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(54) Titre : PROCEDE CHIMIQUE PAR VOIE HUMIDE POUR FORMER DES OXYDES DIESEL PTPD STABLES
(54) Title: WET CHEMICAL METHOD OF FORMING STABLE PTPD DIESEL OXIDATION

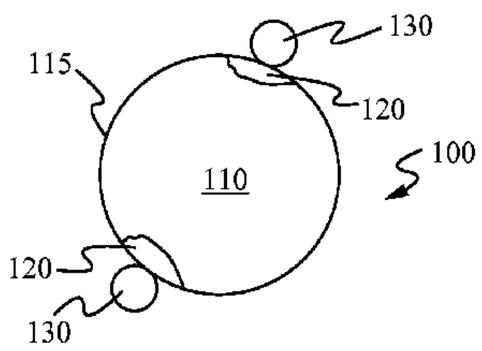

Fig. 1

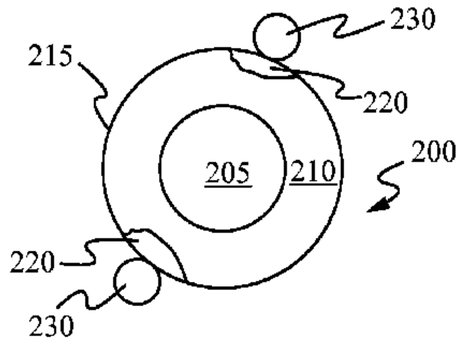

Fig. 2

(57) Abrégé/Abstract:
A nanoparticle comprising: an interior region comprising a mixed-metal oxide; and an exterior surface comprising a pure metal. In some embodiments, the mixed-metal oxide comprises aluminum oxide and a metallic pinning agent. In some embodiments, the pure metal at the exterior surface is the same as the metallic pinning agent in the mixed-metal oxide in the interior region. In some embodiments, a catalytic nanoparticle is bonded to the pure metal at the exterior surface. In some embodiments, the interior region and the exterior surface are formed using a plasma gun. In some embodiments, the interior region and the exterior surface are formed using a wet chemistry process. In some embodiments, the catalytic nanoparticle is bonded to the pure metal using a plasma gun. In some embodiments, the catalytic nanoparticle is bonded to the pure metal using a wet chemistry process.

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(54) Title: WET CHEMICAL METHOD OF FORMING STABLE PtPd DIESEL OXIDATION

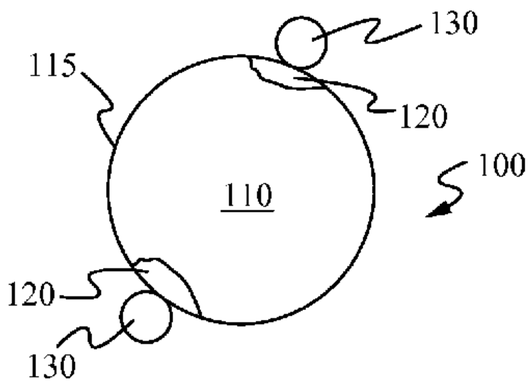

Fig. 1

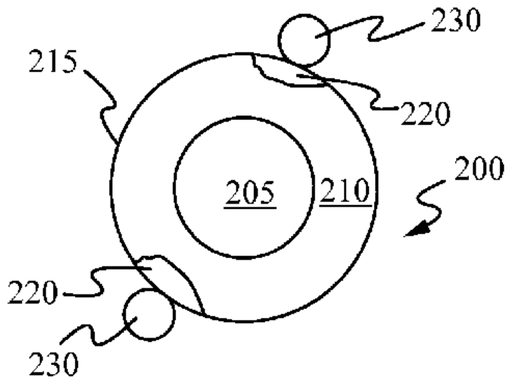

Fig. 2

(57) Abstract: A nanoparticle comprising: an interior region comprising a mixed-metal oxide; and an exterior surface comprising a pure metal. In some embodiments, the mixed-metal oxide comprises aluminum oxide and a metallic pinning agent. In some embodiments, the pure metal at the exterior surface is the same as the metallic pinning agent in the mixed-metal oxide in the interior region. In some embodiments, a catalytic nanoparticle is bonded to the pure metal at the exterior surface. In some embodiments, the interior region and the exterior surface are formed using a plasma gun. In some embodiments, the interior region and the exterior surface are formed using a wet chemistry process. In some embodiments, the catalytic nanoparticle is bonded to the pure metal using a plasma gun. In some embodiments, the catalytic nanoparticle is bonded to the pure metal using a wet chemistry process.

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WET CHEMICAL METHOD OF FORMING STABLE PtPd DIESEL OXIDATION

FIELD OF THE INVENTION

The present invention relates to the field of catalysts. More specifically, the present invention relates to a method of producing a catalyst.

BACKGROUND OF THE INVENTION

Wet chemistry can be used to put small platinum particles on a ceramic surface in order to create a catalyst. However, when the catalyst is heated up and used, the platinum particles are not sufficiently bonded to the ceramic surface. As a result, they are mobile, allowing them to move around and find other platinum particles. Upon the platinum particles' discovery of each other, they agglomerate into bigger platinum particles.

For example, even if prior art processes start with ½ nanometer platinum particles, by the time the aging test has been performed (equivalent to 10 years on a car), the platinum particles have agglomerated to become 16 nanometer particles.

This agglomeration of platinum particles results in a massive reduction in catalytic surface area. As a result, the performance of the catalyst significantly decreases.

SUMMARY OF THE INVENTION

The present invention employs a pinning agent to securely bond catalytic material to a core nano-particle. The pinning agent securely anchors the catalytic material to the core nano-particle, thereby minimizing or eliminating the mobility and agglomeration of the catalytic material. As a result, the present invention is able to provide a minimally-aging or even a non-aging catalyst that substantially maintains the extremely small size of the catalytic particles on each core nano-particle, thereby maximizing the catalytic surface area.

In one aspect of the present invention, a nano-particle is comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide, wherein the mixed-metal oxide comprises aluminum oxide and palladium. The exterior surface comprises pure metallic palladium.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the nano-particle comprises a diameter of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle. In some embodiments, the center core comprises silica.

In another aspect of the present invention, a method of forming nano-particles comprises: loading a quantity of aluminum oxide material and a quantity of palladium material into a plasma gun in a desired ratio; vaporizing the quantity of aluminum oxide material and the quantity of palladium material using the plasma gun, thereby forming a vapor cloud comprising vaporized aluminum oxide and vaporized palladium; and quenching the vapor cloud, thereby condensing the vaporized aluminum oxide and the vaporized palladium into nano-particles. Each nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium.

In some embodiments, the step of vaporizing the quantity of aluminum oxide material and the quantity of palladium material comprises: flowing a working gas into a reactor of the plasma gun; delivering energy to the working gas, thereby forming a plasma stream; and flowing the quantity of aluminum oxide material and the quantity of palladium material into the plasma stream.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the nano-particles have an average grain size of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In yet another aspect of the present invention, a method of forming nano-particles comprises: providing a quantity of aluminum material and a quantity of palladium material in a desired ratio; and forming a plurality of nano-particles from the quantity of aluminum material and the quantity of palladium material using a wet chemistry process. Each formed nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium.

In some embodiments, the aluminum material is a salt or an organic compound, the palladium material is a salt, and the wet chemistry process comprises forming the mixed-

metal oxide from the aluminum material and the palladium material via a chemical reaction. In some embodiments, the mixed-metal oxide forms a center core of each nano-particle, and the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the step of forming the plurality of nano-particles using a wet chemistry process comprises forming a monolayer around a center core of each nano-particle, the monolayer comprising the mixed-metal oxide. In some embodiments, the center core of each nano-particle comprises silica.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the nano-particles have an average grain size of approximately 10 nanometers or less.

In yet another aspect of the present invention, a catalytic carrier nano-particle comprises a carrier nano-particle comprising an interior region and an exterior surface. The interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium. A catalytic nano-particle is bonded to the pure metallic palladium at the exterior surface. The catalytic nano-particle is platinum.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the carrier nano-particle comprises a diameter of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the carrier nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the mixed-metal oxide forms a monolayer that surrounds a center core of the carrier nano-particle. The nano-particle of Claim 26, wherein the center core comprises silica.

In yet another aspect of the present invention, a method of forming nano-particles comprises: loading a quantity of aluminum oxide material, a quantity of palladium material, and a quantity of platinum material into a plasma gun in a desired ratio; vaporizing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material using the plasma gun, thereby forming a vapor cloud comprising vaporized aluminum oxide, vaporized palladium, and vaporized platinum; and quenching the vapor cloud, thereby condensing the vaporized aluminum oxide, the vaporized palladium, and the vaporized platinum into carrier nano-particles. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide

that includes aluminum oxide and palladium, the exterior surface comprises a pure metallic palladium, and a platinum nano-particle is bonded to the pure metallic palladium at the exterior surface.

In some embodiments, the step of vaporizing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material comprises: flowing a working gas into a reactor of the plasma gun; delivering energy to the working gas, thereby forming a plasma stream; and flowing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material into the plasma stream.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the carrier nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In yet another aspect of the present invention, a method of forming nano-particles comprises: providing a quantity of aluminum material, a quantity of palladium material, and a quantity of platinum material in a desired ratio; and forming a plurality of carrier nano-particles from the quantity of aluminum material, the quantity of palladium material, and the quantity of platinum material using a wet chemistry process. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, the exterior surface comprises a pure metallic palladium, and a platinum nano-particle is bonded to the pure metallic palladium at the exterior surface.

In some embodiments, the aluminum material is a salt or an organic compound, the palladium material is a salt, and the wet chemistry process comprises forming the mixed-metal oxide from the aluminum material and the palladium material via a chemical reaction, and forming a carrier particle suspension comprising a plurality of aluminum-palladium carrier particles. Each aluminum-palladium carrier particle comprises an interior region and an exterior surface. The interior region comprises the mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium. In some embodiments, the wet chemistry process further comprises mixing a platinum solution with the carrier particle suspension, and bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the aluminum-palladium carrier

particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution.

In some embodiments, the aluminum material is a salt or an organic compound, the palladium material is a salt, and the wet chemistry process comprises forming the mixed-metal oxide from the aluminum material and the palladium material via a chemical reaction, forming a monolayer of the mixed-metal oxide around a center core of each carrier nano-particle, and forming a suspension of the carrier nano-particles. The center core comprises a different chemical composition from the monolayer, and each monolayer comprises an interior region and an exterior surface. The interior region comprises the mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium. In some embodiments, the wet chemistry process further comprises mixing a platinum solution with the suspension of nano-particles, and bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the carrier nano-particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution. In some embodiments, the center core of each carrier nano-particle comprises silica.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium-platinum alloy at the exterior surface. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

In yet another aspect of the present invention, a method of forming nano-particles comprises: forming a plurality of aluminum-palladium carrier particles using a plasma gun, wherein each aluminum-palladium carrier particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium; and affixing a platinum nano-particle to the pure metallic palladium on each of the aluminum-palladium carrier particles using a wet chemistry process.

In some embodiments, forming the plurality of aluminum-palladium carrier particles using the plasma gun comprises: loading a quantity of aluminum oxide material and a quantity of palladium material into the plasma gun in a desired ratio; vaporizing the quantity of aluminum oxide material and the quantity of palladium material using the plasma gun, thereby forming a vapor cloud comprising vaporized aluminum oxide and vaporized palladium; and quenching the vapor cloud, thereby condensing the vaporized aluminum oxide and the vaporized palladium into carrier nano-particles. Each carrier nano-particle comprises

an interior region and an exterior surface. The interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium.

In some embodiments, the step of vaporizing the quantity of aluminum oxide material and the quantity of palladium material comprises flowing a working gas into a reactor of the plasma gun, delivering energy to the working gas, thereby forming a plasma stream, and flowing the quantity of aluminum oxide material and the quantity of palladium material into the plasma stream. In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the aluminum-palladium carrier particles have an average grain size of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of each aluminum-palladium carrier particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the wet chemistry process comprises mixing a platinum solution with the aluminum-palladium carrier particles, and bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the aluminum-palladium carrier particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution.

In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the aluminum-palladium carrier particles have an average grain size of approximately 10 nanometers or less.

In yet another aspect of the present invention, a catalyst comprises a support structure, and a plurality of carrier nano-particles bonded to the support structure. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide, and the exterior surface comprises a pure metal. A catalytic particle is bonded with the pure metal at the exterior surface.

In some embodiments, the support structure is a porous ceramic material, and the plurality of carrier nano-particles are disposed within the pores of the ceramic material. In some embodiments, the mixed-metal oxide comprises aluminum oxide and palladium. In

some embodiments, the pure metal is palladium. In some embodiments, the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of each carrier nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the mixed-metal oxide forms a monolayer that surrounds a center core of each carrier nano-particle. In some embodiments, the center core comprises silica.

In some embodiments, the exterior surface comprises a pure metal alloy. In some embodiments, the catalytic particle is platinum. In some embodiments, the catalyst is a non-aging catalyst.

In yet another aspect of the present invention, a method of forming a catalyst comprises: providing a support structure; providing a plurality of carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal, wherein a catalytic particle is bonded with the pure metal at the exterior surface; and bonding the plurality of carrier nano-particles to the support structure.

In some embodiments, the support structure is a porous ceramic material, and the plurality of carrier nano-particles are disposed within the pores of the ceramic material. In some embodiments, the mixed-metal oxide includes aluminum oxide and palladium, and the pure metal comprises pure metallic palladium. In some embodiments, the catalytic particle is platinum. In some embodiments, the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the carrier nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the step of providing a plurality of carrier nano-particles comprises: loading a quantity of a first material, a quantity of a second material, and a

quantity of a third material into a plasma gun in a desired ratio; vaporizing the quantity of the first material, the quantity of the second material, and the quantity of the third material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material, vaporized second material, and vaporized third material; and quenching the vapor cloud, thereby condensing the vaporized first material, the vaporized second material, and the vaporized third material into carrier nano-particles. Each carrier nano-particle comprises the interior region and the exterior surface. In some embodiments, the step of vaporizing the quantity of the first material, the quantity of the second material, and the quantity of the third material comprises flowing a working gas into a reactor of the plasma gun, delivering energy to the working gas, thereby forming a plasma stream, and flowing the quantity of the first material, the quantity of the second material, and the quantity of the third material into the plasma stream.

In some embodiments, the step of providing a plurality of carrier nano-particles comprises: providing a quantity of a first material, a quantity of a second material, and a quantity of third material in a desired ratio; and forming the plurality of carrier nano-particles from the quantity of the first material, the quantity of the second material, and the quantity of the third material using a wet chemistry process. In some embodiments, the mixed-metal oxide comprises the first material and the second material, the exterior surface comprises the second material, and the catalytic particle comprises the third material. In some embodiments, the wet chemistry process comprises forming the mixed-metal oxide from the first material and the second material via a chemical reaction, and forming a carrier particle suspension comprising a plurality of carrier particles. Each carrier particle comprises the interior region and the exterior surface, the mixed-metal oxide of the interior region comprising the first material and the second material, and the pure metal at the exterior surface comprising the second material.

In some embodiments, the first material is an aluminum material, and the second material is a palladium material. In some embodiments, the aluminum material is a salt or an organic compound, and the palladium material is a salt.

In some embodiments, the wet chemistry process further comprises mixing a catalytic solution with the carrier particle suspension, and bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the carrier particles. In some embodiments,

the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution.

In some embodiments, forming the mixed-metal oxide comprises forming a monolayer around a center core of each carrier particle. The center core comprises a different chemical composition from the monolayer, and each monolayer comprises the interior region and the exterior surface. In some embodiments, the mixed-metal oxide includes aluminum oxide and palladium, and the pure metal at the exterior surface comprises palladium. In some embodiments, the wet chemistry process further comprises mixing a catalytic solution with the carrier particle suspension, and bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the carrier particles. In some embodiments, the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution. In some embodiments, the center core of each carrier nano-particle comprises silica.

In some embodiments, the step of providing the plurality of carrier nano-particles comprises: forming a plurality of carrier particles using a plasma gun, wherein each carrier particle comprises the interior region and the exterior surface; and affixing a catalytic nano-particle to the pure metal on each of the carrier particles using a wet chemistry process. In some embodiments, the step of forming the plurality of carrier particles using the plasma gun comprises: loading a quantity of a first material and a quantity of a second material into the plasma gun in a desired ratio; vaporizing the quantity of the first material and the quantity of the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material; and quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into the carrier particles. In some embodiments, the step of vaporizing the quantity of first material and the quantity of second material comprises: flowing a working gas into a reactor of the plasma gun; delivering energy to the working gas, thereby forming a plasma stream; and flowing the quantity of first material and the quantity of second material into the plasma stream. In some embodiments, the step of affixing a catalytic nano-particle to the pure metal using a wet chemistry process comprises mixing a catalytic solution with the core particles, and bonding catalytic nano-particles formed from the catalytic solution to the pure metal on

the carrier particles. In some embodiments, the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution. In some embodiments, bonding the plurality of carrier nano-particles to the support structure comprises performing a calcining process.

In yet another aspect of the present invention, a nano-particle comprises an interior region comprising a mixed-metal oxide, and an exterior surface comprising a pure metal.

In some embodiments, the mixed-metal oxide comprises aluminum oxide and palladium. In some embodiments, the pure metal is palladium. In some embodiments, the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface. In some embodiments, the nano-particle comprises a diameter of approximately 10 nanometers or less.

