



- (51) International Patent Classification:
C09K 8/467 (2006.01)
- (21) International Application Number:
PCT/US2015/022757
- (22) International Filing Date:
26 March 2015 (26.03.2015)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
61/973,150 31 March 2014 (31.03.2014) US
- (71) Applicant (for CA only): **SCHLUMBERGER CANADA LIMITED** [CA/CA]; 125 - 9 Avenue SE, Calgary, Alberta T2G 0P6 (CA).
- (71) Applicant (for FR only): **SERVICES PETROLIERS SCHLUMBERGER** [FR/FR]; 42 rue Saint Dominique, 75007 Paris (FR).
- (71) Applicant (for GB, JP, NL only): **SCHLUMBERGER HOLDINGS LIMITED**; P.O. Box 71, Craigmuir Chambers, Road Town, Tortola, 1110 (VG).
- (71) Applicant (for all designated States except AE, AO, BH, CA, CN, FR, GB, GH, IN, JP, LY, MA, MZ, NA, NG, NL, PH, SD, SY, US, VN): **SCHLUMBERGER TECHNOLOGY B.V.** [NL/NL]; Parkstraat 83-89m, 2514 JG The Hague (NL).
- (71) Applicant (for AE, AO, BF, BH, BJ, CF, CG, CI, CM, CN, GA, GH, GN, GQ, GW, IN, KM, LY, MA, ML, MR, MZ, NA, NE, NG, PH, SD, SN, SY, TD, TG, VN only): **PRAD RESEARCH AND DEVELOPMENT LIMITED**; P.O. Box 71, Craigmuir Chambers, Road Town, Tortola, 1110 (VG).
- (71) Applicant (for US only): **SCHLUMBERGER TECHNOLOGY CORPORATION** [US/US]; 300 Schlumberger Drive, Sugar Land, Texas 77478 (US).
- (72) Inventors: **BULTE-LOYER, Helene**; Etudes et Productions Schlumberger, 1, rue Becquerel, BP 202, 92142 Clamart Cedex (FR). **REGNAULT DE LA MOTHE, Loic**; Etudes et Productions Schlumberger, 1, rue Becquerel, BP 202, 92142 Clamart Cedex (FR).
- (74) Agents: **DAE, Michael** et al.; 10001 Richmond Avenue, IP Administration Center of Excellence, Room 4720, Houston, Texas 77042 (US).
- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, JP, KE, KG, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC,

[Continued on next page]

(54) Title: METHODS FOR MAINTAINING ZONAL ISOLATION IN A SUBTERRANEAN WELL

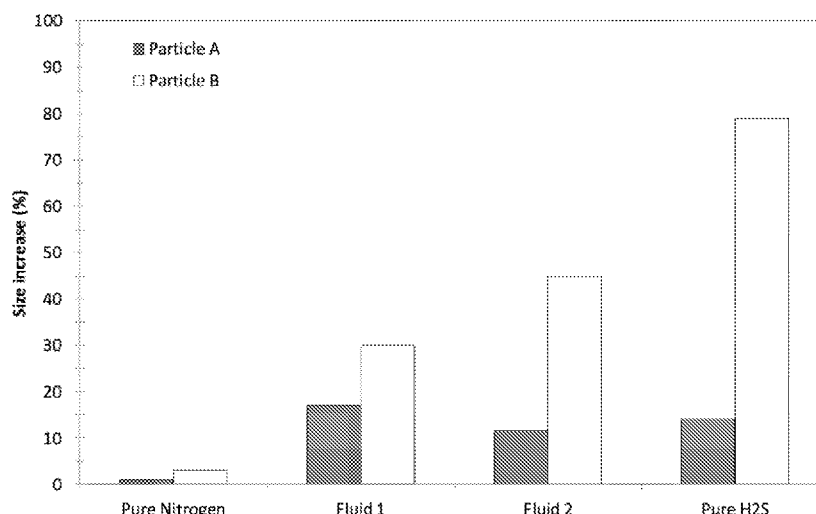


Figure 1

(57) Abstract: A cement for use in wells in which hydrogen sulfide is present, comprises polymer particles. In the event of cement-matrix failure, or bonding failure between the cement/casing interface or the cement/borehole-wall interface, the polymer particles swell when contacted by hydrogen sulfide. The swelling seals voids in the cement matrix, or along the bonding interfaces, thereby restoring zonal isolation.



SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN,
TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (*unless otherwise indicated, for every
kind of regional protection available*): ARIPO (BW, GH,
GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, ST, SZ,
TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU,
TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE,

DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT,
LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE,
SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA,
GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

Published:

— *with international search report (Art. 21(3))*

METHODS FOR MAINTAINING ZONAL ISOLATION IN A SUBTERRANEAN WELL

CROSS-REFERENCE TO RELATED APPLICATION

[0001] This application claims benefit of U.S. Provisional Patent App. Serial No. 61/973,150 filed March 31, 2014, which is herein incorporated by reference in its entirety.

BACKGROUND

[0002] The statements in this section merely provide background information related to the present disclosure and may not constitute prior art.

[0003] This disclosure relates to compositions and methods for treating subterranean formations, in particular, compositions and methods for cementing and completing wells in which hydrogen sulfide is present.

