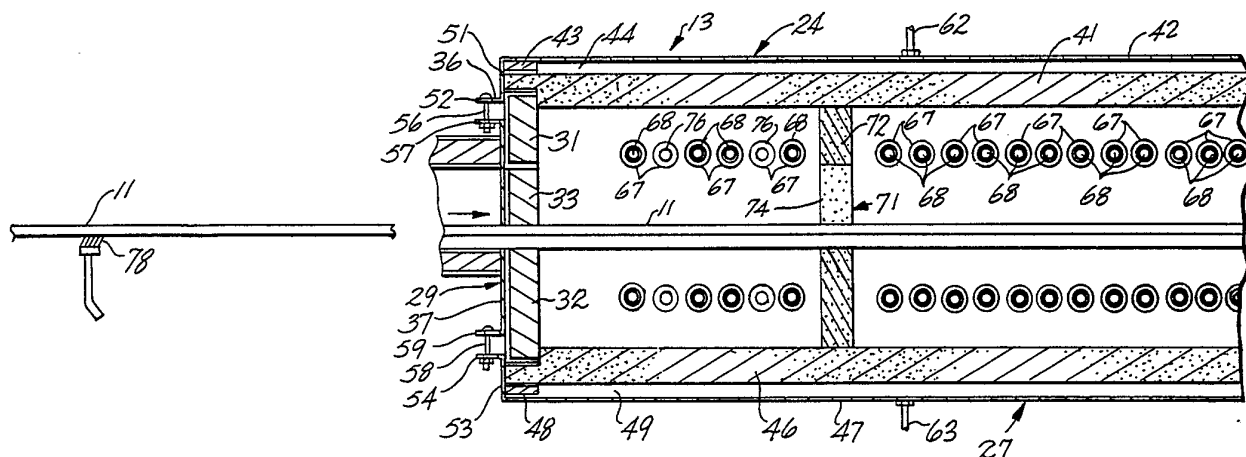


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**(54) Title: APPARATUS AND METHOD FOR RAPIDLY REMOVING ORGANIC MATERIALS FROM FILMS BY HEATING IN AN ELECTRIC FIELD**



**(57) Abstract**

An apparatus and process for rapidly removing organic components from films. The apparatus comprises a heating chamber (13) having one or more heating elements (68). A pair of spaced-apart electrodes (76, 11) are provided for establishing an electric field in the heating chamber. The process comprises heating the film in the presence of the electric field to a temperature sufficient to cause the removal of organic components from the film.

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10 APPARATUS AND METHOD FOR RAPIDLY REMOVING ORGANIC MATERIALS  
FROM FILMS BY HEATING IN AN ELECTRIC FIELD

Cross Reference to Related Applications

15 This application is a continuation-in-part of Serial  
No. 797,629 filed November 13, 1985, the disclosure of  
which is incorporated herein by reference.

Field of the Invention

20 This invention relates to an apparatus and method for  
removing organic components from films, particularly thick  
film compositions, by heating in an electric field.

Background of the Invention

25 Thick films comprise one or more layers of conductive  
and/or dielectric inks which are deposited on insulative  
substrates to form electronic circuit elements such as  
resistors, conductors, capacitors, and the like. Thick  
films are typically applied to the substrate as a paste,  
the pattern for the desired electronic circuit component  
30 being formed by the use of a fine mesh screen utilizing  
silk screening techniques. Once the desired pattern is  
formed, the substrate is dried at low temperatures and  
then fired at high temperatures.

35 Thick film pastes generally include organic-containing  
solvents and a temporary organic binder. The organic

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1 binder serves to provide dimensional stability to the  
dried print and to give it green strength for handling  
through firing. As the thick film heats up in the firing  
cycle, the first thing that occurs is burnout, i.e. the  
5 removal of any remaining solvent and the temporary organic  
binder. Following solvent and binder removal, the conductive  
or dielectric particles of the paste fuse together.

In the past, the firing process was accomplished in a  
conventional furnace by gradually heating the material  
10 which allows slow binder removal. Slow heating however,  
results in the development of glassy layers on the surface  
of the thick films that inhibit later soldering and bonding  
operations. Recently, infrared furnaces have been utilized  
to fire thick film compositions. Infrared furnaces are  
15 characterized by extremely high heating rates and fast  
processing.

The removal of temporary organic binder components,  
from wet or dried thick film compositions is a critical  
and difficult-to-control process in either conventional or  
20 infrared furnaces. For successful firing, the binder must  
be completely removed before any of the conductive or  
dielectric particles fuse together. Binder removal must  
occur cleanly, i.e., no reducing species can be allowed to  
remain to attack the fused conductive or dielectric particles.  
25 Moreover, the decomposed binder components must be removed  
from the process area immediately.

In the case of nitrogen fireable thick films, it has  
been advocated that high velocity, high volume nitrogen  
spargers be used to scrub the decomposing binder away from  
30 the film during burnout and thus minimize the residual  
carbon in the film. This technique is undesirable because  
it greatly increases the nitrogen requirement of the  
firing process and its associated cost, requires preheating  
of the nitrogen to prevent gross variation in temperature  
35 at a substrate surface, and introduces a large and poorly

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- 1 controllable gas velocity and temperature distribution  
across the burnout area.

Summary of the Invention

- 5 The present invention provides a method and apparatus  
for enhancing the removal of organic material from organic-  
containing films such as conductive and dielectric thick  
films. The method comprises establishing an electric  
field around the film while heating the film, preferably  
10 to a temperature sufficient to cause at least some organic  
components of the film to form organic ions, i.e., carbanions  
and carbonium ions.

- The strength of the electric field is sufficient to  
increase the rate at which organic components are removed  
15 from the film and/or decrease the quantity of organic  
components left in the film as a residue after firing.  
The upper limit is that which detrimentally affects the  
chemical or physical properties of the films or at which  
the atmosphere surrounding the thick film, e.g., air or  
20 nitrogen, begins to break down. It is preferred that an  
electric field strength of at least about 200 V/cm, preferably  
at least about 4000 V/cm and preferably no more than about  
7500 V/cm be used.

- The orientation of the electric field is not critical.  
25 However, the electric field is preferably established in  
an orientation normal to the top surface of the film,  
preferably with a polarity that exerts an upward force on  
carbonium ions in the thick film.

- An apparatus for removing organic components from  
30 organic-containing films in accordance with the present  
invention comprises a heating chamber having one or more  
heating elements, preferably infrared lamps. A first voltage  
source is connected to the heating elements to provide energy  
to the heating elements. At least two spaced-apart elec-  
35 trodes are provided which are connected to a second voltage

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1 source for generating an electric field within the chamber.  
Means are also provided for supporting a substrate on  
which an organic-containing film has been deposited within  
the electric field generated by the electrodes. Preferably,  
5 means are also provided for introducing a select, preferably  
non-reactive gas, e.g., nitrogen, into the chamber.

