

PCTWORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁷ : B01D 19/04, D21H 21/12	A1	(11) International Publication Number: WO 00/50147 (43) International Publication Date: 31 August 2000 (31.08.00)
(21) International Application Number: PCT/US00/03975 (22) International Filing Date: 17 February 2000 (17.02.00) (30) Priority Data: 09/256,059 24 February 1999 (24.02.99) US (71) Applicant: BETZDEARBORN INC. [US/US]; 4636 Somerton Road, P.O. Box 3002, Trevese, PA 19053-6783 (US). (72) Inventors: BERZANSKY, Charles, J.; Apartment 202, 10023 Belle River Boulevard #614, Jacksonville, FL 32256 (US). HENDRIKS, William, A.; 1116 Fruit Cove Road, Jacksonville, FL 32259 (US). (74) Agents: BOYD, Steven, D. et al.; BetzDearborn Inc., 4636 Somerton Road, Trevese, PA 19053-6783 (US).		(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i> <i>Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>
(54) Title: HYDROCARBON OIL FREE DEFOAMER (57) Abstract The present invention is directed to foam control compositions and their use in aqueous systems. The foam control compositions are particularly useful in the processing of pulp and paper and the treatment of effluent water.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav	TM	Turkmenistan
BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
CA	Canada	IT	Italy	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
CM	Cameroon		Republic of Korea	PL	Poland		
CN	China	KR	Republic of Korea	PT	Portugal		
CU	Cuba	KZ	Kazakstan	RO	Romania		
CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
DE	Germany	LI	Liechtenstein	SD	Sudan		
DK	Denmark	LK	Sri Lanka	SE	Sweden		
EE	Estonia	LR	Liberia	SG	Singapore		

HYDROCARBON OIL FREE DEFOAMER

FIELD OF THE INVENTION

5

The present invention relates to foam control compositions, their preparation and use in aqueous systems. The foam control compositions are particularly useful in the processing of pulp and paper and the treatment of effluent water.

10

BACKGROUND OF THE INVENTION

Foam can present serious problems in the manufacture of pulp and paper. Lack of adequate foam control in the paper slurry can limit or
15 halt production, as well as affect finished product quality.

Defoamers, when added to a foaming liquid, can prevent bubble formation or cause coalescence of smaller bubbles into larger ones, leading to rupture. Most conventional defoamers include a hydrophobic
20 material having a melting point greater than 40° C, such as a diamide and/or a hydrophobically modified insoluble particle, e.g., silica, dispersed in a hydrocarbon oil phase. Many of these compositions also

contain silicone oil. A problem with these types of compositions is that they quickly become unstable, and a settling of individual components, such as silica, occurs. Hydrophobic materials with melting points above 40° C will precipitate rapidly after the processing temperatures fall below
5 their melting point when dispersed in a hydrocarbon oil phase. Pure organic-based defoamers are not highly effective and require large amounts of material to defoam systems, while silica-based defoamers are expensive because large amounts of the silica particles are required to perform effectively.

10

DETAILED DESCRIPTION OF THE INVENTION

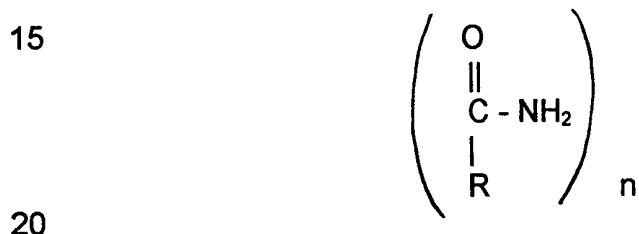
The present invention relates to a water-based defoaming agent containing (1) an ester of a dicarboxylic acid and a monohydric alcohol,
15 (2) at least one predominantly hydrophobic material having a melting point greater than 40° C selected from the group consisting of hydrocarbon waxes, fatty amides, fatty acids and fatty alcohols, (3) a surfactant selected from the group consisting of polyethylene glycol esters, sorbitan/sorbitol esters, ethoxylated fatty alcohols, fatty acids and
20 fatty ethers, and (4) water. In a preferred embodiment, the defoaming agent further includes a polyether modified silicone copolymer, an aqueous polyacrylate and an alkali material.