In some embodiments, the mixed-metal oxide forms a center core of the nano-particle. In some embodiments, the mixed-metal oxide consists only of aluminum oxide and palladium.

In some embodiments, the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle. In some embodiments, the center core comprises silica. In some embodiments, the exterior surface comprises a pure metal alloy. In some embodiments, the pure metal alloy is a palladium-platinum alloy.

In yet another aspect of the present invention, a method of forming nano-particles comprises providing a quantity of a first material and a quantity of a second material, and forming a plurality of nano-particles from the quantity of the first material and the quantity of the second material. Each nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide, and the exterior surface comprises a pure metal.

In some embodiments, the mixed-metal oxide comprises the first material and the second material, and the pure metal comprises the second material. In some embodiments, the mixed-metal oxide comprises aluminum oxide and palladium, and the pure metal comprises palladium. In some embodiments, the mixed-metal oxide comprises aluminum oxide and palladium, and the pure metal comprises a palladium-platinum alloy.

In some embodiments, the step of forming the plurality of nano-particles comprises: loading the quantity of the first material and the quantity of the second material into a plasma gun in a desired ratio; vaporizing the quantity of the first material and the quantity of the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material; and quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into nano-particles, wherein each nano-particle comprises the interior region and the exterior surface. In some embodiments, the step of vaporizing the quantity of the first material and the quantity of the second material comprises: flowing a working gas into a reactor of the plasma gun; delivering energy to the working gas, thereby forming a plasma stream; and flowing the quantity of the first material and the quantity of the second material into the plasma stream. In some embodiments, the mixed-metal oxide comprises the first material and the second material, and the pure metal comprises the second material. In some embodiments, the first material is an aluminum material and the second material is a palladium material.

In some embodiments, the step of forming the plurality of nano-particles comprises performing a wet chemistry process with the quantity of first material and the quantity of second material. In some embodiments, the wet chemistry process comprises forming the mixed-metal oxide from the first material and the second material via a chemical reaction. In some embodiments, the first material comprises an aluminum material, and the second material comprises a palladium material. In some embodiments, the aluminum material is a salt or an organic compound, and the palladium material is a salt. In some embodiments, the mixed-metal oxide forms a center core of each nano-particle, and the mixed-metal oxide consists only of aluminum oxide and palladium. In some embodiments, the wet chemistry process comprises forming a monolayer around a center core of each nano-particle. The monolayer comprises the mixed-metal oxide. In some embodiments, the center core of each nano-particle comprises silica.

In some embodiments, the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface. In some embodiments, the nano-particles have an average grain size of approximately 10 nanometers or less.

In yet another aspect of the present invention, a nano-particle comprises an interior region comprising a mixed-metal oxide, an exterior surface comprising a pure metal, and a

catalytic particle affixed to the pure metal. The catalytic particle has a chemical composition different from the pure metal and the mixed-metal oxide.

In some embodiments, the mixed-metal oxide comprises a first material and a second material, the pure metal comprises the second material, and the catalytic particle comprises a third material. In some embodiments, the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle. In some embodiments, the center core comprises silica.

In some embodiments, the mixed-metal oxide comprises aluminum-oxide and a metallic pinning agent that does not comprise platinum, the pure metal comprises the metallic pinning agent, and the catalytic particle comprises platinum. In some embodiments, the metallic pinning agent comprises copper, molybdenum, or cobalt. In some embodiments, the catalytic particle comprises a metal alloy. In some embodiments, the metal alloy is a palladium-platinum alloy. In some embodiments, the metallic pinning agent comprises copper, molybdenum, or cobalt, and the catalytic particle comprises a palladium-platinum alloy.

In some embodiments, the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface. In some embodiments, the nano-particle comprises a diameter of approximately 10 nanometers or less.

In some embodiments, the catalytic particle is a nano-particle. In some embodiments, the catalytic particle comprises a diameter of approximately 1 nanometer or less. In some embodiments, the catalytic particle comprises a diameter of approximately ½ nanometer.

In yet another aspect of the present invention, a method of forming a nano-particle comprising: providing a carrier nano-particle, wherein the carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal; and affixing a catalytic particle to the pure metal, wherein the catalytic particle has a chemical composition different from the pure metal and the mixed-metal oxide.

In some embodiments, the mixed-metal oxide is formed from a first material and a second material, the pure metal is formed from the second material, and the catalytic particle is formed from a third material. In some embodiments, the mixed-metal oxide forms a

monolayer that surrounds a center core of the carrier nano-particle. In some embodiments, the center core comprises silica.

In some embodiments, the mixed-metal oxide comprises aluminum-oxide and a metallic pinning agent that does not comprise platinum, the pure metal comprises the metallic pinning agent, and the catalytic particle comprises platinum. In some embodiments, the metallic pinning agent comprises copper, molybdenum, or cobalt. In some embodiments, the catalytic particle comprises a metal alloy. In some embodiments, the metal alloy is a palladium-platinum alloy. In some embodiments, the metallic pinning agent comprises copper, molybdenum, or cobalt, and the catalytic particle comprises a palladium-platinum alloy.

In some embodiments, the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface. In some embodiments, the carrier nano-particle comprises a diameter of approximately 10 nanometers or less.

In some embodiments, the catalytic particle is a nano-particle. In some embodiments, the catalytic particle comprises a diameter of approximately 1 nanometer or less. In some embodiments, the catalytic particle comprises a diameter of approximately ½ nanometer.

In some embodiments, the steps of providing the carrier nano-particle and affixing the catalytic particle to the pure metal are performed using a plasma gun. In some embodiments, the steps of providing the carrier nano-particle and affixing the catalytic particle to the pure metal are performed using a wet chemistry process.

In some embodiments, providing the carrier nano-particle comprises loading a quantity of a first material and a quantity of a second material into a plasma gun in a desired ratio, vaporizing the quantity of the first material and the quantity of the second material, thereby forming a vapor cloud comprising vaporized first material and vaporized second material, and quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into the carrier nano-particle. The carrier nano-particle comprises the interior region and the exterior surface. In some embodiments, the step of vaporizing the quantity of the first material material and the quantity of the second material comprises flowing a working gas into a reactor of the plasma gun, delivering energy to the working gas, thereby forming a plasma stream, and flowing the quantity of the first material and the quantity of the second material into the plasma stream. In some embodiments, the

step of affixing the catalytic particle to the pure metal is performed using a wet chemistry process. In some embodiments, the wet chemistry process comprises mixing a catalytic solution with the carrier nano-particle, and bonding a catalytic particle formed from the catalytic solution to the pure metal on the carrier nano-particle. In some embodiments, the catalytic solution is a platinum solution and the catalytic particle is a platinum nano-particle. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution.

In yet another aspect of the present invention, a method of forming catalytic nano-particles comprises: providing a plurality of carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal; and affixing a catalytic nano-particle to the pure metal on each carrier nano-particle.

In some embodiments, the step of affixing the catalytic nano-particle to the pure metal is performed using a plasma gun. In some embodiments, the step of affixing the catalytic nano-particle to the pure metal comprises: loading the plurality of nano-particles and a quantity of catalytic material into a plasma gun in a desired ratio, vaporizing the quantity of catalytic material, thereby forming a vapor cloud comprising vaporized catalytic material, and quenching the vapor cloud, thereby condensing the vaporized catalytic material into catalytic nano-particles that are bonded to the pure metal on the carrier nano-particles.

In some embodiments, the plurality of carrier nano-particles has an average grain size of approximately 10 nanometers or less. In some embodiments, each catalytic nano-particle comprises a diameter of approximately 1 nanometer or less. In some embodiments, each catalytic nano-particle comprises a diameter of approximately ½ nanometer.

In some embodiments, the step of affixing the catalytic nano-particle to the pure metal is performed using a wet chemistry process. In some embodiments, the step of affixing a catalytic nano-particle to the pure metal using a wet chemistry process comprises mixing a catalytic solution with the plurality of carrier nano-particles, and bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the plurality of carrier nano-particles. In some embodiments, the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 illustrates a cross-sectional view of one embodiment of catalytic nano-particles bonded with a carrier nano-particle in accordance with the principles of the present invention.

FIG. 2 illustrates a cross-sectional view of another embodiment of catalytic nano-particles bonded with a carrier nano-particle in accordance with the principles of the present invention.

FIG. 3A illustrates a cross-sectional view of one embodiment of a catalyst in accordance with the principles of the present invention.

FIG. 3B illustrates a cross-sectional view of another embodiment of a catalyst in accordance with the principles of the present invention.

FIG. 4 illustrates one embodiment of a method of producing catalytic carrier nano-particles and a catalyst therefrom in accordance with the principles of the present invention.

FIG. 5 illustrates one embodiment of a method of producing catalytic carrier nano-particles using a plasma gun in accordance with the principles of the present invention.

FIG. 6 illustrates another embodiment of a method of producing catalytic carrier nano-particles using a plasma gun in accordance with the principles of the present invention.

FIG. 7 illustrates one embodiment of a plasma-based particle production system in accordance with the principles of the present invention.

FIG. 8 illustrates another embodiment of a plasma based particle production system in accordance with the principles of the present invention.

FIG. 9 illustrates one embodiment of a method of catalytic carrier nano-particles using a wet chemistry process in accordance with the principles of the present invention.

FIG. 10 illustrates one embodiment of a method of producing catalytic carrier nano-particles using both a plasma gun and a wet chemistry process in accordance with the principles of the present invention.

DETAILED DESCRIPTION OF THE INVENTION

The following description is presented to enable one of ordinary skill in the art to make and use the invention and is provided in the context of a patent application and its requirements. Various modifications to the described embodiments will be readily apparent to those skilled in the art and the generic principles herein may be applied to other

embodiments. Thus, the present invention is not intended to be limited to the embodiment shown but is to be accorded the widest scope consistent with the principles and features described herein.

This disclosure provides several embodiments of the present invention. It is contemplated that any features from any embodiment can be combined with any features from any other embodiment. In this fashion, hybrid configurations of the illustrated and describer embodiments are well within the scope of the present invention.

This disclosure refers to both particles and powders. These two terms are equivalent, except for the caveat that a singular "powder" refers to a collection of particles. The present invention may apply to a wide variety of powders and particles. Powders that fall within the scope of the present invention may include, but are not limited to, any of the following: (a) nano-structured powders (nano-powders), having an average grain size less than 250 nanometers and an aspect ratio between one and one million; (b) submicron powders, having an average grain size less than 1 micron and an aspect ratio between one and one million; (c) ultra-fine powders, having an average grain size less than 100 microns and an aspect ratio between one and one million; and (d) fine powders, having an average grain size less than 500 microns and an aspect ratio between one and one million.

Various aspects of the disclosure may be described through the use of flowcharts. Often, a single instance of an aspect of the present disclosure may be shown. As is appreciated by those of ordinary skill in the art, however, the protocols, processes, and procedures described herein may be repeated continuously or as often as necessary to satisfy the needs described herein. Additionally, it is contemplated that certain method steps of the invention can be performed in alternative sequences to those disclosed in the flowcharts. Accordingly, the scope of the claims should not be limited to any specific order of method steps unless the order is explicitly required by the language of the claims.

FIG. 1 illustrates a cross-sectional view of one embodiment of catalytic nano-particles 130 bonded with a carrier nano-particle 100 in accordance with the principles of the present invention. The carrier nano-particle 100 comprises an interior region and an exterior surface 115. The exterior surface 115 defines the perimeter of the carrier nano-particle 100. In some embodiments, the carrier nano-particle 100 comprises a diameter of approximately 10

nanometers or less, which has been found to be an ideal size. Although, it is contemplated that other diameters are within the scope of the present invention.

The interior region is the region between the center of the carrier particle and the exterior surface 115. This interior region comprises a mixed-metal oxide 110. In some embodiments, the mixed-metal oxide 110 comprises aluminum oxide along with an accompanying metal. In some embodiments, this accompanying metal is palladium. Other accompanying metals that have been found to be suitable are copper, molybdenum, and cobalt. It is contemplated that other oxides and other metals may be used as well.

The exterior surface comprises one or more pure metal regions 120 consisting of pure metal. Metals that have been found to be suitable for these pure metal regions include palladium, copper, molybdenum, and cobalt. However, it is contemplated that other metals may be used as well. For the purposes of this disclosure, the terms "pure metal" and "pure metallic" should be interpreted to mean that the regions consist only of metal and are characterized by the absence of any other type of material. For example, the pure metal regions 120 may comprise a metal or a metal-metal alloy, but they may not comprise any ceramic material. The purpose of these purely metal regions 120 is to provide exposed regions at the exterior surface 115 of the carrier nano-particle 100 to which catalytic nano-particles 130 will have a strong attraction.

The catalytic nano-particles 130 are bonded, or otherwise affixed, to these pure metal regions 120. In some embodiments, these catalytic nano-particles 130 comprise or consist of platinum. In some embodiments, these catalytic nano-particles 130 comprise or consist of a platinum alloy, such as a platinum-palladium alloy. However, it is contemplated that other catalytic materials may be used as well. In some embodiments, the catalytic nano-particles 130 have an average grain size of approximately 1 nanometer or less. In some embodiments, the catalytic nano-particles 130 have an average grain size of approximately ½ nanometer. However, it is contemplated that other sizes are within the scope of the present invention. Although FIG. 1 shows more than one pure metal region 120 on the carrier nano-particle 110 and more than one catalytic nano-particle 130 bonded to the carrier nano-particle 110, it is contemplated that other quantities of these elements are within the scope of the present invention. For example, in some embodiments, the carrier nano-particle 110 comprises only one distinct pure metal region 120 and one distinct catalytic nano-particle 130 bonded to the

carrier nano-particle 110. In some embodiments, the carrier nano-particle 110 comprises more than one distinct pure metal region 120 and more than one distinct catalytic nano-particle 130 bonded to the carrier nano-particle 110.

The pure metal regions 120 are bonded to the mixed-metal oxide 110 and act as pinning agents to securely affix the catalytic nano-particles 130 to the carrier nano-particle100, thereby reducing or eliminating the mobility of the catalytic nano-particles 130 and preventing their agglomeration.

FIG. 2 illustrates a cross-sectional view of one embodiment of catalytic nano-particles 230 bonded with a carrier nano-particle 200 in accordance with the principles of the present invention. The carrier nano-particle 200 comprises an interior region and an exterior surface 215. The exterior surface 215 defines the perimeter of the carrier nano-particle 200. In some embodiments, the carrier nano-particle 200 comprises a diameter of approximately 10 nanometers or less, which has been found to be an ideal size. Although, it is contemplated that other diameters are within the scope of the present invention.

The interior region is the region between the center of the carrier particle and the exterior surface 215. This interior region comprises a mixed-metal oxide 210. In some embodiments, the mixed-metal oxide 210 comprises aluminum oxide along with an accompanying metal. In some embodiments, this accompanying metal is palladium. Other accompanying metals that have been found to be suitable are copper, molybdenum, and cobalt. It is contemplated that other oxides and other metals may be used as well.

The embodiment of FIG. 2 is very similar to the embodiment of FIG. 1. However, in contrast to the interior region of carrier nano-particle 100, which is shown in FIG. 1 as comprising only the mixed-metal oxide 110 from its center to its exterior surface 115 and pure metal regions 120, the interior region of carrier nano-particle 200 is shown in FIG. 2 as having a center core 205 around which a monolayer of the mixed-metal oxide 210 is formed. In some embodiments, this center core 205 is formed from a ceramic material. In some embodiments, this center core 205 is formed from silica. However, it is contemplated that a other materials may be suitable to form the center core 205.

Similar to the exterior surface of carrier nano-particle 100, the exterior surface of carrier nano-particle 200 comprises one or more pure metal regions 220 consisting of pure metal. Metals that have been found to be suitable for these pure metal regions include

palladium, copper, molybdenum, and cobalt. However, it is contemplated that other metals may be used as well. For the purposes of this disclosure, the terms "pure metal" and "pure metallic" should be interpreted to mean that the regions consist only of metal and are characterized by the absence of any other type of material. For example, the pure metal regions 220 may comprise a metal or a metal-metal alloy, but they may not comprise any ceramic material. The purpose of these purely metal regions 220 is to provide exposed regions at the exterior surface 215 of the carrier nano-particle 200 to which catalytic nano-particles 230 will have a strong attraction.

As discussed above with respect to the embodiment of FIG. 1, the catalytic nano-particles 230 are bonded, or otherwise affixed, to these pure metal regions 220. In some embodiments, these catalytic nano-particles 230 comprise or consist of platinum. In some embodiments, these catalytic nano-particles 230 comprise or consist of a platinum alloy, such as a platinum-palladium alloy. However, it is contemplated that other catalytic materials may be used as well. In some embodiments, the catalytic nano-particles 230 have an average grain size of approximately 1 nanometer or less. In some embodiments, the catalytic nano-particles 230 have an average grain size of approximately ½ nanometer. However, it is contemplated that other sizes are within the scope of the present invention. Although FIG. 2 shows more than one pure metal region 220 on the carrier nano-particle 210 and more than one catalytic nano-particle 230 bonded to the carrier nano-particle 210, it is contemplated that other quantities of these elements are within the scope of the present invention. For example, in some embodiments, the carrier nano-particle 210 comprises only one distinct pure metal region 220 and one distinct catalytic nano-particle 230 bonded to the carrier nano-particle 210. In some embodiments, the carrier nano-particle 210 comprises more than one distinct pure metal region 220 and more than one distinct catalytic nano-particle 230 bonded to the carrier nano-particle 210.

