

CORRECTED VERSION

(19) World Intellectual Property Organization
International Bureau(43) International Publication Date
7 May 2009 (07.05.2009)

PCT

(10) International Publication Number
WO 2009/059286 A9(51) International Patent Classification:
B01D 59/44 (2006.01)(21) International Application Number:
PCT/US2008/082229(22) International Filing Date:
3 November 2008 (03.11.2008)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
61/001,597 2 November 2007 (02.11.2007) US
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02205 (US).(81) Designated States (unless otherwise indicated, for every
kind of national protection available): AE, AG, AL, AM,
AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ,
CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ,
EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN,
HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR,
KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME,
MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO,
NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG,
SK, SL, SM, ST, SV, SY, TJ, TM, TN, TR, TT, TZ, UA,
UG, US, UZ, VC, VN, ZA, ZM, ZW.(84) Designated States (unless otherwise indicated, for every
kind of regional protection available): ARIPO (BW, GH,

[Continued on next page]

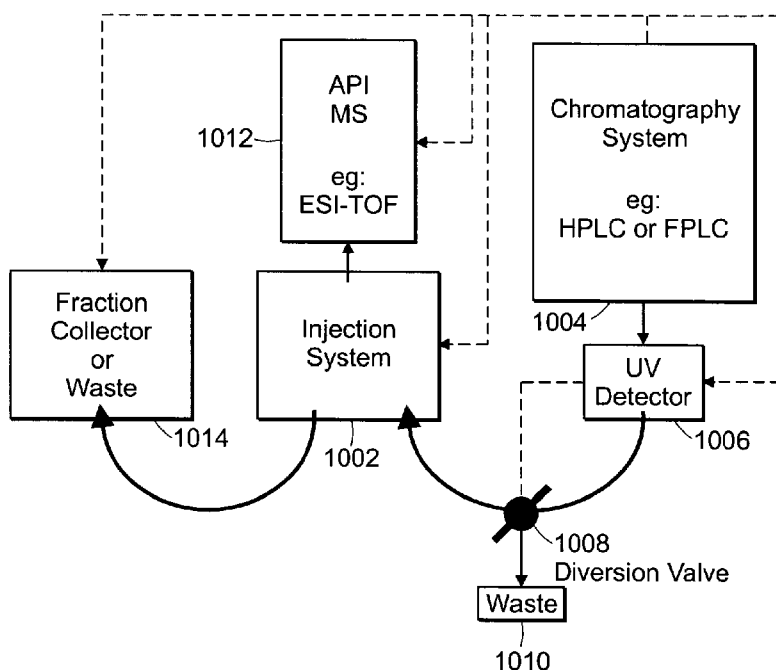
(54) Title: DEVICES AND METHODS FOR COUPLING MASS SPECTROMETRY DEVICES WITH CHROMATOGRAPHY
SYSTEMS

FIG. 10

(57) Abstract: One embodiment of the invention provides a method of preparing an eluted sample containing salts or buffers from a liquid chromatography device for analysis by a mass spectrometry device. The method includes: continuously providing a non-polar solvent to the mass spectrometry device; receiving the eluted sample from the liquid chromatography device; flowing the eluted sample over a solid phase extraction column; flowing the non-polar solvent over the solid phase extraction column; and presenting non-polar solvent and the eluted sample to the mass spectrometry device.



GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MT, NL, NO, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— *with international search report (Art. 21(3))*

(48) Date of publication of this corrected version:

19 November 2009

(15) Information about Correction:

see Notice of 19 November 2009

DEVICES AND METHODS FOR COUPLING MASS SPECTROMETRY DEVICES WITH CHROMATOGRAPHY SYSTEMS

CROSS REFERENCE TO RELATED APPLICATIONS

5 This application claims priority to U.S. Provisional Patent Application
Serial No. 61/001,595, filed November 2, 2007 and U.S. Provisional Patent
Application Serial No. 61/001,597, filed November 2, 2007. Each of these
patent applications is hereby incorporated by reference in its entirety.

10 TECHNICAL FIELD

The present invention relates generally to high throughput screening of fluidic samples, and more particularly, to automated systems and methods for increasing sample throughput of fluidic samples.

15 BACKGROUND ART

In many applications, such as drug discovery and development, environmental testing, and diagnostics, there is a need to analyze a large number of samples in an efficient and reproducible manner. Many of the techniques used to analyze fluidic samples require that the samples be tested in a serial manner. In such applications, the process of serial analysis can be automated through the use of a computer controlled robotics and automation. Such devices are generally called auto-injectors and are commonly interfaced to all manner of serial analysis systems including, but not limited to, chromatography systems, mass spectrometers, and spectroscopic detectors.

25 Typical auto-injectors include a plurality of sample reservoirs, a syringe or syringe-like sample transport system, and an injection valve along with the automation and computer control systems. Auto-injectors commonly mimic cumbersome manual injection methods in which a metered aliquot of a sample is aspirated from a desired sample reservoir into a transfer syringe.

30 The aspiration process is often controlled by pulling back on a plunger or piston to create a negative pressure resulting in aspiration of the sample. The transfer syringe is then moved to and docked with a stationary injection valve. The sample aliquot is then transferred from the syringe to the injection valve

by depressing the transfer syringe plunger or activating the piston. The sample fills an injection loop within the injection valve. Upon actuation of the valve the sample is introduced into the fluidic circuit and diverted to the analysis system.

- 5 The transfer syringe and the injection valve ports are then rinsed with an appropriate buffer or solvent to remove traces of the analyte to minimize contamination between samples. Contamination of the fluidic system with a sample can cause a significant barrier to the successful operation of a serial analysis system resulting in carryover and compromised data. After an
- 10 appropriate cleaning protocol the entire process is repeated for the next sample. Various embodiments of this general approach to auto-injectors are available commercially. Sample reservoirs used in auto-injectors range from glass vials to 96 or 384-well microtiter plates. Sample reservoirs may be sealed with a plastic film or metal foil, or a septum. Some auto-injection
- 15 devices use conventional syringes of various sizes attached to a robotic arm. Other devices use a tube attached to a small piston. The sample is aspirated into this tube and transferred to the injection valve. Some versions of auto-injectors attempt to increase throughput by using multiple syringes such that while an injection is being made by one syringe others are being washed.
- 20 One auto-injector increases throughput with a simultaneous aspiration of eight samples. These samples are then loaded into the sample injection loops of eight separate injection valves. The samples are then sequentially diverted from each of the eight injection valves into the analysis system. Throughput is thus increased through the parallelism of the process, however at increased
- 25 cost and complexity.

- Mass spectrometry (MS) with atmospheric pressure ionization (API) is a commonly used technique for the analysis of complex mixtures. Variations of API-MS include electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI) and atmospheric pressure photoionization (APPI).
- 30 API-MS is used routinely in the pharmaceutical industry, environmental and forensic analysis, materials science, and in scientific applications. Both quantitative and qualitative information about specific compounds in complex mixtures can be obtained with the use of API-MS methods.

However, API-MS has several drawbacks. Traditionally, MS is a serial process in which samples are analyzed sequentially unlike parallel analysis schemes typically employed in many optical analysis systems. Sequential analysis can be impractical and in many cases economically unviable if very large numbers of samples are to be analyzed.

Furthermore, many compounds typically found at high concentrations in complex biological, chemical, or environmental samples, such as salts, buffers, ionic or non-ionic detergents, proteins or enzymes, and other cofactors can cause a significant reduction in the amount of target signal observed in mass spectrometry. Interference from high concentrations of non-volatile components are particularly troublesome because in addition to causing signal suppression non-volatile compounds tend to build up in the source region of the MS and gradually result in a decline in instrument performance.

The inherent expense involved in purchasing and operating mass spectrometers makes it highly desirable to improve productivity by devising methods and devices for increasing the analysis throughput (*i.e.* the number of samples that can be analyzed in a given time). Any method and device that attempts to increase throughput in API-MS must address several key issues such as: (1) a rapid system for delivery of a sample to the mass spectrometer must be designed; (2) the components of complex mixtures that cause suppression of the target signal must be isolated and removed from the analytes of interest; (3) the non-volatile components of complex mixtures that build up in the MS source and result in a decay of instrument performance over time must be isolated and removed; and (4) each sample must be cleaned from the analysis system to an acceptable level before the next sample is analyzed to prevent sample-to-sample carryover that will result in contamination of the data.

Liquid chromatography (LC) can be used to remove the salts, buffers, and other components from complex mixtures that may cause suppression of the MS signal of interest or result in degradation of MS instrument performance. Conventional liquid chromatography (LC) and its variations, such as high performance liquid chromatography (HPLC), typically involve

flowing a liquid sample over a solid, insoluble matrix (generally referred to as Solid Phase Extraction (SPE)) commonly packed in a column format. The liquid sample includes an analyte(s) of interest that has an affinity for the matrix under certain conditions of pH, salt concentration, or solvent composition. Affinity of the analyte(s) of interest to the matrix may be due to hydrophobic or hydrophilic interactions, ionic interactions, molecular size, or coordination chemistry. In a highly specific variation, antibodies immobilized to the matrix are used to selectively capture molecules containing a highly specific epitope from complex mixtures.

As a result of the analyte(s) affinity to the matrix, the analyte(s) binds to the matrix and becomes immobilized while other (undesired) components of the liquid sample flow through the matrix and are removed. The analyte(s) of interest are then eluted away from the matrix by changing the conditions of the flowing liquid, such that the analyte of interest no longer has affinity for the matrix. For example, changes in pH, ionic strength, solvent composition, temperature, and/or other physicochemical parameters may weaken the affinity of the analyte(s) for the matrix.

However, the traditional use of liquid chromatography in high-throughput mass spectrometry has limitations. Very often, the throughput of a serial analysis is limited by the time it takes to collect the signal from an individual sample. In liquid chromatography applications, the matrix output signal from an analyte of interest is in the form of a peak, and the width of this peak in time is the ultimate determinant of the maximum throughput. A key factor in increasing mass spectrometry throughput is the elution of the samples of interest from the insoluble matrix as a tight, sharp band that is presented to the mass spectrometer in the shortest amount of time. For example, to achieve an overall throughput greater than 30 seconds per sample, with baseline resolution of each sample, the peak width must be narrower than 30 seconds. As throughput is increased, more stringent requirements on the peak width must be imposed. If the throughput begins to approach the peak width, the sequential samples begin to overlap, baseline resolution between samples in the MS is lost, and accurate quantification for each sample is no longer possible.

In traditional LC, the analyte(s) of interest that are bound to the insoluble solid matrix (typically packed in a column format) are eluted away from the matrix by changing various properties of the liquid flowing over the matrix such that the analyte(s) are no longer immobilized on the column.

5 However, as the analyte(s) flow through the length of the matrix a phenomenon known as band broadening occurs, in which linear diffusion causes the volume which contains the focused analyte(s) to expand. Consequently, the concentration of the analyte of interest presented to the mass spectrometer (or other analyzer) is decreased, and a broad peak is
10 produced that makes High Throughput Screening (HTS) problematic.

SUMMARY OF THE INVENTION

The current invention describes a system and method for increasing the throughput of analysis of selected components in complex biological,
15 chemical, or environmental matrices with the use of, for example, chromatography and/or mass spectrometry. In various embodiments, throughput rates ranging from 30 seconds per sample to 1 second per sample or faster are achievable, depending on the specific application. Further embodiments of the invention include an auto-injection system that increases
20 throughput and minimizes sample carryover.

In accordance with one aspect of the invention, there is provided a system for high throughput sample preparation and analysis. The system includes a chromatography column including an insoluble matrix. A fluidic circuit is capable of passing a fluid over the insoluble matrix in a first direction
25 such that an analyte in the fluid binds to the insoluble matrix, and back-eluting an elution fluid over the insoluble matrix in a second direction opposite the first direction to output a sample that includes the analyte. A controller controls the fluidic circuit to periodically perform the steps of passing the fluid over the insoluble matrix and back-eluting the elution fluid over the insoluble
30 matrix to output a plurality of samples at a periodic rate.

In accordance with related embodiments of the invention, the periodic rate is 30 seconds/sample or faster. The fluidic circuit may include a valving module capable of alternately directing fluid over the insoluble matrix in the

first direction and back-eluting an elution fluid over the insoluble matrix in the second direction. The valving module may include at least one pneumatically actuated valve and/or have an actuation time of faster than 100 milliseconds.

5 In accordance with further related embodiments of the invention, the system may further include an analyzer for analyzing one or more of the samples. The analyzer may be, for example, an optical analyzer or a mass spectrometer that outputs a signal representative of the one or more samples. The fluidic circuit may include a valve module that is actuated to back-elute the elution fluid over the insoluble matrix, and wherein the controller integrates
10 the signal for a predetermined time after the valve module is actuated to determine a characteristic of the sample. The mass spectrometer may include, without limitation, an electrospray ionization source, an atmospheric pressure chemical ionization source, or an atmospheric pressure photoionization source.

15 In accordance with still further related embodiments of the invention, the fluidic circuit may include tubing having a diameter between 20 μm to 300 μm . The fluidic circuit may include one or more surfaces which contact the fluid, wherein each surface is bioinert, such that it is non-reactive. Each surface may include, for example, poly ether ketone, polyimide, titanium, and/or titanium alloy. The fluidic circuit may include a fluidic pathway made of
20 steel coated with a material to minimize binding with the analyte, such as polytetrafluoroethylene and/or polyethylene glycol. The fluidic circuit may include an aspirator for aspirating an aliquot of the fluid to be passed over the insoluble matrix. The chromatography column may include a first end and a
25 second end, wherein the analyte enters and exits the chromatography column at the first end.

In accordance with another aspect of the invention, a method of high throughput sample preparation and analysis includes passing a fluid over an insoluble matrix in a first direction, the fluid including an analyte that binds to
30 the insoluble matrix. An elution fluid is back-eluted over the insoluble matrix in a second direction opposite the first direction to output a sample that includes the analyte. The steps of passing the fluid and back-eluting the elution fluid are repeated so as to output a plurality of samples at a periodic

rate.

In accordance with related embodiments of the invention, the periodic rate is 30 seconds/sample or faster. Analyzing each sample may include presenting each sample to a mass spectrometer. Back-eluting may include
5 actuating a valving element to initiate flow of the elution fluid over the insoluble matrix, wherein the method further includes integrating an output of the mass spectrometer for a predetermined time after the valve is actuated to determine a characteristic of the sample. Wash solution may be passed over the chromatography matrix prior to passing the fluid over the insoluble matrix,
10 or back eluting the elution fluid. The fluid may be aspirated from a fluid source prior to passing the fluid over an insoluble matrix. The chromatography matrix may be packaged in a column format.

In accordance with still another aspect of the invention, a system for high throughput sample preparation and analysis includes a plurality of
15 chromatography columns and a mass spectrometer. A valve is capable of selectively presenting effluent from one of the plurality of chromatography columns to the mass spectrometer.

In accordance with related embodiments of the invention, the valve may be actuated to present effluent from one of the plurality of
20 chromatography columns to the mass spectrometer. A processor may receive an output signal from the mass spectrometer, and integrate the output for a predetermined time after the valve is actuated to determine a characteristic of the sample.

In still other aspects of the invention, a computer program product is
25 presented for use on a computer system for controlling a high throughput system having a fluidic circuit in fluid communication with a chromatography column. The computer program product includes a computer usable medium having computer readable program code thereon. The computer readable program code includes program code for controlling the fluidic circuit to pass a
30 fluid over the insoluble matrix in a first direction such that an analyte in the fluid binds to the insoluble matrix. The computer readable program code also includes program code for controlling the fluidic circuit to back-elute an elution fluid over the insoluble matrix in a second direction opposite the first direction

to output a sample that includes the analyte; and program code for repeating the passing of the fluid and the back-eluting the elution fluid to output samples at a periodic rate.

In accordance with related embodiments of the invention, the computer
5 readable program code for controlling the fluidic circuit to back-elute the elution fluid includes program code for actuating a valve module that allows the elution fluid to flow through the chromatography column in the second direction. The high throughput system may include a mass spectrometer for analyzing the sample, wherein the computer program product further includes
10 program code for integrating an output of the mass spectrometer upon actuation of the valve module to determine a characteristic of the sample.

In accordance with another embodiment of the invention, an auto-injection system for high throughput screening of fluidic samples includes a sample sipper tube, a sample loop, and an injection valve. The injection valve
15 applies a reduced pressure to the sample sipper tube. When the injection valve is in a first position, the sample loop is in fluid communication with the sample sipper tube.

In related embodiments of the invention, the system may further include a vacuum means for supplying the reduced pressure. The vacuum
20 means may include a vacuum pump for continuous application of reduced pressure and/or a piston for metered application of reduced pressure. A valve may select one of the vacuum pump and the piston pump as a source of the reduced pressure. An inline trap may be positioned between the vacuum means and the injection valve. A cutoff valve, which may be a solenoid valve,
25 may meter an amount of sample fluid to be aspirated into the sample loop via the sample sipper tube, the cutoff valve positioned between the vacuum means and the injection valve. Fluid contacting surfaces of the system may be made of a material from the group of materials consisting of polytetrafluoroethylene (available under the TEFLON® trademark from E. I.
30 Du Pont De Nemours and Company of Wilmington, Delaware), fused silica, and poly ether ketone.

In further related embodiments of the invention, when the injection valve is in a second position, the sample loop is in fluid communication with

an output port of the injection valve. When the injection valve is in the second position, the sample sipper tube may be in fluid communication with a source of the reduced pressure so as to aspirate wash fluid, an inline-trap capturing the wash fluid.

5 In accordance with still another aspect of the invention, an auto-injection system for high throughput screening of fluidic samples includes a source of reduced pressure, a sample loop, a sample sipper tube, and an injection valve. The injection valve includes a first port in fluid communication with the sample sipper tube; a second port in fluid communication with the
10 sample loop; a third port in fluid communication with the sample loop; and a fourth port in fluid communication with the source of reduced pressure.

 In related embodiments of the invention, when the injection valve is in a first position the source of reduced pressure, the sample loop, and the sample sipper tube are in fluid communication. The injection valve may include a fifth
15 port for outputting sample fluid from the sample loop. When the injection valve is in a second position, the sample loop is in fluid communication with the fifth port. The system may include a source of high pressure, and wherein the injection valve further includes a sixth port in fluid communication with the source of high pressure.

20 The source of reduced pressure may include a vacuum pump and/or a piston. A valve may select one of the vacuum pump and the piston pump as a source of the reduced pressure. An inline trap may be positioned between the source of reduced pressure and the injection valve. When the injection valve is in a second position, the sample sipper tube may be in fluid
25 communication with the source of the reduced pressure so as to aspirate wash fluid, the inline-trap capturing the wash fluid. A cutoff valve, such as a solenoid valve, may be used for metering an amount of sample fluid to be aspirated into the sample loop via the sample sipper tube, the cutoff valve positioned between the source of reduced pressure and the injection valve.

30 In accordance with another embodiment of the invention, an autosampler system for repetitive sampling and presentation of samples includes a fluidic circuit. The fluidic circuit includes a sample port in fluid communication with an injection valve. The fluidic circuit further includes

means for applying a reduced pressure to the sample port to load a sample into the fluidic circuit. The sample is presented, via output means, into an analyzer from an output port of the fluidic circuit that is distinct from the sample port. The system further includes automated means for positioning
5 the multiple samples relative to the sample port.

In related embodiments of the invention, the means for applying a reduced pressure may include a trap, and/or may continuously apply a negative pressure to the sample port throughout the presentation of samples. The automated means for positioning multiple samples may include a robotic
10 device for successively presenting wells of microplates. The samples are processed at a rate of greater than one sample every 30 seconds. The analyzer may be a mass spectrometer. The sample may be aspirated intermittently into the sample port, while fluid is continuously injected into the analyzer.

In further related embodiments of the invention, the fluidic circuit may include a resin for purification of the samples. The system may further include means for introduction of a sample to the resin, washing the resin with a wash solution and back-eluting the sample with an elution solution prior to
15 presentation.

In accordance with another embodiment of the invention, a system for high throughput sample preparation and analysis includes a chromatography column including an insoluble matrix. Fluidic circuit means passes a fluid over the insoluble matrix in a first direction such that an analyte in the fluid binds to the insoluble matrix, and passes an elution fluid over the insoluble matrix to
20 output the analyte to an analyzer. A controller controls the fluidic circuit to periodically perform the steps of passing the fluid over the insoluble matrix and passing the elution fluid over the insoluble matrix to output to the analyzer a plurality of samples at a periodic rate, such that the fluidic circuit presents only at least one of the elution fluid and the analyte to the analyzer.

In accordance with another embodiment of the invention, a system for high throughput screening of fluid samples includes a sample aspiration tube, a valving element, sample loop, and an analyzer. A controller controls the valving element to alternatively aspirate a first fluid into the sample loop via
30

the sample aspiration tube, and aspirate a second fluid via the aspiration tube while simultaneously outputting the first fluid in the sample loop to the analyzer.

In another embodiment auto-injection system provides high throughput screening of fluidic samples. A sample injection valve has a first position which applies a reduced pressure to a sample sipper tube for aspirating a fluidic sample into the sample sipper tube, and a second position which delivers the fluidic sample to a sample supply loop. A column control valve has a first position which delivers the fluidic sample from the sample supply loop to a sample chromatography column, and a second position which reverses direction of fluid flow through the sample chromatography column to deliver the fluidic sample to a sample analyzer, *e.g.*, a mass spectrometer. A wash control valve has a first position which supplies a wash buffer solution to the sample chromatography column in a forward fluid flow direction, and a second position which supplies elution solvent to flush the sample supply loop. Positioning means present individual microplate sample wells to the sample sipper tube. Automated control means creates a cycle of repeatedly introducing samples and actuating the sample injection valve, column control valve, and wash control valve.

In a further such embodiment, the first position of the sample injection valve may further deliver wash buffer solution from the wash control valve to the sample chromatography column. The second position of the sample injection valve may further deliver elution solvent from the wash control valve to the sample supply loop. The first position of the column control valve may further supply elution solvent to flush the sample analyzer. The second position of the column control valve may further allow elution solvent to exit from the sample supply loop. The second position of the wash control valve may further deliver wash buffer solution to flush the sample supply loop.

In a further embodiment, a wash buffer supply pump provides wash buffer solution to the wash control valve. A first elution solvent pump may provide elution solvent to the column control valve. And a second elution solvent pump may provide elution solvent to the wash control valve.

Embodiments also include a similar method of performing high-throughput screening of fluidic samples. A sample injection valve, a column control valve, and a wash control valve, each having two operating positions, are provided and arranged to perform a repeating cycle for high throughput screening of fluidic samples. The cycle includes positioning each of the valves in a first operating position in which: (i) a fluidic sample in a microplate sample well is presented to a sample sipper tube and aspirated through the sample sipper tube to the sample injection valve, (ii) elution solvent is supplied by the column control valve to a sample analyzer, and (iii) wash buffer solution is delivered from each of the valves in series to a sample chromatography column for equilibration of the column. The sample injection valve is actuated to a second operating position in which: (i) the sample sipper tube is withdrawn from the fluidic sample and the aspirated sample is delivered by the sample injection valve through a sample supply loop to the column, and (ii) wash buffer solution is delivered from each of the valves in series to the column for purification of the sample. The column control valve and the wash control valve are actuated to respective second operating positions in which: (i) the sample sipper tube aspirates a wash solution for cleaning, (ii) elution solvent is supplied through each of the valves in series to flush the sample supply loop, and (iii) elution solvent is supplied by the column control valve to reverse direction of fluid flow through the column to deliver the fluidic sample to the analyzer. Finally, each of the valves is actuated back to their respective first operating positions to repeat the cycle.

In a further such embodiment, the cycle is performed at a speed of greater than two samples per minute. The sample analyzer may be a mass spectrometer.

In another embodiment a sample injection system includes a vacuum source, a conduit in communication with the vacuum source, a fluid sensor configured to detect the presence of the fluid in the conduit, a sample loop in communication with the conduit; and a sipper in communication with the sample loop.

The above embodiment can include several additional features. The conduit can include a transparent portion. The fluid sensor can be an optical

sensor and configured to detect the presence of fluid in the transparent portion of the conduit. The system can include a trap located between the vacuum source and the conduit. A valve can be coupled to the sipper, the sample loop, and the conduit. The valve can be a multi-port valve. The valve
5 can be pneumatically, electrically, electromechanically, or mechanically actuated. The valve can be configured to interrupt fluid communication between the conduit and the sample loop when fluid is detected by the fluid sensor.

