A self-polishing and/or antifouling coating composition comprising a binder, wherein the binder comprises a copolymer comprising (i) units of A and (ii) units of B and/or C; wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a \( \text{C}_1-\text{C}_8 \) alkyl group or phenyl, \( R_3 \) is \( H \), a \( \text{C}_1-\text{C}_8 \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a \( \text{C}_1-\text{C}_8 \) alkenylene group, \( R'_3 \) is \( H \), a \( \text{C}_1-\text{C}_8 \) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a \( \text{C}_1-\text{C}_8 \) alkenylene group.
A Coating Composition

The present invention relates to a binder (i.e. a film forming binder), particularly to the use of a binder in a self polishing and/or antifouling coating composition suitable for marine applications. The present invention also relates to a self polishing and/or antifouling coating composition comprising a binder, which composition is suitable for marine applications.

Coating compositions formulated to prevent the adhesion of micro-organisms, plants and animals to surfaces are well known in the art. An area where such coatings are of particular interest is in marine applications, where a surface is exposed to water which contains organisms that will adhere to the surface, thus fouling the surface. For example, if the surface is the hull of a ship, the increase in frictional resistance caused by the adhesion of organisms such as barnacles to the surface leads to a drastic reduction in the fuel efficiency of the ship.

Traditionally, there have been three ways that a coating composition can be designed to reduce and/or prevent the adhesion and build up of fouling agents on a surface. Firstly, the coating composition can contain a biocide agent (such as an antifouling biocide agent, also known as an antifoulant agent) which serves to physiologically disrupt or kill the marine organism. This can happen either prior to, during or after adhesion of the organism to the surface such that the organism falls away from the surface. This mode of adhesion reduction/prevention is often referred to as "antifouling" and such coatings often referred to as antifouling coatings.

Secondly, the coating composition may be designed to slowly degrade over time, such that organisms adhered to the surface will gradually fall off the surface with the degradation of the coating. The degradation is often caused by a slow hydrolysation of the coating (usually a binder within the coating). This mode of adhesion reduction/prevention is often referred to as self polishing and such coatings are often referred to as self polishing coatings or ablative coatings. These coatings often work by having a binder that hydrolyses in marine conditions which results in the controlled degradation of the coating and causes adhered marine organisms to fall away from the coated surface.

Finally, coatings have been developed which have a very smooth, slippery, low-friction surface onto which fouling organisms have difficulty attaching. Any which do attach, often do
so only weakly and can usually be easily removed, especially under marine conditions with water washing over the coated surface. Such coatings are often referred to as fouling release coatings.

In order to obtain an effective and efficient removal of organisms from surfaces it is now common practice to produce coating compositions that both contain a biocide agent (such as an antifouling biocide agent/antifoulant agent) and which slowly degrade over time. Such dual functional coatings are often referred to as self polishing antifouling coatings.

One aspect of self polishing and/or antifouling coatings that can greatly affect the polishing rates or antifouling properties is the binder.

The present invention provides a binder for marine self polishing and/or antifouling coating compositions.

According to a first aspect of the present invention, there is provided a self polishing and/or antifouling coating composition comprising a binder. The binder comprises a copolymer comprising (i) units of A and (ii) units of B and/or C:

\[
\begin{align*}
R_1 & \quad R_2 \\
\text{Si} & \quad \text{O} \\
A & \\
\end{align*}
\]

\[
\begin{align*}
R_3 & \quad R_4 \\
\text{O} & \quad \text{O} \\
B & \\
\end{align*}
\]

\[
\begin{align*}
R'_3 & \quad R'_4 \\
\text{O} & \quad \text{O} \\
C & \\
\end{align*}
\]

wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_3 \) is \( H \), a \( \text{CrC}_6 \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a \( \text{C}_{1,3} \) alkylene group, \( R'_3 \) is \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a \( \text{C}_{1,3} \) alkylene group.

In one embodiment according to a first aspect of the present invention, the binder comprises a copolymer comprising units of each of A and B:
wherein $R_1$ and $R_2$ are each independently $H$, a $\text{Ci-C}_6$ alkyl group or phenyl, $R_3$ is $H$, a $\text{CrC}_6$ alkyl group or phenyl, and $R_4$ is optional and, when present, is a $\text{C}_1\text{C}_3$ alkyene group.

In one embodiment, according to a first aspect of the present invention, the binder comprises a copolymer comprising units of each of $A$ and $C$:

wherein $R_1$ and $R_2$ are each independently $H$, a $\text{Ci-C}_6$ alkyl group or phenyl, $R_3'$ is $H$, a $\text{Ci-C}_6$ alkyl group or phenyl, and $R_4'$ is optional and, when present, is a $\text{C}_1\text{C}_3$ alkyene group.

In one embodiment according to a first aspect of the present invention, the binder comprises a copolymer comprising units of each of $A$, $B$ and $C$:

wherein $R_1$ and $R_2$ are each independently $H$, a $\text{Ci-C}_6$ alkyl group or phenyl, $R_3$ is $H$, a $\text{CrC}_6$ alkyl group or phenyl, $R_4$ is optional and, when present, is a $\text{C}_1\text{C}_3$ alkyene group, $R_3'$ is $H$, a $\text{Ci-C}_6$ alkyl group or phenyl, and $R_4'$ is optional and, when present, is a $\text{C}_1\text{C}_3$ alkyene group.
The units A can be considered as siloxane units.

The units B and C can each be considered as ester units.

The copolymer may comprise (i) single units A and (ii) single units B and/or C linked so as to provide an alternating copolymer, a periodic copolymer or a random copolymer. For example, the copolymer may comprise single units A and B linked so as to provide an alternating copolymer (i.e. of the structure \[-\text{A-B-A-B-}\]), a periodic copolymer or a random copolymer.

In one embodiment, the copolymer comprises (i) blocks of the units A and (ii) blocks of the units B and/or C so as to provide a block copolymer.

In one embodiment, the copolymer comprises blocks of the units A and/or blocks of the units B, for example blocks of the units A and blocks of the units B, so as to provide a block copolymer. For example, in the block copolymer the blocks of the units A and the blocks of the units B may be dimers, oligomers or polymers, for example dimers, oligomers or polymers A and B' represented as follows:

\[
\begin{align*}
A' & = \left[ \begin{array}{c}
\text{Si} \\
\text{O} \\
\text{R}_1 \\
\text{R}_2
\end{array} \right]_x \\
B' & = \left[ \begin{array}{c}
\text{O} \\
\text{R}_3 \\
\text{R}_4
\end{array} \right]_y
\end{align*}
\]

wherein \(x\) and \(y\) are each greater than 1. For example, \(x\) and \(y\), which may be the same or different, may each be greater than 10. In one embodiment, \(x\) and \(y\) each independently represent an integer of from 10 to 500.

In one embodiment, the copolymer comprises blocks of the units A and/or blocks of the units C, for example blocks of the units A and blocks of the units C, so as to provide a block copolymer. For example, in the block copolymer the blocks of the units A and the blocks of the units C may be dimers, oligomers or polymers, for example dimers, oligomers or polymers A' and C represented as follows:
wherein x and z are each greater than 1. For example, x and z, which may be the same or different, may each be greater than 10. In one embodiment, x and z each independently represent an integer of from 10 to 500.

The blocks A can be considered as poly(siloxane) units.

The blocks B' and the blocks C can be considered as polyester units.

The copolymer may comprise any suitable combination of units A, units B, units C, blocks A', blocks B' and blocks C.

In one embodiment, the copolymer may comprise any suitable combination of units A, units B, blocks A' and blocks B'.

