cysteine residue (PKC $\alpha$ , CK2 $\alpha$ , INS-R). It should be noted that as these compounds are expected to be irreversible inhibitors, the measured IC<sub>50</sub> will depend on the procedure used and caution should be applied in comparing data from experiments performed under different conditions. To obtain a value which best reflects the rate of inactivation, the substrate and inhibitor were added simultaneously to the kinase solution. The data is shown in

Figure 9 (kinases have been ranked from the highest to least inhibited). As evident from Figure 9, there is very little difference in relative selectivity of kinase inhibition throughout the library under these conditions. VEGF-R2 is the most highly inhibited kinase in this panel for all library members followed by PDGFR- $\alpha$ , VEGF-R3, Flt3, VEGF-R1, MEK1 SESE (which is a constitutively active form of MEK) and KIT. Overall, the trend of structure activity relationship which emerged from this screen is that the chiral methyl group at the ester position, while not essential, does provide higher activity (entry 6 vs. 10, 7 vs. 11 and 8 vs. 12). Substitution at the adjacent position with a hydroxyl group (entry 14 vs. 7) or at the  $\beta$ -position of the enone with a methyl abrogated activity (15-19). Position X of library 1 (Figure 2), as suggested from the diverse natural products, is tolerant of modification and, an oxygen is also tolerated at that position. The trend of activity for the three natural products can be ranked as follows: alkene = epoxide >alkane. The ether in the macrocycle is generally comparable to the alkane. The two modifications of R<sup>2</sup> tested were beneficial in certain combinations. For example, the extra hydroxyl group present in fragment **i** does afford a significant (10 fold) gain of activity (8 vs. 7) in combination with fragment **a**. The same beneficial effect was not observed in the presence of fragment **b** (entry 12 vs. 11). The larger macrocycle (fragment **j**) was particularly beneficial in conjunction with the alkane moiety at the benzylic position (**f**). For example, entry 6, 10 and 25 are amongst the most potent compounds tested whereas a similar gain is not observed for X = CH or oxygen (entry 4, 9, 13). Generally, methylation of the phenol *para* to the ester had marginal changes in activity while oxime formation abolished activity. Interestingly, none of the alcohols **23** lacking the ketone showed significant inhibition attesting to the fairly low affinity of the macrocycle for the ATP-binding pocket and the importance of the Michael acceptor.

Two representative library members (**1afh** and **1bgi**) were then evaluated in a larger panel of kinases (402) using the technology described by Lockhart et al. (M. A. Fabian, et al., *Nat. Biotechnol.* **2005**, 23, 329-336; M. W. Karaman, et al., *Nat. Biotechnol.* **2008**, 26, 127-132). In this panel, 31 kinases out of the 46 kinases bioinformatically identified by Santi and coworkers are present along with numerous other kinases bearing a cysteine residue at other positions within the ATP-binding pocket (J. Zhang, P. L. Yang, N. S. Gray, *Nat. Rev. Cancer* **2009**, 9, 28-39). The two compounds were screened at 1  $\mu$ M (i.e. well above their IC<sub>50</sub> concentration for the most potently inhibited kinases) to have a realistic perspective of their selectivity. Interestingly, only two kinase outside of the set of 46 kinases predicted by Santi *et al* showed significant inhibition (STK36 and PRKD2) attesting to the potential of exploiting cysteine residues within the ATP-binding pocket as selectivity filters (M. S. Cohen, et al.,

*Science* **2005**, 308, 1318-1321). As shown in Figure 10, 11 kinases out of 31 had more than 50% residual activity given this high concentration of **1afh** (MEK4 had 0% residual activity at this concentration). The kinases which showed less than 10% residual activity relatively to control were: Flt1&3; GAK; KIT, MEK1,2,4; MKNK 1&2; PDGFR $\alpha$ &B; TAK1, TGFR2.

5 Interestingly, some kinases in this group (CDKL5, ERK1, GSK3 $\beta$ , NIK, NLK, PRP4) were not inhibited. For **1bgi**, 19 out of the 31 kinases had over 50% residual activity and MEK4 was also the most highly inhibited kinases having 1% residual activity. While **1bgi** was overall less potent, it does appear to show subtle differences in reactivity compared to **1afh**. For example, while both compounds have similar activity against GAK, **1bgi** is ten times less  
10 active against Flt1 and PDGFR $\beta$ .

The two compounds were also evaluated against a series of mutations of Flt3 and KIT. It is known that several mutation in Flt3 lead to a gain of function or constitutive activity. An internal tandem duplication (ITD) in Flt3 (M. Nakao, et al., *Leukemia* **1996**, 10, 1911-1918) is found in 20% of patients with acute myeloid leukemia (AML) while the most abundant  
15 mutation leading to a gain of function was found to be D835Y for which both compounds (**1afh** and **1bgi**, Figure 11) are more active than against the wild type (Y. Yamamoto, et al., *Blood* **2001**, 97, 2434-2439). Other mutants (K663Q (M. Schittenhelm, et al., *Leukemia* **2006**, 20, 2008-2014) and N841L (J. Jiang, et al., *Blood* **2004**, 104, 1855-1858)) leading to a gain of function remained equally sensitive to the inhibitors. Likewise, the compound retained their  
20 activity against KIT mutants which results in a gain of function (D816V (C. Willmore-Payne, et al., *Modern Pathology* **2006**, 19, 1164-1169) L576P (C. Willmore-Payne, et al., *Modern Pathology* **2006**, 19, 1164-1169) and V559D (S. Hirota, et al., *Science* **1998**, 279, 577-580)) however, these compounds are not active against the dual mutations which confers resistance to imatinib (V559D, T670I; V559D, V654A) (T. Carter, et al., *Proc. Natl. Acad. Sci. USA*  
25 **2005**, 102, 11011-11016).

From a therapeutic perspective, the kinases inhibited by the *cis*-enone resorcylic acids contribute to the development, progression and aggressiveness of cancer. More recent in depth characterization of the specificity of kinase inhibitors have shown that all small molecule kinase inhibitors approved for therapeutic intervention or in clinical development do  
30 inhibit multiple targets (M. W. Karaman, et al., *Nat. Biotechnol.* **2008**, 26, 127-132). The pallet of kinases inhibited by the *cis*-enone resorcylic acids all drive tumor growth. The fact that none of the compounds lacking the enone moiety fail to show significant activity testifies to the importance of the Michael acceptor as a selectivity filter. An important question is whether a mutation in the cysteine residue which is so important for the activity would be

viable. However, based on the number of kinases inhibited along the MAP kinase cascade, a concerted evolution of several kinases would be necessary to evade the growth inhibition of these resorcylic acid lactones. The irreversible nature of the inhibition may prove to be important in achieving long lasting inactivation of specific pathways. The EGFR receptor has also been the target of irreversible inhibitors exploiting a cysteine residue. Results from phase I clinical trial have shown that EKB-569 is safe (C. Erlichman, et al., *J. Clin. Oncol.* **2006**, *24*, 2252-2260). From a chemical biology standpoint, compounds which form covalent bonds to their target are particularly useful as they can be used to label the targeted kinases for activity-based profiling and imaging. It is interesting to note how frequently evolution has resorted to irreversible inhibition in the selection of bioactive secondary metabolites (C. Drahl, B. F. Cravatt, E. J. Sorensen, *Angew. Chem. Int. Ed. Engl.* **2005**, *44*, 5788-5809). It should also be noted that irreversible inhibition of kinases leveraged on a conjugate addition is an endogenous mechanism as has been demonstrated for the inhibition of I $\kappa$ B by cyclopentenone prostaglandins (A. Rossi, et al., *Nature* **2000**, *403*, 103-108).

Two analogs bearing an oxygen in the macrocycle and a larger macrocyclic ring (1agh, 1aej) were evaluated for their efficacy in cell using a renal carcinoma line (RENCA) and in an orthopedic mice renal carcinoma. As shown in Figure 12, the *cis*-enone inhibitor performed better than known inhibitor having an IC<sub>50</sub> in the 100 to 500 nM. Evaluation of these analogs in mice showed a clear reduction in tumor growth at 20mg/kg for aej and at 20 and 10 mg/kg for agh (Figure 15).

In conclusion, this library offers important structural activity relationship in the *cis*-enone resorcylic acids. Two modifications were found to independently and synergistically improve the activity of this series of compounds while a modification which dramatically simplifies the synthetic accessibility of these compounds was established. The fact that several members of this family are potent inhibitors of multiple notorious oncogenic kinases and effective in cell as well as in whole animals should heighten their therapeutic interest.

## Example 2

Hypothemycin and related resorcylic acid lactones (RAL) bearing a *cis*-enone moiety have emerged as an alternative pharmacophore to heterocyclic motifs with a unique selectivity filter based on the irreversible reaction with a subset of the kinome bearing a suitably positioned cysteine residue. Two prototypical examples of "edited" RAL were evaluated for antitumoral, antimetastatic and antiangiogenic efficacy in an orthotopic murine

renal cell carcinoma (RENCA) model. While both compounds were good inhibitors of VEGFRs in vitro, and inhibited tumor growth in vivo with comparable efficacy to sunitinib, an FDA-approved VEGFRs inhibitor, one compound did promote lung metastasis to a similar extent as sunitinib while the other strongly inhibited lung metastasis. This study attests to the potential of irreversible kinase inhibitors and molecular editing of natural pharmacophores and provides encouraging results to a clinically significant problem.

## Material and method

### Cell proliferation assay.

Murine RENCA cells were originally obtained from a tumor that arose spontaneously in the kidney of BALB/c mice. Histologically, RENCA consists of granular cell type adenocarcinoma, which is pleomorphic with large nuclei. To obtain RENCA-LN, RENCA cells were transduced using a retrovirus encoding a luciferase-neomycin fusion protein and selected using 1 mg/ml G418. RENCA-LN cells were plated with 1.500 cells/150  $\mu$ l/well in DMEM supplemented with 10% FCS in 96 well culture dishes. The following day the serial diluted test samples (from 1 mM to 0.1  $\mu$ M in DMSO) were mixed with medium (1:25) and 50  $\mu$ l of the mix was added to each well of the cultured cells (1:4) resulting in a final DMSO concentration of 1%. Cells were grown for 72 h at 10 % CO<sub>2</sub> and 37°C. After 72 h RENCA-LC cells were lysed in 100  $\mu$ L luciferase assay buffer, and 10  $\mu$ l of the lysate was assayed for luciferase activity. The mean value of the low controls of each 96 well cell culture plate represents the background which was subtracted from all other data points of the respective plate. Raw data were converted into percent proliferation relative to the high control, which were set to 100%. IC<sub>50</sub> determination was done with GraphPad Prism 5.01 software with constrain of bottom to 0 and top to 100 using a nonlinear regression curve fit with variable hill slope. The equation is a four-parameter logistic equation. Experiments were run in triplicates.

### Divergent synthesis of resorcylic acid lactones 3 and 5.

The gram-scale synthesis of compounds 3 and 5 is shown in Figure 13. Lithium acetylide addition to propylene oxide followed by silylation of the crude reaction product afforded compound 6 which was deprotonated (nBuLi) and added onto aldehyde 7 readily obtained from D-2-deoxyribose in three steps (acetone protection, (Barbat, J., et al. (1983) *Carbohydr. Res.* **116**, 312-16) olefination (Geng, X. & Danishefsky, S. J. (2004) *Org. Lett.* **6**, 413-6) and oxidation). Benzoate protection of the product afforded the key common

intermediate 8 in good overall yield (x% from propylene oxide). Ozonolysis with a reductive workup afforded alcohol 9 wherein the alkene was reduced with Lindlar's catalyst and the alcohol converted to an iodide to obtain intermediate 10. Conversely, the terminal alkene of compound 8 was hydroborated with an oxidative work up to obtain alcohol 11 which was converted to iodide 12 after a Lindlar reduction of the alkyne. Following previously established chemistry, fragment 12 was coupled to the aromatic moiety 13 by standard alkylation of the phenol. Global silyl deprotection (TMSE and TBDPS) and Mitsunobu cyclization afforded 14. Selective hydrolysis of the benzoate, oxidation to reveal the cis-enone and global deprotection afforded macrocycle 3 in x% from 10. The larger macrocycle was prepared by alkylation of 12 with the aryl moiety 14 bearing a benzylic selenide by alkylation followed by oxidation/elimination of the selenide (Dakas, P. Y., et al. (2007) *Angew. Chem. Int. Ed. Engl.* **46**, 6899-902). Using the same procedure as above with the exception of the final deprotection (HF rather than sulfonic acid resin), compound 5 was obtained in x% from 12 via 15.

#### In vivo efficacy

RENCA-LN cells were obtained and cultured as described above. Sixty female BALB/c mice of 5 to 6 weeks were orthotopically implanted RENCA-LN cells into the left kidney ( $4 \times 10^5$  cells in 25  $\mu$ l aliquots into the subcapsular space through a flank incision, Day 0). The mice were randomly divided into 5 groups of 12 mice each. On day 2, therapy was initiated. Animals in the vehicle control group (group 1) received 10 ml/kg Vehicle (PBS w/o Ca/Mg, 10%DMSO, 40% PEG-300) i.p. once daily until day 19. Animals of positive control group (group 2) received 40 mg/kg Sunitinib (Sutent<sup>®</sup>) p.o. once daily until day 20. Animals of treatment group 3 and 4 received 20 mg/kg of **3** once daily and every other day respectively (i.p.). Animals of treatment group 5 received 20 mg/kg of **5** once daily (i.p.). Due to partial weight loss, therapy in Groups 3 and 5 was suspended on days 11 and 12 and stopped on day 17. On day 20, the study was terminated and a necropsy performed. All animals were weighted, anaesthetised by isoflurane and killed by exsanguination. Two animals in group 1 were found dead on day 19 and 1 animal in group 3 and 5 on day 11 and 9 respectively. At necropsy, primary tumor weight and volume, lung weight and number of lung metastases were determined. Primary tumor tissues were collected and divided into two parts. One part of each tumor was snap-frozen in liquid nitrogen and stored at -80°C whereas the other part was analyzed via ex vivo bioluminescence imaging. During the course of the study, the growth of the orthotopically implanted RENCA tumors was monitored on days 7, 14 and 19 using

in vivo bioluminescence imaging. For this purpose, luciferin was injected intraperitoneally (i.p.) into the mice and light emission was measured post injection (10 min) using a NightOWL LB 981 bioluminescence imaging system (Berthold Technologies, Germany). The migration of RENCA tumor cells into the lung (metastasis) was analyzed post necropsy by counting nodules and using the luciferase assay. For this purpose, lungs were collected, homogenized, and assayed for luciferase activity using the kit from Promega (E1501) according to the manufacturer's instructions. Protein concentrations were determined using a Bradford assay and used to normalize luciferase activities (except for the primary tumor). A histological examination of the tumor vasculature was performed. For this purpose, cryosections of primary tumor tissues (thickness = 5-10  $\mu$ m) from the snap-frozen portion of the organ and blood vessels were visualized by immunohistochemical staining for CD 31 (PECAM-1; PharMingen, San Diego, CA). Statistical analysis: All relevant data of this study are graphically displayed as means  $\pm$  SEM. Statistical analysis of efficacy data was done using the *Mann Whitney* test.

#### Kinase selectivity profile

The selectivity profile was measure using the KinomeScan™ technology (<http://www.kinomescan.com/>) based on active-site dependent competition binding assay (Karaman, M. W., et al. (2008) *Nat. Biotechnol.* **26**, 127-32; Fabian, M. A., et al. (2005) *Nat. Biotechnol.* **23**, 329-36). Compounds **3** and **5** were profiled against a panel of 50 kinases including 42 kinases from the group of 48 kinases predicted by Santi et al. (Schirmer, A., et al. (2006) *Proc. Natl. Acad. Sci. USA* **103**, 4234-9) to be targeted by hypothemycin and 5 mutants of FLT3 as well as 3 mutants of KIT. Both compounds were assayed at 1  $\mu$ M. The results are shown in Figure 10. The results are expressed as a percentage of signal between a negative (DMSO) and positive (know binder) control  $[(\text{test compound} - \text{positive control})/(\text{test compound} - \text{negative control}) \times 100]$ .

#### **Results and Discussion**

Multiple strategies for inhibiting the VEGF pathway have been shown to hinder tumor growth and the recent approval of small molecule inhibitors (sorafenib (Escudier, B., et al. (2007) *N. Engl. J. Med.* **356**, 125-134; Abou-Alfa, G. K., et al. (2006) *J. Clin. Oncol.* **24**, 4293-4300) and sunitinib (Motzer, R. J., et al. (2006) *J. Clin. Oncol.* **24**, 16-24; Demetri, G. D., et al. (2006) *Lancet* **368**, 1329-1338)) as well as neutralizing antibodies (bevacizumab (Miller, K., et al. (2007) *N. Engl. J. Med.* **357**, 2666-2676; Hurwitz, H., et al. (2004) *N. Engl.*



*J. Med.* **350**, 2335-2342)) targeting the VEGF receptors have clinically validated this strategy (Folkman, J. (2007) *Nat. Rev. Drug Discov.* **6**, 273-86). However, enduring clinical response are rare and there is mounting evidence that antiangiogenic therapy, while delaying primary tumor growth, can concomitantly promote tumor adaptation and progression to stages of greater malignancy with heighten invasiveness and distant metastasis (Bergers, G. & Hanahan, D. (2008) *Nat. Rev. Cancer* **8**, 592-603; Kerbel, R. S. (2008) *N. Engl. J. Med.* **358**, 2039-2049). It has recently been shown that short-term sunitinib treatment resulted in accelerated metastasis in an mice model (Ebos, J. M., et al. (2009) *Cancer Cell* **15**, 232-9). Substantiating this pharmacological outcome, genetic ablation of *Vegf-A* increased invasiveness (Pacz-Ribes, M., et al. (2009) *Cancer Cell* **15**, 220-31). Furthermore, pretreatment of healthy mice with these VEGF inhibitors prior to inoculation of tumor cells "conditioned" the development of aggressive metastasis with shortened survival (Ebos, J. M., et al. (2009) *Cancer Cell* **15**, 232-9). While several hypotheses have been advanced to rationalize these observations, the mechanism remains speculative (Loges, S., et al. (2009) *Cancer Cell* **15**, 167-70). In most cancers, metastasis, not primary tumor growth, is ultimately fatal (Gupta, G. P. & Massague, J. (2006) *Cell* **127**, 679-695). Yet, most preclinical studies focus on primary tumor growth with less attention on metastasis. While sunitinib and sorafenib are very effective inhibitors of the angiogenic pathways and primary tumor growth, we sought to evaluate resorcylic acids, as the irreversible inhibition of the target kinases may provide different outcome (Copeland, R. A., et al. (2006) *Nat. Rev. Drug Discov.* **5**, 730-9; Swinney, D. C. (2004) *Nat. Rev. Drug Discov.* **3**, 801-8; Potashman, M. H. & Duggan, M. E. (2009) *J. Med. Chem.*). Several programs to develop irreversible kinase inhibitors have already progressed to the clinic (Erlichman, C., et al. (2006) *J. Clin. Oncol.* **24**, 2252-60; Janne, P. A., et al. (2007) *J. Clin. Oncol.* **25**, 3936-44; Eskens, F. A., et al. (2008) *Br. J. Cancer* **98**, 80-5).

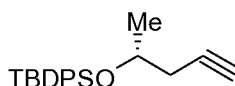
Following our biochemical screening, five cis-enone RAL analogues containing modifications which were found to be tolerated or beneficial in a biochemical assay were tested for their efficacy in inhibiting renal carcinoma cell line. As a bench mark, the natural product LL Z-1640-2 was also included in the assay. As shown in Figure 14, compound 3 containing the benzylic ether was as effective as the natural product while the larger macrocycles were slightly less effective with the best (**5**) being nearly 3 fold less potent in growth inhibition. Compound **3** and **5** were selected as prototypical examples of modified RALs for further investigations. As described in the method section, both of these compounds derived from a common synthetic intermediate and can be readily prepared in gram quantities. While compound **3** was notably more soluble in aqueous buffers than **5**, both compounds

were easily formulated in 10% DMSO, 40% PEG, 60% saline. A first dose finding study indicated systemic toxicity for both compounds at high dose (75 mg/kg) but were tolerated at 25 mg/kg. Animal weights decreased between day 2 and day 5 (e.g. 8% decrease in case of group treated with **3** and 13% decrease for the group treated with **5** (q1d), but remained stable afterwards. Compound **3** was tested with two different schedules (every day dosing and every other day dosing) while compound **5** was tested at a single schedule (every day dosing) and the results were compared to a positive control with sunitinib. Both test compounds (**3** and **5**) showed a highly significant reduction of tumor growth as measure by bioluminescence (Figure 15) and primary tumor volume with P values >0.0010 comparable to sunitinib (P=0.0004) for the daily dosing. Compound **3** administered every other day showed a marginal decrease activity (P=0.0011). It should be noted that we and others have found that cis-enone RAL such as LL Z-1640-2, compound 2 and 5 are inactivated by thiols such as DTT or GST and have short plasma half-life (Du, H., et al. (2009) *Bioorg. Med. Chem. Lett.* **19**, 6196-9). Thus the marginal difference in activity between the two dose schedule of compound **3** suggests that the irreversible inhibition of its target is fast relative to compound inactivation and, that target turn over is slower than 12h. Primary tumor weight confirmed these observations. A key objective of this study was to measure the effect of these compounds on metastasis. Upon termination of the study, the lungs were inspected for metastatic nodules and the ex-vivo luciferase activity of lung homogenates was measured. Concurring other studies and clinical observations, treatment with sunitinib resulted in a significant increase in lung metastatic nodules with 5 animals having a high count of nodules whereas the untreated group had only two animals (Figure 16). The overall level of luciferase activity from lung homogenates was three fold higher in the sunitinib treated group than in the untreated group. A similar result was obtained with compound **3** irrespective of the dosing schedule. However, within the group treated with compound **5**, no animal was found with a high count of lung metastasis (P=0.0445) nor luciferase activity in lung homogenates (P=0.0014). Surprisingly, tumor vasculature histology (CD-31 staining) showed a strong inhibition of angiogenesis for animal treated with compound **3** but not for animals treated with compound **5** (Figure 17). Comparison of the profile of kinase inhibition for both compound **3** and **5** showed a very similar pattern (Figure 10) with VEGFRs being strongly inhibited in both cases. However, considering that both compounds **3** and **5** have a short plasma half life, difference in the kinetics of kinase inactivation *in vivo* relatively to inhibitor inactivation may become significant. It should be noted that these compounds were also shown to inhibit the inflammation pathways which have been implicated in facilitating

metastasis. Finally, while hypothemycin had already been tested in mice bearing colon and lung tumors, only modest growth inhibition was observed (Tanaka, H., et al. (1999) *Jap. J. Cancer Res.* **90**, 1139-1145).