The system can include a robotic system for positioning the sipper to
10 aspirate a fluid sample from a sample reservoir. The robotic system can be configured to lower the sipper into the sample reservoir until fluid is detected by fluid detector. The robotic system can be further configured to prevent the sipper from traveling beyond a defined position. The defined position can be specified by a user. The robotic system can be configured to retract the
15 sipper from the reservoir when fluid is detected by the fluid sensor.

Another embodiment is directed to an auto-injection system for high throughput screening of fluidic samples. The system includes a vacuum source, a sample injection valve, a conduit connecting the vacuum source and the sample injection valve, a fluid sensor configured to detect the presence of
20 the fluid in the conduit, a column control valve configured to facilitate a continuous flow of an elution solvent to a sample analyzer, a wash control valve, and automated control means for creating a cycle of repeatedly introducing samples and actuating the sample injection valve, column control valve, and wash control valve.

25 The sample injection valve has a first position which applies a reduced pressure to a sipper for aspirating a fluidic sample into the sipper, and a second position which delivers the fluidic sample from the sample supply loop.

The column control valve has a first position which simultaneously delivers the fluidic sample from the sample supply loop to a sample
30 chromatography column in a first direction and delivers an elution solvent to the sample analyzer, and a second position which flows the elution solvent over the sample chromatography column in a second direction to deliver the fluidic sample and the elution solvent to the sample analyzer.

The wash control valve has a first position which supplies a wash buffer solution to the sample chromatography column in a forward fluid flow direction, and a second position which supplies elution solvent to flush the sample supply loop.

5 The above embodiment can include several additional features. The conduit can include a transparent portion. The fluid sensor can be an optical sensor and configured to detect the presence of fluid in the transparent portion of the conduit.

10 The system can include a robotic system for positioning the sipper to aspirate a fluid sample from a sample reservoir. The robotic system can be configured to lower the sipper into the reservoir until fluid is detected by fluid detector. The robotic system can be configured to prevent the sipper from traveling beyond a defined position. The defined position can be specified by a user. The robotic system can be configured to retract the sipper from the
15 reservoir when fluid is detected by the fluid sensor.

 Another embodiment is directed to a method of high-throughput sample injection comprising providing a vacuum source, a conduit in communication with the vacuum source, a fluid sensor configured to detect the presence of fluid in the conduit, a sample loop in communication with the conduit, and a
20 sipper in communication with the sample loop; applying suction to the sipper; inserting the sipper into a sample reservoir; and withdrawing the sipper from the sample reservoir upon detection of fluid by the fluid sensor.

 The above embodiment can include several additional features. The method can include withdrawing the sipper from the sample reservoir upon
25 advancement beyond a defined position. The method can also include reporting an error. The method can further include analyzing a sample held in the sample loop.

 Another embodiment of the invention provides a method of preparing an eluted sample containing salts or buffers from a liquid chromatography
30 device for analysis by a mass spectrometry device. The method includes: continuously providing a non-polar solvent to the mass spectrometry device; receiving the eluted sample from the liquid chromatography device; flowing the eluted sample over a solid phase extraction column; flowing the non-polar

solvent over the solid phase extraction column; and presenting non-polar solvent and the eluted sample to the mass spectrometry device.

This embodiment can have several variations. For example, the liquid chromatography device can be an ion exchange chromatography device. The liquid chromatography device can be a cation exchange chromatography device. The liquid chromatography device can be a size exclusion chromatography device. The method can also include the step of optically analyzing the eluted sample from the liquid chromatography device to generate an optical data set. The method can also include the step of associating the optical data set with a data set generated by the mass spectrometry device. The method can also include the step of flowing a polar wash solution over the solid phase extraction column.

Another embodiment of the invention provides a sample injection system for coupling a liquid chromatography device with a mass spectrometry device. The system can include a sample injection valve and a column control valve. The sample injection valve can include (i) a first position that allows sample from the liquid chromatography device to pass through the sample injection system, and (ii) a second position that loads sample from the liquid chromatography device onto a sample supply loop. The column control valve can include (i) a first position that simultaneously delivers the fluidic sample from the sample supply loop to a solid phase extraction column in a first direction and delivers an elution solvent to the sample analyzer, and (ii) a second position that flows the elution solvent over the solid phase extraction column in a second direction to deliver the fluidic sample and the elution solvent to the sample analyzer.

This embodiment can have several variations. The system can also include an optical detector for analyzing the sample from the liquid chromatography device. The system can also include a diversion valve located between the liquid chromatography device and the sample injection valve. The diversion valve can be actuated as a result of signal generated by the optical detector. The system can also include a fraction collector. The elution solvent can be a polar solvent or a non-polar solvent.

Another embodiment of the invention provides a method of preparing

an eluted sample from a liquid chromatography device for analysis by a mass spectrometry device. The method includes: continuously providing a polar solvent to the mass spectrometry device, receiving the eluted sample from the liquid chromatography device; flowing the eluted sample over a HILIC column;
5 flowing the polar solvent over the HILIC column; and presenting the polar solvent and the eluted sample to the mass spectrometry device.

The liquid chromatography device can be one selected from the group consisting of: an ion exchange chromatography device, a cation exchange chromatography device, and a size exclusion chromatography device. The
10 method can also include the step of optically analyzing the eluted sample from the liquid chromatography device to generate an optical data set. The method can also include the step of associating the optical data set with a data set generated by the mass spectrometry device. The method can also include the step of flowing a polar wash solution over the solid phase extraction
15 column.

BRIEF DESCRIPTION OF THE DRAWINGS

The foregoing features of the invention will be more readily understood by reference to the following detailed description, taken with reference to the accompanying drawings, in which:

20 FIG. 1 is a block diagram of a rapid chromatography system, in accordance with an embodiment of the invention;

FIG. 2(a) is a schematic of a rapid chromatography system that includes two injection valves, in accordance with an embodiment of the invention;

25 FIG. 2(b) is a schematic of the rapid chromatography system of FIG. 2(a) when a complex mixture is passed through a matrix in a first direction, in accordance with an embodiment of the invention;

FIG. 2(c) is a schematic of the rapid chromatography system of FIG. 2(a) when elution fluid is passed through the matrix in a second direction, in
30 accordance with an embodiment of the invention; and

FIG. 3 is a schematic of a multiplexed analyzer system, in accordance with an embodiment of the invention.

FIG. 4 is a schematic of an auto-injection device, in accordance with an embodiment of the invention.

FIG. 5(a) is a schematic of the auto-injection device of FIG. 4 during sample aspiration, in accordance with an embodiment of the invention.

5 FIG. 5(b) is a schematic of the auto-injection device of FIG. 4 when aspirated sample is output to a fluidic circuit, in accordance with an embodiment of the invention.

FIGS. 6(A)-6(D) is a schematic of an embodiment using three valves.

FIG. 6(E) depicts an embodiment of a three-valve auto-injective device.

10 FIG. 6(F) is an isometric projection of a bracket for supporting a three-valve auto-injection device.

FIG. 7 is a schematic of an auto-injection device incorporating a fluid sensor, in accordance with an embodiment of the invention.

15 FIG. 8(a) is a schematic of the auto-injection device of FIG. 7 before sample aspiration, in accordance with an embodiment of the invention.

FIG. 8(b) is a schematic of the auto-injection device of FIG. 7 during sample aspiration, in accordance with an embodiment of the invention.

20 FIG. 8(c) is a schematic of the auto-injection device of FIG. 7 when aspirated sample is output to a fluidic circuit, in accordance with an embodiment of the invention.

FIG. 9 is a flow chart illustrating the operation of a high-throughput sample injection system, in accordance with an embodiment of the invention.

FIGS. 10-11(e) are schematics of a system for coupling a mass spectrometry system and a liquid chromatography system.

25 FIG. 12 is a schematic of a method for processing an eluted sample from a liquid chromatography system in a mass spectrometry device

DETAILED DESCRIPTION OF SPECIFIC EMBODIMENTS

30 In illustrative embodiments, an automated system and method for increasing sample throughput and/or analysis of selected components in complex biological, chemical, or environmental matrices is presented. Generally, the system includes a chromatography column and fluidic circuit that is capable of rapidly outputting a plurality of samples to an analyzer, such

as a mass spectrometer. In various embodiments, sample throughput rates ranging from 30 seconds per sample to 1 second per sample or faster are achievable, depending on the specific application. Further embodiments of the invention include an auto-injection system that increases throughput and
5 minimizes sample carryover. Details are discussed below.

FIG. 1 shows a block diagram of a rapid chromatography system 100 for rapidly outputting an analyte of interest while removing undesirable salts, buffers, and other components from a complex mixture, in accordance with one embodiment of the invention. Such undesirable components may, for
10 example, degrade analyzer performance or cause an output signal from an analyzer to be suppressed.

The system 100 relies on the principle of back-elution for the specific purpose of increasing sample throughput. In particular, the complex mixture to be analyzed is delivered to an insoluble matrix 102 in a first direction 109
15 via a fluidic circuit 104. The matrix 102, which may be packed in a chromatography column, is selected such that the analyte(s) of interest is selectively immobilized. The matrix 102 may be, without limitation, various resins known in the art of chromatography. Typically, the analyte binds to the first part of the matrix 102 encountered due to a phenomenon known as
20 focusing. In focusing, a large amount of the analyte may be immobilized in a very small physical space within the head of the matrix 102 due to a strong affinity for that matrix 102.

Non-binding components of the complex mixture, which may include, without limitation, salts, buffers, and/or detergents, are not so immobilized and
25 pass through the insoluble matrix 102. These undesirable components are typically diverted to waste by the fluidic circuit 104. To ensure sufficient removal of the undesirable components, the matrix 102 may be washed for a predetermined period of time, while the analyte(s) of interest is still immobilized and focused on the head of the matrix.

30 Once the undesirable components have been removed from the matrix, elution fluid is passed via the fluidic circuit 104 over the matrix 102 such that the analyte(s) is no longer immobilized by the matrix 102. However, instead of passing the elution fluid over the matrix 102 in the first direction 109, the

elution fluid is passed over the matrix 102 in a second direction 111 that is substantially opposite the first direction 109, in accordance with preferred embodiments of the invention. Thus, the analyte(s) does not travel through the length of the matrix 102, but is instead back-eluted from the matrix 102 in the opposite direction it was loaded. Due to the focusing effect, the analyte(s) does not have to travel through the entire bed of the chromatography matrix 102, and a minimal amount of linear diffusion takes place. Thus, a sharp, concentrated sample peak can be output from the matrix 102 within a minimal bandwidth of time. The sharp sample peak obtained by back-elution is significantly sharper than those obtained when using conventional chromatography. The samples back-eluted from the matrix 102, which contain the analyte(s) of interest, can subsequently be introduced into an analyzer 116 in a rapid and concentrated manner.

In various embodiments of the invention, a controller 125 automatically controls the fluidic circuit 104 to periodically perform the steps of passing the fluid over the matrix 102 in the first direction and back-eluting the elution fluid over the matrix 102 in the second direction, so as to obtain a high sample-throughput rate. The controller 125 may include, without limitation, a processor which may be appropriately pre-programmed or configured to be loaded with an appropriate program. The controller 125 may work in conjunction with a robotic system that samples an aliquot of the complex mixture to be analyzed, and that allows for sequential presentation of each complex mixture to be analyzed.

Various methodologies may be used in which containers of each complex mixture to be analyzed and a sipper tube can be moved relative to one another to allow for sequential sampling. These methodologies include, but are not limited to, systems in which the containers of the liquid to be analyzed (*e.g.*, a microtiter plate or an array of vials) are held in a fixed position and the sipper tube is translocated by means of a robotic arm to sequentially sample each container. Robotic microplate positioning systems are known in the art as in U.S. Patent No. 5,985,214. Preferably, the robotic system should be capable of presenting samples at a rate that does not limit the throughput of the device. In other embodiments, the sipper tube can be

immobilized and each container to be analyzed can be moved into a position where an aliquot can be sampled. In an embodiment, liquid samples can be transported to the sipper tube with the use of a laminated tape or belt system for sequential analysis, as described in the following U.S. patents and patent applications: U.S. Patent Application Publication No. 2002/0001544, U.S. Patent Application Publication No. 2003/0119193, and U.S. Patent No. 6,812,030. Systems that incorporate elements of both approaches (*e.g.*, moving the sample containers in two dimensions and the sipper tube in one dimension) are also possible.

The fluidic circuit 104 may include a valving module 106 that is capable of alternately directing fluid over the matrix in the first direction and back-eluting the elution fluid over the matrix in the second direction. Valving module 106 may include one or more valves. For example, FIGS. 2(a)-(c) are schematics of a chromatography system 200 that includes a chromatography matrix 225 and two injection valves 206 and 207, in accordance with one embodiment of the invention.

FIG. 2(a) shows the position of the valves 206 and 207 when the complex mixture is being loaded into a sample loop 208. A reduced pressure 221 and an increased pressure 222 are continuously applied, by pumps, for example, to a first port and a second of port of the valve 206. The reduced pressure 221 is used to aspirate the complex mixture via a sipper tube 204. The sipper tube 204 may be, without limitation, narrow-bore capillary tubing. Enough of the complex mixture to fill sample loop 208 with a defined volume is aspirated. The amount of complex mixture to be passed over the matrix 225 can thus be controlled by the size of the sample loop 208. Any excess mixture aspirated is collected in a trap 209 that may be positioned, for example, between the injection valve 206 and the reduced pressure source 221.

In various embodiments, a wash solvent or buffer solution is positioned between the region of increased pressure 222 and the injection valve 206. While the complex mixture is being loaded into the sample loop 206, the increased pressure 222 applied to valve 206 pumps wash fluid to valve 207, which passes the wash fluid through the matrix 225 in a first direction 226.

The output of the matrix 225 is diverted to waste by the valve 207.

Additionally, an increased pressure 223 continuously applied to valve 207 pumps elution fluid, which may be positioned between the region of increased pressure 223 and valve 207, to an analyzer 240. In this manner, carryover
5 from previous complex mixture/samples is flushed from the matrix 225 and the analyzer 240 while the complex mixture is being loaded into the sample loop 206.

FIG. 2(b) shows the position of the valves 206 and 207 when the metered complex mixture from the sample loop 206 is passed through the
10 matrix 225, in accordance with one embodiment of the invention. Upon actuation of the injection valve 206, the complex mixture, followed by wash fluid, is passed through the matrix 225 in the first direction 226. Due to the focusing effect, the analyte of interest binds to the first part of the matrix 225, as discussed above. The wash fluid that follows the sample ensures sufficient
15 removal of the undesirable components (*e.g.* salts, buffers, detergents, etc.) from the matrix 225, which are diverted to waste. To clean the sipper tube 204 prior to aspiration of the next sample loop of complex mixture, the sipper tube 104 is dipped into a wash solvent or buffer solution. The reduced pressure 221 applied to the sipper tube 104 passes wash solvent through the
20 sipper tube 204 and into trap 209.

After the analyte(s) of interest has been loaded onto the matrix 225 and the undesirable components removed, the valves 206 and 207 divert the pumping system that loads the complex mixture onto the matrix 225 away from the head of the matrix 225. Simultaneously, an elution fluid is passed
25 through the matrix 225 in substantially the opposite direction 227 from which the complex mixture was loaded. The elution fluid, which may be either a solution or a solvent, dissociates the bound analyte(s) of interest from the matrix 225. In various embodiments, separate pumping systems are used to load the complex mixture, and pump the elution fluid across, the matrix 225.

FIG. 2(c) shows the position of the valves 206 and 207 when elution fluid is back-eluted through the matrix 225, in accordance with one
30 embodiment of the invention. Valve 207 is actuated to pass the elution fluid to the matrix 225 in the second direction 227. Since the analyte is primarily

immobilized within the head 226 of the matrix 225 due to the focusing effect, and does not have to travel the entire length of the matrix 225, thus limiting diffusion, the sample output of the matrix is delivered to the analyzer 240 in a concentrated manner within a small bandwidth of time. While back-eluting, wash solution is passed through the sample loop 206 to clean and prepare the sample loop 206 for subsequent aspiration of complex mixture.

The analyzer 240 may be, for example, an optical interrogator or mass spectrometer. In various embodiments, the sample may be presented directly to a mass spectrometer using a variety of standard systems, including atmospheric pressure chemical ionization (APCI), electrospray ionization (ESI) or atmospheric pressure photoionization (APPI). The mass spectrometer is capable of quantitatively analyzing a large number of compounds based on the mass-to-charge ratio of each compound. Further separation of individual compounds is generally not necessary, since an accurate mass-selective detection and quantification can be performed by mass spectrometry. The output of the MS is analyzed and the amount of compound present in the sample is determined by integrating the area under the MS peak.

After back-eluting, both valves 206 and 207 are actuated as shown in FIG. 2(a). The steps of loading the complex mixture into the sample loop 208 (if implemented), passing the complex mixture over the matrix 225 in the first direction, and back-eluting the elution fluid over the matrix 225 in the second direction are then periodically repeated so as to achieve a high sample-throughput rate.

Minimizing Peak Width of Matrix Sample Output

The sample peak width (at half height) at the output of the matrix 225 can be further minimized by selecting appropriate flow rates from the pumping systems 221, 222, and 223 and by selecting tubing diameters that further minimize linear diffusion as the complex mixture and samples are moved through the fluidic circuit 104. Typically, narrower bore tubing produces sharper peaks enabling higher throughput, but also lead to higher back-pressure in the fluidic pumping system. Similarly, higher flow rates also generally result in sharper peaks, but also lead to higher back-pressure. High

flow rates can also lead to decreased signal intensity in a mass spectrometer due to incomplete sample ionization. Determining the maximum throughput of the system is therefore a compromise between several factors that can be modeled or determined empirically. The various parameters used, including the nature and type of the insoluble matrix, pumping flow rates and pressures, tubing specifications, the nature of the fluids used to perform the rapid chromatography, and the timing of the switching of the fluidic valves 206 and 207 must be optimized for each family of chemical compounds to be analyzed. This set of optimized parameters makes up a compound-specific method for high throughput mass spectrometric analysis. In accordance with various embodiments of the invention, typical ranges for tubing diameters range from 20 μm to 300 μm and flow rates range from 0.1 mL/min to 5 mL/min resulting in backpressures that may reach anywhere from 5 to 6000 psi.

15 Minimizing Carryover

A major concern in maximizing sample throughput is the elimination of sample-to-sample carryover. Referring back to FIG. 1, any sample that is not removed from the fluidic circuit 104, matrix 102, and analyzer interface 130 after one analysis may cause interference with the next sample. If a sample with a low level of analyte is preceded by a sample with a high level of analyte, carryover from the first sample may result in an incorrect analysis in the second, low analyte sample. Minimizing carryover is typically achieved by washing the fluidic circuit 104, matrix 102, and analyzer interface 130 with a solvent that fully solubilizes the analytes of interest so that they are removed from the system 100. Various embodiments of the invention also use this technique, and the fluidic circuit 104, matrix 102, and analyzer interface 130 may be flushed with the elution buffer/wash solution to minimize sample carryover.

Washing of the fluidic circuit 104 and other components of the system 100 which contact the complex mixture and/or sample is conventionally a time consuming step and a long washing step between samples limits the overall throughput of the system. Therefore, a system that requires a minimum amount of washing while producing an acceptably low level of carryover is

highly desirable. This requirement can be achieved, in part, by making those surfaces in the fluidic circuit 104 and other components in the system 100 which contact the complex mixture and/or sample (including the sample loop 206, valving module 106, MS interface 130, etc.) bio-inert so as to minimize the amount of carryover and ease cleaning. Furthermore, due to the high backpressures generated by the pumping system, such surfaces must have a strong mechanical resistance and the ability to resist high pressure liquids without leaks.

A commonly used material for such systems is Poly Ether Ether Ketone (PEEK) that has strong chemical resistance and can be manufactured in a wide range of interior and exterior diameters. However, in preferred embodiments of the invention, the tubing within the fluidic circuit 104 is manufactured from polyimide. Polyimide tubing has exceptionally low carryover of even very highly hydrophobic compounds, can resist high pressures before failing and can be manufactured in the 20-300 micrometer inner diameters that are optimum for minimizing linear diffusion. Use of a polyimide fluidic system allows for very rapid washing steps between samples for a wide range of analytes with minimal carryover. Another option for the construction of the fluidic circuit 104 and other components which contact the complex mixture and/or sample is titanium or titanium alloys that are also known to have low carryover properties. The fluidic circuit 104 may also include a microfluidic biochip that may have, without limitation, channels having a diameter between 20 μm to 300 μm optimized for minimal linear diffusion.

Another embodiment of the invention is to construct the fluidic pathway in full or in part from a material such as stainless steel. Stainless steel is not a particularly bio-inert substrate and tends to strongly adsorb hydrophobic compounds in its surface. However, the surfaces of the fluidic circuit 104 and other components which contact the complex mixture and/or sample may be chemically or physically coated with a hydrophobic or hydrophilic film (*e.g.*, Teflon, polyethylene glycol) by methods known to those familiar to the art in a manner that will minimize the binding of analyte(s), thus minimizing carryover.

Fluidic Valves

In accordance with various embodiments of the invention, fluidic valves in the fluidic circuit 104 are actuated to reverse direction of flow across the chromatography matrix 102. Typically, the flow across the matrix 102 needs to be reversed twice for each sample output from the matrix 102. The complex mixture 102 is first loaded onto the matrix 102 in one direction and the analyte(s) are bound but other components (*e.g.* salts, buffers, detergents, etc.) are not. The flow is then reversed and the analyte(s) are eluted off of the matrix 102 in the opposite direction to which there were loaded and diverted to the analyzer 116 for analysis. Finally, the flow is reversed again in preparation for the next sample. In many fluidic valves used in such microfluidic applications, the flow of liquid through the valve is physically stopped during the time at which the valve is being actuated. Typical electronically actuated valve modules 106 can switch between states in 100 milliseconds or slower. Pneumatically actuated valves may be switched much faster, and may reach actuation times of 30 to 40 milliseconds. This short blockage of flow during the actuation time is not a concern during conventional LC where runs typically last minutes.

However, the blockage of flow becomes a concern at very high throughput rates where the sample throughput time approaches 1 sample/second. Typically, the injection valves that may be used in this system allow for fluidic communication between two ports and have two actuation positions. However, if the valve is adjusted to an intermediate state between the actuation positions, the fluid communication is physically cut and no fluid can pass through the valve. During the actuation procedure there is a finite amount of time as the valve is rotated from one position to the other that the flow of fluid through the valve is cut. The high-pressure pumps that are pushing fluid through the valves continue to operate during this time. The creation of a blockage in flow at the valve during the actuation process results in an increase in the pressure within the fluidic circuit between the valve and the high-pressure pump. If the pressure increase is large enough it will eventually result in a failure of the fluidic system and could result in a leak. With conventional valves systems, the pressure increase is transient and the

increase in pressure is not sufficient to actually cause a failure of the fluidic circuit. However, the blockage in the flow will be observed in the baseline of the mass spectrometer signal. Since the impurities in the solvent that result in the background MS signal are eliminated with a blockage in flow, the baseline
5 tends to drop significantly during the valve actuation. When the valve has finished rotating and a fluidic connection is reestablished, the increased pressure between the pump and the valve is released and a higher than normal flow of solvent is delivered to the mass spectrometer. This results in an increased amount of impurities entering the mass spectrometer and an
10 increase in the background signal. If this event overlaps with analyte signal it can lead to unsymmetrical peaks, distorted baselines, and generally poor quantification.

Ideally, the reversal of flow occurs at such a speed that there is no detectable disturbance in the flow rates and pressures during the flow
15 switching operation. A valve module 106 capable of 100 millisecond actuation times employed in an application where a sample throughput of one sample per second is being performed means that the flow to the analyzer 116 will be physically blocked for 200 milliseconds per second, or 20% of the overall sample analysis time. In various embodiments of the invention, a valve
20 module 106 utilizing, without limitation, pneumatic valves capable of actuation speeds faster than 100 milliseconds, and preferably on the order of 30 milliseconds or less are utilized.

Pneumatic valve actuators are available from VICI Valco Instruments of Houston, Texas. Pneumatic valve actuators can be coupled to any valve
25 (including valves from manufacturers other than VICO Valco Instruments) through a shaft coupling. Suitable valves include valves having porcelain rotors and/or diamond-like coatings, such as NANOPEAK™ valves available from Scivex, Inc. of Oak Harbor, Wisconsin.

An additional advantage of the above-described embodiments over
30 conventional systems is that the fluidic circuit is arranged such that the same solvent is always delivered to the mass spectrometer. Even when doing a step elution, the elution solvent is the only solution that is sprayed in to the mass spectrometer. While the wash solution containing the mass

spectrometer incompatible components of the reaction mixture is diverted to waste, elution solvent is sprayed in to the mass spectrometer inlet. In this manner a stable API spray is always maintained and variation in baseline due to different background signals from wash and elution solutions is eliminated.