In a preferred embodiment of the invention, the apparatus comprises a heating chamber having entrance and exit openings. A motor driven metal conveyor belt runs through  
10 the chamber, passing through each opening. Substrates on  
which an organic-containing film has been applied can be  
carried through the chamber on the conveyor. Heating  
elements, preferably infrared lamps, are mounted in the  
chamber, preferably above and below the conveyor and are  
15 electrically connected to a voltage source. Within the  
chamber and preferably adjacent the entrance opening, one  
or more electrodes are mounted above the conveyor belt,  
which acts as a second electrode. The electrode(s) and the  
metal conveyor belt are electrically connected to a voltage  
20 source for generating an electric field within the chamber.  
Control circuits are provided for regulating the speed of  
the conveyor, the amount of energy radiated by the infrared  
lamps and the strength of the electric field generated  
between the electrodes.

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1 Brief Description of The Drawings

These and other features and advantages of the present invention will be better understood by reference to the following detailed description when considered in conjunction  
5 with the accompanying drawings wherein:

FIG. 1 is a schematic block diagram of an infrared furnace constructed in accordance with the present invention;

FIG. 2 is a perspective view of a heating chamber of an infrared furnace constructed in accordance with the  
10 present invention;

FIG. 3 is a longitudinal cross-sectional view of the heating chamber shown in FIG. 2;

FIG. 4 is a transverse cross-sectional view of the heating chamber shown in FIG. 2 and 3 taken along lines  
15 4-4; and

FIG. 5 is a enlarged cross-sectional view showing the end portion of the electrode mounted in a ceramic holder secured to the side wall of the heating chamber.

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# 1 Detailed Description of The Invention

A particularly preferred applications of the present invention is in providing an apparatus and process for firing thick films. In the process, the thick films and the insulated substrates on which thick films have been applied are heated to a temperature and for a time sufficient to cause the removal of organic components from the film and then to fuse the conductive or dielectric particles present in the film. At least the initial heating of the thick films to remove organic components is performed in the presence of an electric field. The presence of the electric field enhances the removal of the organic components from the thick film, increasing the rate at which the organic components are removed and/or decreasing the residual amount of carbon remaining in the film after firing.

In the firing process, the thick film is heated to a temperature of about 900°C. The temporary organic binder is removed in the initial stages of the process, generally at temperatures below about 600°C. During the initial heating, at least some of the organic components undergo pyrolysis or combustion to form organic ions, e.g., carbonium ions and carbanions. This typically occurs at about 300°C in air and about 350°C in nitrogen.

Accordingly, the strength of the electric field must be sufficient to increase the rate of organic removal and/or quantity of organic components removed from the thick film during heating. For most conventional thick films, an electric field strength of at least about 200 V/cm is required to produce any significant improvement in the rate and/or quantity of organic removal. The maximum electric field strength is that which the atmosphere surrounding the thick film, e.g., air, nitrogen, oxygen, and the like, begins to breakdown. This occurs at about 7500 V/cm for air and nitrogen.



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1        It is believed that there are two separate mechanisms  
involved in the enhancement of organic removal from thick  
films by the use of the electric field. First, the electri-  
cal field induces a force on the carbonium ions or carbanions  
5 formed in the initial stages of the firing process in a  
direction out of the thick film.

Secondly, during the heating of a thick film, a boundary  
layer develops at the surface of the thick film. The  
boundary layer is a static or stagnant layer of the gas in  
10 which the thick film is being heated which is adjacent to  
the thick film surface. The boundary layer resists heat  
conduction to the film and mass diffusion away from the  
film and is thus a controlling factor in a heating process  
for removing organic components from thick films. It has  
15 been found that the presence of an electric field around  
the thick film disturbs the boundary layer and, if sufficiently  
strong, will eliminate or at least reduce the thickness of  
the boundary layer. This permits not only more rapid  
heating of the thick film but more rapid transport of  
20 organic components away from the surface of the thick  
film.

Accordingly, it is preferred that the electric field  
be sufficient to eliminate or at least significantly reduce  
the thickness of the boundary layer. It has been found  
25 that an electric field having a strength of about 1000  
V/cm, is sufficient for this purpose.

The orientation of the electric field i.e., the dir-  
ection of the lines of force while not critical, is prefer-  
ably normal to the upper surface of the thick film and the  
30 top surface of the insulative substrate. Since the thickness  
of conventional thick films is generally much less than  
the width of the thick film, such an orientation generates  
a force acting on charged organic ions, either positive or  
negative ions depending on the polarity of the electric  
35 field, in a direction towards the top surface of the thick

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1 film which for the majority of the charged ions is the  
closest surface.

The polarity of the electric field is not critical.  
However, it has been found that a greater reduction in the  
5 carbon content of thick films is accomplished when the  
electric field has a polarity which induces a force on  
carbonium ions in a direction towards the upper surface of  
the thick film, than when the electric field has a polarity,  
which induces a force on carbanions in a direction towards  
10 the top surface of the thick film.

The reason for this is believed to lay in the pathways  
through which the organic compounds decompose or breakdown  
during the firing process. The decomposition of organic  
material compounds may take place through the cleavage of  
15 carbon-carbon bonds with no charge being developed, i.e.  
through the formation of free-radicals, or it may take  
place through the formation of oppositely charged ions.  
Free-radicals are very short-lived, energetically unfavorable  
species that would not be expected to make up the majority  
20 of the diffusing species. Charged intermediates will be  
subjected to a force when the electric field is applied.  
This force will vary in magnitude and direction depending  
on the strength and polarity of the applied field. If the  
intermediates included charged species, one positive ion  
25 will be created for every negative ion. A random distribution  
of charge over the resulting species would result in a  
population that would show no favor of one polarity over  
the other in terms of diffusion-aiding force. The observed  
difference in result between negative and positive polarities  
30 is believed to be a manifestation of the relative stability  
of carbon ions that may be generated.

The stability of carbon ions depends on the degree of  
substitution at the carbon that carries the charge. Primary,  
secondary, and tertiary carbons are those bonded to one,  
35 two, or three other carbons respectively. The stability

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1 of carbon ions depends on their degree of substitution and  
charge. Positively charged carbon ions, or carbonium  
ions, are most stable when highly substituted. Negatively  
charged carbon ions, or carbanions, are most stable when  
5 not substituted. This relationship is summarized below.

Carbonium ions (+) Primary - Secondary - Tertiary  
Carbanions (-) Tertiary - Secondary - Primary

The relationship between stability and the degree of  
substitution provides an explanation of the observed  
10 difference in residual carbon with polarity of the electric  
field. The more highly substituted and branched fragments  
of the binder are more likely to be positively charged.  
These highly substituted fragments have the smallest  
diffusivity due to their structure. Accordingly, when  
15 the electric field is applied with a polarity that will  
induce a force on these positively charged fragments to  
pull them out of the thick film, the greatest reduction in  
carbon content is seen.