The present invention provides a defoaming agent with improved
25 efficacy in aqueous systems that is particularly effective at controlling foam in the paper making process. The present invention provides a water-based defoamer composition with improved stability and no particle precipitation over a temperature range of about 10° C to 50° C. Furthermore, the invention provides an "environmentally friendly"

defoamer composition comprising low or no volatile organic carbon (VOC) content and no dioxin precursor containing hydrocarbon oils.

The dicarboxylic acid ester is formed from the reaction of a
 5 dicarboxylic acid with a monohydric alcohol. The dicarboxylic acid is selected from the group consisting of adipic acid, suberic acid, azelaic acid, sebacic acid, phthalic acid and isomers thereof. These acids are then reacted with alcohols selected from the group consisting of hexyl, octyl, nonyl, decyl, dodecyl, tetradecyl, hexadecyl, octadecyl alcohol, and
 10 isomers thereof.

The hydrophobic material with a melting point greater than 40° C includes fatty acid amides with the general formula:



wherein n is an integer from 2 to 6, and R is a saturated or unsaturated, straight or branched aliphatic group having from 5 to 22 carbons.

The fatty amides are formed from the reaction of polyamine
 25 containing at least one alkylene group having from 2 to 6 carbon atoms and a fatty acid having from 5 to 22 carbon atoms not including the carboxyl group carbon. Examples of suitable hydrophobic amides formed from this reaction include but are not limited to methylene bis acrylamide, methylene bis lauramide, ethylene bis stearamide, ethylene bis
 30 behenamide, and ethylene bis oleoamide. The aliphatic fatty alcohols

and fatty acids are saturated, linear or branched and contain at least 12 carbon atoms, while the hydrocarbon wax has a melting point of from about 40° C to 120° C.

5 The emulsifying surfactants are preferably nonionic surfactants and include polyethylene glycol esters such as PEG 400 dioleate and PEG 600 dioleate, sorbitan/sorbital esters such as sorbitan monooleate, POE (20) sorbitan monooleate and POE (40) sorbitol hexaoleate, ethoxylated fatty acids, ethoxylated fatty alcohols, and ethoxylated fatty
10 ethers such as POE (40) stearate, POE (20) stearyl alcohol and POE (10) oleyl ether. The polyether-modified silicone copolymer is a silicone copolymer modified with ethylene oxide and propylene oxide which results in a copolymer with a cloud point of from about 40° C to less than about 25° C, and a molecular weight range of from about 10,000 to
15 40,000.

 The polyacrylates that are encompassed by the present invention are aqueous, acidic, acrylic emulsion copolymers that thicken by an uncoiling action of the polymer chain upon neutralization with an alkali
20 material. The preferred polyacrylate is a copolymer of ethylacrylate and methacrylic acid. The alkali material can be sodium hydroxide, potassium hydroxide, ammonium hydroxide, or triethanolamine.

 In a preferred embodiment, the present invention relates to the use
25 of a dicarboxylic acid ester, without organic solvent or hydrocarbon oil, as a carrier for the high melting point hydrophobic material and a surfactant selected from the group consisting of polyethylene glycol esters, sorbitol/sorbitan esters, ethoxylated fatty acids, ethoxylated fatty alcohols and ethoxylated fatty ethers, emulsified in water to form a stable

dispersion of particles that does not precipitate upon cooling below the melting point of the hydrophobic material. The dispersions are stable at temperatures of from about 10° C to 50° C for about 6 weeks.

- 5 Preferred amounts of the components of the defoamer composition of the present invention are as follows:

(Values in percent by weight)

10	<u>Component</u>	<u>Preferred %</u>	<u>Particularly Preferred %</u>
	Dicarboxylic Acid Ester	5 - 60	10 - 40
	Hydrophobic material	0.5 - 20	1 - 10
	Surfactants	0.5 - 30	5 - 20
15	Modified silicone copolymer	0.1 - 5	0.1 - 5
	Water	40 - 95	40 - 80
	Polyacrylate	0.5 - 5	0.5 - 5
	Alkali material	0.1 - 5	0.5 - 5

- 20 The defoamers are prepared by charging all the material excluding the water to a vessel, heating to about 120° C, and holding at this temperature for about 30 minutes. The mixture is added to a vessel that has been charged with the water phase maintained under constant agitation. The dispersion is then cooled to below 45° C, at which time the
- 25 polyacrylate is added and then neutralized with the alkali material.