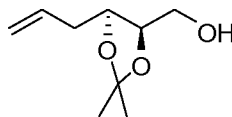
In conclusion, the *cis*-enone resorcylic acid lactones have emerged as highly promising agent in oncology and worthy of further development. The molecular editing of this natural pharmacophore has revealed important differences in biological activity which are the product of subtle differences in biochemical activity and kinetics of inactivation. In most cases, metastasis, not primary tumor growths, kills the cancer patient. The inhibition of lung metastasis in the present study is thus highly interesting considering that current first line therapy based on angiogenesis promotes it.

### Synthesis of compounds

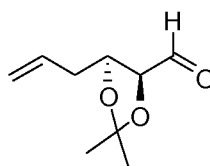


**TBDPS protected chiral alkyne (#):** To a stirred suspension of lithium acetylide.EDA complex (1.3 equiv, 37.6 g, 408.7 mmol) in DMSO (330 mL) was added epoxide (1.0 equiv, 18.2 g, 314.4 mmol) at 0°C and the resulting mixture was stirred for 1h at 0°C and 14 h at 23°C. the reaction mixture was poured on ice (200 g), stirred for 30 min, extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 200 mL), the combined organic layers were washed with brine (200 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the crude solution was submitted for next step without further evaporation of the solvents (*R<sub>f</sub>* = 0.25 (PE/EtOAc 4:1). To the above solution (660 mL) was added sequentially imidazol (2.0 equiv, 42.8 g, 628.8 mmol) and TBDPSCl (1.1 equiv, 89.9 mL, 345.8 mmol) at 23°C. The resulting mixture was stirred for 18 h at the same temp before adding another portion of imidazol (2.0 equiv, 42.8 g, 628.8 mmol) and TBDPSCl (0.5 equiv, 40.8 mL, 157.2 mmol). After stirring the reaction mixture for another 30 h, it was diluted with H<sub>2</sub>O (200 mL), the layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 100 mL). The combined organic layers were washed with brine (300 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, PE, PE/EtOAc 33:1; PE/EtOAc 20:1) to furnish TBDPS protected chiral alkyne # (68.8 g, 68%) as a colorless oil.

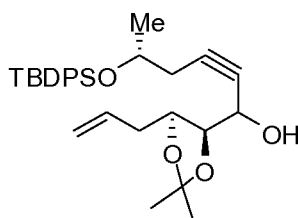
43



**Alcohol (#):** NaHDMS (2.8 equiv, 241.2 mL, 241.2 mmol, 1M in THF) was added to a suspension of methyltriphenylphosphoniumbromide (3.0 equiv, 92.2 g, 258.3 mmol) in THF (375 mL) at  $-78^{\circ}\text{C}$ . The mixture was stirred at  $23^{\circ}\text{C}$  for 30 min before the addition of sugar # (1.0 equiv, 15.0 g, 86.1 mmol) in THF (50 mL) at  $-78^{\circ}\text{C}$ . Later the reaction mixture was stirred at  $23^{\circ}\text{C}$  for 1 h, quenched with sat.aq. $\text{NH}_4\text{Cl}$  (100 mL) before the reaction mixture turns red from yellow solution. The layers were separated, aqueous layer was extracted with EtOAc (3 x 100 mL) and the combined organic layers were dried over  $\text{Na}_2\text{SO}_4$ , filtered and evaporated under reduced pressure. The residue was purified by flash chromatography ( $\text{SiO}_2$ , PE/EtOAc 5:1) to obtain alcohol # (10.7 g, 72%) as a colorless liquid.

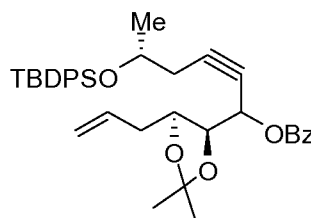


**Aldehyde (#):** To a solution of alcohol # (1.0 equiv, 10.0 g, 58.6 mmol) in  $\text{CH}_2\text{Cl}_2$  (450 mL) and DMSO (115 mL) was added sequentially  $\text{Et}_3\text{N}$  (4.9 equiv, 39.5 mL, 284.5 mmol) and  $\text{SO}_3$ -pyridine complex (3.47 equiv, 32.0 g, 201.5 mmol) at  $0^{\circ}\text{C}$ . The reaction mixture was stirred for 20 min at the same temperature and 1 h at  $23^{\circ}\text{C}$  before it was quenched with  $\text{H}_2\text{O}$  (200 mL). The layers were separated and the aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$  (2 x 100 mL), the combined organic layers were washed with  $\text{H}_2\text{O}$  (200 mL), brine (200 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography ( $\text{SiO}_2$ , PE/EtOAc 4:1) to provide aldehyde # (8.5 g, 86%) as a light yellow liquid.

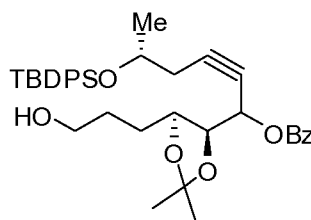


**Allylic alcohol (#):** To a solution of alkyne # (1.1 equiv, 17.7 g, 55.0 mmol) in THF (128 mL) was added  $n\text{-BuLi}$  (1.15 equiv, 42.6 mL, 57.5 mmol, 1.35 M in hexane) at  $-78^{\circ}\text{C}$ . The reaction mixture was stirred for 45 min at the same temperature before aldehyde (1.0

equiv, 8.5 g, 50.0 mmol) in THF (128 mL) was added. After being stirred for 1.5 h at -78 °C, 2 h at 0°C and 30 min at 23°C, the reaction was quenched with Sat. aq. NH<sub>4</sub>Cl (100 mL), warmed to RT, the layers were separated and the aqueous layer was extracted with EtOAc (3 x 50 mL), the combined organic layers were washed with brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated at vacuo. The crude product was purified by flash chromatography (SiO<sub>2</sub>, PE/EtOAc 20:1, 5:1, 4:1) to obtain alcohol # (19.8 g, 81%) as light yellow oil.

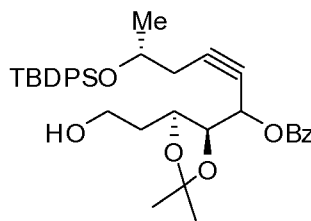


**Benzoate (#):** To a solution of alcohol # (1.0 equiv, 19.2 g, 38.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (230 ml) was added sequentially pyridine (2.5 equiv, 7.87 mL, 97.4 mmol) and benzoyl chloride (2.5 equiv, 11.3 mL, 97.4 mmol) at 0°C. The mixture was stirred for 12 h by warming slowly from 0 °C to RT. Then, the reaction mixture was diluted with H<sub>2</sub>O (100 mL), stirred for 30 min, layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 50 mL). The combined organic layers were washed with sat. aq. NH<sub>4</sub>Cl (100 mL), brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed in vacuo. The residue was purified by flash chromatography (PE, PE/EtOAc 50:1, 25:1) to obtain benzoate # (19.8 g, 85%) as a colorless oil.

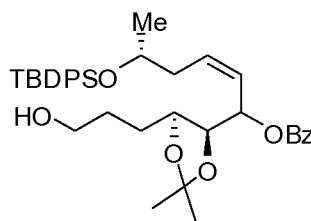


**Alcohol (#):** To a solution of alkene # (1.0 equiv, 21.7 g, 36.4 mmol) in THF (380 mL) was added 9-BBN (2.5 equiv, 182 mL, 91.0 mmol, 0.5M in THF) at 0 °C. The reaction mixture was warmed to 23 °C and stirred for 14 h hours. After the flask was cooled to 0 °C 3N aq. NaOH solution (36.4 mL) followed by 30% aq. H<sub>2</sub>O<sub>2</sub> ( 36.4 mL) was added very slowly, then the mixture was warmed to 23°C and stirred for another 1.5 hours. The reaction mixture was quenched with H<sub>2</sub>O (200 mL), the layers were separated, the aqueous layer was extracted with EtOAc (3 x 100 mL) and the combined organic layers were washed with brine (200 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated under reduced pressure. The residue

was purified by flash chromatography (SiO<sub>2</sub>, PE /EtOAc 3:1, 2:1, 1:1) to provide the alcohol # (21.3 g, 95%) as a colorless liquid.

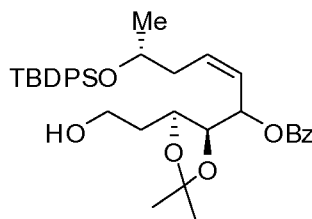


5        **Aldehyde (#):** A solution of alkene (1.0 equiv, 8.0 g, 13.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub>/MeOH (130 mL) was bubbled with ozone for 30 min at -78°C. The excess ozone was removed by bubbling oxygen (the oxygen was bubbled until the blue colour disappeared). Later the mixture was bubbled using N<sub>2</sub> for 15min before NaBH<sub>4</sub> (5 equiv, 2.53 g, 67.0 mmol)) was added. The reaction mixture was stirred for 30 min and added another portion of NaBH<sub>4</sub> (5  
10 equiv, 2.53 g, 67.0 mmol), stirred the resulting mixture for 1 h at the same temp, placed the flask at 0°C, stirred for another 30 min before it was quenched with H<sub>2</sub>O (50 mL), the reaction mixture was warmed to RT, extracted with EtOAc (5 x 50 mL), the combined organic layers were washed with brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography  
15 (SiO<sub>2</sub>, PE/EtOAc 4:1, 3:1, 2:1) to furnish alcohol # (4.0 g, 50%) as a colorless oil.



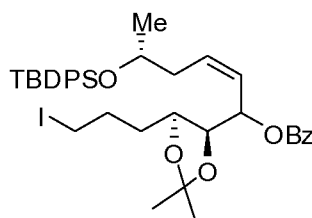
20        **Alcohol #:** To a solution of alcohol # (1.0 equiv, 21.2 g, 34.4 mmol) in MeOH (2.6 L) was added Pd/CaCO<sub>3</sub> (1.72 equiv, 6.3 g, 59.2 mmol) at 23 °C and the reaction mixture was stirred for 25 min under H<sub>2</sub> atmosphere (reaction was monitored by LC-MS). After filtration on celite, the solvents were evaporated under reduced pressure. The crude product was flushed from short pad of silica (PE/EtOAc 2:1) to obtain compound # (21.1 g, 99%) as a colorless oil.

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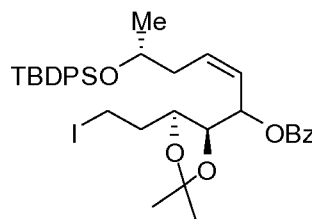


**Compound #:** The procedure of alcohol # was used with alcohol # (1 equiv, 4.41 g, 7.35 mmol) in MeOH (500 mL) and Pd/CaCO<sub>3</sub> (1.72 equiv, 1.34 g, 12.6 mmol) to yield compound # (4.30 g, 97%) as a colorless oil after filtration on short pad of silica.

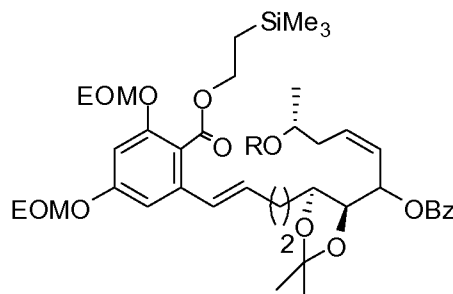
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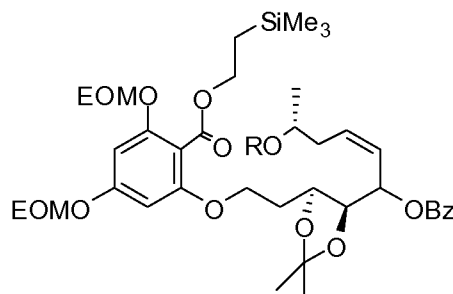
**Alkyl iodide #:** To a solution of alcohol # (1.0 equiv, 21.1 g, 34.2 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (125 mL) was added sequentially imidazole (2.0 equiv, 4.65 g, 68.4 mmol) and triphenylphosphine (1.8 equiv, 16.1 g, 61.6 mmol) at 23°C. The resulting mixture was cooled to 0°C, I<sub>2</sub> (1.8 equiv, 15.6 g, 61.6 mmol) was added and stirred at the same temperature for 10 min. After stirring the reaction mixture for 2.5 h at 23°C, it was quenched with sat. aq. NaHCO<sub>3</sub> (50 mL), the layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 25 mL). The combined organic layers were washed with sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution (50 mL), brine (50 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (PE/EtOAc 10:1) to afford the alkyl iodide # (22.3 g, 90%) as a light yellow oil.



**Iodide #:** The procedure of iodide # was used with alcohol (1 equiv, 4.30 g, 7.13 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 mL), imidazole (3.0 equiv, 1.45 g, 21.4 mmol), triphenylphosphine (1.8 equiv, 3.36 g, 12.83 mmol) and I<sub>2</sub> (1.8 equiv, 3.25 g, 12.83 mmol) to yield iodide # (4.57 g, 90%) as a colorless oil after flash chromatography.



**Procedure for alkylation reaction followed by the oxidation *syn*-elimination #:** To a solution of aromatic seleno-ether (1.0 equiv, 16.3 g, 30.3 mmol) in THF/HMPA 10/1 (200 mL) cooled at -78 °C was added a freshly prepared solution of LDA in THF (2.0 equiv, 116.5 mL, 60.6 mmol, 0.52 M in THF) via cannula. The resulted dark red color reaction mixture was stirred for 10 min at -78°C and a solution of the precooled alkyl iodide # (1.0 equiv, 22.0g, 30.3 mmol) in THF (60 mL) was added dropwise at the same temperature. After stirring for 20 min, the reaction was quenched with sat. aq. NH<sub>4</sub>Cl (100 mL), warmed the flask to room temperature, the layers were separated and the aqueous layer was extracted with EtOAc (3 x 50 mL). The combined organic layers were washed with brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were removed under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, PE/EtOAc, 20:1, 10:1) to obtain desired alkylated compound # (30.9 g, 89%) as mixture of 4 diastereoisomers. *R<sub>f</sub>* = 0.26 (PE/EtOAc 6:1). To a stirred solution of previously prepared seleno-ether # (1.0 equiv, 30.9 g, 27.1 mmol) in THF (230 mL) was added H<sub>2</sub>O<sub>2</sub> 35% (2.0 equiv, 6.5 mL, 61.8 mmol) at 23 °C. After stirring for 13 h, the reaction mixture was concentrated under reduced pressure and the crude product was purified by flash chromatography (SiO<sub>2</sub>, PE/EtOAc, 10:1) to afford compound # (24.2 g, 91%) as yellow oil.

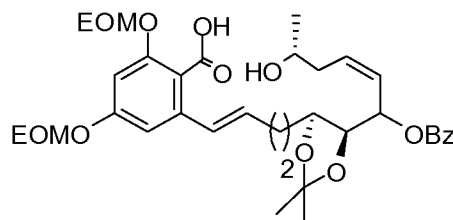


**Compound #:** To a solution of phenol # (1.0 equiv, 2.24 g, 5.81 mmol) in DMF (15 mL) was added K<sub>2</sub>CO<sub>3</sub> (2.0 equiv, 1.55 g, 11.61 mmol) followed by a solution of iodide (1.1 equiv, 4.56 g, 6.39 mmol) in DMF (15 mL) at 23°C. The resulting mixture was heated to 80°C and stirred for 16 h. then it was diluted with H<sub>2</sub>O (25 mL), stirred for 20 min, extracted



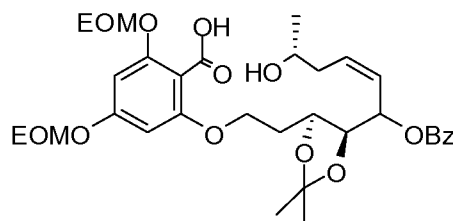
with Et<sub>2</sub>O (4 x 25 mL), the combined organic layers were washed with brine (25 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (PE/EtOAc 10:1, 5:1) to afford the compound # (5.42 g, 96%) as a very light yellow oil.

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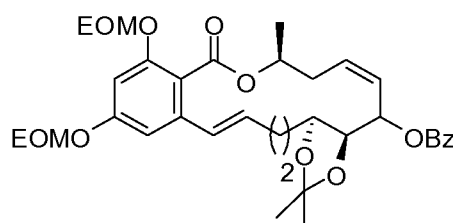
**Acid #:** To a solution of silyl-ester (1.0 equiv, 11.2 g, 11.4 mmol) in THF (155 mL) 23 °C was added tetra-butyl-ammonium fluoride (10.0 equiv, 114 mL, 114.1 mmol, 1M in THF). The resulting mixture was stirred for 56 h, then the reaction was quenched using sat. aq. NH<sub>4</sub>Cl (100 mL), layers were separated, the aqueous layer extracted with EtOAc (5 x 50 mL), the combined organic layers were washed with H<sub>2</sub>O (2 x 100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduce pressure. The resulted crude product was subjected for the next step with out any further purification. LCMS: *m/z* for [M + Na]<sup>+</sup> calcd for C<sub>35</sub>H<sub>46</sub>O<sub>11</sub>Na: 665.293; found: 665.300.

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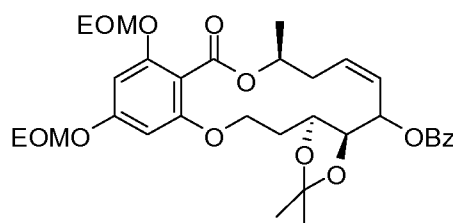


**Acid #:** The procedure of acid # was used with silyl ester # (1 equiv, 5.40 g, 5.56 mmol) in THF (80 mL), TBAF (10 equiv, 55.6 mL, 55.6 mmol, 1M in THF) to obtain acid # as a crude residue. LCMS: *m/z* for [M + Na]<sup>+</sup> calcd for C<sub>33</sub>H<sub>44</sub>O<sub>12</sub>Na: 655.273; found: 655.440.

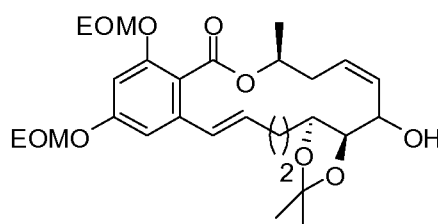
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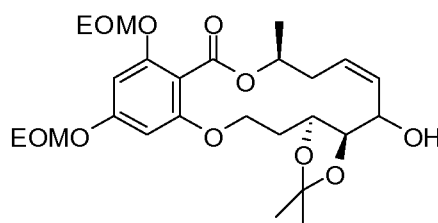
**Macrocycle #:** To a solution of acid # (1.0 equiv, 5.4 g, 8.4 mmol) in toluene (800 mL) at 23°C was added DIAD (2.0 equiv, 3.32 mL, 16.9 mmol) and PPh<sub>3</sub> (2.0 equiv, 4.43 g, 16.9 Mmol) and the resulting mixture was stirred for 2h. The solvents were removed under reduced pressure and the residue was purified by flash chromatography (SiO<sub>2</sub>, PE/EtOAc, 10:1, 6:1) to obtain Mitsunobu product # (3.67 g, 70%) as a colorless oil.



