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Kamagata et al.

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[54] EPOXY RESIN COMPOSITION

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[58] Field of Search 525/113, 504; 528/93, 528/104, 118, 123, 361, 407

[56] References Cited

U.S. PATENT DOCUMENTS

3,242,131	3/1966	Peerman	528/118 X
3,265,664	8/1966	Fulmer et al.	528/118
4,021,403	5/1977	Fujiwara et al.	525/113 X
4,555,532	11/1985	Tanaka et al.	528/118 X

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[57] ABSTRACT

An epoxy resin composition comprising an epoxy resin, either 2-vinyl-4,6-diamino-s-triazine or 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct, and at least one compound selected from dicyandiamide, polyvinyl-p-phenol and specific imidazole compounds. The composition is useful, for example, in inks, insulating paints and adhesives for printed circuit base boards and in copper-clad laminated boards.

7 Claims, No Drawings

EPOXY RESIN COMPOSITION

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to an epoxy resin composition which has the ability to prevent migration of gold, silver, copper, etc. used as circuits in electronic component parts and printed base boards, and possesses excellent heat resistance, chemical resistance, moisture resistance, adhesion and insulation. It is useful, for example, for the production of sealing materials for electronic component parts, inks, insulating paints and adhesives for printed circuit base boards, and copper-clad laminated boards.

2. Description of the Prior Art

Epoxy resins have found widespread use in various industrial fields because they cure with, for example, aliphatic amines, aromatic amines, tertiary amines, polyamide resins or acid anhydrides to give resins having excellent heat resistance, electrical insulation and adhesion. Especially, it is known that blending of a certain imidazole compound, a kind of tertiary amine, with an epoxy resin can give a one-package epoxy resin composition having good workability or handlability and being capable of giving a cured product having excellent heat resistance (see U.S. Pat. No. 4,205,156).

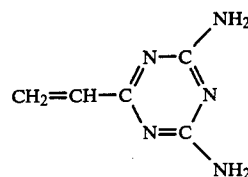
With the recent tendency to the smaller sizes, lighter weights and higher densities of electronic appliances and component parts, epoxy resins for sealing are required to have high insulating property and heat resistance.

Inks and insulating paints for printed circuit base boards and copper-clad laminated boards have been required likewise to have high insulation and heat resistance in fine and multilayered printed circuits. Furthermore, the prevention of migration of silver or copper constituting the circuits has also been rigorously required. Various resins have been developed and marketed in an attempt to meet these requirements, but none have proved to be entirely satisfactory in workability or handlability or in other properties. For example, these resins are not entirely satisfactory in regard to curing conditions and pot lives, or have large shrinkage on curing or low adhesion.

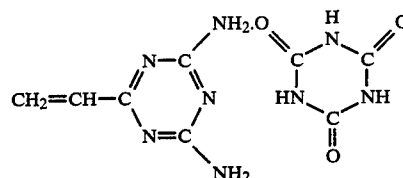
SUMMARY OF THE INVENTION

Extensive investigations of the present inventors conducted in view of the foregoing state of the art have now led to the discovery of a novel epoxy resin composition which has satisfactory heat resistance, electrical insulation, adhesion, moisture resistance, chemical resistance and the ability to prevent migration, can be formulated as a one-package system, and can cure at relatively low temperatures.

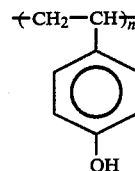
Thus, according to this invention, there is provided an epoxy resin composition comprising an epoxy resin, either 2-vinyl-4,6-diamino-s-triazine of the following formula