- 5 Some advanced conventional systems divert the wash solution away from the MS inlet to avoid a buildup of non-volatile compounds in the source region. However, it can take several seconds to reestablish a stable spray in the MS inlet when the elution solvent is diverted to the MS. If the sample signal overlaps with this region of unstable spray, it can lead to problems with peak symmetry, baseline stability, and poor quantification.
- 10

Software

- Software used to analyze the data generated by the analyzer 116, which may be executed by the controller 125 or another processor, enables many features of a high throughput analysis. For example, the mass spectrometer output at the end of a long analysis at high throughput consists of a series of data point in which time versus intensity values are recorded at each mass channel being analyzed. If plotted in a Cartesian coordinate system, these graphs result in a chromatogram made up of a series of peaks, wherein an integration of the area under each peak can be correlated to the concentration of the sample that was analyzed.
- 15
- 20

- This integration event can be coupled to the switching of various valves in the fluidic circuit 104, in accordance with one embodiment of the invention. The time that a valve was actuated to back-elute the sample from the chromatography matrix into the mass spectrometer (or other analyzer) can be precisely recorded. It is known that until this event takes place no analyte(s) can be delivered to the mass spectrometer. Upon actuation of the valve, the mass spectrometer signal from the analyte(s) being back eluted from the chromatography matrix can be observed. The valve actuation time and the beginning of a mass spectrometer peak can be accurately mapped to one another in time, such that the peak integration algorithm consists of an integration of the mass spectrometer signal for a selected time period after each valve actuation. Even those samples that contain no detectable
- 25
- 30

analyte(s) can be accurately analyzed in this manner since an identical signal window is monitored and integrated in each and every case.

In some cases, an error in the fluidic circuit 104 may lead to no signal being seen in the mass spectrometer. An example of such an error could be a fluidic reservoir in which no sample was present. This would lead to air being injected on to the column 225 rather than an aliquot of sample. In such a case, only baseline signal would be detected for all analytes. Since the final quantification relies on a relative measurement (*i.e.* substrate versus product or analyte versus an internal standard) such an error can be easily detected. If the sum of the two or more analyte signals is below a certain threshold, that sample can be flagged as an error.

Multiplexing

In various embodiments of the invention, the time required for rapid chromatography and inter-sample washing is much larger than the sample peak width (at half-height) at the output of the matrix. In such embodiments, there may be several seconds of baseline mass spectrometer (or other analyzer) signal before the next sample to be analyzed is delivered to the mass spectrometer. This period is effectively a loss in productivity, since the mass spectrometer is not actively quantifying samples.

Because mass spectrometers are large footprint instruments that require a significant capital expense, two or more high throughput mass spectrometry interfaces 303 and 304 are used to feed samples to a single mass spectrometer 302, as shown in FIG. 3 in accordance with one embodiment of the invention. Each mass spectrometry interface 303 and 304 may include, without limitation, a rapid chromatograph system 100 described above. A selection valve 310 is placed between the plurality of high throughput mass spectrometry interfaces 303 and 304 and the mass spectrometer 302. When a sample from a given high throughput mass spectrometry interface 303 or 304 is ready to be analyzed, the selection valve 310 is used to direct that sample to the mass spectrometer 302 while the remaining interfaces 303 or 304 are diverted to waste. By staggering the sample delivery to the mass spectrometer 302 such that while one interface is being actively analyzed the others are in the washing or sample acquisition

steps, a plurality of interfaces 303 and 304 can be used on a single mass spectrometer 302, allowing throughput to be maximized.

Auto-Injection Device

FIG. 4 is a schematic of an auto-injection device 400 that includes a single injection valve 405, in accordance with one embodiment of the invention. The auto-injection device 400 may be used in combination with additional valves as in the above-described embodiment, and may be used to transfer samples from a sample reservoir to, without limitation, a fluidic circuit that may include an analyzer and/or a chromatography column.

As shown in more detail in FIG. 5(a), and similar to valve 206 in FIGS. 2(a-b), when the injection valve 405 is in a first position, (*e.g.* not activated), a source of reduced pressure 411 is used to aspirate a sample 401 through sample sipper tube 407 and into a sample loop 403. Upon actuation of the injection valve 405, the sample is introduced to a fluidic circuit 413 by applying increased pressure, as shown in FIG. 5(b). To clean the sipper tube 407 prior to deactivation of valve 405 and aspiration of the next sample, the aspirator tube 405 may be dipped into a wash solvent or buffer solution, with reduced pressure applied to aspirate wash solvent through the aspirator tube 104 and into trap 109. Thus, the combination of the constant negative pressure and the in-line trap eliminates the need for repetitive aspiration and dispensing of wash solution through a syringe.

Where an excess of sample is available, the reduced pressure source 411 may be, without limitation, a vacuum pump that is capable of applying a continuous vacuum to the distal end of the sample sipper tube 407. When a large enough volume of sample 401 has been aspirated into the sample loop 403 to fill it completely, the injection valve 405 is actuated and the sample is output to the fluidic circuit 413. The trap 409, located between the injection valve 405 and the vacuum pump 411 is used to collect excess sample. Changing the injection volume can be accomplished by changing the length of the sample loop 403.

However, in cases where an excess of sample 401 is not available or the sample is too valuable to expend, a metered amount of sample 401 may be aspirated into the injection valve 405. In a preferred embodiment of the

injection, this metering is performed through the use of a cut-off valve 415 located between the vacuum pump 411 and the injection valve 405. In preferred embodiments, the cutoff valve 415 is a solenoid valve with very rapid response times allowing for accurate and precise actuation in the millisecond time scale. The cutoff valve 415 may be used to aspirate an aliquot of sample into the sample loop 403 through the sipper tube 407 for a very precise and controlled amount of time. The volume of sample aspirated into the sample loop 403 can be precisely calibrated based on the diameters of the sample loop 403, sample sipper tube 407, and the timing of the cutoff valve 415. The longer the cutoff valve 415 is kept in the open position, the longer the aspiration of the sample and the larger the volume of sample aspirated into the injection valve 405.

In accordance with another embodiment of the invention, the continuous vacuum system may be replaced with a piston device in fluid communication with the injection valve 403, particularly in cases where a cutoff valve 415 is impractical, or where the plurality of samples to be analyzed has large differences in viscosity. Changes in viscosity may cause changes in the rate of sample aspiration. The amount of sample aspirated into the sample loop 403 can be metered, for example, by controlling the distance the piston is withdrawn within a cylinder. A sufficient time is allocated to the aspiration process to permit the entire metered amount of sample to be loaded into the sample loop 403. Imprecision in injection volumes due to differences in rates of aspiration caused by sample viscosity can thus be eliminated. The sample is aspirated directly into the sample loop 403 and then injected into the fluidic system. There is no need to apply positive pressure from the piston until it has reached the end of its traverse within the cylinder.

In other embodiments of the invention, various combinations of the above-described approaches for aspirating sample into the injection loop through the sipper tube may be used. For example, a selection valve may be used to select whether a cutoff valve in combination with a continuous vacuum source, or alternatively, a piston device, is placed in fluid communication with the injection valve. Where an excess of sample is

available, the selection valve is operated so as to place the cutoff valve and continuous vacuum in fluid communication with the injection valve, with the cutoff valve left in the open position. If the aspiration of the sample must be metered either the cutoff valve can be activated as described above, or the
5 selection valve can be actuated to use the piston-based aspiration system.

The auto-injection device 400 is advantageous over conventional auto-injectors for several reasons. By directly aspirating the sample into the injection loop rather than into a transfer syringe, the computer controlled robotic motion required for each injection is reduced. In conventional auto-
10 injector systems, the transfer syringe must first be moved into the sample reservoir to aspirate an aliquot of sample. Next the transfer syringe must be moved to the injection valve and the aliquot of sample loaded into the injection loop. After the injection, the syringe must be moved yet another time to one or more cleaning stations. Because the current invention aspirates the
15 sample directly into the injection loop, the need to move the transfer syringe from the sample reservoir to the injection valve is eliminated. Minimizing robotic movement within the device both increases the throughput and the reliability of the system. By repeatedly aspirating and injecting from the same sample, larger sample volumes may be analyzed without undue delay. If the
20 injection is to a chromatographic resin, multiple aliquots of sample may be added to the column prior to additional steps of washing and eluting.

Another advantage of the current invention is realized in the cleaning of the auto-injector between samples. All surfaces that come into contact with sample generally must be thoroughly cleaned before the next sample can be
25 injected. In conventional auto-injectors, this includes the injection valve as well as the transfer syringe. Cleaning of the transfer syringe can be especially challenging and time consuming, especially if a standard syringe is used. Because most syringes are manufactured from glass and stainless steel, certain samples are particularly difficult to remove. Many lipophilic
30 compounds tend to adhere strongly to stainless steel and can lead to sample carryover or leaching. Transfer syringes typically have tubing of various diameters and are composed of multiple materials (*e.g.*, glass and stainless steel) that are more difficult to clean than continuous and smooth bio-inert

tubing. Since the current invention aspirates samples directly into the injection valve through a sipper tube, the invention does not require the cleaning of a transfer syringe. This has the double impact of decreasing sample carryover while also increasing the throughput of the device.

5 Of importance to minimizing sample carryover is the choice of material used for the sipper tube. In a preferred embodiment of the invention, a concentric tube injector is used to provide the ability to pierce sealed sample reservoirs without the need to have the sample come into contact with materials such as stainless steel that are not chemically compatible with a
10 wide range of samples, as described in U.S. Patent No. 7,100,460.

 As described above, cleanup of the device can be achieved by simply aspirating a large volume of fluid through the sipper tube 407 while the sample of interest is being diverted to the fluidic circuit for analysis, as shown in FIG. 5(b). The use of biocompatible materials coupled with the small
15 surface area of the sipper tube and injection valve that needs to be cleaned allows for very efficient reduction of sample carryover while maintaining a rapid throughput.

 The following are examples, without limitation, of high-throughput sampling using various configurations of the above-described embodiments.

20 Example 1 – Drug-Drug Interaction (DDI) Assay

 Many xenobiotic compounds are metabolized *in vivo* by a family of enzymes known as cytochrome P450s, primarily in the liver. The metabolic activity by P450 enzyme also includes a vast majority of small molecule pharmaceuticals. Since the therapeutic activity of many pharmaceutically
25 active compounds is highly dose-dependent it is advantageous to understand the metabolic fate of these chemicals. In many cases, high doses of certain chemicals can be toxic or have long-term deleterious effects.

 Many compounds are known to affect the metabolism of certain P450 enzymes, either acting as inhibitors or activators. This is true for a range of
30 chemicals that are currently used as therapeutics. It is critical to know from a drug-safety standpoint whether or not the metabolic profile of a pharmaceutical compound taken by an individual may be affected by other chemicals that individual may be taking. If an individual is currently taking a

certain drug that inhibits the action of a specific P450 enzyme, taking a second drug that is also metabolized by the same P450 enzyme can have catastrophic consequences. The inhibitory effect of the first drug on the P450 enzyme can lead to the second drug not being metabolized at the predicted rate and result in much higher than expected *in vivo* concentrations. In some cases this can be toxic or even fatal.

To study the possible effects of potential new pharmaceutical compounds a series of *in vitro* assays known as the drug-drug interaction assays have been developed and are familiar to those skilled in the art. The assays use various preparations of P450 enzymes, either as purified recombinant proteins, or as various cellular or sub-cellular (*e.g.* microsomes, S9 fractions, etc.) preparations of liver tissue. The enzyme preparations are allowed to react with known substrates of the P450 enzymes, known as probes, under controlled conditions in the presence of the test compounds. If the test compound is active in the assay it will cause a shift in the expected metabolism of the probe molecule. A wide range of different probes and assays has been described in the scientific literature. These formats include both optically active probes typically used with recombinant enzyme preparations and mass spectrometric approaches that facilitate the use of subcellular liver preparations and highly selective and specific probes. While the throughput of optical assays can be very high, researchers generally prefer to perform mass spectrometry-based assays since more biologically relevant data can be obtained.

The above-described embodiments of the invention can be used to vastly improve the throughput of mass spectrometry-based drug-drug interaction assays. An assay to test the activity of test compounds against cytochrome P450-2D6 (CYP2D6) was performed in a 96-well microtiter plate. A microsomal preparation from human liver tissue was incubated in the presence of dextromethorphan, the test compound, and NADPH in a buffer containing potassium phosphate at pH 7.4 and magnesium chloride. After a 30 minute incubation, the reaction was quenched by acidifying the reaction with the addition of 10% (v/v) 0.1% formic acid. While the human liver microsomes have a variety of different enzymes, dextromethorphan is a

specific substrate of CYP2D6 and is metabolized into dextrorphan. While the remaining dextromethorphan substrate and the dextrorphan product formed in the reaction can be quantified by the use of conventional liquid chromatography-mass spectrometry at throughputs that is on the order of minutes per sample, the above-described embodiments of the invention allow similar analysis to be performed on the order of 5 seconds.

In accordance with an embodiment of the invention, and referring to FIGS. 2 (a)-(c), the sipper tube 204 attached to the valve 206 is moved relative to the first sample to be analyzed in the 96-well microtiter plate. The distal end of the sipper 204 is immersed into the reaction buffer and an aliquot is aspirated into a 5.0 microliter injection loop 208 through the use of a vacuum 221 applied to the distal end of the sipper tube 204. Fifty milliseconds after aspiration has begun, enough fluid has been aspirated into the valve 206 to completely fill the 5.0 microliter injection loop 208. At this time the injection valve 206 is actuated and the sample in the loop 208 is brought into fluid communication with the output from a high-pressure fluidic pump 222 that pushes the 5.0 microliter sample aliquot through the injection loop 208 and onto a chromatographic column 225 containing an insoluble matrix. The matrix consists of impermeable beads that are an average of 40 microns in diameter. The surface of each bead has been derivatized with a 4-carbon long alkane chain to create a hydrophobic environment. Porous frits constrain the insoluble matrix beads within the column 225, however, the nature of the particles allows for fluid to freely move around and between the particles without an unacceptably high increase in pressure.

The high-pressure fluidic pump 222 is used to pump water at a flow rate of 1.2 milliliters per minute. When the sample reaches the column 225 the dextrorphan and dextromethorphan analytes, being lipophilic molecules, interact with the insoluble matrix beads within the column and are adsorbed onto the column 225. Compounds in the reaction buffer that interfere with mass spectrometry, including the potassium phosphate buffer, magnesium chloride salts, NADPH and NADP, are highly hydrophilic, and accordingly, are flushed through the column into a waste container. Insoluble components in the assay buffer that may have been aspirated along with the sample are

small enough to move through the space between the 40 micron beads and are also removed from the analytes.

The total internal volume of the column 225 is 4.0 microliters. To remove the interfering salts at an acceptable level it is necessary to flush the column with several volumes of water. At a flow rate of 1.2 milliliters per minute, a total of 20 microliters of water per second is pumped. Therefore in the one-second wash, a total of 5 column volumes of water were pumped over the bed of matrix to remove the mass spectrometry incompatible components.

During this entire process a second fluidic pump 223 is used to pump a solution of 80% acetonitrile in water at a flow rate of 1.0 milliliters per minute directly on to a triple quadrupole mass spectrometer 240 operating in the electrospray ionization (ESI) mode. The mass spectrometer 240 was optimized to specifically monitor the dextrorphan and dextromethorphan analytes in multiple reaction monitoring (MRM) mode. A stable ESI flow was maintained and a constant baseline from the 80% acetonitrile solution was established. Exactly 1.0 seconds after the valve 222 to push the sample from the injection loop onto the matrix, the second valve 207 was actuated. This valve 207 forces the 80% acetonitrile to enter the column 225 from the direction opposite that from which the sample was loaded. Simultaneously, the output of the first pump 222 was diverted away from the column and to waste. The second valve 207 actuation brought the column 225 into fluidic contact with the second pump 223 and the analytes adsorbed on the column 225 were eluted by the 80% acetonitrile and pushed into the ESI source of the mass spectrometer 240 where they are analyzed. The two analytes were eluted simultaneously and analyzed in the mass spectrometer based on their mass to charge ratios.

Since the elution step is done in the opposite direction of the loading step, the analytes never travel through the column 225. This is an important point, since fluid traveling over a column 225 tends to undergo turbulence that can result in mixing and linear diffusion. Minimizing the linear diffusion is very important since this leads to an increase in the volume of fluid in which the sample is presented to the mass spectrometer. Best analytical data is

obtained when the sample is presented in the smallest possible elution volume in the shortest amount of time. Small elution volumes lead to high local concentrations of analyte and a correspondingly high signal level that can be distinguished from the background signal and shot noise. In 1.5
5 seconds at a flow rate of 1.0 milliliters per minute a total of 25 microliters of elution fluid was pushed over the column 224. This corresponds to over 6 column volumes of elution fluid, more than enough to flush the column and eliminate carryover to the next sample.

At this time both valves 222 and 207 were actuated again to their
10 starting positions. The injection loop 208 was available to aspirate the next sample, the 80% acetonitrile from the second high pressure pump 223 was diverted away from the column 225 and directly to the mass spectrometer 240 and the water from the first high pressure pump 222 was pushed over the column 225 in the original direction. This state was maintained for a minimum
15 of 2 column volumes to (a minimum of 400 milliseconds) to allow the local environment within the matrix of the column 225 to be flushed with water to allow binding of the analytes in the next sample. This process is known as column equilibration and must be performed in sufficient time to allow proper analysis. After the equilibration of the column 225 the next sample was ready
20 for analysis.

While the sample was being analyzed 80% acetonitrile from a reservoir was aspirated through the sipper tube 204 to remove any contamination in the sipper tube and to eliminate carryover into the next sample. In this manner, samples were analyzed at a periodic rate of 5 seconds per sample. It is also
25 possible to analyze more than two analytes simultaneously and therefore to multiplex assays for multiple P450 isoforms.

Example 2 – Metabolic Stability Assay

In accordance with various embodiments of the invention, certain attributes of the system may be advantageously enhanced at the expense of
30 throughput. An example of such an application is the metabolic stability assay. The metabolic stability assay assesses the activity of liver enzymes on a test molecule. It is typically an *in vitro* assay performed by incubating a sub-cellular liver preparation (e.g. liver microsomes or S9 fraction) with a source of

energy (e.g. NADPH) and the test compound in an appropriate buffer system. The liver enzymes may metabolize the test compound, the rate of which can be determined by quantifying the amount of the test compound at controlled times using mass spectrometry.

- 5 This assay is different from the DDI assay in that each and every test compound must be monitored in the mass spectrometer. In the DDI assay, only a specific set of probes needed to be monitored allowing for a full optimization of the system. Given that a very wide range of test compounds needs to be analyzed in the metabolic stability assay, generic methods
- 10 capable of analyzing many different chemical structures are required. In this application, the throughput of the system is lowered slightly to facilitate the analysis of a wider range of analytes.

- To perform the metabolic stability assay, a different approach than the DDI assay is used. The reverse elution (*i.e.*, eluting the analytes from the
- 15 column in the opposite direction to which it was loaded onto the column) is not used. Rather, the analytes are eluted from the column in the same direction as they were loaded on to the column. This results in linear diffusion as the analytes experience turbulent flow as they are pushed over the insoluble matrix beads, causing a broader peak and therefore lower throughput.
- 20 However, many of the aspects of the invention used in the DDI assay can still be applied to the assay and result in a significantly increased throughput over conventional methods without a sacrifice in the sensitivity of the assay. These advantages will be described in detail below.

- The sample aliquot is aspirated into an injection loop and loaded onto
- 25 the column in the same manner as in the DDI assay with a first high-pressure pump that is used to pump a wash solution. This solution is typically water or an aqueous buffer and is used to flush out the salts, buffer components, NADPH, and insoluble components of the reaction mixture to a waste container. During this time a second high pressure pump is used to pump an
- 30 elution solvent in to the ESI or APCI source of the mass spectrometer. When the second valve is actuated, the elution fluid is forced over the column in the same direction that the analytes were loaded on to the column. An elution fluid that is capable of dissolving a very wide range of chemicals but is

compatible with atmospheric pressure ionization mass spectrometry is used. These buffers may include alcohols (*e.g.* methanol, ethanol, or isopropanol), acetonitrile, acetone, tetrahydrofuran or mixtures of these solvents. It is generally desirable to have a small amount of water in the mixture, and
5 additives such as ammonium acetate, ammonium carbonate, or DMSO to the elution solution may result in sharper peaks.

If the mass spectrometric characteristics of the analytes of interest are previously known, the mass spectrometer can be set up to specifically monitor those compounds in MRM mode. However, if no previous information is
10 available about the analytes, it may be desirable to use a mass spectrometer to scan a range of masses. The use of a time-of-flight, ion trap, hybrid quadrupole/ion trap, or hybrid quadrupole/time-of-flight mass spectrometer can facilitate the scanning of a wide range of masses with minimal loss in signal intensity. To obtain good quantitative data an internal standard is
15 added upon the quenching of the reaction and the signal from the analyte(s) is normalized with respect to the internal standard.

The current invention uses a step-elution system to purify samples using column chromatography and analyses them using mass spectrometry. However, the system uses a significant improvement over conventional step
20 elution systems: the same solvent system (the elution solution) is always sprayed into the inlet of the mass spectrometer. In conventional systems, as the wash and elution solutions are alternated for each sample and the two different solutions are alternately sprayed in to the mass spectrometer inlet. This can have a huge impact on the baseline signal observed. The variation
25 is baseline signal may have a significant impact on the quantification of peaks, particularly those that have a low level of signal.

In some of the more advanced conventional systems the wash solution is diverted away from the mass spectrometer inlet to a waste container and only the elution solution is sprayed into the mass spectrometer. However, this
30 also results in a change in the background signal seen in the mass spectrometer since there is no flow during the column loading and washing stages. Furthermore it may take several seconds to reestablish a stable spray in the MS inlet. In a high throughput system such as described here,

the leading edge of the analyte peak may overlap with the region of unstable flow resulting in poor sensitivity, uneven peak shape and increased error in quantification.

A further improvement the current invention provides over conventional systems is in the fast switching valves. Typical electronically actuated valves provide switching times over 100 milliseconds. However the very fast valve switching (*e.g.* 50 milliseconds or less) employed in the current invention allows for a pulse-free spray in the mass spectrometer providing symmetrical peaks with flat baselines and facilitates accurate quantification.

10 Example 3 – Compound Purity Testing

In some applications, the samples to be analyzed are already in a buffer that is compatible with mass spectrometry. Such an application may be the quality control analysis of test compounds in an aqueous or organic buffer that can directly be sprayed in to an API source without the need for any purification. Various aspects of the above-described embodiments can be used to increase the sample throughput for such an application.

In this application only a single injection valve is used. An aliquot of sample is aspirated into the injection loop and upon actuation of that valve the sample is sprayed directly in to the mass spectrometer. The flow from a single fluidic pump is used to push the sample through the injection loop and into the inlet of the mass spectrometer. In preferred embodiments of the invention, the fluid used to push the sample on to the mass spectrometer is one that the analytes are highly soluble in and provides good ionization in the mass spectrometer inlet. These solutions may include alcohols (*e.g.* methanol, ethanol, or isopropanol), acetonitrile, acetone, tetrahydrofuran or mixtures of one or more of these solvents with water.

The system provides increased throughput over conventional systems through several means. The elimination of a transport syringe to move a sample aliquot from a reservoir to the injection valve increases the overall speed of the robotics. The injection valve and the sample reservoir may be moved relative to each other such that an aliquot of sample can be directly aspirated in to the injection loop. Besides facilitating faster robotics, this also eliminates the need to clean the transport syringe between injections. In

conventional systems both the transport syringe and the injection loop need to be thoroughly cleaned between samples. However, in the current invention there is no transport syringe.

Samples that do not need to be purified have been analyzed by
5 atmospheric pressure ionization at throughputs of approximately 1 second per sample using the system with minimal carryover between samples.

Avoiding Sample-To-Sample Carryover

As explained above, the entire fluidic system is properly cleaned
between analyses to ensure that carryover from a given sample does not
10 confound the analysis of the next sample in the cue. Cleaning of the fluidic system is achieved by flushing the fluidic system with a solvent that the confounding compounds are freely soluble in. The fluidic system described herein is composed of two major components, each associated with a valve assembly. A first valve contains an injection loop and a sample aspiration
15 tube through which an aliquot of a fluidic sample to be analyzed is aspirated into the injection loop. This valve arrangement is in fluid communication with the second valve that contains a chromatography system containing an insoluble matrix that is capable of purifying the sample prior to analysis.

