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(54) **Title:** SOLAR CELL MODULES WITH IMPROVED BACKSHEET

(57) **Abstract:** Disclosed is a polyamide-ionomer composition suitable for use in a backsheets in a photovoltaic module comprising a polymer component a polyamide and an anhydride ionomer comprising a copolymer of ethylene, an alpha, beta unsaturated C<sub>3</sub>-C<sub>8</sub> carboxylic acid and an ethylenically unsaturated dicarboxylic acid or derivative thereof selected from the group consisting of maleic acid, fumaric acid, itaconic acid, maleic anhydride, and a C<sub>1</sub>-C<sub>1</sub> alkyl half ester of maleic acid, wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts of one or more alkali metal, transition metal, or alkaline earth metal cations; 0 to 20 weight % of pigment; and 0 to 40 weight % of filler; preferably wherein the combination of pigment and filler comprises 10 to 50 weight % of the composition; and 0 to 5 weight % of weatherability additives.



TITLE

Solar Cell Modules with Improved Backsheet

FIELD OF THE INVENTION

This invention relates to photovoltaic solar cell modules having an improved backsheet.

BACKGROUND OF THE INVENTION

A common form of solar cell module or photovoltaic (PV) module is made by interconnecting individually formed and separate solar cells, e.g., crystalline silicon solar cell, and then mechanically supporting and protecting the cells against environmental degradation by integrating the cells into a laminated solar cell module. The laminated modules usually comprise a stiff transparent protective front panel or sheet, and a rear panel or sheet typically called a "backsheet" or "backskin".

Interconnected solar cells and an encapsulant are disposed between the front and back sheets so as to form a sandwich arrangement. A necessary requirement of the encapsulant (or at least that portion thereof that extends between the front sides of the cells and the transparent front panel) is that it be transparent to solar radiation. The typical mode of forming the laminated module is to assemble a sandwich comprising in order a transparent panel, e.g., a front panel made of glass or a transparent polymer, a front layer of at least one sheet of encapsulant, an array of solar cells interconnected by electrical conductors (with the front sides of the cells facing the transparent panel), a back layer of at least one sheet of encapsulant, a sheet of scrim to facilitate gas removal during the lamination process, and a backsheet or back panel, and then bonding those components together under heat and pressure using a vacuum-type laminator. The back layer of encapsulant may be transparent or any other color, and prior art modules have been formed using a backsheet consisting of a thermoplastic polymer, glass or some other material.

Although the lamination process seals the several layered components together throughout the full expanse of the module, it is common practice to apply a protective polymeric edge sealant to the module so as to assure that moisture will not penetrate the edge portion of the module. The polymeric edge sealant may be in the form of a strip of tape or a caulking-type compound. Another common practice is to provide the module with a perimeter frame, usually made of a metal like aluminum, to provide mechanical edge protection. Those techniques are disclosed or suggested in U.S. Patent 5,741,370. That patent also discloses the concept of eliminating the back layer of encapsulant and bonding a thermoplastic backskin directly to the interconnected solar cells.

A large number of materials have been used or considered for use as the encapsulant in modules made up of individual silicon solar cells. Until at least around 1995, ethylene vinyl acetate copolymer (commonly known as "EVA") was considered the best encapsulant for modules comprising crystalline silicon solar cells. However, EVA has certain limitations: (1) it decomposes under sunlight, with the result that it discolors and gets progressively darker, and (2) its decomposition releases acetic acid which in turn promotes further degradation, particularly in the presence of oxygen and/or heat.

U.S. Patent 5,478,402 discloses use of an ionomer as a cell encapsulant substitute for EVA. The use of ionomer as an encapsulant is further disclosed in U.S. Patent 5,741,370. Ionomers are acid copolymers in which a portion of the carboxylic acid groups in the copolymer are neutralized to salts containing metal ions. U.S. Patent 3,264,272 discloses a composition comprising a random copolymer of copolymerized units of an alpha-olefin having from two to ten carbon atoms, an alpha, beta-ethylenically-unsaturated carboxylic acid having from three to eight carbon atoms in which 10 to 90 percent of the acid groups are neutralized to salts with metal ions from Groups I, II, or III of the Periodic Table, notably, sodium, zinc, lithium, or magnesium, and an optional third mono-ethylenically unsaturated comonomer such as methyl methacrylate or butyl acrylate.

It is known to use a rear panel or backsheet that is made of the same material as the front panel, but a preferred and common practice is to make it of a different material, preferably a material that weighs substantially less than glass, such as a polyvinyl fluoride polymer available under the tradename Tedlar® from E.I. Du Pont de Nemours Co. (DuPont). A widely used backsheet material is a Tedlar®/polyester/ethylene vinyl acetate laminate. Another common backsheet uses a trilayer structure of Tedlar®/Polyester/Tedlar®, also called TPT™, described in WO 94/22172. This structure allows the fluoropolymer to protect

both sides of the polyester from photo-degradation. However, Tedlar® and Tedlar® laminates are not totally impervious to moisture, and as a consequence over time the power output and/or the useful life of modules made with this kind of backsheet material is reduced due to electrical shorting resulting from absorbed moisture.

Due to the price and the supply concern, the PV industry has been gradually evaluating new alternatives, such as backsheets derived from PET, polyamides, etc. For example, WO 2008/138021 discloses PV modules with backsheets based on polyamides derived from linear and/or branched aliphatic and/or cycloaliphatic monomers, which have an average of at least 8 and most 17 carbon atoms, such as nylon 12. However, polyamides are semi-crystalline polymers with high degree of crystallinity, which can lead to brittleness, low flexibility and excessive shrinkage. High moisture absorption is especially a problem for nylon-6 and nylon-66, the most inexpensive polyamides. Water absorption causes dimensional instability, poor weatherability, and, most importantly, reduces insulation capability. While nylon-11 and nylon-12 have better moisture resistance and weatherability, the melting temperature may be too low for use in some lamination processes of PV module assembly.

U.S. Patent 5,741,370 discloses using as the backskin material a thermoplastic olefin comprising a combination of two different ionomers, e.g., a sodium ionomer and a zinc second ionomer, with that combination being described as producing a synergistic effect which improves the water vapor barrier property of the backskin material over and above the barrier property of either of the individual ionomer components. The patent also discloses use of an ionomer encapsulant with the dual ionomer backskin.

It is known that thermoplastic blends or alloys based on ionomers and polyamides have a combination of desirable properties (see U.S. Patents 4,174,358, 5,688,868, 5,866,658, 6,399,684, 6,569,947, 6,756,443 and 7,144,938, 7,592,056, 8,057,910, 8,062,757 and 8,119,235). For example, U.S. Patent 5,866,658 discloses a blend of an ionomer dispersed in a continuous or co-continuous polyamide phase in the range of 60/40 weight % to 40/60 weight % used for molded parts exhibiting toughness, high gloss, abrasion/scratch resistance, and high temperature properties. U.S. Patent 6,399,684 discloses similar blends also containing phosphorous salts such as a hypophosphite salt. See also U.S. Patent Applications 2002/0055006, 2005/007462, 2006/0142489, 2008/0161503, 2009/0298372, 2013/0167966, 2013/0171390, 2013/0171394, 2013/0172470 and 2013/0172488.

U.S. Patents 5,700,890, 5,859,137, 7,267,884 and U.S. Patent Application Publications 2005/0020762A1, and 2006/0142489A1 disclose polyamides toughened with ionomers of ethylene copolymers containing a monocarboxylic acid and a dicarboxylic acid or derivative thereof. U.S. Patent Application Publication 2011/0020573 discloses a blend comprising a polyamide, an ionomer of an ethylene copolymer containing a monocarboxylic acid and a dicarboxylic acid or derivative thereof, and a sulfonamide. U.S. Patent 8,586,663 discloses a blend comprising a polyamide, an ionomer of an ethylene copolymer containing a monocarboxylic acid and a dicarboxylic acid or derivative thereof, and a second ionomer. U.S. Patent 7,592,056 discloses blends of polyamides with mixed ion ionomers, including zinc and sodium mixtures.

U.S. Patent 6,660,930 discloses photovoltaic modules comprising backskins comprising a nylon/ionomer alloy.

Photovoltaic modules can be assessed for moisture permeation and weatherability by cyclic treatment with high moisture and temperature and cold temperature in standardized "stress tests". It is desirable to provide PV modules that are capable of withstanding such stress tests for substantially more than 1000 hours. Thus, it also is desirable to provide backsheet materials that provide PV modules that are capable of withstanding such stress tests.

#### SUMMARY OF THE INVENTION

This invention relates to a polyamide-ionomer composition suitable for use in a backsheet in a photovoltaic module.

The polyamide-ionomer blend composition comprises, or consists essentially of

- (i) A polymer component comprising, or consisting essentially of
  - 1) 53 to 64 weight %, based on the combination of (1) and (2), of a polyamide;

2) 36 to 47 weight %, based on the combination of (1) and (2), of an anhydride ionomer comprising a copolymer of

- (a) ethylene;
- (b) from 5 weight % to 15 weight % of an alpha, beta-unsaturated C<sub>3</sub>-C<sub>8</sub> carboxylic acid;
- (c) from 0.5 weight % to 12 weight % of at least one comonomer that is an ethylenically unsaturated dicarboxylic acid or derivative thereof selected from the group consisting of maleic acid, fumaric acid, itaconic acid, maleic anhydride, and a C<sub>1</sub>-C<sub>4</sub> alkyl half ester of maleic acid; and
- (d) from 0 weight % to 30 weight % of monomers selected from alkyl acrylate and alkyl methacrylate, wherein the alkyl groups have from one to twelve carbon atoms; wherein the carboxylic acid functionalities present are at least partially neutralized by one or more alkali metal, transition metal, or alkaline earth metal cations;
- (ii) 0 to 20 weight % of pigment; and
- (iii) 0 to 40 weight % of filler; preferably wherein the combination of (ii) and (iii) comprises 8 to 50 weight % of the combination of (i), (ii), (iii) and (iv); and
- (iv) 0 to 5 weight % of additives selected from oxidation inhibitors, UV stabilizers, and hindered amine light stabilizers.

The invention provides a backsheet for a photovoltaic module comprising or consisting essentially of the composition described above.

The invention also provides a laminated solar cell module comprising or consisting essentially of a front support layer formed of a light transmitting material and having first and second surfaces; a plurality of interconnected solar cells having a first surface facing the front support layer and a second surface facing away from the front support layer; a transparent encapsulant surrounding and encapsulating the interconnected solar cells, the transparent encapsulant being bonded to the second surface of the front support layer; and a backsheet as described above wherein one surface of the backsheet is bonded to the second surface of the transparent encapsulant.

The invention also provides an assembly for conversion under heat and pressure into a laminated solar cell module, the assembly comprising a front support layer formed of a light transmitting material and having front and back surfaces; a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; a plurality of interconnected solar cells having first and second surfaces adjacent to the first transparent encapsulant layer; a second transparent thermoplastic encapsulant layer disposed adjacent to the solar cells in parallel relation to the first transparent encapsulant layer; and a thermoplastic backsheet as described above.

The invention also provides a method of manufacturing a solar cell module comprising providing a front support layer formed of a light transmitting material and having front and back surfaces; placing a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; positioning a plurality of interconnected solar cells having first and second surfaces so that the first surfaces thereof are adjacent to the first transparent encapsulant layer; placing a second transparent thermoplastic encapsulant layer adjacent to the second surfaces of the solar cells; placing a backsheet as described above adjacent to the second transparent thermoplastic encapsulant layer to thereby form an assembly; subjecting the assembly to heat and pressure so as to melt the encapsulant layers and cause the encapsulant to surround the solar cells, and cooling the assembly so as to cause the encapsulant to solidify and bond to the front support layer, the solar cells and the backsheet, thereby laminating the layers and the solar cells together to form an integrated solar cell module.

#### DETAILED DESCRIPTION OF THE INVENTION

As used herein, the terms “comprises,” “comprising,” “includes,” “including,” “has,” “having” or any other variation thereof, are intended to cover a non-exclusive inclusion. For example, a process, method, article, or apparatus that comprises a

list of elements is not necessarily limited to only those elements but may include other elements not expressly listed or inherent to such process, method, article, or apparatus. Further, unless expressly stated to the contrary, “or” refers to an inclusive or and not to an exclusive or. For example, a condition A or B is satisfied by any one of the following: A is true (or present) and B is false (or not present), A is false (or not present) and B is true (or present), and both A and B are true (or present). As used herein, the terms “a” and “an” include the concepts of “at least one” and “one or more than one”.