References herein to binder(s) and/or copolymer(s) comprising (i) units of A and (ii) units of B and/or C of course include those wherein the units A and B and/or C may be included as blocks of A and B and/or C (i.e. blocks A' and/or B' and/or C). References herein to binder(s) and/or copolymer(s) comprising units of A and B of course include those wherein the units A and/or B may be included as blocks of A and/or B (i.e. blocks A' and/or B'). In other words, copolymers comprising units A and B may comprise these units as blocks of A (i.e. A') and/or as blocks of units B (i.e. B').

In one embodiment, the copolymer is a tri-block copolymer, for example a tri-block copolymer comprising a central block A' with blocks B' either side.

The units A and B and/or C (and the blocks A' and B' and/or C) can be connected to each other using any suitable chemical process, as would be known to persons skilled in the art, so as to form the desired copolymer.
In one embodiment, the units A and B (and the blocks A' and B') can be connected to each other using any suitable chemical process, as would be known to persons skilled in the art, so as to form the desired copolymer. The units A and B (and the blocks A' and B') may be directly bonded to each other so as to form the desired copolymer. The units A and B (and the blocks A and B') may be attached to each other by means of any suitable chemical linker group. Thus, the copolymer may have the following structure, for example where single units A and B are linked:

![Chemical Structure 1](image1)

or the copolymer may have the following structure, for example wherein the copolymer is a block copolymer:

![Chemical Structure 2](image2)

or the copolymer may have the following structure, for example wherein units A are copolymerised with blocks B':

![Chemical Structure 3](image3)

wherein R₁, R₂, R₃, R₄, x and y are as defined herein and wherein L and L' each independently represent a direct bond or a linker group. Each occurrence of the integer y may be the same or different.
For example, when \( L \) and/or \( L' \) is a linker group, \( L \) and/or \( L' \) may each independently represent a group \(-O-(CR_5R_6)_n\) or \(-O-(CR_7R_8)_mSi(R9Rio)\) wherein \( R_5 \) and \( R_6 \) are each independently \( H \) or a \( CrC_6 \) alkyl group, \( n \) is 1 to 5, \( R_7 \) and \( R_8 \) are each independently \( H \) or a \( CrC_6 \) alkyl group, \( m \) is 1 to 5 and \( R_9 \) and \( Rio \) are each independently a \( Cl-C_6 \) alkyl group.

The linker groups \( L \) and \( L' \) may be the same or different. For example, a suitable linker group \( L \) and/or \( L' \) may be \(-OCH_2\). Another suitable linker group may be \(-OCH_2Si(R_5R_6)\) wherein \( R_5 \) and \( R_6 \) are each independently \( H \) or a \( CrC_6 \) alkyl group, such as \(-OCH_2Si(CH_3)_2\). Thus, the copolymer may typically comprise linker groups \( L \) and \( L' \) in addition to the units \( A \) and \( B \) and/or \( C \) (including blocks \( A \) and \( B' \) and/or \( C' \)). In one embodiment, the copolymer comprises linker groups \( L \) and \( L' \) in addition to the units \( A \) and \( B \) (including blocks \( A' \) and \( B' \)).

The copolymers can be terminated by any suitable end group, as would be appreciated by persons skilled in the art. The nature of the end groups depends on the method by which the copolymer is formed and the reagents used. Thus, the copolymer may have the following structure, for example where it is a tri-block copolymer:

![Copolymer Structure](image)

wherein \( R_1, R_2, R_3, R_4, x, y, L \) and \( L' \) are as defined herein and wherein \( M \) and \( M' \) each independently represent a suitable end group. As discussed above, each occurrence of the integer \( y \) may be the same or different.

The end groups \( M \) and \( M' \) may be the same or different. For example, a suitable end group \( M \) and/or \( M' \) is \( H \). Thus, the copolymer may typically comprise end groups \( M \) and \( M' \) in addition to the units \( A \) and \( B \) and/or \( C \) (including blocks \( A' \) and \( B' \) and/or \( C' \)), such as the units \( A \) and \( B \) (including blocks \( A' \) and \( B' \)) and the linker groups \( L \) and \( L' \).

The units \( A \) and \( B \) (and similarly the blocks \( A' \) and \( B' \)), the linker groups \( L \) and \( L' \) described herein and the end groups \( M \) and \( M' \) described herein are represented above with the chemical groups shown in one direction, but could of course be transposed in the
copolymers, for example such that the units A and B (and similarly the blocks A' and B') may also be represented as:

wherein $R_1$, $R_2$, $R_3$ and $R_4$ are as defined herein.

Similarly, the units C may also be represented as:

wherein $R'_3$ and $R'_4$ are as defined herein.

Similarly, the linker group L may be represented as $-(CR_5R_6)_n-0-$ wherein $R_5$, $R_6$ and $n$ are as defined herein or as $-Si(R_9R_{10})(CR_7R_8)_m-0-$ wherein $R_7$, $R_8$, $R_9$, $R_{10}$, $n$ and $m$ are as defined herein.

A binder which is a tri-block copolymer, may have the following structure:
wherein $R_1$, $R_2$, $R_3$, $R_4$, $x$ and $y$ are as defined herein and wherein $L$ represents a direct bond or a linker group as defined herein. As discussed above, each occurrence of the integer $y$ may be the same or different.

For example, where $L$ is -0-(CR$_5$R$_6$)$_n$, the tri-block copolymer may have the structure:

![Tri-block copolymer structure](image)

wherein $R_1$, $R_2$, $R_3$, $R_4$, $R_5$, $R_6$, $M$, $M'$, $x$, $y$ and $n$ are as defined herein. As discussed above, each occurrence of the integer $y$ may be the same or different.

For example, where $L$ is -0-(CR$_5$R$_6$)$_n$ and $L'$ is -0-(CR$_7$R$_8$)$_m$-Si(R$_9$R$_{10}$)-, the tri-block copolymer may have the structure:

![Tri-block copolymer structure](image)

wherein $R_1$, $R_2$, $R_3$, $R_4$, $R_5$, $R_6$, $R_7$, $R_8$, $R_9$, $R_{10}$, $M$, $M'$, $x$, $y$, $n$ and $m$ are as defined herein. As discussed above, each occurrence of the integer $y$ may be the same or different.

For example, where $L$ is -0-(CR$_5$R$_6$)$_n$, $L'$ is -0-(CR$_7$R$_8$)$_m$-Si(R$_9$R$_{10}$)-, $M$ is H and $M'$ is H, the tri-block copolymer may have the structure:
wherein $R_1$, $R_2$, $R_3$, $R_4$, $R_5$, $R_6$, $R_7$, $R_8$, $R_9$, $R_{10}$, $n$, $m$, $x$ and $y$ are as defined herein. As discussed above, each occurrence of the integer $y$ may be the same or different.

In one embodiment, in the units $A$ and blocks $A'$, the groups $R_1$ and $R_2$ are each independently methyl or phenyl. In one embodiment, $R_1$ and $R_2$ both represent methyl. When $R_1$ and $R_2$ both represent methyl, the unit $A$ is a dimethylsiloxane group (i.e. $-[\text{Si(CH}_3)_2\text{O}]_2$) and the block $A'$ is a polydimethylsiloxane group (i.e. $-[\text{Si(CH}_3)_2\text{O}]_x$, where $x$ is as defined herein).

In one embodiment, in the units $B$ and blocks $B'$, the group $R_3$ represents H or a C$_{16}$ alkyl group, for example H or a C$_{13}$ alkyl group, such as H or methyl.

