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(54) Titre : COMPLEXES METALLIQUES CONTENANT DU PERFLUOROALKYLE ET LEUR UTILISATION DANS LE
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(54) Title: METAL COMPLEXES THAT CONTAIN PERFLUOROALKYL, PROCESS FOR THEIR PRODUCTION AND
THEIR USE IN NMR DIAGNOSIS

(57) Abrégé/Abstract:

The invention concerns new monomer perfluoroalkyl-substituted metal complexes, a process for their production, and their use in diagnostics and therapy. The compounds according to the invention to the invention are particularly suited for use as in vivo contrast agents in nuclear spin resonance tomography (MRT). They can be preferably used as blood pool agents and contrast agents for lymphography.

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(54) Title: PERFLUOROALKYL-CONTAINING METAL COMPLEXES AND THEIR USE IN NMR DIAGNOSTICS (54) Bezeichnung: PERFLUORALKYLHALTIGE METALLKOMPLEXE UND IHRE VERWENDUNG IN DER NMR-DIAGNOSTIK (57) Abstract The invention concerns new monomer perfluoroalkyl-substituted metal complexes, a process for their production, and their use in diagnostics and therapy. The compounds according to the invention are particularly suited for use as in vivo contrast agents in nuclear spin resonance tomography (MRT). They can be preferably used as blood pool agents and contrast agents for lymphography. (57) Zusammenfassung Die Erfindung betrifft neue, monomere, perfluoralkylsubstituierte Metallkomplexe, Verfahren zu deren Herstellung und ihre Verwendung in Diagnostik und Therapie. Die erfindungsgemäßen Verbindungen sind insbesondere als in vivo-Kontrastmittel in der Kernspintomographie (MRT) geeignet. Bevorzugt können sie als blood-pool-agents und als Lymphographika eingesetzt werden.		

Metal Complexes that Contain Perfluoroalkyl, Process for their Production and their Use in NMR Diagnosis

The invention relates to the subjects characterized in the claims, i.e., new, monomeric, perfluoroalkyl-substituted, paramagnetic metal complexes and complex salts, pharmaceutical agents containing these metal complexes, processes for their production, and their use as contrast media in ^1H -NMR diagnosis and ^1H -NMR spectroscopy, diagnostic radiology, and radiodiagnosis and as radiotherapeutic agents.

Nuclear magnetic resonance (NMR) is now a very extensively used method of medical diagnosis, used for in vivo imaging, with which vessels of the body and body tissue (including tumors) can be visualized by measuring the magnetic properties of the protons in the body water. To this end, e.g., contrast media are used that produce contrast enhancement in the resulting images or make these images readable by influencing specific NMR parameters of the body protons (e.g., relaxation times T^1 and T^2). Mainly complexes of paramagnetic ions, such as, e.g., gadolinium-containing complexes (e.g., Magnevist^(R)) are used owing to the effect of the paramagnetic ions on the shortening of the relaxation times. A measure of the shortening of the relaxation time is relaxivity, which is indicated in $\text{mM}^{-1} \cdot \text{sec}^{-1}$.

Paramagnetic ions, such as, e.g., Gd^{3+} , Mn^{2+} , Cr^{3+} , Fe^{3+} and Cu^{2+} cannot be administered in free form as solutions since they are too toxic. To make these ions suitable for in vivo use, they

are generally complexed, which is described for the first time in EP 0 071 564 A1 (complexing with aminopolycarboxylic acids, e.g., with diethylenetriamine-pentaacetic acid [DTPA]). The di-N-methylglucamine salt of the Gd-DTPA complex is known under the name Magnevist^(R) and is used, i.a., to diagnose tumors in the human brain and in the kidney.

The meglumine salt of Gd-DOTA (gadolinium(III) complex of 1,4,7,10-tetracarboxymethyl-1,4,7,10-tetraazacyclododecane) that is described in French Patent 25 39 996 is another contrast medium that has demonstrated its value in nuclear spin tomography and was registered under the name Dotarem^(R).

These contrast media cannot be used, however, in a satisfactory manner for all applications. Thus, the contrast media used clinically at present for the modern imaging processes of nuclear spin tomography (MRI) and computer tomography (CT), such as, e.g., Magnevist^(R), Pro Hance^(R), Ultravist^(R) and Omniscan^(R), are dispersed into the entire extracellular space of the body (in the intravascular space and in the interstice).

Contrast media that also are dispersed exclusively in the vascular space when administered in the latter and thus label it (so-called blood-pool agents) are especially desirable, however, for visualizing vessels.

An attempt was made to solve these problems by using complexing agents, which are bonded to macromolecules or biomolecules. Up until now, this has had very limited success.

Thus, for example, the number of paramagnetic centers in the complexes that are described in EP 0 088 695 A1 and EP 0 150 844 A1 is inadequate for satisfactory imaging.

If the number of required metal ions is increased by repeatedly introducing complexing units into a macromolecular biomolecule, this is associated with an intolerable impairment of the affinity and/or specificity of this biomolecule [J. Nucl. Med. 24, 1158 (1983)].

Macromolecular contrast media for angiography, such as albumin-Gd-DTPA, are described in Radiology 1987; 162: 205. Twenty-four hours after intravenous injection in rats, however, albumin-Gd-DTPA shows a concentration in the liver tissue that amounts to almost 30% of the dose. In addition, only 20% of the dose is eliminated within 24 hours.

The macromolecule polylysine-Gd-DTPA (EP 0 233 619 A1) can also be used as a blood-pool agent. For production reasons, however, this compound consists of a mixture of molecules of various sizes. In excretion tests in rats, it was possible to show that this macromolecule is excreted unaltered by glomerular filtration via the kidney. For synthesis reasons, it contains polylysine-Gd-DTPA but also macromolecules that are so large that they cannot pass through the capillaries of the kidney in the case of glomerular filtration and thus remain in the body.

Macromolecular contrast media based on carbohydrates, e.g., dextran, have also been described (EP 0 326 226 A1). The drawback of these compounds lies in the fact that the latter

generally have only about 5% of the signal-enhancing paramagnetic cation.

The object of the invention was therefore to make available new ^1H -NMR contrast media that do not exhibit the above-mentioned drawbacks and especially show a higher proton relaxivity and thus allow a reduction of the dose with increased signal intensity. The contrast media are also to be stable, well compatible and mainly exhibit organ-specific properties, whereby, on the one hand, their retention in the organs to be studied is to be sufficient to obtain the number of images that is necessary for an unambiguous diagnosis at low dosage, but, on the other hand, an excretion of metals from the body that is as fast as possible and very largely complete is then to be ensured.

The object of the invention is achieved with the monomeric, perfluoroalkyl-containing compounds of general formula I according to claim 1, which show a surprisingly high proton relaxivity of 20-50 $[\text{mM}^{-1} \cdot \text{sec}^{-1}]$, 39°C, 0.47 T]. In comparison with this, proton relaxivity for commercially available ^1H -NMR contrast media Magnevist^(R), Dotarem^(R), Omniscan^(R) and Pro Hance^(R) is between 3.5-4.9 $[\text{mM}^{-1} \cdot \text{sec}^{-1}]$, 39°C, 0.47 T].

In addition, the compounds according to the invention are extremely well suited for detecting and locating vascular diseases, since they are also exclusively dispersed in the latter when administered in the intravascular space. The compounds according to the invention make it possible to distinguish tissue that is well-supplied with blood from tissue that is poorly supplied with blood with the aid of nuclear spin tomography and

thus to diagnose an ischemia. Infarcted tissue can also be delimited from surrounding healthy or ischemic tissue due to its anemia, if the contrast media according to the invention are used. This is of special importance if the point is, e.g., to distinguish a myocardial infarction from an ischemia.

Compared to the macromolecular compounds that were previously used as blood-pool agents, such as, for example, Gd-DTPA polylysine, the compounds according to the invention also show a higher T^1 relaxivity (see Tab. 3) and are thus distinguished by a large increase in signal intensity in the case of NMR imaging. Since, in addition, they have an extended retention in the blood space, they can also be administered in relatively small dosages (of, e.g., $\leq 50 \mu\text{mol}$ of Gd/kg of body weight). Mainly, however, the compounds according to the invention, none of which are polymeric compounds, are eliminated quickly and almost completely from the body.

It was also shown that the compounds of this invention are suitable not only as blood-pool agents, but are also excellent lymph-specific MRT contrast media (lymphographic agents).

The visualization of the lymph nodes is of vital importance for early detection of metastatic invasion in cancer patients. The contrast media according to the invention allow small metastases in non-enlarged lymph nodes ($< 2 \text{ cm}$) to be distinguished from lymph node hyperplasias without malignant invasions. In this case, the contrast media can be administered intravascularly or interstitially/intracutaneously. Interstitial/intracutaneous administration has the advantage that

the substance is transported directly from the scattering concentrate table (e.g., primary tumor) via the corresponding lymph tracts into the potentially affected regional lymph node stations. Likewise, a high concentration of contrast medium in the lymph nodes can be achieved with a small dose.

The compounds according to the invention meet all requirements that are called for by contrast media in indirect MRT lymphography: good local compatibility, quick elimination of injection site, quick and very largely complete elimination from the entire organism. They also show a high concentration over several lymph node stations and thus make it possible to make relevant diagnostic statements. Thus, in the guinea pig model, it was possible to show a high concentration over several lymph node stations (popliteal, inguinal, iliac) after s.c. administration (2.5-10 $\mu\text{mol/kg}$ of body weight, injection in the interdigital spaces of the hind paw). In especially suitable cases, gadolinium concentrations of respectively ≥ 200 or ≥ 300 $\mu\text{mol/l}$ were obtained in the second (inguinal) and third (iliac) stations. Usually, lymph node concentrations in the range of 100 to 1000 $\mu\text{mol/l}$ can be obtained with the compounds according to the invention.

It was possible to confirm the special suitability of the compounds according to the invention in MR imaging studies in guinea pigs. Thus, 120 minutes after subcutaneous administration of 10 $\mu\text{mol/kg}$ of body weight of a perfluorine-containing gadolinium complex (guinea pigs, hind paw, interdigital space) in T^1 -weighted spin-echo images (TR 400 ms, TE 15 ms), a clear

enhancement of the popliteal lymph nodes (270%) as well as the inguinal lymph nodes (104%) was observed (cf. Image 1).

In humans, the compounds according to the invention can be injected locally (either subcutaneously or directly percutaneously in the tissue of interest). Several injection sites (weals) with a respective injection volume of 0.2 to 1 ml are grouped around the areas of interest (e.g., tumor). In this case, the injected total volume should not in any case exceed 5 ml. This means that in the formulation, a metal concentration of 75-100 mmol/l must be present, so that a potential clinical dose of 5-10 $\mu\text{mol/kg}$ of body weight can be administered with this volume. The site of administration thereof depends on whether a specific lymph outflow field from the tissue corresponding to it is to be specifically stained (e.g., in the case of gynecological or rectal tumors), or whether the unknown outflow field of a certain lesion (ergo the area for a possible therapeutic intervention, e.g., with melanoma or carcinoma of the breast) is to be visualized.

In normal lymph node tissue, where the concentration of the compound occurs, gadolinium concentrations of at least 50 $\mu\text{mol/l}$ and at most 2500 $\mu\text{mol/l}$ are required for MR imaging. The imaging can be carried out (depending on injection site and tissue) after 30 minutes or up to 4-6 hours after injection of the compounds according to the invention. Since mainly the T^1 relaxation times of the protons of the water of the lymph node tissue are influenced with the compounds of gadolinium complexes according to the invention, T^1 -weighted sequences are best able to identify

an MRT enhancement of the lymph node stations. Since lymph nodes very frequently are embedded in fatty tissue, and the latter has a very high signal intensity in such sequences, fat-suppressed measuring methods suggest themselves. Paramagnetic gadolinium complexes in combination with fat-suppressed, T¹-weighted measuring sequences have the great advantage, compared to the formulations of superparamagnetic iron oxide particles, that they allow MRT images with greater spatial resolution, with fewer distortion artifacts (based on susceptibility artifacts) and with shorter imaging time.

Since positive labeling of the lymph nodes occurs (i.e., signal rise), MRT images without contrast media for comparison are also no longer absolutely necessary, and the total examination time per patient can be shortened.

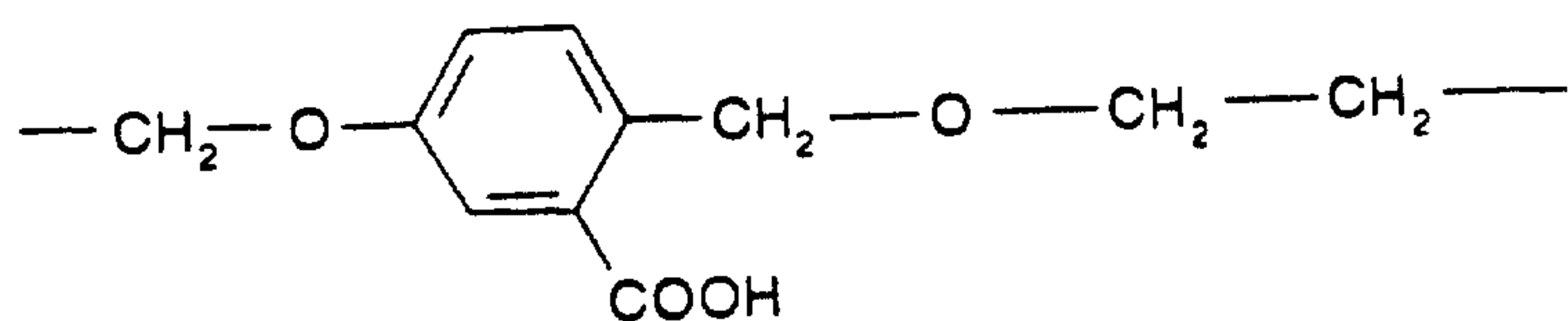
The new perfluoroalkyl-containing compounds of general formula I of claim 1 according to the invention comprise both complexing agents and metal complexes. Compounds of general formula I with Z¹ as a hydrogen atom are referred to as complexing agents and compounds with at least one of possible substituents Z¹ as a metal ion equivalent are referred to as metal complexes.

The compounds of general formula I according to the

invention contain, as preferred radical L, the following:

- CH₂-
- CH₂CH₂-
- (CH₂)_s- s = 3 - 15
- CH₂-O-CH₂CH₂-
- CH₂-(O-CH₂-CH₂)_t t = 2 - 6
- CH₂-NH-CO-
- CH₂-NH-CO-CH₂-N(CH₂COOH)-SO₂-
- CH₂-NH-CO-CH₂-N(C₂H₅)-SO₂-
- CH₂-NH-CO-CH₂-N(C₁₀H₂₁)-SO₂-
- CH₂-NH-CO-CH₂-N(C₆H₁₃)-SO₂-
- CH₂-NH-CO-(CH₂)₁₀-N(C₂H₅)-SO₂-
- CH₂-NH-CO-CH₂-N(-CH₂-C₆H₅)-SO₂-
- CH₂-NH-CO-CH₂-N(-CH₂-CH₂-OH)SO₂-
- CH₂-NHCO-(CH₂)₁₀-S-CH₂CH₂-
- CH₂NHCOCH₂-O-CH₂CH₂-
- CH₂NHCO(CH₂)₁₀-O-CH₂CH₂-
- CH₂-C₆H₄-O-CH₂CH₂-
- CH₂-O-CH₂-C(CH₂-OCH₂CH₂-C₆F₁₃)₂-CH₂-OCH₂-CH₂-
- CH₂-NHCOCH₂CH₂CON-CH₂CH₂NHCOCH₂N(C₂H₅)SO₂C₈F₁₇

$\begin{array}{c} | \\ \text{CH}_2\text{-CH}_2\text{NHCOCH}_2\text{N(C}_2\text{H}_5\text{)-SO}_2\text{-} \end{array}$
- CH₂-O-CH₂-CH(OC₁₀H₂₁)-CH₂-O-CH₂CH₂-
- (CH₂NHCO)₄-CH₂O-CH₂CH₂-
- (CH₂NHCO)₃-CH₂O-CH₂CH₂-
- CH₂-OCH₂C(CH₂OH)₂-CH₂-O-CH₂CH₂-



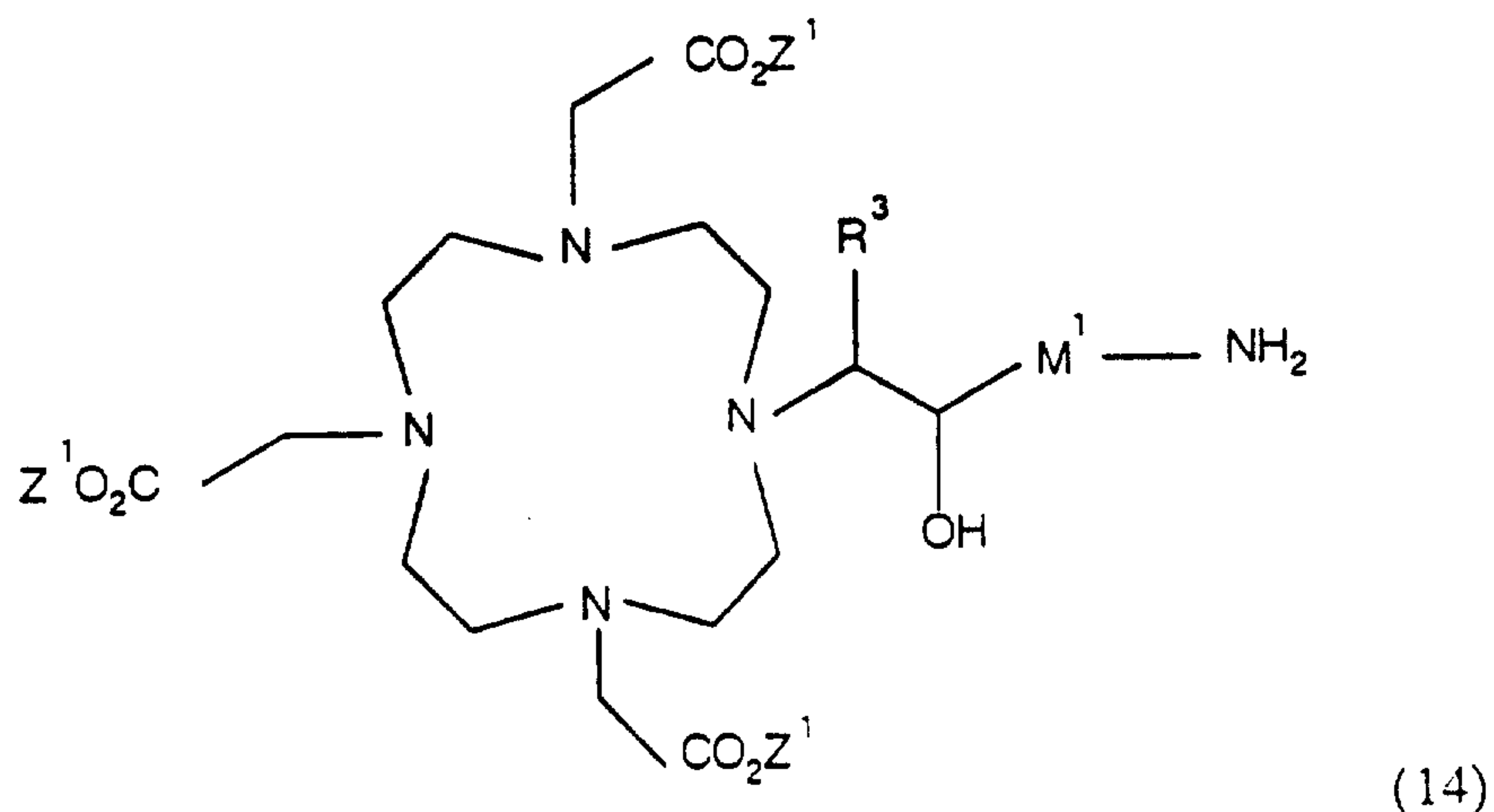
- CH₂NHCOCH₂N(C₆H₅)-SO₂-
- NHCO-CH₂-CH₂-
- NHCO-CH₂-O-CH₂CH₂-
- NH-CO-

-NH-CO-CH₂-N(CH₂COOH)-SO₂-
-NH-CO-CH₂-N(C₂H₅)-SO₂-
-NH-CO-CH₂-N(C₁₀H₂₁)-SO₂-
-NH-CO-CH₂-N(C₆H₁₃)-SO₂-
-NH-CO-(CH₂)₁₀-N(C₂H₅)-SO₂-
-NH-CO-CH₂-N(-CH₂-C₆H₅)-SO₂-
-NH-CO-CH₂-N(-CH₂-CH₂-OH)SO₂-
-NH-CO-CH₂-
-CH₂-O-C₆H₄-O-CH₂-CH₂-
-CH₂-C₆H₄-O-CH₂-CH₂-
-N(C₂H₅)-SO₂-
-N(C₆H₅)-SO₂-
-N(C₁₀H₂₁)-SO₂-
-N(C₆H₁₃)-SO₂-
-N(C₂H₄OH)-SO₂-
-N(CH₂COOH)-SO₂-
-N(CH₂C₆H₅)-SO₂-
-N-[CH(CH₂OH)₂]-SO₂-
-N-[CH(CH₂OH)CH(CH₂OH)]-SO₂-

According to the invention, radicals L of the compounds mentioned in the examples of this description of the invention are quite especially preferred.

Other preferred compounds are those in which X of the formula $-C_nF_{2n}X$ means fluorine, and n stands for the numbers 4 to 15.

Compounds of general formula I with A in the meaning of general formula IX, whereby L contains at least one $-NHCO$ group, can be obtained from compounds of general formula 14

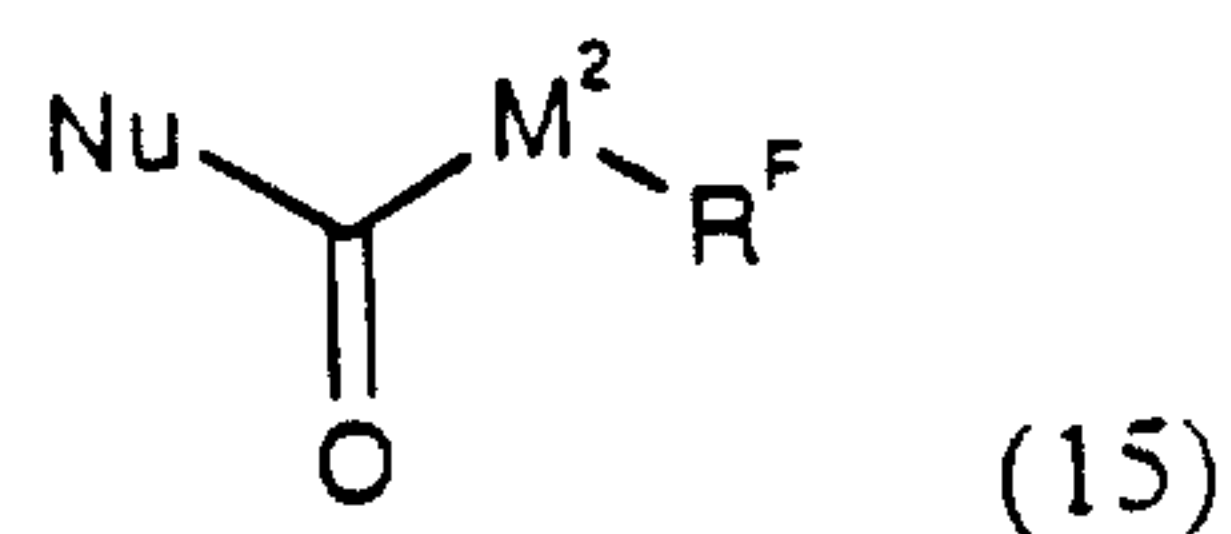


in which

R^3 is in the above-mentioned meaning, Z^1 is in the meaning of a metal ion equivalent of atomic numbers 21-29, 39, 42, 44 or 57-83, and

M^1 is in the meaning of L,

by reaction with compounds of general formula 15



in which

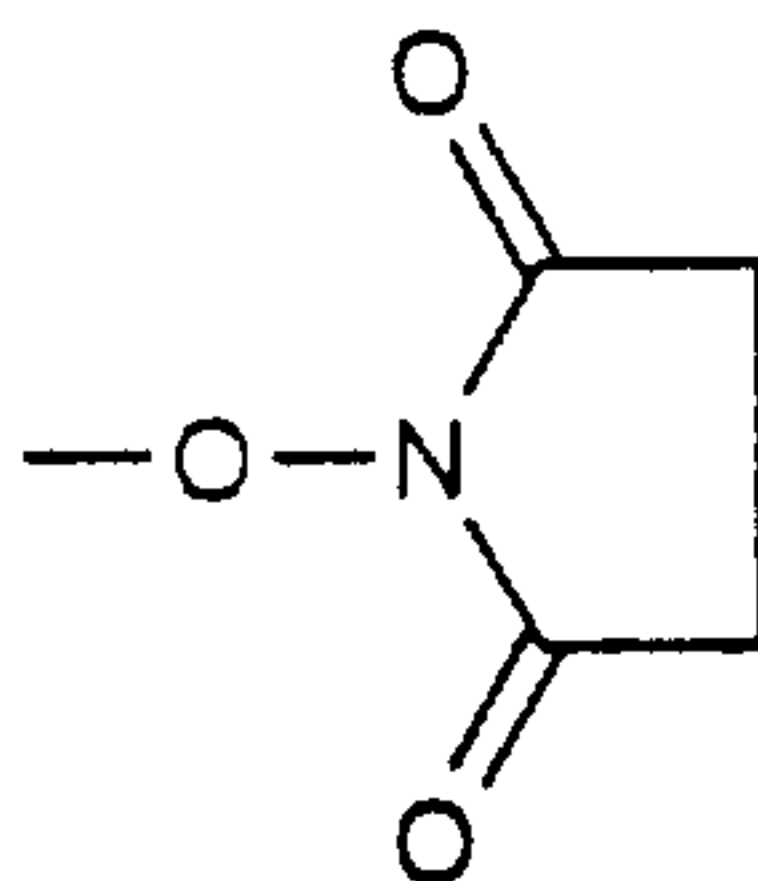
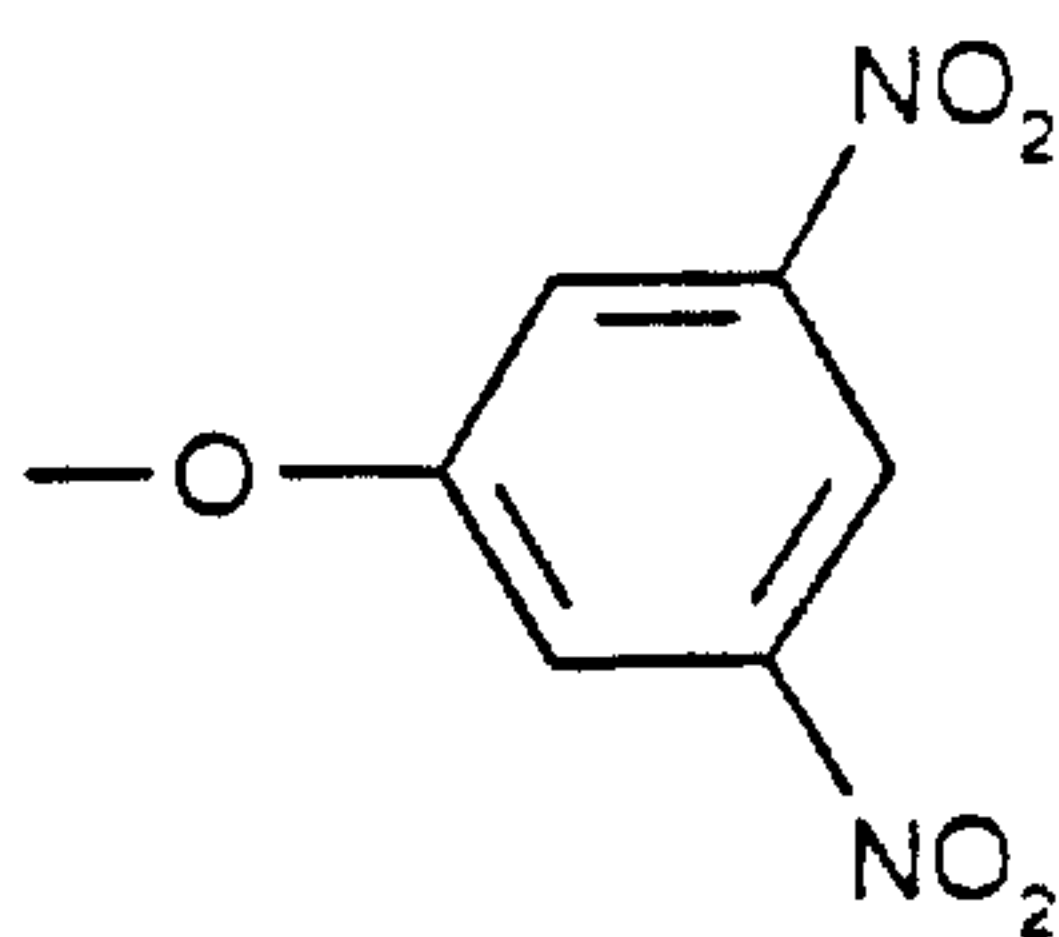
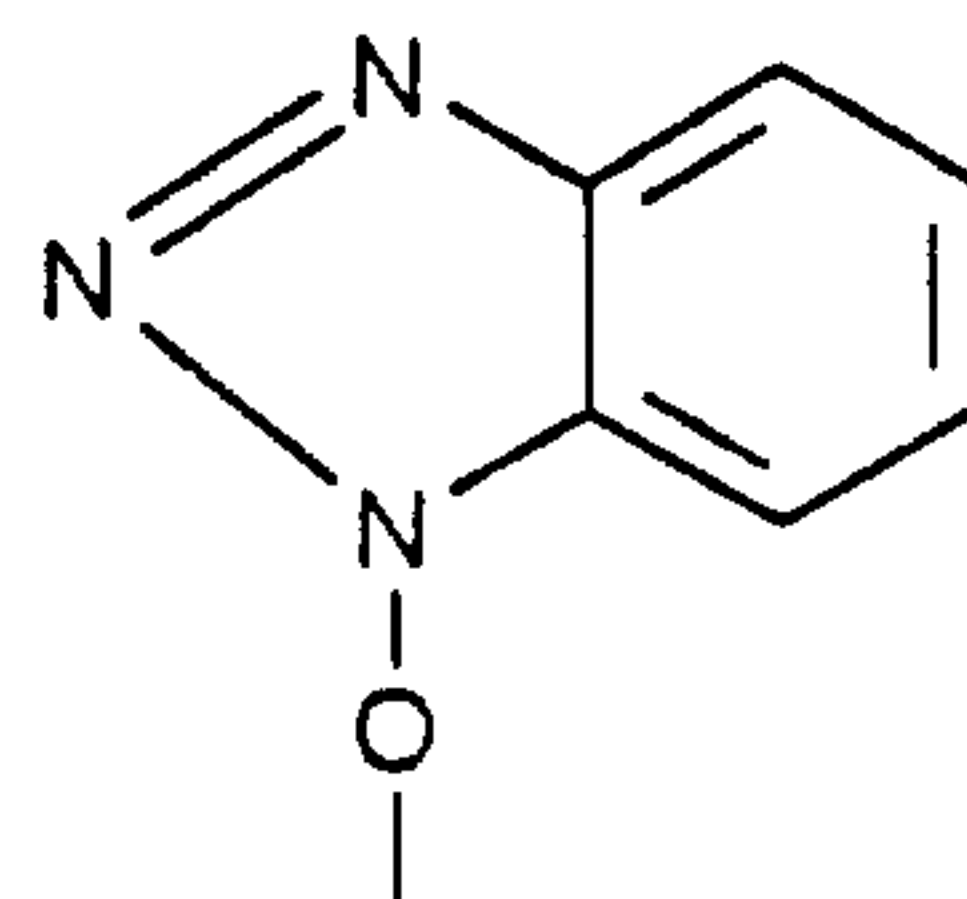
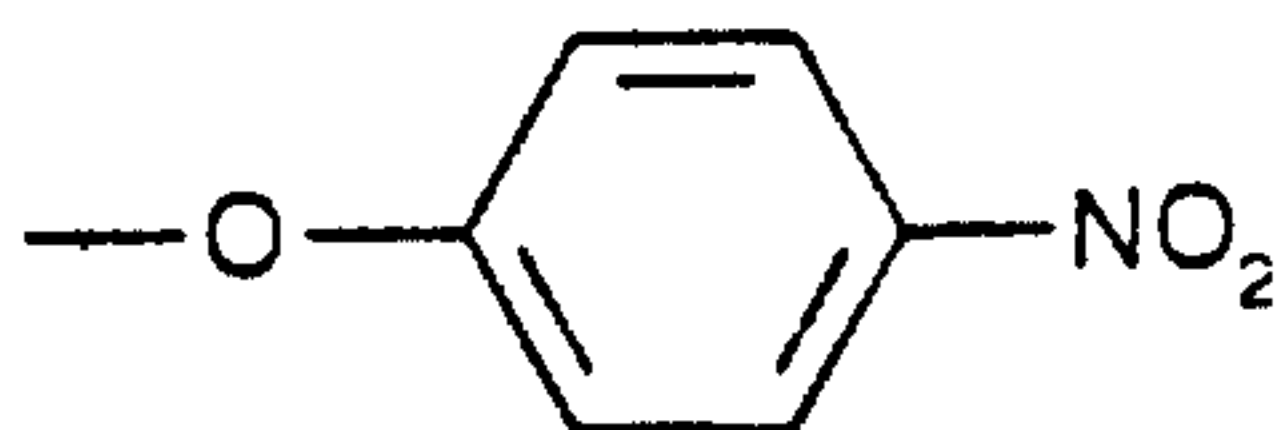
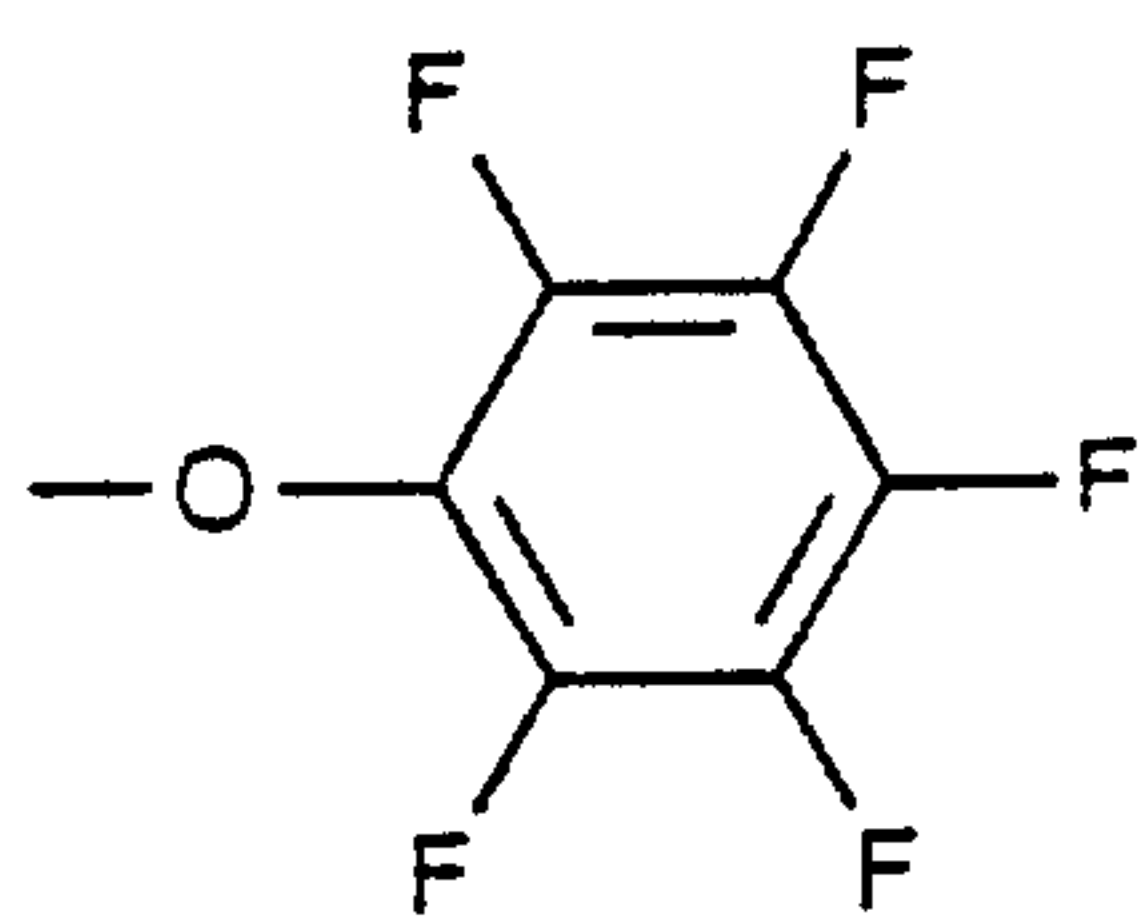
R^f has the above-mentioned meaning,

M^2 is in the meaning of L and

Nu is in the meaning of a nucleofuge.

Advantageously used as nucleofuges are the radicals:

Cl, F, $-OTs$, $-OMs$,



The reaction is carried out in a mixture of water and organic solvents such as: isopropanol, ethanol, methanol, butanol, dioxane, tetrahydrofuran, dimethylformamide, dimethylacetamide, formamide or dichloromethane. Preferred are ternary mixtures that consist of water, isopropanol and dichloromethane.

The reaction is carried out at a temperature interval of between -10°C - 100°C , preferably between 0°C - 30°C .

As acid traps, inorganic and organic bases such as triethylamine, pyridine, N-methylmorpholine, diisopropylethylamine, dimethylaminopyridine, alkali and alkaline-earth hydroxides, their carbonates or bicarbonates such as lithium hydroxide, sodium hydroxide, potassium hydroxide, sodium carbonate, sodium bicarbonate and potassium bicarbonate are used.

The compounds of general formula 15 are obtained from compounds of general formula 16



in which

R^{f} , M^2 have the above-mentioned meaning, according to the processes of acid activation that are generally known to one skilled in the art, such as by reaction of the acid with dicyclohexylcarbodiimide, N-hydroxysuccinimide/dicyclohexylcarbodiimide, carbonyldiimidazole, 2-ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline, oxalic acid dichloride or isobutyl chloroformate according to the processes described in the literature:

- ◆ Aktivierung von Carbonsäuren [Activation of Carboxylic Acids]. Übersicht in Houben-Weyl, Methoden der organischen Chemie [Survey in Houben-Weyl, Methods of Organic Chemistry], Volume XV/2, Georg Thieme Verlag Stuttgart, 19.
- ◆ Aktivierung mit Carbodiimiden [Activation with Carbodiimides]. R. Schwyzer and H. Kappeler, Helv. 46: 1550 (1963).

- ◆ E. Wunsch et al., Volume 100: 173 (1967).
- ◆ Aktivierung mit Carbodiimiden/Hydroxysuccinimid
[Activation with Carbodiimides/Hydroxysuccinimide]: J. Am. Chem. Soc. 86: 1839 (1964) as well as J. Org. Chem. 53: 3583 (1988). Synthesis 453 (1972).
- ◆ Anhydridmethode, 2-Ethoxy-1-ethoxycarbonyl-1,2-dihydrochinolin [Anhydride Method, 2-Ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline]: B. Belleau et al., J. Am. Chem. Soc., 90: 1651 (1968), H. Kunz et al., Int. J. Pept. Prot. Res., 26: 493 (1985) and J. R. Voughn, Am. Soc. 73: 3547 (1951).
- ◆ Imidazolid-Methode [Imidazolidine Method]: B. F. Gisin, R. B. Menifield, D. C. Tosteen, Am. Soc. 91: 2691 (1969).
- ◆ Säurechlorid-Methoden, Thionylchlorid [Acid Chloride Methods, Thionyl Chloride]: Helv., 42: 1653 (1959).
- ◆ Oxalylchlorid [Oxalyl Chloride]: J. Org. Chem., 29: 843 (1964).

The compounds of general formula 16 are commercially available products (Fluorochem, ABCR) or are obtained from compounds of general formula 17



with

M^3 in the meaning of L and

Q in the meaning of oxygen, sulfur, a > CO group, > N-R³,

' R³-N-SO₂ with a binding of a nitrogen atom to a

hydrogen atom,
by reaction with compounds of general formula 18

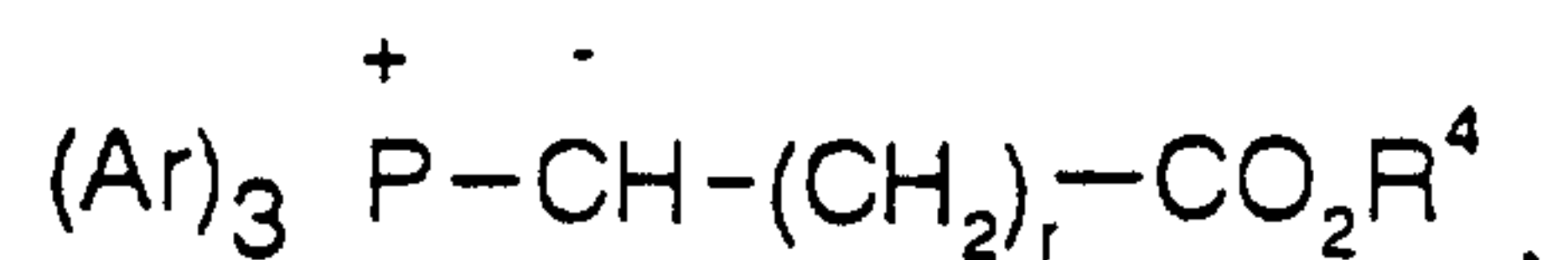


with

Hal meaning Cl, Br, I and

R⁴ meaning H, methyl, ethyl, t-butyl, benzyl, isopropyl, represented, for example, according to C. F. Ward, Soc. 121, 1161 (1922), according to the methods known to one skilled in the art, such as alkylation of alcohols with alkyl halides [Houben-Weyl, Methoden der organischen Chemie, Sauerstoffverbindungen [Oxygen Compounds] I, Part 3, Methoden zur Herstellung und Umwandlung von Ethern [Methods for the Production and Conversion of Ethers], Georg Thieme Verlag, Stuttgart 1965, Alkylierung von Alkoholen mit Alkylhalogeniden [Alkylation of Alcohols with Alkyl Halides], p. 24, Alkylierung von Alkoholen mit Alkylsulfaten [Alkylation of Alcohols with Alkyl Sulfates] p. 33] or N-Alkylierung eines Sulfonamids mit Alkylsulfonaten [N-Alkylation of a Sulfonamide with Alkylsulfonates] [Houben-Weyl, Methoden der organischen Chemie, XI/2 Stickstoffverbindungen [XI/2 Nitrogen Compounds], Georg Thieme Verlag Stuttgart, 1957, p. 680; J. E. Rickman and T. Atkins, Am. Chem. Soc., 96: 2268, 1974, 96: 2268; F. Chavez and A. D. Sherry, J. Org. Chem. 1989, 54: 2990].

If Q means a > CO group, the reaction is performed with a Wittig reagent of the structure

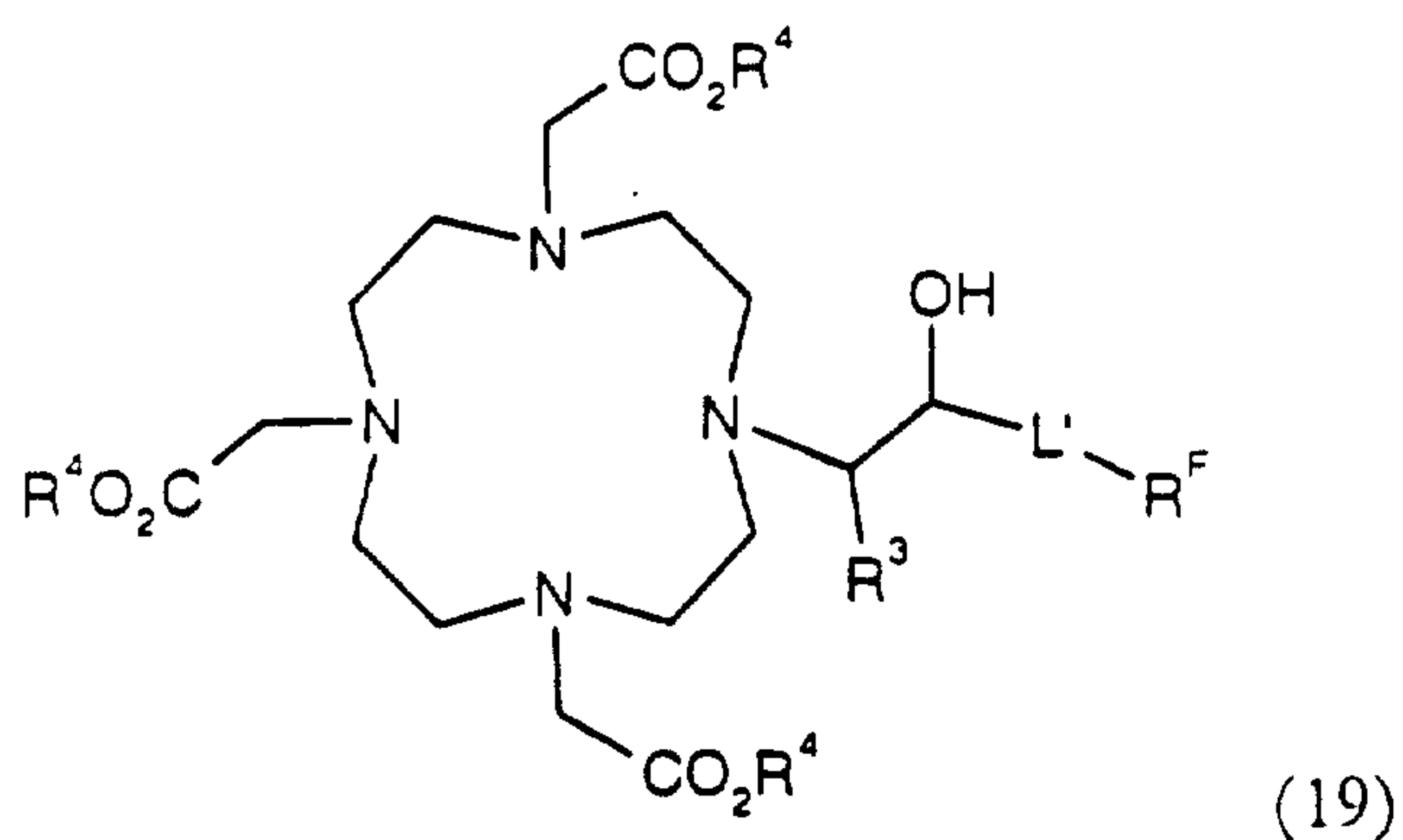


whereby r means numbers 0-16.

The -CH=CH double bond that is produced in this case can remain as a component of the structure or be converted to a -CH₂-CH₂ group by catalytic hydrogenation (Pd 5%/C).

The compounds of general formula 18 are commercially available products (Fluorochem, ABCR).

As an alternative, compounds of general formula I with A in the meaning of general formula IX can be obtained from compounds of general formula 19



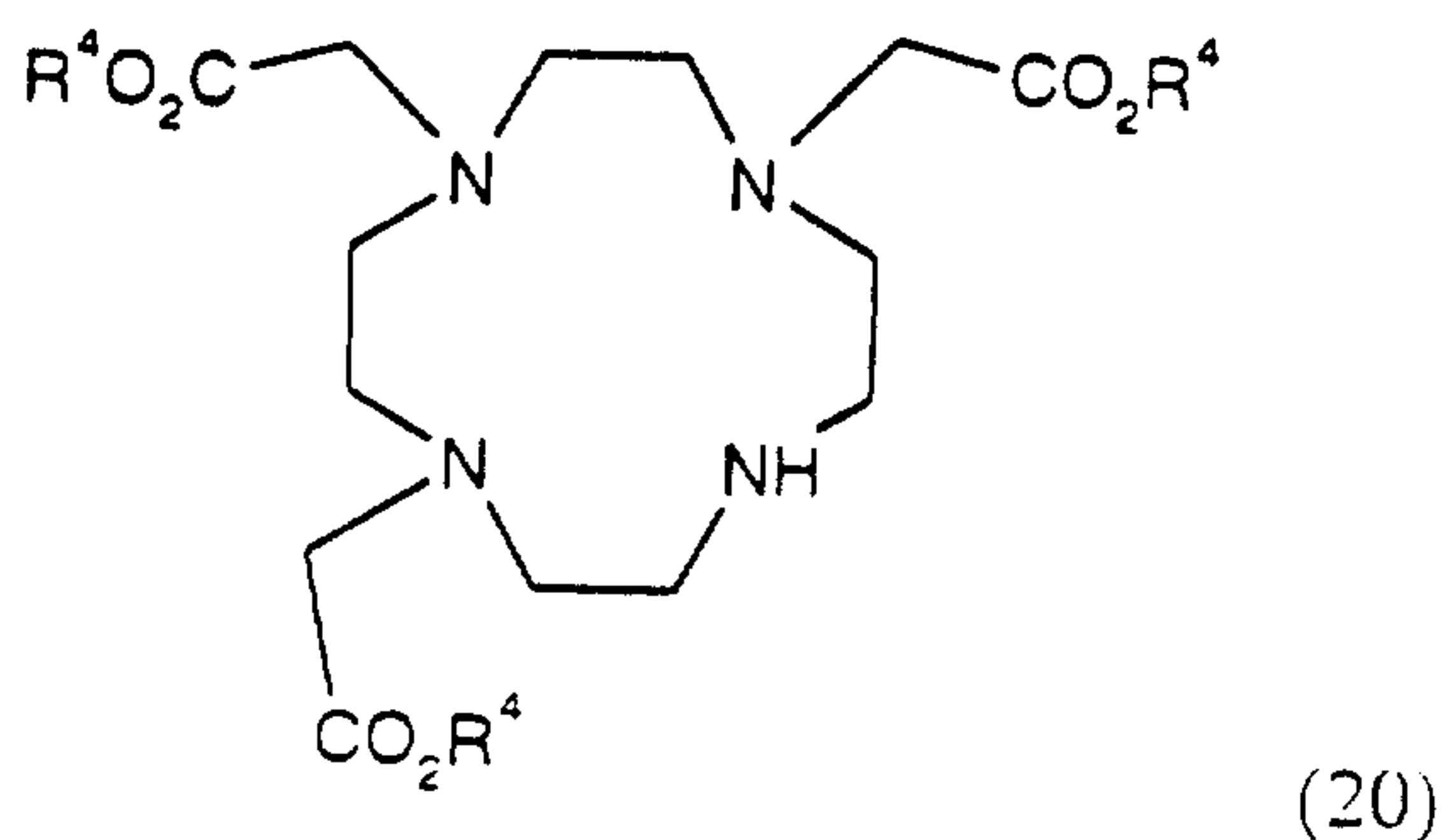
with

R^F, R³ and R⁴ in the above-mentioned meaning and

L' in the meaning of L, optionally with protected hydroxyl or carboxyl functions,

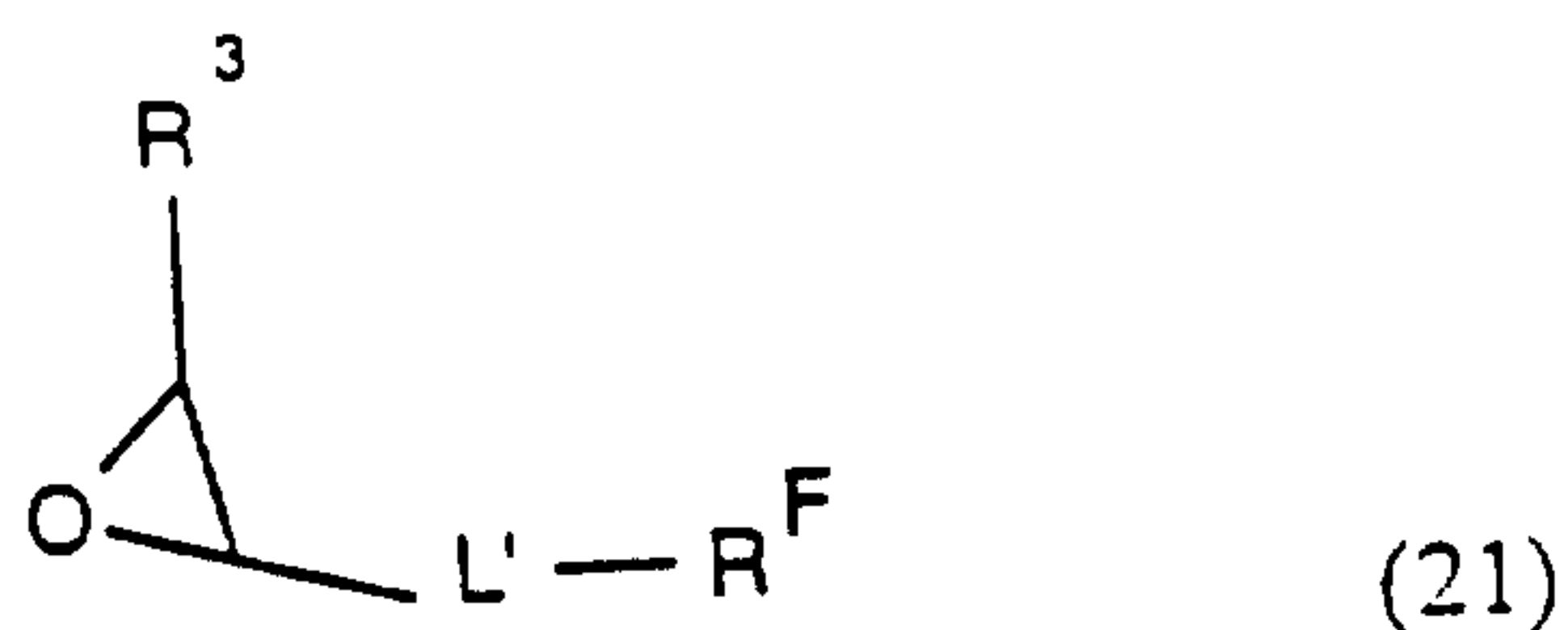
by, if necessary, protective groups that are present being cleaved and the thus obtained complexing agents being reacted with metal oxides or metal salts at room temperature or elevated temperature with the methods known according to one skilled in the art (EP 250358, EP 255471), and then -- if desired -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides.

The compounds of general formula 19 are obtained from compounds of general formula 20 (D03A or the esters)



with

R^4 in the above-mentioned meaning
by reaction with compounds of general formula 21



in which

R^3 has the meaning of R^1 , optionally in protected form, or $-(CH_2)_m-L'-R^F$, whereby m is 0, 1 or 2, and L' and R^F have the above-mentioned meaning. The reaction is carried out in alcohols such as methanol, ethanol, isopropanol, butanol, ethers such as dioxane, tetrahydrofuran, dimethoxy ethers or in water or in mixtures of water and one of the mentioned organic solvents, as well as also acetonitrile, acetone, dimethylformamide, dimethylacetamide or dimethyl sulfoxide, dichloromethane, dichloroethane, chloroform at temperatures of between -10°C and 180°C , preferably at $20^\circ - 100^\circ\text{C}$. The addition of organic or inorganic bases, such as triethylamine, pyridine, dimethylaminopyridine, N-methylmorpholine, diisopropylamine, alkali or alkaline-earth hydroxides or their carbonates or bicarbonates such as lithium hydroxide, sodium hydroxide, potassium hydroxide, sodium carbonate, potassium carbonate, sodium bicarbonate and potassium bicarbonate, has proven advantageous. In the case of low-boiling epoxides, the reaction is carried out in an autoclave.

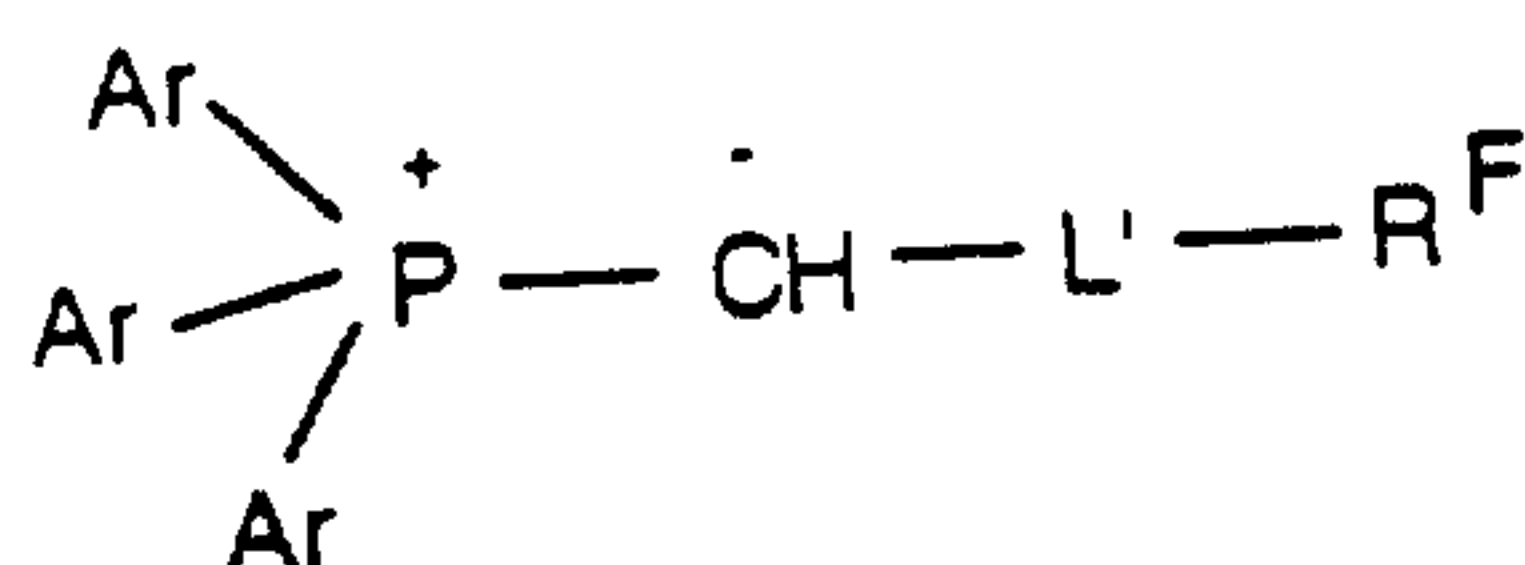
The compounds of general formula 21 are commercially available products (Fluorochem, ABCR) or can be obtained from compounds of general formula 22



by epoxidation according to the methods known to one skilled in the art, for example, the wolframate-catalyzed oxidation with H_2O_2 according to Payne, the cyclization of halohydrins or the alkaline H_2O_2 oxidation in the presence of nitriles.

Especially suitable for this reaction is 3-chloroperbenzoic acid in dichloromethane at room temperature. Houben-Weyl, Methoden der organischen Chemie, Sauerstoffverbindungen I, Part 3, Methoden zur Herstellung und Umwandlung dreigliedriger cyclische Ether (1,2-Epoxide) [Methods for the Production and Conversion of Three-Membered Cyclic Ethers (1,2-Epoxides)], Georg Thieme Verlag, Stuttgart, 1965; G. B. Payne and P. H. Williams, J. Org. Chem., 159, 24: 54; Y. Ogata and Y. Samaki, Tetrahedron 1964, 20: 2065; K. B. Sharpless et al., Pure Appl. Chem. 55, 589 (1983).

Compounds of general formula 22 are preferably obtained by Wittig reaction, or by the variants according to Horner, Schlosser or Bestmann, Houben-Weyl, Methoden der organischen Chemie XII/1, Organische Phosphorverbindungen Teil 1 [Organic Phosphorus Compounds Part 1], Georg Thieme Verlag, Stuttgart, 1963, Phosphoniumsalze [Phosphonium Salts] p. 79, Phosphoniumylide [Phosphonium Ylides] p. 112, Wittig Reaction p. 121; A. W. Johnson, Ylides and Imines of Phosphorus, John Wiley & Sons, Inc., New York, Chichester, Brisbane, Toronto, Singapore, 1993, Wittig Reaction p. 221; Schlosser-Modifikation der Wittig-Reaktion [Schlosser Modification of the Wittig Reaction] p. 240; Wadsworth-Emmons-Reaktion [Wadsworth-Emmons Reaction] p. 313; Horner Reaktion [Horner Reaction] p. 362, by reaction of a triarylphosphonium ylide



(23)

with L' and R^F in the above-mentioned meaning and Ar meaning aryl, especially phenyl, with commercially available methods (Merck, Fluka) or according to the methods known to one skilled in the art, for example, oxidation of primary alcohols with chromium trioxide/pyridine, Houben-Weyl, Methoden der organischen Chemie, Sauerstoffverbindungen II, Part 1, Aldehyde [Aldehydes], Georg Thieme Verlag, Stuttgart, 1954, aldehydes of general formula 20 that can be produced



whereby

R³ can also be H.

Triarylphosphonium ylides 23 are produced from the corresponding halides of general formula 25



with Hal, L' and R^F in the above-mentioned meaning according to the methods known to one skilled in the art, for example by heating the triarylphosphine with the alkylhalide, Houben-Weyl, Methoden der organischen Chemie XII/1, Organische Phosphorverbindungen Teil 1, Georg Thieme Verlag, Stuttgart, 1963 or A. W. Johnson, Ylides and Imines of Phosphorus, John Wiley & Sons, Inc., New York, Chichester, Brisbane, Toronto, Singapore, 1993. The compounds of general formula 25 are commercially available products (Fluorochem, ABCR, 3M).

