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(54) PROCESSES FOR INTERMEDIATES FOR MACROCYCLIC COMPOUNDS

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(57) ABSTRACT

The present invention is directed to novel macrocyclic compounds of formula (I) and their pharmaceutically acceptable salts, hydrates or solvates:

$$\begin{array}{c} R_4 \\ CH_2)n_1 \\ CH_2)m \\ R_2 \\ R_1 \\ T - Z_3 \end{array}$$

wherein R₁, R₂, R₃, R₄, R₅, R₆, n₁, m, p Z₁, Z₂, and Z₃ are as describe in the specification. The invention also relates to compounds of formula (I) which are antagonists of the motilin receptor and are useful in the treatment of disorders associated with this receptor and with or with motility dysfunction.

General Synthetic Strategy to Conformationally-Defined Macrocycles of the Present Invention

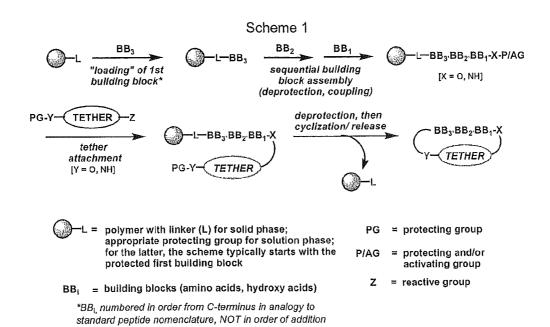


FIG. 1

FIG. 2

The yield of **Ddz-T9** from **T9-0** on a 65 g scale was 60.9 g (91%)

¹H NMR (CDCl₃): 8 7.19-7.01, (m, 2H), 6.92-9.83 (m, 2H), 6.53 (bs, 2H), 6.34 (t, 1H), 5.17 (bt, 1H), 4.08 (m, 2H), 3.98 (m, 2H), 3.79 (s, 6H), 3.01 (bq, 2H), 2.66 (t, 3H), 1.26 (bs, 8H);

¹³C NMR (CDCl₃) 8 160.9, 156.8, 155.6, 149.6, 130.4, 127.5, 121.2, 111.7, 103.2, 98.4, 80., 69.7, 61.6, 55.5, 40.3, 30.5, 29.3, 27.4

FIG. 3

Standard Procedure for the Synthesis of Ddz-propargylamine

FIG. 4

Standard Procedure for the Synthesis of Tether T10 *Method A*

TLC (EtOAc/Hexanes 1:1, detection: UV, ninhydrin; $R_f = 0.17$)

 1 H NMR (CDCl₃) & 7.18, t, 1H,J = 8.2Hz; 6.51, m, 5H; 6.34, t, 1H, J = 2.2Hz; 5.19, s, 1H; 4.05, t, 2H, J = 5.0Hz; 3.94, m, 4H; 3.75, s, 6H; 3.49, d, 2H J = 5.2Hz; 1.73, s, 6H.

¹³C NMR (CDCl₃) δ 160.856; δ 160.152; 160.005; 155.410; 149.305; 130.279; 107.438; 107.310; 103.163; 101.877; 98.517; 69.488; 67.382; 61.595; 55.427; 40.420; 29.427.

HPLC (standard gradient) t_R: 7.25 min

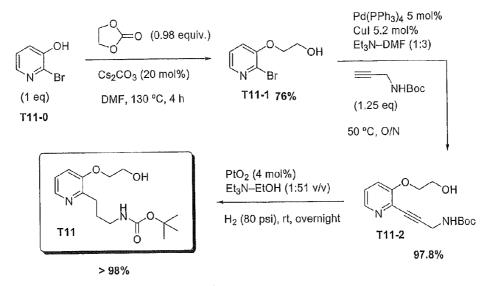
MS: 420 (M+H)

FIG. 5A

Method B

The second synthetic route to T10 is presented in the accompanying scheme.

FIG. 5B



TLC (15:85 THF/DCM; detection: UV; R_f: 0.33).

 1 H NMR (DMSO-d₆) δ 8.00, d, 1H; 7.32, d, 1H; 7.15, m, 1H; 6.44, s, 2H; 6.33, s, 1H; 3.99, t, 2H; 3.71, m, 8H; 2.89, m = 4, 2H; 2.71, t, 2H; 1.71, m = 5, 2H; 1.61, s, 6H.

 $^{13}\text{C NMR},$ solvent DMSO-d₆) & 160.879; 153.275; 151.405; 150.447; 140.773; 122.666; 118.934; 103.347; 98.456; 79.778; 70.449; 60.212; 55.717; 55.599; 29.740; 28.592.

HPLC (standard gradient) t_R: 5.4 min

MS: 419 (M+H)

FIG. 6

HO NH₂ HO NHDdz DdzN₃, DIPEA, TMG DMF,
$$50^{\circ}$$
C Ddz-T12

FIG. 7

Procedure for Synthesis of PPh₃-DIAD Adduct

FIG. 8

Sep. 6, 2012 Sheet 7 of 21

Standard Procedure for Attachment of Tethers via Reductive Amination

FIG. 9

Standard Procedure for the Synthesis of Tether T28

FIG. 10

TLC (100% EtOAc; detection: UV, CMA; $R_f = 0.24$).

 1 H NMR (CDCl₃, ppm): 7.74 (1H, dd), 7.35 (1H, d), 6.72 (1H, d), 6.53-6.49 (2H, m), 3.61-3.29 (1H, m), 5.06 (1H, t), 4.25-4.01 (2H, m), 3.91-3.89 (2H, m), 3.73 (3H, s), 2.99 (2H, dd), 2.63 (2H, t), 1.71 (8H, broad), 1.53 (9H, s).

¹³C NMR (CDCl₃, ppm): 163.8, 162.2, 161.0, 159.7, 155.9, 149.4, 130.0, 129.1, 128.0, 126.8, 110.8, 98.1, 80.9, 79.3, 69.7, 61.3, 55.5, 39.1, 29.3, 28.5, 26.7.

FIG. 11

Standard Procedure for the Synthesis of Tether T33a and T33b

 1 H NMR (CDCl₃) δ (ppm) 7.18-7.11 (m, 2H), 6.90 (m, 2H), 6.52 (m, 2H), 6.33(m, 1H), 5.09 (bt, 1H), 4.52 (m, 1H), 3.77 (s, 6H), 3.08 (bq, 2H), 2.64 (bt, 2H), 1.75 (m, 8H); 1.27 (bd, 3H), 13 C NMR (CDCl₃) δ 160.8, $^{-1}$ 155.5, 149.5, 131.2, 130.6, 127.4, 121.2, 113.3, 103.2, 98,4, 80.7, 74.8, 66.5, 55,4, 40.2, 30.6, 29.3, 29.2, 27.4, 16.1

FIG. 12A

FIG. 12B

TLC (100% EtOAc; detection: CMA, R_f = 0.5).

MW Calc. for $C_{24}H_{35}N_3O_7$, 477.55; MS Found $(M+H)^{\dagger}$ 478.

¹H NMR (CDCl₃) § 1.62 (m, 2H), 1.70 (m, 8H), 2.43 (m, 2H), 2.67 (m, 2H), 3.07 (m, 2H), 3.34 (s, 3H), 3.43 (s, 3H), 3.61 (m, 2H), 3.75 (s, 6H), 5.40 (sb, 1H), 6.31 (s, 1H), 6.49 (s, 2H)

¹⁸C NMR (CDCl₃) &23.25 ($\underline{C}H_2$), 25.97 ($\underline{C}H_2$), 28.56 ($\underline{C}H_3$), 39.31 ($\underline{C}H_3$), 30.09 ($\underline{C}H_3$), 31.25 ($\underline{C}H_2$), 32.19 ($\underline{C}H_2$), 40.16 ($\underline{C}H_2$), 55.47 ($\underline{C}H_3$), 61.38 ($\underline{C}H_2$), 80.65 ($\underline{C}q$), 99.38 ($\underline{C}q$), 103.17 ($\underline{C}q$), 111.01($\underline{C}q$), 149, 60 ($\underline{C}q$), 151.33 ($\underline{C}q$), 152.46 ($\underline{C}q$), 160.80 ($\underline{C}q$).

HPLC (standard gradient) t_R: 6.68 min.

FIG. 13

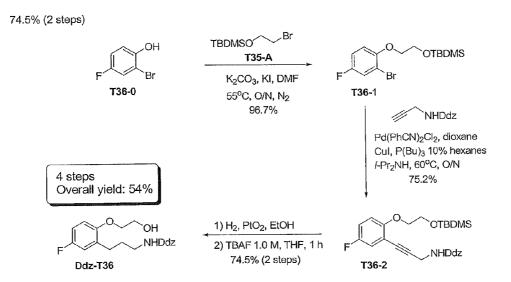
TLC (25/75 EtOAc/Hex; detection: UV, ninhydrin; $R_f = 0.03$)

 1 H NMR (CDCl₃): δ 7.06-7.00 (bt, 1H), 6.61-6.52 (m, 4H), 6.35 (m, 1H), 5.12 (bt, 1H), 4.03 (m, 2H), 3.95 (m, 2H), 3.77 (s, 6H), 3.11-3.04 (bq, 2H), 2.60 (bt, 2H), 1.75 (m, 8H)

¹³C NMR (CDCl₃): 8 163.9, 160.9, 160.6, 157.6, 157.5, 155.6, 149.5, 130.8, 130.6, 125.9, 107.26, 106.9, 103.2, 98,4, 80.8, 77.5, 69.9, 61,3, 60.9, 60.6, 55,4, 40.3, 30.4, 29.3, 26.9,

HPLC (standard gradient): t_R = 8.37 min

FIG. 14



TLC: (25/75 EtOAc/Hex; detection: UV, ninhydrin; $R_f = 0.03$)

 1 H NMR (CDCl₃) 8 (ppm): 6.84-6.75 (m, 3H), 6.52 (bs, 2H), 6.34 (m, 1H), 5.17 (bt, 1H), 4.01 (m, 2H), 3.93 (m, 2H), 3.77 (s, 6H), 3.10 (bq, 2H), 2.63 (bt, 2H), 1.74 (m, 8H)

 $^{13}\text{C NMR (CDCl}_3)$ δ 160.9, 158.9, 155.8, 155.6, 152.9, 152.9, 149.5, 132.4, 132.3, 117.1, 116.8, 112.7, 112.6, 103.2, 98.4, 80.8, 70.4, 61.6, 55.5, 40.2, 30.3, 29.3, 27.4.

HPLC (standard gradient): t_R = 8.29 min

FIG. 15

TLC (25/75 EtOAc/Hex; detection: UV, ninhydrin; $R_f = 0.03$)

 1 H NMR (CDCl₃): δ 7.12-7.08 (bd, 2H), 6.76-6.73 (d, 1H), 6.52 (m, 2H), 6.33 (bs, 1H), 5.15 (bt, 1H), 4.02 (m, 2H), 3.95 (m, 2H), 3.79 (s, 6H), 3.09 (bq, 2H), 2.61 (bt, 2H), 1.74 (m, 8H). 13 C NMR (CDCl₃) δ 160.8, 155.6, 155.4, 149.5, 132.4, 130.1, 127.0, 126.0, 112.8, 103.2, 98.4, 80.8, 70.0, 61.4, 55.5, 40.3, 30.2, 29.3, 24.5, 27.4

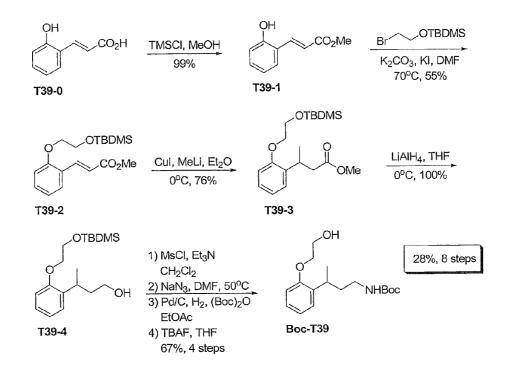
HPLC (standard gradient): t_R = 9.60 min

FIG. 16

 1 H NMR (CDCl₃): δ 7.20-7.10, (m, 2 H), 6.95-6.80 (m, 2 H), 6.55 (bs, 2 H), 6.35 (s, 1 H), 5.18 (bt, 1 H), 4.12 (m, 1 H), 3.95 (m, 2 H), 3.80 (s, 6 H), 3.15 (bq, 2 H), 2.65 (t, 2 H), 1.98 (bs, 2 H), 1.65 (bs, 6 H), 1.25 (m, 3 H).

¹³C NMR (CDCl₃): <u>8</u>160.8, 156.6, 155.8, 149.6, 130.4, 127.5, 121.3, 111.7, 103.2, 98.4, 80.7, 73.5, 66.6, 55.5, 40.2, 30.5, 29.3, 29.1, 27.3, 19.5.

FIG. 17

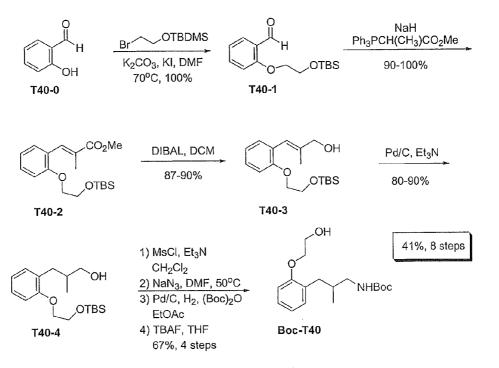


TLC (50% EtOAc, 50% Hex; detection: UV and CMA; $R_{\rm f}$ = 0.25)

¹H NMR (CDCl₃, ppm): 7.11-7.08 (2H, m), 6.86 (1H, t), 6.76 (1H, d), 5.05 (1H, broad), 4.26-3.85 (4H, m), 3.22-3.07 (2H, m), 2.71 (1H, broad), 1.66-1.60 (2H, m), 1.33 (9H, s), 1.17 (3H, d).

¹³C NMR (CDCl₃, ppm): 156.1, 135.0, 127.1, 127.0, 121.4, 111.7, 69.9, 61.5, 39.8, 38.4, 28.7, 20.7.

FIG. 18



TLC (50% EtOAc, 50% Hex; detection: UV and CMA; R_f = 0.25)

¹H NMR (CDCl₃, ppm): 7.11-7.08 (2H, m), 6.86 (1H, t), 6.76 (1H, d), 5.05 (1H, broad), 4.26-3.85 (4H, m), -3.22-3.07 (2H, m), 2.71 (1H, broad), 1.66-1.60 (2H, m), 1.33 (9H, s), 1.17 (3H, d).

¹³C NMR (CDCl₃, ppm): 156.1, 135.0, 127.1, 127.0, 121.4, 111.7, 69.9, 61.5, 39.8, 38.4, 28.7, 20.7.

FIG. 19

TLC (100% EtOAc; detection: CMA; $R_f = 0.5$)

 1 H NMR (CDCl₃) $_{8}$.1.23 (s, 3H), 1.49 (s, 3H), 1.69 (s, 3H), 1.74 (s, 3H), 1.90 (m, 2H), 2.35 (m, 1H), 3.35 (m, 2H), 3.76 (s, 6H), 3.92 (m, 2H), 4.40 (m, 2H), 5.10 (m, 1H), 6.15 (s, 1H), 6.25 (s, 2H).

 $^{13}C \text{ NMR } (\text{CDCl}_3) \text{ } \&25.52 \text{ } (\underline{\text{CH}}_3), 27.53 \text{ } (\underline{\text{CH}}_3), 28.88 \text{ } (\underline{\text{CH}}_3), 29.61 \text{ } (\underline{\text{CH}}_3), 35.92 \text{ } (\underline{\text{CH}}_2), 42.62 \text{ } (\underline{\text{CH}}_2), 55.43 \text{ } (\underline{\text{CH}}_3), 60.60 \text{ } (\underline{\text{CH}}_2), 82.38 \text{ } (\underline{\text{CH}}), 83.33 \text{ } (\underline{\text{CH}}), 83.68 \text{ } (\underline{\text{CH}}), 84.96 \text{ } (\underline{\text{CH}}), 98.26 \text{ } (\underline{\text{CH}}), 103.23 \text{ } (\underline{\text{CH}}), 118.3 \text{ } (\underline{\text{Cq}}), 149.50 \text{ } (\underline{\text{Cq}}), 156.20 \text{ } (\underline{\text{Cq}}), 160, 02 \text{ } (\underline{\text{Cq}})$

HPLC (standard gradient): t_R = 6.64 min

MS: M+H found: 439

FIG. 20

¹H NMR (300 MHz, CDCl₃) δ 6.82-6.98 (m, 2H); 6.80-6.75 (m, 1H); 6.53 (s, 2H); 6.35 (t, 1H, 2 Hz); 5.23 (b, 1H); 4.08 (m, 1H); 3.90-3.68 (m, 8H); 3.20-2.97 (m, 2H); 2.95-53 (m, 4H); 2.0-1.63 (m, 10H).

¹³C NMR (75.5 MHz, CDCl₃) δ160.85; 155.56; 152.55; 149.56; 128.13; 127.77; 120.28; 103.22; 98.43; 80.72; 76.80; 65.76; 55.46; 40.23; 30.45; 29.34; 29.22; 27.10; 24.97; 23.94.