Unless stated otherwise, all percentages, parts, ratios, etc., are by weight. Further, when an amount, concentration, or other value or parameter is given as either a range, preferred range or a list of upper preferable values and lower preferable values, this is to be understood as specifically disclosing all ranges formed from any pair of any upper range limit or preferred value and any lower range limit or preferred value, regardless of whether ranges are separately disclosed. Where a range of numerical values is recited herein, unless otherwise stated, the range is intended to include the endpoints thereof, and all integers and fractions within the range. It is not intended that the scope of the invention be limited to the specific values recited when defining a range. When a component is indicated as present in a range starting from 0, such component is an optional component (i.e., it may or may not be present). When present an optional component may be at least 0.1 weight % of the composition or copolymer.

When materials, methods, or machinery are described herein with the term “known to those of skill in the art”, “conventional” or a synonymous word or phrase, the term signifies that materials, methods, and machinery that are conventional at the time of filing the present application are encompassed by this description. Also encompassed are materials, methods, and machinery that are not presently conventional, but that may have become recognized in the art as suitable for a similar purpose.

As used herein, the term “copolymer” refers to polymers comprising copolymerized units resulting from copolymerization of two or more comonomers and may be described with reference to its constituent comonomers or to the amounts of its constituent comonomers such as, for example “a copolymer comprising ethylene and 15 weight % of acrylic acid”. A description of a copolymer with reference to its constituent comonomers or to the amounts of its constituent comonomers means that the copolymer contains copolymerized units (in the specified amounts when specified) of the specified comonomers.

In this application, the terms “sheet”, “layer” and “film” are used in their broad sense interchangeably to describe articles wherein the compositions are processed into generally planar forms, either monolayer or multilayer. The processing method and/or the thickness may influence whether the term “sheet” or “film” is used herein, but either term can be used to describe such generally planar articles.

A “frontsheet” is a sheet, layer or film positioned as the outermost layer on the side of a photovoltaic module that faces a light source and may also be described as an incident layer. Because of its location, it is generally desirable that the frontsheet has high transparency to the desired incident light to allow efficient transmission of sunlight into the solar cells. It is also desirable that the frontsheet has high moisture barrier properties to prevent entry of moisture into the photovoltaic module. Such moisture intrusion can degrade the photovoltaic module components and/or reduce the electrical efficiency of the module.

A “backsheet” is a sheet, layer or film on the side of a photovoltaic module that faces away from a light source, and is often opaque. In some instances, it may be desirable to receive light from both sides of a device (e.g. a bifacial device), in which case a module may have transparent layers on both sides of the device.

“Encapsulant” layers are layers used to encase the fragile voltage-generating solar cell layer to protect it from damage and hold it in place in the photovoltaic module and are normally positioned between the solar cell layer and the incident layer and the backing layer. Suitable polymer materials for these encapsulant layers typically possess a combination of characteristics such as high transparency, high impact resistance, high penetration resistance, high moisture resistance, good ultraviolet (UV) light resistance, good long term thermal stability, adequate adhesion strength to frontsheets, backsheets, and other rigid polymeric sheets and cell surfaces, and good long term weatherability.

This invention involves the use of a polyamide-ionomer alloy in sheet form as a backsheet material. As used herein, the term "alloy" is used to describe a polymer blend that forms a distinct polymer substance. Various polyamide-ionomer alloys are available. The composition limits as defined in the Summary of the Invention above provide that the fraction of polyamide is sufficient to ensure the polyamide remains the continuous phase during alloying and subsequent converting to film and the fraction of the ionomer is sufficient to ensure adequate toughness as made and after aging as well as good adhesion to the encapsulant.

Thermoplastic resins are polymeric materials that can flow when heated under pressure. Melt index (MI) is the mass rate of flow of a polymer through a specified capillary under controlled conditions of temperature and pressure. It is typically measured according to ASTM 1238.

This invention provides a polymeric blend that is a marriage of a polyamide such as nylon 6 and an ionomer selected from a special family of ionomers (denoted anhydride ionomers) to provide materials that are highly suitable for polymeric backsheets for PV modules. In essence, the new materials overcome some of the major deficiencies of both polyamides and ionomers, while continuing to retain most of the desirable attributes. As indicated above, the ionomers used in this invention are selected from a family of ionomers containing dicarboxylic acid moieties, or derivatives thereof. As used herein, the term "anhydride ionomer" is used to describe an ionomer that includes dicarboxylic acid moieties, derivatives thereof such as anhydrides or other known carboxylic acid derivatives. The presence of dicarboxylic acid moieties in the ionomers enhances the compatibility with polyamides, particularly at higher levels, and provides blends with very good toughness, low temperature impact strength and resistance to hydrolytic delamination. Higher amounts of dicarboxylic acid moieties provide two unique features to blends of such ionomers and a polyamide, such as nylon 6. First, the anhydride ionomer is dispersed in the polyamide in extremely fine particles and second, the particle size distribution is very narrow.

As indicated above, this invention provides a backsheet for a PV module comprising a thermoplastic composition comprising a polyamide and an ionomeric composition comprising a copolymer of ethylene, an  $\alpha,\beta$ -unsaturated  $C_3$ - $C_8$  carboxylic acid, at least one comonomer that is an ethylenically unsaturated dicarboxylic acid or derivative thereof, and optionally at least one comonomer selected from alkyl acrylate and alkyl methacrylate.

Ionomeric resins ("ionomers") are ionic copolymers of an olefin such as ethylene (E) with a metal salt of an unsaturated carboxylic acid, such as acrylic acid (AA), methacrylic acid (MAA), and/or other acids, and optionally softening comonomers. At least one alkali metal, transition metal, or alkaline earth metal cation, such as lithium, sodium, potassium, magnesium, calcium, or zinc, or a combination of such cations, is used to neutralize some portion of the acidic groups in the copolymer resulting in a thermoplastic resin exhibiting enhanced properties. For example, a copolymer of ethylene and acrylic acid can then be at least partially neutralized to salts comprising one or more alkali metal, transition metal, or alkaline earth metal cations to form an ionomer. Copolymers can also be made from an olefin such as ethylene, an unsaturated carboxylic acid and other comonomers such as alkyl (meth)acrylates providing "softer" resins that can be neutralized to form softer ionomers.

The ionomers useful in this invention consist of a family of ionomers containing dicarboxylic acid moieties that can be derived from ethylenically unsaturated derivatives of dicarboxylic acid comonomers, such as maleic anhydride and ethyl hydrogen maleate, at least partially neutralized by one or more alkali metal, transition metal, or alkaline earth metal cations (denoted as anhydride ionomers). They are copolymers of ethylene, an  $\alpha,\beta$ -unsaturated  $C_3$ - $C_8$  carboxylic acid and at least one comonomer that is an ethylenically unsaturated dicarboxylic acid at an amount of from 0.5 weight % to 12 weight %, alternatively from 3 weight % to 12 weight %. Preferably, the dicarboxylic acid comonomer(s) are present in an amount from 4 weight % to 10 weight %. The unsaturated dicarboxylic acid comonomers or their derivatives can be selected from, for example, maleic anhydride (MAH), ethyl hydrogen maleate (also known as maleic acid monoethylester – MAME), and itaconic acid (ITA). More preferably, a copolymer comprises from 4 to 8 weight % of maleic acid monomethylester comonomer in an

ethylene/methacrylic acid/maleic acid monomethylester copolymer wherein from 20 to 70 percent of the total acid groups in the copolymer are neutralized to provide carboxylate salts containing alkali metal, transition metal, or alkaline earth metal cations.

Some non-neutralized ethylene acid copolymers comprising lower amounts of ethylenically unsaturated dicarboxylic acid comonomers are known (see U.S. Patent 5,902,869), as are their ionomeric derivatives (see U.S. Patent 5,700,890).

As indicated above, comonomers such as alkyl (meth)acrylates can be included in the ethylene acid copolymer to form a copolymer that can be neutralized to provide carboxylate salts with alkali metal, alkaline earth metal or transition metal cations. Preferred are comonomers selected from alkyl acrylate and alkyl methacrylate wherein the alkyl groups have from 1 to 8 carbon atoms, and more preferred are comonomers selected from methyl acrylate, ethyl acrylate, *iso*-butyl acrylate (iBA), and *n*-butyl acrylate (nBA). The alkyl (meth)acrylates are optionally included in amounts from 0 to 30 weight % alkyl (meth)acrylate such as 0.1 to 30 weight % when present and preferably from 0.1 to 15 weight % of the copolymer.

Examples of copolymers useful in this invention include copolymers of ethylene, methacrylic acid and ethyl hydrogen maleate (E/MAA/MAME) and copolymers of ethylene, acrylic acid and maleic anhydride (E/AA/MAH).

Neutralization of the ethylene acid copolymer can be effected by first making the ethylene acid copolymer and treating the copolymer with inorganic base(s) with alkali metal, alkaline earth metal or transition metal cation(s). The copolymer can be from 10 to 99.5 % neutralized with at least one metal ion selected from lithium, sodium, potassium, magnesium, calcium, barium, lead, tin, zinc, aluminum; or combinations of such cations. Neutralization may be from 10 to 70 %. Preferably the copolymer has from 20 %, alternatively from 35 %, to 70 % of the available carboxylic acid groups ionized by neutralization with at least one metal ion selected from sodium, zinc, lithium, magnesium, and calcium; and more preferably zinc or sodium. Notably the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising zinc or sodium, preferably zinc. Of particular note is an anhydride ionomer comprising zinc as the neutralizing cation.

Mixed metal ionomers may provide a combination of better properties to the blends with polyamides than ionomers comprising a single type of cation. For example, a zinc/sodium mixed ion ionomer blended with polyamide may provide lower water sorption, better scratch resistance and better processing capability than those provided by a corresponding ionomer containing only an alkali metal such as sodium. The zinc/sodium mixed ion ionomer may also provide higher hardness and higher mechanical strength than provided by a corresponding ionomer containing only zinc. Mixed ion ionomers are conveniently prepared by blending an ionomer composition with a single cation, such as a zinc-containing ionomer, with an ionomer with a different cation, such as a sodium-containing ionomer. Alternatively, mixed ion ionomers may be prepared by neutralizing an acid copolymer with different neutralizing agents, either sequentially or simultaneously.

Methods for preparing ionomers from copolymers are well known in the art.

Of particular note are blends of polyamides and anhydride ionomers further comprising conventional ionomers prepared from acid copolymers not containing a dicarboxylic acid or derivative. Accordingly, compositions of this invention include blends of polyamide with component (2) further comprising in combination with the anhydride ionomer one or more conventional ionomer comprising an E/X/Y copolymer where E is ethylene, X is a C<sub>3</sub> to C<sub>8</sub>  $\alpha,\beta$ -ethylenically unsaturated monocarboxylic acid, and Y is a comonomer selected from alkyl acrylate and alkyl methacrylate wherein the alkyl groups have from 1 to 8 carbon atoms, wherein X is present in from 2 to 30 weight % of the E/X/Y copolymer, Y is present from 0 to 40 weight % of the E/X/Y copolymer, wherein the carboxylic acid functionalities present are at least partially neutralized by one or more alkali metal, transition metal, or alkaline earth metal cations. Preferred  $\alpha,\beta$ -ethylenically unsaturated monocarboxylic acids include acrylic acid and methacrylic acid. Non-limiting, illustrative examples of conventional ionomers include E/15MAA/Na, E/19MAA/Na, E/15AA/Na, E/19AA/Na, E/15MAA/Mg, E/19MAA/Li, and E/15MAA/60Zn (wherein E represents ethylene, MAA represents methacrylic acid, AA represents acrylic acid, the numbers represents either the weight % of comonomer(s) present in the copolymer or the amount of neutralization of the available carboxylic acid groups, and the atomic symbol represents the neutralizing cation).