In the testing of the present invention, various defoamers were tested in a recirculation test cell. The foaming medium is circulated from the bottom of a calibrated graduated column via a pump and returned

down through the top of the column. This action agitates the medium and causes foam which flows up the calibrated tube and allows determination of foam height at different time intervals. The calibrated column is wrapped with heat tape and kept at a constant temperature ($\pm 2^{\circ}\text{F}$) via a probe and temperature controller. The foam test cell holds approximately 1200 ml of medium. The column is filled with medium and recirculated briefly to evacuate all the air from the loop.

Examples

10

The defoamers noted below were tested using the synthetic medium at a temperature of 130°F and a pH of 7.5. The conductivity of the medium was 2400 umhos. Control time was 6.5 seconds to top.

15 Example 1 (Numbers below correspond to entries in Figure 1)

- 1) An emulsion antifoam containing hydrocarbon oil, ethylene bis stearamide, surfactants and water.
- 20 2) An emulsion antifoam of the present invention containing diisononyl adipate, wax, surfactants, silicon copolymer, ethylene bis stearamide, polyacrylate, water and sodium hydroxide.
- 25 3) An emulsion antifoam of the present invention containing dioctyl sebacate, ethylene bis stearamide, wax, surfactants, silicone copolymer, polyacrylate, water, and sodium hydroxide.
- 4) An emulsion antifoam of the present invention containing diisooctyl phthalate, ethylene bis stearamide, surfactants, and water.

5) An emulsion antifoam of the present invention containing diisononyl adipate, ethylene bis stearamide, surfactants, and water.

As shown in Figure 1, an amide particulate antifoam emulsion containing hydrocarbon oil (No. 1) is not as effective as the other examples which are variations of the present invention and contain various dicarboxylic acid esters. All formulations were run at a dosage of 10 ul.

10 Example 2 (Numbers below correspond to entries in Figure 2)

The defoamers listed below were tested using the synthetic medium at a temperature of 125° F and a pH of 7.0. The conductivity of the medium was 4600 umhos. Control time was 8 seconds to top.

15

6) An emulsion antifoam containing hydrocarbon oil, ethylene bis stearamide, surfactants and water.

7) An emulsion antifoam of the present invention containing diisononyl adipate, ethylene bis stearamide, surfactants, and water.

20

8) An emulsion antifoam of the present invention containing diisooctyl phthalate, ethylene bis stearamide, surfactants, and water.

9) An antifoam from U.S. Patent 3,959,175 containing polybutene and ethylene bis stearamide.

25

10) An emulsion antifoam of the present invention containing diisononyl adipate, wax, surfactants, silicon copolymer, ethylene bis stearamide, polyacrylate, water and sodium hydroxide.

5 As shown in Figure 2, tests 7, 8 and 10 provide superior performance when compared to the hydrocarbon oil/amide emulsion and the polybutene/amide based product. All products were tested at a dosage of 12 ul.

10 While this invention has been described with respect to particular embodiments thereof, it is apparent that numerous other forms and modifications of this invention will be obvious to those skilled in the art. The appended claims and this invention generally should be construed to cover all such obvious forms and modifications which are within the true
15 spirit and scope of the present invention.

We claim:

1. A water-based defoaming agent composition comprising:

(a) an ester of dicarboxylic acid and a monohydric alcohol;

5

(b) a hydrophobic material having a melting point greater than 40° C selected from the group consisting of hydrocarbon waxes, fatty amides, fatty acids and fatty alcohols; and

10

(c) a surfactant selected from the group consisting of polyethylene glycol esters, sorbitan/sorbitol esters, ethoxylated fatty alcohols, fatty acids and fatty ethers.

2. The composition as recited in claim 1 further comprising a polyether modified silicone copolymer.

3. The composition as recited in claim 1 further comprising an aqueous polyacrylate.

4. The composition as recited in claim 1 wherein said defoaming agent forms a stable dispersion at from about 10° - 50° C.