In one embodiment, the group $R_3$ represents methyl.

In one embodiment, in the units $B$ and blocks $B'$, the group $R_4$ represents C$_{13}$ alkylene.

In one embodiment, in the units $B$ and blocks $B'$, the group $R_4$ is not present. As the skilled person would appreciate, when the group $R_4$ is not present, $R_4$ represents a direct bond.

In one embodiment, in the units $B$ and blocks $B'$, the group $R_3$ represents H or a C$_{16}$ alkyl group, for example H or a C$_{13}$ alkyl group, such as H or methyl, and the group $R_4$ is not present. Thus, in one embodiment, units $B$ and blocks $B'$ may be represented as one or more of:
wherein \( y \) is as defined herein. These groups \( B \) and \( B' \) are derivable from hydroxycarboxylic acids and poly(hydroxycarboxylic acids), such as poly(glycolic acid), poly(lactic acid), poly((l)-lactic acid), poly(d)-lactic acid) and poly((d, l)-lactic acid), and combinations thereof.

When the group \( R_3 \) represents methyl and the group \( R_4 \) is not present, then the unit \( B \) is a lactic acid residue and the block \( B' \) is poly(lactide). The lactic acid residue may be L-lactic acid residue and/or D-lactic acid residue and the poly(lactide) may be poly(L-lactide) and/or poly(D-lactide). In one embodiment, the molar ratio of the L-lactic acid residue and the D-lactic acid residue is in the range of from 1 to 5. When the block \( B' \) represents poly(lactide), in one embodiment, it is poly(D,L-lactide).

When the group \( R_3 \) represents H and the group \( R_4 \) is not present, then the unit \( B \) is a glycolic acid residue and the block \( B' \) is poly(glycolide).

In one embodiment, in the units \( C \) and blocks \( C' \), the group \( R'_3 \) represents H or a \( C_{1-6} \) alkyl group, for example H or a \( C_{1-3} \) alkyl group, such as H or methyl.

In one embodiment, the group \( R'_3 \) represents methyl.

In one embodiment, in the units \( C \) and blocks \( C' \), the group \( R'_4 \) represents a \( C_{1-3} \) alkylene group.

In one embodiment, in the units \( C \) and blocks \( C' \), the group \( R'_4 \) is not present.
In one embodiment, the copolymer may have a weight ratio of units of B to units of A of from 50:50 to 96:4, for example of from 67:33 to 94:6, such as from 80:20 to 92:8.

In one embodiment, the copolymer may have a weight ratio of units of C to units of A of from 50:50 to 96:4, for example of from 67:33 to 94:6, such as from 80:20 to 92:8.

The copolymer may have a Mw in the range of 2000 to 100000 Dalton, for example of 5000 to 40000 Dalton.

The number-average molecular weight may be measured by any suitable method. Techniques to measure the number-average molecular weight will be well known to a person skilled in the art. Suitably, the Mn may be determined by gel permeation chromatography using a polystyrene standard according to ASTM D6579-11 ("Standard Practice for Molecular Weight Averages and Molecular Weight Distribution of Hydrocarbon, Rosin and Terpene Resins by Size Exclusion Chromatography"). UV detector; 254nm, solvent: unstabilised THF, retention time marker: toluene, sample concentration: 2mg/ml).

Techniques to measure the weight-average molecular weight will be well known to a person skilled in the art. Suitably, the Mw may be determined by gel permeation chromatography using a polystyrene standard.

In one embodiment, the binder is used in such an amount that the proportion thereof in the solid content of the coating composition is from 5 to 35 wt%, of a binder composed of for example from 35 to 5 wt% of A units (such as from 15 to 5 wt% of A units) and from 65 to 95 wt% of B units (such as from 85 to 95 wt% of B units).

The present inventors have surprisingly found that the binder as discussed herein is very flexible. This is unexpected because polymers such as poly(lactic acid) are known to be rigid and have limited uses. Coating compositions comprising binders according to the first aspect of the present invention are also flexible and may result in little or no cracking of the coating when applied to substrates, thereby providing long lasting and effective coatings. The use of units of the type B and/or C (and blocks B' and/or C') in the binders of the coating compositions of the present invention may also be advantageous because the units/binders are typically biodegradable and therefore good for the environment.
Unless otherwise stated, the term "binder" as used herein means a substance (for example a copolymer) that will form a film on a substrate. In other words, the binder may be a film forming copolymer that can be used as a film forming component, for example in a coating composition (such as an antifouling paint).

Unless otherwise stated, the term "alkyl" as used herein includes both straight chain and branched chain alkyl groups, such as propyl, isopropyl and tert-butyl. However, references to individual alkyl groups such as "propyl" are specific for the straight-chain version only and references to individual branched-chain alkyl groups such as "isopropyl" are specific for the branched-chain version only. A CrC₆ alkyl group has from one to six carbon atoms including methyl, ethyl, n-propyl, isopropyl, tert-butyl, n-pentyl, n-hexyl and the like. References to a C1-C3 alkyl group will be understood accordingly to mean a straight or branched chain alkyl moiety having from one to three carbon atoms.

The term "alkylene", as used herein, relates to a bivalent radical alkyl group as defined above. For example, an alkyl group such as methyl which would be represented as -CH₃, becomes methylene, -CH₂-, when represented as an alkylene. Other alkyene groups should be understood accordingly.

The binder may comprise a copolymer consisting essentially of units of each of A and B (wherein reference to units A and B includes blocks A' and B').

The binder may comprise a copolymer consisting essentially of units of each of A and B (wherein reference to units A and B includes blocks A' and B') and of linker groups L and/or L'.

The binder may comprise a copolymer consisting essentially of units of each of A and B (wherein reference to units A and B includes blocks A' and B'), of linker groups L and/or L' and of end groups M and/or M'.

The binder may comprise a copolymer consisting essentially of units of each of A and B (wherein reference to units A and B includes blocks A' and B') and of end groups M and/or M'.
The binder may comprise a copolymer consisting essentially of units of each of A and C (wherein reference to units A and C includes blocks A and C).

The binder may comprise a copolymer consisting essentially of units of each of A and C (wherein reference to units A and C includes blocks A' and C) and of linker groups L and/or L'.

The binder may comprise a copolymer consisting essentially of units of each of A and C (wherein reference to units A and C includes blocks A' and C), of linker groups L and/or L' and of end groups M and/or M'.

The binder may comprise a copolymer consisting essentially of units of each of A and C (wherein reference to units A, B and C includes blocks A', B' and C).

The binder may comprise a copolymer consisting essentially of units of each of A and B and/or C (wherein reference to units A, B and C includes blocks A', B' and C) and of linker groups L and/or L'.

The binder may comprise a copolymer consisting essentially of units of each of A and B and/or C (wherein reference to units A, B and C includes blocks A', B' and C), of linker groups L and/or L' and of end groups M and/or M'.

The binder may comprise a copolymer consisting essentially of units of each of A and B and/or C (wherein reference to units A, B and C includes blocks A', B' and C) and of end groups M and/or M'.

The binder may comprise a copolymer consisting essentially of units of each of A, B and C (wherein reference to units A, B and C includes blocks A', B' and C).
The binder may comprise a copolymer consisting essentially of units of each of A, B and C (wherein reference to units A, B and C includes blocks A', B' and C') and of linker groups L and/or L'.

The binder may comprise a copolymer consisting essentially of units of each of A, B or C (wherein reference to units A, B and C includes blocks A', B' and C'), of linker groups L and/or L' and of end groups M and/or M'.