Samples aspirated into the injection loop are loaded onto the
20 chromatography column and washed with a "wash solution" in a first direction and, after an appropriate purification has been performed, the samples are eluted from the chromatography column with an elution solution in a second direction, opposite to the first direction. Since the elution solution removes the sample from the column by solubilizing it, it has the secondary effect of
25 cleaning the chromatography column and effectively reducing carryover into the next sample. Several column volumes of elution solution may be necessary to reduce carryover to an acceptable level, depending on the exact nature of the analyte of interest and elution solvent used. A larger volume of elution solvent can be delivered to the column by increasing the time that the
30 column is flushed with the elution solution in the second direction.

While the action of eluting the analyte off of the chromatography system has the added benefit of cleaning the column and valve assembly, carryover can also result from traces of analyte left in the injection loop,

sample aspiration tube, or in the portions of initial valve. This portion of the fluidic circuit can also be flushed with elution solution to clean the system between samples and eliminate or minimize carryover effects. In one embodiment of the invention, the sample aspiration tube can be moved to a reservoir containing elution solvent. The valve can be actuated such that the sample loop is in fluid communication with the sample aspiration tube, and a necessary volume of elution buffer can be aspirated through the sample aspiration tube and the injection loop. The solvent so aspirated will be collected in the in-line trap downstream of the valve before the source of vacuum.

In one representative embodiment, the aspiration of the elution solvent through the sample aspiration tube and injection loop will occur simultaneously with the back elution of the analyte from the chromatography column to the analyzer. This allows maximizing of the throughput since the flushing of the first valve containing the sample injection loop will be achieved during the time that the sample is being eluted. However, there may be particularly difficult analytes (*e.g.* very hydrophobic compounds) that despite the use of bioinert materials and surface coatings still cause carryover to be observed. These particularly difficult analytes may require a large volume elution solvent to be aspirated through the sample aspiration tube and the injection loop to eliminate carryover to an acceptable level. In some cases the aspiration of a large volume of elution solution through the sample aspiration tube and the injection loop may take longer than the back-elution of the analytes from the chromatography column to the analyzer. In such a case flushing the fluidic circuit to minimize carryover becomes a limiting factor for system throughput.

Thus another embodiment of the invention addresses the case of those analytes where flushing the fluidic system is a throughput-limiting event. Such an embodiment increases the volume of elution solvent with which a portion of the fluidic system is flushed between samples to further minimize carryover while still maximizing sample throughput. To this end, an additional fluidic valve can be used. This valve can be a selection valve or a 4-port injection valve. In one embodiment, a 6-port injection valve is used which is identical

to the two existing valves where 2 of the ports have been short-circuited with a piece of tubing. The use of identical valves throughout the fluidic circuit has advantages in manufacturing, inventory management, and servicing of the instrument.

5 The fluidic valve associated with the sample injection loop (valve 1) is flushed with elution fluid from a positive pressure source, such as an additional high-pressure fluidic pump, rather than being aspirated through the valve with a vacuum as described above. The use of positive pressure allows for a much larger volume of fluid to be flushed through the valve in a given
10 amount of time as compared to the use of vacuum aspiration. Under normal conditions, the maximum pressure with which a fluid can be aspirated is 1 atmosphere, assuming a perfect vacuum can be applied. In comparison, a standard high-pressure pump can apply tens of atmospheres of fluidic pressure, resulting in a much larger volume of fluid being delivered in an
15 equivalent amount of time.

 In this embodiment, the sample is aspirated into the sample loop on valve 1 as before. The valve is then actuated and the analytes are diverted to the second valve and loaded onto the chromatography column as before. Once the sample is purified through flushing with an appropriate volume of
20 wash solution, valve 2 is actuated and the sample is back-eluted onto the analyzer with elution solution. During this time the additional upstream valve is simultaneously actuated to flush the fluidic circuit in valve 1 (including the sample injection loop) with elution solution. The sample aspiration tube is not in fluid communication with the sample loop at this time, and still must be
25 flushed with aspiration of elution solution from an appropriate reservoir. Before the next sample is aspirated the additional valve is actuated once again and wash solution is flushed through the fluidic system and the chromatography column to equilibrate the system.

Three-Valve Embodiment

30 FIGS. 6(A)-(D) show an embodiment of the present invention using a three-valve and three-pump arrangement as described above. FIG. 6(A) shows the first assay phase in which a liquid sample is loaded into the sample injection loop. A sipper tube 604 is lowered into the sample reservoir and an

aliquot of sample is aspirated into the injection loop via sample injection valve 601 much as described before with respect to two-valve embodiments. Any excess sample aspirated is collected in a vacuum trap 621 downstream of the loop. During this time fresh elution solvent is delivered to the analyzer 640 from first elution solvent pump 624 to establish and maintain stable ESI or APCI spray and a corresponding stable baseline signal from the mass spectrometer. The column 625 is equilibrated by pumping wash buffer (typically an aqueous solution) from wash solvent pump 622. Second elution solvent pump 623 is diverted directly to waste by solvent valve 603.

FIG. 6(B) shows the next phase, in which the valves are aligned to load the sample onto the column 625 and wash the lines. Sample injection valve 601 is actuated and the sipper tube 604 is raised out of the sample reservoir. The sample aspirated into the injection loop is delivered to the column 625 where the analytes of interest bind, but interfering compounds (*e.g.*, salts, detergents, etc.) pass over the column 625 and are sent to waste. An appropriate number of column volumes of wash buffer are pumped to over the column 625 to ensure that the sample is purified properly.

Next, as shown in FIG. 6(C), the sample is back eluted off the column 625 into the analyzer 640. Column control valve 602 and wash control valve 603 are simultaneously actuated while the sipper tube 604 is moved into the "wash solution" reservoir. The wash solution is aspirated through the sipper tube 604 to clean it and eliminate sample-to-sample carryover. Actuation of column control valve 602 results in the elution solvent from first elution pump 624 to be delivered to the column 625 in the opposite direction and the bound sample is back-eluted into the analyzer 640. An appropriate number of column volumes are pumped over the column 625 to fully elute the sample. Actuation of wash control valve 603 causes elution solvent to be pumped by second elution pump 623 over the injection loop and diverted to waste. This allows for a complete flushing of the injection loop with elution solvent between each sample and helps to minimize carryover.

Finally, FIG. 6(D) shows equilibration of the column 625 and aspiration of the next sample. All three valves 601, 602 and 603 are simultaneously actuated to their home positions. Fresh elution solvent is pumped to the

analyzer 640 from first elution solvent pump 624, while wash solvent is pumped over the column 625 in the forward direction. An appropriate number of column volumes are pumped over the column 625 to ensure column equilibration. The sipper tube 604 is then dipped into the next sample, an aliquot is aspirated into the injection loop, and the cycle is repeated.

The term "column volume" is referred to several times. A variety of column geometries, volumes, and packing can be used in various specific applications and may be optimized for each application. Both physical characteristics of the packing material (*i.e.*, particle size, shape, porosity, etc.) and packing chemistry (*i.e.*, C-18 vs. polymeric packings, etc.) are important in optimizing a give embodiment. Typically, column bed volumes under 10 μL are used. At a flow rate of 1.2 mL/min, 20 μL /second is delivered to the column 625. This means that in a typical 1.5 second elution, 3 volumes of a 10 μL column or 6 volumes of a 5 μL column can be realized. Maximum throughput is achieved by using the smallest acceptable column bed volume for a given application.

The elution solvent (usually an organic solvent) is delivered to the analyzer 640 in an uninterrupted manner at a constant flow rate. Even though a step elution is performed, only the elution solvent is delivered to the analyzer 640 in this arrangement. This allows for a stable ESI or APCI flow to be established and a constant baseline to be achieved. Analytes are simply "inserted" into this flow with the actuation of column control valve 602 facilitating very sharp and symmetrical peaks on a constant and stable baseline.

Other methods for performing a fast-step elution may be somewhat problematic. For example, the wash solution and elution solvent can be delivered to the analyzer 640 in a periodic manner. But this can result in large changes in background signal from the two solutions. Attempting to deconvolute a peak on a changing baseline is very troublesome. Also, the wash solution (typically containing the salts and other incompatible components of the sample mixture) can be diverted to waste and only the elution solution diverted to the analyzer 640. But establishing a stable ESI flow can take significant time and if the analytes of interest are delivered to

the analyzer 640 before a stable flow is established, sensitivity and quantification issues are encountered.

Standard electro-mechanically activated valves may not be appropriate in the embodiments just described due to slow actuation times. Since the valves do not permit fluid communication during the actuation process, flow is physically cut. The interruption in flow manifests itself as a “negative” peak in the baseline signal twice per cycle (actuation the valve and then returning to the home position). Given the throughputs of specific embodiments, at least one of these two “negative” peaks interferes with the analyte signal from the analyzer 640 resulting in a double peak that impacts on quantification and data quality. The solution is a very fast actuating valve (*e.g.*, 30 μ sec actuation time or faster) which has been engineered specifically for the task. The actuation time of the valves should be rapid enough that no interruption to the flow can be observed physically or in the data.

Wash control valve 603 can be replaced with a 4-port valve. However, to keep manufacturing and parts inventory simple, a six-port valve identical to valves 601 and 602 may be preferred, and 2 ports can be permanently “short-circuited” with a piece of tubing. To reduce sample carryover and promote valve performance and durability, various valve components (*e.g.* the stator) can be formed from ceramics and can be coated with materials such as polytetrafluoroethylene (PTFE) or diamond-like carbon (DLC). Suitable valves include part number S-15287 available from Upchurch Scientific, Inc. of Oak Harbor, Washington.

Wash control valve 603 also could be eliminated from the system and the injection loop could be cleaned between samples by placing the sipper tube 604 into the wash solution and aspirating solvent through the sample supply valve 601 in the “load” position where the injection loop is in fluid communication with the sipper tube 604. However, a greater amount of fluid can be pumped with positive pressure than can be aspirated with negative pressure in a given amount of time. For analytes where sample-to-sample carryover is problematic, the use of wash control valve 603 can help minimize carryover by washing the loop with a greater amount of solvent while maintaining throughput.

The valves can be actuated by an electric or pneumatic actuation. Suitable actuators include the VEXTA® two-phase stepping motor, available from Oriental Motor Co., Ltd. of Tokyo, Japan under part number P0040-9212KE. The actuator and the valve can be coupled by part number DK/GS 9
5 from GERWAH Drive Components, LP of Fayetteville, Georgia.

In certain embodiments, the valve actuator(s) is controlled with software and/or hardware. The software and/or hardware can control the timing of movement of the valve actuator(s). Additionally, the software and/or hardware can control the velocity and/or acceleration of the valves to achieve
10 optimal performance and/or longevity. For example, in high-speed application, it is desirable to apply a braking or decelerating force as the valve approaches the desired position in order to prevent damage to the valve.

Figure 6(E) depicts an embodiment of the three-valve embodiment. System 600 includes valves 601, 602, and 603 (not visible), and sipper 604,
15 all mounted on a bracket 650. The valves 601, 602, 603 are connected by tubing as depicted in Figure 6(a). Advantageously, by mounting the valves 601, 602, 603, and sipper 604 on the bracket, the length of tubing required is minimized, which enables higher throughput as smaller volumes are held in the tubing.

20 Each motor 651, 652, and 653 (not visible) is also mounted on bracket 650 for actuation of valves 601, 602, and 603. In some embodiments, the sipper 604 is a sipper in accordance with U.S. Patent No. 7,100,460.

The bracket 650, along with the sipper 605 can be moved at least vertically (*i.e.* in the *y* direction) by control assembly 654, which can be an
25 electrical, mechanical, or electromechanical device as known to those of skill in the art. Likewise, reservoir plate 401 can be moved at least in the *x* and *z* directions by control assembly 655, which can be an electrical, mechanical, or electromechanical device as known to those of skill in the art. Control device 655 can interact with plate handling device 656 to obtain pre-loaded plates
30 401 and return plates after samples are extracted.

Figure 6(F) provides an isometric projection of bracket 650, with valves 601, 602, 603 (again, not visible), motors 651, 652, and 653, and sipper 604. As seen more clearly in Figure 6(E), valves 601, 602, and 603 are mounted at

angles so that the tubing distance between each valve can be further reduced.

When used in conjunction with other aspects of this invention, such as the use of bio-inert materials and surface coatings, three valve embodiments
5 result in a minimum amount of carryover for even the most difficult compounds, while allowing for a very high rate of sample analysis.

Improved Fluid Injection Valve Timing

Another embodiment of the present invention provides a device and method for the rapid sequential analysis of a plurality of samples. The device
10 comprises a computer controlled robotic system that aspirates an aliquot of fluidic sample from a sample reservoir directly into an injection loop in a fluidic injection valve. This improvement allows for the device in the current injection to realize higher throughputs while minimizing sample carryover.

As depicted in FIG. 7, activation of the valve is controlled by a
15 feedback mechanism that includes a fluidic sensor 702 located between injection valve 405 and vacuum trap 409. Fluid sensor 702 detects the presence of fluid in conduit 704. The fluid sensor 702 is used to control the precise timing at which the injection valve 405 is actuated as well as determining the position of the sample aspiration tube 407 with respect to the
20 fluidic sample to be analyzed. Several embodiments of the invention are described below.

In accordance with one aspect of the invention, the transfer syringe is completely eliminated from the auto-injection device. The sample to be analyzed is aspirated directly in to the injection loop 403 of the injection valve
25 405 through an aspiration tube 407 attached directly to a port of the injection valve 405. The computer-controlled robotic system of the device allows for the movement of the aspiration tube 407 connected to the injection valve 405 to be moved relative to the sample reservoirs 401.

An aliquot of sample can be aspirated directly into the injection valve
30 405 through the aspiration tube 407 in a number of ways. In cases where an excess of sample is available and continuous vacuum may be applied to the distal end of the injection valve 405 with a vacuum pump 211. When a large enough volume of sample has been aspirated into the injection loop 403 to fill

it completely the injection valve 405 is actuated and the sample is introduced to the fluidic circuit. A trap 409 located between the injection valve 405 and the vacuum pump 211 is used to collect excess sample. In this embodiment of the invention, the amount of sample injected is controlled solely through the
5 volume of the injection loop 403. Changing the injection volume requires changing the injection loop of the device.

During sample analysis it is important that the injection volume is known such that an accurate measurement is achieved. This is particularly important when a plurality of samples is analyzed serially, since an
10 inconsistent injection volume may lead to variability in the measurements. The current invention relates to a device and method that ensures that a full injection loop with a minimum amount of sample is achieved with every sample aspiration, even in cases where the viscosity, amount, temperature, or other physical parameters of a plurality of samples may differ.

15 As the aspiration tube 407 is lowered into the fluidic sample 401, the aspiration tube 407, injection loop 403, and valve 405 will be free of liquid and contain only ambient air. When the aspiration tube 407 is dipped into the fluidic sample 401, the sample 401 will be drawn into the aspiration tube 407 and fill the injection loop 403 and eventually be collected within a vacuum trap
20 409. Actuation of the injection valve 405 results in the volume of fluidic sample located within the injection loop 403 to be introduced into the sample purification or analysis device. The timing of this actuation is critical since actuation of the valve 405 too quickly results in an incompletely full injection loop 403 while actuation of the valve 405 too late results in a waste of sample.
25 In cases where only a small amount of fluidic sample is available for analysis, a late valve actuation may lead to the entire sample traveling through the injection loop 403 and being collected in the vacuum trap 409, resulting in an incomplete or empty injection loop 403 and a loss of the sample.

While it is possible to determine the timing of the proper valve actuation
30 empirically, this is a time consuming and error-prone process that typically results in an excess of sample being aspirated leading to waste. Empirical determination of the valve actuation timing becomes exceedingly difficult when the volume of the fluidic sample to be analyzed is very small.

Furthermore, small changes in the physical characteristics of the fluidic sample, such as the viscosity, temperature, or insoluble materials such as cellular or sub-cellular components can greatly affect the sample aspiration rate leading to inconsistency when a plurality of samples is aspirated. The
5 present invention overcomes these problems.

The present invention provides a device wherein activation of injection valve 405 is controlled by sensor 702. In some embodiments, the sensor 702 also controls certain aspects of the sample aspiration process.

In one embodiment, the device of the current invention comprises a
10 sensor 702 that is placed between the injection loop 403 and the vacuum source 211. When fluidic sample reaches the sensor 702 the interface between air and fluid generates a signal that is used to trigger the actuation of the injection valve 405. The presence of fluid at the sensor 702 can only be achieved when the sample aspiration tube 407 and injection loop 403 are
15 completely full of fluidic sample. By minimizing the volume between the distal end of the injection loop 403 and sensor 702 the amount of sample that is aspirated prior to the actuation of the valve 405 can be minimized, resulting in a minimal amount of sample being wasted during the analysis.

In another preferred embodiment of the present invention, feedback
20 from the sensor 702 is used not only to trigger the actuation of the injection valve 405 when the sample injection loop 403 is full, but also to control the mechanical movement of the sample aspiration tube 407. In this embodiment, as depicted in FIGS. 8(a)-(c), the sample aspiration tube 407 is lowered into the vessel 802 in which the fluidic sample to be analyzed is
25 contained. The level of fluid within this vessel 802 does not need to be known *a priori*. When the aspiration tube 407 is moved to a position below the level of the fluidic sample (FIG. 8(b)), the sample will be aspirated through the aspiration tube 407 into the loop 403. When the loop 403 is full and the fluidic sample reaches and triggers the sensor 702, the injection valve 405 will be
30 actuated and the sample aspiration tube 407 will be moved such that it is removed from the fluidic sample, typically by raising it up and out of the vessel 802 containing the fluidic sample (FIG. 8(c)). In this manner, vessels 802 with differing amounts of fluidic volumes can be accurately interrogated without

any previous knowledge of the volume of sample in each vessel 802. Furthermore, multiple analyses be required from a single vessel 802, the movement of the sample aspiration tube will automatically compensate for the reduced amount of sample after each analysis.

5 In another embodiment of the invention, a “safe” level will be set such that the sample aspiration tube 407 may not travel below a predetermined distance. The “safe” level may be determined at manufacture or may be configurable to reflect the dimension of particular sample reservoirs. This embodiment protects that sample aspiration tube 407 from making contact
10 with the bottom of an empty vessel 802, and in cases in which there is no fluidic sample in a given reaction vessel prevents the sample aspiration tube 407 from moving continuously until the bottom of the vessel 802 is reached and physical damage to the sample aspiration tube 407 and/or the reaction vessel 802 may occur.

15 In a further embodiment of the invention, if the sample aspiration tube 407 is lowered to the “safe” level without the sensor 702 being triggered, the computerized control system will produce an error message indicating that an aspiration of fluidic sample did not occur. This could be due to several reasons, including but not limited to a vessel 802 not containing any fluidic
20 sample, a clog or plug in the sample aspiration tube 407, or a loss of vacuum. An error message generated will permit the user of the invention to pause the analysis and solve the problem before continuing.

Suitable fluid sensors 702 include optical sensors such as those available from OPTEK Technology of Carrollton, Texas.

25 In another embodiment, an optical fluid sensor can be fabricated from a light source, two lengths of fiber optics, and an optical detector. One end of the first fiber optic length is coupled with the light source (e.g. a lamp, a red diode laser, and the like). The other end of the first fiber optic length is coupled with a first optical window on the conduit, for example, with optical
30 glue. One end of the second fiber optic length is coupled with a second optical window on the conduit. The other end of the second fiber optic length is coupled with an optical detector, for example, a visible light detector.

Figure 9 is a flowchart depicting the use of the system described

herein. In step S902, the vacuum source 211, 411 is initiated. In step S904, sipper aspiration tube 407 is introduced in sample reservoir 401, 802. In step S906, sensor 702 detects a sample in conduit 704. In step S908, valve 405 is actuated to release the sample in injection loop 403. In step S910, sipper aspiration tube S910 is retracted from sample reservoir 401, 802. In step S912, wash solution is aspirate through sample aspiration tube 407 before the process is repeated.

One skilled in the art will appreciate that the steps depicted in Figure 9 need not necessarily performed sequentially. Rather, certain steps may be performed concurrently, simultaneously, and/or in parallel. For example, sample aspiration tube 407 can be retracted from reservoir 401, 802 while valve 405 is actuated.

Coupling of Mass Spectrometry Devices with Systems Containing Salts or Buffers

Many biological separations use ion-exchange chromatography (*e.g.* cation exchange or anion exchange) or size-exclusion chromatography. These techniques have particularly important applications in the separations of proteins, peptides, oligonucleotides and many other analytes. The separation techniques have many applications ranging from scientific research and development through the manufacturing of pharmaceutically active compounds.

The techniques typically rely on the selective elution of individual analytes in a complex mixture from a chromatography matrix in response to the variation of one or more biophysical parameters. For example, in cation exchange chromatography, the concentration of cations in the elution buffer is typically increased in a gradual manner. When the concentration of cations in the elution buffer reaches a level at which the affinity of the cations in the elution buffer for the chromatography matrix is stronger than the affinity of the analyte, the analyte is displaced from the chromatography matrix by the cation. The displaced analyte is then eluted from the column. Since different analytes within a complex mixture typically have different affinities for the chromatography matrix a separation can be achieved. Other ion exchange systems rely on a change in pH to enact the desired separation. Many

parameters, such as the selection of chromatography matrix, the selection of the cation used in the elution buffer, the rate at which cation concentration is varied, and others may need to be optimized in order to affect a desirable separation.

5 In size exclusion chromatography (SEC), a separation of analytes is performed based on the relative size of the analytes. Typical SEC separations are performed using a chromatography matrix that consists of porous particles. When a mixture of analytes is introduced on the column, smaller analytes tend to travel through the pores in the chromatography
10 matrix, whereas those analytes that are too large are excluded from the pores and travel through the spaces between the particles. As a result, large particles tend to have a shorter residence time in the chromatography matrix and are eluted early. Smaller particles that travel through the porous matrix have a longer residence on the column and elute later, thereby enacting a
15 size-based separation. As in ion-exchange chromatography, proper selection of the chromatography matrix, buffers used for eluting the sample, the geometry and size of the separation column, and other factors must be optimized to achieve a desirable separation outcome.

Both ion-exchange chromatography and SEC require the presence of
20 salts and/or buffers in the elution fluids. The presence of anions or cations is particularly unavoidable in the case of ion-exchange chromatography where the entire separation is based on the displacement of analytes from the chromatography matrix with an ion. However, even in reversed phase chromatography where the elution is generally performed with an organic
25 solvent and salts are usually not required, there are many cases where the separation may be improved through the addition of certain salts or other compounds to the wash or elution solvents.

Mass spectrometry (MS) is an important analytical technique with applications including research, drug discovery, environmental testing,
30 forensics, quality control, and many others. Mass spectrometry is a mass-selective detector that has the ability to quantitatively detect compounds based on the molecular mass of the analytes. While many different types of mass spectrometry have been described two main basic approaches are

often used, namely quadrupole and time-of-flight (TOF). Many variations on both of these approaches, including hybrid systems comprising both approaches, have been developed. In all forms of mass spectrometry, the analytes of interest must be ionized and transferred to the gas phase. There are a large number of different methodologies that have been employed to achieve this, but most modern systems rely on one of two basic approaches. One approach is atmospheric pressure ionization (API), which is further divided into electrospray ionization (ESI) and atmospheric pressure chemical ionization (APCI). The other approach is matrix-assisted laser desorption ionization (MALDI).

The various approaches to MS (quadrupole vs. TOF) and sample ionization (API vs. MALDI) have their various strengths and weaknesses for specific applications. The one constant in all approaches, however, is that MS is not compatible with analytes that are in solutions that contain high ionic strength, such as those that contain high concentrations of salts or buffers. The presence of high concentrations of ions results in a well-documented phenomenon known as ion suppression. Ion suppression causes the analyte of interest to be ionized inefficiently due to the confounding effect of the high concentration of non-specific ions. A second problem with salts and buffers is that many are not volatile. As a result, the salts tend to deposit on the interior surfaces of the MS source region and will degrade instrument performance until eventually the system is no longer operational.

The incompatibility of MS with samples that contain salts and ions together with the need for salts and ions in separation systems such as ion exchange and most size-exclusion chromatography applications means that the two techniques can not be directly interfaced. There are many cases where it is very advantageous to be able to analyze samples separated by ion-exchange or size exclusion chromatography by MS. Currently the only way by which chromatography systems which require salts or buffers can be interfaced with MS is to collect the eluate from the chromatography in fractions. The fractions are then desalted with a secondary separation process that does not require salts, typically a technique using a reversed-phase chromatography system. Since a large number of fractions may be

collected to maintain the fidelity of the separation process, the secondary desalting process is typically carried out using a fast system such as solid-phase extraction (SPE) and may be performed in parallel (e.g. with the use of a 96-well SPE plate). It may be necessary to concentrate the eluate from the SPE process to increase the concentration of the analyte(s) to achieve the required sensitivity. The purified, concentrated samples are then analyzed serially with the appropriate MS system. This extends the time and cost associated with the analysis.