The length of time during which the thick film is  
20 heated in the electric field is also not critical. Preferably,  
the electric field is maintained around the thick film for  
a time sufficient to maximize the removal of carbon from  
the thick film. Such time will depend on the temperature  
to which the thick film is heated while in the electric  
25 field. For example, it has been found that a time of one  
minute in an electric field of 4000 V/cm yields excellent  
results when the thick film is heated within that electric  
field to a temperature of about 600°C. Beneficial results  
would be seen by the use of shorter times. If desired,  
30 the electric field could be maintained around the thick  
film throughout the entire firing process, which normally  
takes about 10 to about 60 minutes and reaches a temperature  
of up to about 900°C or more.

The electric field is created within the firing chamber  
35 by means of two or more electrodes which are connected to

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1 a voltage source. It is preferred that the voltage source  
be a direct voltage source. This creates an electric field  
with a single non-changing polarity which induces a steady  
force on either the carbonium ions or carbanions in a  
5 direction out of the film.

While not preferred, an alternating voltage source could  
be used in the practice of this invention. An alternating  
voltage source would create an electric field having an  
alternating polarity. Such an electric field would not  
10 induce a steady force on either the carbanions or the  
carbonium ions. However, such an electric field could  
still disrupt and reduce or eliminate the boundary layer  
around the thick film and thereby enhance the removal of  
organics from the thick films by that mechanism. The  
15 drawback to alternating voltage sources is that there is  
no steady force acting on either carbanions or carbonium  
ions pulling them out of the thick film. Moreover, alternating  
voltage sources tend to induce current into the thick film  
circuit which is generally undesirable.

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1       The present invention is applicable to both nitrogen  
and air-fireable thick films. It has been found however,  
that the presence of an electric field generally causes a  
greater increase in the rate and quantity of organic com-  
5       ponents removed from nitrogen-fireable thick films then  
from air-fireable thick films. While not being bound by  
theory, it is believed to stem from a difference in the  
mechanism of organic removal in an oxygen containing  
atmosphere versus a non-oxygen containing atmosphere.

10       Air-fireable material benefit greatly from the presence  
and participation of oxygen in burnout, i.e., the initial  
stage in the firing process wherein the temporary organic  
binder is removed. It provides for a rapid combustion  
pathway that appears not to be diffusion-rate controlled.

15       Burnout in nitrogen however, provides a different  
chemical and kinetic picture. The temporary binders of  
nitrogen fireable thick films are high molecular weight  
polymers, e.g. acrylic, and must be removed prior to the  
development of physical film integrity. In some materials,  
20       the binder or other components of the film are designed to  
supply some oxygen to provide for partial combustion of  
the organic material. Nonetheless, decomposition remains  
the primary mechanism for binder removal.

      Electron Spectroscopy for Chemical Analysis (ESCA)  
25       data has been collected that examines the binding energy  
of residual carbon in dielectric films fired both  
conventionally and in infrared equipment. Although bound  
carbon peaks were the same in both cases, the free carbon  
peak was substantially higher in the infrared-fired samples.  
30       This data indicated that thus, in contrast to air firing,  
burnout in nitrogen appears to be diffusion-rate controlled.

      It is believed that the presence of the electric  
field increases the diffusion rate of charged organic ion  
through and out of the thick film and, if sufficiently  
35       strong, disrupts the boundary layer around the thick film

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1 thereby increasing the rate of diffusion of all components  
to and away from the thick film. Thus, because removal of  
organic components in nitrogen appears to be more diffusion-  
rate controlled than removal of organic components in air,  
5 the former benefits to a greater extent by the presence of  
an electric field.

There is also provided an apparatus for practicing  
the invention which is particularly useful in the firing  
of thick film circuits. With reference to Fig. 1, there  
10 is shown schematically an apparatus constructed in accordance  
with the present invention. The apparatus comprises an  
infrared furnace 10 and a metal conveyor belt 11 for  
transporting thick film circuits to be fired through the  
furnace. Infrared furnace 10 comprises a plurality of  
15 interconnected chambers including an entrance chamber 12,  
a heating or firing chamber 13, a cooling chamber 14 and  
an exit chamber 16. Conveyor belt 11 is driven by a motor  
17 having a speed control 18.

A temperature sensor 19 monitors the temperature in the  
20 firing chamber 13 of furnace 10. Responsive to temperature  
sensor 19, a temperature control circuit 21 adjusts the  
magnitude of the voltage applied to a plurality of infrared  
lamps in the firing chamber 13 of furnace 10 by a voltage  
source 22. Additional temperature control loops identical  
25 to temperature sensor 19, temperature control circuit 21,  
and voltage source 22 could be provided if desired for  
different zones in the firing chamber 13 of furnace 10 so  
as to establish the desired temperature profile for the  
thick film circuits on conveyor belt 11. In any case, the  
30 control loop 21 increases and decreases the energy radiated  
by the infrared lamps so as to maintain a constant temper-  
ature within the firing chamber or zone thereof, as the  
case may be.

An electric field control circuit 23 adjusts the  
35 magnitude of the voltage established between one or more

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1 first electrodes in the firing chamber 13 and the metal  
conveyor belt, which acts as a second electrode, by a  
second voltage source 24 and can be used to increase or  
decrease the strength of the electric field generated  
5 between the first electrodes and the conveyor belt. If  
more than one first electrode is utilized, separate electric  
field control circuits 23 and second voltage source 24  
could be provided, if desired, to establish different  
electrical field strengths in different zones in the firing  
10 chamber 13.

With reference to FIGS. 2, 3, and 4, there is shown  
a particularly preferred heating chamber 13 constructed in  
accordance with the present invention. The heating chamber  
13 is generally rectangular in shape with upper and lower  
15 walls 26 and 27, a pair of side walls 28, and a pair of  
end walls 29.

Each end wall 29 comprises an upper insulation member  
31 and a lower insulation member 32 which are vertically  
spaced apart to provide a rectangular port or opening 33  
20 through which the conveyor belt 11 passes. The lower  
insulation member 32 is provided on its upper edge with  
three spaced semi-circular grooves 34. The upper and  
lower insulation members 31 and 32 of the end walls 29  
are covered by upper and lower end sheet-metal covers 36  
25 and 37.

The side walls 28 comprise an elongated, generally  
rectangular side insulation member 38 which is covered by  
a side sheet-metal cover 39. The upper wall 26 also comprise  
an elongated, generally-rectangular upper insulation  
30 member 41 which is covered by an upper sheet-metal cover  
42. The upper sheet-metal cover 42 is spaced apart from  
the upper insulation member 41 by insulated upper spacers  
43 to thereby form an upper plenum 44 between the upper  
sheet-metal cover 42 and the upper insulation member 41.

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1        Likewise, the lower wall 27 comprises an elongated,  
generally-rectangular lower insulation member 46 which is  
covered by a lower sheet-metal cover 47. The lower sheet-  
metal cover 47 is spaced apart from the lower insulation  
5 member 46 by insulated lower spacers 48 to form a lower  
plenum 49 between the lower sheet-metal cover 47 and the  
lower insulation member 46.