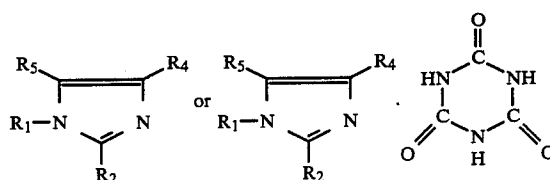
or 2-vinyl-4,6-diamino-s-triazine isocyanuric acid adduct of the following formula



and at least one compound selected from the group consisting of polyvinyl-p-phenol of the following formula



wherein n is 25 to 70, dicyandiamide and imidazole compounds represented by the formula



wherein R₁ represents a hydrogen atom or a cyanoethyl group or 4,6-diaminotriazinyl(-2')ethyl group, R₂ represents a methyl, ethyl, undecyl, heptadecyl or phenyl group, R₄ represents a hydrogen atom or a methyl, hydroxymethyl, benzyl or 2-ethyl-5-methylimidazolyl(-4)-methyl group, and R₅ represents a hydrogen atom or a methyl or hydroxymethyl group.

DETAILED DESCRIPTION OF THE INVENTION

The characteristic feature of the present invention is that a composition having the aforesaid excellent properties can be obtained from an epoxy resin and 2-vinyl-4,6-diamino-s-triazine or 2-vinyl-4,6-diamino-s-triazine isocyanuric acid adduct as essential ingredients and at least one third ingredient selected from the group consisting of certain imidazole compounds, dicyandiamide and polyvinyl-p-phenol.

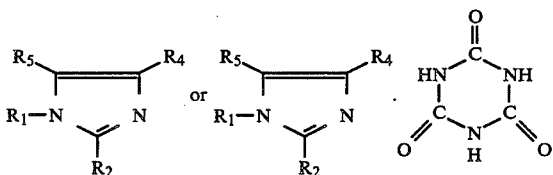
2-Vinyl-4,6-diamino-s-triazine can be synthesized, for example, by reacting biguanide with acryloyl chloride (J. Am. Chem. Soc., 80, 988 (1958)), reacting dicyandiamide with beta-dimethylaminopropionitrile (French Pat. No. 1563255), or heating 1,2-di(4',6'-diamino-s-

triazinyl-(2')-cyclobutane under reduced pressure (Japanese Patent Publication No. 35068/1971).

2-Vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct can be synthesized by heating the 2-vinyl-4,6-diamino-s-triazine prepared as above and isocyanuric acid in water in the presence of a polymerization inhibitor.

Examples of epoxy resins suitably used in this invention include polyglycidyl ethers of polyhydric phenols such as a diglycidyl ether of bisphenol A or a diglycidyl ether of bisphenol F, epoxidized phenolnovolac resins, alicyclic epoxy resins and brominated products of these epoxy resins.

Suitable imidazole compounds for use in this invention are compounds represented by the following formula



wherein R₁ represents a hydrogen atom or a cyanoethyl group or 4,6-diaminotriazinyl(-2')ethyl group, R₂ represents a methyl, ethyl, undecyl, heptadecyl or phenyl group, R₄ represents a hydrogen atom or a methyl, hydroxymethyl, benzyl or 2-ethyl-5-methylimidazolyl(-4)-methyl group, and R₅ represents a hydrogen atom or a methyl or hydroxymethyl group.

Examples of the imidazole compounds include 2,4-diamino-6-(2'-methylimidazolyl-(1'))ethyl-s-triazine, 2,4-diamino-6-(2'-methylimidazolyl-(1'))ethyl-s-triazine/isocyanuric acid adduct, 2-phenyl-4,5-dihydroxymethylimidazole, 2-phenyl-4-methyl-5-hydroxymethylimidazole, 2-phenyl-4-benzyl-5-hydroxymethylimidazole, and 4,4'-methylenebis-(2-ethyl-5-methylimidazole). When these imidazole compounds are blended with liquid epoxy resins, one-package compositions can be obtained which have excellent storage stability and can be cured at relatively low temperatures of about 150° C.