Depending on the need of a particular application, the amount of such conventional ionomer or mixture of conventional ionomers in combination with the anhydride ionomer in component (2) can be manipulated to provide an appropriate balance of toughness, low temperature impact strength and resistance to hydrolytic delamination. For example, highly toughened polyamide compositions can be achieved by using relatively larger amounts of conventional ionomers with smaller amounts of anhydride ionomers (for example, 30 weight % of conventional ionomer and 6 weight % of anhydride ionomer). Toughened polyamide films can be prepared using relatively larger amounts of anhydride ionomers with smaller amounts of conventional ionomers (for example, 30 weight % of anhydride ionomer and 6 weight % of conventional ionomer). Of note are modifier blends comprising equal amounts of anhydride ionomer and conventional ionomer (for example, 18 weight % of anhydride ionomer and 18 weight % of conventional ionomer). When a conventional ionomer is blended with an anhydride ionomer, the blend desirably contains combined amounts of each of the comonomers within the ranges described in the Summary of the Invention for the ionomeric copolymer.

Notably, when the polyamide is poly(caprolactam) (nylon-6), the ionomer component consists of an anhydride ionomer.

The polyamide-ionomer blend may comprise, consist essentially of, consist of, or be produced from, a polyamide in an amount from a lower limit of 53, 58 or 60 weight % to an upper limit of 64 weight % and an ionomer in an amount from a lower limit of 36 or 40 weight % to an upper limit of 42, or 47 weight %, all based on the weight of the combination of polyamide and ionomer.

Polyamides (abbreviated herein as PA), also referred to as nylons, are condensation products of one or more dicarboxylic acids and one or more diamines, and/or one or more aminocarboxylic acids such as 11-aminododecanoic acid, and/or ring-opening polymerization products of one or more cyclic lactams such as caprolactam and laurolactam. Polyamides may be fully aliphatic or semiaromatic.

Polyamides from single reactants such as lactams or amino acids, referred to as AB type polyamides are disclosed in *Nylon Plastics* (edited by Melvin L. Kohan, 1973, John Wiley and Sons, Inc.) and include nylon-6, nylon-11, nylon-12, or combinations of two or more thereof. Polyamides prepared from more than one lactam or amino acid include nylon-6,12.

Other well-known polyamides useful in the composition include those prepared from condensation of diamines and diacids, referred to as AABB type polyamides (including nylon-66, nylon-610, nylon-612, nylon-1010, and nylon-1212), as well as from a combination of lactams, diamines and diacids such as nylon-6/66, nylon-6/610, nylon-6/66/610, nylon-66/610, or combinations of two or more thereof.

Fully aliphatic polyamides used in the composition are formed from aliphatic and alicyclic monomers such as diamines, dicarboxylic acids, lactams, aminocarboxylic acids, and their reactive equivalents. In this context, the term "fully aliphatic polyamide" also refers to copolymers derived from two or more such monomers and blends of two or more fully aliphatic polyamides. Linear, branched, and cyclic monomers may be used.

Carboxylic acid monomers comprised in the fully aliphatic polyamides include, but are not limited to aliphatic dicarboxylic acids, such as for example adipic acid (C6), pimelic acid (C7), suberic acid (C8), azelaic acid (C9), decanedioic acid (C10), dodecanedioic acid (C12), tridecanedioic acid (C13), tetradecanedioic acid (C14), and pentadecanedioic acid (C15). Diamines can be chosen among diamines with four or more carbon atoms, including but not limited to tetramethylene diamine, hexamethylene diamine, octamethylene diamine, decamethylene diamine, dodecamethylene diamine, 2-methylpentamethylene diamine, 2-ethyltetramethylene diamine, 2-methyloctamethylenediamine; trimethylhexamethylenediamine, meta-xylylene diamine, and/or mixtures thereof.

Semi-aromatic polyamides include a homopolymer, a copolymer, a terpolymer or more advanced polymers formed from monomers containing aromatic groups. One or more aromatic carboxylic acids may be terephthalic acid or a mixture of terephthalic acid with one or more other carboxylic acids, such as isophthalic acid, phthalic acid, 2-methyl terephthalic acid and



naphthalic acid. In addition, the one or more aromatic carboxylic acids may be mixed with one or more aliphatic dicarboxylic acids, as disclosed above. Alternatively, an aromatic diamine such as meta-xylylene diamine (MXD) can be used to provide a semi-aromatic polyamide, an example of which is MXD6, a homopolymer comprising MXD and adipic acid.

Preferred polyamides disclosed herein are homopolymers or copolymers wherein the term copolymer refers to polyamides that have two or more amide and/or diamide molecular repeat units. The homopolymers and copolymers are identified by their respective repeat units. For copolymers disclosed herein, the repeat units are listed in decreasing order of mole % repeat units present in the copolymer. The following list exemplifies the abbreviations used to identify monomers and repeat units in the homopolymer and copolymer polyamides:

HMD	hexamethylene diamine (or 6 when used in combination with a diacid)
T	Terephthalic acid
6	$\epsilon$ -Caprolactam
AA	Adipic acid
DDA	Decanedioic acid
DDDA	Dodecanedioic acid
I	Isophthalic acid
MXD	meta-xylylene diamine
TMD	1,4-tetramethylene diamine
6T	polymer repeat unit formed from HMD and T
MXD6	polymer repeat unit formed from MXD and AA
66	polymer repeat unit formed from HMD and AA
610	polymer repeat unit formed from HMD and DDA
612	polymer repeat unit formed from HMD and DDDA
11	polymer repeat unit formed from 11-aminoundecanoic acid
12	polymer repeat unit formed from 12-aminododecanoic acid

In the art the term “6” when used alone designates a polymer repeat unit formed from  $\epsilon$ -caprolactam. Alternatively “6” when used in combination with a diacid such as T, for instance 6T, the “6” refers to HMD. In repeat units comprising a diamine and diacid, the diamine is designated first. Furthermore, when “6” is used in combination with a diamine, for instance 66, the first “6” refers to the diamine HMD, and the second “6” refers to adipic acid. Likewise, repeat units derived from other amino acids or lactams are designated as single numbers designating the number of carbon atoms.

In various embodiments the polyamide comprises one or more polyamides selected from among the following groups (wherein PA is shorthand for polyamide or “nylon”):

Group I polyamides have a melting point of less than 210 °C, and comprise an aliphatic or semiaromatic polyamide such as poly(hexamethylene dodecanediamide/hexamethylene terephthalamide) (PA612/6T). See PCT Patent Application Publication WO2011/94542. Group I polyamides may have semiaromatic repeat units to the extent that the melting point is less than 210 °C and generally the semiaromatic polyamides of the group have less than 40 mole percent of semiaromatic repeat units. Semiaromatic repeat units are defined as those derived from monomers selected from one or more of the group consisting of aromatic dicarboxylic acids having 8 to 20 carbon atoms and aliphatic diamines having 4 to 20 carbon atoms. Other notable Group I polyamides include PA6/66, PA6/610, PA6/66/610, PA6/6T, PA1010, PA11 and PA12.

Group II polyamides have a melting point of at least 210 °C and comprise an aliphatic polyamide. Notable Group II polyamides include PA6, PA66, PA610 and PA612. The RV of PA6 is commonly measured according to ISO Test Method 307 using a solution of 1 % of polymer in 96 % sulfuric acid. The RV of PA66 is commonly measured according to ISO Test Method 307 using a solution of 1 % of polymer in 90 % formic acid.

Group III polyamides have a melting point of at least 210 °C and comprise

(aa) 20 to 35 mole percent semiaromatic repeat units derived from one or more monomers selected from

(i) aromatic dicarboxylic acids having 8 to 20 carbon atoms and aliphatic diamines having 4 to 20 carbon atoms;

and

(bb) 65 to 80 mole percent aliphatic repeat units derived from one or more monomers selected from

(ii) an aliphatic dicarboxylic acid having 6 to 20 carbon atoms and an aliphatic diamine having 4 to 20 carbon

atoms; and

(iii) a lactam and/or aminocarboxylic acid having 4 to 20 carbon atoms.

A preferred Group III polyamide is PA66/6T.

Group IV polyamides have a melting point of greater than 230 °C and comprise

(cc) 50 to 95 mole percent semiaromatic repeat units derived from one or more monomers selected from

(i) aromatic dicarboxylic acids having 8 to 20 carbon atoms and aliphatic diamines having 4 to 20 carbon atoms;

and

(dd) 5 to 50 mole percent aliphatic repeat units derived from one or more monomers selected from

(ii) an aliphatic dicarboxylic acid having 6 to 20 carbon atoms and said aliphatic diamine having 4 to 20 carbon

atoms; and

(iii) a lactam and/or aminocarboxylic acid having 4 to 20 carbon atoms.

A preferred Group IV polyamide is PA6T/66.

Group V polyamides have a melting point of at least 260 °C and comprise

(ee) greater than 95 mole percent semiaromatic repeat units derived from one or more monomers selected from

(i) aromatic dicarboxylic acids having 8 to 20 carbon atoms and aliphatic diamines having 4 to 20 carbon atoms;

and

(ff) less than 5 mole percent aliphatic repeat units derived from one or more monomers selected from

(ii) an aliphatic dicarboxylic acid having 6 to 20 carbon atoms and said aliphatic diamine having 4 to 20 carbon

atoms; and

(iii) a lactam and/or aminocarboxylic acid having 4 to 20 carbon atoms.

A preferred Group V Polyamide is PA6T/DT.

Group VI polyamides have no melting point and include poly(hexamethylene isophthalamide/hexamethylene terephthalamide) (PA6I/6T) and poly(hexamethylene isophthalamide/hexamethylene terephthalamide/hexamethylene hexanediamide) (PA6I/6T/66).

In various embodiments the polyamide is a Group I polyamide, Group II polyamide, Group III polyamide, Group IV polyamide, Group V polyamide or Group VI polyamide, respectively.

Preferred polyamides include PA6, PA66, PA610, PA612, PA6/66, PA6/610, PA6/66/610, PA6/6T, PA610/6T, P612/6T, PA1010, PA11, PA12 and combinations thereof. More preferred polyamides include PA6, PA66, PA610, PA612, PA610/6T, PA612/6T, PA1010, PA11, PA12 and combinations thereof, including PA6, PA612, P612/6T or PA12, with PA6 most preferred.

The polyamide component may also be a blend of two or more polyamides. Preferred blends include those selected from the group consisting of Group I and Group II polyamides, Group I and Group III polyamides, Group I and Group VI polyamides, Group II and Group III polyamides, Group II and Group IV polyamides, Group II and Group V polyamides, Group II and Group VI polyamides, Group III and Group VI polyamides, and Group IV and Group V polyamides. A notable blend is a blend of PA612 and PA612/6T, especially in a blend ratio of 1:3 of PA612:PA612/6T. See U.S. Patent Application Publication 2012/0196973.

Polyamides and processes for making them are well known to those skilled in the art, so the disclosure of which is omitted in the interest of brevity.

The polyamide may have a relative viscosity (RV) of 2.5 to 4.0, preferably from 2.6 to 3.5. Relative viscosity is related to melt viscosity. Varied methods may be used for measured RV values, and not all commercial polyamides list the RV values.

RV is determined by comparing the time required for a specific volume of polymer solution to flow through a capillary tube with the corresponding flow time of the same volume of pure solvent. Different solvents may be used, depending on the polyamide of interest. Common solvents include 96 % sulfuric acid and 90 % formic acid. For example, the RV of nylon-6 is measured using 1% in 96% sulfuric acid according to ISO Test Method 307. A similar method for determining RV is according to ASTM D789.

Grades of nylon-6 targeted for extrusion (such as Ultramid® B33 from BASF) with RV of around 3.3 are suitable.

Molding grades of nylon-6 (such as Ultramid® B27 from BASF) with RV of around 2.7 are also suitable for this application.

The polyamide-ionomer blend further contains 0 to 20 weight % of pigment; and 0 to 40 weight % of filler; such that the pigment and/or filler comprises 8 to 50 weight % of the total composition.

As used herein pigments have refractive indices greater than 1.8, preferably greater than 2, and particle size less than 0.5 microns such as 0.2 to 0.4 microns. The compositions may comprise inorganic pigments such as oxide pigments, e.g., titanium dioxide, zinc oxide, and antimony oxide. Other pigments include lithopone, chromomolybdic acid, sulfide selenium compound, ferrocyanide and carbon black pigments. Notably, the pigment comprises titanium dioxide, zinc oxide, or antimony oxide, preferably titanium dioxide. The compositions containing inorganic pigments maintain good flowability and color the molded article even when used in a small amount.