The compounds of general formula 21 with R³ = H are preferably obtained from compounds of general formula 17

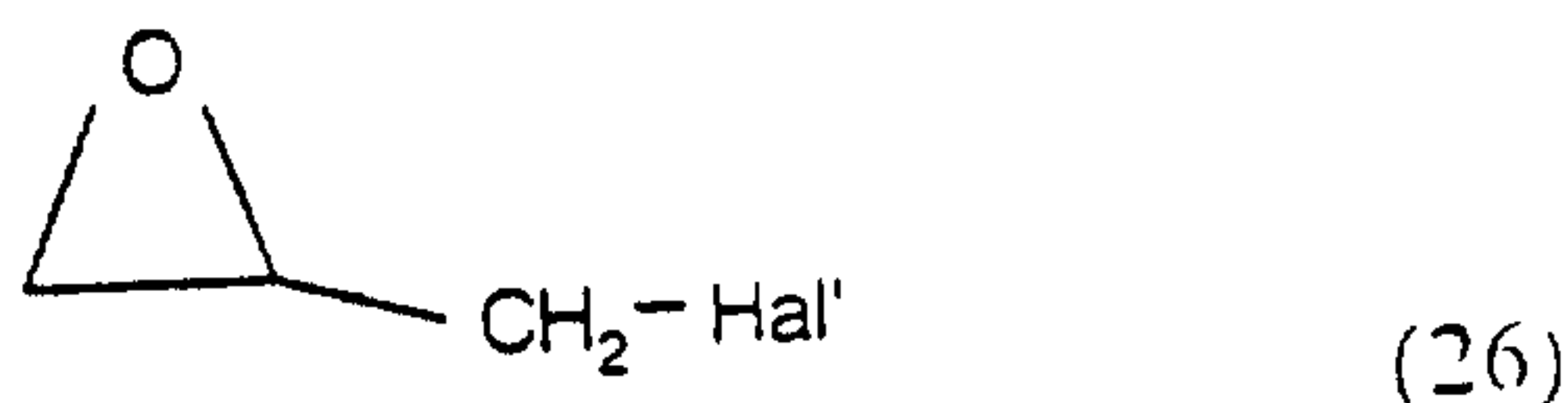


in which

Q' is in the meaning of Q , but cannot mean any $> CO$ group,
 M^3 has the meaning of L with the exception of the direct
 bond and

R^f has the above-mentioned meaning,

by reaction according to the way of etherification or
 sulfonamidealkylation with epihalohydrins that is known to one
 skilled in the art: (Houben-Weyl, Methoden der organischen
 Chemie, Sauerstoffverbindungen I, Part 3, Methoden zur
 Herstellung und Umwandlung von Ethern, Georg Thieme Verlag,
 Stuttgart, 1965, Alkylierung von Alkoholen [Alkylation of
 Alcohols], pp. 24, 33; Houben-Weyl, Methoden der organischen
 Chemie, XI/2 Stickstoffverbindungen, Georg Thieme Verlag,
 Stuttgart, 1957, p. 680; J. E. Rickman and T. J. J. Atkins, Am.
 Chem. Soc. 1974, 96: 2268; F. Chavez and A. D. Sherry, 1989, 54:
 2990) of general formula 26



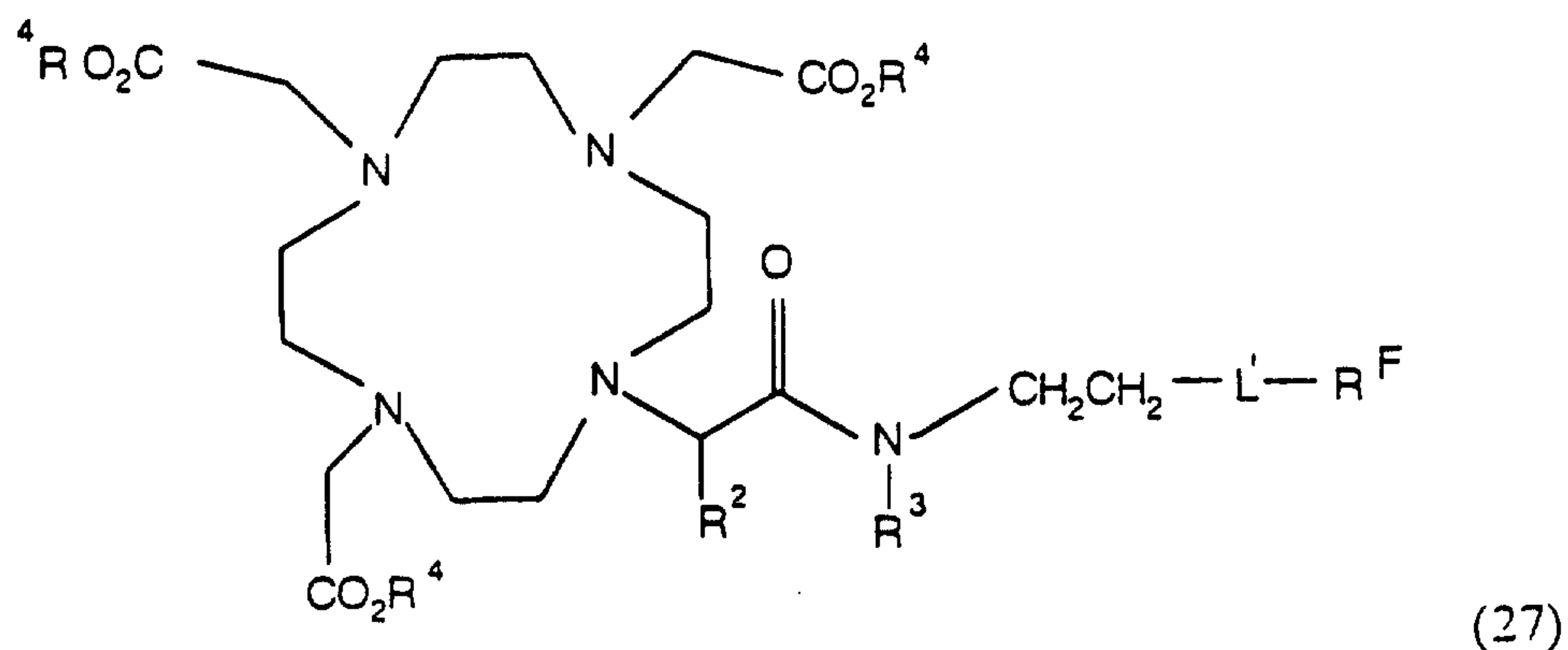
with

Hal' in the meaning of Hal , F , $-OTs$, OMs .

In the case of low-boiling epoxides, the reaction is carried
 out in an autoclave.

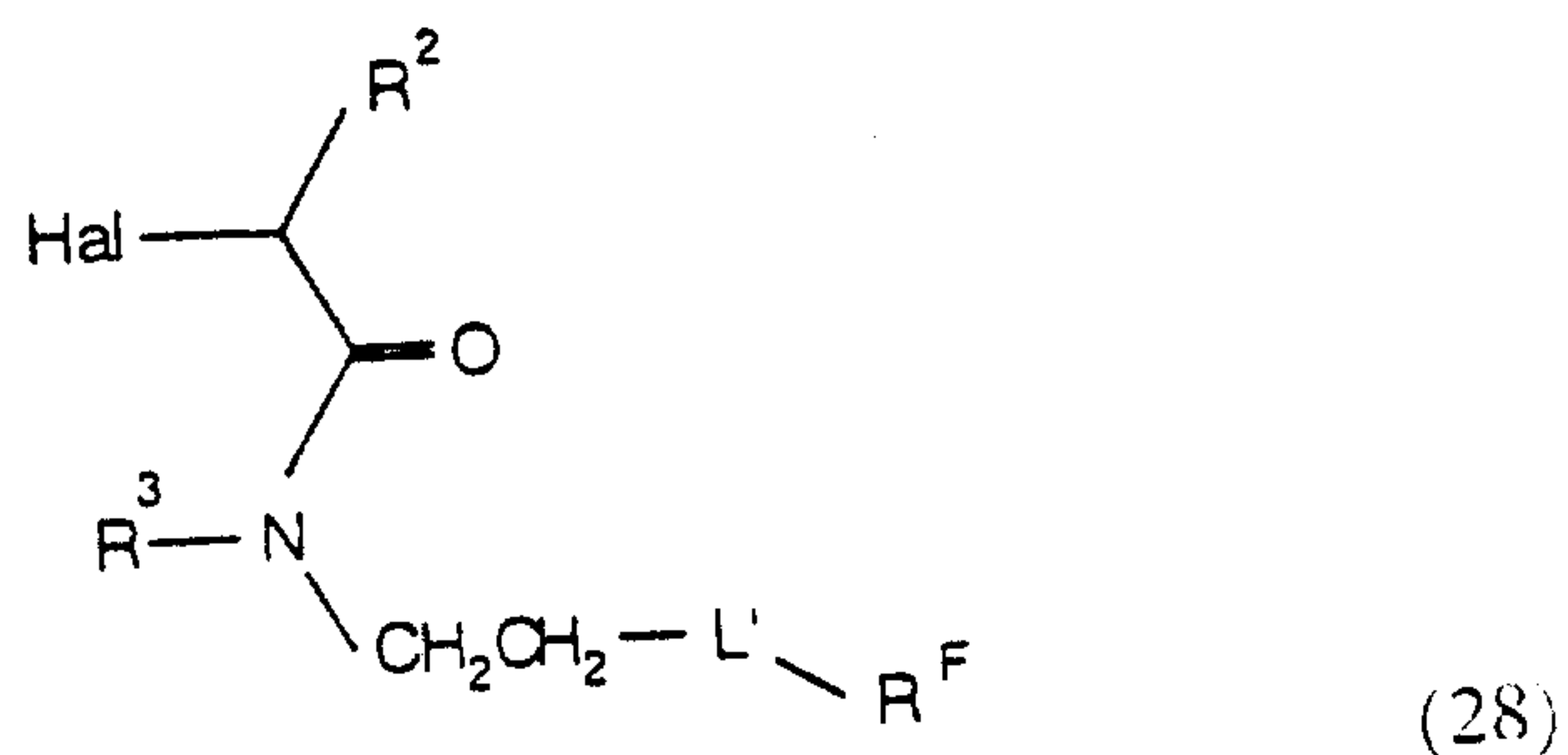
Compounds of general formula I with A in the meaning of
 general formula VIII are obtained from compounds of general

formula 27



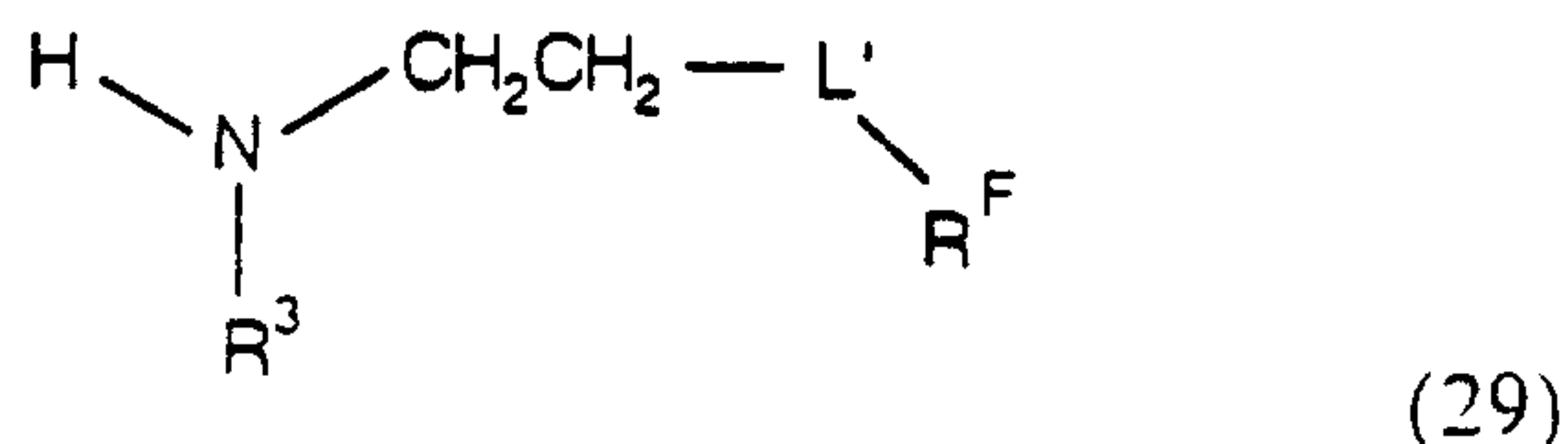
with R^2 , R^3 , R^4 , L' and R^F in the above-mentioned meaning, by cleavage of optionally present protective groups and complexing in the way known to one skilled in the art.

Compounds of general formula 27 are obtained by alkylation of the compounds of general formula 20 with compounds of general formula 28

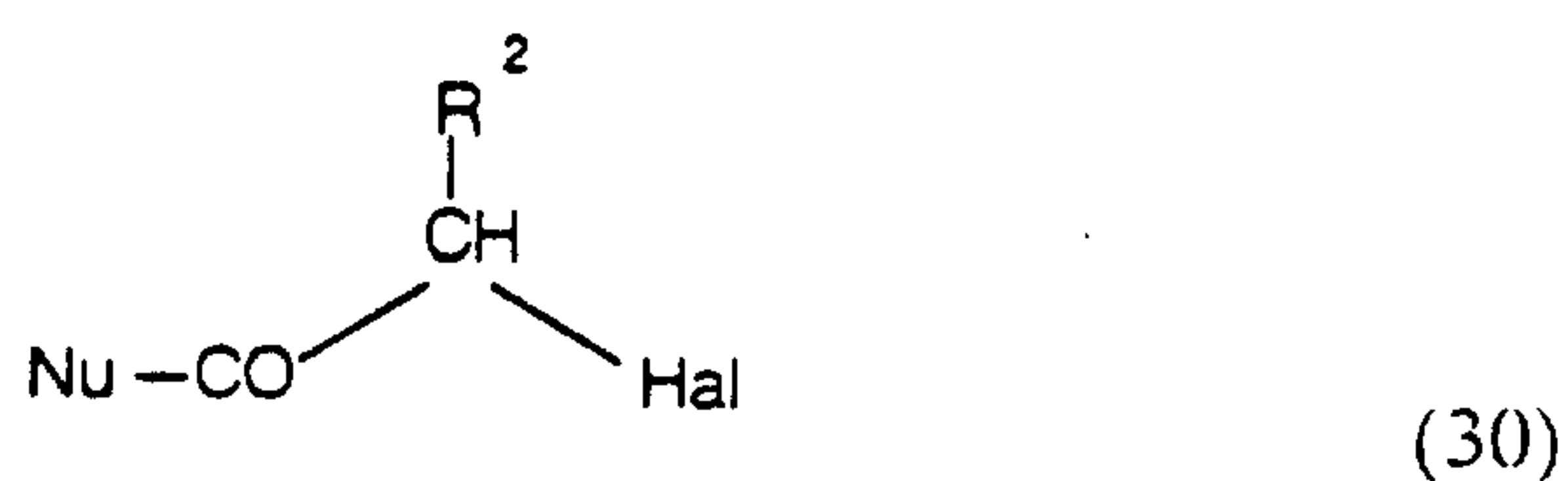


in which Hal, R^2 , R^3 , L' and R^F have the above-mentioned meaning, in a way known in the art, for example as described under EP 0 232 751 B1 (Squibb).

Compounds of general formula 28 are produced from compounds of general formula 29



with L', R³ and R^F in the above-mentioned meaning and an activated halocarboxylic acid of general formula 30



with Nu, R² and Hal in the above-mentioned meaning according to the methods of amide formation via activated carboxylic acids that are known to one skilled in the art [cf. Lit. p. 11].

Compounds of general formula 30 can be obtained from the acids according to C. Hell, Vol. 14: 891 (1881); J. Volhard, A 242, 141 (1887); N. Zelinsky, Vol. 20: 2026, (1887) or from the haloacids according to the activation methods as they are described in general formula 15.

The compounds of general formula 29 can be easily produced according to the methods of amine synthesis that are known to one skilled in the art [Houben-Weyl, Methoden der organischen Chemie, Stickstoffverbindungen II, Amino, 1st Run, Georg Thieme Verlag, Stuttgart, 1957] from the commercially available compounds

(Fluorochem, ABCR) of general formula 31

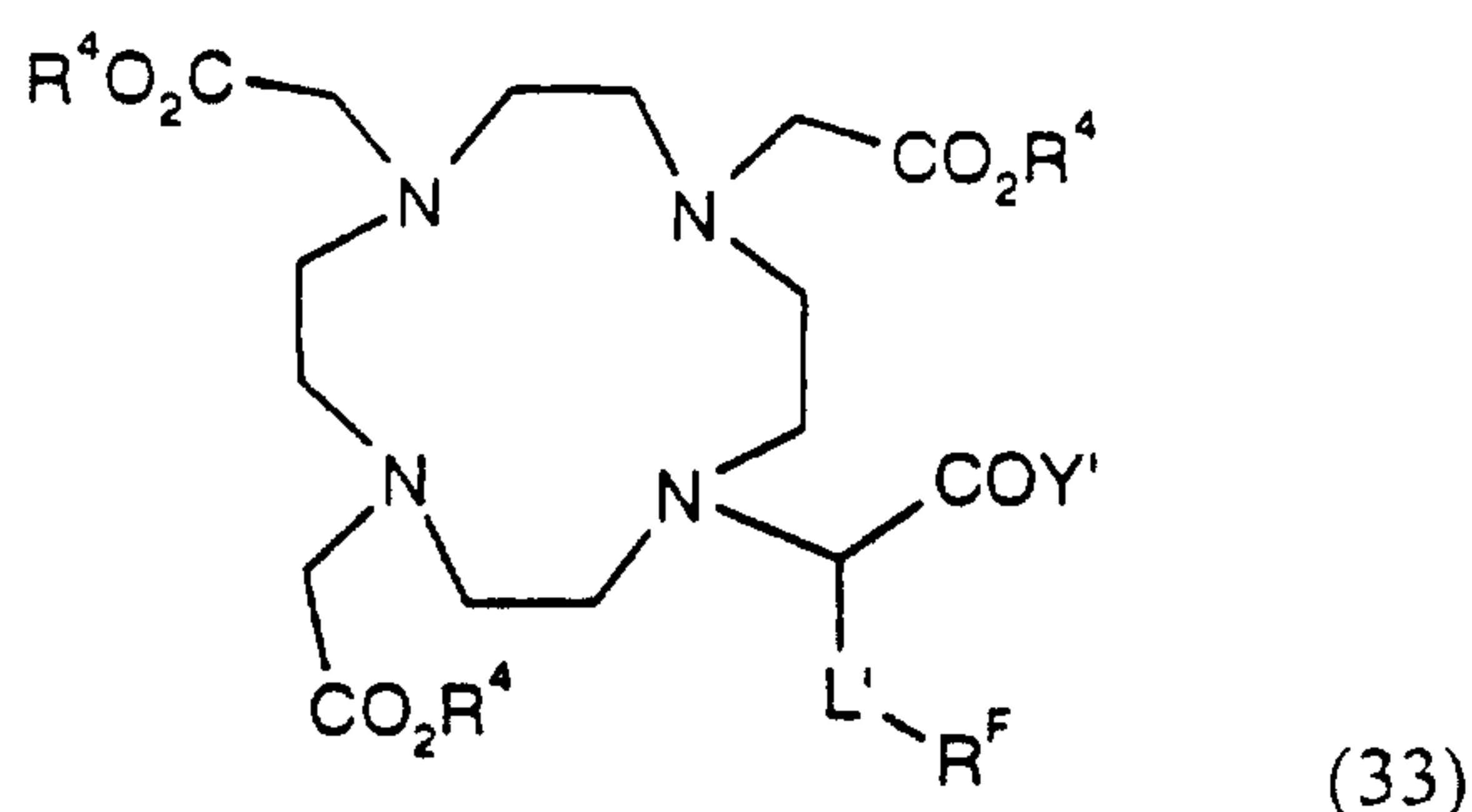


or 32



for example by alkylation of a compound 31 with an amine $\text{PhCH}_2\text{NHR}^3$ and subsequent deprotection of the amino group by catalytic hydrogenation or by Mitsunobu reaction [H. Loibner and E. Zbiral, *Helv.* 59, 2100 (1976), A. K. Bose and B. Lal, *Tetrahedron Lett.* 3973 (1973)] of a compound 32 with potassium phthalimide and deprotection with hydrazine hydrate.

Compounds of general formula I with A in the meaning of general formula VII are obtained from compounds of general formula 33



with

L' , R^{F} and R^4 in the above-mentioned meaning and

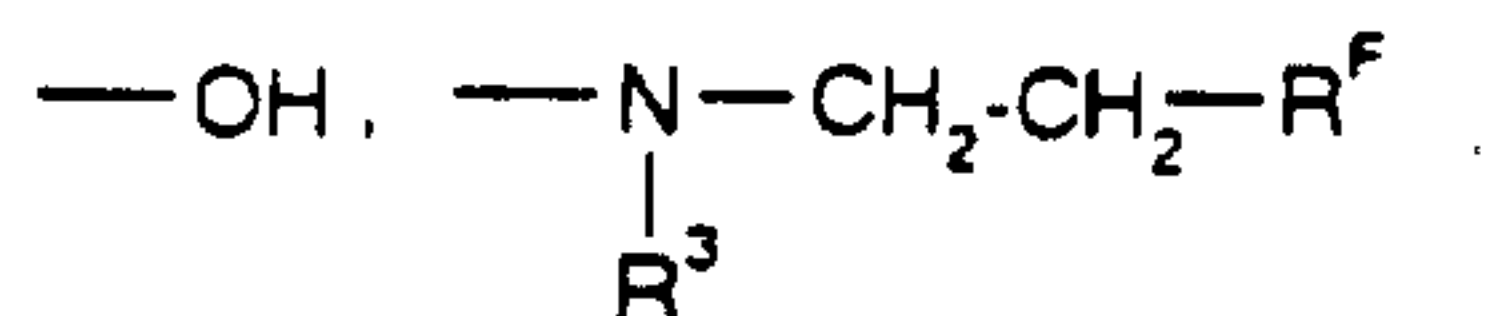
Y' in the meaning of Y, optionally with protective groups, by cleavage of optionally present protective groups and complexing according to the methods that are known to one skilled in the art (Protective Groups in Organic Synthesis, 2nd Edition, T. W. Greene and P. G. M. Wuts, John Wiley & Sons, Inc., New York, 1991; EP 0 130 934, EP 0 250 358).

Compounds of general formula 33 are obtained from compounds of general formula 20 and compounds of general formula 34



in which

Hal', L', R^F have the above-mentioned meaning and Y' stands for the radical

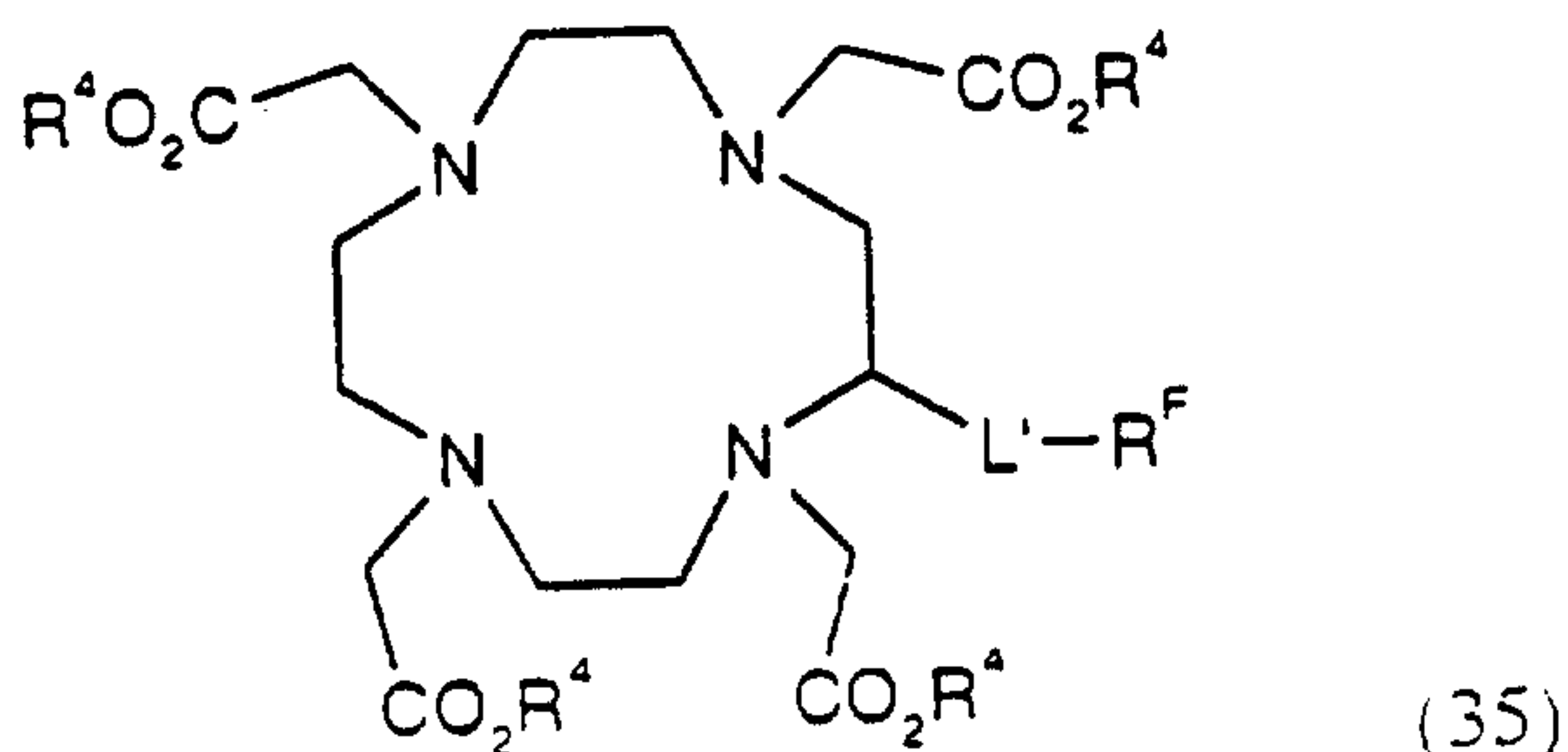


in a way known in the art, for example as described in EP 0 232 751 B1, EP 0 292 689 A2 (both Squibb) or EP 0 255 471 A1 (Schering).

The production of compounds of general formula 34 is carried out according to known methods, for example, according to Hell-Volhard-Zelinsky from commercially available precursors (ABCR).

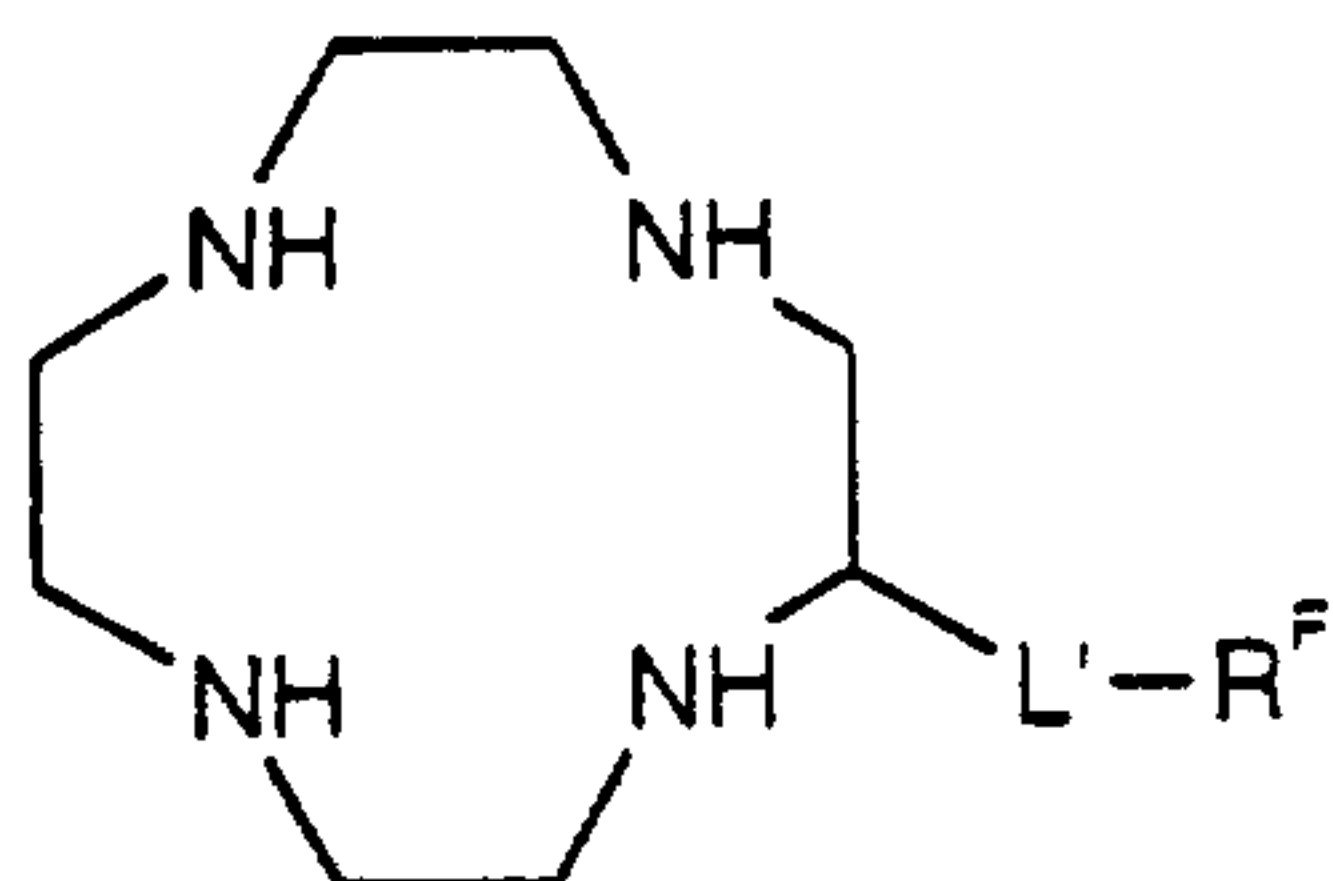
Compounds of general formula I with A in the meaning of general formula VI are obtained from compounds of general formula

35



in which L' , R^4 and R^f have the above-mentioned meaning, by, if appropriate, cleavage of protective groups and complexing in a way known in the art [Protective Groups in Organic Synthesis, 2nd Edition, T. W. Greene and P. G. M. Wuts, John Wiley & Sons, Inc., New York, 1991 (EP 0 130 934, EP 0 250 358)].

Compounds of general formula 35 are obtained by reacting α -halocarboxylic acid esters or α -halocarboxylic acids of general formula 18 with compounds of general formula 36

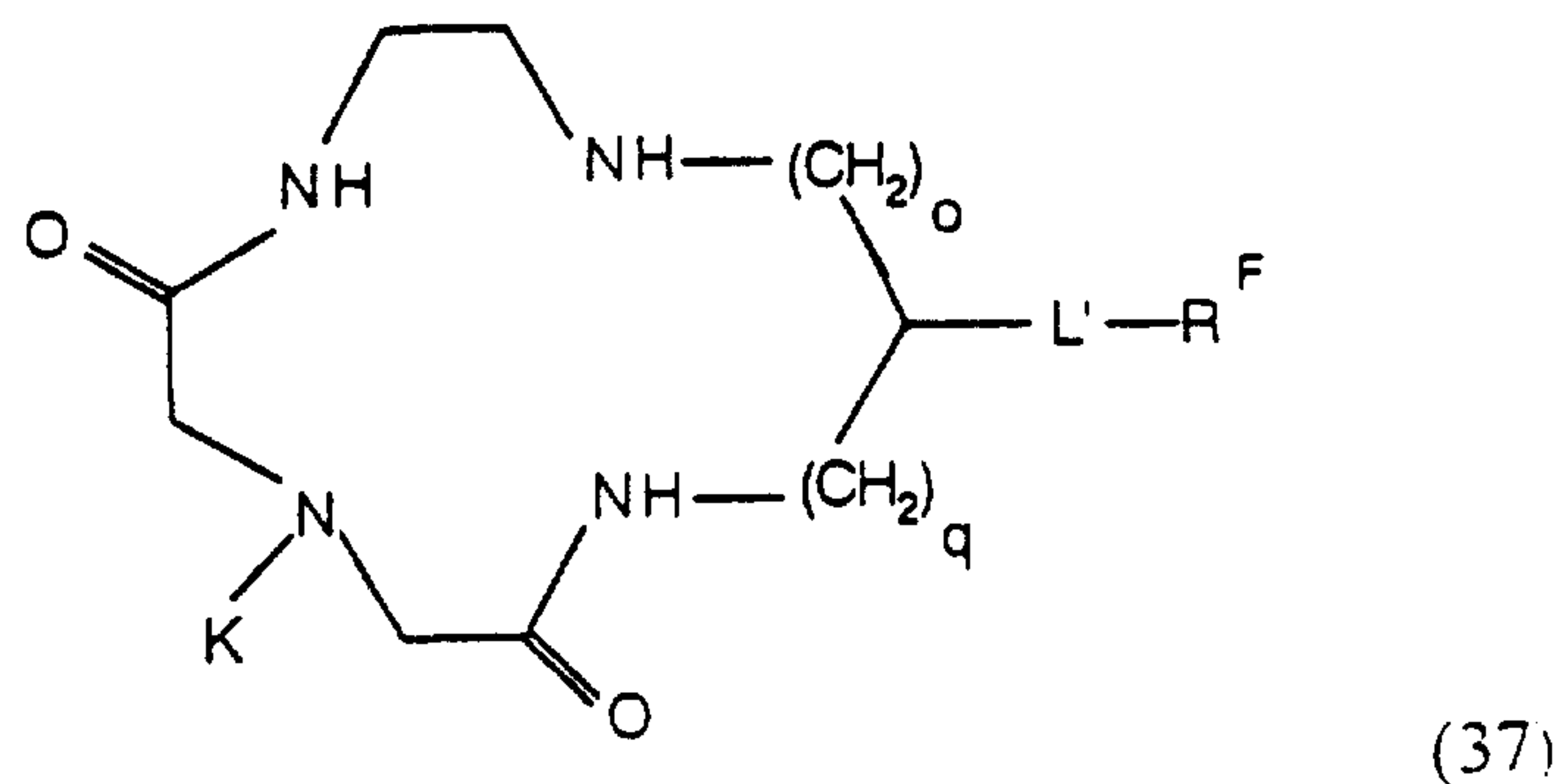


(36)

with L' and R^f in the above-mentioned meaning, according to the methods that are known to one skilled in the art, as described in, for example, EP 0 255 471 or US 4,885,363.

Compounds of general formula 36 can be obtained by cleavage of optionally present protective groups and subsequent reduction with diborane according to the known processes from compounds of

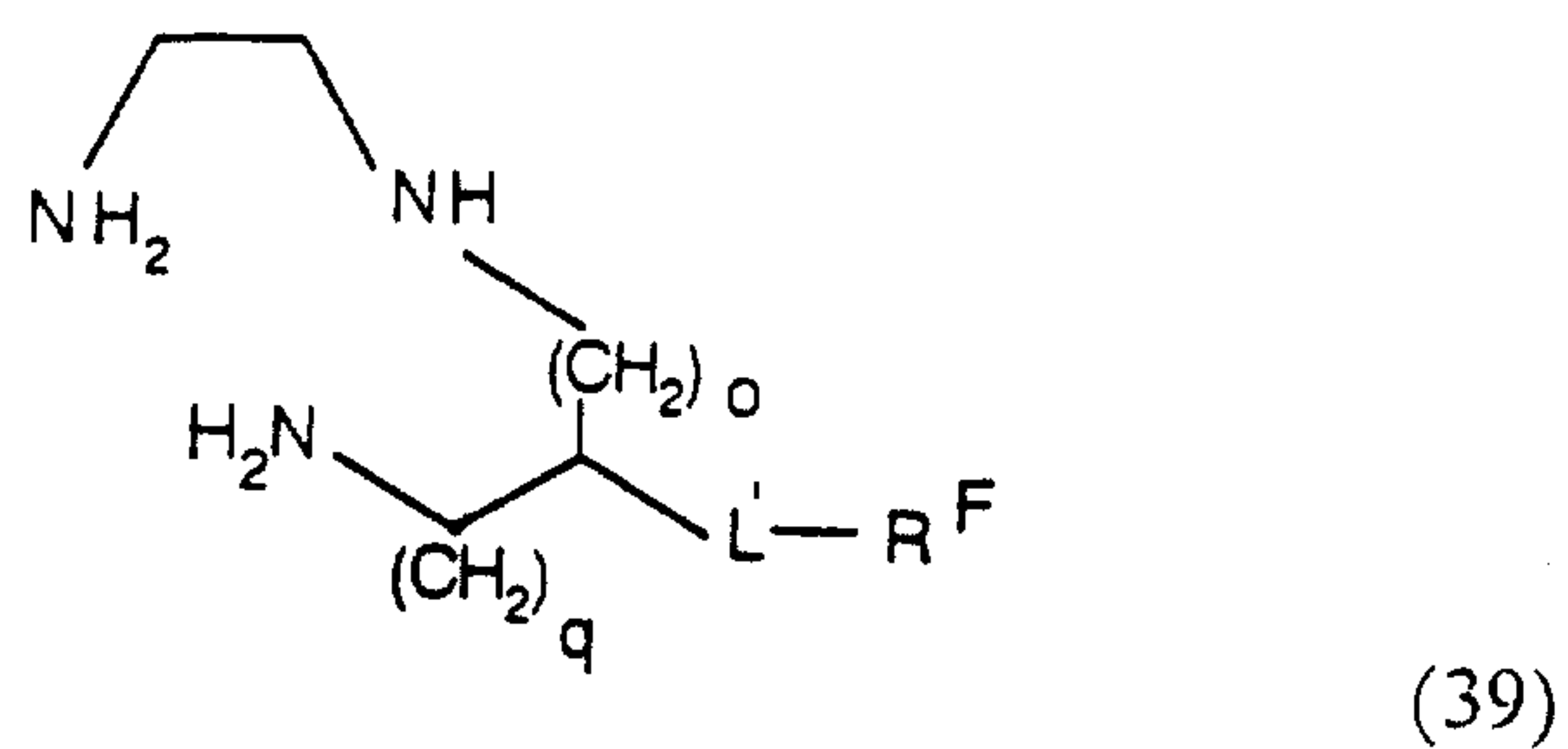
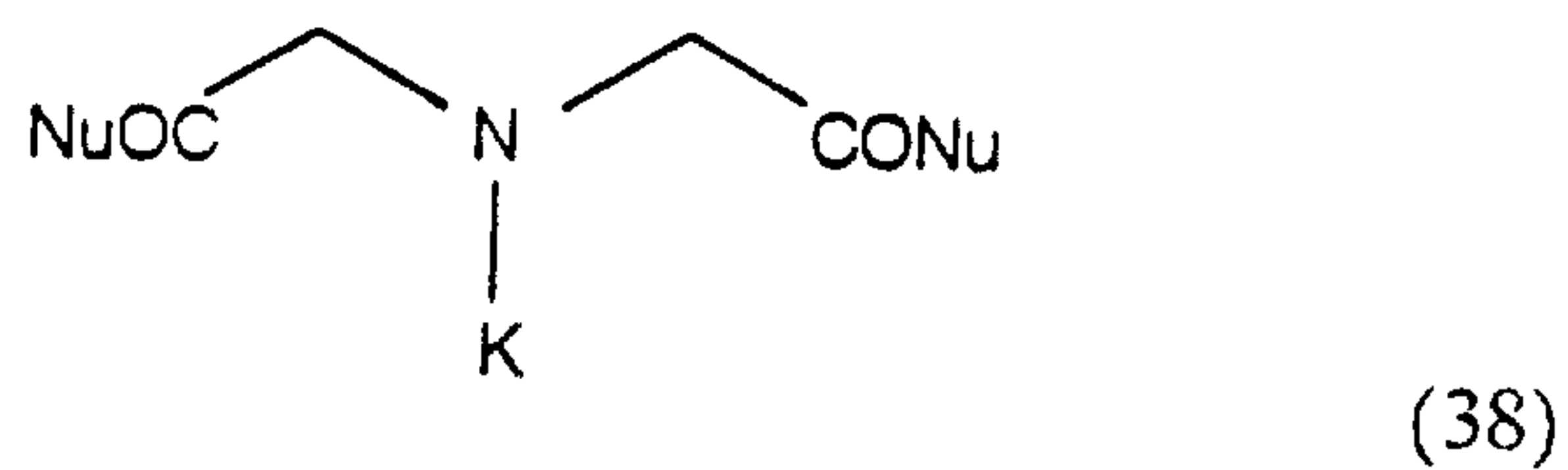
general formula 37



in which

L' , R^F , o , q , have the above-mentioned meaning and
 K has the meaning of a protective group.

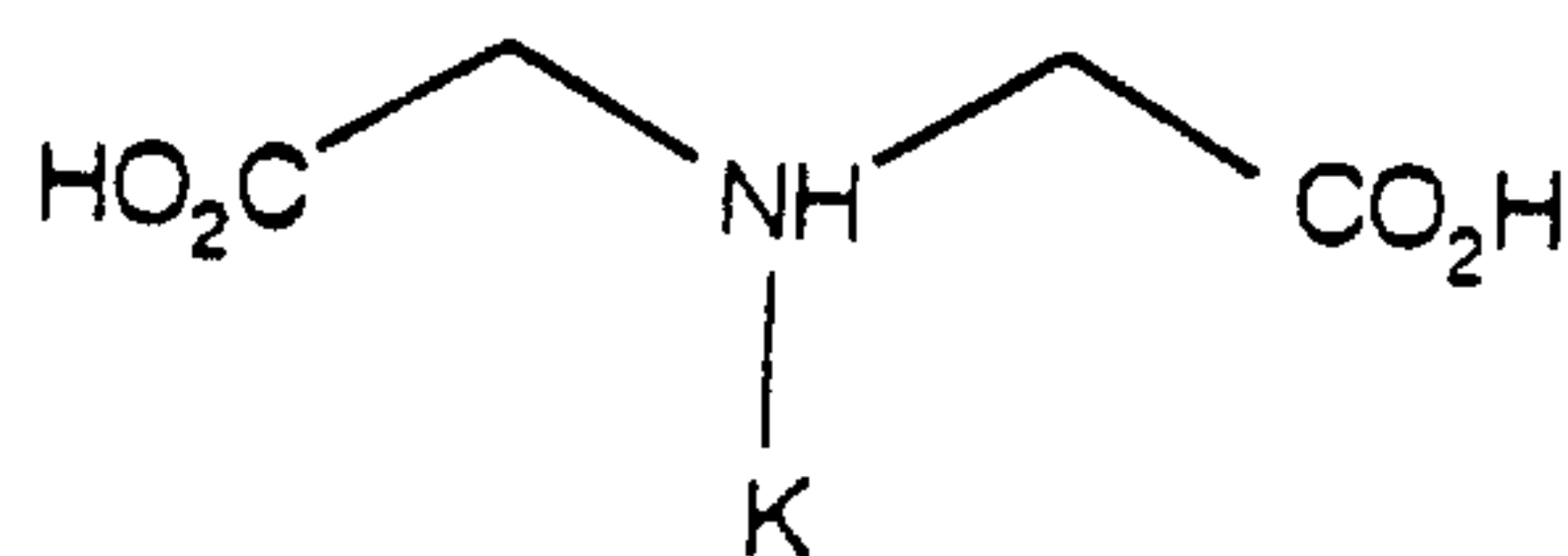
The compounds of general formula 37 are available by a condensation reaction from an activated, N-protected iminodiacetic acid 38 and amine 39:



in which

L' , R^f , o , q , Nu and K have the above-mentioned meaning. As a nucleofuge, preferably the N-hydroxysuccinimide is used; as a protective group, the benzyloxycarbonyl, trifluoroacetyl or t-butyloxycarbonyl group is used.

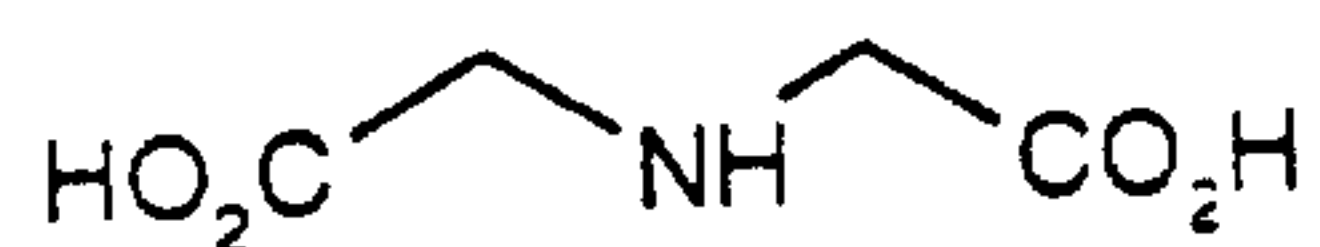
Compounds of general formula 38 can be obtained according to the processes of protecting the amino group and of activating carboxylic acid that are known to one skilled in the art [Protective Groups, Aktivierung von Carboxylgruppen [Activation of Carboxyl Groups], p. 11] with protected iminodiacetic acid 40



(40)

in which

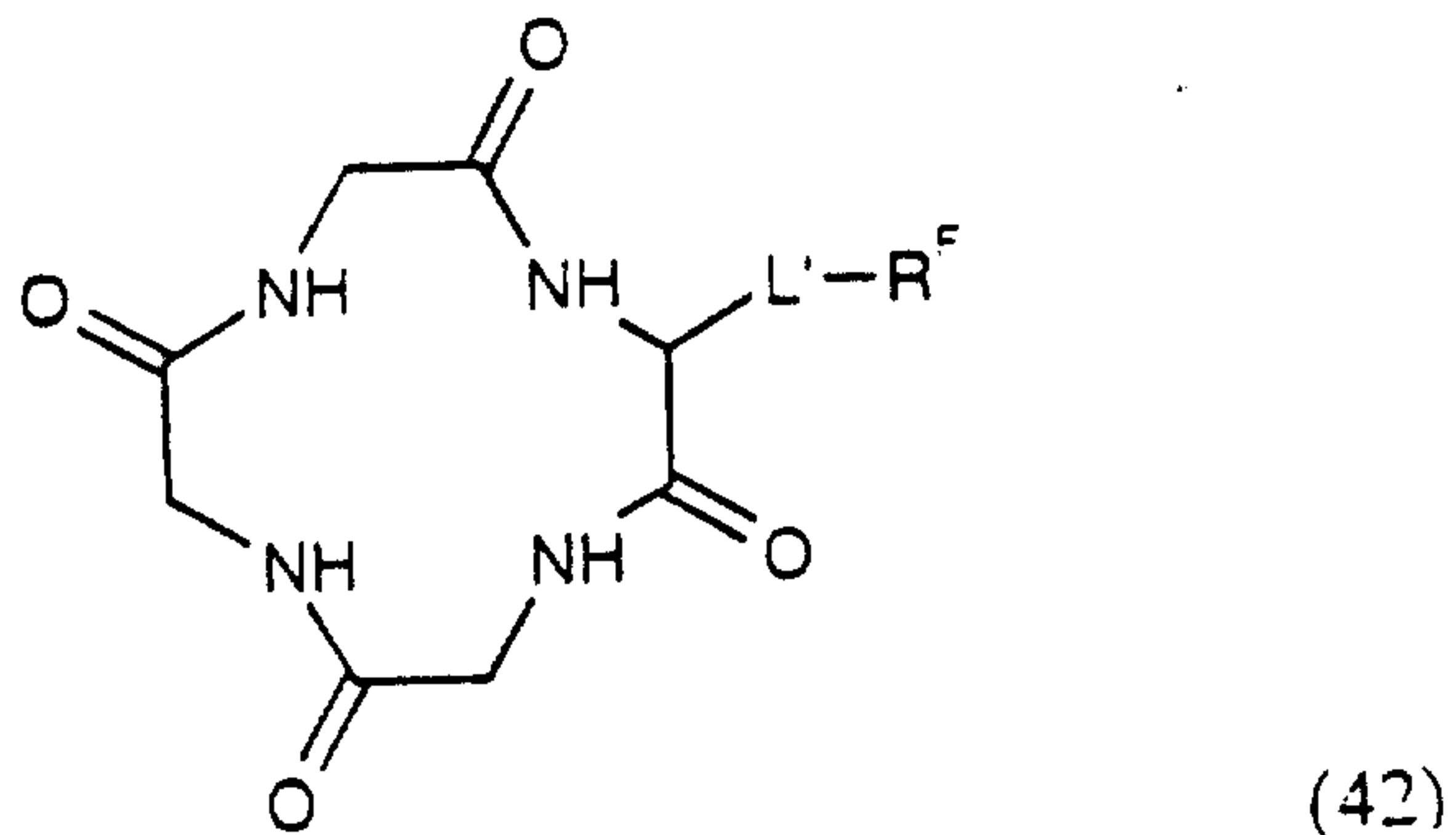
K has the meaning of a protective group, from iminodiacetic acid 41



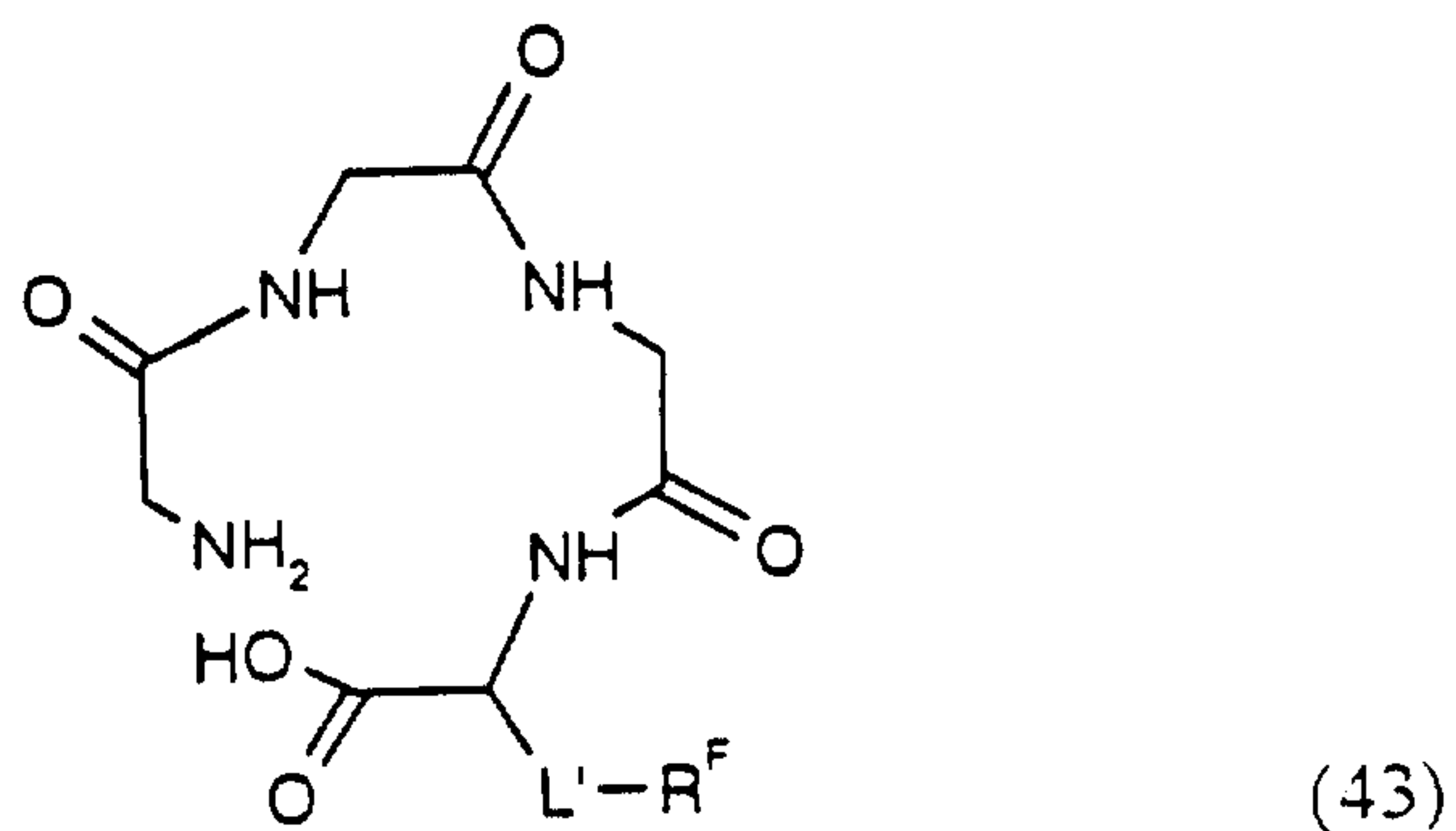
(41)

As an alternative, compounds of general formula 36 are available by, if appropriate, cleavage of protective groups and reduction with diborane according to the process described in 37

from compounds of general formula 42

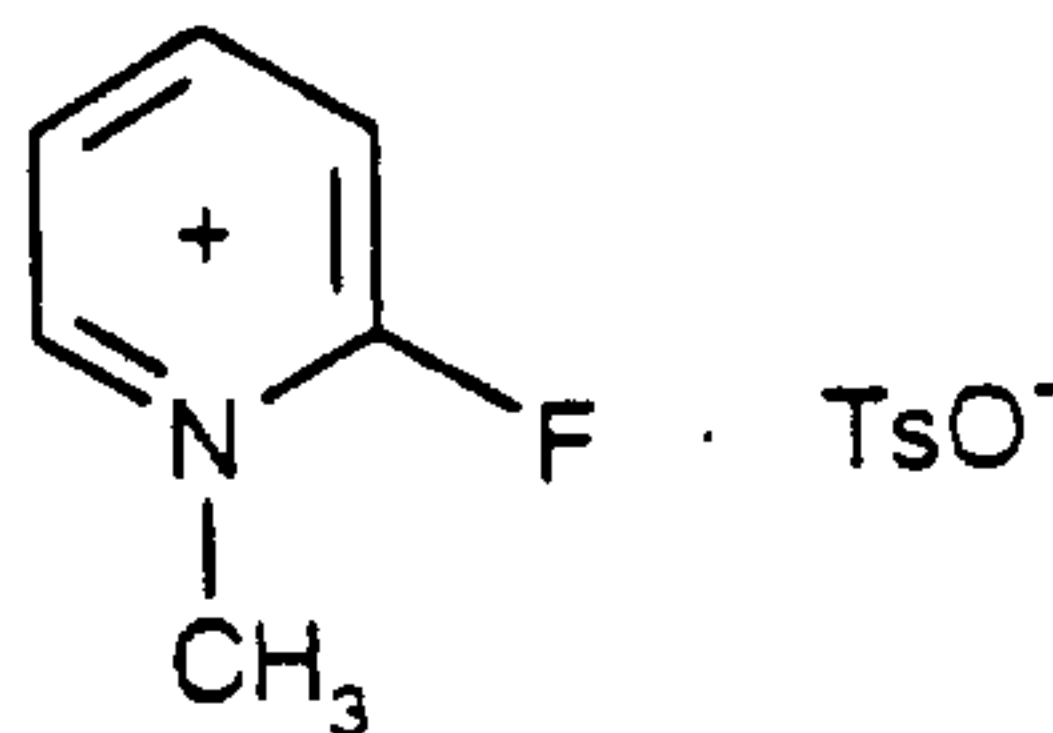


Compounds of general formula 42 can be obtained by closing the rings of Secco compounds 43

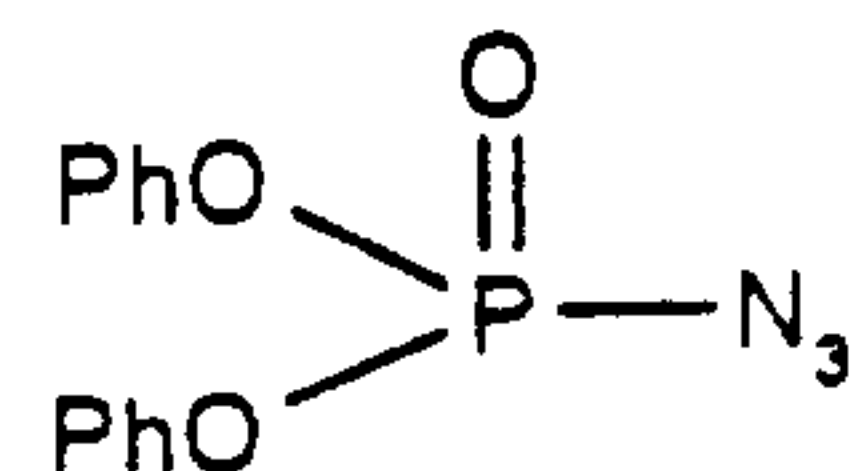


in which

L' and R^F have the above-mentioned meaning, according to standard processes; for example, by reaction with the Mukaiyama reagent 2-fluoro-1-methylpyridinium-tosylate

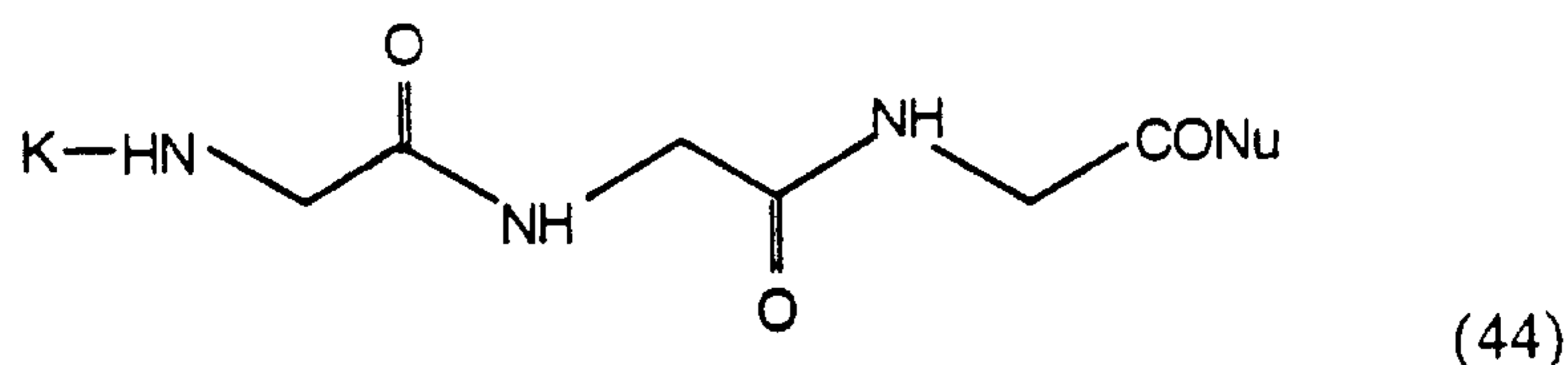


[J. Org. Chem. 1994, 59, 415: Synthetic Communications 1995, 25, 1401] or with the phosphoric acid diphenylester-azide

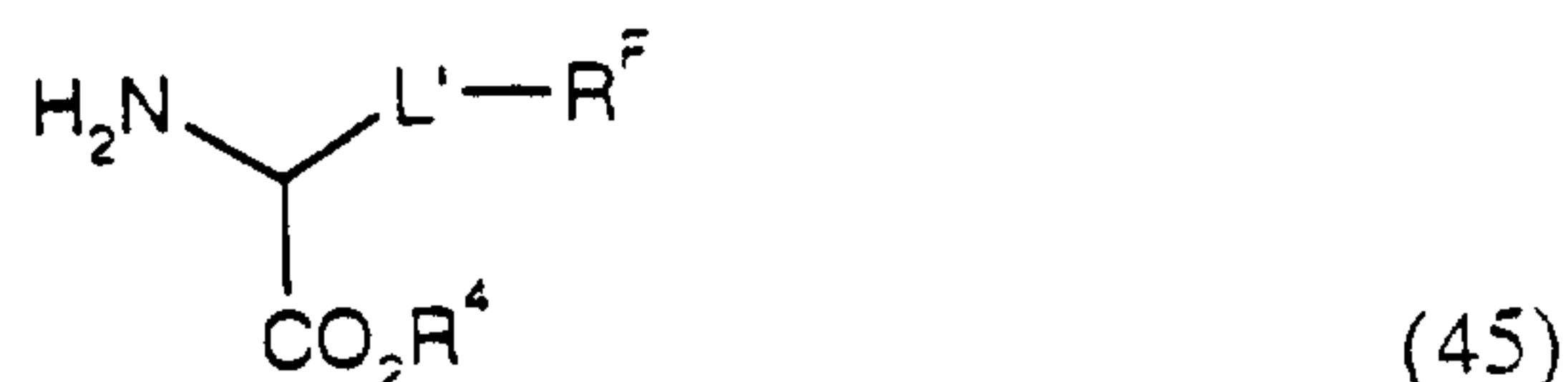


[J. Am. Chem. Soc. 1993, 115, 3420; WO 94/15925].

Compounds of general formula 43 are available according to the described processes by condensation of activated acid 44



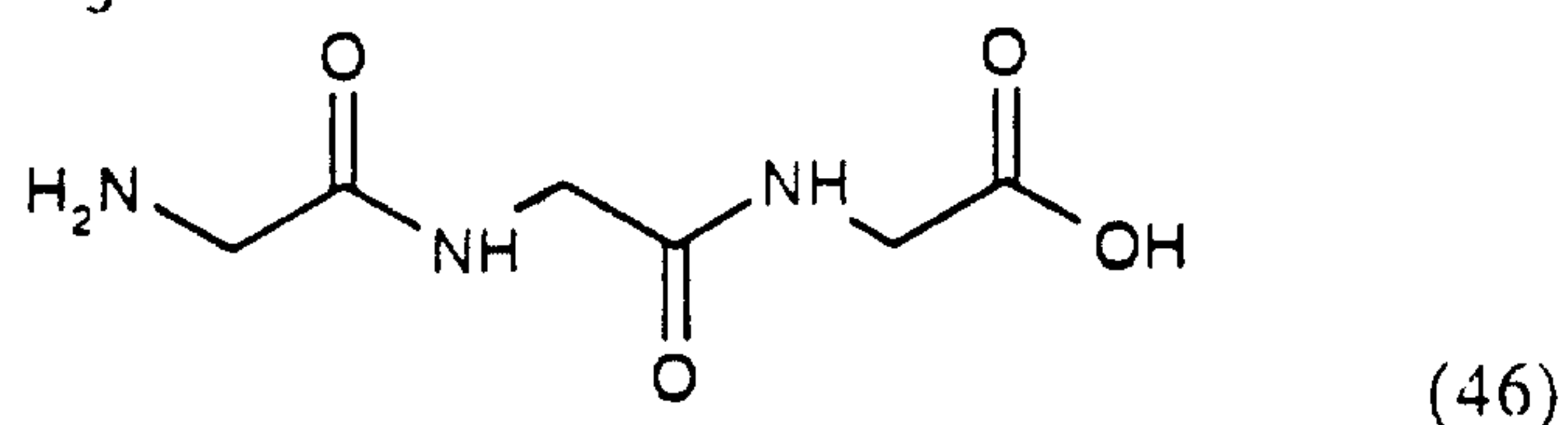
with Nu and K in the above-mentioned meaning, with a compound of general formula 45



in which

L', R⁴ and R^f have the above-mentioned meaning.