As previously discussed with respect to the embodiment of FIG. 1, the pure metal regions 220 are bonded to the mixed-metal oxide 210 and act as pinning agents to securely affix the catalytic nano-particles 230 to the carrier nano-particle200, thereby reducing or eliminating the mobility of the catalytic nano-particles 230 and preventing their agglomeration.

Whether it is with respect to the embodiments of FIGS. 1-2 or some other embodiment within the scope of the present invention, the chemical and structural compositions of the elements can vary. For example, in some embodiments, the mixed-metal oxide comprises or consists of aluminum oxide, the pure metal regions comprise or consist of palladium, and the catalytic particles comprise or consist of platinum. In some embodiments, the interior region consists only of the mixed-metal oxide. In some embodiments, the interior region comprises a ceramic center core surrounded by a monolayer of mixed-metal oxide. In some embodiments, the pure metal comprises or consists of a metal other than palladium, such as copper, molybdenum, or cobalt. In some embodiments, the catalytic particles comprise or consist of a palladium-platinum alloy. In some embodiments, the pure metal comprises or consists of a metal other than palladium, such as copper, molybdenum, or cobalt, and the catalytic particles comprise or consist of a palladium-platinum alloy.

Whether it is with respect to the embodiments of FIGS. 1-2 or some other embodiment within the scope of the present invention, the bonding of one or more of the catalytic nano-particles with the carrier nano-particle defines the formation of catalytic carrier nano-particles. These catalytic carrier nano-particles can be impregnated or otherwise affixed onto a catalyst support in order to form a usable catalyst. In some embodiments, the catalyst support is a ceramic material. In some embodiments, the catalyst support is an oxide material. In some embodiments, the catalyst support is a monolith. In some embodiments, the catalyst support is an extrudate. Such supports have found utility due to their highly accessible and large surface area, as high as 250 m^2/g. In some embodiments, the catalyst support is a macroscopic support particle. In such an embodiment, the size of the macroscopic support particle is selected to provide maximum surface area to which the catalytic carrier nano-particles are bonded or fixed. In some embodiments, the catalyst support is a porous structure.

In some embodiments, the mixed-metal oxide 210 can be anything between a partial monolayer and multiple monolayers. In some embodiments, pure metal regions 220 can be anything between a metal atom partially bonded to oxygen atoms in the matrix, with at least one bonding site to anchor a metal particle, and a metal atom cluster with one or more atoms incorporated in the mixed-metal oxide and one or more pure metal atoms available to alloy with a metallic catalysis particle (e.g., a platinum nano-particle).

FIG. 3A illustrates a cross-sectional view of one embodiment of a catalyst 300A in accordance with the principles of the present invention. The catalyst 300A comprises a catalyst support 310A, which is a porous structure, impregnated with catalytic carrier nano-particles 330A, such as the catalytic carrier nano-particles discussed above. As a porous structure, the catalyst support 310A comprises a plurality of pores 320. A close up view of the impregnated porous support 310A is illustrated in FIG. 3A, showing catalytic carrier nano-particles 330A fixed within the pores 320.

FIG. 3B illustrates a cross-sectional view of another embodiment of a catalyst 300B in accordance with the principles of the present invention. The catalyst 300B comprises a catalyst support 310B, which is a macro support, impregnated with catalytic carrier nano-particles 330B, such as the catalytic carrier nano-particles discussed above.

It is contemplated that a variety of different methods can be employed to impregnate the catalyst supports 310A, 310B with the catalyst carrier nano-particles 330A, 330B. In some embodiments, a liquid dispersion of the catalyst carrier nano-particles is applied to the catalyst support. The catalyst support is allowed to dry and/or a drying process is performed on the catalyst support. As the liquid in the dispersion evaporates, the catalyst carrier nano-particles settle onto the surface of the support and/or into the pores within the support. When the impregnated catalyst support dries, electrostatic interactions and other forces between the catalyst carrier nano-particles and the catalyst support effectuate some adhesion. Advantageously, such forces cause the nano-particles to stick onto the surfaces and pores of the support.

In some embodiments, a calcining step is performed to form oxide-oxide bonds between the nano-particles and the catalyst support, exposing them to heat, pressure, or a combination thereof. The calcining temperature is generally from 350 to 1000 degrees centigrade, and the pressure is on the order of ambient atmosphere to several atmospheres. For optimum oxide-oxide bonds, a portion of the catalytic carrier nano-particles is chosen to correspond to the material of which the catalyst support is comprised. By way of example, if a portion of the catalyst carrier nano-particles comprises alumina, then the catalyst support preferably comprises alumina, although dissimilar oxides are also contemplated. Due to the physical and chemical bond between the catalyst support and the nano-particles, islands of nano-particles that are bonded, fixed or otherwise pinned to the surfaces of the catalyst

support will not migrate and coalesce during catalytic conversion. The surface area for catalysis remains high, and therefore the catalytic activity remains high. In effect, operations such as fine chemical plants and oil refineries will not be required to stop operations and swap out ineffective catalyst supports with fresh catalyst supports with the same frequency as existing processes, thereby increasing throughput at the plants and refineries and reducing their overall cost of operation.

FIG. 4 illustrates one embodiment of a method of producing catalytic carrier nano-particles and a catalyst therefrom in accordance with the principles of the present invention.

At step 410, a plurality of carrier nano-particles is formed. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide, while the exterior surface comprises a pure metal. The carrier nano-particles can comprise any of the elements discussed above with respect to FIGS. 1-2.

At step 420, at least one catalytic particle is bonded to each of the carrier nano-particles, thereby forming catalytic carrier nano-particles. The catalytic particle is bonded to the pure metal at the exterior surface of the carrier particle. The catalytic particle can comprise any of the elements discussed above with respect to FIGS. 1-2.

At step 430, the plurality of carrier nano-particles are bonded to a support structure. One example of such a support structure is a porous ceramic material. However, it is contemplated that other types of support structures can be employed. The support structure can comprise any of the elements discussed above with respect to FIGS. 3A-B.

FIG. 5 illustrates one embodiment of a method of producing catalytic carrier nano-particles using a plasma gun in accordance with the principles of the present invention.

At step 510, a first material, a second material, and a third material are loaded into a plasma gun in a desired ratio. In some embodiments, the first material is an aluminum oxide material, the second material is a palladium material, and the third material is a platinum material. However, it is contemplated that other materials can be used in addition or as an alternative to these materials.

At step 520, the first material, the second material, and the third material are vaporized using the plasma gun, thereby forming a vapor cloud that comprises vaporized first material, vaporized second material, and vaporized third material.

At step 530, the vapor cloud is quenched, thereby condensing the vaporized material into carrier nano-particles with catalytic nano-material bonded to the carrier nano-particles. Each nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide. The exterior surface comprises a pure metal. The catalytic nano-material is bonded to the pure metal at the exterior surface.

It is contemplated that the present invention can employ a variety of different types of plasma gun systems to vaporize and quench the loaded material into nano-particles. In a preferred embodiment, the nano-particles are substantially uniform in size. The nano-particles can be formed by introducing micron-sized material into a plasma process, such as described and claimed in the co-owned and co-pending Application Serial No. 11/110,341, filed April 19, 2005, and titled "High Throughput Discovery of Materials Through Vapor Phase Synthesis," and the co-owned and co-pending Application Serial No. 12/151,935, filed May 8, 2008, and titled "Highly Turbulent Quench Chamber," both of which are hereby incorporated by reference as if set forth herein.

FIG. 7 illustrates one embodiment of a plasma-based particle production system 700 that can be used in accordance with the principles of the present invention. System 700 uses a plasma process and a highly turbulent quench chamber 745 to produce nano-particles. The system 700 comprises a precursor supply device 710 and a working gas supply device 720 fluidly coupled to a plasma production and reaction chamber 730. An energy delivery system 725 is also coupled with the plasma production and reactor chamber 730. The plasma production and reactor chamber 730 includes an injection port 740 that communicates fluidly with the constricting quench chamber 745. One or more ports 790 can also allow fluid communication between the quench chamber 745 and a controlled atmosphere system 770. The quench chamber 745 is also fluidly coupled to an outlet 765.

Generally, the chamber 730 operates as a reactor, producing an output comprising particles within a gas stream. Production includes the basic steps of combination, reaction, and conditioning as described later herein. The system combines precursor material supplied from the precursor supply device 710 and working gas supplied from the working gas supply device 720 within the energy delivery zone of the chamber 730.

In some embodiments, the precursor material comprises a powdered substance. In some embodiments, the precursor material is micron-sized. In some embodiments, the

precursor material comprises an average grain diameter of 500-600 nanometers. In some embodiments, the precursor material comprises an average grain diameter of one micrometer. In some embodiments, the precursor material comprises an average grain diameter greater than or equal to 5 microns.

The system energizes the working gas in the chamber 730 using energy from the energy supply system 725, thereby forming a plasma. The plasma is applied to the precursor material within the chamber 730 to form an energized, reactive mixture. This mixture comprises one or more materials in at least one of a plurality of phases, which may include vapor, gas, and plasma. The reactive mixture flows from the plasma production and reactor chamber 730 into the quench chamber 745 through an injection port 740.

The quench chamber 745 preferably comprises a substantially cylindrical surface 750, a frusto-conical surface 755, and an annular surface 760 connecting the injection port 740 with the cylindrical surface 750. The frusto-conical surface 760 narrows to meet the outlet 765. The plasma production and reactor chamber 730 includes an extended portion at the end of which the injection port 740 is disposed. This extended portion shortens the distance between the injection port 740 and the outlet 765, reducing the volume of region in which the reactive mixture and the conditioning fluid will mix, referred to as the quench region. In a preferred embodiment, the injection port 740 is arranged coaxially with the outlet 765. The center of the injection port is positioned a first distance d_1 from the outlet 765. The perimeter of the injection port is positioned a second distance d_2 from a portion of the frusto-conical surface 755. The injection port 740 and the frusto-conical surface 755 form the aforementioned quench region therebetween. The space between the perimeter of the injection port 740 and the frusto-conical surface 755 forms a gap therebetween that acts as a channel for supplying conditioning fluid into the quench region. The frusto-conical surface 755 acts as a funneling surface, channeling fluid through the gap and into the quench region.

While the reactive mixture flows into the quench chamber 745, the ports 790 supply conditioning fluid into the quench chamber 745. The conditioning fluid then moves along the frusto-conical surface 755, through the gap between the injection port 740 and the frusto-conical surface 755, and into the quench region. In some embodiments, the controlled atmosphere system 770 is configured to control the volume flow rate or mass flow rate of the conditioning fluid supplied to the quench region.

As the reactive mixture moves out of the injection port 740, it expands and mixes with the conditioning fluid. Preferably, the angle at which the conditioning fluid is supplied produces a high degree of turbulence and promotes mixing with the reactive mixture. This turbulence can depend on many parameters. In a preferred embodiment, one or more of these parameters is adjustable to control the level of turbulence. These factors include the flow rates of the conditioning fluid, the temperature of the frusto-conical surface 755, the angle of the frusto-conical surface 755 (which affects the angle at which the conditioning fluid is supplied into the quench region), and the size of the quench region. For example, the relative positioning of the frusto-conical surface 755 and the injection port 740 is adjustable, which can be used to adjust the volume of quench region. These adjustments can be made in a variety of different ways, using a variety of different mechanisms, including, but not limited to, automated means and manual means.

During a brief period immediately after entering the quench chamber 745, particle formation occurs. The degree to which the particles agglomerate depends on the rate of cooling. The cooling rate depends on the turbulence of the flow within the quench region. Preferably, the system is adjusted to form a highly turbulent flow, and to form very dispersed particles. For example, in preferred embodiments, the turbidity of the flow within the quench region is such that the flow has a Reynolds Number of at least 1000.

Still referring to FIG. 7, the structure of the quench chamber 745 is preferably formed of relatively thin walled components capable of dissipating substantial quantities of heat. For example, the thin-walled components can conduct heat from inside the chamber and radiate the heat to the ambient.

Substantial heat is emitted, mostly in the form of radiation, from the reactive mixture following its entry into the quench chamber 745. The quench chamber 745 is designed to dissipate this heat efficiently. The surfaces of the quench chamber 745 are preferably exposed to a cooling system (not shown). In a preferred embodiment, the cooling system is configured to control a temperature of the frusto-conical surface 755.

Following injection into the quench region, cooling, and particle formation, the mixture flows from the quench chamber 745 through the outlet port 765. Suction generated by a generator 795 moves the mixture and conditioning fluid from the quench region into the conduit 792. From the outlet port 765, the mixture flows along the conduit 792, toward the

suction generator 795. Preferably, the particles are removed from the mixture by a collection or sampling system (not shown) prior to encountering the suction generator 795.

Still referring to FIG. 7, the controlled atmosphere system 770 is fluidly coupled to a chamber 785, which is fluidly coupled to the quench region through port(s) 790, into which conditioning fluid is introduced from a reservoir of the controlled atmosphere system 770 through a conduit 780. As described above, the conditioning fluid preferably comprises argon. However, other inert, relatively heavy gases are equally preferred. Also, as discussed above, the preferable mechanism of providing the conditioning fluid into the quench chamber 745 is the formation of a pressure differential between the quench chamber 745 and the outlet 765. Such pressure differential will draw the conditioning fluid into the quench chamber 745 through the ports 790. Other methods of providing the conditioning fluid include, but are not limited to, forming positive pressure within the chamber 785.

The angle of the frusto-conical surface affects the angle at which the conditioning fluid is supplied into the quench region, which can affect the level of turbulence in the quench region. The conditioning fluid preferably flows into the quench region along a plurality of momentum vectors. The greater the degree of the angle between the momentum vectors, the higher the level of turbulence that will be produced. In a preferred embodiment, the high turbulent quench chamber comprises a frusto-conical surface that is configured to funnel at least two conditioning fluid momentum vectors into the quench region such that there is at least a 90 degree angle between the two momentum vectors. It is contemplated that other angle degree thresholds may be applied as well. For example, attention may also be paid to the angle formed between at least one of the conditioning fluid momentum vectors and the momentum vector of the reactive mixture. In one embodiment of a highly turbulent quench chamber, a reactive mixture inlet is configured to supply the reactive mixture into the quench region along a first momentum vector, the frusto-conical surface is configured to supply the conditioning fluid to the quench region along a second momentum vector, and the second momentum vector has an oblique angle greater than 20 degrees relative to the first momentum vector.

The size of the quench region also affects the level of turbulence in the quench region. The smaller the quench region, the higher the level of turbulence that will be produced. The

size of the quench region can be reduced by reducing the distance between the center of the injection port 740 and the outlet 765.

The high turbulence produced by the embodiments of the present invention decreases the period during which particles formed can agglomerate with one another, thereby producing particles of more uniform size, and in some instances, producing smaller-sized particles. Both of these features lead to particles with increased dispersibility and increased ratio of surface area to volume. While the plasma process described above is extremely advantageous in producing the nano-particles, it is contemplated that the nano-particles can be produced in other ways as well.

FIG. 6 illustrates another embodiment of a method of producing catalytic carrier nano-particles using a plasma gun in accordance with the principles of the present invention.

At step 610, a first material and a second material are loaded into a plasma gun in a desired ratio. In some embodiments, the first material is an aluminum oxide material and the second material is a palladium material. However, it is contemplated that other materials can be used in addition or as an alternative to these materials.

At step 620, the first material and the second material are vaporized using the plasma gun, thereby forming a vapor cloud that comprises vaporized first material and vaporized second material.

At step 630, the vapor cloud is quenched, thereby condensing the vaporized first material and vaporized second material into carrier nano-particles. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide. The exterior surface comprises a pure metal.

At step 640, the carrier nano-particles and a third material are loaded into a plasma gun in a desired ratio. In some embodiments, the third material is a platinum material. However, it is contemplated that other materials can be used in addition or as an alternative to a platinum material.

At step 650, the third material is vaporized using the plasma gun, thereby forming a vapor cloud that comprises vaporized third material.

At step 660, the carrier nano-particles are mixed with the vaporized third material in the vapor cloud.

At step 670, the vapor cloud is quenched, thereby condensing the vaporized third material into catalytic nano-particles and bonding them to the pure metal at the exterior surface of the carrier nano-particles.

As discussed above, it is contemplated that the present invention can employ a variety of different types of plasma gun systems to vaporize and quench the loaded material into nano-particles. The nano-particles can be formed by introducing micron-sized material into a plasma process, such as described and claimed in the co-owned and co-pending Application Serial No. 12/152,111, filed May 9, 2008, and titled "Powder Coating System and Method Including Gun Extension," which is hereby incorporated by reference as if set forth herein.

FIG. 8 illustrates another embodiment of a plasma based particle production system 800 in accordance with the principles of the present invention. System 800 comprises a plasma production chamber 820, a coating chamber 830, a cooling conduit 870, a sampling system 880, and a motive pump 890. The plasma production chamber 820 and coating chamber 830 are preferably enclosed in a motive gas supply chamber 810.

The motive gas supply chamber 810 is fluidly coupled to a motive gas supply system 815, which is configured to supply a motive gas. Preferably, the motive gas is an inert gas, such as argon. Furthermore, the chamber 810 preferably include airtight inlets and outlets configured to permit conduits to pass through to supply the plasma production chamber 820 and coating chamber 830. The motive gas supply chamber 810 can include airtight couplings that permit these conduits to pass through while preventing leakage either into or out of the chambers. Furthermore, output from the coating chamber 830 can flow into the cooling conduit 870, which can also passes through a wall of the motive gas supply chamber 810. Preferably, the cooling conduit 870 also exits the motive gas supply chamber through an airtight coupling.