FIG. 21

Scheme 2: Thioester Strategy for Macrocycl'ic Compounds of the Present Invention

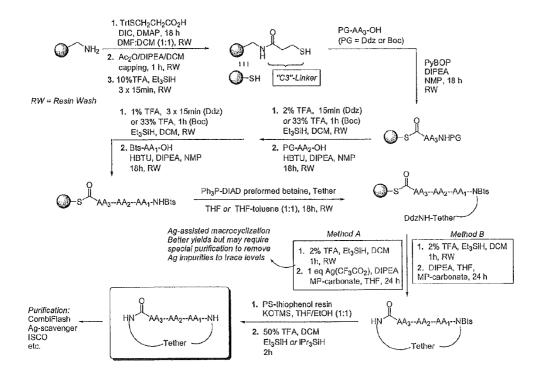
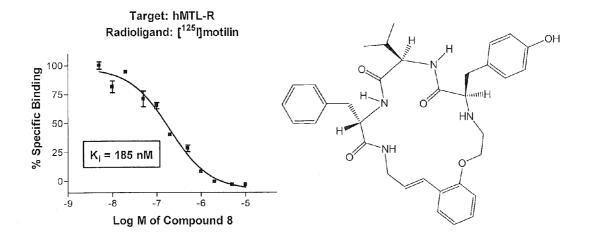
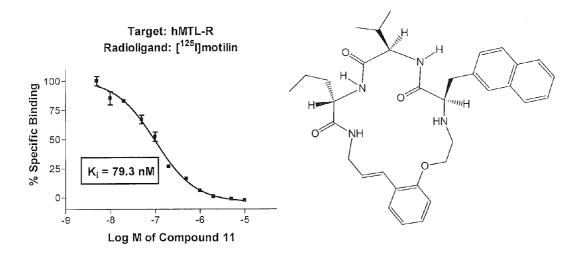


FIG. 22



Compound 8

FIG. 23



Compound 11

FIG. 24

PROCESSES FOR INTERMEDIATES FOR MACROCYCLIC COMPOUNDS

RELATED APPLICATION INFORMATION

[0001] This application is a divisional application of U.S. patent application Ser. No. 12/273,648, filed Nov. 19, 2008, now U.S. Pat. No. 8,129,561, which is a continuation application of U.S. patent application Ser. No. 12/273,638, filed Nov. 19, 2008, which is a continuation application of U.S. patent application Ser. No. 10/872,142, filed Jun. 18, 2004, now U.S. Pat. No. 7,521,420, which claims the benefit of U.S. Patent Application Ser. No. 60/479,223, filed Jun. 18, 2003. The disclosure of each application is incorporated herein by reference in its entirety.

FIELD OF THE INVENTION

[0002] The present invention relates to novel conformationally-defined macrocyclic compounds, pharmaceutical compositions comprising same and intermediates used in their manufacture. More particularly, the invention relates to macrocyclic compounds that have been demonstrated to selectively antagonize the activity of the motilin receptor. The invention further relates to macrocyclic compounds useful as therapeutics for a range of gastrointestinal disorders, in particular those in which malfunction of gastric motility or increased motilin secretion is observed, such as hypermotilinemia, irritable bowel syndrome and dyspepsia.

BACKGROUND OF THE INVENTION

[0003] A number of peptide hormones are involved in the control of the different functions in the gastrointestinal (GI) tract, including absorption, secretion, blood flow and motility (Mulvihill, et al. in *Basic and Clinical Endocrinology, 4th* edition, Greenspan, F. S.; Baxter, J. D., eds., Appleton & Lange: Norwalk, Conn., 1994, pp 551-570). Since interactions between the brain and GI system are critical to the proper modulation of these functions, these peptides can be produced locally in the GI tract or distally in the CNS.

[0004] One of these peptide hormones, motilin, a linear 22-amino acid peptide, plays a critical regulatory role in the GI physiological system though governing of fasting gastrointestinal motor activity. As such, the peptide is periodically released from the duodenal mucosa during fasting in mammals, including humans. More precisely, motilin exerts a powerful effect on gastric motility through the contraction of gastrointestinal smooth muscle to stimulate gastric emptying, decrease intestinal transit time and initiate phase III of the migrating motor complex in the small bowel (Itoh, Z., Ed., Motilin, Academic Press: San Diego, Calif., 1990, ASIN: 0123757304; Nelson, D. K. Dig. Dis. Sci. 1996, 41, 2006-2015; Peeters, T. L.; Vantrappen, G.; Janssens, J. Gastroenterology 1980, 79, 716-719).

[0005] Motilin exerts these effects through receptors located predominantly on the human antrum and proximal duodenum, although its receptors are found in other regions of the GI tract as well (Peeters, T. L.; Bormans, V.; Vantrappen, G. *Regul. Pept.* 1988, 23, 171-182). Therefore, motilin hormone is involved in motility of both the upper and lower parts of the GI system (Williams et al. *Am. J. Physiol.* 1992, 262, G50-G55). In addition, motilin and its receptors have been found in the CNS and periphery, suggesting a physiological role in the nervous system that has not yet been definitively elucidated (Depoortere, I.; Peeters, T. L. *Am. J.*

Physiol. 1997, 272, G994-999 and O'Donohue, T. L et al. Peptides 1981, 2, 467-477). For example, motilin receptors in the brain have been suggested to play a regulatory role in a number of CNS functions, including feeding and drinking behavior, micturition reflex, central and brain stern neuronal modulation and pituitary hormone secretion (Itoh, Z. Motilin and Clinical Applications. Peptides 1997, 18, 593-608; Asakawa, A.; Inui, A.; Momose, K.; et al., M. Peptides 1998, 19, 987-990 and Rosenfeld, D. J.; Garthwaite, T. L. Physiol. Behay. 1987, 39, 753-756). Physiological studies have provided confirmatory evidence that motilin can indeed have an effect on feeding behavior (Rosenfeld, D. J.; Garthwaite, T. L. Phys. Behay. 1987, 39, 735-736).

[0006] The recent identification and cloning of the human motilin receptor (WO 99/64436) has simplified and accelerated the search for agents which can modulate its activity for specific therapeutic purposes.

[0007] Due to the critical and direct involvement of motilin in control of gastric motility, agents that either diminish (hypomotility) or enhance (hypermotility) the activity at the motilin receptor, are a particularly attractive area for further investigation in the search for new effective pharmaceuticals towards these indications.

[0008] Peptidic agonists of the motilin receptor, which have clinical application for the treatment of hypomotility disorders, have been reported (U.S. Pat. Nos. 5,695,952; 5,721, 353; 6,018,037; 6,380,158; 6,420,521, U.S. Appl. 2001/0041791, WO 98/42840; WO 01/00830 and WO 02/059141). Derivatives of erythromycin, commonly referred to as motilides, have also been reported as agonists of the motilin receptor (U.S. Pat. Nos. 4,920,102; 5,008,249; 5,175,150; 5,418, 224; 5,470,961; 5,523,401, 5,554,605; 5,658,888; 5,854,407; 5,912,235; 6,100,239; 6,165,985; 6,403,775).

[0009] Antagonists of the motilin receptor are potentially extremely useful as therapeutic treatments for diseases associated with hypermotility and hypermotilinemia, including irritable bowel syndrome, dyspepsia, gastroesophogeal reflux disorders, Crohn's disease, ulcerative colitis, pancreatitis, infantile hypertrophic pyloric stenosis, diabetes mellitus, obesity, malabsorption syndrome, carcinoid syndrome, diarrhea, atrophic colitis or gastritis, gastrointestinal dumping syndrome, postgastroenterectomy syndrome, gastric stasis and eating disorders leading to obesity.

[0010] A variety of peptidic compounds have been described as antagonists of the motilin receptor (Depoortere, I.; Macielag, M. J.; Galdes, A.; Peeters, T. L. Eur. J. Pharmacol. 1995, 286, 241-247; U.S. Pat. Nos. 5,470,830; 6,255,285; 6,586,630; 6,720,433; U.S. 2003/0176643; WO 02/64623). These peptidic antagonists suffer from the known limitations of peptides as drug molecules, in particular poor oral bioavailability and degradative metabolism.

[0011] Cyclization of peptidic derivatives is a method employed to improve the properties of a linear peptide both with respect to metabolic stability and conformational freedom. Cyclic molecules tend to be more resistant to metabolic enzymes. Such cyclic tetrapeptide motilin antagonists have been reported (Haramura, M. et al *J. Med. Chem.* 2002, 45, 670-675, U.S. 2003/0191053; WO 02/16404).

[0012] Other motilin antagonists, which are non-peptidic and non-cyclic in nature have also been reported (U.S. Pat. Nos. 5,972,939; 6,384,031; 6,392,040; 6,423,714; 6,511,980; 6,624,165; 6,667,309; U.S. 2002/0111484; 2001/041701; 2002/0103238; 2001/0056106, 2002/0013352; 2003/0203906 and 2002/0002192)

(I)

[0013] The macrocyclic motilin antagonists of the present invention comprise elements of both peptidic and non-peptidic structures in a combination which has not been pursued for this application previously.

[0014] Indeed, the structural features of antagonists of the present invention are different. In particular, within the known motilin antagonists which are cyclic peptides, it was found that such derivatives containing D-amino acids were devoid of activity. In contrast, for the tripeptidomimetic compounds of the present invention, the D-stereochemistry is required for two of the three building elements.

[0015] The motilin antagonists of the present invention are also distinct from the prior art in that they comprise a tether element to fulfill the dual role of controlling conformations and providing additional sites for interaction either through hydrophobic interactions, hydrogen bonding or dipole-dipole interactions.

SUMMARY OF THE INVENTION

[0016] In a first aspect, the present invention is directed to compounds of formula (I):

$$R_4$$
 R_3
 R_4
 R_4
 R_5
 R_5
 R_6
 R_7
 R_7

and pharmaceutically acceptable salts, hydrates or solvates thereof wherein:

 $Z_t Z_2$ and Z_3 are independently selected from the group consisting of O, N and NR₁₀, wherein R₁₀ is selected from the group consisting of hydrogen, lower alkyl, and substituted lower alkyl;

R₁ is independently selected from the group consisting of lower alkyl substituted with aryl, lower alkyl substituted with substituted aryl, lower alkyl substituted with heteroaryl and lower alkyl substituted with substituted heteroaryl;

R₂ is hydrogen;

R₃ is independently selected from the group consisting of alkyl and cycloalkyl with the proviso that when Z₁ is N, R₃ can form a four, five, six or seven-membered heterocyclic ring together with Z_1 ;

R₄ is hydrogen;

R₅ and R₆ are independently selected from the group consisting of hydrogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl and substituted heteroaryl, with the proviso that at least one of R₅ and R₆ is hydrogen;

X is selected from the group consisting of O, NR_8 , and $N(R_9)$

[0017] wherein R_8 is selected from the group consisting of hydrogen, lower alkyl, substituted lower alkyl, formyl, acyl, carboxyalkyl, carboxyaryl, amido, sulfonyl, sulfonamido and amidino; and

[0018] R_o is selected from the group consisting of hydrogen, lower alkyl, and substituted lower alkyl;

m, n_1 and p are independently selected from 0, 1 or 2; and T is a bivalent radical of formula II:

$$-U-(CH_2)_d-W-Y-Z-(CH_2)_e-$$
 (II)

[0019] wherein d and e are independently selected from 0, 1, 2, 3, 4 or 5;

[0020] wherein U is bonded to X of formula (I) and is -CH₂-- or --C(=-O)--;

[0021] wherein Y and Z are each optionally present;[0022] W, Y and Z are independently selected from the group consisting of: -O-, -NR₂₈-, -S-, $-SO_{-}$, $-SO_{2}^{-}$, $-C(=O)_{-}$, $-\tilde{O}-C(=O)_{-}$, $-C(=O)-NH^-$, -NH-C(=O)-, $-SO_2-NH-$, $-NH-SO_2-$, $-CR_{29}R_{30}-$, -CH=CH- with a configuration Z or E, and -C=C-, or from a ring structure independently selected from the group:

$$R_{32}$$
 R_{31}
 G_{1}
 G_{2}
 G_{1}
 G_{2}
 G_{3}
 G_{4}
 G_{5}
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 G_{3}
 G_{4}
 G_{5}
 G_{5}
 G_{1}
 G_{5}
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 G_{7}
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 G_{2}
 G_{3}
 G_{4}
 G_{5}
 G_{5}
 G_{5}
 G_{7}
 G_{7}
 G_{7}
 G_{7}
 G_{8}
 $G_$

[0023] wherein any carbon atom contained within said ring structure, can be replaced by a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen atoms and if said ring structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

[0024] G1 and G2 each independently represent a covalent bond or a bivalent radical selected from the group consisting of -O—, $-NR_{41}$ —, -S—, -SO—, $-SO_2$ —, -C(=O)—, -C(=O)—-O—. —C(=O)NH—, —NH—C(=O)—, (=O)—, $-NH-SO_2-$, $-CR_{42}R_{43}-$ —CH—CH— with a configuration Z or E, and -C = C; with the proviso that G_1 is bonded closer to U than G₂;

[0025] K_1 , K_2 , K_3 , K_4 , K_6 , K_{15} and K_{16} are independently selected from the group consisting of O, NR₄₄ and S;

[0026] f is selected from 1, 2, 3, 4, 5 or 6;

[0027] R_{31} , R_{32} , R_{38} , R_{39} , R_{48} and R_{49} are independently selected from hydrogen, halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido; and

[0028] R₃₃, R₃₄, R₃₅, R₃₆, R₃₇, R₄₇, R₅₀ and R₅₁ are independently selected from hydrogen, halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido.

[0029] In a second aspect, the invention also proposes compounds of formula (I) which are antagonists of the motilin receptor.

[0030] In a third aspect, the invention proposes a method of treating a disorder associated with the motilin receptor or motility dysfunction in humans and other mammals, comprising administering a therapeutically effective amount of a compound of formula (I).

[0031] While the invention will be described in conjunction with example embodiments, it will be understood that it is not intended to limit the scope of the invention to such embodiment. On the contrary, it is intended to cover all alternatives, modifications and equivalents as may be included as defined by the appended claims.

DETAILED DESCRIPTION OF THE INVENTION

[0032] Preferably in formula (I), as depicted hereinabove, R_1 is selected from the group consisting of —(CH₂) $_qR_{11}$, and —CHR $_{12}R_{13}$

[0033] wherein q is 0, 1, 2 or 3; and

[0034] R_{11} and R_{12} are independently selected from a ring structure from the following group:

$$F$$

$$F$$

$$F$$

$$A_{1}$$

$$A_{2}$$

$$A_{3}$$

$$A_{4}$$

$$A_{2}$$

$$A_{3}$$

$$A_{4}$$

$$A_{4}$$

$$A_{5}$$

$$A_{5}$$

$$A_{5}$$

$$A_{6}$$

$$A_{1}$$

$$A_{2}$$

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$$A_{8}$$

$$A_{9}$$

-continued A_{1} A_{2} A_{3} A_{4} A_{3} A_{4} A_{4} A_{4} A_{5} A_{1} A_{2} A_{4} A_{4} A_{4} A_{4} A_{5} A_{6} A_{7} A_{1} A_{2} A_{3} A_{4} A_{4} A_{5} A_{6} A_{7} A_{8} A_{1} A_{2} A_{3} A_{4} A_{4} A_{5} A_{6} A_{7} A_{8} A_{1} A_{2} A_{3} A_{4} A_{5} A_{6} A_{7} A_{8} A_{9} A_{1} A_{2} A_{1} A_{2} A_{3} A_{4} A_{5} A_{7} A_{8} A_{9} A_{1} A_{2} A_{1} A_{2} A_{3} A_{4} A_{5} A_{7} A_{8} A_{1} A_{2} A_{3} A_{4} A_{5} A_{7} A_{8} A_{8}

[0035] wherein any carbon atom in said ring structure can be replaced a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen atoms and if said ring structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

[0036] A₁, A₂, A₃, A₄ and A₅ are each optionally present and are independently selected from the group consisting of halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido;

[0037] B₁, B₂, B₃, and B₄ are independently selected from NR₁₄, S or O, wherein R₁₄ is selected from the group consisting of hydrogen, alkyl, substituted alkyl, formyl, acyl, carboxyalkyl, carboxyaryl, amido, sulfonyl and sulfonamido;

[0038] R_{13} is as defined for as R_{11} and R_{12} or is selected from the group comprising lower alkyl, substituted lower alkyl, hydroxy, alkoxy, aryloxy, amino, carboxy, carboxyalkyl, carboxyaryl, and amido.

wherein A_1 , A_2 , A_3 , A_4 and A_5 are most preferably selected from halogen, trifluororomethyl, C_{1-6} alkyl or C_{1-6} alkoxy.

[0039] Preferably, R_{11} , R_{12} and R_{13} are selected from the group consisting of:

$$R_a$$
 R_a R_a

-continued

wherein R_a and R_b are chosen from the group consisting of Cl, F, CF₃, OCH₃, OH, and C(CH₃)₃ and CH₃.