Compounds of general formula 44 are available from commercially available triglycine (Bachem, Fluka) 46

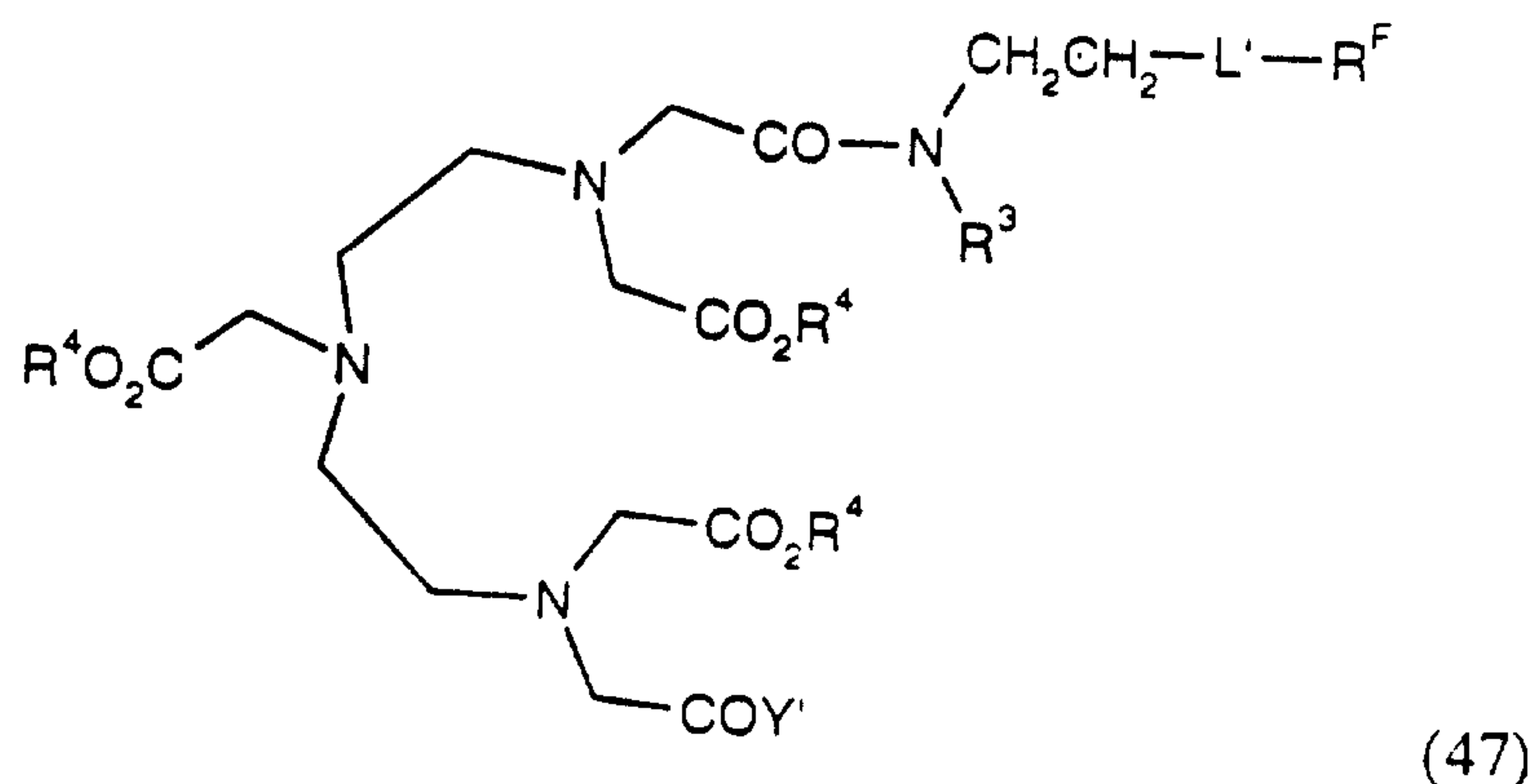


by protection of the amino group with subsequent activation of the acid function according to the processes for amine protection

and carboxylic acid reactivation that are known to one skilled in the art (Lit. p. 12).

The compounds of general formula 45 can be easily obtained from compounds of general formula 62 by introducing protective group R^4 according to the methods known to one skilled in the art -- for example, re-esterification of a sulfite ester.

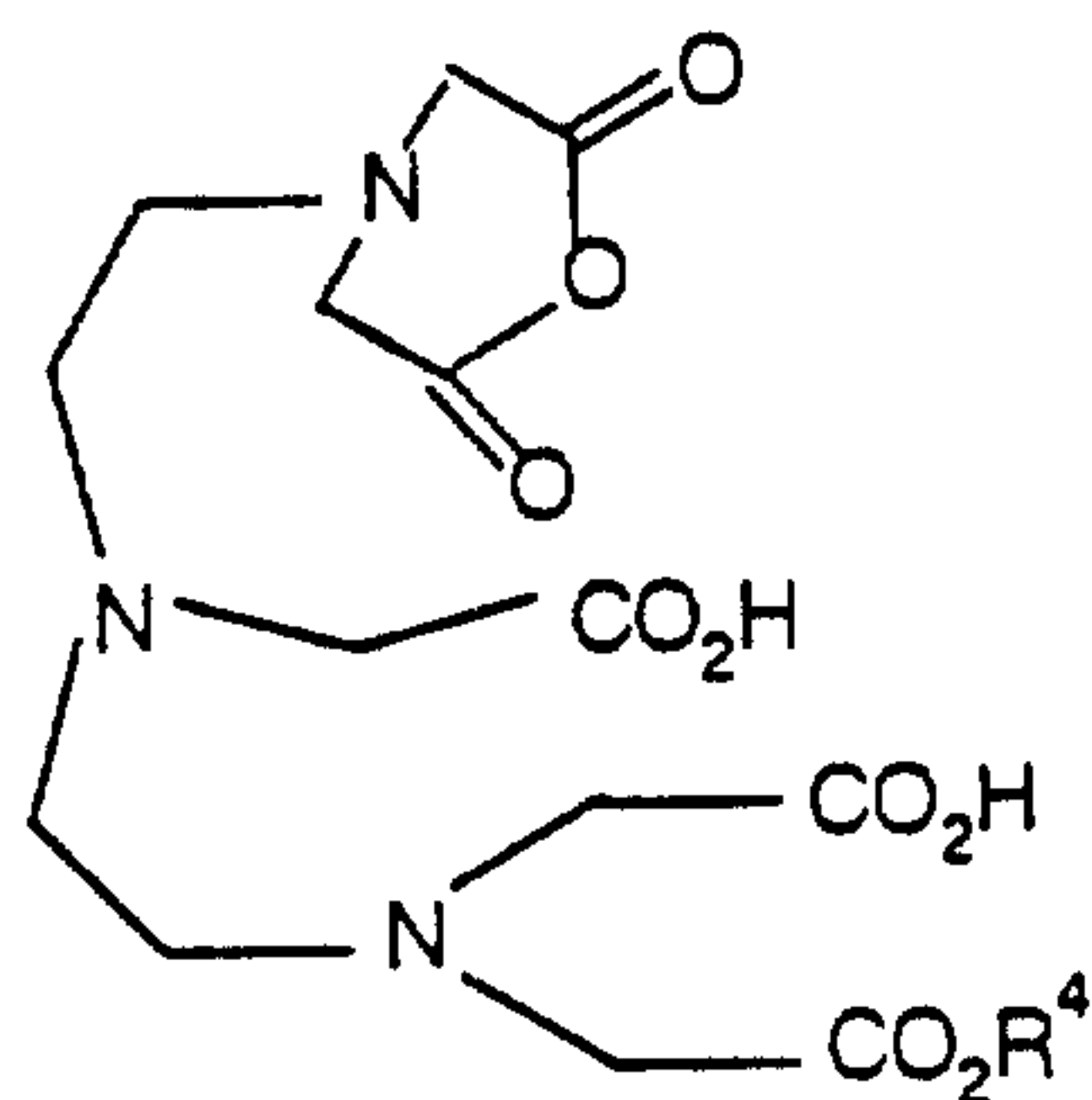
Compounds of general formula I with A in the meaning of general formula II are obtained from compounds of general formula 47



with L' , R^3 , R^4 , R^f and Y' in the above-mentioned meaning, by, if appropriate, cleavage of protective groups and complexing in a way that is well-known to one skilled in the art (Protective Groups, EP 0 250 358, EP 0 130 934).

If Y' in general formula 47 means an OH group, the compounds

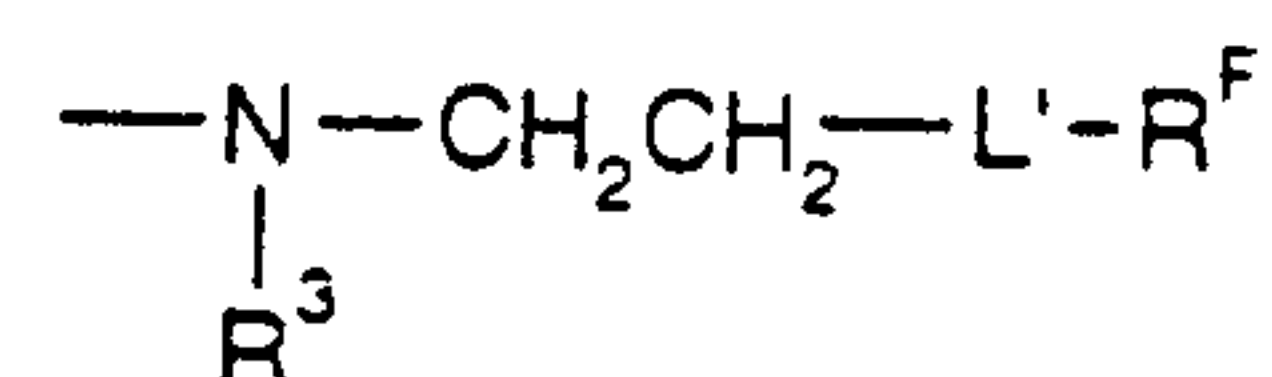
are obtained by reacting a compound 48



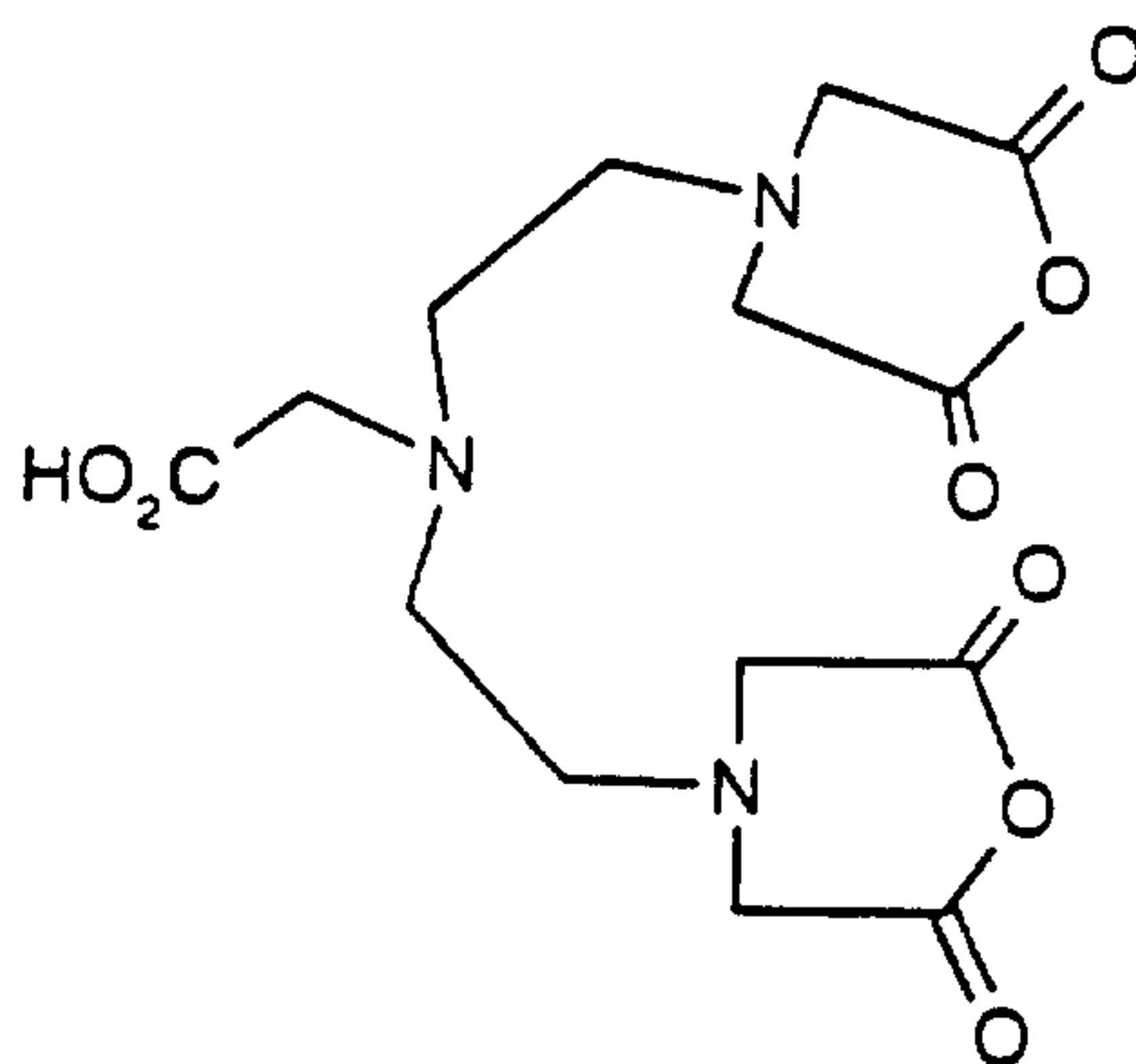
(48)

with R^4 in the above-mentioned meaning, produced according to DE 3 633 243, with an amine of general formula 29 under the conditions already described and subsequent cleavage of the protective groups.

If Y' in formula 47, however, is the group



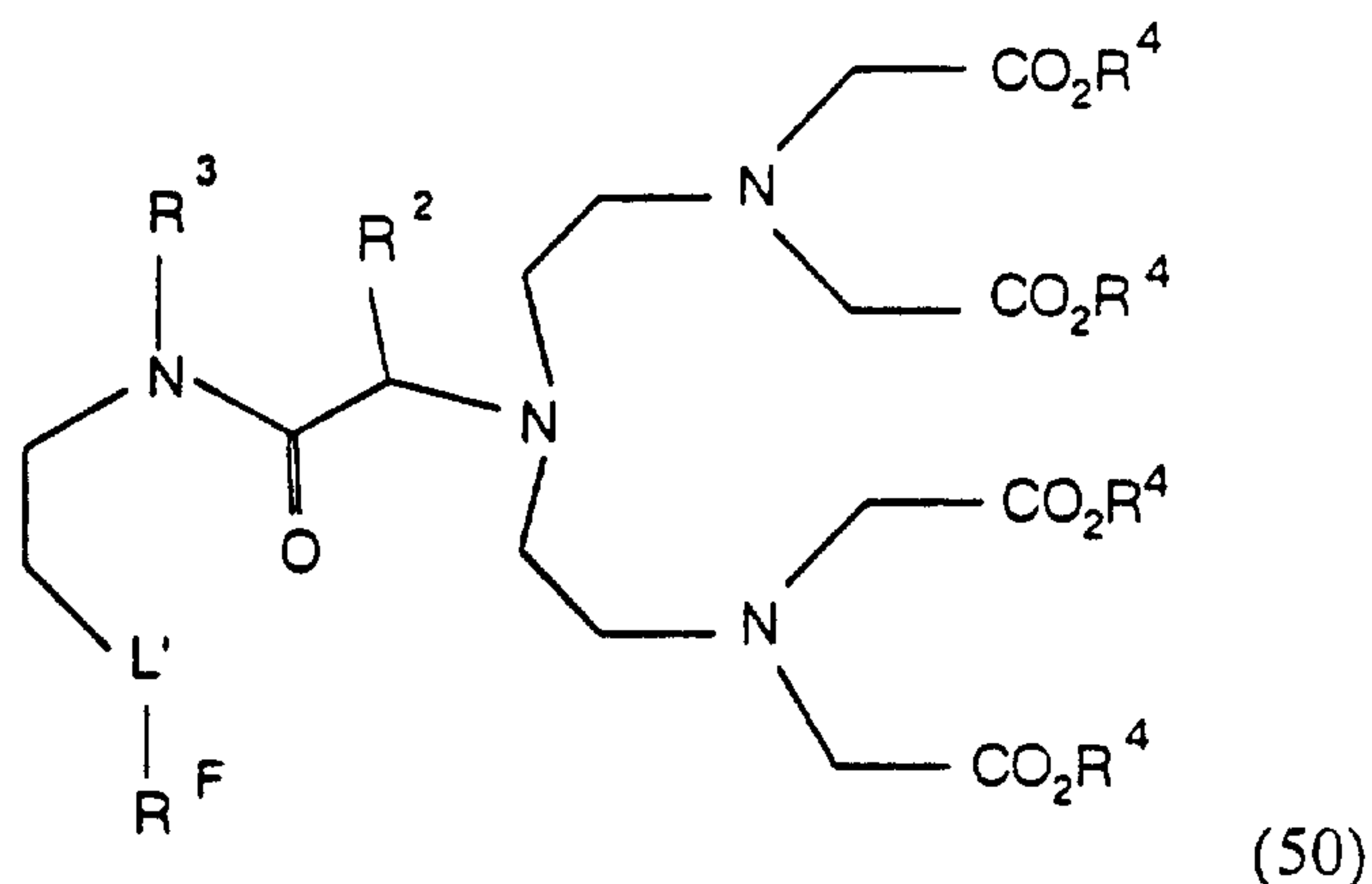
then the reaction is performed under analogous conditions with DTPA-bisanhydride (commercially available product, Merck) 49



(49)

Compounds of general formula I, with A in the meaning of general formula III, are obtained from compounds of general

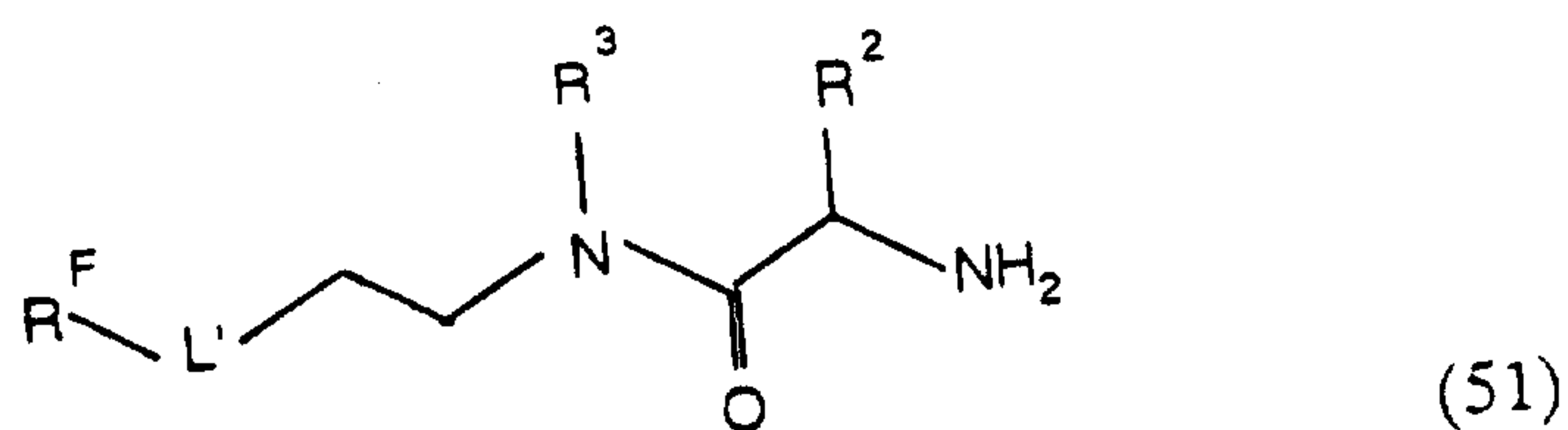
formula 50



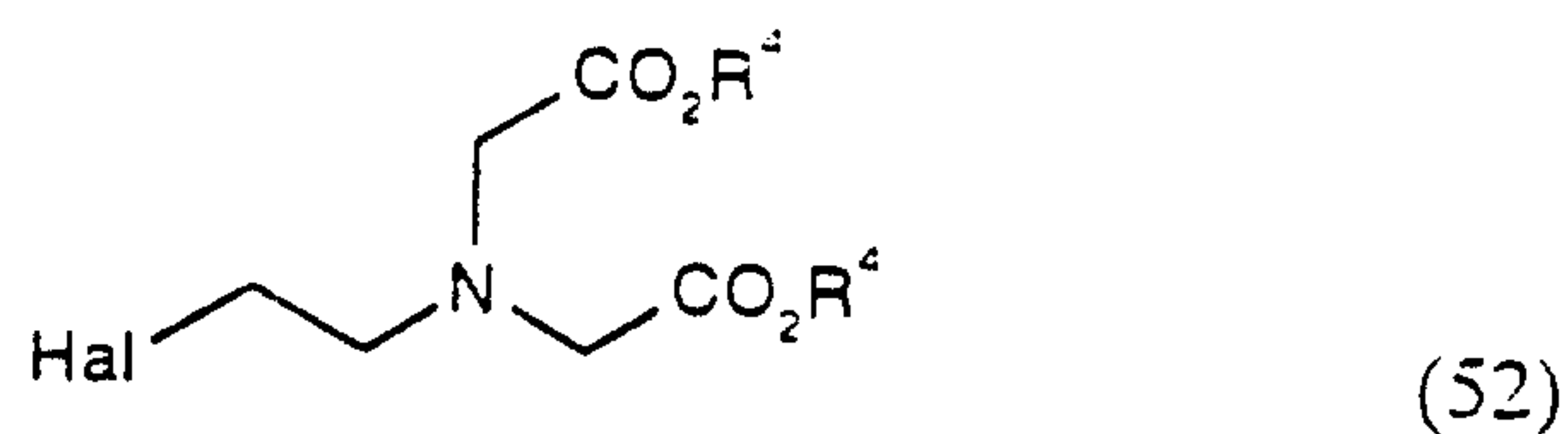
in which

L' , R^2 , R^3 , R^4 and R^F have the above-mentioned meaning, by, if appropriate, cleavage of protective groups and complexing in a way that is well-known to one skilled in the art [Protective Groups, EP 0 071564, EP 0 130 934, DE-OS 3 401 052].

Compounds of general formula 50 are obtained according to the process described in J. Org. Chem. 1993, 58: 1151 from compounds of general formula 51



and halocarboxylic acid derivatives of formula 52

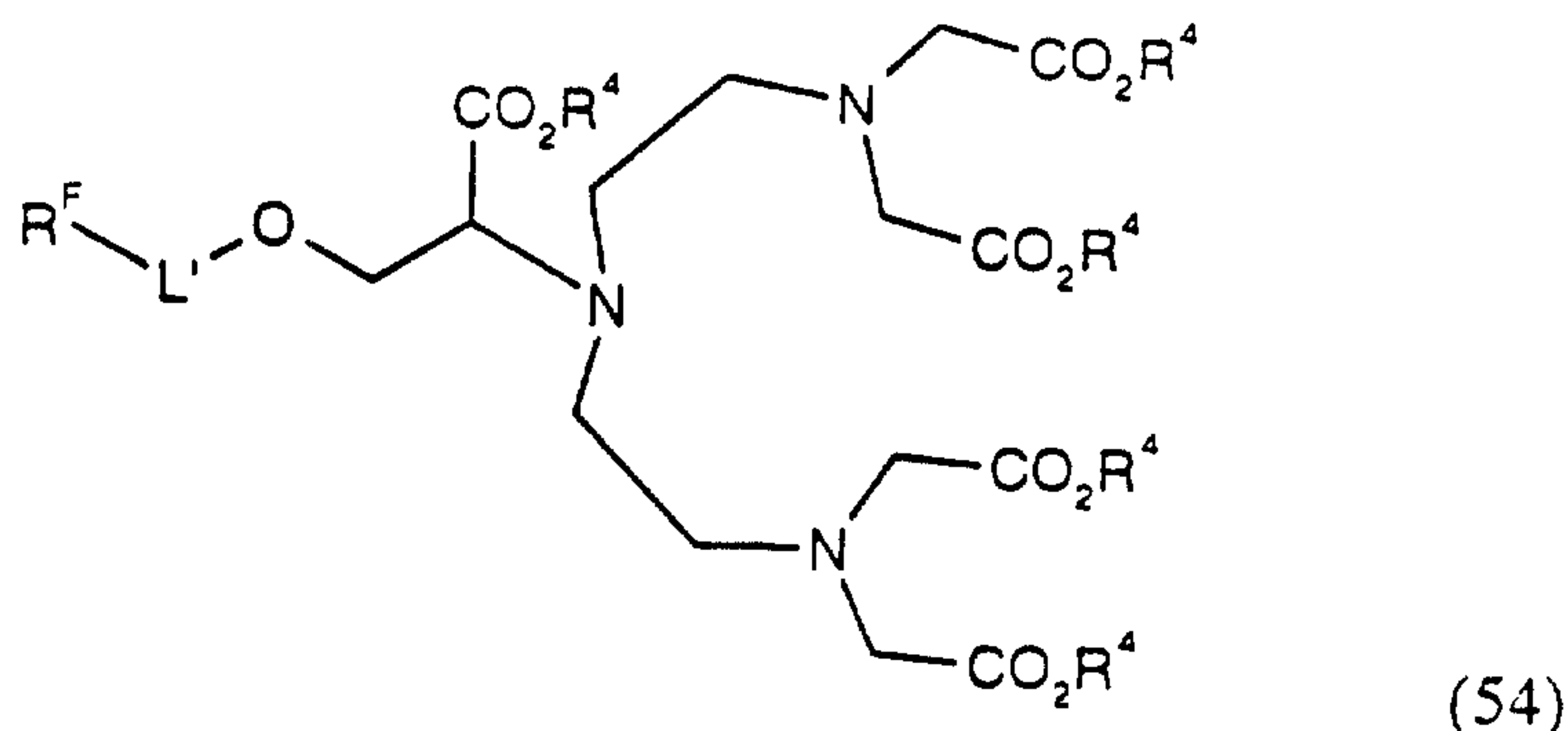


in which R^4 and Hal have the already described meaning. The compounds of general formula 51 are produced by acylation of an amine of general formula 29 with an activated N-protected amino acid of general formula 53



in which Nu has the above-mentioned meaning and K is in the meaning of a protective group such as Z, -BOC, FMOC, -COCF₃, and subsequent cleavage of the protective group.

Compounds of general formula I with A in the meaning of general formula IV are obtained from compounds of general formula 54



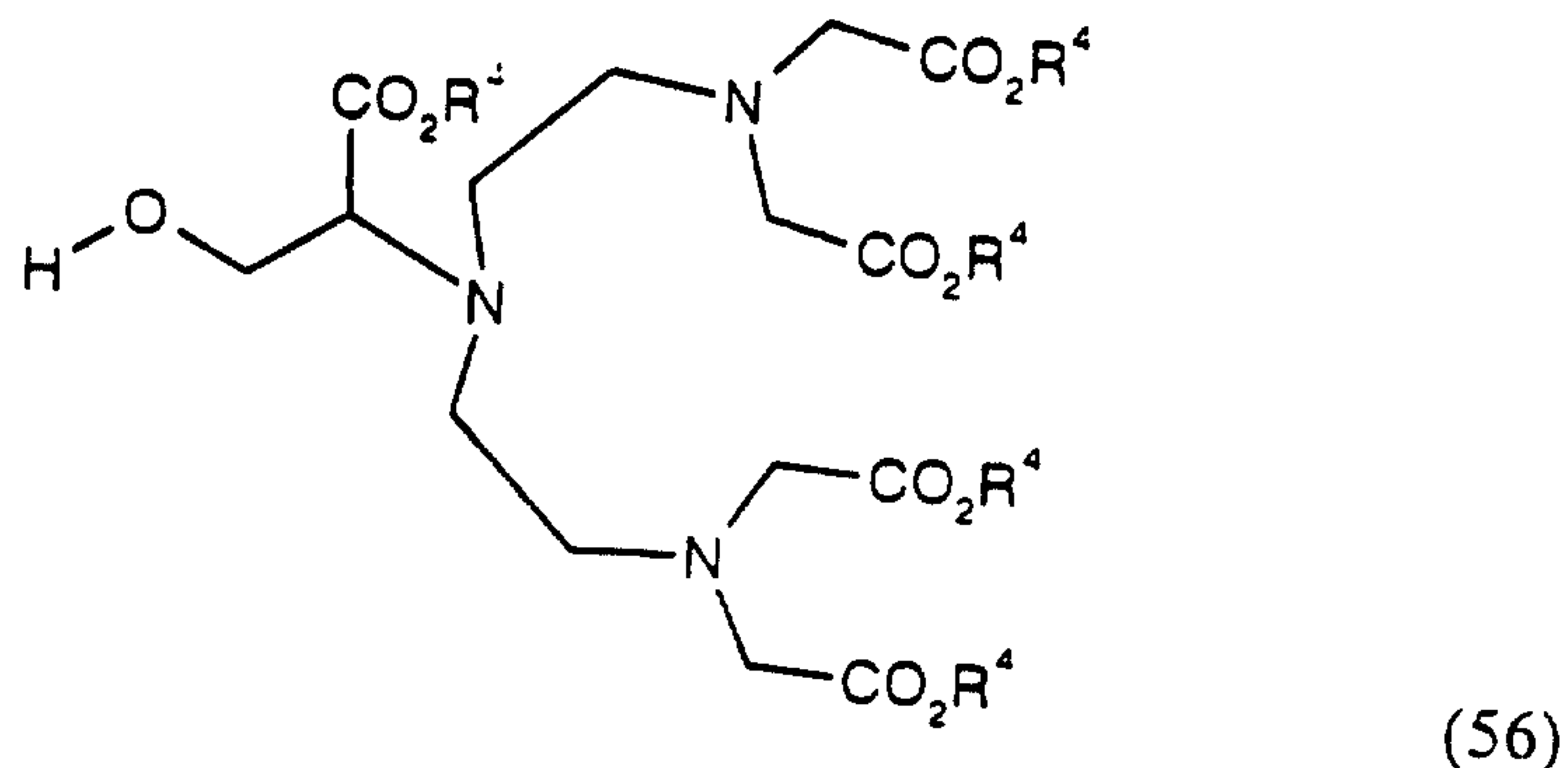
in which

L' , R^F and R^4 have the above-mentioned meaning, by, if appropriate, cleavage of the protective groups and complexing according to a method that is known to one skilled in the art, as already described [Protective Groups, EP 0 071 564, EP 0 130 934, DE-OS 3 401 052].

Compounds of general formula 54 can be obtained in a known way from the halogen compounds of general formula 55



that can be obtained as commercially available products (Fluorochem, ABCR) by reaction with hydroxy acids 56

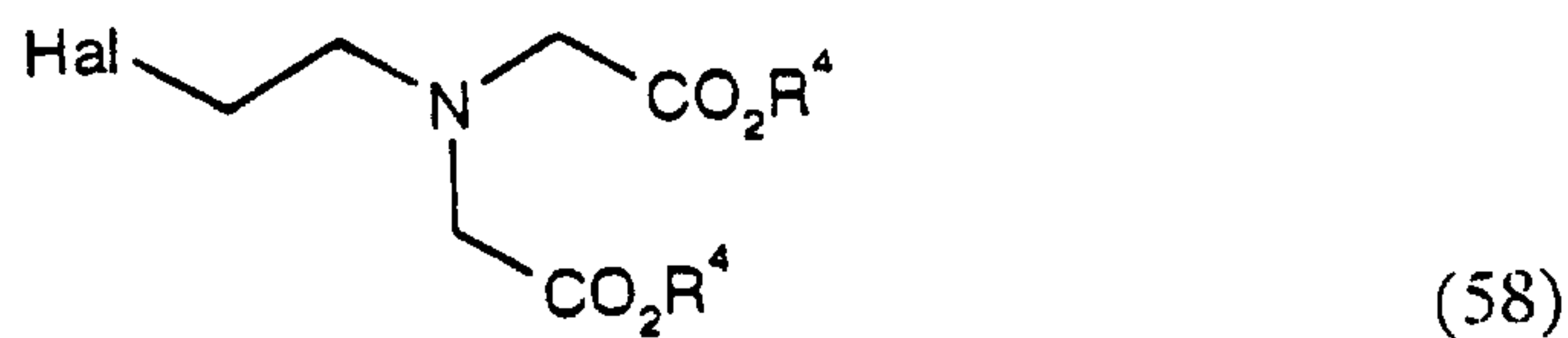


in which

R⁴ has the above-mentioned meaning. The compounds of formula 56 can be obtained in a way known in the art according to J. Org. Chem. 58, 1151 (1993) from commercially available serine ester 57 (Bachem, Fluka)

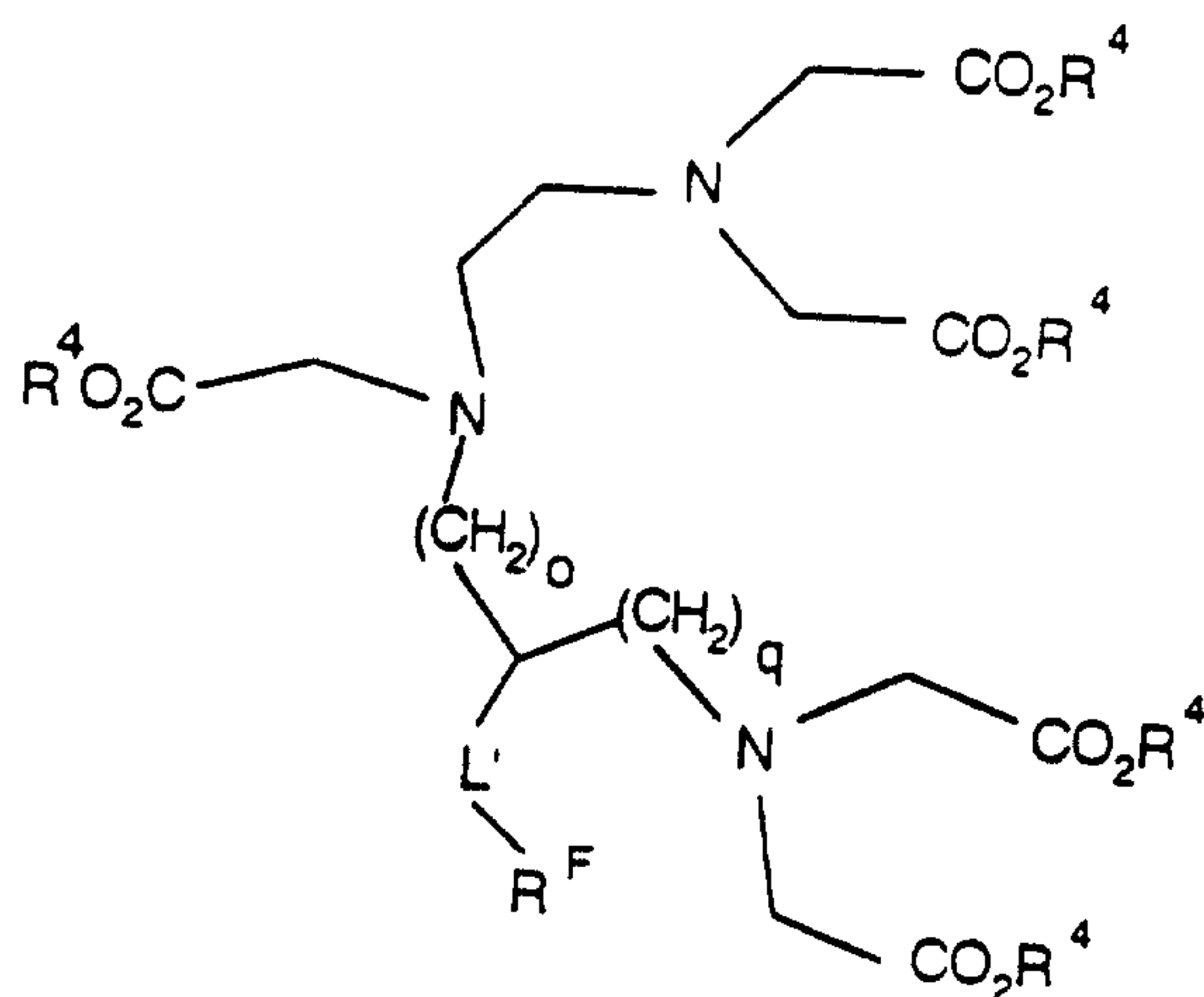


with R⁴ in the above-mentioned meaning and halocarboxylic acid esters 58



Compounds of general formula I with A in the meaning of

general formula V are obtained from compounds of general formula 59



(59)

in which

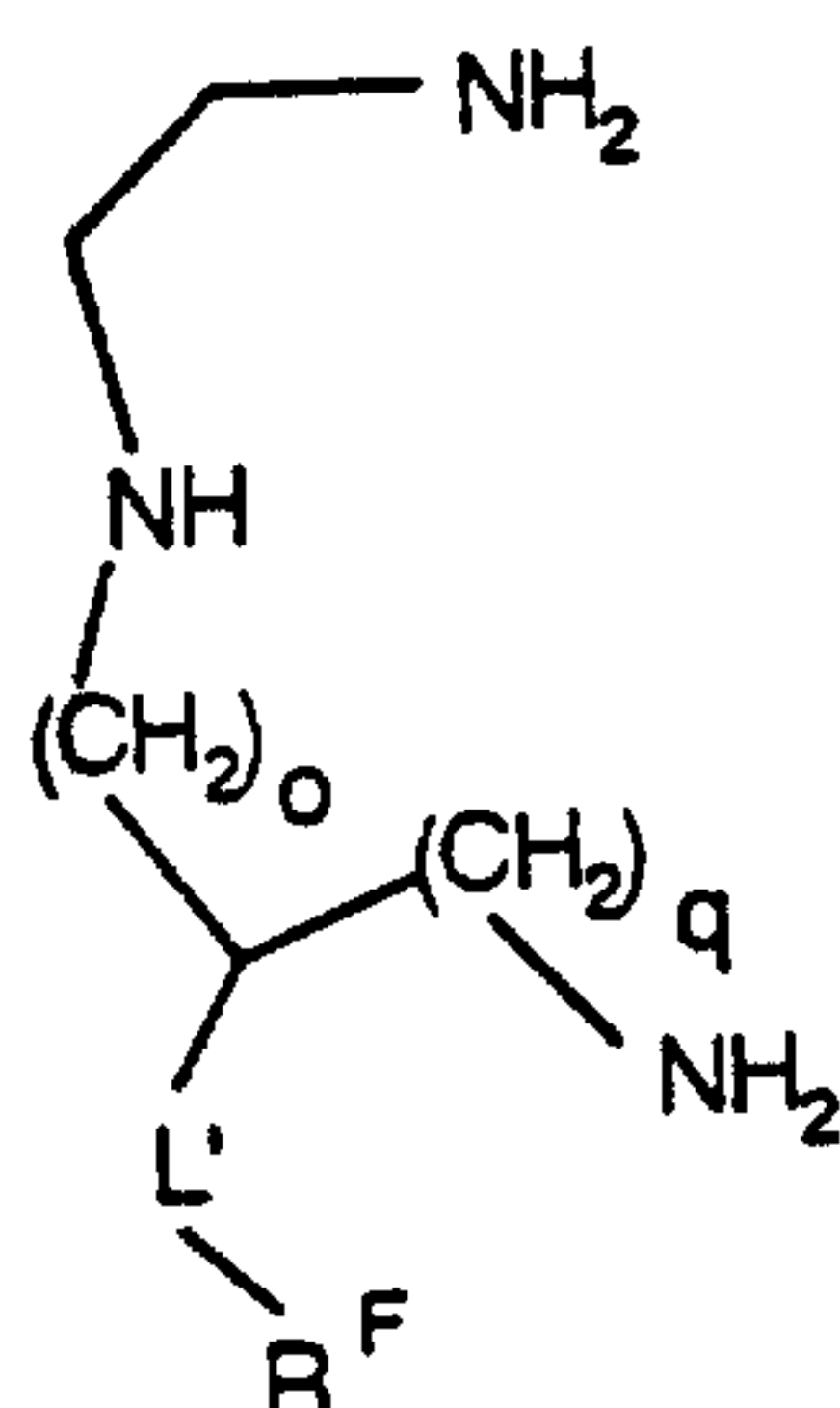
L' , o , q , R^4 and R^F have the above-mentioned meaning, by, if appropriate, cleavage of protective groups and complexing according to a method that is known to one skilled in the art. [Protective Groups, EP 0 071 564, EP 0 130 934, DE-OS 3 401 052].

Compounds of general formula 59 can be produced in a known way, for example, according to J. Org. Chem., 58, 1151 (1993), by reaction of halocarboxylic acid esters 18



with Hal and R^4 in the above-mentioned meaning, and a compound of

general formula 39

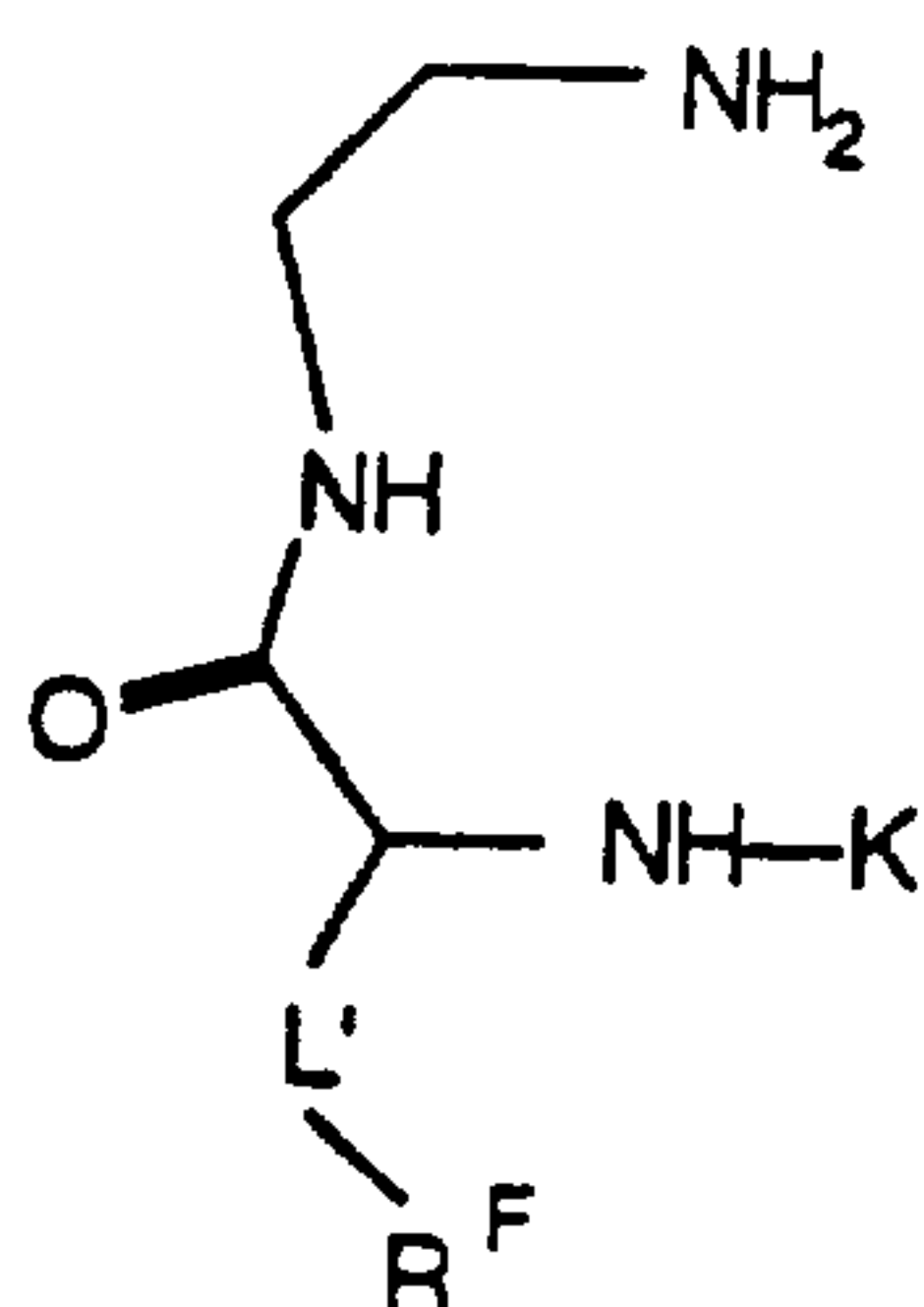


(39)

in which

L' , o , q , and R^F have the above-mentioned meaning.

The compounds of general formula 39 are obtained for the case $q = 0$ from the compounds of general formula 60



(60)

with

L' , R^F and K in the above-mentioned meaning,
 in a way known in the art [Helv. Chim. Acta, 77: 23 (1994)] by
 reduction with diborane and cleavage of the protective groups.
 The compounds of general formula 60 are obtained with
 ethylenediamine by aminolysis of the activated compounds of

general formula 61



in which

L' , Nu , R^{F} and K have the above-mentioned meaning.

The compounds of general formula 61 are produced according to the known methods of protective group chemistry [Protective Groups] from the unprotected acid of general formula 62



specifically the amino group is protected in a first step, followed by the activation of the acid group in the second step.

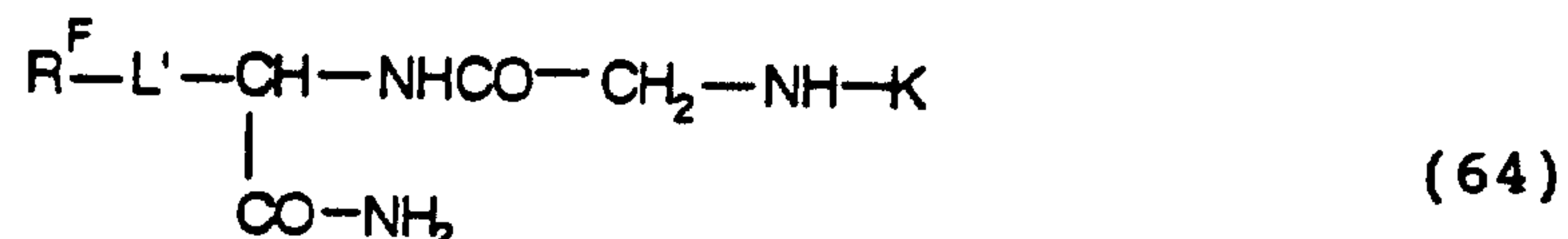
The compounds of general formula 62 can be produced according to the methods of amino acid synthesis [Houben-Weyl, Methoden der organischen Chemie, XI/2 Stickstoffverbindungen II and III, II Aminosäuren [II Amino Acids]; Georg Thieme Verlag Stuttgart, 1958, Strecker-Reaktion [Strecker Reaction], p. 305; Erlenmeyer-Reaktion [Erlenmeyer Reaction], p. 306; Aminolyse von α -Halogencarbonsäuren [Aminolysis of α -Halocarboxylic Acids], p. 309] from the commercially available aldehydes of general formula 63



for example, according to Strecker, via the azlactone or via the cyanohydrin.

The compounds of general formula 39 are obtained for the

case $o = 0$ from the compounds of general formula 64



with R^F , L' and K in the mentioned meanings,
in a way known in the art by cleavage of the protective groups
and reduction with diborane.

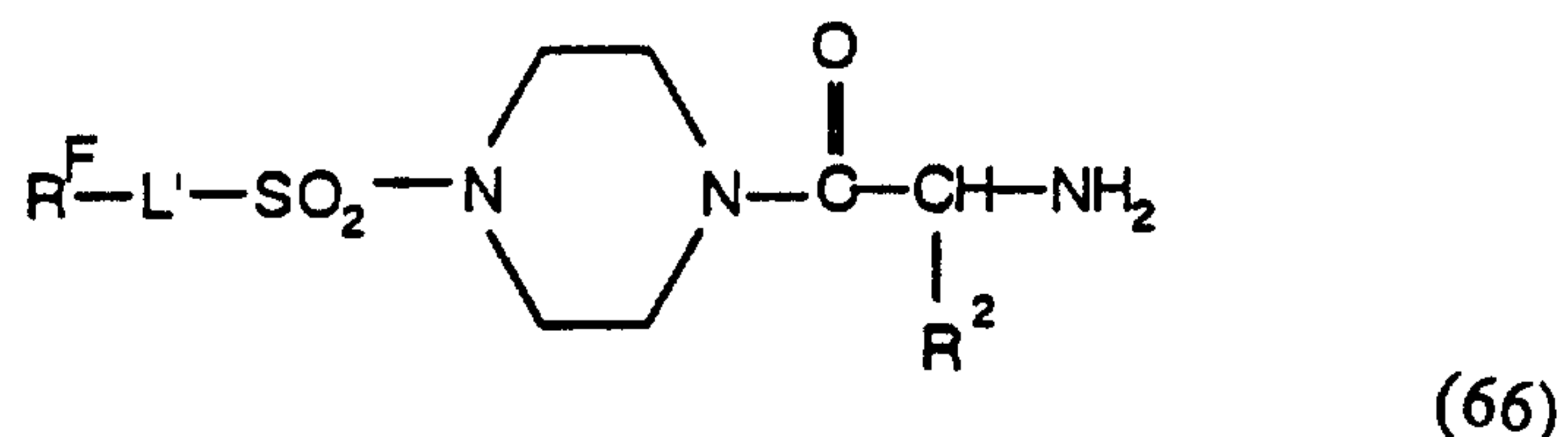
Compounds of general formula 64 are available by aminolysis
of N-protected, activated glycines 53 with compounds of general
formula 65



in which R^F and L' have the mentioned meanings.

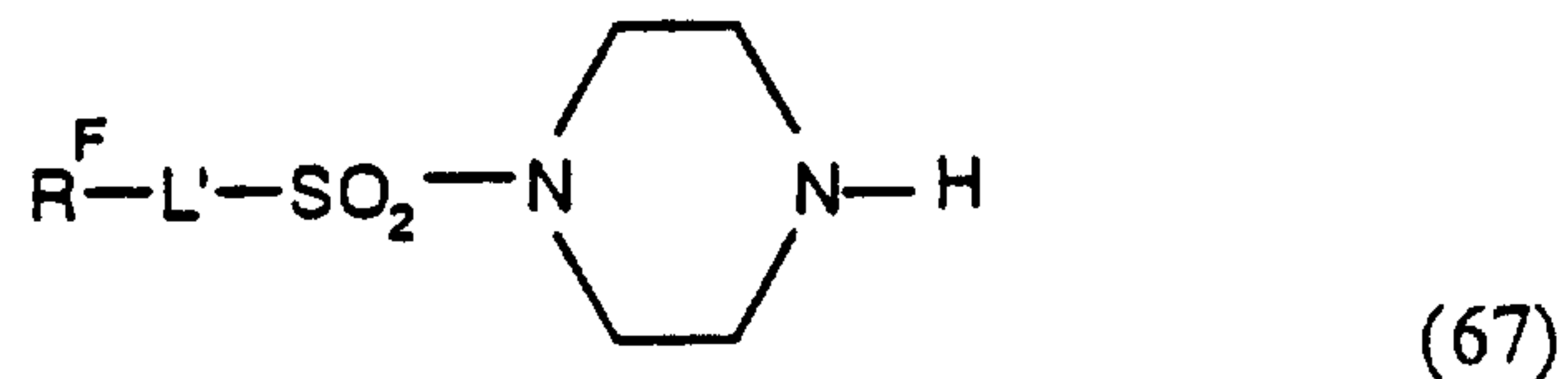
The compounds of general formula 65 can be obtained in a
simple way from compounds of general formula 61 by amide
formation with ammonia and subsequent cleavage of the protective
group.

Compounds of general formula XIII can be produced
analogously to the compounds of general formula III, by
halocarboxylic acid derivatives of general formula 52 being
reacted with a compound of general formula 66



in which R^F , L' and R^2 have the above-mentioned meanings.

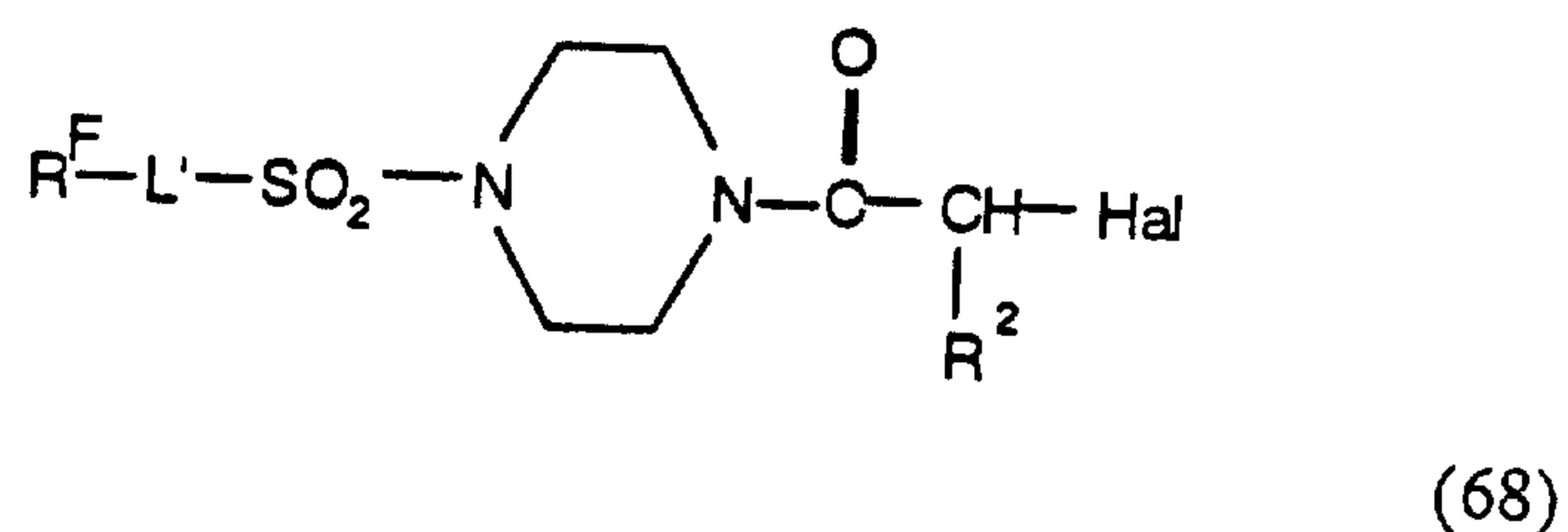
The compounds of general formula 66 are produced by reaction of a compound of general formula 67



with the activated, N-protected amino acid of general formula 53 analogously to the reaction of amine 29 with compound 53.

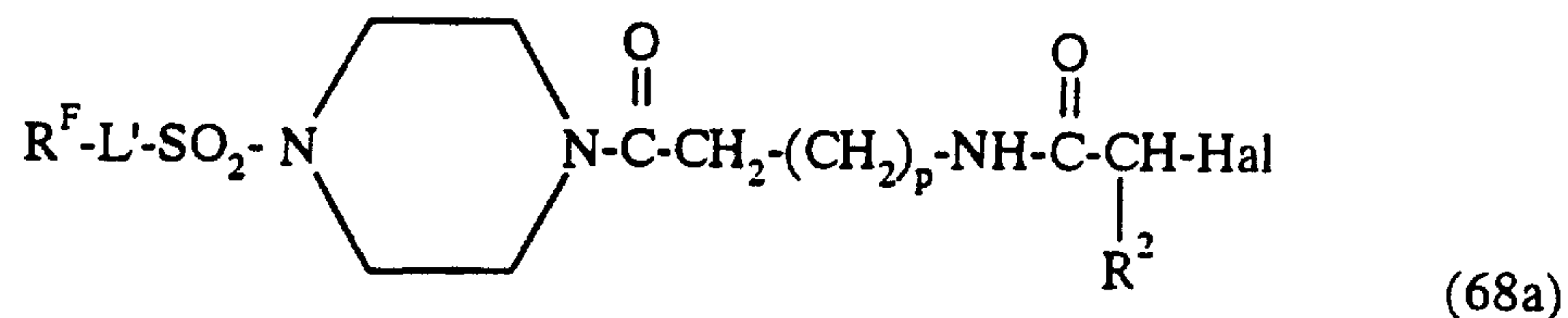
The compounds of general formula 67 can be obtained by reaction of piperazine -- freely or optionally partially protected -- with perfluoroalkylsulfonic acid fluorides or -chlorides. (The sulfonamide formation from amine and sulfofluoride is described in DOS 2 118 190, DOS 2 153 270, both Bayer AG).

Compounds of general formula XI with q meaning number 0 or 1 are produced analogously to compounds of general formula VIII, by compounds of general formula 20 being reacted with compounds of general formula 68



in which R^{F} , L' , R^2 and Hal have the above-mentioned meaning,

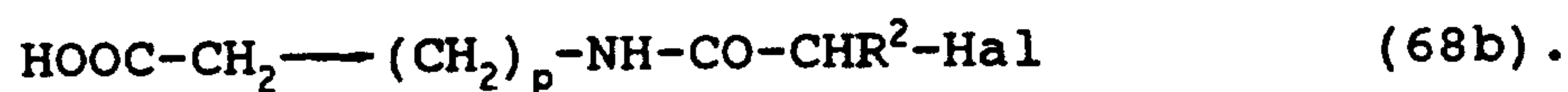
or with compounds of general formula 68a



in which R^{F} , L' , R^2 , p and Hal have the above-mentioned meanings.

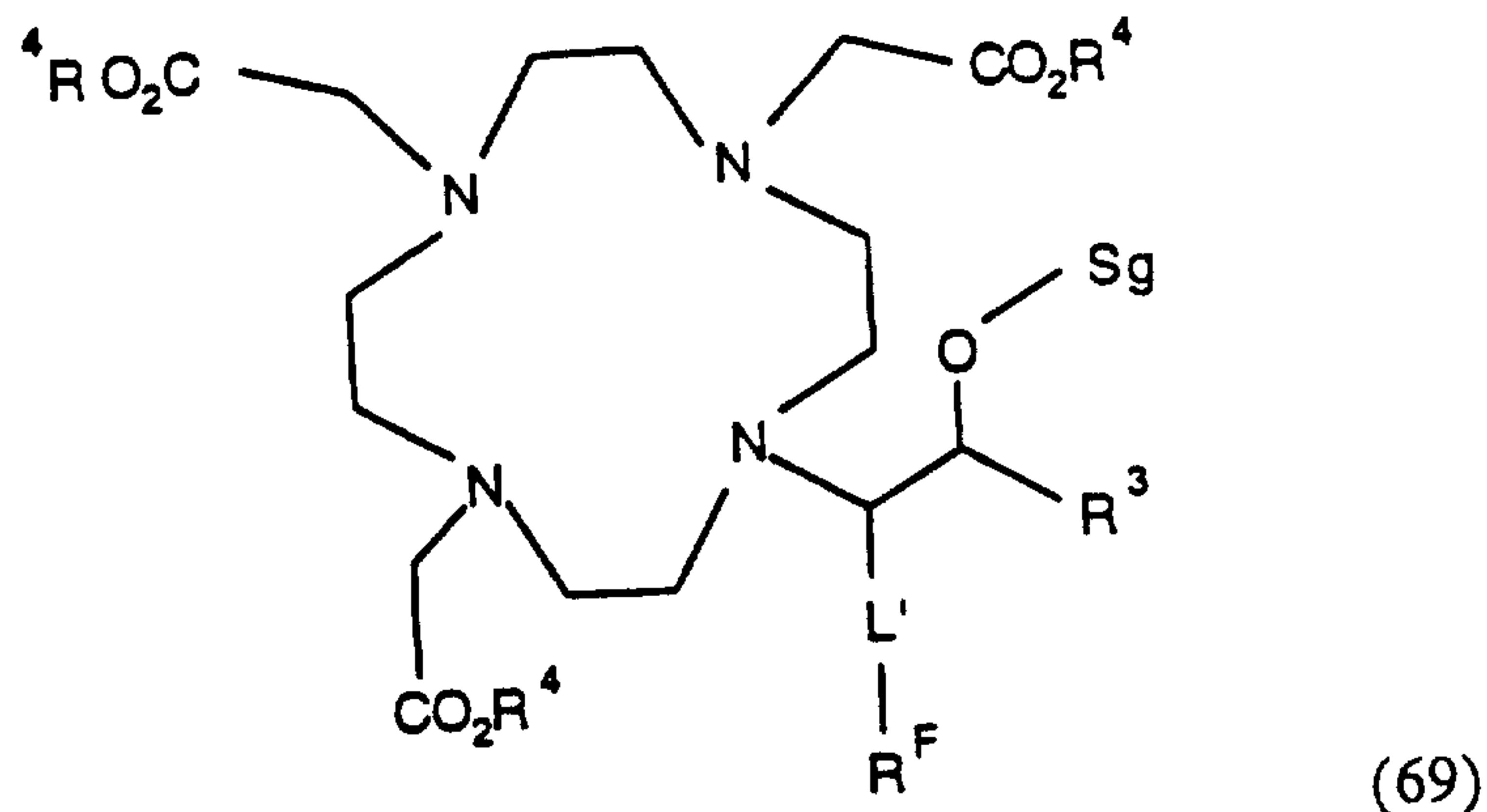
Compounds of general formula 68 can be obtained from compounds of general formula 30 and piperazine derivatives of general formula 67 in a way known in the art.

Compounds of general formula 68a can be obtained from compounds of general formula 67 by amide coupling with compounds of general formula 68b



Compounds of general formula XII are produced analogously to compounds of general formula II, e.g., by reaction of compounds of formula 49 with piperazine derivatives of general formula 67.

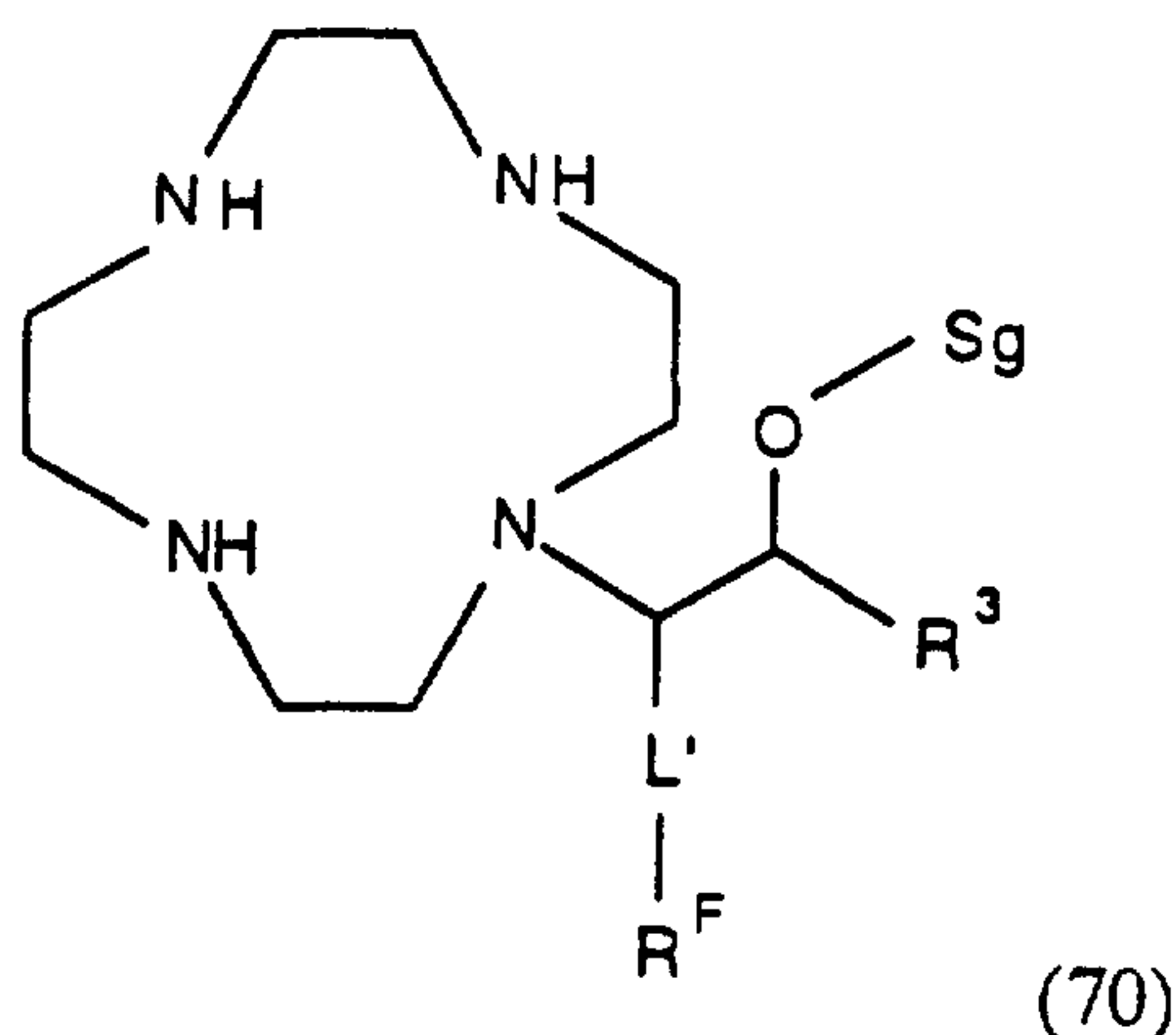
Compounds of general formula I with A in the meaning of general formula X are obtained from compounds of general formula 69



in which

L' , R^3 , R^4 and R^F have the above-described meaning and Sg is in the meaning of a protective group, by, if appropriate, cleavage of protective groups and complexing in a way known in the art [Protective Groups in Organic Synthesis, 2nd Edition, T. W. Greene and P. G. M. Wuts, John Wiley & Sons, Inc., New York, 1991 (EP 0 130 934, EP 0 250 358)].