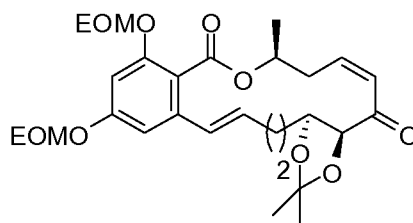
**Macrocycle #:** The procedure of macrocycle # was used with acid # (1 equiv, 2.58 g, 4.07 mmol) in toluene (380 mL), PPh<sub>3</sub> (2.0 equiv, 2.13 g, 8.15 mmol), DIAD (2.0 equiv, 1.60 mL, 8.15 mmol) to obtain Mitsunobu product # (2.05 g, 80%) as a colorless oil after flash chromatography.



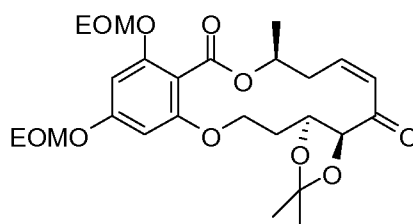
**Allylic alcohol #:** A solution of benzoate # (1.0 equiv, 3.67 g, 5.87 mmol) in 1% NaOH in MeOH (75 mL) was heated to 50°C and stirred for 4 h. The solvents were removed under reduced pressure and the residue was purified by flash chromatography (SiO<sub>2</sub>, PE/EtOAc, 6:1) to obtain allylic alcohol # (1.95 g, 64%) as colorless oil.



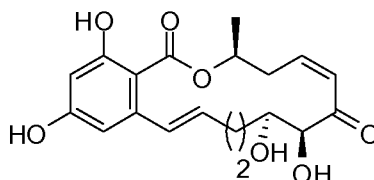
**Allylic alcohol #:** The procedure of macrocycle # was used with benzoate # (1 equiv, 2.0 g, 3.20 mmol) in 1% NaOH in MeOH (40 mL) to yield allylic alcohol # (1.13 g, 68%) as a colorless oil after flash chromatography.



**Cis-enone #:** To a solution of allylic alcohol # (1.0 equiv, 1.90 g, 3.65 mmol) in  $\text{CH}_2\text{Cl}_2$  (75 mL) was added DMP (1.5 equiv, 2.32 g, 5.47 mmol) at 23°C. The resulting mixture was stirred for 14 h, diluted with  $\text{Et}_2\text{O}$  (50 mL), quenched the reaction with 10% aq.  $\text{Na}_2\text{S}_2\text{O}_3$  (20 mL). After stirring the reaction mixture for 10 min, the layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  (3 x 10 mL), the combined organic layers were washed with sat. aq.  $\text{NaHCO}_3$  (25 mL) and brine (25 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvents were evaporated under reduced pressure. The residue was filtered through short pad of silica ( $\text{SiO}_2$ , PE/ $\text{EtOAc}$ , 2:1) to obtain cis-enone # (1.62 g, 86%) as a light yellow oil.

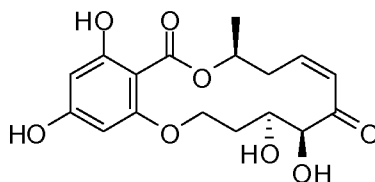


**Cis enone #:** The procedure of cis-enone # was used with alcohol # (1 equiv, 1.12 g, 2.15 mmol) in  $\text{CH}_2\text{Cl}_2$  (45 mL) and DMP (1.5 equiv, 1.36 g, 3.22 mmol) to yield cis-enone # (910 mg, 83%) as a very light yellow oil.



**Macrocycle #:** The cis-enone # (1.0 equiv, 1.0 g, 1.93 mmol) was diluted with HF 40% in  $\text{H}_2\text{O}/\text{MeCN}$  (1/10, 100 mL) at 23 °C. The solution was stirred at 23 °C for 6.5 h, diluted with  $\text{H}_2\text{O}$  (100 mL), and then dried by lyophilization. The residue was purified by flash chromatography ( $\text{SiO}_2$  was washed atleast for 5 times with eluent before loading the compound,  $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 25:1) to afford the desired macrocycle # (490 mg, 70%) along with the corresponding *para*-mono-EOM compound that were resubmitted for a second deprotection in the same conditions..  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ , 400 MHz):  $\delta$  = 6.97 (d,  $J$  = 15.9 Hz,

1H), 6.47 (dd,  $J = 11.7, 3.1$  Hz, 1H), 6.44 (d,  $J = 2.4$  Hz, 1H), 6.28 (d,  $J = 2.4$  Hz, 1H), 6.22-6.08 (m, 2H), 5.53-5.48 (m, 1H), 4.63 (d,  $J = 2.4$  Hz, 1H), 4.15 (dt,  $J = 9.7, 3.0$  Hz, 1H), 3.14 (dt,  $J = 17.7, 10.9$  Hz, 1H), 2.75 (dq,  $J = 17.7, 2.7$  Hz, 1H), 2.47-2.34 (m, 1H), 2.31-2.16 (m, 1H), 1.90-1.82 (m, 1H), 1.48 (d,  $J = 6.2$  Hz, 3H), 1.32-1.25 (m, 1H) ppm, 4OH signals not visible.  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz, 25 °C):  $\delta = 202.2$  (C=O), 172.9 (C=O), 166.7 (C), 164.0 (C), 146.1 (CH), 145.2 (C), 132.7 (CH), 132.5 (CH), 127.5 (CH), 109.4 (CH), 103.4 (C), 102.8 (CH), 82.1 (CH), 74.1 (CH), 69.6 (CH), 38.3 ( $\text{CH}_2$ ), 30.0 ( $\text{CH}_2$ ), 26.7 ( $\text{CH}_2$ ), 20.9 ( $\text{CH}_3$ ); HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{19}\text{H}_{22}\text{O}_7\text{H}$ : 363.1444; found: 363.1443.



**Oxygen Macrocycle #:** To a solution of cis-enone # (1.0 equiv, 1.05 g, 2.06 mmol) in MeOH (185 mL) was added sulfonic acid resin (5.0 equiv, 3.33 g, 10.3 mmol, 3.0 mmol/g) at 23°C. The reaction mixture was shaken for 9 h, then it was filtered washed with  $\text{CH}_2\text{Cl}_2$  (5 x 50 mL) and the solvents were removed under reduced pressure. The residue was precipitated by  $\text{Et}_2\text{O}$  to obtain desired macrocycle # (435 mg, 60%) as a very light red solid.  $R_f = 0.34$  ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 9:1); 10% trans isomer was observed in  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 400 MHz):  $\delta = 6.56$  (d,  $J = 15.6$  Hz, 0.1H), 6.50 (dd,  $J = 11.7, 3.0$  Hz, 0.9H), 6.21 (td,  $J = 11.6, 2.4$  Hz, 1H), 6.00 (d,  $J = 2.1$  Hz, 1H), 5.97 (d,  $J = 2.1$  Hz, 1H), 5.43-5.36 (m, 1H), 4.67 (bs, 1H), 4.27-4.17 (m, 2H), 3.91 (dt,  $J = 9.1, 3.0$  Hz, 1H), 3.21 (dt,  $J = 16.4, 11.6$  Hz, 1H), 2.53-2.42 (m, 1H), 1.87 (ddd,  $J = 14.5, 10.5, 3.5$  Hz, 1H), 1.67-1.55 (m, 1H), 1.45 (d,  $J = 6.2$  Hz, 0.3H), 1.38 (d,  $J = 6.2$  Hz, 2.7H) ppm, 4OH signals not visible.  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz, 25 °C):  $\delta = 202.3$  (C=O), 172.6 (C=O), 167.1 (C), 165.4 (C), 163.7 (C), 145.7 (CH), 128.3 (CH), 96.5 (CH), 96.4 (C), 92.7 (CH), 82.9 (CH), 73.4 (CH), 69.9 (CH), 65.7 ( $\text{CH}_2$ ), 37.8 ( $\text{CH}_2$ ), 31.4 ( $\text{CH}_2$ ), 20.9 ( $\text{CH}_3$ ); HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{17}\text{H}_{20}\text{O}_8\text{H}$ : 353.1237; found: 353.1243.

### Example 3

Based on the fact that these compounds react with the cysteine residue through a Michael addition, modulation of the electronic properties of the enone may further enhance the inhibitory properties of this pharmacophore. We had shown that substitution at the  $\beta$ -

position of the enone with a methyl group abrogated activity (Jogireddy, R., et al. (2009) *Chem. Eur. J.* **15**, 11498-506). These results were concurrently confirmed by researchers from Eisai (Du, H., et al. (2009) *Bioorg. Med. Chem. Lett.* **19**, 6196-9). To further pursue modulation of the Michael acceptor, we reasoned that a fluorine at the  $\alpha$ -position of the enone may accentuate the Michael acceptor properties while constituting a fairly neutral steric permutation. This "molecular editing" (Wilson, R. M. & Danishefsky, S. J. (2006) *J. Org. Chem.* **71**, 8329-51) may provide further benefits from a synthetic perspective as well as favorably impede the propensity of these *cis*-enones to isomerize to the thermodynamically more favorable and biologically less active *trans*-enone. As shown in Figure 18, the synthesis could capitalize on our previously established strategy (Dakas, P. Y., et al. (2007) *Angew. Chem. Int. Ed. Engl.* **46**, 6899-902; Dakas, P. Y., et al. (2009) *Chem. Eur. J.* **15**, 11490-7; Barluenga, S., et al. (2006) *Angew. Chem. Int. Ed. Engl.* **45**, 3951-3954), namely, disconnection of the ester and benzylic position, however, the difference of reactivity between a fluoroalkene and an alkene could be harnessed to introduce the diol moiety by dihydroxylation chemistry.

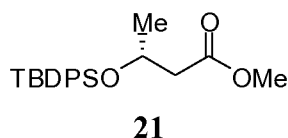
The synthesis commenced with the straightforward conversion of ester **5** into fluoroenal **7** via aldehyde **6** relying on a *Horner-Wadsworth-Emmons reaction* with triethyl 2-fluoro-2-phosphonoacetate to install the desired *E*-fluoroenal (Figure 19). Reaction of aldehyde **7** with alkyl lithium **8** afforded the desired alcohol as a 1:1 diastomeric mixture which was reduced to the *cis*-alkene using Lindlar's catalyst ( $H_2$ , Pd/CaCO<sub>3</sub>) to afford compound **9**. As the hydroxyl group will ultimately be oxidized to the ketone, the lack of selectivity is inconsequential. Benzoyl protection of alcohol **9** thus afforded **10** wherein the terminal OPMB was converted to an iodide **11** (DDQ; I<sub>2</sub>, PPh<sub>3</sub>, imidazole). Attempts to engage this compound in alkylation chemistry with a phenol led to rapid elimination of the iodide to give a conjugated triene (product not shown). To circumvent this elimination which was thought to be facilitated by the presence of the alkene, compound **10** was dihydroxylated (2.5:1 *dr* in favor of the desired isomer **12**) and converted to acetone **13**. Conversion of the terminal protected PMB ether into an iodide afforded the key intermediate **14**. Following previously established chemistry, fragment **14** was coupled to the aromatic moiety **15** bearing a benzylic selenide by alkylation followed by oxidation/elimination of the selenide to afford **17**. Conversely, fragment **14** was alkylated with phenol **16** to obtain the ether analogue **18** in excellent yield. Global deprotection of the silyl groups (TMSE and TBDPS) followed by Mitsunobu macrocyclization, selective benzoate deprotection and oxidation of the resulting alcohol afforded products **19** and **20** from **17** and **18** respectively in good overall yield. Global

deprotection with aqueous HF or sulfonic acid resin afforded the desired fluoroenone resorcylic acids **1** and **2**. We and others had previously noted the sensitivity of the *cis*-enone system of LL-Z1640-2 and related analogues to acid conditions which could lead to isomerisation (Dakas, P. Y., et al. (2009) *Chem. Eur. J.* **15**, 11490-7; Hofmann, T. & Altmann, K.-H. (2008) *Synlett*, 1500-1504; Selles, P. & Lett, R. (2002) *Tet. Lett.* **43**, 4627-4631). As anticipated based on the higher thermodynamic stability of *E*-fluoroalkene, analogues **1** and **2** proved to be more resistant to epimerization and no trace of isomerization was observed even with prolonged reaction times. Interestingly, the different diastereoisomers originating from the unselective dihydroxylation proved to have significant difference in the kinetics of deprotection which offered a convenient means to separate them.

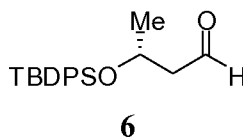
To gage the indiscriminate reactivity of these enone resorcylic acids with thiols, fluoroenone **2** and the parent analogue lacking the fluoride were treated with one equivalent of DTT (2 mM) as a prototypical thiol nucleophile in PBS (pH 7.4) containing 1% DMSO. Interestingly, a clear difference in reactivity was observed with the fluoroenone **2** reacting slower. With fluoroenone **1** and **2** in hand, we tested their efficacy against VEGF-R2 which was frequently found to be the most inhibited kinase in our profile of *cis*-enone resorcylic acid library (Dakas, P. Y., et al. (2007) *Angew. Chem. Int. Ed. Engl.* **46**, 6899-902; Jogireddy, R., et al. (2009) *Chem. Eur. J.* **15**, 11498-506) and KIT which represents a less inhibited kinase. As shown in Figure 20, the fluoroenone **1** and **2** are less active than the natural product LL-Z1640-2 however maintain an inhibition level which is interesting (6.8 nM and 60.5 nM for **1** and **2** respectively against VEGF-R2 vs 2.63 nM for LL-Z1640-2) and their selectivity for VEGF-R2 relative to KIT remains unaffected. The products originating from the minor diastereoisomers of the dihydroxylation were significantly less active. Their cellular activity against VEGF-R2 did parallel the biochemical activity. As shown in figure 21, the natural product LL-Z1640-2 is a very potent inhibitor with an IC<sub>50</sub> of 6.5 nM. The direct comparison with the fluoroenone **2** revealed an IC<sub>50</sub> of 17 nM while the analogue containing the synthetically more accessible ether macrocycle remained very potent with an IC<sub>50</sub> of 70 nM. Thus the fluoroenone modification is well tolerated but does not enhance the cellular efficacy of VEGF-R2 inhibition. During the course of our investigation, a similar modification was reported by researchers from Esai (Du, H., et al. (2009) *Bioorg. Med. Chem. Lett.* **19**, 6196-9) however, their medicinal chemistry efforts focused the inhibition of inflammation related pathways (MEKs and their down-stream regulatory effect on cytokines) (Du, H., et al. (2009) *Bioorg. Med. Chem. Lett.* **19**, 6196-9; Goto, M., et al. (2009) *J. Pharmacol. Exp. Ther.* **331**, 485-95). A selected analogue (E6201) indeed shows a >60 and >200 fold selectivity (Goto,

M., et al. (2009) *J. Pharmacol. Exp. Ther.* **331**, 485-95) for MEK1 relatively to VEGF-R2 and PDGFR $\beta$ , which is a significantly different pattern relatively to the parent natural product (LL-Z1640-2) for which VEGF-R2 and PDGFR $\beta$  are most inhibited (Jogireddy, R., et al. (2009) *Chem. Eur. J.* **15**, 11498-506.). It was further shown that the addition of a methyl substituent at the  $\delta$  position to the enone enhanced the metabolic stability with a tolerable loss of activity (10 fold) (Du, H., et al. (2009) *Bioorg. Med. Chem. Lett.* **19**, 6196-9). It should be noted however that several *cis*-enone had already shown to be effective *in vivo*. As the inhibition is irreversible, the relative rate of metabolic instability and irreversible inhibition coupled to target turn over should be considered.

## Synthesis of compounds

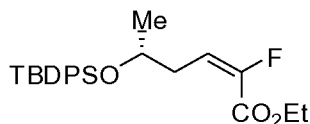


**TBDPS protected chiral ester (21):** To a stirred solution of alcohol **5** (1.0 equiv, 10.0 g, 84.6 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (150 mL) were added sequentially imidazol (2.0 equiv, 11.5 g, 169.2 mmol) and TBDPSCl (1.2 equiv, 26.5 mL, 101.5 mmol) at 0 °C. The resulting mixture was stirred for 13 hours at 23 °C. Then the mixture was diluted with ether (300 mL), washed with sat.aq.NH<sub>4</sub>Cl (200 mL) and brine (200 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 9:1) to obtain **21** (27.2 g, 90%) as a colorless oil.

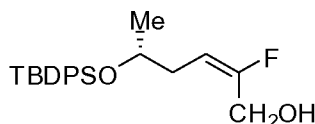


**Aldehyde (6):** To a solution of ester **21** (1.0 equiv, 10.0 g, 28.0 mmol) in toluene (150 mL) at -78 °C was added dropwise a 1M solution of DIBAL in toluene (1.1 equiv, 30.8 mL, 30.8 mmol) over a period of 30 min. The reaction mixture was stirred for 1 hour at -78 °C before it was quenched with MeOH (3.3 mL) and sat.aq.NH<sub>4</sub>Cl (5.5 mL). The mixture was then diluted with Et<sub>2</sub>O (200 mL), poured to a flask containing 50% aq. NaK tartarate (200 mL), stirred for 1 hour at room temperature, the layers were separated and the aqueous layer was washed with Et<sub>2</sub>O (2 x 100 mL). The organic layers were combined and dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was

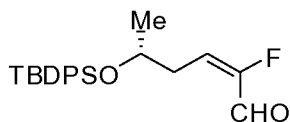
purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 9:1) to afford aldehyde **6** (7.82 g, 85%) as a colorless oil.

**22**

**Allylic ester (22):** To a solution of the phosphonate ester (1.2 equiv, 5.0 g, 20.6 mmol) in THF (200 mL) at 0 °C was added n-BuLi (1.2 equiv, 14.7 mL, 20.6 mmol, 1.4 M in hexane). The reaction mixture was stirred for 1 hour at the same temperature before the solution of aldehyde **6** (1.0 equiv, 5.60 g, 17.1 mmol) in THF (20 mL) was added. After 20 min the flask was placed at room temperature and stirred for 12 hours. The reaction was then quenched with 0.5 N aq. HCl (100 mL), the layers were separated and the aqueous layer was extracted with EtOAc (3 x 50 mL). The combined organic layers were washed with brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were removed under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 19:1) to afford ester **22** (5.92 g, 83%) as a light yellowish oil. It contains 6% Z-isomer.

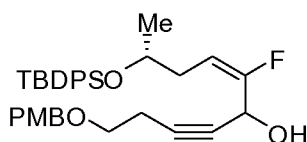
**23**

**Allylic alcohol (23):** To a solution of allylic ester **22** (1.0 equiv, 5.86 g, 14.1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (160 mL) at 0 °C was added DIBAL (2.5 equiv, 35.3 mL, 35.3 mmol, 1M solution in toluene). The reaction mixture was stirred for 40 min at 0 °C and 30 min at room temperature, quenched with 50% aq. NaK tartarate solution (100 mL) and stirred for 2 hours more at the same temperature. The layers were separated and the aqueous layer was extracted with EtOAc (2 x 100 mL), the combined organic layers were washed with brine (200 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 9:1) to afford allylic alcohol **23** (4.48 g, 85%) as a colorless oil.

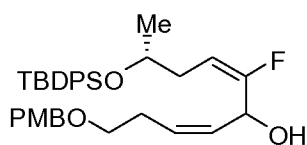




**Aldehyde (7):** To a solution of oxalyl chloride (1.5 equiv, 1.52 mL, 17.7 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (35 mL) at -78 °C was added DMSO (2.45 equiv, 2.2 mL, 28.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (7 mL). The mixture was stirred for 30 min at the same temperature before a solution of alcohol **23** (1.0 equiv, 4.4 g, 11.8 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added. After stirring the reaction for 45 min at -78 °C, Et<sub>3</sub>N (4.0 equiv, 6.56 mL, 47.2 mmol) was added and the mixture was allowed to warm to room temperature over a period of 2 hours. Then it was diluted with H<sub>2</sub>O (25 mL), layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 20 mL), the combined organic layers were washed with 1N HCl (25 mL), sat. aq. NaHCO<sub>3</sub> (25 mL), and brine (25 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 9:1) to furnish aldehyde **7** (4.30 g, 98%) as a colorless oil.

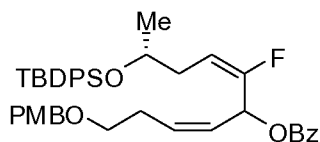
**24**

**Alcohol 24:** To a solution of the alkyne (2.06 equiv, 4.54 g, 23.9 mmol) in THF (227 mL) at -78 °C was added *n*-BuLi (2.0 equiv, 19.3 mL, 23.2 mmol, 1.2 M in hexane). The reaction mixture was stirred for 1 hour at the same temperature before adding aldehyde **23** (1.0 equiv, 4.30 g, 11.6 mmol) in THF (140 mL). After stirring the mixture for 1 hour at -78 °C, the reaction was quenched with sat. aq. NH<sub>4</sub>Cl (100 mL) and warmed to 23 °C. Once a room temperature the layers were separated and the aqueous layer was extracted with EtOAc (3 x 50 mL), the combined organic layers were washed with brine (100 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under vacuo. The crude product was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc 9:1, 4:1) to obtain alcohol **24** (4.05 g, 62%, 1:1 ratio of isomers) as a light yellow oil.