[0040] Also preferably, R_3 in formula (I), is selected from the group consisting of:

 $\begin{array}{lll} \textbf{[0041]} & -(\text{CH}_2)_s \text{CH}_3, & -\text{CH}(\text{CH}_3)(\text{CH}_2)_t \text{CH}_3, & -\text{CH} \\ & (\text{OR}_{15})\text{CH}_3, & -\text{CH}_2 \text{SCH}_3 - \text{CH}_2 \text{CH}_2 \text{SCH}_3, & -\text{CH}_2 \text{S} \\ & (=\!\text{O})\text{CH}_3, & -\text{CH}_2 \text{CH}_2 \text{S} (=\!\text{O})\text{CH}_3, & \text{CH}_2 \text{S} (=\!\text{O})_2 \text{CH}_3, \\ & -\text{CH}_2 \text{CH}_2 \text{S} (=\!\text{O})_2 \text{CH}_3, & -(\text{CH}_2)_u \text{CH}(\text{CH}_3)_2, \\ & -\text{C}(\text{CH}_3)_3, & \text{and} -(\text{CH}_2)_y - \text{R}_{21}, & \text{wherein:} \end{array}$

[0042] s and u are independently selected from 0, 1, 2, 3, 4 or 5;

[0043] t is independently selected from 1, 2, 3 or 4;

[0044] y is selected from 0, 1, 2, 3 or 4;

[0045] R₁₅ is selected from the group consisting of hydrogen, alkyl, substituted alkyl, formyl and acyl;

[0046] R₂₁ is selected from a ring structure selected from the following group:

$$E_1$$
 E_2
 E_3
 E_3
 E_4
 E_3
 E_4
 E_5
 E_5

[0047] wherein any carbon atom in said ring structure can be replaced by a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen

atoms and if said ring structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

[0048] z is selected from 1, 2, 3, 4 or 5;

[0049] E₁, E₂ and E₃ are each optionally present and are independently selected from the group consisting of halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido; and

[0050] J is optionally present and is selected from the group consisting of alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, mercapto, sulfinyl, sulfonyl and sulfonamido.

[0051] The tether portion (T) of formula (I) is preferably selected from the group consisting of:

$$(X) \qquad \qquad L_1 \qquad \qquad (Z_3) \qquad \qquad (X) \qquad \qquad (Z_3) \qquad$$

R₅₆ -continued
$$(Z_3)$$

$$R_{55}$$

$$R_{57}$$

$$(X)$$

$$R_{54}$$

wherein L_1 is O, NH or NMe; L_2 is CH or N; L_3 is CH or N; L_4 is O or CH₂; L_5 is CH or N L_6 is CR₅₂R₅₃ or O; R₄₆ is H or CH₃:

 $R_{52},\,R_{53},\,R_{54},\,R_{55},\,R_{56}$ and R_{57} are independently selected from hydrogen, lower alkyl, substituted lower alkyl, hydroxy, alkoxy, aryloxy, amino, and oxo; or R_{52} together with R_{53} or R_{54} together with R_{55} or R_{56} together with R_{57} can independently form a three to seven-membered cyclic ring comprising carbon, oxygen, sulfur and/or nitrogen atoms;

(X) is the site of a covalent bond to X in formula (I); and (Z_3) is the site of a covalent bond to Z_3 in formula (I).

[0052] In a particularly preferred embodiment of the invention, there are provided compounds of formula (I) wherein m, n and p are $0, X, Z_1, Z_2$ and Z_3 are NH and R_2, R_4 and R_5 are hydrogen, represented by formula (III):

[0053] According to another aspect of the invention, there are provided compounds of formula (I) wherein when Z_1 is a nitrogen atom, R_3 forms a four, five, six or seven-membered heterocyclic ring together with Z_1 , represented by formula (IV):

$$(CH_2)_m$$

$$R_2$$

$$R_1$$

$$T - Z_3$$

$$(CH_2)_p$$

$$T - Z_3$$

$$(IV)$$

wherein said heterocyclic ring may contain a second nitrogen atom, or an oxygen, or sulfur atom;

n₂ is selected from 0, 1, 2 or 3

 R_7 is optionally present and is selected from the group consisting of alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy,

alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, mercapto, sulfinyl, sulfonyl and sulfonamido.

[0054] It is to be understood, that in the context of the present invention, the terms amino, guanidine, ureido and amidino encompass substituted derivatives thereof as well.

[0055] Preferably, the invention provides a method of treating a disorder associated with hypermotility or hypermotilinemia in humans and other mammals comprising administering a therapeutically effective amount of a compound of formula (I).

BRIEF DESCRIPTION OF THE DRAWINGS

[0056] FIG. 1 depicts Scheme 1 presenting a general synthetic strategy to conformationally-defined macrocycles of the present invention.

[0057] FIG. 2 depicts the standard procedure for the synthesis of tether T8 of Example 16.

[0058] FIG. 3 depicts the standard procedure for the synthesis of tether T9 of Example 17.

[0059] FIG. 4 depicts the standard procedure for the synthesis of Ddz-propargylamine of Example 18.

[0060] FIG. 5A depicts the standard procedure for the synthesis of tether T10 of Example 19.

[0061] FIG. 5B depicts the second synthetic route to tether T10 of Example 19.

[0062] FIG. 6 depicts the standard procedure for the synthesis of Tether T11 of Example 20.

[0063] FIG. 7 depicts the standard procedure for the synthesis of tether T12 of Example 26.

[0064] FIG. 8 depicts the procedure for synthesis of PPh₃-DIAD adduct of Example 29-C.

[0065] FIG. 9 depicts the standard procedure for attachment of tethers via reductive amination of Example 30.

[0066] FIG. 10 depicts the standard procedure for the synthesis of tether T28 of Example 32.

[0067] FIG. 11 the standard procedure for the synthesis of tether T32 of Example 36.

[0068] FIGS. 12A, 12B depict the standard procedure for the synthesis of tether T33a and T33b of Example 37.

[0069] FIG. 13 depicts the standard procedure for the synthesis of tether T34 of Example 38.

[0070] FIG. 14 depicts the standard procedure for the synthesis of tether T35 of Example 39.

[0071] FIG. 15 depicts the standard procedure for the synthesis of tether T36 of Example 40.

[0072] FIG. 16 depicts the standard procedure for the synthesis of tether T37 of Example 41.

[0073] FIG. 17 depicts the standard procedure for the synthesis of tether T38 of Example 42. Chiral T38 can be accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature. HPLC (standard gradient) t_R =8.46 min

Chiral material can be accessed by starting with the chiral epoxide. For example, the (S)-isomer of T38 was constructed in 89% overall yield from (S)-propylene oxide.

[0074] FIG. 18 depicts the standard procedure for the synthesis of tether T39 of Example 43. Chiral T39 can be accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature.

[0075] FIG. 19 depicts the standard procedure for the synthesis of tether T40 of Example 44. Chiral T40 can be

accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature.

[0076] FIG. 20 depicts the standard procedure for the synthesis of tether T41 of Example 45.

[0077] FIG. 21 depicts the standard procedure for the synthesis of tether T42 of Example 46.

[0078] FIG. 22 depicts Scheme 2 of the thioester strategy for macrocyclic compounds of the present invention.

[0079] FIG. 23 depicts the competitive binding curve for compound 8.

[0080] FIG. 24 depicts the competitive binding curve for compound 11

DESCRIPTION OF PREFERRED EMBODIMENTS

[0081] Although preferred embodiments of the present invention have been described in detail herein and illustrated in the accompanying structures, schemes and tables, it is to be understood that the invention is not limited to these precise embodiments and that various changes and modifications may be effected therein without departing from the scope or spirit of the present invention.

[0082] Specifically preferred compounds of the present invention, include, but are not limited to:

156

155

-continued

-continued

ONH HN NH NH 15

ONH HN

-continued

[0083] In addition to the preferred tethers (T) illustrated previously, other specific tethers employed for compounds of the invention are shown hereinbelow:

-continued

T33a [(R)-isomer] T33b [(S)-isomer]

[0084] In a preferred embodiment, the present invention is directed to a method of treating irritable bowel syndrome, dyspepsia, Crohn's disease, gastroesophogeal reflux disorders, ulcerative colitis, pancreatitis, infantile hypertrophic pyloric stenosis, carcinoid syndrome, malabsorption syndrome, diarrhea, diabetes mellitus, obesity, postgastroenterectomy syndrome, atrophic colitis or gastritis, gastric stasis, gastrointestinal dumping syndrome, celiac disease and eating disorders leading to obesity in humans and other mammals comprising administering a therapeutically effective amount of a compound of formula (I).

with the synthetic protocol, such as Boc, Ddz, Fmoc, or Alloc

Synthetic Methods

A. General Information

[0085] Reagents and solvents were of reagent quality or better and were used as obtained from various commercial suppliers unless otherwise noted. DMF, DCM and THF used are of DriSolv® (EM Science, now EMD Chemicals, Inc., part of Merck KgaA, Darmstadt, Germany) or synthesis grade quality except for (i) deprotection, (ii) resin capping reactions and (iii) washing. NMP used for the amino acid (AA) coupling reactions is of analytical grade. DMF was adequately degassed by placing under vacuum for a minimum of 30 min prior to use. Tyr(3tBu) was synthesized following the method reported in JP2000 44595. Cpa was made using literature methods (Tetrahedron: Asymmetry 2003, 14, 3575-3580) or obtained commercially. Boc- and Fmoc-protected amino acids and side chain protected derivatives, including those of N-methyl and unnatural amino acids, were obtained from commercial suppliers or synthesized through standard methodologies known to those in the art. Ddz-amino acids were either synthesized by standard procedures or obtained commercially from Orpegen (Heidelberg, Germany) or Advanced ChemTech (Louisville, Ky., USA). Bts-amino acids were synthesized as described in Example 6. Hydroxy acids were obtained from commercial suppliers or synthesized from the corresponding amino acids by literature methods. Analytical TLC was performed on pre-coated plates of silica gel 60F254 (0.25 mm thickness) containing a fluorescent indicator. The term "concentrated/evaporated under reduced pressure" indicates evaporation utilizing a rotary evaporator under either water aspirator pressure or the stronger vacuum provided by a mechanical oil vacuum pump as appropriate for the solvent being removed. "Dry pack" indicates chromatography on silica gel that has not been pre-treated with solvent, generally applied on larger scales for purifications where a large difference in R_f exists between the desired product and any impurities. For solid phase chemistry processes, "dried in the standard manner" is that the resin is dried first in air (1 h), and subsequently under vacuum (oil pump usually) until full dryness is attained (~30 min to O/N).

B. Synthetic Methods for Building Blocks of the Invention

Example 6

Standard Procedure for the Synthesis of Bts-Amino Acids

[0086]

$$\bigoplus_{\text{H}_3\text{N}} \bigoplus_{\text{CO}_2} \bigoplus_{\text{S}} \bigoplus_{\text{SO}_2\text{CI}} \bigoplus_{\text{BtsNH}} \bigoplus_{\text{CO}_2\text{H}} \bigoplus_{\text{R}_{44}} \bigoplus_{\text{R}_{44}} \bigoplus_{\text{CO}_2\text{H}} \bigoplus_{\text{CO}_2\text{$$

[0087] To a solution of the amino acid or amino acid derivative (0.1 mol, 1.0 eq) in 0.25 N sodium hydroxide (0.08 mol, 0.8 eq) with an initial pH of approximately 9.5 (pH meter) at rt, solid Bts-Cl (0.11 mol, 1.1 eq) was added in one portion. The resulting suspension was stirred vigorously for 2-3 d. The pH of the reaction should be adjusted with 5.0 N sodium hydroxide as required to remain within the range 9.5-10.0 during this time. Typically, the pH has to be adjusted every 20-30 min during the first 5 h. Once the pH stops dropping, it is an indication that the reaction is almost complete. This can be confirmed by TLC (EtOAc:MeOH, 95:5). Upon completion, the reaction mixture was washed with Et₂O. Washing is continued until the absence of non-polar impurities in the aqueous layer is confirmed by TLC (typically 3×100 mL). The aqueous solution was then cooled to 0° C., acidified to pH 2.0 with 1 N HCl until no additional cloudiness forms, and extracted with EtOAc (3×100 mL). Alternatively, a mixture of DCM and EtOAc may be used as the extraction solvent, depending on the solubility of the product obtained from different amino acids or derivatives. Note that DCM cannot be used solely as solvent because of the emulsion formed during extraction. The combined organic phases were washed with brine (2×150 mL), dried over MgSO₄, filtered and evaporated under reduced pressure. DCM $(1\times)$ and hexanes (2x) were evaporated from the residue in order to ensure complete removal of the EtOAc and give the desired compound as a solid in 55-98% yield.

[0088] The following are modifications that have proven useful for certain amino acids:

[0089] Gly, Ala, D-Ala, β -Ala and GABA: Use 1.5 eq of amino acid per eq of Bts-Cl, in order to prevent dibetsylation.

[0090] Met: Carry out the reaction under N_2 to prevent oxidation.

[0091] Gln and Asn: Due to the solubility of Bts-Gln and Bts-Asn, the work-up required is modified from the standard procedure: Upon completion of the reaction, the reaction mixture was washed with diethyl ether. Washing is continued until the absence of non-polar impurities in the aqueous layer is confirmed by TLC (typically 3×100 mL). The aqueous phase was then cooled to 0° C. and acidified to pH 2.0 with 6 N HCl. 6 N HCl was employed to minimize the volume of the solution due to the water solubility of Bts-Gln and Bts-Asn. (They are, in contrast, difficult to dissolve in DCM, EtOAc or chloroform.) The solution was maintained at 0° C. for 10 min and the product was collected by filtration as a white precipitate. The solid was washed with cold water $(1\times)$, cold brine $(2\times)$ and water $(1\times, 25^{\circ} \text{ C.})$. The pH of this wash was taken, if it is not approximately 4, the solid was washed again with water. Finally, the solid was washed with cold EtOAc, then with cold Et₂O (2x), and finally dried under vacuum (oil pump) (83-85% yield).

C. General Synthetic Strategy to Conformationally-Defined Macrocycles of the Present Invention

[0092] The compounds of Formula I can be synthesized using traditional solution synthesis techniques or solid phase chemistry methods. In either, the construction involves four phases: first, synthesis of the building blocks, including one to four moieties, comprising recognition elements for the biological target receptor, plus one tether moiety, primarily for control and definition of conformation. These building blocks are assembled together, typically in a sequential fashion, in a second phase employing standard chemical transformations. The precursors from the assembly are then cyclized in the third stage to provide the macrocyclic structures. Finally, a post-cyclization processing stage involving removal of protecting groups and optional purification then provides the desired final compounds (see FIG. 1). This method has been previously disclosed in WO 01/25257 and U.S. patent application Ser. No. 09/679,331. A general synthetic strategy is shown in FIG. 1.

D. Procedures for the Synthesis of Representative Tethers of the Present Invention

[0093] The important tether component required for compounds of the invention are synthesized as described in WO01/25257, U.S. Provisional Pat. Appl. Ser. No, 60/491, 248 or herein. A standard procedure for the synthesis of tether B is shown in FIG. 2.

Step T8-1

[0094] Chlorotrimethylsilane (116 mL, 0.91 mol, 1.5 eq) was added to a suspension of 2-hydroxycinnamic acid (100 g, 0.61 mol, 1.0 eq) in MeOH (500 mL, HPLC grade) over 30 min at 0° C. The resulting mixture was stirred at rt O/N. The reaction was monitored by TLC (EtOAc/MeOH:98/2). Heating the reaction mixture in a hot water can accelerate the process if necessary. After the reaction was completed, the reaction mixture was evaporated under reduced pressure to afford methyl 2-hydroxycinnamate as a white solid (108.5 g) in quantitative yield. The identity of this intermediate com-

pound is confirmed by NMR. This reaction can be carried out on larger (kg) scale with similar results

Step T8-2

[0095] 3,4-Dihydro-2H-pyran (DHP, 140 mL, 1.54 mol, 2.52 eq) was added dropwise to 2-bromoethanol (108 mL, 1.51 mol, 2.5 eq) in a 2 L three-neck flask with mechanical stirring at 0° C. over 2 h. The resulting mixture was stirred for additional 1 h at rt. Methyl 2-hydroxycinnamate from Step T8-1 (108 g, 0.61 mol, 1.0 eq), potassium carbonate (92.2 g, 0.67 mol, 1.1 eq), potassium iodide (20 g, 0.12 mol, 0.2 eq) and DMF (300 mL, spectrometric grade) were added to the above flask. The reaction mixture was stirred at 70° C. (external temperature) for 24 h. The reaction was monitored by TLC (DCM/Et₂O: 95/5). The reaction was allowed to cool to rt and Et₂O (450 mL) was added. The inorganic salts were removed by filtration and washed with Et_2O (3×50 mL). The filtrate was diluted with hexanes (400 mL) and washed with water (3×500 mL), dried over MgSO $_4$, filtered and the filtrate evaporated under reduced pressure. The crude ester (desired product and excess Br—C₂H₄—OTHP) was used for the subsequent reduction without further purification.

Step T8-3

[0096] DIBAL (1.525 L, 1.525 mol, 2.5 eq, 1.0 M in DCM) was added slowly to a solution of the above crude ester from Step T8-2 (0.61 mol based on the theoretical yield) in anhydrous DCM (610 mL) at -35° C. with mechanical stirring over 1.5 h. The resulting mixture was stirred for 1.5 h at -35 C., then 1.5 h at 0° C. The reaction was monitored by TLC (hex/EtOAc:50/50). When complete, Na₂SO₄.10 H₂O (100 g, 0.5 eq) was slowly added; hydrogen evolution was observed, when it subsided water was added (100 mL). The mixture was warmed to rt and stirred for 10 min, then warmed to 40° C. with hot water and stirred under reflux for 20 min. The mixture was cooled to rt, diluted with DCM (600 mL), and the upper solution decanted into a filter. The solid that remained in the flask was washed with dichloromethane (5×500 mL) with mechanical stirring and filtered. The filtrate from each wash was checked by TLC, and additional washes performed if necessary to recover additional product. In an alternative work-up procedure, after dilution with DCM (600 mL), the mixture was filtered. The resulting solid was then continuously extracted with 0.5% TEA in dichloromethane using a Soxhlet extractor. Higher yield was typically obtained by this alternative procedure, although it does require more time. The filtrate was concentrated under reduced pressure and the residue purified by dry pack (EtOAc/hex/Et₃N: 20/80/0.5) to give the product alcohol as a yellowish oil (yield: 90%). The identity and purity were confirmed by NMR.