As used herein fillers have refractive indices of 1.6 or less and particle size of 0.8 micron or greater. Suitable fillers include mineral fillers such as inorganic oxides, carbonates, sulfates, silicas, alkali and alkaline earth metal silicates, and barytes of a metal of Groups IA, IIA, IIIA, IIB, VIB or VIII of the periodic table of the elements, including magnesium silicates such as talc ( $\text{Mg}_3\text{Si}_4\text{O}_{10}(\text{OH})_2$ ), wollastonite ( $\text{CaSiO}_3$ ), phyllosilicates (mica) and calcium carbonate. Notably, the filler comprises an inorganic oxide, carbonate, sulfate, silica, alkali and alkaline earth metal silicate, or baryte of a metal of Groups IA, IIA, IIIA, IIB, VIB or VIII of the periodic table of the elements, preferably wherein the filler comprises calcium carbonate, barium sulfate, wollastonite and talc, more preferably talc. Fillers may optionally be coated such as with silane treatments to improve wetting between the filler and the polymer matrix. The shape, size, and size distribution of the filler all impact its effectiveness as filler, though, at high levels, the particular characteristics of the filler become less important. Preferably the filler particles have a ratio of the largest dimension to the smallest dimension greater than 5. Fillers also include glass fibers. Desirably, fillers provide stiffening (improving Young's modulus ASTM D882) and reduce the coefficient of linear thermal expansion (ASTM E831) for the composition while maintaining good elongation to break (ASTM D882). Large particles such as those having a particle size in at least one dimension greater than 200 microns such as mica or glass fibers provide good stiffening but can reduce elongation to break. Particles with at least one dimension less than 20 microns or less than 5 microns are preferred. Talc (plate like fillers) and Wollastonite (rod like fillers) provide for minimizing co-efficient of linear thermal expansion, maximizing stiffness, while still maintaining a reasonable amount of elongation to break. Preferably the fillers are either transparent, such as glass fibers, or white to produce whiter compositions. For example, talc commercially available as Jetfine® 3CA is whiter than Jetfine® 3CC, which when incorporated into the polymer matrix leads to a whiter composition. This is particularly desirable when used in combination with a white pigment such as titanium dioxide.

For example but not limitation, the composition may comprise 8 to 20 weight % of pigment such as  $\text{TiO}_2$  and 0 weight % filler, or the composition may comprise 8 to 40 weight % of filler such as talc and 0 weight % of pigment, or the composition may comprise 10 to 40 weight % of filler such as talc and 8 to 15 weight % of pigment. Notable compositions comprise or include 8 to 12 weight % of pigment and 12 to 18 weight % of filler such as 10 weight % of pigment and 15 weight % of filler.

The polyamide-ionomer blend may further contain additional additives that provide weatherability, stability or improved processing. The polyamide-ionomer composition or blend can comprise 0.1 to 5 weight% of optional additives, based on the weight of the total composition. Such additives include stabilizers, antioxidants, ultraviolet ray absorbers, hydrolytic stabilizers, antistatic agents, fire-retardants, processing aids such as lubricants, antiblock agents, release agents, or combinations of two or more thereof. Of particular note are oxidation inhibitors (antioxidants), UV stabilizers and hindered amine light stabilizers. The relative percentages of these additives may be varied depending upon the particular use of the object desired. The additives can be added to the polymer blend in typical melt compounding equipment.

Suitable stabilizers include antioxidants, such as the Irganox<sup>®</sup> family produced by Ciba-Geigy (now a part of BASF), and UV stabilizers such as those sold under the Tinuvin<sup>®</sup> tradename by Ciba-Geigy or Cyasorb<sup>®</sup> light stabilizer and light absorber produced by Cytec. Preferred antioxidants are based on hindered phenols, and preferred UV stabilizers are based on hindered amine light stabilizers (HALS) such as those sold under the Chimassorb<sup>®</sup> tradename from BASF.

Lubricants of note include salts of fatty acids such as sodium stearate or zinc stearate, which may be added at 0.1 to 1 weight % of the total composition.

The blend may also contain phosphorous salts such as a hypophosphite salt. Suitable phosphorous salts for use in the blends are described in greater detail in U.S. Patent 6,399,684. The salts, including sodium, lithium, or potassium hypophosphite may be added to the blend composition in 0.1 to 3 weight % of the composition. Hypophosphite salts may provide improved morphological or physical properties to the blend such as increased Vicat temperature and/or improved tensile properties. Of note is a composition as described herein consisting essentially of (1) a polyamide as described above; (2) an ionomer as described above; and (3) hypophosphite salt.

The polymeric blend composition may be mixed with pigment, filler and/or additional additives using well known melt mixing methods employing extruders or other suitable mixers such as Banbury or Farrel continuous mixers or roll mills.

Embodiments of the composition include:

The composition wherein the polyamide component comprises nylon 6, nylon 12, nylon 610, nylon 612, nylon 610/6T, nylon 612/6T, or combinations thereof.

The composition wherein the polyamide component comprises nylon 6.

The composition wherein the comonomer of (c) is a C<sub>1</sub>-C<sub>4</sub> alkyl half ester of maleic acid.

The composition wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising zinc or sodium, preferably zinc.

The composition wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising zinc or sodium.

The composition wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising zinc.

The composition wherein the anhydride ionomer component further comprises in combination with the anhydride ionomer one or more ionomer comprising an E/X/Y copolymer where E is ethylene, X is a C<sub>3</sub> to C<sub>8</sub>  $\alpha,\beta$ -ethylenically unsaturated monocarboxylic acid, and Y is a comonomer selected from alkyl acrylate and alkyl methacrylate wherein the alkyl groups have from 1 to 8 carbon atoms, wherein X is present in from 2 to 30 weight % of the E/X/Y copolymer, Y is present from 0 to 40 weight % of the E/X/Y copolymer, wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising one or more alkali metal, transition metal, or alkaline earth metal cations.

The composition comprising 8 to 20 weight % of pigment and 0 weight % filler.

The composition comprising 8 to 40 weight % of filler and 0 weight % of pigment.

The composition comprising 10 to 40 weight % of filler and 8 to 15 weight % of pigment.

The composition comprising 8 to 12 weight % of pigment and 12 to 18 weight % of filler.

The composition wherein the pigment comprises titanium dioxide, zinc oxide, or antimony oxide, preferably titanium dioxide.

The composition wherein the pigment comprises titanium dioxide.

The composition wherein the filler comprises an inorganic oxide, carbonate, sulfate, silica, alkali and alkaline earth metal silicate, or baryte of a metal of Groups IA, IIA, IIIA, IIB, VIB or VIII of the periodic table of the elements, preferably wherein the filler  
5 comprises calcium carbonate, barium sulfate, wollastonite and talc, more preferably talc.

The composition wherein the filler comprises an inorganic oxide, carbonate, sulfate, silica, alkali and alkaline earth metal silicate, or baryte of a metal of Groups IA, IIA, IIIA, IIB, VIB or VIII of the periodic table of the elements.

The composition wherein the filler comprises calcium carbonate, barium sulfate, wollastonite or talc.

The composition wherein the filler comprises talc.

10 The composition comprising 10 to 40 weight % of talc and 8 to 15 weight % of titanium dioxide.

The composition embodying any combination of the above embodiments.

Once the polyamide-ionomer blends are prepared as described above, they can be further processed into monolayer or multilayer structures useful as a backsheet for a photovoltaic module. Molten extruded thermoplastic compositions can be converted into film or sheet using any techniques known to one skilled in the art. Suitable additional processes include without  
15 limitation blown film extrusion, cast film extrusion, cast sheet extrusion, lamination, coextrusion, extrusion coating, and the like. A notable multilayer backsheet structure may comprise a layer comprising a blend of polyethylene and ionomer adjacent to the encapsulant layer of the photovoltaic module.

Embodiments of the backsheet comprise any of the compositions described above. Notable embodiments include:

The backsheet wherein the polyamide component comprises nylon 6.

20 The backsheet wherein the pigment comprises titanium dioxide.

The backsheet wherein the filler comprises calcium carbonate, barium sulfate, wollastonite or talc.

The backsheet wherein the filler comprises talc.

The backsheet wherein the polyamide-ionomer blend composition comprises 10 to 40 weight % of talc and 8 to 15 weight % of titanium dioxide.

25 A film or sheet can be further oriented beyond its immediate quenching or casting. The process comprises the steps of (co)extruding a laminar flow of molten polymers, quenching the (co)extrudate and orienting the quenched (co)extrudate in at least one direction. The film may be uniaxially oriented, or it can be biaxially oriented by drawing in two mutually perpendicular directions in the plane of the film to achieve a satisfactory combination of mechanical and physical properties. Orientation and stretching are well known to one skilled in the art and the description of which is omitted herein for the interest of brevity.

30 For multilayer structures, the layers may be coextruded or they may be formed independently and then adhesively attached to one another to form the backsheet. A backsheet can be made by (co)extrusion optionally followed by lamination onto one or more other layers. The backsheet may be fabricated by extrusion coating or laminating some or all of the layers onto a substrate. For example, a sheet or film of a core layer may be produced, to which skin layers and optional tie layers are adhered. Some back-sheet structures contain "e-layers" or layers that have a special affinity to adhere to the encapsulant. The e-layers can  
35 be co-extruded, or laminated to the subject back-sheet composition through the use of a coextrudable adhesive. However, additional "e-layers" are not required to be coated or laminated onto the instant backsheet structures. The polyamide-ionomer alloys that comprise the backsheet have strong adhesion to encapsulant layers, such as the standard commercial EVA encapsulant materials without additional "e-layers"

40 A sheet could be further processed by thermoforming into a shaped article. In thermoforming, a flat sheet is heated above its softening point and stretched to the desired shape. For example, a sheet comprising the polyamide-anhydride ionomer composition could be thermoformed into a shape that conforms to the shape of the photovoltaic elements in the photovoltaic cell.

For use as a backsheet in a photovoltaic module, the thickness of the sheet is 8 to 20 mils (200 to 500 microns).

A laminated solar cell module of the invention comprises or consists essentially of a frontsheet providing a front support layer formed of a light transmitting material and having first and second surfaces; a plurality of interconnected solar cells having a first surface facing the front support layer and a second surface facing away from the front support layer; a transparent encapsulant surrounding and encapsulating the interconnected solar cells, the transparent encapsulant being bonded to the second surface of the front support layer; and a backsheet as described above wherein the one surface of the backsheet is bonded to the second surface of the transparent encapsulant.

The frontsheet or incident layer may be derived from any suitable sheets or films. Suitable sheets may be glass or polymeric sheets, such as those comprising a polymer selected from polycarbonates, acrylics, polyacrylates, cyclic polyolefins (e.g., ethylene norbornene polymers), polystyrenes (preferably metallocene-catalyzed polystyrenes), polyamides, polyesters, fluoropolymers, or combinations of two or more thereof.

The term "glass" includes not only window glass, plate glass, silicate glass, sheet glass, low iron glass, tempered glass, tempered CeO-free glass, and float glass, but also colored glass, specialty glass (such as those containing ingredients to control solar heating), coated glass (such as those sputtered with metals (e.g., silver or indium tin oxide) for solar control purposes), E-glass, Toroglass, Solex<sup>®</sup> glass (PPG Industries, Pittsburgh, PA) and Starphire<sup>®</sup> glass (PPG Industries). Such specialty glasses are disclosed in, e.g., U.S. Patents 4,615,989; 5,173,212; 5,264,286; 6,150,028; 6,340,646; 6,461,736; and 6,468,934. It is understood, however, that the type of glass to be selected for a particular module depends on the intended use.

For example, fluoropolymer films, such as ethylene-tetrafluoroethylene copolymer (ETFE) films, may be used as frontsheets in photovoltaic modules instead of the more common glass layers. Another alternative is a film made from a perfluorinated copolymer resin such as a tetrafluoroethylene-hexafluoropropylene copolymer (FEP).

The light-receiving side of the solar cell layer may sometimes be referred to as a front side and in actual use conditions would generally face a light source. The non-light-receiving side of the solar cell layer may sometimes be referred to as a lower or back side and in actual use conditions would generally face away from a light source.