[0004] During the construction of subterranean wells, it is common, during and after drilling, to place a tubular body in the wellbore. The tubular body may comprise drillpipe, casing, liner, coiled tubing or combinations thereof. The purpose of the tubular body is to act as a conduit through which desirable fluids from the well may travel and be collected. The tubular body is normally secured in the well by a cement sheath. The cement sheath provides mechanical support and hydraulic isolation between the zones or layers that the well penetrates. The latter function is important because it prevents hydraulic communication between zones that may result in contamination. For example, the cement sheath blocks fluids from oil or gas zones from entering the water table and polluting drinking water. In addition, to optimize a well's production efficiency, it may be desirable to isolate, for example, a gas-producing zone from an oil-producing zone.

[0005] The cement sheath achieves hydraulic isolation because of its low permeability. In addition, intimate bonding between the cement sheath and both the tubular body and borehole is necessary to prevent leaks. However, over time the cement sheath can deteriorate and become permeable. Alternatively, the bonding between the cement sheath and the tubular body or borehole may become compromised. The principal causes of deterioration and debonding include physical stresses associated with tectonic movements, temperature changes, pressure changes inside the casing and chemical deterioration of the cement.

[0006] Some oil and gas fields have formations whose fluids contain acid gases such as carbon dioxide and hydrogen sulfide. Such wells may be challenging from a zonal isolation point of view.

[0007] Oil and gas that contains elevated amounts of hydrogen sulfide are called "sour." It has been estimated that 15 to 25% of natural gas in the United States may contain hydrogen sulfide. Worldwide, the percentage could be as high as 30%. Hydrogen sulfide is a toxic substance; therefore, it is important to prevent it from escaping through the cement sheath into aquifers or to the surface. Furthermore, hydrogen sulfide is corrosive to steel, and maintaining a competent cement sheath is essential to prevent casing deterioration during the life of the well.

SUMMARY

[0008] The present disclosure presents improvements by describing compositions that form a sustainable cement sheath even in a hydrogen-sulfide environment, and methods by which they may be prepared and applied in subterranean

wells.

[0009] In an aspect, embodiments relate to methods for maintaining zonal isolation in a subterranean well having a borehole in which hydrogen sulfide is present. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0010] In a further aspect, embodiments relate to methods for cementing a subterranean well having a borehole in which hydrogen sulfide is present. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0011] In yet a further aspect, embodiments relate to methods for completing a subterranean well having a borehole in which hydrogen sulfide is present. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0012] In yet a further aspect, embodiments relate to methods for cementing or completing a subterranean well having a borehole. The borehole is tested, analyzed or otherwise determined to contain hydrogen sulfide or is likely to develop hydrogen sulfide

in the future. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

BRIEF DESCRIPTION OF THE DRAWINGS

[0013] Figure 1 shows the swelling behavior of polypropylene particles and natural rubber/styrene-butadiene particles in the presence of nitrogen, pure H₂S and reservoir fluids containing H₂S.

[0014] Figure 2 is a section view of a test cell for the implementation of the test described in Example 2.

[0015] Figure 3 is a section view of a core holder placed in the test cell.

[0016] Figure 4 is a section view of a test cell provided with a sample of set cement wherein a crack has been created parallel to the axis of the core by fracturing in tension.

[0017] Figure 5 shows the ability of a cracked cement core containing polypropylene particles to reduce the flow of an invading gaseous reservoir fluid containing H₂S.

[0018] Figure 6 shows the swelling behavior of polypropylene, ground rubber, acrylonitrile-butadiene and styrene-isoprene-styrene in the presence of a H₂S-nitrogen mixture.

DETAILED DESCRIPTION

[0019] At the outset, it should be noted that in the development of any such actual embodiment, numerous implementation—specific decisions must be made to achieve the developer's specific goals, such as compliance with system related and business related constraints, which will vary from one implementation to another. Moreover, it will be appreciated that such a development effort might be complex and time consuming but would nevertheless be a routine undertaking for those of ordinary skill in the art having the benefit of this disclosure. In addition, the composition used/disclosed herein can also comprise some components other than those cited. In the summary and this detailed description, each numerical value should be read once as modified by the term "about" (unless already expressly so modified), and then read again as not so modified unless otherwise indicated in context. Also, in the summary and this detailed description, it should be understood that a concentration range listed or described as being useful, suitable, or the like, is intended that any and every concentration within the range, including the end points, is to be considered as having been stated. For example, "a range of from 1 to 10" is to be read as indicating each and every possible number along the continuum between about 1 and about 10. Thus, even if specific data points within the range, or even no data points within the range, are explicitly identified or refer to only a few specific, it is to be understood that the Applicant appreciate and understands that any and all data points within the range are to be considered to have been specified, and that the Applicant possessed knowledge of the entire range and all points within the range.

[0020] Cement systems that form durable cement sheath in an environment

containing hydrogen sulfide would be positively received by the industry. The Applicant has determined that cement compositions comprising materials that swell in the presence of hydrogen sulfide would respond to the industry challenges. When incorporated into a cement composition, such hydrogen sulfide swelling compounds may enable the cement sheath to close its own voids and/or cracks that may appear.

[0021] The Applicant has determined that certain polymers may fulfill the required swellable capacity in the presence of hydrogen sulfide. Such polymers may comprise natural rubber, nitrile rubber, styrene-butadiene rubber, polymers or copolymers comprising ethylene and/or propylene, butyl (isobutene-isoprene) rubber, hydrogenated nitrile butadiene rubber, acrylonitrile-butadiene copolymer, or combinations thereof. In some embodiments, such polymers comprise polypropylene, a blend of natural rubber and styrene-butadiene rubber or a combination thereof. In some embodiments, such polymers are polypropylene. In some embodiments, such polymers are a blend of natural rubber and styrene-butadiene rubber.