The plasma production chamber 820 preferably includes several types of inputs for receiving power, gas, and target materials. Inputs are provided to the chamber 820 by a variety of supply systems through means described below. Functional aspects of these supply systems are also described below.

Various input conduits fluidly couple the plasma gas supply mixer 840 with a first gas supply vessel 842 and a second gas supply vessel 844. The plasma gas supply system 840 includes an outlet fluidly coupled with an inlet of the plasma production chamber 820. The

gas supply system 840 and the production chamber 820 are coupled by a conduit that passes through the motive gas supply chamber 810. Preferably, but not shown in the figure, an overall system controller provides control signals to the plasma gas supply system 840.

The power supply system 825 is also coupled to the plasma production chamber 820 through conduits that pass through the motive gas supply chamber 810. Preferably, but not shown in the figure, an overall system controller provides control signals to the power supply system 825.

The coater supply system 852 houses precursor material and includes an externally controllable delivery system fluidly coupled to a materials inlet of the plasma production chamber 820. The supply system 852 is coupled with the production chamber 820 by a conduit that passes through the motive gas supply chamber 810. Preferably, but not shown in the figure, an overall system controller provides control signals to the supply system 852.

Preferably, the conduits that run between the plasma gas supply system 840 and the production chamber 820, the coater supply system 852 and the plasma production chamber 820, and the power supply system 825 and the plasma production chamber 820, all enter the motive gas supply chamber 810 through air-tight seals. In some embodiments, the conduits of the various supply systems all have dedicated airtight entrances to the motive gas supply chamber 810. In some embodiments, a single airtight entrance accommodates multiple conduits.

The coating chamber 830 preferably includes inlets for motive gas, coater material-plasma mixture, and coatee material, as well as an outlet to provide an output to the cooling conduit 870. The inlets for motive gas couple the motive gas supply chamber 810 to the interior of the coating chamber 830. Preferably, these inlets are channels of adjustable size that directly coupled the two chambers, yet permit for controlled flow from the supply chamber 810 into the coating chamber 830.

Coatee material is stored within the coatee material supply system 854, which preferably sits outside the motive gas supply chamber 810. A conduit passes from the supply system 854 through the motive gas supply chamber 810 and also through a wall of the coating chamber 830. The coatee material supply system 854 includes a controllable delivery system that provides coatee material to the conduit. Preferably, the conduit enters both chambers though airtight seals. The conduit terminates within the coating chamber at a selected

location. Preferably, the location of the terminus is selected based on parameters of operation. Also preferably, but not shown, an overall system controller is configured to supply control signals to the supply system 854.

The cooling conduit 870 connects the coating chamber 830 with the sampling system 880. The conduit 870 exits the motive gas supply chamber 810 through airtight means. The cooling conduit 870 includes a first section 872, a second section 874, third section 876, and fourth section 878. The sections are joined by gas input couplings that contain gas input features. The first section 872 is joined to the second section 874 by the gas input coupling 862. In turn, the second section 874 is joined to the third section 876 by the gas input coupling 864. The gas input coupling 866 joins the third section 876 to the fourth section 878.

In the illustrated embodiment, no gas is shown being supplied to the gas input couplings 862, 864, 866 for input into the cooling conduit 870. However, gas can be supplied through one or more of the gas input couplings.

The sampling system 880 is fluidly coupled between the cooling conduit 870 and the motive pump 890. The sampling system 880 is configured to receive output from the cooling conduit 870, sample material having appropriate characteristics from the output, and permit remains of the output to flow to the motive pump 890, which is fluidly coupled through a conduit to the sampling system 880.

During operation, the supply systems 840, 852, and 825 provide plasma gas, coater material, and power, respectively, to the plasma production chamber 820. Power from the delivery system 825 is used to energize gas from the supply system 840 to produce a plasma within the production chamber 820. The coater material supply system 852 provides coater material in metered amounts into the plasma production chamber 820, exposing the coater material to plasma formed therein.

The overall control system (not shown) sends signals to the plasma gas supply system 840, coater material supply system 852, and power supply system 825 to set operational parameters. The plasma gas supply system 840 determines the ratio of mixing for the first and second gasses to produce plasma gas, as well as the rate at which the plasma gas feeds into the plasma production chamber 820. In a preferred embodiment, the first gas is hydrogen and the second gas is an inert gas, such as argon. The coater material supply system 852 determines the rate at which the coater material is supplied into the plasma production

chamber 820. The power supply system 825 determines the voltage and amperage at which power is supplied to the plasma production chamber. In combination, these parameters determine the characteristics of the plasma produced within the plasma production chamber 820, as well as the characteristics of the plasma-coatee material mixture also produced within the chamber 820. Furthermore, although the coater supply system is described as providing only a single coater material into the plasma production chamber 820 at a single location, in some embodiments of the present invention, the coater supply system 852 supplies a plurality of materials into the plasma production chamber 820 at one or more locations.

The motive gas supply chamber 810 receives motive gas, typically an inert gas such as argon, from a preferably dedicated motive gas supply system 815. The motive gas supply chamber 810 provides an airtight enclosure around the plasma production chamber 820 and coating chamber 830. The motive gas supply system 815 preferably maintains a pressure within the motive gas supply chamber 810 that slightly exceeds the ambient pressure of the environment in which the system 800 is housed, regardless of any variation in suction force generated by the motive pump 890.

The coating chamber 830 receives a coater material and plasma mixture from the plasma production chamber 820. The coating chamber 830 also receives motive gas through input features. Preferably, these input features provide for an adjustable flow rate of motive gas into the coating chamber 830. Motive gas flow is preferably motivated by the motive pump 890 by pulling a negative pressure on the conduit 870, motivating mass flow through the outlet of the coating chamber 870. However, the flow rate of the motive gas into the coating chamber 830 is preferably controlled by the overall control system.

Furthermore, the coatee material supply system 854 provides a metered stream of coater material through the coatee material conduit to the conduit's terminus location within the coating chamber 830. The rate at which coatee material is provided into the chamber 830 preferably is determined by the overall control system. Furthermore, although the terminus of the coatee material supply conduit is shown to deposit material only at one location within the coating chamber 830, in some embodiments of the present invention, the terminus deposits coatee material at a plurality of locations within the conduit (e.g., in an annular configuration surrounding output of the plasma production chamber 820).

Preferably, coating chamber 830 is shaped and the operational parameters of the apparatus 800 are controlled so that the coater material and plasma mixture enters the coating chamber and reaches a maximum enthalpy shortly thereafter. Most preferably, this maximum of enthalpy occurs within a defined region of the coating chamber 830, with the average enthalpy of the mixture falling as it moves away from that region with its minimum (within the coating chamber 830) coming at the outlet to the cooling conduit 870. For example, as illustrated in FIG. 8, the maximum enthalpy occurs in region 1. In these embodiments, the location of the terminus for delivery of the coatee material is chosen to be outside the maximum region and between the maximum region and the outlet.

Furthermore, in certain embodiments of the present invention, the coating chamber 830 is shaped and the operational parameters of the apparatus 800 are chosen so that the coater material begins to condense from the coater material and gas mixture within a defined region of the coating chamber 830. For example, in FIG. 8, the region 2 is such an initial condensation region. Typically, the initial condensation region lies between the maximum enthalpy region and the outlet of the coating chamber 830. In some embodiments, the coatee material is delivered into the initial condensation region. Preferably in these embodiments, sufficient enthalpy remains in the condensing mixture to vaporize the coatee material. Thus, as the coater material is condensing to form particles, the coatee material is vaporized mixed with the condensing particles. As the coater material gas particle mixture mixes with the vaporized coatee material and moves towards the outlet, the coater condenses on the coatee particles, forming a plurality of coated particles.

The cooling conduit 870 receives the coated particle and gas mixture from the coating chamber 830. Preferably, the mixture is pulled into the cooling conduit by the motive pump 890. However, in some embodiments, a motive pump or other system within the sampling system 880 provides some motive force to pull the mixture. Of course, to some extent, pressure provided by the plasma production chamber 820 and the motive gas supply 815 motivate the movement of the mixture into the cooling conduit 870.

In some embodiments, the cooling conduit 870 is equipped with an active cooling system. In some embodiments, a gas is supplied into the gas input couplings 862, 864, or 866. In some of these embodiments, the gas is a cooling and entraining gas. In some of these

embodiments, the gas is a passivating gas configured to reduce the reactivity of the condensed particles within the mixture.

As mentioned above, the sampling system 880 preferably permits the motive pump 890 to provide a motive force therethrough. However, in some embodiments the sampling system 880 provides additional motive force. In some embodiments the sampling system 880 supplants the motive force provided by the motive pump 890 and provides a substitute motive force to the cooling conduit 870.

It is contemplated that the sampling system 880 can be configured in a variety of ways. In one embodiment, the sampling system 880 comprises a sampling structure, at least one filled aperture formed in the sampling structure, and at least one unfilled aperture formed in the sampling structure. Each filled aperture is configured to collect particles from the mixture stream, such as by using a filter. The sampling structure is configured to be adjusted between a pass-through configuration and a collection configuration. The pass-through configuration comprises an unfilled aperture being fluidly aligned with a conduit, such as conduit 870, thereby allowing the unfilled aperture to receive the mixture stream from the conduit and the mixture stream to flow through the sampling structure without substantially altering the particle content of the mixture stream. The collection configuration comprises a filled aperture being fluidly aligned with the conduit, thereby allowing the filled aperture to receive the mixture stream and collect particles while the mixture stream is being flown through the filled aperture.

It is contemplated that the sampling structure can be adjusted between the pass-through configuration and the collection configuration in a variety of ways. In one embodiment, the sampling structure is a disk-shaped structure including an annular array of apertures, wherein the annular array comprises a plurality of the filled apertures and a plurality of the unfilled apertures. The sampling structure is rotatably mounted to a base, wherein rotational movement of the sampling structure results in the adjustment of the sampling structure between the pass-through configuration and the collection configuration. In another embodiment, the sampling structure is a rectangular-shaped structure including a linear array of apertures, wherein the linear array comprises a plurality of the filled apertures and a plurality of the unfilled apertures. The sampling structure is slideably mounted to a base,

wherein sliding of the sampling structure results in the adjustment of the sampling structure between the pass-through configuration and the collection configuration.

It is contemplated that the particles of the present invention can be formed and bonded using wet chemistry methods in addition or as an alternative to the use of a plasma-based system. Such methods include, but are not limited to, the methods described in the following references, which are hereby incorporated by reference as if set forth herein: Adschiri et al., "Hydrothermal synthesis of metal oxide nanoparticles at supercritical conditions" (*Journal of Nanoparticle Research* 3: 227-235, 2001); Kratohvil et al., "Preparation and properties of coated, uniform, inorganic colloidal particles: I. Aluminum (hydrous) oxide on hematile, chromia, and titania" (*Advanced Ceramics Materials* 2:4, 1987); Matijevic, "Monodispersed Metal (Hydrous) Oxides - A Fascinating Field of Colloid Science" *(Acc. Chem. Res.* 14, 22-29, 1981); Lakshmi et al., "Sol-Gel Template Synthesis of Semiconductor Oxide Micro- and Nanostructures" (*Chem. Mater.* 9, 2544-2550, 1997); Carnes et al., "Synthesis, Characterization, and Adsorption Studies of Nanocrystalline Aluminum Oxide and a Biometallic Nanocrystalline Aluminum Oxide/Magnesium Oxide" (*Chem. Mater.* 14, 2922-2929, 2002); Zhu et al., "γ-Alumina Nanofibers Prepared from Aluminum Hydrate with Poly(ethylene oxide) Surfactant" (*Chem. Mater.* 14, 2086-2093, 2002); Alexander et al., "PROCESS FOR MODIFYING THE PROPERTIES OF A SILICA SOL AND PRODUCT THEREOF" (U.S. Patent No. 2,892,797, Issued June 30, 1959); Alexander et al., "AQUASOLS OF POSITIVELY-CHARGED COATED SILICA PARTICLES AND THEIR PRODUCTION" (U.S. Patent No. 3,007,878, Issued Nov. 7, 1961); Moore, Jr., "STABLE POSITIVELY CHARGED ALUMINA COATED SILICA SOLS AND THEIR PREPARATION BY POST-NEUTRALIZATION" (U.S. Patent No. 3,719,607, Issued March 6, 1973); Moore, Jr., "STABLE POSITIVELY CHARGED ALUMINA COATED SILICA SOLS" (U.S. Patent No. 3,745,126, Issued July 10, 1973); and Kovarik, "METAL OXIDE SOLS" (U.S. Patent No. 3,864,142, Issued Feb. 4, 1975).

FIG. 9 illustrates one embodiment of a method 900 of catalytic carrier nano-particles using a wet chemistry process in accordance with the principles of the present invention.

At step 910, a first material and a second material are provided in a desired ratio. In some embodiments, the first material is an aluminum material and the second material is a palladium material. In some embodiments, the aluminum material is a salt or an organic

compound, and the palladium material is a salt. However, it is contemplated that other materials can be used in addition or as an alternative to these materials.

At step 920, a plurality of carrier nano-particles are formed from the first material and the second material via a chemical reaction using a wet chemistry process. Each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprises a mixed-metal oxide. The exterior surface comprises a pure metal. In some embodiments, the center core of each carrier nano-particle comprises silica.

At step 930, a suspension of the carrier nano-particles is formed and mixed with a catalytic solution (a solution comprising particles having catalytic properties). In some embodiments, the catalytic solution is a platinum solution. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution. However, it is contemplated that other types of catalytic solutions can be used.

At step 940, the catalytic solution reacts with the carrier nano-particles, thereby pinning catalytic nano-particles to the pure metal at the exterior surface of the carrier nano-particles. For example, in one embodiment, platinum ions from the catalytic solution react with the oxide surface of the carrier nano-particles, resulting in discreet platinum atoms ending up on the surface of the carrier nano-particles. These atoms can diffuse over the oxide surface and agglomerate with other platinum atoms to form platinum nano-particles. When a platinum nano-particle reaches a palladium atom cluster bonded in the oxide surface, the platinum cluster alloys with the palladium cluster, thereby pinning the platinum cluster. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less. In some embodiments, the catalytic nano-particles have an average grain size of approximately 1 nanometer or less. In some embodiments, the catalytic nano-particles have an average grain size of approximately ½ nanometer. However, it is contemplated that other sizes are within the scope of the present invention.

FIG. 10 illustrates one embodiment of a method 1000 of producing catalytic carrier nano-particles using both a plasma gun and a wet chemistry process in accordance with the principles of the present invention.

At step 1010, a first material and a second material are loaded into a plasma gun in a desired ratio. In some embodiments, the first material is an aluminum oxide material and the

second material is a palladium material. However, it is contemplated that other materials can be used in addition or as an alternative to these materials.

At step 1020, the first material and the second material are vaporized using the plasma gun, thereby forming a vapor cloud that comprises vaporized first material and vaporized second material.

At step 1030, the vapor cloud is quenched, thereby condensing the vaporized first material and vaporized second material into carrier nano-particles. Each carrier nano-particle comprises an interior region and an exterior surface. The interior region comprises a mixed-metal oxide. The exterior surface comprises a pure metal. One objective of the present invention is to mix an easily oxidizable and reducible metal into a carrier oxide (e.g., aluminum oxide, ceria, zirconia, silica, etc.). The idea is that the metal (e.g., Pd, Cu, Zn, Cr, Ni, etc.) gets incorporated into the oxide matrix. However, by making this an oxygen deficient environment, such as by adding H_2 into the carrier gas, some of the metal is in its non-oxidized state on the surface of the particle. One method that can be used is to form mixed-metal oxide nano-particles, and then expose them to a reducing environment, such as H_2 at temperature or a H_2 plasma, and partially reduce the metal-oxide on the surface. The carrier-oxide would not easily reduce in this environment. However, the oxidation state of the metal buried below the surface of the mixed metal oxide is not crucial. The result is a mixed-metal oxide particle with metal atoms on the surface that are not fully oxidized. These metal atoms form the anchor for the metallic catalysis particles added later, such as described in steps 930 and 940 of FIG. 9 and steps 1040 and 1050 of FIG. 10.

At step 1040, the carrier nano-particles are mixed with a catalytic solution. In some embodiments, the catalytic solution is a platinum solution. In some embodiments, the platinum solution is a platinum nitrate solution or a platinum chloride solution. However, it is contemplated that other types of catalytic solutions can be used.

At step 1050, the catalytic solution reacts with the carrier nano-particles, thereby pinning catalytic nano-particles to the pure metal at the exterior surface of the carrier nano-particles. For example, in one embodiment, platinum ions from the catalytic solution react with the oxide surface of the carrier nano-particles, resulting in discreet platinum atoms ending up on the surface of the carrier nano-particles. These atoms can diffuse over the oxide surface and agglomerate with other platinum atoms to form platinum nano-particles. When a

platinum nano-particle reaches a palladium atom cluster bonded in the oxide surface, the platinum cluster alloys with the palladium cluster, thereby pinning the platinum cluster. In some embodiments, the carrier nano-particles have an average grain size of approximately 10 nanometers or less. In some embodiments, the catalytic nano-particles have an average grain size of approximately 1 nanometer or less. In some embodiments, the catalytic nano-particles have an average grain size of approximately ½ nanometer. However, it is contemplated that other sizes are within the scope of the present invention.

