



(86) **Date de dépôt PCT/PCT Filing Date:** 2006/11/22  
(87) **Date publication PCT/PCT Publication Date:** 2007/06/21  
(45) **Date de délivrance/Issue Date:** 2014/03/18  
(85) **Entrée phase nationale/National Entry:** 2008/06/03  
(86) **N° demande PCT/PCT Application No.:** EP 2006/011382  
(87) **N° publication PCT/PCT Publication No.:** 2007/068344  
(30) **Priorité/Priority:** 2005/12/16 (EP05077906.5)

(51) **Cl.Int./Int.Cl. C11D 1/00** (2006.01),  
**A23L 1/00** (2006.01), **A23L 1/0534** (2006.01),  
**A23L 1/308** (2006.01), **B01F 17/00** (2006.01)

(72) **Inventeurs/Inventors:**  
CAI, YA, CN;  
CAO, JIAN, CN;  
GOLDING, MATTHEW D., NL;  
LIU, WEICHANG, CN;  
PAUNOV, VESSELIN NIKOLAEV, GB;  
PELAN, EDWARD G., NL;

(73) **Propriétaire/Owner:**  
UNILEVER PLC, GB

(74) **Agent:** RIDOUT & MAYBEE LLP

(54) **Titre : MATERIAU TENSIOACTIF ET SON APPLICATION**  
(54) **Title: SURFACE-ACTIVE MATERIAL AND ITS APPLICATION**

(57) **Abrégé/Abstract:**

There is provided a surface-active material that comprises fibres which have been modified so as to impart surface-active properties onto said fibres and giving it a contact angle between 60° and 120°, wherein the fibres have an aspect ratio of more than 10 to 1,000. The surface-active material can be used for foam and emulsion formation and stabilisation, coatings, encapsulation and drug delivery. It can for example be used in the following industries: foods, home and personal care, oilfield, agriculture, textile, construction, emulsion polymerisation, leather, plastic, pulp, paper and pharma.



(72) **Inventeurs(suite)/Inventors(continued):** STOYANOV, SIMEON D, NL; ZHOU, WEIZHENG, CN

## (12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
21 June 2007 (21.06.2007)

PCT

(10) International Publication Number  
**WO 2007/068344 A1**

## (51) International Patent Classification:

*CIID 1/00* (2006.01)      *A23L 1/308* (2006.01)  
*A23L 1/00* (2006.01)      *B01F 17/00* (2006.01)  
*A23L 1/0534* (2006.01)

## (21) International Application Number:

PCT/EP2006/011382

## (22) International Filing Date:

22 November 2006 (22.11.2006)

## (25) Filing Language:

English

## (26) Publication Language:

English

## (30) Priority Data:

05077906.5      16 December 2005 (16.12.2005)      EP

(71) Applicant (for all designated States except AE, AG, AU, BB, BW, BZ, CA, CY, EG, GB, GD, GH, GM, IE, IL, IN, KE, KN, LC, LK, LS, LY, MN, MW, MY, NA, NG, NZ, OM, PG, SC, SD, SG, SL, SZ, TT, TZ, UG, US, VC, ZA, ZM, ZW): UNILEVER N.V. [NL/NL]; Weena 455, NL-3013 AL Rotterdam (NL).

(71) Applicant (for AE, AG, AU, BB, BW, BZ, CA, CY, EG, GB, GD, GH, GM, IE, IL, KE, KN, LC, LK, LS, LY, MN, MW, MY, NA, NG, NZ, OM, PG, SC, SD, SG, SL, SZ, TT, TZ, UG, VC, ZA, ZM, ZW only): UNILEVER PLC [GB/GB]; UNILEVER HOUSE, Blackfriars, London, Greater London EC4P 4BQ (GB).

(71) Applicant (for IN only): HINDUSTAN LEVER LIMITED [IN/IN]; Hindustan Lever House, 165/166 Backbay Reclamation Maharashtra, Mumbai 400 020 (IN).