[0022] In an aspect, embodiments relate to methods for maintaining zonal isolation in a subterranean well having a borehole, in which hydrogen sulfide is present. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0023] In a further aspect, embodiments relate to methods for cementing a subterranean well having a borehole, in which hydrogen sulfide is present. A cement

slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0024] In yet a further aspect, embodiments relate to methods for completing a subterranean well having a borehole, in which hydrogen sulfide is present. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0025] In yet a further aspect, embodiments relate to methods for cementing or completing a subterranean well having a borehole. The borehole is tested, analyzed or otherwise determined to contain hydrogen sulfide or is likely to develop hydrogen sulfide in the future. A cement slurry, containing a material that swells when contacted by hydrogen sulfide, is pumped into the borehole. The cement slurry is allowed to set and harden. In the event of cement-matrix or bonding failure, the set cement is exposed to wellbore fluids that contain hydrogen sulfide. The material is allowed to swell, thereby restoring zonal isolation.

[0026] For all aspects, the borehole penetrates at least one fluid-containing reservoir, the reservoir containing fluid with a hydrogen sulfide concentration greater than about 5 moles per liter of fluid. The hydrogen sulfide may be supercritical, wet, dry or dissolved in oil or water. Hydrogen sulfide may also be liberated from the dissociation (aquathermolysis) of asphaltic or heavy hydrocarbons in the presence of steam in thermal

recovery wells. Such heavy crude oils may contain organic sulfides ($R-S-R'$), polysulfides ($R-S_x-R'$), thiophenes and thiols ($R-S-H$). Therefore, the present disclosure is applicable to wells in which the steam assisted gravity drainage (SAGD) or cyclic steam stimulation (CSS) processes are employed. For CSS applications, the swellable material may be a thermoset or thermoplastic material with a melting point higher than the steam injection temperature.

[0027] For all aspects, the material may comprise natural rubber, nitrile rubber, styrene-butadiene rubber, polymers or copolymers comprising ethylene and/or propylene, butyl (isobutene-isoprene) rubber, hydrogenated nitrile butadiene rubber, acrylonitrile-butadiene copolymer, or combinations thereof. Such polymers may comprise polypropylene, a blend of nitrile rubber and styrene-butadiene rubber or a combination thereof. In some embodiments, such polymers are polypropylene. In some embodiments, such polymers are a blend of natural rubber and styrene-butadiene rubber. The concentration of the material may be between about 5% and 50% by volume of solids in the cement slurry, also known as “by volume of blend (BVOB).” Or the range may be between about 10% and 40% BVOB. For optimal performance, the particle-size distribution of the material may be such that the average particle size is between about 10 μm and about 1000 μm . The average particle size may also be between about 100 μm and 900 μm .

[0028] Persons skilled in the art will recognize that the present use of polymers is different and distinct from their use as cement extenders (i.e., to reduce the amount of cement or to reduce the cement-slurry density) or as materials to improve cement flexibility.

[0029] The polymers may have Young's moduli between about 1 and 4000 MPa, or the polymers may have Young's moduli between about 1 and 2000 MPa. The Poisson's ratio of the polymers may be higher than 0.4, or Poisson's ratio of the polymers may be higher than 0.45.

[0030] For all aspects, the cement may additionally comprise one or more members of the list comprising Portland cement, calcium aluminate cement, fly ash, blast furnace slag, lime-silica blends, zeolites, geopolymers, Sorel cements, chemically bonded phosphate ceramics, polymer resins and mixtures thereof. The polymer resins may comprise epoxy resins, furan resins, phenolic resins and combinations thereof. The composition shall be pumpable. Those skilled in the art will recognize that a pumpable fluid in the context of well cementing has a viscosity lower than about 1000 mPa-s at a shear rate of 100 s^{-1} at the temperatures to which the fluid is exposed during a cementing operation, and during the time necessary to place the composition in the well. Also, the tubular body may comprise one or more members of the list comprising drillpipe, casing, liner and coiled tubing. In addition, the borehole may penetrate at least one fluid-containing reservoir, the reservoir preferably containing fluid with a hydrogen sulfide concentration greater than about five moles per liter.

[0031] The cement slurry may further comprise dispersing agents, fluid-loss-control agents, set retarders, set accelerators, foaming agents, gas generating agents, antifoaming agents, extenders, weighting agents, lost-circulation control agents, flexible particles, self-healing additives and combinations thereof. Other compounds may also be present such as coal, petroleum coke, graphite or gilsonite and mixtures thereof. Further, the hydrogen sulfide swellable polymers may be coupled to water super absorbent

polymers such as polymethacrylate, polyacrylamide and non-soluble acrylic polymers. A further association may be with one or more compounds from the list comprising an aqueous inverse emulsion of polymer comprising a betaine group, poly-2, 2, 1-bicyclo heptene (polynorbornene), alkylstyrene, crosslinked substituted vinyl acrylate copolymers, diatomaceous earth, vulcanized rubber, polyisoprene rubber, vinyl acetate rubber, polychloroprene rubber, acrylonitrile butadiene rubber, hydrogenated acrylonitrile butadiene rubber, ethylene propylene diene monomer, ethylene propylene monomer rubber, styrene-butadiene rubber, styrene/propylene/diene monomer, brominated poly(isobutylene-co-4-methylstyrene), butyl rubber, chlorosulphonated polyethylenes, polyacrylate rubber, polyurethane, silicone rubber, brominated butyl rubber, chlorinated butyl rubber, chlorinated polyethylene, epichlorohydrin ethylene oxide copolymer, ethylene acrylate rubber, ethylene propylene diene terpolymer rubber, sulphonated polyethylene, fluoro silicone rubbers, fluoroelastomers, substituted styrene acrylate copolymers and bivalent cationic compounds or any other particles such as those described in WO2004/101951 that swells when exposed to liquid hydrocarbons, the international application being incorporated herein by reference in its entirety. Further combinations may be made with thermoplastic block polymers including for example styrene-isoprene-styrene (SIS), styrene-butadiene-styrene (SBS) and mixtures thereof.