Monocrystalline silicon (c-Si), poly- or multi-crystalline silicon (poly-Si or mc-Si) and ribbon silicon are the materials used most commonly in forming the more traditional wafer-based solar cells. Photovoltaic modules derived from wafer-based solar cells often comprise a series of self-supporting wafers (or cells) that are soldered together. The wafers generally have a thickness of between 180 and 240  $\mu\text{m}$ .

The solar cell layer may be significantly thicker than the other layers and irregular in shape and/or thickness, including spaces between and around the solar cells and other components of the solar cell layer. In this connection, it should be noted that the conductors that interconnect the solar cells commonly are arranged to form stress relief loops to compensate for expansion and contraction caused by temperature changes. Those loops need to be encapsulated with the cells. However, when a polymeric backsheet is used, care must be taken to make certain that the stress loops will not pierce the backsheet when the several layers are compressed under heat to form the laminated module. Penetration of the backsheet by one or more stress loops will promote early failure of the module, e.g., by short-circuiting resulting from ingress of moisture at the point(s) of stress loop penetration the backsheet.

Therefore, portions of the backsheet laminate will contact the encapsulant layer outside the perimeter of the solar cell layer and can be adhered when heat is applied. As used herein, the perimeter of the solar cell layer is the outline of the outer limits of the area encompassed by the solar cell layer. In many cases, it is desirable that the encapsulant material flows into the spaces and closely encapsulates the solar cells and other components to physically consolidate the photovoltaic module. Thus, it may be necessary to apply heat for a period of time sufficient to allow such flow, which may be longer than that needed for adhering thinner layers of a more regular shape. For example, heat may be applied in such a manner that the assembly is maintained above the softening point of the encapsulant layer for 5 to 30 minutes to effectively consolidate the photovoltaic module.

The encapsulant layers used in preparing photovoltaic modules described herein may each comprise a polymeric material independently selected from olefin unsaturated carboxylic acid copolymers, ionomers of olefin unsaturated carboxylic acid copolymers, ethylene vinyl acetate copolymers, poly(vinyl acetals) (including acoustic grade poly(vinyl acetals)), polyurethanes, polyvinylchlorides, polyethylenes (e.g., linear low density polyethylenes), polyolefin block copolymer elastomers, copolymers of  $\alpha$ -olefins and ethylenically unsaturated carboxylic acid esters (e.g., ethylene methyl acrylate copolymers and ethylene butyl acrylate copolymers), silicone elastomers, epoxy resins, or combinations of two or more thereof.

The encapsulant layer may preferably comprise a thermoplastic polymer including ethylene vinyl acetate copolymers, olefin unsaturated carboxylic acid copolymers, ionomers of olefin unsaturated carboxylic acid copolymers, and combinations thereof (for example, a combination of two or more olefin unsaturated carboxylic acid copolymers, a combination of two or more ionomers of olefin unsaturated carboxylic acid copolymers, or a combination of at least one unsaturated carboxylic acid copolymer with one or more ionomers of unsaturated carboxylic acid copolymers).

The solar cell module and assembly to prepare it may optionally further comprise other functional film or sheet layers (e.g., dielectric layers or barrier layers) embedded within the module. Such functional layers may be derived from any of the above mentioned polymeric films or those that are coated with additional functional coatings. For example, poly(ethylene terephthalate) films coated with a metal oxide coating, such as those disclosed in U.S. Patents 6,521,825 and 6,818,819 and European Patent EP1182710, may function as oxygen and moisture barrier layers in the laminates.

If desired, a layer of nonwoven glass fiber (scrim) may also be included between the solar cell layers and the encapsulants to facilitate deaeration during the lamination process or to serve as reinforcement for the encapsulants. The use of such scrim layers is disclosed in, e.g., U.S. Patents 5,583,057; 6,075,202; 6,204,443; 6,320,115; and 6,323,416 and European Patent EP0769818.

In addition, metal films, such as aluminum foil or metal sheets, such as aluminum, steel, galvanized steel, or ceramic plates may be utilized in addition to the polymeric backsheet described herein as backing layers for the photovoltaic module.

A special film or sheet may be included to serve both the function of an encapsulant layer and an outer layer. It is also conceivable that any of the film or sheet layers included in the module may be in the form of a pre-formed single-layer or multilayer film or sheet.

If desired, one or both surfaces of the incident layer films and sheets, the backsheet films and sheets, the encapsulant layers and other layers incorporated within the solar cell module may be treated prior to the lamination process to enhance the adhesion to other laminate layers. This adhesion enhancing treatment may take any form known in the art and includes those set forth in U.S. Patent Application Publication 2010/0108126.

Manufacturing the solar cell module comprises providing a front support layer formed of a light transmitting material and having front and back surfaces; placing a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; positioning a plurality of interconnected solar cells having first and second surfaces so that the first surfaces thereof are adjacent to the first transparent encapsulant layer; placing a second transparent thermoplastic encapsulant layer adjacent to the second surfaces of the solar cells; placing a backsheet as described above adjacent to the second transparent thermoplastic encapsulant layer to thereby form an assembly; subjecting the assembly to heat and pressure so as to melt the encapsulant layers and cause the encapsulant to surround the solar cells, and cooling the assembly so as to cause the encapsulant to solidify and bond to the front support layer, the solar cells and the backsheet, thereby laminating the layers and the solar cells together to form an integrated solar cell module.

The photovoltaic module is prepared by providing an assembly for conversion under heat and pressure into a laminated solar cell module, the assembly comprising a front support layer formed of a light transmitting material and having front and back surfaces; a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; a plurality of interconnected solar cells having first and second surfaces adjacent to the first transparent encapsulant layer; a second

transparent thermoplastic encapsulant layer disposed adjacent to the solar cells in parallel relation to the first transparent encapsulant layer; and a thermoplastic backsheet layer as described above.

A vacuum laminator may be used to adhere the layers of the assembly together to provide the photovoltaic module. The laminator comprises a platen base, on which the layers of the assembly are placed in overlaying fashion for lamination. The laminator also comprises an enclosure that covers and completely surrounds the platen base. The region enclosed by the platen and enclosure may be evacuated. The laminator also comprises a flexible bladder within the enclosure attached to the top inner surface of the enclosure, which may be inflated to a pressure greater than the pressure in the evacuated region. For example, the pressure above the bladder may be atmospheric and the laminate may be held under vacuum beneath the bladder to remove air. When the bladder is inflated, the flexible surface of the bladder is pushed from the top of the enclosure toward the platen and applies a surface pressure to the multilayer assembly to ensure a good thermal contact between the assembly and the platen. For lamination of the module, the laminator is preheated to a temperature above the softening temperature of the encapsulant layer(s) and held at that temperature throughout the lamination process.

Heat-resistant sheets may be placed under the assembly to retard heat flow and allow deaeration and devolatilization of the sample. Release sheets may be placed under the and/or over the assembly to prevent the sample layers from adhering to parts of the laminator. The assembly is placed on the platen and the enclosure of the laminator is lowered into place and sealed. Next, the region surrounding the assembly between the platen and enclosure of the laminator is evacuated (e.g. to a pressure of 1 mbar) to help further with the prevention of voids, defects, and air pockets. Next, the rubber bladder is inflated (e.g. to a pressure of 999 mbar) so that it presses against the assembly and ensures good thermal contact with the platen. The pressure and heat are maintained for a sufficient period of time (for 1 to 10 minutes) to soften the encapsulant layers and adhere to solar cells and the adjoining layers.

When the heating step is complete, the bladder is depressurized to 0 mbar so that it may be removed from contact with the multilayer film laminate, the enclosure is vented to atmospheric pressure and the enclosure is unsealed and opened. The multilayer film laminate is removed from the platen and allowed to cool to room temperature.

The lamination methodology described here is by no means the only possible way to carry out such laminations. For example, more advanced laminators have retractable pins that hold the multilayer laminate structure above the heat source until the desired time to effect contact and heating. This would obviate the need for heat resistant layers in most cases.

### EXAMPLES

#### Materials Used

AI-1: A copolymer of ethylene, 11 weight % of methacrylic acid and 6 weight % of ethyl hydrogen maleate, neutralized with Zn cations to a level of 50-60%, MI of 0.1 g/10 min.

AI-2: an anhydride ionomer terpolymer comprising ethylene, 13 weight % of acrylic acid and 4 weight % of ethyl hydrogen maleate neutralized with Zn cations to a level of 50%.

ION-1: A copolymer of ethylene and 19 weight % of methacrylic acid, neutralized with Zn cations to a level of about 36 %, MI of 1.3 g/10 min.

ION-2: A copolymer of ethylene and 15 weight % of methacrylic acid, neutralized with Zn cations to a level of about 60%, MI of 0.7 g/10 min.

PA-6: nylon-6 homopolymer available commercially as Ultramid® B27E from BASF.

PA-12A: nylon-12 homopolymer available commercially as Rilsan® AMNO from Arkema.

PA-12B: nylon-12 homopolymer available commercially as Rilsan® AESNO from Arkema.

PA-612/6T: nylon-612/6T copolymer available from DuPont under the tradename Zytel®.

Tie-1: a maleic anhydride modified linear low density polyethylene (LLDPE), with density of 0.91 g/cm<sup>3</sup> and melt index of 1.7 g/10 min, commercially available from DuPont.



Tie-2: a maleic anhydride modified linear low density polyethylene (LLDPE), with density of 0.91 g/cm<sup>3</sup> and melt index of 2.7 g/10 min, commercially available from DuPont.

Tie-3: a maleic anhydride modified linear low density polyethylene (LLDPE), with density of 0.91 g/cm<sup>3</sup> and melt index of 3.1 g/10 min, commercially available from DuPont.

5 TiO<sub>2</sub>: titanium dioxide commercially available from DuPont as Ti-Pure<sup>®</sup> R105 or comparable material.

TiO<sub>2</sub> Concentrate: 70 weight % titanium dioxide pre-dispersed in ethylene methacrylate copolymer commercially available as 111676 White COP MB from Ampacet (660 White Plains Road, Tarrytown, NY 10591).

ZnO Concentrate: 45 weight % zinc oxide pre-dispersed in ethylene methacrylic acid copolymer.

Fillers used are summarized in the following table.

Filler	material	Commercial designation	Particle shape	Diameter (μm)	Length (μm)	bulk density (kg/m <sup>3</sup> )	true density (kg/m <sup>3</sup> )
F1	wollastonite	Nyglos <sup>®</sup> 8	rod	12	156	480	2900
F2	coated wollastonite	Nyglos <sup>®</sup> 8 10012	rod	12	156	480	2900
F3	wollastonite	Nyglos <sup>®</sup> G	rod	55	825	720	2900
F4	mica	Suzorite <sup>®</sup> 60S	platy	150-500		176-291	2700
F5	glass fibers	Chopvantage <sup>®</sup> HP3660	fiber	10	3000-4000		2460
F6	talc	Jetfine <sup>®</sup> 3CA	platy	1			
F7	talc	Jetfine <sup>®</sup> 3CC	platy	1			
F8	talc	Luzenac <sup>®</sup> HAR-T84	platy	2			

10 Additives used are summarized in the following table.

Additive	function	material	Commercial designation
Add1	lubricant	zinc stearate	Commercial grade
Add2	UV absorber	oxanilide	Tinuvin <sup>®</sup> 312
Add3	UV light stabilizer	HALS	Tinuvin <sup>®</sup> 770
Add4	UV light stabilizer	HALS	Chimassorb <sup>®</sup> 440
Add5	antioxidant	phenolic antioxidant and phosphite	Irganox <sup>®</sup> B1171
Add6	antioxidant	phenolic antioxidant	Irganox <sup>®</sup> 1010
Add7	antioxidant	phenolic antioxidant and phosphite	Irganox <sup>®</sup> B215
Add8	processing stabilizer	trisarylphosphite	Irgafos <sup>®</sup> 168
Add9	antioxidant	phenolic antioxidant	Irganox <sup>®</sup> 1098
Add10	UV light stabilizer	HALS	Chimassorb <sup>®</sup> 944
Add11	UV absorber	Oxanilide	Tinuvin <sup>®</sup> 234
Add12	processing stabilizer	Sodium hypophosphite	Commercial grade
Add13	lubricant	Zinc stearate	Commercial grade

Several commercial backsheet structures were evaluated as standards:

HRPET: Hydrolysis resistant PET, 12 mil thickness, commercially available as PYE3000 from Coveme SPA, Bologna, Italy.