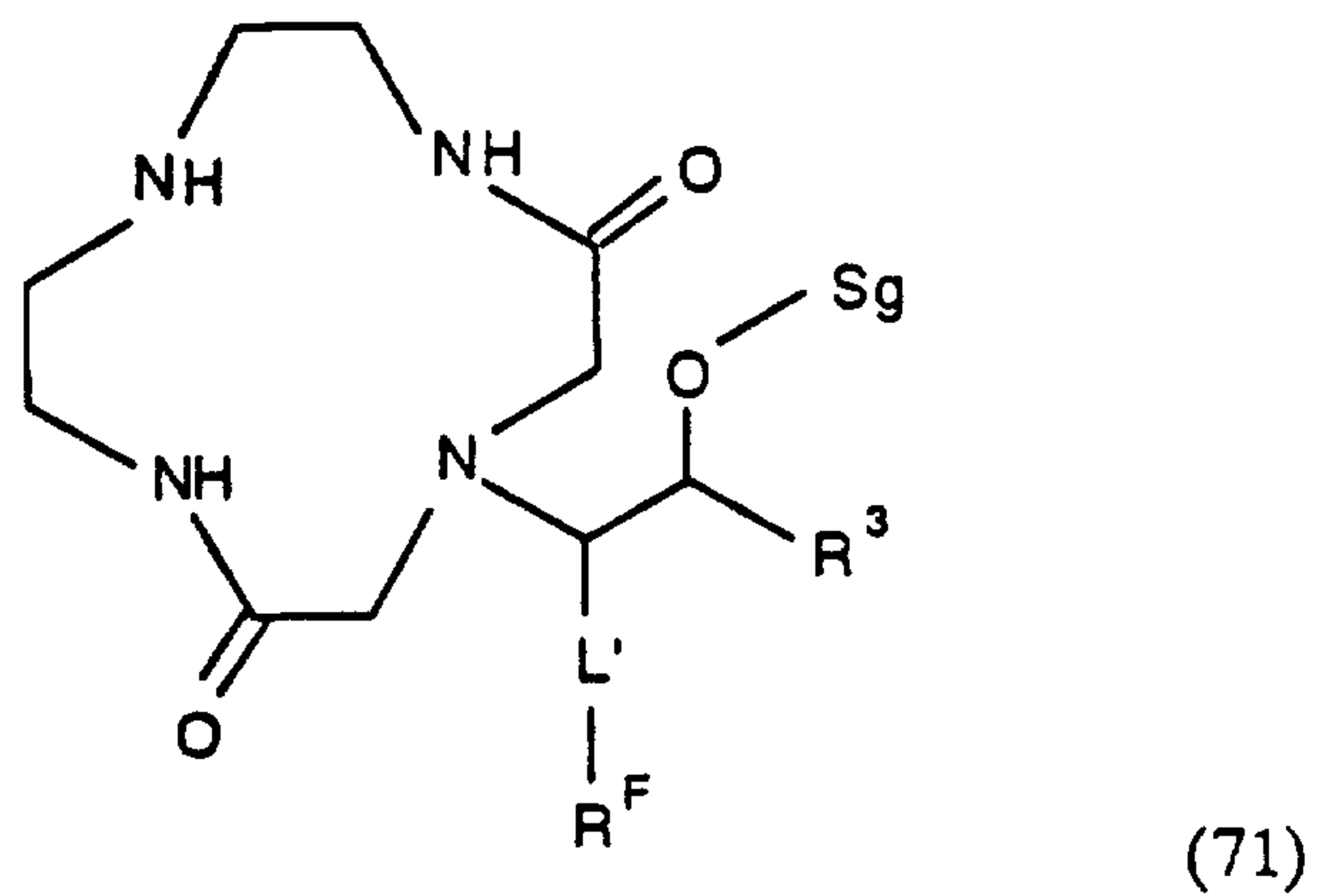
Compounds of general formula 69 are obtained by reaction of α -halocarboxylic acid esters or α -halocarboxylic acids of general formula 18 with compounds of general formula 70



with L' , R^F , R^3 and Sg in the above-mentioned meaning according to the methods that are known to one skilled in the art, as described in, for example, EP 0 255 471 or US 4,885,363.

Compounds of general formula 70 can be obtained by cleavage of optionally present protective groups and subsequent reduction with diborane according to the known processes from compounds of

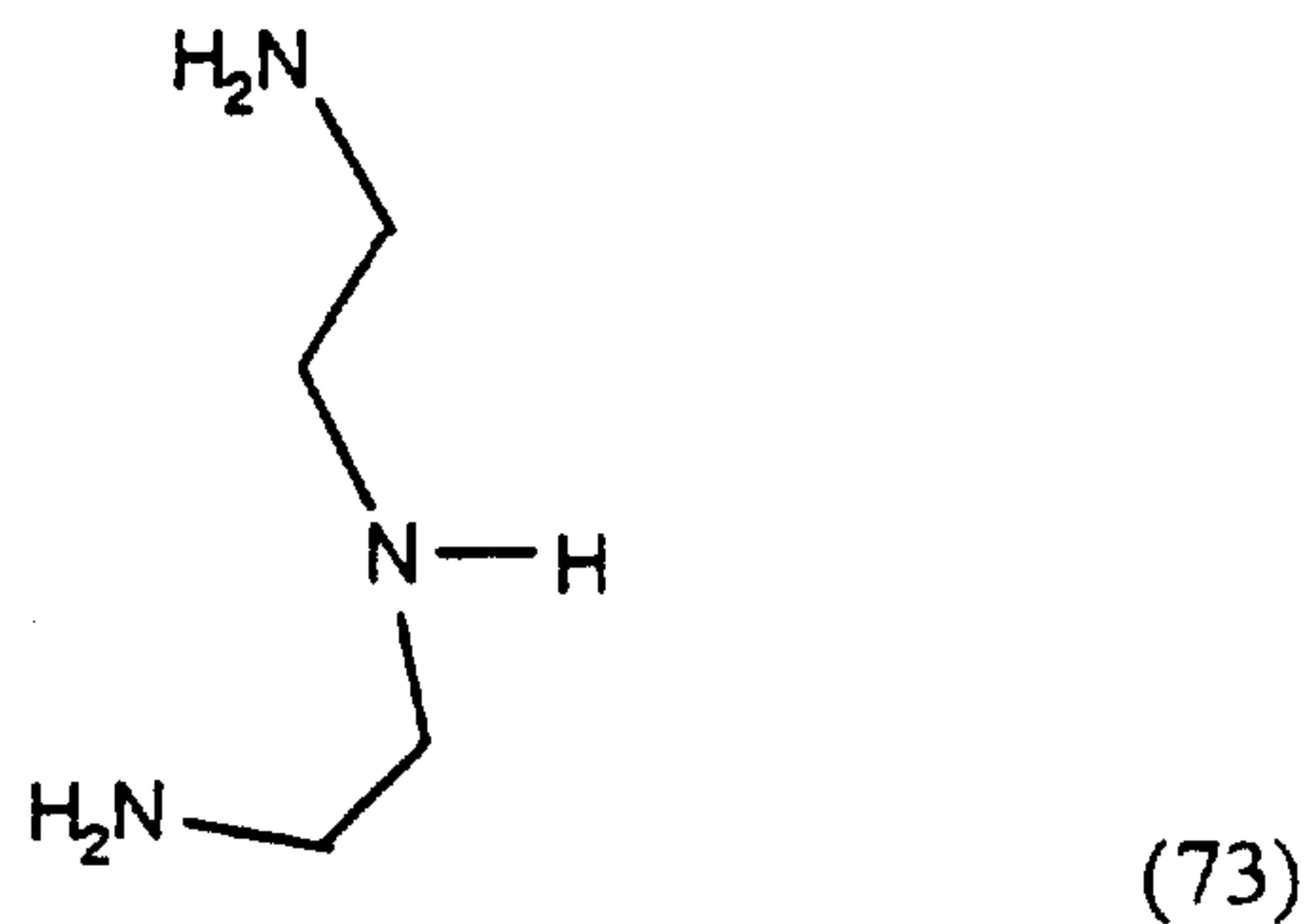
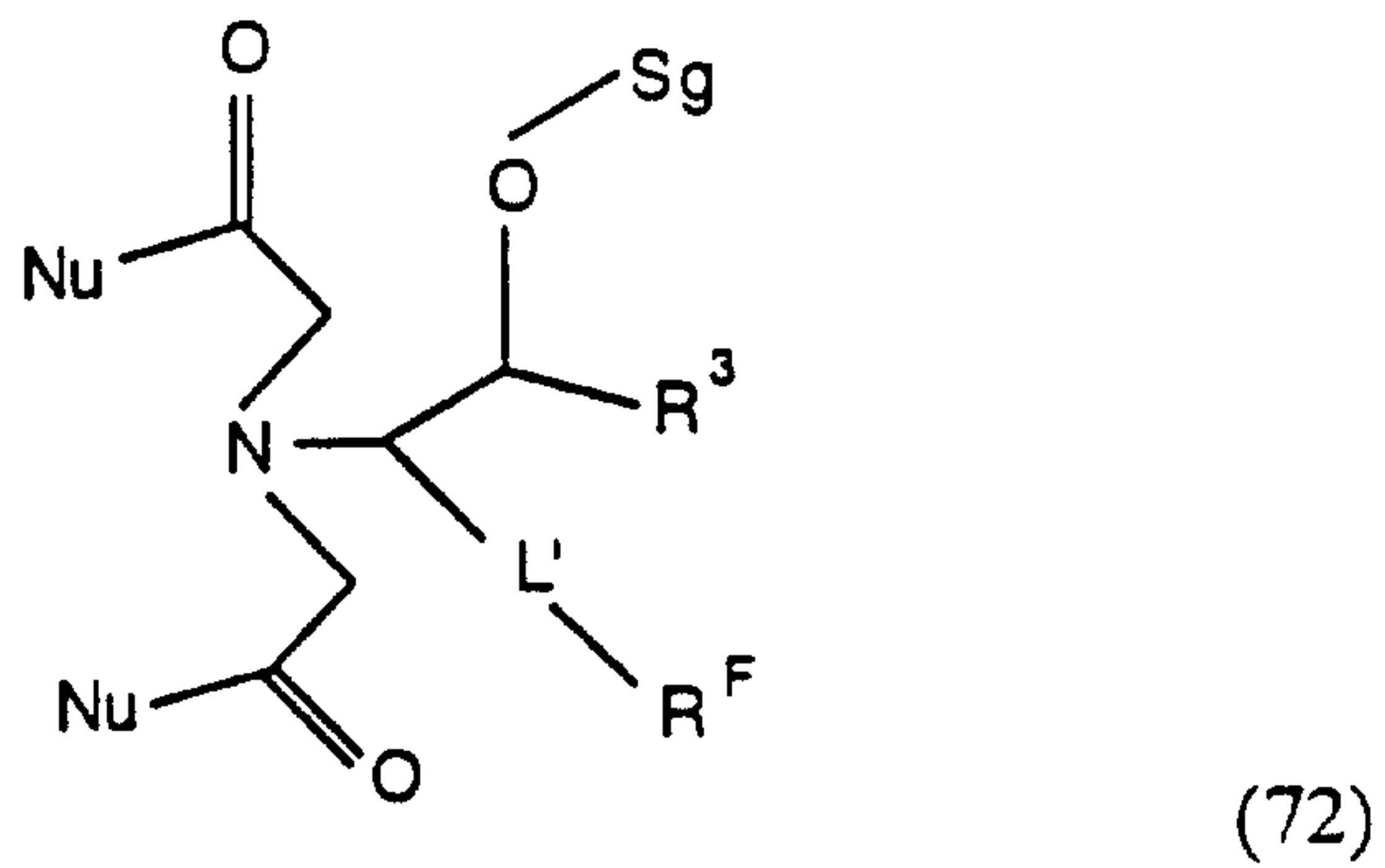
general formula 71



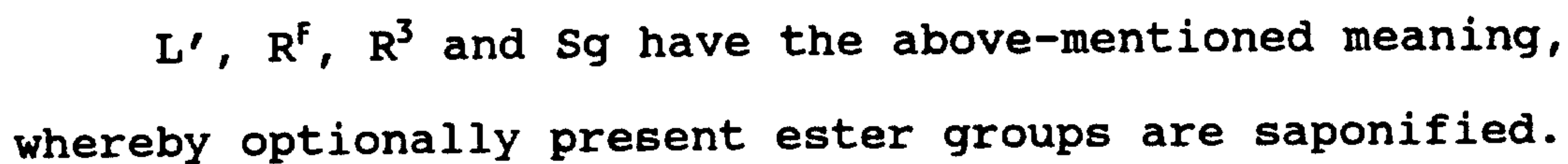
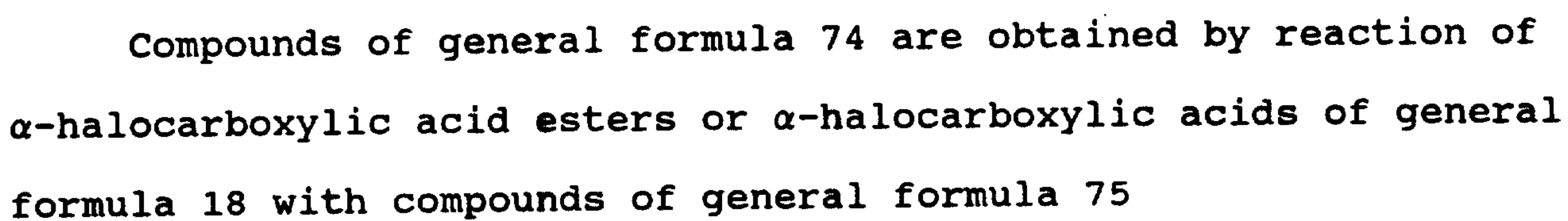
in which

L' , R^F , R^3 and Sg have the above-mentioned meaning.

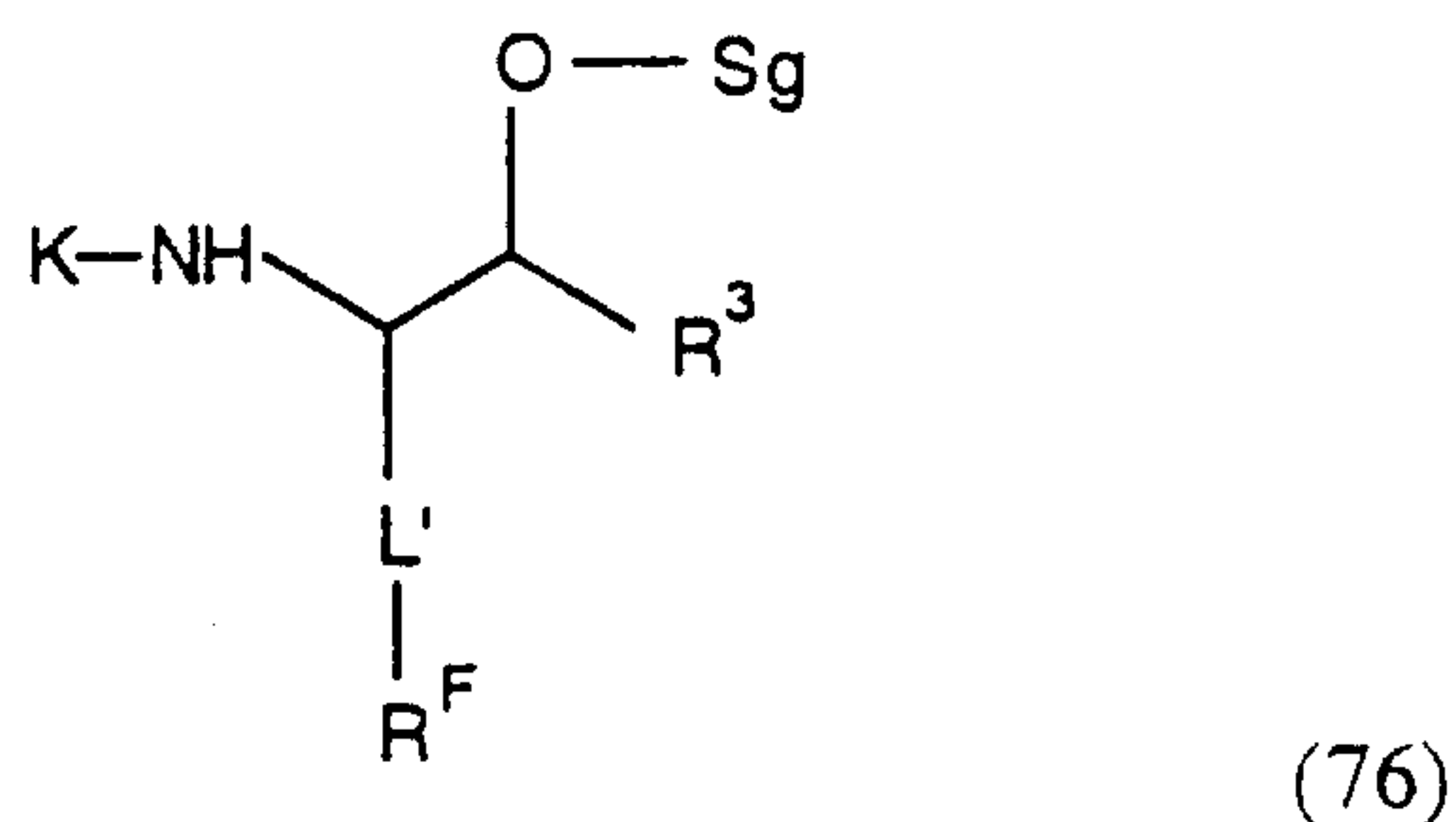
The compounds of general formula 71 can be obtained by a condensation reaction from an activated iminodiacetic acid derivative of general formula 72 and the diethylenetriamine of formula 73



Compounds of general formula 72 can be obtained from compounds of general formula 74



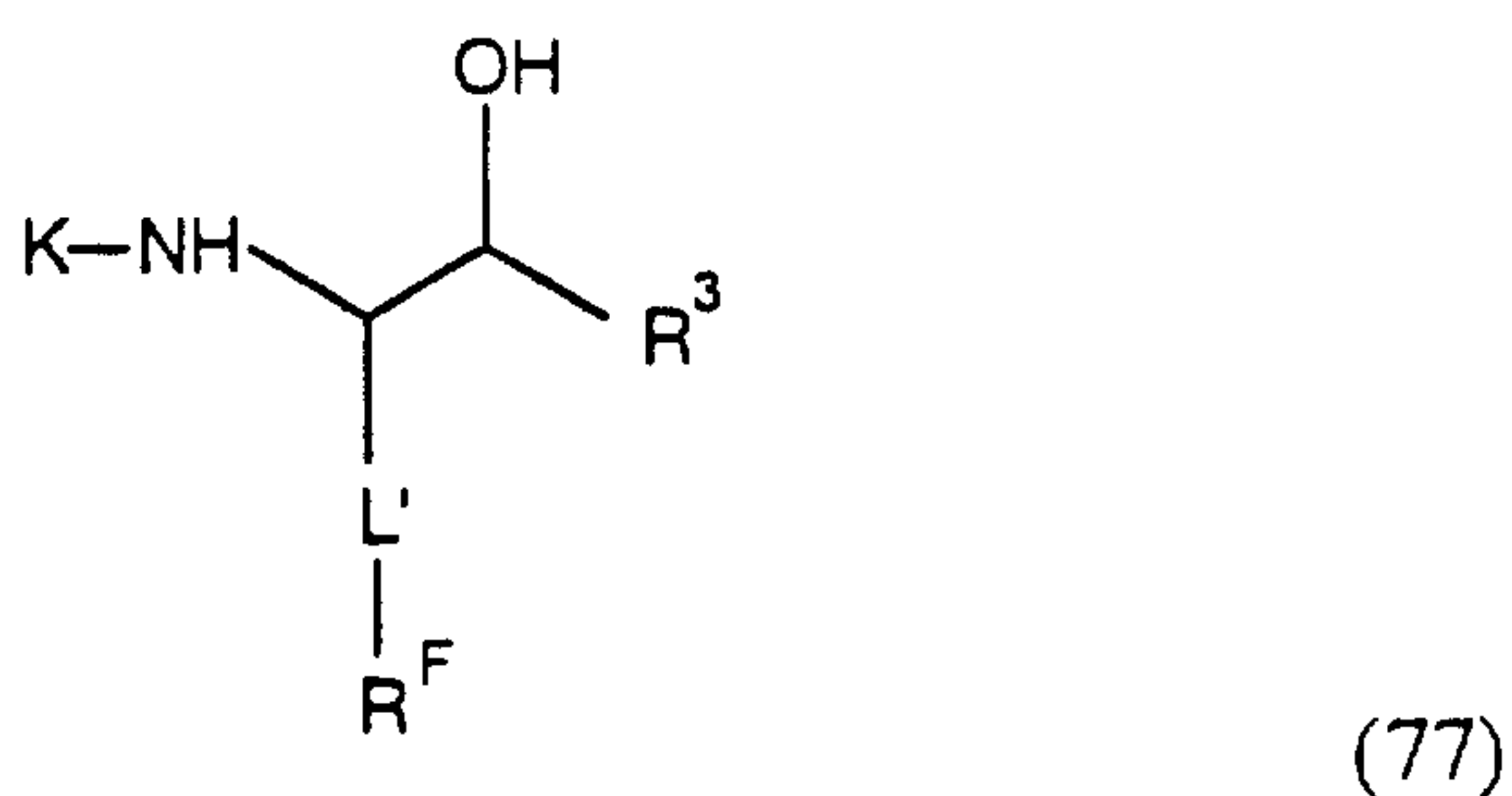
Compounds of general formula 75 are obtained from compounds of general formula 76



in which

L' , R^F , R^3 , Sg and K have the above-mentioned meaning, by cleavage of protective group K according to the known processes.

Compounds of general formula 76 are obtained from compounds of general formula 77

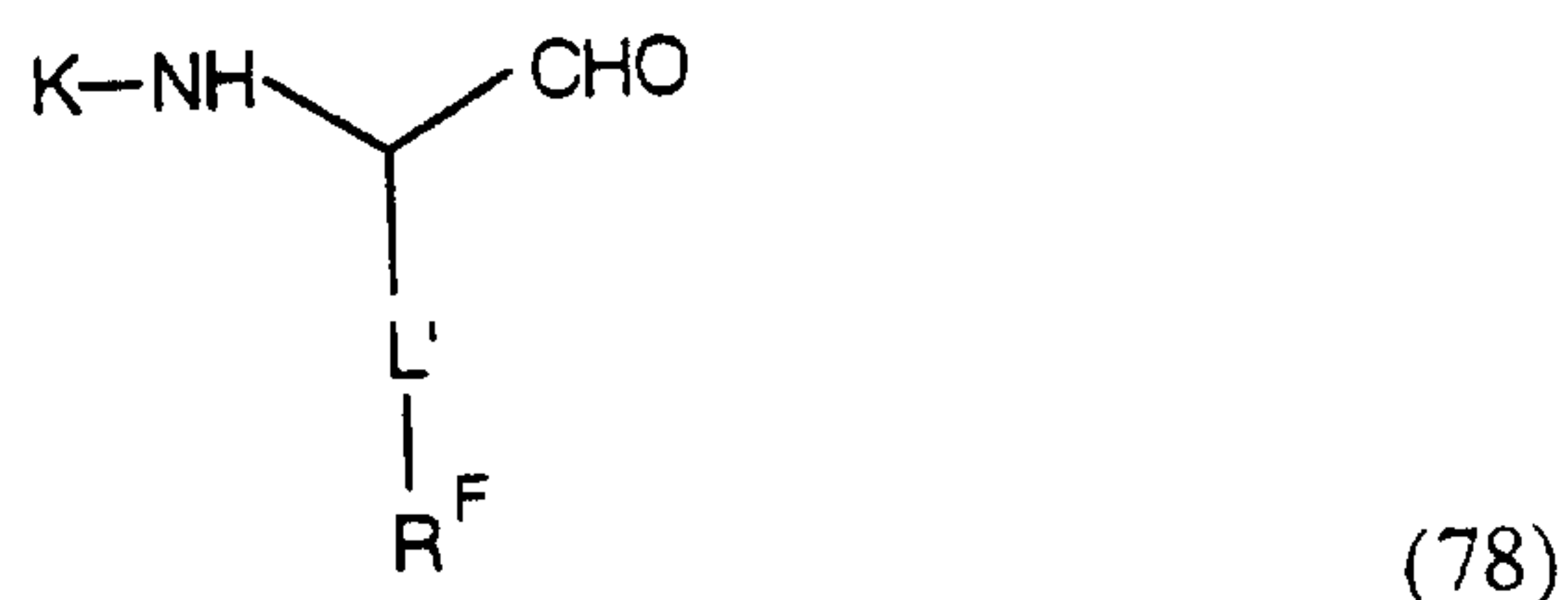


in which

L' , R^F , R^3 and K have the above-mentioned meaning, by introduction of a protective group Sg in the way known to one skilled in the art.

Compounds of general formula 77 are obtained from the

compounds of general formula 78



in which

L', R^F and K have the above-mentioned meaning, according to the methods that are well-known to one skilled in the art (Houben-Weyl, Methoden der organischen Chemie, XIII 2a, Metallorganische Verbindungen [Organometallic Compounds], Georg Thieme Verlag Stuttgart, 1973, p. 285 ff, Umsetzung magnesiumorganischer Verbindungen mit Aldehyden [Reaction of Magnesium-Organic Compounds with Aldehydes]); p. 809 ff, Umsetzung von zinkorganischen Verbindungen mit Aldehyden [Reaction of Zinc-Organic Compounds with Aldehydes]; Houben-Weyl, Methoden der organischen Chemie XIII/1, Metallorganische Verbindungen, Georg Thieme Verlag Stuttgart, 1970; p. 175 ff, Umsetzung lithiumorganischer Verbindungen mit Aldehyden [Reaction of Lithium-Organic Compounds with Aldehydes] by reaction with the organometallic compounds, such as magnesium, lithium or zinc compounds, that can be obtained from compounds of general formula 79

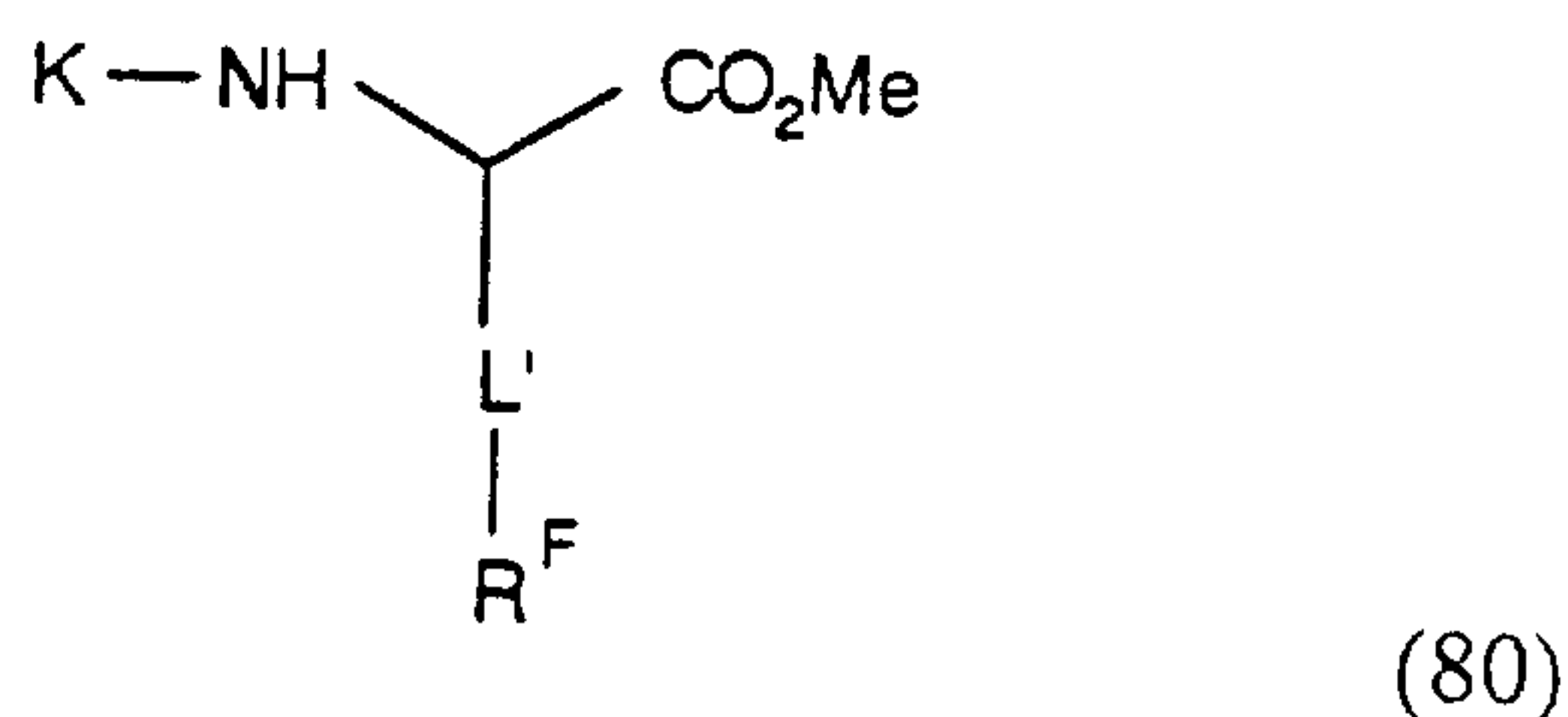


in which

Hal and R³ have the above-mentioned meaning.

Compounds of general formula 79 are commercially available products (ABCR, Fluka).

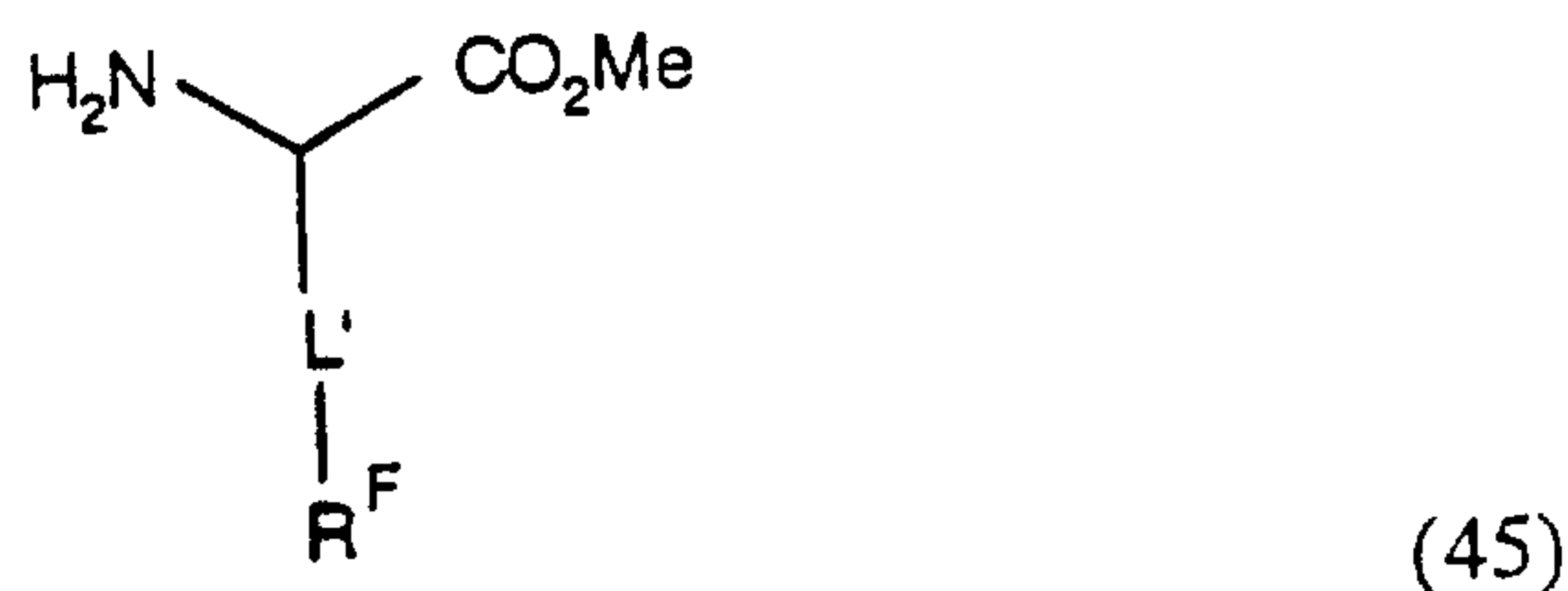
Compounds of general formula 78 are produced from compounds of general formula 80



in which

L' , R^{F} and K have the above-mentioned meaning, by reduction with diisobutylaluminum hydride (Tett. Lett., 1962, 619; Tett. Lett., 1969, 1779; Synthesis, 1975, 617).

Compounds of general formula 80 are produced from compounds of general formula 45



in which

L' and R^{F} have the above-mentioned meaning, in a way known to one skilled in the art by introducing protective group K .

The neutralization of optionally still present free carboxy groups is done with the aid of inorganic bases (for example, hydroxides, carbonates or bicarbonates) of, for example, sodium, potassium, lithium, magnesium or calcium and/or organic bases

such as, i.a., primary, secondary and tertiary amines, such as, for example, ethanolamine, morpholine, glucamine, N-methylglucamine and N,N-dimethylglucamine, as well as basic amino acids, such as, for example, lysine, arginine and ornithine or of amides of originally neutral or acidic amino acids.

To produce neutral complex compounds, enough of the desired bases can be added to, for example, the acidic complex salts in aqueous solution or suspension to ensure that the neutral point is reached. The solution obtained can then be evaporated to the dry state in a vacuum. Often, it is advantageous to precipitate the neutral salts formed by adding water-miscible solvents, such as, for example, lower alcohols (methanol, ethanol, isopropanol, etc.), lower ketones (acetone, etc.), polar ethers (tetrahydrofuran, dioxane, 1,2-dimethoxyethane, etc.) and to obtain easily isolated and readily purified crystallizates. It has proven especially advantageous to add the desired base as early as during the complexing of the reaction mixture and thus to save a process step.

Pharmaceutical agents that contain at least one physiologically compatible compound of general formula I, optionally with the additives that are commonly used in galenicals, are also the subject of the invention.

The production of the pharmaceutical agents according to the invention is carried out in a way known in the art, by the complex compounds according to the invention -- optionally with the addition of the additives that are commonly used in galenicals -- being suspended or dissolved in aqueous medium and

then the suspension or solution optionally being sterilized. Suitable additives are, for example, physiologically harmless buffers (such as, for example, tromethamine), additions of complexing agents or weak complexes (such as, for example, diethylenetriaminepentaacetic acid or the Ca complexes that correspond to the metal complexes according to the invention) or -- if necessary -- electrolytes, such as, for example, sodium chloride or -- if necessary -- antioxidants such as, for example, ascorbic acid.

If suspensions or solutions of the agents according to the invention in water or physiological salt solution are desired for enteral or parenteral administration or other purposes, they are mixed with one or more adjuvant(s) that are commonly used in galenicals [for example, methyl cellulose, lactose, mannitol] and/or surfactant(s) [for example, lecithins, Tween^(R), Myrj^(R)] and/or flavoring substance(s) for taste correction [for example, ethereal oils].

In principle, it is also possible to produce the pharmaceutical agents according to the invention without isolating the complexes. In any case, special care must be used to undertake the chelation so that the complexes according to the invention are practically free from noncomplexed metal ions that have a toxic effect.

This can be ensured, for example, with the aid of color indicators such as xylenol orange by control titrations during the production process. The invention therefore also relates to the process for the production of complex compounds and their

salts. As a final precaution, there remains purification of the isolated complex.

The pharmaceutical agents according to the invention preferably contain $0.1 \mu\text{mol} - 1 \text{ mol/l}$ of the complex and are generally dosed in amounts of $0.0001-5 \text{ mmol/kg}$. They are intended for enteral and parenteral administration. The complex compounds according to the invention are used

1. for NMR diagnosis and diagnostic radiology in the form of their complexes with the ions of elements with atomic numbers 21-29, 39, 42, 44 and 57-83;
2. for radiodiagnosis and radiotherapy in the form of their complexes with the radioisotopes of the elements with atomic numbers 27, 29, 31, 32, 37-39, 43, 49, 62, 64, 70, 75 and 77.

The agents according to the invention meet the varied requirements for suitability as contrast media for nuclear spin tomography. Thus, after oral or parenteral administration and by increasing the signal intensity, they are extremely well suited to improve in its informative value the image obtained with the aid of the nuclear spin tomograph. They also show the high effectiveness that is necessary to burden the body with the smallest possible amounts of foreign substances, and the good compatibility that is necessary to maintain the noninvasive nature of the studies.

The good water-solubility and low osmolality of the agents according to the invention make it possible to produce highly-concentrated solutions, thus to keep the volume burden of the

circulatory system within justifiable limits and to offset the dilution by the bodily fluid. In addition, the agents according to the invention exhibit not only high stability in vitro, but also surprisingly high stability in vivo, so that a release or an exchange of the ions that are bonded to the complexes -- and toxic in themselves -- is carried out only extremely slowly within the time in which the new contrast media are completely eliminated again.

In general, the agents according to the invention for use as NMR diagnostic agents are dosed in amounts of 0.0001 - 5 mmol/kg, preferably 0.005 - 0.5 mmol/kg. Especially low dosages (below 1 mg/kg of body weight) of organ-specific NMR diagnostic agents can be used, for example, for detecting tumors and myocardial infarctions.

The complex compounds according to the invention can also be used advantageously as susceptibility reagents and as shift reagents for in vivo NMR spectroscopy.

The agents according to the invention are also suitable as radiodiagnostic agents owing to their advantageous radioactive properties and the good stability of the complex compounds that are contained in them. Details of such use and dosage are described in, e.g., "Radiotracers for Medical Applications," CRC Press, Boca Raton, Florida.

The compounds and agents according to the invention can also be used in positron emission tomography, which uses positron-emitting isotopes such as, e.g., ^{43}Sc , ^{44}Sc , ^{52}Fe , ^{55}Co and ^{68}Ga

(Heiss, W. D.; Phelps, M. E.; Positron Emission Tomography of Brain, Springer Verlag Berlin, Heidelberg, New York 1983).

The compounds according to the invention are also suitable, surprisingly enough, for differentiating malignant and benign tumors in areas without blood-brain barriers.

They are also distinguished in that they are eliminated completely from the body and are thus well compatible.

Since the substances according to the invention accumulate in malignant tumors (no diffusion in healthy tissue, but high permeability of tumor vessels), they can also assist the radiation therapy of malignant tumors. The latter are distinguished from the corresponding diagnosis only by the amount and type of isotope used. In this case, the aim is the destruction of tumor cells by high-energy shortwave radiation with as small a range of action as possible. For this purpose, interactions of the metals that are contained in the complexes (such as, e.g., iron or gadolinium) with ionizing radiations (e.g., x rays) or with neutron rays are employed. The local radiation dose at the site where the metal complex is located (e.g., in tumors), is significantly enhanced by this effect. To produce the same radiation dose in malignant tissue, the radiation exposure for healthy tissue can be considerably reduced and thus burdensome side effects for the patients can be avoided when such metal complexes are used. The metal complex conjugates according to the invention are therefore also suitable as radiosensitizing substances in the case of radiation therapy of malignant tumors (e.g., use of Mössbauer effects or in the case

of neutron capture therapy). Suitable β -emitting ions are, for example, ^{46}Sc , ^{47}Sc , ^{48}Sc , ^{72}Ga , ^{73}Ga and ^{90}Y . Suitable α -emitting ions that exhibit short half-lives are, for example, ^{211}Bi , ^{212}Bi , ^{213}Bi and ^{214}Bi , whereby ^{212}Bi is preferred. A suitable photon- and electron-emitting ion is ^{158}Gd , which can be obtained from ^{157}Gd by neutron capture.

If the agent according to the invention is specified for use in the variant of radiation therapy that is proposed by R. L. Mills et al. (Nature Vol. 336, (1988), p. 787], the central ion must be derived from a Mössbauer isotope, such as, for example, ^{57}Fe or ^{151}Eu .

In the case of in vivo administration of the agents according to the invention, the latter can be administered together with a suitable vehicle, such as, for example, serum or physiological common salt solution and together with another protein such as, for example, human serum albumin. In this case, the dosage is dependent on the type of cellular impairment, the metal ion used and the type of imaging method.

The agents according to the invention are usually administered parenterally, preferably i.v. They can also be administered -- as already discussed -- intravascularly or interstitially/intracutaneously, depending on how the vessels or tissue of the body are to be examined.

The agents according to the invention are extremely well suited as x-ray contrast media, in which case it is especially to be emphasized that with them, no signs of the anaphylaxis-like reactions that are known from the iodine-containing contrast

media can be detected in biochemical-pharmacological studies. They are especially valuable because of the advantageous absorption properties in the ranges of higher tube voltages for digital subtraction techniques.

In general, the agents according to the invention for use as x-ray contrast media are dosed, for example, analogously to meglumine-diatrizoate, in amounts of 0.1 - 5 mmol/kg, preferably 0.25 - 1 mmol/kg.

Taken overall, it has been possible to synthesize new complexing agents, metal complexes and metal complex salts to open up new possibilities in diagnostic and therapeutic medicine.

The following examples are used for a more detailed explanation of the subject of the invention:

Example 1**a) N-Ethyl-N-(perfluorooctylsulfonyl)-amino-acetic acid-t-butyl ester**

20 g (37.94 mmol) of N-ethylperfluorooctylsulfonamide and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 14.80 g (75.87 mmol) of bromoacetic acid-tert-butyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. The salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 21.66 g (89% of theory) of a waxy, colorless solid

Elementary analysis:

Clcd: C 29.96 H 2.51 F 50.36 N 2.18 S 5.00

Fnd: C 29.81 H 2.70 F 50.15 N 2.30 S 4.83

b) N-Ethyl-N-(perfluorooctylsulfonyl)-amino-acetic acid

20 g (31.18 mmol) of the title compound of Example 1a) is dissolved in 200 ml of trifluoroacetic acid and stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from methanol/ether.

Yield: 17.34 g (95% of theory) of a colorless,
crystalline solid

Elementary analysis:

Cld: C 24.63 H 1.38 F 55.19 N 2.39 S 5.48

Fnd: C 24.48 H 1.50 F 55.01 N 2.17 S 5.59

- c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (17.09 mmol) of the title compound of Example 1b) and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. At 0°C, 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added and stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 16.37 g (78% of theory) of a colorless, vitreous solid

Water content: 7.1%

T_1 -relaxivity (L/mmol·sec) at 20 MHz, 37°C:

41 (water)

49 (human plasma)

Elementary analysis (relative to anhydrous substance):

Cld: C 30.58 H 3.18 F 28.31 Gd 13.78 N 7.37 S 2.81

Fnd: C 30.40 H 3.29 F 28.14 Gd 13.55 N 7.28 S 2.65

- d) 10-[2-Hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (8.76 mmol) of the title compound of Example 1c) is dissolved in a mixture of 100 ml of water/100 ml of ethanol, and 1.73 g (13.71 mmol) of oxalic acid-dihydrate is added. It is heated for 8 hours to 80°C. It is cooled to 0°C, and precipitated gadolinium oxalate is filtered out. The filtrate is evaporated to the dry state, and the residue is purified on RP-18 (RP-18/mobile solvent: gradient consisting of water/i-propanol/acetonitrile).

Yield: 8.96 g (94% of theory) of a vitreous solid

Water content: 9.3%

Elementary analysis (relative to anhydrous substance):

Cld: C 35.30 H 3.98 F 32.73 N 8.52 S 3.25

Fnd: C 35.10 H 4.15 F 32.51 N 8.35 S 3.15

- e) Manganese complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (as sodium salt)

5 g (5.07 mmol) of the title compound of Example 1d) is dissolved in 100 ml of water, and 0.58 g (5.07 mmol) of manganese(II) carbonate is added. It is stirred for 3 hours at 80°C. The solution is filtered, and the filtrate is set at pH 7.2 with 1N sodium hydroxide solution, then it is freeze-dried.

Yield: 5.87 g (quantitative) of a colorless, amorphous powder

Water content: 8.4%

T₁-relaxivity (L/mmol·sec) at 20 MHz, 37°C:

2.7 (water)

4.2 (human plasma)

Elementary analysis (relative to anhydrous substance):

Clcd: C 32.81 H 3.42 F 30.42 Mn 5.17 N 7.92 Na 2.17 S 3.02

Fnd: C 32.62 H 3.57 F 30.21 Mn 5.06 N 7.80 Na 2.01 S 2.90

- f) Ytterbium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctyl-sulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

1.33 g (2.53 mmol) of ytterbium carbonate is added to 5 g (5.07 mmol) of the title compound of Example 1d) in 100 ml of water/30 ml of ethanol, and it is stirred for 3 hours at 80°C. The solution is filtered, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 6.36 g (quantitative) of a vitreous solid.

Water content: 7.8%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.11 H 3.14 F 27.92 N 7.27 S 2.77 Yb 14.96

Fnd: C 30.02 H 3.27 F 27.80 N 7.10 S 2.68 Yb 14.75

- g) Dysprosium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

0.95 g (2.53 mmol) of dysprosium oxide is added to 5 g (5.07 mmol) of the title compound of Example 1d) in 100 ml of water/30 ml of ethanol, and it is stirred for 3 hours at 80°C. The solution is filtered, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 6.35 g (quantitative) of a colorless, vitreous solid.

Water content: 8.5%.

Elementary analysis (relative to anhydrous substance):

Cld: C 30.39 H 3.17 F 28.18 N 7.33 S 2.80 Dy 14.18

Fnd: C 30.17 H 3.25 F 28.03 N 7.21 S 2.65 Dy 14.00

Example 2

- a) 13,13,13,12,12,11,11,10,10,9,9,8,8,7,7,6,6-Heptadecafluoro-3-oxatridecanoic acid-t-butyl ester

10.51 g (53.9 mmol) of bromoacetic acid-tert-butyl ester is added in drops to a mixture of 10 g (21.55 mmol) of 1H,1H,2H,2H-

perfluorodecan-1-ol and 0.73 g (2.15 mmol) of tetrabutylammonium hydrogen sulfate in 100 ml of 60% potassium hydroxide solution/50 ml of toluene while being stirred vigorously at 0°C. It is stirred for 1 hour at 0°C. 200 ml of toluene is added, the aqueous phase is separated and extracted twice with 50 ml of toluene each. The combined organic phases are dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 20/10/1).

Yield: 9.72 g (78% of theory) of a colorless, viscous oil

Elementary analysis:

Cld: C 33.23 H 2.61 F 55.85

Fnd: C 33.09 H 2.78 F 55.71

b) 13,13,13,12,12,11,11,10,10,9,9,8,8,7,7,6,6-Heptadecafluoro-3-oxatridecanoic acid

9.0 g (15.56 mmol) of the title compound of Example 2a) is dissolved in 180 ml of trifluoroacetic acid and stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from methanol/ether.

Yield: 7.80 g (96% of theory) of a colorless solid

Elementary analysis:

Cld: C 27.60 H 1.35 F 61.85

Fnd: C 27.48 H 1.49 F 61.66

c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-oxa-10,10,11,11,12,12,13,13,14,14,15,15,16,16,17,17,17-heptadecafluoroheptadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

7.0 g (13.41 mmol) of the title compound of Example 2b) and 1.70 g (14.75 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 30 ml of dimethylformamide/20 ml of chloroform. At 0°C, 3.04 g (14.75 mmol) of dicyclohexylcarbodiimide is added and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 4.48 g (44.25 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 8.46 g (14.75 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane, dissolved in 40 ml of water, is added and stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 100 ml of methanol/30 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 11.8 g (75% of theory) of a colorless, vitreous solid

Water content: 8.2%

T₁-relaxivity (L/mmol·sec) at 20 MHz, 37°C:

19 (water)

33 (human plasma)

Elementary analysis:

Cld: C 32.32 H 3.27 F 29.96 Gd 14.59 N 6.50

Fnd: C 32.16 H 3.42 F 29.78 Gd 14.39 N 6.40

Example 3

a) 1,2-Epoxy-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorotetradecane

7.97 g (86.18 mmol) of epichlorohydrin is added in drops to a mixture of 20 g (43.09 mmol) of 1H,1H,2H,2H-perfluorodecan-1-ol and 0.79 g (2.32 mmol) of tetrabutylammonium hydrogen sulfate in 200 ml of 60% potassium hydroxide solution/100 ml of toluene while being stirred vigorously at 10°C, and care is taken to ensure that the temperature of the reaction solution is not higher than 20°C. It is allowed to stir for 2 hours at 15°C, and then 3.99 g (43.09 mmol) of epichlorohydrin is added in drops as described above. Then, it is stirred overnight at room temperature. 100 ml of methyl-tert-butyl ether is added, and the aqueous phase is separated. The latter is extracted twice more with 50 ml of toluene each. The organic phases are combined, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/hexane/acetone = 20/10/1).

Yield: 19.05 g (85% of theory) of a colorless oil

Elementary analysis:

Cld: C 30.02 H 1.74 F 62.09

Fnd: C 29.87 H 1.95 F 61.81

- b) 10-[-2Hydroxy-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorotetradecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

8.3 g (207.6 mmol) of sodium hydroxide is added to 12.0 g (34.60 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 50 ml of water. A solution of 18.0 g (34.60 mmol) of the title compound of Example 3a), dissolved in 60 ml of n-butanol/60 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 70°C. It is evaporated to the dry state in a vacuum, the residue is taken up in 300 ml of water and set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 200 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 26.61 g (79% of theory)

Water content: 11.0%

Elementary analysis (relative to anhydrous substance):

Clc: C 37.42 H 4.07 F 37.27 N 6.47

Fnd: C 37.25 H 4.19 F 37.08 N 6.30

- c) Gadolinium complex of 10-[-2hydroxy-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorotetradecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (11.54 mmol) of the title compound of Example 3b) is dissolved in a mixture of 100 ml of water/50 ml of 2-propanol,

and 2.09 g (5.77 mmol) of gadolinium oxide is added. It is stirred for 3 hours at 80°C. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 12.48 g (quantitative) of a vitreous solid

Water content: 5.6%

T_1 -relaxivity (L/mmol·sec) at 20 MHz, 37°C:

15.2 (water)

27.5 (human plasma)

Elementary analysis (relative to anhydrous substance):

Clcd: C 31.77 H 3.16 F 31.64 Gd 15.40 N 5.49

Fnd: C 31.55 H 3.30 F 31.49 Gd 15.28 N 5.35

Example 4

a) 1,2-Epoxy-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorododecane
10.17 g (109.9 mmol) of epichlorohydrin is added in drops to a mixture of 20 g (54.93 mmol) of 1H,1H,2H,2H-perfluorooctan-1-ol and 1.87 g (5.5 mmol) of tetrabutylammonium hydrogen sulfate in 200 ml of 60% aqueous potassium hydroxide solution/100 ml of toluene while being stirred vigorously at 10°C, and care is taken to ensure that the temperature of the reaction solution is not higher than 20°C. It is allowed to stir for 2 hours at 15°C, and then 5.08 g (54.93 mmol) of epichlorohydrin is added in drops as described above. Then, it is stirred overnight at room temperature. 100 ml of toluene and 100 ml of methyl-tert-butyl ether are added, and the aqueous phase is separated. The latter is extracted twice more with 50 ml of toluene each. The organic

phases are combined, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/hexane/acetone = 20/10/1).

Yield: 19.15 g (83% of theory) of a colorless oil

Elementary analysis:

Clcd: C 31.44 H 2.16 F 58.78

Fnd: C 31.40 H 2.29 F 58.55

- b) 10-[2-Hydroxy-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorododecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10.3 g (257 mmol) of sodium hydroxide is added to 14.84 g (42.84 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (D03A) in 70 ml of water. A solution of 18 g (42.84 mmol) of the title compound of Example 4a), dissolved in 80 ml of n-butanol/60 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 70°C. It is evaporated to the dry state in a vacuum, the residue is taken up in 300 ml of water and set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 200 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 27.4 g (75% of theory) of a vitreous solid

Water content: 10.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 39.17 H 4.60 F 32.22 N 7.31

Fnd: C 39.05 H 4.85 F 32.05 N 7.19

c) Gadolinium complex of 10-[2-hydroxy-4-oxa-

1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorododecyl]-1,4,7-

tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (13.04 mmol) of the title compound of Example 4b) is dissolved in a mixture of 100 ml of water/50 ml of 2-propanol, and 2.36 g (6.52 mmol) of gadolinium oxide is added. It is stirred for 3 hours at 80°C. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 12.77 g (quantitative) of a vitreous solid

Water content: 6.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 32.61 H 3.50 F 26.82 Gd 17.08 N 6.08

Fnd: C 32.43 H 3.69 F 26.67 Gd 16.85 N 5.91

Example 5

a) 9,9,9,8,8,7,7,6,6-Nonafluoro-3-oxa-nonanoic acid-t-butyl ester

29.54 g (151.5 mmol) of bromoacetic acid-tert-butyl ester is added in drops to a mixture of 20 g (75.73 mmol) of 1H,1H,2H,2H-perfluorohexan-1-ol and 2.57 g (7.57 mmol) of tetrabutylammonium hydrogen sulfate in 300 ml of 60% aqueous potassium hydroxide solution/200 ml of toluene while being stirred vigorously at 0°C.

It is stirred for 1 hour at 0°C. 100 ml of toluene is added, the aqueous phase is separated and extracted twice with 50 ml of toluene. The combined organic phases are dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 20/10/1).

Yield: 21.48 g (75% of theory) of a colorless oil

Elementary analysis:

Cld: C 38.11 H 4.00 F 45.21

Fnd: C 37.95 H 4.18 F 45.03

b) 9,9,9,8,8,7,7,6,6-Nonanefluoro-3-oxa-nonanoic acid

20 g (52.88 mmol) of the title compound of Example 5a) is dissolved in 300 ml of trifluoroacetic acid and stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from hexane/ether.

Yield: 14.82 g (87% of theory) of a colorless,
crystalline solid

Elementary analysis:

Cld: C 29.83 H 2.19 F 53.08

Fnd: C 29.71 H 2.40 F 52.90

c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-oxa-10,10,11,11,12,12,13,13,13-nonafluoro-tridecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

7.41 g (23.01 mmol) of the title compound of Example 5b) and 2.91 g (25.31 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 40 ml of dimethylformamide/20 ml of chloroform. At 0°C, 5.22 g (25.31 mmol) of dicyclohexylcarbodiimide is added and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 6.98 g (69 mmol) of triethylamine/30 ml of 2-propanol is added. Then, 13.2 g (23.01 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane, dissolved in 40 ml of water, is added and stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/50 ml of chloroform, and cyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 15.20 g (71% of theory) of a colorless, vitreous solid

Water content: 5.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 34.21 H 4.02 F 19.48 Gd 17.91 N 7.98

Fnd: C 34.09 H 4.18 F 19.31 Gd 17.74 N 7.87

Example 6

a) N-Ethyl-N-(perfluorooctylsulfonyl)-amino-acetic acid-N-(2-aminoethyl)-amide

15 g (25.63 mmol) of the title compound of Example 1b) and 3.24 g (28.19 mmol) of N-hydroxysuccinimide are dissolved in 80 ml of dimethylformamide, and 5.82 g (28.19 mmol) of dicyclohexylcarbodiimide is added at 0°C. It is stirred for 1 hour at 0°C, then for 2 hours at room temperature. Precipitated dicyclohexylurea is filtered out, and the filtrate is added in drops within 30 minutes to a solution of 46.21 g (768.9 mmol) of ethylenediamine in 300 ml of dichloromethane. It is stirred for 5 hours at room temperature. 1000 ml of H₂O is added, and the organic phase is separated. The latter is washed twice with 500 ml of water each, then dried on magnesium sulfate and evaporated to the dry state in a vacuum. The purification is carried out by chromatography on silica gel. (Mobile solvent: dichloromethane/2-propanol = 15/1).

Yield: 11.79 g (75% of theory) of a colorless, waxy solid

Elementary analysis:

Clc: C 27.42 H 2.30 F 52.66 N 4.57 S 5.23

Fnd: C 27.20 H 2.41 F 52.48 N 4.38 S 5.10

b) N-Ethyl-N-(perfluorooctylsulfonyl)-amino-acetic acid-N-[2-(bromoacetyl)-aminoethyl]-amide

10 g (16.3 mmol) of the title compound of Example 6a) and 2.02 g (20 mmol) of triethylamine are dissolved in 40 ml of dichloromethane. At -10°C, 3.29 g (16.3 mmol) of bromoacetyl bromide is added in drops within 30 minutes and stirred for 2 hours at 0°C. The solution is poured into 300 ml of 1N

hydrochloric acid and stirred thoroughly. The organic phase is separated, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/acetone = 20/1).

Yield: 11.1 g (91% of theory) of a slightly yellow-colored waxy solid

Elementary analysis:

Clcd: C 25.68 H 2.02 Br 10.68 F 43.16 N 5.62 S 4.29

Fnd: C 25.47 H 2.18 Br 10.45 F 43.29 N 5.47 S 4.10

c) 10-[2-Oxo-3-aza-6-aza-7-oxo-9-aza-9-(perfluorooctylsulfonyl)-undecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

4.63 g (13.36 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (D03A) and 18.5 g (133.6 mmol) of potassium carbonate are added to 10 g (13.36 mmol) of the title compound of Example 6b) in 180 ml of methanol. It is refluxed for 12 hours. The inorganic salts are filtered off, and the filtrate is evaporated to the dry state. The residue is taken up in 100 ml of water and set at pH 3 with 5N hydrochloric acid. It is extracted twice with 150 ml of n-butanol. The combined organic phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent = gradient consisting of water/n-butanol/acetonitrile).

Yield: 10.43 g (67% of theory) of a colorless solid

Water content: 13.0%

Elementary analysis (relative to anhydrous substance):

Cld: C 35.55 H 3.98 F 31.86 N 9.67 S 3.16

Fnd: C 35.37 H 3.75 F 31.64 N 9.78 S 3.25

- d) Gadolinium complex of 10-[2-oxo-3-aza-6-aza-7-oxo-9-aza-9-(perfluorooctylsulfonyl)-undecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (9.86 mmol) of the title compound of Example 6c) is dissolved in a mixture of 50 ml of water/20 ml of ethanol, and 1.79 g (4.93 mmol) of gadolinium oxide is added. It is stirred for 4 hours at 80°C. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 12.4 g (quantitative)

Water content: 7.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.85 H 3.19 F 27.65 Gd 13.46 N 8.39 S 2.75

Fnd: C 30.64 H 3.35 F 27.58 Gd 13.29 N 8.28 S 2.65

Example 7

a) 1H,1H,2H,2H-Perfluorodecan-1-ol-p-toluenesulfonic acid ester
12.57 g (65.93 mmol) of p-toluenesulfonic acid chloride is added to 30 g (64.64 mmol) of 1H,1H,2H,2H-perfluorodecan-1-ol in 300 ml of dichloromethane and 10.12 g (100 mmol) of triethylamine at 0°C. It is stirred for 2 hours at 0°C, then for 2 hours at room temperature. The solution is poured into 500 ml of cold 2N hydrochloric acid and stirred vigorously. The organic phase is separated, dried on magnesium sulfate and evaporated to the dry state. The residue is recrystallized from a little methanol.

Yield: 39.97 (95% of theory) of a colorless, crystalline powder

Elementary analysis:

Cld: C 33.02 H 1.79 F 52.23 S 5.19

Fnd: C 32.81 H 1.93 F 52.04 S 5.05

b) 10-[(1-Hydroxymethyl-1-carboxy)-methyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane
37.2 g (173.4 mmol) of 2-chloro-3-benzyloxy-propanoic acid is added to a solution of 20 g (57.78 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (D03A), 31.21 g (780 mmol) of sodium hydroxide and 2 g (12 mmol) of potassium iodide in 100 ml of dimethylformamide, and it is stirred for 3 days at 60°C. It is evaporated to the dry state, and the residue is dissolved in 300 ml of water. Then, it is set at pH 3 with 3N hydrochloric acid and extracted twice with 250 ml of

dichloromethane each. 4 g of palladium catalyst (10% Pd/C) is added to the aqueous phase and hydrogenated for 5 hours at 60°C. The catalyst is filtered off, and the filtrate is evaporated to the dry state. The residue is purified by RP-chromatography (RP-18/mobile solvent = gradient consisting of water/2-propanol/acetonitrile).

Yield: 5.92 g (21% of theory relative to D03A) of a colorless, vitreous solid

Water content: 11.1%

Elementary analysis (relative to anhydrous substance):

Clcd: C 47.00 H 6.96 N 12.90

Fnd: C 46.81 H 6.78 N 12.99

- c) 10-[1-Hydroxymethyl-1-(methoxycarbonyl)-methyl]-1,4,7-tris(methoxycarbonylmethyl)-1,4,7,10-tetraazacyclododecane
- 9.53 g (80 mmol) of thionyl chloride is added in drops to 200 ml of methanol at 0°C. Then, 5.8 g (13.35 mmol) of the title compound of Example 7b) is added and stirred for 1 hour at 0°C. Then, it is heated for 6 hours to 60°C. It is evaporated to the dry state, the residue is taken up in 150 ml of methylene chloride and extracted 3 times with 200 ml of 8% aqueous soda solution each. The organic phase is dried on magnesium sulfate and evaporated to the dry state. 6.09 g (93% of theory) of the title compound is obtained as a slightly yellowish-colored oil.