Other examples of the imidazole compound used in this invention are 2-methyl-imidazole, 2-ethyl-4-methylimidazole, 2-phenylimidazole, 2-undecylimidazole, 2-heptadecylimidazole, 2-phenyl-4-methylimidazole, 1-cyanoethyl-2-methylimidazole, 1-cyanoethyl-2-ethyl-4-methylimidazole, and 1-cyanoethyl-2-phenylimidazole. A group of these compounds are preferably added to epoxy resins immediately before use because blends of these with liquid epoxy resins have poor storage stability.

The curing mechanism of the composition of this invention is not certain, but the present inventors theorize as follows. The epoxy resin polymerizes by the action of the imidazole compound, polyvinyl-p-phenol or dicyandiamide. Simultaneously, the double bonds of 2-vinyl-4,6-diamino-s-triazine or 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct are cleaved to induce radical polymerization. Furthermore, the two amino groups on the triazine ring react with the epoxy resin. It is presumed that these three reactions take place almost simultaneously to give a cured product having excellent heat resistance and electrical properties.

Now, the invention will be described below with reference to inks and insulating paints for printed circuit

base boards and copper-clad laminated boards for the production of which the epoxy resin composition of this invention is useful.

The inks and insulating paints for printed circuit base boards are used for the following purposes.

In so-called thick film hybrid integrated circuit base boards obtained by forming the desired circuit including a silver-, platinum- or palladium-type conductor containing a glass frit as a binder, a resistor, a condenser etc. on an insulating base board such as a ceramic or glass board by screen printing, and firing the resulting board at high temperatures or printed circuit base boards obtained by forming the desired circuit by etching a copper foil on a paper, phenolic resin or glass-epoxy resin laminated board, the required circuit is normally exposed flat on the surface, and is therefore liable to be corroded by the adhesion of corrosive substances or moisture contained in the atmosphere.

In order to prevent such an accident by corrosion and to increase the reliability of the circuit, it is the general practice to form a protective coating, called a permanent resist, on the circuit by screen printing, etc.

The protective coating used in this case is required to have various properties such as electrical insulation, moisture resistance, chemical resistance, adhesion and hardness. Moreover, since the circuits in the aforesaid thick film hybrid integrated circuit base boards or printed circuit base boards tend to be minute and complex, plating of nickel, gold, copper, etc. or soldering might cause short-circuiting. It is known that an attempt is made to prevent this type of accident by coating a protective paint called a resist ink by screen printing, etc. on those parts which do not require plating, curing the coated film, and thereafter, plating the unprotected parts. The protective paint is required to have the functions of ink such as viscosity, roping property and thixotropic property suitable for screen printing, etc. Discoloration and migration of a conductor circuit of silver or copper which occur upon application of a voltage under high temperature and high pressure environmental conditions have become a problem as printed circuit base boards and the like have been required to have smaller sizes, higher densities and higher reliability so as to provide small-sized and light-weight electronic appliances having high reliability.

Except a triazine resin and a bismaleimide-triazine resin either singly or in combination, scarcely any resins have been found to be useful for protective paints.

Known conventional protective paints having the aforesaid properties and functions include, for example, paints of the fireable glass type, those of the heat-curable epoxy resin, phenol resin, melamine resin, epoxy-melamine and silicone resin types, and paints of the ultraviolet-curable epoxy-acrylate resin type. Except the paints of the glass type, these protective paints have only an insufficient ability to prevent discoloration and migration of silver or copper conductors. The aforesaid triazine resin-type and bismaleimide-triazine resin-type protective paints are not satisfactory with regard to curing conditions and pot lives.