Step T8-4

[0097] To a mixture of the allylic alcohol from Step T8-3 (28 g, 0.100 mol, 1.0 eq) and collidine (0.110 mol, 1.1 eq) in 200 mL of anhydrous DMF under $\rm N_2$ was added anhydrous LiCl (4.26 g, 0.100 mol, 1.0 eq.) dissolved in 100 mL of anhydrous DMF. The mixture was then cooled to 0° C., and MsCl (12.67 g, 0.110 mol, 1.1 eq., freshly distilled over $\rm P_2O_5$), was added dropwise. The reaction was allowed to warm to rt and monitored by TLC (3:7 EtOAc/hex). When the reaction was complete, NaN $_3$ (32.7 g, 0.500 mol, 5.0 eq.) was added. The reaction mixture was stirred at rt O/N with progress followed by NMR. When the reaction was complete,

the mixture is poured into an ice-cooled water bath, and extracted with diethyl ether $(3\times)$. The combined organic phases were then washed sequentially with citrate buffer $(2\times)$, saturated sodium bicarbonate $(2\times)$, and finally with brine $(1\times)$. The organic layer was dried with MgSO₄, filtered and the filtrate concentrated under reduced pressure. The allylic azide was obtained in 90% combined yield, and was of sufficient quality to use as such for the following step.

Step T8-5

[0098] PPh₃ (25.9 g, 0.099 mol, 1.5 eq) was added at 0° C. to a solution of the allylic azide from Step T8-4 (20.0 g, 0.066 mol, 1.0 eq.) in 100 mL of THF. The solution was stirred for 30 min at 0° C. and 20 h at rt. Water (12 mL) was then added and the resulting solution was heated at 60° C. for 4 h. The solution was cooled to rt, 2N HCl (15 mL) added and the mixture stirred for 90 min at 50° C. The separated organic phase was extracted with 0.05 N HCl (2×100 mL). The combined aqueous phase was washed with $Et_2O(5\times150 \text{ mL})$ and toluene (4×150 mL) (more extraction could be necessary, follow by TLC), which were combined and back-extracted with 0.05 N HCl (1×100 mL). This acidic aqueous phase from back-extraction was combined with the main aqueous phase and washed with ether (5×150 mL) again. The pH of the aqueous phase was then adjusted to 8-9 by the addition of sodium hydroxide (5 N). Care must be exercised to not adjust the pH above 9 due to the reaction conditions required by the next step. The aqueous phase was concentrated under reduced pressure (aspirator, then oil pump) or lyophilized to dryness. Toluene (2x) was added to the residue and then also evaporated under reduced pressure to remove traces of water. The crude product (desired amino alcohol along with inorgnic salt) was used for the next reaction without further purification.

Step T8-6

[0099] A mixture of the crude amino alcohol from Step T8-5 (0.5 mol based on the theoretical yield), Ddz-OPh (174 g, 0.55 mol, 1.1 eq) and Et₃N (70 mL, 0.5 mol, 1.0 eq) in DMF (180 mL) was stirred for 24 h at 50° C. Additional DMF is added if required to solubilize all materials. The reaction was monitored by TLC (hex/EtOAc:50/50, ninhydrin detection). After the reaction was complete, the reaction mixture was diluted with Et₂O (1.5 L) and water (300 mL). The separated aqueous phase was extracted with Et₂O (2×150 mL). The combined organic phase was washed with water (3×500 mL) and brine (1×500 mL), dried over MgSO₄, filtered and the filtrate concentrated under reduced pressure. The layers were monitored by TLC to ensure no product was lost into the aqueous layer. If so indicated, perform one or more additional extractions with Et₂O of the aqueous phase to recover this material. The crude product was purified by dry pack (recommended column conditions: EtOAc/hex/Et₃N: 35/65/0.5 to 65/35/0.5) to give the tether Ddz-T8 as a pale yellow syrup (yield: ~40%). The identity and purity of the product was confirmed by NMR.

[0100] ¹H NMR (DMSO-d₆): 1.6 ppm (s, 6H, 2×CH3), 3.6-3.8 ppm (wide s, 10H, 2×OCH₃, 2×OCH₂), 3.95 ppm (triplet, 2H, CH₂N), 6-6.2 ppm (m, 2H, 2×CH), 6.2-6.5 ppm (m, 3H, 3×CH, aromatic), 6.6-7.6 ppm (m, 5H, aromatic).

[0101] A standard procedure for the synthesis of tether T9 is shown in FIG. 3.

[0102] Tether T9 can also be synthesized from T8 by reduction as in step T9-3 or with other appropriate hydrogenation catalysts known to those in the art.

[0103] A standard procedure for the synthesis of Ddz propargylamine is shown in FIG. 4.

[0104] In a dried three-neck flask, a solution of propargylamine (53.7 g, 0.975 mol, 1.5 eq) in degassed DMF (Drisolv, 388 mL) was treated with Ddz-N₃ (170.9 g, 0.65 mol, 1.0 eq), tetramethylguanidine (TMG, 81.4 mL, 0.65 mol, 1.0 eq) and DIPEA (113.1 mL, 0.65 mol, 1.0 eq) and stirred at 50° C., O/N. The reaction was monitored by TLC (conditions:25/75 EtOAc/hex. R: 0.25; detection: UV, ninhydrin). Upon completion, DMF was evaporated under reduced pressure until dryness and the residue dissolved in Et₂O (1 L). The organic solution was washed sequentially with citrate buffer (pH 4.5, 3×), saturated aqueous sodium bicarbonate (2×), and brine (2x), then dried with MgSO₄, filtered and the filtrate evaporated under reduced pressure. A pale orange solid was obtained. This solid was triturated with 1% EtOAc in hex, then collected by filtration and dried under vacuum (oil pump) to provide the desired product (153.4 g, 85.2%).

[0105] A standard procedure for the synthesis of tether T10 is shown in FIG. 5A.

[0106] Two alternative routes to this tether have been developed. The first synthetic approach proceeded starting from the commercially available monobenzoate of resorcinol (T10-0). Mitsunobu reaction under standard conditions with the protected amino alcohol from Example 9, followed by saponification of the benzoate provided T10-1 in good yield after recrystallization. Alkylation of the phenol with 2-bromoethanol using the optimized conditions shown permitted the desired product Ddz-T10 to be obtained after dry pack purification in 42% yield.

[0107] A second synthetic route to T10 is shown in FIG. 5B.

[0108] From resorcinol, two successive Mitsunobu reactions are conducted with the appropriate two carbon synthons illustrated, themselves derived from 2-aminoethanol and ethylene glycol, respectively, through known protection methodologies. Lastly, deprotection of the silyl ether, also under standard conditions provided Boc-T10.

[0109] Although the yields in the two methods are comparable, the first required less mechanical manipulation and is preferred for larger scales.

[0110] A standard procedure for the synthesis of tether T11 is shown in FIG. 6.

[0111] A standard procedure for the synthesis of tether T12 is shown in FIG. 7.

[0112] In a 3-L flame-dried three-neck flask, a solution of (aminomethyl)phenylthiobenzyl alcohol (12-0, 96 g, 0.39 mol) in degassed DMF (1 L, 0.4 M) was prepared. To this was added DdzN₃ (0.95 eq), followed by TMG (0.39 mol, 49 mL). The reaction was stirred for 10 min, then DIPEA (68 mL, 0.39 mol) added. The mixture was heated at 50° C. under N₂ until TLC indicated no DdzN₃ remained (48 h typically). (TLC eluent: EtOAc:Hex 50:50; detection: ninhydrin). Upon completion, to the reaction mixture was added 3 L citrate buffer and the separated aqueous layer extracted with Et₂O (3×1500 mL). The combined organic phase was washed sequentially with citrate buffer (2×200 mL), water (2×200 mL) and brine (2×200 mL). The organic layer was dried over MgSO₄, filtered and the filtrate evaporated under reduced pressure. A dark orange oil was obtained, which was purified by dry-pack. For this procedure, the oil was first dissolved in

EtOAc:Hex:DCM:TEA (20:80:1:0.5, v/v/v/v). At this point, a little extra DCM was sometimes required to ensure complete dissolution. The solution was loaded onto the column, then the column eluted with EtOAc:Hex:DCM:Et₃N (20:80: 1:0.5) until all the impurities were separated out as indicated by TLC, paying particular attention to that closest to the desired product. The elution was then continued with EtOAc: Hex:Et₃N 30:70:0.5 (v/v/v) and finally with EtOAc:hexanes: Et₃N (50:50:0.5) to elute the desired product. After removal of the solvent from the fractions containing the product under reduced pressure, the residue was dissolved in the minimum amount of DCM, a three-fold larger volume of hexanes added, then the solvents again evaporated under reduced pressure. This treatment was repeated until an off-white foam was obtained. The latter solidified while drying under vacuum (oil pump). Alternatively, the material yielded a solid after sequential concentration with DCM $(1\times)$ and hexanes $(2\times)$. Tether Ddz-T12 was obtained as an off-white solid (85-90% yield).

Example 29

Standard Procedure for Attachment of Tethers Utilizing the Mitsunobu Reaction

Example 29-A

Using PPh₃-DIAD Isolated Adduct

[0113] To a 0.2 M solution of the appropriate tether (1.5 eq) in THF or THF-toluene (1:1) was added the PPh₃-DIAD (pre-formed by mixing equivalent amounts of the reagents and isolated by evaporation of solvent, see Example 29-C) adduct (1.0 eq.). The resultant mixture was manually agitated for 10 sec (the solution remained turbid), then added to the resin. Alternatively, the resin was added to the solution. The reaction suspension was agitated O/N (after ~5 min the mixture becomes limpid). The resin was filtered and washed 2×DCM, 1×toluene, 1×EtOH, 1×toluene, 1×(DCM/MeOH), 1×(THF/MeOH), 1×(DCM/MeOH), 1×(THF/MeOH), 2×DCM, then dried in the standard manner.

Example 29-B

Using "PPh3-DIAD In Situ Procedure"

[0114] To a 0.2 M solution of the appropriate tether (4 eq) in THF or THF-toluene (1:1) was added triphenylphosphine (4 eq). The resultant mixture was manually shaken until a homogenous solution was obtained, then added to the resin. Alternatively, the resin (or IRORITM MiniKans® (NEXUS Biosystems, Poway, Calif.), miniaturized microreactors, containing resin) was added to the solution. To this suspension was then added DIAD (3.9 eq) and the reaction agitated O/N. Note: Since the reaction is exothermic, for larger scales, the reaction should be cooled in an ice bath. In addition, an appropriate vent must be supplied to allow any pressure build-up to be released. The resin was filtered and washed DCM (2×), toluene (1×), EtOH (1×), toluene (1×), DCM/MeOH (1×), 1×THF/MeOH (1×), DCM/MeOH (1×), THF/MeOH (1×), 2×DCM, then dried in the standard manner.

[0115] A procedure for the synthesis of PPh₃-DIAD adduct is shown in FIG. **8**. DIAD (1 eq) was added dropwise to a well-stirred solution of triphenylphosphine (1 eq) in THF (0.4 M) at 0° C. under nitrogen. The mixture was then maintained at 0° C. with stirring for 30 min. The white solid obtained was collected by filtration (use medium sized fritted filters),

washed with cold anhydrous THF until the washes were colorless, and lastly washed once with anhydrous Et₂O. The white solid product was then vacuum-dried (oil pump) and stored under nitrogen. (Note: The PPh₃-DIAD adduct can be made in larger than immediately required quantity and stored under nitrogen; it is very important to store this reagent under anhydrous conditions.)

Example 30

Standard Procedure for Attachment of Tethers via Reductive Amination as shown in FIG. 9

[0116] In certain instances, the Mitsunobu process of Example 29 cannot be applied or is not efficient for incorporation of the tether. Hence, reductive amination has been developed as an alternative that can be employed for tether incorporation as illustrated hereinbelow for one of the preferred tethers. Similar chemistry can be used to incorporate other tethers of the present invention.

[0117] The Tether (30-2) with the amine protected as its Ddz derivative was efficiently oxidized to the corresponding aldehyde 30-2 using SO₃.pyr in DMSO-Et₃N-DCM. This aldehyde (0.14 mmol, 56 mg, 1.5 eq based upon loading of resin support) was dissolved in a 1:3 mixture of TMOF-MeOH (DriSolv, 4 mL) at rt. To this was added the resin containing the tripeptide (30-1, as its trifluoroacetic acid salt from the deprotection of the terminal amine), the mixture was agitated briefly to wet the resin, and then borane-pyridine complex (as the commercially available 8 M solution, 23 µL, 2 eq) was introduced to the suspension. The reaction was agitated O/N, then the resin filtered, washed with DCM $(2\times)$, THF (1x), DCM/MeOH [3:1] (1x), THF/MeOH [3:1] (1x), DCM (2x) and dried in the standard manner. Care must be taken to ensure that the desired resin bound product 30-3 is not contaminated with the dialkylated material. However, even if the reaction does not proceed to completion or if a small amount of the dialkylation side product is present, the material is of sufficient purity for the macrocyclization reac-

[0118] A standard procedure for the synthesis of tether T28 is shown in FIG. 10.

[0119] Henry reaction of 2-hydroxybenzaldehyde 28-0 provided 28-1 in 79% yield. This was followed by reduction first with sodium borohydride, then with catalytic hydrogenation, to give the amine, which was then protected as its Boc derivative, 28-2. Yields of these first two steps were lower on larger scales. Alkylation of 28-2 with the TBDMS ether of 2-bromoethanol, itself synthesized by standard methods, gave 28-3 in 74% yield. Deprotection of the silyl ether under standard conditions yielded the desired protected tether, Boc-T28. Alternative use of ethylene carbonate for the phenol alkylation to avoid the protection/deprotection steps, gave 73% yield.

[0120] A standard procedure for the synthesis of tether T32 is shown in FIG. 11.

[0121] A standard procedure for the synthesis of tether T33a and T33b is shown in FIGS. 12A and 12B.

[0122] The construction to the (R)-isomer of this tether (T33a) was accomplished from 2-iodophenol (33-0) and (S)-methyl lactate (33-A). Mitsunobu reaction of 33-0 and 33-A proceeded with inversion of configuration in excellent yield to give 33-1. Reduction of the ester to the corresponding alcohol (33-2) also occurred in high yield and was followed by Sonagashira reaction with Ddz-propargylamine. The

alkyne in the resulting coupling product, 33-3, was reduced with catalytic hydrogenation. Workup with scavenger resin provided the desired product, Ddz-T33a.

[0123] The synthesis of the (S)-enantiomer (Ddz-T33b) was carried out in an identical manner in comparable yield starting from (R)-methyl lactate (33-B). See FIG. 12B.

[0124] Standard procedures for the synthesis of various tethers are shown in the figures: tether T34 (FIG. 13), tether T35 (FIG. 14), tether T36 (FIG. 15), tether T37 (FIG. 16), tether T38 (FIG. 17), tether T39 (FIG. 18), tether T40 (FIG. 19), tether T41 (FIG. 20) and tether T42 (FIG. 21).

E. Examples of Synthetic Strategies for the Macrocyclic Compounds of the Invention

[0125] FIG. 22 presents a scheme depicting a thioester strategy for macrocyclic compounds of the present invention.
[0126] It should be noted that one or more of the amino acids indicated can be replaced by corresponding hydroxy acids and coupled to the next building block utilizing methods known to those in the art.

Example 47

Standard Procedure for Macrocyclization with Thioester Linker

[0127] The resin containing the cyclization precursor is combined in an appropriate vessel with pre-washed MP-carbonate resin [Argonaut Technologies, Foster City, Calif., commercially supplied MP-carbonate resin was treated with 3×THF (1 L per 400 g) and dried O/N at 30° C. in a vacuum oven] (1.4 to 1.6 eq relative to the initial loading of the synthesis resin). A 0.2 M DIPEA solution in THF was then added to the combined resins (1 mL/60 mg MP-carbonate resin) and the suspension agitated O/N at rt. Subsequently, the resin was filtered and rinsed 2×THF. The combined filtrates are collected together in an appropriate vessel, then the volatile contents evaporated under reduced pressure [in addition to the standard methods, solvent can also be removed in vacuo using centrifugal evaporation (ThermoSavant Discovery®, SpeedVac® or comparable) (Thermo Electron Corporation, Waltham, Mass.)] to provide the crude macrocycles.

Example 48

Standard Procedure for Silver-Assisted Macrocyclization with Thioester Linker

[0128] Except for the cyclization itself and subsequent work-up, this procedure is identical to that of Example 47.