Elementary analysis:

Cld: C 51.42 H 7.81 N 11.42

Fnd: C 51.20 H 7.95 N 11.28

- d) 10-[1-(Methoxycarbonyl)-3-oxa-1H,2H,2H,4H,4H,5H,5H-perfluorotridecyl]-1,4,7-tris(methoxycarbonylmethyl)-1,4,7,10-tetraazacyclododecane

0.44 g (14.68 mmol) of sodium hydride (80% suspension in mineral oil) is added to 6 g (12.23 mmol) of the title compound of Example 7c) in 40 ml of dimethylformamide and stirred for 30 minutes at -10°C. Then, 8.32 g (13.45 mmol) of the title compound of Example 7a) is added and stirred for 8 hours at room temperature. 400 ml of ice water is carefully added and extracted twice with 300 ml of ethyl acetate each. The combined ethyl acetate phases are washed with saturated aqueous common salt solution and dried on magnesium sulfate. It is evaporated to the dry state in a vacuum, and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/methanol = 20/1).

Yield: 7.68 g (67% of theory) of a viscous yellow oil

Elementary analysis:

Cld: C 39.75 H 4.41 F 34.48 N 5.98

Fnd: C 39.58 H 4.60 F 34.27 N 5.75

e) 10-[1-Carboxy-3-oxa-1H,2H,2H,4H,4H,5H,5H-perfluorotridecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane
7.5 g (8.01 mmol) of the title compound of Example 7d) is suspended in a mixture of 50 ml of water/30 ml of ethanol, and then 3.84 g (96 mmol) of sodium hydroxide is added. It is refluxed overnight. It is cooled to room temperature and set at pH 3 with 3N hydrochloric acid. It is evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent = gradient consisting of water/n-butanol/acetonitrile).

Yield: 6.84 g (87% of theory) of a vitreous solid

Water content: 10.3%

Elementary analysis (relative to anhydrous substance):

Clcd: C 36.83 H 3.78 F 36.68 N 6.36

Fnd: C 36.67 H 3.90 F 36.49 N 6.25

f) Gadolinium complex of 10-[1-carboxy-3-oxa-1H,2H,2H,4H,4H,5H,5H-perfluorotridecyl]1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (as sodium salt)

6 g (6.81 mmol) of the title compound of Example 7e) is suspended in 80 ml of water, and 1.23 g (3.4 mmol) of gadolinium oxide is added. It is heated for 3 hours to 90°C. It is allowed to cool to room temperature and set at pH 7.2 with 2N sodium hydroxide solution. The solution is filtered and then freeze-dried.

Yield: 7.83 g (quantitative) of a colorless, flocculent powder

Water content: 8.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.69 H 2.77 F 30.56 Gd 14.88 N 5.30 Na 2.18

Fnd: C 30.48 H 2.85 F 30.37 Gd 14.69 N 5.17 Na 1.95

Example 8

a) 2H,2H-Perfluorooctanal

30 g (82.4 mmol) of 1H,1H,2H,2H-perfluorooctan-1-ol is dissolved in 500 ml of dichloromethane, and 17.76 g (82.4 mmol) of pyridinium chlorochromate is added. It is stirred overnight at room temperature. The solution is filtered with a short column, filled with aluminum oxide (neutral), the filtrate is evaporated to the dry state and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/hexane/acetone = 10/10/1).

Yield: 26.55 g (89% of theory) of a waxy solid

Elementary analysis:

Cld: C 26.54 H 0.84 F 68.21

Fnd: C 26.47 H 1.05 F 68.10

b) 2-Amino-2H,3H,3H-perfluorononanoic acid (as hydrochloride)

7.04 g (143.6 mmol) of sodium cyanide and 8.45 g (158 mmol) of ammonium chloride are dissolved in 30 ml of water. 40 ml of

ethanol and 26 g (71.8 mmol) of the title compound of Example 8a) are added to this solution. It is heated for 2 hours to 45°C. 300 ml of water is added, and it is extracted 3 times with 200 ml of benzene each. The combined benzene phases are washed 3 times with 200 ml of water each, and the organic phase is evaporated to the dry state in a vacuum. The residue is taken up in 100 ml of 6N aqueous hydrochloric acid/50 ml of methanol and refluxed for 2 hours. It is evaporated to the dry state in a vacuum. The residue is recrystallized from a little 2-propanol/methyl-tert-butyl ether.

Yield: 11.15 g (35% of theory) of a crystalline solid

Elementary analysis:

Clcd: C 24.37 H 1.59 Cl 7.99 F 55.68 N 3.16

Fnd: C 24.15 H 1.72 Cl 7.65 F 55.51 N 3.05

c) 2-[(N-Benzyloxycarbonyl)-triglycidyl]-amino-2H,3H,3H-perfluorononanoic acid

8.37 g (24.8 mmol) of N-benzyloxycarbonyl-triglycine and 3.14 g (27.28 mmol) of N-hydroxysuccinimide are dissolved in 80 ml of dimethylformamide, and 5.63 g (27.28 mmol) of dicyclohexylcarbodiimide is added at 0°C. It is stirred for 1 hour at 0°C, then for 2 hours at room temperature. It is cooled to 0°C, 7.53 g (74.4 mmol) of triethylamine and 11 g (24.8 mmol) of the title compound of Example 8b are added and then stirred overnight at room temperature. It is evaporated to the dry state in a vacuum, the residue is taken up in 300 ml of 5% aqueous

citric acid and extracted 3 times with 200 ml of ethyl acetate each. The combined organic phases are dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/n-propanol=20/1).

Yield: 11.83 g (67% of theory) of a colorless, sheetlike solid

Elementary analysis:

Clcd: C 38.78 H 2.97 F 34.67 N 7.86

Fnd: C 38.59 H 2.85 F 34.48 N 7.91

d) 2-[Triglycidyl]-amino-2H,3H,3H-perfluorononanoic acid

11.5 g (16.14 mmol) of the title compound of Example 8c) is dissolved in 200 ml of 2-propanol, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. Catalyst is filtered out, and the filtrate is evaporated to the dry state.

Yield: 9.33 g (quantitative) of a colorless solid

Elementary analysis:

Clcd: C 31.15 H 2.61 F 42.71 N 9.69

Fnd: C 31.29 H 2.80 F 42.53 N 9.48

- e) 2-(1H,1H-Perfluoroheptyl)-1,4,7,10-tetraaza-3,6,9,12-tetraoxo-cyclododecane

9.2 g (15.91 mmol) of the title compound of Example 8d) is dissolved in 1000 ml of dimethylformamide, and 3.93 g (15.91 mmol) of 2-ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline is added. It is stirred for 3 days at room temperature. It is evaporated to the dry state, and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/2-propanol = 20/1).

Yield: 4.54 g (51% of theory) of a waxy solid

Elementary analysis:

Clcd: C 32.16 H 2.34 F 44.08 N 10.00

Fnd: C 32.05 H 2.47 F 43.87 N 9.89

- f) 2-(1H,1H-Perfluoroheptyl)-1,4,7,10-tetraazacyclododecane (as tetrahydrochloride)

200 ml of 1M borane-tetrahydrofuran complex solution is added to 4.4 g (7.85 mmol) of the title compound of Example 8e) and refluxed for 2 days. It is evaporated to the dry state in a vacuum, and the residue is taken up in 50 ml of concentrated hydrochloric acid. 100 ml of ethanol is added, and it is refluxed for 8 hours. It is evaporated to the dry state in a vacuum, and the residue is recrystallized from ethanol.

Yield: 4.75 g (93% of theory) of a colorless, crystalline powder

Elementary analysis:

Cld: C 27.71 H 3.88 Cl 21.81 F 37.99 N 8.62

Fnd: C 27.65 H 3.95 Cl 21.40 F 37.69 N 8.41

g) 2-(1H,1H-Perfluoroheptyl)-1,4,7,10-tetra(carboxymethyl)-
1,4,7,10-tetraazacyclododecane

4.6 g (7.07 mmol) of the title compound of Example 8f) and 4.0 g (42.4 mmol) of chloroacetic acid are dissolved in 40 ml of water, and the pH is set at 10 by adding 30% aqueous potassium hydroxide solution. It is heated for 8 hours to 70°C and in this case, the pH is kept between 8 and 10 (by adding 30% aqueous potassium hydroxide solution). The solution is cooled to room temperature, set at pH 2 with concentrated hydrochloric acid and evaporated to the dry state. The residue is taken up in 150 ml of methanol, the salts are filtered off, and the filtrate is evaporated to the dry state in a vacuum. The residue is purified by RP-18 chromatography (RP-18/mobile solvent: gradient consisting of water/2-propanol/acetonitrile).

Yield: 5.03 g (87% of theory) of a vitreous solid

Water content: 10.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 37.51 H 3.97 F 33.53 N 7.61

Fnd: C 37.35 H 4.12 F 33.40 N 7.45

- h) Gadolinium complex of 2-(1H,1H-perfluoroheptyl)-1,4,7,10-tetra(carboxymethyl)-1,4,7,10-tetraazacyclododecane (as sodium salt)

4.5 g (6.11 mmol) of the title compound of Example 8g) is suspended in 100 ml of water, and 1.107 g (3.05 mmol) of gadolinium oxide is added. It is heated for 3 hours to 90°C. It is allowed to cool to room temperature and set at pH 7.2 with 2N sodium hydroxide solution. The solution is filtered and then freeze-dried.

Yield: 6.03 g (quantitative) of a colorless powder

Water content: 7.5%

Elementary analysis (relative to anhydrous substance):

Clcd: C 30.23 H 2.87 F 27.03 Gd 17.21 N 6.13 Na 2.52

Fnd: C 30.10 H 3.05 F 26.81 Gd 17.15 N 5.95 Na 2.30

Example 9

- a) 10-[2-Hydroxy-1H,1H,2H,3H,3H-perfluorononyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

13.85 g (346.4 mmol) of sodium hydroxide is added to 15 g (43.3 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 50 ml of water. A solution of 27.68 g (64.95 mmol) of 1,2-epoxy-1H,1H,2H,3H,3H-perfluorononane, dissolved in 50 ml of n-butanol/50 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 80°C. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water and set at pH 3 with 3N hydrochloric acid.

Then, it is extracted twice with 200 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 30.34 g (78% of theory) of a vitreous solid

Water content: 13.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 37.32 H 4.04 F 36.89 N 7.25

Fnd: C 37.15 H 4.21 F 36.70 N 7.19

- b) Gadolinium complex of 10-[2-hydroxy-1H,1H,2H,3H,3H-perfluorononyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (12.94 mmol) of the title compound of Example 9a) is dissolved in 100 ml of water/50 ml of ethanol, and 2.34 g (6.47 mmol) of gadolinium oxide is added. It is stirred for 3 hours at 80°C. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 13.16 g (quantitative) of a colorless, vitreous solid

Water content: 9.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 31.11 H 3.05 F 30.75 Gd 16.97 N 6.05

Fnd: C 31.01 H 3.19 F 30.55 Gd 16.71 N 5.88

Example 10

a) 9H,9H,10H,11H,12H,12H-Perfluoroeicos-10-ene

24.77 g (52.26 mmol) of 1H,1H,2H,2H-perfluorodecyl-1-iodide and 13.71 g (52.26 mmol) of triphenylphosphine are heated to 70°C in 500 ml of acetone while being stirred. The initially clear solution quickly turns milky, and the colorless phosphonium salt is precipitated. The phosphonium salt is filtered off and dried in a vacuum at 40°C.

Yield: 38.9 g (89% of theory)

This phosphonium salt is used directly in the following reaction without purification: 5.22 g (46.5 mmol) of potassium-tert-butyrate, 0.20 g (0.75 mmol) of 18-crown 6 and 19.54 g (42.28 mmol) of 2H,2H-perfluorodecanol are added to the above-produced phosphonium salt, 38.9 g (46.5 mmol) in 250 ml of dichloromethane, and it is stirred for 10 hours at room temperature. It is evaporated to the dry state, and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/n-hexane/diethyl ether = 10/20/1).

Yield: 30.3 g (65% of theory relative to the iodide used)
of a colorless, waxy solid

Elementary analysis:

Cld: C 26.92 H 0.68 F 72.40

Fnd: C 26.81 H 0.79 F 72.20

b) 10,11-Epoxy-9H,9H,10H,11H,12H,12H-perfluoroeicosane

10.47 g (36.42 mmol) of 3-chloroperoxybenzoic acid (about 60%) is added to 25 g (28.02 mmol) of the title compound of Example 10a), dissolved in 250 ml of dichloromethane, at 0°C, and it is stirred overnight at room temperature. 300 ml of 5% aqueous sodium carbonate solution is added and stirred thoroughly. The organic phase is separated, dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: n-hexane/dichloromethane/diethyl ether = 10/10/1).

Yield: 24.17 g (95% of theory) of a colorless solid

Elementary analysis:

Cld: C 26.45 H 0.67 F 71.12

Fnd: C 26.25 H 0.88 F 71.35

c) 10-[1-(1H,1H-Perfluorononyl)-2-hydroxy-1H,2H,3H,3H-perfluoroundecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

7.04 g (0.176 mmol) of sodium hydroxide is added to 7.63 g (22.02 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 35 ml of water. A solution of 20 g (22.02 mmol) of the title compound of Example 10b), dissolved in 50 ml of n-butanol/40 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 120°C in an autoclave. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water and set at pH 3 with 3N hydrochloric

acid. Then, it is extracted twice with 300 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 9.79 g (31% of theory) of a colorless, vitreous solid

Water content: 12.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 32.55 H 2.57 F 51.49 N 4.47

Fnd: C 32.38 H 2.75 F 51.29 N 4.28

d) Gadolinium complex of 10-[1-(1H,1H-perfluorononyl)-2-hydroxy-1H,2H,3H,3H-perfluoroundecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

8 g (6.38 mmol) of the title compound of Example 10c) is dissolved in 50 ml of water/40 ml of ethanol/20 ml of chloroform, and 1.16 g (3.19 mmol) of gadolinium oxide is added. It is stirred for 4 hours at 90°C in an autoclave. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 9.47 g (quantitative) of a vitreous solid

Water content: 5.2%

Elementary analysis (relative to anhydrous substance):

Cld: C 28.99 H 2.07 F 45.85 Gd 11.16 N 3.98

Fnd: C 28.81 H 2.19 F 45.71 Gd 11.03 N 4.12

Example 11

a) 7H,7H,8H,9H,10H,10H-Perfluorohexadec-8-ene

18.7 g (50 mmol) of 1H,1H,2H,2H-perfluorooctyl-1-iodide and 13.11 g (50 mmol) of triphenylphosphine are heated to 70°C in 400 ml of acetone while being stirred. The initially clear solution quickly turns milky, and the colorless phosphonium salt is precipitated. The phosphonium salt is filtered off and dried in a vacuum at 40°C.

Yield: 28.95 g (91% of theory)

This phosphonium salt is used directly in the following reaction without purification: 5.05 g (45.5 mmol) of potassium-tert-butyrate, 0.20 g (0.75 mmol) of 18-crown 6 and 14.98 g (41.36 mmol) of the title compound of Example 8a) are added to the above-produced phosphonium salt, 28.95 g (45.5 mmol) in 200 ml of dichloromethane, and it is stirred for 10 hours at room temperature. It is evaporated to the dry state, and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/n-hexane/diethyl ether = 10/20/1).

Yield: 19.65 g (61% of theory) of a colorless, waxy solid

Elementary analysis:

Clcd: C 22.38 H 0.94 F 76.69

Fnd: C 22.20 H 0.99 F 76.51

b) 8,9-Epoxy-7H,7H,8H,9H,10H,10H-perfluorohexadecane

11.03 g (38.35 mmol) of 3-chloroperoxybenzoic acid (about 60%) is added to 19 g (29.5 mmol) of the title compound of

Example 11a), dissolved in 200 ml of dichloromethane, at 0°C, and it is stirred overnight at room temperature. 300 ml of 5% aqueous sodium carbonate solution is added and stirred thoroughly. The organic phase is separated, dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: n-hexane/dichloromethane/diethyl ether = 10/10/1).

Yield: 19.43 g (93% of theory) of a colorless solid

Elementary analysis:

Clcd: C 27.14 H 0.85 F 69.75

Fnd: C 27.01 H 0.97 F 69.60

c) 10-[1-(1H,1H-Perfluoroheptyl)-2-hydroxy-1H,2H,3H,3H-perfluorononyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

8.59 g (214.6 mmol) of sodium hydroxide is added to 9.3 g (26.83 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 50 ml of water. A solution of 19 g (26.83 mmol) of the title compound of Example 11b), dissolved in 70 ml of n-butanol/60 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 120°C in an autoclave. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water and set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 300 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-

18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 9.4 g (29% of theory) of a vitreous solid

Water content: 12.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 34.17 H 3.06 F 46.84 N 5.31

Fnd: C 33.98 H 3.18 F 46.65 N 5.20

d) Gadolinium complex of 10-[1-(1H,1H-perfluoroheptyl)-2-hydroxy-1H,2H,3H,3H-perfluorononyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

9 g (8.53 mmol) of the title compound of Example 11c) is dissolved in 60 ml of water/40 ml of ethanol/30 ml of chloroform, and 1.54 g (4.27 mmol) of gadolinium oxide is added. It is stirred for 4 hours at 90°C in an autoclave. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 11.45 g (quantitative) of a colorless, vitreous solid

Water content: 10.2%

Elementary analysis (relative to anhydrous substance):

Cld: C 29.81 H 2.42 F 40.86 Gd 13.01 N 4.63

Fnd: C 29.60 H 2.60 F 40.63 Gd 12.84 N 4.51

Example 12**a) 7,12-Dioxa-5H,5H,6H,6H,8H,8H,9H,10H,11H,11H,13H,13H,14H,14H-perfluorooctadec-9-ene**

30 g (91.74 mmol) of 1H,1H,2H,2H-perfluorohexyl-1-bromide is dissolved in 100 ml of toluene, then 3.23 g (36.7 mmol) of cis-1,4-butene-diol and 1 g (2.95 mmol) of tetrabutylammonium hydrogen sulfate are added. It is cooled to 0°C, and 16 g (400 mmol) of finely powdered sodium hydroxide is added. Then, it is stirred for 1 hour at 0°C and overnight at room temperature. Solid is filtered out, the filtrate is washed twice with 200 ml of water each, the organic phase is dried on magnesium sulfate and then evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/n-hexane/acetone = 15/15/1).

Yield: 11.71 g (55% of theory relative to diol) of a waxy solid

Elementary analysis:

Clcd: C 33.12 H 2.43 F 58.93

Fnd: C 33.05 H 2.61 F 58.73

b) 9,10-Epoxy-7,12-dioxa-

5H,5H,6H,6H,8H,8H,9H,10H,11H,11H,13H,13H,14H,14H-perfluorooctadecane

7.08 g (24.64 mmol) of 3-chloroperoxybenzoic acid (about 60%) is added to 11 g (18.96 mmol) of the title compound of Example 12a), dissolved in 100 ml of dichloromethane, at 0°C, and

it is stirred overnight at room temperature. 150 ml of 5% aqueous sodium carbonate solution is added and stirred thoroughly. The organic phase is separated, dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: n-hexane/dichloromethane/diethyl ether = 10/10/1).

Yield: 10.74 g (95% of theory) of a colorless solid

Elementary analysis:

Clcd: C 32.23 H 2.37 F 57.35

Fnd: C 32.13 H 2.51 F 57.20

c) 10-[1-(2-Oxa-1H,1H,3H,3H,4H,4H-perfluorooctyl)-2-hydroxy-4-oxa-1H,2H,3H,3H,5H,5H,6H,6H-perfluorodecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

5.63 g (141 mmol) of sodium hydroxide is added to 6.1 g (17.61 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 40 ml of water. A solution of 10.5 g (17.61 mmol) of the title compound of Example 12b), dissolved in 50 ml of n-butanol/40 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 120°C in an autoclave. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water and set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 300 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-

18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 4.96 g (27% of theory) of a colorless, vitreous solid

Water content: 9.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 38.27 H 4.17 F 36.32 N 5.95

Fnd: C 38.12 H 4.20 F 36.20 N 5.81

- d) Gadolinium complex of 10-[1-(2-oxa-1H,1H,3H,3H,4H,4H-perfluorooctyl)-2-hydroxy-4-oxa-1H,2H,3H,3H,5H,5H,6H,6H-perfluorodecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

4.7 g (5 mmol) of the title compound of Example 12c) is dissolved in 30 ml of water/30 ml of ethanol/20 ml of chloroform, and 0.90 g (2.5 mmol) of gadolinium oxide is added. It is stirred for 3.5 hours at 90°C in an autoclave. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 5.89 g (quantitative) of a colorless, vitreous solid

Water content: 7.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 32.88 H 3.31 F 31.21 Gd 14.35 N 5.11

Fnd: C 32.67 H 3.45 F 31.04 Gd 14.18 N 5.02

Example 13

- a) 1-Phenyl-2,6-dioxo-1H,1H,3H,3H,4H,5H,5H,7H,7H,8H,8H-perfluorohexa-decan-4-ol

1 g (2.94 mmol) of tetrabutylammonium hydrogen sulfate and 15.6 g (390 mmol) of finely powdered sodium hydroxide are added to 7.14 g (39.2 mmol) of glycerol-1-monobenzylether and 25 g (43.55 mmol) of 1H,1H,2H,2H-perfluorodecyl-1-iodide in 100 ml of toluene. It is stirred for 24 hours at room temperature. The organic phase is separated from the solid and washed twice with 5% aqueous hydrochloric acid each. The organic phase is dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: n-hexane/acetone = 15/1).

Yield: 19.95 g (81% of theory) of a colorless oil

Elementary analysis:

Cld: C 38.23 H 2.73 F 51.40

Fnd: C 38.10 H 2.89 F 51.25

- b) 1-Phenyl-4-(decyloxy)-2,6-dioxo-1H,1H,3H,3H,4H,5H,5H,7H,7H,8H,8H-perfluorohexadecane

1.12 g (37.24 mmol) of sodium hydride (80% suspension in mineral oil) is added in portions to 19.5 g (31.03 mmol) of the title compound of Example 13a), dissolved in 100 ml of dimethylformamide, and it is stirred for 2 hours at room temperature. Then, 8.24 g (37.24 mmol) of n-decyl bromide is added and stirred overnight at 50°C. 150 ml of ice water is

added and extracted twice with 150 ml of ethyl acetate each. The combined organic phases are washed twice with 150 ml of water each, dried on magnesium sulfate and evaporated to the dry state. The residue is chromatographed on silica gel (mobile solvent: n-hexane/acetone = 20:1).

Yield: 22.66 g (95% of theory) of a waxy solid

Elementary analysis:

Cld: C 46.88 H 4.85 F 42.02

Fnd: C 46.64 H 4.97 F 41.87

c) 2-(Decyloxy)-4-oxa-1H,1H,2H,3H,3H,5H,5H,6H,6H-perfluorotetradecan-1-ol

20 g (26.02 mmol) of the title compound of Example 13b) is dissolved in 200 ml of isopropanol, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. The catalyst is filtered off, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 17.65 g (quantitative) of a colorless solid

Elementary analysis:

Cld: C 40.72 H 4.61 F 47.60

Fnd: C 40.55 H 4.76 F 47.43

d) 1,2-Epoxy-4-oxa-6-(decyloxy)-8-oxa-
1H,1H,2H,3H,3H,5H,5H,6H,7H,7H,9H,9H,10H,10H-
perfluorooctadecane

9.25 g (100 mmol) of epichlorohydrin is added in drops to a mixture of 17 g (25.06 mmol) of the title compound of Example 13c) and 2 g (5.89 mmol) of tetrabutylammonium hydrogen sulfate in 300 ml of 60% aqueous potassium hydroxide solution/100 ml of toluene while being stirred vigorously at 10°C, and care is taken to ensure that the temperature of the reaction solution does not exceed 20°C. It is allowed to stir for 2 hours at 15°C, and then 4.63 g (50 mmol) of epichlorohydrin is added in drops as described above. Then, it is stirred overnight at room temperature. 100 ml of toluene and methyl-tert-butyl ether are added, and the aqueous phase is separated. The latter is extracted twice more with 100 ml of toluene each. The organic phases are combined, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/hexane/acetone = 20/10/1).

Yield: 14.91 g (81% of theory) of a colorless solid

Elementary analysis:

Clcd: C 42.51 H 4.80 F 43.97

Fnd: C 42.37 H 4.96 F 43.68

e) 10-[2-Hydroxy-4,8-dioxa-6-(decyloxy)-
1H,1H,2H,3H,3H,5H,5H,6H,7H,7H,9H,9H,10H,10H-
perfluorooctadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-
tetraazacyclododecane

6.11 g (152.8 mmol) of sodium hydroxide is added to 6.6 g (19.06 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 60 ml of water. A solution of 14 g (19.06 mmol) of the title compound of Example 13d), dissolved in 80 ml of n-butanol/40 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 80°C in an autoclave. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water, and it is set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 300 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 17.88 g (76% of theory) of a vitreous solid

Water content: 12.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 44.49 H 5.60 F 29.91 N 5.19

Fnd: C 44.31 H 5.75 F 29.70 N 5.03

- f) Gadolinium complex of 10-[2-hydroxy-4,8-dioxa-6-(decyloxy)-1H,1H,2H,3H,3H,5H,5H,6H,7H,7H,9H,9H,10H,10H-perfluorooctadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (9.26 mmol) of the title compound of Example 13e) is dissolved in 30 ml of water/100 ml of ethanol/30 ml of chloroform, and 1.68 g (4.63 mmol) of gadolinium oxide is added. It is stirred for 3.5 hours at 90°C in an autoclave. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 12.39 g (quantitative) of a colorless, vitreous solid

Water content: 7.8%

Elementary analysis (relative to anhydrous substance):

Clcd: C 38.93 H 4.66 F 26.17 Gd 12.74 N 4.54

Fnd: C 38.71 H 4.82 F 26.01 Gd 12.55 N 4.38

Example 14

- a) 1-Phenyl-2-oxa-4,4,4-tris(2-oxa-1H,1H,3H,3H,4H,4H-perfluorodecyl)-butane

2 g (5.89 mmol) of tetrabutylammonium hydrogen sulfate and 22.48 g (562 mmol) of finely powdered sodium hydroxide are added to 4.24 g (18.74 mmol) of pentaerythritol monobenzyl ether and 40 g (93.7 mmol) of 1H,1H,2H,2H-perfluorooctyl-1-bromide in 150 ml of toluene. It is stirred for 24 hours at room temperature. The organic phase is separated from the solid and washed twice with 5% aqueous hydrochloric acid each. The organic phase is dried on

magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: n-hexane/acetone = 25/1).

Yield: 14.45 g (61% of theory relative to the benzyl ether) of a colorless, waxy solid

Elementary analysis:

Cld: C 34.19 H 2.15 F 58.59

Fnd: C 34.02 H 2.31 F 58.41

b) 2,2,2-Tris(2-oxa-1H,1H,3H,3H,4H,4H-perfluorodecyl)-ethan-1-ol

14 g (11.07 mmol) of the title compound of Example 14a) is dissolved in 100 ml of isopropanol/100 ml of tetrahydrofuran, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. The catalyst is filtered off, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 13 g (quantitative) of a colorless solid

Elementary analysis:

Cld: C 29.66 H 1.80 F 63.09

Fnd: C 29.45 H 1.97 F 62.91

- c) 1,2-Epoxy-4-oxa-6,6,6-tris(2-oxa-1H,1H,3H,3H,4H,4H-perfluorodecyl)-hexane

3.94 g (42.57 mmol) of epichlorohydrin is added in drops to a mixture of 12.5 g (10.64 mmol) of the title compound of Example 14b) and 1 g (2.95 mmol) of tetrabutylammonium hydrogen sulfate in 150 ml of 60% aqueous potassium hydroxide solution/50 ml of toluene while being stirred vigorously at 10°C, and care is taken to ensure that the temperature of the reaction solution does not exceed 20°C. It is allowed to stir for 2 hours at 15°C, and then 1.97 g (21.29 mmol) of epichlorohydrin is added in drops as described above. Then, it is stirred overnight at room temperature. 100 ml of toluene and 100 ml of methyl-tert-butyl ether are added, and the aqueous phase is separated. The latter is extracted twice more with 50 ml of toluene each. The organic phases are combined, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/hexane/acetone = 20/10/1).

Yield: 8.12 g (62% of theory) of a colorless solid

Elementary analysis:

Clcd: C 31.24 H 2.05 F 60.22

Fnd: C 31.09 H 2.19 F 60.10

- d) 10-[2-Hydroxy-4-oxa-6,6,6-tris(2-oxa-1H,1H,3H,3H,4H,4H-perfluorodecyl)-hexyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

2.08 g (52 mmol) of sodium hydroxide is added to 2.25 g (6.50 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane in 30 ml of water. A solution of 8.0 g (6.50 mmol) of the title compound of Example 14c), dissolved in 50 ml of n-butanol/30 ml of 2-propanol, is added dropwise to it, and the solution is heated overnight to 100°C in an autoclave. It is evaporated to the dry state in a vacuum, the residue is taken up in 200 ml of water and set at pH 3 with 3N hydrochloric acid. Then, it is extracted twice with 100 ml of n-butanol. The combined butanol phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-butanol/acetonitrile).

Yield: 7.79 g (67% of theory) of a colorless, vitreous solid

Water content: 11.9%

Elementary analysis (relative to anhydrous substance):

Clc: C 35.06 H 3.20 F 47.02 N 3.56

Fnd: C 34.90 H 3.38 F 46.86 N 3.47

- e) Gadolinium complex of 10-[2-hydroxy-4-oxa-6,6,6-tris(2-oxa-1H,1H,3H,3H,4H,4H-perfluorodecyl)-hexyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

7 g (4.44 mmol) of the title compound of Example 14d) is dissolved in 30 ml of water/50 ml of ethanol/50 ml of chloroform, and 0.80 g (2.22 mmol) of gadolinium oxide is added. It is stirred for 5 hours at 90°C in an autoclave. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 8.34 g (quantitative) of a colorless, vitreous solid

Water content: 8.1%

Elementary analysis (relative to anhydrous substance):

Clcd: C 31.94 H 2.74 F 42.83 Gd 9.09 N 3.24

Fnd: C 31.74 H 2.91 F 42.67 Gd 8.85 N 3.15

Example 15

- a) 1,7-Bis[acetyl-(2-(N-ethyl-N-perfluorooctylsulfonylamino))]-1,4,7-triazaheptane

20 g (34.17 mmol) of the title compound of Example 1b) and 4.33 g (37.59 mmol) of N-hydroxysuccinimide are dissolved in 150 ml of dimethylformamide. 7.76 g (37.59 mmol) of dicyclohexylcarbodiimide is added at 0°C and stirred for 3 hours at room temperature. Dicyclohexylurea is filtered out, and the filtrate is added in drops to a solution of 1.76 g (17.09 mmol) of diethylenetriamine and 13.83 g (136.7 mmol) of triethylamine in 200 ml of dimethylformamide at room temperature. It is

stirred overnight at room temperature. It is evaporated to the dry state in a vacuum, and the residue is taken up in 200 ml of 5% aqueous soda solution. It is extracted twice with 150 ml of dichloromethane each, the combined organic phases are dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/2-propanol = 20/1).

Yield: 16.5 g (78% of theory) of a waxy solid

Elementary analysis:

Cld: C 27.17 H 2.04 F 52.19 N 5.66 S 5.18

Fnd: C 27.03 H 2.17 F 52.04 N 5.49 S 5.07

b) 4-(3-Carboxy-propanoyl)-1,7-bis-{acetyl-[2-(N-ethyl-N-perfluorooctylsulfonylamino)]}-1,4,7-triazaheptane

3.92 g (38.78 mmol) of triethylamine is added to 16 g (12.93 mmol) of the title compound of Example 15a) in 100 ml of methylene chloride, and the solution is cooled to 0°C. Then, 2.59 g (25.86 mmol) of succinic acid anhydride is added and stirred for 3 hours at 0°C, overnight at room temperature. 200 ml of 5% aqueous hydrochloric acid is added and shaken well. The organic phase is separated and dried on magnesium sulfate. It is evaporated to the dry state in a vacuum, and the residue is chromatographed on silica gel (mobile solvent: dichloromethane/2-propanol = 15/1).

Yield: 15.74 g (91% of theory) of a colorless solid

Elementary analysis:

Cld: C 28.73 H 2.19 F 48.29 N 5.24 S 4.79

Fnd: C 28.58 H 2.40 F 48.17 N 5.17 S 4.65

- c) 10-[7-Hydroxy-5-aza-4-oxo-octanoic acid-N,N-bis(3-aza-4-oxo-6-aza-6-(perfluorooctylsulfonyl)-octyl)-amide]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

15 g (11.21 mmol) of the title compound of Example 15b) and 1.42 g (12.33 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 80 ml of dimethylformamide/30 ml of chloroform. 2.54 g (12.33 mmol) of dicyclohexylcarbodiimide is added at 0°C and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 4.05 g (40 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 7.07 g (12.33 mmol) of the gadolinium complex of 10-[2-hydroxy-3-amino-propyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane, dissolved in 30 ml of water, is added and stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 100 ml of methanol/50 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 17.76 g (78% of theory) of a colorless, vitreous solid

Water content: 6.8%

Elementary analysis (relative to anhydrous substance):

Cld: C 31.08 H 3.03 F 34.12 Gd 8.31 N 7.40 S 3.39

Fnd: C 30.89 H 3.15 F 34.01 Gd 8.14 N 7.25 S 3.24

Example 16

Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-(2-hydroxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane)-1,4,7,10-tetraazacyclododecane

a) 16,16,17,17,18,18,19,19,20,20,21,21,22,22,22-
Heptafluoro-3,6,9,12-tetra-oxa-docosan-1-ol

A mixture of 20 g (32.35 mmol) of 1-p-toluenesulfonyloxy-1H,1H,2H,2H-perfluorodecane [see Example 7a], 1 g of tetrabutylammonium hydrogen sulfate, 62.83 g (323.5 mmol) of tetraethylene glycol, 300 ml of dichloromethane and 100 ml of 50% sodium hydroxide solution is stirred intensively at about 5°C for 24 hours. It is then diluted with 200 ml of dichloromethane, the phases are separated, and the dichloromethane phase is washed with water. The organic phase is dried on magnesium sulfate and concentrated by evaporation in a vacuum. 18.5 g of the desired title compound is obtained as a light yellow oil.

b) 1,2-Epoxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,
26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane

A mixture of 17 g (26.5 mmol) of 16,16,17,17,18,18,19,19,20,20,21,21,22,22,22-heptafluoro-3,6,9,12-tetra-oxa-docosan-1-ol, 0.5 g of tetrabutylammonium hydrogen sulfate, 2.94 g of

epichlorohydrin, 200 ml of dichloromethane and 50 ml of 50% sodium hydroxide solution is stirred intensively at room temperature for 8 hours. The phases are separated, the aqueous phase is shaken with 100 ml of dichloromethane, the organic phases are combined, shaken with 50 ml of water, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel with hexane/5-50% ethyl acetate, and 12.92 g of the title compound is obtained as an oil.

Elementary analysis:

Cld: C 36.22 H 3.62 F 46.38

Fnd: C 36.00 H 3.78 F 46.20

- c) 1,4,7-Tris(carboxylatomethyl)-10-(2-hydroxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane)-1,4,7,10-tetraazacyclododecane

A solution of 12.05 g (17.3 mmol) of 1,2-epoxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane in 50 ml of tetrahydrofuran is added to a solution of 6 g (17.3 mmol) of 1,4,7-(triscarboxylatomethyl)-1,4,7,10-tetraazacyclododecane and 4 g of sodium hydroxide in 30 ml of water. It is stirred overnight at 70°C, then largely concentrated by evaporation in a vacuum, the residue is taken up in 150 ml of water and set at pH 3 with 6N hydrochloric acid and extracted several times with n-

butanol. The combined extracts are concentrated by evaporation in a vacuum, and the residue is purified by chromatography on RP-18 with a gradient consisting of water/n-butanol/acetonitrile. 13.71 g of the title compound is obtained as a yellow viscous oil.

Elementary analysis:

Cld: C 40.31 H 4.93 F 30.97 N 5.37

Fnd: C 40.08 H 5.21 F 30.77 N 5.29

- d) Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-(2-hydroxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane)-1,4,7,10-tetraazacyclododecane

A mixture of 5 g (4.79 mmol) of 1,4,7-tris(carboxylatomethyl)-10-(2-hydroxy-19,19,20,20,21,21,22,22,23,23,24,24,25,25,26,26,26-heptafluoro-4,7,10,13,16-penta-oxa-hexacosane)-1,4,7,10-tetraazacyclododecane, 50 ml of water and 30 ml of ethanol is mixed with 869 mg (2.397 mmol) of gadolinium oxide, and it is refluxed for 5 hours. The hot solution is filtered and concentrated by evaporation in a vacuum. 5.60 g of the title compound is obtained as a vitreous, solid substance with a water content of 4.1%.

Elementary analysis (relative to anhydrous substance):

Cld: C 35.12 H 4.04 F 26.98 Gd 13.14 N 4.68

Fnd: C 34.90 H 4.38 F 26.70 Gd 13.10 N 4.62

Example 17

Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-(4-aza-2-hydroxy-26,26,26,25,25,24,24,23,23,22,22,21,21,20,20,19,19-heptafluoro-5-oxo-16-thia-hexacosyl)-1,4,7,10-tetraazacyclododecane

a) 22,22,22,21,21,20,20,19,19,18,18,17,17,16,16,15,15-
Heptafluoro-12-thia-docosanoic acid

A solution of 10 g (37.71 mmol) of 11-bromoundecanoic acid in 150 ml of dichloromethane is mixed with 11.43 g of triethylamine and 18.11 g (37.71 mmol) of 1H,1H,2H,2H-perfluorodecylmercaptan, and it is stirred overnight at room temperature. The solution is extracted several times with 2N hydrochloric acid, washed with common salt solution, dried on magnesium sulfate and concentrated by evaporation in a vacuum. 21.5 g of the title compound is obtained as a yellow oil.

Elementary analysis:

Clcd: C 37.96 H 3.79 F 48.61 S 4.83

Fnd: C 38.30 H 4.01 F 48.40 S 5.20

b) Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-(4-aza-2-hydroxy-26,26,26,25,25,24,24,23,23,22,22,21,21,20,20,19,19-heptafluoro-5-oxo-16-thia-hexacosyl)-1,4,7,10-tetraazacyclododecane

5 g (7.52 mmol) of the title compound of Example 17a) and 0.95 g of N-hydroxysuccinimide are dissolved in a mixture of 25 ml of dimethylformamide and 15 ml of chloroform. 1.71 g of

dicyclohexylcarbodiimide is added at 0°C and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is then cooled again to 0°C and mixed with 3 ml of triethylamine and 20 ml of n-propanol. Then, 4.75 g (8.27 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxylatomethyl)-1,4,7,10-tetraazacyclododecane, dissolved in 25 ml of water, is added and stirred for 3 hours at 20°C. It is evaporated to the dry state, the residue is taken up in a mixture of 55 ml of methanol and 20 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by chromatography on RP-18 with a gradient consisting of water/n-propanol/acetonitrile. 6.15 g of the title compound is obtained as a vitreous solid, with a water content of 2.3%.

Elementary analysis (relative to anhydrous substance):

Clcd: C 37.41 H 4.38 F 26.47 Gd 12.89 N 5.74 S 2.63

Fnd: C 37.08 H 4.60 F 26.30 Gd 12.68 N 5.91 S 2.49

Example 18

Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-[1-(1,2-dihydroxyethyl)3-oxa-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro]undecane-1,4,7,10-tetraazacyclododecane

a) 1-p-Toluenesulfonyloxy-1H,1H,2H,2H-perfluorooctane

20 ml of pyridine is added to a solution of 25 g (68.7 mmol) of 1H,1H,2H,2H-perfluorooctan-1-ol in 300 ml of dichloromethane at 0°C, and 13.49 g (70.76 mmol) of p-toluenesulfonic acid chloride is added in portions while being stirred. It is stirred

for 3 more hours at 0°C, and the dichloromethane is drawn off at room temperature in a vacuum. The remaining pyridine solution is mixed with ice water, whereby the desired product precipitates. The residue is decanted and dissolved in dichloromethane, the solution is washed with water, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is purified by chromatography on silica gel with hexane/5-40% ethyl acetate. 29.2 g of the title compound is obtained as a viscous foam.

Elementary analysis:

Cld: C 34.76 H 2.14 F 47.65 S 6.19

Fnd: C 34.98 H 2.38 F 47.39 S 6.42

b) 1,4,7-Tris(benzyloxycarbonyl)-10-[1-(2,2-dimethyl-1,3-dioxolan-4-yl)-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro-3-oxa]-undecane-1,4,7,10-tetraazacyclododecane
20 ml of 50% sodium hydroxide solution, 0.5 g of tetrabutylammonium hydrogen sulfate and 5.18 g (10 mmol) of 1-p-toluenesulfonyloxy-1H,1H,2H,2H-perfluorooctane [see Example 18a)] are added in succession to 7.33 g (10 mmol) of 1,4,7-tris(benzyloxycarbonyl)-10-[2-hydroxy-1-(2,2-dimethyl-1,3-dioxolan-4-yl)]-ethyl-1,4,7,10-tetraazacyclododecane [J. Mag. Res. Imag. 5: 7-10, (1955)], dissolved in 100 ml of dichloromethane, and the mixture is stirred intensively overnight at room temperature. The phases are separated, the organic phase is washed several times with water, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is

purified by chromatography on silica gel with dichloromethane/1-10% ethanol. 8.02 g of the title compound is obtained as a viscous oil.

Elementary analysis:

Cld: C 53.01 H 5.02 F 23.19 N 5.26

Fnd: C 53.30 H 5.39 F 23.01 N 5.40

c) 1-[1-(2,2-Dimethyl-1,3-dioxolan-4-yl)-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro-3-oxa]-undecane-1,4,7,10-tetraazacyclododecane

A solution of 7 g (6.57 mmol) of 1,4,7-tris(benzyloxycarbonyl)-10-[1-(2,2-dimethyl-1,3-dioxolan-4-yl)-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro-3-oxa]-undecane-1,4,7,10-tetraazacyclododecane in 100 ml of isopropyl alcohol is mixed with 0.7 g of palladium on carbon (10%), and it is shaken for 3 hours under hydrogen atmosphere. Catalyst is filtered out, and the solution is concentrated by evaporation in a vacuum. 4.20 g of the title compound is obtained as a vitreous foam.

Elementary analysis:

Cld: C 41.70 H 5.32 F 37.28 N 8.46

Fnd: C 41.61 H 5.57 F 37.10 N 8.59

- d) 1,4,7-Tris(carboxylatomethyl)-10-[1-(1,2-dihydroxy-ethyl)-3-oxa-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro]undecane-1,4,7,10-tetraazacyclododecane

3.36 g (24.15 mmol) of bromoacetic acid in 50 ml of water is dissolved and mixed with 6N sodium hydroxide solution until pH 7 is reached. A solution of 4 g (6.04 mmol) of 1-[1-(2,2-dimethyl-1,3-dioxolan-4-yl)-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro-3-oxa]-undecane-1,4,7,10-tetraazacyclododecane, dissolved in 20 ml of isopropyl alcohol, and enough 6N sodium hydroxide solution are added in drops at 40°C while simultaneously being stirred so that the pH is kept at 9-10. Then, it is mixed with semiconcentrated hydrochloric acid up to pH 1 and stirred for another 3 hours at 60°C. It is cooled to room temperature, and the solution is extracted several times with n-butanol. The organic extract is concentrated by evaporation, and the residue is purified by chromatography on RP-18 with a gradient consisting of water/n-butanol/acetonitrile. 3.85 g of the title compound is obtained as a yellow oil with a water content of 3.9%.

Elementary analysis (relative to anhydrous substance):

Cld: C 39.20 H 4.68 F 31.00 N 7.03

Fnd: C 39.08 H 4.98 F 30.72 N 7.29

- e) Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-[1-(1,2-dihydroxy-ethyl)-3-oxa-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro]undecane-1,4,7,10-tetraazacyclododecane

A mixture of 1.59 g (2 mmol) of 1,4,7-tris(carboxylatomethyl)-10-[1-(1,2-dihydroxy-ethyl)-3-oxa-6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro]undecane-1,4,7,10-tetraazacyclododecane, 25 ml of water and 15 ml of ethanol is mixed with 363 mg (1 mmol) of gadolinium oxide, and it is refluxed for 5 hours. The hot solution is filtered, concentrated by evaporation in a vacuum, and 1.85 g of the title compound is obtained as a vitreous, solid substance with a water content of 4.2%.

Elementary analysis (relative to anhydrous substance):

Clcd: C 32.84 H 3.60 F 25.98 Gd 16.54 N 5.89

Fnd: C 32.53 H 3.71 F 25.72 Gd 16.39 N 5.93

Example 19

Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-{2-hydroxy-4-oxa-4-[4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)]-phenyl}-but-1-yl-1,4,7,10-tetraazacyclododecane

- a) 1-Hydroxy-4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)-benzene

5 g (45.41 mmol) of hydroquinone is mixed with 100 ml of acetone and mixed while being stirred in succession with 13.8 g of potassium carbonate and 14.04 g (22.7 mmol) of 1-p-toluenesulfonyloxy-1H,1H,2H,2H-perfluorodecane [see Example 7a)]. It is refluxed for 6 hours, then largely concentrated by

evaporation in a vacuum, diluted with 200 ml of water, set at pH 3 with citric acid and extracted several times with dichloromethane. The organic extract is dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is purified by chromatography on silica gel with hexane/5-30% ethyl acetate. 8.20 g of the desired title compound is obtained as a viscous oil.

Elementary analysis:

Cld: C 34.55 H 1.63 F 58.07

Fnd: C 34.31 H 1.79 F 58.01

- b) 1-(3,4-Epoxy-1-oxa-but-1-yl)-4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)-benzene

A mixture of 8 g (14.38 mmol) of 1-hydroxy-4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)-benzene, 0.4 g of tetrabutylammonium hydrogen sulfate, 1.60 g (17.26 mmol) of epichlorohydrin, 150 ml of dichloromethane and 30 ml of 50% sodium hydroxide solution is stirred intensively for 30 minutes in an ice bath, then for 5 hours at room temperature. The phases are separated, the organic phase is washed with water, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is purified by chromatography on silica gel with hexane/5-30% ethyl acetate, and 6.60 g of the title compound is obtained as a viscous oil.

Elementary analysis:

Cld: C 37.27 H 2.41 F 52.75

Fnd: C 37.10 H 2.66 F 52.80

- c) 1,4,7-Tris(carboxylatomethyl)-10-{2-hydroxy-4-oxa-4-[4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)]-phenyl}-but-1-yl-1,4,7,10-tetraazacyclododecane

A solution of 6.12 g (10 mmol) of 1-(3,4-epoxy-1-oxa-but-1-yl)-4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)-benzene in 25 ml of tetrahydrofuran is added to a solution of 3.46 g (10 mmol) of 1,4,7-tris(carboxylatomethyl)-1,4,7,10-tetraazacyclododecane and 2.5 g of sodium hydroxide in 25 ml of water, and it is refluxed for 24 hours, then largely concentrated by evaporation in a vacuum, the residue is dissolved in 100 ml of water, set at pH 3 with 6N hydrochloric acid and extracted several times with n-butanol. The combined extracts are concentrated by evaporation in a vacuum. The residue is purified by chromatography on RP-18 with a gradient consisting of water/n-butanol/acetonitrile. 6.71 g of the title compound is obtained as a viscous oil.

Elementary analysis:

Cld: C 41.35 H 4.10 F 33.69 N 5.84

Fnd: C 41.58 H 4.38 F 33.50 N 5.91

d) Gadolinium complex of 1,4,7-tris(carboxylatomethyl)-10-{2-hydroxy-4-oxa-4-[4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)]-phenyl}-but-1-yl-1,4,7,10-tetraazacyclododecane

A mixture of 4.79 g (5 mmol) of 1,4,7-tris(carboxylatomethyl)-10-{2-hydroxy-4-oxa-4-[4-(2H,2H,3H,3H-1-oxa-perfluoroundec-1-yl)]-phenyl}-but-1-yl-1,4,7,10-tetraazacyclododecane, 50 ml of water and 30 ml of ethanol is mixed with 906 mg (2.5 mmol) of gadolinium oxide and refluxed for 5 hours. The hot solution is filtered and concentrated by evaporation in a vacuum. 5.50 g of the title compound is obtained as a vitreous solid substance with a water content of 4.9%.

Elementary analysis (relative to anhydrous substance):

Cld: C 35.62 H 3.26 F 29.02 Gd 14.13 N 5.03

Fnd: C 35.40 H 3.50 F 28.81 Gd 14.01 N 5.18

Example 20

Gadolinium complex, disodium salt of 3,9-bis(carboxymethyl)-6-[(1-carboxy)-1H,2H,2H,4H,4H,5H,5H-3-oxa-perfluorotridecyl]-3,6,9-triazaundecanedioic acid

a) N-t-Butoxycarbonyl-serine-(1H,1H,2H,2H-perfluorodecyl)-ether-benzyl ester

300 mg (10 mmol) of sodium hydride (80% in oil) is added in portions to a solution of 2.953 g (10 mmol) of N-t-butyloxycarbonyl-serine-benzyl ester (Bachem commercially available products) in 30 ml of dry dimethylformamide. After

dissolving is completed, it is mixed with 6.072 g (10 mmol) of the tosylate produced under 7a). It is stirred for 12 hours at room temperature. Then, it is poured into 500 ml of ice water, the product is taken up in dichloromethane, the organic solution is washed with water, dried on sodium sulfate and evaporated to the dry state. The residue is purified by chromatography on silica gel. A mixture of dichloromethane with increasing addition of methanol is used as eluant.

The title compound is obtained as a syrup.

Yield: 5.902 g (79.6% of theory)

Elementary analysis:

Cld: C 40.50 H 3.26 F 43.56 N 1.89

Fnd: C 40.64 H 3.37 F 43.49 N 1.83

b) Serine-(1H,1H,2H,2H-perfluorodecyl)-ether-benzyl ester (as a salt of trifluoroacetic acid

7.414 g (10 mmol) of the N-protected compound that is produced under 20a) is dissolved in 50 ml of a mixture of trifluoroacetic acid and dichloromethane at a 2:1 ratio, and it is stirred overnight at room temperature. It is evaporated to the dry state, and the remainder of the trifluoroacetic acid is removed by codistillation with ethanol. The title compound is isolated as a salt of trifluoroacetic acid.

Yield: 7.418 g (98.2% of theory)

Elementary analysis:

Cld: C 34.98 H 2.27 F 50.30 N 1.85

Fnd: C 34.89 H 2.31 F 50.39 N 1.80

- c) 3,9-Bis(t-butoxycarbonylmethyl)-6-[(1-benzyloxycarbonyl)-1H,2H,2H,4H,4H,5H,5H-3-oxa-perfluorotridecyl)-3,6,9-triazaundecanedioic acid-di(t-butyl)-ester

3.777 g (5 mmol) of the amine-trifluoroacetate that is produced under 20b) and 3.523 g (10 mmol) of N,N-bis(t-butyloxycarbonylmethyl)-2-(bromoethyl)-amine are added to a mixture of 10 ml of acetonitrile and 20 ml of phosphate buffer of pH 8.0, and it is stirred intensively at room temperature for 2 hours. Then, the buffer phase is separated, extracted with 10 ml of acetonitrile, and the latter is added to the organic phase. After 20 ml of fresh buffer is added, it is stirred for 20 more hours at room temperature. The organic phase is separated, concentrated by evaporation, and the residue is dispersed between 100 ml of phosphate buffer (pH 8.0) and 100 ml of ethyl acetate. The organic phase is washed with saturated common salt solution, dried on sodium sulfate and concentrated by evaporation. The title compound is purified by chromatography on silica gel. Dichloromethane is used as eluant with increasing addition of methanol. The title compound is obtained as a glass-like solid.

Yield: 3.162 g (53.4% of theory)

Elementary analysis:

Cld: C 48.69 H 5.62 F 27.28 N 3.55

Fnd: C 48.82 H 5.72 F 27.37 N 3.50

d) 3,9-Bis(carboxymethyl)-6-[(1-carboxy)-1H,2H,2H,4H,4H,5H,5H-3-oxa-perfluorotridecyl]-3,6,9-triazaundecanedioic acid

5.920 g (5 mmol) of the compound that is produced under 20c) is added to a mixture of 25 ml of trifluoroacetic acid/dichloromethane at a 2:1 ratio. It is allowed to stir overnight at room temperature, then evaporated to the dry state, the residue is taken up in 100 ml of 3N hydrochloric acid, refluxed for 3 hours, then evaporated to the dry state in a vacuum and taken up in 160 ml of a mixture of water, ethanol and chloroform (10:5:1). The solution is set at a constant pH (about 3) by adding ion exchanger IRA 67 (OH⁻ form). It is quickly suctioned out, concentrated by evaporation, and the title compound is obtained as a vitreous solid.

Yield: 3.080 g (71.3% of theory)

Water content: 11.3%

Elementary analysis (relative to anhydrous substance):

Cld: C 34.53 H 3.25 F 37.15 N 4.83

Fnd: C 34.41 H 3.32 F 37.29 N 4.90

- e) Gadolinium complex, disodium salt of 3,9-bis(carboxymethyl)-6-[(1-carboxy)-1H,2H,2H,4H,4H,5H,5H-3-oxa-perfluorotridecyl]-3,6,9-triazaundecanedioic acid

2.941 g (3.0 mmol, relative to 11.3% water content) of the acid that is produced under 20d) is added to a mixture of 60 ml of distilled water and 30 ml of ethanol. 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. After addition is completed, it is stirred until dissolved. The pH of the solution is then set at 7.2 by adding sodium hydroxide solution. The solution is then concentrated by evaporation, whereby strong foaming can be observed. The residue is codistilled with distilled water. The title compound is obtained as a vitreous solid.

Yield: 3.489 g (quantitative)

Water content: 8.2%

Elementary analysis (relative to anhydrous substance):

Cld: C 28.12 H 2.17 F 30.25 Gd 14.73 N 3.94 Na 4.31

Fnd: C 28.25 H 2.26 F 30.40 Gd 14.85 N 3.99 Na 4.38

Example 21

Gadolinium complex, monosodium salt of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-mono-N-{ethyl-2-amino-[carbonylmethyl-amino-(N-ethyl-N-perfluorooctylsulfonyl)]}-amide

a) 3,6,9-Tris(carboxylatomethyl)-3,6,9-triazaundecanedioic acid-mono-N-{ethyl-2-amino-[carbonylmethyl-amino-(N-ethyl-N-perfluorooctylsulfonyl)]}-amide

17.87 g (50 mmol) of diethylenetriaminepentaacetic acid-bis-anhydride is suspended in 200 ml of a mixture of dimethylformamide and dichloromethane at a 4:1 ratio and mixed in portions with the mixture of 3.137 g (5 mmol) of [N-(2-aminoethyl)-N-perfluorooctylsulfonyl]-aminoacetic acid-N-(2-aminoethyl)-amide and 6.50 g (64.2 mmol) of triethylamine while being stirred vigorously. It is allowed to stir for 5 more hours, evaporated to the dry state, mixed with 300 ml of ice water, and the pH of the batch is set at about 3 with 3N hydrochloric acid. It is extracted twice with 200 ml of n-butanol each, the organic solutions are combined and concentrated by evaporation. The product is purified by chromatography on silica gel RP-18. Water and tetrahydrofuran are used as eluants. The title compound is obtained as a vitreous solid.

Yield: 2.722 g (54.3% of theory)

Water content: 9.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 33.54 H 3.52 F 32.21 N 8.38 S 3.20

Fnd: C 33.65 H 3.60 F 32.14 N 8.51 S 3.29

- b) Gadolinium complex, monosodium salt of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-mono-N-{ethyl-2-amino-[carbonylmethyl-amino-(N-ethyl-N-perfluorooctylsulfonyl)]}-amide

3.259 g (3 mmol, relative to 9.7% water) of the compound that is produced under 21a) is added to 90 ml of a mixture of distilled water and ethanol (2:1). While being stirred, 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions. It is stirred until dissolved, then the pH of the solution is set at 7.2 by adding sodium hydroxide solution, and it is concentrated by evaporation, whereby strong foaming occurs. The residue is codistilled with distilled water. The title compound is obtained as a glass-like solid.

Yield: 3.861 g (quantitative)

Water content: 8.4%

The elementary analysis is calculated based on anhydrous substance:

Cld: C 28.53 H 2.65 F 27.40 Gd 13.34 N 7.13 Na 1.95 S 2.72

Fnd: C 28.61 H 2.68 F 27.48 Gd 13.40 N 7.08 Na 1.99 S 2.76

Example 22

Gadolinium complex, monosodium salt of 3,9-bis(carboxymethyl)-6-1H,1H,4H,4H,5H,5H,8H,8H,10H,10H,11H,11H-2,7-dioxo-3,6-diaza-9-oxa-perfluoromonodecyl)-3,6,9-triazaundecanedioic acid

a) Glycolic acid-(1H,1H,2H,2H-perfluorodecyl)-ether-N-(2-aminoethyl)-amide

10.44 g (20 mmol) of compound 2b) is dissolved in 80 ml of dichloromethane and mixed with 2.30 g (20 mmol) of N-hydroxysuccinimide as well as 4.13 g (20 mmol) of dicyclohexylcarbodiimide. It is allowed to stir overnight, dicyclohexylurea is filtered out, and the filtrate is stirred in a solution of 60.1 g (1000 mmol) of ethylenediamine in 100 ml of dichloromethane. It is allowed to stir overnight, mixed with 1.5 l of water, and the organic phase is separated. The dichloromethane solution is washed with water, dried on sodium sulfate, evaporated to the dry state, and the residue is purified by chromatography on silica gel. A mixture of dichloromethane with increasing addition of isopropanol is used as eluant.

Yield: 9.615 g (85.2% of theory)

Elementary analysis:

Cld: C 29.80 H 2.32 F 57.24 N 4.96

Fnd: C 29.96 H 2.37 F 57.12 N 5.01

- b) Glycolic acid-(1H,1H,2H,2H-perfluorodecyl)-ether-N-[ethyl-2-(benzyloxycarbonyl-aminomethylcarbonylamino)]-amide

2.092 g (10 mmol) of benzyloxycarbonylglycine is dissolved in 15 ml of dichloromethane and mixed with 1.151 g (10 mmol) of N-hydroxysuccinimide as well as 2.063 g (10 mmol) of dicyclohexylcarbodiimide. It is allowed to stir overnight, dicyclohexylurea is filtered out and evaporated to the dry state. The residue is purified on silica gel by column chromatography. A mixture of dichloromethane and ethanol is used as eluant. The title compound is obtained as a vitreous solid.

Yield: 6.905 g (91.4% of theory)

Elementary analysis:

Clcd: C 38.16 H 2.94 F 42.75 N 5.56

Fnd: C 38.28 H 2.98 F 42.82 N 5.50

- c) Glycolic acid-(1H,1H,2H,2H-perfluorodecyl)-ether-N-[ethyl-(2-aminomethyl-carboxylamino)]-amide

3.777 g (5 mmol) of the compound that is produced under 22b) is hydrogenated in 100 ml of a mixture of tetrahydrofuran and ethanol at a 2:1 ratio in the presence of 0.2 g of Pearlman's catalyst (Pd 20%/C) until 112 ml of hydrogen is taken up. Catalyst is suctioned out, rewashed well with ethanol and evaporated to the dry state. The title compound is obtained as a glass-like solid.

Yield: 3.097 g (99.7% of theory)

Elementary analysis:

Cld: C 30.93 H 2.60 F 51.98 N 6.76

Fnd: C 30.87 H 2.64 F 52.11 N 6.82

d) 3,9-Bis(t-butyloxycarbonylmethyl)-6-

(1H,1H,4H,4H,5H,5H,8H,8H,10H,10H,11H,11H,-2,7-dioxo-3,6-diaza-9-oxa-perfluorononadecyl)3,6,9-triazaundecanedioic acid-bis(t-butylester)

3.107 g (5 mmol) of the amine that is produced under 22c) and 3.523 g (10 mmol) of N,N-bis(t-butyloxycarbonylmethyl)-2-(bromoethyl)-amine are added to a mixture of 10 ml of acetonitrile and 20 ml of phosphate buffer of pH 8 and stirred intensively for 2 hours at room temperature. Then, the buffer phase is separated, it is extracted with 10 ml of acetonitrile, and the latter is added to the organic phase. After 20 ml of fresh buffer is added, it is stirred for 20 more hours at room temperature. The organic phase is separated, it is concentrated by evaporation, and the residue is dispersed between 100 ml of phosphate buffer (pH 8.0) and 100 ml of ethyl acetate. The organic phase is washed with saturated common salt solution, dried on sodium sulfate and concentrated by evaporation. The compound is purified on silica gel by chromatography. Dichloromethane with increasing addition of methanol is used as eluant. The title compound is obtained as a glass-like solid.

Yield: 3.044 g (52.3% of theory)

Elementary analysis:

Cld: C 45.40 H 5.71 F 27.75 N 6.02

Fnd: C 45.47 H 5.78 F 27.68 N 6.10

e) 3,9-Bis(carboxymethyl)-6-

(1H,1H,4H,4H,5H,5H,8H,8H,10H,10H,11H,11H-2,7-dioxo-3,6-diaza-9-oxa-perfluoromonodecyl)-3,6,9-triazaundecanedioic acid

5.820 g (5 mmol) of the compound that is produced under 22d) is added to a mixture of 120 ml of trifluoroacetic acid/dichloromethane at a 2:1 ratio. It is allowed to stir overnight at room temperature, evaporated to the dry state, the remainder of trifluoroacetic acid is removed by codistillation with ethanol and taken up in 240 ml of a mixture of water, ethanol and chloroform. The solution is set at a constant pH (about 3) by adding ion exchanger IRA-67 (OH⁻ form). It is quickly suctioned out, concentrated by evaporation, and the title compound is obtained as a vitreous solid.

Yield: 3.214 g (68.4% of theory)

Water content: 10.3%

Elementary analysis (relative to anhydrous substance):

Cld: C 35.79 H 3.65 F 34.37 N 7.45

Fnd: C 35.90 H 3.72 F 34.31 N 7.51

- f) Gadolinium complex, monosodium salt of 3,9-bis(carboxymethyl)-6-(1H,1H,4H,4H,5H,5H,8H,8H,10H,10H,11H,11H-2,7-dioxo-3,6-diaza-9-oxa-perfluorononadecyl)-3,6,9-triaza-undecanedioic acid

3.143 g (3.0 mmol, relative to 10.3% water content) of the acid that is produced under 22e) is added to a mixture of 60 ml of distilled water and 30 ml of ethanol. 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. After addition is completed, it is stirred until dissolved. Then, the pH of the solution is set at 7.2 by adding sodium hydroxide solution, the solution is concentrated by evaporation, whereby strong foaming can be observed. The residue is codistilled with distilled water. The title compound is obtained as a vitreous solid.

