## UNITED STATES PATENT OFFICE.

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## REMOVAL OF OXIDS FROM FERROUS METAL.

No Drawing.

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To all whom it may concern:

Be it known that we, RUDYARD PORTER and JOSHUA C. WHETZEL, citizens of the United States, and residents of Pittsburgh, in the 5 county of Allegheny and State of Pennsyl- longer time of immersion. In other words, from Ferrous Metal, of which the following tion. is a specification.

particularly to a method of treating ferrous may be noted after a few minutes immersion. metal articles having an oxid or scale thereand, therefore, is not readily removable by

first so treat the articles containing the acid necessary to remove the oxid. resisting oxid or scale as to break down the 25 acid resisting qualities of the oxid prior to less steel to remove the oxid or scale, rela-30 obtaining a considerable saving in acid and cles are substantially free from etching. producing a superior quality of oxid or scale

oxid is especially resistant to acid removal.

In carrying out our invention the metal approximately 170 degrees Fahrenheit.

In the interpretation of the metal approximately 170 degrees Fahrenheit.

It will be understood that the percentages 40 containing the oxid is first immersed in a solution may vary but we have found that particularly described in connection with a 30 per cent solution of sodium hydroxid the treatment of stainless steel, it is not limis very efficient when heated to a temperature of approximately 235 degrees Fahren- be used in the treatment of any or all ferheit. The action of this alkaline solution rous metals. is such as to break down the resistance of 50 the oxid or scale to acid so that it is readily

found that immersion of the oxid or scale solution. coated article in the above solution for ap-

proximately one hour is sufficient to break 55 down the resistance of the oxid to the acid. The same results may be accomplished however with a weaker or cooler solution with vania, have invented certain new and useful the time of immersion varies in proportion 60 Improvements in the Removal of Oxids to the strength and temperature of the solu-

It has also been found that with oxids less This invention relates to the treating of resistant than those found on stainless steel, ferrous metal to remove the oxid or scale such long time immersions are not required, 65 and while not limited thereto relates more in fact some effect on the oxid of plain steel

After the article has been treated by suf-15 on which is particularly resistant to acid ficient immersion in the above described alkaline solution, it is then immersed in an 70 the ordinary or well-known acid pickling acid pickling bath to completely remove the without the use of excessively strong acid oxid or scale, and due to the fact that the solutions and long time immersions which resistance of the oxid to the acid has been broken down by the prior alkaline solution, faces on the articles.

a comparatively short immersion in a com-75 The object of the present invention is to paratively weak acid solution is all that is

Heretofore in the acid pickling of staintheir immersion in the acid pickling bath, tively strong acid solutions have been necesthus eliminating the use of excessively strong sary with long immersions, which resulted acid solutions and materially reducing the in badly etched surfaces, while with the prestime of immersion in the acid bath, thereby ent method the surfaces of the pickled arti-

In carrying out our method, we have ob- 85 free articles, the faces of which are not tained clean surfaces with various strengths etched by the acid.

Our invention is particularly adaptable

of muriatic and nitric acids, from 5 to 15

our invention is particularly adaptable

per cent, by varying the temperature and

to removing the oxid or scale formed on time of immersion, using the higher strength

stainless steel, especially the oxid resulting acid at the lower temperature of approxi
90 from annealing stainless steel, since such mately 100 degrees Fahrenheit and lower strength acid at the higher temperature of

strong alkaline solution, a solution of a and temperatures given herein are only given 95 caustic alkali such as sodium hydroxid for purposes of illustration and not as limi-(NaOH) is preferable. The strength of the tations, and also while the method has been particularly described in connection with ited to the treatment of such steel, but may 100

We claim-

1. A method of removing oxid from ferrous metal articles which includes as a step 105 removable by acid pickling.

In the treatment of stainless steel we have

rous metal articles which includes as a step
the immersion of the articles in an alkaline

2. A method of removing oxid from fer-

rous metal articles which includes as a step the immersion of the articles in a strong caustic alkaline solution.

3. A method of removing oxid from fer-5 rous metal articles which includes as a step the immersion of the articles in a strong caustic soda solution.

4. A method of removing oxid from ferrous metal articles which includes as a step 10 the immersion of the articles in an alkaline

solution prior to acid pickling.

5. A method of removing oxid from ferrous metal articles consisting in first immersing the articles in an alkaline solution to break down the resistance of the oxid to acid, and then immersing the article so treated in an acid pickling bath to remove the oxid.

6. A method of removing oxid from ferrous metal articles consisting in first immersing the articles in a caustic alkaline
solution to break down the resistance of the
oxid to acid, and then immersing the article
so treated in an acid pickling bath to remove the oxid.

7. A method of removing oxid from ferrous metal articles consisting in first immersing the articles in a hot alkaline solution to break down the resistance of the oxid to acid, and immersing the articles so treated in a hot acid pickling bath to remove the oxid.

8. A method of removing oxid from ferrous metal articles which consists in first immersing the articles in a hot bath consisting of approximately a 30 per cent solution of a caustic alkali to break down the resistance

of the oxid to acid, and then immersing the articles so treated in a hot acid pickling bath consisting of a solution of approximately 40 from 5 to 15 per cent of one of the usual pickling acids such as muriatic or nitric acid.

9. A method of removing oxid from ferrous metal articles which consists in first immersing the articles in a caustic alkaline so-45 lution to break down the resistance of the oxid to acid, the time of said immersion being varied according to the strength and heat of said solution, since either the heating or strengthening of the solution or both 50 together will quicken its action on the oxid, and then immersing the articles so treated in an acid pickling bath to remove the oxid.

in an acid pickling bath to remove the oxid.

10. A method of removing oxid from ferrous metal articles which consists in first immersing the articles in a caustic alkaline solution to break down the resistance of the oxid to acid, the time of said immersion being varied according to the strength and heat of said solution, since either the heating or strengthening of the solution or both together will quicken its action on the oxid, and then immersing the articles so treated in a hot acid pickling bath to remove the oxid, the time of immersion in the acid pickling bath being varied according to its temperature and strength, since either the heating or strengthening of the acid bath or both together will quicken its action on the oxid.

In testimony whereof we have hereunto 70

set our hands.

RUDYARD PORTER. JOSHUA C. WHETZEL.