[0032] The cement slurry may further comprise materials that swell in the presence of another acid gas. The materials may be an elastomer comprising chlorofluorocarbons, tetrafluoroethylene-propylene copolymers, ethylene-propylene copolymers, isobutene-isoprene rubbers, nitrile rubbers, hydrogenated nitrile butadiene rubbers, or tetrafluoroethylene-perfluorovinyl methyl ether copolymers and combinations

thereof.

[0033] Persons skilled in the art will recognize that these methods may be performed during a primary cementing operation or a remedial cementing operation. The primary cementing operation comprises the installation of a tubular body inside the borehole of a well, or inside a tubular body. The pumping of the cement slurry may be performed the traditional way (i.e., the slurry is pumped down the casing and up the annulus) or by "reverse cementing," which consists of pumping the slurry down the annulus. Remedial cementing techniques for which the disclosed cement slurries may be applicable include plug cementing and squeeze cementing.

EXAMPLES

[0034] The following examples serve to further illustrate the disclosure. During the disclosed experiments, polymer particles were exposed to various types of fluids. The fluids were nitrogen, pure H₂S and four fluid mixtures (Fluids 1–4). The fluid-mixture compositions are given in Table 1.

	Fluid 1	Fluid 2	Fluid 3	Fluid 4
Component (mol%)				
CO ₂	4	4.954	4.94	0
H ₂ S	15	17.983	0	17
N ₂	0.8	0.978	18.76	83
C ₁	47.5	56.467	56.67	0
C ₂	7.4	8.778	8.60	0
C ₃	4.1	4.893	4.88	0
i-C ₄	0.9	1.027	1.04	0
n-C ₄	2.1	2.336	2.48	0
i-C ₅	0.9	0.912	0.90	0
n-C ₅	1.1	0.87	0.88	0
C ₆	1.3	0.803	0.85	0
C ₇	1.5	0	0	0
Toluene	0.2	0	0	0
C ₈	1.7	0	0	0

m&p-Xylene	0.25	0	0	0
o-Xylene	0.125	0	0	0
C ₉	1.3	0	0	0
C ₁₀	9.825	0	0	0
Total	100	100.001	99.94	100

Table 1. Compositions of Fluid Mixtures Employed in Particle Swelling Tests

EXAMPLE 1

[0035] Polymer particles were placed inside a pressure cell equipped with a window that allows one to observe the behavior of materials within the cell. The cell supplier is Temco Inc., located in Houston, Texas USA. The cell temperature is also adjustable. A camera captures images from inside the pressure cell, and image-analysis software is employed to interpret the behavior of materials inside the cell. After the polymer particles were introduced into the cell, the cell was sealed. Various types of fluids were introduced into the cell, and the camera recorded size changes of the particles when exposed to 60°C (140°F) and 34.5 MPa (5000 psi) for 360 min.

[0036] Two types of particles were tested: polypropylene (Particle A) and a blend of natural rubber and styrene-butadiene rubber (Particle B). The particles were exposed to four different test fluids: nitrogen, pure H₂S, and two reservoir fluids (Fluids 1 and 2) that contained H₂S. Their compositions are given in Table 1. Fluid 1 was liquid and Fluid 2 was gaseous under the test conditions.

[0037] The results, shown in Fig. 1, indicate that both particles swell significantly in the presence of H₂S, while swelling is minimal in the presence of nitrogen.

EXAMPLE 2

[0038] The following example demonstrates the ability of H₂S swellable particles embedded in a cement matrix to stop the flow of H₂S through cracks or

microannuli.

[0039] Experiments were performed in a sample test cell. As shown in Fig. 2, the test cell **10** is made of stainless steel and is substantially cylindrical. The outside diameter **11** of the cylinder is 38.1 mm, its inside diameter **12** is 21.2 mm and its length **13** is 180 mm. Test cement specimens are fabricated to fit inside the test cell. The test cell is positioned inside a core holder **20**, shown in Fig. 3. The core holder is a core flood system based on a Hassler sleeve that uses samples of 38.1-mm diameter. Such sleeves are well known in the art. Two volumetric pumps with a combined capacity of 32 mL/min deliver test fluids to the cell. A back-pressure regulator maintains a constant pressure at the outlet face of the cement sample.

[0040] The first cement composition was Class G cement + 10% BVOB Particle A + 45% BVOB silica + 10% BVOB microsilica + 1% by weight of blend (BWOB) magnesium oxide + 2L/tonne polyglycol antifoam agent + 1.3% BWOB polynaphthalene sulfonate dispersant + 0.65% BWOC lignosulfonate retarder + 85 L/tonne latex fluid-loss additive. Sufficient water was added to achieve a slurry density of 1950 kg/m³.

[0041] As shown in Fig. 4 cylindrical set-cement sample **40** was prepared and damaged such that cracks **43** extended along its longitudinal axis. The test cell **10** contained the cement sample in a thick metallic cylinder **41**. The core was held in place by surrounding sleeve **42**.