The present invention has been described in terms of specific embodiments incorporating details to facilitate the understanding of principles of construction and operation of the invention. Such reference herein to specific embodiments and details thereof is not intended to limit the scope of the claims appended hereto. It will be readily apparent to one skilled in the art that other various modifications may be made in the embodiment chosen for illustration without departing from the spirit and scope of the invention as defined by the claims.

CLAIMS

What is claimed is:

1. A nano-particle comprising:

an interior region comprising a mixed-metal oxide, wherein the mixed-metal oxide comprises aluminum oxide and palladium; and

an exterior surface comprising pure metallic palladium.

2. The nano-particle of Claim 1, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

3. The nano-particle of Claim 1, wherein the nano-particle comprises a diameter of approximately 10 nanometers or less.

4. The nano-particle of Claim 1, wherein the mixed-metal oxide forms a center core of the nano-particle.

5. The nano-particle of Claim 4, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

6. The nano-particle of Claim 1, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle.

7. The nano-particle of Claim 6, wherein the center core comprises silica.

8. A method of forming nano-particles, the method comprising:

loading a quantity of aluminum oxide material and a quantity of palladium material into a plasma gun in a desired ratio;

vaporizing the quantity of aluminum oxide material and the quantity of palladium material using the plasma gun, thereby forming a vapor cloud comprising vaporized aluminum oxide and vaporized palladium; and

quenching the vapor cloud, thereby condensing the vaporized aluminum oxide and the vaporized palladium into nano-particles,

wherein each nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium.

9. The method of Claim 8, wherein the step of vaporizing the quantity of aluminum oxide material and the quantity of palladium material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of aluminum oxide material and the quantity of palladium material into the plasma stream.

10. The method of Claim 8, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

11. The method of Claim 8, wherein the nano-particles have an average grain size of approximately 10 nanometers or less.

12. The method of Claim 8, wherein the mixed-metal oxide forms a center core of the nano-particle.

13. The method of Claim 12, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

14. A method of forming nano-particles, the method comprising:

providing a quantity of aluminum material and a quantity of palladium material in a desired ratio; and

forming a plurality of nano-particles from the quantity of aluminum material and the quantity of palladium material using a wet chemistry process,

wherein each formed nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium.

15. The method of Claim 14, wherein:

the aluminum material is a salt or an organic compound;

the palladium material is a salt; and

the wet chemistry process comprises forming the mixed-metal oxide from the aluminum material and the palladium material via a chemical reaction.

16. The method of Claim 15, wherein:

the mixed-metal oxide forms a center core of each nano-particle; and

the mixed-metal oxide consists only of aluminum oxide and palladium.

17. The method of Claim 14, wherein the step of forming the plurality of nano-particles using a wet chemistry process comprises forming a monolayer around a center core of each nano-particle, the monolayer comprising the mixed-metal oxide.

18. The method of Claim 17, wherein the center core of each nano-particle comprises silica.

19. The method of Claim 14, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

20. The method of Claim 14, wherein the nano-particles have an average grain size of approximately 10 nanometers or less.

21. A catalytic carrier nano-particle comprising:

a carrier nano-particle comprising an interior region and an exterior surface, wherein the interior region comprises a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprises pure metallic palladium; and

a catalytic nano-particle bonded to the pure metallic palladium at the exterior surface, wherein the catalytic nano-particle is platinum.

22. The catalytic carrier nano-particle of Claim 21, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

23. The catalytic carrier nano-particle of Claim 21, wherein the carrier nano-particle comprises a diameter of approximately 10 nanometers or less.

24. The catalytic carrier nano-particle of Claim 21, wherein the mixed-metal oxide forms a center core of the carrier nano-particle.

25. The catalytic carrier nano-particle of Claim 24, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

26. The catalytic carrier nano-particle of Claim 21, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of the carrier nano-particle.

27. The catalytic carrier nano-particle of Claim 26, wherein the center core comprises silica.

28. A method of forming nano-particles, the method comprising:

loading a quantity of aluminum oxide material, a quantity of palladium material, and a quantity of platinum material into a plasma gun in a desired ratio;

vaporizing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material using the plasma gun, thereby forming

a vapor cloud comprising vaporized aluminum oxide, vaporized palladium, and vaporized platinum; and

quenching the vapor cloud, thereby condensing the vaporized aluminum oxide, the vaporized palladium, and the vaporized platinum into carrier nano-particles,

wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, the exterior surface comprising a pure metallic palladium, and a platinum nano-particle is bonded to the pure metallic palladium at the exterior surface.

29. The method of Claim 28, wherein the step of vaporizing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of aluminum oxide material, the quantity of palladium material, and the quantity of platinum material into the plasma stream.

30. The method of Claim 28, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

31. The method of Claim 28, wherein the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

32. The method of Claim 28, wherein the mixed-metal oxide forms a center core of the carrier nano-particle.

33. The method of Claim 32, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

34. A method of forming nano-particles, the method comprising:

providing a quantity of aluminum material, a quantity of palladium material, and a quantity of platinum material in a desired ratio; and

forming a plurality of carrier nano-particles from the quantity of aluminum material, the quantity of palladium material, and the quantity of platinum material using a wet chemistry process,

wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, the exterior surface comprising a pure metallic palladium, and a platinum nano-particle is bonded to the pure metallic palladium at the exterior surface.

35. The method of Claim 34, wherein:

the aluminum material is a salt or an organic compound;

the palladium material is a salt; and

the wet chemistry process comprises forming the mixed-metal oxide from the aluminum material and the palladium material via a chemical reaction, and forming a carrier particle suspension comprising a plurality of aluminum-palladium carrier particles, wherein each aluminum-palladium carrier particle comprises an interior region and an exterior surface, the interior region comprising the mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium.

36. The method of Claim 35, wherein the wet chemistry process further comprises:

mixing a platinum solution with the carrier particle suspension; and

bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the aluminum-palladium carrier particles.

37. The method of Claim 36, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

38. The method of Claim 34, wherein:

the aluminum material is a salt or an organic compound;

the palladium material is a salt; and

the wet chemistry process comprises forming the mixed-metal oxide from the aluminum material and the palladium material via a chemical reaction, forming a monolayer of the mixed-metal oxide around a center core of each carrier nano-particle, and forming a suspension of the carrier nano-particles, wherein the center core comprises a different chemical composition from the monolayer, and each monolayer comprises an interior region and an exterior surface, the interior region comprising the mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium.

39. The method of Claim 38, wherein the wet chemistry process further comprises:

mixing a platinum solution with the suspension of nano-particles; and

bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the carrier nano-particles.

40. The method of Claim 39, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

41. The method of Claim 38, wherein the center core of each carrier nano-particle comprises silica.

42. The method of Claim 34, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium-platinum alloy at the exterior surface.

43. The method of Claim 34, wherein the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

44. A method of forming nano-particles, the method comprising:

forming a plurality of aluminum-palladium carrier particles using a plasma gun, wherein each aluminum-palladium carrier particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium; and

affixing a platinum nano-particle to the pure metallic palladium on each of the aluminum-palladium carrier particles using a wet chemistry process.

45. The method of Claim 44, wherein forming the plurality of aluminum-palladium carrier particles using the plasma gun comprises:

loading a quantity of aluminum oxide material and a quantity of palladium material into the plasma gun in a desired ratio;

vaporizing the quantity of aluminum oxide material and the quantity of palladium material using the plasma gun, thereby forming a vapor cloud comprising vaporized aluminum oxide and vaporized palladium; and

quenching the vapor cloud, thereby condensing the vaporized aluminum oxide and the vaporized palladium into carrier nano-particles,

wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide that includes aluminum oxide and palladium, and the exterior surface comprising pure metallic palladium.

46. The method of Claim 45, wherein the step of vaporizing the quantity of aluminum oxide material and the quantity of palladium material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of aluminum oxide material and the quantity of palladium material into the plasma stream.

47. The method of Claim 45, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

48. The method of Claim 44, wherein the aluminum-palladium carrier particles have an average grain size of approximately 10 nanometers or less.

49. The method of Claim 44, wherein the mixed-metal oxide forms a center core of each aluminum-palladium carrier particle.

50. The method of Claim 49, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

51. The method of Claim 44, wherein the wet chemistry process comprises:

mixing a platinum solution with the aluminum-palladium carrier particles; and

bonding platinum nano-particles formed from the platinum solution to the pure metallic palladium on the aluminum-palladium carrier particles.

52. The method of Claim 51, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

53. The method of Claim 44, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

54. The method of Claim 44, wherein the aluminum-palladium carrier particles have an average grain size of approximately 10 nanometers or less.

55. A catalyst comprising:

a support structure; and

a plurality of carrier nano-particles bonded to the support structure, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal, wherein a catalytic particle is bonded with the pure metal at the exterior surface.

56. The catalyst of Claim 55, wherein:

the support structure is a porous ceramic material; and

the plurality of carrier nano-particles are disposed within the pores of the ceramic material.

57. The catalyst of Claim 55, wherein the mixed-metal oxide comprises aluminum oxide and palladium.

58. The catalyst of Claim 55, wherein the pure metal is palladium.

59. The catalyst of Claim 55, wherein the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface.

60. The catalyst of Claim 55, wherein the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

61. The catalyst of Claim 55, wherein the mixed-metal oxide forms a center core of each carrier nano-particle.

62. The catalyst of Claim 61, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

63. The catalyst of Claim 55, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of each carrier nano-particle.

64. The catalyst of Claim 63, wherein the center core comprises silica.

65. The catalyst of Claim 55, wherein the exterior surface comprises a pure metal alloy.

66. The catalyst of Claim 55, wherein the catalytic particle is platinum.

67. The catalyst of Claim 55, wherein the catalyst is a non-aging catalyst.

68. A method of forming a catalyst, the method comprising:

providing a support structure;

providing a plurality of carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal, wherein a catalytic particle is bonded with the pure metal at the exterior surface; and

bonding the plurality of carrier nano-particles to the support structure.

69. The method of Claim 68, wherein:

the support structure is a porous ceramic material; and

the plurality of carrier nano-particles are disposed within the pores of the ceramic material.

70. The method of Claim 68, wherein the mixed-metal oxide includes aluminum oxide and palladium, and the pure metal comprises pure metallic palladium.

71. The method of Claim 68, wherein the catalytic particle is platinum.

72. The method of Claim 68, wherein the palladium in the interior region extends to and is bonded with the pure metallic palladium at the exterior surface.

73. The method of Claim 68, wherein the carrier nano-particles have an average grain size of approximately 10 nanometers or less.

74. The method of Claim 68, wherein the mixed-metal oxide forms a center core of the carrier nano-particle.

75. The method of Claim 74, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

76. The method of Claim 68, wherein the step of providing a plurality of carrier nano-particles comprises:

loading a quantity of a first material, a quantity of a second material, and a quantity of a third material into a plasma gun in a desired ratio;

vaporizing the quantity of the first material, the quantity of the second material, and the quantity of the third material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material, vaporized second material, and vaporized third material; and

quenching the vapor cloud, thereby condensing the vaporized first material, the vaporized second material, and the vaporized third material into carrier nano-particles, wherein each carrier nano-particle comprises the interior region and the exterior surface.

77. The method of Claim 76, wherein the step of vaporizing the quantity of the first material material, the quantity of the second material, and the quantity of the third material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of the first material, the quantity of the second material, and the quantity of the third material into the plasma stream.

78. The method of Claim 68, wherein the step of providing a plurality of carrier nano-particles comprises:

providing a quantity of a first material, a quantity of a second material, and a quantity of third material in a desired ratio; and

forming the plurality of carrier nano-particles from the quantity of the first material, the quantity of the second material, and the quantity of the third material using a wet chemistry process.

79. The method of Claim 78, wherein the mixed-metal oxide comprises the first material and the second material, the exterior surface comprises the second material, and the catalytic particle comprises the third material.

80. The method of Claim 78, wherein the wet chemistry process comprises:

forming the mixed-metal oxide from the first material and the second material via a chemical reaction; and

forming a carrier particle suspension comprising a plurality of carrier particles, wherein each carrier particle comprises the interior region and the exterior surface, the mixed-metal oxide of the interior region comprising the first material and the second material, and the pure metal at the exterior surface comprising the second material.

81. The method of Claim 80, wherein:

the first material is an aluminum material; and

the second material is a palladium material.

82. The method of Claim 81, wherein:

the aluminum material is a salt or an organic compound; and

the palladium material is a salt.

83. The method of Claim 80, wherein the wet chemistry process further comprises:

mixing a catalytic solution with the carrier particle suspension; and

bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the carrier particles.

84. The method of Claim 83, wherein the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles.

85. The method of Claim 84, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

86. The method of Claim 80, wherein forming the mixed-metal oxide comprises forming a monolayer around a center core of each carrier particle, wherein the center core comprises a different chemical composition from the monolayer, and each monolayer comprises the interior region and the exterior surface.

87. The method of Claim 86, wherein the mixed-metal oxide includes aluminum oxide and palladium, and the pure metal at the exterior surface comprises palladium.

88. The method of Claim 86, wherein the wet chemistry process further comprises:

mixing a catalytic solution with the carrier particle suspension; and

bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the carrier particles.

89. The method of Claim 88, wherein the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles.

90. The method of Claim 89, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

91. The method of Claim 86, wherein the center core of each carrier nano-particle comprises silica.

92. The method of Claim 68, wherein the step of providing the plurality of carrier nano-particles comprises:

forming a plurality of carrier particles using a plasma gun, wherein each carrier particle comprises the interior region and the exterior surface; and

affixing a catalytic nano-particle to the pure metal on each of the carrier particles using a wet chemistry process.

93. The method of Claim 92, wherein the step of forming the plurality of carrier particles using the plasma gun comprises:

loading a quantity of a first material and a quantity of a second material into the plasma gun in a desired ratio;

vaporizing the quantity of the first material and the quantity of the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material; and

quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into the carrier particles.

94. The method of Claim 93, wherein the step of vaporizing the quantity of first material and the quantity of second material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of first material and the quantity of second material into the plasma stream.

95. The method of Claim 92, wherein the step of affixing a catalytic nano-particle to the pure metal using a wet chemistry process comprises:

mixing a catalytic solution with the core particles; and

bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the carrier particles.

96. The method of Claim 95, wherein the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles.

97. The method of Claim 96, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

98. The method of Claim 68, wherein bonding the plurality of carrier nano-particles to the support structure comprises performing a calcining process.

99. A nano-particle comprising:

an interior region comprising a mixed-metal oxide; and

an exterior surface comprising a pure metal.

100. The nano-particle of Claim 99, wherein the mixed-metal oxide comprises aluminum oxide and palladium.

101. The nano-particle of Claim 99, wherein the pure metal is palladium.

102. The nano-particle of Claim 99, wherein the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface.

103. The nano-particle of Claim 99, wherein the nano-particle comprises a diameter of approximately 10 nanometers or less.

104. The nano-particle of Claim 99, wherein the mixed-metal oxide forms a center core of the nano-particle.

105. The nano-particle of Claim 104, wherein the mixed-metal oxide consists only of aluminum oxide and palladium.

106. The nano-particle of Claim 99, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle.

107. The nano-particle of Claim 106, wherein the center core comprises silica.

108. The nano-particle of Claim 106, wherein the exterior surface comprises a pure metal alloy.

109. The nano-particle of Claim 108, wherein the pure metal alloy is a palladium-platinum alloy.

110. A method of forming nano-particles, the method comprising:

providing a quantity of a first material and a quantity of a second material; and

forming a plurality of nano-particles from the quantity of the first material and the quantity of the second material, wherein each nano-particle comprises an interior region and an exterior surface, and wherein the interior region comprises a mixed-metal oxide, and the exterior surface comprises a pure metal.

111. The method of Claim 110, wherein the mixed-metal oxide comprises the first material and the second material, and the pure metal comprises the second material.

112. The method of Claim 110, wherein the mixed-metal oxide comprises aluminum oxide and palladium, and the pure metal comprises palladium.

113. The method of Claim 110, wherein the mixed-metal oxide comprises aluminum oxide and palladium, and the pure metal comprises a palladium-platinum alloy.

114. The method of Claim 110, wherein the step of forming the plurality of nano-particles comprises:

loading the quantity of the first material and the quantity of the second material into a plasma gun in a desired ratio;

vaporizing the quantity of the first material and the quantity of the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material; and

quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into nano-particles, wherein each nano-particle comprises the interior region and the exterior surface.

115. The method of Claim 114, wherein the step of vaporizing the quantity of the first material material and the quantity of the second material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and

flowing the quantity of the first material and the quantity of the second material into the plasma stream.

116. The method of Claim 114, wherein the mixed-metal oxide comprises the first material and the second material, and the pure metal comprises the second material.

117. The method of Claim 116, wherein the first material is an aluminum material and the second material is a palladium material.

118. The method of Claim 110, wherein the step of forming the plurality of nano-particles comprises performing a wet chemistry process with the quantity of first material and the quantity of second material.

119. The method of Claim 118, wherein the wet chemistry process comprises forming the mixed-metal oxide from the first material and the second material via a chemical reaction.

120. The method of Claim 119, wherein:

the first material comprises an aluminum material; and

the second material comprises a palladium material.

121. The method of Claim 120, wherein:

the aluminum material is a salt or an organic compound; and

the palladium material is a salt.