The resin containing the cyclization precursor was combined in an appropriate vessel with pre-washed MP-carbonate resin [Argonaut Technologies, commercially supplied MP-carbonate resin was treated with THF (3x, 1 L per 400 g) and dried O/N at 30° C. in a vacuum oven] (1.4 to 1.6 eq relative to the initial loading of the synthesis resin). To this was added THF (1 mL per 100 mg resin) and silver trifluoroacetate (1 eq relative to the initial loading of the resin). Finally, an amount of DIPEA sufficient to obtain a 0.2 M solution was added. The reaction mixture was agitated at rt O/N. The solution was then filtered and the resins washed 2×THF. The filtrates are collected together in an appropriate vessel, then evaporated under reduced pressure [(the volatile contents could also be removed in vacuo using centrifugal evaporation (ThermoSavant Discovery®, SpeedVac® or comparable)] to provide the crude macrocycles. For this procedure, silver trifluoroacetate should be stored in a dessicator between uses. In addition, it is recommended to use a new bottle of THF (or a bottle that has been recently opened under N₂ or Ar) to minimize formation of silver oxide.

[0129] Additionally, a ring-closing metathesis (RCM) strategy, as developed by Grubbs et al. can also be used to access some of the macrocyclic compounds of the invention (see for example U.S. Pat. No. 5,811,515; Grubbs, R. H. et al. *J. Org. Chem.* 2001, 66, 5291-5300; Fürstner, A. Angew. Chem. Int. Ed. 2000, 39, 3012-3043).

[0130] To access certain derivatives of compounds of the present invention, additional reactions from those in the general scheme were required. For some, it was advantageous to react the functionality to be derivatized prior to the formation of the macrocyclic ring. The cyclic structure can restrict access of reagents to that functionality. For example, in the synthesis of N-methyl and N-acyl derivatives of macrocycles, where the secondary nitrogen atom of the ring is the site of derivatization, the reaction is preferred to be performed prior to the application of the appropriate cyclization protocol.

[0131] In other cases, for example the derivatization of side chain functionality, the reaction was best performed after formation of the macrocyclic ring. For example, further reaction of amino moieties on side chains examples was typically efficiently done by reaction of the partially protected macrocycle. In this manner, acylation, sulfonylation, alkylation (via reductive amination), guanidine and urea formation were performed via standard methods.

[0132] Table 1, hereinbelow, shows a representative, but by no means exclusive, summary of the chemical synthesis of several representative compounds of the invention.

TABLE 1

					Tether	Additional
	AA1	AA2	AA3	Tether	Attachment	Steps
1	Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T8	Example 29	none
	Tyr(tBu)					
2	Bts-D-Phe	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none
3	Bts-D-Phe	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
4	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
5	Bts-D- Tyr(tBu)	Boc-D-Ala	Boc-Nva	Ddz-T8	Example 29	none
6	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Met	Ddz-T8	Example 29	none

TABLE 1-continued

	S	Synthesis of Represer	ntative Compound	s of the Present l	nvention	
	AA1	AA2	AA3	Tether	Tether Attachment	Additional Steps
7	Bts-D-	Boc-D-Val	Boc-Nle	Ddz-T8	Example 29	none
8	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Phe	Ddz-T8	Example 29	none
9	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Val	Ddz-T8	Example 29	none
10	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Leu	Ddz-T9	Example29	none
11	Tyr(tBu) Bts-D-2-Nal	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none
12	Bts-D-	Boc-D-Val	Boc-Abu	Ddz-T8	Example 29	none
4.0	Tyr(tBu)	D D 1111	D 1	D	E 1.00	
13	Bts-D-Phe	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
14	Bts-D-2-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
15	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
16	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
17	Bts-D- Trp(Boc)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
18	Bts-D- Tyr(tBu)	Boc-D-2-Abu	Boc-Nva	Ddz-T9	Example 29	none
19	Bts-D- Phe(4F)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
20	Bts-D-Phe	Boc-D-Val	Boc-Leu	Boc-T8	Example 29	none
21	Bts-D-2-Nal	Boc-D-Val	Boc-Leu	Boc-T8	Example 29	none
22	Bts-D-2-Nai	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Tyr(OMe)				•	
23	Bts-D-1-Nal	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
24	Bts-D-2-Thi	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
25	Bts-D- Phe(2Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
26	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Cpa	Ddz-T9	Example 29	none
27	Bts-D-4-Thz	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
28	Bts-D-3-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
29	Bts-D- Tyr(tBu)	Boc-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
30	Bts-D- Tyr(tBu)	Hnva(THP)	Boc-Nva	Ddz-T9	Example 29	none
34	Bts-D- Tyr(tBu)	Ddz-D-Tyr(tBu)	Boc-Nva	Ddz-T8	Example 29	None
38	Bts-D-	Boc-D-Val	Boc-Ala	Ddz-T8	Example 29	none
39	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-□-Ala	Ddz-T8	Example 29	none
40	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Gly	Ddz-T8	Example 29	none
41	Tyr(tBu) Bts-D-	Boc-DPhe	Boc-Nva	Ddz-T8	Example 29	none
52	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Phg	Ddz-T8	Example 29	none
55	Tyr(tBu) Bts-D-	Ddz-D-Val	Ddz-Lys(Boc)	Ddz-T8	Example 29	none
56	Tyr(tBu) Bts-D-	Ddz-D-Val	Ddz-Orn(Boc)	Ddz-T8	Example 29	none
	Tyr(tBu)				•	
57	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T8	Example 29	none
58	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Tyr(tBu)	Ddz-T8	Example 29	none
59	Bts-D- Tyr(tBu)	DdzD-Val	Ddz-Trp(Boc)	Ddz-T8	Example 29	none
60	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Tyr(OMe)	Ddz-T8	Example 29	none
65	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T2	Example 29	none
71	Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T10	Example 29	none
72	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-2-Nal	Ddz-T8	Example 29	none
76	Tyr(tBu) Bts-D-	Boc-D-2-Nal	Boc-Nva	Ddz-T8	Example 29	none
	Tyr(tBu)					

TABLE 1-continued

		Synthesis of Repres	entative Compound	s of the Present 1	Invention	
	AA1	AA2	AA3	Tether	Tether Attachment	Additional Steps
77	Bts-D-	Boc-D-Nle	Boc-Nva	Ddz-T8	Example 29	none
80	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Ile	Ddz-T8	Example 29	none
85	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-D-Nva	Ddz-T8	Example 29	none
87	Bts-D-Bip	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
89	Bts-D-Hfe	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D-Dip	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
91	Bts-D- Tyr(tBu)	Boc-D-Nva	Boc-Nva	Ddz-T9	Example 29	none
92	Bts-D- Tyr(tBu)	Boc-D-Tle	Boc-Nva	Ddz-T9	Example 29	none
96	Bts-D- Tyr(tBu)	Boc-β-Ala	Boc-Nva	Ddz-T9	Example 29	none
97	Bts-D- Tyr(tBu)	Boc-D-Chg	Boc-Nva	Ddz-T9	Example 29	none
98	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T18	Example 29	none
99	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T15	Example 29	none
109	Bts-D- Tyr(tBu)	Boc-D-Val	Ddz-Dab(Boc)	Ddz-T9	Example 29	none
110	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T11	Example 29	none
111	Bts-D- Tyr(tBu)	Boc-D-Val	Hval(THP)	Ddz-T9	Example 29	none
112	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
120	Bts-D- Tyr(tBu)	Boc-D-Pro	Boc-Nva	Ddz-T8	Example 29	none
121	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ac-T8-NH2	Example 29	none
122	Boc-D-3-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 30	none
	Boc-D-2-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 30	none
124	Boc-D-4-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 30	none
125	Bts-D- Tyr(tBu)	Boc-D-Cpg	Boc-Nva	Boc-T9	Example 29	none
126	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-NMeLeu	Boc-T9	Example 29	none
127	Boc-D- His(Mts)	Boc-D-Val	Boc-Nva	Boc-T12	Example 30	none
128	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
129	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
130	Bts-D-2-Thi	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
131	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
132	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
133		Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	none
134	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Leu	Boc-T2	Example 29	none
135	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
136	Bts-D-1Nal	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
137	Bts-D-2-Thi	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
139	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
140	Bts-D-	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
141	Phe(4F) Bts-D-	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none
142	Tyr(OMe)	Pec D 17-1	Pac Cas	Dog TO	Example 20	none
142		Boc-D-Val	Boc-Cpa	Boc-T9	Example 29 Example 29	none
143	Bts-D-2-Thi	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none

TABLE 1-continued

	Sy	nthesis of Represen	tative Compounds	of the Present In	vention	
	AA1	AA2	AA3	Tether	Tether Attachment	Additional Steps
44	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
45	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none
46	Bts-D- Phe(4F)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
17	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
48	Bts-D-1-Nal	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
	Bts-D-	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
	Phe(3Cl)					
50	Bts-D- Phe(4Cl)	Boc-D-Val	Вос-Сра	Boc-T11	Example 29	none
51	Bts-D- Phe(4F)	Boc-D-Val	Вос-Сра	Boc-T11	Example 29	none
52	Bts-D- Tyr(OMe)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
53	Bts-D-1-Nal	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Bts-D-2-Thi	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Phe(3Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Phe(4Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Phe(4F) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T11	Example 29	none
59	Phe(3Cl) Bts-D-	Boc-D-Ile	Boc-Nva	Boc-T9	Example 29	none
50	Tyr(But) Bts-D-	Boc-D-allolle	Boc-Nva	Вос-Т9	Example 29	none
51	Tyr(But) Boc-D-	Boc-D-Val	Boc-Nva	Вос-Т9	Example 30	none
52	Phe(4CH2NHFmoc) Bts-D-	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
53	Phe(2Me) Bts-D-	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
54	Phe(3Me) Bts-D- Phe(4Me)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
55	Bts-D- Phe(3OMe)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
56	Bts-D- Phe(2OMe)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
57	Bts-D-3- benzothienyl	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
58	Bts-D-3-Thi	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D-□- HomoPhe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
70	Bts-D- Phe(3,4diCl)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
71	Bts-D- Phe(3,4diF)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
72	Bts-D- Phe(3,4diOMe)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
73	Bts-D-1Nal	Hnva(THP)	Boc-Nva	Boc-T9	Example 29	none
	Bts-D- Tyr(OMe)	Hnva(THP)	Boc-Nva	Boc-T9	Example 29	none
75	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Boc-T33b	Example 29	none
76	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Boc-T33a	Example 29	none
77	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Boc-T28	Example 29	none
78	Bts-D- Tyr(OMe)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none
79	Bts-D-1-Nal	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none
80	Bts-D-2-Thi	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none
81	Bts-D- Phe(3Cl)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none
32	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none

TABLE 1-continued

		Synthesis of Repres				
	AA1	AA2	AA3	Tether	Tether Attachment	Additional Steps
183	Bts-D- Phe(4F)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none
	Bts-D-1-Nal Bts-D-	Ddz-D-Val Ddz-D-Val	Ddz-Dap(Boc) Ddz-Dap(Boc)	Ddz-T11 Ddz-T11	Example 29 Example 29	none none
186	Phe(4Cl) Ddz-D-	Ddz-D-Val	Ddz-His(Mts)	Ddz-T9	Example 30	none
187	Tyr(tBu) Bts-D- Phe(3CF3)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
188	Bts-D- Phe(3F)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D- Phe(4NO2)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D-3- benzothienyl	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
	Bts-D- Phe(3OMe)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
	Bts-D- Phe(3,4diCl)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
	Bts-D- Phe(3,4diF)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Nva	Boc-T34	Example 29	none
	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Nva	Boc-T38	Example 29	none
	Bts-D- Phe(3Cl) Bts-D-	Boc-D-Val Boc-D-Val	Boc-Cpa Boc-Cpa	Ddz-T32(Boc) Boc-T34	Example 29 Example 29	none
	Phe(3Cl) Bts-D-	Boc-D-Val	Вос-Сра	Boc-T38	Example 29	none
	Phe(3Cl) Bts-D-	Boc-D-Val	Вос-Сра	Boc-T41	Example 29	none
00	Phe(3Cl) Bts-D-	Boc-D-Val	Вос-Сра	Boc-T8	Example 29	none
	Phe(3Cl) Bts-D-1-Nal Bts-D-	Boc-D-Val Boc-D-Val	Boc-Nva Boc-Nva	Boc-T8 Boc-T8	Example 29 Example 29	none none
03	Phe(3OMe) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	acetylation
04	Phe(4Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	guanidinylatic
05	Phe(4Cl) Bts-D-	Boc-D-Val	Boc-NMeLeu	Boc-T9	Example 29	none
06	Phe(3Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	mesylation
207	Phe(4Cl) Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	TMS- isocyanate followed by dilute acid
808	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	guanidinylatio
09	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	acetylation
10	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination wit
11	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	acetone reductive amination wit excess formaldehyde
212	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination wit acetone
213	Bts-D- Tyr(3,5dil)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
214	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Hse(Bzl)	Вос-Т9	Example 29	hydrogenolysi for protecting group remova

TABLE 1-continued

	AA1	AA2	AA3	Tether	Tether Attachment	Additional Steps
215	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination wit excess formaldehyde
216	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T40	Example 29	none
217	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T36	Example 29	none
218	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T39	Example 29	none
219	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T37	Example 29	none
220	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T39	Example 29	none
221	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T35	Example 29	none
222	Bts-D- Tyr(3tBu)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
223	Bts-D- Tyr(But)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	acetylation
224	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	reductive amination wit formaldehyde
225	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	acetylation
226	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	reductive amination wit aldehyde
227	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	reductive amination wit benzaldehyde

Notes

Any amino acid or tether designated as the Boc derivative could be substituted with the corresponding Ddz derivative.

D. Analytical Data for Selected Compounds of the Invention ¹H and ¹³C NMR spectra were recorded on a Varian Mercury 300 MHz spectrometer (Varian, Inc., Palo Alto, Calif.) and are referenced internally with respect to the residual proton signals of the solvent. Information about the conformation of the molecules in solution can be determined utilizing appropriate two-dimensional NMR techniques known to those skilled in the art. HPLC purifications were run on a Waters Xterra® MS C18 column, using the Waters FractionLynx® system (Waters Corporation, Milford, Mass.). Automated medium pressure chromatographic purifications were performed on an Isco CombiFlash® 16x system with disposable silica or C18 cartridges that permitted up to sixteen (16) samples to be run simultaneously (Teledyne Isco, Inc., Lincoln, Nebr.). MS spectra were recorded on a Waters Micro $mass \circledR \ Platform \ II \ or \ ZQ^{TM} \ system. \ HRMS \ spectra \ were$ recorded with a VG Micromass ZAB-ZF spectrometer. Chemical and biological information were stored and analyzed utilizing the ActivityBase® database software (ID Business Solutions Ltd., Guildford, Surrey, UK).

General Methods for Analytical HPLC Analyses

[0133] HPLC analyses are performed on a Waters Alliance® system 2695 running at 1 mL/min using an Xterra MS C18 column 4.6×50 mm (3.5 $\mu m)$. A Waters 996 PDA provided UV data for purity assessment (Waters Corporation, Milford, Mass.). An LCPackings (Dionex Corporation, Sunnyvale, Calif.) splitter (50:40:10) allowed the flow to be separated in three parts. The first part (50%) went to a Micro-

mass® Platform II MS equipped with an APCI probe for identity confirmation. The second part (40%) went to an evaporative light scattering detector (ELSD, Polymer Laboratories, now part of Varian, Inc., Palo Alto, Calif., PL-ELS-1000TM) for purity assessment and the last portion (10%) to a chemiluminescence nitrogen detector (CLND, Antek® Model 8060, Antek Instruments, Houston, Tex., part of Roper Industries, Inc., Duluth, Ga.) for quantitation and purity assessment. Data was captured and processed utilizing the most recent version of the Waters Millenium® software package (Milford, Mass.).

[0134] An example LC method suitable for compounds of the present invention uses MeOH as solvent A, $\rm H_2O$ as solvent B and 1% TFA/ $\rm H_2O$ as solvent D. Initial mobile-phase composition is 5% A, 85% B and 10% D. Details of the standard gradient method are shown below:

Time	Α%	В%	D %	Curve
0.00	5	85	10	6
1.00	5	85	10	6
6.00	50	40	10	6
9.00	50	40	10	6
14.00	90	0	10	6
17.00	90	0	10	6
17.50	5	85	10	6
20.00	5	85	10	6

[0135] Compounds 2-6, 8-10, 56, 65 and 144 are as defined in Table (3), hereinbelow.

Compound 2

[0136] Yield: 12 mg pure macrocycle was obtained (CLND quantification).

[0137] 1 H NMR (300 MHz, DMSO-d₆) δ 8.83 (m, 1H); 8.53 (m, 1H); 7.63 (m, 1H); 7.4-7.08 (m, 7H); 7.00-6.84 (m, 2H); 6.60 (d, 15 Hz, 1H); 6.41 (dt, 15 Hz, 5.4 Hz, 1H); 4.35 (m, 1H); 4.25-4.05 (m, 3H); 3.94 (dt, 1H, 6 Hz, 15 Hz); 3.79 (dd, 1H, 3.6 Hz, 8.4 Hz); 3.60 (m, 1H); 3.52-3.40 (bd, 1H); 3.22-3.06 (m, 4H); 1.88 (m, 2H); 1.54-1.28 (m, 2H); 1.25 (d, 3H, 4.8 Hz); 1.22 (d, 3H, 2.7 Hz); 0.92-0.80 (m, 6H).

[0138] HRMS calc. for $\rm C_{30}H_{40}N_4O_4$: 520.3049. found 520. 3057 ± 0.0016

[0139] HPLC [standard gradient method (refers to that presented in General Methods for Analytical HPLC Analyses)] t_R =9.55 min.