## (72) Inventors; and

(75) Inventors/Applicants (for US only): CAI, Ya [GB/CN]; Unilever Research China, 3rd Floor, Xinmao Building, 99 Tianzhou Road, Cao-he-jing Hi-Tech Park, Shanghai 200 233 (CN). CAO, Jian [CN/CN]; Unilever Research China, 3/F XinMao Building, 99 TianZhou Road, Shanghai 200233 (CN). GOLDING, Matthew, D. [GB/NL];

Unilever R & D Vlaardingen B.V., Olivier van Noortlaan 120, NL-3133 AT Vlaardingen (NL). LIU, Weichang [CN/CN]; Unilever Research China, 3/F XinMao Building, 99 TianZhou Road, Shanghai 200233 (CN). PAUNOV, Vesselin Nikolaev [GB/GB]; University of Hull, Department of Chemistry, Cottingham Road, Hull Hull HU6 8RX (GB). PELAN, Edward, G. [GB/NL]; Unilever R & D Vlaardingen B.V., Olivier van Noortlaan 120, NL-3133 AT Vlaardingen (NL). STOYANOV, Simeon, D [BG/NL]; Unilever R & D Vlaardingen B.V., Olivier van Noortlaan 120, NL-3133 AT Vlaardingen (NL). ZHOU, Weizheng [CN/CN]; Unilever Research China, 3/F XinMao Building, 99 TianZhou Road, Shanghai 200233 (CN).

(74) Agent: KAN, Jacob, H; Unilever Patent Group, Olivier van Noortlaan 120, NL-3133 AT Vlaardingen (NL).

(81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

## Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: SURFACE-ACTIVE MATERIAL AND ITS APPLICATION

(57) Abstract: There is provided a surface-active material that comprises fibres which have been modified so as to impart surface-active properties onto said fibres and giving it a contact angle between 60° and 120°, wherein the fibres have an aspect ratio of more than 10 to 1,000. The surface-active material can be used for foam and emulsion formation and stabilisation, coatings, encapsulation and drug delivery. It can for example be used in the following industries: foods, home and personal care, oilfield, agriculture, textile, construction, emulsion polymerisation, leather, plastic, pulp, paper and pharma.

WO 2007/068344 A1

**SURFACE-ACTIVE MATERIAL AND ITS APPLICATION****FIELD OF THE INVENTION**

5 The invention relates to a new surface-active material and its applications in the area of foam and emulsion formation and stabilisation, coatings, encapsulation and drug delivery. More in particular, it relates to a surface-active material that comprises surface-active fibres and to a method for preparing  
10 said surface-active material, as well as to products comprising said surface-active material.

**BACKGROUND TO THE INVENTION**

15 A surface-active agent or surfactant is a substance that lowers the surface tension of the medium in which it is dissolved, and/or the interfacial tension with other phases. Accordingly, it is positively adsorbed at the liquid/vapour and/or at other interfaces.

20

Surface-active agents are widely used industry, for instance in foods, cleaning compositions and personal care products. In foods, they are used to achieve emulsions of oily and water-phases, such as in fat spreads or mayonnaise. In laundry  
25 cleaning applications, they are used to solubilise dirt and keep it in solution, so that it can be effectively removed from the fabric.

For cleaning applications, the surface-active compounds may be  
30 chosen from anionic, cationic, nonionic, amphoteric and zwitterionic surfactants. Many suitable surface-active compounds are available and are fully described in the literature, for example, in "Surface-Active Agents and Detergents", Volumes I and II, by Schwartz, Perry and Berch.

The most commonly used detergent-active compounds are soaps and synthetic non-soap anionic and nonionic compounds.

Examples of anionic surfactants include alkylbenzene sulphonates, particularly linear alkylbenzene sulphonates  
5 having an alkyl chain length of C8-C15; primary and secondary alkylsulphates, particularly C8-C15 primary alkyl sulphates; alkyl ether sulphates; olefin sulphonates; alkyl xylene sulphonates; dialkyl sulpho-succinates; and fatty acid ester sulphonates. Sodium salts are generally preferred.

10

Examples of nonionic surfactants include the primary and secondary alcohol ethoxylates, especially the C<sub>8</sub>-C<sub>20</sub> aliphatic alcohols ethoxylated with an average of from 1 to 20 moles of ethylene oxide per mole of alcohol, and more especially the  
15 C<sub>10</sub>-C<sub>15</sub> primary and secondary aliphatic alcohols ethoxylated with an average of from 1 to 10, preferably 3 to 7 moles of ethylene oxide per mole of alcohol. Non-ethoxylated nonionic surfactants include alkylpolyglycosides, glycerol monoethers, and polyhydroxy-amides (glucamide).