TPT: Tedlar<sup>®</sup>/PET/Tedlar<sup>®</sup>, 12 mil thickness, commercially available as Icosolar<sup>®</sup> 2442 from Isovoltaic AG, Leibring, Austria or 1200 Dun-solar TPT backsheet, commercially available from DUNMORE Corporation, 145 Wharton Rd., Bristol, PA 19007.

APA: A three-layer sheet comprising a core layer comprising polyester and two skin layers comprising modified polyamide, commercially available as Icosolar<sup>®</sup> APA 4004 from Isovoltaic AG, Liebring, Austria.

AAA: A three-layer sheet comprising a core layer comprising modified polyamide and two skin layers comprising modified polyamide, commercially available as Icosolar<sup>®</sup> AAA 3554 from Isovoltaic AG, Liebring, Austria.

5 Blends of materials listed above were prepared by melt blending following the procedure described or similar processes. Compounding was done using a 25mm 38/1 L/D ZSK-25 World Lab twin-screw extruder comprised of nine 100mm long barrels manufactured by Krupp Werner & Pfleiderer (Coperion) or similar processes. The polymers were pre-blended and then fed to the throat of the extruder (barrel 1) using a K-tron<sup>®</sup> loss-in-weight feeder. The fillers were fed using a second K-tron<sup>®</sup> feeder to the extruder using a side feeder at Barrel 4. There was a vacuum pulled on the melt before and after addition of the  
10 filler (at barrel 4 and barrel 8). The melt blend exiting the extruder die face (after barrel 9) was die face-cut using a Gala cutter.

Operating conditions for Comparative Example C4 in Table 2 are shown in Table 1. Other examples were prepared similarly.

Table 1

	Set Point (°C)	Actual (°C)
Temperature Control Zone 1 (Barrel 1 feed)	uncontrolled	
Temperature Control Zone 1 (Barrel Zones 2 and 3)	260	260
Temperature Control Zone 2 (Barrel Zones 4 and 5)	260	260
Temperature Control Zone 3 (Barrel Zones 6 and 7)	260	260
Temperature Control Zone 4 (Barrel Zones 8 and 9)	260	257
Temperature Control Zone 5 (Die)	260	257
Screw RPM		300
Torque %		53%
Die pressure (Mpa)		5.3 (770 psig)
Melt Temperature °C		297
Feed rate polymer		95 gpm
Feed rate filler		40 gpm
Vacuum (mm Hg) Zone 4 and 8		51KPa (15 in Hg)

The compositions of the melt blends are shown in Table 2.

Table 2

Example	Filler Loading	PA6	ION-2	AI-1	F1	F2	F3	F4
C1	0%	60%	40%					
C2	0%	60%		40%				
C3	15%	51%	34%			15%		
1	15%	51%		34%		15%		
C4	30%	42%	28%		30%			
2	30%	42%		28%		30%		
3	40%	36%		24%		40%		
4	30%	42%		28%			30%	
5	40%	36%		24%				40%
C5	15%	51%	34%					15%
6	15%	51%		34%				15%
C6	30%	42%	28%					30%
7	30%	42%		28%				30%

The collected pellets were dried overnight at 70 to 85 °C in an air-circulating Blue M tray dryer oven that was fitted with a nitrogen purge. Each of the dried polymer samples were used to cast 8-inch (228 mm) wide, nominally 0.33 to 0.35 mm thick sheets. Sheets were cast using a 31.75-mm diameter 30/1 L/D single screw extruder, built by Wayne Machine (Totowa, New Jersey), fitted with a 3/1 compression ratio, single-flight screw with 5 L/D of a melt mixing section. The extruder die was a 203-mm wide coat hanger type flat film die with a 0.35mm die gap. The molten polymer film exiting from the die was cast onto a 203-mm wide by 203-mm diameter double shell spiral baffle casting roll fitted with controlled temperature cooling water. The casting roll and die were built by Killion Extruders (Davis Standard, Cedar Grove, New Jersey). Extruder conditions typical for the compositions are provided in Table 3.

Table 3

Extruder Conditions	Set Point (°C)	Actual (°C)
Barrel Zone 1	240	240
Barrel Zone 2	240	240
Barrel Zone 3	240	240
Barrel Zone 4	240	240
Filter Flange	240	240
Adapter	240	240
Die End	245	245
Flat Die	240	240
Melt Temp (before filter)		237
Melt Temp (after filter)		240
Melt Press (MPa) before filter		4.6 (670 psig)

The 0.35-mm thick polymer sheets were used to test properties relevant to PV backsheets.

#### Test Methods

**Ash:** Samples were weighed into a crucible and heated for 15minutes in an 800 °C muffle furnace. The reported number represents the % of sample remaining in the crucible. This test was used to verify that the proper loading of filler was achieved.

**Tensile Properties (ASTM D882-12)** were measured on 25 mm by 150 mm coupons die cut from the sheet. Five coupons were oriented so that the long direction was in the machine direction (MD) and five coupons were oriented so that the long direction was in the transverse direction (TD). Coupons were conditioned at 50% RH and 23°C for at least seven days prior to testing at 50% RH and 23 °C. The gage length was 25mm and the cross-head speed was 508mm/min. The reported results are the average of five coupons. A combination high Young's Modulus (measure of stiffness) and at least 100% elongation (higher elongation suggests better toughness) to break is preferred.

**Coefficient of Linear Thermal Expansion (ASTM E813-13)** was measured on 4.9 by 65 mm coupons die cut from the sheet. Three coupons were oriented so that the long direction was in the MD and three coupons were oriented so that the long direction was in the TD. Coupons were conditioned at 50% RH and 23°C for at least seven days prior to testing. The thermal mechanical analyzer was set up with the film fiber probe and a 0.1 N preload force was applied. Specimens were cooled to -60 °C prior to the start of the run and the heated a rate of 5 °C/min to 90 °C. The slope of the best fit linear line between -60°C and 90°C was taken as CLTE ( $\mu\text{m}/\text{m}/^\circ\text{C}$ ). Two test results in each orientation are reported. A lower coefficient of linear thermal expansion is preferred so that the thermoplastic backsheet expands and contacts the encapsulant layer.

**Moisture Vapor Transmission Rate (MVTR) at 38°C and 100% RH** was measured on coupons cut from sheet according to ASTM F1249.

**Moisture Vapor Transmission Rate (MVTR) at 85°C and 100% RH** was measured on a Permatran 3/33 model using a Yamato DKN402 oven. ASTM F1249-06, was followed except the temperature was 85°C.

Moisture Uptake at 50% RH and 23 °C or water immersion at 85 °C This test measured the amount of water absorbed into a sheet sample (typically 25mm wide by 150mm long) die cut from the sheet after the indicated exposure condition.

Moisture absorbed was determined by Karl Fischer titration as per ASTM D6869-03 (150°C oven temperature).

Shrinkage after heat treating for 30minutes in air-circulating oven at 150 °C Rectangles were scissor cut from the sheet. Samples were allowed to condition at 50% RH and 23 °C for at least seven days prior to measuring sheet dimension in MD and TD. Samples were then suspended by a hook in a pre-heated air circulating oven at 150 °C for 30 minutes after which the sheet samples were removed from the oven allowed to condition for at least 48 hours at 50% RH and 23 °C before the MD and TD lengths were measured again. Changes in dimension are reported as the percent reduction in the dimension as a result of conditioning at 150 °C.

Color Measurement (ASTM E1347-06) Color of samples was measured using a Hunter Lab Colorquest XE Colorimeter (L\*, a\*, b\*). L\* is a measure of whiteness; whiter materials have higher L\* values.

Summaries of the test results are given in Tables 4 and 5.

Table 4

	Tensile properties (ASTM D882-12)				CLTE (µm/m/°C)			
	Young's Modulus (ksi)		Strain at break (%)		MD		TD	
	MD	TD	MD	TD	Sample 1	Sample 2	Sample 1	Sample 2
C1	82.8	83.6	507	451	248	232	232	225
C2	100	79	484	451				
C3	154	102	458	428	152	144	224	244
1	118	101	431	298				
C4	249	108	162	19				
2	181	125	12.6	16.2				
3	194	138	3.8	5.4				
4	261	114	64	20.8				
5	171	161	3.8	4.8				
C5	117.9	114	73.7	36	142	126	136	152
6	114.9	107.8	91	50				
C6	178.8	177	9	6.6				
7	163	141	9	8.2				

The test results in Table 4 show that adding fillers increased stiffness as measured by a higher Young's modulus and reduced toughness as measured by elongation or strain at break. Fillers also reduced the Coefficient of Linear Thermal Expansion and on a weight basis, bigger rodlike or platelike fillers reduce CLTE more. Comparative Example C1 contained no filler and had CLTE greater than 200 µm/m/°C. Comparative Example C3 containing 15 weight % of wollastonite had an MD CLTE of less than 155µm/m/°C and Comparative Example C5 containing 15 weight % mica had an MD CLTE of less than 142µm/m/°C.

Table 5

	MVTR at 38°C		Water Uptake at 23°C		shrinkage at 150°C	
	g-mm/m <sup>2</sup> /day	g-mm/m <sup>2</sup> /day	at 50% RH	in water	MD	TD
	Sample 1	Sample 2	%	%	%	%
C1	95	99	2.5	7.3	0	0.3
C2	64	62	2.4	6.4	0.5	0.6
C3	72	70		5.6	0.2	1
1	62	61		5.3	0.5	0.7
C4	48	40	1.6	5.3	0.1	0.7
2	67	69		4.7	0.2	0.3

3				4.8	0.1	0.7
4	43	41	1.7	4.9	0	0.3
5	330	366	1.4	4.7	0	0
C5	54	51	2	5.4	0.5	0.3
6	63	63	2.1	5.6	0.2	0.3
C6	32	32	1.6	4.7	0.4	0
7	65	67	1.6	4.7	0	0

As shown in Table 5, the addition of fillers tended to reduce the moisture vapor transmission rate, with some exceptions. Example 5 contained 40 weight % of mica and had a very high MVTR (higher than the sample that contained no filler). With small rod-shaped fillers, the polyamide-anhydride ionomer examples had lower MVTR than comparable polyamide-conventional ionomer compositions. With mica filler, the opposite was observed. The addition of filler can result in a reduction of the moisture uptake. In terms of shrinkage of the sheet after 30 minutes at 150 °C, all of the samples had very low shrinkage (with or without filler). These shrinkage numbers are very close to the error in the measuring device.

The results in Tables 4 and 5 suggested bigger filler types (like fiberglass or mica) were better at reducing sheet CLTE and increasing sheet stiffness but resulted in very low elongation to break and had negative effect on moisture transmission.

Additional blends were prepared with additives as shown in Table 6.

Table 6

Blend Material	A	B	C*	D* Weight %	E	F*	G	H	I	J
Blend A					83.9				73.9	
Blend C				90		85				
PA-6	60	0	60				60			39.23
PA-12A	0	22	0							
PA-12B	0	33	0							
AI-1	40	45	0				40			
ION-1			40							31.36
ION-2								35.7		
HDPE								53.54		
ZnO Conc.										1.7
TiO <sub>2</sub>					15	15		10	10	15
Talc				10					15 F6	10
Add1					0.7					
Add2									0.2	
Add3								0.15	0.6	
Add5					0.2				0.3	0.28
Add6					0.1			0.15		0.15
Add7								0.15		
Add8					0.1					
Add10								0.31		0.7
Add11										0.35
Add12										0.13
Add18										1.1

\* Commercially available from LTL Color Compounds, Inc. 20 Progress Drive, Morrisville, PA 19067.

Additional UV stabilizer and antioxidant additives may also be present (not included in the weight %).

The blends in Table 6 were combined and melt blended with additional materials to provide polyamide-ionomer compositions summarized in Table 7 useful for PV backsheets. Other fillers were included such as talc. Example 10 has no filler but has 21.5 weight % of TiO<sub>2</sub> Concentrate as a pigment.

Table 7

	Filler Loading	Blend			TiO <sub>2</sub> Concentrate	F6	F7	Add2	Add3	Add5	L*	a*	b*
		A	C	G									
C7	0		100										
C8	30		70				30						
8	30	70					30						
C9	15		85				15						
9	15	85					15						
10				77.7	21.5			0.2	0.6	0.3	89.3	-0.2	3.3
11	15			84.2			15	0.2	0.6	0.3	47.5	1.1	3.1
12	25			74.3		25		0.2	0.6	0.3	58.6	-1.3	3.9
13	15			70.1	14.2	15		0.2	0.6	0.3	84.6	-0.9	3.3

The color of pellets of Compositions 10-13 was measured and the values indicated in Table 7. Compositions using talc filler F6 were whiter than with talc filler F7 as indicated by L\*.