Compound 4

[0140] Yield: 12 mg pure macrocycle was obtained (CLND quantification).

[0141] 1 H NMR (300 MHz, DMSO-d₆) 89.35 (b, 1H); 8.98 (b, 1H); 5.52 (d, 1H, 8.4 Hz); 8.38 (b, 1H); 7.25 (b, 1H); 7.13-7.07 (m, 4H); 6.86 (t, 2H, 7.5 Hz); 6.57 (d, 2H, 8.7 Hz); 4.33 (b, 1H); 4.21-4.02 (m, 3H); 3.78 (dd, 1H, 3.3 Hz; 8.1 Hz); 3.65-3.54 (m, 1H); 3.31-3.23 (m, 1H); 3.13-3.02 (m, 4H); 2.78-2.2.28-2.18 (m, 1H); 2.0-1.80 (m, 2H); 1.50-1.30 (m, 3H); 1.25 (d, 3H, 4.5 Hz); 1.22 (d, 3H, 4.5 Hz); 1.01 (d, 3H, 6.6 Hz); 0.90 (d, 3H, 6.6 Hz); (t, 3H, 7.5 Hz).

 $\begin{array}{lll} \textbf{[0142]} & ^{13}\text{C NMR } (75.5 \text{ MHz}, \text{DMSO-d}_6) \, \delta \, 172.22; \, 171.37; \\ 157.77; \, 157.44; \, 156.04; \, 131.76; \, 130.80; \, 130.70; \, 127.88; \, 121. \\ 82; \, 115.83; \, 111.71; \, 62.13; \, 60.62; \, 54.21; \, 52.81; \, 47.13; \, 42.47; \\ 33.31; \, 29.69; \, 29.30; \, 28.61; \, 20.36; \, 19.44; \, 18.72; \, 17.60; \, 13.97. \\ \textbf{[0143]} & \text{HRMS calc. for } \, \text{C}_{30}\text{H}_{42}\text{N}_4\text{O}_6; \, \, 538.3155. \, \, \text{found:} \\ 538.3145 \pm 0.0016 \end{array}$

[0144] HPLC (standard gradient) t_R =8.12 min.

Compound 5

[0145] Yield: 17 mg pure macrocycle was obtained (CLND quantification).

[0146] 1 H NMR (300 MHz, DMSO-d₆) δ 9.02 (b, 1H); 8.47 (d, 1H, 8.4 Hz); 7.7 (b, 1H); 7.58 (d, 1H, 5.4 Hz); 7.28 (dd, 1H, 7.8 Hz, 0.8 Hz); 7.20 (t, 1H, 9.0 Hz, 0.8 Hz); 7.14 (d, 2H, 8.4 Hz); 6.98-6.91 (m, 3H); 6.66 (d, 8.7 Hz); 6.63 (d, 1H, 15.0 Hz); 6.43 (dt, 1H, 6.0 Hz, 15.0 Hz); 4.28-3.86 (m, 6H); 3.60-3.40 (m, 2H); 3.22-3.12 (m, 1H0; 3.05 (d, 2H, 5.4 Hz); 1.92-1.80 (m, 1H); 1.56-1.40 (m, 1H); 1.36-1.20 (m, 2H); 1.25 (d, 3H, 6.6 Hz); 0.84 (t, 3H, 7.2 Hz).

[0147] ¹³C NMR (75.5 MHz, DMSO-d₆) δ 172.54; 171.86; 158.97; 158.56; 127.39; 155.84; 131.62; 129.73; 129.20; 129.02; 128.43; 126.30; 124.51; 122.01; 115.85; 112.88; 61.23; 52.90; 51.23; 47.08; 42.66; 36.13; 33.30; 21.14; 19.57; 17.07; 14.14; 11.49.

[0148] HRMS calc. for $C_{28}H_{36}N_4O_6$: 508.2685. found: 508.2681 \pm 0.0015

[0149] HPLC (standard gradient) t_R =7.67 min.

Compound 6

[0150] Yield: 16 mg pure macrocycle was obtained (CLND quantification).

[0151] ¹H NMR (300 MHz, DMSO-d₆) δ 9.37 (b, 1H); 8.87 (b, 1H); 8.61 (d, 1H, 8.7 Hz); 7.62 (b, 1H); 7.27 (d, 1H, 7.8 Hz); 7.21 (t, 1H, 8.4 Hz); 7.14 (d, 2H, 8.4 Hz); 6.98-6.87 (m, 3H); 6.64 (d, 2H, 8.1 Hz); 6.70 (d, 1H, 15.6 Hz); 6.39 (dt, 1H,

6.3 Hz, 15.6 Hz); 4.44-4.36 (m, 1H); 4.34-4.08 (m, 2 Hz); 4.45-3.92 (dt, 1H, 6.9 Hz, 15.6 Hz); 3.74 (dd, 1H, 3.6 Hz, 8.4 Hz); 3.54-3.26 (m, 3H); 3.22-3.02 (m, 3H); 2.60-2.36 (m, 4H); 2.24-2.14 (m, 1H); 2.02 (s, 3H); 1.96-1.89 (m, 1H); 1.80-1.66 (m, 1H); 1.01 (d, 3H, 6.3 Hz); 0.90 (d, 3H, 6.6 Hz). [0152] $^{13}{\rm C}$ NMR (75.5 MHz, DMSO-d₆) δ 171.51; 171.26; 158.90; 158.49; 157.38; 155.86; 131.63; 129.82; 129.21; 128. 86; 128.63; 126.21; 121.98; 115.83; 112.83; 62.11; 61.06; 51.97; 47.10; 42.78; 30.91; 30.67; 29.34; 20.37; 19.39; 15.06. [0153] HRMS calc. for ${\rm C_{30}H_{40}N_4O_6S}$: 568.2719. found: 568.2711±0.0017

[0154] HPLC R, (general method) 7.92 min.

Compound 8

[0155] Yield: 27 mg pure macrocycle was obtained (CLND quantification).

[0156] ¹H NMR (300 MHz, DMSO-d₆) 8 9.05 (b, 1H); 8.43 (b, 1H); 8.34 (d, 1H, 9.3 Hz); 7.40 (b, 1H); 6.97 (d, 1H, 7.5 Hz); 6.92-6.74 (m, 9H); 6.67-6.54 (m, 2H); 6.33-6.25 (m, 3H); 6.10 (dt, 1H, 5.7 Hz, 16.2 Hz); 4.22 (dt, 1H, 0.9 Hz, 12 Hz); 3.94-6.66 (m, 4H); 3.30 (dd, 1H, 3.6 Hz, 7.8 Hz); 3.24 (m, 1H); 3.18 (m, 1H); 2.85-2.68 (m, 3H); 2.44-2.23 (m, 2H); 1.32 (o, 1H, 7.5 Hz); 0.97-0.89 (m, 1H); 0.42 (d, 3H, 6.6 Hz); 0.01 (d, 3H, 6.6 Hz).

[0157] 13 C NMR (75.5 MHz, DMSO-d₆) δ 171.20; 157.35; 155.88; 139.12; 131.61; 130.87; 129.74; 129.21; 128.77; 128. 88; 126.85; 126.19; 121.97; 115.82; 112.84; 62.04; 61.10; 55.07; 50.01; 47.09; 42.85; 37.42; 29.11.

[0158] HRMS calc. For $C_{34}H_{42}N_4O_6$: 586.3155. found: 586.3145 \pm 0.0017

[0159] HPLC R, (general method) 9.34 min.

Compound 9

[0160] Yield: 17 mg pure macrocycle was obtained (CLND quantification).

[0162] 1.96-1.88 (m, 1H); 1.25 (dd, 2H, 4.5 Hz; 6 Hz); 1.01 (d, 3H, 6.3 Hz); 0.91 (d, 3H, 6.6 Hz); 0.86 (d, 3H, 7.2 Hz); 0.81 (d, 3H, 6.6 Hz).

[0163] 13 C NMR (75.5 MHz, DMSO-d₆) δ 171.85; 171.17; 157.37; 155.87; 131.59; 129.88; 129.18; 128.97; 128.78; 128.51; 126.16; 121.97; 115.83; 112.85; 61.55; 61.18; 58.15; 54.22; 47.08; 42.89; 36.32; 29.35; 29.00; 20.34; 19.56; 18.73; 17.44.

[0164] HRMS calc. for $\rm C_{30}H_{40}N_4O_6$ 536.2998. found: 536. 2990±0.0017.

[0165] HPLC (standard gradient) t_R =8.15 min.

Compound 10

[0166] Yield: 24 mg pure macrocycle was obtained (CLND quantification).

[0167] 1 H NMR (300 MHz, DMSO-d₆) δ 9.33 (b, 1H); 8.82 (b, 1H); 8.56 (d, 1H, 8.3 Hz); 7.60 (b, 1H); 7.27 (d, 2H, 7.8 Hz); 7.20 (t, 1H, 7.8 Hz); 7.13 (d, 2H, 8.4 Hz); 6.95 (t, 2H, 7.8 Hz); 6.64 (d, 2H, 8.4 Hz); 6.57 (d, 1H, 15.4 Hz); 6.38 (dt, 1H, 15.4 Hz, 5.8 Hz); 4.26-4.10 (m, 3H); 3.96 (dt, 1H, 5.4 Hz, 8.4 Hz); 3.77 (dd, 1H, 3.7 Hz, 7.8 Hz); 3.51-3.24 (m, 3H); 3.18-3.02 (m, 3H); 1.90 (h, 1H, 6.4 Hz); 1.73-1.54 (m, 2H); 1.45

(dt, 1H, 6.7 Hz, 0.9 Hz); 0.99 (d, 3H, 6.6 Hz); 0.89 (d, 3H, 6.3 Hz); 0.87 (d, 3H, 6.0 Hz); 0.80 (d, 3H, 6.3 Hz).

[0168] ¹³C NMR (75.5 MHz, DMSO-d₆) 8 172.23; 171.17; 157.37; 155.88; 131.62; 129.82; 129.19; 128.95; 128.59; 126. 24; 121.99; 115.84; 112.88; 64.23; 61.98; 61.14; 51.43; 61.14; 51.43; 47.07; 42.81; 29.38; 24.85; 24.11; 21.00; 20.32; 19.30.

[0169] HRMS calc. for $O_{31}H_{42}N_4O_5$ 550.3155. found: 550. 3150 \pm 0.0016.

[0170] HPLC (standard gradient) t_R =8.91 min.

Compound 56

[0171] Yield: 16 mg pure macrocycle was obtained (CLND quantification).

 $\begin{array}{ll} \textbf{[0172]} & ^{1}\text{H NMR (300 MHz, DMSO-d}_{6}) \, \&\, 9.39 \, (b, 1\text{H}); \, 8.90 \\ (b, 1\text{H}); \, 8.67 \, (d, 1\text{H}, 8.4 \, \text{Hz}); \, 7.74 \, (b, 4\text{H}); \, 7.29\text{-}7.08 \, (m, 4\text{H}); \\ 6.99\text{-}6.87 \, (m, 2\text{H}); \, 6.64 \, (d, 2\text{H}, 8.1 \, \text{Hz}); \, 6.61 \, (d, 1\text{H}, 16.5 \, \text{Hz}); \\ 6.40 \, (dt, 1\text{H}, 5.7 \, \text{Hz}, 16.5 \, \text{Hz}); \, 4.40\text{-}4.06 \, (m, 4\text{H}); \, 4.02\text{-}3.95 \\ (m, 1\text{H}); \, 3.79 \, (dd, 1\text{H}, 3.6 \, \text{Hz}, 7.8 \, \text{Hz}); \, 3.55\text{-}3.30 \, (m, 2\text{H}); \\ 3.16\text{-}3.05 \, (m, 3\text{H}); \, 2.82\text{-}2.69 \, (m, 2\text{H}); \, 2.02\text{-}1.85 \, (m, 2\text{H}); \\ 1.64\text{-}1.43 \, (m, 3\text{H}); \, 1.29\text{-}1.23 \, (m, 1\text{H}); \, 1.01 \, (d, 3\text{H}, 6.3 \, \text{Hz}); \\ 0.91 \, (d, 3\text{H}, 6.3 \, \text{Hz}); \, 0.86\text{-}0.84 \, (m, 2\text{H}). \end{array}$

[0173] HPLC (standard gradient) $t_p=5.71$ min.

Compound 65

[0174] Yield: 17 mg pure macrocycle was obtained (CLND quantification).

[0175] 1 H NMR (300 MHz, DMSO-d₆) δ 9.60 (b, 1H); 9.39 (b, 1H); 8.88 (b, 1H); 8.70 (d, 1H, 7.5 Hz); 8.57 (d, 1H, 4.2 Hz); 7.27 (t, 6 Hz); 6.96 (d, 2H, 8.4 Hz); 6.66 (d, 2H, 8.4 Hz); 5.78-5.68 (m, 1H); 5.42-5.33 (m, 1H); 3.96-3.89 (m, 1H); 3.80-3.57 (m, 5H); 3.41-3.34 (m, 1H); 3.10-2.90 (m, 1H); 2.78-2.66 (m, 1H); 2.21-2.10 (m, 1H); 2.06-1.93 (m, 1H); 1.70-1.60 (m, 1H); 1.52-1.41 (m, 1H); 1.39-1.26 (m, 1H); 1.25 (d, 3H, 4.8 Hz); 1.23 (d, 3H, 4.5 Hz); 0.83 (dd, 3H, 3 Hz, 4.5 Hz).

[0176] 13 C NMR (75.5 MHz, DMSO-d₆) δ 172.68; 172.63; 159.15; 158.73; 157.38; 157.25; 130.89; 124.99; 116.03; 62.51; 62.12; 54.29; 49.27; 42.47; 32.77; 30.43; 28.85; 20.46; 19.59; 18.72; 17.39; 13.90; 13.09.

[0177] HRMS calc. for $C_{24}H_{36}N_4O_4$: 444.2736. found: 444.2726 \pm 0.0013

[0178] HPLC (standard gradient) t_R =6.80 min.

Compound 144

[0179] 1 H NMR (300 MHz, CD₃OD) δ 7.4 (m, 1H); 7.27 (dt, 1H, 1.5 Hz, 6.6 Hz); 7.22-7.14 (m, 2H); 7.08-6.98 (m, 2H); 6.78 9t, 2H, 6.6 Hz); 4.45-4.39 (m, 2H); 4.15 (d, 2H, 8.1 Hz); 7.74 (d, 1H, 9.3 Hz); 3.54 (d, 1H, 10.8 Hz); 3.35-3.22 (m, 2H); 3.20 (q, 1H, 1.5 Hz); 2.82-2.71 (m, 1H); 2.61-2.55 (m, 1H); 2.21-2.11 (m, 1h); 2.02-1.94 (m, 1H); 1.74-1.40 (m, 5H); 1.04 (d, 3H, 6.6 Hz); 0.93 (d, 3H, 6.6 Hz); 0.74-0.64 9m, 1H); 0.45-0.28 (m, 2H); 0.15-0.08 (m, 1H); 0.06-0.02 (m, 1H).

[0180] ¹³C NMR (75.5 MHz, CD₃OD) & 173.29; 172.14; 167.51; 155.47; 134.86; 134.81; 130.38; 130.31; 128.81; 128. 25; 127.44; 121.63; 110.39; 107.71; 105.02; 67.10; 66.66; 62.81; 62.06; 60.10; 53.99; 41.44; 36.07; 31.91; 30.01; 29.18; 28.94; 27.79; 23.68; 23.15; 19.08; 18.25; 8.17; 4.98; 3.16.