20

The choice of the surface-active material (surfactant), and the amount present, will depend on the intended use of the detergent composition. For fabric washing compositions, different surfactant systems may be chosen, as is well known  
25 to the skilled formulator, than for handwashing products or mechanical dishwashing products.

In foods, surface-active materials are commonly used to prepare emulsions. Edible emulsions are used as a base for  
30 many types of food products. Mayonnaise compositions, for example, comprise edible oil-in-water emulsions that typically contain between 80 to 85% by weight oil, and egg yolk, salt, vinegar and water. Mayonnaise compositions are enjoyed by many consumers, and particularly, on sandwiches, in dips, with fish

and other food applications. The oil present in the edible emulsions used in such food products is generally present as droplets dispersed  
5 in the water phase. In addition to droplet size and the amount of droplets dispersed, the close packing of the oil droplets results in the characteristic rheological behaviour of the emulsions used to make the desired food product, such as mayonnaise or margarine.

10 The surface-active agents that are most commonly used in food applications comprise low molecular weight emulsifiers that are primarily based on fatty acid derivatives. Examples include: lecithin's, monoglycerides (saturated and unsaturated), polysorbate esters (Tweens™), sorbitan esters (Spans™), polyglycerol esters,  
15 propylene glycol monostearate, sodium and calcium stearyl lactylates, sucrose esters, organic acid (lactic, acetic, tartaric, succinic) esters of monoglycerides. Proteins and other surface-active biopolymers can also be used for this purpose. Typical examples of food proteins include milk proteins (caseins and whey proteins), soy  
20 protein, egg protein, lupin protein, pea protein, wheat protein. Examples of other surface-active biopolymers include gum Arabic, modified surface active pectin and OSA modified starch.

Recently, the interest in the study of solid particles as emulsifiers  
25 of dispersed systems has been re-awakened. Much of this activity has been stimulated by the research of Binks and co-workers (Binks, B. P. Curr. Opin. Colloid Interface Sci. 2002, 7, 21), though the principles of such stabilisation were observed at least 100 years ago (Ramsden, W. Proc. R. Soc. London 1903, 72, 156). The advantage of  
30 particle stabilisation is that it is almost impossible to displace an adsorbed particle once adsorbed to an interface. This gives particle stabilised emulsions and foams excellent stability, especially

with respect to ripening mechanisms such as disproportionation.

5 Whilst the use of particles to stabilise o/w, w/o and duplex  
emulsions and foams has been amply demonstrated in recent years, much  
less research has been carried out on non-spherical structures with  
respect to the stabilisation of interfaces. Furthermore, it has  
recently been demonstrated by Alargova et. al. (Langmuir, 2004, 20,  
10 10371), that epoxy rods can be used to provide interfacial  
stabilisation to emulsions and foams.

DE 196 17 568 C1 discloses a dispersion aid that consists of  
cellulose fibres, which are milled to lengths less than 150  
15 micrometer and thicknesses less than 50 micrometer and treated during  
or after milling with additive(s) imparting hydrophobic and anti-  
aggregation properties.

EP 0 238 330 A2 discloses a modified emulsifier for use in food  
20 applications. The modified emulsifier consists of binding an  
emulsifier, e.g. lecithin, to a polysaccharide.

WO 03/040190 A1 discloses modified dialdehyde polysaccharides which  
act as wet-strength agents.

25

Notwithstanding the fact that many surface-active materials are known  
and available, there is a continuous need for new alternative or  
improved surface-active materials, especially environmentally  
friendly surface-active materials having good biodegradability  
30 properties. It is therefore an object of the present invention to  
provide such surface-active materials. It is a further object to  
provide surface-active materials that are useful in the stabilisation  
of emulsions and foams.

35

4a

Surprisingly, it has now been found that one or more of the above-mentioned objects can be achieved by the surface-active material  
5 according to the invention, which is characterised in that it comprises fibres which have been modified so as to impart surface-active properties onto said fibres and giving it a contact angle between  $60^\circ$  and  $120^\circ$ .