It has now been found in accordance with this invention that the aforesaid defects can be eliminated by uniformly dispersing 10 to 40 parts by weight, per 100 parts by weight of the epoxy resin, of polyvinyl-p-phenol having an average degree of polymerization of 25 to 70 and a particle diameter of 200 to 300 mesh and 20 to 100 parts by weight, per 100 parts by weight of the

TABLE 2-continued

Composition (parts by weight)	Example										Comparative Example			
	3	4	5	6	7	8	9	10	11	12	2	3	4	
2-Vinyl-4,6-diamino-s-triazine/ isocyanuric acid adduct						10	10	20	10	10				
4. Polyvinyl-p-phenol (Resin M, Maruzen Oil Co., Ltd.)	20	10	10	10	10	20	10	10	10	10				
5. Filler:														
Talc	20	20	20			20	20	20						
Barium sulfate	60	60	60	60	30	60	60	60	60	30	30	30	60	
Mica				20	10					20	10	10	20	
6. Dispersant:														
Disparon NS-30 (Kusumoto Chemical Co.)	2	2	2			2	2	2						
Disparon NS-30 #230 (Kusumoto Chemical Co.)				0.3	0.3					0.3	0.3	0.3	0.3	
Disparon NS-30 #1900 (Kusumoto Chemical Co.)												0.3		
Disparon NS-30 #1930 (Kusumoto Chemical Co.)				0.3	0.3					0.3	0.3	0.3	0.3	
7. Flow adjusting agent: Aerosil #300 (Japan Aerosil Co.)				2	2					2	2	2	2	
8. Flow adjusting aid:														
Butyl carbitol acetate	20	10	10			20	10	10						
Cellosolve acetate				5						5	5	5		
Xylene					5						5		5	
9. Defoamer:														
KS603 (Shin-etsu Chemical Co., Ltd)	0.5	0.5	0.5			0.5	0.5	0.5						

TABLE 3

Example (Ex.) or Comparative Example (CEX.)	Gel time (seconds) (*1)	Storage stability (*2)		30
		25° C. (months)	40° C. (days)	
Ex. 3	95	2	10	
Ex. 4	95	2	10	
Ex. 5	105	2	10	
Ex. 6	70	1	6	
Ex. 7	50	1	6	
Ex. 8	95	2	10	
Ex. 9	95	2	10	
Ex. 10	90	2	10	
Ex. 11	80	1	6	
Ex. 12	60	1	6	

TABLE 3-continued

Example (Ex.) or Comparative Example (CEX.)	Gel time (seconds) (*1)	Storage stability (*2)	
		25° C. (months)	40° C. (days)
CEX. 2	80	2	20
CEX. 3	75	1	10
CEX. 4	80	1	6

35 (*1): Measured by the hot plate method. About 0.3 g of the composition was placed on a hot plate kept at 150° C., and spread thin by a metallic spatula, and the time which elapsed until no roping occurred between the spatula and the composition was determined and defined as the gel time.

(*2): The time which elapsed until the viscosity of the composition measured by a B-type rotary viscometer reached twice that in the initial stage was measured, and defined as the storage stability.

TABLE 4

Ex. or CEX.	Resistance to migration(*3)			Electrical insulation (ohm-cm)(*7)	Heat resis- tance to solder(*8)	Chemical resis- tance(*9)		Hard- ness (H)(*10)
	Cu conductor circuit on XPC substrate(*4)	Ag conductor circuit on XPC substrate(*5)	Ag conductor circuit on ceramic substrate(*6)			10% H ₂ SO ₄	10% NaOH	
Ex. 3	no change	no change	no change	>10 ⁴	no change	no change	no change	>7
Ex. 4	"	"	"	"	"	no change	no change	"
Ex. 5	"	"	"	"	"	no change	no change	"
Ex. 6	"	"	"	"	"	no change	no change	"
Ex. 7	"	"	"	"	"	no change	no change	"
Ex. 8	"	"	"	"	"	no change	no change	"
Ex. 9	"	"	"	"	"	no change	no change	"
Ex. 10	"	"	"	"	"	no change	no change	"
Ex. 11	"	"	"	"	"	no change	no change	"
Ex. 12	"	"	"	"	"	no change	no change	"
CEX. 2	Cathode side circuit discol- ored 100 hours later	Anode side cir- cuit discolored 24 hours later, and the discolored part disappeared 50 hours later	Anode side circuit discolored 200 hours later	>10 ⁴	no change	peeling occurred	no change	>7
CEX. 3	Cathode side circuit disco-	Anode side cir- cuit discolored 24	Anode side circuit discolored 200 hours	"	"	peeling occurred	no change	"

