



Office de la Propriété
Intellectuelle
du Canada

Un organisme
d'Industrie Canada

Canadian
Intellectual Property
Office

An agency of
Industry Canada

CA 2028782 C 2003/10/14

(11)(21) **2 028 782**

(12) **BREVET CANADIEN
CANADIAN PATENT**

(13) **C**

(22) Date de dépôt/Filing Date: 1990/10/29

(41) Mise à la disp. pub./Open to Public Insp.: 1991/05/08

(45) Date de délivrance/Issue Date: 2003/10/14

(30) Priorité/Priority: 1989/11/07 (432,707) US

(51) Cl.Int.⁵/Int.Cl.⁵ C07C 227/22, C07C 229/30

(72) Inventeurs/Inventors:
GORALSKI, CHRISTIAN T., US;
HOOPS, JOHN F., US;
RAMANARAYANAN, KUTTANCHERY A., US

(73) Propriétaire/Owner:
MERRELL PHARMACEUTICALS INC., US

(74) Agent: OSLER, HOSKIN & HARCOURT LLP

(54) Titre : METHODE DE PRODUCTION DE VINYLE-GABA

(54) Title: PROCESS FOR THE PRODUCTION OF VINYL-GABA

(57) Abrégé/Abstract:

The present invention is directed to a method for producing vinyl GABA by submitting 5-vinylpyrrolidin-2-one to a basic hydrolysis.



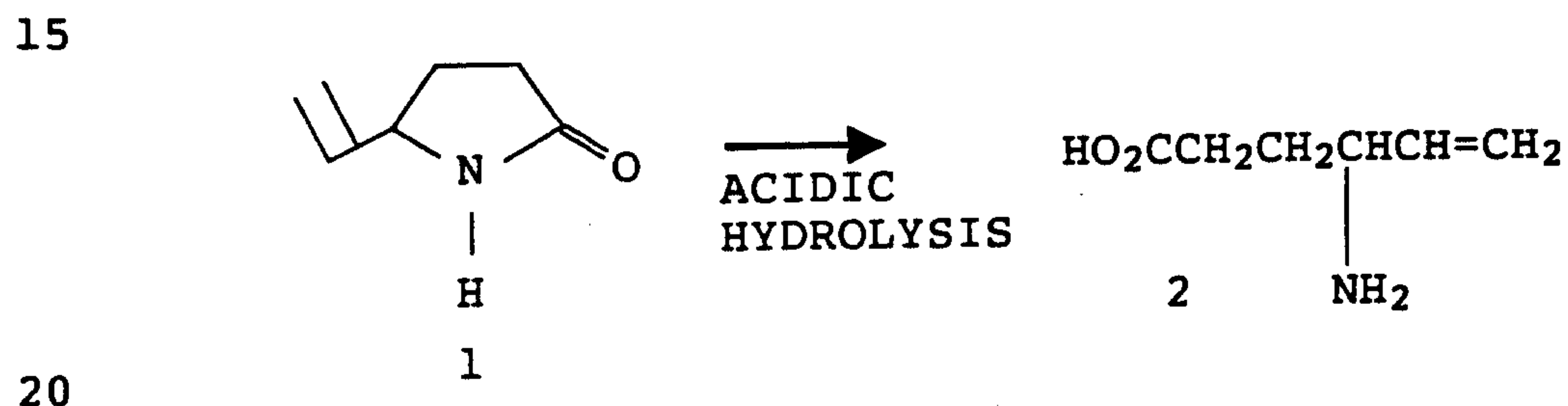
ABSTRACT OF THE DISCLOSURE

The present invention is directed to a method for producing vinyl GABA by submitting 5-vinylpyrrolidin-2-one to a basic hydrolysis.