The binder may comprise a copolymer consisting essentially of units of each of A, B and C (wherein reference to units A, B and C includes blocks A', B' and C') and of end groups M and/or M'.

There is also provided a method of forming a binder, particularly suitable for a self-polishing and/or antifouling coating composition, the method comprising copolymerising sources of (i) units of A and (ii) units of B and/or C. The units A and B and/or C may of course be individual units A and B and/or C and/or blocks A and B' and/or C.

A method of forming a binder, particularly suitable for a self-polishing and/or antifouling coating composition, may comprise copolymerising sources of units of A and B. The units A and B may of course be individual units A and B and/or blocks A' and B'.

The method may comprise the copolymerisation of a suitable cyclic monomer such as glycolide or lactide with a suitable polysiloxane (such as a bishydroxy terminated polydialkylsiloxane, for example bishydroxyterminated polydimethylsiloxane) in the presence of a suitable ring-opening polymerisation catalyst. Such copolymerisation reactions typically require heating under an inert atmosphere, such as an argon or nitrogen atmosphere. The copolymer is typically then isolated once the copolymerisation reaction is complete.

Another method may comprise the copolymerisation of a suitable aliphatic hydroxycarboxylic acid such as lactic acid or glycolic acid, or of a suitable polymer such as polylactide or polyglycolide, and a suitable polysiloxane (such as a bishydroxy terminated polydialkylsiloxane, for example bishydroxyterminated polydimethylsiloxane) in the presence of a suitable polymerisation catalyst. Such reactions are typically conducted at room temperature.
As the skilled person would appreciate, suitable click chemistry may be used to prepare the desired copolymers using suitable reagents.

For example, when L is a group -O-(CR₅R₆)ₙ, the method may comprise copolymerising a poly(hydroxycarboxylic acid) and a bis(hydroxyalkyl) terminated polysiloxane.

The copolymerisation reaction may be conducted using any suitable method. For example, to prepare a copolymer having a unit B wherein the group R₃ represents methyl and the group R₄ is not present, the copolymerisation reaction may be conducted by a ring opening polymerisation reaction of a bis(hydroxyalkyl) terminated poly(dialkylsiloxane) and 3,6-dimethyl-1,4-dioxane-2,5-dione using a suitable catalyst. A suitable catalyst is a tin catalyst, such as tin octanoate (also referred to as di(2-ethylhexanoate)). Typically, the reaction mixture must be heated to a temperature in the range 70 to 160°C.

For example, the copolymerisation reaction may be conducted by a ring opening polymerisation reaction of bis(hydroxyalkyl) terminated poly(dimethylsiloxane) and 3,6-dimethyl-1,4-dioxane-2,5-dione using a suitable catalyst, such as tin octanoate, to provide a tri-block copolymer:

\[
\begin{align*}
\text{H} & \quad \text{O} \quad \text{O} \\
\text{O} & \quad \text{Si} \quad \text{Si} \\
\text{O} & \quad \text{O} \\
\text{O} & \quad \text{O} \\
\end{align*}
\]

wherein \( R_5, R_6, R_7, R_s, R_9, R_{10}, n, m, x \) and \( y \) are as defined herein. As discussed above, each occurrence of the integer \( y \) may be the same or different.

The binder as discussed herein may be included in the coating composition of the invention either as the only binder content or in combination with other co-binders as mentioned below.

The coating composition typically is a marine coating composition.

In one embodiment, the coating composition is a self polishing antifouling coating composition, such as a self polishing antifouling marine coating composition.
In one embodiment, the binder is used in such an amount that the proportion thereof in the solid contents of the coating composition is 5 to 40 wt%, for example 15 to 30 wt%, such as 17 to 25 wt%.

Typically, the coating composition comprises one or more components in addition to the binder as discussed herein. A person skilled in the art would readily understand what components are typically included in a coating composition, especially a marine coating composition such as a self polishing antifouling coating composition.

Suitably, the coating composition comprises one or more biocides, for example one or more antifoulant agents. Examples of suitable antifoulant agents include any of one or more conventionally known antifoulant agents. The known antifoulant agents are roughly divided into inorganic compounds, metal-containing organic compounds, and metal-free organic compounds.

Examples of the inorganic compounds include copper compounds (for example copper sulphate, copper powder, cuprous thiocyanate, copper carbonate, copper chloride, and the traditionally preferred cuprous oxide), zinc sulphate, zinc oxide, and copper nickel alloys.

Examples of the metal-containing organic compounds include organo-copper compounds, and organo-zinc compounds. Also usable are manganese ethylene bis dithiocarbamate (maneb), propineb, and the like. Examples of the organo-copper compounds include copper nonylphenol-sulphonate, copper bis(ethylenediamine) bis(dodecylbenzene sulphonate), copper acetate, copper naphthenate, copper pyrithione and copper bis(pentachlorophenolate). Examples of the organo-zinc compounds include zinc acetate, zinc carbamate, bis(dimethylcarbamoyl) zinc ethylene-bis(dithiocarbamate), zinc dimethyl dithiocarbamate, zinc pyrithione, and zinc ethylene-bis(dithiocarbamate). As an example of mixed metal-containing organic compound, one can cite (polymeric) manganese ethylene bis dithiocarbamate complexed with zinc salt (mancozeb).

Examples of the metal-free organic compounds include N-trihalomethylthiophthalimides, trihalomethylthiosulphamides, dithiocarbamic acids, N-arylmaleimides, 3-(substituted amino)-1,3 thiazolidine-2,4-diones, dithiocyanato compounds, triazine compounds, oxathiazines and others.
Examples of the N-trihalomethylthiophthalimides include N-trichloromethylthiophthalimide and N-fluorodichloromethylthiophthalimide.

Examples of the dithiocarbamic acids include bis(dimethylthiocarbamoyl) disulphide, ammonium N-methylthiocarbamate and ammonium ethylene-bis(dithiocarbamate).

Examples of trihalomethylothiosulphamides include N-(dichlorofluoromethylthio)-N,N'-dimethyl-N-phenylsulphamide and N-(dichlorofluoromethylthio)-N,N'-dimethyl-N-(4-methylphenyl)sulphamide.

Examples of the N-arylmaleimides include N-(2,4,6-trichlorophenyl)maleimide, N-4-tolylmaleimide, N-3-chlorophenylmaleimide, N-(4-n-butylphenyl)maleimide, N-(anilinophenyl)maleimide, and N-(2,3-xylyl)maleimide.

Examples of the 3-(substituted amino)-1,3-thiazolidine-2,4-diones include 2-(thiocyanomethylthio)-benzothiazole, 3-benzylideneamino-1,3-thiazolidine-2,4-dione, 3-(4-methylbenzylideneamino)-1,3-thiazolidine-2,4-dione, 3-(2-hydroxybenzylideneamino)-1,3-thiazolidine-2,4-dione, and 3-(2,4-dichlorobenzylideneamino)-1,3-thiazolidine-2,4-dione.

Examples of the dithiocyan compounds include dithiocyanomethane, dithiocyanatoethane, and 2,5-dithiocyanothiophene.

Examples of the triazine compounds include 2-methylthio-4-butylamino-6-cyclopropylaminotriazine.