TABLE 4-continued

	lored 100 hours later	hours later, and the discolored part disappeared 50 hours later	later					
CEx. 4	Cathode side circuit discolored 100 hours later	Anode side circuit discolored 24 hours later, and the discolored part disappeared 50 hours later	Anode side circuit discolored 200 hours later	"	"	peeling occurred	no change	"

(*3) Migration in the test specimen was examined every predetermined time at a temperature of 60° C., a humidity of 95% and an applied voltage (DC) of 30 V.

(*4) A copper clad XPC laminate in accordance with the NEMA standards was used. The copper foil was etched away to form a comb-like electrode circuit with a conductor line in width of 0.5 mm and an interconductor spacing of 0.3 mm. The composition was screen-printed on the circuit and cured under heat. The resulting test specimen was left to stand for 500 hours under the aforesaid conditions.

(*5) By screen printing, a comb-like electrode circuit with a conductive line in width of 0.5 mm and an interconductor spacing of 0.3 mm was formed from a silver powder-phenolic resin paste on an XPC laminate in accordance with the NEMA standards, and baked at 150° C. for 30 minutes. Then, the composition was screen-printed on the circuit, and cured under heat. The resulting test specimen was left to stand for 100 hours under the conditions indicated in (*3) above.

(*6) By screen printing, a comb-like electrode circuit with a conductor line in width of 1 mm and an interconductor spacing of 0.5 mm was formed from a silver-palladium powder glass paste on a ceramic substrate, and fired at 700° C. for 2 hours. The composition was screen-printed on the circuit and cured under heat. The resulting test specimen was left to stand for 5 hours under the conditions indicated in (*3) above.

(*7) The test specimen was prepared by screen-printing the composition on an aluminum plate to form a protective film in accordance with JIS-C-2103.

(*8) The test specimen prepared by the method of (*4) above was dipped for 30 seconds in a molten solder bath at 260° C., and then observed for blisters, peeling, etc. on the surface of the coated film. Then, the specimen was subjected to a crosscut peeling test in accordance with JIS-D-0202.

(*9) The test specimen prepared by the method of (*4) above was immersed for 120 hours in a 10% sulfuric acid solution or a 10% sodium hydroxide solution, and then a change in the coated film was observed.

(*10) The pencil hardness of the test specimen prepared by the method (*4) was measured in accordance with JIS-D-0202.

EXAMPLES 13 TO 15 AND COMPARATIVE EXAMPLE 5

Two-package inks for printed circuit base boards were prepared in accordance with the formulations indicated in Table 5.

The mixing ratio of package A and package B should be determined according to the desired performance. In these examples, the weight ratio of package A to package B was adjusted to 100:13.6.

For comparison, a commercial ultraviolet-curable composition (Solder Resist UR3000, a trade name for a product of Sanwa Chemical Industry Co., Ltd.) was used (Comparative Example 5).

TABLE 5

Package	Composition (parts by weight)	Example			Comparative Example 5
		13	14	15	
A	1. Epoxy resin (EPIKOTE #828)	100	100	100	Commercial ultraviolet-curable composition (Solder Resist UR3000)
	2. 2-Vinyl-4,6-diamino-s-triazine	15	15	15	
	3. Fillers: Talc	20	20	20	
	Barium sulfate	60	60	60	
B	4. Dispersant: Disparon NS-30	2	2	2	
	5. Flow adjusting aid: butyl-carbitol acetate	20	10	10	
	6. Polyvinyl-p-phenol (Maruzen Resin M)	20	10	10	
	7. Curing agent				
	2,4-Diamino-6-(2'-methylimidazolyl-(1'))-ethyl-s-triazine	7.5	—	—	
	2,4-Diamino-6-(2'-methylimidazolyl-(1'))-ethyl-s-triazine/isocyanuric acid adduct	—	7.5	—	
	2-Ethyl-4-methyl-imidazole	—	—	3	
8.	Defoamer KS603 (Shin-etsu Chemical)	0.5	0.5	0.5	

The properties of the ink compositions were tested and the results are shown in Table 6. In Table 6, the gel time and the storage stability (pot life) were measured in the same way as indicated in the footnotes to Table 3.