PROCESS FOR THE PRODUCTION OF VINYL-GABA

The present invention is directed to a method for the
 5 production of 4-amino-5-hexenoic acid. 4-Amino-5-hexenoic
 acid is known in the art as an anti-epileptic agent and is
 described in United States Patent No. 3,960,927. It is also
 known as vinyl-GABA and is currently available from Merrell
 Dow Pharmaceuticals, Inc.

10 United States Patent No. 4,621,145 describes
 one method for synthesizing this compound. The
 last step in the reaction sequence is depicted
 below:



In this reaction 5-vinyl-2-pyrrolidinone (structure 1)
 is subjected to an acidic hydrolysis thereby producing the
 desired compound, 4-amino-5-hexenoic acid (structure 2).
 25 This acidic hydrolysis is carried out using techniques known
 in the art. Typically, the 5-vinyl-2-pyrrolidinone is
 contacted with a strong acid such as hydrochloric acid or
 trifluoroacetic acid at a temperature above 60°C in an
 aqueous solvent system. The 4-amino-5-hexenoic acid is

recovered by concentration as is known in the art. The acidic hydrolysis followed by recrystallization produces a yield of 4-amino-5-hexenoic acid of 57-67%. The spent
5 recrystallization liquor can be saved, concentrated, redissolved in ethanol and a second recrystallization can be carried out on this residue which increases the yield of final product by another 10-11%. The yield of 4-amino-5-hexenoic acid which is produced by this process is from 57-
10 67% with one recrystallization and from 70-79% when a second recrystallization is carried out.

It has been discovered that when 5-vinyl-2-pyrrolidinone is subjected to a basic hydrolysis, the yield of purified 4-
15 amino-5-hexenoic acid that is obtained ranges from 80-87% with only one recrystallization being required. Thus the yield of vinyl-GABA is increased by a factor of from 19-52% compared to the acidic hydrolysis in which only one recrystallization is utilized.

20

This basic hydrolysis can be carried out using techniques known in the art. The 5-vinyl-2-pyrrolidinone is contacted with a molar excess of potassium hydroxide. Typically from about 1.1 to about 1.5 equivalents are
25 utilized. The basic hydrolysis is carried out at a temperature range of from about 60°C to 140°C, and more preferably from about 75°C to 130°C. The reaction is typically carried out for a period of time ranging from about 0.5 hours to about 24 hours.

30

The basic hydrolysis can either be carried out in water or in a mixed solvent system containing water and a lower alkanol such as isopropanol or ethanol. If a mixed solvent system is used, the relative proportions of the lower
35 alkanol and water can vary widely and are not critical to the hydrolysis. It is preferred that the water be present in a minimum quantity of at least 1.5 v/v% in order to expediate the rate of hydrolysis. Carrying out the reaction

in water alone allows the hydrolysis to be completed in a shorter period of time.

5 The resulting 4-amino-5-hexenoic acid can be recovered from the reaction medium using techniques known in the art. Although the relative proportion of lower alkanol to water is not critical to the hydrolysis reaction, it is necessary for a sufficient quantity of the lower alkanol to be present
10 in the reaction zone prior to commencing the recovery of the desired product. The reaction medium should contain from about 60 v/v% to about 90 v/v% of the lower alkanol and more preferably about 85 v/v% of the lower alkanol. If necessary, this quantity of lower alkanol can be added to
15 the reaction medium after the hydrolysis is completed. Once a sufficient quantity of the lower alkanol is present in the reaction medium, the 4-amino-5-hexenoic acid can be recovered by adding approximately one equivalent of an acid such as glacial acetic acid or propionic acid to the
20 reaction. The 4-amino-5-hexenoic acid will be converted to its free base and will precipitate from solution. The precipitate is then recovered by filtration, washed and dried as is known in the art. The 4-amino-5-hexenoic acid can then be purified by recrystallization from a solvent
25 system such as isopropanol/ water.

The following Examples are being presented in order to exemplify the invention but they should not be construed as limiting the scope of the invention in any manner.