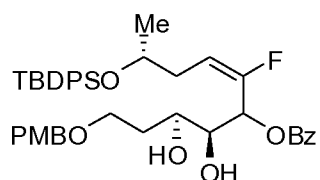
**9**

**Compound 9:** A solution of alcohol **24** (1.0 equiv, 3.39 g, 6.04 mmol) in MeOH (470 mL) at 23 °C was treated with Pd/CaCO<sub>3</sub> (1.72 equiv) and the reaction mixture was stirred for

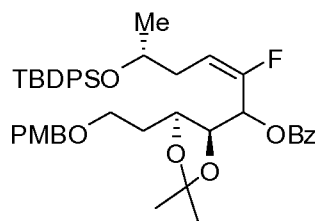
25 min under H<sub>2</sub> atmosphere. After filtration on celite, the solvents were evaporated under reduced pressure. The crude product was flushed through a short pad of silica (petroleum ether/EtOAc 3:1) to obtain compound **9** (3.30 g, 97%) as a colorless oil.

**10**

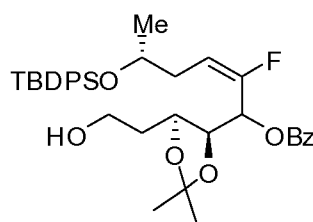
**Benzoate 10:** To a solution of alcohol **9** (1.0 equiv, 3.30 g, 5.86 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (35 ml) at 0°C were added sequentially pyridine (2.5 equiv, 1.18 mL, 14.65 mmol) and benzoyl chloride (1.5 equiv, 1.02 mL, 8.79 mmol) and the mixture was stirred for 14 hours while warming slowly from 0 °C to room temperature. Then, the reaction was diluted with H<sub>2</sub>O (20 mL), the layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 20 mL). The combined organic layers were washed with sat. aq. NH<sub>4</sub>Cl (20 mL), brine (20 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed under vacuo. The crude product was purified by flash chromatography (petroleum ether/EtOAc 20:1, 15:1, 10:1) to obtain benzoate **10** (3.56 g, 91%) as a colorless oil.

**12**

**Diol 12:** To a solution of olefin **10** (1.0 equiv, 702 mg, 1.05 mmol) in THF (7 mL) were added sequentially NMO.H<sub>2</sub>O (2.0 equiv, 284 mg, 2.10 mmol) in H<sub>2</sub>O (7 mL) and OsO<sub>4</sub> (0.013 equiv, 86 µL, 0.0136 mmol, 4% wt in H<sub>2</sub>O) at 23 °C. The resulted mixture was stirred for 15 hours. Then the reaction was quenched with a 1:1 mixture of sat. aq. NaHCO<sub>3</sub> and 10% aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution (5 mL) and stirred at room temperature for 1 hour. The layers were separated and the aqueous layer was extracted with EtOAc (3 x 5 mL), the combined organic layers were washed with brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The crude product was purified by flash chromatography (petroleum ether/EtOAc 6:1, 4:1, 3:1) to afford the diol **12** (577 mg, 78% based on isolated yield and 87% based on recovered starting material, 2.5:1 ratio of isomers from the dihydroxylation) as a colorless oil.

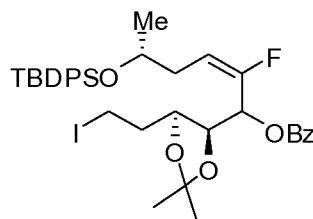
**13**

**Acetal 13:** To a solution of diol **12** (1.0 equiv, 577 mg, 0.823 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added 2-methoxy propene (2.0 equiv, 0.15 mL, 1.64 mmol) followed by PPTS (0.1 equiv, 20.7 mg, 0.082 mmol) at 23 °C and the resulted solution was stirred 1 hour . Then the reaction was quenched with sat. aq. NaHCO<sub>3</sub> soln. (5 mL), stirred for 10 min, the layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 5 mL). The combined organic layers were washed with brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The crude product was purified by flash chromatography (petroleum ether/EtOAc 4:1) to afford the acetal **13** (566 mg, 93%) as a colorless oil.

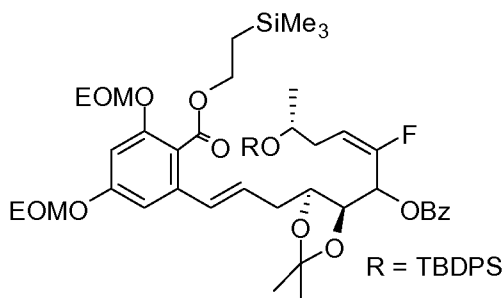
**25**

**Alcohol 25:** To a solution of PMB-ether **13** (1.0 equiv, 566 mg, 0.763 mmol) in CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O 2/1 (35 mL) was added DDQ (1.2 equiv, 208 mg, 0.916 mmol) at 23 °C. After stirring for 3 hours, the reaction mixture was diluted with Et<sub>2</sub>O (10 mL), quenched with sat. aq. NaHCO<sub>3</sub> (10 mL) and stirred for 10 min. Then the layers were separated and the aqueous layer was extracted with Et<sub>2</sub>O (3 x 5 mL). The combined organic layers were washed with H<sub>2</sub>O (2 x 10 mL) and brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were removed under reduced pressure. The residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc, 9:1, 3:1) to obtain desired alcohol **25** (434 mg, 92%) as light yellow oil.

59

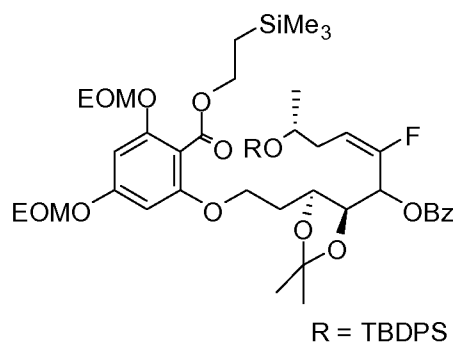
**14**

**Alkyl iodide 14:** To a solution of alcohol **25** (1.0 equiv, 434 mg, 0.699 mmol) in  $\text{CH}_2\text{Cl}_2$  (5 mL) were added sequentially imidazole (3.0 equiv, 142.7 mg, 2.09 mmol), triphenylphosphine (1.8 equiv, 330 mg, 1.26 mmol) and  $\text{I}_2$  (1.8 equiv, 319.2 mg, 1.26 mmol) at 23 °C and the resulted mixture was stirred for 1.5 hours. Then, the reaction was quenched with sat. aq.  $\text{NaHCO}_3$  (5 mL), the layers were separated and the aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$  (3 x 5 mL). The combined organic layers were washed with sat. aq.  $\text{Na}_2\text{S}_2\text{O}_3$  solution (10 mL), brine (10 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (petroleum ether/EtOAc 9:1) to afford the alkyl iodide **14** (460 mg, 90%) as a light yellow oil.

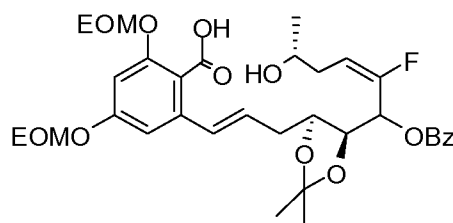
**17**

**Synthesis of compound 17:** To a solution of aromatic seleno-ether **15** (1.0 equiv, 334 mg, 0.62 mmol) in THF/HMPA 10/1 (5 mL) cooled at -78 °C was added a freshly prepared solution of LDA in THF (2.0 equiv, 2.58 mL, 1.24 mmol, 0.48 M in THF). The resulted dark red color mixture was stirred for 10 min at -78°C and a solution of the precooled alkyl iodide **14** (1.0 equiv, 453 mg, 0.62 mmol) in THF (1 mL) was added dropwise at the same temperature. After stirring for 20 min, the reaction was quenched with sat. aq.  $\text{NH}_4\text{Cl}$  (5 mL), warmed the flask to room temperature, the layers were separated and the aqueous layer was extracted with EtOAc (3 x 5 mL). The combined organic layers were washed with brine (10 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvents were removed under reduced pressure. The residue was purified by flash chromatography ( $\text{SiO}_2$ , petroleum ether/EtOAc, 10:1) to obtain desired alkylated compound (569 mg, 80%) as mixture of diastereoisomers.  $R_f = 0.28$

(petroleum ether/EtOAc 5:1). To a stirred solution of previously prepared seleno-ether (1.0 equiv, 569 mg, 0.498 mmol) in THF (4.2 mL) was added H<sub>2</sub>O<sub>2</sub> 35% (2.0 equiv, 0.12 mL, 0.996 mmol) at 23 °C. After stirring for 3 hours, the reaction mixture was concentrated under reduced pressure and the crude product was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc, 5:1) to afford compound **17** (455 mg, 92%) as a yellow oil.

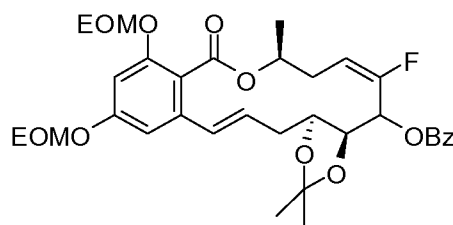
**18**

**Compound 18:** To a solution of phenol **16** (1.0 equiv, 192.3 mg, 0.497 mmol) in DMF (1.2 mL) was added K<sub>2</sub>CO<sub>3</sub> (2.0 equiv, 137.3 mg, 0.994 mmol) followed by a solution of iodide (1.1 equiv, 400 mg, 0.547 mmol) in DMF (1.2 mL) at 23 °C. The resulting mixture was heated to 80 °C and stirred for 12 hours. The reaction was then diluted with H<sub>2</sub>O (5 mL), stirred for 20 min, extracted with Et<sub>2</sub>O (5 x 5 mL), the combined organic layers were washed with brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvents were evaporated under reduced pressure. The residue was purified by flash chromatography (petroleum ether/EtOAc 9:1, 5:1) to afford compound **18** (490 mg, 99%) as a very light yellow oil.

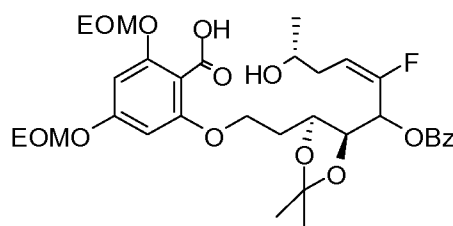
**26**

**Acid 26:** To a solution of silyl-ester **17** (1.0 equiv, 455 mg, 0.461 mmol) in THF (6.2 mL) was added tetra-butyl-ammonium fluoride (10.0 equiv, 4.6 mL, 4.6 mmol, 1M in THF) at 23 °C. The resulting mixture was stirred for 44 hours. The reaction was then quenched using sat. aq. NH<sub>4</sub>Cl (5 mL), the layers were separated, the aqueous layer extracted with EtOAc (5 x 5 mL), the combined organic layers were washed with H<sub>2</sub>O (2 x 5 mL), dried over Na<sub>2</sub>SO<sub>4</sub>,

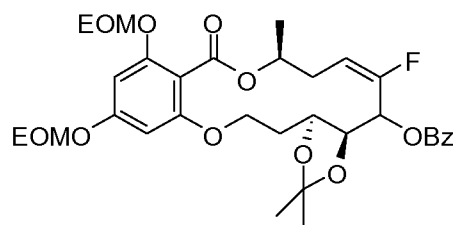
filtered and concentrated under reduce pressure. The resulted crude product was subjected for the next step with out any further purification.

**27**

**Macrocycle 27:** To a solution of acid **26** in toluene (45 mL) were added DIAD (2.0 equiv, 0.18 mL, 0.922 mmol) and  $\text{Ph}_3\text{P}$  (2.0 equiv, 241.8 mg, 0.922 mmol) at 23 °C and the resulting mixture was stirred for 1.5 hours. Then, the solvents were removed under reduced pressure and the residue was purified by flash chromatography ( $\text{SiO}_2$ , petroleum ether/EtOAc, 10:1, 6:1) to obtain macrocycle **27** (216 mg, 74%) as a light yellow oil.

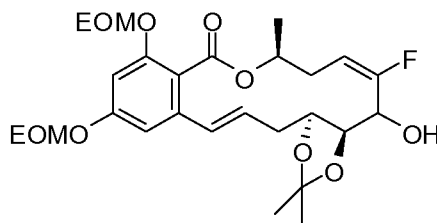
**28**

Following the same procedure as describe above silyl ester **18** (1.0 equiv, 490 mg, 0.495 mmol) in THF (6.7 mL), TBAF (15 equiv, 7.43 mL, 7.43 mmol, 1M in THF) was deprotected to obtain acid **28** that was submitted to the next step without any further purification.

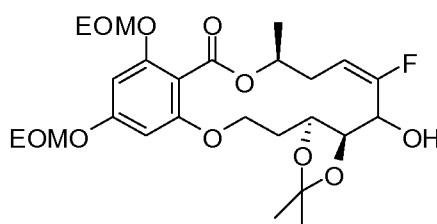
**29**

Following the same procedure as describe above acid **28** (1.0 equiv, 322 mg, 0.495 mmol) in toluene (47 mL) was treated with  $\text{PPh}_3$  (2.0 equiv, 259.6 mg, 0.99 mmol) and DIAD

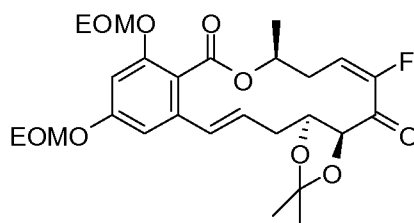
(2.0 equiv, 0.19 mL, 0.99 mmol) to obtain, after flash chromatography, macrocycle **29** (220 mg, 70%) as a colorless oil.

**30**

**Allylic alcohol 30:** A solution of benzoate **27** (1.0 equiv, 166 mg, 0.264 mmol) in 1% NaOH in MeOH (4 mL) was heated to 50 °C and stirred for 2 hours. Then, the solvents were removed under reduced pressure and the residue was purified by flash chromatography (SiO<sub>2</sub>, petroleum ether/EtOAc, 5:1, 4:1, 2:1) to obtain allylic alcohol **30** (105 mg, 76%) as light yellow oil.

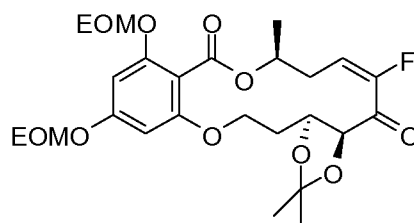
**31**

Following the same procedure as describe above alcohol **29** (1.0 equiv, 156 mg, 0.246 mmol) was treated with 1% NaOH in MeOH (3.2 mL) to yield allylic alcohols **31** (90 mg, 69%) as a colorless oil after flash chromatography.

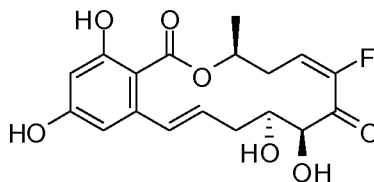
**19**

**Cis-enone 19:** To a solution of allylic alcohol **30** (1.0 equiv, 51 mg, 0.097 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) was added DMP (1.5 equiv, 61.5 mg, 0.145 mmol) at 23 °C. The resulting mixture was stirred for 12 hours, diluted with Et<sub>2</sub>O (5 mL), and the reaction was finally

quenched with 10% aq.  $\text{Na}_2\text{S}_2\text{O}_3$  (2 mL). After stirring the mixture for 10 min, the layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  (3 x 1 mL), the combined organic layers were washed with sat. aq.  $\text{NaHCO}_3$  (2 mL) and brine (2 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered and the solvents were evaporated under reduced pressure. The residue was purified by  
 5 flash chromatography ( $\text{SiO}_2$ , petroleum ether/ $\text{EtOAc}$ , 5:1) to obtain cis-enone **19** (45 mg, 88%) as a light yellow oil.

**20**

**Cis enone 20:** Following the procedure described above compound **31** (1.0 equiv, 30 mg, 0.056 mmol) in  $\text{CH}_2\text{Cl}_2$  (2 mL) was oxydized with DMP (1.5 equiv, 35.6 mg, 0.084 mmol) to yield cis-enone **20** (25 mg, 83%) as a very light yellow oil.

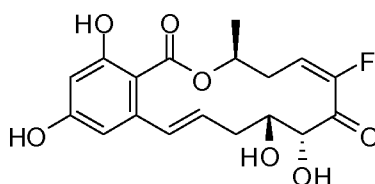
**1**

**Macrocycle 1:** Compound **19** (1.0 equiv, 40 mg, 0.076 mmol) was diluted with HF 40% in  $\text{H}_2\text{O}/\text{MeCN}$  (1/10, 4 mL) at 23 °C. The solution was stirred for 6.5 hours, diluted with  $\text{H}_2\text{O}$  (4 mL), and then dried by lyophilization. The residue was purified by flash chromatography ( $\text{SiO}_2$  was washed at least for 5 times with eluent prior to loading the  
 20 compound,  $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 25:1) affording desired macrocycle **1** (14 mg, 50%) along with a mixture of compounds at different stages of deprotection. This mixture was submitted for a second deprotection reaction using this time PS- $\text{SO}_3\text{H}$  resin/ $\text{MeOH}$  (10 equiv) at 50 °C. After 7 hours the resin was filtered and was washed with  $\text{CH}_2\text{Cl}_2$  (5 x 2 mL), the solvents were removed under reduced pressure and the residue was purified by HPLC [Supelco C18 (5cm x  
 25 10 mm, 5uM) 16 to 20 % acetonitrile in water, 2 mL/min] to afford macrocycle **32** (5.5 mg, 20%) as a colorless compound as well as 4.5 mg (15%) of **1**.

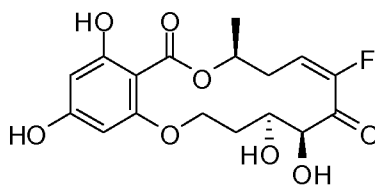
**1:**  $R_f$  = 0.19 ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  9:1);  $^1\text{H-NMR}$  ( $\text{MeOD}$ , 400 MHz):  $\delta$  = 6.99 (d,  $J$  = 15.3 Hz, 1H), 6.40 (d,  $J$  = 2.4 Hz, 1H), 6.28 (d,  $J$  = 2.4 Hz, 1H), 6.09 (ddd,  $J$  = 15.1, 8.9, 6.4 Hz,



1H), 5.95 (ddd,  $J = 21.9, 11.6, 2.7$  Hz, 1H), 5.28-5.20 (m, 1H), 4.90 (bs, 1H), 4.09-4.04 (m, 1H), 3.50-3.39 (m, 1H), 2.62-2.53 (m, 1H), 2.31-2.17 (m, 2H), 1.50 (d,  $J = 6.2$  Hz, 3H) ppm, 4OH signals not visible;  $^{13}\text{C}$  NMR (MeOD, 100 MHz, 25 °C):  $\delta = 196.4$  (d,  $^2J_{\text{C,F}} = 36.8$  Hz, C=O), 172.8 (C=O), 166.9 (C), 164.1 (C), 155.2 (2xd,  $^1J_{\text{C,F}} = 254.6$  Hz, C), 145.0 (C), 133.9 (CH), 132.3 (CH), 123.9 (d,  $^2J_{\text{C,F}} = 18.4$  Hz, CH), 109.4 (CH), 103.5 (C), 102.9 (CH), 81.7 (d,  $^3J_{\text{C,F}} = 3.8$  Hz, CH), 74.8 (CH), 74.2 (CH), 36.6 (CH<sub>2</sub>), 33.4 (d,  $^3J_{\text{C,F}} = 6.1$  Hz, CH<sub>2</sub>), 20.9 (CH<sub>3</sub>); HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{Na}]^+$  calcd for C<sub>18</sub>H<sub>19</sub>FO<sub>7</sub>Na: 389.1013; found: 389.1024.

**32**

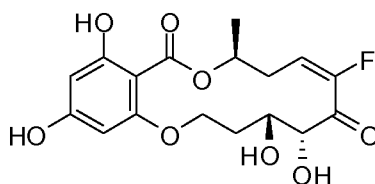
$^1\text{H}$ -NMR (MeOD, 400 MHz):  $\delta = 6.63$  (d,  $J = 15.3$  Hz, 1H), 6.49 (d,  $J = 2.4$  Hz, 1H), 6.30 (ddd,  $J = 15.3, 8.6, 6.7$  Hz, 1H), 6.26 (d,  $J = 2.4$  Hz, 1H), 6.19 (ddd,  $J = 22.8, 11.0, 4.0$  Hz, 1H), 5.39-5.28 (m, 1H), 4.90 (bs, 1H), 4.15-4.09 (m, 1H), 3.55-3.43 (m, 1H), 2.59 (dq,  $J = 17.2, 4.4$  Hz, 1H), 2.34-2.21 (m, 2H), 1.54 (d,  $J = 6.2$  Hz, 3H) ppm, 4OH signals not visible;  $^{13}\text{C}$  NMR (MeOD, 100 MHz, 25 °C):  $\delta = 196.7$  (d,  $^2J_{\text{C,F}} = 36.7$  Hz, C=O), 171.9 (C=O), 164.4 (C), 163.4 (C), 155.4 (2xd,  $^1J_{\text{C,F}} = 253.8$  Hz, C), 142.7 (C), 132.7 (CH), 130.3 (CH), 117.5 (d,  $^2J_{\text{C,F}} = 19.1$  Hz, CH), 108.0 (CH), 106.2 (C), 102.7 (CH), 80.8 (d,  $^3J_{\text{C,F}} = 3.7$  Hz, CH), 73.8 (d,  $^3J_{\text{C,F}} = 2.2$  Hz, CH), 72.7 (CH), 37.0 (CH<sub>2</sub>), 31.8 (d,  $^3J_{\text{C,F}} = 5.9$  Hz, CH<sub>2</sub>), 17.8 (CH<sub>3</sub>); HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{Na}]^+$  calcd for C<sub>18</sub>H<sub>19</sub>FO<sub>7</sub>Na: 389.1013; found: 389.1015.