Tables 8 and 9 list the property tests on the sheet samples.

5

Table 8

	Tensile properties (ASTM D882-12)				CLTE ( $\mu\text{m}/\text{m}/^\circ\text{C}$ )			
	Young's Modulus (MPa)		Strain at break (%)		MD		TD	
	MD	TD	MD	TD	Sample 1	Sample 2	Sample 1	Sample 2
C7	548	566	560	533	212	202	194	205
C8	1426	1197	200	186	80	77	96	97
8	1257	1248	326	297				
C9	963	856	377	395	126	112	151	138
9	944	783	453	440				
HRPET	2141	2158	135	114	588.1	584.5	205	224.8
TPT	3123	3241	101	81	33.08	33.96	25.15	25.8

The test results in Table 8 show that smaller platelike fillers like talc can be added to the polyamide-ionomer blend at loadings as high as 30 weight % to reduce the CLTE and still maintain at least 100% elongation to break. Polyamide-anhydride ionomer compositions had roughly comparable stiffness and better strain at break than comparable polyamide-conventional ionomer compositions. The compositions had lower Young's modulus and higher strain at break than commercial backsheet materials.

10

Table 9

	Ash Test		MVTR at 38°C (g-mm/[m <sup>2</sup> -day])		Water uptake at 23 °C		shrinkage at 150°C	
					at 50% RH	in water	MD	TD
	Sample 1	Sample 2	Sample 1	Sample 2	%	%	%	%
C7	2	2	9.24	9.62	2.2	6.3	0.7	0.7
C8	31	31	1.88	1.88	1.4	4.2	0.2	0.5
8	30	30	1.75	1.75	1.5	4.4	0.1	0.5
C9	16	16	3.88	4.11	1.8	5.0	0.2	0.3
9	16	16	3.15	3.20	1.8	4.6	0.6	0.5
HRPET			0.786		0.2	1.4	1.0	0.3
TPT			0.786		0.3	0.5	0.2	0.0

In Table 9, significant reductions in MVTR and moisture absorption were seen when fillers were added to the polyamide-ionomer blends. Example 8 and Comparative Example C8 contained 30 weight% talc and had MVTR of 75 or less

and moisture absorption of less than 4.5 weight % at 85 °C compared to the non-filled blend (C7) that had an MVTR of over 300 g-mil/m<sup>2</sup>/day and moisture absorption at 85 °C of 6.3%. Polyamide-anhydride ionomer compositions had roughly comparable water uptake and lower MVTR than comparable polyamide-conventional ionomer compositions.

The next comparison illustrates that under equivalent compounding conditions, better mechanical and thermal properties were obtained when the mineral filler was added to a nylon-6 polyamide-anhydride ionomer alloy compared to adding the mineral filler to a nylon-6 polyamide-standard ionomer alloy.

Table 10

	Weight %						Thickness of sheet (mm)
	PA-6	ION-2	AI-1	Add8	Add9	F7	
C10	44.80	30		0.10	0.1	25	0.35
10	44.80		30	0.10	0.1	25	0.34

The tensile properties were measured on film samples conditioned for one week at 23°C and 50% RH and summarized in Table 11. Moisture permeation, uptake and damp heat aging are also summarized in Table 11.

Table 11

		C10		10	
Tensile properties (ASTM D882-12)	Young's Modulus (MPa)	MD	TD	MD	TD
		1113	938	1150	896
		Strain at break (%)	42	319	280
		CLTE (µm/m/°C)	83.5	72.5	102
MVTR at 38 °C	Transmittance (g/[m <sup>2</sup> -day])	6.05		5.84	
	Permeation (g-mm/[m <sup>2</sup> -day])	2.13		1.97	
Water uptake at 23 °C	At 50% RH (%)	1.4		1.3	
	1 week in water (%)	4.6		4.5	
Damp heat aging at 85 °C and 100 % RH		Young's Modulus	Strain at break	Young's Modulus	Strain at break
		After 72 hours	929	186	853
		After 1000 hours	1001	51	1020
		After 2000 hrs	1028	50	1252

As shown in Table 11, Young's Modulus was comparable for both compositions but the strain at break was significantly higher for the polyamide-anhydride ionomer composition. The coefficient of linear thermal expansion was lower in both MD and TD for the polyamide-anhydride ionomer composition. The permeation rate was lower with the polyamide-anhydride ionomer composition. The moisture pick-up after 1 week at 50% RH or 1 week immersed in water was slightly lower with the polyamide-anhydride ionomer composition.

Table 11 also records the stiffness (Young's modulus) and elongation (strain at break) for the MD-oriented coupons after 72, 1000 and 2000 hours of damp heat conditioning. The mineral-filled alloy based on the polyamide-anhydride ionomer composition produced better hydrolysis resistance to damp heat aging than the one prepared with a conventional ionomer. The polyamide portion of the polyamide-ionomer alloy was presumably undergoing further crystallization under the 85 °C and 100% RH conditioning so the observed stiffness (Young's Modulus) increased with time. Because hydrolysis and/or thermal degradation were also occurring there was a reduction in the amount of elongation in the coupon before it broke. The coupons from the anhydride ionomer-polyamide composition exhibited much better retention of elongation to break over this 2000-hour conditioning period.

Three-layer coextruded backsheets were prepared with the structures summarized in Table 12. The term "outer skin" refers to the layer of the backsheet structure that would face outward in the photovoltaic module, the term "core" refers to a layer inside the backsheet structure, and the term "inner skin" refers to the layer of the backsheet that would face the encapsulant layer of the photovoltaic module. The three-layer sheets were prepared from 15 weight % TiO<sub>2</sub>-filled and 10 weight

% talc-filled polyamide-ionomer compositions. The sheets were cast on a three layer coextrusion sheet line with nominally 0.002-inch (0.05 mm) thick skin layers and 0.010-inch (0.25 mm) thick core layers. The sheet structures are given in Table 12.

Table 12

	Outer skin layer	Core layer	Inner skin layer
C11	Blend F	Blend C	Blend F
C12	Blend F	Blend D	Blend F
11	Blend F	Blend A	Blend F
C13	Blend C	Blend D	Blend C
12	Blend C	Blend A	Blend C
13	Blend E	Blend B	Blend E

Tables 13 and 14 report the test results on the three-layer sheets.

5

Table 13

	Tensile properties (ASTM D882-12)				CLTE ( $\mu\text{m}/\text{m}/^\circ\text{C}$ )	
	Young's Modulus (MPa)		Strain at break (%)		MD	TD
	MD	TD	MD	TD		
C11	717	703	495	505	144.6	153.6
C12	800	727	446	459	114.1	138.3
11	624	591	414	470	186.9	210.6
C13	816	770	462	444	48.8	125.5
12	619	639	502	472		
13	627	577	395	503	175.5	199.2
HRPET	2141	2158	135	114	586.3	227.9
TPT	3123	3241	101	81	33.5	25.5

Table 14

	Thickness		MVTR at 85°C		Water uptake at 23 °C		shrinkage at 150°C	
	(mil)	(mm)	g/[m <sup>2</sup> -day]	(g-mm/[m <sup>2</sup> -day])	At 50% RH	At 100% RH	MD	TD
					%	%	%	%
C11	14.5	0.368	250.93	92.33	1.7	5.6	0.5	0.9
C12	13.9	0.353	213.68	75.43	1.6	4.8	0.1	0.5
11					1.8	5.8	0.0	0.7
C13					1.6	4.8	0.2	0.3
12					1.9	6.1	0.0	0.7
13	15.6	0.396	85.84	33.99	0.9	2.1	0.1	0.7
HRPET	12	0.305			0.2	1.4	1.0	0.3
TPT	13	0.330	36.5	12.04	0.3	0.5	0.2	0.0

Example 13 showed very good (low) water vapor transmission and uptake compared to Comparative Examples C11 and C12.

10 The following General Procedure was used to laminate backsheet materials to glass and encapsulant sheets to test the properties of a photovoltaic module. For these tests no photovoltaic layer was included, but laminations with photovoltaic layers could be performed similarly.

Glass: 4-mm thick annealed glass from Kingston Plate and Glass (Kingston, Ontario Canada)

Encapsulant sheet: 0.015 inch thick EVA (ethylene vinyl acetate copolymer encapsulant sheet (Photocap<sup>®</sup> 15420 or 15295 sold by STR Corporation, 18 Craftsman Road, East Windsor, CT 06088 USA)



General Procedure for fabricating glass/encapsulant/backsheet laminates

Rectangles of Glass/Encapsulant/Backsheet [2.5 inch by 5 inch (62 by 125mm) or 5 inch by 5 inch (125 by 125mm)] were laminated. The backsheet samples were dried overnight under vacuum at 30 °C prior to laminating. The glass was washed with soapy water and the layers assembled as follows:

5 glass/encapsulant/backsheet/10 mil thick fluoropolymer coated cloth/0.25-inch silicone rubber. If the backsheet contained an inner skin layer, that layer was in contact with encapsulant. One-inch (25mm) wide tabs of fluoropolymer-coated release sheet were placed between the glass and the encapsulant and the encapsulant and the backsheet. The inclusion of fluoropolymer release sheets provided a tab between glass and encapsulant or backsheet and encapsulant to initiate subsequent peel testing on laminated glass. The assembly was placed in a 12 inch by 9 inch (300mm by 225mm) vacuum bag (Tyvek<sup>®</sup> barrier bag Part  
10 number SI-BA-7033, supplied by Smith Induspac Ltd 140 Iber Road, Stittsville, Ontario, Canada K2S 1E9). The assembly was vacuum sealed in the bag using a Promarks TC-420LC vacuum sealer (Promarks, Inc., 1915 E. Acacia Street, Ontario, CA 91761, USA) with 60 seconds of vacuum prior to heat seal, seal time 2 seconds, cooling time 2.1 seconds. The layers were thermally bonded together by placing the vacuum-bagged assembly in a preheated air-circulating oven set at 150 °C for 30 minutes. A 2.6 kg aluminum block was placed on top of the vacuum bag to improve the consistency of the glass laminate. The  
15 glass side of the laminate was down during the thermal bonding. After 30 minutes in the oven, the vacuum bag assembly was removed and allowed to cool 10 minutes before cutting open the bag and removing the laminate.

Adhesion: Peel tests were performed using 4 inch/minute (100 mm/min crosshead speed after conditioning at 23 °C, 50% RH. The glass is held in place by the stationary jaw on the test machine. A tab from the backsheet is gripped in the moving jaw. Test results are reported as the average of four peels for each material. Table 15 reports the average peel strength between  
20 the weakest interface (backsheet to encapsulant) or (encapsulant to glass).

Table 15

	Adhesion to Encapsulant		Location of peel
	lb-f/in	N/cm	
C11	32	56	Backsheet to encapsulant
C12	40	70	Combination of backsheet to encapsulant and glass to encapsulant
11	40.5	71	
C13	13	23	Backsheet to encapsulant
HRPET	10	18	Backsheet to encapsulant
TPT	21	37	Glass-encapsulant

These peel strength numbers demonstrate a strong bond between backsheet and encapsulant which confirms the polyamide/ionomer alloy sheet does not require any special surface treatment to achieve a strong bond to the EVA based encapsulant.

25 Test sheets were also fabricated into small photovoltaic modules having a 2 x 2 array of solar cells. The components are listed in Table 16. The materials were laid up in the proper positions to prepare a functional photovoltaic module and laminated using a Meier vacuum laminator with lamination conditions of 145 °C, for fifteen minutes, evacuation 3 minutes and press for 11 minutes.

Table 16

Material	Brand	Code
EVA	Revax	R767-0.45 mm thickness
Solar cell	JA	Mono-125SOR22B 17.6-17.8%

Solder Ribbon	Sveck	Sn62 Pb36 Ag2
Glass	AGC Japan	Solar grade --temper glass--low-Iron
Al Frame	Haida	304 x 284 x 25mm
Junction Box	Renhe	PV-RH06-60
Seal silicon	Tonsan	PV1527
Flux	Asahi	ANX-3133

#### Testing after sheets were used to fabricate Mini Solar Modules

Damp heat: Two modules of each state were exposed to 85 °C/85% RH for after 0, 1000 and/or 2000 hours of damp heat exposure. Changes in color on the front side (color through the glass) and back side of module were measured as described above.