One embodiment of the current invention relates to devices and methods which interface a chromatography system that relies on high ionic strength to achieve separation (such as ion-exchange chromatography) with mass spectrometry. The invention provides a direct and fully automated connection between the chromatography system and the MS and eliminates the labor-intensive steps of collecting fractions from the chromatography system, enacting a parallel purification with SPE, sample concentration.

The eluate from the chromatography system is connected to an injection valve. The injection valve is used to capture an aliquot of the eluate from the chromatography system and to divert it to a fast and automated sample purification system, such as the RAPIDFIRE® system, available from BioTrove, Inc. of Woburn, Massachusetts, which has been described U.S. Patent 6,309,600 to Hunter, U.S. Patent Publication 2002/0001544 of Hess, *et al.*, U.S. Patent Publication 2003/0119193 of Hess, *et al.*, U.S. Patent Publication 2005/0123970 of Ozbal, *et al.*, U.S. Patent 6,812,030 to Ozbal, *et al.*, U.S. Patent 6,932,939 to Ozbal, *et al.*, and U.S. Patent Publication 2005/0194318 of Ozbal, *et al.*. The contents of the above patents and publications are each incorporated here in its entirety by reference. The RAPIDFIRE® high throughput mass spectrometry system is capable of solid-phase extraction based purification at throughputs on the order of 5 seconds per sample. With such a system it is possible to take a mass spectrometric reading of the eluate from the chromatography system every 5 seconds. The remaining sample may be collected in fractions for additional analysis or further fractionation.

The general layout of the system is shown in FIG. 10. An injection system 1002 is directly connected to the chromatography system 1004 (e.g. a high-pressure liquid chromatography system). Optionally, the chromatography system 1002 may have an optical detector 1006 immediately after the chromatography column to monitor and quantify the analytes as they elute from the chromatography column. In one embodiment of the invention, a diversion valve 1008 is placed after the optical detector 1006 that may be used to direct the eluate from the chromatography column away from the downstream instrumentation. The diversion valve 1008 may be electronically controlled by the optical detector 1006 such that if certain signal criteria are met the valve will be actuated. One application of the diversion valve 1008 may be to divert chromatography eluate to waste 1010 if the concentration of analytes is too high in order to protect the injection system 1002 and/or mass spectrometry system 1012 from contamination.

In one embodiment of the invention, starting a separation with the chromatographic system generates a trigger signal (e.g. a TTL (transistor-transistor logic) pulse) that is detected by the injection system, the mass spectrometer, the optical detector, and the fraction collector and is used to synchronize the start of all of the devices. The electronic communication circuitry between the various components of the preferred embodiment of the invention is shown as dashed lines in Figure 10.

Referring now FIG. 11(a), the invention includes two fluidic injection valves 1102, 1104 and two high pressure fluidic pumps 1106, 1108. The first pump 1102 is used to flow an aqueous wash solution while the second pump 1104 is used to flow an organic elution solution over a SPE cartridge 1110. The details of the SPE cartridge 1112 and its application in sample purification for high-throughput mass spectrometry have been described previously. See, e.g., Nigel J.K. Simpson, Solid Phase Extraction: Principles, Strategies & Applications (2000); E.M. Thurman & M.S. Mills, Solid-Phase Extraction: Principles & Practices (1998). The eluate from the chromatography system 1004 is connected directly to one port of fluidic injection valve 1102 as shown in FIG. 11(a). Another tube connected to a second port of the same valve 1102 is used to carry the eluate to waste or to a fraction collector 1014,

depending on the application. The aqueous fluid from pump 1106 is flowed over the SPE cartridge 1110 in a first direction to condition and equilibrate the cartridge 1110. In the meantime, the organic solvent from pump 1108 is flowed directly to the source of an API-MS to establish a stable spray in ESI or APCI mode. This initial fluidic circuit is shown in FIG. 11(a).

Referring now to FIG. 11(b), once the chromatographic separation is begun, valve 1102 is electromechanically actuated to the position shown in FIG. 11(b). In this position the eluate from the chromatography system 1002 is diverted over an injection loop 1114.

Valve 1102 is actuated a second time to the position shown in FIG. 11(c) after enough time has been allowed to ensure that the injection loop 1114 is completely full of sample. The determination of the amount of time before valve 1102 is actuated the second time can be determined by calculating the flow rate of the chromatography eluate and the volume of the injection loop 1114. For example, if the chromatography solvent is pumped at 0.6 mL/min and a 10 μ L injection loop is used, the system will require 1 second to completely fill the injection loop ($0.6 \text{ mL/min} = 10 \text{ } \mu\text{L/sec}$). Alternatively, an optical sensor can be coupled with tube 1112 as described above.

Reactuation of valve 1102 after an aliquot of sample has been allowed to fill the loop 1114 as shown in FIG. 11(c) will result in the sample being pushed from the loop 1114 onto the SPE cartridge 1110. The analytes of interest (*e.g.* proteins or oligonucleotides) will adsorb on the SPE cartridge 1110 while the salts and other ions used in the chromatographic separation will pass through the cartridge 1110 and will be collected in a waste container 1116. Waste container 1116 and 1014 can, in various embodiments, be the same or separate vessels.

After approximately 10 SPE column volumes of wash solution have been flowed over the SPE cartridge 1110, valve 1104 is actuated to the position shown in FIG. 11(d). In this position, the organic solvent from the second pump 1108 is diverted over the SPE cartridge 1110 in the opposite direction to the sample loading and washing. The purified and desalted analyte(s) of interest are solubilized by the organic solvent, desorbed from the

SPE cartridge 1110, and flowed onto the API-MS 1118 for mass spectrometric analysis. Typically, 10 SPE column volumes of organic solvent are sufficient to achieve a near complete elution of analytes. In a preferred embodiment of the invention, the timing of the wash and elution steps adjustable and may be optimized for each specific application. In typical applications, a SPE cartridge with a 4.0 μL column bed volume is used. At a flow rate of 1.2 mL/min (or 20 $\mu\text{L}/\text{sec}$) 10 column volumes of wash or elution solvent can be delivered to the SPE cartridge 1110 in as little as 2 seconds.

After the analytes have been delivered to the MS 1118, valve 1104 is actuated again to the position shown in FIG. 11(e). This is the initial position of the fluidic system and facilitates the reconditioning and equilibration of the SPE cartridge 1110. In a preferred embodiment of the invention, the cycle shown in FIGS. 11(a) through 11(e) can be repeated at a rate that is selected by the user through a software interface. For example, the user may select a desalting and MS analysis cycle of 10 seconds. Longer cycle times may be selected, however the minimum cycle time will depend on the timing of the valve cycles. Typical peak widths from ion-exchange or size-exclusion chromatography systems are in the 10-30 second range meaning that at least one MS analysis will be available for every peak that is eluted from the chromatography system 1002. Following from the previous example of a 10 μL injection loop and a HPLC flow rate of 0.6 mL/min, 10 $\mu\text{L}/\text{sec}$ will be flowed over valve 1102 in the desalting apparatus. If 10 μL of sample is removed via the injection loop at a rate of once per every 10 seconds, a total of 10% of the total eluate from the HPLC column 1002 will be diverted for rapid desalting and MS analysis. The remaining 90% of eluate may be collected in a fraction collector or disposed of in a waste container depending on the application.

At the conclusion of the HPLC experiment, a second electronic signal will direct the injection system 1002, the mass spectrometer 1012, the optical detector 1006, and the fraction collector 1014 to switch to a standby mode. At the end of the chromatographic separation, the user will have a continuous optical trace (such as a UV chromatogram) obtained from the optical detector 1006, a series of non-continuous MS data at the cycling frequency selected, and a series of fractions collected by the fraction collector. There will be an

offset between the optical detector and the MS data based on the internal volume of the tubing between the optical detector and the MS source and the wash time selected in the injection system 1002 valving. The offset may be calculated or empirically determined, however, once the offset is known it will
5 be possible to correctly align the MS data with the optical data and the appropriate fraction.

With this invention, it is possible to directly collect MS data from a separation system 1004 that contains MS-incompatible buffers without needing to perform labor intensive and time consuming steps of fraction
10 collection and off-line sample preparation.

FIG. 12 illustrates a method of processing of an eluted sample from a liquid chromatography system in a mass spectrometry device. In step 1202, a flow of a non-polar solvent to the mass spectrometry device is initiated. In step 1204, an eluted sample is received from the liquid chromatography
15 system. The eluted sample is flowed over an SPE column in step 1206. In step 1208, the SPE column is washed with a polar solution. In step 1210, a non-polar solvent is flowed over the SPE column. In step 1212, the non-polar solvent and the eluted sample are presented to the mass spectrometry device.

20 The systems and methods described above can also be reversed so such that an eluted sample is non-polar, while the wash solvent is polar. In such an embodiment, the column can be a HILIC (Hydrophilic Interaction Liquid Chromatography) column.

Example 4

25 Cation exchange chromatography is used in the quality control step of the manufacturing of a pharmacologically active protein. For each lot that is manufactured, a 60 minute HPLC separation using an established standard operating procedure must be performed. It is known that the protein of interest elutes off of the HPLC column between 28 and 30 minutes. The
30 entire chromatographic run is monitored by the optical detector 1006 at a wavelength of 220 nm. The HPLC 1004 is run at 0.6 mL/min and a gradient from 0.1M sodium chloride to 1M sodium chloride is used to enact the separation.

If any other chromatographic peaks are detected other than the main protein itself it is possible that these are contaminants or breakdown products. Because this protein is meant to be administered to patients, it is required that a full characterization of all potential contaminant peaks be completed before the lot can be approved. Traditionally this characterization would involve collecting fractions from the HPLC, performing a sample preparation step to remove the MS-incompatible salts, and running the MS measurement. Typically, an aliquot of the fraction must also be re-injected in the HPLC separation to ensure that the correct fraction was used in the MS characterization.

The present invention eliminates many of the time consuming and labor intensive steps described above. The device 1002 is placed between the optical detector 1006 and the fraction collector 1014 as shown in Figure 10. A 10 μ L injection loop 1114 is used along with a SPE cartridge 1110 that contains a polymeric matrix with a 4.0 μ L bed volume. Pump 1102 is used to deliver a wash solvent consisting of water with 0.02% trifluoroacetic acid while pump 1104 is used to deliver an elution solvent of 80% acetonitrile with 0.02% trifluoroacetic acid. A time of 1 second is selected to completely fill the injection loop 1114, 2 seconds to wash the salts away from the analytes, 2 seconds to elute the analytes off of the SPE cartridge 1110, and 1 second to fully recondition the SPE cartridge 1110. For this application it is decided to run the system at the fastest cycle time, which is 6 seconds.

When the cation exchange HPLC run is initiated, a TTL pulse also triggers the start of the UV detector 1006, the MS 1012, the injection system 1002, and the fraction collector 1014. Over the 60 minute, run a total of 600 high throughput mass spectrometry system cycles will be performed (3600 seconds at 6 seconds/cycle = 600 cycles). At the end of the experiment, the MS data, consisting of a time trace with 600 injections can be aligned with the optical detector either through the identification of a landmark (e.g. the main protein in the assay) or through the calculation of the delay within the system. Using the invention described, the equivalent experiment to collecting and preparing 600 individual fractions from the HPLC can be performed in a

completely automated fashion and obviate the need for any additional validation experiments.

The embodiments of the invention described herein can be controlled by a variety of electronic devices including hardware and software as is known to those of skill in the art. Electrical-mechanical components such as valves 206, 207, 601, 602, 603, 1102, 1104 can be controlled according to interfaces described in related literature can communicate with control devices according to a variety of standard and proprietary technologies and protocols including, but not limited to, transistor-transistor logic (TTL), serial, parallel, FireWire, USB, Ethernet, and the like.

The functions of several elements may, in alternative embodiments, be carried out by fewer elements, or a single element. Similarly, in some embodiments, any functional element may perform fewer, or different, operations than those described with respect to the illustrated embodiment. Also, functional elements (*e.g.*, modules, databases, computers, clients, servers and the like) depicted as distinct for purposes of illustration may be incorporated within other functional elements, separated in different hardware or distributed in a particular implementation.

While certain embodiments according to the invention have been described, the invention is not limited to just the described embodiments. Various changes and/or modifications can be made to any of the described embodiments without departing from the spirit or scope of the invention. Also, various combinations of elements, steps, features, and/or aspects of the described embodiments are possible and contemplated even if such combinations are not expressly identified herein.

INCORPORATION BY REFERENCE

The entire contents of all patents, published patent applications, and other references cited herein are hereby expressly incorporated herein in their entireties by reference.

CLAIMS

1. A method of preparing an eluted sample containing salts or buffers from a liquid chromatography device for analysis by a mass spectrometry device, the method comprising:
 - 5 continuously providing a non-polar solvent to the mass spectrometry device;
receiving the eluted sample from the liquid chromatography device;
flowing the eluted sample over a solid phase extraction column;
flowing the non-polar solvent over the solid phase extraction column;
 - 10 and
presenting the non-polar solvent and the eluted sample to the mass spectrometry device.
2. The method of claim 1, wherein the liquid chromatography device is an
15 ion exchange chromatography device.
3. The method of claim 1, wherein the liquid chromatography device is a cation exchange chromatography device.
- 20 4. The method of claim 1, wherein the liquid chromatography device is a size exclusion chromatography device.
5. The method of claim 1, further comprising:
optically analyzing the eluted sample from the liquid chromatography
25 device to generate an optical data set.
6. The method of claim 5, further comprising:
associating the optical data set with a data set generated by the mass
spectrometry device.
30
7. The method of claim 1 further comprising:
flowing a polar wash solution over the solid phase extraction column.

8. A sample injection system for coupling a liquid chromatography device with a mass spectrometry device, the system comprising:
a sample injection valve having:
i. a first position that allows sample from the liquid chromatography device to pass through the sample injection system,
5 and
ii. a second position that loads sample from the liquid chromatography device onto a sample supply loop; and
a column control valve configured to facilitate a continuous flow of an
10 elution solvent to a sample analyzer, the column control valve having:
i. a first position that simultaneously delivers the fluidic sample from the sample supply loop to a solid phase extraction column in a first direction and delivers an elution solvent to the sample analyzer, and
15 ii. a second position that flows the elution solvent over the solid phase extraction column in a second direction to deliver the fluidic sample and the elution solvent to the sample analyzer.
9. The system of claim 8, further comprising:
20 an optical detector for analyzing the sample from the liquid chromatography device.
10. The system of claim 8, further comprising:
a diversion valve located between the liquid chromatography device
25 and the sample injection valve.
11. The system of claim 10, wherein the diversion valve is actuated as a result of signal generated by the optical detector.
12. The system of claim 8, further comprising a fraction collector.
13. The system of claim 8, wherein the elution solvent is a polar solvent.

14. The system of claim 8, wherein the elution solvent is a non-polar solvent.
15. A method of preparing an eluted sample from a liquid chromatography device for analysis by a mass spectrometry device, the method comprising:
5 continuously providing a polar solvent to the mass spectrometry device;
receiving the eluted sample from the liquid chromatography device;
flowing the eluted sample over a HILIC column;
10 flowing the polar solvent over the HILIC column; and
presenting the polar solvent and the eluted sample to the mass spectrometry device.
16. The method of claim 15, wherein the liquid chromatography device is
15 one selected from the group consisting of: an ion exchange chromatography device, a cation exchange chromatography device, and a size exclusion chromatography device.
17. The method of claim 13, further comprising:
20 optically analyzing the eluted sample from the liquid chromatography device to generate an optical data set.
18. The method of claim 17, further comprising:
associating the optical data set with a data set generated by the mass
25 spectrometry device.
19. The method of claim 1 further comprising:
flowing a polar wash solution over the solid phase extraction column.

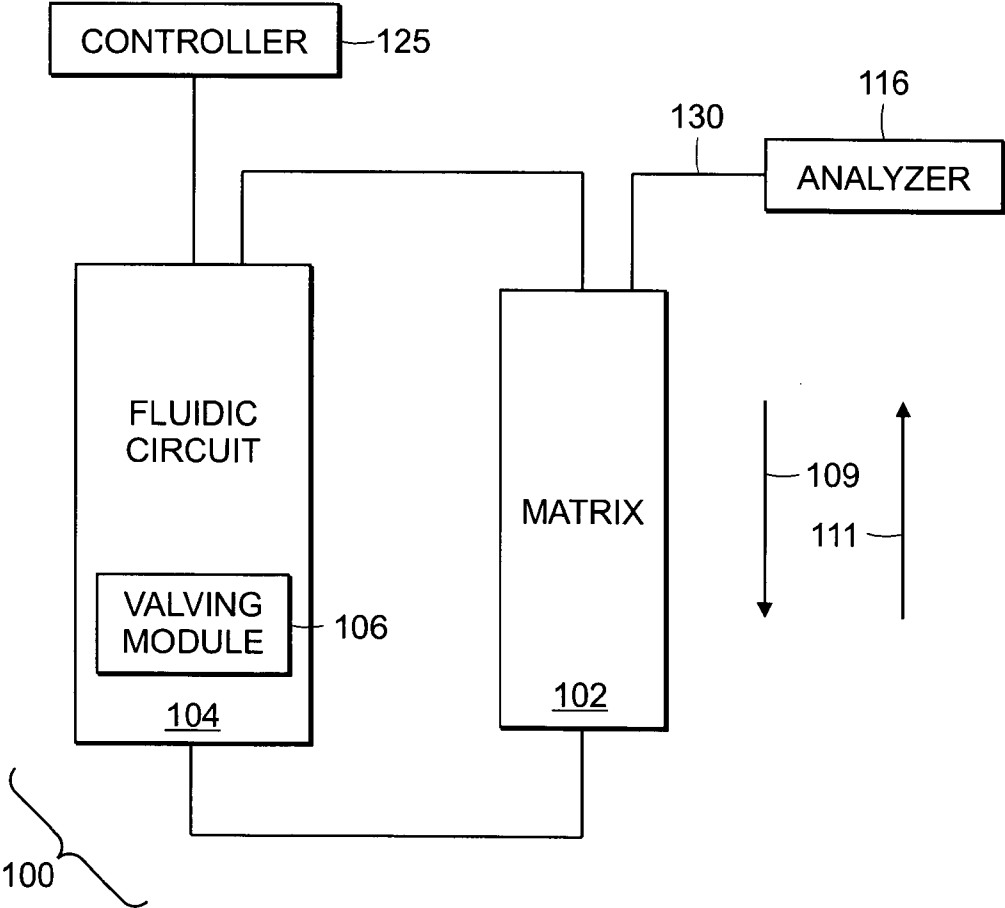


FIG. 1

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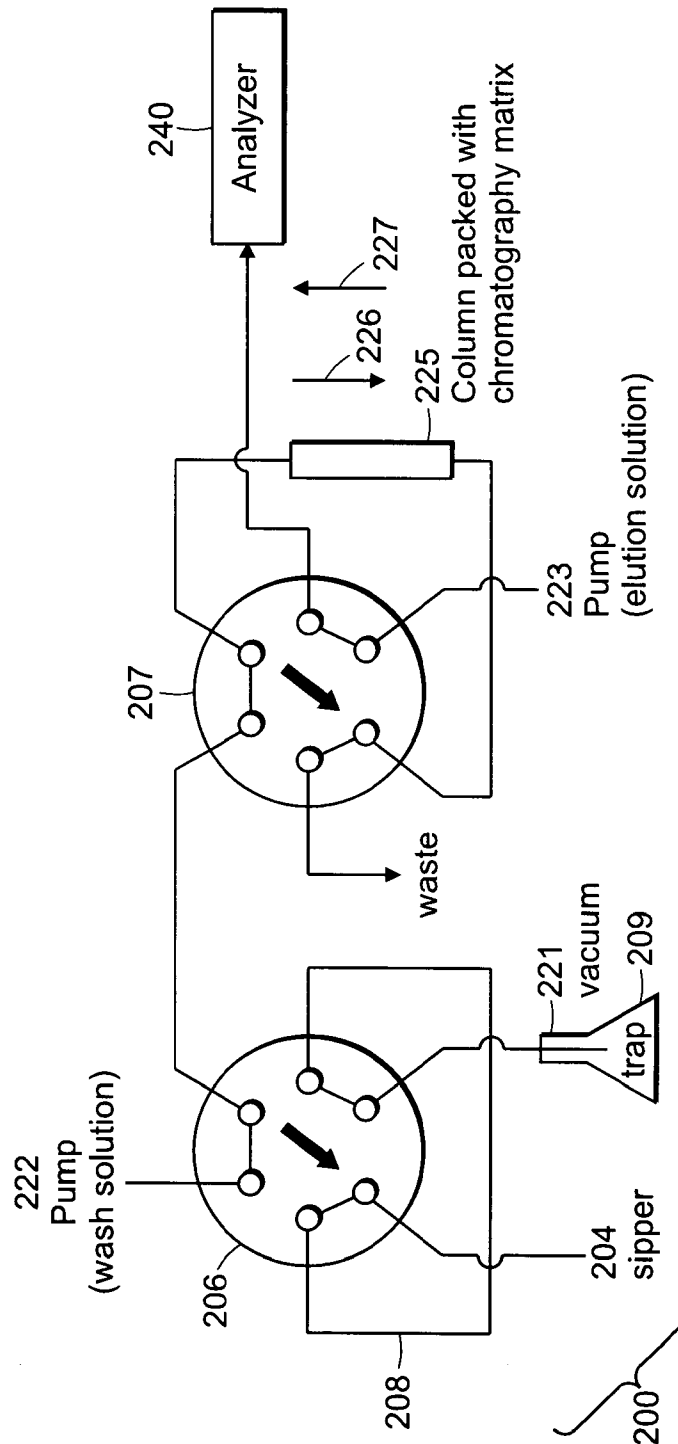


FIG. 2(a)

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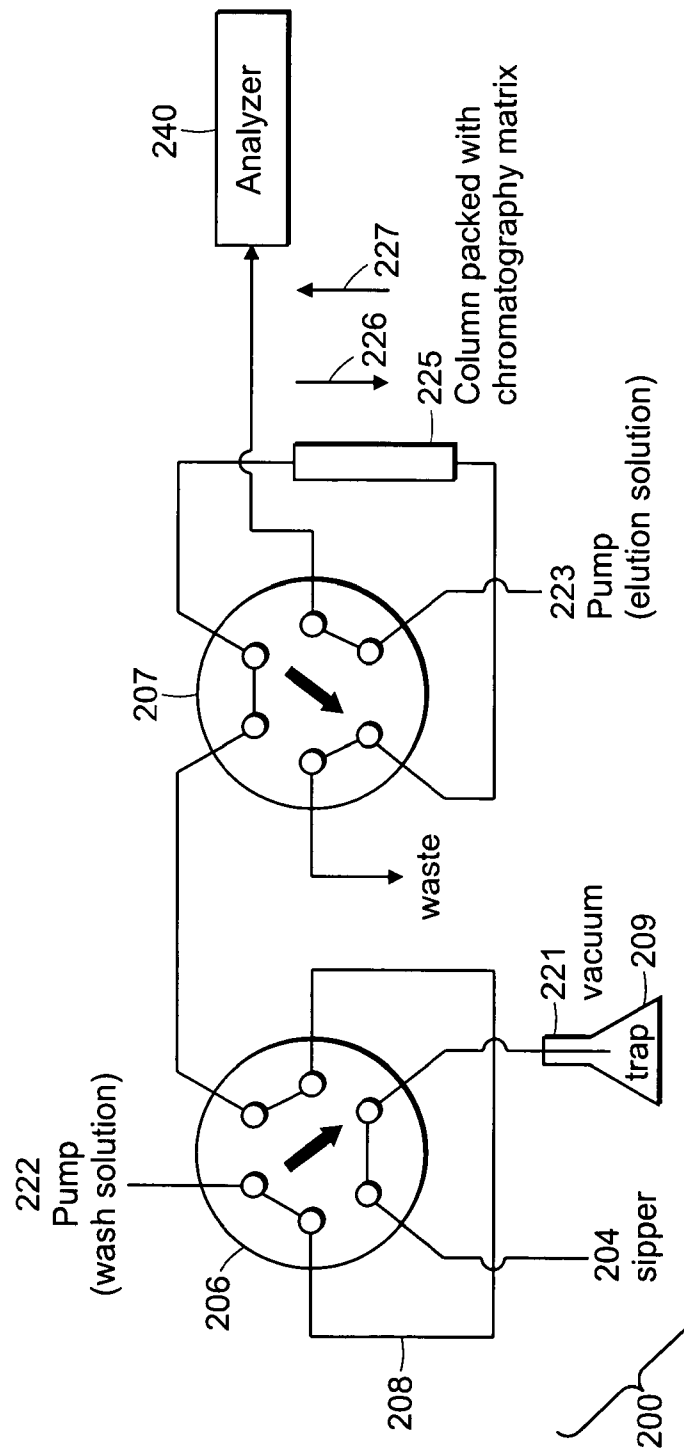