**2**

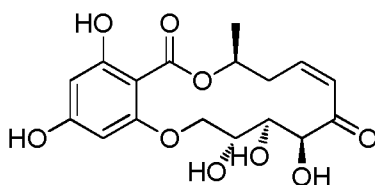
**Macrocycle 2:** To a solution of cis enone **20** (1.0 equiv, 26 mg, 0.049 mmol) in MeOH (3 mL) was added PS-SO<sub>3</sub>H resin (10.0 equiv, 164 mg, 0.493 mmol, 3.0 mmol/g) at 23 °C. The reaction mixture was shaken for 7 hours, then it was filtered washed with CH<sub>2</sub>Cl<sub>2</sub> (5 x 2 mL) and the solvents were removed under reduced pressure. The residue was purified

by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 25:1) to obtain desired macrocycle as a mixture of isomers (435 mg, 16 mg, 88%), which were separated by HPLC [Supelco C18 (5cm x 10 mm, 5uM) 16 to 20 % acetonitrile in water, 2 mL/min] to obtain pure macrocycles **2** and **33** in 60 and 20 % yield respectively.

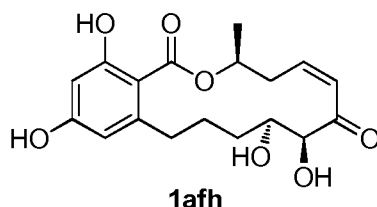
- 5        **2**:  $R_f = 0.18$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 9:1); <sup>1</sup>H-NMR (MeOD, 400 MHz):  $\delta = 6.02$  (d,  $J = 1.9$  Hz, 1H), 5.98 (d,  $J = 1.9$  Hz, 1H), 5.88 (ddd,  $J = 21.8, 11.8, 2.7$  Hz, 1H), 5.44-5.29 (m, 1H), 5.03 (d,  $J = 4.3$  Hz, 1H), 4.24-4.18 (m, 2H), 3.97-3.88 (m, 1H), 3.11-3.00 (m, 1H), 2.55-2.44 (m, 1H), 1.94-1.86 (m, 1H), 1.74-1.61 (m, 1H), 1.38 (d,  $J = 5.9$  Hz, 3H) ppm, 4OH signals not visible; <sup>13</sup>C NMR (MeOD, 100 MHz, 25 °C):  $\delta = 196.8$  (d,  $^2J_{C,F} = 35.9$  Hz, C=O), 172.6 (C=O), 167.1 (C), 165.4 (C), 163.6 (C), 155.6 (2xd,  $^1J_{C,F} = 255.3$  Hz, C), 119.5 (d,  $^2J_{C,F} = 17.6$  Hz, CH), 96.6 (CH), 96.4 (C), 92.8 (CH), 80.8 (d,  $^3J_{C,F} = 2.9$  Hz, CH), 73.1 (CH), 69.5 (CH), 65.8 (CH<sub>2</sub>), 33.1 (d,  $^3J_{C,F} = 5.9$  Hz, CH<sub>2</sub>), 31.3 (CH<sub>2</sub>), 20.8 (CH<sub>3</sub>); HRMS (MALDI-TOF)  $m/z$  [M + Na]<sup>+</sup> calcd for C<sub>17</sub>H<sub>19</sub>FO<sub>8</sub>Na: 393.0962; found: 393.0942.

**33**

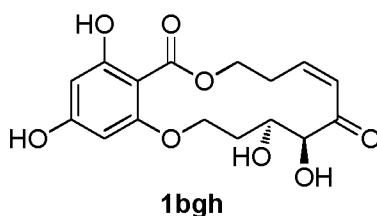
- 15        <sup>1</sup>H-NMR (MeOD, 400 MHz)  $\delta = 6.21$  (ddd,  $J = 23.4, 11.3, 4.6$  Hz, 1H), 6.04 (d,  $J = 2.1$  Hz, 1H), 5.99 (d,  $J = 2.1$  Hz, 1H), 5.34-5.25 (m, 1H), 5.05 (dd,  $J = 4.3, 1.6$  Hz, 1H), 4.36-4.25 (m, 2H), 3.89-3.82 (m, 1H), 3.21 (ddd,  $J = 15.3, 11.1, 2.9$  Hz, 1H), 2.54 (dq,  $J = 16.4, 4.4$  Hz, 1H), 1.89 (ddd,  $J = 15.0, 10.4, 4.2$  Hz, 1H), 1.68-1.62 (m, 1H), 1.32 (d,  $J = 6.2$  Hz, 3H) ppm, 4OH signals not visible; <sup>13</sup>C NMR (MeOD, 100 MHz, 25 °C):  $\delta = 196.3$  (d,  $^2J_{C,F} = 35.3$  Hz, C=O), 172.0 (C=O), 165.5 (C), 164.7 (C), 163.0 (C), 155.3 (2xd,  $^1J_{C,F} = 252.3$  Hz, C), 117.1 (d,  $^2J_{C,F} = 19.2$  Hz, CH), 98.3 (CH), 96.8 (C), 93.7 (CH), 80.7 (d,  $^3J_{C,F} = 2.3$  Hz, CH), 72.2 (CH), 69.9 (CH), 66.9 (CH<sub>2</sub>), 31.7 (d,  $^3J_{C,F} = 6.1$  Hz, CH<sub>2</sub>), 31.6 (CH<sub>2</sub>), 17.9 (CH<sub>3</sub>); HRMS (MALDI-TOF)  $m/z$  [M + Na]<sup>+</sup> calcd for C<sub>17</sub>H<sub>19</sub>FO<sub>8</sub>Na: 393.0962; found: 393.0985.

**1agi**

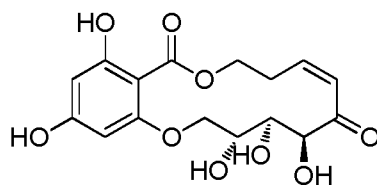
**1agi:**  $R_f = 0.6$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 4/1); <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz, 25 °C) δ 6.45 (dd,  $J = 11.8, 2.7$  Hz, 1H), 6.15 (td,  $J = 11.8, 2.7$  Hz, 1H), 5.93 (d,  $J = 2.54$  Hz, 1H), 5.92 (d,  $J = 2.4$  Hz, 1H), 5.36-5.29 (m, 1H), 4.74 (d,  $J = 1.3$  Hz, 1H), 4.13-4.09 (m, 1H), 4.03 (d,  $J = 1.4$  Hz, 1H), 3.82 (dd,  $J = 9.7, 2.7$  Hz, 1H), 3.56 (dd,  $J = 8.6, 2.7$  Hz, 1H), 3.31 (dt,  $J = 16.5, 11.3$  Hz, 1H), 2.43 (dq,  $J = 16.6, 2.4$  Hz, 1H), 1.38 (d,  $J = 5.9$  Hz, 3H), 5 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$  [M + H]<sup>+</sup> calcd for C<sub>17</sub>H<sub>20</sub>O<sub>9</sub>H: 369.1186; found: 369.1180.



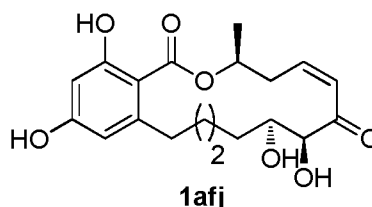
**1afh:**  $R_f = 0.29$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 10/1); <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz, 25 °C) δ 6.47 (dd,  $J = 11.6$  Hz, 2.4 Hz, 1H), 6.24 (td,  $J = 11.3$  Hz, 2.4 Hz, 1H), 6.21 (d,  $J = 2.1$  Hz, 1H), 6.15 (d,  $J = 2.4$  Hz, 1H), 5.43-5.35 (m, 1H), 4.49 (d,  $J = 1.6$  Hz, 1H), 3.93-3.87 (m, 1H), 3.04-2.92 (m, 1H), 2.69-2.49 (m, 2H), 1.85-1.72 (m, 1H), 1.70-1.56 (m, 2H), 1.39 (d,  $J = 6.4$  Hz, 3H), 1.39-1.35 (m, 2H), 4 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$  [M + H]<sup>+</sup> calcd for C<sub>18</sub>H<sub>22</sub>O<sub>7</sub>H: 351.1444; found: 351.1429.



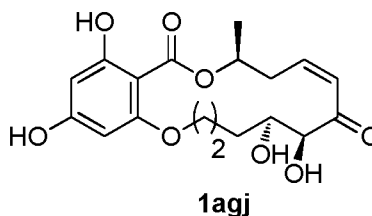
**1bgh:**  $R_f = 0.51$  (EtOAc/MeOH 10/1); <sup>1</sup>H NMR (CD<sub>3</sub>OD, 400 MHz, 25 °C) δ 6.48 (dd,  $J = 11.6, 2.8$  Hz, 1H), 6.15 (td,  $J = 11.2, 2.4$  Hz, 1H), 5.94 (d,  $J = 2.0$  Hz, 1H), 5.90 (d,  $J = 2.0$  Hz, 1H), 4.63 (s, 1H), 4.56 (t,  $J = 11.6$  Hz, 2H), 4.22-4.14 (m, 2H), 3.81-3.78 (m, 1H), 2.49-2.44 (m, 1H), 1.85-1.78 (m, 1H), 1.61-1.55 (m, 2H), 4 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$  [M + H]<sup>+</sup> calcd for C<sub>16</sub>H<sub>18</sub>O<sub>8</sub>H: 339.1080; found: 339.1056



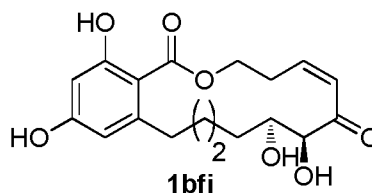
**1bgi:**  $R_f = 0.64$  ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  4/1);  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ , 400 MHz, 25 °C)  $\delta$  6.50 (dd,  $J = 11.8, 2.7$  Hz, 1H), 6.17 (td,  $J = 11.3, 2.7$  Hz, 1H), 5.95-5.92 (m, 2H), 4.77 (s, 1H), 4.62-4.52 (m, 2H), 4.20-4.12 (m, 2H), 3.88 (dd,  $J = 9.7, 2.7$  Hz, 1H), 3.54 (dd,  $J = 8.0, 2.7$  Hz, 1H), 2.54-2.45 (m, 1H), 1.99-1.85 (m, 1H), 5 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{16}\text{H}_{18}\text{O}_9\text{H}$ : 355.1029; found: 355.1054.



**1afj:**  $R_f = 0.26$  ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  10/1);  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ , 400 MHz, 25 °C)  $\delta$  6.46 (dd,  $J = 11.6$  Hz, 2.4 Hz, 1H), 6.29 (td,  $J = 10.4$  Hz, 2.4 Hz, 1H), 6.22 (d,  $J = 2.4$  Hz, 1H), 6.15 (d,  $J = 2.4$  Hz, 1H), 5.55-5.49 (m, 1H), 4.48 (d,  $J = 1.6$  Hz, 1H), 4.05-4.02 (m, 1H), 3.38-3.18 (m, 1H), 2.66-2.61 (m, 1H), 2.35-2.28 (m, 1H), 1.82-1.74 (m, 1H), 1.66-1.56 (m, 2H), 1.40 (d,  $J = 6.0$  Hz, 3H), 1.36-1.29 (m, 4H), 4 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{19}\text{H}_{24}\text{O}_7\text{H}$ : 351.1444; found: 351.1469.

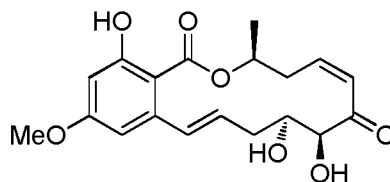


**1agj:**  $R_f = 0.43$  (15%  $\text{MeOH}/\text{CH}_2\text{Cl}_2$ );  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 400 MHz, 25 °C)  $\delta$  12.24 (s, 1H), 6.26 (dd,  $J = 11.2, 2.9$  Hz, 1H), 6.10 (td,  $J = 11.3, 2.1$  Hz, 1H), 6.03 (s, 1H), 5.93 (d,  $J = 2.1$  Hz, 1H), 5.49-5.43 (m, 1H), 4.57 (d,  $J = 3.5$  Hz, 1H), 4.05-3.92 (m, 3H), 2.99-2.89 (m, 1H), 2.63-2.57 (m, 1H), 1.92-1.83 (m, 2H), 1.59-1.43 (m, 2H), 1.38 (d,  $J = 5.9$  Hz, 3H), 3 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$   $[\text{M} + \text{H}]^+$  calcd for  $\text{C}_{18}\text{H}_{22}\text{O}_8\text{H}$ : 367.1393; found: 367.1378.

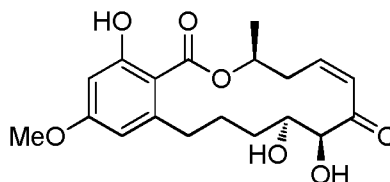


**1bfj:**  $R_f = 0.24$  ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  9/1);  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ , 400 MHz, 25 °C)  $\delta$  6.61 (d,  $J = 12.0$  Hz, 1H), 6.34-6.28 (m, 1H), 6.23 (d,  $J = 2.0$  Hz, 1H), 6.14 (d,  $J = 2.4$  Hz, 1H), 4.66-4.39 (m, 3H), 4.01-3.98 (m, 1H), 2.92-2.71 (m, 3H), 2.61-2.53 (m, 1H), 1.67-1.58 (m, 2H),

1.47-1.38 (m, 2H), 1.36-1.29 (m, 2H), 4 OH signals are not visible. HRMS (MALDI-TOF)  $m/z$   $[M + H]^+$  calcd for  $C_{18}H_{22}O_7H$ : 351.1444; found: 351.1469.

**3aejk**

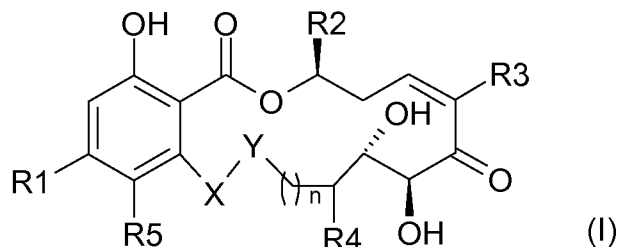
- 5 **3aejk.**  $R_f$  = 0.44 (3% MeOH in EtOAc);  $^1H$  NMR ( $CDCl_3$ , 500 MHz, 25 °C) 12.15 (s, 1H), 6.87 (d,  $J$  = 15.1 Hz, 1H), 6.40 (d,  $J$  = 2.4 Hz, 1H), 6.38 (d,  $J$  = 2.4 Hz, 1H), 6.33 (dd,  $J$  = 11.4, 2.6 Hz, 1H), 6.20 (td,  $J$  = 11.1, 2.6 Hz, 1H), 5.98 (ddd,  $J$  = 15.1, 8.4, 4.1 Hz, 1H), 5.27-5.21 (m, 1H), 4.50 (bs, 1H), 4.00-3.98 (m, 1H), 3.81 (s, 3H), 3.58 (dt,  $J$  = 17.0, 11.3 Hz, 1H), 2.50 (dq,  $J$  = 17.0, 2.1 Hz, 1H), 2.23-2.19 (m, 1H), 2.13 (ddd,  $J$  = 15.7, 10.8, 3.0 Hz, 1H), 1.47 (d,  $J$  = 6.1 Hz, 3H), 2 OH signals no visible;  $^{13}C$  NMR ( $CDCl_3$ , 125.75 MHz, 25 °C) 199.2, 171.5, 166.1, 164.4, 147.5, 143.1, 132.9, 130.2, 125.3, 108.2, 103.5, 100.4, 80.8, 73.8, 73.7, 55.6, 37.6, 37.2, 20.9; HRMS (MALDI-TOF)  $m/z$  385.1260 ( $[M+Na]^+$ ,  $C_{19}H_{22}O_7Na$  requires 385.1264).
- 10

**3afjk**

- 15 **3afjk.**  $^1H$  NMR ( $CDCl_3$ , 400 MHz, 25 °C) 12.18 (s, 1H), 6.39 (dd,  $J$  = 11.6, 3.2 Hz, 1H), 6.36 (d,  $J$  = 2.4 Hz, 1H), 6.29 (dd,  $J$  = 11.6, 2.4 Hz, 1H), 6.28 (d,  $J$  = 2.4 Hz, 1H), 5.48-5.40 (m, 1H), 4.56-4.54 (m, 1H), 4.21 (ddd,  $J$  = 16.4, 12.0, 4.8 Hz, 1H), 3.93 (bs, 1H), 3.82 (s, 3H), 3.75 (d,  $J$  = 6.8 Hz, 1H), 3.68-3.65 (m, 2H), 3.37 (dt,  $J$  = 17.5, 11.5 Hz, 1H), 2.96 (tm,  $J$  = 12.0 Hz, 1H), 2.57 (dq,  $J$  = 16.8, 2.8 Hz, 1H), 2.52-2.44 (m, 1H), 2.04-2.00 (m, 1H), 1.45 (d,  $J$  = 6.0 Hz, 3H), 1.12 (m, 1H); HRMS (MALDI-TOF)  $m/z$  365.1637 ( $[M+H]^+$ ,  $C_{19}H_{24}O_7H$  requires 365.1600).
- 20

## Claims

1- A compound of formula I, a tautomer or pharmaceutically acceptable salt, solvate, or ester thereof,



n is 0 or 1;

R1 is selected from the group consisting of hydrogen, halogen, cyano, -OR, -NRR', -NRS(O)R', -NRS(O<sub>2</sub>)R', -SR, -S(O)R, -S(O<sub>2</sub>)R, -OC(O)R, -C(O)R, -C(O)OR, -NRC(O)R', -C(O)NRR', -OC(O)OR, aliphatic, heteroaliphatic, acyl, aryl, heteroaryl, alkylaryl, arylalkyl, alkylheteroaryl and heteroarylalkyl;

R2 is hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

R3 is hydrogen or halogen;

R4 is hydrogen or hydroxyl;

R5 is hydrogen, halogen, OR, NHR, NH-COR,

R and R', independently, are hydrogen, or a C<sub>1</sub>-C<sub>5</sub> alkyl;

X-Y is CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, or CH<sub>2</sub>-S, and,

provided that the compound is not a compound of Formula I with

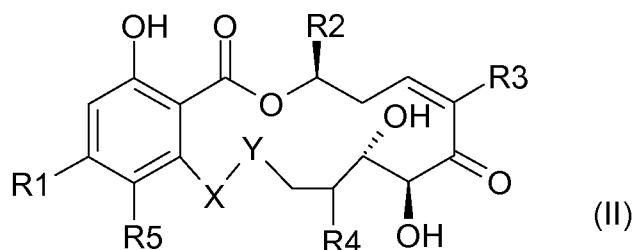
n is 0, R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, X-Y is O-CH<sub>2</sub> or CH<sub>2</sub>-S; or

n is 0, R1 is hydroxyl, R4 and R5 are hydrogen, R2 is methyl, R3 is fluorine, X-Y is O-CH<sub>2</sub>; or

n is 0, R1 is methoxy, R2 is methyl, R3 is fluorine, R4 and R5 are hydrogen, X-Y is CH<sub>2</sub>-S.

2- Compound of claim 1, wherein R1 is selected from the group consisting of -OR, -NHR and -SR.

3- Compound of claim 1 or 2, wherein the compound has the formula II,



R1, R2, R3, R4, R5 and X-Y being as defined in Claim 1 or 2.

4- Compound of claim 3, wherein the compound has one or several of the following features:

- R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl, preferably hydrogen; and/or
- R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>; and/or

provided that the compound does not have concomitantly a hydrogen in R2 and O-CH<sub>2</sub> in X-Y.

5- Compound of claim 3, wherein the compound has one or several of the following features:

- R1 is hydroxyl or methoxy; and/or
- R2 is hydrogen or methyl; and/or
- R3 is hydrogen; and/or
- R4 is hydrogen; and/or
- R5 is hydrogen; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>;

provided that the compound does not have concomitantly a hydrogen in R2 and O-CH<sub>2</sub> in X-Y.