      The upper sheet-metal cover 42 comprises a side flange  
51 which extends downwardly around periphery of the upper  
10 sheet-metal cover 42. An upper bracket is provided on the  
side flange 51 of the upper sheet-metal cover 42.

      The lower sheet-metal cover 47 also comprises a side  
flange 53 which extends upwardly around periphery of the  
lower sheet-metal cover 47 and comprises a lower bracket  
15 54. To hold the insulating walls of the heating chamber  
13 in assembled position, the side flange 51 of the upper  
sheet-metal cover 42 is drawn down over the side and upper  
end sheet-metal covers 39 and 36 and is held by tie rods  
56 which pass through openings in the upper bracket 52 of  
20 the upper sheet-metal cover 42 and aligned openings in  
corresponding upper brackets 57 of the side and upper end  
sheet-metal covers 39 and 36. Likewise, the side flange  
53 of the lower sheet-metal cover 47 is drawn up over the  
side and lower end sheet-metal covers 39 and 36 and is  
25 held by tie rods 58 which pass through openings in the  
lower bracket 54 of the lower sheet-metal cover 47 and  
aligned openings in a lower bracket 59 carried by the side  
and lower end sheet-metal covers 39 and 36.

      The conveyor belt 11 is supported within the heating  
30 chamber 13 on three quartz tubes 61 which extend the length  
of the heating chamber 13 and rest on the three semi-circular  
grooves 34 provided on the lower insulation members 32 of  
the end walls 29.

      A cover gas, which may be air or nitrogen, for example,  
35 may be fed under a low pressure through upper and lower



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1 tubular connectors 62 and 63 mounted on the respective  
upper and lower sheet-metal covers 42 and 47. The cover  
gas slowly and evenly filters through the porous upper and  
lower insulating members 41 and 46 which form each of the  
5 top and bottom walls of the heating chamber 13, thus causing  
the interior of the heating chamber 13 to be at a slightly  
higher pressure than the atmosphere surrounding the infrared  
furnace.

Each of the side walls 28 of the heating chamber 13  
10 is provided with an upper series of spaced apart circular  
holes 64 above the conveyor belt 11 and a lower series of  
spaced circular holes 66 below the conveyor belt 11. Each  
of the circular holes 64 and 66 extend through the side  
sheet-metal cover 39 and the side insulation member 38  
15 forming each of the side walls 28. The inner half portion  
of each the holes 64 and 66, and the side insulation member  
38 is provided with a 45° chamfer 67.

A plurality of elongated infrared lamps 68 are mounted  
with their end portions passing through circular openings  
20 64 or 66 and the opposing side walls 28. The infrared  
lamps 68 may be mounted in any suitable manner such as  
that described in U.S. Patent No. 4,406,994 which is fully  
incorporated herein by reference. Infrared lamps 68 are  
thus arranged side by side transverse to the direction of  
25 movement of conveyor belt 11 above and below conveyor belt  
11. Suitable infrared lamps may comprise a transparent or  
translucent elongated quartz tube in the center of which  
is located in electrically heated infrared ray emitting  
filament preferably made of tungsten. The quartz tube is  
30 hermetically sealed at its ends. The ends of the quartz  
tubes are provided with a metal terminal which connects to  
the respective ends of the tungsten filament. Leads are  
connected to each metal terminal. Such infrared lamps are  
well known and are commercially available products, the

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1 tubes generally being charged with an inert gas such as argon.

The energy emitted by such infrared lamps is concentrated in the near infrared band extending approximately from 0.7  
5 microns to approximately 2.5 microns. The exact distribution and the wave-length of the peak power vary as a function of the temperature of the lamp filament. Generally, sufficient electrical voltage is applied to the lamp filament to maintain a filament temperature typically in  
10 the range of 1400° to 2000°K, depending upon the desired operating temperature and the quantity of heat that needs to be transferred to the thick films within the heating chamber 13.

The entrance portion 66 of the heating chamber 13 is  
15 separated from the remaining portion by a vertical wall 71 formed of upper and lower high temperature insulation sheet members 72 and 73 spaced to provide a central rectangular opening 74 for the traveling conveyor belt 11. The series of upper and lower holes 64 and 66 in the side  
20 walls 28 are more closely spaced in the entrance portion 69 of the heating chamber 13 than in the remaining portion.

In the embodiment shown there are six upper holes 64 and six lower holes 66 in the entrance portion 69 of the heating chamber 13. Infrared lamps 68 as previously  
25 described are mounted in the first, third, fourth, and sixth of the upper and lower holes 64 and 69. A pair of upper electrodes 76 are mounted in the remaining two upper holes 66, i.e. the second and fifth holes. The second and fifth lower holes of the entrance portion 69 of the heating  
30 chamber 13 are sealed by ceramic plugs 77.

In this embodiment, the metal conveyor belt 11 forms a second or ground electrode.

The upper electrode 72 and the metal conveyor belt 11 are electrically connected to a voltage source (not shown)  
35 which can be adjusted to establish the desired potential

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1 difference between the upper electrode 72 and the metal  
conveyor belt 11, which is preferably maintained at ground  
potential. As best shown in FIG. 3, electrical contact is  
made between the voltage power source and the metal conveyor  
5 belt 11 by a metal brush 78 which continuously engages the  
conveyor belt 11 at a location outside of the heating  
chamber 13.

With reference to FIG. 5, each upper electrode 76  
comprises a metal wire 79 encapsulated in a quartz tube  
10 81. The wire 79 may be made of any metal which is stable  
at the operating temperatures of the furnace, i.e., up to  
about 1,000°C. The thickness of the wire 79 is also not  
critical. A presently preferred metal wire 74 is made of  
high-temperature stainless steel and has a diameter of  
15 about one millimeter.

The electrode wire 79 must be insulated to prevent  
arcing which may otherwise occur between the electrode  
wire 69 and conductive element at ground potential such as  
the side sheet-metal covers 39 of the sidewall 28, the  
20 metal conveyor belt 11, or even the thick film being  
processed if it contains any metallic particles. The wall  
thickness of the quartz tube 81 must be sufficient to  
prevent such arcing. The actual quartz tube thickness  
will therefore depend on the strength of the electric field  
25 being generated. For example, for an electric field having  
a strength of about 2000 V/cm and requiring an electrode  
potential of 15kV, a quartz tube having a wall thickness of  
about 5 millimeters is suitable. If the electric field  
strength is about 4000 V/cm and requires an electrode  
30 potential of 30,000 V, a quartz tube having a wall thickness  
of about 10 millimeters would be preferred.

As long as the metal conveyor belt 11 is maintained  
at ground potential, i.e. the same potential as the frame  
of the furnace, it need not be insulated.