TABLE 6

Example (Ex.) or Comparative Example (CEx.)	Gel time (*1) (seconds)	Storage stability (*2)		
		5° C. (months)	25° C. (days)	40° C. (days)
Ex. 13	25	4 (*b)	4 (*c)	2 (*c)
Ex. 14	35	4 (*b)	5 (*c)	2 (*c)
Ex. 15	20	4 (*b)	1 (*c)	2 (*c)
CEx. 5	20 (*a)	3	—	—

(*a): Three 80 W/cm ozoneless high-pressure mercury lamps were used for irradiation. The oven passing rate was 4 m/min.

(*b): Packages A and B were stored independently.

(*c): The pot life of a mixture of predetermined amounts of packages A and B.

Each of the compositions was coated to a film thickness of 20 microns by screen printing, and then cured at

150° C. for 15 minutes. The ultraviolet-curable ink of Comparative Example 5 was cured by using three 80 W/cm ozoneless high-pressure mercury lamps (speed 4 m/min.).

The properties of the cured products were tested in the same way as indicated in the footnotes to Table 4. The results are summarized in Table 7.

TABLE 7

Ex. or CEx.	Cu conductor	Ag conductor	Ag conductor circuit on ceramic substrate(*6)	Electrical insulation (ohm-cm)(*7)	Heat resis- tance to solder(*8)	Chemical resis- tance(*9)		Hardness (H)(*10)
	circuit on XPC substrate(*4)	circuit on XPC substrate(*5)				10% H ₂ SO ₄	10% NaOH	
Ex. 13	no change	no change	no change	>10 ⁴	no change	no change	no change	>7
Ex. 14	"	"	"	"	"	no change	no change	"
Ex. 15	"	"	"	"	"	no change	no change	"
CEX. 5	Cathode side circuit dis- colored 100 hours later	Anode side circuit discolored 24 hours later, and the discolored part disappeared 50 hours later	Anode side circuit discolored 200 hours later	"	"	peeled	peeled	>3

EXAMPLES 16 AND 17 AND COMPARATIVE EXAMPLE 6

In each run, a composition was prepared in accordance with the formulation shown in Table 8.

TABLE 8

Ingredients (parts by weight)	Exam- ple 16	Exam- ple 17	Comparative Example 6
Epomic R301-M80 (Mitsui Petro- chemical Industries (*d))	37.5		100
Epikote DX2480B70 (Yuka Shell) (*e)		100	
Epikote #828 (Yuka Shell)	20	10	
2-Vinyl-4,6-diamino-s- triazine (*f)	10	8	
Dicyandiamide		6.4	6.4
2-Ethyl-4-methylimidazole	0.7	0.8	0.8
Methyl ethyl ketone	71.2	40	
Methyl Cellosolve		53.1	62.8

(*d): Epomic R301-M80 is a 80% by weight methyl ethyl ketone solution of an epi-bis type solid epoxy resin.

(*e): Epikote DX248-B70 is a 70% by weight methyl ethyl ketone solution of brominated epoxy resin (bromine content 24-26%).

(*f): 2-Vinyl-4,6-diamino-s-triazine was kneaded in advance with Epikote #828 on a three-roll mill and dispersed until the particle diameter became less than 5 microns.

The composition in solution was impregnated with glass woven cloths (0.18 mm thick) treated with aminosilane, and dried in the air to remove the solvent.