Examples of oxathiazines include 1,4,2-oxathiazines and their mono- and di-oxides such as disclosed in WO 98/05719: mono- and di-oxides of 1,4,2-oxathiazines with a substituent in the 3 position representing (a) phenyl; phenyl substituted with 1 to 3 substituents which are independently hydroxyl, halo, C1-C12 alkyl, C5-C6 cycloalkyl, trihalomethyl, phenyl, C1-C5 alkoxy, C1-C5 alkylthio, tetrahydroprpyranyloxy, phenoxy, C1-C4 alkyl carbonyl, phenyl carbonyl, C1-C4 alkyl sulfinyl, carboxy or its alkali metal salt, C1-C4 alkoxy carbonyl, C1-C4 alkylaminocarbonyl, phenylaminocarbonyl, tolylaminocarbonyl, morpholinocarbonyl, amino, nitro, cyano, dioxolanyl or C1-C4 alkyl oxyiminomethyl; naphthyl; pyridinyl; thieryl; furanyl; or thienyl or furanyl substituted with one to three substituents which are independently C1-C4.
alkyl, C₁₋₄ alkoxy, C₁₋₄ alkylthio, halo, cyano, formyl, acetyl, benzoyl, nitro, C₁₋₄ alkoxy carbonyl, phenyl, phenylaminocarbonyl or C₁₋₄ alkyl oxyiminomethyl; or (b) a substituent of generic formula:

\[
\begin{align*}
&\text{X} \quad \text{Y} \\
&\text{C₆ ring may have one C₁₋₄ alkyl substituent; a second substituent which is C₁₋₄ alkyl or benzyl being optionally present in position 5 or 6.}
\end{align*}
\]

wherein X is oxygen or sulphur; Y is nitrogen, CH or C(CrC₄ alkoxy); and the C₆ ring may have one C₁₋₄ alkyl substituent; a second substituent which is C₁₋₄ alkyl or benzyl being optionally present in position 5 or 6.

Among the fouling organisms, barnacles have proved to be the most troublesome, because they resist to most biocides. Accordingly, the coating composition may also include at least an effective amount of antifoulant agent which includes at least one specific barnaclecide, such as cuprous oxide or thiocyanate. Another suitable barnaclecide is ECONEA (2-(p-chlorophenyl)-3-cyano-4-bromo-5-trifluoromethyl pyrrole) disclosed in EP-A-831 134 and commercially available from Janssen Pharmaceutica. EP-A-831 134 discloses the use of from 0.5 to 9.9 wt%, based on the total weight of the dry mass of the composition, of at least one 2-trihalogenomethyl-3-halogeno-4-cyano pyrrole derivative substituted in position 5 and optionally in position 1, the halogens in positions 2 and 3 being independently fluorine, chlorine or bromine, the substituent in position 5 being CrC₈ alkyl, CrC₈ monohalogenoalkyl, C₅-C₆ cycloalkyl, C₅-C₆ monohalogenocycloalkyl, benzyl, phenyl, mono- or di-halogenobenzyl, mono- or di-halogenophenyl, mono- or di-CrC₄ alkyl benzyl, mono- or di-CrC₄ alkyl phenyl, monohalogeno mono-CrC₄ alkyl benzyl or monohalogeno mono-CrC₄ alkyl phenyl, any halogen on the substituent in position 5 being chlorine or bromine, the optional substituent in position 1 being C₁₋₄ alkyl or C₁₋₄ alkoxy C₁₋₄ alkyl.
Another suitable barnaclecid is SELEKTOPE (also known as medetomidine or 4-[1-(2,3-dimethylphenyl)ethyl]-1/-imidazole), which may be used in such an amount that the proportion thereof in the solid contents of the coating composition is from 0.05 to 0.5 wt%. SELEKTOPE is commercially available from I-Tech AB.

The antifoulant agents are used in such an amount that the proportion thereof in the solid contents of the coating composition is usually from 0.01 to 90 wt%, such as from 0.05 to 80 wt%, for example from 0.5 to 60 wt%. Too small antifoulant agent amounts do not produce an antifouling effect, while too large antifoulant agent amounts result in the formation of a coating film which is apt to develop defects such as cracking and peeling and thus becomes less effective in antifouling property.

The coating composition may comprise one or more primary antifoulant agents such as cuprous oxide (Cu₂O), copper thiocyanate (CuSCN), ECONEA and/or SELEKTOPE.

The coating composition may comprise one or more antifoulant agents such as cuprous oxide and/or copper thiocyanate used in such an amount that the proportion thereof in the solid contents of the coating composition is from 5 to 50 wt%, such as from 7 to 30 wt%, for example from 10 to 20 wt%.

The coating composition may comprise the antifoulant agent ECONEA used in such an amount that the proportion thereof in the solid contents of the coating composition is from 0.5 to 5 wt%, such as from 1 to 4 wt%.

The coating composition may additionally comprise one or more organic co-antifoulant agents dichlofluaninide, tolyfluanide, zinc pyrithione, copper pyrithione, zineb, Irgarol 1051 and/or Sea-nine (with DCOIT as active). Such an organic co-antifoulant agent may be used in such an amount that the proportion thereof in the solid contents of the coating composition is from 0.5 to 5 wt%, for example from 1 to 4 wt%. The addition of an organic co-antifoulant agent is believed to enhance the performance of the primary antifoulant agent.

The present inventors have surprisingly found that the amount of biocide (for example antifoulant agent) useful in the coating compositions of the present invention may be significantly less than that used in other known coating compositions having similar service.
life times. This offers advantages in use of lower costs associated with manufacturing the compositions and also is good for the environment in terms of using less biocide.

The coating composition may comprise one or more co-binders in addition to the binder as discussed herein. Any suitable co-binder may be included on the coating composition, such as rosin or a metal resinate (for example a metal resinate, such as zinc resinate, copper resinate, calcium resinate or magnesium resinate).

The coating composition may contain other components such as solvents, pigments, thixotropic agents, fillers etc.

Any suitable solvent(s) may be used. The solvent may be an organic solvent, such as xylene, toluene, ketones (for example methyl isobutyl ketone (MIBK), methyl amyl ketone (MAK), methyl isamyl ketone (MIAK)), high boiling aromatic solvents, ethyl acetate and butyl acetate, for example.

Any suitable pigment(s) may be used. Suitable pigments may be one or more of iron (III) oxide, titanium dioxide, zinc oxide and/or carbon black for example.

Any suitable thixotropic agent(s) may be used. Suitable thixotropic agents may be one or more of Crayvallac®-types, castor oil derivatives, bentonites and/or Disparion-type agents for example.

Any suitable filler(s) may be used. Suitable fillers may be one ore more of calcium carbonate, barium sulphate, talcum, quartz and/or silicas etc.

According to a second aspect of the present invention, there is provided the use of a copolymer comprising (i) units of A and (ii) units of B and/or C:
wherein R₁ and R₂ are each independently H, a C₁₋₆ alkyl group or phenyl, R₃ is H, a C₆ alkyl group or phenyl, R₄ is optional and, when present, is a C₁₋₃ alkyne group, R₃ is H, a C₁₋₆ alkyl group or phenyl, and R₄ is optional and, when present, is a C₁₋₃ alkyne group, as a binder in a self polishing and/or antifouling coating composition.

In one embodiment according to a second aspect of the present invention, there is provided the use of a copolymer comprising units of A and C:

![Chemical structure](image1)

wherein R₁ and R₂ are each independently H, a C₁₋₆ alkyl group or phenyl, R₃ is H, a C₆ alkyl group or phenyl, and R₄ is optional and, when present, is a C₁₋₃ alkyne group, as a binder in a self polishing and/or antifouling coating composition.

In one embodiment according to a second aspect of the present invention, there is provided the use of a copolymer comprising units of A and B:

![Chemical structure](image2)

wherein R₁ and R₂ are each independently H, a C₁₋₆ alkyl group or phenyl, R₃ is H, a C₆ alkyl group or phenyl, and R₄ is optional and, when present, is a C₁₋₃ alkyne group, as a binder in a self polishing and/or antifouling coating composition.