FIG. 2(b)

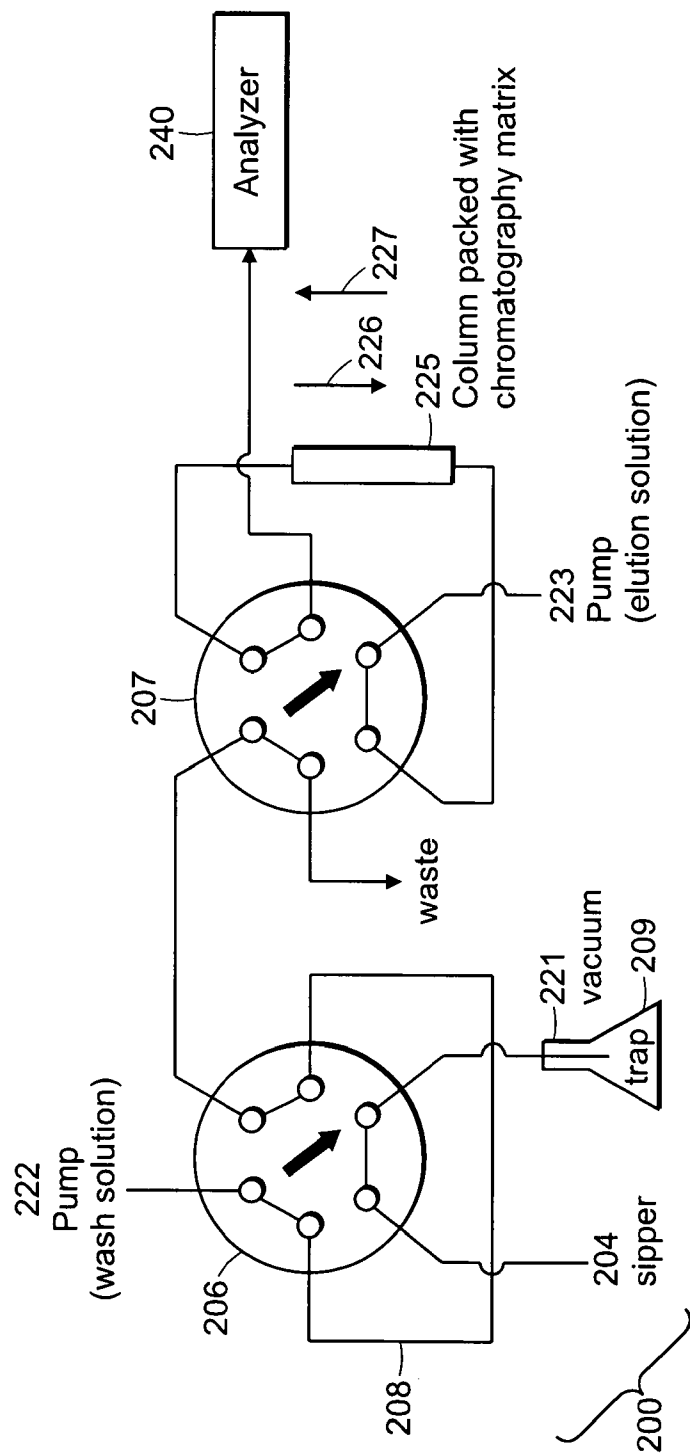


FIG. 2(c)

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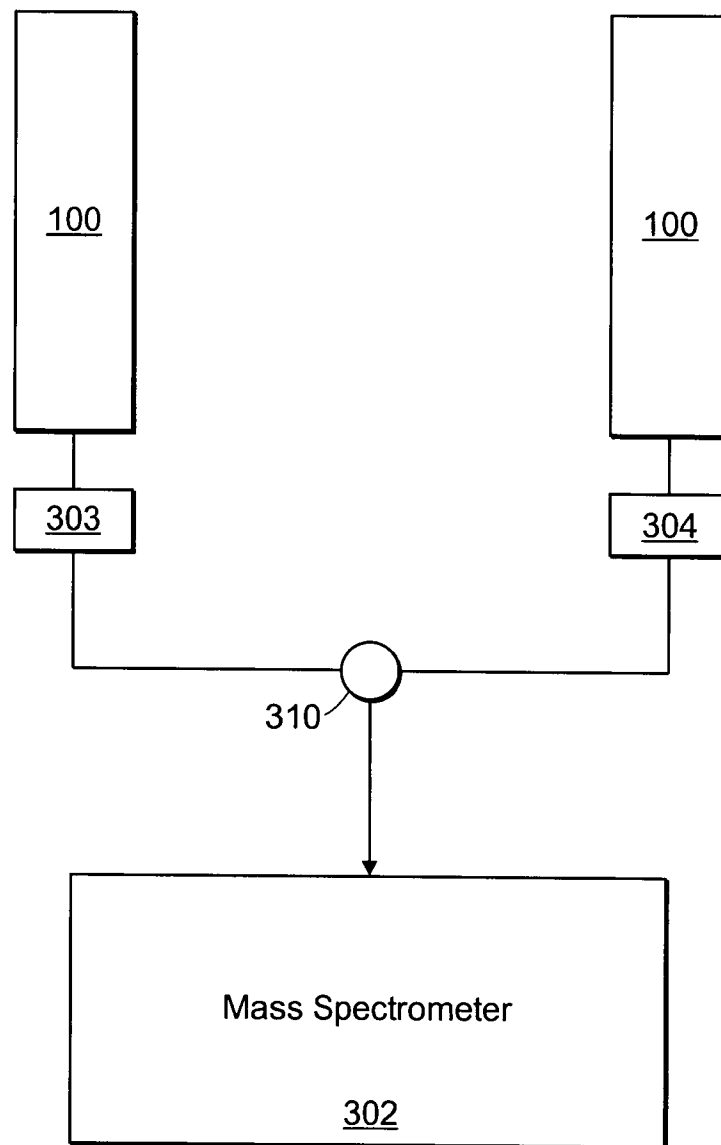


FIG. 3

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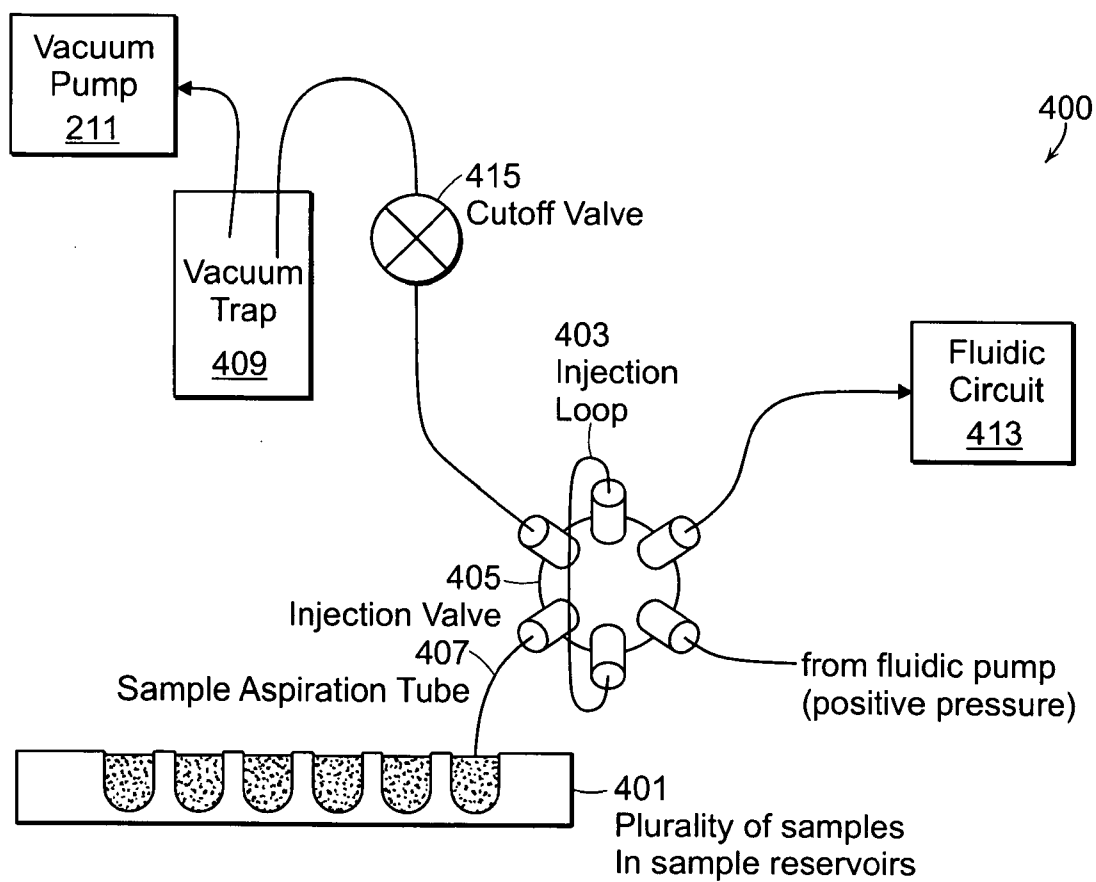


FIG. 4

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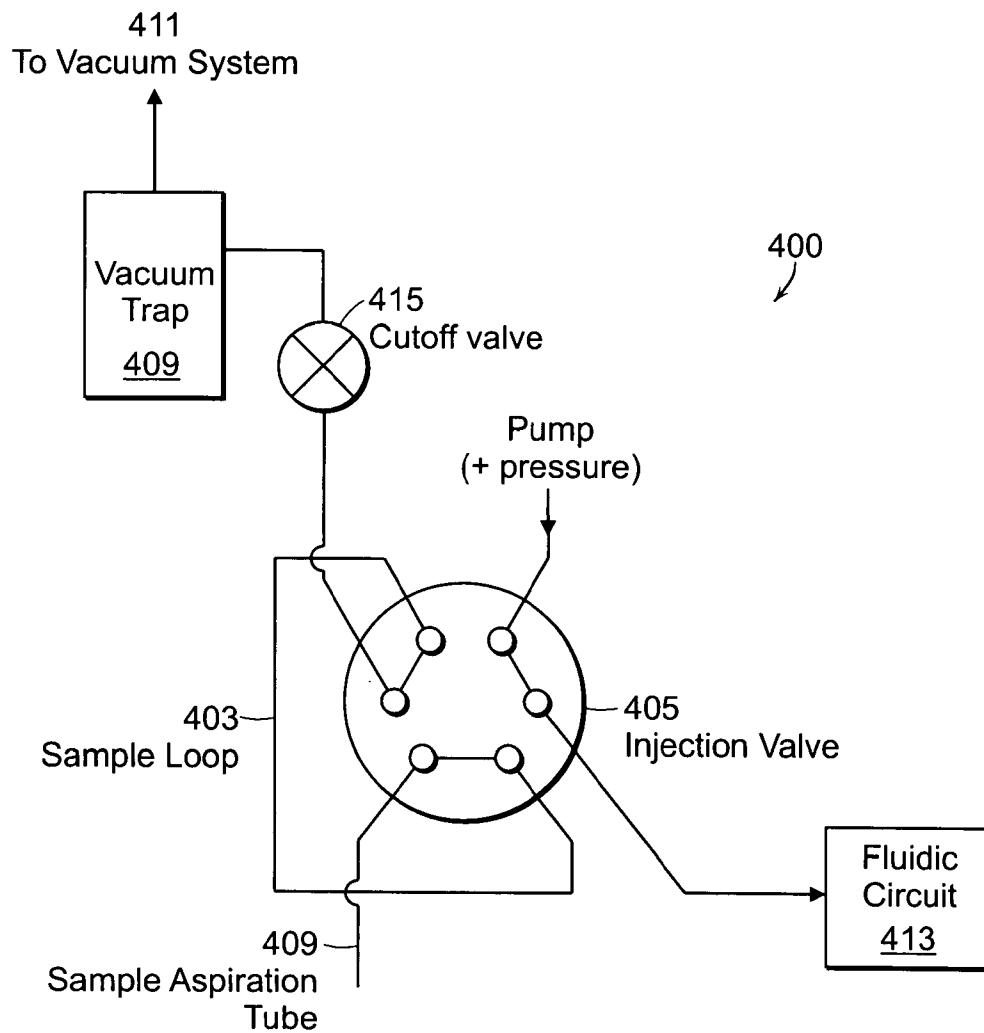


FIG. 5(a)

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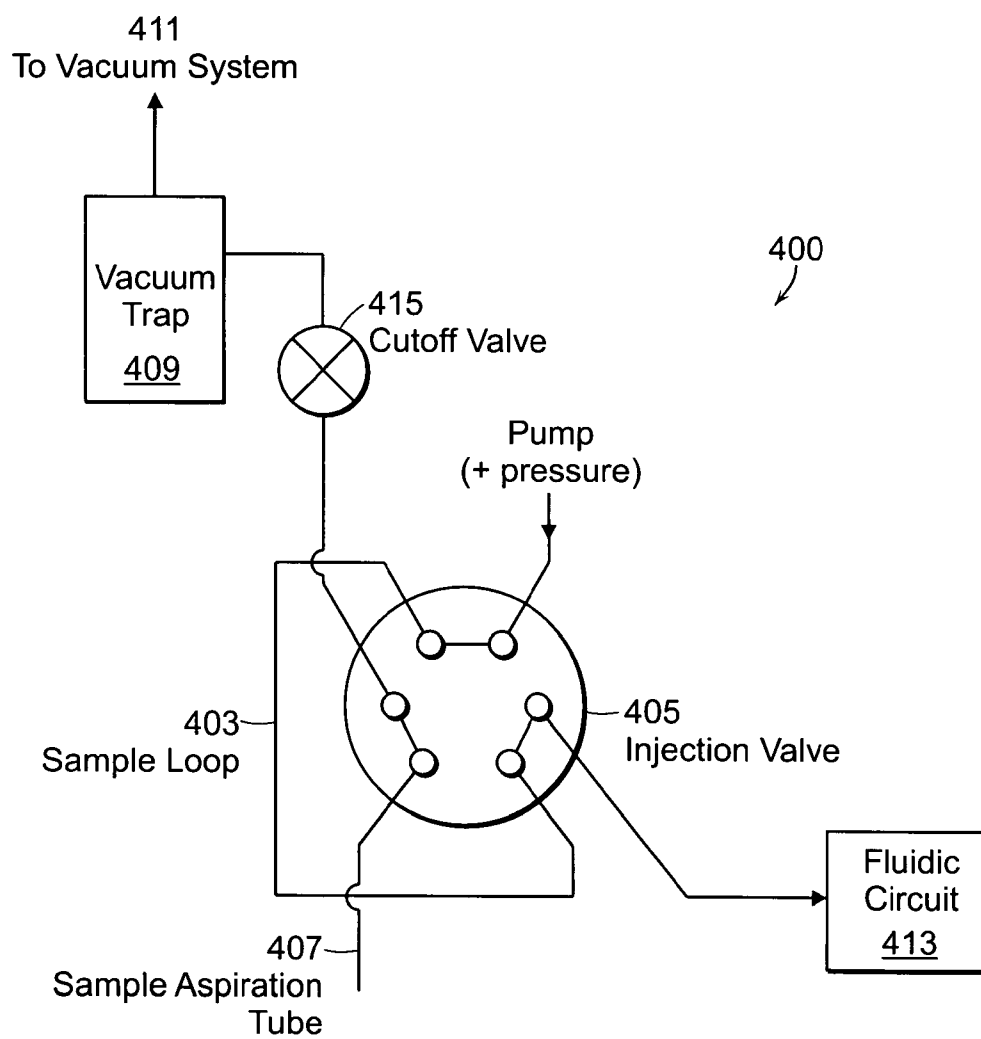


FIG. 5(b)

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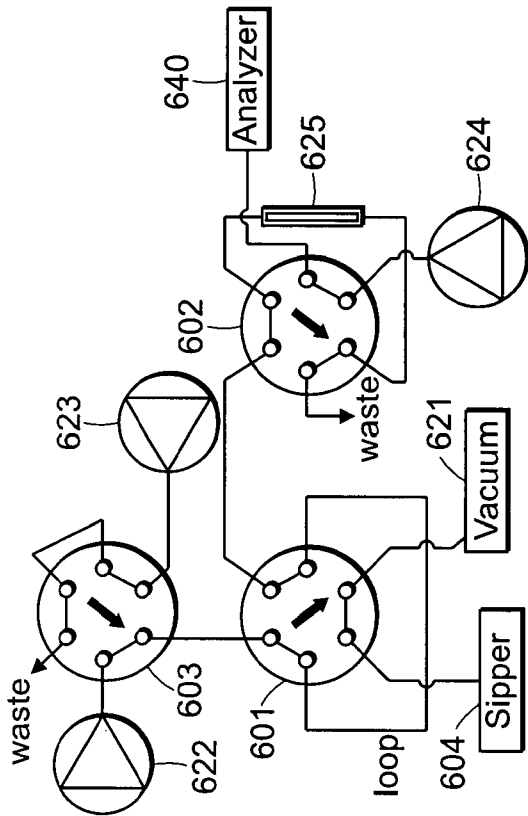


FIG. 6(b)

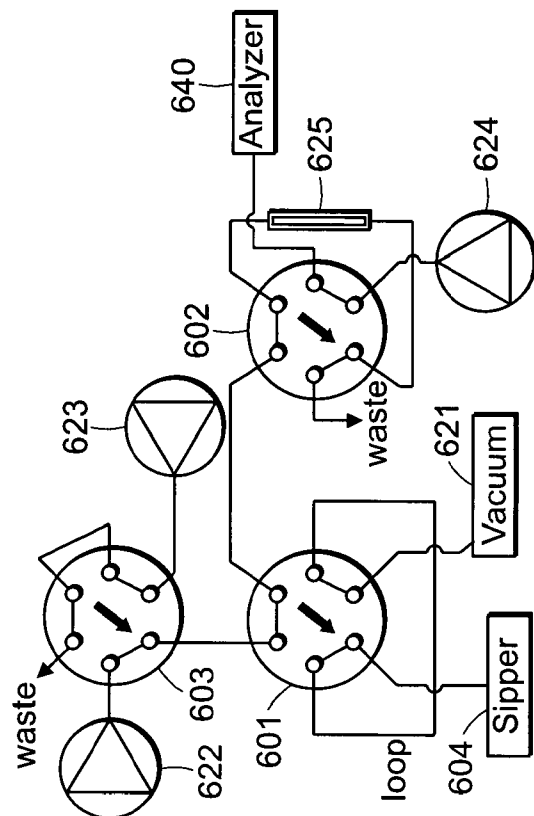


FIG. 6(d)

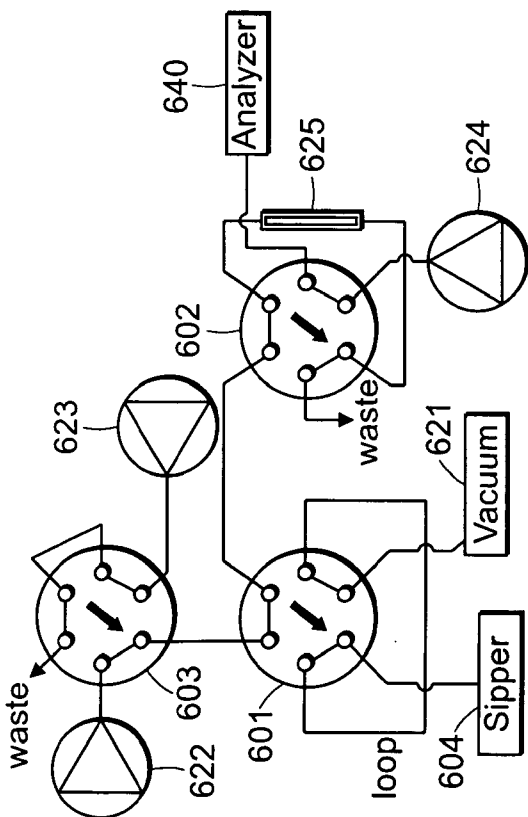


FIG. 6(a)

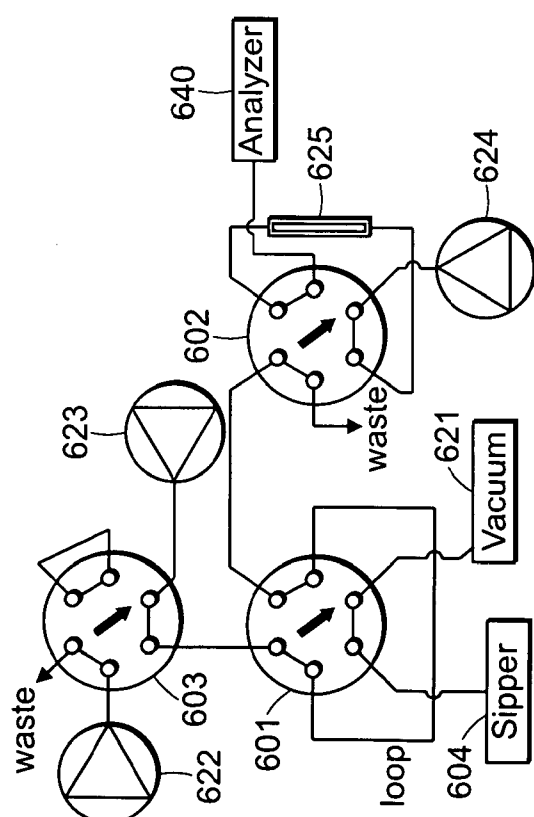


FIG. 6(c)

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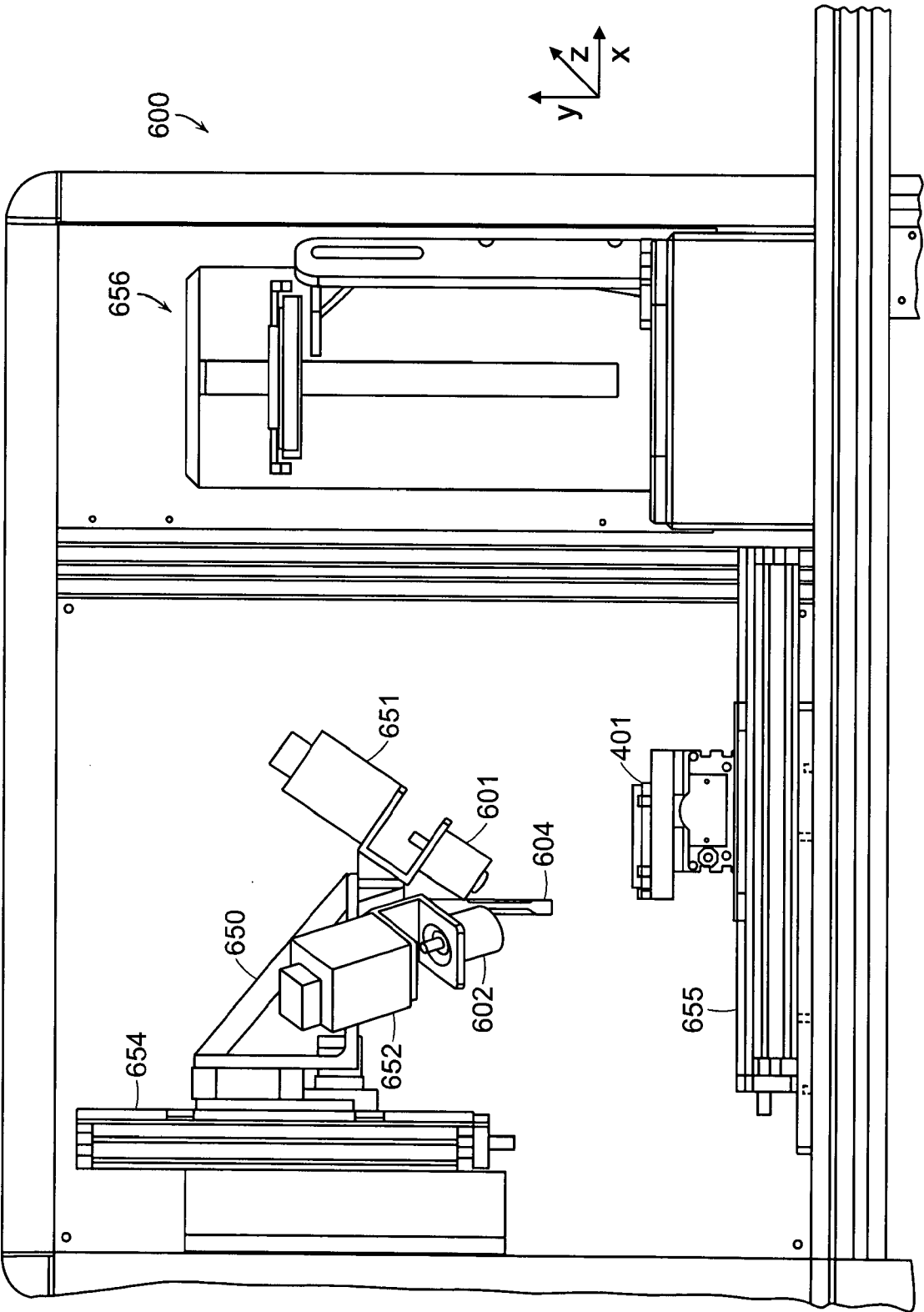