122. The method of Claim 119, wherein:

the mixed-metal oxide forms a center core of each nano-particle; and

the mixed-metal oxide consists only of aluminum oxide and palladium.

123. The method of Claim 119, wherein the wet chemistry process comprises forming a monolayer around a center core of each nano-particle, the monolayer comprising the mixed-metal oxide.

124. The method of Claim 123, wherein the center core of each nano-particle comprises silica.

125. The method of Claim 110, wherein the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface.

126. The method of Claim 110, wherein the nano-particles have an average grain size of approximately 10 nanometers or less.

127. A nano-particle comprising:

an interior region comprising a mixed-metal oxide;

an exterior surface comprising a pure metal; and

a catalytic particle affixed to the pure metal, wherein the catalytic particle has a chemical composition different from the pure metal and the mixed-metal oxide.

128. The nano-particle of Claim 127, wherein:

the mixed-metal oxide comprises a first material and a second material;

the pure metal comprises the second material; and

the catalytic particle comprises a third material.

129. The nano-particle of Claim 128, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of the nano-particle.

130. The nano-particle of Claim 129, wherein the center core comprises silica.

131. The nano-particle of Claim 127, wherein:

the mixed-metal oxide comprises aluminum-oxide and a metallic pinning agent;

the pure metal comprises the metallic pinning agent, wherein the metallic pinning agent does not comprise platinum; and

the catalytic particle comprises platinum.

132. The nano-particle of Claim 131, wherein the metallic pinning agent comprises copper, molybdenum, or cobalt.

133. The nano-particle of Claim 131, wherein the catalytic particle comprises a metal alloy.

134. The nano-particle of Claim 133, wherein the metal alloy is a palladium-platinum alloy.

135. The nano-particle of Claim 131, wherein:

the metallic pinning agent comprises copper, molybdenum, or cobalt; and

the catalytic particle comprises a palladium-platinum alloy.

136. The nano-particle of Claim 127, wherein the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface.

137. The nano-particle of Claim 127, wherein the nano-particle comprises a diameter of approximately 10 nanometers or less.

138. The nano-particle of Claim 127, wherein the catalytic particle is a nano-particle.

139. The nano-particle of Claim 138, wherein the catalytic particle comprises a diameter of approximately 1 nanometer or less.

140. The nano-particle of Claim 139, wherein the catalytic particle comprises a diameter of approximately ½ nanometer.

141. A method of forming a nano-particle, the method comprising:

providing a carrier nano-particle, wherein the carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal; and

affixing a catalytic particle to the pure metal, wherein the catalytic particle has a chemical composition different from the pure metal and the mixed-metal oxide.

142. The method of Claim 141, wherein:

the mixed-metal oxide is formed from a first material and a second material;

the pure metal is formed from the second material; and

the catalytic particle is formed from a third material.

143. The method of Claim 142, wherein the mixed-metal oxide forms a monolayer that surrounds a center core of the carrier nano-particle.

144. The method of Claim 143, wherein the center core comprises silica.

145. The method of Claim 141, wherein:

the mixed-metal oxide comprises aluminum-oxide and a metallic pinning agent;

the pure metal comprises the metallic pinning agent, wherein the metallic pinning agent does not comprise platinum; and

the catalytic particle comprises platinum.

146. The method of Claim 145, wherein the metallic pinning agent comprises copper, molybdenum, or cobalt.

147. The method of Claim 145, wherein the catalytic particle comprises a metal alloy.

148. The method of Claim 147, wherein the metal alloy is a palladium-platinum alloy.

149. The method of Claim 145, wherein:
the metallic pinning agent comprises copper, molybdenum, or cobalt; and
the catalytic particle comprises a palladium-platinum alloy.

150. The method of Claim 141, wherein the mixed-metal oxide in the interior region extends to and is bonded with the pure metal at the exterior surface.

151. The method of Claim 141, wherein the carrier nano-particle comprises a diameter of approximately 10 nanometers or less.

152. The method of Claim 141, wherein the catalytic particle is a nano-particle.

153. The method of Claim 152, wherein the catalytic particle comprises a diameter of approximately 1 nanometer or less.

154. The method of Claim 153, wherein the catalytic particle comprises a diameter of approximately ½ nanometer.

155. The method of Claim 141, wherein the steps of providing the carrier nano-particle and affixing the catalytic particle to the pure metal are performed using a plasma gun.

156. The method of Claim 141, wherein the steps of providing the carrier nano-particle and affixing the catalytic particle to the pure metal are performed using a wet chemistry process.

157. The method of Claim 141, wherein providing the carrier nano-particle comprises:

loading a quantity of a first material and a quantity of a second material into a plasma gun in a desired ratio;

vaporizing the quantity of the first material and the quantity of the second material, thereby forming a vapor cloud comprising vaporized first material and vaporized second material; and

quenching the vapor cloud, thereby condensing the vaporized first material and the vaporized second material into the carrier nano-particle, wherein the carrier nano-particle comprises the interior region and the exterior surface.

158. The method of Claim 157, wherein the step of vaporizing the quantity of the first material material and the quantity of the second material comprises:

flowing a working gas into a reactor of the plasma gun;

delivering energy to the working gas, thereby forming a plasma stream; and flowing the quantity of the first material and the quantity of the second material into the plasma stream.

159. The method of Claim 157, wherein the step of affixing the catalytic particle to the pure metal is performed using a wet chemistry process.

160. The method of Claim 159, wherein the wet chemistry process comprises:

mixing a catalytic solution with the carrier nano-particle; and

bonding a catalytic particle formed from the catalytic solution to the pure metal on the carrier nano-particle.

161. The method of Claim 160, wherein the catalytic solution is a platinum solution and the catalytic particle is a platinum nano-particle.

162. The method of Claim 161, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

163. A method of forming catalytic nano-particles, the method comprising:

providing a plurality of carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal; and

affixing a catalytic nano-particle to the pure metal on each carrier nano-particle.

164. The method of Claim 163, wherein the step of affixing the catalytic nano-particle to the pure metal is performed using a plasma gun.

165. The method of Claim 164, wherein the step of affixing the catalytic nano-particle to the pure metal comprises:

loading the plurality of nano-particles and a quantity of catalytic material into a plasma gun in a desired ratio;

vaporizing the quantity of catalytic material, thereby forming a vapor cloud comprising vaporized catalytic material; and

quenching the vapor cloud, thereby condensing the vaporized catalytic material into catalytic nano-particles that are bonded to the pure metal on the carrier nano-particles.

166. The method of Claim 164, wherein the plurality of carrier nano-particles has an average grain size of approximately 10 nanometers or less.

167. The method of Claim 166, wherein each catalytic nano-particle comprises a diameter of approximately 1 nanometer or less.

168. The method of Claim 167, wherein each catalytic nano-particle comprises a diameter of approximately ½ nanometer.

169. The method of Claim 163, wherein the step of affixing the catalytic nano-particle to the pure metal is performed using a wet chemistry process.

170. The method of Claim 169, wherein the step of affixing a catalytic nano-particle to the pure metal using a wet chemistry process comprises:

mixing a catalytic solution with the plurality of carrier nano-particles; and

bonding catalytic nano-particles formed from the catalytic solution to the pure metal on the plurality of carrier nano-particles.

171. The method of Claim 170, wherein the catalytic solution is a platinum solution and the catalytic nano-particles are platinum nano-particles.

172. The method of Claim 171, wherein the platinum solution is a platinum nitrate solution or a platinum chloride solution.