6- Compound of any one of claims 3 to 5, wherein X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

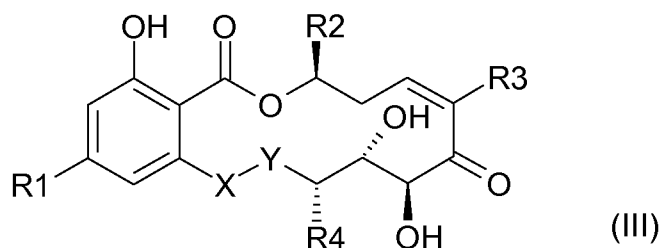
7- Compound of any one of claims 3 to 6, wherein R1 is hydroxyl or methoxy, R2, R3, R4 and R5 are hydrogen, and X-Y is CH=CH or CH<sub>2</sub>-CH<sub>2</sub>.

8- Compound of any one of claims 3 to 6, R1 is hydroxyl or methoxy, R2 is methyl, R3, R4 and R5 are hydrogen, and X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

9- Compound of claim 3, wherein the compound is selected from the group consisting of:

- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R3, R4 and R5 are hydrogen, R2 is methyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula II, wherein R1 is hydroxyl, R2, R3, R4 and R5 are hydrogen, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH=CH;
- a compound of formula II, wherein R3, R4 and R5 are hydrogen, R1 is methoxy, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>; and,
- a compound of formula II, wherein R2, R3, R4 and R5 are hydrogen, R1 is methoxy and X-Y is CH<sub>2</sub>-CH<sub>2</sub>.

10- Compound of claim 1 or 2, wherein the compound has the formula III,



R1, R2, R3, R4 and X-Y being as defined in Claim 1 or 2.

11- Compound of claim 10, wherein the compound has one or several of the following features:



- R1 is -OR and R is selected from hydrogen, methyl, ethyl and isopropyl, more preferably R1 is hydroxyl or methoxy; and/or
- R2 is selected from hydrogen, methyl, ethyl and isopropyl, more preferably from hydrogen or methyl, still more preferably is a methyl; and/or
- 5      - R3 is hydrogen or fluorine, preferably hydrogen; and/or
- R4 is hydrogen or hydroxyl; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, O-CH<sub>2</sub>, CH<sub>2</sub>-O, S-CH<sub>2</sub>, and CH<sub>2</sub>-S, preferably from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

10      12- Compound of claim 10, wherein the compound has one or several of the following features:

- R1 is hydroxyl; and/or
- R2 is hydrogen or methyl, preferably methyl; and/or
- R3 is hydrogen or fluorine, preferably hydrogen; and/or
- 15      - R4 is hydrogen or hydroxyl; and/or
- X-Y is selected from CH=CH, CH<sub>2</sub>-CH<sub>2</sub>, and O-CH<sub>2</sub>.

13- Compound of any one of claims 10 to 12, wherein R2 is methyl, ethyl and isopropyl, preferably methyl, and R4 is hydroxyl.

20

14- Compound of claim 10, wherein the compound is selected from the group consisting of:

- a compound of formula III, wherein R1 is hydroxyl, R3 and R4 are hydrogen, R2 is methyl, and X-Y is CH<sub>2</sub>-CH<sub>2</sub>;
- 25      - a compound of formula III, wherein R1 is hydroxyl, R3 is hydrogen, R2 is methyl, , R4 is hydroxyl, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R2, R3 and R4 are hydrogen, and X-Y is O-CH<sub>2</sub>;
- a compound of formula III, wherein R1 is hydroxyl, R2 and R3 are hydrogen, R4
- 30      is hydroxyl, and X-Y is O-CH<sub>2</sub>; and,
- a compound of formula III, wherein R1 is hydroxyl, R4 is hydrogen, R2 is methyl, , R3 is fluorine, and X-Y is CH=CH.

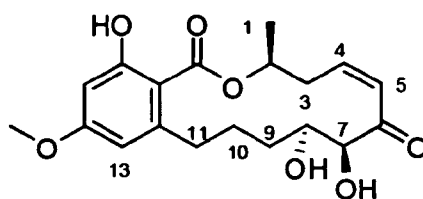
15- Compound of any one of claims 1-14 as a drug.

16- A pharmaceutical composition comprising a compound of any one of claims 1-14 and a pharmaceutically acceptable carrier.

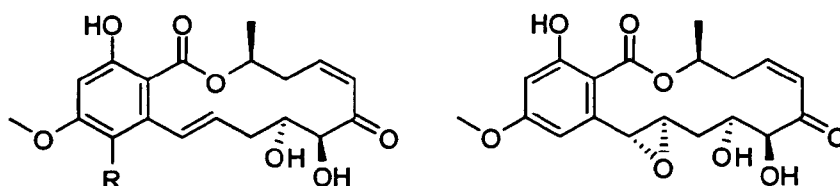
5           17- Compound of any one of claims 1-14 for use in the treatment of cancer.

18- Compound of any one of claims 3-9 for use in the treatment of cancer while preventing metastasis occurrence.

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L-783277 (FR265082)



R = H: LL-Z1640-2  
(5Z-7-oxozeaenol or FR148083)  
R = OMe: Radicicol A

hypothemycin

FIGURE 1

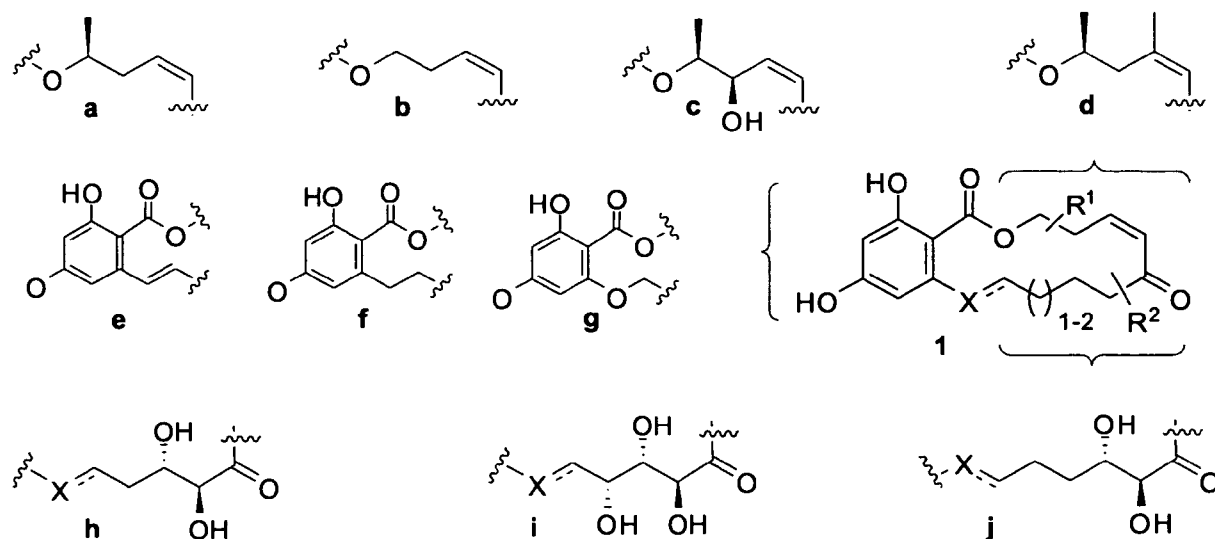
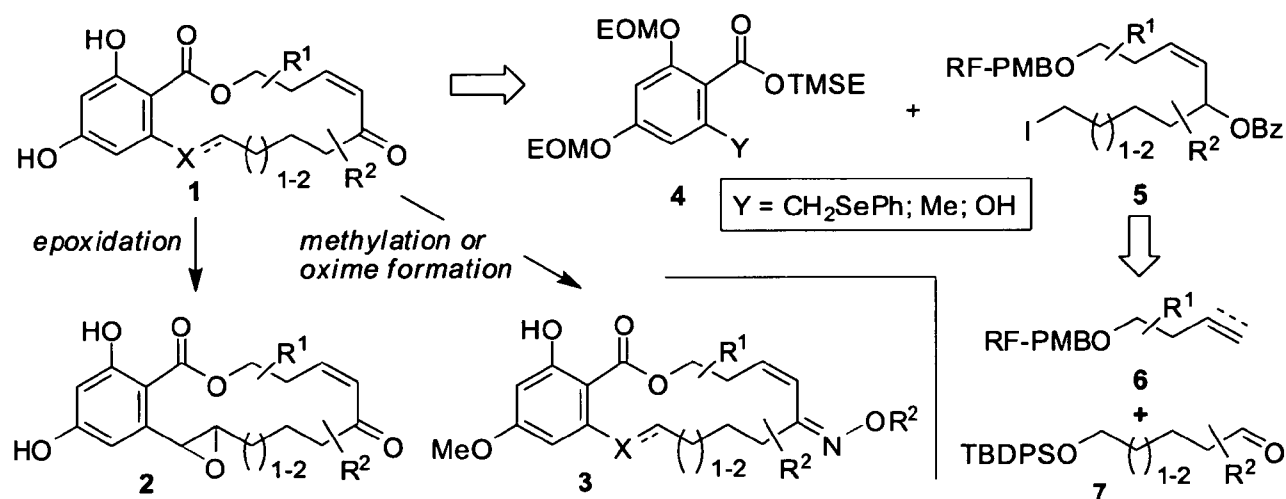


FIGURE 2

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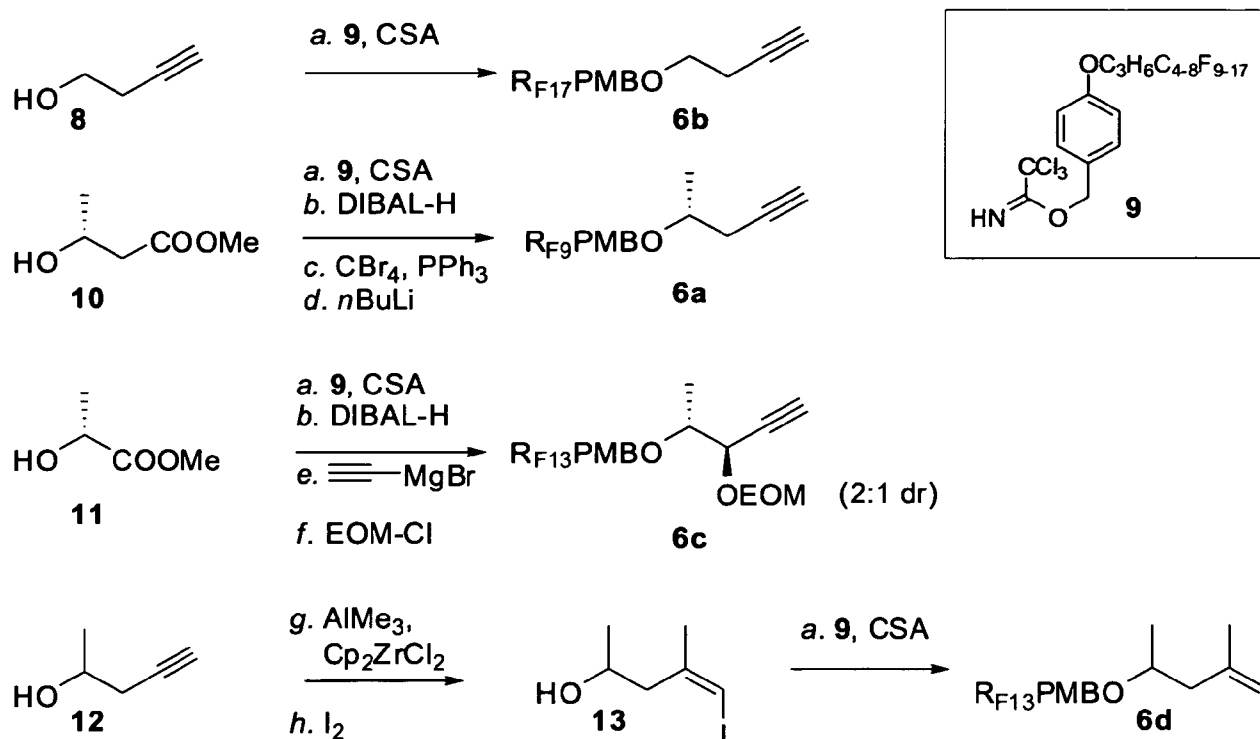


FIGURE 3

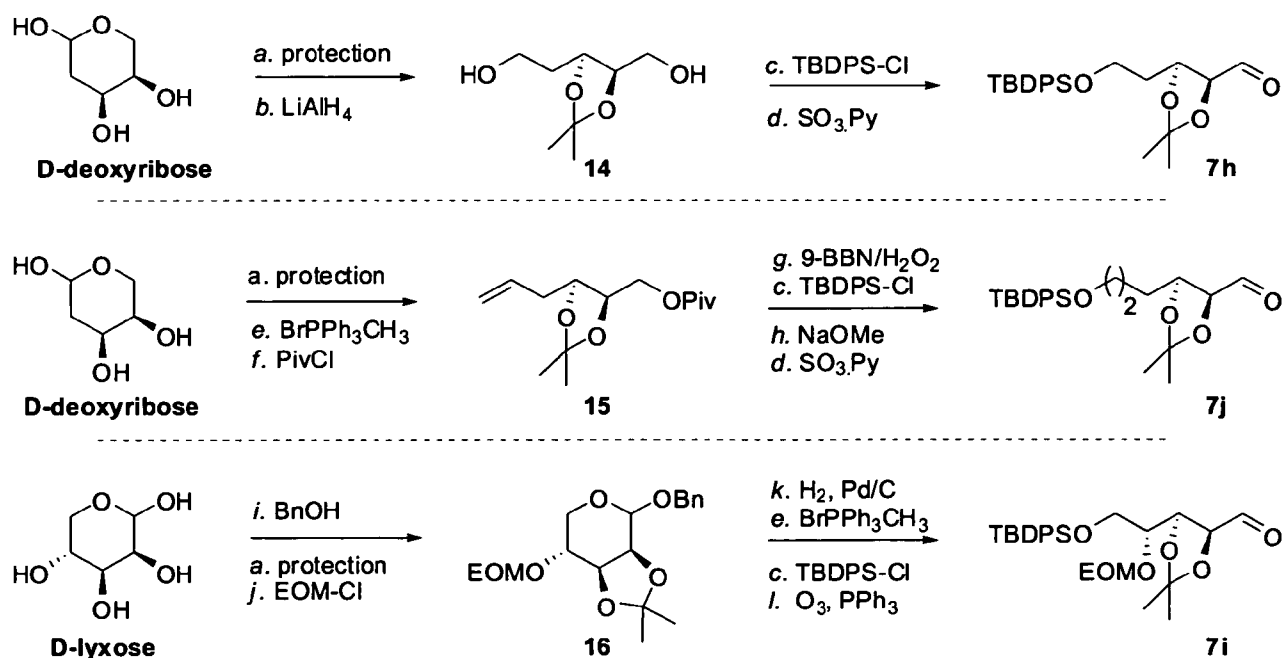


FIGURE 4

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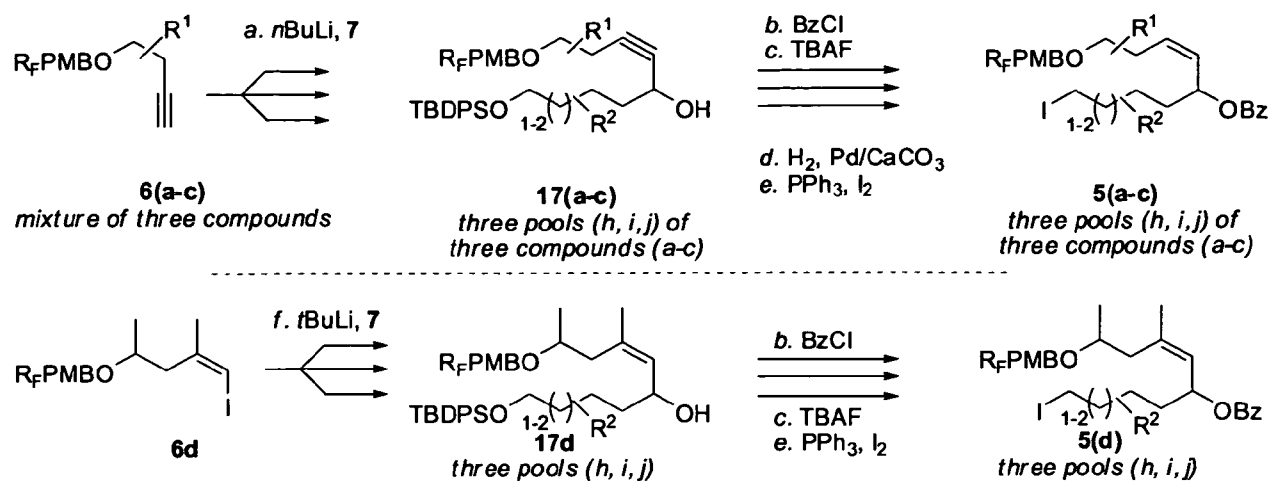


FIGURE 5

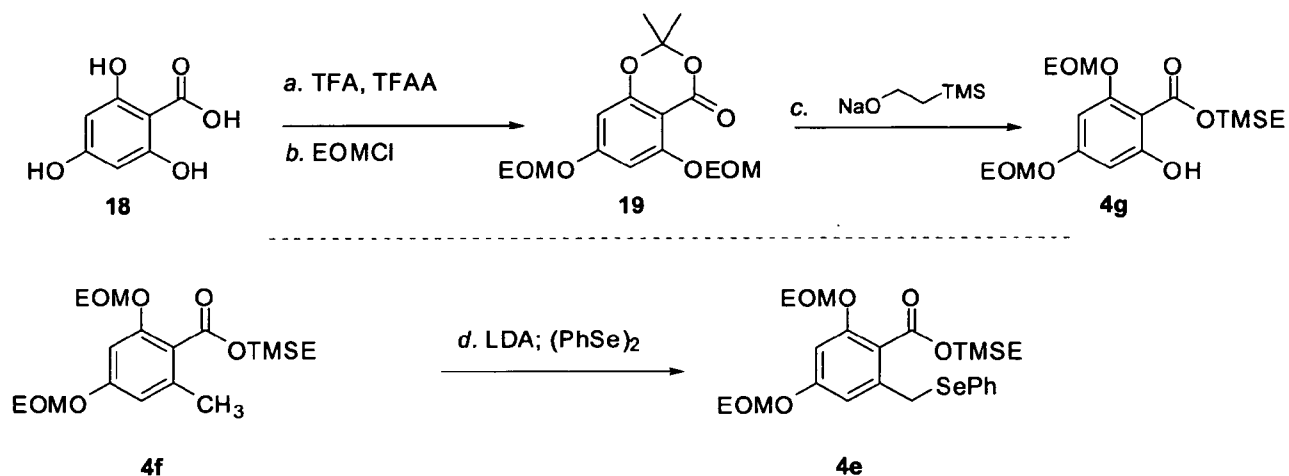


FIGURE 6

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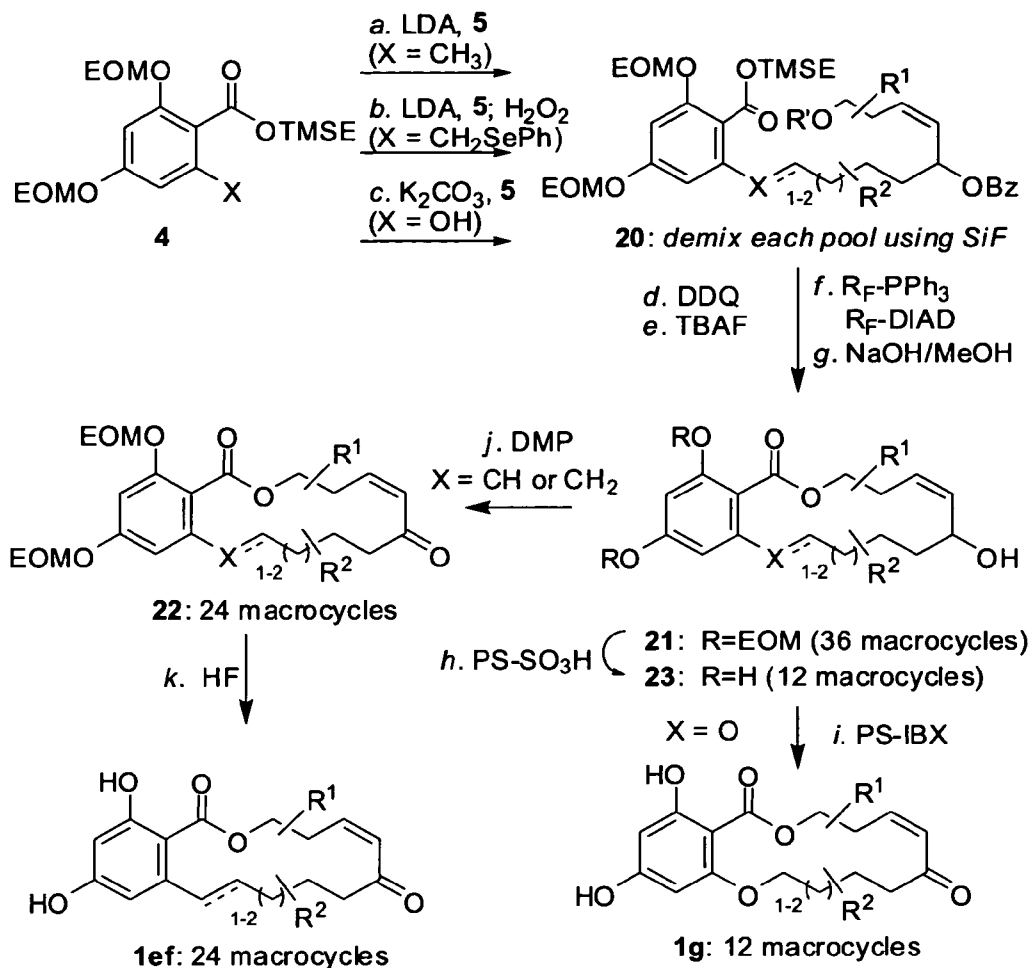