10 The present inventors have found that the shape and size are of critical importance for the colloidal stability of foams and emulsions. Rod like (fibril) shapes are much more efficient than spherical particles. Another key factor for good foam and emulsion  
15 stabilisation is particle contact angle at oil/water or air/water interface, which must be as close to

90° as possible. The rod-like structures must therefore be amphiphathic in design (o/w and w/o stabilisation depends on the relative balance between hydrophobicity and hydrophilicity).

Rod- and fibre-like structures such as microcrystalline cellulose (MCC) and natural plant fibres have been used in food systems. Such structures often find application as bulk structuring agents, providing a contribution to the rheology of a formulation, without showing a tendency to adsorb at interfaces or exhibiting surface-active properties.

#### 15 SUMMARY OF THE INVENTION

According to a first aspect, the invention provides a surface-active material that which is characterised in that it comprises organic fibres which have been modified by physical means by coating the fibres with hydrophobic materials, so as to impart surface-active properties onto said fibres and giving it a three-phase contact angle between 60° and 120°, wherein the fibres have an insoluble, particulate structure, and wherein the fibres have an aspect ratio of more than 10 to 1,000.

25 According to a second aspect, there is provided a process for preparing such a surface-active material.

According to a third aspect, there is provided a product comprising said surface-active material for the purpose of foam and emulsion formation and stabilisation, coatings, encapsulation and drug delivery. The fourth aspect involves the application of said surface-active material in home and personal care, foods, oilfield, agriculture, textile, construction, emulsion polymerisation, leather, plastic, pulp, paper and pharma.

**DETAILED DESCRIPTION OF THE INVENTION**

In its first aspect, the invention relates to a surface-active material comprising natural fibres that have been modified. By the word "fibre", we mean any structure, especially an insoluble, particulate structure, wherein the ratio between the length and the diameter ranges from 10 to infinite. Here, the diameter means the largest distance of the cross-section. The materials of the "fibre" substance can be organic, inorganic, polymeric and macromolecular. The fibre topology might be liner or branched (star-like). The aspect ratio in this case is defined as aspect ratio of the longest branch.

The fibres used in the present invention have a length of 0.1 to 100 micrometer, preferably from 1 to 50 micrometer. Their diameter is in the range of 0.01 to 10 micrometer. The aspect ratio (length / diameter) is generally more than 10, preferably more than 20 up to 1,000.

The fibres may be of organic or inorganic origin. In particular, organic, natural fibres made of a crystalline, insoluble form of carbohydrates, such as microcrystalline cellulose, can be used. Such fibres have the advantage that they are very biodegradable, which is favourable for the environment. Very often organic fibres are also food-grade. One example of a suitable source is the microcrystalline cellulose obtainable from *Acetobacter*. Other examples are citrus fibres, onion fibres, silk, stearic acid, polyhydroxybutyrate-valerate, PVP (polyvinyl pyrrolidone), PCL (polycaprolactone), their derivatives and copolymers, and other polymers that can be spun with diameter ranging from 0.01 to 30 micrometers.

Examples of inorganic fibres are  $\text{CaCO}_3$ ,  $\text{CaSO}_4$ ,  $\text{ZnO}$ ,  $\text{TiO}_2$ ,  $\text{MgO}$ ,  $\text{MgSO}_4$ ,  $\text{Mg}(\text{OH})_2$ ,  $\text{Mg}_2\text{B}_2\text{O}_5$ , aluminium borate, potassium titanate,

barium titanate, hydroxyapatite, attapulgite, but other inorganic crystals with fibre-like morphology could also be used.

5 The fibres that are used in the present invention are not used as such, but in modified form. As a consequence of the modification, the contact angle is modified such that is in the range of between  $60^\circ$  and  $120^\circ$ , preferably between  $70^\circ$  and  $110^\circ$ , more preferably between  $80^\circ$  and  $100^\circ$ . By contact angle  
10 we mean the three-phase contact angle at the fibre/air/water interface or the fibre/oil/water interface, depending on the type product in which the surface-active material of the present invention is used. For foams this will be the fibre/air/water contact angle, for emulsions, the  
15 fibre/oil/water contact angle.