5. The composition as recited in claim 1 wherein said ester is a diisononyl adipate.

6. The composition as recited in claim 1 wherein said ester is a dioctyl sebacate.

7. The composition as recited in claim 1 wherein said ester is a diisooctyl phthalate.

8. The composition as recited in claim 1 wherein said amide is ethylene bis stearamide.

FIGURE 1

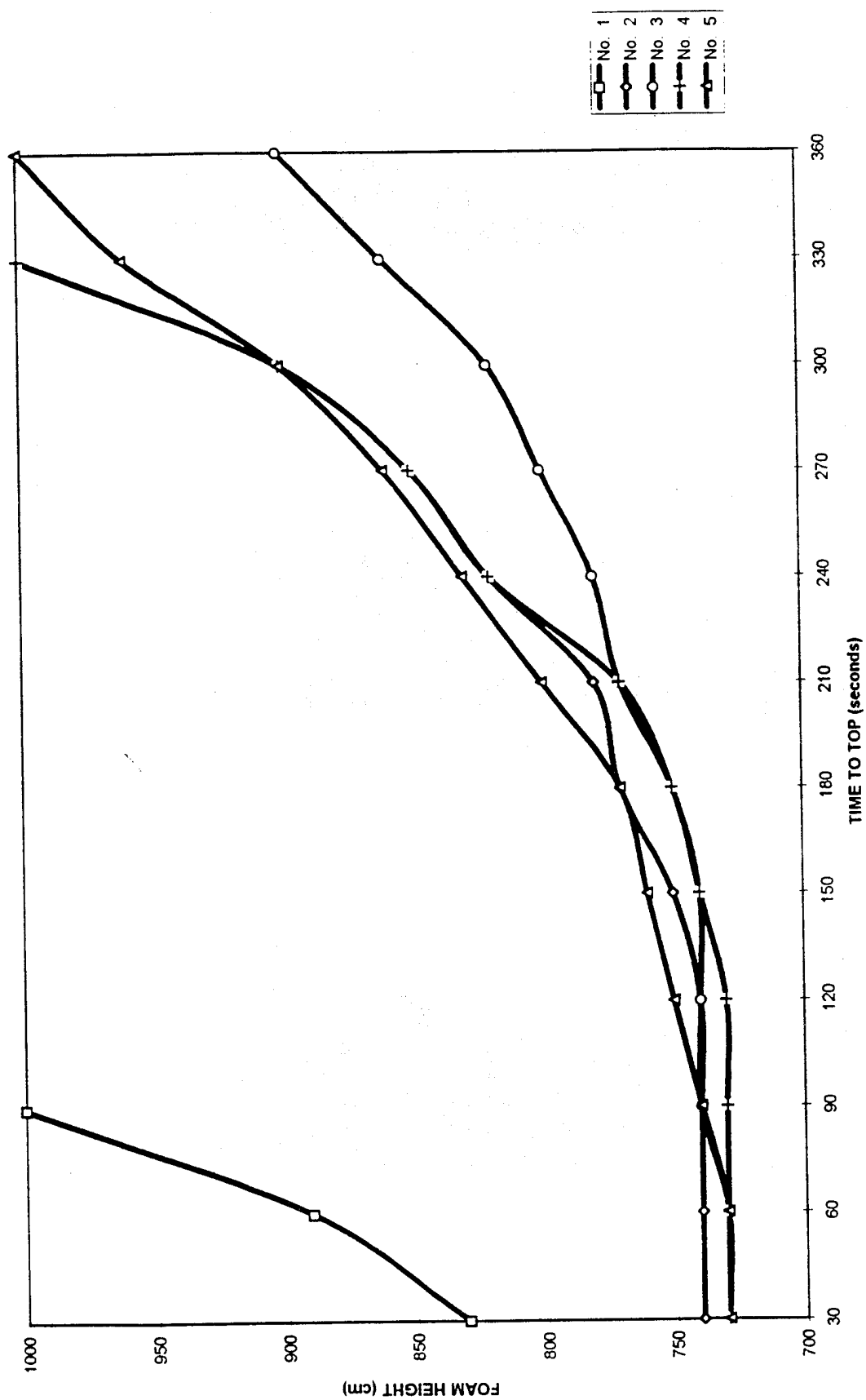
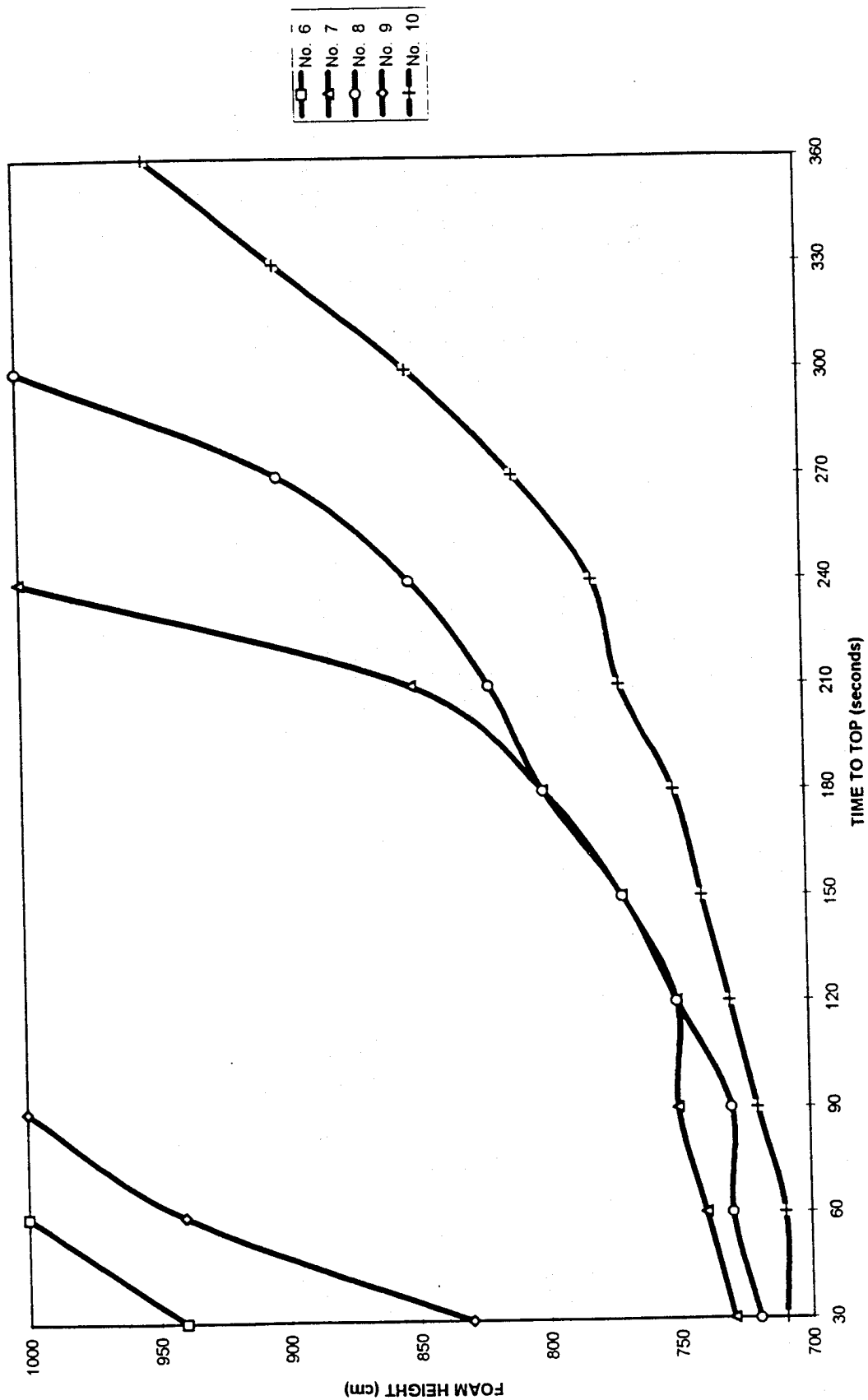