[0181] HRMS: calc. for $C_{31}H_{41}N_4O_4C1568.2816$. found 568.2802 ± 0.0017

F. Mass Spectral Data for Selected Compounds of the Invention

[0182]

TABLE 2

Analysis of selec	cted compounds o	f the invention	
	Molecular Weight	Monoisotopic	M + H
Molecular Formula	(calculated)	Mass	Found
1 C30H40N4O5	536.7	536	537
2 C30H40N4O4	520.7	520	521
3 C30H42N4O4	522.7	522	523
4 C30H42N4O5	538.7	538	539
5 C28H36N4O5	508.6	508	509
6 C30H40N4O5S	568.7	568	569
7 C31H42N4O5	550.7	550	551
8 C34H42N4O5 9 C30H40N4O5	586.7 536.7	586 536	587 537
10 C31H42N4O5	550.7	550	551
11 C34H44N4O4	572.7	572	573
12 C29H38N4O5	522.6	522	523
13 C31H44N4O4	536.7	536	537
14 C35H46N4O4	586.8	586	587
15 C30H41N4O4Cl	557.1	556	557
16 C30H41N4O4Cl	557.1	556	557
17 C32H43N5O4	561.7	561	562
18 C29H40N4O5	524.7	524	525
19 C30H41N4O4F 20 C31H42N4O4	540.7 534.7	540 534	541 535
21 C35H44N4O4	584.7 584.7	584	585
22 C31H44N4O5	552.7	552	553
23 C34H44N4O4	572.7	572	573
24 C28H40N4O4S	528.7	528	529
25 C30H41N4O4Cl	557.1	556	557
26 C31H42N4O5	550.7	550	551
27 C27H39N5O4S	529.7	529	530
28 C29H41N5O4	523.7	523	524
29 C28H39N5O5	525.6	525	526
30 C30H41N3O6 34 C34H40N4O6	539.7 600.7	539 600	540 601
38 C28H36N4O5	508.6	508	509
39 C28H36N4O5	508.6	508	509
40 C27H34N4O5	494.6	494	495
41 C34H40N4O5	584.7	584	585
52 C33H38N4O5	570.7	570	571
55 C31H43N5O5	565.7	565	566
56 C30H41N5O5	551.7	551	552
57 C28H36N4O6	524.6	524	525
58 C34H40N4O6	600.7	600	601
59 C36H41N5O5 60 C35H42N4O6	623.7 614.7	623 614	624 615
65 C24H36N4O4	444.6	444	445
71 C29H40N4O6	540.7	540	541
72 C38H42N4O5	634.8	634	635
76 C38H42N4O5	634.8	634	635
77 C31H42N4O5	550.7	550	551
80 C31H42N4O5	550.7	550	551
85 C30H40N4O5	536.7	536	537
87 C36H46N4O4	598.8	598	599
88 C34H50N4O5	594.8	594 536	595 527
89 C31H44N4O4 90 C36H46N4O4	536.7 598.8	536 598	537 599
91 C30H42N4O5	538.7	538	539
92 C31H44N4O5	552.7	552	553
96 C28H38N4O5	510.6	510	511
97 C33H46N4O5	578.7	578	579
98 C24H39N5O4	461.6	461	462
99 C24H39N5O4	461.6	461	462
109 C29H41N5O5	539.7	539	540
110 C29H41N5O5	539.7	539	540
111 C30H41N3O6	539.7	539	540

TABLE 2-continued					TABLE 2-continued					
Analysis	s of selected cor	npounds c	f the invention		Analysis of selected compounds of the invention					
Molecular Fort	W	olecular Veight culated)	Monoisotopic Mass	M + H Found		Molecular Formula	Molecular Weight (calculated)	Monoisotopic Mass	M + H Found	
112 C31H44N4O5	5	552.7	552	553	189	C30H41N5O6	567.7	567	568	
120 C30H38N4O5	5	34.6	534	535	190	C33H42N4O4S	590.8	590	591	
121 C32H45N5O6	5	95.7	595	596	191	C32H44N4O5	564.7	564	565	
122 C31H43N4O4		571.2	570	571		C31H40N4O4Cl2	603.6	602	603	
123 C29H41N5O4		523.7	523	524		C31H40N4O4F2	570.7	570	571	
124 C29H41N5O4		523.7	523	524		C32H48N6O6	612.8	612	613	
125 C30H40N4O5		36.7	536	537		C32H46N4O5	566.7	566	567	
126 C32H46N4O5 127 C30H38N6O3		566.7 562.7	566 562	567 563		C32H43N6O4Cl C32H45N6O5Cl	611.2 629.2	610 628	611 629	
128 C32H46N4O5		66.7	566	567		C32H43N4O4Cl	583.2	582	583	
129 C35H46N4O4		86.8	586	587		C27H39N4O6Cl	551.1	550	551	
130 C29H42N4O4		42.7	542	543		C31H39N4O4Cl	567.1	566	567	
131 C31H43N4O40		571.2	570	571		C34H42N4O4	570.7	570	571	
132 C31H43N4O40	CI 5	71.2	570	571	202	C31H42N4O5	550.7	550	551	
133 C31H43N4O4	F 5	554.7	554	555	203	C30H40N5O5Cl	586.1	585	586	
134 C25H37N4O3	Cl 4	177.0	476	477	204	C29H40N7O4Cl	586.1	585	586	
135 C31H45N5O5		67.7	567	568		C32H45N4O4Cl	585.2	584	585	
136 C34H45N5O4		87.8	587	588	206		622.2	621	622	
137 C28H41N5O4		543.7	543	544	207		587.1	586	587	
138 C30H42N5O40		572.1	571	572		C29H41N7O5	567.7	567	568	
139 C30H42N5O40 140 C30H42N5O40		572.1 555.7	571	572 556	209 210	C30H41N5O6 C31H45N5O5	567.7 567.7	567 567	568	
141 C32H44N4O5		564.7	555 564	565		C30H42N5O4Cl	572.1	571	568 572	
141 C32H44N4O4		584.7	584	585		C31H44N5O4Cl	586.2	585	586	
143 C29H40N4O4		40.7	540	541	213	C30H40N4O512	790.5	790	791	
144 C31H41N4O4		69.1	568	569		C30H42N4O6	554.7	554	555	
145 C31H41N4O40		69.1	568	569		C30H43N5O5	553.7	553	554	
146 C31H41N4O4	F 5	552.7	552	553	216	C32H43N4O4Cl	583.2	582	583	
147 C31H43N5O5	5	65.7	565	566	217	C31H40N4O4FCl	587.1	586	587	
148 C34H43N5O4		85.7	585	586		C31H43N4O4Cl	571.2	570	571	
149 C30H40N5O4		70.1	569	570	219	C30H40N4O4Cl2	591.6	590	591	
150 C30H40N5O4		570.1	569	570		C31H43N4O4F	554.7	554	555	
151 C30H40N5O4		553.7	553	554 540	221	C30H40N4O4FCl	575.1	574 594	575 505	
152 C29H41N5O5 153 C32H41N5O4		539.7 559.7	539 559	540 560		C34H50N4O5 C32H44N4O6	594.8 580.7	580	595 581	
154 C26H37N5O4		515.7	515	516		C36H48N4O4	600.8	600	601	
155 C28H38N5O4		544.1	543	544	225	C37H48N4O5	628.8	628	629	
156 C28H38N5O4		544.1	543	544		C39H49N5O4S	683.9	683	684	
157 C28H38N5O4		527.6	527	528		C42H52N4O4	676.9	676	677	
158 C27H37N6O4	CI 5	45.1	544	545						
159 C31H44N4O5	5	552.7	552	553	Notes					
160 C31H44N4O5		552.7	552	553	1. Mole	ecular formulas and molecul re via ActivityBase ® softwa	ar weights (MW) are	calculated automatic	ally from the	
161 C31H45N5O4		551.7	551	552		ware program Molecular W			w only, nom	
162 C31H44N4O4		36.7	536	537		H obtained from LC-MS an			oed	
163 C31H44N4O4 164 C31H44N4O4		536.7 536.7	536 536	537 537	3. All a	nalyses conducted on mater	ial after preparative I	IPLC purification		
165 C31H44N4O5		552.7	552	553						
166 C31H44N4O5		552.7	552	553					~	
167 C32H42N4O4		78.8	578	579		BIOLOGICAL	METHODS A	AND RESULT	S	
168 C28H40N4O4		528.7	528	529						
169 C31H43N4O40		71.2	570	571	[018	The compound	ds of the presen	nt invention we	re evalu-	
170 C30H40N4O4	C12 5	591.6	590	591		for their ability to i				
171 C30H40N4O4		558.7	558	559	utiliz	zing a competitive r	adioligand bin	ding assay as d	escribed	
172 C32H46N4O6		82.7	582	583		ethod B1. Further o				
173 C34H43N3O5		573.7	573	574		erformed utilizing				
174 C31H43N3O6		553.7	553 553	554		nods B2, B3 and				
175 C31H44N4O5 176 C31H44N4O5		552.7	552 552	553 553		ucted, if so desire				
176 C31H44N4O3 177 C29H40N4O5		552.7 524.7	552 524	525		it the simultaneous				
177 C29H40N4O6		540.7	540	541						
179 C32H40N4O5		60.7	560	561		r assays have also				
100 62(112(1)1465)	~ ~	1.67	51.6	517	ш18	, such as that bas	eu upon the s	siabie express	ion oi a	

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529

545

577

541

180 C26H36N4O5S

181 C28H37N4O5Cl

182 C28H37N4O5Cl

183 C28H37N4O5F

185 C27H37N6O4Cl

187 C31H41N4O4F3

188 C30H41N4O4F

184 C31H40N6O4

186 C31H40N6O5

were evaluilin receptor as described eraction can lescribed in ods can be manner to compounds. suitable for HTS, such as that based upon the stable expression of a synthetic gene for the human motilin receptor.

[0184] Results for the examination of representative compounds of the present invention using Method B1 are presented in Table 3. The binding activity is listed as ranges with the following levels: $A=0.001-0.10 \mu M$; $B=0.10-1.0 \mu M$; C=1.0-10.0 µM. In addition, the assay results of two additional compounds using this Method are shown below. As can be observed, this demonstrates the activity of a representative

bicyclic compound of Formula IV of the invention, which resulted from incorporation of D-proline as the second recognition building block. Significantly, the lack of binding activity obtained with compound 121, which is the linear analogue of compound 1 (K_i=level B), illustrates the critical importance of the cyclic structure to attaining the desired interaction.

HO
$$K_i$$
 = level B

Compound 121

$$HN$$
 HN
 NH_2
 NH
 $K_t > 10 \text{ mM}$

[0185] Competitive binding curves for two representative compounds of the invention (Compounds 8 and 11) are presented in FIG. 23 and FIG. 24, respectively.

[0186] For determination of functional significance of the binding, the compounds are preferably tested in the Aequorin assay as described in Method B2, although the procedure of Method B3 is also applicable. As can be seen from the data presented in Table 4, the representative compounds examined act as antagonists at the motilin receptor and are devoid of agonist activity at the concentrations studied. The functional activity is listed as ranges with the following levels: A=0.001- $0.10 \,\mu\text{M}$; B=0.10-1.0 μM . The higher sensitivity of the assay of Method B2, almost 100 times that of Method C, makes it the preferred one for this assessment. This is evident in the EC_{50} values obtained in each for the positive agonist standard, motilin. Additionally, Method B2 measures the actual signaling event, which makes it more relevant to the effect that is desired, whereas the assay of Method B3 simply measures GTP turnover.

TABLE 4

Demendaden errina	Aequorin (Method B2) ¹			
Compound	Binding (K_i)	IC ₅₀		
142	A	В		
149	A	В		
167	A	A		
168	A	A		
212	A	A		
Motilin (human, porcine) ²	0.6	not applicable		

 $^{1}Activity$ is listed as ranges with the following levels: A = 0.001-0.10 $\mu M;$ B = 0.10-1.0 μM $^{2}Human$ and poreine motilin are the same peptide.

[0187] In addition, a common and scientifically-accepted ex vivo assay for the measurement of agonist or antagonist activity at the motilin receptor is the contraction of rabbit duodenum or other gastrointestinal smooth muscle tissue. A2-A4 Agonists are defined as compounds that induce >50% contraction relative to the motilin peptide, whereas antagonists are defined as compounds that cause >50% inhibition of the response to motilin. Compounds of the present invention have shown significant antagonist activity in this assay. For example, compound 144 exhibited a pA₂=6.95, while compound 165 had a pA₂=7.17, as calculated from the Schild plots of the response obtained at various concentrations as described in Method B4.

[0188] Gastric motility is generally measured in the clinical setting as the time required for gastric emptying and subsequent transit time through the GI tract. Gastric emptying scans are well known to those skilled in the art an, briefly, comprise use of an oral contrast agent, such as barium, or a radiolabeled meal. Solid and liquids can be measured independently. A test food or liquid is radiolabeled with an isotope (99mTc) and after ingestion or administration, transit time through the GI tract and gastric emptying are measured by visualization using gamma cameras. These studies are performed before and after the administration of the therapeutic agent to quantify the efficacy of the compound.

Example Method B1

Competitive Radioligand Binding Assay (Motilin Receptor)

Materials:

[0189] Membranes were prepared from CHO cells stably transfected with the human motilin receptor and utilized at a quantity of 1.5 μg/assay point. [PerkinElmerTM SignalScreen® Product #6110544, PerkinElmer, Inc., Wellesley, Mass.]

[0190] [125I]-Motilin (PerkinElmer, #NEX-378); final concentration: 0.04-0.06 nM

[0191] Motilin (BachemTM, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.); final concentration: 1 uM

[0192] Multiscreen® Harvest plates-GF/B (Millipore™, #MAHFB1H60, Billerica, Mass.)

[0193] Deep-well polypropylene titer plate (Beckman CoulterTM, #267006, Fullerton, Calif.)

[0194] TopSeal-ATM (PerkinElmer, #6005185, Wellesley, Mass.)

[0195] Bottom seal (Millipore™, #MATAH0P00, Billerica, Mass.)

[0196] MicroScint-0TM (PerkinElmer, #6013611, Wellesley, Mass.)

[0197] Binding Buffer: 50 mM Tris-HCl (pH 7.4), 10 mM MgCl₂, 1 mM EDTA, 0.1% BSA

Assay Volumes:

[0198] 150 μL of membranes diluted in binding buffer

[0199]

 $10\,\mu L$ of compound diluted in binding buffer $10\,\mu L$ of radioligand ([^125]]-Motilin) diluted in [0200]binding buffer

[0201] Final Test Concentrations (N=11) for Compounds: **[0202]** 10, 5, 2, 1, 0.5, 0.2, 0.1, 0.05, 0.02, 0.01, 0.005 μM.

Compound Handling:

[0203] Compounds were provided frozen on dry ice at a stock concentration of 10 mM diluted in 100% DMSO and stored at -20° C. until the day of testing. On the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.5%.

Assay Protocol:

[0204] In deep-well plates, diluted cell membranes (1.5 μg/mL) are combined with 10 μL of either binding buffer (total binding, N=5), 1 μM motilin (non-specific binding, N=3) or the appropriate concentration of test compound. The reaction is initiated by addition of 10 µl of [125] motilin (final conc. 0.04-0.06 nM) to each well. Plates are sealed with TopSeal-A, vortexed gently and incubated at room temperature for 2 hours. The reaction is arrested by filtering samples through pre-soaked (0.3% polyethyleneimine, 2 h) Multiscreen Harvest plates using a Tomtec® Harvester (Tomtec, Hamden, Conn.)), washed 9 times with 500 μL of cold 50 mM Tris-HCl (pH 7.4), and than plates are air-dried in a fumehood for 30 minutes. A bottom seal is applied to the plates prior to the addition of 25 µL of MicroScint-OTM to each well. Plates are then sealed with TopSeal-A® and counted for 30 sec per well on a TopCount® Microplate Scintillation and Luminescence Counter (PerkinElmer, Wellesley, Mass.) where results are expressed as counts per minute (cpm).

[0205] Data are analyzed by GraphPadTM Prism (GraphPad Software, San Diego, Calif.) using a variable slope non-linear regression analysis. K_i values were calculated using a K_d value of 0.16 nM for [125 I]-motilin (previously determined during membrane characterization).

test concentration with maximal displacement-

$$D_{max} = 1 - \frac{\text{non-specific binding}}{\text{total binding-non-specific binding}} \times 100$$

where total and non-specific binding represent the cpm obtained in the absence or presence of 1 µM motilin, respectively.

Example Method B2

Aequorin Functional Assay (Motilin Receptor)

Materials:

[0206] Membranes were prepared using AequoScreenTM (EUROSCREEN, Belgium) cell lines expressing the human motilin receptor (cell line ES-380-A; receptor accession #AF034632). This cell line is constructed by

transfection of the human motilin receptor into CHO-K1 cells co-expressing $G_{\alpha 16}$ and the mitochondrially targeted Aequorin (Ref #ES-WT-A5).

[0207] Motilin (BachemTM, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.)

[0208] Assay buffer: DMEM-F12 (Dulbeccoe's Modified Eagles Medium) with 15 mM HEPES and 0.1% BSA (pH 7.0)

[0209] Coelenterazine (Molecular ProbesTM, Leiden, The Netherlands)

[0210] Final Test Concentrations (N=5) for Compounds: [0211] 10, 3.16, 1, 0.316, 0.1 μ M.

Compound Handling:

[0212] Compounds were provided as dry films at a quantity of approximately 1.2 µmol in pre-formatted 96-well plates. Compounds were dissolved in 100% DMSO at a concentration of 10 mM and stored at -20° C. until further use. Daughter plates were prepared at a concentration of 500 µM in 30% DMSO with 0.1% BSA and stored at -20° C. until testing. On the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.6%.

[0213] Cell Preparation:

[0214] Cells are collected from culture plates with Ca²⁺ and Mg²⁺-free phosphate buffered saline (PBS) supplemented with 5 mM EDTA, pelleted for 2 minutes at 1000xg, resuspended in assay buffer (see above) at a density of 5×10⁶ cells/mL and incubated overnight in the presence of 5 µM coelenterazine. After loading, cells were diluted with assay buffer to a concentration of 5×10^5 cells/mL.

Assay Protocol:

[0215] For agonist testing, 50 µl of the cell suspension was mixed with 50 µl of the appropriate concentration of test compound or motilin (reference agonist) in 96-well plates (duplicate samples). The emission of light resulting from receptor activation was recorded using the Functional Drug Screening System 6000 'FDSS 6000' (Hamamatsu Photonics K.K., Japan).

[0216] For antagonist testing, an approximate EC80 concentration of motilin (i.e. 0.5 nM; 100 µL) was injected onto the cell suspension containing the test compounds (duplicate samples) 15-30 minutes after the end of agonist testing and the consequent emission of light resulting from receptor activation was measured as described in the paragraph above.

[0217] Results are expressed as Relative Light Units (RLU). Concentration response curves were analyzed using GraphPad™ Prism® (GraphPad Software, San Diego, Calif.) by non-linear regression analysis (sigmoidal dose-response) based on the equation $E=E_{max}/(1+EC_{50}/C)n$ where E is the measured RLU value at a given agonist concentration (C), E_{max} is the maximal response, EC₅₀ is the concentration producing 50% stimulation and n is the slope index. For agonist testing, results for each concentration of test compound were expressed as percent activation relative to the signal induced by motilin at a concentration equal to the EC_{80} (i.e. 0.5 nM). For antagonist testing, results for each concentration of test compound were expressed as percent inhibition relative to the signal induced by motilin at a concentration equal to the EC₈₀ (i.e. 0.5 nM).