The contact angle can be measured using the gel-trapping technique as described by technique as described by Paunov (Langmuir, 2003, 19, 7970-7976) or alternatively by using  
20 commercial contact angle measurement apparatus such as the Dataphysics OCA20.

The modification of the fibres can be achieved by chemical or physical means. The chemical modification involves  
25 esterification or etherification, by means of hydrophobic groups, such like stearate and ethoxy groups, using well-known techniques. The physical modification includes coating of the fibres with hydrophobic materials, for example ethylcellulose or hydroxypropyl-cellulose. One can also use waxes, such as  
30 shellac, carnauba or bees wax. Fat and fatty acids such as stearic acid may also be used. The coating can be done using colloidal precipitation using solvent or temperature change, for instance. The physical modification may also involve

"decoration" of rod like materials using hydrophobic nanoparticles, for instance silica.

According to the invention, one can use the process of  
5 controlled esterification of Microcrystalline cellulose  
(Antova et. al, Carbohyd. Polym., 2004, 57 (2), 131) as  
possible route for controlled hydrophobicity modification and  
therefore obtaining particles with surface-active properties.  
Based on this principle, it will be understood that the  
10 skilled person can easily find other routes to modify the  
hydrophobicity of other types of fibres of organic or  
inorganic origin.

According to another aspect of the present invention, there is  
15 provided a surface-active material, obtainable by modification  
of fibres so as to impart surface-active properties onto said  
fibres and giving it a contact angle between  $60^\circ$  and  $120^\circ$ ,  
wherein the fibres have an aspect ratio of more than 10 to  
1,000.

20

Possible applications for the surface-active materials of the  
present invention are in the area of foam and emulsion  
formation and stabilisation, for instance as foam stabilisers  
for ice cream, as emulsion stabilisers for mayonnaise or  
25 margarine, as foam formation agent and stabiliser for home and  
personal care products such as toothpaste, and as flotation  
agent, for instance in the mining industry. They may also be  
used for coatings, encapsulation and drug delivery.

30 The industries where these surface-active materials of the  
present invention could be applied include home and personal  
care, foods, oil industry, agriculture, textile, construction,  
polymer industry (emulsion polymerisation), leather, plastic,  
pulp, paper and pharma.

The invention will now be further illustrated by means of the following non-limiting examples.

5 **Example 1**

In a 50-ml beaker, 0.05g ethyl cellulose (EC, Aldrich product, viscosity: 10cps) was added into 20 ml of acetone. Then under ultrasonication (Branson Ultrasonics Corporation, 5510E-DTH) and magnetic stirring (IKA™, RCT basic™), the ethyl cellulose gradually dissolved to form a homogenous solution. Next 0.2g of Microcrystalline cellulose (MCC, rod-like, Diameter: ~20nm, Length: several to tens of microns) was added into the system and ultrasonication was applied for 10 minutes to induce the homogenous dispersion of the MCC. As a non-solvent of ethyl cellulose, 10 ml of water was dropped into the above system to induce coacervation of ethyl cellulose, during which the coacervated ethyl cellulose particles were attached to MCC fibers. Subsequently, the acetone was completely removed by stirring or under reduced pressure at an elevated temperature. The obtained MCC/ethyl cellulose water dispersion was used to investigate the foamability and foam stability. The foams were prepared at room temperature by hand-shaking for a period of 40s. The foams stabilized by this material are stable at ambient conditions for more than two weeks.

25 **Example 2**

4.0 g of rod-like CaCO<sub>3</sub> (provided by Qinghai Haixing Science and Technology Development Co., Ltd, China, Diameter: ~2 microns, Length: ~50 microns) was dispersed into 40 ml acetone solution containing 0.20 g of ethyl cellulose (EC, Aldrich product, viscosity: 10cps). Ultrasonication (Branson Ultrasonics Corporation, 5510E-DTH™) was used for 10 minutes to induce the homogenous dispersion of the CaCO<sub>3</sub>. Then 160ml of water was quickly poured into the dispersion to make the ethyl