FIGURE 2



INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 00/03975

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 B01D19/04 D21H21/12

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B01D D21H

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

WPI Data, EP0-Internal, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 988 463 A (K. WALZ) 29 January 1991 (1991-01-29) the whole document ---	1,4
X	US 4 880 564 A (H. ABEL) 14 November 1989 (1989-11-14) column 2, line 18 - line 26 column 4, line 62 - line 64; claims 1,8,10 ---	1,4,8
X	US 4 341 656 A (H. ABEL) 27 July 1982 (1982-07-27) page 6, line 23 - line 26; claims 1,2,11,12 --- -/--	1



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

° Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"&" document member of the same patent family

Date of the actual completion of the international search

24 July 2000

Date of mailing of the international search report

31/07/2000

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Hilgenga, K

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 00/03975

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DE 32 42 202 A (SANDOZ PATENT) 1 June 1983 (1983-06-01) page 5, last paragraph -page 6, paragraph 1 page 15, line 1 page 8, paragraph 2 ---	1
A	US 5 431 853 A (YASUTAKA TSUDA) 11 July 1995 (1995-07-11) column 5, line 33; claims 1,7 ---	1,2
A	US 5 562 862 A (C.J. BERZANSKY) 8 October 1996 (1996-10-08) column 3, line 20 - line 22; claims 1,4 ---	1,7
A	EP 0 687 725 A (DOW CORNING) 20 December 1995 (1995-12-20) page 3, line 9 page 2, line 42 - line 43 page 3, line 50 ---	1,8
A	US 4 559 162 A (H. ABEL) 17 December 1985 (1985-12-17) column 12, line 34 -column 13, line 31; claims 1,19 ---	1-3
A	DATABASE WPI Section Ch, Week 199348 Derwent Publications Ltd., London, GB; Class E14, AN 1993-383978 XP002037175 & SU 1 775 125 A (KUSKOVO PLASTMASSY ASSOC CHEM WKS), 15 November 1992 (1992-11-15) abstract ---	1
A	DE 196 21 595 C (HENKEL) 11 September 1997 (1997-09-11) claim 1 ---	
A	WO 96 11733 A (HENKEL CORPORATION) 25 April 1996 (1996-04-25) claims 1,9,18 -----	

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 00/03975

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4988463 A	29-01-1991	DE 3810426 A AT 97012 T CA 1330029 A DE 58906129 D EP 0335173 A ES 2060681 T JP 1274806 A JP 2799180 B	12-10-1989 15-11-1993 07-06-1994 16-12-1993 04-10-1989 01-12-1994 02-11-1989 17-09-1998
US 4880564 A	14-11-1989	AT 91914 T AU 588132 B AU 7903987 A CA 1289435 A DE 3786738 A DK 510087 A EP 0263069 A ES 2058137 T IL 84013 A JP 1703895 C JP 3069563 B JP 63091106 A MX 168376 B ZA 8707275 A	15-08-1993 07-09-1989 31-03-1988 24-09-1991 02-09-1993 30-03-1988 06-04-1988 01-11-1994 15-04-1991 14-10-1992 01-11-1991 21-04-1988 20-05-1993 29-03-1988
US 4341656 A	27-07-1982	CH 637304 A AT 375838 B AT 705279 A BE 879772 A CA 1124613 A DE 2943754 A FI 793446 A FR 2440216 A GB 2038793 A, B IT 1126821 B JP 55067305 A NL 7907952 A NO 793535 A, B, SE 440027 B SE 7909066 A ZA 7905887 A	29-07-1983 10-09-1984 15-02-1984 30-04-1980 01-06-1982 14-05-1980 04-05-1980 30-05-1980 30-07-1980 21-05-1986 21-05-1980 07-05-1980 06-05-1980 15-07-1985 04-05-1980 29-10-1980
DE 3242202 A	01-06-1983	BR 8206812 A CH 665743 A FR 2516807 A GB 2112767 A, B HK 98585 A IT 1189426 B JP 1705664 C JP 3069562 B JP 58137408 A MY 13388 A ZA 8208697 A	04-10-1983 15-06-1988 27-05-1983 27-07-1983 13-12-1985 04-02-1988 27-10-1992 01-11-1991 15-08-1983 31-12-1988 27-06-1984
US 5431853 A	11-07-1995	CA 2094891 A CN 1080008 A JP 2537460 B JP 6039207 A	28-10-1993 29-12-1993 25-09-1996 15-02-1994

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 00/03975

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
US 5562862	A	08-10-1996	CA	2130888 A	07-10-1995
EP 687725	A	20-12-1995	AU	687539 B	26-02-1998
			AU	2171795 A	04-01-1996
			BR	9502640 A	05-03-1996
			CA	2151005 A	18-12-1995
			FI	952987 A	18-12-1995
			JP	8024512 A	30-01-1996
			US	5693256 A	02-12-1997
US 4559162	A	17-12-1985	CH	651581 A	30-09-1985
			DE	3208309 A	07-10-1982
			FR	2501523 A	17-09-1982
			GB	2094330 A, B	15-09-1982
SU 1775125	A	15-11-1992	NONE		
DE 19621595	C	11-09-1997	WO	9746648 A	11-12-1997
WO 9611733	A	25-04-1996	AU	3633895 A	06-05-1996
			CA	2202552 A	25-04-1996
			EP	0785815 A	30-07-1997
			FI	971495 A	10-06-1997
			NO	971680 A	10-06-1997
			US	5645762 A	08-07-1997