[0042] Fluid 2, described in Table 1, was pumped into the test cell at 8 mL/min. The cell conditions were 60°C and 34.5 MPa. The results, shown in Fig. 5, demonstrate that the cement sample self-healed within about 50 minutes exposure to the

flowing Fluid 2. The normalized flow rate through the cell fell by 97%. The experiment was repeated with Fluid 3, described in Table 1. Fluid 3 was gaseous at the cell conditions. Note that Fluid 3 does not contain H₂S. This time, the normalized flow rate through the cell increased by 10%.

[0043] The second cement composition was Class G cement + 10% BVOB Particle B + 35% BVOB silica + 10% BVOB microsilica + 1% by weight of blend (BWOB) magnesium oxide + 6L/tonne polyglycol antifoam agent + 0.5% BWOB polynaphthalene sulfonate dispersant + 0.57% BWOC lignosulfonate retarder + 85 L/tonne latex fluid-loss additive. Sufficient water was added to achieve a slurry density of 1950 kg/m³.

[0044] Fluid 2 was pumped through the sample as described above at 16 mL/min. The normalized flow rate through the cell fell by 82%. The experiment was repeated with Fluid 3. This time, the normalized flow rate through the cell decreased by 5%.

[0045] The results show that the presence of H₂S in the reservoir fluids was necessary for the polymer particles to provide adequate flow reduction.

EXAMPLE 3

[0046] Polymer particles were placed in the pressure cell described in Example 1. Five types of particles were tested: polypropylene (PP) (Icorene 9013P from ICO Polymers), ground rubber (RU) (RNM 45 from Rubber Resources), acrylonitrile-butadiene copolymer (ABC) (Chemigum P86F from Omnova Solutions), styrene-isoprene-styrene (SIS1) (D1114P from Kraton) and styrene-isoprene-styrene (SIS2) (D1119G from Kraton). Fluid 4 introduced into the cell, and the camera recorded size

changes of the particles when exposed to 60°C (140°F) and 34.5 MPa (5000 psi) for 360 min. The fluid was gaseous under the test conditions. The results, shown in Fig. 6, indicate that the particles swell significantly in the presence of H₂S.

[0047] Although various embodiments have been described with respect to enabling disclosures, it is to be understood that the preceding information is not limited to the disclosed embodiments. Variations and modifications that would occur to one of skill in the art upon reading the specification are also within the scope of the disclosure, which is defined in the appended claims.

CLAIMS

1. A method for maintaining zonal isolation in a subterranean well having a borehole in which hydrogen sulfide is present, comprising:
 - (i) pumping a cement slurry comprising a material that swells when contacted by hydrogen sulfide into the borehole;
 - (ii) allowing the cement slurry to set and harden;
 - (iii) in the event of cement-matrix or bonding failure, exposing the set cement to wellbore fluids that contain hydrogen sulfide; and
 - (iv) allowing the material to swell, thereby restoring zonal isolation.
2. The method of claim 1, wherein the material comprises natural rubber, nitrile rubber, styrene-butadiene rubber, polymers or copolymers comprising ethylene and/or propylene, butyl (isobutene-isoprene) rubber, hydrogenated nitrile butadiene rubber, acrylonitrile-butadiene copolymer, or styrene-isoprene-styrene, or combinations thereof.
3. The method of claim 1 or 2, wherein the concentration of the material in the cement slurry is between about 5 percent and about 50 percent by volume of solid blend.
4. The method of any one of claims 1–3, wherein the average particle size of the material is between about 10 μm and about 1000 μm .
5. The method of any one of claims 1–4, wherein the hydrogen sulfide is supercritical, wet, dry or dissolved in oil or water.
6. The method of any one of claims 1–5, wherein the borehole penetrates at least one fluid-containing reservoir, the reservoir containing fluid with a hydrogen sulfide concentration higher than about five moles per liter.

7. The method of any one of claims 1–6, wherein the cement slurry comprises one or more members selected from the list consisting of Portland cement, calcium aluminate cement, fly ash, blast furnace slag, lime-silica blends, zeolites, geopolymers, Sorel cements, chemically bonded phosphate ceramics and polymer resins.
8. The method of any one of claims 1–7, wherein the tubular body comprises one or more members selected from the the list consisting of drillpipe, casing, liner and coiled tubing.
9. A method for cementing a subterranean well having a borehole in which hydrogen sulfide is present, comprising:
 - (i) pumping an aqueous cement slurry comprising a material that swells when contacted by hydrogen sulfide into the borehole;
 - (ii) allowing the cement slurry to set and harden;
 - (iii) in the event of cement-matrix or bonding failure, exposing the set cement to wellbore fluids that contain hydrogen sulfide; and
 - (iv) allowing the material to swell, thereby restoring zonal isolation.
10. The method of claim 9, wherein the material comprises natural rubber, nitrile rubber, styrene-butadiene rubber, polymers or copolymers comprising ethylene and/or propylene, butyl (isobutene-isoprene) rubber, hydrogenated nitrile butadiene rubber, acrylonitrile-butadiene copolymer, or styrene-isoprene-styrene, or combinations thereof.
11. The method of claim 9 or 10, wherein the concentration of the material in the cement slurry is between about 5 percent and about 50 percent by volume of solid blend.

12. The method of any one of claims 9–11, wherein the average particle size of the material is between about 10 μm and about 1000 μm .
13. The method of any one of claims 9–12, wherein the hydrogen sulfide is supercritical, wet, dry or dissolved in oil or water.
14. The method of any one of claims 9–13, wherein the borehole penetrates at least one fluid-containing reservoir, the reservoir containing fluid with a hydrogen sulfide concentration higher than about five moles per liter.
15. The method of claim 9–14, wherein the cement comprises one or more members of the list comprising Portland cement, calcium aluminate cement, fly ash, blast furnace slag, lime-silica blends, zeolites, geopolymers, Sorel cements, chemically bonded phosphate ceramics and polymer resins.