cellulose deposit fast on the surface of  $\text{CaCO}_3$  particles. After magnetic stirring (IKA, RCT basic) for 2 minutes, the dispersion was filtrated, and the filter cake was immediately dried in vacuum oven at  $80^\circ\text{C}$ . Finally  $\text{CaCO}_3$ /ethyl cellulose composite was obtained. Then the powder was put into water to investigate foamability and foam stability. The foams were prepared at room temperature by hand-shaking for a period of 40s. The foams stabilized by these materials are stable for more than one month. The initial volume of the foam linearly increased with the amount of material added. It is interesting to note that initial foam volume of the foams stabilized by these materials passes through a maximum at a ratio of EC: $\text{CaCO}_3$  of about 1:20 (which was chosen in this example).

### 15 Example 3

4.0 g of rod-like ZnO (tetrapod-like, provided by Chengdu Advanced Technologies and Crystal-Wide Co., Ltd, China, Diameter: ~ 2 microns, Length: several tens of micron) was dispersed into 40 ml of acetone solution containing 0.20 g of ethyl cellulose (EC, Aldrich product, viscosity: 10cps). Ultrasonication (Branson Ultrasonics Corporation, 5510E-DTH) was used for 10 minutes to induce the homogenous dispersion of the ZnO. Then 160ml of water was quickly poured into the dispersion to make ethyl cellulose deposit fast on the surface of ZnO particles. After magnetic stirring (IKA, RCT basic) for 2 minutes, the dispersion was filtrated, and the filter cake was immediately dried in vacuum oven at  $80^\circ\text{C}$ . Finally, a ZnO/ethyl cellulose composite was obtained. Then the powder was put into water to investigate foamability and foam stability. The foams were prepared at room temperature by hand-shaking for a period of 40s. The foams stabilized by this material are stable at ambient conditions for more than two weeks.

What is claimed is:

1. A surface-active material comprising organic fibres which have been modified by physical means by coating of the fibres with hydrophobic materials,  
so as to impart surface-active properties onto said fibres and giving the surface active material a three-phase contact angle at the fibre/air/water interface or the fibre/oil/water interface between  $60^\circ$  and  $120^\circ$ ,  
wherein the fibres have an insoluble, particulate structure, and  
wherein the fibres have an aspect ratio of more than 10 to 1,000.
2. The surface-active material according to claim 1, wherein the fibres have a length of 0.1 to 100 micrometer.
3. The surface-active material according to claim 1 or 2, wherein the hydrophobic materials comprise ethylcellulose or hydroxypropyl-cellulose.
4. The surface-active material according to any one of claims 1 to 3, wherein the fibres are cellulose fibres.
5. The surface-active material according to claim 4, wherein the fibres are microcrystalline cellulose fibres.
6. The surface-active material according to claim 5, wherein the fibres are microcrystalline cellulose fibres obtainable from *Acetobacter*.

7. A process for the preparation of a surface-active material according to any one of claims 1 to 6, comprising the steps of: selecting said organic fibres; and modifying the organic fibres using physical means by coating the fibres with hydrophobic materials, so as to impart surface-active properties onto said fibres and giving the surface active material a three-phase contact angle at the fibre/air/water interface or the fibre/oil/water interface between  $60^\circ$  and  $120^\circ$ , wherein the fibres have an insoluble, particulate structure, and wherein the fibres have an aspect ratio of more than 10 to 1,000.
8. The process according to claim 7, wherein the hydrophobic materials comprise ethylcellulose or hydroxypropyl-cellulose.
9. A food product comprising the surface-active material according to any one of claims 1 to 6, wherein the food product is selected from aerated products or non-aerated products.
10. A food product according to claim 9, in the form of a liquid product selected from the group consisting of sauces, soups and drinks.
11. A process for the preparation of a food product according to claim 9 or 10, comprising the step of adding the surface-active material according to any one of claims 1 to 6 to a food product.

12. Use of the surface-active material according to any one of claims 1 to 6 for foam and emulsion formation and stabilisation, coatings, encapsulation and drug delivery.

13. Use according to claim 12, in the following industries: home and personal care, foods, oilfield, agriculture, textile, construction, emulsion polymerisation, leather, plastic, pulp, paper and pharma.