In one embodiment according to a second aspect of the present invention, there is provided the use of a copolymer comprising (i) units of A and (ii) units of B and C:
wherein \( R_2 \) and \( R_3 \) are each independently \( H \), a \( \text{C}_6\text{H}_{12} \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a \( \text{C}_1\text{H}_2 \) alkyene group, \( R'_3 \) is \( H \), a \( \text{C}_6\text{H}_{12} \) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a \( \text{C}_1\text{H}_2 \) alkyene group.

The copolymer for use in the second aspect of the present invention typically is as defined above, for example with reference to the first aspect of the invention.

According to a third aspect of the present invention, there is provided the use of a binder in a marine self polishing and/or antifouling coating composition, wherein the binder comprises a copolymer comprising (i) units of A and (ii) units of B and/or C:

wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a \( \text{C}_6\text{H}_{12} \) alkyl group or phenyl, \( R_3 \) is \( H \), a \( \text{C}_6\text{H}_{12} \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a \( \text{C}_1\text{H}_2 \) alkyene group, \( R'_3 \) is \( H \), a \( \text{C}_6\text{H}_{12} \) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a \( \text{C}_1\text{H}_2 \) alkyene group.

The binder for use in the third aspect of the present invention typically is as defined above, for example with reference to the first aspect of the invention.

According to a fourth aspect of the present invention, there is provided a method of reducing and/or preventing the adhesion of marine organisms on at least a portion of a surface, the
method comprising the steps of applying a coating composition according to the first aspect of the present invention to at least a portion of the surface.

According to a fifth aspect of the present invention, there is provided the use of a coating composition according to the first aspect of the present invention in reducing and/or preventing the adhesion of marine organisms on at least a portion of a surface to which the coating composition is applied.

The coating composition of the present invention acts to reduce and/or prevent the adhesion of marine organisms on at least a portion of a surface to which the coating composition is applied by acting as an antifouling composition, i.e. wherein a biocide (for example antifoulant agent) in the composition poisons the organisms attached to the portion of the surface. The coating composition of the present invention also acts as a self polishing composition, i.e. by slowly degrading over time. The degradation is believed to be caused by slow hydrolysation of the binder within the coating. In particular, the units B and/or blocks B' (and/or the units C and/or blocks C) of the binder may hydrolyse so as to cause degradation.

According to a sixth aspect of the present invention, there is provided a marine vessel or marine structure coated on at least a portion thereof with a coating composition according to the first aspect of the present invention.

The present invention also extends to a marine vessel or marine structure coated with a coating composition according to the above aspects of the present invention.

All of the features contained herein may be combined with any of the above aspects and in any combination.

The invention will be further discussed with reference to the following non-limiting Examples.

**Examples**

**Examples 1 to 3** Laboratory synthesis of tri-block copolymers by ring opening polymerisation with tin di(2-ethylhexylhexanoate)

Examples 1 to 3 were conducted using the reagents and quantities set out below in Table 1.
The reaction was conducted in a three-necked reaction vessel of 100 ml, equipped with a magnetic stirrer bar, a cooler with bubbler and an inlet for dry nitrogen gas.

3,6-dimethyl-1,4-dioxane-2,5-dione, Tegomer® H-Si (i.e. bishydroxy terminated polydimethylsiloxane, available commercially and purchased from Evonik) and tin octanoate (also referred to as di(2-ethylhexanoate)) were added to toluene in the amounts shown in Table 1. The reaction mixture was then de-aerated under a constant stream of nitrogen and cooled with an ice bath for 30 minutes. The reaction mixture was then stirred and heated to 120°C for 24 hours. Then the toluene was removed by rotational evaporation. The residue was dissolved in tetrahydrofuran (10 ml). The polymer was precipitated by addition of sufficient cold petroleum ether, filtered and dried to yield a white powder.

Prior to formulation as coating compositions, the white powder product for each of Examples 1 to 3 was dissolved in xylene (see Table 1 below for amounts of xylene).

**Example 4 Large Scale synthesis of tri-block copolymer by ring opening polymerisation with tin di(2-ethylhexylhexanoate)**

Example 4 was conducted using the reagents and quantities set out below in Table 1.

Tegomer HSi 231 1 (128g, 0.0009 mol), tin(ll)2-ethylhexanoate (3.9g, 0.0001 mol), 3,6-dimethyl-1,4-dioxane-2,5-dione (1174g, 0.13 mol) and toluene (695g) were added to a 2-liter, 4-necked flask equipped with a motor driven stainless steel stir blade, a water-cooled condenser, a nitrogen inlet, and a heating mantle with a thermometer connected through a temperature feedback control device. The contents of the flask were heated to 75°C and held at that temperature for 10 to 12 hours, until all of the monomers had reacted (i.e. so that the practical solids (determined at 110°C) matched with theoretical solids). The temperature was then decreased to room temperature and poured into a can. The resulting PDMS-lactate copolymer was characterized and a sample of it was placed in a 50°C hot room for 4 weeks, and the copolymer remained liquid.

Prior to formulation as coating compositions, the product of Example 4 was added to xylene (see Table 1 below for amount of xylene).
The products were analysed using the following techniques and the results are shown below and in Table 1:

1H NMR

Bruker 500 MHz.

Samples were dissolved in deuterated chloroform (CDCl₃).

The NMR 1H values are:

- for the poly(D,L)lactide: 5.2 ppm (0-CH<, 1 H), 1.65 ppm (CH3-CH<, 3H).
- for the polydimethylsiloxane unit: 0.06 ppm (CH3-Si, 6H).
- for one linker group: 0.55 ppm (Si-CH2-CH2-, 2H), 1.4 ppm (Si-CH2-(CH2)4-CH2, 8H), 1.55 ppm (-CH2-CH2-CH2-0, 2H), 4.05 ppm (-CH2-CH2-0-, 2H).

Gel Permeation Chromatography

Merck pump attached to 2 columns of PLgel of Polymer Laboratories (Mixed-E, 3 µη and Mixed-D, 5 µη).

Eluent solvent: Tetrahydrofuran at 1 mL/min.
Injection of samples: 20 µL of circa 1 mg/L.
Polymer for calibrations was Easical PS-2 from Agilent Technologies

Infrared

Bruker Tensor 27 with a Harrick MVP 2 series cell

DSC (for crystallinity)

Mettler Toledo DSC 822
10 mg samples
Heating cycles: 25°C to 100°C at 20 °C/min, cooling to 100°C at 20 °C/min, 2 min at 100°C and reheating to + 100°C at 20 °C/min.
The copolymers were obtained with a good control on the molecular weight and on the molecular ratio.