1/4

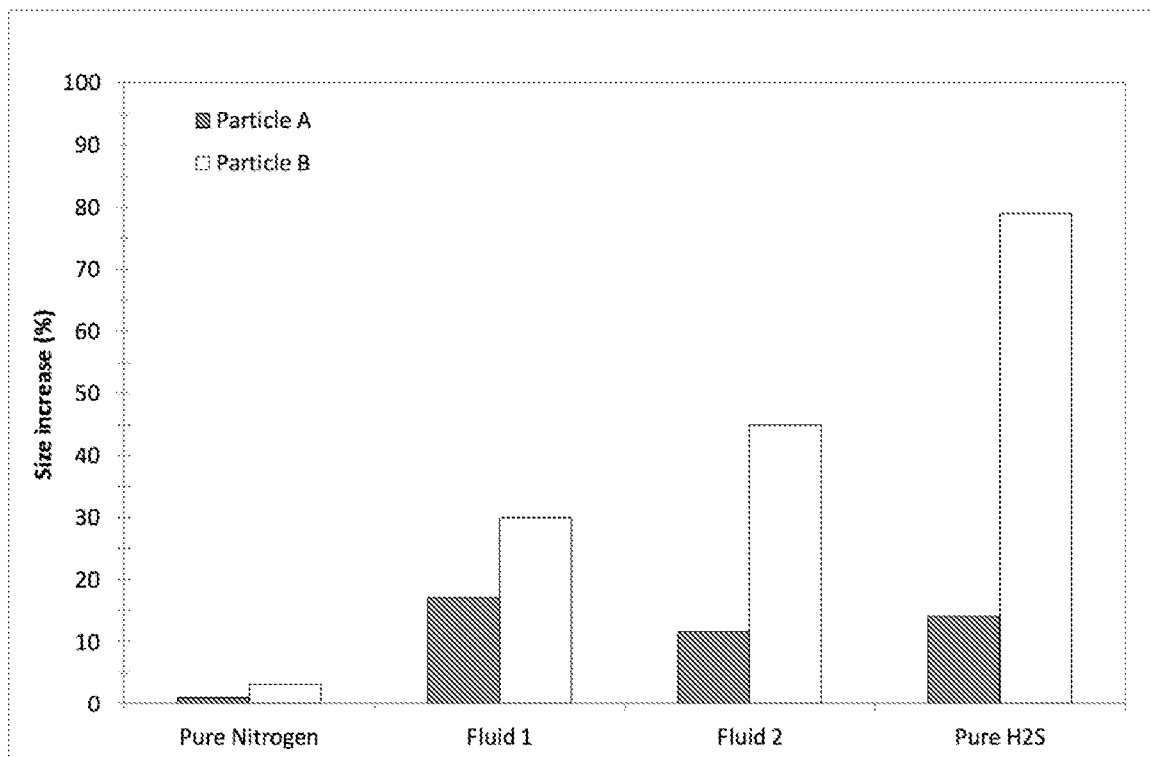


Figure 1

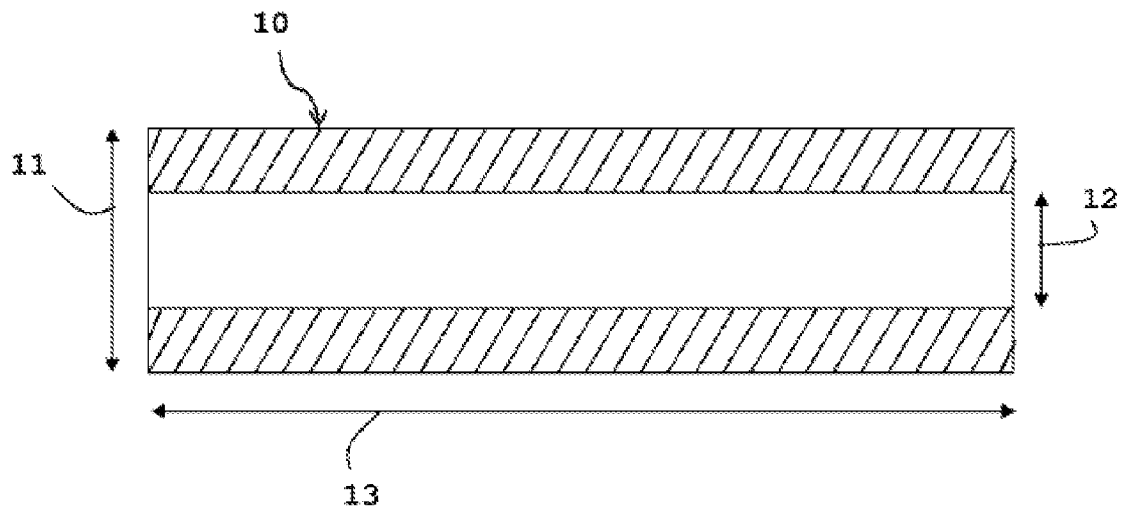


Figure 2

2/4

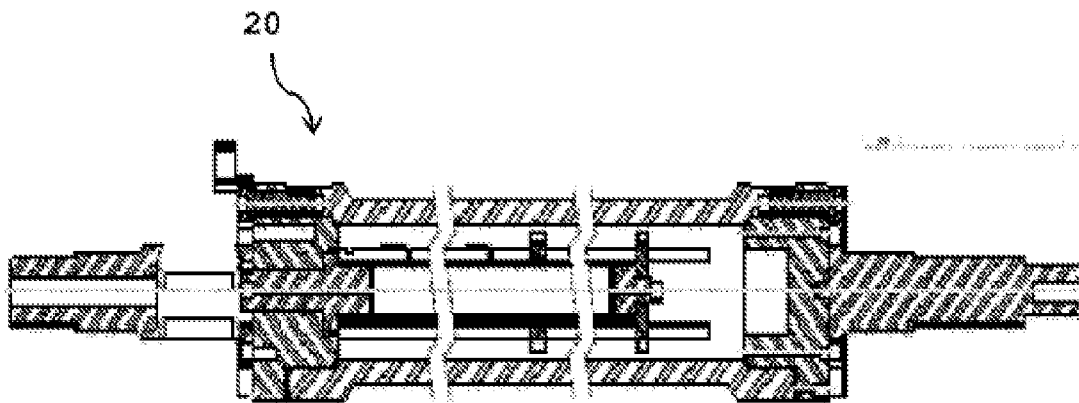


Figure 3

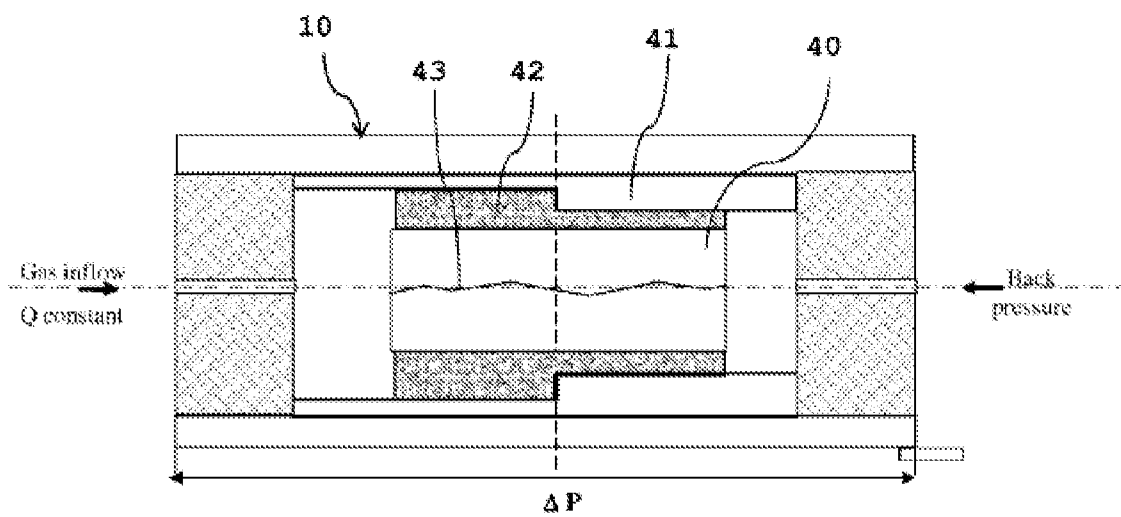


Figure 4

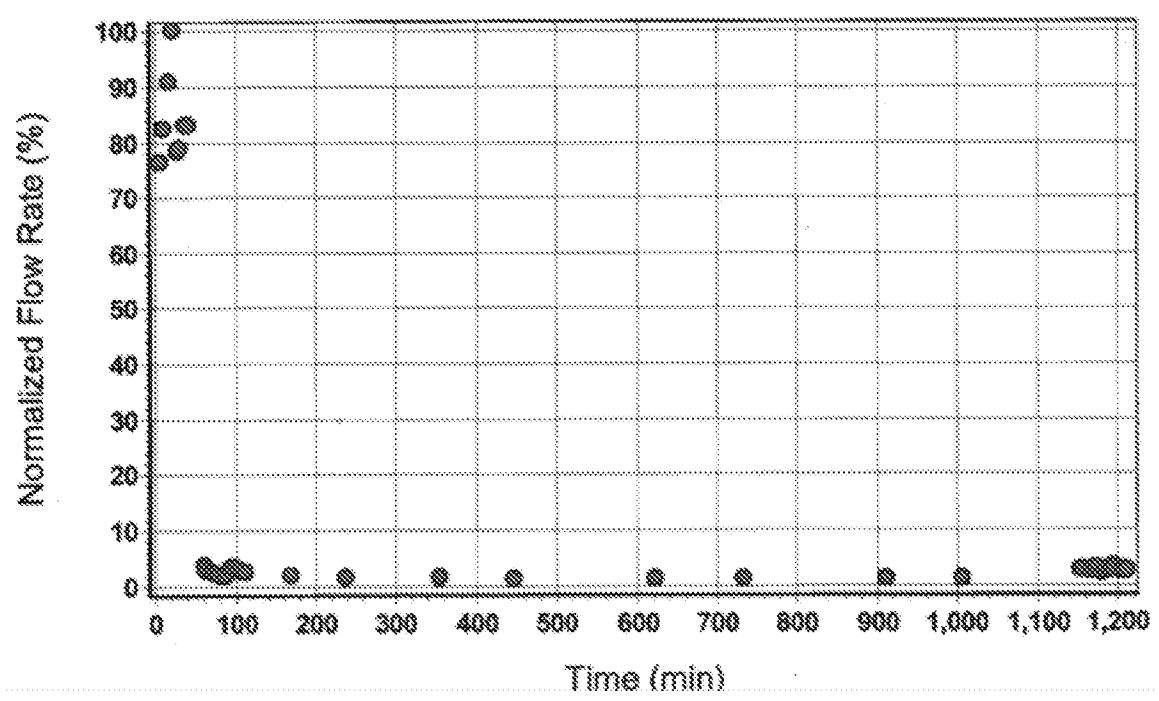


Figure 5

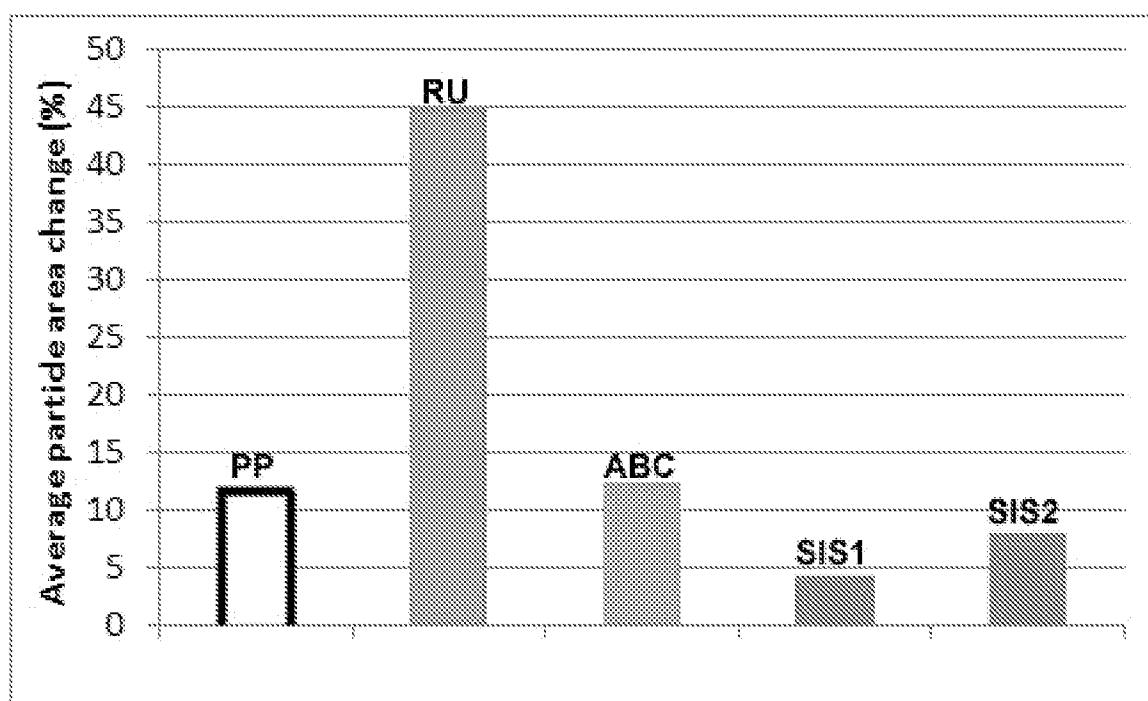


Figure 6

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US15/22757

A. CLASSIFICATION OF SUBJECT MATTER

IPC(8) - C09K 8/467 (2015.01)

CPC - C09K 8/467

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC(8): C09K 8/467; C04B 24/24, 26/02, 28/00, 28/02, 28/04, 28/06, 28/08, 28/18, 28/32, 28/34 (2015.01)

CPC: C09K 8/467; C04B 24/24, 26/02, 28/00, 28/02, 28/04, 28/06, 28/08, 28/18, 28/32, 28/34; USPC: 166/285, 292

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