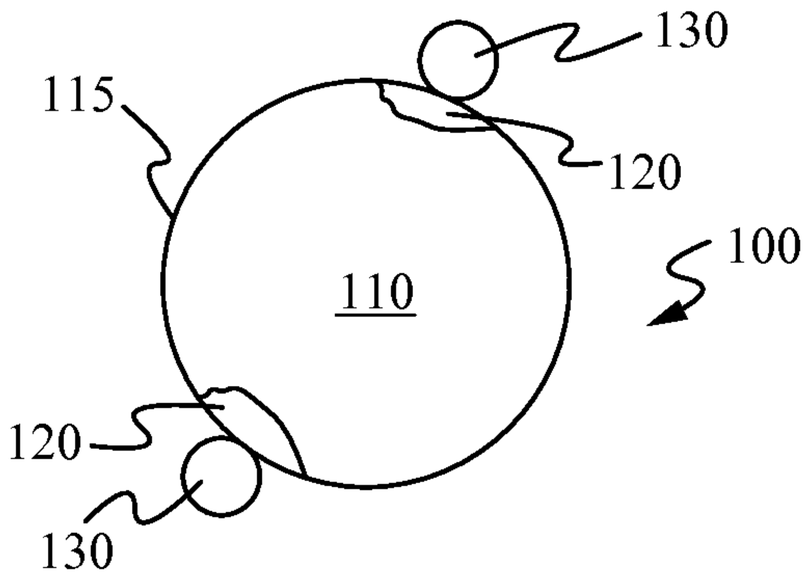

Fig. 1

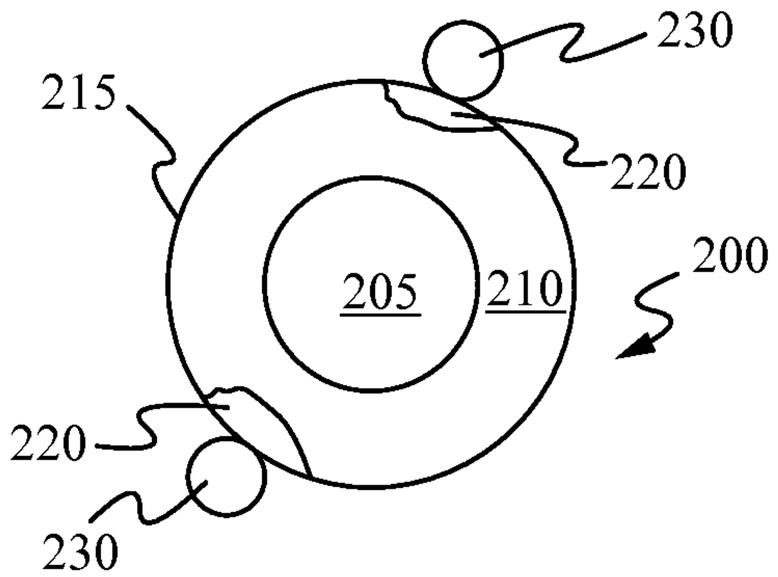

Fig. 2

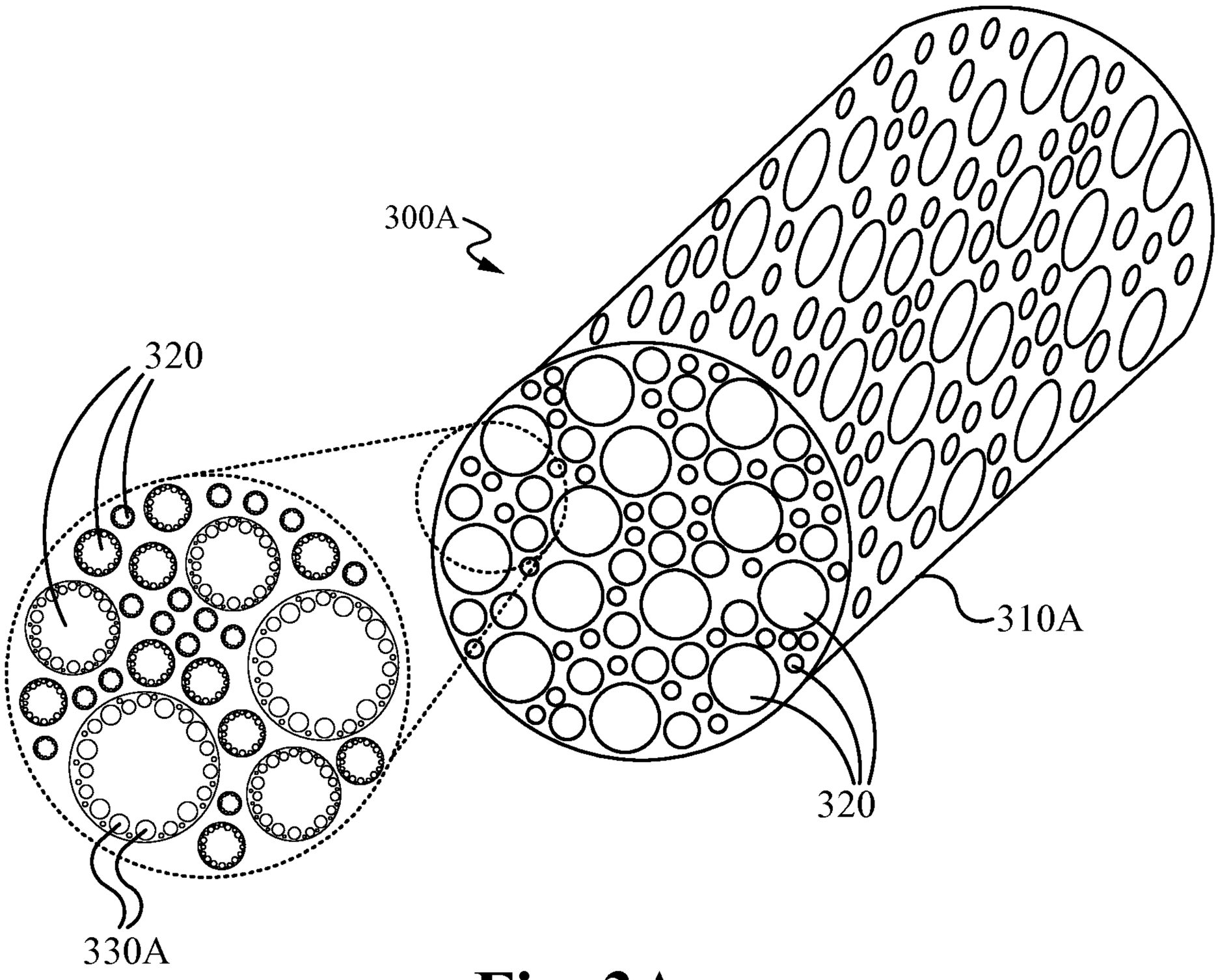

Fig. 3A

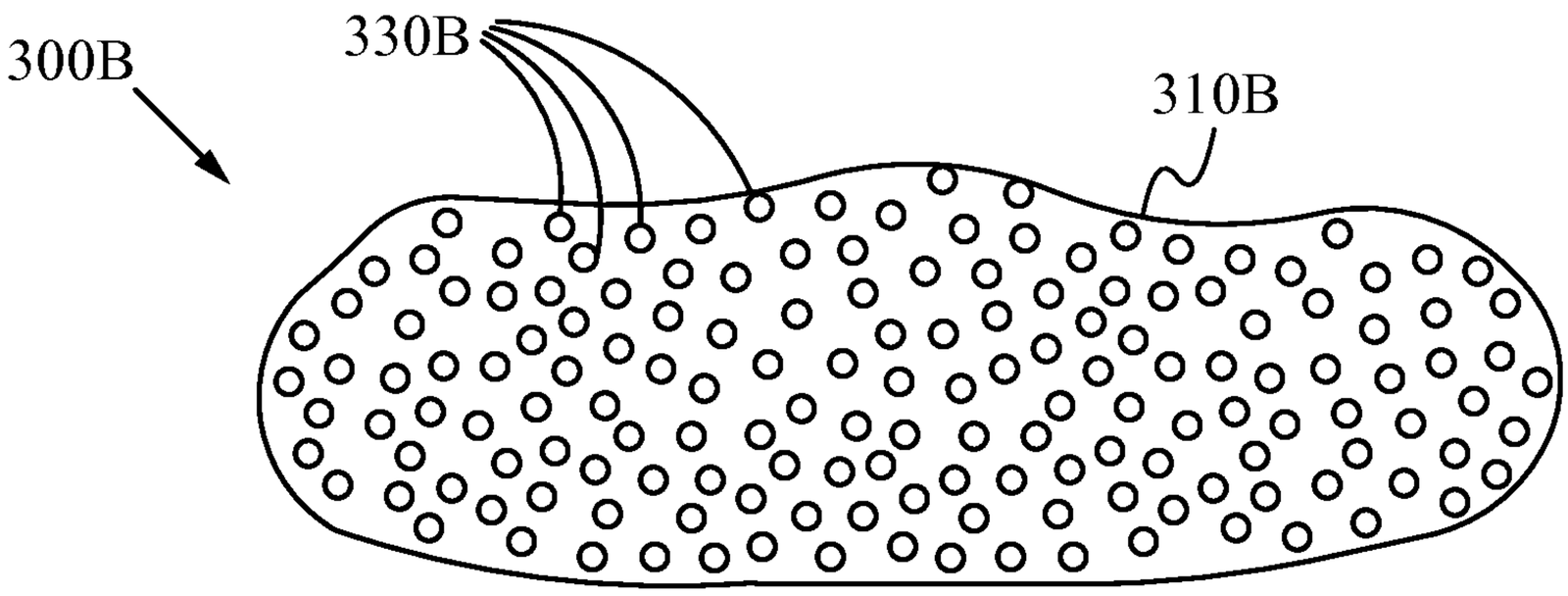

Fig. 3B

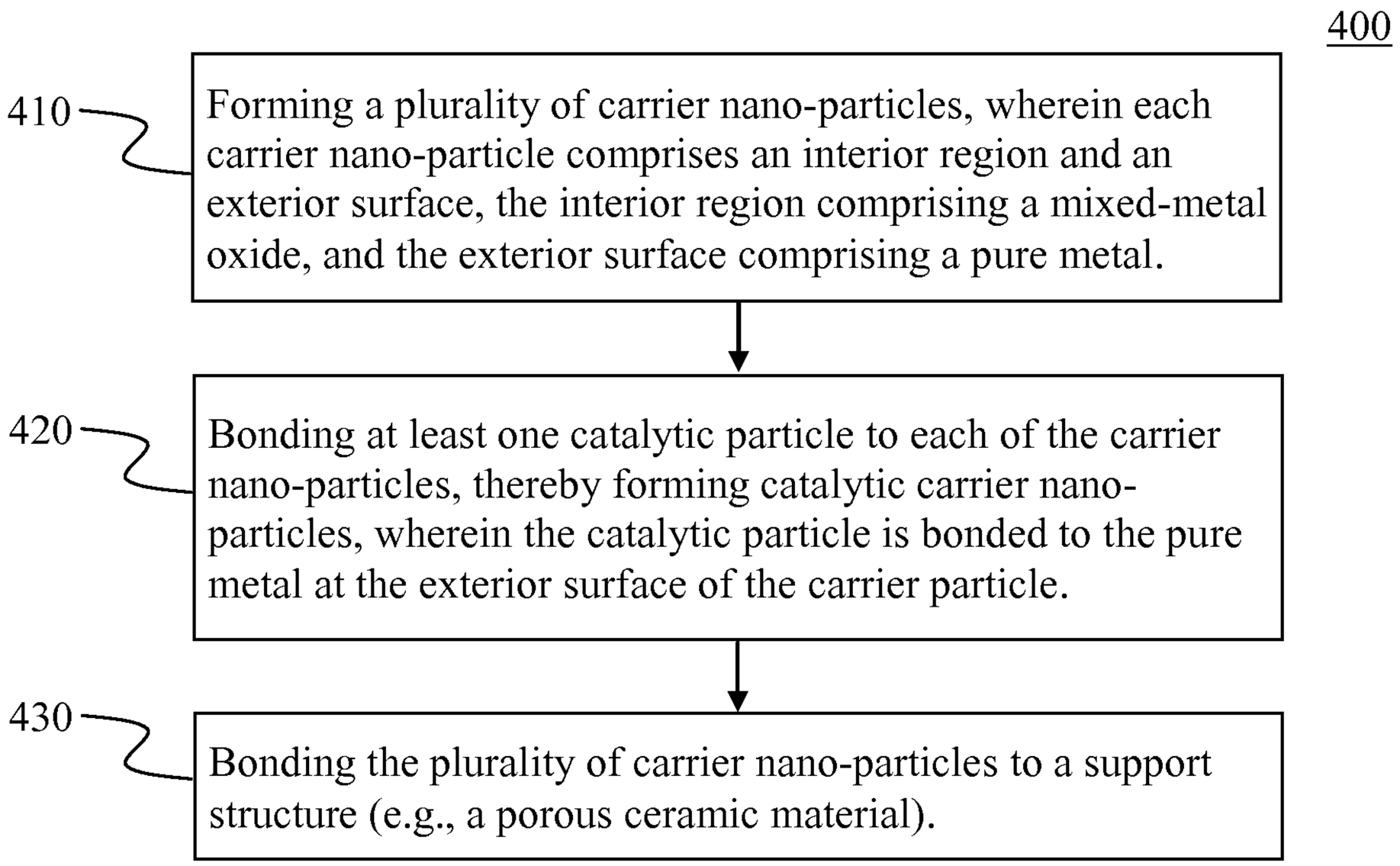
400
410
Forming a plurality of carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal.
420
Bonding at least one catalytic particle to each of the carrier nano-particles, thereby forming catalytic carrier nano-particles, wherein the catalytic particle is bonded to the pure metal at the exterior surface of the carrier particle.
430
Bonding the plurality of carrier nano-particles to a support structure (e.g., a porous ceramic material).

Fig. 4

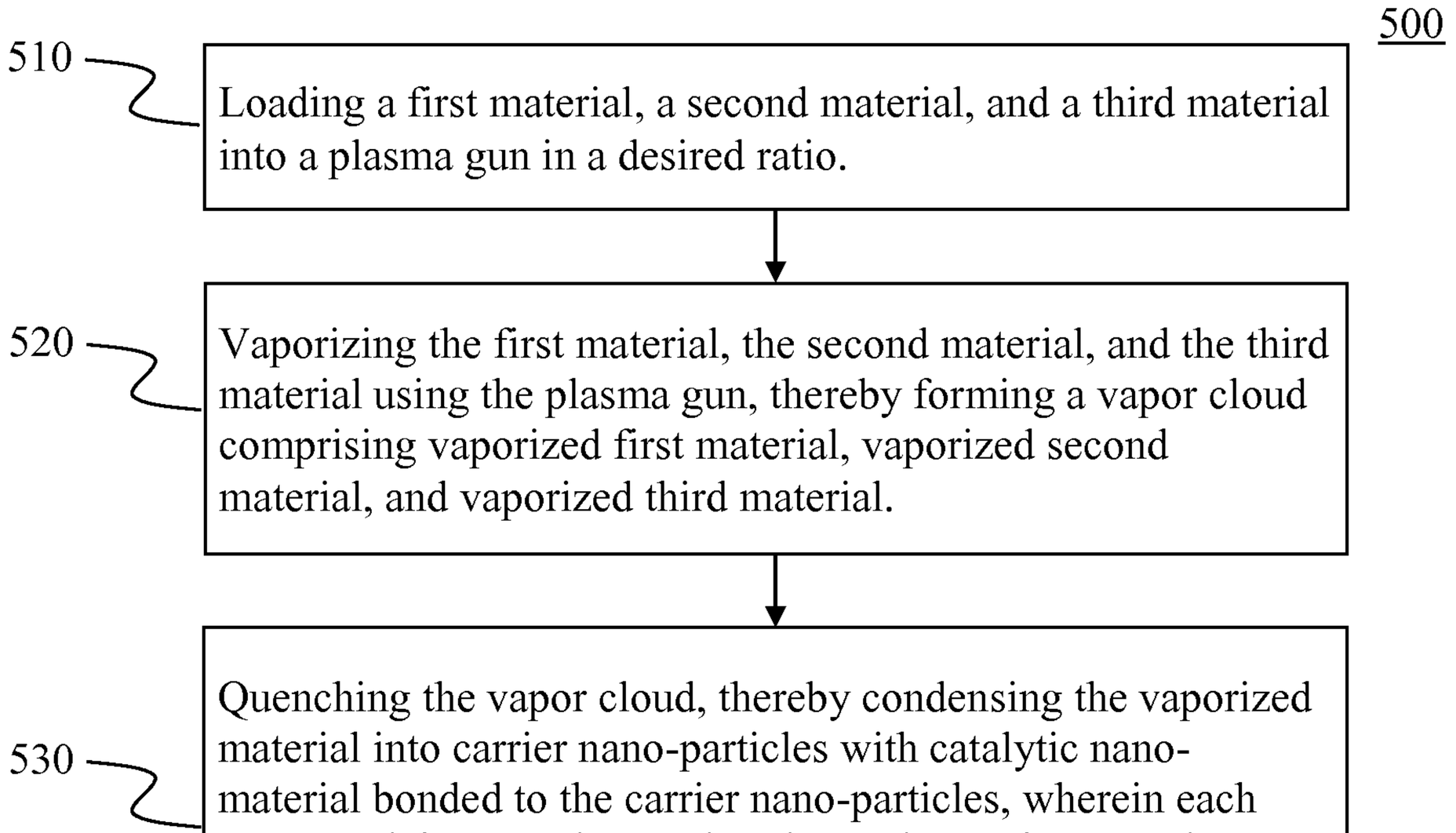
500
510
Loading a first material, a second material, and a third material into a plasma gun in a desired ratio.
520
Vaporizing the first material, the second material, and the third material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material, vaporized second material, and vaporized third material.
530
Quenching the vapor cloud, thereby condensing the vaporized material into carrier nano-particles with catalytic nano-material bonded to the carrier nano-particles, wherein each nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, the exterior surface comprising a pure metal, and the catalytic nano-material bonded to the pure metal at the exterior surface.

Fig. 5

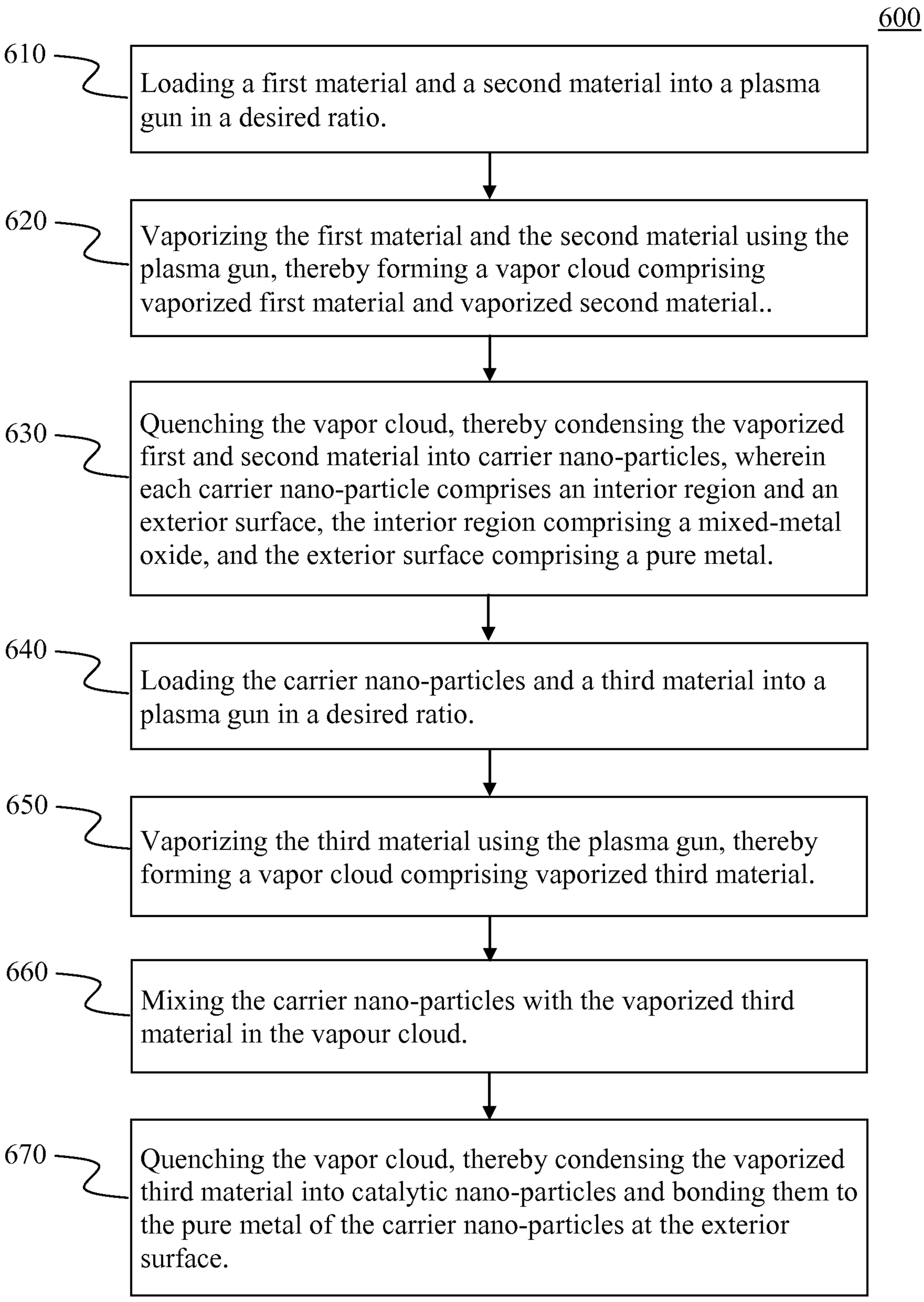
600
610
Loading a first material and a second material into a plasma gun in a desired ratio.
620
Vaporizing the first material and the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material..
630
Quenching the vapor cloud, thereby condensing the vaporized first and second material into carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal.
640
Loading the carrier nano-particles and a third material into a plasma gun in a desired ratio.
650
Vaporizing the third material using the plasma gun, thereby forming a vapor cloud comprising vaporized third material.
660
Mixing the carrier nano-particles with the vaporized third material in the vapour cloud.
670
Quenching the vapor cloud, thereby condensing the vaporized third material into catalytic nano-particles and bonding them to the pure metal of the carrier nano-particles at the exterior surface.

Fig. 6

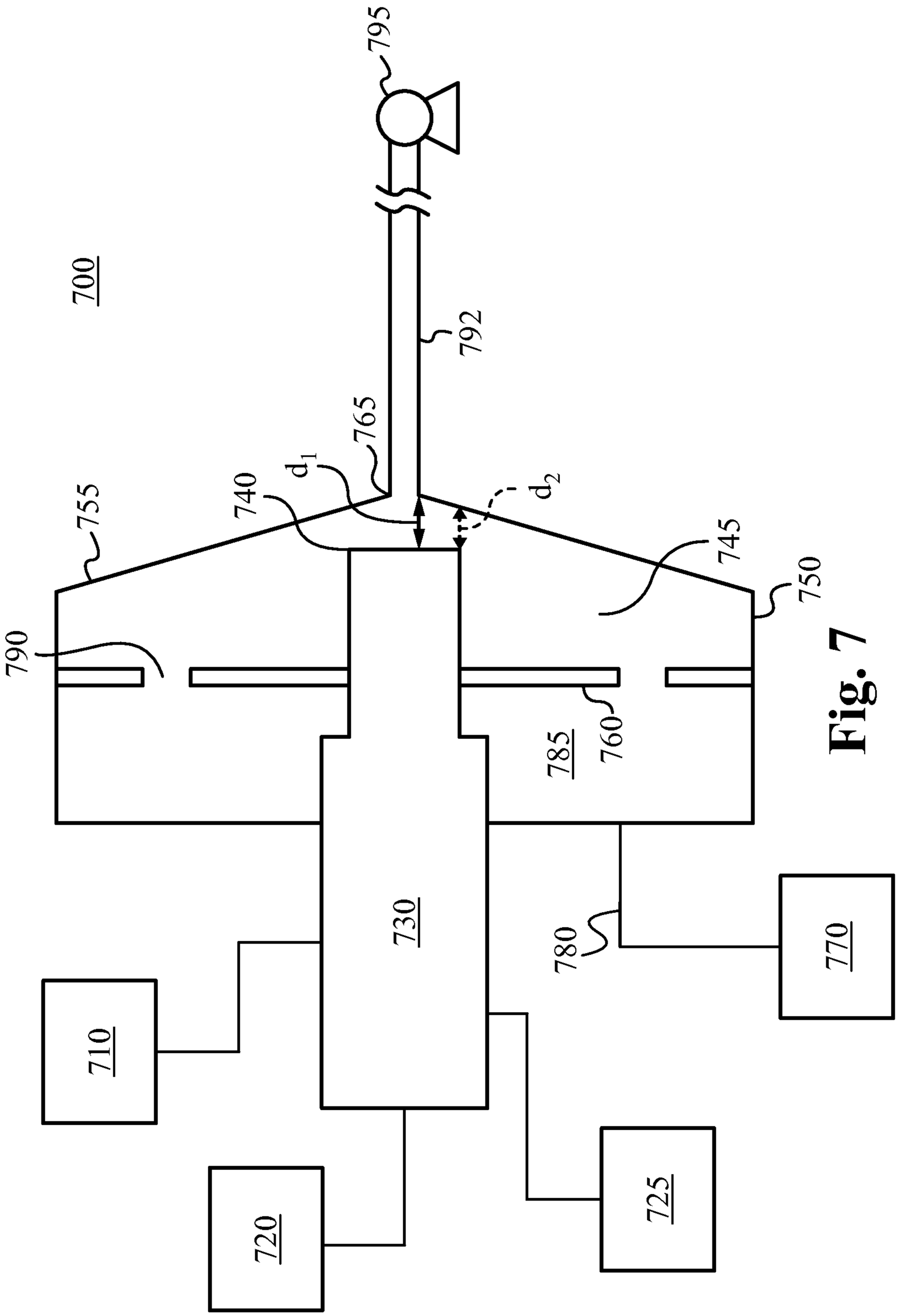
700
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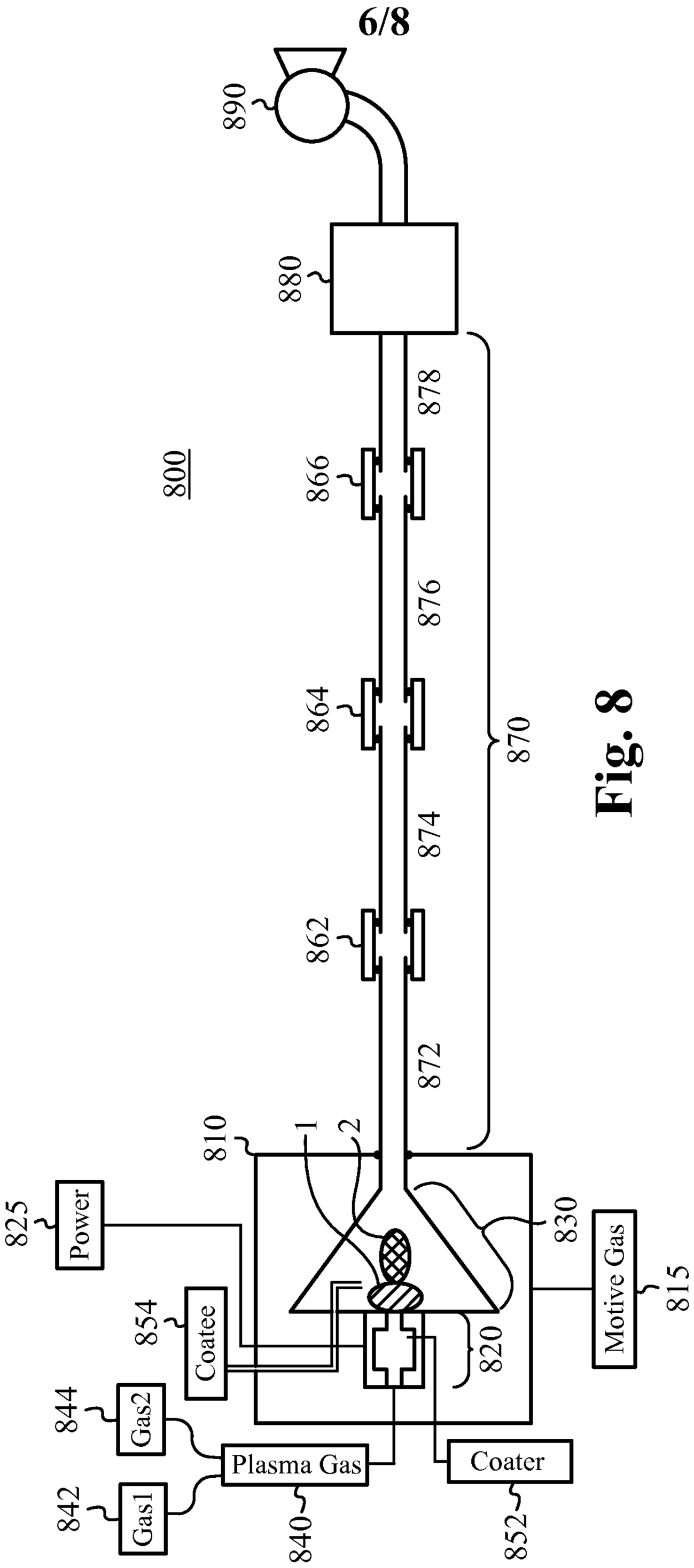
Fig. 7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2
825
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830
Motive Gas
815
844
Gas2
Plasma Gas
Coater
842
Gas1
840
852