FIGURE 7

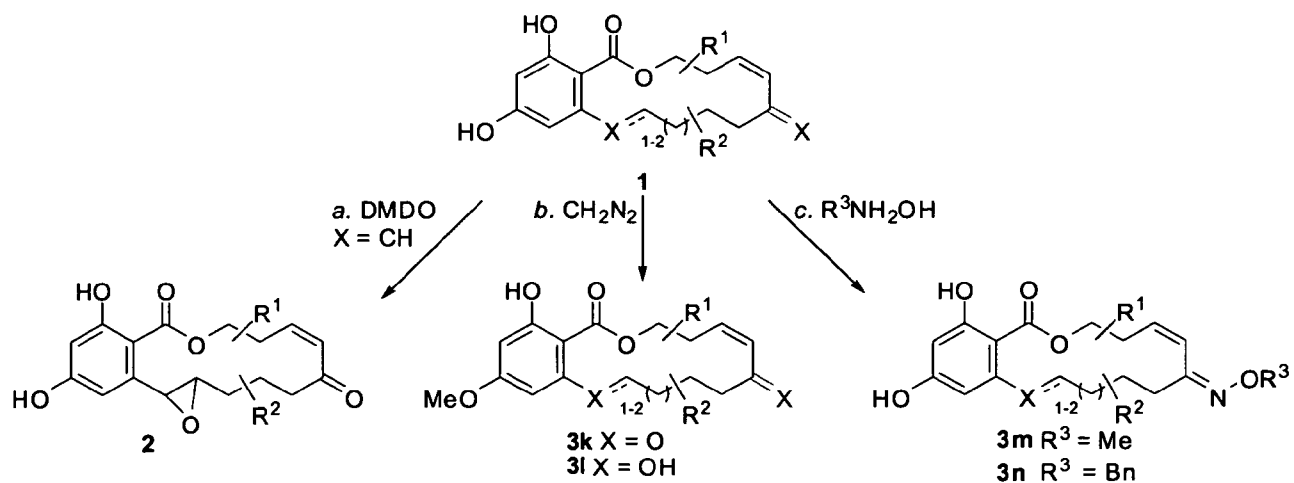


FIGURE 8

	VEGF-R2*	PDGFR- $\alpha$ *	VEGF-R3*	FLT3*	VEGF-R1*	MEK1	ISESEL*	KIT*	NLK*	GSK-3 $\alpha$ *	MAP KAPK5*	EGFR-3**	JNK3**	NEK2**	NIK**	SRC**	ZAP70**	CK2 $\alpha$	PKC- $\delta$	INSR
1	hypothemycin	0,009	0,012	0,018	0,033	0,053	0,089	0,189	1,212	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
2	LL-Z1640	0,010	0,015	0,021	0,024	0,042	0,110	0,150	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
3	LL-783277	0,169	0,227	0,253	0,369	2,113	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
4	1aej	0,016	0,022	0,035	0,026	0,058	>3	0,185	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
5	1afh	0,030	0,031	0,069	0,102	0,132	0,152	0,922	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
6	1afj	0,009	0,011	0,017	0,018	0,052	0,142	0,214	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
7	1agh	0,468	0,568	0,974	1,182	1,678	1,749	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
8	1agi	0,028	0,055	0,047	0,102	0,17	0,069	0,702	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
9	1agj	0,229	0,23	0,47	0,379	1,261	0,373	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
10	1bfj	0,044	0,152	0,11	1,271	0,198	1,511	0,676	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
11	1bgh	0,176	0,437	0,366	1,137	1,507	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
12	1bgi	0,208	0,647	0,379	2,213	1,279	0,75	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
13	1bgj	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
14	1cgh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
15	1deh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
16	1dfh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
17	1dfj	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
18	1dgh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
19	1dgi	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
20	2aj	1,142	1,565	1,66	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
21	3aejk	0,919	1,259	1,482	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
22	3aejkl	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
23	3aeim	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
24	3afik	0,013	0,020	0,022	0,033	0,061	0,749	0,192	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
25	3aghl	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
26	3agil	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
27	3agjl	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
28	3bfjk	0,040	0,074	0,072	0,939	0,195	2,34	0,701	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
29	3dejik	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
30	23agh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3
31	23dgh	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3	>3

\* part of the group of kinase described by Santi et al containing a cysteine in the same position as ERK2

\*\* kinases containing a cysteine in the ATP-binding pocket but at a different site than ERK2

FIGURE 9

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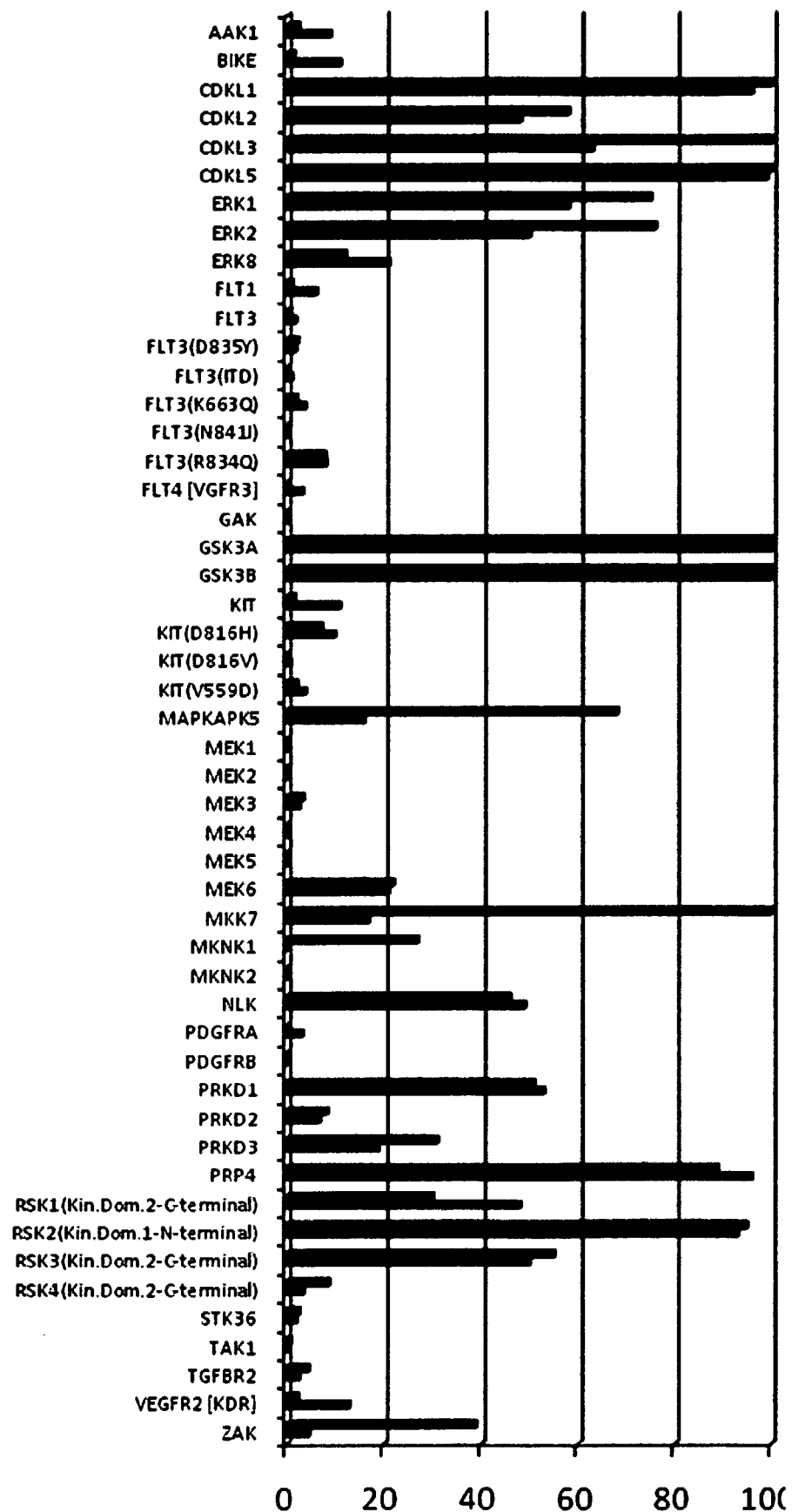


FIGURE 10



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kinase	1afh	1bgi
<b>FLT3</b>	<b>7</b>	<b>40</b>
<b>FLT3(D835H)</b>	<b>9</b>	<b>12</b>
<b>FLT3(D835Y)</b>	<b>0.4</b>	<b>2.8</b>
<b>FLT3(ITD)</b>	<b>3</b>	<b>26</b>
<b>FLT3(K663Q)</b>	<b>6</b>	<b>19</b>
<b>FLT3(N841I)</b>	<b>0</b>	<b>4.8</b>
<b>KIT</b>	<b>8.8</b>	<b>62</b>
<b>KIT(D816V)</b>	<b>0.9</b>	<b>13</b>
<b>KIT(L576P)</b>	<b>5.2</b>	<b>57</b>
<b>KIT(V559D)</b>	<b>7.8</b>	<b>38</b>
<b>KIT(V559D,T670I)</b>	<b>68</b>	<b>84</b>
<b>KIT(V559D,V654A)</b>	<b>66</b>	<b>100</b>

FIGURE 11

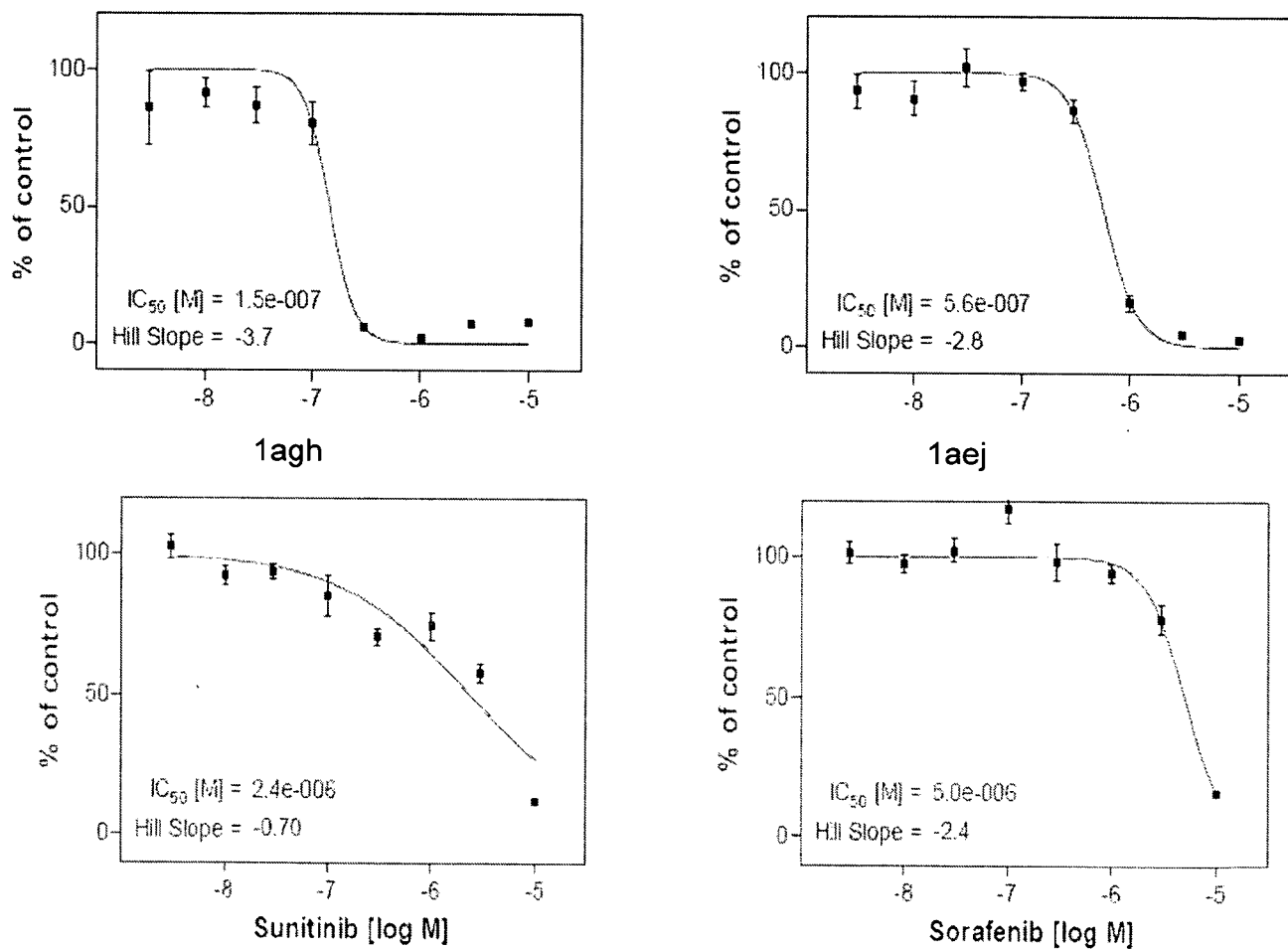


FIGURE 12

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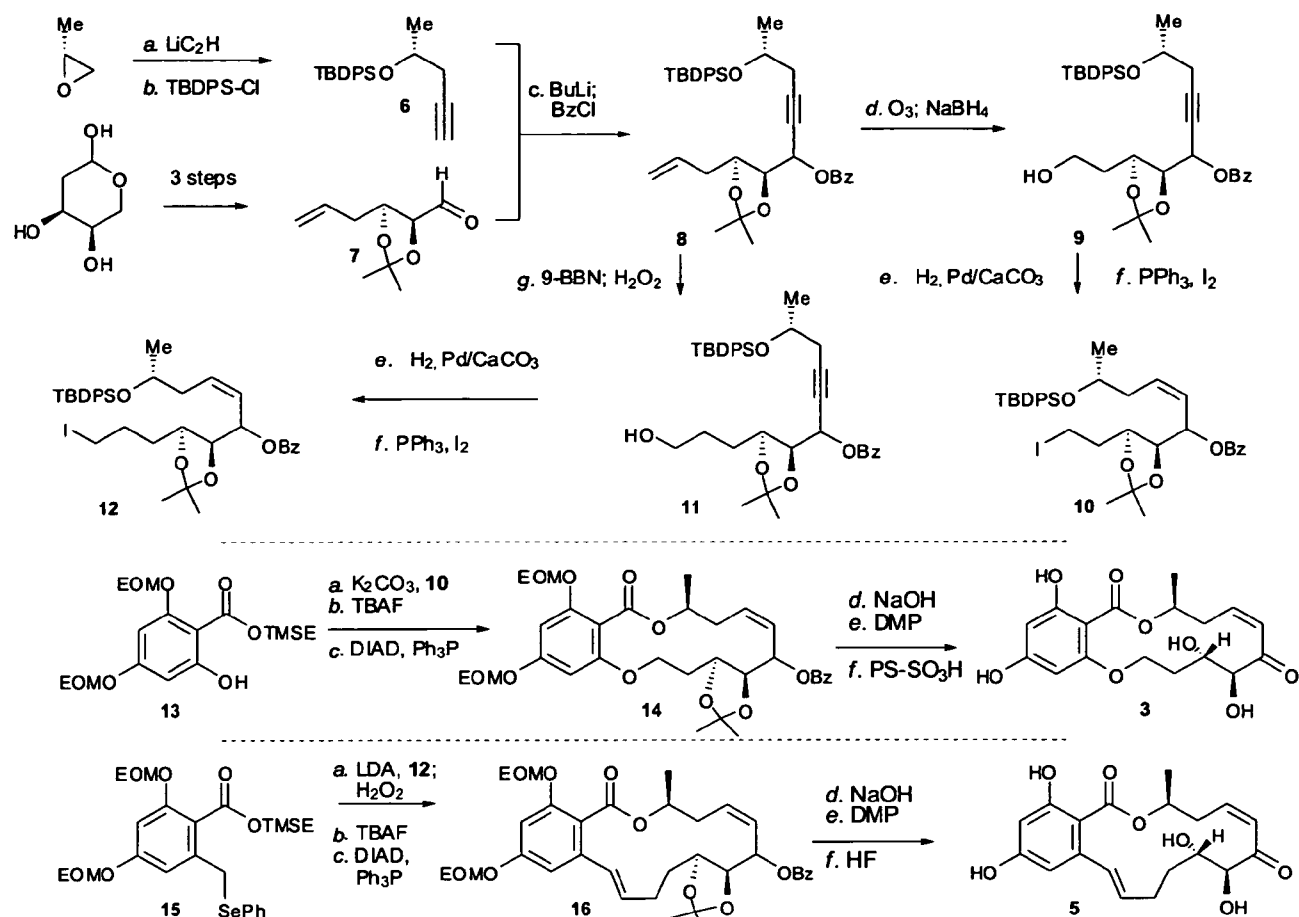


FIGURE 13

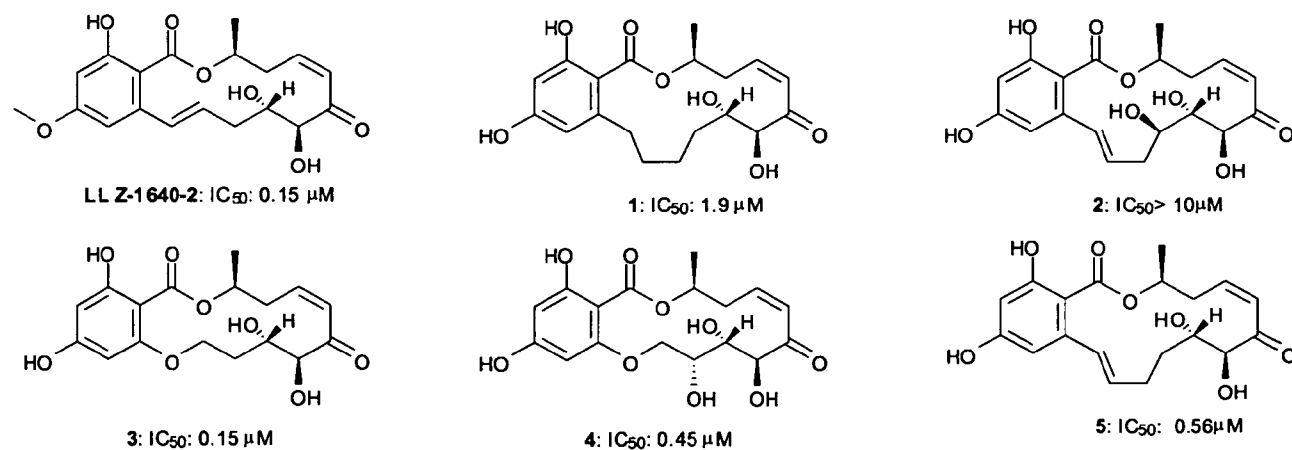


FIGURE 14

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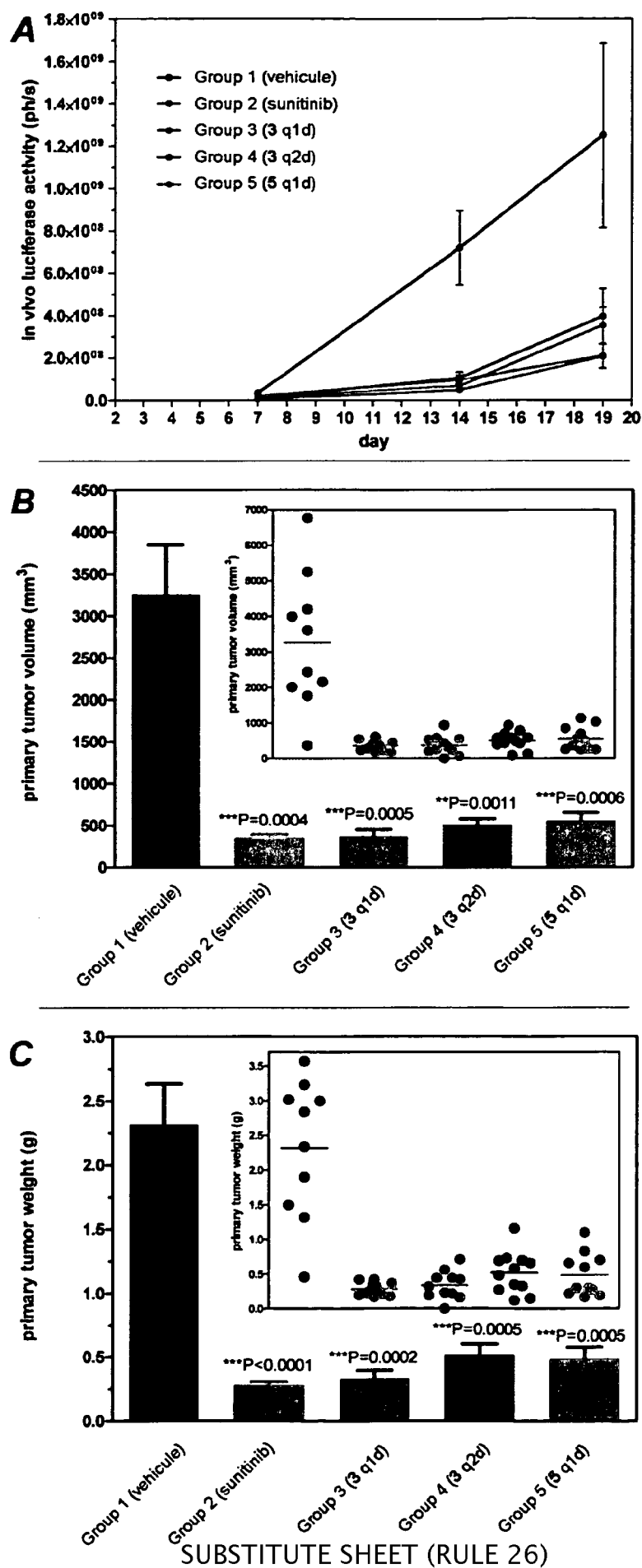