Yield: 3.635 g (quantitative)

Water content: 7.9%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.14 H 2.71 F 28.95 Gd 14.09 N 6.28 Na 2.06

Fnd: C 30.21 H 2.78 F 29.03 Gd 14.16 N 6.22 Na 2.11

Example 23

Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-[2-aminoethyl-(N-ethyl-N-perfluorooctylsulfonyl)-amide]}

a) N-Ethyl-(2-benzyloxycarbonylamino-ethyl)-perfluorooctylsulfonic acid amide

5.272 g (10 mmol) of perfluorooctylsulfonic acid-N-ethylamide is dissolved in 30 ml of dimethylformamide. With exclusion of moisture, it is mixed with 330 mg (11 mmol) of sodium hydride (80% in oil). After gas generation is completed, the solution of 2.093 g (10 mmol) of N-benzyloxycarbonylaziridine is added dropwise to it. It is poured into 300 ml of ice water, extracted with dichloromethane, the organic solution is washed with water, it is dried on sodium sulfate and evaporated to the dry state. The residue is chromatographed on silica gel with dichloromethane/methanol. The title compound is a glass-like solid.

Yield: 6.149 g (87.3% of theory)

Elementary analysis:

Clcd: C 34.10 H 2.43 F 45.85 N 3.98 S 4.55

Fnd: C 34.00 H 2.49 F 45.97 N 4.06 S 4.49

b) N-Ethyl-N-2-(aminoethyl)-perfluorooctylsulfonamide

3.522 g (5 mmol) of the compound that is produced under 23a) is hydrogenated in 100 ml of a mixture of tetrahydrofuran and ethanol at a 2:1 ratio in the presence of 0.2 g of Pearlman's

catalyst (Pd 20%/C) until 112 ml of hydrogen is taken up. Catalyst is suctioned out, it is rewashed well with ethanol and evaporated to the dry state. The title compound is obtained as an amorphous solid.

Yield: 2.814 g (98.7% of theory)

Elementary analysis:

Cld: C 25.27 H 1.94 F 56.64 N 4.91 S 5.62

Fnd: C 25.39 H 1.99 F 56.57 N 4.96 S 5.53

c) 3,6,9-Tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-[2-aminoethyl-(N-ethyl-N-perfluorooctylsulfonyl)]-amide}

5.703 g (10 mmol) of the compound that is produced under 23b) as well as 1.518 g (15 mmol) of triethylamine are dissolved in 30 ml of dry dimethylformamide and mixed in portions with 1.787 g (5 mmol) of diethylenetriaminepentaacetic acid-bisanhydride while being stirred and with exclusion of moisture. It is allowed to stir overnight, then concentrated by evaporation, mixed with water, the pH is set at about 3 with 3N hydrochloric acid, and it is extracted twice with 100 ml of n-butanol each. The organic solutions are combined, concentrated by evaporation and subjected to a chromatography on silica gel RP-18. Water and tetrahydrofuran are used as eluant. The title compound is obtained as a glass-like solid.

Yield: 6.172 g (82.4% of theory)

Water content: 9.8%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.47 H 2.76 F 43.12 N 6.55 S 4.28

Fnd: C 30.59 H 2.81 F 43.00 N 6.61 S 4.33

- d) Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-[2-aminoethyl-(N-ethyl-N-perfluorooctylsulfonyl)]-amide}

6.570 g (4 mmol, relative to 9.8% water content) of the compound that is produced under 23c) is added to a mixture of 120 ml of distilled water, 60 ml of ethanol and 20 ml of chloroform. 725 mg (82.0 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. It is stirred until dissolved, then concentrated by evaporation, whereby strong foaming occurs, and the residue is subjected to codistillation with distilled water. The codistillation is repeated twice. The title compound is obtained as a glass-like solid.

Yield: 7.191 g (quantitative)

Water content: 8.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 27.63 H 2.32 F 39.10 Gd 9.52 N 5.93 S 3.88

Fnd: C 27.50 H 2.37 F 39.22 Gd 9.61 N 5.85 S 3.95

Example 24

Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-<2-aminoethyl-[glycolic acid-(1H,1H,2H,2H-perfluorodecyl-ether)-amide]>-amide}

a) 3,6,9-Tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-<2-aminoethyl-glycolic acid-(1H,1H,2H,2H-perfluorodecyl-ether)-amide]>-amide}

6.771 g (12 mmol) of the compound that is produced under Example 22a) as well as 1.821 g (18 mmol) of triethylamine are dissolved in 40 ml of dry dimethylformamide and mixed in portions with 2.144 g (6 mmol) of diethylenetriaminepentaacetic acid-bisanhydride while being stirred and with exclusion of moisture. It is allowed to stir overnight, then concentrated by evaporation, mixed with 20 ml of water, the pH is set at about 3, and it is extracted with 3N hydrochloric acid twice with 150 ml of butanol each. The organic solutions are combined, concentrated by evaporation, and the residue is subjected to a chromatography on silica gel RP-18. Water and tetrahydrofuran are used as eluant. The title compound is obtained as a glass-like solid.

Yield: 6.989 g (78.4% of theory)

Water content: 7.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 33.95 H 3.05 F 43.47 N 6.60

Fnd: C 34.06 H 3.11 F 43.40 N 6.67

- b) Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis{N-<2-aminoethyl-[glycolic acid-(1H,1H,2H,2H-perfluorodecyl-ether)-amide]>-amide}

4.798 g (3 mmol, relative to 7.1% water) of the compound that is produced under 24a) is added to a mixture of 100 ml of distilled water, 50 ml of ethanol and 20 ml of chloroform. 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. It is stirred until dissolved, then concentrated by evaporation, whereby strong foaming occurs. The residue is codistilled several times with distilled water. The title compound is obtained as a glass-like solid.

Yield: 5.285 g (quantitative)

Water content: 6.9%

The elementary analysis is calculated based on anhydrous substance.

Clcd: C 30.76 H 2.58 F 39.39 Gd 9.59 N 5.98

Fnd: C 30.87 H 2.65 F 39.51 Gd 9.69 N 6.11

Example 25

Gadolinium complex, sodium salt of 3,9-bis(carboxymethyl)-6-[N-(1H,1H,2H,2H-perfluorodecyl)-aminocarbonylmethyl]-3,6,9-triazaundecanedioic acid

- a) N-Benzylloxycarbonylglycine-N-(1H,1H,2H,2H-perfluorodecyl)-amide

7.877 g (15 mmol) of 1H,1H,2H,2H-perfluorodecylamine (J. Fluor. Chem. 55, 85 (1991)) is dissolved in 70 ml of

dichloromethane and mixed with 1.726 g (15 mmol) of N-hydroxysuccinimide, 3.095 g (15 mmol) of dicyclohexylcarbodiimide and 3.138 g (15 mmol) of N-benzyloxycarbonylglycine (commercially available products, Bachem). It is allowed to stir overnight, the dicyclohexylurea is filtered off, concentrated by evaporation, and the residue is subjected to column chromatography on silica gel. Mixtures of dichloromethane and ethanol are used as eluant. The title compound is obtained as a solid.

Yield: 8.951 g (91.2% of theory)

Elementary analysis:

Cld: C 36.71 H 2.31 F 49.36 N 4.28

Fnd: C 36.87 H 2.39 F 49.51 N 4.37

b) Glycine-N-(1H,1H,2H,2H-perfluorodecyl)-amide

7.594 g (10 mmol) of the compound that is produced under 28a) is dissolved in 150 ml of a mixture of tetrahydrofuran and ethanol at a 2:1 ratio and hydrogenated in the presence of 0.25 g of Pearlman's catalyst (Pd 20%/C) until 224 ml of hydrogen is taken up. Catalyst is suctioned out, rewashed well with ethanol and evaporated to the dry state. The title compound is obtained as an amorphous solid.

Yield: 6.21 g (99.3% of theory)

Elementary analysis:

Cld: C 25.37 H 1.60 F 56.84 N 4.93

Fnd: C 25.28 H 1.65 F 56.92 N 4.99

- c) 3,9-Bis(t-butyloxycarbonylmethyl)-6-N-[1H,1H,2H,2H-perfluorodecyl)-aminocarbonylmethyl-3,6,9-triazaundecanedioic acid-di(t-butylester)

2.841 g (5 mmol) of the amine that is produced under 25b) and 3.875 g (11 mmol) of N,N-bis(t-butyloxycarbonylmethyl)-2-(bromoethyl)-amine are added to a mixture of 10 ml of acetonitrile and 20 ml of phosphate buffer of pH 8.0, and it is stirred intensively at room temperature for 2 hours. Then, the buffer phase is separated, it is extracted with 10 ml of acetonitrile, and the latter is added to the organic phase. After 20 ml of fresh buffer is added, it is stirred for 20 more hours at room temperature. The organic phase is separated, concentrated by evaporation, and the residue is dispersed between 100 ml of phosphate buffer (pH 8.0) and 100 ml of ethyl acetate. The organic phase is washed with saturated common salt solution, dried on sodium sulfate and concentrated by evaporation. The title compound is purified by chromatography on silica gel. Dichloromethane with increasing addition of methanol is used as eluant. The title compound is obtained as a glass-like solid.

Yield: 4.161 g (78.3% of theory)

Elementary analysis:

Cld: C 45.20 H 5.59 F 30.39 N 5.27

Fnd: C 45.35 H 5.67 F 30.47 N 5.34

- d) 3,9-Bis(carboxymethyl)-6-N-(1H,1H,2H,2H-perfluorodecyl)-aminocarbonylmethyl-3,6,9-triazaundecanedioic acid

4.783 g (4.5 mmol) of the compound that is produced under 25c) is added to a mixture of 100 ml of trifluoroacetic acid/dichloromethane at a 2:1 ratio. It is allowed to stir overnight at room temperature, then evaporated to the dry state, the remainder of trifluoroacetic acid is removed by codistillation with ethanol and taken up in 160 ml of a mixture of water, ethanol and chloroform (10:5:1). A pH of about 3 (pH constant) is set by adding ion exchanger IRA-67 (OH⁻ form). It is suctioned out quickly, concentrated by evaporation, and the title compound is obtained as a vitreous solid.

Yield: 3.007 g (79.7% of theory)

Water content: 10.9%

Elementary analysis (relative to anhydrous substance):

Cld: C 34.38 H 3.25 F 38.52 N 6.68

Fnd: C 34.29 H 3.33 F 38.65 N 6.77

e) Gadolinium complex, monosodium salt of 3,9-bis(carboxymethyl)-6-N-(1H,1H,2H,2H-perfluorodecyl)-aminocarbonylmethyl)-3,6,9-triazaundecanedioic acid

2.823 g (3.0 mmol, relative to 10.9% water content) of the acid that is produced under Example 25d) is added to a mixture of 60 ml of distilled water and 30 ml of ethanol. 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. After addition is completed, it is stirred until dissolved. Then, the pH of the solution is set at 7.2 by adding sodium hydroxide solution. The solution is concentrated by evaporation. In this case, strong foaming occurs. The residue is codistilled twice with distilled water. The title compound is obtained as a vitreous solid.

Yield: 3.353 g (quant)

Water content: 9.2%

The elementary analysis is calculated based on anhydrous substance.

Cld: C 28.41 H 2.28 F 31.83 Gd 15.50 N 5.52 Na 2.27

Fnd: C 28.51 H 2.33 F 31.76 Gd 15.57 N 5.46 Na 2.35

Example 26

Gadolinium complex, disodium salt of 3,6,9-tris(carboxymethyl)-4-[N-1H,1H,2H,2H-perfluorodecyloxy)-benzyl]-3,6,9-triazaundecanedioic acid

a) 3,6,9-Tris-(t-butyloxycarbonylmethyl)-4-[4-(1H,1H,2H,2H-perfluorodecyloxy)-benzyl]-3,6,9-triazaundecanedioic acid-di(t-butylester)

6.131 g (5 mmol) of 3,6,9-tris(t-butyloxycarbonylmethyl)-4-(4-hydroxybenzyl)-3,6,9-triazaundecanedioic acid-di(t-butylester), produced according to PCT WO 88/07521, is added to 50 ml of dry dimethylformamide, and it is mixed in portions with 150 g (5 mmol) of sodium hydride (80% in oil) while being stirred and with exclusion of moisture. After dissolving is completed, it is mixed with 3.092 g (5 mmol) of the tosylate that is produced under Example 7a). It is stirred for 12 hours at 40°C. Then, it is poured into 500 ml of ice water, the product is taken up in dichloromethane, the organic solution is washed with water, dried on sodium sulfate and evaporated to the dry state. The residue is purified by chromatography on silica gel. A mixture of dichloromethane, isopropanol, hexane at a 20:1:5 ratio is used as eluant.

The title compound is obtained as an amorphous solid.

Yield: 5.015 g (81.8% of theory)

Elementary analysis:

Clcd: C 49.96 H 5.92 F 26.34 N 3.43

Fnd: C 50.11 H 6.00 F 26.43 N 3.38

b) 3,6,9-Tris(carboxymethyl)-4-[4-(1H,1H,2H,2H-perfluorodecyloxy)-benzyl]-3,6,9-triazaundecanedioic acid
3.678 g (3 mmol) of the compound that is produced under Example 26a) is dissolved in 100 ml of a mixture of trifluoroacetic acid and dichloromethane at a 2:1 ratio, and it is stirred overnight at room temperature. It is evaporated to the dry state, and the remainder of trifluoroacetic acid is removed by codistillation with ethanol. The residue is taken up in 160 ml of a mixture of water, ethanol and chloroform (10:5:1). By adding ion exchanger IR A-67 (OH⁻ form), a pH of about 3 (constant pH) is set. It is quickly suctioned out, concentrated by evaporation, and the title compound is obtained as a vitreous solid.

Yield: 2.357 g (83.1% of theory)

Water content: 11.3%

The elementary analysis is calculated based on anhydrous substance.

Clcd: C 39.38 H 3.41 F 34.16 N 4.44

Fnd: C 39.52 H 3.47 F 34.32 N 4.36

c) Gadolinium complex, disodium salt of 3,6,9-tris(carboxymethyl)-4-[N-(1H,1H,2H,2H-perfluorodecyloxy)-benzyl]-3,6,9-triaza-undecanedioic acid
3.145 g (3.0 mmol, relative to 11.3% water content) of the acid that is produced under Example 26b) is added to a mixture of 60 ml of distilled water and 30 ml of ethanol. 543.8 mg (1.5

mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. After addition is completed, it is stirred until dissolved. Then, the pH of the solution is set at 7.2 by adding sodium hydroxide solution, and it is concentrated by evaporation. In this case, strong foaming occurs. The residue is codistilled twice with distilled water. The title compound is obtained as a vitreous solid.

Yield: 3.804 g (quantitative)

Water content: 9.8%

Elementary analysis (relative to anhydrous substance):

Cld: C 32.55 H 2.38 F 28.24 Gd 13.75 N 3.67 Na 4.02

Fnd: C 32.44 H 2.43 F 28.30 Gd 13.66 N 3.71 Na 4.10

Example 27

Gadolinium complex of 10-[(perfluorooctylsulfonyl)-piperazin-1-yl-carbonylmethyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

a) 1-Perfluorooctylsulfonyl-piperazine

34.39 g (398.3 mmol) of piperazine, 50 g (99.6 mmol) of perfluorooctylsulfonyl fluoride and 10.12 g (100 mmol) of triethylamine are heated for 24 hours to 85°C. 500 ml of water is added, and it is extracted twice with 200 ml of dichloromethane each. The organic phase is dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/2-propanol = 25:1).

Yield: 17.55 g (31% of theory) of a colorless, amorphous solid

Elementary analysis:

Cld: C 25.36 H 1.60 F 56.84 N 4.93 S 5.64

Fnd: C 25.15 H 1.80 F 56.65 N 4.81 S 5.70

b) 1-(2-Bromoacetyl)-4-perfluorooctylsulfonyl-piperazine

17 g (29.9 mmol) of the title compound of Example 27a) and 5.1 g (50 mmol) of triethylamine are dissolved in 100 ml of dichloromethane. 9.1 g (44.9 mmol) of bromoacetyl bromide is added in drops at -10°C within 30 minutes, and it is stirred for 2 hours at 0°C. The solution is poured into 200 ml of 2N hydrochloric acid and stirred thoroughly. The organic phase is separated, dried on magnesium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/acetone = 20/1).

Yield: 18.55 g (90% of theory) of a slightly yellow-colored waxy solid

Elementary analysis:

Cld: C 24.40 H 1.46 F 46.86 N 4.06 S 4.65 Br 11.59

Fnd: C 24.22 H 1.60 F 46.75 N 3.97 S 4.48 Br 11.41

- c) 10-[(Perfluorooctyl-sulfonyl)-piperazin-1-yl-carbonylmethyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

4.63 g (13.36 mmol) of 1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (= D03A) and 18.5 g (133.6 mmol) of potassium carbonate are added to 17.78 g (20 mmol) of the title compound of Example 27b) in 180 ml of methanol. It is refluxed for 12 hours. The inorganic salts are filtered off, and the filtrate is evaporated to the dry state. The residue is taken up in 100 ml of water and set at pH 3 with 5N hydrochloric acid. It is extracted twice with 150 ml of n-butanol. The combined organic phases are evaporated to the dry state in a vacuum, and the residue is purified by RP-chromatography (RP-18/mobile solvent = gradient consisting of water/n-butanol/acetonitrile).

Yield: 12.79 g (67% of theory) of a colorless solid

Water content: 8.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 35.23 H 3.70 F 33.83 N 8.80 S 3.36

Fnd: C 35.17 H 3.81 F 33.67 N 8.65 S 3.18

- d) Gadolinium complex of 10-[(perfluorooctyl-sulfonyl)-piperazin-1-yl-carbonylmethyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (10.47 mmol) of the title compound of Example 27c) is dissolved in a mixture of 50 ml of water/20 ml of ethanol, and 1.90 g (5.23 mmol) of gadolinium oxide is added. It is stirred

for 4 hours at 80°C. The solution is filtered and evaporated to the dry state in a vacuum.

Yield: 12.2 g (quantitative)

Water content: 5.1%

Elementary analysis (relative to anhydrous substance):

Clc: C 30.33 H 2.91 F 29.13 Gd 14.18 S 2.89

Fnd: C 30.39 H 2.81 F 29.02 Gd 14.01 S 2.78

Example 28

Gadolinium complex, monosodium salt of 3,9-bis(carboxymethyl)-6-[(4-perfluorooctylsulfonyl)-piperazine-1-carbonylmethyl]-3,6,9-triazaundecanedioic acid

a) 1-(2-Benzyloxycarbonylamino)-methyl-carbonyl-4-(perfluorooctylsulfonyl)-piperazine

8.524 g (15 mmol) of the piperazine derivative that is produced under 27a) is dissolved in 80 ml of dichloromethane and mixed with 1.726 g (15 mmol) of N-hydroxysuccinimide, 3.095 g (15 mmol) of dicyclohexylcarbodiimide and 3.138 g (15 mmol) of N-benzyloxycarbonylglycine (commercially available products, Bachem). It is allowed to stir overnight, the dicyclohexylurea is filtered off, concentrated by evaporation, and the residue is subjected to column chromatography on silica gel. Mixtures of dichloromethane and ethanol are used as eluant. The title compound is obtained as a solid.

Yield: 10.16 g (89.2% of theory)

Elementary analysis:

Cld: C 34.79 H 2.39 F 42.53 N 5.53 S 4.22

Fnd: C 34.60 H 2.43 F 42.65 N 5.66 S 4.17

b) 1-(2-Amino)-acetyl-4-(perfluorooctyl)-sulfonyl-piperazine
7.594 g (10 mmol) of the compound that is produced under 28a) is dissolved in 150 ml of a mixture of tetrahydrofuran and ethanol at a 2:1 ratio, and it is hydrogenated in the presence of 0.25 g of Pearlman's catalyst (Pd 20%/C) until 224 ml of hydrogen is taken up. Catalyst is suctioned out, rewashed well with ethanol and evaporated to the dry state. The title compound is obtained as an amorphous solid.

Yield: 6.21 g (99.3% of theory)

Elementary analysis:

Cld: C 26.89 H 1.93 F 51.65 N 6.72 S 5.13

Fnd: C 27.03 H 1.97 F 51.77 N 6.58 S 5.20

c) 3,9-Bis(t-butyloxycarbonylmethyl)-6-[(4-perfluorooctylsulfonyl)-piperazine-1-carbonylmethyl]-3,6,9-triazaundecanedicarboxylic acid-di(t-butylester)
3.127 g (5 mmol) of the amine that is produced under 28b) and 3.875 g (11 mmol) of N,N-bis(t-butyloxycarbonylmethyl)-2-(bromoethyl)-amine are added to a mixture of 10 ml of acetonitrile and 20 ml of phosphate buffer of pH 8.0, and it is stirred intensively at room temperature for 2 hours. Then, the buffer is separated, extracted with 10 ml of acetonitrile, and

the latter is added to the organic phase. After 20 ml of fresh buffer is added, it is stirred for 20 more hours at room temperature. The organic phase is separated, concentrated by evaporation, and the residue is dispersed between 100 ml of phosphate buffer (pH 8.0) and 100 ml of ethyl acetate. The organic phase is washed with saturated common salt solution, dried on sodium sulfate and concentrated by evaporation. The title compound is purified by chromatography on silica gel. Dichloromethane with increasing addition of methanol is used as eluant. The title compound is obtained as a glass-like solid.

Yield: 4.481 g (76.3% of theory)

Elementary analysis:

Clcd: C 43.71 H 5.42 F 27.99 N 4.85 S 2.78

Fnd: C 43.84 H 5.47 F 28.10 N 5.00 S 2.69

- d) 3,9-Bis(carboxymethyl)-6-[(4-perfluorooctyl-sulfonyl)-piperazin-1-yl-carbonylmethyl]-3,6,9-triazaundecanedioic acid

5.193 g (4.5 mmol) of the compound that is produced under 28c) is added to a mixture of 100 ml of trifluoroacetic acid/dichloromethane at a 2:1 ratio. It is allowed to stir overnight at room temperature, then evaporated to the dry state, the remainder of the trifluoroacetic acid is removed by codistillation with ethanol and taken up in 160 ml of a mixture of water, ethanol and chloroform (10:5:1). A pH of about 3 (constant pH) is set by adding ion exchanger IRA-67 (OH⁻ form).

It is quickly suctioned out, concentrated by evaporation, and the title compound is obtained as a vitreous solid.

Yield: 3.718 g (79.2% of theory)

Water content: 10.9%

Elementary analysis (relative to anhydrous substance):

Clc: C 33.59 H 3.25 F 34.74 N 6.03 S 3.45

Fnd: C 33.69 H 3.36 F 34.82 N 6.10 S 3.38

- e) Gadolinium complex, monosodium salt of 3,9-bis(carboxymethyl)-6-[(4-perfluorooctylsulfonyl)-piperazine-1-carbonylmethyl]-3,6,9-triazaundecanedioic acid

3.13 g (3.0 mmol, relative to 10.9% water content) of the acid that is produced under Example 28d) is added to a mixture of 60 ml of distilled water and 30 ml of ethanol. 543.8 mg (1.5 mmol) of gadolinium oxide is added in portions while being stirred and heated to 50°C. After addition is completed, it is stirred until dissolved. Then, the pH of the solution is set at 7.2 by adding sodium hydroxide solution, and it is concentrated by evaporation. In this case, strong foaming occurs. The residue is codistilled twice with distilled water. The title compound is obtained as a vitreous solid.

Yield: 3.678 g (quantitative)

Water content: 9.2%

Elementary analysis (relative to anhydrous substance):

Cld: C 28.24 H 2.37 F 29.21 Gd 14.22 N 5.07 Na 2.08 S 2.90

Fnd: C 28.36 H 2.41 F 29.14 Gd 14.30 N 5.15 Na 2.12 S 2.83

Example 29

Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis[(4-perfluorooctylsulfonyl)-piperazine]-amide

a) 3,6,9-Tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis[(4-perfluorooctylsulfonyl)-piperazine]-amide

5.683 g (10 mmol) of the compound produced under 27a) as well as 1.518 g (15 mmol) of triethylamine are dissolved in 30 ml of dry dimethylformamide and mixed in portions with 1.787 g (5 mmol) of diethylenetriaminepentaacetic acid-bisanhydride while being stirred and with exclusion of moisture. It is allowed to stir overnight, then concentrated by evaporation, mixed with water, the pH is set at about 3 with 3N hydrochloric acid, and it is extracted twice with 100 ml of n-butanol each. The organic solutions are combined, concentrated by evaporation and subjected to a chromatography on silica gel RP-18. Water and tetrahydrofuran are used as eluant. The title compound is obtained as a glass-like solid.

Yield: 6.741 g (81.4% of theory)

Water content: 9.8%

Elementary analysis (relative to anhydrous substance):

Cld: C 30.55 H 2.50 F 43.24 N 6.56 S 4.29

Fnd: C 30.67 H 2.55 F 43.33 N 6.49 S 4.21

- b) Gadolinium complex of 3,6,9-tris(carboxymethyl)-3,6,9-triazaundecanedioic acid-bis[(4-perfluorooctylsulfonyl)-piperazine]-amide

6.570 g (4 mmol, relative to 9.8% water content) of the compound that is produced under 23c) is added to a mixture of 120 ml of distilled water, 60 ml of ethanol and 20 ml of chloroform. While being stirred and heated to 50°C, 725 mg (82.0 mmol) of gadolinium oxide is added in portions. It is stirred until dissolved, then concentrated by evaporation, whereby strong foaming occurs, and the residue is subjected to codistillation with distilled water. Codistillation is repeated twice. The title compound is obtained as a glass-like solid.

Yield: 7.191 g (quantitative)

Water content: 8.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 27.69 H 2.08 F 39.19 Gd 9.54 N 5.95 S 3.89

Fnd: C 27.83 H 2.15 F 39.10 Gd 6.91 N 6.03 S 3.88

Example 30

a) 11-[N-Ethyl-N-(perfluorooctylsulfonyl)-amino]undecanoic acid
benzyl ester

20 g (37.94 mmol) of N-ethyl-N-perfluorooctylsulfonamide and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 26.96 g (75.87 mmol) of 11-bromoundecanoic acid benzyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. Salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 26.46 g (87% of theory) of a colorless, crystalline powder.

Elementary analysis:

Cld: C 41.95 H 4.02 N 1.75 F 40.29 S 4.00

Fnd: C 41.78 H 4.17 N 1.68 F 40.12 S 3.88

b) 11-[N-Ethyl-N-(perfluorooctylsulfonyl)-aminoundecanoic acid

20 g (24.95 mmol) of the title compound of Example 30a) is dissolved in 300 ml of isopropanol/200 ml of dichloromethane, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. Catalyst is filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is recrystallized from ether/hexane.

Yield: 16.69 g (94% of theory) of a colorless, crystalline solid.

Elementary analysis:

Cld: C 35.45 H 3.68 N 1.97 F 45.39 S 4.51

Fnd: C 35.31 H 3.81 N 1.85 F 45.25 S 4.42

c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-16-aza-16-(perfluorooctylsulfonyl-octadecyl)]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane
12.16 g (17.09 mmol) of the title compound of Example 30b) and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. At 0°C, 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added and stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/N-propanol/acetonitrile).

Yield: 16.82 g (71% of theory) of a colorless, vitreous solid.

Water content: 8.6%

Elementary analysis (relative to anhydrous substance):

Cld: C 36.02 H 4.30 F 25.49 Gd 12.41 N 6.63 S 2.53

Fnd: C 35.87 H 4.45 F 25.28 Gd 12.29 N 6.50 S 2.41

d) 10-[2-Hydroxy-4-aza-5-oxo-16-aza-16-(perfluorooctylsulfonyl-octadecyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

11.1 g (8.76 mmol) of the title compound of Example 30c) is dissolved in a mixture of 100 ml of water/100 ml of ethanol, and 1.73 g (13.71 mmol) of oxalic acid-dihydrate is added. It is heated for 8 hours to 80°C. It is cooled to 0°C, and precipitated gadolinium oxalate is filtered out. The filtrate is evaporated to the dry state, and the residue is purified on RP-18 (RP-18/ mobile solvent: gradient consisting of water/i-propanol/acetonitrile).

Yield: 9.80 g (92% of theory) of a vitreous solid.

Water content: 8.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 41.01 H 5.16 F 29.02 N 7.55 S 2.88

Fnd: C 40.87 H 5.31 F 28.85 N 7.40 S 2.73

- e) Ytterbium complex of 10-[2-hydroxy-4-aza-5-oxo-16-aza-16-(perfluorooctylsulfonyl-octadecyl)]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

1.33 g (2.53 mmol) of ytterbium carbonate is added to 5.64 g (5.07 mmol) of the title compound of Example 30d) in 100 ml of water/50 ml of ethanol, and it is stirred for 3 hours at 80°C.

The solution is filtered, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 7.08 g (quantitative) of a vitreous solid.

Water content: 8.1%

Elementary analysis (relative to anhydrous substance):

Clcd: C 35.58 H 4.24 F 25.17 N 6.55 S 2.50 Yb 13.49

Fnd: C 35.43 H 4.37 F 25.05 N 6.48 S 2.39 Yb 13.35

- f) Dysprosium complex of 10-[2-hydroxy-4-aza-5-oxo-16-aza-16-(perfluorooctylsulfonyl-octadecyl)]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

0.95 g (2.53 mmol) of dysprosium oxide is added to 5.64 g (5.07 mmol) of the title compound of Example 30d) in 100 ml of water/50 ml of ethanol, and it is stirred for 3 hours at 80°C.

The solution is filtered, and the filtrate is evaporated to the dry state in a vacuum.

Yield: 7.10 g (quantitative) of a colorless, vitreous solid.

Water content: 9.1%

Elementary analysis (relative to anhydrous substance):

Cld: C 35.87 H 4.28 F 25.38 N 6.60 S 2.52 Dy 12.77

Fnd: C 35.69 H 4.39 F 25.18 N 6.49 S 2.43 Dy 12.70

Example 31

a) 11,11,11,10,10,9,9,8,8,7,7-Tridecafluoro-3-oxaundecanoic acid-tert-butyl ester

19.51 g (100.0 mmol) of bromoacetic acid-tert-butyl ester is added in drops to a mixture of 27.57 g (75.73 mmol) of 1H,1H,2H,2H-perfluorooctan-1-ol and 2.57 g (7.57 mmol) of tetrabutylammonium hydrogen sulfate in 300 ml of 60% aqueous potassium hydroxide solution/200 ml of toluene while being stirred vigorously at 0°C. It is stirred for one hour at 0°C, the organic phase is separated, and the aqueous phase is extracted twice with 50 ml of toluene. The combined organic extracts are dried on sodium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane).

Yield: 28.97 g (80% of theory) of a colorless oil.

Elementary analysis:

Cld: C 35.16 H 3.16 F 51.64

Fnd: C 35.08 H 3.20 F 51.70

b) 11,11,11,10,10,9,9,8,8,7,7-Tridecafluoro-3-oxaundecanoic acid

25.29 g (52.88 mmol) of the title compound of Example 1a) is dissolved in 300 ml of trifluoroacetic acid, and it is stirred overnight at room temperature. It is evaporated to the dry state in a vacuum, and the residue is recrystallized from hexane/diethyl ether.

Yield: 20.54 g (92% of theory) of a colorless, crystalline solid.

Elementary analysis:

Clc: C 28.45 H 1.67 F 58.51

Fnd: C 28.36 H 1.60 F 58.62

c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-oxa-10,10,11,11,12,12,13,13,14,14,15,15,15-tridecafluoropentadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

7.21 g (17.09 mmol) of the title compound of Example 31b and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-

hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 12.68 g (71% of theory) of a colorless, vitreous solid.

Water content: 6.4%

Elementary analysis (relative to anhydrous substance):

Cld: C 33.16 H 3.61 F 25.26 Gd 16.08 N 7.16

Fnd: C 32.85 H 3.84 F 25.01 Gd 15.87 N 7.03

Example 32

a) 15,15,15,14,14,13,13,12,12,11,11,10,10,9,9,8,8-7,7-

Henicosafuoro-3-oxapenta-decanoic acid-tert-butyl ester

19.51 g (100.0 mmol) of bromoacetic acid-tert-butyl ester is added in drops to a mixture of 42.72 g (75.73 mmol) of 1H,1H,2H,2H-perfluorooctan-1-ol and 2.57 g (7.57 mmol) of tetrabutylammonium hydrogen sulfate in 300 ml of 60% aqueous potassium hydroxide solution/200 ml of toluene while being stirred vigorously at 0°C. It is stirred for one hour at 0°C, the organic phase is separated, and the aqueous phase is extracted twice with 50 ml of toluene. The combined organic

extracts are dried on sodium sulfate and concentrated by evaporation in a vacuum. The residue is chromatographed on silica gel (mobile solvent: dichloromethane).

Yield: 42.12 g (82% of theory) of a colorless oil.

Elementary analysis:

Cld: C 31.87 H 2.23 F 58.82

Fnd: C 31.73 H 2.20 F 58.90

b) 15,15,15,14,14,13,13,12,12,11,11,10,10,9,9,8,8,7,7-

Henicosafuoro-3-oxapentadecanoic acid-tert-butyl ester

35.87 g (52.88 mmol) of the title compound of Example 1a) is dissolved in 300 ml of trifluoroacetic acid, and it is stirred overnight at room temperature. It is evaporated to the dry state in a vacuum, and the residue is recrystallized from hexane/diethyl ether.

Yield: 30.60 g (93% of theory) of a colorless, crystalline solid.

Elementary analysis:

Cld: C 27.03 H 1.13 F 64.12

Fnd: C 26.91 H 1.20 F 64.02

c) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-oxa-10,10,11,11,12,12,13,13,14,14,15,15,16,16,17,17,18,18,19,19,19-henicosafuoro-nonadecyl]-1,4,7-tris(carboxymethyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane 10.63 g (17.09 mmol) of the title compound of Example 32b and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 14.73 g (69% of theory) of a colorless, vitreous solid.

Water content: 5.7%

Elementary analysis (relative to anhydrous substance):

Cld: C 31.61 H 2.99 F 33.87 Gd 13.35 N 5.95

Fnd: C 31.49 H 3.15 F 33.68 Gd 13.21 N 6.01

Example 33

a) N-(2-Bromopropionyl)glycine-benzyl ester

55.9 g (326.1 mmol) of 2-bromopropionic acid chloride is added in drops to 100 g (296.4 mmol) of glycine benzyl ester-p-toluenesulfonic acid salt and 33.0 g (326.1 mmol) of triethylamine in 400 ml of methylene chloride at 0°C. The temperature is not allowed to exceed 5°C. After the addition is completed, it is stirred for one hour at 0°C, then for 2 hours at room temperature. 500 ml of ice water is added, and the water phase is set at pH 2 with 10% aqueous hydrochloric acid. The organic phase is separated and washed once each with 300 ml of 5% aqueous soda solution and 400 ml of water. The organic phase is dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is recrystallized from diisopropyl ether.

Yield: 68.51 g (75% of theory) of a colorless, crystalline powder

Melting point: 69-70°C

Elementary analysis:

Cld: C 48.02 H 4.70 N 4.67 Br 26.62

Fnd: C 47.91 H 4.82 N 4.51 Br 26.47

b) 1-[4-(Benzyloxycarbonyl)-1-methyl-2-oxo-3-azabutyl]-
1,4,7,10-tetraazacyclododecane

50 g (162.2 mmol) of the title compound of Example 1a) is added to 55.8 g (324.4 mmol) of 1,4,7,10-tetraazacyclododecane, dissolved in 600 ml of chloroform, and it is stirred overnight at room temperature. 500 ml of water is added, the organic phase is separated, and it is washed twice with 400 ml of water in each case. The organic phase is dried on magnesium sulfate and evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: chloroform/methanol/aqueous 25% ammonia = 10/5/1).

Yield: 40.0 g [63% of theory relative to 1a) used] of a light yellowish viscous oil.

Elementary analysis:

Cld: C 61.36 H 8.50 N 17.89

Fnd: C 61.54 H 8.68 N 17.68

c) 10-[4-(Benzyloxycarbonyl)-1-methyl-2-oxo-3-azabutyl]-1,4,7-
tris(tert-butoxycarbonylmethyl)-1,4,7,10-
tetraazacyclododecane (sodium bromide complex)

33 g (169 mmol) of bromoacetic acid-tert-butyl ester is added to 20 g (51.08 mmol) of the title compound of Example 1b) and 17.91 g (169 mmol) of sodium carbonate in 300 ml of acetonitrile, and it is stirred for 24 hours at 60°C. It is cooled to 0°C, salts are filtered out, and the filtrate is evaporated to the dry state. The residue is chromatographed on

silica gel (mobile solvent: ethyl acetate/ethanol: 15/1). The fractions that contain the product are concentrated by evaporation, and the residue is recrystallized from diisopropyl ether.

Yield: 34.62 g (81% of theory) of a colorless, crystalline powder

Melting point: 116-117°C

Elementary analysis:

Cld: C 54.54 H 7.59 N 8.37 Na 2.74 Br 9.56

Fnd: C 54.70 H 7.65 N 8.24 Na 2.60 Br 9.37

- d) 10-(4-Carboxy-1-methyl-2-oxo-3-azabutyl)-1,4,7-tris(tert-butoxycarbonylmethyl)-1,4,7,10-tetraazacyclododecane (sodium bromide complex)

30 g (35.85 mmol) of the title compound of Example 1c is dissolved in 500 ml of isopropanol, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. Catalyst is filtered out, the filtrate is evaporated to the dry state in a vacuum and recrystallized from acetone.

Yield: 22.75 g (85% of theory) of a colorless, crystalline powder

Melting point: 225°C (decomposition)

Elementary analysis:

Cld: C 49.86 H 7.69 N 9.38 Na 3.07 Br 10.71

Fnd: C 49.75 H 7.81 N 9.25 Na 2.94 Br 10.58

- e) 10-[1-Methyl-2-oxo-3-aza-5-oxo-5-{4-perfluorooctylsulfonyl-piperazin-1-yl}-pentyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10 g (13.39 mmol) of the title compound of Example 33d and 7.61 g (13.39 mmol) of the title compound of Example 27a are dissolved in 150 ml of tetrahydrofuran. 3.97 g (16.07 mmol) of N-ethoxycarbonyl-2-ethoxy-1,2-dihydroquinoline (EEDQ) is added at 0°C, stirred for 3 hours at 0°C, then for 12 hours at room temperature. It is evaporated to the dry state in a vacuum. The residue is taken up in 150 ml of trifluoroacetic acid and stirred for 12 hours at room temperature. It is evaporated to the dry state, the residue is dissolved in water and set at pH 3.2 with 10% aqueous sodium hydroxide solution. For purification, it is chromatographed on RP-18 (gradient consisting of water/acetonitrile/tetrahydrofuran).

Yield: 9.67 g (63% of theory) of a hygroscopic solid.

Water content: 10.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 36.30 H 3.93 N 9.56 F 31.49 S 3.13

Fnd: C 36.14 H 3.98 N 9.40 F 31.67 S 3.02

- f) Gadolinium complex of 10-[1-methyl-2-oxo-3-aza-5-oxo-5-{4-perfluorooctylsulfonyl-piperazin-1-yl}-pentyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

5 g (4.87 mmol) of the title compound of Example 33e is dissolved in 60 ml of water, and 0.883 g (2.44 mmol) of gadolinium oxide is added. It is stirred for 3 hours at 90°C. The solution is filtered, and the filtrate is freeze-dried.

Yield: 6.47 g (quantitative) of a voluminous, amorphous powder

Water content: 11.3%

Elementary analysis (relative to anhydrous substance):

Clcd: C 31.56 H 3.16 N 8.31 F 27.37 S 2.72 Gd 13.33

Fnd: C 31.37 H 3.35 N 8.18 F 27.19 S 2.92 Gd 13.05

Example 34

- a) 4-Perfluorooctanesulfonylpiperazin-1-ylpentanediamic acid

A solution of 10.62 g (105.0 mmol) of triethylamine and 59.67 g (105.0 mmol) of the title compound of Example 27a) in 50 ml of tetrahydrofuran are added in drops to a suspension of 11.41 g (100.0 mmol) of glutaric anhydride in 100 ml of tetrahydrofuran while being stirred vigorously at 0°C, and it is allowed to come to room temperature overnight. The reaction mixture is acidified with 100 ml of 2N HCl and extracted three times with 100 ml of tetrahydrofuran. The combined organic extracts are dried with

sodium sulfate, filtered and concentrated by evaporation. The residue is recrystallized from 2-propanol/ethyl acetate.

Yield: 52.30 g (73% of theory) of a colorless, crystalline solid.

Elementary analysis:

Cld: C 29.92 H 2.22 N 4.11 F 47.33 S 4.70

Fnd: C 29.90 H 2.18 N 4.07 F 47.42 S 4.79

- b) Gadolinium complex of 10-[2-hydroxy-4-aza-5,9-dioxo-9-{4-perfluorooctyl)-piperazin-1-yl}-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7-tetraazacyclododecane

11.66 g (17.09 mmol) of the title compound of Example 34a and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and

dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18/mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 16.7 g (73% of theory) of a colorless, vitreous solid.

Water content: 7.5%

Elementary analysis (relative to anhydrous substance):

Cld: C 32.99 H 3.50 F 26.09 Gd 12.70 N 7.92 S 2.59

Fnd: C 32.75 H 3.68 F 25.88 Gd 12.55 N 7.84 S 2.63

Example 35

a) N-Benzylperfluorooctanesulfonamide

50.21 g (100.0 mmol) of perfluorooctanesulfonyl fluoride is added in drops to a mixture of 10.62 g (105.0 mmol) of triethylamine and 10.72 g (100.0 mmol) of benzylamine at 80°C while being stirred vigorously. It is stirred for 2 days at 80°C, the reaction mixture is mixed with 300 ml of water and extracted 3 times with ethyl acetate. The combined organic extracts are dried on sodium sulfate, filtered and concentrated by evaporation. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/methanol = 4/1).

Yield: 45.96 g (78% of theory) of a colorless liquid

Elementary analysis:

Cld: C 30.57, H 1.37, N 2.38, S 5.44, F 54.81

Fnd: C 30.49 H 1.30, N 2.42, S 5.50, F 54.90

b) N-Benzyl-N-(perfluorooctylsulfonyl)-aminoacetic acid-t-butyl ester

22.4 g (37.94 mmol) of the title compound of Example 35a and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 14.80 g (75.87 mmol) of bromoacetic acid-tert-butyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. Salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 24.02 g (90% of theory) of a waxy, colorless solid

Elementary analysis:

Cld: C 35.86, H 2.58, N 1.99, S 4.56, F 45.91

Fnd: C 35.67 H 2.71, N 2.13, S 4.45, F 45.83

c) N-Benzyl-N-(perfluorooctylsulfonyl)-aminoacetic acid

20 g (28.43 mmol) of the title compound of Example 35b is dissolved in 200 ml of trifluoroacetic acid, and it is stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from methanol/ether.

Yield: 17.48 g (95% of theory) of a colorless, crystalline solid

Elementary analysis:

Cld: C 31.54, H 1.56, N 2.16, S 4.95, F 49.89

Fnd: C 31.38 H 1.70, N 2.05, S 4.87, F 49.71

- d) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-8-phenyl-octyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

11.06 g (17.09 mmol) of the title compound of Example 35c and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added and stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18 mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 16.49 g (75% of theory) of a colorless, vitreous solid

Water content: 6.5%

Elementary analysis:

Cld: C 33.95, H 3.18, N 6.99, S 2.67, F 26.85, Gd 13.07

Fnd: C 33.81 H 3.24, N 6.82, S 2.54, F 26.64 Gd 12.91

Example 36

a) N-Decylperfluorooctanesulfonamide

50.21 g (100.0 mmol) of perfluorooctanesulfonyl fluoride is added in drops to a mixture of 10.62 g (105.0 mmol) of triethylamine and 15.73 g (100.0 mmol) of decylamine at 80°C while being stirred vigorously. It is stirred for 2 days at 80°C, the reaction mixture is mixed with 300 ml of water and extracted three times with ethyl acetate. The combined organic extracts are dried on sodium sulfate, filtered and concentrated by evaporation. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/methanol = 4/1).

Yield: 43.48 g (68% of theory) of a colorless, viscous liquid

Elementary analysis:

Cld: C 33.81, H 3.47, N 2.19, S 5.02, F 50.51

Fnd: C 33.71 H 3.39, N 2.15, S 4.93, F 50.31

b) N-Decyl-N-(perfluorooctylsulfonyl)-aminoacetic acid-t-butyl ester

24.26 g (37.94 mmol) of the title compound of Example 36a and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 14.80 g (75.87 mmol) of bromoacetic acid-tert-butyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. Salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 24.87 g (87% of theory) of a waxy, colorless solid

Elementary analysis:

Cld: C 38.25, H 4.28, N 1.86, S 4.26, F 42.86

Fnd: C 38.09 H 4.41, N 1.74, S 4.10, F 42.67

c) N-Decyl-N-(perfluorooctylsulfonyl)-aminoacetic acid

20 g (26.54 mmol) of the title compound of Example 36b is dissolved in 200 ml of trifluoroacetic acid and stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from methanol/ether.

Yield: 17.22 g (93% of theory) of a colorless, crystalline solid

Elementary analysis:

Cld: C 34.44, H 3.47, N 2.01, S 4.60, F 46.31

Fnd: C 34.28 H 3.30, N 1.95, S 4.65, F 46.28

d) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-heptadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane
11.92 g (17.09 mmol) of the title compound of Example 36c and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18 mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 16.76 g (71% of theory) of a colorless, vitreous solid

Water content: 6.5%

Elementary analysis:

Cld: C 35.46, H 4.18, N 6.71, S 2.56, F 25.77 Gd 12.55

Fnd: C 35.28 H 4.33, N 6.80 S 2.61, F 25.65 Gd 12.41

Example 37**a) N-Hexylperfluorooctanesulfonamide**

50.21 g (100.0 mmol) of perfluorooctanesulfonyl fluoride is added in drops to a mixture of 10.62 g (105.0 mmol) of triethylamine and 10.12 g (100.0 mmol) of benzylamine at 80°C while being stirred vigorously. It is stirred for 2 days at 80°C, the reaction mixture is mixed with 300 ml of water and extracted three times with ethyl acetate. The combined organic extracts are dried on sodium sulfate, filtered and concentrated by evaporation. The residue is chromatographed on silica gel (mobile solvent: dichloromethane/methanol = 4/1).

Yield: 45.50 g (78% of theory) of a colorless liquid

Elementary analysis:

Cld: C 28.83, H 2.42, N 2.40, S 5.50, F 55.37

Fnd: C 28.29 H 2.39, N 2.44, S 5.55, F 55.50

b) N-Hexyl-N-(perfluorooctylsulfonyl)-aminoacetic acid-t-butyl ester

22.13 g (37.94 mmol) of the title compound of Example 37a and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 14.80 g (75.87 mmol) of bromoacetic acid-

tert-butyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. Salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 23.02 g (87% of theory) of a waxy, colorless solid

Elementary analysis:

Cld: C 34.44, H 3.47, N 2.01, S 4.60, F 46.31

Fnd: C 34.31 H 3.61, N 1.97, S 4.65, F 46.25

c) N-Hexyl-N-(perfluorooctylsulfonyl)-aminoacetic acid

20 g (28.43 mmol) of the title compound of Example 37b is dissolved in 200 ml of trifluoroacetic acid, and it is stirred overnight at room temperature. It is evaporated to the dry state in a vacuum. The residue is recrystallized from methanol/ether.

Yield: 16.74 g (91% of theory) of a colorless, crystalline solid

Elementary analysis:

Cld: C 29.96, H 2.51, N 2.18, S 5.00, F 50.36

Fnd: C 29.87 H 2.70, N 2.05, S 4.84, F 50.17

d) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-tridecyl]-1,4,7-

tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10.96 g (17.09 mmol) of the title compound of Example 37c and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxy-propyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18 mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 16.46 g (75% of theory) of a colorless, vitreous solid

Water content: 6.8%

Elementary analysis:

Cld: C 33.11, H 3.70, N 7.02, S 2.68, F 26.98 Gd 13.14

Fnd: C 33.01 H 3.84, N 6.95, S 2.57, F 26.85 Gd 13.03

Example 38**a) 11-[N-Ethyl-N-(perfluorooctylsulfonyl)-amino]-hexanoic acid benzyl ester**

20 g (37.94 mmol) of N-ethyl-N-perfluorooctylsulfonylamide and 15.73 g (113.8 mmol) of potassium carbonate are suspended in 200 ml of acetone, and 21.64 g (75.87 mmol) of 6-bromohexanoic acid benzyl ester is added in drops at 60°C. It is stirred for 3 hours at 60°C. Salts are filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is chromatographed on silica gel (mobile solvent: hexane/dichloromethane/acetone = 10/10/1). After the product-containing fractions are concentrated by evaporation, the residue is recrystallized from methanol/ether.

Yield: 25.26 g (91% of theory) of a colorless, crystalline powder

Elementary analysis:

Cld: C 37.77, H 3.03, N 1.91, S 4.38, F 44.15

Fnd: C 37.61 H 3.18, N 1.84, S 4.27, F 44.01

b) 11-[N-Ethyl-N-(perfluorooctylsulfonyl)-amino]-hexanoic acid

20 g (27.34 mmol) of the title compound of Example 38b is dissolved in 300 ml of isopropanol/200 ml of dichloromethane, and 3 g of palladium catalyst (10% Pd/C) is added. It is hydrogenated overnight at room temperature. Catalyst is filtered out, and the filtrate is evaporated to the dry state in a vacuum. The residue is recrystallized from ether/hexane.

Yield: 16.13 g (92% of theory) of a colorless, crystalline solid

Elementary analysis:

Cld: C 29.96, H 2.51, N 2.18, S 5.00, F 50.36

Fnd: C 29.81 H 2.70, N 2.09, S 4.93, F 50.14

d) Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-11-aza-11-(perfluorooctylsulfonyl)-tridecyl]-1,4,7-

tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane

10.96 g (17.09 mmol) of the title compound of Example 38b and 1.97 g (18.79 mmol) of N-hydroxysuccinimide are dissolved in a mixture of 50 ml of dimethylformamide/50 ml of chloroform. 3.88 g (18.79 mmol) of dicyclohexylcarbodiimide is added at 0°C, and it is stirred for 1 hour at 0°C, then for 3 hours at room temperature. It is cooled again to 0°C, and 5.19 g (51.27 mmol) of triethylamine/50 ml of 2-propanol is added. Then, 10.78 g (18.79 mmol) of the gadolinium complex of 10-(3-amino-2-hydroxypropyl)-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane (WO 95/17451), dissolved in 50 ml of water, is added, and it is stirred for 3 hours at room temperature. It is evaporated to the dry state, the residue is taken up in a mixture of 200 ml of methanol/100 ml of chloroform, and dicyclohexylurea is filtered out. The filtrate is evaporated to the dry state and purified by RP-chromatography (RP-18 mobile solvent: gradient consisting of water/n-propanol/acetonitrile).

Yield: 15.0 g (69% of theory) of a colorless, vitreous solid

Water content: 5.9%

Elementary analysis:

Cld: C 33.11, H 3.70, N 7.02, S 2.68, F 26.98 Gd 13.14

Fnd: C 33.01 H 3.83, N 6.91, S 2.49, F 26.83 Gd 13.05

Example 39

Blood Elimination Kinetics of Contrast Media

The blood elimination kinetics of contrast media was examined in rats (Han Wistar, Schering SPF, $\approx$ 250 g of body weight). For this purpose, after one-time intravenous administration (via a caudal vein) of the substances (dose: 50-100 μ mol of Me per kg of body weight), the substance concentration in the blood (based on the Gd or Dy content) was determined over a period of up to 300 minutes p.i. with the aid of ICP-AES. The pharmacokinetic parameters: distribution volume (V_{ss}), total clearance (CL_{tot}) and elimination half-life (t_{β}) were calculated with a special computer program (TOPFIT 2.0; Thomae, Schering, Gödecke), whereby a one- or two-compartment distribution model was used as a basis.

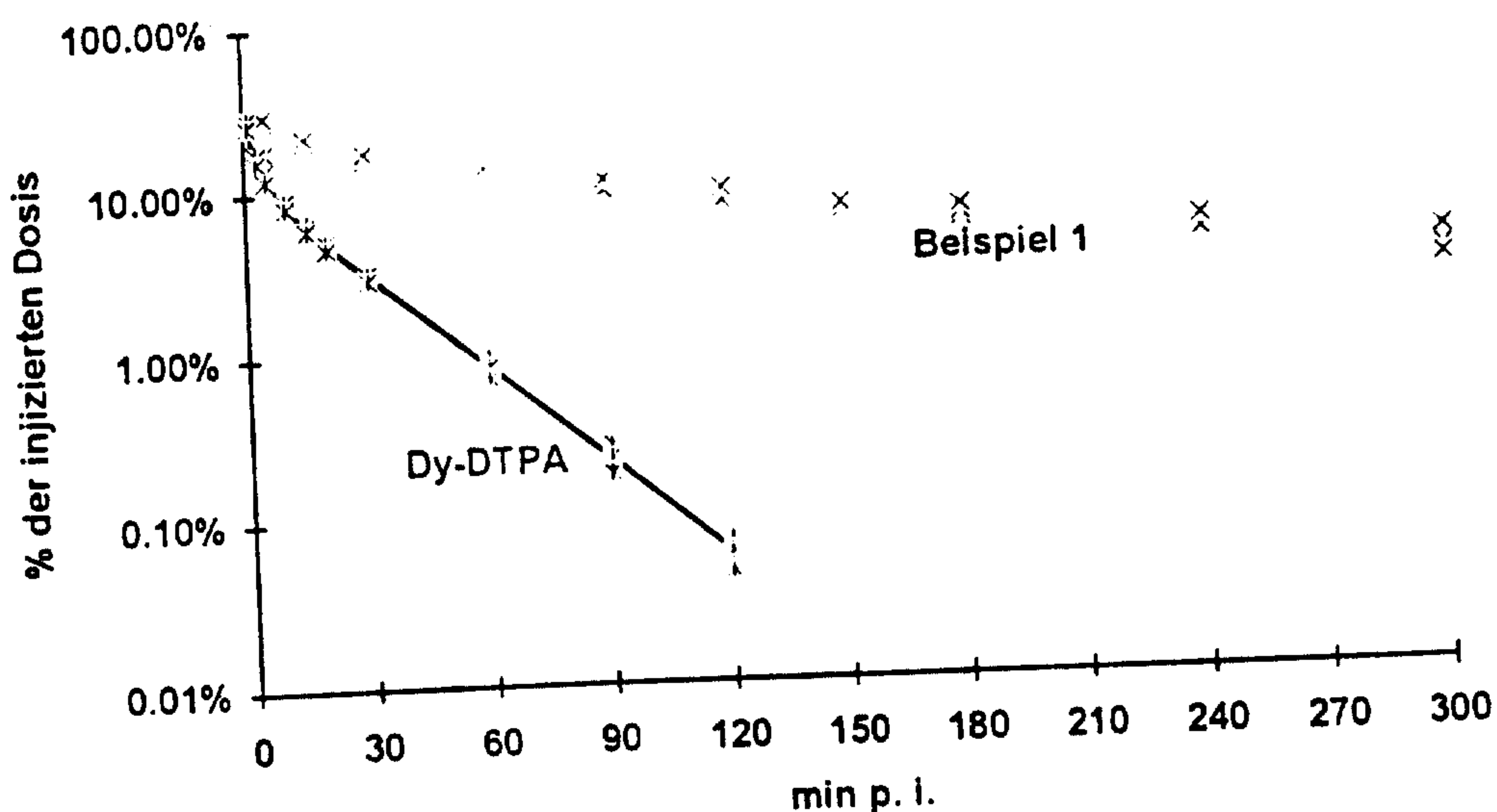
In comparison to Dy-DTPA (the dysprosium analogue of Magnevist^(R)), the fluorine compounds according to the invention (e.g., Example 1c) showed considerably slower elimination from the blood and, in addition, a smaller distribution volume (see also Figure 1 and Table 1).

It has been determined, surprisingly enough, that these compounds have an extended retention in the blood space and therefore are suitable as "blood-pool contrast media" -- e.g., for the visualization of blood vessels with suitable techniques -- even at relatively small dosages of $\leq 50 \mu\text{mol}$ of Gd per kg of body weight.

Figure 1:

Elimination from the blood (in % of the injected dose) of Dy-DTPA (dose: $100 \mu\text{mol}$ of Dy per kg of body weight, $n = 3$) and of the fluorine compounds of Example 1c according to the invention (dose: $50 \mu\text{mol}$ of Gd per kg of body weight, $n = 2$) after one-time intravenous administration of substances in rats (Han Wistar, Schering SPF, $\approx 250 \text{ g}$ of body weight).

The Gd and Dy contents in the blood were determined with the aid of ICP-AES.



[Key to Table:]

% der injizierten Dosis = % of injected dose

Beispiel 1c = Example 1c

Table 1:

Pharmacokinetic parameters: distribution volume (Vss), total clearance (CLtot) and elimination half-life (t β) of Dy-DTPA and the fluorine compounds of Example 1c according to the invention (calculated with TOPFIT 2.0; one- or two-compartment model).

	Vss (l/kg)		CLtot(ml/(min.kg))		t β (min)	
	MEAN	SD	MEAN	SD	MEAN	SD
Dy-DTPA	0.17	0.00	9.27	0.60	14.98	0.73
Example 1c	0.14	0.02	1.07	0.09	95.01	10.37

For further details see text for Figure 1

Example 40

Lymph Node Concentration in Guinea Pigs

Various fluorine-containing gadolinium and manganese complexes were examined 90 minutes to 24 hours after subcutaneous administration (2.5-10 μ mol of total gadolinium/kg of body weight, hind paw s.c.) to stimulated guinea pigs (complete Freund's adjunct; 0.1 ml i.m. respectively in the right and left thigh and lower leg; 2 weeks before administration of test substances) to determine their lymph node concentration in three successive lymph node stations (popliteal, inguinal, iliac). In this case, the results (determination of gadolinium concentration with the aid of ICP-AES) that are listed below in Table 2 were obtained:

Table 2

Substance Example No.	Time of lymph node removal (dose)	Gadolinium or manganese concentration in three successive lymph node stations [$\mu\text{mol/l}$] [% dose/g of tissue]			
		popliteal	inguinal	iliac	ratio
1c)	4 h (2.5 $\mu\text{mol/kg}$)	120 $\mu\text{mol/l}$ 17.2%	29 $\mu\text{mol/l}$ 4.2%	40 $\mu\text{mol/l}$ 5.6%	10:2.4: 3.3
2c)	4 h (10 $\mu\text{mol/kg}$)	435 $\mu\text{mol/l}$ 10.5%	84 $\mu\text{mol/l}$ 2.0%	150 $\mu\text{mol/l}$ 3.6%	10:2.0: 3.5
1e)	90 min (10 $\mu\text{mol/kg}$)	559 $\mu\text{mol/l}$ 15.0%	224 $\mu\text{mol/l}$ 6.0%	290 $\mu\text{mol/l}$ 7.8%	10:4.0: 5.2
3e)	90 min (10 $\mu\text{mol/kg}$)	880 $\mu\text{mol/l}$ 21.4%	277 $\mu\text{mol/l}$ 6.7%	339 $\mu\text{mol/l}$ 8.3%	10:3.1: 3.9

Table 2 shows that a high contrast medium concentration above three successive lymph node stations is to be noted.

Example 41**Lymph node visualization (MRT) after interstitial administration of the contrast medium**

Figure 1 shows MR recordings of popliteal and inguinal lymph nodes both before (left side: precontrast) and 120 minutes after (right side) subcutaneous administration (guinea pigs, hind paw, interdigital space) of the Gd complex of Example 2c (referred to in the figure as Gd-D03A-g-aminoamide-perfluorooctylether) (10 μ mol of Gd/kg of body weight). The T¹-weighted spin echo recordings (TR 400 ms, TE 15 ms) illustrate the strong signal rise in the popliteal and inguinal lymph nodes of the injected side of the body (straight arrow) in comparison to the non-injected side of the body (curved arrow), or to the precontrast image.

FIGURE 1: MRT-Lymph Node Visualization After Interstitial Administration of the Contrast Medium



[Key to Figure:]

MR-Bildgebung vor und nach interstitieller Applikation von 10 μmol Gd/kg = MR Imaging before and after interstitial administration of 10 μmol of Gd/kg

Gd-D03A- γ -aminoamid-perfluorooctylether = Gd-D03A- γ -aminoamide-perfluorooctylether

Präkontrast = precontrast

T1-gewichtete Spin-Echo-Sequenz (TR 400/TE 15) Meerschweinchen, Injektionsort: Zwischenzehenraum Hinterpfote (einseitig) Pfeile: poplitealer und inguinal profunder Lymphknoten der injizierten Seite = T¹-weighted spin echo sequence (TR 400/TE 15) guinea pigs, injection site: interdigital space of the hind paw (on one side); arrow: popliteally and inguinally profound lymph nodes of the injected side

Example 42**Elimination of the Contrast Medium after i.p. Administration**

After administration of a perfluorinated gadolinium complex according to the invention (100 μ mol of total gadolinium/kg of body weight) in the intraperitoneal space of rats, the retention of metal in the liver as well as in the remainder of the body was examined 14 days after administration. In this test, fluorine-containing compound 2c) was used. After 14 days p.i., the gadolinium concentration was 0.22% of the administered dose in the liver and 1.1% of the administered dose in the remainder of the body.

In comparison with this, Gd-DPTA-polylysine as polymeric material is not completely eliminated. After 14 days, the body still contains 7% of the original dose.

Example 43**Determination of T¹-Relaxivity of Selected Compounds**

The relaxivity of the following compounds was determined with a Minispec pc 20 (20 MHz, 0.47T) at 37°C in water and human plasma and compared to that of Gd-DTPA-polylysine and Magnevist^(R) as comparison substances.