5 UV testing: Using the IEC 61215 UV preconditioning standard (3% UVB), one module was exposed from the front and one from the back. Tests were conducted after 0, 1, 2, 3, 4 and 5 times the IEC standard duration.

Thermal cycling (TC): Two modules of each state were exposed to cycles of -40 °C to 85 °C per the IEC 61215 standard. Standard properties (see below) were measured after 0, 200, 400 and 600 cycles.

10 Thermal cycling-humidity freeze (TC-HF): Two modules of each state were exposed to the IEC 61215 protocol of 50 cycles thermal cycling (-40 °C to 85 °C) followed by 10 cycles of humidity freeze (-40 °C to 85 °C at 85% RH). Standard properties after 0, 1, 2 and 3 intervals of 50 thermal and 10 humidity freeze cycles were measured. Testing was done after the humidity freeze portion of the cycling. In the Tables below, the number of each type of cycle is indicated.

Table 17 reports testing on mini modules that were fabricated using the backsheets in Table 12.

Table 17

Change in module color after 1000 hours damp heat treatment

Crack behavior

	$\Delta L^*$ front	$\Delta L^*$ back	$\Delta b^*$ front	$\Delta b^*$ back	after 150TC/20HF	after TC400	after TC150/HF30
C11	3.6	0.6	-18.0	-3.6	Yes	No	Yes
C12	4.0	0.3	-20.1	-3.0	No	No	No
11	2.1	0.4	-10.1	-2.7	Yes	Yes	Yes
C13	4.0	0.4	-18.6	-3.5	No	No	No
12							
13	1.8	0.2	-2.5	-3.3	Yes	Yes	Yes
HRPET	0.1	0.3	-1.8	-0.4	No	No	No
TPT	0.8	0.0	-1.3	-0.4	No	No	No

15 After 1000 hours of damp heat conditioning at 85 °C and 85% RH, it was found that looking through the front glass, certain modules had yellowed. It was found the yellowness was due to yellowing of the encapsulant and not the backsheet. As shown in Table 17, small changes in the b color value were noted on the back side (-3 b shift) but large changes in the b value were noted for the front side on certain modules. It was also found that cracks were appearing on selected backsheets in the modules treated with periods of thermal cycling and freezing after high humidity exposure.

20 Comparing the available MVTR data in Table 14, there is a correlation between the observed yellowing of the encapsulant and the MVTR. Lower MVTR corresponds to less yellowing. Cracking was not observed for mini modules fabricated from sheets with polyamide-ionomer formulations that contained talc.

25 Additional backsheets were prepared according to Table 18. Blend K in Table 18 is a mixture of 85 weight % of a 65:35 blend of high density polyethylene and ION-1 and 15 weight % of TiO<sub>2</sub> and UV stabilizers available from Mosaic Color and Additives, 110 Sulphur Springs Road, Greenville, SC under the commercial designation M002132WTPEP. The sheets were

tested for MVTR at 85 °C and 100 % relative humidity as described in the following procedure. MVTR was estimated by measuring the weight loss of water filled, flanged aluminum cups lidded with the backsheet in an 85°C air circulating oven. To minimize bulging of the backsheet from the water vapor pressure at 85°C, an 80 mesh stainless steel screen supported the backsheet on the non-water side of the cup. Backsheet and mesh screen were fixed in place by a retaining ring that bolted to the flange of the aluminum cup. The weight loss (water loss) from the cup was measured each day over the course of seven days. Typically after 24 hours of conditioning at 85°C, the daily mass loss was consistent. The average daily mass loss was then corrected for the surface area of the lidding backsheet to estimate the transmission rate. Each backsheet was tested in duplicate and the average of two measures reported. The nominal dimensions of the inside of the flanged aluminum cups, was diameter 76mm and the depth 50mm. The cups were typically half filled with water (150 ml of water added to the cup).

Table 18

	Outer skin layer	Tie layer	Core layer	Tie layer	Inner skin layer	MVTR (g/[m <sup>2</sup> -day])
14	50 µm Blend E	--	250 µm Blend 10	30 µm Tie-2	50 µm Blend K	56
15	50 µm Blend E	--	250 µm Blend 12	30 µm Tie-2	50 µm Blend K	46
16	300 µm Blend 13	--	--	30 µm Tie-2	50 µm Blend K	44
C14	300 µm Blend J	--	--	30 µm Tie-2	50 µm Blend K	49
17	300 µm Blend 13	--	--	30 µm Tie-1	50 µm Blend K	57
C15	300 µm Blend J	--	--	30 µm Tie-1	50 µm Blend K	66
18	200-250 µm Blend I	--	--	25-50 µm Tie-2	50-75 µm Blend H	56
19	50-55 µm Blend I	40-50 µm Tie-3	150-200 µm Blend I	40-50 µm Tie-3	50-55 µm Blend I	

CLAIMS

1. A polyamide-ionomer blend composition comprising
  - (i) A polymer component comprising
    - 5 1) 53 to 64 weight %, based on the combination of (1) and (2), of a polyamide component;
    - 2) 36 to 47 weight %, based on the combination of (1) and (2), of an anhydride ionomer component comprising a copolymer of
      - (a) ethylene;
      - (b) from 5 weight % to 15 weight % of an alpha, beta-unsaturated C<sub>3</sub>-C<sub>8</sub> carboxylic acid;
      - 10 (c) from 0.5 weight % to 12 weight % of at least one comonomer that is an ethylenically unsaturated dicarboxylic acid or derivative thereof selected from maleic acid, fumaric acid, itaconic acid, maleic anhydride, or a C<sub>1</sub>-C<sub>4</sub> alkyl half ester of maleic acid; and
      - (d) from 0 weight % to 30 weight % of monomers selected from alkyl acrylate or alkyl methacrylate, wherein the alkyl groups have from one to twelve carbon atoms; wherein the carboxylic acid
 15 functionalities present are at least partially neutralized to carboxylate salts comprising one or more alkali metal, transition metal, or alkaline earth metal cations;
      - (ii) 0 to 20 weight % of pigment; and
      - (iii) 0 to 40 weight % of filler; wherein the combination of (ii) and (iii) comprises 8 to 50 weight % of the combination of (i), (ii), (iii) and (iv); and
      - 20 (iv) 0 to 5 weight % of weathering additives selected from oxidation inhibitors, UV stabilizers and hindered amine light stabilizers.
2. The composition of claim 1 wherein the combination of (ii) and (iii) comprises 8 to 50 weight % of the combination of (i), (ii), (iii) and (iv).
3. The composition of claim 1 or claim 2 wherein the polyamide component comprises nylon 6, nylon 12, nylon 610,
 25 nylon 612, nylon 610/6T, nylon 612/6T, or combinations thereof.
4. The composition of any of claims 1, 2 or 3 wherein the comonomer of (c) is a C<sub>1</sub>-C<sub>4</sub> alkyl half ester of maleic acid.
5. The composition of any of claims 1, 2, 3 or 4 wherein the carboxylic acid functionalities present are at least partially neutralized to carboxylate salts comprising zinc or sodium, preferably zinc.
6. The composition of any of claims 1, 2, 3, 4 or 5 comprising 8 to 20 weight % of pigment and 0 weight % filler.
- 30 7. The composition of any of claims 1, 2, 3, 4 or 5 comprising 8 to 40 weight % of filler and 0 weight % of pigment.
8. The composition of any of claims 1, 2, 3, 4 or 5 comprising 10 to 40 weight % of filler and 8 to 15 weight % of pigment.
9. The composition of any of claims 1, 2, 3, 4 or 5 comprising 8 to 12 weight % of pigment and 12 to 18 weight % of filler.
- 35 10. The composition of any of claims 1, 2, 3, 4, 5, 6, 7, 8, or 9 wherein the pigment comprises titanium dioxide, zinc oxide, or antimony oxide, preferably titanium dioxide.
11. The composition of any of claims 1, 2, 3, 4, 5, 6, 7, 8, 9 or 10 wherein the filler comprises an inorganic oxide, carbonate, sulfate, silica, alkali and alkaline earth metal silicate, or baryte of a metal of Groups IA, IIA, IIIA, IIB, VIB or VIII of the periodic table of the elements, preferably wherein the filler comprises calcium carbonate, barium sulfate, wollastonite and
 40 talc, more preferably talc.
12. A backsheet for a photovoltaic module comprising the composition of any of claims 1, 2, 3, 4, 5, 6, 7, 8, 9, 10 or 11.

13. A laminated solar cell module comprising a front support layer formed of a light transmitting material and having first and second surfaces; a plurality of interconnected solar cells having a first surface facing the front support layer and a second surface facing away from the front support layer; a transparent encapsulant surrounding and encapsulating the interconnected solar cells, the transparent encapsulant being bonded to the second surface of the front support layer; and a backsheet of claim 12  
5 wherein one surface of the backsheet is bonded to the second surface of the transparent encapsulant.

14. An assembly for conversion under heat and pressure into a laminated solar cell module, the assembly comprising a front support layer formed of a light transmitting material and having front and back surfaces; a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; a plurality of interconnected solar cells having first and second surfaces adjacent to the first transparent encapsulant layer; a second transparent thermoplastic encapsulant layer disposed  
10 adjacent to the solar cells in parallel relation to the first transparent encapsulant layer; and a thermoplastic backsheet of claim 12.

15. A method of manufacturing a solar cell module comprising providing a front support layer formed of a light transmitting material and having front and back surfaces; placing a first transparent thermoplastic encapsulant layer adjacent to the back surface of the front support layer; positioning a plurality of interconnected solar cells having first and second surfaces so that the first surfaces thereof are adjacent to the first transparent encapsulant layer; placing a second transparent thermoplastic  
15 encapsulant layer adjacent to the second surfaces of the solar cells; placing a backsheet of claim 12 adjacent to the second transparent thermoplastic encapsulant layer to thereby form an assembly; subjecting the assembly to heat and pressure so as to melt the encapsulant layers and cause the encapsulant to surround the solar cells, and cooling the assembly so as to cause the encapsulant to solidify and bond to the front support layer, the solar cells and the backsheet, thereby laminating the layers and the solar cells together to form an integrated solar cell module.

## INTERNATIONAL SEARCH REPORT

International application No  
PCT/US2015/027917

## A. CLASSIFICATION OF SUBJECT MATTER

INV. C08L77/02 H01L31/049 C08L23/08  
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

H01L C08L C08K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, WPI Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 2011/011577 A1 (DU PONT [US]; CHOU RICHARD T [US]; BENDLER HERBERT VERNON [US]) 27 January 2011 (2011-01-27)	1-5,7-9
Y	claims 1,11,12 examples 8,12,16,22,26	1-15
Y	----- EP 2 196 489 A1 (ARKEMA FRANCE [FR]; ETIMEX SOLAR GMBH [DE] ARKEMA FRANCE [FR]; SOLUTIA) 16 June 2010 (2010-06-16) example 1 claims 11-19	1-15
Y	----- US 2009/298372 A1 (CHOU RICHARD T [US] ET AL) 3 December 2009 (2009-12-03) examples 2,6; table 1 paragraph [0036] -----	1-5



Further documents are listed in the continuation of Box C.



See patent family annex.

## \* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

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"&" document member of the same patent family

Date of the actual completion of the international search

23 June 2015

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# INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 2011011577	A1	27-01-2011	CN 102471573 A 23-05-2012
			EP 2456821 A1 30-05-2012
			JP 2013500356 A 07-01-2013
			KR 20120037496 A 19-04-2012
			US 2011020573 A1 27-01-2011
			WO 2011011577 A1 27-01-2011
-----			
EP 2196489	A1	16-06-2010	CN 102257043 A 23-11-2011
			EP 2196489 A1 16-06-2010
			ES 2433720 T3 12-12-2013
			JP 5532054 B2 25-06-2014
			JP 2012512523 A 31-05-2012
			KR 20110118629 A 31-10-2011
			PT 2196489 E 30-09-2013
			US 2011315199 A1 29-12-2011
			WO 2010069546 A2 24-06-2010
-----			
US 2009298372	A1	03-12-2009	NONE
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