Fig. 8

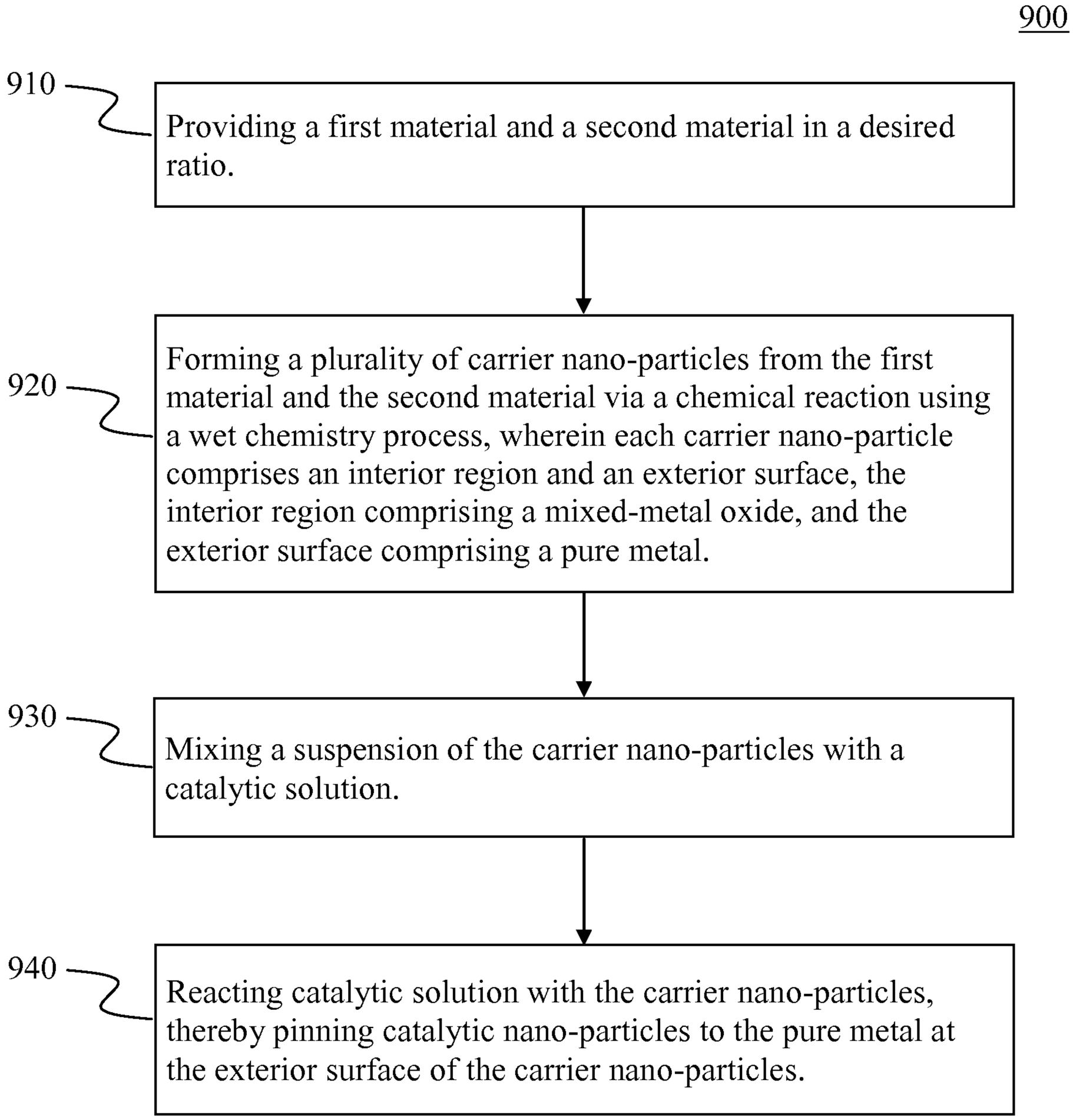
900
910
Providing a first material and a second material in a desired ratio.
920
Forming a plurality of carrier nano-particles from the first material and the second material via a chemical reaction using a wet chemistry process, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal.
930
Mixing a suspension of the carrier nano-particles with a catalytic solution.
940
Reacting catalytic solution with the carrier nano-particles, thereby pinning catalytic nano-particles to the pure metal at the exterior surface of the carrier nano-particles.

Fig. 9

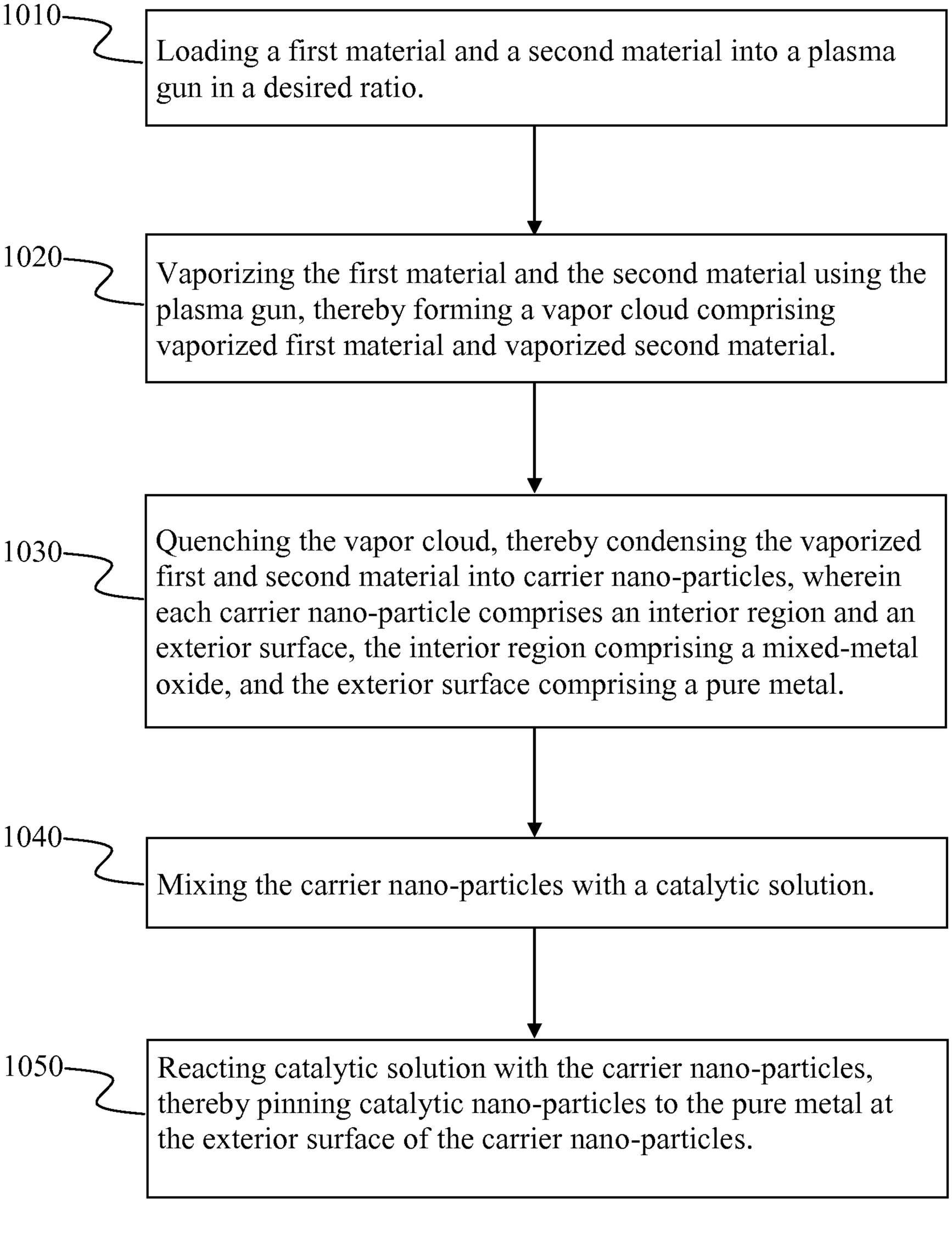
1000
1010
Loading a first material and a second material into a plasma gun in a desired ratio.
1020
Vaporizing the first material and the second material using the plasma gun, thereby forming a vapor cloud comprising vaporized first material and vaporized second material.
1030
Quenching the vapor cloud, thereby condensing the vaporized first and second material into carrier nano-particles, wherein each carrier nano-particle comprises an interior region and an exterior surface, the interior region comprising a mixed-metal oxide, and the exterior surface comprising a pure metal.
1040
Mixing the carrier nano-particles with a catalytic solution.
1050
Reacting catalytic solution with the carrier nano-particles, thereby pinning catalytic nano-particles to the pure metal at the exterior surface of the carrier nano-particles.

Fig. 10

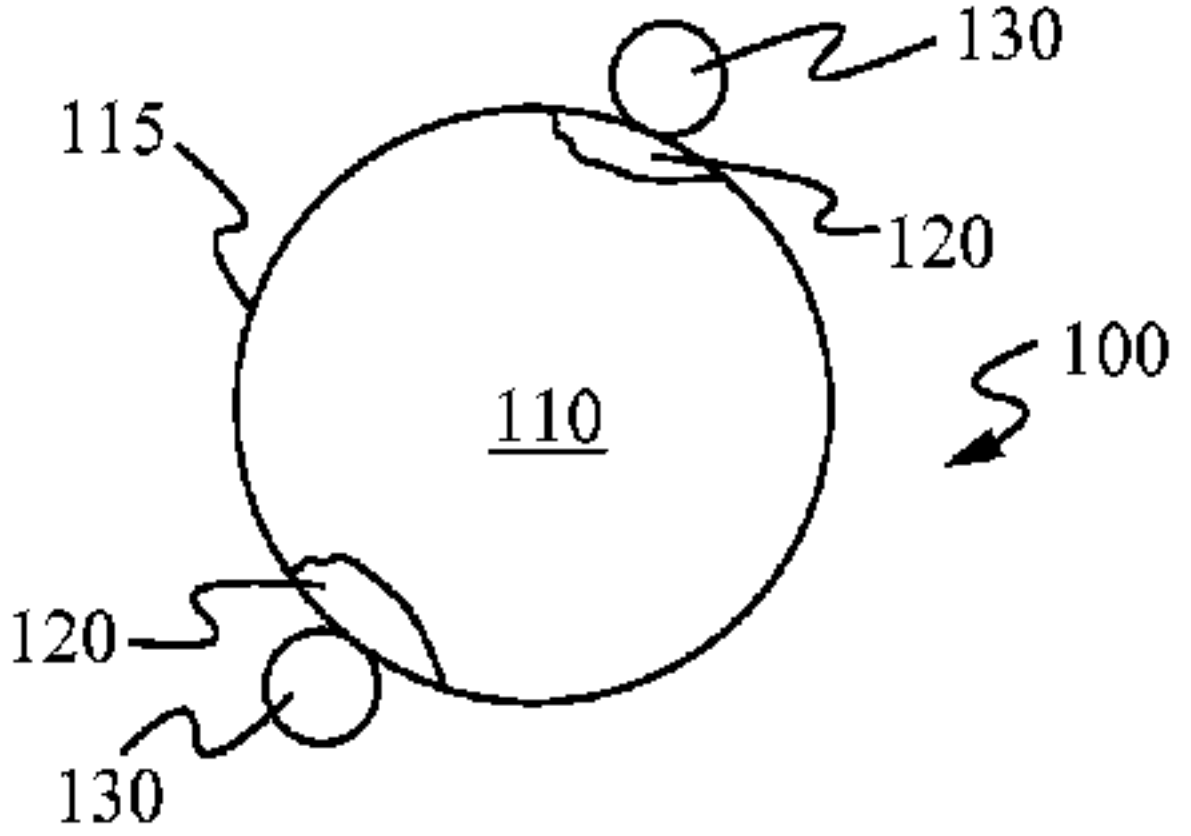

Fig. 1

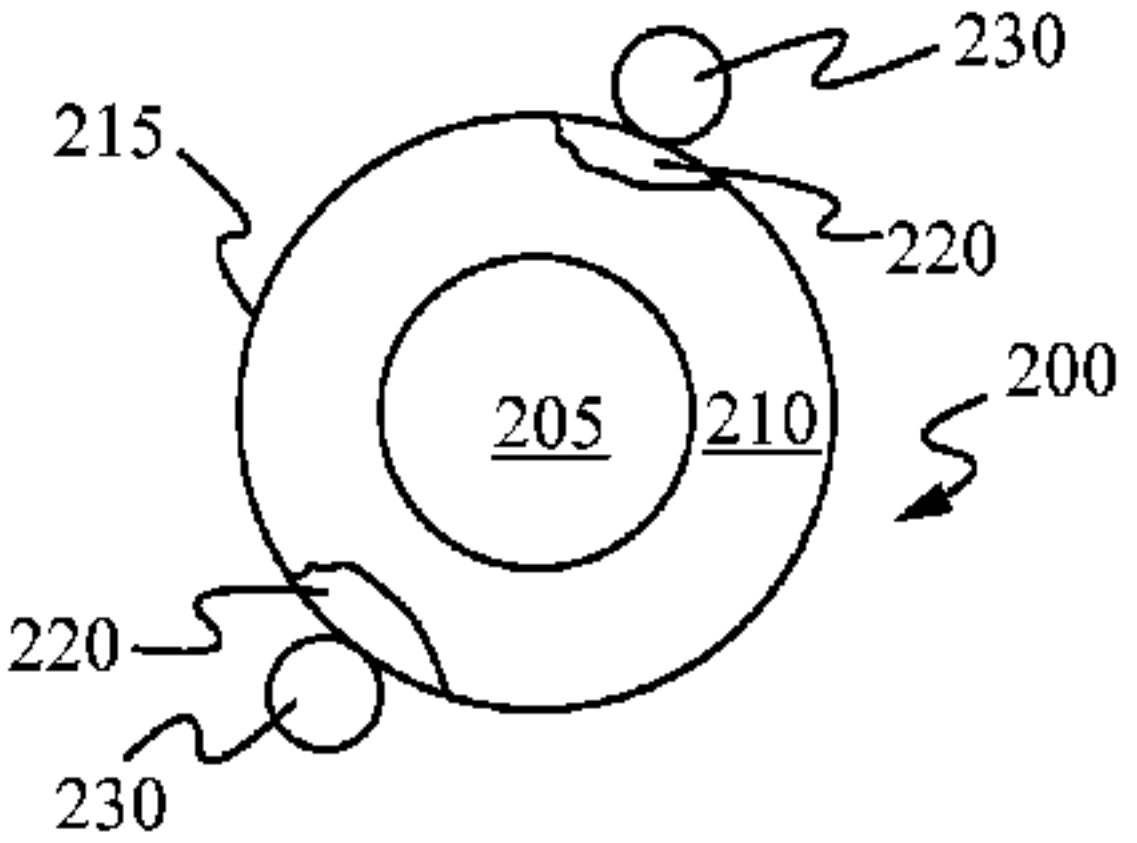

Fig. 2