The composition was pre-cured at 90° C. for 3 minutes (for Examples 16 and 17) or 150° C. for 3 minutes (for Comparative Example 6) to prepare prepreps.

Eight such prepreps were laid one on top of another, and copper foils (35 microns) were placed over the top and bottom of the assembly. The entire assembly was compressed at 170° C. and 50 to 60 kg/cm² for 90 minutes, and then heat-treated at 200° C. for 60 minutes to cure the epoxy composition.

The properties of the resulting copper-clad laminated board were measured, and the results are shown in Table 9.

TABLE 9

Properties	Example 16	Example 17	Comparative Example 6
Glass transition point (°C.) (*11)	185	180	150
Solder heat resistance (seconds) (*12)	>180	>180	30
Peel strength of the copper foils (kg/cm) (*13)	1.8	1.7	1.8

TABLE 9-continued

Properties	Example 16	Example 17	Comparative Example 6
Flame retardance (seconds) (*14)	burned	6	burned

(*11): The glass transition point of the copper-clad laminated board was measured by the TBA method after the copper foils were removed from it by etching and a thin piece was cut out from the board.

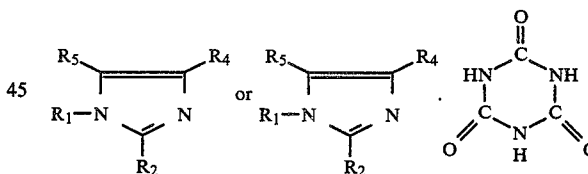
(*12): A square piece with each side measuring 25 mm was prepared from the copper-clad laminated board, and caused to float on a solder bath at 260° C. It was observed for changes such as blisters.

(*13): The peel strength of the copper foils is the strength measured when the copper foil was peeled perpendicularly from a 10 mm width test piece of the copper-clad laminated board at a speed of 5 mm/min.

(*14): Measured by the vertical method of UL Standards 94.

What is claimed is:

- 35 1. An epoxy resin composition comprising an epoxy resin, either 2-vinyl-4,6-diamino-s-triazine or 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct, and at least one compound selected from the group consisting of dicyandiamide, polyvinyl-p-phenol, and imidazole compounds represented by the following formula



- 50 wherein R₁ represents a hydrogen atom or a cyanoethyl group or 4,6-diaminotriazinyl(-2')ethyl group, R₂ represents a methyl, ethyl, undecyl, heptadecyl or phenyl group, R₄ represents a hydrogen atom or a methyl, hydroxymethyl, benzyl or 2-ethyl-5-methylimidazolyl(-4)-methyl group, and R₅ represents a hydrogen atom or a methyl or hydroxymethyl group.

- 55 2. An epoxy resin composition according to claim 1 which comprises 100 parts by weight of the epoxy resin, 5 to 50 parts by weight of either 2-vinyl-4,6-diamino-s-triazine or 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct, and at least one compound selected from the group consisting of 1 to 20 parts by weight of dicyandiamide, 10 to 40 parts by weight of polyvinyl-p-phenol, and 0.025 to 10 parts by weight of the imidazole compounds.

- 60 3. An epoxy resin composition according to claim 1 which comprises the epoxy resin, 2-vinyl-4,6-diamino-s-triazine and the imidazole compound.

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4. An epoxy resin composition according to claim 1 which comprises the epoxy resin, 2-vinyl-4,6-diamino-s-triazine, the imidazole compound and dicyandiamide.

5. An epoxy resin composition according to claim 1 which comprises the epoxy resin, 2-vinyl-4,6-diamino-s-triazine, the imidazole compound and polyvinyl-p-phenol.

6. An epoxy resin composition according to claim 1

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which comprises the epoxy resin, the 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct and the imidazole compound.

7. An epoxy resin composition according to claim 1 which comprises the epoxy resin, the 2-vinyl-4,6-diamino-s-triazine/isocyanuric acid adduct, the imidazole compound and polyvinyl-p-phenol.

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