FIGURE 15

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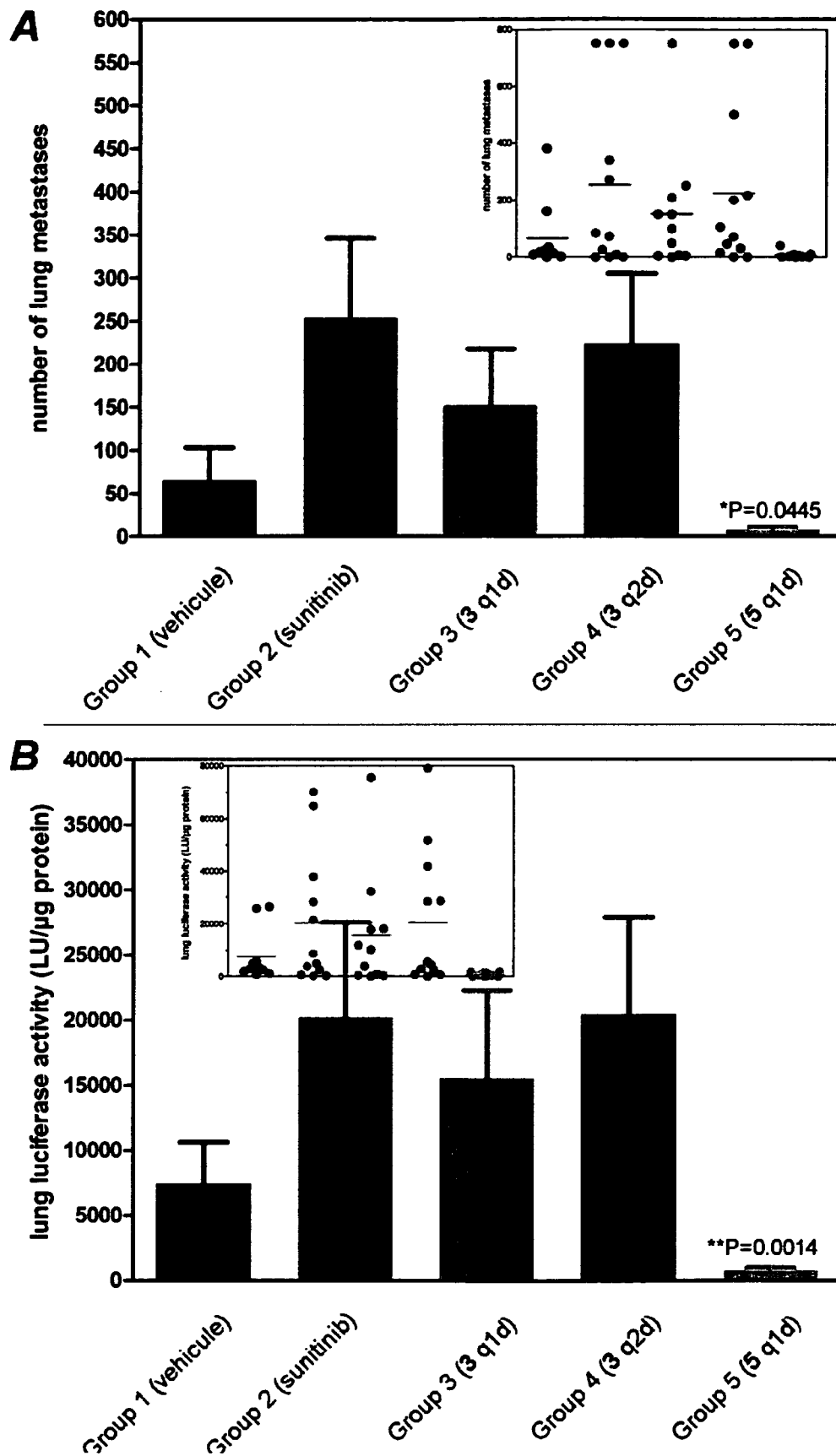


FIGURE 16

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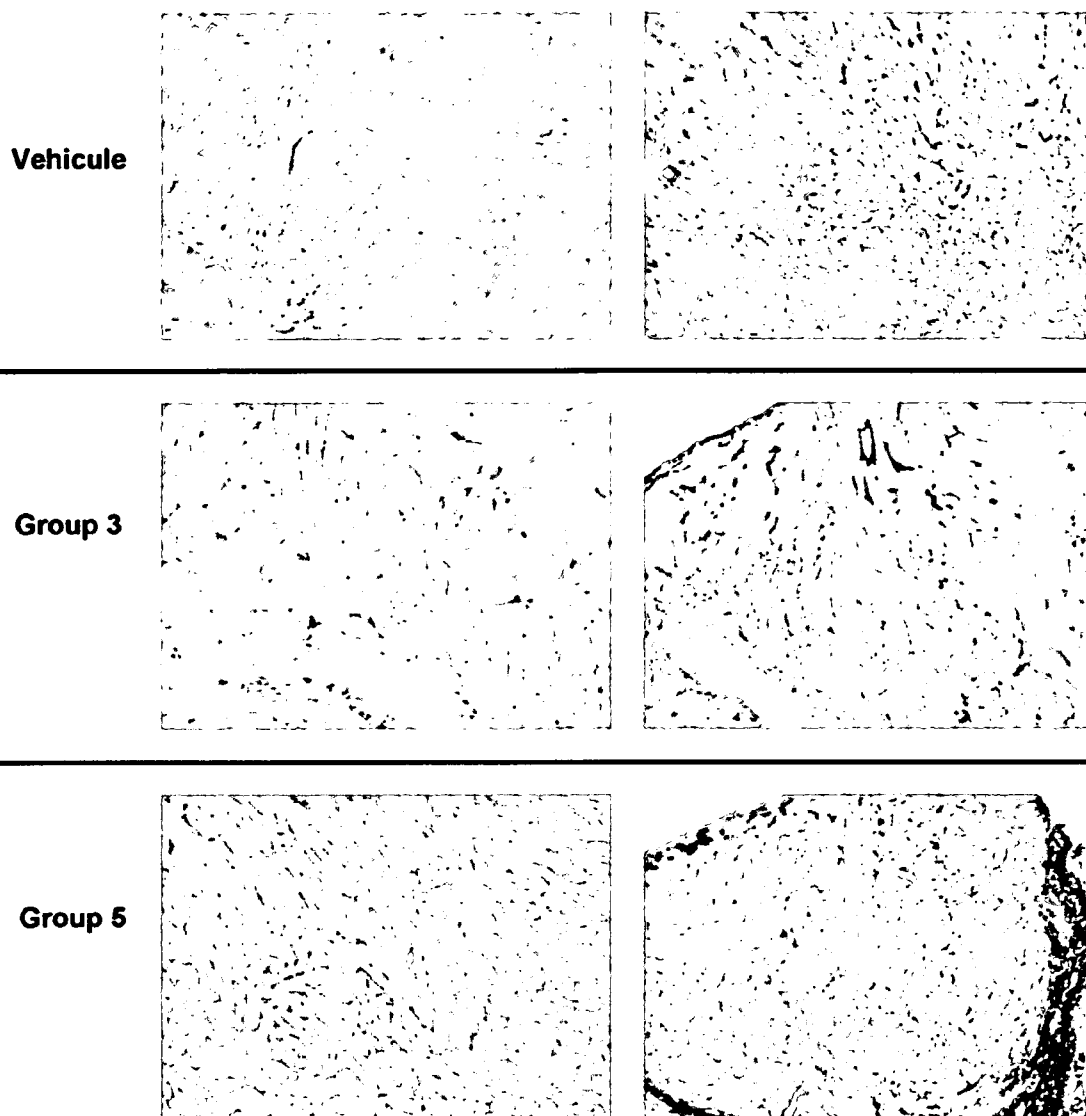


FIGURE 17

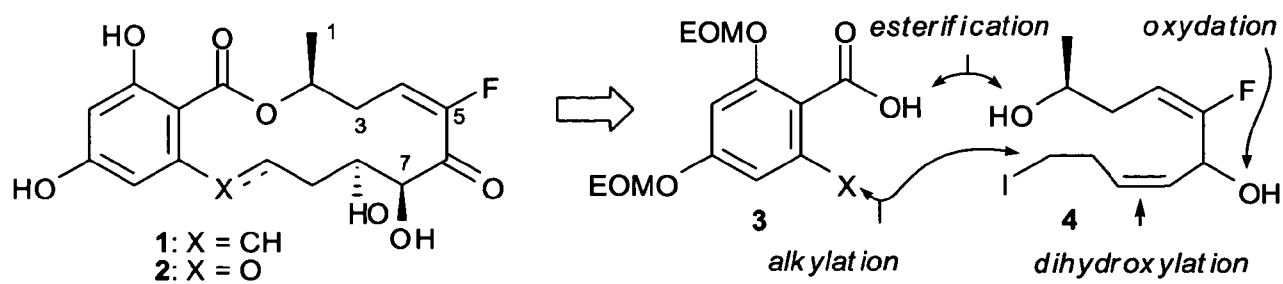


FIGURE 18

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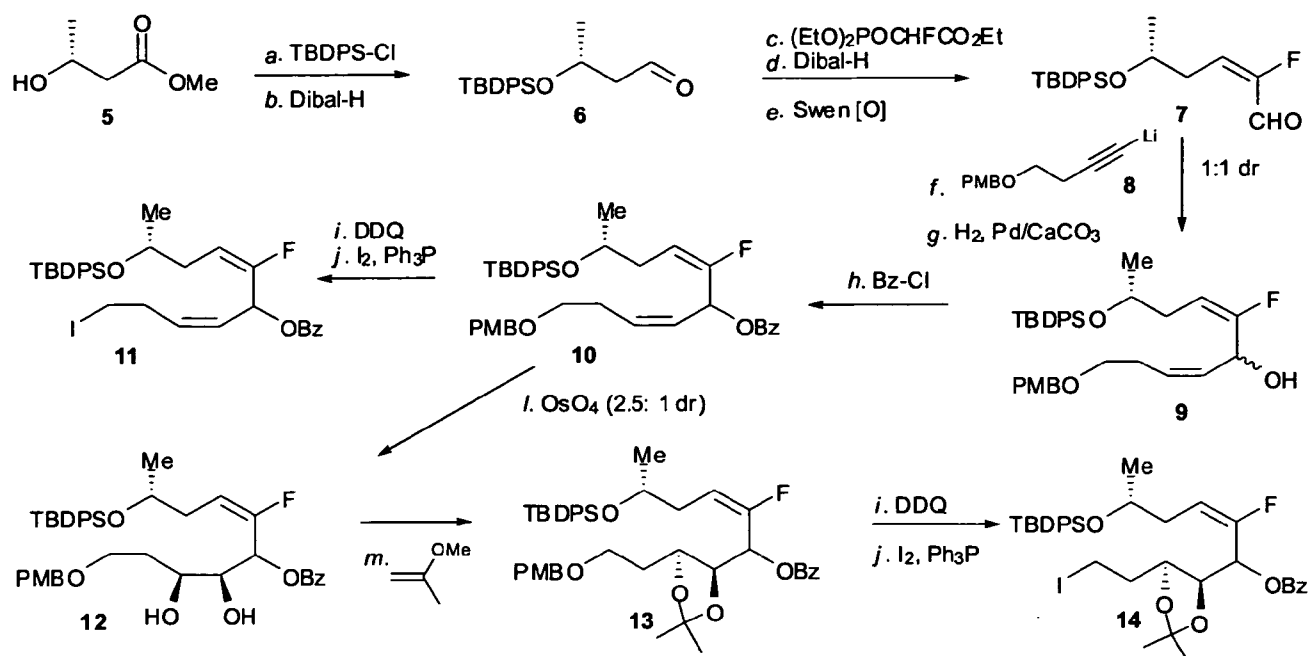


FIGURE 19A

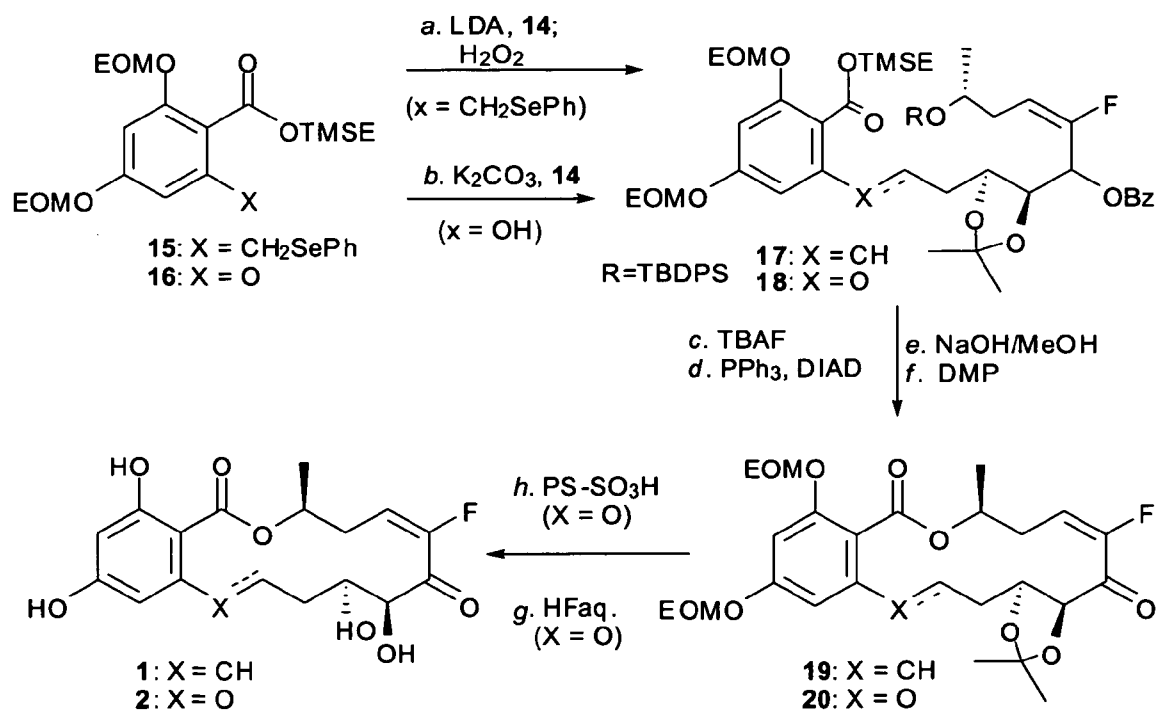


FIGURE 19B

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Compound	VEGF-R2 [nM]	KIT [nM]
LL-Z1640-2	2.63	57.8
fluoroenone 1	6.8	221
fluoroenone 2	60.5	1600

FIGURE 20

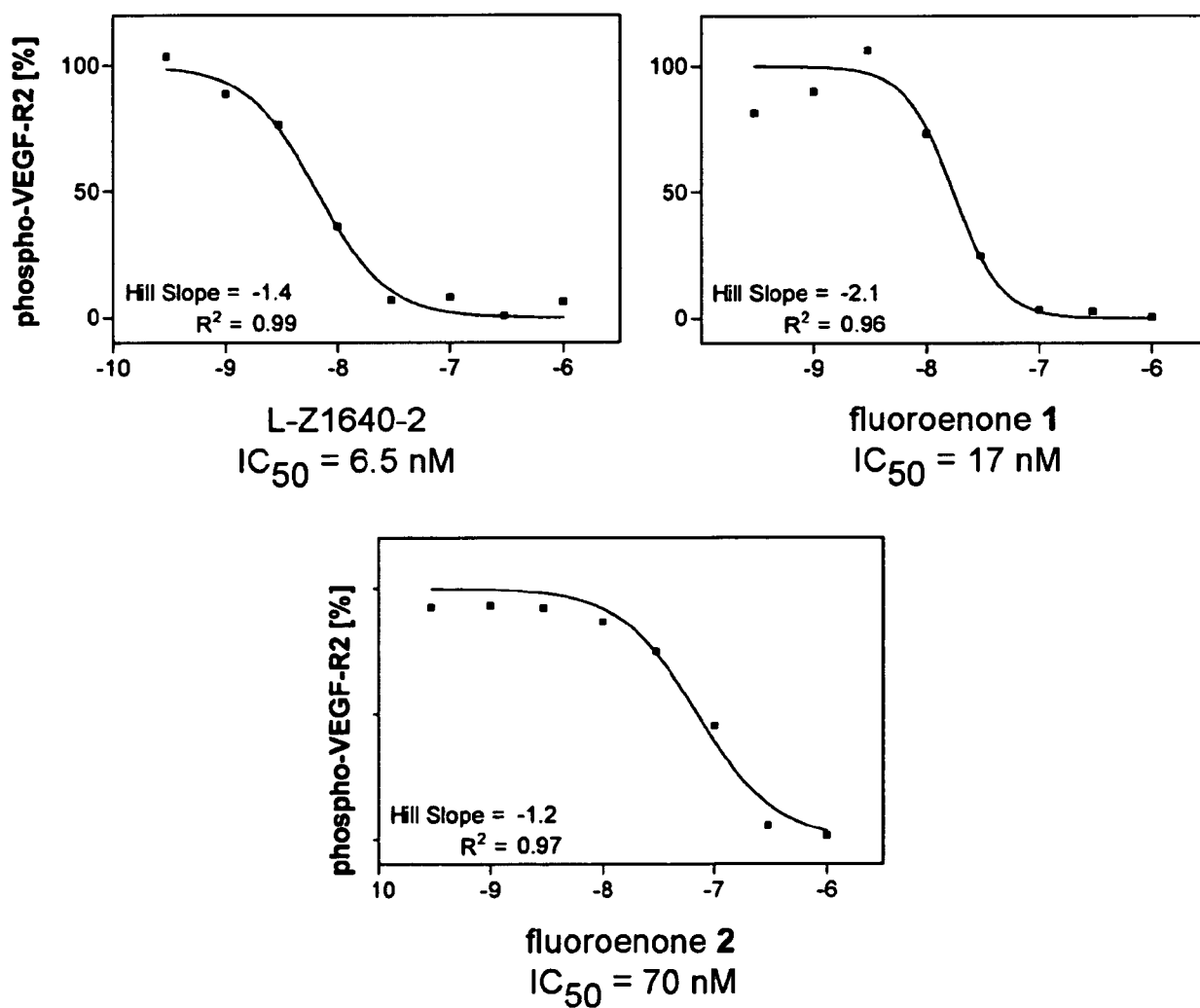


FIGURE 21

## INTERNATIONAL SEARCH REPORT

International application No

PCT/EP2010/064305

## A. CLASSIFICATION OF SUBJECT MATTER

INV. C07D313/00 A61K31/335 A61P35/00  
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, BEILSTEIN Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 02/48135 A1 (CHUGAI PHARMACEUTICAL CO LTD [JP]; TSUCHIYA MASAYUKI [JP]; OHTOMO TOSH) 20 June 2002 (2002-06-20) claim 2	1-18
X	WO 2006/036941 A2 (KOSAN BIOSCIENCES INC [US]; SANTI DANIEL V [US]; REID RALPH C [US]; HU) 6 April 2006 (2006-04-06) cited in the application claims 5,7-10,35  ----- -/--	1-18



Further documents are listed in the continuation of Box C.



See patent family annex.

## \* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance  
 "E" earlier document but published on or after the international filing date  
 "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)  
 "O" document referring to an oral disclosure, use, exhibition or other means  
 "P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

29 October 2010

Date of mailing of the international search report

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International application No

PCT/EP2010/064305

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