Example Method B3

FlashPlate® Motilin [35S]-GTPyS Functional Assay Materials:

[0218] Membranes were prepared from CHO cells stably transfected with the human motilin receptor and utilized at a quantity of 1.5 µg/assay point. [PerkinElmerTM SignalScreen® Product #6110544, PerkinElmer, Inc. Wellesley, Mass.]

[0219] GTPγS Guanosine 5'-[γ-thio]triphosphate tetralithium salt (Sigma, #G-8634, Sigma-Aldrich, St. Louis, Mo.)

[**0220**] [³⁵S]-GTPγS (PerkinElmer, #NEX-030H)

[0221] Motilin (BachemTM, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.)

[0222] 96-well FlashPlate® white polystyrene microplates (PerkinElmer, #SMP200, Wellesley, Mass.)

[0223] Deep-well polypropylene titer plate (Beckman CoulterTM, #267006, Fullerton, Calif.)

[0224] TopSeal-ATM (PerkinElmer, #6005185, Wellesley, Mass.)

[0225] Assay Buffer: 50 mM Tris (pH 7.4), 100 mM NaCl, 10 mM MgCl $_2$, 1 mM EDTA, 1 μ M GDP, 0.1% BSA

Assay Volumes:

[0226] 25 μL of compound diluted in assay buffer

[0227] 25 μL of assay buffer (agonist assay) or 0.6 μM motilin (0.1 μM final concentration) diluted in assay buffer (antagonist assay)

[0228] $100 \,\mu\text{L}$ of [^{35}S]-GTP γ S diluted in assay buffer [0229] Final Test Concentrations (N=12) for Compounds: [0230] $50, 20, 10, 5, 2, 1, 0.5, 0.2, 0.1, 0.05, 0.02, 0.01 <math>\mu$ M.

Compound Handling:

[0231] Compounds were provided frozen on dry ice at a stock concentration of 10 mM diluted in 100% DMSO and stored at -20° C. until the day of testing. On the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.5%.

Assay Protocol:

[0232] CHO membranes were immobilized into 96-well FlashPlate® microplates. Test compound, GTPγS, motilin and [35S]-GTPγS were combined in each well according to the Assay Volumes described above.

[0233] For the assay to measure agonist activity, an additional 25 μ L of buffer was added to each well in addition to 25 μ L of either buffer (basal value, N=4), 1 μ M (final conc.) motilin (E_{max} value, N=3), 25 μ M (final conc.) GTP γ S (nonspecific value, N=4), or the appropriate concentration of test compound (N=3).

[0234] For the assay to measure antagonist activity, an additional 25 μ L of either buffer (unstimulated control) or motilin (0.1 μ M final conc.) is added to each well, in addition to either 25 μ L of buffer (basal value, N=3), 1 μ M (final conc.) motilin (E_{max} value, N=3), 25 μ M (final conc.) GTP γ S (nonspecific value, N=4), or the appropriate concentration of test compound (N=3).

[0235] The reaction is initiated by addition of 100 mL of $[^{35}S]$ -GTP γS to each well. Each plate is sealed (TopSeal-ATM) and incubated in the dark at room temperature for 150 min. Then, plates are counted for 30 seconds per well on the TopCount® NXT.

[0236] Data were analyzed by GraphPadTM Prism® 3.0 (GraphPad Software, San Diego, Calif.) using non-linear regression analysis (sigmoidal dose-response) for the calculation of IC_{50}/EC_{50} values.

$$E_{max}$$
(agonist) or D_{max} (antagonist) = $\frac{\text{Top} - \text{Bottom}}{\text{Bottom}} \times 100$

[0237] Where Top and Bottom correspond to the top and bottom values of the dose-response curve calculated by GraphPadTM Prism®).

Example Method B4

Rabbit Duodenum Contractility Assay

[0238] Duodenal segments were vertically suspended in organ chambers of 10 mL filled with Krebs buffer and connected to an isotonic force transducer, with a preload of 1 g. After a stabilization period, the muscle strips were challenged with 10^{-4} M acetylcholine and washed. This was repeated until a stable maximal contraction was obtained (2-3 times), with an interval of at least 20 minutes.

[0239] After a stable base line was reached, test compounds were added to the bath. After 15 min incubation, a dose response to motilin was recorded by adding logarithmically increasing concentrations of motilin to the bath (final concentration 10^{-9} to 10^{-6} M). A blank experiment (no test compound present) was also performed. At the end of the dose response curve, a supramaximal dose of acetylcholine (10^{-4} M) was given and this response was used as a reference (100% contraction).

[0240] The results of experiments at different concentrations of test compound were combined and analyzed to derive the pA_2 value from the Schild plot.

[0241] It is appreciated that although specific experimental methods have been described herein for the purposes of illustration, various modifications to these experimental methods as well as alternate methods of experimentation may be used without departing from the scope of this invention.

TABLE 3

	Binding activ	ity of selected compounds		
R_1	R_3	R_6	Т	$K_i^{1, 2}$
OH OH			Z_3	В

TABLE 3-continued

	Binding activ	vity of selected compounds		
R_1	R_3	R_6	T	$K_i^{1,2}$
2		- Sand	$\sum_{0}^{Z_3}$	A
3			$\sum_{0}^{Z_3} x$	В
4 AND OH			$\sum_{0}^{Z_3}$	A
5 OH	СН3		$\sum_{0}^{Z_3}$	В
OH OH		Now S	Z_3	В
OH	-	- See	Z_3	В
OH OH		www.	$\sum_{0}^{Z_3}$	В
OH OH			$\sum_{0}^{Z_3}$	В
10 OH	-	· · · · · · · · · · · · · · · · · · ·	$\sum_{0}^{Z_3}$	A
11		- Source	Z_3	A
′				

TABLE 3-continued

Binding activity of selected compounds						
R_1	R_3	R_6	T	$K_i^{1, 2}$		
12 miles	-он	- Section 1	Z_3	В		
13	>	- Annual Control of the Control of t	Z_3	В		
14			\sum_{0}^{Z3}	В		
15			Z_3	A		
16	-CI		Z_3	A		
17 NH			Z_3	В		
18 product	-он		$\sum_{0}^{Z_3}$	В		
19	—F		Z_3	A		
20	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	and	Z_3	В		
21		and and a second	Z_3	A		

TABLE 3-continued

Binding activity of selected compounds						
R ₁	R_3	R_6	Т	$K_i^{1,2}$		
22	_o_ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	- man	Z_3	A		
23			Z_3	A		
24 222	>		Z3 X	A		
25 CI		- Andrews	Z_3	В		
26 renter	_он		Z_3	A		
27 27	\)		Z_3	В		
28			Z3 X	В		
29 mm	_он		Z3 X	В		
30	_он		Z3 X	В		
34 rrr	OH OH	- man	X	В		

TABLE 3-continued

Binding activity of selected compounds					
R_1	R_3	R_6	T	$K_i^{1,2}$	
OH OH	-	$\mathrm{CH_3}$	X	С	
OH 39		н	\sum_{0}^{Z3}	В	
40 AM		Н	\sum_{0}^{Z3}	С	
41 OH	why.	- And - Control	\sum_{0}^{Z3}	С	
52 OH			\sum_{0}^{Z3}	В	
OH OH	- ***	NH ₂	\sum_{0}^{Z3}	В	
oH	-	λ_{NH_2}	X	В	
57 OH	- ***	OH	\sum_{0}^{Z3}	В	
OH OH		www.	$\sum_{i=1}^{N} Z_{i}^{2}$	В	
OH COH	- w	NH NH	$\bigcup_{0}^{Z3} X$	В	

TABLE 3-continued

Binding activity of selected compounds					
	R_1	R_3	R_6	T	$K_i^{1, 2}$
60	Not the second s		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Z_3	С
65	W. OH			$X \longrightarrow Z3$	В
71	Note that the second se			$X \longrightarrow O \longrightarrow Z3$	В
72	W. OH	-		Z3 X	В
76	www.			Z_3	С
77	Varyang OH		- Sandara	Z_3	С
80	HO	-		$\sum_{0}^{Z_3} X$	В
85	www.	-	Н	$\sum_{0}^{Z_3} X$	В
87				Z_3	В
88	ryhy O	-		Z_3	С

TABLE 3-continued

	Binding activi	ty of selected compounds		
R_1	R_3	R_6	T	$K_i^{1, 2}$
89			Z_3	С
90			Z_3	С
91 OH			Z_3	С
92 North OH	- \		$\sum_{0}^{Z_3}$	В
96 Purkun OH	Н		Z_3	С
97 ANNON OH	No. of the second secon		$\sum_{0}^{Z_3}$	С
OH OH			$X \longrightarrow N \longrightarrow Z3$	С
OH OH	- \$ - <		$X \longrightarrow N \longrightarrow Z3$	С
109 VILLE	- \$ - \	NH ₂	Z_3	В
OH OH	***		X	В

TABLE 3-continued

	Binding activity of selected compounds						
	R_1	R_3	R_6	Т	$K_i^{1,2}$		
111	Very OH			Z_3	В		
112	OH OH		- Source of the second of the	Z_3	В		
122	N			Z_3	В		
123	N			Z_3	В		
124	N			Z_3	В		
125	ОН		- Sandara	Z_3	В		
126	ОН			\sum_{0}^{Z3}	В		
127	N. H.			X Z3 S	В		
128				Z_3	В		

TABLE 3-continued

		Binding activ	ity of selected compounds		
	R_1	R_3	R_6	T	$K_i^{1, 2}$
129		- Annual Control of the Control of t		Z3 X	A
130	s			Z_3	В
131	CI	- And a		\sum_{0}^{Z3}	A
132	CI			$\sum_{0}^{Z_3} x$	A
133	F			\sum_{0}^{Z3}	A
134	CI			CI	С
135					В
136		- Anna		$ \begin{array}{c c} & Z3 \\ \hline & X \end{array} $	В
137	s ·····			$ \begin{array}{c c} & Z3 \\ & X \end{array} $	В

TABLE 3-continued

		Binding activ	rity of selected compounds		
	R_1	R_3	R_6	Т	$K_i^{1,2}$
138	Cl	- A		X	В
139	CI	-		X	В
140	F			$ \begin{array}{c} $	В
141				Z^3	A
142				Z3 X	A
143		- Anna		Z_3	В
144	s ·····/····			Z_3	A
145				Z_3	A
146	CI			Z_3	A

TABLE 3-continued

Binding activity of selected compounds					
	R_1	R_3	R_6	T	$K_i^{1,2}$
147	CI			X	В
148	F			X	В
149				$\sum_{i=1}^{N} z_{i}$	A
150	CI			$ \begin{array}{c c} & Z3 \\ & X \end{array} $	В
151	F			$ \begin{array}{c c} & Z3 \\ & X \end{array} $	В
152			NH ₂	Z_3	В
153			NH ₂	$\sum_{0}^{Z_3} X$	В
154	s indicate the second s	- ****	NH ₂	Z_3	В
155	CI			Z_3	A
156	CI		NH ₂	Z_3	A

TABLE 3-continued

Binding activity of selected compounds					
	R_1	R_3	R_6	T	$K_i^{1,2}$
157	F	-	NH ₂	Z_3	В
158	CI		NH ₂	X	A
159	OH	· · · · · · · · · · · · · · · · · · ·		$\sum_{0}^{Z_3} X$	В
160	OH			X	В
161	NH ₂		Source	Z_3	В
162		- www.	- Andrews	$\sum_{0}^{Z_3} x$	В
163		-		Z_3	A
164				Z_3	В
165				Z_3	A
166		- www.		Z_3	В

TABLE 3-continued

		Binding activ	rity of selected compounds		
	R_1	R_3	R_6	T	$K_i^{1, 2}$
167	S S			Z_3	A
168				Z_3	A
169	CI	- \		$\sum_{0}^{z_3}$	В
170	Cl	- Marie 1		$\sum_{0}^{z_3} x$	A
171	F			Z_3	A
172				Z_3	A
173				Z_3	В
174				Z_3	В
175	OH			\sum_{0}^{Z3}	В
176	ОН			Z_3	В

TABLE 3-continued

Binding activity of selected compounds					
	R_1	R_3	R_6	T	$K_i^{1,2}$
177	ОН			$\sum_{0}^{Z3} X$	В
178			ОН	Z_3	В
179			OH	Z3 X	В
180	s ········	-	OH	Z_3	В
181	CI		ОН	$\sum_{0}^{Z_3}$	A
182	CI	-	ОН	\sum_{0}^{Z3}	A
183	F		OH	Z_3	В
184			NH ₂	X	В
185	CI		NH ₂	$\sum_{i=1}^{N} z_{i}$	В

TABLE 3-continued

	Binding activ	vity of selected compounds		
R_1	R_3	R_6	T	$K_i^{1, 2}$
186 OH		NH	$\sum_{0}^{Z_3} X$	В
187 F F	***		\sum_{0}^{Z3}	A
188 F	-		\sum_{0}^{Z3}	A
189 N-O	- www.		Z_3 X	В
190 S			Z_3	A
191			$\sum_{0}^{Z_3}$	A
192 CI			$\sum_{0}^{Z_3} X$	A
193 F			$\sum_{0}^{Z_3} X$	A
194			O X X X	В

TABLE 3-continued

Binding activity of selected compounds						
R_1	R_3	R_6	T	$K_i^{1,2}$		
195	- day		Z^3	A		
196 CI			H_2N X X			
197 CI	- Anna		Z_3			
198 CI	- Land		Z^3	A		
199 CI			Z3 OH X	В		
200 CI			Z_3	A		
201	- Announce of the second		Z_3	В		
202	-		\sum_{0}^{Z3}	A		
203 CI		HN	Z_3	В		

TABLE 3-continued

Binding activity of selected compounds					
I	$t_{\scriptscriptstyle \mathrm{I}}$	R_3	R_6	Т	$K_i^{1, 2}$
204	CI	- Annor	NH NH2	Z_3	A
205	Cl			Z_3	В
206	CI		HN — S — O	Z_3	В
207	CI		HN NH ₂	Z_3	В
208	ОН		NH NH ₂	Z_3	В
209	ОН	-	HN—O	$\sum_{0}^{Z_3}$	С
210	ОН	-	HN	Z_3	
211	CI	-	N	X	A
212	CI		NH	$\sum_{0}^{Z_3} X$	A
213	OH			$\sum_{i=1}^{n} z_{i}$	В

TABLE 3-continued

Binding activity of selected compounds						
R ₁	R_3	R_6	T $K_i^{1,2}$			
214		ОН	Z_3 Z_3 Z_3			
215 OH			Z_3 B X			
216 CI			Z_3 A X			
217 CI			F $Z3$ X			
218 CI	-		Z_3 X			
219 CI	-		CI $Z3$ X			
220 F	-		Z_3 X			
221	-		Z_3 B			
2222 OH			Z_3 A X			
223 ОН			Z_3 C X			

TABLE 3-continued

Binding activity of selected compounds						
	R_1	R_3	R_{6}	Т	$K_i^{1,2}$	
224				Z3 X	В	
225				Z_3	В	
226				Z_3 X	С	
227				Z3 X	В	

Notes

Radioligand competitive binding assays performed using Method B1 Values reported as ranges: A = 0.001-0.100 μ M; B = 0.100-1.0 μ M; C = 1.0-10.0 μ M

X is NH except for:

Compound 223 and 225, X is:



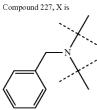
Compound 224, X is NMe

Compound 226, X is:



TABLE 3-continued

	Binding activ	ity of selected compounds		
R_1	R_3	R_6	T	$K_i^{1, 2}$



 Z_1, Z_2 and Z_3 are NH except for compounds 30, 173 and 174 and where Z1 is O and compound 111 wherer Z_2 is O.

 R_2 , R_4 and R_5 are hydrogen except for compound 85 where it is:

1-34. (canceled)

35. A compound of formula (III):

$$\bigcap_{X} OPG_4, \qquad (III)$$

wherein R₁ is C₁-C₄ alkyl; X is halogen or triflate; and PG₄ is hydrogen or a hydroxy protecting group.

36. The compound of claim 35, wherein X is iodine.

37. The compound of claim 35, wherein R_1 is methyl.

38. The compound of claim **35**, wherein PG_4 is hydrogen.

39. The compound of claim 35 represented by the following structures:

$$\bigcap_{I} \operatorname{OPG_4}$$

$$\bigcap_{I} \operatorname{OPG_4}$$

$$\bigcap_{I} \operatorname{OPG_4}$$

$$\bigcap_{I} \operatorname{OPG_4}$$

wherein PG_4 is hydrogen or a hydroxy protecting group. **40-46**. (canceled)

48. The compound of claim 39, wherein PG₄ is a hydroxy protecting group.

49. A process of using a compound of claim 35 to make a compound of formula (I);

$$\begin{array}{c} R_1 \\ O \\ \end{array} \\ NH_2 \end{array}$$

wherein R_1 is C_1 - C_4 alkyl.

50. A process of using a compound of claim 35 to make a compound of formula (G);

$$\begin{array}{c} R_{I} \\ O \\ \end{array} \\ NHPG_{I} \end{array} \tag{G}$$

wherein R_1 is C_1 - C_4 alkyl and PG_1 is an amine protecting