FIG. 6(e)

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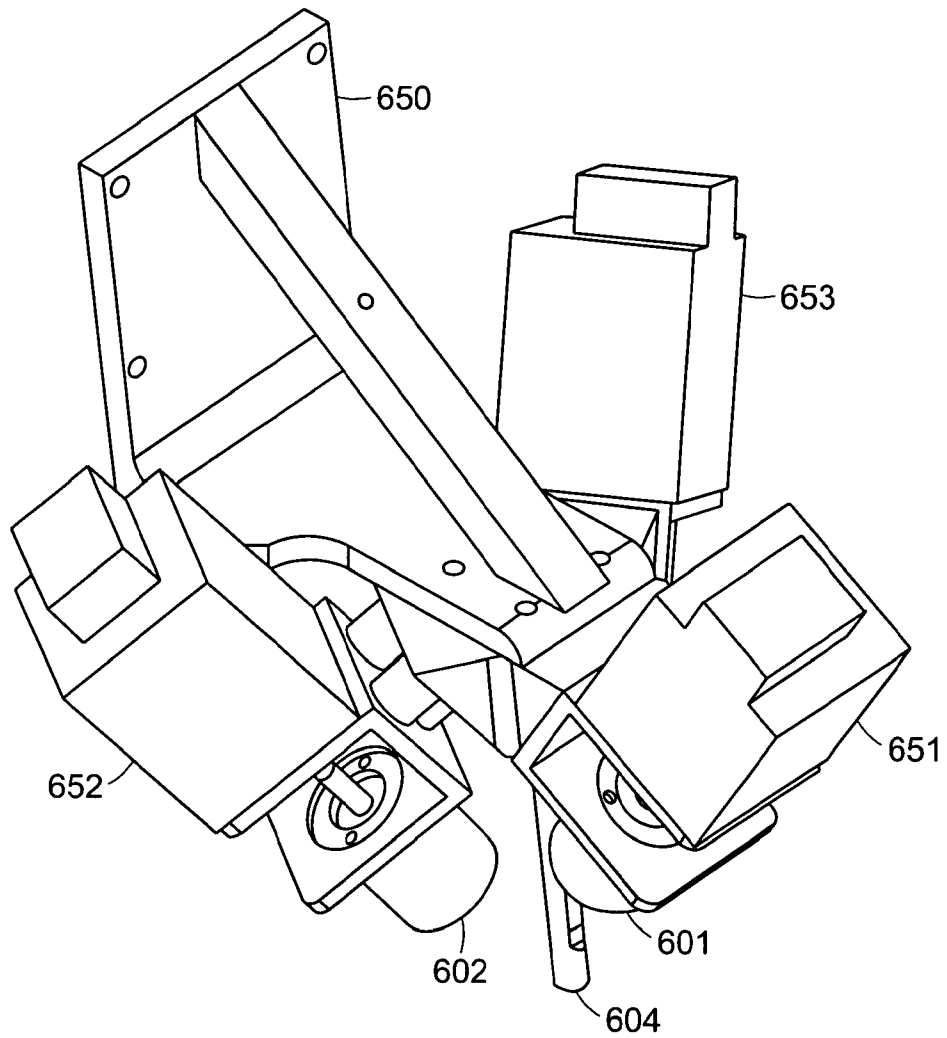


FIG. 6(f)

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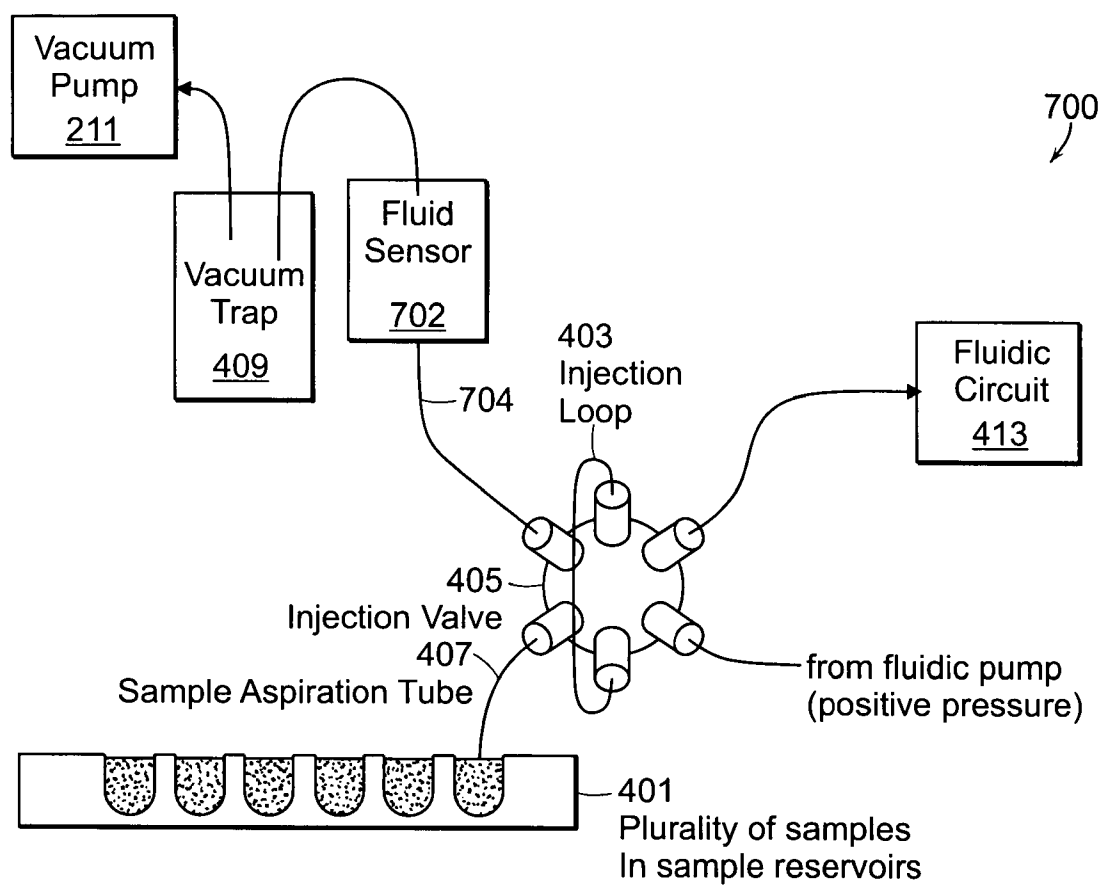


FIG. 7

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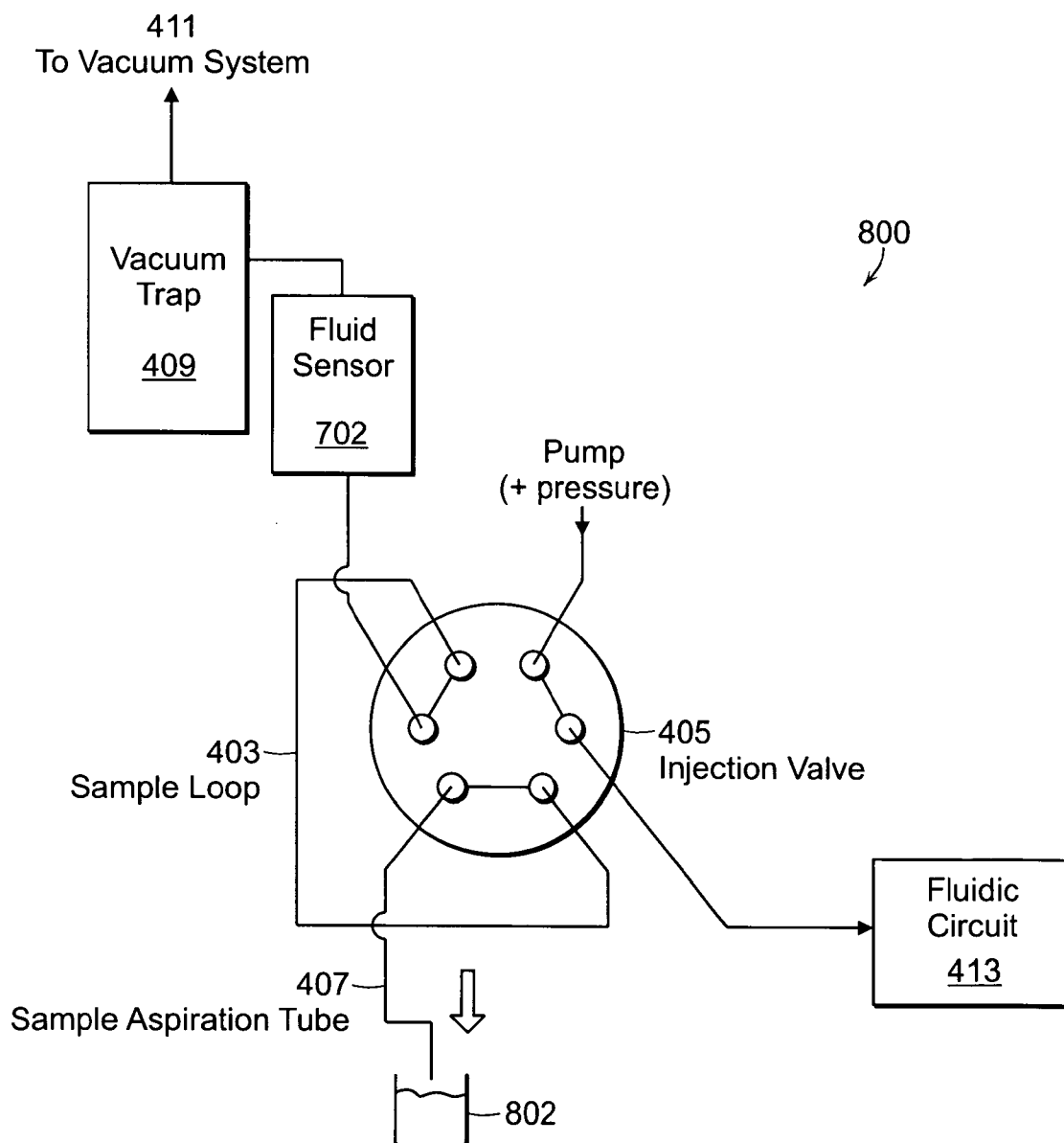


FIG. 8(a)

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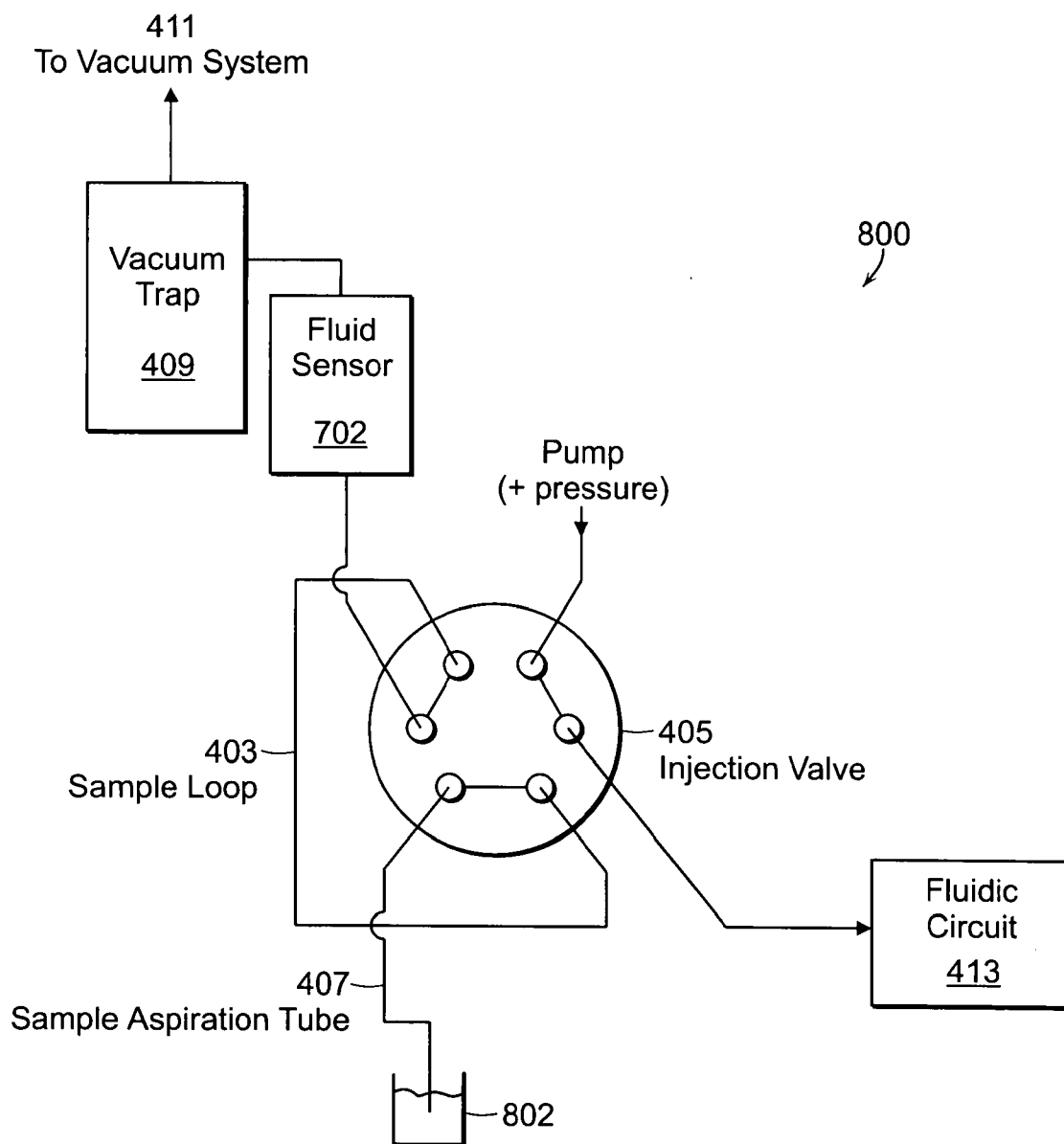


FIG. 8(b)

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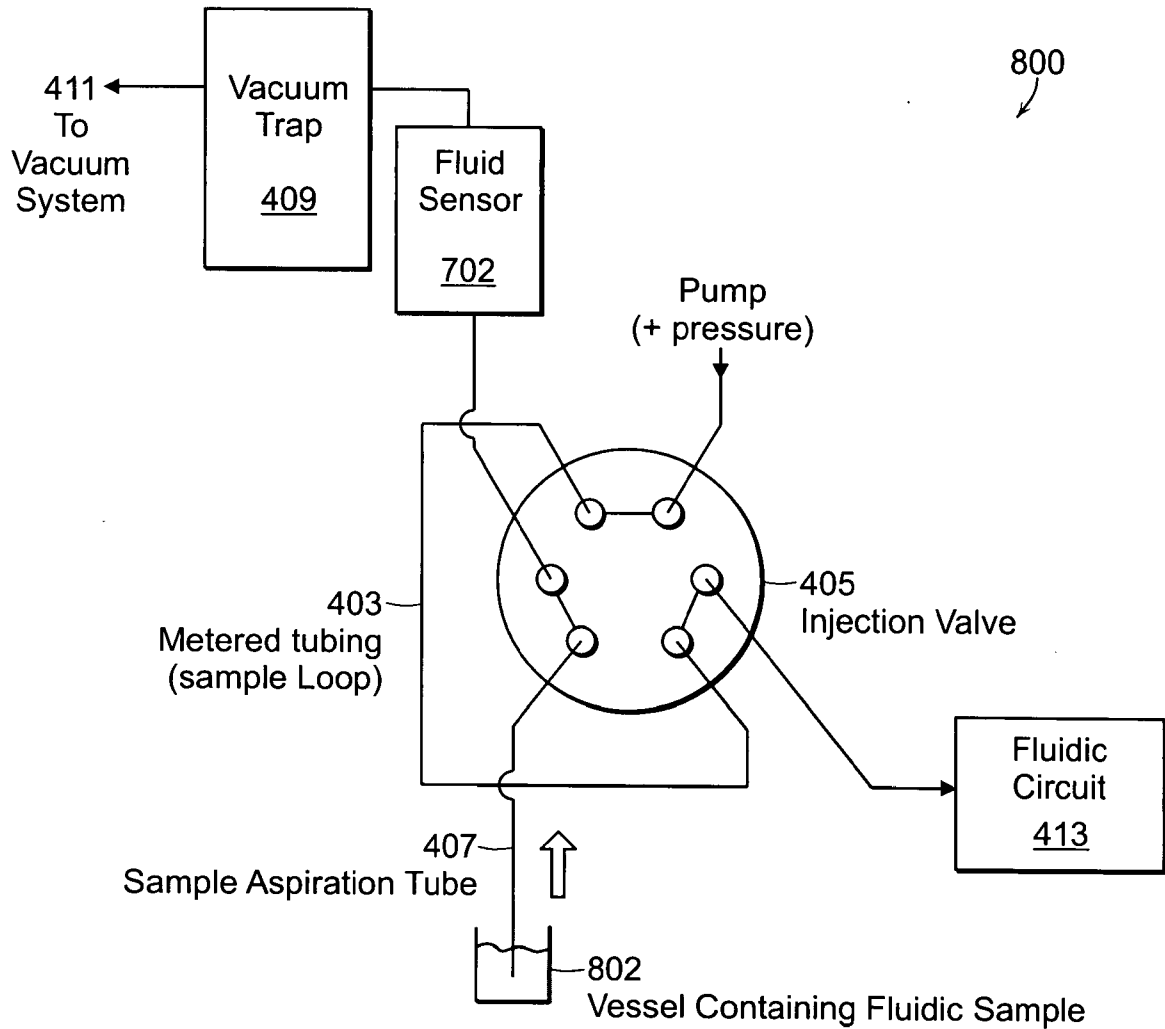


FIG. 8(c)

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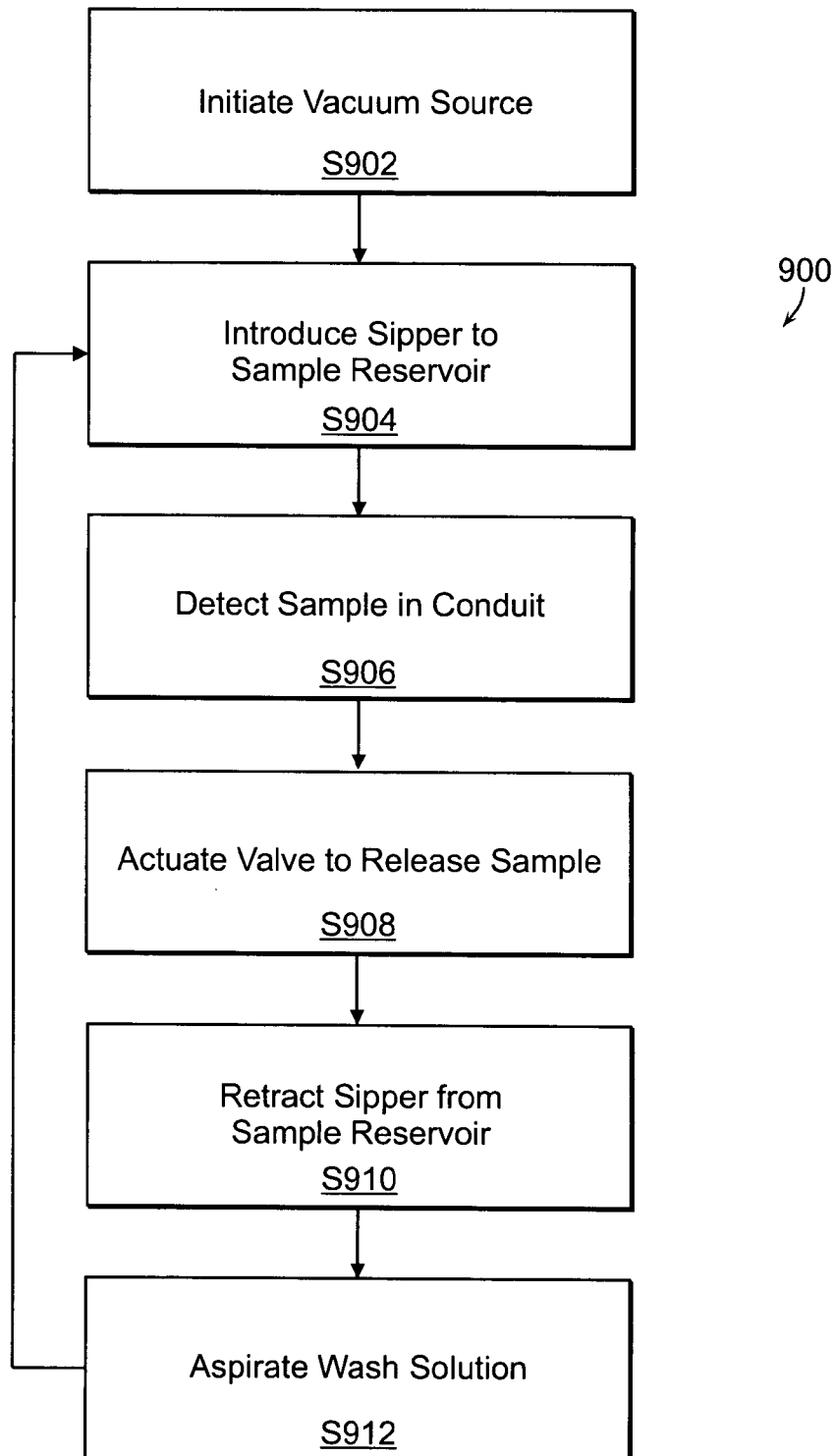


FIG. 9

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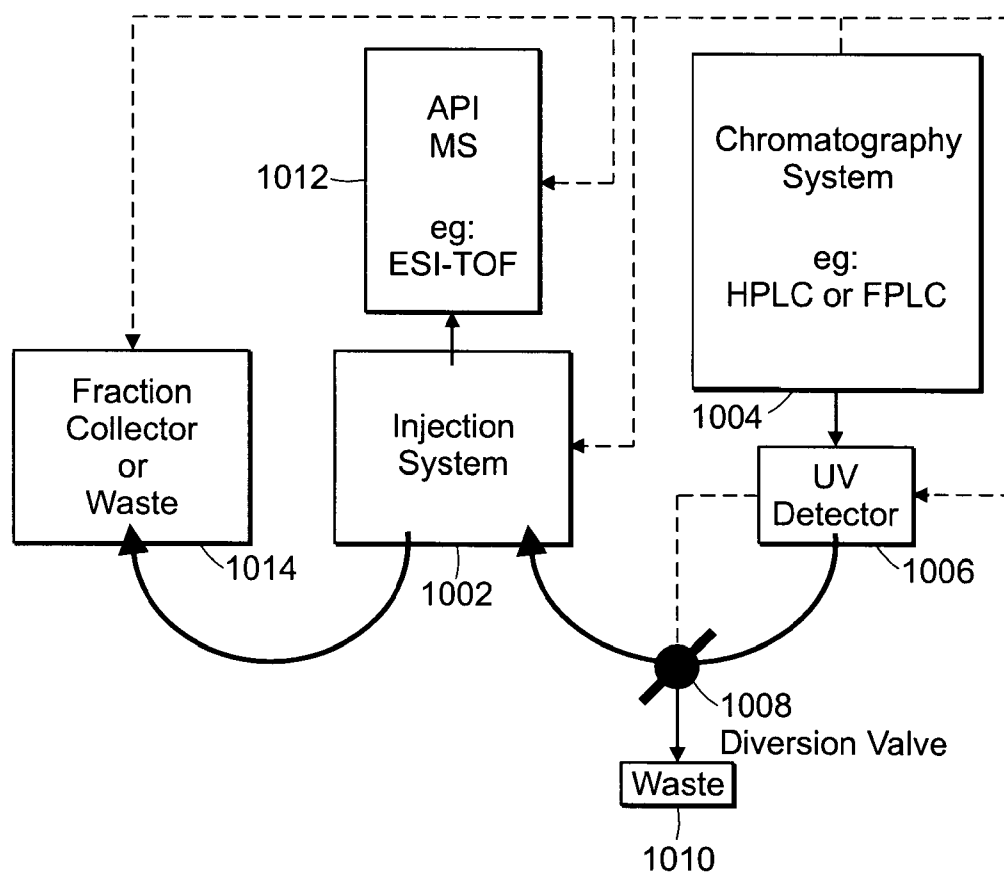