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1 With reference to FIG. 5, the upper electrodes 76 are  
mounted in the upper holes 61 and in the side walls 28 by  
means of ceramic holders 82. The upper electrodes 76 pass  
through openings in the ceramic holders 82. At one end of  
5 the upper electrode 76, the end of the electrode wire 79  
protrudes from the quartz tube 81 and is connected to a  
generally cylindrical metal male terminal 83 which protrudes  
axially from the ends of the quartz tube 81. The electrode  
wire 79 is connected to the voltage power supply by means  
10 of an insulated lead 84, which, in the embodiment shown is  
generally similar to a spark plug wire. At its end, the  
lead 84 has an insulative rubber cap 86 which fits over  
the male terminal 83 and a portion of the end of the quartz  
tube 81 which protrudes from the sidewall 28. Within the  
15 cap 86, a female connector 87 is provided which receives  
the male terminal 83 of the electrode 76.

The ceramic holder 82 has a hollow cylindrical body 88  
with a shoulder 89 on its outer end and a bottom wall 91  
on its inner end. The bottom wall 91 is provided with a  
20 concentric circular opening 92. The length of the body 88  
of the ceramic holder 82 is approximately one-half the  
thickness of the side insulating member.

The ceramic holder 82 is positioned with its cylindrical  
body 88 having a relatively close fit in the circular hole  
25 66 of the side wall 78 and with its shoulder 89 lying against  
the side sheet metal cover 39 and permanently sealed in  
position by use of a silicone sealant.

To mount an electrode 76 in a pair of opposing ceramic  
holders 82 provided in the upper holes 64 on the sidewalls  
30 28 of the heating chamber 13, the closed end of the electrode  
76, i.e. the end opposite the male terminal 83 is inserted  
into the ceramic holder 82 secured in hole 66 on one of  
the sidewalls 28. The electrode 76 is then pushed, and  
rotated as needed, until it passes through the ceramic  
35 holder 82 in the opposing sidewall 28. The chamber 67 on

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1 the inner portion of the hole 66 aids in guiding the  
electrode 76 into the opening of the opposing ceramic  
holder 82. A gasket 93 formed of resilient refractory  
material, such as alumina fiber, and in the shape of a  
5 short tube, is then fitted over each end of the electrode  
76 and packed tightly within the ceramic holders 82. It  
may be desirable in some cases to apply a silicone sealant  
on the outer end of the cylindrical opening of the ceramic  
holder 82 to provide further sealing.

10 It may be desirable in other cases to saturate the  
short tube-like alumina gasket 93 in a solution of soluble  
refractory material such as sodium fluosilicate just prior  
to inserting it over the ends of the electrodes 76 and  
packing it into the cavity of the ceramic holder 82 around  
15 the quartz tube 81. Upon drying, the alumina gasket 93 is  
effectively bonded to the wall of the ceramic holder 82  
and the wall of the quartz tube 81. This assures that the  
alumina gasket 93 does not work loose in the ceramic holder  
82 as a result of the expansion and contraction of the  
20 sidewalls of the heating chamber 13 and that joint will  
remain gas-tight.

In operation, an insulated substrate on which a thick  
film has been applied is set on the conveyor belt 11 which  
causes the substrate and thick film through the furnace.  
25 Within the entrance portion of the heating chamber 13,  
infrared lamps 68 heat the substrate and thick film,  
preferably to a temperature of at least about 350°C and  
typically to a temperature of about 600°C. Also in the  
entrance portion of the heating chamber 13 an electric  
30 field is established between the upper electrode 76 and  
the metal conveyor belt 11. As the substrate and the  
thick film pass through the entrance portion of the heating  
chamber 13, the temporary organic binder decomposes and  
volatilizes. After removal of the temporary organic binder,  
35 the substrate and thick film passes through the remainder

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1 of the heating chamber wherein the temperature of the  
thick film increases to about 900°C to cause the conductive  
or dielectric particles of the thick film to fuse together.  
The substrate then passes to the cooling and exit chambers.

5 The preceding description of an apparatus has been  
presented with reference to a presently preferred embodiment  
of the invention shown in the accompanying drawings.  
Workers skilled in the art and technology to which this  
invention pertains will appreciate that alterations and  
10 changes in the described apparatus and structure can be  
practiced without meaningfully departing from the principles,  
spirit and scope of this invention.

For example, the preceding description was directed  
to an infrared furnace of a particular design, it is apparent  
15 that any infrared furnace or, indeed, any conventional  
furnace may be used. Particularly preferred furnaces to  
which the present invention is applicable are manufactured  
by Radiant Technology Corporation of Cerritos, California  
and sold as models CU300, CU600, L900 and L1200.

20 While, infrared lamps are preferred, it is apparent  
that any type of heating element may be used. Moreover,  
it is clear that as the apparatus need not comprise a  
conveyor belt which runs through the heating chamber.  
Rather, a conventional furnace simply having a door and a  
25 stationary supporting means for holding the insulative  
substrate and thick film may be employed.

Likewise, the size, shape, location and number of  
electrodes may vary as desired. For example, rather than  
a single quartz encapsulated wire, the upper electrode may  
30 be an insulated plate or screen. Such shapes are not  
preferred for use in the described apparatus because they  
tend to restrict the flow of cover gas within the chamber.  
For other heating chamber designs, such shapes may be  
preferred.

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1 While it is convenient that the lower electrode be  
the metal conveyor belt, electrodes similar to the described  
upper electrodes or electrodes of another shape and design,  
e.g., a plate or screen, may be used, as desired.

5 While quartz is the preferred insulation material  
around the upper electrode because of its high thermal  
shock resistance, any suitable insulative material may be  
used.

10 It is not so apparent that the lower electrode need  
not be at ground potential. However, if at a potential  
other than ground potential, the lower electrode should be  
encapsulated in an insulative material.

The preceding description was also presented with  
reference to a particularly preferred application that-  
15 being the firing of thick films. It is understood that the  
invention is useful in the applications as well. Such  
other applications include the drying and removal of organic  
material from other films, such as solder paste, paint,  
glue, glass or silver glass die attach materials and, in  
20 general, any organic or inorganic film or adhesive from  
which solvents, reaction products or binders need to be  
removed. With respect to the specific applications of thick  
film, the invention is equally applicable to the firing of  
multi-layered films as well as single layered films.

25

## EXAMPLE

An infrared furnace was modified to generate an electric  
field in the entrance portion of the heating chamber. The  
infrared furnace employed was a Model Cu-610 furnace  
manufactured by Radiant Technology Corporation of Cerritos,  
30 California. Within the entrance portion of the infrared  
furnace, the second and fifth upper and lower infrared  
lamps were disconnected and replaced with four electrodes.

Each electrode consisted of a 1mm stainless steel  
wire surrounded by an 11mm diameter quartz tube. The ends  
35 of the electrode protruded out of the heating chamber. At

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1 one end, the quartz tube completely surrounded the end of  
the electrode wire. At the other end, the electrode wire  
protruded from the quartz tube and was connected to a  
cylindrical male connector having a length of about 1cm  
5 and a diameter of about 5mm. A spark plug wire was connected  
to the male connector of each electrode, the fibber cap of  
the spark plug wire fitting over the end of the quartz  
tube. The spark plug wires were connected to a direct  
voltage source, specifically a 0-30kV DC power supply.