<table>
<thead>
<tr>
<th>Table 1</th>
</tr>
</thead>
<tbody>
<tr>
<td>Example (EG)</td>
</tr>
<tr>
<td>reactor size (L)</td>
</tr>
<tr>
<td>Toluene (g)</td>
</tr>
<tr>
<td>3,6-dimethyl-1,4-dioxane-2,5-dione (g)</td>
</tr>
<tr>
<td>Tegomer H-Si (g)</td>
</tr>
<tr>
<td>tin (II)octanoate (g)</td>
</tr>
<tr>
<td>yield (g)</td>
</tr>
<tr>
<td>addition xylene at 50 wt%</td>
</tr>
<tr>
<td>total 50 wt% binder solution</td>
</tr>
</tbody>
</table>

**Polymer Characterisation**

| Mn theoretical (kD) | 4.5 | 6.6 | 20.3 | 20 |
| Mn measured by gas permeation chromatography (kD) | 7.3 | 9.4 | 22.4 | 20 |
| Mn measured by 1H NMR (kD) | 4.3 | 6.9 | 18.1 | n/a |
| Mw/Mn | 1.4 | 1.4 | 1.3 | n/a |
| Theoretical molar ratio polylactide/polysiloxane | 1.0 | 2.0 | 9.0 | 9.0 |
| molar ratio polylactide/polysiloxane by 1H NMR after synthesis | 0.89 | 2.1 | 7.3 | n/a |

**physical properties after 250 days immersion in fresh water**

| % Decrease of Mn | n/a | 37.0 | 34.0 | n/a |
| water contact angle fresh (°) | n/a | n/a | 98 | n/a |
| water contact angle after 250 days (°) | n/a | n/a | 92# | n/a |

#varies in time

n/a = not available
Examples 5 to 8

Coating compositions were prepared by addition of the binders of Examples 2 to 4 to other components as listed in Table 2 below, under high speed dispersing at 1500 rpm for 30 minutes. For further study, the coating compositions were applied to a polycarbonate sheet (laboratory studies) or PVC panels (raft immersion test). The properties of the coating compositions were then studied as follows:

The water absorption of immersed paint films was determined by a Karl-Fisher Colorimeter technique. The coating compositions based on the binders of Examples 2 and 3 absorbed 34 to 40 wt% water after 150 days. That level was reached in 150 days for the coating composition based on the binder of Example 3.

The maximum loss of mass weight (Mn) of the binder in the immersed coating compositions is about 26% for the binder of Example 2 and 34% for the binder of Example 3. This shows that the coating compositions comprising binders according to the present invention degrade well in water.

The erosion of the coating compositions was studied in freshwater. The coating composition comprising a binder according to the present invention showed fast, consistent and constant erosion in time (see Table 2).

The coating compositions comprising a binder according to the present invention showed a 40 \( \mu \eta \) decrease in 280 days. PVC panels were coated with the coating compositions and were static immersed in the harbor of Lorient. The antifouling resistance, erodability and film integrity were judged at regular intervals. For paints on the raft the coating compositions comprising binders according to the present invention showed constant erosion after 7 months, with the coating composition comprising the binder of Example 3 being the fastest. The coating composition with the best erosion also showed the best antifouling performance and stayed fouling free over the whole test period of 7 months in the port of Lorient (Bretagne).
### Table 2

<table>
<thead>
<tr>
<th>Example</th>
<th>Average particle size (Mm)</th>
<th>Ranges in wt%</th>
<th>5 wt%</th>
<th>6 wt%</th>
<th>7 wt%</th>
<th>8 wt%</th>
</tr>
</thead>
<tbody>
<tr>
<td>Binder</td>
<td>EG 2</td>
<td>EG 3</td>
<td>EG 4</td>
<td>EG 4</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Mn</td>
<td>(kD)</td>
<td>6.9</td>
<td>18.1</td>
<td>18.1</td>
<td>16</td>
<td></td>
</tr>
<tr>
<td>Binder</td>
<td>17-22</td>
<td>22.0</td>
<td>20.6</td>
<td>20.6</td>
<td>21.0</td>
<td></td>
</tr>
<tr>
<td>TiO2</td>
<td>0.2</td>
<td>15 - 20</td>
<td>20.0</td>
<td>18.7</td>
<td>18.7</td>
<td>9.8</td>
</tr>
<tr>
<td>ZnO</td>
<td>30</td>
<td>5 - 10</td>
<td>10.0</td>
<td>9.4</td>
<td>9.4</td>
<td>14.7</td>
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<td>CaC03</td>
<td>2</td>
<td>3 - 5</td>
<td>5.0</td>
<td>4.7</td>
<td>4.7</td>
<td>0</td>
</tr>
<tr>
<td>Dichlofluanide</td>
<td>-</td>
<td>1 - 3</td>
<td>3.0</td>
<td>2.8</td>
<td>2.8</td>
<td>0</td>
</tr>
<tr>
<td>China clay</td>
<td>-</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>14.7</td>
<td></td>
</tr>
<tr>
<td>Econea</td>
<td>-</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>3.9</td>
<td></td>
</tr>
<tr>
<td>Zinc Pyrithione</td>
<td>-</td>
<td>1 - 3</td>
<td>3.0</td>
<td>2.8</td>
<td>2.8</td>
<td>3.9</td>
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<tr>
<td>CuSCN</td>
<td>25</td>
<td>10 - 15</td>
<td>15.0</td>
<td>14.1</td>
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<tr>
<td>Crayvallac</td>
<td>-</td>
<td>0.5 - 1.5</td>
<td>1.0</td>
<td>0.9</td>
<td>0.9</td>
<td>1.0</td>
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<tr>
<td>Toluene</td>
<td>-</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>11.3</td>
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<tr>
<td>Xylene</td>
<td>-</td>
<td>15 - 30</td>
<td>18.0</td>
<td>21.0</td>
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<td>7.0</td>
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<tr>
<td>MIBK</td>
<td>-</td>
<td>3 - 7</td>
<td>3.0</td>
<td>5.0</td>
<td>5.0</td>
<td>8.0</td>
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<tr>
<td>Total (wt%)</td>
<td>100.0</td>
<td>100.0</td>
<td>100.0</td>
<td>100</td>
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<tr>
<td>Total solids</td>
<td>79.0</td>
<td>74.0</td>
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<tr>
<td>Color</td>
<td>Plum</td>
<td>Grey-white</td>
<td>Grey-white</td>
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<tr>
<td>Film defects?</td>
<td>Yes</td>
<td>Yes</td>
<td>Yes</td>
<td>Yes</td>
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<tr>
<td>Paint study</td>
<td>Yes</td>
<td>Yes</td>
<td>Yes</td>
<td>Yes</td>
<td></td>
<td></td>
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<tr>
<td>In distilled water</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td></td>
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<tr>
<td>Karl Fisher</td>
<td>35 days</td>
<td>wt%- water</td>
<td></td>
<td></td>
<td></td>
<td></td>
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<tr>
<td>------------</td>
<td>---------</td>
<td>------------</td>
<td>----</td>
<td></td>
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<tr>
<td></td>
<td>50 days</td>
<td>wt%- water</td>
<td></td>
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<tr>
<td></td>
<td>180 days</td>
<td>wt%- water</td>
<td>40</td>
<td>34</td>
<td></td>
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<tr>
<td>Max Mn loss</td>
<td>240-360</td>
<td>%</td>
<td>26</td>
<td>34</td>
<td></td>
<td></td>
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<tr>
<td>Paint erosion</td>
<td>over 280</td>
<td>μηι</td>
<td>40</td>
<td>40</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

In seawater on raft ↓

<table>
<thead>
<tr>
<th>Raft performance</th>
<th>After 7 months</th>
<th>Erosion (relative)</th>
<th>+</th>
<th>++</th>
<th>++ (3 months)</th>
<th>++</th>
</tr>
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<td>Efficacy against</td>
<td></td>
<td>Slime/ biofilm</td>
<td>+</td>
<td>++</td>
<td>++ (3 months)</td>
<td>+</td>
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<tr>
<td>Efficacy against</td>
<td></td>
<td>Macro-fouling</td>
<td>+</td>
<td>++</td>
<td>++ (3 months)</td>
<td>+</td>
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Remarks: Dispersion time: 30 min at 1500 rpm
$erosion stops after 150 days
EG = example

As is clearly shown in Table 2, the self-polishing erosion rate is greatly increased by using the coating composition and binder according to the present invention.