Table 3

Substance Example No.	R ¹ [L/mmol*sec] at 0.47 T and 37°C	
	Water	Plasma
1c)	41	49
2c)	19	33
3c)	15.2	27.5
22f)	6.9	20.5
30c)	21.1	26.9
31c)	5.2	29.1
32c)	19.4	24.8
33f)	31.5	35.7
34b)	25.9	24.9
35d)	23.1	34.0
37d)	19.9	n.d.
38c)	23.3	30.5
Comparison substances:		
Magnevist ^(R)	3.8	4.8
Gd-DTPA-polylysine ¹⁾	13.1	16.8

n.d. = not determined

¹⁾ from Invest. Radiol. 1992, 346

Claims

1. Perfluoroalkyl-containing compounds of general formula I

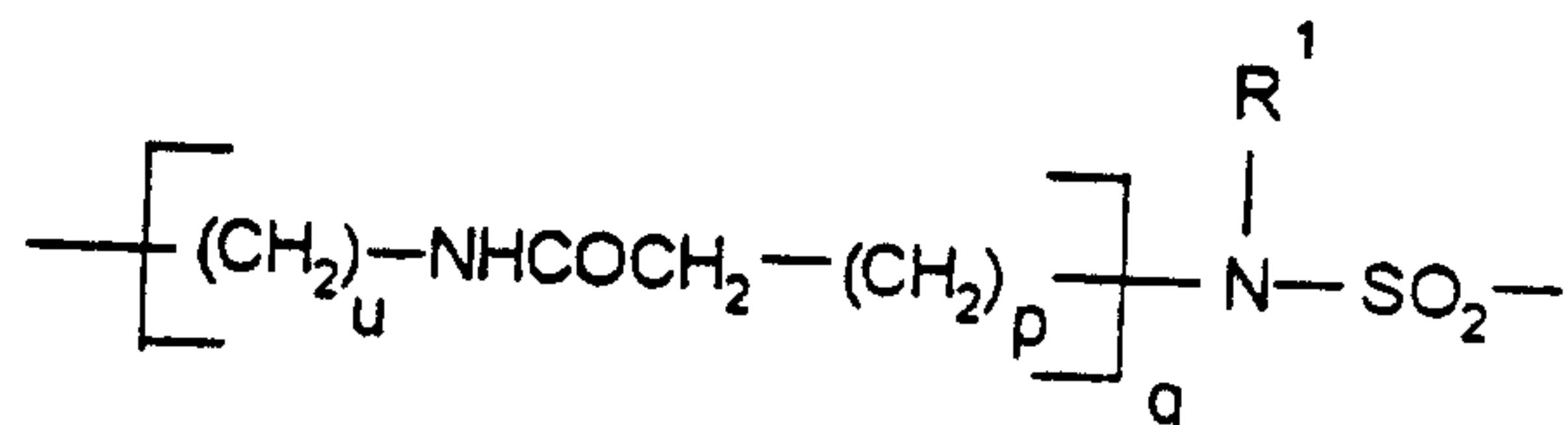


in which

R^{F} is a perfluorinated, straight-chain or branched carbon chain with formula $-\text{C}_n\text{F}_{2n}\text{X}$, in which

X represents a terminal fluorine, chlorine, bromine, iodine or hydrogen atom, and n stands for numbers 4-30,

L means a direct bond, a methylene group, an $-\text{NHCO}$ group, a group

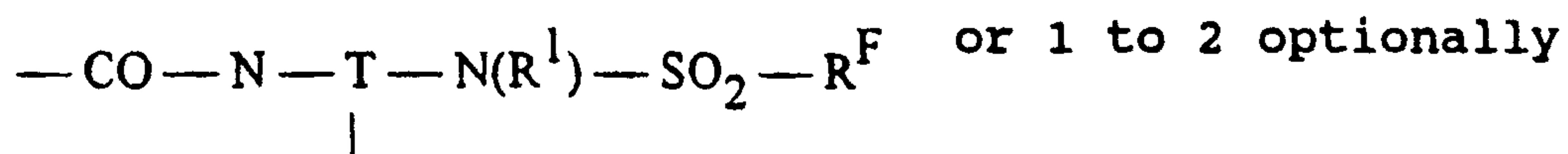


whereby p means numbers 0 to 10, q and u , independently of one another, mean numbers 0 or 1 and

R^1 is a hydrogen atom, a methyl group, a $-\text{CH}_2-\text{OH}$ group, a $-\text{CH}_2-\text{CO}_2\text{H}$ group or a C_2-C_{15} chain, which optionally is interrupted by 1 to 3 oxygen atoms, 1 to 2 $>\text{CO}$ groups or an optionally substituted aryl group and/or is substituted with 1 to 4 hydroxyl groups, 1 to 2 C_1-C_4 alkoxy groups, 1 to 2 carboxy groups, a group $-\text{SO}_3\text{H}$,

or L is a straight-chain, branched, saturated or unsaturated C_2-C_{30} carbon chain, which optionally contains 1 to 10 oxygen atoms, 1 to 3 $-\text{NR}^1$ groups, 1 to 2 sulfur atoms, a piperazine, a $-\text{CONR}^1$ group, an $-\text{NR}^1\text{CO}$ group,

an $-\text{SO}_2$ group, an $-\text{NR}^1-\text{CO}_2$ group, 1 to 2 $-\text{CO}$ groups, a group



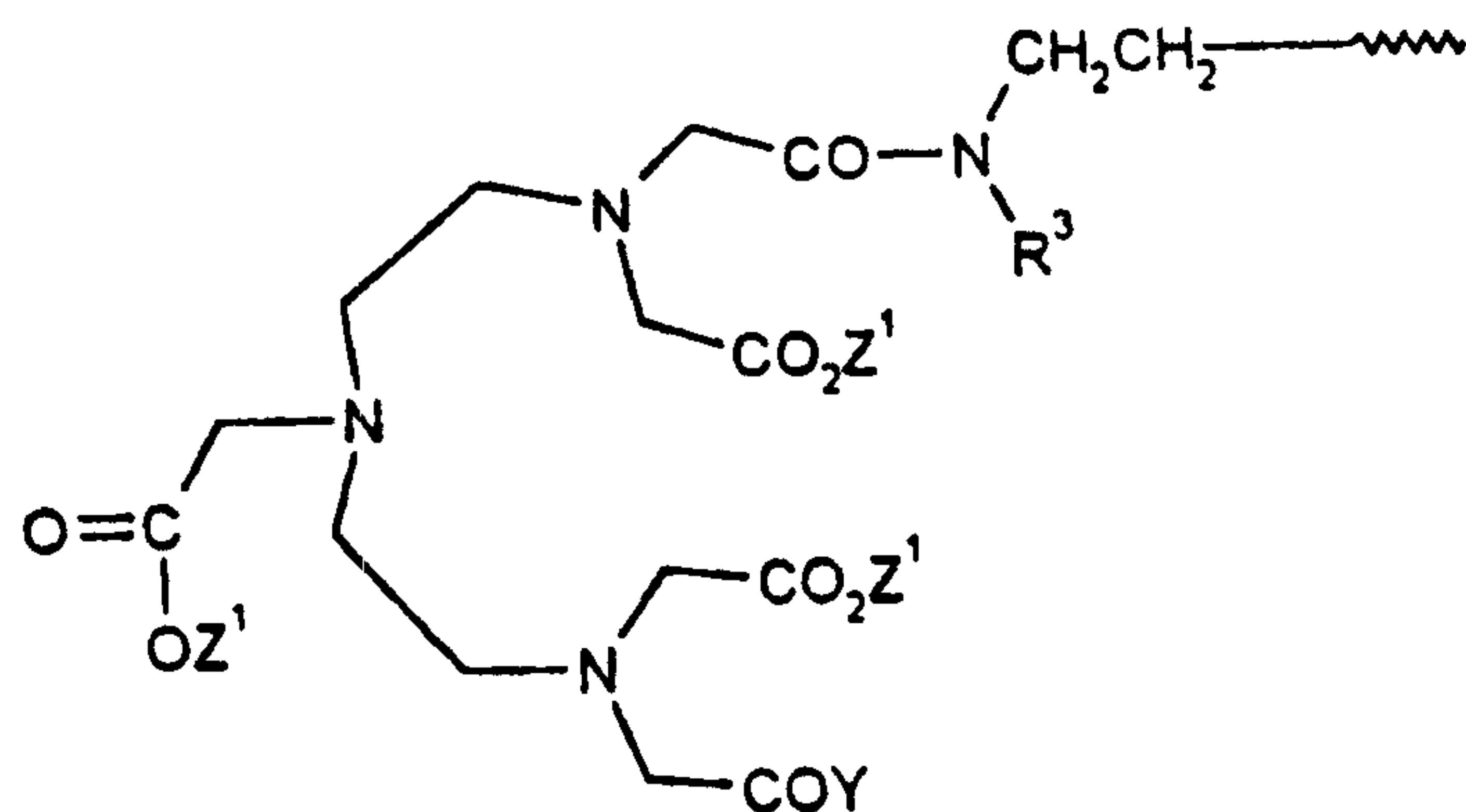
substituted aryls and/or is interrupted by these groups and/or is optionally substituted with 1 to 3 $-\text{OR}^1$ groups, 1 to 2 oxo groups, 1 to 2 $-\text{NH}-\text{COR}^1$ groups, 1 to 2 $-\text{CONHR}^1$ groups, 1 to 2 $-(\text{CH}_2)_p-\text{CO}_2\text{H}$ groups, 1 to 2 groups of $-(\text{CH}_2)_p-(\text{O})_q-\text{CH}_2\text{CH}_2-\text{R}^F$,

whereby

R^1 , R^F and p and q have the above-indicated meanings, and

T means a C_2-C_{10} chain, which optionally is interrupted by 1 to 2 oxygen atoms or 1 to 2 $-\text{NHCO}$ groups,

A stands for a complexing agent or metal complex or their salts of organic and/or inorganic bases or amino acids or amino acid amides, selected from a complexing agent or complex of general formula II



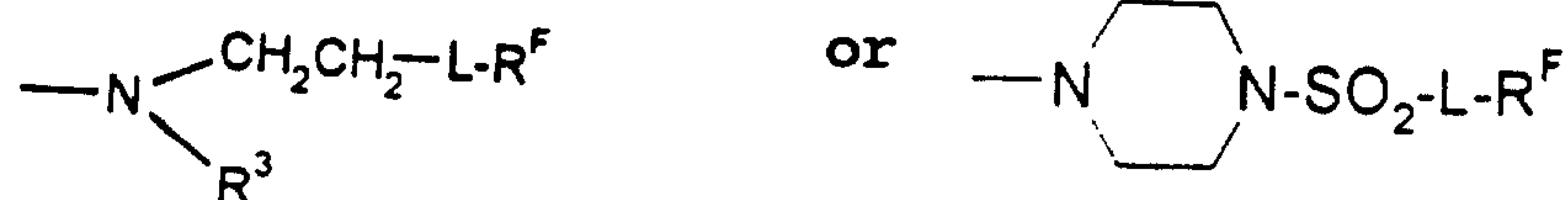
(II)

in which R^3 , Z^1 and Y are independent of one another,
and

R^3 has the meaning of R^1 or means $-(CH_2)_m-L-R^F$,
whereby m is 0, 1 or 2 and L and R^F have the
above-mentioned meaning,

Z^1 independently of one another, mean a hydrogen atom
or a metal ion equivalent of atomic numbers 21-29,
39, 42, 44 or 57-83,

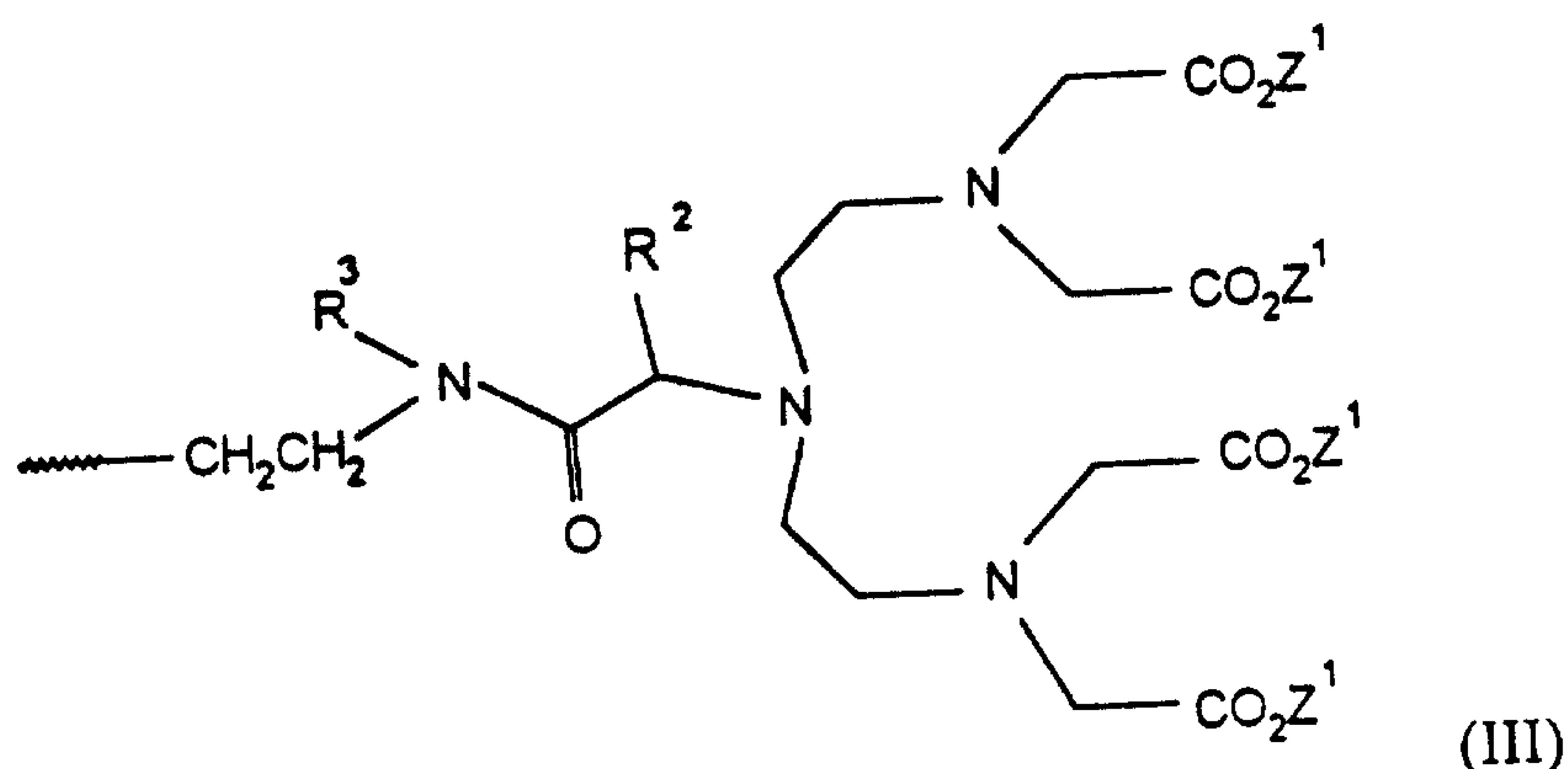
Y means $-OZ^1$ or



whereby Z^1 , L , R^F and R^3 have the above-mentioned
meanings,

or

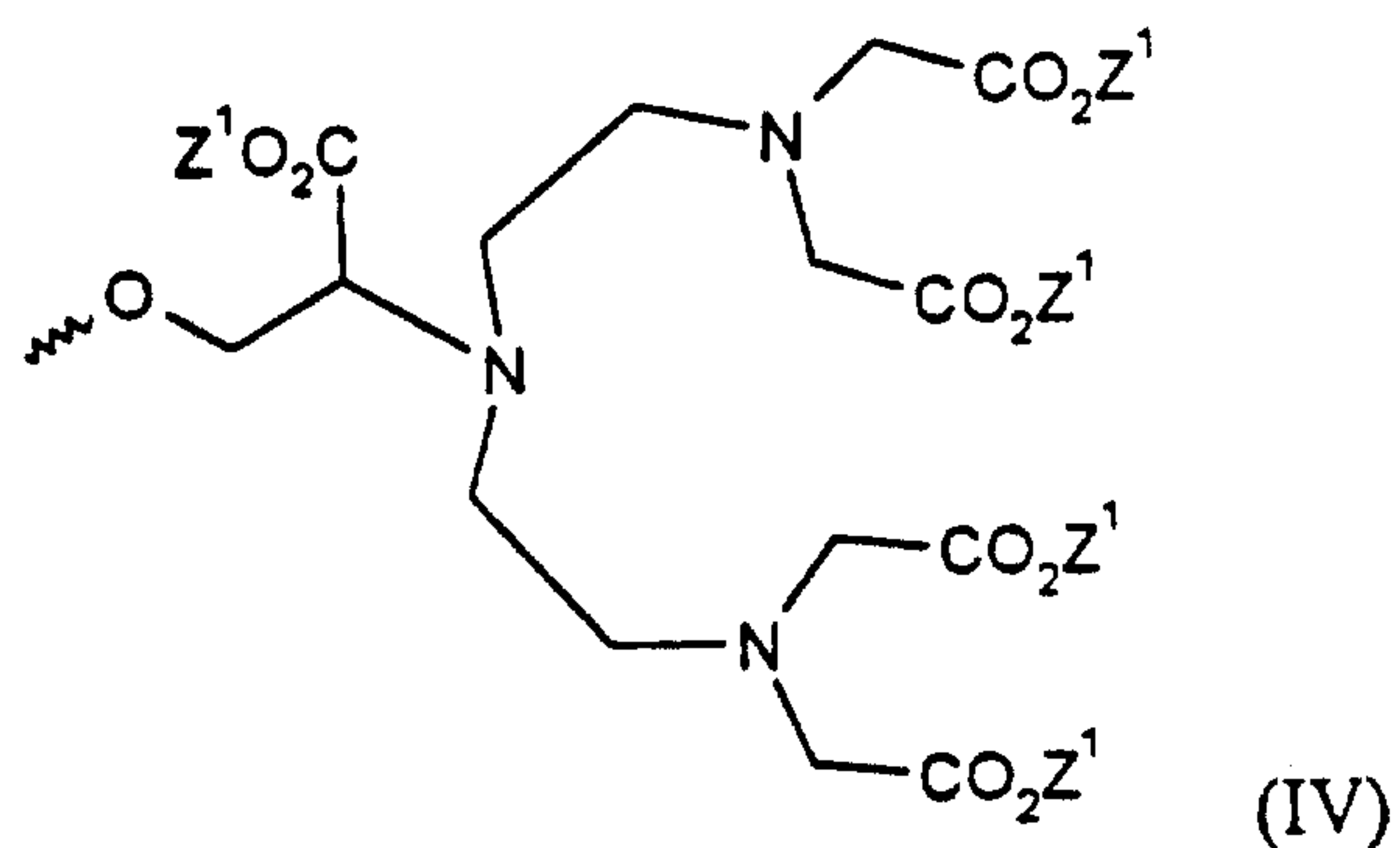
A stands for a complexing agent or complex of general
formula III



in which R^3 and Z^1 have the above-mentioned
meanings and R^2 has the meaning of R^1 ,

or

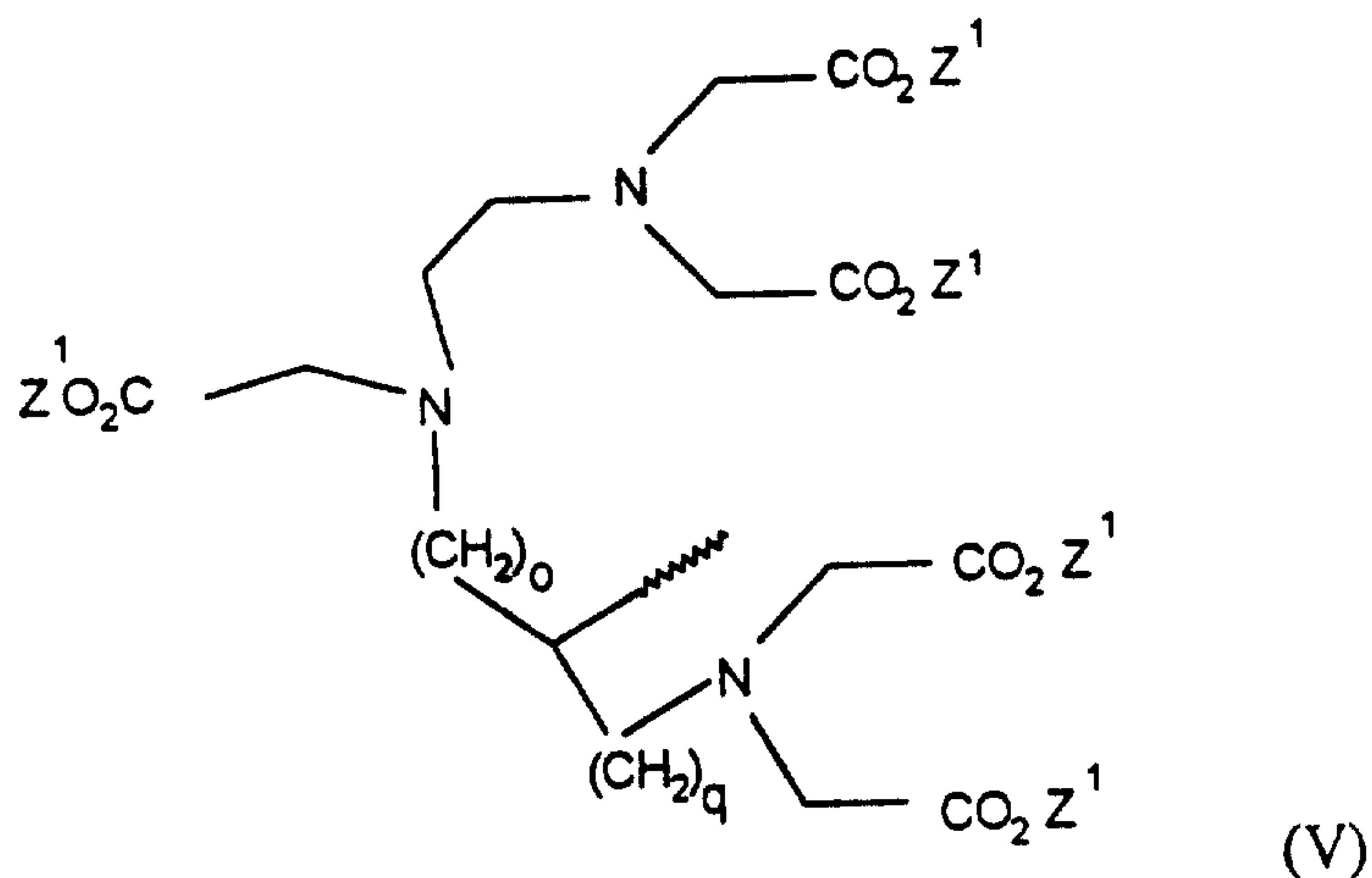
A stands for a complexing agent or complex of general formula IV



in which Z¹ has the above-mentioned meaning,

or

A stands for a complexing agent or complex of general formula V

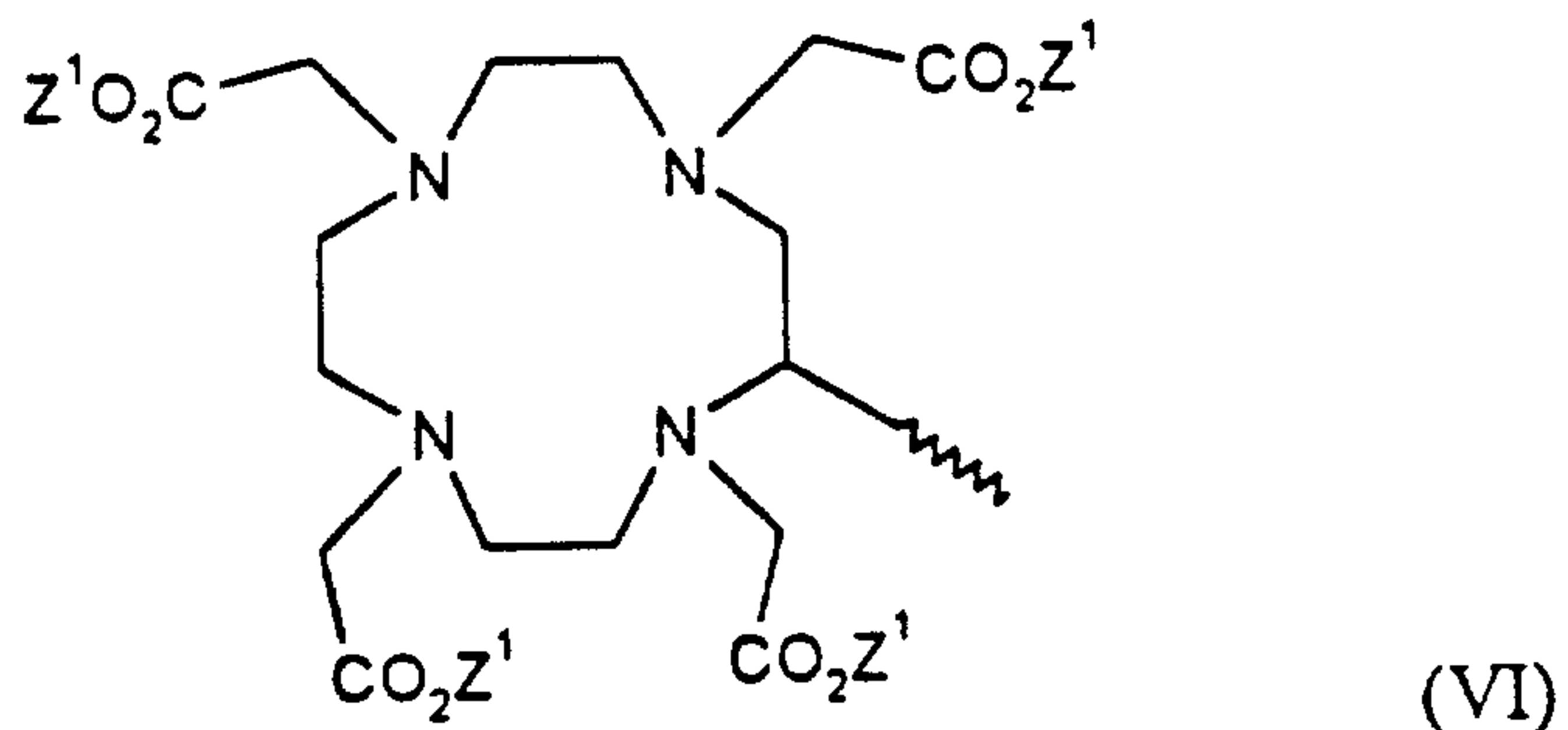


in which Z¹ has the above-mentioned meaning, and o and q stand for numbers 0 or 1 and the sum

$o + q = 1$ results,

or

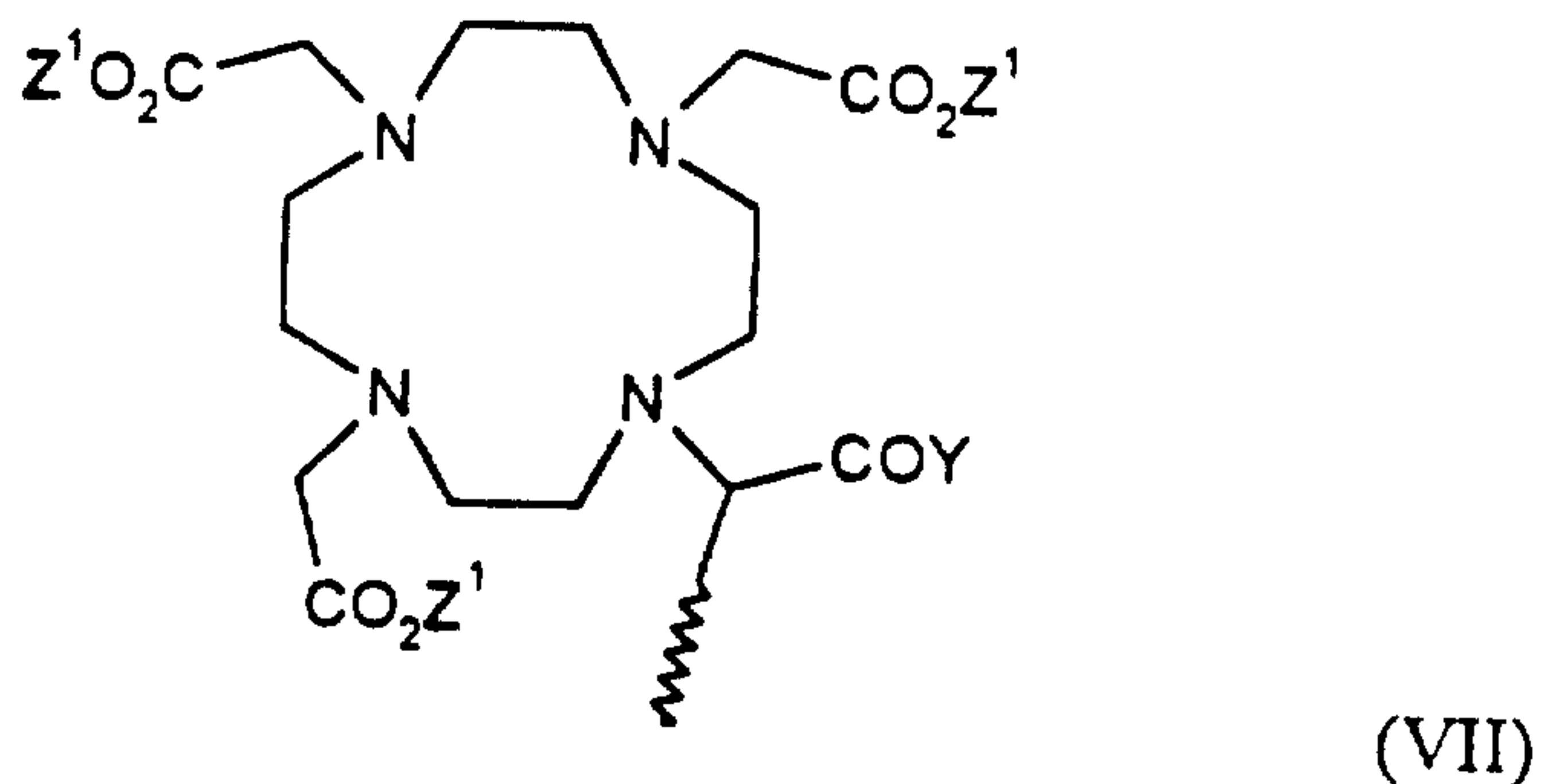
A stands for a complexing agent or complex of general formula VI



in which Z^1 has the above-mentioned meaning

or

A stands for a complexing agent or complex of general formula VII

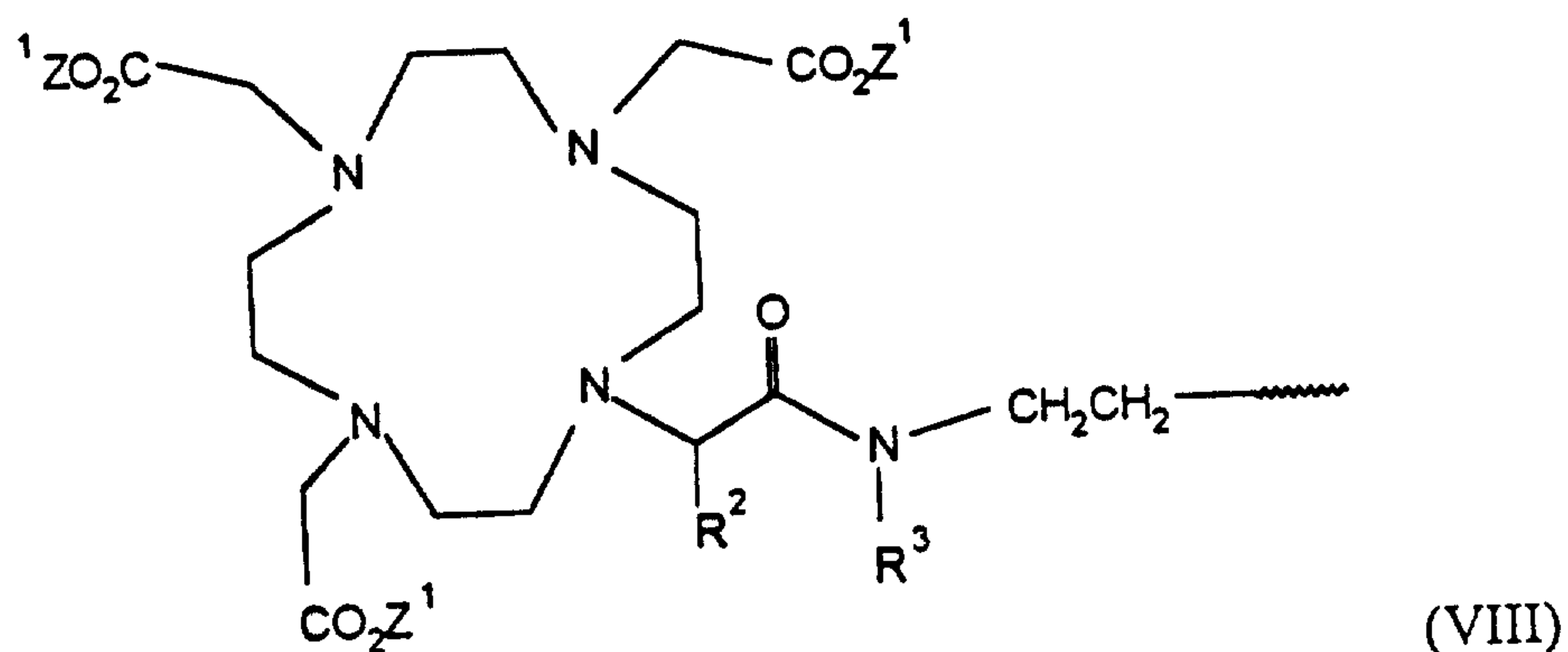


in which Z^1 and Y have the above-mentioned meanings

or

A stands for a complexing agent or complex of general

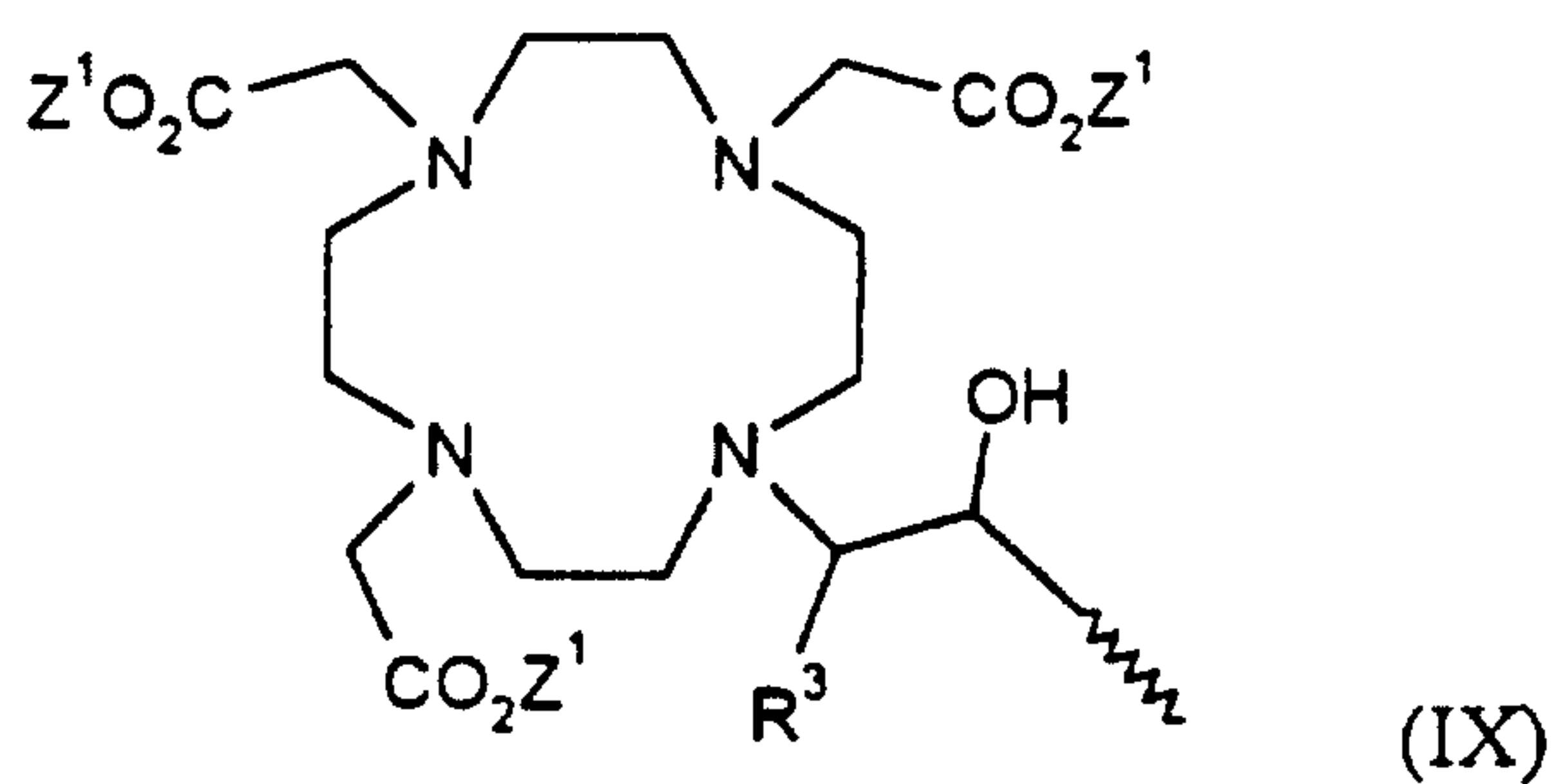
formula VIII



in which R^3 and Z^1 have the above-mentioned meanings, and R^2 has the above-mentioned meaning of R^1 ,

or

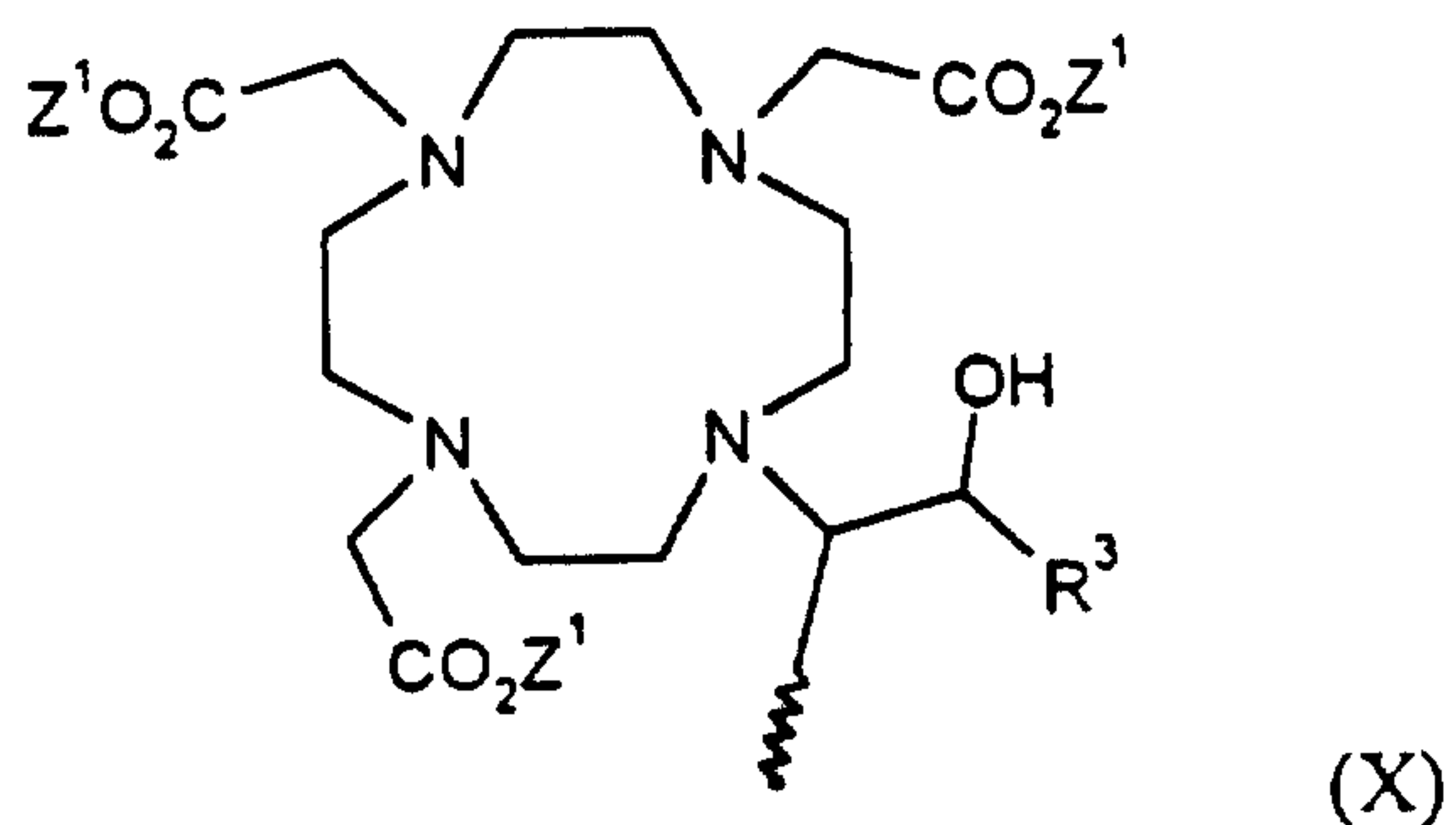
A stands for a complexing agent or complex of general formula IX



in which R^3 and Z^1 have the above-mentioned meanings,

or

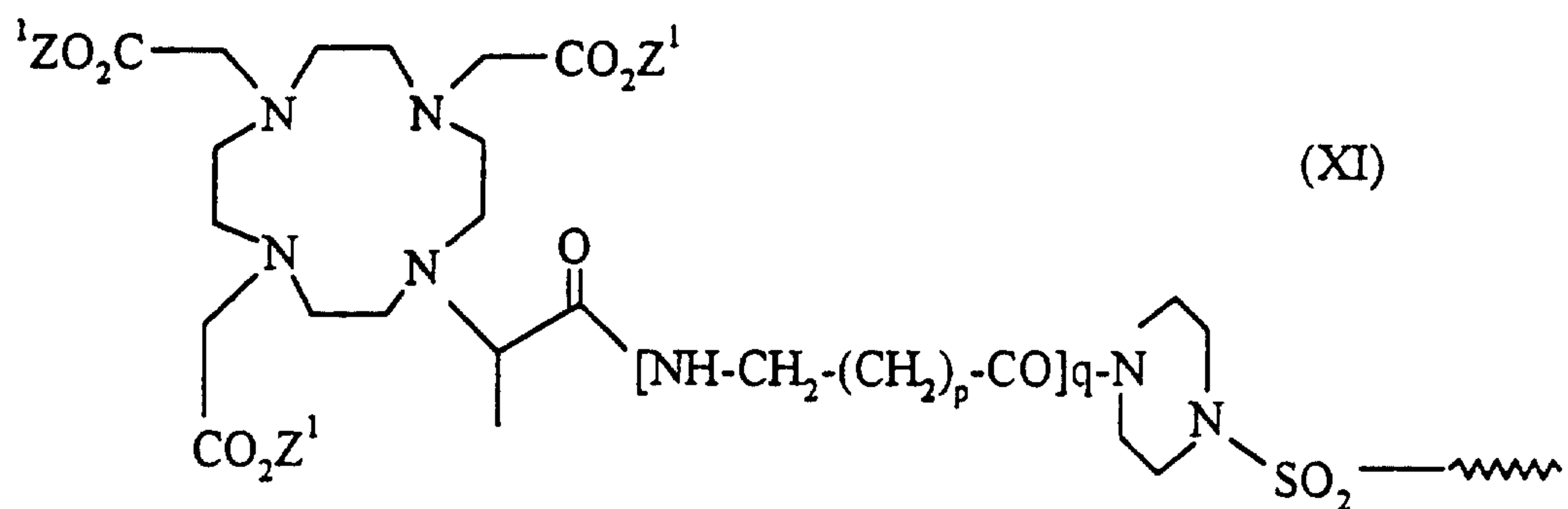
A stands for a complexing agent or complex of general formula X



in which R^3 and Z^1 have the above-mentioned meanings,

or

A stands for a complexing agent or complex of general formula XI

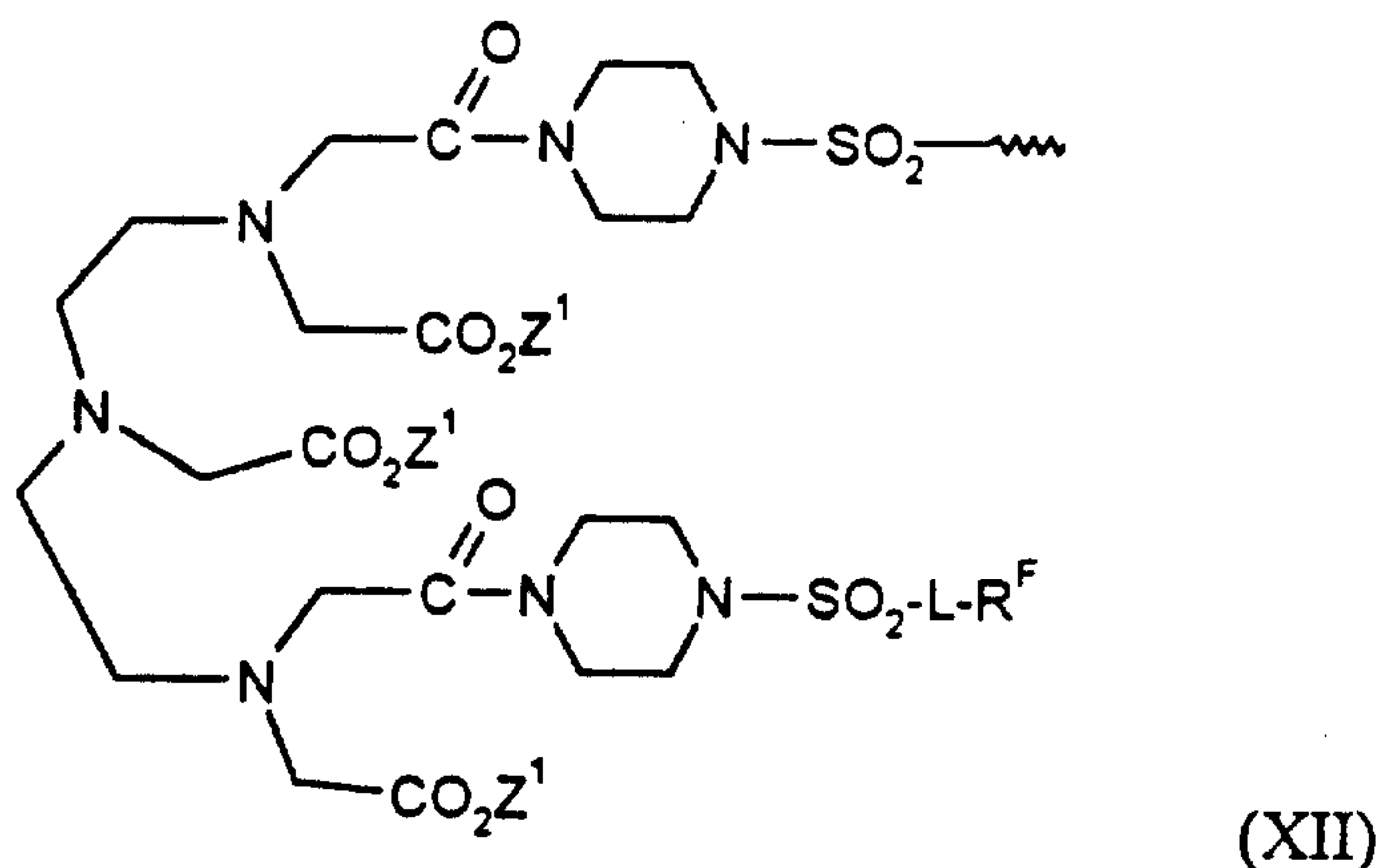


in which Z^1 , p and q have the above-mentioned meanings and R^2 has the meaning of R^1 ,

or

A stands for a complexing agent or complex of general

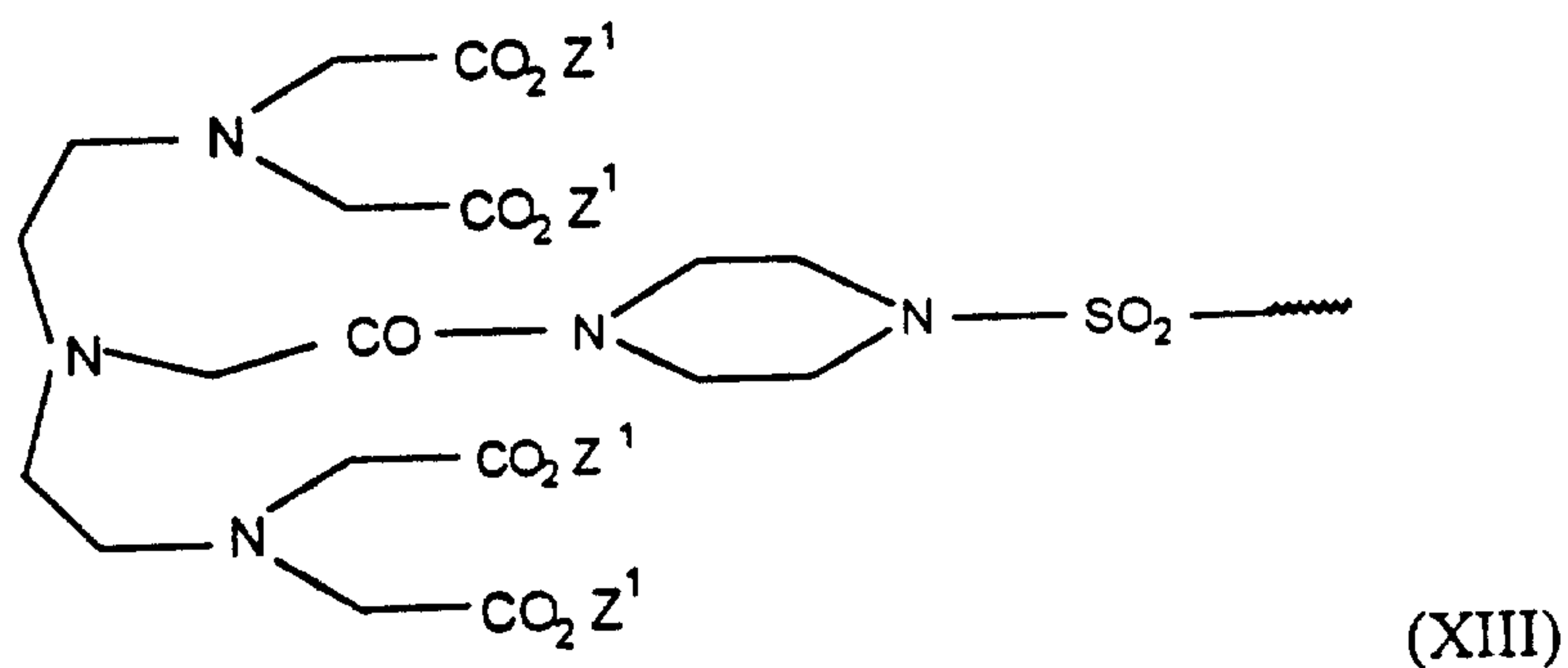
formula XII



in which L, R^F and Z¹ have the above-mentioned meanings,

or

A stands for a complexing agent or complex of general formula XIII



in which Z¹ has the above-mentioned meaning.

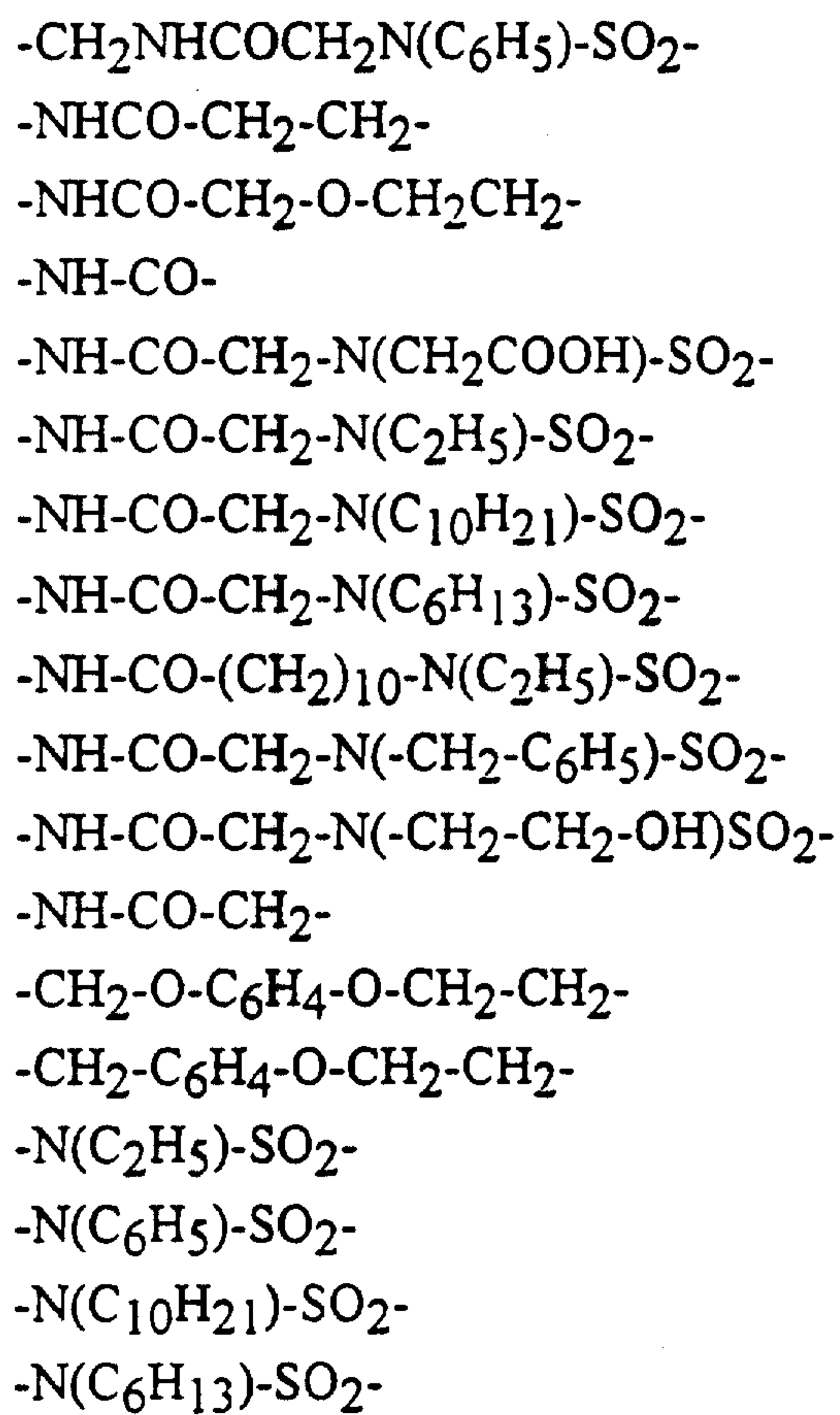
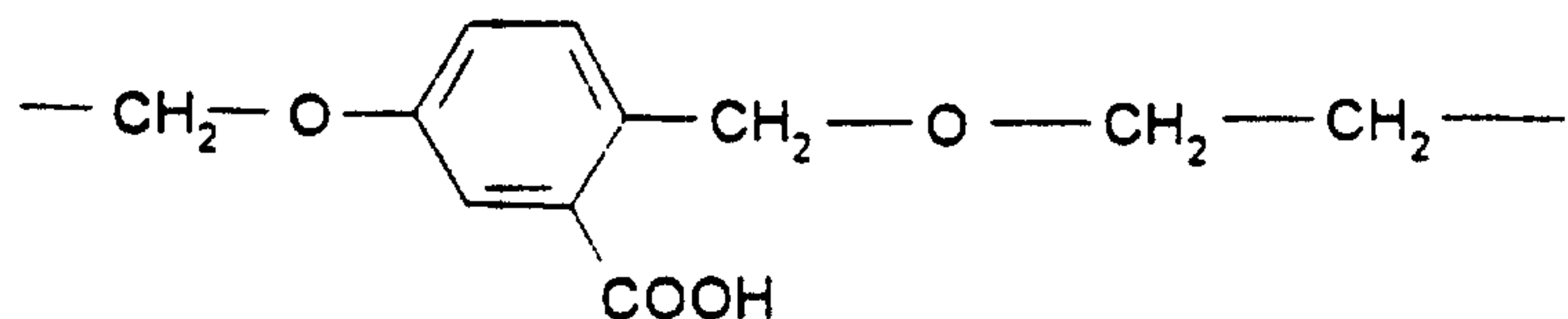
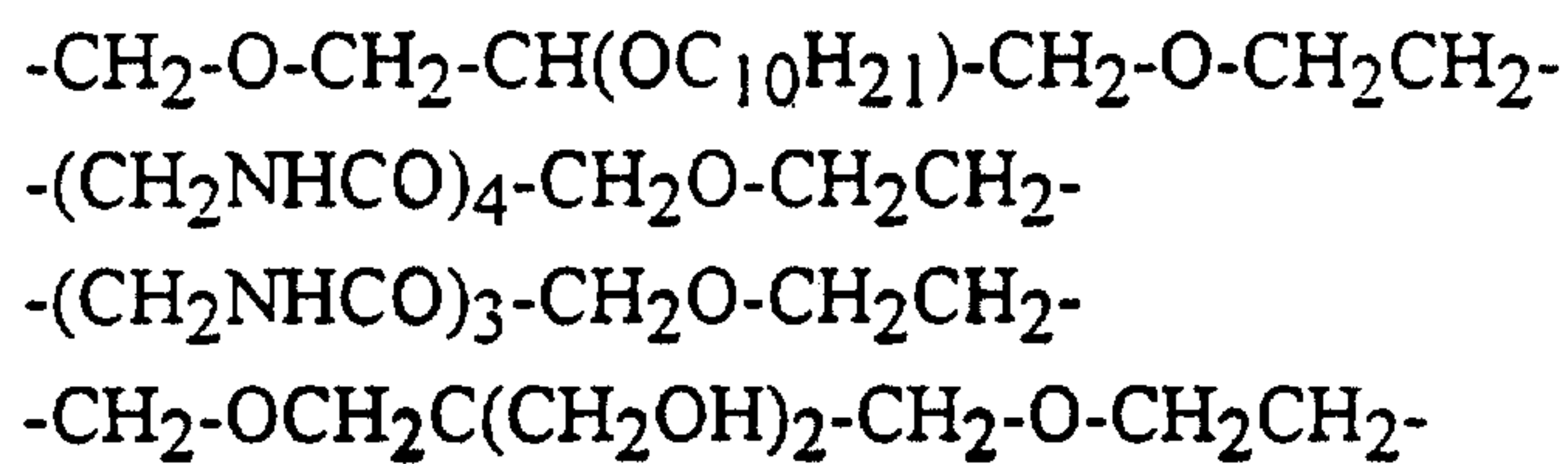
2. Compounds according to claim 1, characterized in that Z¹ stands for a hydrogen atom.

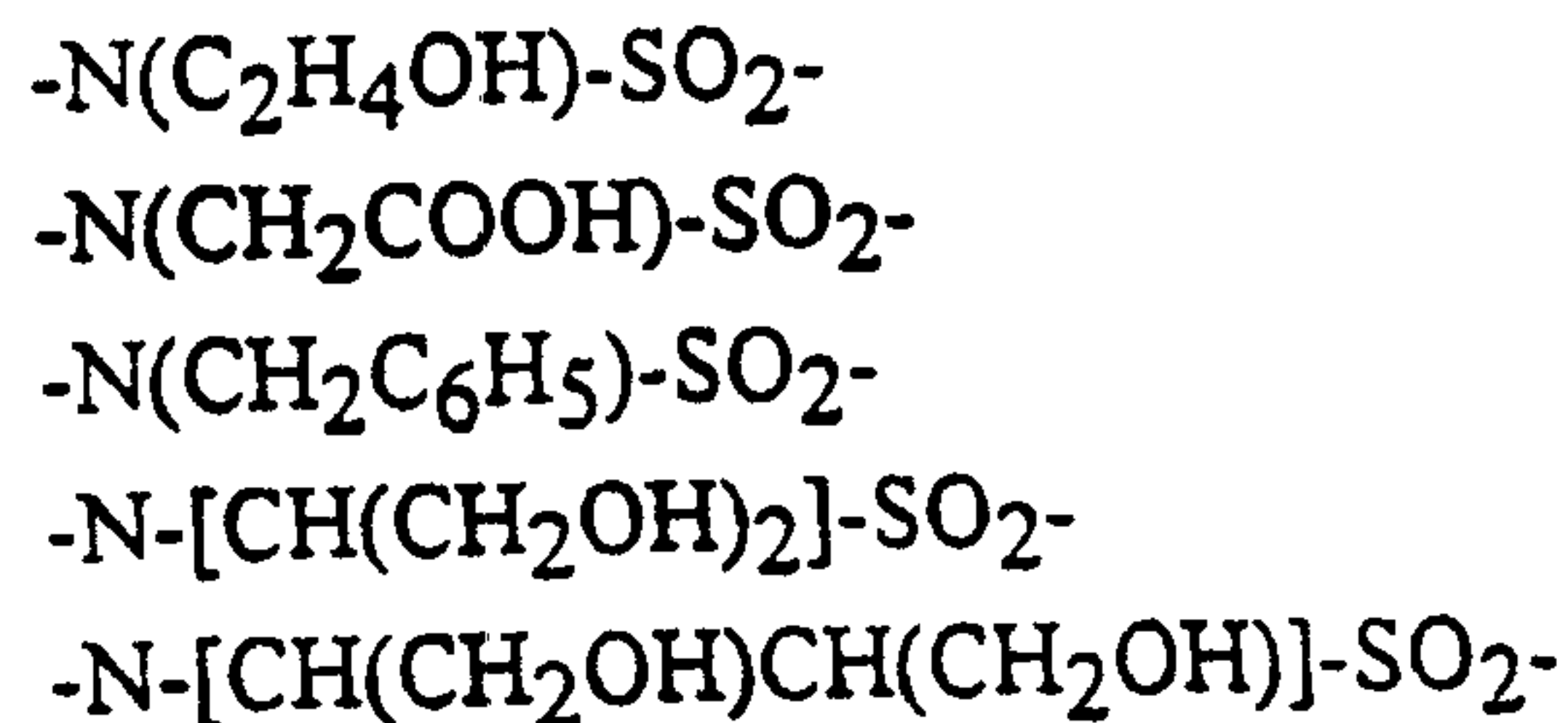
3. Compounds according to claim 1 or 2, wherein n in formula -C_nF_{2n} X stands for numbers 4-15.