10 The configuration implemented had sufficient electrical  
integrity to generate a field of 700 V/cm. This strength  
is insufficient to produce the onset of significant boundary  
layer instability, but was investigated for its ability to  
modify diffusion kinetics sufficiently to produce a significant  
15 change in residual carbon concentrations. Nitrogen-fireable  
dielectric, DuPont 4575, was printed and dried three times  
to a thickness of 125 micrometers before being fired. The  
material was fired in a pure nitrogen atmosphere (<10 ppm  
O<sub>2</sub>) ambient at a speed of 15 ipm. These conditions gave a  
20 12 minute overall cycle with approximately 2.5 minutes at  
900°C. Residual carbon concentrations were measured by  
milling 200A below the surface of the dielectric with Ar<sup>+</sup>  
ions and using small spot ESCA analysis to determine  
atomic concentrations. The results are summarized below:

25

## ELECTRIC FIELD STRENGTH

	0 V/cm	+ 700 V/cm	- 700 V/cm
ATOMIC % CARBON	7.2%	3.4%	2.5%
% DECREASE	-	-53%	-65%

30

For the above reasons, the foregoing description  
should not be read as pertaining only to the precise  
structures and procedures described, but rather should be  
read consistent with and as support for the following  
35 claims which are to have their fullest fair scope.



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1 What Is Claimed Is:

1. A method for removing organic compounds from an organic-compound-containing film comprising, introducing  
5 the film into an electric field and heating the film to a temperature sufficient to decrease the organic content of the film.
2. A method as claimed in claim 1, wherein the  
10 strength of the electric field is at least about 200 V/cm.
3. A method as claimed in claim 1, wherein the strength of the electric field is sufficient to reduce the boundary layer surrounding the film.  
15
4. A method as claimed in claim 1, wherein the strength of the electric field is at least about 4000V/cm.
5. A method as claimed in claim 1, wherein the  
20 orientation of the electric field is normal to the top surface of the film.
6. A method as claimed in claim 1, wherein the temperature to which the film is heated is sufficient to  
25 cause the formation of carbonium ions and carbanions.
7. A method as claimed in claim 6, wherein the polarity of the field is selected to induce a force on carbonium ions in a direction toward the top surface of  
30 the film.
8. A method as claimed in claim 1, wherein the film is a thick film.

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1           9. A method as claimed in claim 8, wherein the film  
is a nitrogen fireable thick film and wherein the film is  
heated in the presence of nitrogen.

5           10. A method for firing thick films comprising;  
              heating the thick film for a predetermined  
period of time to a temperature sufficient to cause the  
non-organic particles of the film to fuse together; and  
              establishing an electric field around the  
10 thick film at the beginning of the predetermined period of  
time and maintaining the electric field for at least a  
portion of the predetermined period of time.

              11. A method as claimed in claim 10, wherein the  
15 strength of the electric field is at least about 200 V/cm.

              12. A method as claimed in claim 10, wherein the  
strength of the electric field is sufficient to reduce the  
boundary layer surrounding the film.

20           13. A method as claimed in claim 10, wherein the  
strength of the electric field is at least about 4000V/cm.

              14. A method as claimed in claim 10, wherein the  
25 orientation of the electric field is normal to the top  
surface of the film.

              15. A method as claimed in claim 10, wherein the  
temperature to which the film is heated is sufficient to  
30 cause the formation of carbonium ions and carbanions.

              16. A method as claimed in claim 10, wherein the  
polarity of the field is selected to induce a force on  
carbonium ions in a direction toward the top surface of  
35 the film.

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1           17. A method as claimed in claim 10, wherein the film  
is a thick film.

5           18. A method as claimed in claim 10, wherein the film  
is a nitrogen fireable thick film and wherein the film is  
heated in the presence of nitrogen.

10           19. An apparatus for decreasing the carbon content  
of a film comprising;  
a heating chamber;  
at least one heating element within the  
heating chamber;

15           a first voltage source electrically connected  
to the heating element for supplying energy to the heating  
element;

a first control circuit for regulating the  
amount of energy supplied by the first voltage source to  
the heating element;

20           a pair of electrodes in the heating chamber;  
a second voltage source electrically connected  
to the electrodes for supplying energy to the electrodes  
to thereby generate an electric field within the heating  
chamber;

25           a second control circuit for regulating the  
amount of the energy supplied to the electrodes from the  
second voltage source;

30           means for supporting a substrate on which a  
film has been applied within the electric field generated  
by the electrodes.

20. An apparatus as claimed in claim 19, wherein the  
heating elements are infrared lamps.

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1        21. An apparatus as claimed in claim 19, wherein the  
electrodes are spaced-apart above and below the supporting  
means.

5        22. An apparatus as claimed in claim 19, where the  
supporting means comprises a conductive platform and wherein  
the conductive platform forms one of the electrodes.

10       23. An apparatus as claimed in claim 19, wherein the  
heating chamber comprises entrance and exit openings and  
wherein the supporting means comprises a conveyor belt  
which passes through the entrance and exit openings and  
the heating chamber.

15       24. An apparatus as claimed in claim 23, wherein the  
conveyor belt is metal and forms one of the electrodes.

25. An infrared furnace for firing thick films  
comprising;

20           a heating chamber having an entrance opening  
and an exit opening;

          a conveyor belt passing through the entrance  
and exit openings and the heating chamber for transporting  
a substrate on which a thick film has been applied through  
25 the heating chamber;

          a plurality of infrared lamps above and  
below the conveyor belt for heating a substrate on which a  
thick film has been applied carried by the conveyor belt;

          at least two spaced-apart electrodes in the  
30 heating chamber at a location adjacent the entrance opening  
for generating an electric field around a substrate on  
which a thick film has been applied carried by the conveyor  
belt.

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1        26. An apparatus as claimed in claim 25, where the  
supporting means comprises a conductive platform and wherein  
the conductive platform forms one of the electrodes.

5        27. An apparatus as claimed in claim 25, wherein the  
conveyor belt is metal and forms one of the electrodes.