Furthermore, the antifouling rate is either maintained or enhanced by using the coating composition and binder according to the present invention.

As can be seen from the above examples, a coating composition and binder according to the present invention is particularly suitable for marine self-polishing (erodible) and/or antifouling coatings and offers particular advantages with regard to self-polishing rates, resistance to cracking and effective biocidal activity.

Attention is directed to all papers and documents which are filed concurrently with or previous to this specification in connection with this application and which are open to public inspection with
this specification, and the contents of all such papers and documents are incorporated herein by reference.

All of the features disclosed in this specification (including any accompanying claims, abstract and drawings), and/or all of the steps of any method or process so disclosed, may be combined in any combination, except combinations where at least some of such features and/or steps are mutually exclusive.

Each feature disclosed in this specification (including any accompanying claims, abstract and drawings) may be replaced by alternative features serving the same, equivalent or similar purpose, unless expressly stated otherwise. Thus, unless expressly stated otherwise, each feature disclosed is one example only of a generic series of equivalent or similar features.

The invention is not restricted to the details of the foregoing embodiment(s). The invention extends to any novel one, or any novel combination, of the features disclosed in this specification (including any accompanying claims, abstract and drawings), or to any novel one, or any novel combination, of the steps of any method or process so disclosed.

As used herein, unless otherwise expressly specified, all numbers such as those expressing values, ranges, amounts or percentages may be read as if prefaced by the word "about", even if the term does not expressly appear. Any numerical range recited herein is intended to include all sub-ranges subsumed therein. Plural encompasses singular and vice versa. When ranges are given, any endpoints of those ranges and/or numbers within those ranges can be combined within the scope of the present invention. Including and like terms means "including but not limited to". Similarly, as used herein, the terms "on", "applied on/over", "formed on/over", "deposited on/over", "overlay" and "provided on/over" mean formed, overlay, deposited, or provided on but not necessarily in contact with the surface. For example, a coating layer "formed over" a substrate does not preclude the presence of one or more other coating layers of the same or different composition located between the formed coating layer and the substrate. Including, for example, and like terms means including but not limited to, for example, but not limited to, and the like.
CLAIMS

1. A self-polishing and/or antifouling coating composition comprising a binder, wherein the binder comprises a copolymer comprising (i) units of A and (ii) units of B and/or C:

   ![Chemical structure A](image)
   ![Chemical structure B](image)
   ![Chemical structure C](image)

   wherein \( R_{-1} \) and \( R_2 \) are each independently H, a \( \text{Cl-C}_6 \) alkyl group or phenyl, \( R_3 \) is H, a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a \( \text{C}_1-3 \) alkyne group, \( R'_3 \) is H, a \( \text{C}_1-6 \) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a \( \text{d}^{\text{a}} \) alkyne group.

2. A self-polishing and/or antifouling coating composition according to claim 1, wherein the copolymer comprising units of A and B:

   ![Chemical structure A](image)
   ![Chemical structure B](image)

   wherein \( R_{-1} \) and \( R_2 \) are each independently H, a \( \text{Cl-C}_6 \) alkyl group or phenyl, \( R_3 \) is H, a \( \text{Ci-C}_6 \) alkyl group or phenyl and \( R_4 \) is optional and, when present, is a \( \text{C}_{1-3} \) alkyne group.

3. A self-polishing and/or antifouling coating composition according to claim 2, wherein the copolymer comprises blocks of A and blocks of B.
4. A self polishing and/or antifouling coating composition according to claim 3, wherein the copolymer is a tri-block copolymer.

5. A self polishing and/or antifouling coating composition according to any of claims 1 to 4, wherein \( R_1 \) and \( R_2 \) are both methyl.

6. A self polishing and/or antifouling coating composition according to any of claims 1 to 5, wherein \( R_3 \) is methyl and \( R_4 \) is not present.

7. A self polishing and/or antifouling coating composition according to claim 6 wherein the unit B represents poly(D,L-lactide).

8. A self polishing and/or antifouling coating composition according to any of claims 1 to 5, wherein \( R_3 \) is H and \( R_4 \) is not present.

9. A method of reducing and/or preventing the adhesion of marine organisms on at least a portion of a surface, the method comprising the steps of applying a coating composition according to any of claims 1 to 8 to at least a portion of the surface.

10. Use of a coating composition according to any of claims 1 to 8 in reducing and/or preventing the adhesion of marine organisms on at least a portion of a surface to which the coating composition is applied.

11. A marine vessel or marine structure coated on at least a portion thereof with a coating composition according to any of claims 1 to 8.

12. Use of a copolymer comprising (i) units of A and (ii) units of B and/or C:

\[
\begin{align*}
&A \quad \text{B} \quad \text{C} \\
&\begin{array}{c}
R_1 \\
\text{Si} \\
R_2 \\
\end{array} \quad \begin{array}{c}
R_3 \\
\text{O} \\
R_4 \\
\end{array} \quad \begin{array}{c}
R' \quad \text{R'} \\
\text{O} \\
R' \quad \text{R'} \\
\end{array}
\end{align*}
\]
wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_3 \) is \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a d-3 alkylene group, \( R'_3 \) is \( H \), a d-C\(_6\) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a C\(_{1-3}\) alkylene group, as a binder in a self polishing and/or antifouling coating composition

13. Use of a binder in a marine self polishing and/or antifouling coating composition, wherein the binder comprises a copolymer comprising (i) units of \( A \) and (ii) units of \( B \) and/or \( C \):

![Chemical Structures](image)

wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_3 \) is \( H \), a \( \text{Ci-C}_6 \) alkyl group or phenyl, \( R_4 \) is optional and, when present, is a d-3 alkylene group, \( R'_3 \) is \( H \), a d-C\(_6\) alkyl group or phenyl, and \( R'_4 \) is optional and, when present, is a d-3 alkylene group.

14. Use according to claim 12 or 13, wherein the copolymer comprising units of \( A \) and \( B \):

![Chemical Structures](image)

wherein \( R_1 \) and \( R_2 \) are each independently \( H \), a d-C\(_6\) alkyl group or phenyl, \( R_3 \) is \( H \), a d-C\(_6\) alkyl group or phenyl and \( R_4 \) is optional and, when present, is a d-3 alkylene group.

15. Use according to claim 14, wherein the copolymer comprises blocks of \( A \) and blocks of \( B \).
INTERNATIONAL SEARCH REPORT

A. CLASSIFICATION OF SUBJECT MATTER

INV. C08G77/445 C08G63/08 C08G63/695 C09D167/04 C08L67/04

ADD.

According to International Patent Classification (IPC) or to both national classification and IPC.

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

C08G C09D C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched.

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

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See patent family annex.

* Special categories of cited documents:
  *A* document defining the general state of the art which is not considered to be of particular relevance
  *E* earlier application or patent but published on or after the international filing date
  *L* documents which may throw doubts on priority claim(s) or on which the international search was carried out
  *O* document referring to an oral disclosure, use, exhibition or other special reason (as specified)
  *P* document published prior to the international filing date but later than the priority date claimed

*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

*X* document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

*Y* document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

*Z* document member of the same patent family

Date of the actual completion of the international search

10 April 2015

Date of mailing of the international search report

23/04/2015

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
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Tel. (+31-70) 340-2040,
Fax: (+31-70) 340-3016

Authorized officer

Kolitz, Roderich

Form PCT/ISA210 (second sheet) (April 2005)
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