5. Compounds according to one of claims 1 to 4, wherein L stands for

-CH₂-
-CH₂CH₂-
-(CH₂)_s- s = 3 - 15
-CH₂-O-CH₂CH₂-
-CH₂-(O-CH₂-CH₂-)_t t = 2 - 6
-CH₂-NH-CO-
-CH₂-NH-CO-CH₂-N(CH₂COOH)-SO₂-
-CH₂-NH-CO-CH₂-N(C₂H₅)-SO₂-
-CH₂-NH-CO-CH₂-N(C₁₀H₂₁)-SO₂-
-CH₂-NH-CO-CH₂-N(C₆H₁₃)-SO₂-
-CH₂-NH-CO-(CH₂)₁₀-N(C₂H₅)-SO₂-

$$\begin{aligned}
& -\text{CH}_2\text{-NH-CO-CH}_2\text{-N(-CH}_2\text{-C}_6\text{H}_5\text{)-SO}_2\text{-} \\
& -\text{CH}_2\text{-NH-CO-CH}_2\text{-N(-CH}_2\text{-CH}_2\text{-OH)SO}_2\text{-} \\
& -\text{CH}_2\text{-NHCO-(CH}_2\text{)}_{10}\text{-S-CH}_2\text{CH}_2\text{-} \\
& -\text{CH}_2\text{NHCOCH}_2\text{-O-CH}_2\text{CH}_2\text{-} \\
& -\text{CH}_2\text{NHCO(CH}_2\text{)}_{10}\text{-O-CH}_2\text{CH}_2\text{-} \\
& -\text{CH}_2\text{-C}_6\text{H}_4\text{-O-CH}_2\text{CH}_2\text{-} \\
& -\text{CH}_2\text{-O-CH}_2\text{-C(CH}_2\text{-OCH}_2\text{CH}_2\text{-C}_6\text{F}_{13}\text{)}_2\text{-CH}_2\text{-OCH}_2\text{-CH}_2\text{-} \\
& -\text{CH}_2\text{-NHCOCH}_2\text{CH}_2\text{CON-CH}_2\text{CH}_2\text{NHCOCH}_2\text{N(C}_2\text{H}_5\text{)SO}_2\text{C}_8\text{F}_{17} \\
& \quad \quad \quad | \\
& \quad \quad \quad \text{CH}_2\text{-CH}_2\text{NHCOCH}_2\text{N(C}_2\text{H}_5\text{)-SO}_2\text{-}
\end{aligned}$$





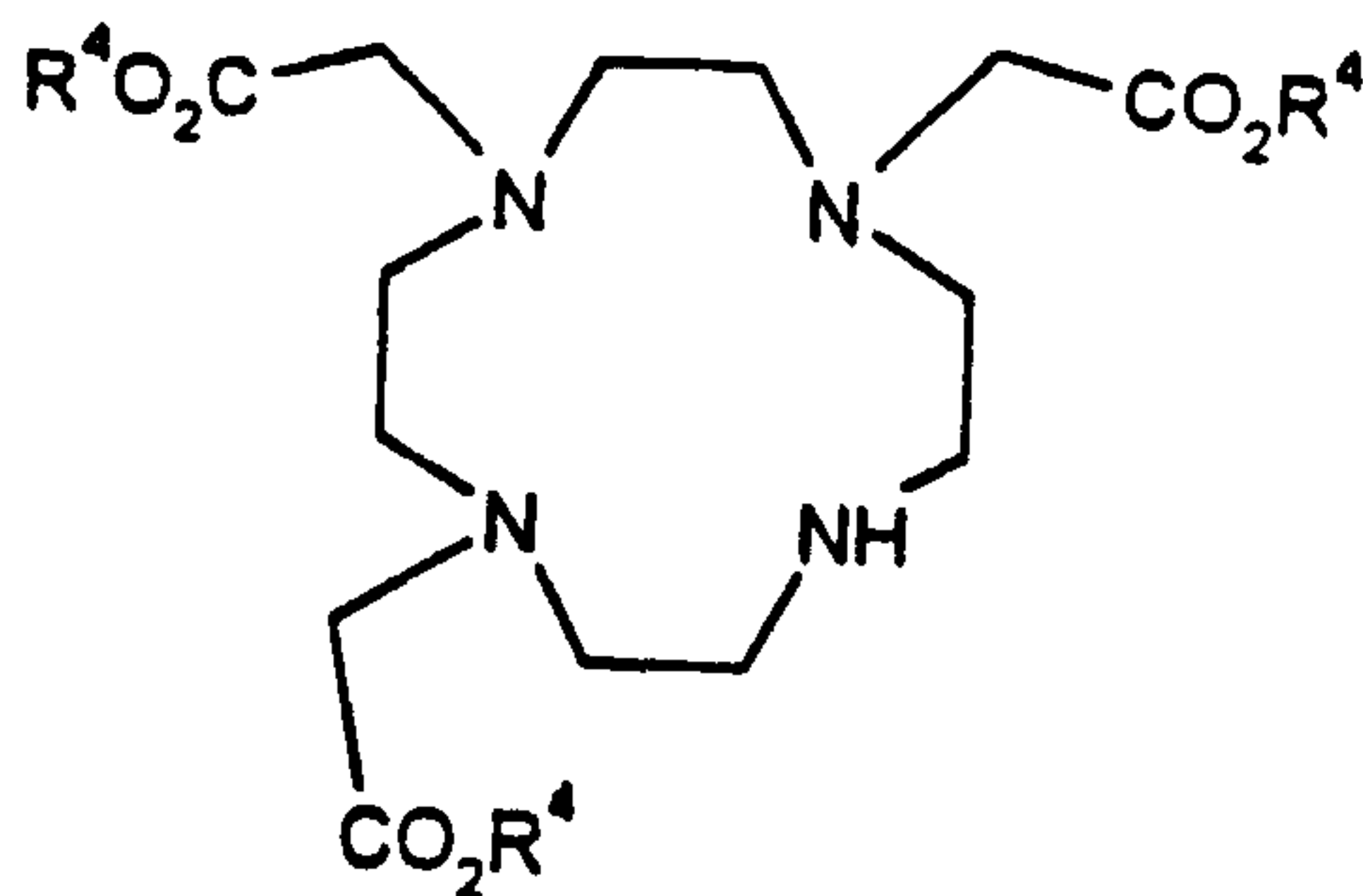
6. The compounds according to claim 1:

Gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-aza-7-(perfluorooctylsulfonyl)-nonyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane,

gadolinium complex of 10-[2-hydroxy-4-aza-5-oxo-7-oxa-10,10,11,11,12,12,13,13,14,14,15,15,16,16,17,17,17-heptadecafluoro-heptadecyl]-1,4,7-tris(carboxymethyl)-1,4,7,10-tetraazacyclododecane.

7. Process for the production of perfluoroalkyl-containing compounds of general formula I as defined in claim 1, wherein

- a) Compounds of general formula I with A in the meaning of general formula IX are produced by compounds of general formula 20

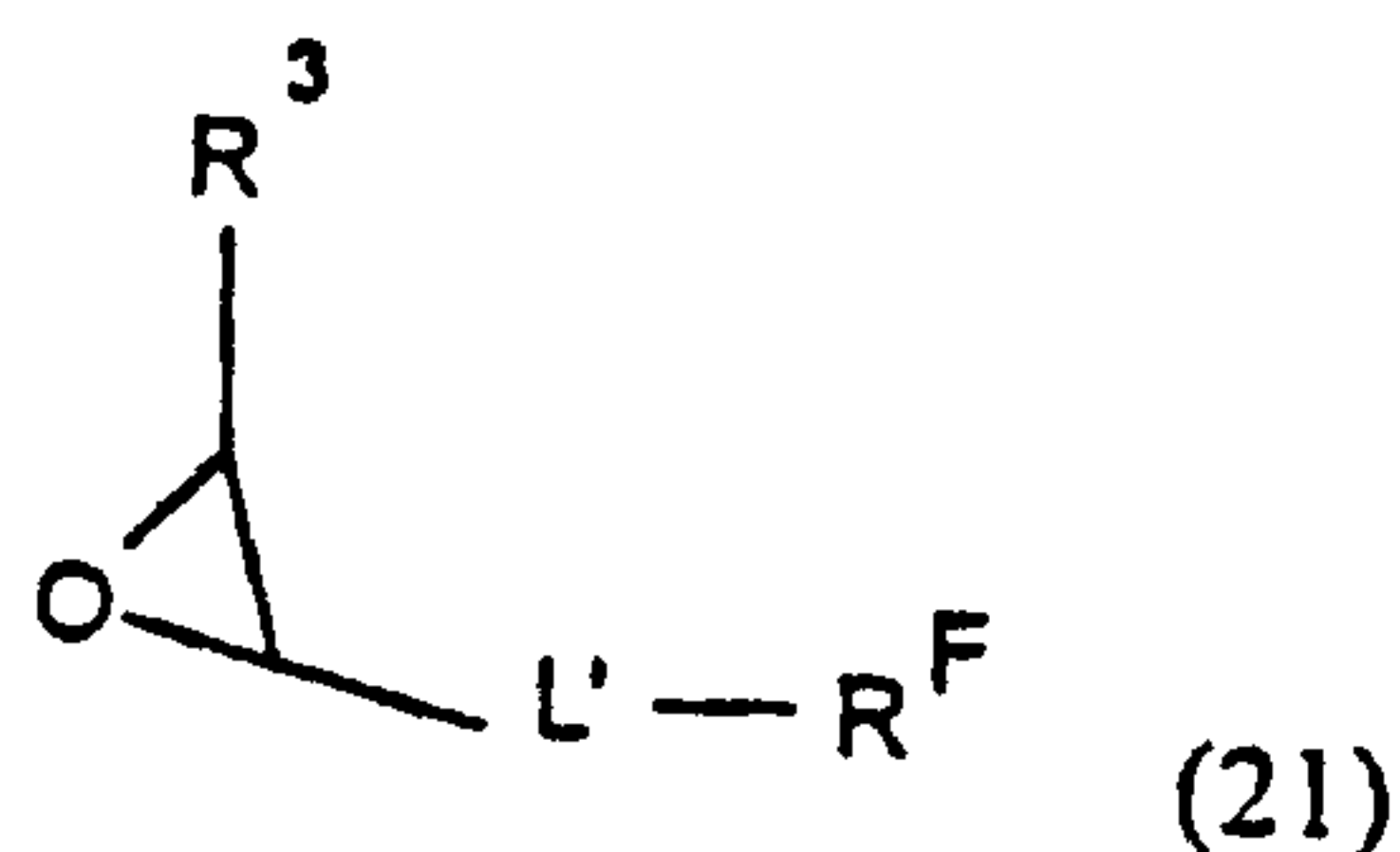


(20)

in which

R^4 means hydrogen, methyl, ethyl, isopropyl, t-butyl or benzyl,

being reacted with epoxides of general formula 21



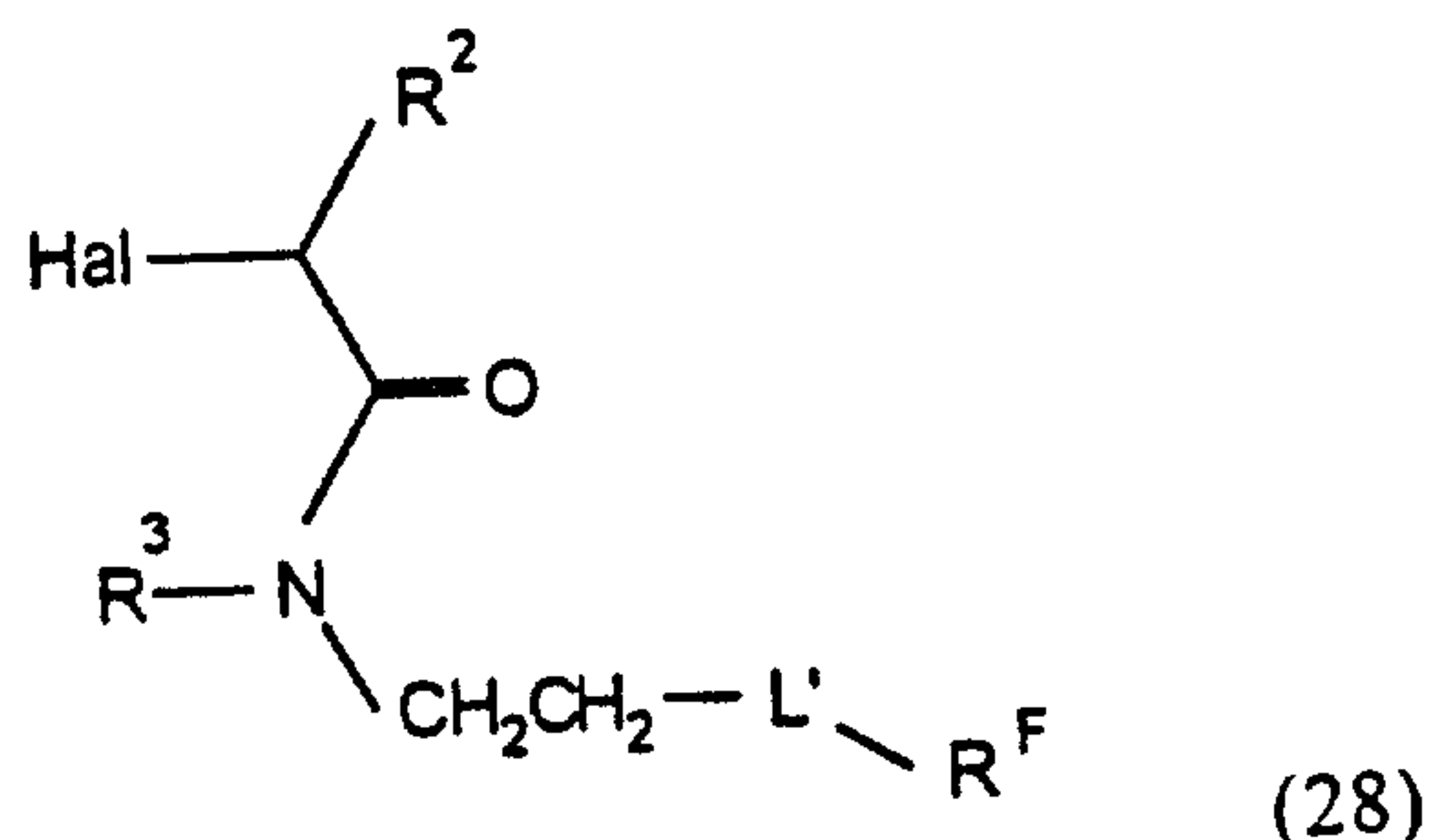
in which

R^3 has the meaning of R^1 , optionally in protected form, or means $-(CH_2)_m-L-R^F$, whereby m can be 0, 1 or 2, L' has the meaning of L , optionally in protected form, and R^F is a perfluorinated carbon chain,

in alcohols, ethers, water or in mixtures of water and an organic solvent at temperatures of between -10°C and 180°C with the addition of inorganic and/or organic bases, then optionally present protective groups being cleaved, the thus obtained complexing agents being reacted with at least one metal oxide or metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature, and then -- optionally -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides,

b) compounds of general formula I with A in the meaning of general formula VIII are produced by compounds of

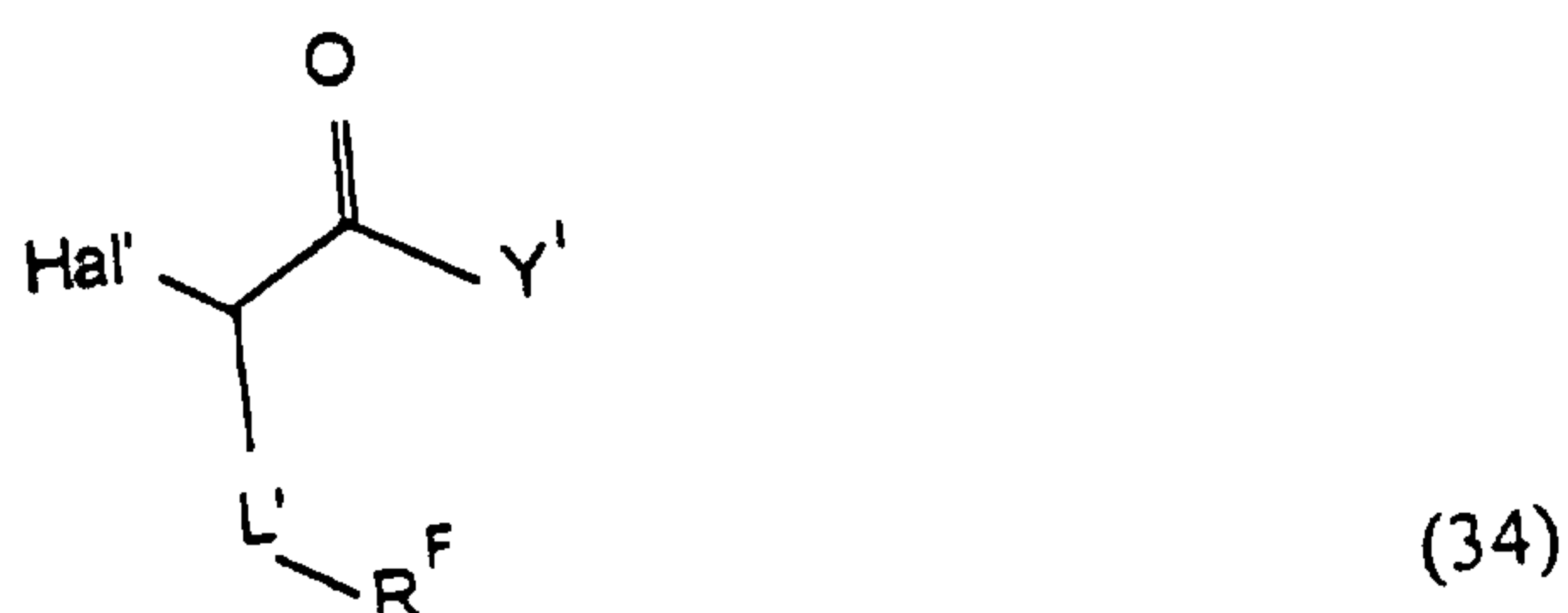
general formula 20 being alkylated in a way known in the art with compounds of general formula 28



in which R^2 has the meaning of R^1 , Hal means chlorine, bromine and iodine, and R^F , L' and R^3 have the above-mentioned meanings,

then optionally present protective groups being cleaved and the procedure further being performed as under a) with the thus obtained complexing agents,

- c) compounds of general formula I with A in the meaning of general formula VII are produced by compounds of general formula 20 being reacted in a way known in the art with compounds of general formula 34



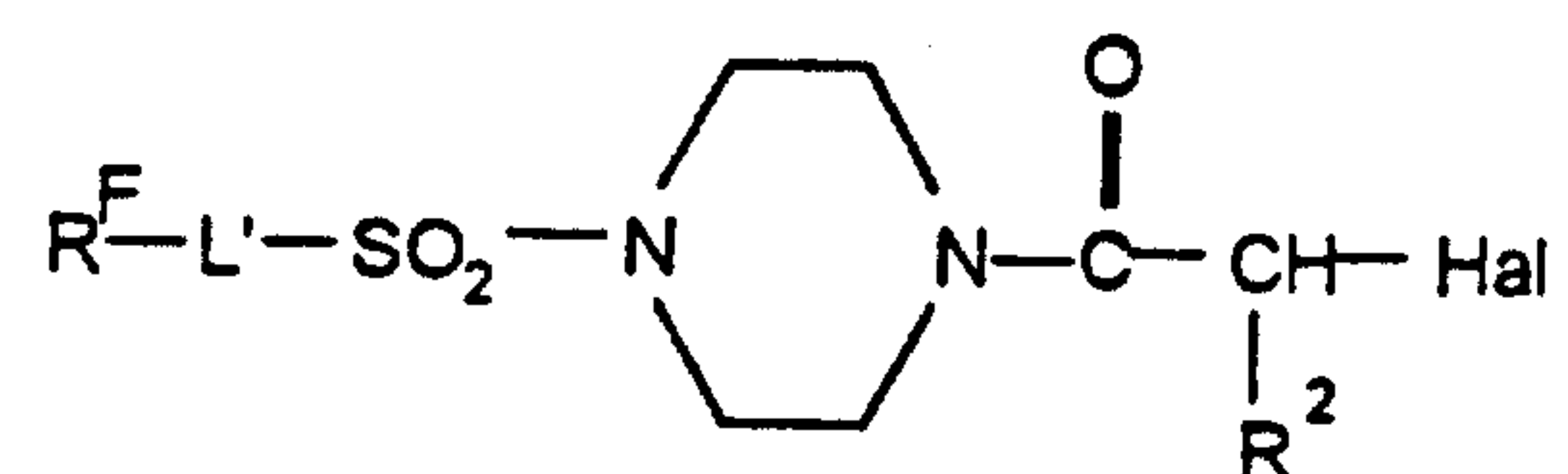
in which Hal' has the meaning of Hal, F, -OTs, OMs, Y' stands for radicals -OH and $-\text{N}-\text{CH}_2-\text{CH}_2-\text{R}^F$

$\begin{array}{c} | \\ \text{R}^3 \end{array}$

and L' and R^F have the above-mentioned meaning,

then optionally present protective groups being cleaved and the procedure further being performed as under a) with the thus obtained complexing agents,

- d) compounds of general formula I with A in the meaning of general formula XI with q in the meaning of number 0 are produced by compounds of general formula 20 being reacted with compounds of general formula 68

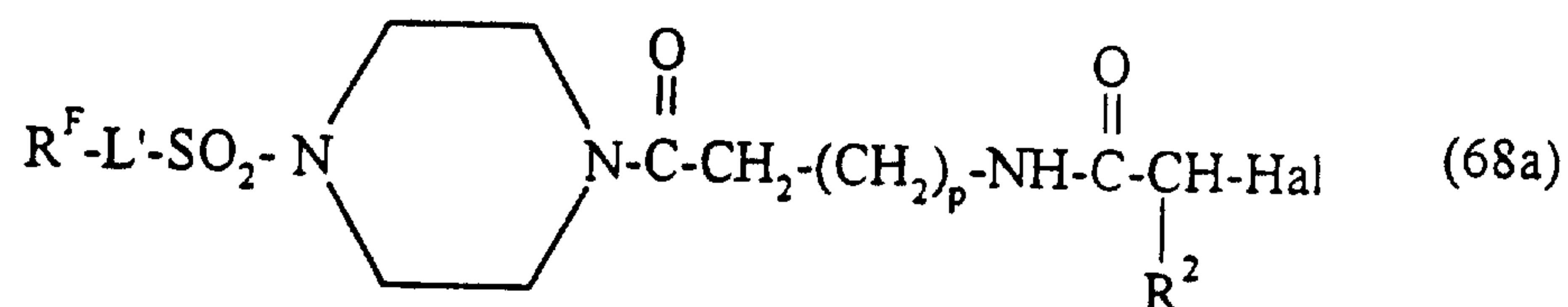


(68)

in which R^{F} , L' , R^2 and Hal have the above-mentioned meanings,

in an organic solvent at elevated temperatures for several hours, then optionally present protective groups being cleaved and the procedure further being performed as under a) with the thus obtained complexing agents,

- e) compounds of general formula I with A in the meaning of general formula XI with q in the meaning of number 1 are produced by compounds of general formula 20 being reacted with compounds of general formula 68a

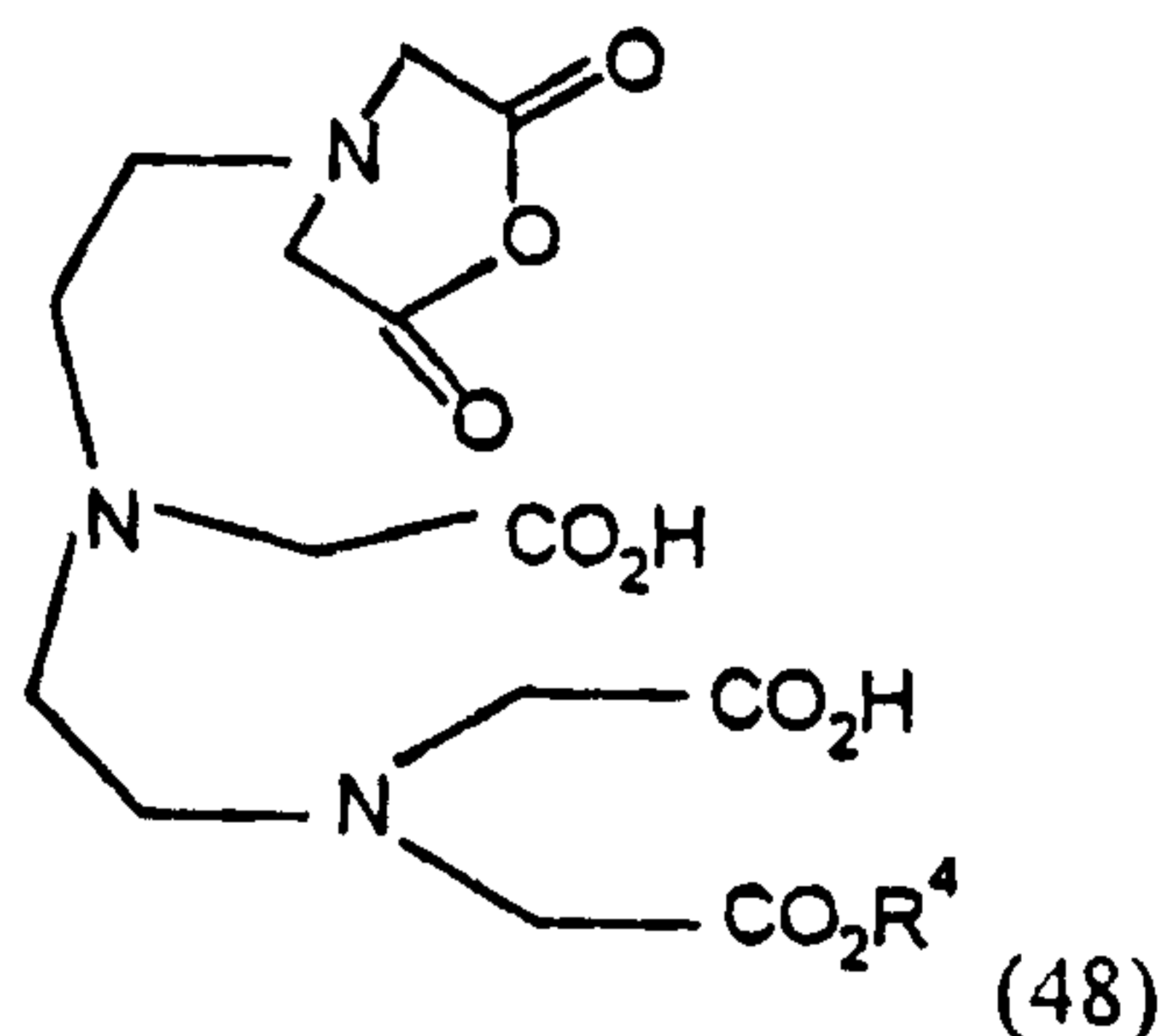


in which R^F , L' , R^3 , p and Hal have the above-mentioned meanings,

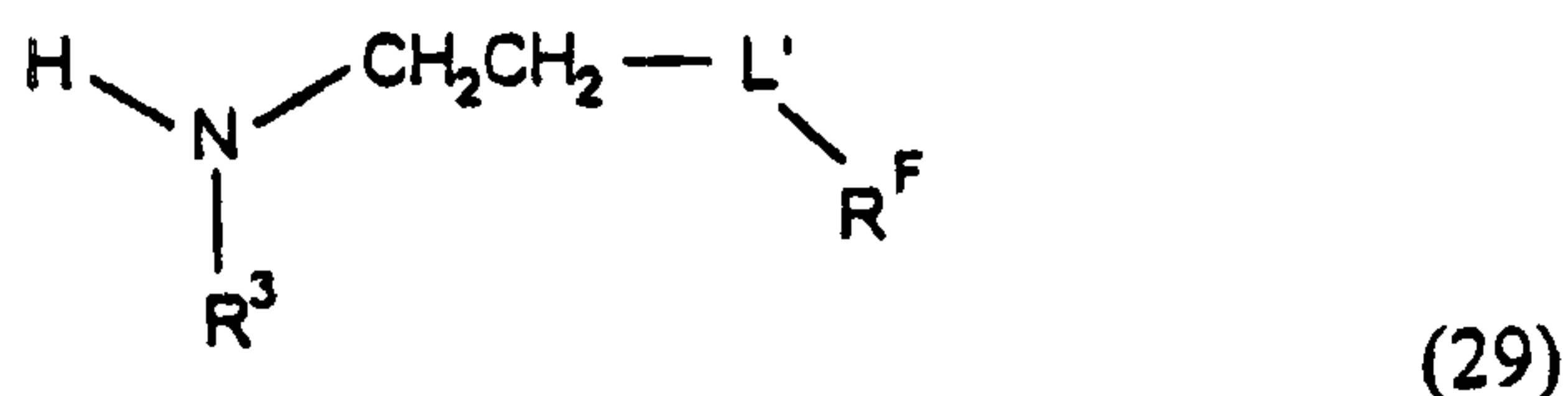
in an organic solvent at elevated temperatures for several hours, then optionally present protective groups being cleaved and the procedure being performed as under a) with the thus obtained complexing agents.

8. Process for the production of perfluoroalkyl-containing compounds of general formula I as defined in claim 1, wherein

- a) compounds of general formula I with A in the meaning of general formula II are produced by, if Y in general formula II stands for an OH group, compounds of general formula 48



in which R^4 has the above-mentioned meaning, being reacted with an amine of general formula 29

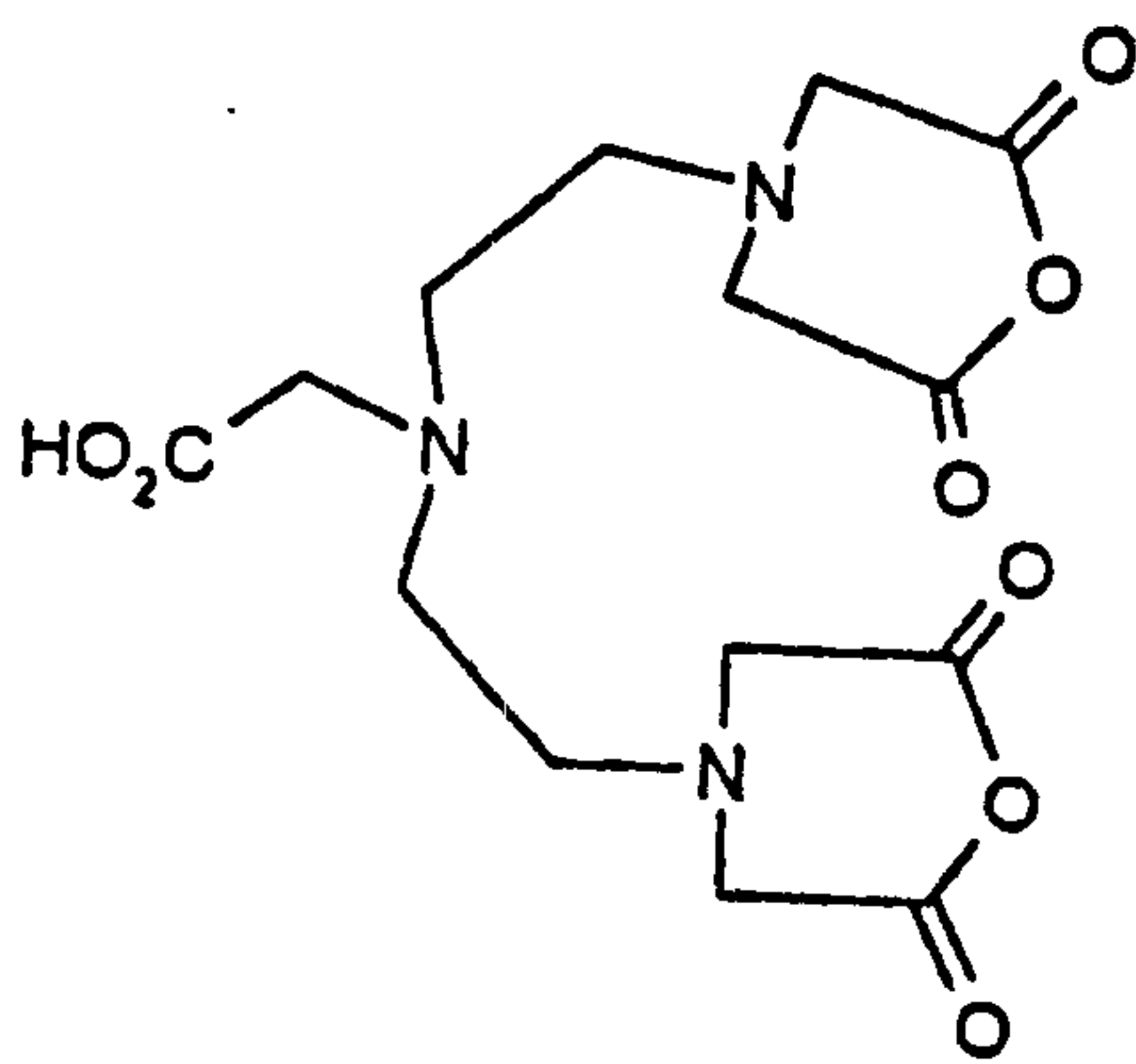


in which R^3 , L' and R^F have the above-mentioned meanings,

in an organic solvent, optionally with the addition of inorganic and/or organic bases, at elevated temperatures, then optionally present protective groups being cleaved, the thus obtained complexing agents being reacted with at least one metal oxide or metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature, and then -- optionally -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides, or, if Y in general formula II stands for group $-N-CH_2-CH_2-L'-R^F$

R^3 ,

the bisanhydride of the diethylenetriamine-pentaacetic acid of general formula 49

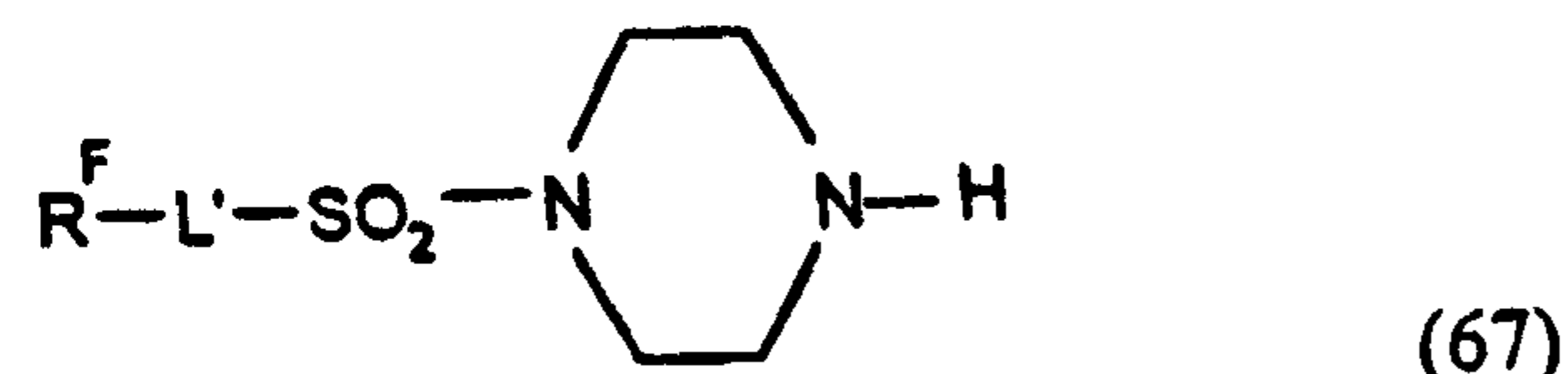


(49)

being reacted under analogous conditions with an amine of formula 29

in an organic solvent, optionally with the addition of inorganic and/or organic bases, at elevated temperatures, then optionally present protective groups being cleaved, the thus obtained complexing agents being reacted with at least one metal oxide or metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature, and then – optionally – acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides,

- b) compounds of general formula I with A in the meaning of general formula XII are produced by bisanhydride 49 being reacted with piperazine derivatives of general formula 67

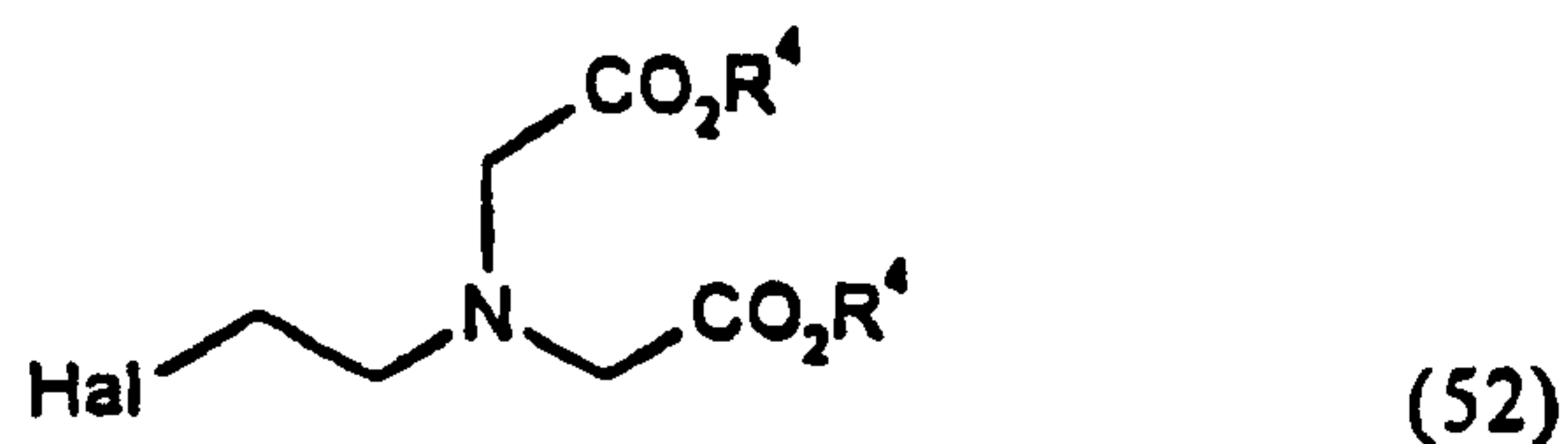


in which R^F and L' have the above-mentioned meaning,

under the same conditions as under a), then optionally present protective groups being cleaved and the procedure further being performed as under a).

9. Process for the production of perfluoroalkyl-containing compounds of general formula I as defined in claim 1, wherein

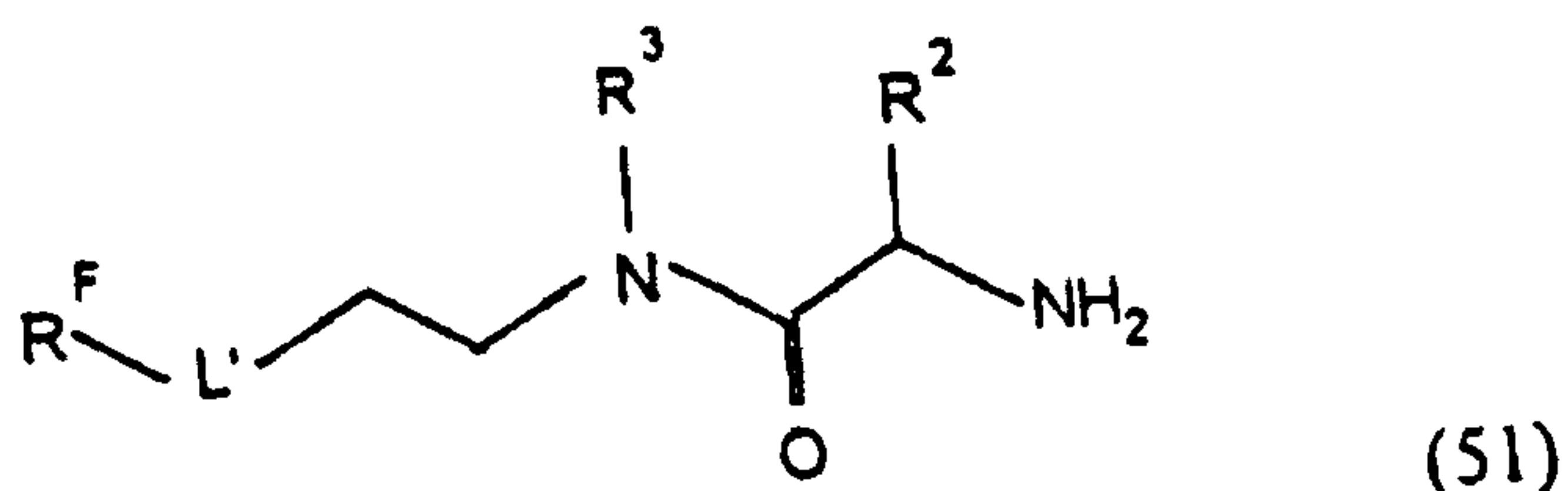
- a) Compounds of general formula I with A in the meaning of general formula III are produced by halocarboxylic acid derivatives of general formula 52



in which Hal and R⁴ have the above-mentioned meaning,

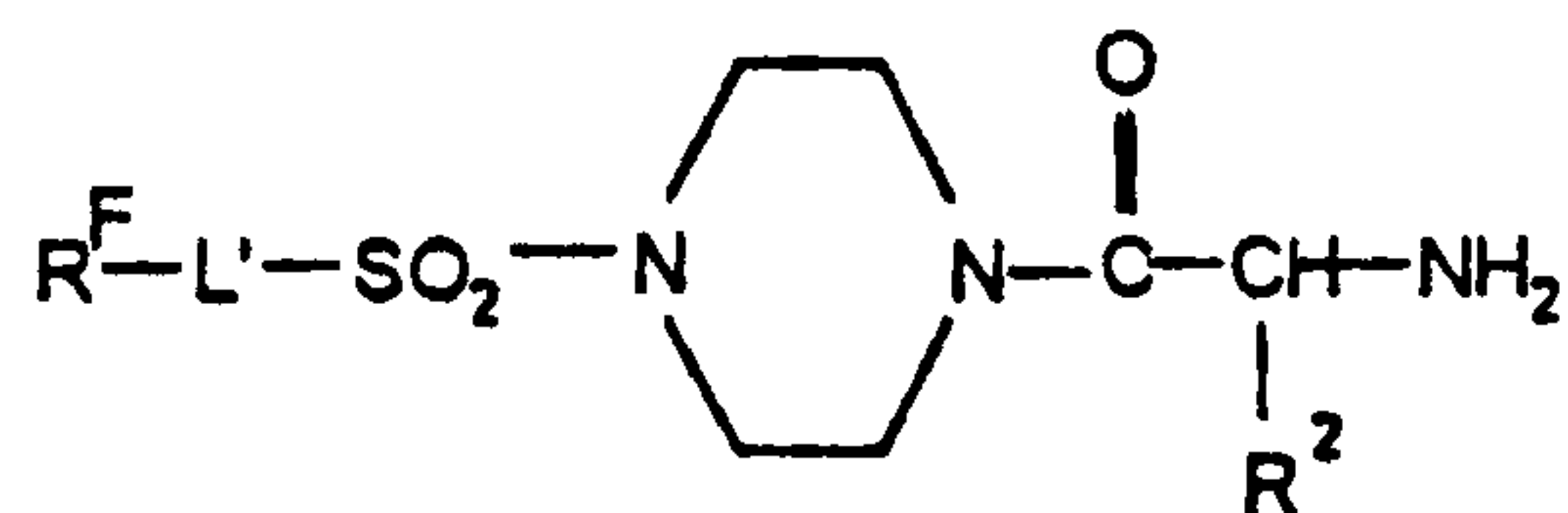
being reacted in a way known in the art with compounds

of general formula 51



in which R^{F} , L' , R^2 and R^3 have the above-mentioned meaning,

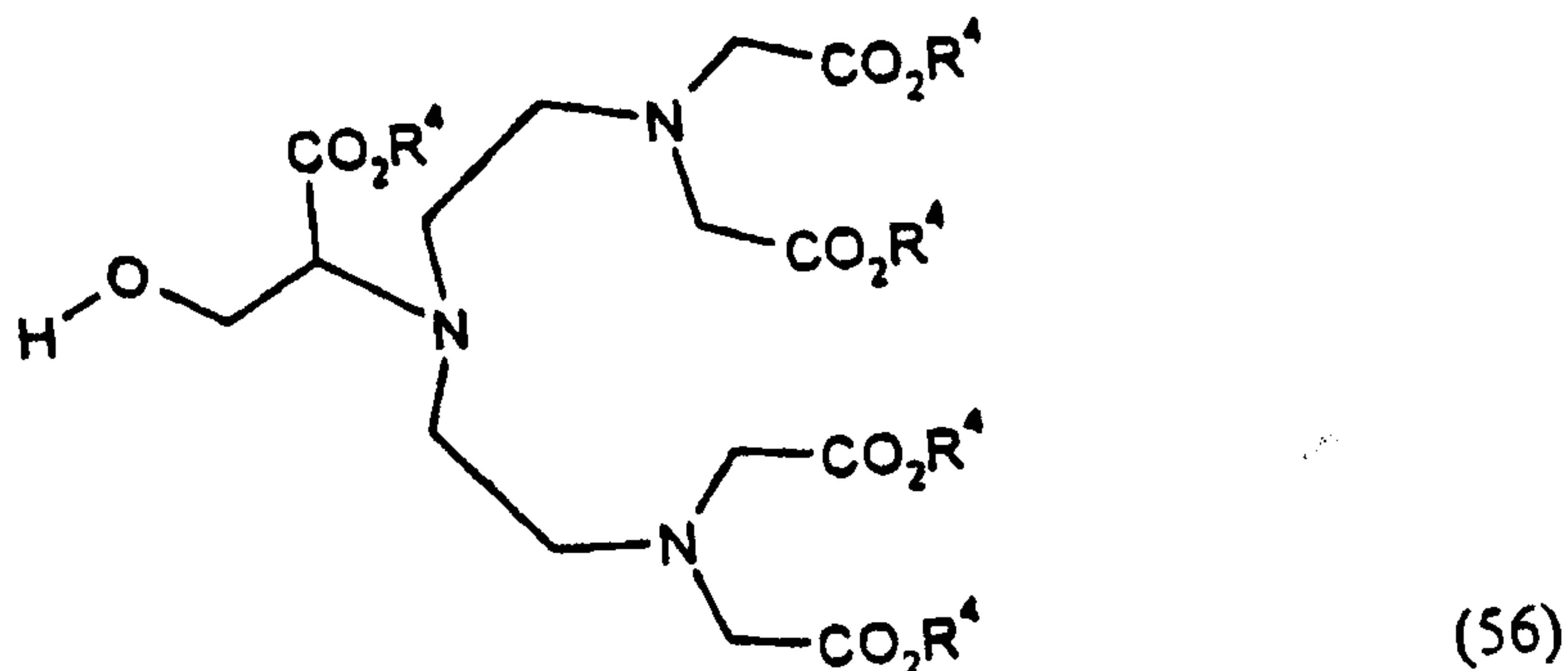
- then optionally present protective groups being cleaved, the thus obtained complexing agents being reacted with at least one metal oxide or metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature and then -- optionally -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides,
- b) compounds of general formula I with A in the meaning of general formula XIII are produced by halocarboxylic acid derivatives of general formula 52 being reacted analogously to a) with piperazine derivatives of general formula 66



in which R^f , L' and R^2 have the above-mentioned meaning,

then optionally present protective groups being cleaved and the procedure further being performed as under a).

10. Process for the production of perfluoroalkyl-containing compounds of general formula I as defined in claim 1, wherein compounds with A in the meaning of general formula IV are produced by hydroxy acids or hydroxy esters of general formula 56



in which R^4 has the above-mentioned meaning, being reacted with halogen compounds of general formula 55



in which R^f , L' and Hal have the above-mentioned meanings,

in a mixture consisting of an organic solvent and a buffer at slightly alkaline pH at room temperature for several hours, then optionally present protective groups being cleaved, the thus obtained complexing agents being reacted with at least one metal oxide or

metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature and then -- optionally -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides.

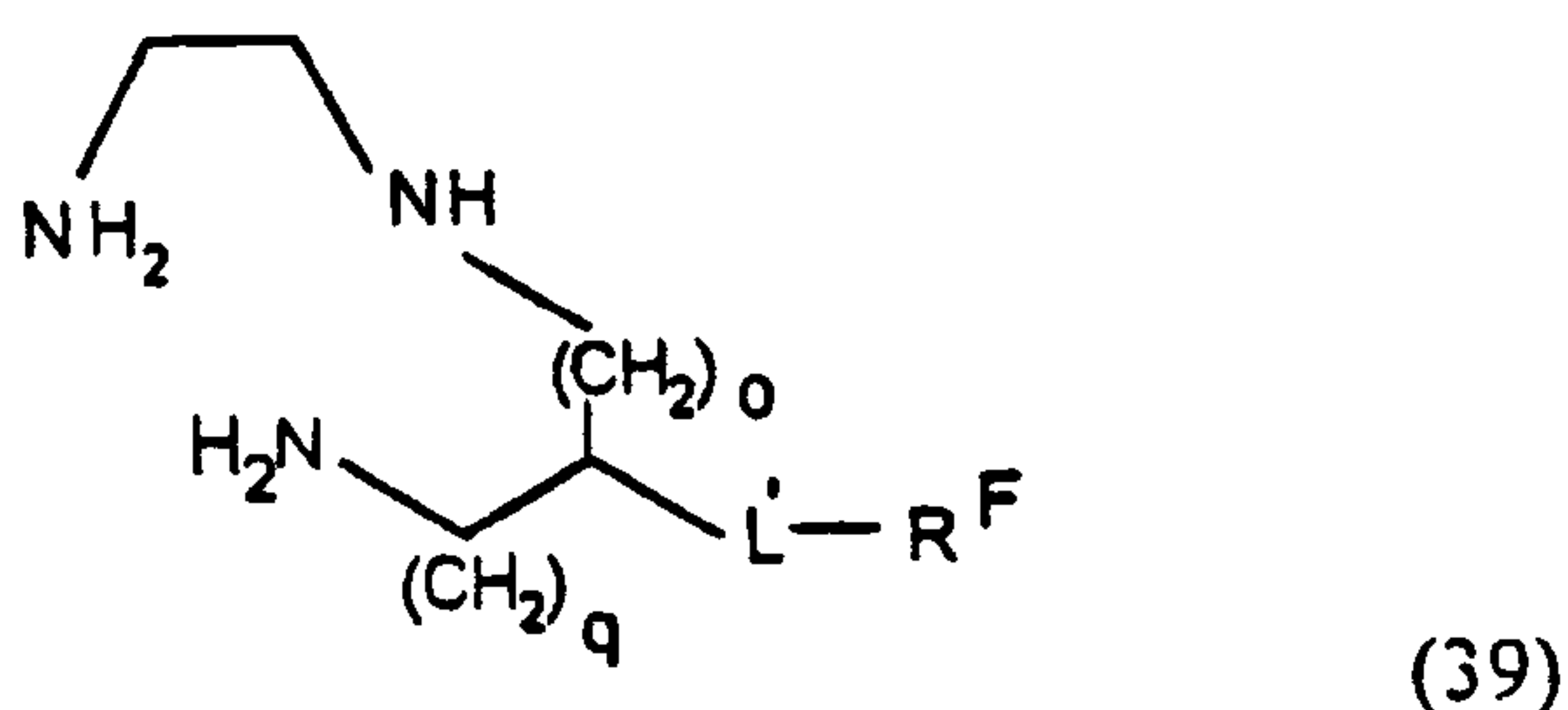
11. Process for the production of perfluoroalkyl-containing compounds of general formula I as defined in claim 1, wherein

- a) Compounds of general formula I with A in the meaning of general formula V are produced by α -halocarboxylic acid esters or α -halocarboxylic acids of general formula 18



in which Hal and R^4 have the above-mentioned meaning,

being reacted in a way known in the art with amines of general formula 39

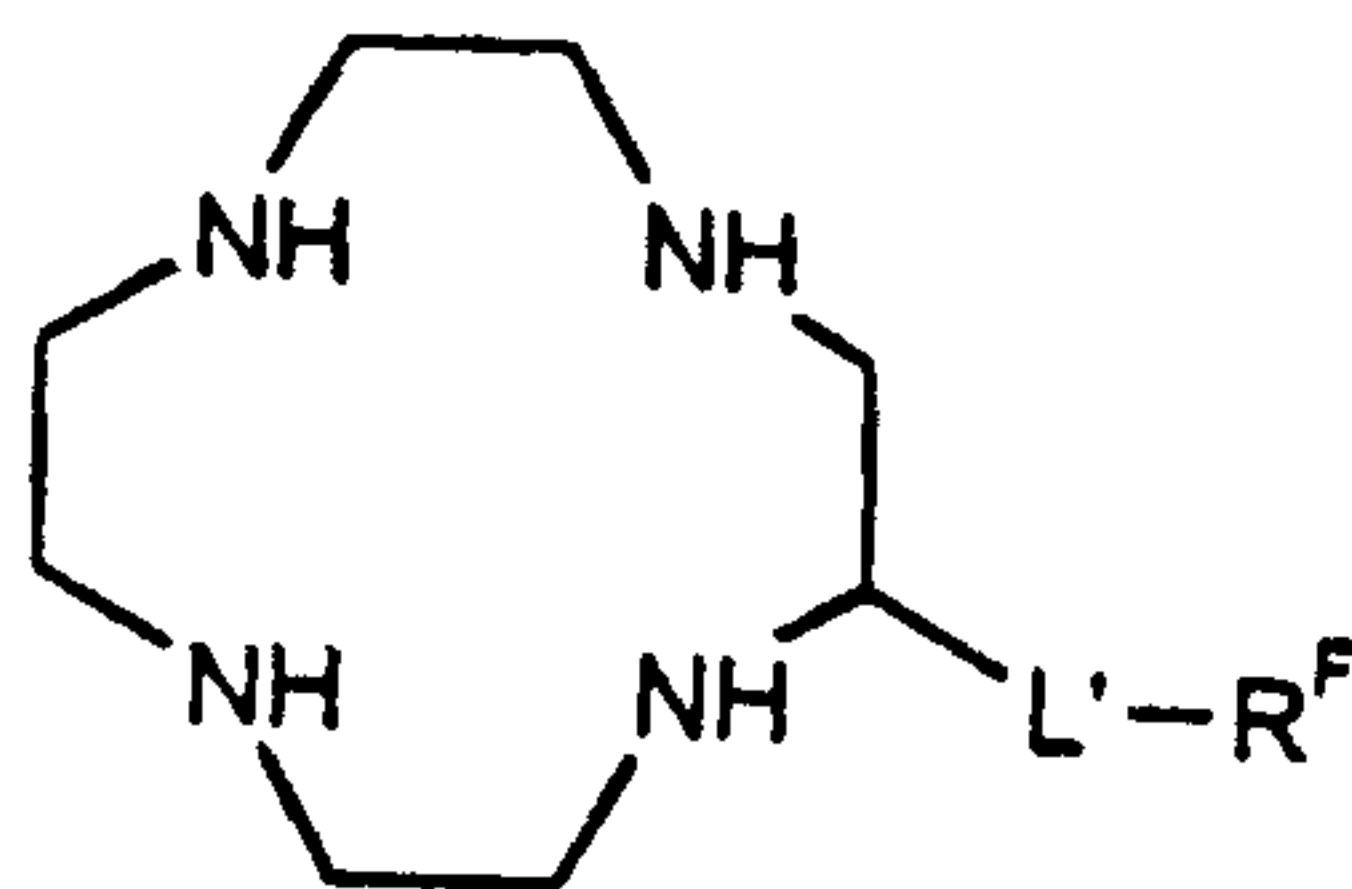


in which L' , R^F , o and q have the mentioned meanings,

then optionally present protective groups being cleaved, the thus obtained complexing agents being

reacted with at least one metal oxide or metal salt of an element of atomic numbers 21-29, 39, 42, 44 or 57-83 at room temperature or elevated temperature and then -- optionally -- acid hydrogen atoms that are present being substituted by cations of inorganic and/or organic bases, amino acids or amino acid amides,

- b) compounds of general formula I with A in the meaning of general formula VI are produced by α -halocarboxylic acid esters or α -halocarboxylic acids of general formula 18 being reacted in a way known in the art with compounds of general formula 36



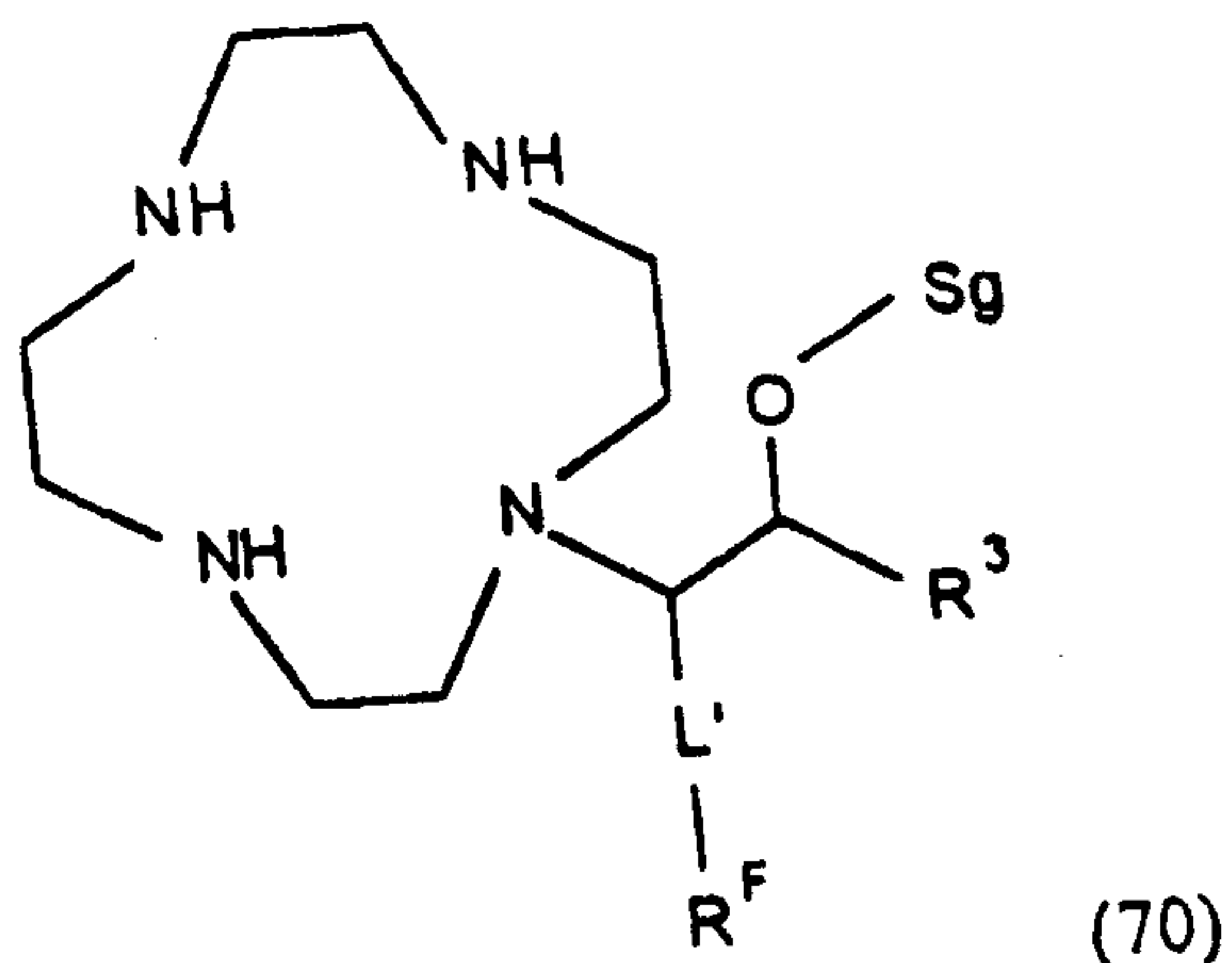
(36)

in which L' and R^F have the above-mentioned meanings,

and the procedure further is performed as under a),

- c) compounds of general formula I with A in the meaning of general formula X are produced by α -halocarboxylic acid esters or α -halocarboxylic acids of general formula 18 being reacted in a way known in the art with compounds

of general formula 70



in which R^f , L' , R^3 have the above-mentioned meaning, and Sg is in the meaning of a protective group,

and the procedure further is performed as under a).

12. Pharmaceutical agents that contain at least one physiologically compatible compound according to claim 1, optionally with the additives that are commonly used in galenicals.

13. Use of at least one physiologically compatible compound according to claim 1 or a pharmaceutical agent according to claim 12 as a contrast medium in 1H -NMR diagnosis and 1H -NMR spectroscopy.

14. Use of at least one physiologically compatible compound according to claim 1 or a pharmaceutical agent according to claim 12 as a contrast medium in diagnostic radiology.

15. Use of at least one physiologically compatible compound according to claim 1 or a pharmaceutical agent according to claim 12 as a pharmaceutical agent for radiodignosis and radiotherapy.

16. Use according to claim 13 or 14 as blood-pool agents.

17. Use according to claim 13 or 14 as lymphographic agents.

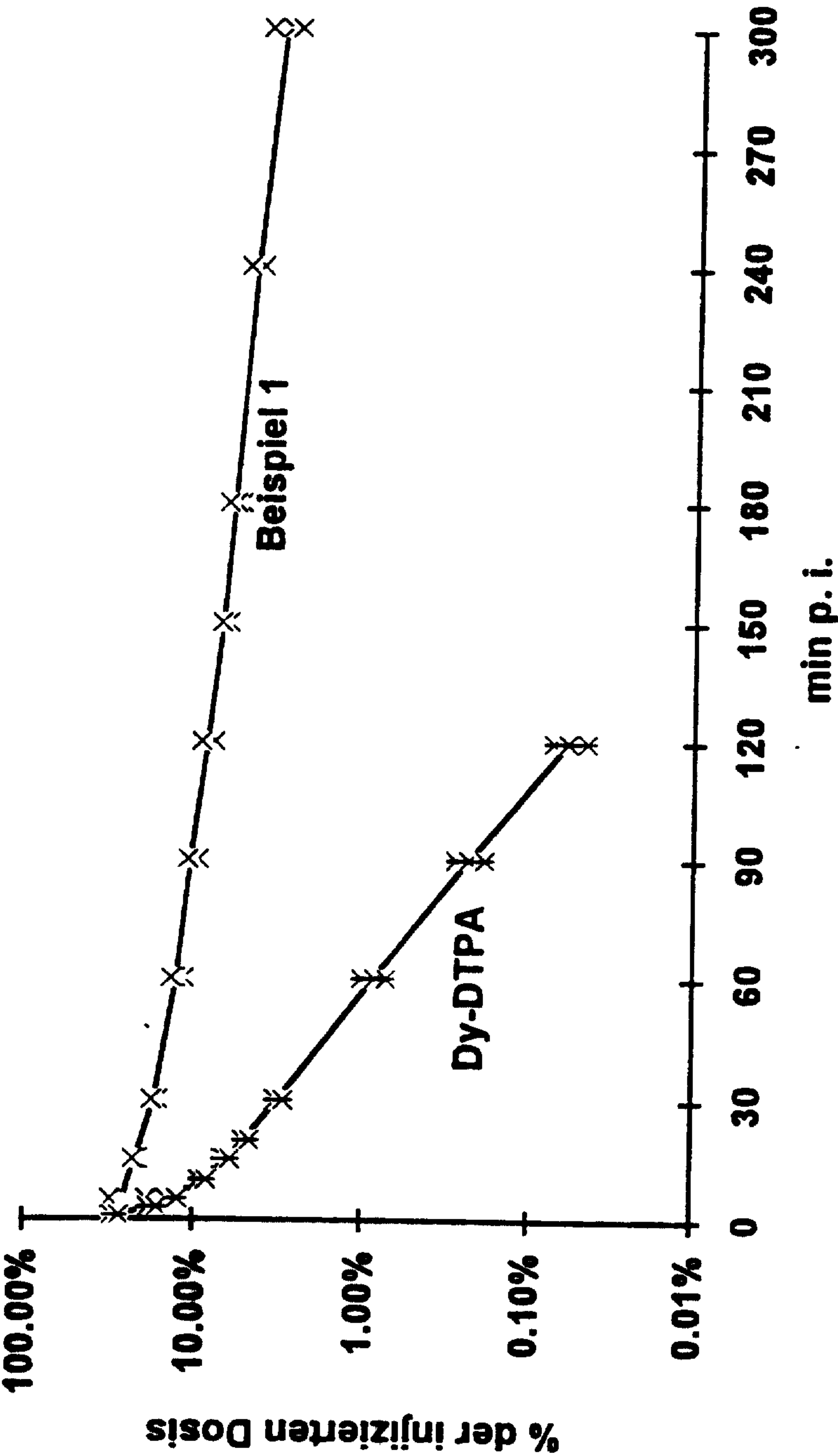


Abbildung 1

Bild 1: MRT-Lymphkonotendarstellung nach interstitieller Gabe des Kontrastmittels