10

15

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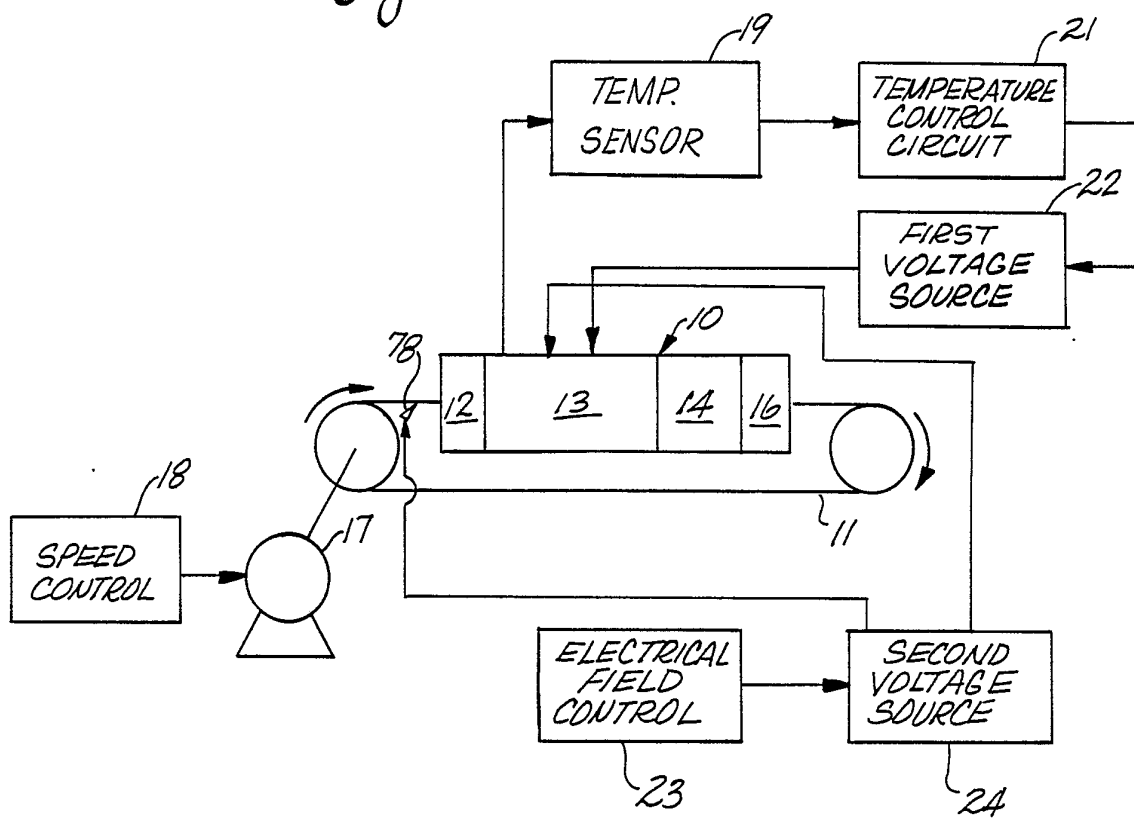
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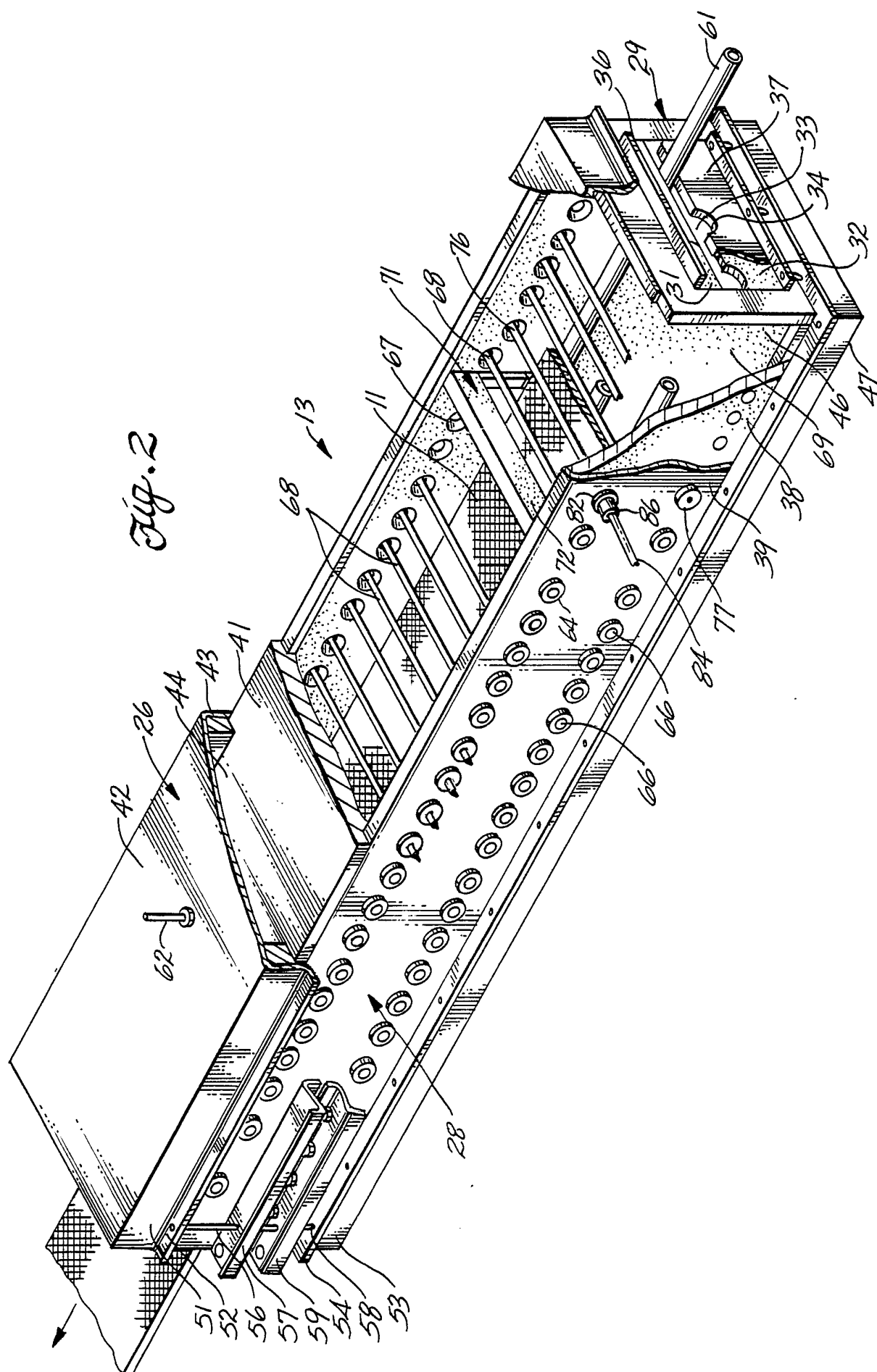
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1/4