PatSeer (US, EP, WO, JP, DE, GB, CN, FR, KR, ES, AU, IN, CA, INPADOC Data); Google; Google Scholar; ProQuest; IP.com; keywords: rubber, polymer, hydrogen sulfide, cement, wellbore, bore, isolate, swell, pump

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WO 2011/144433 A1 (SERVICES PETROLIERS SCHLUMBERGER) 24 November 2011 (24.11.2011); paragraphs [0009], [0021]	1, 2, 3/1, 9, 10, 11/9
Y	US 4,524,982 A (HERTZ, JR. DL) June 25, 1985; page 1; column 3, lines 1-20	1, 2, 3/1, 9, 10, 11/9
P, X	WO 2014/187566 A1 (SERVICES PETROLIERS SCHLUMBERGER) 27 November 2014 (27.11.2014); entire document	1, 2, 3/1, 3/2, 9, 10, 11/9, 11/10
A	US 2012/0073814 A1 (ALGU DR et al.) March 29, 2012; entire document	1, 2, 3/1, 3/2, 9, 10, 11/9, 11/10
A	US 2008/0023205 A1 (CRASTER B et al.) January 31, 2008; entire document	1, 2, 3/1, 3/2, 9, 10, 11/9, 11/10

☐ Further documents are listed in the continuation of Box C.

☐ See patent family annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

04 June 2015 (04.06.2015)

Date of mailing of the international search report

06 JUL 2015

Name and mailing address of the ISA/

Mail Stop PCT, Attn: ISA/US, Commissioner for Patents

P.O. Box 1450, Alexandria, Virginia 22313-1450

Facsimile No. 571-273-8300

Authorized officer

Shane Thomas

PCT Helpdesk: 571-272-4300

PCT OSP: 571-272-7774

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US15/22757

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. ☐ Claims Nos.:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:

3. ☒ Claims Nos.: 4-8, 12-15
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying additional fees, this Authority did not invite payment of additional fees.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.
- ☐ The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.
- ☐ No protest accompanied the payment of additional search fees.