FIG. 10

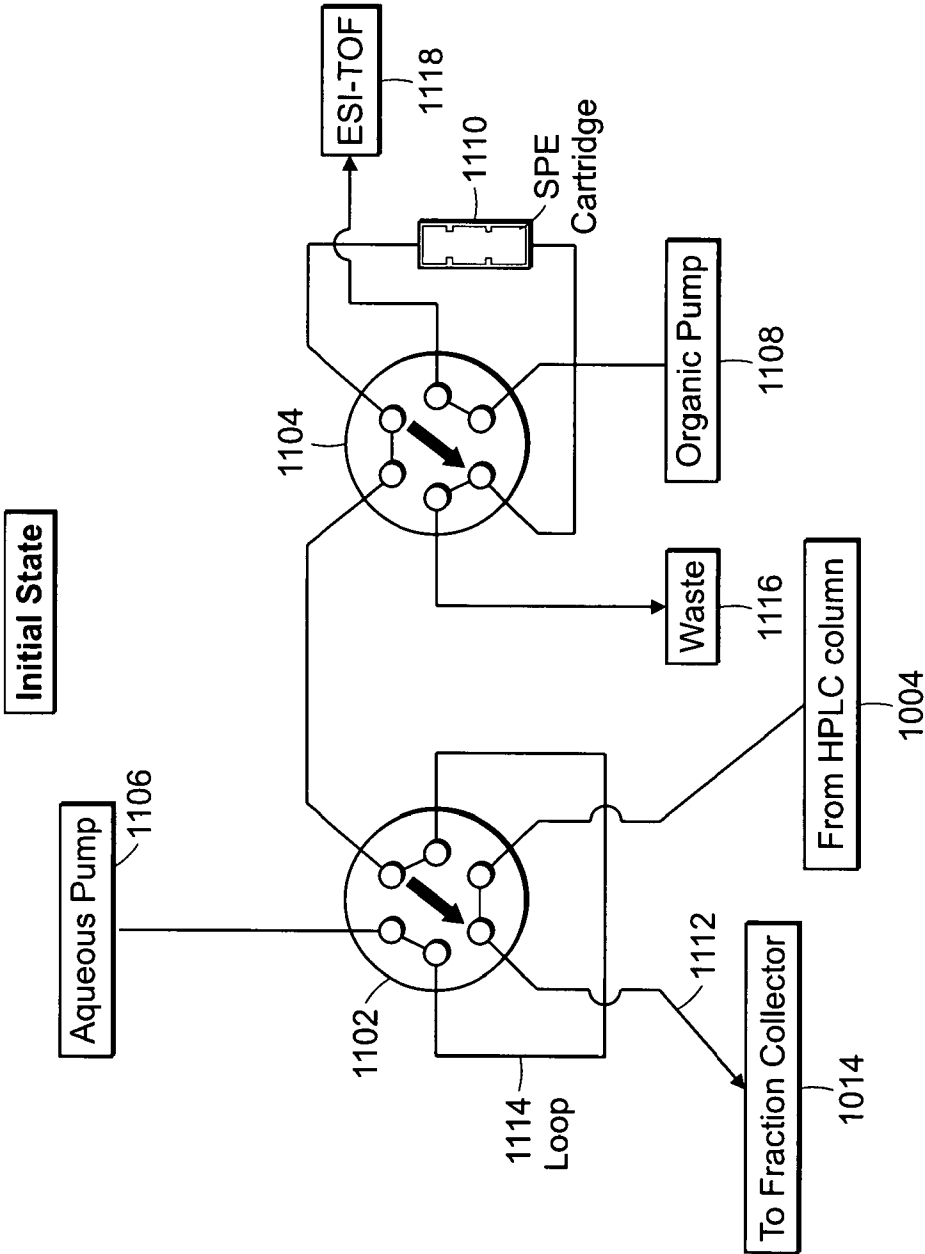


FIG. 11(a)

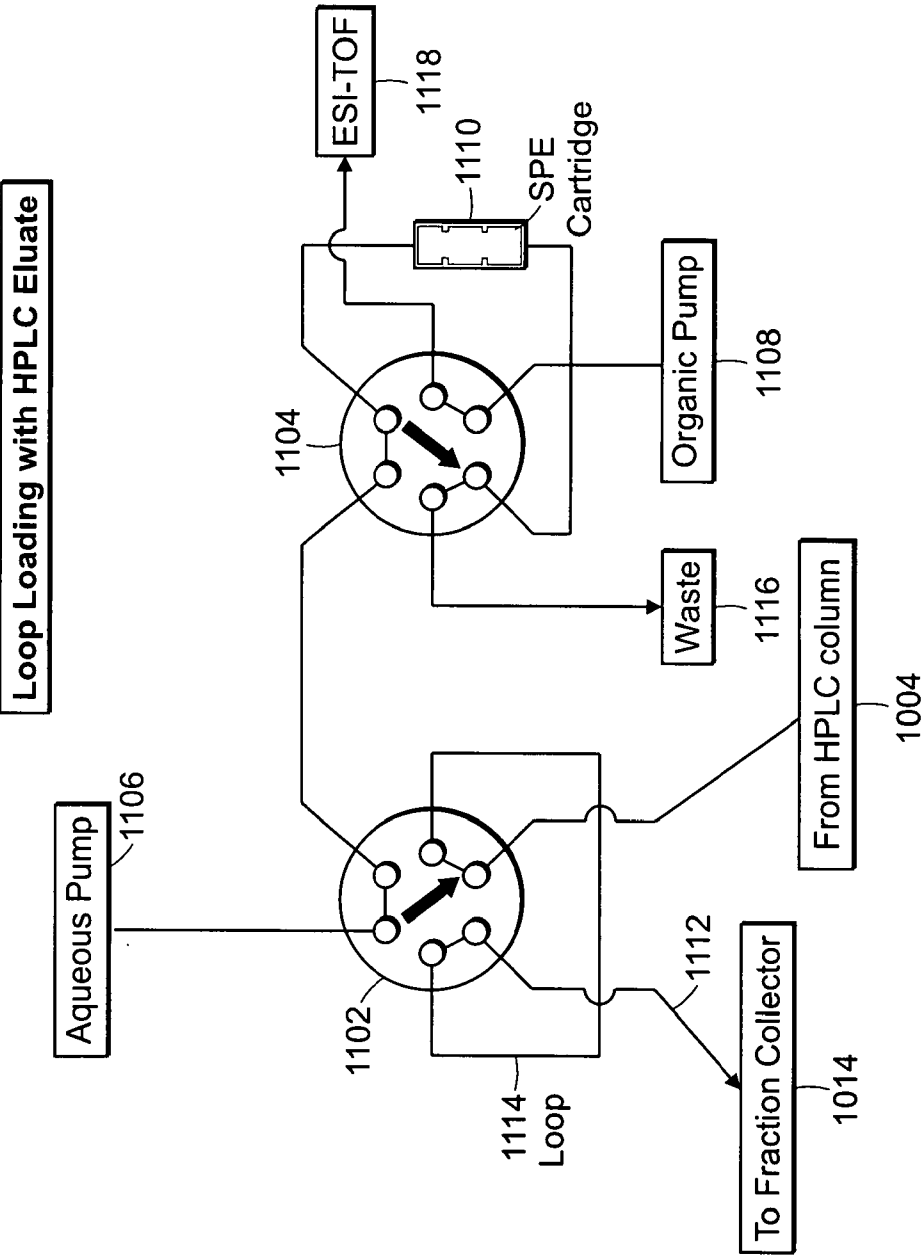


FIG. 11(b)

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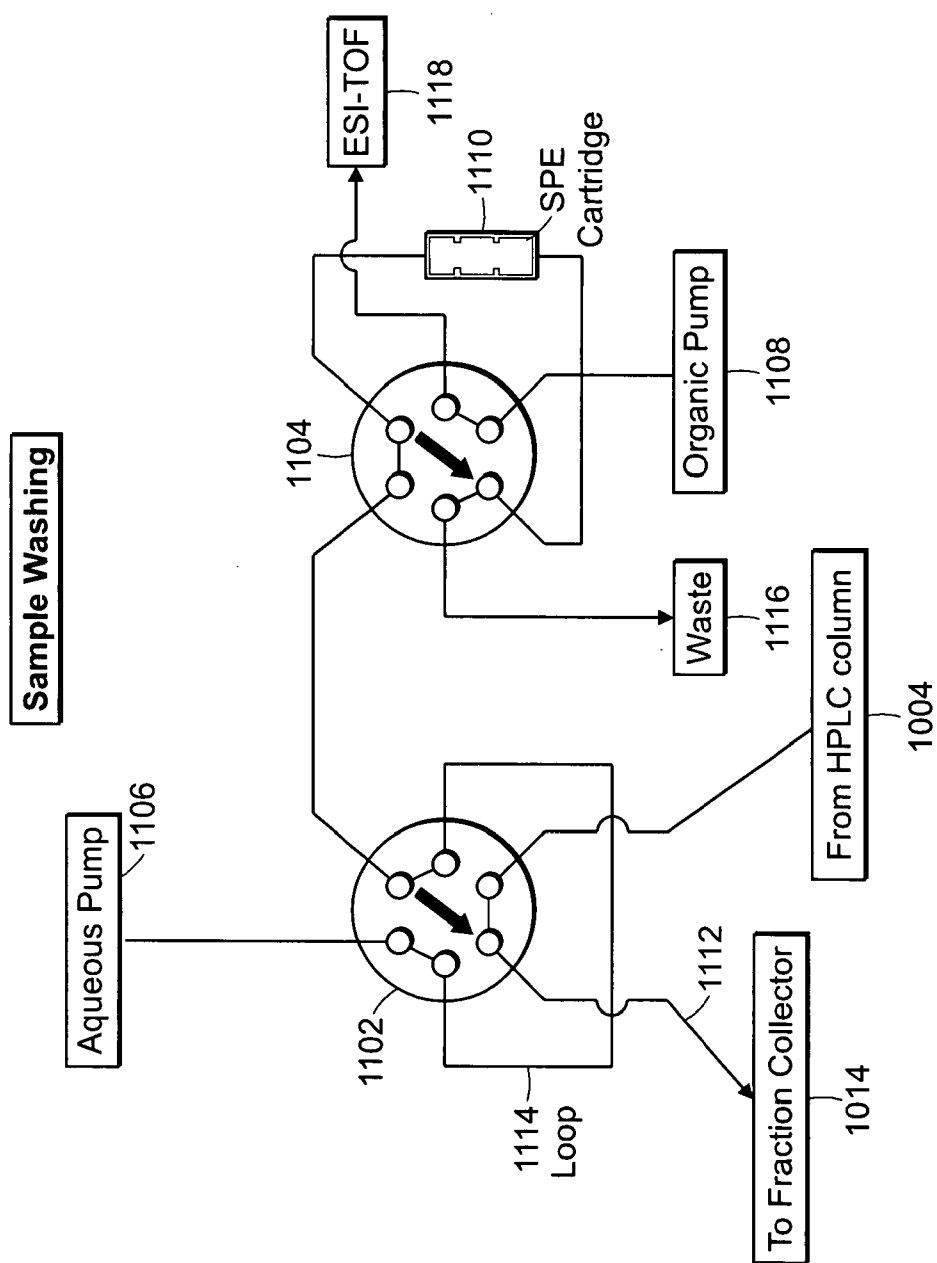


FIG. 11(c)

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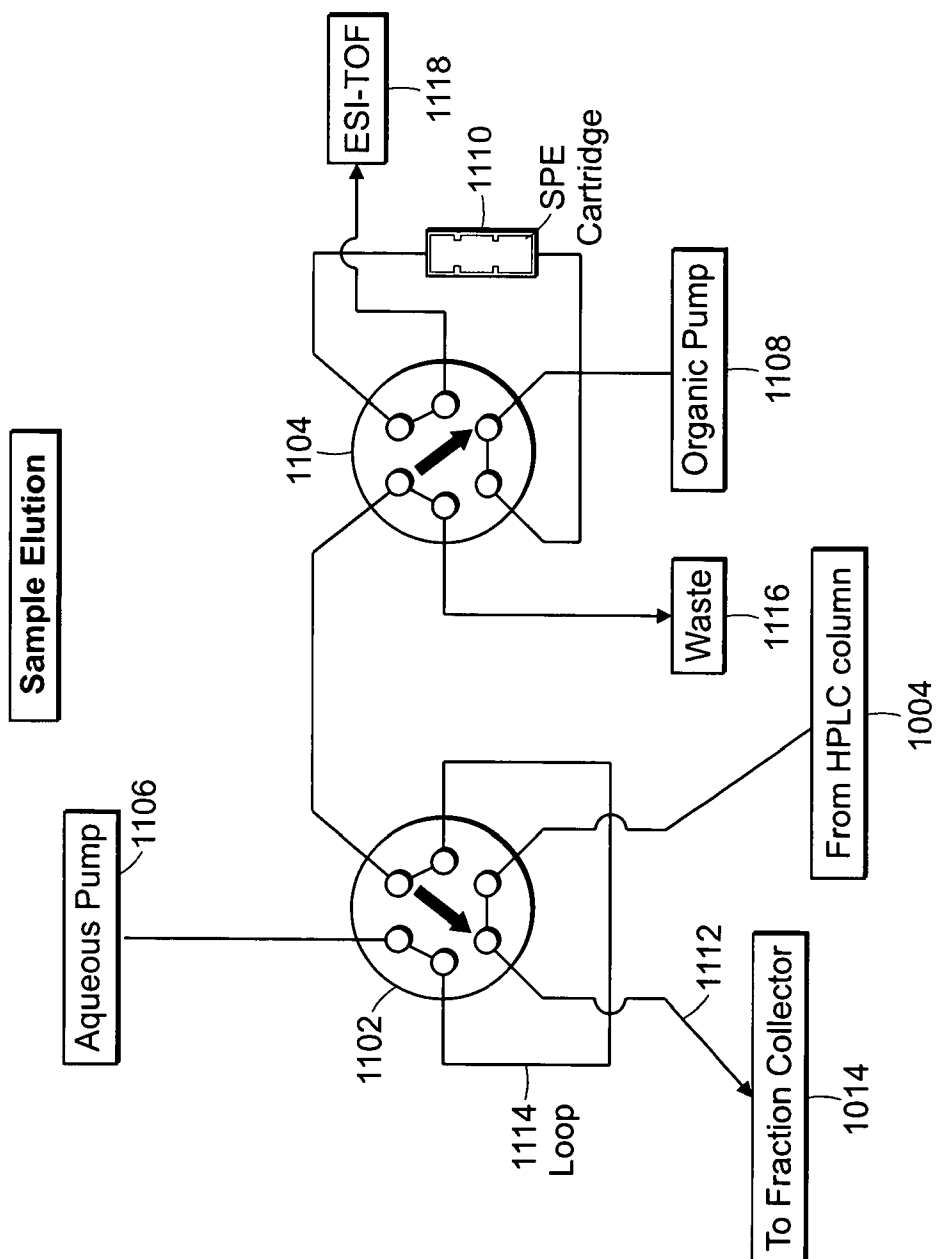


FIG. 11(d)

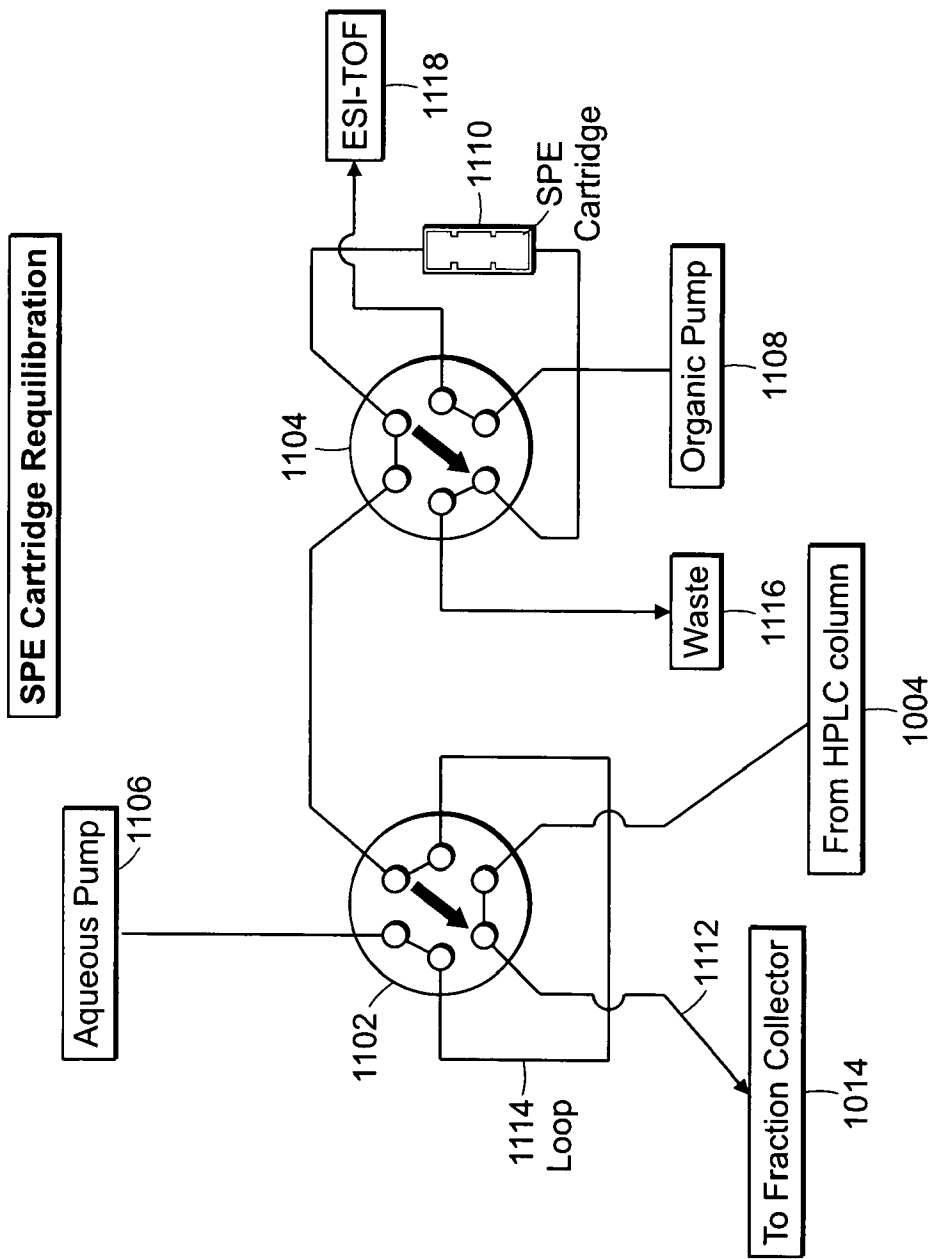


FIG. 11(e)

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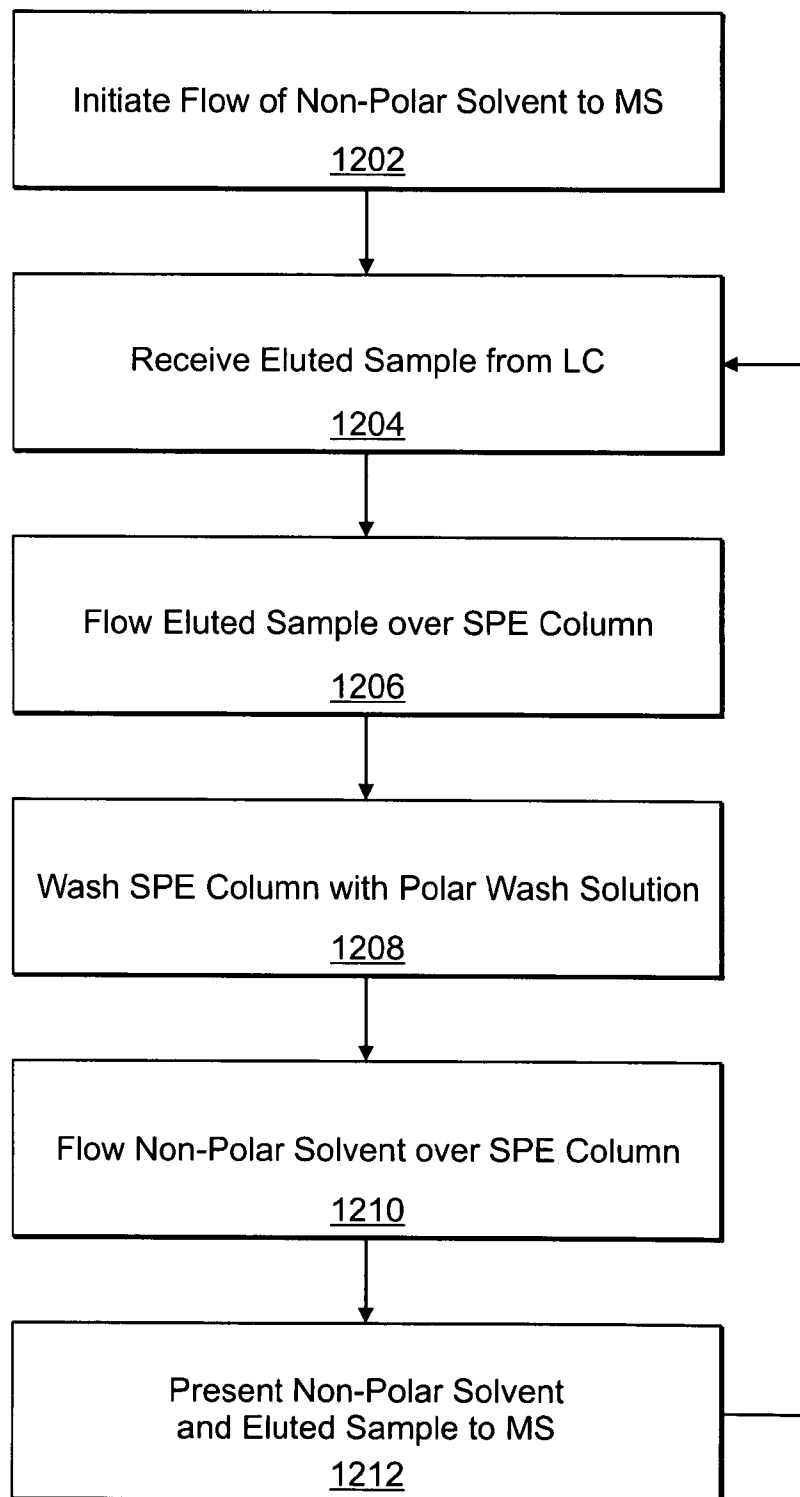


FIG. 12

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US2008/082229

A. CLASSIFICATION OF SUBJECT MATTER

IPC(8) - B01D 59/44 (2008.04)

USPC - 250/281

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC(8) - B01D 59/44, 59/50 (2008.04)

USPC - 250/281, 282, 288

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

PatBase, Google Patent Search

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 7,189,504 B2 (MARNETT et al) 13 March 2007 (13.03.2007) entire document	1, 7, 19
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Y		2-6, 8-18
Y	WO 2007/059312 A2 (MIAO et al) 24 May 2007 (24.05.2007) entire document	2-6, 16
Y	US 6,150,172 A (SCHMERR et al) 21 November 2000 (21.11.2000) entire document	15-16
Y	US 7,214,320 B1 (GREGORI et al) 08 May 2007 (08.05.2007) entire document	8-14, 17-18

☐ Further documents are listed in the continuation of Box C.

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"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

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"&" document member of the same patent family

Date of the actual completion of the international search

08 December 2008

Date of mailing of the international search report

27 JAN 2009

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