30

EXAMPLE I

Basic Hydrolysis of 5-Vinyl-2-pyrrolidinone with potassium Hydroxide in Isopropanol/Water.

35

A 2-L, three-neck flask equipped with a magnetic stirrer, a thermowell, and a nitrogen bubbler was charged with 85.0 g (0.765 mol) of 5-vinyl-2-pyrrolidinone, 75 mL of deionized water, and 900 mL of isopropanol. To this was

added 75.0 g (1.136 mol) of 85% potassium hydroxide, and the mixture heated to reflux. After 24 hours, the reflux was stopped and the reaction cooled to ambient temperature. The
5 reaction was treated with 69.0 g (1.149 mol) of glacial acetic acid. The resulting solution was seeded with vinyl GABA and a heavy slurry of white solid formed. The slurry was cooled to ambient temperature and then stirred with ice-bath cooling for 2 hours. The solid was separated by
10 filtration, washed with two 100 mL portions of ice-cold isopropanol, air dried, and vacuum dried at 35-40°C to give 92.22 g (93% yield) of crude vinyl GABA as a white, crystalline solid.

15 An 84.30 g sample of this crude vinyl GABA was mixed with 120 mL of deionized water and the slurry heated until a cloudy solution formed. The solution was filtered through a medium (M) glass frit and washed through with 23 mL of boiling water followed by 20 mL of boiling water. The
20 filtrate was heated to near reflux and 756 mL of isopropanol was slowly added. A slurry of white solid formed. The slurry was cooled to ambient temperature and then stirred with ice-bath cooling for 2 hours. The solid was separated by filtration, washed with two 50 mL portions of ice-cold
25 isopropanol, air dried, and vacuum dried at 35-40°C to give 78.68 g of vinyl GABA as a white, crystalline solid, mp 178.5-179°C. The final yield was 87%.

EXAMPLE II

30

Basic Hydrolysis of 5-Vinyl-2-pyrrolidinone with Potassium Hydroxide in Isopropanol/Water. Preparation of Vinyl GABA.

35

A 2-L, three-neck flask equipped with a magnetic stirrer, a thermowell, two glass stoppers, and a reflux condenser fitted with a nitrogen bubbler was charged with 85.0 g (0.765 mol) of 5-vinyl-2-pyrrolidinone, 75 mL of deionized water, and 900 mL of isopropanol. To this

mixture, 75.0 g (1.136 mol) of 85% potassium hydroxide was added and the mixture heated to reflux. After 24 hours, the reaction was cooled to approximately 50°C and 69.0 g (1.149
5 mol) of glacial acetic acid was added. The slurry of white solid which formed was cooled to ambient temperature and then stirred with ice-bath cooling for 2 hours. The solid was separated by filtration and washed with two 100 mL portions of ice-cold isopropanol. The solid was air dried
10 and then vacuum dried to give 88.19 g (89% yield) of crude vinyl GABA as a white, crystalline solid.

An 83.10 g sample of this vinyl GABA was mixed with 120 mL of deionized water and the mixture heated to reflux to
15 give a cloudy solution. The solution was filtered through filter paper into a 2-L, single-neck flask equipped with a magnetic stirrer. The beaker and the filter were rinsed with two 25 mL portions of hot water and the filtrate heated to reflux. To the hot filtrate was added 750 mL of
20 isopropanol. The solution was cooled to ambient temperature to give a slurry of white solid. The slurry was stirred for 2 hours with ice-bath cooling. The solid was separated by filtration, washed with two 100 mL portions of ice-cold isopropanol, air dried, and vacuum dried at 40°C to give
25 75.08 g of vinyl GABA as a white, crystalline solid, mp 179-180°C. The final yield was 81%.