Fig. 1





**SUBSTITUTE SHEET**

Fig. 3

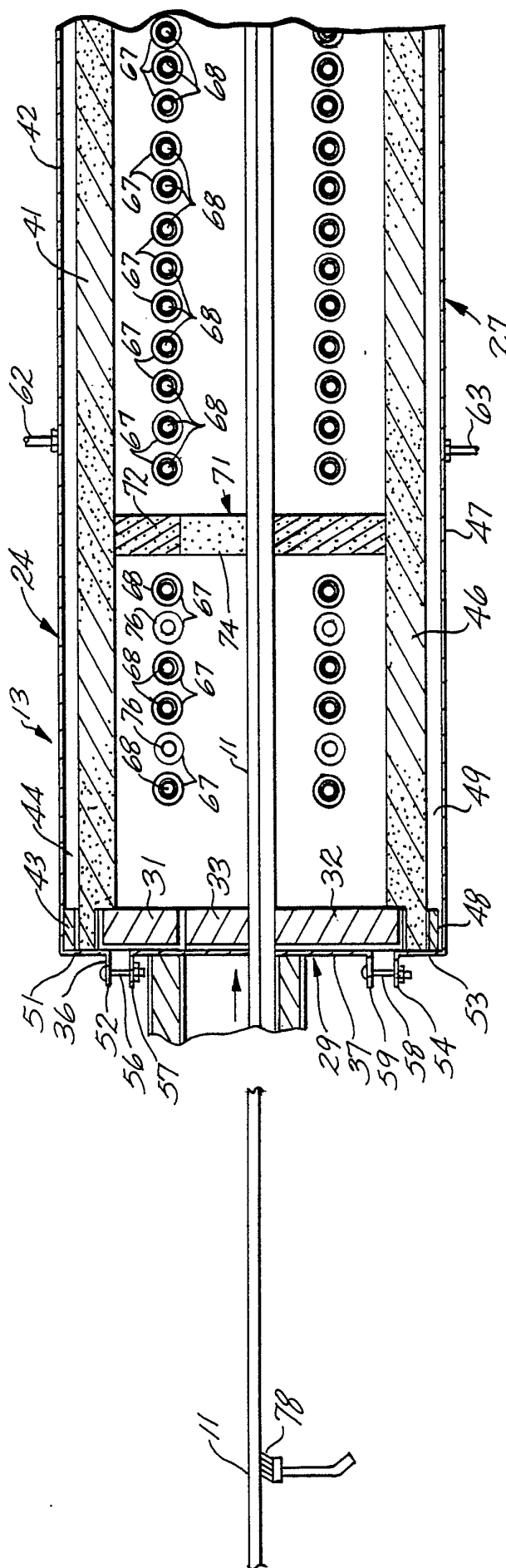




Fig. 4

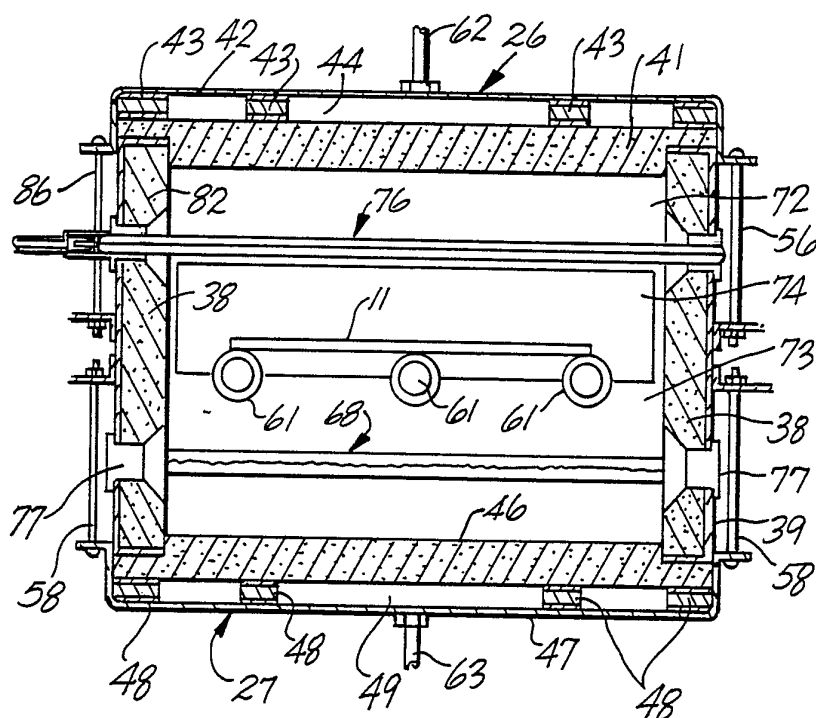
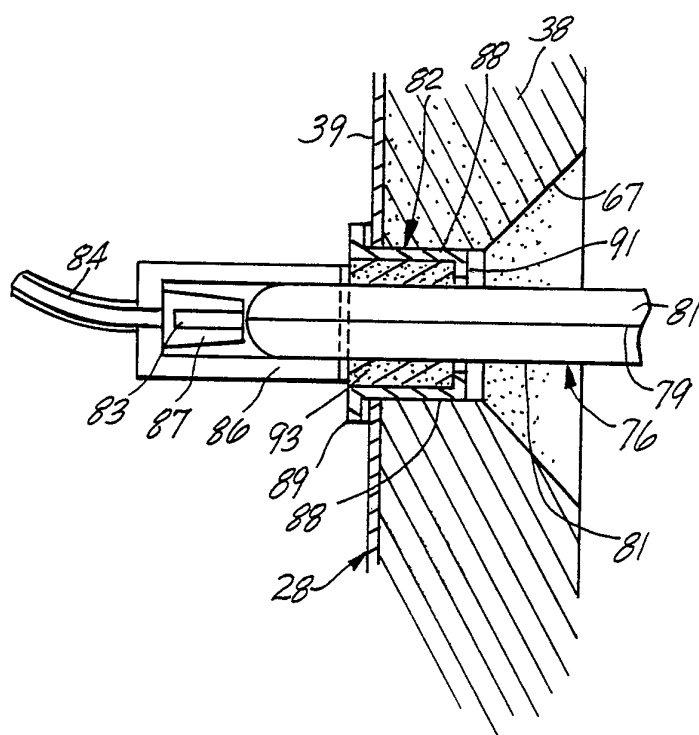


Fig. 5



# INTERNATIONAL SEARCH REPORT

International Application No PCT/US86/02500

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (if several classification symbols apply, indicate all) <sup>3</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC IPC(4): F27B 9/16 U.S. CL. 219/388		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>4</sup>		
Classification System	Classification Symbols	
U.S.	219/350, 354, 388 427/12, 13, 226, 228, 372.2, 384 204/180.1, 164, 210 264/330, 345	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>5</sup>		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT</b> <sup>14</sup>		
Category *	Citation of Document, <sup>16</sup> with indication, where appropriate, of the relevant passages <sup>17</sup>	Relevant to Claim No. <sup>18</sup>
Y	U.S., A, 3,725,114, (Warneke), 3 April 1973 See entire document	1-7, 10-15, 17
Y	U.S., A, 4,155,735, (Ernsberger), 22 May 1979 See entire document	1-5, 10-15, 17-23
Y	U.S., A, 3,931,446 (Murayama et al.), 6 Jan. 1976 See entire document	1-5, 10-15, 17-23
A, P	U.S., A, 4,578,313 (Ito et al.), 25 March 1986 See entire document	1
Y	U.S., A, 4,435,637, (de Vries), 6 March 1984 See entire document.	19-27
Y	U.S., A, 4,404,754, (Candor), 20 Sept. 1983 See entire document.	19-27
<div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p><sup>15</sup> * Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p> </div> </div>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search <sup>2</sup>		Date of Mailing of this International Search Report <sup>3</sup>
2 February 1986		18 FEB 1987
International Searching Authority <sup>1</sup>		Signature of Authorized Officer <sup>20</sup>
ISA/US		M.H. Paschall 