EXAMPLE III

30 Basic Hydrolysis of 5-Vinyl-2-pyrrolidinone with Potassium Hydroxide in Ethanol/Water. Preparation of Vinyl GABA.

A 500-mL, single-neck flask equipped with a magnetic
35 stirrer and a reflux condenser fitted with a nitrogen bubbler was charged with 27.07 g (0.244 mol) of 5-vinyl-2-pyrrolidinone, 25 mL of deionized water, 300 mL of ethanol, and 24.10 g (0.366 mol) of 85% potassium hydroxide. The mixture was heated to reflux and held there for 24 hours.

The reaction was cooled to ambient temperature and 22.17 g (0.369 mol) of glacial acetic acid was added. A white solid rapidly began to crystallize, and the slurry was stirred at room temperature over night. The slurry was cooled with an ice bath. The solid was separated by filtration, washed with two 50 mL portions of cold ethanol, air dried, and vacuum dried at 35°C to give 24.73 g (78% yield) of crude vinyl GABA as a white, crystalline solid.

10

EXAMPLE IV

Hydrolysis of 5-Vinyl-2-pyrrolidinone with Potassium Hydroxide in Water.

15

A 500-ml, single-neck flask equipped with a magnetic stirrer and a reflux condenser fitted with a nitrogen bubbler was charged with 85.0 g (0.765 mol) of 5-vinyl-2-pyrrolidinone, 75 mL of deionized water, and 53.0 g (0.874 mol) of 87% potassium hydroxide. The resulting mixture was heated to reflux and held there for 1.0 h. The reaction mixture was cooled to ambient temperature and homogeneous orange solution resulted. The solution was diluted with 700 ml of isopropanol and 52.50 g (0.874 mol) of glacial acetic acid was added. A slurry of white solid formed. The slurry was cooled to 0-5°C and stirred at that temperature for 2 h. The solid was separated by filtration, washed with two 50-mL portions of cold isopropanol, air dried, and vacuum dried at 55°C to give 89.72 g (90.8% yield) of vinyl GABA as a white solid. The vinyl GABA was mixed with 130 mL of deionized water and the mixture heated to reflux to give a hazy solution. The solution was filtered through a coarse (C) glass frit. The frit was washed with 20 mL of refluxing deionized water. The filtrate was diluted with 750 ml of isopropanol and a slurry of white solid formed. The slurry was cooled to 0-5°C and stirred at that temperature for 2 hours. The solid was separated by filtration, washed with two 50-ml portions of cold isopropanol, air dried, and

20

25

30

35

vacuum dried at 55°C to give 85.5 g of vinyl GABA as a white, crystalline solid, mp 179-180°C.

5

EXAMPLE V (COMPARATIVE EXAMPLE)

Acidic Hydrolysis of 5-Vinyl-2-pyrrolidinone to Vinyl GABA.

10 A 1-L, three-neck flask equipped with a mechanical stirrer, a thermometer, and a reflux condenser fitted with a nitrogen bubbler was charged with 50.00 g (0.45 mol) of 5-vinyl-2-pyrrolidinone, 300 ml of deionized water, and 50 mL of concentrated hydrochloric acid. The reaction mixture was
15 heated to 95°C during 30 min. and stirred at that temperature for 5 h. The reaction mixture was concentrated *in vacuo* (85°C/10-15 mm) to afford 83.8 g of red-orange oil. This material was transferred, along with 500 mL of ethanol, to a 1-L, three-neck flask equipped with a mechanical
20 stirrer. To the rapidly stirred solution, 70 mL (50.82 g, 0.50 mol) of triethylamine was added in portions over 30 min. The reaction temperature was between 32°C and 34°C and the final pH was approximately 7 to 8. The precipitated product was collected by filtration, washed with two 50-mL
25 portions of ethanol, and air dried to give 51.28 g of crude vinyl GABA as a white-pink solid. The crude vinyl GABA was dissolved in 70 mL of deionized water and 2.0 g of Darco activated carbon was added. The resulting mixture was heated at 90°C for 30 min. with occasional swirling. The
30 mixture was filtered hot through a Celite pad. The filtrate was transferred to a 1-L flask and diluted with 420 mL of ethanol. The resulting solution was cooled with an ice bath and stirred for 3 h. The solid was collected by filtration, washed with two 50mL portions of ethanol, air dried, and
35 vacuum dried for 24 h at 36-38°C and 60-70 mm to give 32.00 g (57% yield) of vinyl GABA as a white solid, mp 174-174.5°C.

The embodiments of the invention in which an exclusive property or privilege is claimed are defined as follows:

1. A process for producing 4-amino-5-hexenoic acid
5 which comprises subjecting 5-vinyl-2-pyrrolidinone to a basic hydrolysis wherein said hydrolysis is carried out with potassium hydroxide.

2. A process for producing 4-amino-5-hexenoic acid
10 comprising:

a) subjecting 5-vinyl-2-pyrrolidinone to a basic hydrolysis with potassium hydroxide;

b) precipitating the resulting 4-amino-5-hexenoic acid from the reaction mixture, by treating the reaction
15 mixture with a lower alkanol and either acetic acid or propanionic acid, and;

c) recovering the precipitated 4-amino-5-hexenoic acid.

20 3. The process according to claim 1 or 2 wherein the hydrolysis is carried out in water.

4. The process according to claim 1 or 2 wherein the hydrolysis is carried out in a mixed solvent system
25 containing water and a lower alkanol.

5. The process according to claim 4 wherein said lower alkanol is ethanol.

30 6. The process according to claim 4 wherein said lower alkanol is isopropanol.

7. The process according to claim 1, 2, 3 or 4 wherein said potassium hydroxide is present in an amount of from about 1.1 to about 1.5 equivalents.

5 8. The process according to claim 4 wherein said mixed solvent system contains at least about 1.5 v/v% of water and a lower alkanol.

9. The process according to claim 8 wherein said
10 lower alkanol is ethanol.

10. The process according to claim 8 wherein said lower alkanol is isopropanol.

15 11. The process according to claim 1, 2, 3 or 4 wherein said hydrolysis is carried out at a temperature of from about 60°C to about 140°C.

12. The process according to claim 1, 2, 3 or 4
20 wherein said hydrolysis is carried out at a temperature of from about 75°C to about 130°C.

13. The process according to claim 1, 2, 3 or 4 wherein said hydrolysis is carried out for a period of time
25 in the range of from about 0.5 hour to about 24 hours.

14. The process according to claim 1, 2, 3 or 4 wherein the hydrolysis is carried out for about 24 hours.

30 15. The process according to claim 4 wherein the mixed solvent system contains at least 7.7 v/v% of water and isopropanol.

16. The process according to claim 1, 2, 3 or 4 wherein the hydrolysis is carried out at a temperature of from about 75°C to about 130°C for about 24 hours.

5 17. The process according to claim 1, 3 or 4 wherein said resulting 4-amino-5-hexenoic acid is recovered.

18. The process according to claim 2 or claim 17 wherein the recovered 4-amino-5-hexenoic acid is purified.
10

19. The process according to claim 2 in which the order of addition of the lower alkanol is modified by adding said lower alkanol at the initiation of or during said basic hydrolysis reaction and not at the initiation of
15 or during said precipitation.

20. A process according to claim 19 wherein said potassium hydroxide is present in the amount of from 1.1 to 1.5 equivalents.
20

21. A process according to claim 19 wherein said hydrolysis is carried out in a solvent containing a lower alkanol and at least 1.5 v/v % of water.

25 22. A process according to claim 19 wherein said lower alkanol is isopropanol.

23. A process according to claim 19 in which the hydrolysis is carried out at a temperature range of from
30 75° to 130°C for a period of time ranging from 0.5-24 hours.

24. A process according to claim 19 or 23 wherein said resulting 4-amino-5-hexenoic acid is recovered and purified.

5